



1    **Measurement report: Year-long chemical composition,**  
2    **optical properties, and sources of atmospheric aerosols in**  
3    **the northeastern Tibetan Plateau**

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5    **Kemei Li<sup>1,3</sup>, Yanqing An<sup>1</sup>, Jianzhong Xu<sup>1,2\*</sup>, Miao Zhong<sup>1</sup>, Wenhui Zhao<sup>1</sup>, Xiang Qin<sup>1</sup>**

6  
7    <sup>1</sup>State Key Laboratory of Cryospheric Science and Frozen Soil Engineering, Northwest Institute of  
8    Eco-Environment and Resources, Chinese Academy of Sciences, Lanzhou 730000, China

9    <sup>2</sup>School of Oceanography, Shanghai Jiao Tong University, Shanghai 200030, China

10    <sup>3</sup>University of Chinese Academy of Sciences, Beijing 100049, China

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15    Corresponding author: Jianzhong Xu ([jzxu78@sjtu.edu.cn](mailto:jzxu78@sjtu.edu.cn); [jzxu@lzb.ac.cn](mailto:jzxu@lzb.ac.cn))

16



## 17 Abstract

18 Due to significant climatic effects, brown carbon (BrC) aerosol has received much attention in  
19 recent years. In this study, a year-long fine particular-matter ( $PM_{2.5}$ ) samples were collected at  
20 Waliguan Baseline Observatory in the northeast of the Tibet Plateau to investigate the optical  
21 properties of water-soluble BrC and its source. The average concentration of  $PM_{2.5}$  throughout the  
22 year was  $10.3 \pm 7.4 \mu g m^{-3}$ , with maximum in spring ( $14.0 \pm 1.6 \mu g m^{-3}$ ) and winter ( $12.5 \pm 1.6 \mu g m^{-3}$ )  
23  $m^{-3}$ ) and minimum in fall ( $7.95 \pm 0.9 \mu g m^{-3}$ ) and summer ( $7.14 \pm 0.9 \mu g m^{-3}$ ). Organic aerosol (OA)  
24 was the major component accounting for 37.7% on average, followed by sulfate (21.3%), nitrate  
25 (12.1%), and other species. OA and nitrate peaked during winter, while sulfate increased  
26 significantly during summer. Backward trajectory analysis on air mass reveals that the sources of  
27 the polluted air mass were mainly transported from the northeast and east of the sampling site. The  
28 seasonally average carbon-based mass absorption efficiency (MAE) of WS-BrC at 365nm were  $0.92$   
29  $\pm 0.54 m^2 g^{-1}$  in spring,  $0.40 \pm 0.24 m^2 g^{-1}$  in summer,  $0.81 \pm 0.46 m^2 g^{-1}$  in fall,  $0.97 \pm 0.49 m^2 g^{-1}$  in  
30 winter, respectively. Comparison with other results, BrC in this study is weakly absorbed throughout  
31 the year, with that during the summer being the most photobleaching BrC. The chemical  
32 compositions of BrC are further investigated by parallel factorization analysis on the three-  
33 dimensional excitation-emission matrix and positive matrix factorization analysis on OA.



## 1 Introduction

Aerosols, tiny particulate matters suspended in the atmosphere, are critical climate forcing factor, such as on atmospheric radiation and the water cycle (Forster et al., 2021). Crucially, the influence of aerosols on climate is dictated by their physical and chemical properties, including mass concentration, number concentration, and chemical composition, which vary widely and unpredictably. This variability makes in-situ measurements essential for accurately assessing their impact. This is especially important for remote regions where aerosol loading is extremely low and strongly interacted with ambient conditions during transport. For example, during long-range transport, aerosols can interact with gas phase pollutants like nitrogen oxides (NO<sub>x</sub>) and volatile organic compounds (VOCs), which can initiate photochemical reactions leading to the formation of secondary aerosols (Schnitzler and Abbatt, 2018; Fan et al., 2024). These transformations are further influenced by topographic and meteorological conditions that can mitigate the formation and evolution of aerosol characteristics including aerosols' optical properties (Schnitzler et al., 2022).

Aerosol optical properties are key parameters for evaluating their climatic effect. Brown Carbon (BrC) and Black Carbon (BC) represent two key optically sensitive components. BrC is particularly notable for its strong wavelength-dependent light absorption properties, distinct from the more uniform absorption characteristics of BC (Laskin et al., 2015). Originating from a variety of both anthropogenic and natural sources, BrC contributes significantly to the complexity of aerosol interactions within the atmosphere and remine the major uncertainty of aerosol light absorption estimation (Laskin et al., 2015; Yan et al., 2018).

The Tibetan Plateau (TP) is the highest and largest plateau on Earth. Aerosols in this receptor region mainly undergo long-range transport from the source region. Recent studies found that WS-BrC absorptivity in this remote and cold region has longer half-life than those in low altitude regions due to their lower decay-rate during transport (Choudhary et al., 2022). The higher aerosol loading and contribution of BrC on the TP mainly locates its margin due to the short distance from the source regions (Xu et al., 2024b). Qilian Mountains (QLM), situated on the northeastern margin of the TP represent a background region of inland of China. The importance of this region is represented by



63 its crucial hydrological resource for the arid northwestern region, which is essential for the  
64 sustenance of downstream communities and the ecological balance (Chen and Wang, 2009; Liu et  
65 al., 2017; Li et al., 2019). Precipitation in the mountain areas, through aerosol-cloud interaction, is  
66 the major origination (Qi et al., 2022). Xu et al. (2024a) emphasize the anthropogenic emission from  
67 the inland of China significantly increase the concentration of cloud condensation nuclei (CCN) in  
68 the QLM. However, the physical and chemical properties of aerosol in this background region is  
69 limited understood.

70

71 Research focusing aerosol in the QLM has been aroused increased interesting during last ten years  
72 (Che et al., 2011; Zhao et al., 2012; Zheng et al., 2015; Dai et al., 2021; Xie et al., 2022). It was  
73 found that inorganic components, especially for sulfate, accounted for a large proportion in the  
74 aerosol of QLM (Xu et al., 2014; Xu et al., 2015; Zhang et al., 2019; Zhang et al., 2020). Moreover,  
75 organic aerosol (OA) constitutes a significant fraction of the aerosol mass and exhibits significant  
76 chemical aging (Zhang et al., 2019; Zhang et al., 2020). Aerosol concentrations exhibit a notable  
77 seasonal variation. In spring, the QLM is predominantly affected by the prevalence of mineral dust,  
78 while during summer, the region experiences an influence of polluted air masses, which are  
79 conveyed from the northern and northeastern sectors of the TP (Xu et al., 2013). However, previous  
80 studies at the QLM are either short-term or discontinuous which are limited to represent the whole  
81 picture of aerosol properties in this region.

82

83 Located in the southeastern edge of the QLM, Waliguan Baseline Observatory (WLG) stands as a  
84 pivotal research site for understanding the atmospheric environmental variations both locally and  
85 regionally. To gain a deeper insight into the effects of human activities on aerosol in this region, this  
86 study conducted a year-long observation of aerosols at WLG to obtain the chemical composition,  
87 optical characteristics, seasonal variations, and sources of aerosols.



## 88 2 Sample collection and analysis

### 89 2.1 Sampling site

90 The WLG (36°17' N, 100°54' E; 3816 m a.s.l.) located at the top of the Waliguan Mountain in the  
91 northeastern TP, which is belong to the Global Atmosphere Watch (GAW) program of the World  
92 Meteorological Organization (WMO) (Figure 1). The Waliguan Mountain, with a relative elevation  
93 difference of about 600m above the ground (Figure 1b), is an ideal location for studying the  
94 background characteristics of atmospheric environment of inner Asia. The WLG is about 90km west  
95 of Xining, the capital of Qinghai Province with an elevation of ~2300 m a.s.l. The climate at this  
96 region is dominated by a distinct plateau continental climate, marked by pronounced Asian summer  
97 monsoon weather during summer and East Asian winter monsoon during winter. Spring and fall  
98 serve as transitional seasons between these two climatic systems.

