



Sea ice melt drives vertical pCO₂ variability modulating air-sea gas exchange

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Key Points:

- Spring melting of sea ice and snow introduces distinct heterogeneity in surface water conditions within coastal Arctic oceans.
- Standard bulk parameterizations for air-sea CO₂ flux calculations, based on subsurface pCO₂ measurements, may misrepresent flux direction and magnitude during melt periods.
- Vertical near-surface temperature and CO₂ gradients must be considered to improve flux estimates in stratified Arctic fjords.



30 Abstract

31 Strong spatial and temporal gradients in salinity, temperature, and carbonate chemistry in Arctic
32 coastal surface waters complicate the estimation of air-sea CO₂ exchange, particularly during sea
33 ice breakup. This study evaluates the applicability of the widely used bulk flux model under such
34 conditions. The bulk approach assumes homogeneous surface conditions and linear vertical
35 pCO₂ gradients. However, our observations in a stratified Arctic fjord reveal pronounced vertical
36 variability in pCO₂ within the upper water column, including non-linear gradients near the air-
37 sea interface. Micrometeorological measurements captured episodic upward CO₂ fluxes even
38 when waters 1 m and below were CO₂-undersaturated. We hypothesize that transient, high-pCO₂
39 layers at ~0.1 m depth intermittently decouple the atmospheric exchange from subsurface waters,
40 reversing the expected flux direction. These findings highlight the importance of resolving near-
41 surface variability during the transition from ice-covered to open water conditions. We
42 recommend incorporating micrometeorological techniques and high-resolution vertical profiling
43 in Arctic fjords to improve flux estimates of CO₂ in this rapidly changing region.

44 Plain Language Summary

45 Sea ice melt adds less-saline water to the surface ocean. This creates vertical gradients in
46 salinity, temperature, and partial pressures of carbon dioxide (pCO₂). The concentration
47 difference of pCO₂ across the air-ocean boundary is used to estimate gas transfer. Thus, the
48 depth that we measure will impact our estimates. Directly measuring gas transfer showed CO₂
49 release from the ocean during sea ice breakup. This means ocean layering during ice melt may
50 briefly reverse CO₂ transfer.

52 1 Introduction

53 High latitude coastal oceans are strong sinks for atmospheric carbon dioxide (CO₂), absorbing
54 more CO₂ per unit area than lower latitude regions (Dai et al., 2022; Roobaert et al., 2019). This
55 strong uptake results from both the high solubility of gases in cold water and the intense
56 biological activity typical of these regions. However, climate change is rapidly transforming this
57 carbon sink. The Arctic is warming more than twice as fast as the global average, and sea-ice
58 extent has been shrinking by over 13% per decade (Perovich et al., 2020). The loss of sea ice
59 increases CO₂ uptake by exposing larger areas of open water for longer periods, which can
60 further stimulate biological productivity (Arrigo and van Dijken, 2015; Bates and Mathis, 2009;
61 Perovich et al., 2020). However, at the same time, melting sea ice freshens the surface layer and
62 strengthens stratification, limiting vertical mixing with deeper water. Freshwater from melting
63 sea ice and terrestrial run-off creates pronounced gradients in physical properties such as salinity
64 and temperature, as well as chemical properties like dissolved inorganic carbon (DIC) and total
65 alkalinity (TA) (e.g. Henson et al., 2025). As a result, the partial pressure of CO₂ (pCO₂) can
66 vary markedly with depth under melt conditions.

68



This vertical variability in $p\text{CO}_2$ poses a challenge for air-sea CO_2 flux estimation. The transfer of gases between the atmosphere and ocean depends on the difference in concentration between the two as well as the efficiency of the transfer process. Therefore, the bulk flux of CO_2 across the air-sea interface is commonly described as the product of the gas transfer velocity, k (m s^{-1}), CO_2 solubility s ($\text{mol kg}^{-1} \text{atm}^{-1}$), and the partial pressure gradient (μatm) across the air-sea interface (Wanninkhof et al., 2009):

$$F = ks(p\text{CO}_{2_{\text{sea}}} - p\text{CO}_{2_{\text{air}}}) \quad (1)$$

While widely applied, this formulation simplifies a complex process influenced by surfactants on the water surface, bubble-mediated gas exchange, and turbulence. Furthermore, surface water heterogeneity, driven by sea ice melt and freshwater runoff from land, complicate the physical and chemical processes governing air-sea CO_2 exchange. As a result, simplified parameterizations commonly used in global carbon flux estimates may be inadequate in these settings.

In many studies, $p\text{CO}_2$ is measured several meters below the surface, assuming vertical homogeneity under well-mixed condition (Jørgensen et al., 2020). However, in stratified waters, where temperature, salinity, and pH can vary with depth, this assumption may lead to substantial errors in flux estimates (Ahmed et al., 2020; Dong et al., 2021; Miller et al., 2019; Watts et al., 2022). Although Arctic surface waters are often undersaturated with respect to atmospheric CO_2 levels and act as CO_2 sinks (e.g., Burgers et al., 2017; Dai et al., 2022; Henson et al., 2024; Laruelle et al., 2014; Roobaert et al., 2019), such assessments typically rely on sparse data collected from 0.5-5 m depth during limited periods. Dong et al. (2021) illustrate that high latitude fluxes of CO_2 calculated from the bulk method (based on measurements sampled at 6 m depth) differ significantly from those measured using direct eddy covariance.

Gas transfer velocity (k) is often parameterized as a function of wind speed. However, the true driver is mixing in the surface waters, which governs k . Fick's first law of diffusion, which underlies Equation (1), assumes a linear concentration gradient within the diffusive sublayer (Fig. 1) and steady-state conditions (Garbe et al., 2014). Jørgensen et al. (2020) argued that, due to seawater's high buffer capacity, chemical gradients do not significantly affect CO_2 equilibration, supporting the use of measurements at 3-4 m depth. However, this conclusion relies on the assumption of horizontal and vertical homogeneity and neglects the effects of shallow surface stratification, particularly when alkalinity dilution is involved.

