

# Electrostatics and Collision Dynamics of Ice and Anthropogenic Smoke Particles in the Mesosphere/Lower Thermosphere

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**Abstract.** The increase in satellite launches raises the anthropogenic influx of various elements into the Mesosphere and Lower Thermosphere (MLT), comparable to the natural influx caused by meteoric ablation. This study investigates the electrostatic interactions between **nano-meter-sized** ice particles and remnants of space debris using a classical electrostatic framework. Aside from the Coulomb interaction, the attractive force between two particles at short distances, arising from polarization, is taken into account. Collision outcomes, the effective velocity regime for collisions, and the subsequent aggregation probability are estimated. Aggregation is limited to a specific range of collision velocities between minimum and maximum values. This range varies depending on factors such as particle size, mass density, and dielectric constant. ~~For most particles, the aggregation velocities range from a few  $\text{m s}^{-1}$  to several tens of  $\text{m s}^{-1}$ , where smaller particles may need significantly higher velocities to form stable aggregates.~~ **For most particles, the velocity range of aggregation is from a few  $\text{m s}^{-1}$  to several tens of  $\text{m s}^{-1}$ , where smaller particles may need significantly higher velocities to form stable aggregates.** When considering the collisions of particles in thermal motion, it is found that  $\text{Al}_2\text{O}_3$  (due to its greater abundance) and  $\text{TiO}_2$  (due to its higher dielectric constant), both originating from anthropogenic sources, may dominate in the formation of ice-anthropogenic particle aggregates. In the MLT region, the formation of stable aggregates from the collision of ice with particles from space debris, which one may denote as anthropogenic smoke particles (ASPs), is similar to that from collisions with meteoric smoke particles (MSPs).

## 1 Introduction

Due to the increasing number of space program projects, perhaps 75,000 additional satellites will be in orbit by 2030 (Schulz and Glassmeier, 2021; Howard and Ah, 2022; Murphy et al., 2023; Schulz et al., 2026). Defunct satellites, spent rocket stages, fragmentation events, and mission-related ejecta lead to the rapid accumulation of anthropogenic space debris. They pose a major concern for both spacecraft safety and the long-term sustainability of the near-Earth space environment. These pieces of space debris re-enter the Earth's upper atmosphere, along with the natural influx from meteoroids, undergo intense aerodynamic heating, ablation, and fragmentation, and inject metal atoms to form metal layers (Schulz and Glassmeier, 2021).

The space debris is composed of many elements, particularly Li, Al, Cu, Nb, Hf, Pb, Be, Mg, Ti, Cr, Fe, Ni, Zr, Ga, Mo, In, Sn, Ba, Ce, Ta, W, Sb, Bi, Cd, Pr, and Nd (Murphy et al., 2023; Schulz and Glassmeier, 2021; Schulz et al., 2026). Schulz and Glassmeier (2021) reported anthropogenic materials contribute about 2.8% compared to the annual influx of interplanetary

25 dust particles. They further suggested that atmospheric re-entry of space debris may increase by 13% to 40% in the near future. Recently, Schulz et al. (2026) updated the estimates of annual anthropogenic influx into the Earth's atmosphere. Confirming the findings from the earlier work, they further report that the injected mass doubled between 2020 and 2024, reaching  $887 \pm 123$  tonnes in 2024. This results in a 16% increase in anthropogenic metal injection into the mesosphere/lower thermosphere (MLT) region (between 70 to 120 km) compared to the natural meteoric metal injection as of 2024 (Schulz et al., 2026). It is anticipated that, in the future, over 75,000 large constellation satellites could increase this value to 70% (Schulz et al., 2026). As a result, the mass influx of elements from space debris is becoming significant and comparable to the natural input by meteoroid ablation. Additionally, some metals, such as Al and Cu, are dominantly originating from human space activities. In a 75,000-satellite constellation future scenario, Schulz et al. (2026) predicted that the annual elemental mass injection for Al and Ti will be increased by factors of  $13.1 \pm 4.4$  and  $22.9 \pm 7.6$  respectively. The detailed estimates for other elements are given in Table 5 of Schulz et al. (2026). Thus, space debris may significantly influence the formation of high-altitude clouds, introduce new chemical pathways, affect radiative forcing, and produce particles with unique optical properties (Murphy et al., 2023; Schulz et al., 2026).

In the MLT region, the presence of highly reactive atomic oxygen leads to the formation of metal oxides, which later form nm-sized meteoric smoke particles (MSPs) (Plane, 2003). The interactions and growth of solid particles in the MLT have been studied in detail for MSPs, focusing on their transport and growth (Megner et al., 2008; Bardeen et al., 2008), their influence on the charge balance (Baumann et al., 2013, 2015), and their origin from meteors (Plane et al., 2015). Additionally, there is evidence that charged MSPs influence the radar diagnostics of the D-region ionosphere using incoherent scatter (Gunnarsdottir et al., 2024). Charged MSPs may interact electrostatically, influencing the chemical and microphysical framework of the MLT through processes such as electron depletion, ion balance modification, and ice-particle coagulation (Baumann et al., 2015). These MSPs may also act as condensation nuclei for heterogeneous ice particle formation. In the Polar Region, during the summer, due to summer-to-winter meridional circulation, air experiences adiabatic cooling in the mesosphere, causing temperatures to drop significantly to around 120 K. The low temperatures, combined with atmospheric water, allow ice particles to form. During polar summer, the aggregation of the ice particles forms polar mesospheric clouds, which give rise to noctilucent clouds (NLC) observed from the ground and contribute to the formation of radar echoes known as polar mesospheric summer echoes. Both phenomena, which are observed from the Earth, have increased in recent years (Latteck et al., 2021). The increase in water in the MLT is likely a decisive factor for this (Lübken et al., 2021), while studies on ice formation indicate that the formation of ice particles must occur through heterogeneous condensation (Tanaka et al., 2022). These are of considerable interest on account of their potential relevance to global warming. Therefore, the influence of anthropogenically originated particles on the formation of NLC is possibly significant, and it is crucial to thoroughly investigate their role in the growth process. The formation process of these particles is poorly constrained. Due to their similarity with MSPs, we refer to these particles as anthropogenic smoke particles (ASPs).

In the MLT region, which consists of a weakly ionized, complex dusty plasma, the space debris, MSPs, and ice particles become electrically charged through mechanisms such as electron and ion accretion, photoemission, and other ionization processes (Rapp and Thomas, 2006; Mann et al., 2014). The description of the interactions is largely based on theoretical

60 considerations, the physical properties depend on the dust composition, and in addition, small size effects become important for particle sizes of a few nm (Mann et al., 2014). As the surface charge on the particles significantly influences their agglomeration, it should be considered to investigate the growth of ice onto the ASPs, and electrostatic interactions should be taken into account. Given that the frequency of ice particles in the MLT appears to have increased in recent decades, it is reasonable to investigate the extent to which anthropogenic particles contribute to this increase.

65 In this paper, we investigate electrostatic interactions between charged ice and a component of the solid particles predicted to form from space debris, that is, the ASP. ~~The~~The model of Bichoutskaia et al. (2010) has been applied to quantify the electrostatic interaction energy between charged ice and oxides that are found in space debris. The study by Baptiste et al. (2021) demonstrated how electrostatic interactions influence the growth of MSPs and ice particles, but did not take into account space debris particles. ~~The model of Bichoutskaia et al. (2010) has been applied to quantify the electrostatic interaction energy between charged ice and oxides. Due to its dielectric nature, two particles with charges of the same sign can experience a strong attractive interaction at small separation distances. This attractive interaction is determined by the polarization of surface charge, resulting in regions of positive and negative surface charge density on one particle due to the other.~~ Two particles with charges of the same sign will experience a repulsive Coulomb force at certain separation distances. However, at small separations, their dielectric properties lead to a strong attractive interaction. This attraction arises from the polarization of surface charge between like-charged particles, leading to regions of positive and negative surface charge density on one particle due to the influence of the other. Therefore, the collision outcomes and the subsequent formation of a stable aggregate differ significantly among oppositely charged, likely charged, and charge-neutral particle collisions.

75 The remainder of this manuscript is organized as follows: Section 2 outlines the applied model framework of electrostatics and collisions, along with the parameters assumed for particles in the MLT. Section 3 presents the numerical results, and Section 4 provides a concluding discussion. Additionally, Appendix A & B contain a series of tables summarizing the numerical results obtained.

## 2 Electrostatics and Collision Dynamics

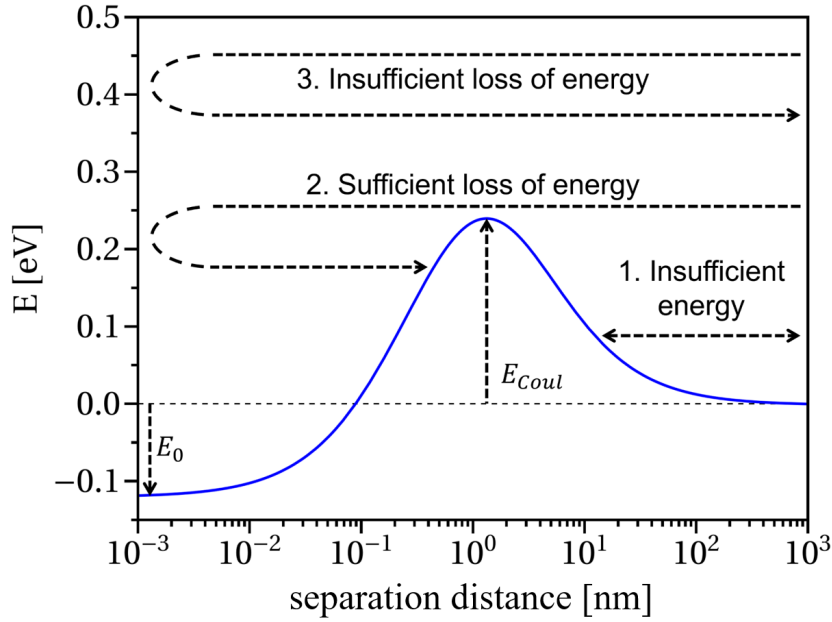
The MLT region constitutes a weakly ionized dusty plasma environment composed of electrons, positive ions, neutral species (Mann et al., 2019). At MLT altitudes, the temperature is extremely low (110–150 K). The ambient plasma density varies 85 strongly between day and night conditions, with typical electron densities ranging from  $\sim 35 \text{ cm}^{-3}$  at night to  $\sim 4.5 \times 10^3 \text{ cm}^{-3}$  during daytime conditions (Havnes et al., 2001; Sodha et al., 2011). The ice, ASPs, and MSPs are embedded within this plasma medium. They undergo electrostatic charging via electron and ion collection currents at the particle surface. Further, under the exposure to extreme EUV/UV fluxes, photoelectrons are also emitted from the ice and dust particles. However, the degree of photoemission is significantly dependent on the work function of the particles. The work function 90 of ice varies in a wide range, from as high as 8.7 eV for pure ice to as low as 2.3 eV for dirty ice (Misra et al., 2011). Since the dust number density is comparable to that of ambient plasma electrons and ions, we can estimate the dust charge by examining the collective variation in plasma electron and ion densities. This assessment should consider the effects of plasma particle capture, photodetachment, and photoionization

**Table 1.** Dielectric constants and mass densities of ice and metal oxides considered in this study

| Particle                                        | Dielectric constant, $\kappa$ | Density, $\rho$ (g/cc) |
|-------------------------------------------------|-------------------------------|------------------------|
| Ice, H <sub>2</sub> O                           | 100                           | 0.92                   |
| Iron oxide, FeO                                 | 14.2                          | 5.74                   |
| Silicon dioxide, SiO <sub>2</sub>               | 3.9                           | 2.65                   |
| Aluminium oxide, Al <sub>2</sub> O <sub>3</sub> | 9                             | 3.50                   |
| Titanium dioxide, TiO <sub>2</sub>              | 95                            | 4.25                   |
| Magnesium oxide, MgO                            | 9.6                           | 3.58                   |

(Sodha et al., 2011; Misra et al., 2011; Baumann et al., 2013, 2015). In the MLT region, the density of dust particles is comparable to that of the ambient plasma electrons and ions. The dust can act both as individual particles and collectively. Under the plasma conditions present in the MLT region, the dust acquires charges through plasma particle capture, photodetachment, and photoionization (Sodha et al., 2011; Misra et al., 2011; Baumann et al., 2013, 2015). For mesospheric ice particles and MSPs, the typical charge state varies around  $(0, -1e, -2e)$  (Sodha et al., 2011). Since the dust/ice charging is independent of the origin of the particles in the MLT region, the charge state of ASPs will vary similarly to that of MSPs. Overall, the mesospheric ice and dust particles in the MLT usually are believed to have a charge of  $-1e$  (Rapp and Lübken, 2001). And a large fraction of nanometer-sized dust particles also remains neutral. Depending on mechanisms such as electron bite-out or dominant photoemission from dirty ice, ice and dust particles may also acquire positive charges.

