https://doi.org/10.5194/egusphere-2025-4822 Preprint. Discussion started: 20 October 2025 © Author(s) 2025. CC BY 4.0 License.





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A mechanistic model of hypoxia-driven benthic carbon cycling integrating microbial energetics and faunal mortality

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Abstract. Hypoxia reduces the mineralization of organic detritus and increases mortality in benthic fauna, both of which alter carbon storage through complex changes in organic matter and calcium carbonate (CaCO₃) dynamics. To mechanistically assess these processes, we developed a new model that links oxic, suboxic, and anoxic mineralization pathways to microbial ATP production efficiency. This formulation was incorporated into the benthic-pelagic coupled model EMAGIN-B.C., resulting in an extended version designated EMAGIN-B.C.-MR (MR: mineralization rate). The model also includes revised mortality and metabolic suppression functions for benthic fauna under oxygen-deficient conditions and explicitly couples suspension-feeding benthos biomass with CaCO₃ production and burial fluxes. We applied EMAGIN-B.C.-MR to Tokyo Bay, a eutrophic coastal system prone to seasonal hypoxia, to simulate long-term changes in carbon cycling under hypoxic (0 mg L⁻¹) and non-hypoxic (5 mg L⁻¹) summer conditions. Results showed that hypoxia enhanced detritus storage and burial by both suppressing microbial degradation and reducing bioturbation and grazing due to suspension-feeding benthos mortality. Conversely, CaCO3 production and burial declined owing to inhibited shell formation. These dynamics revealed that total carbon storage is shaped by interacting biogeochemical and ecological feedbacks, resulting in nonlinear trajectories under repeated hypoxic stress over decadal timescales. By integrating microbial energetics and oxygen-sensitive faunal responses, the EMAGIN-B.C.-MR model provides a mechanistic framework for assessing benthic carbon cycling under deoxygenation. This framework offers biogeochemical insights into the regulation of organic and inorganic carbon burial balance by oxygen availability – with implications for coastal carbon budgets, blue carbon management, and climate feedbacks – and is applicable to other oxygen-deficient environments such as eutrophic estuaries and semi-enclosed seas.

1 Introduction

According to the Sixth Assessment Report of the Intergovernmental Panel on Climate Change under the SSP5-8.5 scenario, the pathway projecting the highest temperature increase, the global average temperature is predicted to rise by up to 5.7 °C between 2081 and 2100 relative to the pre-industrial period (1850–1900) (IPCC, 2023). This highlights the urgency of reducing greenhouse gas emissions, particularly CO₂. The ocean, functioning as a major carbon sink, plays a pivotal role in mitigating climate change via carbon sequestration (Archer and Brovkin, 2008; Falkowski et al., 2000; IPCC, 2007; McLeod et al., 2011), which is governed by the following three primary mechanisms: (1) air—sea gas exchange of atmospheric CO₂ (carbon capture), (2) biological fixation of dissolved inorganic carbon via photosynthesis and calcification (carbon fixation), and (3) long-term burial of fixed carbon in sediments (carbon storage) (Sohma et al., 2018) (Fig. 1). Coastal areas, characterized by high





- biological productivity, are particularly efficient in carbon capture, fixation (Nellemann et al., 2009), and storage, thereby
 identifying these areas as key regions for carbon cycle research (Alongi et al., 2016; Chen et al., 2013; Chmura et al., 2003;
- Donato et al., 2011; Fourqurean et al., 2012; Frankignoulle et al., 1998; Kubo and Kanda, 2017; Kuwae et al., 2016; Murdiyarso
- 36 et al., 2015).

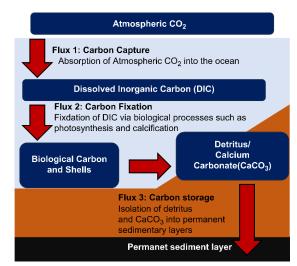


Figure 1. A schematic representation of climate change mitigation functions.

In enclosed coastal bays, such as Tokyo Bay, seasonal hypoxia in bottom waters has become an increasingly frequent occurrence during summer months (Breitburg et al., 2018; Diaz, 2001; Hanyu, 2020; Ishii and Ohata, 2010; Okubo et al., 2016; Rabalais et al., 2009). Hypoxia influences carbon storage via two major pathways. First, it suppresses aerobic mineralization, which promotes the preservation of organic matter (Andersen, 1996; Anderson et al., 1986; Bastviken et al., 2003; Benner et al., 1984; Blackburn, 1991; Canfield, 1989; Canfield, 1993; Canfield, 1994; Cowie et al., 1995; Emerson and Hedges, 1988; Fallon and Brock, 1979; Hansen and Blackburn, 1991; Rodger Harvey et al., 1995; Hedges, 1999; Hedges and Keil, 1995; Hollander et al., 1992; Hulthe et al., 1998; Ingall et al., 1993; Kristensen and Holmer, 2001; Kristensen and Holmer, 2001; Nguyen and Harvey, 1997; Paropkari et al., 1993; Sun et al., 1993; Sun et al., 1997; Wilson et al., 1985; Zsolnay, 1971). Second, it induces mortality in benthic fauna, which reduces dissolved inorganic carbon fixation via the food web originating from primary producers and the formation of shells (i.e., CaCO₃) by benthic fauna (Kristensen and Holmer, 2001; Rabalais et al., 2009). In this context, the storage of detritus due to suppressed oxic mineralization enhances carbon burial into permanent sedimentation layers (Burdige, 2007; Carey et al., 2018; Hartnett et al., 1998; Isidorova et al., 2019; Katsev and Crowe, 2015; LaRowe and Van Cappellen, 2011; Sobek et al., 2009), whereas the reduced production of benthic organisms and the decreased





