



## Litter biomass as a driver of soil VOC fluxes in a Mediterranean forest

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**Abstract.** Soil biogenic volatile organic compound (BVOC) emissions have been studied in different biomes, showing that their emissions are not negligible. However, previous studies have so far neglected the role of litter accumulation on soil BVOC fluxes, and most of them refer to coniferous and evergreen forests, while litter emissions from Mediterranean deciduous forests remain poorly explored. To fill these gaps, the present work aimed to study BVOC fluxes in a Mediterranean deciduous forest, with a particular attention to the relationship between soil BVOC fluxes and litter biomass accumulation on soils. Measurements were performed in southern France, in the downy oak (*Quercus pubescens* Willd.) forest of the Observatoire de Haute Provence (O3HP), during the late spring of 2023, using dynamic chambers coupled to an on-line PTR-ToF-MS. We investigated in-situ daily BVOC fluxes from soil alone and different litter biomasses mimicking low, current, high, and very high litter production, respectively, as both decreases and increases of litter accumulation are expected in the Mediterranean region under the current context of climate change and greening management policies. The results showed a high BVOC diversity with more than 135 emitted compounds. For a large majority of the measured compounds, the BVOC fluxes were negative, suggesting that soil (bare soil covered by litter) uptakes compounds through biochemical and/or physical processes. Some compounds, such as acetone, methanol or sesquiterpenes, increased with increasing litter biomass, suggesting the importance of considering litter accumulation when assessing soil BVOC emissions from Mediterranean deciduous forests. Microbial abundance was highlighted as a potential driver of this relation between litter biomass and VOC fluxes.

### 1- Introduction

25 Biogenic volatile organic compounds (BVOCs) are key components of the atmosphere's oxidative capacity through their influence on the OH radical, O<sub>3</sub> and NO<sub>x</sub> budgets, among others. They also play a critical role in the formation of the secondary organic aerosol (SOA) (Kulmala et al., 2004; Hallquist et al., 2009; Seinfeld and Pandis, 2016; Mahilang et al., 2021), contributing to health and climate impacts (Seinfeld and Pandis, 2016; Thornhill et al., 2021). Numerous studies have documented BVOC emissions from the aerial parts of terrestrial plants - at the canopy, branch and leaf scales (Artaxo et al., 2022; Gros et al., 2022; Mu et al., 2022; Rinnan, 2024), and they are estimated to release between 300 and 1000 Tg (C) yr<sup>-1</sup> on a global scale (Guenther et al., 2012; Sindelarova et al., 2014; Wang et al., 2024). However, these estimates are likely to be underestimated as they neglect the contribution of soil (bare soil covered by litter) due to the scarcity of studies. The recent interest on soil BVOCs shows bi-directional VOC fluxes with positive net fluxes from soils covered by litter (Peñuelas et al., 2014; Bourtsoukidis et al., 2018; Viros et al., 2020; Yang et al., 2024a; Legros et al., 2025). Soil could thus be a significant source of BVOCs affecting atmospheric chemistry and related climate impacts (Kramshøj et al., 2019; Yang et al., 2024b).

30 Soil-atmosphere exchanges have been measured in different regions of the world, such as tropical (Jardine et al., 2015, 2017; Bourtsoukidis et al., 2018; Artaxo et al., 2022), boreal (Aaltonen et al., 2013; Mäki et al., 2019; Artaxo et al., 2022), temperate (Leff and Fierer, 2008; Svendsen et al., 2018; Mäki, 2019; Isidorov and Zaitsev, 2020; Isidorov et al., 2024), and Mediterranean forests (Asensio et al., 2008; Viros et al., 2020; Rezaie et al., 2023; Yang et al., 2024a). Recent studies have reviewed most VOCs emitted from soil compartments (roots, bare soil, soil covered by litter or litter alone) (Tang et al., 2019; Yang et al., 2024b). Most reported VOCs are isoprene, monoterpenes, sesquiterpenes, and oxygenated VOCs such as methanol (Gray et al., 2010). Methanol is both emitted and consumed by soil bacteria and rhizosphere microorganisms (Asensio et al., 2007a). Other studies have reported important VOC deposition fluxes of hexanal or acetone (Peñuelas et al., (2014), and references therein).

45 Abiotic (soil moisture and temperature) and biotic (e.g. microbial community composition and biomass, soil and litter nutrient availability) processes can contribute to VOC degradation (Cleveland and Yavitt, 1998; Rinnan and Albers, 2020; Trowbridge et al., 2020). Processes and sources related to soil VOC emissions are reviewed in (Tang et al., 2019; Isidorov and Zaitsev, 2022) and highlight the importance of both biotic and abiotic parameters, as



55 evaluated in several studies (Wilkins, 1996; Mackie and Wheatley, 1999; Asensio et al., 2007a; Leff and Fierer, 2008; Abis et al., 2020; Jiao et al., 2023a). BVOC emissions are known to increase with microbial activity, through microbial decomposition of litter or soil organic carbon, and evaporation of litter-stored VOCs (Stahl and Parkin, 1996; Leff and Fierer, 2008; Insam and Seewald, 2010; Aaltonen et al., 2013). Plant roots are also known to be a source of VOCs, acting as mediators of belowground interactions between plants and soil microorganisms (Asensio et al., 2007a; Wenke et al., 2010; Yang et al., 2024b).

60 Soil also acts as a sink for BVOCs, as they can be assimilated through microbial metabolism as a source of carbon and energy for certain heterotrophic microbes (Shennan, 2006; Kramshøj et al., 2018; McGenity et al., 2018; Zhang et al., 2020; Jiao et al., 2023b). BVOCs can also diffuse through soil pores, where they can be adsorbed onto soil particles or dissolved in soil water until equilibrium is reached (Ruiz et al., 1998; Ahn et al., 2020). Although these uptake processes are less well understood, VOC uptake is considered to be a widespread process in soil (Rinnan and Albers, 2020; Jiao et al., 2023b) as well as in the canopy (Niinemets et al., 2014). Among the different drivers of soil VOC fluxes, temperature and drought are important abiotic factors (Asensio et al., 2007a; Trowbridge et al., 2020; Rezaie et al., 2023; Legros et al., 2025), suggesting that these fluxes will be affected by climate change.

70 Indeed, Mediterranean forests are strongly affected by increasing warming and aridity (Peñuelas, 2008), which limit soil functioning (e.g. litter decomposition, microbial activity, Santonja et al., 2017; Quer et al., 2022) and thus BVOC exchanges between soil and atmosphere (Peñuelas et al., 2017; Yang et al., 2024a). While bare soil appears to be a sink for VOCs in Mediterranean ecosystems and other biomes (Asensio et al. 2007), the soil surface covered by litter and litter alone have been highlighted as a source of BVOCs, with negligible to moderate fluxes compared to leaf emissions (Peñuelas et al., 2014; Viros et al., 2020; Legros et al., 2025). However, the influence of litter biomass on soil VOC fluxes in Mediterranean forests has never been investigated. On the one hand, increasing litter accumulation is expected in this region, as intensive drought reduces litter mixture interactions and decomposition, leading to litter accumulation on the soil (Santonja et al., 2015, 2017). Moreover, greening policies and gradual abandonment (around 1855-1870) of wood in shipbuilding have contributed to an increase in forest area, which has doubled since the mid-19th century. On the other hand, chronic limited precipitation expected in the coming decades in the Mediterranean region will lead to lower leaf production and thus a lower litter fall in these ecosystems.

80 In this context, the present study aimed to investigate *in situ* BVOC emissions from soils in a Mediterranean deciduous forest dominated by the downy oak (*Quercus pubescens* Willd.), the major deciduous forest in the Southern Mediterranean part of France. We focussed on the relationships between litter BVOC fluxes and litter biomass accumulation on the soil surface and associated microbial communities.

## 2- Material and methods

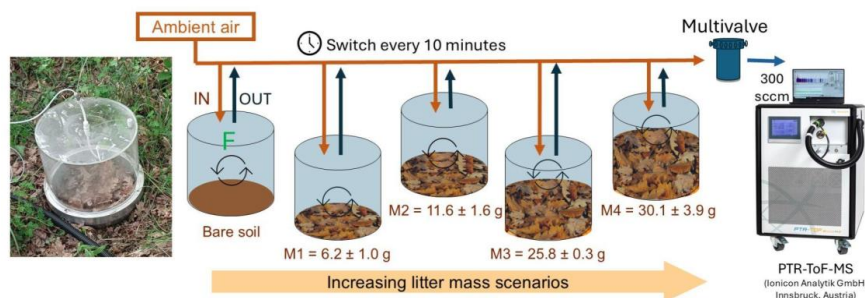
### 2-1 Experimental site and sampling strategy

85 Litter BVOC measurements were performed at the O<sub>3</sub>HP experimental site, an AnaEE *in situ* platform located in the research center Observatoire de Haute Provence (5°42 E, 43°55 N, 650 m a.s.l., Saint-Michel l'Observatoire, France). The climate is typical of the Mediterranean region, characterized by a dry and hot summer. The O<sub>3</sub>HP site was created in 2009 in order to study the *Q. pubescens* forest ecosystem ( $\approx 90$  % of the biomass and  $\approx 75$  % of the trees) at the soil and tree scales, under both natural and accentuated water stress conditions induced by a rainfall exclusion device (an automated monitored roof deployed during rain events) set up over a part of the O<sub>3</sub>HP canopy. Litter production of *Q. pubescens* within the site ranges between 1.4 and 1.6 t ha<sup>-1</sup> yr<sup>-1</sup>, that is 166 g.m<sup>-2</sup> yr<sup>-1</sup> (Genard-Zielinski et al., 2015; Viros et al., 2020). A dense network of sensors in the soil, under and above the canopy, continuously recorded the climatic and edaphic parameters (air and soil temperatures and relative humidity, photosynthetically active radiation or PAR). More details can be found in (Garnier et al., 2021).

