

Culturing experiments reveal mechanisms of daily trace element incorporation into *Tridacna* shells

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Abstract

Giant clams such as *Tridacna* sp. are exceptionally well suited for studying past environmental changes on various timescales, from daily to multidecadal timescales. The visible growth bands in their shells, which can be yearly, seasonal or even daily, are accompanied by changes in the elemental composition of the shell and provide insights into their growth and environmental history. The daily elemental cycles, particularly in Mg/Ca and Sr/Ca, can be used to determine age and growth rates. However, the mechanisms creating the visible day and night banding and the associated elemental cycles, remain unclear. To better understand the mechanisms of El/Ca incorporation into the shells of *Tridacna* shell during day and night growth, we performed controlled growth experiments using ¹³⁵Ba-labelled seawater. The isotope spike was alternately applied in 12-hour intervals in order to individually and unequivocally mark day and night growth segments in *Tridacna*. These experiments show that *Tridacna* calcification rates are nearly five times higher during the day than at night. In addition, based on the observed changes in shell composition we deduce that the bivalve's extrapallial fluid (EPF) reacts to changes in seawater chemistry within tens of minutes, both during day and night. A, with full compositional replenishment is achieved after approximately one day, assuming a similar residence time for all elements. During daytime, El/Ca (for El = B, Mg, Sr, Ba) decrease, while Na/Ca increases. The opposite behaviour occurs at night. The night peak in El/Ca occurs in the earliest morning, shortly before the change between spiked and non-spiked water at 7:30. Daily El/Ca cycles are likely dominantly driven by variations in active Ca²⁺ and HCO₃⁻ transport into the EPF, influenced by light availability, circadian rhythms and/or energy availability (from both photosymbionts and filter feeding), rather than a closed-system Rayleigh fractionation process driven by contrasting El-distribution coefficients alone. We propose that active Ca²⁺ and HCO₃⁻ pumping into the EPF might also drive diurnal changes of growth rate, shell structure and possibly organic content.

1 Introduction

Giant clams serve as archives recording palaeoenvironmental changes in (sub)tropical reefs since their emergence in the middle Eocene, with *Tridacna* emerging in the early Miocene (Harzhauser et al., 2008). Their large, dense aragonitic shells are less prone to diagenetic recrystallisation compared to other reef organisms such as corals (Griffiths et al., 2013; Veech and Chappell, 1970; Welsh et al., 2011). In addition, *Tridacna* clams live for several decades (Arndt et al., 2025; Rosewater, 1965; Watanabe et al., 2004) and grow quickly, at rates ranging

from millimetres to centimetres per year (Arndt et al., 2023; Bonham, 1965; Elliot et al., 2009; Fousiya et al., 40 Fursman et al., 2025; Ma et al., 2020; Mills et al., 2023; Rosewater, 1965; Warter et al., 2018). This combination of excellent preservation potential~~preservability~~ and rapid~~quick~~ growth makes them valuable for reconstructing past environmental changes on timescales from daily to multiannual timescales (Warter et al., 2015; Warter and Müller, 2017; de Winter et al., 2023; Yan et al., 2020, 2021; Zhao et al., 2021)(de Winter et al., 2023; Warter et al., 2015; Warter and Müller, 2017; Yan et al., 2021, 2020; Zhao et al., 2021), even for pre-Pleistocene 45 'deep time' periods. Decade-long, sub-daily-resolved records can be used to examine phenomena such as (palaeo-)ENSO, seasonal aspects of palaeoclimate as well as short-term extreme weather events (Arndt et al., 2025).

Tridacna are mixotrophic clams that can obtain nutrition via filter feeding and photosynthesis from the photosymbionts hosted in the soft tissue (Jantzen et al., 2008; Klumpp et al., 1992; Kunzmann, 2008). However, phototrophy is thought to be the main source of energy for the host (Klumpp and Griffiths, 1994), although the 50 The balance between phototrophy and heterotrophy might vary between~~be adapted depending on~~ species (Jantzen et al., 2008) and turbidity of the habitat (Mills et al., 2023). Overall, light availability is known to have~~has~~ a positive impact on calcification (Rossbach et al., 2019; Sano et al., 2012; Warter et al., 2018).

The shells of *Tridacna* contain visible patterns that provide insights into their growth and environmental history, with growth~~Growth~~ bands representing~~within the shell can represent~~ different time intervals. Banding patterns 55 on the millimetre to centimetre scale are visible to the naked eye and may~~an~~ be yearly growth bands (Ayling et al., 2015; Pätzold et al., 1991; Warter et al., 2015; Welsh et al., 2011) or seasonal, with growth bands, such as two bands per year being observed (Arndt et al., 2025; Ma et al., 2020). Bands on the micrometre scale, visible under a microscope, show daily growth (Aharon and Chappell, 1986; Arndt et al., 2023; Hori et al., 2015; Pätzold et al., 60 1991; Warter et al., 2018; Watanabe and Oba, 1999; Yan, 2020). All of these banding~~Banding~~ patterns within the shell (both seasonal and daily) are linked to changes in the shell's microstructure (Brosset et al., 2025; Mills et al., 2023, 2024)(Brosset et al., 2025; Mills et al., 2024, 2023).

By counting the visible daily growth bands, the age and growth rates of giant clams can be determined (Duprey et al., 2015; Fursman et al., 2025; Gannon et al., 2017; Sano et al., 2012; Zhao et al., 2023)(Duprey et al., 2015; Gannon et al., 2017; Sano et al., 2012; Zhao et al., 2023). *Tridacna* shells also exhibit daily compositional cycles, 65 most prominently in Mg/Ca and Sr/Ca (Brosset et al., 2025; Hori et al., 2015; Sano et al., 2012; Warter et al., 2018; Warter and Müller, 2017; Yan, 2020). These elemental cycles can be detected even if daily bands are poorly visible, such as in fossil giant clams (Arndt et al., 2023). Therefore, elemental ratios can be used to help quantifying growth rates and constrain~~determining~~ the age of the specimen, e.g. using the Daydacna Python script (Arndt et al., 2023; Arndt and Coenen, 2023). However, the causes for the visible day and night banding as well as the 70 related cycles in El/Ca ratios remainsing unclear.

A difference~~Between~~ areas grown at day and night, differences in growth rate (Sano et al., 2012; Warter et al., 2018), shell structure (Agbaje et al., 2017; Brosset et al., 2025; Mills et al., 2024) and organic content (Liu et al., 75 2022) have been observed between areas of the shell grown at day versus night. The varying incorporation of elements into the shell between day and night might be linked to these observations and caused by associated physiological, environmental or chemical factors. However, there~~There~~ is a lack of studies that unequivocally discern El/Ca changes between day and night growth in *Tridacna* under controlled experimental conditions. To the best of our knowledge, the study by Warter et al. (2018) is the only one to date that demonstrated an increase in Sr/Ca and Mg/Ca during nighttime calcification in *Tridacna*, using isotopically-labelled seawater during an

unintended nighttime culturing interruption. However, the study could not rule out potential stress-related factors
80 that might have overprinted the results.

In this study, we expand on the previous work of Warter et al. (2018) and present the results of a specifically
85 designed culturing experiment with *Tridacna* where day- and night-growth periods were individually and
alternatingly marked using the isotopic tracer ^{135}Ba . Via spatially/time-resolved elemental analysis by LA-ICPMS
90 at ~~an unprecedented~~ $<2\text{ }\mu\text{m}$ resolution, the respective identification of day and nighttime trace elemental signatures
of the shell was possible. Isotopic tracers such as ^{135}Ba and ^{87}Sr ~~isotopic tracers~~ have previously been used to
identify specific growth domains in foraminifera (Evans et al., 2016; Fehrenbacher et al., 2017; Hauzer et al., 2018,
2021; Levi et al., 2019) (Evans et al., 2016; Fehrenbacher et al., 2017; Hauzer et al., 2021, 2018; Levi et al., 2019)
and giant clams (Warter et al., 2018), such that this approach is expected to yield new insights into calcification
dynamics. With this tracer experiment, we therefore aim to provide gain unprecedented insights into changes in
growth rate, elemental uptake from the surrounding seawater and element incorporation into *Tridacna* shell the
shell between day and nighttime.

2 Material and Methods

2.1 Culturing

Eight juvenile giant clams with lengths of 3 to 4 cm were purchased from an aquarium supplier and transported to
95 the Hebrew University of Jerusalem, where all culturing experiments took place. The clams' species were
identified by Dan Killam based on ~~Prior to~~ the dichotomous key of Neo (2023), with seven clams being *Tridacna*
maxima and one *Tridacna squamosa* (see Tab. S1). Prior to the actual onset of the controlled day-night culturing

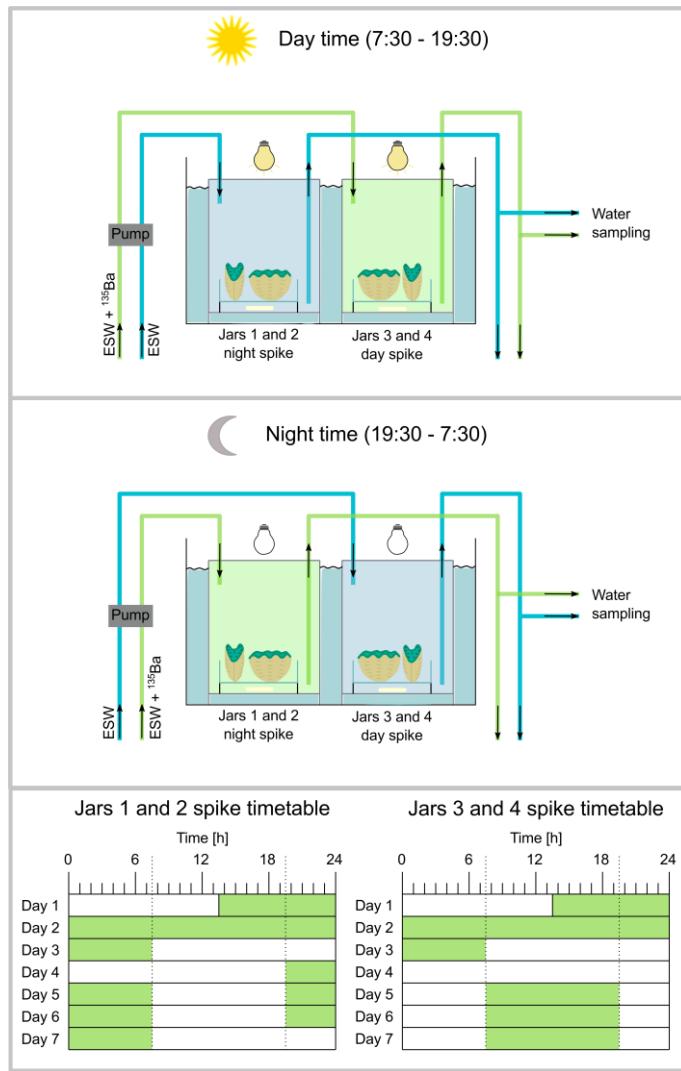
100 experiments (see below), the clams' calcification performance was monitored over 15 days. The culture and
conditions (detailed below) were optimised for high calcification rates accordingly by adjusting water temperature,
alkalinity, light availability and feeding during this interval.