### 99 2.2 Aerosol sampling

100 Fine particulate matter (PM<sub>2.5</sub>) filter samples was collected on 47mm diameter quartz filter (PALL  
101 Life Sciences, USA) using a low flow aerosol sampler (Wuhan Tianhong Instrument Co. LTD, TH-  
102 16E) at a flow rate of 16.7 L·min<sup>-1</sup>. Before sampling, the quartz filters were baked in a Muffle oven  
103 at 550°C for 4h to remove the carbonaceous material. After the sampling, each filter was stored in  
104 a filter box packaged with clean aluminum foil. Subsequently, the box was saved in a ziplock bag  
105 and stored at -18°C. A total of 48 filter samples and three blank samples were collected during June  
106 14, 2019 and May 6, 2020. Each sample was collected for 48h every seven days. The blank filters  
107 were obtained by being placed in the sampler for 10min without pumping. In this study, we divided  
108 the sampling period into different seasons as summer (6 June, 2019 to 28 August, 2019), fall (4  
109 September, 2019 to 27 November, 2019), winter (11 December, 2019 to 26 February, 2020), and  
110 spring (4 March, 2020 to 6 May, 2020). The real-time meteorology data monitored by a Vantage  
111 Pro2 (Davis Instruments Corp., Hayward, CA, USA) weather station, including ambient  
112 temperature (T), relative humidity (RH), wind speed (WS) and wind direction (WD) were also  
113 obtained.



## 114 2.3 Chemical analysis

115 A 0.5 cm<sup>2</sup> piece of quartz filter was punched and used to determine organic carbon (OC) and  
116 elemental carbon (EC) content in PM<sub>2.5</sub>. The rest of the filter was extracted by ultrasonication with  
117 22 mL Milli-Q water (18.2 MΩcm) for 40 minutes and filtered by a 0.45μm PTFE filter (PALL Life  
118 Sciences, Ann Arbor, MI, USA). A suite of advanced instruments were employed to analyze the  
119 filtrate, including Ion Chromatography (IC) for water soluble ion speciation, Ultraviolet-Visible  
120 (UV-Vis) spectroscopy for absorbance spectrum of WSOC, Excitation-Emission Matrix (EEM)  
121 fluorescence spectroscopy for assessing fluorescence dissolved organic matter, Total Organic  
122 Carbon (TOC) analyzer for quantifying carbonaceous content, and offline analysis using a High-  
123 Resolution Time-of-Flight Aerosol Mass Spectrometer (HR-ToF-AMS) for detailed aerosol  
124 composition.

### 125 2.3.1 OC&EC

126 The OC/EC analysis was performed using a Thermal/optical Carbon Analyzer (DRI Model 2001;  
127 Desert Research Institute, Las Vegas, NV, USA) with the IMPROVE-A method (Chow et al., 2007).  
128 The 0.5 cm<sup>2</sup> quartz filter was loaded into the instrument, and then incrementally heated to 140°C  
129 (OC1), 280°C (OC2), 480°C (OC3) and 580°C (OC4), respectively, to vaporize OC. Then it was  
130 heated at 580°C (EC1), 740°C (EC2), 840°C (EC3) with oxidizing gas and 98% He/2% O<sub>2</sub> as carrier  
131 gas to vaporize EC. At each designated temperature stage, carbon is oxidized to CO<sub>2</sub> and then  
132 reduced to CH<sub>4</sub> by H<sub>2</sub> catalyzed by MnO<sub>2</sub>. Ultimately, the hydrogen flame ionization detector is  
133 utilized to quantify the concentration of the resulting CH<sub>4</sub>. In some samples, the EC concentration  
134 was lower than the minimum detection limit, and the EC content was not detected.

### 135 2.3.2 Ion chromatography

136 Eight water-soluble ionic species (WSIs) (Na<sup>+</sup>, NH<sub>4</sub><sup>+</sup>, K<sup>+</sup>, Ca<sup>2+</sup>, Mg<sup>2+</sup>, Cl<sup>-</sup>, NO<sub>3</sub><sup>-</sup>, SO<sub>4</sub><sup>2-</sup>) were  
137 determined using two 881 ion chromatography systems (Metrohm, Herisau, Switzerland). The  
138 cation system is facilitated by a Metrosep A Supp 5-250/4.0 column (Metrohm). The eluent  
139 composition consists of 3.2 mM Na<sub>2</sub>CO<sub>3</sub> and 1.0 mM NaHCO<sub>3</sub>, delivered at a flow rate of 0.7 mL



min<sup>-1</sup>. The anionic chromatographic system is facilitated by a Metrosep C4-250/2.0 column. The mobile phase comprises 1.7 mM nitric acid and 0.7 mM dipicolinic acid (DPA), administered at a flow rate of 0.3 mL min<sup>-1</sup>. To ensure optimal separation efficiency, the two columns temperature are maintained at 30°C, respectively. Prior to sample analysis, the instrument undergoes a rigorous calibration process. The IC measurements are determined by analyzing the retention time of the peaks and the integrated peak areas. These analytical parameters are then correlated with the calibration curve, which has been previously established using a series of standard solutions, to ensure accurate quantification of the analytes (Xu et al., 2015).

### 2.3.3 UV-vis

The ultraviolet-visible (UV-Vis) absorption spectrum of the samples was measured over a wavelength range of 200 nm–900 nm, with a resolution of 1 nm, using a dual-beam UV spectrometer (UV-2700, Shimadzu, Kyoto, Japan). Samples were positioned in a quartz cuvette with an optical path length of 1 cm and scanned at a rate of 5 nm s<sup>-1</sup>, utilizing a dual light source system comprising deuterium and tungsten lamps. To correct baseline, the spectra of all samples were subtracted from the mean absorption value of the corresponding sample at 695 nm – 705 nm.

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The absorption coefficient ( $Abs_{\lambda}$ ) is calculated by Eq. (1) (Murphy et al., 2010).

$$Abs_{\lambda} = (A_{\lambda} - A_{700}) \frac{V_l}{V_a \cdot l} \cdot \ln(10) \quad (1)$$

where  $A_{\lambda}$  (M m<sup>-1</sup>) is the absorption coefficient at a specific wavelength;  $A_{700}$  is the mean absorption value at 695 nm–705 nm;  $l$  is the light distance of the samples during the determination;  $V_l$  is the volume of water used in extraction;  $V_a$  is the volume of gas that passes through the quartz filter. In general, the absorption coefficient at wavelength 365 nm is used to refer to the absorption of brown carbon. The wavelength dependence of brown carbon absorption can be expressed by Eq. (2).

$$Abs_{\lambda} = K \cdot \lambda^{-AAE} \quad (2)$$

where  $K$  is a constant related to aerosol mass concentration;  $AAE$  is the absorption Ångström exponent of particulate matter, which is obtained by linear fitting the natural logarithm of the wavelength (300 nm–400 nm) to the natural logarithm of the corresponding  $Abs_{\lambda}$ . To calculate the



168 light absorption intensity of unit mass WSOC at a certain wavelength, the mass absorption cross  
 169 section (MAE) is calculated by Eq. (3).

$$170 \quad MAE = \frac{Abs_{\lambda}}{C_{WSOC}} \quad (3)$$

171 where  $C_{WSOC}$  ( $\mu\text{gC m}^{-3}$ ) is the concentration of water-soluble organic carbon in the atmosphere.

### 172 2.3.4 EEM

173 The three-dimensional excitation-emission matrix (3D-EEM) fluorescence of samples was scanned  
 174 by an F-7100 fluorescence spectrometer (Hitachi High-Technologies, Tokyo, Japan), using 700-V  
 175 xenon arc lamp as the excitation source. During the scanning process, the excitation (Ex)  
 176 wavelengths ranged from 200 to 450 nm with intervals of 5 nm, while the emission (Em)  
 177 wavelengths spanned from 250 to 600 nm with intervals of 1 nm. In this study, Milli-Q water ( $18.2$   
 178  $\text{M}\Omega \text{ cm}^{-1}$ ) was used as a blank value for reference. Blank reference was subtracted from the EEM  
 179 fluorescence spectra of the samples to eliminate the impacts of instrument to mitigate the  
 180 instrumental effects. Once the EEMs were corrected, they were converted to Raman units (R.U.),  
 181 after which the EEMs were subjected to modeling analysis (Murphy et al., 2013). Parallel factor  
 182 analysis (PARAFAC), which is a three-way method, divides organic matter into different  
 183 components based on the similarity of fluorescence characteristics. In addition to knowing the  
 184 relative contribution of each component to the total fluorescence of organic matter, this method also  
 185 provides information on the biochemical composition, origin, and biogeochemical action of the  
 186 samples (Fellman et al., 2010). PARAFAC modeling was performed using DOMfluor and drEEM  
 187 toolboxes installed on Matlab R2019a in this study. The entire process encompassed several stages:  
 188 data preprocessing and preliminary analysis, followed by model construction and validation,  
 189 culminating in the presentation of the final results (Stedmon and Bro, 2008).

