



## Analysis of inorganic chlorine chemistry in the midlatitude summer stratosphere using aircraft and satellite observations

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**Abstract.** In recent years, significant perturbations to stratospheric inorganic chlorine have been observed following  
25 wildfires or volcanic eruptions, as well as modeled in simulations with elevated water vapor from overshooting  
convection or organic species in biomass burning aerosol. A detailed evaluation of inorganic chlorine in the  
stratosphere sampled under various conditions is presented here using high precision in situ aircraft measurements of  
chlorine monoxide (ClO) and chlorine nitrate (ClONO<sub>2</sub>). Data were obtained over North America during the NASA  
Dynamics and Chemistry of the Summer Stratosphere (DCOTSS) mission. The vertical distributions of ClO and  
30 ClONO<sub>2</sub> from all 29 research flights of the 2-year mission are found to be relatively compact and lacking substantial  
outliers. The peak mixing ratios are approximately 30 ppt and 400 ppt for ClO and ClONO<sub>2</sub>, respectively. No chlorine  
activation was observed in the lower stratosphere in the presence of low temperatures and elevated water vapor from  
convective injection. Steady-state calculated ClONO<sub>2</sub> is found to be in good agreement with measured values,  
suggesting that a reduction in the uncertainty of the recommended rate coefficient for ClONO<sub>2</sub> production is possible.  
35 HCl is calculated at high time resolution throughout the mission using satellite data, and the resulting evaluation of  
the inorganic chlorine budget shows excellent agreement, with the ratio of ClO + ClONO<sub>2</sub> + HCl to total inorganic  
chlorine equal to 0.95 for mixing ratios greater than 500 ppt. No evidence of inorganic chlorine activation was  
observed during DCOTSS when aerosol organic mass fraction and biomass burning fraction were elevated.

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## 1 Introduction

Extraordinary perturbations to Earth's stratospheric composition have occurred in just the past few years. These  
45 significant and unforeseen changes to stratospheric constituents have in some cases exposed gaps in our understanding  
and highlighted the need for ongoing study of stratospheric chemical and dynamical processes (Salawitch et al., 2025).  
For example, the Australian New Year's (ANY) fires in 2019-2020 led to substantial changes in inorganic chlorine  
concentrations in the midlatitude stratosphere (Santee et al., 2022; Strahan et al., 2022; Salawitch and McBride, 2022;  
Solomon et al., 2023; Chipperfield and Bekki, 2024), with previously unconsidered heterogeneous chemistry on  
50 organic aerosols appearing to be the likely cause (Solomon et al., 2023; Stone et al., 2025). The Hunga volcanic  
eruption in 2022 caused extraordinary enhancements in stratospheric water vapor with more modest increases in  
sulfate and led to large changes in inorganic chlorine partitioning, ozone, and other compounds in the stratosphere  
(Millán et al., 2022; Santee et al., 2023; Wilmouth et al., 2023). In 2023, a serendipitous discovery was made that  
metals vaporized upon spacecraft reentry to Earth are now observable in sulfuric acid particles in the stratosphere  
55 (Murphy et al., 2023). Given the rapid growth in the space industry, there will likely be orders of magnitude increases  
in space debris in the stratosphere, with unknown potential impacts on stratospheric chemistry (Murphy et al., 2023;  
Revell et al., 2025).

Against this backdrop of ongoing change, it is important to evaluate our understanding of stratospheric chlorine  
chemistry, as chemical cycles involving inorganic chlorine exert significant control on Earth's protective ozone layer.  
60 The recent NASA Earth Venture Suborbital (EVS-3) campaign—Dynamics and Chemistry of the Summer  
Stratosphere (DCOTSS)—provided this opportunity, with 29 research flights in summers 2021 and 2022 over North  
America (Bowman et al., 2026). The Harvard Halogens (HAL) instrument made highly sensitive in situ measurements  
of chlorine monoxide (ClO) and chlorine nitrate (ClONO<sub>2</sub>) from the ER-2 aircraft throughout the campaign. In  
addition to examining inorganic chlorine chemistry in the background stratosphere, the DCOTSS mission also  
65 provided the first opportunity to test, using in situ measurements, the possibility of heterogeneous chlorine activation  
and subsequent ozone loss occurring over North America in summertime. As shown by several modeling studies  
(Anderson et al., 2012; Anderson et al., 2017; Robrecht et al., 2019), this chemistry could become significant under  
conditions of low temperature and elevated water vapor found in the lower stratosphere following tropopause-  
overshooting convective injection, which is frequent in the North American Monsoon Anticyclone (NAMA) region.

70 For the past two decades, the atmospheric sciences community has benefited immeasurably from continuous daily  
satellite measurements of the inorganic halogens ClO and HCl (hydrogen chloride) from the Microwave Limb Sounder  
(MLS) instrument on Aura, as well as the Atmospheric Chemistry Experiment—Fourier Transform Spectrometer  
(ACE-FTS) satellite instrument on SCISAT-1, which also measures ClONO<sub>2</sub>. The in situ chlorine measurements from  
HAL during DCOTSS complement these satellite data and also provide an improved capability, in combination with  
75 other concurrent aircraft measurements, to test our understanding of chlorine photochemistry and kinetics due to the  
superior precision of the in situ data and targeted aircraft sampling. The DCOTSS mission addresses a surprising void  
in the in situ record in which the chemistry of the stratosphere over the contiguous United States in summertime has  
been relatively undersampled.



In this study, we analyze the stratospheric in situ chemical data from the DCOTSS mission to establish the dominant factor controlling inorganic chlorine mixing ratios, to explore the possibility of heterogeneous chlorine activation in stratospheric air influenced by tropopause-overshooting convection, and to quantitatively test the established photochemistry and kinetics of ClONO<sub>2</sub> formation and loss. We further calculate HCl for the time and location of aircraft sampling using a technique that combines satellite and in situ data. This calculated HCl is used with measured ClO, measured ClONO<sub>2</sub>, and calculated total inorganic chlorine to test closure of the inorganic chlorine budget in the stratosphere. In situ data are compared with relevant satellite data throughout the analysis.

Anthropogenic climate disruption has resulted in an unprecedented increase in wildfire frequency and intensity globally, with pyroconvection from these severe wildfires potentially injecting biomass burning products into the lower stratosphere, increasing aerosol loading and altering stratospheric composition (Ellis et al., 2022; Fromm et al., 2022; Ansmann et al., 2022). In this study, we look for observational evidence from DCOTSS of the heterogeneous chlorine chemistry mechanism that was identified by Solomon et al. (2023) as being responsible for elevated ClO and ClONO<sub>2</sub> mixing ratios following the ANY fires, in which the presence of enhanced organic aerosol led to heterogeneous uptake of HCl at relatively warm temperatures.

## 2 Experimental

All in situ data analyzed in this study were obtained from instruments flown onboard the high-altitude ER-2 aircraft during the NASA DCOTSS mission based out of Salina, KS and Palmdale, CA. There were 31 total flights in the mission, with 29 of those representing science research flights, which took place in June–August 2021 and May–July 2022. The operational base of Salina, KS in summertime was chosen to optimize sampling of strong convective storms in the NAMA region that regularly penetrate deep into the lower stratosphere (e.g., Anderson et al., 2012; Smith et al., 2017; Cooney et al., 2018; Liu et al., 2020; Homeyer and Bowman, 2021; Homeyer et al., 2023). The primary goals of the DCOTSS project were to improve understanding of the effects of deep convection and large-scale dynamics on the composition and chemistry of the lower stratosphere over North America and to extensively sample the background stratosphere. A detailed overview of the DCOTSS mission, including a description of each ER-2 flight and the instrumentation onboard, was presented by Bowman et al. (2026).

### 2.1 Inorganic Chlorine Measurements

The HAL instrument uses thermal dissociation / atomic resonance fluorescence detection to make highly sensitive measurements of inorganic halogens at four independent axes. The technique used for measurement of ClO involves chemical titration via a rapid bimolecular reaction with NO, followed by resonance fluorescence detection of the resulting Cl atoms in the vacuum ultraviolet (Anderson et al., 1977; Brune et al., 1989; Anderson et al., 1991; Stimpfle et al., 2004; Wilmouth et al., 2009). Measurement of ClONO<sub>2</sub> requires an initial step of rapid thermal dissociation to break the weak ClO–NO<sub>2</sub> bond, followed by detection of the dissociated ClO in the same manner as ambient ClO (Stimpfle et al., 1999; Stimpfle et al., 2004; Wilmouth et al., 2006). Each independent detection axis contains a custom-built resonance fluorescence lamp (Hare et al., 2024), optimized for Cl detection at 118.9 nm, and a photomultiplier



115 tube detector positioned perpendicular to the lamp. A description of the Harvard Halogens flight instrument, along with a schematic diagram and photo, are included in Fig. S1 and in the accompanying Supplement text.

The instrument was calibrated in the laboratory under pressures and flow velocities typical of flight. The absolute sensitivity of each detection axis in flight was determined at systematic intervals from the observed Rayleigh scatter as a function of air density. The estimated accuracy of the ClO measurement is 17% (all uncertainties reported herein are  $1\sigma$ ) with a precision of 2 ppt (parts per trillion; all reported mixing ratios herein are by volume) in a sampling interval of 35 seconds. The data rate of 35 seconds is set by the timing of the NO titration cycle. The estimated accuracy of the ClONO<sub>2</sub> measurement is 20% with a precision of 5 ppt in 35 seconds, with data only being reported when the dissociation heater cycle is at the highest temperatures (~500 K). The detection limit of the HAL instrument with additional data averaging was very low during DCOTSS, with discrimination of approximately 0.2 ppt for ClO and 2 ppt for ClONO<sub>2</sub>. The measurement precision achieved during the DCOTSS mission for both ClO and ClONO<sub>2</sub> was approximately a factor of two higher than previous flight values (Stimpfle et al., 2004) due to improvements in the optical system. The excellent agreement between the ClO data from multiple independent HAL detection axes provides additional confidence in the accuracy of the measurements.

