

# Measurement reports: Chemical characterization of cloud water at Monte Cimone (Italy). Impact of air mass origin and assessment of atmospheric processes.

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## Abstract:

In this article, we present the results of the chemical and microbiological characterization of clouds water collected at Monte Cimone (CMN) in Italy at 2165 m a.s.l. during the MC3 (Molecular Composition of Clouds at mt. Cimone) campaign, which took place in October 2024. Twenty-six cloud samples are analyzed. Chemical analyses, including ions, oxidants, trace metals, and microbiological analyses with cell counting, are performed. The chemical characterization and back-trajectories analysis reveal that Mt. Cimone is a site under the influence of marine air masses, coming mainly from southern Europe and from the Mediterranean region. During the measurement campaign, 3 sampling periods are identified: period (1) October 07-10, with air masses mainly originating from Spain and Atlantic Ocean with a majority of  $\text{Cl}^-$  and  $\text{Na}^+$  that are characteristics of marine origin; period (2) October 16-18 with air masses originating from North of Africa, impacted by a Saharan dust event with a high concentration in  $\text{Ca}^{2+}$ ; period (3) October 22-23 marked by air masses originated from southern Italy under polluted influence with a high concentration in  $\text{NO}_3^-$ . Across the campaign, DOC ranged from 1.6 to 4.3  $\text{mgC L}^{-1}$ , while  $\text{H}_2\text{O}_2$  reached its maximum during period 3 ( $71.8 \pm 35.4 \mu\text{mol L}^{-1}$ ), highlighting enhanced oxidant levels under polluted conditions. This study paves the way to further scientific campaigns intended to better comprehend cloud water composition at Mt. Cimone.

## 1 Introduction

The chemical composition of the atmosphere is modulated by complex reactions and has a direct impact on the global climate, air pollution, and thus human health ([Pöschl, 2005](#)). The effect of clouds on atmospheric chemical composition is poorly understood. Clouds cover almost 70 % of the Earth's surface and are one of the greatest sources of uncertainty in climate models ([Ceppi et al., 2017](#)). Due to their multiphasic nature, clouds are characterized as a complex, dynamic, and transformative medium ([Herrmann et al., 2015](#)).

Inorganic and organic compounds from anthropogenic or natural sources are abundant in the atmosphere as gases and aerosol particles. Once emitted in the atmosphere, these compounds undergo multiple transformations, including photochemical reactions and gas to particle conversion (Herrmann et al.,

2015). Soluble gaseous compounds dissolve in cloud droplets, and soluble aerosol particles act as cloud condensation nuclei (CCN), contributing to cloud droplet composition. This medium therefore participates to the atmospheric transport of compounds emitted on the Earth's surface depending on their solubility and lifetime. In addition, the multiphase nature of cloud can favor chemical reactions that would not take place in the gas phase, unless the time scale is sufficiently long ([Herrmann, 2003](#)). The transformations in cloud can lead to the modification of chemical and physical properties of aerosols such as oxidation state and hygroscopicity ([Dominutti et al., 2022](#); [Herrmann et al., 2005, 2015](#)). Due to their high reactivity, they can be a source of secondary organic aerosol ("aqSOA") and inorganic compounds ([Ervens, 2015](#)).

Similarly, in addition to the chemical reactions that occur naturally in clouds, microbiological processes also can impact its composition ([Deguillaume et al., 2008](#)). Cloud droplets contain biological matter such as dead cells, cell fragments, but also living microorganisms such as fungi and yeast ([Amato et al., 2005](#); [Bauer et al., 2002](#); [Liu et al., 2023](#)). These living microorganisms come from different environments, mainly natural sources such as soil, plants, and sea surfaces ([Mohler et al., 2007](#)), and are suspended in the air by aerosolization. Once in the atmosphere, they may interact with clouds as ice nucleating particles ([Joly et al., 2015](#)), contributing to the formation and lifetime of clouds, through their potential role in triggering precipitation.

Although previous studies have been carried out to investigate the chemical composition of cloud water, the cloud chemistry knowledge is limited by the event-driven nature of cloud. In particular, dissolved organic matter (DOM) concentration and composition in cloud water represent a matter of interest in the frame of a changing atmosphere. Efforts to understand the chemical composition and reactivity of cloud water DOM have been made since the late 1980s ([Weathers et al., 1988](#)), with studies carried out on different continents such as Europe ([Brege et al., 2018](#); [Brüggemann et al., 2005](#); [Deguillaume et al., 2014](#); [Herckes et al., 2013](#); [Renard et al., 2020](#); [Van Pinxteren et al., 2005](#); [van Pinxteren et al., 2016](#)), Asia ([J. Li et al., 2017](#); [T. Li et al., 2020](#); [Xu et al., 2019](#)) and America ([Collett et al., 2008](#); [Desyaterik et al., 2013](#)). Based on long term database, efforts have been made to classify clouds into different categories such as polluted, continental or marine ([Deguillaume et al. 2014](#); [Renard et al., 2020](#)), using

80 in particular content of inorganic ions. The recent work from Lawrence et al. (2023) highlighted that sulphate is no longer the driving factor for cloud water chemistry, which is now dominated by base cations, reactive nitrogen species, and organic compounds. To date, there are very few infrastructures dedicated to the long term study of bio-physico-chemical processes. In this frame, the puy de Dôme observatory (PUY) in the Auvergne region of France is one of the rare such site in Europe (Baray et al., 2020). A few others exist worldwide, including the Schmücke Mountain in Germany (Van Pinxteren et al., 2005; van Pinxteren et al., 2016), Whiteface Mountain in New York (Lance et al., 2020; C. E. Lawrence et al., 2023) and Mt Tai in China (Guo et al., 2012; X. Shen et al., 2012).

The present study focuses on the aqueous phase of clouds collected during the MC3 campaign at Monte Cimone (CMN) in Italy. This is the first cloud collection campaign ever conducted at this site, with the  
90 aim of investigating how the origin of air masses influences both the chemical composition of clouds and the microbial biomass present in cloud water.

## 2 Materials and method

### 2.1 *Monte Cimone site description*

The MC3 (Molecular Composition of Clouds at Mt. Cimone) field campaign was performed from  
95 October 7 to 23 at the Italian Climate Observatory “Ottavio Vittori”, located at Monte Cimone, (44.19°N, 10.69 °E, 2165 m asl). Due to its location and high altitude, CMN is representative of Southern European / Mediterranean free troposphere. It is strongly influenced by the long-range transport of air masses including Saharan dust events (Duchi et al., 2016; Vogel et al., 2025), but it is also influenced by local and regional processes, such as vertical transport of Po valley pollution during summer months  
100 (Cristofanelli et al., 2021; Giulianelli et al., 2014; Unsworth & Fowler, 1988) and frequent new particle formation events (Mazzini et al., 2025). It is the only high mountain station for atmospheric research both South of the Alps and the Po basin (<https://www.isac.cnr.it/cimone>). In collaboration with the Italian Air Force (CAMM), this observatory is managed by the Institute of Atmospheric Sciences and Climate of the National Research Council (CNR-ISAC). It is a global station of the Global Atmospheric  
105 Watch (GAW) program by the World Meteorological Organization (WMO) as well as part of ACTRIS

(Aerosols, Clouds, and Traces gases Research Infrastructure) and ICOS (Integrated Carbon Observation System) Research Infrastructures.

## 2.2 *Cloud water sampling procedures*

Each sample is labelled with the date in the format *dd/mm* and a letter to indicate the sequential sample.

110 Details are reported in Table S1. Clouds are formed at the summit of the site during the advection of frontal systems or by orographic uplift of moist air. Sampling is restricted to non-precipitating clouds (except for sample 23/10 A), under warm cloud condition, with temperature always above 2.2 °C. Two cloud impactors BOOGIE ([Vaitilingom et al., 2025](#)) are used: one for the physico-chemical characterization and the other for microbiological analysis. Before each sampling both collectors are  
115 cleaned using ethanol 70% and then rinsed with MilliQ water. The cloud collector dedicated to microbial measurements is sterilized by high pressure and temperature. During the campaign, a blank is collected on 15/10 by spreading MilliQ water on the clean cloud collector in sunny conditions, to determine possible contamination already present on the surface of the impactor. After collection, cloud water from the non-sterilized impactor is directly filtered using a 0.22 µm H-PTFE filter (Whatman) to remove  
120 particles and microorganisms, and then split into different aliquots and stored in the fridge (4°C) or freezer (-25°C). pH is also measured immediately after collection and checked with a Hanna pH meter after the campaign. These collection procedures are similar to previously conducted, validated and published measurement campaigns ([Deguillaume et al., 2014](#); [Dominutti et al., 2022](#)).

## 2.3 *Physico-chemical analyses*

125 The concentrations of the major organic (formic, acetic, and oxalic acids) and inorganic ions ( $\text{Ca}^{2+}$ ,  $\text{K}^+$ ,  $\text{Mg}^{2+}$ ,  $\text{Na}^+$ ,  $\text{NH}_4^+$ ,  $\text{Cl}^-$ ,  $\text{SO}_4^{2-}$  and  $\text{NO}_3^-$ ) are measured by ion chromatography, using a Thermo ICS5000+ instrument, more details are available in Text S1. Dissolved organic carbon is measured with a Shimadzu TOC-VCPH/CPN analyzer without dilution. TOC measurements are obtained by calculating the difference between total carbon (TC) and inorganic carbon (IC) measurements. As the samples are  
130 filtered prior to analysis, the particulate organic carbon fraction is removed; consequently, the total organic carbon (TOC) is equivalent to the dissolved organic carbon (DOC). The hydrogen peroxide

concentration is measured with the spectrofluorimetric method based on the reactivity of p-hydroxyphenylacetic acid with horseradish peroxidase (Bader et al., 1988). Trace metals analysis is conducted using Inductively Coupled Plasma Mass Spectrometry instrument (ICP-MS, Agilent 7500).  
135 As the cloud impactor is made in aluminum, this element is not quantified in the samples. The analyzed trace metals are Ag, As, Be, Bi, Cd, Co, Cr, Cu, Fe, Ga, Li, Mg, Mn, Mo, Ni, Pb, Pd, Rb, Sb, Sn, Sr, Tl, U, V, W, and Zn. The result in section 3.5 presents the comparison between blank and cloud water samples: as some samples had values below the blank, the blank was not subtracted. As the blank is prepared in clear sky conditions, atmospheric particles, with concentration of  $766.6 \pm 24.0 \text{ \#/cm}^3$ , may  
140 dissolve into MilliQ water, thereby increasing the concentrations of certain trace metals. More details about the physico-chemical analysis can be found in Text S1.

