



1 On the presence of high nitrite (NO_2^-) in coarse particles at Mt.

2 Qomolangma

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22 **Abstract**

23 Atmospheric reactive nitrogen cycling is crucial for maintaining the atmospheric oxidation capacity
24 of background atmosphere on the Tibetan Plateau, with nitrous acid (HONO) and particulate nitrite
25 (NO_2^-) as important intermediates. During an eleven-day field campaign at the Base Camp of Mt.
26 Qomolangma in spring of 2022, we observed significant enrichment of NO_2^- in total suspended
27 particulate (TSP) with a mean concentration of $375 \pm 386 \text{ ng m}^{-3}$, while NO_2^- was absent in fine
28 particles ($\text{PM}_{2.5}$). The comparison revealed that NO_2^- predominately exists in coarse particles. Local
29 surface soil at the sampling site also exhibited high levels of NO_2^- , with $\delta^{15}\text{N}$ value similar to NO_2^-
30 in TSP. This similarity suggests that wind-blown soil is probably the primary source of NO_2^- in TSP,
31 accounting for the background levels. While concentration changes of water-soluble inorganic ions
32 in TSP and $\text{PM}_{2.5}$ in response to shifts in air mass back-trajectories imply that atmospheric pollutants
33 transported from South Asia may further elevate the NO_2^- , the specific mechanisms of long-range
34 transport resulting in NO_2^- accumulation in TSP rather than $\text{PM}_{2.5}$ remain unknown and need to be
35 investigated. Our results reveal an overlooked source of atmospheric NO_2^- , i.e., soil NO_2^- , and
36 highlight in remote regions such as Tibet where other sources are limited, wind-blown soil may
37 serve as an important source of atmospheric NO_2^- . Once lofted into the atmosphere, NO_2^- may
38 readily participate in atmospheric reactive nitrogen cycling through gas-particle partitioning or
39 photolysis, leading to the production of HONO, OH and NO and thereby influencing oxidation
40 chemistry.



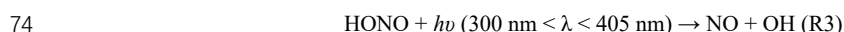
41 1 Introduction

42 The Himalayas-Tibetan Plateau (HTP) represents one of the most important
43 geomorphologically cryospheric regions, boasting high abundance of alpine glaciers and extensive
44 areas of high-altitude snow cover (Kehrwald et al., 2008; Yao et al., 2012; Kang et al., 2019). The
45 melting of snow cover and glacier ice holds immense importance as crucial freshwater resources for
46 over 1.4 billion people in Asia, earning HTP the title as the “Water Tower of Asia” (Immerzeel et
47 al., 2010). However, the enhanced warming rate in the Tibetan Plateau (TP, $\sim 0.30^{\circ}\text{C}$ per decade)
48 has resulted in a rapid and alarming increase in glacier melting over the past decade, significantly
49 affecting the climate, hydrological cycles, and ecosystems at local and global scales (Immerzeel et
50 al., 2010; Xu et al., 2009; Lau et al., 2010). Persistently increased aerosol loadings and greenhouse
51 gas in TP region account for the increased warming rate (Kang et al., 2019; Lau et al., 2010; Lüthi
52 et al., 2015). More importantly, once deposited on the surface of snow cover and glacier ice, the
53 aerosol, especially these light-absorbing components (i.e., black carbon, dust) contribute
54 significantly to the rapid glacial retreat (Xu et al., 2009; Zhao et al., 2020).

55 Atmospheric oxidation capacity (AOC) regulates the formation of secondary aerosol and the
56 removal of trace gases including CH_4 (Wang et al., 2023; Ye et al., 2023; Ye et al., 2016; Andersen
57 et al., 2023), therefore acting as a critical link between atmospheric pollution and cryospheric
58 changes. Previous study have suggested strong solar radiation, high O_3 and relatively high water
59 vapor dominate the relatively strong AOC over the TP (Lin et al., 2008). Recent field campaign
60 further highlighted the rapid reactive nitrogen cycling, with N(III) species (i.e., HONO) as the
61 intermediate, also plays an important role in maintaining the strong AOC in TP (Wang et al., 2023).
62 For example, *Wang et al.* reported high-than-expected HONO ($\sim 30 \pm 13$ pptv) in the Namco station,
63 a typical background site in the middle of TP, with soil emission and particulate nitrate photolysis
64 as the dominant sources (Wang et al., 2023). Interestingly, relatively high levels of nitrite (NO_2^-) in
65 total suspended particulate (TSP) have also been reported from remote sites of TP, i.e., in a forest
66 site in the Southeast Tibet ($\sim 140 \text{ ng m}^{-3}$) and at the Qomolangma monitoring station (QOMS, ~ 60
67 ng m^{-3}) (Bhattarai et al., 2019; Bhattarai et al., 2023). Such high levels of particulate NO_2^- may also
68 contribute to the strong AOC in TP, either via directly photolysis to produce NO_x (Jacobi et al., 2014)



69 or indirectly serve as an important source of HONO through gas-particle partitioning (R1). However,
70 the sources and formation mechanisms for the relatively high level of atmospheric NO_2^- observed
71 in the TP remain unclear.



75 The stable nitrogen and oxygen isotopic compositions ($\delta^{15}\text{N}$, $\delta^{18}\text{O}$, and $\Delta^{17}\text{O}$; where $\delta =$
76 $(R_{\text{sample}}/R_{\text{reference}} - 1) \times 1000\text{‰}$ and with R denoting the $^{15}\text{N}/^{14}\text{N}$, $^{18}\text{O}/^{16}\text{O}$, and $^{17}\text{O}/^{16}\text{O}$ ratios; $\Delta^{17}\text{O} =$
77 $\delta^{17}\text{O} - 0.52 \times \delta^{18}\text{O}$) may provide diagnostic information regarding the sources and formation
78 pathways of atmospheric nitrite. Similar isotopic approaches have been widely used to explore the
79 nitrate (NO_3^-) sources (Morin et al., 2008; Zong et al., 2020; Geng et al., 2014; Fang et al., 2011;
80 Hastings et al., 2003; Zhang et al., 2022; Zhang et al., 2021b; Liu et al., 2018; Felix and Elliott,
81 2014; Miller et al., 2018). Considering that atmospheric NO_2^- may share similar sources and
82 formation pathways with NO_3^- , the specific NO_2^- formation pathways are expected to be
83 characterized by distinct oxygen or nitrogen isotopic endmembers, despite reports on the
84 atmospheric NO_2^- isotopic compositions are rare. For instance, NO_2^- produced from the photolysis
85 of particulate NO_3^- may possess very negative $\delta^{15}\text{N}$ values compared to NO_3^- , analogous to the
86 pronounced nitrogen isotope fractionation effects associated with snow nitrate photolysis (Erbland
87 et al., 2013), while the $\Delta^{17}\text{O}$ of NO_2^- is expected to closely resemble that of NO_3^- as the oxygen atom
88 in NO_2^- is imparted from NO_3^- , unless significant oxygen atom exchange between NO_2^- and aerosol
89 water occurs. The $\Delta^{17}\text{O}$ of NO_2^- (and HONO) from primary emission sources is expected to be
90 negligible, while that generated from heterogeneous reactions of NO_2 on the aerosol surface would
91 be characterized by positive $\Delta^{17}\text{O}$ values depending on the degree of NO_2^- and aerosol water oxygen
92 isotope exchange. These unique isotopic fingerprints may be utilized in distinguishing the sources
93 and formation pathways of atmospheric NO_2^- .

