



1 Microbial Interactions with Dissolved Organic Matter (DOM) from
2 Immature Cretaceous Marine Black Shale: *Implications for subsurface*
3 *carbon flux and geothermal energy systems*

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10 **Abstract**

11 Microbial transformation of dissolved organic matter (DOM) is a central process in subsurface
12 carbon cycling, yet its long-term dynamics in shale environments remain poorly constrained. We
13 conducted an 810-day incubation experiment using DOM leached from immature Cretaceous
14 marine black shale to investigate fluid–rock–microbe interactions and their role in climate-active
15 gas generation. Liquid chromatography–organic carbon detection (LC-OCD) revealed
16 progressive microbial degradation of labile DOM pools and the accumulation of recalcitrant
17 fractions, accompanied by sustained CO₂ and CH₄ release. These findings demonstrate that shale
18 DOM can fuel persistent microbial metabolism, restructure DOM composition, and contribute to
19 subsurface carbon fluxes over extended timescales. Our results highlight the ecological
20 significance of shale DOM biodegradation in unlocking deep carbon reservoirs and provide new
21 insights into the microbial mediation of subsurface biogeochemical cycles.

22 **1.0 Introduction**

23 The biogeochemical degradation and transformation of dissolved organic matter (DOM) are
24 critical for its contributions to the global carbon (C) cycle and the C flux in ecosystems
25 (Schlesinger and Andrews, 2000; Davidson and Janssens, 2006; Heimann and Reichstein, 2008).
26 Microorganisms are important intermediaries in the formation, mobilisation, transformation,
27 mineralisation, and storage of DOM in various environments, such as shale, soil, sediment, and
28 marine and freshwater (Ji et al., 2023; Young et al., 2004; Jiao et al., 2010; Lin et al., 2014; Zhu et
29 al., 2016; Wu et al., 2018; Spray et al., 2021). Understanding the dynamic interactions among
30 substrate/rock, DOM, and microbial activities is critical not only to delineate the routes of DOM
31 mobilisation, transformation, and flux in the surface and subsurface environments but also to
32 determine the release of climate-active gases (CO₂ and CH₄) from these processes. Identifying
33 DOM groups accessible to microbial activity and understanding microbial transformation and
34 mineralisation of DOM are central to this challenge. However, these processes and interactions



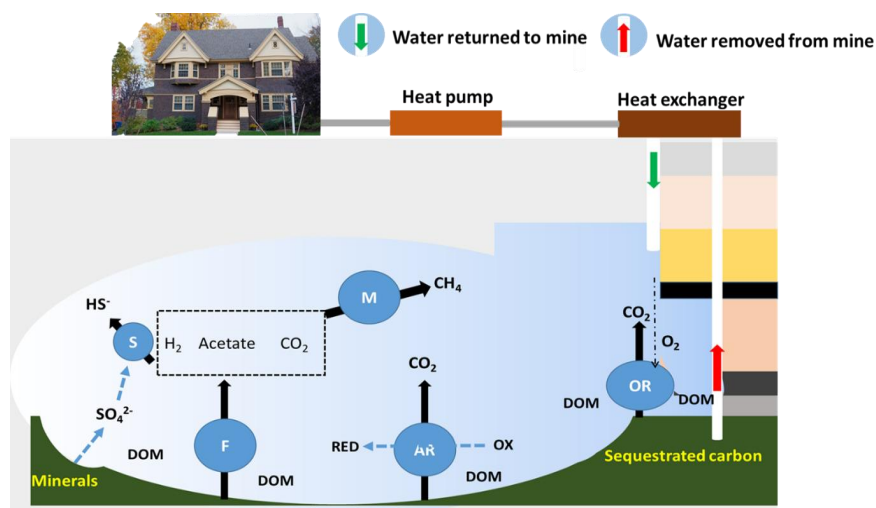
35 in natural and engineered subsurface environments are commonly studied independently (Wu et
36 al., 2018) and so lack a comprehensive and integrated understanding. This knowledge is pressing
37 as drilling activities for geothermal and unconventional energy using hydraulic fracturing are
38 increasing globally (Mouser et al. 2016), and are likely to stimulate biophysical-chemical
39 perturbations in the deep subsurface, with unknown consequences. A key component in this
40 process is the injection of fluids into the sub-surface, which are not sterile (Mouser et al. 2016).
41 Thereby, surface microbes and chemicals can be introduced into the subsurface while (re-
42)activating subsurface microbial communities that can exist up to 3–5 km in the deep subsurface
43 (Kadnikov et al., 2020). It is therefore possible, if not likely, that highly pressurised injection
44 fluids mobilise natural organic matter and other inorganic nutrients from the sub-surface rock,
45 thereby creating a complex mixture of in situ (autochthonous) and surface (allochthonous) DOM,
46 mineral, and microbial communities that can interact with sub-surface fluid flow. This
47 autochthonous source of DOM may be more biologically available, despite its lower
48 concentration compared to the larger and possibly more stable allochthonous DOM (Findlay and
49 Parr, 2017). Consequences for engineering are well known and include biogenic CO₂/CH₄
50 generation, reservoir/borehole souring leading to corrosion of the installed facilities, biofilm
51 formation from rapid microbial bloom growth, which may constrain fluid flow, as well as
52 borehole fouling (Ren et al., 2024; Vetter et al., 2012). This study aims to investigate whether the
53 leached aged OM from a subsurface rock could stimulate microbial activity leading to GHG
54 generation.

55 The deep subsurface hosts a large and diverse microbial community (Mouser et al., 2016).
56 Pioneering works in mines, boreholes/wells, and springs have confirmed that the deep subsurface
57 is a habitable setting rather than vast expanses of sterile rock (Colwell and D'Hondt, 2013). The
58 deep terrestrial (or continental) biosphere is structurally complex, resulting in heterogeneous
59 geological environments (Osburn et al., 2020). These environments and geological gradients with
60 their unique microbial habitats and sources of energy and carbon (such as DOM) are modified by
61 hydrological connectivity in aquifers, or, for engineering operations, fractured- (or drilled-) based
62 mixing (Osburn et al., 2020). Large open questions remain relating to the stability and
63 connectivity of these in-situ (subsurface) and introduced-terrestrial microbial taxa, a plethora of
64 complex DOM (both autochthonous and allochthonous), their resultant interactions in the sub-
65 surface, and their engineered and natural connectivity with the surface.

66 To understand these sub-surface processes and interactions, the UK government initiated the
67 United Kingdom Geoenergy Observatory Site Project (UKGEOS) to explore the deep subsurface
68 environment for geothermal energy production as part of its strategy towards decarbonisation of
69 energy by 2050 (HM Government, 2021). Geothermal energy is a promising alternative energy
70 source that may allow for faster, wider, and more efficient cleaner energy (EU (CORDIS), 2020).



71 However, the process of deep energy extraction involves perturbation of the deep subsurface,
 72 which may mobilise bedrock carbon and other elements (including nutrients and metals) and
 73 introduce or reactivate microbial taxa in the subsurface. These processes may potentially stimulate
 74 microbial activities producing greenhouse gases (GHG) in the sub-surface and leakage to the
 75 surface environment (Fig 1). These DOM-microbial interactions might also pose challenges to
 76 the exploitation and operation of geothermal boreholes by altering the chemical properties of the
 77 fluids through biodegradation products, corrosion, and pipeline clogging, leading to well fouling
 78 (Leins et al. 2022). Hence, it is critical to document and understand these deep subsurface
 79 interactions to de-risk geothermal energy production.



80

81 Figure 1: An ecological model of the deep subsurface aquifer for geothermal energy production, indicating
 82 the inflow and outflow of cold and hot water, resulting in the mobilisation and deposition of DOM due to
 83 temperature and pH differences, the introduction of O₂, and the mobilisation of sequestered C substrates
 84 and mineral nutrients, which may support microbial activity. Key: OR, aerobic respiration; F, fermentation;
 85 SR, sulphate reduction; AR, anaerobic respiration (other than SR); M, methanogenesis. (Modified from
 86 Kadnikov et al. 2020). Note: the model does not represent the real dimension/size of the aquifer, but a rather
 87 idealised cartoon of engineered mobilised DOM-microbial interactions in the subsurface aquifers and how
 88 it influences biogeochemical changes and generation of climate active gases (CO₂/CH₄).

89 To contribute to these wider challenges, we conducted a set of long-term microcosm experiments
 90 using organic-carbon-rich immature Cretaceous source rocks as a substrate to test the potential of
 91 geologically old DOM to stimulate microbial activities, before exploring the more complex, OM-
 92 rich coal-bearing environment of Carboniferous rock lithologies recorded from the UKGEOS
 93 study site. This is critically needed to understand and better predict fluid-rock-microbial
 94 interaction in new geothermal systems, and the modelling and operation of such systems.

95 The DOM chemistry strongly affects the microbial community structure and their metabolic
 96 potential, as shown for marine (McCarren et al., 2010), soil (Ding et al., 2015), groundwater



97 environments (Zhang et al., 2015), and sediment leachates (Wu et al., 2018). Tracing and
98 investigating DOM changes with microbial activity is complex, as the change in microbial
99 community structure and function coupled with the contribution of microbial products (i.e.
100 bacterial lysates released from dead cells, extracellular metabolites, and microbially produced
101 aromatic amino acids e.g., phenylamine, tyrosine, tryptophane) to the DOM pool can affect the
102 transformation of DOM and contribute to changes in DOM composition over time (Goodwin et
103 al., 2015; Turnbull et al., 2016; Wu et al., 2018). These microbially derived products and
104 biodegradation residues are, for example, polysaccharides, proteins, and cell wall polymers,
105 together with several uncharacterized molecules that can become an integral component of OM
106 during the process (Kogel-Knabner, 2002; Kelleher and Simpson, 2006; Koch et al., 2014;
107 Osterholz et al., 2015; Kallenbach et al., 2016). Therefore, the bioavailability of C substrate in
108 DOM, microbial community composition, and the ability of microbes to metabolise DOM are
109 closely interconnected, but such interactions are not well documented for the subsurface
110 environment.

111 The DOM leached from rocks serves as the main source of C to groundwater (Aiken, 2002) and
112 contributes to the permanent groundwater DOC pool that is decoupled from seasonal shifts in
113 organic C content at shallower depths due to fluid flow and recharge (Awoyemi et al., 2014). The
114 highly pressurised drilling fluids induced by drilling and hydro-fracking facilitate the release of
115 DOM from rocks in the deep subsurface (Vieth-Hillebrand et al., 2017). The DOM signature and
116 function are linked to the kerogen type of the host rock and therefore may interact in different
117 ways with the autochthonous (groundwater/fluid-flow) and allochthonous (drilling fluid) origins
118 of microbial communities in the subsurface (Ding et al., 2015). Investigations that link rock DOM
119 turnover to microbial activities and biogenic CO₂/CH₄ production in subsurface settings are rare
120 (Wu et al., 2018). Therefore, in this study, the surface water microbes were exposed to water-
121 leached DOM from thermally immature Cretaceous marine black shale and biogenic CO₂ and
122 CH₄ production was monitored using a greenhouse gas analyser (GHG) during 810 days of
123 incubation. This approach will account for longer-term interaction effects and thereby
124 significantly extend previous studies (Horstmann, L. (2018); Wu et al., 2018) that focus on shorter
125 (2-month) incubation periods. To document the change in DOM composition before and after our
126 incubations, the LC-OCD was used. This research hypothesised that H1: Water-extracted DOM
127 may stimulate microbial activity and generate GHG. H2: The microbial activity will gradually
128 transform the black shale-derived DOM pool towards more recalcitrant DOM (less labile lipid
129 and lignin-like macro-molecules). H3: The amount and composition of DOM pools directly
130 dictate the amount of GHG (CO₂/CH₄) generated by microbial activity.