In Arctic spring, the upper ocean is often strongly stratified due to freshwater input from glacier melt, snowmelt, river runoff, and sea ice meltwater (Ahmed et al., 2020; Granskog et al., 2011; Meire et al., 2017; Miller et al., 2019). These inputs can extend vertical CO_2 gradients beyond the diffusive sublayer, complicating flux estimates during ice break-up and early open-water

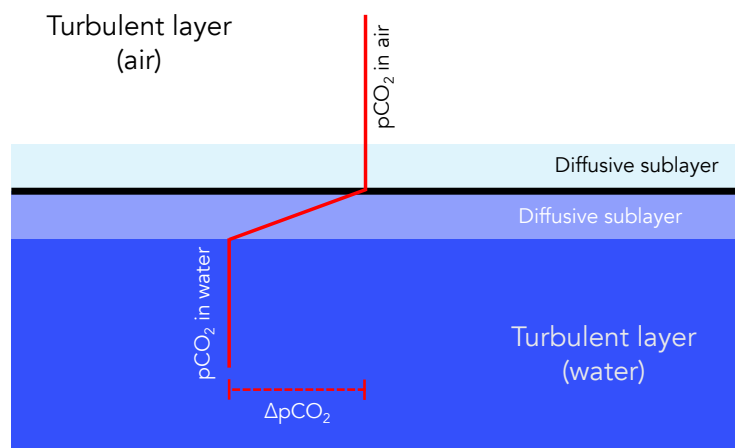


109 periods. Several studies have demonstrated strong vertical heterogeneity in $p\text{CO}_2$ in Arctic
 110 coastal waters, with implications for air-sea flux calculations (Ahmed et al., 2020; Dong et al.,
 111 2021; Miller et al., 2019).

112 Surface freshening from ice melt and runoff strongly influences carbonate chemistry in Arctic
 113 coastal waters, which can either suppress or enhance oceanic CO_2 uptake. For example, Burgers
 114 et al. (2017) reported large horizontal variability in surface $p\text{CO}_2$ (144–364 μatm) linked to
 115 riverine input in the Eastern Canadian Arctic. Similarly, Sejr et al. (2011) observed strong
 116 surface $p\text{CO}_2$ gradients associated with salinity and temperature in Young Sound, and later
 117 documented a long-term decline in surface salinity (Sejr et al., 2017). Freshwater-induced
 118 stratification has also been shown to create vertical gradients in $p\text{CO}_2$ and pH with important
 119 implications for flux calculations (Miller et al., 2019). Finally, Bates et al. (2014) demonstrated
 120 that sea ice meltwater and melt ponds exhibit extreme variability in $p\text{CO}_2$ (<10 to >1500 μatm)
 121 and pH (6.1 to >10.8), highlighting the complex chemical landscape of ice-influenced waters.
 122 Together, these studies underscore the high spatial and temporal variability of carbonate
 123 chemistry in freshened waters across the Arctic.

124 To project future CO_2 uptake or outgassing in the Arctic, we must better understand the physical
 125 and chemical drivers of near-surface carbonate variability. In this study, we investigate the
 126 vertical and temporal variations in $p\text{CO}_2$ in a stratified Arctic fjord during sea ice breakup. By
 127 combining micrometeorological flux measurements with water-column $p\text{CO}_2$ profiles across the
 128 transition from ice-covered to open water, we evaluate the performance of the bulk flux model
 129 under Arctic seasonal transitions.

130



131
 132 **Figure 1.** Schematic illustrating the interface between the air and the water in conjunction with
 133 $p\text{CO}_2$ concentration gradients. In equation 1, the concentration gradient is assumed to occur in the
 134 diffusive layer between the air and water, and the concentrations are assumed to be vertically
 135 constant in the turbulent layers. (Adapted from Wanninkhof et al. 2009)



2. Study Site and Measurement Methods

2.1 Study Site

This study was conducted in Young Sound, a high Arctic fjord system located near the Daneborg Research Station in Northeast Greenland (Fig. 2). The fjord system comprises the Tyrolerfjord (inner fjord) and Young Sound (outer fjord), extending approximately 90 km from Tyroler River to the Greenland Sea. A sill at about 45 m depth separates Young Sound from the open ocean. Young Sound is 2 to 7 km wide, with an average depth of 100 m (maximum 350 m), and a total surface area of ~390 km². Tidal amplitudes range from 0.8 to 1.5 m, with mean current velocities of approximately 2 cm s⁻¹ (Rysgaard et al., 2003). Freshwater inputs are primarily derived from Greenland Ice Sheet runoff, local glaciers, precipitation, and snowmelt from adjacent ice-free terrain. The drainage basin of the Tyrolerfjord/Young Sound system spans 2846 km², of which 33% is glaciated.

Sampling was conducted from 12 to 31 July 2017. Sampling occurred during and immediately after a period of sea ice breakup. On 15 July, ice coverage was approximately 30%, decreasing to less than 10% by 16 July. Water sampling was conducted both from an inflatable boat and via sea ice leads, all in close proximity to the Greenland Ecosystem Monitoring (GEM) program's standard station (Fig. 2).

2.2 pCO₂ Measurements Using the HydroC Sensor

Surface water pCO₂ was measured with a CONTROS® HydroC CO₂ sensor, which utilizes a membrane equilibrator coupled with a non-dispersive infrared detector. The instrument is equipped with a built-in water pump that provides flow rate of 35 ml s⁻¹ across the membrane. At each sampling depth, the sensor was allowed to equilibrate for 10 to 20 minutes, and values were recorded once stable for at least two minutes. The sensor operates over a range of 200-1000 µatm and temperatures of -2 to 35°C. Annual calibration has been conducted using a certified 400 ± 2% ppm CO₂ gas that was traceable to WMO standards. The sensor showed remarkable stability (397-401 ppm), supporting a measurement uncertainty of ± 2 µatm.

2.3 pCO₂ Estimation from TA and DIC

In addition to direct measurements, pCO₂ was calculated from total alkalinity (TA) and dissolved inorganic carbon (DIC) using the Seacarb package (Gattuso et al., 2024) in R. Due to the low salinity and cold temperatures characteristic of Arctic coastal waters, no universally accepted set of equilibrium constants (K₁ and K₂) exists. For consistency with previous studies in the region (Henson et al., 2023), we used the refitted constants from (Lueker et al., 2000). The selection of equilibrium constants introduces assumptions regarding seawater composition. (Raimondi et al., 2019) showed that different constants can lead to discrepancies between measured and calculated pCO₂ values, ranging from -3.1 to -35.8 µatm, with Lueker et al. (2000) demonstrating the best internal consistency under polar conditions. Still, (Sulpis et al., 2020) found that the calculation



of $p\text{CO}_2$ from DIC and TA can lead to uncertainty up to 15% under cold conditions, which is far greater than when $p\text{CO}_2$ is measured directly.

2.4 Sea Ice TA and DIC Sampling

TA and DIC in sea ice were assessed using three ice cores. Each core was sectioned into 5-10 cm segments and sealed in gas-tight NEN/PE bags with sampling valves (Hansen et al., 2000). Samples were transported in thermally insulated boxes to a nearby field laboratory. Cold (1°C) deionized water of known mass and carbonate composition (10 - 30 ml) was added to each bag, which was then resealed after removing air and weighted.

The samples were melted in the dark over ~48 hours. Meltwater was transferred to 12 mL Exetainer vials (Labco, UK) pre-dosed with 20 μl of saturated HgCl_2 solution (5% w/v) to prevent microbial alteration. DIC was measured by on Apollo SciTech®'s AS-C3 analyzer while TA was determined via potentiometric titration on an Apollo SciTech AS-ALK2 total alkalinity titrator (Haraldsson et al., 1997).