## 130 2.2 Other In Situ Data

The ER-2 carried a payload of 12 instruments to measure trace gases, aerosol properties, and meteorology during the DCOTSS flights (Bowman et al., 2026). A complete list of the in situ measurements with specified precision and accuracy appears in Table 1 of Howar et al. (2025). The trace gas measurements used in the present study include: ozone (O<sub>3</sub>) from the Rapid Ozone Experiment (ROZE), reported with a precision of 1 ppb and accuracy of 6% (Hannun et al., 2020); nitrogen dioxide (NO<sub>2</sub>) from the Compact Airborne Nitrogen dioxide Experiment (CANOE), reported with a precision of 50 ppt and accuracy of 10% (St. Clair et al., 2019); and water vapor (H<sub>2</sub>O) from the Harvard Herriott Hygrometer (HHH) as part of the Harvard Water Vapor (HWV) instrument, reported with a precision of 0.1 ppm and accuracy of 5% (Sargent et al., 2013). Temperature (T), pressure (P), and potential temperature ( $\theta$ ) were measured by the Meteorological Measurement System (MMS) with accuracies of 0.3 K, 0.3 hPa, and 0.5–1.5 K, respectively (Scott et al., 1990).

Aerosol fractional composition products were reported by the Particle Analysis by Laser Mass Spectrometry-Next Generation (PALMS-NG) instrument as a value from 0 to 1, with approximate uncertainty of 1–10%. PALMS-NG detects and obtains size information on single particles via light scattering and then determines chemical composition using a one-step Laser Desorption and Ionization technique combined with dual time-of-flight mass spectrometers (Jacquot et al., 2024). This Single Particle Mass Spectrometry technique allows for measurement of refractory and non-refractory particle components over a size range from ~0.1 to 10 micrometer diameter.

Three values that were not measured are calculated for all flights of the mission: (a) HCl was determined using a combination of satellite and in situ data, as discussed in detail in Section 3.5. (b) Total inorganic chlorine (Cl<sub>y</sub>) was determined by Howar et al. (2025) as the difference between organic chlorine measured in the stratosphere during DCOTSS by the Advanced Whole Air Sampler (Smith et al., 2024) and the total organic chlorine burden in the troposphere at the time of stratospheric entry. We use the average of the Howar et al. (2025) upper and lower limit



estimates of  $Cl_y$ , which include both long-lived and Very Short-Lived (VSL) chlorine species. This analysis provides total inorganic chlorine with an approximate uncertainty of 6% when  $Cl_y > 500$  ppt, with larger uncertainties at smaller mixing ratios, as discussed in Section 3.5. (c) Photolysis rates for  $ClONO_2$  ( $J_{ClONO_2}$ ) were calculated using a radiative transfer model that accounts for Rayleigh scattering (Salawitch et al., 1994) constrained by surface reflectivity from the TROPospheric Monitoring Instrument (TROPOMI) and vertical profiles of  $O_3$  derived from a combination of ozonesonde and MLS retrievals with total column measurements from TROPOMI, as described by Howar et al. (2025). The absorption cross sections for  $ClONO_2$  are taken from the NASA JPL-19 recommendation (Burkholder et al., 2019). The estimated uncertainty in  $J_{ClONO_2}$  values used here is 10%.

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### 2.3 Satellite Data

MLS measurements of HCl,  $O_3$ , and ClO are used in this study, as are ACE-FTS (hereafter, ACE) measurements of  $ClONO_2$  and ClO. MLS measures microwave thermal emission from the limb of Earth's atmosphere to obtain vertical profiles of trace gases at various pressure levels with daily global coverage. MLS Version 5, Level 2 data are used. All data screening recommendations from the MLS Version 5 Data Quality document (Livesey et al., 2022) were applied to each variable. For HCl, the individual-profile precision / accuracy are estimated to be 300 ppt / 100 ppt, 200 ppt / 100 ppt, and 200 ppt / 100 ppt at 100, 68, and 46 hPa, respectively, and the vertical resolution is 3 km over that range. For  $O_3$ , the individual-profile precision / accuracy are estimated to be 40 ppb / (+2.5 ppb + 4%), 40 ppb / 50 ppb, and 60 ppb / 100 ppb at 100, 68, and 46 hPa, respectively, and the vertical resolution is 2.5 km over that range. For ClO, the individual-profile precision and vertical resolution are estimated to be 100 ppt and 3 km, respectively, over the 100 to 46 hPa pressure range. The ClO measurement has known biases at the 100 and 68 hPa retrieval pressure levels, and the appropriate MLS-recommended bias corrections (Livesey et al., 2022) were implemented.

ACE measures  $ClONO_2$  and ClO using Fourier Transform infrared spectroscopy at sunrise and sunset. ACE has relatively sparse observational coverage outside of the high latitudes, and ClO is a very weak absorber, which makes it difficult to measure when ClO concentrations are not elevated as in chlorine processing events in polar spring. To our knowledge, pressure-dependent precision and accuracy data comparable to those listed for MLS measurements are not available for ACE measurements. The version 5.2 data used here were filtered for outliers beyond 3 standard deviations from the mean in accordance with the ACE team recommendations, where appropriate (Boone et al., 2023).

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## 3 Results and Discussion

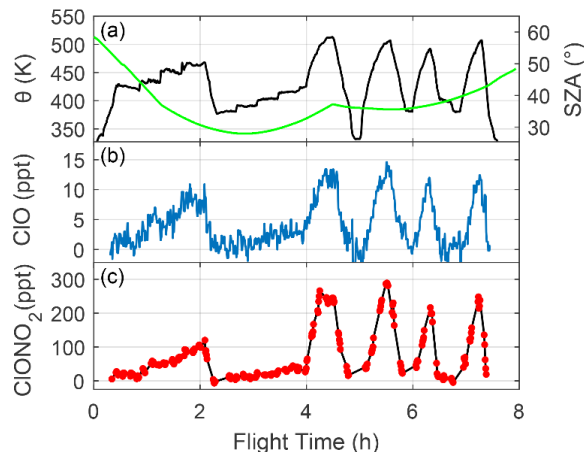
### 3.1 Inorganic Chlorine

The DCOTSS mission extensively sampled the stratosphere over much of the contiguous United States, as well as north into Canada and over the Pacific Ocean off the west coasts of the US and Mexico. Example flight data are shown for the DCOTSS flight of 16 July 2021 in Fig. 1. The primary objective of this flight was to survey the Northern Hemisphere midlatitude summer stratosphere. On this 8-hour flight, the ER-2 took off at 14:13 UT (9:13 AM CDT) from Salina, KS, flew southeast to southern IL, then turned north, flying up to Hudson Bay, Canada, then back to Salina, KS. Potential temperature and solar zenith angle (SZA) for the flight are shown in Fig. 1a, with mixing ratios

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190 of ClO and ClONO<sub>2</sub> as a function of flight time plotted in Figs. 1b and 1c, respectively. The vertical coordinate is shown as potential temperature rather than altitude because, in the absence of diabatic processes, air parcels in the stratosphere are transported along surfaces of constant potential temperature. There were many level flight legs in the first half of the flight, of approximately 20-minute duration, followed by several extensive vertical profiles in the second half. The peak altitudes reached by the ER-2 represented by the maximum potential temperatures shown in Fig. 1 were approximately 20.5 – 21.1 km above mean sea level. Other than takeoff, landing, and the deepest descents, 195 the flight was dominated by stratospheric air, evidenced by ozone mixing ratios exceeding 200 ppb with water mixing ratios below 6 ppm (not shown).



**Figure 1.** In situ measurements from the DCOTSS flight on 16 July 2021 over North America on the NASA ER-2. (a) Potential temperature (black) as measured by MMS and solar zenith angle (green). Mixing ratios of (b) ClO and (c) ClONO<sub>2</sub> as measured by HAL in units of parts per trillion by volume. The black line in panel (c) connects the measured points to guide the eye.

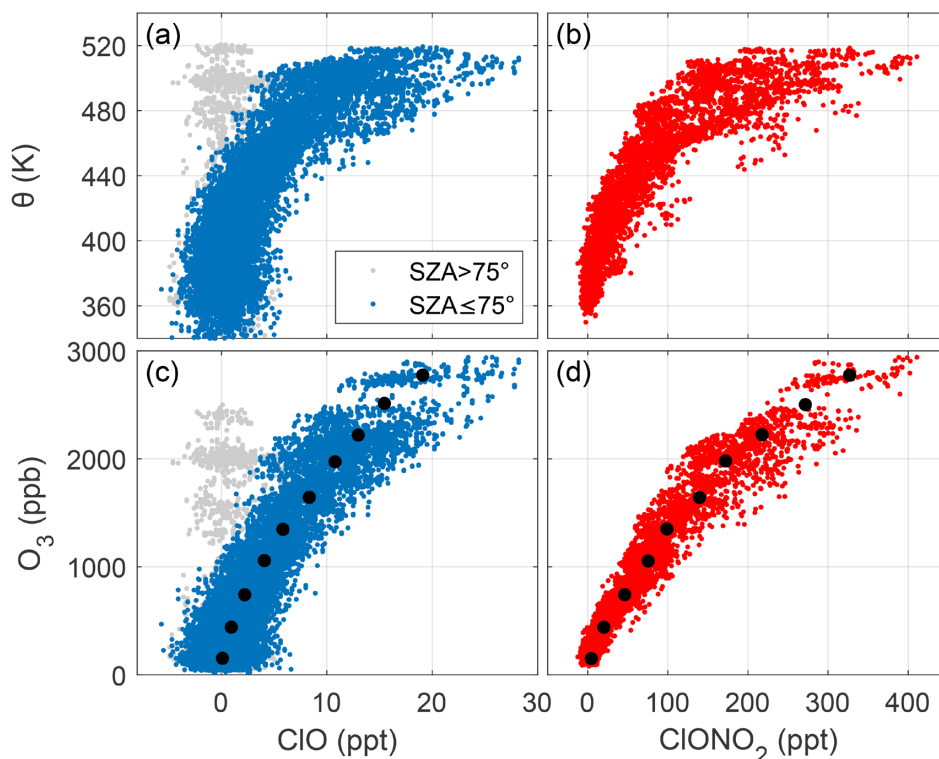
200 Strong correlations between the inorganic chlorine species and potential temperature are evident in Fig. 1. Mixing ratios of ClO and ClONO<sub>2</sub> slowly increase as  $\theta$  gradually rises across the level flight legs in the first half of the flight, then mixing ratios more rapidly increase and decrease in concert with the vertical profiles in  $\theta$  during the second half of the flight. The peak ClO and ClONO<sub>2</sub> mixing ratios observed on this flight were approximately 15 ppt and 300 ppt, respectively. The red points in Fig. 1c represent the ClONO<sub>2</sub> measurements, which as indicated earlier, are less frequent than the other observations shown due to the required synchronization of the HAL NO titration cycle and dissociation heater cycle. As shown by the solar zenith angle in Fig. 1a, this entire survey flight was during daylight hours, so the vertical structure in ClO and ClONO<sub>2</sub> mixing ratios was unaffected by nighttime impacts to photochemistry.