53 deposition of CaCO3 due to hypoxia-induced mortality act to lower it. In addition, the dissolution of CaCO3 contributes to 54 altering the alkalinity of seawater and pCO2, thereby influencing the capacity of oceans to absorb atmospheric CO2. Thus, the 55 net effect of hypoxia on carbon storage is dependent on complex biogeochemical and ecological interactions. 56 Although the coastal ecosystem models developed to date, including EMAGIN-B.C., incorporate multiple redox pathways for 57 organic matter mineralization, they often represent these processes using simplified first-order kinetics, without explicitly 58 considering microbial energetic constraints (Aumont et al., 2003; Burdige, 2007; Carey et al., 2018; Cerco et al., 2006; Cowie 59 et al., 1995; Fennel et al., 2011; Flynn, 2010; Hartnett et al., 1998; Hulthe et al., 1998; Isidorova et al., 2019; Katsev and 60 Crowe, 2015; LaRowe and Van Cappellen, 2011; Meire et al., 2013; Reed et al., 2011; Sobek et al., 2009; Soetaert and 61 Middelburg, 2009; Soetaert et al., 1996; Sohma et al., 2018; Stock et al., 2014; Wild-Allen et al., 2010; Yakushev et al., 2017; 62 Yool et al., 2013). Moreover, although some models account for the oxygen-dependent mortality of benthic fauna (Butenschön 63 et al., 2016; Nagao et al., 2005; Sohma et al., 2008), the interactions among mineralization efficiency, microbial growth, and 64 faunal mortality under conditions of hypoxic stress remain poorly integrated. 65 In this study, we addressed these limitations by developing a novel model module that explicitly links redox-sensitive 66 mineralization rates to microbial ATP production, grounded in microbial energetics and thermodynamics. Additionally, we 67 refine hypoxia-induced mortality functions for benthic fauna based on threshold dissolved oxygen (DO) saturation levels derived from empirical observations. Our approach is novel given our representation of microbial responses to hypoxia as well 68 69 as capturing the synergistic interplay between microbial metabolism and benthic faunal dynamics. These interactions 70 determine the net carbon storage outcome in hypoxic sediments. 71 By integrating these advances into the established EMAGIN-B.C. benthic-pelagic coupled ecosystem model, we formulated 72 an extended framework, designated as EMAGIN-B.C.-MR (mineralization rate), which can be applied to simulate 73 biogeochemical feedback of hypoxia on benthic carbon storage. Our application of this model in Tokyo Bay enabled us to 74 mechanistically analyze long-term changes in carbon burial and storage, thereby offering new insights into the role of hypoxia 75 in coastal carbon dynamics.

2. Methods

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In this study, we define "detritus storage" as the temporary accumulation of organic matter within the upper 10 cm of surface

78 sediments, whereas "carbon burial" refers to the permanent sequestration of organic carbon within deeper sediment layers



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with their respective respiratory reactions.

(Here, M means carbon mass, T means time)

The rate of microbial growth (M_{BAC} T⁻¹) is proportional to the rate of ATP acquisition (M_{ATP} T⁻¹)



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79 beyond the active biogeochemical zone. To prevent ambiguity, we have intentionally avoided use of the term "detritus 80 accumulation," which is replaced with either "storage" or "burial," depending on the context. 81 2.1 Modeling of mineralization 82 To quantify the effects of hypoxia on benthic carbon cycling, we extended the benthic-pelagic coupled ecosystem model 83 EMAGIN-B.C. by incorporating a novel formulation of organic matter mineralization that explicitly accounts for microbial 84 energetics (Zehnder and Stumm, 1988). The enhanced model, referred to as EMAGIN-B.C.-MR, simulates redox-sensitive 85 mineralization rates based on adenosine triphosphoric acid (ATP) production efficiencies under oxic, suboxic, and anoxic 86 conditions. This formulation draws on established microbial physiological principles and thermodynamic considerations, thereby enabling the mechanistic representation of energy-constrained microbial decomposition processes. 87 88 Although in numerous conventional ecosystem models, including EMAGIN-B.C., MEDUSA-2.0 (Yool et al., 2013), and 89 BROM (Yakushev et al., 2017), mineralization is represented as a first-order function of detritus concentration, this approach 90 neglects important redox-dependent constraints on microbial growth and metabolism. Our approach introduces microbial ATP 91 production rates and biomass dynamics as key mediators of organic matter degradation, thereby contributing to more realistic 92 simulations of mineralization processes under conditions of hypoxic stress. 93 To capture the variation in microbial growth rates under different redox conditions (oxic > suboxic > anoxic) and to incorporate 94 pathway-specific differences in mineralization efficiency, we made the following assumptions: The rate of detritus mineralization (M_{DET} T⁻¹) by microorganisms is proportional to the amount of detritus (M_{DET} T⁻¹). 95 96 The rate of detritus mineralization (M_{DET} T⁻¹) by microorganisms is proportional to microbial biomass (M_{BAC}). 97 The rate of microbial growth (M_{BAC} T⁻¹) is proportional to microbial biomass (M_{BAC}). 98 The rate of ATP acquisition (MATP T-1) from mineralization by microorganisms is proportional to the rate of microbial 99 mineralization (M_{DET} T⁻¹). 100 The rate of ATP generation (MATP T-1) by microorganisms is proportional to the rate of Gibbs free energy production associated





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- 104 Under these assumptions, we formulated the rate of detritus mineralization (Table 1 and Fig. 2). This formulation introduces
 105 the following three parameters:
- 106 1. ATP acquisition per unit organic matter mineralized (α_{ATP}):
- $\alpha_{ATP} [M_{ATP} M_{DET}^{-1}]$ represents the molar concentration of ATP (M_{ATP}) generated per unit of organic matter (detritus, $M_{DET})$ mineralized. Compared with anaerobic microorganisms, this value is greater for oxic (aerobic) microbes, reflecting the higher
- energy acquisition efficiency of the latter and the more rapid rate of oxic mineralization compared with anoxic processes. The
- value of α_{ATP} for each class of microorganism was estimated based on the Gibbs free energy (δ_{ene}) (Canfield et al., 2005)
- 111 generated during each type of mineralization (oxic and anoxic), the energy consumed by microorganisms during ATP
- 112 generation (ε_{ATP}) (Bailey and Ollis, 1986) and the proportion of energy available for ATP generation (β_{ATP}) from
- 113 mineralization energy (Payne, 1970).
- 114 2. Microbial biomass per ATP generated (Y_{ATP}) :
- 115 $Y_{ATP}[M_{BAC}M_{ATP}^{-1}]$ represents the proliferation of bacterial biomass per mole of ATP generated, the value of which is
- assumed to be constant in different classes of microorganism (oxic and anoxic), as previously indicated (Jin, 2012).
- 117 3. Ratio of the rate of detritus consumption to the rate of microbial growth (R):
- 118 $R[M_{DET}T^{-1}/M_{BAC}T^{-1}]$ is treated as a tuning parameter that represents the ratio between the rates at which detritus is
- 119 consumed and microbial biomass grows.

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- 120 By incorporating these three parameters as multiplicative factors into the organic matter mineralization rate equation of existing
- models, we developed a detritus mineralization rate equation for EMAGIN-B.C.-MR.
- 122 Consequently, EMAGIN-B.C.-MR facilitated a more realistic evaluation of the reduction in detritus mineralization rates under
- 123 hypoxic conditions and the corresponding increase in carbon burial.

