

Measurement report: Formation and brownness of aqueous secondary organic aerosol from the aged biomass-burning emissions in the Sichuan Basin, China

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20 **Abstract.** Secondary organic aerosol (SOA), formed via complex chemical
21 mechanisms, is the major contributor to atmospheric aerosol pollution and climate
22 forcing worldwide. Aqueous-phase oxidation serve as an important pathway for SOA
23 formation, with aqueous SOA (aqSOA) exhibiting light absorption across the
24 ultraviolet to visible range. Here, we reported the formation and absorption properties
25 of aqSOA in the Sichuan Basin, China. The results show that aqSOA mainly
26 originated from aged biomass-burning emissions via aqueous-phase reactions rather
27 than from photochemical reactions, particularly under high aerosol liquid water
28 content (ALWC) conditions, during the polluted period. Substantial enhancement of
29 brown carbon (BrC) absorption by SOA was observed between 370 nm and 660 nm
30 (27.5%–43.2%). These findings highlighted the significant contribution of aqSOA
31 formation from aged biomass-burning emissions to the BrC budget and absorption,
32 especially at night. The mean aerosol absorption Ångström exponents between 370
33 nm and 880 nm ($AAE_{370-880}$) was 1.95, higher than values reported for fresh and
34 photochemically aged biomass-burning emissions. This study elucidates the formation
35 and light-absorbing characteristics of aqSOA derived from aged biomass-burning
36 emissions and highlights the important role of aqueous-phase reactions in aerosol
37 pollution and radiative absorption.
38 **Keywords:** Particulate matter; Secondary organic aerosol; Aqueous-phase oxidation;
39 Aged biomass-burning emissions; Brown carbon.

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40 1 Introduction

41 Organic aerosol (OA) constitutes a dominant fraction (20 to 90%) of atmospheric
42 aerosol, with significant implications for air quality and climate forcing (Jimenez et
43 al., 2009). Field observations consistently indicate that secondary OA (SOA), formed
44 by atmospheric oxidation of volatile organic compounds (VOCs) and primary OA
45 (POA), accounts for most OA worldwide (Ervens et al., 2011; Huang et al., 2014;
46 Kourtchev et al., 2016). Recent results demonstrate that aqueous-phase oxidation
47 serves as an important pathway for SOA formation and these SOA products (aqSOA)
48 exhibit light absorption across the ultraviolet (UV) to visible (Vis) range (Gilardoni et
49 al., 2016; Lim et al., 2010; McNeill 2015; Powelson et al., 2014; Sun et al., 2010).
50 However, the formation mechanisms and absorption properties of aqSOA remain
51 poorly understood, which hinders efforts to improve air quality and reduce
52 uncertainties in global climate estimates.

53 An increasing number of studies suggest that aqSOA represents a major
54 component of SOA formed in fogs, clouds, and aerosol water (Ervens et al., 2011;
55 Ortiz-Montalvo et al., 2012; Tan et al., 2012; Xu et al., 2022). Oxygenated VOCs
56 (OVOCs) with high water solubility and low Henry's law constants (i.e.,
57 methylglyoxal and glycolaldehyde) are important aqSOA precursors (Ortiz-Montalvo
58 et al., 2012; Tan et al., 2012). A few laboratory studies have also shown that
59 levoglucosan and phenolic species produced from biomass burning could also act as
60 aqSOA precursors (Yu et al., 2016; Zhao et al., 2014). Gilardoni et al. (2016) reported
61 direct ambient observations of aqSOA formation from biomass-burning emissions in
62 fog water and wet aerosol. Additionally, recent studies indicate that aqSOA with high
63 molecular weight (i.e., 4-ethylphenol) formed by aqueous-phase photochemical
64 oxidation can exhibit strong light absorptivity within the UV range (Herrmann et al.,
65 2015; Ye et al., 2019). Previous laboratory studies also demonstrate that certain
66 aqSOA components, such as π -conjugated compounds and imidazole with C=N bonds
67 produced by aldol condensation and aqueous-phase carbonyl compound reactions
68 respectively, would strongly absorb light at the near-UV region (Drozd and McNeill,

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69 2014; Kampf et al., 2012; Nozière and Esteve, 2007; Powelson et al., 2014). Despite
70 numerous studies on the formation and optical properties of aqSOA, limited ambient
71 observations still hinder a better understanding of its role in atmospheric chemistry
72 and climate.

73 China has experienced severe PM_{2.5} pollution under stagnant high-humidity
74 conditions, during which SOA, a major PM_{2.5} component, originated largely from
75 fossil fuel combustion and biomass burning (Huang et al., 2014; Wang et al., 2016;
76 Wang et al., 2021; Xu et al., 2022). Field observations indicate that highly oxidized
77 SOA could form through aqueous-phase processing driven by acid-catalyzed
78 oxidation (Meng et al., 2020; Xu et al., 2017), and aqSOA derives from
79 biomass-burning OA (BBOA) and fossil-fuel OA via aqueous-phase reactions (Wang
80 et al., 2021; Zhao et al., 2019). A few laboratory studies have found that
81 aqueous-phase reactions serve as an important oxidation pathway for nitrophenol
82 products (i.e., 5-nitrovanillin and 4-nitroguaiacol) with exhibit strong UV absorption
83 and higher formation/transformation rates in more acidic solutions (Kroflc et al.,
84 2015; Li et al., 2023; Pang et al., 2019; Yang et al., 2021). However, observations of
85 aqSOA formation and optical properties in China remain limited, with most research
86 focusing on the North China Plain (NCP). Similar to the NCP, the Sichuan Basin
87 (SCB) characterized by high humidity and frequent biomass burning is also the main
88 region with severe aerosol pollution in China (Tian et al., 2019; Wang et al., 2018;
89 Yang et al., 2011).

90 Previous research has indicated that aqSOA from different regions can exhibit
91 distinct formation mechanisms and optical properties due to varying emission sources
92 and ambient conditions (Bao et al., 2023; Bao et al., 2024; Wang et al., 2021; Xu et al.,
93 2017). For instance, Wang et al. (2021) revealed fast aqueous-phase conversion of
94 fossil-fuel primary organic aerosol (FF-POA) to aqSOA under high-humidity
95 conditions during a winter haze event in Beijing, noting that the resulting aqSOA
96 exhibited much lower light absorption than its primary precursor due to decreased
97 aromaticity. Similarly, Huang et al. (2023) illustrated that the aqueous-phase oxidation
98 of fossil fuel combustion emissions played a critical role in SOA formation under high

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99 RH conditions during winter. Xu et al. (2022) indicated that biomass burning serves
100 as a significant non-fossil source of aqSOA under high RH and high aerosol liquid
101 water content (ALWC) conditions, especially during fall-to-winter period when open
102 burning of post-harvest agricultural crop residues is widespread in China. While
103 aqSOA formation has been extensively studied during winter, autumn—despite
104 featuring both high ALWC and strong biomass burning emissions—has received
105 considerably less attention (Feng et al., 2022; Qiu et al., 2016; Zhao et al., 2019). In
106 contrast to these studies in NCP, research in the SCB has shown that the effect of
107 aqueous-phase reactions on oxygenated OA (OOA) formation was significant when
108 ALWC was below $200 \mu\text{g m}^{-3}$, but became insignificant at $\text{ALWC} > 200 \mu\text{g m}^{-3}$.
109 Additionally, aqueous-phase oxidation, probably did not contribute to the decay of
110 BrC during summer in the SCB, (Bao et al., 2024). It should be pointed out that
111 autumn is the typical biomass burning season following the harvest of rape and rice in
112 the SCB, and the contribution of biomass burning to OA is significantly higher in
113 autumn than in other seasons (Chen et al., 2017; Tao et al., 2014; Yang et al., 2019). In
114 summary, the intensive biomass burning emissions and high ALWC during autumn in
115 the SCB likely result in aqSOA formation pathways that differ from those in other
116 seasons. To date, few studies have explored the dynamic evolution and optical
117 properties of aqSOA in the SCB, especially during autumn, leaving ambient aqSOA
118 processing poorly understood. Therefore, a more detailed characterization of aqSOA
119 formation and optical properties is of great importance to reveal the key factors
120 contributing to haze formation in this region.

121 Here, a time-of-flight aerosol chemical speciation monitor (ToF-ACSM) and a
122 series of collocated instruments were deployed to characterize the dynamic evolution
123 of aqSOA from biomass burning under real ambient conditions. The study was
124 conducted in a typical city with relatively serious air pollution in SCB from October
125 21 to November 23, 2022. Observations revealed, that haze formation was largely
126 driven by BBOA and aqSOA. We demonstrated that aqSOA originated from aged
127 BBOA via aqueous-phase reactions. Furthermore, results showed that aqSOA derived
128 from aged BBOA exhibited strong UV absorption and contributed to positive

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129 radiative forcing. These results elucidate the formation and brownness of aqSOA from
130 aged biomass-burning emissions, providing insights for better simulating its impacts
131 on atmospheric chemistry and climate.

132 2 Methods

133 2.1 Sampling site

134 An intensive field campaign investigating the chemical and physical properties
135 of aerosols was conducted at a site in Yongchuan (29°21'25" N, 105°54'6" E), a city
136 experiencing severe aerosol pollution from October 21 to November 23, 2022. This
137 site is representative of an urban environment, surrounded by restaurants, shopping
138 malls, and residential buildings, and is located in a parallel ridge-and-valley area
139 between the two megacities in SCB (Chongqing center and Chengdu) (Fig. S1). It is
140 primarily influenced by multiple local emission sources, including traffic (arterial
141 roads to the east 600 m and west 300 m) and a variety of residential activities (i.e.,
142 biomass burning and fossil fuel combustion). The absence of dynamic interference,
143 from neighboring buildings, allowed the measurements to clearly capture the
144 characteristics and evolution of haze pollution at this location.

145 2.2 Instrumentation

146 During the campaign, the non-refractory aerosol (NR-PM_{2.5}) species, including
147 OA, ammonium (NH₄), nitrates (NO₃), sulfates (SO₄), and chlorides (Chl), were
148 measured on-line by ToF-ACSM (Aerodyne Research Inc.). Ambient aerosols were
149 pumped into ToF-ACSM at a flow rate of 3 L min⁻¹ through a PM_{2.5} cyclone
150 (URG-2000-30ED) and a Nafion dryer (MD-110-48S, Perma Pure, Inc.) reducing the
151 relative humidity to below 30%. The measurement principle was described in detail in
152 the previous studies (Fröhlich et al., 2013; Ng et al., 2011c).

153 A seven-wavelength Aethalometer (AE33, Magee Scientific) was used to
154 measure the aerosol light absorption (Abs_λ) and equivalent black carbon (BC_λ) mass
155 concentrations in real time at 370, 470, 520, 590, 660, 880, and 950 nm. The sampled

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156 particles were dried by a Nafion dryer (MD-70024S-3, Perma Pure, Inc.) before
157 entering into AE33. The light attenuation coefficients were converted to Abs_{λ} based
158 on the real-time compensation parameter, and the nonlinear loading effects of quartz
159 filters were dealt with on-line by the parallel measurements of attenuation values
160 (ATN1 and ATN2) (Coen et al., 2010; Drinovec et al., 2015). The scattering effects of
161 quartz filters were modified automatically by a fixed multiple scattering parameter
162 (2.14). Detailed measurement methods and principles of AE33 can be found in
163 Drinovec et al. (2015).

164 During the campaign, the gaseous species (including O_3 , NO_2 , and CO) were
165 continuously measured by gas analyzers (49i, 42i, and 48i, Thermo Scientific), that
166 were maintained and calibrated weekly. Hourly meteorological parameters data
167 including temperature (T), relative humidity (RH) and $PM_{2.5}$ mass concentrations
168 were obtained on-line from the measurements at the National Environmental
169 Monitoring Station, which was close to our sampling site (<http://www.cnemc.cn/>).

