The Critical Role of Volatile Organic Compounds Emission in Nitrate Formation in Lhasa, Tibetan Plateau: Insights from Oxygen Isotope Anomaly Measurements

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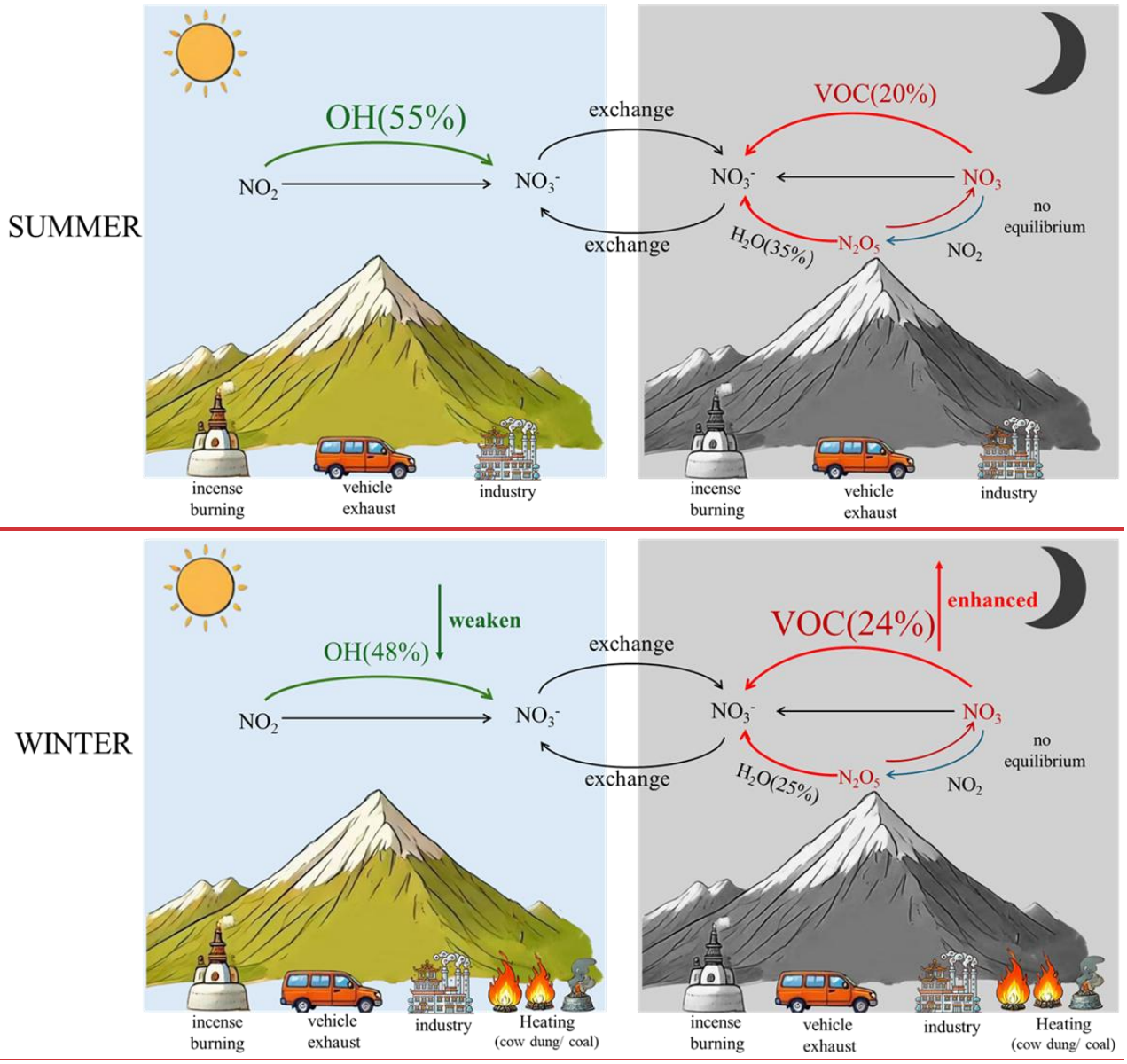
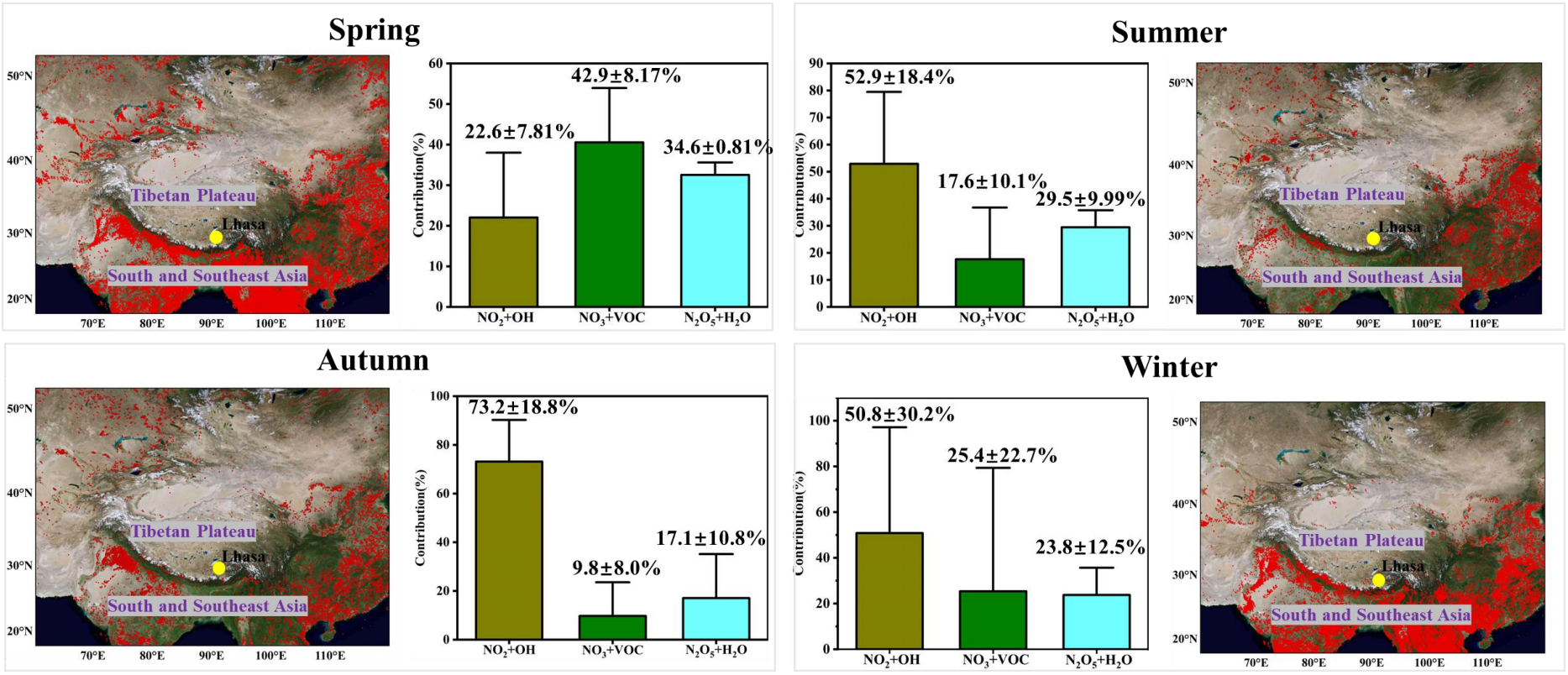

SUMMER
OH(55%)
NO2
NO3-
exchange
exchange
VOC(20%)
NO3-
NO3
N2O5
H2O(35%)
NO2
no equilibrium
incense burning
vehicle exhaust
industry
incense burning
vehicle exhaust
industry
WINTER
weaken
OH(48%)
NO2
NO3-
exchange
exchange
enhanced
VOC(24%)
NO3-
NO3
N2O5
H2O(25%)
NO2
no equilibrium
incense burning
vehicle exhaust
industry
Heating (cow dung/ coal)
incense burning
vehicle exhaust
industry
Heating (cow dung/ coal)


Spring
Tibetan Plateau
Lhasa
South and Southeast Asia
50°N
40°N
30°N
20°N
70°E
80°E
90°E
100°E
110°E
Contribution(%)
22.6±7.81%
42.9±8.17%
34.6±0.81%
NO2+OH
NO3+VOC
N2O5+H2O
Summer
52.9±18.4%
17.6±10.1%
29.5±9.99%
Autumn
73.2±18.8%
9.8±8.0%
17.1±10.8%
Winter
50.8±30.2%
25.4±22.7%
23.8±12.5%

TOC art

Abstract

Atmospheric particulate nitrate aerosol (NO_3^-), produced via the oxidation of nitrogen oxides (NO_x = NO + NO_2), plays an important role in atmospheric chemistry and air quality, yet its formation mechanism remains~~still~~ poorly constrained in the plateau region. In this study, we ~~first~~ reported for the first time the yearly variation ~~of~~ in the signatures ~~for~~ of the stable oxygen isotope anomaly ($\Delta^{17}O = \delta^{17}O - 0.52 \times \delta^{18}O$) in NO_3^- collected in the urban ~~region of Lhasa city~~ Lhasa(3650 m a.s.l), on the Tibetan Plateau, China. Our results show that NO_2 + OH ~~is~~ was the largest contributor to NO_3^- formation (46 ± 26%), followed by NO_3 + VOC (26 ± 18%), and N_2O_5 + H_2O (28 ± 11%) using the Bayesian Isotope Mixture Model. Notably, there ~~are~~were significant differences in the NO_2 + OH, NO_3 + VOC, and N_2O_5 + H_2O pathways between spring and the other three seasons (T test, $p < 0.05$). By Hybrid Single-Particle Lagrangian Integrated Trajectory (HYSPLIT) dispersion model, we~~Our results~~ highlighted the influence of VOC emissions from regions such as Afghanistan and northern India, which enhanced NO_3^- concentrations in Lhasa during spring. Furthermore, the diurnal distribution of NO_3^- oxidation pathways varied distinctly across seasons, suggesting that these differences in NO_3^- pathways are attributed to aerosol liquid water content (ALWC), volatile organic compounds (VOC) concentrations, and ~~pollution level~~atmospheric lifetime of NO_3^-~~s~~.

Keywords: nitrate, $\Delta^{17}O$-NO_3^-, oxidation pathways, Lhasa, VOC

1. Introduction

Nitrate aerosol (NO_3^-) is a key component regulating the mass concentration of atmospheric fine particulate matter ($PM_{2.5}$), which is highly related with air quality (Colmer et al., 2020), public health (Zhang et al., 2019; Zhang et al., 2017; Geng et al., 2021), and climate system (Clark and Tilman, 2008). Globally, the mass contribution of NO_3^- in $PM_{2.5}$ is in the range of 5-30% (Huang et al., 2014; Xu et al., 2019; Salameh et al., 2015; Espina-Martin et al., 2024; Bell et al., 2007; Sun et al., 2022), depending on the locations and the severities of air pollution. For example, it was reported that NO_3^- accounts for 22%, 27% and 26% of $PM_{2.5}$ in ~~the $PM_{2.5}$ of~~ megacities in China (Zong et al., 2020), Europe (Espina-Martin et al., 2024) and U.S. (Sun et al., 2022), respectively. In addition, some studies found that the contribution of NO_3^- would increase by 3-8 times with the occurrence of the particular-derived haze pollution (Ge et al., 2024; Song et al., 2019; Yin et al., 2022; Walters et al., 2024).