95 The experiment was conducted in spring, from 23<sup>rd</sup> to 26<sup>th</sup> May 2023. Five dynamic soil chambers (14 L volume) consisting of polycarbonate cylinders with a stainless-steel basement were used (for a full description see Legros et al., 2025). Each chamber covered 0.067 m<sup>2</sup> of soil surface. Chamber 1 covered bare soil, while chambers 2 to 5 circled bare soil fully covered by 8 g, 16 g, 32 g and 48 g of fresh mass of *Q. pubescens* litter, respectively, corresponding to litter dry mass ranging from 6.2 to 30.01 g (Table S1, Figure 1). These added litter masses are hereafter referred to as LM1, LM2, LM3 and LM4 in the text. All chambers were flushed using ambient air at 0.6 L min<sup>-1</sup>, and an inlet to measure ambient air was placed at 50 cm above ground level, at a central position from the



5 chambers. Emissions from each chamber were monitored over a 24 h period and repeated three times using renewed fresh litter each day while the bare soil remained unchanged.



105 Figure 1. Experimental setup for bare soil and soil covered by different litter accumulations.

## 2-2 PTR-ToF-MS parameters, data treatment and flux calculation

BVOCs were monitored by a PTR-ToF-MS (Proton Transfer Reaction – Time of Life – Mass Spectrometer 600X2, Ionicon Analytik GmbH, Innsbruck, Austria). Each chamber and the ambient inlet were connected to a flow through multivalve (VICI VALCO) placed inside the PTR-ToF-MS through ~20 m long 1/8" teflon lines, with a flow around 0.3 L min<sup>-1</sup> in each line. Parameters of PTR-ToF-MS were set up as the reaction chamber pressure was fixed at 2.6 ± 0.001 mbar, drift tube voltage at 450 V and temperature at 120 °C, corresponding to an E/N ratio (electric field strength over buffer gas number density) of ≈125 Td (1 Td = 10–17 V cm<sup>-2</sup>). Each sample (bare soil, LM1 to LM4 and ambient air) was sequentially monitored every 10 min during 24 h, leading to a one-hour cycle. Measurements were repeated for three days using a new pool of litter each day within each chamber (i.e. day 1, 2 and 3, chamber one encircled a soil surface containing each day a different litter pool of 8 g of fresh litter biomass). BVOC concentrations on a large range of mass to charge (m/z) up to m/z 500.

PTR-ToF-MS data were post-processed with IDA software (Ionicon Analytik GmbH, Innsbruck, Austria). First, mass calibration was performed based on H<sub>3</sub>O<sup>+</sup> isotope (H<sub>2</sub><sup>18</sup>O.H<sup>+</sup>, m/z 21.022), H<sub>3</sub>O<sup>+</sup>.H<sub>2</sub>O isotope (H<sub>2</sub>O.H<sub>2</sub><sup>18</sup>O.H<sup>+</sup>, m/z 39.033), diiodobenzene (an internal calibrant) parent ion (C<sub>6</sub>H<sub>4</sub>I<sub>2</sub>H<sup>+</sup>, m/z 330.848) and its main main fragment (C<sub>6</sub>H<sub>4</sub>I.H<sup>+</sup>, m/z 203.943). Then peak fitting and high-resolution integration were performed. Peak identification was done based on peak position and the most probable combination of C, H, N, and O atoms. When possible, molecular formulas were assigned to organic compounds based on previous references on BVOC emissions (Yáñez-Serrano et al., 2021). BVOC mixing ratios were finally calculated using the proton transfer theory where the rate constant *k* (cm<sup>3</sup> s<sup>-1</sup>) was determined for each molecular formula based on the method proposed by Cappellin et al. (2012). The relative ion transmission efficiency was calculated using a standard gas calibration mixture containing 14 different VOCs at 100 ± 10 ppb in nitrogen (TO-14A Aromatic Mix, Restek Corporation, Bellefonte, USA).

BVOC fluxes (*F*) from dynamic chamber were calculated as described in (Yang et al., 2024b):

$$F = \frac{(C_{chamber} - C_{ambient}) \times Q}{A} [\mu\text{g m}^{-2} \text{h}^{-1}] \quad (1)$$

130 Where *C*<sub>chamber</sub> and *C*<sub>ambient</sub> are BVOC concentrations in ng/L at the chamber outlet and inlet respectively, *Q* is the flow rate inside the chamber (= 0.3 L min<sup>-1</sup>) and *A* the soil surface (0.0616 m<sup>2</sup>) circled by the chamber.

## 2-3 Phospholipid fatty acid analysis (PLFA) measurements

After the end of BVOC measurements, the soil and litter samples were collected, frozen, lyophilized during 72 h and then ground into powder prior to microbial analyses. The phospholipid fatty acids (PLFAs) are essential components of all living cells and are used as biomarkers of soil microbial communities (Frostegård and Bååth 1996). The PLFA extraction, identification and quantification followed the protocol described by Aupic-Samain et al. (2021) and Biryol et al. (2024). Litter lipid contents were extracted with Bligh-Dyer solution containing a quantification standard (C19:0; MIDI, Inc., Newark, DE, USA). The lipids were then separated using a 96-well solid-phase extraction (SPE) plate (SilactSPE Silica Affinisep®). To identify and quantify the lipids present, a gas

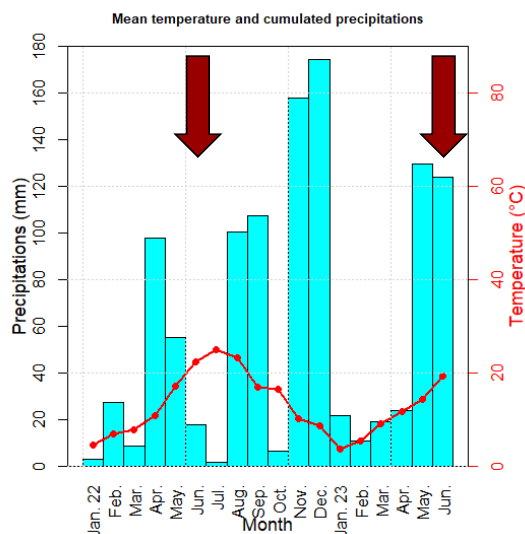


140 chromatography mass spectrometer (a GC-MS Agilent 7890 system with a MSD5977A Network mass detector,  
an ALS7693 automatic injector, and HP5-MS apolar column) was employed with Mass-Hunter and Sherlock  
software (MIDI, Inc., Newark, DE, USA). Among the PLFAs identified in the samples, 10 were analyzed because  
they are markers Gram-positive bacteria, Gram-negative bacteria, Actinobacteria, saprotrophic fungi and  
145 arbuscular mycorrhizal (AM) fungi (Frostegård and Bååth 1996; Biryol et al. 2024). Microbial biomasses were  
obtained by converting the peak areas into  $\mu\text{g/g}$  of litter or soil.

#### 2-4 Measurements of environmental variables

Ambient meteorological parameters (temperature, relative humidity and precipitation, wind speed and direction,  
rain, solar radiation) were recorded using a Sentinel weather station. Temperature and precipitation within the plot  
during one year prior to this study are shown in the ombrothermic diagram in Figure 2 (for more details on climate  
150 at O<sub>3</sub>HP, referring to Garnier et al., 2021). It illustrates that 2022 was very dry, with 5 months of drought (January,  
March, June, July and October 2022), while May 2023 (measurement period for this study) was rainy with an  
accumulated precipitation of 120 mm.

Temperature and humidity inside the chambers were continuously monitored by i-buttons (Whitewater, WI 53190  
USA). Litter humidity was estimated from measurement of fresh and dry litter using 3 aliquots of 8 g each day  
155 (Table S.1). Litter aliquots were dried using oven-drying at 65 °C for 3 days. Litter humidity calculated as  
 $\frac{\text{dry mass (g)} \times 100}{\text{fresh mass (g)}}$ , ranged from 14 % to 41 % (Table S.1).



