

The Oxygen Valve on Hydrogen Escape Since the Great Oxidation Event

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Abstract. The Great Oxidation Event (GOE) was a 200 Myr transition circa 2.4 billion years ago that converted the Earth's anoxic atmosphere to one where molecular oxygen (O₂) was abundant (volume mixing ratio > 10⁻⁴). This significant rise in O₂ is thought to have substantially throttled hydrogen (H) escape and the associated water (H₂O) loss. Atmospheric estimations from the GOE onward place O₂ concentrations ranging between 0.1% to 150% PAL, where PAL is the present atmospheric level of 21% by volume. In this study we use WACCM6, a three-dimensional Earth System Model to simulate Earth's atmosphere and predict the diffusion-limited escape rate of hydrogen due to varying O₂ post-GOE. We find that O₂ indirectly acts as a control valve on the amount of hydrogen atoms reaching the homopause in the simulations: less O₂ leads to decreased O₃ densities, reducing local tropical tropopause temperatures by up to 18 K, which increases H₂O freeze-drying and thus reduces the primary source of hydrogen in the considered scenarios. The maximum differences between all simulations in the total H mixing ratio at the homopause and the associated diffusion-limited escape rates are a factor of 3.2 and 4.7, respectively. The prescribed CH₄ mixing ratio (0.8 ppmv) sets a minimum diffusion escape rate of $\approx 2 \times 10^{10}$ mol H yr⁻¹, effectively a negligible rate when compared to pre-GOE estimates ($\sim 10^{12} - 10^{13}$ mol H yr⁻¹). Because the changes in our predicted escape rates are comparatively minor, our numerical predictions support geological evidence that the majority of Earth's hydrogen escape occurred prior to the GOE. Our work demonstrates that estimations of how the tropical tropopause layer and the associated hydrogen escape rate evolved through Earth's history requires 3D chemistry-climate models which include a global treatment of water vapour microphysics.

1 Introduction

A key component of Earth's atmospheric and biological history is its water (H₂O). The relative amount of H₂O that was present during the Earth's formation versus the quantity delivered by volatile-rich bodies (e.g., comets and asteroids; Dauphas et al., 2000; Izidoro et al., 2013), as well as the H₂O lost to space through hydrogen escape (Catling et al., 2001; Genda and Ikoma, 2008), or exchanged into the interior (Hallis et al., 2015; Korenaga et al., 2017), has had an important impact on how the Earth's surface and the atmosphere have developed. Beyond Earth, the solar system's terrestrial planet atmospheres are desiccated and consequently, Venus and Mars are both apparently desolate environments. It's possible that Venus was never habitable (Turbet et al., 2021; Warren and Kite, 2023; Constantinou et al., 2024), but Mars was likely habitable early in its history before its

25 water was isolated to the subsurface and the crust (di Achille and Hynek, 2010; Orosei et al., 2018; Scheller et al., 2021; Lauro et al., 2021; Sauterey et al., 2022; Zheng et al., 2024; Wright et al., 2024). Venus is now too hot for liquid H₂O to persist and probably lost much of its natal H₂O through atmospheric hydrogen escape (Ingersoll, 1969; Kulikov et al., 2006; Kane et al., 2019; Chaffin et al., 2023; Gu et al., 2025).

In contrast, Earth has mostly remained temperate throughout its history, with liquid water persisting on the surface for over 4 billion years (Feulner, 2012, and references therein), although there were periods of widespread glaciation at either end of the Proterozoic eon (2.4–0.541 Gyr ago; Young, 2013; Warke et al., 2020). A mostly temperate Earth has been possible despite a fainter Sun in Earth's past (Bahcall et al., 2001; Feulner, 2012; Charnay et al., 2020), in combination with fluctuations in albedo (Charnay et al., 2020; Goldblatt et al., 2021) and atmospheric composition, in particular the greenhouse gases CO₂ (Sheldon, 2006, 2013; Kanzaki and Murakami, 2015) and CH₄ (Daines and Lenton, 2016; Zhao et al., 2018; Laakso and Schrag, 2019; Fakhraee et al., 2019). Geochemical proxies suggest a broad range but generally greater past abundance for CO₂, whereas only loose constraints have been placed on CH₄ (Catling and Zahnle, 2020, and references therein), with debates regarding whether CH₄ during the Proterozoic was higher or lower in abundance still ongoing (see Discussion section).

As all known life requires liquid H₂O, the history of Earth's H₂O is inextricably tied to the evolution of Earth's life. From isotopic evidence and computational modelling, since its formation, the Earth may have lost between 0.13 and 2 times the present ocean volume (Pope et al., 2012; Kurokawa et al., 2018; Zahnle et al., 2019). Indeed, the early Archean Earth, before 3.2 Gyr ago, may have been fully covered in a global ocean (Dong et al., 2021). The current hydrogen escape rate is $\approx 4 \times 10^{10}$ mol of H yr⁻¹, corresponding to roughly 1 metre of global water loss per 1 billion years (Zahnle et al., 2013). Given that the Earth's ocean can be up to 11 km deep (Gardner et al., 2014), and its mean depth is ≈ 3.6 km (Charette and Smith, 2010), the loss rate is insignificant. This suggests a different past for hydrogen escape than the constant modern rate would imply.

Therefore, an important but uncertain component of Earth's water loss is determining the amount of hydrogen (H) that has escaped to space. As the lightest element, H escapes atmospheres more readily than other elements and chemical compounds. On Earth, hydrogen atoms exist in many different chemical species in the lower and middle atmosphere (Brasseur and Solomon, 2005). The four major chemical constituents that contribute hydrogen atoms to the upper atmosphere, and therefore the species we focus on in this work, are H, H₂, H₂O, and CH₄. Atmospheric escape of hydrogen can occur from a planet through several mechanisms, including: Jeans escape, hydrodynamic flow, sputtering, impacts, photochemical escape, and charge exchange escape (Lundin et al., 2007; Johnson, 2010; Shematovich and Marov, 2018; Gronoff et al., 2020). For these mechanisms, with the exception of impacts and hydrodynamic flow, the hydrogen escapes from the exosphere at the top of the atmosphere. In such cases, hydrogen cannot escape faster than it can be delivered to the exosphere through molecular diffusion. This is known as 'diffusion-limited hydrogen escape' (Hunten, 1973; Hunten and Donahue, 1976; Walker, 1977; Kasting and Catling, 2003; Zahnle et al., 2013). The diffusion-limited rate of hydrogen escape is set by two controlling bottlenecks, the upward diffusion of hydrogen atoms (mostly in H₂O and CH₄) through the tropopause 'cold trap', and the upward diffusion of hydrogen atoms through molecular diffusion above the homopause¹ (Catling et al., 2001; Catling and Kasting, 2017).

¹The homopause is the atmospheric point above which molecular diffusion dominates and below which turbulent diffusion dominates

The primary hydrogen bottleneck, the cold trap, is where the saturation mixing ratio of H₂O and ice is at a minimum, such that it modulates the amount of H₂O propagating upwards into the stratosphere (Kasting et al., 2015). On Earth, the cold trap is typically described as the tropical tropopause layer (TTL). The TTL is three-dimensional and is defined as the transition region between the Hadley circulation in the troposphere and the start of the deep branch of the Brewer-Dobson circulation (Brasseur and Solomon, 2005; Fueglistaler et al., 2009; Glanville and Birner, 2017). Despite being located above the tropics (at 12–16 km in altitude, which is 70–150 hPa in pressure), which have on average greater surface temperatures than higher latitudes, the TTL is defined by its cold temperatures (Brasseur and Solomon, 2005; Fueglistaler et al., 2009). The cause of this is adiabatic cooling as air is transported upwards from the convective troposphere into the stratosphere, and this cold temperature region leads to the removal of water vapor from the atmosphere through condensation (Brewer, 1949; Newell and Gould-Stewart, 1981; Danielsen, 1982; Pommereau and Held, 2007). The location, temperatures, and dynamics of the TTL are affected by seasons (Kim and Son, 2012) and climate change (Randel and Jensen, 2013; Lin et al., 2017; Zou et al., 2023; Zolghadrshojaee et al., 2024). Of the major H-bearing species, only H₂O is significantly affected by temperatures in the TTL, whereas H₂ and CH₄, which don't condense in Earth's atmosphere, are unaffected by the cold trap mechanism.

In the stratosphere, hydrogen atoms are interchanged between CH₄, H₂O and H₂, until reaching the homopause (Catling and Kasting, 2017). At altitudes exceeding the homopause, through diffusive separation, the lighter atmospheric constituents with lower molecular masses increase in relative abundance with decreasing pressure (e.g., H, He, O, and N). Note that H₂ would also increase but it is rapidly photodissociated into H. Hence, above the homopause, H dominates the total hydrogen budget (Catling and Kasting, 2017).

The Archean (4–2.4 Gyr ago) Earth was likely weakly reducing with greater quantities of CH₄, CO, and H₂ (Kasting et al., 2001; Kharecha et al., 2005; Kadoya and Catling, 2019; Catling and Zahnle, 2020). Midway through Earth's history, approximately 2.4 billion years ago, a rise in oxygen known as the Great Oxidation Event (GOE; Farquhar et al., 2000; Holland, 2002; Hodgskiss et al., 2019; Warke et al., 2020; Goldblatt et al., 2021) has been proposed to have significantly reduced hydrogen escape from $\sim 10^{12} - 10^{13}$ mol H yr⁻¹ to $\sim 10^{10}$ mol H yr⁻¹ (Catling et al., 2001; Zahnle et al., 2013). This is because an oxygenated atmosphere destroys reducing species (e.g., H₂ and CH₄), resulting in lower amounts of such gases which contribute hydrogen atoms to the upper atmosphere.