It is well-known that atmospheric NO_3^- is formed by the oxidation of nitrogen oxides (NO_x=NO+NO_2) with different oxidants such as O_3, OH and RO_2 (Text S1). In general, atmospheric chemical transportation models are employed to depict the detailed oxidation pathways of NO_3^- formation. However, there remains considerable uncertainty in modelling the contribution of individual oxidation pathways to NO_3^- formation, particularly the N_2O_5 + H_2O pathway, due to the wide variability of key parameters such as the N_2O_5 uptake coefficient, which has been shown to vary significantly depending on aerosol composition, relative humidity, and temperature.~~However, there are a large uncertainty in modeling the contribution of oxidation pathways to NO_3^- formation using this methodology due to the variable parameters in the real atmosphere.~~ For example, it was reported that the predicted N_2O_5 uptake to NO_3^- formation in Beijing, as estimated using WRF-Chem, ranges from 5% to 21% (Su et al., 2017). Higher contributions between 66% and 85% have been observed when applying the CMAQ model in Beijing (Qiu et al., 2019). Therefore, the application of alternative techniques is crucial for providing more reliable estimates and enhancing our understanding of NO_3^- formation mechanisms, in addition to the insights gained from atmospheric chemical transportation models.

Stable oxygen isotope anomaly ($\Delta^{17}O = \delta^{17}O - 0.52 \times \delta^{18}O$) is recognized as a powerful tool to track formation pathways of atmospheric NO_3^- (Zhang et al., 2024; Feng et al., 2023). This is because the oxygen atoms in the terminal positions of O_3 exhibit an elevated $\Delta^{17}O$ ($\Delta^{17}O = 39 \pm 2‰$)~~,~~ (Vicars

and Savarino, 2014), whereas the $\Delta^{17}O$ values of other atmospheric oxidants (e.g., H_2O, OH, and RO_2) that can be incorporated to NO_3^- are very close to 0‰. (Dubey et al., 1997; Barkan and Luz, 2003; Alexander et al., 2020) Therefore, $\Delta^{17}O(NO_3^-)$ serves as a unique tracer of O_3 involvement in its formation pathways, offering valuable insights into the relative contributions of individual reactions. In recent years, the use of $\Delta^{17}O(NO_3^-)$ to elucidate NO_3^- formation has garnered considerable attention. Walters et al. (2024) reported that the major formation pathways of annual HNO_3 production in the northeastern U.S. ~~are~~ were NO_2+OH (46%), N_2O_5 uptake (34%), and organic nitrate hydrolysis (12%), with notable seasonal variability. Additionally, Zhang et al. (2022) observed that the contribution of nocturnal chemistry to NO_3^- formation ~~increases~~ increased at night, peaking at 72% around midnight. In contrast, the contribution of NO_2+ OH ~~rises~~ rose with sunrise, reaching its highest fraction (48%) around noon. However, nearly all current $\Delta^{17}O$-related observations have been conducted in the plain cities, with little attention given to plateau cities, where atmospheric conditions generally suffer from distinct energy consumption patterns and unique climatic factors (e.g., intense solar radiation). In this study, we present detailed results from comprehensive field observations conducted in Lhasa (3650 m a.s.l), one of the highest cities in the world, located on the Tibetan Plateau, China. For the first time, we quantify the relative contribution of three oxidation pathways to NO_3^- formation in Lhasa on the basis of ambient measurements for $\Delta^{17}O$ signatures in NO_3^-.

2. Materials and methods

2.1 Sampling campaign

$PM_{2.5}$ samples were collected on the roof of a building (~15 m above ground) at the Meteorological Bureau of Lhasa (91.08°E, 29.40°N; Figure 1) in China. Lhasa, the capital of the Tibet Autonomous Region, is a rapidly developing city with a population of ~ 950000 and an urban area of ~ 30000 km^2 (Lhasa). The sampling site is surrounded by mixed land use, including residential areas, government offices, religious temples and commercial zones, with minimal heavy industry. The strong solar radiation and large diurnal temperature variations in this sampling site can lead to pronounced changes in boundary layer height, which in turn significantly influence vertical mixing and the transport of air pollutants. ~~This site is located in Chengguan, a typical urban area of Lhasa characterized by a dense population and nearby temples.~~

The sampling campaign was conducted from June 2022 to July 2023 using a high-volume $PM_{2.5}$ sampler, which operated at a flow rate of 1.0 m^3/min. Samples were collected once a week, with each sampling session lasting 48 hours, except during intensive sampling periods in the summer (June 30 to July 14, 2022) and winter (January 28 to February 7, 2023). During these intensive periods, each sample was collected for 12 hours, from 8:00 to 20:00 and 20:00 to 8:00 on the following day, respectively. During the autumn of 2022, Lhasa experienced intermittent COVID-19 control measures, including restricted movement, reduced traffic activity, and temporary lockdowns in urban areas (Daily). Before sampling, all quartz filters (8 in. × 10 in., Pallflex) were calcined in a muffle furnace at 450 °C for 6 h to prevent impurities from contaminating the collected $PM_{2.5}$ samples. After sampling, the samples were collected and stored in a freezer at -20°C.

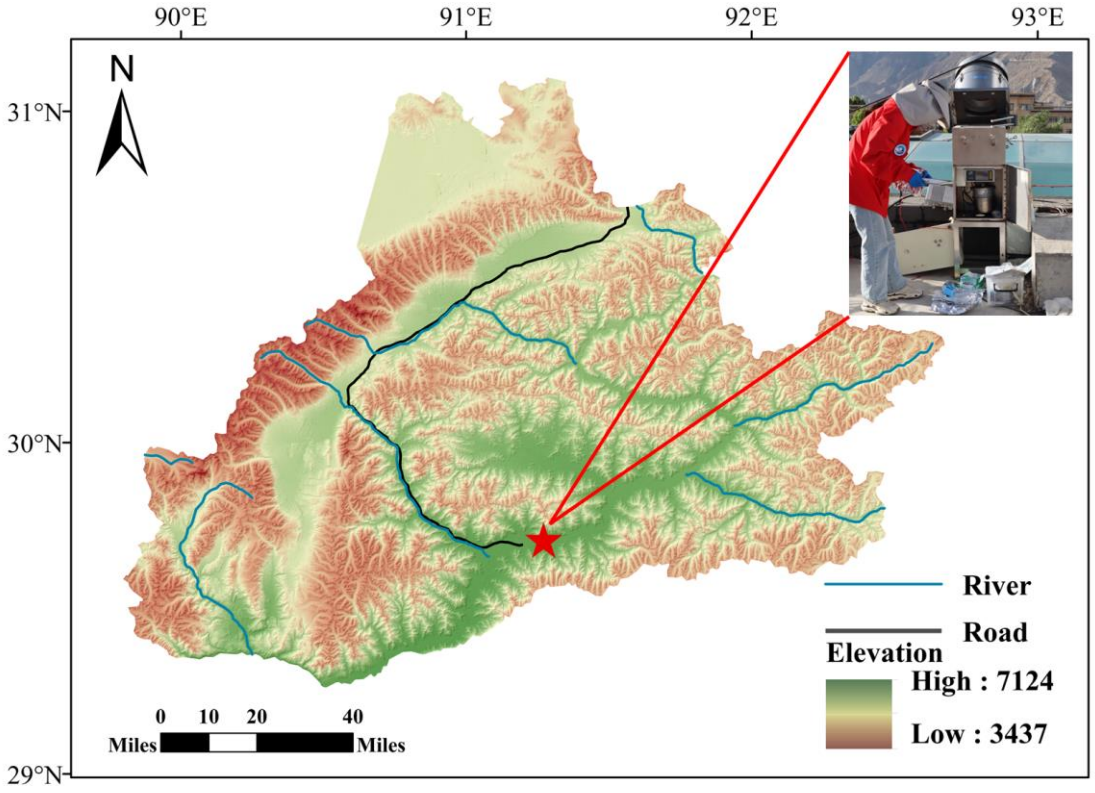


Figure 1. Geographic position of sampling site in Lhasa, China.

2.2 Measurements of water-soluble ions and isotopes

Water-soluble ions were measured by an ion chromatography (Dionex ICS-5000, Thermo Scientific Inc.) (Chen et al., 2022). In brief, a part of filter membranes (4.54 cm^2) was cut using a 17 mm diameter punch and placed in a 15 mL centrifuge tube with 10 mL of 18.2 MΩ ultrapure water. The tube was then subjected to ultrasonic treatment in an ice-water bath for 30 min to prevent ion

volatilization. The extract was filtered through a 0.22 μm filter into a 30 mL sample bottle. This process was repeated with an additional 10 mL of water to ensure full extraction. The final extract was analyzed by an ion chromatography. The method detection limits (MDLs) ~~of~~ for Cl^-, NO_3^-, SO_4^{2-}, Na^+, NH_4^+, K^+, Mg^{2+}, ~~,~~ and Ca^{2+} ~~was~~ were ~~0.003~~0.001 mg/L, 0.001 mg/L, 0.003 mg/L, 0.02 mg/L, 0.01 mg/L, 0.02 mg/L, 0.006 mg/L, and 0.02 mg/L, respectively.

Stable oxygen isotopes ($\delta^{17}O$, $\delta^{18}O$, $\Delta^{17}O$, and $\Delta^{17}O = \delta^{17}O - 0.52 \times \delta^{18}O$) of NO_3^- were determined using an isotope ratio mass spectrometer (MAT253, Thermo Fisher Scientific, USA) at Nanjing University of Information Science and Technology (Fan et al., 2021; Zhang et al., 2022). Briefly, NO_3^- from filter extractions (containing at least 0.8 μg N) ~~was~~were converted into gaseous N_2O using the bacterial denitrifier method. N_2O was then further thermally decomposed into O_2 and N_2 in a gold tube heated to 800°C. The produced O_2 was analyzed for oxygen isotopes by isotope ratio mass spectrometer. The duplicated analysis showed that the errors were within 1.32% for $\Delta^{17}O$-NO_3^-.

2.3 Primary data sources

Meteorological parameters, including ambient temperature (T), relative humidity (RH), rainfall, radiation, wind direction (WD) and wind speed (WS) during the sampling campaign, were obtained from the Meteorological Bureau of Lhasa. Additionally, NO_2 and O_3 during the sampling campaign were downloaded from the National Meteorological Information Center (https://air.cnemc.cn:18007/~~http://air.cnem.cn:18007/~~).