The seawater used for the culturing of the clams was retrieved from the Gulf of Eilat. Its salinity was lowered to
37 on the practical scale ~~psu~~ by mixing 10 l of Eilat sea water (ESW), characterized by a salinity of 40.7 ~~psu~~, with
1 l of distilled water for each of ~~the~~ four 1 l seawater reservoirs, kept in separate plastic containers. Two of the
11 l reservoirs were spiked with 0.55 ml of dissolved $^{135}\text{BaCO}_3$ (240 $^{135}\text{BaCO}_3$ $\mu\text{g}/\text{ml}$; 93.5% enriched; Oak Ridge
105 National Laboratory, USA). In ~~to~~ create the 'spiked' seawater, ~~in which~~ the naturally minor Ba-isotope ^{135}Ba
(6.59% natural abundance) was enriched ~~is~~ more than tenfold ~~enriched~~. Based on a [Ba] value of ESW of 9.9 ng/g
(Evans et al., 2015b) (Evans et al., 2015), this yields a $^{135}\text{Ba}/^{138}\text{Ba}$ of approximately 1.24 in the 11 l of spiked,
modified ESW. However, we stress that ~~No~~ accurate knowledge of this ratio in the culture seawater was not ~~is~~
110 required (other than being approximately ~~ever~~ tenfold higher than the natural $^{135}\text{Ba}/^{138}\text{Ba}$ of 0.092 for ease of
measurement), as it solely serves to ~~distinguish~~ unequivocally distinguish between the respective growth
increments of the cultured clams. Assuming a natural 10% inter-annual variability ~~range~~ of [Ba] in ESW, the
associated $^{135}\text{Ba}/^{138}\text{Ba}$ range of spiked, modified ESW is ~ 1.13 ~~to~~ 1.35, with Ba mass bias contributing another
~1.5% uncertainty only on this isotopic ratio, which is negligible in view of the ESW [Ba] uncertainty. Two further
115 1 l plastic containers with normal, non-spiked seawater were used. Night and day water was separated, hence the
requirement for four different seawater reservoirs, namely 'night ^{135}Ba -spiked water', 'night non-spiked water',
'day non-spiked water' and 'day ^{135}Ba -spiked water'.

In order to supply nutrients to the organisms, food (Fauna Marin, type “coral sprint”) was added to the reservoir water. This fine powder consists of 85% protein, 11% fat, 3% fibre and 1% ash, as well as the following additives per 1 kg: 600 i.u. vitamin D3 (E671), 50 mg iron sulphate monohydrate, 2.2 mg calcium iodate, 6 mg copper sulphate, 17 mg manganese monohydrate, 120 mg zinc monohydrate, and 57 mg antioxidants. The recommended dosage of one measurement cup every two days for 500 l of water was scaled down accordingly for the 11 l water reservoirs. Initially, after purchase, all clams grew in natural ESW in a 400 l large aquarium but were in turn then transferred to culturing jars. To the large aquarium filled with artificial sea water, 10 ml of “Reef Energy Plus” (Red Sea), containing carbohydrates, amino acids, fatty acids, and vitamins, are added daily during the week specific culturing jars.

All clams underwent a 78-hour ‘spiking’ procedure. This included growing in the culturing jars with modified ESW enriched with ^{135}Ba isotope tracer for 42-hours under optimized growth conditions, i.e. minimal stress, after which they were transferred and transferring them back to an ESW aquarium to grow in non-spiked water for 36 hours. The optimized growth conditions included initial alkalinity above 2 mEq/kg, a temperature of 28 °C and coral food being present in added to the water.

During the main experimental period, all eight clams were cultured in pairs in four jars for a duration of 72 hours. The absence of other organisms in the experimental setup is relevant for growth rate analysis via alkalinity measurements, not only to exclude calcification from other organisms but also to avoid the nitrogen cycle impacting alkalinity measurements. The clam sample IDs include the jar number, the number of the clam within the jar and whether a specimen was exposed to spiked seawater during the it is day (DAS) or night (NIS) spike treatment (Tab. 2). For example, sample “1.1 NIS” refers to the first jar, first clam that received ^{135}Ba -spiked water at night. We Please note that the laboratory sample ID (see Tab. S1) is based on markings drawn on the shells for differentiation during culturing, and therefore it differs from the sample ID used here in the paper (see Table S1). The culture jars have with a water capacity of 630 ml, were sealed airtight and placed in a temperature bath. Water was pumped through the jars at with a rate of 210 ml/h, resulting in a 3 h residence time. This design, in which there was a continuous inflow of reservoir water, ensured continuous nutrient availability. All jars were kept under the same conditions regarding light, temperature and salinity. The clams were exposed to light with a photon flux density an intensity of 400 $\mu\text{mol}/(\text{m}^2 \cdot \text{s})$ μE (measured on top of the jars) for 12 h, from 7:30 to 19:30. The temperature was maintained kept at 28 °C. The four jars were exposed to an alternating sequence of ^{135}Ba -spiked and non-spiked water (Fig. 1). Jars 1 and 2 were exposed with seawater spiked with ^{135}Ba during the night (19:30 to 7:30), while non-spiked seawater was present in jars 3 and 4 at night. During the day (7:30 to 19:30) jars 1 and 2 were filled with non-spiked seawater, while jars 3 and 4 were exposed to seawater spiked with ^{135}Ba .



150 **Figure 1: Experimental setup of the culturing tanks: A: Setup during daytime, jars number 1 and 2 (represented by one jar in the sketch) contain Eilat Sea Water (ESW) while ESW spiked with the ^{135}Ba tracer is introduced into jars number 3 and 4 from 7:30 to 19:30. During this time, the lights over the culturing jars are on. B: During nighttime (19:30 to 7:30) the source is reversed: ESW spiked with the ^{135}Ba tracer is introduced into jars number 3 and 4, while jars number 1 and 2 receive non-spiked ESW and the lights remain off. A transition period where water is pumped in and out of the jars takes place in the mornings and evenings from ~7:00 to 7:30 and 19:00 to 19:30, respectively. Each jar contains two clams sitting on ~~one~~ petri dish ~~each~~. The water within the jars is stirred with a magnetic stir bar. C: Timing of the alternating seawater exposure between ^{135}Ba -spiked and non-spiked water (the green shaded blocks indicate ^{135}Ba -enriched seawater) for all four jars: After the initial spiking period of 72 hours, jars 3 and 4 received spiked water during the day from 7:30 to 19:30— as indicated by the dashed line. Jars 1 and 2 received the opposite spiked seawater treatment.**

155 The initial reservoir water was recycled, by mixing the outflow from the jars with the remaining respective reservoir water, and aerated for 15 min. By reintegrating the outflow (~2 liters) to the respective 11 l reservoirs, nutrient concentrations within the reservoir waters were slightly diluted over the 3 day experiment, although the food within the reservoir water was sufficient throughout the culturing interval according to the aquarium food provider. The reservoir waters were sampled every morning and evening, before the new water source was connected. Final outflow water samples were taken directly from the jar outflow before the lights turned on in the morning and off in the evening.

160 Oxygen, pH and alkalinity measurements were performed every 12 hours on both the initial and final water samples. The resulting Resulting differences, namely Δoxygen , ΔpH and $\Delta\text{alkalinity}$, for each jar are displayed in

170 Tab.table 2. Oxygen was monitored via an YSI ProODO meter, pH was measured using a WTW pH 340i pH meter, with measurements reported on the NIST scale, while alkalinity measurements were performed using a Metrohm 716 DMS titrino titratordevice. The alkalinity difference between the beginning and end value of the experiments was used to calculate the CaCO₃ uptake and the resulting percentage of growth per day and clam (see Tab.le 2).

175 After the clams were sacrificed, the shells were separated from the soft tissue and brushed clean using tap water. CleanedThe clean shells were embedded in resin and cut along the maximum growth axis. Thin sections with a thickness of 50 μm were prepared and polished using a 3 μm diamond suspension.

2.2 Spatially-resolved elemental analysis by LA-ICPMS

180 Laser-ablation inductively-coupled-plasma mass spectrometry (LA-ICPMS) was performed on thin sections (50 μm thick) using a novel custom-built dual-wavelength (157 & 193 nm) LA-ICP-MS/MS system, operated at 193

nm, coupled to an inductively coupled plasma tandem mass spectrometer, operated in single-quadrupole mode (Erhardt et al., 2025)(Erhardt et al., in revision, JAAS). More specifically, this comprises a modified RESolution-

185 SE LA-system (Applied Spectra Inc., formerly Resonetics LLC), featuring a Coherent Excistar 500 excimer laser (instead of an Atlex ATL laser) and a Laurin Technic S-155 two-volume LA cell (Müller et al., 2009) that is linked

to an Agilent 8900 ICP-MS/MS. Ablation took place in an He atmosphere (0.35 l/min) to which Ar (1.00 l/min) was added, withbesides N₂ (3.5 ml/min) added downstream of the ablation cell to enhance sensitivity and plasma stability (see Tab.le S2). LA-ICPMS instrument tuning used a 50 μm round spot, with 10 Hz and 3 J/cm² on NIST

190 SRM612 to achieve tuning conditions characterized by parameters of 0.07% oxide rate (ThO⁺/Th⁺), 0.5% doubly-charge rate (m/z=22/44), 0.18 ³⁸Ar/⁸⁰Ar₂ and 99% ²³²Th/²³⁸U ratio with an ²³⁸U-signal of around 1 Mcps. The

analyses were broadly based on the methodology in Warter et al. (2018) by performing laser ablation in slow continuous profiling mode with a rotatable slit and LA tracks that were set perpendicular to the daily banding, in

195 direction of growth starting at the final growth segments of the outermost shell and progressing directly opposite to the direction of growth (Fig. 2). To maximize spatial resolution while maintaining suitable instrument sensitivity

(that is crucial in view of the anticipated low Ba concentrations of these samples of around 2 $\mu\text{g/g}$ _{LS}, we used a

200 rotating rectangular mask that resulted in a 1.25 \times 50 μm laser spot on the sample. The ablation area is equivalent to a 9 μm round laser spot and thus provides more than sevenfold improved spatial resolution in the direction of

growth, which is essential to resolve the daily growth segments ranging between \sim 5 to 20 μm . The rotatable slit was aligned to be parallel to the daily banding throughout the measurements (Fig. 2). Differing from Warter et al.