125 Table 1. Model formulation for the rate of detritus mineralization in EMAGIN-B.C.-MR

$$Ddet(i, j) = \alpha_{ATP}(j) \times Y_{ATP} \times R(j) \times \alpha(i) \times g(j) \times f(T, i) \times DET(i)$$

where i = 1: Labile detritus, i = 2: Quasi labile detritus, i = 3: Refractory detritus

j = 1: Oxic mineralization (oxygen consumption), j = 2: Suboxic mineralization (nitrate consumption), j = 3: Anoxic mineralization (iron, manganese, and sulfate consumption)

Symbol	Description	Unit	Value	Reference	





				/
$lpha_{ ext{ATP}}(j)$	ATP energy required for the decomposition of detritus $per\ unit\ mass.\ \divideontimes\alpha_{ATP}(j) = \sigma_{ene}(j)/\beta_{ATP} \times \epsilon_{ATP}$	$M_{ATP}M_{DET}^{-1}$	cal	-
$\sigma_{\text{ene}}(j=1)$	The amount of energy obtained via oxic (aerobic) respiration	kJ _{ENE} M _{DET} -1	402	(Zehnder and Svensson, 1986)
$\sigma_{ene}(j=2)$	The amount of energy obtained via suboxic respiration	kJ _{ENE} M _{DET} -1	359	(Zehnder and Svensson, 1986)
$\sigma_{\rm ene}(j=3)$	The amount of energy obtained via anoxic (anaerobic) respiration	$kJ_{ENE}\;M_{DET}^{-1}$	223	(Zehnder and Svensson, 1986)
$\epsilon_{ m ATP}$	The proportion of σ_{ene} used for ATP synthesis	$kJ_{ATP}\;M_{ATP}^{-1}$	0.4	(Fenchel and Finlay, 1995)
Ватр	The amount of energy required to synthesize one molecule of ATP	kJ _{ATP} kJ _{ENE} -1	30.88	(Russell and Cook, 1995)
Y_{ATP}	Microbial growth per unit of ATP	$M_{BAC} mol_{ATP}^{-1}$	5	(Canfield et al., 2005)
R(j=1)	The rate of organic matter consumption per unit of oxic (aerobic) microbial growth rate	$M_{DET}M_{BAC}^{-1}$	cal	tuning
R(j = 2)	The rate of organic matter consumption per unit of suboxic microbial growth rate	$M_{DET}M_{BAC}^{-1}$	cal	tuning
R(j=3)	The rate of organic matter consumption per unit of anaerobic (anoxic) microbial growth rate	$M_{DET}M_{BAC}^{-1}$	cal	tuning
α(i)	Rate constant	T-1	-	(Sohma et al., 2018)
g(i)	Oxygen-nitrate concentration-dependent function	-	-	(Sohma et al., 2018)
f(T, i)	Water temperature-dependent function	-	-	(Sohma et al., 2018)





DET(i) Amount of detritus (organic matter) storage $M_{DET}L^{-3}$ - 2018

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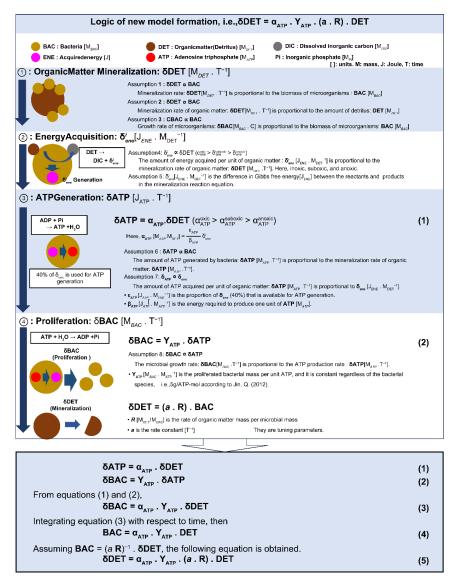


Figure 2. Logic of the new modeling of mineralization.



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2.2 Modeling of hypoxia tolerance for benthic fauna

When using both EMAGIN-B.C. and EMAGIN-B.C.-MR models, two types of benthic fauna are considered, namely, suspension (SFB) and deposit (DFB) feeding benthos, the mortality of which due to hypoxia is represented using the functions of temperature and DO concentrations (Table 2).

Compared with the EMAGIN-B.C. model, the primary enhancement of EMAGIN-B.C.-MR lies in the re-definition of the values of parameters deDO51 and deDO52, which represent the DO saturation thresholds (based on 24 h averages) for hypoxia tolerance. These threshold values were set in the range from 0.30 to 0.75 based on fisheries laboratory reports [Hypoxia Information of Ise and Mikawa Bay (R5-11), https://www.pref.aichi.jp/uploaded/attachment/472338.pdf], previous modeling studies (Sasaki et al., 2009), and laboratory experiments that examined hypoxia tolerance in benthic organisms (Shimo et al., 2004).

Here, the DO thresholds for benthic invertebrates are defined in terms of percent oxygen saturation, following standard practice in experimental studies (e.g., Diaz and Rosenberg, 1995; Gray et al., 2002). In the model, DO saturation (DO_sat) is dynamically calculated at each time step based on water temperature using empirical solubility functions (e.g., Garcia and Gordon, 1992). Percent saturation is then computed as the ratio of actual DO concentration to DO_sat. This method facilitates ecologically meaningful and temperature-consistent threshold application across time and space. Thus, even when DO concentrations (mg L⁻¹) are input, they are internally converted to percent saturation to determine benthic faunal responses.

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Table 2. Hypoxia tolerance equations for benthic fauna in EMAGIN-B.C.-MR

Mortality of SFB (suspension feeders): $Dsfb4 = -v511 \times u512 \times SFB$

Mortality of DFB (deposit feeders): $Ddfb4 = -v521 \times u522 \times DFB$

v511: Temperature-dependent function for SFB mortality

u512: Dissolved oxygen (DO) concentration-dependent function for SFB mortality

v521: Temperature-dependent function for DFB mortality

u522: DO concentration-dependent function for DFB mortality

SFB: Biomass of suspension feeders

DFB: Biomass of deposit feeders



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 $u512 = Ade511 + Ade512 \times [1 - min(1.0, dorat/deDO51)]$

 $u522 = Ade521 + Ade522 \times [1 - min(1.0, dorat/deDO52)]$

dorat: 24-h average value of oxygen saturation

Ade511: Background mortality rate of SFB at 0 °C (applies at all DO levels)

Ade512: Hypoxia-induced incremental mortality rate of SFB at 0 °C (zero when dorat ≥ deDO51

deDO51: Oxygen saturation threshold causing hypoxia for SFB

Ade521: Background mortality rate of DFB at 0 °C (applies at all DO levels)

Ade522: Hypoxia-induced incremental mortality rate of DFB at 0 °C (zero when dorat ≥ deDO52dorat)

deDO52: Oxygen saturation threshold causing hypoxia for DFB

147 Note. Ade511 (Ade521) is background mortality; Ade512 (Ade522) is the hypoxia-induced increment that vanishes at or above

the DO-saturation threshold deDO51(deDO52).