In conjunction, atmospheric escape of hydrogen may have played a significant role in shaping Earth's atmosphere through time. It is thought to at least be partially responsible for oxidising the Earth's surface: when H₂O is photodissociated above the troposphere, the liberated hydrogen (H) can escape to space, and the oxygen (O) left behind oxidises the Earth (Catling et al., 2001; Claire et al., 2006; Zahnle et al., 2013). Alongside photosynthetic production of O₂ from cyanobacteria, this process is hypothesized to have contributed to the occurrence of the GOE because the process of hydrogen escape and subsequent oxidation is irreversible, leading to a reduced sink of O₂ over time (Catling et al., 2001; Claire et al., 2006; Zahnle et al., 2013). Nevertheless, considerable uncertainty persists regarding the precise temporal evolution of Earth's atmosphere (Kasting, 2025), with hydrogen escape a key aspect of the holistic picture.

There is no direct way to determine the amount of hydrogen that has escaped in Earth's history. As mentioned, several measurements can be used to infer this quantity, and the chronology of one of the heaviest gases, Xe, may be informative.

Xe is more easily ionised compared to the other noble gases which do not exhibit the same fractionation (the so-called Xenon paradox; Anders and Owen, 1977; Zhu et al., 2013). It was proposed that Xe^+ could escape Earth's atmosphere when propelled by H^+ ions, and this has been shown to be a feasible physical process (Zahnle et al., 2019). Accordingly, the timing of Earth's Xe fractionation (see figure 3 in Zahnle et al., 2019) may trace high levels of atmospheric hydrogen escape during the Archean eon. There has been relatively small changes in Xe fractionation since the GOE (Avicé et al., 2018), indicating the vast majority of historical hydrogen escape occurred prior to this time (i.e., before ~ 2.4 Gyr ago).

100 Whilst hydrogen escape rates may have been much lower in the last ~ 2.4 Gyr, to our knowledge, no work has quantitatively estimated escape rates since the GOE occurred using a 3D model. In this work, we build upon a previous study (Cooke et al., 2022) which simulated a range of O_2 concentrations that may have been present on the Earth for the last 2.4 Gyr and found that lower O_3 columns were predicted compared to previous work (Segura et al., 2003; Rugheimer et al., 2015; Rugheimer and Kaltenecker, 2018; Kaltenecker et al., 2020; Kozakis et al., 2022; Ji et al., 2023). More recent work has in part supported the results of lower O_3 for a given O_2 mixing ratio (Yassin Jaziri et al., 2022; Ji et al., 2024; Liu et al., 2025). O_3 and its associated UV heating is responsible for the temperature inversion that creates the Earth's stratosphere, and this heating affects the temperatures in the TTL. Therefore, the O_2 fluctuations following the GOE may have modulated the upward transport of H_2O through the TTL and consequently, hydrogen escape too.

Using the 3D chemistry-climate model WACCM6, we first explore the affect of changing the O_2 abundance on the TTL region, before investigating the influence it may have had on hydrogen escape during the Proterozoic eon. Our aim is to estimate how changes to atmospheric O_2 alone can affect the diffusion-limited hydrogen escape rate. In terms of hydrogen escape fluctuations, our focus is on H_2O , and we hold the mixing ratios of H_2 and CH_4 constant at the surface. We note that CO_2 , CH_4 and H_2 are likely to have varied during the past 2.4 billion years (Catling and Zahnle, 2020), as will have the luminosity of the Sun (Feulner, 2012). These variables are important because CH_4 and H_2 are key hydrogen carriers to the upper atmosphere, and additionally CH_4 and CO_2 provide greenhouse warming which can affect TTL temperatures. The TTL will also have been affected from the reduced total insolation from a younger Sun which would act to cool the troposphere. The purpose of this work is, however, to present the mechanism by which O_2 affects hydrogen escape. We calculate such H escape deviations, paving the way for future work on the topic to investigate other physical, chemical, and biological influences.

2 Numerical methods

2.1 WACCM6 model

120 The simulations we analyse are those from Cooke et al. (2022), where the Earth System Model WACCM6 (Gettelman et al., 2019) was employed to model different oxygenation levels. WACCM6 is a configuration of the Coupled Earth System Model (CESM), and we use version 2.1.3 (CESM2.1.3²) with minor code alterations in the initial conditions³, which we describe in this section.

²<https://www.cesm.ucar.edu/models/cesm2>

³ExoCESM GitHub: https://github.com/exo-cesm/CESM2.1.3/tree/main/O2_Earth_analogues

The first simulation we perform commences with initial conditions which are intended to be representative of Earth's pre-
 125 industrial (PI) atmosphere circa 1850, alongside contemporary ocean and land settings. The model's spatial resolution was set
 at 1.875° by 2.5° (96 latitude points and 144 longitude points), encompassing 70 vertical atmospheric layers spanning from
 1000 hPa to 4.5×10^{-6} hPa (which is up to approximately 150 km in altitude). **In total, eight simulations were performed for the
 Earth system with different mixing ratios of atmospheric O_2 , where the atmospheric pressure was kept the same by changing
 the abundance of the background gas, N_2 . O_2 mixing ratios are given in terms of present atmospheric level (PAL), which is 21%
 130 by volume. The O_2 surface mixing ratios span values between 0.1–150% PAL, which are within estimates for the Earth over the
 last 2.4 billion years (Lyons et al., 2014; Large et al., 2019; Catling and Zahnle, 2020; Steadman et al., 2020). **These temporal
 estimates do not always agree, so we remain agnostic and perform simulations covering the whole range. The simulation names
 are PI, 150% PAL, 50% PAL, 10% PAL, 5% PAL, 1% PAL, 0.5% PAL, and 0.1% PAL. The mixing ratios or fluxes of other
 gases specified at the surface were kept constant as O_2 was altered. This assumption is likely not realistic for the entirety of the
 135 last 2.4 Gyr, but it allows us to isolate the role O_2 changes make to the chemistry and dynamics of the atmosphere. Water vapor
 feedback, in terms of evaporation and rainout, is included in WACCM6, but hydrogen escape and water loss are not explicitly
 simulated. Instead, we estimate these values once quasi-steady state has been reached in the simulation (see section 2.2).****

Both atmospheric and oceanic components of the model are interactive, allowing dynamic responses to environmental stimuli
 like temperature changes. Our simulations incorporated middle atmosphere chemistry as described by Emmons et al. (2020),
 140 featuring 98 chemical species and 298 reactions, including O_3 chemistry. The atmospheric time step, Δt , was set at 30 minutes.
 The concentrations of 75 species were computed implicitly, while 22 long-lived species were computed explicitly (Sandu et al.,
 1997; Brasseur and Solomon, 2005) and N_2 is considered invariant.

We average the outputs of the WACCM6 simulations over the last four years of the simulation once the middle atmospheric
 (defined here as between 12–100 km in altitude; Brasseur and Solomon, 2005) trend in total hydrogen atoms has halted, and
 145 this required between 30 and 90 simulation years for the different scenarios.

2.2 Diffusion-limited hydrogen escape rate

The diffusion-limited hydrogen escape rate (Hunten, 1973; Kasting and Catling, 2003), Φ_{esc} , **is proportional to the total mixing
 ratio of hydrogen-bearing species, $f_T(\text{H})$, at the homopause:**

$$\Phi_{\text{esc}} \propto f_T(\text{H}), \tag{1}$$

150 where $f_T(\text{H})$, can be written as

$$f_T(\text{H}) = f(\text{H}) + 2f(\text{H}_2) + 2f(\text{H}_2\text{O}) + 4f(\text{CH}_4)\dots, \tag{2}$$

and so on, accounting for all hydrogen-bearing species. **We focus on evaluating the abundances of H, H_2 , H_2O , and CH_4 . In
 modern Earth's atmosphere, at the tropopause, the H_2 and CH_4 mixing ratios are ≈ 0.55 ppmv (Catling and Kasting, 2017) and**

155 ≈ 1.9 ppmv (Saunois et al., 2025), respectively. There, the H_2O mixing ratio varies between $\approx 2.5 - 4.5$ ppmv, with a mean
 stratospheric entry value of $\sim 3.5 - 4$ ppmv (Fueglistaler et al., 2009; Gettelman et al., 2019). This puts the stratospheric entry
 value of total hydrogen at ~ 17 ppmv, although with CH_4 at a lower concentration during the pre-industrial period, the entry
 value would be closer to ~ 12 ppmv. In our simulations, the lower boundary mixing ratios are instead mixing ratios of 0.5
 160 ppmv and 0.8 ppmv for H_2 and CH_4 , respectively, corresponding to pre-industrial values. Consequently, H_2O is the primary
 carrier of hydrogen atoms to the middle atmosphere. Other H-bearing species such as OH, HO_2 , and H_2O_2 are included in the
 WACCM6 model but are not sufficiently abundant to contribute significantly to the overall hydrogen budget.

