

Sea ice melt drives vertical pCO₂ variability modulating air-sea gas exchange

Henry C. Henson^{1,2}, Dorte H. Søgaard^{2,3,7}, Bjarne Jensen⁶, Kunuk Lennert⁴, Tim Papakyriakou⁵, Mikael K. Sejr^{1,2}, Jakob Sievers⁶, Søren Rysgaard^{2,7}, and Lise Lotte Sørensen^{2,6}

¹Department of Ecoscience, Aarhus University, Aarhus, 8000, Denmark

²Arctic Research Center, Aarhus University, Aarhus, 8000, Denmark

³Greenland Climate Research Cluster, Greenland Institute of Natural Resources, Nuuk, 3900, Greenland

⁴UiT, The Arctic University of Norway, Tromsø, 9037, Norway

⁵Centre for Earth Observation Science, University of Manitoba, Winnipeg, MB, R3T 2N2, Canada

⁶Department of Environmental science, Aarhus University, Roskilde, 4000, Denmark

⁷Department of Biology, Center for Ice-free Arctic Research, Aarhus University, Aarhus, 8000, Denmark

Corresponding author: Henry C. Henson (hch@ecos.au.dk)

Key Points:

- Spring melting of sea ice and snow introduces distinct heterogeneity in surface water conditions within coastal Arctic oceans.
- Standard bulk parameterizations for air-sea CO₂ flux calculations, based on subsurface pCO₂ measurements, may misrepresent flux ~~direction and~~ magnitude during melt periods.
- Vertical near-surface ~~temperature and~~ CO₂ and temperature gradients must be considered to improve flux estimates in stratified Arctic fjords.

30 Abstract

31 Strong spatial and temporal gradients in salinity, temperature, and carbonate chemistry in Arctic
32 coastal surface waters complicate the estimation of air-sea carbon dioxide (CO₂) exchange,
33 particularly during sea ice breakup. ~~Our~~ The present study evaluates the applicability of the
34 widely used bulk flux model under such conditions. This approach assumes homogeneous
35 surface conditions and no vertical pCO₂ gradients in the bulk seawater. However, our
36 observations in a stratified Arctic fjord reveal pronounced vertical variability in pCO₂ within the
37 upper water column, including non-linear gradients near the air-sea interface. This results in,
38 widely varying flux estimates depending upon the depth of the pCO₂ measurement used to
39 establish air-sea disequilibrium. Micrometeorological flux estimates suggested variability in the
40 direction of gas transfer, even when waters at 1 m depth and below were CO₂-undersaturated.
41 We hypothesize that transient, high-pCO₂ layers within the upper decimeters intermittently
42 decouple the atmospheric exchange from subsurface waters and have the potential to reverse the
43 expected flux direction. These findings highlight the importance of resolving near-surface
44 variability during the transition from ice-covered to open water conditions. We recommend
45 incorporating both micrometeorological techniques and high-resolution vertical profiling in
46 Arctic fjords to improve flux estimates of CO₂ in this rapidly changing region.

47

48 Plain Language Summary

49 Sea ice melt adds less-saline water to the surface ocean. This creates vertical gradients in
50 salinity, temperature, and partial pressures of carbon dioxide (pCO₂). The concentration
51 difference of pCO₂ across the air-ocean boundary is used to estimate gas transfer. Thus, the
52 depth that we measure will impact our estimates. Directly measuring gas transfer showed both
53 CO₂ uptake and release from the ocean during sea ice breakup. This means ocean layering during
54 ice melt may slow or briefly reverse the direction of CO₂ transfer.

55

56 1 Introduction

57 High latitude coastal oceans are strong sinks for atmospheric carbon dioxide (CO₂), absorbing
58 more CO₂ per unit area than lower latitude regions (Dai et al., 2022; Roobaert et al., 2019). This
59 strong uptake results from both the high solubility of gases in cold water and the intense
60 biological activity typical of these regions. However, climate change is rapidly transforming this
61 carbon sink. The Arctic is warming more than twice as fast as the global average, and sea ice
62 extent has been shrinking by over 13% per decade (Perovich et al., 2020). The loss of sea ice
63 increases CO₂ uptake by exposing larger areas of open water for longer periods, which can
64 further stimulate biological productivity (Arrigo and van Dijken, 2015; Bates and Mathis, 2009;
65 Perovich et al., 2020). However, at the same time, melting sea ice freshens the surface layer and
66 strengthens stratification, limiting vertical mixing with deeper water. Freshwater from melting
67 sea ice and terrestrial run-off creates pronounced gradients in physical properties such as salinity
68 and temperature, as well as chemical properties like dissolved inorganic carbon (DIC) and total
69 alkalinity (TA) (e.g. Henson et al., 2025). As a result, the partial pressure of CO₂ (pCO₂) can

70 vary markedly with depth under melt conditions (Miller et al., 2019).

71

72 This vertical variability in pCO_2 poses a challenge for air-sea CO_2 flux estimation. The transfer
73 of gases between the atmosphere and ocean depends on the difference in concentration between
74 the two as well as the efficiency of the transfer process. Therefore, the bulk flux of CO_2 across
75 the air-sea interface is commonly described as the product of the gas transfer velocity, k ($m\ s^{-1}$),
76 CO_2 solubility s ($mol\ kg^{-1}\ atm^{-1}$), and the partial pressure gradient (μatm) across the air-sea
77 interface (Wanninkhof et al., 2009):

78

$$79 \quad F = ks(pCO_{2_{sea}} - pCO_{2_{air}}) \quad (1)$$

80

81 While widely applied, this formulation simplifies a complex process influenced by surfactants on
82 the water surface, bubble-mediated gas exchange, and turbulence (Wanninkhof et al., 2009).
83 Furthermore, surface water heterogeneity, driven by sea ice melt and freshwater runoff from
84 land, complicate the physical and chemical processes governing air-sea CO_2 exchange. As a
85 result, simplified parameterizations commonly used in global carbon flux estimates may be
86 inadequate in these settings.

87

88 In most studies, pCO_2 is measured several meters below the surface, assuming vertical
89 homogeneity under well-mixed condition (Jørgensen et al., 2020). However, in stratified waters,
90 where temperature, salinity, and pH can vary with depth, this assumption may lead to substantial
91 errors in flux estimates (Ahmed et al., 2020; Dong et al., 2021; Miller et al., 2019; Watts et al.,
92 2022). Although Arctic surface waters are often undersaturated with respect to atmospheric CO_2
93 levels and act as CO_2 sinks (e.g., Burgers et al., 2017; Dai et al., 2022; Henson et al., 2024;
94 Laruelle et al., 2014; Roobaert et al., 2019), such assessments typically rely on sparse data
95 collected from 0.5-5 m depth during limited periods. Dong et al. (2021) illustrate that high
96 latitude fluxes of CO_2 calculated using the bulk method (based on measurements sampled at 6 m
97 depth) differ significantly from those measured using direct eddy covariance in sea ice melt
98 regions.

99

100 Gas transfer velocity (k) is often parameterized as a function of wind speed. However, the true
101 driver is mixing in the surface waters, which governs k . Fick's first law of diffusion, which
102 underlies Equation (1), assumes a linear concentration gradient within the diffusive sublayer
103 (Fig. 1) and steady-state conditions (Garbe et al., 2014). Jørgensen et al. (2020) argued that, due
104 to seawater's high buffer capacity, chemical gradients do not significantly affect CO_2
105 equilibration, supporting the use of measurements at 3-4 m depth. However, this conclusion
106 relies on the assumption of horizontal and vertical homogeneity and neglects the effects of
107 shallow surface stratification, particularly when alkalinity dilution is involved.

