

Part 2: Quantitative contributions of cyanobacterial alkaline phosphatases to biogeochemical rates in the subtropical North Atlantic

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Abstract

Microbial enzymes alter marine biogeochemical cycles by catalyzing chemical transformations that bring elements into and out of particulate organic pools. These processes are often studied through enzyme rate-based estimates and nutrient-amendment bioassays, but these approaches are limited in their ability to resolve species-level contributions to enzymatic rates. Molecular methods including proteomics have the potential to link the contributions of specific populations to the overall community enzymatic rate; this is important because organisms will have distinct enzyme characteristics, feedbacks, and responses to perturbations. Integrating molecular methods with rate measurements can be achieved quantitatively through absolute quantitative proteomics. Here, we use the subtropical North Atlantic as a model system to probe how [a combination of traditional bioassays and absolute quantitative proteomics](#) can provide a more comprehensive understanding of nutrient limitation in marine environments. The experimental system is characterized by phosphorus stress and potential metal-phosphorus co-limitation due to dependence of the organic phosphorus scavenging enzyme alkaline phosphatase on metal cofactors. We performed nutrient amendment incubation experiments to investigate how alkaline phosphatase [absolute](#) abundance and activity is affected by trace metal additions [and develop an inventory of cyanobacterial alkaline phosphatases](#). We show that

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38 the two most abundant picocyanobacteria, *Prochlorococcus* and *Synechococcus* are minor contributors
39 to total alkaline phosphatase activity as assessed by a widely used enzyme assay, with
40 *Prochlorococcus* accounting for 3-35% and *Synechococcus* contributing 0.5-5% of alkaline
41 phosphatase activity depending on location and metal cofactor. This was true even when trace metals
42 were added, despite both species having the genetic potential to utilize both the Fe and Zn containing
43 enzymes, PhoX and PhoA respectively. Serendipitously, we also found that the alkaline phosphatases
44 responded to cobalt additions suggesting possible substitution of the metal center by Co in natural
45 populations of *Prochlorococcus* (substitution for Fe in PhoX) and *Synechococcus* (substitution for Zn
46 in PhoA). This integrated approach allows for a nuanced interpretation of how nutrient limitation
47 affects marine biogeochemical cycles and highlights the benefit of building quantitative connections
48 between rate and “-omics” based measurements.

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Introduction

Microbial enzymes alter marine biogeochemical cycles by catalysing chemical transformations and facilitating the movement of elements through planetary reservoirs. On one hand, enzyme contributions from different groups of microbes can be considered collectively, for instance in rate-based or bioassay incubation experiments where the activities of the entire microbial community are aggregated. On the other hand, we anticipate that the enzymes of different organisms will have different activities and responses to perturbations; this means that resolving enzyme provenance could enhance the quantitative connection between microbial activity and biogeochemical rates (e.g. the goals of the fledgling Biogeoscapes program (Saito et al., 2024)). “-Omics” based methods, particularly proteomics which directly resolves protein/enzyme concentrations, can provide a window into the relationships between microbial abundance, enzyme concentration, and biogeochemical rates.

In this work we use quantitative proteomics to constrain the relative contributions of different microbes (*Synechococcus* and *Prochlorococcus*) to biogeochemical rates of alkaline phosphatase activity in the oligotrophic subtropical North Atlantic gyre. In this region, primary production is constrained by availability of dissolved inorganic nitrogen (DIN) and phosphorus (DIP), but inputs of atmospherically derived iron (Fe) from Saharan desert dust create a niche for nitrogen fixation, partially alleviating nitrogen limitation but driving the system to DIP depletion (Martiny et al., 2019; Moore et al., 2013). Lack of DIP then drives a shift towards the acquisition of the abundant yet less bioavailable dissolved organic phosphorus (DOP) by phytoplankton (Lomas et al., 2010; Mather et al., 2008). The DOP pool includes relatively labile phosphomono- and diesters (together ~75 to 85 % of DOP) that derive from ribonucleic acids, adenosine phosphates and phospholipids (Kolowitz et al., 2001; Young and Ingall, 2010). These compounds cannot be directly assimilated but require the phosphate group to be cleaved from the ester moiety first. Cleaving is catalysed by a range of hydrolytic enzymes, such as alkaline phosphatases, which are common in marine microbes, including bacterial as well as eukaryotic phytoplankton (Dyhrman and Ruttenberg, 2006; Luo et al., 2009; Shaked et al., 2006). Reflecting this, alkaline phosphatase activity (APA) is high across the oligotrophic gyres (Browning et al., 2017; Davis et al., 2019; Duhamel et al., 2010; Mahaffey et al., 2014; Wurl et al., 2013).

Alkaline phosphatase activity is commonly regulated by intracellular phosphate levels (Santos-Beneit, 2015) and appears to be closely linked to low ambient DIP concentrations (Mahaffey et al., 2014). However, these enzymes also have a metal dependence, as metal co-factors are involved in the hydrolysis process at the active site. Different alkaline phosphatases exist that, while sharing function, evolved independently and have distinct metal requirements. For example, in *Escherichia coli* (*E. coli*) the alkaline phosphatase PhoA has two Zn^{2+} (zinc) or Co^{2+} (cobalt) ions and one Mg^{2+} (magnesium) ion at each active site per homodimer (Coleman, 1992), and in *Pseudomonas fluorescens*

the monomeric alkaline phosphatase PhoX has two Fe³⁺ ions and three Ca³⁺ (calcium) ions(Yong et al., 2014). The active sites of PhoA and PhoX in marine microbes have yet to be characterized but based on sequence homology are presumed to be like these model organisms, leading to the hypothesis that alkaline phosphatase activity to be limited by scarce Fe, Zn, or Co trace metals in the marine environment (Lohan and Tagliabue, 2018).

Global change is predicted to intensify phosphorus stress and alter trace metal and nutrient cycles in the ocean (Hoffmann et al., 2012; Kim et al., 2014). Throughout the North Atlantic, the utilisation of DOP is widespread(Mather et al., 2008) and whole community rates of APA are high compared with other oceanic regions (Duhamel et al., 2010; Mahaffey et al., 2014). At this time, it is not known which microbes and enzyme types are responsible for bulk APA in the North Atlantic and elsewhere. Resolving this could lead to a more quantitative understanding of how APA activity is regulated in the modern ocean, allowing better predictions of future changes in enzyme abundance and activity and the resulting influence on carbon export. In this study, we use field-based quantitative proteomics to develop an inventory of alkaline phosphatase activity and to identify nutrient-related regulatory controls on alkaline phosphatase that are distinct for different organisms. We use this as a proof of concept for developing quantitative connections between biogeochemical rates and “-omics” based measurements of microbial enzymes, a topic that is of interest to ongoing international efforts to characterize ocean metabolism.

Methods

Shipboard bioassays

All samples for this study were collected on board the *RRS James Cook* during research cruise JC150 (GEOTRACES process study GApr08), on a zonal transect at 22 °N leaving Guadeloupe on June 26th and arriving in Tenerife on August 12th, 2017, with multiple stations occupied for bioassays. A detailed description of the bioassays and analysis of environmental parameters is presented in Mahaffey et al. (submitted as a companion to this article).

Briefly, surface seawater was collected and processed according to trace metal clean protocols and before dawn. For each location, duplicate or triplicate 24 L polycarbonate (Nalgene) carboys were filled and spiked with additions of Fe, Zn or Co, as detailed in Table 1. The seawater was incubated at ambient sea surface temperature and 50 % surface light level for 48 h from dawn to dawn with a 12:12 h simulated light cycle using white daylight LED panels.