94 To gain insight into the sources and/or formation mechanisms of atmospheric NO_2^- in the pristine
95 environment of TP, we collected the TSP and fine particulate matter ($\text{PM}_{2.5}$) synchronously at the
96 Base Camp, the north slope of the Mt. Qomolangma during the campaign of “Earth Summit



97 Mission-2022” scientific expedition from April 24th to May 6th, 2022, with additional surface soil
98 samples collected in May, 2023. The NO₂⁻ concentration and multi-isotopic signatures ($\delta^{15}\text{N}$, $\delta^{18}\text{O}$,
99 and $\Delta^{17}\text{O}$) in aerosol and surface soil were then determined in order to evaluate the potential sources
100 of atmospheric NO₂⁻. Additionally, the potential environmental implication of atmospheric nitrite
101 was explored in the term of atmospheric oxidation capacity at this pristine environment.

102 **2 Material and Method**

103 **2.1 Site description**

104 The Base Camp is located in the middle of the Rongbuk valley (86.85 °E, 28.14 °N), situated
105 ~5200 m above sea level (m a.s.l) on the north slope of the Mt. Qomolangma (Zou et al., 2008; Zhu
106 et al., 2006). The surrounding surface consists of loosed soil, gravel, broken rocks of various sizes,
107 with sparse vegetation due to the semi-arid status (Ming et al., 2007; Zou et al., 2008). Rongbuk
108 valley is characterized by a depth of ~1000 m and a floor width of ~1000 m, with elevations of the
109 surrounding mountains exceeding 6000 m on both sides (Zou et al., 2008). Attributed to the unique
110 topography, the local air circulation is dominated by mountain and valley breezes. The predominant
111 wind regime is the katabatic flow of southerly and southeasterly, which is typically persists
112 from noon to midnight (Zhu et al., 2006; Zou et al., 2008; Zhou et al., 2011). The nearest
113 accessible area for residents and visitors is at least 2 km north of the Base Camp. During the
114 campaign, electricity and natural gases were routinely used for cooking and hot water production.
115 There were intermittent vehicle exhaust emissions around the station during daytime for the daily
116 necessities supporting, i.e., water and food. To minimize the influence of local anthropogenic
117 activities on sampling, the instruments were set in the southwest (upwind direction) and
118 approximately 100 m away from the living space of the Base Camp. The anthropogenic influence
119 on the sampling is expected to be minimal.

120 **2.2 Field campaign and sample collection**

121 From April 26th to May 6th, 2022, TSP samples were collected simultaneously with NO₂ using
122 a homemade denuder-filter system (Zhou et al., 2022). A polytetrafluoroethylene (PTFE) sleeve is
123 used to assemble the homemade denuder with the filter pack, flowmeter, and pump. The filter pack
124 was placed in the front of the denuder. All connections between the various parts of the sampling



125 apparatus are made using 3/8" Teflon tubing. A detailed description of the sampling apparatus can
126 be found in our previous report (Zhou et al., 2022). Whatman quartz filter (circles, diam. 47mm,
127 pre-heated at 400 °C for 3 h before use) was placed into the filter pack to collect TSP sample. In the
128 present study, the collected bulk aerosol can be regarded as total suspended particulate since no
129 size-selective inlet was employed. The flow rate was controlled at 30 L/min using a flowmeter,
130 corresponding to a filter face velocity of approximately 0.288 m s⁻¹. Previous reports have indicated
131 that the face velocity has negligible effect on the concentrations of sulphate, nitrate, and ammonium
132 when using quartz filter for sampling (Keck and Wittmaack, 2005). To minimize the potential
133 influence of the loose ground surface on the TSP collection, a mountain tent was used to separate
134 the pump (out the tent) with the denuder-filter system (in the other side of tent), and the inlet Teflon
135 tube was stretched out of the tent for ~1.5 m height straight.

136 From April 24th to May 6th, 2022, PM_{2.5} were sampled using a high-volume aerosol sampler
137 (TH-1000F; Wuhan Tianhong Instruments Co. Ltd., China) equipped with PM_{2.5} inlet and Whatman
138 quartz-fiber filters (sheets, 203 mm × 254 mm) at a flow rate of 1.5 m³/min. All the quartz filters
139 were pre-heated at 400 °C for 3 h before use. In general, TSP and PM_{2.5} samples were collected with
140 diurnal resolution during this campaign, with daytime samples from approximately 09:00–20:00
141 and nighttime samples from 21:00–08:00 (local time), respectively. From May 2nd to May 4th, we
142 collected the daytime TSP and PM_{2.5} samples in the morning (09:00–14:00) and afternoon (14:00–
143 20:00), respectively. A total of 24 TSP samples (including 2 blanks) and 29 PM_{2.5} (including 3 blanks)
144 were collected during this campaign. After each sampling period, quartz filters were removed from
145 the filter pack/PM_{2.5} inlet and then stored in frozen until analysis. During the campaign, snowfall
146 events occurred on the night of April 29th and during the daytime of April 30th.

147 Surface soil samples (n = 9) were collected on both sides of the Rongbuk valley (i.e., east and
148 west) as well as on the south side in May, 2023. Soil samples were transported into laboratory using
149 a cold chain. Upon arrival at our laboratory, soil samples were passed through a 60-mesh screen
150 (~0.25 mm).