190

191 In this study, humification index (HIX) and biological index (BIX) were used to analyze  
 192 fluorescence spectral characteristics (Yang et al., 2020; Zhai et al., 2022). Notably, the HIX values  
 193 exhibit disparities due to the distinct origins and transformation pathways of aerosol and aquatic  
 194 samples. Therefore, the emission wavelength selected for HIX calculation was adjusted (from the  
 195 commonly used 300-345nm to 325-365nm, and 435-480nm to 410-450nm) (Wen et al., 2021; Wu





et al., 2021). The above two optical indices can be calculated from Eq. (4) and Eq. (5), respectively (Zsolnay et al., 1999).

$$HIX = \frac{\sum SFI(410nm \leq \lambda_{Em} \leq 450nm)}{\sum SFI(325nm \leq \lambda_{Em} \leq 365nm)} (\lambda_{Ex} = 225nm) \quad (4)$$

$$BIX = \frac{SFI(\lambda_{Em} = 380nm)}{SFI(\lambda_{Em} = 430nm)} (\lambda_{Ex} = 310nm) \quad (5)$$

where  $\lambda_{Em}$  is the emission wavelength;  $\lambda_{Ex}$  is the excitation wavelength.

### 2.3.5 WSOC

Water-soluble organic carbon (WSOC) was measured by a total organic carbon analyzer (Elementar vario TOC cube, Hanau, Germany). The measurement was conducted by applying the total carbon (TC) and total inorganic carbon (TIC) method ( $TOC = TC - TIC$ ). With oxygen as the carrier gas and platinum as the catalyst, inorganic carbon was transformed into CO<sub>2</sub> gas following acidification with 4% phosphoric acid. The concentration of CO<sub>2</sub> was determined using a non-infrared gas detector integrated within the instrument. Prior to the measurement, the total organic carbon (TOC) analyzer was calibrated with standard solutions of potassium hydrogen phthalate and sodium carbonate to ensure accurate quantification (Zhang et al., 2017).

### 2.3.6 HR-ToF-AMS off-line analysis and PMF source decomposition

The High-Resolution Time-of-flight Aerosol Mass Spectrometer (HR-ToF-AMS, Aerodyne Inc., Billerica, MA, USA) can obtain the information of chemical composition and particle size of non-refractory aerosol in real Time. HR-ToF-AMS mainly measures particles in the particle size range of 40–1000 nm. The instrument can not only observe aerosols online, but also analyze atomized aerosol extracts offline (Xu et al., 2015). Using argon as carrier gas, the samples were aerosolized and collected. The aerosol particles enter the HR-ToF-AMS through an aerodynamic lens, pass through a vacuum chamber and reach the hot surface at 600°C where they are vaporized instantly. Finally, it is bombarded with a 70 eV electron source and ionized into positively charged ion fragments, which enter the mass spectrometer for the detection of chemical components. According



221 to the different shapes of ion flight paths in the mass spectrum, HR-ToF-AMS has two operating  
 222 modes, namely V mode and W mode. By comparing the data of W mode and V mode, we choose  
 223 the data of V mode for the subsequent analysis. The data is processed using standard ToF-AMS data  
 224 analysis software (Igor Pro 6.37). The software includes standard data processing toolkits  
 225 SQUIRREL (v1.56) and PIKA (v1.15c). The processed matrix data were employed to investigate  
 226 the sources of WSOA by positive matrix factorization (PMF). PMF source analysis is usually  
 227 processed using the standard PMF evaluation tool (PET v2.03) developed based on Igor Pro  
 228 software and the PMF2.exe algorithm (Ulbrich et al., 2009). Based on the Improved Ambient (I-A)  
 229 method, relevant information of elemental analysis including oxygen-carbon ratio (O/C), hydrogen-  
 230 carbon ratio (H/C), nitrogen-carbon ratio (N/C) and ratio of organic matter to organic carbon  
 231 (OM/OC) can be obtained. The mass concentration of OM can be calculated using Eq. (6).

$$232 \quad OM = OC \times (OM/OC) \quad (6)$$

233 where  $OC$  is the mass concentration of OC measured by Thermal/optical Carbon Analyzer,  
 234  $OM/OC$  is the ratio obtained from the above.

### 235 2.3.7 Backward Trajectory Model

236 To understand the possible source of the air mass during the sampling, the HYbrid Single-Particle  
 237 Lagrangian Integrated Trajectory (HYSPLIT) model developed by the National Ocean and  
 238 Atmospheric Administration (NOAA) and the Australian Bureau of Meteorology was used to  
 239 calculate and analyze the backward transport trajectory of the air mass (Stein et al., 2015). The  
 240 meteorological data used in this study are Global Data Assimilation System (GDAS) from the  
 241 National Centers for Environmental Prediction (NECP), with  $1^\circ \times 1^\circ$  horizontal resolution. In the  
 242 calculation, the height of observation station is set as 500 meters above the ground from the  
 243 sampling site. Thereafter, hourly backward trajectories were performed for a duration of 72 hours  
 244 to trace the air mass movements. Furthermore, the average backward trajectory cluster of the air  
 245 mass during the sampling period was determined by assessing the spatial distribution similarities  
 246 across all calculated trajectories.

247

248 The concentration-weighted trajectory (CWT) was used to analyze the source of pollution to the



249 sampling site. The CWT is a mixed-trajectory receptor model that combines meteorological  
 250 trajectory nodes (residence time) and pollutant concentrations to trace their contributions to the  
 251 pollution of a recipient site. After the study area was firstly gridded with a resolution of  $0.25^\circ \times$   
 252  $0.25^\circ$ , the CWT value of Grid (i, j) was calculated as follows:

$$253 \quad CWT_{ij} = \frac{\sum_{l=1}^M C_l t_{ijl}}{\sum_{l=1}^M t_{ijl}} W_{ij} \quad (7)$$

$$254 \quad W_{ij} = \begin{cases} 1.0(n_{ij} > 4n_{ave}); \\ 0.7(4n_{ave} > n_{ij} > n_{ave}); \\ 0.42(n_{ave} > n_{ij} > 0.5n_{ave}); \\ 0.05(n_{ij} < n_{ave}) \end{cases} \quad (8)$$

255 where  $CWT_{ij}$  is the average weighted concentration in the cell  $ij$ ;  $M$  is the total number of  
 256 trajectories;  $C_l$  is the pollutant concentration when the trajectory  $l$  through the grid  $ij$ ;  $t_{ijl}$  is the  
 257 time that the trajectory  $l$  stayed in the grid  $ij$ ;  $W_{ij}$  is the weight factor used to reduce the  
 258 uncertainty of the calculation;  $n_{ij}$  is the number of trajectory endpoints of grid  $ij$ , and  $n_{ave}$  is the  
 259 average number of trajectory endpoints. In this way, the CWT model is able to reveal regions that  
 260 contribute significantly to the concentration of pollutants at the receptor site.