160 **Figure 2.** Ombrothermic diagram from January 2022 to June 2023. Red arrows refer to the field campaign in June 2022 (Legros et al., 2025) performed within the natural and accentuated water stress conditions plots and May 2023 (this study).

#### 2-5 Statistical analysis

Data were analyzed, statistically computed and plotted using RStudio. A holistic approach was used to investigate  
emissions, meaning that all VOCs detected by PTR-ToF-MS were considered in the data analysis steps described  
as follows. First, a *Welsh* t-test was used to select only ions showing significant differences between ambient and  
165 chamber measurements. Dataset was filtered to keep ions with  $p_{\text{value}} > 0.001$ . BVOC targeted in this study are  
summarized in Table S.2. Statistics were carried out using filtered VOC data, that is, the most abundant VOC  
fluxes in each chamber with fluxes  $> 0.002 \mu\text{g m}^{-2} \text{h}^{-1}$  (in absolute value).

To investigate the link between these parameters with main measured VOC fluxes, a correlation matrix has been  
built based on Pearson correlations by merging all chamber data. A heatmap correlation plot was classified into 5



170 clusters of VOC fluxes, environmental and biological parameters. The optimal number of clusters was estimated using a k-means clustering method technique (Steinley, 2006).

### 3- Results and discussion

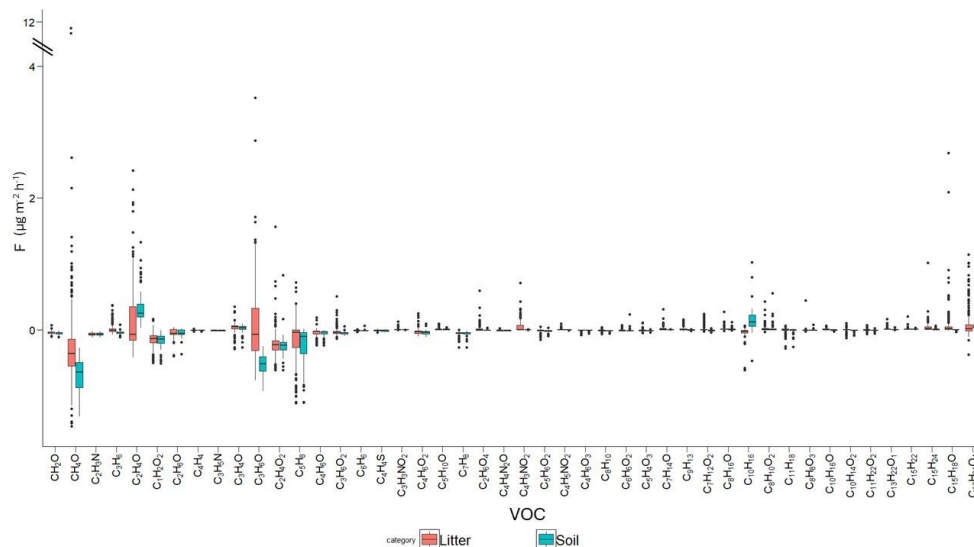
#### 3-1 Diversity of VOC fluxes

175 Considering all treatments, more than 135 compounds were identified, which is much more than in previous studies focusing VOC fluxes from litter soil (Asensio et al., 2007b; Viros et al., 2020; Yang et al., 2024a; Legros et al., 2025), reflecting highly diverse VOC fluxes from *Q. pubescens* litter. This finding might be the consequence of contrasting conditions compared to other studies. Furthermore, the holistic approach used in this study tends to include a higher diversity of ions. Altogether, these compounds represent a total negative flux (or immision) of -0.85  $\mu\text{g m}^{-2} \text{h}^{-1}$ . Considering the most abundant VOC fluxes (compounds with absolute value of average fluxes above 0.01  $\mu\text{g m}^{-2} \text{h}^{-1}$ , a total of 46 compounds) in each chamber, the average total flux of VOC reached -0.72  $\mu\text{g m}^{-2} \text{h}^{-1}$ , representing 90 % of the total flux (Figure 3, Table S.2).

180 In comparison, Legros et al. (2025) reported emissions of  $9.11 \pm 0.64 \mu\text{g m}^{-2} \text{h}^{-1}$  when considering total VOC fluxes. (Yang et al., 2024a) showed that fluxes are dependent on seasonality. Indeed, they found negative fluxes in summer and winter ( $-25.9 \pm 9.36$  and  $-2.77 \pm 3.84 \mu\text{g m}^{-2} \text{h}^{-1}$ , respectively) and positive fluxes in autumn and spring ( $4.54 \pm 7.18$  and  $3.86 \pm 2.59 \mu\text{g m}^{-2} \text{h}^{-1}$ , respectively).

185 Regarding individual compounds, positive and negative fluxes occurred from bare soil and soil covered by litter, as often reported earlier (Yang et al., 2024a; Legros et al., 2025). Figure 3 presents the positive and negative net fluxes for major compounds, while results for all VOCs are reported in Supplementary data. Almost 48 % of the compounds showed negative fluxes, meaning they were immitted by the soil and soil covered by litter through assimilation or adsorption processes, 37 % were both emitted and immitted and 15 % were only emitted (Supplementary data).

195 Soil covered by litter and bare soil mainly differed in terms of methanol ( $\text{CH}_4\text{O}$ ), acetone ( $\text{C}_3\text{H}_6\text{O}$ ) and isoprene ( $\text{C}_5\text{H}_8$ ) fluxes, which were bidirectional (emitted and immitted) in presence of litter, whereas they were only emitted or immitted by bare soil (Figure 3). For monoterpenes ( $\text{C}_{10}\text{H}_{16}$ ) and acetaldehyde ( $\text{C}_2\text{H}_4\text{O}$ ) fluxes, higher net positive emissions were observed for bare soil than soil covered by litter. This suggests that soil can be a source of monoterpenes and acetaldehyde to the atmosphere. The presence of litter on the soil physically reduces the transfer to the atmosphere, or favours the presence of microorganisms that consumes these two VOCs (Marmulla and Harder, 2014; McBride et al., 2023).





200 **Figure 3.** Most abundant compounds and their measured VOC fluxes (in  $\mu\text{g m}^{-2} \text{h}^{-1}$ ) during the field campaign. The line that divides the box represents the median of measured fluxes, ends of the box indicate the upper (Q3) and lower (Q1) quartiles, up and down whiskers represent max and min values and outliers are represented by dots. VOC fluxes for this figure were obtained from the 24-h measurements of all replicates for both soil and the four LM in the case of litter.

### 205 *VOC emissions*

VOCs are only considered as emitted in this section, when they have positive average VOC fluxes for each modality (bare soil and soil covered by litter), and conversely for immission (next section). The compounds with the highest emissions were acetaldehyde ( $\text{C}_2\text{H}_4\text{O}$ ), acrolein ( $\text{C}_3\text{H}_4\text{O}$ ), 1H-pyrrole-2,5-dione ( $\text{C}_4\text{H}_3\text{NO}_2$ ), lindestrene ( $\text{C}_{15}\text{H}_{18}\text{O}$ ), creosol ( $\text{C}_8\text{H}_{10}\text{O}_2$ ), sesquiterpenes ( $\text{C}_{15}\text{H}_{24}$ ), octanal ( $\text{C}_8\text{H}_{16}\text{O}$ ) and dibutyl phthalate ( $\text{C}_{16}\text{H}_{20}\text{O}_4$ ).

210 On average, we measured positive acetaldehyde fluxes of  $0.056 \pm 0.192$  and  $0.186 \pm 0.226 \mu\text{g m}^{-2} \text{h}^{-1}$  from soil covered by litter and bare soil, respectively. In both cases, the fluxes occurred with a high variability, resulting in positive and negative fluxes with respect to the diurnal cycle (sections 3.3 and 3.4). Emissions of acetaldehyde have been previously measured during litter emissions experiments (Asensio et al., 2007a; Gray et al., 2010, 2014).  
215 The origin of acetaldehyde can be attributed to root and leaf litter emissions (Warneke et al., 1999; Schade and Goldstein, 2001) formed by the oxidation of VOCs, but this second source is expected to be minor considering the residence time in dynamic chambers.

Although acrolein is mostly considered as an anthropogenic compound originating from combustion sources (Koss et al., 2018; Schieweck et al., 2021), positive acrolein fluxes were observed in all chambers with an average flux  
220 of  $0.009 \pm 0.033 \mu\text{g m}^{-2} \text{h}^{-1}$ . Acrolein has previously been associated with the decomposition of litter biomass in a mixed broadleaf and coniferous forest (Ehrlich and Cahill, 2018). Indeed, higher concentrations were measured from fresh leaves compared to senescent or decomposing leaves. Acrolein was also previously detected in litter of *Pinus halepensis* Mill. by Viros et al. (2021) and in litter of *Q. pubescens* litter in a previous study performed at O<sub>3</sub>HP (Legros et al., 2025), which reported an average flux 100 times higher than in the present study ( $0.79 \pm 0.04 \mu\text{g m}^{-2} \text{h}^{-1}$ ). Drier conditions during Legros et al. (2025) study compared to our experiment (Figure 2) could explain the differences observed between the two studies.