131 **2.0 Methodology**

132 **2.1 Sample collection**

133 **2.1.1 Cretaceous marine black shale sample**

134 Two Cretaceous black shale samples (A and B) were selected from study well S75 at the deep part
135 of the Tarfaya basin in southern Morocco, offshore North-West Africa. See Kolonic et al. (2002)
136 for further details on the location and geochemistry. The stratigraphic section represented in the
137 study interval belongs to the Cenomanian-Turonian transition (upper *Rotalipora cushmani* to
138 middle *H. Helvetica biozone*) of the Oceanic Anoxic Event 2. Samples of 2 and 5 cm thickness
139 were taken for this study. The two samples were selected for the experiment based on their OM
140 richness, kerogen type, and thermal immaturity. Sample A was obtained from 60 cm well depth,
141 while sample B was collected from 120 cm, both representing high TOC palaeo-shelf marine
142 environments (Kuhnt et al., 1990; 1997). In this study, we used elemental and Rock-Eval pyrolysis
143 data obtained by Kolonic et al. (2002). The total organic carbon content of samples A and B is 7.5
144 and 10.5 (w %), respectively. The OM in the samples is thermally immature (Tmax; 411 °C and
145 407 °C, respectively) with pyrolytic kerogen yields (S2) of 55.23 and 83.3 mg HC/g rock,
146 respectively. Both samples have very high Hydrogen Indices (HI; 733 and 795 mg HC/g TOC,
147 respectively) and low Oxygen Index (OI; 53 and 43 mg CO₂/g TOC, respectively), qualifying the
148 kerogen as marine/microbial type I to IIs.

149 **2.1.2 Inoculum – Freshwater sample**

150 The microbial community used as inoculum for the microcosm experiment of this study was
151 obtained from the Black Burn River, Auchencorth Moss field, located in Southeast Scotland
152 (55°48'21.5" N; 3°13'36.0" W). The catchment has a long-term data set relating both terrestrial
153 and aquatic carbon species, which includes a complete, inter-annual C and GHG budget from
154 2008 onwards (Dinsmore et al. 2010). The vegetation that covers the main body of the peatland
155 consists of grasses, rushes, and sedges (Drewer et al. 2010). The peat catchment is drained by the
156 Black Burn, a small stream with a width of ~1 m and a depth from peat surface to channel bed of
157 ~1.2 m. The Burn River is characterised by low pH (~3-6), low-temperature conditions, and a
158 dynamic hydrological response to rainfall events (Billett and Harvey, 2013; Dinsmore and Billett,
159 2008). Stream water is commonly highly coloured and mean long-term DOC concentrations are
160 28.4 ± 1.07 mg L⁻¹ (Dinsmore et al. 2013).

161 The freshwater sample was obtained in a 10 L High-Density Polyethylene (HDPE) carboy one
162 day before incubation and stored at room temperature. The carboy and 1L HDPE wide-neck bottle
163 were pre-cleaned with 18.2 MΩ cm⁻¹ decarbonised water before the sampling. Both the carboy
164 and bottle were rinsed three times with river freshwater before sampling and were sampled from



165 mid to the bottom of the river (the soil from the bottom of the river was also collected with the
166 water sample).

167 **2.2 Extraction of DOM from the Cretaceous shale samples**

168 The PTFE bottles (50 mL) used for leaching of the shale samples were pre-cleaned by soaking in
169 a 10% HCl/18.2 M Ωcm^{-1} water (v/v) for 30 minutes, transferred to an 18.2 M Ωcm^{-1} water bath,
170 and then rinsed with 18.2 M Ωcm^{-1} water and dried up on a drying rag covered with ashed
171 aluminium foil. The ground shale samples (A and B; three 3g each) were measured into pre-
172 cleaned Teflon tubes, and 37.5 mL 18.2 M Ωcm^{-1} decarbonised water (1:12.5 wt./v) was added
173 and placed in a shaking incubator, at a temperature of 50 °C for 48 hours (Zhu et al. 2015). The
174 samples were filtered with GF/F membrane filters, diameter 70 mm, pore size 0.7 μm (Whatman)
175 in a 250 ml glass bottle and rinsed with an additional 50 mL 18.2 M Ωcm^{-1} decarbonised water.
176 The samples were immediately re-filtered using a Polyether sulfone (PES) filter 0.45 μm for
177 injection to LC–OCD, which is suitable for DOM analysis (Karanfil et al., 2002). The resulting
178 filtrates were analysed using LC-OCD (according to Huber et al. (2011) and Zhu et al (2015)) to
179 identify different DOM fractions. The post-incubation inoculum samples were also filtered with
180 a 0.45 μm PES filter before LC-OCD analyses to determine the biotransformation of DOM at the
181 end of the incubation.

182 **2.3 Incubation of surface water microbes with black shale-derived DOM**

183 **2.3.1 Pre-incubation preparation**

184 The microcosms were set up in 100 mL glass serum bottles. All the bottles were cleaned and
185 rinsed with 18.2 M Ωcm^{-1} decarbonised water before being ashed in a furnace at 450 °C for 24
186 hours to avoid any contamination from residual organic carbon. The clean bottles were autoclaved
187 before use.

188 **2.3.2 Preparation of organic and inorganic nutrients**

189 The inorganic nutrient mix was prepared by adding 100 mL of 4.67 mM NH_4Cl , 100 mL of 1.47
190 mM KH_2PO_4 , 5 mL of non-chelated trace elements, and 5 mL of selenite-tungstate (SeW) solution
191 in a 250 mL beaker on a magnetic stirrer at 300 rpm.

192 The non-chelated trace element solution was prepared according to Widdel et al. (1983).
193 Deionised water (987 mL) was measured into a 1L ashed glass bottle on a hotplate magnetic stirrer
194 at 300 rpm and 12.5 mL of 25% HCl (100 mM), $\text{FeSO}_4 \cdot 7\text{H}_2\text{O}$ (7.5 mM), H_3BO_3 (0.5 mM),
195 $\text{MnCl}_2 \cdot 4\text{H}_2\text{O}$ (0.5 mM), $\text{CoCl}_2 \cdot 6\text{H}_2\text{O}$ (0.8 mM), $\text{NiCl}_2 \cdot 6\text{H}_2\text{O}$ (0.1 mM), $\text{CuCl}_2 \cdot 2\text{H}_2\text{O}$ (0.01 mM),
196 $\text{ZnSO}_4 \cdot 7\text{H}_2\text{O}$ (0.5 mM) and $\text{Na}_2\text{MoO}_4 \cdot 2\text{H}_2\text{O}$ (0.15 mM) were sequentially added. The trace
197 element mixture was autoclaved in bottles tightly closed with rubber-fitted screw caps or fixed
198 stoppers; a headspace of approximately 1/3 of the volume was left.



199 The SeW solution was prepared as reported by Widdel and Bak (1992). The NaOH (10 mM) was
 200 weighed into a 1 L bottle containing 1 L DH₂O on a hotplate magnetic stirrer stirring at 300 rpm,
 201 and added Na₂SeO₃ · 5H₂O (0.02 mM) and Na₂WO₄ · 2H₂O (0.02 mM) with constant stirring. The
 202 solution was then autoclaved.

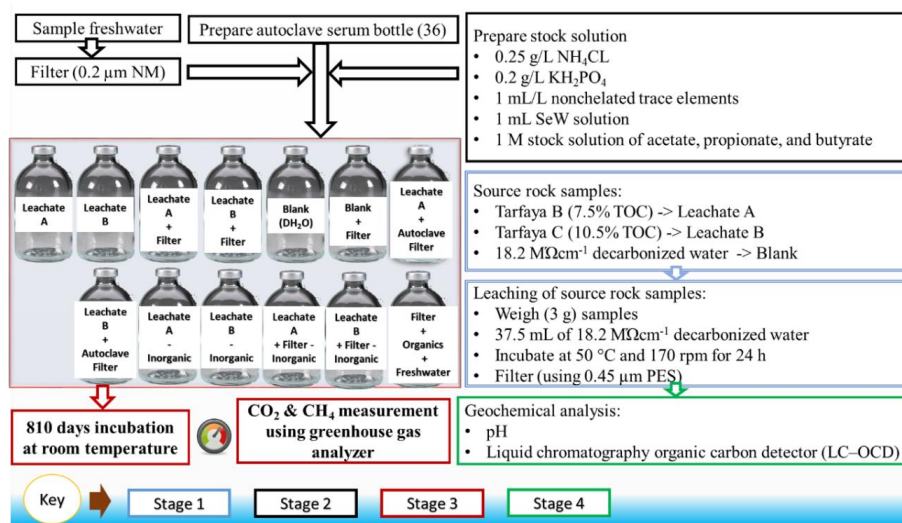
203 The NH₄Cl (4.67 mM) and KH₂PO₄ (1.47 mM) were also prepared as reported by Widdel and
 204 Bak (1992). The solutions were autoclaved.

205 2.3.3 Filtration of surface water sample and concentration of microbes

206 The surface water was first filtered using a vacuum pump mounted with a 0.7 μm GF/F, 47 mm
 207 diameter to remove the particulate organic matter. Microbes were subsequently collected on the
 208 0.2 μm nylon membrane (NM) filter by re-filtration of 0.7 μm GF/F filtrate. To ensure an
 209 approximately equal number of microbial cells were collected on each filter, an equal volume
 210 (100 mL of 0.7 μm GF/F filtrate) was used on each NM filter. The 0.2 μm NM filters were used
 211 as inoculum for the experiment.

212 2.3.4 Setting up the microcosm

213 The serum bottles were prepared and labelled as indicated in Fig 2 and Table 1:



214 Figure 2: Schematic overview of the experimental setup (Modified from Horstmann (2018)).

215 The microcosm set-up consisted of bottles with testing materials (leachate A and B) as a source
 216 of C substrate, plus the inoculum filter and six control groups, as summarised in Table 1. Each
 217 microcosm setup was prepared in triplicate to measure variation in the experiment so that
 218 statistical tests could be applied to evaluate differences and assess reproducibility.
 219



220 Filters (0.2 μm NM) were placed in all microcosm bottles except uninoculated controls.
 221 Autoclaved control bottles with wet filters were autoclaved before the addition of leachate. The
 222 10 mL of the given leachate was added to all microcosms except the blank and positive controls.
 223 The 10 mL of the blank (18.2 $\text{M}\Omega\text{cm}^{-1}$ decarbonised water) was added only to blank controls.
 224 The 4 mL inorganic nutrients mix was added to all microcosms except A-inorg, B-inorg, A + F-
 225 inorg and B + F-inorg, in which instead 4 mL of 18.2 $\text{M}\Omega\text{cm}^{-1}$ decarbonised water was added to
 226 bring about the same volume as the other microcosms with an inorganic amendment. 28 μL each
 227 of acetate (2 mM), propionate (2 mM), butyrate (2 mM), and 10 mL filtered freshwater (filtered
 228 with a 0.45 μm PES filter) was added to the positive control. The rubber stopper was placed onto
 229 every serum bottle and sealed with an aluminium cap. The microcosms were stored in the box
 230 (away from sunlight to avoid photo mineralisation or degradation) at room temperature.