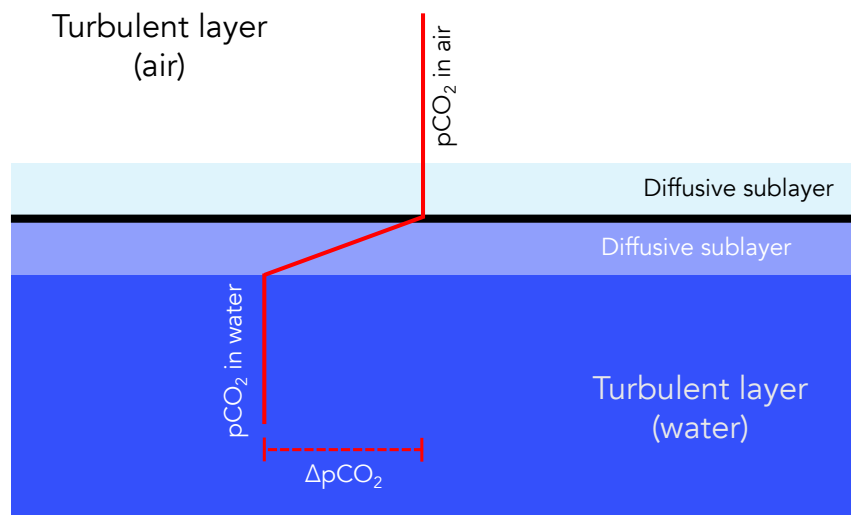
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109 In Arctic spring, the upper ocean is often strongly stratified due to freshwater input from glacier
110 melt, snowmelt, river runoff, and sea ice meltwater (Ahmed et al., 2020; Granskog et al., 2011;
111 Meire et al., 2017; Miller et al., 2019). These inputs can extend vertical CO_2 gradients beyond
112 the diffusive sublayer, complicating flux estimates during ice break-up and early open-water
113 periods. Several studies have demonstrated strong vertical heterogeneity in pCO_2 in Arctic
114 coastal waters, with implications for air-sea flux calculations (Ahmed et al., 2020; Dong et al.,
115 2021; Miller et al., 2019).

116 Surface freshening from ice melt and runoff strongly influences carbonate chemistry in Arctic
117 coastal waters, which can either suppress or enhance oceanic CO_2 uptake. For example, Burgers
118 et al. (2017) reported large horizontal variability in surface pCO_2 (144–364 μatm) linked to
119 riverine input in the Eastern Canadian Arctic. Similarly, Sejr et al. (2011) observed strong
120 surface pCO_2 gradients associated with salinity and temperature in Young Sound, and later
121 documented a long-term decline in surface salinity (Sejr et al., 2017). Freshwater-induced
122 stratification has also been shown to create vertical gradients in pCO_2 and pH with important
123 implications for flux calculations (Miller et al., 2019). Finally, Bates et al. (2014) demonstrated
124 that sea ice meltwater and melt ponds exhibit extreme variability in pCO_2 (<10 to >1500 μatm)
125 and pH (6.1 to >10.8), highlighting the complex chemical landscape of ice-influenced waters.
126 Together, these studies underscore the high spatial and temporal variability of carbonate
127 chemistry in freshened waters across the Arctic.

128 To project future CO_2 uptake or outgassing in the Arctic, we must better understand the physical
129 and chemical drivers of near-surface carbonate variability. In this study, we investigate the
130 vertical and temporal variations in pCO_2 in a stratified Arctic fjord during sea ice breakup. By
131 ~~combining-examining~~ water-column pCO_2 profiles ~~with micrometeorological flux measurements~~
132 across the transition from ice-covered to open water, we evaluate ~~the applicability~~usage of the
133 bulk flux model under Arctic seasonal transitions.

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135

136 **Figure 1.** Schematic illustrating the interface between the air and the water in conjunction with
137 pCO₂ concentration gradients. In equation 1, the concentration gradient is assumed to occur in the
138 diffusive layer between the air and water, and the concentrations are assumed to be vertically
139 constant in the turbulent layers. (Adapted from Liss and Slater, 1974; Wanninkhof et al., 2009)

141 2. Study Site and Measurement Methods

143 2.1 Study Site

144 This study was conducted in Young Sound, a high Arctic fjord system located near the Daneborg
145 Research Station in Northeast Greenland (Fig. 2). The fjord system comprises the Tyrolerfjord
146 (inner fjord) and Young Sound (outer fjord), extending approximately 90 km from Tyroler River
147 to the Greenland Sea. A sill at about 45 m depth separates Young Sound from the open ocean.
148 Young Sound is 2 to 7 km wide, with an average depth of 100 m (maximum 350 m), and a total
149 surface area of ~390 km². Tidal amplitudes range from 0.8 to 1.5 m, with mean current velocities
150 of approximately 2 cm s⁻¹ (Rysgaard et al., 2003). Freshwater inputs are primarily derived from
151 Greenland Ice Sheet runoff, local glaciers, precipitation, and snowmelt from adjacent ice-free
152 terrain. The drainage basin of the Tyrolerfjord/Young Sound system spans 2846 km², of which
153 33% is glaciated.

154
155 Sampling was conducted from 12 to 31 July 2017. Sampling occurred during and immediately
156 after a period of sea ice breakup. On 15 July, ice coverage was approximately 30%, decreasing to
157 less than 10% by 16 July. Water sampling was conducted both from an inflatable boat and via
158 sea ice leads, all in close proximity to the Greenland Ecosystem Monitoring (GEM) program's
159 standard station (Fig. 2).

161 2.2 pCO₂ Measurements Using the HydroC Sensor

162 Surface water pCO₂ was measured with a CONTROS® HydroC CO₂ sensor, which utilizes a
163 membrane equilibrator coupled with a non-dispersive infrared detector. The instrument is
164 equipped with a built-in water pump that provides flow rate of 35 ml s⁻¹ across the membrane. At
165 each sampling depth, the sensor was allowed to equilibrate for 10 to 20 minutes, and values were
166 recorded once stable for at least two minutes. The sensor operates over a range of 200-1000 μatm
167 and temperatures of -2 to 35°C. Annual calibration has been conducted using a certified 400 ±
168 2% ppm CO₂ gas that was traceable to WMO standards. The sensor showed remarkable stability
169 (397-401 ppm), supporting a measurement uncertainty of ± 2 μatm.

171 2.3 pCO₂ Estimation from TA and DIC

172 In addition to direct measurements, pCO₂ was calculated from total alkalinity (TA) and dissolved
173 inorganic carbon (DIC) using the Seacarb package (Gattuso et al., 2024) in R. Due to the low
174 salinity and cold temperatures characteristic of Arctic coastal waters, no universally accepted set
175 of equilibrium constants (K1 and K2) exists. For consistency with previous studies in the region
176 (Henson et al., 2023), we used the refitted constants from (Lueker et al., 2000). The selection of

177 equilibrium constants introduces assumptions regarding seawater composition. (Raimondi et al.,
178 2019) showed that different constants can lead to discrepancies between measured and calculated
179 pCO₂ values, ranging from -3.1 to -35.8 μatm, with Lueker et al. (2000) demonstrating the best
180 internal consistency under polar conditions. Still, (Sulpis et al., 2020) found that the calculation
181 of pCO₂ from DIC and TA can lead to uncertainty up to 15% under cold conditions, which is far
182 greater than when pCO₂ is measured directly.

183

184 **2.4 Sea Ice TA and DIC Sampling**

185 TA and DIC in sea ice were assessed using three ice cores. Each core was sectioned into 5-10 cm
186 segments and sealed in gas-tight NEN/PE bags with sampling valves (Hansen et al., 2000).
187 Samples were transported in thermally insulated boxes to a nearby field laboratory. Cold (1°C)
188 deionized water of known mass and carbonate composition (10 - 30 ml) was added to each bag,
189 which was then resealed after removing air and weighted.

190

191 The samples were melted in the dark over ~48 hours. Meltwater was transferred to 12 mL
192 Exetainer vials (Labco, UK) pre-dosed with 20 μl of saturated HgCl₂ solution (5% w/v) to
193 prevent microbial alteration. DIC was measured by on Apollo SciTech®'s AS-C3 analyzer while
194 TA was determined via potentiometric titration on an Apollo SciTech AS-ALK2 total alkalinity
195 titrator (Haraldsson et al., 1997).

196

197 **2.5 Physical Parameters**

198 Vertical profiles of conductivity, temperature, and depth (CTD) were obtained using a Seabird®
199 SBE19plus CTD. On 16 July 2017, additional surface conductivity measurements were taken
200 using a Thermo Orion-Star® instrument with an Orion 013610MD conductivity cell. Surface
201 water temperatures were independently measured with a Testo® thermometer.