Table 1 Bioassay details at each station, showing the types of treatments, the amount of metal added, and the number of replicates per treatment for which proteomics analyses were conducted. Note that one of the three replicates of the Fe addition at the Station at 31 °W (*) was removed as an outlier from all further analysis.

	Station at 54 °W	Station at 50 °W	Station at 45 °W	Station at 31 °W
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Treatment	Control	-	-	-	-
	Fe	+ 1.0 nM	+ 1.0 nM	+ 1.0 nM	+ 1.0 nM
	Zn	+ 1.0 nM	+ 1.0 nM	+ 0.5 nM	+ 1.0 nM
	Co	+ 50 pM	+ 50 pM	+ 50 pM	+ 20 pM
Replicates per treatment		2	2	2	3*

After the incubation period, subsamples for proteins were collected into acid cleaned 10 L polycarbonate carboys (Nalgene) and immediately filtered, collecting the >0.22 µm fraction on polyethersulfone membrane filter cartridges (Millipore, Sterivex) and recording the filtered volume. Any remaining water was pressed out with an air-filled syringe, the filtration unit was sealed with clay and then frozen at -80 °C. This procedure was repeated for the second (and third where applicable) replicate of each treatment.

Alkaline phosphatase activity (APA) rate measurements

Total APA was measured in unfiltered seawater samples using the synthetic fluorogenic substrate 4- methylumbeliferoyl-phosphate (MUFP, Sigma Aldrich, Ammerman 1993, Davis et al 2019). MUFP stock solutions (100 mM in 2-methoxyethanol) were diluted with Milli-Q deionized water (200 µM stock). Unfiltered seawater was spiked with the MUFP substrate to final concentrations of 500 nM or 2000 nM MUFP for single substrate additions, or a series of replicates were incubated over a final MUFP concentration range from 100 nM to 2000 nM for the determination of enzyme kinetic parameters, Vmax and Km. Once spiked, samples were incubated in polycarbonate bottles in triplicate in the temperature and light adjusted reefer container for up to 12 hours.

MUFP hydrolysis to the fluorescent product, 4- methylumbelliferone (MUF), was measured at regular intervals (typically every 90 minutes) over a period of up to 8 - 12 hours using a Turner 10Au field fluorometer (365 nm excitation, 455 nm emission) after the addition of a buffer solution (3 : 1 sample: 50 mM sodium tetraborate solution, pH 10.5). A calibration was produced at the start and end of the cruise using MUF standards (concentration range 0–1000 nM) to ensure linearity of the fluorescence of MUF over the expected concentration range. Fluorescence response factors were determined daily using freshly prepared 200 nM MUF stocks and was used to convert the rate of change in fluorescence to MUFP hydrolysis rate, here considered to be synonymous with volumetric APA (nM P h⁻¹). Boiled seawater blanks (500 nM MUFP) were incubated in parallel with samples to ensure that there was no significant change in fluorescence due to abiotic degradation or hydrolysis over time. Enzyme kinetic parameters were determined using a range of MUFP concentrations. Michaelis-Menten equation was transformed to produce substrate-response curves or linear regression plots and the maximum hydrolysis rates (Vmax) and half saturation constant (Km)

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148 were determined using the Hanes-Woolf plot graphical linearization of the Michaelis-Menten
149 equation following Duhamel et al., 2011.

151 **Protein extraction and digestion**

152 All plastics materials were washed with ethanol and dried before usage. All samples from one station
153 were processed together in one extraction and digestion cycle. The frozen Sterivex filter cartridges
154 were transported to the laboratory on ice and cut open with a tube cutter. The filters were cut out from
155 their holders with razor blades and placed into 2 ml microfuge tubes (Eppendorf). Following
156 previously established protocols(Held et al., 2020; Saito et al., 2014)Click or tap here to enter text.,
157 proteins were extracted in a 1 % sodium dodecyl sulfate (SDS) buffer for 15 min at 20 °C, followed
158 by 10 min at 95 °C for denaturation, and 1 h at 20 °C while shaking at 350 rpm. The protein extract
159 was then centrifuged at 13.5 rpm for 20 min, with the impurities-free supernatant collected and then
160 spin-concentrated for 1 h in 5 kD membrane filters (Vivaspin, GE Healthcare). Total protein
161 concentrations were then measured by bicinchoninic assay (BCA) (Pierce) on a Nanodrop ND-1000
162 spectrophotometer (ThermoScientific). Proteins were left to precipitate in a 50:50 solvent mixture of
163 methanol and acetone (Fisher) with 0.004 % concentrated HCl (Sigma, ACS 37 %) for 5 days at -20
164 °C. At the end of the precipitation period, samples were centrifuged at 13.5 rpm at 4 °C, supernatants
165 were removed, and the remaining protein pellets were vacuum-dried (DNA110 Savan SpeedVac,
166 ThermoFisher). Pellets were redissolved in 50 µl SDS buffer, and the post-precipitation total protein
167 concentrations were measured via a second BCA assay to assess recovery. The protein extracts were
168 digested with the proteolytic enzyme trypsin (1 µg per 20 µg protein; Promega #V5280) in a
169 polyacrylamide tube gel(Lu and Zhu, 2005). The digested samples were concentrated by vacuum
170 drying and stored at -20 °C until analysis. The final volume was recorded to calculate the total protein
171 concentration in the processed sample, typically ~1 µg µl⁻¹.

172 **Target protein selection**

173 Protein biomarkers for *Synechococcus* and *Prochlorococcus* were chosen to detect DIP stress (PstS)
174 and related coping mechanisms via DOP hydrolysis (PhoA and PhoX) in our samples (Table 2). PstS
175 is the substrate-binding protein of the high-affinity phosphate ABC (ATP-Binding Cassette)
176 transporter, which is upregulated under low intracellular phosphate concentrations via the *pho* regulon
177 and has previously been used as an indicator of DIP stress (Cox and Saito, 2013; Martiny et al., 2006;
178 Scanlan et al., 1993). PhoA and PhoX are the Zn/Co-dependent and Fe-dependent alkaline
179 phosphatases, respectively, which facilitate the acquisition of phosphorus from the DOP pool.

180 Table 2 Details on the quantified peptide biomarkers that are used to represent each protein in the subsequent plots and
181 discussions. For *Prochlorococcus* strains, 'HL' and 'LL' refer to high-light and low-light adapted strains, respectively.

Protein	Quantified peptide (amino acid sequence)	Isolate strains with this peptide
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<i>Synechococcus</i>	PhoA	HYIAVALER	WH8102 (clade III)
	PhoX	SQAGAEFR	WH8102 (clade III)
	PstS	WFQELAAAGGPK	RCC307 (clade X)
<i>Prochlorococcus</i>	PhoA	IYVIDPSSSPALLER	MIT9311 (clade HL II) MIT9312 (clade HL II) MIT9314 (clade HL II)
	PhoX	GNLWIQTDGK	MIT9314 (clade HL II)
	PstS	LSGAGASFPK	MIT9301 (clade HL II) MIT9302 (clade HL II) MIT9311 (clade HL II) MIT9312 (clade HL II) MIT9314 (clade HL II) SB (clade HL II) NATL1A (clade LL I) NATL2A (clade LL I)

185

186 The criteria for a peptide of the protein biomarker to be used for quantification were as
 187 follows. Firstly, we attempted to minimise the presence of methionine and cysteines because they are
 188 subject to oxidation and cause modifications of the mass-to-charge ratio (m/z) during the analyses.
 189 Secondly, the specificity and least common ancestor of each tryptic peptide was assessed using
 190 METATRYP (<https://metatryp.who.edu/>) (Saunders et al., 2020). It has been demonstrated that
 191 carefully selected tryptic peptides, screened by using tryptic peptides databases made from genome
 192 sequences like METATRYP, can be used to identify specific proteins in mixed microbial assemblages
 193 to the species or even sub-species (ecotype) taxonomic resolution (Saito et al., 2015). Finally, the
 194 performance of each precursor ion was visually inspected in Skyline (MacLean et al., 2010) for peak
 195 shape and signal to noise-ratio during uncalibrated test measurements using a target list containing
 196 many peptides of cyanobacterial alkaline phosphatases on a subset of the incubation samples.

