Technical note: Quantifying the Nitrogen Isotope Difference between Ammonium in the Atmosphere and Ammonia Emitted from Sources

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Abstract. The difference ($\delta^{15}N_{4a-3s}$) in nitrogen isotopes ($\delta^{15}N$) between NH_4^+ and source - emitted NH_3 is a crucial factor influencing the source apportionment of atmospheric NH_4^+. This $\delta^{15}N_{4a-3s}$ is mainly due to isotopic fractionation during NH_3 - NH_4^+ gas - particle conversion and atmospheric deposition. The impact of isotope fractionation on $\delta^{15}N_{4a-3s}$ had been well quantified by simplified method, but that of atmospheric deposition had often been overlooked. This study developed a model to assess $\delta^{15}N_{4a-3s}$ variations by considering both the atmospheric deposition and isotope fractionation. The results of six model scenarios showed the difference between $\delta^{15}N_{4a-3s}$ values under both influences and under isotope fractionation alone increased with the rise of ξ_A (the molar fraction of NH_4^+ to NH_x in the atmosphere). At 20°C, when $\xi_A = 0.9$, the maximum gap could reach 10.7‰. $\delta^{15}N_{4a-3s}$ was insensitive to NH_3 and NH_4^+ deposition rates, NH_4^+ generation rate, and temperature, but it was sensitive to ξ_A. A prediction function for $\delta^{15}N_{4a-3s}$ was constructed and applied for atmospheric NH_4^+ source apportionment in the Yellow River Delta. Compared with the simplified method, the fitted equation could more reasonably reflect the contribution of agricultural sources (e.g., fertilizer application). The constructed equation could be used for tracing atmospheric NH_4^+ origin, thus improving the accuracy of atmospheric NH_4^+ source apportionment.

1. Introduction

Ammonia (NH_3) is the primary alkaline gas in the atmosphere, and its cycle plays a crucial role in the geological and biological nitrogen cycles on Earth's surface. NH_3 can neutralize atmospheric acids such as sulfuric acid (H_2SO_4), nitric acid (HNO_3), and hydrochloric acid (HCl), forming particulate ammonium (NH_4^+) aerosols. This process deteriorates air quality and affects the acidity of airborne particulate matter, precipitation, and cloud water (Ianniello et al., 2010;Pan et al., 2016;Chang et al., 2019a). NH_3 and NH_4^+ (collectively abbreviated as NH_x) are deposited back to the surface through dry and wet deposition processes.Excess NH_x in the atmosphere not only leads to eutrophication of terrestrial and aquatic ecosystems and threatens biodiversity, but also increases public health risks (Bobbink et al., 2010;Liu et al., 2019;Tan et al., 2020;Bouwman et al., 2002). In recent decades, rapid industrialization,

urbanization, and agricultural development have significantly increased NH_3 emission worldwide, particularly in Asia, Africa, and South America. This resulted in a substantial rise in severe haze events

45   and eutrophication risks in these regions (Zhao et al., 2017;Bouwman et al., 2002;Kim et al., 2011). In order to develop effective strategies to reduce NH_x levels in the atmosphere, source apportionment of NH_x has garnered considerable attention as a foundational research focus in recent years (Behera et al., 2013;Shen et al., 2011;Hu et al., 2014;Schiferl et al., 2016;Sun et al., 2016).

Numerous methods has been developed and used to trace sources of NH_x in the atmosphere. Among

50   these, the utilization of the stable nitrogen isotopic composition of gaseous NH_3 ($\delta^{15}N$-NH_3) or aerosol NH_4^+ ($\delta^{15}N$-NH_4^+) has emerged as a highly promising tool for source apportionment of NH_3 in the atmosphere (Gu et al., 2025;Elliott et al., 2019). However, it is important to note that neither aerosol $\delta^{15}N$-NH_4^+ nor gaseous $\delta^{15}N$-NH_3 can be directly used for this purpose alone. This is because their isotopic values do not correspond directly to the $\delta^{15}N$-NH_3 of mixed NH_3 emissions from various sources

55   (Zhang et al., 2020). The discrepancy arises from nitrogen isotope exchange between NH_3 and NH_4^+ in the atmosphere (Walters et al., 2019;Kawashima and Ono, 2019). Observations typically show higher $\delta^{15}N$ values in NH_4^+ compared to NH_3, which can be attributed to equilibrium isotopic fractionation (Walters et al., 2019). The mass and isotope balance for NH_x in the atmosphere is given by the following equation:

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$$\xi_A \delta^{15}N\text{-}NH_4^+ + \left(1-\xi_A\right)\delta^{15}N\text{-}NH_3 = \delta^{15}N\text{-}NH_x \text{ ,} \tag{1}$$

where ζ_A is the molar fraction of NH_4^+ to NH_x in the atmosphere, and $\delta^{15}N$-NH_4^+, $\delta^{15}N$-NH_3 and $\delta^{15}N$-NH_x are the isotopic composition of NH_4^+, NH_3 and NH_x in the atmosphere, respectively. After the introduction of the isotopic fractionation factor (α) between NH_4^+ and NH_3, the Eq. (1) can be rewritten as:

65
$$\delta^{15}N\text{-}NH_4^+ = \frac{\alpha}{1+\left(\alpha-1\right)\xi_A}\delta^{15}N\text{-}NH_x + \frac{1000\left(\alpha-1\right)\left(1-\xi_A\right)}{1+\left(\alpha-1\right)\xi_A} \text{ ,} \tag{2}$$