#### 2.4 Complementary atmospheric observations

A set of complementary atmospheric observations is implemented in the data analysis to characterize, comprehensively, the cloud events. These observations include meteorological parameters, trace gases  
145 ( $\text{O}_3$ ,  $\text{CO}$ ,  $\text{SO}_2$ ,  $\text{NO}_x$ ), and aerosol particles concentration (Cristofanelli & Montagnuti, 2025; Cristofanelli & Trisolino, 2021). The data, measured at a time resolution of 1 minute are averaged according to the duration of the cloud events. The corresponding measurements are reported in Figure S1.

##### 2.4.1 Trace gases and meteorology

An UV-absorption analyzer (49i, Thermo Scientific) is used to measure *in-situ* atmosphere mole fraction  
150 of surface  $\text{O}_3$ . The UV analyzer is regularly calibrated (roughly every 3 months) with a laboratory transfer standard (Tei 49i-PS, Thermo Scientific) referred to the WMO calibration scale SRP#2 and the standard uncertainty of unbiased measurements is usually below  $\pm 1 \text{ nmol mol}^{-1}$ .

Atmospheric nitric oxide (NO), nitrogen dioxide ( $\text{NO}_2$ ), and  $\text{NO}_x$  (defined as  $\text{NO} + \text{NO}_2$ ), are measured by a chemiluminescence analyzer (Teledyne T200UP) equipped with a photolytic converter (Blue Light  
155 Converter) with a detection limit of  $0.01 \text{ nmol mol}^{-1}$  for both NO and  $\text{NO}_2$ . Roughly every 7 days, zero and span calibrations are carried out for NO by using an external zero air source and dilution of NO

mixture in N<sub>2</sub> (5.0 ppm ±3.5%). To determine the efficiency of the NO<sub>2</sub> converter, a gas phase titration (GPT) module is used after NO calibration.

160 Atmospheric CO measurements ([Cristofanelli & Montaguati, 2025](#)) are carried out in the framework of ICOS by using a Cavity Ring Down Spectrometer analyzer (G5310, Picarro). Within ICOS, atmospheric observations of CO are carried out in a standardized way for measurement set-up, used materials, quality assurance strategy and data creation workflow (see [Hazan et al., 2016](#)). As deduced by the calculation of measurement bias, long-term and continuous repeatability, the CO measurement uncertainty is below ±1 nmol mol<sup>-1</sup>.

165 At CMN, standard meteorological variables ([Cristofanelli & Trisolino, 2021](#)) are observed by an ultrasonic anemometer (WMT-702, Vaisala), air-temperature and relative humidity by a sensor HMP-155 (Vaisala) and atmospheric pressure by a barometer PTB-210 (Vaisala).

#### 2.4.2 Dust transport identification with an optical particle counter

The aerosol sampling is performed through a heated whole air inlet (WAI) in stainless steel, which  
170 ensures efficient sampling of aerosol below 20 µm. The optical particle counter (OPC, Grimm® model 1.108) is used to measure the particle size distribution in the optical diameter range of 0.3-20 µm with a time resolution of 1 min. The resulting size distribution is divided into fine (diameter < 1µm) and coarse (diameter > 1 µm) particles. Starting from 24 h averages of the coarse particle number concentration, the dust transport days are identified following the methodology presented in [Duchi et al., 2016](#) and [Vogel et al., 2025](#). From the number size distribution of coarse particles, we calculated the  
175 mass concentration of coarse particles (PM<sub>COAR</sub>) with 1 minute time resolution assuming a size dependent particle density ([Wittmaack, 2002](#)).

#### 2.4.3 Aerosol chemical composition

180 Measurements of non-refractory (NR) PM1 chemical components (i.e. organics, SO<sub>4</sub><sup>2-</sup>, NO<sub>3</sub><sup>-</sup>, NH<sub>4</sub><sup>+</sup>, Cl<sup>-</sup>) are continuously performed by a Time-of-Flight Aerosol Chemical Speciation Monitor (ToF-ACSM) (Aerodyne Research Inc.) ([Fröhlich et al., 2013](#)), operating at 6 min time resolution and equipped with a PM1 lens cut-off. The inlet used is the same described in Paragraph 2.4.2. In brief, PM1

enters the instrument passing through a three-way valve system (switching between sampling and background signal), a 120  $\mu\text{m}$  critical orifice and aerodynamic lenses which beam the PM1 to the vaporizer ( $\sim 600^\circ\text{C}$ ), placed at the end of the vacuum chamber. Here the NR-PM1 components vaporize and are then impacted by a 70 eV electron beam which generates ion fragments, detected by the Time-of-Flight mass analyzer (ToFwerk ETOF) depending on their mass-to-charge ratio. Compensation of particle losses due to bouncing after impacting onto vaporizer is performed by correcting the ACSM data for the composition-dependent collection efficiency (CDCE) (Middlebrook et al., 2012). Ionization efficiency of nitrate in ions  $\text{pg}^{-1}$  and relative ionization efficiencies (RIEs) for sulphate and ammonium are obtained by calibrations; the final mean values are 44.6, 0.615 and 3.65 ions  $\text{pg}^{-1}$ , respectively. In this work, the ACSM is used to quantify the aerosol particle mass concentration in the sub micrometric range (PM1). To easily differentiate the aerosol phase from the ions liquid concentration, the contribution of the various ions to PM1 are called:  $\text{Org}_{\text{PM}}$  for organics,  $\text{SO}_4^{2-}_{\text{PM}}$  for sulphate,  $\text{NO}_3^-_{\text{PM}}$  for nitrate,  $\text{NH}_4^+_{\text{PM}}$  for ammonium and  $\text{Cl}^-_{\text{PM}}$  for chloride. The total particle concentration ( $\text{PM}_{\text{TOT}}$ ) is then calculated as the sum of PM1, measured by the ACSM, and  $\text{PM}_{\text{COAR}}$  derived with the OPC. Similarly in the aqueous phase, they are called:  $\text{SO}_4^{2-}_{\text{CW}}$  for sulphate,  $\text{NO}_3^-_{\text{CW}}$  for nitrate,  $\text{NH}_4^+_{\text{CW}}$  for ammonium and  $\text{Cl}^-_{\text{CW}}$  for chloride, where “CW” means cloud water. The mass concentration of equivalent black carbon (eBC) is derived from the aerosol absorption coefficient measured with a Multi-Angle Absorption Photometer (MAAP, model 5012, Thermo Scientific) using a fixed mass absorption cross-section ( $10 \text{ m}^2 \text{ g}^{-1}$ ; Zanatta et al., (2016). While the operating principle of the MAAP and its limitations are provided by Petzold and Schönlinner (2004) and Petzold et al. (2005), its use and corrections at CMN is described by Renzi et al. (2025).

### 2.5 *Air masses back-trajectory analysis*

CAT (Computing Atmospheric Trajectory Tool) is a 3D model which simulates the trajectories of air masses (forward/backward) using as input wind fields from ECMWF ERA-5 (European Centre for Medium-Range Weather Forecast) reanalysis (Baray et al., 2020). For this work, wind fields are extracted every 3h with a spatial resolution of  $0.25^\circ$  in latitude ( $70^\circ\text{N}$ ,  $-20^\circ\text{S}$ ) and longitude ( $60^\circ\text{E}$ ,  $-60^\circ\text{W}$ ), on 23 vertical pressure levels between 200 and 1000 hPa. CAT integrates a topography matrix

210 at a spatial resolution of around 10 km. Clusters of 216 back-trajectories are calculated every hour during the cloud sampling using starting points distributed within  $\pm 0.2^\circ$  in latitude and longitude around the measurement site. The vertical starting altitude of the back-trajectories is deduced from the pressure measured at CMN summit considering the hydrostatic equilibrium. In the present work, the temporal resolution of back-trajectory clusters is 15 min and their total duration is 72 h.

## 215 *2.6 Statistical analyses: Principal Component Analysis*

A principal component analysis (PCA) is performed using 28 cloud samples (including blank and 23/10 rain) collected at the Monte-Cimone site. The PCA matrix is centered and standardized using mean and standard deviation. The correlation matrix is calculated directly on R-studio software (Lê et al., 2008) using the function `res.pca <- PCA(Data,scale.unit=TRUE)`, where Data is our dataset. The option  
220 `scale.unit=TRUE` centres and scales the variables, which means that PCA is performed on the correlation matrix. The software used is R-studio with the programming language R. Concentrations of  $\text{SO}_4^{2-}$ ,  $\text{Cl}^-$ ,  $\text{NO}_3^-$ ,  $\text{Na}^+$ ,  $\text{NH}_4^+$ ,  $\text{Mg}^{2+}$ , and  $\text{Ca}^{2+}$  presented in Table S2 are treated manually to obtain categories (polluted, marine, continental clouds) based on ion concentration dissimilarities.

