

Selective accumulation of dissolved organic matter in the sea surface microlayer: Insights from CDOM and FDOM characterisation at a Mediterranean coastal site.

Selective Accumulation of Dissolved Organic Matter in the Sea Surface Microlayer: Integrated Multi-Spectral CDOM and FDOM Characterization and a novel FDOM/CDOM index at a Mediterranean Coastal Site

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Abstract

The aims of this work are: i) To understand potential transformations and dynamics within the broad chromophoric dissolved organic matter (CDOM) pool in the sea surface microlayer (SML) and the underlying water (ULW) at a Mediterranean coastal site using high-resolution (biweekly, one-year) temporal data and by applying established absorbance indices across multiple spectral regions along with the novel fluorescent dissolved organic matter (FDOM) to CDOM (FDOM/CDOM) index. ii) To evaluate the relative contributions of in situ processes, upward flux from the ULW, and atmospheric (rainwater) inputs in the shaping of the dissolved organic matter (DOM) and CDOM pools in the SML. The sea surface microlayer (SML) is an extremely thin (~1 mm) boundary between the ocean and the atmosphere, forming a dynamic microenvironment that regulates air-sea gas exchange. Owing to its unique position, the SML is enriched with surface active and hydrophobic organic compounds, typically more concentrated than in the underlying water (ULW), and can therefore modulate gas transfer across the interface. Despite its importance in ocean-atmosphere interactions, the processes governing its functioning remain insufficiently understood. This study investigates the biogeochemical coupling between the SML and ULW by examining the dynamics of dissolved organic matter (DOM)—including its chromophoric (CDOM) and fluorescent (FDOM) fractions—at a coastal Mediterranean site from 2016 to 2017. Twenty-two paired SML–ULW samples and fourteen rainwater samples were analyzed for dissolved organic carbon (DOC), UV–visible (250–700 nm) absorption spectra, and 3D fluorescence excitation emission matrices (EEMs). The SML was consistently enriched in DOC, CDOM, and FDOM relative to the ULW throughout the study. Enrichment factors (EFs) for long-wavelength absorption coefficients (a_{300} , a_{370}) exceeded 5 indicating a preferential accumulation of high molecular, aromatic absorbing DOM. Relationships between DOC , a_{300} and the spectral slope ($S_{275-295}$) indicated that specific processes in the SML modulate optical quality of DOM beyond bulk DOC quantity. pP photodegradation was apparent in both layers, though more pronounced in the ULW. In the SML, photodegradation effects

appeared to be partially counterbalanced by in situ production or aggregation of hydrophobic, optically active, higher-molecular-weight material. Parallel Factor Analysis (PARAFAC) identified four FDOM components: two humic-like (A-C, A-M) and two protein-like (T, B). Terrestrial humic-like (A-C) and tryptophan-like (T) fluorophores were dominant and strongly enriched in the SML ($EF > 4$). However, humic-like components exhibited SML enrichment linked to both a_{300} absorption and layer, while protein-like components were enriched independently of a_{300} and mainly reflected layer effects. By introducing a new FDOM/CDOM index, a decoupling between fluorescent and non-fluorescent chromophoric organic fractions was revealed: the SML exhibited higher fluorescence in the UV-C/UV-B excitation regions (A, B, T peaks) but lower fluorescence in the UV-A/near-visible excitation region (C peak), suggesting selective accumulation of absorbing but non-fluorescent CDOM in longer wavelengths. Potential drivers of this decoupling include biological transformations, rapid microlayer reorganization, and atmospheric inputs. Rainwater showed DOC concentrations and absorption features comparable to the SML but distinct fluorescence characteristics. PARAFAC modeling of rainwater did not resolve the tryptophan-like fluorophore and revealed blue-shifted humic-like components, consistent with photochemically aged, low molecular weight DOM of mixed marine-terrestrial origin. Overall, the results indicate that wavelength-dependent enrichment of CDOM and FDOM in the SML is primarily driven by photodegradation, biological activity, rapid molecular reorganization, and atmospheric deposition, rather than upward DOM flux from the ULW, collectively shape the wavelength-dependent enrichment of CDOM and FDOM in the SML, highlighting its active role in air-sea biogeochemical exchange.

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1. Introduction

The sea surface microlayer (SML) is represents the boundary interface between the ocean and the atmosphere. Having a thickness of less than 1 mm, it ubiquitously covers the ocean surface. The formation and persistence of the SML have historically been linked to are generally considered to be strongly influenced by wind speed (Liss & Duce, 1997). However, recent studies findings suggest that show that this the relationship between SML stability and wind conditions is not straightforward, as enrichment can occur even at under high wind speeds above 10 m s^{-1} , mainly due to increased bubble fluxes that return surfactants to the SML (Sabbaghzadeh et al., 2017; 2024). Beyond wind effects, other factors—including solar radiation, which drives photochemical and microbial processes; biological activity, which produces and modifies surface-active substances; and atmospheric deposition—play critical roles in determining the composition and persistence of the SML (Upstill-Goddard, 2006; Wurl & Holmes, 2008; Cunliffe et al., 2013; Engel et al., 2017; Stolle et al., 2020 Upstill-Goddard, 2006). These physical, chemical, and biological interactions collectively define the dynamic nature of the SML and its influence on air-sea exchange processes (Liss & Duce, 1997; Pereira et al., 2016; 2018). Enriched with surfactants, the SML can significantly impede the air-sea transfer of greenhouse gases such as CO_2 , CH_4 , and N_2O by acting as a physical barrier (Liss & Duce, 1997; Pereira et al., 2018). Studies suggest that the presence of the SML can reduce CO_2 fluxes by 15–50%. However, most gas exchange models

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neglect this effect, leading to potential errors of up to 20% in regions with elevated surfactant concentrations (Asher, 1997; Mustafa et al., 2020). Integrating SML dynamics into climate models is therefore essential for refining global gas flux estimates and improving our understanding of ocean-atmosphere interactions (Milinković et al., 2022). The sea surface microlayer (SML) is characterized by a high content of amphiphilic hydrophobic, surface-active compounds such as carbohydrates, proteins, lipids, and humic substances, as well as particulate organic matter and a diverse community of autotrophic and heterotrophic microorganisms (Hunter, 1981; Cunliffe et al., 2009; Cunliffe et al., 2013). The enrichment of dissolved organic matter (DOM) in the SML is a complex process (Mustafa et al., 2018), influenced primarily driven by the continuous supply of organic material from the underlying seawater through biological activity and selective scavenging mechanisms (Liss & Duce, 1997; Obernosterer et al., 2008; Pereira et al., 2016; Penezić et al., 2022). Additional sources of organic matter include terrestrial and atmospheric inputs, as well as in situ production and release by primary producers and bacteria (Hardy, 1982; Wurl & Obbard, 2005; Wurl et al., 2009; Stolle et al., 2010; Ebling & Landing, 2017). The SML often exhibits spatial heterogeneity due to the accumulation of floating particulate material at the surface (Hunter, 1980), resulting in patchy distributions or slicks enriched in surfactants and potentially characterized by elevated biological productivity (Ploug, 2008; Ribas-Ribas et al., 2017). Chlorophyll-a (chl-a) is widely used as a proxy for phytoplankton biomass and primary productivity and is closely linked to the production and transformation of DOM in surface waters. Phytoplankton blooms have been identified as important drivers of DOM enrichment in SML, with phytoplankton-derived organic matter fueling microbial activity and influencing biogeochemical cycling at the air-sea interface (Wurl et al., 2016; Bibi et al., 2025). Chromophoric dissolved organic matter (CDOM) represents the fraction of DOM capable of absorbing light in the ultraviolet (UV) and visible regions. A subset of CDOM, known as fluorescent dissolved organic matter (FDOM), can additionally emit light upon excitation in these spectral ranges. In marine environments, CDOM and FDOM are is produced in situ following the same mechanisms as bulk DOM (Rochelle-Newall and Fisher 2002; Nelson et al., 2004; Lonborg et al., 2009; Ortega Retuerta et al., 2009; Romera-Castillo et al., 2010; Osburn et al., 2012). through processes such as primary production, bacterial activity, cell metabolism and lysis, and the leakage of material from particulate organic matter. External inputs also contribute to CDOM levels, including terrestrial sources delivered via riverine and land runoff, sewage discharges, and atmospheric deposition through precipitation. The primary removal mechanisms of CDOM in surface waters are photodegradation driven by UV radiation and bacterial mineralization. Over the past decades, CDOM has received considerable attention due to its key role in carbon cycling, the bioavailability of trace elements, and the air-sea exchange of trace gases (Mopper et al., 1991; Zepp, 1998; Coble, 2007). Moreover, CDOM strongly influences light attenuation in the ocean (Kirk, 1994; Siegel & Michaels, 1996; Nelson et al., 1998), thereby affecting primary productivity. Consequently, the optical properties of CDOM at the ocean surface have significant implications for remote sensing applications, particularly in relation to chlorophyll a retrievals and ocean colour observations (Tassan, 1988; Hoge et al., 1995).

Enrichment of CDOM and FDOM in the SML relative to the underlying water (ULW) has been reported in several studies (Obernosterer et al., 2008; Mustafa et al., 2017, 2018). Both CDOM and FDOM

consists of large, complex molecules rich in unsaturated bonds and aromatic moieties, structural features that impart hydrophobic properties favouring its accumulation within the SML. In coastal regions, terrestrial inputs can contribute substantially to the overall CDOM pool (Opsahl and Benner, 1998; Blough and Del Vecchio, 2002), while intense photochemical processes at the sea surface can significantly modify its composition. Interaction with sunlight can lead to CDOM (or FDOM) mineralization and removal, ~~accompanied by CO₂ and CO emissions~~, or to the breakdown and formation of less complex organic compounds that are more susceptible to bacterial transformation and degradation (Miller and Zepp, 1995; Gao & Zepp, 1998; Moran et al., 2000; Vahatalo and Wetzel, 2004; Zhang et al., 2013; Helms et al., 2013; Logozzo et al., 2021). Spatially, CDOM distribution within the SML is often patchy, with elevated concentrations occurring in slicks and other surface accumulations (Wurl et al., 2009; Mustafa et al., 2017). while significant correlation has been found between CDOM and surfactant activity (Rickard et al., 2022). Despite its ecological and biogeochemical importance, CDOM dynamics in the SML remain insufficiently explored, particularly under the distinct physical and chemical conditions characteristic of this boundary layer.

CDOM and particularly FDOM optical properties have proven to be a valuable tool for investigating the chemical composition, sources and transformation state of DOM (Stedmon et al., 2007; Hansen et al., 2016; Pitta et al., 2017). Fluorescence techniques are more sensitive than absorption spectroscopy, and fluorescence spectra provide greater detail and more information on the chemical characteristics of FDOM (Coble, 2007). Several indices based solely on FDOM properties, such as the humification, biological and fluorescence indices (HIX, BIX, FI), or on absorbance and DOC parameters, including $a(\lambda)$, spectral slope (S), and SUVA₂₅₄, are widely used to assess the quality of absorbing or fluorescent material separately (Weishaar et al., 2003; Helms et al., 2008; Huguet et al., 2009). Here, we introduce a new index, the FDOM/CDOM index, which links these two pools (CDOM and FDOM) and provides integrated insights into the chemical structure of optically active DOM as a whole.

The aims of this work are: i) To understand potential transformations and dynamics within the broad CDOM pool in the sea surface microlayer (SML) and the underlying water (ULW) at a Mediterranean coastal site using high-resolution (biweekly, one-year) temporal data, and by applying established absorbance indices across multiple spectral regions and the novel FDOM/CDOM index. ii) To evaluate the relative contributions of in situ processes, upward flux from the ULW, and atmospheric (rainwater) inputs in the shaping of the DOM and CDOM pools in the SML.

The present study investigates the nature, sources, and transformation processes of CDOM/FDOM in the SML compared to the ULW in a Mediterranean coastal site, using high-resolution temporal data obtained from biweekly sampling over a one year period. In addition, we examine rainwater CDOM/FDOM in order to investigate potential linkages between the optical signatures of rainwater and those observed in the SML. We anticipate that the results will provide new insights regarding CDOM dynamics in the SML.

2. Materials and Methods

2.1. Study area and sampling strategy

The sampling site is located in the south-east Saronikos Gulf about 1km from the coast in the eastern Mediterranean (Fig. 1). Due to its proximity to the coast the site is affected by urban runoff while only a small urban stream outflows in the area. Two phytoplanktonic blooms in the spring-early summer and autumn-early winter are characteristic for the area (Kitsiou and Karydis, 2000; Zervoudaki et al., 2022).

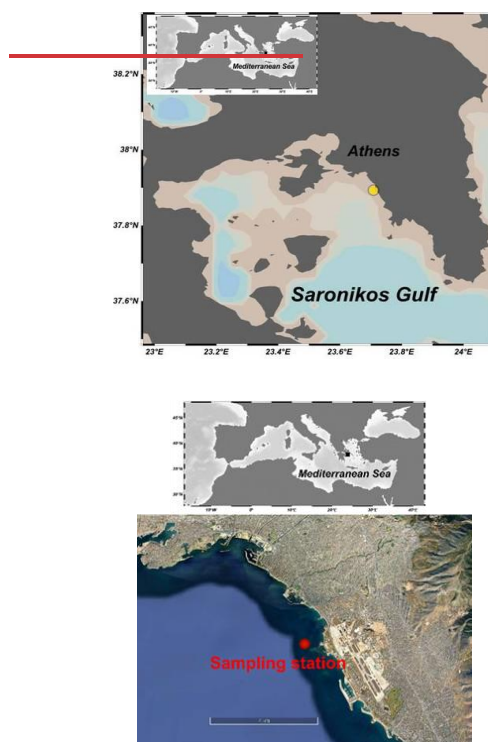


Figure 1: Study area, the red yellow dot marks the sampling site in the Saronikos Gulf

The study covered an annual cycle from June 2016 to June 2017. During this period data for wet precipitation (mm), and solar radiation from the National Observatory of Athens (www.meteo.gr) and Saharan dust episodes and deposition ($\mu\text{g m}^{-3}$) from Skiron Forecasting system (<https://forecast.uoa.gr/en/forecast-maps/skiron>) were have been gathered. Saharan dust events were also confirmed via the EOSDIS Worldview (<https://worldview.earthdata.nasa.gov/>) (Figure 2(a), Table S1). Seawater samplings were performed almost twice per month under low wind speed conditions ($< 3.5 \text{ m s}^{-1}$). Seawater temperature (T) and salinity (S) were recorded in situ using a CTD (Seabird electronics, Bellevue, WA, USA) instrument. Wind speed was recorded with a portable anemometer Benetech GM816. Samples were collected from SML and from 1m depth representing the UWL, in total 22 pairs. SML was sampled following the glass plate technique (Harvey and Burzell, 1972) yielding a microlayer thickness of $50 \pm 10 \mu\text{m}$ (Zhang et al., 2003) (Table S1). A 30 x 30 cm glass plate sampler with a width

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of 4 mm was used. Samples from the ULW were collected using a Niskin sampling bottle. Samples for DOC and CDOM and FDOM analysis were collected in amber glass containers with teflon caps while samples for chl-a determination in HDPE containers. All samples were refrigerated and immediately transported to the laboratory, where they were filtered through 0.22 µm polycarbonate filters previously rinsed with Milli-Q water. Filtered samples for DOC analysis were kept in precombusted (450 °C, 12 h) glass ampoules, acidified with HCl 2N at pH~2, sealed and stored at ~4 °C until analysis. Filtered samples for absorption and fluorescence analysis were kept in amber glass bottles with teflon caps and stored at ~4 °C until analysis. The glass containers used during sampling and for the storage of CDOM and FDOM samples were previously acid cleaned (HCl 10% (v/v) diluted HCl, 12 h) and rinsed thoroughly with Milli-Q water. Filters for chl-a determination were preserved frozen (-20 °C) in the dark until analysis.

For the period December 2016 to May 2017 rainwater samples were collected (14 samplings), via a simple system made with a Nalgene bottle connected to a HDPE funnel for wet deposition (Azimi et al., 2003) situated on the roof of a building at the coast of the study site. This was an open system, so it is assumed it was affected by dry deposition passively collected when not raining. Details on the study site and sampling procedure are described in Tzempelikou et al., (2025).

2.2. Chlorophyll-a (chl-a)

Chl-a was determined according to the Holm-Hansen et al., (1965) method using fluorescence detection in a TURNER fluorometer model TD 700. An initial sample volume equal to 500mL was filtered for SML and 1.5L for ULW. The lower filtration volumes used for the SML were due to sampling constraints associated with the glass plate technique. Laboratory tests evaluating different filtration volumes showed no significant differences in chl-a concentrations.

2.3. Dissolved Organic Carbon (DOC) analysis

The determination of DOC (mg L⁻¹) in the samples was performed using a Shimadzu TOC-L organic carbon analyzer following the High Temperature Catalytic Oxidation, HTCO. The system was standardized prior analysis using a potassium hydrogen phthalate standard solution series. Each sample was injected 3 to 5 times and DOC concentration was estimated as the average value of three replicates that yielded standard deviation < 2%. Analytical precision and accuracy were tested daily prior and at the end of the analysis using Deep Atlantic Seawater Reference Material provided by the DOC-CRM program (University of Miami - D.A. Hansell). The certified value of the reference material is 0.480-0.528 mg L⁻¹ (Batch 19 Lot 10-19) and the measured values (n=22) during the analysis of the samples were between 0.478 - 0.530 mg L⁻¹ indicating that no drift correction was necessary.

2.4. Absorption analysis

Absorption analysis was performed following the IOCCG protocol (Mannino et al., 2019). Samples for absorption determination were allowed to reach room temperature before analysis. Absorption spectra were obtained between 250 and 700 nm at 1 nm increments using a dual beam UV-visible spectrophotometer (Perkin Elmer, Lambda 25) equipped with 5 cm quartz cells and referenced to Milli-Q water. A blank scan (Milli-Q) was measured frequently to assure the stability of the instrument and it was also subtracted from each sample spectra for baseline correction. The average absorbance between 680-700 nm was subtracted from each sample in order to correct for the residual scattering by fine particles or colloidal material present in the sample. Absorption units were converted to absorption

coefficients through the relationship:

$$a(\lambda) = 2.303 \cdot A(\lambda) / l$$

where $a(\lambda)$ is the absorption coefficient (m^{-1}) at wavelength λ , $A(\lambda)$ is the absorbance at the certain wavelength and l is the light path length in meters.

Absorption coefficient at 300 nm or 350 nm-wavelength was commonly used for the representation of bulk CDOM quantity. In this study, absorption coefficient at 300 nm was selected as a more reliable proxy, since absorption at 350 nm was strongly attenuated, particularly in ULW samples. However, 3 more wavelengths (265 nm, 280 nm, 370 nm) were used in order to investigate whether there are differences in CDOM profiles in the different regions of the spectrum between the SML and ULW. The wavelengths selected provide a good coverage of the spectrum and correspond to excitation wavelengths inducing of maximum fluorescence intensities of the samples (see section 2.5). Wavelength 265 nm falls in the UV-C region, wavelength 280 nm is the border between UV-C and UV-B region, wavelength 300 nm belongs to the UV-B region while wavelength 370 nm falls in the UV-A region and it is also close to the visible light preserving though high absorption values. No wavelength of the visible region (> 400 nm) was chosen due to the low absorption values especially in the ULW.

