

Contributions of primary anthropogenic sources and rapid secondary transformations to organic aerosol pollution in Nanchang, Central China

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Abstract

Due to the complex composition of organic aerosols (OAs), identifying their sources and understanding their dynamics remain challenging, particularly in urban environments of China where natural and anthropogenic influences to OAs intersect. This study aimed to clarify the relative contributions of primary emissions and secondary formation to urban OAs and confirm the sources and influencing factors of OA pollution. We analyzed major polar organic compounds in fine particulate matter (PM_{2.5}) collected over one year in Nanchang, Central China. Specific biomarkers and diagnostic ratios were applied to characterize OA sources and distribution patterns, while chemical mass balance (CMB) models and tracer-based approaches were used to estimate source contributions. Statistical analyses were conducted to investigate OA characteristics and drivers during winter pollution episodes. Notably, fatty acids, fatty alcohols, and saccharides predominated over lignin, resin products, sterols, glycerol, hydroxy acids, and aromatic acids, with molecular profiles indicating both anthropogenic and biogenic origins. Source apportionment results showed that primary organic carbon (POC) and primary OAs (POAs) contributed 58% of total organic carbon and 23% of PM_{2.5} mass, respectively, compared with 8% and 4% from secondary organic carbon (SOC) and secondary OAs (SOAs). Anthropogenic sources dominated, accounting for approximately 90% of POC and POAs as well as 60% of SOC and SOAs. Seasonal patterns revealed stronger biogenic influences in spring–summer, whereas

anthropogenic emissions dominated in autumn–winter. Short-term winter episodes were characterized by rapid secondary formation, facilitated by elevated primary emissions and favorable oxidation conditions, including enhanced light intensity and nitrogen oxides.

1 Introduction

Fine particulate matter (PM), specifically PM with an aerodynamic diameter of $<2.5\ \mu\text{m}$ ($\text{PM}_{2.5}$), affects the environment and climate, posing severe threats to human health (Huebert et al., 2003; Kanakidou, 2005; Riipinen et al., 2012; Shiraiwa et al., 2017; Yazdani et al., 2021). This is evident in China, which has become a focal point for $\text{PM}_{2.5}$ pollution, characterized by recurrent haze episodes that have intensified since 2011 (Cao et al., 2012; Zhang et al., 2012; Huang et al., 2014). To tackle this pressing challenge, the Chinese government has implemented various measures, including the Air Pollution Prevention and Control Action Plan in 2013, revisions to the Air Pollution Prevention and Control Law in 2014, and the Three-year Action Plan for Winning the Blue Sky Defense War in 2018. Such initiatives have remarkably reduced $\text{PM}_{2.5}$ levels, with concentrations below $35\ \mu\text{g m}^{-3}$ as of 2020 (Figure S1a). However, the challenge persists, particularly during winter, as exemplified by the winter of 2023–2024 when $\text{PM}_{2.5}$ concentrations in many cities exceeded the national air quality standards (Figures S1b and S1c). This underscores an urgent need for effective air quality management strategies (An et al., 2019; Cao and Cui, 2021; Chen et al., 2021; Wu et al., 2021; Zhang et al., 2024). To address urban $\text{PM}_{2.5}$ pollution, comprehensive investigation of pollutant components, identification of pollution sources, and evaluation of influencing factors are required.

Organic components constitute an important part of $\text{PM}_{2.5}$, accounting for 30%–50% of its mass, with primary and secondary sources (Huang et al., 2014; Zhang et al., 2016; Haque et al., 2022). Primary organic carbon (POC) or primary organic aerosol (POA) is directly emitted from various sources, including biomass burning, coal combustion, vehicle exhaust, cooking, plant debris, and fungal spores (Guo et al., 2012; Yazdani et al., 2021). Secondary organic carbon (SOC) or secondary organic aerosol (SOA) forms in the atmosphere through the photooxidation of biogenic and anthropogenic volatile organic compounds (BVOCs and AVOCs, respectively) (Lewandowski et al., 2008; Stone et al., 2009). Common biogenic precursors include hemiterpenes, monoterpenes, and sesquiterpenes, whereas typical anthropogenic precursors include toluene

and polycyclic aromatic hydrocarbons (PAHs) (Claeys et al., 2004; Al-Naiema and Stone, 2017; Jaoui et al., 2019). Owing to the complex composition of OAs, their diverse emission sources, and the impact of meteorological conditions and photochemical oxidation processes, OA source identification is complicated (Ding et al., 2013; Zhang et al., 2024). For years, research on the composition and source apportionment of OAs has attracted considerable attention. However, a unified conclusion regarding OA sources, particularly in China's intricate urban environments influenced by natural and anthropogenic factors, remains elusive (Wang et al., 2006; Fu et al., 2008; Guo et al., 2012; Ding et al., 2014; Xu et al., 2022). Various source apportionment methods have been employed. Such methods include the elemental carbon-based and water soluble OC-based methods (EC-based and WSOC-based methods, respectively; Xu et al., 2021), the compound tracer method (tracer-based method; Ding et al., 2014; Ren et al., 2021; Haque et al., 2023), the isotope signature method (Tang et al., 2022; Xu et al., 2022 and 2023; Zhang et al., 2024), and the use of source apportionment models, such as the chemical mass balance (CMB) and positive matrix factorization models (Xu et al., 2021; Zhang et al., 2023). However, these methods yielded different source contributions. The relative significance of primary emissions versus secondary formation in urban OAs continues to be controversial. Further investigation is required to clarify the sources of OA pollution and the influencing factors.

Thus, this research presents a comprehensive assessment of OAs in Nanchang, a city in Central China characterized by moderate $PM_{2.5}$ pollution levels that reflect broad urban atmospheric conditions across China (Figure S1). Employing the CMB model and tracer-based method, as well as EC-based method, we quantitatively evaluated the contributions of POC and SOC to OC, as well as the contributions of the corresponding POA and SOA to the mass of $PM_{2.5}$. Results showed that urban OAs were predominantly influenced by anthropogenic sources, with primary contributions exceeding secondary contributions. Based on a continuous year-long observation (November 1, 2020 to October 31, 2021, 365 daily samples), we discovered that the anthropogenic contributions significantly increased in autumn and winter. However, the biogenic contributions increased in spring and summer. Short-term winter pollution episodes were promoted by rapid secondary transformation, primarily due to high primary emissions and favorable oxidation conditions, including increased light intensity and nitrogen oxides (NO_x). This study integrates multiple source apportionment methods and accounts for the seasonal characteristics of OA pollution, providing a robust

framework for understanding urban air quality dynamics. The findings have implications for governmental strategies for mitigating air pollution and preventing haze episodes in Nanchang and its environs. The insights obtained serve as a future reference for investigating the impacts of organic compounds in PM_{2.5} on visibility, public health, and climate change.

2 Materials and methods

2.1 Sampling sites and sample collection

The study area was Nanchang, Central China, with sampling at the East China University of Technology located in the northwest region of the city (28.72°N, 115.83°E). As detailed in our previous investigations (Guo et al., 2024a and 2024b), the sampling site was in a mixed-use area characterized by educational, commercial, transportation, and residential activities, devoid of significant local pollution sources. The sampler was strategically placed on the rooftop of a six-story building, approximately 20 m in height, ensuring an unobstructed sampling environment.

Sampling was conducted from November 1, 2020, to October 31, 2021, with daily PM_{2.5} sample collections. An 8-inch × 10-inch quartz fiber filter (Pall Tissuquartz, USA) was employed in a high-volume air sampler at a flow rate of $1.05 \pm 0.03 \text{ m}^3 \text{ min}^{-1}$ for sample collection. The meteorological parameters and gaseous pollutant data for the sampling period were obtained from publicly available online monitoring platforms (<https://weatherandclimate.info> and <http://www.aqistudy.cn/>). In particular, the meteorological data were retrieved from the ground meteorological observation station at Nanchang Changbei Airport, and gaseous pollutant data were obtained from the air quality monitoring station operated by the Jiangxi Academy of Forestry. These stations are the closest to the sampling site, located approximately 10 km and 2 km away, respectively. Both stations are ground-based, equipped with online monitoring instruments, and situated on open, flat terrain without surrounding buildings or obstructions. A detailed summary of the prevailing meteorological conditions and air quality during the sampling period is provided in Text S1 and Figure S2.

2.2 Chemical analyses

The OC and EC concentrations in the PM_{2.5} samples were quantified using the Desert Research Institute Model 2001 Carbon Analyzer, following the thermal/optical reflectance protocol established by the Interagency Monitoring of Protected Visual Environments (IMPROVE) (Chow et al., 2007). A 1.0-cm² filter sample was

placed in a quartz boat in the analyzer and subjected to incremental heating at predetermined temperatures: 140 °C for OC1, 280 °C for OC2, 480 °C for OC3, and 580 °C for OC4 in a non-oxidizing helium atmosphere; and 580 °C for EC1, 740 °C for EC2, and 840 °C for EC3 in an oxidizing atmosphere containing 2% oxygen in helium. Repeated analyses were performed, demonstrating an analytical uncertainty of $\pm 10\%$.

We employed methodologies outlined by Wang et al. (2005 and 2006) and Fu et al. (2008 and 2010) for the extraction and derivatization of organic compounds. The involved filter samples underwent three consecutive ultrasonic extractions using a dichloromethane–methanol mixture (2:1 v/v), followed by the concentration and drying of the resulting extracts. Prior to instrumental analyses, N,O-bis-(trimethylsilyl) trifluoroacetamide and pyridine were introduced to derivatize the polar compounds in the extract. The extract species included polar and non-polar organic compounds, and they were simultaneously detected by the involved instrument. We used non-polar C13 n-alkanes as internal standards for quantitative analyses.

The identification and quantification of organic compounds were achieved via gas chromatography–mass spectrometry (GC–MS) using a Thermo Scientific TRACE GC system coupled to a Thermo Scientific ISQ QD single quadrupole mass spectrometer. GC separation was performed using a DB-5MS fused silica capillary column. Text S2 and Table S1 present the parameters for GC–MS analysis, including temperature elevation procedures, qualitative and quantitative methods for the compounds, and quality assurance and control protocols. Polar and nonpolar organic compounds in the extracts were simultaneously analyzed; however, this study focused on the results regarding the polar compounds; the findings related to nonpolar compounds are provided in our earlier publications (Guo et al., 2024a and 2024b).