### 261 3 Results and discussion

262 During the sampling period, the meteorological conditions exhibited notable seasonal variations  
 263 (Figure 2a). The average air temperature ( $\pm 1\sigma$ ) was  $1.8 \pm 8.3^\circ\text{C}$ , with a daily maximum of  $13.8^\circ\text{C}$   
 264 recorded on July 27, 2019, and a minimum of  $-15.8^\circ\text{C}$  on December 26, 2019. Relative humidity  
 265 (RH) ranged from 10% to 99%, with an average of  $57 \pm 28.1\%$ . Seasonally, the average air  
 266 temperatures were  $-2.7 \pm 5.1^\circ\text{C}$  in spring,  $9.1 \pm 3.5^\circ\text{C}$  in summer,  $-1.6 \pm 6.1^\circ\text{C}$  in fall, and  $-10.0 \pm$   
 267  $3.9^\circ\text{C}$  in winter. Similarly, the average RH values were  $47.0 \pm 29.4\%$  in spring,  $88.3 \pm 12.4\%$  in  
 268 summer,  $68.3 \pm 18.5\%$  in fall, and  $32.0 \pm 16.1\%$  in winter. Wind patterns were predominantly from  
 269 the west during winter, with a step increase from the east during spring, reaching the predominance  
 270 from the east in summer. Fall represented a transitional period (Figure 2a and 2b). Precipitation  
 271 occurred primarily in summer (66.9%), followed by fall (17.2%) and spring (15.0%), with winter  
 272 experiencing the least precipitation (0.9%).



### 273 3.1 Chemical speciation of PM<sub>2.5</sub>

274 The total mass concentration of all species (WSIs + OM + EC) ranged from 2.0  $\mu\text{g m}^{-3}$  to 41.8  $\mu\text{g}$   
 275  $\text{m}^{-3}$  during the study period, with a mean of  $10.3 \pm 7.4 \mu\text{g m}^{-3}$  (Figure 2d). OM was the major  
 276 contributor to aerosol mass concentration with an average contribution of 37.7%, followed by  
 277 sulfate (21.3%), nitrate (12.1%), EC (1.1%), and other inorganic ions, which together accounted for  
 278 29.0% (including 7.5%  $\text{Na}^+$ , 7.6%  $\text{NH}_4^+$ , 1.8%  $\text{K}^+$ , 6.7%  $\text{Ca}^{2+}$ , 0.8%  $\text{Mg}^{2+}$ , and 3.6%  $\text{Cl}^-$ ) (Figure  
 279 2c). The mass concentrations were higher during spring ( $14.0 \mu\text{g m}^{-3}$ ) and winter ( $12.5 \mu\text{g m}^{-3}$ ),  
 280 while relatively lower values were observed in summer ( $7.1 \mu\text{g m}^{-3}$ ) and fall ( $8.0 \mu\text{g m}^{-3}$ ) (Figure  
 281 2c). These seasonal patterns were driven by increased transport of polluted air masses from the east  
 282 in winter and prevalent mineral dust storms in spring. The natural mineral dust reached its peak in  
 283 spring (7.5% of  $\text{Ca}^{2+}$ ) and its minimum in summer (4.1% of  $\text{Ca}^{2+}$ ). The anthropogenic pollution  
 284 markers ( $\text{SO}_4^{2-} + \text{NO}_3^-$ ) accounted for 33.2% of the mass in spring and 32.8% in winter. Among the  
 285 secondary inorganic ions (sulfate, nitrate, and ammonium), sulfate was the most abundant,  
 286 especially in summer, when its proportion reached 28.6%, similar to observations made by our group  
 287 in July 2017 (Zhang et al., 2019). Sulfate formation during summer was mainly attributed to strong  
 288 solar radiation, high humidity, and the heterogeneous reaction of  $\text{SO}_2$  (Luo et al., 2022). In contrast,  
 289 nitrate showed its minimum in summer (10.9%) and its maximum in winter (15.5%), which was  
 290 mainly controlled by temperature-dependent partitioning. The average nitrate concentrations was  
 291  $1.4 \mu\text{g m}^{-3}$  with  $2.0 \mu\text{g m}^{-3}$  in spring,  $0.8 \mu\text{g m}^{-3}$  in summer,  $0.9 \mu\text{g m}^{-3}$  in fall,  $1.9 \mu\text{g m}^{-3}$  in winter  
 292 in this study, which are comparable to measurements at WLG in July 2017 ( $0.7 \mu\text{g m}^{-3}$ ) (Zhang et  
 293 al., 2019) and at sites around the region, such as Qinghai Lake in the summer of 2010 ( $0.8 \pm 0.5 \mu\text{g}$   
 294  $\text{m}^{-3}$ ) (Li et al., 2013) and Menyuan in autumn 2013 ( $1.7 \mu\text{g m}^{-3}$ ) (Han et al., 2020). However, these  
 295 concentrations are significantly higher than those recorded in the western Qilian Mountains, such  
 296 as the summer 2012 observation at the Qilian Shan Station of Glaciology and Ecologic Environment  
 297 (QSS) ( $0.6 \mu\text{g m}^{-3}$ ) (Xu et al., 2015).

298

299 Ion balance, represented by the ratio of cation equivalent concentration (CE,  $\text{neq m}^{-3}$ ) to anion  
 300 equivalent concentration (AE,  $\text{neq m}^{-3}$ ), was used to assess potential missing ions or the acid-base  
 301 properties of aerosols (Xu et al., 2014; Xu et al., 2015). The CE/AE ratio calculated in this study



was 1.43 (Figure 3a), suggesting the potential presence of acidic aerosols, although carbonate and bicarbonate ions were not measured in the IC analysis. Assuming  $2[\text{HCO}_3^-] = [\text{Ca}^{2+}]$ , the estimated CE/AE is still 1.35. In addition, the ratio of  $[\text{SO}_4^{2-} + \text{NO}_3^-]$  to  $[\text{NH}_4^+]$  was 1.94, indicating that there was an excess of sulfuric and nitric acids. The acidic property in the aerosol of our study can be further supported by a significant number of organic acids, such as oxalic acid (Figure 3b). Oxalic acid is a product of atmospheric photochemical aging and is closely associated with sulfate and liquid water (Yang et al., 2009; Huang et al., 2019; Xu et al., 2020b; Boreddy et al., 2023). A moderate correlation was found between oxalic acid peak area and sulfate during summer ( $R^2=0.4$ ) (Figure 3c).

Air mass backward trajectory analysis enables the initial tracing of potential sources and transport pathways of atmospheric aerosols throughout the observation period (Figure 4). Air mass origination varied from east to west seasonally, with the east mainly occurred during the summer transported with a shorter distance and the west during winter with a longer distance. Specifically, the fraction of the air mass from the east was up to 50.5% in spring and 66.0% in summer and the potential source areas for pollutants were predominantly associated with these air masses (Figure 5). The less important source areas are also observed from the north and west, especially during the fall, when the climatic systems of summer monsoon and the westerlies interacted. In these directions, widely distributed mineral dust source areas and sparse urban cities are located. Overall, anthropogenic emissions located in the east of WLG emerge as the most significant sources to the WLG.

### 3.2 Optical properties of WS-BrC

The average absorption coefficient ( $\text{Abs}_{365}$ ) of WS-BrC at 365nm was  $1.15 \pm 0.97 \text{ Mm}^{-1}$ . The  $\text{Abs}_{365}$  was much higher in spring and winter than in summer and fall ( $1.55 \pm 1.30 \text{ Mm}^{-1}$  in winter and  $1.45 \pm 0.54 \text{ Mm}^{-1}$  in spring vs.  $0.88 \pm 0.70 \text{ Mm}^{-1}$  in fall and  $0.36 \pm 0.21 \text{ Mm}^{-1}$  in summer), which is consistent with the distribution of OM mass concentration. The average absorption efficiency of WS-BrC at unit WSOC content (MAE) during the summer at 365nm ( $\text{MAE}_{365}$ ,  $0.40 \pm 0.24 \text{ m}^2\text{g}^{-1}$ ) is significantly lower than that of the other three seasons ( $0.92 \pm 0.54 \text{ m}^2\text{g}^{-1}$  in spring,  $0.81 \pm 0.46$



330  $\text{m}^2\text{g}^{-1}$  in fall and  $0.97 \pm 0.49 \text{ m}^2\text{g}^{-1}$  in winter) (Figure 6a), suggesting highly photobleaching of BrC.  
 331  $\text{MAE}_{365}$  in summer is comparable to that at WLG ( $0.48 \text{ m}^2\text{g}^{-1}$ ) in July 2017 (Xu et al., 2020a), Nam  
 332 Co ( $0.38 \text{ m}^2\text{g}^{-1}$ ) from May 13 to July 1, 2015 (Zhang et al., 2017) and the regional background  
 333 points of North China Plain ( $0.38 \text{ m}^2\text{g}^{-1}$ ) in summer of 2017 (Luo et al., 2020). But the  $\text{MAE}_{365}$  in  
 334 spring of this study ( $0.92 \pm 0.54 \text{ m}^2\text{g}^{-1}$ ) is at a high level over the TP and even higher than the  
 335 Qomolangma Station (QOMS) ( $0.81 \text{ m}^2\text{g}^{-1}$ ) which is frequently impacted by biomass burning  
 336 emission (Xu et al., 2020a).

