

Mercury contamination in staple crops impacted by Artisanal Small-scale Gold Mining (ASGM): Stable Hg isotopes demonstrate dominance of atmospheric uptake pathway for Hg in crops.

Excellent O. Eboigbe¹, Nimelan Veerasamy¹, Abiodun M. Odukoya², Nnamdi C. Anene³, Jeroen E. Sonke⁴, Sayuri Sagisaka Méndez⁵, David S. McLagan^{1,6,*}

¹Dept. of Geological Sciences and Geological Engineering, Queens University, 36 Union St, Kingston ON, Canada, K7L3N6; ²Department of Geosciences, University of Lagos, G97X+XJC, Yaba, Oworonshoki 101245, Lagos, Nigeria; ³Ministry of Solid Minerals Development, Government of Nigeria, P.M.B. 107, Luanda Crescent, Wuse II, Abuja 904101, Federal Capital Territory, Nigeria; ⁴Géosciences Environnement Toulouse, CNRS/IRD/University of Toulouse, 14 Av. Edouard Belin, 31400 Toulouse, France; ⁵Department of Environmental Sciences, University of Toronto, 1265 Military Trail, Toronto, ON, M1C1A4, Canada; ⁶School of Environmental Studies, Queen's University, 116 Barrie St, Kingston, ON, K7L3N6, Canada.

* Contact author: david.mclagan@queensu.ca.

Summary

High mercury concentrations are observed in cassava, maize, and peanuts (groundnuts) near an ASGM site. Stable Hg isotopes indicate atmospheric uptake (foliar assimilation) is the dominant mercury uptake mechanism and transfer pathway to other crop tissues in these unsaturated soil crops.

Abstract

This study investigates mercury (Hg) biogeochemical cycling and Hg uptake mechanisms in three common staple crops at a contaminated farm (Farm1) $\approx$ 500m from an artisanal and small-scale gold mining (ASGM) processing site (PS) and a background farm (Farm2; $\approx$ 8km upwind) in Nigeria. We examine air, soil, and various crop tissues using total Hg (THg), Hg stable isotope, Hg speciation, and methyl-Hg (MeHg) analyses.

Results show elevated gaseous elemental Hg (GEM) levels in the air (mean concentrations: $1200 \pm 400 \text{ ng m}^{-3}$) and soil (mean THg concentration: $2470 \pm 1640 \text{ } \mu\text{g kg}^{-1}$) at the PS, significantly higher than those at Farm1 (GEM: $54 \pm 19 \text{ ng m}^{-3}$; THg: $76.6 \pm 59.7 \text{ } \mu\text{g kg}^{-1}$), which are in turn significantly higher

than background site, Farm2 (GEM: 1.7 ng m^{-3} ; THg: $11.3 \pm 8 \text{ } \mu\text{g kg}^{-1}$). These data confirm the ASGM-derived Hg contamination at the PS and the exposures of crops at Farm1 to elevated levels of Hg in both air and soil. Aligning with Hg concentrations in air and soil, Farm1 had significantly higher THg concentrations in all crop tissues compared to Farm2. At Farm1, foliage exhibits the highest THg concentrations in tissues, particularly for peanuts and cassava (up to $385 \pm 20 \text{ } \mu\text{g kg}^{-1}$ in peanuts).

These data, along with highly negative $\delta^{202}\text{Hg}$ values in foliage and other crop tissues (indicative of light Hg isotope enrichment imparted during stomatal assimilation of Hg) demonstrate atmospheric uptake of GEM as the primary uptake pathway for Hg in these crops. We observe air-to-foliage mass dependent enrichment factors ($\epsilon^{202}\text{Hg}$) of -2.60 ± 0.35 , -2.54 ± 0.35 , and $-1.28 \pm 0.43\text{‰}$ for cassava, peanuts, and maize, respectively. While our two-endmember mixing model shows Hg in crop roots is influenced by both soil (59-74%) and atmospheric (26-41%) uptake pathways, we suggest soil Hg in roots is largely associated with root epidermis/cortex (external root tissues) and little soil-derived Hg is transferred to above ground tissues ($<7\%$ across all crops). The lower THg concentrations in edible parts (with the exception of cassava leaves, commonly eaten in Nigeria) indicate that even translocation from foliage to other tissues is a relatively slow process. MeHg concentrations were $<1\%$ across all tissues and probable dietary intakes (PDI) for both MeHg and THg based on typical diets in Nigeria are all below reference dose thresholds, indicating these crops are generally lower health risk to the local population.

1 Introduction

Artisanal and small-scale gold mining (ASGM) is generally defined as mining activities related to the extraction of gold that involves minimal (or no) mechanisation undertaken by individuals or small groups/cooperatives whose participation in these activities ranges from regulated to informal (illegal); specific definitions can vary between jurisdictions (Hentschel et al., 2002; Seccatore et al., 2014). In recent years, the ASGM sector has grown exponentially, driven by rising gold prices and the ease of selling gold (World Gold Council, 2019; Verbrugge and Geenen, 2019; Achina-Obeng and Aram, 2022). Currently, ASGM contributes $\approx 20\text{-}30\%$ of global gold production (PlanetGOLD, 2022), especially in emerging economies where it serves as a vital source of livelihood for many communities. Despite attempts to regulate mercury (Hg) use in ASGM under the Minamata Convention (UNEP, 2013), elemental Hg (Hg(0)) use remains a fundamental part of gold processing in ASGM due to the effectiveness and simplicity of the Hg-gold amalgamation process (Vieira et al.,

2006; Bugmann et al., 2022) and the general preference of miners for Hg-amalgamation (Hinton, 2003; Jönsson et al., 2013).

ASGM is now considered the largest global source of anthropogenic Hg emissions (Streets et al., 2019; Munthe et al., 2019; Yoshimura et al., 2021). Recent estimates suggest that ASGM emits 838 ± 163 Mg of Hg to air (almost entirely as gaseous Hg(0): GEM) and releases 1221 ± 637 Mg of inorganic Hg forms, Hg(0) and divalent Hg (Hg(II)) to land and rivers annually (Munthe et al., 2019). The continued rapid growth of the sector in the decade since the ratification and implementation of the Minamata Convention raises questions about the effectiveness of the measures introduced by the Convention in reducing the use and impacts of Hg in ASGM. Such concerns are largely driven by growth in illegal mining, a thriving illicit, international trade market of Hg, and the criminal networks tied to both issues (Verité, 2016; UNEP, 2017; Lewis et al., 2019; Marshall et al., 2020; Cheng et al., 2022). These security issues also present a major barrier to the implementation of more effective and holistic study of Hg use and impacts in ASGM areas (Moreno-Brush, 2020).

GEM has a long atmospheric residence time (≈ 6 -18 months), and long-range atmospheric transport is the dominant mechanism for the global redistribution of Hg (Ariya et al., 2015). Hence, Hg emitted from sources such as ASGM can have impacts on human and environmental health in areas great distances from these activities (Bose-O'Reilly et al., 2010; Weinhouse et al., 2021). Work in recent decades has shown that terrestrial plants play a critical role in the global Hg cycle, acting as the primary sink of atmospheric Hg in terrestrial systems through the assimilation of GEM into foliar stomata during photosynthesis (a mechanism of GEM dry deposition) and subsequent storage within plant tissues (Jiskra et al., 2018; Obrist et al., 2021). This is observed in trees, grasses (e.g. Millhollen et al., 2006; Mao et al., 2013; Assad et al., 2016), and crops like rice, wheat, and corn (e.g. Niu et al., 2011; Yin et al., 2013; Sun et al., 2019). Total Hg (THg) concentrations in plant foliage (and other above ground tissues) are proportional to local GEM concentrations (Millhollen et al., 2006; Fu et al., 2016; Sun et al., 2019; Wang et al., 2020). Another potential uptake pathway of atmospheric Hg in plants is sorption to and transfer through the foliage cuticle. Yet wash-off by precipitation, revolatilisation of sorbed Hg, and the likely slow transfer through the cuticle result in this uptake mechanism being minor compared to the stomatal assimilation pathway (Rea et al., 2000; Rutter et al., 2011a; 2011b; Laacouri et al., 2013). While there is potential for plants to also take up Hg from soil via roots, a large body of research indicates that $>90\%$ of Hg in the above-ground biomass of plants is derived from the air-foliage pathway with the root epidermis/cortex providing an effective barrier for the less

bioavailable forms of Hg found in soils (Beauford et al., 1977; Rutter et al., 2011b; Zhou et al., 2021). A major exception to this is the uptake of the more bioaccumulative and toxic methyl-Hg (MeHg) in species growing in saturated soils such as rice (Qui et al., 2007).

Critical to the advancements that have been made in understanding the importance of this GEM uptake mechanism by vegetation is the use of Hg stable isotope analyses in air, plants, soils, and precipitation samples. Hg has seven stable isotopes, which undergo both mass-dependent fractionation (MDF) and mass-independent fractionation (MIF) in the environment (Blum and Bergquist, 2007). MDF occurs during biogeochemical transformations, while processes causing MIF are rarer and linked largely to photochemical processes and some dark abiotic reactions; both MDF and MIF enable researchers to track Hg sources and identify *in-situ* transformation processes (Bergquist and Blum, 2009). For example, MDF is useful in tracking plant uptake, where foliage often shows large negative MDF shifts (-1 to -3 ‰ in $\delta^{202}\text{Hg}$) during stomatal assimilation compared to the $\delta^{202}\text{Hg}$ isotope values of GEM in the surrounding air (Zhou et al., 2021, and references therein). After being taken up by leaf stomata, GEM is rapidly oxidised to divalent forms and can then translocate to other plant tissues, including stems, branches, bark, and seeds, supported by negative $\delta^{202}\text{Hg}$ values in stem and seeds that closely resemble values of observed in foliage (Yin et al., 2013; Sun et al., 2019; Liu et al., 2020; McLagan et al., 2022a).

Significant gaps remain in our understanding of Hg uptake mechanisms, internal cycling, and associated health risks from Hg in crops particularly as this relates to the largest global anthropogenic emitter of Hg: ASGM. The limited number of studies examining Hg in crops affected by ASGM activities have primarily focussed on THg and occasionally also MeHg analyses, which may not fully capture the complexity of Hg dynamics in these systems. In addition, such studies have typically assumed soil-roots as the dominant Hg uptake mechanism (Suhadi et al., 2021; Addai-Arhin et al., 2023).

In this study, we employ a multidisciplinary, total systems approach to comprehensively examine the dominant Hg uptake pathway, and internal translocation and storage of Hg in three staple crops in a Nigerian farm adjacent to ASGM activities. Air, soil, and a range of crop tissues (foliage, stems, roots, and tubers/grains) were assessed using a series of Hg analyses (THg, MeHg, Hg stable isotopes, and Hg speciation) to provide critical data on the biogeochemical cycling of Hg in agricultural regions impacted by ASGM and preliminary assessment of the potential risks to human health from consuming these crops.

2 METHODS

We note, methods described in this section are abbreviated due the need for conciseness and to the broad multidisciplinary approach utilised. Full details of the study site, sampling approaches, analytical methods, and quality assurance and quality control (QAQC) are provided in the supplementary information.

2.1 Study Area

The study is based around a mine (8.87126° N, 7.71828° E) and ASGM processing site (8.9012° N, 7.7061° E) separated by ≈ 3.5 km and situated near the town of Uke (population: $\approx 20,000$) in the Karu Local Government Area of Nasarawa State, ≈ 51 km southeast of Abuja, the capital city of Nigeria (Figure 1). ASGM in this area began around 2015 following the discovery of gold, which attracted a significant influx of miners and includes mining, ore processing with Hg, and amalgam burning (further details of the ASGM activities are described in Section S1).