Apart from the surface charge, the electrostatic interactions between charged spherical ice particles and oxides depend on their particle size and dielectric constants. To investigate the electrostatics, collision dynamics, and coalescence of ice and space debris, we consider charged ice particles with radii ranging from 1 to 100 nm and various oxides with radii ranging from 1 to 5 nm. Although the actual ASP size distribution is currently unknown, we assume various representative oxides with radii ranging from 1 to 5 nm. Aluminum being the most dominant, Al<sub>2</sub>O<sub>3</sub> has been selected for this study. TiO<sub>2</sub> has been selected for its very high dielectric constant compared to other compounds found in space debris. We also select three other oxides, FeO, MgO, and SiO<sub>2</sub>, because Fe, Mg, and Si are the next most common elements in space debris. However, these three elements are also dominant in sources from natural influx. Note that the study of all individual oxides is computationally expensive, so we restrict our study to these five oxides. To study particle growth, collision outcomes, and the subsequent formation of a stable aggregate, it is necessary to consider the surface charges, dielectric constants, and mass densities of the two colliding particles. The dielectric constants and mass densities are given in Table 1. The selected parametric regime in this study encompasses the dielectric constants and mass densities expected for various oxides related to space debris. Figure 1 shows an example of an electrostatic interaction energy curve between like-charged spherical ice and SiO<sub>2</sub>. The distinct features are: i) at large separations, there is no interaction. ii) As the separation decreases to a certain amount, they will feel a repulsive Coulomb barrier (hill) ( $E_{Coul}$ ). iii) At a very small separation (which may be considered as the point of contact), the electrostatic energy becomes negative ( $E_0 < 0$ ).  $E_0$  can be considered as the binding energy. Note that, depending on the particle sizes and dielectric constants,  $E_0$  can be negative, positive, or zero, but always less than or equal to the Coulomb



**Figure 1.** Illustration of the electrostatic interaction energy between a singly negatively charged ice particle ( $r_{H_2O} = 1$  nm) and a singly negatively charged  $SiO_2$  particle ( $r_{SiO_2} = 0.2$  nm) for varying separations derived using the model of Bichoutskaia et al. (2010).

barrier, i.e.,  $E_{Coul} \geq E_0$ . Because at a small separation, the polarization effect of one particle on the other reduces the repulsion by inducing opposite surface charge. For a system with one charged particle and another neutral, the Coulomb barrier is absent,  $E_{Coul} = 0$ , and an attractive well will be created between infinity with zero energy and at the point of contact with  $E_0 < 0$ .

The electrostatic interaction energy plays a crucial role in determining the outcome of particle collisions. It determines the effective velocity regime for the formation of a stable aggregate following a collision. Based on the Figure 1, during a collision event between two like-charged particles, the particles have to overcome the Coulomb hill ( $E_{Coul}$ ). So, the relative kinetic energy with respect to the center of mass must be higher than  $E_{Coul}$ . Therefore, the minimum relative initial velocity of the colliding particles required to overcome the Coulomb barrier is

$$u_{\min} = \sqrt{\frac{2eE_{Coul}}{\mu}}, \quad (1)$$

where  $\mu = \frac{m_{H_2O} m_{Oxide}}{m_{H_2O} + m_{Oxide}}$  is the reduced mass of the colliding ice (with mass  $m_{H_2O}$ ) and oxides (with mass  $m_{Oxide}$ ), and  $e$  is the electronic charge. After overcoming the Coulomb hill, the collision occurs. The kinetic energy just before the collision is

$$130 \quad KE_{\text{Before}} = \frac{1}{2} \mu u^2 - e E_0. \quad (2)$$

After the collision, the energy is reduced by

$$KE_{\text{After}} = k_r^2 \times KE_{\text{Before}}, \quad (3)$$

where  $k_r$  is the coefficient of restitution, which quantifies how much kinetic energy remains after a collision. It determines the elasticity of collision, which varies according to the composition of a particle. Following Baptiste et al. (2021), we have  
 135 taken  $k_r = 0.9$  in this study. If this excess energy  $KE_{\text{After}}$  is less than the barrier  $E_{\text{Coul}} - E_0$ , then coalescence is possible. Otherwise, the particles will move apart after the collision. Hence, the maximum velocity beyond which coalescence is not possible can be derived as

$$KE_{\text{After}} = k_r^2 \times KE_{\text{Before}} = e(E_{\text{Coul}} - E_0), \quad (4)$$

or

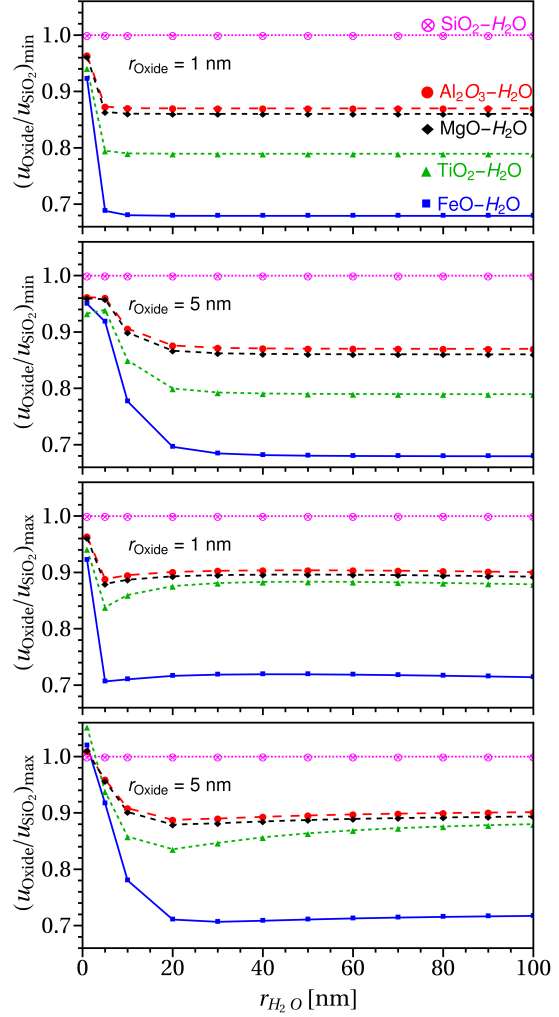
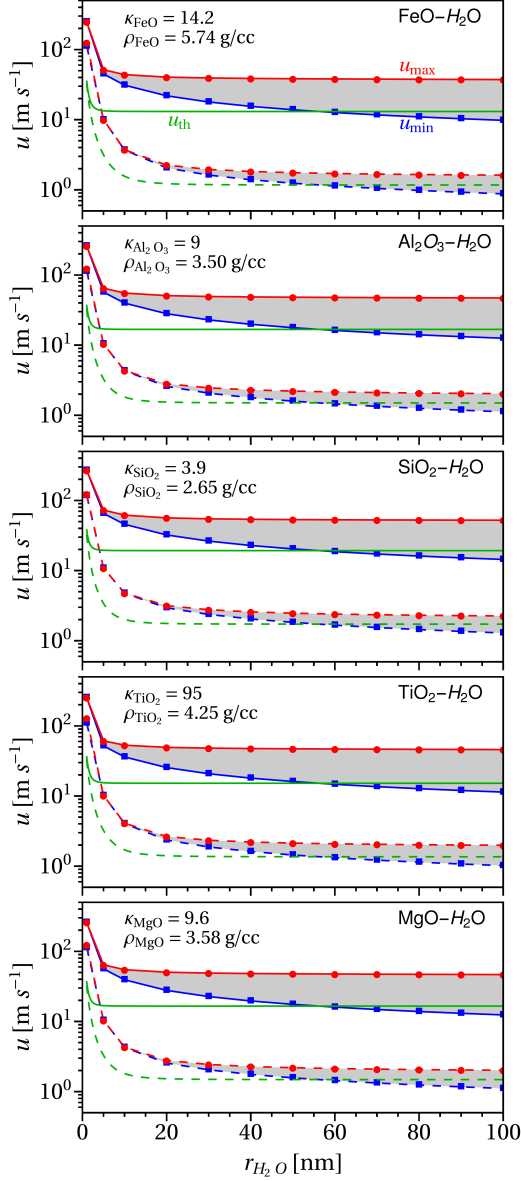
$$140 \quad u_{\text{max}} = \sqrt{\frac{2e[(E_{\text{Coul}} - E_0)/k_r^2 + E_0]}{\mu}}. \quad (5)$$

Therefore, particles with relative velocities between  $(u_{\text{min}}, u_{\text{max}})$  can form stable aggregates after collision. In the case of a collision between charged-neutral and oppositely charged particles, the minimum speed becomes zero.

Note that the aggregation of FeO, MgO, and SiO<sub>2</sub> with H<sub>2</sub>O ice has been discussed earlier by Baptiste et al. (2021). They highlighted that the specific location of the point charge on the particle's surface can influence the electrostatic interaction  
 145 between two like-charged particles. This influence is significant for sub-nanometer-sized particles. For particle sizes of  $r_{\text{Oxide}} \geq 1$  nm, we observe that the difference between the model outcomes of aggregation probability of Baptiste et al. (2021) and Bichoutskaia et al. (2010) is only about 5% in the case of H<sub>2</sub>O ice and metal oxide aggregation. Baptiste et al. (2021) conducted their analysis of the aggregation of like-charged FeO and MgO with ice up to  $r_{\text{H}_2\text{O}} = 30$  nm. In the present study, we are extending this analysis to  $r_{\text{H}_2\text{O}} = 100$  nm. Additionally, we have further examined the outcomes of collisions in scenarios  
 150 involving charged ice and neutral oxide aggregation.

