



1 **A new biogeochemical modelling framework (FLaMe v1.0) for lake methane**
2 **emissions on the regional scale: Development and application to the European**
3 **domain**

4 Manon Maisonnier¹, Maoyuan Feng^{1*}, David Bastviken², Sandra Arndt¹, Ronny Lauerwald³, Aidin
5 Jabbari⁴, Goulven Gildas Laruelle¹, Murray D. MacKay⁵, Zeli Tan⁶, Wim Thiery⁷, Pierre Regnier¹

6 ¹Biogeochemistry and Modelling of the Earth System-BGEOSYS, Department of Geoscience,
7 Environment and Society, Université Libre de Bruxelles, Brussels, Belgium

8 ²Linköping University, Department of Thematic Studies, Tema Environmental Change, Sweden

9 ³Université Paris-Saclay, INRAE, AgroParisTech, UMR Ecosys, Palaiseau, France

10 ⁴Environmental Fluid Dynamics Laboratory, Department of Civil Engineering, Queen's University,
11 Kingston, ON, Canada

12 ⁵Science and Technology Branch, Environment and Climate Change Canada, Toronto, M3H5T4,
13 Canada

14 ⁶Pacific Northwest National Laboratory, Richland, WA, USA

15 ⁷Department of Water and Climate, Vrije Universiteit Brussel, Brussels, Belgium

16

17 *Correspondence to Maoyuan Feng (maoyuan.feng@ulb.be)



18 Abstract

19 This study presents a new physical-biogeochemical modelling framework for simulating lake
20 methane (CH₄) emissions at regional scales. The new model, FLaMe v1.0 (Fluxes of **L**ake **M**ethane),
21 rests on an innovative, computationally efficient lake clustering approach that enables the simulation
22 of CH₄ emissions across a large number of lakes. Building on the Canadian Small Lake Model (CSLM)
23 that simulates the lake physics, we develop a suite of biogeochemical modules to simulate transient
24 dynamics of organic Carbon (C), Oxygen (O₂), and CH₄ cycling. We first test the performance of
25 FLaMe by analyzing physical and biogeochemical processes in two representative lakes (an
26 oligotrophic, deep lake driven by cold climate *versus* a trophic, shallow lake driven by warm climate).
27 Next, we evaluate the model by comparing simulated and observed timeseries of CH₄ emissions in
28 four well-surveyed lakes. We then apply FLaMe at the European scale to evaluate simulated diffusive
29 and ebullitive lake CH₄ fluxes against *in-situ* measurements in both boreal and central European
30 regions. Finally, we provide a first assessment of the spatio-temporal variability in CH₄ emissions
31 from European lakes smaller than 1000 km² (n=108407, total area = 1.33x10⁵ km²), indicating a total
32 emission of 0.97±0.23 Tg CH₄ yr⁻¹, with the uncertainty constrained by combining FLaMe and
33 machine learning techniques. Moreover, 30% and 70% of these CH₄ emissions are through diffusive
34 and ebullitive pathways, respectively. Annually averaged CH₄ emission rates per unit lake area during
35 2010–2016 have a South-to-North decreasing gradient, resulting in a mean over the European domain
36 as 7.39 g CH₄ m⁻² yr⁻¹. Our simulations reveal a strong seasonality in European lake CH₄ emissions,
37 with late summer emissions nearly ten times higher than winter values. This pronounced seasonal
38 variation highlights the importance of accounting for the sub-annual variability in CH₄ emissions to
39 accurately constrain regional CH₄ budgets. In the future, FLaMe could be embedded into Earth
40 System Models to investigate the feedback between climate warming and global lake CH₄ emissions.



41 1. Introduction

42 Methane (CH₄) is the second most important greenhouse gas after carbon dioxide (CO₂), with a
43 Global Warming Potential (GWP) per mass ~28 times higher than that of CO₂ over a 100-year horizon
44 (Saunois *et al.*, 2020). Over the last centuries, the atmospheric CH₄ concentration has increased from
45 722 ppb in the pre-industrial period (year 1750) to 1923 ppb in year 2023 (Saunois, *et al.*, 2020;
46 Dlugokencky, 2022; Forster *et al.*, 2024) and this increase is expected to continue in the future. The
47 critical role of CH₄ in global warming has called for the establishment of a comprehensive global
48 CH₄ budget, which embraces both natural and anthropogenic sources (Saunois *et al.*, 2016; 2020;
49 2024). This budget identified inland waters (wetlands, lakes, reservoirs, ponds, rivers, etc.) as an
50 important, yet highly uncertain atmospheric CH₄ source (Jackson *et al.*, 2020, 2024; Saunois, *et al.*,
51 2020, Canadell *et al.*, 2021). Global lake CH₄ emissions, which has been estimated to account for ~5
52 to 20% of total CH₄ emissions (576 TgCH₄ yr⁻¹), are the largest contributors to this inland water
53 source (Jackson *et al.*, 2020; Saunois *et al.*, 2020). However, estimates of its magnitude vary
54 depending on the assessment methods, with discrepancies of up to a factor of four (Saunois *et al.*,
55 2020; DelSontro and John 2018; Rosentreter *et al.*, 2021; Bastviken *et al.*, 2011; Deemer *et al.*, 2016;
56 Johnson *et al.*, 2021; Holgerson and Raymond 2016; Stavert *et al.* 2022). This variability in global
57 estimates also manifests itself at the continental scale. For instance, estimates of European lake
58 methane emissions range from 0.9 to 2.5 Tg CH₄ yr⁻¹ (Petrescu *et al.* 2021, 2023; Lauerwald *et al.*,
59 2023).

60 Observation-based upscaling approaches are highly dependent on the availability and quality of
61 *in-situ* measurements, which are unevenly distributed across the globe and biased towards summer
62 months (Canadell *et al.*, 2021; Johnson *et al.*, 2022). Although the number of CH₄ emission



63 measurements from lakes has increased considerably in recent decades, the two largest current
64 databases together contain only 1081 records from 575 lakes worldwide (Rosentreter *et al.*, 2021;
65 Johnson *et al.*, 2022). This relatively small data compilation is unlikely to capture the full diversity
66 of physical and biogeochemical patterns of >1.4 million lakes worldwide, which vary by morphology,
67 climate, trophic status, and underlying sediment characteristics (Rinta *et al.*, 2017; Bastviken 2004,
68 2022; Deemer and Holgerson 2021; Johnson *et al.*, 2022). Even more critically, the underlying data
69 collection was not designed to capture the inter-annual and decadal variability in CH₄ emissions
70 driven by climate change and nutrient dynamics, hence rendering the decomposition of the total lake
71 CH₄ fluxes into natural and human-induced components challenging (Saunois *et al.*, 2020). Finally,
72 although current instruments and techniques can effectively capture CH₄ fluxes through diffusive
73 (driven by gradients of aqueous CH₄ concentrations) and ebullitive (via gas bubbles in the sediments
74 due to oversaturation) emission pathways, measurements related to lake turnover events (release of
75 previously accumulated CH₄ due to stratification and ice cover) remain highly challenging (Denfeld
76 *et al.*, 2018; Mayr *et al.*, 2020; Zimmermann *et al.*, 2019). These limitations induce large uncertainties
77 in observation-based upscaling methods. In this context, process-based modelling approaches – that
78 rely on detailed representations of lake physical and biogeochemical processes informed and tested
79 with the available observational data – offer complementary strategies to help reduce these
80 uncertainties.

81 Process-based biogeochemical models provide powerful tools to upscale scarce observations,
82 both in space and in time. In combination with the available observational datasets, they can help
83 deliver regional and global estimates of lake CH₄ emissions from daily to decadal timescales, as well
84 as future projections. These mechanistic models can also help identify the drivers such as climate,



85 land-use and atmospheric composition changes responsible for the complex temporal dynamics of
86 lake CH₄ emissions. Over the last decades, several process-based models have thus emerged, e.g.,
87 LAKE 2.0 (Stepanenko *et al.* 2016), bLake4Me (Tan *et al.*, 2015), and ALBM (Tan *et al.*, 2018;
88 2024), to estimate lake CH₄ emissions to the atmosphere. These models explicitly account for the
89 physical and biogeochemical processes that govern lake CH₄ dynamics and resulting emissions. For
90 instance, using ALBM, Zhuang *et al.* (2023) recently estimated that global lakes (larger than 0.1 km²)
91 emit 24.0 ± 8.4 Tg CH₄ yr⁻¹, which is at the lower end of the range reported by Saunois *et al.* (2020)
92 and represents 11% of total global CH₄ emissions from natural sources as estimated from atmospheric
93 inversions. Yet, these process-based models also have limitations that need to be addressed. A central
94 limitation is the omission of lake phytoplankton productivity, which is one of the most reactive
95 organic C sources and thus substrates for CH₄ production. In most of existing models, this key process
96 and the associated microbial degradation of organic C is not simulated explicitly but taken as
97 prescribed model inputs. If phytoplankton productivity and associated contributions of methane
98 substrates can be incorporated in lake CH₄ models, this would allow capturing the impacts of
99 environmental conditions beyond the commonly included direct temperature effects on organic
100 matter decomposition and CH₄ production. Such additional important impacts include feedback of C
101 metabolism on lake oxygen (O₂) cycling along with eutrophication effects on CH₄ emissions (Del
102 Sontro *et al.*, 2018; Rosentreter *et al.*, 2021; Stavert *et al.*, 2022). However, including an explicit
103 description of these processes is challenging, because it requires to account for a suite of key physical
104 and biogeochemical processes controlling the coupled C-O₂-CH₄ cycles while at the same time
105 maintaining model complexity, as well as the needs for observational data and computational costs
106 for regional and global scale applications with tractable bounds. In addition, it also requires the



107 quantification of nutrient inputs from the surrounding catchments, which exert a key control on lake
108 productivity.

109 To tackle with these challenges, we here develop a new process-based model framework of
110 intermediate complexity, FLaMe v1.0 (Fluxes of **L**ake **M**ethane version 1.0, hereafter referred as
111 FLaMe) that couples the C-O₂-CH₄ cycles in lakes using a one-dimensional representation.
112 Specifically, FLaMe builds upon the existing physical lake model CSLM (Canadian Small Lake
113 Model–MacKay, 2012; MacKay *et al.*, 2017), and extends with a novel biogeochemical module that
114 captures the production, oxidation, storage, transport and emissions of CH₄ in/from lakes. Importantly,
115 FLaMe introduces lake primary production and turnover of autochthonous organic carbon as a major
116 driver of lake O₂ and CH₄ dynamics. The coupled, mechanistic lake model is then embedded in a
117 computationally efficient clustering approach that allows for the application of the new, coupled,
118 mechanistic lake model for (i) large parameter/input ensemble runs on regional/global scales for
119 uncertainty assessments, (ii) long-term model projections for the assessment of future climate change
120 and its feedback on the Earth system, (iii) a potential coupling to ESMs in subsequent stages of its
121 development.

122 The structure of this paper is described as follows. In section 2, we provide a general description
123 of the lake model with a focus on a detailed description of the novel biogeochemical modules, as well
124 as the parameter choices and numerical solutions. In section 3, we first test the overall behavior of
125 FLaMe using two representative lakes (an oligotrophic, deep lake driven by cold climate *versus* a
126 trophic, shallow lake driven by warm climate), and then evaluate the simulated temporal variations
127 of CH₄ fluxes against observational data at four well-surveyed lakes in real world. Next, we apply



FLaMe at the European scale and evaluate the results against *in-situ* measurements in boreal and central European lakes compiled by Rinta *et al.* (2017). Finally, we provide a spatio-temporally resolved estimate of CH₄ emissions from European lakes (2010–2016), assess their sensitivity to key model parameters, and constrain their uncertainty range using a machine-learning approach. In section 4, we discuss model limitations and potential directions for further research. Main conclusions and outlooks are drawn in section 5.

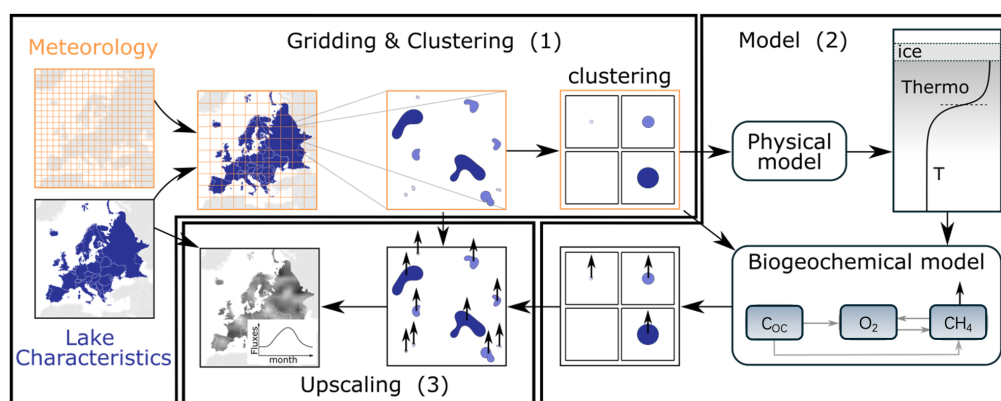
2. Methodology

2.1 General model approach

We developed a new process-based physical-biogeochemical model, FLaMe v1.0 (Fluxes of Lake Methane), to simulate lake CH₄ production and emission at large spatial scales. FLaMe resolves the interplay of physical and biogeochemical processes that governs organic matter ($C_{OC,auto}$), oxygen (O₂), and methane (CH₄) dynamics to estimate (diffusive and ebullitive) lake CH₄ emissions, as well as CH₄ storage fluxes due to lake turnover and ice melting. To enable a continental-scale application of FLaMe (e.g., in Europe, $n=108407$ and total area = 1.33×10^5 km² for lakes with $0.1 \leq A_0 \leq 1000$ km² according to Messenger *et al.*, 2016; A_0 is the lake surface area), we here propose a lake clustering strategy to reduce the computational and data/input costs (Fig. 1) while resolving the variability in lake sizes, morphology, and trophic status as well as climate conditions across Europe. Within each grid cell ($2.5^\circ \times 2.5^\circ$), lakes are binned into four classes according to surface area (0.1–1 km², 1–10 km², 10–100 km², 100–1000 km²). We then run a FLaMe set-up for one representative lake per size class within each grid cell, using the arithmetic means of lake area, depth and trophic status of all lakes pertaining to the respective size class across the respective grid cell. The total emission flux



149 from a given size class can be obtained by multiplying the CH₄ emission rates simulated by FLame
150 with the total lake area of this size class (Fig. 1).



151
152 **Fig. 1. Illustration of the lake clustering and upscaling strategy for the continental application of FLame (Europe**
153 **as an example). (1) Gridding and clustering:** The European domain was divided into grid cells at a coarse spatial
154 **resolution of 2.5°×2.5°. Within each grid cell, lakes are clustered into four classes according to their surface areas.**
155 **(2) FLame parallelization:** FLame simulates the lake metabolic dynamics, vertically resolved concentration and
156 **rate profiles of the coupled O₂-CH₄ cycle as well as diffusive and ebullitive CH₄ fluxes through the water-air**
157 **interface. The model was parallelized under transient conditions for each grid cell and each lake class. (3)**
158 **Upscaling:** The areal rates (i.e., fluxes per unit lake surface area) simulated by FLame were then multiplied by the
159 **total surface area of each lake class within each grid cell (available from HydroLAKES) and aggregated at the**
160 **monthly timescale. The arrows pertaining to clustered and original lakes represent the CH₄ emissions.**

161 2.2 Model description

162 FLame builds on an online coupling approach between the Canadian Small Lake Model (CSLM;
163 MacKay, 2012; MacKay *et al.*, 2017) – a widely used lake physics model (Garnaud *et al.*, 2022;
164 Verseghe and MacKay, 2017; William *et al.*, 2014) – and a set of newly developed biogeochemical



modules that resolve lake OC, O₂ and CH₄ dynamics. CSLM simulates the physical variables of temperature profile (T), thermocline depth (h_{mix}), photic depth (h_{phot}), and ice cover, which will be used to force the biogeochemical modules (Fig. 2). In turn, the biogeochemical module will later modify the photic depth simulated by CSLM to account for the effect of phytoplankton growth and self-shading on light penetration, thus resolving the feedback between lake biogeochemical processes and lake physical dynamics, hence forming a complete feedback loop. A detailed description of the well-established CSLM model can be found in MacKay (2012) and MacKay *et al.* (2017) and is also briefly presented in Supplementary Text S1. In what follows, we provide a detailed description of the vertically resolved 1D model set-up (section 2.2.1) used here, as well as of the novel biogeochemical modules (section 2.2.2 and Table 1 for selected model parameter values).

2.2.1 Model Scope: Idealized representation of lake morphology

Figure 2 illustrates the vertically resolved, one-dimensional model set-up of FLaMe that is used for both the physical and biogeochemical modules. The original version of CSLM usually adopts a “bucket” shaped morphology which assumes a constant area (A) versus water depth (z), i.e., $A(z) = A_0$, where A_0 is the lake surface area at $z = 0$ m. However, this morphology is unsuitable for the simulation of biogeochemical processes, especially when the variations in water depth within each lake are important. Therefore, we, instead, adopted a “valley” shaped lake morphology, with lake area $A(z)$ given by:

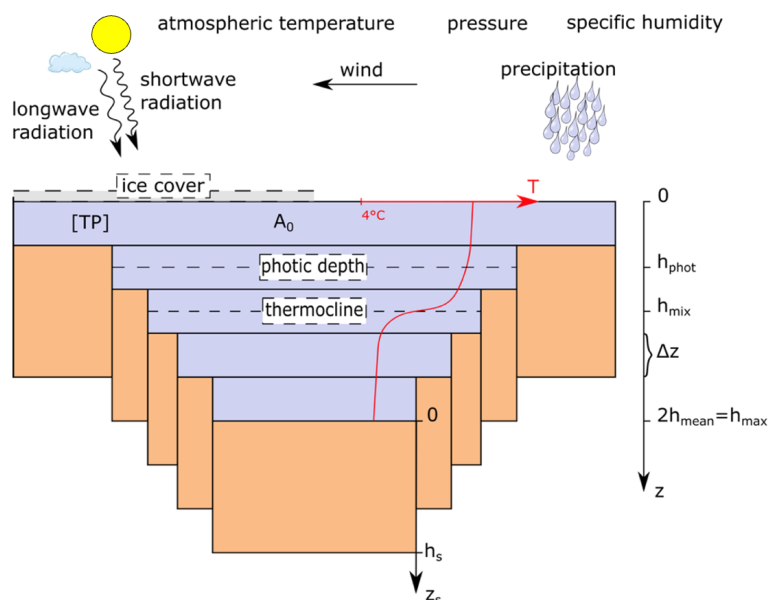
$$A(z) = \frac{A_0}{2s} (s + \arctanh((1 - 2(z/h_{\max}))\tanh(s))) \quad (1)$$

where A is the lake area (m²), z is the water depth (m), s is a shape parameter that controls the slope of the lakebed (a larger s indicates a steeper slope), and $h_{\max}$ is the maximum lake depth. To ensure



186 that the volume (and hence heat exchange) is conserved between the “bucket” and “valley” shape set-
187 ups, the maximum depth of the valley-shape lake, $h_{\max}$, must be twice that of the mean depth of the
188 bucket-shape lake, h_{mean} (i.e., $h_{\max} = 2h_{\text{mean}}$), which was extracted from the global HydroLAKES
189 database compiled by Messenger *et al.* (2016). The bottom temperature profiles simulated by CSLM
190 were then extended to the maximal depth of the valley shape lake.