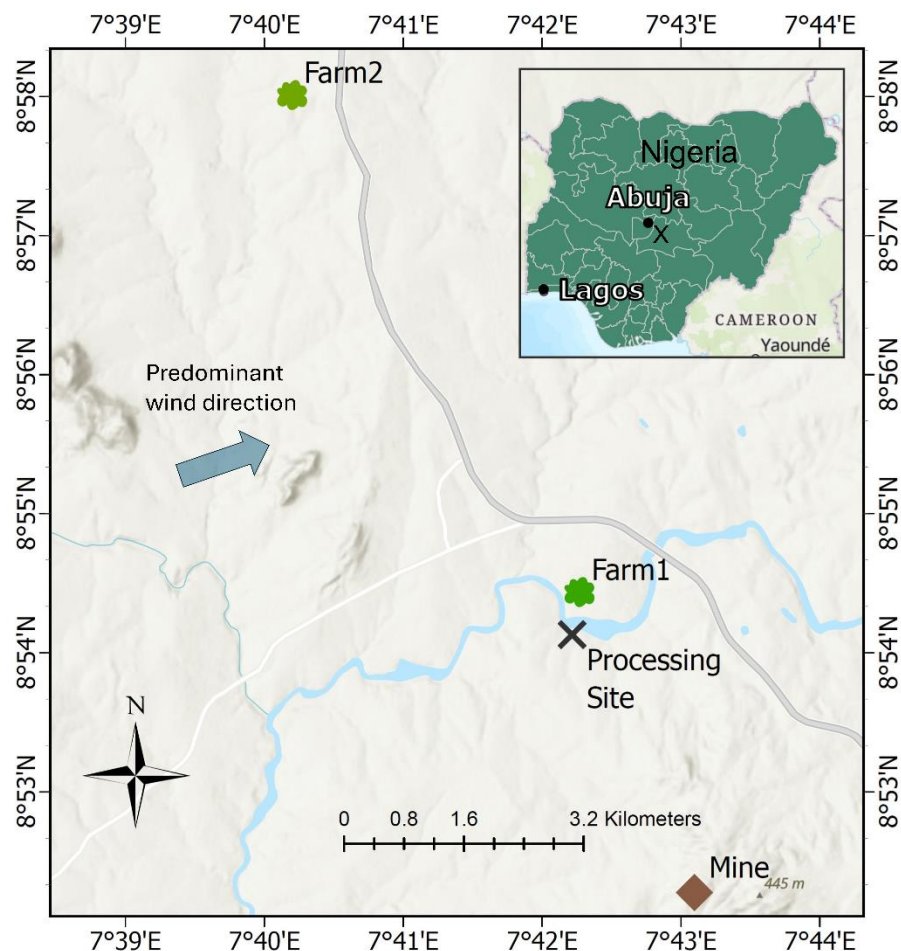


Figure 1: Map showing the study area showing mine, processing site (PS), Farm1, Farm2, predominant winds in the region (Iorhemba and Mijinyawa, 2021), and an inset map indicating site and major Nigerian cities and the processing site marked by the 'x' (Basemap: ©OpenStreetMap Foundation).

In July 2023, sampling was conducted at three sites: (1) the ASGM processing site (Site name: PS), (2) Farm1: situated ≈ 0.5 km north of the processing site, and (3) Farm2: a control farm located ≈ 8 km north-northwest of the processing site in another town with no known sources of Hg, ASGM or otherwise (Figure 1). Predominant winds in this region are from the west-southwest (Iorhemba and Mijinyawa, 2021); thus, there is potential for emissions from the processing site to impact Farm1 (but unlikely to influence Farm2). Critically, Farm1 is on the opposite side to the processing site to ensure surface or groundwater flows from the processing site were not contaminating the Farm1 soils. We note that the informal nature of ASGM restricted sampling to the scope outlined below, and this was by invitation and allowable concession of the operators of the legally operating site, local and

national governmental authorities, and the farmers themselves. This was particularly limiting to the number Hg stable isotope samples that could be collected for each crop and crop tissue type.

2.2 Soil sampling

Surface soil samples (0 – 10 cm) were collected from the PS (n=13), Farm1 (n=8), and Farm2 (n=5) (farm soils were sampled directly around sampled plants) using standard sampling procedures (USEPA, 2023). Full details of the soil sampling procedures are provided in Section S2.

2.3 Air sampling

Considering the substantial emissions of Hg to air from ASGM and the potential for GEM uptake by vegetation (crops), it was critical to assess GEM concentrations. Hg passive air samplers (MerPAS; Tekran Instruments Corp.) were deployed according to the guidelines of the developers (McLagan et al., 2016), and modifications were made for deployments in highly contaminated areas (shorter deployments and extreme care in sampler transport and storage) (McLagan et al., 2019; Tekran Instruments, 2020; Si et al., 2020). MerPAS were deployed for ≈72 hours as PS (n=3), ≈144 hours at Farm1 (n=6), and ≈100 days at (control) Farm2 (n=1). See Section S2 for full details on MerPAS deployments (including blank details) and Table S7.1 for specific sampling periods.

2.4 Crop sampling

Foliage, stem, grains/tubers, and roots samples of three crops cultivated in both Farm1 and Farm2: maize (*Zea mays*), cassava (*Manihot esculenta*), and peanuts (*Arachis hypogaea*) were collected. We note that in cassava, tubers are true roots, but we classify “roots” as undeveloped (no tuber) adventitious roots/rootlets (Rees et al., 2012). Cassava and maize were nearing maturity at the time of sampling and ≈1-2 months (or less) before harvest, while peanuts were fully mature and being harvested at the time of sampling. Three whole plants of each crop at each farm were removed from the soil for sampling with care made to consolidate below ground tissues with each selected plant. Due to sampling availability and access restrictions placed by farmers, the community, and mine owners we could not deviate from this sample timing (and quantity). Full Details of crop sampling procedures are provided in Section S2.

2.5 Analytical Procedures

2.5.1 Solid-phase THg analyses

THg analysis for soil and plant samples (0.01 – 0.2 g aliquots) was carried out by thermal desorption, gold amalgamation, and atomic adsorption spectrometry according to USEPA Method 7473 (USEPA, 2007) using a MA-3000 direct Hg analyser (Nippon Instruments Corp.). All tissue and soil samples were measured in triplicate for THg and full details of the analytical method are provided in Section S3. THg concentration analysis of MerPAS samples also utilised the MA-3000 Direct Hg Analyser (with ≈ 0.1 g additions of pre-cleaned sodium carbonate) and followed methods developed by McLagan et al. (2016) with some refinements (including a verified 400°C maximum combustion temperature) detailed in Section S3. Calculations of GEM concentrations (ng m^{-3}) followed methods described by McLagan et al. (2016) and further details are provided in Section S3. MerPAS samples used for Hg stable isotope analyses of GEM were corrected for the MDF offset (measured $\delta^{202}\text{Hg} + 1.1 \pm 0.2 \text{ ‰}$) as posited by Szponar et al. (2020).

Quality assurance and quality control (QAQC) were exercised using replication of all THg sample analyses (2-3 replicates), and regular (after every 10 sample runs) analyses of (matrix matched) certified reference materials (CRMs) and the internal liquid Hg standard. All recoveries were within accepted ranges and specific details on the CRMs used and recovery data are presented in Table S3.1. All data are presented on a dry-weight (dw) basis.

2.5.2 Hg stable isotopes: extractions and analyses

All samples analysed for Hg stable isotopes were trapped off the exhaust of the MA3000 detector during solid-phase THg analyses of the same matrix for sample preconcentration. This required combining some tissue samples for each crop species from individual plants at Farm1 due to low THg concentrations (Farm2 samples were not considered due to low concentrations and analytical capacities of the MC-ICP-MS). The combust and trap method broadly followed methods by Enrico et al. (2021) and McLagan et al. (2022b) with some modifications. This method allows accumulation of Hg from matrix matched samples measured for solid-phase THg concentrations within a single 5 or 10 mL inverse aqua regia (3:1 concentrated HNO_3 :HCl diluted to 40% v/v with DI water) trap. MA-3000 combustion method followed the matrix specific methods described in Section S3. A heated ($\approx 60^\circ\text{C}$) polytetrafluoroethylene (PTFE) tube connected a coarse frit gas dispersion sparger (6 mm outer

diameter; modified to 25 cm length) was connected to the MA-3000 detector outlet. Trapping samples analysed for THg concentrations allows for direct recovery testing between the measured THg solid-phase concentrations to the liquid-phase THg concentrations within the trap, which removes the uncertainty of assuming homogenous THg solid-phase concentrations. Any sample with recovery below 80% was not considered for Hg stable isotope analyses due to concerns that losses during extraction/analyses could artificially induce isotope fractionations. Recoveries of all samples analysed for Hg stable isotopes are listed in Table S3.3. All samples were diluted to a 20% acid strength (by volume) for Hg stable isotope analyses.

An online cold vapor generator (CETAC-HGX-200) was used to reduce Hg(II) in the trapping solutions into Hg(0) vapor by SnCl_2 (3% w/v in 1 M HCl). Using this system, gas-phase Hg(0) is then introduced into a multi-collector inductively coupled mass spectrometry (MC-ICP-MS, Thermo-Finnigan Neptune) for analyses of Hg stable isotopes at the Observatoire Midi-Pyrenees (Toulouse, France). Full details of instrument setup can be found in Sun et al., 2013. Sample isotope ratios were corrected for mass bias by sample-standard bracketing using NIST 3133 (Blum and Bergquist, 2007; Sun et al., 2013). Results are reported as δ -values in per mil (‰) by referencing to NIST 3133, representing Hg mass-dependent fractionation (MDF), while MIF is reported in “capital delta” notation (Δ), which is the difference between the measured values and those predicted by the kinetic MDF law using equations previously stated (Blum and Bergquist, 2007). The quality control of Hg isotope measurements was assessed by analysing ETH-Fluka, and UM Almaden reference standards and these data are presented in Table S3.3. Uncertainties on sample δ and Δ -signatures were conservatively estimated as the larger of the two standard deviation uncertainties on weekly ETH-Fluka, UM-Almaden or sample replicates (Table S3.3). Due to the limited number of Faraday cups on the Neptune MC-ICP-MS, analysis of ^{204}Hg was preferred to ^{203}Tl or ^{205}Tl (used in some studies for mass-bias corrections). However, we confirm this follows the same method implemented in Jiskra et al. (2021) (further discussion on this is provided in Section S3).

2.5.3 Hg Speciation/Fractionation Analyses

Solid-phase speciation/fractionation analyses were performed using the pyrolytic thermal desorption (PTD) method developed by Biester and Scholz (1996) adapted for use on a Lumex 915M with PYRO-915+ Module (Lumex Instruments Corp.) by Mashyanov et al. (2017). Due to the inherent uncertainties and challenges in peak identification of Hg(II) species, these analyses are considered qualitative and complementary (McLagan et al., 2022b). Further details of this method are provided

in Section S3.

2.5.4 Methyl-Hg (MeHg) Analyses

MeHg concentrations were determined using isotope dilution method and followed methods described in Mitchell and Gilmour (2008). A detailed description of this method is provided in Section S3. All QAQC data for MeHg analyses are presented in Table S3.2.

2.6 Two endmember mixing model to identify Hg sources within crops

A two-endmember mixing model was used to quantitatively determine source pathways for Hg in internal crop tissues according to Equation S4.1. The $\delta^{202}\text{Hg}$ values of foliage for each crop are used as the first endmember: air-foliage uptake pathway. Similarly, the soil-root endmember must be the $\delta^{202}\text{Hg}$ signature of Hg immediately after uptake to the roots. Hence, the mean $\delta^{202}\text{Hg}$ value for Farm1 soils minus the soil-to-shallow roots (roots above 150cm) MDF ($\epsilon^{202}\text{Hg}$: -0.35) taken from Yuan et al. (2022) is used. Details of this two-endmember mixing model and the data used in the derivation of the endmembers are provided in Section S4.

2.7 Estimates of annual Hg dry deposition rates to crops

Hg(0) dry deposition rates ($F_{\text{Hg}(0):AGB}$; g km^{-2}) via the foliar uptake pathway to the aboveground biomass (AGB) for each crop were calculated using Equations S5.1 and S5.2 adapted from Casagrande et al. (2020) with additions relating to the transfer of Hg to other above ground tissues and the transfer to edible below ground parts (for peanuts and cassava). This was achieved using annual edible biomass yields and the fraction of Hg in tubers/nuts derived from a foliage-based two-endmember mixing model (Table 1). Details of these calculations are provided in Section S5.

2.8 Probable daily intake calculations

We also calculated the probable daily intake (PDI) of MeHg and Hg(II) using the method from Zhao et al. (2019) adapted for MeHg and Hg(II) and the concentrations measured in the examined crops and their mean dietary intake data for adults in Nigeria. Specific equations and data used in these calculations and the PDI data generated are presented in Section S6.

2.9 Statistical Analyses

Statistical tests (Welch's T-test; unequal variances and sample sizes) used to assess differences in

THg concentrations and Hg stable isotopes, data analyses, and figure generations were all generated in R-Studio (Boston, USA).