151 **2.3 Ionic concentration analysis**

152 Water-soluble inorganic ions (WSIs, including Na⁺, NH₄⁺, K⁺, Mg²⁺, Ca²⁺, Cl⁻, NO₂⁻, NO₃⁻, and



153 SO_4^{2-}) in TSP (entire filter, $\sim 16.6 \text{ cm}^2$) and $\text{PM}_{2.5}$ (1/32 section, $\sim 13.0 \text{ cm}^2$) were extracted using 20
154 mL Milli-Q water ($18.2 \Omega \text{ cm}$) in an ultrasonic bath at room temperature for 30 min. After filtration
155 through a $0.22 \mu\text{m}$ pore size syringe filter which was pre-cleaned with purewater, the extract was
156 subjected to inorganic species analysis using ion chromatography (Dionex Aquion) (Zhang et al.,
157 2020). Blank samples were pretreated and analyzed along with the aerosol samples. The ionic
158 concentrations reported here were blank corrected (volatile components (i.e., NH_4^+ , NO_2^- , NO_3^-)
159 and K^+ in blank are always below the detection limits; SO_4^{2-} , Mg^{2+} and Ca^{2+} in blank are generally
160 lower than samples by at least five times (the lower end); Na^+ in the blank is comparable to samples
161 and was not presented in this study). The analysis uncertainties were typically within 5% standard
162 deviation based on replicate measurements of standards ($n = 10$). The detection limit for the
163 inorganic species is general less than 6 ng mL^{-1} .

164 For soil ionic concentration analysis, 4.0 g sieved soil was extracted using 20 mL Milli-Q
165 purewater, shaken for 30min at room temperature. After centrifugation, the extract was passed
166 through 0.45 mm filters before ions analysis. The concentration determination of water-soluble
167 inorganic ions in soil extract was similar to that of aerosol samples. Detailed procedure for the
168 extraction of soil inorganic ions, especially the NO_2^- has been described in previous report (Homyak
169 et al., 2015).

170 **2.4 Isotopic analysis**

171 After concentration measurements, isotopic analyses ($\delta^{15}\text{N}$, $\delta^{18}\text{O}$, and $\Delta^{17}\text{O}$) of NO_2^- in TSP
172 were conducted using the azide method (Casciotti et al., 2007). The azide method can reduce nitrite
173 ion in solution into N_2O in a single step, while nitrate ion remains unchanged, ensuring no
174 interference on the nitrite isotopic analysis. The azide reagent is prepared by mixing 2 M sodium
175 azide with 40% acetic acid at a 1:1 ratio by volume in our laboratory. NO_2^- standards and samples
176 were pretreated under identical conditions concerning the total volume, nitrite amount, water isotope,
177 and matrix. The $\delta^{15}\text{N}$, $\delta^{18}\text{O}$, and $\delta^{17}\text{O}$ of N_2O reduced from NO_2^- in the TSP samples and standards
178 were determined using a Finnigan[®] MAT253 plus isotope ratio mass spectrometer (IRMS) equipped
179 with a GasBench II and preconcentration system. The data calibration followed the procedures
180 described in *Albertin et al.*, 2021, using three international KNO_2 salt standards (RSIL-N10219,
181 RSIL-N7373, and RSIL-N23 with respective $\delta^{15}\text{N}$ and $\delta^{18}\text{O}$ values of 2.8/88.5 ‰, -79.6/4.2 ‰, and



182 3.7/11.4‰). The $\Delta^{17}\text{O}$ of RSIL-N7373 and RSIL-N23 are suggested to be zero (Albertin et al., 2021),
183 while the $\Delta^{17}\text{O}$ of RSIL-N10219 is determined to be $(-9.3 \pm 0.2) \text{‰}$ in our laboratory. The ^{17}O -
184 excess in samples is then calculated as $\Delta^{17}\text{O} = \delta^{18}\text{O} - 0.52 \times \delta^{17}\text{O}$. The standard deviations for $\delta^{15}\text{N}$,
185 $\delta^{18}\text{O}$, $\Delta^{17}\text{O}$ of reference materials ($n = 10$) were determined to be less than 0.1‰, 0.6‰, and 0.4‰,
186 respectively.

187 For soil NO_2^- isotopic analysis, ~30.0g sieved soil was extracted using 150 mL Milli-Q
188 purewater. The soil extract was then pre-concentrated into 10 mL using ion-exchange resin before
189 isotopic analysis. The pre-concentration approach was widely used for nitrate isotopic analysis in
190 snow and ice samples, and the detailed procedures and the performance have been provided in
191 *Erbland et al.*, 2013. The concentrated soil extracts (50 nmol NO_2^-) was then subjected to soil NO_2^-
192 isotopic analysis by converting into N_2O via the azide method. The remaining procedures of soil
193 NO_3^- and NO_2^- analysis were same as those for TSP samples, as aforementioned.

194 **2.5 Complementary analyses of air mass backward trajectory**

195 To evaluate the possible impact of biomass burning emissions or other pollution sources from
196 South Asia, the Hybrid Single-Particle Lagrangian Integrated Trajectory (HYSPLIT) model
197 (performed using TrajStat plugin of the MeteoInfo software) and archived Global Data Assimilation
198 System (GDAS) of meteorological data were used to model the air mass back trajectories (Wang,
199 2014). In this study, six-day air mass backward trajectories with arriving height of 1000 m above
200 ground level were simulated to identify the most likely pathway and potential source regions of the
201 air masses at the Base Camp. Moreover, the Fire Information and Resource Management System
202 (FIRMS) developed by Moderate Resolution Imaging Spectrometer (MODIS)
203 (<https://worldview.earthdata.nasa.gov>) was employed to identify the distribution of active fire spots
204 during the sampling period.

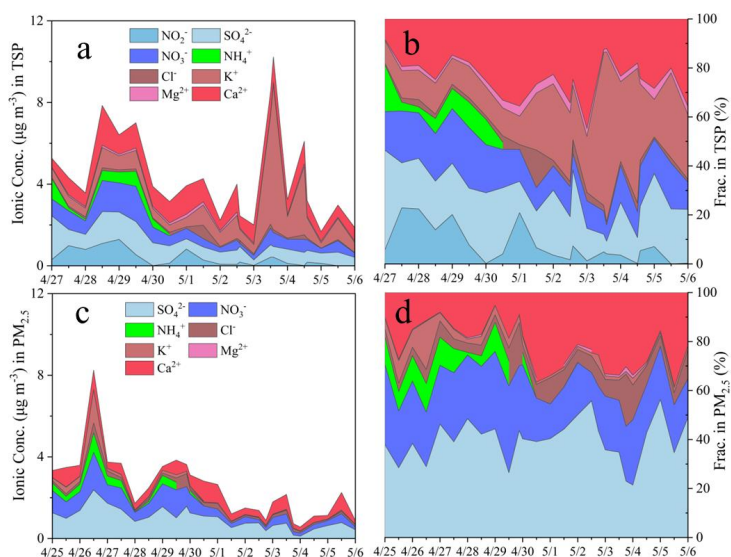
205 **3 Results**

206 **3.1 Mass concentrations of water-soluble inorganic ions in TSP and $\text{PM}_{2.5}$**

207 Figure 1 displays the chemical compositions of water-soluble inorganic ions, their
208 corresponding time series and fractional contributions in TSP and $\text{PM}_{2.5}$. Throughout the campaign,
209 substantial variations of total WSIs in $\text{PM}_{2.5}$ were observed, with a general decline after May 1st.