202

203 **2.6 Historical Data**

204 For contextual comparison, pCO₂ time series data from the Greenland Ecosystem Monitoring
205 program are also included in the analysis. pCO₂ data from 2007-2023 was measured using the
206 same HydroC CO₂ sensor in August each year.

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208 **2.7 Eddy Covariance**

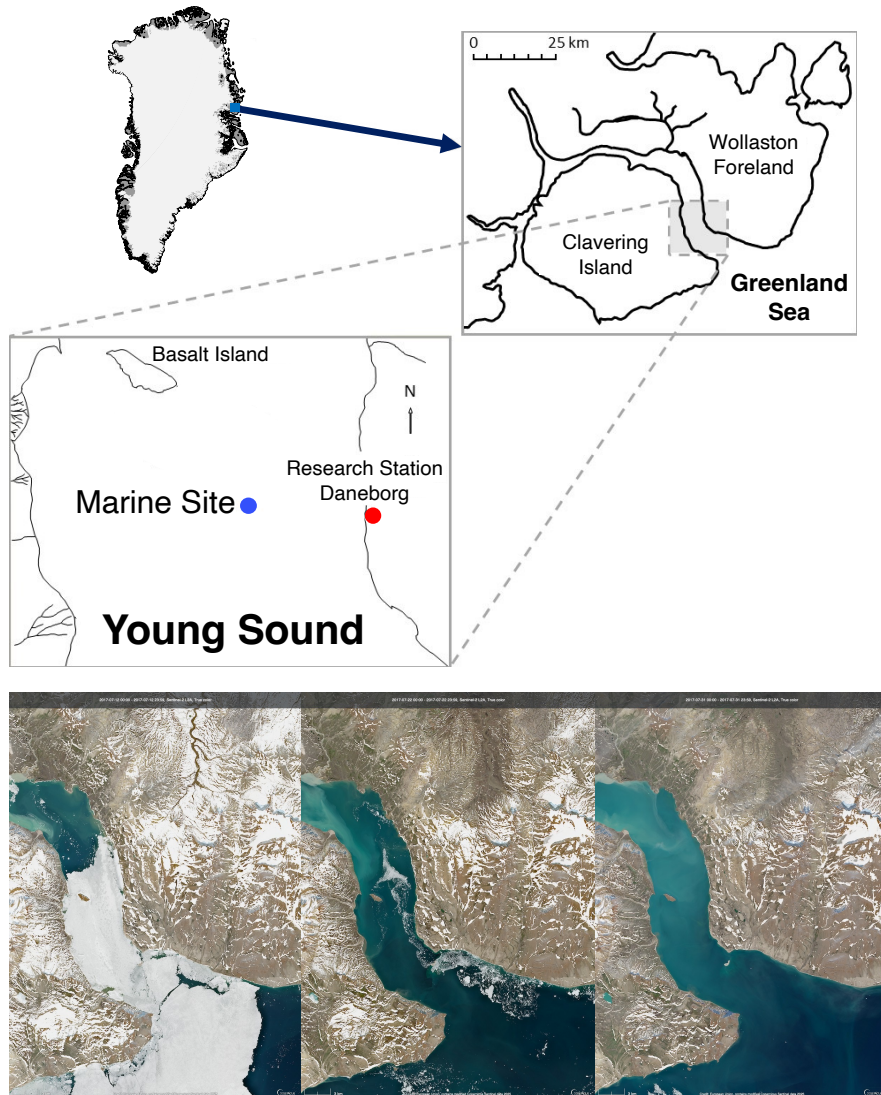
209 ~~Air-sea CO₂ fluxes, as well as s~~Sensible and latent heat fluxes, were estimated using
210 micrometeorological instrumentation mounted on a 3-meter mast positioned approximately 0.5
211 meters from the waterline. ~~CO₂ concentrations were measured with a LI-COR® 7500 open-path~~
212 ~~gas analyzer, and t~~Three-dimensional wind vectors were recorded using a METEK® uSonic-
213 Scientific sonic anemometer. ~~The open-path analyzer was used at the time due to its low power~~
214 ~~consumption, suitable for operation on battery systems in remote Arctic environments.~~
215 ~~However, the authors recognize that open-path sensors over polar, marine environments can lead~~
216 ~~to larger errors due to the cross-sensitivity between humidity and CO₂ for near infrared gas~~

217 ~~analyzers (Blomquist et al., 2014; Landwehr et al., 2014). As a result, we do not aim to precisely~~
218 ~~constrain CO₂ flux magnitudes throughout the study but provide support for the profiles~~
219 ~~measured throughout this seasonal transition.~~

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222 To enhance reliability, we applied complementary analysis techniques for flux estimation: (1) the
223 standard eddy covariance (EC) method using EddyPro software (Version 7.0.6, LI-COR Inc.,
224 2019); (2) the ogive optimization method (OGM) (Sievers et al., 2015a). Among these, the OGM
225 was deemed most robust due to its ability identify and filter out low-frequency noise, sensor
226 dampening, and large-scale turbulent motions that can bias flux measurements. These issues
227 often introduce large relative bias associated with flux measurement over ~~surfaces~~
228 ~~characteristically exhibiting low CO₂ fluxes, such as Arctic~~ marine surfaces (Sievers et al.,
229 2015b). OGM's superior ability to isolate relevant turbulent scales and reduce contamination
230 from mesoscale variability is based on the accumulation and modelling of each cospectra over
231 each 20 min averaging period (Fig. S1 and S2). Uncertainty in ~~CO₂-sensible and latent heat~~
232 fluxes was estimated directly from the OGM procedure. The reported values correspond to the
233 standard error associated with the fitted ogive tail and reflect random uncertainty in flux
234 integration.

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238 **Figure 2.** Map of Greenland and the sampling area at the coast of Young Sound in Northeast
 239 Greenland. The red circle indicates the location of the Eddy Covariance tower
 240 micrometeorological measurement tower at research station Daneborg while the mMarine
 241 sampling site (Standard Station in the Greenland Ecosystem Monitoring program) is indicated as
 242 a blue circle (74.310, -20.300). Three Copernicus Sentinel true-color images of the fjord on July
 243 12, 22, and 31 illustrate the transition between sea ice cover and open water.

244

245 **3 Data and results**

246

247 **3.1 CO₂ and Heat Fluxes**

248 Surface air-sea CO₂ fluxes were measured using micrometeorological techniques between July
 249 16 and July 31, 2017 (Fig. 3a). However, only a limited number of flux estimates passed the
 250 quality control criteria defined by OGM. This method uses a Haar wavelet analysis to assess the

251 continuity of high-frequency CO₂ and vertical wind velocity signals, rejecting fluxes when either
252 variable fails to meet spectral continuity thresholds. In addition to the automated filtering,
253 manual inspection of the cospectra was performed to evaluate fluxes that were soft-flagged by
254 the Haar analysis. Only fluxes that passed both stages of evaluation were retained for further
255 analysis and are shown in Fig. 3a (Fig. S1).

256
257 Fluxes measured using eddy covariance (EC) were highly variable throughout the period of sea
258 ice melt, exhibiting both upward and downward fluxes. Positive values indicate net efflux of CO₂
259 from the ocean to the atmosphere, implying temporary oversaturation of surface waters with
260 respect to atmospheric CO₂. EC-based effluxes were observed during and shortly after the sea
261 ice breakup period. These estimates contrast with prior studies in Young Sound, which describe
262 the fjord as a net CO₂ sink throughout the year (Sejr et al., 2011). However, historic estimates are
263 based on pCO₂ measurements from the month of August and taken at 1 m depth; not from
264 vertical pCO₂ profiles that capture salinity gradients. Similar episodic outgassing events have
265 been documented in other Arctic coastal systems under variable sea ice conditions, though
266 particularly during or following sea ice melt (Butterworth et al., 2025; Else et al., 2011; Miller et
267 al., 2011; Papakyriakou and Miller, 2011; Prytherch and Yelland, 2021; Sievers et al., 2015c).

268
269 In addition to CO₂ fluxes, eddy covariance measurements of sensible and latent heat fluxes were
270 also quantified during the same period and are presented in Fig. 3b and 3c. For all scalar
271 quantities, negative values represent downward fluxes directed toward the ocean surface. These
272 heat flux data provide important context for interpreting variability in CO₂ exchange, as they
273 reflect changes in atmospheric forcing and surface stratification. Corresponding meteorological
274 variables, including wind speed and air temperature, are shown in Fig. 3d-f.