205 (2018), the laser and ICPMS settings were optimized to match the 800 ms washout time of the sample introduction system (full width at 1% maximum of the single pulse response). To do so, the overall ICPMS sweep time was set to 206 ms for the following monitored isotopes (and dwell times): m/z ¹¹B (37 ms), ²³Na (16 ms), ²⁴Mg (18 ms),

⁴³Ca (16 ms), ⁸⁸Sr (16 ms), ¹³⁵Ba (47 ms) and ¹³⁸Ba (42 ms). In turn, the laser beam was scanned at 1.517 $\mu\text{m/s}$ to yield four sweeps per laser beam width, with the laser triggered five times during every sweep using a QuadLock

210 device (Norris Scientific (Norris et al., 2021)), resulting in a repetition rate of 24.27 Hz. The QuadLock aligns the laser firing rate to the ICPMS, eliminating any aliasing artifacts in the data (Müller et al., 2009; Norris et al., 2021)(Norris et al., 2021). The on sample fluence was set to \sim 4.5 J/cm². Each ablation track was pre-cleaned with

215 100 Hz repetition rate with a spot overlap of 90%, i.e. ten shots deep, equivalent to 1.6 μm at this fluence (Coenen et al., 2024).

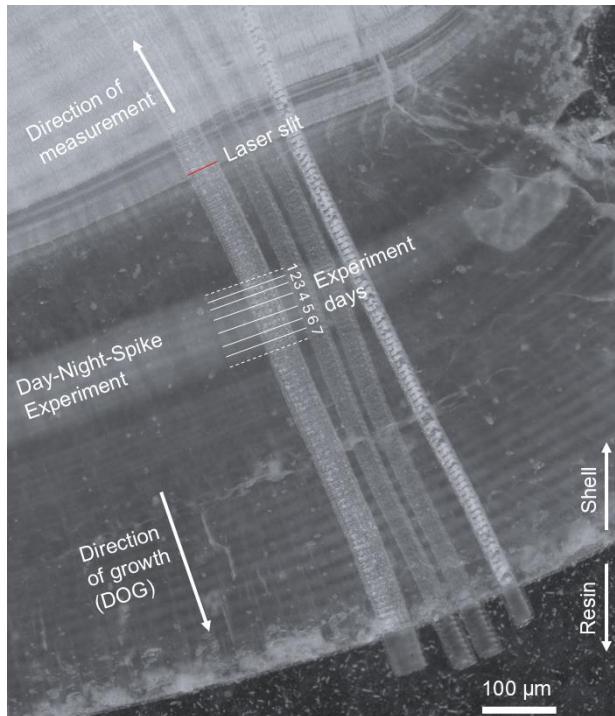


Figure 2: Microscope image of the thin section of sample 3.1 DAS with laser-ablation paths, of which the leftmost reflects the data presented in this study. The brighter shaded domain that represents the day-night bands is indicated, including experiment days; following the experiment reported herein, the shells continued growing in a different experimental setup whose results are beyond the scope of this study⁵ and which will be reported elsewhere. The rotatable 1.25 \times 50 μm laser slit, indicated by the red bar, is set parallel to the daily bands, with the laser-ablation path progressing parallel (but opposite to) the direction of growth.

The measurements were quantified following Longerich et al. (1996) using NIST SRM612 as [the](#) external standard, with the reference values from Jochum et al. (2011), except for Mg for which the updated value of 62.4 $\mu\text{g/g}$ from Evans and Müller (2018) was used. ^{43}Ca was used as [the](#) internal standard.

Processing and visualization of LA-ICPMS data were performed utilizing the Iolite 4 software (Paton et al., 2011) and Python, with the support of libraries such as NumPy (Harris et al., 2020), Pandas (Reback et al., 2022) and Matplotlib (Caswell et al., 2022).

To evaluate data quality, i.e. accuracy and precision, measurements were taken of the carbonate standard MACS-3 nanopellet and the MPI-DING glass standard K2L-G were performed. In the case of For MACS-3, boron reference values are taken from previous LA-ICPMS analyses^{data} (Jochum et al., 2012), while the Na, Mg, Sr, Ba reference values are solution ICPMS-based USGS-data from Stephen Wilson (Jochum et al., 2012) (pers. eomm, 2010). For KL2-G we use the preferred values from the GeoReM database version 658 (Jochum et al., 2006). These standards were measured and processed in the same way as the samples. Accuracies range from -2.2 to -13.0% for MACS-3NP and -7.4 to 15.5% for KL2-G, while precision ranged between 1.5 and 10.8% 2 RSD for MACS-3NP and 1.8 and 22.2% 2 RSD for KL2-G (see Tab. 1).

Table 1: Assessment of analytical accuracy and precision based on repeated measurements of the carbonate standard MACS-3 and the glass standard KL2-G. Reported values include measured means and associated ± 2 SD, compared to published reference values from literature (Jochum et al., 2012 (for MACS-3 B); Jochum et al., 2006 (for KL2-G); Stephen Wilson, see Jochum et al., 2012 *pers. eomm.*, 2010 (for MACS-3 Na, Mg, Sr, Ba)). Accuracy is expressed as the deviation from the reference value, and precision as the reproducibility (± 2 % RSD) of the measurements.

MACS-3	Measured mean [$\mu\text{g/g}$]	± 2 SD [$\mu\text{g/g}$]	Reference value [$\mu\text{g/g}$]	± 2 SD [$\mu\text{g/g}$]	Accuracy [%]	Precision [% 2 RSD]
B	8.63	0.47	8.90		-3.0	10.8
Na	5133	240	5900	800	-13.0	9.2
Mg	1717	45	1756	272	-2.2	5.2
Sr	6396	140	6760	700	-5.4	4.5
Ba	56.3	0.6	58.7	4.0	-4.1	2.3
KL2-G	Measured mean [$\mu\text{g/g}$]	± 2 SD [$\mu\text{g/g}$]	Reference value [$\mu\text{g/g}$]	± 2 SD [$\mu\text{g/g}$]	Accuracy [%]	Precision [% 2 RSD]
B	3.15	0.35	2.73	0.28	15.5	22.2
Na	16136	820	17434	593	-7.4	10.2
Mg	42628	2000	44263	543	-3.7	9.4
Sr	352	3	356	8	-1.0	1.8
Ba	121.1	1.8	123.0	5.0	-1.5	3.0

3 Results

3.1 Water parameters

240 A clear difference pattern between night and daytime calcification is seen in the difference in seawater carbonate chemistry and dissolved $[\text{O}_2]$ measurements performed measured in the evening and morning. During daytime, pH increases by ~ 0.14 units while oxygen concentrations rise by ~ 1 mg/l. At the same time, water-alkalinity is reduced by 80 $\mu\text{mol/kg}$ on average, equivalent to a calcification rate of 0.09 ± 0.01 (2SD) wt% growth (weight increase) per clam for this 12 h interval. The shell weight was estimated as 60% of the measured weight of the clam with soft tissue, with and the resulting average shell weight of 5 g was used to calculate the calcification rates. In contrast, during the night, pH decreased by 0.17 units while oxygen concentrations are ~ 1.35 mg/l lower than in the morning, compared to the previous evening. The average water-alkalinity reduction was only 16 $\mu\text{mol/kg}$, resulting in an average calcification rate of 0.02 ± 0.01 (2SD) wt% per clam per 12 h. While the clams calcify both during day and night, the daytime calcification rates are almost fivefold (4.5 to 5 times) higher.

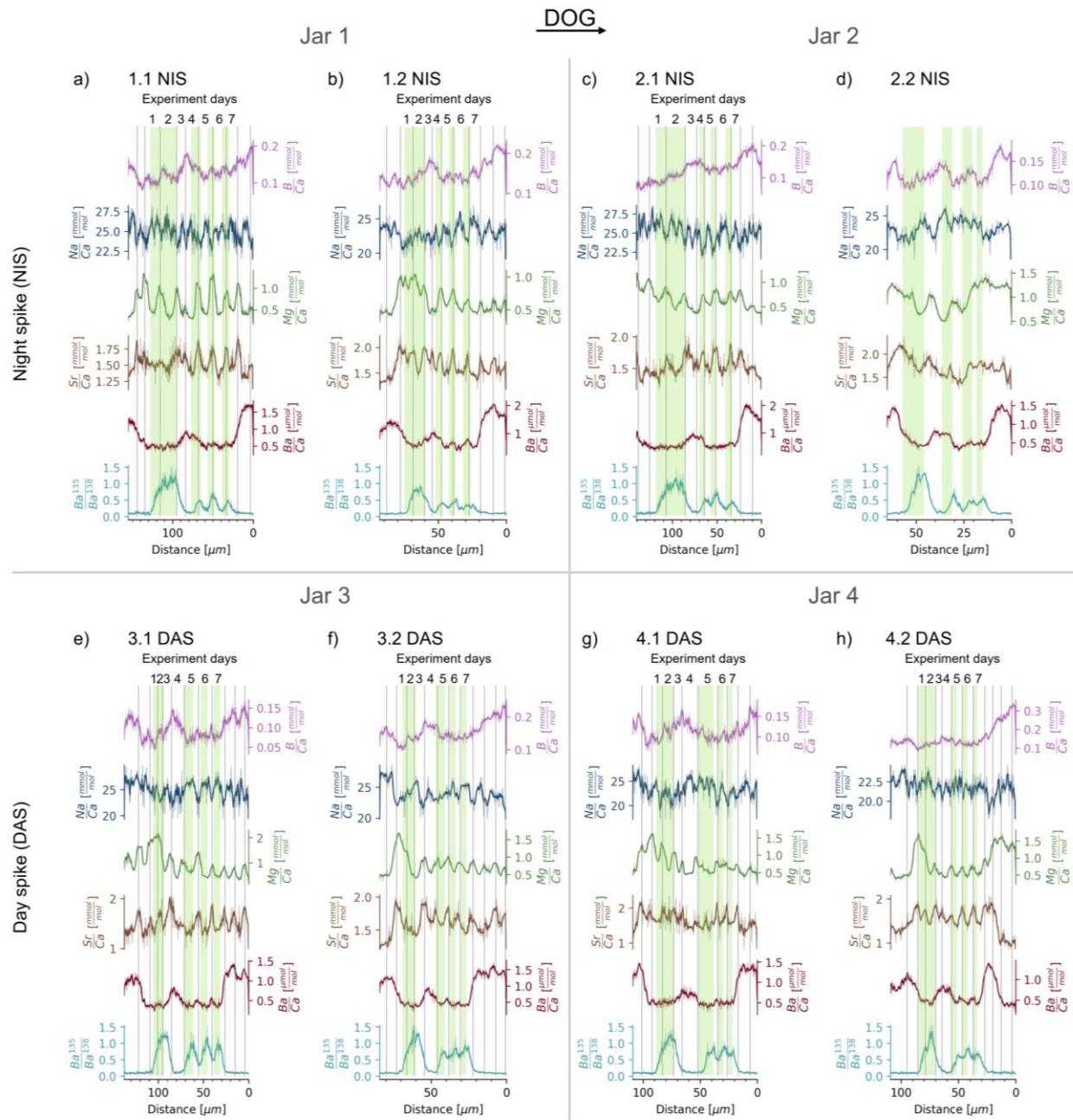
Table 2: Changes in water chemistry and calcification rates of the clams that were alternately exposed to ^{135}Ba -spiked and non-spiked seawater for 12 h each during daytime and nighttime. The table indicates which clams were cultured in each jar, the spike condition (presence or absence of ^{135}Ba), and the measured differences in pH, oxygen, and alkalinity over each 12-hour interval (note that all alkalinity anomalies were negative, thus implying calcification took place in all cases). 12-hour intervals. Calcification rates were calculated from alkalinity changes and are expressed as the percentage of growth per clam per 12 hours. The average calcification of all eight clams is provided for every day and night interval.