3 Results and Discussion

3.1 Crop exposure levels: Hg concentrations in air and soils

Highly elevated GEM concentrations ($1200 \pm 400 \text{ ng m}^{-3}$) were observed at the ASGM processing site (PS) (Figure 2). These concentrations are $\approx 1000\times$ background concentrations and consistent with levels observed in other ASGM regions where Hg is used in ASGM activities (González-Carrasco et al., 2011; Kawakami et al., 2019; Nakazawa et al., 2021; Snow et al., 2021) and indicate substantial Hg use and emissions at PS. At Farm1 ($\approx 500\text{m}$ distance from PS), GEM concentrations were significantly lower than at PS ($p=0.014$), but remained elevated ($54 \pm 19 \text{ ng m}^{-3}$) above background, confirming exposure of Farm1 to GEM emissions from ASGM activities at PS. In contrast, GEM concentrations at Farm2 (1.7 ng m^{-3} , $n=1$) were consistent with the Northern Hemisphere background concentrations ($\approx 1.5 - 2 \text{ ng m}^{-3}$; Sprovieri et al., 2016).

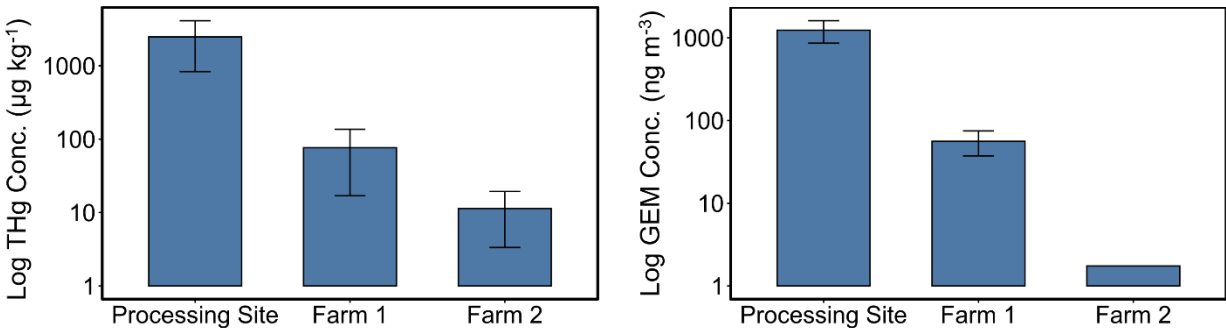


Figure 2: THg concentrations in soils (left; $\mu\text{g kg}^{-1}$) and GEM concentrations in air (right; ng m^{-3}) for all sampling sites.

Stable isotope measurements of GEM at sites exposed to elevated GEM concentrations were indicative of more negative MDF and more positive MIF signatures (PS: $\delta^{202}\text{Hg}$: $-1.38 \pm 0.21 \text{ ‰}$, $\Delta^{199}\text{Hg}$: $0.07 \pm 0.03 \text{ ‰}$; Farm1: $\delta^{202}\text{Hg}$: $-0.94 \pm 0.19 \text{ ‰}$, $\Delta^{199}\text{Hg}$: $0.08 \pm 0.08 \text{ ‰}$) (Figure 4), which is typical of anthropogenic Hg emitted into air from industrial (Sonke et al., 2010; Fu et al., 2021; McLagan et al., 2022b) and ASGM (Gerson et al., 2022; Szponar et al., 2025) sources. Considering the burning of Hg-gold amalgams emits Hg(0) directly into the atmosphere, we deem the mean stable isotope values for Hg(0) in air in the contaminated areas (PS and Farm1) to be signal most representative of the

ASGM source. Contrastingly, GEM at Farm2 was relatively enriched in heavier isotopes and had a slightly negative MIF signal ($\delta^{202}\text{Hg}$: $-0.01 \pm 0.19 \text{ ‰}$; $\Delta^{199}\text{Hg}$: $-0.12 \pm 0.08 \text{ ‰}$) typical for GEM samples in background areas (Si et al., 2020; Szponar et al., 2020). We suspect that the rapid decline in GEM concentrations as we move away from the contamination source is influenced both by dilution with background air and by wind direction. Although GEM was not measured in areas downwind of the processing site, McLagan et al. (2019) observed a more gradual decline in GEM in areas downwind (compared to upwind sites) of a former Hg mine that remained heavily contaminated by elemental Hg; we suggest a similar pattern is likely at our study site. The vegetation of the area itself may play a role in reducing GEM concentrations around the ASGM area by removing GEM from the atmosphere during plant growing seasons. An assessment of this flux is described below in Section 3.4.

Similar to GEM, soil samples at PS were heavily contaminated (mean THg concentration: $2470 \pm 1640 \text{ } \mu\text{g kg}^{-1}$). Elevated soil Hg at the processing site is likely due to rapid GEM deposition from the atmosphere after emissions from amalgam burning and direct spills from improper handling of liquid Hg (Telmer and Viega, 2009). The large variation in soil THg concentrations at this site (see Table S4.1) also reflects the spatial heterogeneity of processing activities at PS where Hg-Au amalgamation, ore washing, amalgam burning, and other activities occur.

Farm1 soils were also contaminated (mean THg concentration: $80.9 \pm 60.1 \text{ } \mu\text{g kg}^{-1}$), but similar to differences in GEM concentrations, Farm1 was 1-2 orders of magnitude (and significantly; $p < 0.001$) lower than PS $\approx 500\text{m}$ away. Hence, we suggest emissions of GEM from the PS and deposition directly to soils or via GEM assimilation into vegetation, litterfall, and decomposition as the dominant contamination pathway at Farm1, a mechanism now well described in the literature (Jiskra et al., 2015; Obrist et al., 2017; Zhou and Obrist, 2021). Our results fall into the range of concentration ($2 - 5570 \text{ } \mu\text{g kg}^{-3}$) recorded by Odukoya et al. (2020) from farmlands adjacent to an ASGM region in Niger state of Nigeria and farms impacted by ASGM in Brazil ($81.7 \pm 13.5 \text{ } \mu\text{g kg}^{-1}$) (Casagrande et al., 2020). The latter study by Casagrande et al. (2020), is one of the only other studies examining Hg in ASGM impacted agricultural areas directly attributing elevated concentrations in farm plants and soils to the atmospheric transfer pathway. In contrast, Farm2 soils were significantly lower again than Farm1 ($p = 0.029$) and at background levels with a mean concentration of $11.3 \pm 8.0 \text{ } \mu\text{g kg}^{-1}$.

All soil samples exhibited relatively small (compared to other contaminated soils) variation in $\delta^{202}\text{Hg}$ (PS: $0.29 \pm 0.98 \text{ ‰}$; Farm1 $-0.26 \pm 0.43 \text{ ‰}$) and $\Delta^{199}\text{Hg}$ MIF signal (PS: $-0.09 \pm 0.12 \text{ ‰}$; Farm 1 $-0.07 \pm 0.03 \text{ ‰}$) (Grigg et al., 2018; McLagan et al., 2022; Vaňková et al., 2024). Nonetheless, the mean

$\Delta^{199}\text{Hg}$ for GEM is slightly (but not significantly; $p=0.073$) higher than the mean value for soils. This suggests minor evidence of some MIF induced by photochemical reduction from the soils (Rose et al., 2015).

3.2 Hg contamination and distribution in crops grown in ASGM impacted areas

With the confirmation of exposure levels in both air and soils at Farm1 and Farm2 (control), it was then critical to assess the degree to which this contamination is affecting staple crops grown in this area and determine the predominant uptake mechanism of Hg in these plants. The mean THg concentration in peanut ($p=0.021$), cassava ($p=0.015$), and maize ($p=0.004$) tissues were all significantly elevated at Farm1 compared to Farm2 (Figure 3 and Table S4.3). The highest THg concentrations were detected in foliage of peanuts and cassava at Farm1 (peanut: $385 \pm 20 \mu\text{g kg}^{-1}$; cassava: $320 \pm 116 \mu\text{g kg}^{-1}$) and Farm2 (peanut: $7.06 \pm 2.74 \mu\text{g kg}^{-1}$; cassava: $13.2 \mu\text{g kg}^{-1}$) (Figure 3), which suggests that the stomatal assimilation pathway is likely the dominant uptake mechanism of Hg in cassava, and peanuts. While foliage THg concentrations were also elevated in maize (Farm1: $182 \pm 44 \mu\text{g kg}^{-1}$; Farm2: $5.16 \mu\text{g kg}^{-1}$), again demonstrating the likelihood of foliar Hg uptake, concentrations in maize roots were slightly higher (Farm1: $202 \pm 136 \mu\text{g kg}^{-1}$; Farm2: 5.74 ± 3.73) than foliage. Roots also had the second highest concentration in peanuts and cassava. These data suggest potential for soil-to-root Hg uptake in all three crops, but the contribution of the uptake mechanisms will be examined in more detail in Section 3.3.

Adjorololo-Gasokpoh et al. (2012) measured similar THg in cassava foliage (up to $177 \mu\text{g kg}^{-1}$) and tuber (up to $185 \mu\text{g kg}^{-1}$). Even though they did not observe any significant trends or differences between tissues, a novel component of their study was the division of cassava tuber into flesh, inner peel, and outer peel (Adjorololo-Gasokpoh et al., 2012). Division of root tissues into epidermis, cortex, and vascular bundle (or stele) and then analysis for THg and stable Hg isotopes in crops impacted by ASGM would provide critical insight into the effectiveness of epidermis and/or cortex in restricting Hg uptake into the vascular bundle of inner root, as has been suggested elsewhere (Rutter et al., 2011b; Lomonte et al., 2020; Yuan et al., 2022).

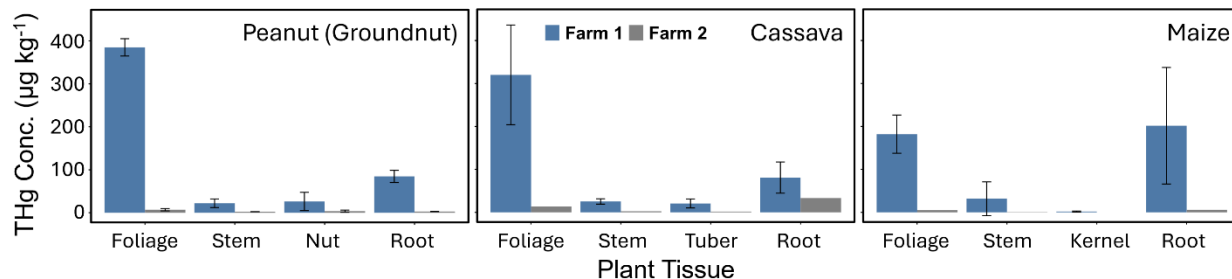


Figure 3: THg concentration ($\mu\text{g kg}^{-1}$) for all crop tissues at Farm1 and Farm2

Similar results showing the highest crop tissue THg concentrations in ASGM affected farms are common in the literature (i.e., [cassava] Golow and Adzei, 2002; [cassava] Nyanza, et al., 2014; [soy: *Glycine max*] Casagrande et al., 2020). However, we note the challenges of comparing absolute THg concentrations even of samples from the same species as distance from ASGM activities is likely a major determinant controlling observed levels of crop contamination and, in many cases, little specific information on source-receptor distances is provided (i.e., Essumang et al., 2007; Adjorololo-Gasokpoh et al., 2012; Nyanza et al., 2014).

Of the three crops we studied, maize foliage exhibited lower THg concentrations than other crops, which may be attributed to maize's C4 photosynthetic pathway. An earlier study by Browne and Fang (1983) found C3 plants to take up five times more atmospheric Hg than C4 species due to differences in leaf surface area, stomatal conductance (C3 plants exchange gases more readily with the atmosphere, allowing for greater uptake of atmospheric Hg), and internal resistance to Hg vapor uptake within the plant. Furthermore, maize kernels ($1.78 \pm 1.22 \mu\text{g kg}^{-1}$) contained the lowest concentration in all crops at Farm1 suggesting minimal translocation from foliage or roots as observed by Wang et al. (2024), which again may be attributable to physiological differences in C4 species – a hypothesis that requires direct exploration in future research. Glauser et al. (2022) also suggest non-stomatal pathways can result in Hg sorption in certain maize tissues such as maize tassels and silk. Although cassava is an intermediate between C4 and C3 plants (exhibiting some properties of both C3 and C4 photosynthetic pathways (El-Sharkawy and Cook, 1987; Bräutigam and Gowik, 2016; Xia et al., 2023), THg concentrations in cassava were higher than those of maize. This phenomenon is, however, not yet fully understood and warrants further research.

The concentration patterns for the crops were as follows: for maize, roots > foliage > stem > kernel; for peanuts, foliage > roots > nut > stem; and for cassava, foliage > roots > stem > tubers. All crops exhibited the lowest concentrations in their edible parts (kernel, nuts and tubers), except for the

peanut stem, which was slightly lower than the nuts. This finding aligns with studies on rice in China, where the stem and seeds consistently showed the lowest THg concentrations across all measured sites (Yin et al., 2013).