231 Table 1: Microcosms treatment groups, description, and rationale behind each control group

Treatment Group	Treatment (microcosm) ID	Description	Reasons
Inoculated sample	A+F and B+F	These contained testing materials (leachate A and B) as sources of C substrate, plus the inoculum filter.	To study whether the introduced microbes can biodegrade the DOM derived from the leachates.
Filter control	Blk+F	The set contained only decarbonized (18.2 $\text{M}\Omega\text{cm}^{-1}$) water and an inoculum filter.	To investigate the filter response due to incubation. To assess whether changes in A+F and B+F are associated with the leachates or the filter itself due to OM carryover from the freshwater sample. Note: We assumed the filter might not collect only microbes, but with particulate OM that may get mobilized in the decarbonised water and potentially utilised by the microbes.
Blank Control	Blk	Decarbonised (18.2 $\text{M}\Omega\text{cm}^{-1}$) water without any C amendment (i.e., without Tarfaya black shale-derived DOM, surface water-derived DOM or any added nutrients) or inoculation.	To observe any response from the decarbonised water.
Positive control	F+Org+FW	Contained freshwater-derived DOM and added organic substrates (acetate (2 mM), propionate (2 mM) and butyrate (2 mM) acid) instead of Tarfaya black shale-derived DOM as a C source to the microbes. Note: the freshwater was autoclaved before addition.	To provide the microcosm with excess organic and inorganic nutrients to observe whether the filter inoculum can produce CO_2 and CH_4 .
Non-inoculated control	A and B	Abiotic controls: Contained leachate only (no inoculum filter).	To investigate whether the shale samples contain indigenous microbes which may be activated through the process or to monitor any abiotic changes of DOM throughout incubation.
Autoclaved control	A+aut.F and B+aut.F	The inoculum filters were autoclaved before the addition of leachate.	To assess whether changes in A+F and B+F are associated with the presence of the filter itself, rather than living microbes in the filter.
Inorganic nutrient control	A-inorg, B-inorg A+F-inorg, and B+F-inorg	The inorganic nutrients were not added to this group.	To assess whether shale-derived DOM (A+F and B+F) contain sufficient inorganic nutrients to sustain biodegradation.

232

233 Key: A and B =leachate A and B, F= filter, Blk = blank (18.2 $\text{M}\Omega\text{cm}^{-1}$ decarbonized water), FW = freshwater, aut.F=
 234 autoclaved filter, inorg=inorganic amendments, org= organic amendments.

235 Changes in CO_2 and CH_4 concentrations in the microcosms' headspace were measured using a
 236 GHG analyser at 1, 5, 9, 23, 54, 97, 275, 580, and 810 days of incubation and recorded. For quality
 237 assurance, a new calibration was performed after each GHG analyser's maintenance service or
 238 longer measurement intervals. All calibration correlation coefficients (R^2) fall within the range of
 239 0.9989 to 1.0000 at a 99.5 % confidence interval. The wider gaps in measurement days between
 240 97, 275, 580, and 810 days were due to the laboratory's inaccessibility during the subsequent



241 COVID-19 lockdown. This might favour the anaerobic microbes to grow as there is no ventilation
242 (no oxygen supply) of the headspace within the interval time. Upon termination of the experiment
243 on day 810, the inoculum medium of microcosm A + F and B + F, and other filter and positive
244 controls (Blk + F and F + Org + FW) in which we observed substantial microbial response
245 (considerable generation of biogenic CO₂) were further analysed for biotransformation of DOM.
246 The medium was split into two, 6 mL + ¼ filter fractions, which were stored at -80 °C for potential
247 DNA extraction and qPCR, while ~7 mL of the remaining sample was filtered using a 0.45 µm
248 PES filter and analysed using LC-OCD for potential biotransformation/degradation of DOM.

249 **2.4 Analytical Procedures**

250 Several analytical techniques were applied to characterise the % TOC and kerogen type of the
251 black shale samples, and the leachate's DOM before and following incubations. The % TOC,
252 thermal maturity, and kerogen type were analysed using elemental analyses and Rock-Eval
253 pyrolysis performed on bulk sediments (data from Kolonic et al. 2005). LC-OCD was applied in
254 this study to investigate the trajectory chemical changes of DOM in microcosms. A GHG analyser
255 was used to measure the change in CO₂ and CH₄ in the microcosm headspace over time.

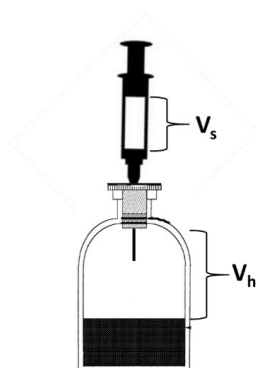
256 **2.4.1 Liquid chromatography – organic carbon detector – organic nitrogen detector** 257 **(LC-OCD-OND)**

258 The LC-OCD-OND model 9 was used to measure the change in DOM quantity and compositional
259 organic fractions change with incubation. All glass vials were ashed, and for reproducibility, each
260 sample replicate was run twice on the SEC from which the mean value and relative per cent
261 difference were determined. Initially, the aliquot of 400 µL filtered leachate was diluted with 7.6
262 mL 18.2 MΩ cm⁻¹ decarbonized water (1:20 v/v) in an ashed glass vial (the dilution was to ensure
263 the measured DOM concentration is within the instrument calibration range, 0.5 – 5000 µg/L),
264 and prepared for LC–OCD analysis. After 810 days of incubation, the experiment was terminated.
265 The serum bottle was de-crimped, a sterile syringe (60 mL) with PES 0.45 µm was rinsed with
266 18.2 MΩ cm⁻¹ decarbonized water, and then 7 mL of the incubation sample was poured from the
267 top after removing the syringe piston and then the piston was inserted back. The first 2 mL that
268 passed the filter was discarded to avoid contamination if residual OM hang in the syringe piston.
269 The samples were then filtered into the glass vial. As the post-incubation samples still contain a
270 higher amount of DOM beyond the LC-OCD calibrated measuring range, the A+F, B+F and
271 Blk+F were appropriately diluted (1:15 v/v %) and F+Org+FW by 1:20 v/v % with 18.2 MΩ cm⁻¹
272 decarbonized water before analysing them in LC–OCD. ChromeCalc software was used to
273 manually integrate sample peaks as presented in section 3.6.4 (Fig 3.4; Table 3.1)



274 **2.4.2 GHG-GC analysis**

275 The gas chromatograph with a flame ionisation detector (GC – FID) was used to measure the
276 concentration of CO₂ (via a methanizer) and CH₄ in the headspace of the microcosm during the
277 incubation period of 810 days. To measure the GHG from the headspace, the cap of the serum
278 bottle was wiped with ethanol, and a volume of 5 - 10 mL was taken off the headspace using a 10
279 mL syringe mounted with a needle (Fig 3). The needle was flushed with 2 mL of the sample gas
280 before being injected under constant pressure (19-21 ml/min equal to the injection pressure of
281 calibration) into the GHG analyser. The headspace was ventilated with air (re-introduction of
282 oxygen) on 54 and 275 days of incubation by injecting atmospheric air equivalent to the total
283 volume of the samples taken out. The syringe filter (0.2 µm) was mounted on the syringe during
284 the injection of atmospheric air into the microcosm. The change in the concentration of the CO₂
285 and CH₄ with re-ventilation was calculated using the derived equation 1.



286

287 Figure 3: Diagram of the microcosm system

288
$$C = C_1 - \frac{V_s}{V_h} (C_1 - C_2) \quad 1$$

289 Where C is the change in concentration of CO₂/CH₄ in the headspace, C₁ is an initial concentration
290 of CO₂/CH₄ in the headspace, C₂ be a concentration of CO₂/CH₄ in air, V_h is the volume of the
291 headspace and V_s be the volume of the amount taken in the syringe.

292 A requirement for the quantification of GHGs produced from the black shale leachate microcosm
293 headspace required appropriate gas-phase calibration standards to determine absolute
294 concentrations. Following the very high CO₂ production observed in the initial microcosm test
295 phase the calibration was extended to up to 30% (300000 ppm).



296 3.0 Results

297 3.1 Molecular composition of DOM in the leachates

298 The LC-OCD results showed that black shale sample B (10.5 % TOC and of kerogen Type I)
299 released more DOM (41.34 mg/g rock) than sample A (7.5 % TOC and kerogen Type I/II) which
300 released 17.78 mg/g rock. Both samples contain 9 identifiable compound groups (Fig 4), which
301 range from macromolecules (X1, X2, and X3; large groups of compounds with molecular weight
302 >350 Da) to low molecular weight groups (X4 – X9; MW <350 Da). The DOM of the shale
303 leachates was characterized by one prominent peak, X4, eluted at a retention time range of 23.35
304 – 30.82 minutes in the chromatogram, contributing 38.3 and 42.0 % in leachate A and B to the
305 total DOC. This peak is considered to represent low molecular weight organic acids fraction (X4;
306 Huber et al. (2011); Zhu et al. (2015)). OM mobilised from kerogen is generally believed as the
307 source of LMWOAs and their generation is considered to result from an early stage of thermal
308 maturation; Diagenesis (Zhu et al., 2015) or hydrolysis of cross-linked esters within the
309 macromolecular network of kerogen (Siskin and Katritzky 1991). The concentrations of DOM
310 acids extracted from the shales were validated against the initial kerogen oxygen content of the
311 Tarfaya shale samples. The samples are thermally immature. The immature kerogen contains
312 significant amounts of aliphatic components and oxygen ((Zhu et al., 2015; Leins et al. 2022).
313 The oxidation of n-alkanes during geological times might also lead to the formation of LMWOAs.
314 Thus, the higher relative proportion of X4 in B compared to leachate A (17.36 vs 6.81 mg/g rock)
315 likely reflects an elevated release from the high pyrolytic kerogen S2 yields, expressed in the
316 S2/S3 ratio (18.55 vs 13.77).

317 Table 2 indicates the different compound groups eluting at different time ranges and molecular
318 sizes from the LC-OCD column. The LC-OCD results of the sample leachates (A and B) showed
319 the first peak at 15.18 – 18.62 minutes retention time, representing macromolecule X₁ with the
320 largest molecular size. The X₁ fraction contributed 1.9 and 0.9 % to the total DOM. The peak X₂
321 observed at 18.62– 20.87 minutes contributed 5.9 and 8.9 % and had a lower MW than X₁. The
322 peak eluted at 20.87 – 23.35 minutes represents macromolecule X₃ (MW < X₂), which contributed
323 19.7 and 12.7 % to the total DOM. We identified low molecular fractions (X₅ – X₉), which are
324 composed of high molecular weight neutrals (X₅ and X₆), intermediate neutrals (X₇ and X₈), and
325 very low molecular weight neutral (X₉) in both leachates. The two peaks X₅ and X₆, which
326 cumulatively contributed 24.9 and 17.8 % in leachate A and B to the total DOM, had a residence
327 time range of 30.82 – 40.07 minutes. The intermediary neutrals observed at 40.07 – 46.95 minutes
328 contributed only 10.0 % and 4.8 % to the total DOM. The very labile fraction X₉ eluting at the
329 tail end of the chromatogram at residence time range 46.95- 70 minutes, contributed 7.0 and 6.0%
330 to the total DOM.



331 Table 2: Description of LC-OCD fractions; Modified from Huber et al. (2011), Penru et al. (2013), and Zhu
332 et al. (2015).

This study fractions		Fraction (Huber et al., 2011)	Molecular mass range (Da)	Properties	Description
Macro-1	X1	Biopolymers	>10,000	Not/little UV-absorbable, hydrophilic	Polysaccharides and proteins
Macro-2	X2	Humic substances	>1000	Highly UV-absorbable, hydrophobic	Calibration based on Suwannee River standard from IHSS
Macro-3	X3	Building blocks	350 – 500	UV-absorbable	Break down products of humic substances
Acids	X4	Low molecular weight acids	<350	Negatively charged	Low molecular weight aliphatic acids
Neutrals	X5 – X9	Intermediate – low molecular weight neutrals	<350	Weakly or uncharged hydrophilic, amphiphilic	Alcohols, aldehydes, ketones, amino acids

333

334 The compound groups X1 – X4 (in black) are reported by Zhu et al. (2015), while X5 – X6 (in red) are
335 additional peaks (compounds) reported in this work.

336 Both leachates have moderate UV absorbance with a relatively higher response from leachate B
337 (UV absorbance: A = 11.67 AU, B = 24.48 AU), indicating the presence of aromatic/unsaturated
338 compounds. Both leachates show similar concentrations of total dissolved nitrogen (A = 3.27
339 mg/g ON, B = 3.43 mg/g ON).