	Jar	Clams	Spike	ΔpH	Δoxygen [mg/l]	$\Delta \text{Alkalinity}$ [μMol] \pm 2SD	Calcification [%/clam and 12 h] \pm 2SD12h	Average calcification [%/clam and 12 h] \pm 2SD12h
Night 4	J1	1.1 NIS, 1.2 NIS	^{135}Ba	-0.189	-1.82	11 \pm 4	0.013 \pm 0.005	0.02 \pm 0.01
	J2	2.1 NIS, 2.2 NIS	^{135}Ba	-0.188	-1.84	13 \pm 4	0.017 \pm 0.004	
	J3	3.1 DAS, 3.2 DAS		-0.208	-0.73	20 \pm 5	0.023 \pm 0.006	
	J4	4.1 DAS, 4.2 DAS		-0.2055	-1.37	16 \pm 2	0.019 \pm 0.003	
Day 5	J1	1.1 NIS, 1.2 NIS		0.1325	0.83	89 \pm 4	0.103 \pm 0.004	0.10 \pm 0.01
	J2	2.1 NIS, 2.2 NIS		0.1395	1.18	86 \pm 3	0.106 \pm 0.004	
	J3	3.1 DAS, 3.2 DAS	^{135}Ba	0.1405	1.03	80 \pm 2	0.092 \pm 0.003	
	J4	4.1 DAS, 4.2 DAS	^{135}Ba	0.1785	1.07	85 \pm 2	0.101 \pm 0.002	
Night 5	J1	1.1 NIS, 1.2 NIS	^{135}Ba	-0.1865	-1.67	24 \pm 4	0.028 \pm 0.004	0.02 \pm 0.01
	J2	2.1 NIS, 2.2 NIS	^{135}Ba	-0.2035	-1.86	21 \pm 5	0.026 \pm 0.006	
	J3	3.1 DAS, 3.2 DAS		-0.2055	-1.38	16 \pm 4	0.019 \pm 0.005	
	J4	4.1 DAS, 4.2 DAS		-0.1755	-1.26	10 \pm 4	0.012 \pm 0.005	
Day 6	J1	1.1 NIS, 1.2 NIS		0.1885	0.23	85 \pm 3	0.098 \pm 0.004	0.09 \pm 0.01
	J2	2.1 NIS, 2.2 NIS		0.1685	0.31	70 \pm 4	0.087 \pm 0.004	
	J3	3.1 DAS, 3.2 DAS	^{135}Ba	0.1455	0.95	74 \pm 6	0.085 \pm 0.007	
	J4	4.1 DAS, 4.2 DAS	^{135}Ba	0.1465	1.08	72 \pm 3	0.085 \pm 0.003	
Night 6	J1	1.1 NIS, 1.2 NIS	^{135}Ba	-0.1335	-0.91	19 \pm 3	0.022 \pm 0.003	0.02 \pm 0.01
	J2	2.1 NIS, 2.2 NIS	^{135}Ba	-0.1105	-0.82	13 \pm 1	0.016 \pm 0.001	
	J3	3.1 DAS, 3.2 DAS		-0.1235	-1.31	21 \pm 2	0.024 \pm 0.002	
	J4	4.1 DAS, 4.2 DAS		-0.1215	-1.18	12 \pm 3	0.015 \pm 0.003	
Day 7	J1	1.1 NIS, 1.2 NIS		0.118	0.95	95 \pm 5	0.110 \pm 0.00644	0.09 \pm 0.01
	J2	2.1 NIS, 2.2 NIS		0.121	1.12	64 \pm 4	0.080 \pm 0.00508	
	J3	3.1 DAS, 3.2 DAS	^{135}Ba	0.127	1.31	79 \pm 1	0.091 \pm 0.001	
	J4	4.1 DAS, 4.2 DAS	^{135}Ba	0.0945	0.89	79 \pm 3	0.094 \pm 0.003	

3.2 Spatially-resolved compositional data (LA-ICPMS)

The elemental ratio data from the day-night spiking experiment can be found in Tab_{1e} S3 and are displayed in Fig_{ure} 3. Overall, the B/Ca values span from 0.04 mmol/mol to 0.34 mmol/mol, with an average of 0.13 mmol/mol. Na/Ca values range between 17.2 to 29.8 mmol/mol, with an average of 24.0 mmol/mol. Mg/Ca values

vary nearly tenfold between 0.25 and 2.21 mmol/mol, with an average of 0.77 mmol/mol. The average Sr/Ca value is 1.57 mmol/mol, with a minimum of 0.82 and maximum at 2.4 mmol/mol. Ba/Ca values vary from 0.21 to 2.09 $\mu\text{mol/mol}$, with an average of 0.77 $\mu\text{mol/mol}$. (Fig. 3).



265
Figure 3: Spatially-resolved El/Ca (B, Na, Mg, Sr, Ba) and $^{135}\text{Ba}/^{138}\text{Ba}$ isotope ratios for all cultured clams. The latter
270 **varies between the natural ratio of 0.092 and approaches that of the spiked water of ~ 1.24 . Datasets a-d) represent the**
275 **shells grown with ^{135}Ba -spiked water during the night at experiment days 4 to 6 ('NIS'), with two jars and two clams**
per jar. Datasets e-h) represent the shells that were exposed to given the ^{135}Ba -spiked water during the day from days 5
to 7 ('DAS'), also with two jars and two clams per jar. Areas grown during exposure to ^{135}Ba -spiked water are indicated
by green bars. $^{135}\text{Ba}/^{138}\text{Ba}$ intake during the 12 h interval is higher for day-spiked clams than night-spiked clams. Daily
cycles are visible in Mg/Ca and Sr/Ca for 7 out of 8 clams. Clam 2.2 NIS, the only *Tridacna squamosa*, grew only about
half compared to its 'partner clam' 2.1 NIS and does not display daily cycles. For the other 7 shells, vertical lines indicate
Mg/Ca maxima, and if uncertain, Sr/Ca maxima, which thus delineate the daily growth increments. Note that except
for $^{135}\text{Ba}/^{138}\text{Ba}$ the x-y axis scaling is not uniform between the clams but optimized for best data visibility, with the
exception of $^{135}\text{Ba}/^{138}\text{Ba}$.

The initial introduction of the clams to ^{135}Ba -spiked water fore 42 h on experiment day 1 and 2 shows that most shells do reach a $^{135}\text{Ba}/^{138}\text{Ba}$ plateau close to the calculated $^{135}\text{Ba}/^{138}\text{Ba}$ ratio of the water of ~ 1.24 after 24 h.

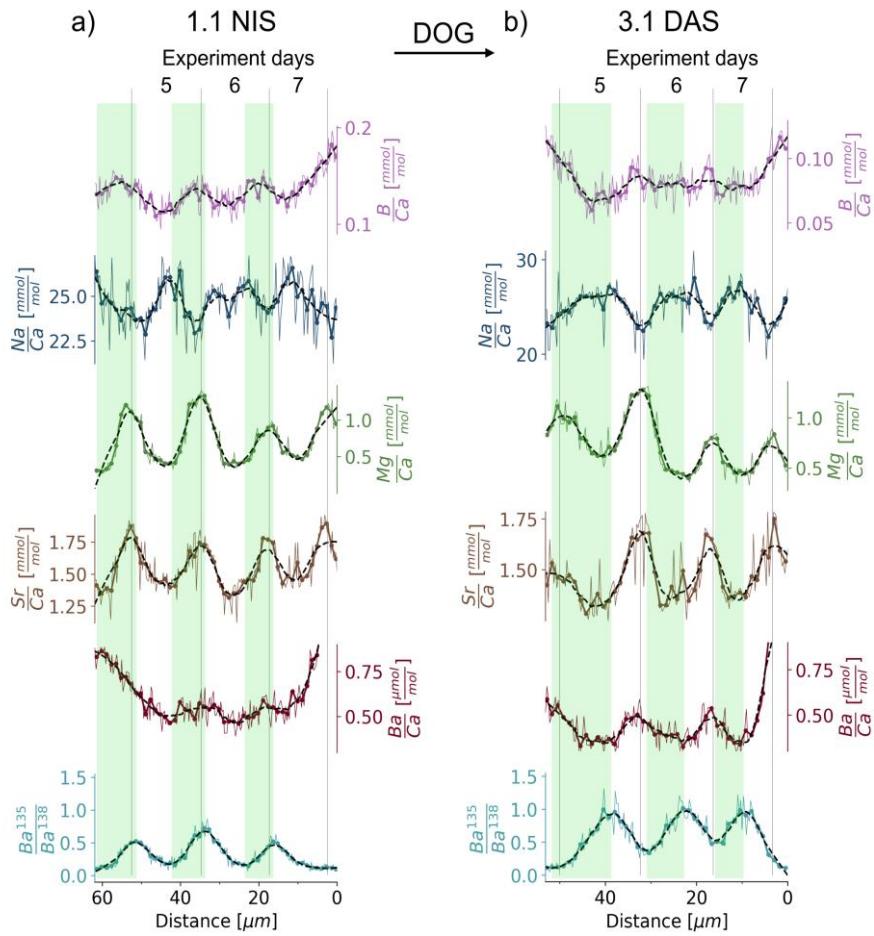
However, shells NIS 1.2 and NIS 2.1 only reach $^{135}\text{Ba}/^{138}\text{Ba}$ plateau values of ~0.8 and 1.0, while DAS 3.2 and