1H-pyrrole-2,5-dione ( $\text{C}_4\text{H}_3\text{NO}_2$ ) was identified here for the first time in litter VOC fluxes. This compound is a monopyrrolic derivative of the maleimide class/family corresponding to a transformation product of chlorophylls (Naeyer et al., 2016). Maleimides are commonly studied in paleoecological and human-based studies and possess  
230 various biological activities such as antibacterial, and antifungal properties (Ma et al., 2022).

Two other compounds not previously measured were identified in this study as  $\text{C}_{15}\text{H}_{18}\text{O}$ , proposed to be lindestrene (average flux of  $0.006 \pm 0.014 \mu\text{g m}^{-2} \text{h}^{-1}$ ), and creosol ( $\text{C}_8\text{H}_{10}\text{O}_2$ ,  $0.003 \pm 0.006 \mu\text{g m}^{-2} \text{h}^{-1}$ ). To our knowledge, no previous study documented lindestrene emissions (or immission) from soil, but it has been reported in emissions from tree resins (Taiti et al., 2018). Regarding creosol, it is a natural organic compound degradation in creosote, which can be further degraded by several fungal species (Lee et al., 2005; Atagana et al., 2006). This degradation  
235 process highlights the role of fungi in the decomposition of complex organic compounds, potentially contributing to soil and ecosystem health by removing phenols from contaminated soils (Atagana, 2004; Atagana et al., 2006).

Sesquiterpenes were also detected with average fluxes of  $0.003 \pm 0.006 \mu\text{g m}^{-2} \text{h}^{-1}$  and are frequently reported in litter of terpene storing species ((Asensio et al., 2008; Viros et al., 2021); (Yang et al., 2024a). Viros et al. (2020) reported VOC fluxes from *Q. pubescens* litter, but they did not observe sesquiterpenes, and it was only reported as minor emissions in a previous field study at O<sub>3</sub>HP (Legros et al., 2025). Since no storage organs are present in *Q. pubescens* leaves, the observed emission can be related to fungi, known to release high amounts of sesquiterpenes (Weigl et al., 2016; Mäki et al., 2017). However, no studies in the Mediterranean region report a relationship between fungi and sesquiterpenes fluxes.

245 Octanal ( $\text{C}_8\text{H}_{16}\text{O}$ , detected at m/z 129.132) fluxes reached on average  $0.002 \pm 0.004 \mu\text{g m}^{-2} \text{h}^{-1}$  for litter. Octanal has been previously found in Mediterranean litters for *Q. pubescens*, *Quercus suber* L. and *Ulex parviflorus* Pourr. (Viros et al., 2020; Legros et al., 2025), and more specifically in tundra soils where it accounts for about 75% of the total VOC emissions (Kramshøj et al., 2016). The processes responsible for octanal emissions remain unexplored.





- 250 Dibutyl phthalate ( $C_{16}H_{20}O_4$ ) fluxes were similar for both bare soil and soil with litter (on average:  $0.007 \pm 0.019 \mu\text{g m}^{-2} \text{h}^{-1}$ ). The occurrence of this compound in a rural environment such as  $O_3$ HP seems surprising as it is known as plasticizer compound, but emission can be related to microbial taxa, as shown by (Sillo et al., 2024), who reported a significant correlation between dibutyl phthalate fluxes and the abundance of *Paenarthrobacter* genus (actinomycete class such as Micrococcaceae).
- 255 **VOC immission**
- Immission was frequently observed for oxygenated compounds, such as alcohols (methanol,  $CH_4O$ , ethanol,  $C_2H_6O$ ), carboxylic acids (acetic,  $C_2H_4O_2$ , propanoic,  $C_3H_6O_2$  and formic  $CH_2O_2$  acids) and also for alkenes (isoprene,  $C_5H_8$  and toluene,  $C_7H_8$ ) and two nitrogenous compounds (acetonitrile,  $C_2H_3N$  and propanenitrile,  $C_3H_5N$ ).
- 260 Average fluxes for both bare soil and soil covered by litter were negative for methanol ( $-0.033 \pm 0.038$  and  $-0.044 \pm 0.047 \mu\text{g m}^{-2} \text{h}^{-1}$ , respectively) and ethanol ( $-0.024 \pm 0.035$  and  $-0.027 \pm 0.040 \mu\text{g m}^{-2} \text{h}^{-1}$ , respectively). In contrast, most of previous studies suggest that soil is a source of alcohols (Asensio et al., 2008; Gray et al., 2010; Yang et al., 2024b), except for (Asensio et al., 2007a) who reported both methanol and ethanol deposition in a Mediterranean *Quercus ilex* L. forest of northern Spain (from  $-95.04 \mu\text{g m}^{-2} \text{h}^{-1}$  to  $-71.28 \text{ng m}^{-2} \text{h}^{-1}$ ). Thus, whether
- 265 these compounds are primarily emitted from soils or deposited is strongly dependent on both biotic (microbial activity) and abiotic (meteorological parameters) factors. For example, (Asensio et al., 2007a) showed that VOC immission of alcohol is related to soil moisture. More generally, lignocellulose, which is the most widespread and abundant source of carbon in nature for microorganisms (Abdeshahian et al., 2020), reached concentrations of about  $187.6 \pm 1.5 \text{mg g}^{-1}_{\text{DW}}$  in *Q. pubescens* litter (Santonja et al., 2015), is considered the preferred consumed
- 270 biomass by bacteria for the production of ethanol (Du et al., 2015).
- Carboxylic acids showed negative fluxes, with acetic acid having the highest deposition rate ( $-0.088 \pm 0.115 \mu\text{g m}^{-2} \text{h}^{-1}$  for LM and  $-0.096 \pm 0.117 \mu\text{g m}^{-2} \text{h}^{-1}$  for bare soil), followed by formic acid ( $-0.082 \pm 0.099 \mu\text{g m}^{-2} \text{h}^{-1}$  for LM and  $-0.089 \pm 0.105 \mu\text{g m}^{-2} \text{h}^{-1}$  for bare soil) and propanoic acid ( $-0.011 \pm 0.022 \mu\text{g m}^{-2} \text{h}^{-1}$  for LM and  $-0.017 \pm 0.22 \mu\text{g m}^{-2} \text{h}^{-1}$  for bare soil). Previous studies showed contrasted features: while in agreement with our results
- 275 some studies observed deposition in Mediterranean forests (Asensio et al., 2007a, 2007b), Legros et al. (2025) observed emissions of acetic and formic acids. These compounds are sensitive to soil moisture, which favors microbial uptake and deposition of VOC on wet surfaces (Asensio et al., 2007a). Propanoic acid is known to be released during microbial fermentation (Wheatley et al., 1996; Leff and Fierer, 2008), while acetic or formic acids are reported to be produced by bacteria and fungi (Wheatley et al., 1996; Mielnik et al., 2018). Indeed, considering
- 280 the environmental conditions (i.e. humid) we hypothesize that carboxylic acid deposition processes linked to the soil moisture are dominating, leading to net deposition (further discussed in section 3.4).
- Isoprene showed an average immission rate of  $-0.048 \pm 0.106 \mu\text{g m}^{-2} \text{h}^{-1}$  for LM and  $-0.082 \pm 0.128 \mu\text{g m}^{-2} \text{h}^{-1}$  for bare soil. Accordingly, most studies have shown negative fluxes for isoprene (Cleveland and Yavitt, 1997; Gray et al., 2014, 2015; McGenity et al., 2018; Legros et al., 2025). Higher negative fluxes were estimated by Legros et al., (2025), with  $-0.92 \pm 0.20 \mu\text{g m}^{-2} \text{h}^{-1}$ , and Gray et al. (2014) with values up to  $-138.04 \pm 89.76 \mu\text{g m}^{-2} \text{h}^{-1}$ . The isoprene fluxes are driven by the microbial activity, as this compound is consumed by bacterial and fungal taxa (Gray et al., 2015), and it can be also released by root exudates (Asensio et al., 2007a). In addition isoprene is well known to be emitted by the canopy of *Q. pubescens* (Genard-Zielinski et al., 2015), and it is possible that specific bacteria or fungi that can use isoprene as carbon source developed in soil and litter at  $O_3$ HP.
- 285
- 290 Toluene ( $C_7H_8$ ) immission occurred in both bare soil and soil covered by litter ( $-0.013 \pm 0.014 \mu\text{g m}^{-2} \text{h}^{-1}$  on average), indicating that litter and soil are a sink for this compound. This VOC is well known to be of anthropogenic origin, associated with fossil fuel and biomass burning (Hanif et al., 2021) and, to a lesser extent, of plant stress (Heiden et al., 1999; Misztal et al., 2015). This compound may thus have been transported in the air masses affected by anthropogenic emissions and deposited on this rural ecosystem.
- 295
- As toluene, acetonitrile ( $C_2H_3N$ ) and propanenitrile ( $C_3H_5N$ ) are mainly known to be emitted from anthropogenic sources, and more particularly from biomass burning (Holzinger et al., 1999; Sarkar et al., 2016; Yang et al., 2016). However, acetonitrile emission has been reported in some plant species such as *Gynandropsis gynandra* L. (Briq.), where it is a repellent against spider mites (Nyalala et al., 2011, 2013). In our study, the averaged fluxes of acetonitrile and propanenitrile were negative with  $-0.038 \pm 0.040 \mu\text{g m}^{-2} \text{h}^{-1}$  and  $-0.005 \pm 0.006 \mu\text{g m}^{-2} \text{h}^{-1}$ , respectively. Acetonitrile deposition has been already measured in Mediterranean forest soils, ranging from  $-75.6$
- 300



