

Supporting information of

Comparing Secondary Organic Aerosols Schemes Implemented in Current Chemical Transport Models and the Policy Implications of Uncertainties

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List of Abbreviations

ALK4	Alkanes $5 \times 10^3 < k_{OH} < 1 \times 10^4$ (lumped class)
ALK5	Alkanes $1 \times 10^4 < k_{OH}$ (lumped class)
ARO1	Aromatics $k_{OH} < 2 \times 10^4$ (lumped class)
ARO2	Aromatics $k_{OH} > 2 \times 10^4$ (lumped class)
ASOA	SOA formed from AVOC
AVOC	Anthropogenic VOC
BENZ	Benzene (explicit)
BSOA	SOA formed from BVOC
BVOC	Biogenic VOC
CG	Condensable gases
CTM	Chemical transport model
HUMULE	Humulene (lumped sesquiterpene class)
ISOP	Isoprene (explicit)
IVOC	Intermediate volatility organic compounds
IV-SOA	SOA formed from IVOC
LIMO	Limonene (lumped class)
OA	Organic aerosol
OLE1	Alkenes $k_{OH} < 7 \times 10^4$ (lumped class)
OLE2	Alkenes $k_{OH} > 7 \times 10^4$ (lumped class)
POA	Primary organic aerosol
PM _{2.5}	Particulate matter with a diameter of 2.5 micrometers or less
SESQ	Sesquiterpenes (lumped class)
SOA	Secondary organic aerosol
SVOC	Semivolatile organic compounds
SV-SOA	SOA formed from SVOC
TERP	Terpenes (lumped class)
TOL	Toluene (lumped class)
VOC	Volatile organic compounds
VBS	Volatility basis set
XYL	Xylenes (lumped class)

Table S1 Summary of SOA schemes implemented in CTMs reviewed in this study

Model (version)	SOA scheme	SOA precursors	Aging treatment	POA treatment	SOA photolysis	References
CAMx v7.20	SOAP2		No aging effect	Non-volatile, no further reactions		
	1.5D VBS	BENZ/TOL/XYL ISOP/TERP/SESQ/IVOC	Gas-phase OH-oxidation aging for SOA formation from AVOC and IVOC	Semivolatile; gas-phase undergoes further oxidation	$J_{SOA} = 0.1\% \times J_{NO_2}$	Ramboll (2022) Koo et al. (2014)
	AERO7	BENZ/TOL/XYL ISOP/TERP/SESQ/IVOC	Particle-phase of semivolatile products forms oligomers; products generated by TERP + NO ₃ undergo hydrolysis to form low-volatile products.	Semivolatile; gas-phase undergoes further oxidation	NA	https://github.com/USEPA/CMAQ
CMAQ v5.4	CRACMM	BENZ/TOL/XYM/XYE APIN/LIM/SESQ IEPOX/IVOC	OH oxidation aging resulting in functionalization and fragmentation based on modified 2-D VBS framework	Semivolatile; gas-phase undergoes further oxidation	NA	Pye et al. (2023)
	Simple	ISOP/TERP/SESQ Anthropogenic precursor scaled based on CO emissions	No aging effect	Non-volatile; 50% of POA is directly emitted; 50% is formed with a lifetime of 1.15 days, without dependence on local oxidation levels.	Described in literature but not found in source code	Pai et al. (2020), Pye et al. (2010) https://geos-chem.readthedocs.io/en/stable/
GEOS-Chem v14.3.0	Complex	BENZ/TOL/XYL/IVOC ISOP/MTPO/LIMO/SESQ		Semivolatile; gas-phase undergoes further oxidation with OH to form oxidized POA with lower volatility.		
	VBS	ARO1/ARO2/ALK4/ALK5 /OLE1/OLE2/IVOC ISOP/TERP/HUMULE	OH oxidation aging with both functionalization and fragmentation; the aerosol phase undergoes oligomerization to form non-volatile products.	Semivolatile; gas-phase undergoes further oxidation.	Not available	Zhang et al. (2013); Shrivastava et al. (2015); Couvidat et al. (2018); CHIMERE (2023)