## *2.7 Microbial analysis*

225 Total microbial concentration is determined using flow cytometry. Immediately after sampling, 5% glutaraldehyde (final concentration: 0.5%) is added to cloud water. Triplicate subsamples are stored at  $4^\circ\text{C}$  (< 15 days) before “SYBR green I” staining and analysis with a LSR FORTRESSA X-20 cytometer (BD Biosciences).

Culturable microbial cells are investigated spreading 100  $\mu\text{L}$  of cloud water on Petri dishes containing  
230 R2A medium (3 replicates per sampling). Petri dishes are incubated at  $20^\circ\text{C}$  during 4 days under aerobic conditions, in the dark. Colony forming units (CFUs) are then counted differentiating bacteria-like CFUs (yeast or bacteria) from mycelial fungi.

## **3 Results and discussion**

### **3.1 Characterization of cloud samples: air mass history and chemical composition**

235 The study of cloud composition, coupled with dynamical analyses, such as air masses back-trajectory, gives useful information on the possible contribution of anthropogenic and biogenic sources. During the campaign, the air masses arrived at CMN from three main different sectors, namely West (W), South (S) and South-East (S-E). This enables the definition of three periods: (1) *Period 1*, from 07 to 10/10, with air masses mainly coming from Spain and the Atlantic Ocean, crossing the Iberian peninsula; (2)  
240 *Period 2*, from 16 to 18/10, with air masses coming from North Africa, strongly impacted by Saharan dust; (3) *Period 3*, from 22 to 23/10, marked by air masses coming from Ionian and Levantine basin and travelling above the Italian peninsula (Figure 1 and Figure S2).

### 3.1.1 Period 1: marine influence from the Atlantic sector

Combining back trajectory analysis with the chemical composition of cloud water, a clear marine  
245 influence is evident. During *Period 1*, concentrations of chloride ( $\text{Cl}^-_{\text{CW}}$ ) and sodium ( $\text{Na}^+_{\text{CW}}$ ), which are indicators of marine influence, were  $56.0 \pm 80.3 \mu\text{mol L}^{-1}$  and  $50.6 \pm 67.8 \mu\text{mol L}^{-1}$ , respectively (Figure S5, Table S2). These values are comparable to those observed in marine clouds at PUY, (Renard et al., 2020) and lower than those reported for other marine and continental sites (Dominutti et al., 2022;  
250 J. Li et al., 2017; Reyes-Rodríguez et al., 2009; Triesch et al., 2021). The marine influence is even more pronounced in single cloud samples. For example, samples 09/10 B and 10/10 B show  $\text{Cl}^-_{\text{CW}}$  and  $\text{Na}^+_{\text{CW}}$  concentrations similar to highly marine clouds collected at PUY (range of  $12.2 - 394.0 \mu\text{mol L}^{-1}$  for  $\text{Cl}^-_{\text{CW}}$  and  $18.5$  to  $678.6 \mu\text{mol L}^{-1}$  for  $\text{Na}^+_{\text{CW}}$ ), as well as to those collected in Puerto Rico (range of  $151.0 - 841.0 \mu\text{mol L}^{-1}$  for  $\text{Cl}^-_{\text{CW}}$  and  $158.0$  to  $738.0 \mu\text{mol L}^{-1}$  for  $\text{Na}^+_{\text{CW}}$ ) (Reyes-Rodríguez et al., 2009) or at the Reunion Island (Dominutti et al., 2022). When focusing on the TIC (total ion content), defined as  
255 the calculated sum of the major anions (inorganic and organic) and cations present in clouds, the average for *Period 1* is around  $176 \pm 188 \mu\text{eq L}^{-1}$ . This value is similar to the TIC reported in (Deguillaume et al., 2014) for clouds collected in PUY with a marine origin. The marine influence is also confirmed by the sea-salt sulphate [ $\text{ss-SO}_4^{2-}$ ] fraction in cloud water, which averaged around 40% and reached up to 70% in individual samples during *Period 1*. This is remarkably higher than in the other periods, where  
260 the average contribution remained below 10%, as shown in Figure S3. The reliability of the sea salt sulphate calculation (Keene et al., 1986) is confirmed by the high correlation with the  $\text{Cl}^-$  concentration

in the aqueous phase. Complementary atmospheric observations such as low concentration of eBC, CO, NO and NO<sub>2</sub> support the occurrence of “clean” marine air masses at the measurement site (Figure 2). The distribution of inorganic ions (Cl<sup>-</sup>, NH<sub>4</sub><sup>+</sup>, NO<sub>3</sub><sup>-</sup>, and SO<sub>4</sub><sup>2-</sup>) in cloud water (CW) is similar in proportions to those observed in particulate matter (PM). We observed a depletion of Cl<sup>-</sup><sub>PM</sub> and NO<sub>3</sub><sup>-</sup><sub>PM</sub>, in the aerosol phase compared to the cloud aqueous phase, particularly visible for samples 09/10 A and B, 10/10 A and B, 17/10 B, 22/10 and 23/10 B and C. The air masses corresponding to these samples are characterized by a strong marine influence, with Cl<sup>-</sup><sub>CW</sub> and NO<sub>3</sub><sup>-</sup><sub>CW</sub> mainly present as inorganic ions (NaCl and NaNO<sub>3</sub>), which are not detected by ACSM, since the temperature of the ionization source (600°C) is below the temperature of volatilization of these inorganic salts (Ng et al., 2011).

### 3.1.2 Period 2: Saharan dust influence from north Africa

Intense dust events are not unusual at CMN in October (Vogel et al., 2025), providing a valuable opportunity to study the impact of Saharan dust on cloud water composition, which remains poorly documented (M. Shen et al., 2024; Valle-Díaz et al., 2016).

Saharan dust events observed in the Mediterranean are characterized by a large fraction of super micrometric particles (Denjean et al., 2016) and contains a large amount of Calcium (Pérez-Martínez et al., 2020). The aerosol observations allowed identifying Saharan dust event on October 16<sup>th</sup> and 17<sup>th</sup> (Figure S2e), when the concentration of coarse particles represented the major component of the particulate matter mass with the highest concentration of all periods (3.28 µg m<sup>-3</sup>). Air mass transport from northern Africa is confirmed by low O<sub>3</sub> and CO values (38.9 nmol mol<sup>-1</sup> and 98.5 ppb, respectively; see Figure S6), typical of dust-rich air masses with low contributions from combustion sources (e.g., Duchi et al., 2016). Cloud samples collected during this period present high concentrations of Ca<sup>2+</sup><sub>CW</sub>, up to 62.3 µmol L<sup>-1</sup>, and low concentrations of DOC, with an average of 1.6 ± 0.4 mgC L<sup>-1</sup>. Notably, the concentration of Ca<sup>2+</sup><sub>CW</sub> is likely underestimated, as some Ca<sup>2+</sup> may be present as insoluble mineral aerosols and thus remain on the filter. The concentration of Ca<sup>2+</sup><sub>CW</sub> is higher than those observed for polluted clouds collected in Poland (Błaś et al., 2008) or in China (J. Li et al., 2017). All ion concentrations of *Period 2* in cloud water are low, confirming that other sources are negligible, besides Saharan dust, for these samples.

On October 16<sup>th</sup>, the first day of the Saharan event, all ion concentrations decreased over the day. The  
290 maximum concentration of  $\text{Ca}^{2+}_{\text{CW}}$  is  $62.3 \mu\text{mol L}^{-1}$  for sample 16/10 A and decreased to  $12.7 \mu\text{mol L}^{-1}$   
for the last sample 16/10 G (around 5 hours later). A decrease of  $\text{NO}_3^-_{\text{CW}}$  and  $\text{SO}_4^{2-}_{\text{CW}}$  concentrations  
is also observed, passing from 16.6 to  $4.5 \mu\text{mol L}^{-1}$  for  $\text{NO}_3^-_{\text{CW}}$  and from 15.6 to  $4.2 \mu\text{mol L}^{-1}$  for  $\text{SO}_4^{2-}_{\text{CW}}$ .  
Studying a cloud sample influenced by a Saharan event over time provides new insights into the  
impact of such events on cloud water chemistry, as these analyses remain limited.

295 The Saharan origin of  $\text{Ca}^{2+}_{\text{CW}}$  is confirmed by its high correlation with the concentration of coarse  
particles, which is enriched during Saharan dust events ([Vogel et al., 2025](#)) (Figure S7). Notably, the x  
intercept is not zero, suggesting that a consistent fraction of calcium may remain in the fine fraction  
mass. The samples G and F have very low concentrations in both the PM1 and PM<sub>COAR</sub> modes.

After a clear sky period of 6 hours, on October 17<sup>th</sup>, two cloud water samples are collected consecutively  
300 in the morning. The concentration of  $\text{Ca}^{2+}_{\text{CW}}$  increases, compared to samples 16/10 F and 16/10 G, with  
an average of  $49.1 \pm 5.5 \mu\text{mol L}^{-1}$ .

The analysis of the images on the webcam from CMN shows the absence of clouds between 2:31:00  
UTC to 8:22:00 UTC on October 17<sup>th</sup> (Figure S8). The trace gases, ACSM and OPC data strengthen the  
fact that clouds collected on 16/10 and on 17/10 belong to two different events, both from Saharan air  
305 masses, but with a different intensity: the concentration of supermicron particles measured on 17/10 ( $3 \mu\text{g m}^{-3}$ )  
is less than a quarter of the peak concentration on 16/10 ( $13 \mu\text{g m}^{-3}$ ) (Figure S4). There is a  
slight variation of the back-trajectory of the air mass between the two events: in the back-trajectory  
corresponding to samples 17/10 A and B, the air masses travelled through Tunisia at low altitude, in the  
boundary layer, before crossing the Tyrrhenian Sea to reach the CMN. As also testified by the higher  
310 atmospheric CO, NO<sub>2</sub> and O<sub>3</sub>, this can result in a continental/polluted influence and lead to higher  
concentrations of  $\text{NO}_3^-$ ,  $\text{SO}_4^{2-}$ , and  $\text{NH}_4^+$ , comparable to those observed for continental samples in PUY  
([Deguillaume et al., 2014](#)).