191 Physical processes in the water column are simulated by CSLM, on a one-dimensional, vertically
192 resolved, evenly distributed grid with a grid spacing of 50 cm. Each water layer is connected to a
193 vertically integrated lake sediment column of 5 m depth (h_s , m) (Fig. 2). Since the CH_4 production
194 rate decreases exponentially with sediment depth, it is typically negligible at 5 m within the sediment
195 column, thus ensuring that the total, depth-integrated benthic CH_4 production becomes insensitive to
196 this arbitrary choice (Langenegger *et al.*, 2019). The surface area of each sediment column in contact
197 with the water column is determined by difference in the widths of two adjacent water layers $A(z)$
198 (Eq. (1)). In addition, it should be noted that we assume no **horizontal** material exchanges (O_2 and
199 CH_4) between the sediments and water columns (i.e., across the interface where left and right edges
200 of a water layer touch the sediment box; Fig. 2), while only the vertical exchanges are simulated in
201 this first version of the model (see details in the following sections).



202

203 **Fig. 2. Schematic representation of the lake morphology in FLaMe. The lake is represented by a “valley” shape**
 204 **(denoted by Eq. (1)). A_0 denotes the lake surface area, A is the area of each water layer, and h_{max} is the maximal**
 205 **water column depth. z represents the depth of a water column down to the surface of a sediment column while z_s**
 206 **stands for the depth inside a sediment column ($z_s = 0$ at the sediment water interface). The physical model is forced**
 207 **by longwave and shortwave radiation, near-surface wind, precipitation, atmospheric temperature, pressure, and**
 208 **specific humidity.**

209 2.2.2 Biogeochemical Modules

210 2.2.2.1 Organic carbon module

211 Following the approach of Maavara *et al.* (2017), FLaMe does not resolve the vertical distribution
 212 of labile (i.e., microbial degradable) organic carbon (OC) concentrations ($[C_{OC,auto}]$) produced by in-
 213 lake primary production, but only simulates the temporal dynamics of the volume-integrated
 214 autochthonous OC stock ($\overline{C_{OC,auto}}$, g C) (the overbar here indicates a volume-integrated value). $\overline{C_{OC,auto}}$



215 should be understood as a simple indicator or an overall reflection of the resulting lake trophic status,
216 itself controlled by the combined effects of climate and nutrient loads from the catchment. The
217 allochthonous C inputs delivered from surrounding catchments are more refractory and thus perform
218 as less important substrates for CH₄ production (Guillemette *et al.*, 2017; DelSontro *et al.*, 2018).
219 Thus, the autochthonous primary production is considered as the only labile OC source, neglecting
220 the allochthonous OC contribution in this first version of FLame.

221 The temporal evolution of volume-integrated labile OC stock is determined by the interplay
222 between autochthonous primary production, pelagic and benthic mineralization and burial fluxes
223 (Maavara *et al.*, 2017):

$$224 \quad \frac{\partial \overline{C_{OC,auto}}}{\partial t} = \overline{F_{PP}} - \overline{F_{Min}} - \overline{F_{Bur}} \quad (2)$$

225 where t is time (day), and $\overline{C_{OC,auto}}$ is the volume-integrated OC stock (g C). $\overline{F_{PP}}$, $\overline{F_{Min}}$ and $\overline{F_{Bur}}$ are
226 the volume-integrated primary production, mineralization, and sedimentary burial fluxes (g C d⁻¹),
227 respectively. Following Maavara *et al.* (2017), we assume that autochthonous primary production
228 rates are controlled by the light regime, water temperature, and the lake total dissolved phosphorus
229 (TDP) concentration ([TDP], g P m⁻³) (Reynolds, 2006). The volume-integrated $\overline{F_{PP}}$ can then be
230 expressed using a classical Michaelis-Menten formulation (Mavaara *et al.*, 2017):

$$231 \quad \overline{F_{PP}} = B P_{Chl,max} M(T_{mean}) \frac{[TDP]}{K_{s,P} + [TDP]} V_{phot} \quad (3)$$

232 where B is the phytoplankton biomass (expressed as chlorophyll-a concentration, g Chl-a m⁻³) in the
233 photic zone (Eq. (5)), $P_{Chl,max}$ is the maximum carbon fixation rate per unit of chlorophyll-a (g C (g
234 Chl-a)⁻¹ h⁻¹), M is a dimensionless metabolic correction factor that depends on the mean lake water
235 temperature in photic zone T_{mean} (°C) (see Eq. (4)), $K_{s,P}$ is the half-saturation constant for phosphorus



236 limitation (g P m^{-3}), and V_{phot} is the water volume above the photic depth (m^3). Parameters $P_{Chl,max}$
237 and $K_{s,P}$ are constrained based on published observations (see section 2.3), while the metabolic factor
238 M is given by:

$$239 \quad M(T_{mean}) = \begin{cases} 1, & T_{mean} \geq 28^\circ\text{C} \\ Q_{10,prod}^{\frac{T_{mean}-28}{10}}, & T_{mean} < 28^\circ\text{C} \end{cases} \quad (4)$$

240 where $Q_{10,prod}$ is the temperature sensitivity for primary production, quantifying the increases of the
241 metabolic factor per 10 degree increase in temperature. Surface water phytoplankton biomass, B , is
242 approximated by a function of the photosynthetically active radiation (PAR), which is determined
243 by shortwave radiation and light extinction in the water column:

$$244 \quad B = \left(\frac{1}{k_c}\right) \left(0.75 \left(\frac{PP}{RP}\right) \ln\left(\frac{0.7PAR_0}{0.5PAR_k}\right) \left(\frac{1}{h_{prod}}\right) - (K_{dw} + K_{dp} + K_{dg})\right) \quad (5)$$

245 where k_c is the absorbance of PAR per unit of chlorophyll-a ($\text{m}^2 (\text{g Chl-a})^{-1}$), and PP/RP is the ratio
246 of maximum gross photosynthesis to respiration per unit chlorophyll-a. PAR_0 is the PAR at the lake
247 surface ($\mu\text{mol m}^{-2} \text{s}^{-1}$), determined by the incoming shortwave radiation, as well as the daytime that
248 is specified by lake latitude and phenology (represented by the day of the year). PAR_k is the PAR at
249 the onset of photosaturation ($\mu\text{mol m}^{-2} \text{s}^{-1}$). The productive depth h_{prod} is determined as the
250 maximum of the thermocline and the photic depth simulated by the physical model. K_{dw} , K_{dp} , and
251 K_{dg} represent nonalgal PAR attenuations, due to pure water, inorganic suspended particulate matter,
252 and labile carbon, respectively. Following Lewis (2011), K_{dg} is calculated as a function of $[C_{OC,auto}]$
253 as:

$$254 \quad \ln(K_{dg}) = -4.44 + 1.80\ln([C_{OC,auto}]) - 0.149(\ln([C_{OC,auto}]))^2. \quad (6)$$



Eq. (5) was derived based on the assumption of a balance between production and respiration (Reynolds, 2006; Lewis, 2011). Here we slightly relax this assumption and assume near-equilibrium conditions for given climate conditions at the monthly timescale, allowing us to simulate seasonal variations of primary production and associated biogeochemical processes. Note that this assumption is only used for biogeochemical variables related to primary production, while physical variables simulated by CSLM are resolved at a sub-daily time step.

Following Hanson *et al.* (2011; 2014) and Maavara *et al.* (2019), the volume-integrated mineralization rate is simulated as a function of temperature and labile OC availability:

$$\overline{F_{Min}} = k_{20} \theta^{T_{mean}-20} \overline{C_{OC,auto}} \quad (7)$$

where k_{20} is a first-order rate constant for the mineralization of $\overline{C_{labile}}$ at 20 °C (d^{-1}). T_{mean} is the mean water temperature (°C) in photic zone, and θ is the temperature dependence of mineralization of organic matter (Hanson *et al.*, 2014).

Following Maavara *et al.* (2019), the burial flux $\overline{F_{Bur}}$ is represented by a first order process driven by the labile OC stock $\overline{C_{OC,auto}}$:

$$\overline{F_{Bur}} = k_{bur} \overline{C_{OC,auto}} \quad (8)$$

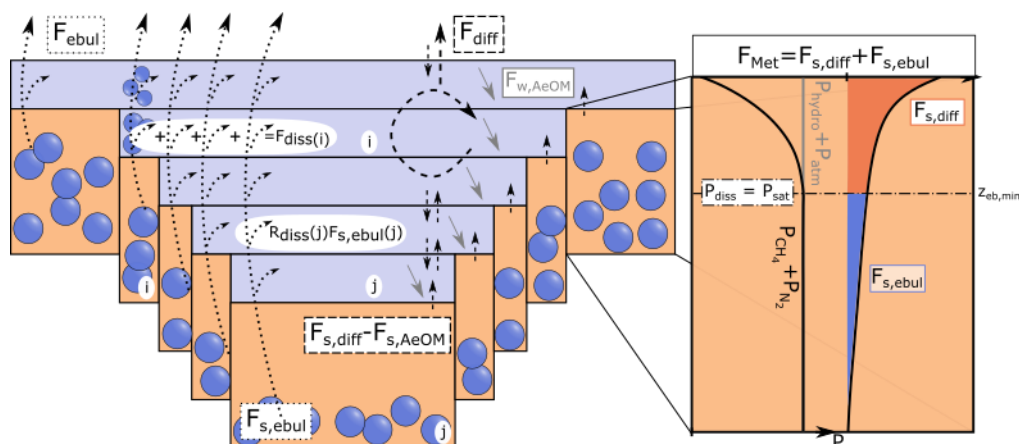
where k_{bur} is the burial rate constant and here set as half of the mineralization rate constant following the ratios between these two processes reported in the global lake dataset ($n=230$) assembled by Mendonça *et al.* (2017). This ratio is likely an upper bound because it accounts for contributions of both autochthonous and allochthonous carbon sources in the dataset, while allochthonous inputs



274 should have distinct (but unquantified) ratios from autochthonous ones (Mendonça *et al.*, 2017;
275 Guillemette *et al.*, 2017).

276 2.2.2.2 Methane module

277 The methane module simulates the dynamics of CH₄ concentration in both sediment and water
278 columns as controlled by benthic CH₄ production, aerobic CH₄ oxidation, and diffusive and
279 ebullitive transport from sediment to water and atmosphere (Fig. 3). Since the observational
280 evidence suggests that benthic CH₄ production is the dominant CH₄ source in lakes (Tan *et al.*, 2015;
281 Bastviken, 2022), we neglect the CH₄ production within the lake's water column in the model.



282
283 **Fig. 3. Illustration of the methane (CH₄) module of FLaMe with a zoom into benthic CH₄ dynamics (zoom modified**
284 **from Langenegger *et al.*, 2019). Benthic CH₄ production (zoom) assumes an exponential decrease in CH₄**
285 **production rate (F_{Met}) with depth. The CH₄ and N₂ partial pressures ($P_{CH_4} + P_{N_2}$) is mainly driven by the CH₄**
286 **production and follows the black curve profile, which starts to exceed the sum of the hydrostatic and atmospheric**
287 **pressure ($P_{hydro} + P_{atm} - P_{H_2O}$, grey line) at $z_{eb,min}$. Thus, this depth ($z_{eb,min}$) divides F_{Met} into a diffusive ($F_{s,diff}$, red**
288 **filled region) and an ebullitive ($F_{s,ebul}$, cyan filled region) flux. $F_{s,AeOM}$ and $F_{w,AeOM}$ are the CH₄ oxidation in the**



289 sediment and water column, respectively. F_{diss} is the dissolution of the gas bubbles during transport through the
290 water column. F_{diff} and F_{ebul} are diffusive and ebullitive CH_4 fluxes through the water-air interface, respectively.

291 Within the lake sediment, CH_4 dynamics are determined by the balance between CH_4 production
292 via methanogenesis and CH_4 migration to the water column through diffusive and ebullitive
293 transport:

$$294 \quad \frac{\partial \widetilde{\text{CH}_{4,s}}(z)}{\partial t} = \widetilde{F_{\text{Met}}}(z) - \widetilde{F_{s,\text{diff}}}(z) - \widetilde{F_{s,\text{ebul}}}(z) \quad (9)$$

$$295 \quad \widetilde{F_{\text{Met}}}(z) = f_{\text{mm}} \frac{M_{\text{CH}_4}}{M_{\text{C}}} \widetilde{F_{\text{Min}}} \frac{V_s(z)}{V_{s,\text{tot}}} \quad (10)$$

296 where the tilde overbar here represents the volume-integrated stocks or fluxes in the sediment
297 column, which is different from the straight overbar for volume-integrated values in the water
298 column. Note that we have sediment columns at different water depths, such that the stocks and
299 fluxes are represented as a function of water depth z , which is characterized by the valley-shape
300 model set-up and different from the conventional bucket shape set-up. $\widetilde{\text{CH}_{4,s}}(z)$ is thus the volume-
301 integrated CH_4 stock for the sediment column with the sediment-water interface positioned at depth
302 z (g CH_4). $\widetilde{F_{\text{Met}}}(z)$ is the volume-integrated flux of CH_4 production through the entire sediment
303 column with a sediment-water interface at depth z (g $\text{CH}_4 \text{ d}^{-1}$), and $\widetilde{F_{s,\text{diff}}}(z)$ and $\widetilde{F_{s,\text{ebul}}}(z)$ are
304 volume-integrated diffusive and ebullition fluxes (g $\text{CH}_4 \text{ d}^{-1}$) through the sediment-water interface
305 at depth z , respectively. f_{mm} denotes the fraction of organic matter mineralization that proceeds via
306 methanogenesis according to data compiled by Hanson *et al.* (2014) and Bastviken (2022). $M_{\text{CH}_4}/M_{\text{C}}$
307 is a conversion factor corresponding to the molar ratio of CH_4 to COC_{auto} . As $f_{\text{mm}} \frac{M_{\text{CH}_4}}{M_{\text{C}}} \widetilde{F_{\text{Min}}}$ is the
308 total CH_4 production flux integrated over the whole water column, we assume that the fractions of



309 CH₄ production occurring in different sediment columns are set according to their volume
310 proportions, i.e., $\frac{V_s(z)}{V_{s,tot}}$.

311 The partitioning of CH₄ production into ebullitive and diffusive fluxes depends on the porewater
312 CH₄ concentration or its partial pressure, which relies mainly on the vertical distribution of CH₄
313 production rate in the sediment as well as the oxygen concentration (but is of second-order
314 importance). Based on the observation-based assumption that the organic carbon concentration and
315 thus mineralization rates exponentially decrease with sediment depth, we here assume an
316 exponentially decreasing relationship between methanogenesis rate versus depth (Fig. 3), following
317 Langenegger *et al.* (2019):

$$318 \quad f_{met}(z, z_s) = F_{Met,0}(z) \exp(-\alpha z_s) \quad (11)$$

319 where $f_{met}(z, z_s)$ is the methanogenesis rate (g CH₄ m⁻³ d⁻¹) at sediment depth z_s for the sediment
320 column with the sediment-water interface positioned at depth z . $F_{Met,0}(z)$ is the maximum CH₄
321 production at the sediment-water interface (g CH₄ m⁻³ d⁻¹) at depth z , and α is a shape parameter
322 (m⁻¹) that controls the decrease of methanogenesis rate with depth. As the shape of this curve
323 typically depends on the flux of labile carbon settling on the lake bottom, and thus, lake trophic
324 status, the parameter α is here scaled by the F_{PP} empirically:

$$325 \quad \alpha = \alpha_{min} + \beta F_{PP} \frac{V_w}{V_{phot}} \quad (12)$$

326 where α_{min} is the minimum or base value, and β is the dependence of α on F_{PP} . The values of α_{min}
327 and β are determined based on the measurements in lakes of different trophic status reported by
328 Langenegger *et al.* (2019).



329 To determine the maximum CH₄ production $F_{Met,0}(z)$, the integral of CH₄ production rate over
330 sediment column should equal to the volume-integrated CH₄ production flux $\widetilde{F_{Met}}(z)$ as specified
331 by Eq. (10):

$$332 \quad A_s(z) \int_0^{h_s} f_{met}(z, z_s) dz_s = \widetilde{F_{Met}}(z) \quad (13)$$

333 where $A_s(z)$ is the surface area of sediment column in contact with the water layer at lake depth z
334 and is determined by difference in the areas of two adjacent water layers $A(z)$ (Eq. (1)). The
335 maximum CH₄ production at depth z , $F_{Met,0}(z)$, can be obtained by combining Equations (10), (11)
336 and (13):

$$337 \quad F_{Met,0}(z) = \frac{\widetilde{F_{Met}}(z)}{A_s(z)} \frac{\alpha}{1 - \exp(-\alpha h_s)} \quad (14)$$

338

339 Since CH₄ production increases the *in-situ* CH₄ concentration as the sediment depth increases,
340 the CH₄ concentration may exceed its solubility concentration and methane gas bubbles may start
341 forming (Fig. 3). To constrain the partitioning of CH₄ production between diffusion and ebullition,
342 the threshold sediment depth, $z_{eb,min}$, at which CH₄ concentration starts to exceed the solubility limit,
343 is determined based on the equilibrium pressure condition following Langenegger *et al.* (2019) (see
344 details in Supplementary Text S2). In the upper portion of the sediment column ($z_s < z_{eb,min}$), the
345 produced CH₄ will diffuse into water; however, a fraction of the diffusing CH₄ will be oxidized in
346 the transit through the upper sediment column, and only the remaining CH₄ will reach the sediment-
347 water interface. The volume-integrated CH₄ oxidation in the sediment at depth z , $\widetilde{F_{s,AeOM}}(z)$, is here
348 assumed to be controlled by the O₂ concentration in the overlying bottom water, and is represented
349 by a Michaelis-Menten function:



$$\widetilde{F_{s,AeOM}}(z) = \widetilde{F_{Met}}(z) \frac{[O_2]_z}{K_{s,O_2} + [O_2]_z} \quad (15)$$

where K_{s,O_2} is the half-saturation constant of O_2 for the sedimentary CH_4 oxidation. As a result, the diffusive flux passing through the sediment-water interface is determined as follows:

$$\widetilde{F_{s,diff}}(z) = A_s(z) \int_0^{z_{eb,min}} F_{Met,0}(z) \exp(-\alpha z_s) dz_s - \widetilde{F_{s,AeOM}}(z) \quad (16)$$

In the lower portion of the sediment column ($z_s > z_{eb,min}$; where oversaturation occurs), the produced CH_4 feeds the ebullitive flux, with the volume-integrated value $\widetilde{F_{s,ebul}}(z)$ ($g CH_4 d^{-1}$) as given by:

$$\widetilde{F_{s,ebul}}(z) = A_s(z) \int_{z_{eb,min}}^{h_s} F_{Met,0}(z) \exp(-\alpha z_s) dz_s \quad (17)$$

Note that Equations. (16) and (17) implicitly imply that, at the monthly resolution of our model, the CH_4 dynamics in the sediment is at steady state and all the CH_4 produced during this time interval is either oxidized or released through the water column via diffusive and ebullitive pathways.