Model (version)	SOA scheme	SOA precursors	Aging treatment	POA treatment	SOA photolysis	References
WRF-Chem v4.4	MOSAIC	ALK4/ALK5/ARO1/ARO2 OLE1/OLE2/IVOC ISOP/TERP/SESQ	No aging for AVOC and BVOC; gas-phase OH-oxidation of IVOC with 15% of mass added for each generation; no fragmentation.	Semivolatile; gas-phase undergoes further oxidation	Described in literature (Zawadowicz et al. 2020) but turned off in source code	Shrivastava et al. 2011

Table S2 Parameters for SOA formation in CAMx SOAP2 modified two-product scheme (adapted from Table 5-5 of Ramboll, 2022)

<i>SOA mass-based yields (g/g) from anthropogenic precursors under low (top)/high (bottom) NO_x conditions</i>			
C* [μg/m ³] @ 300K	0	0.31	14
CG/SOA MW [g/mol]	220	150	160
ΔH _{vap} [kJ/mol]	-	147	116
Benzene	0	0.167	0.487
	0	0.391	0.248
Toluene	0.262	0.345	0.663
	0.044	0.293	0.304
Xylene	0.294	0.306	0.291
	0.025	0.049	0.084
IVOC	0.277	0.275	0
	0.129	0.225	0.012
<i>SOA mass-based yields (g/g) from biogenic precursors under low (top)/high (bottom) NO_x</i>			
C* [μg/m ³] @ 300K	0	0.45	26
CG/SOA MW [g/mol]	220	180	180
ΔH _{vap} [kJ/mol]	-	123	118
Isoprene	0.011	0.029	0.156
	0	0.023	0.076
Monoterpene	0.070	0.090	0.150
	0.070	0.045	0.075
Sesquiterpene	0.270	0.400	0.136
	0.175	0.328	0.092

Table S3 Parameters for SOA formation in CAMx 1.5-D VBS (values adopted from CAMx source code)

C* [$\mu\text{g}/\text{m}^3$] @ 298K	0	1	10	100	1000
CG/SOA MW [g/mol]	172	167	163	158	153
ΔH_{vap} [kJ/mol]	35	35	35	35	35
<i>SOA molar yield (ppm/ppm) under high (top)/low (bottom) NO_x</i>					
Benzene	0	0.001	0.079	0.148	0.222
	0	0.035	0.108	0.185	0.268
Toluene	0	0.006	0.145	0.281	0.432
	0	0.006	0.145	0.437	0.281
Xylene	0	0.001	0.127	0.201	0.301
	0	0.048	0.195	0.252	0.364
Isoprene	0	0	0.009	0.006	0
	0	0.004	0.013	0.006	0
Monoterpenes	0	0.01	0.101	0.173	0.451
	0	0.087	0.077	0.309	0.54
Sesquiterpenes	0	0.092	0.188	0.968	0.679
	0	0.092	0.188	0.968	0.679
<i>SOA molar yield (ppm/ppm) for IVOC</i>					
IVOG	0.022	0.098	0.373	0.699	0
IVOD	0.081	0.135	0.800	0.604	0
IVOA	0.081	0.135	0.800	0.604	0
IVOB	0.081	0.135	0.800	0.604	0
<i>SOA molar yield (ppm/ppm) for NO₃ oxidation</i>					
Monoterpenes	0.314	0.029	0	0.862	0

Table S4 Parameters for SOA formation from aromatics under high- (top row) and low- (bottom row) NO_x conditions in CMAQ AERO7

C* [μg/m ³] @ 298K	0.01	1	10	100
ΔH _{vap} [kJ/mol]	18	18	18	18
CG/SOA MW [g/mol]	198	179	169	158
Precursor	SOA molar yields			
Benzene	0	0.034	0	0.392
	0.146	0	0	0
Toluene	0	0.016	0.051	0.047
	0.14	0	0	0
Xylene	0	0.015	0.023	0.06
	0.193	0	0	0

Table S5 Parameters for SOA formation from monoterpenes and α-pinene in CMAQ AERO7

C* [μg/m ³] @ 298K	0.01	0.1	1	10	100	1000	10000
ΔH _{vap} [kJ/mol]	102	91	80	69	58	47	36
CG/SOA MW [g/mol]	300	200	186	184	170	168	/
SOA molar yields	0.04	0.032	0.032	0.103	0.143	0.285	0.16