205 To extend this analysis across the entire mission, Fig. 2 shows potential temperature versus mixing ratio of ClO and ClONO<sub>2</sub> in panels 2a and 2b, respectively, for all 29 DCOTSS research flights. The vertical axis includes measurements obtained in the upper troposphere up through and including all stratospheric measurements from the



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mission, reaching potential temperatures up to approximately 520 K. The highest sampling altitude achieved by the ER-2 during the DCOTSS mission was ~21.3 km. Supplement Fig. S2 provides a useful reference to aid in comparison of DCOTSS data over a variety of vertical coordinates, displaying pressure, potential temperature, and ozone versus altitude. Because ClO decreases to zero at night, we separate daytime and nighttime data, with the ClO mixing ratios color coded in Fig. 2 based on the solar zenith angle at the time of measurement. Specifically, a threshold of 75° and below is used to represent daytime ClO in Fig. 2a and in subsequent figures of this work (although the exact threshold SZA used is not critical). Approximately 91% of the ClO data as represented in Fig. 2 are categorized as daytime values.



**Figure 2.** Measurements of ClO and ClONO<sub>2</sub> as a function of (a), (b) potential temperature and (c), (d) ozone, respectively, for all 29 research flights of the NASA DCOTSS mission. ClO data are color-coded based on the solar zenith angle at the time of measurement: blue indicates SZA at or below 75° (daytime), and gray indicates SZA greater than 75°. All ClONO<sub>2</sub> data from the mission are in red. The larger black circles in (c) and (d) are averages of ClO and ClONO<sub>2</sub>, respectively, in 300 ppb O<sub>3</sub> bins. The number of data points plotted is 16369 for ClO (14845 in blue, 1524 in gray) and 5129 for ClONO<sub>2</sub>.

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Mixing ratios of ClO and ClONO<sub>2</sub> increase with increasing potential temperature (Figs. 2a and 2b), rising more slowly in the lowest portion of the stratosphere and subsequently increasing more rapidly and generally with greater variability at  $\theta$  values above ~460 K. The observed correlation between inorganic chlorine mixing ratios and potential temperature arises because both generally increase with age of air in the midlatitude stratosphere. Nearly all chlorine

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enters the stratosphere in organic form in the tropics in the rising part of the Brewer-Dobson circulation and is slowly converted to inorganic form as the air parcel ages during its transit through the stratosphere. Higher values of inorganic chlorine at a given potential temperature in DCOTSS indicate a larger fraction of air that has descended from higher altitudes or been transported from higher latitudes, where air has been more photochemically processed. The exact correlations between potential temperature and ClO and ClONO<sub>2</sub> are not universal in the atmosphere, e.g., the vertical gradient at midlatitudes as shown in Figs. 2a and 2b is gradual relative to the sharper increase that would be observed inside the wintertime polar vortex.

Figures 2c and 2d show ClO and ClONO<sub>2</sub> mixing ratios, respectively, now plotted with O<sub>3</sub> as the vertical coordinate. The blue, gray, and red points represent the same mixing ratios as in the upper panels, while the larger black points represent averages of ClO and ClONO<sub>2</sub> mixing ratios in 300 ppb O<sub>3</sub> bins. Relatively compact correlations between ozone as a vertical coordinate and daytime ClO and ClONO<sub>2</sub> mixing ratios are observed. This relation occurs because inorganic chlorine and ozone both have a strong positive correlation with age of air (outside the polar vortex). The formation of both inorganic chlorine and ozone involves processes that require photolysis of precursor molecules by solar UV radiation, with air at progressively higher altitudes and more polar latitudes on average having experienced greater cumulative UV exposure.

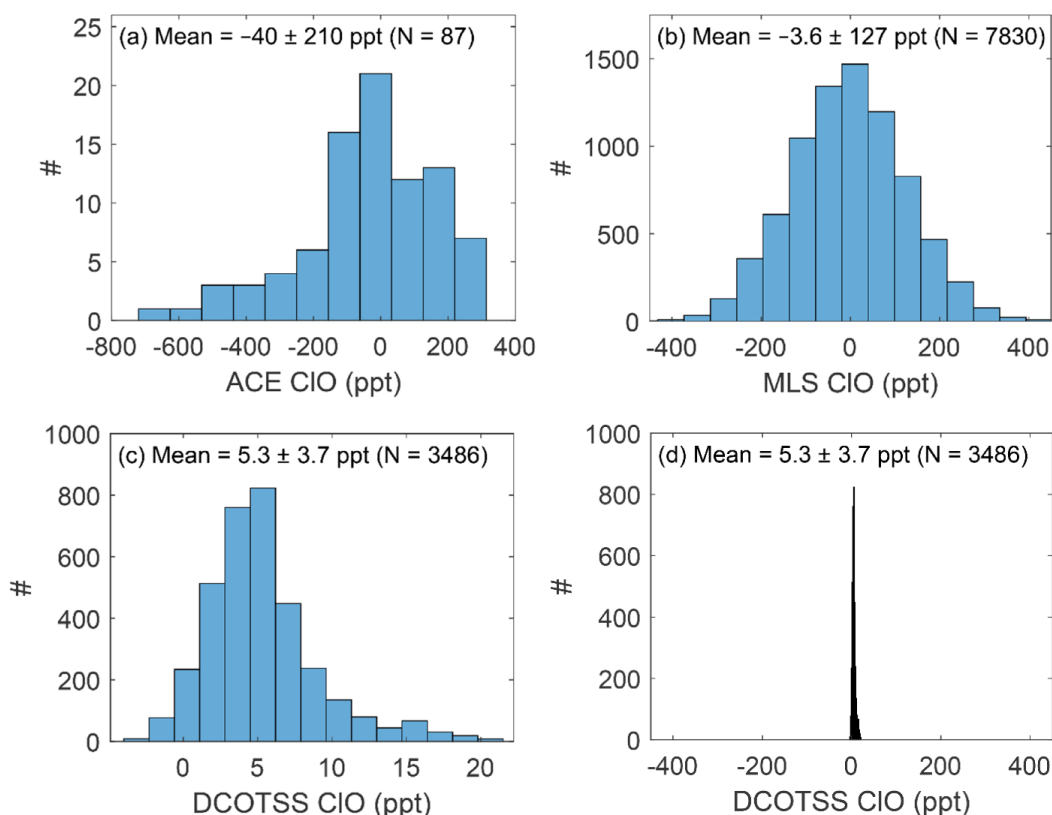
The peak mixing ratios observed during the DCOTSS mission were just under 30 ppt for ClO and just over 400 ppt for ClONO<sub>2</sub>. Peak values for both species were observed in May 2022 and were unrelated to convection. The high values of ClONO<sub>2</sub> that lie to the right of the primary envelope of points in Fig. 2b (at  $\theta \approx 410 - 430$  K and  $440 - 490$  K) are similarly from three flights in May 2022 and were also unrelated to convection. These points only appear as outliers when plotted versus potential temperature; they are unremarkable when plotted against O<sub>3</sub> in Fig. 2d, suggesting that large-scale dynamics is responsible. The variability in the observed mixing ratios of ClO and ClONO<sub>2</sub> for a given O<sub>3</sub> value in Fig. 2 is due to a combination of natural atmospheric variability and HAL instrument precision. The correlations are very tight, considering that the sampling took place over two different years and four different calendar months, over which inorganic chlorine is expected to vary. Despite extraordinary changes in the composition of the stratosphere in 2022 due to the Hunga volcanic eruption, no substantial perturbations were observed during the DCOTSS mission because the impacts of the Hunga eruption at that time occurred primarily at higher altitudes and farther south (Wilmouth et al., 2023; Santee et al., 2023; Østerstrøm et al., 2025).

### 3.2 Satellite Comparison

One useful way to examine the in situ measurements of ClO and ClONO<sub>2</sub> from the DCOTSS mission is to compare the aircraft data with existing satellite measurements. Figure 3 shows histograms of ClO data at 68 hPa for summers 2021 and 2022 from the ACE satellite instrument (Fig. 3a), the MLS satellite instrument (Fig. 3b), and the HAL instrument onboard the ER-2 during DCOTSS (Fig. 3c). The ACE and MLS satellite data represent all reported measurements after implementing the recommended filtering for June–August 2021 and 2022 over latitudes and longitudes approximately representing the contiguous United States. The 68 hPa level was chosen for comparison because this is a pressure level at which MLS data are reported (with vertical resolution of  $\sim 3$  km), and 68 hPa is globally within the stratosphere, corresponding to approximately 19 km or 460 K during DCOTSS sampling. The



majority of the ACE measurements near 68 hPa are reported at 62 hPa or 73 hPa; for the purposes of this analysis, we include in Fig. 3 all reported ACE data in the 61–75 hPa pressure range. For the DCOTSS data from HAL, all in situ CIO data obtained in the 58–78 hPa pressure range are shown. The exact pressure range included is not critical to the overall comparison in Fig. 3.



**Figure 3.** CIO measurements at the approximate 68 hPa pressure level during the summers of 2021 and 2022 over the contiguous U.S. (approximated as 30–50° N latitude, 70–125° W longitude) from satellite instruments and from the ER-2 during DCOTSS. The CIO data shown are from (a) the ACE satellite instrument using data in the  $68 \pm 7$  hPa range for JJA in 2021 and 2022, (b) the MLS satellite instrument using reported data at 68 hPa for JJA in 2021 and 2022 and employing the recommended bias correction, and (c) the HAL instrument using all in situ data obtained at  $68 \pm 10$  hPa from the 29 research flights of the DCOTSS mission 2021–2022. Panel (d) shows the same DCOTSS CIO data as in (c) but with the x-axis rescaled to match that of the MLS data in (b). At the top of each panel are listed the mean CIO value, the 1-sigma standard deviation, and the number of measurements (N) obtained over the 6 months of data collection.