170 **2.3 Data analysis**

171 **2.3.1 ToF-ACSM data analysis**

172 The raw mass spectra data measured by ToF-ACSM were analyzed using
173 Tofware v2.5.13 (Tofwerk AG) in Igor Pro 6.37 (WaveMetrics, Inc.). The ionization
174 efficiency (IE) and relative ionization efficiency (RIEs) were regularly calibrated
175 using a scanning mobility particle sizer with a differential mobility analyzer (SMPS
176 3081A, TSI) and a condensation particle counter (CPC 3775, TSI). The
177 comprehensive overview of the operation and calibration procedures of ToF-ACSM
178 can be found in Bao et al. (2023). In accordance with previous studies, the default
179 RIEs values for OA, NO_3 , and Chl were set to 1.4, 1.1, and 1.3, respectively
180 (Canagaratna et al., 2007; Elser et al., 2016). The IE value (236 ions pg^{-1}) and RIEs of
181 SO_4 (1.2) and NH_4 (4.3) were estimated from the calibrations of pure ammonium
182 nitrate and ammonium sulfate, respectively. Meanwhile, a particle collection
183 efficiency (CE) was introduced to compensate for the particle loss, as the acidity, the

184 contribution of ammonium nitrate (ANMF) and phase state changed the particle
185 bounce effects at the vaporizer (Matthew et al., 2008). Middlebrook et al. (2012)
186 developed a CE algorithm for ToF-ACSM to quantify the aerosol species. Their
187 results indicated that a constant CE value of 0.45 should be used when: (1) the ANMF
188 is below 40%, or (2) particles are partially or fully neutralized. In this study, aerosol
189 particles were dried by Nafion dryer (RH < 30%) before sampling by ToF-ACSM,
190 and the ANMF was always below 40%. As shown in Fig. S2, the average ratio of the
191 measured NH₄ to the predicted NH₄ needed to fully neutralize the SO₄, NO₃ and Chl
192 was approximately 1. All of these conditions did not affect the CE value that had
193 usually been used at this site. The typical default CE value (0.5) was applied during
194 the whole sampling period, which was consistent with previous research (Bao et al.,
195 2025; Peng et al., 2025; Sun et al., 2016a; Sun et al., 2016b; Zhao et al., 2019). While
196 the typical default CE is 10% higher than 0.45, the difference is small considering the
197 30% uncertainty determined for CE (Bahreini et al., 2009). Additionally, the strong
198 correlation between NR-PM_{2.5} and PM_{2.5} mass concentrations supported that the CE
199 value was reasonable (Fig. S3).

200 The mass spectral matrix of OA for m/z 10–120 was analyzed by positive matrix
201 factorization (PMF) and multilinear engine (ME2) implemented via, SoFi (Source
202 Finder) (version 6.3, Canonaco et al., 2013; Paatero 1999; Paatero and Tapper 1994).
203 Briefly, an unconstrained PMF analysis was performed to determine the number and
204 types of source factors. Then, the constrained ME2 approach was applied to minimize
205 rotational ambiguity by testing α -values from 0 to 1 with a step of 0.1 (Elser et al.,
206 2016; Wang et al., 2019b; Zhong et al., 2021). Ions with a signal-to-noise ratio (S/N)
207 lower than 0.2 were discarded, while those with S/N between 0.2–2 were
208 downweighted by a factor of 2 (Bao et al., 2023; Paatero and Hopke 2003). Finally,
209 five OA factors were resolved with the rotational parameter set to zero ($f_{\text{peak}} = 0$),
210 including three POA factors (i.e., BBOA, coal-combustion OA (CCOA), and
211 hydrocarbon-like OA (HOA)) and two SOA factors (i.e., OOA and aqSOA) (Fig. S9
212 and S10). We present detailed diagnostic plots of the PMF results in the supporting
213 information (Fig. S4–S10), and the details of OA source apportionment procedures

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214 are described in SI Text S1.

215 2.3.2 Aerosol liquid water content

216 ALWC is controlled by meteorological conditions (T and RH) ~~as well as by~~
217 inorganic and organic components. During the campaign, ~~ALWC controlled by~~
218 ~~inorganic fraction~~ was estimated by the ISORROPIA-II ~~thermodynamic~~ model, based
219 on ~~mass concentrations of~~ ammonium, nitrates, sulfates, and chlorides ~~measured by~~
220 ToF-ACSM, ~~together with~~ the meteorological parameters (T and RH) from National
221 Environmental Monitoring Station (Fountoukis and Nenes, 2007). ~~The calculations~~
222 ~~were performed in~~ forward type and metastable mode ~~within~~ ISORROPIA-II,
223 (Hennigan et al., 2015). The ~~model simulated the~~ thermodynamic equilibrium of the
224 $\text{NH}_4^+ - \text{SO}_4^{2-} - \text{NO}_3^- - \text{Cl}^- - \text{H}_2\text{O}$ system and ~~subsequently calculated~~ ALWC. ~~Following~~
225 ~~established methods~~, the organic contribution ~~to~~ ALWC was ~~estimated using the~~
226 Zdanovskii–Stokes–Robinson (ZSR) mixing rule as discussed in SI Text S2 (Guo et
227 al., 2015; Huang et al., 2020; Nguyen et al., 2016; Xu et al., 2022). In this study, the
228 ALWC with organic species ranged from 0.1 to 35.2 $\mu\text{g m}^{-3}$, with an average of $1.9 \pm$
229 $3.0 \mu\text{g m}^{-3}$, taking up $3.7 \pm 2.2\%$ of total ALWC. As organic species had minor effects
230 on total ALWC ($< 5\%$), the ALWC was determined only considering inorganic species
231 (Chen et al., 2021; Guo et al., 2015; Liu et al., 2017).

232 2.3.3 Light absorption measurements

233 The Abs_λ was ~~partitioned into contributions from~~ BC and brown carbon (BrC, a
234 group of colored OA compounds) (~~$\text{Abs}_{\lambda, \text{BC}}$ and $\text{Abs}_{\lambda, \text{BrC}}$~~) (~~$\text{Abs}_\lambda = \text{Abs}_{\lambda, \text{BC}} + \text{Abs}_{\lambda, \text{BrC}}$~~) and
235 characterized by the absorption Ångström exponents (AAE) (Laskin et al., 2015).
236 Here, Abs_λ was ~~derived from~~ BC_λ mass concentrations ~~at each wavelength using the~~
237 ~~mass absorption cross-section (MAC_λ)~~ (~~$\text{Abs}_\lambda = \text{BC}_\lambda \times \text{MAC}_\lambda$~~), ~~with~~ MAC_λ values
238 ~~assumed as~~ 18.47, 14.54, 13.14, 11.58, 10.35, 7.77, and 7.19 $\text{m}^2 \text{g}^{-1}$ at 370, 470, 520,
239 590, 660, 880, and 950 nm, respectively (Drinovec et al., 2015; Zhu et al., 2017).
240 ~~Absorbance at 880 nm~~ (Abs_{880}) was ~~attributed solely to~~ BC, then the following

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241 formula was used to determine $Abs_{\lambda,BC}$ values: $Abs_{\lambda,BC}=Abs_{880}\times(880/\lambda)^{-AAE_{BC}}$
242 (Drinovec et al., 2015; Kirchstetter and Novakov, 2004; Moosmüller et al., 2009; Qin
243 et al., 2018; Zhu et al., 2017). The AAE of BC (AAE_{BC}) was obtained from the
244 equality: $AAE_{BC}=-\log_{10}(Abs_{880}/Abs_{950})\div\log_{10}(880/950)$ (Wang et al., 2021). A
245 detailed description of $Abs_{\lambda,BC}$ and $Abs_{\lambda,BrC}$ calculations is provided in SI Text S3.
246 Previous studies have demonstrated that AAE_{BC} is sensitive to the refractive index,
247 size distribution, and coating state of carbonaceous aerosols (Gyawali et al., 2009;
248 Lack and Langridge, 2013; Li et al., 2019). In this study, the uncertainties in Abs_{BC}
249 and Abs_{BrC} estimations were analyzed (Text S3). The relative uncertainty ranges of
250 Abs_{BC} and Abs_{BrC} were [-46%, +21%] and [-112%, +42%] at 370 nm, respectively.
251 $Abs_{\lambda,BrC}$ comprises contributions from primary and secondary BrC ($Abs_{\lambda,BrC,pri}$ and
252 $Abs_{\lambda,BrC,sec}$). $Abs_{\lambda,BrC,sec}$ was calculated by a minimum R-squared (MRS) method at
253 each wavelength (Wang et al., 2019b; Wu and Yu, 2016; Wu et al., 2024). Further
254 details on the MRS method and $Abs_{\lambda,BrC,sec}$ estimation are provided in SI Text S3.

255 The light absorption contribution of different OA components at each
256 wavelength as analyzed using multiple linear regression (MLR), expressed as:
257 $Abs_{BrC}=a\times[OOA]+b\times[BBOA]+c\times[CCOA]+d\times[aqSOA]+e\times[HOA]$ (Qin et al., 2018;
258 Xie et al., 2019). Here, [OOA], [BBOA], [CCOA], [aqSOA], and [HOA] denote the
259 mass concentrations of the corresponding OA components. The coefficients a–e are
260 constants, used to optimize the Abs_{λ} of each OA component, and equivalent to MAC
261 values at each wavelength (i.e., a–e at 370 nm represented $MAC_{370,OOA}$, $MAC_{370,BBOA}$,
262 $MAC_{370,CCOA}$, $MAC_{370,aqSOA}$, and $MAC_{370,HOA}$, respectively). Here, the normalized
263 mean bias (NMB), root mean square error (RMSE), and index of agreement (IOA)
264 were used to evaluate the performance of the MLR method (SI Text S4) (Li et al.,
265 2011). The IOA values for $Abs_{370,BrC}$ and $Abs_{470,BrC}$ (0.99 and 1.00) exceeded 0.95.
266 The slopes of the linear regression between the AE33-measured and MLR-estimated
267 absorption coefficients were 0.81 and 0.96 for $Abs_{370,BrC}$ and $Abs_{470,BrC}$ (close to unity),
268 respectively. These results indicate good agreement between AE33 measurement and
269 the MLR-reconstructed $Abs_{370,BrC}$.

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270 **3 Results and discussion**

271 **3.1 Enhanced OA formation from BBOA and aqSOA during pollution periods**

272 The temporal variations of PM_{2.5} species concentrations, meteorological
273 parameters, Abs_{370,BrC} and MAC_{370,BrC} during the campaign are shown in Fig. 1. Wind
274 speeds were consistently low ($0.3 \pm 0.2 \text{ m s}^{-1}$), indicating stagnant atmospheric
275 conditions throughout the study period. The total PM_{2.5} (BC+NR-PM_{2.5}) mass
276 concentrations ranged from 7.0 to 175.5 $\mu\text{g m}^{-3}$, with an average of $48.4 \pm 27.8 \mu\text{g}$
277 m^{-3} . The average concentrations of OA, NO₃, SO₄, NH₄, Chl, and BC were $24.1 \pm$
278 $18.1, 8.3 \pm 6.2, 6.2 \pm 3.4, 5.2 \pm 2.7, 0.2 \pm 0.1,$ and $4.7 \pm 2.9 \mu\text{g m}^{-3}$, accounting for
279 $46.6 \pm 10.7\%, 17.7 \pm 8.0\%, 13.2 \pm 4.4\%, 11.2 \pm 2.7\%, 0.3 \pm 0.2\%,$ and $10.1 \pm 5.5\%$ of
280 total PM_{2.5}, respectively. OA constituted the largest fraction of PM_{2.5}, highlighting its
281 importance in PM_{2.5} pollution in the SCB (Bao et al., 2023; Wang et al., 2018). The
282 OA fractions of PM_{2.5} in this study are as high as those reported in wintertime studies
283 in the SCB, and significantly higher than those observed in winter in other regions of
284 China (Table S1). Meanwhile, elevated levels of Abs_{370,BrC} and MAC_{370,BrC}, ranging
285 from 5.8 to 210.2 Mm^{-1} ($42.4 \pm 28.5 \text{ Mm}^{-1}$) and from 0.6 to 7.0 $\text{m}^2 \text{g}^{-1}$ ($2.1 \pm 0.9 \text{ m}^2$
286 g^{-1}) respectively, were observed during the campaign.