2.4 Evaluation of NO_3^- oxidation pathways

In our study, we aimed to quantify the relative contribution of different oxidation pathways to NO_3^- production based on $\Delta^{17}O$-NO_3^-. Due to the low Cl^- concentrations observed in Lhasa, the NO_3^- formation pathways considered in this study are limited to NO_2 + OH, NO_3 + VOC, and N_2O_5 + H_2O. Although NO_3 + VOC was generally considered a minor pathway in continental regions (Alexander et al., 2009), we included it because elevated VOC concentrations were observed at our sampling site in Lhasa, influenced by both biogenic emissions (e.g. incense burning) and anthropogenic sources (e.g. vehicle emissions) (Tang et al., 2022). The relative contributions of the three pathways were determined using a $\Delta^{17}O$-based mass balance approach (Michalski et al., 2003), as shown in Equations (1) and (2):

$$\Delta^{17}O\text{-}NO_3^- = (\Delta^{17}O\text{-}NO_3^-)_{NO2+OH} \times f_{NO2+OH} + (\Delta^{17}O\text{-}NO_3^-)_{NO3+VOC} \times f_{NO3+VOC} + (\Delta^{17}O\text{-}NO_3^-)_{N2O5+H2O} \times f_{N2O5+H2O} \quad (1)$$

$$f_{NO2+OH} + f_{NO3+VOC} + f_{N2O5+H2O} = 1 \quad (2)$$

where $\Delta^{17}O\text{-}NO_3^-$ value is the $\Delta^{17}O$ value of NO_3^- in $PM_{2.5}$. The $(\Delta^{17}O\text{-}NO_3^-)_{NO2+OH}$, $(\Delta^{17}O\text{-}NO_3^-)_{NO3+VOC}$, and $(\Delta^{17}O\text{-}NO_3^-)_{N2O5+H2O}$ correspond to the $\Delta^{17}O$ values from NO_2+OH, NO_3+VOC and N_2O_5+H_2O, respectively. The $\Delta^{17}O$ values for each pathway were calculated using Equations (3), (4), and (5) (Savarino et al., 2016; Alexander et al., 2009):

$$(\Delta^{17}O\text{-}NO_3^-)_{NO2+OH}\ (‰) = 2/3\alpha \times \Delta^{17}O\text{-}O_3^* \quad (3)$$

$$(\Delta^{17}O\text{-}NO_3^-)_{NO3+VOC}\ (‰) = 2/3\alpha \times \Delta^{17}O\text{-}O_3^* + 1/3 \times \Delta^{17}O\text{-}O_3^* \quad (4)$$

$$(\Delta^{17}O\text{-}NO_3^-)_{N2O5+H2O}\ (‰) = 1/3\alpha \times \Delta^{17}O\text{-}O_3^* + 1/2(2/3\alpha \times \Delta^{17}O\text{-}O_3^* + 1/3 \times \Delta^{17}O\text{-}O_3^*) \quad (5)$$

Previous studies have demonstrated a linear correlation between $\Delta^{17}O\text{-}O_3$ and $\Delta^{17}O\text{-}O_3^*$, with $\Delta^{17}O(O_3)$ values ranging from 20% to 40% in tropospheric O_3 (Vicars and Savarino, 2014; Ishino et al., 2017). The equations are shown as follows (Vicars et al., 2012):

$$\Delta^{17}O\text{-}O_3^* = 1.5 \times \Delta^{17}O\text{-}O_3 \quad (6)$$

Based on previous observations of tropospheric O_3, $\Delta^{17}O\text{-}O_3^*$ average value was approximately 39‰. The α value represents the proportional contribution of O_3 to the NO oxidation pathway and can be estimated using the following equations (7) (Alexander et al., 2009). When NO*x* is in photochemical steady state, $\Delta^{17}O\text{-}NO_2$ can be represented using the following equation (10):

$$\alpha = K_{P1}[O_3] \times [NO] / (K_{P1} \times [O_3] \times [NO] + K_{P2} \times [NO] \times [HO_2] + K_{P3} \times [NO] \times [RO_2]) \quad (7)$$

$$K_{P1} = 3.0 \times 10^{-12} \times e^{(-1500/T)} \quad (8)$$

$$K_{P2} = K_{P3} = 3.5 \times 10^{-12} \times e^{(270/T)} (cm^3 \cdot molecule^{-1} \cdot s^{-1}) \quad (9)$$

$$\Delta^{17}O\text{-}NO_2 = \alpha \Delta^{17}O\text{-}O_3^* \quad (10)$$

where T represents the ambient temperature (K) (Kunasek et al., 2008). The HO_2 mixing ratios were estimated using empirical equations in the absence of direct HO_2 observations (Kanaya et al., 2007). Due to the lower temperatures in Lhasa during non-summer seasons, HO_2 concentrations were assessed using a formula derived from winter conditions.

Winter

$$[HO_2\cdot]/ppt = \exp(5.7747 \times 10^{-2} [O_3]\ (ppb) - 1.7227) \text{ for daytime} \quad (11)$$

$$[HO_2\cdot]/ppt = \exp(7.7234 \times 10^{-2} [O_3]\ (ppb) - 1.6363) \text{ for nighttime} \quad (12)$$

Summer

$$[HO_2\cdot]/pptv = \exp(2.0706 \times 10^{-2} [O_3]\ (ppb) + 1.0625) \text{ for daytime} \quad (13)$$

$$[HO_2\cdot]/pptv = 0.2456 + 0.1841\ [O_3]\ (ppb) \text{ for nighttime} \quad (14)$$

2.4 5 Stable isotope analysis in the R (SIAR) model

In this study, stable isotope analysis in the R (SIAR) model was employed to estimate the relative contributions of three main pathways to NO_3^- (Parnell et al., 2010). The SIAR model is well-suited for analyzing multiple formation pathways, as it effectively incorporates uncertainties and parameter variability, leading to more reliable estimates. Specifically, this model allows for a detailed analysis of oxygen isotope ($\Delta^{17}O$), enabling accurate modeling of NO_3^- formation pathways based on oxygen isotope measurements. The SIAR model is a Bayesian mixture model, mathematically formulated as follows:

$$X_i = \sum_{j=1}^{K} p_j \times f_{ij}$$

$$p_1 + p_2 + \cdots + p_k = 1$$

$$f_{ij} \sim N(\mu_j, \omega_j^2)$$

Where X_i is the observed $\Delta^{17}O$ values for sample i ($i = 1, 2, 3, ...,N$) and p_j is the proportional contribution of each NO_3^- formation pathway j to the sample i. f_{ij} is the $\Delta^{17}O$ values of formation pathway j for sample i and follows a normal distribution with mean (μ_j) and variance (ω_j^2). Within the Bayesian framework, prior distributions are assigned to each p_j, and these are updated with the observed data X_i to obtain posterior distributions, allowing for inference of the proportional contributions p_j of each pathway.

2.5 6 Aerosol liquid water content (ALWC) and the Hybrid Single-Particle Lagrangian Integrated Trajectory (HYSPLIT)

To evaluate the influence of ~~aerosol liquid water content (~~ALWC~~)~~ on the NO_3^- formation, ALWC was calculated using the ISORROPIA II model developed by Fountoukis and Nenes (Fountoukis and Nenes, 2007). The ISORROPIA II model includes two modes: the forward mode, which requires the concentrations of both particulate and gaseous pollutants concentrations as inputs, and the reverse mode, which only requires the concentrations of particulate pollutants concentrations. The model computes the ALWC in both modes based on particulate pollutant concentrations (e.g., NH_4^+, Na^+, Ca^{2+}, K^+ and Mg^{2+}), as well as ambient RH and T. In this study, the reverse mode was employed due to the lack of gaseous pollutant concentrations observations.

Additionally, the Hybrid Single-Particle Lagrangian Integrated Trajectory (HYSPLIT) model was

utilized to compute 72-hours back trajectories during the sampling campaign. HYSPLIT, developed by the National Oceanic and Atmospheric Administration Air Resources Laboratory (NOAA/ARL), is available on the their website (https://www.ready.noaa.gov/HYSPLIT.phphttp://www.arl.noaa.gov/ready/hysplit4.html),). This model has been widely used for simulating the transport and dispersion trajectories of pollutants such as $PM_{2.5}$, VOC_s, O_3, and NO_x, among others (He et al., 2022; Zhao et al., 2015; Cao et al., 2023). Backward trajectories for each sampling day were calculated at an altitude of 3650 meters using meteorological data from the Global Data Assimilation System (GDAS), available through the US Air Resources Laboratory (NOAA ARL) (https://www.ready.noaa.gov/data/archives/gdas1/).

3. Results

3.1 Overview of the meteorological parameters in Lhasa during the sampling campaign

Figure 2a presents the daily variations in meteorological parameters, including temperature, relative humidity (RH), rainfall and solar radiation. During the sampling campaign, the annual average temperature was 11.5℃, ranging from -2.83 to 24.2℃. The highest average temperature was observed in summer (19.7℃), while the lowest (3.11℃) was recorded in winter. Relative humidity (RH) varied between 6.67 and 66.8%, with the lowest average RH occurring in winter (17.1%) and the highest in summer (35.6%). The near-surface layer of Lhasa is influenced by a thermal low-pressure system, and the southwest monsoon, active between June and September, transports moisture-laden air from the Indian Ocean, resulting in increased rainfall during summer. Solar radiation intensity exhibited a seasonal trend consistentopposite to those of temperature and RH, peaking in summer (394 W/m^2) and reaching its lowest levels in winter (220 W/m^2). The dominant wind direct (WD) was southeast in spring, but southwest in the other three seasons (Figure 3). Wind speed (WS) was highest in spring but lowest in autumn.

3.2 NO_3^- concentration

NO_3^- mass concentrations ranged from 0.10 to 1.72 μg/m^3, with an average value of 0.62 ± 0.31 μg/m^3. NO_3^- concentrations exhibited distinct seasonal patterns. As shown in Figure S1, the equivalent concentrations of [SO_4^{2-} + NO_3^-] were considerably higher than those of [NH_4^+], indicating that NH_4^+ was insufficient to fully neutralize NO_3^-. This suggests that a portion of NO_3^- may have existed in other

forms, such as KNO_3 and $Ca(NO_3)_2$. This inference is supported by the strong positive correlations between NO_3^- and K^+ ($r = 0.64$, $p < 0.1$) and Ca^{2+} ($r = 0.43$, $p < 0.01$), especially in spring, as shown in Figure S2. In contrast, NO_3^- showed relatively weak negative correlations with T ($r = -0.27$, $p < 0.01$) and RH ($r = -0.22$, $p < 0.1$), indicating that under the specific atmospheric conditions in Lhasa, meteorological parameters might not be the dominant factors controlling the gas-particle partitioning of NO_3^-. ~~Specifically, t~~The maximum monthly average values of NO_3^- concentration occurred in spring (0.83 ± 0.35 μg/m^3) with the instantaneous maximum reaching 1.72 μg/m^3, whereas the lowest was recorded in autumn (0.23 ± 0.13 μg/m^3) with an instantaneous minimum of only 0.09 μg/m^3 (Table 1). The elevated NO_3^- concentrations in spring could be attributed to biomass burning emitted from south and Southeast Asia (Figure S3/Figure S4). The strong between NO_3^- and K^+ in spring further this explanation. ~~+The difference in NO_3^- concentrations between autumn and other three seasons was statistically significant ($p < 0.05$). The considerable seasonal variation in NO_3^- concentrations in Lhasa is associated with varying emission sources and meteorological conditions.~~

In spring, high NO_3^- concentrations were associated with weak southeasterly winds (< 3 m/s) in the bivariate polar plot, suggesting probable impacts from local~~/regional~~ emissions (Figure 3). The southeasterly sector of sampling site includes residential areas, agriculture land and major transportation routes, which are potential NO*x* sources. In spring, intensified agriculture activities (e.g., fertilization, biomass burning) might increase NO*x* emissions. Meanwhile, low wind speeds likely limit atmospheric dispersion, promoting the local accumulation of precursors and enhancing NO_3^- production. During the rainy summer, shorter NO_3^- lifetimes indicated a weak influence from regional transport, with a more pronounced contribution from local~~/regional~~ emissions. In autumn, NO_3^- concentrations were relatively low, which coincided with strict local COVID-19 restrictions in Lhasa. These measures significantly reduced human activity and traffic, leading to suppressed local emissions.~~partly due to the reduced local emissions caused by the COVID-19 pandemic, which led to decreased human activity and transportation.~~ Despite ~~Low~~ low wind speeds typically favor pollutant accumulation, NO_3^- concentrations remained low, suggesting that both reduced local sources and seasonal meteorological conditions constrained NO_3^- production. Nevertheless, the persistence of measurable NO_3^- under such stagnant conditions also implied a potential contribution from regional transport during this period. ~~, the lower NO_3^- levels during this period suggest that the pandemic, combined with seasonal meteorological conditions, played a role in limiting local/regional emissions.~~

In winter, elevated NO_3^- concentrations under low wind speeds (< 3 m/s) emphasized the significant contribution of local~~/regional~~ emissions. These findings underscored that~~, in addition to~~ both regional transport and~~,~~ local emissions ~~are~~were~~–~~ ~~the primary~~important contributors to NO_3^- concentrations in Lhasa~~, especially under low wind conditions, with the pandemic further influencing local emission levels~~. Furthermore, based on our day-night sampling scheme, no nychthemeral (day-night) differences in NO_3^- concentrations were detected (Table S1). A similar day-night pattern of NO_3^- concentrations also has been observed in in Beijing (Luo et al., 2020).