275
276 The flux uncertainties shown in Fig. S3 quantify random uncertainty from the OGM integration
277 procedure, and illustrate CO₂ uncertainties were typically below 5 mmol m⁻² d⁻¹. Low
278 uncertainties during both uptake and efflux events demonstrated a good signal to noise ratio and
279 provide support for the integration of fluxes during these variable conditions. The highest
280 uncertainties that exceed 5 mmol m⁻² d⁻¹ corresponded to near-zero fluxes, where precise flux-
281 estimation becomes more difficult. However, these calculated uncertainties do not account for
282 potential systematic bias related to water vapor cross-sensitivity, which must be evaluated
283 through other metrics.

286 3.2 Surface Water pCO₂

287 Vertical profiles of surface water pCO₂ were measured using the CONTROS® HydroC CO₂
288 sensor across three distinct periods in July 2017 (Fig. 35a-c). Each observational period
289 corresponded to different sea ice conditions: before, during and after sea ice breakup (Fig. 2).
290 These high-resolution profiles revealed substantial vertical variability within the upper 2 to 3

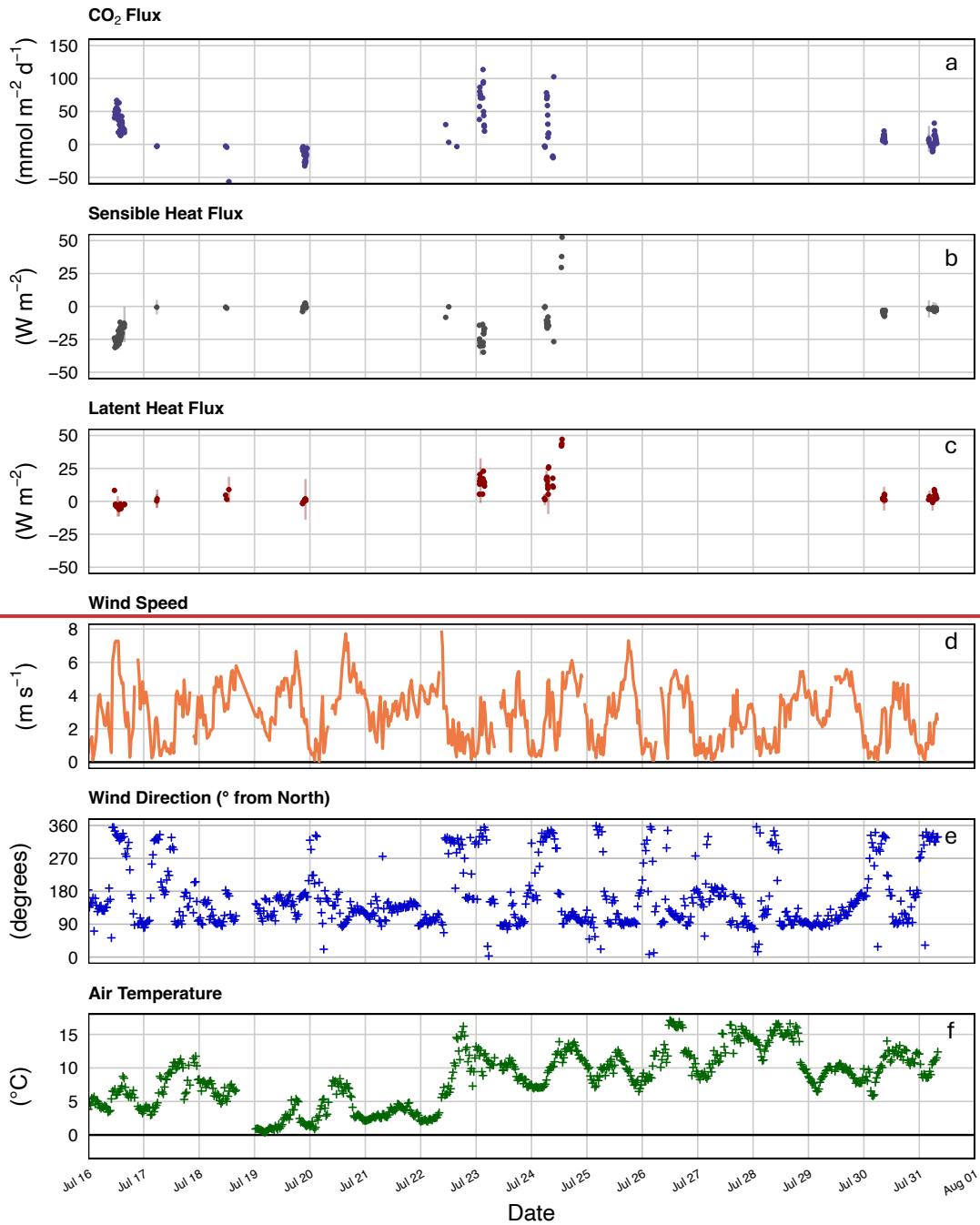
291 meters of the water column. Under ice-covered conditions, pCO₂ measurements were taken
292 through an open melt pond. At this time, elevated CO₂ concentrations were observed at the very
293 surface (0.1 m), followed by a sharp decrease to approximately 1 meter depth, coinciding with
294 the ice-water interface. Below this depth, pCO₂ increased again, though remained well below
295 atmospheric concentrations (Fig. 34a).

296
297 During the period of sea ice breakup, when ice coverage ranged from approximately 30% to
298 10%, the vertical distribution of pCO₂ exhibited a similar structure. Concentrations were highest
299 near the surface, declined to a local minimum at 1 to 2 meters, and then stabilized below 3
300 meters (Fig. 34b). Following, the complete breakup of sea ice, pCO₂ showed a more gradual
301 decrease from the surface down to about 3 meters, beneath which concentrations remained
302 relatively constant (Fig. 34c). Across all three observational periods, a shallow surface layer
303 approximately 5 m thick was identified, within which most of the pCO₂ variability occurred.
304 Below this depth, pCO₂ remained relatively constant.

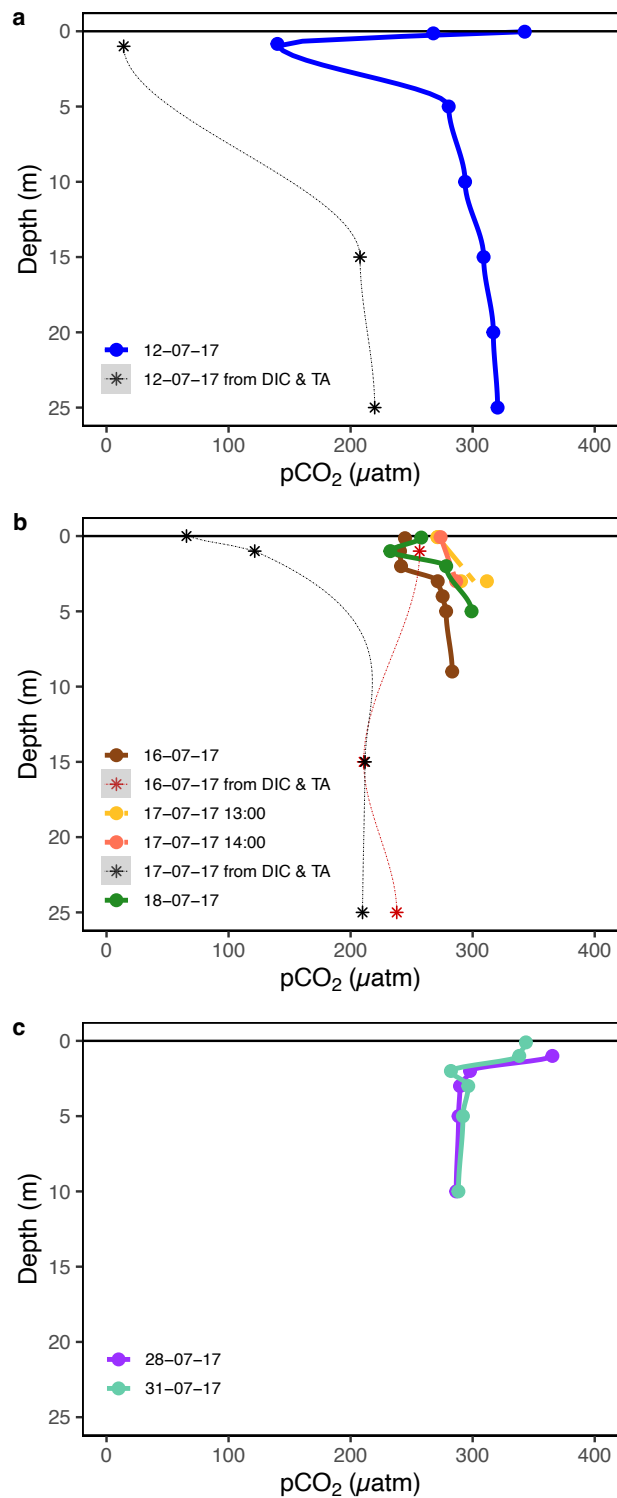
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306 These vertical structures are consistent with strong physical stratification, likely driven by
307 freshwater input from glacial melt and surface heating. Temperature and salinity profiles
308 collected concurrently support the presence of sharp vertical gradients in the upper water
309 column, with salinity ranging from 1.4 to 29.6 PSU and temperature from -0.4°C to 6.2°C. These
310 physical profiles, shown in Fig. 45, confirm that vertical mixing was strongly suppressed during
311 the observational period. The small tides present in Young Sound, combined with the continued
312 input of freshwater from first sea ice and then glacial melt, lead to a fjord system with distinct
313 surface stratification throughout the months of July and August.