3.3 Tracing Uptake and Translocation of Hg Within Crops using Hg Stable Isotopes

While THg quantifies the extent of overall Hg exposure in the examined crops, we have thus far neither been able to fully confirm the Hg uptake pathway nor explain translocation processes within the crop. We therefore applied Hg stable isotope analyses to investigate these processes in greater detail. Foliage samples across all crops displayed highly negative MDF values with $\delta^{202}\text{Hg}$ values being $-3.83 \pm 0.19 \text{ ‰}$ (2SD) for cassava, $-3.77 \pm 0.19 \text{ ‰}$ (2SD) for peanuts, and $-2.51 \pm 0.32 \text{ ‰}$ (1SD) for maize (Figure 4). MIF values were all near zero ($\Delta^{199}\text{Hg}$ range: $-0.04 - 0.03 \text{ ‰}$) (Figure 4). Sun et al. (2020) measured similar MDF values in maize foliage ($-2.68 \pm 0.28 \text{ ‰}$ (1SD)). Yet they also observed more negative $\delta^{202}\text{Hg}$ in higher concentration samples (Sun et al., 2020), which is likely linked to GEM influenced by anthropogenic sources having more negative $\delta^{202}\text{Hg}$.

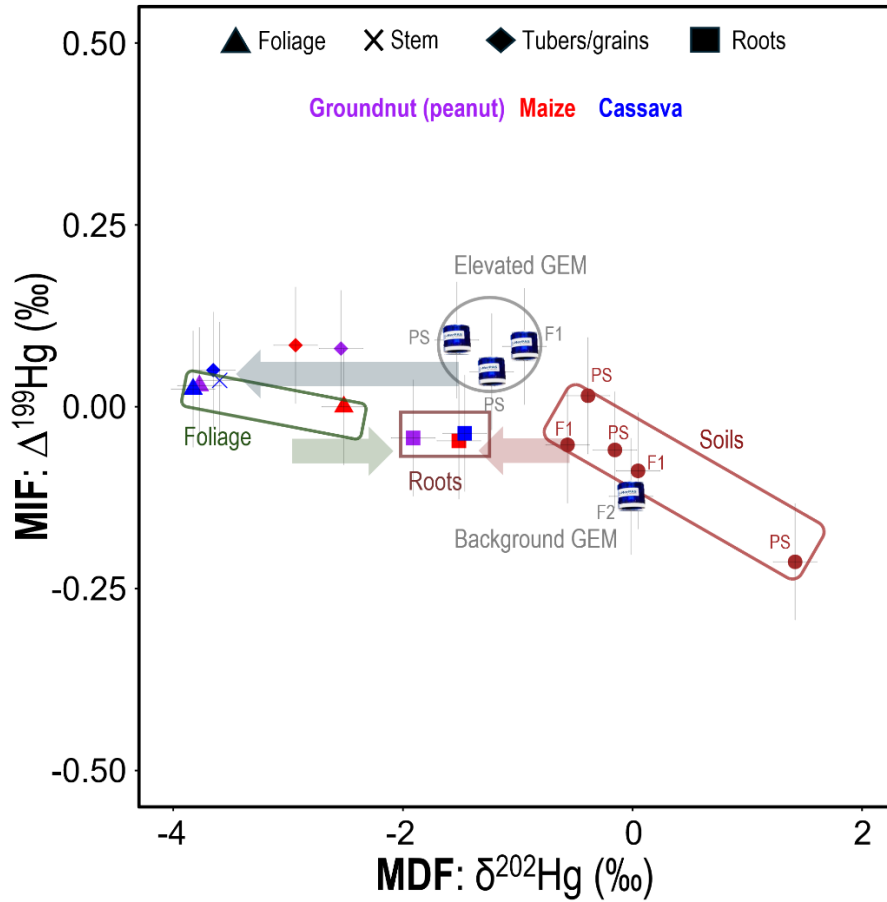


Figure 4. MDF ($\delta^{202}\text{Hg}$) and MIF ($\Delta^{199}\text{Hg}$) for GEM at PS, Farm1 (F1), Farm2 (F2), soil samples at PS and Farm1 (F1), and crop tissues for Farm1 (F1) only. The direction of the arrows shows approximate MDF from air to foliage (stomatal assimilation; blue arrow), foliage to roots (green arrow), and soil to roots (red arrow). Note that cassava tubers are flesh only (no peel).

Assuming all the Hg within foliage for these crops grown in unsaturated soils is derived from stomatal assimilation, we can use the mean Hg stable isotope values for GEM (mean MerPAS values from both PS and Farm1 to provide estimate variance) to estimate the enrichment factors (indicated by ϵ for MDF and E for MIF) associated with the stomatal assimilation process for the stomatal assimilation process in these crops. We calculate $\epsilon^{202}\text{Hg}$ for stomatal assimilation to be -2.60 ± 0.35 , -2.54 ± 0.35 , and -1.28 ± 0.43 ‰ for cassava, peanuts, and maize, respectively, and these data represent the first time such fractionation factors have been calculated for any agricultural crops. These crop stomatal assimilation MDF enrichment factors fall within the range reported (-1 to -3 ‰) in other vegetation (Zhou et al., 2021; Liu et al., 2024; and references therein). There was a small MIF between GEM and foliage ($E^{199}\text{Hg}$: cassava: -0.05 ‰; peanuts: -0.05 ‰; maize: -0.11 ‰), which is similar to the small

range observed during this process elsewhere (Demers et al., 2013). The small negative MIF shift can be attributed to minor in-plant photochemical Hg(II) reduction (Demers et al., 2013) and is typically accompanied by an enrichment in the heavy Hg isotopes, and positive shift in $\delta^{202}\text{Hg}$; while speculative, this could again be attributable to differing physiological processes (i.e., C4 photosynthesis).

Crop root samples exhibited negative $\delta^{202}\text{Hg}$ values (cassava: -1.46 ‰; peanuts; -1.91 ‰; maize: ‰) that were distinct from both soil and foliage samples, while $\Delta^{199}\text{Hg}$ values were similar to those of soil and slightly more negative than GEM and foliage samples (Figure 4). We suggest that the Hg in roots reflects a combination of inputs from both soil and foliage (via stomatal assimilation). Since foliage is a source of energy to plants in the form of the photoassimilates they generate, which are exported to growth (meristems, cambium) and storage (roots, fruits, seeds) tissues via the phloem (Turgeon, 2006), Hg is likely translocated similarly as has been posited in other studies (Zhou et al., 2021; McLagan et al., 2022a). For maize and peanut roots, the *qualitative* PTD analyses (see Figure S8.1) revealed the presence of two distinct Hg pools, which we suggest could be representative of distinct fractions of Hg in (i) the epidermis/cortex (likely derived from surrounding soils) and (ii) inner root tissues: vascular bundle/stele (likely derived from air/foliage).

We also apply a two-endmember mixing model using the mean $\delta^{202}\text{Hg}$ values of (i) GEM at Farm1 and PS, and (ii) soils at Farm1, minus the soil-to-shallow roots (roots above 150cm) MDF ($\epsilon^{202}\text{Hg}$: -0.35) from Yuan et al. (2022), as endmembers to introduce more quantitative assessment of the Hg uptake mechanisms (air/foliage vs soil) (Equation S4.1). Data reveal that between 47% (peanuts) and 66% (cassava) of Hg found in the roots of these crops is derived from air/foliage (Table 1). There is precedence for the transfer of Hg from foliage to roots with a previous study using Hg stable isotopes to indicate 44-83% of Hg in roots is derived from air in selected tree and shrub species (Wang et al., 2020). These data support the hypothesis that the majority of Hg transferred from soil to roots is likely bound to outer root tissue (epidermis/cortex) as suggested elsewhere (Rutter et al., 2011b; Lomonte et al., 2020). Again, root tissue sectioning and analysis for concentrations and stable Hg isotopes would provide the most conclusive evidence of such processes.

Hg stable isotope data for other crop tissues provided further evidence of translocation of Hg away from foliage. Due to the lower concentration and limited available sample, stems could only be analysed for stable isotopes in cassava, and this one sample displayed almost identical MDF and MIF signatures to cassava foliage (93% derived from air/foliage; Table 1). Edible parts of maize

(kernel; 100% from air/foliage) and cassava (tuber flesh; 95% from air/foliage) displayed similar patterns (Table 1). Studies on cassava in ASGM regions, such as those by Nyanza et al. (2014), have similarly demonstrated that Hg concentrations in cassava tubers remain low even in highly contaminated soils, emphasising the influence of foliar pathways. This contrasts with the common assumption that tubers accumulate Hg mainly through soil uptake (Adjorlolo-Gasokpoh et al., 2012).

Table 1: Hg uptake source apportionment in crop tissues using two-endmember, Hg stable isotope mixing model. Foliage is not included as $\delta^{202}\text{Hg}$ values for foliage represent one of the source endmembers for each crop. All estimates come with a $\pm 16\%$ uncertainty derived from propagating uncertainty terms through calculations.

Source	Cassava			Maize		Peanut	
	Stem	Tuber flesh	Root	Kernel	Roots	Nut	Roots
Air/foliage	93%	95%	26%	100%	47%	61%	41%
Soil/roots	7%	5%	74%	0%	53%	39%	59%

The only exception to this was peanut nuts, which had a $\delta^{202}\text{Hg}$ value more similar to roots (39% of nut THg derived from soil) and the most positive $\Delta^{199}\text{Hg}$ value (0.08‰) of any sample across all studied matrices. While we do not have a clear explanation for the anomalous $\Delta^{199}\text{Hg}$ value of the nut sample, we link the similar $\delta^{202}\text{Hg}$ values between peanut roots and nuts to the subsurface growth of the nut. Cassava tubers also grow underground, but their physiological function is as a plant energy storage tissue (Rees et al, 2012) as opposed to the peanut nut, which is a seed used for reproduction (Basuchaudhuri, 2022). The transfer of Hg from soils, through peanut shells and into the nut is still somewhat surprising as other studies have shown the shell to be an effective barrier at preventing metal uptake to the nut (Tang et al., 2024) including for Hg (Namasivayam and Periasamy, 1993; Cobbina et al., 2018). Nonetheless, Liu et al. (2010) observed lower Hg adsorption efficiency by natural (compared to chemically modified) peanut shells; though, we note this was in a laboratory study of just shells (nuts removed). Multi-method analyses of peanut shells would be a useful addition to future work.

3.4 Foliage as an important sink of GEM in ASGM areas

We also used Equations S4.2 and S4.3 to assess the annual GEM dry deposition flux from the atmosphere to these crops via stomatal assimilation. The GEM deposition rates from the atmosphere to leaves for peanuts, maize, and cassava were estimated to be 110 ± 80 , 690 ± 130 , and 1170 ± 180 g

km⁻² yr⁻¹, respectively. The relatively high GEM deposition rate observed for cassava shows the higher vulnerability of cassava to Hg uptake despite it being grown once annually. This may be due to its larger biomass compared to maize and peanuts. These data were substantially higher than the 13-25 g km⁻² yr⁻¹ estimate for soybeans by Casagrande et al. (2020). While this is partly attributable to Casagrande et al. (2020) not accounting for transfer from foliage to other above or below ground tissues, our estimates were dominated by Hg storage in foliage (90-92% of total; see Table S5.1). Hence, we attribute the differences between the two estimates to differences in distances between ASGM activities and farms/crops, scale of ASGM operations, and different crop physiological uptake mechanisms. These data provide crucial insight into the role of crop foliage in sequestering atmospheric Hg in regions impacted by ASGM and can be useful in environmental monitoring and risk assessment. In regions with ongoing ASGM activities, these estimates could be used as baselines to monitor shifts in atmospheric Hg concentrations over time. For instance, if emissions from ASGM were to decline due to policy interventions, a corresponding reduction in annual Hg deposition rates would be expected. Conversely, if emissions increase, these crops could serve as bioindicators for heightened atmospheric contamination.

3.5 Human health implications

MeHg concentration data is typically considered the major endpoint for assessing human and environmental health impacts of Hg. MeHg concentrations were consistently below 1% of total Hg (THg) in all samples (Table S7.5), suggesting low methylation rates of Hg(II) in these soils. Moreover, the probable daily intake (PDI) for MeHg in these crops (<0.001 µg kg⁻¹ day⁻¹) or the sum of their dietary intakes (0.0010±0.0016 µg kg⁻¹ day⁻¹) are two orders of magnitude below the USEPA reference dose (RfD) for MeHg of 0.1 µg kg⁻¹ day⁻¹ (USEPA, 2001). This contrasts data from rice grown in Hg contaminated areas, which is known to accumulate MeHg (via root uptake) due to the capacity of rice paddies to host anoxic conditions known to produce this highly bioaccumulative species (Zhao et al., 2016; 2020).