Table 1 Average values of water-soluble ions and $\Delta^{17}O\text{-}NO_3^-$ during the sampling campaign

		Na^+ μg/m^3	NH_4^+ μg/m^3	K^+ μg/m^3	Mg^{2+} μg/m^3	Ca^{2+} μg/m^3	Cl^- μg/m^3	NO_3^- μg/m^3	SO_4^{2-} μg/m^3	$\Delta^{17}O\text{-}NO_3$- ‰
Annual	**Minmum**	0.02	0	0.004	0.004	0.004	0.004	0.09	0.06	18.3
	Maximum	0.68	1.22	0.29	0.08	3.52	0.51	1.72	2.37	34.1
	Average	0.16	0.3	0.07	0.02	1.09	0.08	0.62	0.74	26.3
	Std.Dev	0.14	0.26	0.06	0.01	0.7	0.1	0.31	0.45	3.13
spring	**Minmum**	0.04	0.16	0.04	0.01	1.02	0.01	0.45	0.6	27.2
	Maximum	0.16	1.22	0.2	0.05	2.56	0.05	1.72	2.14	30.4
	Average	0.09	0.52	0.09	0.02	1.67	0.03	0.83	1.11	28.8
	Std.Dev	0.03	0.3	0.04	0.01	0.51	0.01	0.35	0.52	0.99
summer	**Minmum**	0.02	0	0.01	0.01	0.03	0.003	0.13	0.18	20.2
	Maximum	0.4	1.08	0.09	0.04	2.4	0.13	1	2.37	28.5
	Average	0.09	0.18	0.03	0.02	1.15	0.3	0.5	0.72	25.5
	Std.Dev	0.08	0.17	0.02	0.01	0.5	0.3	0.23	0.45	2.2
autumn	**Minmum**	0.02	0.003	0.004	0.01	0.004	0.01	0.09	0.06	21.2
	Maximum	0.17	0.11	0.1	0.03	0.24	0.17	0.51	0.55	24.9
	Average	0.09	0.04	0.3	0.02	0.13	0.05	0.23	0.31	23.05
	Std.Dev	0.05	0.04	0.3	0.01	0.08	0.05	0.13	0.14	1.44
winter	**Minmum**	0.06	0.09	0.02	0.01	0.05	0.04	0.21	0.32	18.3
	Maximum	0.56	0.87	0.29	0.08	3.52	0.51	1.46	1.57	34.1
	Average	0.19	0.44	0.12	0.03	1.04	0.16	0.75	0.73	25.9
	Std.Dev	0.12	0.21	0.08	0.02	0.78	0.13	0.28	0.34	3.86

3.3 Oxygen isotopes of NO_3^-

To explore the three major oxidation pathways of NO_3^- formation, 53 samples representing varying NO_3^- concentrations across different seasons were selected for oxygen isotope measurements (Figure 2b). The $\Delta^{17}O\text{-}NO_3^-$ values ranged from 18.3 to 34.1‰, with an average of 26.3 ± 3.13‰, which is slightly lower than the global average of 28.6 ± 4.5‰ simulated by the Global Chemical Transport Model (Alexander et al., 2020). As shown in Table ~~S3~~S2, the observed $\Delta^{17}O\text{-}NO_3^-$ values in this study were similar to most mid- and low-latitude regions, but lower than those in polar regions (~ 32‰). ~~Clear seasonal variations in $\Delta^{17}O\text{-}NO_3^-$ values were seen in Lhasa (Figure 2b).~~ As listed in Table S1, the average $\Delta^{17}O\text{-}NO_3^-$ values in spring, summer, autumn and winter were 28.8 ± 8.0‰, 25.5 ± 2.20‰, 25.6 ± 1.35‰~~,~~ and 25.9 ± 3.56‰, respectively. The differences in $\Delta^{17}O\text{-}NO_3^-$ values between spring and summer, as well as between spring and winter, were statistically significant ($p < 0.05$). The elevated $\Delta^{17}O\text{-}NO_3^-$ values in spring could be attributed to a higher proportion of nocturnal pathways that enrich $\Delta^{17}O\text{-}NO_3^-$ values, such as NO_3 + VOC and N_2O_5 + H_2O pathway. In contrast, the lower $\Delta^{17}O\text{-}NO_3^-$ values in other three seasons suggested a greater production of NO_3^- formation via NO_2 + OH pathway, leading to more negative $\Delta^{17}O\text{-}NO_3^-$ values. Diurnal variation in $\Delta^{17}O\text{-}NO_3^-$ values also differed across season (Figure S5). In summer, the average of $\Delta^{17}O\text{-}NO_3^-$ values during the day (25.3 ± 2.39‰) was lower than at night (26.7 ± 1.03‰). Conversely, in winter, the average of $\Delta^{17}O\text{-}NO_3^-$ values during the day (28.0 ± 3.79‰) was significantly higher than at ~~the night~~night (24.4 ± 3.85‰). Similar diurnal patterns, with higher daytime $\Delta^{17}O\text{-}NO_3^-$values and lower nighttime values, have also been observed in winter in the U.S. (Vicars et al., 2013) and other cities in China (He et al., 2018).

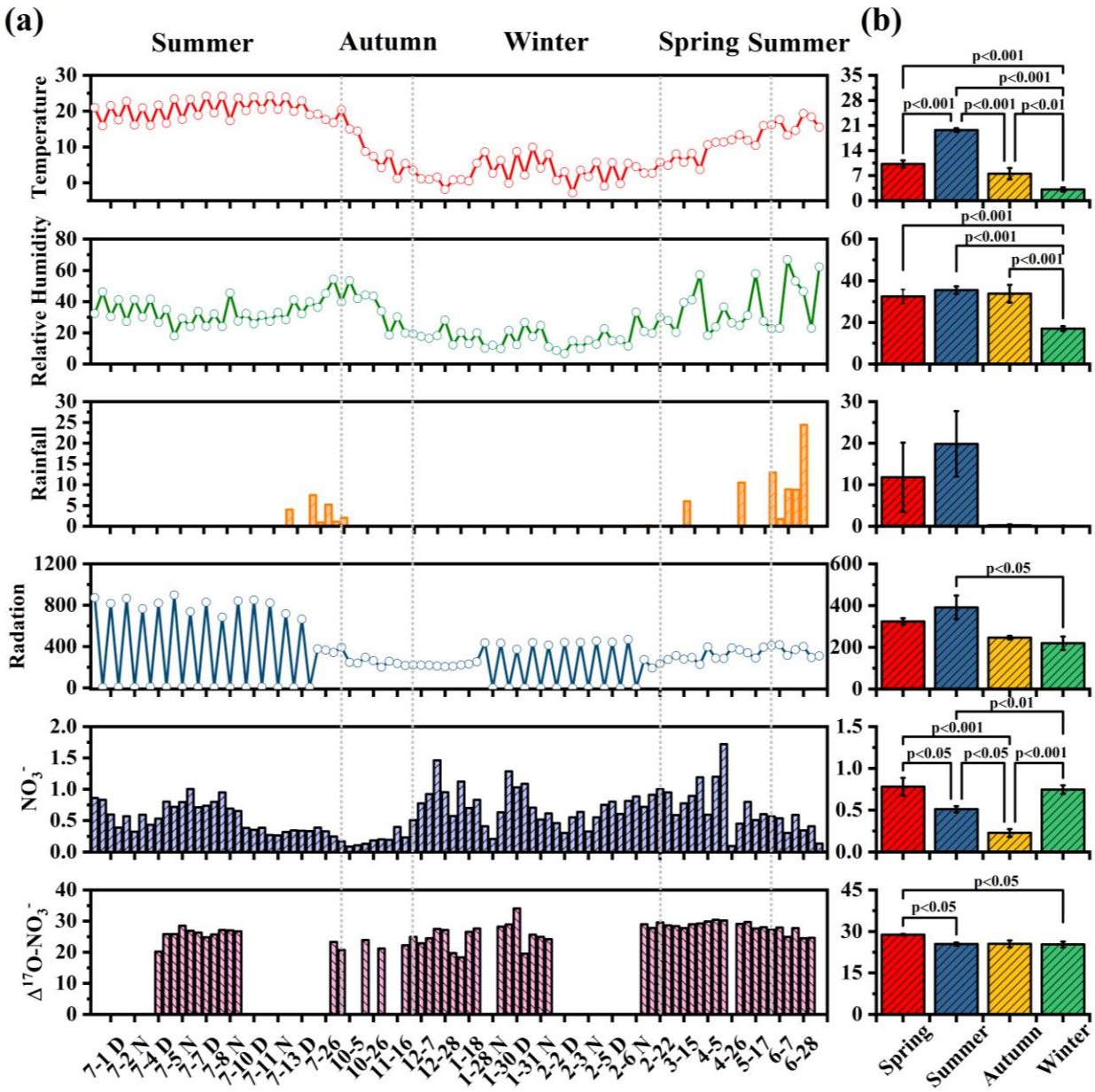
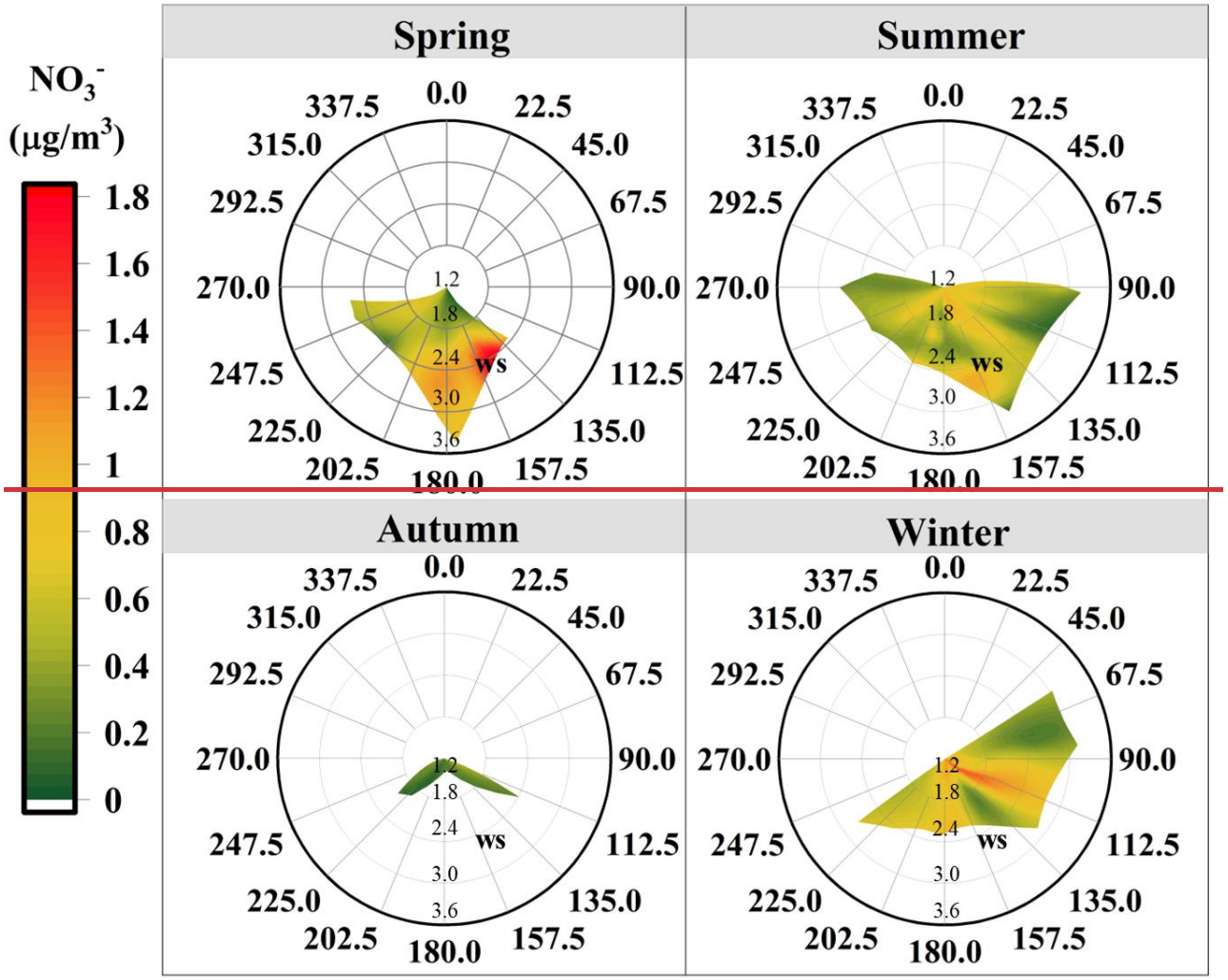