### 3 Numerical Results

The framework presented in Bichoutskaia et al. (2010) has been utilized to derive the electrostatic interaction energy between charged ice and various oxides. The basic equations to derive  $E_{\text{Coul}}$  and  $E_0$  have been solved using the method explained in Brazil Lindgren (2017).  
 The basic equations and method to derive the electrostatic interaction energy have been briefly discussed in the Appendix

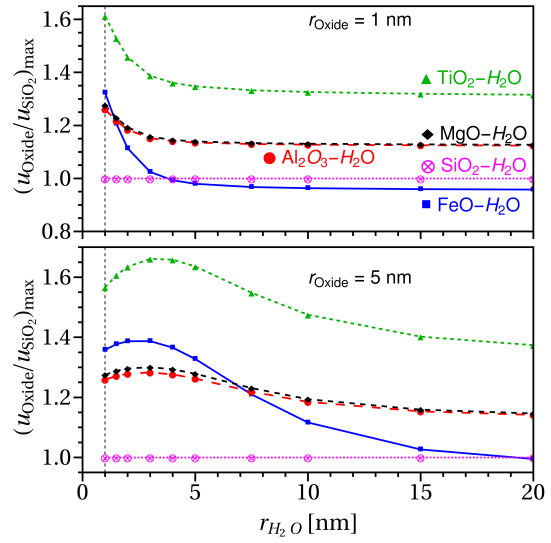
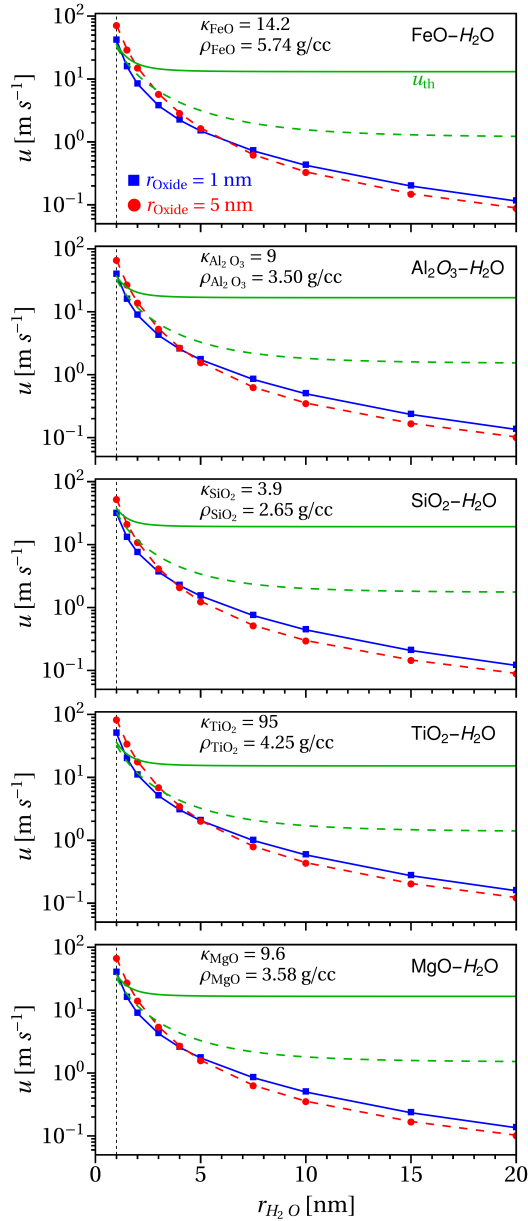


**Figure 2.** Left Panels: Variation of  $u_{\min}$  (blue),  $u_{\max}$  (red), and  $u_{th} = (2k_B T / \mu)^{1/2}$  (green) as a function of  $r_{H_2O}$ , needed for collision between ice and various metal oxides. The solid and dashed lines are for  $r_{\text{Oxide}} = 1$  nm and 5 nm, respectively. The colors are explained in the uppermost panel. Right Panels: Variation of  $u_{\min}$  and  $u_{\max}$  normalized with the same for  $\text{SiO}_2$  as a function of  $r_{H_2O}$ . Other parameter used:  $q_{H_2O} = q_{\text{Oxide}} = -1e$ ,  $\kappa_{H_2O} = 100$  and  $\rho_{H_2O} = 0.92$  g/cc.

155 [A and explained in detail in Brazil Lindgren \(2017\)](#). Thereafter  $u_{\min}$  and  $u_{\max}$  have been derived using Eqs. (1) and (5). Figure 2 illustrates the variation  $u_{\min}$  (blue) and  $u_{\max}$  (red) derived from the electrostatic interaction energy between a singly negatively charged ice particle and a singly negatively charged metal oxide. The variation of relative thermal speed  $u_{th} = (2k_B T / \mu)^{1/2}$  (green) is also shown for different  $r_{H_2O}$  and  $r_{Oxide}$ . In the mesospheric condition,  $T$  is taken to be 150 K following Baptiste et al. (2021). As  $u_{\min}$ ,  $u_{\max}$ , and  $u_{th}$  are inversely proportional to the square root of the reduced mass of the system, i.e.,  $(\propto 1/\sqrt{\mu})$ , and larger particles have higher masses, these velocities reduce with increasing radius of the ice and oxide particles. The estimated  $(u_{\min}, u_{\max})$  are around  $\sim (250, 250) \text{ m s}^{-1}$  and  $\sim (110, 125) \text{ m s}^{-1}$  for  $(r_{H_2O} = 1 \text{ nm}, r_{Oxide} = 1 \text{ nm})$  and  $(r_{H_2O} = 1 \text{ nm}, r_{Oxide} = 5 \text{ nm})$  respectively. While increasing the  $r_{H_2O}$  to 100 nm, the velocity range decreases to  $\sim (10, 40) \text{ m s}^{-1}$  and  $\sim (1, 2) \text{ m s}^{-1}$  for  $r_{Oxide} = 1 \text{ nm}$  and  $r_{Oxide} = 5 \text{ nm}$ , respectively. The detailed quantitative estimates of  $(u_{\min}, u_{\max})$  for different like-charged ice-ASP collisions are given in the Tables of Appendix A. It is further observed that the velocity interval, i.e.,  $u_{\max} - u_{\min}$ , increases with increasing  $r_{H_2O}$  for  $r_{H_2O} \geq r_{Oxide}$ . This can be attributed to the fact that charged smaller ice particles create significantly more Coulomb repulsion than the induced polarization effect on like-charged ASPs, due to their smaller available area for polarization. This results in  $|E_{Coul}| \geq |E_0|$  and  $E_{Coul}$  dominates in the estimates of  $(u_{\min}, u_{\max})$  for smaller  $r_{H_2O} \leq 10 \text{ nm}$ . Here  $|a|$  represents the magnitude of  $a$ . On the contrary, a higher polarization effect is created for larger  $r_{H_2O}$  leading to  $|E_0| \geq |E_{Coul}|$ , which makes  $E_0$  dominate in  $u_{\max}$ . Furthermore, the estimated  $u_{\min}$  and  $u_{\max}$  for  $r_{H_2O} \lesssim 55$  nm, are larger than the characteristic  $u_{th}$  of the particle population. This implies that the collision velocities exceeding the thermal regime are physically accessible for the growth process. However, for  $r_{H_2O} \gtrsim 55 \text{ nm}$ , the thermal population can contribute to the aggregation process due to  $u_{\min} \leq u_{th} < u_{\max}$ .

TiO<sub>2</sub> and SiO<sub>2</sub> exhibit the lowest and highest  $E_{Coul}$  values, respectively, because of their higher and lower dielectric constants, which influence the polarization effect. Additionally, at low density, SiO<sub>2</sub> has higher values of  $u_{\min}$  and  $u_{\max}$  (as shown in the third panel in Figure 2). On the contrary, FeO with a higher density has the lowest values of  $u_{\min}$  and  $u_{\max}$  (as shown in the top panel in Figure 2). Whereas with a higher dielectric constant but lower density, TiO<sub>2</sub> has the next lowest values of  $u_{\min}$  and  $u_{\max}$  (as shown in the fourth panel from the top in Figure 2). Lastly, with comparable values of dielectric constant and density, the  $u_{\min}$  and  $u_{\max}$  values are observed to be similar for Al<sub>2</sub>O<sub>3</sub> and MgO (as shown in the second and fifth panels from the top in Figure 2). For the case of fixed singly negative charge ice and oxides with  $(r_{H_2O} = 1 \text{ nm}, r_{Oxide} = 1 \text{ nm})$  and  $(r_{H_2O} = 5 \text{ nm}, r_{Oxide} = 5 \text{ nm})$ , we observed that  $E_{Coul} \approx E_0$  for all the metal oxides. For these cases, the polarization effect is not strong enough to create the attractive energy well. [Aggregation is not possible in these cases –when considering only electrostatic interactions.](#)

The  $u_{\max}$  for the case of charged ice and neutral metal oxide aggregation is shown in Figure 3. In this case,  $u_{\max}$  decreases approximately three orders of magnitude within  $r_{H_2O} = 20 \text{ nm}$ . Note that the charged ice particle induces polarization on the neutral oxides. The degree of polarization depends on the surface charge density on the ice. For a singly charged ice, the surface charge density decreases as  $\propto 1/r_{H_2O}^2$ , leading to a rapid decrease of  $E_0$  in magnitude. This results in a decrease in  $u_{\max}$  with increase in  $r_{H_2O}$ . For smaller values of  $r_{H_2O} < 5 \text{ nm}$ , we observe an increase in  $u_{\max}$  with increase in  $r_{Oxide}$ . An increase in  $r_{Oxide}$  increases the effective area on the neutral oxides, which results in a higher area available for polarization, leading to an increase in  $E_0$  and subsequently  $u_{\max}$ . However, for  $r_{H_2O} \geq 5 \text{ nm}$ , a slight decrease in  $u_{\max}$  is observed for larger  $r_{Oxide}$ . This



**Figure 3.** Left Panels: Variation of  $u_{\max}$  and  $u_{th}$  (green) as a function of  $r_{H_2O}$  needed for collision between charged ice and various neutral metal oxides. The solid and dashed lines are for  $r_{Oxide} = 1$  nm (blue) and 5 nm (red), respectively. The colors are explained in the uppermost panel. Right Panels: Variation of  $u_{\max}$  normalized with the same for  $SiO_2$  as a function of  $r_{H_2O}$ . Other parameter used:  $q_{H_2O} = -1e$ ,  $q_{Oxide} = 0$ ,  $q_{Oxide} = \theta k_{H_2O} = 100$  and  $\rho_{H_2O} = 0.92$  g/cc.

190 can be attributed to a smaller degree of polarization from larger ice particles to relatively heavier oxides, effectively lowering  $u_{\max}$ .  $\text{TiO}_2$  with the highest dielectric constant among the oxides used in this study exhibits the highest  $u_{\max}$  due to a higher degree of polarization. For smaller values of  $r_{H_2O}$ ,  $\text{SiO}_2$ , with the lowest value of dielectric constant, leads to the lowest  $u_{\max}$  values. However,  $\text{FeO}$  has the lowest  $u_{\max}$  for higher  $r_{H_2O}$  due to its higher mass density. Overall,  $u_{\max}$  ranges from few tens of  $\text{m s}^{-1}$  for  $r_{H_2O} < 5 \text{ nm}$  to few times of  $0.1 \text{ m s}^{-1}$  for  $r_{H_2O} \geq 5 \text{ nm}$ . The detailed quantitative estimates of  $u_{\max}$  for different charged-neutral ice-ASP collisions are given in the Tables of Appendix B.