197 **Isotopically labelled standard peptides**

198 The absolute quantitation of the target peptides was achieved using heavy nitrogen isotope-labelled
 199 peptide standards (Saito et al., 2020). Briefly, DNA was synthesized containing the reverse-translated
 200 gene sequences for our target peptides interspaced with spacer sequences and ligated with a
 201 PET30a(+) plasmid vector using the BAMHI 5' and XhoI 3 restriction sites (Novagen; obtained
 202 through PriorityGENE, Genewiz). Different nucleotide sequences were used to encode for the spacer
 203 (amino acid sequence: TPELFR) to avoid repetition. As per manufacturer instructions, the plasmid
 204 was suspended in TE buffer (10 mM Tris-HCl, 1 mM ethylenediaminetetraacetic acid) to 10 ng μl^{-1}
 205 and of this 1 μl was added to 20 μl competent Tuner(DE3)pLysS *E. coli* cells on ice. The cells were
 206 heated to 42 °C for 30 sec to initiate transformation, followed by 2 min on ice. At room temperature,
 207 80 μl ^{15}N -enriched (98 %, Cambridge Isotope Laboratories), kanamycin-containing (50 μl ml^{-1}) SOC
 208 medium was added, and cells were incubated for 30 min at 37 °C at 300 rpm. Subsequently, 25 μl
 209 were transferred to pre-heated (37 °C) 50 μg ml^{-1} agar plates and incubated overnight. One colony was

210 added to 500 μl ^{15}N -enriched SOC medium containing 50 $\mu\text{l ml}^{-1}$ kanamycin as a starter culture and
211 incubated for 3 h at 37 °C at 350 rpm. Next, 200 μl of the starter culture were transferred into 50 ml
212 flat incubation flasks with 10 ml SOC medium and incubated for approximately 3 h at 37 °C and 350
213 rpm until the optical density at 600 nm reached 0.6. Protein production was induced by the addition of
214 100 mM isopropyl β -D-1- thiogalactopyranoside to the culture and incubating at 25 °C overnight.
215 Inclusion bodies were initially harvested using BugBuster protein extraction protocols (Novagen).
216 The remaining pellet containing the inclusion bodies, i.e. the insoluble protein fraction, was
217 resuspended in 400 μl 6 M urea, left on the shaker table at 350 rpm at room temperature for 3 h, and
218 then moved to the fridge overnight. The next morning, the proteins were reduced, alkylated, and
219 digested with trypsin as outlined above for the bioassay samples, and stored frozen at -20 °C until use.

220 **Absolute protein quantitation**

221 To determine the absolute concentration of the peptides in the heavy peptide mixture, commercial
222 standard peptides of known concentration were used. In addition to the peptides of interest, a range of
223 tryptic peptide sequences from commercially available standards (apomyoglobin, Sigma; Pierce
224 Bovine Serum Albumin, ThermoFisher) were included in the original plasmid design. Using these, the
225 calibrated concentration of the heavy peptide mixture had a relative standard deviation of 57 %, with
226 the standard deviation resulting from the cross-peptide and cross-replicate variability ($n=3$) (Fig. S1).

227 A systematic method-focused study addressing the precision and accuracy of these measurements as
228 well as the development of reference materials will be essential for using absolute quantitative
229 proteomics in the marine environment in the future(Saito et al., 2024). The linear performance range
230 of each heavy peptide standard was assessed using standard curves of the peptide mixture. Targeted
231 proteomic measurements were made by high pressure liquid chromatography with tandem mass
232 spectrometry (HPLC-MS/MS) on an Orbitrap Fusion Tribrid Mass Spectrometer (ThermoFisher).
233 Two μg of each sample diluted to 10 μl in buffer B (0.1 % formic acid in acetonitrile) was spiked with
234 10 fmol μL^{-1} of the heavy peptide mixture and injected into the Dionex nanospray HPLC system at a
235 flow rate of 0.17 $\mu\text{l min}^{-1}$. The chromatography consisted of a nonlinear gradient from 5 to 95 % of
236 buffer B with the remaining concentration consisting of buffer A (0.1 % formic acid in LC-grade
237 H_2O). Precursor (MS^1) ions were scanned for the m/z of the heavy peptide standards and their natural
238 light counterparts. The mass spectrometer was run in parallel reaction monitoring mode and only
239 peptides included in the precursor inclusion list were selected for fragmentation. Absolute peptide
240 concentrations were calculated from the ratio of the peak areas of the product ions (MS^2) of the heavy
241 peptide of known concentration to the natural light peptide (calculated in Skyline (MacLean et al.,
242 2010)). Manual validation of peak shapes was performed for each peptide and sample. Differences
243 between samples with regards to filtration volume, initial protein mass and recovery after
244 precipitation were accounted for. Final peptide concentrations will hereafter be used to represent
245 corresponding protein concentrations, with the caveat that the measurements are not able to discern

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250 active versus non-active proteins. The status of metalation and if the protein is correctly folded or
251 functions as a polymeric complex cannot be determined from this method.

252 **Identification of significant responses to metal additions**

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253 Changes in protein concentrations in response to metal additions were compared relative to the
254 unamended control treatment after 48 h. This approach accounts for any bottle effects. Due to the
255 unique challenges of ocean proteomics sampling and large-scale trace-metal clean bioassays, treatment
256 replication was limited to $n = 2$ at 54 °W, 50 °W and 45 °W and to $n = 3$ at 31 °W. Many statistical
257 tests assume normal distributions, which for $n = 2$ is not assessable. Therefore, in our case, significant
258 differences in protein concentrations were evaluated using a two-fold change criterion, in which the
259 concentrations in all replicates of the metal treatments must lie outside a two-fold change in the
260 average \pm one standard deviation of the control to be deemed a significant response. The fold-change
261 in expression and in particular the two-fold change is a commonly used metric to identify proteins that
262 are significantly more or less expressed across different conditions (Carvalho et al., 2008; Lundgren et
263 al., 2010; Zhang et al., 2006).

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264 For the biogeochemical parameters measured in the bioassays, i.e. Chl-*a*, APA and cell counts
265 replication was not limited to $n=2$ in most cases. Where $n=3$, ANOVA ($\alpha=0.05$) followed by Tukey
266 posthoc tests were applied to compare the Control treatments with other treatments.

267 **Results and Discussion**

268 **Biogeochemical setting**

269 The oligotrophic subtropical North Atlantic is marked by high deposition of Saharan desert dust,
270 delivering large amounts of Fe and other lithogenic trace metals to the surface ocean (Kunde et al.,
271 2019). During JC150, contrasting biogeochemical regimes existed in the western and eastern basin
272 with high-metal, low-phosphorus, low-nitrogen surface waters at the 54 °W and lower-metal, higher-
273 phosphorus, higher-nitrogen surface waters 31 °W Mahaffey et al. (2025; companion article).
274 Furthermore, *Synechococcus* was two-fold more abundant in the west than in the east, whilst
275 *Prochlorococcus* was more than six-fold more abundant in the east than in the west and numerically
276 more abundant than *Synechococcus* throughout. Overall, the stations at 50 °W and 45 °W exhibited
277 biogeochemical intermediates to the conditions in the east and west. The confluence of gradients in
278 both DIP and trace element availability, as well as clear shifts in microbial community structure,
279 provide a natural field laboratory to probe how environmental drivers differentially influence the
280 contributions of dominant microbes to whole-ecosystem enzyme activity.