210 This decline was predominately driven by significant reductions in secondary inorganic species, i.e.,
 211 SO_4^{2-} , NO_3^- and NH_4^+ , with the magnitude by more than 60%. In particular, NH_4^+ in $\text{PM}_{2.5}$ declined
 212 from $(322 \pm 243) \text{ ng m}^{-3}$ before May 1st to negligible after May 1st. Therefore, the fractional
 213 contribution of secondary inorganic species in $\text{PM}_{2.5}$ also decreased. Similarly, K^+ in $\text{PM}_{2.5}$, a good
 214 tracer of biomass burning (Ma et al., 2003), also declined significantly after May 1st. The elevated
 215 concentrations of WSIs before May 1st are comparable to previous reports at remote sites over the
 216 TP (i.e., Namco, QOMS) in the spring (Wang et al., 2020; Lin et al., 2020; Lin et al., 2021; Decesari
 217 et al., 2010). In comparison, concentrations of Ca^{2+} and Mg^{2+} , tracers of wind-blown dust (Wang et
 218 al., 2002), decreased by less than 20% after May 1st, with a smaller degree than secondary species.
 219 In general, SO_4^{2-} , NO_3^- , and Ca^{2+} are the most abundant species in $\text{PM}_{2.5}$, accounting for the majority
 220 of the mass of total WSIs. In addition, despite the average daytime concentrations of the inorganic
 221 species were generally higher than those at night, no clear diurnal variation was observed in this
 222 study.



223

224 **Figure 1.** The chemical compositions and time series of mass concentrations of water-soluble
 225 inorganic species (NO_2^- , SO_4^{2-} , NO_3^- , Ca^{2+} , etc.), as well as the corresponding mass fractions in
 226 respective TSP (a, b) and $\text{PM}_{2.5}$ (c, d) samples collected at Base Camp of Mt. Qomolangma in spring
 227 2022.



228 The variations of WSIs in TSP generally followed a similar pattern to that in PM_{2.5} (Figure 1).
229 For example, after May 1st, the secondary inorganic species in TSP declined considerably by over
230 50% (i.e., NH₄⁺ in TSP declined by more than tenfold), while Ca²⁺ in TSP decreased with a smaller
231 degree (< 15%) and Mg²⁺ remained stable. In contrast, TSP K⁺ (from both crustal and biomass
232 burning sources)(Hsu et al., 2009) drastically surged on May 3rd and May 4th. Note that other species
233 in TSP, i.e., SO₄²⁻ and NO₃⁻ also increased to some extent on May 3rd.

234 The most distinct feature of chemical compositions in TSP was the elevated level and
235 significant variation of NO₂⁻, ranging from 0.2 ng m⁻³ to 1291 ng m⁻³ in air and with an average of
236 375 ± 386 ng m⁻³. In comparison, NO₂⁻ consistently remained below the detection limit in PM_{2.5}
237 samples. Note during the laboratory measurements of ionic concentrations, TSP and PM_{2.5} filters
238 were extracted with purewater and it was the extraction analyzed by ion chromatography. To ensure
239 fair comparisons, similar areas of filters were extracted with same volume of purewater, so that the
240 extractions from the PM_{2.5} filters should be more concentrated in atmospheric particulate species
241 compared to that from the TSP filters, since PM_{2.5} samples were collected at a much faster sampling
242 speed (1.5 m³ min⁻¹ vs. 30 L min⁻¹) over the same sampling duration. Nevertheless, NO₂⁻ was
243 detectable only in the extractions of TSP filter. The determined NO₂⁻ concentrations in TSP in this
244 study (375 ± 386 ng m⁻³) were higher than previous reports conducted in various remote sites, such
245 as at QOMS station (~60 ng m⁻³ for TSP) (Bhattarai et al., 2023), at a forest site in the Southeast
246 Tibet Plateau (~140 ng m⁻³ for TSP) (Bhattarai et al., 2019), in the middle hills of the central
247 Himalayas (~210 ng m⁻³ for TSP) (Tripathee et al., 2021).

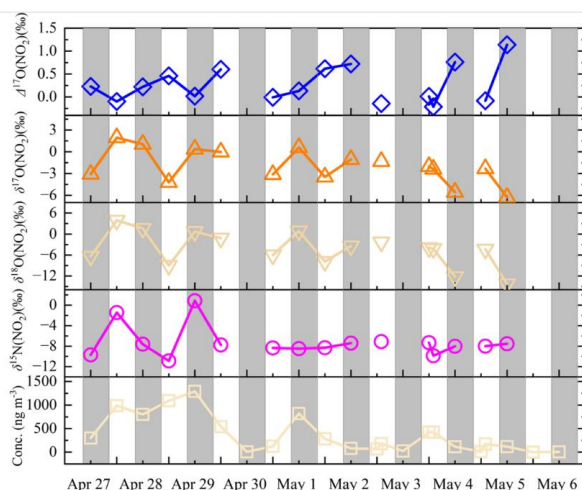
248 In particular, there was a dramatic decrease in TSP NO₂⁻ after May 1st, from a mean of (625 ±
249 457) ng m⁻³ to (147 ± 145) ng m⁻³, in line with the declines in other secondary inorganic species.
250 Over the course of the campaign, NO₂⁻ comprised approximately 8% of the total WSIs mass in TSP,
251 while its contribution reached maximum of ~20% on April 27th and April 28th, being one of the most
252 abundant components on the two days. In addition, there was a strong correlation between NO₂⁻ and
253 NO₃⁻ throughout the campaign ($r = 0.75$, $p < 0.05$. Figure S1). Meanwhile, the mean mass ratio of
254 NO₂⁻ to NO₃⁻ was ~50% throughout the campaign, but on several days (i.e., April 27th) NO₂⁻
255 concentrations significantly exceeded that of NO₃⁻. Previous study also reported comparable NO₂⁻
256 and NO₃⁻ concentrations in at a forest site in the Southeast Tibet (summer:100 ng m⁻³ vs. 110 ng m⁻³



257 ³; winter: 180 ng m⁻³ vs. 270 ng m⁻³) (Bhattarai et al., 2019).