The calculation of Φ_{esc} at the homopause can be made by considering the binary diffusion coefficients which depend on the
 temperature of the atmosphere and the chemical species. Φ_{esc} , in units of $\text{atoms cm}^{-2} \text{ s}^{-1}$, is given by

$$\Phi_{\text{esc}} = \sum_{i=1}^n \frac{N_i f(i) b_i}{H_h}, \quad (3)$$

165 where H_h is the atmospheric scale height at the homopause, N_i is the number of hydrogen atoms in each chemical species,
 i , and b_i is the binary diffusion parameter for each constituent. The binary diffusion parameters we use are from Hunten
 (1973) and Banks and Kockarts (1973). The binary diffusion coefficient depends on the temperature of the atmosphere at the

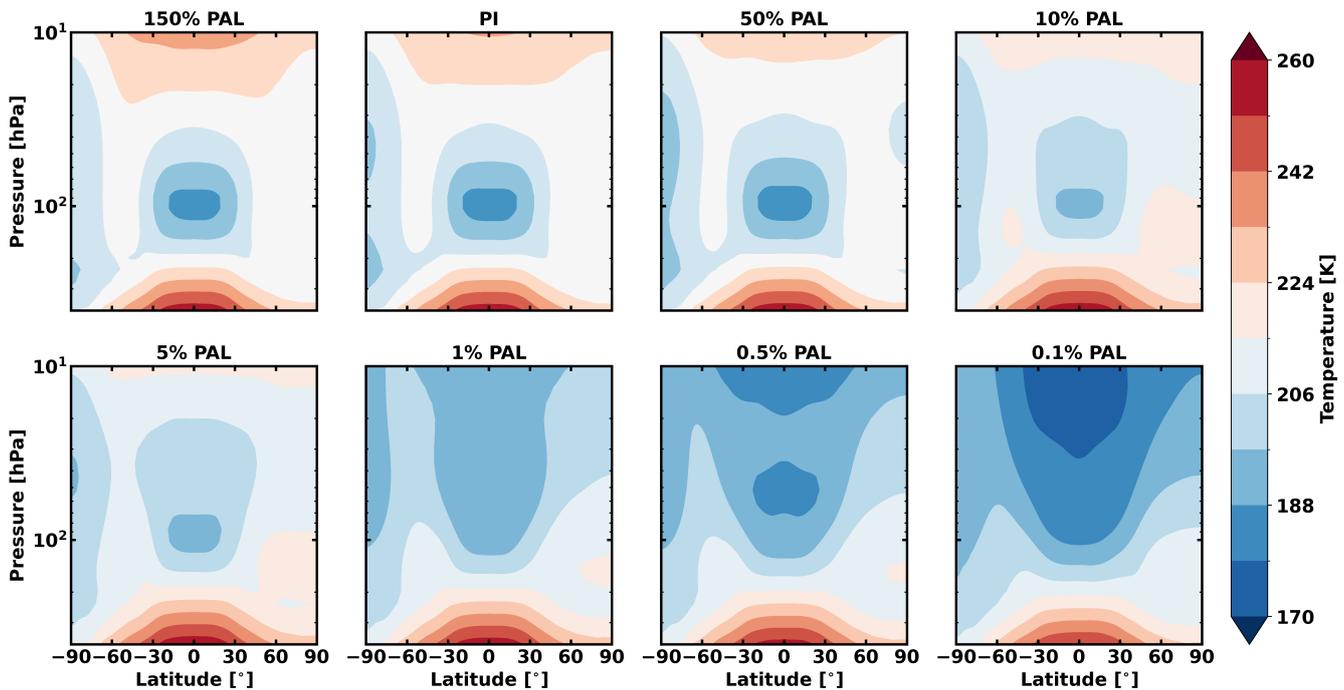


Figure 1. The zonal mean of temperature (in K) is shown for O_2 mixing ratios between 150% PAL and 0.1% PAL, where O_2 decreases in
 abundance from the top left panel to the bottom right panel. The colour bar shows hotter temperatures in red and colder temperatures in blue.

homopause (T_h), the masses of the species, and their collision cross-sections. For each species considered here, the binary diffusion coefficients are $b_H = 6.5 \times 10^{17} T_h^{0.7}$, $b_{H_2} = 2.67 \times 10^{17} T_h^{0.75}$, $b_{H_2O} = 0.137 \times 10^{17} T_h^{1.072}$, $b_{CH_4} = 0.756 \times 10^{17} T_h^{0.747}$. The calculations are made for the whole atmosphere (latitude and longitude at the homopause) before calculating a longitudinally-averaged and latitudinally-weighted mean.

From Earth's exosphere, hydrogen escapes faster than it is delivered, such that throughout this work, we will assume that the loss rate of hydrogen atoms is diffusion-limited (ignoring hydrodynamic escape and impacts). Note that it is possible for hydrogen to escape slower than this upper limit, provided the rate of loss from the exosphere is the limiting factor (e.g., energy-limited escape, photon-limited escape; Tian et al., 2005; Salz et al., 2016; Owen and Alvarez, 2016; Lehmer and Catling, 2017).

3 Results

As part of the hydrological cycle, liquid water evaporates from the surface and warm, moist air rises due to buoyancy. Hence, rivers, lakes, and predominantly the ocean, act as a sources of hydrogen atoms to the atmosphere. We will present the results in terms of how the total hydrogen mixing ratio starts at the surface and how it changes with decreasing atmospheric pressure (increasing altitude). This includes tracking the proportion of hydrogen atoms bound in different species, such as H_2O and CH_4 . We will show how the O_2 mixing ratio affects the total hydrogen mixing ratio through the atmosphere and ultimately the amount of hydrogen atoms diffusing up to the exosphere. In the following subsection, we consider how O_2 affects clouds, condensation, and the cold trap at the top of the troposphere. These processes act to keep hydrogen atoms in the troposphere that would otherwise mix into the stratosphere and upwards to eventually escape to space.

3.1 Water vapour through the cold trap

In every simulation, the global mean mixing ratio of each species at the surface for H_2O , CH_4 , H_2 , and H is $\sim 10,000$ ppmv, ≈ 0.8 ppmv, ≈ 0.5 ppmv and $< 10^{-8}$ ppmv respectively. As temperature and pressure decline with altitude in the troposphere, the atmosphere cannot hold as much H_2O vapor. This is because the drop in pressure causes the air to cool (adiabatic cooling), and the saturation vapor pressure of water, which determines the air's capacity for water vapor, is strongly dependent on temperature. Fig. 1 presents the zonal mean (averaged over longitude) of the atmospheric temperature between the surface and 10 hPa (~ 30 km in altitude) for all the WACCM6 simulations. The temperature structure of the atmosphere is set by the strength and spectral shape of the incoming radiation, the moist adiabat in the troposphere, atmospheric composition (providing heating and cooling in different layers), clouds and ice, and atmospheric dynamics. Between 150% PAL and 50% PAL, the temperature structure at pressures greater than (altitudes below) 10 hPa is similar. At 10% and 5% PAL, the temperature increases in the middle atmosphere, and this is noticeable in the tropics between 200–30 hPa and at northern ($45-90^\circ$) latitudes between 200–100 hPa. At $\leq 1\%$ PAL of O_2 , the stratosphere begins to become significantly cooler, with the 10 hPa pressure level ~ 60 K cooler in the 0.1% PAL simulation compared to the 150% PAL simulation. A globally averaged stratospheric

temperature inversion of ≈ 2 K and ≈ 1 K exists in the 1% PAL and 0.1% PAL simulations, respectively, which is significantly smaller than the ≈ 60 K temperature inversion in the PI atmosphere.

200 Fig. 2 shows the mixing ratio of atmospheric O_2 plotted against the O_3 column density, as well as the mean O_3 mixing ratio between 50 and 450 hPa in the tropical region. Whilst the global O_3 column density increases with increasing O_2 mixing ratio, this is not the case for the O_3 mixing ratio around the TTL region: there is instead a peak at $\approx 10\%$ PAL. O_3 in this region absorbs UV light and heats the TTL, which is the cause of the hotter temperatures in the higher oxygenated scenarios in Fig. 1 compared to scenarios with $< 1\%$ PAL. This is also why there is an asymmetry in temperatures because of hemispherical
 205 asymmetries in the O_3 distribution (there is more O_3 in the northern hemisphere) as a result of the Brewer-Dobson circulation (Dobson and Harrison, 1926; Brewer, 1949; Butchart, 2014).

In Section 1, we described how H_2O moves through the TTL to get into the stratosphere on modern Earth. The cold trap mechanism is indirectly sensitive to O_2 concentrations resulting from the aforementioned difference in O_3 and hence UV heating. Due to the difference in time-averaged zonal mean TTL temperatures between the PI and 0.1% PAL simulations of
 210 up to ≈ 18 K, more H_2O is frozen out in the form of water clouds, ice, and ice clouds before crossing ≈ 100 hPa in the 0.1% PAL case compared to the standard PI case (Wang and Dessler, 2012). Stratospheric O_3 and H_2O are greenhouse gases, and with less H_2O reaching the stratosphere as a consequence of reduced O_3 , this decreases tropospheric heating (Deitrick and Goldblatt, 2023), causing a positive feedback effect, albeit a small one: there is some cooling at the surface, but this is limited to < 3 K in terms of global averaged surface temperatures.

