

# Measurement report: Simultaneous measurement on gas- and particle-phase water-soluble organics in Shanghai: Enhanced light absorption of transported Asian dust

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**Abstract:** To better understand the physicochemical evolution of Asian dust particles during long-range transport, water-soluble organic compounds (WSOCs) in gas- (WSOC<sub>g</sub>) and particle-phase (WSOC<sub>p</sub>) in the spring atmosphere of Shanghai during the 2023 dust storm period (DS) and haze events (HE) were simultaneously measured with a 3-h time resolution, and characterized for their optical properties and size distribution. Our results showed that gas-to-particle-phase partitioning coefficients ( $F_p$ ) of WSOCs in DS ( $0.30 \pm 0.06$ ) was comparable to that in HE ( $0.32 \pm 0.06$ ), although both temperature and relative humidity in DS were not favorable for the partitioning, indicating a promoting role of dust particles in the transformation process of WSOC<sub>g</sub> from gas to particle phase.  $F_p$  variation was largely driven by aerosol liquid water content in HE but by aerosol acidity in DS. WSOC<sub>p</sub> and its light absorption at  $\lambda_{365\text{nm}}$  dominated at the fine mode ( $< 2.1 \mu\text{m}$ ) in non-DS period and the coarse mode ( $> 2.1 \mu\text{m}$ ) in DS, respectively. Mass absorption coefficient (MAC) of the coarse mode of WSOC<sub>p</sub> at  $\lambda_{365\text{nm}}$  in DS was  $0.8 \text{ m}^2 \text{ g}^{-1}$ , which is four times that ( $0.20 \pm 0.09 \text{ m}^2 \text{ g}^{-1}$ ) in the source region of Tengger Desert, suggesting a remarkably increase in light absorbing ability of Asian dust during long-range transport. Sharp co-increases of nitroaromatics, imidazoles, and water-soluble organic nitrogen at the coarse mode in the DS period further revealed that such an increasing MAC is mainly caused by adsorption and heterogeneous formation of light absorbing nitrogen-containing organics on the dust surface during long-range transport.

## 1 Introduction

Dust storm (DS) release up to 2 billion tons of dust particles into the global atmosphere annually, exerting significant impacts on visibility, climate, and ecological environment (Li et al., 2014; Tang et al., 2016; Wu et al., 2020; Chen et al., 2023; Ge et al., 2024). During a long range transport physicochemical properties of dust particles can significantly be changed by a series of chemical reactions especially in the downwind region of East Asia, where anthropogenic gas pollutants, such as O<sub>3</sub>, NO<sub>x</sub>, SO<sub>2</sub> and NH<sub>3</sub>, are abundant (Wang et al., 2014; Wang et al., 2015; Zhang et al., 2015; Wu et al., 2020; Wang et al., 2025).

Since the majority of oxidation products of atmospheric volatile organic compounds (VOCs) are water-soluble, gas-to-particle-phase partitioning of water-soluble organic compounds (WSOCs) is a major formation pathway of SOA in the atmosphere (Ervens et al., 2011; Liu et al., 2012; Gkatzelis et al., 2021; Lv et al., 2022b; Lv et al., 2022a). The gas-to-particle-phase partitioning process is regulated by a variety of factors such as concentrations of VOCs, NH<sub>3</sub> and NO<sub>x</sub> and their compositions, in addition to temperature and relative humidity (RH) (Hennigan et al., 2009; El-Sayed et al., 2018). In the past decade, atmospheric environment in East Asia has changed significantly due to sharp reductions in SO<sub>2</sub> and black carbon (BC) emissions in China, which is currently dominated by NO<sub>x</sub>, VOCs and NH<sub>3</sub> along with an increasing level of O<sub>3</sub> (Zhang et al., 2020; Li et al., 2021; Liu et al., 2022; Xiao et al., 2022; Wang et al., 2023; Chen et al., 2024). Dust storm from East Asia desert regions is one of the major contributors to the global atmospheric aerosols (Wang et al., 2020). Such reductions in SO<sub>2</sub> and BC in the East Asia atmospheric environment could also significantly alter the physicochemical evolution process of dust particles during long range transport by adsorption and reactions such as gas-to-particles

partitioning of WSOCs and SOA formation on the dust particle surface. Therefore, field observations on the gas-to-particle-phase partitioning of WSOCs is necessary for better understanding on SOA formation process in the current East Asia atmosphere.

A few studies have indicated that gas-to-particle phase partitioning of WSOCs in China is much more efficient than that in developed countries such as the United States, mainly due to a high level of  $\text{NH}_3$  in China (Zhang et al., 2012; Lv et al., 2022b; Lv et al., 2022a). Our recent observation found that such an enhanced gas-to-particle phase partitioning in China not only increase SOA production but also enhance atmospheric aerosol light absorption (Liu et al., 2023). Up to date, most of field investigations on WSOCs partitioning and organic aerosol light absorption have been conducted in winter with a focus on fine particle pollution (Cheng et al., 2016; Yuan et al., 2020; Lv et al., 2022b; Li et al., 2023a; Li et al., 2023b; Liu et al., 2024). In contrast, WSOC partitioning behavior and organic aerosol light absorption in East Asia during spring has rarely been investigated, where dust particles and  $\text{NH}_3$  in the troposphere are much more abundant in spring than in winter. In this study we performed a simultaneous measurement on the atmospheric gas- and aerosol-phase WSOCs in Shanghai with a focus on the heterogenous formation of SOA on dust surface and its impact on dust particle optical properties. Our work for the first time unveiled that light absorbing ability of Asian dust is remarkably enhanced during long range transport by formation of light absorbing SOA, which is termed as brown carbon, on dust particle surface.

## 2 Materials and methods

### 2.1 Samples collection

The field observation was conducted from 27 March to 21 April 2023 on the campus of East China Normal University ( $31.03^\circ$ ,  $121.45^\circ$ ), which is situated on the southern area of

Shanghai. All instruments were set on the roof of the Atmospheric Observation Station with the air inlet 10 meters above the ground. A modified in-situ gas and aerosol compositions monitor (IGAC, Model 63GA, Fortelice International Co., Ltd., Taiwan, China) was used to synchronously measure gaseous and particulate species with a 3 h interval for WSOCs and a 1-h interval for inorganics species and small organic acids, respectively (Lv et al., 2022b; Lv et al., 2022a). The IGAC monitor is coupled with ICS-5000<sup>+</sup> ion chromatography (Thermo) and a TOC/TON analyzer (model TOC/TON-L CPH, Shimadzu, Inc., Japan). Due to the difference in the online instrument detection limits, gas-phase WSOC (WSOC<sub>g</sub>) and particle phase WSOC (WSOC<sub>p</sub>) and particle-phase water-soluble organic nitrogen (WSON) were measured with a time resolution of 3 h, while inorganic species organic acids were measured with a time resolution of 1 h. In this study, the concentrations of WSON were calculated as the difference between total nitrogen (TN), which was determined by the online TOC/TON analyzer, and total inorganic nitrogen ( $\text{IN} = \text{NH}_4^+ + \text{NO}_2^- + \text{NO}_3^-$ ), which was determined by the IGAC-IC, namely,  $[\text{WSON}] = [\text{TN}] - [\text{IN}]$ . The collection efficiency of IGAC system for both gas- and aerosol- phase species is >73% (Lv et al., 2022, 2023).

PM<sub>2.5</sub> filter samples were collected on a day/night basis by using a high-volume sampler (1.13 m<sup>3</sup> min<sup>-1</sup>, TISCH Environmental, Inc.) in the spring of 2023 from 27 March to 21 April 2023 ( $N = 50$ ), while size-segregated aerosols were also collected using an Anderson 9-stage sampler (Thermo Electronic Corporation, USA) at an airflow rate of 28.3 L min<sup>-1</sup> with cutoff points of 0.43, 0.65, 1.1, 2.1, 3.3, 4.7, 5.8 and 9.0 μm, respectively. In this study four sets of the size-resolved aerosol samples were collected with two sets collected in 11-13 April, 2023, which is a dust storm period (DS) (see discussion later), while other two sets of the size-resolved aerosol samples were collected in other non-dust storm periods (NDS) (7-10 April). In addition, PM<sub>10</sub> samples during a spring dust event of

2023 were also collected in Tengger Desert ( $N = 3$ ), which is one of the major dust source regions in East Asia. Field blanks of each types of samples were also collected every 3-5 days. All of the filter samples were collected onto the pre-baked (450 °C) quartz filters and stored in a freezer (−18 °C) prior to analysis.

Hourly PM<sub>2.5</sub> and PM<sub>10</sub> mass concentrations were obtained from the Shanghai Environmental Monitoring Center. The meteorological factors including temperature and RH were monitored by an automatic weather station (MILOS520, Vaisala, Inc., Finland).

## 2.2 Chemical analysis and light absorption measurement

For the offline-PM<sub>2.5</sub> samples and 9-stage size-segregated filter samples analysis, a half of each filter sample was cut into pieces and ultrasonically extracted with 45 mL Milli-Q water. The extracts were subsequently filtered with a 0.45 μm syringe filter, and analyzed using ion chromatography (ICS-5000<sup>+</sup>, Thermo Scientific) for inorganic ions (Na<sup>+</sup>, NH<sub>4</sub><sup>+</sup>, K<sup>+</sup>, Mg<sup>2+</sup>, Ca<sup>2+</sup>, Cl<sup>−</sup>, NO<sub>2</sub><sup>−</sup>, NO<sub>3</sub><sup>−</sup>, SO<sub>4</sub><sup>2−</sup>), organic acids (formic, acetic, oxalic, pyruvic (Pyr) and methanesulfonic acids (MSA). Water-soluble organic (WSOC) and inorganic carbon (WSIC) and WSON of the filter samples were determined by the TOC/TON analyzer above. Meanwhile, optical absorptions of WSOC, i.e., water-soluble brown carbon (BrC), in the size-resolved samples were measured using a liquid waveguide capillary cell UV–Vis spectrometer (LWCC, World Precision Instrument, Inc., USA) coupled with a long effective path length. In our application, light from a light source (DH-Mini, Ocean Optics, Inc., USA) is introduced into the LWCC through a fiber optic cable and, after passing through the LWCC, is collected to the detector (Ocean Optics, Inc., USA). Each instrument was calibrated weekly using authentic standards containing the target compounds. The detection limits of TC, IC and TN were 0.04, 0.05 and 0.01 mg/L, and other species detection limits were less than 0.01 ml/L.

Nitroaromatics (NACs) in the filter samples were extracted by a mixture of methanol and dichloromethane (2:1, v:v), derivatized with N,O-bis-(trimethylsilyl)trifluoroacetamide (BSTFA) and quantified by a gas chromatography (GC, HP7890B) coupled with mass spectroscopy (MS, HP5977B) (Wang et al., 2009). In this study, four NACs were determined, which are 4-nitrophenol (4NP), 2-methoxy-4-nitrophenol (4NGA), 2-methoxy-5-nitrophenol (5NGA), and 5-nitrosalicylic acid (5NSA), respectively. For imidazoles (IMS), an aliquot of the filter samples was extracted with 5 mL of ultrapure water with 1% (v/v) methanol and analyzed using high-performance liquid chromatography (HPLC, Thermo Scientific) coupled with orbitrap-mass spectrometry (MS, Thermo Scientific). A total of eight imidazole compounds (IMs) were quantified in this study, which are 2-methylimidazole (2MI), 4(5)-methylimidazole (4MI), 1-ethylimidazole (1EI), 2-ethylimidazole (2EI), 1-phenylimidazole (1PhI), 2-phenylimidazole (2PhI), 2-imidazolcarboxaldehyde (2IC) and 4-imidazolcarboxaldehyde (4IC), respectively. Recoveries of all the detected compounds were better than 85%. The detailed method for the organic aerosol analysis can be found elsewhere (Li et al., 2020; Li et al., 2023a; Liu et al., 2023).