2.3 Source apportionment methods

To quantify the contributions of various primary sources to OC and PM_{2.5}, i.e., the POC and POA, we utilized the CMB model (version 8.2), a widely accepted source apportionment method developed by the United States Environmental Protection Agency (Lewandowski et al., 2008; Stone et al., 2009; Guo et al., 2012; Wu et al., 2020; Xu et al., 2021). The model assumes that the chemical composition of pollutants remains unchanged during transport, allowing measured chemical species at the receptor to be expressed as the linear sum of contributions from individual source categories. Accurate CMB results depend critically on the selection of representative emission sources and chemical tracers. These must include all major contributors and feature

chemical markers that are stable during transport and distinct among sources (Stone et al., 2009). To meet these requirements, we first identified four major primary OC sources: biomass burning, coal combustion, vehicular exhaust, and cooking emissions, based on China's atmospheric PM and gaseous pollutant emission inventories (Huang et al., 2015; Li et al., 2019; Tong et al., 2020). We also considered the contributions of plant debris and fungal spores to OC, based on the regional studies on OC sources (Fan et al., 2020; Xu et al., 2020; Wu et al., 2020). Representative source profiles and tracers of these sources were compiled from recent local and regional reports. For example, biomass burning releases substantial amounts of levoglucosan, accounting for ~8.3% of OC in local profiles (Zhang et al., 2007), and was therefore chosen as its marker compound. For coal combustion and vehicular exhaust, both characterized by complex mixtures include n-alkanes, polycyclic aromatic hydrocarbons (PAHs), and hopanes (Rogge et al., 1993a; Oros and Simoneit, 2000; Zhang et al., 2008; Cai et al., 2017). Local emission profiles indicated coal combustion tends to emit a higher proportion of 3- to 4-ring PAHs (Zhang et al., 2008), whereas vehicular exhaust is characterized by higher contributions from C₂₀–C₂₂ n-alkanes (Cai et al., 2017). Consequently, we selected 3- to 4-ring PAHs and C₂₀–C₂₂ n-alkanes as the representative organic markers for local coal combustion and vehicle emissions, respectively. Cooking emissions are dominated by fatty acids, particularly saturated palmitic acid (C_{16:0}) and stearic acid (C_{18:0}), as well as unsaturated fatty acids like palmitoleic acid (C_{16:1}) and oleic acid (C_{18:1}). Given the instability of unsaturated fatty acids in ambient air (Kawamura and Gagosian, 1987; Rudich et al., 2007), only the more stable saturated fatty acids—palmitic and stearic acids—were used as characteristic markers for local cooking emissions (He et al., 2004; Zhao et al., 2007 and 2015). For plant debris, emission profiles from Los Angeles area indicate that plants release considerable quantities of long-chain odd-carbon-number n-alkanes (e.g., C₂₅, C₂₇, C₂₉) (Rogge et al., 1993b); however, many local studies indicate that biomass burning, coal combustion, and vehicle exhaust also emit long-chain n-alkanes to some extent (Zhang et al., 2007, 2008; Cai et al., 2017). Overlap among sources diminishes the diagnostic value of long-chain n-alkanes, some investigations have consequently proposed glucose as a more selective tracer for plant debris (Fan et al., 2020; Xu et al., 2021). Puxbaum and Tenze-Kunit (2003) report that glucose comprises approximately 5.2% of plant-derived organic carbon, whereas other sources other than plants rarely emit glucose. On this basis, glucose was adopted here as the plant emission biomarker. Additionally, certain fungal

spores contribute to OC in ambient air; literature reports indicate that particles originating from fungal spores contain abundant mannitol and arabitol, accounting for roughly 13% and 19% of OC, respectively (Bauer et al., 2002, 2008). Accordingly, mannitol and arabitol were chosen as the marker compounds representing fungal spores in this study. Detailed source profile information used in CMB model is provided in Fig. S3 of the supplementary material. It should be noted that the local source profile dataset subdivides major sources—such as coal combustion, vehicle exhaust, and cooking—into multiple categories (for example, industrial versus residential coal burning, gasoline versus diesel exhaust, and regional cooking styles such as Guangdong versus Sichuan). For each major source category, we represented typical source characteristics by using either the mean profile of its subcategories or the average profile derived from multiple observational studies. Similarly, using the proportions of characteristic species in PM_{2.5} from these emission source profiles, the CMB model subsequently estimated each source's contribution to PM_{2.5}. In the primary emission profiles applied here, the PM-to-OC ratio ranged from 1.42 to 2.15.

To assess the contributions of various secondary sources to OC and PM_{2.5}, i.e., SOC and SOA concentrations, we employed tracer-based method, a well-established method for SOC and SOA source apportionment. The tracer-based method operates on principles analogous to those involved in the CMB model, relying on the mass fractions of characteristic tracers in SOC (f_{SOC}) and SOAs (f_{SOA}) from emission sources to ascertain the contributions of different sources. The SOC and SOA tracer-based method was first proposed and utilized by Kleindienst et al. (2007 and 2012). Its specific calculation procedure is as follows:

$$[\text{SOC}] = \frac{\sum_i [\text{tri}_i]}{f_{\text{SOC}}}, \quad (1)$$

$$[\text{SOA}] = \frac{\sum_i [\text{tri}_i]}{f_{\text{SOA}}}, \quad (2)$$

where $\sum_i [\text{tri}_i]$ is the total concentration of the selected tracers in the sample, denoting representative compounds from specific emission categories; f_{SOC} and f_{SOA} are the mass fractions of the tracers in OC and PM_{2.5} from secondary emissions, respectively. The values of f_{SOC} and f_{SOA} are determined according to the chamber experiments conducted, in which gaseous precursors are transformed into SOC and SOAs under oxidative and illuminative conditions. Herein, we employed six compounds associated with hemiterpenes, four compounds related to monoterpenes, as well as β -caryophyllinic acid, 2,3-dihydroxy-4-oxopentanoic acid, and

phthalic acid as marker compounds to indicate the contributions of different biogenic (hemiterpenes, monoterpenes, and sesquiterpenes) and anthropogenic (toluene and naphthalene) sources to SOC and SOA. We adopted the f_{SOC} and f_{SOA} values used by Kleindienst et al. (2007 and 2012). to calculate the SOC and SOA concentrations originating from various sources. All marker selections, f -values, and the conversion coefficients between f_{SOA} and f_{SOC} derived from chamber studies are listed in Table S2, and the calculations assume that the chamber-derived SOC and SOA relationships are representative of the relevant atmospheric oxidation regimes, acknowledging associated uncertainties.

Furthermore, we employed the EC-based method to estimate the total POC and SOC concentrations, enabling a comparison with the POC and SOC concentrations obtained using the tracer-based method. This method uses EC as a tracer for POC and approximates the observed minimum ratio of OC to EC $(\text{OC}/\text{EC})_{\text{min}}$ as an equivalent to the primary emission OC/EC value $(\text{OC}/\text{EC})_{\text{pri}}$ (Turpin and Huntzicker, 1995; Castro et al., 1999). The equation used for estimating SOC is

$$\text{SOC}_{\text{EC-based}} = \text{OC} - \text{EC} \times (\text{OC}/\text{EC})_{\text{min}}, \quad (3)$$

where the minimum OC/EC values observed in spring, summer, autumn, and winter were 2.01, 2.20, 2.15, and 2.31, respectively.

3 Results and discussion

3.1 Carbonaceous components

The OC and EC concentrations were in ranges of 1.01–13.66 (mean: $5.88 \pm 2.57 \mu\text{g m}^{-3}$) and 0.27–4.55 $\mu\text{g m}^{-3}$ (mean: $1.75 \pm 0.88 \mu\text{g m}^{-3}$), respectively. Seasonal variations in the OC and EC concentrations closely mirrored those observed for $\text{PM}_{2.5}$, with elevated levels in autumn and winter and diminished levels in spring and summer (Figure 1a). Owing to its stability and resistance to chemical transformations in the atmosphere, EC is frequently employed as a tracer for primary emissions. The OC/EC ratio is a useful tool for assessing the relative contributions of primary and secondary sources (Castro et al., 1999; Turpin and Huntzicker, 1995). Generally, an OC/EC value above 2 indicates a predominant contribution of secondary sources to OC, whereas values below 2 suggest greater contributions from primary sources (Kunwar and Kawamura, 2014). Here, the majority of the OC/EC values exceeded 2 (Figure 1b), implying that the OC may have been significantly influenced by secondary sources. The OC/EC values were relatively high in summer (4.33 ± 2.07) and winter

(4.18 ± 2.57) compared with those observed in spring (3.36 ± 1.98) and autumn (3.59 ± 1.22). This indicates an increased contribution to OC from secondary sources during the summer and winter.