270 Figure 3 shows that the range of CIO observations represented on the x-axes is dramatically narrower for the in situ data in panel 3c relative to the satellite data in panels 3a and 3b. The satellite data are at times hundreds of ppt negative or positive, while the total range from lowest to highest in situ observation is only about 25 ppt. ACE has very limited sampling over the northern midlatitudes in summertime, with only 87 data points over the 6 months total represented in Fig. 3a, while MLS has extensive coverage with 7830 total data points represented in Fig. 3b. Both

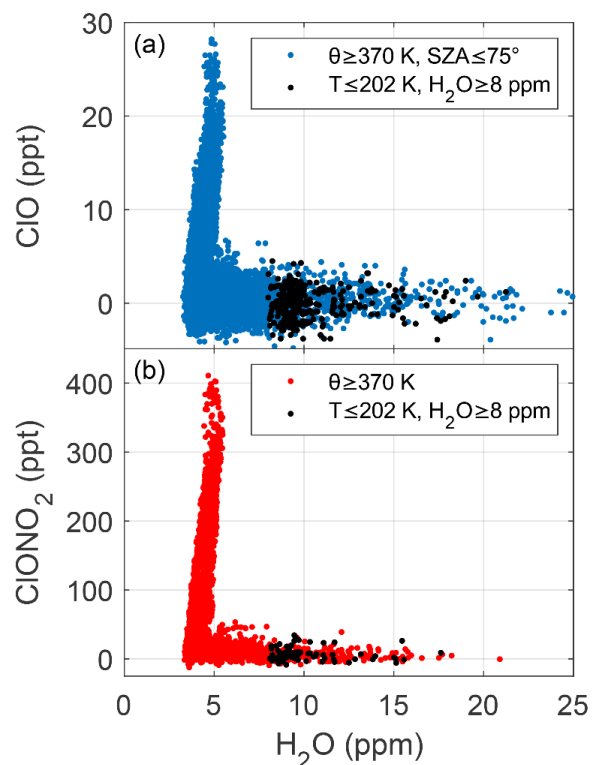


275 ACE and MLS ClO data for these 6 summer months have a negative mean and a very large standard deviation, as  
shown at the top of each figure panel. The in situ ClO data obtained by HAL during the DCOTSS mission include  
3486 total data points and have a mean value of 5.3 ppt with a standard deviation of 3.7 ppt. Much of the spread in the  
in situ data is due to real atmospheric variability at  $68 \pm 10$  hPa rather than instrumental uncertainty. To better visualize  
the precision of the in situ data relative to the satellite data, Fig. 3d replots the same DCOTSS ClO data shown in Fig.  
280 3c with the x-axis rescaled to match the x-axis of the MLS data in Fig. 3b. The precision of the in situ ClO data is  
more than an order of magnitude higher than that of the satellite data. Similar results are seen when comparing the  
ClO satellite data from ACE or MLS with the in situ ClO data from DCOTSS at higher and lower pressure levels  
relevant to ER-2 sampling (~50 – 150 hPa).

The intent of this satellite data analysis is to highlight that the in situ data obtained in DCOTSS are unique and  
285 that the excellent data quality allows stratospheric analyses that would otherwise not be possible. The limitations of  
the ClO satellite measurements are well known by the MLS and ACE data teams, and the results in Fig. 3 reflect the  
known accuracy and individual-profile precision of the data (Livesey et al., 2022; Boone et al., 2023). However, it is  
important to note that the satellite data at higher altitudes and deeper in the stratosphere, particularly those altitudes  
beyond the range of the ER-2, are significantly improved relative to the lower stratospheric ClO satellite data discussed  
290 here. Moreover, the higher mixing ratios of ClONO<sub>2</sub> relative to ClO allow the ACE measurements of ClONO<sub>2</sub> to  
become usable at a lower altitude than the ACE measurements of ClO. Supplement Fig. S3 shows a vertical profile  
comparison of ClONO<sub>2</sub> data from HAL during DCOTSS with all of the ACE satellite measurements of ClONO<sub>2</sub> for  
similar conditions as in Fig. 3. The preponderance of ClONO<sub>2</sub> data points from ACE are negative (below the detection  
limit) for pressures greater than ~80 hPa, but at lower pressures (higher altitudes, ~18 km and above), the ACE and  
295 DCOTSS data show good agreement. In the subsequent sections, we take advantage of the excellent precision and  
accuracy of the HAL data, using the in situ inorganic chlorine data from DCOTSS as the centerpiece of several  
analyses to examine our understanding of lower stratospheric chlorine chemistry.

### 3.3 Convection

300 One of the questions the DCOTSS mission sought to answer is whether perturbed inorganic chlorine mixing ratios  
would be observed in the stratosphere in the presence of elevated stratospheric water vapor following tropopause-  
overshooting convection in the NAMA. Non-catalytic reservoir species (HCl and ClONO<sub>2</sub>) typically comprise more  
than 95% of total inorganic chlorine in the midlatitude lower stratosphere. However, enhanced water vapor in  
combination with cold temperatures in the NAMA can potentially enable repartitioning of Cl<sub>y</sub> via heterogeneous  
305 reactions on ubiquitous binary sulfate-water aerosols (Anderson et al., 2012; Anderson et al., 2017; Robrecht et al.,  
2019). Under these conditions, HCl mixing ratios would be expected to decrease, while ClONO<sub>2</sub> mixing ratios would  
initially increase before later falling (Anderson et al., 2012). If elevated water vapor and low temperatures persisted  
long enough such that NO<sub>x</sub> (NO + NO<sub>2</sub>) was removed, elevated ClO would result, which in turn would lead to  
enhanced O<sub>3</sub> loss rates (Anderson et al., 2017). Two prior instances where ClO was measured in the lowermost  
310 stratosphere under conditions of elevated water vapor and aerosol surface area both showed substantially elevated ClO  
(Keim et al., 1996; Thornton et al., 2007).



**Figure 4.** Stratospheric measurements of CIO and ClONO<sub>2</sub> relative to measured water vapor for all 29 research flights of the NASA DCOTSS mission. (a) CIO data are shown in blue for potential temperature  $\geq 370$  K and solar zenith angle  $\leq 75^\circ$ , and (b) ClONO<sub>2</sub> data are shown in red for potential temperature  $\geq 370$  K. The black points in (a) and (b) represent the subset of data in each panel where observed temperature was  $\leq 202$  K and observed H<sub>2</sub>O was  $\geq 8$  ppm.

315 Elevated water vapor in the stratosphere due to convective injection was targeted and frequently sampled during the DCOTSS mission (Homeyer et al., 2023; Homeyer et al., 2024; Bowman et al., 2026). Indeed, significantly more sampling of convectively influenced stratospheric air was achieved in this field campaign than in all previous in situ field campaigns combined (~28 hours; Homeyer et al., 2023). The relationship between water vapor up to 25 ppm and inorganic chlorine species as sampled over the entire DCOTSS mission is shown in Fig. 4. The data are filtered to focus on stratospheric measurements ( $\theta \geq 370$  K), and the CIO data are additionally filtered to only include daytime values ( $SZA \leq 75^\circ$ ). Both CIO (Fig. 4a) and ClONO<sub>2</sub> (Fig. 4b) are observed to have a similar L-shaped relationship with H<sub>2</sub>O, for which the vertical portion represents the background stratosphere, with H<sub>2</sub>O mixing ratios of typically 4–5 ppm and the horizontal portion represents stratospheric air influenced by convection, with elevated H<sub>2</sub>O. There were 23 additional measurements of CIO when H<sub>2</sub>O was between 25 and 50 ppm (not shown for figure clarity), but these similarly lie near zero CIO and were acquired at temperatures exceeding 205 K.

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The stratospheric conditions required to produce a significant change in inorganic chlorine partitioning following convection are a function of the water vapor mixing ratio and temperature, as well as the pressure and aerosol surface area (Anderson et al., 2012). In Fig. 4, we show black points overlaid in both panels indicating the subset of data where  $\text{H}_2\text{O} \geq 8$  ppm and  $T \leq 202$  K. The black points represent 2.2% of the overall data collected during the DCOTSS mission. As is evident in both panels, there were no observed instances in DCOTSS where either ClO or ClONO<sub>2</sub> was elevated in the presence of low temperatures and elevated water vapor. Even when H<sub>2</sub>O was at the highest observed levels, thereby providing some relaxation to the low temperature constraint, perturbations to ClO and ClONO<sub>2</sub> were not observed. Multiple factors contributed to the lack of observed perturbations to inorganic chlorine partitioning in DCOTSS, but most importantly, most of the air that was sufficiently cold and wet during the mission tended to be very low in the stratosphere where inorganic chlorine mixing ratios are small. Also, the vast majority of DCOTSS sampling was of recent convection, so less time had elapsed for heterogeneous reactions to take place prior to observation. Further explanation for the absence of observed chlorine activation is provided in an extensive modeling study of DCOTSS data (Howar et al., 2025).

### 3.4 ClONO<sub>2</sub> Photochemistry and Kinetics

Chlorine nitrate is particularly important in stratospheric chemistry, as it is one of the two relatively long-lived reservoir species for inorganic chlorine (HCl being the other) and because it links the chlorine- and nitrogen-containing radical families (Stimpfle et al., 1999). The formation of ClONO<sub>2</sub> proceeds via a termolecular reaction of ClO with NO<sub>2</sub>,



This reaction is also responsible for ClONO<sub>2</sub> being the primary reservoir for ClO at night outside of winter polar regions.

During the daytime, the primary loss mechanism for ClONO<sub>2</sub> is photolysis. Both Cl + NO<sub>3</sub> and ClO + NO<sub>2</sub> products are possible (Burkholder et al., 2019), but Cl and NO<sub>3</sub> products are rapidly converted to ClO and NO<sub>2</sub> in the stratosphere via reaction with O<sub>3</sub> and photolysis, respectively, so the net reaction of ClONO<sub>2</sub> photolysis for both product channels is



With Reactions (R1) and (R2) dominating the formation and loss of ClONO<sub>2</sub> in the stratosphere, the concentration of ClONO<sub>2</sub> can be calculated using the photochemical steady-state approximation (Kawa et al., 1992; Stimpfle et al., 1999) as shown in Eq. (1).