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281 *Table 3 Date, location and biogeochemical conditions at 40 m depth at the start (t_0) of the bioassays. Biogeochemical*
282 *parameters are presented as the average \pm one standard deviation of replicate t_0 samples, except for the singlet samples of*

286 DOP at Station 4 and dCo at all stations. Mixed layer depths (MLD; defined after ⁵²) averages over multiple days, as these
 287 were not always determined on the exact day of bioassay set-up.

	Parameter	Station at 54 °W	Station at 50 °W	Station at 45 °W	Station at 31 °W
General	Date	11 th July 2017	15 th July 2017	19 th July 2017	5 th August 2017
	Location	22 °N 54 °W	22 °N 50 °W	23 °N 45 °W	22 °N 31 °W
	SST (°C)	27	27	26	25
	MLD (m)	24 ± 3 (5 th to 8 th July)	33 ± 1 (12 th to 15 th July)	42 ± 9 (17 th to 20 th July)	51 ± 8 (4 th to 8 th August)
Macronutrients	DIP (nM)	3.7 ± 2.1	3.7 ± 1.0	3.4 ± 0.8	14 ± 0.70
	DOP (nM)	87 ± 7.5	137 ± 39	112	129 ± 29
	DIN (nM)	1.5 ± 1.9	1.66 ± 0.56	3.36 ± 1.0	6.2 ± 0.0
	APA (nM h ⁻¹)	2.8 ± 0.21	2.86	2.48 ± 0.10	1.15 ± 0.08
Trace metals	dFe (nM)	1.26 ± 0.06	0.53 ± 0.06	0.83 ± 0.00	0.23 ± 0.05
	dZn (nM)	0.25 ± 0.14	0.46 ± 0.09	0.14 ± 0.01	0.04 ± 0.01
	dCo (pM)	11.0	11.1	13.0	13.9
Phytoplankton community	<i>Synechococcus</i> (cells ml ⁻¹)	3.4 ± 0.55 · 10 ³	-	-	1.6 ± 0.26 · 10 ³
	<i>Prochlorococcus</i> (cells ml ⁻¹)	29 ± 0.37 · 10 ⁴	-	-	181 ± 0.37 · 10 ⁴
	Chl- <i>a</i> (µg L ⁻¹)	0.064 ± 0.01	0.055 ± 0.01	0.110 ± 0.06	0.149 ± 0.005

Alkaline phosphatase responded differently in the traditional vs proteomic bioassays

We measured responses in alkaline phosphatase activity (traditional bioassay) and enzyme identity and provenance (proteomic assay) to metal additions across the North Atlantic gyre. Both assays were performed on matched samples at four locations (St 2, 3, 4, 7; Figure 1), allowing direct comparison of the results. In all cases for the traditional assay (shown as circles in Figure 1), bulk alkaline phosphatase activity did not increase significantly upon metal additions (see also Figure 2). However, the proteomic assay revealed that specific alkaline phosphatases did respond positively to the metal additions (squares in Figure 1), depending on the location and metal added. The discrepancy between the traditional bioassay (no response) and proteomic assay (specific, albeit patchy responses) merited additional exploration of the two approaches, which we detail below. One important explanation emerges from the fact that the APA assay covers the entire microbial community (i.e. everything from bacteria to eukaryotes) but the proteomics measurements are specific to a subset of *Prochlorococcus* and *Synechococcus*.

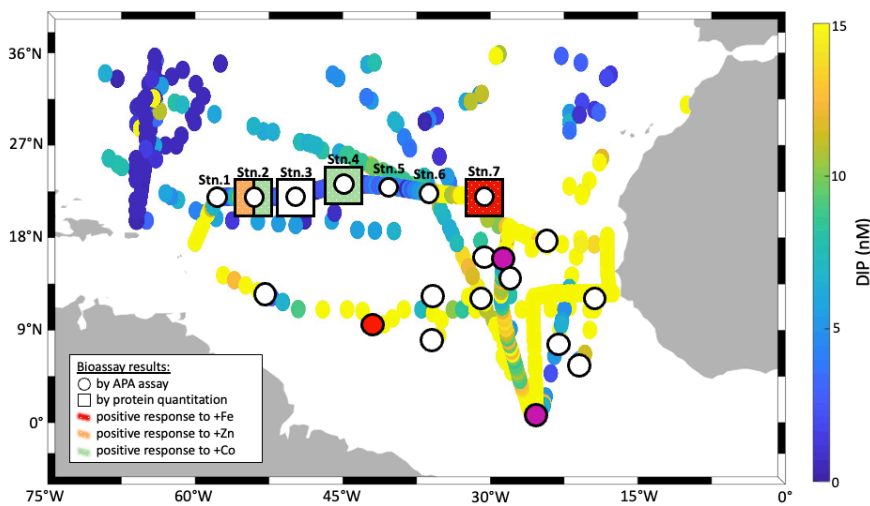


Figure 1 Summary of traditional and molecular bioassay results from this manuscript and others. Map of the North Atlantic showing surface phosphate concentrations (compiled by Martiny et al., 2019 and augmented with data from Browning et al., 2017). Overlain are locations of bioassays, where the response of APA to metal additions was tested (circles), and of bioassays, where the absolute concentration of the alkaline phosphatase proteins was measured in response to metal additions (squares). Bioassays of the present study include longitudinal station labels. The others are from Mahaffey et al. (2014) and Browning et al. (2017) as well as from additional bioassays during JC150 Mahaffey et al. (submitted as a companion to this article), but where no protein measurements were made. Symbols at bioassay locations are coloured in orange, green or red, if a positive response was observed upon addition of Fe, Zn or Co respectively.

Variability in the response to metal additions in the traditional bioassays

As mentioned above, there was no significant response in bulk alkaline phosphatase activity to metal additions at any of the stations. Here we focus on Stations 2, 3, 4, and 7, where matched proteomic assays were also conducted. However, alkaline phosphatase assays were conducted on incubations at all seven stations, and there was no response to metal addition at any of them, nor in APA rates normalized to chl-a (Figures S2-S5 and see Table S6). Given this, we sought to address whether there were other observable shifts in microbial activity as a result of the metal additions, including in Chl-A (a proxy for phytoplankton growth) and cell counts for *Prochlorococcus* and *Synechococcus* (Figure 2). While there were no statistically significant changes, either positive or negative, in any of these conventional assays, there were differences between replicate incubation bottles and within the basal conditions across the stations. Despite our efforts to homogenize the incubations and work in large volumes, this variation seems to result from stochasticity of sampling the low biomass system of the subtropical North Atlantic gyre, particularly since there is clear variation among the control bottles as well as in the amended conditions. It is also consistent with past literature including Browning et al., 2017 in which only one in eight experiments showed a metal driven response in APA and Mahaffey et al., 2014 in which there was a positive response of APA to