### 2.3 Calculations on ALWC and pH of PM<sub>2.5</sub>

ALWC and pH were estimated using the thermodynamic model ISORROPIA-II, based on the hourly measured NH<sub>3</sub>, RH, T and PM<sub>2.5</sub>-related chemical compounds. A forward metastable mode was chosen for the calculation (Fountoukis and Nenes, 2007; Song et al., 2018; Wang et al., 2018). Figure S1 compares the concentrations of NH<sub>3</sub> and NO<sub>3</sub><sup>-</sup> measured by the IGAC instrument with those predicted by ISORROPIA-II model. As shown in Figure S1, the predicted are very close to the measured, suggesting that the metastable mode we used can remarkably reproduce the measured and the model results including ALWC and

acidity (pH) are reliable. It should be noted that the samples collected under  $RH > 95\%$  conditions were excluded in this study, because  $RH > 95\%$  conditions could increase ALWC and pH estimation uncertainties.

### 3 Results and discussion

#### 3.1 Chemical characteristics of different aerosol pollution events

During the campaign  $PM_{2.5}$  varied from 3.0 to  $87 \mu g m^{-3}$  with an average of  $30 \pm 14 \mu g m^{-3}$  (Figure 1 and Table 1), which is slightly lower than the National Air Quality I Grade Standard of China ( $35 \mu g m^{-3}$ ). However, particle pollution with a daily  $PM_{2.5} > 35 \mu g m^{-3}$  were frequently observed during the campaign (Figure 1). In this study, a pollution event lasting for more than two consecutive days with a daily  $PM_{2.5}$  levels  $> 35 \mu g m^{-3}$  was defined as a haze episode. As shown in Figure 1, a dust storm event had occurred in Shanghai from 11 to 13 April with a  $PM_{10}$  concentration sharply increased to  $986 \mu g m^{-3}$ , resulting in the  $PM_{2.5}/PM_{10}$  mass ratio down to less than 0.1 and  $Ca^{2+}$  up to  $2.5 \mu g m^{-3}$  (Table 1 and Figure 1). Thus, here we classified the observation periods of 5-9, 11-13, April as a haze event (HE) and dust storm event (DS), respectively. While those periods with a daily  $PM_{2.5} < 35 \mu g m^{-3}$  was defined as a clean period (CP), which occurred from 18-22, April (Figure 1).

As seen in Table 1, secondary ions (sulfate, nitrate, ammonium, SNA) were the dominant species among the detected components. The mass contribution of SNA to  $PM_{2.5}$  reached up to ~69 % in HE, which was 1.6-fold of that in CP, indicating a key role of SNA in the haze formation process. During the DS period, SNA loadings in Shanghai were comparable to that in CP, which only accounted for ~17 % of  $PM_{2.5}$ . However, the proportion is still remarkably higher than those in Asian desert sources and upwind regions (Table S1), indicating a significant formation of SNA during the dust long-transport ( $t$  test,

$p < 0.01$ ).  $\text{Ca}^{2+}$ ,  $\text{Na}^+$  and  $\text{Mg}^{2+}$  were 4–8 times higher in DS than in HE. Such enhancements in metal cations affected the aerosol acidity, resulting in the pH of fine particles ( $\text{PM}_{2.5}$ ) rising from 3.3 in non-dust storm period (NDS) to 7.5 in DS (Figure 1c).

As shown in Table 1, there was a large variability in  $\text{WSOC}_g$  concentration, which ranged from 0.7 to  $19.6 \mu\text{gC m}^{-3}$  with an average loading of  $8.2 \pm 3.3 \mu\text{gC m}^{-3}$  in the CP period and was higher than those in the HE and DS periods (Table 1). The partitioning coefficient of WSOC ( $F_p = \text{WSOC}_p / (\text{WSOC}_p + \text{WSOC}_g)$ ) ( $0.32 \pm 0.06$ , Table1) in HE was 33% higher than that in CP ( $0.24 \pm 0.06$ , Table1) and robustly correlated with  $\text{WSOC}_p$  ( $r = 0.68, p < 0.01$ ) (Figure S2a). A similar strong correlation of  $\text{PM}_{2.5}$  with  $\text{WSOC}_p$  was also observed in HE but not in DS (Figure S2b). Since most components of SOA are water-soluble, such a strong correlation between  $F_p$  and  $\text{WSOC}_p$  suggests that gas-to-particle-phase partitioning of WSOC is a major formation pathway of SOA in the HE period. WSOC level and  $F_p$  in DS ( $0.3 \pm 0.06$ , Table1) was similar to those in HE and exhibited a well correlation each other ( $r = 0.61, p < 0.01$ ) (Figure S2a), indicating that gas-to-particle partitioning of  $\text{WSOC}_g$  was also an important formation pathway for SOA in the DS period. As seen in Table 1, SNA accounted for  $68 \pm 13\%$   $\text{PM}_{2.5}$  in HE but only  $16 \pm 12\%$   $\text{PM}_{2.5}$  in DS, because  $\text{PM}_{2.5}$  was dominated by secondary species in HE but by mineral components in DS. Thus, unlike the case in HE, there was no correlation between  $\text{PM}_{2.5}$  and  $\text{WSOC}_p$  in DS.

### 3.2 Factors driving WSOC partitioning during aerosol pollution events

As shown in Table 1 and Figure 2a, ALWC in HE was 2.5-fold higher than that in CP period and well correlated with  $F_p$  ( $r = 0.66, p < 0.01$ ), because  $\text{PM}_{2.5}$  aerosols in the HE period were enriched in hygroscopic SNA, resulting in a high level of ALWC under the humid conditions (Table 1). Thus, ALWC took a key role controlling the partitioning process of WSOC, similar to the results observed during winter in Shanghai, China and Los Angeles

and Atlanta, United States (Hennigan et al., 2009; Zhang et al., 2012; Lv et al., 2022b). In contrast, ALWC in the DS period was much lower than that in the HE period with no correlation with  $F_p$  (Figure 2a), because SNA concentrations were much less in the DS period than those in the HE period. Moreover, RH in the DS period was  $44 \pm 18\%$ , which is lower than the deliquescent point of SNA (Wang et al., 2016). Therefore, WSOC partitioning process during the DS was not regulated by ALWC. In this study, we calculated the contributions of organic matter (OM) and water-soluble inorganic ions to ALWC in the HE period. It can be seen from Figure S3a,  $\text{NH}_4\text{NO}_3$  was the largest contributor to ALWC during the haze pollution period, followed by  $(\text{NH}_4)_2\text{SO}_4$  and OM, with the contributions being  $56.4 \pm 12.2\%$ ,  $29.1 \pm 7.1\%$ , and  $6.1 \pm 3.8\%$ , respectively. Figure 2b shows that ALWC increased exponentially along with the  $\text{NH}_3$ -partitioning coefficient ( $\epsilon(\text{NH}_4^+) = \text{NH}_4^+ / (\text{NH}_3 + \text{NH}_4^+)$ ) in the HE period and  $\text{WSOC}_p$  increased in tandem with an increase in ALWC in HE, which increased along with  $\text{TNH}_x$  concentrations (Figure 2c). Moreover, as seen in Figure 2d,  $\text{WSOC}_p$  well correlated with  $\text{TNH}_x$  ( $= \text{NH}_3 + \text{NH}_4^+$ ) in HE ( $r = 0.72$ ,  $p < 0.01$ ), indicating that abundant ammonia in the atmosphere of Shanghai is favorable for WSOC partitioning and SOA formation. Na et al., (2007) and Liu et al., (2021) reported that  $\text{NH}_3$  can be neutralized with organic acids to form ammonium salts, thus induce rapid increase in SOA yields. It can be seen from Figure S3b that particulate carboxylates exponentially increased along with an increase of  $\text{TNH}_x$  concentration, further demonstrating that  $\text{NH}_3$  neutralization with organic acids was also responsible for the strong correlations of  $\text{TNH}_x$  with  $F_p$  in the HE period.

During the DS period,  $F_p$  of WSOCs robustly correlated with pH ( $r = 0.98$ ,  $p < 0.01$ , Figure 2e), suggesting that pH was a key factor controlling the WSOC partitioning. As we discussed before, ammonia played an important role in  $\text{WSOC}_p$  formation during the HE

period with a strong correlation with  $\text{TNH}_x$  in HE (Figure 2d). However, such a correlation was not observed in DS (Figure 2d), which indicates that ammonia was not an effective factor driving the WSOC partitioning process in the DS period. As shown in Figure 2f, the sum of  $\text{Ca}^{2+}$ ,  $\text{Na}^+$  and  $\text{Mg}^{2+}$  concentrations showed a significantly correlation with particulate organic acids ( $r = 0.88$ ,  $p < 0.01$ ), including formic, acetic and oxalic acids, during the DS period, because the lower acidity of aerosols in the DS period was favorable for the reactive uptake of acidic organics by dust surface (Table S2).

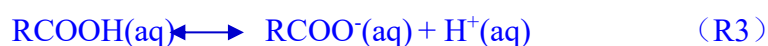
To further analyze the factors controlling the WSOC partitioning, we investigated the partitioning process of formic and acetic acids, which are abundant species of WSOCs in the atmosphere (Table S2). As shown in Figure S4,  $F_p$  of formic and acetic acids in HE presented strong correlations with ALWC, pH and T. In DS, however, compared with ALWC and T,  $F_p$  of formic and acetic acids exhibited a stronger correlation with pH, further indicating that the partitioning of WSOC was more significantly affected by the particle acidity in the presence of dust storm.