3.2 Major polar components

3.2.1 Fatty acids

A range of homologous straight-chain C10:0–C32:0 fatty acids (FAs) and unsaturated C16:1 and C18:1 FAs were detected in the PM_{2.5} samples (Figures 1c and 2a). Their distribution exhibited a strong even-carbon-number predominance, as indicated by a carbon preference index (CPI) of 7.59 ± 7.42 (Figure S4a), with peaks at C16:0 and C18:0. The total FA concentration was in a range of 3–658 ng m⁻³, with an average concentration of 197 ± 111 ng m⁻³. Similar molecular distribution patterns and concentrations have been documented in urban aerosol studies across China (59 – $2,090$ ng m⁻³; He et al., 2004 and 2006; Wang et al., 2006; Haque et al., 2019; Fan et al., 2020). In these Chinese cities, FAs are primarily influenced by biogenic sources, including plant releases and biomass burning, as well as fossil fuel combustion and residential cooking. This average concentration generally exceeds those observed for coastal and marine aerosol samples (0.1 – 160 ng m⁻³; Kawamura et al., 2003; Wang et al., 2007; Fu et al., 2011), which are primarily influenced by marine biological activities and long-range transport of continental sources. High-molecular-weight FAs (HFAs, $\geq$ C20:0) are typically derived from the waxes of terrestrial higher plants or biomass burning, whereas low-molecular-weight FAs (LFAs, $<$ C20:0) originate from more diverse sources, including vascular plants, microorganisms, marine phytoplankton, and kitchen emissions (Fu et al., 2008 and 2010). Consequently, the ratio of LFAs to HFAs (LFA/HFA ratio) is an indicator of the relative contributions from various sources. Here, the LFA/HFA range was 0.34 – 10.67 , with an average of 2.45 ± 1.53 (Figure S4b), suggesting that the FAs may have originated from a mixture of natural and anthropogenic sources. The specific FAs C16:0 and C18:0 are associated with different sources. C16:0 is primarily derived from biomass burning or plant emissions, whereas C18:0 predominantly originates from vehicle exhaust, cooking emissions, or road dust (Wang et al., 2007; Fu et al., 2010). Thus, the ratio of C18:0 to C16:0 (C18:0/C16:0) is frequently employed to assess the FA source. A C18:0/C16:0 ratio below 0.25 suggests a primary contribution from biomass burning or plant emissions, whereas ratios between 0.25 and 0.5 indicate a predominant influence from vehicle exhaust. C18:0/C16:0 ratios within a range of 0.5 – 1 suggest a significant contribution from cooking emissions or road dust (Rogge et

al., 2006). Here, the C18:0/C16:0 ratio varied from 0.05 to 9.53, with an average value of 0.59 ± 0.88 (Figure S3c), indicating that the FAs in Nanchang originated from a mixture of sources.

The average concentrations of unsaturated FAs, specifically C16:1 and C18:1, were 7 ± 4 and 4 ± 3 ng m^{-3} , respectively. Such unsaturated FAs are primarily derived from biomass burning, plant leaf emissions, cooking activities, and release from marine organisms (Kawamura and Gagosian, 1987; Rogge et al., 1993b; Nolte et al., 1999). Upon emission into the atmosphere, unsaturated FAs undergo rapid oxidation by ozone (O_3), hydrogen peroxide (H_2O_2), or hydroxyl (OH) radicals. Consequently, the ratio of unsaturated FAs to saturated FAs, represented as $(\text{C16:1} + \text{C18:1})/(\text{C16:0} + \text{C18:0})$, is an indicator of the reactivity and degree of aging of unsaturated FAs (Rudich et al., 2007; Kawamura and Gagosian, 1987). Here, this ratio was 0.12 ± 0.06 (Figure S3d), indicating that the unsaturated FAs underwent significant photochemical degradation and that secondary organic compounds may be common in $\text{PM}_{2.5}$.

The average concentrations of most FAs were significantly ($p < 0.05$) higher in autumn and winter than in spring and summer (Figure 2a), indicating a substantial increase in fatty acid emissions during autumn and winter. The LFA/HFA and C18:0/C16:0 ratios were lower in autumn and winter than in summer (Figures S4b and S4c). This indicated that in autumn and winter, FAs were significantly influenced by terrestrial plant or biomass burning, whereas in summer, they were more influenced by vehicular emissions, marine phytoplankton, and cooking emissions. Conversely, the $(\text{C16:1} + \text{C18:1})/(\text{C16:0} + \text{C18:0})$ values were higher in winter than in summer (Figure S4d). This suggested that unsaturated FAs experienced a greater degree of photochemical degradation during the warm seasons. Unsaturated FAs can be rapidly oxidized by ozone or OH radicals. The higher temperatures, relatively stronger solar radiation, and higher concentrations of O_3 and OH radicals in the summer season will facilitate the oxidative decomposition of unsaturated FAs (Wang et al., 2006; Fan et al., 2020).

3.2.2 Fatty alcohols

Several straight-chain n-alkanols (C14–C32) were detected (Figures 1c and 2b), exhibiting a predominance of even carbon numbers, as indicated by a CPI of 10.28 ± 17.05 (Figure S4e). Low-molecular-weight alcohols ($\text{LMW}_{\text{alc}}, \leq \text{C20}$) exhibited peaks at C16 and C18, and high-MW alcohols ($\text{HMW}_{\text{alc}}, > \text{C20}$) exhibited peaks at C26, C28, and C30 (Figure 2b). The total concentration of n-alkanols was in a range of 6–572 ng m^{-3} (average

= $114 \pm 93 \text{ ng m}^{-3}$). This concentration fell within the range of reported n-alkanol concentrations in urban aerosols in China ($3\text{--}1,301 \text{ ng m}^{-3}$; Wang et al., 2006; Haque et al., 2019; Fan et al., 2020) and generally exceeded that of coastal and marine aerosol samples ($0.1\text{--}19.7 \text{ ng m}^{-3}$; Kawamura et al., 2003; Fu et al., 2011). In urban aerosols of China, fatty alcohols are primarily derived from vegetation releases, biomass burning, and resuspension of soil particles. The fatty alcohols in coastal and marine aerosols are primarily attributed to the long-range transport of terrestrial soil and biomass burning as well as marine biogenic sources. HMW_{alc} primarily originates from higher plant leaf waxes, loess deposits, or biomass burning, and LMW_{alc} primarily originates from soil and marine microorganisms (Simoneit, 2002; Kawamura et al., 2003). Here, the LMW_{alc} concentration ($75 \pm 61 \text{ ng m}^{-3}$) exceeded that of HMW_{alc} ($39 \pm 34 \text{ ng m}^{-3}$), yielding an average LMW/HMW ratio of 2.50 ± 1.73 (Figure S4f), indicating a mixture of n-alkanols from soil, marine organisms, and biomass burning.

The average concentrations of most fatty alcohols were significantly ($p < 0.05$) higher in autumn and winter than in summer (Figure 2b), indicating a substantial increase in fatty alcohols emissions during autumn and winter. The LMW/HMW values were relatively low in autumn and winter and higher in spring and summer (Figure S4f). This suggests an increased contribution of soil and marine organisms to n-alkanols in spring and summer. Contributions from plant or biomass burning increased in autumn and winter. In Chinese cities, rising temperatures in spring increase soil microbial activity, with marine air masses exerting a stronger influence during summer, potentially increasing emissions from soil and marine organisms in these seasons. In contrast, autumn and winter coincide with the crop harvest and increased winter heating, leading to likely increases in biomass burning emissions during these periods (Wang et al., 2006; Haque et al., 2019; Fan et al., 2020).

3.2.3 Saccharides

We investigated three saccharide classes: anhydrosugars, primary sugars, and sugar alcohols. Among the anhydrosugars, levoglucosan had the highest concentration ($102 \pm 79 \text{ ng m}^{-3}$). We observed markedly lower concentrations of mannosan ($4 \pm 4 \text{ ng m}^{-3}$) and galactosan ($2 \pm 3 \text{ ng m}^{-3}$) (Figure 2c). Levoglucosan—a prominent biomarker for biomass burning—has been extensively investigated in urban aerosol samples, with concentrations in a range of $22\text{--}2,706 \text{ ng m}^{-3}$ (Wang and Kawamura, 2005; Wang et al., 2006; Fu et al., 2010; Haque et al., 2019; Fan et al., 2020), indicating the significant influence of biomass burning on the urban

atmosphere. The widespread detection of levoglucosan in various environments—including suburban (10–482
ng m⁻³; Yttri et al., 2007; Fu et al., 2008), marine (0.2–30 ng m⁻³; Simoneit et al., 2004a; Fu et al., 2011), and
polar regions (0–3 ng m⁻³; Stohl et al., 2007; Fu et al., 2009)—indicates its potential for long-range
atmospheric transport. Compared to those of the major polar organic compounds found in marine
environments, such as FAs, fatty alcohols, and saccharides, the concentrations of these substances in urban
PM_{2.5} are substantially higher. This is primarily attributed to the substantial impact of anthropogenic activities
in urban environments—such as vehicle emissions, industrial processes, and residential heating and cooking—
all of which can release these major polar components (Wang et al., 2006; Fu et al., 2008; Haque et al., 2019;
Fan et al., 2020). In contrast, the organic matter in marine PM_{2.5} originates from relatively simpler sources with
lower emissions, mainly influenced by natural processes, including marine biological activity and the long-
range transport of continental pollutants (Kawamura et al., 2003; Simoneit et al., 2004a; Wang et al., 2007; Fu
et al., 2011).

Seasonal variations of anhydrosugars exhibited consistent patterns, with elevated concentrations in autumn
and winter and reduced levels in spring and summer (Figure 2c). This indicated increased biomass burning
during the cold months. The sources were characterized using levoglucosan-to-mannosan (L/M) and
mannosan-to-galactosan (M/G) ratios. Research suggests distinct L/M ranges for different biomass burning
sources: softwood (3–10), hardwood (13–35), and agricultural crop burning (40–56) (Oros and Simoneit, 2001;
Sheesley et al., 2003; Engling et al., 2014). Coal combustion can produce significant amounts of levoglucosan
(Sheesley et al., 2003; Fabbri et al., 2009), typically with higher L/M values (12–189) because of the lower
cellulose content compared with that of plant materials (Rybicki et al., 2020a and 2020b). Conversely, M/G
values are generally higher for softwood and hardwood combustion at 3.6–7.0 and 1.2–2.0, respectively.
However, M/G values are relatively low for agricultural crop burning (0.3–0.6) (Oros and Simoneit, 2001;
Sheesley et al., 2003). Here, most samples exhibited relatively high L/M ratios (4.63–466.61, average = 40.22
± 49.82) and relatively low M/G ratios (0.30–9.41, average = 2.34 ± 1.36). The L/M and M/G ratios were
similar to those of previous observations in the urban area of Nanchang (L/M = 7.9–359.1; average = 59.9;
Zhu et al., 2022) and fell within the ranges for crop residue burning and coal combustion. This suggests that
the dehydrosugars in the PM_{2.5} of Nanchang were likely primarily derived from crop residue burning and coal

combustion. In autumn and winter, the contributions of crop residue burning and coal combustion were more prominent, as evidenced by the higher L/M ratios and lower M/G ratios observed compared with those in spring and summer (Figures S4g and S4h).