Table S6 Parameters for SOA formation from isoprene, sesquiterpenes, and IVOC in CMAQ AERO7

Precursor	Isoprene		Sesquiterpenes	IVOC
C* [μg/m ³] @ 298K	0.617	116.01	24.984	1E-05
ΔH _{vap} [kJ/mol]	40	40	40	40
CG/SOA MW [g/mol]	132	133	273	170
SOA mass yields	0.0288	0.232	1.537	1.176

Table S7 Aging effect for different SOA precursors in CMAQ AERO7

SOA precursors	Aging effect treatment
Aromatics	Under high-NOx conditions, oxidation products form non-volatile oligomers (AOLGA) at a rate of $9.49 \times 10^{-6} \text{ s}^{-1}$; under low-NOx conditions, oxidation products do not form oligomers, i.e., no aging effect.
Isoprene/Sesquiterpene	Aging is NOx-independent; oxidation products form non-volatile oligomers (AOLGB) at a rate of $9.49 \times 10^{-6} \text{ s}^{-1}$
Monoterpenes	There is no aging effect for OH and O ₃ oxidation products; NO ₃ oxidation products undergo hydrolysis to form a non-volatile product (AMTHYD) at a rate of $9.26 \times 10^{-5} \text{ s}^{-1}$
IVOC	No aging effect

Table S8 SOA yield¹ parameterization of GOES-Chem Complex scheme (adopted from Pye et al. 2010)

C* [μg/m3] @ 298K	Non-volatile	Mass yields (g/g) under high (top)/low (bottom) NOx conditions				ΔH _{vap} [kJ/mol]
		0.1	1	10	100	
ISOP	0	0	0.0306	0.0000	0.0945	42
MTPA ²	0	0.04	0.0095	0.0900	0.0150	42
	0	0.08	0.0190	0.1800	0.0300	
LIMO	0	0	0.4743	0.1174	1.4190	42
	0	0	0.3661	0.3214	0.8168	
SESQ	0	0	0.0005	1.1463	2.9807	42
	0	0	0.0000	0.5738	1.4893	
BENZ	0	0	0.0778	0.0000	0.7932	42
	0.37	0	0.0000	0.0000	0.0000	
TOL	0	0	0.0315	0.0944	0.0800	42
	0.30	0	0	0.0000	0.0000	
XYL	0	0	0.0250	0.0360	0.0899	42
	0.36	0	0	0.0000	0.0000	
IVOC ³	0	0	0.0390	0.2960	0.2350	42
	0.73	0	0	0.0000	0.0000	

¹SOA yield for OH and O₃ oxidation. SOA yields from NO₃ oxidation with isoprene and terpenes are not presented in this table.

² α-pinene and similar monoterpenes

³ Use naphthalene as a proxy

Table S9 SOA yield parameterization in CHEMIERE VBS (Zhang et al., 2013; CHIMERE, 2023)

C* [µg/m ³] @ 300K	Mass yields under high (top)/low (bottom) NOx conditions (molar-based) ²				ΔH _{vap} [kJ/mol]	CG/SOA MW [g/mol]
	1	10	100	1000		
ISOP	0.0003	0.0225	0.015	0	30	180
	0.009	0.03	0.015	0		
TERP ¹	0.012	0.1215	0.201	0.507	30	180
	0.1073	0.0918	0.3587	0.6075		
HUMULE	0.075	0.15	0.75	0.9	30	180
	0.075	0.15	0.75	0.9		
ARO1	0.003	0.165	0.3	0.435	30	150
	0.075	0.225	0.375	0.525		
ARO2	0.0015	0.195	0.3	0.435	30	150
	0.075	0.3	0.375	0.525		
ALK4	0	0.0375	0	0	30	120
	0	0.075	0	0		
ALK5	0	0.15	0	0	30	150
	0	0.3	0	0		
OLE1	0.0008	0.0045	0.0375	0.15	30	120
	0.0045	0.009	0.06	0.225		
OLE2	0.003	0.0225	0.0825	0.27	30	120
	0.0225	0.0435	0.129	0.375		

¹ Same mass yields are used for APINEN, BPINEN, OCIMEN, and LIMONE. Use TERP to represent all monoterpene species.