### 3.1 Aggregation probability

Based on the estimated effective relative velocity between oxides (of anthropogenic origin) and ice particles, we can establish initial constraints on the likelihood of aggregation in the MLT region. However, the actual velocity distribution of these anthropogenic particulates in the MLT region is presently unknown. The velocity regime may depend on the deceleration histories and processes during re-entry to the MLT. If the particles are moving at a roughly constant speed within the range of ( $u_{\min}$ ,  $u_{\max}$ ), the aggregation probability will be 100%. However, for a first-order approximation, assuming these debris behave similarly to naturally generated MSPs, the effective velocity distribution may be represented by a Maxwell-Boltzmann distribution in the MLT temperature regime. Following Baptiste et al. (2021), the distribution can be written as

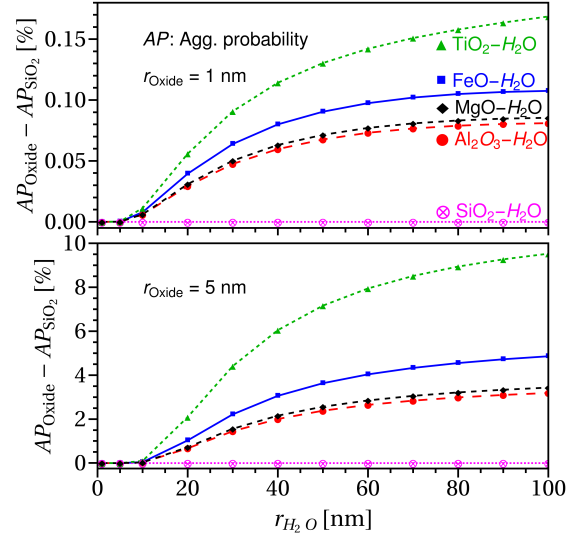
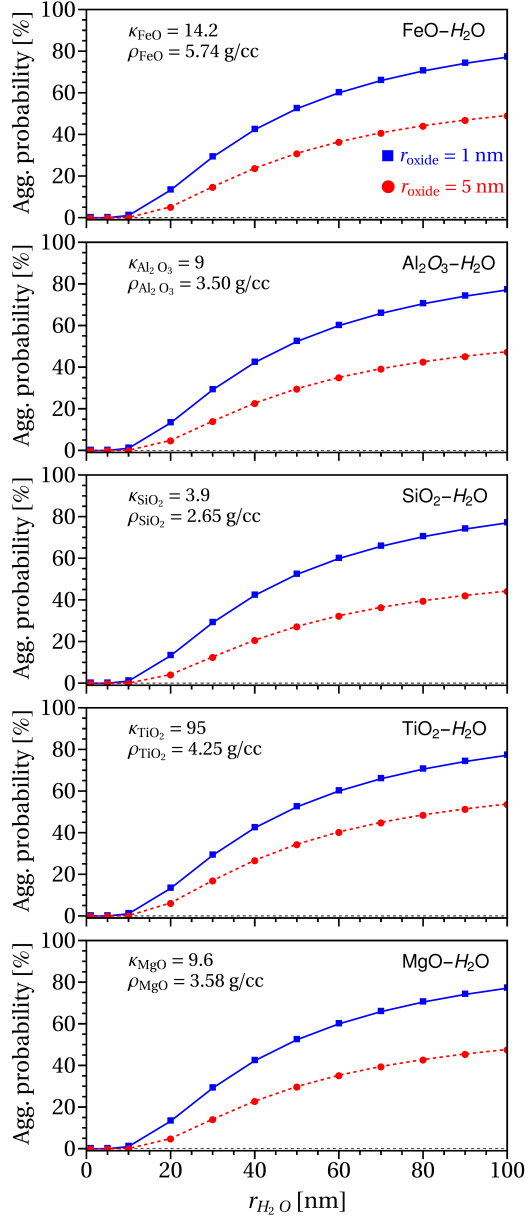
$$f(u) = \sqrt{\frac{2}{\pi}} \left( \frac{\mu}{k_B T} \right)^{3/2} u^2 \exp \left( -\frac{\mu u^2}{2k_B T} \right), \quad (6)$$

205 where  $k_B$  is the Boltzmann constant and  $T$  is the temperature in the MLT region. Then the aggregation probability can be written as:

$$\text{Agg. probability [\%]} = \int_{u_{\min}}^{u_{\max}} f(u) du. \quad (7)$$

It is observed from Figure 2 that for the like-charged ice-oxide aggregation case, for  $r_{H_2O} \lesssim 55 \text{ nm}$ ,  $u_{\min}$  and  $u_{\max}$  remain greater than  $u_{th}$ , suggesting only the high-energy Maxwellian tail of the distribution forms stable aggregates after collision. For  $r_{H_2O} \gtrsim 55 \text{ nm}$ ,  $u_{\min}$  becomes lower than  $u_{th}$ , indicating that a significant fraction of particles near the Maxwellian peak can actively participate in the collision process. Figure 4 illustrates the aggregation probability for like-charged ice-oxide collision, where  $r_{H_2O}$  increased to 100 nm for fixed  $r_{\text{Oxide}} = 1$  (blue) and 5 (red) nm. Similar to Baptiste et al. (2021), we also observed a one order of magnitude increase in the aggregation probability with an increase in  $r_{H_2O}$  from 10 to 100 nm. Our further observations revealed no significant differences in the aggregation probability profile across different ice-oxide aggregations. Figure 4 also highlights that the maximum aggregation probability decreases from 77% to approximately 50% upon an increase in  $r_{\text{Oxide}} = 1$  (blue) to 5 (red) nm for  $r_{H_2O} = 100 \text{ nm}$  due to the increase in  $m_{\text{Oxide}}$ .

As noted earlier, for the case of fixed singly negative charge ice and oxides with ( $r_{H_2O} = 1 \text{ nm}$ ,  $r_{\text{Oxide}} = 1 \text{ nm}$ ) and ( $r_{H_2O} = 5 \text{ nm}$ ,  $r_{\text{Oxide}} = 5 \text{ nm}$ ),  $E_{Coul} \approx E_0$  and aggregation is not possible in these cases when considering only electro-



**Figure 4.** Left Panels: Aggregation probability [%] for collision between ice ( $q_{H_2O} = -1e$ ) and different metal oxides ( $q_{Oxide} = -1e$ ).  $r_{H_2O}$  varies from 1 to 100 nm. The solid and dashed lines are for  $r_{Oxide} = 1$  nm (blue) and 5 nm (red), respectively. The colors are explained in the uppermost panel. Right Panels: Difference of Aggregation probability with the same for SiO<sub>2</sub> as a function of  $r_{H_2O}$ . Other parameter used:  $\kappa_{H_2O} = 100$  and  $\rho_{H_2O} = 0.92$  g/cc.

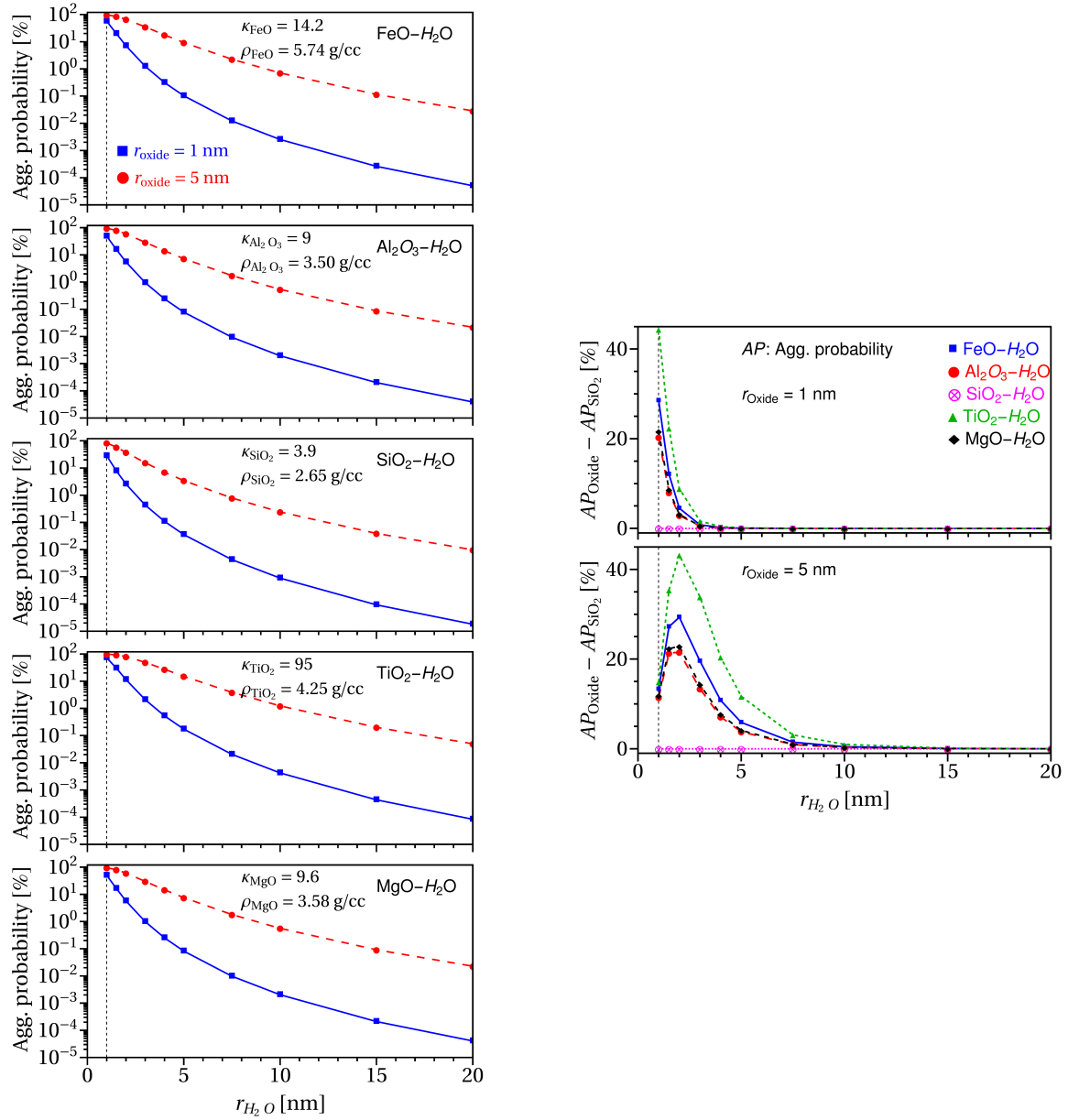
static interactions. Here, the van der Waals force is generally comparable to the electrostatic interaction at short separation distances of 0.1 nm to a few nanometers (Israelachvili, 2011; Filippov, 2022), leading to stronger binding near the point of contact, which may increase  $u_{\max}$ . The Van der Waals interaction may reduce the  $E_{Coul}$ . However, the reduced value of  $u_{\min}$  will remain significantly higher than  $u_{th}$ . As a result, the aggregation probability may not vary significantly, because most of the velocity regime of the Maxwellian distribution is not able to overcome the  $E_{Coul}$ . However, the Van der Waals Interactions may affect the aggregation scenario.

Furthermore, the contact electrification resulting from the collision may also contribute to the formation of a bound aggregate. Laboratory experiments suggested a typical charge transfer of the order of  $10^{-5}$  C m $^{-2}$  (Shen et al., 2016). With a typical collisional area  $A \sim \pi r_{Oxide}^2$ ,  $|Q| \sim 0.005e$  and  $0.123e$  for  $r_{Oxide} = 1$  nm and 5 nm, respectively, which is significantly low. Furthermore, collision event in the MLT region is not frequent enough for significant charge transfer via contact electrification. So for nanometer-sized particles, the charge transfer due to contact electrification is marginal in the MLT region. However, the contact electrification or triboelectric charging may become significant in the environment with significant collision frequency, where dust/particle density is large, such as Earth's lower atmospheric clouds and dusty atmospheres of Mars (Di Renzo and Urzay, 2018; Johnson et al., 2025).