287 According to the Chinese National Ambient Air Quality Standard (NAAQS) (GB
288 3095-2012) (MEP, 2012), the Grade I and Grade II 24-hour limits for PM_{2.5} are $35 \mu\text{g}$
289 m^{-3} and $75 \mu\text{g m}^{-3}$, respectively. The Grade II limit, based on WHO Phase-1 interim
290 target (IT-1), is higher than the WHO Air Quality Guideline (AQG) value ($15 \mu\text{g m}^{-3}$),
291 the EU daily limit ($25 \mu\text{g m}^{-3}$), and U.S. 24-hour standard ($35 \mu\text{g m}^{-3}$). During the
292 campaign, the average of PM_{2.5} concentration was 1.4 times of the Grade I limit (35
293 $\mu\text{g m}^{-3}$). Pollution periods (PP) were defined as days with daily PM_{2.5} mass
294 concentrations exceeding the Grade II limit ($75 \mu\text{g m}^{-3}$), while days with PM_{2.5} below
295 $75 \mu\text{g m}^{-3}$ were classified as clean periods (CP). During PP, the average mass
296 concentrations of BC+NR-PM_{2.5} and OA were 102.3 ± 26.9 and $57.4 \pm 22.5 \mu\text{g m}^{-3}$,
297 2.5 and 3.1 times of those, during CP, respectively. As shown in Fig. 2, the chemical

298 composition of PM_{2.5} differed substantially between PP and CP. Compared with other
299 species, a significantly higher contribution of OA was observed during PP (56.6%)
300 than CP (46.6%) (Student's t-test, $p < 0.001$) (Fig. 2).

301 Here, five OA factors were resolved by PMF model with detailed information
302 provided in SI Text S1, and the mass spectra of these factors are shown in Fig. S9.

303 HOA was characterized by alkyl fragment ion series at $C_nH^{+}_{2n-1}$ and $C_nH^{+}_{2n+1}$ (i.e.,
304 m/z 41, 43, 55, and 57), typical of primary combustion emissions (Elser et al., 2016;
305 Lanz et al., 2007). BBOA was identified by prominent signals at m/z 60 (mainly
306 $C_2H_4O_2^+$) and m/z 73 (mainly $C_3H_5O_2^+$), which are fragments of levoglucosan and
307 mannosan from incomplete biomass burning (Alfarra et al., 2007). CCOA showed
308 high signals of unsaturated hydrocarbon ion fragments such as PAH-related ion
309 fragments (i.e., m/z 77, 91, 115), indicative of traditional coal combustion (Sun et al.,

310 2016a). OOA was distinguished by a dominant signal at m/z 44 (mainly CO_2^+) and
311 strong correlation with oxygenated ions (Ng et al., 2011b). aqSOA also correlated
312 strongly with oxygenated ions (i.e., m/z 43 (mainly $C_2H_3O^+$) and m/z 44) and
313 exhibited a notably higher signal at m/z 29 (mainly CHO^+), compared with other OA

314 factors, consistent with previous reports (Sun et al., 2016a; Xu et al., 2019; Zhao et al.,
315 2019; Zhong et al., 2021). Moreover, BBOA showed significant correlations with m/z
316 60 (mainly $C_2H_4O_2^+$) and m/z 73 (Pearson's r^2 (r^2) = 0.85, 0.80, $p < 0.001$); CCOA
317 was strongly correlated with Chl and m/z 115 (r^2 = 0.56, 0.48, $p < 0.001$); HOA was
318 correlated with NO_2 and m/z 41 (r^2 = 0.47, 0.59, $p < 0.001$); OOA and aqSOA were
319 significantly correlated with NO_3/NH_4 (r^2 = 0.77, 0.75, $p < 0.001$) and $SO_4/ALWC$ (r^2
320 = 0.67, 0.85, $p < 0.001$), respectively (Fig. S10). These results confirm the robustness
321 and physicochemical plausibility of the five OA factors identified in this study.

322 Overall, the average concentration of BBOA was $8.6 \pm 7.7 \mu g m^{-3}$ and accounted for
323 the highest proportion of OA ($34.8 \pm 11.2\%$), followed by OOA ($5.5 \pm 3.5 \mu g m^{-3}$,
324 $21.7 \pm 11.4\%$), CCOA ($4.0 \pm 3.3 \mu g m^{-3}$, $15.7 \pm 7.1\%$), HOA ($3.5 \pm 2.8 \mu g m^{-3}$, 14.6
325 $\pm 8.1\%$), and aqSOA ($3.3 \pm 2.9 \mu g m^{-3}$, $13.2 \pm 5.9\%$) (Fig. 1). These results
326 demonstrated BBOA as the dominant component of OA in autumn in the SCB. It is
327 worth noting that fractions of BBOA to OA in this study were much higher than those

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328 [reported in wintertime studies in China \(Table S1\).](#)

329 It should be noted that the contributions of BBOA and aqSOA to OA increased

330 from CP (31.7% and 12.6%) to PP (38.6% and 14.1%), while **the contributions of**

331 CCOA, HOA, and OOA **decreased.** **Significantly higher RH and ALWC were**

332 observed during PP ($58.5 \pm 12.4\%$ and $69.4 \pm 30.3 \mu\text{g m}^{-3}$) than CP ($49.8 \pm 8.9\%$ and

333 $37.1 \pm 20.8 \mu\text{g m}^{-3}$) ($p < 0.001$), **whereas temperature showed no significant**

334 **difference** ($p > 0.1$). The **average wind speed** was $0.32 \pm 0.18 \text{ m s}^{-1}$ during CP, 1.3

335 times that during PP. These results indicated that the atmosphere **during PP** was

336 **characterized by stagnation,** **higher RH,** and **elevated ALWC,** which **likely** led to the

337 **distinct** sources and chemical processing of OA **compared with CP.**

338 Compared with CP, **OA concentration exhibited an** obvious diurnal variation,

339 during PP. As shown in Fig. 2, the OA concentration **peaked** was observed at 12:00

340 local time (LT) (**$82.7 \mu\text{g m}^{-3}$**) **during** daytime **in PP,** while **in CP the peak occurred** at

341 21:00 LT. **During PP,** OA concentration **increased rapidly** at a rate of $7.8 \mu\text{g m}^{-3} \text{ hr}^{-1}$

342 from 09:00 to 12:00 LT, **accompanied by** a significant decrease **in** NO_3 , BBOA and

343 aqSOA concentrations showed similar diurnal patterns to OA, with high values **during**

344 daytime and **a** rapidly increased from 09:00 to 12:00 LT during PP. Previous **studies**

345 **have** indicated that aqSOA **mass** spectrum showed **a** higher **signal at** m/z 29 (CHO^+)

346 than other OA factors (Gilardoni et al., 2016; Meng et al., 2020; Wang et al., 2021).

347 During PP, the **tracer ion fragments for** BBOA (**m/z 60**) and aqSOA (**m/z 29**), **peaked**

348 at 12:00 LT ($1.2 \mu\text{g m}^{-3}$) and 13:00 LT ($4.3 \mu\text{g m}^{-3}$), respectively. Additionally,

349 **ALWC correlated more strongly with** aqSOA concentration ($r^2 = 0.86$, $p < 0.001$) **than**

350 **with** BBOA concentration ($r^2 = 0.58$, $p < 0.001$), **Both** ALWC and aqSOA

351 concentration **peaked** at 13:00 LT, **later** than BBOA peak (12:00 LT), supporting that

352 ALWC might play a significant role in the chemical processing **that converts BBOA**

353 **to** aqSOA **during PP.** In contrast, **odd oxygen** ($\text{O}_x = \text{O}_3 + \text{NO}_2$) showed weak

354 correlations with both OOA and aqSOA concentrations **throughout** the campaign ($p >$

355 0.1 , **data** not shown). Although the average O_x concentration was higher during PP

356 ($51.1 \pm 19.6 \text{ ppb}$) than CP ($36.9 \pm 14.0 \text{ ppb}$), no significant correlation **was** observed

357 in either period. These results suggested that photochemical reactions might play a

358 limited role in SOA formation in this study.

359 In summary, these results suggested that OA was the dominant component of
360 PM_{2.5}, especially during PP in the SCB. During PP, BBOA and aqSOA contributed
361 significantly to the daytime, increasing in OA concentration. Previous studies have
362 shown, that autumn harvest period – specifically October and November – is
363 characterized by intensified biomass burning in the SCB, primarily due to
364 post-harvest crop-residue burning (Chen et al., 2017; Tao et al., 2014). Thus, daytime
365 aqSOA formation, during PP might be related to high aerosol water content and BBOA
366 emissions (Bao et al., 2023; Chen et al., 2017; Chen et al., 2019; Tao et al., 2014).
367 Additionally, Gilardoni et al. (2016) also found that aqSOA such as guaiacol dimer
368 (C₁₄H₁₄O₄⁺) could be formed from aged biomass-burning emissions in, both fog water
369 and wet aerosol, especially under high ALWC conditions. To further explore the
370 formation of aqSOA, from biomass-burning emissions via aqueous-phase reactions,
371 the next section will discuss the dynamic evolution of aqSOA in relation to BBOA.

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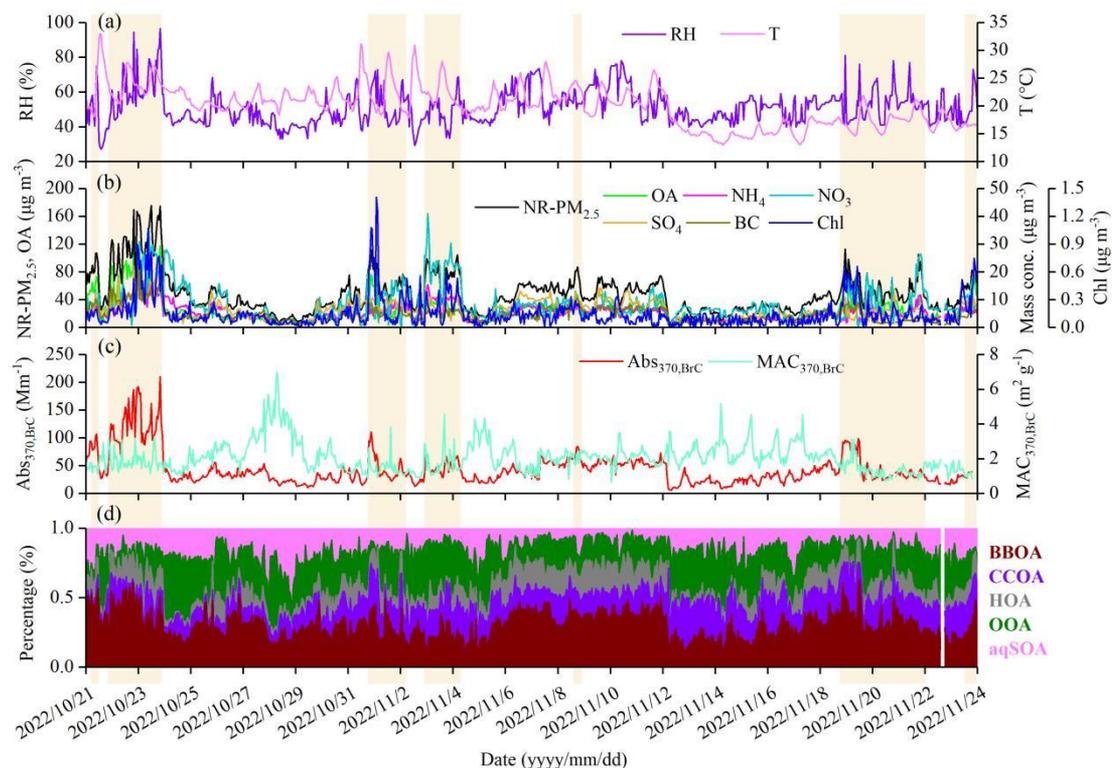
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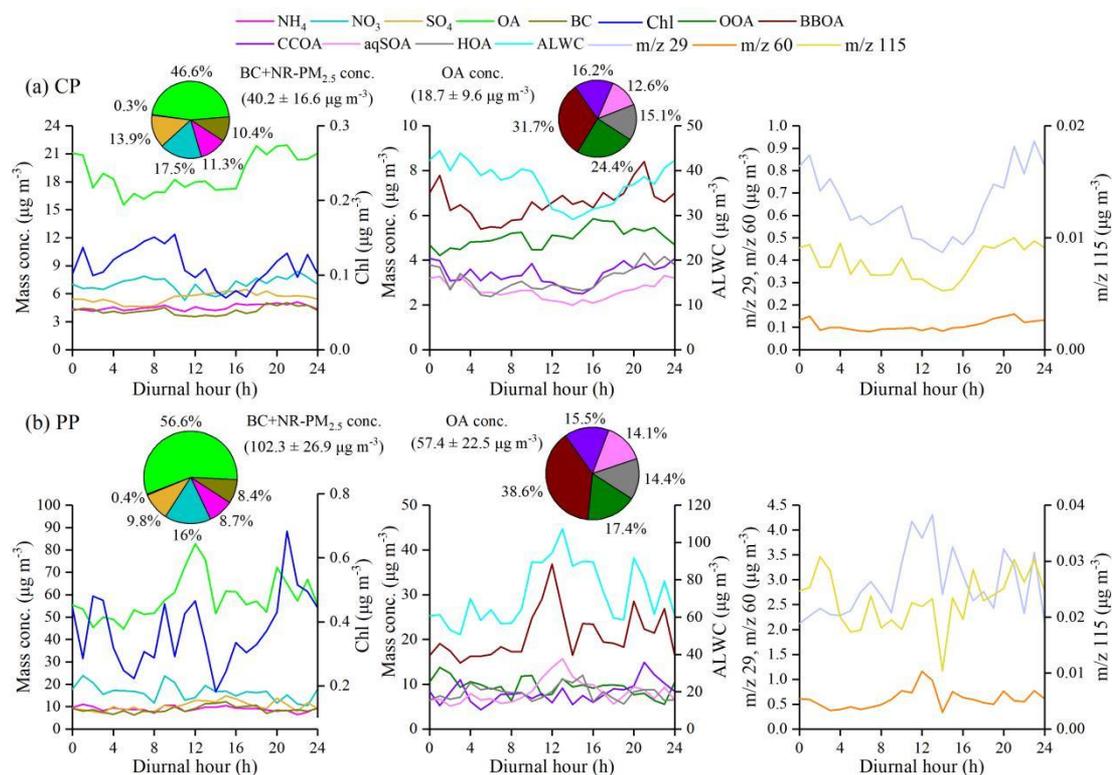