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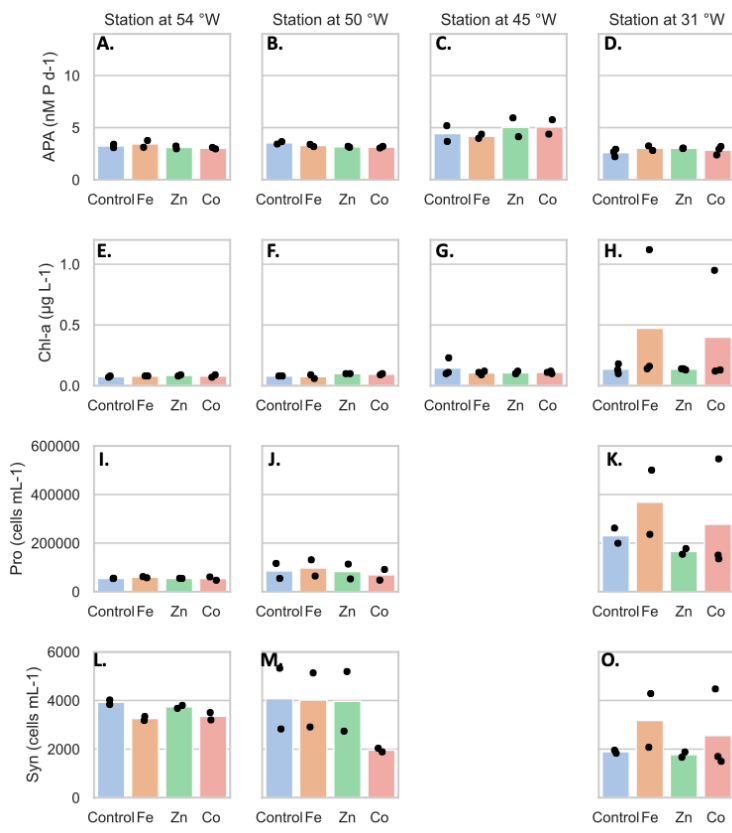
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327 Zn only in the eastern basin (see Figure 1). One possible explanation for the presence of the many null
 328 responses across the basin is that organisms could be re-allocating metals towards use in alkaline
 329 phosphatases when under phosphorus stress. Supporting this idea, a comparable re-allocation
 330 mechanism of cellular Fe between metalloproteins involved in biological N₂ fixation and
 331 photosynthesis has previously been demonstrated in the diel cycle of *Crocosphaera watsonii* (Saito et
 332 al., 2011b). Regardless, the absence of significant responses in the biogeochemical parameters
 333 contrasted notably with the observed responses in protein data detailed below.



335 Figure 2. Mean concentrations (bars) of the bioassay parameters after the addition of Fe, Zn and Co at the four stations),
 336 specifically concentrations of chlorophyll a ($\mu\text{g L}^{-1}$), (A-D) rates of alkaline phosphatase (nM d^{-1}), (E-H) Prochlorococcus
 337 abundance (cells mL^{-1}) (I-K) and Synechococcus abundance (cells mL^{-1}) (L-O). Dots represent the concentrations of each
 338 replicate. Note the data gap for cell counts at the Station at 45 °W.

339 **Strain-resolved cyanobacterial alkaline phosphatases did respond to metal additions**

340 In contrast to the bioassay results, there were clear changes in proteomically-resolved alkaline
 341 phosphatase concentrations after metal additions. We focused on the enzymes PhoA and PhoX and

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used peptides that were specific to one or more strains of either *Prochlorococcus* or *Synechococcus* (2) and represent a subset of the population of alkaline phosphatase enzymes in the ocean. We note that marine alkaline phosphatases are found at different subcellular localizations and are also known to be secreted to the environment (i.e. into the dissolved phase)(Li et al., 1998; Luo et al., 2009). Our measurements focus on the alkaline phosphatase associated with microbial cells (i.e. the particulate phase). Coming from an overview of the enzyme concentrations across isoforms, taxa and bioassays, we will discuss how these compare to the APA assay involving fluorogenic substrates.

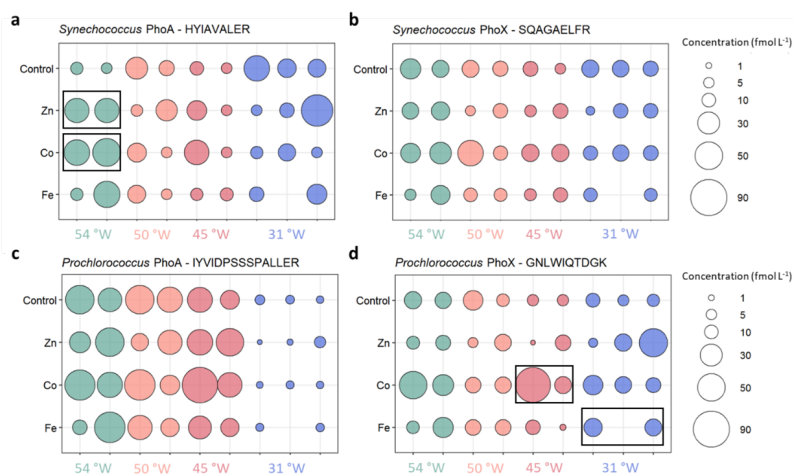


Figure 3 Absolute concentrations of the alkaline phosphatases PhoA (left column) and PhoX (right column) of *Synechococcus* (top) and *Prochlorococcus* (bottom) in the different metal treatments or the unamended control at the four probed stations. Bubbles of the same colour are replicates of the same treatment and show the concentrations as fmol enzyme per L seawater. Black boxes indicate significant change from Control treatment.

The results of all measured alkaline phosphatase concentrations are shown in Fig. 3 and all data is compiled in Table S5. *Synechococcus* PhoA and PhoX concentrations in the control treatments ranged from 6 to 43 fmol L⁻¹ and 6 to 26 fmol L⁻¹, respectively, with no clear cross-basin trend despite a strong west-to-east decreasing gradient in *Synechococcus* cell abundance (Table 3). Similarly, *Prochlorococcus* PhoA and PhoX concentrations in the control treatments ranged from 2 to 55 fmol L⁻¹ and 6 to 23 fmol L⁻¹, respectively, but with elevated PhoA at in the west and the lowest concentrations at 31 °W, which is opposite to the west-to-east increasing gradient in *Prochlorococcus* cell abundance. This suggests that we observed a gradient in DIP/trace metal nutrient stress for *Prochlorococcus*, but not for *Synechococcus*.

Our measured alkaline phosphatase concentrations were similar, albeit at the lower end, to concentrations reported for other cyanobacterial enzymes and nutrient regulators from the North Pacific (~10⁻¹ to 10³ fmol L⁻¹)(Saito et al., 2014). Interestingly, our alkaline phosphatase

concentrations occurred at the same concentration range as other macronutrient stress indicators (response regulator protein PhoP, sulfolipid biosynthesis protein SqdB, nitrogen regulatory protein P-II), all of which did not exceed tens of fmol L⁻¹ (Saito et al., 2014). In contrast, concentrations of the *Prochlorococcus* PstS transporter protein were higher, ranging from 95 to 472 fmol L⁻¹ (Table S5). This is within the concentration range of other cyanobacterial nutrient transporters, such as the urea transporter UrtA, measured previously (Saito et al., 2014). In mediating nutrient stress, particularly phosphorus stress, the relative role of transporter proteins (such as PstS) versus other strategically deployed enzymes like alkaline phosphatase in the oligotrophic specialists *Synechococcus* and *Prochlorococcus*, represents an interesting avenue for future research.

Evidence for direct biochemical regulation of certain alkaline phosphatases by metals

Our strain-specific, quantitative proteomics approach allowed us to resolve contrasting responses across the sites. The responses differed with varying phytoplankton species, alkaline phosphatase form and stimulating metal addition, consistent with differences in the biogeochemical regimes (Table 3). At the iron-rich westernmost station (Station 4; 54 °W), the *Synechococcus* PhoA concentration increased six- and seven-fold upon addition of Zn (to 38 ± 0.56 fmol L⁻¹) and Co (to 47 ± 6.8 fmol L⁻¹) relative to the control (6.7 ± 1.5 fmol L⁻¹), respectively. At one intermediate Station (45 °W), the *Prochlorococcus* PhoX concentration increased 8-fold upon addition of Co relative to the control. Notably, a direct response of alkaline phosphatase to an addition of Co has not been shown in the field before. In contrast, at the low iron easternmost station, the *Prochlorococcus* PhoX increased over two-fold upon Fe addition (to 18 ± 2.6 fmol L⁻¹) relative to the control (8.2 ± 2.4 fmol L⁻¹).