215 The fractional coverage of high clouds (high clouds are defined as clouds at pressures < 400 hPa and are therefore relevant to the TTL) versus latitude is shown in Fig. 3. The coverage of high clouds for the PI, 1% PAL, and 0.1% PAL is also shown in

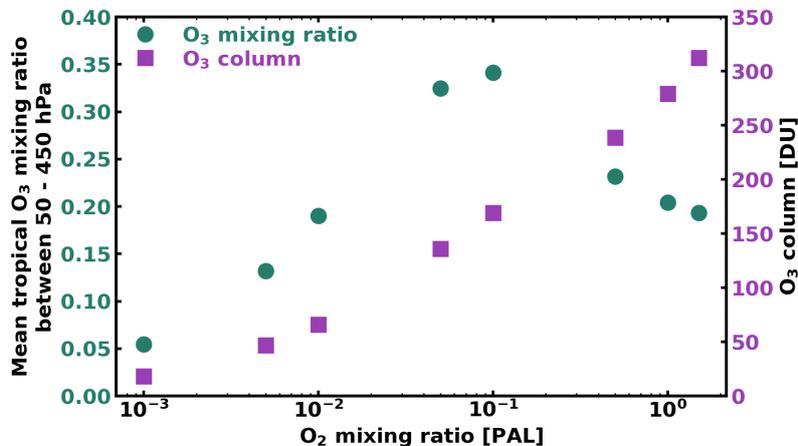


Figure 2. The mean tropical (defined at latitudes $\pm 24^\circ$ from the equator) O_3 mixing ratio between 50–450 hPa is plotted on the left vertical axis in teal circles against the atmospheric mixing ratio of O_2 at the surface in terms in terms of the present atmospheric level (PAL), which is 21% by volume. The total global ozone (O_3) column density in Dobson Units (DU, where $1 \text{ DU} = 2.69 \times 10^{20} \text{ molecules m}^{-2}$) is also plotted in purple squares on the right vertical axis against the atmospheric mixing ratio of O_2 .

the same figure. The O₂ mixing ratio affects high cloud coverage across the entire planet, but especially in the tropics. There is a $\approx 30\%$ increase in the peak high cloud fraction at the equator (0° latitude) from the 150% PAL simulation to the 0.1% PAL simulation, although the peak high cloud fraction for the simulations between the O₂ concentrations of 150% PAL to 1% PAL at the equator all fall within 6% of each other. Low clouds (surface–700 hPa) and medium clouds (700–400 hPa) exhibit very little deviation between the simulations.

The total hydrogen mixing ratio bound in ice (ice water content and ice clouds) and H₂O for the PI and 0.1% PAL cases is shown in Fig. 4 between -60° to $+60^\circ$ latitude. At 142 hPa (roughly 14 km in altitude), the 0.1% PAL scenario has a greater amount of ice content and a lower amount of H₂O. Thus, due to the freeze-drying of the atmosphere, fewer hydrogen atoms are able to contribute to the stratospheric total hydrogen mixing ratio, $f_T(\text{H})$, when compared to the PI case. Both simulations show a larger build-up of total ice water, ice clouds, and H₂O content over the western pacific (e.g., between $90-150^\circ$), as is found on modern Earth (Newell and Gould-Stewart, 1981; Dessler, 1998; Dong et al., 2020).

In the cases we have simulated, the interplay between UV radiation, O₂ mixing ratio, and 3D transport, is especially important for predicting the O₃ distribution. The amount of O₃ and the incoming UV affect temperatures around the tropopause, and ultimately how much H₂O can reach the stratosphere. Here we are exploring the impact of O₂ levels alone on the TTL temperatures and thus atmospheric escape. Other factors will have influenced TTL temperatures for the past 2.4 billion years, and these will be discussed in Section 4.

3.2 Tropical tape recorder

The tropical “tape recorder” (Mote et al., 1995; Liu et al., 2007; Glanville and Birner, 2017) for H₂O is the phenomenon where seasonal temperature fluctuations in the TTL modulate the mixing ratio of H₂O that enters the stratosphere over an annual cycle. The temperature of the TTL itself is affected by several processes, including radiative heating and cooling, deep convection, sea surface temperatures, and the Quasi-Biennial Oscillation (Fueglistaler et al., 2009; Paulik and Birner, 2012; Garfinkel et al., 2013; Tegtmeier et al., 2020). Note that the tape recorder effect exists for some other molecules too, including CO (Schoeberl et al., 2006; Liu et al., 2007; Pumphrey et al., 2008). The Brewer-Dobson circulation (Dobson and Harrison, 1926; Brewer, 1949; Butchart, 2014), over a period of months, transports H₂O upward (Glanville and Birner, 2017), causing the distinct tape recorder effect (WACCM6 is able to generate this effect - see the PI simulation in Fig. 5 and also Gettelman et al., 2019). The seasonally-varying temperatures in the TTL imprint a record on the observed mixing ratio of H₂O which is altered in the middle atmosphere in the subsequent months.

In the PI simulation, the tape recorder is easily visible over a 4-year time frame with a period of 1 year (see Fig. 5). For the PI case, the peak seasonal differences in the tape recorder effect are of order 1.5 ppmv at ≈ 90 hPa. Note that this amplitude is much smaller than the tape recorder variation Liu et al. (2023) found for an Earth-like exoplanet with a higher eccentricity of 0.4.

At low oxygenation states ($< 1\%$ PAL), seasonal differences in temperature and the Brewer-Dobson circulation are still present, but their magnitude is smaller so the tropical tape recorder effect is muted. Whilst the H₂O mixing ratios still vary, the tape recorder effect is no longer visible from the scale shown in Fig. 5 at 0.1% PAL because there is no clear periodic signal.

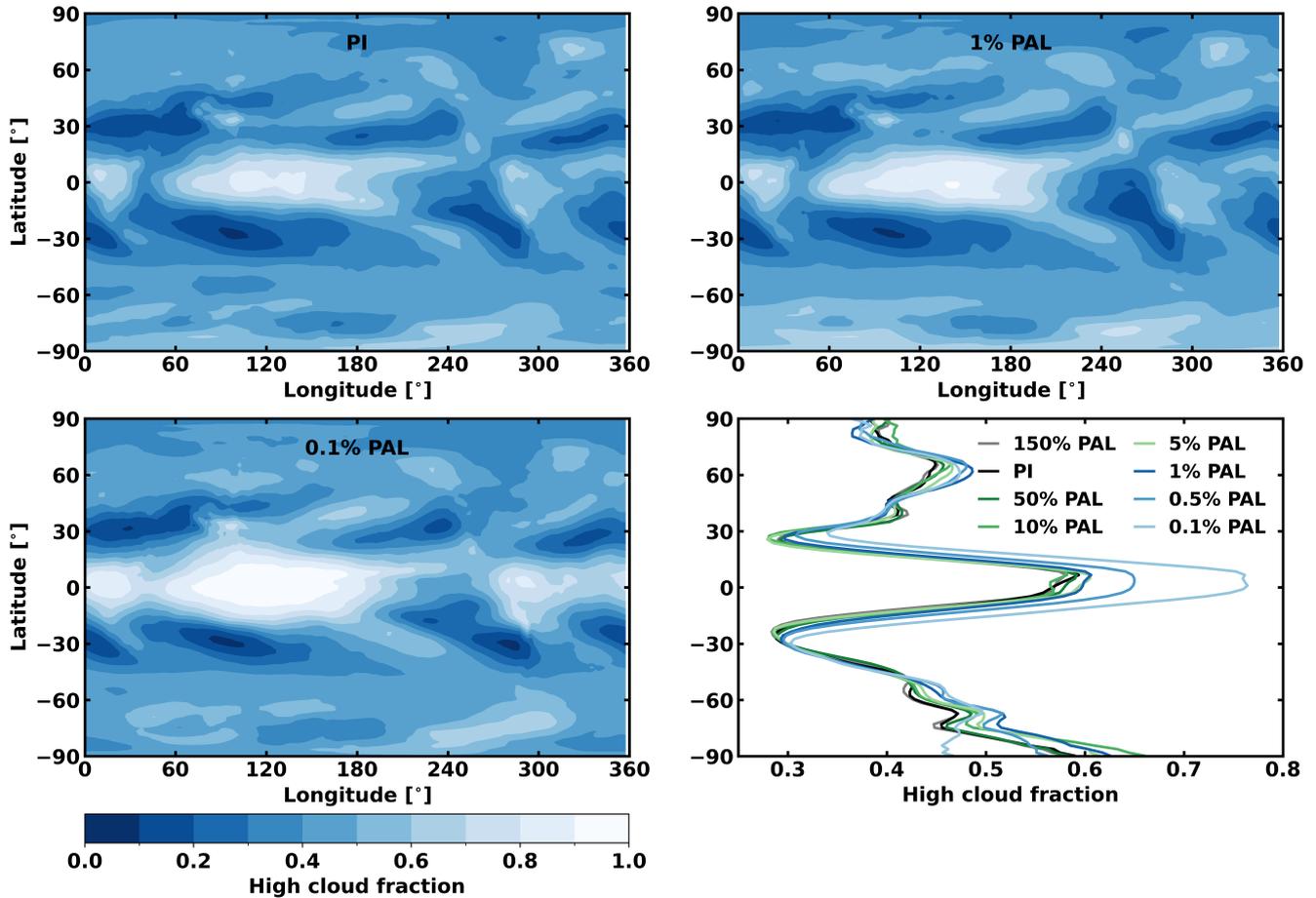


Figure 3. The **top left**, **top right**, and **bottom left** panels show the high cloud fraction as a function of longitude and latitude for the PI, 1% PAL, and 0.1% PAL scenarios, respectively. **Bottom right:** The high cloud fraction (high clouds are defined as clouds at pressures < 400 hPa) is averaged over longitude and shown as a function of latitude in the top panel for all the WACCM6 simulations.

We verified that the PI simulation has a tape recorder and the 0.1% PAL simulation does not have one using a Fourier analysis. Therefore, the seasonal mechanism which transports more water vapour into the stratosphere during warmer periods in the TTL is no longer effective in the 0.1% PAL case.

3.3 Total hydrogen mixing ratio

255 After H₂O passes through the cold trap, the first bottleneck for the total hydrogen mixing ratio of temperate Earth-like atmospheres has been passed. We now assess the quantitative effect of this bottleneck, and the relative contributions of each of the four major hydrogen bearing species in our simulations.