Figure 2. (a) shows the time series of temperature (℃), relative humidity (%), Rainfall (mm), Radiation (W/m^2), NO_3^- concentration ($\mu g/m^3$), and $\Delta^{17}O$- NO_3^- (‰) from June ~~30~~ 2022 to July~~,~~ 2023. (b) shows the average values significance at different seasons with their statistical.


NO3-
(μg/m3)
1.8
1.6
1.4
1.2
1
0.8
0.6
0.4
0.2
0
Spring
Summer
Autumn
Winter
0.0
22.5
45.0
67.5
90.0
112.5
135.0
157.5
180.0
202.5
225.0
247.5
270.0
292.5
315.0
337.5
1.2
1.8
2.4
3.0
3.6
ws

319

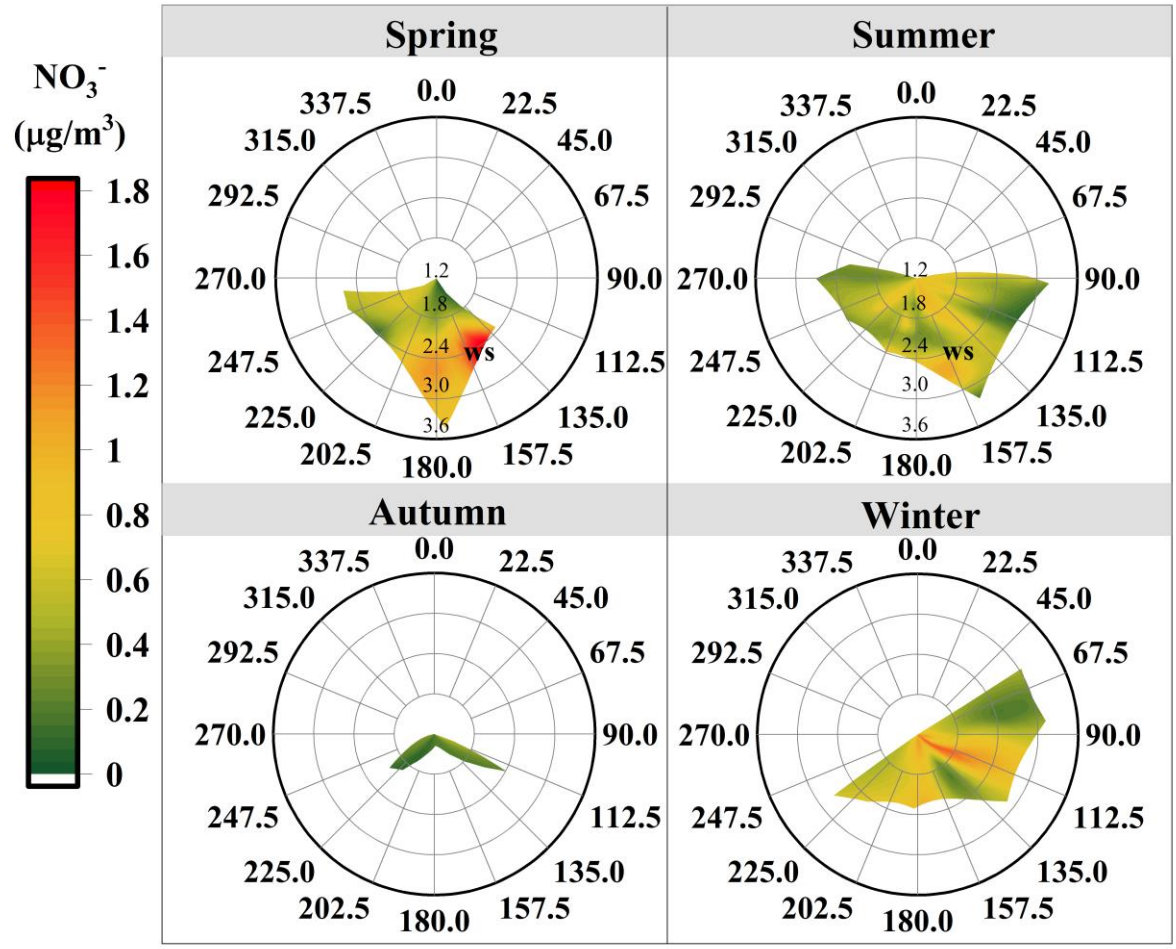


Figure 3. Bivariate polar plot illustrates the seasonal variation in the mass concentration of NO_3^- in relation to wind speed (WS, m/s) and wind direction (WD, degrees).

4. Discussion

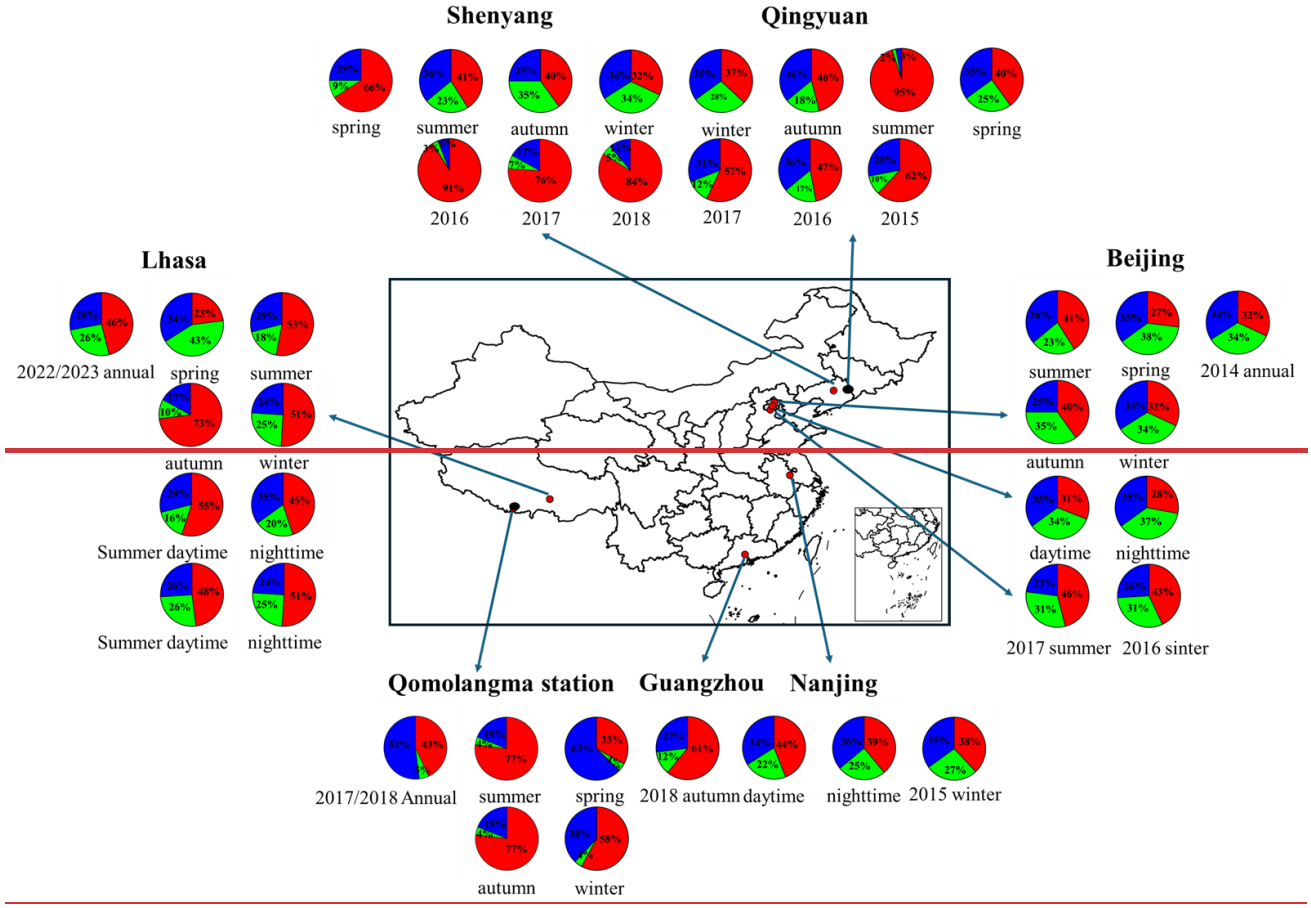
4.1 A comparison of NO_3^- oxidation pathways in Lhasa with other megacities in plain regions

Typically, observations of $\Delta^{17}O$-NO_3^- and estimated α (the proportion of O_3 oxidation in NO_2 production rate) values are employed to quantify the contributions of major NO_3^- oxidation pathway in conjunction with a Bayesian model. The α value ranged from 0.63 to 0.93, with an average of 0.83 ± 0.06, suggesting the significance of O_3 participation in NO oxidation during the sampling campaign. On the other hand, our α values were lower than those (0.85-1) for other midlatitude regions (Alexander et al., 2009). The α values are influenced by the relative amount of O_3, HO_2 and RO_2 in NO*x* cycling. Due to the generally high O_3 concentrations (O_3 > 50 ppb) observed in Lhasa, nearly all α values exceeded 0.8 (Figure S6). To evaluate the impact of key parameters on the estimated contributions of different NO_3^- formation pathways, we conducted a sensitivity analysis by assumed

the α values and $\Delta^{17}O$ value of the terminal oxygen atoms of O_3 ($\Delta^{17}O$-O_3*). As listed in Table S3, the assumption of α and $\Delta^{17}O$-O_3* have an impact on the NO_3^- formation mechanisms. When $\Delta^{17}O$-O_3* was fixed at 39‰, increasing α from 0.7 to 0.9 led to a notable increase in the relative contribution of the NO_2 + OH pathway from 25% to 46%, while that of the NO_3 + VOC pathway decreased from 46% to 25%. The N_2O_5 + H_2O pathway remained nearly constant, with contributions ranging between 28% and 29%, indicating that this pathway is relatively insensitive to changes in α values. Similarly, when α was varied within a reasonable range (0.68-0.93), increasing the $\Delta^{17}O$-O_3* value from 37‰ to 39‰ led to an increase in the NO_2 + OH contribution from 37% to 46%, and a corresponding decrease in the NO_3 + VOC contribution from 35% to 26%. Again, the N_2O_5 + H_2O contribution remained stable at ~ 28%. These results suggest that the estimated contributions of NO_2 + OH and NO_3 + VOC pathways are sensitive to assumptions about α and $\Delta^{17}O$-O_3*, whereas the contribution of the N_2O_5 + H_2O pathway is relatively robust under the tested conditions. Because Lhasa is characterized by relatively high VOC concentrations and $\Delta^{17}O$-O_3* is generally close to 39‰, we consider our parameter assumptions reasonable for further estimating NO_3^- formation pathways for each sample.~~, likely due to stronger UV radiation enhancing OH radical production and competitive oxidation pathways involving VOC and RO_2 in Lhasa~~.