280 DAS 4.2 reach 1.3 and 1.4, respectively. The latter might indicate that the $^{135}\text{Ba}/^{138}\text{Ba}$ ratio of 1.24, calculated for the spiked water, was slightly underestimated by overestimating the initial ESW [Ba]. After~~At~~ experiment day 7, after the last exposure to the spiked water (experiment day 7), the $^{135}\text{Ba}/^{138}\text{Ba}$ ratio returns to a ratio of ~0.1, i.e., approaching the natural ratio of 0.092, within 24 h in the night-spiked and within 12 h in the day-spiked clam shells.

285 The data from experiment days 4 to 7 show that ^{135}Ba is incorporated into the shells of clams subjected to the tracer both during the day (7:30 to 19:30) as well as during the night (19:30 to 7:30). The $^{135}\text{Ba}/^{138}\text{Ba}$ ratio delineates the relative growth between day and night, with daytime ^{135}Ba uptake resulting in measured shell $^{135}\text{Ba}/^{138}\text{Ba}$ ~being on average 1.7 times higher than those exposed to ^{135}Ba at night. Then~~night~~ ^{135}Ba uptake. This is because of the contrasting cyclical $^{135}\text{Ba}/^{138}\text{Ba}$ ratio variations are 0.2 to 0.7 and 0.4 to 1.1 for night and day-spiked clams, 290 respectively.

In most clams, Mg/Ca and Sr/Ca show a regular cyclicity that can be clearly identified as daily, based on the timing of the isotopic tracer introduction. Unequivocally, both elemental ratios show the highest values at the end of the night~~during nighttime~~ (Fig. 4a). Daily cyclicity is less clear in Na/Ca, for which the nighttime growth seems to be characterised by domains with decreasing~~low~~ Na/Ca, while higher Na/Ca is increasing~~reached~~ during the day. 295 Daily cycles in~~In~~ B/Ca and Ba/Ca daily cycles~~are~~ are less clear, while~~and~~ the ratios vary strongly between the portions of the shell parts grown in the aquarium and the culturing jars. The specimens grew~~shells were growing~~ in a large artificial seawater~~ESW~~ aquarium with other clams and corals before experiment day 1 and after experiment day 7, residing; they resided in a smaller separate aquarium with ESW during~~at~~ experiment days 3 to 4 and in the culturing jars with modified (reduced salinity) ESW during experiments days 1 to 2 and 5 to 7 (Fig. 1). The 300 observed changes in B/Ca and Ba/Ca (Fig. 3) may therefore be~~are~~ likely caused by the slightly different seawater compositions in these different growth environments.

We observe that Mg/Ca and Sr/Ca, and less clearly B/Ca and Ba/Ca, are in phase with $^{135}\text{Ba}/^{138}\text{Ba}$ for night-spiked shells (e.g. in NIS 1.1, Fig. 4a). In contrast, day-spiked shells display an anti-phased relationship (e.g. in DAS 3.1, Fig. 4b). Therefore, a ^{135}Ba increase at night is seen in the shell, approximately coinciding with the minimum to maximum of both Mg/Ca and Sr/Ca cycles. In contrast, the increase in ^{135}Ba during the day coincides with (or shortly follows~~post~~dates) the shift from the Mg/Ca (or Sr/Ca) maximum to respective minimum (or shortly thereafter in each case).



310 **Figure 4: Expanded view of El/Ca and $^{135}\text{Ba}/^{138}\text{Ba}$ ratios for the sections of the shell grown during the day-night-spiking experiment for two representative examples, namely 1.1 NIS and 3.1 DAS. A 4-point running mean is plotted in addition to the original data. Each 4-point-mean data point is indicated by a circle, resulting in one circle every 1.25 μm which is the same as the width of the laser beam. The red dashed line represents a quadratic polynomial fit to the 4-point-mean signal using a Savitzky-Golay filter (Savitzky and Golay, 1964), which smooths the data while preserving important features of the signal. Areas grown during exposure to ^{135}Ba -spiked water are indicated by green bars.**

315 **4 Discussion**

4.1 Reproducibility

All in the experimental setup all eight clams grew in the same environmental conditions in terms of such as temperature, light and food availability, and only the timing of the introduction of ^{135}Ba -spiked water varied introduction was systematically varied. Four clams in two jars received a ^{135}Ba -spike at night and daytime, respectively. With two clams per jar The initial water reservoir, later divided into spiked and two jar per treatment we can compare whether the elemental ratio patterns are non-spiked, was the same between clams grown in the same jar and in different jars with the same treatment. Furthermore, this provides a replication of the water chemistry and estimated growth rates, which, in our experimental setup, reflect the mean of the two clams inhabiting the same jar. Some patterns consistently emerge between the two sets of four replicates, such as the overall incorporation of the isotopic tracer, daily elemental cycles and overall changes in shell chemistry between aquarium and culturing water (Fig. 3). On a more detailed level, there are clear differences between the individual clams regarding overall growth, growth pattern and element incorporation even between pairs kept within the same jar (see 1.1 vs 1.2 NIS, 2.1 vs 2.2 NIS, 3.1 vs. 3.2 DAS), presumably related to their individual

330 **physiology.** During the seven days of experiment, the clams' shells grew between 55 to 115 μm in the direction of growth. Similarly, daily growth rates range from 6 to 21 $\mu\text{m}/\text{day}$ and vary by up to 9 $\mu\text{m}/\text{day}$ even within one clam (Fig. 3e). As environmental factors were identical for the two clams in one jar, such variability is presumably related to the individual physiology of the clams.

335 The culturing conditions of this experiment were based on the findings of Warter et al. (2018) regarding temperature, light and food availability, which were optimized for high calcification rates through several experiments. The resulting calcification patterns may therefore not be uniformly applicable to all species, sizes and temperature regimes, although the parameters utilised here are within the range of those that *Tridacna* experience in their natural environment. In addition, we note that the daily cycle amplitudes and absolute El/Ca values from the cultured *Tridacna* of this study overlap well with data from naturally grown samples, namely those from the late Miocene *Tridacna* of Arndt et al. (2025) as well as the recent and fossil clams from Warter and 340 Müller (2017), which were all measured using a similar analytical approach. However, we do observe that the Ba/Ca minima in all clams are below the average minus 2SD of the values reported in the literature. Three clams (3.1 DAS, 3.2 DAS, 4.2 DAS) also exhibit unusually high Mg/Ca maxima with values 52, 20 and 17 % above the average of the literature values, respectively. Two clams (4.1 DAS, 4.2 DAS) are characterised by very low Sr/Ca 345 minima, 34 and 32 % below the literature average (see Tab. S4). Nonetheless, the overall comparability of the data from cultured, recent and fossil clams grown in different environmental settings, combined with the variability observed in El/Ca ratios among clams grown under identical conditions in this study, underscores the significant biological influence on the El/Ca composition of clam shells, as well as the applicability of laboratory culture results to naturally grown samples.

4.2 Analytical resolvability

350 Given a sweep time of 0.206 s and a scan speed of 1.517 $\mu\text{m}/\text{s}$, the resulting sampling frequency is 0.313 μm . With a slit width of 1.25 μm this results in four sampling points per slit width. If we use the 4-point mean values, i.e. one data point per slit width, to get independent data points, we can safely detect cycles with a above 3 μm wavelength of greater than 3 μm . As daily cycle wavelengths range from 6 to 20 μm , all compositional cycles should analytically be well resolved.

355 The full signal rise takes 0.2 s (i.e. 0.3 μm) in the sampling direction and signal washout to 10% and 1% takes place in 0.5 s and 1.2 s (0.8 and 1.8 μm in sampling distance), respectively. Therefore, data points in 1.25 μm steps (~ 0.8 s measurement time) are not completely independent but can contain $\sim 5\%$ of the signal from the previous data point. As the resulting measured signal consists of around 95% of the actual signal at that distance, we do not expect washout to lead to substantialsignificant signal alteration.

360 Using a narrow but wide laser slit (1.25 \times 50 μm) helps to obtain suitably high elemental signals for low [Ba] and [B] while maintaining very high spatial resolution ($< 2 \mu\text{m}$). With a narrow, long slit optical alignment to daily banding is facilitated as misalignment is easy to see, however slight variations from the alignment might result in minor signal mixing between growth layers. In some shell areas the daily band structures are either slightly curvedbent or not visible at all (Fig. 2). We aimed to position the laser slit as parallel as possible to the daily banding, avoiding thesebent structures. Quantitatively, if the slit is 2° misaligned, it would cover $\sim 23 \mu\text{m}$ in direction of growth, while 5° result in almost 56 μm , i.e., almostmore than four times the laser slit width. We evaluated the microscope pictures of the laser ablation paths on the shells. As the day-night-experiment is well visible through a lighter shading in the shell (likely due to higher temperatures and faster growth rates compared

to the aquarium; Fig. 2) an accurate alignment was possible. We therefore demonstrated that we measured parallel to the daily banding with occasional offset between laser slit and growth banding direction of below 2°. While limited, analytical signal smoothing may have occurred in some areas, the associated signal smoothing but would be on the order of 23 µm or less, which is still below half the minimum daily growth width.