Because the coefficients of $\dfrac{\alpha}{1+\left(\alpha-1\right)\xi_A}$ and $1+\left(\alpha-1\right)\xi_A$ are approximately equal to one, the Eq. (1) is often simplified as:

$$\delta^{15}N_{4a\text{-}3x} = \delta^{15}N\text{-}NH_4^+ - \delta^{15}N\text{-}NH_x = 1000\left(\alpha-1\right)\left(1-\xi_A\right) \text{ ,} \tag{3}$$

where $\delta^{15}N_{4a\text{-}3x}$ is difference between $\delta^{15}N$-NH_4^+ and $\delta^{15}N$-NH_x in the atmosphere. The difference is often

70   used to correct the $\delta^{15}N$ of NH_4^+ in the atmosphere, so as to apportion sources of NH_4^+ (Pan et al., 2018;Pan et al., 2016;Chang et al., 2016). The underlying assumption is that the atmosphere is a well-mixed closed system, implying that $\delta^{15}N$-NH_x values in the atmosphere are equivalent to those of the $\delta^{15}N$-NH_3 emitted from various sources. However, it is a well-established fact that both NH_3 and NH_4^+ can exit the atmosphere through deposition processes. These deposition processes may introduce

75   substantial discrepancies between the $\delta^{15}N$-NH_x values observed in the atmosphere and the $\delta^{15}N$-NH_3

values emitted from sources. This bias could further compromise the accuracy of δ^{15}N-based source apportionment (Zhang et al., 2020).

To more accurately identify the source of atmospheric NH$_x$, this study develops a model to quantify the difference in δ^{15}N-NH$_4^+$ in the atmosphere and source – emitted δ^{15}N-NH$_3$ (δ^{15}N$_{4a-3s}$) under combined

80   atmospheric deposition and isotope fractionation. It should be noted that the model primarily addresses the isotopic mass balance shift induced by deposition processes, rather than detailed chemical mechanisms. The objectives of this study are (1) to understand the variation pattern of δ^{15}N$_{4a-3s}$ and the key influencing factors, (2) to construct a fitting equation for the δ^{15}N$_{4a-3s}$ and the key influencing factors used for NH$_4^+$ source apportionment, and (3) to evaluate the application effects of the fitted equation in

85   the source apportionment of NH$_4^+$ through practical examples.

2. Methods and Theory

2.1 Model Development

Similar to the Eq. (3), the δ^{15}N$_{4a-3s}$ could be calculated as follows:

$$\delta^{15}\text{N}_{4a-3s} = \left(\delta^{15}\text{N-NH}_4^+ - \delta^{15}\text{N-NH}_{3s}\right) + 1000\left(\alpha - 1\right)\left(1 - \xi_A\right) \quad , \tag{4}$$

90   where δ^{15}N-NH$_{3s}$ is the nitrogen isotope of NH$_3$ from various types of sources and other items are the same as that in Eq. (1) and (2). The equation indicates that the variation in δ^{15}N$_{4a-3s}$ depends on the change of α, ξ_A and δ^{15}N-NH$_x$. The parameter α is defined as (δ^{15}N-NH$_4^+$+1000)/ (δ^{15}N-NH$_3$+1000) (Coplen, 2011). This value of α could be empirically determined by fitting experimental data (Urey, 1947;Li et al., 2012) and computational quantum chemistry method (Walters et al., 2019) based on the ambient

95   temperature. The present study used the computational quantum chemistry method (Walters et al., 2019) to calculate the α values as the following method:

$$1000\left(\alpha - 1\right) = \frac{12522}{\text{T}\left(\text{K}\right)} - 11.31 \quad , \tag{5}$$

where T is ambient temperature (in Kelvin). The values of ξ_A and δ^{15}N-NH$_x$ vary synchronously over time. When considering the atmosphere as a well-mixed open system, the continuous variations in the

100   mass fractions of NH$_3$ and NH$_4^+$ within the atmosphere, along with their deposition process, can be discretized as follows:

$$\left[\text{NH}_{3a}\right]^t = \left(1 - G_4 - D_3\right)\left[\text{NH}_{3a}\right]^{t-1}$$
$$\left[\text{NH}_{4a}^+\right]^t = G_4\left[\text{NH}_{3a}\right]^{t-1} + \left(1 - D_4\right)\left[\text{NH}_{4a}^+\right]^{t-1}$$
$$\left[\text{NH}_{3d}\right]^t = D_3\left[\text{NH}_{3a}\right]^{t-1}$$
$$\left[\text{NH}_{4d}^+\right]^t = D_4\left[\text{NH}_{4a}^+\right]^{t-1} \quad , \tag{6}$$

where G_4, D_3, and D_4 represent the transformation rate of NH$_4^+$, and the deposition rate of NH$_3$ and the deposition rate of NH$_4^+$. The superscript of t indicates the iteration times. [NH$_{3a}$], [NH$_{3d}$], [NH$_{4a}^+$], and

[NH_{4d}^+] are mass fractions of NH_3 and NH_4^+ in the atmosphere and deposition, respectively. The value of ξ_A at the tth time interval can be expressed as:

$$\xi_A^t = \frac{\left[NH_{4a}^+\right]^t}{\left[NH_x\right]^t} = \frac{\left[NH_{4a}^+\right]^t}{\left[NH_{4a}^+\right]^t + \left[NH_{3a}\right]^t} \quad , \tag{7}$$

where [NH_x]^t is the mass fraction of NH_x in the atmosphere at the tth time interval, and the meanings of other items are the same as those in Eq. (6). Correspondingly, [$\delta^{15}N$-NH_x] in the atmosphere at the tth time interval can be written as:

$$\left[\delta^{15}N\text{-}NH_x\right]^t = \xi_A^t \left[\delta^{15}N\text{-}NH_{4a}^+\right]^t + \left(1-\xi_A^t\right)\left[\delta^{15}N\text{-}NH_{3a}\right]^t \quad , \tag{8}$$