Spectral slope of the narrow region 275-275, $S_{275-295}$ (nm^{-1}) of the spectra was calculated using linear regression of the log-transformed spectra. This is a useful indicator of the molecular weight and source composition of CDOM and can be determined with high precision, even in highly photobleached open-ocean environments (Helms et al., 2008). Consequently, it is often favored over other spectral slope indices. Photodegradation processes typically lead to a reduction in DOM molecular weight, resulting in elevated $S_{275-295}$ slope values, whereas microbial transformation generally decreases this spectral slope through the production of higher-molecular-weight organic compounds that exhibit strong absorption in longer wavelengths within the 275–295 nm range. (Helms et al., 2008; Fichot and Benner, 2012; Ortega-Retuerta et al., 2009)

2.5. Fluorescence Analysis

Fluorescence was measured using a Horiba Aqualog-UV-800 spectrofluorometer equipped with a 150 W ozone-free xenon arc-lamp and a 1 cm quartz cell. Excitation- Emission Matrices (EEMs) were obtained between 240 and 450 nm (5 nm increments) for excitation and between 250 and 800 nm (~0.6 nm increments) for emission. The integration time of the measurement was adjusted according to the fluorescence intensity in the SML and ULW samples. A blank EEM (Milli-Q water) was measured daily

and subtracted from each sample in order to correct Raman and Rayleigh scattering. All EEMs were corrected for inner filter effect following the Aqualog Operations Manual and were normalized to Raman Units (R.U.) using the Raman scatter emission peak of Milli-Q (Lawaetz and Stedmon, 2009) obtained after the measurement of each sample.

The collected EEMs were further analyzed using Parallel Factor Analysis (PARAFAC). PARAFAC was performed following Murphy et al. (2013) using MATLAB R2015a and the "drEEM toolbox". Prior modeling, the emission region between 550 and 800 nm was removed since no significant CDOM fluorescence signal was detected. The subtraction of Milli-Q water from the EEMs did not remove sufficiently the Rayleigh and Raman scatter peaks and thus both first and second order Rayleigh and Raman scatter bands were interpolated using the "smootheem" function of the drEEM package. All EEMs were then normalized to their total signal to tackle the large concentration gradients that were observed between the SML and ULW. The model was run five times with random initial values and model validation was carried out through split half analysis using four splits yielding six different combinations and it was repeated five times ($S_4C_6T_5$). The convergence criterion was 10^{-6} . Visual inspection of the residuals was also implemented along with split half analysis to assure randomness of residuals (Pitta and Zeri, 2021). PARAFAC analysis was performed on the seawater EEMs (pairs of samples ($n=22$) from the SML and ULW (SML-ULW, $n=22$)). An additional PARAFAC model was conducted for the rainwater EEMs ($n=14$). Spectral characteristics of the components resolved by the seawater and rain water models are given in Table S2.

2.6. FDOM/CDOM index

For the calculation of the FDOM/CDOM index discussed in Section 3.4.4.3-numerical fluorescence intensity values were extracted from scattering corrected EEMs using the 'smootheem' function of the drEEM package. Fluorescence intensity (I_{ij}) at each excitation wavelength (Ex_i) was normalized by the corresponding absorption coefficient (a_i) derived from absorption measurements. This normalization was applied across all emission wavelengths (Em_j), yielding the ratio I_{ij}/a_i i.e., the FDOM/CDOM (R.U.m) index, in the form of a 3-dimensional matrix spanning excitation wavelengths of 250-450 nm and emission wavelengths of 260-550 nm (Suppl Fig. S1). To obtain mean FDOM/CDOM index for each layer, the FDOM/CDOM matrices of all samples were grouped separately for the SML and ULW. For each dataset the average ($n = 22$) ratio matrix was calculated and plotted. All analyses were conducted in MATLAB R2015a using in-house functions (Pitta and Zeri, 2021).

we extracted the numeric values of modelled intensities of the scattering corrected EEMs derived from the 'smootheem' function of the drEEM package. The fluorescence intensity (I_{ij}) at each excitation wavelength (Ex_i) was then divided by the corresponding absorption coefficient (a_i) obtained from the absorption measurements. This was repeated for all emission wavelengths (Em_j), thus yielding the ratio I_{ij}/a_i i.e., the FDOM/CDOM (R.U.m) index, in the form of a 3-dimensional matrix where $Ex=250-450$ nm and $Em=260-550$ nm (Suppl Fig. S1). We then calculated the average FDOM/CDOM index for all the samples in the SML and ULW separately. First step was the construction of (number of emission wavelengths) $n \times 41$ matrices (where n is the number of samples in each dataset and 41 is the number of excitation wavelengths) using the FDOM/CDOM matrices of the samples in each layer. Next, we

calculated the average of each column of the 291 ($n \times 41$) matrices resulting into 291 1×41 vectors. Finally, 291 1×41 vectors i.e. the average FDOM/CDOM were illustrated in a 3 dimensional plot. This analysis was performed in MATLAB R2015a using in-house functions.

2.6.7. Enrichment factor (EF)

The Enrichment Factor, EF, of the various measured parameters in the SML is calculated as the ratio of the measured values in the SML to the measured values in the ULW:

$$EF = C_{SML} / C_{ULW}$$

where C_{SML} is the concentration of any parameter in the SML and C_{ULW} its corresponding concentration in the ULW. EF values > 1 indicate enrichment in the SML, while EF values ≤ 1 indicate depletion.

2.8.7. Statistical analysis

The variability of the measured parameters between SML and ULW was evaluated using the Mann–Whitney U test, as the data did not meet the assumptions required for parametric analysis. For the same reason, deviations of enrichment factors from unity were assessed using the one-sample Wilcoxon signed-rank test. ~~Linear regression and Spearman correlation was applied to evaluate relationships among the measured parameters. All reported regression and correlation coefficients (R^2 and r respectively) are statistically significant ($p < 0.05$). Generalized linear models (GLMs) were applied to examine relationships between the response variables (a_{300} , spectral slope and PARAFAC component intensities) and predictors (DOC, a_{300} and layer). A Gamma error distribution with a log link function was selected because the response variables were continuous, positive and exhibited right-skewed distributions, making this framework more appropriate than Gaussian linear models. GLMs were preferred over simple linear regression because they allow the simultaneous evaluation of continuous predictors (i.e., DOC, a_{300}) and a categorical factor (layer), as well as their interaction terms, thereby explicitly testing whether the strength or direction of relationships differed between layers. Spearman rank correlation analysis was additionally applied to assess monotonic relationships among variables.~~

All analyses were conducted using IBM SPSS Statistics, Version 20.0.

3. Results and discussion

3.1. Environmental and physicochemical conditions

Figure 2(a) presents the timing and intensity of rain and dust events recorded between 2016 and 2017, illustrating that the study site was influenced by both wet and dry atmospheric deposition. Over the 376-day observation period, Saharan dust was detected on 103 days (27%), comprising 77 dry dust pulses and 26 dusty rain events, while dust-free rainfall occurred on 31 days (8%). Solar irradiance exhibited pronounced seasonality, peaking in summer–autumn ($600\text{--}900 \text{ W m}^{-2}$) and reaching minima during

winter (130–500 W m⁻²). Seawater temperature (T) varied between 17 and 26 °C from June 2016 to January 2017, dropped to ~14 °C in February–March 2017, and increased again ~~to 17 °C~~ by May. Salinity remained relatively stable (38.06–39.61 psu) throughout the study period (Table S1). Chl-a concentrations in the SML and ULW are shown in Fig. 2(b). Values ranged from 0.21 to 0.88 µg L⁻¹ (average = 0.51 µg L⁻¹) in the SML and from 0.13 to 0.83 µg L⁻¹ (average = 0.44 µg L⁻¹) in the ULW, consistent with the region’s meso-oligotrophic character. No significant difference was detected between layers (Mann-Whitney U test, $p > 0.05$), and SML enrichment was ~~weak minimal~~ (EF = 1.2 ± 0.3) ~~with considerable variability and several samples showing no enrichment (EF < 1)~~. The annual ~~c~~Chl-a distribution revealed an unexpected ~~for the region~~ summer peak in July 2016 followed by a secondary maximum in autumn, indicative of an autumn bloom. Thereafter, concentrations declined in winter and increased again in early spring. It should be noted that ~~C~~chl-a was determined using a finer filter (0.22 µm instead of the standard ~0.87 µm GF/F), likely capturing smaller phytoplankton cells and yielding marginally higher values (Tzempelikou et al., 2025). Moreover, the occurrence of multiple Saharan dust pulses during our study period is expected to enhance primary production, consistent with observations in the Mediterranean ecosystem (Astrahan et al., 2016; Gallisai et al. 2014).

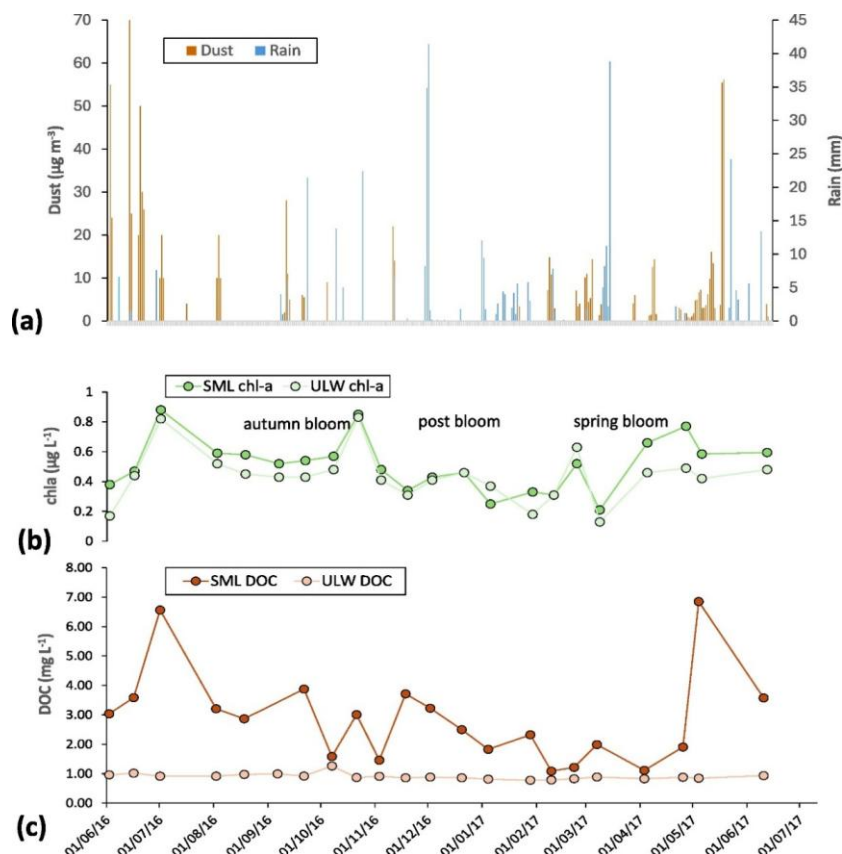


Figure 2: Monthly variation of (a) Saharan dust ($\mu\text{g m}^{-3}$) and rain (mm) deposition during the whole study period, (b) chl-a ($\mu\text{g L}^{-1}$) and (c) DOC (mg L^{-1}) measured during the 22 samplings period. (a) Dust is given in orange and rain in blue (b,c) dark dots correspond to SML concentrations and light colored to ULW ones.

The environmental conditions during our samplings study period (low wind speed, $< 3.5 \text{ m s}^{-1}$), relatively elevated dust particles and elevated chl-a ($> 0.13 \mu\text{g L}^{-1}$) favored pronounced SML formation, as evidenced macroscopically by apparent surface slicks in 12 of the 22 samplings (Table S1). Whenever slicks were present, SML samples were collected directly from the slicks. These gel-like surface slicks are formed primarily by surfactants, are rich in hydrophobic organic matter and can host microorganisms and particles forming a distinct ecosystem in the SML compared to the ULW (Zancker et al., 2017; Wurl et al., 2016). Indeed, accumulation of phytoplankton exudates such as gel-like transparent exopolymer particles (TEPs) has been recorded for the area during the same period (Tzempelikou et al., 2025).

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3.2. Dissolved organic carbon in the SML and UWL

Annual variability of DOC concentrations in the SML and ULW is shown in Fig. 2(c). In the SML, DOC ranged from 1.09 to 8.33 mg L⁻¹, with an average concentration of 3.13 mg L⁻¹, significantly higher than that in the ULW (average: 0.93 mg L⁻¹; range: 0.78–1.23 mg L⁻¹; Mann-Whitney U test, $p = 0.001$) while the enrichment factor ($EF = 2.7 \pm 1.0$; Wilcoxon signed-rank test, $p = 0.001$; Fig. 3(a)) confirms substantial accumulation of DOM components in the SML. DOC enrichment in the SML has been previously reported (Chen et al., 2016; 2022; van Pinxteren et al., 2017), yet is not a consistent feature across marine environments with some studies observing no significant (Obenosterer et al., 2008) or weak enrichment (Wurl and Holmes, 2008; van Pinxteren et al., 2012; Stolle et al., 2020; Barthelmeß and Engel, 2022). Two exceptionally high DOC values observed in the SML (01 July 2016 and 04 May 2017, apparent slicks) were not reflected in the ULW, suggesting episodic enrichment events. No significant relationships were observed between chl-a and either DOC concentrations or DOC enrichment factors indicating that a direct linkage between primary production and DOC dynamics or accumulation in the SML is not evident in this study. Weak or absent relationship between chl-a and DOC concentrations or SML enrichment has been previously reported (Chen et al., 2013; van Pinxteren et al., 2017) despite phytoplankton blooms being recognized as important drivers for DOC accumulation in the SML (Wurl et al., 2016; Bibi et al., 2025). Moreover, no significant relationship was observed between DOC in the two layers ($r^2 = 0.028$, $p = 0.492$), indicating a clear decoupling and suggesting that DOC in the ULW does not exert a measurable influence on DOC in the SML (Fig. 3(a)). This observation contrasts with previous studies in which a strong correlation was reported between the DOC in the SML and the ULW indicating that upward transport of material plays a dominant role in the composition of the SML in DOM (Chen et al., 2013; 2016; 2022; Engel and Galgani, 2016; Engel et al., 2018).

DOC dynamics in the SML tracked those in the ULW during winter but diverged substantially during summer and autumn (Table S1). Consequently, DOC concentrations between the two layers were weakly but significantly correlated ($r = 0.463$; $p = 0.046$, excluding the two extreme values) indicating that apart from upward transport of DOM other processes affect the DOC pool in the SML.

Significant correlation of several DOM compounds between the SML and ULW is a common feature across studies indicating that upward transport of material plays a dominant role in the composition of the SML in DOM (Chen et al., 2013; 2016; Engel and Galgani, 2016). Furthermore, the DOC enrichment factor ($EF = 2.7 \pm 1$; $p = 0.001$; Fig. 3(a)) confirms substantial accumulation of DOM components in the SML.

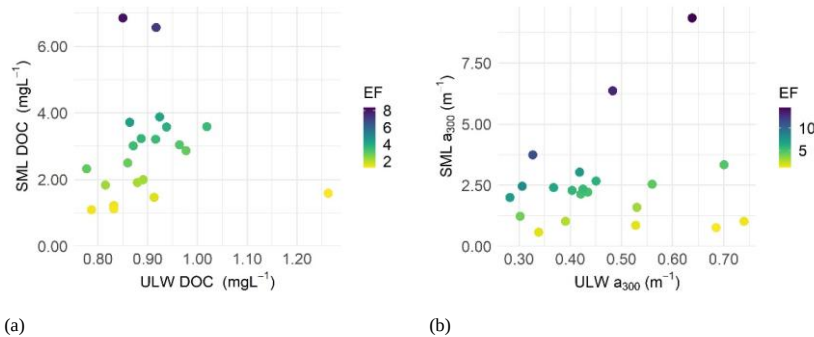


Figure 3: Correlation graphs between the SML (y axis) and ULW (x axis) of (a) DOC (mg L⁻¹); (b) absorption coefficient a_{300} (m⁻¹). The z axis illustrates the enrichment factors

3.3. CDOM dynamics within the DOM pool in SML and ULW

The distribution of the absorption coefficients at 300nm, a_{300} (m⁻¹) in the SML and ULW is presented in Fig. 3 (b), (see Table S3 for wavelengths 265, 280, 370 nm). In the SML, a_{300} ranged from 0.569 to 9.347 m⁻¹, with an average of 2.565 m⁻¹. The distribution of a_{300} (m⁻¹) closely mirrored the one of DOC, displaying maxima on 01 July 2016 and 04 May 2017. As observed in the case of DOC, a_{300} values in the ULW were significantly lower (Mann-Whitney U Test, $p = 0.001$), and the high SML peaks were not reflected in the ULW. The lack of any significant relationship correlation between a_{300} in the two layers (Fig. 3(b)) suggests that CDOM either originates from different sources and/or undergoes distinct processes rather than being influenced primarily by upward flux, highlighting the additional roles of in situ photochemical or biological transformations and/or direct atmospheric inputs to the optically active DOM pool. Penezic et al., (2022) reported lack of CDOM correlation between the SML and ULW even with similar absorption coefficient values in the two layers. Contrary, other studies (Galgani and Engel, 2016; Yang et al., 2022) reported strong correlation of CDOM between the SML and ULW implying a connection of CDOM in the two layers. Enrichment factors (EFs) of $a(\lambda)$ were consistently and significantly greater than unity (Wilcoxon signed-rank test, $p < 0.05$), confirming systematic enrichment of CDOM in the SML as observed in previous works (Obenosterer et al., 2008; Wurl et al., 2009; Pereira et al., 2016; Mustafa et al., 2017; Rickard et al., 2022; Xu et al., 2025). Moreover, EFs increased progressively from the UV-C (a_{265}) (EF = 3.6) to the UV-A/near-visible regions (a_{370}) (EF = 7.4) (Table S4), indicating wavelength-dependent accumulation of chromophoric material at the surface. Higher CDOM absorption in longer wavelengths in the SML compared to the ULW have been reported in previous studies. Pereira et al. (2016), reported higher absorption at 365 nm over 250 nm in the SML in the North East UK coast while Yang et al., (2022) reported higher EFs in longer wavelengths in the eastern marginal seas of China. UV-A/near-visible chromophores are disproportionately enriched at the surface relative to UV-B and UV-C, highlighting a compositional differentiation of CDOM in the SML with higher contribution of HMW DOM compared to the underlying water.

Spectral slope values $S_{275-295}$ in the SML ranged from 0.015 to 0.034 nm⁻¹ with an average of 0.025 nm⁻¹. In the ULW, $S_{275-295}$ values were considerably (Mann-Whitney U test, $p=0.001$) higher than those in the SML fluctuating between 0.029-0.039 nm⁻¹ with an average of 0.034 nm⁻¹.

In the SML, DOC was strongly correlated with a_{300} indicating that the variations in CDOM are largely driven (by 77%) by those in the bulk DOM. In contrast, no significant correlation between DOC and CDOM was observed in the ULW (Table 1). The overall strong coupling between the optically active fraction of DOM and bulk DOM in the SML suggests shared sources and transformation dynamics, whereas their decoupling in the ULW implies that DOM and CDOM respond differently to biogeochemical and photochemical processes taking place in the ULW.