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373 **Figure 1.** Time series of (a) RH and T, (b) NR-PM_{2.5} species measured by ToF-ACSM and BC, (c)

374 Abs_{370,BrC} and MAC_{370,BrC}, and (d) mass fraction of OA factors during the campaign. The pollution

375 period (BC+NR-PM_{2.5} > 75 µg m⁻³) is highlighted by the shaded areas.



376

377 **Figure 2.** Diurnal variations of PM_{2.5} species, BC, OA factors, m/z 29, m/z 60, and m/z 115 mass
 378 concentrations during **(a)** clean period (CP) (BC+NR-PM_{2.5} < 75 μg m⁻³) and **(b)** polluted period
 379 (PP) (BC+NR-PM_{2.5} > 75 μg m⁻³). The pie charts in the left side of **(a)** and **(b)** show the average
 380 mass contributions of different chemical compositions to BC+NR-PM_{2.5} during CP and PP,
 381 respectively. Meanwhile, the average mass contributions of OOA, BBOA, CCOA, aqSOA, and
 382 HOA in OA are shown in the pie charts in the middle of **(a)** and **(b)**, respectively.

383 3.2 Biomass-burning emissions as precursors for aqSOA

384 **Figure 3** shows a strong positive correlation between the mass fraction (%) of
 385 aqSOA in total PM_{2.5} and ALWC during the campaign ($r^2 = 0.64$, $p < 0.001$). The
 386 contribution of aqSOA increased with increasing f_{29} values (the normalized signal at
 387 m/z 29). It was important to note that the aqSOA factor showed significantly higher
 388 f_{29} and f_{60} values (0.167 and 0.011) than the OOA factor (0.017 and 0.002) (Fig. S9).
 389 Moreover, both aqSOA concentrations and f_{29} were strongly correlated with ALWC
 390 ($r^2 = 0.85$, 0.73 , $p < 0.001$) (Fig. 3). The average oxygen-to-carbon ratio (O:C) ratio of
 391 aqSOA factor (0.85) was 2.7 times that of BBOA factor (0.31), whereas the
 392 hydrogen-to-carbon (H:C) ratios were similar (1.74 for aqSOA and 1.81 for BBOA,

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393 respectively), indicating **replacement of a** hydrogen atom by a OH moiety (Lim et al.,
394 2010; Ng et al., 2011a). These results were similar to aqSOA observed in Italy and
395 Beijing (Gilardoni et al., 2016; Zhao et al., 2019).

396 **Previous studies indicated that aqueous-phase processes could play a role in SOA**
397 **formation from fossil fuel emissions (Ervens et al., 2011; Huang et al., 2023; Wang et**
398 **al., 2021; Xu et al., 2022; Yan et al., 2017).** For example, Wang et al. (2021) and Xu et
399 al. (2022) have highlighted the **role of** aqueous-phase reactions **in** SOA formation,
400 particularly in regions with **substantial** anthropogenic emissions. In this study, a
401 strong anticorrelation **was observed** between the mass fraction of fossil-fuel related
402 OA components (sum of CCOA, HOA and OOA) and ALWC at high f_{29} values ($r^2 =$
403 $0.48, p < 0.001$, **data** not shown), consistent with recent **findings** (Wang et al., 2021).
404 This indicated that aqSOA might also be produced by aqueous-phase reactions of
405 fossil-fuel related OA components.

406 **Figure 4 shows the relationships between ALWC and the concentrations of OA**
407 **factors as well as f_{29} during the campaign. The mass concentrations of all five**
408 **resolved** OA factors **generally** increased with **rising** ALWC. However, compared with
409 other OA factors, aqSOA and BBOA **increased more significantly**, from 1.1 and 4.9
410 $\mu\text{g m}^{-3}$ to 5.2 and $10.8 \mu\text{g m}^{-3}$ **respectively**, when $20 \mu\text{g m}^{-3} < \text{ALWC} < 60 \mu\text{g m}^{-3}$.
411 **Notably**, only aqSOA concentration **continued to rise** under high ALWC conditions ($>$
412 $100 \mu\text{g m}^{-3}$). **This enhancement likely results from the formation of more**
413 **water-soluble organic species, such as glyoxal and methylglyoxal, which are**
414 **subsequently** oxidized to aqSOA via aqueous-phase reactions in aerosol liquid water
415 (Carlton et al., 2007; Ervens et al., 2011; Tan et al., 2012).

416 As shown in Fig. 4b, the mass fraction of aqSOA **increased significantly** from
417 less than 5% at $\text{ALWC} < 20 \mu\text{g m}^{-3}$ to 17–22% at $\text{ALWC} > 60 \mu\text{g m}^{-3}$ **accompanied**
418 **by** a corresponding decrease in **the OOA fraction**. **In contrast, the contributions of**
419 **POA (BBOA+CCOA+HOA) and SOA (OOA+aqSOA) remained relatively stable**,
420 across different ALWC levels (58–68% and 32–42% **for POA and SOA, respectively**).
421 This result suggested a more intensive formation of aqSOA than OOA via
422 aqueous-phase reactions, although aqSOA might **also be** formed from OOA,

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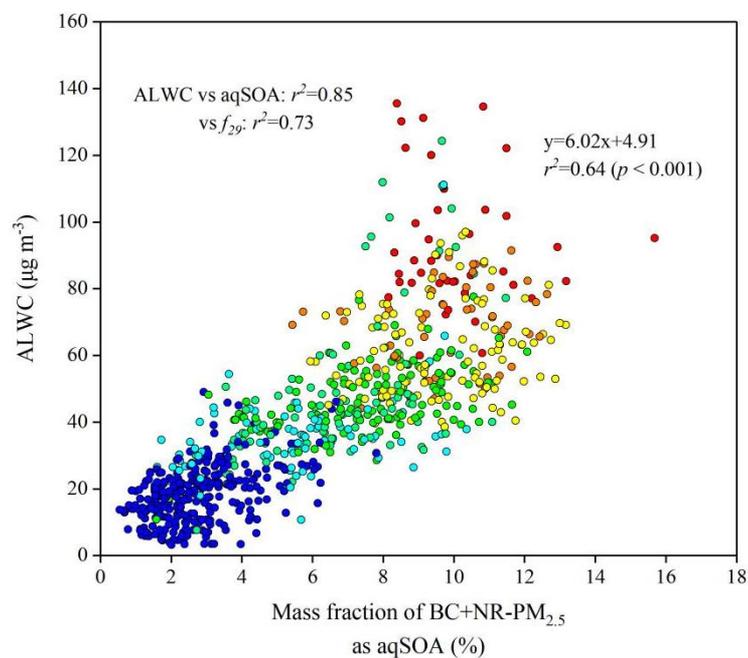
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423 consistent with the recent studies in northwestern China (Zhao et al., 2019; Zhong et
424 al., 2021).

425 Furthermore, f_{29} (CHO^+) increased from 0.010 to 0.227 as a function of ALWC
426 (Fig. 4b). A pronounced rise in f_{29} from 0.055 to 0.210 occurred when ALWC
427 increased from $60 \mu\text{g m}^{-3}$ to $100 \mu\text{g m}^{-3}$ ($p < 0.001$), tracking the concurrent increase
428 in OA mass concentrations ($13.2\text{--}109.1 \mu\text{g m}^{-3}$). Previous laboratory analysis of
429 organic standards have found that species without alcohol groups showed low f_{29} ($<$
430 0.05), while polyols and species with non-acid OH groups, common products of
431 biomass-burning emissions, display high f_{29} values (0.05–0.15) (Canagaratna et al.,
432 2015; Gilardoni et al., 2016; Zhao et al., 2014). This further supports the potential
433 formation of organic compounds with hydroxyl groups (i.e., glyoxal and
434 methylglyoxal) under high ALWC conditions. Overall, these results pointed to the fact
435 that the observed aqSOA could be formed predominantly from biomass-burning
436 emissions via aqueous-phase reactions, reinforcing the important role of BBOA in
437 increasing $\text{PM}_{2.5}$ mass concentration.