where ξ_A^t is the value calculated by Eq. (7), and [$\delta^{15}N$-NH_{4a}^+]^t, [$\delta^{15}N$-NH_{3a}]^t and [$\delta^{15}N$-NH_x]^t are the $\delta^{15}N$ values of NH_4^+, NH_3 and NH_x in the atmosphere at the tth time interval, respectively. The value of [$\delta^{15}N$-NH_{4a}^+] and [$\delta^{15}N$-NH_{3a}] at the tth time interval can be expressed as:

$$\left[\delta^{15}N\text{-}NH_{4a}\right]^t = \left[\delta^{15}N\text{-}NH_x\right]^{t-1} + 1000(\alpha-1)\left(1-\xi_A^{t-1}\right)$$
$$\left[\delta^{15}N\text{-}NH_{3a}\right]^t = \left[\delta^{15}N\text{-}NH_x\right]^{t-1} - 1000(\alpha-1)\xi_A^{t-1} \quad , \tag{9}$$

It can be assumed that $\delta^{15}N$ values of NH_3 and NH_4^+ ([$\delta^{15}N$-NH_{4d}^+] and [$\delta^{15}N$-NH_{3d}]) deposited to the surface are equal to $\delta^{15}N$-NH_3 and $\delta^{15}N$-NH_4^+ in the atmosphere at the tth time interval, respectively, as the following:

$$\left[\delta^{15}N\text{-}NH_{4d}\right]^t = \left[\delta^{15}N\text{-}NH_{4a}\right]^t$$
$$\left[\delta^{15}N\text{-}NH_{3d}\right]^t = \left[\delta^{15}N\text{-}NH_{3a}\right]^t \quad , \tag{10}$$

From Eq. (9), it can be noted that [$\delta^{15}N$-NH_x] equals [$\delta^{15}N$-NH_{3s}] when t = 1, which is the starting point for the iteration of this model. The following iterative procedure is used to obtain the value of $\delta^{15}N_{4a\text{-}3s}$:

(1) Set up [$\delta^{15}N$-NH_{3s}];

(2) Set up G_4, D_3, and D_4 to calculate [NH_{3a}], [NH_{3d}], [NH_{4a}^+] and [NH_{4d}^+] from Eq. (6);

(3) Calculate ξ_A from Eq. (7);

(4) Input T to calculate α from Eq. (5);

(5) Calculate [$\delta^{15}N$-NH_{4a}^+], [$\delta^{15}N$-NH_{3a}], [$\delta^{15}N$-NH_{4d}^+] and [$\delta^{15}N$-NH_{3d}] from Eq. (9) and (10);

(6) Calculate [$\delta^{15}N$-NH_x] from Eq. (8);

(7) Calculate $\delta^{15}N_{4a\text{-}3s}$ from Eq. (4).

The model was developed by R 4.1.3 software.

2.2 Parameter Identification

As described in the previous section, the value of $\delta^{15}N_{4a\text{-}3s}$ could be predicted by the developed model if the values of G_4, D_3, D_4, T, and [$\delta^{15}N$-NH_{3s}] were available. The G_4, D_3, and D_4 had strong spatiotemporal variability, primarily driven by the difference in the content and composition of acidic gases in the

atmosphere, as well as the variation of surface roughness, wind speed, and etc (Baek and Aneja,
2004;Schrader and Brummer, 2014). This is a systematic project that cannot be validated experimentally
by an article. So, here we constructed six simulation scenarios to assess the extent to which changes in
the three parameters affected the simulation results of $\delta^{15}N_{4a\text{-}3s}$ as listed in Table 1.

These six simulation scenarios were composed of combining G_4 being equal to 0.5 times, 1.0 times and
2.0 times D_3, as well as D_4 being 0.2 times and 0.5 times D_3 respectively. The detailed basis for this
proportion setting could be found in the Text S1 and Tables S1-S2 in the Supporting Information (SI).
Briefly, the G_4 is mainly attributed to the comprehensive neutralization reaction rate of ammonia (NH$_3$)
with acid gases (sulfuric acid (H$_2$SO$_4$), nitric acid (HNO$_3$) and hydrochloric acid (HCl)) in the atmosphere
(Baek and Aneja, 2004). Under normal atmospheric conditions, the relatively abundant acidic gas in the
atmosphere could make the G_4 faster, and the G_4 between NH$_3$ and H$_2$SO$_4$ was faster than that of NH$_3$
with HNO$_3$ and HCl, as listed in Text S1 of SI (Schrader and Brummer, 2014). The atmospheric G_4/D_3
ratio was estimated to be 3% based on the mean and median values, indicating that D_3 generally exceeds
G_4 in the atmosphere. Based on this relationship, we established three G_4 levels: $0.5\times D_3$ (low acid-gas
content), $1.0\times D_3$ (moderate acid-gas content), and $2.0\times D_3$ (high acid-gas content). Surface roughness is
an important parameter affecting the rate of atmospheric deposition of NH$_3$ and NH$_4^+$, with a general
pattern of decreasing deposition rate from mountainous regions to flat terrain areas (Zhang et al., 2001).
Numerous studies had also shown that the D_4 was significantly lower than D_3 in the atmosphere(Shen et
al., 2009). Therefore, we set two levels for D_4, one was $0.2\times D_3$ and the other was $0.5\times D_3$, representing
the deposition differences in flat regions and mountainous areas, respectively.

Table 1. Parameter settings, characteristics and representative regions of the six model scenarios.