Primary sugars and sugar alcohols serve as tracers for primary biological aerosol particles, originating from biogenic emissions, including the release of microorganisms, plants, and flowers, as well as the resuspension of surface soils and unpaved road dust containing biological materials (Graham et al., 2003; Simoneit et al., 2004b; Yttri et al., 2007). Furthermore, biomass burning is a notable source of primary sugars and sugar alcohols (Fu et al., 2012). We identified nine primary sugars (ribose, fructose, galactose, glucose, sucrose, lactulose, maltose, turanose, and trehalose; Figure 2c). Glucose ($15 \pm 7 \text{ ng m}^{-3}$) and sucrose ($15 \pm 6 \text{ ng m}^{-3}$) exhibited the highest concentrations, followed by fructose ($11 \pm 5 \text{ ng m}^{-3}$), trehalose ($8 \pm 4 \text{ ng m}^{-3}$), and maltose ($5 \pm 3 \text{ ng m}^{-3}$). The concentrations of galactose ($2 \pm 1 \text{ ng m}^{-3}$), ribose ($3 \pm 2 \text{ ng m}^{-3}$), turanose ($1 \pm 1 \text{ ng m}^{-3}$), and lactulose ($1 \pm 1 \text{ ng m}^{-3}$) were comparatively low. We detected four sugar alcohols (mannitol, arabitol, pinitol, and inositol). Mannitol ($10 \pm 5 \text{ ng m}^{-3}$) exhibited the highest concentration, followed by pinitol ($6 \pm 4 \text{ ng m}^{-3}$) and arabitol ($5 \pm 3 \text{ ng m}^{-3}$). The concentration of inositol ($2 \pm 2 \text{ ng m}^{-3}$) was low.

Levoglucosan serves as a specific biomarker for biomass burning, enabling the assessment of biomass burning contributions to sugar compounds through the ratio of levoglucosan to OC (Lev/OC) and the proportion of levoglucosan in the total sugar compounds (Lev%). Typically, a Lev/OC ratio of >0.048 and a Lev% of $>68\%$ indicate a dominance of biomass burning. Conversely, values below these thresholds suggest that in addition to biomass burning, other sources, such as biogenic emissions, significantly contribute to the presence of sugar compounds (Yan et al., 2019 and references therein). Here, the Lev% and Lev/OC values were $53.57\% \pm 17.93\%$ and 0.016 ± 0.009 (Figures S4i and S4j), respectively, indicating that the sugar compounds were influenced by biomass burning and biogenic emissions, particularly the primary sugars and sugar alcohols. Primary sugars, such as glucose and trehalose, as well as the mannitol in sugar alcohol, positively correlated with levoglucosan (Figure S5a). This confirmed that biomass burning may be a potential source of primary sugars and sugar alcohols. However, the correlation was not statistically significant, suggesting that biological sources remain an important contributor to these compounds in Nanchang. The Lev/OC and Lev% values were higher in autumn and winter than in spring and summer (Figures S4i and S4j),

further indicating that biomass burning contributions increased during the cold months. As previously noted, the primary reason for increased biomass burning emissions during the cold season is that autumn marks the crop harvest in China, leading to more straw burning. Furthermore, the temperature decrease in winter results in additional biomass burning for heating purposes.

3.3 Minor polar components

3.3.1 Lignin, resin products, and sterols

Here, we detected four types of lignin and resin-derived compounds: 4-hydroxybenzoic acid, dehydroabietic acid, vanillic acid, and syringic acid (Figures 1c and 3a). These are generally considered to originate from natural plant release and burning. The burning of coniferous trees, such as pine, releases higher concentrations of dehydroabietic acid than other lignin and resin-derived compounds (Simoneit, 2002). Here, the concentrations of 4-hydroxybenzoic acid ($5 \pm 4 \text{ ng m}^{-3}$) and dehydroabietic acid ($3 \pm 2 \text{ ng m}^{-3}$) were high, whereas those of vanillic acid ($2 \pm 1 \text{ ng m}^{-3}$) and syringic acid ($2 \pm 1 \text{ ng m}^{-3}$) were relatively low. A significant positive correlation was observed between the total concentration of lignin and resin-derived compounds and levoglucosan ($r = 0.80$, $p < 0.01$, Figure S5a), suggesting that biomass burning was a potential source of lignin and resin-derived compounds. The concentration of dehydroabietic acid was significantly lower than that of 4-hydroxybenzoic acid, similar to the pattern observed for Nanjing aerosols (Haque et al., 2019). This implies that coniferous tree combustion may not be the primary source of lignin and resin acids in this region.

We identified several sterols (cholesterol, stigmasterol, and β -sitosterol; average concentrations = 5 ± 3 , 5 ± 4 , and $10 \pm 8 \text{ ng m}^{-3}$, respectively; Figure 3a). These sterols originate from distinct sources. Cholesterol, an animal-derived sterol, is primarily associated with meat cooking in the atmosphere (Rogge et al., 1991; Nolte et al., 1999; He et al., 2004). Contrarily, stigmasterol and β -sitosterol are plant-derived sterols, typically originating from plant leaves, cooking, or biomass burning (Nolte et al., 2001; He et al., 2004; Zhao et al., 2007). Here, stigmasterol and β -sitosterol exhibited a significant positive correlation with levoglucosan (Figure S5a). This suggests that biomass burning may be a potential source of plant-derived sterols. The concentrations of stigmasterol and β -sitosterol were higher in autumn and winter than in spring and summer, indicating an increased contribution of biomass burning to sterols in autumn and winter.

3.3.2 Glycerol, hydroxy acids, and aromatic acids

Glycerol and three hydroxy acids, glycolic acid, malic acid, and citric acid, were detected in the PM_{2.5} samples (Figures 1c and 3b). The concentration of glycerol was relatively high ($10 \pm 4 \text{ ng m}^{-3}$), whereas the three hydroxy acids exhibited comparable and relatively low concentrations: glycolic acid ($7 \pm 2 \text{ ng m}^{-3}$), malic acid ($7 \pm 2 \text{ ng m}^{-3}$), and citric acid ($6 \pm 2 \text{ ng m}^{-3}$). These concentrations aligned with reported ranges for urban aerosols (2–146 and 1–180 ng m^{-3} for glycerol and hydroxy acids, respectively; Wang et al., 2006; Fu et al., 2010; Haque et al., 2019; Fan et al., 2020). The source of glycerol in urban aerosols is similar to that of sugar alcohols, primarily originating from the emission from plants, the resuspension of surface soils and unpaved road dust containing biological materials, and biomass burning (Simoneit et al., 2004a and 2004b; Fu et al., 2008), whereas glycolic, malic, and citric acids mainly originate from the secondary photooxidation of organic compounds in the atmosphere, such as biogenic unsaturated FAs, cyclic olefins, and isoprene (Kawamura and Ikushima, 1993; Kawamura and Sakaguchi, 1999; Claeys et al., 2004, Haque et al., 2019). No significant correlation was observed between glycerol and the hydroxy acids; however, a notable correlation was observed among the hydroxy acids ($r = 0.49\text{--}0.66$, $p < 0.01$, Figure S5b). This confirmed the divergent sources of glycerol and hydroxy acids and similar origins for the hydroxy acids. In summer, the concentrations of the hydroxy acids exceeded those in other seasons (Figure 3b). A positive correlation was observed between the polyacids and temperature ($r = 0.35\text{--}0.51$, $p < 0.01$), suggesting enhanced secondary photochemical oxidation in the warm months. The secondary formation of hydroxy acids in the atmosphere is suggested to be related to the ozone-driven oxidation of organic compounds. Elevated ozone levels and stronger solar radiation during the summer season create more favorable conditions for the secondary production of hydroxy acids (Kawamura and Ikushima, 1993; Kawamura and Sakaguchi, 1999; Claeys et al., 2004).

We detected four aromatic acids in the PM_{2.5} samples: benzoic acid, phthalic acid, isophthalic acid, and terephthalic acid at concentrations of 0.8 ± 0.3 , 6.4 ± 4.1 , 1.1 ± 0.7 , and $10.1 \pm 3.1 \text{ ng m}^{-3}$, respectively (Figure 3b). Aromatic acids significantly contribute to the formation of atmospheric particles. Benzoic acid is recognized as a major pollutant in vehicle exhaust and the secondary photochemical oxidation product of aromatic hydrocarbons from vehicle emissions (Kawamura and Kaplan, 1987; Rogge et al., 1993a; Kawamura et al., 2000). Phthalic acid typically originates from the secondary transformation of PAHs, including naphthalene and other PAHs. Terephthalic acid is primarily produced through terephthalate hydrolysis during

the combustion of urban plastics (Fu et al., 2010; Haque et al., 2019). Here, phthalic and terephthalic acids jointly accounted for approximately 90% of the total detected aromatic acids. A significant positive correlation was observed between phthalic acid and PAHs ($r = 0.83$, $p < 0.01$), as well as between terephthalic acid and the total phthalates ($r = 0.72$, $p < 0.01$). The PAH and phthalate data have been published by Guo et al. (2024a and 2004b). This suggests that the secondary conversion of PAHs and the hydrolysis of phthalates (specifically terephthalate) during plastic burning significantly contribute to aromatic acid levels. The phthalic acid concentrations were higher in autumn and winter, whereas terephthalic acid levels peaked in spring and summer (Figure 3b). This indicates that the secondary transformation of PAHs was more pronounced during the cold months, whereas high temperatures in spring and summer promoted the volatilization and transformation of phthalates from plastic sources. Phthalates are employed as plasticizers in resins and polymers. Because they are not chemically bound to the polymer matrix, they can be easily released into the air through evaporation. Higher ambient temperatures in spring and summer favor phthalate release from the matrix (Wang et al., 2006; Fu et al., 2010).