258 3.2 Isotopic signatures of nitrite in TSP

259 Figure 2 presents the times series of $\delta^{15}\text{N}$, $\delta^{18}\text{O}$ and $\Delta^{17}\text{O}$ of NO_2^- in TSP, along with the NO_2^-
 260 concentrations. Similar to the variation trend of NO_2^- concentrations, NO_2^- isotopes varied in a wider
 261 range before May 1st, but became more stable afterward. For example, $\delta^{15}\text{N}(\text{NO}_2^-)$ ranged from -
 262 10.9 ‰ to 0.8 ‰, with a relatively large standard deviation before May 1st compared to that after
 263 May 1st (-6.4 ± 4.3 ‰ vs. -8.0 ± 0.7 ‰). The large variability in $\delta^{15}\text{N}(\text{NO}_2^-)$ before May 1st is
 264 predominately attributed to the two high values observed in daytime of April 27th and night of April
 265 28th. Note that the two high $\delta^{15}\text{N}(\text{NO}_2^-)$ samples were also associated with relatively high NO_2^- mass
 266 concentrations but relatively low $\delta^{18}\text{O}(\text{NO}_2^-)$. In contrast to the declining trend of NO_2^-
 267 concentrations, the mean $\delta^{15}\text{N}(\text{NO}_2^-)$ values were comparable before and after May 1st. Relatively
 268 large variability was observed in TSP $\delta^{18}\text{O}(\text{NO}_2^-)$, ranging from -9.0‰ to 3.9‰ and with an average
 269 of (-3.4 ± 3.8) ‰. TSP $\Delta^{17}\text{O}(\text{NO}_2^-)$ varied within a narrow range from -0.2‰ to +0.7‰ and with a
 270 mean of (0.2 ± 0.3) ‰. During the campaign, no significant correlations were observed between
 271 $\delta^{15}\text{N}(\text{NO}_2^-)$ and the NO_2^- concentrations; while $\delta^{18}\text{O}(\text{NO}_2^-)$ appeared to be moderately correlated
 272 with the NO_2^- concentrations (Figure S2).



273

274 **Figure 2.** Time series of multiple isotopic signatures of NO_2^- ($\delta^{15}\text{N}$, $\delta^{18}\text{O}$, $\delta^{17}\text{O}$ and $\Delta^{17}\text{O}$) as
 275 well as corresponding concentrations in TSP samples during “Earth Summit Mission” scientific



276 expedition in spring 2022. The gray shaded area denotes local nighttime.

277 **Table 1.** The measured nitrite (and nitrate) concentration and isotopic signatures in surface soil
278 collected in the Rongbuk Valley.

Soil sample ID	NO ₂ ⁻				NO ₃ ⁻			
	Conc.(ng g ⁻¹)	δ ¹⁵ N	δ ¹⁸ O	Δ ¹⁷ O	Conc.(ng g ⁻¹)	δ ¹⁵ N	δ ¹⁸ O	Δ ¹⁷ O
East-1	67.7	-12.0	6.1	3.3	1127.3	-2.7	23.4	5.7
East-2	76.1	-11.9	2.7	2.2	1098.3	1.3	18.1	3.8
East-3	82.3	-14.6	6.0	3.3	2978.1	-0.3	25.4	6.6
West-1	88.6	-10.1	8.9	1.6	3176.5	-3.1	44.5	13.4
West-2	106.2	-7.0	11.4	1.5	2880.3	-1.2	22.1	6.2
West-3	179.3	-5.2	12.7	1.4	7686.9	-3.4	35.8	11.4
South-1	53.8	-13.2	12.7	6.7	3651.2	-0.6	32.6	10.1
South-2	42.3	-9.0	18.1	7.3	1683.2	-2.2	49.3	14.8
South-3	48.9	-10.3	12.4	6.7	385.5	-2.4	45.7	14.9

279

280 3.3 Surface soil nitrite concentration and isotopic signature

281 The concentration of soil NO₂⁻ (and NO₃⁻) as well as the corresponding isotopic signatures are
282 displayed in Table 1. High soil NO₂⁻ and NO₃⁻ concentrations were observed on the west and east
283 slopes of the Rongbuk valley relative to that on the south side of the sampling site. The mean surface
284 soil NO₂⁻ and NO₃⁻ in the Rongbuk Valley were 82.8 and 2740.8 ng g⁻¹, respectively. The soil NO₃⁻
285 concentrations are significantly higher than the NO₂⁻ by a factor of 8 ~ 40. In general, the measured
286 soil NO₃⁻ concentrations at the Rongbuk valley are significantly lower than other remote regions of
287 TP (i.e., Naqu, Bangda), while soil NO₂⁻ concentrations are comparable to these reports (Wang et
288 al., 2019). Soil δ¹⁵N(NO₂⁻) values ranged from -13.2‰ to -5.2‰ (on average -10.4‰), which are
289 comparable to TSP δ¹⁵N(NO₂⁻). In comparison, we observed positive soil δ¹⁸O(NO₂⁻) and
290 Δ¹⁷O(NO₂⁻), ranging from 2.7‰ to 18.1‰ (on average 10.5‰) and 1.4‰ to 7.3‰ (on average
291 3.8‰), respectively, in contrast to the negative δ¹⁸O(NO₂⁻) and near-zero Δ¹⁷O(NO₂⁻) observed in
292 TSP samples. The determined soil δ¹⁸O(NO₂⁻) is comparable to that in laboratory incubated soil
293 (11.8‰ ~ 12.5‰) (Lewicka-Szczebak et al., 2021).