3. Application of the model: construction of the Base Case

To evaluate the effects of hypoxia on carbon storage, we applied the extended model "EMAGIN-B.C.-MR," which incorporates oxygen-dependent mineralization processes and revised hypoxia tolerance of benthic fauna, to Tokyo Bay. The objectives of the application were to: (1) verify the model's validity, and (2) construct a calculation scenario (Base Case) for comparison with two analytical scenarios (Cases 1 and 2). The Base Case was designed to reproduce the representative seasonal dynamics of the Tokyo Bay ecosystem around the year 2000 (1998 to 2002), for which abundant observational data are available for both pelagic and benthic systems. The model's validity was assessed by comparing the simulation results of the Base Case with the observations during this period.

3.1. Geomorphological and temporal settings

In the model configuration, Tokyo Bay was divided into 26 horizontal segments. The pelagic system (water column) was vertically divided into 2 m intervals. The benthic system (sediment layer) was divided into 30 vertical layers covering the upper 10 cm, with thinner layers closer to the sediment surface (Fig. 3). The surface layers of the benthic system were represented with millimeter-scale resolution to resolve sharp gradients associated with oxic, suboxic, and anoxic mineralization. The time resolution was set to 0.2 h to capture both diurnal and seasonal variations.





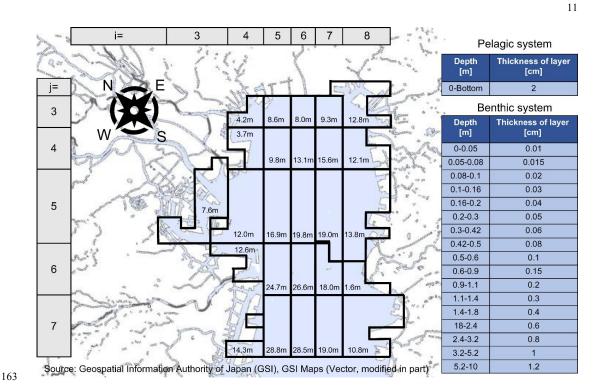


Figure 3. Calculation domain of EMAGIN-B.C.-MR (area: 728 km²). Left: grid coordinates (i, j) and water depth. Right: vertical layer thickness.

3.2. Boundary conditions, initial values, and biogeochemical parameters

EMAGIN-B.C.-MR is an extension of EMAGIN-B.C. (Sohma et al., 2018) and ECOHYM (Sohma et al., 2008). Therefore, the following components are shared among the three models: (1) model variables (model components) across pelagic and benthic systems (e.g., phytoplankton [PP], zooplankton, benthic microalgae, benthic invertebrates [suspension and deposit feeders], fast-/slow-/refractory detritus, dissolved organic matter, NH₄-N, NO₃-N, PO₄-P, DO, reduced substances [total of SO₄²⁻, Mn⁴⁺, and Fe³⁺], dissolved inorganic carbon, total alkalinity, pH, and pCO₂); (2) driving forces (e.g., river discharge, nutrient loading, solar radiation, wind speed, tidal height, temperature, salinity, and boundary values of model variables); (3) initial values of model components; and (4) biogeochemical parameters (except for those specifically related to oxygen-dependent mineralization and hypoxia tolerance of benthic fauna).

In the Base Case, the boundary conditions, initial values, and biogeochemical parameters, excluding those related to oxygen-dependent mineralization and hypoxia tolerance of benthic fauna, were set identically to those used in previous applications of EMAGIN-B.C. to Tokyo Bay (Sohma et al., 2008, 2018). Specifically, the boundary conditions were constructed as annual



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cyclic functions based on observational data from 1998 to 2002, facilitating the reproduction and simultaneous simulation of
seasonal variations that are representative of the Tokyo Bay ecosystem around the year 2000. Initial values were set to the
estimated average values on April 1st based on data from 1998 to 2002.

3.3. Execution and interpretation of the Base Case

Using the above settings, the EMAGIN-B.C.-MR model was run until all model variables reached an annual periodic steady state, where seasonal and diurnal variations are repeated each year. Approximately 20 years of simulation were required to reach this state, which corresponds to the time needed for organic particles settling from the pelagic layer to be buried to a depth of 10 cm in the sediment. The resulting Base Case thus represents the steady-state ecosystem of Tokyo Bay under conditions in which the forcing functions (boundary conditions) characteristic of around the year 2000 were applied repeatedly for 20 years. A more realistic reconstruction of ecosystem dynamics around the year 2000 would ideally involve setting the model's initial conditions to values from 1980 and defining the temporal changes in boundary conditions from 1980 to 2000. However, such datasets are not currently available. Moreover, seasonal patterns from 1998 to 2002 showed little interannual variability. Therefore, the Base Case was considered to be a representative approximation of the seasonal dynamics in Tokyo Bay around the year 2000.

4. Model validation: assessment using the Base Case

193 To evaluate the performance of EMAGIN-B.C.-MR, we compared its output with observed values for key ecological variables 194 from 1998 to 2002. Specifically, we compared DO, particulate organic carbon (POC), PP, nitrate (NO₃), ammonium (NH₄), 195 and phosphate (PO₄) (Table 3 and Fig. 4). We determined the correlation coefficient (R), root mean square error (RMSE), and 196 P-value for different depths and time points (Table 3). Observational data were obtained from representative stations in Tokyo 197 Bay between 1999 and 2002. 198 The results indicate that the model can successfully reproduce the seasonal reduction in bottom-layer DO and trends in POC. 199 A representative comparison is highlighted at grid point (i, j) = (6, 4) (Fig. 4), indicating the model's ability to capture seasonal 200 hypoxia in the bottom layer, with an R of 0.83 and an RMSE of 1.64 mg L⁻¹. 201 Overall, the modeled outputs for DO, POC, PP, NO₃, NH₄, and PO₄ reflect the integrated effects of biological, chemical, and

physical processes across the pelagic-benthic system. This confirms that EMAGIN-B.C.-MR meets the necessary conditions





for realistically representing the Tokyo Bay ecosystem, supporting its application to mechanistic analysis of hypoxia-driven carbon dynamics.