Apart from the above factors that influence the  $F_p$  variations, aerosol mass loading may also affect the partitioning of WSOC (Lutz et al., 2019). According to the Raoult's law, the field gas-particle partitioning coefficient can be described as follows (Pankow, 1994):

$$K_i = \frac{i_{\text{particle}}}{i_{\text{gas}} M_{\text{org}}} \quad (\text{R1})$$

where  $K_i$  is the partitioning constant of  $i$  compound,  $i_{\text{particle/gas}}$  are the particle and gas phase concentrations of compound  $i$ , and  $M_{\text{org}}$  is the concentration of aerosol organic mass (WSOC<sub>p</sub> are used here). In this section, our main focus is the  $F_p$  of WSOC in  $\text{PM}_{2.5}$ . Thus,  $i$  and  $M_{\text{org}}$  are the concentrations of formic and acetic acids and WSOC in  $\text{PM}_{2.5}$ . By definition, the  $K_i$  is governed by the organic aerosol mass loading. However, as shown in

Figure S5, the partitioning constants ( $K_i$ ) of formic and acetic acids did not correlate with WSOC<sub>p</sub> in both HE and DS periods, indicating that the effect of aerosol organic mass loading on the uptake of WSOC during the campaign was not significant. As shown in Table 1, the PM<sub>2.5</sub> and WSOC loadings in DS were comparable to those in HE, although PM<sub>10</sub> loading was much higher in the dust storm period. Moreover, the pH ( $4.8 \pm 1.5$ , Table 1) of PM<sub>2.5</sub> in DS was 1.4 units higher than that ( $3.4 \pm 0.3$ ) in HE, which means that H<sup>+</sup> concentration in DS was one order of magnitude lower than that in HE. According to the equations R2 and R3, the low H<sup>+</sup> concentration in DS was favorable for the organic acid equilibria shifting toward the aerosol aqueous phase. Thus, the impact of pH variation on  $F_p$  was much more significant than organic aerosol mass loading in DS, taking a key role in WSOC partitioning process during the dust storm event. Previous studies found that diffusivity of organics in particles strongly depends on the phase state of particles (Gkatzelis et al., 2021; Ma et al., 2022). Here we further investigated the impact of particle state on WSOC partitioning during the campaign by calculating the phase state of fine particles (see the detailed method in SI). As shown in Figure S6, fine aerosols were mainly in liquid state during the HE period and mainly in solid or semi-solid phase during the DS period. Thus,  $F_p$  linearly and negatively correlated with  $T_g/T$  along with an increasing ALWC in HE. However, such a trend was not observed in DS, again suggesting that the key factors controlling the gas-to-particle phase partitioning process of WSOC were different during the two particle pollution events, which was ALWC in HE and pH in DS, respectively.



### 3.3 Gas-to-particle partitioning of WSOC on size-resolved particles

Figure 3 shows the size distributions of water-soluble organic species, including WSOC<sub>p</sub>, NACs, IMs and WSON<sub>p</sub>, in the non-dust storm (NDS) and dust storm (DS) periods in Shanghai during the campaign. As shown in Figure 3a, WSOC<sub>p</sub> showed a bimodal size distribution pattern in both periods. The relative abundance of WSOC<sub>p</sub> in the coarse mode (> 2.1 μm) accounted for 59 % of the total during the DS period, which is 25 % higher than that in the NDS period. Oxalic, pyruvic and methanesulfonic acids are typical SOA tracers in the atmosphere (Wang et al., 2012; Huang et al., 2014; Wang et al., 2015; Wang et al., 2016; Tang et al., 2019), which displayed a bimodal pattern in the NDS period with two peaks distributing in the fine and coarse modes, respectively (Figure S7). In contrast, those secondary SOA presented a monomodal size distribution pattern in the DS period with a predominant peak in the coarse mode only. As seen in Table 2, their concentrations are substantially higher than those in the Tengger Desert region and mostly stayed in the coarse mode (> 2.1 μm) during the DS period, clearly revealing an enhanced SOA formation on the dust particles during their transport from the Asian desert source region to the downwind city. As shown in Figure S8, F<sub>p</sub> of WSOC in the fine and coarse modes were larger than those in the NDS period, especially for the coarse mode of F<sub>p</sub>, which was 60 % higher in the DS period than that in the NDS period, indicating that dust surface is favorable for the uptake of WSOC<sub>g</sub>.

Previous studies reported that NACs can be formed secondarily via both gas-phase and aqueous-phase reaction pathways (Wang et al., 2019; Hems et al., 2020; Ren et al., 2023). For example, benzene reacts with OH radicals and generates phenol. Then, phenol further reacts with NO<sub>2</sub> and yields gaseous nitrophenol (Liang et al., 2020). The latter subsequently partitions to the condensed phase (Lauraguais et al., 2014). Moreover, phenol and catechol can also react with NO<sub>2</sub> radical in the aerosol aqueous phase to form the corresponding

NACs (Vione et al., 2004). Our recent study in Shanghai found that the gaseous NACs (i.e., nitrophenol and methylnitrophenols) concentrations were 2 orders of magnitude higher than the particulate ones (Liu et al., 2023). As seen in Figure 3b, NACs showed a bimodal pattern with a relatively larger peak in the coarse mode during the NDS period. In contrast, NACs almost entirely stayed in the coarse mode during the DS event. Such distinct size distribution patterns indicate that NACs in Shanghai in the DS event were mainly formed by a gas-phase oxidation and subsequent partitioning onto the dust surface.

IMs can be directly emitted from biomass burning process and secondarily produced from reactions of carbonyls with amines and ammonia in aerosol aqueous phase (Lian et al., 2021; Liu et al., 2023). As shown in Figure 3c, IMs predominantly stayed in the fine mode in the NDS period. However, in the DS period IMs showed a bimodal pattern in the DS period with two equivalent peaks in the fine and coarse modes, respectively. Since IMs were not detected in the Desert source samples (Table 2), the peak of IMs in the coarse particles in Shanghai during the DS period are expected to be secondarily formed on the dust particles via heterogeneous reactions such as the reactions of glyoxal and methylglyoxal with free  $\text{NH}_3$  molecules in aerosol aqueous phase (Liu et al., 2023). Size-resolved chemistry analysis results showed that  $\text{NO}_3^-$  in the coarse mode poorly correlated with  $\text{NH}_4^+$  ( $> 2.1 \mu\text{m}$ ) in the NDS period ( $r = 0.29$ ,  $p < 0.05$ ) but robustly correlated with  $\text{NH}_4^+$  ( $r = 0.76$ ,  $p < 0.01$ ) in the DS period (Figure S9a). Field observations found that  $\text{NO}_3^-$  in the coarse mode during the NDS period is formed via reaction uptake of gaseous  $\text{HNO}_3$  by coarse particles and exists mostly as  $\text{Ca}(\text{NO}_3)_2$  (Tobo et al., 2010; Wu et al., 2020). Therefore,  $\text{NO}_3^-$  in coarse particles poorly correlates with  $\text{NH}_4^+$  (Figure S9a), because coarse particles are usually enriched with  $\text{CaCO}_3$ , which is more basic than  $\text{NH}_3$  (Laskin et al., 2005; Li and Shao, 2009; Tobo et al., 2010). However, in a typical Asian dust storm period the coarse mode of  $\text{NO}_3^-$  usually

strongly correlates with  $\text{NH}_4^+$ , because there are large amounts of mineral dust particles in Asian dessert region such as  $\text{Na}_2\text{SO}_4$ ,  $\text{CaSO}_4$  and  $\text{NaCl}$  (Wu et al., 2020). Those mineral dust particles are hygroscopic and can promote the hydrolysis of  $\text{N}_2\text{O}_5$  on dust particle surface and produce  $\text{HNO}_3$ , which subsequently neutralize with  $\text{NH}_3$  as  $\text{NH}_4\text{NO}_3$  (Wang et al., 2014; Wu et al., 2020). Therefore, a linear correlation between  $\text{NO}_3^-$  and  $\text{NH}_4^+$  in the coarse mode was observed in Shanghai in DS during the campaign (Figure S9a). As we discussed in section 3.2,  $\text{NH}_4\text{NO}_3$  was the biggest contributor to ALWC. Therefore, the formation of hygroscopic species such as  $\text{NH}_4\text{NO}_3$  can enhance gas-to-particle phase partitioning of  $\text{WSOC}_g$  and the heterogeneous reactions on the dust surface. Similarly, the coarse mode of oxalic acid robustly and linearly correlated with  $\text{NH}_4^+$  in the DS period but not correlated with  $\text{NH}_4^+$  in the NDS period ( $r = 0.97$ ,  $p < 0.01$ ) (Figure S9b), which again suggests an enhanced heterogeneous reaction of SOA on the mineral particle surface during Asian dust long-range transport.

### 3.4 Enhanced light absorption ability of Asian dust particles

As seen in Figure 3d,  $\text{WSON}_p$  presented a monomodal size distribution pattern in Shanghai, peaking at the fine and coarse mode in the NDS and DS periods, respectively. The relative abundance of  $\text{WSON}_p$  in the coarse mode in the DS period is 0.8 times higher than that in the NDS period. Such an increased  $\text{WSON}_p$  in the coarse mode in the DS period is in good agreement with that of NACs and IMs and indicates an enhanced formation of nitrogen-containing organic compounds on dust surface (Figure 3b-d).

Previous researches have shown that many of nitrogen-containing organic compounds in the troposphere are light absorbing brown carbon (BrC) (Laskin et al., 2015; Liu et al., 2024; Laskin et al., 2025). Thus, here we further investigate the optical properties of the dust samples. As seen in Table 2,  $\text{Abs}_{365}$  ( $1.9 \text{ M m}^{-1}$ ) and  $\text{MAC}_{365}$  ( $1.0 \text{ m}^2 \text{ g}^{-1}$ ) of BrC in the fine

mode during the NDS period were comparable to those during the DS period ( $2.2 \text{ M m}^{-1}$  and  $0.9 \text{ m}^2 \text{ g}^{-1}$ , respectively). However,  $\text{Abs}_{365}$  and  $\text{MAC}_{365}$  ( $2.6 \text{ M m}^{-1}$ ,  $0.8 \text{ m}^2 \text{ g}^{-1}$ , Table 2) in the coarse mode in the DS period were 2–3 times higher than those ( $0.8 \text{ M m}^{-1}$ ,  $0.4 \text{ m}^2 \text{ g}^{-1}$ , Table 2) in the NDS period, respectively, suggesting an enhanced light absorption of coarse particle in the Asian dust storm events. To reveal the nature of such an increase in light absorption of dust particles, we measured the optical properties of atmospheric  $\text{PM}_{10}$  collected in Tengger Desert. As shown in Figure 4a, MAC of water-soluble BrC in total suspended particles (TSP) in the wavelength range of  $\lambda = 250\text{--}600 \text{ nm}$  in Shanghai during the DS period is slightly higher than that in the NDS period, but it is more than two times that in Tengger Desert region, suggesting a remarkable increase in light absorption ability of TSP in the downwind city during both NDS and DS periods.  $\text{MAC}_{365}$  of BrC in the coarse mode during the DS period in Shanghai ( $0.8 \text{ m}^2 \text{ g}^{-1}$ , Table 2) is four times that in Tengger Desert regions ( $0.2 \pm 0.09 \text{ m}^2 \text{ g}^{-1}$ , Table 2). As shown in Figure 4b,  $\text{MAC}_{250-600}$  of water-soluble BrC in the fine mode during both sampling periods were comparable, but the  $\text{MAC}_{250-600}$  in the coarse mode during DS period was significantly higher than that in NDS period. Such an increase in light absorption that only occurred in the coarse mode clearly demonstrates that light absorption ability of Asian dust particles in the DS period was significantly strengthened during the long-range transport, which is mostly due to an efficient formation of nitrogen-containing organic compounds on the dust surface.