314
315 Measurements from a different fjord in East Greenland on June 4, 2025, revealed strikingly
316 similar vertical pCO₂ heterogeneity (Fig. 56). Elevated pCO₂ at 0.1 m decreased to a minimum
317 around 1-1.5 m before increasing again and stabilizing near 3 m depth. Extreme stratification in
318 the upper few meters caused pCO₂ levels in each profile to vary by more than 100 µatm between
319 the surface and 1 m. This repeated observation of comparable vertical pCO₂ heterogeneity 8
320 years later and in a different fjord system suggest this is not an isolated phenomenon. Indeed,
321 Arctic surface stratification during sea ice breakup induces chemical changes that may influence
322 the way we estimate air-sea exchange of CO₂.

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324
 325 **Figure 3.** The 5 min averages of measured fluxes and meteorological conditions over Young
 326 Sound during July 2017. This time period reflects the transition between sea ice break up (30%
 327 ice cover) and open water (no sea ice present) from 16 July 2017 to 1 August 2017. Air-sea
 328 exchange of (a) CO₂ (b) sensible heat and (c) latent heat were estimated using the ogive
 329 optimization method with estimated uncertainty shown as vertical error bars. (d) Wind speed, (e)
 330 wind direction and (f) air temperature are shown for the same period. Note: Use of an undried
 331 air stream in the open path CO₂ sensor likely inflates the magnitude of the illustrated CO₂ fluxes
 332 due to water vapor cross sensitivity.

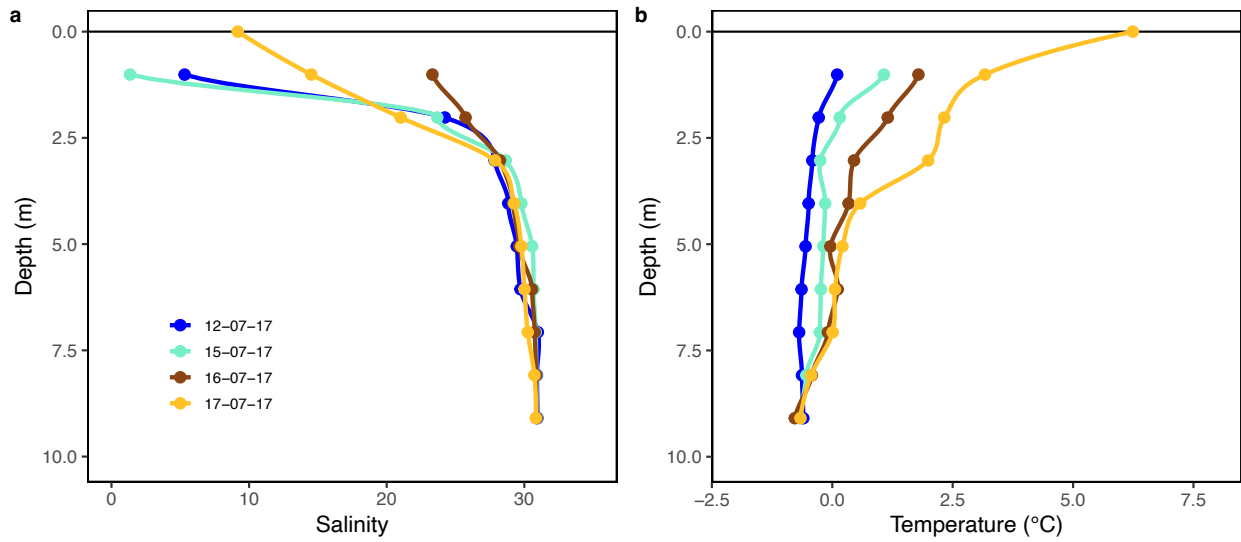


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336 **Figure 34.** Measured Young Sound $p\text{CO}_2$ profiles (a) prior to sea ice breakup (measured
 337 through open melt pond), (b) during sea ice breakup and (c) after sea ice break up measured
 338 through CO_2 equilibration and calculation from carbonate chemistry parameters (DIC & TA).

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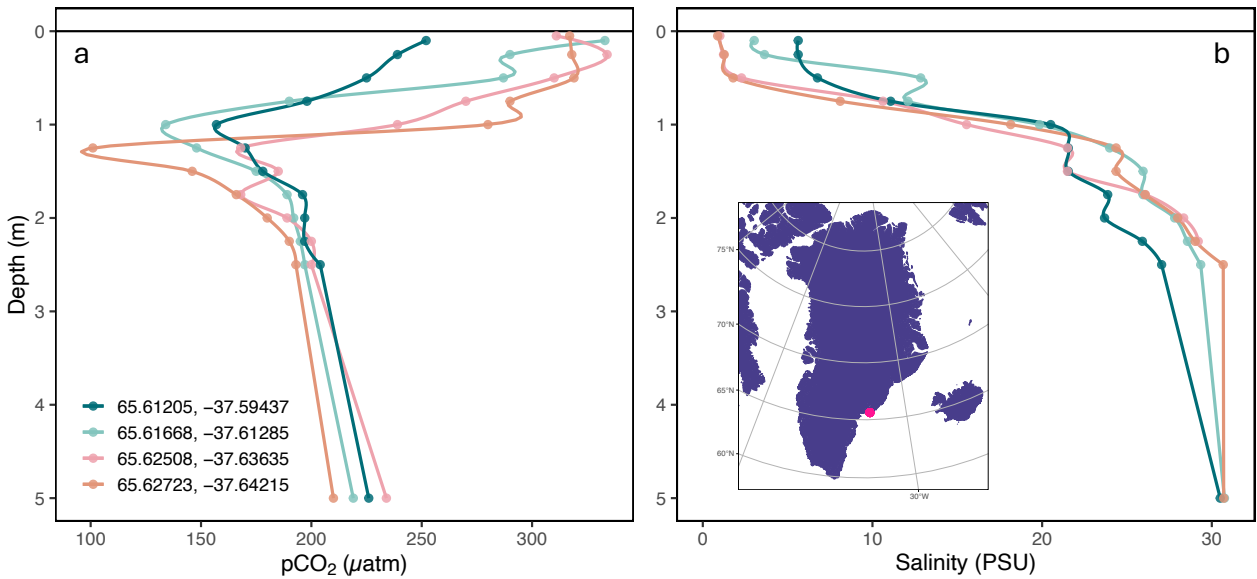


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342 **Figure 45.** Measured Young Sound profiles of under-ice water and open water salinity and
343 temperature.