Enrichment factors (EFs) of $a(\lambda)$ were consistently and significantly greater than unity ($p < 0.05$), confirming systematic enrichment of CDOM in the SML as observed in previous works (Obernosterer et al., 2009; Wurl et al., 2009; Lechtenfeld et al., 2014; Mustafa et al., 2017). Moreover, EFs increased progressively from the UV-C (a_{265}) (EF=3.6) to the UV-A/near-visible regions (a_{320}) (EF=7.4) (Table S4), indicating wavelength-dependent accumulation of chromophoric material at the surface. Yang et al., 2022 also reported higher EFs in longer wavelengths in the eastern marginal seas of China. UV-A/near-visible chromophores are disproportionately enriched at the surface, highlighting a decoupling of CDOM constituents in the SML relative to the underlying water.

Table 1. Linear regression analysis of DOC vs absorption coefficients in SML, ULW and rain waters (excluding the outlier values of 01 July 206 and 04 May 2017).

	equation	R^2	p
SML (n=22)			
DOC vs a_{300}	$y=0.844x-0.094$	0.774	0
ULW (n=22)			
DOC vs a_{300}	-	0.073	0.224
Rain (n=14)			
DOC vs a_{300}	$y=1.225x+0.534$	0.878	0

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Table 1: Results of Generalized Linear Model analysis assessing the influence of DOC, a_{300} and water layer on CDOM optical properties and PARAFAC intensities.

Response	Predictor	B	Std. Error	Wald χ^2	p
a_{300}	DOC	0.682	0.537	4.791	0.029
	Layer	0.761	0.522	2.125	0.145
	(SML vs ULW)				
	DOC x Layer	-0.179	0.542	0.109	0.741
$S_{275-295}$	DOC	-0.070	0.242	0.467	0.494
	Layer	-0.079	0.231	0.119	0.730

	(SML vs ULW)				
	DOC x Layer	-0.027	0.234	0.012	0.912
S ₂₇₅₋₂₉₅	a ₃₀₀	-0.618	0.129	32.62	<0.001
	Layer	-0.286	0.075	14.65	<0.001
	(SML vs ULW)				
	a ₃₀₀ x Layer	0.493	0.130	14.37	<0.001
I ₁	a ₃₀₀	1.006	0.383	11.89	0.001
	Layer	1.064	0.224	22.52	<0.001
	(SML vs ULW)				
	a ₃₀₀ x Layer	-0.676	0.388	3.041	0.081
I ₂	a ₃₀₀	0.606	0.456	3.205	0.073
	Layer	1.277	0.270	22.49	<0.001
	(SML vs ULW)				
	a ₃₀₀ x Layer	-0.386	0.462	0.698	0.403
I ₃	a ₃₀₀	1.105	0.494	7.924	0.005
	Layer	0.987	0.295	11.21	0.001
	(SML vs ULW)				
	a ₃₀₀ x Layer	-0.801	0.501	2.559	0.110
I ₄	a ₃₀₀	0.691	1.186	0.961	0.327
	Layer	0.278	0.780	0.127	0.721
	(SML vs ULW)				
	a ₃₀₀ x Layer	-0.193	1.213	0.025	0.874

To investigate potential drivers on CDOM dynamics, generalized linear models were applied as described in Table 1. The relationship between DOC and CDOM was first examined using a GLM with a₃₀₀ as the response variable and DOC and layer as predictors. DOC concentrations significantly explained variation in CDOM absorption at 300 nm across both layers (Wald $\chi^2 = 4.79$, $p = 0.029$). The effect of layers was not significant ($p = 0.145$) and there was not significant interaction between DOC and layer ($p = 0.741$) indicating that the DOC vs a₃₀₀ relationship did not differ between layers. These results suggest that the significantly higher a₃₀₀ values in the SML are primarily attributed to DOC enrichment rather than a layer specific change in CDOM bulk quantity. To further explore CDOM qualitative relationships, we conducted two separate GLMs between spectral slope (S₂₇₅₋₂₉₅), a₃₀₀ and DOC, (Table 1) using either DOC or a₃₀₀ as predictors alongside layer. This approach was adopted in order to avoid multicollinearity and ensure stable parameter estimates since DOC and a₃₀₀ are correlated and the sample size is relatively small ($n = 44$). When DOC and layer were used as predictors and S₂₇₅₋₂₉₅ as response variable, neither DOC ($p = 0.494$) nor the DOC – layer interaction ($p = 0.912$) were significant, indicating that bulk DOC concentration does not explain variation in spectral slope between layers. In contrast, when a₃₀₀ and layer were included as predictors and S₂₇₅₋₂₉₅ as response variable, the GLM revealed significant effects of layer ($p < 0.001$), a₃₀₀ ($p < 0.001$) and the layer – a₃₀₀ interaction ($p < 0.001$). Particularly, S₂₇₅₋₂₉₅ values were significantly lower in the SML (Wald

$\chi^2 = 14.65$, $p < 0.001$), $S_{275-295}$ decreased with increasing a_{300} in the ULW (Wald $\chi^2 = 32.62$, $p < 0.001$) while the $S_{275-295}$ – a_{300} relationship differed between layers (Wald $\chi^2 = 14.37$, $p < 0.001$). These results indicate that $S_{275-295}$ is modulated by CDOM quantity in a layer specific manner, with the SML showing a weaker negative relationship between $S_{275-295}$ and a_{300} than the underlying water. Thus, the GLM analysis of spectral slope suggest that specific processes in the SML, rather than DOC concentration alone, control the optical quality of DOM. Overall, the combined GLM results for DOC and CDOM indices indicate that CDOM enrichment in the SML is not layer specific in terms of quantity, but is layer specific with respect to optical quality.

The negative relationship between spectral slope and absorption coefficients revealed by GLM has been previously reported (Chin et al., 1994; Weishaar et al., 2003; Helms et al., 2008; Tzortziou et al., 2011). This relationship is commonly used to interpret changes in CDOM resulting from photodegradation processes in surface waters (Helms et al., 2008; Fichot and Benner, 2012). The relationship tends to be linear within a compositionally uniform DOM source (Helms et al., 2008; Twardowski and Donaghay, 2001) while it becomes nonlinear (often exponential) across natural gradients encompassing a wide range of CDOM concentrations or mixed sources (e.g., riverine and marine), due to the coexistence of DOM with differing molecular compositions and reactivities (Fichot and Benner, 2012; Nelson and Siegel, 2013). In Fig. 4, this relationship is presented against a_{300} (m^{-1}).

Spectral slope values $S_{275-295}$ in the SML ranged from 0.015 to 0.034 nm^{-1} with an average of 0.025 nm^{-1} . In the ULW $S_{275-295}$ values were considerably ($p=0.001$) higher than those in the SML fluctuating between 0.029–0.039 nm^{-1} with an average of 0.034 nm^{-1} . The increase in spectral slope $S_{275-295}$ (nm^{-1}) i.e. a steeper absorption spectra, usually reflects a decrease in molecular weight, and when plotted against the absorption coefficients a_{300} (m^{-1}), typically exhibits a negative trend (Chin et al., 1994; Weishaar et al., 2003; Helms et al., 2008; Tzortziou et al., 2011). This relationship is commonly used to interpret changes in CDOM resulting from photodegradation processes in surface waters (Helms et al., 2008; Fichot and Benner, 2012). The relationship tends to be linear within a compositionally uniform DOM source (Helms et al., 2008; Twardowski and Donaghay, 2001) while it becomes nonlinear (often exponential) across natural gradients encompassing a wide range of CDOM concentrations or mixed sources (e.g., riverine and marine), due to the coexistence of DOM with differing molecular compositions and reactivities (Fichot and Benner, 2012; Nelson and Siegel, 2013). In Fig. 4, this relationship is presented against a_{300} (m^{-1}). It is evident that for the ULW, the relationship between the two parameters exhibits a steeper slope compared to the SML.

Furthermore, for both layers combined, the a_{300} vs $S_{275-295}$ relationship is well described by an exponential decay equation $R^2=0.791$ ($y=0.020*e^{-1.572*x}+0.023$) indicating different CDOM composition in the two layers. These observations indicate photodegradation of CDOM in both the SML and ULW. Yet, the less steep/moderate slopes found for the a_{300} vs $S_{275-295}$ relationship in the SML suggest that other processes taking place in the SML probably compensate the effect of photodegradation.

When both layers are combined, the a_{300} vs $S_{275-295}$ relationship is well described by an exponential decay equation $R^2 = 0.791$ ($y=0.020*e^{-1.572*x}+0.023$) indicating photodegradation of CDOM in the two layers. However, it is obvious that the relationship in the ULW presents steeper slope (slope = -0.02, $R^2 = 0.533$, $p = 0.000$) than in the SML (slope = -0.003, $R^2 = 0.699$, $p = 0.000$) indicating more intense

photodegradation in the ULW. The less steep slope in the SML suggests that other processes taking place there probably compensate for the effect of photodegradation; leading to the layer specific dependency of spectral slope highlighted by the GLM. Previous studies have also reported persistent CDOM concentrations in the SML hindering photodegradation effect (Obenosterer et al., 2008; Wurl et al., 2009; Xu et al., 2025). Particularly, Xu et al., (2025) reported 16% loss in a_{350} in the SML compared to 30% loss in a_{350} in surface waters. Possible explanations are the continuous reorganization of the hydrophobic DOM in the SML, (Wurl et al., 2009; Lechtenfeld et al., 2014; Mustafa et al., 2017; Penezic et al., 2022) and/or the in-situ production of optically active material in the UV-A near-visible region in the SML. Possible explanations for the less profound photodegradation of CDOM in the SML compared to the ULW are the continuous reorganization of the amphiphilic DOM in the SML (Wurl et al., 2009; Mustafa et al., 2017, Penezic et al., 2022). Rickard et al. 2022, suggested the production of relatively low molecular weight surfactants through CDOM photodegradation in surface waters. On the other hand, biological in-situ production of optically active material in the UV-A/ near-visible region in the SML is also possible. Galgani and Engel (2016) have observed a decrease in spectral slope in the SML due to the abundance of bacterial and phytoplankton cells in parallel to elevated gelatinous material (Coomassie particles and TEPs), thus corroborating to SML specific processes leading to the production of higher molecular weight CDOM.

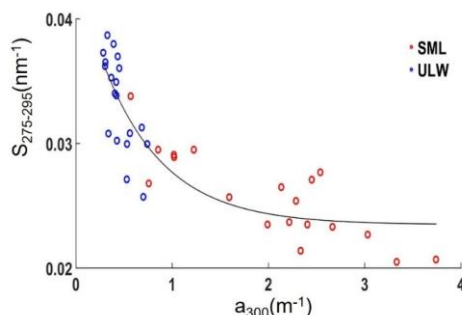


Figure 4: Relationship between $a_{300}(m^{-1})$ and $S_{275-295}(nm^{-1})$ in the SML (red circles) and ULW (blue circles). When both layers are combined the relationship is well approximated by an exponential decay equation denoting multiple chemical structures of CDOM compounds

As previously noted, absorption was noticeably higher in the SML on 01 July 2016 and 04 May 2017, following the trends observed in DOC concentrations. Alongside these elevated absorption values, the SML spectra exhibited two distinct **bands shoulders** at approximately 330 nm and 360 nm, features that were absent in the respective spectra in the ULW (Fig. 5).

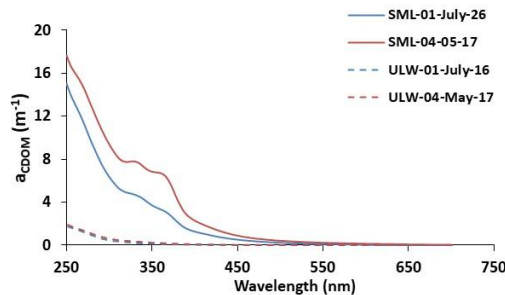


Figure 5: **CDOM A**bsorption spectra in the SML (solid lines) and ULW (dashed lines) on 01 July 2016 (blue) and 04 May 2017 (red). Spectra **bands shoulders with maxima** at 330 nm, 360 nm are indicative of MAA absorption

Tilstone et al. (2010) reported pronounced UV absorption shoulders at 300–340 nm, coinciding with elevated concentrations of mycosporine-like amino acids (MAAs), particularly during surface slicks. These high MAAs levels were linked to coastal phytoplankton blooms and associated exudation, which protect cells from solar irradiance and tend to accumulate in the SML. Similarly, Sabbaghzadeh et al. (2024) observed sporadic UV absorption **bands shoulders** across an Atlantic-wide survey, attributing them also to in situ MAAs production by phytoplankton and bacteria in **high-UV environments exposed to high – UV radiation**. During both sampling events (01 July 2016 and 04 May 2017), surface slicks were clearly observed. Solar irradiance reached some of its annual maximum values (approximately 800 and 750 W m⁻², respectively), while consecutive Saharan dust episodes (30 and 40 µg m⁻³, of dust flux respectively) occurred concurrently. These conditions favoured intense phytoplankton activity as reflected by the elevated **chlorophyll-a** concentrations (0.88 and 0.58 µg L⁻¹, respectively) and corroborate to the release of MAAs into the consolidated SML, contributing to the UV absorption features observed in the CDOM spectra of Fig. 5. Although distinct spectral signatures of MAAs were not discernible during the remaining samplings, their occurrence provides clear evidence of biological processes within the SML that influence CDOM dynamics.

3.3. FDOM composition in the SML and ULW resolved by PARAFAC modelling

In Fig. 6 the components resolved by the PARAFAC model on SML and ULW EEMs are shown. Hereafter C1, C2, C3 and C4 will refer to the components and I₁, I₂, I₃ and I₄ to the **respective fluorescence modelled** intensities **of the respective components** in the SML and ULW samples.

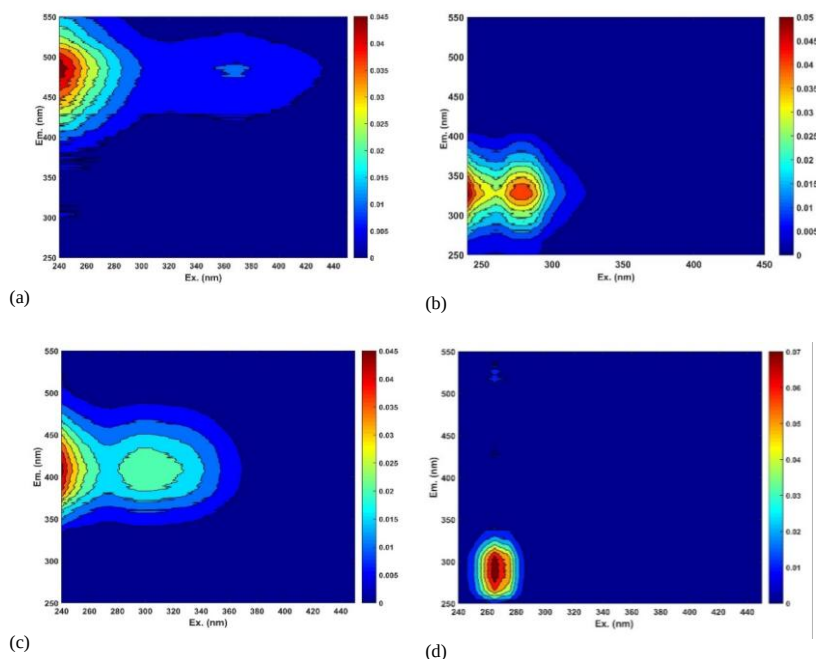


Figure 6: The four components resolved in the SML and ULW by the seawater PARAFAC model, (a) Component C1, peaks A-C, (b) Component C2, peak T, (c) Component C3, peaks A-M, (d) Component C4, peak B

Component C1 presented two excitation maxima at 240 nm and 375 nm and a single emission maximum at 486 nm that resembles peaks A and C described by Coble et al. (1990; 1998). Peak A (ex/em 240/486 nm) has been associated with aromatic highly conjugated humic material origin (McKnight et al., 2001), while peak C (ex/em 375/486 nm) reflects more mature, condensed aromatic humics also of terrestrial origin. The fact that peak C presents longer excitation wavelength compared to peak A implies that the organic substances that fluoresce in the region of peak C are of relatively higher molecular weight and aromaticity compared to the substances that fluoresce in the region of peak A (McKnight et al., 2001). Component C2 presented excitation maxima at 240 nm and 280 nm and emission maximum at 338 nm and is associated with peak T (Coble et al., 1998). The spectral characteristics of component C2 indicate proteinaceous material that resembles tryptophan amino acid (Coble, 1996; Coble et al., 1998), indicative of autochthonous biological production in aquatic systems. The third component, C3 showed two excitation maxima at 240 nm and 295 nm and a single emission maximum at 408 nm. These two peaks resemble peaks A and M respectively. In this case peak A (ex/em 240/408 nm) is resolved in shorter wavelengths. Peak M (ex/em 295/408 nm) is indicative of lower molecular weight, low less aromatic humic substances predominately associated with in situ bacterial humification processes, while there are also references attributing to peak M terrestrial or riverine origin especially in estuaries and coastal

environments (Stedmon and Markager, 2005; Fellman et al., 2010; Catala et al., 2013). A common feature for PARAFAC modelling of FDOM in aquatic systems is that peak A may be deconvoluted in several components. This likely reflects the fact that the fluorescence associated with the humic-like peaks A, C, and M originates from aromatic chromophores that differ in their degree of conjugation, oxidation state, and source. Moreover, in coastal environments, dissolved organic matter is often enriched in lignin-derived compounds, originating from terrestrial plant material. Lignin exhibits strong UV absorption around 280 nm, which closely aligns with the excitation wavelength of fluorescence peak A, and fluoresces in the 420–470 nm range, overlapping with the emission regions of both fluorescence peaks A and C (Del Vecchio and Blough, 2004). Component C4 shows a single excitation and emission maximum at ex/em 270/290 nm similar to the maximum of peak B (Coble et al. 1998). Peak B is representative of proteinaceous material that resembles tyrosine amino acids. This organic material is associated to bacterial degradation processes in contrast to the proteinaceous peak T which is associated with primary production processes. All four components have been previously reported in the SML in the Mediterranean Sea and the global ocean (Galgani and Engel 2016; Pitta et al., 2017; Martínez-Pérez et al., 2019; Drozdowska et al., 2018). In marine environments, humic-like fluorescent components are generally more susceptible to photodegradation than protein-like components. Several studies have shown that exposure to solar radiation leads to a marked decrease in the intensity of humic-like peaks (A, C, M), while the protein-like peaks (T, B) are comparatively less affected (Retelletti Brogi et al., 2020; Romera-Castillo et al., 2011; Zhang et al., 2013). Among the humic-like components, peak C typically exhibits the greatest sensitivity to photobleaching, followed by peak A (Helms et al., 2013; Retelletti Brogi et al., 2020).