	G_4	D_4	Characteristics and representative regions
MS1	$0.5D_3$	$0.2D_3$	less acidic gas and flat surfaces, open oceans.
MS2	$0.5D_3$	$0.5D_3$	less acidic gas and rough surfaces, mountainous forests.
MS3	$1.0D_3$	$0.2D_3$	moderate acidic gas and flat surfaces, plain background regions.
MS4	$1.0D_3$	$0.5D_3$	moderate acidic gas and rough terrains, mountainous background regions.
MS5	$2.0D_3$	$0.2D_3$	higher acidic gases and flat terrains, plain cities.
MS6	$2.0D_3$	$0.5D_3$	higher acidic gas and rough terrains, mountainous cities.

A total of five temperature levels of -10ºC, 0ºC, 10ºC, 20ºC, 30ºC were set for the above six simulation
scenarios. In addition, the [δ^{15}N-NH$_{3s}$] was set to 0 ‰ in these simulations.

2.3 Sensitivity Analysis

Sensitivity analysis is a statistical technique used to quantify the extent to which variations in different
inputs influence the variability of the outputs. In present study, sensitivity analysis was conducted to
examine the impact of individual parameters on the $\delta^{15}N_{4a\text{-}3s}$ values. This analysis was achieved by
assessing how sensitive the model was to alterations in its input parameters. To do this, the model was
run with each parameter individually scaled to 0.9 and 1.1 times its original value(Cao et al., 2007). The

evaluation involved computing sensitivity coefficients (*SC*), which quantify the relative changes in the primary output estimates in response to changes in the input parameters, as outlined below:

$$SC = abs\left(\frac{OUT_{1.1} - OUT_{0.9}}{0.2 \times OUT_{1.0}}\right), \tag{11}$$

where $OUT_{1.1}$, $OUT_{1.0}$, and $OUT_{0.9}$ are the model output results when the input parameter is 1.1, 1.0, and 0.9 times of its original value, respectively. The sensitivity of various parameters can be directly compared because the *SC* values obtained through Eq. (11) are dimensionless. The absolute magnitude of *SC* represents the extent to which input parameters affect the output results. Furthermore, the positive or negative sign of *SC* reveals the direction of this influence: positive values signify that an increase in input parameters leads to an increase in output results, whereas negative values imply the opposite relationship(Cao et al., 2007;Dong et al., 2010).

3. Results and Discussion

3.1 The Variation in $\delta^{15}N_{4-3}$ with ξ_A

Figure 1 shows the variation of $\delta^{15}N_{4a-3s}$ with respect to the ξ_A value at temperatures of -10°C, 0°C, 10°C, 20°C, and 30°C, respectively, by the six model scenarios. In order to aid in comparisons, Fig.1 also displays that the change in $\delta^{15}N_{4a-3x}$ obtained using Eq. (4), which can be thought of as the $\delta^{15}N_{4a-3s}$ value calculated using the simplified method. Under different temperature conditions, the six simulation scenarios all obtained similar change characteristics, that is, the $\delta^{15}N_{4a-3s}$ value decreased with the increase of ξ_A value.

The following took the results at 20°C as an example to further illustrate. Specifically, when ξ_A was 0.1, the $\delta^{15}N_{4a-3s}$ values across the six simulated scenarios average around 28.8‰. Conversely, when ξ_A reached 0.9, the $\delta^{15}N_{4a-3s}$ values decreased to a range of 5.89-14.0‰. Among the six model scenarios, the variation in $\delta^{15}N_{4a-3s}$ values was narrower in regions with lower generation rates (G_4) and deposition rates (D_4) of NH_4^+(such as MS1 and MS2), whereas it was broader values in regions with higher rates (such as MS5 and MS6). The finding suggested that there was a minor fluctuation in $\delta^{15}N_{4a-3s}$ values in areas with less acidic gas and flat landscapes, like flat land surfaces and open oceans, whereas a more significant variation was observed in $\delta^{15}N_{4a-3s}$ values in regions characterized by higher acidic gas levels and rugged terrains, such as mountainous cities with high air pollution load.

The change in $\delta^{15}N_{4a-3s}$ values calculated by the simplified method was larger than those $\delta^{15}N_{4a-3s}$ values yielded by all six model scenarios, and their difference was a gradual increase as ξ_A values increase. The difference indicated that when the simple method was used for source apportionment of NH_4^+ in the atmosphere, the deviation would be larger with the increase of ξ_A value. This phenomenon originated from NH_3 exhibiting a higher deposition rate compared to NH_4^+(Behera and Sharma, 2011, 2012), which results in a greater proportion of NH_x components with lower $\delta^{15}N$ values being removed from the atmosphere. Consequently, the $\delta^{15}N-NH_x$ in the atmosphere gradually increased. When using the $\delta^{15}N_{4a-3s}$ values calculated by the simplified method to correct the $\delta^{15}N$ of NH_4^+ for the purpose of apportioning sources of NH_4^+, overcorrection may occur, leading to an overestimation of the contribution proportion

of NH_3 emission sources with relatively negative $\delta^{15}N$ values, like non-agricultural sources (e.g., vehicle exhaust and NH_3 slip) (Zong et al., 2023;Feng et al., 2023). The larger $\delta^{15}N_{4a-3s}$ value was in the case of lower ambient temperature, indicating that the difference between $\delta^{15}N_{4a-3s}$ calculated by the model scenarios and the simplified method had more obvious influence on the source apportionment of NH_4^+