438 **Figure 3.** Scatter plot of the mass fraction of aqSOA in BC+NR- $\text{PM}_{2.5}$ versus ALWC colored by
439 the f_{29} (normalized signal at m/z 29) during the campaign. f_{29} (mainly CHO^+) is a tracer for alcohol
440 compounds and used to monitor the aqueous-phase oxidation of organic compounds (i.e., glyoxal).
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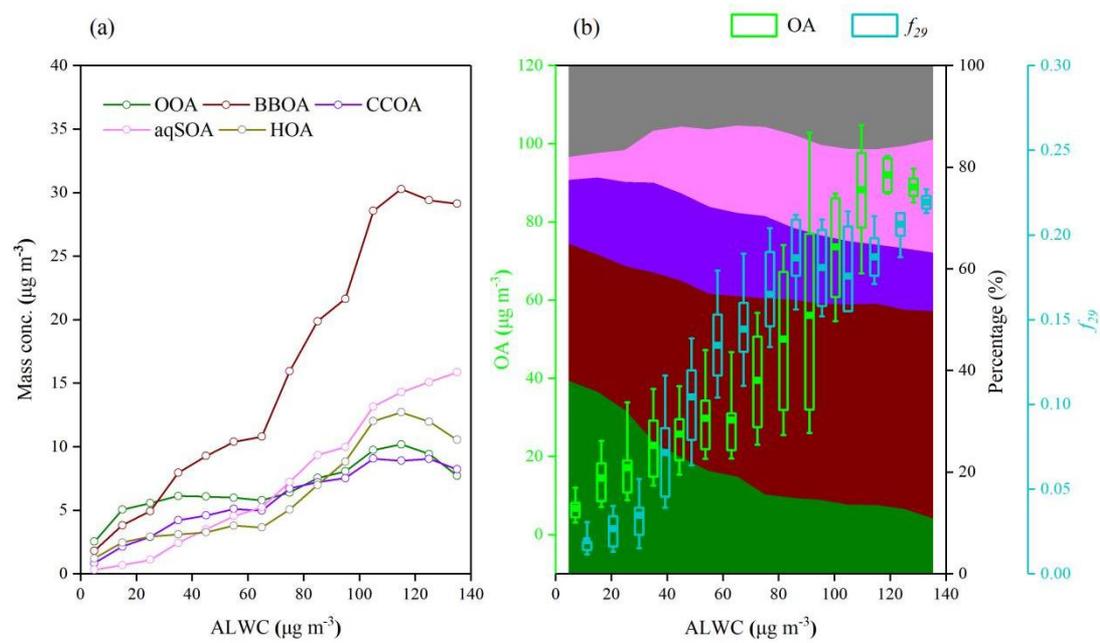
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443 **Figure 4.** Variations of (a) OA factors mass concentrations, and (b) OA mass concentrations, f_{29} (a
 444 tracer for alcohol compounds), and mass fraction of OA factors as a function of ALWC. Data were
 445 grouped into different bins according to a $10 \mu\text{g m}^{-3}$ increment of ALWC.

446 To identify the formation of aqSOA and its precursors under different $\text{PM}_{2.5}$
 447 pollution levels, the relationships between aqSOA and BBOA or OOA mass
 448 concentrations, as well as key ion-fragment tracers, were analyzed separately during
 449 CP and PP. The correlation (r^2) between aqSOA and BBOA, was stronger during PP
 450 (0.64) than during CP (0.54) ($p < 0.001$, Fig. 5a and c). Although both aqSOA and
 451 BBOA concentrations increased with rising ALWC during both periods, the
 452 correlations of ALWC with aqSOA and BBOA were relatively stronger during PP
 453 than during CP ($p < 0.001$). As shown in Fig. 5b and d, f_{29} was highly correlated with
 454 aqSOA formation during both CP and PP. A subset of data points with high aqSOA
 455 and OOA concentrations displayed relatively low f_{29} values (0.071–0.102) in Fig. 5d.
 456 For these points, the average value of f_{44} (normalized signal of m/z 44), (0.103 ± 0.024)
 457 was 1.3 times that of the overall dataset (0.080 ± 0.035) during PP, likely due to the
 458 formation of more-oxidized OOA under high ALWC values ($> 80 \mu\text{g m}^{-3}$) (Xu et al.,
 459 2017). Previous studies have found that f_{29} values for polyols and species with
 460 non-acid OH groups from biomass-burning emissions were typically lower than 0.15
 461 (Canagaratna et al., 2015; Gilardoni et al., 2016; Zhao et al., 2014). Moreover, the

462 mass fraction of aqSOA showed a stable increasing trend and remained **elevated** (from
463 18% to 22%) at ALWC > 80 $\mu\text{g m}^{-3}$, **accompanied by** a corresponding decrease in
464 OOA (from 15% to 10%) (Fig. 4b). **In contrast to** OOA ($p > 0.1$), aqSOA
465 concentration **showed a** strong positive correlation with ALWC ($r^2 = 0.73$, $p < 0.001$)
466 when ALWC > 80 $\mu\text{g m}^{-3}$ during PP. **However, no such correlations were observed**
467 during CP ($p > 0.1$). **Notably**, a strong anticorrelation between aqSOA and OOA
468 concentrations was observed during PP at ALWC > 80 $\mu\text{g m}^{-3}$ when $f_{29} > 0.15$ ($r^2 =$
469 0.76 , $p < 0.001$), but not during CP ($p > 0.1$) (Fig. 5b and d). These results indicated
470 that aqSOA formation was more intensive than OOA at high ALWC levels during PP.

471 Previous research demonstrated that f_{44} could be used as a tracer **for** aged SOA,
472 f_{43} (normalized signal at m/z 43) **for** POA and fresh SOA, and f_{60} (presence of
473 anhydrosugars) **for** BBOA (Cubison et al., 2011; Ng et al., 2010). Additionally, m/z 44
474 and 43 are usually from different functional groups and their ratio changes as a
475 function of atmospheric aging. The triangle plot of f_{44} versus f_{43} has been widely used
476 to characterize OA evolution, **while** f_{44} versus f_{60} is commonly used to **track** the aging
477 of BBOA (Ortega et al., 2013; Paglione et al., 2020; Xu et al., 2017; Xu et al., 2019).

478 **In this study**, the bottom region of the triangle (Fig. 6) was dominated by BBOA,
479 CCOA, and HOA, **which exhibited** low f_{44} (0.040, 0.017, and 0.016, respectively),
480 indicating **fresh, less-oxidized emissions**. **In contrast**, the f_{44} of SOA factors (i.e., OOA
481 and aqSOA) (0.118 and 0.117) were observably higher, **reflecting their** oxidized
482 **nature**. **The** f_{44} of aqSOA was close to that observed in fogs (Gilardoni et al., 2016;
483 Kim et al., 2019), highlighting the **role** of aqueous-phase reactions in this study.

484 **Moreover**, the relative abundance of m/z 45 (mainly HCO_2^+), a tracer ion for
485 carboxylic acids, was higher in the aqSOA **spectrum** than in the OOA spectrum (Fig.
486 S9), **consistent** with previous **findings**, that aqueous-phase reactions were important
487 sources of oxygenated organic compounds, including organic acids (Ervens et al.,
488 2011; Kim et al., 2019; McNeill, 2015; Sun et al., 2010; Yu et al., 2014). **As shown in**
489 Fig. 6b, BBOA and aqSOA **displayed** higher f_{60} values (0.019 and 0.011) than CCOA
490 (0.009) and HOA (0.008). The f_{60} value of OOA was 0.002, lower than the typical
491 background **level** (0.003) **observed** in the atmospheric **unaffected by** biomass burning,

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(Cubison et al., 2011). The mass spectrometry feature of aqSOA, characterized by
elevated f_{44} and f_{60} values, placed it in the schematic region of aged BBOA reported in
previous studies (Cubison et al., 2011; Ortega et al., 2013). Additionally, BBOA
contains abundant water-soluble organic compounds (WSOC) (i.e., sugars, phenols,
and organic acids) that can efficiently form aqSOA via aqueous-phase reactions (i.e.,
oxidation and oligomerization reactions) (Ervens et al., 2011; Gilardoni et al., 2016;
Lee et al., 2013; Lei et al., 2024; Li et al., 2020; Powelson et al., 2014). In contrast,
OOA formation primarily relies on gas-phase oxidation of VOCs with high-reactivity
(i.e., aromatics and long-chain alkanes, by OH radicals, which have low
concentrations in BBOA plumes (Akagi et al., 2011; Jimenez et al., 2009; Shrivastava
et al., 2017; Yokelson et al., 2007). This suggested that BBOA acted as the important
precursors for aqSOA instead of OOA via aqueous-phase reactions. These results
were consistent with previous studies and most of the observed data fall within the
triangle space (Bao et al., 2023; Kim et al., 2019; Paglione et al., 2020). It should be
noted that these processes of aqSOA formation could be more intense and important
during autumn due to elevated precursor concentrations (i.e., BBOA), ALWC, and RH
values, though the underlying chemical pathways are robust and can occur
year-round (Bao et al., 2023; Tang et al., 2025; Zeng et al., 2025).

During PP, f_{44} values ranging from 0.022 to 0.140 (0.080 ± 0.035) were
significantly higher than those during CP (0.021–0.150, 0.064 ± 0.019) ($p < 0.001$),
while f_{43} was slightly lower with an average of 0.062 ± 0.027 . Compared with CP (r^2
= 0.17, slope = -0.53), f_{44} showed a more pronounced inverse relationship with f_{43}
during PP, characterized by a higher r^2 value (0.70) and a regression slope closer to -1
(-1.09). This pattern indicated a greater abundance of aged SOA in the atmosphere
during PP (Fig. 6a and c). It should be noted that the data points of in the f_{44} vs f_{43}
space during PP fell within the upper boundary of the triangle region, with most
points located outside the bottom boundary. These results suggested that less oxidized
SOA were predominantly formed via aqueous-phase reactions instead of
photochemical pathways during PP (Kim et al., 2019; Zhao et al., 2019). Moreover,
points outside the bottom boundary of the triangle region, characterized by higher f_{44}

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522 (> 0.05) and lower f_{43} (< 0.06), were associated with relatively higher ALWC during
523 PP, a feature, not observed during CP.

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524 Here, the triangle plots of f_{44} versus f_{60} colored by ALWC under different PM_{2.5}
525 pollution levels were analyzed (Fig. 6b and d), when the link between aqSOA and
526 BBOA was further stressed by a schematic representation of aged BBOA. With the
527 exception of a few outliers, f_{60} values were consistently higher than 0.003, and most
528 data points fell within the triangular region, suggesting a widespread influence of
529 biomass burning on OA. During PP, f_{60} ranged from 0.005 to 0.019 (0.010 ± 0.004),
530 similar to the range observed in CP (from 0.004 to 0.019, 0.010 ± 0.003). However,
531 the correlation between f_{44} and f_{60} was higher during PP ($r^2 = 0.72$) than during CP (r^2
532 $\equiv 0.31$) ($p < 0.001$). Moreover, data points within the schematic space of aged BBOA
533 showed relatively higher ALWC compared with overall dataset during PP, a pattern
534 that was not evident during CP.