On average, the relative contributions of NO_2 + OH (f_{NO2+OH}), NO_3 + VOC ($f_{NO3+VOC}$) and N_2O_5 + H_2O ($f_{N2O5+H2O}$) to NO_3^- formation in Lhasa during the sampling campaign were 46 ± 26%, 26 ± 19% and 28 ± 11%, respectively. To better understand the characteristics of NO_3^- formation mechanism in Lhasa, we performed a detailed comparison around the China for the relative contributions of key oxidation pathways using the $\Delta^{17}O$ methodology (Figure 4). Overall, similar to most Chinese cities, NO_3^- formation in Lhasa was predominantly driven by the NO_2 + OH pathway, exhibiting distinct seasonal and regional variations. In particular, the average $f_{NO3+VOC}$ values were generally several times higher in spring in Lhasa than in other urban cities. Compared to rural/remote areas, the average $f_{NO3+VOC}$ values showed higher fractions in Lhasa, ~~revelingrevealing~~ the influence of anthropogenic emission, i.e., vehicle exhaust and heating, on NO_3^- formation. In Lhasa, the Capital of Tibet, field measurements among different years showed a substantial increase in VOC concentrations in urban areas of the Tibet Plateau, comparable to those in North China (Tang et al., 2022), revealing the importance of the active NO_3 + VOC pathway for NO_3^- pollution formation in Lhasa. In fact, ~~Recent~~ recent studies have ~~emphasized~~ recognized ~~that~~ NO_3 + VOC ~~is a~~as a major formation mechanism ~~of~~

for NO_3^- production. For instance, Fan et al. (2021) found that $f_{NO3+VOC}$ in Beijing increased from 17% in summer to 32% in winter based on $\Delta^{17}O$-NO_3^- measurements. ~~used $\Delta^{17}O$-NO_3^- to reveal that the average $f_{NO3+VOC}$ value was 17% in summer and increased to 32% in winter based on $\Delta^{17}O$-NO_3^- observations in Beijing~~. He et al. (2018) estimated the relative contributions of NO_3 + VOC and N_2O_5 + Cl^- to NO_3^- formation and found that NO_3 + VOC and N_2O_5 + Cl^- were in the range of 16-56%, underscoring the significant roles of these pathways during haze events in Beijing. ~~Meanwhile~~Similarly, Feng et al. (2023) also reported that the $f_{NO3+VOC}$ values were up to 49.6% in winter in northern China. In Guangzhou,~~Additionally~~, Wang et al. (2023) noted that ~~in Guangzhou,~~ the average $f_{NO3+VOC}$ value was at the 488m (25%) higher than that at the ground (12%). Furthermore, Li et al. (2022) reported that $f_{NO3+VOC}$ increased from 5% in urban to 13.5% in rural regions in Northeast China.~~utilized $\Delta^{17}O$-NO_3^- to explore the oxidation pathway of NO_3^- in both urban and rural atmosphere in Northeast China and found that the $f_{NO3+VOC}$ values increased from 5% in urban areas to 13.5% in rural areas.~~ Although the specific nighttime RO_2 production mechanism in Lhasa remains unclear, studies in other cities have demonstrated that NO_3+VOC pathway was the dominant channel for nighttime RO_2 (Fisher et al., 2016), which in turn leads to the formation of alkyl and multifunctional nitrates ($RONO_2$) and eventually NO_3^-. In such cases, the RO_2 concentration is expected to be correlated with NO_3 radical production, which depends on the reaction rate of O_3 and NO_2 (Brown and Stutz, 2012). Given the relatively high nighttime O_3 concentrations in Lhasa, it is plausible that O_3-driven nighttime NO_3 chemistry plays an important role, thereby enhancing NO_3+VOC derived from RO_2 production and NO_3^- formation. Global modelling studies also support the significant of this pathway. For instance, Alexander et al. (2020) reported that the NO_3 + VOC pathway via the $RONO_2$ mechanism accounted for 3% of global NO_3^- formation on average. The relatively high $f_{NO3+VOC}$ values observed in Lhasa are broadly consistent with these findings, especially under conditions of high VOC concentrations and strong nighttime oxidant levels.


Shenyang
Qingyuan
spring
summer
autumn
winter
winter
autumn
summer
spring
2016
2017
2018
2017
2016
2015
Lhasa
Beijing
2022/2023 annual
spring
summer
summer
spring
2014 annual
autumn
winter
autumn
winter
Summer daytime
nighttime
daytime
nighttime
Summer daytime
nighttime
2017 summer
2016 sinter
Qomolangma station
Guangzhou
Nanjing
2017/2018 Annual
summer
spring
2018 autumn daytime
nighttime
2015 winter
autumn
winter

389

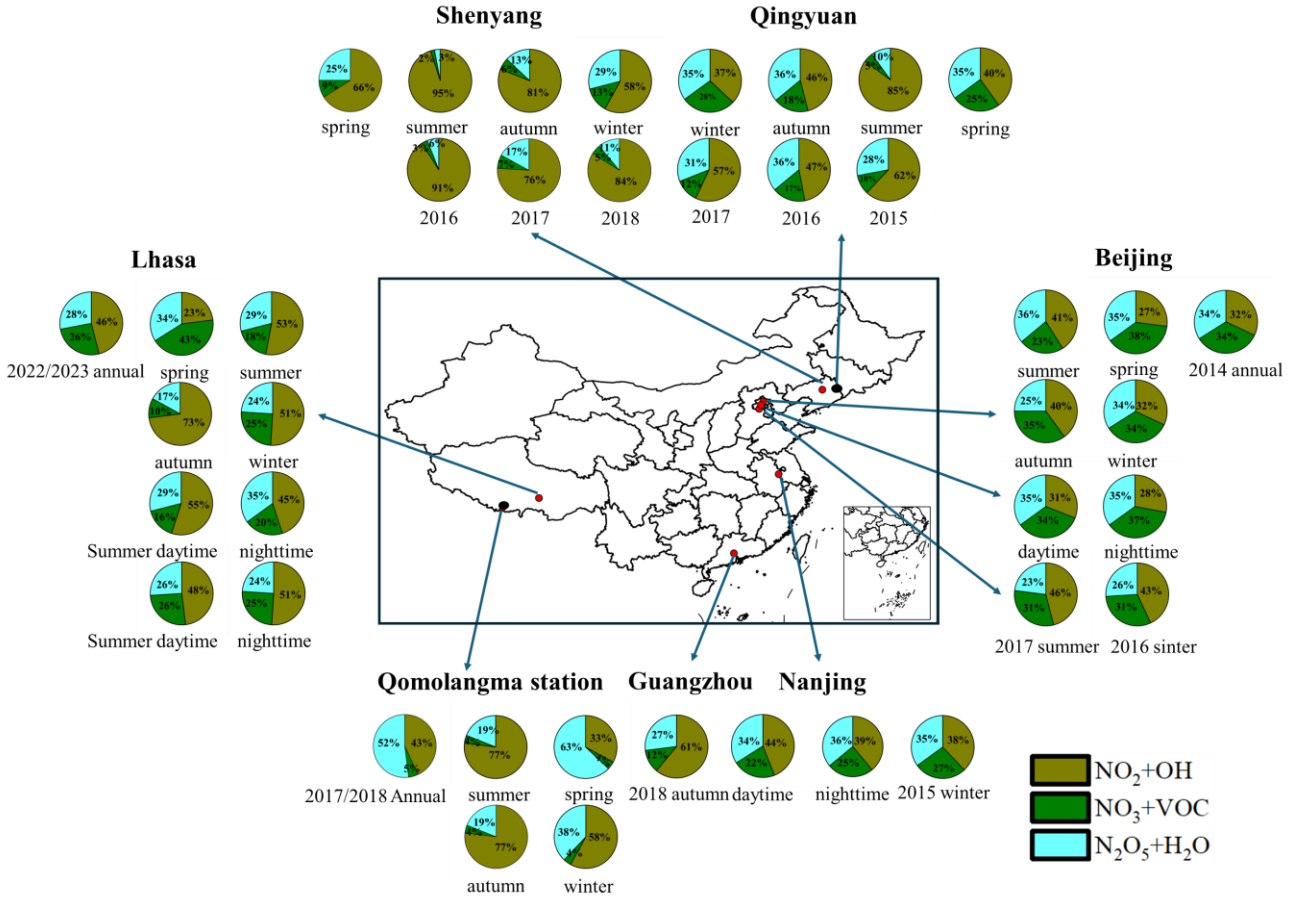


Figure 4. Summary of the relative contributions of key oxidation pathways using the $\Delta^{17}O$ methodology around the China (data given in Table ~~S3~~ S4 in the Supplement). Colors for the study labels indicate the type of sampling location: urban areas (red), and rural/remote areas (black). The pie charts show the relative contribution of different pathways to NO_3^- formation: f_{NO2+OH} (~~red~~ deep yellow), $f_{NO3+VOC}$ deep green~~(green)~~, and $f_{N2O5+H2O}$ (light blue~~blue~~).

4.2 Seasonal and diurnal variations of NO_3^- oxidation pathways

Figure ~~S1~~ S7 illustrates the seasonal variations in the relative contributions of the three main oxidation pathways to NO_3^- formation. When comparing different seasons, the f_{NO2+OH} values were lower ($p < 0.01$) in spring (22.6%) than in winter (50.8%), summer (52.9%) and autumn (73.2%). The dominance of NO_2 + OH pathway in autumn is consistent with observations at Mt. Everest during the autumn seasons of 2017 and 2018, suggesting that NO_3^- formation on the Tibetan Plateau in autumn may be mainly driven by NO_2 + OH pathway (Lin et al., 2021; Wang et al., 2020c). ~~The NO_2 concentration was 12.7 ± 1.81 μg/m^3 in spring, which was higher than in autumn (11.3 ± 5.83 μg/m^3). Therefore, the reduced f_{NO2+OH} in spring cannot be attributed to lower NO_2 levels. Typically, high solar radiation enhances NO_3^- formation via the NO_2 + OH pathway. However, this pathway has been shown~~

~~to be suppressed in clean atmospheric conditions with low particulate matter loadings (Zhang et al., 2022; Fan et al., 2021). The mean NO_3^- concentration was higher in spring (0.83 ± 0.35 μg/m^3) compared to autumn (0.23 ± 0.13 μg/m^3), which might account for the higher f_{NO2+OH} in autumn.~~