While less toxic than MeHg, inorganic Hg has been associated with health effects on gastrointestinal, renal, and nervous systems (Ha et al., 2017; Basu et al., 2023). Due to the low MeHg concentrations, inorganic Hg exposures were assessed using THg concentrations and these values were adjusted for the lower absorption rates of inorganic Hg (7%; WHO, 1990); this adjustment allows direct comparison of inorganic Hg/THg exposures to the MeHg RfD value (Zhao et al., 2019). While PDI values for inorganic Hg/THg were higher in all edible crop tissues examined (individual crop range:

0.0001 $\mu\text{g kg}^{-1} \text{ day}^{-1}$ in maize to 0.016 $\mu\text{g kg}^{-1} \text{ day}^{-1}$ in cassava leaves; dietary sum: $0.023 \pm 0.007 \mu\text{g kg}^{-1} \text{ day}^{-1}$; Table S6.1), they are again below the MeHg RfD value. Hence, there is little health risk to the local population from Hg levels ingested during typical consumption of the studied tubers/grains/nuts.

Cassava leaves are commonly consumed in various regions including Nigeria (El-Sharkawy, 2003); hence, the high inorganic Hg in cassava leaves we observed in particular could pose some risk. Without direct data for estimated dietary intake of cassava leaves in Nigeria, we chose to assume a conservative daily intake rate (50 g day^{-1} ; Section S6). Latif and Müller (2015), report that cassava leaves are consumed up to 500 g day^{-1} in countries such as Zaire and the Democratic Republic of Congo, which is an order of magnitude higher than our assumed daily intake rate. Consumption of contaminated cassava leaves at the observed levels of contamination and at rates of 500 g day^{-1} would surpass RfD levels. Despite the reported health benefits of eating cassava leaves (Latif and Müller, 2014), we suggest dietary caution when consuming cassava foliage grown in close proximity to ASGM, and this likely applies for other edible plant foliage.

Although maize and peanut leaves are not typically consumed by humans, they are frequently used as fodder for livestock (Samkol, 2018; Abdul Rahman et al., 2022). This introduces an additional layer of concern, as the ingestion of Hg-laden plants by livestock can lead to the accumulation of Hg in livestock kidneys and liver (Verman et al, 1986; Crout et al., 2004). Human exposure through the consumption of Hg contaminated meat and dairy/egg products is an additional understudied potential human exposure pathway. Adding Hg stable isotope analyses to any future work around Hg in livestock in ASGM areas could provide valuable insight into the biogeochemical processes involved.

3.6 Limitations and Future Work

As noted in Section 2.1, the scope of our sampling was limited by the social and geopolitical complexity of the ASGM issue. While it would have been optimal to assess larger crop sampling sizes at each site, we had to respect the wishes of the site operators, the community, and the farmers for whom these crops are their livelihood. Despite the lower-than-optimal sampling sizes, we achieved robustness through a thorough experimental design that captured samples from all the critical environmental compartments (and different plant tissues) and multi-method analyses. With that, we are confident in our data and the findings made with those data. Future studies should expand upon

this work by adding dissected crop tissues (i.e., roots, edible parts) to improve the assessment of uptake pathways, internal cycling of Hg by plants, and translocation into edible tissues. Hg stable isotope analyses should remain a key part of future studies of this nature. Other studies have determined more elevated concentrations of Hg in edible parts of crops near ASGM areas (i.e., Adjorololo-Gasokpoh et al., 2012; Addai-Arhin et al., 2023); hence, if it is feasible, similar structured studies to our own should attempt to assess ASGM sites of differing (larger) scales and/or the proximity of farms to these sites.

This site was chosen due to existing partnerships that were built through discourse and trust. As described in Moreno-Brush et al. (2020), these partnerships between the communities, miners, local researchers, and international collaborators are critical to the success of Hg biogeochemical assays in ASGM areas. Security and research safety are considerable issues of research conducted in ASGM areas. While this should highlight the need for strong local partnerships, we stress that flexibility and adaption are vital components of such work, work which becomes increasingly important as ASGM continues to expand in the Global South.

Supplement link

The supplementary information is available at XXXX and contains all complementary information and all data used within study as well as extended details on the study methods.

Code/Data Availability

No specialised modelling code was used to perform the calculations used in this study and all data can be recreated from the raw data and equations provided in the supplement. Please contact the corresponding author for any additional information or questions relating to the data.

Author Contributions

E.O.E. built research partnerships in Nigeria, contributed to experimental design, led field work and sample collection, undertook sample preparations and analysis, generated figures, wrote the manuscript draft. N.V. worked on the development of analytical methods (including Hg isotope combust and trap), undertook sample preparation and analysis (including Hg stable isotope analyses in Toulouse), and provided reviews of the manuscript. A.O.M. built connections with Uke community and mining stakeholders, contributed to experimental design and fieldwork, provided lab space for sample preparations at the University of Lagos, and provided reviews of the manuscript.

N.C.A. built connections with Uke community and mining stakeholders, contributed to experimental design and fieldwork (including follow-up sampling and MerPAS collections at Farm2), and provided reviews of the manuscript. J.E.S. provided guidance on Hg stable isotope sample extraction methods, led Hg stable isotopes analyses, and provided reviews of the manuscript. S.S.M. undertook all MeHg analyses and provided reviews of the manuscript. D.S.M. conceptualised the study, built research partnerships in Nigeria, worked on the development of analytical methods, contributed to figures, provided manuscript draft guidance, and reviewed the manuscript.

Competing Interests

D.S.M. is a member of the editorial board of the journal *Biogeosciences*. The authors declare that they have no other conflict of interest.

Acknowledgements

We thank the support from Zinariya Kyautar Allah (Mining) Venture Ltd. and the community of Uke, Nigeria for welcoming us into their community and being active and willing contributors to this work. We acknowledge contributions of the Analytical Services Unit (ASU) team at Queen's University (Meesha Thompson, Paula Whitely, Dale Marecak, and Graham Cairns) for their support and feedback on Hg analyses. A big thank you to Dr. Jason Demers and Dr. Bridget Bergquist for their direction and guidance in setting up our lab for Hg stable isotope sample preparations and extractions. We also thank D.K. Olukoya Central Research and Reference Laboratories from the University of Lagos for providing support in sample drying in Nigeria. We acknowledge Dr. Carl Mitchell for providing the use of their MeHg analysis methods to assess MeHg concentrations in plant and soil samples. Dr. Jerome Chmeleff is thanked for assistance with Hg isotope analyses. Finally, we thank the three reviewers and associate editor for their contributions in bringing this work to publication.

Financial Support

We acknowledge an internal, alumni-based, graduate scholarship/award (Maria Nathanson/IAMGOLD Fund: 50540) from Queen's University for funding this work.

References

Abdul Rahman, N., Larbi, A., Addah, W., Sulleyman, K. W., Adda, J. K., Kizito, F., and Hoeschle-

573 Zeledon, I.: Optimizing food and feed in maize–livestock systems in northern Ghana: The effect of
 574 maize leaf stripping on grain yield and leaf fodder quality, *Agriculture*, 12, 275,
 575 <https://doi.org/10.3390/agriculture12020275>, 2022.

576 Achina-Obeng, R., and Aram, S. A.: Informal artisanal and small-scale gold mining (ASGM) in Ghana:
 577 Assessing environmental impacts, reasons for engagement, and mitigation strategies, *Resour.*
 578 *Policy*, 78, 102907, <https://doi.org/10.1016/j.resourpol.2022.102907>, 2022.

579 Addai-Arhin, S., Novirsa, R., Jeong, H., Phan, Q. D., Hirota, N., Ishibashi, Y., Shiratsuchi, H., and
 580 Arizono, K.: Mercury waste from artisanal and small-scale gold mining facilities: a risk to farm
 581 ecosystems—a case study of Obuasi, Ghana, *Environ. Sci. Pollut. Res.*, 30, 4293–4308,
 582 <https://doi.org/10.1007/s11356-022-22456-4>, 2023.

583 Adjorlolo-Gasokpoh, A., Golow, A. A., and Kambo-Dorsa, J.: Mercury in the surface soil and cassava,
 584 *Manihot esculenta* (flesh, leaves, and peel) near goldmines at Bogoso and Prestea, Ghana, *Bull.*
 585 *Environ. Contam. Toxicol.*, 89, 1106–1110, <https://doi.org/10.1007/s00128-012-0849-7>, 2012.

586 Afandi, Y., Tejowulan, R. S., and Baiq, D. K.: Mercury uptake by *Zea mays* L. grown on an inceptisol
 587 polluted by amalgamation and cyanidation tailings of small-scale gold mining, *J. Degrad. Min. Lands*
 588 *Manag.*, 6, 1821, <https://doi.org/10.15243/jdmlm.2019.063.1821>, 2019.

589 Ariya, P. A., Amyot, M., Dastoor, A., Deeds, D., Feinberg, A., Kos, G., Poulain, A., Ryjkov, A., Semeniuk,
 590 K., Subir, M., and Toyota, K.: Mercury physicochemical and biogeochemical transformation in the
 591 atmosphere and at atmospheric interfaces: a review and future directions, *Chem. Rev.*, 115, 3760–
 592 3802, <https://doi.org/10.1021/cr500667e>, 2015.

593 Assad, M., Parelle, J., Cazaux, D., Gimbert, F., Chalot, M., and Tatin-Froux, F.: Mercury uptake into
 594 poplar leaves, *Chemosphere*, 146, 1–7, <https://doi.org/10.1016/j.chemosphere.2015.11.103>, 2016.

595 Basu, N., Bastiansz, A., Dórea, J. G., Fujimura, M., Horvat, M., Shroff, E., Weihe, P., and Zastenskaya,
 596 I.: Our evolved understanding of the human health risks of mercury, *Ambio*, 52, 877–896,
 597 <https://doi.org/10.1007/s13280-023-01831-6>, 2023.

598 Basuchaudhuri, P.: *Physiology of the Peanut Plant*, CRC Press, Boca Raton, USA, p. 430,
 599 <https://doi.org/10.1201/9781003262220>, 2022.

600 Bergquist, B. A., and Blum, J. D.: Mass-dependent and -independent fractionation of Hg isotopes by

601 photoreduction in aquatic systems, *Science*, 318, 417–420,
 602 <https://doi.org/10.1126/science.1148050>, 2007.

603 Bergquist, B. A., and Blum, J. D.: The odds and evens of mercury isotopes: applications of mass-
 604 dependent and mass-independent isotope fractionation, *Elements*, 5, 353–357,
 605 <https://doi.org/10.2113/gselements.5.6.353>, 2009.

606 Biester, H., and Scholz, C.: Determination of mercury binding forms in contaminated soils: mercury
 607 pyrolysis versus sequential extractions, *Environ. Sci. Technol.*, 31, 233–239,
 608 <https://doi.org/10.1021/es960369h>, 1996.

609 Blum, J. D., and Bergquist, B. A.: Reporting of variations in the natural isotopic composition of
 610 mercury, *Anal. Bioanal. Chem.*, 388, 353–359, <https://doi.org/10.1007/s00216-007-1236-9>, 2007.

611 Bose-O'Reilly, S., Drasch, G., Beinhoff, C., Rodrigues-Filho, S., Roeder, G., Lettmeier, B., Maydl, A.,
 612 Maydl, S., and Siebert, U.: Health assessment of artisanal gold miners in Indonesia, *Sci. Total*
 613 *Environ.*, 408, 713–725, <https://doi.org/10.1016/j.scitotenv.2009.10.070>, 2009.

614 Bräutigam, A., and Gowik, U.: Photorespiration connects C3 and C4 photosynthesis, *J. Exp. Bot.*, 67,
 615 2953–2962, <https://doi.org/10.1093/jxb/erw056>, 2016.

616 Browne, C. L., and Fang, S. C.: Differential uptake of mercury vapor by gramineous C3 and C4 plants,
 617 *Plant Physiol.*, 72, 1040–1042, <https://doi.org/10.1104/pp.72.4.1040>, 1983.

618 Bugmann, A., Brugger, F., Zongo, T., and Van der Merwe, A.: “Doing ASGM without mercury is like
 619 trying to make omelets without eggs”: Understanding the persistence of mercury use among
 620 artisanal gold miners in Burkina Faso, *Environ. Sci. Policy*, 133, 87–97,
 621 <https://doi.org/10.1016/j.envsci.2022.03.009>, 2022.