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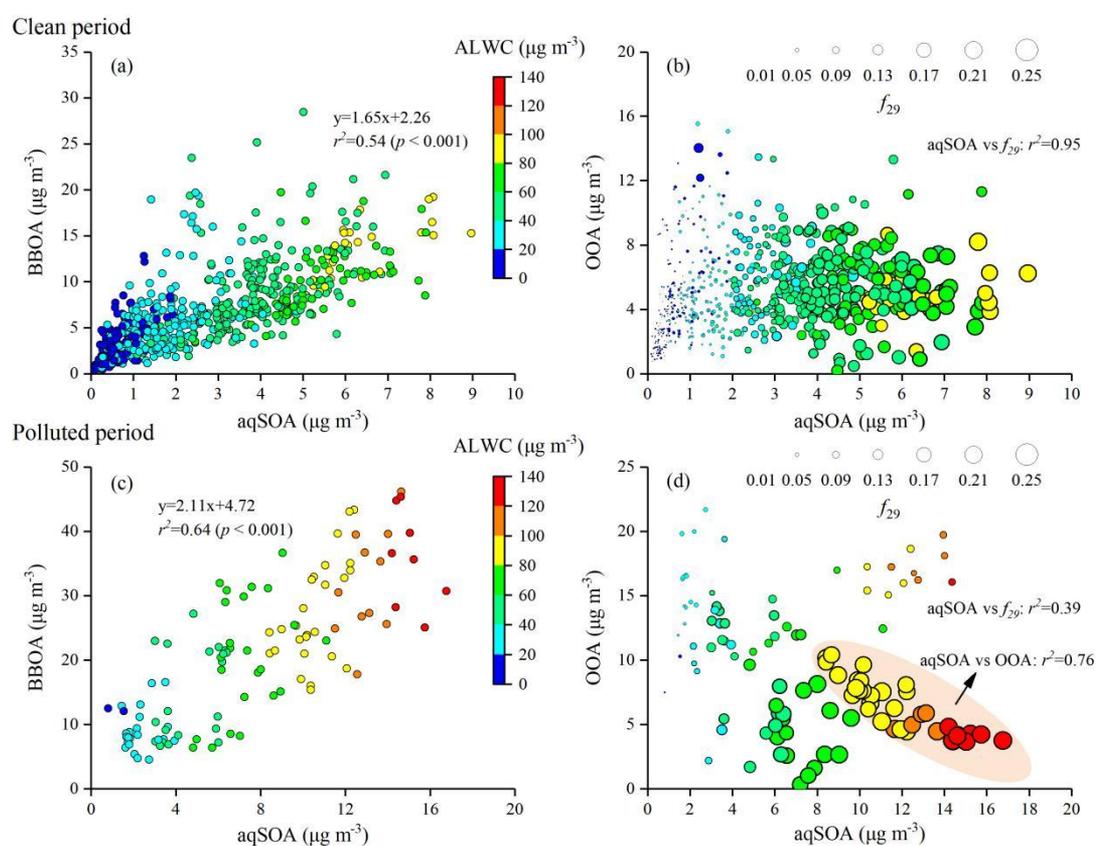
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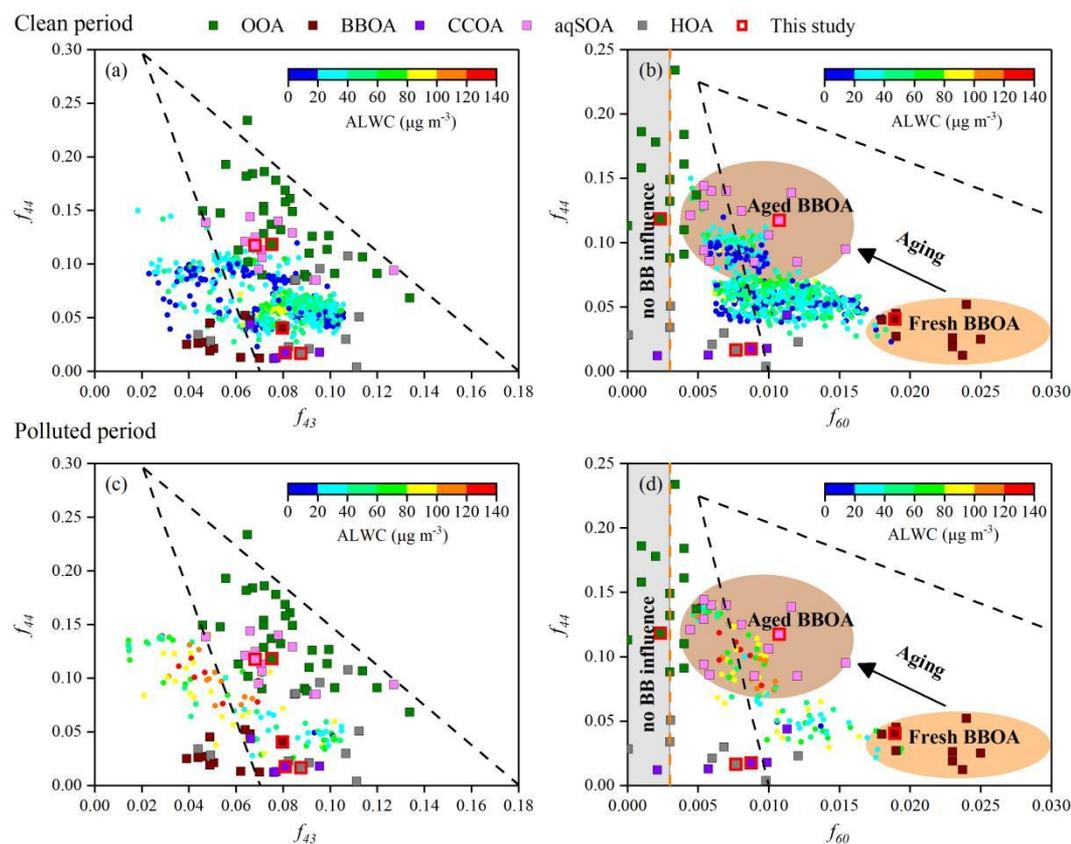
535 Regional transport significantly influenced the aging of BBOA in the SCB.
536 Previous studies have demonstrated that northeast winds prevail during autumn in this
537 region, facilitating the transport of pollutants along the Dazhou→Guang'an→Hechuan
538 pathway, and this northeast-southwest transport pathway had a significant impact on
539 Chongqing (Peng et al., 2019; Wang et al., 2018). As shown in Fig. S11, air masses
540 during the campaign predominantly originated from northeastern Chongqing, an area
541 with widespread agricultural burning activities (He et al., 2015; Luo et al., 2020), and
542 were transported over relatively short distances. Compared to other air mass clusters,
543 the highest contribution and concentration of BBOA to total PM_{2.5} were observed
544 when air masses passed through northeastern Chongqing. The percentage of air mass
545 trajectories that passed through biomass-burning-influenced regions was higher
546 during PP (~57%) than CP (~35%). Unlike cluster 3 during CP, air masses originating
547 from northeastern Chongqing (cluster 2) during PP showed significantly higher
548 contributions and concentrations of BBOA (24.7%, $27.8 \mu\text{g m}^{-3}$) and aqSOA (9.4%,
549 $10.6 \mu\text{g m}^{-3}$) than other clusters ($p < 0.001$). In addition, cluster 2 exhibited notably
550 higher values of ALWC ($93.7 \mu\text{g m}^{-3}$), $\text{Abs}_{370, \text{BrC}, \text{sec}}$ (105.1 Mm^{-1}), f_{44} (0.113), and the

551 NO₃/SO₄ ratio (2.1, a tracer for BBOA aging (Liu et al., 2024; Zhang et al., 2025))
 552 than other clusters during PP ($p < 0.001$). During PP, approximately 68% of the
 553 trajectories in cluster 2 passed through biomass-burning-influenced regions, with
 554 transport times to the sampling site ranging from 12 to 48 hours, consistent with the
 555 typical timescale for aqueous-phase aging of biomass burning emissions reported in
 556 earlier works (Cubison et al., 2011; Hennigan et al., 2010; Ortega et al., 2013; Zhu et
 557 al., 2023). These results suggested that regional transported BBOA, primarily
 558 originated from northeastern Chongqing, underwent aging to form aqSOA within
 559 approximately 12 to 48 hours.

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560
 561 **Figure 5.** Scatter plots of aqSOA versus (a, b) BBOA and (c, d) OOA mass concentrations
 562 colored by ALWC during clean period and polluted period. The size of the symbols in (b) and (d)
 563 increases with the increase of the f_{29} value, which is a tracer for alcohol compounds.



564

565 **Figure 6.** Triangle plots of **(a, c)** f_{44} (normalized mass spectrum signal at m/z 44) versus f_{43}
 566 (normalized mass spectrum signal at m/z 43), and **(b, d)** f_{44} versus f_{60} (normalized mass spectrum
 567 signal at m/z 60) colored by ALWC (circles) during clean period and polluted period. The dashed
 568 lines in **(a)** and **(c)** were derived from Ng et al. (2010) and used to follow the aging of OA
 569 components in the atmosphere. The background space ($f_{60} < 0.003$) without biomass burning
 570 influence was also shown by the grey shaded area. The background value of secondary aged OA
 571 (brown dashed line) and the black dashed lines characterizing the aging of BBOA in **(b)** and **(d)**
 572 were derived from Cubison et al. (2011). The data points (squares) included the measurements in
 573 this study (bordered in red) and previous research (Bao et al., 2023; Gilardoni et al., 2016; Kim et
 574 al., 2019; Ng et al., 2011a; Paglione et al., 2020; Xu et al., 2015; Xu et al., 2017; Xu et al., 2019;
 575 Zhao et al., 2017; Zhao et al., 2019). f_{43} (mainly $C_2H_3O^+$) is a tracer for POA and fresh SOA. f_{44} is
 576 a proxy of the OA oxygenation degree and used as a tracer for aged SOA. f_{60} is a proxy of
 577 anhydrosugars emitted from biomass burning.

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578 3.3 Evolution of BrC Absorption

579 Previous research indicated that OA from **both** fresh and aged biomass-burning
580 emissions, exhibited **light** absorption, across UV to Vis range, with significantly higher
581 AAE value than BC, **and** might contribute to a net positive radiative forcing (Laskin
582 et al., 2015). **In this work**, the absorption properties **of BrC** and their relationships
583 with **the** five OA factors were analyzed. The values of $Abs_{\lambda, BrC, pri}$ and $Abs_{\lambda, BrC, sec}$ were
584 obtained by MRS method, and MLR method was used to estimate Abs of **each** OA
585 factor, at each wavelength (SI Text S3 and S4). The average value of $Abs_{370, BrC}$ was
586 $42.4 \pm 28.5 \text{ Mm}^{-1}$ (accounting for 49.2% of Abs_{370}), much higher than $Abs_{660, BrC}$ (2.6
587 $\pm 1.3 \text{ Mm}^{-1}$, 10.5%), suggesting high absorption efficiency for BrC in the near-UV
588 **region. From 370 nm to 660 nm**, $Abs_{\lambda, BrC, pri}$ and $Abs_{\lambda, BrC, sec}$ accounted for
589 56.8%–72.5% and 27.5%–43.2% of $Abs_{\lambda, BrC}$, respectively, indicating **that** primary
590 emissions were the **dominant** contributors to BrC absorption (Fig. S13). However, the
591 contribution of $Abs_{\lambda, BrC, sec}$ to $Abs_{\lambda, BrC}$ increased with wavelength, suggesting **that** the
592 impact on Abs_{BrC} from SOA should not be ignored. **The following analysis**
593 **demonstrates** that aqSOA **formed** from aged BBOA contributed **substantially** to the
594 BrC budget and **exhibited** strong **light** absorption across UV to Vis range.

595 **Data at 370 nm, which exhibit** higher signal-to-noise ratios and **a greater**
596 **contribution of** Abs_{BrC} , **were selected** to further analyze the correlations of BrC
597 absorption with various OA components. As described in section 2.3.3, the Abs of **the**
598 five OA factors at each wavelength were obtained by MLR method (Table S2). **The**
599 **contributions of** BBOA ($Abs_{370, BBOA}$) and aqSOA ($Abs_{370, aqSOA}$) **to** $Abs_{370, BrC}$ **were**
600 51.9% and 16.4%, **respectively, higher than those of CCOA ($Abs_{370, CCOA}$, 11.5%),**
601 **HOA ($Abs_{370, HOA}$, 9.1%), and OOA ($Abs_{370, OOA}$, 11.1%). This pattern is** consistent
602 with the higher MAC values **of BBOA and aqSOA** (Fig. S15).

603 **Figure S16** presents correlations between $Abs_{370, BrC}$ and the mass concentrations
604 of OOA, BBOA, CCOA, aqSOA, HOA, and **the tracer ion m/z 60 (ion fragments**
605 **tracers of BBOA)**. $Abs_{370, BrC}$ showed the strongest positive correlations with BBOA
606 and m/z 60, ($r^2 = 0.77$, $p < 0.001$), followed by aqSOA, ($r^2 = 0.69$, $p < 0.001$). In

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607 contrast, correlations with HOA ($r^2 = 0.36$), CCOA ($r^2 = 0.25$), and OOA ($r^2 = 0.09$,
608 $p > 0.1$) were much weaker. Among all OA factors (excluding BBOA), aqSOA
609 contributed relatively more to $Abs_{370,BrC}$ (Table S2), and also displayed a stronger
610 correlation with $Abs_{370,BrC}$. These results might be related to the formation of aqSOA
611 from aged BBOA via aqueous-phase reactions. Gilardoni et al. (2016) demonstrated
612 that aqSOA derived from aged BBOA via aqueous-phase reactions in the ambient
613 atmosphere contributed to the BrC budget and exhibited slightly higher $AAE_{467-660}$
614 (AAE of aerosols from 467 nm to 660 nm) values than fresh or processed
615 biomass-burning emissions in laboratory experiments.