A significant increase in the $f_{NO3+VOC}$ values was observed in spring ($p < 0.05$). First, O_3 and NO_2 are precursors of NO_3. In this work, ~~highest~~the highest concentrations of O_3 were found in spring (114.9 ± 18.1 μg/m^3), likely leading to elevated NO_3 concentrations. Additionally, ~~The~~ the low temperature and reduced OH radical concentrations in spring facilitate the reaction of NO_2 and O_3 to synthesize NO_3. This might be an appropriate reason for the $f_{NO3+VOC}$ values in spring. High-altitude locations such as Nepal (5079 m a.s.l.) and Qomolangma Station (4300 m a.s.l.) have experienced stratospheric ozone intrusions, especially in spring and winter, as reported in previous studies (Zhang et al., 2025; Cristofanelli et al., 2010; Morin et al., 2007; Zhang et al., 2022; Lin et al., 2016; Yin et al., 2017; Wang et al., 2020c). Notably, such intrusions in spring may elevate tropospheric O_3 levels in Lhasa, resulting in a mixture of tropospheric and stratospheric O_3 that enhances NO_3^- production. Second, previous study has indicated that the Afghanistan-Pakistan-Tajikistan region, the Indo-Gangetic Plain, and Meghalaya-Myanmar region could transport industrial VOC to various zones in Tibet from west to east. Additionally, agricultural areas in northern India could contribute biomass burning-related VOC to the middle-northern and eastern regions of Tibet (Li et al., 2017). During our sampling campaign, ~~The W and S~~South and Southeast Asia air clusters were notably prevalent in the springtime, coinciding with intensive fire spots observed in Afghanistan, Pakistan, India, Nepal, and Bhutan (Figure ~~S2~~S3/~~S3~~S4). These observations, combined with the prevailing South and Southeast Asia air mass trajectories in spring, strongly suggest that long-range transported VOC from South Asia were delivered to Lhasa and likely participated in local NO_3^- production via NO_3 + VOC pathway.~~Air masses originating from and traversing these regions are likely to pick up VOC emitted from biomass burning and transport them to the sampling site~~ Moreover, recent studies have shown that ambient VOC concentrations in the urban areas on the Qinghai-Tibet Plateau were comparable to those in the North China Plain (Tang et al., 2022). The input of VOC through long-range transport might further elevate VOC concentrations, thereby promoting NO_3^- formation via NO_3 + VOC pathway and contributing to the enhanced $f_{NO3+VOC}$ values observed in spring. While VOC appears to play a dominant role in the process, it should be noted that other nitrogen species (e.g., NO, NO_2) associated with biomass burning emissions may also be transported over long distances and influence NO_3^-

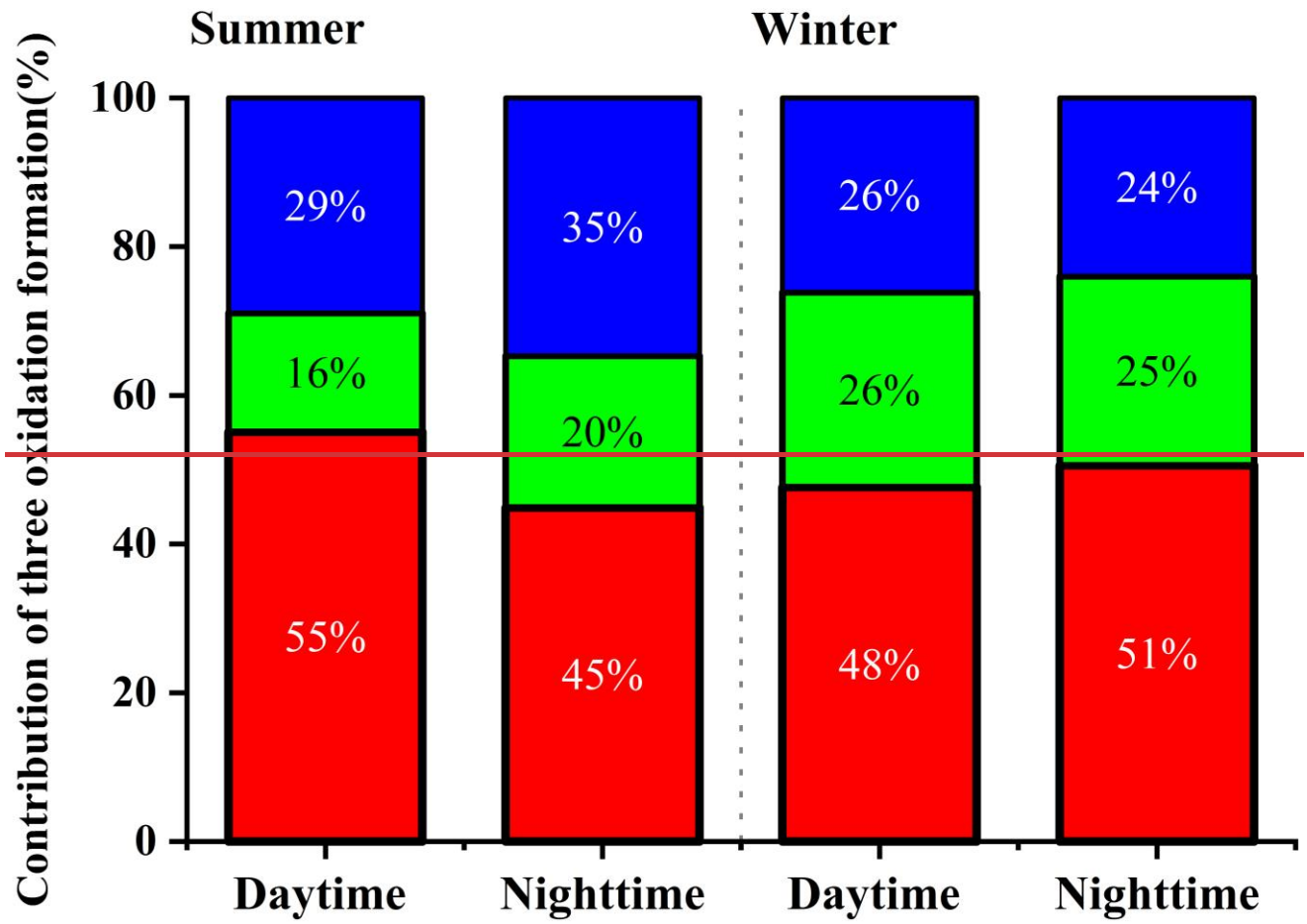
formation in Lhasa. These co-transported nitrogen compounds, although not directly quantified in this study, could further contribute to NO_3^- production in spring. Taken together, these findings provide strong evidence that long-range transport of biomass burning emissions, particularly from South Asia, can substantially influence springtime NO_3^- formation in Lhasa.

Similarly, the $f_{N2O5+H2O}$ values exhibited its highest contributions during spring, with significant seasonal differences ($p < 0.05$) except when compared to summer ($p > 0.05$). Typically, high RH enhances NO_3^- formation via N_2O_5 + H_2O pathway. However, studies have revealed that during sandstorm events, a significant large N_2O_5 uptake coefficient is observed on urban aerosols in spring (Xia et al., 2019). In this study, the mean Ca^{2+} concentration in $PM_{2.5}$ was found to be the highest in spring, suggesting a possible role of dust in facilitating N_2O_5 uptake. Additionally, N_2O_5 + H_2O pathway has been reported to be promoted by elevated NO_3^- concentrations~~,~~ (Lin et al., 2021), which were also highest in spring. Therefore, the increased $f_{N2O5+H2O}$ values during spring might be attributed to the combined effects of lower RH, elevated Ca^{2+} levels, and high NO_3^- concentrations.

Interestingly, distinct diurnal patterns of NO_3^- oxidation pathways were observed during the sampling campaign (Figure 5). In summer, NO_2 + OH pathway showed a significantly higher contribution during the daytime (55.1%) compared to nighttime (44.9%), which is attributed to increased OH radical synthesis during longer days and higher temperatures in Lhasa (Rohrer and Berresheim, 2006). A previous study indicated that lower NO_2 and higher O_3 concentrations enhance the relative contribution of OH pathway to NO_3^- formation (Wang et al., 2019). Additionally, the concentration of ~~aerosol liquid water content~~ALWC (~~ALWC,~~ the detailed information is given in Text S3) was higher at night than during the day in summer, favoring NO_3^- formation through nocturnal formation. In winter, f_{NO2+OH}, $f_{NO3+VOC}$ and $f_{N2O5+H2O}$ were similar during both day and night. Typically, photolytic destruction and chemical reactions with NO are rapid sinks during the daytime, with lifetimes generally less than 5 seconds and resulting in extremely low concentrations. Similarly, the atmospheric lifetime of N_2O_5 under sunlight is also very short (Wang et al., 2018). Thus, daytime NO_3 and N_2O_5 chemistry is often considered negligible. However, a recent study revealed that a non-negligible amount of NO_3 radicals can persist during the daytime in cold months, owing to the limited solar radiation (Hellén et al., 2018). Wang et al. (2020b) found that the daytime production rate of NO_3 can be substantial due to elevated concentrations of O_3 and NO_2, suggesting that the mixing ratios of NO_3 and N_2O_5 during the day may not be negligible.~~daytime NO_3 and N_2O_5 chemistry is considered~~

~~less important due to rapid photolysis of NO_3 and the titration reaction initiated by NO. However, (Wang et al., 2020a). demonstrated that the daytime production rate of NO_3 can be substantial due to increased concentrations of NO_2 and O_3 in winter. Although NO_3 titration occurs rapidly and the atmospheric lifetimes of NO_3 and N_2O_5 are short, their daytime concentrations are not negligible. A recent study found that in winter, the amount of NO_3 radical during the day was comparable to that at night, whereas in summer, daytime NO_3 radicals were lower than those at night (Brown et al., 2011). This discrepancy in NO_3 radical levels might explain the differences in NO_3^- formation pathways between summer and winter.~~ Furthermore, in winter, lower temperatures and elevated NO_2 concentrations facilitate a quasi-steady-state equilibrium between NO_3 and N_2O_5, slowing the overall reactivity of the NO_3^- precursors (Brown et al., 2003). This equilibrium condition minimizes diurnal fluctuations in precursor concentrations, resulting in relatively stable nocturnal and daytime NO_3^- formation pathways, including NO_3 + VOC and N_2O_5 + H_2O. Nevertheless, we acknowledge that the exact role of daytime NO_3/N_2O_5 chemistry remains uncertain in Lhasa and should be further assessed using concurrent filed observations or chemical transport models. Moreover, when interpreting the diurnal differences in $\Delta^{17}O$-NO_3^- values, the atmospheric lifetime of NO_3^- must be considered. Given the atmospheric lifetime of NO_3^- is generally more than 12 hours, each sample might reflect both daytime and nighttime NO_3^- production impacting on $\Delta^{17}O$-NO_3^- values (Park et al., 2004; Vicars et al., 2013).