From Figure 4b one may further see that the high  $\text{MAC}_{250-600}$  values at the coarse mode in the DS period largely occurred at the 250–350 nm. As reported by previous studies (Yuan et al., 2020; Liu et al., 2023), light absorption of NACs mainly peaks at a wavelength range large than 350 nm while IMs absorb light mostly at a wavelength less than 300 nm. Thus, here we quantified the contribution of IMs light absorption in the coarse mode during DS

period. Based on its concentration and light absorption properties, the light absorption contribution of each individual chromophore to the total BrC in the wavelength range of 250–400 nm was calculated (Text S2). As seen in Figure 5, IMs show two large absorption peaks at wavelengths of 260 and 290 nm, respectively, which are mainly associated with the absorption of 2IC and 4IC. Moreover, it can also be seen that IMs light absorption contribution to light absorption of water-soluble BrC in DS is almost twice that of NDS, indicating a significant contribution of aqueous phase formation to BrC in DS. This result further suggests that an enhanced heterogeneous formation of nitrogen-containing organics such IMs on the dust particles surface are responsible for the increased light absorbing of Asian dust during a long-range transport.

#### 4. Summary and implications

In this study we synchronously measured the hourly concentrations of gaseous and particulate WSOC in Shanghai, a downwind city of East Asian desert sources, along with measurements on organic tracers and inorganic ions in the gas- and aerosol- phases. We found that the gas-to-aerosol-phase partitioning coefficient ( $F_p$ ) of WSOC on the dust storm days ( $0.3 \pm 0.06$ ) in the city is comparable to that on the haze days ( $0.32 \pm 0.06$ ), although the meteorological conditions were drier and hotter during the dust storm period and thus not favorable for the WSOC partitioning, suggesting an enhancing effect of dust particles on the SOA formation on the dust particle surface. In this study, we found the  $F_p$  variation was influenced by ALWC in HE but by pH in DS. Moreover,  $WSOC_p$  and its light absorption at  $\lambda_{365nm}$  in the coarse mode was increased significantly in DS period than in NDS period. Notably, by comparing with the aerosol optical properties in Tengger desert, we further found that Asian dust aerosols became much more light absorbing with a MAC four times that in the Asian dust source region, which is caused by the enhanced SOA formation on the

dust surface during the long-range transport through adsorption and heterogenous formation of light-absorbing brown carbon including nitrogen-containing organics such as nitroaromatics and imidazoles.

Atmospheric NACs and IMs are mostly produced from the reactions of aromatic with  $\text{NO}_x$  in the gas phase and carbonyls with free  $\text{NH}_3$  in the aerosol aqueous phase, respectively (Liu et al., 2023; Liu et al., 2024). Although atmospheric  $\text{SO}_2$  level in China has been decreased significantly due to strict emission control, the concentrations of  $\text{NO}_x$ , VOCs and  $\text{NH}_3$  in the country are still very high, resulting in SOA in the country much more light absorbing than those in USA and other developed countries (Hecobian et al., 2010; Liu et al., 2024; Li et al., 2025). Our previous studies found that secondary brown carbon is much more favorably formed under a high pH conditions (Liu et al., 2023; Zhang et al., 2024), because (1) nitrophenols are acidic species and thus easier to be adsorbed onto dust surface, and (2) a high pH condition favors  $\text{NH}_4^+$  dissociate into  $\text{NH}_3 + \text{H}^+$  and thus can promote the reaction of carbonyl with  $\text{NH}_3$ .

Currently, global aerosol model ensemble estimate that the global dust direct radiative effects is about  $-0.4 \text{ W m}^2$  (Kok et al., 2017). However, by using an analysis on the size and abundance of dust aerosols to constrain the global model, Abedibi and Kok (2020) found that dust aerosols in the atmosphere is coarser than the size estimated by the current global model ensemble and the dust direct radiative effect is  $+0.15 \text{ W m}^2$ , which means that dust cause a net warming of the planet. Based the results given by this study, we believe that if the secondary formation of light absorbing brown carbon is accounted for the global direct radiative effect of dust aerosol is possibly more warming.

**Data Availability.**

The data including meteorological data, gaseous pollutants, and major chemical components in PM<sub>2.5</sub> are freely available at <https://doi.org/10.5281/zenodo.14883402> (Li, 2025).

#### **Author Contributions.**

GW designed the experiment and supervised the research. ZL, BX, RL and FW collected the samples. ZL, CW, BX, and GW conducted the sample analysis. ZL, BX, CW, RL, and GW performed the data interpretation. ZL, CW and GW wrote the paper. All the authors contributed to the paper with useful scientific discussions.

#### **Competing Interests.**

The authors declare no competing financial interest.

#### **Acknowledgement.**

This work was funded by the National Key Research and Development Program of China (2023YFC3706302) and the National Natural Science Foundation of China (No. 42130704, U23A2030).

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**Table 1.** Meteorological parameters and concentrations of gaseous and particulate pollutants in PM<sub>2.5</sub> in Shanghai during the spring 2023 campaign

	Whole campaign	Dust storm	Haze event	Clean period
I Meteorological parameters and gaseous pollutants				
RH (%)	59 ± 14	44 ± 18	61 ± 13	63 ± 9
T (°C)	17 ± 4	17 ± 3	13 ± 3	21 ± 3
WSOC <sub>g</sub> <sup>a</sup> (μgC m <sup>-3</sup> )	8.0 ± 3.4	6.1 ± 1.5	6.0 ± 1.4	8.2 ± 3.3
II Major components of PM <sub>2.5</sub>				
PM <sub>2.5</sub> (μg m <sup>-3</sup> )	30 ± 14	39 ± 16	43 ± 16	20 ± 7
PM <sub>10</sub> (μg m <sup>-3</sup> )	76 ± 89	301 ± 192	58 ± 15	43 ± 19
WSOC <sub>p</sub> <sup>a</sup> (μgC m <sup>-3</sup> )	3.1 ± 1.3	2.6 ± 0.7	3.0 ± 0.8	2.8 ± 1.7
NH <sub>4</sub> <sup>+</sup> (μg m <sup>-3</sup> )	3.7 ± 2.5	1.3 ± 1.3	6.8 ± 2.1	2.2 ± 0.9
NO <sub>3</sub> <sup>-</sup> (μg m <sup>-3</sup> )	7.7 ± 6.5	3.2 ± 2.5	15.2 ± 6.2	2.9 ± 1.5
SO <sub>4</sub> <sup>2-</sup> (μg m <sup>-3</sup> )	4.2 ± 2.6	2.1 ± 2.2	6.8 ± 2.9	3.7 ± 1.8
SNA (μg m <sup>-3</sup> ) <sup>b</sup>	13.3 ± 11.1	6.5 ± 5.6	25.6 ± 13.1	8.5 ± 3.9
SNA/PM <sub>2.5</sub> (%) <sup>c</sup>	55 ± 22	16 ± 12	68 ± 13	44 ± 12
Ca <sup>2+</sup> (μg m <sup>-3</sup> )	0.1 ± 0.2	0.6 ± 0.5	0.08 ± 0.05	0.01 ± 0.01
ALWC <sup>c</sup> (μg m <sup>-3</sup> ) <sup>d</sup>	9.6 ± 12.1	2.7 ± 2.2	18.5 ± 14.4	5.3 ± 2.5
pH of PM <sub>2.5</sub>	3.4 ± 0.8	4.8 ± 1.5	3.4 ± 0.3	2.8 ± 0.5
F <sub>p</sub> <sup>f</sup>	0.28 ± 0.08	0.3 ± 0.06	0.32 ± 0.06	0.24 ± 0.06

<sup>a</sup>WSOC<sub>g</sub>, is the gas-phase organics while WSOC<sub>p</sub>, is the organics in PM<sub>2.5</sub>. <sup>b</sup>SNA is the sum of sulfate, nitrate and ammonium. <sup>c</sup>R ratio of sum of NH<sub>4</sub><sup>+</sup>, NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup> to PM<sub>2.5</sub>. <sup>d</sup>ALWC: aerosol liquid water

content of PM<sub>2.5</sub>: 
$$F_p = \frac{WSOC_p}{WSOC_g + WSOC_p}$$

**Table 2.** Concentrations of WSOC<sub>p</sub>, WSON<sub>p</sub>, organic acids, and inorganic ions and optical properties of WSOC<sub>p</sub> in airborne particles in Shanghai and Tengger Desert region

	Non-dust storm		Dust storm		Tengger Desert
	Fine mode ( $<2.1\mu\text{m}$ )	Coarse mode ( $>2.1\mu\text{m}$ )	Fine mode ( $<2.1\mu\text{m}$ )	Coarse mode ( $>2.1\mu\text{m}$ )	PM <sub>10</sub>
WSOC <sub>p</sub> ( $\mu\text{gC m}^{-3}$ )	$2.0 \pm 0.02$	$1.8 \pm 0.02$	$2.4 \pm 0.01$	$3.4 \pm 0.01$	$14 \pm 7.5$
WSON <sub>p</sub> ( $\mu\text{gN m}^{-3}$ )	$1.1 \pm 0.02$	$0.3 \pm 0.01$	$0.7 \pm 0.01$	$0.5 \pm 0.01$	ND <sup>c</sup>
PM ( $\mu\text{g m}^{-3}$ )	19	48	52	367	$402 \pm 75$
Oxalic acid ( $\mu\text{g m}^{-3}$ )	$0.7 \pm 0.04$	$0.4 \pm 0.02$	$0.3 \pm 0.02$	$1.4 \pm 0.1$	$0.7 \pm 0.08$
Pyr <sup>a</sup> ( $\mu\text{g m}^{-3}$ )	$0.06 \pm 0.01$	$0.09 \pm 0.01$	$0.03 \pm 0.01$	$0.2 \pm 0.01$	$0.03 \pm 0.01$
MSA <sup>b</sup> ( $\mu\text{g m}^{-3}$ )	$0.1 \pm 0.01$	$0.1 \pm 0.02$	$0.09 \pm 0.01$	$0.5 \pm 0.04$	ND <sup>c</sup>
NACs ( $\text{ng m}^{-3}$ )	$0.5 \pm 0.02$	$0.9 \pm 0.03$	$0.5 \pm 0.01$	$1.7 \pm 0.01$	ND <sup>c</sup>
IMs ( $\text{ng m}^{-3}$ )	$0.7 \pm 0.02$	$0.1 \pm 0.01$	$0.5 \pm 0.02$	$0.2 \pm 0.01$	ND <sup>c</sup>
Abs <sub>365</sub> ( $\text{M m}^{-1}$ )	$1.9 \pm 0.2$	$0.8 \pm 0.08$	$2.2 \pm 0.2$	$2.6 \pm 0.3$	$2.5 \pm 0.4$
MAC <sub>365</sub> ( $\text{m}^2 \text{g}^{-1}$ )	$1.0 \pm 0.04$	$0.4 \pm 0.02$	$0.9 \pm 0.04$	$0.8 \pm 0.04$	$0.2 \pm 0.09$

<sup>a</sup> Pyr: pyruvic acid. <sup>b</sup> MSA: methanesulfonic acid. <sup>c</sup>ND: Not Detected

## Figure Caption

**Figure 1.** Temporal variations in meteorological parameters and concentrations of major components in PM<sub>2.5</sub> during the spring of 2023 in Shanghai, China.