205 in the northern cold region and the winter period when the temperature was lower(Sun et al., 2021).
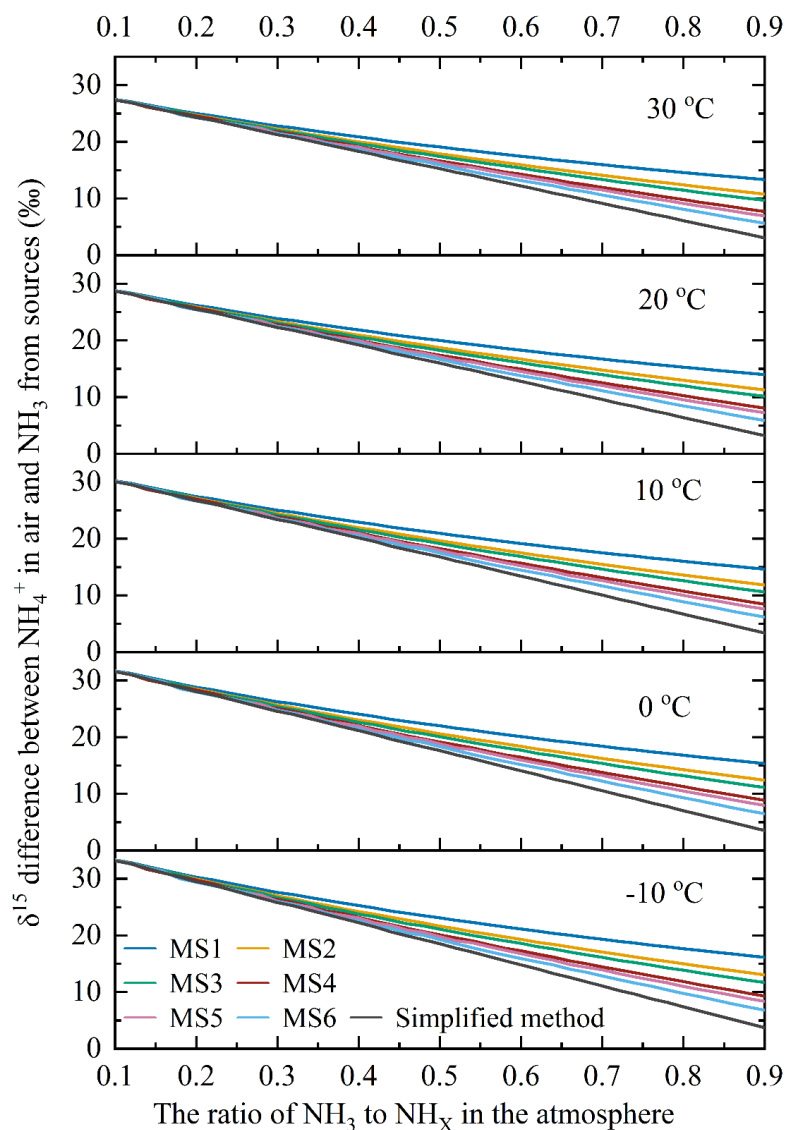
Figure 1. The variation in δ15N4a-3s with ξA at -10°C, 0°C, 10°C, 20oC, and 30°C simulated by the six model scenarios and the simplified method. (note: δ15N4a-3s is the difference between δ15N-NH4+ in the atmosphere and δ15N-NH3 emitted from sources, MS1-MS6 are the six model scenarios as listed in Table 1,

210 **the simplified method is showed in Eq. (4)).**

3.2 Sensitivity of $\delta^{15}N_{4a-3s}$ To Input Parameters

The sensitivity coefficients of each input parameters (D_3, D_4, G_4, and T) to the $\delta^{15}N_{4a-3s}$ values at 20°C were calculated by using Eq. (11) across the six model scenarios (see Fig. 2). The sensitivity coefficient of D_3 variation on the simulation results ranged between 2.55×10^{-5} and 9.32×10^{-2}. The sensitivity

215 coefficient of D_4 and G_4 variations on the $\delta^{15}N_{4a-3s}$ ranged respectively from 1.77×10^{-5} to 0.474 and from 3.62×10^{-4} to 0.494, showing a similar variation signature, which intensified as the value of ξ_A increases. Additionally, the range of variation in D_4's sensitivity coefficient also widened as the value of ξ_A increased. However, G_4 did not exhibit this characteristic. The variation range of the sensitivity coefficient of temperature (T) variation on the simulation results was narrow, falling between 0.092 and

220 0.094, and this influence diminished as the value of ξ_A increased. Generally, the sensitivity coefficient exceeding 0.5 is considered that the corresponding input parameter has a sensitive influence on the output result (Cao et al., 2007). It suggested that the influence of each individual input parameter on the simulation output result ($\delta^{15}N_{4a-3s}$) did not show obvious sensitivity in the six model scenarios.

In fact, the degree to which input parameters affect the simulation results is not only related to the

225 sensitivity coefficient evaluated based on a single input parameter, but also to the synergistic effect of changes in all input parameters (Wang et al., 2023). In this model simulation, the changes in the $\delta^{15}N_{4a-3s}$ and ξ_A were the results of the input parameter changes. Taking ξ_A as a comprehensive input parameter of D_3, D_4, G_4, and T, the sensitivity of ξ_A to output results based on the six simulation scenarios was calculated by formula (11). The variation range of the calculated corresponding sensitivity coefficient

230 with ξ_A is also shown in Fig. 2. The variation characteristic of this sensitivity coefficient is similar to that of the input parameter D_4, showing that the variation range becomes larger and larger with the increase of ξ_A. When the ξ_A value was greater than 0.32, the maximum of the sensitivity coefficients obtained by the six simulation scenarios began to be greater than 0.5, that is, the changes in the input parameters (ξ_A) were generally considered to have sensitive effects on the output results ($\delta^{15}N_{4a-3s}$) (Cao et al., 2007).