3.4. FDOM dynamics within the CDOM pool in the SML and ULW

As illustrated from Y-axes in Fig. 7 and Table S5, in the SML during most months, components C1 (peak A–C) and C2 (peak T) exhibited higher fluorescence intensities (I1: 0.033–0.398 R.U. and I2: 0.038–0.429 R.U.) than components C3 (peak A–M) (I3: 0.016–1.249 R.U.) and C4 (peak B) (I4: 0–0.196 R.U.) consistent with their order of deconvolution by the PARAFAC model. Moreover, fluorescence intensities show roughly equal contributions of the humic-like C1 (average 30.6 %) and tryptophan-like C2 (average 31.7 %) to the FDOM pool. Component C3 (peak A–M) had the highest average fluorescence intensity (0.122 R.U.) and the widest range; however, this was largely due to an outlier recorded on 01 July 2016 (Fig. 7(c)). Excluding this outlier, C3 intensities remained considerably lower than those of C1 and C2 contributing on average 21.1 % to the total FDOM. Component C4 (peak B) displayed the lowest values with only 14.8 % contribution to total FDOM, but higher variability (70.4%) with highest values during summer 2016 and lowest occurred from late winter (February) to mid-spring 2017 (Table S5). The two samplings on 01 July 2016 and 04 May 2017, which exhibited high DOC and $a(\lambda)$ absorption values in the SML, also demonstrated exceptionally strong fluorescence intensities across all four PARAFAC components (2.5, 4.6, 18 and 2.8 times higher than the average of I1, I2, I3 and I4 respectively). Other than that, no clear seasonal trend was observed in any of the four PARAFAC components in the SML. Likewise, in the ULW, the four components showed little seasonal variation.

peaking during autumn bloom on 07 October 2016 (Table S5). In the ULW, components C1, C2, and C3 showed minimal variation overall, peaking during autumn bloom on 07 October 2016. In contrast, component C4 was more variable, with higher intensities from June to October 2016 (Table S5).

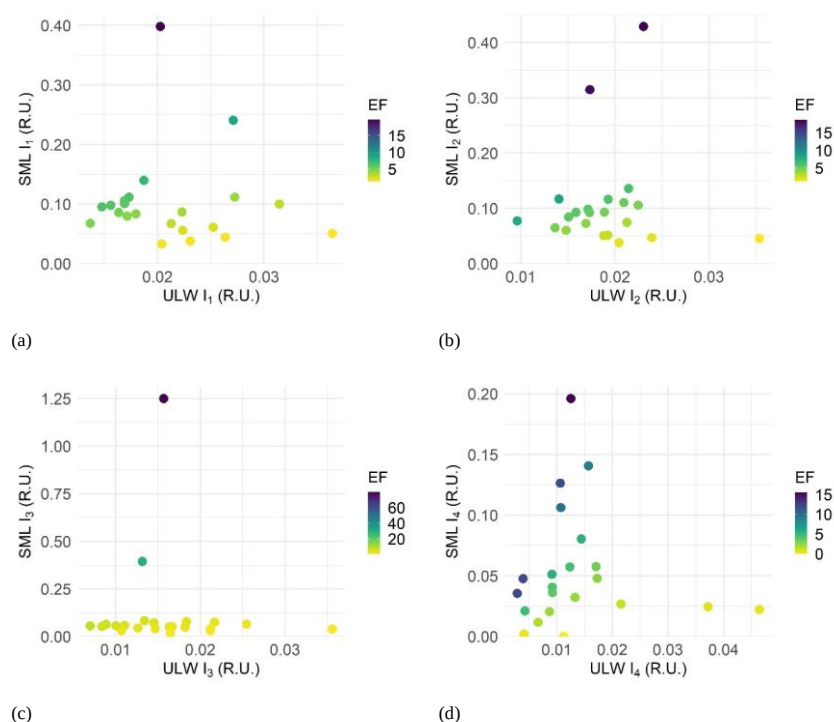


Figure 7: Correlation graphs between the SML (y axis) and ULW (x axis) of PARAFAC maximum fluorescence intensities (R.U.) of (a) I₁; (b) I₂; (c) I₃; (d) I₄. The z axis illustrates the enrichment factors

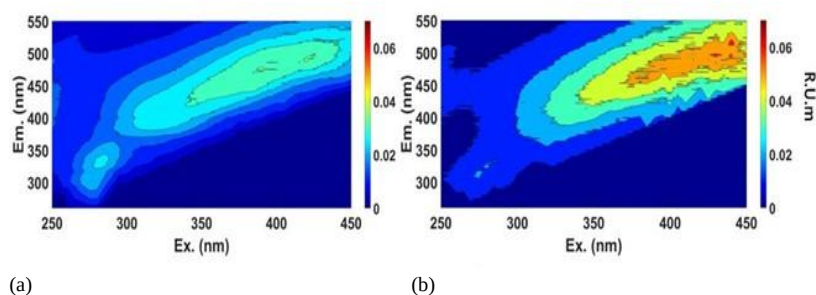
The lack of correlation of the PARAFAC intensities I₁₋₄ between the SML and ULW (Spearman, $p > 0.05$), denotes a decoupling of processes acting upon FDOM in the two water layers (Fig. 7). As with DOC and CDOM, the SML is found enriched in FDOM with most of the EFs fluctuating well above 1 (Wilcoxon test, $p < 0.05$ in all cases) (Fig. 7, Table S4). In particular, the humic like fluorophores (peak A₂/C) and the tryptophan like one (peak T) appear the most enriched in the SML (average EF: 4.2 and 4.5 respectively). Significant enrichment of aromatic amino acids in the SML has been previously documented (Galgani and Engel, 2016; Engel et al., 2018) and has been attributed both to their amphiphilic nature and to in situ production within the SML. Moreover, Yang et al. 2022, reported clearly higher EFs of the tryptophan-like peak in the SML compared to the humic-like and tyrosine-like peaks, indicating a greater contribution of marine autochthonous DOM in the SML relative to the ULW. In our study, the FDOM enrichment within the SML appears to be more evenly distributed between humic (terrestrial) DOM and proteinaceous (autochthonous) DOM, indicating a relatively balanced influence of terrestrial and autochthonous inputs. Miranda et al. (2018) reported enrichment of the humic-like C

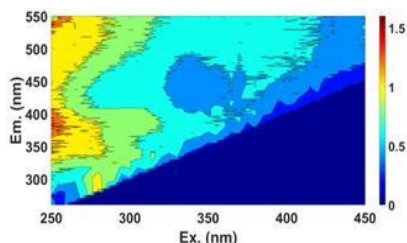
fraction in the SML, which followed the pattern of solar radiation. This enrichment was attributed to photolysis of DOM compounds and enhanced microbial activity, leading to reprocessing of photodegraded DOM into FDOM and thus suggesting that humic-like FDOM can be produced in-situ in the SML under the influence of solar radiation.

Four separate GLMs (Table 1) were fitted for each fluorescent component identified by PARAFAC, with a_{300} and layer included as predictors. I1 and I3 intensities (humic-like components) were significantly higher in the SML than in the ULW (I1: Wald $\chi^2 = 22.52$, $p < 0.001$; I3: Wald $\chi^2 = 11.21$, $p = 0.001$), and both were positively associated with a_{300} (I1: Wald $\chi^2 = 11.89$, $p = 0.001$; I3: Wald $\chi^2 = 7.92$, $p = 0.005$). I2 was primarily layer (SML) enriched (Wald $\chi^2 = 22.49$, $p < 0.001$) with only a weak, non-significant trend with a_{300} (Wald $\chi^2 = 3.21$, $p = 0.073$). In contrast, I4 showed no significant effects of layer, a_{300} , or their interaction ($p > 0.3$). Interactions between a_{300} and layer were not significant for any component (I1: $p = 0.081$; I2: $p = 0.403$; I3: $p = 0.110$; I4: $p = 0.874$). Overall, these results highlight the distinct relationships between humic-like and protein-like components with bulk CDOM as well as layer dependent patterns. The humic-like components C1 and C3 are layer enriched but also positively associated with a_{300} , the tryptophan-like component C2 is primarily layer enriched and independent of a_{300} while the tyrosine-like C4 appears independent of both layer and a_{300} .

As described previously, PARAFAC modelling resolved the terrestrial humic-like fluorophores (peak A) into two components—C1 (peak A-C) and C3 (peak A-M)—reflecting overlapping emission regions typical of marine samples. However, this separation makes it difficult to directly assess the relationship between FDOM and CDOM compounds across the full spectral range. To investigate the relative contribution of FDOM to the bulk CDOM, we calculated the FDOM to CDOM index (FDOM/CDOM, R.U.m) by dividing the full excitation-emission matrix (EEM) of each sample by the corresponding absorption coefficients at the respective excitation wavelengths, as described in Section 2.6 (Fig. S1). The average ratios for all SML and ULW samples are presented in the plots of Fig. 8. This approach provides qualitative insight on the relative importance of the fluorescent to absorbing chromophores present in the two water layers.

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(c)

Figure 8: The FDOM/CDOM index (R.U.-m) in the (a) SML, (b) ULW, (c) the ratio $(\text{FDOM}/\text{CDOM})_{\text{SML}} / (\text{FDOM}/\text{CDOM})_{\text{ULW}}$. (a,b) Depletion of the fluorescence intensity excited in the UV-A/near visible range fluorescence is apparent in the SML relative to the ULW. (c) Higher contribution of FDOM excited at short wavelength (UV-C/UV-B) FDOM to CDOM is apparent in the SML

Figure 8 (a, b) show that FDOM/CDOM index values are elevated at longer excitation wavelengths (> 350 nm) relative to shorter wavelengths in both the SML and ULW, indicating a relatively higher contribution of UV-A/near-visible FDOM to bulk CDOM compared to UV-B and UV-C FDOM. Between layers, the SML appears less enriched in UV-A/near-visible FDOM compared to CDOM, likely reflecting the higher bulk CDOM content in the SML at these wavelengths, as discussed previously based on CDOM indices. CDOM enrichment factors (EFs) increased more than threefold from the UV-C to the UV-A region, consistent with selective enrichment of chromophores absorbing at the UV-A/near visible light chromophores (Table S4). It appears that within the SML, the humic-like FDOM fraction of CDOM is more susceptible to photodegradation than the corresponding humic CDOM fraction, and/or that humic CDOM is produced at a higher rate than humic FDOM within the SML. Photodegradation of terrigenous DOM, leads to the formation of lower molecular weight, less complex DOM with altered optical characteristics (Opsahl & Benner, 1998; Opsahl & Zepp, 2001; Obernosterer & Benner, 2004), while contrary plankton-derived DOM (mostly protein-like and low molecular weight substances) presents minimal photochemical alteration (Obernosterer and Benner, 2004). Previous studies have also demonstrated that FDOM exhibits stronger photodegradation at longer excitation wavelengths than at shorter ones (Helms et al., 2013; Gonsior et al., 2013). Particularly, Helms et al. (2013) observed that irradiation caused the emission maxima of humic-like peaks A and C to shift toward shorter wavelengths, with the greater shift occurring for peak C, while its excitation maximum shifted toward longer wavelengths. These findings, explain the more intense FDOM photodegradation in long wavelengths in the SML, in contrast to CDOM. Moreover At the same time, the SML may be influenced by atmospheric inputs of water-soluble organic carbon (WSOC) with distinct optical signatures. For example, nitro-aromatic compounds present in atmospheric WSOC exhibit strong absorption at 300-500 nm (Huang et al., 2021, Lin et al., 2017) but show minimal fluorescence likely due to the significant reduction in the electron density of the benzene ring by the nitro group (Chen et al., 2020, Cao et al. 2023). Due Because of to the hydrophobic nature of these aromatic compounds forming amphiphilic organic molecules, they are expected to preferentially accumulate in the SML. Overall, the various sources and complex DOM transformation processes occurring in the SML appear to promote a more pronounced accumulation of absorbing but not fluorescent CDOM at longer wavelengths.

In the UV-B and UV-C [excitation](#) regions, where humic-like peak A and protein-like peaks T and B are observed in the EEMs (Ex/Em = 250/485 nm), the FDOM/CDOM index is relatively low in both layers, suggesting that highly absorbing material dominates over fluorescent material in this spectral range in both layers. To examine differences between the SML and ULW in the low-wavelength [excitation](#) region, we calculated the ratio $(\text{FDOM/CDOM})_{\text{sml}} / (\text{FDOM/CDOM})_{\text{ulw}}$ (Fig. 8(c)). This analysis revealed that the SML has a relatively higher FDOM fraction at UV-C and UV-B [excitation](#) wavelengths, indicating that the proportion of highly aromatic, terrestrial humic material and freshly produced protein-like material is greater in the SML than in the ULW. Terrestrial humics and lignin degradation products absorbing at the short UV-B and UV-C regions can be small sized allochthonous molecules, resistant to photodegradation and are expected to enter the coastal environments either via land runoff or via the atmosphere. (Hernes and Benner, 2003; Jaffe et al., 2014)

The preferential enrichment of nitrogen organic compounds, mostly amino-acids, in the SML is documented for diverse coastal and open waters (Liss and Duce, 1997; Kuznetsova and Lee, 2002; Kuznetsova et al., 2004; van Pixteren et al., 2012; Cunliffe et al., 2013; Chen et al., 2022). It is documented that amino acid enrichment in the SML is likely facilitated by other factors rather than excess in situ production (Kuznetsova and Lee, 2002; Kuznetsova et al., 2004). Within the FDOM pool in particular, the accumulation of amino acid like fluorophores (peaks T and B) has been attributed to microbial release by photoprotection mechanisms and/or cell lysis (Galgani and Engel, 2016; Yang et al., 2022). Our study provides indication of episodic MAAs production in the SML. Increased bacterial production in the SML compared to the ULW has been reported in the Mediterranean Sea under the influence of dry atmospheric deposition (Astrahan et al., 2016). As already stated, several dust events occurred during the sampling period, likely stimulating bacterial activity and the subsequent production of [amino acid humic](#)-like compounds in the SML.

Overall, the SML and ULW display distinct FDOM to CDOM profiles. The SML exhibits a higher fluorescent fraction [relative to bulk CDOM](#) in the UV-C and UV-B regions (protein-like peaks and peak A) but a lower fraction in the UV-A/near-visible range (peak C) compared to the ULW. Based on CDOM and FDOM indices, the SML appears enriched in high molecular weight, non-fluorescent [optically active](#) organic compounds, together with low molecular weight fluorescent material of terrestrial and amino acid-like origin. This pattern suggests that the SML acts as a selective accumulation layer, concentrating photoreactive and surface-active DOM fractions, while bulk UV-A fluorescent material is comparatively diluted, reflecting differential contributions of in situ production, atmospheric deposition, and selective photochemical or biological transformations.

3.4. Influence of Wet Deposition on SML Enrichment

Rain events are significant sources of dissolved organic carbon in surface seawater as it was estimated that they contribute about $90 \times 10^{12} \text{ g C yr}^{-1}$ (Willey et al., 2000). Rainwater can be exceptionally rich in CDOM (Kieber et al., 2006) while increased DOM concentrations in the SML during rain events have been reported (Wurl and Obbard 2005; Lim et al., 2007; Ribas-Ribas et al. 2017). Several sources contribute to water soluble organic carbon (WSOC) forms in the atmosphere, such as biomass burning,

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fossil fuel combustion, wind driven soil particulates, including Saharan dust, volatile organic compounds over urban areas, and sea spray over coastal and marine sites (Mladenov et al., 2011; Miyazaki et al., 2018). Studies on the optical properties of WSOC in the atmosphere have shown an overall similarity with the fluorophores of organic molecules found in the marine environment, i.e., humic-like and protein-like substances, yet with distinct differences in their spectral position (Wu et al., 2021 and references therein).

Given the coastal location of the rainwater sampler near an urban area and the prevailing meteorological conditions, the collected samples are expected to reflect contributions from multiple sources. The objective of the CDOM and FDOM analysis in rainwater was to investigate potential linkages between the optical signatures of rainwater and those observed in the SML.

DOC ranged from 0.60 to 6.26 mg L⁻¹ (average 2.49 mg L⁻¹). Absorption coefficients at 300 nm (a_{300}) varied between 0.931 and 8.070 m⁻¹ (average 3.395 m⁻¹), while the spectral slope $S_{275-295}$ exhibited a narrow range 0.021–0.030 nm⁻¹ (average 0.024 nm⁻¹) (Table S6). Rainwater DOM was highly enriched in optically active material, with a strong coupling observed between DOC and a_{300} ($r^2 = 0.878$, $p = 0.001$) (Table 1). Compared to the SML, rainwater exhibited comparable DOC concentrations, slightly higher a_{300} values, and similar average spectral slope $S_{275-295}$, though with a narrower range. PARAFAC analysis of the rainwater EEMs resolved three fluorescent components (R1, R2, and R3) (Table S6, Fig. 9). Component R1, with excitation maxima at 240 and 290 nm and an emission maximum at 394 nm, corresponds to peaks A and M and is analogous to seawater component C3, albeit slightly blue-shifted in emission, suggesting a smaller molecular weight. This component was dominant with fluorescence intensities I_1 ranging from 0.108 R.U. to 1.09 R.U. Component R2 exhibited excitation maxima at 245 and 335 nm and an emission maximum at 456 nm, representing peaks A and C, and was similarly blue-shifted in emission relative to seawater component C1. These spectral shifts indicate differences in the composition of aromatic, high-molecular-weight humic substances between seawater and rainwater. Intensities of component R2 ranged from 0.025 R.U. to 0.389 R.U. Component R3, with excitation maxima at 240 and 270 nm and an emission maximum at 299 nm, corresponds to peak B, whose spectral characteristics closely match those observed in seawater. I_3 fluorescence intensities fluctuated in levels comparable to those of I_2 , from 0.056 R.U. to 0.457 R.U., indicating an equal contribution of components R2 (peak A-C) and R3 (peak B) in rainwater FDOM. All identified rainwater components have been previously reported in studies of CDOM in precipitation (Müller et al., 2008; Santos et al., 2012; Yang et al., 2019). Notably, peak T, resolved in seawater, was not resolved in the rainwater PARAFAC model.

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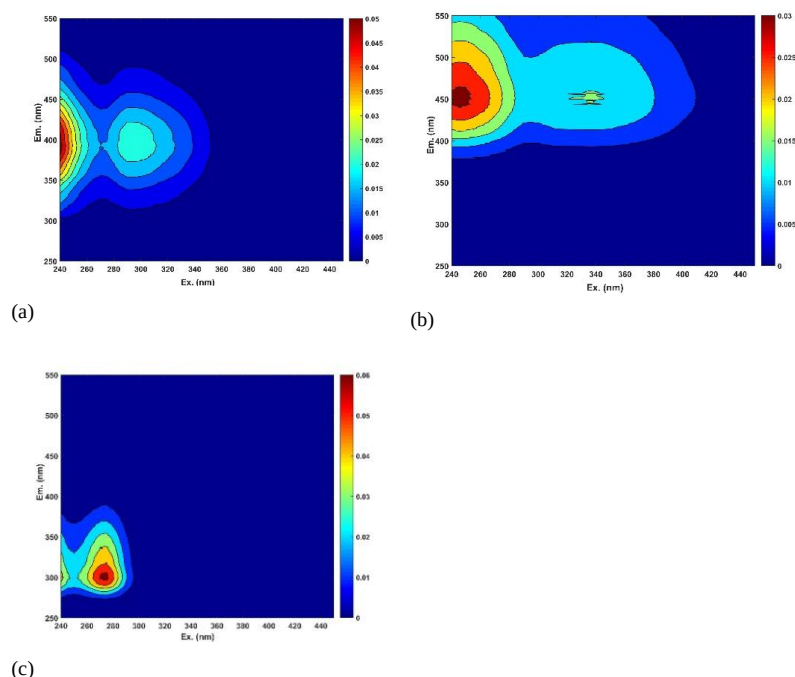


Figure 9: The three components resolved by the rainwater PARAFAC model, (a) Component R1, peaks A-M, (b) Component R2, peaks A-C, (c) Component R3, peak B

The predominance of humic-like fluorophores over protein-like ones in marine aerosols has been reported by Miyazaki et al. (2019), particularly within the spectral range of peak M fluorophores (Ex/Em = 300/400 nm). These authors suggest that during the transfer of DOC to the atmosphere, protein-like compounds undergo substantial decomposition and/or there is preferential formation of humic-like substances in atmospheric aerosols. They also propose that the absence of protein-like fluorophores in sea spray may result from the strong association of these molecules with gel-like exopolymer colloids. Based on laboratory oxidation studies of atmospheric organic carbon, Fan et al. (2019) reported that various oxidation pathways—beyond simple photodegradation—affect atmospheric WSOC, including reactions with hydroxyl radicals (-OH), ozone (O_3), and nitrogen oxides (NO_x). In particular, tryptophan-like fluorophores were found to be highly susceptible to -OH oxidation, leading to their transformation into humic-like and tyrosine-like compounds. Moreover, several works on atmospheric humic-like substances provide evidence that during atmospheric transport degradation processes (photo- and oxidative ones) generate low molecular weight molecules, and of more aliphatic nature (Wu et al., 2021 and references therein).