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Table 3. Analysis of correlation coefficients (R) and root mean square errors (RMSE) between observed and calculated values in EMAGIN-B.C.-MR

Item	Water depth	Correlation	RMSE	Observed averaged	P-value	Number of data
		coefficient (R)		value		points
DO	Surface layer	0.47	1.96	8.85	P < 0.01	232
	Bottom layer	0.83	1.64	5.43	P < 0.01	232
NO ₃	Surface layer	0.73	0.22	0.41	P < 0.01	232
	Bottom layer	0.60	0.14	0.27	P < 0.01	232
NH4	Surface layer	0.59	0.25	0.34	P < 0.01	232
PO ₄	Surface layer	0.66	0.03	0.05	P < 0.01	232
	Bottom layer	0.51	0.03	0.05	P < 0.01	232
PP	Surface layer	0.46	0.93	0.76	P < 0.01	232
POC	Surface layer	0.80	0.81	2.99	P < 0.01	100
	Bottom layer	0.41	0.59	1.85	P < 0.01	78



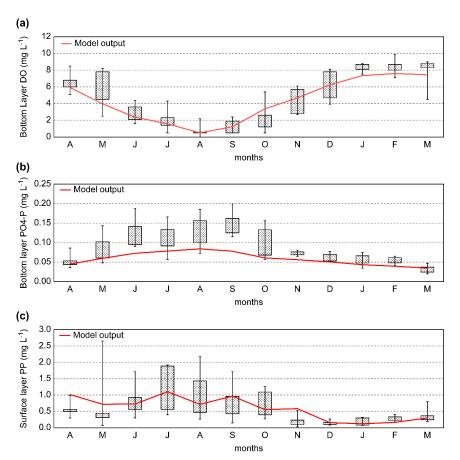


Figure 4. Comparison of calculated (model output) and observational data [region (i, j) = (6, 5) in Figure 3]. a. Comparison of dissolved oxygen (DO) concentrations in the bottom layer. b. Comparison of phytoplankton (PP) in the surface layer. c. comparison of PO₄-P in the bottom layer.

5. Sensitivity analysis

To evaluate the long-term effects of summer hypoxia using EMAGIN-B.C.-MR, we conducted a sensitivity analysis based on the verified Base Case. Specifically, two scenarios were defined in which bottom-layer DO in the pelagic system was forcibly set to 0 mg L⁻¹ (Case 1: hypoxic) and 5 mg L⁻¹ (Case 2: non-hypoxic) for 31 days from August 1st to 31st each year (Table 4, Fig. 5).

Both scenarios were initialized using April 1st values from the Base Case steady state, and simulations were run for 100 years.

During each August, the bottom-layer DO was overwritten to 0 mg L⁻¹ or 5 mg L⁻¹ depending on the scenario, whereas all other conditions, including DO, except during August, remained consistent with the Base Case. These scenarios thus represent





hypothetical systems in which Tokyo Bay undergoes annual, recurring hypoxic, or non-hypoxic conditions every August for a century.

These scenarios were not intended to reproduce specific years but rather to represent extreme end-member conditions (0 and 5 mg L⁻¹) to evaluate the long-term sensitivity of benthic carbon dynamics to oxygen availability. In this context, Case 1 (0 mg L⁻¹) and Case 2 (5 mg L⁻¹) were designed as contrasting end-member cases, not as realistic representations of summer DO conditions, to isolate and clarify the mechanistic impacts of oxygen availability on benthic ecosystems and carbon cycling.

In both scenarios, the Banzu tidal flat (i, j = 8, 6) was maintained at 5 mg L⁻¹ throughout the 100-year simulation, as this area is considered unlikely to become hypoxic. This setting ensured a continuous supply of benthic invertebrate larvae from Banzu to other parts of the bay, even under the prolonged hypoxic conditions in Case 1. For both scenarios, annual average values were extracted for simulation Years 1, 10, 30, and 100, and the time-series results were compared among Case 1, Case 2, and the Base Case to quantify the effects of hypoxia on carbon storage. The Base Case had already reached an annual steady state; therefore, the annual averages for Years 1, 10, 30, and 100 were identical to those at Year 0. This serves as the reference in subsequent figures.

Table 4. Sensitivity analysis scenarios

Analysis scenario	Dissolved oxygen (DO) concentration in the bottom waters		
	in August		
Base Case	Average annual variation from 1999 to 2002		
Case 1: When the bottom waters become anoxic during the	$DO = 0 \text{ mg } L^{-1}$		
summer (hypoxic conditions)			
Case 2: When the bottom waters do not become anoxic during	$DO = 5 \text{ mg L}^{-1}$		
the summer (non-hypoxic conditions)			



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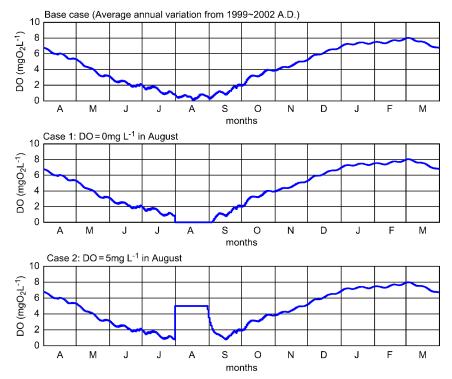


Figure 5. Time series of analysis scenarios, showing dissolved oxygen (DO) concentrations within the bottom layers of the pelagic system. Example: (i, j) = (6, 5).

6 Results and discussion

6.1 Long-term changes in carbon storage function

Figure 6 shows the year-on-year changes in the carbon storage function across the entire calculation domain of Tokyo Bay (728 km², shown in Fig. 3). The figure presents the annual average values for the rate at which carbon moves from a sediment depth of 10 cm to deeper layers (Carbon storage, Flux3 in Fig. 1, M T⁻¹), and the amount of carbon stored within the sediment layer from 0 to 10 cm depth (t C).





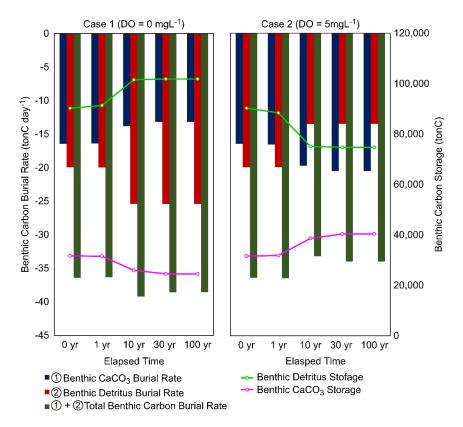


Figure 6. Time-series of changes in detritus storage and burial rates in the benthic system. Burial rate (Flux 3 in Figure 1) represents the annual average total value across the entire bottom layer of the benthic system (the square of the entire bottom layer = 728 km^2 , Figure 3). The amount of stored carbon indicates the annual average total value across the entire benthic system (the volume of the entire benthic system = $728 \text{ km}^2 \times 10 \text{ cm}$ depth). Positive rates represent processes that contribute to an increase in detritus storage, whereas negative rates represent processes that lead to a decline. Note: Base Case values remain constant over time and are therefore not shown separately.

Compared with the Base Case (i.e., the values at 0 yr, since the Base Case remained unchanged over time), the results revealed that under Case 1 (hypoxic conditions), the amount of stored detritus in the benthic system increased over time, whereas there were reductions in the storage of CaCO₃. Similarly, the rate at which detritus was buried showed an increasing trend over time, whereas the rate of benthic CaCO₃ burial declined. The total benthic burial rate, which is the sum of the benthic detritus and benthic CaCO₃ burial rates, peaked in Year 10 and subsequently declined in Years 30 and 100.