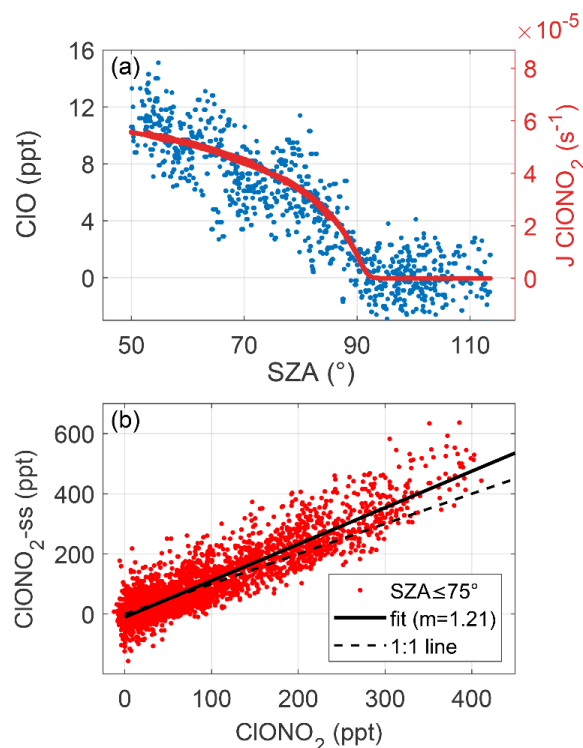
$$[\text{ClONO}_2]_{ss} = \frac{k_{\text{ClO}+\text{NO}_2}[\text{ClO}][\text{NO}_2]}{J_{\text{ClONO}_2}} \quad (1)$$

360

Here square brackets denote concentration in molecule cm<sup>-3</sup>, k is the second order rate coefficient for ClONO<sub>2</sub> production (R1) in cm<sup>3</sup> molecule<sup>-1</sup> s<sup>-1</sup> and J is the photolysis rate coefficient (R2) in s<sup>-1</sup>.



Figure 5a provides a demonstration of the interconnectedness between the mixing ratios of ClO and the photolysis of ClONO<sub>2</sub> in the stratosphere. Shown are the HAL ClO data from all DCOTSS research flights that included observations at solar zenith angle  $\geq 60^\circ$  at some point during the flight. The data are also filtered to include only higher altitudes ( $O_3 \geq 1500$  ppb) to restrict the analysis to similar air masses, although there is still scatter in the ClO data due to natural variability. Overall, mixing ratios of ClO clearly decrease with increasing solar zenith angle as expected, reaching 0 in darkness when SZA is approximately  $92^\circ$ . Overlaid on the ClO data is the photolysis rate of ClONO<sub>2</sub>, which is in excellent qualitative agreement.



**Figure 5.** Inorganic chlorine photochemistry from measurements during the NASA DCOTSS mission. (a) Measured ClO as a function of solar zenith angle (blue) with the photolysis rate of ClONO<sub>2</sub> (red) overlaid. The data shown are from all the research flights that included observations at solar zenith angle  $\geq 60^\circ$  and were selected for higher altitudes with  $O_3 \geq 1500$  ppb. (b) Comparison of steady-state calculated ClONO<sub>2</sub> versus measured ClONO<sub>2</sub>. The individual points (red) represent all available data from the 29 research flights of the DCOTSS mission filtered for solar zenith angle  $\leq 75^\circ$ . The black solid line is a fit to the data, with slope of 1.21, and the black dashed line is the 1:1 line for reference. The number of data points plotted in (b) is 3602, which is less than the total number of ClONO<sub>2</sub> measurements at  $SZA \leq 75^\circ$  due to the requirement for simultaneous measurements of ClO, NO<sub>2</sub>, O<sub>3</sub>, and T.

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To quantitatively examine the ClO–ClONO<sub>2</sub> relationship, Fig. 5b shows steady-state ClONO<sub>2</sub> calculated as in Eq. (1) versus measured ClONO<sub>2</sub> from HAL for all 29 research flights of the DCOTSS mission. Measurements of ClO and NO<sub>2</sub> from the HAL and CANOE instruments on the ER-2 are used in the steady-state calculation, along with the



375 recommended values of  $k_{\text{ClO}+\text{NO}_2}$  from the latest NASA JPL compendium (Burkholder et al., 2019) and values of  $J_{\text{ClONO}_2}$ , calculated as described in Section 2.2. The only filtering of the measured data is for  $\text{SZA} \leq 75^\circ$  to remove nighttime ClO. Overlaid on the data are a linear fit and the 1:1 line.

There is an excellent correlation between steady-state  $\text{ClONO}_2$  and measured  $\text{ClONO}_2$  mixing ratios during the DCOTSS mission. The relationship as shown in Fig. 5b is relatively compact and effectively linear. The slope of the fit line is 1.21 with y-intercept of  $-10$ . For  $\text{ClONO}_2$  mixing ratios 25 ppt and greater, the maximum difference between the fit line and the 1:1 line is 19%, peaking at the highest measured mixing ratio of 411 ppt; the percent difference is 16% at 200 ppt and 11% at 100 ppt. The primary limitation on the accuracy of determining  $\text{ClONO}_2$  from Eq. (1) is not the steady-state approximation itself, rather the uncertainties associated with the four variables that go into the equation. As discussed in Sections 2.1 and 2.2, the reported uncertainties of the ClO and  $\text{NO}_2$  measurements are 17% and 10%, respectively, and the uncertainty in  $J_{\text{ClONO}_2}$  is 10%. The largest uncertainty is in the JPL-recommended value of the production rate coefficient of  $\text{ClONO}_2$ ,  $k_{\text{ClO}+\text{NO}_2}$ , which is approximately 40% at the stratospheric temperatures considered here (Burkholder et al., 2019). Accounting for the uncertainties in the four variables on the righthand side of Eq. (1), the overall uncertainty in the steady-state calculated  $\text{ClONO}_2$  mixing ratios is approximately 45%. By comparison, the uncertainty in measured  $\text{ClONO}_2$  is 20%.

390 Given these uncertainties applicable to the x-axis (20%) and y-axis (45%) of Fig. 5b, the 19% or better level of agreement between steady-state calculated and measured  $\text{ClONO}_2$  at all mixing ratios  $>25$  ppt represents very good agreement within the total uncertainty. While the discrepancy between steady-state calculated and measured  $\text{ClONO}_2$  is not statistically significant, we note that the direction of change needed for improvement is by increasing measured  $\text{ClONO}_2$  or  $J_{\text{ClONO}_2}$  or by decreasing measured ClO, measured  $\text{NO}_2$ , or  $k_{\text{ClO}+\text{NO}_2}$ . The only individual value in the steady-state calculation of  $\text{ClONO}_2$  (Eq. 1) with an uncertainty larger than 19% is  $k_{\text{ClO}+\text{NO}_2}$ , suggesting that the JPL-recommended value may have a small high bias. Nevertheless, what can be stated with relatively high confidence, given the overall very good agreement shown in Fig. 5b, is that the large uncertainty in the JPL-recommended  $\text{ClONO}_2$  production rate coefficient can be reduced.

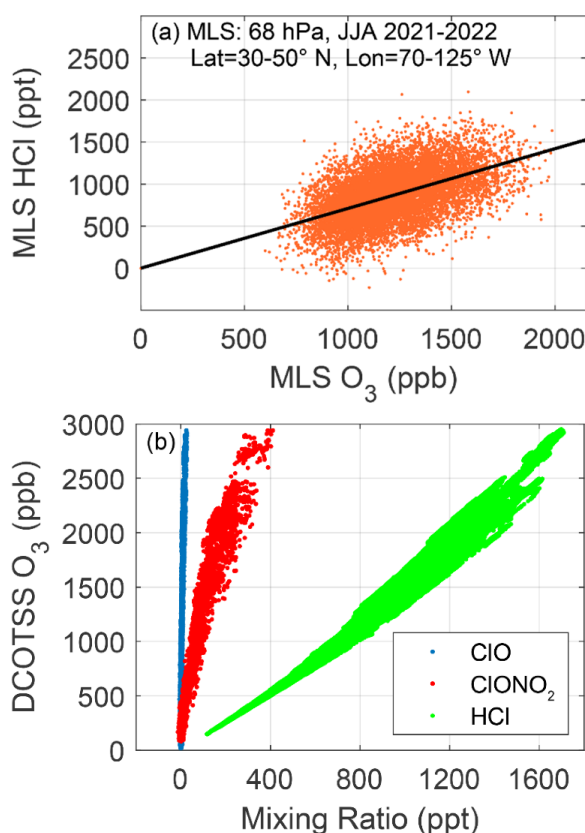
### 400 3.5 Inorganic Chlorine Partitioning and Budget

HCl is a long-lived reservoir and the dominant form of inorganic chlorine in the stratosphere. Mixing ratios of HCl and  $\text{O}_3$  have been observed to correlate linearly in the lower stratosphere, since both are produced primarily in the stratosphere and have long photochemical lifetimes, such that variations in their distributions are dominated by dynamical processes (Plumb and Ko, 1992; Marcy et al., 2004). Although HCl was not measured in situ during DCOTSS, we are able to calculate its mixing ratio for all flights of the mission using a technique that takes advantage of consistent relationships between HCl and  $\text{O}_3$  in the lower stratosphere at altitudes accessible by the ER-2. MLS satellite data were utilized for this purpose. One difference here from past studies that focused on a fixed linear  $\text{HCl}:\text{O}_3$  relationship in the upper troposphere and lowermost stratosphere (Marcy et al., 2004; Thornton et al., 2005; Marcy et al., 2007; Froidevaux et al., 2008; von Hobe et al., 2011; Jurkat et al., 2014; Wilkerson et al., 2021) is that we extend this analysis deeper into the stratosphere. More specifically, most previous studies did not exceed approximately 1000 ppb  $\text{O}_3$  in a linear  $\text{HCl}:\text{O}_3$  correlation plot, with 1450 ppb  $\text{O}_3$  from a single balloon flight (Wilkerson et al., 2021)



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representing the previous maximum to our knowledge, whereas for this study we examine values up to approximately 3000 ppb O<sub>3</sub>. To do this, we determine HCl:O<sub>3</sub> ratios at the available MLS pressure levels but do not require that those ratios be the same at all pressure levels, as is implicit in previous studies at lower altitudes over a smaller vertical range. Using the HCl:O<sub>3</sub> ratios determined from each MLS retrieval pressure level as reference values, HCl mixing ratios during DCOTSS were calculated for each second of every flight using DCOTSS pressure and O<sub>3</sub> measurements. This approach works well, as demonstrated below.