As this study focuses on quantifying the effect of electrostatic interactions on particle collisions and aggregation, other interactions that could bring particles close together, such as van der Waals forces and contact electrification, are not considered, which could be taken up in future work.

The aggregation probability for charged ice-neutral oxide collision has been shown in Figure 5. For smaller  $r_{H_2O}$ , Figure 3 shows that  $u_{\max} \geq u_{th}$ , which leads to greater coverage of the Maxwell-Boltzmann distribution. However, a higher  $r_{H_2O}$  results in a significantly smaller  $u_{\max}$  compared to  $u_{th}$ , indicating that particles with high speed move apart after a collision. This results in a decrease in aggregation probability with increasing  $r_{H_2O}$ . For  $r_{Oxide} = 1$  and 5 nm, approximately seven and four orders of magnitude decrease is observed, respectively, while increasing  $r_{H_2O}$  from 1 to 20 nm. TiO $_2$  and SiO $_2$  exhibit the highest and lowest aggregation probability, respectively, because of their higher and lower dielectric constants. The probability of aggregation increases with the dielectric constants and radius of oxides due to the enhanced strength and effective area of the polarization effect, respectively. For a charged ice-neutral oxide collision, the Van der Waals force may influence the estimates of  $u_{\max}$ . This may lead to a different probability of aggregation. The detailed estimate of the collision dynamics considering both electrostatics and Van der Waals interaction will be taken up in future studies.

The calculation presented so far is for  $q_{H_2O} = -1e$ . For a like-charged case, increasing the charge from  $-1e$  to  $-2e$  increases the Coulomb barrier  $E_{Coul}$  and reduces the magnitude of  $E_0$ . As a result, it reduces the effective regime of  $(u_{\min}, u_{\max})$ , thereby reducing the probability of aggregation. For like-charged  $r_{H_2O} = 100$  nm,  $r_{Oxide} = 1$  nm, changing  $q_{H_2O} = -1e$  from  $-2e$ , reduces the aggregation probability from 77% to 46%. For the oppositely charged case,  $E_{Coul} = 0$ , and due to polarization effect,  $|E_0|$  is significantly stronger than the Coulomb attraction, which increases  $u_{\max}$ , leading to an enhancement of aggregation probability. As particle size increases with a fixed charge, surface charge density decreases. This reduction lowers the binding energy and the probability of aggregation. However, in the parametric regime used in the present analysis, the aggregation probability exceeds 99.5%. Overall, the collision outcome between



**Figure 5.** Left Panels: Aggregation probability [%] for collision between ice ( $q_{H_2O} = -1e$ ) and different neutral metal oxides ( $q_{Oxide} = 0$ ).  $r_{H_2O}$  varies from 1 to 20 nm. The solid and dashed lines are for  $r_{Oxide} = 1$  nm (blue) and 5 nm (red), respectively. The colors are explained in the uppermost panel. Right Panels: Difference of Aggregation probability with the same for SiO<sub>2</sub> as a function of  $r_{H_2O}$ . Other parameter used:  $\kappa_{H_2O} = 100$  and  $\rho_{H_2O} = 0.92$  g/cc.

255 ASPs and mesospheric ice particles depends on their surface charges, sizes, dielectric constants, and mass densities. The models of electrostatic interaction are independent of the origin of the ASPs or MSPs. Hence, ASPs exhibit aggregation behavior similar to that of the ice-MSP aggregation regime reported by Baptiste et al. (2021). The aggregation further depends on the concentrations of specific elements in the MLT region, along with the other parameters discussed above. The observed agreement suggests that this distinct anthropogenic particle population can be detected and systematically studied within the  
260 well-established naturally originated MSP framework.

## 4 Discussion

The present study investigates the electrostatic interaction between charged ice and anthropogenic smoke particles (ASPs) in the MLT region. This estimation is essential for evaluating the outcomes of collisions and subsequent aggregation of ice and ASPs. Like-charged ice and ASPs feel strong Coulomb repulsion at large separations. Additionally, at small separations, they  
265 feel a strong attraction due to the induced polarization of the particle's surface charge. Hence, the electrostatic interaction between ice and ASPs offers important insights into the minimum (to overcome the Coulomb barrier) and maximum (to form a stable aggregate after contact) relative velocities necessary for the formation of stable aggregates. For ice and ASPs in thermal motion at  $T = 150$  K, aggregation depends on particle size, mass density, and dielectric constant. In the case of a like-charged ice-ASP collision, the velocity range for growth varies from several tens of meters per second to a few meters per second. For  
270 ice particles of  $r_{H_2O} \lesssim 55$  nm, the high-energy tail of the Maxwellian distribution leads to the formation of stable aggregates. For larger particles, a significant fraction near the Maxwellian peak can actively contribute to collision and aggregation. As a result, for like-charged ice-ASP, the collision aggregation probability increases with increasing  $r_{H_2O}$ . On the contrary, the present study found that in a charged-neutral ice-ASP collision, the aggregation probability reduces to approximately four to seven orders of magnitude with increasing  $r_{H_2O}$  to 20 nm. Larger ice particles have a low surface charge density, which prevents  
275 them from creating a sufficient polarization effect at short distances from neutral ASPs. Consequently, these particles tend to move apart after a collision, leading to aggregation primarily involving the lower part of the Maxwellian distribution. However, very small ice particles (a few nm) have a higher surface charge density because of their smaller surface area. This increased charge density can generate significant binding energy, allowing them to form stable aggregates when their relative collision velocity is comparable to the thermal velocity,  $u_{th}$ . For charged-neutral ice-ASP collision, the maximum velocity ranges from  
280 a few tens of  $\text{m s}^{-1}$  for  $r_{H_2O} < 5$  nm to a few times  $0.1 \text{ m s}^{-1}$  for  $r_{H_2O} \geq 5$  nm. Therefore, the present study suggests that like-charged ice-ASP collisions are more favorable to form stable aggregates than charged-neutral ice-ASP collisions for larger ice particles.

The aggregation of ice and ASPs has significant implications for the formation of polar mesospheric clouds. These phenomena are of considerable interest as they serve as indicators of climate change. Therefore, understanding the composition, origin,  
285 and evolution of ASPs is essential for comprehending these processes. Natural influx from meteoric ablation is believed to be the dominant source of meteoric smoke particles (MSPs). However, a significant increase in atmospheric re-entry of space

debris may enhance the presence of ASPs. As a result, ASPs enriched in specific elemental species (such as  $\text{Al}_2\text{O}_3$  and  $\text{TiO}_2$ ) may exceed the abundance of their naturally occurring MSPs.

The present study suggests that for like-charged ice-ASP aggregation, there are no significant differences in the aggregation probability across different ice-oxide combinations. Therefore, particle growth will depend on the concentration of specific elemental species in the ASPs. In contrast, for the charged-neutral ice-ASP collision, the probability of ice- $\text{TiO}_2$  aggregation is observed to be higher due to the large dielectric constant of  $\text{TiO}_2$ , which creates a stronger polarization effect. Schulz et al. (2026) predicted that the post-ablation annual elemental mass injection of anthropogenic elements (especially Al and Ti) is going to be dominated over the natural meteoric injection in the near future. We suggest that  $\text{Al}_2\text{O}_3$  (due to its greater abundance) and  $\text{TiO}_2$  (due to its higher dielectric constant), both originating from anthropogenic sources, may dominate in the formation of ice-ASP aggregates. In the near future, based on the prediction of Schulz et al. (2026), we may expect a higher concentration of Al and Ti in the in-situ detected ASPs. The findings of this study are useful for future work on the impact of human space activities in the MLT region. Additionally, these results could be significant for sounding rocket experiments focused on detecting remnants of space debris in the MLT region.

## 300 Appendix A: Calculation of electrostatic interactions

The total electrostatic force between two dielectric spherical particles with dielectric constants ( $\kappa_1$  and  $\kappa_2$ ), radii ( $r_1$  and  $r_2$ ), charges ( $q_1$  and  $q_2$ ), and center-to-center separation,  $h$  can be written as (Bichoutskaia et al., 2010; Brazil Lindgren, 2017)

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$$F_{12} = K \frac{q_1 q_2}{h^2} + \left( -q_1 \sum_{m=1}^{\infty} \sum_{l=0}^{\infty} A_{1,l} \frac{(\kappa_2 - 1)m(m+1)}{(\kappa_2 + 1)m+1} \times \frac{(l+m)!}{l!m!} \frac{r_2^{2m+1}}{h^{2m+l+3}} - \frac{1}{K} \sum_{l=1}^{\infty} A_{1,l} A_{1,l+1} \frac{(\kappa_1 + 1)(l+1)+1}{(\kappa_1 - 1)r_1^{2l+3}} \right), \quad (\text{A1})$$

305 where  $K = 1/4\pi\epsilon_0$  and  $\epsilon_0$  is permittivity of free space. The first term of the above equation is Coulomb's Research at UiT was supported by the Research Council of Norway through grant number NFR 275503. TS and SKM acknowledge the support from the Department of Space, Government of India. TS is thankful to Prof. Elena Besley and Mr. Cameron Reeve for their suggestions regarding the calculation of electrostatic interaction energy-law. The second term (terms inside parentheses) is the polarization term derived from the multipole expansion of the electric potential, and  $A_{i,l}$  is the  $l$ th moment coefficient of particle  $i$ . The multipole moment coefficients can be derived from the following equations

$$4\pi K r_1 \sigma_{f,1} \delta_{l,0} = \frac{A_{1,l}}{r_1^{l+1}} + \frac{(\kappa_1 - 1)l}{(\kappa_1 + 1)l + 1} \sum_{m=0}^N A_{2,m} \frac{(l+m)!}{l!m!} \frac{r_1^l}{h^{l+m+1}}, \quad (\text{A2})$$

and

$$4\pi K r_2 \sigma_{f,2} \delta_{l,0} = \frac{A_{2,l}}{r_2^{l+1}} + \frac{(\kappa_2 - 1)l}{(\kappa_2 + 1)l + 1} \sum_{m=0}^N A_{1,m} \frac{(l+m)!}{l!m!} \frac{r_2^l}{h^{l+m+1}}. \quad (\text{A3})$$

Here,  $\sigma_{f,i} = q_i/4\pi r_i^2$  is the free surface charge density of the  $i$ th particle. The infinite series has been truncated at an upper bound  $N$ , yielding a convergent solution. Eqs. A2 and A3 can be converted to matrix representation of linear system:

$$\begin{pmatrix} C_1 & D_1 \\ D_2 & C_2 \end{pmatrix} \begin{pmatrix} A_1^N \\ A_2^N \end{pmatrix} = \begin{pmatrix} b_1 \\ b_2 \end{pmatrix}, \quad (\text{A4})$$

where  $A_i^N$  is the vector of dimension  $N$  with coefficient  $A_{i,l}$ . The detailed expression of  $C_i$  and  $D_i$  can be found in Brazil Lindgren (2017). The system of linear equations can be solved by iterative methods, and  $A_{i,l}$  can be derived. After deriving the moment coefficients for each  $h$  the electrostatic force given in Eq. A1 can be derived. The force can then be numerically integrated to derive the electrostatic interaction energy as a function of separation distance.