616 The MAC values of the five resolved OA components (equivalent to
617 coefficients a–e in the MLR method) at different wavelengths were shown in Fig. S15.
618 At 370 nm, BBOA showed the highest MAC ($2.37 \text{ m}^2 \text{ g}^{-1}$), followed by aqSOA (1.23
619 $\text{m}^2 \text{ g}^{-1}$), indicating that oxidation of BBOA to aqSOA reduced light absorption at
620 shorter wavelengths. Previous research found that the MAC of BBOA at 365 nm was
621 twice that of SOA, which was associated with the water-soluble BrC (Lorenzo et al.,
622 2017; Washenfelder et al., 2015). The AAE values of the OA factors, calculated by a
623 power-law fitting of their Abs from 370 nm to 660 nm (Qin et al., 2018; Wang et al.,
624 2019b), were shown in Fig. S15. It should be noted that aqSOA had the lowest
625 $AAE_{370-660,aqSOA}$ value (3.54), while BBOA had the highest $AAE_{370-660,BBOA}$ value
626 (4.93). Moreover, the contribution of aqSOA to Abs_{BrC} increased from 16.4% at 370
627 nm to 26.7% at 660 nm, while the contribution of BBOA decreased from 51.9% to
628 39.1% over the same wavelength range. These findings suggested aqSOA formed
629 from aged BBOA might play an important role in the light absorption of BrC across
630 the UV to Vis range.

631 Figure 7 shows a ternary contour map that quantifies the contributions of BBOA,
632 CCOA, and HOA to $Abs_{370,BrC,pri}$. A strong positive correlation ($p < 0.001$) and a steep
633 linear regression slope (1.80) were observed between BBOA mass concentration and
634 $Abs_{370,BrC,pri}$. Among these POA factors, high mass fractions of BBOA relative to POA
635 were consistent with high values of $Abs_{370,BrC,pri}$ (Fig. 7a). For example, most data
636 points with $Abs_{370,BrC,pri}$ higher than 49.1 Mm^{-1} (90th percentile of $Abs_{370,BrC}$) fell

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637 within the region of high BBOA/POA values (> 0.5). Moreover, $Abs_{370,BrC,pri}$
638 increased significantly with rising BBOA and m/z 60 mass concentrations, exhibiting
639 higher correlation coefficients, ($r^2 = 0.63$ and 0.55) than HOA ($r^2 = 0.19$) and CCOA
640 ($r^2 = 0.14$) (Fig. 7b). These results indicated the major contribution of BBOA to
641 primary BrC light absorption at 370 nm.

642 To understand the relationship between secondary BrC absorption and its
643 chromophores, the correlation between $Abs_{370,BrC,sec}$ and the mass concentrations of
644 SOA factors, was analyzed. As shown in Fig. S17, $Abs_{370,BrC,sec}$ increased significantly
645 with rising aqSOA concentration, ($r^2 = 0.44$, $p < 0.001$), and high $Abs_{370,BrC,sec}$ values
646 were associated with elevated ALWC. In contrast, no such relationship was found for
647 OOA ($p > 0.1$). The slope of the linear regression, between aqSOA mass
648 concentrations and $Abs_{370,BrC,sec}$ (3.50) was steeper than that for OOA (Fig. S17), and
649 the MAC values of aqSOA was also higher across UV to Vis range (Fig. S15).

650 To further characterize the evolution of secondary BrC absorption, $Abs_{370,BrC,sec}$
651 was normalized by ΔCO (background-corrected CO mixing ratios) to minimize the
652 effect of boundary layer height (Fig. 8) (DeCarlo et al., 2010). Here, the background
653 CO value (400 ppb) was defined as the lowest 1.25th percentile of CO values during
654 the campaign (Kondo et al., 2006). Figure 8 shows that the ratios of $Abs_{370,BrC,sec}/\Delta CO$
655 increased with both aqSOA concentration and ALWC, from 17:00 to 03:00 LT ($r^2 =$
656 0.63 , 0.57 , $p < 0.001$), while $Abs_{370,BrC,pri}/\Delta CO$ decreased slightly with increasing
657 BBOA and m/z 60 concentrations over the same period ($r^2 = 0.35$, 0.33 , $p < 0.001$).
658 Additionally, the mass concentrations of NO_3 , NH_4 , and NO_2 from 17:00 to 03:00 LT
659 were 1.2, 1.2, and 1.3 times those from 04:00 to 16:00 LT, respectively. These results
660 were consistent with previous winter observations in the SCB (Peng et al., 2025; Wu
661 et al., 2024).

662 As discussed in section 3.2, SOA with hydroxyl groups (i.e., glyoxal and
663 methylglyoxal) could form from aged BBOA via aqueous-phase reactions under high
664 ALWC. Previous studies have shown that oligomers (involving two glyoxal
665 molecules) formed via aqueous reactions of glyoxal with NH_3 , which contain C=C or
666 C=N bonds, exhibit strong absorption at near-UV (Laskin et al., 2015; Lee et al., 2013;

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667 Nozière et al., 2009; Powelson et al., 2014). This suggested that secondary BrC
668 chromophores with strong absorption at 370 nm were formed under high ALWC from
669 17:00 to 03:00 LT, which might be related to aqSOA generated from aged BBOA via
670 aqueous-phase reactions. The low values of $Abs_{370, BrC, sec} / \Delta CO$ at 12:00–14:00 LT
671 could be related to photolysis and/or photooxidation causing BrC photobleaching
672 (Sareen et al., 2013; Zhao et al., 2015). Overall, our findings suggested that aqSOA
673 formed from biomass-burning emissions contributed importantly to BrC absorption,
674 especially at night during the campaign.

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675 Previous studies indicated that biomass-burning activity is negligible in summer,
676 and although it may also experience in spring and winter, its intensity is typically less
677 intense than in autumn (Chen et al., 2014; Chen et al., 2017; Tao et al., 2014; Yang et
678 al., 2019). For example, Tao et al. (2014) found that the contribution of biomass
679 burning to PM_{2.5} during autumn (19 ± 11%) was significantly higher than in other
680 seasons. Meanwhile, the highest fraction of organic matter (OM) in PM_{2.5} was also
681 observed during autumn (33.4%), with biomass burning as the dominant contributor.
682 Additionally, the concentration of BBOA and its fraction in OA during autumn in this
683 study (8.6 ± 7.7 μg m⁻³ and 34.8 ± 11.2%, respectively) were respectively higher than
684 those observed during summer (0.41 μg m⁻³ and 5.7%, respectively) (Zeng et al.,
685 2025) and winter in the SCB (Tang et al., 2025; Zhang et al., 2023) (Table S1). It
686 should be noted that the ALWC during autumn in this study (41.6 ± 24.9 μg m⁻³) was
687 substantially higher than that reported for summer (18.6 ± 35.3 μg m⁻³) (Zeng et al.,
688 2025) and winter (27.4 ± 9.7 μg m⁻³) (Tang et al., 2025) in the SCB. Wang et al. (2018)
689 observed that RH values at Chengdu and Chongqing in autumn were also higher than
690 those in other seasons. Previous research indicated that while aqueous-chemistry
691 pathways in spring were comparable to those in autumn, photochemical bleaching of
692 BrC was potentially stronger in spring (Wang et al., 2019a). Although winter features
693 lower biomass-burning emissions, secondary BrC could still form, from BBOA
694 through aqueous-phase reactions under high NO_x and NH₄ concentrations and
695 stagnant nighttime conditions, as observed during winter in the SCB (Peng et al., 2025;
696 Wu et al., 2024). In summer, elevated temperature and O_x (O_x = NO₂ + O₃) levels

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697 could enhance photochemical oxidation, promoting secondary BrC formation, while
698 also intensifying BrC photobleaching (Wu et al., 2024). In summary, while secondary
699 BrC can form from BBOA through aqueous-phase reactions in all seasons in the SCB
700 (with the possible exception of spring), the elevated ALWC and BBOA concentrations
701 during autumn are particularly favorable for its aqueous-phase formation.

702 AAE₃₇₀₋₈₈₀ is another key parameter for characterizing aerosol absorption
703 properties. Figure 9 displays its correlations with the mass fraction of aqSOA (f_{aqSOA})
704 and BBOA (f_{BBOA}) to OA, and the BC-to-OA ratio. During the campaign, a strong
705 positive correlation ($r^2 = 0.49$, $p < 0.001$) was observed between AAE₃₇₀₋₈₈₀ and f_{aqSOA} ,
706 with AAE₃₇₀₋₈₈₀ reaching values up to 2.65. In contrast, AAE₃₇₀₋₈₈₀ increased only
707 slightly with rising f_{BBOA} ($r^2 = 0.21$, $p < 0.001$) (Fig. 9a and c). AAE was calculated
708 using a power-law fitting of aerosol absorption values (Qin et al., 2018; Wang et al.,
709 2019b). While BC concentration was linearly dependent on Abs_{BC}, OA concentration
710 did not follow the same pattern with Abs_{BrC}. The mixing state of BC and OA,
711 influenced by combustion conditions, can also affect AAE. Previous studies have
712 shown that biomass-burning emissions can impact absorption properties, which is
713 reflected in the relationship between AAE and the BC-to-OA ratio (a measure of the
714 combustion conditions) (Lu et al., 2015; Saleh et al., 2014). Thus, the relationship
715 observed in Fig. 9b reflected the influence of biomass-burning emissions during this
716 campaign. The parameterized curve obtained here (black) was consistent with prior
717 research (red) derived for wavelengths from 370 nm to 880 nm (Lu et al., 2015).
718 When 950 nm was used as the highest wavelength instead of 880 nm, AAE₃₇₀₋₉₅₀
719 values differed by less than 10% from AAE₃₇₀₋₈₈₀. It should be noted that data points
720 with high AAE₃₇₀₋₈₈₀ generally corresponded to low BC-to-OA ratios and large f_{aqSOA}
721 values. Moreover, the average value of AAE₃₇₀₋₈₈₀ observed in this study (1.95) was
722 higher than AAE₃₇₀₋₉₅₀ observed in the laboratory experiments on fresh and
723 photochemically aged biomass-burning emissions (i.e., 1.38 and 1.48 for fresh oak
724 and pocosin pine, 1.42 and 1.73 for aged oak and pocosin pine) (Saleh et al., 2013).