At least three scenarios are possible to explain the increased alkaline phosphatase concentrations of *Synechococcus* and *Prochlorococcus* in seawater in these treatments – two biochemical and one growth driven hypotheses. First, the metal addition may stimulate the production of the alkaline phosphatase enzyme via a direct or indirect metal regulation on the expression of this enzyme, as was previously observed for PhoA with Zn additions in *Synechococcus* cultures (Cox and Saito, 2013). Second, the metal addition may prevent the degradation of the existing alkaline phosphatases by filling empty metal co-factor sites (Bicknell et al., 1985), with the caveat that PhoA is likely to be periplasmic and hence unlikely to be actively degraded (Luo et al., 2009). Both biochemical scenarios allow for increased alkaline phosphatase concentrations at a constant cell abundance. The third explanation is that the alkaline phosphatase concentration increases because the metal addition stimulates overall cell growth, resulting in higher phosphorus demands and hence more production of alkaline phosphatase proteins by the cell. This could manifest itself as higher cell abundances in addition to increased alkaline phosphatase concentration per unit biomass.

While the different scenarios are not mutually exclusive, our quantitative proteomic approach allowed us to discern between biochemical and growth mechanisms by normalising the alkaline

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phosphatase concentrations to the total cell counts of *Prochlorococcus* and *Synechococcus*, caveating that the cell counts are not strain-specific, unlike the peptide-based protein measurements. Cell counts did not change significantly across these treatments Mahaffey et al. (submitted as a companion to this article). which means that the trends of increased alkaline phosphatase concentration per L seawater persisted in bioassays (i.e. +Zn and +Co at 54 °W and +Fe at 31 °W; cell counts do not exist for 45 °W) even when converted to the number of alkaline phosphatase enzymes per cell, indicating biochemical regulation as opposed to simply growth of the responsible organism. Specifically, the concentration of *Synechococcus* PhoA increased to 8418 ± 673 enzymes cell⁻¹ upon Co addition and to 6057 ± 48 enzymes cell⁻¹ upon Zn addition relative to 1025 ± 257 enzymes cell⁻¹ in the control at 54 °W, while the concentration of the *Prochlorococcus* PhoX increased to 59 enzymes cell⁻¹ upon Fe addition relative to 19 ± 7 enzymes cell⁻¹ in the control at 31 °W. Therefore, a direct biochemical metal control on the alkaline phosphatase concentrations during the bioassays is plausible (i.e. either of the first two explanations) and adds weight to the hypothesis for the localised metal-phosphorus co-limitation in the subtropical North Atlantic (Browning et al., 2017; Jakuba, R. Wisniewski et al., 2008; Mahaffey et al., 2014; Saito et al., 2017; Shaked et al., 2006).

These estimates of enzyme copies per cell are potentially underestimates as multiple *Prochlorococcus* and *Synechococcus* ecotypes co-exist and the alkaline phosphatase peptide sequences probed here do not encompass all of them (Table 2). Moreover, it is also possible that there are additional isoforms of alkaline phosphatase present in these organisms that have yet to be identified (Bradshaw et al., 1981). Yet in these marine cyanobacteria, the cellular concentration of alkaline phosphatase was much higher compared to a measurement in the model bacterium *E. coli*, which contained ~4 PhoA copies cell⁻¹ (Wiśniewski and Rakus, 2014). This underscores the ecological demand for alkaline phosphatases due to the significant depletion of phosphorus in the marine environment. It is yet to be determined whether the per-cell estimates of alkaline phosphatases presented here are the norm for marine cyanobacteria, or whether these estimates are exceptionally high due to the prevalence of phosphorus stress in our study region.

While PhoX enzymes are unknown to use Co as a metal co-factor and the response at 45 °W warrants further investigation, the substitution of Zn with Co in PhoA has been hypothesised previously based on the distributions of trace metals and phosphate in the Sargasso Sea (Jakuba, R. Wisniewski et al., 2008; Saito et al., 2017). The results from 54 °W support this hypothesis as the addition of both Zn and Co were associated with almost equal increases of the *Synechococcus* PhoA concentration relative to the control. It is thought that while Zn is the preferred metal centre for PhoA, it is possible to substitute Co for Zn in the protein, such as occurs in *Thermotoga maritima* (Wojciechowski et al., n.d.) *in vivo* and *in vitro* in *E. coli* (Gottesman et al., 1969). Metabolic substitution capabilities between Zn and Co in carbonic anhydrases have previously been identified in marine phytoplankton, with similar or slightly reduced growth rates for a range of marine diatoms and

coccolithophores, when Zn was replaced with Co in carbonic anhydrases (Dupont et al., 2006;
 Kellogg et al., 2020; Morel et al., 2020; Price and Morel, 1990; Sunda and Huntsman, 1995;
 Timmermans et al., 2001; Xu et al., 2007; Yee and Morel, 1996). However, in certain organisms such
 as in the coccolithophore *Emiliana huxleyi*, it is possible that Co is the preferred metal co-factor since
 the growth rate was higher under replete Co than under replete Zn. One reason could be the
 simultaneous development of ocean chemistry and cyanobacterial metabolism under the Co- and Fe-
 replete, but Zn-deplete conditions of the ancient ocean ~2.5 Gyr ago (Dupont et al., 2006; Johnson et
 al., 2024; Saito et al., 2003). Another explanation for Co use in alkaline phosphatases may require
 maintaining low intracellular availability of Zn to avoid toxicity through inhibition of Co insertion by
 high Zn into cobalamin (Hawco and Saito, 2018). Supporting this, the cyanobacterium *Synechococcus*
bacillaris and *Prochlorococcus* were found to have absolute Co requirements for growth (Sunda and
 Huntsman, 1995). Together with these aspects, our insights from the bioassay response at 54 °W
 merits further investigations into whether *Synechococcus* can interchange Zn and Co in PhoA, and
 indeed which metal is preferred. This would be an important insight for considerations of
 stoichiometric plasticity and niche partitioning across the vast Zn- and Co-depleted regions of the
 ocean, especially where dZn can become depleted to levels similar or below dCo (Kellogg et al.,
 2020).

Across all bioassays, the addition of Zn did not increase the concentration of any presumably
 Fe-dependent PhoX, and the addition of Fe did not increase the concentration of any presumably Zn-
 or Co-dependent PhoA. In other words, no significant unexpected responses were observed.
 Nevertheless, there are some non-significant trends upon the addition of Co that warrant further study.
 For example, the addition of Co increased the putative Fe-containing *Prochlorococcus* PhoX protein
 concentration dramatically in one of the replicates at 45 °W and hence, Co could be an efficient metal
 co-factor in PhoX (as in the bacterium *Pasteurella multocida* (Wu, Jin-Ru et al., 2007)), or at least,
 directly or indirectly stimulate production of PhoX. This contrasts with the results of Kathuria &
 Martiny (Kathuria and Martiny, 2011), who hypothesized an enzyme inhibiting role of Co (and Zn;
 with Fe untested) for the activity of both *Synechococcus* and *Prochlorococcus* PhoX by replacing
 Ca²⁺ at the active site.