2.5 Physical Parameters

Vertical profiles of conductivity, temperature, and depth (CTD) were obtained using a Seabird® SBE19plus CTD. On 16 July 2017, additional surface conductivity measurements were taken using a Thermo Orion-Star® instrument with an Orion 013610MD conductivity cell. Surface water temperatures were independently measured with a Testo® thermometer.

2.6 Historical Data

For contextual comparison, $p\text{CO}_2$ time series data from the Greenland Ecosystem Monitoring program are also included in the analysis. $p\text{CO}_2$ data from 2007-2023 was measured using the same HydroC CO_2 sensor in August each year.

2.7 Eddy Covariance

Air-sea CO_2 fluxes, as well as sensible and latent heat fluxes, were estimated using micrometeorological instrumentation mounted on a 3-meter mast positioned approximately 0.5 meters from the waterline. CO_2 concentrations were measured with a LI-COR® 7500 open-path gas analyzer, and three-dimensional wind vectors were recorded using a METEK® uSonic-Scientific sonic anemometer. The authors recognize that open-path sensors over polar, marine environments can lead to larger errors due to the cross-sensitivity between humidity and CO_2 for near infrared gas analyzers. (Blomquist et al., 2014; Landwehr et al., 2014). However, the open-path analyzer was used at the time due to its low power consumption, making it suitable for operation on battery systems in remote Arctic environments.

To enhance reliability, we applied three complementary analysis techniques for flux estimation: (1) the standard eddy covariance (EC) method using EddyPro software (Version 7.0.6, LI-COR



Inc., 2019); (2) the dissipation technique (DT) (Sørensen and Larsen, 2010); and (3) the ogive optimization method (OGM) (Sievers et al., 2015a). Among these, the OGM was deemed most robust due to its ability identify and filter out low-frequency noise, sensor dampening, and large-scale turbulent motions that can bias flux measurements. These issues often introduce large relative bias associated with flux measurement over surfaces characteristically exhibiting low CO₂ fluxes, such as marine surfaces (Sievers et al., 2015b). OGM's superior ability to isolate relevant turbulent scales and reduce contamination from mesoscale variability is based on the accumulation and modelling of each cospectra over each 20 min averaging period (Fig. S1 and S2). Uncertainty in CO₂ fluxes was estimated directly from the OGM procedure. The reported values correspond to the standard error associated with the fitted ogive tail and reflect random uncertainty in flux integration.

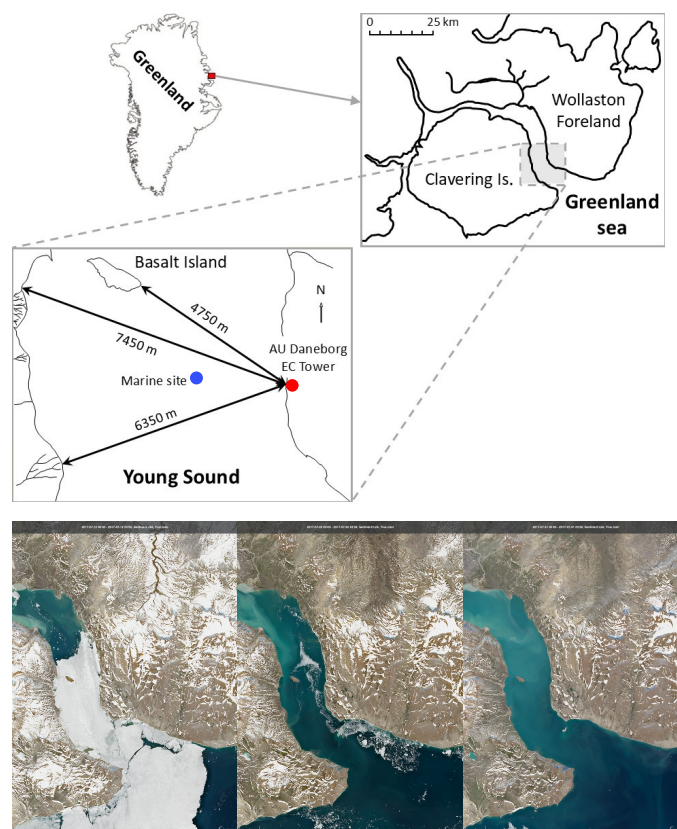


Figure 2. Map of Greenland and the sampling area at the coast of Young Sound in Northeast Greenland. The red circle indicates the location of the Eddy Covariance tower while the Marine sampling site (Standard Station in the Greenland Ecosystem Monitoring program) is indicated as a blue circle (74.310, -20.300). Three Copernicus Sentinel true-color images of the fjord on July 12, 22, and 31 illustrate the transition between sea ice cover and open water.



3 Data and results

3.1 CO₂ and Heat Fluxes

Surface air-sea CO₂ fluxes were measured using micrometeorological techniques between July 16 and July 31, 2017 (Fig. 3a). However, only a limited number of flux estimates passed the quality control criteria defined by OGM. This method uses a Haar wavelet analysis to assess the continuity of high-frequency CO₂ and vertical wind velocity signals, rejecting fluxes when either variable fails to meet spectral continuity thresholds. In addition to the automated filtering, manual inspection of the cospectra was performed to evaluate fluxes that were soft-flagged by the Haar analysis. Only fluxes that passed both stages of evaluation were retained for further analysis and are shown in Fig. 3a (Fig. S1).

Twenty-minute flux averages ranged from -25 to 110 mmol m⁻² day⁻¹, with both upward and downward fluxes observed. Positive values indicate net efflux of CO₂ from the ocean to the atmosphere, implying temporary oversaturation of surface waters with respect to atmospheric CO₂. These elevated fluxes occurred during and shortly after the sea ice breakup period. This finding contrasts with prior studies in Young Sound, which have described the fjord as a net CO₂ sink throughout the year. However, historic estimates are based on pCO₂ measurements from the month of August and taken at 1 m depth; not from vertical pCO₂ profiles that capture salinity gradients. Similar episodic outgassing events have been documented in other Arctic coastal systems under variable sea ice conditions, though particularly during or following sea ice melt (Butterworth et al., 2025; Else et al., 2011; Miller et al., 2011; Papakyriakou and Miller, 2011; Prytherch and Yelland, 2021; Sievers et al., 2015c).

In addition to CO₂ fluxes, eddy covariance measurements of sensible and latent heat fluxes were also quantified during the same period and are presented in Fig. 3b and 3c. For all scalar quantities, negative values represent downward fluxes directed toward the ocean surface. These heat flux data provide important context for interpreting variability in CO₂ exchange, as they reflect changes in atmospheric forcing and surface stratification. Corresponding meteorological variables, including wind speed and air temperature, are shown in Fig. 3d-f.