Pelagic, dissolved CH_4 diffuses in the water column and its concentration is determined by the diffusive CH_4 flux passing through the sediment-water interface (acting as a source for each water layer), by aerobic CH_4 oxidation in the water column, and by the re-dissolution of the ebullitive CH_4 fluxes during transit through the water column. The mass conservation equation of dissolved CH_4 is then given by:

$$\frac{\partial [CH_4]_w}{\partial t} = \frac{\partial}{\partial z} \left(K_{diff} \frac{\partial [CH_4]_w}{\partial z} \right) + \widetilde{F_{s,diff}}(z) \frac{1}{A(z) dz} - F_{w,AeOM}(z) + F_{diss}(z) \quad (18)$$



where $[CH_4]_w$ is the pelagic CH_4 concentration ($g\ CH_4\ m^{-3}$) and K_{diff} is the eddy diffusion coefficient of CH_4 in water ($m^2\ d^{-1}$). $\widetilde{F_{s,diff}}(z) \frac{1}{A(z)dz}$ is the change of CH_4 concentration induced by diffusive inputs from the sediment columns, the term $A(z)dz$ being the volume of the water layer connected to the corresponding sediment column. $F_{w,AeOM}(z)$ is the aerobic CH_4 oxidation rate in the water column, and is described through double Michaelis-Menten reaction kinetics (Stepanenko *et al.*, 2016; Liikanen *et al.*, 2002; Thottathil and Prairie, 2019):

$$F_{w,AeOM}(z) = k_{max} Q_{10,ox}^{\frac{T-T_r}{10}} \frac{[CH_4]_{w,z}}{K_{s,CH_4} + [CH_4]_{w,z}} \frac{[O_2]_z}{K_{s,O_2} + [O_2]_z} \quad (19)$$

where k_{max} is the maximum CH_4 oxidation rate (Liikanen *et al.* 2002), T is the water temperature, T_r is the reference temperature, and $Q_{10,ox}$ expresses the temperature dependency of the CH_4 oxidation rate. K_{s,CH_4} and K_{s,O_2} are the half-saturation constants for CH_4 and O_2 , respectively.

To constrain the redissolution of gas bubbles ($F_{diss}(z)$), we follow the approach proposed by McGinnis *et al.* (2006) where a function ($f_{bdiss}(z)$) is used to represent the fraction of the benthic ebullitive CH_4 flux that redissolves in the water column during gas ascent. This fraction is a function of water depth and gas bubble diameter, and the latter was set to 5 mm following Delwiche and Hemond (2017). With this function $f_{bdiss}(z)$, the redissolved CH_4 fluxes from sediment column at depth z are assumed to be evenly redistributed in the water layers above the sediment, i.e.,

$$f_{rediss}(z) = \frac{f_{bdiss}(z) \widetilde{F_{s,ebul}}(z)}{\int_0^z A(z)dz} \quad (20)$$

where $\int_0^z A(z)dz$ is the water volume above the sediment layer at the depth of interest z . Then, at this particular depth z , the redissolution flux (F_{diss} , $g\ CH_4\ m^{-3}\ d^{-1}$) is thus determined as follows:

$$F_{diss}(z) = \int_z^{h_{max}} f_{rediss}(z)dz \quad (21)$$



388 where $\int_z^{h_{max}} f_{rediss}(z)dz$ represents the integral of all re-dissolved ebullitive fluxes from sediment
389 columns below z .

390 By deducing this dissolution flux from the ebullitive flux released from lake sediments, the
391 resultant ebullitive flux reaching the atmosphere (F_{ebul} ; $\text{g CH}_4 \text{ m}^{-2} \text{ d}^{-1}$) is calculated as:

392
$$F_{ebul} = \frac{1}{A_0} \int_0^{h_{max}} (1 - f_{bdiss}(z)) \widetilde{F_{s,ebul}}(z) dz \quad (22)$$

393 where A_0 is the lake surface area, and $(1 - f_{bdiss}(z)) \widetilde{F_{s,ebul}}(z)$ is the component of ebullitive flux at
394 depth z that reaches the atmosphere.

395 In addition to diffusive and ebullitive pathways, FLaMe also calculates a storage flux (F_{stor}) that
396 encapsulates the changes in the total CH_4 mass stored in hypolimnion due to the weakening of lake
397 stratification or turnover events when the lake surface temperature crosses the threshold of 4°C
398 (MacKay, 2012; MacKay *et al.*, 2017). This results in a full mixing of the lake that releases the
399 previously accumulated CH_4 in the anoxic portion of the lake and concomitantly fully aerates the
400 water column. Lake turnovers thus lead to a complete homogenization of O_2 and CH_4 concentration
401 across the vertically resolved water column. Upon full mixing, remobilization of larger CH_4 stocks
402 that accumulated in the hypolimnion abruptly increase the CH_4 concentration near the lake surface,
403 and hence strongly enhance the diffusive flux through the air-water interface. That is, the storage
404 flux in FLaMe is not simulated separately but it is implicitly incorporated into the diffusive flux F_{diff}
405 which increases dramatically following the formation of a very sharp CH_4 concentration gradient at
406 the lake surface.

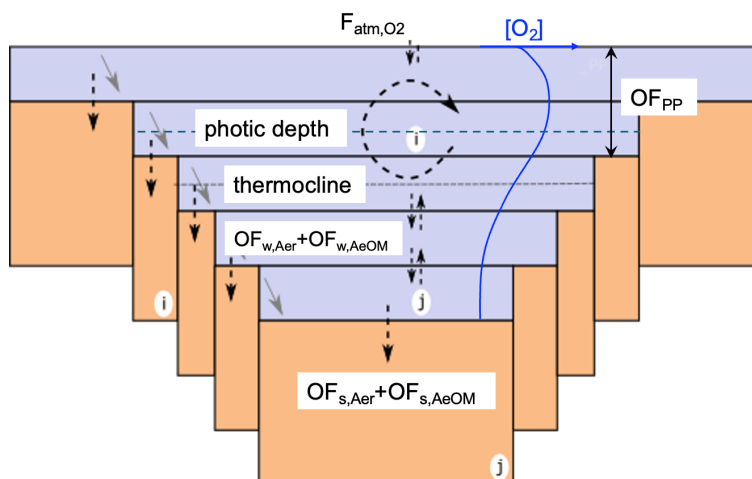


407 2.2.2.3 Oxygen module

408 The oxygen module is needed to simulate the lake methane processes (section 2.2.2.2). It
409 represents the O₂ cycle within the water column, driven by O₂ production by photosynthesis, O₂
410 consumption by pelagic and benthic OC mineralization, and aerobic pelagic and benthic CH₄
411 oxidation. These processes are coupled to the vertical diffusive transport of O₂ through water column
412 (Fig. 4). The one-dimensional conservation equation for O₂ concentration in the water column is
413 thus given by:

$$414 \quad \frac{\partial [O_2]}{\partial t} = \frac{\partial}{\partial z} (K_{diff} \frac{\partial [O_2]}{\partial z}) + OF_{PP}(z) - OF_{w,Aer}(z) - \frac{1}{A(z)dz} \widetilde{OF_{s,Aer}}(z) - OF_{w,AeOM}(z) - OF_{s,AeOM}(z) \quad (23)$$

415 where [O₂] is the O₂ concentration in the water (g O₂ m⁻³), and K_{diff} is the eddy diffusion coefficient
416 of O₂ (m² d⁻¹), assumed identical to that of CH₄. $OF_{PP}(z)$ is the oxygen production through primary
417 production (g O₂ m⁻³ d⁻¹) at depth z . $OF_{w,Aer}(z)$ is the O₂ consumption by heterotrophic respiration
418 (g O₂ m⁻³ d⁻¹) in the water column at depth z , while $\widetilde{OF_{s,Aer}}(z)$ is the volume-integrated O₂
419 consumption by heterotrophic respiration in the sediment (g O₂ m⁻³ d⁻¹), and $A(z)dz$ is the volume
420 of the water layer connected to the corresponding sediment column. $OF_{w,AeOM}(z)$ and $OF_{s,AeOM}(z)$ are
421 the aerobic CH₄ oxidation in the water column and sediment (g O₂ m⁻³ d⁻¹), respectively, at depth
422 z .



423

424 **Fig. 4. Illustration of the oxygen (O_2) module in the FLAME model. The O_2 production due to primary production**
 425 **occurs only in the photic zone (OF_{PP}), while the O_2 consumption by heterotrophic respiration occurs in both the**
 426 **entire pelagic zone and benthic zone ($OF_{w,Aer}$ and $OF_{s,Aer}$). The O_2 consumption due to CH_4 oxidation occurs also**
 427 **in both pelagic and benthic zones ($OF_{w,AeOM}$ and $OF_{s,AeOM}$). In this figure, the dotted arrows crossing the sediment-**
 428 **water interface represent the O_2 demands in sediments ($OF_{s,Aer}$ and $OF_{s,AeOM}$), the dashed arrows represent the**
 429 **eddy diffusion of O_2 between water layers and through the water-air interface, and the tilted grey arrows represent**
 430 **the aerobic oxidation of CH_4 in the water column. As a result, the blue curve depicts a typical vertical profile of**
 431 **O_2 concentration under lake water stratification.**

432 Photosynthesis occurs only in the photic zone, and the amount of O_2 produced by primary
 433 production $\overline{OF_{PP}}$ (volume-integrated value; $g\ O_2\ d^{-1}$) can be determined according to the
 434 stoichiometric ratio M_{O_2}/M_C , where M_{O_2} and M_C are the molar masses of oxygen and carbon,
 435 respectively. To resolve the vertical O_2 profile, the O_2 produced during primary production is
 436 assumed to be evenly redistributed within the water layers above the photic depth (Fig. 4):



$$OF_{PP}(z) = \begin{cases} \overline{F_{PP}} \frac{1}{V_{phot}} \frac{M_{O_2}}{M_C}, & z < z_{phot} \\ 0, & z \geq z_{phot} \end{cases} \quad (24)$$

where V_{phot} is the photic volume.

The oxygen consumption induced by CH_4 oxidation in the sediment and water column can be calculated from corresponding CH_4 fluxes (Eqs. (15) and (19), respectively) and the stoichiometry of the reactions involved:

$$OF_{s,AeOM}(z) = \frac{2M_{O_2}}{M_{CH_4}} F_{s,AeOM}(z) \quad (25)$$

$$OF_{w,AeOM}(z) = \frac{2M_{O_2}}{M_{CH_4}} F_{w,AeOM}(z) \quad (26)$$

As in Eq. (10), a fraction of the mineralized organic carbon (represented by f_{mm}) is channeled into the methanogenesis pathway according to the data compiled by Hanson *et al.* (2014) and Bastviken (2009). Thus, the remaining fraction ($1-f_{mm}$) of the total mineralization $\overline{F_{Min}}$ is channeled into the aerobic metabolic pathway (F_{Aer}). As a result, the *bulk volumetric rate* of oxygen consumption due to the aerobic metabolic activity (OF_{Aer}) can be represented by the fraction $1-f_{mm}$ and the volume-integrated mineralization $\overline{F_{Min}}$:

$$OF_{Aer} = (1-f_{mm}) \overline{F_{Min}} \frac{1}{V_w} \frac{M_{O_2}}{M_C} \quad (27)$$

In the sediment, the aerobic mineralization occurs only in the upper oxic layer. The thickness of this aerobic layer is limited by the oxygen penetration depth z_{ox} . Following Ruardij and Van Raaphorst (1995), this depth z_{ox} can be derived by solving the steady-state reaction-diffusion equation for O_2 in the sediment:

$$z_{ox} = \sqrt{\frac{2K_{s,diff}}{OF_{s,AeOM} + OF_{Aer}}} \quad (28)$$



456 where $K_{s,diff}$ is the molecular diffusion coefficient within the sediment, which is dependent on the
457 temperature (Ruudij and Van Raaphorst, 1995). The amount of O_2 consumed within the oxic layers
458 of the sediment can thus be determined as:

$$459 \quad \widetilde{OF}_{s,Aer}(z) = OF_{Aer} A_s(z) z_{ox} \quad (29)$$

460 where $A_s(z)$ is the area of the corresponding sediment column at depth z . To ensure a mass balance,
461 the volumetric rate of O_2 consumption due to aerobic metabolism in water can then be calculated
462 as follows:

$$463 \quad OF_{w,Aer}(z) = OF_{Aer} - \widetilde{OF}_{s,Aer}(z) \frac{1}{A(z)dz} \quad (30)$$

464 where $A(z)dz$ is the volume of the water layer connected to the corresponding sediment column, and
465 it is used here to convert the sedimentary O_2 consumption into a volumetric rate in the water column.
466 Furthermore, following Martin *et al.* (1987), Carlson *et al.* (1994) and Arístegui *et al.* (2003), we
467 redistribute the respiration ($OF_{w,Aer}$) within the water column, assuming that 80% of the respiration
468 occurs in the photic zone, with the remaining 20%, sustained by the export production, occurs in the
469 deeper water layers where it can further degrade.

470 2.2.3 Boundary conditions for the transport module

471 The partial differential equations (18) and (23) require boundary conditions to constrain the
472 diffusive transport (i.e., the first term on the right-hand side of both equations). At the sediment-
473 water interface, a zero-flux boundary condition is imposed, because the diffusive exchanges of CH_4
474 and O_2 between the sediment columns and the overlying waters are already included as source/sink
475 terms in Eq. (18) and (23). This choice was guided by the valley-shape configuration of our lake set-
476 up, and thus by the presence of diffusive CH_4 and O_2 exchange fluxes with sediment in each water



layer of our model, a situation in stark contrast from a bucket shape model where only a single sediment column would be connected to the bottom water layer.

At the lake surface ($z = 0$ m), the boundary conditions are determined by the CH_4 and O_2 exchange fluxes with the atmosphere, as given by (Wanninkhof et al., 2009; Cole and Caraco, 1998):

$$F_{\text{atm},\text{CH}_4} = k_{\text{ge}}([\text{CH}_4] - f_{\text{CH}_4,\text{atm}} P_{\text{atm}} M_{\text{CH}_4} K_{H,\text{CH}_4} \exp(\frac{\partial \ln(K_{H,\text{CH}_4})}{\partial T} (\frac{1}{T_1} - \frac{1}{298.15}))) \quad (31)$$

$$F_{\text{atm},\text{O}_2} = k_{\text{ge}}([\text{O}_2] - f_{\text{O}_2,\text{atm}} P_{\text{atm}} M_{\text{O}_2} K_{H,\text{O}_2} \exp(\frac{\partial \ln(K_{H,\text{O}_2})}{\partial T} (\frac{1}{T_1} - \frac{1}{298.15}))) \quad (32)$$

where $F_{\text{atm},\text{CH}_4}$ and $F_{\text{atm},\text{O}_2}$ are diffusive fluxes of CH_4 ($\text{g CH}_4 \text{ m}^{-2} \text{ d}^{-1}$) and O_2 ($\text{g O}_2 \text{ m}^{-2} \text{ d}^{-1}$) through the air-water interface of the lake, respectively. $f_{\text{CH}_4,\text{atm}}$ and $f_{\text{O}_2,\text{atm}}$ are molar fractions of CH_4 and O_2 in the atmosphere, respectively, and P_{atm} is the atmospheric pressure. K_{H,CH_4} and K_{H,O_2} are Henry's constants of CH_4 and O_2 at 298.15 K and k_{ge} is the piston velocity (m s^{-1}), here constrained from the empirical equation reported by Cole and Caraco (1998), as in Tan *et al.* (2015; 2018) and Stepanenko *et al.* (2016):

$$k_{\text{ge}} = (C_{k1} + C_{k2} v_{a,10}^n) \sqrt{\frac{600}{S_{CX}}} \quad (33)$$

where C_{k1} , C_{k2} and n are empirical constants (Cole and Caraco, 1998). $v_{a,10}$ is the absolute wind velocity measured at 10 m above the lake surface (m s^{-1}), and S_{c,CH_4} and S_{c,O_2} are the Schmidt number of CH_4 and O_2 , respectively (Wanninkhof et al. 2009). Note that more recently new formulations of k_{ge} have been proposed (McIntire et al., 2020) but we here choose to use Eq. (33) following previous modelling studies (Tan et al., 2015; Stepanenko *et al.* 2016; Tan *et al.*, 2018).



495 2.3 Parameter values

496 Table 1 summarizes all physical and biogeochemical parameters, their values, as well as the
497 original references from which they were extracted. Most of these parameters were either directly
498 taken from relevant modelling studies or constrained based on comprehensive literature reviews. In
499 addition, several key parameters of the FLame model, highlighted in Table 1, were adjusted by
500 calibrating the model based on observations of lake C fluxes (i.e., F_{PP} , diffusive and ebullitive CH_4
501 emissions). For instance, the parameters $P_{\text{Chl,max}}$ and $K_{s,P}$ control the lake primary production and
502 were tuned to reproduce broad global patterns of primary production rates across the full range of
503 lake trophic status (Wetzel, 2001). The mineralization k_{20} and burial constants k_{bur} were adjusted
504 based on the observed fraction of $C_{\text{OC,auto}}$ that settles onto the lake sediment, either to be decomposed
505 in anerobic or oxic conditions or accumulated in the sediment (Hanson *et al.*, 2011, 2014; Maavara
506 *et al.*, 2019; Mendonça *et al.*, 2017). The temperature dependence of mineralization θ was fine-
507 tuned to reproduce the observational ranges of temperature dependence of net- CH_4 emissions (Aben
508 *et al.*, 2017). f_{mm} specifies the fraction of mineralization that channels to the methanogenesis pathway,
509 which is adjusted to produce the observational patterns of CH_4 emissions. α_{min} is the base value of
510 the exponentially decreasing rate of CH_4 production versus sediment depth, and controls the split of
511 CH_4 production between diffusive and ebullitive pathways, which was calibrated to reproduce
512 observed broad trends of F_{tot} , F_{ebul} and F_{diff} from the literature (Rinta *et al.*, 2017). The parameter
513 values listed in Table 1 provide the reference setup for the simulation of lake CH_4 emissions, and
514 the sensitivity and uncertainty analyses regarding the key model parameters (listed in Table 3) is
515 carried out using wide ranges of values covering most possible lake conditions from the real world
516 (see section 3.3.3).