Taken together, the results of our bioassays suggest that alkaline phosphatase enzymes are
 affected by trace metal concentrations, and that the response to Zn, Co or Fe may be species or strain
 specific. The metal effects differed between the responsive enzyme type (PhoA versus PhoX) and
 phytoplankton species (*Prochlorococcus* versus *Synechococcus*) at contrasting biogeochemical
 settings across the basin. It is plausible that the significant changes in protein concentration can result
 directly from the metal addition triggering more alkaline phosphatase production per cell. This
 demonstrates that the cycling of macronutrients and metals are intermittently linked and that the
 nature of that linkage depends on microbial community composition

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Towards a quantitative, in situ marine metalloproteome of *Synechococcus*

An advantage of absolute quantitative measurements over relative proteomics data is the ability to relate the absolute protein concentrations to other data types, including biological rate measurements and cellular metal stoichiometry. To this end, the concentrations of the Zn, Co or Fe-dependent alkaline phosphatases measured in this study naturally lead to two questions: First, how much metal is allocated as alkaline phosphatase co-factors in the cell, and how does this compare to total cellular metal content? Second, how does the APA estimated from enzyme abundance compare to assay-based APA?

To address these questions, model calculations were performed. Variables other than the absolute concentrations of the alkaline phosphatases were either measured concomitantly during the bioassays, such as cellular metal quotas, cell abundance, APA and DOP concentration, or sourced from the literature such as strain specific contribution to cell abundance, phosphoester contribution to the DOP pool, enzyme kinetics parameters and subcellular enzyme localization (Table S7). We chose to use the *Synechococcus* PhoA concentrations in the control treatments at 54 °W after 48 h in these calculations for three reasons: First, cellular metal quotas of *Synechococcus* but not of *Prochlorococcus* were measured in this treatment, due to limited sampling capacity and small cell sizes for *Prochlorococcus* (Sofen et al., 2022). Second, enzyme kinetics parameters of the PhoA rather than the PhoX isoform are well documented in the literature (Lazdunski and Lazdunski, 1969). Third, estimates for the contribution of *Synechococcus* strain WH8102 (to which our measured PhoA is specific) to total *Synechococcus* counts exist from previous studies nearest to 54 °W (Ohnemus et al., 2016). A similar reasoning applied to the phosphoester contribution to the DOP pool. For ease, more detailed explanations, all values, and assumptions are in Tables S3 and S4.

Equation 1a approximates the cellular Zn allocation towards the *Synechococcus* PhoA from the replicate-averaged PhoA concentration in seawater normalised to cell abundance, assuming full metalation of the enzyme with four metal ions per dimer (Coleman, 1992). The cell abundance is a function of *Synechococcus* cell counts and the fractional abundance of strain WH8102, to which the measured PhoA is specific. Equation 1b expresses the results of Equation 1a as a fraction of the total cellular Zn content.

$$\text{Allocated Zn}_{\text{PhoA}} = \text{metalated co-factors}_{\text{PhoA}} \times \text{PhoA}_{\text{SW}} / (\text{Syn. abundance} \times \text{WH8102 fraction}) \quad (1a)$$

$$\text{Fractional allocated Zn}_{\text{PhoA}} = \text{Zn}_{\text{PhoA}} / \text{Zn}_{\text{total cell}}. \quad (1b)$$

The amount of metal allocated to PhoA in *Synechococcus* is 3,054 atoms cell⁻¹. This translates to a maximum fractional contribution towards the total cellular Zn content of 0.66 % after dividing by cellular Zn measured using SXRF (Table S1). If the co-factor in PhoA is assumed to be occupied by

Deleted: Alkaline phosphatase abundances in the context of bulk community APA

Alkaline phosphatase activity and phytoplankton biomass (by Chl-*a* proxy) did not increase significantly upon the metal additions in the bioassays, neither together with the responses observed in the absolute enzyme concentrations at 54 °W, 45 °W and 31 °W, nor in any other treatments or locations (also Mahaffey et al.; submitted). A quantitative explanation for this – and hence a demonstration of the power of proteomics on the organism level – emerges from estimates of enzyme abundance-based enzyme rates, where the APA assay covers the entire microbial community (i.e. everything from bacteria to eukaryotes) but the proteomics measurements are specific to a subset of *Prochlorococcus* and *Synechococcus*.

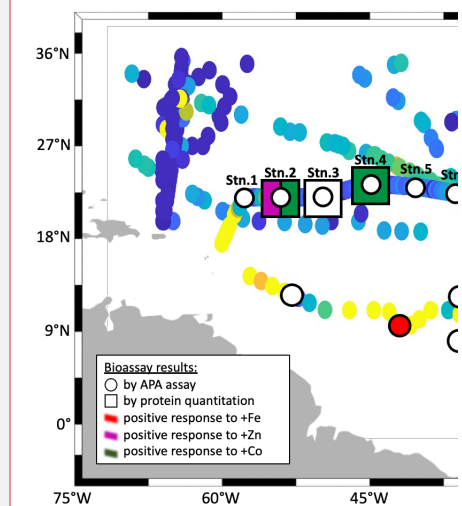


Figure 3 Map of the North Atlantic showing surface phosphate concentrations (compiled by Martiny et al., 2019 (Martiny et al., 2019); augmented with data from Browning et al. 2017 (Browning et al., 2017)). Overlain are locations of bioassays, where the response of APA to metal additions was tested (circles), and of bioassays, where the absolute concentration of the alkaline phosphatase proteins was measured in response to metal additions (squares). Bioassays of the present study include longitudinal station labels. The others are from Mahaffey et al. (2014) (Mahaffey et al., 2014) and Browning et al. (2017) (Browning et al., 2017) as well as from additional bioassays during JC150 Mahaffey et al. (submitted as a companion to this article). but where no protein measurements were made. Symbols at bioassay locations are coloured in red, purple or green, if a positive response was observed upon addition of Fe, Zn or Co respectively.

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613 Co^{2+} instead of Zn^{2+} , the fractional contribution to the cellular Co content is 38 %, due to the lower
 614 total cellular Co content of *Synechococcus* compared to Zn (Table S1). It is possible that the active
 615 sites of PhoA are occupied by a mixture of Zn and Co, incompletely metalated, or under competition
 616 by metals other than Zn or Co. Nevertheless, these low fractional contributions of PhoA-allocated Zn
 617 appear biochemically reasonable, as the majority of Zn in *Synechococcus* appears to be stored in
 618 metallothioneins to maintain Zn homeostasis and potentially supply alkaline phosphatases with Zn as
 619 needed (Cox and Saito, 2013; Mikhaylina et al., 2022). However, our bioassay results also suggest Zn
 620 may not be the preferred co-factor in *Synechococcus* PhoA: The larger response of this enzyme to Co
 621 additions at 54 °W suggests the effective substitution of or even preference for Co (Fig. 2). This
 622 aligns with evolutionary arguments (see ‘Metal control on alkaline phosphatases’) and imply that
 623 PhoA is a potential major sink of cellular Co. This also implies that *Synechococcus* growth may be
 624 sensitive towards Co-phosphorus co-limitation in the oligotrophic ocean.

625 Equation 2a approximates the *Synechococcus* PhoA-abundance based hydrolysis rate as a
 626 function of the PhoA concentration (converting molarity units to grams using its molecular weight),
 627 phosphoester substrate concentration, and Michaelis-Menten kinetics parameters V_{\max} and K_m , the
 628 maximum reaction rate and half-saturation constant, respectively, derived from an *E. coli* homologue
 629 of PhoA (Table S2). Equation 2b expresses the results of Equation 2a as a fraction of the ‘total APA’,
 630 a function of the measured MUF-P assay-based APA with a correction applied for the subcellular
 631 localisation of marine alkaline phosphatases, of which only the periplasmic-outwards fraction (~20 to
 632 80 %) is detectable via the MUF-P assay. In other words, the calculated ‘total APA’ accounts for both
 633 the dissolved and particulate activity. Details on made assumptions are in the supplement (Table S2).