Consistent with these findings, our data show a clear predominance of humic-like peaks A and M (component R1) in rainwater samples, the absence of tryptophan-like fluorophores among the PARAFAC components, and blue shifted FDOM components R1 (peak A-M) and R2 (peak A-C). These features suggest: (i) a significant contribution of marine aerosols to rainwater over our study area,

and (ii) effective photodegradation and oxidation processes leading to the depletion of tryptophan-like fluorophores and the formation of relatively low-molecular-weight, humic-like compounds. The optical signatures of the rainwater samples further indicate that wet deposition contributes to the CDOM and FDOM pool in the SML primarily through degraded, humic-like components of terrestrial origin (peaks A and C) and tyrosine-like amino acids. At the same time, the prevalence of marine humic-like fluorescence (peak M) points to an efficient recirculation of marine FDOM through sea spray derived marine aerosols. In parallel, the similarities in a_{300} and spectral slope values $S_{275-295}$ with the SML suggest that high molecular weight, light-absorbing but non-fluorescent compounds are present in both rainwater and the SML.

4. Conclusion

This study demonstrates that the sea surface microlayer (SML) functions as a highly dynamic biogeochemical boundary characterized by the selective accumulation and transformation of dissolved organic matter. Although the SML was consistently enriched in DOC, CDOM, and FDOM relative to the ULW, the absence of clear coupling between the two layers indicates that upward transport alone cannot explain DOM enrichment in the SML. Instead, the optical and compositional signatures point to the dominance of in situ processes and atmospheric inputs in shaping microlayer DOM. The strong relationship between DOC and CDOM absorption suggests that bulk carbon availability controls chromophoric material in respect of quantity; however, the layer-dependent behavior of spectral slope reveals that the optical quality of DOM is further modified by processes specific to the SML. Additionally, the investigation of absorption coefficients in various wavelengths covering the UV-C, UV-B, UV-A and near-visible regions of the spectrum, revealed information that could not be discerned through the investigation of absorption at a single wavelength, i.e., 300 nm. A preferential enrichment of higher-molecular-weight, aromatic compounds was observed. This together with the weaker photodegradation signal compared to the ULW, support the view that photochemical alteration is partially counterbalanced by rapid biological production, aggregation, and molecular reorganization at the air-sea interface.

Fluorescence analyses further indicate that the accumulated DOM reflects mixed sources. The concurrent enrichment of humic-like and protein-like components suggests a balanced contribution from terrestrial inputs and marine autochthonous production, while the distinct relationships between fluorescent fractions and CDOM highlight compositional differentiation within the microlayer. The application of the novel FDOM/CDOM index which covers the whole absorption/emission spectrum provided an additional dimension for resolving compositional variability, offering a sensitive optical metric for detecting shifts in bulk CDOM quality and transformation processes at the air-sea interface. Overall, the CDOM and FDOM analyses highlight that integrating optical indices spanning different spectral regions—such as absorption coefficients at multiple wavelengths, spectral slope, and the FDOM/CDOM index—provides comprehensive insights into CDOM–FDOM dynamics and interactions.

Rainwater signatures reinforce the importance of atmospheric deposition as an additional pathway supplying optically altered, humic-like material to the SML, emphasizing the tight connectivity between atmospheric processes and surface ocean biogeochemistry.

Overall, these findings underscore that the SML is not merely an enriched extension of the underlying water but a distinct reactive environment where photochemical, biological, and atmospheric processes interact to regulate DOM composition. Given the central role of surface-active organic matter in air-sea exchange, such transformations may have important implications for carbon cycling and climate-relevant processes.

The SML exhibited consistent enrichment in DOM, CDOM and FDOM relative to the ULW. Upward transport from the ULW to the SML was found more important for bulk DOM while different processes seem to act upon the CDOM and FDOM pools within each layer. Although bulk CDOM was closely coupled to DOM in the SML, based on CDOM and FDOM indices we were able to discriminate variable accumulation patterns of the optically active organic material. Surprisingly the impact of photodegradation processes on the $S_{275-295}/a_{300}$ relationship was less evident in the SML relatively to the ULW. This was attributed to the selective enrichment of high molecular weight optically active material in the UV-A/near visible region, as evidenced by the increase in $a(\lambda)$ EFs for longer wavelengths, and the higher $S_{275-295}$ values relatively to the ULW. By introducing the FDOM/CDOM ratio we further revealed wavelength-dependent enrichment patterns: SML exhibited lower fraction of the more conjugated fluorophores at the UV-A/near visible range relatively to the ULW and higher fractions of low molecular weight terrestrial humics and protein like fluorophores in the UV-C, UV-B regions. In combination with CDOM indices, these observations suggest that the SML favors the selective accumulation of high molecular weight, non-fluorescent organic compounds alongside low molecular weight fluorescent material of terrestrial and amino acid like origin. Several parallel processes may contribute to these patterns including: i) biological transformation of DOM within the SML generating molecules of variable molecular weight, ii) fast reorganisation of the SML macromolecular structure, iii) atmospheric inputs of optically active DOM. For the latter our analysis in rainwater provided evidence of the presence of relatively low molecular weight fluorescent humic-like material (blue shifted peaks) while at the same time low spectral slopes point to the presence of non-fluorescent high molecular weight CDOM in wet deposition. In contrast, CDOM in the ULW exhibited weaker coupling with DOC, higher spectral slopes, and clear evidence of photodegradation, emphasizing the distinct processes governing DOM dynamics between the two layers.

Collectively, these findings highlight the SML as a dynamic, selectively enriched interface where specific DOM fractions accumulate relative to the underlying water, underscoring its chemical and biological significance in marine ecosystems.

Data availability: Detailed data are provided in the Supplementary material of this work.

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References

Astrahan, P., Herut, B., Paytan, A., Rahav, .: The Impact of Dry Atmospheric Deposition on the Sea-Surface Microlayer in the SE Mediterranean Sea: An Experimental Approach, *Front. Mar. Sci.*, 3:222, doi: 10.3389/fmars.2016.00222, 2016

Azimi, S., Ludwig, A., Thévenot, D.R., Colin, J-L.: Trace metal determination in total atmospheric deposition in rural and urban areas, *Sci. Total Environ.*, 308, 247–256, [https://doi.org/10.1016/S0048-9697\(02\)00678-2](https://doi.org/10.1016/S0048-9697(02)00678-2), 2003

Barthelmeß, T., Engel, A.: How biogenic polymers control surfactant dynamics in the surface microlayer: insights from a coastal Baltic Sea study, *Biogeosciences*, 19, 4965–4992, <https://doi.org/10.5194/bg-19-4965-2022>, 2022

Bibi, R., Ribas-Ribas, M., Jaeger, L., Lehnert, C., Gassen, L., Cortés-Espinoza, E. F., Wollschläger, J., Thölen, C., Waska, H., Zöbelein, J., Brinkhoff, T., Athale, I., Röttgers, R., Novak, M., Engel, A., Barthelmeß, T., Karnatz, J., Reinthaler, T., Spriahailo, D., Friedrichs, G., Schäfer, F. A., Wurl, O.: Biogeochemical dynamics of the sea-surface microlayer in a multidisciplinary mesocosm study, *Biogeosciences*, 22, 7563–7589, <https://doi.org/10.5194/bg-22-7563-2025>, 2025.

Blough, N.L., Del Vecchio, R.: Chromophoric DOM in coastal environment, in: *Biogeochemistry of marine dissolved organic matter*, edited by: Hansell, D.A., Carlson, C.A., Academic., 509–546, 2002

Cao, T., Li, M., Xu, C., Song, J., Fan, X., Li, J., Jia, W., and Peng, P.: Technical note: Chemical composition and source identification of fluorescent components in atmospheric water-soluble brown carbon by excitation–emission matrix spectroscopy with parallel factor analysis – potential limitations and applications, *Atmos. Chem. Phys.*, 23, 2613–2625, <https://doi.org/10.5194/acp-23-2613-2023>, 2023.

Catala, T.S., Mladenov, N., Echevarria, F., Reche, I.: Positive trends between salinity and chromophoric and fluorescent dissolved organic matter in a seasonally inverse estuary, *Estuar. Coast. Shelf Sci.*, 133, 206–216, <http://dx.doi.org/10.1016/j.ecss.2013.08.030>, 2013

Formatted: English (United States)

889 [Chen, Y., Yang, G.P., Wu, G.W., Gao, X.C., Xia, Q.Y.: Concentration and characterization of dissolved](#)
890 [organic matter in the surface microlayer and subsurface water of the Bohai Sea, Cont. Shelf Res., 52, 97](#)
891 [– 107, <https://doi.org/10.1016/j.csr.2012.11.007>, 2013](#)

892 [Chen, M., Jaffé, R.: Photo- and bio-reactivity patterns of dissolved organic matter from biomass and soil](#)
893 [leachates and surface waters in a subtropical wetland, Water Research, 61, 181–190,](#)
894 [https://doi.org/10.1016/j.watres.2014.03.075, 2014](#)

895 [Chen, Y., Yang, G.P., Xia, Q.Y., Wu, G.W.: Enrichment and characterization of dissolved organic matter](#)
896 [in the surface microlayer and subsurface water of the South Yellow Sea, Mar. Chem., 182, 1 – 13,](#)
897 [https://doi.org/10.1016/j.marchem.2016.04.001, 2016](#)

898 [Chen, Q., Li, J., Hua, X., Jiang, X., Mu, Z., Wang, M., Wang, J., Shan, M., Yang, X., Fan, X., Song, J.,](#)
899 [Wang, Y., Guan, D., and Du, L.: Identification of species and sources of atmospheric chromophores by](#)
900 [fluorescence excitation-emission matrix with parallel factor analysis, Sci. Total Environ., 718, 137322,](#)
901 [https://doi.org/10.1016/j.scitotenv.2020.137322, 2020.](#)

902 [Chen, Y., Zhang, N., Hu, C., Chen, R., and Yang, G.-P.: Controls on the distribution, enrichment, and](#)
903 [degradation of dissolved organic matter in the sea surface microlayer of the Yellow Sea and East China](#)
904 [Sea, J. Geophys. Res. Oceans, 127, <https://doi.org/10.1029/2022JC019122>, 2022](#)

905 [Chin, Y. P., G. Aiken, O'Loughlin, E.: Molecular-weight, polydispersity, and spectroscopic properties](#)
906 [of aquatic humic substances, Environ. Sci. Technol., 28, 1853–1858, doi: 10.1021/es00060a015, 1994](#)

907 [Coble, P. G., Green, S. A., Blough, N. V., and Gagosian, R. B.: Characterisation of dissolved organic](#)
908 [matter in the Black Sea by fluorescence spectroscopy, Nature, 348, 432–435, doi: 10.1038/348432a0,](#)
909 [1990.](#)

910 [Coble, P. G.: Characterization of marine and terrestrial DOM in seawater using excitation emis5699 sion](#)
911 [matrix spectroscopy, Mar. Chem., 51, 325–346, doi: 10.1016/0304-4203\(95\)00062-3, 1996.](#)

912 [Coble, P. G., Del Castillo, C. E., and Avril, B.: Distribution and optical properties of CDOM in the](#)
913 [Arabian Sea during the 1995 southwest monsoon, Deep-Sea Res. Pt. II, 45, 2195–2223,](#)
914 [https://doi.org/10.1016/S0967-0645\(98\)00068-X, 1998.](#)

915 [Coble, P. G.: Marine optical biogeochemistry: the chemistry of ocean color, Chem. Rev., 107, 402–418,](#)
916 [https://doi.org/10.1021/cr050350+, 2007](#)

917 [Cunliffe, M., Salter, M., Mann, P.J., Whiteley, A.S., Upstill-Goddard, R.C., Murrell, J.C.: Dissolved](#)
918 [organic carbon and bacterial populations in the gelatinous surface microlayer of a Norwegian fjord](#)
919 [mesocosm, FEMS Microbiol. Lett., 299, 248–254, <https://doi.org/10.1111/j.1574-6968.2009.01751.x>,](#)
920 [2009](#)

921 [Cunliffe, M., Engel, A., Frka, S., Gašparović, B., Guitart, C., Murrell, J.C., Salter, M., Stolle, C., Upstill-](#)
922 [Goddard, R., Wurl, O.: Sea surface microlayers: A unified physicochemical and biological perspective](#)

Field Code Changed

923 of the air–ocean interface, *Prog. Oceanogr.*, 109, 104–116, <https://doi.org/10.1016/j.pocean.2012.08.004>,
924 2013

925 Del Vecchio, R., Blough, N.V.: On the origin of the optical properties of humic substances. *Environ. Sci.*
926 *Technol.*, 38, 3885–3891, doi: 10.1021/es049912h, 2004

927 Drozdowska, V., Kowalczyk, P., Konik, M., Dzierzbicka-Glowacka, L.: Study on Different Fractions of
928 Organic Molecules in the Baltic Sea Surface Microlayer by Spectrophot- and Spectrofluorimetric
929 Methods, *Front. Mar. Sci.* 5, 456, doi.org/10.3389/fmars.2018.00456, 2018

930 Ebling, A. and Landing, W.M.: Trace elements in the sea surface microlayer: rapid responses to changes
931 in aerosol deposition, *Elementa: Sci. Anthropocene*, 5:42, <https://doi.org/10.1525/elementa.237>, 2017

932 Engel, A. and Galgani, L.: The organic sea-surface microlayer in the upwelling region off the coast of
933 Peru and potential implications for air–sea exchange processes, *Biogeosciences*, 13, 989–1007,
934 <https://doi.org/10.5194/bg-13-989-2016>, 2016.

935 Engel, A., Bange, H.W., Cunliffe, M., Burrows, S.M., Friedrichs, G., Galgani, L., Herrmann, H.,
936 Hertkorn, N., Johnson, M., Liss, P.S., Quinn, P.K., Schartau, M., Soloviev, A., Stolle, C., Upstill-
937 Goddard, R.C., van Pinxteren, M., Zäncker, B.: The Ocean’s Vital Skin: Toward an Integrated
938 Understanding of the Sea Surface Microlayer, *Front. Mar. Sci.* 4:165, doi: 10.3389/fmars.2017.00165,
939 2017

940 Engel, A., Sperling, M., Sun, C., Grosse, J., Friedrichs, G.: Organic matter in the surface microlayer:
941 Insights from a wind wave channel experiment, *Front. Mar. Sci.*, 5, 182,
942 doi.org/10.3389/fmars.2018.00182, 2018.

943 Fan, X., Yu, X., Wang, Y., Xiao, X., Li, F., Xie, Y., Wei, S., Song, J., Peng, P.: The aging behaviors of
944 chromophoric biomass burning brown carbon during dark aqueous hydroxyl radical oxidation processes
945 in laboratory studies, *Atmos. Environ.*, 213, 391–400, doi:10.1016/j.atmosenv.2019.116909, 2019

946 Fellman, J.B., Hood, E., Spencer, R.G.M.: Fluorescence spectroscopy opens new windows into dissolved
947 organic matter dynamics in freshwater ecosystems: A review, *Limnol., Oceanogr.*, 55, 2452–2462,
948 <https://doi.org/10.4319/lo.2010.55.6.2452>, 2010.

949 Fichot, C.G. and Benner, R.: A novel method to estimate DOC concentrations from CDOM absorption
950 coefficients in coastal waters, *Geophys. Res. Lett.*, 38, doi:10.1029/2010GL046152, 2012.

951 Galgani, L., Engel, A.: Changes in optical characteristics of surface microlayers hint to photochemically
952 and microbially mediated DOM turnover in the upwelling region off the coast of Peru, *Biogeosciences*,
953 13, 2453–2473, <https://doi.org/10.5194/bg-13-2453-2016>, 2016.

954 Gallisai, R., Peters, F., Volpe, G., Basart, S., Baldasano, J.M.: Saharan Dust Deposition May Affect
955 Phytoplankton Growth in the Mediterranean Sea at Ecological Time Scales, *PLoS ONE*, 9,
956 doi:10.1371/journal.pone.0110762, 2014

Field Code Changed

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Formatted: English (United States)

Field Code Changed

Gao, H., Zepp, R. G.: Factors influencing photoreactions of dissolved organic matter in a coastal river of the Southeastern United States, *Environ. Sci. Technol.*, 32, 2940 – 2946, <https://doi.org/10.1021/es9803660>, 1998

Gonsior, M., Schmitt-Kopplin, P., Bastviken, D.: Depth-dependent molecular composition and photo-reactivity of dissolved organic matter in a boreal lake under winter and summer conditions, *Biogeosciences*, 10, 6945–6956, <https://doi.org/10.5194/bg-10-6945-2013>, 2013

Hansen, A.M., Kraus, T.E.C., Pellerin, B.A., Fleck, J.A., Downing, B.D., Bergamaschi, B.A.: Optical properties of dissolved organic matter (DOM): Effects of biological and photolytic degradation, *Limnol. Oceanogr.*, 61, 1015-1032, doi: 10.1002/lno.10270, 2016

Hardy, J.T.: The sea surface microlayer: Biology, chemistry and anthropogenic enrichment, *Prog. Oceanogr.*, 11, 307-328, [https://doi.org/10.1016/0079-6611\(82\)90001-5](https://doi.org/10.1016/0079-6611(82)90001-5), 1982

Harvey, G.W. and Burzell, L.A.: A Simple Microlayer Method for Small Samples, *Limnol. Oceanogr.*, 11, 608-614, doi:10.4319/lno.1966.11.4.0608, 1972

Helms, J.R., Stubbins, A., Ritchie, J.D., Minor, E.C., Kieber, D.J., Mopper, K.: Absorption spectral slopes and slope ratios as indicators of molecular weight, source, and photobleaching of chromophoric dissolved organic matter, *Limnol. Oceanogr.*, 53, 955-969, <https://doi.org/10.4319/lno.2008.53.3.0955>, 2008

Helms, J. R., Stubbins, A., Perdue, E. M., Green, N. W., Chen, H., Mopper, K.: Photochemical bleaching of oceanic dissolved organic matter and its effect on absorption spectral slope and fluorescence, *Mar. Chem.*, 155, 81–91, doi:10.1016/j.marchem.2013.05.015, 2013

Hernes, P. J., and Benner, R.: Photochemical and microbial degradation of dissolved lignin phenols: Implications for the fate of terrigenous dissolved organic matter in marine environments, *J. Geophys. Res. Oceans*, 108(C9), 3291, doi:10.1029/2002JC001421, 2003

Holm-Hansen, O., Lorenzen, C.J., Holmes, R.W., Strickland, J.D.H.: Fluorometric determination of chlorophyll, *Journal de Conseil*, 30 (1), 3–15, <https://doi.org/10.1093/icesjms/30.1.3>, 1965

Huang, R.-J., Yang, L., Shen, J., Yuan, W., Gong, Y., Ni, H., Duan, J., Yan, J., Huang, H., You, Q., and Li, Y. J.: Chromophoric Fingerprinting of Brown Carbon from Residential Biomass Burning, *Environ. Sci. Technol. Lett.*, 9, 102–111, <https://doi.org/10.1021/acs.estlett.1c00837>, 2021.