**Figure 6.** Measured HCl from MLS and calculated HCl for DCOTSS. (a) Scatter plot of HCl and O<sub>3</sub> measurements from the MLS satellite instrument obtained at 68 hPa during the summers of 2021 and 2022 over the approximate contiguous U.S. (30 – 50° N latitude, 70 – 125° W longitude). The slope of the fit line as shown is 0.71, or  $7.1 \times 10^{-4}$  when computed with HCl and O<sub>3</sub> in the same mixing ratio units (see text). The total number of points plotted is 10,100. (b) Inorganic chlorine data for all flights of the DCOTSS mission with DCOTSS O<sub>3</sub> as the vertical coordinate. HCl (green) is calculated as described in the text by determining the mean HCl:O<sub>3</sub> ratios from MLS at 100, 68 (shown in panel a), and 46 hPa and combining the interpolated slopes with high-resolution O<sub>3</sub> measurements from DCOTSS to produce HCl data for all DCOTSS flights. All ClO measurements (blue) for solar zenith angles  $\leq 75^\circ$  and ClONO<sub>2</sub> measurements (red) from DCOTSS are shown for reference. Calculated HCl is shown only for the P range 46 – 100 hPa. The number of HCl points shown is 355,876.



Figure 6a shows all concurrent measurements of HCl and O<sub>3</sub> acquired from MLS at the 68 hPa pressure level over the contiguous US in summers 2021 and 2022. Mean HCl at 68 hPa is 860 ppt and mean O<sub>3</sub> is 1210 ppb, yielding an HCl:O<sub>3</sub> ratio of  $7.1 \times 10^{-4}$  when in the same units. This value is the same as the slope acquired by fitting a line to the data plotted in Fig. 6a and forcing the small intercept of the fit line (39 ppt) to zero. Mean HCl:O<sub>3</sub> ratios and corresponding scatter plots with fit lines of HCl and O<sub>3</sub> were similarly generated for the other relevant MLS pressure levels, resulting in mean HCl:O<sub>3</sub> ratios of  $8.0 \times 10^{-4}$  at 100 hPa and  $5.6 \times 10^{-4}$  at 46 hPa. The estimated uncertainties in the mean ratios, based on the reported pressure-dependent accuracies of the MLS HCl and O<sub>3</sub> data (Livesey et al., 2022), are  $0.9 \times 10^{-4}$ ,  $1.6 \times 10^{-4}$ , and  $0.6 \times 10^{-4}$  at 68, 100, and 46 hPa, respectively. The HCl:O<sub>3</sub> ratios for all pressure levels between 46 and 100 hPa were produced by interpolating between the three available MLS pressure levels. Supplement Fig. S4 shows scatter plots of HCl versus O<sub>3</sub> mixing ratios for the three MLS pressure levels discussed here as well as 32 hPa, where the HCl:O<sub>3</sub> relationship is substantially different. It is evident from this plot that O<sub>3</sub> mixing ratios rise faster than HCl mixing ratios deeper in the stratosphere, necessitating using a different HCl:O<sub>3</sub> ratio for the lower pressures of this study. The 46 – 100 hPa pressure range covers the vertical extent of nearly all stratospheric data obtained during the DCOTSS mission (see Fig. S2).

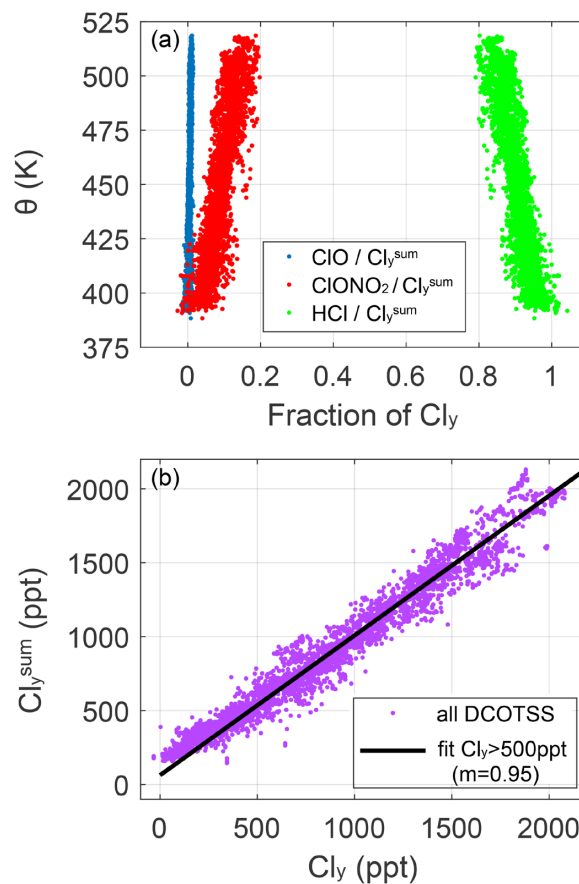
Using the MLS-derived conversion factors described above, in combination with in situ O<sub>3</sub> mixing ratios and pressure data from instrumentation on the ER-2, mixing ratios of HCl at high time resolution were calculated for all DCOTSS flights, as shown in Fig. 6b. DCOTSS in situ O<sub>3</sub> is the vertical coordinate, with HCl data shown for the pressure range 46 – 100 hPa. Also shown for reference are ClO (filtered for daytime,  $SZA \leq 75^\circ$ ) and ClONO<sub>2</sub> measured in situ from the HAL instrument, as previously plotted in Figs. 2c and 2d. The inorganic chlorine mixing ratios in Fig. 6b appear as expected, with the radical ClO having the smallest mixing ratios, followed by the reservoir species ClONO<sub>2</sub>, followed by the longer-lived dominant reservoir HCl. Consistent with previously discussed increases in ClO and ClONO<sub>2</sub> that correlate with increases in potential temperature or O<sub>3</sub>, HCl values increase dramatically with higher O<sub>3</sub> mixing ratios in Fig. 6b. The scatter in the data is low, with a relatively tight spread in the individual variables. There are a range of pressure levels represented by each O<sub>3</sub> mixing ratio, which is driving the spread in the calculated HCl data.

To quantitatively examine inorganic chlorine partitioning in the midlatitude lower stratosphere, we ratio measured ClO, measured ClONO<sub>2</sub>, and calculated HCl to the sum of these three mixing ratios,  $Cl_y^{\text{sum}} \approx ClO + ClONO_2 + HCl$ . Fig. 7a shows this partitioning for all research flights of the DCOTSS mission as a function of potential temperature. Due to the more limited vertical range of calculated HCl data, the lower limit of the potential temperatures shown is approximately 390 K, which represents exclusively stratospheric air during DCOTSS. HCl is the dominant form of inorganic chlorine at all potential temperatures in Fig. 7a, especially at the lowest levels. Complementary to Fig. 6b, which shows universal increases in ClO, ClONO<sub>2</sub>, and HCl mixing ratios as organic chlorine-containing compounds are converted to inorganic form deeper in the stratosphere, Fig. 7a shows that the fractional contribution of HCl to total inorganic chlorine decreases with  $\theta$  in the stratosphere due to repartitioning. Over the 390 – 520 K potential temperature range, the fraction of total inorganic chlorine in the form of HCl decreases from approximately 0.96 ( $\pm 0.03$ ) to 0.84 ( $\pm 0.03$ ), while the fraction of ClONO<sub>2</sub> increases from approximately 0.04 ( $\pm 0.02$ ) to 0.15 ( $\pm 0.02$ ), and the fraction of ClO increases from approximately 0.001 ( $\pm 0.006$ ) to 0.01 ( $\pm 0.002$ ). Listed uncertainties represent the



calculated standard deviations. Despite the relatively small fractional contribution of ClO to total inorganic chlorine, ClO exerts significant control over stratospheric ozone due to the catalytic nature of ozone loss reactions.

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**Figure 7.** Inorganic chlorine partitioning and budget in the stratosphere during the NASA DCOTSS mission. (a) Ratios of measured ClO (blue), measured ClONO<sub>2</sub> (red), and calculated HCl (green) to the sum of the mixing ratios of these inorganic chlorine compounds,  $Cl_y^{sum}$ . The ClO measurements are only shown for solar zenith angle  $\leq 75^\circ$ . (b) Sum of all ClO, ClONO<sub>2</sub>, and HCl data from DCOTSS versus total inorganic chlorine,  $Cl_y$ , independently calculated using measurements of organic chlorine compounds. The fit line (black) is for data with  $Cl_y > 500$  ppt and has a slope of 0.95. The vertical range represented corresponds to 46 – 100 hPa, the pressures over which calculated HCl mixing ratios are available. The total number of points plotted in (b) is 3666, a number set by the requirement for concurrent availability of all measurements.

An important test of the accuracy of the inorganic chlorine mixing ratios determined in this study, and a critical evaluation of our understanding of midlatitude chlorine chemistry more generally, is to evaluate the chlorine budget during the DCOTSS mission. Specifically, we compare agreement between the sum of the individual mixing ratios ( $Cl_y^{sum}$ , as defined above) and total inorganic chlorine ( $Cl_y$ ), which is independently calculated from organic chlorine

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compounds measured on the ER-2, as described in Section 2.2. Fig. 7b shows this comparison, using all data from the 29 DCOTSS research flights, filtered to include only daytime ClO ( $SZA \leq 75^\circ$ ). The agreement in Fig. 7b between  $Cl_y^{sum}$  and calculated  $Cl_y$  is highly robust. The relationship is linear, and a fit of all data points where  $Cl_y$  is greater than 500 ppt produces a slope of 0.95. The uncertainty in calculated  $Cl_y$  is relatively small, so this level of agreement is a testament to the accuracy of the ClO and ClONO<sub>2</sub> in situ measurements as well as the approach used here for calculating HCl. Moreover, this result supports the accuracy of measured ClONO<sub>2</sub> and ClO as it pertains to the differences discussed in Section 3.4 between measured and steady-state calculated ClONO<sub>2</sub>. To our knowledge, the agreement of  $Cl_y^{sum}$  and  $Cl_y$  to within 5% represents the best inorganic chlorine budget agreement ever obtained using in situ ER-2 data, including missions where the budget was deemed to be in agreement, such as POLARIS at northern high latitudes in summertime (Bonne et al., 2000) and SOLVE-THESEO in the Arctic vortex when inorganic chlorine was nearly fully activated in wintertime (Wilmouth et al., 2006).