517 **2.4 Numerical solution**

518 In FLAME, the physical (i.e., CSLM) and biogeochemical (OC, CH₄ and O₂) modules are
519 coupled online. For the dynamics of volume-integrated OC and CH₄ in sediments, the involved
520 ordinary differential equations are solved using a forward Euler scheme. For the dynamics of
521 dissolved O₂ and CH₄ concentrations in the water column, the partial differential equations (Eqs.
522 (18) and (23)) are solved numerically using an explicit central difference scheme for depth and Euler
523 forward scheme for time. The diffusion coefficient K_{diff} for both O₂ and CH₄ is depth dependent
524 (Table 1) to capture the reduced transport when the temperature gradient from the epilimnion,
525 metalimnion and hypolimnion is well pronounced (Dong et al. 2020; Imboden and Wuest 1995;
526 Imberger 1985; Bohrer and Schultze 2008).



527

Table 1. Model parameters of FLAME v1.0 and the choice of their values

Main processes	Key model parameters	Physical meanings (units)	Values	Equations	References
Lake morphology	s	Steepness of lakebed (-)	2	(1)	-
	$P_{chl,max}$	Maximum carbon fixing rate per unit of Chlorophyll-a (mg C (mg Chl-a) ⁻¹ h ⁻¹)	0.5*	(3)	Behrenfeld and Falkowski (1997)
Primary production	$K_{s,P}$	Half saturation coefficient of total dissolved phosphorus for the primary production (g m ⁻³)	0.09*	(3)	Maavara <i>et al.</i> , (2017)
	$Q_{10,prod}$	Temperature sensitivity for the primary production	2	(4)	Lewis (2001) and Reynolds (2006)
	k_c	Absorbance of PAR per unit of chlorophyll-a (m ² (g Chl-a) ⁻¹)	0.014 × 10 ³	(5)	Lewis (2001) and Reynolds (2006)
	PP/RP	ratio of maximum gross photosynthesis to respiration per unit chlorophyll-a (-)	15	(5)	Lewis (2001) and Reynolds (2006)
	PAR_k	PAR at the onset of photo saturation (μmol m ⁻² s ⁻¹)	120	(5)	Lewis (2001) and Reynolds (2006)
	K_{dw}	PAR attenuations due to pure water	0.13	(5)	Lewis (2001) and Reynolds (2006)
	K_{dp}	PAR attenuations due to suspended particulate matter	0.06	(5)	Lewis (2001) and Reynolds (2006)
Mineralization and burial of organic carbon	k_{20}	Mineralization rate at a reference temperature of 20 °C (d ⁻¹)	0.008*	(7)	Maavara <i>et al.</i> , (2017)
	θ	Temperature dependence of mineralization	1.02*	(7)	Maavara <i>et al.</i> , (2017)



	k_{bur}	Carbon burial rate in the lake (d^{-1})	0.004*	(8)	Mendonca <i>et al.</i> , (2017)
	f_{mm}	Fraction of mineralization that channels to the methanogenesis pathway	1/4*	(10) and (27)	Hanson <i>et al.</i> (2014); Bastviken (2009)
CH₄ oxidation	k_{max}	Maximal rate of CH ₄ oxidation ($g\ CH_4\ m^{-3}\ d^{-1}$)	0.69	(19)	Liikanen <i>et al.</i> (2002)
	$Q_{10,ox}$	Temperature dependence of CH ₄ oxidation (-)	1.2	(19)	Liikanen <i>et al.</i> (2002)
	K_{s,CH_4}	Half-saturation constant for CH ₄ ($g\ CH_4\ m^{-3}$)	0.6	(19)	Stepanenko <i>et al.</i> (2016)
	K_{s,O_2}	Half-saturation constant for O ₂ ($g\ O_2\ m^{-3}$)	0.67	(19)	Liikanen <i>et al.</i> (2002)
Shape parameter of sedimentary CH₄ production	α_{min}	Base value of the exponentially decreasing rate of CH ₄ production versus sediment depth (m^{-1})	10*	(12)	Langenegger <i>et al.</i> , (2019)
Gas transport in the water column and exchange with air	K_{diff}	Depth-dependent eddy-diffusion coefficient ($m^2\ d^{-1}$)	8.64 (epilimnion), 8.64×10^{-3} at the thermocline, and 8.64×10^{-1} (hypolimnion)	(18) and (23)	Stefan and Fang (1994)
	C_{k1}	Empirical constant for piston velocity ($m\ s^{-1}$)	5.75×10^{-6}	(33)	Cole and Caraco, (1998)
	C_{k2}	Empirical constant for piston velocity ($m\ s^{-1}$)	5.97×10^{-7}	(33)	Cole and Caraco, (1998)
	n	Empirical constant for piston velocity	1.7	(33)	Cole and Caraco, (1998)
	S_{c,CH_4}	Schmidt number of CH ₄ (-)	677	(33)	Wanninkhof <i>et al.</i> (2009)



	S_{c,O_2}	Schmidt number of O ₂ (-)	589	(33)	Wanninkhof <i>et al.</i> (2009)
	$f_{CH_4,atm}$	Atmospheric molar fractions of CH ₄	0.18×10^{-13}	(31)	Lan <i>et al.</i> (2024)
	$f_{O_2,atm}$	Atmospheric molar fractions of O ₂	0.2095	(32)	Gatley <i>et al.</i> (2008)

528 * indicates that the original parameter values are from the literature, and further adjusted by calibration

529 versus observations.



530 2.5 Case studies

531 We implemented three case studies to assess the performance of FLaMe in simulating lake CH₄
532 emissions, as well as its application to the European scale. First, we present theoretical simulations
533 for two representative cases (methodological details in section 2.5.1) to assess the general behaviors
534 of FLaMe in capturing the physical-biogeochemical patterns of contrasted lakes. Then, we perform
535 the simulations for four well-surveyed real lakes to assess the model's capability in capturing the
536 observed temporal variations of CH₄ fluxes (section 2.5.2). Next, we apply FLaMe to the entire
537 European domain to assess the model's capability in reproducing the spatial patterns and seasonal
538 variations of CH₄ fluxes at continental scale (section 2.5.3). The European scale application can be
539 considered as a “proof of concept” in support of our proposed modeling strategy. Finally, we examine
540 the sensitivity to key model parameters and assess the uncertainty of the continental-scale emissions
541 using the samples produced by sensitivity analysis, combined with a machine learning approach
542 (section 2.5.4).

543 2.5.1 Two theoretical representative lakes for testing FLaMe performance

544 To test if the FLaMe model can capture the contrast in physical-biogeochemical behaviors across
545 shallow vs. deep, eutrophic vs. oligotrophic and warm vs. cold lakes, we set-up the model for two
546 theoretical representative lakes: a “deep oligotrophic” lake ($h_{max} = 35$ m or $h_{mean} = 17.5$ m and $[TP] =$
547 $3 \mu\text{g P L}^{-1}$) driven by a “cold” climate (63.75°N , 26.25°E ; Fig. S1) and a “shallow eutrophic” lake
548 ($h_{max} = 10$ m or $h_{mean} = 5$ m and $[TP] = 80 \mu\text{g P L}^{-1}$) driven by a “warm” climate (43.75°N , -6.25°E ;
549 Fig. S2). For these two theoretical representative cases, FLaMe simulates the spatio-temporal
550 evolutions of physical and biogeochemical variables and fluxes, including primary production and
551 mineralization fluxes, and labile autochthonous OC stocks as well as the vertically resolved gradients



of temperature, CH₄ and O₂ concentrations. Furthermore, we also compared the seasonal patterns of CH₄ productions and emissions for these two contrast lakes. To investigate further how environmental factors affect the FLame model behavior, we further decompose the collective responses of shallow and deep lakes into individual effects induced by trophic level, climate (Fig. S1–S3) and lake depth using hypothetical numerical simulations, i.e., (i) changing the maximal lake depth (h_{max}) from 5 to 25 m; (ii) increasing the [TP] levels from 8 to 80 $\mu\text{g P L}^{-1}$; and (iii) changing the climate from warm (43.75°N, -6.25°E; Fig. S1) to cold conditions (63.75°N, 26.25°E; Fig. S2).

2.5.2 Simulations of temporal patterns for four well-surveyed lakes

To evaluate the ability of FLame to reproduce the observed temporal patterns of CH₄ fluxes, we selected four lakes from the Inter-Sectoral Impact Model Intercomparison Project (ISIMIP) lake datasets for which monthly resolved temporal CH₄ fluxes were available. These lakes cover different lake depths, areas, climate conditions and trophic statuses, as summarized in Table 2. Since *in-situ* measurements of climatic drivers are not available for these lakes, we extracted them from the 0.5°x0.5° GSWP3-W5E5 global climate forcings released by the ISIMIP3a project as an approximation. The measurements of CH₄ fluxes for these lakes were mostly collected during the first 20 years of the 21st century, and we thus selected the climate forcings for the period 1991–2019, using the period 1991–1999 as spin-up phase. Since the lack of *in-situ* measured climatic drivers, as well as observations of variations in lake water levels and areas, affect the model’s ability to capture the full variability in the observed CH₄ emission time series, we here focus only on the magnitudes and broad seasonal patterns in observed CH₄ emissions. Thus, we evaluated the simulated statistics (mean and SD represented by boxplots) of CH₄ fluxes within the year against the observational data.



Table 2. Characteristic information for the four well-surveyed lakes from ISIMIP datasets

Lake	Coordinates	Lake depth	Lake area (km ²)	Climate	Trophic status	Data
Klöntal	47.026N, 8.981E	21.4m (mean), 45m (max)	2.25	Temperate	Oligotrophic	Annual mean
Erssjön	58.371N, 12.162E	1.3m (mean), 4.75m (max)	0.062	Temperate- Boreal	Mesotrophic	2012–2013
Upper Mystic	42.434N, 71.150W	11.7m (mean), 25m (max)	0.58	Temperate	Eutrophic	2007–2008
Villasjön	68.35N, 19.03E	1.3 m (max)	0.17	Boreal	Oligotrophic	2010–2017

2.5.3 Implementation of FLame at continental scale

To implement the model at the scale of Europe (25°W–60°E, 36°–71°N), we extracted the lakes within this domain from the HydroLAKES database (Messenger *et al.*, 2016; $n=108407$, total area = 1.33×10^5 km² for lakes with $0.1 \leq A_0 \leq 1000$ km² within the European domain). Following our clustering strategy, we subdivided, within each grid cell, all lakes into four classes based on their surface area ($0.1 < A_0 < 1$ km², $1 < A_0 < 10$ km², $10 < A_0 < 100$ km², and $100 < A_0 < 1000$ km²). As FLame was derived from the small lake physics model CSLM, we here only considered the lakes with an area smaller than 1000 km², and excluded the very large lakes ($A_0 > 1000$ km²) that account for 40% of the total European lake surface area (but only consist of 21 lakes). Within our model domain, we have 108407 lakes with a surface area larger than 0.1 km², which at spatial resolution of 2.5 degree (Fig. S4–S5) result in 365 grid cells and 953 representative lakes (hence reducing computation cost by more than a factor of 100 compared to a case where each individual lake would be simulated). By parallelizing the model simulations on a high-performance cluster, the implementation of FLame



589 for the entire European domain consumes approximately 365 CPU hours for a single run covering
590 10 years.

591 The FLame model was forced by meteorological conditions from the GSWP3-W5E5 reanalysis
592 product under ISIMIP3a (Frieler *et al.*, 2024) (Fig. S6), including shortwave solar radiation (W m^{-2}),
593 longwave solar radiation (W m^{-2}), precipitation (mm s^{-1}), near surface air temperature (at 10 m
594 height, $^{\circ}\text{C}$), specific humidity (kg kg^{-1}), near surface wind velocity (at 10m, m s^{-1}), and atmospheric
595 pressure (Pa). As these forcings were provided at a finer spatial resolution of 0.5° , we only applied
596 the values in the central 0.5° grid cell of our larger 2.5° grid. In addition, the FLame model was also
597 driven by the TDP in the representative lakes (Fig. S7–S8), which was estimated by dividing the
598 TDP mass outflow by the water discharge reported in HydroLAKES, hence assuming that the lake
599 water is well mixed. The TP mass outflow from each lake in HydroLAKES was obtained by routing
600 the TP loads (extracted from the GlobalNEWS model; Mayorga *et al.*, (2010)) from the watershed
601 (point and non-point terrestrial sources) into the river network, following the procedure outlined in
602 Lauerwald *et al.* (2019) and topological information provided by the HydroSHEDS drainage
603 network. More details related to the TP routing can be found in Bouwman and Billen (2009), Van
604 Drecht *et al.* (2009), and Mayorga *et al.* (2010).

605 **2.5.4 Sensitivity and uncertainty analysis**

606 To explore how model parameterization affects the European-scale assessments of lake CH_4
607 emissions, we conducted a sensitivity analysis encompassing the parameters whose variations
608 induce the largest changes in lake CH_4 dynamics and listed in Table 3. The sensitivity was conducted
609 by varying a parameter once at a time: only one parameter is varied with the other parameters kept



610 unchanged. All these parameters were assumed to have Gaussian distributions, with their SDs
611 specified as 50% of their original values, except the temperature dependency $Q_{10,ox}$ and θ whose
612 SDs were specified as 50% of their deviation to unity. More specifically, we tested the sensitivity
613 within the ranges of $\text{mean} \pm \text{SD}$ at four points, i.e., $+\text{SD}$, $+0.5\text{SD}$, -0.5SD , and $-\text{SD}$.

614 To constrain uncertainties in European scale CH_4 emissions, we complemented the sensitivity
615 analysis ($n=36$) with another 28 scenarios under several extreme cases and different combinations
616 of variations in key parameters. With these 64 assessments taken as samples, we then used a machine
617 learning approach to assess the uncertainty associated to our estimation of European lake CH_4 fluxes.
618 Specifically, we trained a Random Forest (RF) model that capture nonlinear relationships between
619 our key model parameters and European lake CH_4 emissions, i.e., the key parameters are taken as
620 predictors and the European lake CH_4 emissions are taken as target variable. Next, we produced
621 1000 Gaussian-distributed random samples within the parameter space and estimated an ensemble
622 of CH_4 emissions using the trained RF model.



623

Table 3 Key model parameters selected for sensitivity and uncertainty analysis

Main processes	Key model parameters	Physical meanings (units)	Values	Ranges	Eq.	References
Primary production	$P_{chl,max}$	Maximum carbon fixing rate per unit of Chlorophyll-a (mg C (mg Chl-a) ⁻¹ h ⁻¹)	0.5	0.5–6	(3)	Behrenfeld and Falkowski (1997)
	$K_{s,P}$	Half saturation coefficient of total dissolved phosphorus for the primary production (g m ⁻³)	0.09	0.006–0.189	(3)	Maavara <i>et al.</i> , (2017)
Mineralization and burial of organic carbon	k_{20}	Mineralization rate at a reference temperature of 20 °C (d ⁻¹)	0.008	0.003–0.015	(7)	Maavara <i>et al.</i> , (2017)
	θ	Temperature dependence of mineralization	1.02	1.01–1.07	(7)	Maavara <i>et al.</i> , (2017)
	k_{bur}	Carbon burial rate in the lake (d ⁻¹)	0.004	1/2 k_{20}	(8)	Mendonca <i>et al.</i> , (2017)
	f_{mm}	Fraction of mineralization that channels to the methanogenesis pathway	1/4	1/6–1/2	(10) (26)	Hanson <i>et al.</i> (2014); Bastviken (2009)
CH₄ oxidation	k_{max}	Maximal rate of CH ₄ oxidation (g CH ₄ m ⁻³ d ⁻¹)	0.69	0.19–7.68	(18)	Liikanen <i>et al.</i> (2002)
	$Q_{10,ox}$	Temperature dependence of CH ₄ oxidation (-)	1.2	1.1–2.0	(18)	Liikanen <i>et al.</i> (2002)
Base value of the shape parameter	α_{min}	Exponentially decreasing rate of CH ₄ production versus sediment depth (m ⁻¹)	10	10–70	(12)	Langenegger <i>et al.</i> , (2019)

624



625 3. Results

626 3.1 Assessing the performance of FLame in capturing patterns of CH₄ dynamics across 627 different lake types

628 The FLame model is shown to be able to well capture the typically observed, contrasting physical
629 and biogeochemical behaviors for two representative cases (Fig. 5 and Fig. S9–17; more details in
630 Supplementary Text S3): shallow vs. deep, eutrophic vs. oligotrophic and warm vs. cold lakes. In the
631 deep oligotrophic lake, the mean temperature reveals a lower and narrower seasonal variability ($\sim 3\text{--}8^\circ$
632 C) compared to the shallow eutrophic lake ($5\text{--}15^\circ\text{C}$) (Fig. 5a vs. 5b). Large temperature variations
633 in the latter are mainly driven by the smaller water volume and thus faster mean temperature response
634 to fluctuations in atmospheric temperature. In addition, the annual averaged F_{PP} in the shallow
635 eutrophic lake ($490\text{ gC m}^{-2}\text{ yr}^{-1}$) is approximately 38 times higher than that in the deep oligotrophic
636 lake ($13\text{ gC m}^{-2}\text{ yr}^{-1}$) (Fig. 5c vs. 5d). This difference can be explained by the differences in water
637 volume (energy exchange), trophic status, and climate forcings. The higher F_{PP} of the shallow
638 eutrophic lake also translates into higher $C_{OC,auto}$ concentration (~ 110 times) which persist over longer
639 periods (Fig. 5e vs. 5f), despite substantially higher F_{min} rates.