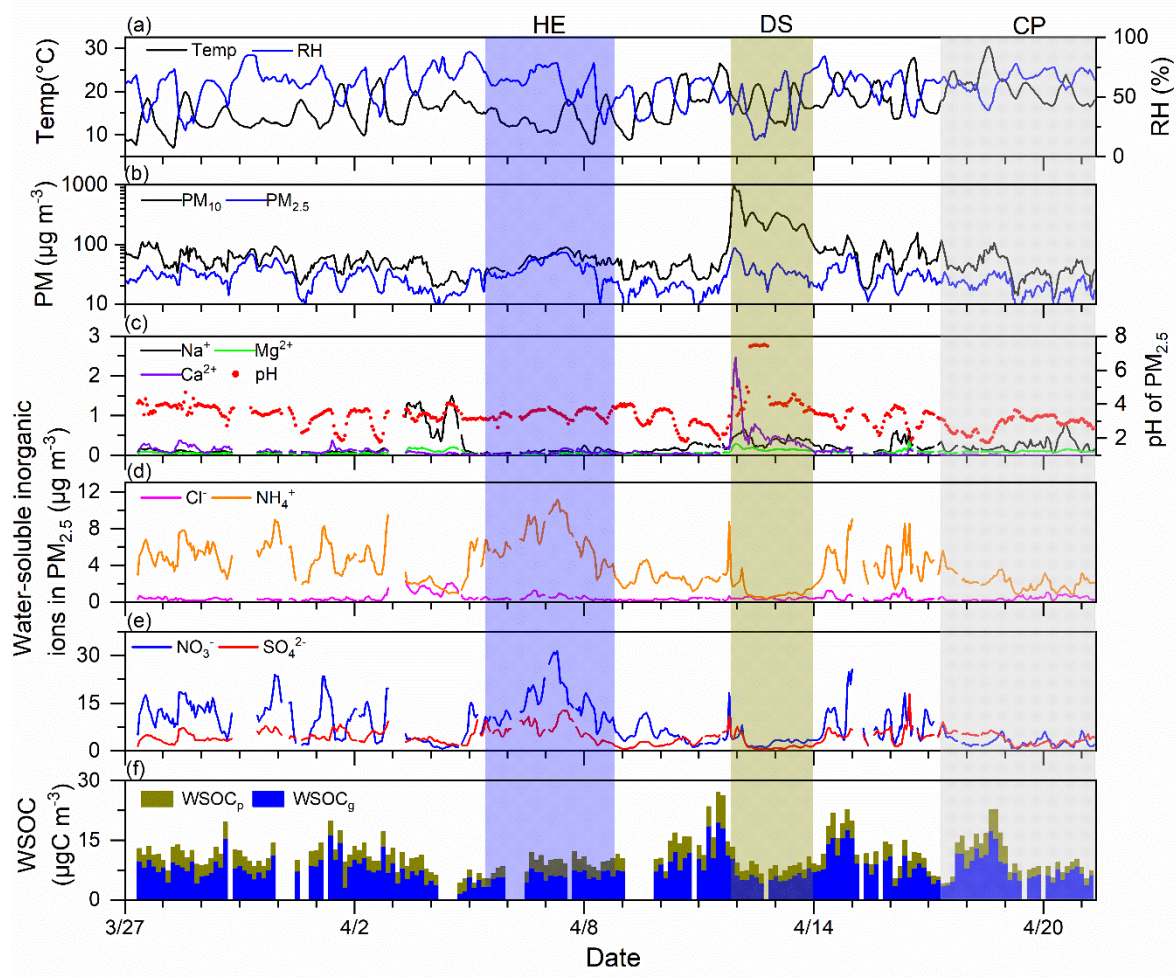
**Figure 2.** Factors controlling the gas-to-particle phase partitioning of WSOC in haze (HE) and dust storm (DS) events. (a)  $F_p$  as a function of ALWC in HE and DS. (b) ALWC as a function of ammonia partitioning coefficient ( $\epsilon(\text{NH}_4^+) = \text{NH}_4^+ / (\text{NH}_3 + \text{NH}_4^+)$ ) in HE. (c) Dependence of WSOC<sub>p</sub> concentrations on ALWC in HE. (d) Linear regression fit for  $\text{TNH}_x$  versus WSOC<sub>p</sub> concentration during HE and DS periods. (e)  $F_p$  of WSOCs as a function of pH during HE and DS periods. (f) Linear regression fit for particulate organic acids and metal cations in HE and DS periods.

**Figure 3.** Size distributions of WSOC<sub>p</sub>, WSON<sub>p</sub>, nitro-aromatic compounds (NACs) and imidazoles (IMs) in Shanghai during the non-dust storm (NDS) and dust storm (DS) periods.

**Figure 4.** Optical properties of WSOC<sub>p</sub> of PM<sub>10</sub> in Tengger Desert regions and TSP in Shanghai during the dust storm (DS) and non-dust storm (DS) periods. (a) Mass absorption coefficient (MAC) of WSOC<sub>p</sub>. (b) MAC of WSOC<sub>p</sub> in the fine (<2.1 μm) and coarse (>2.1 μm) modes of atmospheric particles in Shanghai.

**Figure 5.** Light absorption contributions of IMs to water-soluble BrC over the wavelength range of 250-400 nm in the coarse mode in NDS and DS periods.

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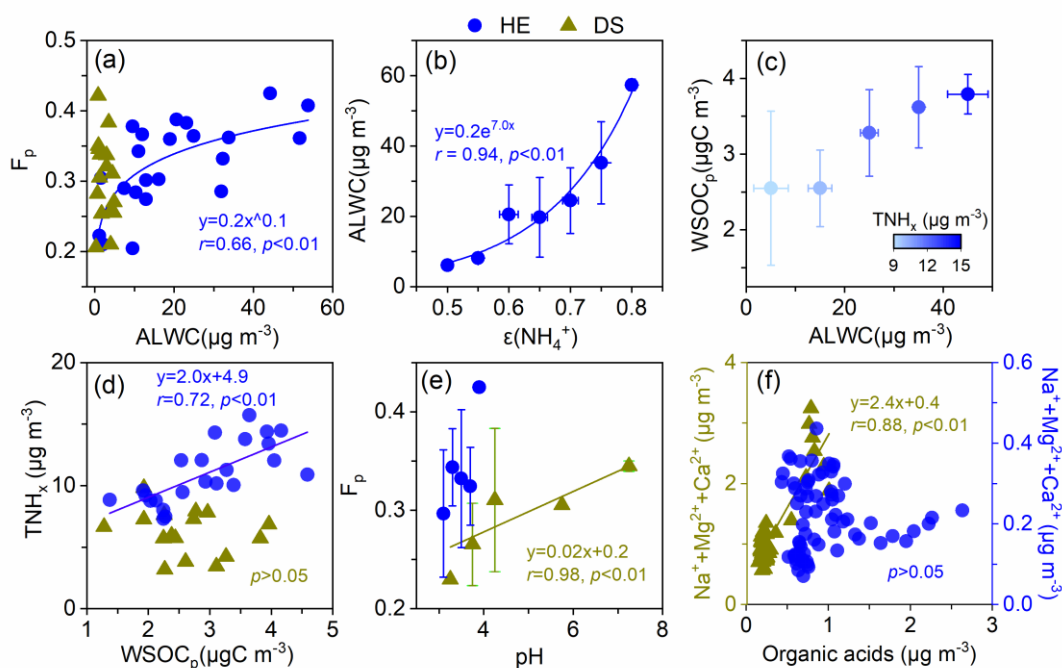
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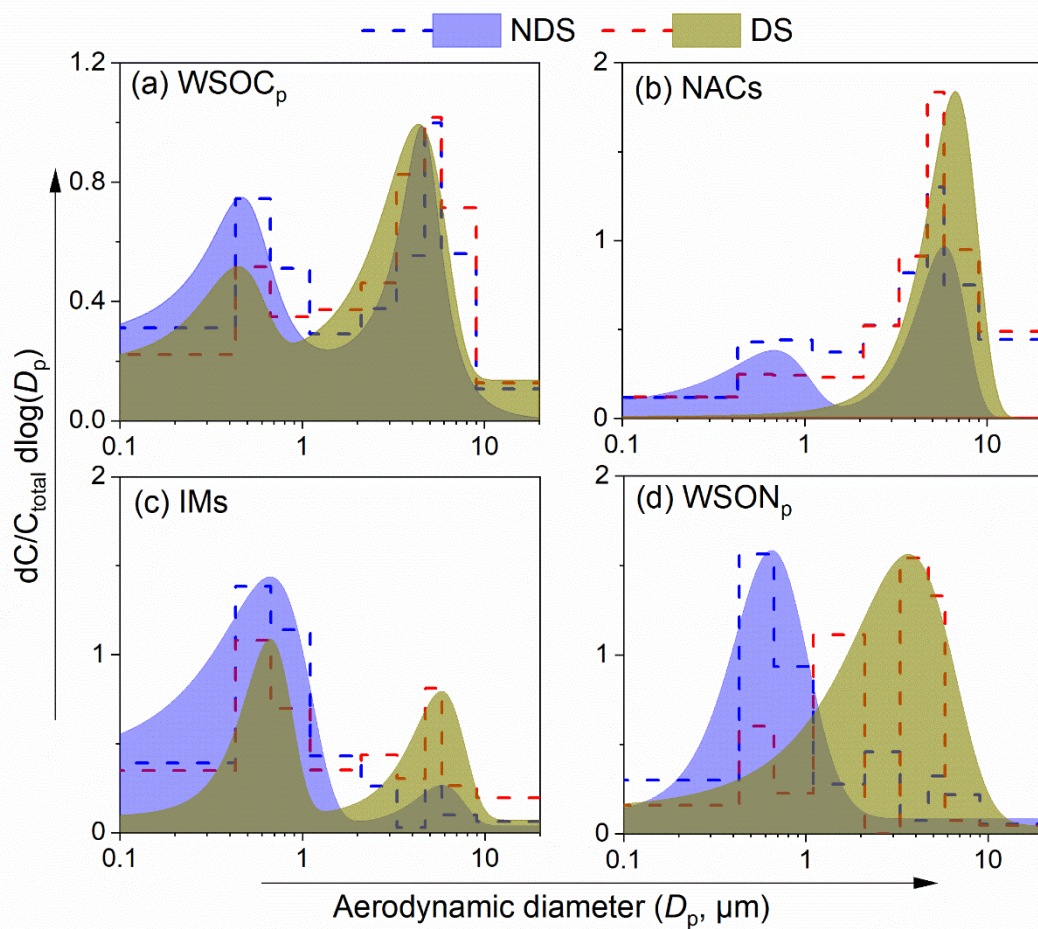
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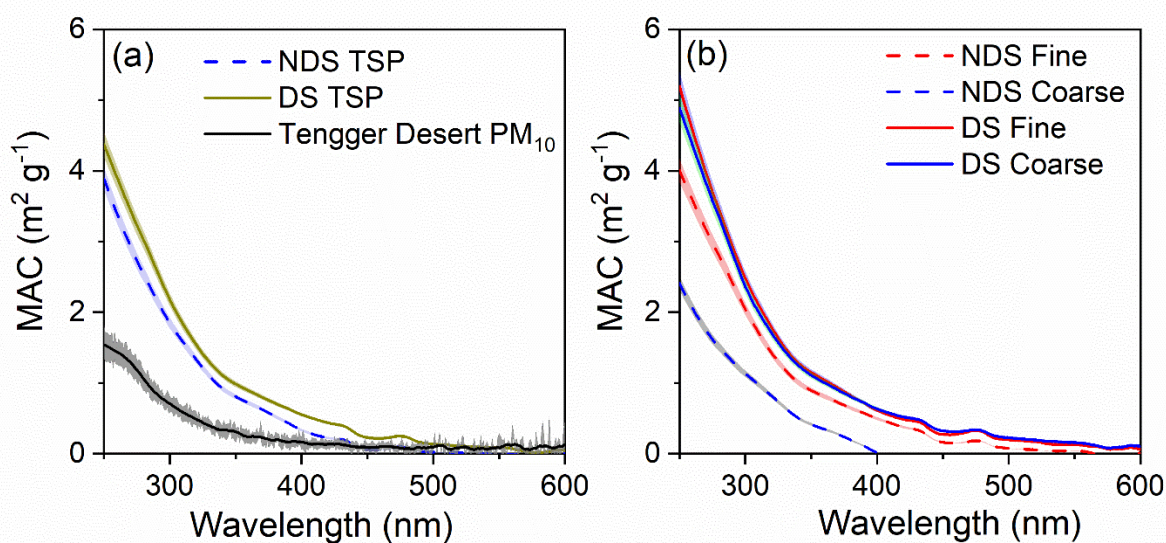
**Figure 1.** Temporal variations in meteorological parameters and concentrations of major components in PM<sub>2.5</sub> during the spring of 2023 in Shanghai, China.