337

338 AAE of light absorption spectrum is an important optical parameter to check the containing of BrC  
 339 in aerosols. In the 300–400 nm range, a high AAE value indicates significant aerosol absorption of  
 340 shortwave ultraviolet light, with a relatively higher contribution from BrC. This phenomenon is  
 341 typically observed in cases from biomass burning emission, secondary organic aerosols (SOA), and  
 342 anthropogenic pollutant emissions (Siemens et al., 2022; Tao et al., 2024). The AAE (300 nm – 400  
 343 nm) in this study ranges from 3.06 to 8.42, with an annual average of  $5.42 \pm 1.26$  peaking in summer  
 344 at  $6.21 \pm 1.50$ , followed by  $5.48 \pm 0.96$  in winter,  $5.19 \pm 1.00$  in fall, and  $5.14 \pm 1.46$  in spring  
 345 (Figure 6a). The average annual AAE is comparable with the observation at Lulang ( $5.39 \pm 1.22$ )  
 346 during August 2014 to August 2015 at the southeastern TP (Li et al., 2016). The summertime AAE  
 347 is similar to those at other stations in TP, such as Nam Co ( $5.91 \pm 2.14$ ) from May 13 to July 1, 2015  
 348 (Zhang et al., 2017) and WLG ( $5.96$ ) from July 2017, but lower than those observed at QOMS ( $6.83$ )  
 349 from April 12 to May 12, 2016 (Xu et al., 2020a).

350

351 Figure 6b illustrates the comparison of optical properties of WSOA in the map space of AAE (300  
 352 nm–400 nm) versus the logarithm of  $\text{MAE}_{365}$  proposed by Saleh (2020). This map can be  
 353 categorized into four classes as  $\text{MAE}_{365}$  increase and  $\text{AAE}_{300-400}$  decreases, which are associated  
 354 with increased molecular sizes, decreased volatility, reduced solubility in water/organic solvents,  
 355 and lower susceptibility to photobleaching. All our collected samples fall within the weakly  
 356 absorbing BrC (W-BrC) category. This result is consistent with previous findings from QOMS and  
 357 WLG reported by our group (Xu et al., 2022). In addition, the samples collected from other stations  
 358 across the TP, including Nam Co station during summer (Zhang et al., 2017), Lulang station during



summer and winter (Wu et al., 2020), and Xining urban station during winter (Zhong et al., 2023), were also distributed in the W-BrC category. These results suggest that the samples at WLG during four seasons are aged BrC. Note that lower AAE and higher MAE<sub>365</sub> observed in spring were closer to moderately absorptive brown carbon (M-BrC) suggesting less oxidization.

### 3.3 Fluorescent components and fluorescence indices

PARAFAC analysis identify four components (C1-C4) in this study (Figure 7a). The chemical properties of each component are determined based on the comparison with previous studies (Chen et al., 2016a; Chen et al., 2016b; Chen et al., 2020; Yu et al., 2023; Zhong et al., 2023). C1 is determined as high-oxidation humus (HULIS-1) with the peaks of Ex and Em at 240 nm and 413 nm (Ex/Em = 240/413 nm) (Tang et al., 2024). C2 (Ex/Em = 225/375 nm) is classified as low-oxidation humus (HULIS-2), which is generally associated with combustion source (Li et al., 2022; Afsana et al., 2023). Both C3 (Ex/Em = 280 /358 nm) and C4 (Ex/Em = 225(270)/297 nm) were classified as protein-like organic matter (PLOM) (Wang et al., 2024). C3 is probably a fossil fuel-related substance (Wu et al., 2019) and C4 has a main peak and a secondary peak similar to the characteristics of tyrosine-like chromophore (Chen et al., 2016b; Chen et al., 2021b). HULIS compounds (C1 and C2) dominated the annual average contribution by 57.9%, of which C1 accounted for 22.9% and C2 accounted for 35.0%. PLOM contributed an average of 42.1%, with C4 accounting for 27.0% and C3 being 15.1% (Figure 7b). C1 presents a weak seasonal variation peaking in summer (23.54%) corresponding to the highest intensity of photochemical oxidation and contributing the least in spring (21.8%). The average relative contribution of C2 was 37.0%, 35.0%, 33.6% and 34.4% in spring, summer, fall and winter, respectively. The average relative contribution of C3 in spring (17.8%) and winter (17.0%) is higher than that in summer (12.6%) and fall (13.1%), which may be related to frequent coal-burning emissions during heating period. In contrast, the contribution of C4 is significantly more pronounced during summer (28.9%) and fall (30.5%) than that in spring (23.4%) and winter (25.1%), Corresponding to enhanced activities in agriculture and ecology (Zheng et al., 2016; Zhang et al., 2020).

An elevated degree of aging in WSOA is associated with an increased HIX value (Fan et al., 2020;



387 Wu et al., 2021; Ma et al., 2022) and a decreased BIX value (Wen et al., 2021). In this study, the  
 388 average HIX and BIX values are  $1.11 \pm 0.18$  and  $1.29 \pm 0.14$ , respectively, with seasonal variations  
 389 of  $1.04 \pm 0.16$  and  $1.39 \pm 0.24$  in spring,  $1.24 \pm 0.11$  and  $1.26 \pm 0.13$  in summer,  $1.13 \pm 0.20$  and  
 390  $1.23 \pm 0.09$  in fall, and  $1.02 \pm 0.17$  and  $1.29 \pm 0.09$  in winter. The spring samples exhibit the greatest  
 391 variability, indicating their fresher properties (Figure 8b). Summer season is characterized by the  
 392 highest HIX and the low BIX, suggesting a high degree of aging and oxidation of WS-BrC. These  
 393 values are positioned in the upper left corner of HIX versus BIX space (Figure 8a). Comparing with  
 394 the results of previous study, the properties of aerosols in this study are more consistent with those  
 395 in the northwestern China (Figure 8) (Chen et al., 2021a; Zhang et al., 2021a; Zhong et al., 2023),  
 396 which is less humified than that in the eastern China.

### 397 3.4 Chemical components of WSOA and their absorption

398 PMF decomposes the WSOA into two factors, i.e., a more oxidized oxygenated OA (MO-OOA) and  
 399 a less oxidized oxygenated OA (LO-OOA) (Figure 9a). The spectra of these two OOAs in this study  
 400 are consistent with those of online measurement at Nam Co Station in the TP during the summer  
 401 (Xu et al., 2018). The average mass contribution of LO-OOA and MO-OOA were 47% and 53%  
 402 (Figure 9c), respectively. The mass contribution of MO-OOA across the four seasons (spring to  
 403 winter) was 55.4%, 54.9%, 61.7% and 42.0%, respectively. The time series of LO-OOA correlated  
 404 well with nitrate ( $R^2=0.47$ ) during winter and less well with sulfate ( $R^2=0.39$ ), while MO-OOA  
 405 correlated poorly with sulfate and nitrate (Figure 9b).