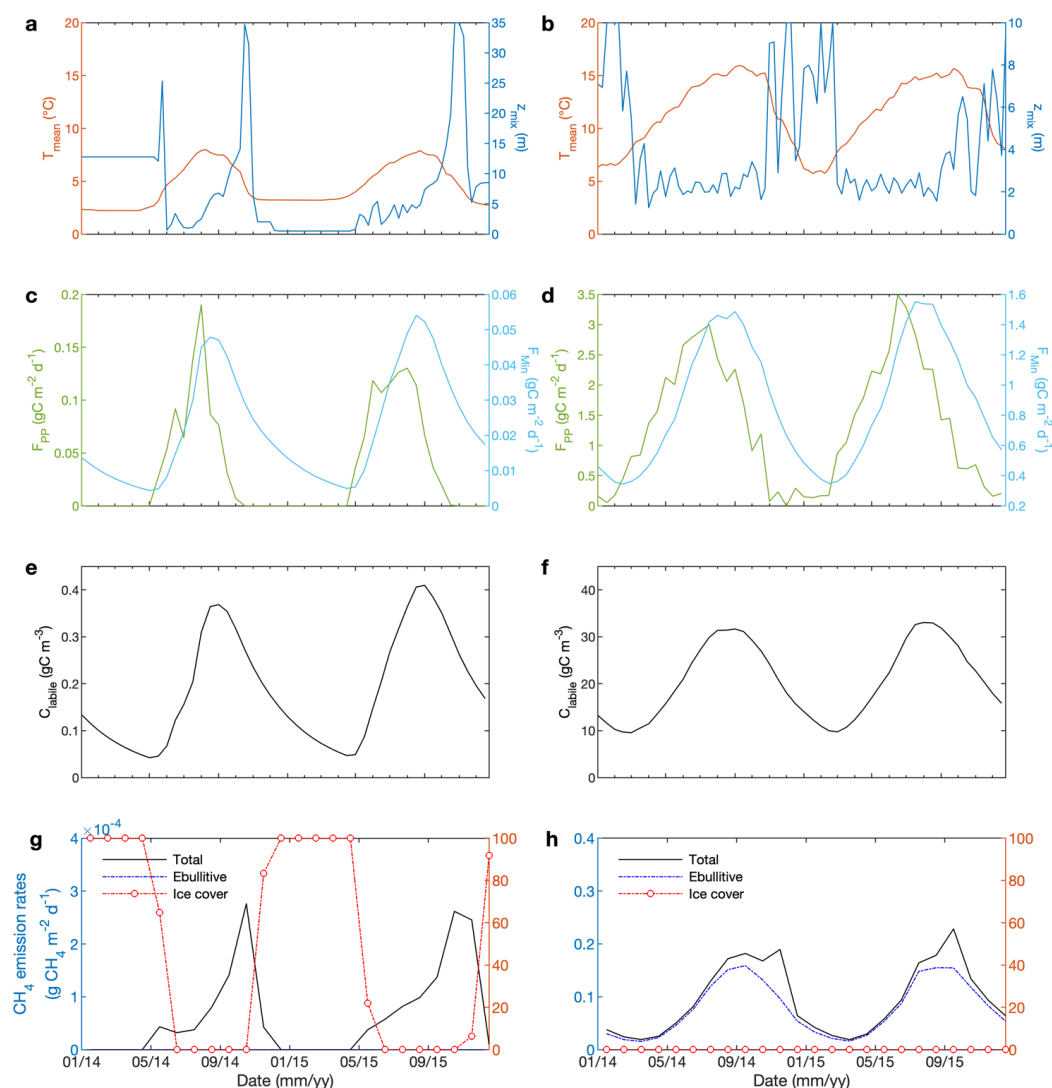


Fig. 5. Depth-integrated temporal evolution of variables and processes in two theoretical representative lakes.

The deep oligotrophic lake (left) has a maximal depth of 35 m and [TDP] of $3 \mu\text{gP L}^{-1}$, and is driven by the climate forcings at the location of 63.75°N , 26.25°E . The shallow eutrophic lake (right) has a maximal depth of 10 m and [TDP] of $80 \mu\text{gP L}^{-1}$, and is driven by the climate forcings at the location 43.75°N , -6.25°E . (a)

and (b) show the evolution of lake mean temperature and mixing depth; (c) and (d) show the evolution of primary production (F_{PP}) and mineralization rate (F_{Min}); (e) and (f) show the evolution of concentration of



647 **autochthonous organic carbon ($C_{OC,auto}$); (g) and (h) show the evolution of CH_4 emission rates and ice cover.**
648 **Note the difference scales between the left and right panels.**

649 In the deep oligotrophic lake, the simulated vertical temperature profiles indicate an almost
650 permanently maintained stratification that is only interrupted by short but intense turnover events
651 during late falls (Fig. S9a). Lake stratification (e.g., lake turnover and O_2 concentrations that depend
652 mostly on solubility and hence, temperature) dominates the spatio-temporal pattern of O_2 such that
653 O_2 concentration is near-saturated during most of the year (Fig. S9c). The oligotrophic status, together
654 with the well oxygenated conditions, results in extremely low CH_4 concentrations. Higher CH_4
655 concentrations are only simulated near the lake bottom following the productive season, i.e., late
656 summer/fall transition (Fig. S9e). In contrast, in the shallow eutrophic lake, the weaker stratification
657 results in a less pronounced vertical temperature gradient (Fig. S9b). The vertical lake O_2 profile is
658 not only controlled by the lake physics (temperature and O_2 solubility) but also by intense
659 biogeochemical processes (Fig. S9d). During summer, O_2 concentrations in the upper portion of the
660 lake are slightly supersaturated due to photosynthetic activity, followed by a gradual decrease in O_2
661 concentration as mineralization rates exceed primary production rates. Due to the high primary
662 production in the eutrophic lake, large amounts of OC are exported below the thermocline, where
663 heterotrophic activity progressively depletes O_2 , leading to the development of anoxic conditions in
664 the hypolimnion. The combination of high F_{Min} and low O_2 concentrations drive the accumulation of
665 CH_4 in late summer at the bottom of the lake (Fig. S9f), with maximal CH_4 concentration ($3.0 \text{ g } CH_4$
666 m^{-3}) exceeding those simulated in the deep oligotrophic lake by a factor of 600 (Fig. S9e).



667 By aggregating CH₄ fluxes over time, we obtained distinct seasonal patterns of CH₄ production
668 and emission for these two representative lakes (Fig. 5g and 5h; Fig. S10). In the cold, deep
669 oligotrophic lake (Fig. 5g and Fig. S10a), winter to early spring ice cover (December–April) blocks
670 CH₄ emissions such that lake CH₄ emissions are limited to the period between May and November.
671 CH₄ production is highest (8.0×10^{-4} g CH₄ m⁻² d⁻¹) in August and lowest (8.0×10^{-5} g CH₄ m⁻² d⁻¹) in
672 April. Almost all the produced CH₄ escapes the sediment via diffusion as gas bubbles do not form
673 due to low CH₄ production rates and high-water pressure. However, the benthic CH₄ flux is
674 subsequently largely oxidized in the well oxygenated deep water column. As a result, total lake CH₄
675 emissions are low (0 to 2.4×10^{-4} g CH₄ m⁻² d⁻¹) with a slight peak in October. In the shallow eutrophic
676 lake (Fig. 5h and Fig. S10b), the warmer climate prevents ice formation on the lake surface, leading
677 to an emission season about twice as long as under colder climatic conditions. CH₄ production (0.02
678 to 0.35 g CH₄ m⁻² d⁻¹) is >1000 times higher than that in cold, deep oligotrophic lake due to the higher
679 nutrient loads, lower O₂ levels, higher irradiance as well as higher temperature (Fig. 5b). Higher CH₄
680 production rates, together with lower water pressure, drive the formation of gas bubbles, leading to a
681 higher fraction of CH₄ emissions via the ebullitive pathway. The weaker stratification and the shorter
682 transport time scale in the shallow lake limits CH₄ oxidation during diffusive transport, leading to
683 ~900 times higher total CH₄ emissions compared to the deep, oligotrophic lake. Total lake CH₄
684 emissions are highest (0.21 g CH₄ m⁻² d⁻¹) in September and lowest (0.02 g CH₄ m⁻² d⁻¹) in February.

685 By decomposing the collective responses of shallow and deep lakes into individual effects
686 induced by trophic level, climate and lake depth using additional theoretical numerical simulations,
687 we found that the trophic level exerts the most important control on CH₄ dynamics, followed by
688 climate, and finally, lake depth (Fig. S11–S14). Specifically, the yearly mean CH₄ production is



689 increased by a factor of 30 (from 0.003 to 0.089 g CH₄ m⁻² d⁻¹), and the yearly mean CH₄ emission is
690 increased by a factor of 44 (from 0.0013 to 0.057 g CH₄ m⁻² d⁻¹) from oligotrophic to trophic status
691 (i.e., [TDP] increased by 10 times) (Fig. S12). From cold to warm climate, the yearly mean CH₄
692 production and emission increase by a factor of 6 (0.0094 to 0.059 g CH₄ m⁻² d⁻¹) (Fig. S13), and a
693 factor of 5 (0.0057 to 0.03 g CH₄ m⁻² d⁻¹), respectively. By increasing lake depth from 15 m to 35 m
694 (Fig. S14), the CH₄ production rates remain almost the same, i.e., 0.02 g CH₄ m⁻² d⁻¹ for the yearly
695 mean and 0.06 g CH₄ m⁻² d⁻¹ for the peak, while the CH₄ emissions are overall lower for the deeper
696 lake.

697 **3.2 Evaluation of simulated temporal lake CH₄ emissions against observations from four well-** 698 **surveyed lakes**

699 In Klöntal and Erssjön Lakes (Table 2, Fig. 6a and 6b), FLaMe captures the observed seasonal
700 cycles of CH₄ emissions well, albeit with almost a one-month delay. As a result, the simulated CH₄
701 fluxes are slightly lower in the first half of the year and slightly higher in the second half. This lag
702 between observations and model results is likely due to the use of idealized climate forcings but could
703 also be attributed to the unresolved changes in water levels and in-lake TDP dynamics. In the Klöntal
704 Lake (Fig. 6a), the observed CH₄ fluxes are exceptionally high in April (1.64 g CH₄ m⁻² d⁻¹) and July
705 (5.03 g CH₄ m⁻² d⁻¹), interrupting the normal seasonal cycles. These abrupt observed emissions might
706 reflect the contributions from storage fluxes that are not well captured by FLaMe. Apart from these
707 two months with exceptionally high fluxes, the observational data indicates peak emissions of 3.18 g
708 CH₄ m⁻² d⁻¹ in August and no emissions during the ice-covered period. FLaMe simulates similar
709 fluxes, with a peak of 3.4 g CH₄ m⁻² d⁻¹ in September (and 3.17 g CH₄ m⁻² d⁻¹ in August), and a null
710 flux in January–February when the model predicts ice formation. In the Erssjön Lake (Fig. 6b),



711 observational data report a peak in CH₄ emission reaching 13.48 g CH₄ m⁻² d⁻¹ in July and no
712 emissions during the ice-covered period, whereas FLaMe simulates a peak emission of 18.76 g CH₄
713 m⁻² d⁻¹ in August (and 12.82 g CH₄ m⁻² d⁻¹ in July), and no flux in February. Moreover, the simulated
714 CH₄ fluxes are exceptionally high in April (11.10 g CH₄ m⁻² d⁻¹) due to the release of a storage fluxes
715 that does not seem to be recorded by the observations. These high CH₄ fluxes attributed to storage
716 and lake turnover are usually associated with large variability, i.e., in Klöntal Lake (Fig. 6a), the
717 observed variability (standard deviation, SD) in CH₄ flux in July is almost 8-fold larger than the
718 simulated one, whereas in Erssjön Lake (Fig. 6b), the simulated SD in CH₄ flux in April is almost 6-
719 fold larger than that of the observed one. This suggests that both *in-situ* measurements and FLaMe
720 struggle to accurately capture the storage fluxes. Apart from these storage fluxes, we found that the
721 SDs of CH₄ fluxes simulated by FLaMe are lower than those observed for most months, indicating a
722 more stable behavior in the simulations compared to the observations across the multi-year period
723 considered here.

724 For the Upper Mystic and Villasjön Lakes (Fig. 6c and 6d), the observed temporal patterns of
725 CH₄ fluxes appear more erratic, either due to the dominant role of short-term water level fluctuations
726 or due to the complex ice cover dynamics. For the Upper Mystic Lake (Fig. 6c), the observed CH₄
727 fluxes are irregular or fluctuating (0–17.6 g CH₄ m⁻² d⁻¹) over the year, a pattern which was explained
728 by dynamic variations of lake water levels (Varadharajan, 2009). Since *in-situ* water level
729 measurements are lacking, the simulated temporal variations cannot capture these observed erratic
730 patterns well. Our model produces a smoother seasonal cycle of monthly-mean CH₄ fluxes over the
731 year, i.e., high fluxes (10.02–13.38 g CH₄ m⁻² d⁻¹) during the productive season (August–October),
732 and low fluxes (0.02–7.56 g CH₄ m⁻² d⁻¹) during the other months. Moreover, the model predicts a



733 weak storage flux occurring in November ($10.20 \text{ g CH}_4 \text{ m}^{-2} \text{ d}^{-1}$). For the Villasjön Lake (Fig. 6d), the
734 observed CH_4 fluxes are limited to the period of June–October, due to the long ice cover period
735 induced by the cold climate. FLaMe captures the observed ice-cover period well and produces similar
736 seasonal cycles of CH_4 fluxes. The simulated means and SDs are very close to observations in June
737 and July, but both, means and SDs, are much lower than observations in August, September, and
738 October.

739 In summary, despite the use of idealized climatic forcing and neglecting variations in lake area
740 and water level, FLaMe broadly captures the observed temporal patterns of monthly mean emissions,
741 albeit sometimes with small delays or diverging extents of high emissions periods. The SDs of
742 simulated CH_4 fluxes are also usually lower than the observed values, which is to be expected
743 considering that our model is not designed to capture high-frequency fluctuations of CH_4 fluxes. The
744 largest biases can be found in the estimations of storage fluxes (timing and magnitude), probably due
745 to 1) the difficulty of capturing these fluxes with existing measurement instruments and techniques,
746 2) the possibility of methane oxidation with greater than expected values during turnover and ice-out
747 (Mayr *et al.*, 2020; Zimmermann *et al.*, 2019; Pajala *et al.*, 2022) and 3) the lack of *in-situ*
748 measurements of climate conditions, dynamical water levels, and dynamic TDP concentrations
749 (Denfeld *et al.*, 2018). Resolving these issues will require to assemble a much larger dataset of
750 observed long time-series of CH_4 fluxes and associated physical and biogeochemical variables. To
751 help further calibrate and evaluate the model, this much larger pool of observations should span a
752 broader range of environmental conditions to be more representative of the lake CH_4 dynamics on
753 the continental to global scales. Overall, given the scarce spatiotemporal observations and the limited
754 possibility to validate current knowledge on process regulation in fields, it is difficult for all existing



models to produce the details of the CH₄ dynamics in specific single lakes. Hence, the temporal patterns of CH₄ fluxes simulated by FLaMe v1.0 are seen as acceptable, as its main focus is to capture the broad spatio-temporal patterns of CH₄ emissions across the thousands of lakes that need to be accounted for in large-scale applications (see section 3.3).

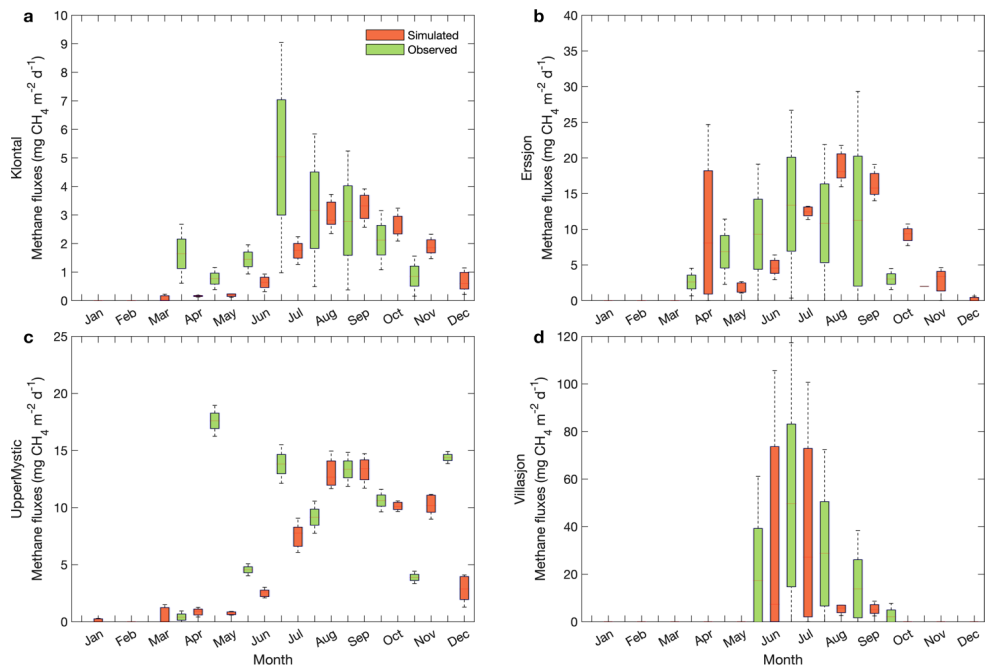


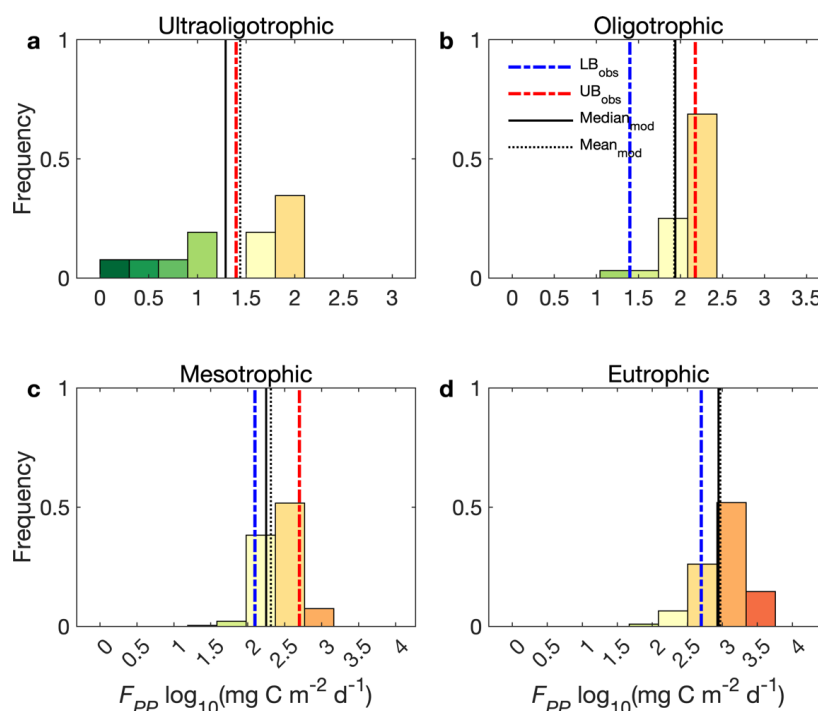
Fig. 6. Evaluation of FLaMe against monthly mean CH₄ fluxes recorded in long time-series of observations in four real lakes. (a) Klöntal, (b) Erssjön, (c) Upper Mystic, and (d) Villasjön. The detailed lake characteristics are listed in Table 2. The climate forcings for these four lakes are extracted from GSWP3-W5E5 model from ISIMIP3a. Note the different scales of CH₄ emissions in each lake.