## Appendix B: ~~Appendix A:~~ Detail tabulated results of like-charged ice-ASP collisions

325  $E_{Coul}$ ,  $E_0$ ,  $u_{\min}$ ,  $u_{\max}$ , and the percentage of aggregation values for ice-ASP collisions for different oxides. The relevant dielectric constants and mass density values are taken from Table 1. Other parameters used:  $q_{H_2O} = q_{Oxide} = -1e$ ,  $T = 150$  K, and  $k_r = 0.9$ .

| $r_{FeO} = 1$ nm |                 |            |                              |                              |         |
|------------------|-----------------|------------|------------------------------|------------------------------|---------|
| $r_{H_2O}$ (nm)  | $E_{Coul}$ (eV) | $E_0$ (eV) | $u_{\min}$ m s <sup>-1</sup> | $u_{\max}$ m s <sup>-1</sup> | Agg(%)  |
| 1                | 0.6423          | 0.6423     | 248.7598                     | 248.7598                     | 0.0000  |
| 5                | 0.1447          | -0.0545    | 44.9713                      | 51.7234                      | 0.0053  |
| 10               | 0.0724          | -0.2338    | 31.1315                      | 43.9439                      | 1.0698  |
| 20               | 0.0362          | -0.3352    | 21.9535                      | 40.5233                      | 13.2876 |
| 30               | 0.0241          | -0.3705    | 17.9199                      | 39.4119                      | 29.1812 |
| 40               | 0.0181          | -0.3879    | 15.5179                      | 38.8381                      | 42.3481 |
| 50               | 0.0145          | -0.3978    | 13.8792                      | 38.4665                      | 52.3978 |
| 60               | 0.0121          | -0.4037    | 12.6697                      | 38.1890                      | 60.0286 |
| 70               | 0.0103          | -0.4072    | 11.7296                      | 37.9615                      | 65.9031 |
| 80               | 0.0090          | -0.4092    | 10.9719                      | 37.7627                      | 70.5071 |
| 90               | 0.0080          | -0.4101    | 10.3443                      | 37.5816                      | 74.1801 |
| 100              | 0.0072          | -0.4102    | 9.8134                       | 37.4118                      | 77.1587 |

| $r_{FeO} = 5$ nm |                 |            |                              |                              |         |
|------------------|-----------------|------------|------------------------------|------------------------------|---------|
| $r_{H_2O}$ (nm)  | $E_{Coul}$ (eV) | $E_0$ (eV) | $u_{\min}$ m s <sup>-1</sup> | $u_{\max}$ m s <sup>-1</sup> | Agg(%)  |
| 1                | 0.1502          | -0.0148    | 111.7681                     | 125.3414                     | 0.0034  |
| 5                | 0.1285          | 0.1285     | 9.9652                       | 9.9652                       | 0.0000  |
| 10               | 0.0722          | 0.0663     | 3.6980                       | 3.7331                       | 0.1024  |
| 20               | 0.0362          | 0.0047     | 2.0562                       | 2.2565                       | 5.2363  |
| 30               | 0.0241          | -0.0225    | 1.6256                       | 1.9597                       | 14.8717 |
| 40               | 0.0181          | -0.0375    | 1.3963                       | 1.8320                       | 23.8121 |
| 50               | 0.0145          | -0.0471    | 1.2452                       | 1.7599                       | 30.9591 |
| 60               | 0.0121          | -0.0537    | 1.1352                       | 1.7135                       | 36.5066 |
| 70               | 0.0103          | -0.0585    | 1.0503                       | 1.6809                       | 40.8240 |
| 80               | 0.0090          | -0.0621    | 0.9821                       | 1.6568                       | 44.2250 |
| 90               | 0.0080          | -0.0650    | 0.9257                       | 1.6382                       | 46.9430 |
| 100              | 0.0072          | -0.0674    | 0.8781                       | 1.6235                       | 49.1463 |

| $r_{\text{Al}_2\text{O}_3} = 1 \text{ nm}$ |                                |                    |                             |                             |         |
|--------------------------------------------|--------------------------------|--------------------|-----------------------------|-----------------------------|---------|
| $r_{\text{H}_2\text{O}} \text{ (nm)}$      | $E_{\text{Coul}} \text{ (eV)}$ | $E_0 \text{ (eV)}$ | $u_{\min} \text{ m s}^{-1}$ | $u_{\max} \text{ m s}^{-1}$ | Agg(%)  |
| 1                                          | 0.6445                         | 0.6445             | 259.9699                    | 259.9699                    | 0.0000  |
| 5                                          | 0.1447                         | -0.0394            | 57.0547                     | 65.0123                     | 0.0052  |
| 10                                         | 0.0724                         | -0.2158            | 39.8196                     | 55.3755                     | 1.0677  |
| 20                                         | 0.0362                         | -0.3163            | 28.1099                     | 50.9490                     | 13.2774 |
| 30                                         | 0.0241                         | -0.3518            | 22.9476                     | 49.5124                     | 29.1648 |
| 40                                         | 0.0181                         | -0.3697            | 19.8723                     | 48.7888                     | 42.3278 |
| 50                                         | 0.0145                         | -0.3802            | 17.7739                     | 48.3377                     | 52.3750 |
| 60                                         | 0.0121                         | -0.3868            | 16.2250                     | 48.0151                     | 60.0043 |
| 70                                         | 0.0103                         | -0.3911            | 15.0212                     | 47.7605                     | 65.8777 |
| 80                                         | 0.0090                         | -0.3939            | 14.0509                     | 47.5447                     | 70.4811 |
| 90                                         | 0.0080                         | -0.3956            | 13.2472                     | 47.3521                     | 74.1538 |
| 100                                        | 0.0072                         | -0.3966            | 12.5673                     | 47.1735                     | 77.1322 |

| $r_{\text{Al}_2\text{O}_3} = 5 \text{ nm}$ |                                |                    |                             |                             |         |
|--------------------------------------------|--------------------------------|--------------------|-----------------------------|-----------------------------|---------|
| $r_{\text{H}_2\text{O}} \text{ (nm)}$      | $E_{\text{Coul}} \text{ (eV)}$ | $E_0 \text{ (eV)}$ | $u_{\min} \text{ m s}^{-1}$ | $u_{\max} \text{ m s}^{-1}$ | Agg(%)  |
| 1                                          | 0.1538                         | 0.0211             | 113.1416                    | 124.0604                    | 0.0025  |
| 5                                          | 0.1289                         | 0.1289             | 10.4233                     | 10.4233                     | 0.0000  |
| 10                                         | 0.0722                         | 0.0673             | 4.3124                      | 4.3463                      | 0.0853  |
| 20                                         | 0.0362                         | 0.0074             | 2.5871                      | 2.8181                      | 4.8735  |
| 30                                         | 0.0241                         | -0.0192            | 2.0703                      | 2.4677                      | 14.1033 |
| 40                                         | 0.0181                         | -0.0339            | 1.7839                      | 2.3086                      | 22.7568 |
| 50                                         | 0.0145                         | -0.0434            | 1.5927                      | 2.2169                      | 29.7084 |
| 60                                         | 0.0121                         | -0.0499            | 1.4528                      | 2.1570                      | 35.1185 |
| 70                                         | 0.0103                         | -0.0546            | 1.3444                      | 2.1148                      | 39.3356 |
| 80                                         | 0.0090                         | -0.0583            | 1.2573                      | 2.0835                      | 42.6608 |
| 90                                         | 0.0080                         | -0.0611            | 1.1852                      | 2.0593                      | 45.3199 |
| 100                                        | 0.0072                         | -0.0635            | 1.1243                      | 2.0400                      | 47.4762 |

| $r_{\text{SiO}_2} = 1 \text{ nm}$ |                         |                    |                             |                             |         |
|-----------------------------------|-------------------------|--------------------|-----------------------------|-----------------------------|---------|
| $r_{H_2O} \text{ (nm)}$           | $E_{Coul} \text{ (eV)}$ | $E_0 \text{ (eV)}$ | $u_{\min} \text{ m s}^{-1}$ | $u_{\max} \text{ m s}^{-1}$ | Agg(%)  |
| 1                                 | 0.6501                  | 0.6501             | 269.6650                    | 269.6650                    | 0.0000  |
| 5                                 | 0.1447                  | -0.0124            | 65.3341                     | 73.1839                     | 0.0051  |
| 10                                | 0.0724                  | -0.1832            | 45.7412                     | 61.8524                     | 1.0618  |
| 20                                | 0.0362                  | -0.2822            | 32.3032                     | 56.5449                     | 13.2480 |
| 30                                | 0.0241                  | -0.3176            | 26.3719                     | 54.8287                     | 29.1172 |
| 40                                | 0.0181                  | -0.3357            | 22.8378                     | 53.9810                     | 42.2680 |
| 50                                | 0.0145                  | -0.3467            | 20.4264                     | 53.4732                     | 52.3073 |
| 60                                | 0.0121                  | -0.3540            | 18.6464                     | 53.1311                     | 59.9311 |
| 70                                | 0.0103                  | -0.3591            | 17.2630                     | 52.8803                     | 65.8010 |
| 80                                | 0.0090                  | -0.3628            | 16.1479                     | 52.6836                     | 70.4021 |
| 90                                | 0.0080                  | -0.3656            | 15.2242                     | 52.5202                     | 74.0734 |
| 100                               | 0.0072                  | -0.3676            | 14.4428                     | 52.3776                     | 77.0511 |

| $r_{\text{SiO}_2} = 5 \text{ nm}$ |                         |                    |                             |                             |         |
|-----------------------------------|-------------------------|--------------------|-----------------------------|-----------------------------|---------|
| $r_{H_2O} \text{ (nm)}$           | $E_{Coul} \text{ (eV)}$ | $E_0 \text{ (eV)}$ | $u_{\min} \text{ m s}^{-1}$ | $u_{\max} \text{ m s}^{-1}$ | Agg(%)  |
| 1                                 | 0.1662                  | 0.1013             | 117.6359                    | 122.9040                    | 0.0008  |
| 5                                 | 0.1316                  | 0.1300             | 10.8504                     | 10.8649                     | 0.0004  |
| 10                                | 0.0722                  | 0.0691             | 4.7597                      | 4.7841                      | 0.0561  |
| 20                                | 0.0362                  | 0.0122             | 2.9529                      | 3.1740                      | 4.2001  |
| 30                                | 0.0241                  | -0.0132            | 2.3743                      | 2.7723                      | 12.6509 |
| 40                                | 0.0181                  | -0.0276            | 2.0483                      | 2.5846                      | 20.7493 |
| 50                                | 0.0145                  | -0.0368            | 1.8296                      | 2.4752                      | 27.3228 |
| 60                                | 0.0121                  | -0.0431            | 1.6691                      | 2.4034                      | 32.4677 |
| 70                                | 0.0103                  | -0.0478            | 1.5448                      | 2.3528                      | 36.4918 |
| 80                                | 0.0090                  | -0.0514            | 1.4448                      | 2.3152                      | 39.6715 |
| 90                                | 0.0080                  | -0.0542            | 1.3620                      | 2.2861                      | 42.2176 |
| 100                               | 0.0072                  | -0.0565            | 1.2920                      | 2.2630                      | 44.2841 |