to  $-15 \text{ ng m}^{-2} \text{ h}^{-1}$  (Asensio et al., 2008). In the present study, the origin of the acetonitrile and propanenitrile is probably regional rather than local, since these compounds are well known to have a long lifetime (more than one year, (Gouw et al., 2003; Andersen et al., 2018)), and deposition of this compound occurred in our forest soil.

### VOCs both emitted and immitted

305 Many oxygenated compounds showed bidirectional fluxes, or fluxes of opposite magnitude when comparing bare soil average fluxes to soil covered by litter, such as acetone ( $\text{C}_3\text{H}_6\text{O}$ ), cyclohexanecarboxylic acid (CHC acid,  $\text{C}_7\text{H}_{12}\text{O}_2$ ), monoterpenes ( $\text{C}_{10}\text{H}_{16}$ ) and oxygenated monoterpenes ( $\text{C}_{10}\text{H}_{16}\text{O}$ ).

Among these compounds, acetone is one of the most commonly detected volatiles released from soils (Asensio et al., 2008; Gray et al., 2010), with an average immission of  $-0.217 \pm 0.220 \text{ } \mu\text{g m}^{-2} \text{ h}^{-1}$  from bare soil, while by  
310 contrast a positive flux of  $0.019 \pm 0.195 \text{ } \mu\text{g m}^{-2} \text{ h}^{-1}$  from soil covered by litter. The positive flux is lower than Legros et al. (2025) who reported an average acetone flux of  $1.15 \pm 0.30 \text{ } \mu\text{g m}^{-2} \text{ h}^{-1}$ . This can be attributed to the different environmental conditions encountered during measurements.

Cyclohexanecarboxylic acid (CHC acid,  $\text{C}_7\text{H}_{12}\text{O}_2$ ) was found to have average positive flux in LM chambers ( $0.003 \pm 0.006 \text{ } \mu\text{g m}^{-2} \text{ h}^{-1}$ ) and negative for bare soil ( $-0.002 \pm 0.003 \text{ } \mu\text{g m}^{-2} \text{ h}^{-1}$ ). Like lindestrene and 1H-pyrrole-2,5-  
315 dione, cyclohexanecarboxylic acid (CHC acid) has not yet been reported in other studies using either bare soil or litter alone or bare soil covered by litter. CHC acid is produced naturally as part of the shikimate pathway, a major metabolic pathway for the biosynthesis of aromatic compounds in bacteria (*Alicyclobacillus acidocaldarius* and *Streptomyces collinus*), plants, and fungi (Shende et al., 2024). This pathway is key to the formation of phenolic compounds (Laoué et al., 2022). Thus, this novel finding highlights the potential for further exploration of its  
320 sources and ecological role.

Monoterpene fluxes were negative when the soil was covered by litter ( $-0.007 \pm 0.015 \text{ } \mu\text{g m}^{-2} \text{ h}^{-1}$  on average). By contrast to soil covered by litter, bare soil showed positive fluxes of monoterpenes ( $0.026 \pm 0.043 \text{ } \mu\text{g m}^{-2} \text{ h}^{-1}$  on average), as a previous study already observed (Leff and Fierer, 2008). The negative fluxes when the soil was covered by litter suggest that microorganisms consume monoterpenes, as these compounds have been reported as  
325 a carbon and energy source for microorganisms (White, 1994; Marmulla and Harder, 2014). Bacteria also consume monoterpenes in order to protect themselves from the monoterpene-related toxic effects (Marmulla and Harder, 2014), and fungi are known to transform monoterpenes during growth (Farooq et al., 2004).

Fluxes of oxygenated monoterpenes were measured of  $0.001 \pm 0.002 \text{ } \mu\text{g m}^{-2} \text{ h}^{-1}$  for litters emissions and  $-0.001 \pm 0.002 \text{ } \mu\text{g m}^{-2} \text{ h}^{-1}$  for bare soil, on average. While it has not been reported for *Q. pubescens* forests, previous study has shown concentrations of camphor from *Rosmarinus officinalis* in the Mediterranean region ( $3.18 \pm 0.42 \text{ mg g}^{-1} \text{ DW}$  by (Ormeño et al., 2008) and  $9.9 \pm 3.4 \text{ mg g}^{-1} \text{ DW}$  by (Staudt et al., 2017). Oxygenated monoterpenes ( $\text{C}_{10}\text{H}_{16}\text{O}$ , such as camphor, fenchone or carveol) may originate from litter decomposed by fungi, as proposed by (Isidorov and Jdanova, 2002).

### 3-2 VOC variation with different litter masses

335 Correlations of VOC fluxes with the litter masses inside the chambers were performed (Figure 4). Surprisingly, only a few compounds showed significant relationships between litter masses and fluxes. Correlations were positive only for methanol, acetone, oxygenated monoterpenes, CHC acid and sesquiterpenes. For other compounds, the lack of correlation may be due to a high variability, or other environmental factors that are more important drivers of emission/deposition.

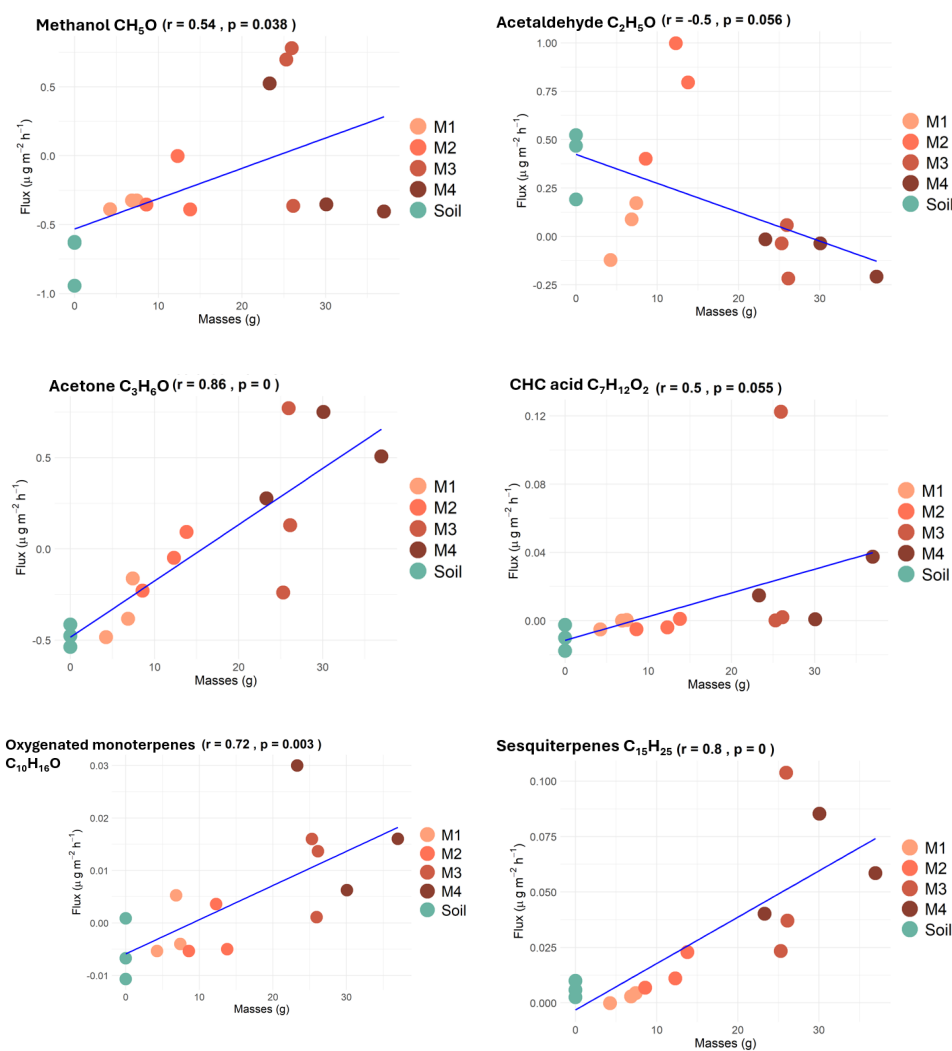
340 Methanol, acetone, oxygenated monoterpenes and CHC acid fluxes showed a positive correlation with litter masses (Figure 4), but average bare soil fluxes were negative. For methanol and CHC acid, the correlation with LM is weaker ( $r = 0.54$ ,  $p_{\text{value}} = 0.038$  for methanol,  $r = 0.50$ ,  $p_{\text{value}} = 0.055$  for CHC acid) than acetone and oxygenated monoterpenes ( $r = 0.63$ ,  $p_{\text{value}} < 0.001$  for acetone,  $r = 0.86$ ,  $p_{\text{value}} < 0.001$  for oxygenated monoterpenes). For these 4 compounds, the net flux is increasing with litter mass, from a negative flux in bare soil to a positive flux at high  
345 LM. As a result, it suggests that the soil alone acts as a sink for these 4 compounds, while litter mostly emits them, the resulting net fluxes depending on the amount of LM and maybe environmental conditions. However, a deeper study dedicated to separate the respective contribution of soil and litter will be necessary to fully elucidate this point.