Huguet, A., Vacher, L., Relexans, S., Saubusse, S., Froidefond, J.M., Parlanti, E.: Properties of fluorescent dissolved organic matter in the Gironde Estuary, *Org. Geochem.*, 40 (6), 706-719, <https://doi.org/10.1016/j.orggeochem.2009.03.002>, 2009

Hunter, K.A.: Processes affecting particulate trace metals in the sea surface microlayer, *Mar. Chem.*, 9, 49–70, [https://doi.org/10.1016/0304-4203\(80\)90006-7](https://doi.org/10.1016/0304-4203(80)90006-7), 1981

Kieber, R. J., Whitehead, R. F., Reid, S. N., Willey, J. D., and Seaton, P. J.: Chromophoric dissolved organic matter (CDOM) in rainwater, Southeastern North Carolina, USA, *J. Atmos. Chem.*, 54, 21–41, doi:10.1007/s10874-005-9008-4, 2006

Kitsiou, D, Karydis, M.: Categorical mapping of marine eutrophication based on ecological indices, *Sci. Total Environ.*, 255 (1-3), 113–27, doi: 10.1016/s0048-9697(00)00457-5. PMID: 10898399, 2000

Kuznetsova, M. and Lee, C.: Dissolved free and combined amino acids in nearshore seawater, sea surface microlayers and foams: influence of extracellular hydrolysis, *Aquat. Sci.*, 64, 252–268, <https://doi.org/10.1007/s00027-002-8070-0>, 2002

Kuznetsova, M., Lee, C., Aller, J., Frew, N.: Enrichment of amino acids in the sea surface microlayer at coastal and open ocean sites in the North Atlantic Ocean, *Limnol. Oceanogr.*, 49, 1605–1619, <https://doi.org/10.4319/lo.2004.49.5.1605>, 2004

Lawaetz, A.J., Stedmon, C.A.: Fluorescence intensity calibration using the Raman scatter peak of water, *Appl Spectrosc.* 63, 936–940. doi: 10.1366/000370209788964548, 2009

Lim, L., Wurl, O., Karupiah, S., and Obbard, J. P.: Atmospheric wet deposition of PAHs to the sea-surface microlayer, *Mar. Pollut. Bull.*, 54(8), 1212–1219, doi:10.1016/j.marpolbul.2007.03.023, 2007

Lin, P., Bluvshtein, N., Rudich, Y., Nizkorodov, S. A., Laskin, J., and Laskin, A.: Molecular Chemistry of Atmospheric Brown Carbon Inferred from a Nationwide Biomass Burning Event, *Environ. Sci. Technol.*, 51, 11561–11570, <https://doi.org/10.1021/acs.est.7b02276>, 2017.

Liss, P.S. and Duce, R.A. (Eds.): The sea surface and global change. Cambridge University Press, 519 pp., ISBN 9780511525025, <https://doi.org/10.1017/CBO9780511525025>, 1997

Lonborg, C., Davidson, K., Alvarez-Salgado X.A.: Bioavailability and bacterial degradation rates of dissolved organic matter in a temperate coastal area during an annual cycle, *Mar. Chem.*, 2009, 113, 219–226, <https://doi.org/10.1016/j.marchem.2009.02.003>, 2009

Logozzo, L., Tzortziou, M., Neale, P., Clark, B. J.: Photochemical and microbial degradation of chromophoric dissolved organic matter exported from tidal marshes, *J. Geophys. Res.: Biogeosciences*, 126, <https://doi.org/10.1029/2020JG005744>, 2021

Mannino, A., Novak, M.G., Nelson, N.B., Belz, M., Berthon, J.-F., Blough, N.V., Boss, E., Bricaud, A., Chaves, J., Del Castillo, C., Del Vecchio, R., D'Sa, E.J., Freeman, S., Matsuoka, A., Miller, R.L., Neeley, A.R., Röttgers, R., Tzortziou, M., Werdell, P.J.: Measurement protocol of absorption by chromophoric dissolved organic matter (CDOM) and other dissolved materials, in: *Inherent Optical Property Measurements and Protocols: Absorption Coefficient*, edited by: Mannino, A. and Novak, M. G., IOCCG Ocean Optics and Biogeochemistry Protocols for Satellite Ocean Colour Sensor Validation, Volume 1.0, IOCCG, Dartmouth, NS, Canada, <http://dx.doi.org/10.25607/OBP-119>, 2019

Field Code Changed

Formatted: English (United States)

Martínez-Pérez, A. M., Catalá, T. S., Nieto-Cid, M., Otero, J., Álvarez, M., Emelianov, M., Reche, I.,
 Álvarez-Salgado, X. A., and Aristegui, J.: Dissolved organic matter (DOM) in the open Mediterranean
 Sea. II: Basin-wide distribution and drivers of fluorescent DOM, *Prog. Oceanogr.*, 170, 93–106,
 doi:10.1016/j.pocean.2018.10.019, 2019

McKnight, D., M., Boyer, E.W., Westerhoff, P.K., Doran, P.T., Thomas Kulbe, T., Andersen, D.T.:
 Spectrofluorometric characterization of dissolved organic matter for indication of precursor organic
 material and aromaticity, *Limnol. Oceanogr.*, 46, 38–48, <https://doi.org/10.4319/lo.2001.46.1.0038>, 2001

Miller, W.L., Zepp, R.G.: Photochemical production of dissolved inorganic carbon from terrestrial
 organic matter: significance to the oceanic organic carbon cycle, *Geophys. Res. Lett.*, 22, 417–420,
<https://doi.org/10.1029/94GL03344>, 1995

Miranda, M.L., Mustaffa, N.I.H., Robinson, T.B., Stolle, C., Ribas-Ribas, M., Wurl, O., Zielinski, O.:
 Influence of solar radiation on biogeochemical parameters and fluorescent dissolved organic matter
 (FDOM) in the sea surface microlayer of the southern coastal North Sea, *Elementa*, 6, 15,
<https://doi.org/10.1525/elementa.278>, 2018

Miyazaki, Y., Yamashita, Y., Kawana, K., Tachibana, E., Kagami, S., Mochida, M., Suzuki, K.,
 Nishioka, J.: Chemical transfer of dissolved organic matter from surface seawater to sea-spray
 water-soluble organic aerosol in the marine atmosphere, *Sci. Rep.*, 8, 14861, doi:10.1038/s41598-018-
 32864-7, 2019

Mladenov, N., Sommaruga, R., Morales-Baquero, R., Laurion, I., Camarero, L., Diéguez, M.C.,
 Camacho, A., Delgado, A., Torres, O., Chen, Z., Felip, M., and Reche, I.: Dust inputs and bacteria
 influence dissolved organic matter in clear alpine lakes, *Nat. Commun.*, 2, 405,
 doi:10.1038/ncomms1411, 2011

Moran, M.A., Sheldon, W.M., Zepp, R.G.: Carbon loss and optical property changes during long term
 photochemical and biological degradation of estuarine dissolved organic matter. *Limnol. Oceanogr.*, 45,
 1254–1264, <https://doi.org/10.4319/lo.2000.45.6.1254>, 2000

Müller, C. L., Baker, A., Hutchinson, R., Fairchild, I. J., Kidd, C.: Analysis of rainwater dissolved
 organic carbon compounds using fluorescence spectrophotometry, *Atmospheric Environment*, 42, 8036–
 8045, <https://doi.org/10.1016/j.atmosenv.2008.06.042>, 2008

Murphy, K., Stedmon, C.A., Graeber, D., Bro, R., Fluorescence spectroscopy and multi-way techniques,
 PARAFAC, *Anal. Methods*, 5, doi: 10.1039/c3ay41160e, 2013

Mustaffa, N.I.H., Ribas-Ribas M., Wurl O.: High-resolution variability of the enrichment of fluorescence
 dissolved organic carbon in the sea surface microlayer of an upwelling region, *Elementa-Sci. Anthropol.*,
 5:242, <https://doi.org/10.1525/elementa.242>, 2017

Field Code Changed

Mustaffa, N.I.H., Badewien, T.H., Ribas-Ribas, M., Wurl, O.: High-resolution observations on enrichment processes in the sea-surface microlayer, *Sci. Rep.*, 8, 13122, <https://doi.org/10.1038/s41598-018-31465-8>, 2018

Nelson, N.B., Carlson, C.A., Steinberg, D.K.: Production of chromophoric dissolved organic matter by Sargasso Sea microbes, *Mar. Chem.*, 89, 273–287, <https://doi.org/10.1016/j.marchem.2004.02.017>, 2004

Nelson, N.B. and Siegel, D.A.: The global distribution and dynamics of chromophoric dissolved organic matter, *Ann. Rev. Mar. Sci.*, 5, 447–476, doi: 10.1146/annurev-marine-120710-100751, 2013

Obernosterer, I., and Benner, R.: Competition between biological and photochemical processes in the mineralisation of dissolved organic carbon, *Limnol. Oceanogr.*, 49, 117–124, doi:10.4319/lo.2004.49.1.0117, 2004

Obernosterer, I., Catala, P., Lami, R., Caparros, J., Ras, J., Bricaud, A., Dupuy, C., van Wambeke, F., Lebaron, P.: Biochemical characteristics and bacterial community structure of the sea surface microlayer in the South Pacific Ocean, *Biogeosciences*, 5, 693–705, <https://doi.org/10.5194/bg-5-693-2008>, 2008

Opsahl, S., and Benner, R.: Photochemical reactivity of dissolved lignin in river and ocean waters, *Limnol. Oceanogr.*, 43, 1297–1304, doi:10.4319/lo.1998.43.6.1297, 1998

Opsahl, S. P., and Zepp, R. G.: Photochemically-induced alteration of stable carbon isotope ratios ($\delta^{13}\text{C}$) in terrigenous dissolved organic carbon, *Geophys. Res. Lett.*, 28, 2417–2420, doi:10.1029/2000GL012686, 2001

Ortega-Retuerta, E., Frazer, T.K., Duarte, C.M., Ruiz-Halpern, S., Tovar-Sanchez, A., Arrieta, J.M., Reche, I.: Biogeneration of chromophoric dissolved organic matter by bacteria and krill in the Southern Ocean, *Limnol. Oceanogr.*, 54, 1941–1950, 2009

Osburn, C.L., Handsel, L.T., Mikan, M.P., Paerl, H.W., Montgomery, M.T.: Fluorescence tracking of dissolved and particulate organic matter quality in a river-dominated estuary, *Environ. Sci. Technol.*, 46, 8628–8636, <https://doi.org/10.1021/es3007723>, 2012

Penezić, A., Drozdowska, V., Novak, Gašparović, B.: Distribution and characterization of organic matter within the sea surface microlayer in the Gulf of Gdańsk, *Oceanologia*, 4, 631–650, <https://doi.org/10.1016/j.oceano.2022.05.003>, 2022

Pereira, R., Schneider-Zapp, K., Upstill-Goddard, R. C.: Surfactant control of gas transfer velocity along an offshore coastal transect: results from a laboratory gas exchange tank, *Biogeosciences*, 13, 3981–3989, <https://doi.org/10.5194/bg-13-3981-2016>, 2016.

Pereira, R., Ashton, I., Sabbaghzadeh, B., Shutler, J.D., Upstill-Goddard, R.C.: Reduced air–sea CO₂ exchange in the Atlantic Ocean due to biological surfactants, *Nat. Geosci.*, 11, 492–496, <https://doi.org/10.1038/s41561-018-0136-2>, 2018

Pitta, E., Zeri, C., Tzortziou, M., Mousdis, G., Scoullou, M.: Seasonal variations in dissolved organic matter composition using absorbance and fluorescence spectroscopy in the Dardanelles Straits - North Aegean Sea mixing zone, *Cont. Shelf Res.*, 149, 82-95, <https://doi.org/10.1016/j.csr.2016.07.013>, 2017

Pitta, E., & Zeri, C.: The impact of combining data sets of fluorescence excitation-emission matrices of dissolved organic matter from various aquatic sources on the information retrieved by PARAFAC modelling, *Spectrochim. Acta A Mol. Biomol. Spectrosc.*, 258, 119800, 2021.

Retelletti Brogi, S., Charrière, B., Gonnelli, M., Vaultier, F., Sempéré, R., Vestri, S., Santinelli, C.: Effect of UV and Visible Radiation on Optical Properties of Chromophoric Dissolved Organic Matter Released by *Emiliania huxleyi*, *J. Mar. Sci. Eng.*, 8(11), 888, [doi:10.3390/jmse8110888](https://doi.org/10.3390/jmse8110888), 2020

Ribas-Ribas, M., Mustafa, N.I.H., Rahlff, J., Stolle, C. and Wurl, O.: Sea Surface Scanner (S3): A catamaran for high-resolution measurements of biochemical properties of the sea surface microlayer, *J. Atmos. Oceanic. Technol.*, 34, 1433–1448, <https://doi.org/10.1175/JTECH-D-17-0017.1>, 2017

Rickard, P.C., Uher, G., Upstill-Goddard, R.: Photo-reactivity of surfactants in the sea-surface microlayer and subsurface water of the Tyne estuary, UK, *Geophys. Res. Lett.*, 49, [doi:10.1029/2021GL095469](https://doi.org/10.1029/2021GL095469), 2022

Rochelle-Newall, E.J., Fisher, T.R.: Production of chromophoric dissolved organic matter fluorescence in marine and estuarine environments: an investigation into the role of phytoplankton, *Mar. Chem.*, 77, 7–21, [https://doi.org/10.1016/S0304-4203\(01\)00072-X](https://doi.org/10.1016/S0304-4203(01)00072-X), 2002

Romera-Castillo, C., Sarmento, H., Álvarez-Salgado, X.A., Gasol J.M., Marrase C.: Production of chromophoric dissolved organic matter by marine phytoplankton, *Limnol. Oceanogr.* 55, 446–454, <https://doi.org/10.4319/lo.2010.55.1.0446>, 2010

Romera-Castillo, C., Nieto-Cis, M., Castro, C.G., Marrase, C., Largier, J., Barton, E.D., Alvarez-Salgado, X.A.: Fluorescence: absorption coefficient ratio – tracing photochemical and microbial degradation processes affecting coloured dissolved organic matter in a coastal system, *Mar. Chem.*, 125, 26-38, <https://doi.org/10.1016/j.marchem.2011.02.001>, 2011.

Sabbaghzadeh, B., Upstill-Goddard, R.C., Beale, R. Pereira, R. Nightingale, P.D.: The Atlantic Ocean surface microlayer from 50°N to 50°S is ubiquitously enriched in surfactants at wind speed up to 13 m s⁻¹, *Geophys. Res. Lett.*, 44, 2852-2858, <https://doi.org/10.1002/2017GL072988>, 2017

Sabbaghzadeh, B., Uher, G., Upstill-Goddard, R.: Dynamics of chromophoric dissolved organic matter in the Atlantic Ocean: unravelling province-dependent relationships, optical complexity, and environmental influences, *Front. Mar. Sci.*, 11, <https://doi.org/10.3389/fmars.2024.1432133>, 2024

Santos, P. S. M., Santos, E. B. H., Duarte, A. C.: First spectroscopic study on the structural features of dissolved organic matter isolated from rainwater in different seasons, *Sci. Total Environ.*, 426, 172–179, [doi:10.1016/j.scitotenv.2012.03.023](https://doi.org/10.1016/j.scitotenv.2012.03.023), 2012

Field Code Changed

Field Code Changed

Stedmon C.A. and Markager, S.: Tracing the production and degradation of autochthonous fractions of dissolved organic matter by fluorescence analysis, *Limnol. Oceanogr.*, 50, 1415-1426, <https://doi.org/10.4319/lo.2005.50.5.1415>, 2005

Stedmon, C.A., Thomas, D.N., Granskog, M.A., Kaartokallio, H., Papadimitriou, S., Kuosa, H.: Characteristics of dissolved organic matter in Baltic coastal sea ice: Allochthonous or Autochthonous origins?, *Environ. Sci. Technol.*, 41, 7273-7279, doi:10.1021/es071210f, 2007

Stolle, C., Nagel, K., Labrenz, M., Jurgens, K.: Succession of the sea-surface microlayer in the coastal Baltic Sea under natural and experimentally induced low-wind conditions, *Biogeosciences*, 7, 2975-2988, <https://doi.org/10.5194/bg-7-2975-2010>, 2010

Stolle, C., Ribas-Ribas, M., Badewien, T. H., Barnes, J., Carpenter, L. J., Chance, R., Damgaard, L. R., Quesada, A. M. D., Engel, A., Frka, S., Galgani, L., Gašparović, B., Gerriets, M., Mustaffa, N. I. H., Herrmann, H., Kallajoki, L., Pereira, R., Radach, F., Revsbech, N. P., Rickard, P., Saint, A., Salter, M., Striebel, M., Triesch, N., Uher, G., Upstill-Goddard, R. C., van Pinxteren, M., Zäncker, B., Zieger, P., Wur, O.: The Milan Campaign: Studying diel light effects on the air-sea interface, *Bull. Am. Meteorol. Soc.* 101, 146-166. <https://doi.org/10.1175/BAMS-D-17-0329.2>, 2020

Tilstone, G.H., Ains, R.L., Vicente, V.M., Widdicombe, C., Llewellyn, C.: High concentrations of mycosporine-like amino acids and colored dissolved organic matter in the sea surface microlayer off the Iberian Peninsula, *Limnol. Oceanogr.* 55, 1835–1850. doi: 10.4319/lo.2010.55.5.1835, 2010

Twardowski, M.S. and Donaghay, P.L.: Separating in situ and terrigenous sources of absorption by dissolved materials in coastal waters, *J. Geophys. Res.*, 106, 2545-2560, <https://doi.org/10.1029/1999JC000039>, 2001

Tzempelikou, E., Zeri, C., Pitta, E., Assimakopoulou, G., Pavlidou, A., Rousselaki, E.: Temporal dynamics of transparent exopolymer particles in surface waters and sea surface microlayer in the coastal Eastern Mediterranean Sea, *Reg. Stud. Mar. Sci.* 85, <https://doi.org/10.1016/j.rsma.2025.104163>, 2025

Tzortziou, M., Neale, P.J., Megonigal, J.P., Pow, C.L., Butterworth, M.: Spatial gradients in dissolved carbon due to tidal marsh outwelling into a Chesapeake Bay estuary, *Mar. Ecol. Prog. Ser.*, 426, 41–56, doi: 10.3354/meps09017, 2011

Upstill-Goddard, R.C.: Air-sea gas exchange in the coastal zone, *Estuar. Coast. Shelf Sci.*, 70, 388–404, <https://doi.org/10.1016/j.ecss.2006.05.043>, 2006