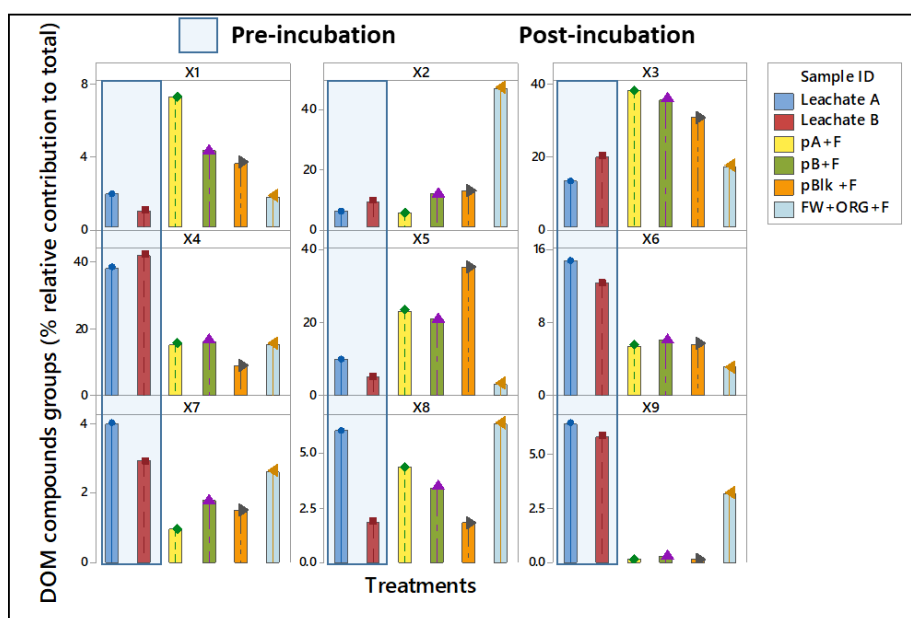
340 **3.2 Changes in the molecular composition of DOM related to microbial** 341 **activity**

342 The LC-OCD was used to investigate the change in DOM compositions in the microcosms over
343 810 days of incubation (Fig 4). The results of pre-incubation leachates indicated that the relative
344 contribution of lower molecular weight (LMW) compound groups (X₄ – X₉) to total DOM was ~
345 80 % and 70 % for leachate A and B, respectively. However, after the incubation, the detected
346 compound groups showed a marked decrease in the % contribution of LMW compounds to ~50
347 % and 52 % for inoculated leachates A and B microcosms, with LMWOA (X₄) decreasing by 23
348 % and 25.5 %. Also, the very labile carbon group (X₉) decreased from 6.5 % to 0.2 % and 5.9 %
349 to 0.3 % in inoculated leachates A and B microcosms (Fig 4). The % relative contribution of the
350 humic-like macromolecules (X₁, X₂, and X₃) stand as 7.2±0.7, 5.0±0.2 and 37.9±0.5 % in
351 inoculated leachates A, and 4.2±1.0, 11.2± 3.2 and 35.5±1.9 % in inoculated leachates B
352 microcosms, which are relatively higher than in pre-incubation leachates A and B (1.9, 5.6 and
353 12.7 % and 0.9, 8.9 and 19.7 %).

354 The filter control microcosm (Blk+F) displayed a % relative contribution of the humic-like
355 macromolecules X₁ and X₃ (3.6 and 30.4 %), lower than the samples inoculated microcosms with
356 X₂ (12.4 %) slightly higher.



357 The positive control (F+Org+FW) exhibited a similar macromolecular pattern to the filter control,
 358 showing lower X₁ and X₃ (1.7±0.14 and 17.1±0.4%) and prominently higher X₂ (46.7±1.0%) than
 359 the microcosm with leachates. The % relative contribution to the total of high molecular weight
 360 neutral group (X₅) increased after incubation in both leachates inoculated microcosms (23.2±2.8%
 361 and 21.0±3.0% for A+F and B+F) and filter control (35.3 %) compared to pre-incubation
 362 leachates (10.1 and 5.4% for A and B). Unlike the sample inoculated microcosms, where lower
 363 molecular fractions disappeared with the incubation, we observed a marked increase of X₈
 364 (6.4±0.6 %) and a slight change of X₉ (3.2±0.1%) in the positive control.



365

366 Figure 4: DOM composition of pre- and post-incubation samples (% relative contribution to total DOM)
 367 measured by LC-OCD. The relative proportion was the mean value of the triplicates. Key: P, post-
 368 incubation samples. A and B represent sample leachates, blank (Blk); deionised water, F; filter, FW;
 369 freshwater, aut.F; autoclave filter, Org; organic, and inorg; inorganic amendments (Table 4.1). X₁, X₂, and
 370 X₃; large groups of compounds with molecular weight >350 Da), and low molecular weight groups (X₄ –
 371 X₉; MW <350 Da) (refer to Table 3). This data suggests the degradation of labile compounds (X₄ –X₉)
 372 with the formation/increase in macromolecules X₁ – X₃ and LMW X₅.

373 Stoichiometric calculations were applied to determine how much organic carbon was degraded to
 374 produce the observed CO₂ and CH₄, and how much total organic C was lost from the leachates.
 375 The following assumptions were made: no external organic carbon source in addition to what was
 376 present in the leachates, and an inoculated filter (which carries over freshwater OM) in the sample
 377 inoculated microcosms. Therefore, the organic feedstock for the microbes should be either C from
 378 the leachates, inoculated filter or contributions of necromass (dead cells).



379 We already know the amount of dissolved C in our leachates; therefore, we need to determine the
380 amount contributed by the necromass and filter.

381 For the necromass contributions, we need to know the actual amount of cells that we have in the
382 microcosms and C cell⁻¹. Braun et al. (2016) quantified amino acids in Baltic Sea sediments and
383 calculated cell-specific carbon content on the order of 19 to 31 fg C cell⁻¹. The abundance of cells
384 in both oil sands and formation water has been previously estimated to be between 10⁴ and 10⁷
385 cells g⁻¹, with the highest cell densities found at the oil–water transition zone (Bennett et al.,
386 2013). Although we don't have the quantitative data of the cells in our microcosms, we assumed
387 the highest values obtained by Bennett et al. (2013). The reaction mixture in our microcosms was
388 ~ 18 mL, therefore, the estimated number of the cells in our inoculated microcosms is 1.8 x 10⁸
389 cells per microcosm. If we assumed 75 % of the cells were utilized by saprotrophs, then the
390 amount of C necromass contributions can be estimated as 4.19 µg C.

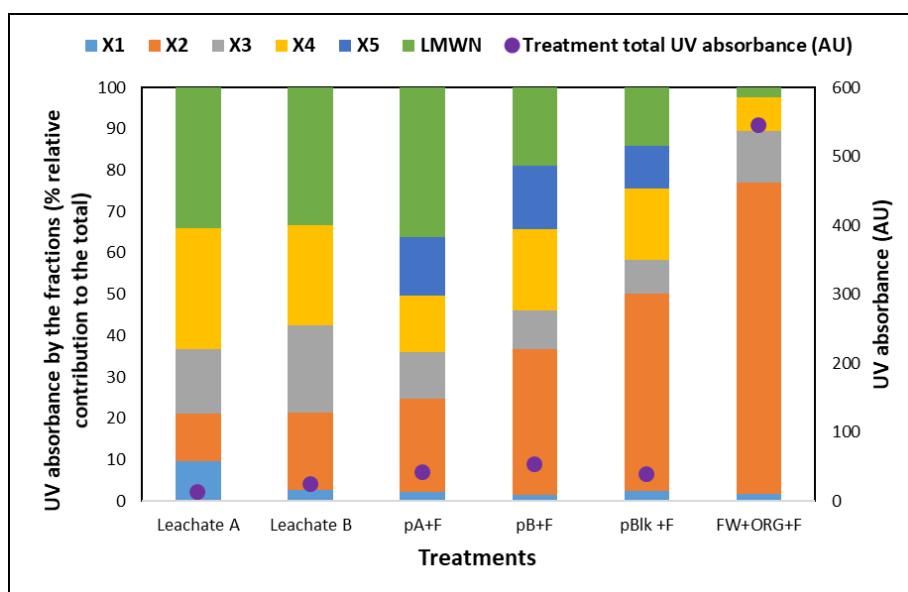
391 For the filter control contributions, we need to know the actual amount of C carried over by the
392 filter from freshwater. Since we observed the breakdown of the filter during incubation, we might
393 suggest also that there was a degradation of the C substrate from the nylon inoculum filter.
394 Therefore, to estimate the inoculum filter control contributions we need to determine the amount
395 of carbon the inoculum filter released in the microcosms, whether it continuously released DOM
396 with depletion of DOM due to microbial degradation in the microcosm and the amount of carbon
397 that remained on the filter (undissolved) after the incubation. These might be very complex and
398 difficult to analyse. However, in this research work, since all the inoculum filters underwent
399 similar treatments, we assumed; the filter contributed a relatively similar amount of C in the
400 inoculum leachates and filter microcosms. To eliminate the filter contributions, we subtract the
401 filter microcosms from the sample microcosms. This approach has previously been used by Miles
402 and Doucette, (2001).

403 DOM concentration of the leachates subsequently decreased after the incubation (A; 17.78 to 4.01
404 mg/g; B; 41.34 to 9.66 mg/g; Note; the post-incubation concentrations were calculated after filter
405 correction; to get the residual DOM of leachates after the incubation period, the post-incubation
406 concentrations in filter control (Blk + F) was subtracted from the inoculated samples (A+F and
407 B+F). The results suggest mineralization of 13.77 and 31.68 mg/g DOM of leachate A and B.

408 Biodegradation does not only mineralize DOM but also bio-transform and generates other OM
409 compounds as by-products, i.e., monomer carbohydrates, low molecular alcohols, polypeptides,
410 to low-molecular-weight monomers, e.g., amino acids and urea (Antia et al., 1991; Zehr and Ward,
411 2002). Some of these by-products contain aromatic and unsaturated compounds. The total UV
412 absorbance was used as a quantitative measure to infer aromatic/ unsaturated compounds formed
413 or utilised in the DOM compound groups by microbial activity at the end of the incubation. The



414 composition of the pre-incubation leachates was generally lower (11.67 and 24.48 Absorption
 415 Unit {AU}) compared to post-incubation samples (Fig 5). The inoculated samples' microcosms
 416 exhibited 40.58±2.67 and 53.38±15.38 AU for A+F and B+F, while the positive control
 417 (F+Org+FW) showed relatively significant UV absorbance (544.84±1.10 AU; in which X₂
 418 contributed to 75.4±0.8 % relative to total) and Blk+F, which had a 37.7 AU in which X₂
 419 contributed 46.7 %, respectively. The relative UV absorbance increases in the macromolecule X₂
 420 across the post-incubation microcosms with positive and filter controls revealed more significant
 421 absorbance. The % relative increase in the contribution of high molecular weight neutrals (X₅)
 422 by ~15% to the total DOC coincided with the increase in their UV absorbance in the sample
 423 inoculated microcosm (14.0±0.7% for A+F and 15.4±1.4% for B+F), and 10.2 % in filter control
 424 samples while showing no UV absorbance response in the pre-incubation leachates and positive
 425 control. The Lower molecular weight neutrals (LMWN; X₅ – X₉) in pre-incubated leachates
 426 showed elevated UV absorbance (34.1 and 33.3 % relative to the total absorbance) compared to
 427 the inoculated sample (B+F; 18.8±3.0 %), filter control (Blk ± F; 14.4 %), with the positive
 428 control (FW+Org+F; 2.3±0.4 %) showing the least absorbance.