634 $\text{Rate}_{\text{PhoA}} = \text{PhoA}_{\text{SW}} \times \text{molecular weight} \times V_{\max} \times \text{substrate} / (\text{substrate} + K_m)$ (2a)

635 where substrate = DOP \times phosphoester fraction

636 $\text{Fractional Rate}_{\text{PhoA}} = \text{Rate}_{\text{PhoA}} \times \text{periplasmic-outwards fraction} / \text{assayed APA}$ (2b)

637 The protein abundance-based rates range from 0.00517 nM h^{-1} to 0.213 nM h^{-1} for the Zn-
 638 dependent *Synechococcus* PhoA and from 0.00428 nM h^{-1} to 0.0187 nM h^{-1} for the Co-dependent
 639 PhoA (Fig. 4a and b), using the *E. coli* kinetics parameters for each metal that are slower under Co
 640 coordination. In terms of fractional contributions to total APA, the rate estimates translate to
 641 maximally 5.2 % for Zn-PhoA and 0.46 % for the Co-PhoA. Regardless of the choice of enzyme
 642 kinetics, it appeared that *Synechococcus* PhoA contributed a small component to the total APA in our
 643 bioassays. This concurs with the observed increase in concentration of *Synechococcus* PhoA upon
 644 metal addition versus the null response in APA. Applying the same calculation and kinetics
 645 parameters for the case of the *Prochlorococcus* PhoA yields abundance-based rates between 0.035

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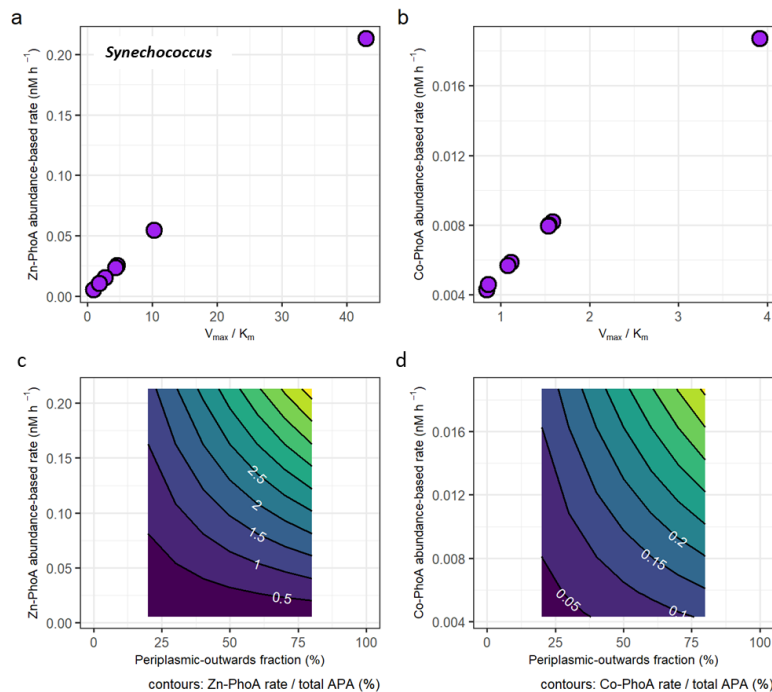
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655 nM h⁻¹ and 1.4 nM h⁻¹ for the Zn-PhoA and between 0.029 nM h⁻¹ and 0.13 nM h⁻¹ for the Co-PhoA,
 656 which translate to maximal contributions to the total APA of 35 % and 3.1 %, respectively. These
 657 higher rates and fractional contributions compared to the *Synechococcus* PhoA are due to the higher
 658 concentrations of the *Prochlorococcus* PhoA than the *Synechococcus* PhoA in the chosen samples.
 659 The taxon-specific alkaline phosphatase concentrations illustrate the challenge of interpreting bulk
 660 enzyme activities when the functional enzyme class is produced by many biological taxa
 661 (cyanobacteria, heterotrophic bacteria, diatoms etc.). In essence, the different bioassay responses
 662 shown here demonstrate the need to further develop a “meta-biochemistry” capability to understand
 663 biogeochemical reactions at the mechanistic level.

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665 **Figure 4.** (a) Protein abundance-based APA estimates of *Synechococcus* Zn-dependent PhoA as a function of different
 666 enzyme kinetic parameters V_{max} and K_m . (b) Same as (a), but with enzyme parameters for the less efficient Co-dependent
 667 PhoA. (see Table S2). (c) The fraction of the Zn-PhoA abundance-based APA from (a) over the total APA. (d) Same as (c)
 668 but using the Co-PhoA rates from (b). Note the scale difference between (a,c) and (b,d).

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669 The enzyme-based rates calculated here may be below that of bulk activity (APA assay)
 670 because our protein analysis focused on few species and only in the particulate phase. Alkaline
 671 phosphatase is known to be more abundant in the dissolved phase, for example as much as 72% of the
 672 APA was observed in Red Sea samples to be in the dissolved phase (Li et al., 1998). Moreover, the

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680 periplasmic location of alkaline phosphatase has been observed to result in loss during preservation.
681 In a preservation study, PhoA was notably the protein with the lowest recovery in *Synechococcus*
682 WH8102 after a month in storage compared to $101\% \pm 27\%$ for the fifty most abundant proteins(Saito
683 et al., 2011a). The combination of multiple abundant and rare microbial sources of alkaline
684 phosphatases together contribute to the particulate, and when secreted or lost, dissolved reservoirs that
685 make up the bulk APA.

686 Conclusions

687 This study performs taxon-specific alkaline phosphatase isoform analysis via absolute quantitative
688 proteomics on *Prochlorococcus* and *Synechococcus*, coupled to enzyme bioassays. This approach
689 supports the use of Zn, Co and Fe in alkaline phosphatases in the natural oceanic environment, but
690 also adds complexity to our understanding of how these enzymes are regulated in a biogeochemical
691 context. Our mechanistic perspective revealed that these two highly abundant microbes are only
692 minor contributors to bulk APA, which carries important implications for the interpretation of the
693 widely used fluorescent APA assay. Additionally, within this picocyanobacterial class, we observed
694 heterogeneous responses of the alkaline phosphatase enzymes depending on the protein, taxonomy,
695 biogeochemical context, and treatment. This indicates that there is significant biological diversity in
696 the responses of individual marine organisms to experimental treatments that can be resolved by
697 combining enzyme assay measurements with quantitative proteomics. Our results indicate a need for
698 biochemical characterisation of key marine alkaline phosphatases, particularly with regards to their
699 kinetics and metal co-factors as highlighted by the potential importance of Co as a metal co-factor in
700 PhoA and possibly PhoX. Future efforts to understand the biochemical properties of marine microbes
701 will benefit the connected interpretation of molecular, enzymatic, and biogeochemical assays, and in
702 turn our understanding of nutrient cycling in the ocean system.

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718 experiments at sea. BST and ELM conducted the cell quota measurements. MCL, CM, AT and MAS
719 led the research campaign. All authors commented on the manuscript.

720 **Competing interests:** The authors declare no competing interests.

721 **Data Availability:** Source data for all main and supplementary figures are provided in the
722 supplement. The mass spectrometry proteomics data have been deposited to the ProteomeXchange
723 Consortium via the PRIDE [1] partner repository with the dataset identifier PXD053717.

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