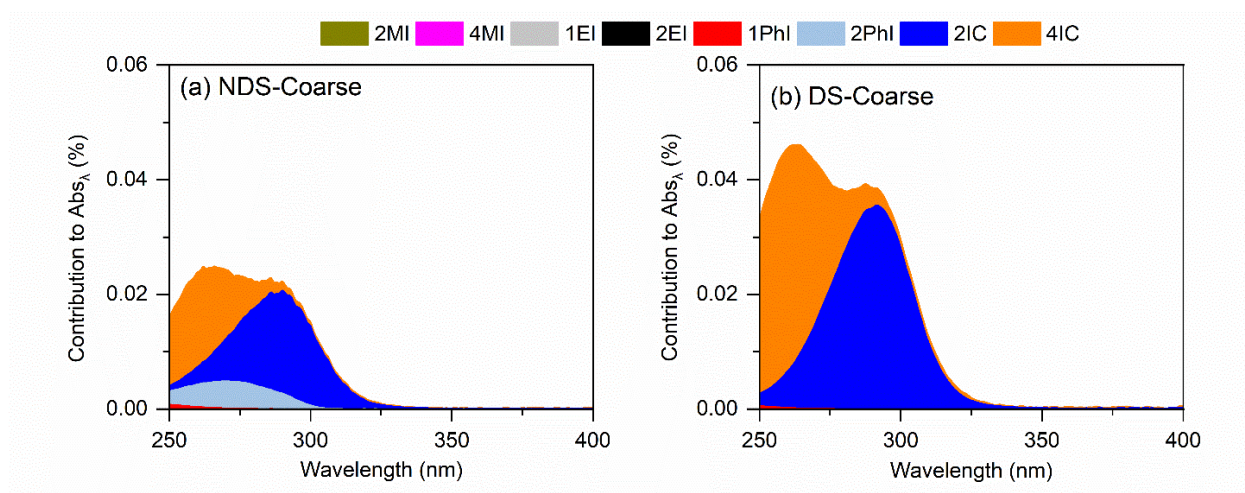
**Figure 2.** Factors controlling the gas-to-particle phase partitioning of WSOC in haze (HE) and dust storm (DS) events. (a)  $F_p$  as a function of ALWC in HE and DS. (b) ALWC as a function of ammonia partitioning coefficient ( $\epsilon(NH_4^+) = NH_4^+ / (NH_3 + NH_4^+)$ ) in HE. (c) Dependence of  $WSOC_p$  concentrations on ALWC in HE. (d) Linear regression fit for  $TNH_x$  versus  $WSOC_p$  concentration during HE and DS periods. (e)  $F_p$  of WSOCs as a function of pH during HE and DS periods. (f) Linear regression fit for particulate organic acids and metal cations in HE and DS periods.



**Figure 3.** Size distributions of WSOC<sub>p</sub>, WSON<sub>p</sub>, nitro-aromatic compounds (NACs) and imidazoles (IMs) in Shanghai during the non-dust storm (NDS) and dust storm (DS) periods. The dashed lines and filled areas are the measured size distribution and fitting results, respectively.  $C_{\text{total}}$  is the sum of concentration on all the 9-stages.



**Figure 4.** Optical properties of WSOC<sub>p</sub> of PM<sub>10</sub> in Tengger Desert regions and TSP in Shanghai during the dust storm (DS) and non-dust storm (DS) periods. (a) Mass absorption coefficient (MAC) of WSOC<sub>p</sub>. (b) MAC of WSOC<sub>p</sub> in the fine (<2.1μm) and coarse (>2.1μm) modes of atmospheric particles in Shanghai.



**Figure 5.** Light absorption contributions of IMs to water-soluble BrC over the wavelength range of 250-400 nm in the coarse mode in NDS and DS periods.

1 *Supporting Information*

2  
3 **Measurement report: Simultaneous measurement on gas-**  
4 **and particle-phase water-soluble organics in Shanghai:**  
5 **Enhanced light absorption of transported Asian dust**

6  
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30 This PDF file includes:

31 1. Four texts, Text S1-S4

32 2. Two tables, Table S1-S2

33 3. Nine figures, Figure S1-S9

34 4. References

## **Text S1 Detailed description of the instruments**

The ambient air was sampled and separated by a PM<sub>2.5</sub> sharp cut cyclone with a flow rate of 16.7 L/min. Then, the air was drawn through the passageway between the inner and outer Pyrex glass tubes in wet annular denuder (WAD) system. In WAD system, gaseous samples were wetted with pure water and collected. Scrub and Impact Aerosol Collector (SIC) was placed downstream of the WAD for collecting aerosol phase samples. The gas and aerosol samples were subsequently injected into the ion chromatography (ICS-5000+, Thermo Scientific) for analysis of water-soluble inorganic ions and small molecular organic acids after removing the insoluble species and bubbles.

The gas and particle-phase water-soluble organic carbon and water-soluble organic nitrogen (WSOC/WSON) were simultaneous determined by using a TOC/TN analyzer (TOC-L CPH, Shimadzu, Japan). The concentrations of WSOC were calculated as the difference between water-soluble total carbon (WSTC) and water-soluble inorganic carbon (WSIC). Similar, the concentrations of WSON were calculated as the difference between water-soluble total nitrogen (WSTN) and water-soluble inorganic nitrogen (WSIN).

Organic carbon (OC) and elemental carbon (EC) were determined by DRI Model 2015 Carbon Analyzer (Atmoslytic, Inc., Calabasas, USA) with IMPROVE\_A protocol (Chow et al., 2007).

## **Text S2 Absorption spectra of water-soluble BrC analysis**

The light absorption spectra of the water-soluble BrC was measured by a UV-vis spectrometer (T6 New Century, Persee) over a wavelength range of 200 – 900 nm (Wu et al., 2024). The light absorption coefficients ( $Abs_{\lambda}$ , M m<sup>-1</sup>) of the water extracts was calculated by eq (1).

$$Abs_{\lambda} = (A_{\lambda} - A_{700}) \frac{V_l}{V_a \times l} \times \ln(10) \quad (1)$$

where  $A_{\lambda}$  and  $A_{700}$  represent the light absorption at the wavelengths of  $\lambda$  and 700 nm measured by the UV-vis spectrometer.  $V_l$  refers to the volume of solvent extract.  $V_a$  refers to the sampling volume and  $l$  corresponds to the path length of the cell (1 cm).

The mass absorption coefficient ( $MAC_{\lambda}$ ,  $m^2 g^{-1}$ ) of the extracts at the wavelength of  $\lambda$  can be quantified as eq (2)

$$MAC_{\lambda} = \frac{Abs_{\lambda}}{M} \quad (2)$$

where  $M$  represent the concentration of WSOC.

The MAC of IMs standards in the water solvent at a wavelength of  $\lambda$  can be calculated as in Laskin et al (2015):

$$MAC_{i, \lambda} = \frac{A_{\lambda} - A_{700}}{l \times C_i} \ln(10) \quad (3)$$

where  $C_i$  ( $mg L^{-1}$ ) is the concentration of the  $i$  compound standards in the water solvent.

The light absorption contribution of IMs to BrC at a wavelength of  $\lambda$  can be obtained using eq (4):

$$Cont_{i/BrC, \lambda} = \frac{MAE_{i, \lambda} \times C_i}{Abs_{BrC, \lambda}} \quad (4)$$

where the  $C_i$  ( $\mu g m^{-3}$ ) is the atmospheric concentration of  $i$  compound, and the  $Abs_{BrC, \lambda}$  is the absorption coefficient of water-soluble BrC at a wavelength of  $\lambda$ .

### Text S3 Contribution of organic matter to aerosol liquid water content (ALWC)

Here we used the ALWC calculation method reported by Lv et al (2022b; 2022a). The contribution of organic matter (OM) to ALWC ( $ALWC_{org}$ ) were defined as the following eq (3)

$$ALWC_{org} = \frac{[OM] \rho_w}{\rho_{org}} \frac{\kappa_{org}}{\frac{1}{RH} - 1} \quad (5)$$

where OM is the mass concentration of organics,  $\rho_w$  is the density of water,  $\rho_{org}$  is the

83 density of OM ( $1.4 \text{ g cm}^{-3}$ ).  $\kappa_{\text{org}}$  is the hygroscopicity parameter of OM (0.06).

#### 84 **Text S4 Particle-phase-state calculations**

85 The phase-state can be inferred using the inverse ambient temperature ( $1/T$ ) scaled by  
86 the glass transition temperature of SOA ( $T_g$ ). The glass transition temperature ( $T_g$ ) is  
87 calculated based on the Gordon-Taylor equation (Koop et al., 2011):

$$88 \quad T_g = \frac{(1-w_{\text{org}})T_{g,w} + \frac{1}{K_{GT}}w_{\text{org}}T_{g,\text{org}}}{(1-w_{\text{org}}) + \frac{1}{K_{GT}}w_{\text{org}}} \quad (6)$$

89 where  $W_{\text{org}}$  is the mass fraction of SOA in the organic-water mixture,  $T_{g,w}$  is the glass  
90 transition temperature of pure water (136 K),  $K_{GT}$  is the Gordon-Taylor constant (2.5),  $T_{g,\text{org}}$   
91 is the glass transition temperature of SOA under dry conditions, calculated as follow  
92 (Shiraiwa et al., 2011):

$$93 \quad T_{g,\text{org}} = A + B M + C M^2 + D(O:C) + E M(O:C) \quad (7)$$

94 where  $A = -21.57 \text{ K}$ ,  $B = 1.51 \text{ K mol g}^{-1}$ ,  $C = -1.7 \times 10^{-3} \text{ K mol}^2 \text{ g}^{-2}$ ,  $D = 131.4 \text{ K}$  and  $E = -0.25$   
95  $\text{K mol g}^{-1}$ , respectively. The mass concentration of water in the mixture can be calculated  
96 as follow (Petters and Kreidenweis, 2007):

$$97 \quad m_{H_2O} = \frac{(0.18 \times O:C + 0.03)p_w m_{SOA}}{p_{SOA}(\frac{1}{a_w} - 1)} \quad (8)$$

98 where  $p_w$  is the density of water ( $1 \text{ g cm}^{-3}$ ),  $p_{SOA}$  is the density of SOA ( $1.4 \text{ g cm}^{-3}$ ),  
99  $m_{SOA}$  is the observed SOA mass concentration,  $a_w$  is the water activity, equal to  
100  $RH/100$ . In this study, when  $T_g/T > 1$ , SOA is solid, while when  $T_g/T < 1$ , SOA exists in  
101 semi-solid or liquid states. Here we assume that the threshold between the semi-solid  
102 and liquid states is  $T_g/T = 0.8$  (Petters and Kreidenweis, 2007; Gkatzelis et al.,  
103 2021). Worg and O:C values were measured by HR-ToF-AMS (High-Resolution Time-  
104 of-Flight Aerosol Mass Spectrometry, Aerodyne Research, Inc., USA).