765 3.3 FLame application on the European domain

766 3.3.1 Evaluation of FLame in European lakes

767 In the European scale application of FLame, we first evaluated the simulated F_{PP} against the
768 empirical ranges reported by Wetzel (2001) for lakes under ultraoligotrophic ($0-5 \mu\text{gP L}^{-1}$),
769 oligotrophic ($5-10 \mu\text{gP L}^{-1}$), mesotrophic ($10-30 \mu\text{gP L}^{-1}$), and eutrophic ($>30 \mu\text{gP L}^{-1}$) conditions
770 (Fig. 7 and Fig. S18). Figure 7 shows that, under different trophic status, the means and medians of
771 F_{PP} simulated by FLame (for 953 representative lakes) fall well within the reported ranges. Slight
772 deviations could only be observed in ultraoligotrophic lake for which the model tends to slightly
773 overestimate F_{PP} (Fig. 7a). Ultraoligotrophic and oligotrophic lakes reveal very similar mean and
774 median of F_{PP} that fall at the higher ends of the ranges specified by Wetzel (2001) or even exceed it
775 in the case of ultraoligotrophic lakes. In turn, mesotrophic and eutrophic lakes reveal mean and
776 median F_{PP} that fall at the lower ends of the ranges specified by Wetzel (2001). This slight difference
777 of simulated versus observed F_{PP} in lakes with different trophic conditions can be explained by the
778 relatively low value of $K_{s,P}$ ($90 \mu\text{g L}^{-1}$) compared to the concentration of [TDP] (Fig. S7–S8), as well
779 as the simplified representation of lake primary production in our model. When extending the
780 representative lakes to all real lakes in the European domain ($n=108407$), the median and mean of
781 simulated F_{PP} are still within the specified ranges but are reduced slightly for all trophic status (Fig.
782 S18), attributed to the positively skewed distribution of [TDP] (Fig. S8), i.e., many lakes have a low
783 [TDP].



784

785 **Fig. 7. Comparison of simulated primary production (F_{PP}) with empirical estimates reported by Wetzel**
786 **(2001). The histograms show the frequency distributions of simulated F_{PP} (log scale) for 953 representative**
787 **lakes that are grouped into ultraoligotrophic (0–5 $\mu\text{gP L}^{-1}$), oligotrophic (5–10 $\mu\text{gP L}^{-1}$), mesotrophic (10–**
788 **30 $\mu\text{gP L}^{-1}$), and eutrophic (>30 $\mu\text{gP L}^{-1}$) lakes. In the figure, blue and red dashed lines are the lower and**
789 **upper bounds (LB_{obs} and UB_{obs}), respectively, of empirical ranges reported by Wetzel (2001) in this class**
790 **of lakes; Black solid and dotted lines are the median_{mod} and mean_{mod}, respectively, of simulated F_{PP} for this**
791 **class of lakes.**

792

793 Next, we evaluated the simulated diffusive and ebullitive CH_4 emission rates against
794 measurements in boreal and central European regions during late summer (August–September, 2010–
795 2011) synthesized by Rinta *et al.* (2017) (Fig. 8 and Fig. S19). As Rinta *et al.* (2017) compiled *in-situ*



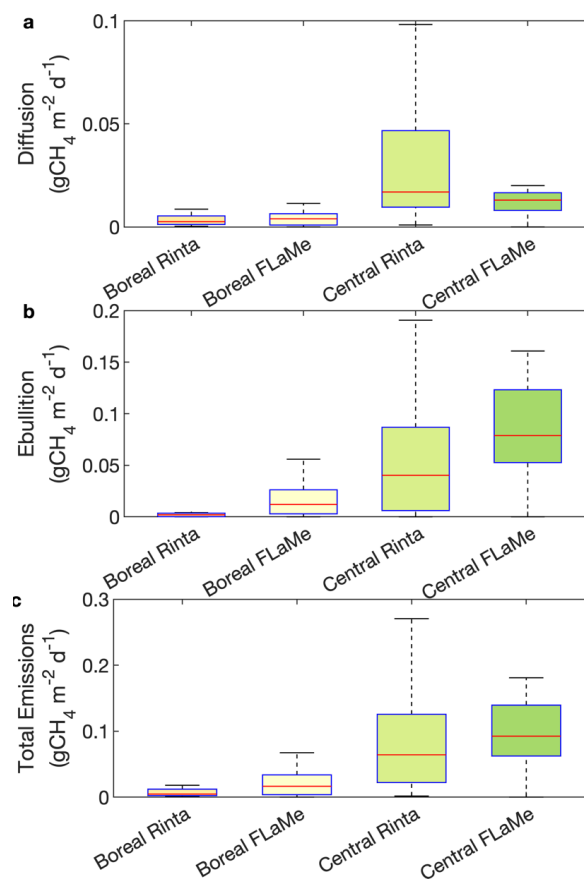
796 measurements of diffusive and ebullitive CH₄ emission rates from 17 boreal lakes (in southern
797 Finland and Sweden) and 30 lakes of central European lakes (in The Netherlands, Germany and
798 Switzerland), we extracted the mean CH₄ emission rates during August–September for representative
799 lakes located in the grid cells corresponding to these two regions. Results indicate that the simulated
800 diffusive CH₄ emissions for boreal European lakes (Fig. 8) agree well with the observations; yet the
801 simulated ebullitive CH₄ emissions are slightly higher than the observations, leading to slightly higher
802 total emissions. For central European lakes, the simulated diffusive CH₄ emissions are slightly lower
803 than the observations, while the simulated ebullitive CH₄ emissions are slightly higher, leading to a
804 good agreement in the total emissions (Fig. 8). Moreover, Rinta *et al.* (2017) reported 6 and 27 times
805 higher diffusive and ebullitive fluxes in central Europe, respectively, while our model simulates a
806 smaller contrast of a 3- and 7-fold difference. This smaller contrast in the simulation can likely be
807 explained by the higher variability in measurements, reflecting diverse climate, light and catchment
808 properties in real lakes, while the variabilities in the simulated fluxes are significantly lower, probably
809 due to more homogeneous representations of environmental conditions in the FLAME simulations.
810 Specifically, the large differences in measured CH₄ emissions in boreal and central European lakes
811 are attributed to their distinct characteristics, including climate (colder and dryer in the boreal region),
812 light regime (larger absorbance in the boreal region) and catchment properties, in particular land-use
813 (dominance of forests and smaller fraction of managed agricultural land in the boreal region).
814 However, in FLAME, the catchment properties are not fully captured by our sole, simplified indicator
815 of [TDP], such that the differences between boreal and central European lakes are underestimated.
816 The coarse resolution of our model also likely reduces the represented range of climate conditions in
817 our simulations compared to those experienced by the sampled lakes. In the meantime, observations



818 are also associated with uncertainties, because measurements were not continuous in time and might
819 thus not be fully representative of the late summer-early fall period, as well as sampling and
820 measuring CH₄ emissions, in particular via the ebullitive pathway, is all but a trivial task.
821 Nevertheless, the above evaluation of FLaMe against observations overall reveals the ability of our
822 model to reproduce broadly observed patterns in primary production and CH₄ emissions observed
823 across distinct trophic status and landscapes.



824



825

826 **Fig. 8. Comparison of simulated diffusive (top), ebullitive (middle) and total (bottom) CH₄ emission rates with the**
 827 **measurements compiled by Rinta *et al.* (2017). The datasets reported by Rinta *et al.* (2017) comprises the diffusive,**
 828 **ebullitive and total emission rates from 17 boreal lakes in Finland and Sweden and 30 lakes of central European**
 829 **lakes in The Netherlands, Germany and Switzerland. The boxes represent the 25% and 75% quartiles, and the**
 830 **whiskers cover the 95% confidence intervals. The same figure with a log scale is presented in Fig. S19.**



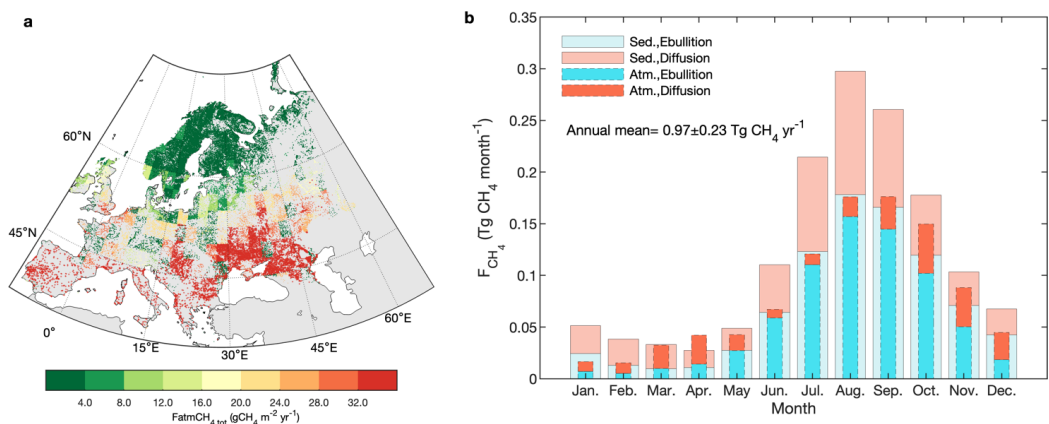
831 3.3.2 European scale assessment of lake CH₄ emissions

832 The continental-scale assessment indicates that European lakes smaller than 1000 km² have an
833 annual mean emission of 0.97 Tg CH₄ yr⁻¹ from autochthonous phytoplankton production during the
834 period of 2010–2016, of which 30% and 70% are through diffusive and ebullitive transport pathways,
835 respectively (Fig. 9 and Fig. S20). The mean CH₄ emission rates per unit lake area amounts to 7.39 g
836 CH₄ m⁻² yr⁻¹, while the mean CH₄ emission rates per unit land surface area amounts to 0.054 g CH₄
837 m⁻² yr⁻¹. Both emission rates decrease from South to North, despite the larger number of lakes and
838 lake surface area in Northern Europe (Messenger *et al.*, 2016; Fig. S4). This south to north decrease
839 can be explained by a much higher CH₄ emission rate in the South of Europe (reaching 109.6 g CH₄
840 m⁻² yr⁻¹) driven by much higher eutrophic status of southern lakes (together with higher temperatures),
841 which outcompetes the effect of the larger lake area in the Scandinavian region and Finland (which
842 contribute to ~30% of the European lake area). This latitudinal pattern of CH₄ emissions per unit lake
843 area is broadly consistent with that reported by Johnson *et al.* (2022) based on observations.

844 In terms of seasonal variability, our model results are in full agreement with the sparse data set
845 of seasonally resolved observations (Tan *et al.*, 2015) and show that European lakes as a whole act
846 as a continuous CH₄ source including during the winter months. Moreover, the simulated CH₄
847 production and emission reveal a sharp 10-fold increase from late Spring to late Summer that is
848 largely driven by the increase in ambient temperature and F_{PP} rates. These findings underscore the
849 importance of accounting for seasonal variations in CH₄ emissions when refining regional methane
850 budgets (Tan *et al.*, 2015; Guo *et al.*, 2020; Johnson *et al.*, 2022; Stavert *et al.*, 2022). A simple
851 extrapolation of observed summer emissions to the yearly timescale would thus lead to an
852 overestimation of yearly mean fluxes. In addition, model results also reveal a slight time-lag between



853 the most favorable climate conditions (air temperature and light) and the maximum CH₄ production.
854 This time lag in the model can be explained by the cascade of biogeochemical reactions (primary
855 production, mineralization, O₂ depletion and onset of CH₄ production) that ultimately control benthic
856 CH₄ fluxes, and the timescale of heat transfer from the lake surface to the deepest portion of our
857 valley-shape lake bottom. This slight time-lag is further amplified by the time required for the benthic
858 CH₄ to reach the water-air interface, although this effect is secondary due to the dominance of shallow
859 lakes (with mean depth <10 m for 98.7% of lakes) within the European domain. Finally, the broad
860 seasonal pattern in CH₄ emissions is complicated by the episodic releases of storage fluxes during
861 lake turnovers which occur during spring (March and April; emissions>production) and fall (October
862 and November; emission circa 85% of the production). Lake turnovers amplify total emissions for
863 the duration of these short-lived events.



864 **Fig. 9. Methane (CH₄) emissions from European lakes. (a) Spatial distribution of annual mean total CH₄ emissions**
865 **(sum of diffusion and ebullition) for the period of 2010-2016, expressed in per unit of lake area. (b) Seasonality of**
866 **total CH₄ production (dashed colors) and emission (full colors) fluxes and their split between ebullitive and**
867 **diffusive pathways (period 2010-2016).**
868



869 3.3.3 Sensitivity and uncertainty analysis

870 The sensitivity analysis of annual mean CH₄ emissions from European lakes to key model
871 parameters (listed in Table 3) are summarized in Table 4. Table 4 indicates that the fraction of benthic
872 organic matter mineralization channeled to methanogenesis (f_{mm}) is the most sensitive parameter, and
873 the increase (decrease) of f_{mm} by one SD leads to an increase (decrease) of European lake CH₄
874 emissions by 0.92 Tg CH₄ yr⁻¹ or 95% (0.67 Tg CH₄ yr⁻¹ or 69%). This is intuitive as a higher fraction
875 of carbon channeled to methanogenesis will increase the continental scale CH₄ emissions, although
876 the response is nonlinear. The second and third most sensitive parameters are the maximum carbon
877 fixation rate per unit of Chlorophyll-a ($P_{\text{chl,max}}$) and the half saturation constant of phosphorus ($K_{\text{s,P}}$).
878 An increase (decrease) of $P_{\text{chl,max}}$ by one SD could increase (decrease) the European lake CH₄
879 emissions by 0.77 Tg CH₄ yr⁻¹ or 79% (0.63 Tg CH₄ yr⁻¹ or 65%). This is again logical as a higher
880 $P_{\text{chl,max}}$ indicates a stronger capacity of phytoplankton to assimilate carbon, thus resulting in higher
881 amounts of organic carbon available for CH₄ production and emissions. The increase (decrease) of
882 $K_{\text{s,P}}$ by one SD decreases (increases) the European lake CH₄ emissions by 0.46 Tg CH₄ yr⁻¹ or 48%
883 (0.22 Tg CH₄ yr⁻¹ or 22%), a result which can be explained by a stronger TDP limitation of primary
884 production when $K_{\text{s,P}}$ increases, resulting in lower CH₄ production and emissions. The next most
885 sensitive parameters are the mineralization and burial rates (k_{20} and k_{bur}), for which an increase
886 (decrease) in k_{20} by one SD result in an increase (decrease) of European lake CH₄ emissions by 0.19
887 Tg CH₄ yr⁻¹ or 20% (0.39 Tg CH₄ yr⁻¹ or 40%), while an increase (decrease) of k_{bur} by one SD leads
888 to a decrease (increase) of European lake CH₄ emissions by 0.35 Tg CH₄ yr⁻¹ or 36% (0.21 Tg CH₄
889 yr⁻¹ or 22%). This is straightforward to interpret as a higher mineralization rate (k_{20}) will channel
890 more mineralization into methanogenesis (and also via lower O₂ levels in the lake), while a higher



891 burial rate (k_{bur}) translates to a lower relative amount of organic matter degradation, and thus lower
892 CH_4 production and emissions.

893 The other parameters (including the shape parameter of the CH_4 production rate versus sediment
894 depth α_{min} , the temperature dependence of mineralization θ , as well as the maximum CH_4 oxidation
895 rate k_{max} and its temperature dependence $Q_{10,ox}$) are less sensitive, with their relative effects on
896 European lake CH_4 emissions ranging from 1–20%. The shape parameter α_{min} can affect the CH_4
897 emissions as it determines the split between diffusive and ebullitive pathways, i.e., a higher α_{min} favors
898 a higher fraction of CH_4 emitted to water and atmosphere through the diffusive pathway, a pathway
899 that is more prone to oxidation thus lowering total CH_4 emissions. We also find that a higher
900 temperature dependence of mineralization (θ) results in a lower CH_4 emission. This can be explained
901 by the reference temperature of 20°C in the expression of the θ function, higher than the mean water
902 temperature in most lakes, leading to a faster drop in mineralization for a larger θ when temperature
903 is lower than 20°C. The parameter k_{max} barely impacts the total CH_4 emissions, as this parameter
904 mostly influences the thickness of the water layers where the profiles of oxygen and methane overlap
905 and the oxidation occurs, while the volume-integrated rates remain essentially unaltered (Regnier *et*
906 *al.*, 2011; Thullner and Regnier, 2019). As for the temperature dependence of oxidation ($Q_{10,ox}$), the
907 sensitivity is even weaker because changing the $Q_{10,ox}$ value has a lower impact on the oxidation rates
908 than changing k_{max} .