3.4 SOA tracers in PM_{2.5}

3.4.1 Biogenic SOA tracers

Biogenic and anthropogenic SOAs are critical in influencing the atmospheric radiation balance and regional air quality (Ding et al., 2014). Biogenic SOA tracers include oxidation products from isoprene, monoterpenes, sesquiterpenes, and other oxygenated hydrocarbons. Isoprene is the dominant component of BVOC emissions, accounting for ~45% of the total emissions (annual release estimated at 600 Tg C) (Piccot et al., 1992; Guenther et al., 1995; Sharkey et al., 2008). Here, we detected six isoprene oxidation products (2-methylglyceric acid (2-MGA), three C5-alkene triols, and two 2-methyltetrols (MTLs) at concentrations of 1.8 ± 1.1 , 2.8 ± 1.9 , and 4.1 ± 3.3 ng m⁻³, respectively; Figure 4). The concentrations of the C5-alkene triols and MTLs exceeded that of 2-MGA. Generally, These isoprene-derived oxidation products are formed via acid-catalyzed ring-opening reactions and isomerization of isoprene epoxydiols (IEPOX), a reactive intermediate produced by the gas-phase oxidation of biogenic isoprene emitted by vegetation (Surratt et al., 2006 and 2010; Fu et al., 2010 and 2014; Shrivastava et al., 2017). Throughout the sampling period, a strong linear correlation was observed between the MTLs and C5-alkene triols ($R^2 = 0.82$, $P < 0.01$), indicating their

430 similar sources. Both species are sensitive to aerosol acidity and NO_x levels, with lower aerosol pH and
431 reduced NO_x concentrations favoring their formation (Surratt et al., 2010; Lin et al., 2013). Nevertheless,
432 certain differences were observed in the formation processes of C5-alkene triols and MTLs. For example, some
433 studies have reported that, compared to MTLs, C5-alkene triols are more susceptible to degradation under high
434 temperature and intense solar radiation, resulting in reduced C5/MTLs ratios in summer versus winter (Yee et
435 al., 2020; Haque et al., 2023; Zhang et al., 2024). A similar seasonal decrease in the C5/MTLs ratio is observed
436 in summer in our dataset (Figure S6a). C5-alkene triols and MTLs are believed to form via isoprene oxidation
437 under low-NO_x conditions. Contrarily, high concentrations of NO_x favor further isoprene oxidation to yield 2-
438 MGA (Lin et al., 2013; Nguyen et al., 2015). Here, the 2-MGA/MTL ratios in winter exceeded those observed
439 in other seasons (Figure S6b), indicating that elevated NO_x concentrations in winter (Figure S2) may enhance
440 2-MGA formation. In MTLs, 2-methylthreitol and 2-methylerythritol (MTL1 and MTL2) exhibited a
441 significant linear correlation ($R^2 = 0.68$, $P < 0.01$), indicating that they may originate from similar sources
442 and/or share similar formation pathways. The ratio of MTL2 to MTLs (MTL2/MTL ratio) can reflect the
443 transformation pathways of MTLs. The ratio of MTL2/MTLs = 0.35, 0.61, 0.76, and 0.90 corresponded to
444 isoprene secondary transformation from biogenic sources, OH-rich conditions (low NO_x), NO_x-rich conditions,
445 and liquid-phase oxidation by H₂O₂, respectively (Kleindienst et al., 2009; Nozière et al., 2011). The results
446 showed that the MTL2/MTL values in spring (0.64 ± 0.12) and summer (0.59 ± 0.11) were relatively low and
447 closer to the values for biogenic sources and OH-rich secondary transformation. However, the values in
448 autumn and (0.66 ± 0.10) winter (0.69 ± 0.06) were relatively high, approaching those for NO_x-rich secondary
449 transformation and liquid-phase oxidation by H₂O₂ (Figure S6c). This indicates that in spring and summer,
450 MTLs were more likely derived from biogenic sources and OH-rich secondary transformation, whereas in
451 autumn and winter, a shift was observed toward NO_x-rich secondary transformations and liquid-phase
452 oxidation by H₂O₂. This relatively high C5/MTL, 2-MGA/MTL, or MTL2/MTL ratio observed during the cold
453 season compared to that in the warm season is also commonly reported in urban environments across China,
454 including Beijing (Liu et al., 2023), Tianjin (Wang et al., 2019; Fan et al., 2019), Nanjing (Yang et al., 2022),
455 Shanghai (Yang et al., 2022), and Guangzhou (Yuan et al., 2018). Notably, the C5-alkene triols detected in our
456 study predominantly exist as double-bonded “triol” compounds, including cis-2-methyl-1,3,4-trihydroxy-1-

butene, 3-methyl-2,3,4-trihydroxy-1-butene, and trans-2-methyl-1,3,4-trihydroxy-1-butene. In fact, acid-catalyzed ring-opening reactions and isomerization of particle-phase IEPOX can also produce various “diol” compounds, which typically exist as cyclic structures, such as trans-3-methyltetrahydrofuran-3,4-diol and cis-3-methyltetrahydrofuran-3,4-diol. These “diols” are also important components of C5-alkene triols (Li et al., 2013; Frauenheim et al., 2022). Recent work, however, indicates that C5-alkene triols detected via GC–MS are unlikely to originate solely from the acid-catalyzed ring-opening reactions and isomerization of IEPOX; instead, they may largely be artifacts produced by thermal decomposition during GC–MS analysis, with roughly 90% of the detected “triol” signal attributable to such artifacts (Frauenheim et al., 2022). If such artifacts indeed exist in the GC–MS measurements, it implies that the C5-alkene triol levels reported in our results are overestimated, potentially leading to an overall overestimation of isoprene-derived SOA tracers by approximately 30%. Despite the potential artifact issue in GC–MS detection, the contribution of C5-alkene triols to SOA should not be underestimated (Frauenheim et al., 2022 and 2024).

Monoterpenes represent a crucial component of BVOC emissions, accounting for ~11% of the annual emissions, with an annual emission of 110 Tg C (Guenther et al., 1995). Here, we identified four oxidation products of monoterpene: pinonic acid (PNA); pinic acid (PA); 3-methyl-1,2,3-butanetricarboxylic acid (MBTCA); and 3-hydroxyglutaric acid (3-HGA), at concentrations of 1.7 ± 1.0 , 1.5 ± 0.9 , 0.8 ± 0.6 , and 0.8 ± 0.5 ng m⁻³, respectively (Figure 4). The concentrations of PNA and PA exceeded those of MBTCA and 3-HGA. PNA and PA are produced via the oxidation of α/β -pinene via reactions with O₃ and OH radicals, and the α/β -pinene detected in the aerosol samples is primarily derived from biomass burning and higher plant release (Hallquist et al., 2009; Eddingsaas et al., 2012; Zhang et al., 2015; Iyer et al., 2021). The predominance of PNA over PA is attributable to its relatively high vapor pressure, consistent with previous findings (Fu et al., 2008 and 2010). PNA and PA are recognized as first-generation oxidation products of α/β -pinene, whereas MBTCA represents a second-generation oxidation product formed via the further photooxidation of PNA and PA with OH radicals. The relative concentrations of such first- and second-generation oxidation products (M/P) reflect the oxidation degree and aging status of monoterpene compounds (Ding et al., 2014). Here, the M/P values were higher in spring (0.25 ± 0.11) and summer (0.32 ± 0.10) than in autumn (0.16 ± 0.09) and winter (0.24 ± 0.10). This suggests that higher temperatures and/or stronger radiation in spring and summer

promoted the oxidation of monoterpene compounds (Figure S6d). The formation of 3-HGA is considered to occur via a ring-opening mechanism, probably linked to heterogeneous reactions of monoterpenes with irradiation in NO_x-rich environments (Jaoui et al., 2005; Claeys et al., 2007). The ratio of MBTCA to 3-HGA (MBTCA/3-HGA) can be employed to distinguish monoterpenes as the secondary transformation of α -pinene yields MBTCA at significantly higher rates relative to 3-HGA compared with β -pinene (Jaoui et al., 2005). Here, the annual MBTCA/3-HGA value was 1.18 ± 0.65 (Figure S6e), close to those observed in urban environments in the United States (0.81 ± 0.32) and other cities in China (0.68 ± 0.65) (Lewandowski et al., 2013; Ding et al., 2014). This suggests that monoterpenes in these urban environments have similar sources. The MBTCA/3-HGA value was higher in spring (1.31 ± 0.66) and summer (1.45 ± 0.70) than in autumn (1.09 ± 0.72) and winter (0.88 ± 0.32) (Figure S6e). This suggests that the contribution of α -pinene to monoterpene was higher in spring and summer than in autumn and winter. This phenomenon can be attributed to the association of 3-HGA formation with NO_x-rich environments, where the relatively high NO_x concentrations in autumn and winter (Figure S2) may enhance 3-HGA production. The ratio of the total isoprene to monoterpene oxidation products (Iso/Pine) was lower in autumn and winter than in summer (Figure S6f). This pattern is also commonly observed in other urban environments of China, suggesting that the higher temperatures in summer favored isoprene release (Li et al., 2018; Fan et al., 2020; Liu et al., 2023).

BVOC emissions include a class of sesquiterpene compounds, among which β -caryophyllene is the most abundant and frequently reported (Ding et al., 2014; Fan et al., 2020). β -caryophyllinic acid, a product of the ozonolysis or photooxidation of β -caryophyllene, predominantly originates from biomass burning and natural plant emissions, including those from pine and birch trees (Helmig et al., 2006; Duhl et al., 2008). Here, β -caryophyllinic acid was detected at a concentration of $2.7 \pm 1.9 \text{ ng m}^{-3}$. The β -caryophyllinic acid concentrations in autumn ($3.2 \pm 1.8 \text{ ng m}^{-3}$) and winter ($4.6 \pm 1.6 \text{ ng m}^{-3}$) exceeded those observed in spring ($1.5 \pm 0.7 \text{ ng m}^{-3}$) and summer ($1.1 \pm 0.5 \text{ ng m}^{-3}$) (Figure 4). The higher concentrations of caryophyllinic acid in autumn and winter may be associated with increased biomass burning and subsequent secondary transformation of β -caryophyllene. During the sampling period, a significant positive linear correlation was observed between β -caryophyllinic acid and levoglucosan, supporting this inference ($R^2 = 0.66$, $P < 0.01$).