The calculated flux uncertainties are shown in Fig. S3 and illustrate CO₂ uncertainties were typically below 5 mmol m⁻² d⁻¹. These uncertainties were used to evaluate the robustness of positive efflux events and to interpret flux magnitudes relative to measurement precision. Low uncertainties during both high uptake and efflux events demonstrated a good signal to noise ratio and provide support for the existence of variable uptake and outgassing. The highest uncertainties that exceed 5 mmol m⁻² d⁻¹ corresponded to near-zero fluxes, where precise flux-estimation becomes more difficult.



3.2 Surface Water pCO₂

Vertical profiles of surface water pCO₂ were measured using the CONTROS® HydroC CO₂ sensor across three distinct periods in July 2017 (Fig. 5a-c). Each observational period corresponded to different sea ice conditions: before, during and after sea ice breakup (Fig. 2). These high-resolution profiles revealed substantial vertical variability within the upper 2 to 3 meters of the water column. Under ice-covered conditions, pCO₂ measurements were taken through an open melt pond. At this time, elevated CO₂ concentrations were observed at the very surface (0.1 m), followed by a sharp decrease to approximately 1 meter depth, coinciding with the ice-water interface. Below this depth, pCO₂ increased again, though remained well below atmospheric concentrations (Fig. 4a).

During the period of sea ice breakup, when ice coverage ranged from approximately 30% to 10%, the vertical distribution of pCO₂ exhibited a similar structure. Concentrations were highest near the surface, declined to a local minimum at 1 to 2 meters, and then stabilized below 3 meters (Fig. 4b). Following, the complete breakup of sea ice, pCO₂ showed a more gradual decrease from the surface down to about 3 meters, beneath which concentrations remained relatively constant (Fig. 4c). Across all three observational periods, a shallow surface layer approximately 5 m thick was identified, within which most of the pCO₂ variability occurred. Below this depth, pCO₂ remained relatively constant.

These vertical structures are consistent with strong physical stratification, likely driven by freshwater input from glacial melt and surface heating. Temperature and salinity profiles collected concurrently support the presence of sharp vertical gradients in the upper water column, with salinity ranging from 1.4 to 29.6 PSU and temperature from -0.4°C to 6.2°C. These physical profiles, shown in Fig. 5, confirm that vertical mixing was strongly suppressed during the observational period.

Measurements from a different fjord in East Greenland on June 4, 2025, revealed strikingly similar vertical pCO₂ heterogeneity (Fig. 6). Elevated pCO₂ at 0.1 m decreased to a minimum around 1-1.5 m before increasing again and stabilizing near 3 m depth. Extreme stratification in the upper few meters caused pCO₂ levels in each profile to vary by more than 100 µatm between the surface and 1 m. This repeated observation of comparable vertical pCO₂ heterogeneity 8 years later and in a different fjord system suggest this is not an isolated phenomenon. Indeed, Arctic surface stratification induces chemical changes that may influence the way we estimate air-sea exchange of CO₂.

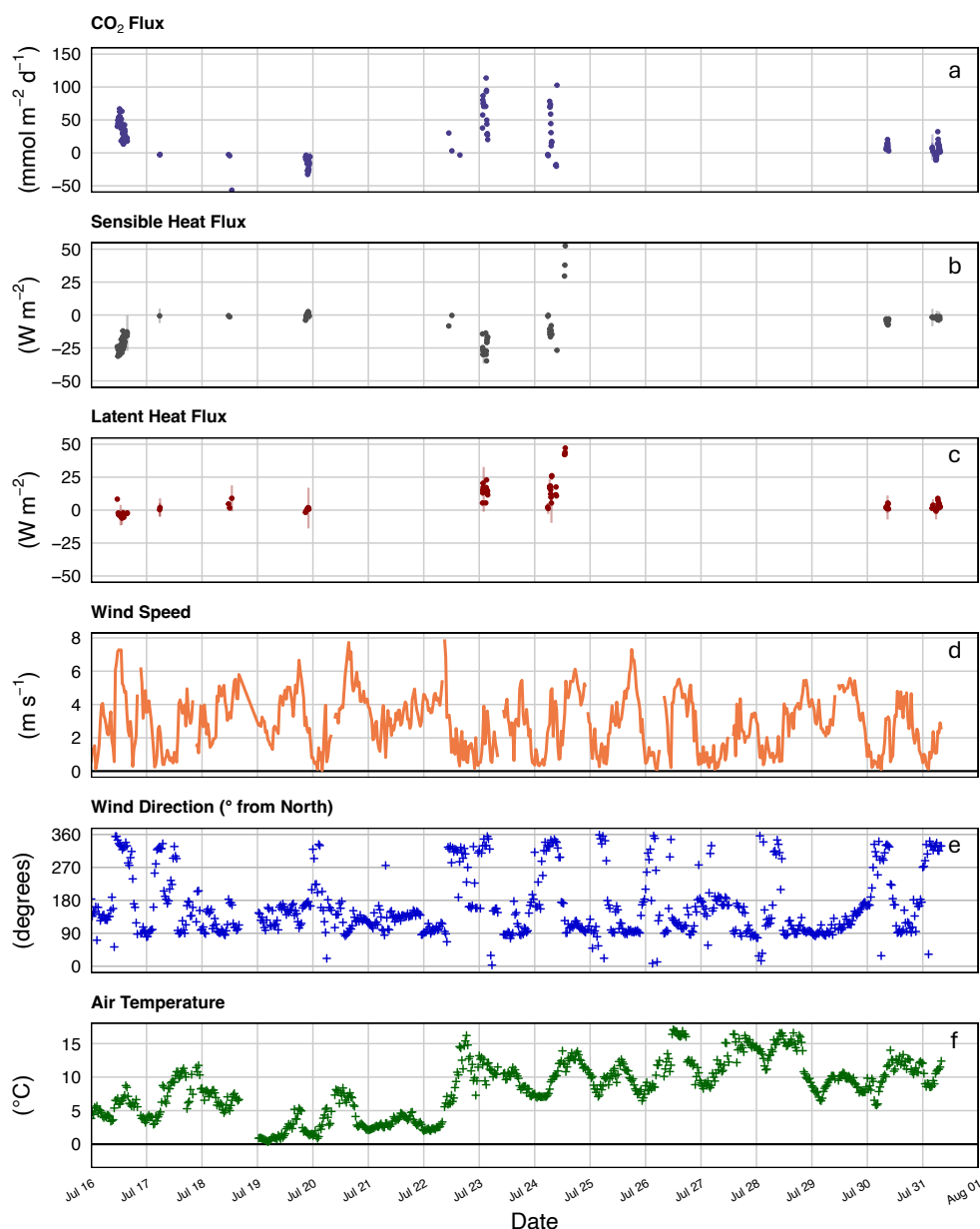
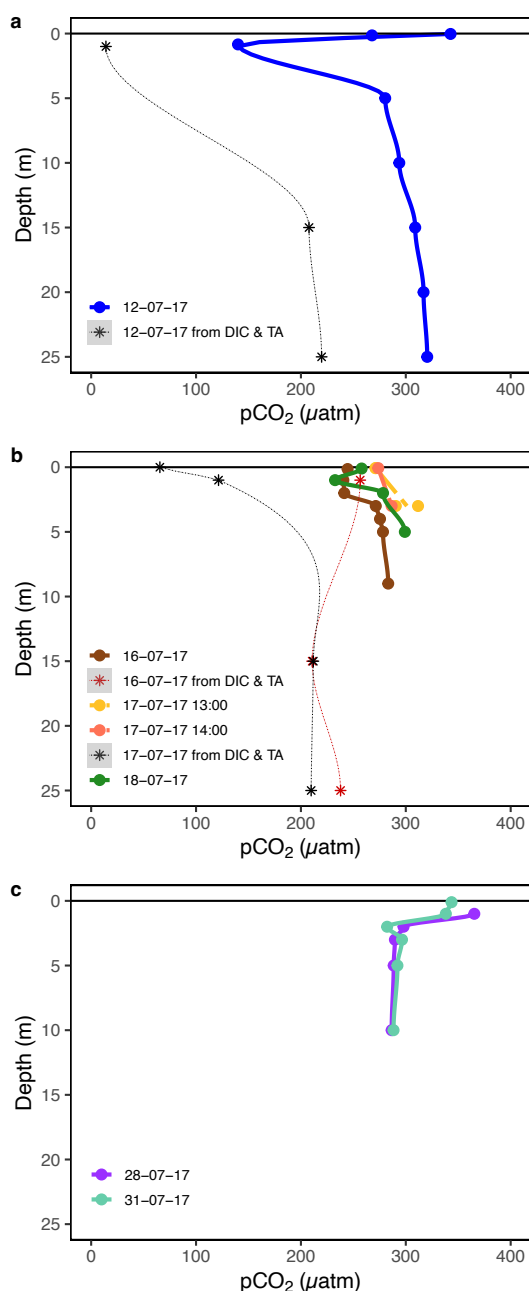