The Saharan dust episode ended on October 18<sup>th</sup>, and the concentration of the different ions returned to  
lower values, with average concentrations of  $10.6 \pm 4.6 \mu\text{mol L}^{-1}$  for  $\text{Ca}^{2+}_{\text{CW}}$ ,  $8.2 \pm 1.7 \mu\text{mol L}^{-1}$  for

315  $\text{NO}_3^-_{\text{CW}}$ ,  $2.1 \pm 0.4 \mu\text{mol L}^{-1}$  for  $\text{SO}_4^{2-}_{\text{CW}}$  and  $8.2 \pm 0.4 \mu\text{mol L}^{-1}$  for  $\text{NH}_4^+_{\text{CW}}$ . This fast decrease after a Saharan dust event is also observed in a study of rainwater collected in Spain (Oduber et al., 2020), reaching a concentration of  $\text{Ca}^{2+}_{\text{CW}}$  of around  $180.0 \mu\text{mol L}^{-1}$  in rainwater collected during the Saharan dust event, and below  $40.0 \mu\text{mol L}^{-1}$  the day later. However, cloud and rain samples are not equivalent, as they may be influenced by different accumulation mechanisms and/or washout processes. The aerosol  
320 particles measurements mirror the results obtained for cloud water: the concentrations of sub and super micrometric aerosol particles decrease drastically on 18/10 compared to the previous days. Also the atmospheric gases consistently traced a change in the features of the collected air masses, with a marked increase of  $\text{O}_3$  and  $\text{NO}_2$ .

### 3.1.3 Period 3: mixed anthropogenic influence from the Western mediteranean

325 Finally, the cloud and aerosol particles concentrations in *Period 3* showed a completely different behavior. This period is represented by only three cloud samples and two events: October 22<sup>nd</sup> and 23<sup>rd</sup>, presenting different back-trajectories. The air masses are mainly coming from S-E sector, with origin in the Ionian Sea at low altitudes and crossing the Italian Peninsula from S-E to N-W at higher altitudes, but in the boundary layer.

330 This is confirmed by an average concentration of  $\text{NO}_3^-_{\text{CW}}$  of  $60.5 \pm 39.9 \mu\text{mol L}^{-1}$  and  $\text{SO}_4^{2-}_{\text{CW}}$  of  $42.7 \pm 38.4 \mu\text{mol L}^{-1}$ . The average concentration of  $\text{NH}_4^+_{\text{CW}}$  is also higher than the first two periods:  $49.8 \pm 30.2 \mu\text{mol L}^{-1}$ . Also, the submicrometric aerosol concentration showed the highest campaign values with the PM1 ( $1.67 \mu\text{g m}^{-3}$ ), eBC ( $0.07 \mu\text{g m}^{-3}$ ) and organic carbon ( $0.42 \mu\text{g m}^{-3}$ ) concentration being roughly four times higher than period 1 and 2.

335 Looking deeper into the different samples, the average concentration is clearly driven by 22/10. This sample presents concentrations of  $\text{Cl}^-_{\text{CW}}$  and  $\text{Na}^+_{\text{CW}}$  of  $254.7 \mu\text{mol L}^{-1}$  and  $224.7 \mu\text{mol L}^{-1}$ , respectively, but also high concentrations of  $\text{NO}_3^-_{\text{CW}}$ ,  $\text{SO}_4^{2-}_{\text{CW}}$ , and  $\text{NH}_4^+_{\text{CW}}$ , reaching  $115.7 \mu\text{mol L}^{-1}$ ,  $99.1 \mu\text{mol L}^{-1}$  and  $91.6 \mu\text{mol L}^{-1}$ , respectively.

This is supported by the back-trajectory of 22/10, as the air masses come firstly from lower altitude of  
340 the Mediterranean Sea, before crossing the S-E of Italy at low altitude and being enriched by ions

resulting from anthropogenic activities such as  $\text{NO}_3^-$ ,  $\text{SO}_4^{2-}$  and  $\text{NH}_4^+$ . The aerosol particles and trace gases measurements agree with these results. Similarly, particulate matter concentration in 22/10 is the highest in the entire sample series, with a remarkable example of  $\text{SO}_4^{2-}\text{-PM}$   $1.91 \mu\text{g m}^{-3}$ , confirming a strong anthropogenic influence. The low  $\text{NO}/\text{NO}_2$  ratio (Figure S6) and  $\text{NO}_3^-\text{-PM}$  average concentrations (0.08  $\pm$  0.01  $\mu\text{g m}^{-3}$ , the highest of the campaign) would suggest the occurrence of relatively aged anthropogenic emissions.

In the past, pH was a determinant criterion to classify the air masses, being more acidic for more polluted clouds (Herckes et al., 2013). However, this criterion has changed in the past few years, as pH has become more homogeneous in the cloud water samples with less variability (Pye et al., 2020). In cloud water samples from CMN, the average pH is  $6.1 \pm 0.8$ , and quite homogeneous between the samples. However, it decreases to 4.1 on 22/10, coinciding with peak  $\text{SO}_4^{2-}\text{-PM}$  and a low neutralization degree, which confirms that this sample is under continental influence with anthropogenic input.

### 3.2 Statistical analysis

Principal component analysis (PCA) is used to explore patterns and relationships among cloud water constituents (variables) and samples (objects) collected at CMN, by identifying combinations of variables (loadings) that explain the largest variance in the dataset (Figure 3). In this PCA the first two factors represent 92.1% of the variability of the dataset. PC1, which represents 72.9% of the information, is driven by the concentration of  $\text{Mg}^{2+}\text{-CW}$ ,  $\text{Cl}^-\text{-CW}$ ,  $\text{Na}^+\text{-CW}$ ,  $\text{NH}_4^+\text{-CW}$ ,  $\text{SO}_4^{2-}\text{-CW}$  and  $\text{NO}_3^-\text{-CW}$ , while PC2, which represents 19.5% of the information, is mainly driven by the concentration of  $\text{Ca}^{2+}\text{-CW}$ . The analysis of the hierarchical clustering combined to the PCA, reveals 5 main categories of samples statistically different described as follows: (1) samples classified as “plain” with a low ions content, (2) samples classified as “Saharan event”, (3) samples classified as “polluted/continental”, (4) samples classified as “highly marine” and; (5) samples classified as “highly marine/polluted” (Figure 3).

The total ion concentration of the various samples in the first group (in green) is comparable to the blank's ion concentration. Sample 23/10 A falls within this category: it is the only sample obtained during rainy conditions, which affected the total ion concentration through dilution processes. Samples 23/10

B and C, which are collected during the same cloud event without rain, are categorized as polluted/continental, do not exhibit dilution, and present higher concentrations of  $\text{NO}_3^-_{\text{CW}}$ ,  $\text{SO}_4^{2-}_{\text{CW}}$ , and  $\text{NH}_4^+_{\text{CW}}$ . Moving to the second category (in yellow), samples from 16/10 A to E are driven by the  $\text{Ca}^{2+}_{\text{CW}}$  concentration. As the concentration of  $\text{Ca}^{2+}_{\text{CW}}$  decreases over time through dilution, it is possible to plainly observe a moving tendency (red arrow) from the second category to the first category, which contains samples 16/10 F and G. In the third category, samples 17/10 A and B are grouped with samples 23/10 B and C. According with the back-trajectories corresponding to these four samples (4 events), they both travel above the Italian peninsula before reaching CMN. However, on the PCA, samples 17/10 A and B are driven by  $\text{Ca}^{2+}_{\text{CW}}$  concentration. This is consistent with the associated back-trajectories, which both start from northern Africa, bringing dust to CMN (Figure S2f). Samples 23/10 B and C, on the other hand, do not have the same origin and, thus, they are not directly correlated to  $\text{Ca}^{2+}_{\text{CW}}$  concentration. Finally, the last two categories include samples 09/10 A and B, which are under marine influence and mainly driven by  $\text{Mg}^{2+}_{\text{CW}}$ ,  $\text{Na}^+_{\text{CW}}$ , and  $\text{Cl}^-_{\text{CW}}$  concentrations. Two isolated samples completed the dataset: sample 10/10 B, which has back-trajectory that is comparable but not identical to 09/10, and mainly driven by  $\text{Na}^+_{\text{CW}}$  and  $\text{Cl}^-_{\text{CW}}$  concentrations, attesting a strong marine influence. Sample 22/10 has similar concentrations of  $\text{Cl}^-_{\text{CW}}$ ,  $\text{Na}^+_{\text{CW}}$ , and  $\text{Mg}^{2+}_{\text{CW}}$ , but higher concentrations of  $\text{NO}_3^-_{\text{CW}}$ ,  $\text{SO}_4^{2-}_{\text{CW}}$ , and  $\text{NH}_4^+_{\text{CW}}$ , which can be explained by the back-trajectory of the air mass, coming from S-E and travelling above Italy in the boundary layer.