3.4.2 Anthropogenic SOA tracers

Anthropogenic SOA tracers include hydroxy and aromatic acids, which primarily originate from the secondary transformation of AVOCs. Although global AVOC emissions are relatively modest at 110 Tg C yr^{-1} (Piccot et al., 1992) compared with BVOC emissions, which reach $1,150 \text{ Tg C yr}^{-1}$ (Guenther et al., 1995), the contribution of anthropogenic sources to SOA is frequently more pronounced in urban environments because of the impact of human activities (von Schneidemesser et al., 2010; Ding et al., 2012; Li et al., 2019). AVOC emissions in urban settings can enhance BVOC oxidation, promoting SOA formation (Carlton et al., 2010; Hoyle et al., 2011). Here, we identified two primary anthropogenic SOA tracers, 2,3-dihydroxy-4-oxopentanoic acid (DHOPA) and phthalic acid, at concentrations of 1.7 ± 1.1 and $6.4 \pm 4.1 \text{ ng m}^{-3}$, respectively (Figures 3 and 4). DHOPA and phthalic acid are recognized as important markers for anthropogenic SOAs, produced via the oxidation of toluene and PAHs, such as naphthalene (Kawamura and Ikushima, 1993; Kleindienst et al., 2007 and 2012). Their concentrations were higher in autumn and winter than in spring and summer (Figures 3 and 4). This indicated a marked increase in the contribution of anthropogenic sources to SOA during the cold months. As typical anthropogenic SOA tracers, DHOPA and phthalic acid have been reported to exhibit significantly higher concentrations in urban areas than in suburban and remote regions, with higher levels in the cold season than in the warm season in China (Ding et al., 2014; Fan et al., 2019; Yang et al., 2022). SOAs in urban environments and in the cold season in China are more strongly influenced by anthropogenic emissions, such as those from fossil fuel combustion and biomass burning (Ding et al., 2012; Yuan et al., 2018; Fan et al., 2019).

3.5 Source apportionment of OC and aerosol

Employing the CMB model and tracer-based approach, we quantified the contributions of POC and SOC to the total OC (Figure 5). The results showed that the POC and SOC concentrations were 3.66 ± 2.47 and $0.50 \pm 0.32 \text{ } \mu\text{g m}^{-3}$ (Figures 5a–5c), accounting for $57.66\% \pm 18.35\%$ and $8.25\% \pm 3.23\%$ of the measured OC, respectively (Figure 5f). Our results indicate that POC continues to be the main contributor to OC. These results were within the reported ranges of POC (5%–76%) and SOC (3%–56%) (Stone et al., 2009; Guo et al., 2012; Fan et al., 2020; Xu et al., 2021; Haque et al., 2023). Notably, anthropogenic sources, including biomass burning, coal combustion, motor vehicle emissions, and cooking, contributed the majority of POC ($3.35 \pm 2.38 \text{ } \mu\text{g m}^{-3}$, accounting for $88.93\% \pm 6.88\%$). Natural sources, such as plant debris and fungal spores, contributed

relatively little ($0.32 \pm 0.17 \mu\text{g m}^{-3}$, accounting for $11.07\% \pm 6.88\%$). For SOC, anthropogenic contributions ($0.33 \pm 0.25 \mu\text{g m}^{-3}$, accounting for $60.67\% \pm 21.29\%$), including toluene and naphthalene, exceeded biogenic contributions ($0.17 \pm 0.09 \mu\text{g m}^{-3}$, accounting for $39.33\% \pm 21.15\%$), which included isoprene, α/β -pinene, and β -caryophyllene. The relative abundances of anthropogenic POC and SOC were higher in autumn and winter, whereas biogenic POC and SOC exhibited greater relative abundances during spring and summer (Figures 5d–5f). These results underscore the significant influence of biogenic emissions during the warm months, whereas anthropogenic emissions exert a pronounced effect during the cold months. This situation is primarily attributed to the elevated temperatures, and increased solar radiation during the warm months facilitates VOC release from vegetation, thereby promoting biogenic POC emissions and SOC formation. Contrarily, the rise in anthropogenic POC emissions during the cold months fosters the development of anthropogenic SOC in the atmosphere. During winter, including both pollution episodes (Winter-P) and the remainder period (Winter-R), the contribution proportions of three main anthropogenic POC sources (biomass burning, coal combustion, and vehicle exhaust, 55%–62%) and anthropogenic SOC precursors (toluene and naphthalene, 6%–7%) were consistently higher than the annual averages (40% for these anthropogenic POC and 5% for anthropogenic SOC) (Figure 5f). This trend was even more pronounced during individual pollution episodes (Figure S7), with these anthropogenic POC contributions reaching 51%–71% and anthropogenic SOC contributions increasing to 6%–8% (Figure S7f). These findings suggest that POC from biomass burning, coal combustion, and vehicle exhaust, along with SOC originating from toluene and naphthalene, may be key drivers of wintertime organic aerosol pollution.

The fraction of OC that remains unexplained by the CMB model and tracer-based methods and is therefore classified as “other OC”. This component is defined as the difference between measured OC and the summed contributions of POC and SOC from all apportioned sources estimated by the CMB model and tracer approaches. On average, other OC accounts for ~34% of measured OC (Figures. 5c, 5f). The presence of other OC likely reflects limitations in source identification: some OC forms (e.g., liquid-phase OC or aged primary OC) are difficult to capture with either the CMB model or tracer techniques (Ding et al., 2014; Fan et al., 2019; Wu et al., 2020; Xu et al., 2021). Although absolute concentrations of other OC are lower in spring and summer than in autumn and winter, its fractional contribution shows the opposite seasonal pattern—about

45%–50% in spring and summer versus 12%–29% in autumn and winter (Figures. 5c, 5f). Similar patterns have been reported elsewhere in other cities of China, e.g., Beijing, where estimated concentrations of other OC are higher in winter but its proportion is greater in summer (~44% vs. 22–25% in winter) (Wang et al., 2009; Wu et al., 2020; Xu et al., 2021). Previous studies indicate that this “other OC” is likely dominated by SOC (Guo et al., 2012; Wu et al., 2020; Xu et al., 2021). To assess whether other OC is predominantly primary or secondary, we used an EC-based method to estimate total POC and SOC; unlike tracer methods, the EC approach simply partitions OC into POC and SOC and therefore can include unapportioned components. Accordingly, the POC calculated via EC-based method minus the POC calculated via CMB model yields the “other POC” (unapportioned POC), and similarly, SOC calculated via EC-based method minus the tracer-based SOC yields the “other SOC” (Figure. S8). Our results show that, across all four seasons, the concentration of other SOC exceeds that of other POC (Figures. S8a, S8b), indicating that these unidentified other OC components are likely predominantly SOC. The concentration of other SOC is substantially higher in autumn and winter—especially during pollution periods in winter—highlighting the critical role of SOC in aerosol pollution during these seasons (Figures. S8c, S8d). Conversely, the proportion of other SOC during summer is the highest among all seasons (around 32%) (Figure. S8b), suggesting that elevated temperatures and intense radiation during summer enhance SOC formation efficiency (Wu et al., 2020; Xu et al., 2021). Although the other SOC fraction exceeds other POC, EC-based estimates of total SOC remain smaller than total POC in all seasons. Similarly, SOC estimated by the tracer method is markedly lower than POC apportioned by the CMB model. These results together indicate that primary emissions play a significant role in urban OC pollution, with substantial POC contributions forming the basis for elevated OC concentrations. However, the influence of SOC should not be overlooked. Numerous studies have shown that even minor increases in SOC during winter pollution episodes can exacerbate OA pollution (Li et al., 2019; Xu et al., 2021; Haque et al., 2023).

Using the CMB model and tracer-based methods, we also quantified the contributions of POA and SOA to $\text{PM}_{2.5}$ (Figure S9). The findings indicated that POA and SOA contributed 6.86 ± 4.55 and $1.02 \pm 0.63 \mu\text{g m}^{-3}$, accounting for $22.84\% \pm 9.71\%$ and $3.47\% \pm 1.41\%$ of the observed $\text{PM}_{2.5}$ mass in Nanchang, respectively. The anthropogenic POA ($6.26 \pm 4.37 \mu\text{g m}^{-3}$, accounting for $89.38\% \pm 8.26\%$) and SOA ($0.65 \pm 0.48 \mu\text{g m}^{-3}$,

accounting for $58.79\% \pm 21.30\%$) exceeded the natural POA ($0.60 \pm 0.33 \mu\text{g m}^{-3}$, accounting for $10.34\% \pm 6.83\%$) and SOA ($0.37 \pm 0.20 \mu\text{g m}^{-3}$, accounting for $41.21\% \pm 21.30\%$), respectively. The POA and SOA exhibited patterns analogous to those of POC and SOC, with greater contributions recorded in autumn and winter compared with spring and summer.