622 Casagrande, G. C. R., Franco, D. N. D. M., Moreno, M. I. C., de Andrade, E. A., Battirola, L. D., and de
 623 Andrade, R. L. T.: Assessment of atmospheric mercury deposition in the vicinity of artisanal and
 624 small-scale gold mines using Glycine max as bioindicators, *Water Air Soil Pollut.*, 231, 1–14,
 625 <https://doi.org/10.1007/s11270-020-04918-y>, 2020.

626 Castilhos, Z. C., and Domingos, L. M. B.: A picture of artisanal and small-scale gold mining (ASGM)
 627 in Brazil and its mercury emissions and releases, *Environ. Geochem. Health*, 46, 101,
 628 <https://doi.org/10.1007/s10653-024-01881-z>, 2024.

629 Cobbina, S. J., Duwiejuah, A. B., and Quainoo, A. K.: Single and simultaneous adsorption of heavy
630 metals onto groundnut shell biochar produced under fast and slow pyrolysis, *Int. J. Environ. Sci.*
631 *Technol.*, 16, 3081–3090, <https://doi.org/10.1007/s13762-018-1910-9>, 2019.

632 Crout, N. M. J., Beresford, N. A., Dawson, J. M., Soar, J., and Mayes, R. W.: The transfer of ⁷³As, ¹⁰⁹Cd
633 and ²⁰³Hg to the milk and tissues of dairy cattle, *J. Agr. Sci.*, 142, 203–212,
634 <https://doi.org/10.1017/S0021859604004186>, 2004.

635 Cui, L., Feng, X., Lin, C. J., Wang, X., Meng, B., Wang, X., and Wang, H.: Accumulation and
636 translocation of ¹⁹⁸Hg in four crop species, *Environ. Toxicol. Chem.*, 33, 334–340,
637 <https://doi.org/10.1002/etc.2443>, 2014.

638 Demers, J. D., Blum, J. D., and Zak, D. R.: Mercury isotopes in a forested ecosystem: Implications for
639 air-surface exchange dynamics and the global mercury cycle, *Global Biogeochem. Cy.*, 27, 222–238,
640 <https://doi.org/10.1002/gbc.20021>, 2013.

641 El-Sharkawy, M. A.: Cassava biology and physiology, *Plant Mol. Biol.*, 53, 621–641,
642 <https://doi.org/10.1023/B:PLAN.0000019109.01740.c6>, 2003.

643 El-Sharkawy, M. A. and Cock, J. H.: C3-C4 intermediate photosynthetic characteristics of cassava
644 (*Manihot esculenta* Crantz) I. Gas exchange, *Photosynth. Res.*, 12, 219–235,
645 <https://doi.org/10.1007/BF00055122>, 1987.

646 Enrico, M., Roux, G. L., Maruszczak, N., Heimbürger, L., Claustres, A., Fu, X., Sun, R., and Sonke, J. E.:
647 Atmospheric mercury transfer to peat bogs dominated by gaseous elemental mercury dry
648 deposition, *Environ. Sci. Technol.*, 50, 2405–2412, <https://doi.org/10.1021/acs.est.5b06058>, 2016.

649 Enrico, M., Balcom, P., Johnston, D. T., Foriel, J., and Sunderland, E. M.: Simultaneous combustion
650 preparation for mercury isotope analysis and detection of total mercury using a direct mercury
651 analyzer, *Anal. Chim. Acta*, 1154, 338327, <https://doi.org/10.1016/j.aca.2021.338327>, 2021.

652 Estrade, N., Carignan, J., and Donard, O. F.: Tracing and quantifying anthropogenic mercury sources
653 in soils of northern France using isotopic signatures, *Environ. Sci. Technol.*, 45, 1235–1242,
654 <https://doi.org/10.1021/es1026823>, 2011.

655 Fu, X., Zhu, W., Zhang, H., Sommar, J., Yu, B., Yang, X., Wang, X., Lin, C., and Feng, X.: Depletion of
656 atmospheric gaseous elemental mercury by plant uptake at Mt. Changbai, Northeast China, *Atmos.*

657 Chem. Phys., 16, 12861–12873, <https://doi.org/10.5194/acp-16-12861-2016>, 2016.

658 Fu, X., Liu, C., Zhang, H., Xu, Y., Zhang, H., Li, J., Lyu, X., Zhang, G., Guo, H., Wang, X., Zhang, L., and
 659 Feng, X.: Isotopic compositions of atmospheric total gaseous mercury in 10 Chinese cities and
 660 implications for land surface emissions, Atmos. Chem. Phys., 21, 6721–6734,
 661 <https://doi.org/10.5194/acp-21-6721-2021>, 2021.

662 Gerson, J. R., Szponar, N., Zambrano, A. A., Bergquist, B., Broadbent, E., Driscoll, C. T., Erkenwick,
 663 G., Evers, D. C., Fernandez, L. E., Hsu-Kim, H., Inga, G., Lansdale, K. N., Marchese, M. J., Martinez,
 664 A., Moore, C., Pan, W. K., Purizaca, R. P., Sánchez, V., Silman, M., Ury, E. A., Vega, C., Watsa, M., and
 665 Bernhardt, E. S.: Amazon forests capture high levels of atmospheric mercury pollution from artisanal
 666 gold mining, Nat. Commun., 13, <https://doi.org/10.1038/s41467-022-27997-3>, 2022.

667 Glauser, E., Wohlgemuth, L., Conen, F., and Jiskra, M.: Total mercury accumulation in aboveground
 668 parts of maize plants (*Zea mays*) throughout a growing season, J. Plant Interact., 17, 239–243,
 669 <https://doi.org/10.1080/17429145.2022.2028914>, 2022.

670 González-Carrasco, V., Velasquez-Lopez, P. C., Olivero-Verbel, J., and Pájaro-Castro, N.: Air mercury
 671 contamination in the gold mining town of Portovelo, Ecuador, Bull. Environ. Contam. Toxicol., 87,
 672 250–253, <https://doi.org/10.1007/s00128-011-0345-5>, 2011.

673 Gratz, L. E., Keeler, G. J., Blum, J. D., and Sherman, L. S.: Isotopic composition and fractionation of
 674 mercury in Great Lakes precipitation and ambient air, Environ. Sci. Technol., 44, 7764–7770,
 675 <https://doi.org/10.1021/es100383w>, 2010.

676 Greger, M., Wang, Y., and Neuschütz, C.: Absence of Hg transpiration by shoot after Hg uptake by
 677 roots of six terrestrial plant species, Environ. Pollut., 134, 201–208,
 678 <https://doi.org/10.1016/j.envpol.2004.08.007>, 2005.

679 Grigg, A.R., Kretzschmar, R., Gilli, R.S. and Wiederhold, J.G., 2018. Mercury isotope signatures of
 680 digests and sequential extracts from industrially contaminated soils and sediments. *Science of the*
 681 *Total Environment*, 636, pp.1344-1354. <https://doi.org/10.1016/j.scitotenv.2018.04.261>

682 Ha, E., Basu, N., Bose-O'Reilly, S., Dórea, J. G., McSorley, E., Sakamoto, M., and Chan, H. M.: Current
 683 progress on understanding the impact of Mercury on human health, Environ. Res., 152, 419–433,
 684 <https://doi.org/10.1016/j.envres.2016.06.042>, 2017.

685 Hinton, J., Veiga, M. M., and Veiga, A.: Clean artisanal gold mining: a utopian approach?, *J. Clean.*
686 *Prod.*, 11, 99–115, [https://doi.org/10.1016/s0959-6526\(02\)00031-8](https://doi.org/10.1016/s0959-6526(02)00031-8), 2003.

687 Jiskra, M., Wiederhold, J. G., Skyllberg, U., Kronberg, R. M., Hajdas, I., and Kretzschmar, R.: Mercury
688 deposition and re-emission pathways in boreal forest soils investigated with Hg isotope signatures,
689 *Environ. Sci. Technol.*, 49, 7188–7196, <https://doi.org/10.1021/acs.est.5b00742>, 2015.

690 Jiskra, M., Sonke, J. E., Obrist, D., Bieser, J., Ebinghaus, R., Myhre, C. L., Pfaffhuber, K. A., Wängberg,
691 I., Kyllönen, K., Worthy, D., Martin, L. G., Labuschagne, C., Mkololo, T., Ramonet, M., Magand, O., and
692 Dommergue, A.: A vegetation control on seasonal variations in global atmospheric mercury
693 concentrations, *Nat. Geosci.*, 11, 244–250, <https://doi.org/10.1038/s41561-018-0078-8>, 2018.

694 Jiskra, M., Heimbürger-Boavida, L.E., Desgranges, M.M., Petrova, M.V., Dufour, A., Ferreira-Araujo, B.,
695 Masbou, J., Chmeleff, J., Thyssen, M., Point, D. and Sonke, J.E.: Mercury stable isotopes constrain
696 atmospheric sources to the ocean. *Nature*, 597(7878), 678-682. [https://doi.org/10.1038/s41586-](https://doi.org/10.1038/s41586-021-03859-8)
697 [021-03859-8](https://doi.org/10.1038/s41586-021-03859-8), 2021.

698 Jönsson, J. B., Charles, E., and Kalvig, P.: Toxic mercury versus appropriate technology: Artisanal gold
699 miners’ retort aversion, *Resour. Policy*, 38, 60–67, <https://doi.org/10.1016/j.resourpol.2012.09.001>,
700 2013.

701 Kawakami, T., Konishi, M., Imai, Y., and Soe, P. S.: Diffusion of mercury from artisanal small-scale
702 gold mining (ASGM) sites in Myanmar, *GEOMATE J.*, 17, 228–235,
703 <https://doi.org/10.21660/2019.61.4823>, 2019.

704 Laacouri, A., Nater, E. A., and Kolka, R. K.: Distribution and uptake dynamics of mercury in leaves of
705 common deciduous tree species in Minnesota, USA, *Environ. Sci. Technol.*, 47, 10462–10470,
706 <https://doi.org/10.1021/es401357z>, 2013.

707 Latif, S., and Müller, J.: Potential of cassava leaves in human nutrition: A review, *Trends Food Sci.*
708 *Technol.*, 44, 147–158, <https://doi.org/10.1016/j.tifs.2015.04.006>, 2015.

709 Lewis, D., McNeill, R., and Shabalala, Z.: Gold worth billions smuggled out of Africa, Reuters,
710 <https://www.reuters.com/article/us-gold-africa-smuggling-exclusive-idUSKCN1S00V4> (last access:
711 11 November 2023), 2019.

712 Liu, Y., Lin, C. J., Yuan, W., Lu, Z., and Feng, X.: Translocation and distribution of mercury in biomasses

713 from subtropical forest ecosystems: Evidence from stable mercury isotopes, *Acta Geochim.*, 40, 42–
 714 50, <https://doi.org/10.1007/s11631-020-00441-3>, 2021.

715 Liu, Y., Sun, X., and Li, B.: Adsorption of Hg^{2+} and Cd^{2+} by ethylenediamine modified peanut shells,
 716 *Carbohydr. Polym.*, 81, 335–339, <https://doi.org/10.1016/j.carbpol.2010.02.020>, 2010.

717 Lomonte, C., Wang, Y., Doronila, A., Gregory, D., Baker, A. J., Siegele, R., and Kolev, S. D.: Study of the
 718 spatial distribution of mercury in roots of vetiver grass (*Chrysopogon zizanioides*) by micro-PIXE
 719 spectrometry, *Int. J. Phytoremediat.*, 16, 1170–1182,
 720 <https://doi.org/10.1080/15226514.2013.821453>, 2014.

721 Mao, Y., Li, Y., Richards, J., and Cai, Y.: Investigating uptake and translocation of mercury species by
 722 sawgrass (*Cladium jamaicense*) using a stable isotope tracer technique, *Environ. Sci. Technol.*, 47,
 723 9678–9684, <https://doi.org/10.1021/es400546s>, 2013.

724 Marshall, B., Camacho, A. A., Jimenez, G., and Veiga, M. M.: Mercury challenges in Mexico:
 725 regulatory, trade and environmental impacts, *Atmosphere*, 12, 57,
 726 <https://doi.org/10.3390/atmos12010057>, 2020.

727 Mashyanov, N. R., Pogarev, S. E., Panova, E. G., Panichev, N., and Ryzhov, V.: Determination of
 728 mercury thermospecies in coal, *Fuel*, 203, 973–980, <https://doi.org/10.1016/j.fuel.2017.03.085>,
 729 2017.