294 4 Discussion

295 The significant contrast in NO₂⁻ concentrations between TSP and PM_{2.5} samples, as shown in
296 Figure 1, suggests that at the sampling site atmospheric particle NO₂⁻ overwhelmingly exists in
297 coarse particles. This observation is consistent with previous studies across the TP, which also



298 reported the absence of NO_2^- in fine mode particles ($\text{PM}_{2.5}$ and $\text{PM}_{1.0}$) using either online real-time
 299 instrument or offline filter sampling (Decesari et al., 2010; Xu et al., 2020; Xu et al., 2023; Zhao et
 300 al., 2020), while relatively high levels of TSP NO_2^- have been reported (Bhattarai et al., 2019;
 301 Bhattarai et al., 2023; Tripathee et al., 2017). In general, the chemical sources of particle NO_2^- in
 302 the atmosphere encompass the uptake of HONO, particulate nitrate photolysis, and the NO_2 -related
 303 reactions (i.e., photo-enhanced uptake of NO_2 on mineral dust, heterogeneous reaction of NO_2 on
 304 the surface of aerosol) (Nie et al., 2012; Vandenboer et al., 2014a; Chen et al., 2019; Shang et al.,
 305 2021), as summarized in Table 2. In addition to these in-situ atmospheric processes, growing
 306 evidence has revealed that long-range transport of atmospheric pollutants from South Asia also
 307 contributes considerably to aerosol loadings in TP in the spring (Kang et al., 2019; Bhattarai et al.,
 308 2023; Zhao et al., 2020), which may also bring nitrite along with other pollutants. Moreover, the
 309 lifting of surface dust can also contribute to the atmospheric coarse particles and significantly
 310 influence the chemical composition of TSP (Zhang et al., 2021a; Pokharel et al., 2019), and therefore
 311 soil nitrite could also be a potential source for TSP NO_2^- . In the following discussion, we examine
 312 the potential importance of the abovementioned processes to the observed high NO_2^- content in
 313 coarse particle and discern the most likely ones.

314 **Table 2.** Particulate nitrite concentration and formation pathways/sources compiled in the literature.

315

Site	Period	NO_2^- Conc. (mean, ng m^{-3})	Formation pathways/Sources	Reference
QOMS station	April 2017	60	Biomass burning emission transported from South Asia	Bhattarai et al., 2023
Bakersfield, California	May–July 2010	150	HONO uptake on lofted alkaline soil particles.	VandenBoer et al., 2014
Jinan, China	November 2013 – January 2014	2080	Heterogeneous reactions of NO_2	Wang et al., 2015
Seoul, Korea	May–July, 2005	1410	Heterogeneous reactions of NO_2	Song et al., 2009
Shanghai, China	June 2020	210	Heterogeneous reactions of NO_2 , reduction of NO_2 by S(IV)	Shang et al., 2021
Mt.Heng, China	April 2009	2500	Surface TiO_2 photocatalysis of NO_2	Nie et al., 2012

316 4.1 The potential effects of atmospheric chemistry on NO_2^- in TSP

317 Increasing evidence supports particulate nitrate photolysis as an important source of
 318 atmospheric HONO especially in pristine atmosphere, with NO_2^- serving as the intermediate in the



319 subsequent gas-particle partition process (Andersen et al., 2023; Ye et al., 2016). In theory,
320 particulate NO_2^- (and HONO) produced from particulate nitrate photolysis might be associated with
321 extremely negative $\delta^{15}\text{N}$ values, due to the significant nitrogen isotopic fractionation effect during
322 nitrate photolysis (Erbland et al., 2013). For example, NO_2^- in water of hypersaline ponds and soil
323 of McMurdo Dry Valleys, Antarctica, produced from the NO_3^- photolysis, were characterized by
324 significantly negative $\delta^{15}\text{N}$ values ($< -80\%$) (Peters et al., 2014). However, $\delta^{15}\text{N}(\text{NO}_2^-)$ in this study
325 was only $\sim 2\%$ lower than the $\delta^{15}\text{N}(\text{NO}_3^-)$ in TSP samples collected during this campaign (on
326 average $(-5.3 \pm 3.3)\%$, Text S1) and that at the QOMS stations (annual average of $(-5.1 \pm 2.3)\%$)
327 (Wang et al., 2020). The similarity in $\delta^{15}\text{N}$ isotopes between NO_2^- and NO_3^- suggests particulate
328 nitrate photolysis is unlikely to be the primary source of the TSP NO_2^- .

329 In addition to nitrate photolysis, the absorption of HONO on alkaline aerosols (i.e., lofted dust
330 and road salt particles) can also result in accumulation of NO_2^- into the particle phase (Vandenboer
331 et al., 2014a; Chen et al., 2019). For example, *VandenBoer et al.* observed a synchronous
332 enhancement of fine particle NO_2^- (as high as 730 ng m^{-3}) alongside the buildup of HONO (up to
333 1.37 ppbv) after sunset in an agricultural site (Vandenboer et al., 2014a). However, the levels of
334 HONO in terrestrial background environments, typically on the order of dozens of pptv (Ye et al.,
335 2023), are obviously too low to support the observed unexpectedly high levels of particulate NO_2^-
336 (up to 1300 ng m^{-3}). What is more, previous studies indicted the HONO uptake predominately
337 occurs on fine particles (Wang et al., 2015; Chen et al., 2019; Shang et al., 2021), while our
338 observations indicated NO_2^- only exists in coarse particles.

339 The uptake of NO_2 on mineral dust has also been identified as a significant route for the
340 formation of particulate NO_2^- or gas-phase HONO (Nie et al., 2012; Ndour et al., 2008). For example,
341 *Nie et al.* found a significantly enhanced NO_2^- in coarse particle during daytime in a dust storm
342 event in Mt. Heng (up to $4.5 \mu\text{g m}^{-3}$) (Nie et al., 2012). The proposed mechanism is initiated by
343 photocatalysis of NO_2 to NO_2^- via photo-produced electrons on surface of dust. Nevertheless, given
344 the relatively small NO_2 uptake coefficients (generally lower than 10^{-6}) (Yu et al., 2021; Bao et al.,
345 2022) and the low concentration of NO_2 in pristine environment of TP (e.g., $\sim 140 \text{ pptv}$ at Namco
346 (Wang et al., 2023), and would be even lower at Mt. Qomolangma), such high levels of particulate
347 NO_2^- are beyond the capacity of NO_2 heterogeneous reactions.



348 Other than the above-mentioned rationales, the atmospheric physicochemical processes leading
349 to NO_2^- production would influence both fine and coarse-mode particles, and some of the processes
350 (e.g., HONO uptake) preferentially interact with fine-mode particles. However, the observation
351 indicated NO_2^- only exists in TSP but not $\text{PM}_{2.5}$, suggesting atmospheric physicochemical processes
352 are unlikely to account for the elevated levels of NO_2^- in TSP.