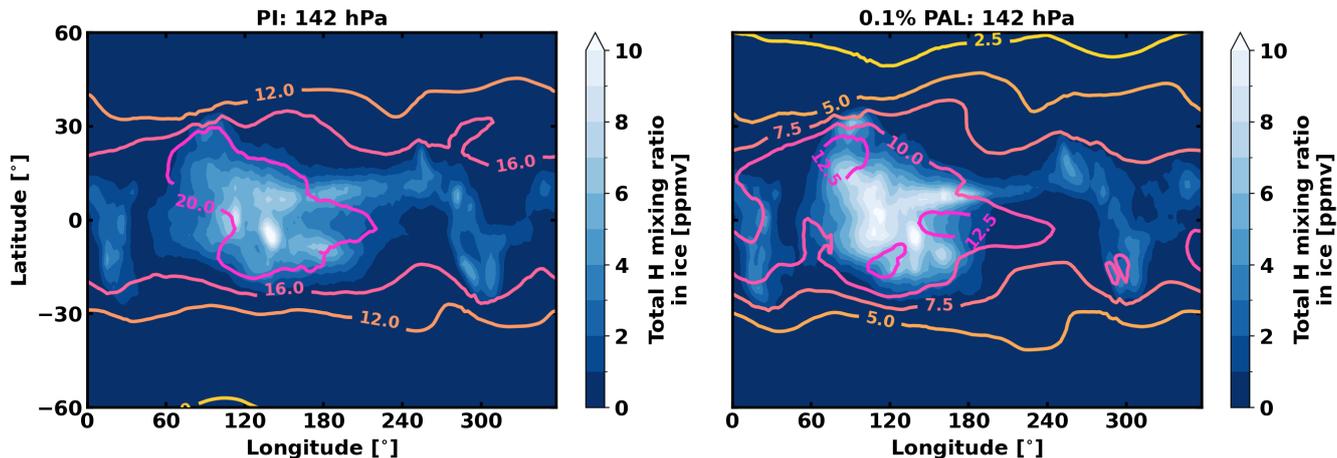


Figure 4. The atmospheric ice content (blue-white shading) and the amount of atmospheric H₂O (coloured contours) is shown on the latitude-longitude grid of the Earth. Both are given in terms of ppmv. The PI simulation (left) and the 0.1% PAL simulation are shown (right). Both scenarios are shown for a pressure level of 142 hPa, which corresponds to an altitude of ≈ 14 km near the base of the TTL.

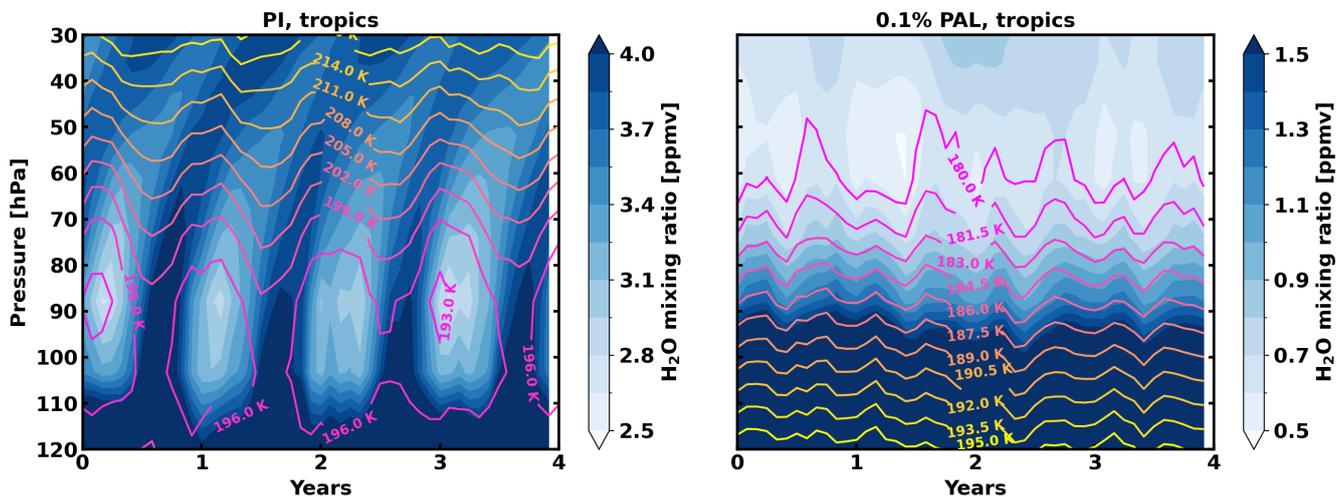


Figure 5. The water vapour mixing ratio in ppmv is shown for the PI scenario (left) and the 0.1% PAL scenario (right) between pressures of 120 – 30 hPa. Each panel shows four years in a row of each simulation, displaying the zonally averaged and latitudinally weighted H₂O mixing ratio in the tropics between $\pm 24^\circ$. Whites show the lowest mixing ratios and darker blues show progressively larger mixing ratios. Note the different scales on each colour bar. The contours show the atmospheric temperature in kelvin, with yellow indicating higher temperatures and magenta indicating lower temperatures.

Fig. 6 displays the warmest atmospheric temperature reached in each simulation at a pressure of 88 hPa (≈ 18 km for Earth's atmosphere)⁴ in the tropics (within $\pm 24^\circ$ from the equator) against the globally averaged $f_T(H)$ at the same pressure

⁴88 hPa is the pressure level in our simulations closest to the winter climatic tropopause of 86-88 hPa, although note that the summer climatic tropopause is closer to 100 hPa (Kim and Son, 2012; Pilch Kedzierski et al., 2016)

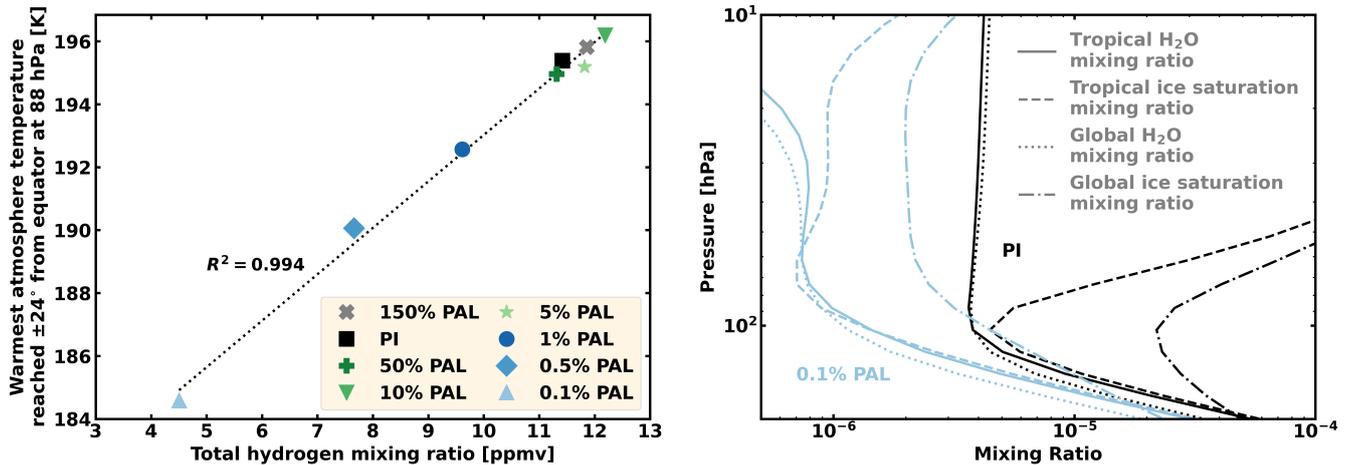


Figure 6. Left: The globally averaged total hydrogen mixing ratio ($f_T(\text{H})$) at 50 hPa is plotted against the warmest temperature reached within $\pm 24^\circ$ (tropical latitudes) from the equator at 88 hPa for each simulation. **Linear regression gives a coefficient of determination (R^2) of 0.994, which is shown by the dotted black line.** The different O₂ mixing ratios are shown by various markers: 150% PAL (grey cross), PI (black square), 50% PAL (dark green plus), 10% PAL (green downward triangle), 5% PAL (light green star), 1% PAL (dark blue circle), 0.5% PAL (blue diamond), and 0.1% PAL (light blue upward triangle). **Right:** The H₂O mixing ratio is plotted against pressure for the PI (black) and 0.1% PAL (light blue) simulations, in terms of its tropical (unbroken line) and global (dotted line) average. Also shown is the ice saturation mixing ratio which depends on temperature and pressure and is shown for the tropical (dashed) and global (dash-dotted) average. The tropical average is for latitudes $\pm 24^\circ$ from the equator.

260 level. There is a **strong positive correlation with a coefficient of determination of $R^2 = 0.994$** between these two variables. **This freeze-drying relationship arises because the warmest temperatures in the tropical atmosphere effectively control the maximum amount of water vapor that can be transported upward through to the stratosphere.** As a result, there is a **weaker correlation** present when comparing $f_T(\text{H})$ with the global mean temperature **at the same pressure level, and worse correlation still when comparing to the warmest temperature found anywhere at 88 hPa.**

265 **More importantly,** Fig. 6 also demonstrates how it is the ice saturation vapor pressure (Murphy and Koop, 2005) in the tropical atmosphere that acts to limit the transport of H₂O upwards. The tropical ice saturation mixing ratio curve is within a factor of 1.18 and 1.01 when compared to the globally averaged H₂O mixing ratio at 100 hPa (**≈ 16 km in altitude**) for the PI and 0.1% PAL atmospheres, respectively. When comparing the global ice saturation mixing ratio curve, the discrepancy is a factor of 5.8 and 3.9 times, respectively. This means that 1D models that are based on globally averaged temperatures may
 270 overestimate the stratospheric H₂O abundance for Earth-like oxygenated simulations by up to a factor of 6 (accounting for all simulations here, the discrepancy range is 3.8–6.0).