350 Negative response has only been observed between acetaldehyde fluxes and LM ( $r = -0.50$ ,  $p_{\text{value}} = 0.056$ ), although average positive fluxes were reported for LM 2 ( $0.343 \pm 0.379 \mu\text{g m}^{-2} \text{h}^{-1}$ , Figure 4). Opposite to the previous 4 compounds, emissions are observed for bare soil while the net fluxes turned negative with increasing LM, suggesting that the soil is a source of acetaldehyde, further deposited on litter. It may be due to the presence of microorganisms on leaf litter uptaken acetaldehyde, or different environmental conditions.

355 Finally, sesquiterpenes were the most significant VOC linked to the LM ( $r = 0.80$ ,  $p_{\text{value}} < 0.001$ ), with positive fluxes for all modalities increasing together with litter mass. If no storage organs are present in *Q. pubescens* leaves, the abundance of potential fungi associated with sesquiterpene emissions, as suggested by (Weigl et al., 2016; Mäki et al., 2017) should depend on LM.



**Figure 4.** Correlation plots of methanol, acetaldehyde, acetone, oxygenated monoterpenes, CHC acid and sesquiterpenes fluxes ( $\mu\text{g m}^{-2} \text{h}^{-1}$ ) with litter masses (0 in bare soil).



360 **3-3 VOC diurnal variations**

The diurnal variations of the main VOC fluxes ( $F > 0.005 \mu\text{g m}^{-2} \text{h}^{-1}$ , 13 % of the total compounds = 21 compounds, ~75 % of the total fluxes) and temperature and relative humidity are shown in Figure 5 and Figure S.1. The temperature was highest between 8 h and 15 h (local time), ranging from 15 °C to 20 °C, and exceptionally up to 30 °C in LM 3. By contrast, the humidity decreased during the same hours up to 75 % for LM 3 and reached 100 % during nighttime.

Most of the compounds showed an increase of their fluxes between 10 h and 15 h local time. Strong diurnal variations were measured with high fluxes during daytime in LM 3 for methanol ( $0.797 \pm 2.355 \mu\text{g m}^{-2} \text{h}^{-1}$ ), 1H-pyrrole-2,5-dione ( $0.050 \pm 0.040 \mu\text{g m}^{-2} \text{h}^{-1}$ ), CHC acid ( $0.009 \pm 0.014 \mu\text{g m}^{-2} \text{h}^{-1}$ ) and acetone ( $0.165 \pm 0.402 \mu\text{g m}^{-2} \text{h}^{-1}$ ).

370 Patterns can vary between compounds within the LM. Indeed, similar patterns can be highlighted for 1H-pyrrole, CHC acid, lindrestrene and sesquiterpenes with higher emissions in LM 3 and 4 during the day and very low during nighttime. The 1H-pyrrole-2,5-dione, lindrestrene and CHC acid fluxes were only significant in the two highest LM scenarii (Supplementary Information, averaged fluxes). These fluxes are dependent on the litter mass but not linearly, and include a threshold effect, explaining why they were not discussed in the previous section.

375 Monoterpenes were released from bare soil and LM 1 between 12 h and 17 h, and then uptaken from 18 h to midnight (Figure 5). A specific pattern was shown for isoprene, formaldehyde (HCHO) and formic acid. Indeed, these fluxes decreased between 5 h and 10 h and increased around 12 h, when isoprene emissions from trees were maximal, as they are linked to temperature and solar radiation (Monson et al., 1992; Kesselmeier et al., 1998; Owen et al., 2002). Around midnight, isoprene fluxes were stable in all LM and no exchange occurred between the soil and the atmosphere (i.e. concentrations inside the chamber were identical to ambient concentrations). As expected, isoprene oxidation products (methacrolein, MACR; methyl vinyl ketone, MVK and Isoprene Hydroxy Hydroperoxide, ISOPOOH) showed the same pattern as isoprene, with an uptake of this compound in bare soil and all LM. Their fluxes increased between 10 h and 12 h with averaged fluxes up to  $-0.021 \pm 0.022 \mu\text{g m}^{-2} \text{h}^{-1}$  during daytime.

385 Formic acid, acetone and formaldehyde were especially sinked by the soil during nighttime when the humidity increased and reached 100 % ( $-0.052 \pm 0.023 \mu\text{g m}^{-2} \text{h}^{-1}$ ,  $-0.059 \pm 0.073 \mu\text{g m}^{-2} \text{h}^{-1}$  and  $-0.027 \pm 0.008 \mu\text{g m}^{-2} \text{h}^{-1}$ , respectively). Propanoic and acetic acids also showed negative fluxes ( $-0.008 \pm 0.027 \mu\text{g m}^{-2} \text{h}^{-1}$  and  $-0.077 \pm 0.114 \mu\text{g m}^{-2} \text{h}^{-1}$ ) during daytime. These differences can be attributed to the fact that fluxes of these oxygenated compounds are influenced by abiotic factors, particularly soil moisture and temperature. (Jacob et al., 2002) have demonstrated that both acetone fluxes and concentrations increase with rising temperatures. Furthermore, the physicochemical degradation of VOCs can act as a sink for VOCs at the soil level (Peñuelas et al., 2014). Our results are in line with Legros et al., (2025) for daytime acetone and isoprene fluxes where they measured positive and negative fluxes for acetone and isoprene ( $3.52 \pm 0.40$  and  $-1.45 \pm 0.41 \mu\text{g m}^{-2} \text{h}^{-1}$  in Legros et al., 2025, respectively).

395 Negative methanol fluxes were recorded for all chambers and essentially during nighttime (Figure 5) and slightly positive during daytime except for LM 2 and LM 3 where the flux is higher than the other days (up to  $0.797 \pm 2.354 \mu\text{g m}^{-2} \text{h}^{-1}$ ). For LM 3, the environmental conditions showed few differences with higher temperature and then lower humidity during the measurements, which can explain differences in observed fluxes. The role of environmental parameters in the processes of consumption or emission of methanol by microorganisms have been already studied (Cleveland and Yavitt, 1998; Schade and Goldstein, 2001; Asensio et al., 2007b; Gray et al., 2015; Jiao et al., 2023b).

400 Positive fluxes were observed during the day for acetaldehyde ( $0.184 \pm 0.245 \mu\text{g m}^{-2} \text{h}^{-1}$ ), except for LM4 and negative acetaldehyde fluxes were observed during nighttime for LM 3 and LM 4 ( $-0.076 \pm 0.029 \mu\text{g m}^{-2} \text{h}^{-1}$  and  $-0.066 \pm 0.033 \mu\text{g m}^{-2} \text{h}^{-1}$ , respectively). This result is in line with previous studies that showed that acetaldehyde can be both emitted and uptaken by soil (Asensio et al., 2008; Peñuelas et al., 2014; Jiao et al., 2023b). Ethanol fluxes showed the same pattern as acetaldehyde with higher fluxes between 10h and 12h ( $-0.017 \pm 0.022 \mu\text{g m}^{-2} \text{h}^{-1}$  on average daytime).