Figure 3. The 5 min averages of measured fluxes and meteorological conditions over Young Sound during July 2017. This time period reflects the transition between sea ice break up (30% ice cover) and open water (no sea ice present) from 16 July 2017 to 1 August 2017. Air-sea exchange of (a) CO₂ (b) sensible heat and (c) latent heat were estimated using the ogive optimization method with estimated uncertainty shown as vertical error bars. (d) Wind speed, (e) wind direction and (f) air temperature are shown for the same period.



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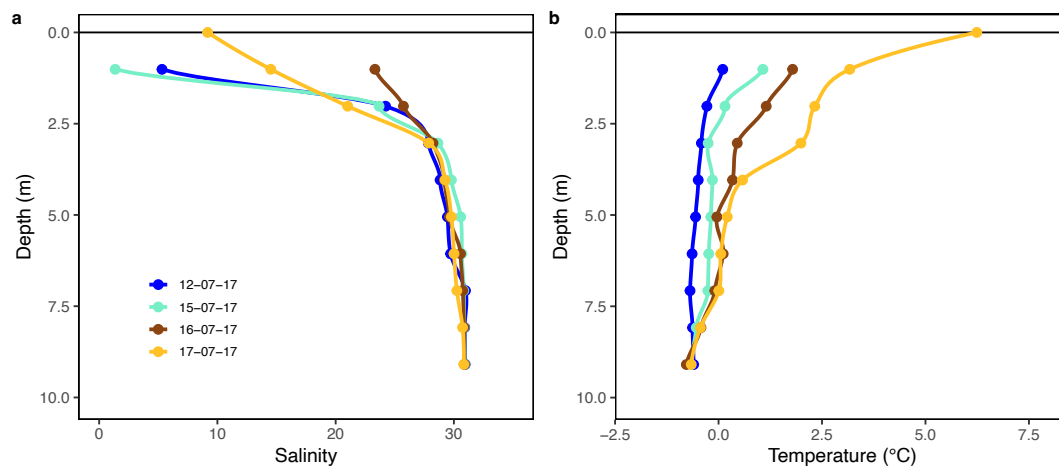
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323 **Figure 4.** Measured Young Sound $p\text{CO}_2$ profiles (a) prior to sea ice breakup (measured through
 324 open melt pond), (b) during sea ice breakup and (c) after sea ice break up measured through CO_2
 325 equilibration and calculation from carbonate chemistry parameters (DIC & TA).



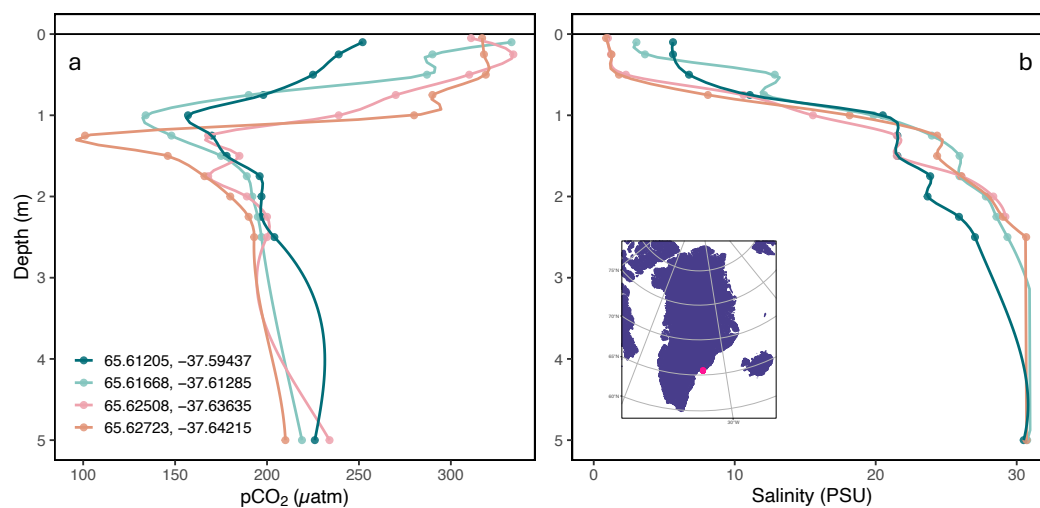
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328

329 **Figure 5.** Measured Young Sound profiles of under-ice water and open water salinity and
 330 temperature.
 331



332

333 **Figure 6.** Measured $p\text{CO}_2$ (a) and salinity (b) profiles at 4 locations in Tasiilaq Bay. Profiles
 334 were measured on June 4, 2025 following the method in Sejr et al. (2011).