353 **4.2 Potential effect of biomass burning emissions in South Asia via long-range transport**

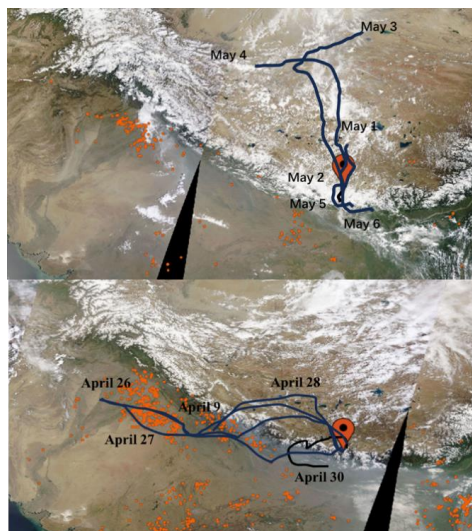
354 There is a growing body of compelling evidence indicating that the elevated aerosol loadings
355 and chemical species in TP during spring, i.e., black carbon (Cong et al., 2015; Kang et al., 2019)
356 and soluble components (Dasari et al., 2023; Bhattarai et al., 2023; Lin et al., 2021; Wang et al.,
357 2020; Zhao et al., 2020) are significantly linked to biomass burning emissions from South Asia,
358 which can penetrate into TP via long range transport. Recently, *Bhattarai et al.* observed
359 synchronously elevated water-soluble nitrogen compounds (i.e., NO_2^- , NH_4^+ , NO_3^-), levoglucosan
360 (a molecule marker for biomass burning) and bulk $\delta^{15}\text{N}$ signatures in TSP at QOMS station, once
361 upon the arrival of biomass burning plumes from South Asia (Bhattarai et al., 2023). Specially, they
362 found that TSP NO_2^- averaged 60 ng m^{-3} during the spring biomass burning influenced episodes,
363 while NO_2^- was always below the detection limit in other seasons (Bhattarai et al., 2023). This
364 suggests that biomass burning events in South Asia and the subsequent long-range transport could
365 contribute to accumulation of NO_2^- in aerosols in Tibet.

366 To evaluate the potential effects of South Asian pollutants on our observations, we further
367 analyzed the air masses origins during the sampling campaign. As shown in Figure 3, during the
368 first-half of the campaign, i.e., before May 1st, air masses mainly originated from or pass through
369 northern India and Nepal with intensive human activities and numerous fire hotspots, while from
370 May 1st to May 6th the air masses originated from the inside of the TP with rare open fires.
371 Accompanied by this shift in air mass origins, the concentrations of NH_4^+ , NO_3^- and SO_4^{2-} in both
372 $\text{PM}_{2.5}$ and TSP, as well as the TSP NO_2^- apparently decreased, suggesting that these species in $\text{PM}_{2.5}$
373 and TSP should be controlled or closely related to biomass burning events in South Asia and the
374 subsequent long-range transport of biomass burning pollutants.

375 However, although biomass burning activities indeed emit various nitrogen compounds into



376 atmosphere, particle nitrite has not yet been detected in biomass burning plumes (Lindaas et al.,
377 2021; Juncosa Calahorrano et al., 2021; Li et al., 2003). It is also difficult for the long-range
378 transport to explain why NO_2^- is predominately present in coarse particles, as fine mode particle is
379 typically easier to be transported in principle. Moreover, transport of pollutants from South Asia is
380 also difficult to explain the presence of NO_2^- in TSP after May 1st when air mass mainly originated
381 from clean regions with minor anthropogenic activities, i.e., the Central Tibet.



382

383 **Figure 3.** The modelled six-day air-mass back trajectories during “Earth Summit Mission” scientific
384 expedition in spring 2022 (a: from May 1st to May 6th; b: April 26th to April 30th). The active fire
385 spots captured by MODIS (<https://worldview.earthdata.nasa.gov>) are also presented.

386 4.3 The potential effects of lofted dust

387 Previous reports have confirmed that wind-blown mineral dust contributed significantly to
388 coarse mode aerosols in TP (Kang et al., 2016; Zhang et al., 2021a). Notably, surface soil collected
389 from the Rongbuk valley is characterized with elevated NO_2^- concentration (up to 180 ng g^{-1} , Table
390 1). The surface soil NO_2^- is expected to mainly reside in the coarse mode after suspended in the
391 atmosphere (Drakaki et al., 2022), consistent with our observations that particle NO_2^- was
392 predominately confined into coarse particle. The unique environment of Rongbuk valley,
393 characterized by exposed surface soil and strong wind (reaching as high as 9 m/s during this



394 campaign) could facilitate the resuspension of soil components into the atmosphere. Furthermore,
395 small localized tornadoes were frequently observed before May 1st, while the snow events occurred
396 in April 30th would reduce the soil-derived dust emission by increasing the snow coverage and
397 enriching the soil moisture. In fact, concentrations of Ca²⁺ in TSP, which predominately originate
398 from dust emission, also declined to some extent after May 1st.

399 In addition, the similarity in $\delta^{15}\text{N}$ of NO_2^- between TSP and the surface soil ($-7.3 \pm 3.1\%$ vs. -
400 $10.3 \pm 3.0\%$) also supports that locally emitted surface soil may contribute to the observed high
401 levels of TSP NO_2^- . But one should note that the oxygen isotopes of TSP and soil NO_2^- are different.
402 This could be explained by the potential oxygen isotope exchanges between NO_2^- and aerosol liquid
403 water (fractionation effect of $^{18}\epsilon_{\text{eq}} \approx 16\%$ at local temperature, $T = 270\text{K}$, $^{18}\epsilon_{\text{eq}} = -0.12 T + 48.79$
404 (Buchwald and Casciotti, 2013)), which will deplete the initial soil $\delta^{18}\text{O}(\text{NO}_2^-)$ once resuspended
405 into atmosphere. The atmospheric water vapor $\delta^{18}\text{O}$ isotope in TP is determined to be significantly
406 negative (approximately -35% to -15% at a remote site in TP with altitude of $\sim 4200\text{m}$) (Yu et al.,
407 2015). The oxygen isotope exchange between NO_2^- and H_2O would also homogenize and erase
408 original soil $\Delta^{17}\text{O}(\text{NO}_2^-)$ signals, because aerosol liquid water content is at least three orders of
409 magnitude higher than NO_2^- (Text S2) and characterized by negligible $\Delta^{17}\text{O}$ values (Luz and Barkan,
410 2005).

411 We also noted that NO_2^- in surface soil is significantly lower than NO_3^- , by up to 40 times
412 (Table 1), contrasting with the chemical compositions in TSP. Soil samples were sieved to less than
413 60 mesh ($\sim 250 \mu\text{m}$), to be in similar order with the diameters of TSP. We speculate that the size
414 distribution of NO_2^- in surface soil maybe different with NO_3^- if resuspended surface soil was the
415 primary source of TSP.