Fig. 6 suggests that the warmest TTL temperatures are the controlling factor in each atmospheric scenario, instead of the atmospheric composition. Yet because composition (i.e., the O₂ mixing ratio) is the variable that is altered in each scenario, the atmospheric composition is the controlling factor for the TTL temperatures. Hence, the oxygenation state of the atmosphere is

275 indirectly controlling the upward flow of hydrogen atoms and affecting the diffusion-limited hydrogen escape rate, and this is what we refer to as the ‘oxygen valve’.

Fig. 7 presents the globally averaged $f_T(\text{H})$ vertical profile throughout the atmosphere, which is calculated from Eq. 2 for all the simulations. The mixing ratio of total hydrogen in the stratosphere (at 1 hPa) is $f_T(\text{H}) = 11.7$ ppmv in the PI simulation. Turbulent mixing causes $f_T(\text{H})$ to remain roughly constant between the lower stratosphere and homopause. For the 10%
280 PAL, 1% PAL, and 0.1% PAL simulations, the stratospheric $f_T(\text{H})$ values at 1 hPa are approximately 12.4, 6.9, 3.0 ppmv, respectively.

Fig. 7 also shows the total hydrogen contribution from the four species that carry the majority of hydrogen atoms (H, H₂, H₂O, and CH₄). These panels show that for the PI simulation (black lines), H₂O is the primary carrier of hydrogen atoms in the lower atmosphere. Above the cold trap in the TTL, the H₂O mixing ratio increases until it reaches a maximum
285 mesospheric value (≈ 5 ppmv) due to CH₄ reacting with OH. At a pressure of approximately 0.01 hPa, H₂ then becomes dominant. Photolysis of H₂O, CH₄, and H₂, as well as diffusive separation, cause atomic H to become the primary carrier at pressures less than 10^{-4} hPa.

Decreasing levels of O₂ act to shift the pressures at which particular species are the principal hydrogen carrier. H₂ begins to increase in abundance at lower altitudes as the atmospheric O₂ concentration is decreased. For instance, with 1000 times
290 less O₂ (0.1% PAL case, light blue lines), H₂ is dominant at pressures less than 10 hPa, due to both H₂O condensation and photolysis at the higher pressure levels. Again, resulting from diffusive separation, H always dominates towards the top of the model in the lower thermosphere. CH₄ is increasingly lost in the troposphere due to enhanced H₂O photolysis and OH production, acting to reduce CH₄ lifetimes (see the following papers for a more in depth discussion on CH₄ lifetimes: Cooke et al., 2022; Ji et al., 2023; Ji et al., 2024). This loss adds to tropospheric H₂O which is ultimately halted by the cold trap. If
295 the CH₄ mixing ratio at the TTL is greater than half that of H₂O, then CH₄ is the primary contributor to hydrogen escape. However, CH₄ is never the main H carrier in the scenarios we present. As we will discuss later, it is unclear whether this would have been the case for some portions of the Proterozoic.

3.4 Diffusion-limited escape

Now we have assessed $f_T(\text{H})$ throughout the atmosphere we can predict how the rate of diffusion-limited hydrogen escape
300 varies between the different simulated atmospheres with WACCM6. We present the predicted escape rates for each simulation in Table 1.

The $f_T(\text{H})$ profile shows that for the cases with $\leq 1\%$ PAL of O₂, the rate of hydrogen escape is lower than that found in the PI case ($\Phi_{\text{esc,PI}}$). The rates for the 0.1% PAL, 0.5% PAL, and 1% PAL simulations are: $0.26 \Phi_{\text{esc,PI}}$, $0.57 \Phi_{\text{esc,PI}}$, and $0.68 \Phi_{\text{esc,PI}}$,
305 respectively. We also predict that the escape rate could have been slightly higher when O₂ levels were both greater (e.g. 150% PAL) and lesser (e.g. 5–10% PAL) than the present day value, with the maximum predicted hydrogen escape rate in the 10% PAL simulation at $1.20 \Phi_{\text{esc,PI}}$. This shows that the TTL heating and predicted escape rate has a non-linear behavior with O₂ concentration. Following the Great Oxidation Event, Earth’s O₂ levels are thought to have fluctuated between 0.1% PAL and 150% PAL, although the precise timings and magnitude of these fluctuations are debated (Catling and Zahnle, 2020; Steadman

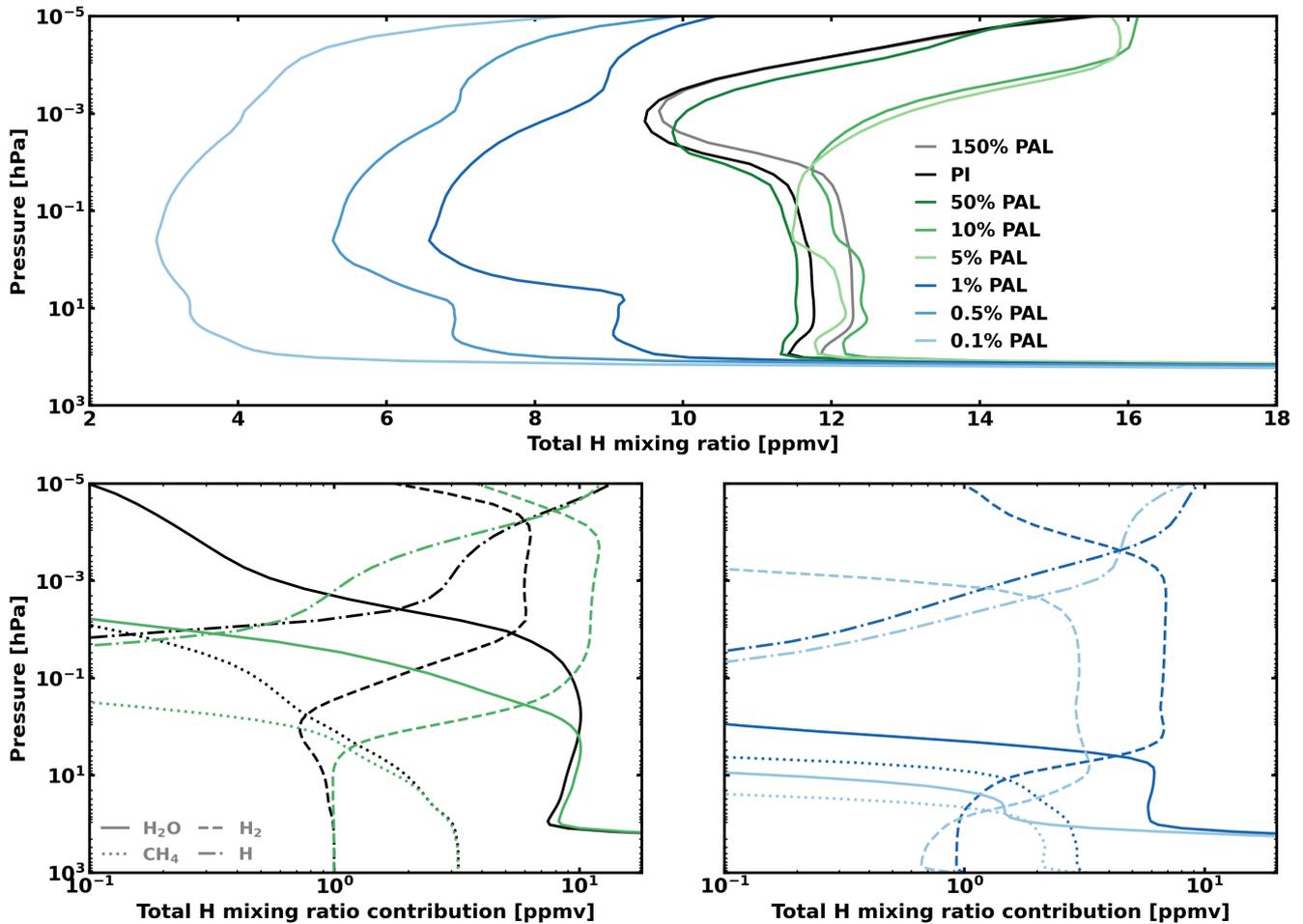


Figure 7. Top: $f_T(H)$, the total hydrogen mixing ratio is given in parts per million by volume (ppmv) and is shown against pressure for the 150% PAL (grey), PI (black), 50% PAL (dark green), 10% PAL (green), 5% PAL (light green), 1% PAL (dark blue), 0.5% PAL (blue), and 0.1% PAL (light blue) simulations. $f_T(H)$ is calculated using Eq. (2). Hence, the mixing ratio contribution from H, H₂O, H₂, and CH₄ is multiplied by a factor of 1, 2, 2, and 4, respectively. **Bottom left:** The total hydrogen mixing ratio contribution from H (dash dotted), H₂ (dashed), H₂O (unbroken), and CH₄ (dotted) is shown for the PI and 10% PAL simulations and **bottom right:** the 1% PAL and 0.1% PAL simulations. In Eq. (2), the mixing ratios of H₂, H₂O, and CH₄, are multiplied by 2, 2, and 4, respectively, to determine their contribution to $f_T(H)$.

et al., 2020; Planavsky et al., 2020; Yierpan et al., 2020; Krause et al., 2022; Mills et al., 2023; Xie et al., 2023; Fakhraee
 310 and Planavsky, 2024; Stockey et al., 2024), with these fluctuations being due to various physical factors, such as carbonate
 amassing in the crust (Alcott et al., 2024), and large igneous province volcanism combined with weathering (Luo et al., 2024).