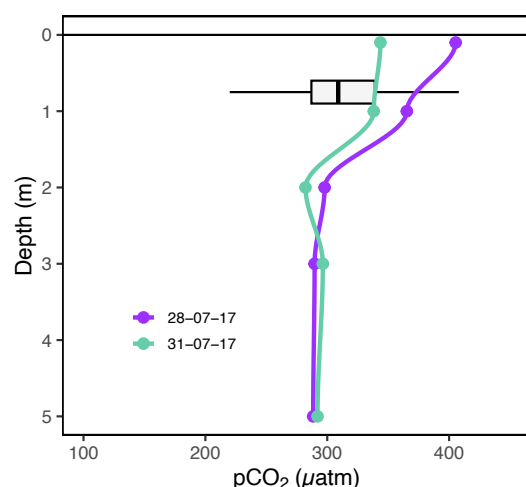


Figure 7. Measured Young Sound $p\text{CO}_2$ profile after ice break up in 2017 compared with historical variation in $p\text{CO}_2$ at 1 m depth in the same location.

4. Discussion

Air-sea CO_2 fluxes in Arctic coastal areas are generally estimated using bulk parameterization models (Henson et al., 2024; Meire et al., 2015; Roobaert et al., 2019; Sejr et al., 2011). These models rely on several key assumptions, including unstratified surface conditions, a linear $p\text{CO}_2$ gradient within the diffusive boundary layer, and a vertically uniform $p\text{CO}_2$ profile within the mixed layer. Our observations challenge the applicability of these assumptions in Arctic coastal waters in several important ways.

4.1. Evaluating the Constant $p\text{CO}_2$ Assumption

Vertical $p\text{CO}_2$ profiles collected during July 2017 revealed pronounced non-linear behavior in the upper 3 to 5 meters of the water column (Fig. 4). This directly contradicts the assumption that the $\Delta p\text{CO}_2$ accurately represents the difference between the atmosphere and the “well-mixed bulk fluid” below the diffusive layer (Wanninkhof et al., 2009). Under ice-covered conditions, the lowest $p\text{CO}_2$ values (~ 150 ppm) were consistently observed just beneath the sea ice, with concentrations increasing with depth and stabilizing around 5 m (Fig. 4a). During the ice breakup stage, a similar pattern emerged, although the minimum $p\text{CO}_2$ was higher (~ 250 ppm).

More recent measurements from Tasiilaq Bay in June 2025 demonstrate very similar vertical $p\text{CO}_2$ profiles. Indeed, 4 high-resolution profiles with measurements every 0.25 m reveal the same C-shaped $p\text{CO}_2$ variation. Like in Young Sound, the most elevated $p\text{CO}_2$ levels were observed near the surface, and $p\text{CO}_2$ minimums occurred near 1-2 meters depth before increasing and becoming stable. This repeated observation in a different fjord system, but during the period



362 of sea-ice breakup indicates this vertical variability may be representative during stratified Arctic
 363 conditions.

364
 365 Several interacting processes influence surface water chemistry during ice breakup. Low surface
 366 water $p\text{CO}_2$ values reflect the influence of low-salinity meltwater from snow and sea ice or
 367 glacial meltwater found in freshened Arctic waters (Geilfus et al., 2015; Henson et al., 2025).
 368 However, surface water chemistry during the ice breakup period is further complicated by
 369 processes such as ikaite ($\text{CaCO}_3 \cdot 6\text{H}_2\text{O}$) dissolution (Miller et al., 2011; Rysgaard et al., 2013;
 370 Sogaard et al., 2013) and high under-ice primary production (Sogaard et al., 2021). Additionally,
 371 snowmelt, characterized by low pH and ionic strength (de Caritat et al., 2005), may further alter
 372 carbonate system dynamics in the upper water column.

373
 374 Two mechanisms may explain the nonlinear C-shaped trend in $p\text{CO}_2$ observed in the top few
 375 meters. First, as demonstrated by Henson et al. (2025) mixing between glacial meltwater and
 376 seawater can result in non-conservative behavior in $p\text{CO}_2$, even when DIC and TA behave
 377 conservatively. In such cases, initial freshwater dilution leads to dramatically reduced $p\text{CO}_2$, but
 378 at very low salinities, the diminished buffering capacity can cause acidification to occur and
 379 $p\text{CO}_2$ to increase again. Although, Henson et al. focused on glacial meltwater, our results suggest
 380 similar processes could occur in systems influenced by sea ice and snowmelt.

381
 382 Both glacial meltwater and sea ice have low DIC concentrations and act to dilute the inorganic
 383 carbon of the surface ocean (Fig. S4). However, changes in alkalinity can also impact the
 384 buffering capacity of the water mixture, leading to nonlinear effects. If the meltwater has a lower
 385 TA:DIC ratio than seawater, due to the absence of ikaite, acidification and a shift in carbonate
 386 equilibria at very low salinities could lead to higher $p\text{CO}_2$ values at the surface. During July
 387 2017, Young Sound showed both diluted DIC and TA levels in upper few meters, suggesting pH
 388 change during sea-ice break up could occur more easily (Fig. S4). Indeed, calculated pH profiles
 389 indicated variable surface conditions between periods of sea ice cover and sea ice breakup (Fig.
 390 S5). In this very fresh surface layer, diminished pH may elevate $p\text{CO}_2$ relative to waters around 1
 391 m depth, where freshwater-seawater mixing ratios are more moderate and seawater buffering
 392 leads to very low CO_2 concentrations.

393
 394 A second, but less likely, explanation involves atmospheric equilibration of sea ice melt ponds
 395 before draining into open leads. The relatively elevated $p\text{CO}_2$ observed at ~ 0.1 m depth could
 396 reflect such partial equilibration. While chamber-based studies (e.g. Geilfus et al., 2012, 2015;
 397 Nomura et al., 2010; Semiletov et al., 2004) have demonstrated both uptake and efflux of CO_2 in
 398 melt ponds, equilibrium times between melt-pond water and atmosphere depend upon pond
 399 depth, wind speed, and carbonate chemistry. For example, a 0.1 m deep pond under low wind
 400 conditions ($\sim 2 \text{ m s}^{-1}$) may reach atmospheric equilibrium in 1-4 days. However, in our case,



401 $p\text{CO}_2$ values calculated from TA and DIC in melt ponds did not indicate equilibrium with the
 402 atmosphere, making this explanation less likely than the freshwater mixing mechanism.

403
 404 Across all conditions, $p\text{CO}_2$ values calculated from TA and DIC using R-Seacarb were
 405 consistently lower than in situ measurements from the CONTROS sensor (Fig. 5b, c). This
 406 discrepancy supports the conclusion that chemical equilibrium is not achieved in the upper 2-4 m
 407 during the melt season in Young Sound. Standard carbonate system calculations rely on
 408 equilibrium constants that represent a set of reversible chemical reactions, assuming *equilibrium*
 409 has been reached for each (Emerson and Hedges, 2008). However, in a dynamic surface
 410 environment influenced by stratification, meltwater input, biological activity, and variable air-sea
 411 exchange, this assumption of a steady state is likely violated, resulting in mismatches between
 412 measured and calculated $p\text{CO}_2$.