² Yields for OH and O₃ oxidation are the same for biogenic VOCs.

Table S10 SOA yield (g/g) parameterization of WRF-Chem MOSAIC (adopted from Shrivastava et al. 2011).

C* ($\mu\text{g}/\text{m}^3$) @ 298K	Mass yields (g/g) of each bin under high (top)/low (bottom) NOx conditions				MW [g/mol]
	1	10	100	1000	
ISOP	0.001	0.023	0.015	0	250
	0.009	0.03	0.15	0	
TERP	0.012	0.122	0.201	0.5	250
	0.107	0.092	0.359	0.6	
SESQ	0.075	0.15	0.75	0.9	250
	0.075	0.15	0.75	0.9	
ARO1	0.01	0.24	0.45	0.7	250
	0.01	0.24	0.7	0.7	
ARO2	0.01	0.24	0.45	0.7	250
	0.01	0.24	0.7	0.7	
ALK4	0	0.38	0	0	250
	0	0.075	0	0	
ALK5	0	0.15	0	0	250
	0	0.3	0	0	
OLE1	0.001	0.005	0.038	0.15	250
	0.005	0.009	0.06	0.225	
OLE2	0.003	0.026	0.083	0.27	250
	0.023	0.044	0.129	0.375	

Table S11 Ranking (from lowest to highest) initial SOA yield from different precursors by each scheme under high and low NO_x conditions
 (ref. Table 1 for specific numbers)

Ranking	CAMx SOAP2	CAMx VBS	CMAQ AERO7	CMAQ CRAMM	GOES-Chem Simple	GEOS-Chem Complex	CHIMERE VBS	WRF- Chem MOSAIC
<i>High NO_x conditions</i>								
1	ISOP	ISOP	ISOP	XYL	ISOP	ISOP	IVOC	IVOC
2	XYL	TERP	XYL	TOL	TERP	XYL	ISOP	ISOP
3	TERP	BEZN	TOL	SESQ	SESQ	TOL	TERP	TERP
4	IVOC	XYL	BENZ	BENZ		TERP	XYL	BENZ
5	TOL	TOL	TERP	TERP	BENZ	TOL	BENZ	TOL
6	BENZ	SESQ	SESQ			IVOC	TOL	XYL
7	SESQ	IVOC	IVOC			SESQ	SESQ	SESQ
<i>Low NO_x conditions</i>								
1	ISOP	ISOP	ISOP	TERP	ISOP	ISOP	IVOC	IVOC
2	TERP	TERP	TERP	TOL	TERP	TERP	ISOP	ISOP
3	BENZ	TOL	TOL	BENZ	SESQ	TOL	SESQ	TERP
4	IVOC	SESQ	XYL	XYL		XYL	TERP	BENZ
5	SESQ	BENZ	BENZ	SESQ	BENZ	TOL	XYL	TOL
6	XYL	XYL	SESQ			SESQ	BENZ	XYL
7	TOL	IVOC	IVOC			IVOC	TOL	SESQ

Table S12 Initial and aged SOA yields (g/g) for CMAQ CRACMM IVOC types at 298 K and CoA of 10 $\mu\text{g}/\text{m}^3$. The aging effect is for 24-hour exposure to OH concentration of 3×10^6 molecules/cm 3 .

SOA yield	NOx condition	Alkane IVOC ¹	Emitted oxygenated IVOC	Aromatic IVOC ¹	Average ¹
Initial	High	0.296	0.121	0.019	0.209
	Low	0.245	0.121	0.035	0.183
Aged	High	0.077	0.121	0.015	0.099
	Low	0.039	0.121	0.190	0.080

¹ Weighted average of ~C12 (ROCP6ALK), ~C14 (ROCP5ALK), ~C18 (ROCP4ALK), and ~C21 (ROCP3ALK) IVOC yields. Weighting factors for each IVOC category are adopted from Table S1 of Zhang et al., (2013).

² Arithmetic average of less volatile (ROCP5ARO) and more volatile aromatic (ROCP6ARO) IVOC yields.