Table 1. This table shows the eight simulations where the mixing ratio of O₂, given in terms of present atmospheric level (PAL), was varied. The total hydrogen mixing ratio, $f_T(\text{H})$, in ppmv at the homopause, which is taken to be at a pressure of 10^{-3} hPa. The diffusion-limited hydrogen escape rate is given by Φ_{esc} in mol yr⁻¹ and compared to the predicted escape rate in the PI simulation ($\Phi_{\text{esc,PI}}$).

Simulation	O ₂ [PAL]	Homopause $f_T(\text{H})$ [ppmv]	Φ_{esc} [mol yr ⁻¹]	$\Phi_{\text{esc}}/\Phi_{\text{esc,PI}}$
150% PAL	1.500	9.68	7.71×10^{10}	1.02
PI	1.000	9.52	7.54×10^{10}	1.00
50% PAL	0.500	10.05	7.79×10^{10}	1.03
10% PAL	0.100	12.74	9.04×10^{10}	1.20
5% PAL	0.050	13.07	8.95×10^{10}	1.19
1% PAL	0.010	8.45	5.11×10^{10}	0.68
0.5% PAL	0.005	6.92	3.89×10^{10}	0.57
0.1% PAL	0.001	4.09	1.94×10^{10}	0.26

4 Discussion

When the Great Oxidation Event occurred approximately 2.4 billion years ago, O₂ increased by several orders of magnitude in the atmosphere in a time frame of approximately 200 Myr (Gumsley et al., 2017). For the first time, these higher levels of O₂ resulted in the generation of a significant O₃ layer which may have created a stratospheric thermal inversion (Kasting et al., 1985; Segura et al., 2003; Cooke et al., 2022). **The O₃ layer provided an additional shield for Earth’s surface from ultraviolet radiation. However, the exact thickness of the O₃ layer over geological time is difficult to determine. Many uncertainties** exist upon the atmospheric pressure and composition throughout the Proterozoic (Lyons et al., 2014; Steadman et al., 2020; Catling and Zahnle, 2020), as well as **there being** differences in model predictions under similar assumptions (Way et al., 2017; Cooke et al., 2022; Yassin Jaziri et al., 2022; Ji et al., 2023, 2024). With an oxygenated biosphere, the conditions were set for the evolution of oxygen-dependent animals (Cole et al., 2020) and the eventual colonization of land (Dunn, 2013). **Additionally, the GOE** may have coincided with the end of relatively high levels of hydrogen escape (Zahnle et al., 2013, 2019). Whether other terrestrial worlds go through a similar path of geological and atmospheric evolution is unknown, although it has been modelled under different conditions and assumptions (Gebauer et al., 2018; Olson et al., 2018; Ozaki and Reinhard, 2021; Krissansen-Totton et al., 2021; Krissansen-Totton and Fortney, 2022).

Our work represents an initial step towards a quantitative estimate for hydrogen escape since the Proterozoic began 2.4 billion years ago. Overall, our results suggest that hydrogen escape during the Proterozoic could have been between 3.8 times lower and up to 1.2 times greater than the calculated pre-industrial escape rate when accounting for O₂ changes only. The present day loss of hydrogen is insignificant on geological timescales, such that the fluctuations in O₂ from the start of the Proterozoic to the present day likely did not in itself cause substantial levels of hydrogen escape. In what follows, we discuss how our simulations can be used as a platform for future work towards the goal of reconstructing Earth’s past atmospheric hydrogen escape.

4.1 The oxygen-ozone valve

When only changing the atmospheric O_2 (and N_2) mixing ratio, as O_2 increases from 0.1% PAL to 150% PAL, the total atmospheric O_3 column increases (Cooke et al., 2022). Note that increasing quantities of total O_3 does not always lead to an increase in temperatures in the TTL, because the location of the O_3 abundance peak in the atmosphere, and the magnitude of the peak, can move with O_2 mixing ratio. This effect is illustrated by the greater total hydrogen mixing ratio at the homopause in the 10% and 5% PAL cases when compared to the PI case (e.g., see Fig. 1, Fig. 6, and Fig. 7).

Our work shows that O_2 levels during the Proterozoic may have partially controlled the diffusion-limited flux for hydrogen escape. However, this will depend on several other factors, such as the continental distribution and the Earth's obliquity which can affect atmospheric dynamics and seasonal cycles, respectively. The fainter Sun, the albedo of the Earth through time, and various greenhouse gases (including CH_4) which can act to cool the stratosphere but warm the surface (Lin and Emanuel, 2024), will have also impacted TTL temperatures and hydrogen escape rates. For instance, a cooling of the stratosphere can increase the amount of O_3 , just like the current increasing atmospheric content of CO_2 is aiding a possible, albeit complex, recovery of the O_3 layer (Rosenfield et al., 2002; Fahey et al., 2018; Ivanciu et al., 2022; Maliniemi et al., 2021).

There exist estimates of many of these aforementioned variables through Earth's history, but each of them come with uncertainties and vary throughout the Proterozoic. Liu et al. (2025) suggest that the O_3 layer could have been kept at low levels due to protracted atmospheric iodine concentrations throughout the Proterozoic. Atmospheric iodine would destroy O_3 through catalytic cycles (WACCM6 does not currently include iodine in its chemical mechanism). The computed WACCM6 O_3 columns are lower than the baseline estimates from Liu et al. (2025), but when Liu et al. (2025) included substantial atmospheric iodine (e.g. $20\times$ modern iodine concentrations), then their O_3 columns are lower than the WACCM6 results. If we incorporated iodine, then this could lead to a reduced rate of predicted hydrogen escape at a given O_2 concentration due to diminished O_3 heating.

4.2 Future work for Earth

Our atmospheric predictions are relevant for both Earth's past and future climate states. However, **this statement comes with several caveats**. The moist-adiabatic lapse rate (the rate at which the temperature of a parcel of saturated air decreases with altitude as it rises in the atmosphere under adiabatic conditions), strongly depends on temperature. Thus, the total hydrogen mixing ratio in the troposphere of the Earth is determined by the incoming radiation, albedo, and greenhouse effect. For modern Earth, these parameters are relatively well known. We did not attempt to fully account for all climatic changes since the Proterozoic began; instead, as a first step, we focused on the effect on the escape rate by changing O_2 only. **The range of O_2 concentrations we accounted for are relevant to the past 2.4 billion years and are not relevant to the Archean eon (4–2.4 Gyr ago) when molecular oxygen was much lower than present day and Proterozoic (2.4–0.541 Gyr ago) concentrations (Catling and Zahnle, 2020). The different climate states that might have existed throughout the Proterozoic and Phanerozoic⁵ (0.541**

⁵By climate states, here we refer to the variety of factors that would have affected weather, atmospheric circulation systems, and surface temperature. These include glaciations and hotter periods, as well as the various continental land masses, fainter Sun, and greenhouse gas concentrations.

Gyr ago–present day) should be simulated in future work. A hotter troposphere may have existed due to increased amounts of
365 CH₄ or CO₂, or both. In this case, the warmer TTL would have increased hydrogen escape rates by elevating the amount of
H₂O reaching the stratosphere. On the other hand, colder temperatures during Earth’s glacial periods would cool the tropopause
(Graham et al., 2019) and reduce the stratospheric H₂O concentration.

Even with a much colder TTL, CH₄ bypasses the control of the oxygen valve or a colder troposphere because CH₄ does not
condense in Earth’s atmosphere like H₂O. Thus, the Proterozoic and Phanerozoic hydrogen escape rate could have instead been
370 controlled by the CH₄ abundance rather than the O₂ abundance, assuming that its mixing ratio reaching the TTL was greater
than half the mixing ratio of H₂O in the TTL - see Eq. 2. The current problem with evaluating this effect is that estimates
for Proterozoic CH₄ concentrations vary by several orders of magnitude (Pavlov et al., 2003; Daines and Lenton, 2016; Zhao
et al., 2018; Olson et al., 2018; Laakso and Schrag, 2019; Catling and Zahnle, 2020; Cadeau et al., 2020) and may be greater
or lower than the present day mixing ratio. The thinner simulated O₃ columns found from 3D modeling (Cooke et al., 2022;
375 Yassin Jaziri et al., 2022) and updated 1D calculations (Ji et al., 2024), as well as the inclusion of iodine chemistry (Liu et al.,
2025), suggest that more UV radiation would have penetrated deeper into the atmosphere for much of the Proterozoic and
potentially resulted in lower CH₄ lifetimes at O₂ concentrations of $\leq 1\%$ PAL. In this situation, greater surface-to-atmosphere
fluxes of CH₄ would have been required to sustain current ~ 1 ppmv mixing ratios (such fluxes may not be plausible; Daines
and Lenton, 2016; Olson et al., 2016; Laakso and Schrag, 2019), or the higher estimates that some studies suggest in the region
380 of ~ 10 – 100 ppmv (Pavlov et al., 2003; Fiorella and Sheldon, 2017; Zhao et al., 2018; Fakhraee et al., 2019). Closer in time to
the present day, CH₄ mixing ratios in the Phanerozoic (0.51 Gyr ago - present day) have been simulated to be between 0.1–12
ppmv (Beerling et al., 2009).

Clearly, there is much to still learn about why the different models predict specific atmospheric abundances for molecules
such as O₃, which has significant horizontal and vertical variation (Cooke et al., 2022; Yassin Jaziri et al., 2022), and CH₄
385 (Kasting, 2025). The entire conceivable parameter space has certainly not been fully explored; such endeavors utilising 3D
models are welcomed but will likely prove computationally expensive. Attempts to define the CH₄ concentrations in the
various geological eons are an important part of the history of hydrogen escape, but for now, the mixing ratio of CH₄ through
time remains poorly constrained. Ascertaining its production rate in the Proterozoic will help to establish its concentrations in
that eon (Kasting and Ji, 2025).