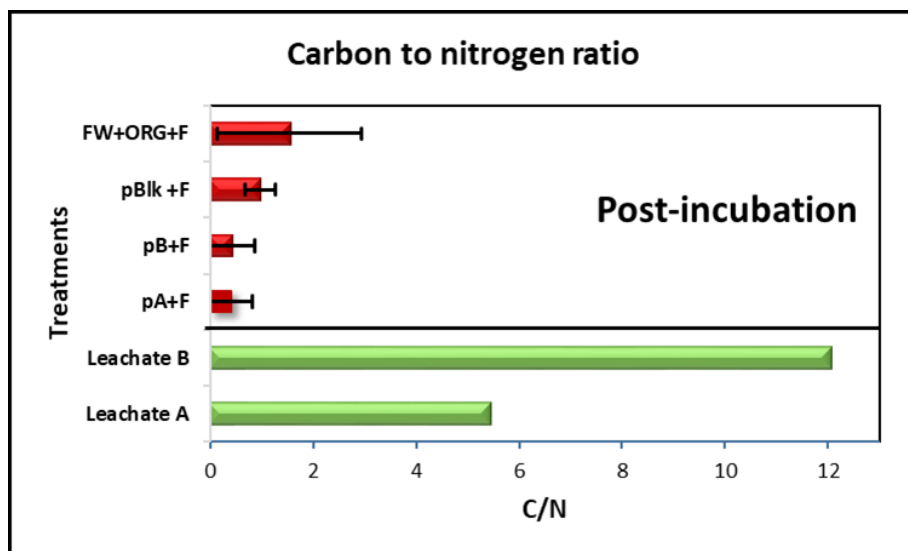
429

430 Figure 5: UV absorbance of samples and % relative contribution to the total by the detected compound
 431 groups. These reflect the formation of LMW condensed aromatic group X₅ and biotransformation of DOM
 432 toward more recalcitrant and highly aromatized/unsaturated macromolecule group X₂.

433 A higher C/N ratio of the samples before the incubation was observed (5.44 and 12.05 for leachate
 434 A and B, respectively) (Fig 6). The C/N was notably lower (compared to leachates) after the
 435 incubation in the inoculated leachates microcosm and controls (filter and positive control): 0.41,
 436 0.43, 0.97 and 1.54 for A+F, B+F, Blk+F and F+Org+FW, respectively. This might suggest the



437 substantial mineralization of the DOC or there was a lot more DON in the surface water which
438 was carried over by the filter in pA+F, pB+F, and pBlk+F.



439

440 Figure 6:: Carbon to nitrogen ratio of pre and post-incubation samples indicating a drastic drop in C/N ratio,
441 which signified mineralization of C substrate from the DOM pool. Note: The C/N was calculated by
442 dividing the total DOC by the total DON.

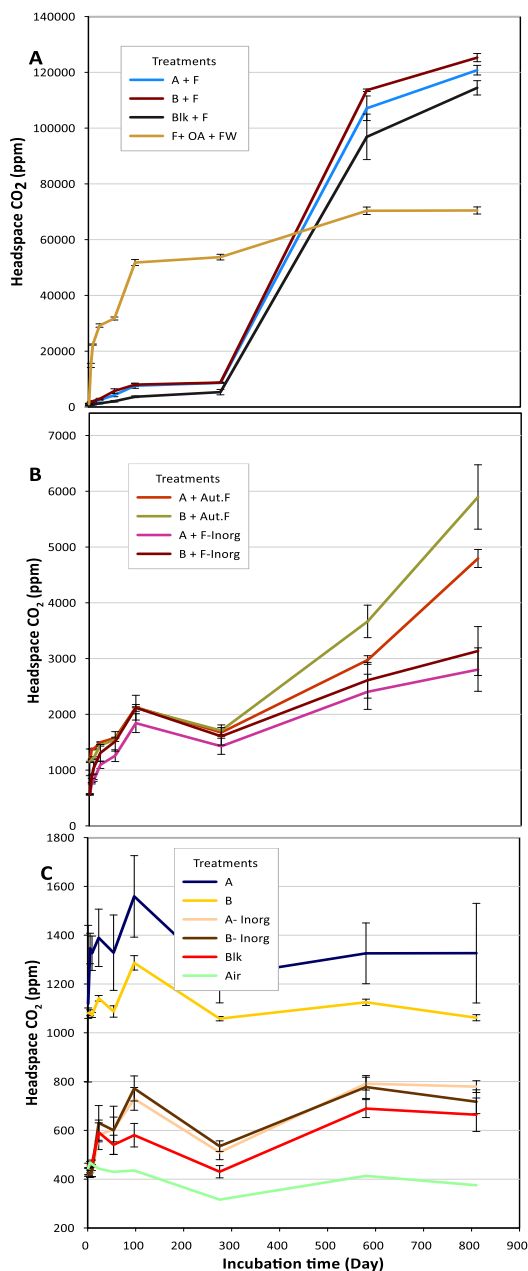
443 3.3 Microbial Growth and GHG Production

444 3.3.1 Carbon dioxide - CO₂

445 The production of CO₂ and CH₄ in the headspace of the triplicate microcosms was monitored for
446 810 days with recurrent measurements of the headspace of CO₂ and CH₄ using a GHG-GC
447 analyser. The cumulative mean CO₂ and CH₄ concentration over time (days) and calculated
448 standard errors from the triplicate measurements from microcosm headspace samples are
449 presented in Figs 7 and 8 and fully reported in Appendix I and II. Fig 7A showed that the
450 inoculated sample microcosms headspace means CO₂ concentrations of 1307±4 and 1195±7 ppm
451 for A±F and B±F on the first day of incubation which was twice the measured laboratory
452 atmospheric concentration (445 ppm) and the blank control. The mean CO₂ concentration doubled
453 after 23 days of incubation and reached 7596±917 ppm for A+F and 8002±449 ppm for B+F at
454 97 days. Substantial production of CO₂ was observed at 580 days of incubation, raising the
455 headspace CO₂ concentration to 107113±4414 ppm and 113585±443 ppm for A+F and B+F. The
456 cumulative mean concentrations reached a peak at 810 days, totalling 120779±1720 and
457 125278±1462 ppm for A+F and B+F microcosms (Fig 7A). These peak concentrations are 92 and
458 104 times the initial concentration (concentration on the first day) of inoculated sample



459 microcosms and 271 times the laboratory atmospheric air CO₂ concentration and blank control
460 (blk).



461

462 Figure 7: Measured headspace CO₂ mean concentrations of the different microcosms set up over time. Key;
463 A; Samples inoculated microcosm, blank, filter, air and positive controls. B; Autoclaved and inorganic
464 nutrients control. C; Uninoculated control group.



465 In the experimental control groups, the first control group (positive control; F + OA + FW)
466 displayed a mean concentration of 1015 ± 8 ppm CO₂ on the first day of the incubation (Fig 7A and
467 appendix I). A rapid increase in CO₂ concentration was observed in the headspace of the
468 microcosm after 5 days producing 14887 ± 75 ppm CO₂ (which is 33 times high than blank control
469 (blk) and measured atmospheric CO₂ concentration which serves as the reference point). The
470 production of CO₂ continued rapidly until 23 days of the incubation reaching 29193 ± 602 ppm
471 CO₂, then slowed down. The F + OA + FW show the cumulative mean amount of CO₂ reached
472 70462 ± 1270 ppm CO₂ after 810 days of incubation.

473 In Fig 7C, the blank (Blk) control showed a relatively similar CO₂ means concentration on the
474 first day (472 ± 16 ppm CO₂) with laboratory atmospheric concentration. The CO₂ concentration
475 remains slightly the same throughout the incubation (810 days) with a cumulative mean
476 concentration of 664 ± 69 ppm CO₂.

477 The filter control (Blk+F) microcosms displayed similar headspace CO₂ concentrations as
478 laboratory atmospheric air (472 ± 16 ppm) on the 1st day of the incubation. However, the CO₂
479 concentration in the microcosm gradually continued to rise and reached 5317 ± 969 ppm at 275
480 days of incubation. The massive CO₂ production was observed at 580 days (96860 ± 8155 ppm)
481 and cumulatively reached a peak at 810 days of incubation, totalling 114441 ± 2565 ppm (Fig 7A).

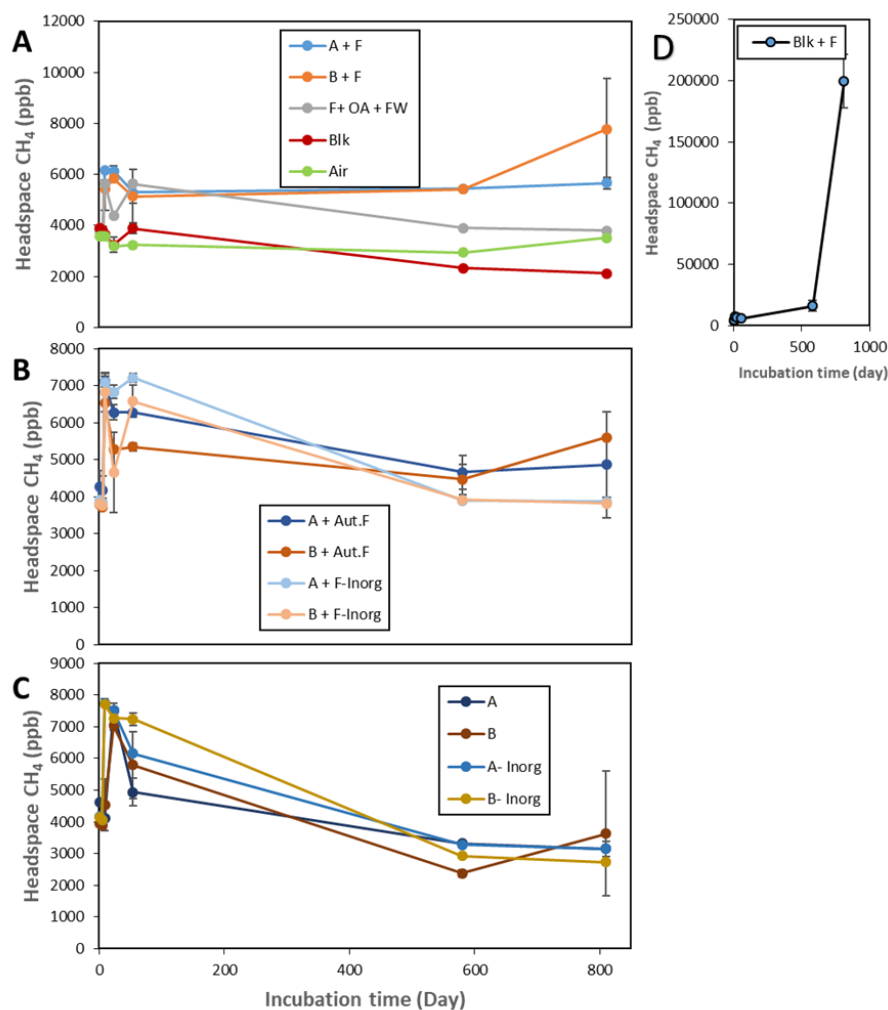
482 The uninoculated control group (A, B, A-inorg, B-inorg) generally showed no significant change
483 in CO₂ concentration in the headspace (from day 1 to 810), and headspace total concentration
484 stands below 1600 ppm CO₂ at 810 days of the incubation (Fig 7C). In the Fig 7B, the autoclaved
485 control (A+aut. F and B+aut. F) revealed a cumulative measured CO₂ mean concentration of
486 4795 ± 162 and 5897 ± 577 ppm for A+aut. F and B+aut. F in their headspace after 810 days of
487 incubation. Also, the inorganic nutrients control group (containing leachates and microbes but
488 without inorganic amendments; A+F – inorg, and B+F – inorg) cumulatively generated only
489 ~ 3000 ppm CO₂ after 810 days of incubation.