730 McLagan, D. S., Mitchell, C. P., Huang, H., Lei, Y. D., Cole, A. S., Steffen, A., Hung, H., and Wania, F.:
 731 A high-precision passive air sampler for gaseous mercury, *Environ. Sci. Technol. Lett.*, 3, 24–29,
 732 <https://doi.org/10.1021/acs.estlett.5b00319>, 2016

733 McLagan, D. S., Monaci, F., Huang, H., Lei, Y. D., Mitchell, C. P., and Wania, F.: Characterization and
 734 quantification of atmospheric mercury sources using passive air samplers, *J. Geophys. Res.-Atmos.*,
 735 124, 2351–2362, <https://doi.org/10.1029/2018JD029373>, 2019.

736 McLagan, D. S., Biester, H., Navrátil, T., Kraemer, S. M., and Schwab, L.: Internal tree cycling and
 737 atmospheric archiving of mercury: examination with concentration and stable isotope analyses,
 738 *Biogeosciences*, 19, 4415–4429, <https://doi.org/10.5194/bg-19-4415-2022>, 2022a.

739 McLagan, D. S., Schwab, L., Wiederhold, J. G., Chen, L., Pietrucha, J., Kraemer, S. M., and Biester, H.:
 740 Demystifying mercury geochemistry in contaminated soil–groundwater systems with

741 complementary mercury stable isotope, concentration, and speciation analyses, *Environ. Sci.-*
742 *Process Impacts*, 24, 1406–1429, <https://doi.org/10.1039/D1EM00368B>, 2022b.

743 Millhollen, A. G., Gustin, M. S., and Obrist, D.: Foliar mercury accumulation and exchange for three
744 tree species, *Environ. Sci. Technol.*, 40, 6001–6006, <https://doi.org/10.1021/es0609194>, 2006.

745 Mitchell, C. P., and Gilmour, C. C.: Methylmercury production in a Chesapeake Bay salt marsh, *J.*
746 *Geophys. Res.-Biogeosci.*, 113, G2, <https://doi.org/10.1029/2008JG000765>, 2008.

747 Molina, J. A., Oyarzun, R., and Esbrí, J. M.: Mercury accumulation in soils and plants in the Almadén
748 mining district, Spain: one of the most contaminated sites on Earth, *Environ. Geochem. Health*, 28,
749 487–498, <https://doi.org/10.1007/s10653-006-9058-9>, 2006.

750 Moreno-Brush, M., McLagan, D. S., and Biester, H.: Fate of mercury from artisanal and small-scale
751 gold mining in tropical rivers: Hydrological and biogeochemical controls. A critical review, *Crit. Rev.*
752 *Environ. Sci. Technol.*, 50, 437–475, <https://doi.org/10.1080/10643389.2019.1629793>, 2020.

753 Munthe, J., Kindbom, K., Parsmo, R., and Yaramenka, K.: Technical Background Report to the Global
754 Mercury Assessment 2018, United Nations Environmental Programme (UNEP), Kenya,
755 [https://www.unep.org/globalmercurypartnership/resources/report/technical-background-report-](https://www.unep.org/globalmercurypartnership/resources/report/technical-background-report-global-mercury-assessment-2018)
756 [global-mercury-assessment-2018](https://www.unep.org/globalmercurypartnership/resources/report/technical-background-report-global-mercury-assessment-2018) (last access: 24 June 2024), 2019.

757 Nakazawa, K., Nagafuchi, O., Kawakami, T., Inoue, T., Elvince, R., Kanefuji, K., Nur, I., Napitupulu, M.,
758 Basir-Cyio, M., Kinoshita, H., and Shinozuka, K.: Human health risk assessment of atmospheric
759 mercury inhalation around three artisanal small-scale gold mining areas in Indonesia, *Environ. Sci.:*
760 *Atmos.*, 1, 423–433, <https://doi.org/10.1039/D0EA00019A>, 2021.

761 Namasivayam, C., and Periasamy, K.: Bicarbonate-treated peanut hull carbon for mercury (II)
762 removal from aqueous solution, *Water Res.*, 27, 1663–1668, [https://doi.org/10.1016/0043-](https://doi.org/10.1016/0043-1354(93)90130-A)
763 [1354\(93\)90130-A](https://doi.org/10.1016/0043-1354(93)90130-A), 1993.

764 Niu, Z., Zhang, X., Wang, Z., and Ci, Z.: Field controlled experiments of mercury accumulation in
765 crops from air and soil, *Environ. Pollut.*, 159, 2684–2689,
766 <https://doi.org/10.1016/j.envpol.2011.05.029>, 2011.

767 Nyanza, E. C., Dewey, D., Thomas, D. S., Davey, M., and Ngallaba, S. E.: Spatial distribution of
768 mercury and arsenic levels in water, soil and cassava plants in a community with long history of gold

769 mining in Tanzania, *Bull. Environ. Contam. Toxicol.*, 93, 716–721, [https://doi.org/10.1007/s00128-](https://doi.org/10.1007/s00128-014-1315-5)
770 014-1315-5, 2014.

771 Obrist, D., Agnan, Y., Jiskra, M., Olson, C. L., Colegrove, D. P., Hueber, J., Moore, C. W., Sonke, J. E.,
772 and Helmig, D.: Tundra uptake of atmospheric elemental mercury drives Arctic mercury pollution,
773 *Nature*, 547, 201–204, <https://doi.org/10.1038/nature22997>, 2017.

774 Obrist, D., Roy, E. M., Harrison, J. L., Kwong, C. F., Munger, J. W., Moosmüller, H., Romero, C. D., Sun,
775 S., Zhou, J., and Commane, R.: Previously unaccounted atmospheric mercury deposition in a
776 midlatitude deciduous forest, *P. Natl. Acad. Sci. USA*, 118, e2105477118,
777 <https://doi.org/10.1073/pnas.2105477118>, 2021.

778 Odukoya, A. M., Uruowhe, B., Watts, M. J., Hamilton, E. M., Marriott, A. L., Alo, B., and Anene, N. C.:
779 Assessment of bioaccessibility and health risk of mercury within soil of artisanal gold mine sites,
780 Niger, North-central part of Nigeria, *Environ. Geochem. Health*, 44, 893–909,
781 <https://doi.org/10.1007/s10653-021-00991-2>, 2022.

782 PlanetGOLD: 2022 Global Forum on Artisanal & Small-Scale Gold Mining,
783 <https://www.planetgold.org/2022-global-forum-artisanal-small-scale-gold-mining>, last access: 21
784 Oct 2024.

785 Qiu, G., Feng, X., Li, P., Wang, S., Li, G., Shang, L., and Fu, X.: Methylmercury accumulation in rice
786 (*Oryza sativa* L.) grown at abandoned mercury mines in Guizhou, China, *J. Agric. Food Chem.*, 56,
787 2465–2468, <https://doi.org/10.1021/jf073391a>, 2008.

788 Rea, A. W., Lindberg, S. E., and Keeler, G. J.: Assessment of dry deposition and foliar leaching of
789 mercury and selected trace elements based on washed foliar and surrogate surfaces, *Environ. Sci.*
790 *Technol.*, 34, 2418–2425, <https://doi.org/10.1021/es991305k>, 2000.

791 Rees, D., Westby, A., Tomlins, K., Van Oirschot, Q., Cheema, M. U., Cornelius, E., and Amjad, M.:
792 Tropical root crops, in: *Crop Post-Harvest: Science and Technology: Perishables*, edited by: Rees, D.,
793 Farrell, G., and Orchard, J., Wiley-Blackwell Publishing, Susse, UK, 392–413,
794 <https://doi.org/10.1002/9781444354652.ch18>, 2012.

795 Rose, C. H., Ghosh, S., Blum, J. D., and Bergquist, B. A.: Effects of ultraviolet radiation on mercury
796 isotope fractionation during photo-reduction for inorganic and organic mercury species, *Chem.*
797 *Geol.*, 405, 102–111, <https://doi.org/10.1016/j.chemgeo.2015.02.025>, 2015.

798 Rutter, A. P., Schauer, J. J., Shafer, M. M., Creswell, J. E., Olson, M. R., Robinson, M., Collins, R. M.,
799 Parman, A. M., Katzman, T. L., and Mallek, J. L.: Dry deposition of gaseous elemental mercury to
800 plants and soils using mercury stable isotopes in a controlled environment, *Atmos. Environ.*, 45, 848–
801 855, <https://doi.org/10.1016/j.atmosenv.2010.11.025>, 2011a.

802 Rutter, A. P., Schauer, J. J., Shafer, M. M., Creswell, J., Olson, M. R., Clary, A., Robinson, M., Parman,
803 A. M., and Katzman, T. L.: Climate sensitivity of gaseous elemental mercury dry deposition to plants:
804 Impacts of temperature, light intensity, and plant species, *Environ. Sci. Technol.*, 45, 569–575,
805 <https://doi.org/10.1021/es102687b>, 2011b.

806 Samkol, P.: Groundnut foliage as feed for Cambodian cattle, PhD thesis, *Acta Univ. Agric. Sueciae*,
807 201:26, 1–50, <https://core.ac.uk/download/pdf/211564208.pdf> (last access: 21 Dec 2024), 2018.

808 Seccatore, J., Veiga, M., Origliasso, C., Marin, T., and De Tomi, G.: An estimation of the artisanal small-
809 scale production of gold in the world, *Sci. Total Environ.*, 496, 662–667,
810 <https://doi.org/10.1016/j.scitotenv.2014.05.003>, 2014.

811 Si, M., McLagan, D. S., Mazot, A., Szponar, N., Bergquist, B., Lei, Y. D., Mitchell, C. P. J., and Wania, F.:
812 Measurement of atmospheric mercury over volcanic and fumarolic regions on the North Island of
813 New Zealand using passive air samplers, *ACS Earth Space Chem.*, 4, 2435–2443,
814 <https://doi.org/10.1021/acsearthspacechem.0c00274>, 2020.

815 Snow, M. A., Darko, G., Gyamfi, O., Ansah, E., Breivik, K., Hoang, C., Lei, Y. D., and Wania, F.:
816 Characterization of inhalation exposure to gaseous elemental mercury during artisanal gold mining
817 and e-waste recycling through combined stationary and personal passive sampling, *Environ. Sci.:*
818 *Process. Impacts*, 23, 569–579, <https://doi.org/10.1039/d0em00494d>, 2021.

819 Sonke, J. E., Schäfer, J., Chmieleff, J., Audry, S., Blanc, G., and Dupré, B.: Sedimentary mercury stable
820 isotope records of atmospheric and riverine pollution from two major European heavy metal
821 refineries, *Chem. Geol.*, 279, 90–100, <https://doi.org/10.1016/j.chemgeo.2010.09.017>, 2010.

822 Sprovieri, F., Pirrone, N., Bencardino, M., D’Amore, F., Carbone, F., Cinnirella, S., Mannarino, V.,
823 Landis, M., Ebinghaus, R., Weigelt, A., Brunke, E., Labuschagne, C., Martin, L., Munthe, J., Wängberg,
824 I., Artaxo, P., Morais, F., De Melo Jorge Barbosa, H., Brito, J., Cairns, W., Barbante, C., del Carmen
825 Diéguez, M., Garcia, P. E., Dommergue, A., Angot, H., Magand, O., Skov, H., Horvat, M., Kotnik, J.,
826 Read, K. A., Mendes Neves, L., Gawlik, B. M., Sena, F., Mashyanov, N., Obolkin, V., Wip, D., Feng, X.

827 B., Zhang, H., Fu, X., Ramachandran, R., Cossa, D., Knoery, J., Maruszczak, N., Nerentorp, M., and
828 Norstrom, C.: Atmospheric mercury concentrations observed at ground-based monitoring sites
829 globally distributed in the framework of the GMOS network, *Atmos. Chem. Phys.*, 16, 11915-11935,
830 <https://doi.org/10.5194/acp-16-11915-2016>, 2016.

831 Streets, D. G., Horowitz, H. M., Lü, Z., Levin, L., Thackray, C. P., and Sunderland, E. M.: Global and
832 regional trends in mercury emissions and concentrations, 2010–2015, *Atmos. Environ.*, 201, 417-
833 427, <https://doi.org/10.1016/j.atmosenv.2018.12.031>, 2019.

834 Suhadi, S., Sueb, S., Muliya, B. K., and Ashoffi, A. M.: Pollution of mercury and cyanide in soils and
835 plants surrounding the Artisanal and Small-Scale Gold Mining (ASGM) at Sekotong District, West
836 Lombok, West Nusa Tenggara, *Biol. Environ. Pollut.*, 1, 30-37,
837 <https://doi.org/10.31763/bioenvipo.v1i1.392>, 2021.