³Aromatic IVOC are excluded from averaging due to disagreement with Pye et al. (2023) values.

Table S13 Ratio of high/low NOx SOA yields for different precursors in each scheme. Numbers in red indicate higher yields under low NOx conditions (ref. Table 1 for specific numbers).

Precursor	CAMx			CMAQ		GEOS-Chem		CHIMERE	WRF-Chem
	SOAP2	VBS	AERO7	CRACMM	Simple	Complex	VBS	MOSAIC	
BENZ	1.25	0.53	0.39	0.35	/	0.39	0.54	0.89	
TOL	0.52	0.90	0.27	0.19	/	0.28	0.54	0.89	
XYL	0.16	0.51	0.14	0.10	/	0.14	0.51	0.89	
IVOC	0.65	1.00	1.00	1.14	/	0.28	1.00	1.00	
ISOP	0.52	0.47	1.00	/	1.00	1.00	0.53	0.56	
TERP	0.67	0.52	1.00	8.02	1.00	0.50	0.53	0.52	
SESQ	0.74	1.00	1.00	0.12	1.00	2.00	1.00	1.00	

Table S14 Ratio of aged SOA yields ([Error! Reference source not found.](#)2) to initial SOA yields ([Error! Reference source not found.](#)1) from different precursors simulated by different schemes. Shaded values indicate no aging processes implemented. Values in red indicate cases when aging decreases SOA yields.

Precursor	NO _x condition	CAMx			CMAQ		GEOS-Chem		CHIMERE	WRF-Chem	Multi-model averages ²
		SOAP2	VBS	AERO7	CRACMM ¹	Simple	Complex	VBS	MOSAIC	WRF-Chem	
BENZ	high	1.0	6.9	1.6	/	1.0	1.0	4.1	1.0	1.0	2.4
	low	1.0	4.9	1.0	/	1.0	1.0	3.1	1.0	1.0	1.9
TOL	high	1.0	6.9	1.5	/	1.0	1.0	4.1	1.0	1.0	2.4
	low	1.0	6.3	1.0	/	1.0	1.0	3.1	1.0	1.0	2.1
XYL	high	1.0	6.3	1.5	/	1.0	1.0	3.9	1.0	1.0	2.2
	low	1.0	4.4	1.0	/	1.0	1.0	2.9	1.0	1.0	1.8
IVOC	high	1.0	2.2	1.0	/	/	1.0	60	4448	4448	645
	low	1.0	2.2	1.0	/	/	1.0	60	4448	4448	645
ISOP	high	1.0	1.0	1.3	/	1.0	1.0	2.3	1.0	1.0	1.2
	low	1.0	1.0	1.3	/	1.0	1.0	1.8	1.0	1.0	1.2
TERP	high	1.0	1.0	1.0	/	1.0	1.0	4.1	1.0	1.0	1.4
	low	1.0	1.0	1.0	/	1.0	1.0	3.2	1.0	1.0	1.3
SESQ	high	1.0	1.0	1.8	/	1.0	1.0	4.2	1.0	1.0	1.6
	low	1.0	1.0	1.8	/	1.0	1.0	4.2	1.0	1.0	1.6

¹ Ratio for CRACMM is not applicable as the initial SOA yields are zero.

² Excluding CRACMM.

Aerosols: Organic Aerosol in CMAQv5.3 (AERO7)