416 5 Conclusion and Implications

417 Unexpectedly high levels of coarse-particle NO_2^- were observed in the pristine environment at
418 Mt. Qomolangma in the spring, 2022. After examining the potential contributions of various NO_2^-
419 sources with assistance from air mass back-trajectory and isotope analyses, we suggest that soil-
420 derived nitrite likely serves as a baseline source of atmospheric NO_2^- , maintaining the background
421 levels of TSP NO_2^- at this pristine site. Moreover, air masses originating from South Asia would



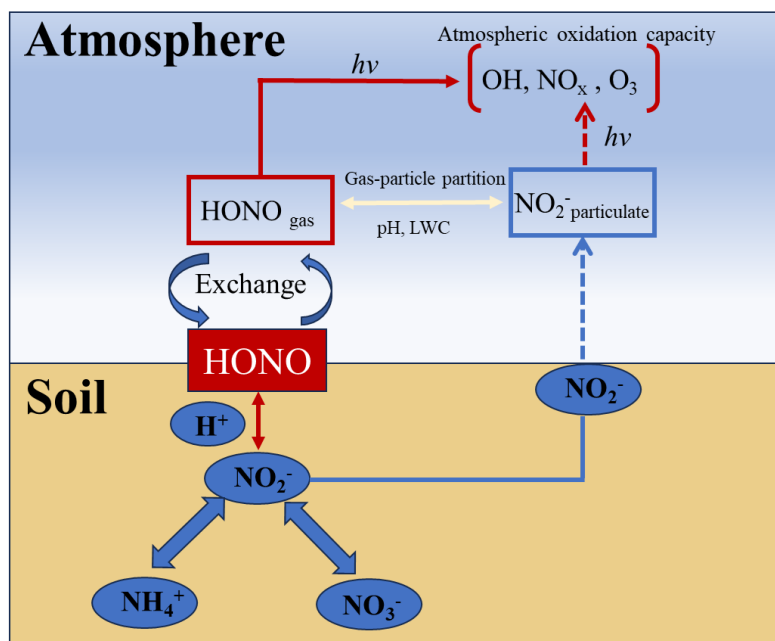
422 also contribute to elevated NO_2^- by bringing biomass burning pollutants, as evidenced by the
423 observed higher TSP NO_2^- before May 1st compared to after that day when air mass origins shifted
424 from South Asia to Central Tibet. The dual-contribution of soil nitrite and long-range transport of
425 pollutants from South Asia to the observed TSP NO_2^- at Mt. Qomolangma is also consistent with
426 the dynamics of NO_2^- isotopic signatures, as the isotopes were more varied during the first half of
427 the campaign when both soil and long-range transport-related nitrite contributed, while in the second
428 half of the campaign the isotopes were more stable when soil nitrite predominated. The effect of
429 long-range transport from South Asia, however, would be likely mainly through the transport of
430 nitrite precursors (e.g., NO_2 , HONO) rather than nitrite itself as there is likely no nitrite in biomass
431 burning plumes (Lindaas et al., 2021; Juncosa Calahorrano et al., 2021; Li et al., 2003). Nevertheless,
432 it is currently unknown why such precursors would selectively interact with coarse mode particles,
433 resulting in nitrite predominately resides in coarse-mode particles. Further investigation should be
434 conducted to clarify the reason.

435 Our results highlight that soil nitrite could also be directly lofted into atmosphere and
436 potentially acting as an important NO_x and HONO source, a previously unrecognized soil–
437 atmosphere interactions which may greatly impact the atmospheric environment (Figure 4). The
438 elevated levels of particle NO_2^- may serve as an important HONO source through the gas-to-particle
439 partition process (Vandenboer et al., 2014a), and the thermodynamic equilibrium between
440 particulate nitrite and HONO ($[\text{pN(III)}]/[\text{HONO}]$ ratio) is primarily governed by the particle acidity
441 and liquid water content (LWC) in theory (Fountoukis and Nenes, 2007; Vandenboer et al., 2014a;
442 Chen et al., 2019). Based on the observed TSP NO_2^- and estimated ratio of $[\text{pN(III)}]/[\text{HONO}]$ (from
443 4.8 to 10.6, Text S2), we can estimate the potential level of atmospheric HONO if the partition ever
444 occurs at this site (Vandenboer et al., 2014b), and result indicates HOHO would be at 8 ~ 15 pptv,
445 on the same order with the observations at a central Tibetan site (~ 30 pptv at Namco (Wang et al.,
446 2023)). Therefore, we suspect that the elevated levels of particle NO_2^- by lofted soil may
447 significantly impact local reactive nitrogen cycling and atmospheric oxidation capacity by acting as
448 source of reactive nitrogen species through photolysis or thermodynamic partitioning processes in
449 the pristine environment of TP (Li, 1994; Song et al., 2009; Wang et al., 2015; Chen et al., 2019).



450

451



452

453 **Figure 4** The conceptual model illustrating the impact of soil nitrite on the atmospheric reactive
454 nitrogen cycling and atmospheric oxidation capacity. Through bi-direction exchange, soil nitrite is
455 an important source of atmospheric HONO. Moreover, the wind-blown soil nitrite can also
456 contribute to atmospheric HONO and NO_x through thermodynamic partition process or photolysis.

457 **Acknowledgments** We thank Dr. Yunhong Zhou for the help in air mass back trajectory analysis.

458 The authors thank all researchers involved in the “Earth Summit Mission-2022” scientific
459 expedition from Peking University, Hefei Institutes of Physical Science Anhui Institute of Optics
460 and Fine Mechanics of Chinese Academy of Sciences, Minzu University of China.

461 **Financial support** L.G. acknowledges financial support from the National Natural Science
462 Foundation of China (Awards: 41822605 and 41871051), and the National Key R&D Program of
463 China (2022YFC3700701). Z. Z acknowledges financial support from the National Natural Science
464 Foundation of China (Awards: 42273001). P. C acknowledges financial support from the National
465 Natural Science Foundation of China (Awards: 42371151).



466 **Conflicts of interest**

467 The authors declare that they have no conflicts of interest.

468 **Author contribution:**

469 L.G designed the research, interpreted the data; L.G and Z.Z prepared the manuscript with
470 contributions from all co-authors; Z.Z., Y.W., C.Y., T.Z., C.Z., Z.J., and P.C., conducted the field
471 sampling and laboratory measurements; L.G, Z.Z., and P.C., acquired funding; L.G., S.K., and C.Y.
472 reviewed and edited the manuscript. All authors have given approval to the final version of the
473 manuscript.

474 **Data availability**

475 The data supporting the findings of this study are available in the archival repository at:
476 <https://doi.org/10.6084/m9.figshare.28188320.v1>

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