Despite the overall excellent budget agreement for the DCOTSS mission, the data below approximately 500 ppt in Fig. 7b deviate from the fit line, and even more so from the 1:1 line. Specifically,  $Cl_y^{sum}$  becomes increasingly high relative to total inorganic chlorine as  $Cl_y$  decreases below 500 ppt, with the value of  $Cl_y^{sum}$  approximately 170 ppt at 0 ppt  $Cl_y$ . We consider three possibilities to explain this discrepancy: (1) Measured ClO or measured ClONO<sub>2</sub> is too high. This cannot be the cause, as ClO and ClONO<sub>2</sub> mixing ratios are not large enough at these  $Cl_y$  levels to resolve the issue even if reduced to zero. (2) Our calculated HCl is biased high at low mixing ratios. This would likely require a high bias in the MLS measurement of HCl or a low bias in the MLS measurement of O<sub>3</sub> in the lowermost stratosphere. The HCl:O<sub>3</sub> slope found here at 100 hPa ( $8.0 \times 10^{-4}$ ) agrees very well with the value ( $\sim 8 \times 10^{-4}$ ) of Jurkat et al. (2014) but is higher than values ( $\sim 5 \times 10^{-4}$ ) from two earlier aircraft studies (Marcy et al., 2004; von Hobe et al., 2011); however, given the measurement uncertainties and expected differences between the studies due to different latitudes being sampled, the results are consistent. We note that it is very unlikely that DCOTSS O<sub>3</sub> used to calculate HCl is biased high at low mixing ratios, because O<sub>3</sub> was measured by two independent instruments (ROZE and UCATS) and the two data sets agree. (3) Finally, we consider that  $Cl_y$  determined from organics might be biased low at low mixing ratios. If true, this could resolve the discrepancy in Fig. 7b below 500 ppt by shifting  $Cl_y$  to the right rather than shifting  $Cl_y^{sum}$  down. The small estimated uncertainty of 6% for  $Cl_y > 500$  ppt increases substantially as mixing ratios decrease, exceeding a factor of 2 at the smallest mixing ratios, due largely to the difficulties in accurately defining the total organic chlorine content at the tropopause of the sampled stratospheric airmasses, as well as the precise contribution to  $Cl_y$  from the decomposition of VSL species (Howar et al., 2025). Ultimately, the overall agreement in Fig. 7b is excellent, and while the disagreement between  $Cl_y^{sum}$  and  $Cl_y$  at the lowest values appears systematic, it is small and generally within the estimated uncertainties.

### 3.6 Wildfire Aerosol

The Australian New Year's (ANY) fires of December 2019 and January 2020 were responsible for injecting a record quantity of biomass burning products into the Southern Hemisphere midlatitude stratosphere and led to unexpected, dramatic perturbations to stratospheric composition in the months following the fires due to changes in dynamics and chemistry (Khaykin et al., 2020; Santee et al., 2022; Strahan et al., 2022; Salawitch and McBride, 2022;



505 Solomon et al., 2023; Ma et al., 2024). In particular, there were substantial changes to the stratospheric chlorine species, namely suppression of HCl and concomitant increases in ClO and ClONO<sub>2</sub>, as measured by the MLS and ACE satellite instruments. These observed changes could not be reproduced in models using typical chlorine heterogeneous reactions known to be important at cold temperatures (e.g., HCl + ClONO<sub>2</sub>) or using the N<sub>2</sub>O<sub>5</sub> hydrolysis reaction known to be enhanced following volcanic eruptions (Strahan et al., 2022, Solomon et al., 2023).

510 Solomon et al. (2023) proposed a new mechanism to explain the perturbed stratospheric inorganic chlorine chemistry following the ANY fires, which was able to reasonably reproduce the MLS and ACE satellite observations of ClO, ClONO<sub>2</sub>, and HCl. In this mechanism, HCl solubility is posited to be greatly increased due to the presence of a high fraction of organic species in aged biomass burning aerosol, allowing heterogeneous chlorine activation at far warmer temperatures than previously considered possible. Stone et al. (2025) subsequently modeled the stratospheric impact of the ANY fires for the 2 years after their occurrence, highlighting better results when HCl solubility in  
515 wildfire organic aerosols is modeled differently from stratospheric background organic aerosols. In theory, if the Solomon et al. (2023) mechanism is correct, the presence of biomass burning organic aerosol could allow for HCl uptake and initiate heterogeneous chemical reactions that reduce HCl and enhance ClONO<sub>2</sub> and ClO at temperatures typical of the midlatitude lower stratosphere during DCOTSS.

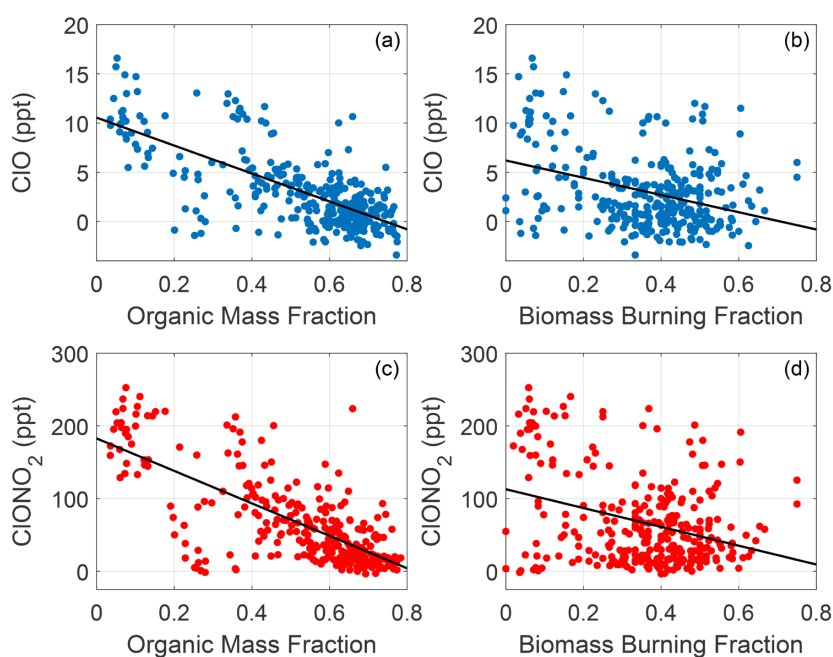
Instrumentation on board the ER-2 during the DCOTSS mission provided several measurements that enable us to  
520 test the relationship between organic aerosol and inorganic chlorine in the midlatitude stratosphere. In particular, aerosol composition measurements from the PALMS-NG instrument were used to generate organic mass fraction (OMF) and biomass burning fraction (BBF) data products. OMF is defined as the ratio of organic mass to the combined mass of organics and sulfate, i.e.,  $\text{Org} / (\text{Org} + \text{Sulf})$ . This quantity is derived from laboratory calibrations and calculated only for non-refractory particle types, including internally mixed sulfate–organic particles, biomass  
525 burning, and meteoric particles. BBF is defined as the number fraction of particles characterized by strong mass spectral signals of potassium and organic species and attributed directly to a biomass burning origin. During past flights, the PALMS instrument acquired mass spectra within biomass burning plumes; particles contained carbon, potassium, organics, and ammonium ions but not pure soot. These mass spectra were used to develop a biomass-burning particle signature for subsequent characterization (Hudson et al., 2004). Particle identification and organic  
530 mass calibration follow methods previously described (Froyd et al., 2019) and have been refined for DCOTSS flights (Shen et al., 2025).

Figure 8 shows scatter plots of ClO and ClONO<sub>2</sub> mixing ratios measured by HAL versus OMF and BBF reported by PALMS-NG. Data are shown for all 16 flights where the aerosol data are available, i.e., the DCOTSS flights in 2022, and are exclusively in the stratosphere ( $\theta > 390$  K). The black line in each panel represents a least-squares fit to  
535 the data. The four panels of Fig. 8 show that ClO and ClONO<sub>2</sub> mixing ratios decrease with increasing organic mass fraction and biomass burning fraction. This result is inconsistent with what might have been expected if the Solomon et al. (2023) mechanism had been significant in the Northern Hemisphere midlatitudes during the sampling period of the DCOTSS mission. It is important to note that the negative correlations in the panels of Fig. 8 are not causal; ClO and ClONO<sub>2</sub> reliably increased as a function of potential temperature (age of air) during DCOTSS, while OMF and  
540 BBF both typically decreased with  $\theta$  (more significantly for OMF), which results in negative correlations when these



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variables are plotted against one another. That said, if significant HCl uptake on organic aerosol and heterogeneous chlorine activation had occurred in DCOTSS 2022 due to high OMF and/or BBF, we would have expected to see anomalously high ClO or ClONO<sub>2</sub> mixing ratios toward the righthand sides of the panels in Fig. 8. The data are noisy, but there are no obvious ClO or ClONO<sub>2</sub> enhancements. Supplement Fig. S5 shows similar results in analogous plots to Fig. 8, with ClO and ClONO<sub>2</sub> plotted as a fraction of Cl<sub>y</sub>. The earlier analyses in this paper (e.g., Fig. 2) confirm that ClO and ClONO<sub>2</sub> were not significantly perturbed during the mission.



**Figure 8.** Scatter plots of ClO mixing ratios from HAL with (a) organic mass fraction and (b) biomass burning fraction (by particle number) from PALMS-NG; (c) and (d) analogous plots for ClONO<sub>2</sub>. Plotted are data from the 16 DCOTSS flights (all in 2022) where aerosol fractional values were reported; data shown are for potential temperature  $\geq 390$  K, and ClO measurements are only shown for solar zenith angle  $\leq 75^\circ$ . Due to the relatively sparse reporting of the aerosol fractional data, the inorganic chlorine data are interpolated to the fractional data; this introduces very little error. The black lines in each panel represent least-squares fits to the data.