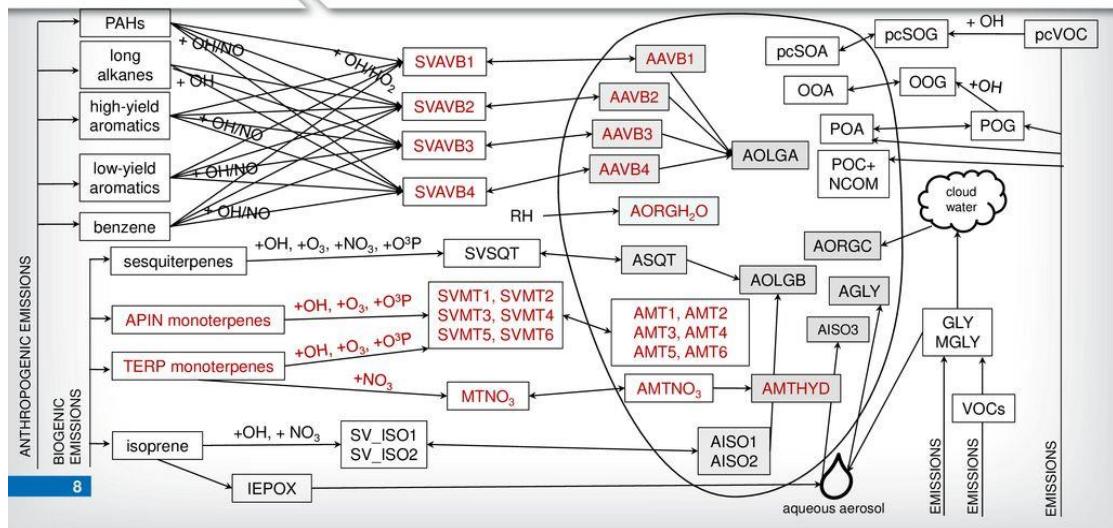


Figure S1 Schematic of AERO7 scheme (Murphy et al. 2018)

References

- CHIMERE Users Guide (2023). <https://www.lmd.polytechnique.fr/chimere/>, accessed on Feb. 15th 2024.
- Couvidat, F., Bessagnet, B., Garcia-Vivanco, M., Real, E., Menut, L., & Colette, A. (2018). Development of an inorganic and organic aerosol model (CHIMERE 2017 β v1. 0): Seasonal and spatial evaluation over Europe. *Geoscientific Model Development*, 11(1), 165-194.
- Pai, S. J., Heald, C. L., Pierce, J. R., Farina, S. C., Marais, E. A., Jimenez, J. L., ... & Vu, K. (2020). An evaluation of global organic aerosol schemes using airborne observations. *Atmospheric Chemistry and Physics*, 20(5), 2637-2665.
- Pye, H. O., Place, B. K., Murphy, B. N., Seltzer, K. M., D'Ambro, E. L., Allen, C., ... & Stockwell, W. R. (2023). Linking gas, particulate, and toxic endpoints to air emissions in the Community Regional Atmospheric Chemistry Multiphase Mechanism (CRACMM). *Atmospheric Chemistry and Physics*, 23(9), 5043-5099.
- Pye, H. O. T., Chan, A. W. H., Barkley, M. P., & Seinfeld, J. H. (2010). Global modeling of organic aerosol: the importance of reactive nitrogen (NO_x and NO₃). *Atmospheric Chemistry and Physics*, 10(22), 11261-11276.
- Ramboll (2022). CAMx User's Guide, Version 7.20. Retrieved from <https://www.camx.com/download/source/>, accessed on Feb. 15th, 2024
- Shrivastava, M., Fast, J., Easter, R., Gustafson Jr, W. I., Zaveri, R. A., Jimenez, J. L., ... & Hodzic, A. (2011). Modeling organic aerosols in a megacity: comparison of simple and complex representations of the volatility basis set approach. *Atmospheric Chemistry and Physics*, 11(13), 6639-6662.
- Shrivastava, M., Easter, R. C., Liu, X., Zelenyuk, A., Singh, B., Zhang, K., ... & Tiitta, P. (2015). Global transformation and fate of SOA: Implications of low-volatility SOA and gas-phase fragmentation reactions. *Journal of Geophysical Research: Atmospheres*, 120(9), 4169-4195.
- Zawadowicz, M. A., Lee, B. H., Shrivastava, M., Zelenyuk, A., Zaveri, R. A., Flynn, C., ... & Shilling, J. E. (2020). Photolysis controls atmospheric budgets of biogenic secondary organic aerosol. *Environmental Science & Technology*, 54(7), 3861-3870.
- Zhang, Q. J., Beekmann, M., Drewnick, F., Freutel, F., Schneider, J., Crippa, M., ... & Perrussel, O. (2013). Formation of organic aerosol in the Paris region during the MEGAPOLI summer campaign: evaluation of the volatility-basis-set approach within the CHIMERE model. *Atmospheric Chemistry and Physics*, 13(11), 5767-5790.