For Case 2 (non-hypoxic conditions), we detected trends that were the opposite to those observed in Case 1, thereby indicating that hypoxic conditions enhance the rates of benthic detritus storage and burial, whilst simultaneously reducing the rates of

benthic CaCO₃ storage and burial, with these effects becoming more pronounced over time.





6.2. Mechanistic analysis of benthic detritus dynamics

To elucidate the influence of changes in bottom-water DO concentrations on the detrital carbon burial flux at a sediment depth of 10 cm (Flux 3 in Fig. 1, M T⁻¹) and on the amount of detritus stored in the benthic system to the same depth (M), we conducted a mechanistic analysis of detritus production and consumption processes.

Figure 7 illustrates a conceptual diagram of the production and consumption pathways of benthic detritus, whereas Fig. 8 presents time-series changes in the annual average rates of production and consumption over the entire Tokyo Bay domain.

Differences between Case 1 (hypoxic condition) and Case 2 (non-hypoxic condition) are shown in Fig. 9. The correspondence between each bar chart item and the respective ecosystem processes is also indicated in Fig. 7.

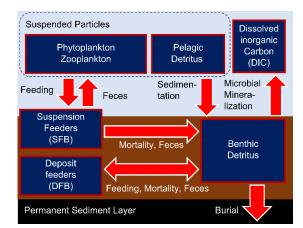


Figure 7. A conceptual diagram illustrating the mechanisms associated with detritus production and consumption in the benthic system.



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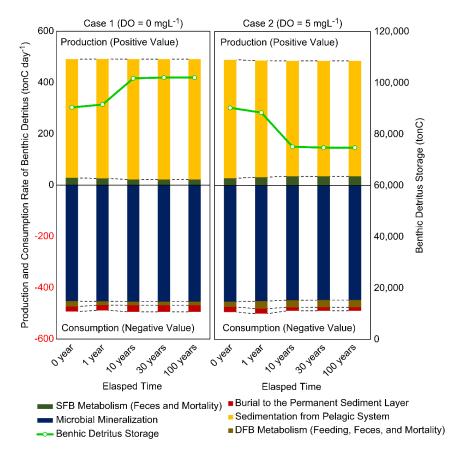


Figure 8. Annual average rates of benthic detritus production and consumption. Total values across the entire benthic system (728 $\rm km^2 \times 10~cm$) (positive for production and negative for consumption). Note: Base Case values remain constant over time and are therefore not shown separately.





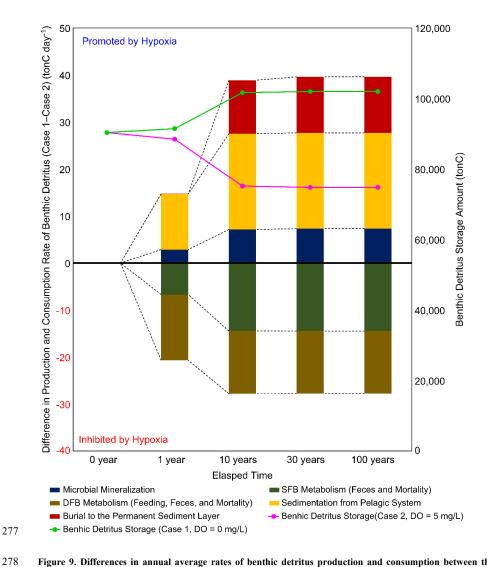


Figure 9. Differences in annual average rates of benthic detritus production and consumption between the Case 1 and Case 2 scenarios (Case 1 - Case 2). Total values across the entire benthic system (728 km² × 10 cm) (positive for production and negative for consumption).

Figure 8 illustrates annual average rates of benthic detritus production and consumption. As noted above, because the Base Case had already reached an annual steady state, the values at Year 0 serve as a constant reference over time. As shown in Fig. 8, the production of benthic detritus is primarily driven by organic matter sedimentation from the pelagic system, whereas consumption is largely governed by microbial mineralization. Biological processes involving SFB and DFB – such as feeding, feces, and mortality – also contribute to detritus production and consumption, but their overall influence is smaller compared to sedimentation and mineralization.





287 Figure 9 illustrates the differences in annual average rates of benthic detritus production and consumption between the Case 1 and Case 2 scenarios (Case 1 - Case 2). The comparison between two cases reveals that, under hypoxic conditions (Case 1), 288 289 detritus production associated with fecal pellet formation and mortality by SFB and DFB (M T-1) was lower than under non-290 hypoxic conditions (Case 2). This reduction is attributable to mass mortality of benthic fauna caused by hypoxia. In contrast, 291 detritus sedimentation from the pelagic layer, mineralization within sediments, and the burial flux of organic matter at a 292 sediment depth of 10 cm (Flux 3 in Fig. 1) all increased under hypoxic conditions. 293 The increase in detritus sedimentation from the pelagic layer (M T-1) is attributed to reduced grazing pressure due to a decline 294 in SFB populations, which led to an accumulation of suspended organic matter in the water column and, consequently, 295 enhanced sedimentation, as well as decreased mineralization efficiency in the pelagic zone under low-oxygen conditions. The increase in mineralization flux within sediments (M T-1) was driven by a decrease in the mineralization rate per unit detritus 296 297 (T-1) under hypoxia, which was offset by an increase in the detritus mass (M) accumulated in the sediment. This seemingly 298 paradoxical outcome - an increased mineralization flux despite a reduced mineralization rate - is explained by the 299 accumulation of detritus in hypoxic sediments. The elevated detritus mass effectively compensates for the reduced per-unit rate, resulting in a net increase in total flux. This relationship is illustrated in Fig. 9, where both individual rate components 300 301 and total fluxes are compared between Case 1 and Case 2. As a result, the total mineralization flux, defined as mineralization flux = mineralization rate × detritus mass (M T⁻¹), was 302 303 greater under Case 1 than Case 2. 304 Furthermore, the combined metabolic changes in microbial and benthic faunal communities under hypoxic conditions 305 enhanced the accumulation of benthic organic matter, leading to an increased burial flux into deeper sediments. 306 These complex mechanisms underlying the hypoxia-induced changes in benthic organic matter storage and burial were made 307 quantifiable for the first time through the integration of an ATP-based microbial metabolic model and a physiological model 308 of benthic faunal activity. This integrated modeling approach represents a key novelty and scientific contribution of the present 309 Figure 10 shows the long-term changes in the rate of mineralization per unit detritus (T-1) and detritus storage in the benthic 310 311 system (sediment depth of 10 cm) at grid point (i, j) = (6, 5). Over the 100-year simulation period, the rate of mineralization 312 per unit detritus was consistently lower in Case 1 (August bottom water DO = 0 mg L⁻¹, representing hypoxic conditions) than 313 in Case 2 (August bottom water $DO = 5 \text{ mg L}^{-1}$, representing non-hypoxic conditions).