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This null result should not be extended to comment on the specific conditions of the ANY fires. However, the result does appear to show that the Solomon et al. (2023) mechanism was not pervasive in the background midlatitude lower stratosphere during DCOTSS, even though organic species in biomass burning aerosol were present and temperatures were sufficiently cold. The presence of biomass burning organic aerosols in the DCOTSS-sampled air masses is supported by independent measurements from multiple instruments (Sharpe et al., 2026). One relevant consideration with respect to chlorine activation is that the DCOTSS sampling was in summertime, and Stone et al. (2025) found a significant seasonal impact of the ANY fires due to photochemistry, with perturbations to inorganic chlorine in wintertime being dominant relative to other seasons. Also, as noted by Chipperfield and Bekki (2024) and

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Stone et al. (2025), the physical state (liquid, glassy, solid) and detailed composition of the aerosol are important factors and were not known for the ANY fires, which prevents us from comparing with DCOTSS. Further study, particularly in the laboratory, is needed to better predict the impact of wildfires and organic aerosol on stratospheric chlorine and the ozone layer.

#### 4 Conclusions

In situ inorganic chlorine measurements of ClO and ClONO<sub>2</sub> were obtained from the ER-2 aircraft over a total of 6 months in the summers of 2021 and 2022 as part of the NASA DCOTSS mission over Northern Hemisphere midlatitudes. The precision of the chlorine data from the HAL instrument was excellent throughout the campaign, to our knowledge the best ever obtained from an aircraft platform at these altitudes: 2 ppt for ClO and 5 ppt for ClONO<sub>2</sub> in 35 sec and, with averaging, a detection limit of under 1 ppt for ClO and 2 ppt for ClONO<sub>2</sub>. In particular, the ClO data from HAL have more than an order of magnitude higher precision than MLS and ACE satellite data for many of the lower stratospheric altitudes studied here, enabling analyses that would not otherwise be possible.

Analysis of the in situ ClO and ClONO<sub>2</sub> measurements shows a relatively compact distribution and increasing mixing ratios with potential temperature or ozone as the vertical coordinate, consistent with all four variables increasing with age of air in the lower stratosphere. Other than ClO mixing ratios going to zero at night as expected, there were no substantial outliers in the vertical profiles of ClO or ClONO<sub>2</sub>, especially when plotted against O<sub>3</sub>. The lack of substantial outliers represents an important scientific result, as discussed below, and also speaks to the quality of the data set for performing tests of stratospheric inorganic chlorine chemistry. The peak ClO and ClONO<sub>2</sub> mixing ratios observed during the DCOTSS mission were approximately 30 ppt and 400 ppt, respectively.

One of the key questions of the DCOTSS campaign is whether chlorine activation would be observed in the lower stratosphere in the presence of low temperatures and elevated water vapor from convective injection, as suggested by several modeling studies. We found no significant difference in measured ClO or ClONO<sub>2</sub> mixing ratios during the mission regardless of whether the sampled air had enhanced water vapor and low temperatures. One factor contributing to this result is that most of the air sampled in DCOTSS that was sufficiently cold and wet to accelerate heterogeneous chlorine reactions was located very low in the stratosphere, with small inorganic chlorine mixing ratios. Also, most DCOTSS sampling of enhanced water vapor followed recent convection because it was more easily targeted in flight planning, but this meant that not much time had elapsed for the relatively slow heterogeneous chlorine reactions to take place prior to ER-2 intercept. While this null result from the two years of DCOTSS sampling suggests that chlorine activation is not pervasive in the Northern Hemisphere midlatitude summer stratosphere following overshooting convection, it cannot be interpreted to mean that inorganic chlorine is never perturbed there: (1) Despite the relatively extensive sampling during the DCOTSS mission, only a small fraction of all convective events in the summers of 2021 and 2022 were sampled; (2) there is significant interannual variability in stratospheric conditions, so sampling other summers could yield different results; and (3) the presence of elevated aerosol surface area beyond 2021–2022 levels in conjunction with enhanced water vapor mixing ratios and low temperatures would make the stratosphere much more vulnerable to chlorine activation following convection, as recently demonstrated by Howar et al. (2025).



595 The production and loss mechanisms used to calculate photochemical steady-state mixing ratios of ClONO<sub>2</sub> in  
the lower stratosphere were found to be in good agreement with in situ measurements of ClONO<sub>2</sub> throughout the  
DCOTSS mission. A scatter plot of steady-state versus measured ClONO<sub>2</sub> is relatively compact and linear across the  
full range of mixing ratios, with a slope of 1.21 and agreement to within 19% or better for ClONO<sub>2</sub> > 25 ppt. This  
level of agreement is much smaller than the uncertainty of the steady-state calculated values (45%), suggesting that  
600 the uncertainty in the JPL-recommended ClONO<sub>2</sub> production rate coefficient,  $k_{\text{ClO}+\text{NO}_2}$ , which represents by far the  
largest uncertainty in the calculation, can be reduced.

HCl, the largest stratospheric reservoir of inorganic chlorine, was not measured on the ER-2 during the DCOTSS  
mission, but we calculate HCl mixing ratios in this study at high time resolution for every research flight. This was  
accomplished by establishing the HCl:O<sub>3</sub> relationship from MLS measurements at the available MLS retrieval  
605 pressure levels (100, 68, and 46 hPa) corresponding to ER-2 stratospheric altitudes, interpolating those ratios against  
pressure to fill in the measurement gaps, and then multiplying the interpolated values as a function of pressure by  
measured in situ O<sub>3</sub>. The HCl calculated in this manner overall appears very well behaved and accurate. HCl dominates  
inorganic chlorine partitioning but decreases approximately 12% as potential temperature rises from 390 to 520 K,  
while ClONO<sub>2</sub> and ClO partitioning increase substantially.

610 Further evaluation of the inorganic chlorine chemistry observed in DCOTSS involves determination of the  
inorganic chlorine budget. We find that the sum of ClO + ClONO<sub>2</sub> + HCl ( $\text{Cl}_y^{\text{sum}}$ ) is in excellent agreement with total  
inorganic chlorine ( $\text{Cl}_y$ ), which is independently determined from measurements of organic chlorine compounds. For  
mixing ratios greater than 500 ppt, the agreement between  $\text{Cl}_y^{\text{sum}}$  and  $\text{Cl}_y$  is within 5% (slope of 0.95 when plotting  
 $\text{Cl}_y^{\text{sum}}$  vs  $\text{Cl}_y$ ), representing the best inorganic chlorine budget agreement ever obtained on the ER-2 to our knowledge.  
615 There is a small disagreement between  $\text{Cl}_y^{\text{sum}}$  and  $\text{Cl}_y$  at mixing ratios below approximately 500 ppt that appears to  
become larger approaching the origin. Despite this disparity generally being within the uncertainty, it appears to be  
systematic. Some potential factors giving rise to this discrepancy can be ruled out entirely, with a high bias in  
calculated HCl or a low bias in  $\text{Cl}_y$  at the lowest mixing ratios being the most likely cause.

Finally, we consider the potential impact of aged organic aerosols on inorganic chlorine mixing ratios. Solomon  
620 et al. (2023) previously proposed a new mechanism to explain the large stratospheric reductions in HCl and increases  
in ClO and ClONO<sub>2</sub> following the ANY fires in 2019–2020 in which HCl solubility is greatly increased due to the  
presence of a high fraction of organic species in the biomass burning aerosol, which would allow heterogeneous  
chlorine activation to occur at temperatures commonly found throughout the midlatitude lower stratosphere. During  
the DCOTSS mission, we did not observe any evidence of this mechanism occurring in the Northern Hemisphere  
625 midlatitude stratosphere, even when organic mass fraction and biomass burning fraction were elevated. This null result  
should not be interpreted as a challenge to the Solomon et al. (2023) mechanism for explaining the changes to  
stratospheric inorganic chlorine observed following the ANY fires. Rather, the DCOTSS observations provide an  
important data point for how pervasive the mechanism may be in the global stratosphere in the presence of organic  
and biomass burning aerosols.

630 Looking ahead, the future of stratospheric chlorine measurements is uncertain (Salawitch et al., 2025). The MLS  
satellite instrument will soon be taken out of service, and while ACE is still producing data, it has relatively sparse



observational coverage outside of the high latitudes and is well beyond its expected lifetime. Opportunities for in situ measurement of stratospheric chemistry like the DCOTSS mission are infrequent. The future holds many unknowns for possible perturbations to inorganic chlorine and ozone in the stratosphere due to influences such as intense wildfires, volcanic eruptions, rocket launches and space metal reentry, VSL chlorine emissions, increasing water vapor, decreasing temperatures, and potential climate intervention strategies. The consistently well-behaved measurements of ClO and ClONO<sub>2</sub> during the DCOTSS mission, despite sampling in many different types of air masses, provide valuable insight into the current stratosphere and serve as a point of reference for future studies in light of potential stratospheric perturbations on the horizon.

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#### **Code and data availability**

All data utilized in this study are publicly available. Data from the DCOTSS project can be obtained from the NASA Atmospheric Science Data Center at [https://doi.org/10.5067/ASDC/DCOTSS-Aircraft-Data\\_1](https://doi.org/10.5067/ASDC/DCOTSS-Aircraft-Data_1). Aura Microwave Limb Sounder (MLS) data are publicly available at <https://search.earthdata.nasa.gov/search?fi=MLS>. ACE-FTS data are available at <https://database.scisat.ca>. Photodissociation frequencies computed along the DCOTSS flight tracks are available at <https://doi.org/10.5281/zenodo.14968099>.

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#### **Author contributions**

DMW conceptualized the paper, performed the data analysis and visualization, and prepared the manuscript with contributions from all co-authors. All authors were involved in data collection essential to this study: DMW, JSH, and JGA for inorganic chlorine; LVH and RJS for total inorganic chlorine and photolysis rates; EJH for total inorganic chlorine and ozone; JBS, DSS, DMW, JSH, and JGA for water vapor; JMSC, ERD, RAH, and TFH for nitrogen dioxide and ozone; DJC and XS for aerosols; MLS for MLS satellite data; TPB for meteorology; CRH, PAN, and KPB for flight planning; KPB and FNK for DCOTSS mission leadership.

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#### **Competing interests**

One of the co-authors is a member of the editorial board of Atmospheric Chemistry and Physics.

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