### 3.3 Oxidants in cloud water

Radicals or other oxidant species induce the transformation of organics in the cloud aqueous phase. Ionic radicals, such as  $\text{SO}_4^{\cdot-}$  (sulphate radical anion) are produced only in the aqueous phase, while neutral radicals, like  $\text{HO}^{\cdot}$  (hydroxyl radical),  $\text{NO}_3^{\cdot}$  (nitrate radical) and  $\text{HO}_2^{\cdot}$ , can be produced in the aqueous phase or transferred from the gas phase into the aqueous phase. The redox potential increases following the order  $\text{NO}_3^{\cdot} \approx \text{SO}_4^{\cdot-} < \text{HO}^{\cdot}$  (Bianco et al., 2020). Although radicals are not measured in samples collected at CMN, some of their precursors, such as hydrogen peroxide ( $\text{H}_2\text{O}_2$ ),  $\text{NO}_3^-$  and  $\text{Fe}^{2+}/\text{Fe}^{3+}$  are quantified. The concentration of total iron is below  $0.3 \mu\text{mol L}^{-1}$ , as confirmed by ICP-MS measurements (Paragraph 3.4, Table S4), and is too low for an efficient photochemical activity and  $\text{HO}^{\cdot}$

production through Fenton and photo-Fenton reactions. The average concentration of H<sub>2</sub>O<sub>2</sub> is 37.8 ±  
395 29.0 μmol L<sup>-1</sup>, higher than the average concentrations measured at other mountain sites in Europe (11-  
13 μmol L<sup>-1</sup> – [Marinoni et al., 2011](#); [Bianco et al. 2015](#); [Valverde-Canossa et al., 2005](#)) and at other non-  
European sites (23-24 μmol L<sup>-1</sup> – [Mohnen & Kadlecek, 1989](#); [J. Li et al., 2017](#); [Benedict et al., 2012](#)).

The relatively low concentration of SO<sub>2</sub> could explain the high concentration of H<sub>2</sub>O<sub>2</sub> in cloud water.  
Besides, the photochemical depletion of H<sub>2</sub>O<sub>2</sub> is inhibited during autumn and winter and could be  
400 responsible of higher concentrations ([Bianco et al., 2020](#)). The highest concentration of H<sub>2</sub>O<sub>2</sub>, measured  
for sample 22/10 (104.1 μmol L<sup>-1</sup>), could be possibly linked to the direct emissions from an industry  
plant in Rosignano Solvay (LV), producing 50.000 tons/year of H<sub>2</sub>O<sub>2</sub>. The industry is located less than  
150 km away from CMN, and the air mass back-trajectory is in agreement with an input from this  
potential source.

405 Although no measurements are performed in the gas phase, if we assume that H<sub>2</sub>O<sub>2</sub> follows the Henry's  
equilibrium in cloud droplets, an average pseudo gas phase concentration can be estimated from the  
partial pressure of H<sub>2</sub>O<sub>2</sub> using Equations (1) and (2)

$$p_{H_2O_2} = \frac{[H_2O_2]_{aq}}{K_H} \quad (1)$$

$$[H_2O_2]_{gas} = \frac{p_{H_2O_2}}{p_{TOT}} \quad (2)$$

410 Considering that K<sub>H</sub> of H<sub>2</sub>O<sub>2</sub> at 278 K is 4.98×10<sup>5</sup> mol L<sup>-1</sup> atm<sup>-1</sup> ([O'Sullivan et al., 1996](#)), and p<sub>TOT</sub> = 0.77  
atm at CMN during the campaign, the resulting gaseous phase concentration is 99 ± 76 ppt, in agreement  
with the values reported by [Laj et al., \(2001\)](#) and [Vione et al., \(2003\)](#).

The main source of H<sub>2</sub>O<sub>2</sub> in the aqueous phase is the mass transfer from the gas phase ([Bianco et al.  
2020](#)), followed by the photochemical production in the aqueous phase. Samples collected during MC3  
415 show a high variability of H<sub>2</sub>O<sub>2</sub> concentrations within a single cloud event, especially concerning  
day/night sampling times. Figure 4a reports in yellow the concentration in day-time samples and in blue  
in night-time. The concentrations in night-time samples are generally higher than in day-time samples,

except for sample 16/10 G, collected at sunset. This result is in agreement with [Marinoni et al., \(2011\)](#), which highlights the importance of photochemical depletion of H<sub>2</sub>O<sub>2</sub> during the day.

420 Figure 4b reveals a linear correlation between the concentrations of NO<sub>3</sub><sup>-</sup> and H<sub>2</sub>O<sub>2</sub>, with a coefficient of determination ( $R^2 = 0.82$ ). The correlation with sulphate is also not negligible ( $R^2 = 0.64$ ). This result is in contrast with the measurements presented by ([Marinoni et al., 2011](#)) which highlights that H<sub>2</sub>O<sub>2</sub> concentration is higher in air masses with anthropogenic influences compared to those from remote areas.

### 425 **3.4 Dissolved organic carbon**

The average concentration of dissolved organic carbon (DOC) is  $2.9 \pm 2.7$  mgC L<sup>-1</sup>, similar to values measured in clouds collected under free tropospheric conditions ([Arakaki et al., 2013](#); [Deguillaume et al., 2014](#); [C. Lawrence et al., 2024](#); [Löflund et al., 2002](#); [Reyes-Rodríguez et al., 2009](#)). The low concentration observed is probably correlated to the high LWC of the cloud sampled. Although the LWC

430 is not measured during the campaign, the LWC for the last samples, collected on 23/10, is  $0.36 \pm 0.17$  g m<sup>-3</sup>. After the campaign, a Gerber PVM-100 measured the LWC for three weeks, corresponding to three cloud events. The average value is  $0.35 \pm 0.1$  g m<sup>-3</sup>, corresponding to stratus or cumulus clouds in the lower free troposphere. [Herckes et al. \(2013\)](#) reported that the DOC concentrations at LWC higher than  $0.25$  g m<sup>-3</sup> are scarce, but always below  $5$  mgC L<sup>-1</sup>, which is consistent with the measurements at CMN.

435 The DOC concentrations during the three periods identified in the previous discussion are  $4.3 \pm 3.9$ ,  $1.6 \pm 0.4$  and  $3.7 \pm 1.2$  mgC L<sup>-1</sup>, respectively. Although *Periods 1* and *3* show similar values, the concentrations in *Period 2*, corresponding to the Saharan dust event, are significantly lower. Looking at the ACSM measurements, the concentration of organic matter in *Period 2* is, in average, slightly higher than in *Period 1*, simultaneous with a remarkable increase of coarse particle. Considering that organic

440 matter may be efficiently internally mixed with Saharan dust particles ([Dall'Osto et al., 2010](#)), the insoluble internally mixed organic matter may be filtered immediately after sampling. Hence, only a small fraction of organic matter dissolves in cloud water, leading to a lower concentration of DOC for cloud samples collected in *Period 2*. Unfortunately, no measurements of the unfiltered samples (total organic carbon) are performed during the MC3 campaign.

445 Short-chain carboxylic acids such as formic, acetic and oxalic acids, and carbonyls such as formaldehyde, acetaldehyde and acrolein (aldehydes) and acetone (ketone) are measured in the samples, as reported in Table S1. The concentrations of all the compounds are homogeneous in all the samples, apart from formic acid, which is significantly lower than the average in *Period 2* and significantly higher in *Period 3*. Likewise, oxalic acid is significantly higher in *Period 1* of the campaign and significantly  
450 lower in *Period 2*. Acetic and formic acid dissolve from the gas phase and their concentrations are the result of the equilibrium, controlled by Henry's law ([Aiyuk et al., 2025](#)). Oxalic acid can be produced in the aqueous phase by oxidation of its precursors (acetic, pyruvic, glyoxylic and malonic acids) and it is considered to be a proxy of aqueous phase processing, in-cloud or in wet particles ([Carlton et al., 2007](#); [Ervens et al., 2004](#); [Lin et al., 2020](#)), leading to the conclusion that the air masses corresponding  
455 to *Period 1* are more aged than the other two. A recent article highlights that oxalic acid concentration is lower during dust episodes for PM<sub>2.5</sub> ([Yang et al., 2022](#)); interestingly, the concentration of oxalic acid is lower during *Period 2*, corresponding to dust events. The concentration of carbonyls is homogeneous for all the samples, with values of  $3.3 \pm 0.9 \mu\text{mol L}^{-1}$ ,  $1.6 \pm 0.4 \mu\text{mol L}^{-1}$ ,  $1.8 \pm 0.5 \mu\text{mol L}^{-1}$  and  $0.3 \pm 0.1 \mu\text{mol L}^{-1}$  for formaldehyde, acetaldehyde, acetone and acrolein, respectively. These  
460 values are in the range of concentrations measured on other sites ([Deguillaume et al., 2014](#); [Ervens et al., 2013](#); [van Pinxteren et al., 2005](#)).

The sum of the concentrations, in mgC L<sup>-1</sup>, of carboxylic acids and carbonyl compounds represents in average  $22 \pm 12 \%$  of the DOC, as reported in Figure 5. This percentage is comparable to values reported in other studies on organic matter ([Dominutti et al., 2022](#); [Wang et al., 2020](#)), which show that these  
465 VOCs (acids and carbonyls), although representing a non-negligible fraction, still constitute only a small part of the overall picture. This highlights the need for further investigations into the true diversity and complexity of organic matter.