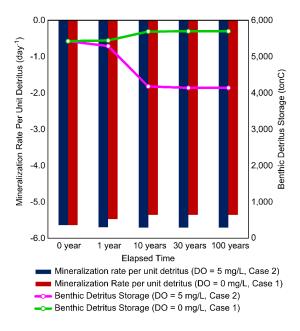


Figure 10. The rates of mineralization per unit weight of benthic detritus. Benthic Detritus Storage Amount represent total value values of benthic system at $(i, j) = (6, 5) (40 \text{ km}^2 \times 10 \text{ cm})$.

In Case 1, detritus storage was characterized by a slight increase from Year 1 to Year 10, followed by a more gradual rise thereafter. In contrast, in Case 2, detritus storage underwent a significant decline from Year 1 to Year 10, with a slower rate of decline thereafter. These results indicate that a reduction in the rate of mineralization per unit detritus promotes an increase in detritus storage, whereas an increase in mineralization rates results in a reduction in detritus storage.

The lack of significant changes in detritus storage after Year 10 can be attributed to the rate of sedimentation in the surface sediments of Tokyo Bay, which is approximately 0.5–1 cm year⁻¹. Consequently, in cases 1 and 2, ecological responses in the vicinity of the sediment—water interface, induced by the changes in bottom water DO concentrations, take approximately 20 years to propagate to a sediment depth of 10 cm, after which the benthic ecosystem reaches equilibrium. The apparent plateau in detritus storage after approximately 10 years reflects a dynamic balance between burial and remineralization, rather than a saturation of storage capacity. Organic matter is gradually transported below the 10 cm layer through sediment mixing and early diagenetic processes, leading to stabilization of vertical fluxes into and out of the upper sediment. Although the carbon in the top 10 cm has not yet reached the permanently buried layer (defined as the permanent sediment layer in the model; see Fig. 7), it actively participates in benthic carbon cycling and determines the near-future burial flux. Therefore, tracking carbon in this transitional layer is essential for assessing long-term carbon dynamics under sustained hypoxic conditions.



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These findings provide evidence to indicate that more severe hypoxia, leading to annual anoxic conditions, would contribute to a decline in SFB due to a deficiency in oxygen. This in turn would reduce the feeding of these organisms on suspended organic matter in the pelagic system, thereby increasing the rate of pelagic detritus sedimentation (M T-1). In addition, within an anaerobic environment, the rate of mineralization per unit detritus (T-1) would decline, further promoting benthic detritus storage, and, as a consequence of a combination of these processes, the rate of detritus burial would increase, ultimately enhancing the carbon storage function. Compared with previous studies, the present model provides a novel process-based formulation of microbial mineralization under varying oxygen conditions. For instance, the model by Sohma et al. (2008) distinguishes between labile and refractory organic matter with different decomposition rates, whereas it did not explicitly incorporate oxygen-dependent microbial respiration mechanisms at that stage. Similarly, Das et al. (2010) represent mineralization as an empirical function of oxygen and temperature, while providing valuable insights without explicitly considering microbial energy metabolism. In contrast, our model explicitly links the rate of organic matter decomposition to the ATP production efficiency under different redox conditions. This enables the simulation of shifts in microbial metabolic strategies e.g., from aerobic to anaerobic pathways, thereby enabling a more realistic representation of carbon mineralization and detritus accumulation under hypoxic stress. Such bioenergetic-based modeling provides deeper insights into benthic carbon cycling and is adaptable to other coastal systems experiencing seasonal hypoxia. Although previous models applied to hypoxia-prone systems such as Chesapeake Bay (e.g., Sturdivant et al., 2013) and the Black Sea (e.g., Yakushev et al., 2007) have explored hypoxia impacts on benthic carbon cycling using empirical or static formulations, our model introduces a mechanistic coupling between microbial respiration pathways and ATP production efficiency. This enables a dynamic representation of organic matter mineralization across redox gradients, thereby advancing beyond existing empirical models in both explanatory depth and cross-system applicability.

6.3 Mechanistic analysis of benthic CaCO₃ dynamics

Similar to the analysis of benthic detritus, the mechanisms associated with the production and consumption of CaCO₃ in the benthic system were analyzed to gain an understanding of the factors contributing to changes in carbon storage function and the long-term variations in carbon storage up to sediment depths of 10 cm (t C). Figure 11 illustrates the processes involved in CaCO₃ production and consumption based on this analysis. Figure 12 illustrates the time-series changes in the annual average



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rates of CaCO3 production and consumption across the entire calculation domain, whereas Fig. 13 shows the differences

between the hypoxic condition (Case 1) and the non-hypoxic condition (Case 2), represented as Case 1 minus Case 2.

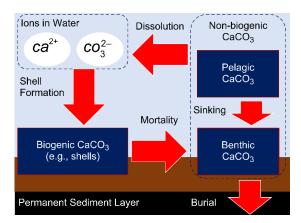


Figure 11. A conceptual diagram of the mechanisms associated with CaCO₃ production and consumption mechanisms in the benthic system.

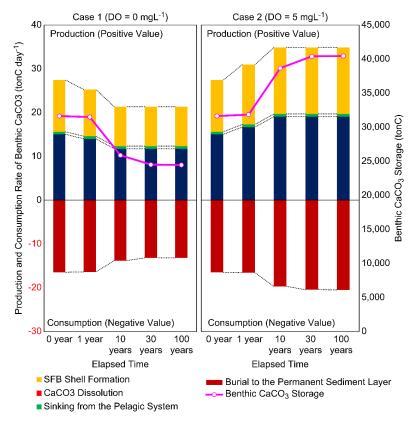


Figure 12. The rates of $CaCO_3$ production and consumption in the benthic system. Total values across the entire benthic system (728 km² × 10 cm) (positive for production and negative for consumption).





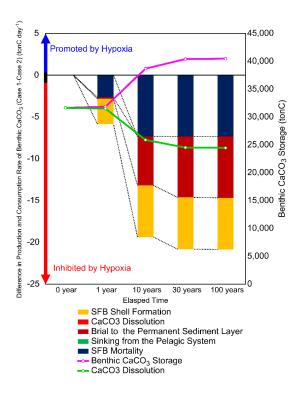


Figure 13. Differences in the rates of CaCO₃ production and consumption in the benthic system between Case 1 and Case 2 (Case 1 – Case 2). Annual average values are provided across the entire benthic system (728 km² × 10 cm), with positive values reflecting increases in production or consumption under specific conditions and negative values showing the converse.

As shown in Fig. 12, it is evident that the rate of benthic CaCO₃ production is largely influenced by SFB shell formation and the deposition of CaCO₃ resulting from SFB mortality, whereas the rate of CaCO₃ consumption is determined primarily by its burial within permanent sediment layers.