413
 414 In addition, the equilibrium constants used in these calculations are typically derived from
 415 laboratory experiments under conditions not representative of Arctic coastal waters. Most
 416 constants were determined at higher salinities and temperatures than those commonly observed
 417 during the Arctic melt season. Yet, temperature strongly impacts these thermodynamic constants
 418 (K_1 and K_2) (Cai et al., 2020). Indeed Sulpis et al. (2020) demonstrated that at temperature below
 419 8°C , the use of constants from Lueker et al. (2000) may lead to underestimation of $p\text{CO}_2$, which
 420 could partially explain the discrepancies observed here.

421
 422 As melt progresses and sea ice recedes, riverine input and vertical mixing become more
 423 influential. Yet even after ice breakup, surface waters often remain fresh, and the resulting low
 424 salinities help maintain stratification. In August 2017, vertical structure remained pronounced,
 425 with elevated $p\text{CO}_2$ at 0.1 m which stabilized below ~ 3 m. In other words, near-surface
 426 conditions remained decoupled from deeper waters. This persistent shallow layer, characterized
 427 by low salinity, higher temperature, and elevated $p\text{CO}_2$, suppresses gas exchange with the colder,
 428 more undersaturated water below, consistent with observations by Dong et al. (2021). In such
 429 environments, bulk flux models that assume homogeneity and linear gradients are likely to yield
 430 biased or inaccurate estimates.

431 To place these 2017 measurements in historical context, we examined long-term surface water
 432 $p\text{CO}_2$ data collected at 0.5-1 m depth by the Greenland Ecosystem Monitoring (GEM) program
 433 between 2007 and 2023. These data, measured using consistent protocols, are presented in Fig. 7
 434 alongside our open-water profiles. Over the 17-year record, August $p\text{CO}_2$ concentrations at ~ 1 m
 435 depth had ranged from 220 to 408 μatm and had consistently remained below atmospheric levels.
 436 This apparent stability has contributed to the perception of sustained CO_2 uptake throughout the
 437 summer season.

438 However, the high-resolution vertical profiles obtained during the 2017 field campaign challenge
 439 this assumption. Elevated $p\text{CO}_2$ levels confined to the uppermost meter of the water column may



go undetected in standard monitoring approaches that rely on fixed-depth sampling. These results suggest that short-lived but significant episodes of CO₂ outgassing can occur during rapid environmental transitions such as sea ice breakup. Consequently, existing sampling protocols may underestimate surface variability and bias flux estimates, especially in stratified conditions where near-surface chemistry is decoupled from subsurface layers.

4.2. Evaluating Bulk Model Flux Estimates

To assess whether bulk models are suitable for estimating CO₂ fluxes in an Arctic fjord influenced by sea ice and snow melt, we calculated fluxes using seawater pCO₂ measurements from multiple depths and two gas transfer velocity parameterizations. Specifically, we computed fluxes throughout July using pCO₂ measured at 0.1, 1, 2, and 4 m. To estimate the surface (interface) pCO₂ at 0 m, we adjusted the 1 m measurements for the thermal skin effect based on skin temperature derived from sensible heat flux data (Fig. 3b), following the parameterization of Smedman et al. (2007). Accounting for this skin layer correction is critical, as Woolf et al. (2016) demonstrated that neglecting the thermal skin and relying only on bulk sea surface temperature can introduce significant errors in flux estimates.

The resulting calculations (Table 1) show that estimated CO₂ fluxes vary significantly depending on the depth of the pCO₂ measurement. Notably, fluxes derived from 0.1 m differ markedly from those based on deeper values. Since many studies rely on pCO₂ measured at a fixed depth (often at 1 m or at a ship's seawater intake below 5 m), these results underscore the potential for misrepresentation of flux direction and magnitude due to vertical heterogeneity in surface water chemistry.

Measured fluxes from eddy covariance (Fig. 3a) also exhibited large temporal variability. While micrometeorological methods integrate fluxes over a horizontal footprint, bulk flux models rely on single point pCO₂ gradients between air and water. Consequently, under heterogeneous or partially ice-covered conditions, the two methods are unlikely to yield identical results.

However, meaningful comparison is still possible. Notably, agreement between micrometeorological and bulk model estimates was only observed in late July, and only when pCO₂ was corrected for the skin temperature. This aligns with the conclusions of Woolf et al. (2016) and Ford et al. (2024), who suggest that using skin-adjusted pCO₂ may improve flux estimates. Accurate application of this correction would require either direct measurements of skin temperature when sampling for the bulk method or high-resolution modeling of heat fluxes.

However, this agreement between methods did not hold before or immediately after ice breakup. During these periods, micrometeorological methods indicated episodes of large upward fluxes (48 mmol m⁻² d⁻¹), while bulk model estimates suggested downward fluxes (-14 mmol m⁻² d⁻¹). Although both methods yielded comparable magnitudes for downward fluxes, only the micrometeorological approach captured large upward events that could not be explained by surface water pCO₂ profiles alone.



481
 482 Measurements from both Young Sound and Tasiilaq demonstrate that during sea-ice breakup,
 483 pCO₂ levels are most elevated at the surface. This may be linked to acidification of the most
 484 freshened 0.5 m and a shift in the marine carbonate system, or partial equilibration due to air-sea
 485 gas transfer. If the air-sea boundary layer warms, for instance, due to solar radiation, pCO₂ may
 486 rise rapidly leading to oversaturation relative to atmospheric concentrations. Indeed, when pCO₂
 487 measurements on July 31 were corrected for skin temperature (to estimate pCO₂ at the boundary
 488 layer), they revealed a transition from undersaturation to oversaturation (Table 1). While we did
 489 not directly observe this oversaturation in the vertical profiles, this likely reflects the inability to
 490 sample at the very surface layer. Moreover, profile measurements represent only single points in
 491 a system characterized by strong spatial and temporal variability during this seasonal transition.
 492 Nevertheless, the occurrence of C-shaped pCO₂ profiles during sea-ice breakup may help explain
 493 the observed reversal of flux direction captured by the micrometeorological approach.