| $r_{\text{TiO}_2} = 1 \text{ nm}$ |                         |                    |                             |                             |         |
|-----------------------------------|-------------------------|--------------------|-----------------------------|-----------------------------|---------|
| $r_{H_2O} \text{ (nm)}$           | $E_{Coul} \text{ (eV)}$ | $E_0 \text{ (eV)}$ | $u_{\min} \text{ m s}^{-1}$ | $u_{\max} \text{ m s}^{-1}$ | Agg(%)  |
| 1                                 | 0.6381                  | 0.6381             | 253.8786                    | 253.8786                    | 0.0000  |
| 5                                 | 0.1447                  | -0.0993            | 51.9398                     | 61.3571                     | 0.0054  |
| 10                                | 0.0724                  | -0.2873            | 36.1503                     | 53.2014                     | 1.0732  |
| 20                                | 0.0362                  | -0.3911            | 25.5106                     | 49.5316                     | 13.3040 |
| 30                                | 0.0241                  | -0.4267            | 20.8250                     | 48.3192                     | 29.2081 |
| 40                                | 0.0181                  | -0.4439            | 18.0339                     | 47.6796                     | 42.3823 |
| 50                                | 0.0145                  | -0.4534            | 16.1296                     | 47.2559                     | 52.4378 |
| 60                                | 0.0121                  | -0.4590            | 14.7240                     | 46.9347                     | 60.0733 |
| 70                                | 0.0103                  | -0.4622            | 13.6316                     | 46.6702                     | 65.9520 |
| 80                                | 0.0090                  | -0.4639            | 12.7510                     | 46.4405                     | 70.5601 |
| 90                                | 0.0080                  | -0.4646            | 12.0216                     | 46.2339                     | 74.2371 |
| 100                               | 0.0072                  | -0.4647            | 11.4046                     | 46.0438                     | 77.2198 |

| $r_{\text{TiO}_2} = 5 \text{ nm}$ |                         |                    |                             |                             |         |
|-----------------------------------|-------------------------|--------------------|-----------------------------|-----------------------------|---------|
| $r_{H_2O} \text{ (nm)}$           | $E_{Coul} \text{ (eV)}$ | $E_0 \text{ (eV)}$ | $u_{\min} \text{ m s}^{-1}$ | $u_{\max} \text{ m s}^{-1}$ | Agg(%)  |
| 1                                 | 0.1448                  | -0.0959            | 109.7412                    | 129.3803                    | 0.0054  |
| 5                                 | 0.1277                  | 0.1277             | 10.1917                     | 10.1917                     | 0.0000  |
| 10                                | 0.0721                  | 0.0629             | 4.0450                      | 4.1056                      | 0.1575  |
| 20                                | 0.0362                  | -0.0041            | 2.3618                      | 2.6525                      | 6.2985  |
| 30                                | 0.0241                  | -0.0331            | 1.8823                      | 2.3481                      | 17.0682 |
| 40                                | 0.0181                  | -0.0489            | 1.6202                      | 2.2146                      | 26.8028 |
| 50                                | 0.0145                  | -0.0588            | 1.4460                      | 2.1385                      | 34.4904 |
| 60                                | 0.0121                  | -0.0656            | 1.3187                      | 2.0891                      | 40.4172 |
| 70                                | 0.0103                  | -0.0705            | 1.2203                      | 2.0542                      | 45.0100 |
| 80                                | 0.0090                  | -0.0742            | 1.1411                      | 2.0282                      | 48.6166 |
| 90                                | 0.0080                  | -0.0771            | 1.0757                      | 2.0081                      | 51.4910 |
| 100                               | 0.0072                  | -0.0795            | 1.0203                      | 1.9919                      | 53.8150 |

| $r_{\text{MgO}} = 1 \text{ nm}$ |                         |                    |                             |                             |         |
|---------------------------------|-------------------------|--------------------|-----------------------------|-----------------------------|---------|
| $r_{H_2O} \text{ (nm)}$         | $E_{Coul} \text{ (eV)}$ | $E_0 \text{ (eV)}$ | $u_{\min} \text{ m s}^{-1}$ | $u_{\max} \text{ m s}^{-1}$ | Agg(%)  |
| 1                               | 0.6441                  | 0.6441             | 259.2941                    | 259.2941                    | 0.0000  |
| 5                               | 0.1447                  | -0.0415            | 56.4326                     | 64.3901                     | 0.0052  |
| 10                              | 0.0724                  | -0.2184            | 39.3738                     | 54.8745                     | 1.0680  |
| 20                              | 0.0362                  | -0.3190            | 27.7942                     | 50.5113                     | 13.2790 |
| 30                              | 0.0241                  | -0.3545            | 22.6898                     | 49.0942                     | 29.1675 |
| 40                              | 0.0181                  | -0.3723            | 19.6490                     | 48.3782                     | 42.3312 |
| 50                              | 0.0145                  | -0.3828            | 17.5742                     | 47.9296                     | 52.3788 |
| 60                              | 0.0121                  | -0.3893            | 16.0427                     | 47.6068                     | 60.0083 |
| 70                              | 0.0103                  | -0.3935            | 14.8524                     | 47.3506                     | 65.8819 |
| 80                              | 0.0090                  | -0.3962            | 13.8930                     | 47.1322                     | 70.4854 |
| 90                              | 0.0080                  | -0.3978            | 13.0983                     | 46.9365                     | 74.1582 |
| 100                             | 0.0072                  | -0.3986            | 12.4261                     | 46.7546                     | 77.1366 |

| $r_{\text{MgO}} = 5 \text{ nm}$ |                         |                    |                             |                             |         |
|---------------------------------|-------------------------|--------------------|-----------------------------|-----------------------------|---------|
| $r_{H_2O} \text{ (nm)}$         | $E_{Coul} \text{ (eV)}$ | $E_0 \text{ (eV)}$ | $u_{\min} \text{ m s}^{-1}$ | $u_{\max} \text{ m s}^{-1}$ | Agg(%)  |
| 1                               | 0.1532                  | 0.0157             | 112.9156                    | 124.2325                    | 0.0026  |
| 5                               | 0.1289                  | 0.1289             | 10.3945                     | 10.3945                     | 0.0000  |
| 10                              | 0.0722                  | 0.0672             | 4.2795                      | 4.3142                      | 0.0877  |
| 20                              | 0.0362                  | 0.0070             | 2.5597                      | 2.7912                      | 4.9259  |
| 30                              | 0.0241                  | -0.0196            | 2.0474                      | 2.4445                      | 14.2150 |
| 40                              | 0.0181                  | -0.0345            | 1.7640                      | 2.2874                      | 22.9105 |
| 50                              | 0.0145                  | -0.0439            | 1.5749                      | 2.1969                      | 29.8908 |
| 60                              | 0.0121                  | -0.0504            | 1.4365                      | 2.1379                      | 35.3209 |
| 70                              | 0.0103                  | -0.0552            | 1.3294                      | 2.0963                      | 39.5527 |
| 80                              | 0.0090                  | -0.0588            | 1.2432                      | 2.0654                      | 42.8890 |
| 90                              | 0.0080                  | -0.0617            | 1.1719                      | 2.0416                      | 45.5567 |
| 100                             | 0.0072                  | -0.0640            | 1.1117                      | 2.0226                      | 47.7200 |

### Appendix C: ~~Appendix B:~~Detail tabulated results of charged-neutral ice-ASP collisions

$E_{Coul}$ ,  $E_0$ ,  $u_{\min}$ ,  $u_{\max}$ , and the percentage of aggregation values for ice-ASP collisions for different oxides. The relevant dielectric constants and mass density values are taken from Table 1. Other parameters used:  $q_{H_2O} = -1e$ ,  $q_{Oxide} = 0$ ,  $T = 150$  K, and  $k_r = 0.9$ .

| $r_{FeO} = 1 \text{ nm}$ |                         |                    |                             |                             |             |
|--------------------------|-------------------------|--------------------|-----------------------------|-----------------------------|-------------|
| $r_{H_2O} \text{ (nm)}$  | $E_{Coul} \text{ (eV)}$ | $E_0 \text{ (eV)}$ | $u_{\min} \text{ m s}^{-1}$ | $u_{\max} \text{ m s}^{-1}$ | Agg(%)      |
| 1                        | 0                       | -0.07594215        | 0                           | 41.42797192                 | 56.88367944 |
| 1.5                      | 0                       | -0.02767035        | 0                           | 15.68680281                 | 19.95644263 |
| 2                        | 0                       | -0.01249221        | 0                           | 8.33154154                  | 7.09087004  |
| 3                        | 0                       | -0.00367904        | 0                           | 3.76027350                  | 1.24531246  |
| 4                        | 0                       | -0.00145103        | 0                           | 2.22969925                  | 0.31597967  |
| 5                        | 0                       | -0.00068381        | 0                           | 1.49711227                  | 0.10307867  |
| 7.5                      | 0                       | -0.00016461        | 0                           | 0.72214555                  | 0.01224326  |
| 10                       | 0                       | -0.00005779        | 0                           | 0.42607981                  | 0.00254984  |
| 15                       | 0                       | -0.00001271        | 0                           | 0.19938935                  | 0.00026315  |
| 20                       | 0                       | -0.00000425        | 0                           | 0.11522100                  | 0.00005087  |

| $r_{FeO} = 5 \text{ nm}$ |                         |                    |                             |                             |             |
|--------------------------|-------------------------|--------------------|-----------------------------|-----------------------------|-------------|
| $r_{H_2O} \text{ (nm)}$  | $E_{Coul} \text{ (eV)}$ | $E_0 \text{ (eV)}$ | $u_{\min} \text{ m s}^{-1}$ | $u_{\max} \text{ m s}^{-1}$ | Agg(%)      |
| 1                        | 0                       | -0.26778015        | 0                           | 72.26674688                 | 97.88024662 |
| 1.5                      | 0                       | -0.15004748        | 0                           | 29.49077690                 | 85.77882537 |
| 2                        | 0                       | -0.09473403        | 0                           | 15.26494270                 | 67.07380653 |
| 3                        | 0                       | -0.04557675        | 0                           | 5.83245386                  | 35.24975139 |
| 4                        | 0                       | -0.02530380        | 0                           | 2.88668998                  | 17.88114770 |
| 5                        | 0                       | -0.01534226        | 0                           | 1.66548962                  | 9.36488041  |
| 7.5                      | 0                       | -0.00557969        | 0                           | 0.63005319                  | 2.27875730  |
| 10                       | 0                       | -0.00251636        | 0                           | 0.33445444                  | 0.71333298  |
| 15                       | 0                       | -0.00074021        | 0                           | 0.15086028                  | 0.11601937  |
| 20                       | 0                       | -0.00029176        | 0                           | 0.08942654                  | 0.02885039  |

| $r_{\text{Al}_2\text{O}_3} = 1 \text{ nm}$ |                                |                    |                                   |                                   |             |
|--------------------------------------------|--------------------------------|--------------------|-----------------------------------|-----------------------------------|-------------|
| $r_{\text{H}_2\text{O}} \text{ (nm)}$      | $E_{\text{Coul}} \text{ (eV)}$ | $E_0 \text{ (eV)}$ | $u_{\text{min}} \text{ m s}^{-1}$ | $u_{\text{max}} \text{ m s}^{-1}$ | Agg(%)      |
| 1                                          | 0                              | -0.06330082        | 0                                 | 39.45961588                       | 48.65964667 |
| 1.5                                        | 0                              | -0.02299548        | 0                                 | 15.82551315                       | 15.86466799 |
| 2                                          | 0                              | -0.01037888        | 0                                 | 8.85486907                        | 5.49176598  |
| 3                                          | 0                              | -0.00306081        | 0                                 | 4.22837430                        | 0.95133035  |
| 4                                          | 0                              | -0.00120911        | 0                                 | 2.56096110                        | 0.24098255  |
| 5                                          | 0                              | -0.00057056        | 0                                 | 1.73497605                        | 0.07866000  |
| 7.5                                        | 0                              | -0.00013766        | 0                                 | 0.84331165                        | 0.00936621  |
| 10                                         | 0                              | -0.00004840        | 0                                 | 0.49874202                        | 0.00195445  |
| 15                                         | 0                              | -0.00001066        | 0                                 | 0.23378085                        | 0.00020218  |
| 20                                         | 0                              | -0.00000357        | 0                                 | 0.13518419                        | 0.00003913  |