Figure 13 shows significant negative values for the rates of shell formation and CaCO₃ deposition due to SFB mortality [representing the rate of CaCO₃ input to sediments (M T⁻¹)]. Compared with that in Case 2, the considerably lower biomass of SFB in Case 1 indicates that hypoxic conditions lead to a substantial reduction in shell formation and a decline in the input of shells to the sediments. Furthermore, the reductions in these rates are more pronounced in Year 10 than in Year 1, which is attributed to the fact that the biomass of SFB (M) is significantly lower after several years of hypoxic exposure, particularly after experiencing repeated hypoxic events each August.

The large negative values for burial rates (M T⁻¹) observed after Year 10 can be ascribed to the near-total extinction of SFB by Year 10 under hypoxic conditions (Case 1), whereas some SFB survived in Year 1. This decline in SFB biomass results in a reduced input of shells (M T⁻¹) to the benthic system, further diminishing the deposition and burial of CaCO₃.



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These findings reveal that hypoxia-induced SFB mortality leads to reductions in shell formation (CaCO₃ production) and the input of shells (CaCO₃) to sediments, thereby resulting in corresponding reductions in the rates of CaCO₃ storage and burial in the benthic system. In many previous models, shell formation and the burial of CaCO₃ have typically been formulated as static or empirical processes, governed by physicochemical parameters such as bottom-water oxygen concentration, sedimentation rate, and carbonate saturation state (e.g., Cartapanis et al., 2018; Dunne et al., 2012; Kanzaki et al., 2021; van Cappellen and Wang, 2002). In contrast, the present model explicitly incorporates dynamic feedback between the biomass of SFB and CaCO3 fluxes - namely, shell formation and burial - enabling a more mechanistic and biologically consistent representation of benthic carbonate cycling. This biologically driven formulation enables the dynamic elucidation of the influence of seasonal or chronic hypoxia-induced variations in benthic faunal populations on carbonate cycling over decadal timescales. By incorporating cumulative hypoxic stress into the suppression of shell formation and SFB recovery, rather than relying on instantaneous DO thresholds, the model enables more realistic projections of long-term carbonate input and burial rates. Moreover, in contrast to empirical diagenetic models that simulate sediment chemistry independently of faunal feedback, our framework explicitly links benthic biological processes with carbon storage functions. These innovations facilitate an integrated understanding of benthic inorganic carbon dynamics, especially under conditions of seasonal or chronic hypoxia, and offer improved mechanistic fidelity compared to earlier models that do not fully resolve oxygen-biomass-carbon feedback.

7. Conclusion

In this study, we describe a novel extension of the benthic–pelagic coupled model EMAGIN-B.C. and demonstrate its application in evaluating the biogeochemical effects of hypoxia on benthic carbon storage. The extended model EMAGIN-B.C.-MR integrates two key mechanisms: (a) redox-sensitive mineralization rates linked to microbial ATP production efficiency, and (b) hypoxia-induced mortality and metabolic suppression in benthic fauna, based on DO saturation thresholds. In addition, the model explicitly couples the biomass of SFB with the production and burial fluxes of CaCO₃, facilitating a biologically driven simulation of carbonate cycling under changing oxygen conditions.

These model enhancements enabled us to capture the compound effects of oxygen depletion on benthic carbon dynamics, showing that detritus and CaCO₃ fluxes are shaped not by any single process but by the interaction of reduced microbial



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degradation and faunal mortality. Application of EMAGIN-B.C.-MR to Tokyo Bay – a eutrophic, hypoxia-prone coastal system – revealed that declining mineralization rates, in combination with increased mortality of SFB, led to enhanced detritus storage and burial. At the same time, shell formation and carbonate deposition were suppressed, resulting in long-term reductions in CaCO₃ burial. These mechanisms were especially pronounced after repeated seasonal hypoxic events and revealed nonlinear, decadal-scale responses in benthic carbon storage.

By explicitly incorporating microbial energetics, DO-sensitive faunal mortality, and faunal-carbonate feedback, the model advances mechanistic understanding of the role of oxygen availability in modulating benthic carbon cycling through intertwined microbial and macrofaunal pathways. These biogeochemical insights are directly relevant for predicting the role of coastal systems in carbon budgets and for informing blue carbon management strategies under progressive deoxygenation.

Moreover, the framework is not limited to Tokyo Bay but is broadly transferable to hypoxia-prone systems worldwide (e.g., Chesapeake Bay, Black Sea, Osaka Bay), thereby supporting cross-system comparisons and future climate change assessments. This transferable framework provides a basis for integrating process-based models into assessments of coastal carbon sequestration under future climate and anthropogenic pressures.

Declaration of generative AI and AI-assisted technologies in the writing process

- During the preparation of this work, the authors used DeepL software to assist in identifying the most appropriate English
- 422 expressions to use. After using this tool/service, the authors reviewed and edited the content as needed and take full
- 423 responsibility for the content of the publication.

Data availability

- 425 Data from the study are available from the corresponding author upon reasonable request. However, any confidential or
- sensitive information will be omitted and cannot be shared.





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427 CRediT authorship contribution statement 428 Akio Sohma: Conceptualization, Data curation, Funding acquisition, Investigation, Methodology, Project administration, 429 Resources, Software, Supervision, Writing - original draft, Writing - review & editing; Kota Ishizuka: Data curation, Formal 430 analysis, Investigation, Software, Validation, Visualization, Writing - original draft, Writing - review & editing. 431 **Declaration of competing interests** 432 The authors declare that they have no known competing financial interests or personal relationships that could have appeared 433 to influence the work reported in this paper. 434 Acknowledgments 435 Part of this study represents Master's thesis research conducted by Kota Ishizuka under the supervision of Akio Sohma. 436 Financial support 437 This study was partially supported by the Environment Research and Technology Development Fund (JPMEERF24S12312) 438 of the Environmental Restoration and Conservation Agency provided by the Ministry of the Environment of Japan, the Fisheries Agency of Japan, and JSPS KAKENHI (18K04409), all granted to A.S. 439 440 References 441 Alongi, D. M., Murdiyarso, D., Fourqurean, J. W., Kauffman, J. B., Hutahaean, A., Crooks, S., Lovelock, C. E., Howard, J., 442 Herr, D., Fortes, M., Pidgeon, E., and Wagey, T.: Indonesia's blue carbon: A globally significant and vulnerable sink for seagrass and mangrove carbon, Wetlands Ecol. Manag., 24, 3-13, https://doi.org/10.1007/s11273-015-9446-y, 443 2016. 444 445 Andersen, F. O.: Fate of organic carbon added as diatom cells to oxic and anoxic marine sediment microcosms, Mar. Ecol.

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