235 The maximum range of variation of this sensitivity coefficient was from 0.831 to 3.90 when the ξ_A value reached 0.9.
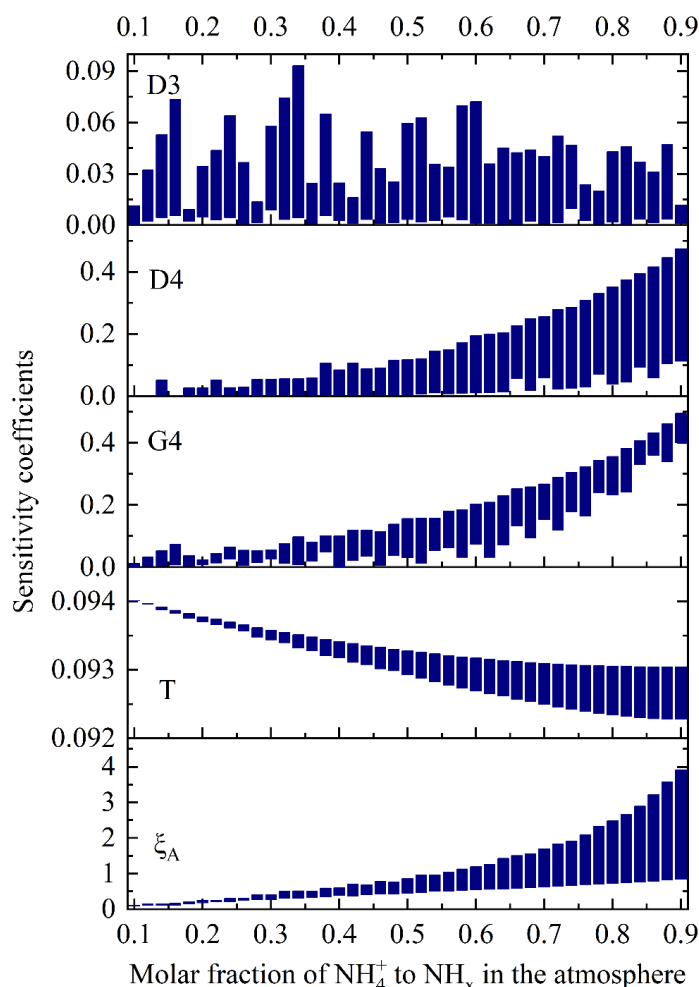
Figure 2. The variation range of sensitivity coefficients of input parameters to the $\delta^{15}N_{4a-3s}$ values at 20ºC obtained from six simulation scenarios.

240   These ranges of sensitivity coefficients were generated by six simulation scenarios. To better understand them, the sensitivity coefficients of ξ_A to output results ($\delta^{15}N4_{a-3s}$) derived from the six simulation scenarios are shown in Fig. S1 of SI. The simulation results of MS1 scenario had the lowest sensitivity to ξ_A, while the simulation results of MS6 scenario had the highest sensitivity to ξ_A, indicating that large D_4 and G_4 had a greater impact on the simulation results. The finding also indicated that in some urban

245   areas with more acidic gases, especially in mountainous areas, more attention should be paid to the influence of ξ_A changes on the source apportionment of NH_4^+ in the atmosphere.

3.3 Construction of the prediction function for $\delta^{15}N_{4a-3s}$

As mentioned earlier, $\delta^{15}N_{4a-3s}$ is generally calculated using Eq. (3) when apportioning sources of NH_4^+ in the atmosphere. The independent variable of this equation included the key influencing parameter ξ_A.

250 Based on the form of Eq. (3) and the calculation results of the $\delta^{15}N_{4a\text{-}3s}$ values, we constructed a calculation function for the $\delta^{15}N_{4a\text{-}3s}$ values through nonlinear regression. The form of this nonlinear function is shown in Eq. (12). The coefficient values of B, C, and D in the equation were iteratively fitted for six simulation scenarios. During the fitting process, each simulation scenario took into account different temperature conditions, including -10℃, 0℃, 10℃, 20℃, and 30℃.

255 $$\delta^{15}N_{4a\text{-}3s} = 1000\left(\alpha^{B}-1\right)\left(1-\xi_{A}^{C}\right)+D \quad ,\tag{12}$$

Table 2 lists the iterative fitting values of the coefficients B, C, and D in Eq. (12) for the six model scenarios (MS1−MS6), as well as the determination coefficient (r^2) and sum of squares error (SSE), which were often used to evaluate the fitting effect (Sun et al., 2023;Xu, 2017). Figure S2 and Figure S3 of SI show the comparison plots and scatter plots of the $\delta^{15}N_{4a\text{-}3s}$ values calculated by the six fitted

260 equations against the six model scenarios (MS) simulated by the developed model. The coefficients B and C are both greater than 0 and less than 1, and the two coefficients gradually increase from MS1 to MS6, indicating that the Eq. (12) for model scenarios from MS1 to MS6 approaches from nonlinear to linear. The determination coefficients obtained from the regression of the six fitting equations with the simulation results range from 0.967 to 0.998, indicating that these equations can fit the calculated $\delta^{15}N_{4a\text{-}}$

265 $_{3s}$ values quite well (see Fig. S3 of SI). As shown in Fig. S2 of SI, the maximum deviation between the $\delta^{15}N_{4a\text{-}3s}$ values obtained from the fitting equation and those calculated by the model occurs in the model scenario of MS1. Under -10ºC condition, the deviation reached its maximum value (1.82‰) when the ξ_{A} value was 0.9. This largest deviation was also significantly smaller than the variation range of $\delta^{15}N$ values of NH_3, which emitted from various types of sources used in the source apportionment of atmospheric

270 NH_4^+(Gu et al., 2022;Zhang et al., 2023;Feng et al., 2022;Li et al., 2023). This indicated that these fitting equations would not significantly increase the uncertainty of the source-resolved assessment of NH_4^+ in the atmosphere when they were used.