~~Consequently, under such equilibrium conditions, the NO_2 + OH pathway remains comparatively steady between day and night, as OH generation is limited. Conversely, in summer, elevated temperatures and extended photoperiod prevent NO_3 and N_2O_5 from achieving steady-state equilibrium, leading to greater variability in NO_3^- formation. Thus, NO_3^- production becomes more sensitive to photochemical and VOC fluctuations, resulting in a pronounced contrast between daytime and nighttime NO_3^- formation pathways (Brown et al., 2003).~~



~~*Figure 5. the relative contributions of NO_2+OH, NO_3+VOC, and N_2O_5+H_2O to NO_3^- formation during the day and night in summer and winter in Lhasa during the sampling campaign. The contributions are represented as f_{NO2+OH} (red), $f_{NO3+VOC}$ (green), and $f_{N2O5+H2O}$ (blue).*~~

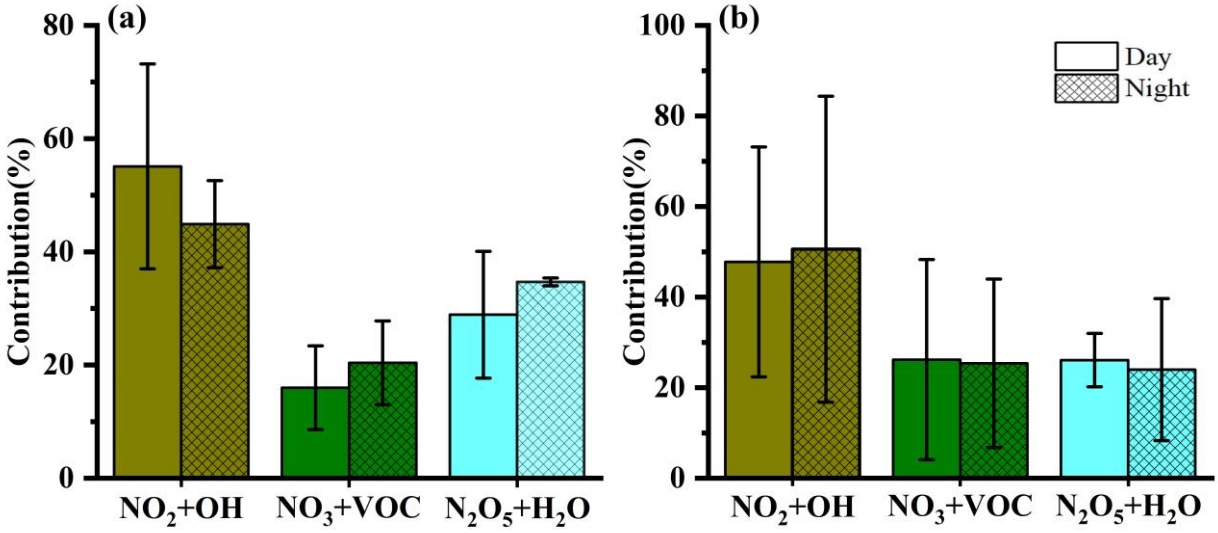


Figure 5. the relative contributions (mean ± SD values) of NO_2 + OH, NO_3 + VOC, and N_2O_5 + H_2O to NO_3^- formation during the day and night (a) in summer and (b) winter in Lhasa during the sampling campaign.

4.3 Integrated analysis of NO_3^- oxidation pathways in Lhasa

As shown in Figure ~~S4~~S8, NO_3 + VOC pathway emerged as the major contributor to NO_3^- formation during periods of high NO_3^- spikes. To elucidate the NO_3^- formation pathways under different NO_3^- concentrations, NO_3^- samples were categorized into different concentration ranges~~To elucidate the mechanisms driving NO_3^- formation under varying conditions, NO_3^- samples were categorized into different concentration ranges~~ (Figure 6). We found the $f_{NO3+VOC}$ values increased and f_{NO2+OH} values decreased with the NO_3^- concentrations. Although recent field radical measurements in urban sites in China found that OH and HO_2 radical during haze period is comparable to clean days (Slater et al., 2020; Yang et al., 2021), our results suggested that NO_3+VOC pathway still played an important role in NO_3^- production under high-NO_3^- concentration in Lhasa, possibly due to enhanced VOC emission. ~~This significant decrease in f_{NO2+OH} values in the air pollution is predictable, as OH radical production tends to decline with increasing air pollution, especially on heavily polluted and hazy days (Bäumer et al., 2008).~~ In addition to concentration effects, meteorological factors typically also regulate the NO_3^- oxidation pathways. Typically, high temperature promotes the NO_3^- formation in f_{NO2+OH} values (Han et al., 2015). However, our study revealed that the relationship between temperature and f_{NO2+OH} values did not consistently show a positive trend. Further analysis indicated that NO_2 and O_3 concentrations were negatively correlated, with lower NO_2 concentrations paired with elevated O_3 levels (Figure ~~S5~~S9). f_{NO2+OH} values reached its minimum when NO_2 was between 15 and 20 μg/m^3 and O_3 was within 100-120 μg/m^3. Although OH radicals exhibit a higher oxidation potential (2.8 V) than O_3 (2.07 V), but atmospheric availability is much lower than that of O_3 (Carslaw et al., 1999; Dubey et al., 1997). Therefore, NO_2 at lower concentrations is more likely to be oxidized by OH than by O_3, even though O_3 concentrations were high. With increasing NO_2 concentrations, the availability of OH radicals for oxidating NO_2 became lower, resulting in a relatively higher proportion of NO_2 being oxidized by O_3 although O_3 concentrations were low. However, when the concentration of O_3 is below 20 μg/m^3, O_3 concentrations were not sufficient to oxidize NO_2 due to the higher NO_2 concentrations and OH radicals for oxidating NO_2 would re-dominate. These observations underscore that in high-altitude urban environments like Lhasa, OH effectiveness is more important on NO_3^- oxidation pathways than that of O_3. Additionally, we identified an intriguing positive correlation between the atmospheric oxidizing capacity (O_x = NO_2 + O_3) and $f_{NO3+VOC}$ values. $f_{NO3+VOC}$ values were lowest when O_x was less than 90 μg/m^3, corresponding to a maximum contribution from the NO_2

+ OH pathway. This suggests that O_x is more indicative of the pathways of NO_3^- formation in the atmosphere compared to either NO_2 or O_3 alone. Typically, High RH and ALWC were also positively correlated with $f_{N2O5+H2O}$. But RH was associated with variable contributions from the N_2O_5+H_2O pathway in our study, while increasing ALWC significantly enhanced this pathway, indicating ALWC as a more reliable indicator of NO_3^- formation.

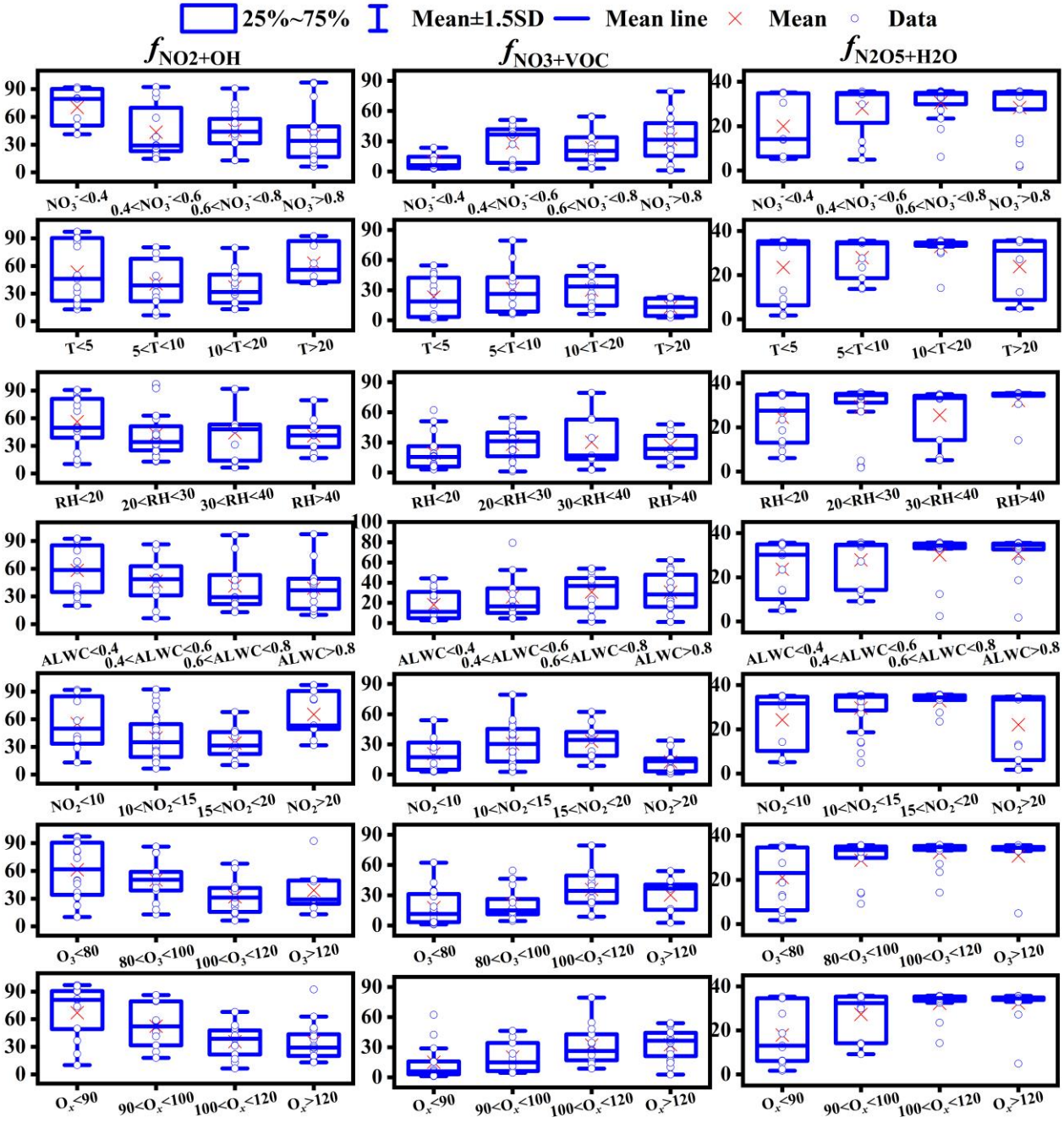


Figure 6 Influence of NO_3^- (μg/m^3), temperature (℃), RH (%), ALWC (μg/m^3), NO_2 (μg/m^3), O_3 and O_x(μg/m^3) on NO_3^- formation pathways (%).

4.4 Implications

The oxidation pathways of NO_3^- in Lhasa, China, were constrained using a full year of $\Delta^{17}O$-NO_3^- measurements from 2022 to 2023. Based on seasonal data, we observed a significant increase in the relative contribution of the NO_3+VOC to NO_3^- formation during ~~the~~ spring. Furthermore, the diurnal distribution of NO_3^- oxidation pathways varied distinctly across seasons. To better understand the factors influencing these pathways, we integrated meteorological conditions, NO*x* precursors, and ALWC for a more comprehensive analysis of NO_3^- formation. The results revealed that O*x* and ALWC are more reliable indicators of NO_3^- oxidation pathways than meteorological factors. Notably, Lhasa's unique high-altitude environment such as strong solar radiation, persistently high O_3, and elevated VOC, promotes active NO_3 + VOC chemistry, especially in spring. Atmospheric ALWC is primarily produced by hygroscopic aerosols such as SO_4^{2-}, NH_4^+, and Cl^-. Therefore, in addition to controlling NO_2, O_3, and VOC, reducing these hygroscopic aerosols is crucial for effective $PM_{2.5}$ pollution control.

Although this study provides valuable insights into NO_3^- formation mechanisms in Lhasa, we must acknowledge the associated uncertainties due to the lack of comprehensive observational constraints in Lhasa. Specifically, the limited understanding of local RO_2 concentrations led us to adopt empirical parameterizations and refer to measurements from other regions, which inevitably introduce uncertainty into the pathway apportionment. In addition, the absence of direct observations of nighttime NO emissions and the NO_2-NO isotope exchange processes in this region further complicates the interpretation of diurnal variations in NO_3^- formation pathways. To improve the robustness of $\Delta^{17}O$-based pathway analysis, future studies should consider synchronous measurements of both NO_2 and NO_3^- isotopes.

Data availability

All data are presented in the main text and/ or the Supplement. For additional data, please contact the corresponding author (liu.junwen@jnu.edu.cn).

Author contributors

JL designed, conceived, and led the research. XZ performed the data analysis and drafted the manuscript. JL, XZ NC and BB planned and carried out the measurements. NC, BB and PD were responsible for measuring the meteorological parameters. JL and PY secured funding for the continuous aerosol sampling and analysis. FC and YZ provided expertise on isotope analysis methods. JL offered guidance on data analysis, and all authors contributed to revising the manuscript.

Competing interests

The authors declare no competing financial interest.

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