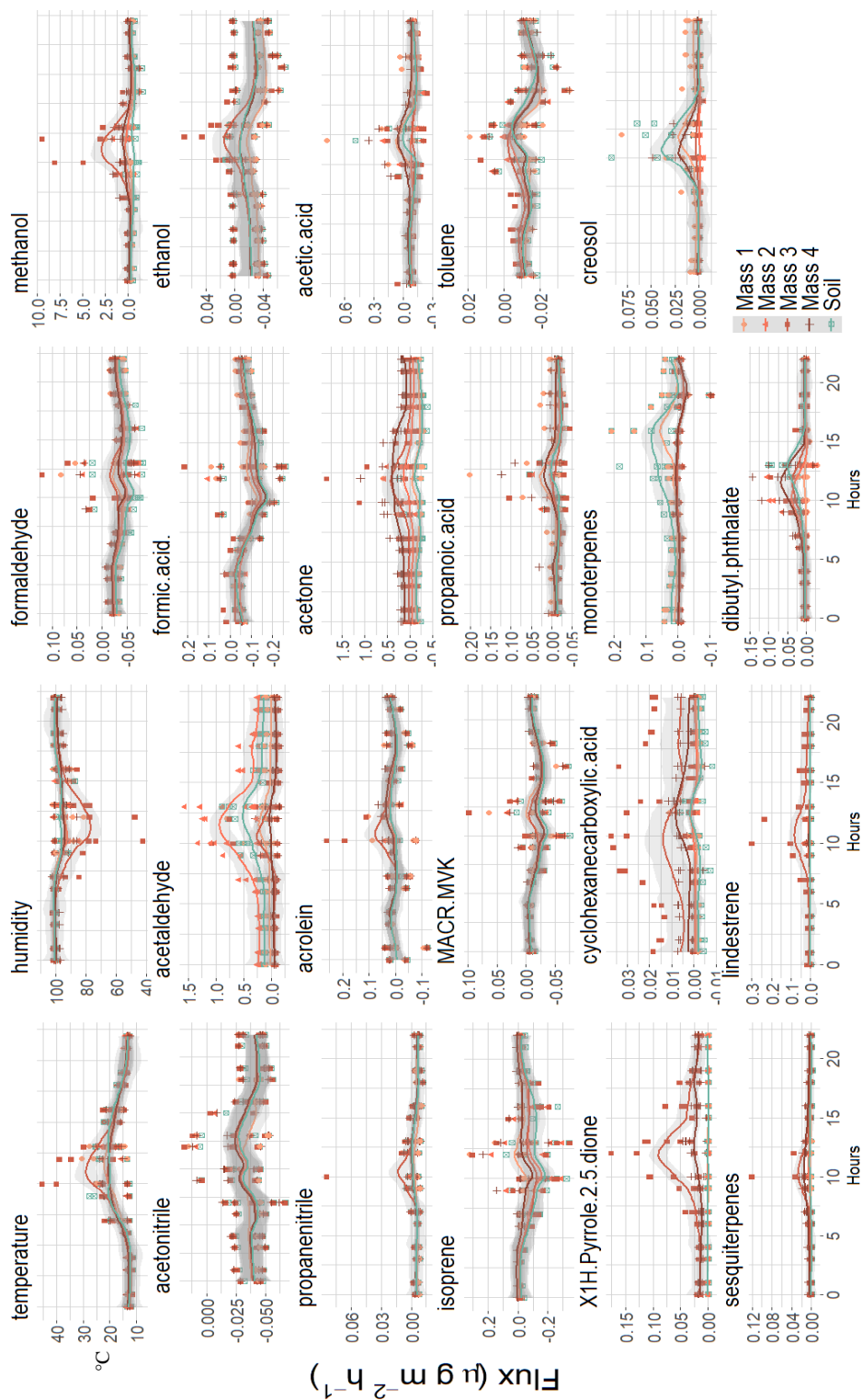
In Legros et al. (2025), acrolein has been shown to have high flux variations during day and night (diurnal cycle). Our measurements were made almost one year after the Legros et al. (2025) study at the same site. However,



410 environmental conditions were completely different and the soil was not submitted to the same constraints explaining the differences observed between our fluxes and this previous study.

Toluene, creosol, propanenitrile and acetonitrile are compounds with long life time ( $> 1$  year). Increases of the magnitude of fluxes during daytime have been reported. However, the fluxes remained negative for toluene ( $-0.012 \pm 0.009 \mu\text{g m}^{-2} \text{h}^{-1}$ ), propanenitrile ( $-0.004 \pm 0.007 \mu\text{g m}^{-2} \text{h}^{-1}$ ) and acetonitrile ( $-0.034 \pm 0.018 \mu\text{g m}^{-2} \text{h}^{-1}$ ) and turned positive for creosol ( $0.006 \pm 0.011 \mu\text{g m}^{-2} \text{h}^{-1}$ ). Dibutyl phthalate fluxes were positive and higher during daytime ( $0.016 \pm 0.025 \mu\text{g m}^{-2} \text{h}^{-1}$ ) than nighttime ( $0.003 \pm 0.003 \mu\text{g m}^{-2} \text{h}^{-1}$ ).

415



**Figure 5.** Temperature and humidity diurnal cycles (24h) and measured BVOC fluxes ( $\mu\text{g m}^{-2} \text{h}^{-1}$ ) for bare soil and soil covered by increasing litter mass (LM 1, 2, 3, 4).



### 3-4 VOC fluxes linked to environmental and biological parameters

420 As the previous section illustrates the strong variability during the hours of the day, but also between the different LM scenarii, we investigated the relation between BVOC fluxes and biotic/abiotic factors to better understand the main drivers of these fluxes. For this purpose, a correlation analysis was performed based on hourly fluxes and presented in Figure 6.

425 First, a strong positive correlation can be observed between monoterpenes and arbuscular mycorrhizal (AM) fungi, forming the first cluster ( $r = 0.48$ ,  $p_{\text{value}} < 0.001$ , Figure 6). We suggested earlier that soil fungi are the main source of monoterpenes, as negative monoterpene fluxes were observed when LM increased. The concentration of AM fungi is observed to be higher in the bare soil compared to LM, which confirms that AM fungi can be the source of monoterpenes from the soil e. This confirms the results of previous studies, showing that fungi are at the origin of monoterpene emissions (Farooq et al., 2004; Marmulla and Harder, 2014).

430 Compounds observed in cluster 2 were positively related to the temperature inside the chambers, and logically negatively related to relative humidity. Most of the compounds are deposited (except acetaldehyde and creosol), and their correlation with temperature indicates that deposition is higher when temperature is lower, or humidity is higher. Acetonitrile\* and propanenitrile were among the compounds with the strongest response to temperature \* ( $r_{\text{acetonitrile/temp}} = 0.50$ ,  $r_{\text{propanenitrile/temp}} = 0.59$ ). As mentioned previously, acetonitrile and propanenitrile main atmospheric sources are considered to be combustion processes, even if a few studies showed biogenic origins (Heiden et al., 1999; Nyalala et al., 2011, 2013; Misztal et al., 2015). This result may indicate that deposition is favoured by lower temperatures or higher humidity, with lower temperatures reducing the volatility of these species, and humidity favouring adsorption and dissolution in water (especially for hydrophilic compounds).\* Acetonitrile fluxes were linked to propanenitrile, propanoic and acetic acids and acetone ( $r_{\text{acetonitrile/propanenitrile}} = 0.55$ ,  $r_{\text{acetonitrile/propanoic acid}} = 0.60$ ,  $r_{\text{acetonitrile/acetic acid}} = 0.55$  and  $r_{\text{acetonitrile/acetone}} = 0.58$ , respectively). As a few positive fluxes were observed for these nitrogen-containing VOCs, it suggests that their potential emissions can be biogenic, as acetone and acids have a microbial fermentation origin. Acetaldehyde and creosol are also included in this second cluster but are mainly emitted compounds (see section 3.1). Acetaldehyde has been described as an oxidation product of ethanol. However, only moderate correlations were measured between these two compounds ( $r = 0.36$ ,  $p_{\text{value}} < 0.001$ ), confirming that there is no strong photochemistry inside the chambers. However, strong relationships were observed with toluene and methanol with ethanol ( $r_{\text{ethanol/toluene}} = 0.64$ ,  $r_{\text{ethanol/methanol}} = 0.56$ ), indicating that these three compounds are likely to share a common origin.