406  
 407 The triangle plot of  $m/z$  44 ( $f_{44}$ ) versus  $m/z$  43 ( $f_{43}$ ) and Van Krevelen diagram of elemental ratios  
 408 are valuable tools for examining ambient evolution of OA (Ng et al., 2010; Zhang et al., 2019;  
 409 Chazeau et al., 2022).  $f_{44}$  is associated highly with oxidized oxygenated OA, while  $f_{43}$  corresponds  
 410 to less oxidized OA. During oxidation, OA transited from a lower to a higher oxidation state,  
 411 characterized by an increase in  $f_{44}$  and a decrease in  $f_{43}$ , moving from the base to the apex of the  
 412 triangular plot (Flores et al., 2014). Most data points in our study locate in the upper section of this  
 413 triangular (Figure 9e) with data points during winter at a lower position and data points during  
 414 summer shifting towards higher position, presenting a distinct oxidation degree at different seasons.





415 The Van Krevelen plot further elucidates the chemical transformations of OA during atmospheric  
 416 aging (Heald et al., 2010; Xu et al., 2018). The transition slope from low to high oxygen states  
 417 typically ranges from  $-1$  to  $-0.5$  (Ng et al., 2011). In our data, the linear regression slope of all data  
 418 points is  $-0.62$  (Figure 9f), higher than winter's  $-0.89$  at Xining and summer's  $-0.76$  at NamCo (Xu  
 419 et al., 2018; Zhong et al., 2023). Seasonal slopes vary, with spring and summer at  $-0.58$ , fall at  $-$   
 420  $0.60$ , and winter at  $-0.66$ , indicating different OA oxidation pathways during each season.

421

422 The light absorption characteristics of different WSOA factors, were evaluated by a multiple linear  
 423 regression (MLR) model to assign the WSOA factors to the  $Abs_{365}$  (Zhang et al., 2021b; Jiang et al.,  
 424 2023). The MLR method can be expressed as Eq. (9).

$$425 \quad Abs_{\lambda} = f_1 \times C_{MO-OOA} + f_2 \times C_{LO-OOA} \quad (9)$$

426 where  $f_n$  is the corresponding fitting coefficients, which can also represent the mass absorption  
 427 cross section (MAC) values of different organic components;  $C_{MO-OOA}$  and  $C_{LO-OOA}$  ( $\mu\text{g m}^{-3}$ )  
 428 are the mass concentration of the organic components;  $f \times C$  is the absorption value of the organic  
 429 component. The  $MAC_{365}$  of the two factors are  $0.41 \text{ m}^2\text{g}^{-1}$  (MO-OOA) and  $0.45 \text{ m}^2\text{g}^{-1}$  (LO-OOA)  
 430 (Figure 6a). The  $MAC_{365}$  value of LO-OOA is slightly higher than that of MO-OOA, which is related  
 431 to the relatively weak photobleaching of LO-OOA. Compared to previous studies,  $MAC_{365, MO-OOA}$   
 432 in this study is lower than  $MAC_{370, MO-OOA}$  ( $0.60 \text{ m}^2\text{g}^{-1}$ ) at the QOMS (Zhang et al., 2021b).  $MAC_{365, LO-OOA}$   
 433 is much lower than that observed in urban stations of Northwest China in winter 2019 ( $1.33$   
 434  $\text{m}^2\text{g}^{-1}$ ) (Zhong et al., 2023), which may be attributed to strong photobleaching of OA in remote areas  
 435 during atmospheric transport.

### 436 3.5 Relationship between oxidation state and optical properties of

#### 437 BrC

438 During the aging process of BrC, changes in its optical properties can reflect alterations in its  
 439 chemical characteristics (Alang and Aggarwal, 2024). In this study, we investigated the relationship  
 440 between  $MAE_{365}$  and the elemental ratios of O/C and H/C across different seasons (Figure10).  
 441  $MAE_{365}$  exhibited a positive correlation with O/C in spring ( $r = 0.63$ ;  $P < 0.01$ ), while an  
 442 insignificant negative correlation was observed in summer and fall ( $r = 0.29$  and  $r = 0.09$ ).



Conversely, the relationship between  $MAE_{365}$  and  $H/C$  showed an opposite pattern in each season (Figure 10b). These results suggest that the light absorption capacity of BrC was enhanced during the oxidation process in spring due to functionalization or oligomerization, while further oxidation in summer and autumn leads to the fragmentation of large molecular weight compounds, resulting photobleaching, which diminishes the light absorption capacity (Jiang et al., 2022).

Furthermore, the optical evolution of WS-BrC during the oxidation process was explored by integrating the PARAFAC components with the WSOA components in EEM plot (Figure 10c). The compounds are categorized based on their correlation analysis among each other: C1 is strongly associated with MO-OOA, whereas C2 and C3 are linked to LO-OOA, and C4 exhibits a weak correlation with these two factors. Through this method, the chemical evolution of different components could be cross-validated and provides additional insights in the plot (Chen et al., 2016b; Zhong et al., 2023). Simply, the transition of less oxidized to highly oxidized OA through photochemical reactions can be applied to the process of BrC. Correspondingly, the optical evolution of BrC can serve as evidence of the oxidative state transition. For our dataset, C1 is likely formed through atmospheric oxidation processes similar to the transition from LO-OOA to MO-OOA, whereas C2 and C3 may originated from primary DOM in less oxidized region. C4 is the protein-like compound and has weak connection with OOA species.

## 4 Conclusions

In this study, atmospheric aerosol samples collected at WLG were analyzed, focusing on their chemical composition, optical properties, and sources. The main conclusions are as follows:

OM is the largest component of  $PM_{2.5}$ , accounting for an average of 37.7% of the mass, followed by sulfate (21.3%) and nitrate (12.1%). Notably, during summer, atmospheric photochemical reactions lead to significant sulfate production. The light absorption capacity of WS-BrC varies seasonally, with the highest levels observed in winter, followed by spring, fall, and summer. In summer, the AAE, and HIX are elevated, likely due to increased oxidation processing of OA. The



471 sources of aerosol to WLG are predominantly from the eastern urban areas.

472

473 Four chromophores are identified based on PARAFAC analysis, with HULIS being the predominant  
474 contributors to fluorescence. PMF analysis on OA revealed two factors of MO-OOA and LO-OOA.  
475 On average, MO-OOA is more dominant in mass concentration; however, its light absorption  
476 capacity is lower than that of LO-OOA. Both factors exhibit reduced light absorption compared to  
477 those in urban studies, indicating a high level of photochemical oxidation at WLG.

478

479 Overall, this study provides valuable insights and serves as a foundational reference for future  
480 research on atmospheric aerosol conditions in the northeastern Tibetan Plateau. The findings will  
481 aid efforts to better understand the background characteristics of aerosols in this region.

482

## 483 Data availability

484 The data used in this study can be accessible at National Cryosphere Desert Data Center  
485 (<https://www.doi.org/10.12072/ncdc.nieer.db6809.2025>).

## 486 Author contributions

487 JX designed the research and KL, MZ, and WZ collected samples. KL and JX processed data, plotted  
488 the figures, and wrote the manuscript when JX and MZ gave constructive discussion. YA and XQ  
489 had an active role in supporting the experimental work. All authors contributed to the discussions  
490 of the results and refinement of the manuscript.

## 491 Competing interests

492 The authors declare that they have no conflict of interests.

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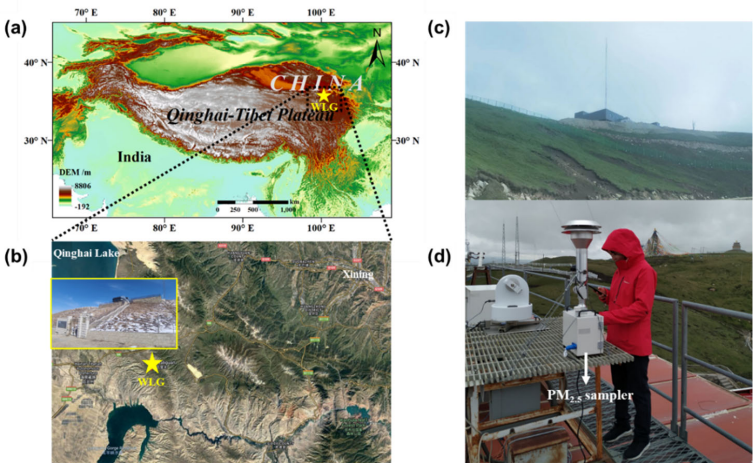