### 3.5 Metals in cloud water

Twenty-six trace metals are analyzed in cloud water samples. Among them, 6 are under the detection  
470 limit: Ag, Be, Bi, Ga, Li and U (Table S4). The most concentrated element is Mg, coming mainly from biogenic sources, such as soil dust or sea salt ([Bianco et al., 2017](#)) and ranging from 1.7 to  $28 \mu\text{mol L}^{-1}$

<sup>1</sup>. These concentrations are higher than those measured for dissolved  $\text{Mg}^{2+}$  by ion chromatography, confirming that Mg is contained in particles with diameter smaller than  $0.2 \mu\text{m}$ , which is the pore size of the filter. Interestingly, the highest concentrations are measured for *Periods 1* and *3*, leading to the  
475 conclusion that Mg in Saharan dust is in the mineral form on coarse particles (Figure 6a). The second highest element is Zn, with concentrations of  $1.4 \pm 0.9 \mu\text{mol L}^{-1}$ . This element is mainly emitted by biogenic sources and shows no trend during the campaign. Fe and Cu are also present in the samples, with concentrations ranging from  $17.3$  to  $283 \text{ nmol L}^{-1}$  for Fe, and from  $25.3$  to  $596 \text{ nmol L}^{-1}$  for Cu. The highest concentration of Fe,  $0.28 \mu\text{mol L}^{-1}$ , is in the range of measurements reported by (Bianco et  
480 al., 2017) in France and (Fomba et al., 2013) in Germany, whereas Cu highest concentration,  $0.6 \mu\text{mol L}^{-1}$ , is closer to those observed in the work of (Hutchings et al., 2009) in Arizona or (Cini et al., 2002) in Italy (Figure 6b). These two compounds can influence the hydroxyl radical budget through their reactions with hydrogen peroxide ( $\text{H}_2\text{O}_2$ ) and peroxy radicals (Deguillaume et al., 2005). Ni is also present in most of the samples. It reaches its highest concentration ( $176.7 \text{ nmol L}^{-1}$ ) on 07/10 A, a sample  
485 that comes mainly from S-E of Spain and that crosses the Mediterranean Sea. Nickel can be emitted by numerous sources, such as combustion of fossil fuel in traffic and industry (Shaltout et al, 2025), but it is also a good tracer of marine influence as this element is present in trace elements for aquatic organisms such as phytoplankton (John et al., 2024).

Other trace metals are also detected: As, Cd, Co, Cr, Mo, Pb, Pd, Rb, Sb, Sn, Sr, Tl, V, and W. Sample  
490 22/10 showed the highest concentration in Sr, As, and Pb:  $137.8 \text{ nmol L}^{-1}$ ,  $7.7 \text{ nmol L}^{-1}$  and  $42.1 \text{ nmol L}^{-1}$ , respectively (Figure 6c). Those values are in the same range of samples analyzed at Mt. Brocken in Germany (Plessow et al., 2001) and slightly higher than in PUY (Bianco et al., 2017). These compounds mainly originated from anthropogenic sources such as fuel combustion, pesticides incineration or wood combustion. This result corroborates the hypothesis of a polluted influence on this sample.

495 Sb, Cd and Cr present maximum concentrations of  $1.6$ ,  $1.3$  and  $5.5 \text{ nmol L}^{-1}$ , respectively. These compounds are all released into the atmosphere by anthropogenic sources, such as industries or automobile exhausts. Nevertheless, the concentrations measured are very low compared to measurements reported on other sites, such as Mt. Brocken (Plessow et al., 2001), which is probably

explained by the overall atmospheric circulation during the campaign that is mainly from marine origin  
500 and thus not strongly affected by pollution emission from anthropogenic sources.

Finally, the Zn/Sb ratio can indicate whether the air masses are more influenced by marine source, or by  
pollution. Wilkinson et al. (1997) define a threshold value for the Zn/Sb ratio : air masses with a ratio  
higher than 80 are more influenced by marine sources, whereas those with a ratio lower than 80 are more  
influenced by pollution. Considering this ratio across the 14 cloud water samples, the Zn/Sb ratio is  
505 always higher than 80 indicating that all of air masses are under marine influence. This is consistent  
with the cloud back-trajectories, all of them always crossed the Mediterranean Sea, strongly impacting  
the samples.

### 3.6 Microbiological analysis

Living microorganisms are ubiquitous in the atmosphere. Most of them originate from vegetation, soil  
510 and sea and can travel long distances (Griffin et al., 2001). It is thus particularly important to better  
understand their impact on cloud water, as they may modify cloud composition and reactivity (Delort et  
al., 2017). As reported in the Section 2.7, cloud samples dedicated to microbiological analysis are  
collected in sterile conditions. The two collectors deployed are operated almost at the same time.  
Nevertheless, some time differences are unavoidable, mainly due to the mounting and dismounting of  
515 the collectors. The precise sampling times are reported in Table S1. Tables S5 and S6 details the  
concentrations of culturable bacteria-like CFUs and mycelial fungi, and total cells for each cloud sample  
in which a microbiological analyses are carried out.

Cell concentrations are highest during october 16<sup>th</sup> and 17<sup>th</sup> sampling events (samples A and B), ranging  
from  $2.4 \pm 0.5 \times 10^5$  to  $1.0 \pm 0.3 \times 10^6$  cells·mL<sup>-1</sup>. This peak is likely explained by a Saharan dust event  
520 occurring on those dates, which transports large quantities of particles and likely increase the number of  
particle-attached microorganisms. Interestingly, culturable cell concentration do not follow this trend,  
with only 470 culturable cells·mL<sup>-1</sup> (including both mycelium and colony-forming units), a value very  
close to the median culturable concentration for the entire campaign (median = 470 cells·mL<sup>-1</sup>; range =  
400–563 cells·mL<sup>-1</sup>). This sample is also one of only two in which bacteria-like CFUs outnumber

525 mycelium-forming units, the other being collected on 07/10. However, this pattern does not appear to be related to air mass origin, as the two samples are under Saharan and marine influence, respectively. Samples influenced by marine air masses contained  $8.0 \times 10^3$  to  $4.1 \times 10^4$  cells·mL<sup>-1</sup>, approximately one order of magnitude lower than in Saharan-influenced clouds. These concentration differences are associated with the altitude of air mass transport over the Mediterranean Sea. In comparison, polluted  
530 clouds contain between  $3.1 \times 10^4$  and  $1.4 \times 10^5$  cells·mL<sup>-1</sup>. These values are within the range typically observed in cloud water at puy de Dôme 1465 m, Central France, a site exposed to air masses from oceanic and continental influences), between  $\sim 3 \times 10^3$  to  $\sim 3 \times 10^5$  cells·mL<sup>-1</sup>. Typically, the abundance of airborne microbes increases with the influence of terrestrial sources (Mayol et al., 2017), and is higher in urban compared to continental and coastal environments (Shaffer & Lighthart, 1997). Values as low  
535 as  $< 10^3$  bacteria cells·m<sup>-3</sup> of air are reported near the surface over oceans (Mayol et al., 2017; Rossi et al., 2025). Consecutive sampling on 17/10 (2-hour interval, same cloud event) reveals marked differences in culturable microbial concentrations (20 vs 63 bacteria-like CFUs·mL<sup>-1</sup>, and 427 vs 163 mycelium-forming units for samples A and B, respectively), despite no significant variation in total cell concentrations ( $2.4 \times 10^5$  vs  $2.9 \times 10^6$  cells·mL<sup>-1</sup>). This illustrates the dynamic nature of clouds, even on  
540 short timescales: while total microbial abundance may remain relatively stable, community viability and culturability can fluctuate rapidly, likely due to environmental change or microphysical processes within the cloud. Consistently with our observations, it is typically reported that less than 1% of the total bacteria in clouds can be recovered by culture under classical conditions, while fungi often maintain higher cultivability rate (e.g., Vaitilingom et al., 2012). The ability of microbial cells to form colonies  
545 (i.e. grow on a culture medium) can be influenced by multiple factors including viability, culture conditions, and others less well-understood mechanisms leading to a viable-but-non-culturable (VBNC) state (Pinto et al., 2015). This VBNC state is often recognized in bacteria as a strategy to tolerate stressful environmental conditions (Dong et al., 2020).

Overall, cell concentrations measured at Monte Cimone are consistent with those reported at other cloud  
550 sampling sites ( $\sim 10^3$  to  $10^5$  cells·mL<sup>-1</sup>) (Amato et al., 2005; Bauer et al., 2002; Sattler et al., 2001). On average, cultivable cells represent only 0.5–6% of the total community, a proportion commonly observed

in atmospheric samples ([Amato et al., 2007](#)). Although they represent a small fraction of the total microorganisms present in samples, cultivable organisms remain essential for downstream applications such as laboratory experiments (e.g. [Vaitilingom et al., 2013](#)) or cloud simulation chamber studies (e.g. [Amato et al., 2015](#)).

#### 4 Conclusions

We present here the characterization of 26 cloud water samples collected for the first time at CMN during the MC3 campaign. This study shows that samples with similar ionic content may have distinct atmospheric origins. Consequently, a combined approach integrating both back-trajectory analysis and measurements of ion concentrations in water and aerosol particles was essential for an accurate and representative classification of cloud water samples being influenced by marine, Saharan and anthropogenic air masses.

The analysis of oxidants reveals a relationship between  $\text{H}_2\text{O}_2$  concentrations with both  $\text{NO}_3^-$  and  $\text{SO}_4^{2-}$ , showing that its concentration is higher in anthropogenic-influenced air masses than in remote ones. Moreover this work confirms that the photochemical depletion of  $\text{H}_2\text{O}_2$  is larger throughout the day, as reported by other cloud water studies.

The dissolved organic carbon (DOC) is measured for most of the samples and we found concentrations and variability similar to those presented for clouds collected in free tropospheric conditions. Short chain carboxylic acids, aldehydes and ketons represent 20% in average of the DOC concentration. An average of 78% remains unknown, strengthening the idea that more research is needed to characterize the organic matter in clouds to understand the chemical processing in the aqueous phase.

During the campaign we measured also the cloud water microbial content, providing cell concentrations and speciation into fungi and bacteria for cultivable microorganisms. The DNA analysis is also performed and will be presented in a future work, providing an additional characterisation of the atmospheric microbiota in the free troposphere.

Overall, the MC3 campaign provides a unique insight into cloud composition in the Mediterranean basin, a region where in situ observations remain scarce, thereby limiting a comprehensive understanding of cloud chemistry and aerosol–cloud interactions.