390 Eventually, as the Sun’s luminosity increases, the Earth will heat up, CO₂ will be sequestered out of the atmosphere through
carbonate-silicate weathering, and it is predicted that the Earth will lose all of its water through the moist greenhouse effect
approximately 2 Gyr in the future (Kasting et al., 1984; Kasting, 1988; Wolf and Toon, 2014; Ozaki and Reinhard, 2021),
although the estimate of the time varies. But before that event occurs, previous studies have predicted that the lack of CO₂ will
have a detrimental effect on the biosphere because plants are unable to survive at very low CO₂ concentrations (Lovelock and
395 Whitfield, 1982; Caldeira and Kasting, 1992).

Several calculations of the lifespan of the Earth’s biosphere have been attempted (Lovelock and Whitfield, 1982; Caldeira
and Kasting, 1992; Franck et al., 2000; Von Bloh et al., 2003; Rushby et al., 2018; Ozaki and Reinhard, 2021; Mello and
Friaça, 2023; Graham et al., 2024). For instance, Ozaki and Reinhard (2021) calculated that in 1.08 ± 0.14 billion years (1 σ),

Earth's atmospheric O₂ will drop to 1% PAL. More recently, Graham et al. (2024) suggested that Earth's biosphere may last
400 for up to 1.6–1.86 Gyr. Most of these estimates for biospheric termination are before the Earth is predicted to lose its water
and become, by the most basic definition, uninhabitable. As O₂ levels drop following the extinction of photosynthesizing
life (Ozaki and Reinhard, 2021), O₃ will decrease, and the upper troposphere and TTL may be colder than under otherwise
equivalent conditions, diminishing the amount of hydrogen available in the upper atmosphere which can escape to space and
cause irreversible water loss (the future rates of hydrogen escape would likely have a negligible effect on oxidizing the Earth
405 to reverse the loss of O₂ from ceasing photosynthesis). Hence, our results suggest that the ocean loss timescale could be
extended. Thus, investigations aiming to assess the lifetime of Earth's future habitability should incorporate a 3D chemistry-
climate model that includes comprehensive atmospheric oxygen chemistry and that can adequately represent the 3D movement
of air parcels through the TTL. Additionally, it is possible that our WACCM6 simulations could be used as input to a model
which has a higher top (lower minimum pressure), such as the WACCM-X model (Liu et al., 2018), or even used as an input
410 into a computational model that explicitly calculates atmospheric escape (e.g., Dong et al., 2018; Schulik and Booth, 2023).

4.3 Future work for exoplanets

Other planets outside of our solar system (exoplanets) may be habitable and evolve life. Whether they can keep liquid water on
their surface is crucial for the persistence of habitability as it is currently defined.

Simulations with other 3D models such as the LMD-g (Yassin Jaziri et al., 2022) or the Unified Model (Braam et al.,
415 2022), or those at higher resolutions (e.g., LFRic-Atmosphere; Sergeev et al., 2023), may yield different quantitative results
due to varying treatments of convection and water vapour microphysics, as well as alternative radiative transfer schemes. The
stratospheric thermal inversion in Earth's atmosphere is due to O₃ which provides sufficient atmospheric heating when O₂
is present in quantities $\gtrsim 0.1\%$ PAL (Way et al., 2017; Cooke et al., 2022). But a terrestrial exoplanet may not require O₂
to modulate the thermal structure: other atmospheric compositions can result in thermal inversions and may similarly act to
420 enhance or reduce water loss. For example, Titan has a thermal inversion due to the presence of hazes which absorb shortwave
radiation (Danielson et al., 1973; Catling and Kasting, 2017), and CN on hot super-Earths can also cause a thermal inversion
(Zilinskas et al., 2021). Additionally, the heating rates will change depending on the distribution of stellar radiation over
wavelengths: O₃ radiative heating significantly affects planets around F and G stars but less so around M and K stars (Godolt
et al., 2015; Kozakis et al., 2022; De Luca et al., 2024). O₂ can also build up on exoplanets without the presence of life under
425 specific conditions due to large amounts of CO₂ (Segura et al., 2007) or H₂O photolysis (Wordsworth and Pierrehumbert,
2014).

There are various planetary system parameters which affect the climate state of exoplanets. Crucial to the atmospheric
dynamics is the planetary rotation rate and whether it is tidally locked (Forget and Leconte, 2014; Turbet et al., 2016; Boutle
et al., 2017; Del Genio et al., 2019b; Braam et al., 2025). The circulation affects the atmospheric temperature structure and
430 how tracers such as H₂O are advected. Other salient factors include the strength and spectral shape of the incoming radiation
(Eager-Nash et al., 2020), the continental distribution or lack thereof (Lewis et al., 2018; Zhao et al., 2021; Macdonald et al.,
2022), obliquity (Kang, 2019), eccentricity (Liu et al., 2023), flares (Chen et al., 2021; Ridgway et al., 2023), simulation

resolution (Sergeev et al., 2020; Lefèvre et al., 2021; Song et al., 2022; Kodama et al., 2022; Yang et al., 2023; Sergeev et al., 2024; Garcia et al., 2024), as well as the planetary Bond albedo and surface composition (Cowan and Agol, 2011; Del Genio et al., 2019a; Madden and Kaltenecker, 2020). Ocean salinity can also be important in modulating the climate of rocky planets (Olson et al., 2022).

By connecting exoplanet simulations to a model that includes the ionosphere (e.g., WACCM-X), one could attempt to answer interesting questions regarding the upper atmospheres of terrestrial exoplanets. For instance, how do the ionospheres of various exoplanets compare to Earth? Just like flares disturb the atmospheres of terrestrial planets (Segura et al., 2010; Tilley et al., 2019; Chen et al., 2021; Ridgway et al., 2023), do flares (especially for active M dwarf stars) impact their ionospheres differently from those that strike the Earth (Leonovich and Tashchilin, 2009; Schillings et al., 2018; Hayes et al., 2021)? And could those differences manifest in a way which is detectable with future instruments (Mendillo et al., 2018; Mendillo, 2019)?

The maximum difference we found in the diffusion-limited hydrogen escape between alternative oxygenation states when all other initial conditions and boundary conditions are kept constant is a factor of 4.7. Whilst this change is not significant for Earth under the assumptions made in this work, it potentially could be significant when the ocean loss timescale for a habitable terrestrial exoplanet is similar to a time that a planet could theoretically spend in the habitable zone (i.e. a planet with a 10 Gyr ocean loss timescale with a different oxygenation state could instead have an ocean loss timescale of ≈ 2 Gyr).

5 Conclusions

Since the dawn of the Proterozoic (2.4 Gyr ago), several properties of the Earth have changed. The continental configuration has altered several times, the Sun's luminosity has increased, and the concentration of atmospheric gases such as O₂, CO₂, and CH₄ have fluctuated.

This work used WACCM6 simulations of the Earth to explore the effect that the changing O₂ mixing ratio had on the total hydrogen mixing ratio at the homopause in order to predict the possible diffusion-limited hydrogen escape rate through geological time. A 3D model with chemistry and a global representation of water vapour microphysics is critical to simulating the transport of H₂O into the stratosphere and thus estimating diffusion-limited hydrogen escape when **other atmospheric constituents (e.g., CH₄ and H₂, which we held constant at the surface)** are not the dominant hydrogen bearing species. **In our simulated scenarios we showed that 1D models may predict a stratospheric water vapour discrepancy of between a factor of 3.8–6.0 depending on the O₂ mixing ratio when compared to 3D calculations.**

We found that the atmospheric O₂ mixing ratio acts as a nonlinear valve on the total hydrogen mixing ratio in the stratosphere. The O₃ mixing ratio (which is affected by the abundance of O₂) around the tropical tropopause layer (TTL) determines the amount of UV heating that warms the TTL and controls the H₂O mixing ratio entering the stratosphere. The obliquity of Earth induces seasonal cycles, which affect phenomena such as the Brewer-Dobson circulation and the tropical tape recorder. We showed that the tropical tape recorder disappeared at 0.1% PAL, which suggests that the present-day seasonal transport of H₂O molecules to the stratosphere may not have taken place at low O₂ mixing ratios.

465 Between 0.1% PAL and 150% PAL of O₂, the upward diffusion of H₂O is modulated in a non-linear manner, with 5% and 10% PAL having the maximum predicted diffusion-limited escape rate of hydrogen due to the peak in O₃ concentrations causing localized heating. The predicted escape rate changes by up to a factor of 4.7 **between all of the simulations**, meaning that whilst the oxygen valve affects diffusion-limited hydrogen escape, in the absence of other factors, it would not have resulted in relatively high levels of hydrogen escape and thus water loss since the GOE.

470 In order to dissect Earth's atmospheric history, future work should investigate multiple other physical, chemical, and biological factors that may affect the hydrogen escape rate whilst also accounting for the oxygen valve we presented.

Code availability. WACCM6 is a publicly available code. The specific release used in this paper was CESM2.1.3, which can be downloaded from the following CESM downloads page. Our code modifications are detailed here in the ExoCESM GitHub.

Data availability. The time-averaged data is currently available in the Dryad data repository. The data with time-variability will be made
475 available upon acceptance of the manuscript.

Author contributions. GJC set up and ran the simulations, performed the analysis, wrote the analysis and plotting code, and wrote the manuscript. DRM and CW supervised the project and provided comments on the manuscript. FSM and MB provided comments on the manuscript.

Competing interests. The authors declare no competing interests

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