Nevertheless, the dataset contains areas with unclear El/Ca signals on scales of tens of µm, e.g. in Mg/Ca days 1 to 2 of DAS 4.1, Sr/Ca days 1 to 2 of NIS 1.1 and in Mg/Ca throughout NIS 2.2 (Fig. 3). Interestingly, at the same distance, the cycles in the other respective ratios (Sr/Ca, Mg/Ca and $^{135}\text{Ba}/^{138}\text{Ba}$) are resolvable. This provides an additional evidence against substantial indicator to exclude analytical smoothing of the signal as the reason for these unclear cycles. We therefore argued that the unclear cyclic patterns are not analytically caused but reflect the clams' individual growth performance and elemental uptake into the shell.

4.3 EPF reaction and replenishment time

The cyclic behaviour of the $^{135}\text{Ba}/^{138}\text{Ba}$ ratio within the shells, contrasts with that of the rapid switch between the endmembers of spiked ($^{135}\text{Ba}/^{138}\text{Ba} = 1.24$) and non-spiked ($^{135}\text{Ba}/^{138}\text{Ba} = 0.092$) water (Fig. 5). In most shells, it takes about one day (i.e. ~5 to 20 µm of growth depending on the specimen) until the Ba isotopic ratio of the spiked water is reached. This smoothing is, as discussed above, again unlikely to be caused by limited analytical smoothing resolvability because it is similar in all clams and larger than the worst case of analytically induced smoothing of ~3 µm. We thus deduce that the $^{135}\text{Ba}/^{138}\text{Ba}$ ratio behaviour indeed reflects predominantly the uptake patterns of elements into the EPF.

During the day-night-spiking, i.e. experiment days 4 to 7, the uptake of the ^{135}Ba -isotopic tracer starts shortly after the Mg/Ca maximum (night spike) or minimum (day spike) is reached after which $^{135}\text{Ba}/^{138}\text{Ba}$ continues to rise until the Mg/Ca minimum (night spike) or maximum (day spike). We therefore infer that during the 12 h of culturing period, in the presence of the ^{135}Ba spike, the concentration amount of ^{135}Ba -isotopic tracer continuously increases in the EPF, such that the $^{135}\text{Ba}/^{138}\text{Ba}$ value of the shell continuously rises. Similarly, washout begins rapidly, as indicated by a reduction of $^{135}\text{Ba}/^{138}\text{Ba}$ right after the Mg/Ca minimum (night spike) or maximum (day spike) is reached. Taken together, we can unambiguously determine that the uptake of the spike occurs more quickly during daytime. Samples spiked during the day are broadly characterised by 1.7 times higher $^{135}\text{Ba}/^{138}\text{Ba}$ ratios, which do not return to the baseline (i.e. natural) $^{135}\text{Ba}/^{138}\text{Ba}$ at night. This indicates that the residence time of Ba in the EPF is substantially shorter during the day than at night. Assuming barium and calcium behave similarly, this aligns with our alkalinity measurements described above, which constrain a fivefold increase in calcification rate.

Despite a quick EPF reaction time, indicated by rapid initial increase or decrease of the tracer within the shell, the time needed to reach the maximum $^{135}\text{Ba}/^{138}\text{Ba}$ value is approximately one day, as also indicated by the initial 42 h tracer exposure (Fig. 5). This further supports the idea of continuous exchange between the ambient seawater and EPF and, for these small specimens, indicates a barium (and calcium) residence time in the EPF of ~24 hours.

Tridacninae have been observed to partially close during the night, likely as a defence mechanism against predators (Killam et al., 2023). A partial closure could affect the overall EPF volume, potentially resulting in a reduced volume present at night and thus shortening the residence time of elements in the EPF. However, data from a controlled environment without predators shows no clear diurnal change in valve opening (Rossbach et al., 2020).

Nonetheless, we explore whether a diurnal change in EPF volume could explain our observations, for the theoretical case that the pumping of ions into the EPF would be equal between day and night. In this scenario, a smaller extrapallial volume should show a faster increase of ^{135}Ba relative to ^{138}Ba within the EPF, such that the $^{135}\text{Ba}/^{138}\text{Ba}$ ratio in the shell should increase faster at night. Conversely, a smaller EPF volume should lead to faster washout/flushing of the EPF during the night, resulting in a faster decrease of $^{135}\text{Ba}/^{138}\text{Ba}$ within the shell. This is contrary to the pattern we observe, such that possible diurnal changes in EPF volume cannot be the main driver of the observed diurnal $^{135}\text{Ba}/^{138}\text{Ba}$ variability. However, this phenomenon could act to dampen the observed signal that we assign to diurnal changes in active ion transport into the EPF. Future work could test whether the barium residence time is longer for larger clams with a larger extrapallial volume, or whether the residence time is maintained as the result of a more efficient replenishment of the more voluminous EPF.

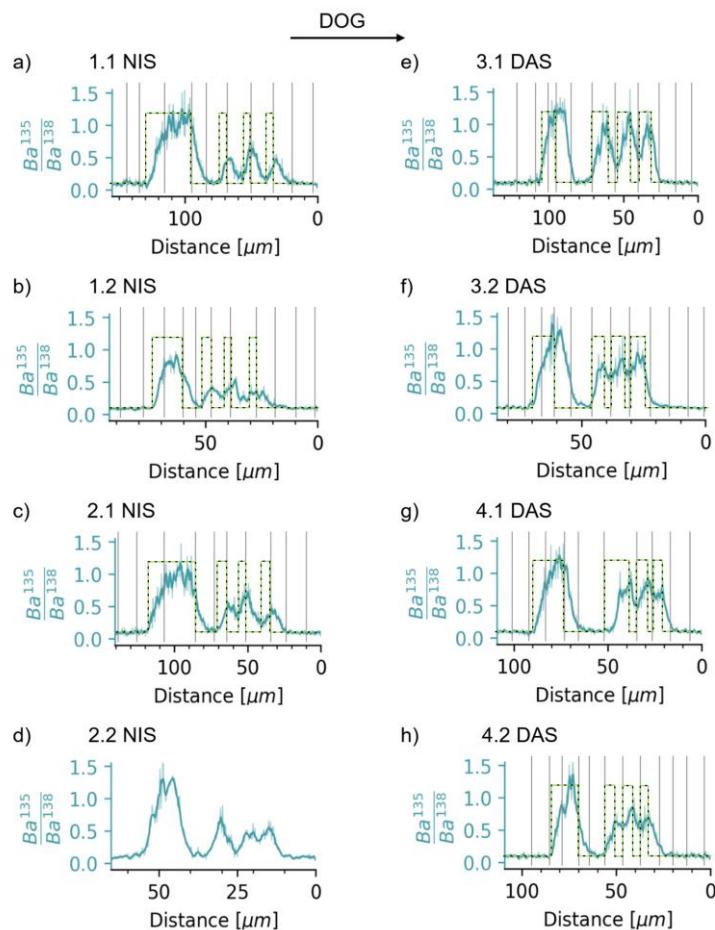


Figure 5: Measured $^{135}\text{Ba}/^{138}\text{Ba}$ ratios of all clams compared with the phases of ^{135}Ba -spiked water introduction, marked as a green-dashed rectangular function. The with the two endmembers showing the natural $^{135}\text{Ba}/^{138}\text{Ba}$ ratio of 0.092 and the calculated $^{135}\text{Ba}/^{138}\text{Ba}$ of the spike water of 1.24. The deviationshape of the measured $^{135}\text{Ba}/^{138}\text{Ba}$ ratio (blue) mostly reflects the uptake behaviour of Ba and other elements from the surrounding seawater into the EPF, with possible maximum analytical smoothing of 2-3 μm .

4.4 Causes for daily El/Ca cycles

In the following sections we discuss which of the possible environmental, biochemical and physiochemical factors have the clearest impact on daily El/Ca ratios. While these factors are inherently interconnected and often exhibit complex interactions, they will be discussed separately to facilitate a clearer and more detailed understanding of their potential contributions.

430 **4.4.1 Environmental factors**

In natural environments, co-varying linked parameter patterns can occur, e.g. temperature and light both positively affect growth rate in *Tridacninea* up to a stress limit (Killam et al., 2021; Schwartzmann et al., 2011; Van Wynsberge et al., 2017; Warter et al., 2018). Besides temperature and light, a further environmental factor that can impact daily elemental cycles is the availability of food that can be acquired through filter feeding. In reef environments, dissolved oxygen concentrations, which broadly reflect net productivity and thus food availability, fluctuate diurnally. Concentrations are lowest at night, rising due to photosynthesis after sunrise, peaking in the afternoon, and then declining due to respiration and oxygen degassing (Silverman et al., 2007). Given However, given that *Tridacna* have photosymbionts and generally rely heavily on phototrophy, albeit with some species and environment-dependent variability (Jantzen et al., 2008; Klumpp and Griffiths, 1994; Mills et al., 2023), light may be a more important environmental driver of shell chemistry. It has been proposed that despite higher nutrient availability in the water at daytime, *Tridacna* rely more on photosynthesis for nutrition during the day and more on filter feeding during the night (Killam et al., 2023). However, giantGiant clams grown in culture experiments in which temperature and nutrient concentrations were not varied between day and night still show daily cycles (Warter et al., 2018; Figs. 3,4). In our setup, two clams grew in one jar with a continuous throughflow of water (residence time 3 h). Thus, we assume that the presence of only two clams and the constant refreshening of the water around the clams strongly reduced the daily nutrient cycle caused by photosynthesis and respiration, compared to a reef environment. We therefore suggested due that neither temperature nor nutrient availability are not the key driver of heterogeneity in shell chemistry in *Tridacna*. Nevertheless, measurements of daily variability in culture water nutrient composition, which was not performed here, may provide further information on the metabolic activity of the clams and their photosymbionts. Future experiments could investigate As this factor explicitly as well as its impact on shell mineralization and chemical composition, ideally in fully closed off culturing jars or jars with reduced water throughflow, containing several giant clams. Given a diurnal change in nutrient availability is not likely to drive the observed geochemical heterogeneity the ease, light availability is a more likely explanation for the daily variability in shell growth and element incorporation, especially given that light enhanced calcification has been observed in *Tridacna* (e.g., Ip et al., 2015; Rossbach et al., 2019; Sano et al., 2012). Recent biochemical studies investigated elemental transportchannelling through the mantle tissue within the context of the light dependent expression of channelling proteins relevant for shell formation. Indeed, multipleMultiple light enhanced calcium and bicarbonate transport mechanisms that could result in this phenomenon are active in shell formation, including voltage-gated calcium channels, $\text{Na}^+/\text{Ca}^{2+}$ exchangers, plasma membrane Ca^{2+} -ATPase, and bicarbonate transporters (Boo et al., 2021, 2022; Cao-Pham et al., 2019; Chan et al., 2021; Chew et al., 2019; Ip et al., 2015, 2017) (Boo et al., 2022, 2021; Cao-Pham et al., 2019; Chan et al., 2021; Chew et al., 2019; Ip et al., 2017, 2015).