### **Data availability**

580 CO and meteorological variables at CMN are available through the ICOS Carbon Portal (<https://data.icos-cp.eu/>). The ozone and NO<sub>x</sub> data can be retrieved from the WMO/GAW World Data Center for Reactive Gases (WDCRG) hosted by NILU (<https://ebas.nilu.no/> (last access: 24 October 2022), 2023.). Dust products are available at DOI: <https://doi.org/10.71763/XDZA-FA77>, while eBC product at DOI: <https://doi.org/10.71763/itineris-hub/nfy7-yz86>. Aerosol chemical  
585 composition is available on EBAS database <https://ebas-data.nilu.no/Pages/DataSetList.aspx?key=FA24E18FDDA640B4A3E2699777225D05>.

Meteorological data are available at PID [https://hdl.handle.net/11676/-23mwxRlF7b\\_gqqmw3KyhAfi](https://hdl.handle.net/11676/-23mwxRlF7b_gqqmw3KyhAfi), and CO at PID: [https://hdl.handle.net/11676/-23mwxRlF7b\\_gqqmw3KyhAfi](https://hdl.handle.net/11676/-23mwxRlF7b_gqqmw3KyhAfi).

### **Author contribution**

590 Conceptualization: AB, AM, MZ  
Methodology: AB, PN, YW, MZ, MJ, JLP, AM, JLB  
Investigation: PN, AB, PC, DP, MZ, FV  
Resources: PN, YW, CV, PC, MJ, FC, SM, LR, AM  
Software: CV, JLB  
595 Visualization: PN, AB, PC  
Funding acquisition: PC, AB, AM, MZ  
Project administration: PC, AB, AM, MZ  
Supervision: AB, MB, LD, MZ

Writing – original draft: PN, MZ, AB

600 Writing – review & editing: AB, PN, YW, MJ, PA, PC, DP, LR, MR, MB, JLB, LD, AM, MZ

### Competing interests

The authors declare no competing interest.

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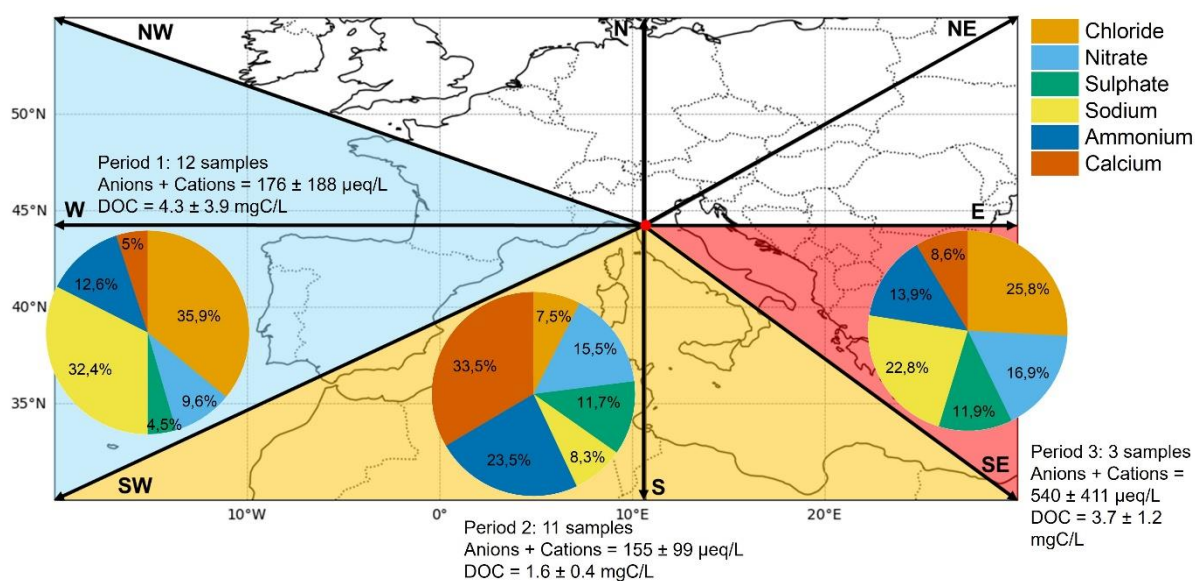
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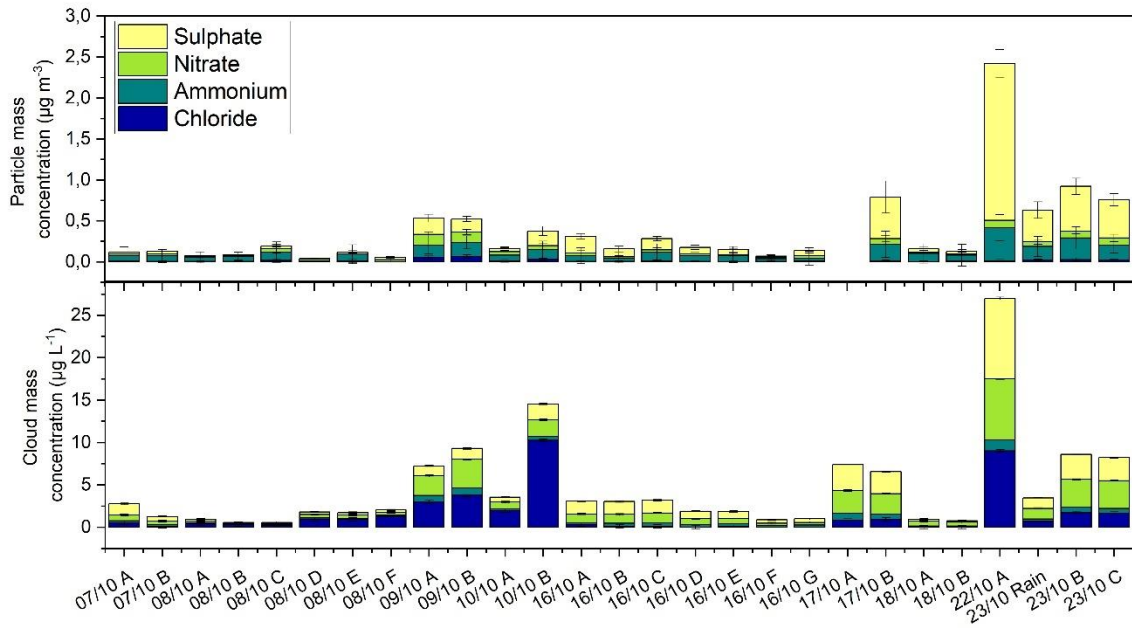
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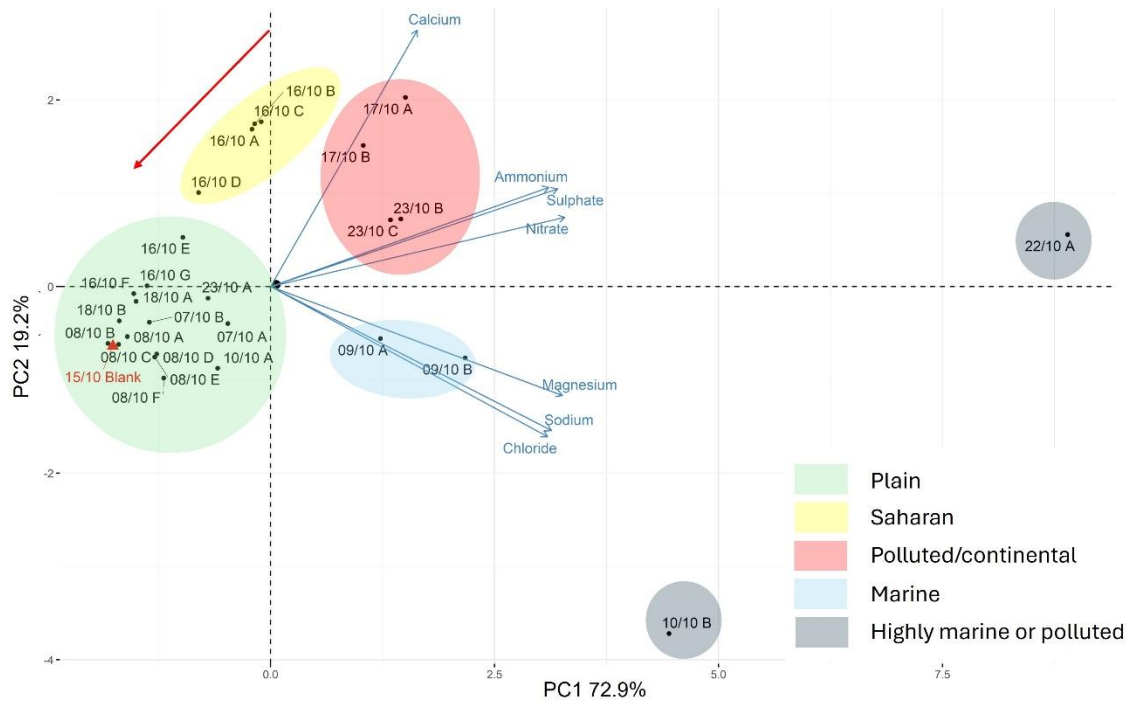


**Figure 1:** Sectors of air masses reaching the CMN observatory. Black lines depict the compass directions; the blue area shows the origin of air masses for *Period 1*, the yellow area for *Period 2*, and the red area for *Period 3*. The pie plots indicate for each period the relative contributions of main inorganic ions and cations. The number of samples collected, the ionic concentration (in  $\mu\text{eq L}^{-1}$ ) and the dissolved organic carbon (DOC) (in  $\text{mgC L}^{-1}$ ) are reported for each period.