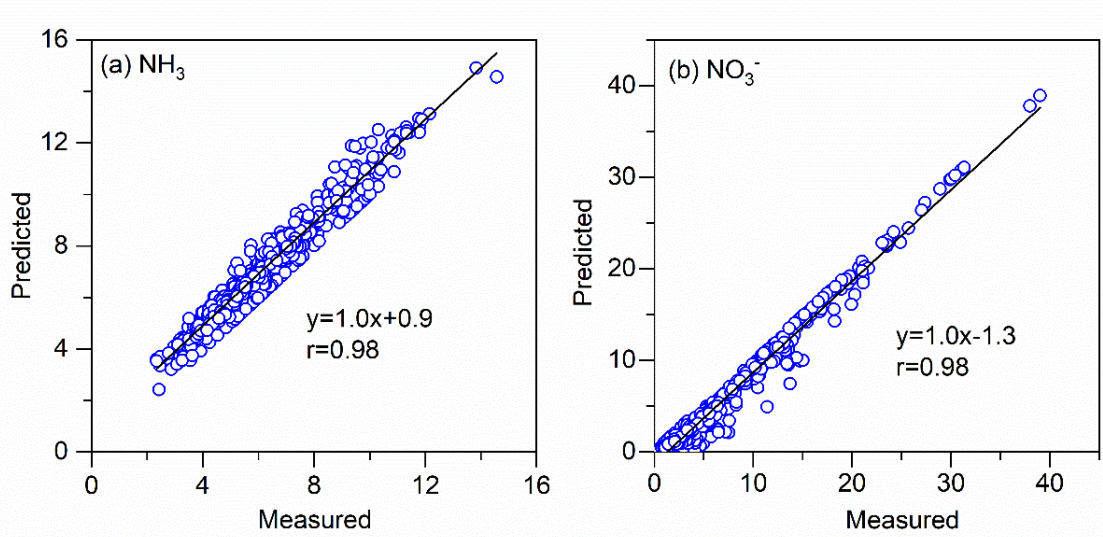
**Table S1.** Relative abundances (%) of ammonium, nitrate, and sulfate in PM<sub>2.5</sub> or PM<sub>10</sub> during dust storm periods in different regions of China.

Sampling site		Sampling time	NH <sub>4</sub> <sup>+</sup>	NO <sub>3</sub> <sup>-</sup>	SO <sub>4</sub> <sup>2-</sup>	Type	Reference
Desert region	Tengger	03, 2023	0.03	0.5	4.1	PM <sub>10</sub>	This study
	Taklimakan	04, 2008	-	0.3	4.2	PM <sub>2.5</sub>	(Wu et al., 2012)
Upwind region	Tongyu	04-06, 2006	0.18	1.26	2.43	PM <sub>2.5</sub>	(Shen et al., 2011)
	Yulin	2006-2008	0.45	0.81	3.53	PM <sub>2.5</sub>	(Wang et al., 2011)
Downwind region	Shanghai	10, 2019	3.8	10.1	6.4	PM <sub>2.5</sub>	(Wu et al., 2020)
	Shanghai	03-04, 2023	3.3	8.1	5.3	PM <sub>2.5</sub>	This study

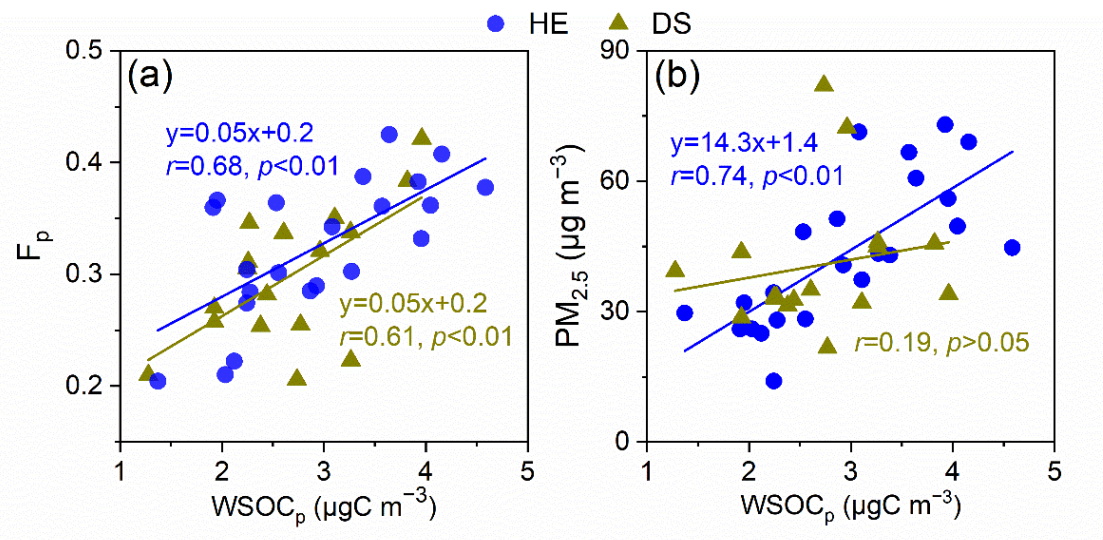
**Table S2.** Concentrations of water-soluble inorganic ions and organic acids in Shanghai during spring of 2023.

	Whole campaign	Dust storm	Haze event	Clean period
I Gaseous pollutants				
Formic acid <sub>g</sub> <sup>a</sup> (μg m <sup>-3</sup> )	3.2 ± 4.4	1.5 ± 2.6	1.3 ± 1.0	3.4 ± 6.7
Acetic acid <sub>g</sub> <sup>a</sup> (μg m <sup>-3</sup> )	1.2 ± 2.3	0.9 ± 2.0	0.5 ± 0.4	1.3 ± 2.8
NH <sub>3</sub> (μg m <sup>-3</sup> )	4.9 ± 1.6	5.1 ± 1.6	4.0 ± 0.8	5.3 ± 1.8
II Major components of PM <sub>2.5</sub>				
OC (μgC m <sup>-3</sup> )	5.0 ± 2.7	6.0 ± 5.0	5.7 ± 1.9	3.8 ± 3.2
WSOC <sub>p</sub> /OC	0.6 ± 0.2	0.5 ± 0.2	0.7 ± 0.1	0.7 ± 0.1
EC (μgC m <sup>-3</sup> )	1.1 ± 0.6	0.7 ± 0.4	1.1 ± 0.5	0.7 ± 0.5
Formic acid <sub>p</sub> <sup>a</sup> (μg m <sup>-3</sup> )	0.2 ± 0.1	0.1 ± 0.1	0.3 ± 0.1	0.2 ± 0.1
Acetic acid <sub>p</sub> <sup>a</sup> (μg m <sup>-3</sup> )	0.06 ± 0.04	0.04 ± 0.04	0.1 ± 0.04	0.06 ± 0.04
Oxalic acid <sub>p</sub> <sup>a</sup> (μg m <sup>-3</sup> )	0.3 ± 0.3	0.2 ± 0.3	0.5 ± 0.4	0.3 ± 0.2
Na <sup>+</sup> (μg m <sup>-3</sup> )	0.2 ± 0.3	0.4 ± 0.1	0.1 ± 0.05	0.2 ± 0.2
Mg <sup>2+</sup> (μg m <sup>-3</sup> )	0.08 ± 0.06	0.2 ± 0.05	0.04 ± 0.01	0.1 ± 0.02

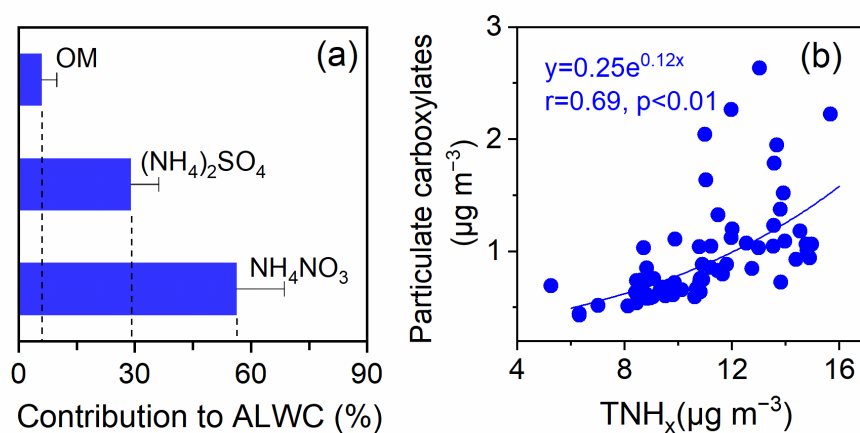
<sup>a</sup>Formic acid<sub>g</sub> and acetic acid<sub>g</sub> are the gas-phase organics while formic acid<sub>p</sub>, acetic acid<sub>p</sub> and oxalic acid<sub>p</sub> are the organics in PM<sub>2.5</sub>.



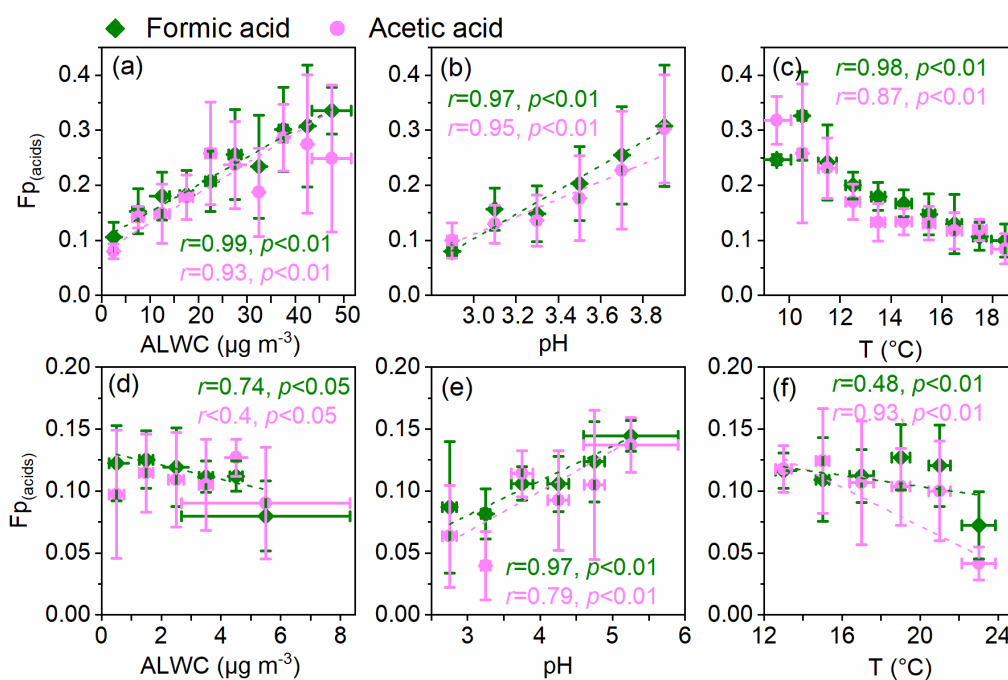
**Figure S1.** Comparison of measured ammonia and nitrate and predicted by ISORROPIA-II model.