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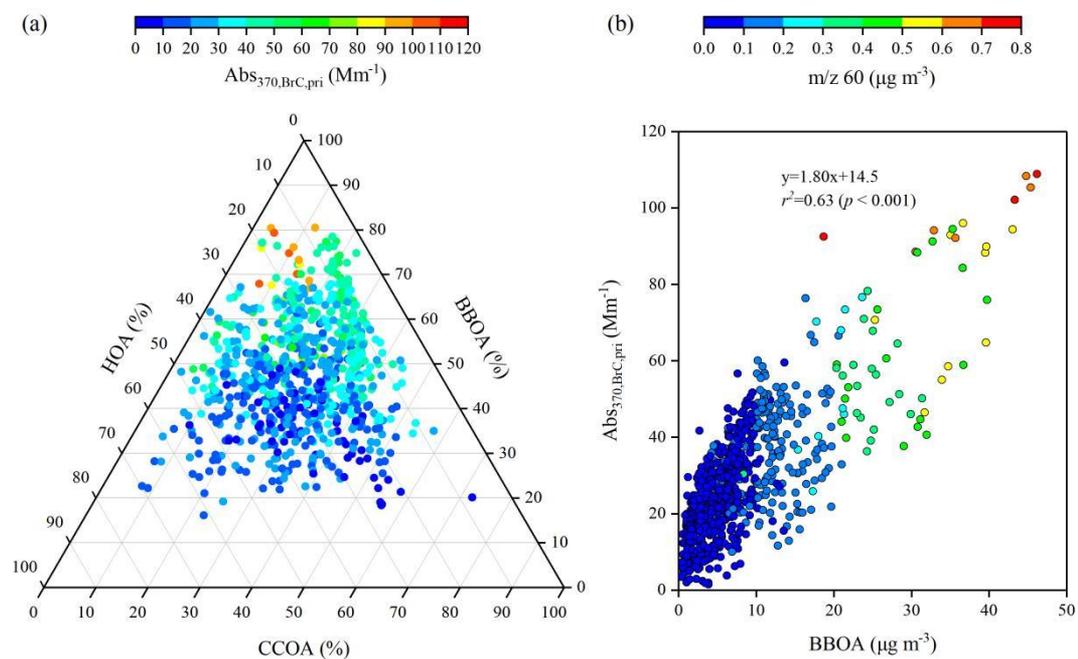
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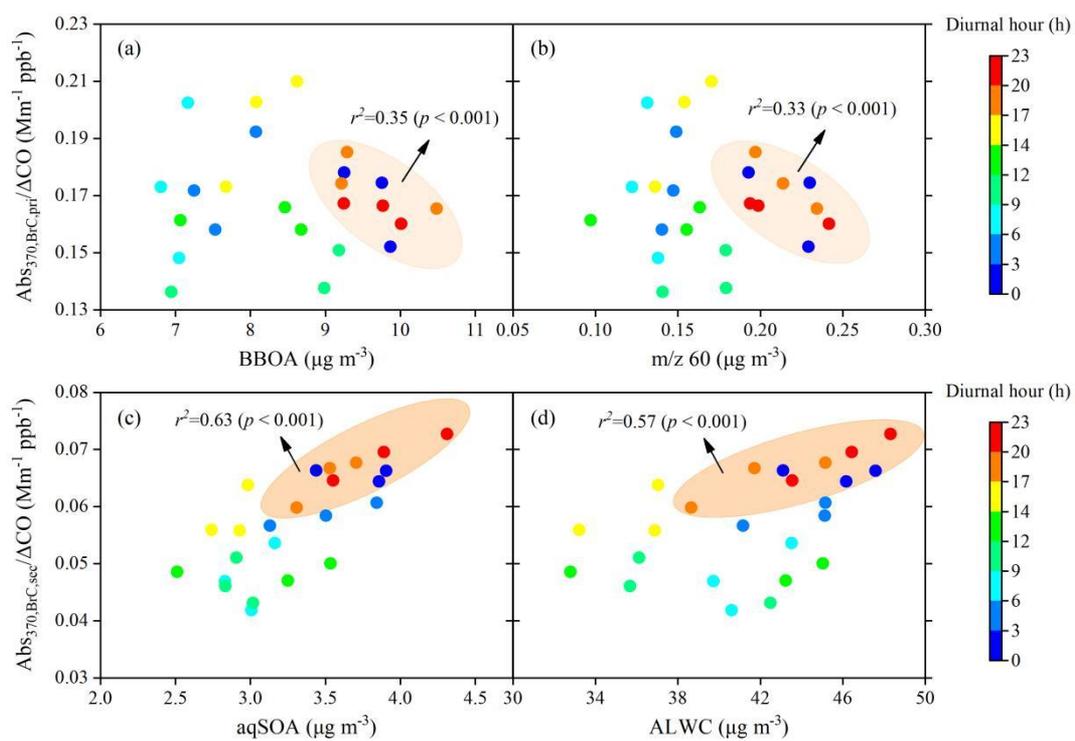
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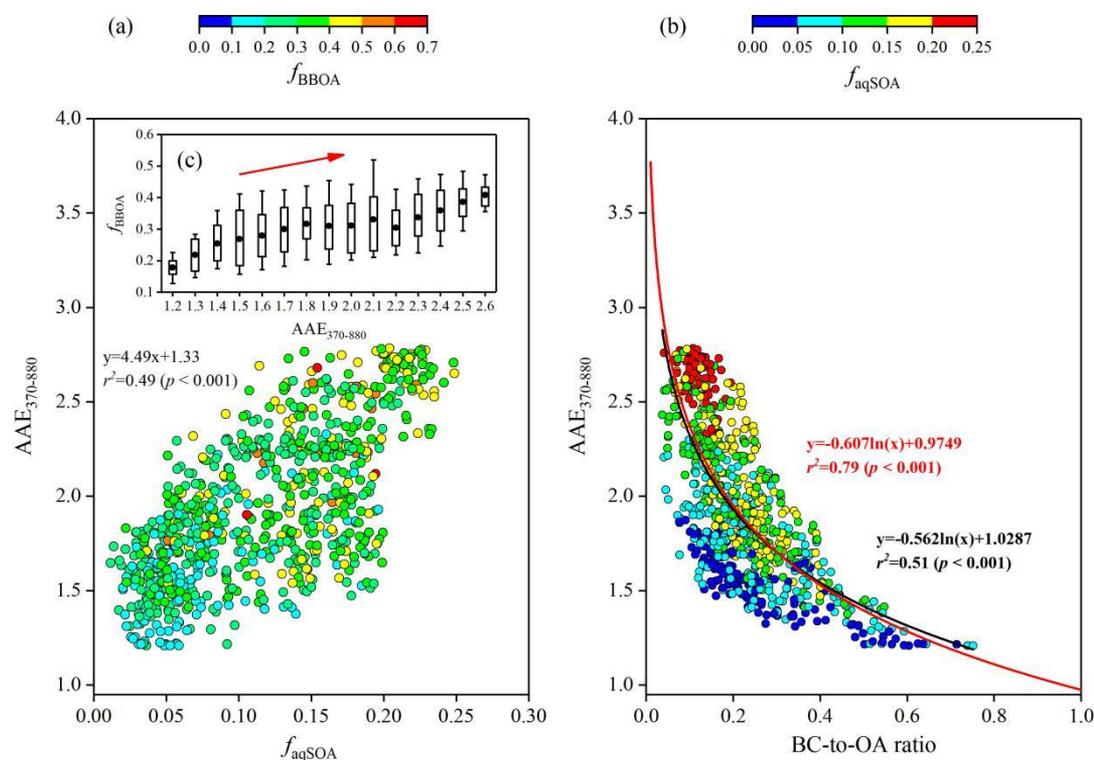
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726 **Figure 7.** (a) Ternary diagram for the mass fractions of BBOA, CCOA, and HOA in POA colored
 727 by Abs_{370,BrC,pri}, and (b) scatter plot of BBOA mass concentrations versus Abs_{370,BrC,pri} colored by
 728 m/z 60 mass concentrations.



729

730 **Figure 8.** Scatter plots of $\text{Abs}_{370,\text{BrC,pri}}/\Delta\text{CO}$ versus **(a, b)** BBOA and m/z 60 mass concentrations
 731 and $\text{Abs}_{370,\text{BrC,sec}}/\Delta\text{CO}$ versus **(c, d)** aqSOA and ALWC colored by the local time.



732
 733 **Figure 9.** Relationship between **(a)** $\text{AAE}_{370-880}$ and the mass fraction of aqSOA ($f_{\text{aqSOA}} =$
 734 aqSOA/OA) colored by the mass fraction of BBOA ($f_{\text{BBOA}} = \text{BBOA/OA}$), and **(b)** BC-to-OA ratios
 735 colored by f_{aqSOA} . **(c)** Variations of f_{BBOA} as a function of $\text{AAE}_{370-880}$. The red curve in **(b)** was the
 736 best fit curve to data taken from Lu et al. (2015) and described the Abs of fresh and aged BBOA.

737 4 Conclusions

738 This study conducted comprehensive real-time measurements of the light
 739 absorption properties and chemical composition of carbonaceous aerosols during
 740 autumn in the Sichuan Basin, China. The findings provided direct ambient evidence
 741 that aqueous secondary organic aerosol (aqSOA) formed from aged biomass-burning
 742 emissions under high aerosol liquid water content ($\text{ALWC} > 60 \mu\text{g m}^{-3}$) conditions
 743 significantly contributes to both aerosol pollution and light absorption. Organic
 744 aerosol (OA) was the dominant component of $\text{PM}_{2.5}$ ($46.6 \pm 10.7\%$), and exhibited

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745 strong absorption, at UV wavelengths ($Abs_{370,BrC} = 42.4 \pm 28.5 \text{ Mm}^{-1}$). During
746 pollution periods (PP), aqSOA (averaged at $7.6 \mu\text{g m}^{-3}$) contributed on average 14.1%
747 to OA, exhibiting enhanced oxidation ($f_{29} = 0.141 \pm 0.062$, $f_{44} = 0.080 \pm 0.035$) and
748 substantial light absorption characteristics ($Abs_{370,BrC} = 91.6 \text{ Mm}^{-1}$, $AAE_{370-880} = 2.1$).
749 Additionally, less-oxidized aqSOA, formed predominantly via aqueous-phase
750 reactions, instead of gas-phase photochemical oxidations of their precursors, played a
751 key role in the dynamic evolution of haze pollution during PP. Backward trajectory
752 analysis further revealed that regionally transported BBOA, originated mainly from
753 northeastern Chongqing, underwent aqueous-phase aging to aqSOA within
754 approximately 12 to 48 hours. These results underscore that aqueous-phase reactions
755 of BBOA—particularly during the transport of biomass-burning plumes—converts
756 primary emissions into strongly light-absorbing aqSOA, thereby substantially
757 influencing regional haze formation and radiative forcing.

758 Our findings align with previous laboratory studies on biomass-burning BrC
759 formation (Lu et al., 2015; Powelson et al., 2014), while providing novel ambient
760 quantification of these processes under realistic atmospheric conditions. The
761 parameterized curve of $AAE_{370-880}$ versus BC-to-OA ratios in this study was
762 consistent with the previous laboratory research on biomass-burning emissions. The
763 mean $AAE_{370-880}$ observed in this study (1.95) were higher than values reported for
764 fresh and photochemically aged biomass-burning emissions in laboratory experiments
765 (Saleh et al., 2013), and increased significantly with rising f_{aqSOA} ($r^2 = 0.49$, $p < 0.001$).
766 Additionally, elevated $Abs_{370,BrC,sec}$ values coincided with high ALWC, NO_3 , and NH_4
767 levels, and correlated strongly with aqSOA concentration ($r^2 = 0.44$, $p < 0.001$). These
768 results suggest that aqueous-phase reactions of BBOA under high- NO_x and high- NH_4
769 conditions produce secondary BrC with particularly strong light absorption. It should
770 be noted that seasonal variations in biomass-burning emissions and the associated
771 chemical processing of carbonaceous aerosols must be adequately represented in
772 climate and air quality models. This is critical to avoid underestimating aerosol
773 impacts in autumn and overestimating them in other seasons. The campaign was
774 conducted during autumn, when biomass-burning activity is intense, and thus may not

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775 fully represent aerosol processes in other seasons. The relative uncertainty of Abs_{BrC}
776 at 370 nm, resulting from the choice of AAE_{BC}, ranged from -112% to 42%.
777 Nevertheless, our results underscore the importance of aqueous-phase processing in
778 transforming biomass-burning emissions, with important implications for climate and
779 air quality modeling. The substantial contribution of aqSOA to both aerosol mass and
780 light absorption highlights the need for improved representation of aqueous processes
781 in models. The linkages established here among aging timescales, transport pathways,
782 and aqSOA formation, provide a transferable framework for understanding aqSOA
783 processing in other humid regions influenced by biomass burning. In general, research
784 on BrC chromophores is still at an early stage, and more studies that quantitatively
785 link the chemical composition and light absorption properties of BrC chromophores to
786 biomass-burning emissions across different seasons are necessary to improve our
787 understanding of their climatic and environmental effects. Future research should
788 prioritize molecular-level characterization of aqSOA precursors and products,
789 quantification of aqueous reaction rates under ambient conditions, and multi-scale
790 modeling to assess regional climate impacts. This study highlights that aqueous
791 processes play an important role in the evolution of biomass-burning emissions and
792 should be adequately considered in both air quality budgets and climate forcing
793 balance on a global scale.

794

795 **Data availability.** The data generated and analysed in this study are available from
796 <https://doi.org/10.5281/zenodo.18635386> (Peng et al., 2026).

797

798 **Author contributions.** CZ, CP, YD, and ZL designed the experiments. Data analysis
799 and interpretation were performed by CP, ZT, HT, KZ, ZL, and GS. CP, XY, and MT
800 wrote the paper. ZT, YC, XL, LZ, YC, and YF contributed to the paper with useful
801 scientific discussions or comments.

802

803 **Competing interests.** The authors declare that they have no conflict of interest.

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