490 3.3.2 Methane - CH₄

491 The CH₄ mean concentrations of the inoculated sample microcosms and controls are presented in
492 Fig 8 and Appendix II. Data in Fig 8A showed that the mean concentration of CH₄ in leachate's
493 inoculated microcosms remains relatively similar to laboratory air (3579 ppb) in the first 5 days
494 of the incubation and slightly increases to 6132 ± 216 and 5830 ± 135 ppb for A+F and B+F, which
495 were twice the atmospheric and blank controls after 23 days of incubation. The cumulative mean
496 methane concentrations lowered to 5298 ± 439 and 5148 ± 1056 ppb (for A+F and B+F) on 54 days
497 of incubation. And slightly increase to 5661 ± 227 and 7775 ± 1966 ppb (2 to 3 times atmospheric
498 concentration) after 810 incubation days.



499 The low headspace CH₄ concentration was observed (relatively similar to atmospheric
 500 concentration) throughout incubation time for the positive control (FW+Org+F) except at 9 and
 501 54 days of incubation, which reached 5642±126 and 5857±56 ppb CH₄, respectively, and returned
 502 to 4031±26 ppb at 810 days of incubation.



503

504 Figure 8: Measured headspace CH₄ mean concentrations of the different microcosms set up over time. Key;
 505 A; Samples inoculated microcosm, and blank, air and positive controls. B; Autoclaved and inorganic
 506 nutrients control. C; Abiotic control group.

507 The autoclaved, abiotic, and inorganic nutrients controls showed slightly similar headspace
 508 methane concentrations (between 4167 – 3755 ppb) as the inoculated samples' microcosms in the
 509 first five days of the incubations (Fig 8B and 8C). The abiotic controls showed no change in



510 methane concentration on the 9th day of the incubation and returned to atmospheric concentration
511 after 810 days of the incubation time (Fig 8B).

512 The autoclave and inorganic controls showed an increase in headspace methane concentration
513 (6545 – 7754 ppb) at 9 days of the incubation and remained slightly similar after 54 days of the
514 incubation. The mean concentration drastically reduced to 3831 ± 881 ppb in A+aut.F, while it rose
515 to 9057 ± 506 ppb in B+aut.F at 810 days of the incubation.

516 Both the filter and blank controls displayed the same mean CH₄ concentration measured in the
517 headspaces of their microcosm with atmospheric conditions in the first and 5 days of incubation.
518 The mean concentration of CH₄ in the filter control rises to 7146 ± 1855 ppb on the 9 days and
519 significantly upsurges to 16211 ± 4417 ppb on 580 days and reaches a peak at 810 days, reaching
520 199413 ± 21971 ppb CH₄ (Fig 8D).

521 **4.0 Discussion**

522 Despite the importance of DOM-microbial interactions in the natural and perturbed environment,
523 little is known about the interaction and availability of rock-derived DOM to subsurface microbes,
524 microbe-stimulated DOM transformations/degradation, and the resultant biogenic GHG
525 formation in subsurface environments. In this study, we investigate the potential of immature
526 marine black shale to stimulate microbial degradation and generate GHG (CO₂ and CH₄), and we
527 applied analytical biogeochemical tools (LC-OCD) to document these changes in DOM chemistry
528 and quantify the biogenic CO₂ and CH₄ generated from microbial activity.

529 For the set of biodegradation experiments, the GHG (CO₂ and CH₄) production in inoculated
530 samples microcosms was compared with six control groups (these are: positive, filter, blank,
531 autoclaved, abiotic and inorganic nutrients controls). The results from identically treated triplicate
532 microcosms show good precision and reproducibility (RSD <15 %).

533 In the positive control microcosm (F+OA+FW), the microbes were introduced to their indigenous
534 freshwater system with additional organic and inorganic nutrients. The massive production of CO₂
535 observed at the earliest stage of the incubation demonstrates that the microbial community
536 collected in the freshwater was active and readily adapted to the microcosm system, which
537 simulated their indigenous environment (freshwater). The rapid production of CO₂ may be related
538 to the rapid utilisation of C substrate supplied to microbes to support their fast growth in the
539 microcosm. Although there is no pre-incubation data for this microcosm, the biodegradation
540 presumably generated highly aromatised humic-like macromolecule X₂ as dominant DOM
541 fractions as observed in the LC-OCD data post-incubation (Figs 5 and 6) and generated very labile
542 carbon (X₈ and X₉) as degradation products. These results are consistent with Wu et al. 2018 who
543 incubated sediment-derived dissolved organic matter (DOM) and groundwater microbes and



544 continually analysed the microbial transformation of DOM over a 50-day incubation. Their results
545 displayed a domination of less labile biopolymers (lipids) and lignin-like compounds at the end
546 of the incubation. The decline in the production of CO₂ observed towards the end of the incubation
547 might be related to either a limiting factor (electron acceptor or electron donor) or any
548 physicochemical condition. It might also probably be due to a shift in the microbial community
549 (natural selection) stimulated by the domination of highly aromatized humic-like macromolecule
550 X₂, or a symbiotic relationship in which the microbially produced CO₂ was continually complexed
551 to form the observed LMW organic compounds (X₈ and X₉) by CO₂ fixing microbes, possibly
552 from a bacterial domain which survive a wide range of aerobic and anaerobic habitats (Saini et
553 al., 2011; Wani et al. 2020).

554 In the earliest stage of the incubation (day 1), all the microcosms containing the leachates and
555 inorganic nutrients (sample inoculated group; A+F and B+F, uninoculated control; A and B, and
556 autoclaved control; A+aut.F and B+aut.F) showed similar headspace CO₂ concentration slightly
557 above 1000 ppm (which is twice as high as atmospheric and blank control concentrations). This
558 likely suggested that the recorded headspace CO₂ concentration might be due to the background
559 respiration of the leachate samples, supported by the addition of inorganic nutrients. Similar
560 results were reported by Horstmann (2018).

561 Unlike in the positive control, where we observed a short lag phase/ acclimation period (within
562 the first 5 days), the leachate-containing inoculated microcosms (A+F and B+F) exhibited longer
563 lag phases. In addition, we observed a slow production of CO₂ for around 3 months (97 days) of
564 the incubation, generating ~8000 ppm. Since the abiotic controls headspace CO₂ concentration
565 remained largely unchanged, suggesting the CO₂ generated in the inoculated samples
566 microcosm's headspace was largely due to microbial activities. The slight production of CO₂
567 observed in the autoclaved controls throughout incubation may suggest that the autoclaving did
568 not completely kill the microbes, and this was consistent with observations from Horstmann
569 (2018). The slight increase in CO₂ concentration in the sample-inoculated treatments without
570 inorganic nutrients (A+F-inorg and B+F-inorg) showed the significance of inorganic nutrients to
571 microbial activities, as electron acceptors and donors must be present in large enough quantities
572 (Nixon et al., 2012; Moore et al., 2025). A similar pattern with much lower headspace CO₂
573 concentrations was observed in the filter control (Blk+F) compared to sample-inoculated
574 microcosms. The similarity in observed patterns might be because the filter supplied a large
575 proportion of DOM (which is of freshwater origin to both microcosms). While the lower
576 concentration might be due to poor carbon substrate availability (as 18.2 MΩ cm⁻¹ decarbonised
577 water was used in place of the leachate) and pH difference. Unlike the filter control, which
578 contained only freshwater DOM, the inoculated samples' microcosms contained DOM of marine
579 origin in addition to freshwater DOM. The aged source-rocks-derived DOM may be more



580 biologically available when mobilised through rock-water interactions, despite its lower
581 concentration compared to the larger and possibly more stable freshwater DOM in the microcosm
582 (Findlay and Parr, 2017). The DOM compound group compositions in marine black shale are
583 highly dominated by LMWOAs and aliphatic LMWNs, while the surface water is majorly
584 reported as dominated by macromolecular compounds (X1, X2, and X3) (Wilke et al. 2015; Yates
585 et al. 2019; Aukes, et al. 2021). The LMWOAs and LMWNs are easily biodegraded by microbes
586 compared to larger molecules (Wu et al. 2018). This might probably be the reason for the higher
587 CO₂ concentrations observed in the inoculated sample microcosms at an early stage of the
588 incubation compared to the filter control, confirming the immediate biodegradation of the
589 leachate's source DOM.

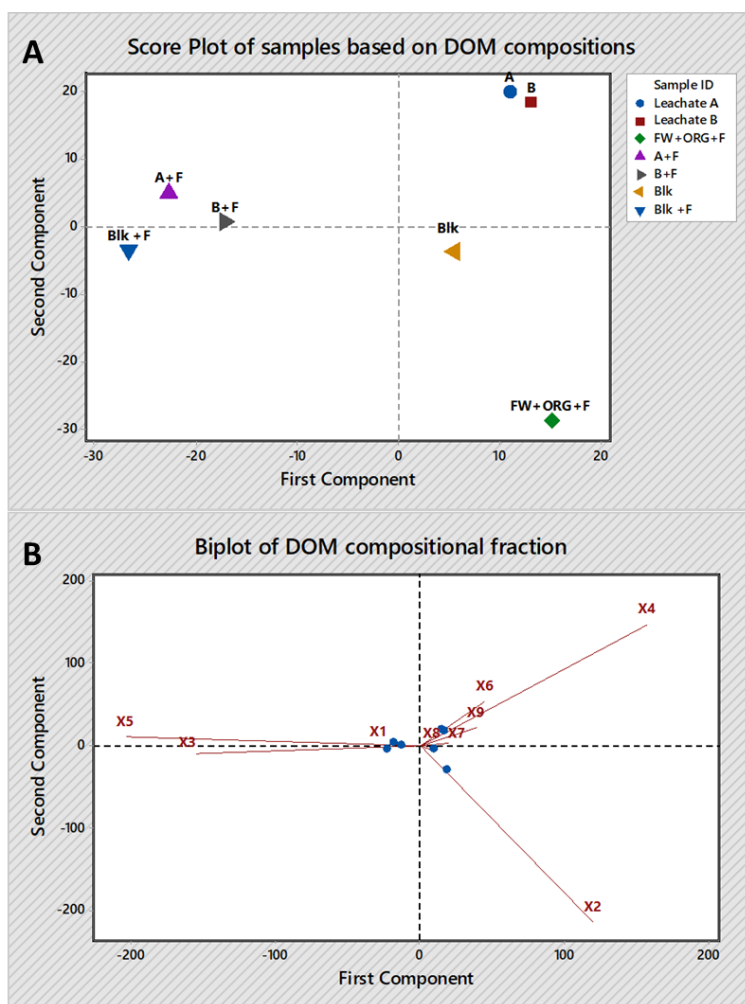
590 The period of the lag phase may vary based on the environments that the microbes came from, as
591 well as the condition of the microbes' cells themselves. If the growing cells are shifted from one
592 type of media into the same type of media, which has similar environmental conditions, they will
593 have the shortest lag period, as we observed in the positive control. Therefore, the slow production
594 in the sample inoculated group may be related to the unusual environment and the C substrate
595 introduced to the microbes. The leachates were derived from Cretaceous marine black shale, while
596 the microbes were collected from the surface water. Therefore, the leachates from aged marine
597 shale substrates might be unsuitable for immediate surface water microbes to grow; hence, they
598 needed to adapt to the new environment (Wani et al., 2022). The microbial adaptation in different
599 environmental conditions is mediated by complex cellular changes that keep physiology and
600 metabolism, thereby ensuring survival (Wani et al., 2022). Adaptation may involve processes such
601 as microbial growth, enzyme induction, or the exchange of genetic or molecular information to
602 give the microbes substrate-degrading capabilities (Miles and Doucette, 2001; Wani et al., 2022).
603 Therefore, an adaptation of the microbes to the new system might be the reason for the slow
604 production. We also speculate that the carbon substrate from the leachates and shift in pH (filtered
605 freshwater pH = 8.34±0.10 and leachates pH = 7.51±0.10) might have stimulated a natural
606 selection, in which the new conditions disfavour the dominant community and favour the rare
607 biosphere, which leads to slow production.