465 **4.4.2 Biochemical factors**

Organic material present at the calcification site or involved in ‘templating’ the growing shell, might be important in driving shell chemistry. During daytime, calcification is faster such thatand less templating organic matter material might be incorporated into the mineral component of *Tridacnid* shells, while thinner bands, presumably grown at nighttime, are characterized by comparatively higher templating organic content (Liu et al., 2022). However, more dissolved organic matter may converselymight be incorporated during faster growth. In the case of inorganic calcite, the addition of certain organic molecules (peptides) has been shown to increase both growth rate and Mg

470 -incorporation by reducing the dehydration enthalpy of Mg^{2+} to a greater degree than Ca^{2+} , resulting in increased
Mg uptake (Stephenson et al., 2008). As organic material in *Tridacna* is present to a greater degree in the lower
growth, nighttime bands with higher Mg/Ca, we assume that the availability of organic material is not a limiting
factor in determining shell growth rate on a daily basis. Irrespective, the incorporation of organic material could
conceivably be important in controlling shell chemistry (to a degree) given that the concentration of key elements,
475 typically considered to primarily substitute for Ca^{2+} in the mineral lattice, may also be present in organics at
relatively high concentration. For example, seasonal growth bands in *Artica islandica* and *Acesta excavata* are
associated with increased organic contents and Mg^{2+} bound to the organic material, leading to increased Mg/Ca
ratios during LA-ICPMS measurements (Schleinkofer et al., 2021; Schöne et al., 2010). It has also been suggested
480 that seasonal Sr/Ca variability in *A. islandica* shells is controlled by the organic matrices at the calcification site
(Shirai et al., 2014). However, given the absence of the (trace) elemental composition of *Tridacnid* organic
material, we cannot assess the degree to which this could drive daily chemical banding.
Additionally, it has been shown that amorphous calcium carbonate (ACC) acts as a precursor phase in the
485 biomineralisation of at least some bivalves (Addadi et al., 2006; Weiss et al., 2002). However, while
organicOrganic material is incorporated into ACC, although the small amount of existing research has shown
opposing impacts on Mg^{2+} incorporation. Wang et al. (2009) found an increase in ACC Mg/Ca when organics are
490 added, while Evans et al. (2020) found that adding amino acids reduced Mg/Ca. The latter observation can be
explained by the reduction in the solution Mg/Ca activity ratio viawith a preferential binding of Mg^{2+} with amino
acids in solution, resulting inandthus less Mg^{2+} being incorporated (Evans et al. 2020). Irrespective, it is agreed
that Mg^{2+} binds with amino acids to ligand-ion complexes to a differential degree to Ca^{2+} , such that this
495 phenomenon has the potential to impact shell chemistry in organisms that utilise an amorphous precursor.
Determining the direction in which this effect may act would require assessing the competing effects of co-
incorporation of metals with organics into ACC and during crystallisation versus the impact that organics have on
the Mg^{2+}/Ca^{2+} solution activity ratio (and therefore ACC and aragonite Mg/Ca), which is beyond the scope of this
study. Studies on *Tridacnid* shell growth have shown that the shell ultrastructure varies between day and night
500 growth (Agbaje et al., 2017; Brosset et al., 2025; Mills et al., 2024). These shell structures are associated with
organic matrix components (Kobayashi and Samata, 2006) and have been interpreted to be caused by the variability
in growth rate and possibly also organic content (Mills et al., 2024).

4.4.3 Physicochemical factors

500 A diurnal change in the physiological performance of the clams, regarding the replenishment of the EPF, indicates
the possibility that elemental cycles are caused by non-uniform depletion of the EPF. This Rayleigh fractionation-
type behaviour (Elderfield et al., 1996; Evans et al., 2018; Ram & Erez 2025) would potentially be visible if the
rate of supply versus consumption of ions to the calcification site differed during the night compared to the day.
In the case that Rayleigh distillation was a key control of shell trace element chemistry, we would expect elements
505 with partition coefficients ($K_d = El/Ca_{aragonite} / El/Ca_{sw}$) below 1 to be characterised by antiphase cyclicity relative
to those with K_d above 1. However, day-night cycles in Mg/Ca are in phase with those in Sr/Ca and Ba/Ca and
opposed to those in Na/Ca, even though Na and Mg have partition coefficients well below 1, while Sr is slightly
above 1 and Ba well above 1 for inorganic precipitation at 25 °C (Gaetani and Cohen, 2006) and in corals (Giri et
al., 2018; Ram and Erez, 2021). The pattern of Mg/Ca varying in phase with Sr/Ca and Ba/Ca indicates that a
differential degree of Rayleigh fractionation at night is not the main cause offer the observed day-night cycles.

510 In addition, we can demonstrate that the EPF is replenished by isotopically-labelled seawater during both day and night and is thus not a closed system at night. However, the differential degree of tracer uptake, with 1.7 times higher $^{135}\text{Ba}/^{138}\text{Ba}$ peaks (i.e. closer to the ratio of the labelled seawater of 1.24) in day-spiked shells compared to those spiked at night, could indicate increased seawater uptake and/or ion transport during the day (see Fig. 5).

515 The observed elemental banding patterns could also be caused by kinetic effects through trace element partitioning into aragonite, driven by changes in the aragonite precipitation rate. Increased precipitation rate causes increased incorporation of Na^+ (Brazier et al., 2024) and Mg^{2+} (Mavromatis et al., 2022), while Sr^{2+} incorporation broadly decreases at higher growth rates (Brazier et al., 2023), although other studies have suggested a more complex response depending on temperature and solution chemistry (AlKhatib and Eisenhauer, 2017). As seen in a previous 520 study (Arndt et al., 2023), Na/Ca in *Tridacna* appears to increase while Mg/Ca and Sr/Ca seem to decrease with increased calcification growth rates. While uncertainties remain regarding kinetically-driven element incorporation, Na/Ca and Sr/Ca therefore appear to follow the suggested trends, while Mg/Ca behaves inversely to is however opposing the trend expected pattern if the banding was driven by kinetics alone. In addition, however, from inorganic growth dynamics. For inorganic calcite precipitation experiments have shown it has also 525 been suggested that higher Mg^{2+} concentrations in solution and/or incorporation into the growing solid phase results in lower structure reduces growth rates, while Sr^{2+} incorporation increases growth (Knight et al., 2023). Therefore, a higher $[\text{Mg}^{2+}]$ in the EPF during nighttime could thus be the cause rather than not the effect of low growth rates, and may reconcile the antiphase relationship between Mg/Ca and Na/Ca .

530 Finally, Another explanation is that increased active Ca^{2+} pumping (possibly light enhanced) into the EPF during daytime would act to reduce the relative concentration of the other elements in the EPF (Sr, Ba, Mg, B) through dilution. This could explain why most elemental ratios (except Na/Ca) - independent of their fractionation factors and inorganic growth rate dynamics - decrease during the day when growth rates increase.

4.4.4 Why is Na different?

535 Na/Ca behaves differently than the other El/Ca as it decreases during the night. This may be because the Na^+ -proton and Na^+ -bicarbonate transporters are involved in controlling the carbonate chemistry of the EPF and maintaining charge balance in channelling Ca^{2+} (Ip and Chew, 2021). In *Tridacna*, the pH of the EPF appears to be dependent, at least to a degree, on light, with pH being controlled through light dependent NH_4^+ channelling (Ip et al., 2006). In addition, the Na^+/H^+ Exchanger (βNHE) is active in pumping out H^+ and regulating pH within the seawater-facing epithelium of the mantle (Cao-Pham et al., 2019a), while light-dependent Na^+/H^+ exchangers 540 are thought to be important in bringing inorganic carbon into the clam and to the photosymbionts (Hiong et al., 2017; Ip and Chew, 2021). Na^+ is also used for metal transport at the EPF-facing epithelium. Specifically, light enhanced activity of the NCX3 enzyme pumping Ca^{2+} into the EPF in exchange for Na^+ has been shown (Boo et al., 2019; Ip and Chew, 2021). NCX3 requires the support of Na^+/K^+ -ATPase, which transports Na^+ back into the EPF and also shows light-enhanced activity (Boo et al., 2017). In addition, bicarbonate transport to the EPF may 545 be an important carbon source for calcification. This, which may be achieved with the electrogenic $\text{Na}^+/\text{HCO}_3^-$ co-transporter, further increasing Na^+ in the EPF during times of increased calcification (Ip and Chew, 2021). All of these processes may lead to a higher Na^+ concentration in the EPF during light exposure, in addition to the kinetic effect described above.

4.4.5 Potential mechanisms explaining elemental banding the patterns

550 Compared to the composition of ESW (Evans et al., 2015a; Steiner et al., 2025) Mg/Ca, Sr/Ca and Ba/Ca are strongly reduced in the shell. Specifically, the observed distribution coefficients are 0.14 for Mg/Ca, 0.19 for Sr/Ca and 0.09 for Ba/Ca (see Tab. S5). This pattern is likely caused by discrimination during ion transport into the EPF. However, we stress that we did not measure the EPF directly, such that this interpretation is based on inference from shell chemistry. With this caveat in mind Overall, light enhanced Na⁺ pumping coupled to Ca²⁺ and HCO₃⁻ transport might overall explain why Na/Ca is elevated during the day unlike other El/Ca (El = B, Mg, Sr, Ba). Together with a (possibly more minor) role of growth rate-related kinetic effects, we propose that light dependent Ca²⁺ and Na⁺ transport and the dilution of other (trace) elements in the EPF can explain the daily cycles in El/Ca seen here (Figs. 3, 4) and previously reported (Arndt et al., 2023; Brosset et al., 2025; Hori et al., 2015; Sano et al., 2012; Warter et al., 2018; Warter and Müller, 2017; Yan, 2020).

555 560 We assume that the same mechanism, namely the intensity of active Ca²⁺ and HCO₃⁻ transport channelling into the EPF, directs day and night changes in El composition, growth rate, shell structure and possibly organic content. An increase in calcification rate during the day, namely fivefold higher at day than at night for the clams cultured in this study (Tab. 2), could affect the crystal structure and might lead to a relatively lower content of template organics in the shell.

