Post-review adjustments

HO• and •OH can be used interchangeably in Chemistry, same as O• and •O, or any other radical. They mean the same thing and the • symbol on O simply indicates that the **free electron** rests on the oxygen atom. In some situations, such as hydrogen bonded complexes where the **hydrogen bonding** is expressed by the ••• symbol, either HO• or •OH can be used in such a way that the structure can be clearly understood by the reader. Putting • and ••• symbols close to each other can be misleading.

Hence, for example, "HO••••CH_3-O-SO_3H" is preferably written as "•OH•••CH_3-O-SO_3H" to highly the presence of both the **free electron** and the **hydrogen bonding**. Originally, such was written as "HO•···CH_3-O-SO_3H", which clearly highlighted both the **free electron** and the **hydrogen bonding**. However, the ··· symbol was changed to ••• symbol during the typesetting and copying-editing processes.

Based on the above, the following changes are required.

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Line 79: Change "HO••••CH_3-O-SO_3H•••$(H_2O)_{n=0-2}$" to "•OH•••CH_3-O-SO_3H•••$(H_2O)_{n=0-2}$".

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Line 15: Change "HO••••CH_3-O-SO_3H•••$(H_2O)_n$" to "•OH•••CH_3-O-SO_3H•••$(H_2O)_n$".

Line 21: Change "HO••• • CH_3-O-SO_3H•••$(H_2O)_{n=0-2}$" to "•OH•••CH_3-O-SO_3H•••$(H_2O)_{n=0-2}$".

Line 25: Change "HO••• •CH_3-O-SO_3H•••$(H_2O)_{n=0-2}$" to "•OH•••CH_3-O-SO_3H•••$(H_2O)_{n=0-2}$".

Line 35: Change "HO••••CH_3-O-SO_3H•••$(H_2O)_n$" to "•OH•••CH_3-O-SO_3H•••$(H_2O)_n$".

Line 53: Change "SO_4^-••••H_3O^+•••$(H_2O)_{n-1}$" to "$SO_4{\bullet}^-$,H_3O^+•••$(H_2O)_{n-1}$". Here, the ••• symbol can be replaced by "**,**" not only to remove the confusion that may arise from ••••, but also to highly the existence of an **ion-pair** between $SO_4{\bullet}^-$ and H_3O^+., which is a common representation in Chemistry.

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Line 39: Change "O_2••••O-H_2C-O-SO_3^-" to "O_2•••O•-H_2C-O-SO_3^-".