3.4 Comparison of Source Apportionment of Atmospheric NH_4^+ in the Yellow River Delta

In order to evaluate the fitting equations as mentioned above using monitoring data, we conducted a

275 source apportionment simulation of NH_4^+ again using the atmospheric $\delta^{15}N\text{-}NH_4^+$ in the Yellow River Delta in the summers of 2013 and 2021. Apart from the difference in the method of calculating $\delta^{15}N_{4a\text{-}3s}$ values, the model, input data, etc. were consistent with those reported in our previous study(Zong et al., 2023). Briefly, the Bayesian mixing model (MixSIAR) developed by the R language was used for the source apportionment of NH_4^+. Four main types of NH_3 emission sources, including fertilizer use (−25.21

280 ± 9.43 ‰), livestock waste (−16.14 ± 7.98 ‰), vehicle exhaust (+6.62 ± 1.89 ‰), and NH_3 slip (−7.12 ± 7.62 ‰), were considered in the model. In the previous study, we used equation (3) (termed as simplified method) to calculate the $\delta^{15}N_{4a\text{-}3s}$ values. In this simulation, we used the MS1 fitting equation to calculate the $\delta^{15}N_{4a\text{-}3s}$ values. This is because these atmospheric particulate matter samples were collected at the Yellow River Delta Ecological Research Station of Coastal Wetland, Chinese Academy

285 of Sciences (37◦45′N, 118◦59′E). The Yellow River Delta is an alluvial plain formed by the Yellow River, featuring a very flat terrain(Li et al., 2022;Zhang et al., 2025). Additionally, there are no obvious emission sources from industrial, transportation, and agricultural activities around the sampling site. Many studies

regarded this sampling site as an atmospheric background point in North China(Zong et al., 2015;Sui et al., 2015), which met the characteristics of low pollution and a flat terrain in the MS1 scenario. Moreover,

290   the deviation between the $\delta^{15}N_{4a\text{-}3s}$ values calculated by the MS1 method and the simplified method is the largest as shown in Fig. 1, which is more conducive to our evaluation of the degree of difference between the source apportionment results obtained by the two methods.

Table 2. The regression coefficients calculated with $\delta^{15}N_{4a\text{-}3s}$ as the dependent variable and the isotopic

295   **fractionation factor (α) and the molar fraction of NH_4^+ to NH_x in the atmosphere (ξ_A) as independent variables.**

	B	C	D	r^2	SSE
MS1	0.654	0.728	12.534	0.967	150
MS2	0.748	0.775	9.468	0.986	86.6
MS3	0.790	0.793	8.138	0.991	64.3
MS4	0.854	0.843	5.747	0.996	32.2
MS5	0.881	0.862	4.787	0.997	23.6
MS6	0.928	0.882	3.267	0.998	11.6

Note: r^2 is the determination coefficient and SSE is sum of squares error.

For ease of comparing the differences, we summarized both the reported previously results and the

300   findings of this study in Fig. 3. For the sources of atmospheric NH_4^+ in 2013, the most noticeable difference between the two source apportionment methods was the significant increase in the contribution proportion of fertilizer application (from 32.9% using the simplified method to 68.8% using the fitting equation method) and the corresponding NH_4^+ concentration (from 1.33 μg/m^3 using the simplified method to 2.78 μg/m^3 using the fitting method). The contributions of the other three emission sources

305   decreased to varying degrees. When we previously used the simplified method for NH_4^+ source apportionment, we found that literature reports indicated similar annual NH_3 emissions from fertilizer application and livestock farming in North China, and even in Shandong Province(Zhang et al., 2010). Based on this evidence, we determined that agricultural sources were the main contributors to atmospheric NH_4^+ in the Yellow River Delta region in 2013, accounting for 67.4%(Zong et al., 2023).

310   Recent NH_3 emission inventory results for Shandong showed that farmland fertilizer application sources were closer to the Yellow River Delta than livestock farming sources(Zhu et al., 2024), and that farmland emissions in summer are significantly higher than those from livestock farming(Li et al., 2021). This was consistent with the higher contribution proportion of fertilizer application to atmospheric NH_4^+ concentration obtained using the fitting equation method in this study, with the proportion also increasing

315   correspondingly to 82.3%. Thus, it could be seen that when considering the impact of atmospheric deposition on $\delta^{15}N\text{-}NH_x$ in the atmosphere, the contribution of agricultural sources would increase to some extent, and correspondingly, the contribution of non-agricultural sources would decrease.
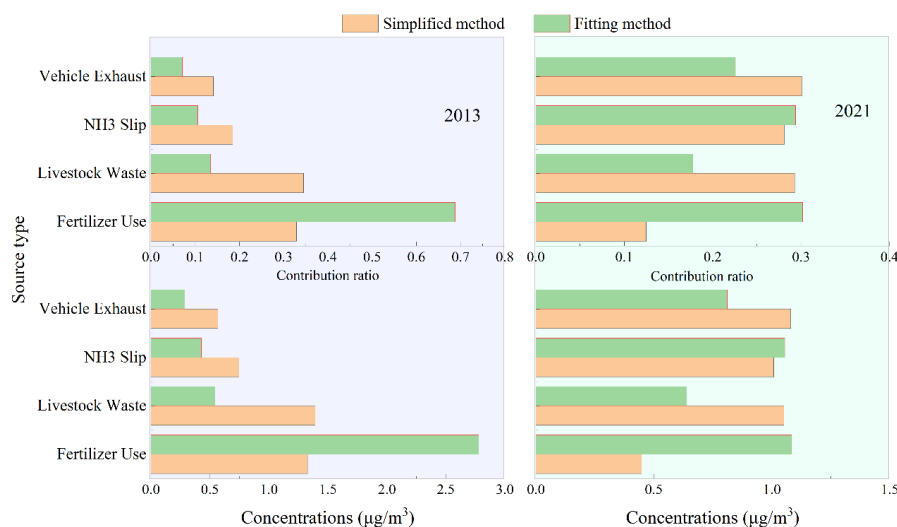
Figure 3. Comparison charts of the source contribution ratios (top) and source contribution concentrations (bottom) of NH₄⁺ in the atmosphere of the Yellow River Delta in summer of 2013 (left) and 2021 (right) from four types of NH₃ emission sources using simplified method and the fitting equation method of this study.