The calculation of POC and SOC contributions based on the CMB model and tracer-based method inherently involves uncertainties. For the CMB model, a primary source of uncertainty is the variability of tracer mass fractions within OC (f_{OC}) from the same primary emission sources across various observational studies. For example, reported levoglucosan/OC fractions range from 12% (Andreae, et al., 2001, 2019) to 13% (Zheng et al., 2002) and 16% (Fine et al., 2004), whereas the local source profile we reference suggests 8.3% (Zhang et al., 2007). Since recent, site-specific emission data are often unavailable, the choice of source profile and corresponding f_{OC} values can strongly influence apportionment outcomes. Secondly, a fundamental assumption of the CMB model is that the selected markers representing various sources remain stable during atmospheric transport and do not undergo significant chemical transformation. However, truly conservative species are rare. For example, levoglucosan, commonly used as a biomass-burning marker and generally considered relatively stable (Wan et al., 2021), has nonetheless been shown to decrease during long-range transport through wet and dry deposition (Fu et al., 2011) and by photodegradation (Holmes and Petrucci, 2006; Stohl et al., 2007; Hoffmann et al., 2010). Analogous loss or alteration of other source tracers can bias CMB results, tending to underestimate contributions from sources whose markers are degraded or removed during transport. For tracer-based methods, the appropriateness of the selected f_{SOC} and f_{SOA} values is also crucial to the accuracy of the results. Compared with primary source profiles, direct observational information on secondary source profiles is sparse. This paucity reflects the experimental difficulty of reproducing atmospheric oxidation and photochemical ageing under representative light and precursor conditions in laboratory or chamber systems. Most studies employing tracer-based methods to estimate SOC and SOA contributions have relied on f_{SOC} and f_{SOA} values reported by Kleindienst et al. (2007, 2012), which are now relatively dated. SOC and SOA formation are influenced not only by oxidants and sunlight but also by factors such as relative humidity, precursor concentrations, NO_x levels, and other ambient variables. The chamber conditions cannot fully reproduce the complexity of real atmospheres, leading to inevitable discrepancies in

f_{SOC} and f_{SOA} values between chamber-derived results and ambient air (Fu et al., 2009; Guo et al., 2012; Ding et al., 2012 and 2014; Haque et al., 2023). Additionally, the f_{SOC} and f_{SOA} values used were average, and their inherent standard deviations could result in deviations of 21% to 48% in the calculated SOC contributions from different sources (Kleindienst et al. 2007 and 2012). Despite the inherent uncertainties associated with both the CMB model and tracer-based method, these approaches remain convenient tools for estimating the contributions of various POC and SOC, yielding relatively reasonable results (Stone et al., 2009; Ding et al., 2012 and 2014; Al-Naiema et al., 2017; Ren et al., 2021; Xu et al., 2021; Haque et al., 2023). However, further efforts are needed to reduce these uncertainties. One critical step will be to conduct more extensive, site-specific observational studies of different types of primary and secondary emission sources. Such data are essential for identifying representative tracers within different sources and accurately determining their mass fractions in OC and OAs. Additionally, when f_{SOC} and f_{SOA} values are applied in both CMB model and tracer-based calculations, using the most recent, locally relevant emission source profile to minimize uncertainties is advisable.

3.6 Characteristics of OAs during winter pollution

From an annual timescale perspective, anthropogenic POC and SOC are synchronous with the measured OC and observed $\text{PM}_{2.5}$ mass (Figure S10). Correlation analysis revealed that anthropogenic POC (biomass burning, coal combustion, and vehicle exhaust) and SOC exhibited a significant positive correlation ($r = 0.72$ – 0.84 , $P < 0.01$) with the measured OC (Figure 6a). Redundancy analysis revealed that anthropogenic POC and SOC significantly contributed (54%–65%, $P < 0.01$) to the variations in OC concentration (Figure 6a). This suggests that anthropogenic OC and SOC are important factors influencing changes in the OC and $\text{PM}_{2.5}$ mass throughout the sampling period. Variations in POC, SOC, OC, and $\text{PM}_{2.5}$ mass were associated with specific meteorological conditions and gaseous pollutant concentrations. For instance, high POC, SOC, OC, and $\text{PM}_{2.5}$ concentrations corresponded to elevated atmospheric pressure, reduced precipitation, and increased NO_2 concentrations (Figure S2). This suggests that meteorological conditions and gaseous pollutant concentrations could also influence POC, SOC, OC, and $\text{PM}_{2.5}$ mass.

However, winter presents a different scenario. Regardless of whether it is during the winter pollution periods or other times in winter, POC did not maintain good consistency with the changes in the total OC

content and $PM_{2.5}$ mass, whereas SOC remained synchronized with their variations (Figure S11). This phenomenon is particularly pronounced during individual pollution episodes in winter (Figure S11). Correlation and redundancy analyses further revealed that source-specific SOC, derived using the tracer methods, and the total SOC, calculated using the EC-based method, exhibited significant positive correlations with the measured OC ($r > 0.6$, $P < 0.05$) and significantly contributed ($> 40\%$, $P < 0.05$) to their variations in entire winter pollution periods and individual pollution episodes (Figures 6b and 6d–6l). Given the relatively small number of nonpolluted days in winter and their scattered distribution across different dates, the correlations and contribution levels were not as evident during the remainder of the winter season (Figures 6c). These findings indicate that on shorter timescales, particularly during brief PM pollution episodes lasting several days, SOC was a critical factor influencing the total OC content and $PM_{2.5}$ mass.

The increased SOC concentration may be associated with elevated temperatures and NO_x concentrations during winter pollution episodes. This inference was supported by a significant linear positive correlation between temperature, NO_2 concentrations, and SOC concentrations observed during entire winter pollution periods and individual pollution episodes (Figures 7b and 7d–7l). Before conducting linear correlation analyses, the Shapiro-Wilk test was used to assess the normality of the data, and linear correlations were conducted only for datasets significantly conforming to a normal distribution ($p > 0.05$). An increase in short-term solar radiation intensity and oxidant levels have been shown to accelerate SOC formation (Fry et al., 2009; Ng et al., 2017; Li et al., 2018; Ren et al., 2019). Nevertheless, the contribution of POC should not be underestimated as its levels remained relatively high throughout the winter. This elevated POC concentration can also promote SOC transformation (Weber et al., 2007; Carlton et al., 2010; Hoyle et al., 2011; Srivastava et al., 2022). Outside pollution periods in winter, variations in temperature and NO_2 concentrations are minimal, which limits their promoting effect on SOC concentration increases (Figure 7c). Additionally, we analyzed the linear correlations between other meteorological factors and SOC. In particular, we examined the relations between wind speed, relative humidity, and SOC (Figure S12). The results indicated that no clear relation existed between wind speed and SOC. Although relative humidity showed some linear association with SOC during a few isolated pollution episodes, this relation was weak, with relatively large P-values and lacking statistical significance. These findings suggest that temperature and NO_2 concentrations are likely the main

meteorological and oxidative factors driving the increase of SOC during winter pollution periods in the Nanchang region.

4 Conclusions

We investigated the composition and concentration of major polar organic compounds in PM_{2.5} samples collected over a year in Nanchang, Central China. The results revealed relatively high concentrations of FAs, fatty alcohols, and saccharides, whereas lignin, resin products, sterols, glycerol, hydroxy acids, and aromatic acids were detected at low levels.

The study findings indicate that the organic components in the PM_{2.5} of Nanchang are predominantly derived from anthropogenic and natural sources. Anthropogenic sources were the primary contributors to OC and OA, and primary sources contributed more than secondary sources. Throughout the sampling period, we observed that the anthropogenic contributions significantly increased during autumn and winter, whereas biogenic contributions increased in spring and summer.

This study highlights the critical role of anthropogenic POC and SOC in influencing atmospheric PM_{2.5} pollution over an annual sampling period. During short-term pollution episodes in winter (lasting several days), the rapid secondary transformation emerged as the primary driver of OC increment and PM_{2.5} pollution. Elevated primary emissions and favorable oxidation conditions, such as increased light intensity and NO_x levels, were identified as key factors facilitating the rapid secondary transformation of OC during the winter pollution episodes.

The study findings underscore the necessity for targeted management strategies that consider primary and secondary anthropogenic emission sources across different seasons and pollution periods. Although the CMB model and tracer-based method provided preliminary insights into the OC and OA from diverse sources, the study encountered several inherent limitations. First, the primary emission source profiles employed in the CMB model exhibit variability across different studies and regions, making it challenging to establish standardized source characteristic parameters and potentially affecting the accuracy of source apportionment. Second, the proportions of SOC tracers obtained from laboratory chamber experiments are influenced by various factors, and incorporating these proportions into tracer-based methods may introduce potential biases or uncertainties in the estimates. To address these issues, future research should prioritize comprehensive

700 observational studies of primary emission sources to obtain high-resolution, region-specific emission data,
701 thereby improving the applicability of source profiles. Combining field observations with laboratory
702 simulations can also provide a more accurate characterization of secondary emissions, ultimately reducing the
703 uncertainties associated with tracer-based estimates of SOC contributions.

704
705 **Data availability.** The dataset associated with this paper is available upon request from the corresponding
706 author (xiaohuayun@sjtu.edu.cn).

707
708 **Competing interests.** The authors declare that they have no conflict of interest.

709
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Figure Captions

Figure 1. One-year time series of organic carbon (OC), elemental carbon (EC), OC/EC ratios, PM_{2.5} levels, and concentrations of polar organic compounds.

Figure 2. Molecular characteristics and seasonal variations of the major polar compounds in PM_{2.5}. The bars in the figure show the average concentrations of different compounds for each season, while the caps on the bars represent the standard deviations of these averages. The letters a, b, c, and d above the caps indicate whether the average concentrations differ significantly between seasons. Data points that do not share the same letter are significantly different, whereas those with the same letter are not significantly different, at the 95% confidence level. The Tukey test was used to assess the statistical significance of these differences. It should be noted that the same method was used to compare the statistical significance of differences between the data in the other figures presented later in the article.

Figure 3. Molecular characteristics and seasonal variations of the minor polar compounds in PM_{2.5}. The explanations for the letters a, b, c, and d on the bars are provided in the caption of Figure 2.

Figure 4. Molecular characteristics and seasonal variations of the SOA tracers in PM_{2.5}. The explanations for the letters a, b, c, and d on the bars are provided in the caption of Figure 2.

Figure 5. Concentrations and relative abundances of primary organic carbon (POC) from biomass burning, coal combustion, vehicle exhaust, cooking, plant debris, and fungal spores, as well as secondary organic carbon (SOC) from isoprene, α/β -pinene, β -caryophyllene, toluene, and naphthalene. POC and SOC concentrations were estimated using a tracer-based method, while other organic carbon (Other OC) was calculated by subtracting the estimated OC from the measured OC. Winter-P refers to winter pollution periods, and Winter-R denotes the winter remainder.