Vahatalo, A.V., Wetzel, R.G.: Photochemical and microbial decomposition of chromophoric dissolved organic matter during long (months-years) exposures, *Mar. Chem.*, 89, 313–326, <https://doi.org/10.1016/j.marchem.2004.03.010>, 2004

van Pinxteren, M., Mav E., Herrmann, H., Herrmann, H.: Chemical characterization of dissolved organic compounds from coastal sea surface microlayers (Baltic Sea, Germany), *Environ. Sci. Technol.*, 46, 10455–10462, doi:10.1021/es204492b, 2012.

van Pinxteren, M., Barthel, S., Fomba, K., W., Muller, K., von Tumpling, W., Herrmann, H.: The influence of environmental drivers on the enrichment of organic carbon in the sea surface microlayer and in submicron aerosol particles – measurements from the Atlantic Ocean, *Elementa*, 5, 35, doi.org/10.1525/elementa.225, 2017

Weishaar, J.L., Aiken, G.R., Bergamaschi, B.A., Fram, M.S., Fujii, R., Mopper, K.: Evaluation of Specific Ultraviolet Absorbance as an Indicator of the Chemical Composition and Reactivity of Dissolved Organic Carbon, *Environ. Sci. Technol.*, 37, 4702–4708, <https://doi.org/10.1021/es030360x>, 2003

Willey, J.D., Kieber, R.J., Eyman, M.S., Avery, G.B.Jr.: Rainwater dissolved organic carbon: concentrations and global flux, *Glob. Biogeochem. Cycles*, 14(1), 139–148, doi:10.1029/1999GB900036, 2000

Wu, G., Fu, P., Ram, K., Song, J., Chen, Q., Kawamura, K., Wan, X., Kang, S., Wang, X., Laskin, A., Cong, Z.: Fluorescence characteristics of water-soluble organic carbon in atmospheric aerosol, *Environ. Pollut.*, 268, 115906, doi:10.1016/j.envpol.2020.115906, 2021

Wurl, O. and Obbard, J.P.: A review of pollutants in the sea-surface microlayer (SML): a unique habitat for marine organisms, *Mar. Pollut. Bull.*, 48, 1016-1030, <https://doi.org/10.1016/j.marpolbul.2004.03.016>, 2005

Wurl, O. and Holmes, M.: The gelatinous nature of the sea-surface microlayer, *Mar. Chem.*, 110, 89-97, <https://doi.org/10.1016/j.marchem.2008.02.009>, 2008

Wurl, O., Miller, L., Röttgers, R., Vagle, S.: The distribution and fate of surface-active substances in the sea-surface microlayer and water column, *Mar. Chem.*, 115, 1–9, <https://doi.org/10.1016/j.marchem.2009.04.007>, 2009

Wurl, O., Stolle, C., Van Thuoc, C., The Thu, P., Mari, X.: Biofilm-like properties of the sea surface and predicted effects on air-sea CO₂ exchange, *Prog. Oceanogr.* doi: 10.1016/j.pocean.2016.03.002, 2016

Xu, G.B., Yan, S.B., Wang, J., Xu, F., Ji, X., Ni, J., Liu, Z., Yang, G.P.: Carbon monoxide cycling in the marginal sea: A case of the yellow sea and the east China sea, *Mar. Environ. Res.*, 210, 107328, <https://doi.org/10.1016/j.marenvres.2025.107328>, 2025

Yang, L., Chen, W., Zhuang, W.-E., Cheng, Q., Li, W., Wang, H., Guo, W., Chen-Tung Arthur Chen, Liu, M.: Characterization and bioavailability of rainwater dissolved organic matter at the southeast coast of China using absorption spectroscopy and fluorescence EEM-PARAFAC, *Mar. Chem.*, 210, 11–22, doi:10.1016/j.marchem.2018.10.021, 2019

Field Code Changed

Formatted: English (United States)

Yang, L., Zhang, J., Engel, A., Yang, G.-P.: Spatio-temporal distribution, photoreactivity and environmental control of dissolved organic matter in the sea-surface microlayer of the eastern marginal seas of China, *Biogeosciences*, 19, 5251–5268, doi:10.5194/bg-19-5251-2022, 2022

Zäncker, B., Bracher, A., Röttgers, R., Engel, A., Variations of the Organic Matter Composition in the Sea Surface Microlayer: A Comparison between Open Ocean, Coastal, and Upwelling Sites Off the Peruvian Coast, *Front. Microbiol.*, 8:2369, doi: 10.3389/fmicb.2017.02369, 2017

Zervoudaki, S., Siokou, I., Krasakopoulou, E., Kontoyiannis, H., Pavlidou, A., Assimakopoulou, G., Katsiaras, N., Reizopoulou, S., Karageorgis, A.P., Kaberi, H., Lardi, P.I., Gerakaris, V., Tsiamis, K., Salomidi, M., Zeri, C., Pitta, E., Strogyloudi, E., Parinos, C., Hatzianestis, I., Christou, E.D., Zoulias, T., Sakavara, A., Pagou, K., Zenetos, A., Panayotidis, P., Simbourn, N.: Biogeochemical Characteristics in the Saronikos Gulf (Aegean Sea, Eastern Mediterranean), in: *The Aegean Sea Environment: Anthropogenic Presence and Impact*, edited by Anagnostou, C.L., Kostianoy, A.G., Mariolakis, I.D., Panayotidis, P., Soilemezidou, M., Tsaltas G., *Hdb Environ. Chem.*, Springer, 142, 511-568, https://doi.org/10.1007/698_2022_898, 2022

Zhang, Z., Cai, W., Liu, L., Liu, C., Chen, F.: Direct determination of thickness of sea surface microlayer using a pH microelectrode at original location, *Sc. China Ser. B-Chem.*, 46, 339–351, <https://doi.org/10.1360/02yb0192>, 2003

Zhang, Y., Liu, X., Osburn, C.L., Wang, M., Qin, B., Zhou, Y.: Photobleaching response of different sources of chromophoric dissolved organic matter exposed to natural solar radiation using absorption and excitation-emission matrix spectra, *PLoS One*, 8(10):e77515. doi: 10.1371/journal.pone.0077515, 2013

Asher, W.: The sea surface microlayer and its effect on global air-sea gas transfer, in: *The sea surface and global change*, edited by: Liss, P.S., and Duce, R.A., Cambridge University Press, 251–286, <https://doi.org/10.1017/CBO9780511525025.009>, 1997

Astrahan, P., Herut, B., Paytan, A., Rahav, .: The Impact of Dry Atmospheric Deposition on the Sea Surface Microlayer in the SE Mediterranean Sea: An Experimental Approach, *Front. Mar. Sci.*, 3:222, doi: 10.3389/fmars.2016.00222, 2016

Azimi, S., Ludwig, A., Thévenot, D.R., Colin, J-L.: Trace metal determination in total atmospheric deposition in rural and urban areas, *Sci. Total Environ.*, 308, 247–256, [https://doi.org/10.1016/S0048-9697\(02\)00678-2](https://doi.org/10.1016/S0048-9697(02)00678-2), 2003

Gao, T., Li, M., Xu, C., Song, J., Fan, X., Li, J., Jia, W., and Peng, P.: Technical note: Chemical composition and source identification of fluorescent components in atmospheric water-soluble brown carbon by excitation-emission matrix spectroscopy with parallel factor analysis—potential limitations and applications, *Atmos. Chem. Phys.*, 23, 2613–2625, <https://doi.org/10.5194/acp-23-2613-2023>, 2023.

Catala, T.S., Mladenov, N., Echevarria, F., Reche, I.: Positive trends between salinity and chromophoric and fluorescent dissolved organic matter in a seasonally inverse estuary, *Estuar. Coast. Shelf Sci.*, **133**, 206–216, <http://dx.doi.org/10.1016/j.eess.2013.08.030>, 2013

Chen, Y., Yang, G.P., Wu, G.W., Gao, X.C., Xia, Q.Y.: Concentration and characterization of dissolved organic matter in the surface microlayer and subsurface water of the Bohai Sea, *Cont. Shelf Res.*, **52**, 97–107, <https://doi.org/10.1016/j.csr.2012.11.007>, 2013

Chen, Y., Yang, G.P., Xia, Q.Y., Wu, G.W.: Enrichment and characterization of dissolved organic matter in the surface microlayer and subsurface water of the South Yellow Sea, *Mar. Chem.*, **182**, 1–13, <https://doi.org/10.1016/j.marchem.2016.04.001>, 2016

Chen, Q., Li, J., Hua, X., Jiang, X., Mu, Z., Wang, M., Wang, J., Shan, M., Yang, X., Fan, X., Song, J., Wang, Y., Guan, D., and Du, L.: Identification of species and sources of atmospheric chromophores by fluorescence excitation emission matrix with parallel factor analysis, *Sci. Total Environ.*, **718**, 137322, <https://doi.org/10.1016/j.scitotenv.2020.137322>, 2020.

Chen, Y., Zhang, N., Hu, C., Chen, R., and Yang, G.-P.: Controls on the distribution, enrichment, and degradation of dissolved organic matter in the sea surface microlayer of the Yellow Sea and East China Sea, *J. Geophys. Res. Oceans*, **127**, <https://doi.org/10.1029/2022JC019122>, 2022

Chin, Y. P., G. Aiken, O'Loughlin, E.: Molecular weight, polydispersity, and spectroscopic properties of aquatic humic substances, *Environ. Sci. Technol.*, **28**, 1853–1858, doi: 10.1021/es00060a015, 1994

Coble, P. G., Green, S. A., Blough, N. V., and Gagosian, R. B.: Characterisation of dissolved organic matter in the Black Sea by fluorescence spectroscopy, *Nature*, **348**, 432–435, doi: 10.1038/348432a0, 1990.

Coble, P. G.: Characterization of marine and terrestrial DOM in seawater using excitation emission matrix spectroscopy, *Mar. Chem.*, **51**, 325–346, doi: 10.1016/0304-4203(95)00062-3, 1996.

Coble, P. G., Del-Castillo, C. E., and Avril, B.: Distribution and optical properties of CDOM in the Arabian Sea during the 1995 southwest monsoon, *Deep-Sea Res. Pt. II*, **45**, 2195–2223, [https://doi.org/10.1016/S0967-0645\(98\)00068-X](https://doi.org/10.1016/S0967-0645(98)00068-X), 1998.

Coble, P. G.: Marine optical biogeochemistry: the chemistry of ocean color, *Chem. Rev.*, **107**, 402–418, <https://doi.org/10.1021/cr050350+>, 2007

Cunliffe, M., Salter, M., Mann, P.J., Whiteley, A.S., Upstill-Goddard, R.C., Murrell, J.C.: Dissolved organic carbon and bacterial populations in the gelatinous surface microlayer of a Norwegian fjord mesocosm, *FEMS Microbiol. Lett.*, **299**, 248–254, <https://doi.org/10.1111/j.1574-6968.2009.01751.x>, 2009

Cunliffe, M., Engel, A., Frka, S., Gašparović, B., Guitart, C., Murrell, J.C., Salter, M., Stolle, C., Upstill-Goddard, R., Wurl, O.: Sea surface microlayers: A unified physicochemical and biological perspective

Field Code Changed

of the air–ocean interface, *Prog. Oceanogr.*, 109, 104–116, <https://doi.org/10.1016/j.pocean.2012.08.004>, 2013

Del Vecchio, R., Blough, N.V.: On the origin of the optical properties of humic substances, *Environ. Sci. Technol.*, 38, 3885–3891, doi: 10.1021/es049912h, 2004

Drozdowska, V., Kowalczyk, P., Konik, M., Dzierzbicka-Glowacka, L.: Study on Different Fractions of Organic Molecules in the Baltic Sea Surface Microlayer by Spectrophotometric and Spectrofluorimetric Methods, *Front. Mar. Sci.*, 5, 456, 2018

Ebling, A. and Landing, W.M.: Trace elements in the sea surface microlayer: rapid responses to changes in aerosol deposition, *Elementa: Sci. Anthropocene*, 5:42, <https://doi.org/10.1525/elementa.237>, 2017

Engel, A. and Galgani, L.: The organic sea surface microlayer in the upwelling region off the coast of Peru and potential implications for air–sea exchange processes, *Biogeosciences*, 13, 989–1007, <https://doi.org/10.5194/bg-13-989-2016>, 2016.

Fan, X., Yu, X., Wang, Y., Xiao, X., Li, F., Xie, Y., Wei, S., Song, J., Peng, P.: The aging behaviors of chromophoric biomass-burning brown carbon during dark aqueous hydroxyl radical oxidation processes in laboratory studies, *Atmos. Environ.*, 213, 391–400, doi:10.1016/j.atmosenv.2019.116909, 2019

Fellman, J.B., Hood, E., Spencer, R.G.M.: Fluorescence spectroscopy opens new windows into dissolved organic matter dynamics in freshwater ecosystems: A review, *Limnol. Oceanogr.*, 55, 2452–2462, <https://doi.org/10.4319/lo.2010.55.6.2452>, 2010.

Fichot, C.G. and Benner, R.: A novel method to estimate DOC concentrations from CDOM absorption coefficients in coastal waters, *Geophys. Res. Lett.*, 38, doi:10.1029/2010GL046152, 2011.

Gallaisai, R., Peters, F., Volpe, G., Basart, S., Baldasano, J.M.: Saharan Dust Deposition May Affect Phytoplankton Growth in the Mediterranean Sea at Ecological Time Scales, *PLoS ONE*, 9, doi:10.1371/journal.pone.0110762, 2014

Gao, H., Zepp, R. G.: Factors influencing photoreactions of dissolved organic matter in a coastal river of the Southeastern United States, *Environ. Sci. Technol.*, 32, 2940–2946, <https://doi.org/10.1021/es9803660>, 1998

Gonsior, M., Schmitt-Kopplin, P., Bastviken, D.: Depth-dependent molecular composition and photoreactivity of dissolved organic matter in a boreal lake under winter and summer conditions, *Biogeosciences*, 10, 6945–6956, doi:10.5194/bg-10-6945-2013, 2013

Hardy, J.T.: The sea surface microlayer: Biology, chemistry and anthropogenic enrichment, *Prog. Oceanogr.*, 11, 307–328, [https://doi.org/10.1016/0079-6611\(82\)90001-5](https://doi.org/10.1016/0079-6611(82)90001-5), 1982

Harvey, G.W. and Burzell, L.A.: A Simple Microlayer Method for Small Samples, *Limnol. Oceanogr.*, 11, 608–614, doi:10.4319/lo.1966.11.4.0608, 1972

Field Code Changed

Helms, J.R., Stubbins, A., Ritchie, J.D., Minor, E.C., Kieber, D.J., Mopper, K.: Absorption spectral slopes and slope ratios as indicators of molecular weight, source, and photobleaching of chromophoric dissolved organic matter, *Limnol. Oceanogr.*, 53, 955–969, <https://doi.org/10.4319/lo.2008.53.3.0955>, 2008

Helms, J. R., Stubbins, A., Perdue, E. M., Green, N. W., Chen, H., Mopper, K.: Photochemical bleaching of oceanic dissolved organic matter and its effect on absorption spectral slope and fluorescence, *Mar. Chem.*, 155, 81–91, doi:10.1016/j.marchem.2013.05.015, 2013

Hernes, P. J., and Benner, R.: Photochemical and microbial degradation of dissolved lignin phenols: Implications for the fate of terrigenous dissolved organic matter in marine environments, *J. Geophys. Res. Oceans*, 108(C9), 3291, doi:10.1029/2002JC001421, 2003

Hoge, F.E., Williams, M.E., Swift, R.N., Yungel, J.K., Vodacek, A.: Satellite retrieval of the absorption coefficient of chromophoric dissolved organic matter in continental marginal, *J. Geophys. Res.*, 100, <https://doi.org/10.1029/95JC02561>, 24847–24854, 1995

Huang, R. J., Yang, L., Shen, J., Yuan, W., Gong, Y., Ni, H., Duan, J., Yan, J., Huang, H., You, Q., and Li, Y. J.: Chromophoric Fingerprinting of Brown Carbon from Residential Biomass Burning, *Environ. Sci. Technol. Lett.*, 9, 102–111, <https://doi.org/10.1021/acs.estlett.1c00837>, 2021.

Hunter, K.A.: Processes affecting particulate trace metals in the sea surface microlayer, *Mar. Chem.*, 9, 49–70, [https://doi.org/10.1016/0304-4203\(80\)90006-7](https://doi.org/10.1016/0304-4203(80)90006-7), 1980

Jaffé, R., Cawley, K. M., and Yamashita, Y.: Applications of Excitation-Emission Matrix Fluorescence with Parallel Factor Analysis (EEM-PARAFAC) in Assessing Environmental Dynamics of Natural Dissolved Organic Matter (DOM) in Aquatic Environments: A Review, *ACS Symp. Ser.*, 1160, 27–73, doi:10.1021/bk-2014-1160.ch003, 2014

Kieber, R. J., Whitehead, R. F., Reid, S. N., Willey, J. D., and Seaton, P. J.: Chromophoric dissolved organic matter (CDOM) in rainwater, Southeastern North Carolina, USA, *J. Atmos. Chem.*, 54, 21–41, doi:10.1007/s10874-005-9008-4, 2006

Kirk, J. T. O.: *Light and Photosynthesis in Aquatic Ecosystems*, 2nd ed., Cambridge Univ. Press, New York, 509 pp, ISBN 9780511623370, 1994

Kitsiou, D., Karydis, M.: Categorical mapping of marine eutrophication based on ecological indices, *Sci Total Environ.*, 255, 113–27, doi: 10.1016/S0048-9697(00)00457-5, 2000

Kuznetsova, M. and Lee, C.: Dissolved free and combined amino acids in nearshore seawater, sea surface microlayers and foams: influence of extracellular hydrolysis, *Aquat. Sci.*, 64, 252–268, <https://doi.org/10.1007/s00027-002-8070-0>, 2002

Field Code Changed

Kuznetsova, M., Lee, C., Aller, J., Frew, N.: Enrichment of amino acids in the sea surface microlayer at coastal and open ocean sites in the North Atlantic Ocean, *Limnol. Oceanogr.*, 49, 1605–1619, <https://doi.org/10.4319/lo.2004.49.5.1605>, 2004

Lawaetz, A.J., Stedmon, C.A.: Fluorescence intensity calibration using the Raman scatter peak of water, *Appl Spectrosc.*, 63, 936–940. doi: 10.1366/000370209788964548, 2009

Lechtenfeld, O. J., Kattner, G., Flerus, R., McCallister, S. L., Schmitt-Kopplin, P., Koch, B. P., Molecular transformation and degradation of refractory dissolved organic matter in the Atlantic and Southern Ocean, *Geochim. Cosmochim. Acta*, 126, 321–337. <https://doi.org/10.1016/j.gca.2013.11.009>, 2014

Lim, L., Wurl, O., Karuppiyah, S., and Obbard, J. P.: Atmospheric wet deposition of PAHs to the sea surface microlayer, *Mar. Pollut. Bull.*, 54(8), 1212–1219, doi:10.1016/j.marpolbul.2007.03.023, 2007

Lin, P., Bluvshstein, N., Rudich, Y., Nizkorodov, S. A., Laskin, J., and Laskin, A.: Molecular Chemistry of Atmospheric Brown Carbon Inferred from a Nationwide Biomass Burning Event, *Environ. Sci. Technol.*, 51, 11561–11570, <https://doi.org/10.1021/acs.est.7b02276>, 2017.