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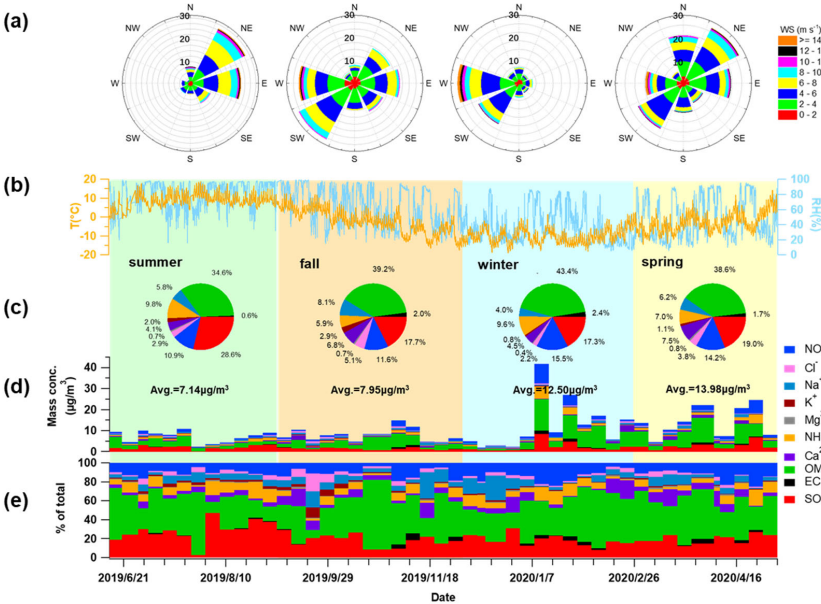
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772 Figure



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774 Figure 1 (a, b) Location map of the Waliguan Baseline Observatory on the TP, adapted from Zhao  
775 et al. (2022) (© Google Maps 2025). (c, d) Photographs of Waliguan Baseline Observatory and *in-*  
776 *situ* PM<sub>2.5</sub> sampling.  
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778

779 Figure 2 (a) Wind rose diagram for four seasons, (b) time series of air temperature (T) and relative  
780 humidity (RH), (c) average chemical composition for four seasons, (d) mass concentration of all



species (WSIs + OM + EC), and (e) percentage of total mass concentration by species.

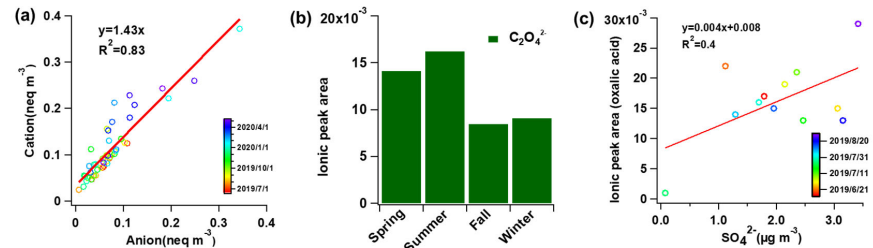


Figure 3 (a) The charge balance between the cations ( $\text{Na}^+ + \text{NH}_4^+ + \text{K}^+ + \text{Ca}^{2+} + \text{Mg}^{2+}$ ) and anions ( $\text{Cl}^- + \text{NO}_3^- + \text{SO}_4^{2-}$ ). (b) Ion peak areas of oxalic acid for four seasons, and (c) correlation between oxalate ion peak area and mass concentration of sulfate during summer.

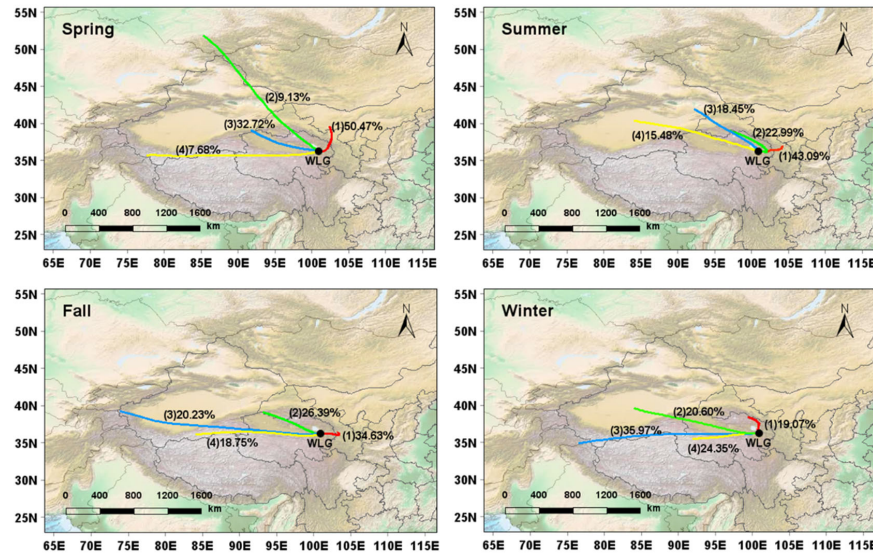


Figure 4 The average backward trajectory clusters and percentage of air mass for the four seasons during the observation period. The map is plotted in Meteoinfo 3.6.0.

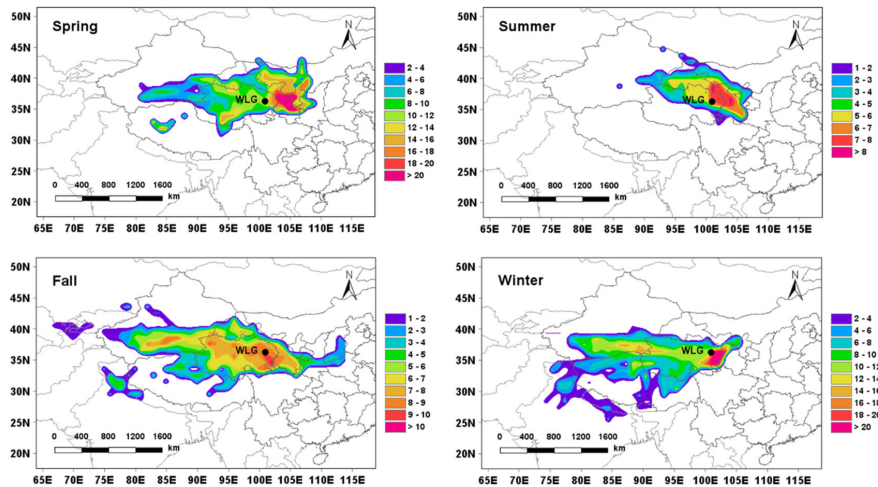


Figure 5 Map of CWT analysis of WLG PM<sub>2.5</sub> in four seasons plotted in MeteoInfo 3.6.0.

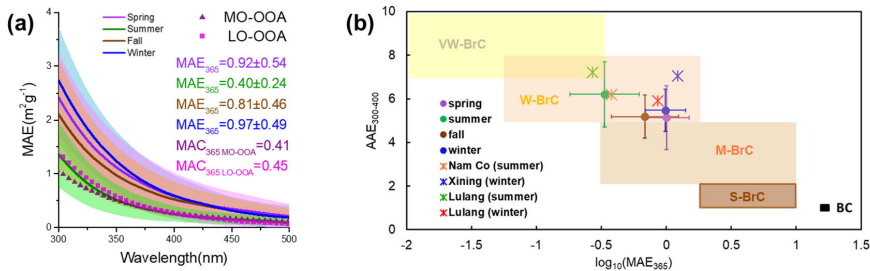


Figure 6 (a) The average MAE spectrum and standard deviations of WS-BrC in different seasons and two factors of WSOA analysed by PMF. (b) Optical-based BrC classification map in AAE- $\log_{10}(\text{MAE}_{365})$  proposed by Saleh (2020). The shaded squares in the map from left to right represent “very weakly” (VW), “weakly” (W), “moderately” (M), and “strongly” (S) absorbing BrC classes and black carbon (BC). The irregular marks are different station data from TP that have been reported by other researchers before.