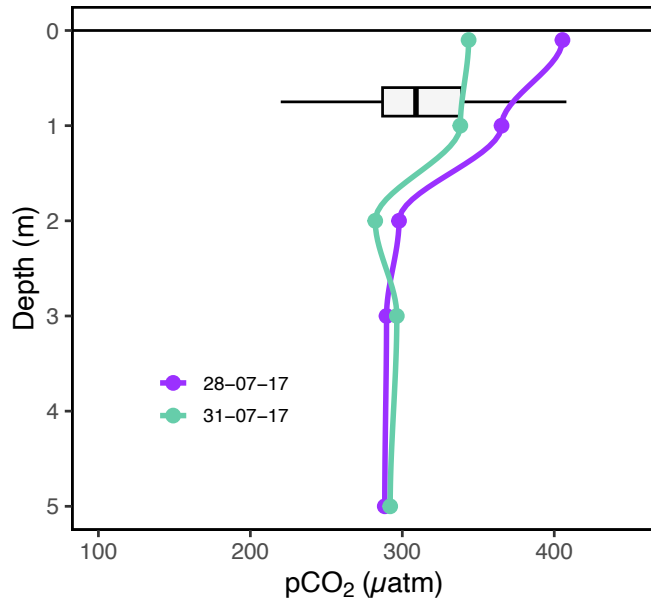
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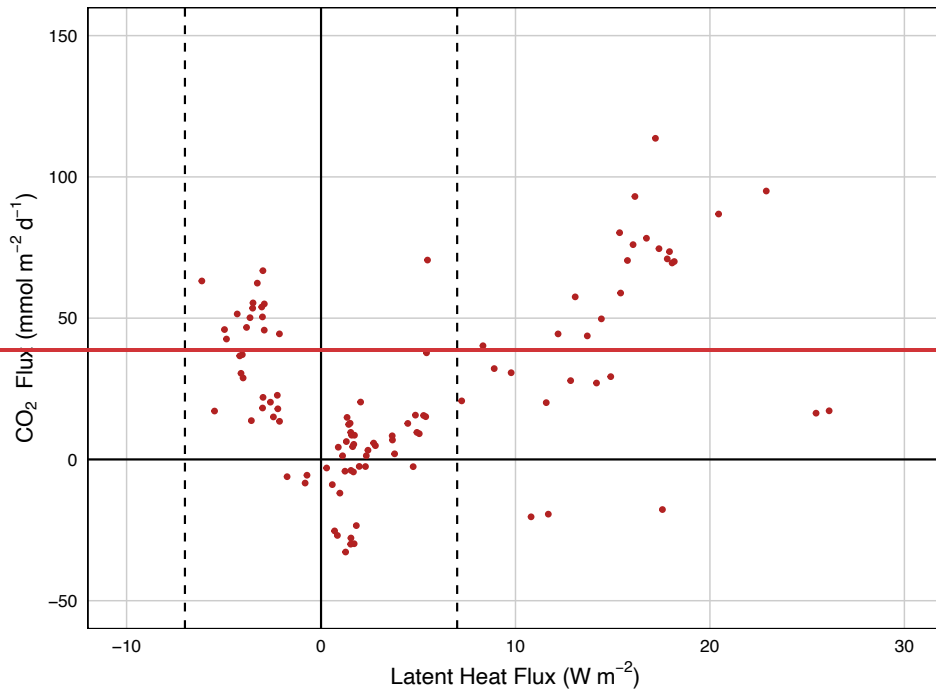
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347 **Figure 56.** Measured pCO₂ (a) and salinity (b) profiles at 4 locations in Tasiilaq Bay. Profiles
348 were measured on June 4, 2025 during the period of sea ice breakup following the method in Sejr
349 et al. (2011).



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Figure 67. Measured Young Sound pCO₂ profile after ice break up in 2017 compared with historical variation in pCO₂ at 1 m depth in the same location.



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Figure 8. Relationship between CO₂ and latent heat fluxes measured using eddy covariance. Vertical dotted lines delineate the threshold ($|LE| < 7 \text{ W m}^{-2}$) where Landwehr et al. (2014) found negligible water vapor-related biases in measurements from the un-dried infrared gas analyzer. Data within this range demonstrate both CO₂ uptake and effluxes, indicating that air-sea CO₂ exchange may change direction during the dynamic period of sea ice melt.

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4. Discussion

Air-sea CO₂ fluxes in Arctic coastal areas are generally estimated using bulk parameterization models (Henson et al., 2024; Meire et al., 2015; Roobaert et al., 2019; Sejr et al., 2011). These models rely on several key assumptions, including unstratified surface conditions, a linear pCO₂ gradient within the diffusive boundary layer, and a vertically uniform pCO₂ profile within the mixed layer. Our observations challenge the applicability of these assumptions in Arctic coastal waters in several important ways. We observe distinct vertical heterogeneity in both physical and chemical parameters, which leads to distinct differences in flux estimates based on the depth of the measurements used.

4.1. ~~Evaluating the Constant pCO₂ Assumption~~ Stratified conditions in the marine CO₂ system

Vertical pCO₂ profiles collected during July 2017 revealed pronounced non-linear behavior in the upper 3 to 5 meters of the water column (Fig. 34). This directly contradicts the assumption that the ΔpCO_2 accurately represents the difference between the atmosphere and the “well-mixed bulk fluid” below the diffusive layer (Wanninkhof et al., 2009). Under ice-covered conditions, the lowest pCO₂ values (~150 ppm) were consistently observed just beneath the sea ice, with concentrations increasing with depth and stabilizing around 5 m (Fig. 34a). During the ice breakup stage, a similar pattern emerged, although the minimum pCO₂ was higher (~250 ppm).

More recent measurements from Tasiilaq Bay in June 2025 demonstrate very similar vertical pCO₂ profiles. Indeed, 4 high-resolution profiles with measurements every 0.25 m reveal the same C-shaped pCO₂ variation. Like in Young Sound, the most elevated pCO₂ levels were observed near the surface, and pCO₂ minimums occurred near 1-2 meters depth before increasing and becoming stable. This repeated observation in a different fjord system, but during the period of sea ice breakup indicates this vertical variability may be representative during stratified Arctic conditions.

Several interacting processes influence surface water chemistry during ice breakup. Low surface water pCO₂ values reflect the influence of low-salinity meltwater from snow and sea ice or glacial meltwater found in freshened Arctic waters (Geilfus et al., 2015; Henson et al., 2025). However, surface water chemistry during the ice breakup period is further complicated by processes such as ikaite (CaCO₃·6H₂O) dissolution (Miller et al., 2011; Rysgaard et al., 2013; Sogaard et al., 2013) and high under-ice primary production (Sogaard et al., 2021). Additionally, snowmelt, characterized by low pH and ionic strength (de Caritat et al., 2005), may further alter carbonate system dynamics in the upper water column.

Two mechanisms may explain the nonlinear C-shaped trend in pCO₂ observed in the top few meters. First, as demonstrated by Henson et al. (2025) mixing between glacial meltwater and

402 seawater can result in nonlinear behavior in pCO₂, even when DIC and TA mix conservatively.
403 In such cases, initial freshwater dilution leads to dramatically reduced pCO₂, but at very low
404 salinities, the diminished buffering capacity can cause acidification to occur and pCO₂ to
405 increase again. In fact, Henson et al. (2025) present a U-shaped pCO₂ curve along a salinity
406 gradient that appears extraordinarily similar to the C-shaped pCO₂ curve we observe with depth.
407 Indeed, the salinity gradient created in their mixing experiments can be observed vertically in the
408 highly stratified surface waters during sea ice breakup (Fig. 3a, 5a). Although, Henson et al.
409 focused on the influence of glacial meltwater, our results suggest similar processes could occur
410 in systems influenced by sea ice and snowmelt.

411
412 Both glacial meltwater and sea ice have low DIC concentrations and act to dilute the inorganic
413 carbon of the surface ocean (Fig. S4). However, changes in alkalinity can also impact the
414 buffering capacity of the water mixture, leading to nonlinear effects. If the meltwater has a lower
415 TA:DIC ratio than seawater, due to the absence of ikaite, acidification and a shift in carbonate
416 equilibria at very low salinities could lead to higher pCO₂ values at the surface. During July
417 2017, Young Sound showed both diluted DIC and TA levels in upper few meters, suggesting pH
418 change during sea ice break up could occur more easily (Fig. S4). Indeed, calculated pH profiles
419 indicated variable surface conditions between periods of sea ice cover and sea ice breakup (Fig.
420 S5). In this very fresh surface layer, diminished pH may elevate pCO₂ relative to waters around 1
421 m depth, where freshwater-seawater mixing ratios are more moderate and seawater buffering
422 leads to very low CO₂ concentrations.