909 With the samples produced by the above sensitivity analysis and complemented by samples from
910 additional tests, we utilized a Random Forest (RF) model to assess the uncertainty of European lake
911 CH_4 emissions (see details in section 2.5.4). The RF model has a R^2 of 0.73 and Root of Mean Square



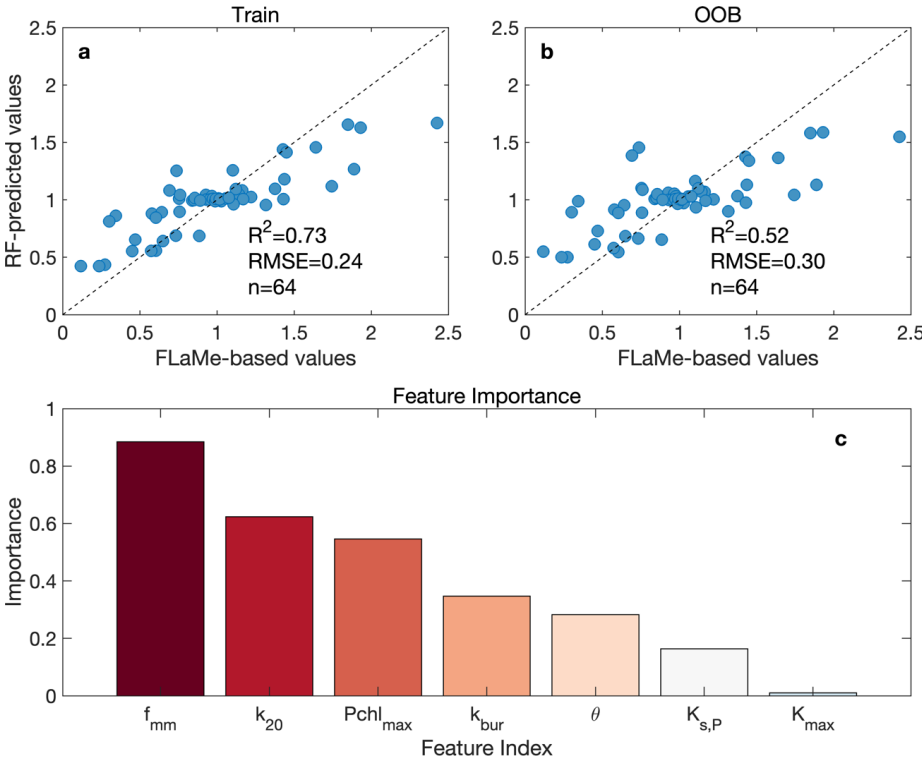
912 Error (RMSE) of $0.24 \text{ Tg CH}_4 \text{ yr}^{-1}$ for the train set (Fig. 10a) and a R^2 of 0.52 and RMSE of 0.30 Tg
913 $\text{CH}_4 \text{ yr}^{-1}$ for the out-of-bag samples (Fig. 10b), suggesting that the it can capture well the relationship
914 between model parameters and European lake CH_4 emissions. Using these ensembles of CH_4
915 emissions, the uncertainty (or SD) of European lake CH_4 emissions associated with the choice of
916 biogeochemical parameter values was estimated as $0.23 \text{ Tg CH}_4 \text{ yr}^{-1}$. Therefore, during the period of
917 2010-2016, the European lakes have an annual mean emission of $0.97 \pm 0.23 \text{ Tg CH}_4 \text{ yr}^{-1}$.

918 With the RF model, we can also identify the importance of key model parameters involved as
919 predictors (Fig. 10c). We noticed that the first four leading parameters are also the most sensitive
920 parameters as identified in Table 4, while the importance of other parameters are slightly different
921 from the sensitivity analysis. This slight difference can be attributed to the interactions of model
922 parameters that are overlooked in the sensitivity analysis. Overall, from the sensitivity and uncertainty
923 analysis, we find that the European lake CH_4 emissions are strongly controlled by the carbon
924 biogeochemical dynamics, which, however, was not fully accounted for in previous lake models.



925 **Table 4 Sensitivity of European lake CH₄ emissions (Tg CH₄ yr⁻¹) to key model**
926 **parameters**

Parameter setting		Mean±SD		Mean±0.5SD	
		−SD	+SD	−0.5SD	+0.5SD
Primary production	P_{chl_max}	0.344	1.743	0.642	1.376
	$K_{s,P}$	1.432	0.754	1.170	0.852
Mineralization and burial rates	k_{20}	0.578	1.164	0.758	1.141
	k_{bur}	1.317	0.761	1.107	0.856
	θ	1.028	0.928	0.989	0.968
	f_{mm}	0.302	1.888	0.605	1.437
Methane oxidation	K_{max}	1.057	0.930	1.009	0.953
	$Q_{10,ox}$	0.992	0.983	0.978	0.973
Base value of the shape parameter	α_{min}	1.222	0.840	1.077	0.891



927
928 **Fig. 10. Random Forest (RF) model for the uncertainty analysis. (a) and (b) are the train and test (Out-of-**
929 **Bag prediction) of the RF model. (c) shows the importance of key model parameters. Note that the**



930 parameters of α_{min} and $Q_{10,ox}$ are excluded from illustration due to their second order of importance
931 (indicated by negative values).

932 4. Model limitations

933 We have illustrated that FLaMe is able to capture complex physical-biogeochemical behaviors
934 for lakes with diverse settings and environmental controls. Specifically, the FLaMe model has been
935 evaluated against (i) observational temporal variations of CH₄ fluxes at four contrasting, well-
936 surveyed real lakes, (ii) the empirical ranges of primary production under different trophic status
937 reported by Wetzel (2001), and (iii) observational patterns of CH₄ emissions against trophic and
938 climate gradients spanning the European domain (Rinta *et al.*, 2017). Moreover, the European scale
939 simulation produces a spatial pattern of lake CH₄ emission rates consistent with observation-based
940 upscaling approaches (Johnson *et al.*, 2022). This continental scale application also demonstrates the
941 power of our modelling framework that rests on a lake clustering approach and on a routing of nutrient
942 (TDP) inputs from surrounding catchments to lakes that allow to account for eutrophication effects.
943 Our results thus suggest that the FLaMe modelling framework performs well in providing reliable
944 spatio-temporal patterns of lake CH₄ emissions at the large scale. However, the results also pinpoint
945 to several key aspects to be improved in the model and highlight critical data gaps that must be
946 addressed in the future.

947 First, the organic carbon module only accounts for autochthonous OC production as the substrate
948 for methanogenesis, but ignores the contribution of allochthonous OC inputs leached from the
949 surrounding catchments, rivers and streamflow. This is based on the distinct reactivity of
950 autochthonous and allochthonous OC inputs, with the latter being more refractory to microbial



951 activities (mineralization and decomposition). As a result, FLame may provide conservative
952 estimates of CH₄ productions and emissions. However, neglecting the allochthonous C inputs may at
953 the same time minimize the feedback of OC on light penetration, leading to systematically biased
954 estimates (section 2.2.2.1). Moreover, transient lake phosphorus dynamics and the co-limitations by
955 nitrogen, albeit assumed to be less important, are neglected and might increase the uncertainty in the
956 estimates of CH₄ production and emission. In addition, our primary production model does not
957 resolve the short-term (e.g., (sub)daily) dynamics of algae growth induced by climate variability,
958 rendering model-data comparison more difficult. In future model developments, these limitations
959 could be addressed by (i) integrating or routing the lake water, carbon and nutrient fluxes along the
960 global river network, which would allow to simultaneously solve the issue of time-invariant lake
961 water levels in current global lake models (Golub *et al.*, 2022), including ours; (ii) refining the carbon
962 module by incorporating more dynamic models for algal growth as well as P and N uptake and
963 recycling processes within lakes.

964 Second, several model assumptions and implementations are based on empirical or theoretical
965 knowledge, which may lead to biases in the estimation of CH₄ fluxes. For instance, the present version
966 of FLame (i.e., v1.0) neglects the plant-mediated emission pathway (through rooted plant) in the
967 littoral zone due to the lack of observational data for model calibration. Moreover, in our model, the
968 lake is assumed to follow a “valley” shape. Although this is an advancement from the “bucket” shape
969 used in previous process-based lake models of CH₄ emissions (e.g., LAKE 2.0, ABLM, and
970 bLake4Me), it remains a simplified assumption that captures important but not all features of a
971 realistic lake geometry. Furthermore, several benthic CH₄ processes are highly parameterized. For
972 instance, the split between aerobic and anaerobic decomposition of organic matter is represented by



973 a single parameter f_{mm} and is determined based on the data compilation from Bastviken (2022). This
974 simplification leads to the same temperature dependence of CH₄ processes occurring in the sediment
975 as that of pelagic and benthic mineralization. This is a shortcoming although it should be noted that
976 the overall temperature dependence of CH₄ emissions, which results from the combined effects of
977 OC production, mineralization, and subsequent CH₄ processes, was found to fall well within the
978 observed ranges reported by Aben *et al.* (2017) (Fig. S21). The split of diffusive and ebullitive CH₄
979 fluxes is also currently captured by an empirically determined threshold depth ($z_{eb,min}$) based on
980 limited observations by Langenegger *et al.* (2019). Moreover, the effects of heat transfer and CH₄
981 bubbles migration in the sediment are not resolved, which may lead to biased simulation of CH₄
982 fluxes especially for the timing. These are simplified representations related to the highly complex
983 pathways of CH₄ production and emission, which needs to be improved by more mechanistic
984 representations of the biogeochemical processes controlling carbon cycling, CH₄ production and
985 transport via diffusion and bubble ascent.

986 Third, different modules of the FLaMe model could benefit from more comprehensive calibration
987 and evaluation but those are limited by data availability. Although FLaMe has been evaluated against
988 several timeseries of observed data collected in four well-surveyed lakes with contrasted dynamics,
989 a full evaluation in the context of large-scale application would benefit from a significantly larger and
990 representative set of observational data. Moreover, the *in-situ* climate conditions may vary greatly
991 from the grid-level forcings, and the lake water dynamics may also affect the CH₄ fluxes significantly
992 (e.g., Upper Mystic Lake; Varadharajan, 2009). Thus, a full comprehensive set of *in-situ*
993 measurements of climate, water level, physical and biogeochemical variables would be highly
994 valuable for the purpose of further model development, calibration and evaluation. At the European



995 scale, we partly circumvented these limitations by evaluating lake primary production against the
996 broad ranges reported by Wetzel (2001), and the simulated diffusive and ebullitive CH₄ fluxes across
997 the environmental (nutrient and climate) gradients compiled by Rinta *et al.* (2017). In this context,
998 complementary time-series of vertically resolved organic carbon, CH₄ and O₂ concentrations, as well
999 as high frequency measurements of CH₄ fluxes capturing short-lived emissions via the storage and
1000 ebullitive pathways and covering heterogeneity of CH₄ fluxes in large lakes (Denfeld *et al.*, 2018;
1001 Mayr *et al.*, 2020; Zimmermann *et al.*, 2019) would help further calibrate and evaluate the FLame
1002 model. These measurements should be performed using a sufficiently large set of representative lakes
1003 covering the full range of lake morphologies, landscape properties, and climate.

1004 **5. Conclusion and outlook**

1005 In this study, we developed and tested a new process-based biogeochemical modeling framework
1006 (FLame) to simulate lake CH₄ fluxes on the large-scale and, as a “proof of concept”, applied the
1007 model to European lakes. The physical lake model builds on the Canadian Small Lake Model (CSLM)
1008 and is coupled to a set of novel biogeochemical modules describing lake organic matter, oxygen and
1009 methane dynamics. We then showcased the abilities and performance of FLame by: (1) analyzing the
1010 overall behaviors of the coupled C-O₂-CH₄ dynamics in two representative cases (a deep oligotrophic
1011 lake driven by cold climate in Northern Europe and a shallow eutrophic lake driven by warm climate
1012 in Southern Europe) as well as their decomposition, and (2) evaluating simulated temporal patterns
1013 of CH₄ fluxes against observations at four well-surveyed lakes with long-term timeseries. Simulation
1014 results were consistent with our common knowledge of lake CH₄ dynamics, suggesting that FLame
1015 can capture the patterns of CH₄ production and emissions across different lake types as well as their
1016 responses to the changes in environment conditions, despite the complexity of underlying



1017 biogeochemical processes. Furthermore, by applying the model to boreal and central European lakes,
1018 we showed that FLame captures well the observed magnitudes of both diffusive and ebullitive CH₄
1019 fluxes as well as the difference between boreal and central lakes. Finally, at the European scale,
1020 FLame estimates total CH₄ emissions from lakes smaller than 1000 km² (n=108407, total area =
1021 1.33x10⁵ km²) as 0.97±0.23 Tg CH₄ yr⁻¹. In addition, the model resolves spatial patterns and seasonal
1022 variations of CH₄ emissions, providing a comprehensive view of their contribution to regional
1023 methane budgets.

1024 Despite some limitations in its current model configuration, this first version of FLame is a
1025 significant step forward in biogeochemical simulations of lake CH₄ dynamics. The model explicitly
1026 incorporates the dynamics of depth-integrated organic carbon cycling, such that the responses of
1027 organic carbon to climate and environmental change can be accounted for in estimating CH₄
1028 emissions. We also have incorporated the primary production as a function of total dissolved
1029 phosphorus loads from the surrounding catchments, allowing us to evaluate for the first time the
1030 impact of eutrophication on CH₄ emissions in a quantitative way. Moreover, our model is of
1031 intermediate complexity, and is thus designed for large scale applications. Although the model was
1032 run here at a coarse spatial resolution, its parallelized version offers the possibility to carry
1033 simulations at a finer resolution in the future. With these advancements, our model can be used to
1034 resolve the spatio-temporal variability of CH₄ emissions at regional and global scales under past and
1035 future climates, and has the potential to be coupled to Earth System Models to investigate the
1036 feedback between climate warming and global lake CH₄ emissions.



1037 **Data availability**

1038 The methane emission data for the four well-surveyed real lakes (Klöntal, Erssjön, Upper Mystic, and
1039 Villasjön) were obtained from Tan *et al.* (2024). The *in-situ* measurements of diffusive and ebullitive
1040 CH₄ emission rates in boreal and central European regions during late summer (August–September
1041 2010–2011) were obtained from Rinta *et al.* (2017). The lake characteristic information within Europe
1042 were obtained from the HydroLAKES database (Messenger et al. 2016):
1043 <https://www.hydrosheds.org/products/hydrolakes>. The meteorological variables from GSWP3-
1044 W5E5 reanalysis product were obtained from Inter-Sectoral Impact Model Intercomparison Project
1045 (ISIMIP3a): <https://www.isimip.org/gettingstarted/input-data-bias-adjustment/>.

1046

1047 **Code availability**

1048 The source codes for FLame (Fluxes of Lake Methane) model version 1.0 are available at:
1049 <https://github.com/myFeng818/FLaMe-model-v1.0.git>. The preprocessing and postprocessing codes
1050 for the model can be obtained upon request.

1051

1052 **Acknowledgements**

1053 This study was supported by the Fonds National de la Recherche Scientifique of Belgium (F.R.S.–
1054 FRNS PDR T.0191.23), by the project of CLIMATE-SPACE RECCAP2: Global Land Carbon
1055 Budget and its Attribution to regional drivers, as well as by the project of ESM2025–Earth System
1056 Models for the Future (101003536). We acknowledge the climate modelling groups involved in
1057 ISIMIP3a for producing and making available their model outputs. Computational resources have



1058 been provided by the Consortium des Équipements de Calcul Intensif (CÉCI), funded by the Fonds
1059 de la Recherche Scientifique de Belgique (F.R.S.-FNRS) under Grant No. 2.5020.11 and by the
1060 Walloon Region.

1061

1062 **Author contributions**

1063 M.M., M.F. and P.R. designed the study. M.M. and M.F. co-developed and tested the FLaMe model.
1064 D.B., S.A., R.L., A.J., G.G.L., M.D.M., and Z.T. provided plenty of valuable suggestions related to
1065 the model development. D.B. and S.A. also provided constructive suggestions on model evaluation
1066 against measurements and manuscript writing. M.D.M. helped us in setting up the CSLM at the
1067 beginning of developing FLaMe, Z.T. provided us the methane emission data from ISIMIP lake
1068 datasets, and W.T. helped us in collecting climate forcings from ISIMIP3a. M.M. and M.F. wrote the
1069 first version of the manuscript, and all coauthors helped in improving the manuscript.

1070

1071 **Competing interests**

1072 The authors declare no competing interests.

1073

1074 **Additional information**

1075 **Correspondence and requests for materials** should be addressed to M.F. (Maoyuan.feng@ulb.be)

1076



1077 **References**

- 1078 Aben, R.C.H., Barros, N., van Donk, E. *et al.*: Cross continental increase in methane ebullition under climate
1079 change. *Nat. Commun.*, 8, 1682. <https://doi.org/10.1038/s41467-017-01535-y>, 2017.
- 1080 Aristegui, J., Agustí, S. and Duarte, C. M.: Respiration in the dark ocean, *Geophys. Res. Lett.*, 30, 1041,
1081 <https://doi.org/10.1029/2002GL016227>, 2003.
- 1082 Bastviken D., Tranvik L. J., Downing, J. A., Crill, P. M., Enrich-Prast, A.: Freshwater methane emissions offset the
1083 continental carbon sink, *Science*, 331(6013), 50. <https://doi.org/10.1126/science.1196808>, 2011.
- 1084 Bastviken, D. (2022). Methane. In T. Mehner & K. Tockner (Eds.), *Encyclopedia of Inland Waters (Second Edition)*
1085 (pp. 136-154). Oxford: Elsevier.
- 1086 Bastviken, D., Cole, J., Pace, M., and Tranvik, L.: Methane emissions from lakes: Dependence of lake
1087 characteristics, two regional assessments, and a global estimate, *Global Biogeochem. Cycles*, 18, GB4009,
1088 <https://doi.org/10.1029/2004GB002238>, 2004.
- 1089 Behrenfeld, M. J. and Falkowski P. G.: Photosynthetic rates derived from satellite-based chlorophyll
1090 concentration, *Limnol. Oceanogr.*, 42, <https://doi.org/10.4319/lo.1997.42.1.0001>, 1997.
- 1091 Bohrer, B. and Schultze, M.: Stratification of lakes, *Rev. Geophys.*, 46, 2006RG000210,
1092 <https://doi.org/10.1029/2006RG000210>, 2008.
- 1093 Bouwman, A. F., Beusen, A. H. W., Billen G.: Human alteration of the global nitrogen and phosphorus soil
1094 balances for the period 1970–2050, *Global Biogeochem. Cycles*, 23, GB0A04,
1095 <https://doi.org/10.1029/2009GB003576>, 2009.
- 1096 Canadell, J. G., Monteiro, P. M. S., Costa, M. H., *et al.*: Global carbon and other biogeochemical cycles and
1097 feedbacks, Climate Change 2021: The Physical Science Basis: Working Group I Contribution to the Sixth
1098 Assessment Report of the Intergovernmental Panel on Climate Change, Cambridge University
1099 Press, Cambridge, United Kingdom and New York, NY, USA, 673-816, 2021.
- 1100 Carlson, C. A., Ducklow, H. W., Michaels, A. F.: Annual flux of dissolved organic carbon from the euphotic zone
1101 in the northwestern Sargasso Sea, *Nature*, 397, 405–408, 1994
- 1102 Cole, J. J., and Caraco, N. F.: Atmospheric exchange of carbon dioxide in a low-wind oligotrophic lake measured
1103 by the addition of SF₆, *Limnol. and Oceanogr.*, 4, <https://doi.org/10.4319/lo.1998.43.4.0647>, 1998.
- 1104 Deemer, B. R., Harrison, J. A., Li, S., *et al.*: Greenhouse Gas Emissions from Reservoir Water Surfaces: A New
1105 Global Synthesis. *Bioscience*, 66(11), 949-964. <https://doi.org/10.1093/biosci/biw117>, 2016.
- 1106 Deemer, B. R., & Holgerson, M. A.: Drivers of methane flux differ between lakes and reservoirs, complicating
1107 global upscaling efforts. *J. Geophys. Res.-Biogeo.*, 126,
1108 e2019JG005600, <https://doi.org/10.1029/2019JG005600>, 2021.
- 1109 DelSontro, T., Beaulieu, J.J., and Downing, J.A.: Greenhouse gas emissions from lakes and impoundments:



- 1110 Upscaling in the face of global change, *Limnol. Oceanogr. Lett.*, 3, 64–75, <https://doi.org/10.1002/lo2.10073>,
1111 2018
- 1112 Delwiche, K. and Hemond, H.F.: An enhanced bubble size sensor for long-term ebullition studies. *Limnol.*
1113 *Oceanogr. Methods*, 15, 821–835. <https://doi.org/10.1002/lom3.10201>, 2017
- 1114 Denfeld, B. A., Baulch, H. M., del Giorgio, P. A., Hampton, S. E., and Karlsson, J.: A synthesis of carbon dioxide
1115 and methane dynamics during the ice-covered period of northern lakes, *Limnol. Oceanogr. Lett.*, 3, 117–
1116 131. <https://doi.org/10.1002/lo2.10079>, 2018.
- 1117 Dlugokencky, E. J., Steele, L. P., Lang, P. M., Masarie, K. A.: The growth rate and distribution of atmospheric
1118 methane, *J. Geophys. Res.*, 99, 17021–17043, <https://doi.org/10.1029/94JD01245>, 1994.
- 1119 Frieler, K., *et al.*: Scenario setup and forcing data for impact model evaluation and impact attribution within the
1120 third round of the Inter-Sectoral Impact Model Intercomparison Project (ISIMIP3a), *Geosci. Model Dev.*, 17,
1121 1–51, <https://doi.org/10.5194/gmd-17-1-2024>, 2024.
- 1122 Forster, P. M., *et al.*: Indicators of Global Climate Change 2023: annual update of key indicators of the state of the
1123 climate system and human influence, *Earth Syst. Sci. Data*, 16, 2625–2658, [https://doi.org/10.5194/essd-16-](https://doi.org/10.5194/essd-16-2625-2024)
1124 2625-2024, 2024.
- 1125 Garnaud, C., MacKay, M., & Fortin, V.: A one-dimensional lake model in ECCC's land surface prediction system.
1126 *J. Adv. Model. Earth Syst.*, 14, e2021MS002861. <https://doi.org/10.1029/2021MS002861>, 2022.
- 1127 Gatley, D. P., Herrmann, S., & Kretzschmar, H. J.: A twenty-first century molar mass for dry air, *HVAC&R*
1128 *Research*, 14(5), 655–662. <https://doi.org/10.1080/10789669.2008.10391032>, 2008.
- 1129 Golub, M., *et al.*: A framework for ensemble modelling of climate change impacts on lakes worldwide: the ISIMIP
1130 Lake Sector, *Geosci. Model Dev.*, 15, 4597–4623, <https://doi.org/10.5194/gmd-15-4597-2022>, 2022.
- 1131 Guillemette, F., von Wachenfeldt, E., Kothawala, D. N., Bastviken, D., Tranvik, L. J.: Preferential sequestration of
1132 terrestrial organic matter in boreal lake sediments, *J. Geophys. Res. Biogeosci.*, 122, 863–874,
1133 <https://doi.org/10.1002/2016JG003735>, 2017.
- 1134 Hanson, P. C., Hamilton, D. P., Stanley, E. H., Preston, N., Langman, O. C., Kara, E. L.: Fate of Allochthonous
1135 Dissolved Organic Carbon in Lakes: A Quantitative Approach. *PLoS One*, 6(7): e21884,
1136 <https://doi.org/10.1371/journal.pone.0021884>, 2011.
- 1137 Hanson, P. C., Buffam, I., Rusak, J. A., Stanley, E. H., Watras, C.: Quantifying lake allochthonous organic carbon
1138 budgets using a simple equilibrium model, *Limnol. Oceanogr.*, 59, <https://doi.org/10.4319/lo.2014.59.1.0167>,
1139 2014
- 1140 Harrison, J. A., Prairie, Y. T., Mercier-Blais, S., Soued, C.: Year-2020 global distribution and pathways of reservoir
1141 methane and carbon dioxide emissions according to the greenhouse gas from reservoirs (G-res) model, *Global*
1142 *Biogeochem. Cycles*, 35, e2020GB006888. <https://doi.org/10.1029/2020GB006888>, 2021.



- 1143 Holgerson, M., and Raymond, P.: Large contribution to inland water CO₂ and CH₄ emissions from very small
1144 ponds. *Nature Geosci.*, 9, 222–226. <https://doi.org/10.1038/ngeo2654>, 2016.
- 1145 Imberger, J.: The diurnal mixed layer, *Limnol. Oceanogr.*, 30, 737–770, <https://doi.org/10.4319/lo.1985.30.4.0737>,
1146 1985.
- 1147 Imboden, D.M., and Wüest, A.: Mixing Mechanisms in Lakes. In: Lerman, A., Imboden, D.M., Gat, J.R. (eds)
1148 Physics and Chemistry of Lakes. Springer, Berlin, Heidelberg. https://doi.org/10.1007/978-3-642-85132-2_4,
1149 1995.
- 1150 Lan, X., Thoning, K.W., and Dlugokencky, E.J.: Trends in globally-averaged CH₄, N₂O, and SF₆ determined from
1151 NOAA Global Monitoring Laboratory measurements. <https://doi.org/10.15138/P8XG-AA10>, 2024
- 1152 Langenegger, T., Vachon, D., Donis, D., McGinnis, D.F.: What the bubble knows: Lake methane dynamics revealed
1153 by sediment gas bubble composition. *Limnol. Oceanogr.*, 64: 1526–1544. <https://doi.org/10.1002/lno.11133>,
1154 2019.
- 1155 Lauerwald, R., Regnier, P., Figueiredo, V., Enrich-Prast, A., Bastviken, D., Lehner, B.: Natural lakes are a minor
1156 global source of N₂O to the atmosphere. *Global Biogeochem. Cycles*, 33, 1564–
1157 1581. <https://doi.org/10.1029/2019GB006261>, 2019.
- 1158 Lauerwald, R., Allen, G. H., Deemer, B. R., Liu, S., Maavara, T., Raymond, P., *et al.*: Inland water greenhouse gas
1159 budgets for RECCAP2: 2. Regionalization and homogenization of estimates. *Global Biogeochemical Cycles*,
1160 37, e2022GB007658, <https://doi.org/10.1029/2022GB007658>, 2023.
- 1161 Lewis, W. M.: Global primary production of lakes: 19th Baldi Memorial Lecture, *Inland Waters*, 1(1), 1–28.
1162 <https://doi.org/10.5268/IW-1.1.384>, 2011.
- 1163 Liikanen, A., Murtoniemi, T., Tanskanen, H., Väisänen, T., *Martikainen, P. J.*: Effects of temperature and oxygen
1164 availability on greenhouse gas and nutrient dynamics in sediment of a eutrophic mid-boreal
1165 lake, *Biogeochemistry*, 59, 269–286. <https://doi.org/10.1023/A:1016015526712>, 2002
- 1166 Maavara, T., Lauerwald, R., Regnier, P., Van Cappellen, P.: Global perturbation of organic carbon cycling by river
1167 damming, *Nat. Commun.* 8, 15347. <https://doi.org/10.1038/ncomms15347>, 2017.
- 1168 Maavara, T., Lauerwald, R., Laruelle, G. G., Akbarzadeh, Z., Bouskill, N. J., Van Cappellen, P., Regnier,
1169 P. Nitrous oxide emissions from inland waters: Are IPCC estimates too high? *Glob Change Biol.*, 25, 473–
1170 448. <https://doi.org/10.1111/gcb.14504>, 2019.
- 1171 MacIntyre, S., Bastviken, D., Arneborg, L., Crowe, A. T., Karlsson, J., Andersson, A., Gålfalk, M., Rutgersson, A.,
1172 Podgrajsek, E., Melack, J. M.: Turbulence in a small boreal lake: Consequences for air-water gas exchange,
1173 *Limnol. Oceanogr.*, 66(3):827–854. <https://doi.org/10.1002/lno.11645>, 2020.
- 1174 MacKay, M. D., Verseghy, D. L., Fortin, V., and Rennie, M. D.: Wintertime simulations of a boreal lake with the
1175 Canadian Small Lake Model, *J. Hydrometeorol.*, 18, 2143–2160, <https://doi.org/10.1175/JHM-D-16-0268.1>,



- 1176 2017.
- 1177 MacKay, M. D.: A process-oriented small lake scheme for coupled climate modeling applications, *J.*
1178 *Hydrometeorol.*, 13, 1911–1924, <https://doi.org/10.1175/JHM-D-11-0116.1>, 2012.
- 1179 McGinnis, D. F., Greinert, J., Artemov, Y., Beaubien, S. E., Wüest, A.: Fate of rising methane bubbles in stratified
1180 waters: How much methane reaches the atmosphere? *J. Geophys. Res.*, 111, C09007,
1181 <https://doi.org/10.1029/2005JC003183>, 2006
- 1182 Mendonça, R., Müller, R.A., Clow, D., Verpoorter, C., Raymond, P., Tranvik, L. J., Sobek, S.: Organic carbon
1183 burial in global lakes and reservoirs, *Nat. Commun.*, 8, 1694, <https://doi.org/10.1038/s41467-017-01789-6>,
1184 2017
- 1185 Messenger, M., Lehner, B., Grill, G., Nedeva, I., Schmitt, O.: Estimating the volume and age of water stored in global
1186 lakes using a geo-statistical approach. *Nat. Commun.*, 7, 13603, <https://doi.org/10.1038/ncomms13603>,
1187 2016. Martin, J. H., Knauer, G. A., Karl, D. M., Broenkow, W. W.: VERTEX: carbon cycling in the northeast
1188 Pacific, *Deep. Sea. Res. A.*, 34, 2, 267–285, [https://doi.org/10.1016/0198-0149\(87\)90086-0](https://doi.org/10.1016/0198-0149(87)90086-0), 1987.
- 1189 Mayorga, E., Seitzinger, S. P., Harrison, J. A., Dumont, E. Beusen, A. H. W., Bouwman, A. F., Fekete, B. M.,
1190 Kroeze, C., Van Drecht, G.: Global Nutrient Export from WaterSheds 2 (NEWS 2): Model development and
1191 implementation, *Environ. Model. Softw.*, 25, 7, 837–853, <https://doi.org/10.1016/j.envsoft.2010.01.007>, 2010.
- 1192 Mayr, M. J., Zimmermann, M., Dey, J., *et al.*: Growth and rapid succession of methanotrophs effectively limit
1193 methane release during lake overturn. *Commun. Biol.*, 3, 108. <https://doi.org/10.1038/s42003-020-0838-z>,
1194 2020.
- 1195 Jackson, R., Saunio, M., Bousquet, P., Canadell, J. G., Poulter, B., Stavert, A. R., Bergamaschi, P., Niwa, Y.,
1196 Segers, A., Tsuruta, A.: Increasing anthropogenic methane emissions arise equally from agricultural and fossil
1197 fuel sources, *Environ. Res. Lett.*, 15, 071002, <https://doi.org/10.1088/1748-9326/ab9ed2>, 2020.
- 1198 Johnson, M. S., Matthews, E., Du, J., Genovese, V., Bastviken, D.: Methane emission from global lakes: New
1199 spatiotemporal data and observation-driven modeling of methane dynamics indicates lower emissions, *J.*
1200 *Geophys. Res.-Biogeo.*, 127, e2022JG006793. <https://doi.org/10.1029/2022JG006793>, 2022.
- 1201 Pajala, G., Sawakuchi, H. O., Rudberg, D., *et al.*: The effects of water column dissolved oxygen concentrations on
1202 lake methane emissions—Results from a whole-lake oxygenation experiment. *J. Geophys. Res.: Biogeo.*,
1203 128(11), e2022JG007185. doi:10.1029/2022jg007185, 2023.
- 1204 Petrescu, A. M. R., Qiu, C., Ciais, P., *et al.*: The consolidated European synthesis of CH₄ and N₂O emissions for
1205 the European Union and United Kingdom: 1990–2017, *Earth Syst. Sci. Data*, 13, 2307–2362,
1206 <https://doi.org/10.5194/essd-13-2307-2021>, 2021.
- 1207 Petrescu, A. M. R., Qiu, C., McGrath, M. J., *et al.*: The consolidated European synthesis of CH₄ and N₂O emissions
1208 for the European Union and United Kingdom, 1990–2019, *Earth Syst. Sci. Data*, 15, 1197–1268,
1209 <https://doi.org/10.5194/essd-15-1197-2023>, 2023.



- 1210 Rinta, P., Bastviken, D. Schilder, J., van Hardenbroek, M., Stötter, T., Heiri, O.: Higher Late Summer Methane
1211 Emission from Central Than Northern European Lakes, *J. Limnol.*, 76 (1),
1212 <https://doi.org/10.4081/jlimnol.2016.1475>, 2016.
- 1213 Regnier, P., Dale, A.W., Arndt, S., LaRowe, D.E., Mogollón, J., Van Cappellen, P.: Quantitative analysis of
1214 anaerobic oxidation of methane (AOM) in marine sediments: A modeling perspective, *Earth-Sci. Rev.*, 106,
1215 1–2, 105–130, <https://doi.org/10.1016/j.earscirev.2011.01.002>, 2011.
- 1216 Reynolds, C. S.: The ecology of phytoplankton, Cambridge University Press, 2006
- 1217 Rosentreter, J. A., Borges, A. V., Deemer, B. R., *et al.*: Half of global methane emissions come from highly variable
1218 aquatic ecosystem sources, *Nat. Geosci.*, 14(4), 225–230, <https://doi.org/10.1038/s41561-021-00715-2>, 2021.
- 1219 Ruardij, P. & Van Raaphorst, W.: Benthic nutrient regeneration in the ERSEM ecosystem model of the North Sea,
1220 *Neth. J. Sea Res.*, 33, 3–4, 453–483, [https://doi.org/10.1016/0077-7579\(95\)90057-8](https://doi.org/10.1016/0077-7579(95)90057-8), 1995.
- 1221 Saunio, M., Bousquet, P., Poulter, B., *et al.*: The global methane budget 2000–2012, *Earth Syst. Sci. Data*, 8, 697–
1222 751, <https://doi.org/10.5194/essd-8-697-2016>, 2016.
- 1223 Saunio, M., Stavert, A. R., Poulter, B., *et al.*: The Global Methane Budget 2000–2017, *Earth Syst. Sci. Data*, 12,
1224 1561–1623, <https://doi.org/10.5194/essd-12-1561-2020>, 2020.
- 1225 Saunio, M., Martinez, A., Poulter, B., *et al.*, Global Methane Budget 2000–2020, *Earth Syst. Sci. Data Discuss.*,
1226 <https://doi.org/10.5194/essd-2024-115>, in review, 2024.
- 1227 Stavert, A. R., Saunio, M., Canadell, J. G., *et al.*: Regional trends and drivers of the global methane
1228 budget, *Glob. Change Biol.*, 28, 182–200. <https://doi.org/10.1111/gcb.15901>, 2021.
- 1229 Stepanenko, V., Mammarella, I., Ojala, A., Miettinen, H., Lykosov, V., and Vesala, T.: LAKE 2.0: a model for
1230 temperature, methane, carbon dioxide and oxygen dynamics in lakes, *Geosci. Model Dev.*, 9, 1977–2006,
1231 <https://doi.org/10.5194/gmd-9-1977-2016>, 2016.
- 1232 Tan, Z., Zhuang, Q., and Anthony, K. W.: Modeling methane emissions from arctic lakes: Model development and
1233 site-level study, *J. Adv. Model. Earth Syst.*, 7, 459–483, <https://doi.org/10.1002/2014MS000344>, 2015.
- 1234 Tan, Z., Yao, H., Melack, J., Grossart, H.-P., Jansen, J., Balathandayuthabani, S., *et al.* (). A lake biogeochemistry
1235 model for global methane emissions: Model development, site-level validation, and global applicability. *J. Adv.*
1236 *Model. Earth Syst.*, 16, e2024MS004275. <https://doi.org/10.1029/2024MS004275>, 2024.
- 1237 Thottathil, S. D., Reis, P. C. J., Prairie, Y. T.: Methane oxidation kinetics in northern freshwater
1238 lakes. *Biogeochemistry*, 143(1), 105–116. <https://www.jstor.org/stable/48701400>, 2019.
- 1239 Thullner, M., Regnier, P.: Microbial controls on the biogeochemical dynamics in the subsurface. *Rev. Mineral.*
1240 *Geochem.*, 85 (1): 265–302. <https://doi.org/10.2138/rmg.2019.85.9>, 2019.
- 1241 Van Drecht, G., Bouwman, A. F., Harrison, J., Knoop, J. M.: Global nitrogen and phosphate in urban wastewater



- 1242 for the period 1970 to 2050, *Global Biogeochem. Cycles*, 23, GB0A03,
1243 <https://doi.org/10.1029/2009GB003458>, 2009.
- 1244 Varadharajan, C.: Magnitude and spatio-temporal variability of methane emissions from a eutrophic freshwater lake,
1245 Thesis (Ph. D.)--Massachusetts Institute of Technology, Dept. of Civil and Environmental Engineering, 2009.
- 1246 Versegny, D. L., and MacKay, M. D.: Offline Implementation and Evaluation of the Canadian Small Lake Model
1247 with the Canadian Land Surface Scheme over Western Canada. *J. Hydrometeor.*, 18, 1563–
1248 1582, <https://doi.org/10.1175/JHM-D-16-0272.1>, 2017.
- 1249 Wanninkhof, R., Asher, W. E., Ho, D. T., Sweeney, C., McGillis, W. R.: Advances in quantifying air-sea gas
1250 exchange and environmental forcing, *Ann. Rev. Mar. Sci.*, 1, 213-44, [https://doi.org/](https://doi.org/10.1146/annurev.marine.010908.163742)
1251 10.1146/annurev.marine.010908.163742, 2009.
- 1252 Wetzel, R.G.: Limnology: Lake and River Ecosystems. Third Edition, Academic Press, San Diego, p389, 2001.
- 1253 William, R., Georgiy, K., Matti, L.: Basin-scale circulation and heat fluxes in ice-covered lakes, *Limnol.*
1254 *Oceanogr.*, 59, <https://doi.org/10.4319/lo.2014.59.2.0445>, 2014.
- 1255 Zhuang, Q., Guo, M., Melack, J.M., Lan, X., Tan, Z., Oh, Y., Leung, L. R.: Current and future global lake methane
1256 emissions: A process-based modeling analysis. *J. Geophys. Res.- Biogeo.*, 128,
1257 e2022JG007137, <https://doi.org/10.1029/2022JG007137>, 2023.
- 1258 Zimmermann, M., Mayr, M. J., Bouffard, D., Eugster, W., Steinsberger, T., Wehrli, B., Brand, A. Bürgmann,
1259 H.: Lake overturn as a key driver for methane oxidation, preprint, <https://doi.org/10.1101/689182>, 2019.