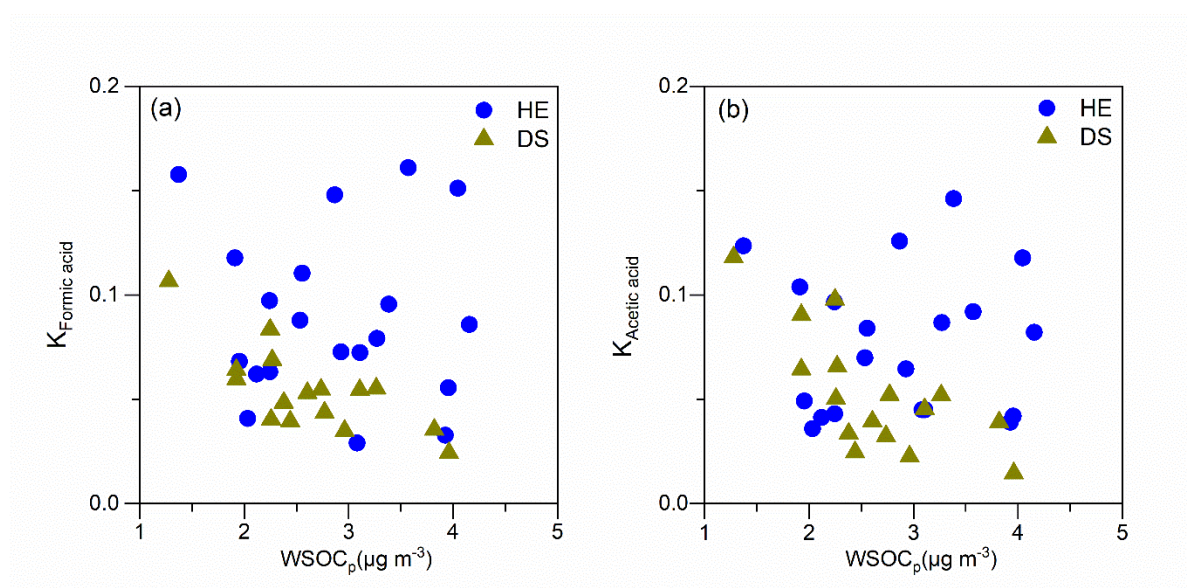
**Figure S2.** Linear regression analysis for WSOC<sub>p</sub> with (a) partitioning coefficient of WSOCs (F<sub>p</sub>) and (b) PM<sub>2.5</sub> during the haze event (HE) and dust storm (DS) event, respectively



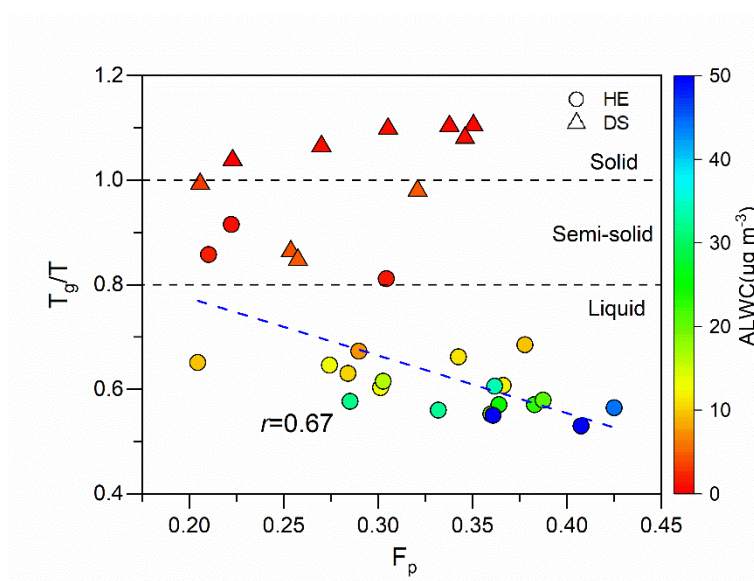
**Figure S3.** (a) Contributions of  $\text{NH}_4\text{NO}_3$ ,  $(\text{NH}_4)_2\text{SO}_4$  and OM to ALWC in the haze event (HE); (b) Particulate carboxylates as a function of  $\text{TNH}_x$  in the haze event (HE).



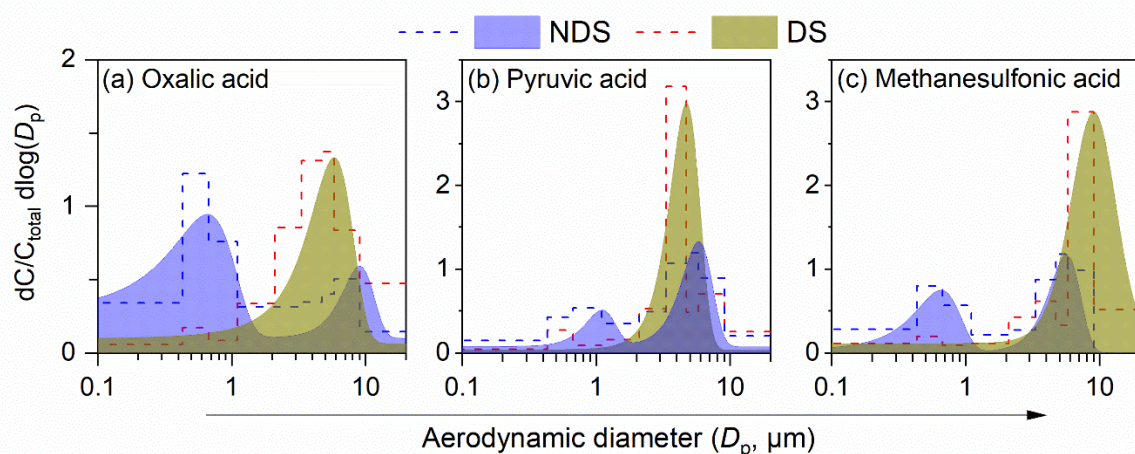
**Figure S4.** Factors affecting the gas-to-particle partitioning coefficients of formic and acetic acids in (a-c) HE and (d-f) DS periods in spring 2023 in Shanghai.



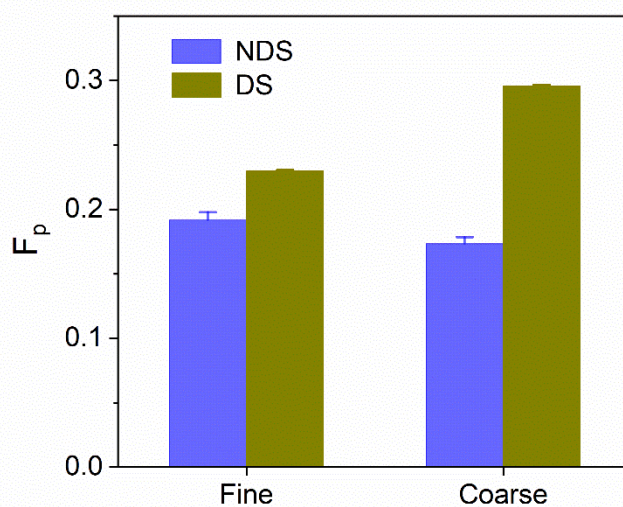
**Figure S5.** Partitioning coefficients of low molecular organic acids (formic and acetic acids) versus particle phase WSOC during HE and DS periods.



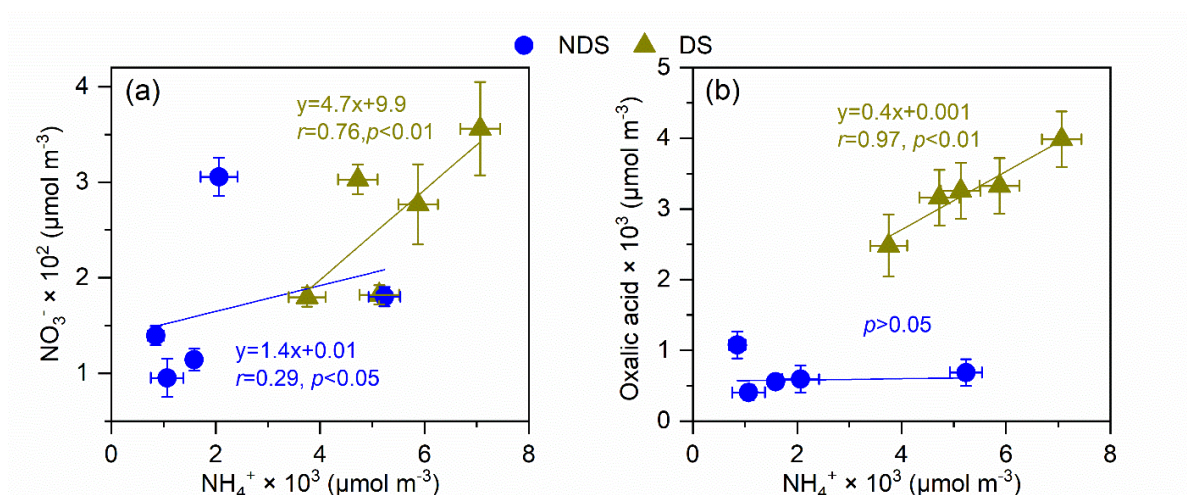
**Figure S6.** Particle-phase-state transition from solid to liquid particles compared to the partitioning coefficient of WSOC during HE and DS periods.



**Figure S7.** Size distribution of (a) oxalic acid, (b) pyruvic acid and (c) methanesulfonic acid during the non-dust storm (NDS) and dust storm (DS) events in Shanghai. The dashed lines and filled areas are the measured size distribution and fitting results, respectively.  $C_{total}$  is the sum of concentration on all the 9-stages.



**Figure S8.** The partitioning coefficients ( $F_p$ ) of WSOCs in the fine ( $<2.1\mu\text{m}$ ) and coarse ( $>2.1\mu\text{m}$ ) in the non-dust storm (NDS) and dust storm (DS) periods in Shanghai during spring of 2023.



**Figure S9.** Liner fit regression for the  $\text{NH}_4^+$  with (a)  $\text{NO}_3^-$  and (b) oxalic acid in coarse mode ( $>2.1 \mu\text{m}$ ) of particles in the non-dust storm (NDS) and dust storm (DS) periods in Shanghai during spring of 2023.

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