For the sources of NH$_4^+$ in the atmosphere in 2021, the most noticeable difference between the two source apportionment methods remained that the contribution proportion of fertilizer application obtained using the fitting equation method was significantly higher (30.2% by the fitting method vs. 12.5% by the simplified method) and also significantly higher for the NH$_4^+$ concentration in the atmosphere (1.08 μg/m^3 by the fitting method vs. 0.45 μg/m^3 by the simplified method). At the same time, the predicted contribution percentage and concentration of NH$_3$ emissions from livestock farming by the fitting method was much lower than those by the simple method, 17.8% and 0.64 μg/m^3 for the former and 29.3% and 1.05 μg/m^3 for the latter, respectively. Thus, due to the offsetting increase and decrease in the contributions of these two types of agricultural sources to atmospheric NH$_4^+$ concentration, the differences in the contributions of agricultural and non-agricultural sources obtained using the two methods were not significant (agricultural sources: 41.8% using the simplified method vs. 48.0% using the fitting method; non-agricultural sources: 58.2% using the simplified method vs. 52.0% using the fitting method).

The difference in the contributions of NH$_4^+$ sources for 2013 resolved by the two methods was significantly greater than that for 2021, primarily due to the decrease in the average percentage of NH$_4^+$ in NH$_x$ in the atmosphere from 71.9% in 2013 to 41.6% in 2021. As shown in Fig. 1, as the value of ζ_A decreases, the gap between the simplified method and the fitting equation method becomes smaller and smaller, leading to a gradual reduction in the deviation between the source apportionment results obtained by the two methods. From this case, it can be seen that the source apportionment results considering the impact of atmospheric deposition are more in line with the objective reality. Using the model parameters (G$_4$, D$_3$, D$_4$) validated against experimental data, the simulation results of source apportionment can be further refined.

4. Implications and Outlook

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Using Bayesian mixing models to apportion sources of atmospheric NH_4^+ based on $\delta^{15}N$ data has garnered widespread attention and become a commonly applied tracing method in recent years. As the importance of this tracking method increases, this methodology is also continuously developing and improving. For instance, $\delta^{15}N$-NH_3 emitted from various sources were continuously reported and

350 supplemented (Chang et al., 2016;Ti et al., 2021;Li et al., 2023), corrections were made for the impact of active and passive sampling on the $\delta^{15}N$ values of atmospheric NH_3 and NH_4^+ (Kawashima et al., 2021;Pan et al., 2020), quantitative assessments were conducted on equilibrium and kinetic isotopic fractionation during the gas-to-particle conversion of NH_3 to NH_4^+ in the atmosphere, and etc (Walters et al., 2019;Gu et al., 2025).

355 NH_3 emitted into the atmosphere undergoes continuous gas-to-particle conversion between NH_3 to NH_4^+ as it is transported and dispersed, ultimately leaving the atmospheric system primarily in the form of NH_3 to NH_4^+ through deposition. The gas-to-particle conversion process of NH_3 to NH_4^+ in the atmosphere exhibits significant isotopic fractionation, and there are marked differences in the deposition rates of NH_3 to NH_4^+, leading to continuous changes in the atmospheric $\delta^{15}N$-NH_x values. This change is widely

360 recognized but has not been fully considered in the source apportionment method based on nitrogen isotope. This study developed a model to quantitatively assess the variation pattern of atmospheric $\delta^{15}N$ values under atmospheric deposition scenarios. Furthermore, a regression equation was constructed through nonlinear fitting to facilitate the application of this research finding in the tracing of atmospheric NH_4^+. A comparative case study revealed that using simplified methods for source apportionment of

365 NH_4^+ could overestimate the contribution of non-agricultural sources. This understanding could partly explain the discrepancies in atmospheric ammonium sources derived from emission inventory methods and nitrogen isotope methods (Chen et al., 2022;Gu et al., 2025;Chang et al., 2019b).

The key parameters in this model (e.g., D_3, D_4, and G_4) are currently set based on a synthesis of literature values. These parameters exhibit significant spatial heterogeneity and temporal dynamics, influenced by

370 factors such as land surface type, meteorological conditions, and atmospheric chemical composition. Future work requires multi-site, multi-season synchronous observations combining micrometeorological methods and isotopic measurements to directly obtain parameter values under different environments. This will enhance the model's empirical foundation and regional applicability.

Data availability.

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The NH_4^+ data and the model developed in this study are available from the corresponding author (Chongguo Tian, cgtian@yic.ac.cn) upon reasonable request.

Author contributions.

380 The manuscript was written through contributions of all authors. CT designed the research, supervised the project, drafted the manuscript, and handled the submission. XHY contributed to data interpretation, figure preparation, and manuscript improvement. XNY and XT assisted with data processing and model simulations. XXY contributed to method development, supported chemical analyses, and ensured data quality control. ZZ and YL were responsible for field sampling design, sample collection logistics, and 385 data validation. RK and YFL provided scientific consultation, helped improve the manuscript structure, and contributed to English language polishing. All authors have read and approved the final version of the manuscript.

Competing interests.

The contact author has declared that none of the authors has any competing interests.

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