494
 495 Another potential explanation for the elevated CO₂ efflux that cannot be fully discounted
 496 involves cross-sensitivity between water vapor and CO₂ in open-path NDIR sensors (Blomquist
 497 et al., 2014). Although the OGM processing method is designed to minimize humidity-induced
 498 artifacts, it cannot entirely remove them. However, several lines of evidence indicate that the
 499 positive CO₂ fluxes are not solely an artifact of water vapor. Elevated CO₂ concentrations
 500 occurred during some, but not all, periods of high relative humidity, and extended intervals with
 501 similarly high humidity exhibited low CO₂ levels (Fig. S6). Moreover, positive CO₂ fluxes were
 502 observed even during negative latent heat flux events (Fig. S7a), which is inconsistent with a
 503 purely humidity-driven bias. In addition, CO₂ fluxes exhibited a strong negative relationship with
 504 sensible heat flux (Fig. S7b), consistent with physical air-sea exchange processes in marine
 505 environments. Taken together, these patterns support the interpretation that the observed effluxes
 506 represent real air-sea CO₂ exchange rather than being dominated by cross-sensitivity artifacts.
 507 Still, future studies using closed-path sensors during Arctic seasonal transitions are encouraged
 508 to verify the precise magnitude of CO₂ outgassing and uptake.

509
 510 Overall, these findings echo those of Miller et al. (2019), who reported pronounced spatial
 511 heterogeneity in Arctic coastal pCO₂ and large differences in estimated flux depending on the
 512 sampling depth. The broader implications of this heterogeneity for seasonal or regional flux
 513 estimates remain unclear. However, if fluxes are upscaled from sparse, single-point
 514 measurements (e.g., once per month, as in Laruelle et al., 2014), substantial errors may result due
 515 to unrecognized spatial and temporal variability. Thus, our results emphasize the need for
 516 continuous, high-resolution observations of air-sea CO₂ fluxes, particularly in Arctic coastal
 517 systems affected by stratification and meltwater input. These observations are essential for
 518 refining flux parameterizations, reducing uncertainty in carbon budget estimates, and improving
 519 the representation of Arctic shelf systems in global carbon models.



Table 1. CO₂ fluxes calculated based on pCO₂ measured at the different depth. The fluxes are calculated using the bulk model of Ho et al., 2006 and Nightingale et al. (2000). We have used locally measured wind speeds for the calculations to match flux measurements captured by eddy covariance. The pCO₂ at 0 m is calculated from the pCO₂ measured at 1 m and then adjusted to the skin temperature, which is derived from the bulk equation and the measured heat fluxes.

Date	Depth (m)	Temperature (°C)	Salinity (psu)	pCO ₂ (µatm)	Wind Speed (m s ⁻¹)	Ho (2006)	Nightingale (2000)	Eddy Covariance
						Flux (mmol CO ₂ m ⁻² day ⁻¹)	Flux (mmol CO ₂ m ⁻² day ⁻¹)	Flux (mmol CO ₂ m ⁻² day ⁻¹)
16-Jul	0.0	3.0	23	252	6.8	-14.88	-13.69	48.3
16-Jul	0.1	3.0	23	244	6.8	-15.78	-14.52	
16-Jul	1.0	1.8	26	240	6.8	-16.12	-14.83	
16-Jul	2.0	1.1	28	241	6.8	-15.93	-14.66	
16-Jul	4.0	0.3	29	275	6.8	-12.19	-11.21	
18-Jul	0.0	6.0	7	262	3.3	-3.45	-3.38	-3.49
18-Jul	0.1	4.3	7	244	3.3	-3.98	-3.90	
18-Jul	1.0	3.2	14	233	3.3	-4.20	-4.11	
18-Jul	2.0	2.3	21	278	3.3	-2.86	-2.79	
18-Jul	4.0	0.6	29	295	3.3	-2.34	-2.29	
28-Jul	0.0	10.0	15	415	2.5	0.53	0.54	—
28-Jul	0.1	10.0	15	405	2.5	0.38	0.38	
28-Jul	1.0	7.0	21	365	2.5	-0.22	-0.23	
28-Jul	2.0	5.0	27	282	2.5	-1.43	-1.45	
28-Jul	4.0	2.0	30	290	2.5	-1.32	-1.34	
31-Jul	0.0	12.0	15	401	2.0	0.20	0.21	5.07
31-Jul	0.1	10.0	15	343	2.0	-0.36	-0.38	
31-Jul	1.0	8.0	21	338	2.0	-0.40	-0.42	
31-Jul	2.0	5.0	27	282	2.0	-0.92	-0.97	
31-Jul	4.0	2.0	30	294	2.0	-0.81	-0.85	



5 Conclusions

During the summer thaw, carbon chemistry and $p\text{CO}_2$ dynamics in Arctic coastal surface waters are significantly altered by the combined effects of snow and sea ice melt, terrestrial runoff, and biological activity. These influences lead to substantial variability in surface temperature, pH, dissolved inorganic carbon (DIC), and total alkalinity (TA), ultimately disrupting carbonate system equilibrium in the upper water column. As a result, estimating air-sea CO_2 fluxes using traditional bulk models becomes highly uncertain during this period.

The sea ice breakup period, typically lasting 2-4 weeks, represents a particularly dynamic and complex phase in the annual cycle. Despite its brevity, this phase may have a disproportionate influence on total summer CO_2 uptake, given that open-water conditions in high Arctic fjords are limited to only 80-120 days per year (Sejr et al., 2011).

Improved flux estimates will require more detailed and spatially resolved investigations aimed at developing and validating gas exchange parameterizations tailored to the highly stratified and ice-affected conditions of Arctic fjords. In particular, new approaches are needed to estimate gas transfer velocities over waters influenced by snow and sea ice melt. Exchange rates depend not only on the $p\text{CO}_2$ gradient between the atmosphere and surface water, but also on rapid, nonlinear changes in surface water chemistry driven by the composition and volume of meltwater and runoff.

Once more suitable parameterizations for gas transfer velocity are established, accurate flux estimation will also require knowledge of the depth at which surface water $p\text{CO}_2$ becomes vertically homogeneous. Profiling $p\text{CO}_2$ in the upper water column is therefore essential to identify this depth and to constrain surface flux estimates reliably.

Our study revealed large upward CO_2 fluxes using the eddy covariance (OGM) method, particularly during the ice breakup period – fluxes that were not captured by the bulk model. These events underscore the need for studies that integrate continuous, direct CO_2 flux measurements with detailed observations of surface water carbonate chemistry, atmospheric forcing, skin temperature, and turbulence at the air-ice-water interface.

Such integrated measurements are critical to improving our understanding of the frequency, drivers, and net effect of upward CO_2 fluxes in Arctic coastal systems. Ultimately, this knowledge is essential to accurately quantify the seasonal and regional uptake of atmospheric CO_2 in the rapidly changing Arctic.



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Author Contribution

Conceptualization: LLS. Formal analysis, writing – original draft preparation: HCH. Funding acquisition: LLS, SR, MKS, TP. Investigation: DS, BJ, KL, TP, MKS, JS, SR, LLS. Writing – review and editing: DS, TP, MKS, SR, LLS. All the authors have read and agreed to the published version of the paper.

Data Availability Statement

Vertical profiles from both Greenlandic fjords can be found in the Zenodo data repository:
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Competing interests

The authors declare no competing interests.



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