608 The production of CO₂ continues to increase throughout the incubation (Fig 7) with a large
609 production peak at 580 days of incubation, indicating that the microbes had already entered the
610 exponential phase, with the development and rapid growth of a new adaptive microbial
611 population, which concurrently increases microbial activity. A slowdown from the increase in
612 production of CO₂ was observed after 580 to 810 days of incubation. This could indicate a change
613 in microbial physiological state from activity and growth to maintenance, following depletion of
614 labile C and LMWOA (X₆, X₇, X₉, and X₄) in the culture (Figs 4, 9, and 11).



615 Furthermore, the key cause for this decline might include a reduction in quantity and quality of
616 inorganic nutrients or competition between microbes for depleted degradable C resources (X_9 and
617 X_4) and with the dominance of the DOM pool by highly aromatised recalcitrant C (X_3 and X_2) as
618 indicated by SEC data (Fig 4 and 11). Also, the shift in the DOM composition might lead to a
619 microbial community shift (Wu et al., 2018). We have no direct evidence to discuss this
620 community shift, but it could be that the populations that were growing on substrates are now
621 exhausted and stopped growing, or growing more slowly, while others growing on less labile
622 substrates have been and continue steadily growing the whole time, but more slowly than others.
623 In a similar observation, Wu et al. (2018) described that the microbial composition was closely
624 related to DOM turnover: microbial community in the initial stages of the incubation was
625 influenced by relatively labile tannins (i.e., organic acids) and protein-like compounds, while in
626 the later stage, the community composition evolved correlated with less labile lipids and lignin
627 like compounds.

628



629

630 Figure 9: Eigenanalysis of the Correlation Matrix (ECM) score plot shows the correlation (90 %) of the
 631 samples based on their DOM compositions. The ECM analysis relates relatively similar samples based on
 632 their compositions. The analysis (panel A) reflected that the post-incubation inoculated sample groups (A+F
 633 and B+F) are slightly different from the filter control (Blk+F). Both are significantly different from the
 634 positive control. This suggests comparable (between the sample inoculated group and filter control) and
 635 different (with the positive control) metabolic by-products in the respective microcosms. The first
 636 component in the biplot (panel B) represents pre-incubation compounds in the leachates, showing the
 637 dominance of LMW groups (with X4 contributing the largest portion) with very low amounts of
 638 macromolecule groups (X1 and X3) and LMW X5. It also displayed post-incubation compounds to the
 639 positive control, showing the dominance of lignin-like macromolecule X2. The post-incubation to the
 640 sample inoculated group in panel B displayed the dominance of macromolecule groups (X1 and X3) and
 641 LMW X5 with incubation. Note: the blue dots in panel B represent groups of samples as indicated in panel
 642 A.

643 Highly aromatized molecular weight neutrals were observed to be dominating the DOM pool as
 644 the experiment progressed to termination in the inoculated sample, filter control and positive
 645 control microcosms. The macromolecules X₁, X₃, and highly aromatised X₂ appear to be



646 recalcitrant to biodegradation, with X_3 continuously dominating the compound groups while X_2
647 dominated compound groups in the positive control (Figs 4, 9 and 11). The observed difference
648 in response of the DOM pool to microbial activity in the sample inoculated microcosm and
649 positive control may be related to the difference in the DOM source (DOM in the positive control
650 is mainly terrestrial, while the DOM in the leachates is marine) and composition. The freshwater
651 DOM composition is generally dominated by macromolecular humic fractions (Aukes et al.,
652 2021), while marine DOM is dominated by LMW fractions (Kye et al., 2021). This suggests the
653 potential for different metabolic pathways, as DOM type modifies the microbial population and
654 their metabolic activities (Wani et al., 2022). The dominance of a recalcitrant C pool (X_2) with
655 admixture LMW compounds (X_8 and X_9) in the positive control and the X_3 in the inoculated
656 sample group, with the addition of microbial-derived recalcitrant (condensed aromatics X_5)
657 compounds, suggest different metabolic pathways followed by microbial communities in these
658 different microcosms.

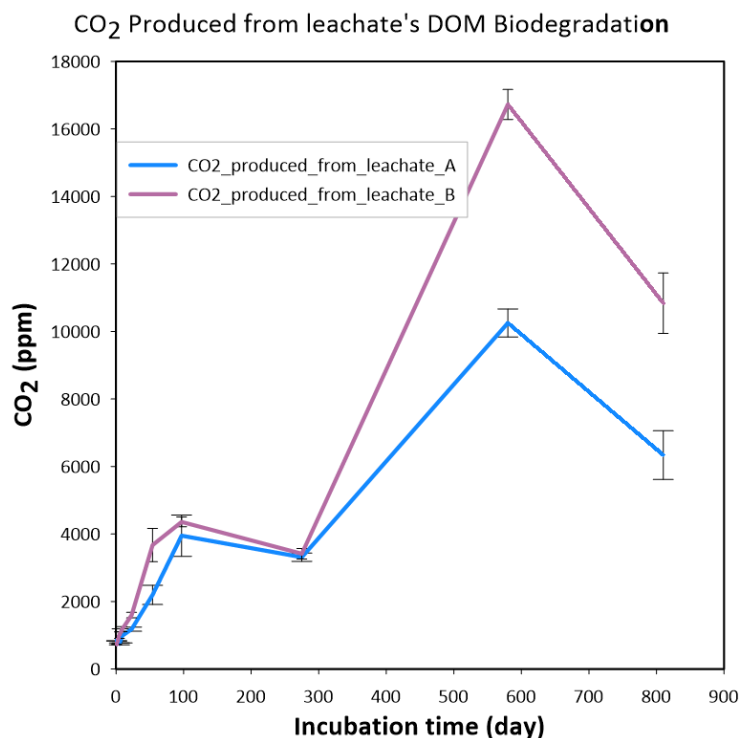
659 The dominance of highly aromatic humic-like macromolecule group X_2 in the filter control
660 microcosm reflects comparable patterns to the positive control (Fig 4), confirming a related
661 freshwater DOM origin. The filter might have contributed the same freshwater DOM to the
662 inoculated samples' microcosm. Based on the degradation of the inoculum filters we observed in
663 the sample's inoculum and filter microcosms, the nylon filter not only carried over the freshwater
664 OM but also supplied C itself. Based on our assumptions, there were three potential sources of C
665 in the microcosms: the inoculum filter, necromass and leachates. The necromass C contribution
666 was calculated/estimated as 4.19 μg . This amount is inconsequential compared to the large
667 amount of CO_2 generated in the leachate microcosms.

668 For the filter contributions, it is difficult to estimate the amount of carbon inoculum filter released
669 in the microcosms and whether it continuously releases DOM with depletion of DOM due to
670 microbial degradation in the microcosm. Furthermore, we need to know the amount of carbon
671 remaining on the filter (Undissolved) after the incubation. In this research, since all the inoculum
672 filters underwent similar treatment in the sample's inoculum and filter control microcosm, we
673 assumed that the filter may contribute a relatively similar amount of C to all inoculated
674 microcosms.

675 To remove the filter contributions, we subtract the filter microcosms from the sample microcosms.
676 The quantity of CO_2 measured in the filter control headspace microcosm was subtracted from the
677 inoculated samples' microcosm. The calculated results are presented in Fig 10, reflecting
678 maximum concentrations of CO_2 of ~ 11000 and 17000 ppm from the degradation of DOM in
679 leachate A and B, respectively. Fig 11 showed filter-corrected DOM fractional compositions,
680 derived from subtracting compound groups' contribution of the filter control from the inoculated

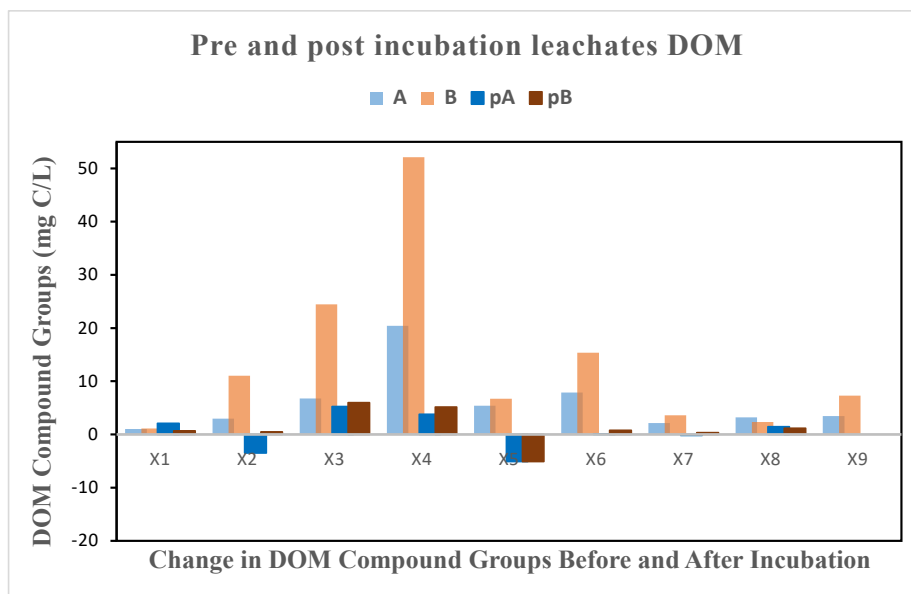


681 leachate microcosms. It exhibited clear utilisation across the DOM groups of the leachates during
682 the entire incubation, indicating the transformation and utilisation of the DOM toward less
683 biodegradable fractions (X_3 and X_1). Also, the X_4 might be the remnant that was yet to be
684 degraded, and X_8 might be biodegradation remnants or by-products (Wu et al, 2018). The negative
685 value of the compound groups X_2 and X_5 observed from the post-incubation filter corrected results
686 (Fig 10), might be DOM groups that solely come from the freshwater DOM collected on the
687 surface of the inoculum filter.



688

689 Figure 10: The amount of CO₂ produced from the leachate's degradation over time (filter control corrected).



690

691 Figure 11: Post-incubation DOM fractional compositions (filter corrected). Key: pA = (A+F – Blk+F) and
 692 pB = (B+F – Blk+F). Note: The data were derived from subtracting the compounds group's contribution
 693 from the filter control from the inoculated leachate microcosms.

694 Nitrogen (N) is an integral component of DOM (Wiegnerand Seitzinger, 2004). Dissolved organic
 695 nitrogen (DON) is encompassed a continuum of compounds ranging from macromolecules (high
 696 molecular weight molecules), e.g., polypeptides, to low-molecular-weight monomers, e.g., amino
 697 acids and urea (Antia et al., 1991; Zehr and Ward, 2002), and hence can provide both C and N for
 698 microbial communities (Antia et al., 1991; Zehr and Ward, 2002; Berman and Bronk, 2003; Ghosh
 699 and Leff, 2013). In this study, the relative proportion of DON macromolecule compounds
 700 decreased gradually during incubation (Fig. 6), indicating that these DON compounds were likely
 701 continuously utilised by microorganisms when labile C was limited/ or simultaneously utilised,
 702 serving as supplementary C and N sources. The drastic decreases in the C/N ratio to incubation
 703 time across the microcosms indicate the mineralisation and biotransformation of the DOM pool
 704 (by removing carbon in the form of CO₂ from the DOM molecules, thereby decreasing carbon
 705 contents).