423
424 A second, but less likely, explanation involves atmospheric equilibration of sea ice melt ponds
425 before draining into open leads. The relatively elevated pCO₂ observed at ~0.1 m depth could
426 reflect such partial equilibration. While chamber-based studies (e.g. Geilfus et al., 2012, 2015;
427 Nomura et al., 2010; Semiletov et al., 2004) have demonstrated both uptake and efflux of CO₂ in
428 melt ponds, equilibrium times between melt-pond water and atmosphere depend upon pond
429 depth, wind speed, and carbonate chemistry. For example, a 0.1 m deep pond under low wind
430 conditions (~2 m s⁻¹) may reach atmospheric equilibrium in 1-4 days. However, in our case,
431 pCO₂ values calculated from TA and DIC in melt ponds did not indicate equilibrium with the
432 atmosphere, making this explanation less likely than the freshwater mixing mechanism.
433 Nevertheless, atmospheric equilibration may play a role after the sea ice barrier is removed.
434 Elevated pCO₂ levels at the surface (0.1 m) post sea ice breakup may result from the
435 combination of the chemical changes described above, heating from solar radiation, and from
436 atmospheric CO₂ uptake (partial equilibration) in the limited volume of this freshwater lens.

437
438 As melt progresses and sea ice recedes, riverine input and vertical mixing become more
439 influential. Yet even after ice breakup, surface waters often remain fresh due to glacial meltwater
440 runoff, and the resulting low salinities help maintain stratification. In August 2017, vertical
441 structure remained pronounced even after sea ice breakup, with elevated pCO₂ at 0.1 m which

442 stabilized below ~3 m. In other words, near-surface conditions remained decoupled from deeper
443 waters. This persistent shallow layer, characterized by low salinity, higher temperature, and
444 elevated pCO₂, suppresses gas exchange with the colder, more undersaturated water below,
445 consistent with observations by Dong et al. (2021). In such environments, bulk flux models that
446 assume homogeneity and linear gradients are likely to yield biased or inaccurate estimates.

447 To place these 2017 measurements in historical context, we examined long-term surface water
448 pCO₂ data collected at 0.5-1 m depth by the Greenland Ecosystem Monitoring (GEM) program
449 between 2007 and 2023. These data, measured using consistent protocols, are presented in Fig.
450 [67](#) alongside our open-water profiles. Over the 17-year record, August pCO₂ concentrations at
451 ~1 m depth had ranged from 220 to 408 μatm and had consistently remained below atmospheric
452 levels. This apparent stability has contributed to the perception of sustained CO₂ uptake
453 throughout the summer season.

454 However, the high-resolution vertical profiles obtained during the 2017 field campaign ~~challenge~~
455 ~~add nuance to~~ this assumption. Elevated pCO₂ levels confined to the uppermost meter of the
456 water column may go undetected in standard monitoring approaches that rely on fixed-depth
457 sampling. These results suggest that ~~the dynamic changes during sea ice melt can induce short-~~
458 ~~lived~~ episodes of ~~slowed uptake, air-sea equilibrium, or~~ CO₂ outgassing ~~can occur during rapid~~
459 ~~environmental transitions such as sea ice breakup~~. Consequently, existing sampling protocols
460 may underestimate surface variability and bias flux estimates, especially in stratified conditions
461 where near-surface chemistry is decoupled from subsurface layers.

462 **4.2. Evaluating Bulk Model Flux Estimates Influence of surface stratification on fluxes**

463 To assess ~~whether how~~ bulk models ~~are suitable for~~ function when estimating CO₂ fluxes in an
464 Arctic fjord influenced by sea ice and snow melt, we calculated fluxes using seawater pCO₂
465 measurements from multiple depths and two gas transfer velocity parameterizations.
466 Specifically, we computed fluxes throughout July using pCO₂ measured at 0.1, 1, 2, and 4 m. To
467 estimate the surface (interface) pCO₂ at 0 m, we adjusted the 1 m pCO₂ measurements to a
468 derived skin temperature (Table 1), estimated from sensible heat fluxes (Fig. [S63b](#)) -following
469 the parameterization of Smedman et al. (2007). Accounting for this skin layer correction is
470 critical, as Woolf et al. (2016) demonstrated that neglecting the thermal skin and relying only on
471 bulk sea surface temperature can introduce significant errors in flux estimates.
472

473
474 The resulting calculations (Table 1) show that estimated CO₂ fluxes vary significantly depending
475 on the depth of the pCO₂ measurement. Notably, fluxes derived from 0.1 m differ markedly from
476 those based on deeper values. Since many studies rely on pCO₂ measured at a fixed depth (often
477 at 1 m or at a ship's seawater intake below 5 m), these results underscore the potential for
478 misrepresentation of flux magnitude and direction ~~and magnitude~~ due to vertical heterogeneity in
479 surface water chemistry.

480

481 ~~Measured fluxes from eddy covariance (Fig. 3a) also exhibited large temporal variability. While~~
482 ~~micrometeorological methods integrate fluxes over a horizontal footprint, bulk flux models rely~~
483 ~~on single point pCO₂ gradients between air and water. Consequently, under heterogeneous or~~
484 ~~partially ice-covered conditions, the two methods are unlikely to yield identical results.~~
485 ~~However, a qualitative comparison between the two approaches remains informative. Notably,~~
486 ~~agreement between micrometeorological and bulk model estimates was only observed in late~~
487 ~~July, and only when in situ pCO₂ was normalized to the derived skin temperature. This aligns~~
488 ~~with the conclusions of Woolf et al. (2016) and Ford et al. (2024), who suggest that using skin-~~
489 ~~adjusted pCO₂ may improve flux estimates. Accurate application of this correction would require~~
490 ~~either direct measurements of skin temperature when sampling for the bulk method or high-~~
491 ~~resolution modeling of heat fluxes.~~

492

493 ~~However, this agreement between methods did not hold before or immediately after ice breakup.~~
494 ~~During these periods, micrometeorological methods indicated episodes of large upward fluxes,~~
495 ~~while bulk model estimates suggested downward fluxes (-14 mmol m⁻² d⁻¹). Although both~~
496 ~~methods yielded comparable magnitudes for downward fluxes, only the micrometeorological~~
497 ~~approach captured large upward events that could not be reconciled with surface water pCO₂~~
498 ~~profiles alone.~~

499

500 Measurements from both Young Sound and Tasiilaq demonstrate that during sea ice breakup,
501 pCO₂ levels are most elevated at the surface. This may be linked to acidification of the most
502 freshened 0.5 m and a shift in the marine carbonate system, or partial equilibration due to air-sea
503 gas transfer. If this acidified freshwater lens warms, for instance, due to solar radiation, pCO₂
504 may rise leading to oversaturation relative to atmospheric concentrations. Indeed, when pCO₂
505 measurements on July 31 were corrected for skin temperature, to estimate pCO₂ at the boundary
506 layer, they suggested a transition from undersaturation to oversaturation (Table 1). While we did
507 not directly observe this oversaturation in the vertical profiles, this likely reflects the inability to
508 sample at the very sea surface layer. ~~Moreover, profile measurements represent only single~~
509 ~~points in a system characterized by strong spatial and temporal variability during this seasonal~~
510 ~~transition.~~ Nevertheless, the occurrence of stratification-related vertical C-shaped pCO₂
511 heterogeneity profiles during sea ice breakup, with levels most elevated at the surface, will slow
512 air-sea gas transfer compared to when conditions are well mixed. Meanwhile, warming or
513 acidification of the thin surface layer may periodically induce a ~~may help explain the apparent~~
514 ~~reversal of flux direction-, as seen in some micrometeorological studies in Arctic coastal~~
515 environments during sea ice breakup (e.g. Butterworth et al., 2025). ~~suggested by the~~
516 ~~micrometeorological approach.~~

517

518 ~~Another potential explanation for the elevated CO₂ efflux that cannot be discounted involves~~
519 ~~cross-sensitivity between water vapor and CO₂ in open-path NDIR sensors (Blomquist et al.,~~