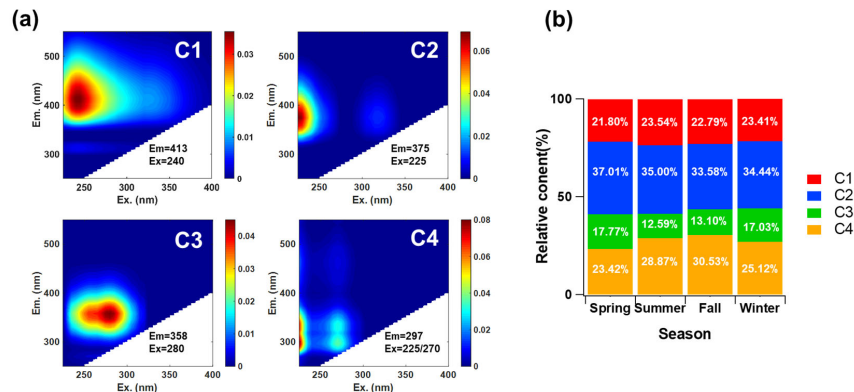


Figure 7 (a) Four EEM components identified by the PARAFAC model for the WSOA and (b) the relative contribution percentage of each component in different seasons.

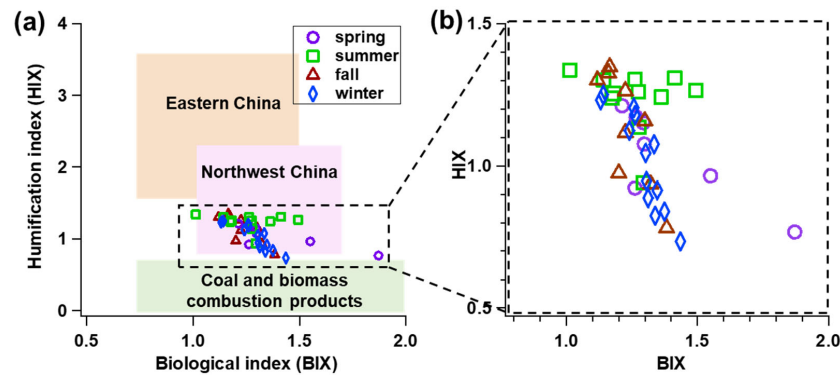


Figure 8 (a) and (b) BIX-HIX distribution map for four seasons, where orange, purple, and green boxes respectively represent the aerosol BIX-HIX range over eastern China, western China, and coal and biomass combustion products summarized by Zhong et al. (2023).



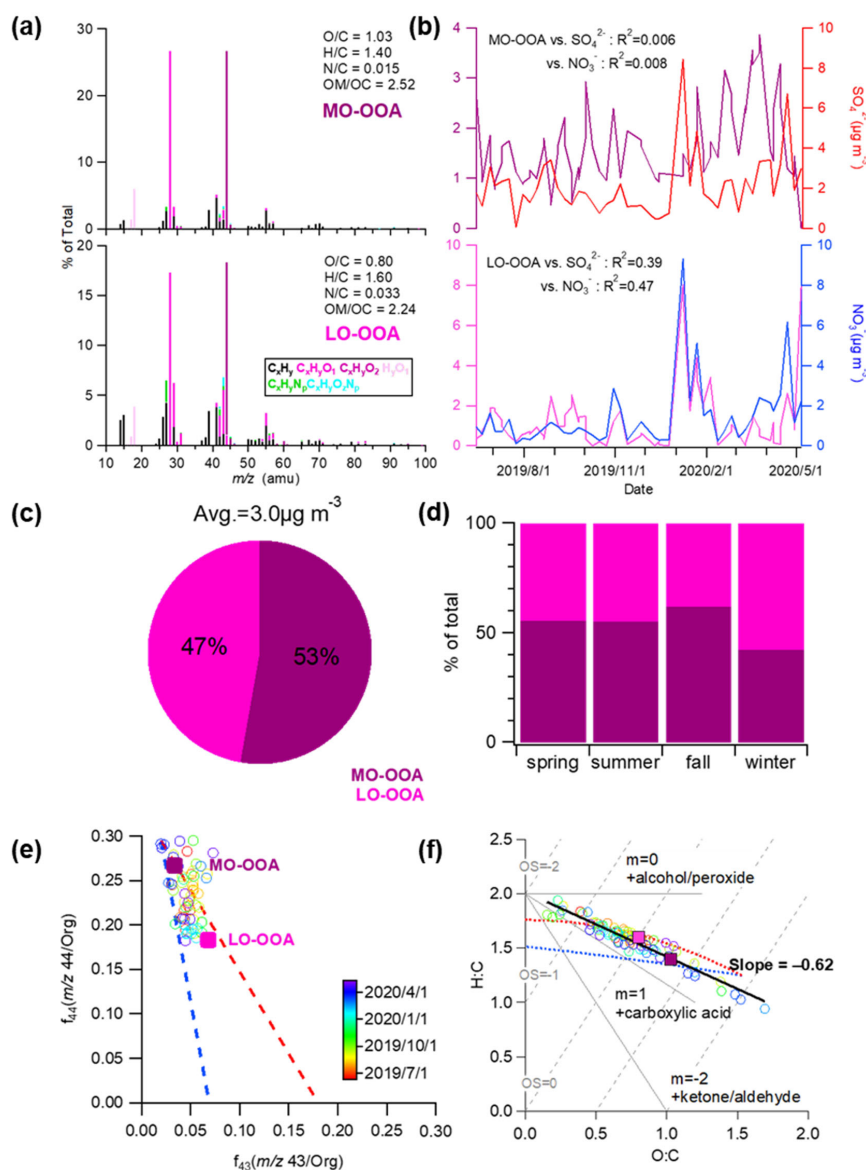
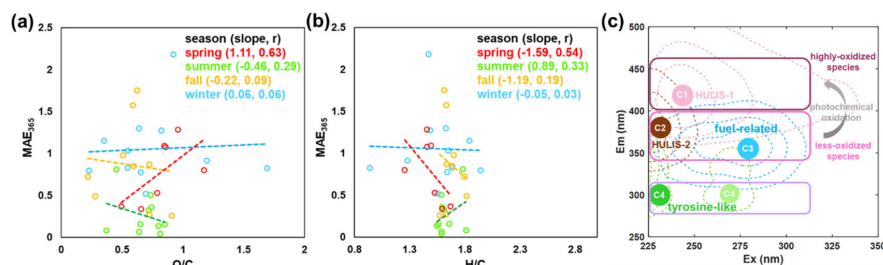


Figure 9 (a) PMF results of high-resolution mass spectra colored by six ion categories for the two OA factors at  $m/z < 120$ , (b) comparison of mass concentration time series changes of the two factors with their correlation of tracer species, (c) the average contribution of mass concentration of each factors to total organics, (d) the contribution percentage of two factors to the total mass in four seasons, (e)  $f_{44}$  vs.  $f_{43}$  triangle plot, and (f) the Van Krevelen diagram (H : C vs. O : C) for the WLG samples and OA components, where the red and blue dashed lines correspond to the same color dashed lines in the  $f_{44}$  vs.  $f_{43}$  triangle plot.



821  
 822 Figure 10 (a) and (b) Scatterplot of  $MAE_{365}$  with O/C and H/C for four seasons. (c) The position of  
 823 fluorescence peak of chromophore and corresponding oxidizing species.





## 824 Table

825 Table 1 Light-absorbing properties of BrC and fluorescence indices of WSOA in four seasons.

Season	Abs <sub>365</sub> (M/m)	AAE <sub>300-400</sub>	MAE <sub>365</sub> (m <sup>2</sup> /g)	HIX	BIX
<b>Spring</b>	1.45 ± 0.54	5.14 ± 1.46	0.92 ± 0.54	1.04±0.16	1.39±0.24
<b>Summer</b>	0.36 ± 0.21	6.21 ± 1.50	0.40 ± 0.24	1.24±0.11	1.26±0.13
<b>Fall</b>	0.88 ± 0.70	5.19 ± 1.00	0.81 ± 0.46	1.13±0.20	1.23±0.09
<b>Winter</b>	1.55 ± 1.30	5.48 ± 0.96	0.97 ± 0.49	1.02±0.17	1.29±0.09

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