838 Sun, G., Feng, X., Yin, R., Zhao, H., Zhang, L., Sommar, J., Li, Z., and Zhang, H.: Corn (*Zea mays* L.): A
839 low methylmercury staple cereal source and an important biospheric sink of atmospheric mercury,
840 and health risk assessment, *Environ. Int.*, 131, 104971,
841 <https://doi.org/10.1016/j.envint.2019.104971>, 2019.

842 Sun, T., Wang, Z., Zhang, X., Niu, Z., and Chen, J.: Influences of high-level atmospheric gaseous
843 elemental mercury on methylmercury accumulation in maize (*Zea mays* L.), *Environ. Pollut.*, 265,
844 114890, <https://doi.org/10.1016/j.envpol.2020.114890>, 2020.

845 Szponar, N., McLagan, D. S., Kaplan, R. J., Mitchell, C. P. J., Wania, F., Steffen, A., Stupple, G. W.,
846 Monaci, F., and Bergquist, B. A.: Isotopic characterization of atmospheric gaseous elemental
847 mercury by passive air sampling, *Environ. Sci. Technol.*, 54, 10533-10543,
848 <https://doi.org/10.1021/acs.est.0c02251>, 2020.

849 Szponar, N., Su, Y., Stupple, G., McLagan, D. S., Pilote, M., Munoz, A., Mitchell, C. P. J., Steffen, A.,
850 Wania, F., and Bergquist, B. A.: Applying passive air sampling and isotopic characterization to assess
851 spatial variability of gaseous elemental mercury across Ontario, Canada, *J. Geophys. Res.-Atmos.*,
852 128, e2022JD037361, <https://doi.org/10.1029/2022JD037361>, 2023.

853 Szponar, N., Vega, C. M., Gerson, J. R., McLagan, D. S., Pillaca, M., Antoni, S., Lee, D., Rahman, N.,
854 Fernandez, L., Bernhardt, E., Kiefer, A., Mitchell, C. P. J., Wania, F., and Bergquist, B.: Tracing
855 atmospheric mercury from artisanal and small-scale gold mining, *Environ. Sci. Technol.*, 59, 5021-

5033, <https://doi.org/10.1021/acs.est.4c10521>, 2025.

Tang, X., Wang, Y., Ding, C., Yin, Y., Zhou, Z., Zhang, T., and Wang, X.: Cadmium found in peanut (*Arachis hypogaea* L.) kernels mainly originates from root uptake rather than shell absorption from soil, *Pedosphere*, 34, 726-735, <https://doi.org/10.1016/j.pedsph.2023.05.009>, 2024.

Telmer, K. H. and Veiga, M. M.: World emissions of mercury from artisanal and small-scale gold mining, in: *Mercury Fate and Transport in the Global Atmosphere: Emissions, Measurements and Models*, Springer, Boston, USA, 131-172, https://doi.org/10.1007/978-0-387-93958-2_6, 2009.

Turgeon, R.: Phloem loading: how leaves gain their independence, *BioScience*, 56, 15-24, [https://doi.org/10.1641/0006-3568\(2006\)056\[0015:PLHLGT\]2.0.CO;2](https://doi.org/10.1641/0006-3568(2006)056[0015:PLHLGT]2.0.CO;2), 2006.

UNEP: Minamata Convention on Mercury, United Nations Environmental Programme, Kenya, NGA, <https://www.unep.org/resources/report/minamata-convention-mercury> (last access: 24 June 2024), 2013.

USEPA: Integrated Risk Information System. Methylmercury (MeHg) (CASRN 22967-92-6), US Environmental Protection Agency, Washington, DC, USA, https://cfpub.epa.gov/ncea/iris/iris_documents/documents/subst/0073_summary.pdf (last access: 17 March 2024), 2001.

USEPA: Method 7473: Mercury in solids and Solutions by Thermal Decomposition, Amalgamation, and Atomic Absorption Spectro-Photometry. Test methods for evaluating solid wastes. Physical/Chemical Methods. SW-846 On-Line, United States Environmental Protection Agency (USEPA), Washington, DC, USA, <https://www.epa.gov/sites/default/files/2015-12/documents/7473.pdf>, 2007.

USEPA: Operating Procedure: Soil Sampling, US Environmental Protection Agency (USEPA), Washington, DC, USA, <https://www.epa.gov/sites/production/files/2015-06/documents/Soil-Sampling.pdf>, 2023.

Vaňková, M., Vieira, A.M.D., Ettler, V., Vaněk, A., Trubač, J., Penížek, V. and Mihaljevič, M., 2024. Tracing anthropogenic mercury in soils from Fe–Hg mining/smeltering area: Isotopic and speciation insights. *Chemosphere*, 357, p.142038. <https://doi.org/10.1016/j.chemosphere.2024.142038>

Veiga, M. M., Maxson, P. A., and Hylander, L. D.: Origin and consumption of mercury in small-scale

884 gold mining, *J. Clean. Prod.*, 14, 436–447, <https://doi.org/10.1016/j.jclepro.2004.08.010>, 2006.

885 Verbrugge, B., and Geenen, S.: The gold commodity frontier: A fresh perspective on change and
886 diversity in the global gold mining economy, *Extr. Ind. Soc.*, 6, 413-423,
887 <https://doi.org/10.1016/j.exis.2018.10.014>, 2019.

888 Verité Inc.: The Nexus of Illegal Gold Mining and Human Trafficking in Global Supply Chains. Lessons
889 from Latin America, Verité Inc. on behalf of Global Initiative Against Transnational Organized Crime,
890 Northampton, USA, [https://globalinitiative.net/wp-content/uploads/2018/01/The-nexus-of-illegal-](https://globalinitiative.net/wp-content/uploads/2018/01/The-nexus-of-illegal-gold-mining.pdf)
891 [gold-mining.pdf](https://globalinitiative.net/wp-content/uploads/2018/01/The-nexus-of-illegal-gold-mining.pdf), 2016.

892 Vreman, K., Van Der Veen, N. G., Van der Molen, E. J., and De Ruig, W. G.: Transfer of cadmium, lead,
893 mercury and arsenic from feed into milk and various tissues of dairy cows: chemical and pathological
894 data, *NJAS*, 34, 129-144, <https://doi.org/10.18174/njas.v34i2.16799>, 1986.

895 Wang, D., Li, Z., and Wang, Q.: Ecological restoration reduces mercury in corn kernel and the
896 distinction of mercury in corn plants in rural China—A case in Wuchuan mercury mining area,
897 *Ecotoxicol. Environ. Saf.*, 271, 115964, <https://doi.org/10.1016/j.ecoenv.2024.115964>, 2024.

898 Wang, X., Yuan, W., Lin, C., Luo, J., Wang, F., Feng, X., Fu, X., and Liu, C.: Underestimated sink of
899 atmospheric mercury in a deglaciated forest chronosequence, *Environ. Sci. Technol.*, 54, 8083-8093,
900 <https://doi.org/10.1021/acs.est.0c01667>, 2020.

901 Weinhouse, C., Gallis, J. A., Ortiz, E., Berky, A. J., Morales, A. M., Diringer, S. E., Harrington, J., Bullins,
902 P., Rogers, L., Hare-Grogg, J., Hsu-Kim, H., and Pan, W. K.: A population-based mercury exposure
903 assessment near an artisanal and small-scale gold mining site in the Peruvian Amazon, *J. Expo. Sci.*
904 *Environ. Epidemiol.*, 31, 126-136, <https://doi.org/10.1038/s41370-020-0234-2>, 2021.

905 WHO: Environmental health criteria 101: Methylmercury, World Health Organization, Geneva, CHE,
906 https://apps.who.int/iris/bitstream/handle/10665/38082/9241571012_eng.pdf, 1990.

907 WHO: Elemental mercury and inorganic mercury compounds: Human health aspects (Concise
908 International Chemical Assessment Document 50), World Health Organization (WHO), Geneva,
909 CHE, [https://www.who.int/publications/i/item/elemental-mercury-and-inorganic-mercury-](https://www.who.int/publications/i/item/elemental-mercury-and-inorganic-mercury-compounds-human-health-aspects)
910 [compounds-human-health-aspects](https://www.who.int/publications/i/item/elemental-mercury-and-inorganic-mercury-compounds-human-health-aspects), 2003.

911 World Gold Council: Gold mine production, World Gold Council,

912 <https://www.gold.org/goldhub/data/historical-mine-production>, 2019.

913 Xia, Z., Du, Z., Zhou, X., Jiang, S., Zhu, T., Wang, L., Chen, F., Carvalho, L., Zou, M., López-Lavalle, L.
914 a. B., Zhang, X., Xu, L., Wang, Z., Chen, M., Feng, B., Wang, S., Li, M., Li, Y., Wang, H., Liu, S., Bao, Y.,
915 Zhao, L., Zhang, C., Xiao, J., Guo, F., Shen, X., Lu, C., Qiao, F., Ceballos, H., Yan, H., Zhang, H., He, S.,
916 Zhao, W., Wan, Y., Chen, Y., Huang, D., Li, K., Liu, B., Peng, M., Zhang, W., Muller, B., Chen, X., Luo,
917 M.C., Xiao, J., and Wang, W.: Pan-genome and Haplotype Map of Cultivars and Their Wild Ancestors
918 Provides Insights into Selective Evolution of Cassava (*Manihot esculenta* Crantz), *bioRxiv*,
919 <https://doi.org/10.1101/2023.07.02.546475>, 2023.

920 Yin, R., Feng, X., and Meng, B.: Stable mercury isotope variation in rice plants (*Oryza sativa* L.) from
921 the Wanshan mercury mining district, SW China, *Environ. Sci. Technol.*, 47, 2238-2245,
922 <https://doi.org/10.1021/es304302a>, 2013.

923 Yoshimura, A., Koyo, S., and Veiga, M. M.: Estimation of mercury losses and gold production by
924 Artisanal and Small-Scale Gold Mining (ASGM), *J. Sustain. Met.*, 7, 1045–1059,
925 <https://doi.org/10.1007/s40831-021-00394-8>, 2021.

926 Yu, B., Fu, X., Yin, R., Zhang, H., Wang, X., Lin, C., Wu, C., Zhang, Y., He, N., Fu, P., Wang, Z., Shang,
927 L., Sommar, J., Sonke, J. E., Maurice, L., Guinot, B., and Feng, X.: Isotopic composition of atmospheric
928 mercury in China: new evidence for sources and transformation processes in air and in vegetation,
929 *Environ. Sci. Technol.*, 50, 9262-9269, <https://doi.org/10.1021/acs.est.6b01782>, 2016.

930 Yuan, W., Wang, X., Lin, C. J., Wu, F., Luo, K., Zhang, H., Lu, Z., and Feng, X.: Mercury uptake,
931 accumulation, and translocation in roots of subtropical forest: implications of global mercury
932 budget, *Environ. Sci. Technol.*, 56, 14154-14165, <https://doi.org/10.1021/acs.est.2c04217>, 2022.

933 Zhang, H., Feng, X., Larssen, T., Qiu, G., and Vogt, R. D.: In inland China, rice, rather than fish, is the
934 major pathway for methylmercury exposure, *Environ. Health Perspect.*, 118, 1183-1188,
935 <https://doi.org/10.1289/ehp.1001915>, 2010.

936 Zhao, L., Anderson, C. W., Qiu, G., Meng, B., Wang, D., and Feng, X.: Mercury methylation in paddy
937 soil: source and distribution of mercury species at a Hg mining area, Guizhou Province, China,
938 *Biogeosciences*, 13, 2429-2440, <https://doi.org/10.5194/bg-13-2429-2016>, 2016.

939 Zhao, H., Yan, H., Zhang, L., Sun, G., Li, P., and Feng, X.: Mercury contents in rice and potential health
940 risks across China, *Environ. Int.*, 126, 406-412, <https://doi.org/10.1016/j.envint.2019.02.055>, 2019.

941 Zhao, L., Meng, B., and Feng, X.: Mercury methylation in rice paddy and accumulation in rice plant: a
942 review, *Ecotoxicol. Environ. Saf.*, 195, 110462, <https://doi.org/10.5194/bg-13-2429-2016>, 2020.

943 Zhou, J., Obrist, D., Dastoor, A., Jiskra, M., and Ryjkov, A.: Vegetation uptake of mercury and impacts
944 on global cycling, *Nat. Rev. Earth Environ.*, 2, 269-284, <https://doi.org/10.1038/s43017-021-00146-y>,
945 2021.

946 Zhou, J., and Obrist, D.: Global mercury assimilation by vegetation, *Environ. Sci. Technol.*, 55, 14245–
947 14257, <https://doi.org/10.1021/acs.est.1c03530>, 2021.