2014; Landwehr et al., 2014). Although the OGM processing method is designed to minimize humidity-induced artifacts and large-scale turbulent motions, it cannot entirely remove them. While many of the extreme eddy covariance measurements from Young Sound are likely influenced by humidity-CO₂ cross-sensitivity, several lines of evidence indicate that some positive CO₂ fluxes may not solely be attributed to this artifact. In their comparative methodological study, Landwehr et al. (2014) identified large biases in open-path NDIR sensors related to water-vapor fluctuations, but reported agreement between sensors operating with dried and undried air streams during periods of low latent heat flux ($|LE| \leq 7 \text{ W m}^{-2}$), where humidity-related bias was negligible. When examining the subset of Young Sound data below this threshold, eddy covariance measurements exhibit both positive and negative CO₂ flux estimates (Fig. 8). Additionally, elevated CO₂ concentrations occur across a wide range of relative humidities (Fig. S6). Taken together, these patterns support the interpretation that some of the measured effluxes may represent real air-sea CO₂ exchange rather than solely resulting from cross-sensitivity artifacts. However, definitive quantification using this experimental setup remains impossible and future studies using closed-path sensors during Arctic seasonal transitions are required to robustly constrain the magnitude of CO₂ outgassing and uptake.

Overall, these findings echo those of Miller et al. (2019), who reported pronounced spatial heterogeneity in Arctic coastal pCO₂ and large differences in estimated fluxes depending on the sampling depth. The broader implications of this heterogeneity for seasonal or regional flux estimates remain unclear. However, if fluxes are upscaled from sparse, single-point measurements (e.g., once per month, as in Laruelle et al., 2014), substantial errors may result due to unrecognized spatial and temporal variability. Thus, our results emphasize the need for continuous, high-resolution observations of air-sea CO₂ fluxes, particularly in Arctic coastal systems affected by stratification and meltwater input. These observations will be essential for refining flux parameterizations, reducing uncertainty in carbon budget estimates, and improving the representation of Arctic shelf systems in global carbon models.

Table 1. CO₂ fluxes calculated based on pCO₂ measured at the different depth. The fluxes are calculated using the bulk model of Ho et al., 2006 and Nightingale et al. (2000). We have used locally measured wind speeds for the calculations to match flux measurements captured by eddy covariance.

Date	Depth (m)	Temperature (°C)	Salinity (psu)	pCO ₂ (µatm)	Wind Speed (m s ⁻¹)	Ho (2006)	Nightingale (2000)
						Flux (mmol CO ₂ m ⁻² day ⁻¹)	Flux (mmol CO ₂ m ⁻² day ⁻¹)
16-Jul	0.0	3.0 [†]	23	252 [‡]	6.8	-14.88	-13.69
16-Jul	0.1	3.0	23	244	6.8	-15.78	-14.52
16-Jul	1.0	1.8	26	240	6.8	-16.12	-14.83
16-Jul	2.0	1.1	28	241	6.8	-15.93	-14.66
16-Jul	4.0	0.3	29	275	6.8	-12.19	-11.21
18-Jul	0.0	6.0 [†]	7	262 [‡]	3.3	-3.45	-3.38
18-Jul	0.1	4.3	7	244	3.3	-3.98	-3.90
18-Jul	1.0	3.2	14	233	3.3	-4.20	-4.11
18-Jul	2.0	2.3	21	278	3.3	-2.86	-2.79
18-Jul	4.0	0.6	29	295	3.3	-2.34	-2.29
28-Jul	0.0	10.0 [†]	15	415 [‡]	2.5	0.53	0.54
28-Jul	0.1	10.0	15	405	2.5	0.38	0.38
28-Jul	1.0	7.0	21	365	2.5	-0.22	-0.23
28-Jul	2.0	5.0	27	282	2.5	-1.43	-1.45
28-Jul	4.0	2.0	30	290	2.5	-1.32	-1.34
31-Jul	0.0	12.0 [†]	15	401 [‡]	2.0	0.20	0.21
31-Jul	0.1	10.0	15	343	2.0	-0.36	-0.38
31-Jul	1.0	8.0	21	338	2.0	-0.40	-0.42
31-Jul	2.0	5.0	27	282	2.0	-0.92	-0.97
31-Jul	4.0	2.0	30	294	2.0	-0.81	-0.85

554

555 † Denotes skin temperatures derived from heat fluxes. ‡ Denotes pCO₂ values estimated from
556 measurements at 1 m depth and adjusted to derived skin temperatures.

557

558

559 5 Conclusions

560

561 During the summer thaw, carbon chemistry and pCO₂ dynamics in Arctic coastal surface waters
562 are significantly altered by the combined effects of snow and sea ice melt, terrestrial runoff, and
563 biological activity. These influences lead to substantial variability in surface temperature, pH,
564 dissolved inorganic carbon (DIC), and total alkalinity (TA), ultimately disrupting carbonate

565 system equilibrium in the upper water column. As a result, estimating air-sea CO₂ fluxes using
566 traditional bulk models becomes highly uncertain during this period.

567
568 The sea ice breakup period, typically lasting 2-4 weeks, represents a particularly dynamic and
569 complex phase in the annual cycle. Despite its brevity, this phase may have a disproportionate
570 influence on total summer CO₂ uptake, given that open-water conditions in high Arctic fjords are
571 limited to only 80-120 days per year (Sejr et al., 2011).

572
573 ~~Improved flux estimates will require more detailed and spatially resolved investigations aimed at~~
574 ~~developing and validating gas exchange parameterizations tailored to the highly stratified and~~
575 ~~ice-affected conditions of Arctic fjords. In particular, new approaches are needed to estimate gas~~
576 ~~transfer velocities over waters influenced by snow and sea ice melt. Air-sea gas eXchange rates~~
577 depend not only on the pCO₂ gradient between the atmosphere and surface water, but also on
578 rapid, nonlinear changes in surface water chemistry driven by the composition and volume of
579 meltwater and runoff.

580
581 ~~Once more suitable parameterizations for gas transfer velocity are established, a~~Accurate flux
582 estimation will also therefore require knowledge of the depth at which surface water pCO₂
583 becomes vertically homogeneous, combined with gas exchange parameterizations tailored to
584 highly stratified and ice-affected conditions. Profiling pCO₂ in the upper water column is
585 therefore essential to identify this depth and to constrain surface flux estimates reliably.

586
587 Several eddy covariance studies in other arctic environments report variable uptake and efflux of
588 CO₂ during the sea ice breakup period (e.g. Butterworth et al., 2025). Building on these findings,
589 our data provides the first step in understanding potential drivers behind this
590 variability. Similarly, eddy covariance measurements in Young Sound exhibited both positive and
591 negative CO₂ flux estimates during this seasonal transition. However, these upward fluxes were
592 not captured by the bulk model. Though, proper quantification of the mechanisms driving
593 nonlinear pCO₂ profiles and the resulting uncertainty of flux estimates will require
594 observations. These variable and sometimes conflicting datasets spanning the air-sea boundary.
595 The current lack of these datasets underscores the need for studies that integrate continuous,
596 direct CO₂ flux measurements with detailed observations of surface water carbonate chemistry,
597 atmospheric forcing, skin temperature, and turbulence at the air-ice-water interface.

598
599 Such integrated measurements are critical to improving our understanding of the frequency,
600 drivers, and net effect of episodic upward sea ice melt-driven changes on CO₂ fluxes in Arctic
601 coastal systems. Ultimately, this knowledge is essential to accurately quantify the seasonal and
602 regional uptake of atmospheric CO₂ in the rapidly changing Arctic.

603
604

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623

624 **Author Contribution**

625 Conceptualization: LLS. Formal analysis, writing – original draft preparation: HCH. Funding
626 acquisition: LLS, SR, MKS, TP. Investigation: DS, BJ, KL, TP, MKS, JS, SR, LLS. Writing –
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629

630 **Data Availability Statement**

631 Vertical profiles from both Greenlandic fjords can be found in the Zenodo data repository:
632 <https://doi.org/10.5281/zenodo.17471918>

633

634 **Competing interests**

635 The authors declare no competing interests.

636

637

638 **References**

639

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