Liss, P.S. and Duce, R.A. (Eds.): The sea surface and global change. Cambridge University Press, 519 pp., ISBN 9780511525025, <https://doi.org/10.1017/CBO9780511525025>, 1997

Logozzo, L., Tzortziou, M., Neale, P., Clark, B. J.: Photochemical and microbial degradation of chromophoric dissolved organic matter exported from tidal marshes, *J. Geophys. Res.: Biogeosciences*, 126, <https://doi.org/10.1029/2020JG005744>, 2021

Martínez-Pérez, A. M., Catalá, T. S., Nieto-Cid, M., Otero, J., Álvarez, M., Emelianov, M., Recche, I., Álvarez-Salgado, X. A., and Aristegui, J.: Dissolved organic matter (DOM) in the open Mediterranean Sea. II: Basin-wide distribution and drivers of fluorescent DOM, *Prog. Oceanogr.*, 170, 93–106, doi:10.1016/j.pocean.2018.10.019, 2019

McKnight, D. M., Boyer, E.W., Westerhoff, P.K., Doran, P.T., Thomas-Kulbe, T., Andersen, D.T.: Spectrofluorometric characterization of dissolved organic matter for indication of precursor organic material and aromaticity, *Limnol. Oceanogr.*, 46, 38–48, <https://doi.org/10.4319/lo.2001.46.1.0038>, 2001

Milinković, A., Penezić, A., Kušan, A.C., Glušević, V., Žužul, S., Skejić, S., Šantić, D., Godec, R., Pehnee, G., Omanović, D., 712 Engel, A., Frka, S.: Variabilities of biochemical properties of the sea surface microlayer: Insights to the atmospheric deposition impacts, *Sci. Total Environ.*, 838, 156440, <https://doi.org/10.1016/j.scitotenv.2022.156440>, 2022

Miyazaki, Y., Yamashita, Y., Kawana, K., Tachibana, E., Kagami, S., Mochida, M., Suzuki, K., Nishioka, J.: Chemical transfer of dissolved organic matter from surface seawater to sea-spray

Field Code Changed

water-soluble organic aerosol in the marine atmosphere, *Sci. Rep.*, 8, 14861, doi:10.1038/s41598-018-32864-7, 2018

Mladenov, N., Sommaruga, R., Morales-Baquero, R., Laurion, I., Camarero, L., Diéguez, M.C., Camacho, A., Delgado, A., Torres, O., Chen, Z., Felip, M., and Reece, I.: Dust inputs and bacteria influence dissolved organic matter in clear alpine lakes, *Nat. Commun.*, 2, 405, doi:10.1038/ncomms1411, 2011

Mopper, K., Zhou, X., Kieber, R. J., Kieber, D.J., Sikorski, R. J., Jones, R.D.: Photochemical degradation of dissolved organic carbon and its impact in the oceanic carbon cycle, *Nature*, 353, 60-62, <https://doi.org/10.1038/353060a0>, 1991

Moran, M.A., Sheldon, W.M., Zepp, R.G.: Carbon loss and optical property changes during long term photochemical and biological degradation of estuarine dissolved organic matter, *Limnol. Oceanogr.*, 45, 1254-1264, <https://doi.org/10.4319/lo.2000.45.6.1254>, 2000

Muller, C. L., Baker, A., Hutchinson, R., Fairchild, I. J., Kidd, C.: Analysis of rainwater dissolved organic carbon compounds using fluorescence spectrophotometry, *Atmospheric Environment*, 42, 8036-8045, <https://doi.org/10.1016/j.atmosenv.2008.06.042>, 2008

Murphy, K., Stedmon, C.A., Graeber, D., Bro, R., Fluorescence spectroscopy and multi-way techniques. PARAFAC, *Anal. Methods*, 5, doi: 10.1039/c3ay41160e, 2013

Mustaffa, N.I.H., Ribas-Ribas M., Wurl O.: High-resolution variability of the enrichment of fluorescence dissolved organic carbon in the sea surface microlayer of an upwelling region, *Elementa Sci. Anthropol.*, 5:242, <https://doi.org/10.1525/elementa.242>, 2017

Mustaffa, N.I.H., Badewien, T.H., Ribas-Ribas, M., Wurl, O.: High-resolution observations on enrichment processes in the sea surface microlayer, *Sci. Rep.*, 8, 13122, <https://doi.org/10.1038/s41598-018-31465-8>, 2018

Mustaffa, N.I.H., Ribas-Ribas, M., Banko-Kubis, H.M., Wurl, O.: Global reduction of in-situ CO₂ transfer velocity by natural surfactants in the sea surface microlayer, *Proc. R. Soc. Math. Phys. Eng. Sci.*, 476, 20190763, <https://doi.org/10.1098/rspa.2019.0763>, 2020

Nelson, N.B., Siegel, D.A., Michaels, A.F.: Seasonal dynamics of colored dissolved material in the Sargasso Sea, *Deep Sea Res., Part I*, 45, 931-957, [https://doi.org/10.1016/S0967-0637\(97\)00106-4](https://doi.org/10.1016/S0967-0637(97)00106-4), 1998

Nelson, N.B. and Siegel, D.A.: The global distribution and dynamics of chromophoric dissolved organic matter, *Ann. Rev. Mar. Sci.*, 5, 447-476, doi: 10.1146/annurev-marine-120710-100751, 2013

Obernosterer, I., and Benner, R.: Competition between biological and photochemical processes in the mineralisation of dissolved organic carbon, *Limnol. Oceanogr.*, 49, 117-124, doi:10.4319/lo.2004.49.1.0117, 2004

Field Code Changed

Field Code Changed

Obernosterer, I., Catala, P., Lami, R., Caparros, J., Ras, J., Bricaud, A., Dupuy, C., van Wambeke, F.,
 Lebaron, P.: Biochemical characteristics and bacterial community structure of the sea surface microlayer
 in the South Pacific Ocean, *Biogeosciences*, 5, 693–705, <https://doi.org/10.5194/bg-5-693-2008>, 2008

Opsahl, S., and Benner, R.: Photochemical reactivity of dissolved lignin in river and ocean waters,
Limnol. Oceanogr., 43, 1297–1304, doi:10.4319/lo.1998.43.6.1297, 1998

Opsahl, S. P., and Zepp, R. G.: Photochemically induced alteration of stable carbon isotope ratios ($\delta^{13}\text{C}$)
 in terrigenous dissolved organic carbon, *Geophys. Res. Lett.*, 28, 2417–2420,
 doi:10.1029/2000GL012686, 2001

Ortega-Retuerta, E., Frazer, T.K., Duarte, C.M., Ruiz-Halpern, S., Tovar-Sanchez, A., Arrieta, J.M.,
 Reche, I.: Biogeneration of chromophoric dissolved organic matter by bacteria and krill in the Southern
 Ocean, *Limnol. Oceanogr.*, 54, 1941–1950, 2009

Osmund, H.H. and Riemann, B.: Chlorophyll a Determination: Improvements in Methodology, *Oikos*,
 30, 438–447, <https://doi.org/10.2307/3543338>, 1978

Penezić, A., Drozdowska, V., Novak, Gašparević, B.: Distribution and characterization of organic matter
 within the sea surface microlayer in the Gulf of Gdańsk, *Oceanologia*, 4, 631–650,
<https://doi.org/10.1016/j.oceano.2022.05.003>, 2022

Pitta, E., Zeri, C., Tzortziou, M., Mousdis, G., Scoullou, M.: Seasonal variations in dissolved organic
 matter composition using absorbance and fluorescence spectroscopy in the Dardanelles Straits—North
 Aegean Sea mixing zone, *Cont. Shelf Res.*, 149, 82–95, <https://doi.org/10.1016/j.csr.2016.07.013>, 2017

Pitta, E., & Zeri, C.: The impact of combining data-sets of fluorescence excitation-emission matrices of
 dissolved organic matter from various aquatic sources on the information retrieved by PARAFAC
 modelling, *Spectrochim. Acta A Mol. Biomol. Spectrosc.*, 258, 119800, 2021.

Pereira, R., Ashton, I., Sabbaghzadeh, B., Shutler, J.D., Upstill-Goddard, R.C.: Reduced air-sea CO₂
 exchange in the Atlantic Ocean due to biological surfactants, *Nat. Geosci.*, 11, 492–496,
<https://doi.org/10.1038/s41561-018-0136-2>, 2018

Ploug, H.: Cyanobacterial surface blooms formed by *Aphanizomenon* sp. and *Nodularia spumigena* in
 the Baltic Sea: small-scale fluxes, pH, and oxygen microenvironments, *Limnol. Oceanogr.*, 53, 914–921,
<https://doi.org/10.4319/lo.2008.53.3.0914>, 2008

Retelletti Brogi, S., Charrière, B., Gonnelli, M., Vaultier, F., Sempéré, R., Vestri, S., Santinelli, C.: Effect
 of UV and Visible Radiation on Optical Properties of Chromophoric Dissolved Organic Matter Released
 by *Emiliania huxleyi*, *J. Mar. Sci. Eng.*, 8(11), 888, doi:10.3390/jmse8110888, 2020

Ribas Ribas, M., Mustaffa, N.I.H., Rahlff, J., Stolle, C. and Wurl, O.: Sea Surface Scanner (S3): A
 catamaran for high-resolution measurements of biochemical properties of the sea surface microlayer, *J.*
Atmos. Oceanic Technol., 34, 1433–1448, <https://doi.org/10.1175/JTECH-D-17-0017.1>, 2017

Field Code Changed

Field Code Changed

Romera-Castillo, C., Nieto-Cis, M., Castro, C.G., Marrase, C., Largier, J., Barton, E.D., Alvarez-Salgado, X.A.: Fluorescence: absorption coefficient ratio — tracing photochemical and microbial degradation processes affecting coloured dissolved organic matter in a coastal system, *Mar. Chem.*, 125, 26–38, <https://doi.org/10.1016/j.marchem.2011.02.001>, 2011.

Sabbaghzadeh, B., Upstill-Goddard, R.C., Beale, R., Pereira, R., Nightingale, P.D.: The Atlantic Ocean surface microlayer from 50°N to 50°S is ubiquitously enriched in surfactants at wind speed up to 13 m s⁻¹, *Geophys. Res. Lett.*, 44, 2852–2858, <https://doi.org/10.1002/2017GL072988>, 2017

Sabbaghzadeh, B., Uher, G., Upstill-Goddard, R.: Dynamics of chromophoric dissolved organic matter in the Atlantic Ocean: unravelling province-dependent relationships, optical complexity, and environmental influences, *Front. Mar. Sci.*, 11, <https://doi.org/10.3389/fmars.2024.1432133>, 2024

Santos, P. S. M., Santos, E. B. H., Duarte, A. C.: First spectroscopic study on the structural features of dissolved organic matter isolated from rainwater in different seasons, *Sci. Total Environ.*, 426, 172–179, [doi:10.1016/j.scitotenv.2012.03.023](https://doi.org/10.1016/j.scitotenv.2012.03.023), 2012

Siegel, D. A., and Michaels, A.F.: Quantification of non-algal light attenuation in the Sargasso Sea: Implications for biogeochemistry and remote sensing, *Deep Sea Res., Part II*, 43, 321–345, [https://doi.org/10.1016/0967-0645\(96\)00088-4](https://doi.org/10.1016/0967-0645(96)00088-4), 1996

Stedmon C.A. and Markager, S.: Tracing the production and degradation of autochthonous fractions of dissolved organic matter by fluorescence analysis, *Limnol. Oceanogr.*, 50, 1415–1426, <https://doi.org/10.4319/lo.2005.50.5.1415>, 2005

Stolle, C., Nagel, K., Labrenz, M., Jurgens, K.: Succession of the sea-surface microlayer in the coastal Baltic Sea under natural and experimentally induced low wind conditions, *Biogeosciences*, 7, 2975–2988, <https://doi.org/10.5194/bg-7-2975-2010>, 2010

Tassan, S.: The effect of dissolved yellow substance on the quantitative retrieval of chlorophyll and total suspended sediment concentration from remote measurements of water colour, *Int. J. Remote Sens.*, 9, 787–797, <https://doi.org/10.1080/01431168808954893>, 1988

Tilstone, G.H., Ains, R.L., Vicente, V.M., Widdicombe, C., Llewellyn, C.: High concentrations of mycosporine like amino acids and colored dissolved organic matter in the sea surface microlayer off the Iberian Peninsula, *Limnol. Oceanogr.*, 55, 1835–1850. [doi: 10.4319/lo.2010.55.5.1835](https://doi.org/10.4319/lo.2010.55.5.1835), 2010

Twardowski, M.S. and Donaghay, P.L.: Separating in situ and terrigenous sources of absorption by dissolved materials in coastal waters, *J. Geophys. Res.*, 106, 2545–2560, <https://doi.org/10.1029/1999JC000039>, 2001

Tzempelikou, E., Zeri, C., Pitta, E., Assimakopoulou, G., Pavlidou, A., Rousselaki, E.: Temporal dynamics of transparent exopolymer particles in surface waters and sea surface microlayer in the coastal Eastern Mediterranean Sea, *Reg. Stud. Mar. Sci.*, 85, <https://doi.org/10.1016/j.rsma.2025.104163>, 2025

Field Code Changed

Field Code Changed

Tzortziou, M., Neale, P.J., Megonigal, J.P., Pow, C.L., Butterworth, M.: Spatial gradients in dissolved carbon due to tidal marsh outwelling into a Chesapeake Bay estuary, *Mar. Ecol. Prog. Ser.*, 426, 41–56, doi: 10.3354/meps09017, 2011

Upstill-Goddard, R.C.: Air–sea gas exchange in the coastal zone, *Estuar. Coast. Shelf Sci.*, 70, 388–404, <https://doi.org/10.1016/j.ecss.2006.05.043>, 2006

van Pinxteren, M., Mav E., Herrmann, H., Herrmann, H.: Chemical characterization of dissolved organic compounds from coastal sea surface microlayers (Baltic Sea, Germany), *Environ. Sci. Technol.*, 46, 10455–10462, doi:10.1021/es204492b, 2012.

Weishaar, J.L., Aiken, G.R., Bergamaschi, B.A., Fram, M.S., Fujii, R., Mopper, K.: Evaluation of Specific Ultraviolet Absorbance as an Indicator of the Chemical Composition and Reactivity of Dissolved Organic Carbon, *Environ. Sci. Technol.*, 37, 4702–4708, <https://doi.org/10.1021/es030360x>, 2003

Willey, J.D., Kieber, R.J., Eyman, M.S., Avery, G.B.Jr.: Rainwater dissolved organic carbon: concentrations and global flux, *Glob. Biogeochem. Cycles*, 14(1), 139–148, doi:10.1029/1999GB900036, 2000

Wu, G., Fu, P., Ram, K., Song, J., Chen, Q., Kawamura, K., Wan, X., Kang, S., Wang, X., Laskin, A., Cong, Z.: Fluorescence characteristics of water soluble organic carbon in atmospheric aerosol, *Environ. Pollut.*, 268, 115906, doi:10.1016/j.envpol.2020.115906, 2021

Wurl, O. and Obbard, J.P.: A review of pollutants in the sea surface microlayer (SML): a unique habitat for marine organisms, *Mar. Pollut. Bull.*, 48, 1016–1030, <https://doi.org/10.1016/j.marpolbul.2004.03.016>, 2004

Wurl, O. and Holmes, M.: The gelatinous nature of the sea surface microlayer, *Mar. Chem.*, 110, 89–97, <https://doi.org/10.1016/j.marchem.2008.02.009>, 2008

Wurl, O., Miller, L., Röttgers, R., Vagle, S.: The distribution and fate of surface active substances in the sea surface microlayer and water column, *Mar. Chem.*, 115, 1–9, <https://doi.org/10.1016/j.marchem.2009.04.007>, 2009

Wurl, O., Stolle, C., Van Thuc, C., The Thu, P., Mari, X.: Biofilm-like properties of the sea surface and predicted effects on air-sea CO₂ exchange, *Prog. Oceanogr.* doi: 10.1016/j.pocean.2016.03.002, 2016

Yang, L., Chen, W., Zhuang, W. E., Cheng, Q., Li, W., Wang, H., Guo, W., Chen-Tung Arthur Chen, Liu, M.: Characterization and bioavailability of rainwater dissolved organic matter at the southeast coast of China using absorption spectroscopy and fluorescence EEM-PARAFAC, *Mar. Chem.*, 210, 11–22, doi:10.1016/j.marchem.2018.10.021, 2019

Field Code Changed

Field Code Changed

Yang, L., Zhang, J., Engel, A., Yang, G.-P.: Spatio-temporal distribution, photoreactivity and environmental control of dissolved organic matter in the sea surface microlayer of the eastern marginal seas of China, *Biogeosciences*, 19, 5251–5268, doi:10.5194/bg-19-5251-2022, 2022

Zäncker, B., Bracher, A., Röttgers, R., Engel, A.: Variations of the Organic Matter Composition in the Sea Surface Microlayer: A Comparison between Open Ocean, Coastal, and Upwelling Sites Off the Peruvian Coast, *Front. Microbiol.*, 8:2369, doi: 10.3389/fmicb.2017.02369, 2017

Zepp, R.G., Callaghan, T.V., Erickson, D.J.: Effects of enhanced solar ultraviolet radiation on biogeochemical cycles, *J. Photochem. Photobiol., B Biol.*, 46, 69–82, [https://doi.org/10.1016/S1011-1344\(98\)00186-9](https://doi.org/10.1016/S1011-1344(98)00186-9), 1998

Zervoudaki, S., Siokou, I., Krasakopoulou, E., Kontoyiannis, H., Pavlidou, A., Assimakopoulou, G., Katsiaras, N., Reizopoulou, S., Karageorgis, A.P., Kaberi, H., Lardi, P.I., Gerakaris, V., Tsiamis, K., Salomidi, M., Zeri, C., Pitta, E., Strogyloudi, E., Parinos, C., Hatzianestis, I., Christou, E.D., Zoulas, T., Sakavara, A., Pagou, K., Zenetos, A., Panayotidis, P., Simbora, N.: Biogeochemical Characteristics in the Saronikos Gulf (Aegean Sea, Eastern Mediterranean), in: *The Aegean Sea Environment: Anthropogenic Presence and Impact*, edited by Anagnostou, C.L., Kostianoy, A.G., Mariolakis, I.D., Panayotidis, P., Soilemezidou, M., Tsaltas, G., *Hdb Environ. Chem.*, Springer, 142, 511–568, https://doi.org/10.1007/698_2022_898, 2022

Zhang, Z., Cai, W., Liu, L., Liu, C., Chen, F.: Direct determination of thickness of sea surface microlayer using a pH microelectrode at original location, *Sci. China Ser. B Chem.*, 46, 339–351, <https://doi.org/10.1360/02yb0192>, 2003

Zhang, X. Y., Chen, X., Deng, H., Du, Y., Jin, H. Y.: Absorption features of chromophoric dissolved organic matter (CDOM) and tracing implication for dissolved organic carbon (DOC) in Changjiang Estuary, China, *Biogeosciences Discuss.*, 10, 12217–12250, <https://doi.org/10.5194/bgd-10-12217-2013>, 2013

Zhang, Y., Liu, X., Osburn, C.L., Wang, M., Qin, B., Zhou, Y.: Photobleaching response of different sources of chromophoric dissolved organic matter exposed to natural solar radiation using absorption and excitation-emission matrix spectra, *PLoS One*, 8(10):e77515, doi: 10.1371/journal.pone.0077515, 2013

Field Code Changed