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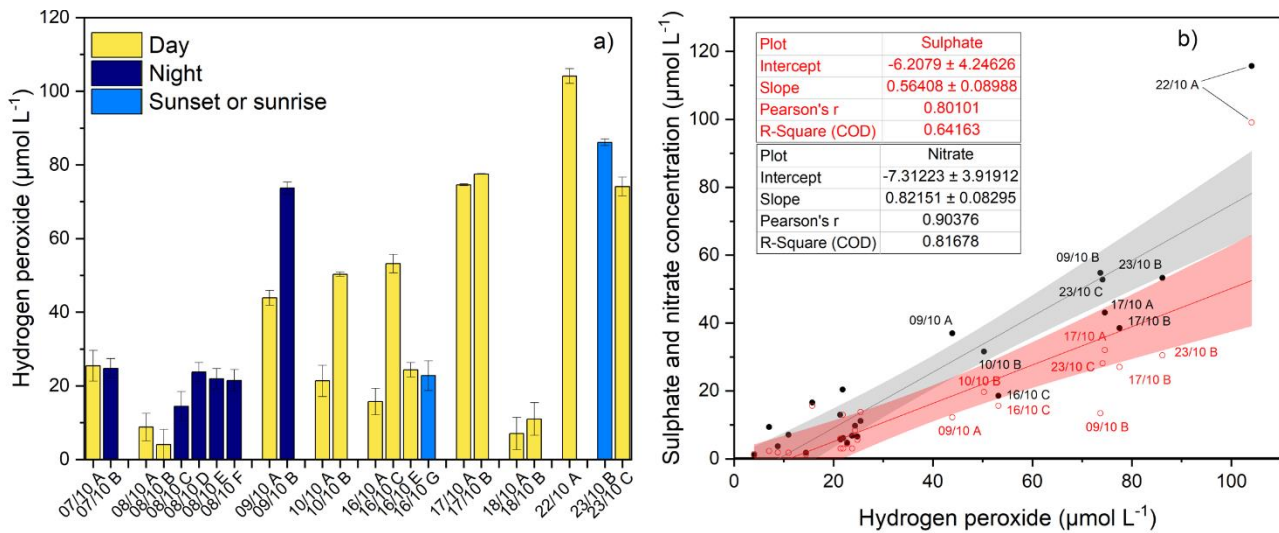


**Figure 2:** Mass concentration of main inorganic ions (chloride, nitrate, sulphate and ammonium) in aerosol (upper panel) and in cloud (lower panel) phases.



**Figure 3:** Biplot of the principal component analysis (PCA) performed on cloud samples collected at CMN. Loadings represent the ions, scores represent the samples. Colored areas depict the different classes of samples: plain, Saharan, polluted/continental, highly marine, and, highly marine or polluted.

990 The blank sample is reported in red. The red arrow highlights the depletion of  $\text{Ca}^{2+}$  during the cloud event 16/10.

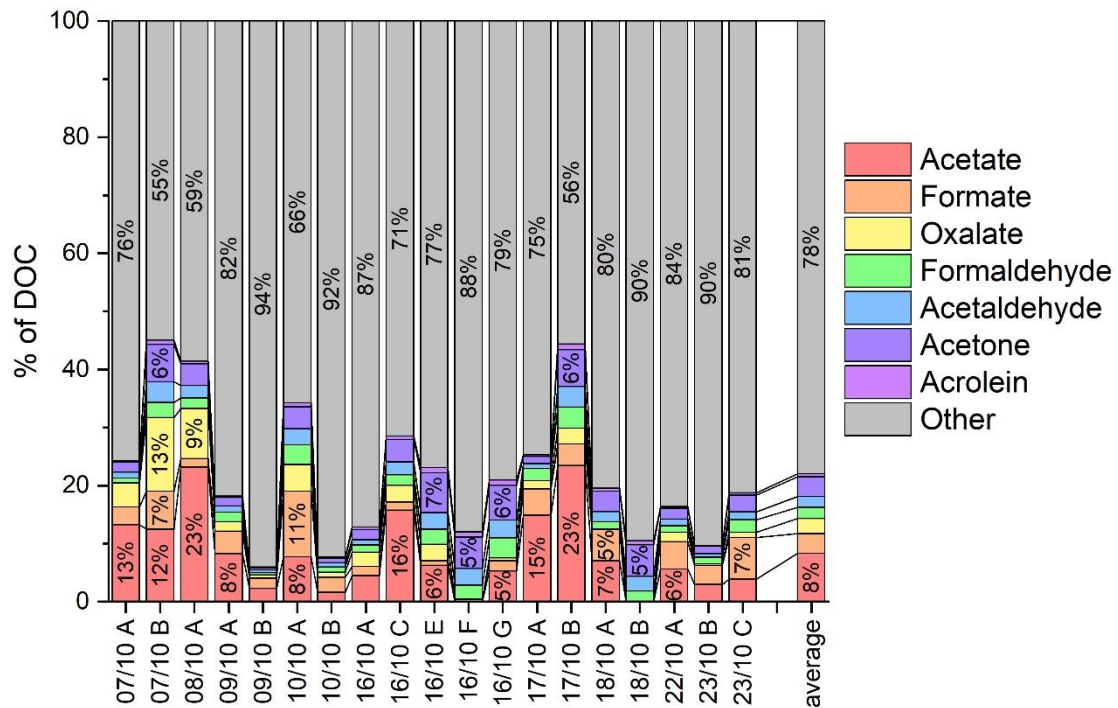


**Figure 4:** a) Boxplot of hydrogen peroxide concentration in cloud water. b) Correlation between

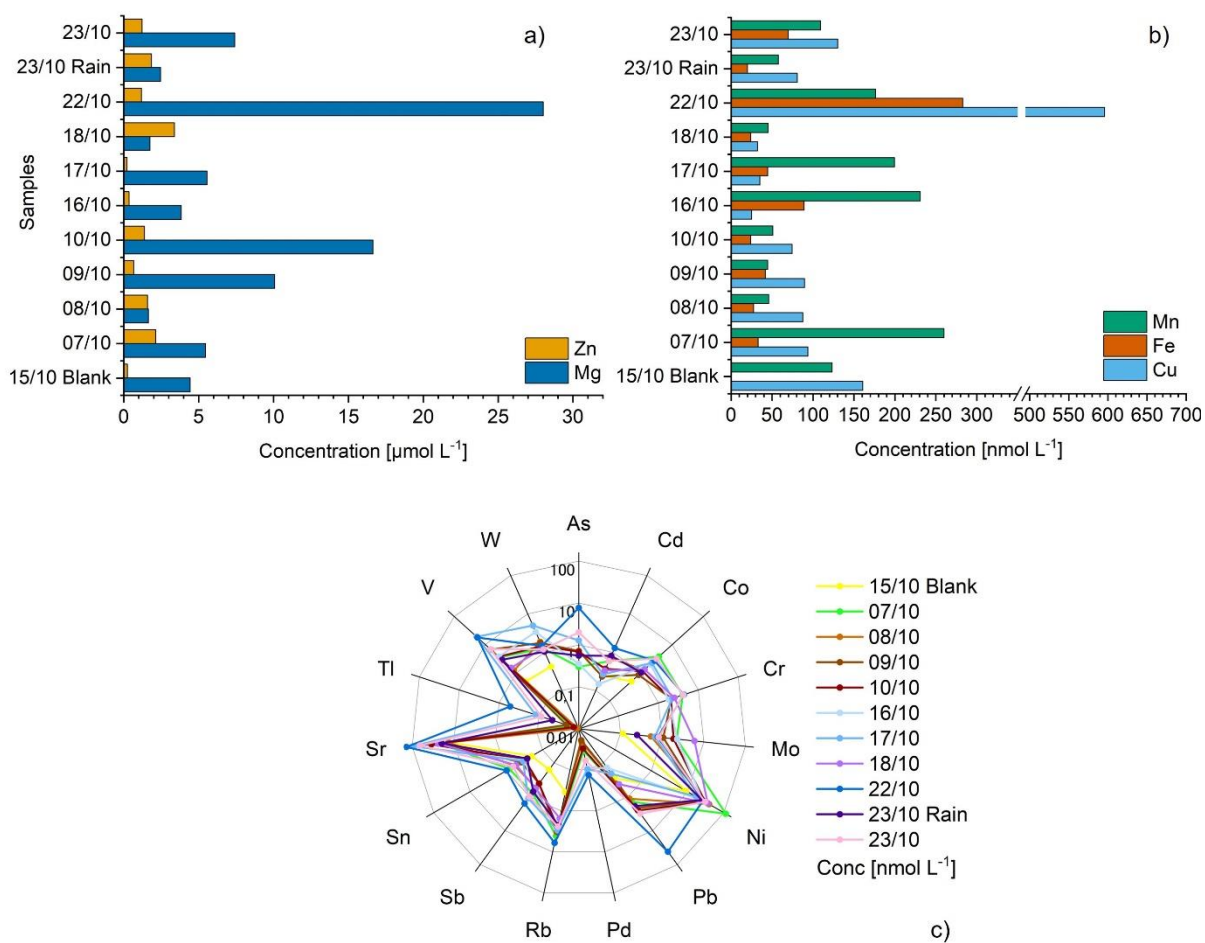
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nitrate concentration (left y-axis, black dots) or sulphate concentration (right y-axis, red circles) and hydrogen peroxide in cloud water. The linear correlation is reported in grey for nitrate and in red for sulphate, with 95% confidence intervals. Equations and coefficient of determinations are reported in the inset tables.

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**Figure 5:** Contributions of main carboxylic (acetic, formic, and oxalic acids) and carbonyl (formaldehyde, acetaldehyde, acetone and acrolein) compounds to the DOC concentration. The last bar on the right depicts the average for all the samples analyzed.



**Figure 6:** a) Bar plot showing the concentrations in  $\mu\text{mol L}^{-1}$  of Zn and Mg in each sample; b) Bar plots showing the concentrations in  $\text{nmol L}^{-1}$  of Mn, Fe and Cu in each sample; c) radar plot showing the concentrations of 15 different trace metals in  $\text{nmol L}^{-1}$  in each sample. The y-axis is reported in logarithmic scale.

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