| $r_{\text{Al}_2\text{O}_3} = 5 \text{ nm}$ |                                |                    |                                   |                                   |             |
|--------------------------------------------|--------------------------------|--------------------|-----------------------------------|-----------------------------------|-------------|
| $r_{\text{H}_2\text{O}} \text{ (nm)}$      | $E_{\text{Coul}} \text{ (eV)}$ | $E_0 \text{ (eV)}$ | $u_{\text{min}} \text{ m s}^{-1}$ | $u_{\text{max}} \text{ m s}^{-1}$ | Agg(%)      |
| 1                                          | 0                              | -0.22987472        | 0                                 | 66.98426860                       | 96.04389737 |
| 1.5                                        | 0                              | -0.12759959        | 0                                 | 27.23293762                       | 79.87600958 |
| 2                                          | 0                              | -0.08001931        | 0                                 | 14.07492611                       | 59.29591781 |
| 3                                          | 0                              | -0.03816322        | 0                                 | 5.39390656                        | 29.06784819 |
| 4                                          | 0                              | -0.02108362        | 0                                 | 2.69818508                        | 14.20686985 |
| 5                                          | 0                              | -0.01274786        | 0                                 | 1.58384176                        | 7.28988493  |
| 7.5                                        | 0                              | -0.00462459        | 0                                 | 0.63477234                        | 1.73722727  |
| 10                                         | 0                              | -0.00208567        | 0                                 | 0.35503769                        | 0.54078703  |
| 15                                         | 0                              | -0.00061456        | 0                                 | 0.16946542                        | 0.08788861  |
| 20                                         | 0                              | -0.00024266        | 0                                 | 0.10261572                        | 0.02189488  |

| $r_{\text{SiO}_2} = 1 \text{ nm}$     |                                |                    |                                   |                                   |             |
|---------------------------------------|--------------------------------|--------------------|-----------------------------------|-----------------------------------|-------------|
| $r_{\text{H}_2\text{O}} \text{ (nm)}$ | $E_{\text{Coul}} \text{ (eV)}$ | $E_0 \text{ (eV)}$ | $u_{\text{min}} \text{ m s}^{-1}$ | $u_{\text{max}} \text{ m s}^{-1}$ | Agg(%)      |
| 1                                     | 0                              | -0.03731667        | 0                                 | 31.29202590                       | 28.34513821 |
| 1.5                                   | 0                              | -0.01347942        | 0                                 | 12.99778768                       | 7.86520505  |
| 2                                     | 0                              | -0.00608175        | 0                                 | 7.47884257                        | 2.57918406  |
| 3                                     | 0                              | -0.00179882        | 0                                 | 3.66900005                        | 0.43450102  |
| 4                                     | 0                              | -0.00071294        | 0                                 | 2.24454265                        | 0.10969937  |
| 5                                     | 0                              | -0.00033735        | 0                                 | 1.52767475                        | 0.03585282  |
| 7.5                                   | 0                              | -0.00008178        | 0                                 | 0.74619765                        | 0.00429148  |
| 10                                    | 0                              | -0.00002884        | 0                                 | 0.44223747                        | 0.00089913  |
| 15                                    | 0                              | -0.00000638        | 0                                 | 0.20772245                        | 0.00009348  |
| 20                                    | 0                              | -0.00000214        | 0                                 | 0.12024451                        | 0.00001815  |

| $r_{\text{SiO}_2} = 5 \text{ nm}$     |                                |                    |                                   |                                   |             |
|---------------------------------------|--------------------------------|--------------------|-----------------------------------|-----------------------------------|-------------|
| $r_{\text{H}_2\text{O}} \text{ (nm)}$ | $E_{\text{Coul}} \text{ (eV)}$ | $E_0 \text{ (eV)}$ | $u_{\text{min}} \text{ m s}^{-1}$ | $u_{\text{max}} \text{ m s}^{-1}$ | Agg(%)      |
| 1                                     | 0                              | -0.14481751        | 0                                 | 53.18435026                       | 84.57081081 |
| 1.5                                   | 0                              | -0.07872104        | 0                                 | 21.41438244                       | 58.54029135 |
| 2                                     | 0                              | -0.04865465        | 0                                 | 11.00424354                       | 37.72658280 |
| 3                                     | 0                              | -0.02278349        | 0                                 | 4.20340502                        | 15.68017381 |
| 4                                     | 0                              | -0.01245966        | 0                                 | 2.11329906                        | 7.06561462  |
| 5                                     | 0                              | -0.00749072        | 0                                 | 1.25397552                        | 3.47266274  |
| 7.5                                   | 0                              | -0.00270356        | 0                                 | 0.52065134                        | 0.79279030  |
| 10                                    | 0                              | -0.00121926        | 0                                 | 0.29951084                        | 0.24399460  |
| 15                                    | 0                              | -0.00036044        | 0                                 | 0.14689812                        | 0.03958601  |
| 20                                    | 0                              | -0.00014282        | 0                                 | 0.08985411                        | 0.00989670  |

| $r_{\text{TiO}_2} = 1 \text{ nm}$     |                                |                    |                                   |                                   |             |
|---------------------------------------|--------------------------------|--------------------|-----------------------------------|-----------------------------------|-------------|
| $r_{\text{H}_2\text{O}} \text{ (nm)}$ | $E_{\text{Coul}} \text{ (eV)}$ | $E_0 \text{ (eV)}$ | $u_{\text{min}} \text{ m s}^{-1}$ | $u_{\text{max}} \text{ m s}^{-1}$ | Agg(%)      |
| 1                                     | 0                              | -0.10748237        | 0                                 | 50.46502358                       | 72.72521962 |
| 1.5                                   | 0                              | -0.03953333        | 0                                 | 19.87068131                       | 30.23094596 |
| 2                                     | 0                              | -0.01787784        | 0                                 | 10.90448837                       | 11.47041056 |
| 3                                     | 0                              | -0.00525197        | 0                                 | 5.09245862                        | 2.08831748  |
| 4                                     | 0                              | -0.00206398        | 0                                 | 3.05461675                        | 0.53249738  |
| 5                                     | 0                              | -0.00096964        | 0                                 | 2.05900229                        | 0.17351332  |
| 7.5                                   | 0                              | -0.00023215        | 0                                 | 0.99475814                        | 0.02048996  |
| 10                                    | 0                              | -0.00008123        | 0                                 | 0.58657321                        | 0.00424776  |
| 15                                    | 0                              | -0.00001780        | 0                                 | 0.27412007                        | 0.00043594  |
| 20                                    | 0                              | -0.00000594        | 0                                 | 0.15824600                        | 0.00008398  |

| $r_{\text{TiO}_2} = 5 \text{ nm}$     |                                |                    |                                   |                                   |             |
|---------------------------------------|--------------------------------|--------------------|-----------------------------------|-----------------------------------|-------------|
| $r_{\text{H}_2\text{O}} \text{ (nm)}$ | $E_{\text{Coul}} \text{ (eV)}$ | $E_0 \text{ (eV)}$ | $u_{\text{min}} \text{ m s}^{-1}$ | $u_{\text{max}} \text{ m s}^{-1}$ | Agg(%)      |
| 1                                     | 0                              | -0.35560100        | 0                                 | 83.29675421                       | 99.51300247 |
| 1.5                                   | 0                              | -0.20408700        | 0                                 | 34.41970968                       | 93.98301309 |
| 2                                     | 0                              | -0.13114600        | 0                                 | 17.99251268                       | 80.94056878 |
| 3                                     | 0                              | -0.06458872        | 0                                 | 6.98378075                        | 49.54957859 |
| 4                                     | 0                              | -0.03634800        | 0                                 | 3.50546588                        | 27.51463309 |
| 5                                     | 0                              | -0.02221110        | 0                                 | 2.05187900                        | 15.18278301 |
| 7.5                                   | 0                              | -0.00813572        | 0                                 | 0.80625787                        | 3.90367138  |
| 10                                    | 0                              | -0.00366938        | 0                                 | 0.44186457                        | 1.24053969  |
| 15                                    | 0                              | -0.00107452        | 0                                 | 0.20602418                        | 0.20218091  |
| 20                                    | 0                              | -0.00042151        | 0                                 | 0.12346793                        | 0.05002817  |

| $r_{\text{MgO}} = 1 \text{ nm}$ |                         |                    |                             |                             |             |
|---------------------------------|-------------------------|--------------------|-----------------------------|-----------------------------|-------------|
| $r_{H_2O} \text{ (nm)}$         | $E_{Coul} \text{ (eV)}$ | $E_0 \text{ (eV)}$ | $u_{\min} \text{ m s}^{-1}$ | $u_{\max} \text{ m s}^{-1}$ | Agg(%)      |
| 1                               | 0                       | -0.06518563        | 0                           | 39.94953690                 | 49.95803091 |
| 1.5                             | 0                       | -0.02369040        | 0                           | 15.97826451                 | 16.47040590 |
| 2                               | 0                       | -0.01069289        | 0                           | 8.91950404                  | 5.72371570  |
| 3                               | 0                       | -0.00315276        | 0                           | 4.24918170                  | 0.99353120  |
| 4                               | 0                       | -0.00124514        | 0                           | 2.57127964                  | 0.25173429  |
| 5                               | 0                       | -0.00058745        | 0                           | 1.74126524                  | 0.08216208  |
| 7.5                             | 0                       | -0.00014169        | 0                           | 0.84602386                  | 0.00977953  |
| 10                              | 0                       | -0.00004980        | 0                           | 0.50026372                  | 0.00204010  |
| 15                              | 0                       | -0.00001097        | 0                           | 0.23445830                  | 0.00021097  |
| 20                              | 0                       | -0.00000367        | 0                           | 0.13556562                  | 0.00004083  |

| $r_{\text{MgO}} = 5 \text{ nm}$ |                         |                    |                             |                             |             |
|---------------------------------|-------------------------|--------------------|-----------------------------|-----------------------------|-------------|
| $r_{H_2O} \text{ (nm)}$         | $E_{Coul} \text{ (eV)}$ | $E_0 \text{ (eV)}$ | $u_{\min} \text{ m s}^{-1}$ | $u_{\max} \text{ m s}^{-1}$ | Agg(%)      |
| 1                               | 0                       | -0.23565386        | 0                           | 67.81945757                 | 96.40095369 |
| 1.5                             | 0                       | -0.13099421        | 0                           | 27.59063615                 | 80.89601887 |
| 2                               | 0                       | -0.08223149        | 0                           | 14.26551627                 | 60.55685789 |
| 3                               | 0                       | -0.03926936        | 0                           | 5.46823176                  | 30.00751117 |
| 4                               | 0                       | -0.02171061        | 0                           | 2.73437990                  | 14.74891163 |
| 5                               | 0                       | -0.01313239        | 0                           | 1.60380908                  | 7.59124317  |
| 7.5                             | 0                       | -0.00476585        | 0                           | 0.64100071                  | 1.81466679  |
| 10                              | 0                       | -0.00214937        | 0                           | 0.35767946                  | 0.56536184  |
| 15                              | 0                       | -0.00063317        | 0                           | 0.17031988                  | 0.09189325  |
| 20                              | 0                       | -0.00024995        | 0                           | 0.10304051                  | 0.02288643  |

330 *Data availability.* The data supporting this study's findings are available within the article and Appendix B and C.

*Author contributions.* TS and IM conceptualized the study. TS implemented and performed the numerical analysis. All authors contributed to the analysis, discussion, writing, and editing of the manuscript and approved the final version for submission.

*Competing interests.* The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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