706 Low methanogenic activity observed in the sample inoculated microcosm with relatively similar
 707 headspace CH₄ concentration compared to the uninoculated control, suggested abiotic CH₄
 708 production, presumably through low-temperature serpentinization and Fischer-Tropsch Type
 709 reactions (Duc, 2010; Okland et al., 2014; Moore et al., 2025). These low-temperature
 710 methanation reactions are described as likely to occur even at low pressure for wide geological
 711 settings (Etiopie and Lollar, 2013). The higher CH₄ produced by the Blk+F microcosm than the
 712 leachate inoculum microcosm may be due to methanogenic activity. However, the produced



713 methane observed in the inoculated samples microcosm may not directly be linked to the
714 biodegradation of the leachates, because of the pronounced methane produced by the inoculated
715 filter control.

716 The important limitation of this microcosm experimental approach used in this research is that it
717 is associated with an inorganic carbon sink. The headspace measurement does not account for the
718 dissolved inorganic carbon (a certain amount of produced CO₂ might get dissolved/reacted in the
719 reaction mixture in the microcosm). And the activities due to lithotrophs and methanotrophs may
720 sink a certain amount of produced CO₂ and CH₄, which the microcosm's headspace measurement
721 cannot account for.

722 Generally, in this experiment, it is observed that the DOM derived from Cretaceous marine black
723 shale contributed to stimulating bioactivity and generating biogenic GHGs. The degradation of
724 the DOM from the leachates generated approximately 11000 and 17000 ppm of CO₂ from the
725 degradation of 13.77 and 31.68 mg/g DOM of leachate A and B (Fig 11). The higher CO₂
726 concentration generated in leachate B may be linked to the amount of DOM mineralised. The
727 biodegradation also transformed the DOM toward less biodegradable fractions (X₃ and X₁). The
728 observed X₄ might be the remnant that was yet to be degraded, and X₈ might be biodegradation
729 leftovers or by-products from the microbial metabolic process.

730 **4.1 Microbial Interactions with DOM in Geothermal Reservoirs:** 731 **Implications for UKGEOS Mine Water Systems**

732 Building on these results, drilling of the subsurface may likely lead to the generation of biogenic
733 GHG from the biodegradation of mobilised aged rock-derived DOM. The microbes introduced to
734 the subsurface during drilling, upon meeting favourable conditions in the borehole, can return to
735 their full metabolic activity. An uncontrolled growth of the microbial community might lead to
736 corrosion of the casing and formation of scales due to their metabolic by-products, changing the
737 fluid chemistry in the perturbed subsurface reservoir (Inagaki et al. 2003; Westphal et al. 2019).
738 The formation of highly recalcitrant macromolecules (X₁) and other remnant products (X₃, X₄,
739 and X₈) might be a good example of the biodegradation products that may contribute to the
740 alteration of the subsurface fluid chemistry and pose a threat to the plants. Biofilm formation from
741 microbial growth can affect the flow rates and injectivity, due to the clogging of pores and filters
742 (Brehme et al. 2020; Ábel et al. 2021), and even heat exchangers might lose their capacity as
743 biofilms within the heat exchanger create an insulating effect at the surfaces where they are
744 formed (Sand 2003). Hence, studying the DOM-microbial interaction in establishing and
745 operating geothermal plants is critical.

746 The DOM type and source in the UKGEOS geothermal mine water site are strikingly complex
747 and heterogeneous. However, this experiment generated insight into how both the terrestrial and



748 aged rock-derived DOM were transformed and mineralised in microbial activities. Hence,
 749 applying this approach to the UKGEOS samples might give more detailed information, which
 750 may be needed in the safe exploitation of this unconventional energy source.

751 5.0 Conclusions

752 The findings of this research have demonstrated that marine black shale-derived DOM stimulates
 753 microbial degradation that generates GHG, and the microbial activities bio-transformed and
 754 changed the property and molecular composition of DOM. SEC analyses of water-leached DOM
 755 from original shale and post-microcosm experiments identify the DOM groups of both sets of
 756 samples that are accessible to immediate microbial activity and recalcitrant C groups and
 757 document trajectory changes in DOM composition. Turnover and mineralisation of DOM have
 758 been observed, suggesting degradation of LMWOA and labile C by surface microorganisms. The
 759 data suggest that microbial metabolic activity affected the transformation of individual DOM
 760 groups (macromolecules X₁, X₂ and X₃, and LMWOA X₄), while producing compounds that
 761 contribute to the X₅ DOM fraction, resulting in dynamic changes in DOM property and
 762 composition induced by microbial interaction. Our results demonstrated that C cycling in an
 763 environment was orchestrated by a microbial activity that utilised specific C pools in available
 764 DOM. In engineered perturbed environments, such interactions between DOM and
 765 microorganisms dictate the ultimate fate of C.

766 6.0 Appendices

767 Appendix A: Mean CO₂ concentration and standard error of the triplicate measured in each
 768 microcosm headspace.

Sample ID	Day(s)																	
	1	5		9		23		54		97		275		580		810		
	$\bar{X}_{CO_2} \pm SE$ (ppm)	$\bar{X}_{CO_2} \pm SE$ (ppm)	$\bar{X}_{CO_2} \pm SE$ (ppm)	$\bar{X}_{CO_2} \pm SE$ (ppm)	$\bar{X}_{CO_2} \pm SE$ (ppm)	$\bar{X}_{CO_2} \pm SE$ (ppm)	$\bar{X}_{CO_2} \pm SE$ (ppm)	$\bar{X}_{CO_2} \pm SE$ (ppm)	$\bar{X}_{CO_2} \pm SE$ (ppm)	$\bar{X}_{CO_2} \pm SE$ (ppm)	$\bar{X}_{CO_2} \pm SE$ (ppm)	$\bar{X}_{CO_2} \pm SE$ (ppm)	$\bar{X}_{CO_2} \pm SE$ (ppm)	$\bar{X}_{CO_2} \pm SE$ (ppm)	$\bar{X}_{CO_2} \pm SE$ (ppm)	$\bar{X}_{CO_2} \pm SE$ (ppm)	$\bar{X}_{CO_2} \pm SE$ (ppm)	
Air	445	-	468	-	462	-	443	-	430	-	436	-	316	-	414	-	375	-
A	1119	321	1345	63	1326	70	1389	117	1328	154	1559	167	1231	108	1326	125	1326	204
B	1080	21	1082	12	1073	10	1142	11	1087	23	1287	29	1058	9	1125	13	1062	12
A + Aut.F	1139	240	1380	6	1377	14	1499	10	1577	14	2126	10	1674	35	2974	77	4795	162
B + Aut.F	1147	12	1192	7	1219	19	1447	16	1545	32	2115	60	1708	102	3666	291	5897	577
A- Inorg	416	4	418	8	423	12	582	60	604	51	729	47	510	30	792	26	779	24
B- Inorg	428	15	422	15	435	25	631	71	600	99	772	51	535	22	778	47	717	49
A + F-Inorg	569	11	765	13	815	16	1095	65	1259	106	1842	169	1427	144	2404	315	2802	390
B + F-Inorg	558	7	884	31	1037	107	1308	151	1515	175	2119	222	1606	156	2611	321	3135	438
A + F	1307	4	1532	27	1873	211	2451	224	4242	577	7596	917	8630	122	107113	4414	120779	1720
B + F	1195	7	1792	97	2068	75	2866	245	5717	785	8002	449	8733	155	113585	4443	125278	1462
Blk	457	9	460	13	464	14	592	39	541	39	580	48	431	25	689	37	664	69
Blk + F	472	16	789	171	890	170	1270	164	2047	292	3644	305	5317	969	96860	8155	114441	2565
F+OA+FW	1015	8	14887	756	22374	202	29193	602	31718	539	51778	1071	53736	1010	70355	1336	70462	1270

769
 770 Where; \bar{x} is the mean CO₂ (ppm) of the triplicates produced in the microcosm headspace and SE is a
 771 standard error calculated from the triplicates.

772 A = leachate A, B= leachate B, F= filter, Blk= blank, Aut. F= autoclave filter, A-inorg = leachate A without
 773 inorganic amendments, B+F-Inorg= leachate B + Filter without inorganic amendment, F+Org+FW=
 774 microcosm containing filter, fresh water with organic amendments.

775



776 Appendix B: Mean CH₄ concentration and standard error of the triplicate measured in each
777 microcosm headspace.

Sample ID	Day(s)													
	1		5		9		23		54		580		810	
	\bar{x} CH ₄ ± SE (ppb)		\bar{x} CH ₄ ± SE (ppb)		\bar{x} CH ₄ ± SE (ppb)		\bar{x} CH ₄ ± SE (ppb)		\bar{x} CH ₄ ± SE (ppb)		\bar{x} CH ₄ ± SE (ppb)		\bar{x} CH ₄ ± SE (ppb)	
Air	3579	-	3565	-	3560	-	3188	-	3227	-	2936	-	3509	-
A	4616	376	4153	129	4120	122	7516	181	4934	49	3314	836	3145	119
B	3958	124	3897	130	4543	79	7027	257	5797	47	2373	24	3630	501
A + F	3856	66	3796	56	6147	47	6132	216	5298	439	5432	73	5661	227
B + F	3886	144	3808	120	5419	815	5830	135	5148	1056	5413	107	7775	1966
Blk	3892	181	3832	135	3641	69	3244	283	3879	196	2323	53	2127	16
Blk + F	4083	134	4225	286	7146	185	6847	44	5965	568	16211	4417	199413	21971
A + Aut.F	4265	441	4167	399	7117	137	6284	215	6275	38	4660	458	4873	1433
B + Aut.F	3781	2	3717	4	6545	37	5275	79	5349	111	4467	419	5599	28
A- Inorg	4170	230	4063	217	7754	184	7512	284	6164	1038	3274	923	3146	1057
B- Inorg	4140	71	4067	63	7702	59	7266	155	7235	30	2930	635	2735	648
A + F-Inorg	3906	123	3838	111	7098	238	6839	177	7215	130	3887	84	3866	106
B + F-Inorg	3808	40	3755	14	6824	529	4657	1082	6576	433	3919	62	3817	20
F+ OA + FW	3796	14	3720	13	5642	126	4375	91	5623	69	3906	62	3797	22

778

779 Where \bar{x} is the mean CO₂ (ppm) of the triplicates produced in the microcosm headspace, and SE is a
780 standard error calculated from the triplicates

781 **7.0 Code and data availability**

782 All data generated and analysed during this study, including DOM composition measurements,
783 greenhouse gas (CO₂ and CH₄) incubation results, and LC-OCD outputs, are available from the
784 corresponding author upon reasonable request. The analytical protocols and microcosm setup
785 procedures are described in detail within the manuscript to ensure reproducibility. No proprietary
786 code was used; all calculations and data processing were performed using standard laboratory
787 software (ChromeCalc for LC-OCD integration and instrument-specific packages). In line with
788 the Copernicus data policy, datasets supporting the findings of this study will be archived in an
789 open repository and linked upon publication.

790 **8.0 Author contribution**

S/N	Author	Contribution
1	Muhammad Sabiu Jibrin	<ul style="list-style-type: none"> ➤ Conceptualization ➤ Data curation ➤ Formal Analysis ➤ Funding Acquisition ➤ Methodology development ➤ Writing and reviewing
2	Abdulhamid Dahiru	<ul style="list-style-type: none"> ➤ Contributed to conceptualisation ➤ Contributed to methodology design ➤ Data analysis ➤ Co in writing and reviewing

791



792 **9.0 Competing interests:**

793 The authors declare that they have no conflict of interest.

794 **10.0 Statement on Inclusivity in Global Research**

795 This research was conducted at the Lyell Centre, Heriot-Watt University, Scotland, using shale
796 samples collected from the Tarfaya Basin, Morocco. We acknowledge the importance of
797 international collaboration and the ethical responsibility of working with geological materials
798 sourced from diverse regions. Our study was carried out with respect for the scientific and cultural
799 contexts of both locations, aiming to generate insights that contribute to a globally inclusive
800 understanding of subsurface carbon flux and microbial interactions relevant to geothermal energy
801 systems.

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808 this project.

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