565 570 575 Light enhanced activation of important channelling enzymes could indicate that the formation of daily elemental cycles and increments is light dependent. It has, however, also been suggested that, independent from environmental factors, circadian rhythms play an important role in are dictating diurnal changes (Liu et al., 2024; Warter et al., 2018; de Winter et al., 2023), with de Winter et al. (de Winter et al., 2023; Liu et al., 2024; Warter et al., 2018), with de Winter et al. (2023) observing that a shell grown under shades does not exhibit significantly different daily cycles. Even under shaded conditions, increased photosymbiont Energy consumption and light availability seem to be strongly coupled in *Tridacna* and thus it remains difficult to distinguish the impact of metabolic activity during daytime could provide excess energy that can be used for active Ca²⁺ and HCO₃⁻ transport. The observed increase in oxygen concentration during daytime indicates that in our culturing setting, more oxygen is produced by the photosymbionts than is consumed by the giant clams during throughout the day. Energy consumption and light availability seem to be strongly coupled in *Tridacna* and thus it remains difficult to distinguish the impact of metabolic activity throughout the day from that of light availability.

4.5 Relative phasing of the daily increments and elemental cycles

580 An increase in trace element concentrations in bands grown at night has previously been predicted based on the assumption that thinner bands likely represent the night growth, while thicker bands represent day growth in *Tridacna* (Brosset et al., 2025; Mills et al., 2024; Sano et al., 2012). While daytime shell growth was previously assumed to be about three times higher than at night (Brosset et al., 2025), our alkalinity measurements (conducted every 12 h) allow us to determine that the eight specimens utilised here grew on average five times faster regarding calcification in volume (Tab. 2). The), resulting in roughly 1.2 times higher accumulation of new shell material in the direction of growth is, however, only roughly 1.2 times higher during daytime versus nighttime (Figs. 3, 4). 585 This could indicate that the day versus night extension of the shells area perpendicular to the direction of growth (adding to shell width) is even higher than extension in direction of growth (adding to shell thickness). If the fivefold increased shell growth at daytime would be uniform in all directions, we would expect the respective extension in direction of growth to be 1.7 (the cube root of five). As shell extension in direction of growth during

daytime is only increased 1.2-fold, the remaining material must be precipitated perpendicular to the direction of 590 growth to make the shell wider (as compared to thicker). If the shell width increases more at daytime than at nighttime, we would expect an incremental growth pattern on the micrometre scale at the outer surface of the shell. While this incremental growth pattern is poorly preserved in the thin sections (likely due to polishing), it is clearly seen as a grooved structure on photographs of the shell (see Fig. S1).

By comparing the differential incorporation of the ^{135}Ba isotope tracer, supplied only during day or during night, 595 to the Mg/Ca and Sr/Ca cycles within the shells, we gain new insights about the detailed timing of the Mg/Ca and Sr/Ca increase and decrease over the course of a day. Based on this, we can unequivocally show that Mg/Ca and Sr/Ca are increasing during nighttime in the narrower low-growth bands, but Mg/Ca and Sr/Ca values are not necessarily higher at nighttime, as previously suggested (Brossel et al., 2025; Hori et al., 2015; Warter et al., 2018). The 600 clams that grew with the tracer introduced at nighttime (from 19:30 to 7:30) begin to incorporate the tracer into the shell approximately during the Mg/Ca and Sr/Ca minimum, with the tracer concentration decreasing within the shell shortly after the Mg/Ca and Sr/Ca maximum is reached. Similarly, the clams that grew in the presence of the tracer during the day (from 7:30 to 19:30) are characterised by a tracer onset in the shell approximately coincident with the Mg/Ca and Sr/Ca maximum. The tracer signal decreases shortly after the Mg/Ca and Sr/Ca 605 minimum is reached. The isotopic tracer administered for 12 h each can only be incorporated as observed if the Mg/Ca and Sr/Ca maxima reflect the early morning hours or the end of the night, while the Mg/Ca and Sr/Ca minima are reached in the evening hours or towards the end of the day.

4.6 Relevance for palaeoclimate applications

Understanding the causes of daily element banding holds relevance for palaeoclimate applications. The daily El/Ca 610 patterns, likely caused by metabolic activity and ion pumping, reflect seasonal patterns that have previously been investigated in fossil specimens, which also exhibit a dark-light shell banding pattern in both day-night bands and seasonal sunny-rainy-season banding (Arndt et al., 2025). While seasonal variability is certainly influenced by various factors on scales not present in daily cycles, insights into the effects of light availability and metabolic 615 activity provide useful additional information to consider alongside environmental factors when interpreting palaeo-seasonality. Together with the analysis of recent clams grown in a natural environment (e.g., Arias-Ruiz et al., 2017; Elliot et al., 2009; de Winter et al., 2023; Yan et al., 2020), information from such short-term culturing experiments helps distinguish – or even link – biological effects and environmental signals, aiding in the production of accurate reconstructions of past conditions. The results presented here facilitate this by providing 620 unambiguous evidence of the portion of the shell that is grown during the day.

Furthermore, daily cyclicity provides the possibility of quantifying daily cycle wavelengths as a useful tool for 625 obtaining a high-resolution age model for geochemical datasets, independent of daily band visibility (Arndt et al., 2023). Having daily resolved palaeoclimate data at this time scale facilitates the identification of cyclic variability, including seasonality and multi-annual atmospheric and oceanographic oscillations such as the El Niño–Southern Oscillation, via spectral analysis (Arndt et al., 2025). In addition, knowledge of how giant clams build their shells daily may facilitate the interpretation of short-term geochemical signals linked to rapid changes in the environment, such as extreme weather events. All of these approaches require the driving factors behind element cyclicity to be understood within the context of the biomineralisation process of the organism. This is only possible when physiological versus kinetic chemical signals can be disentangled, which requires knowledge of how and when different domains within the shell were formed.

630 We gain new insights about the detailed timing of the Mg/Ca and Sr/Ca increase and decrease over the course of a day by comparing the differential incorporation of the ^{135}Ba isotope tracer, supplied only during the day or during nighttime, to the Mg/Ca and Sr/Ca cycles within the shells. We can unequivocally confirm that Mg/Ca and Sr/Ca increase at nighttime in the narrower low growth bands, but the highest Mg/Ca values do not correspond to midnight, as previously suggested (e.g., Brosset et al., 2025). The clams that grew with the tracer introduced at nighttime (from 19:30 to 7:30) begin to incorporate the tracer into the shell approximately during the Mg/Ca and Sr/Ca minimum, with the tracer concentration decreasing within the shell shortly after the Mg/Ca and Sr/Ca maximum is reached. Similarly, the clams that grew in the presence of the tracer during the day (from 7:30 to 19:30) are characterised by a tracer onset in the shell approximately coincident with the Mg/Ca and Sr/Ca maximum. The tracer signal decreases shortly after the Mg/Ca and Sr/Ca minimum is reached. The isotopic tracer administered for 12 h each can only be incorporated as observed if the Mg/Ca and Sr/Ca maxima reflect the early 635 morning hours or the end of the night, while the Mg/Ca and Sr/Ca minima are reached in the evening hours or towards the end of the day.

5 Conclusion

640 Here, we present calcification and geochemical data from eight *Tridacna* specimens grown under controlled conditions in the laboratory in seawater labelled with a ^{135}Ba isotope spike, introduced during either the day or night only. Our findings indicate that calcification rates in *Tridacna* sp. are nearly five times higher during the day compared to those at night. The EPF is replenished by ions from the surrounding seawater during both day and night, resulting in daily elemental heterogeneity of the shell. We demonstrate that with the uptake and washout of elements starting quickly, likely within tens of minutes, but requiring approximately one day to fully replace the barium pool, and assuming similar behaviour, that of calcium inreplenish the EPFs composition. During the 645 daytime, B/Ca, Mg/Ca, Sr/Ca and Ba/Ca decrease while Na/Ca increases, and vice versa during nighttime. Our data also demonstrate that theThe night peak, often best seen in Mg/Ca, happens in the early morning hours rather thannot at midnight, while the day peak reflects the evening hours rather thannot the middle of the day. Daily El/Ca cycles might be affected by growth rate dynamics, while the elemental composition ofeompositions in the EPF could in turn affect growth rates. It is, however, likely that daily El/Ca cycles are primarily caused by changes 650 in the intensity of active Ca^{2+} and HCO_3^- transport-channelling into the EPF. This reduces, reducing most El/Ca ratios by increasing [through relatively higher Ca^{2+}] relative to the other elements⁺ availability during the day, with the exception ofexcept Na^{+} which is strongly involved in Ca^{2+} , HCO_3^- and H^+ transport (e.g. via Na^+ - HCO_3^- co-transport).channelling. The changes in active pumping into and from the EPF could in turn be dependent on light availability, circadian rhythms or simply be an expression of the energy available for transportehannelling. 655 We therefore propose that active pumping of Ca^{2+} and HCO_3^- into the EPF might also cause diurnal changes in growth rate, organic content and shell structure. Overall, this study enhances ourthe knowledge base for utilizing *Tridacna* sp. as a high-resolution palaeoclimate archive, particularly for sub-daily time scale applications, by improving the spatial resolution of LA-ICPMS and deepening ourthe understanding of diurnal growth patterns, variability in elemental composition, and therefore rapid responses to environmental changes.

665 Competing interests

The authors declare that they have no conflict of interest.

Data availability

All data, [supplementary tables](#) and supplementary [figuretables](#) presented in this study ([Fig. Tables S1, Tab. S1 – S5 – S3](#)) are available at Zenodo with licence CC BY 4.0 (*link will be inserted at later stage*).

670 Authors contributions

IA: Conceptualization, Data curation, Formal analysis, Funding acquisition, Investigation, Validation, Visualization, Writing (original draft preparation), Writing (review and editing); JE: Conceptualization, Funding acquisition, Investigation, Methodology, Project administration, Resources, Supervision, Validation, Writing (review and editing); DE: Conceptualization, Data curation, Supervision, Validation, Writing (review and editing);
675 TE: Data curation, Investigation, Methodology, Supervision, Validation, Writing (review and editing); AL: Investigation, Validation, Writing (review and editing); WM: Conceptualization, Data curation, Funding acquisition, Methodology, Project administration, Resources, Supervision, Validation, Writing (review and editing)

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