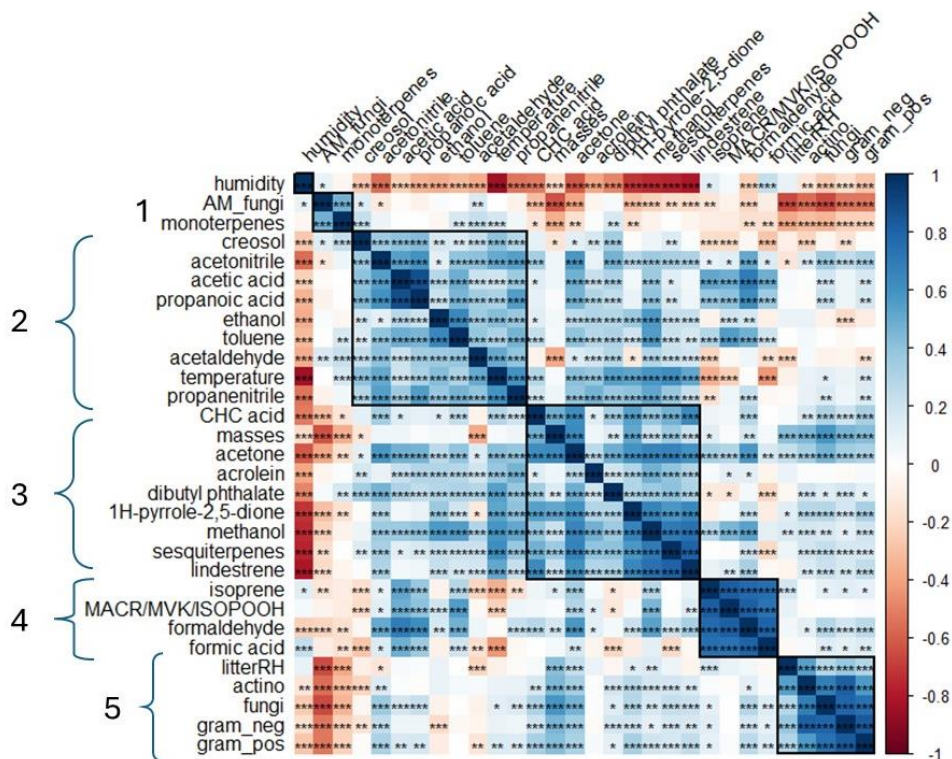
445 Cluster 3 gathers compounds positively correlated with the litter masses highlighted in section 3.2, and some other compounds (CHC acid, acrolein, dibutyl phthalates and 1H-pyrrole-2,5-dione, e.g.  $r_{\text{acetone/LM}} = 0.63$ ,  $r_{\text{CHC acid/LM}} = 0.58$ ). In addition, we showed previously (Figure 4) that these compounds present the same pattern with higher emissions during daytime and in LM3 and LM4. This means that the emissions of these compounds are either directly related to the litter (emission from the dead leaves) or indirectly through the presence of microorganisms. However, our study did not show a link between dibutyl phthalate fluxes and actinomycetes, as suggested by Sillo et al. (2024). This difference may be explained by the type of soil, since their study focused on urban soils that can be more polluted. Studies have demonstrated that both wet and dry conditions may influence microbial biomass, by creating environments unfavorable for aerobic g+ and gram-negative (g-) bacteria, as well as mycorrhizal fungi (Borowik and Wyzkowska, 2016). Most particularly, actinomycete abundance has been shown to increase with decomposing organic matter. However, actinomycetes are highly pH-sensitive organisms, and their abundance decreases significantly at pH levels below 5 (Kovacs et al., 2023). This parameter was not measured in our experiments, so we cannot establish a link with actinomycete abundance.

460 Temperature was negatively related to compounds in cluster 4. Surprisingly, isoprene fluxes were negatively correlated with temperature ( $r = -0.35$ ,  $p_{\text{value}} < 0.001$ ), and so does for their oxidation products MACR/MVK/ISOPOOH ( $r = -0.24$ ,  $p_{\text{value}} < 0.001$ ) and formic acid ( $r = -0.43$ ,  $p_{\text{value}} < 0.001$ ). In other words, as these compounds are mostly deposited, the magnitude of deposition is increasing with temperature. Indeed, *Q. pubescens* is well documented as a strong isoprene emitter, emissions being well known to be higher with high temperatures (Genard-Zielinski et al., 2015). The temperature dependance of the deposition can thus be explained by the increasing production of isoprene by the green parts of *Q. pubescens*, which makes it more available for deposition. All compounds in this cluster are strongly correlated between themselves. MACR/MVK/ISOPOOH, formaldehyde and formic acid are mostly associated as secondary products formed by the oxidation of isoprene. Isoprene fluxes have been found significantly correlated with MACR/MVK/ISOPOOH, formaldehyde and formic acid ( $r_{\text{isop/MACR}} = 0.83$ ,  $r_{\text{isop/HCHO}} = 0.78$  and  $r_{\text{isop/formic acid}} = 0.75$  with  $p_{\text{value}} < 0.001$ ). These compounds can be



attributed to secondary origin from the degradation of isoprene. In addition, regarding the daily fluxes, isoprene, MACR/MVK/ISOPOOH, formaldehyde and formic acid fluxes presented the same pattern supporting the stronger relationship between these compounds. We can hypothesize here that the production of these compounds is following isoprene emissions, and the deposition rate simply increased accordingly. This hypothesis is further supported by the absence of a relationship between these cluster's compounds fluxes and PLFA (g+, g-, actinomycetes and fungi).

Finally, soil moisture (litterRH in Figure 6) showed positive correlations with gram-positive and gram-negative bacteria (g+ and g-), actinomycete and fungi ( $r_{\text{litterRH/g}^+} = 0.29$ ,  $r_{\text{litterRH/g}^-} = 0.37$ ,  $r_{\text{litterRH/actino}} = 0.55$ ,  $r_{\text{litterRH/fungi}} = 0.29$ , with  $p_{\text{value}} < 0.001$ , values are reported in Supplementary file), all forming cluster 5. Negatively significant but moderate relationships of monoterpenes fluxes with fungi have been observed ( $r_{\text{monop/fungi}} = -0.33$ ,  $p_{\text{value}} < 0.001$ ). Monoterpene and acetaldehyde fluxes were likely driven by the abundance of actinomycetes and gram-negative bacteria, and were consumed by these microorganisms. This explains why we observed a predominance of negative fluxes for these compounds when litter accumulation increased inside the chambers.



485 **Figure 6.** Correlation matrix between BVOC fluxes and abiotic (air temperature and humidity) and biotic (litter mass, litter humidity content, PLFA) factors. Positive correlations ( $r > 0$ ) were highlighted in blue and negative ( $r < 0$ ) in red. Stars represent  $p_{\text{values}}$  and their confidence: \*  $\geq 0.05$ , 0.01  $< ** < 0.05$  and 0.001  $< *** < 0.01$ . CHC acid = cyclohexanecarboxylic acid. Cluster's number is at the left of the correlation plot. AM\_fungi = arbuscular mycorrhiza fungi.

490 **4- Conclusion**

A wide range of VOCs were measured from bare soil and litter, with over 135 compounds identified. Positive and negative fluxes were observed, with some fluxes appearing to be independent of litter mass. About 48% of the compounds showed negative fluxes, indicating uptake by the soil and soil covered by litter, 37% showed both emission and uptake, and 15% were emitted. The most uptaken compounds were ethanol, formaldehyde, isoprene, acetonitrile, toluene, formic and acetic acids. Thanks to holistic methodology of measurement, some VOCs such





as lindestrene, cyclohexanecarboxylic acid (CHC acid) and 1H-pyrrole-2,5-dione, were identified from the litter for the first time. The origin of these compounds can be attributed to biological sources, particularly microbial activity.

500 The diurnal variation of the main VOC fluxes showed an increase of their fluxes between 10h and 15h local time. Following the studied VOC, we observed positive or negative fluxes representing respectively emission or immission over bare soil and soil covered by litter. Indeed, temperature and humidity conditions were different in LM 3 that we correlated to higher fluxes for some compounds such as lindestrene, CHC acid or 1H-pyrrole-2,5-dione. We performed correlations between VOC fluxes, environmental parameters and microbial biomasses. Monoterpenes fluxes were negatively correlated with actinomycete and gram-negative bacteria biomasses and are therefore likely to be consumed by these organisms. We also observed higher fluxes of acetonitrile, propanenitrile and acetaldehyde with increasing temperature, showing that these compounds can be biogenically emitted, but we cannot rigorously explain their origins at this time.

510 Based on the results of this study, we can conclude that no clear relationship between litter accumulation and VOC fluxes could be established, except for a few compounds such as acetone, sesquiterpenes, CHC acid or methanol. This study represents only a first step toward understanding the relationship between litter and VOC fluxes. It is not sufficient to conclude that litter mass has no effect on VOC emissions. We recommend that further experiments are carried out under varying environmental conditions and in different locations to gain a deeper understanding of the role and impact of litter accumulation on VOC fluxes.

#### Data availability:

515 All raw data can be provided by the corresponding authors upon request and most of data are provided by the supplementary material.

#### Author contribution:

520 MR and JK analyzed the data. MR, JK, wrote the manuscript draft. EO, BTR, MS reviewed and edited the manuscript. EO, MS, BTR and EO planned the campaign and performed the measurements. CS and CL analyzed PLFA and provided PLFA raw data. EQ and HW reviewed the manuscript and served as the principal investigators responsible for securing funding for this project.

#### Competing interests:

The authors declare that they have no conflict of interest.

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