Figure 6. Pearson correlations between measured OC and each individual POC and SOC source for the annual sampling period (a), winter pollution periods (b), winter remainder (c), and individual pollution episodes in winter (d–l). The size of each circle represents the contribution (%) of POC and SOC to the variation in measured OC, as determined by redundancy analysis. Orange circles indicate P-values from correlation and redundancy analyses less than 0.05, while gray circles denote P-values greater than 0.05. Definitions: POC_{bb} (biomass burning), POC_{cc} (coal combustion), POC_{ve} (vehicle exhaust), POC_c (cooking), POC_{pd} (plant debris),

POC_{fs} (fungal spores), and POC_{EC-based} (based on the EC method). SOC_i, SOC_p, SOC_c, SOC_t, SOC_n, and SOC_{EC-based} correspond to SOC from isoprene, α/β -pinene, β -caryophyllene, toluene, naphthalene, and EC-based methods, respectively.

Figure 7. Linear correlations between temperature, NO₂, and SOC for the annual sampling periods (a), winter pollution periods (b), winter remainder (c), and individual winter pollution episodes (d–l). SOC concentrations were estimated using a tracer-based method. Winter-P refers to the winter pollution periods, while Winter-R denotes the winter remainder. PE1–PE9 correspond to winter pollution episodes 1–9. Before conducting linear correlation analyses, the Shapiro-Wilk test was used to assess the normality of the data, and linear correlations were conducted only for datasets significantly conforming to a normal distribution ($p > 0.05$).

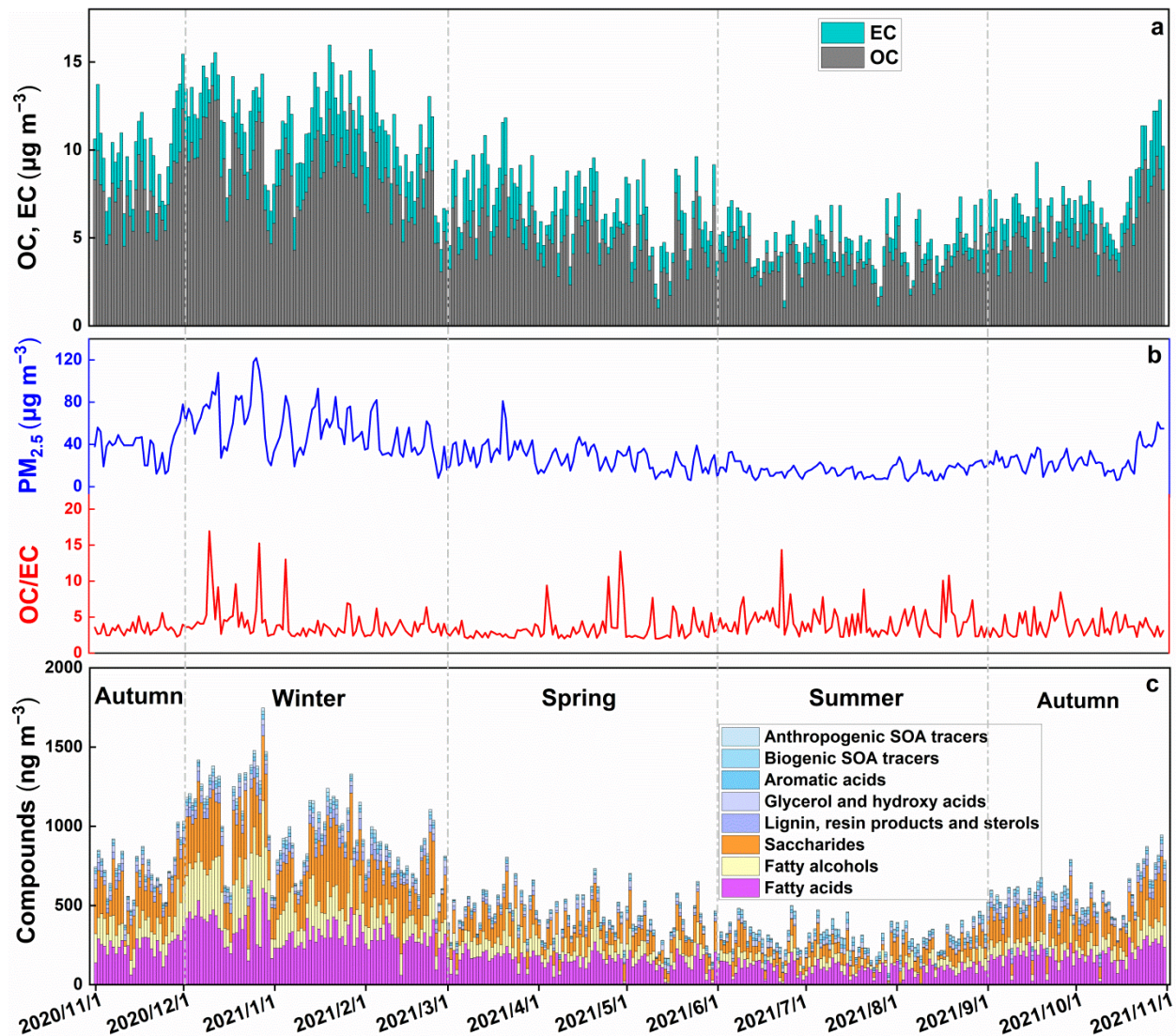


Figure 1. One-year time series of organic carbon (OC), elemental carbon (EC), OC/EC ratios, $\text{PM}_{2.5}$ levels, and concentrations of polar organic compounds.



Figure 2. Molecular characteristics and seasonal variations of the major polar compounds in PM_{2.5}. The bars in the figure show the average concentrations of different compounds for each season, while the caps on the bars represent the standard deviations of these averages. The letters a, b, c, and d above the caps indicate whether the average concentrations differ significantly between seasons. Data points that do not share the same letter are significantly different, whereas those with the same letter are not significantly different, at the 95% confidence level. The Tukey test was used to assess the statistical significance of these differences. It should be noted that the same method was used to compare the statistical significance of differences between the data in the other figures presented later in the article.

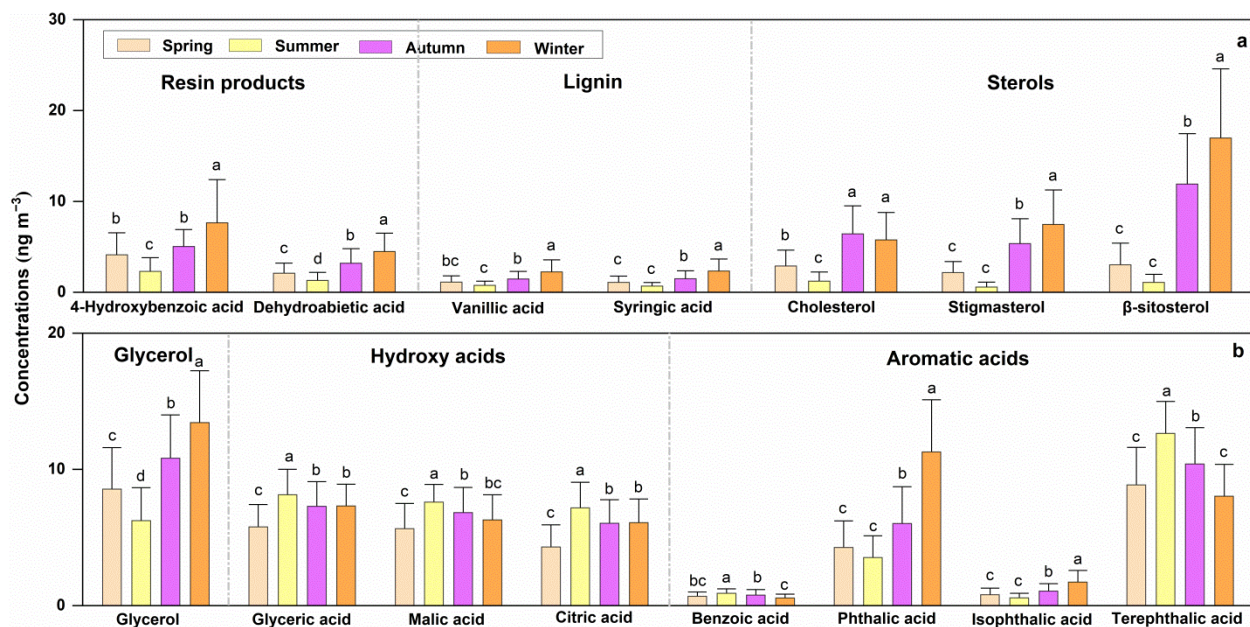


Figure 3. Molecular characteristics and seasonal variations of the minor polar compounds in PM_{2.5}. The explanations for the letters a, b, c, and d on the bars are provided in the caption of Figure 2.

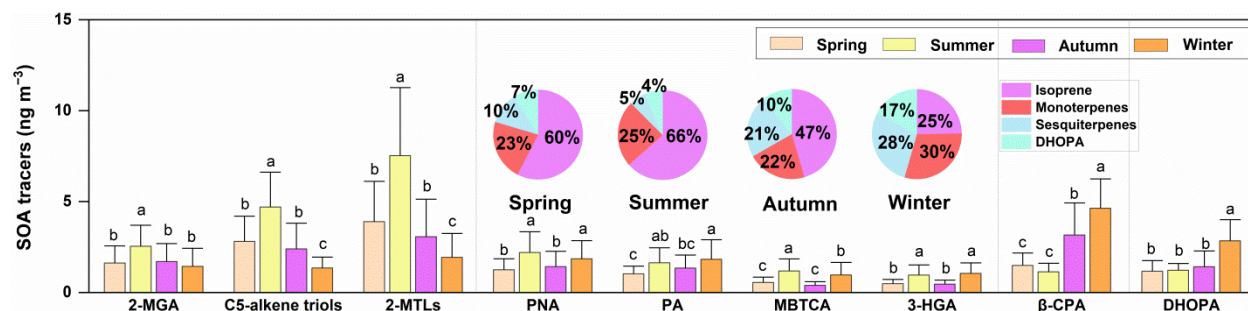


Figure 4. Molecular characteristics and seasonal variations of the SOA tracers in PM_{2.5}. The explanations for the letters a, b, c, and d on the bars are provided in the caption of Figure 2.

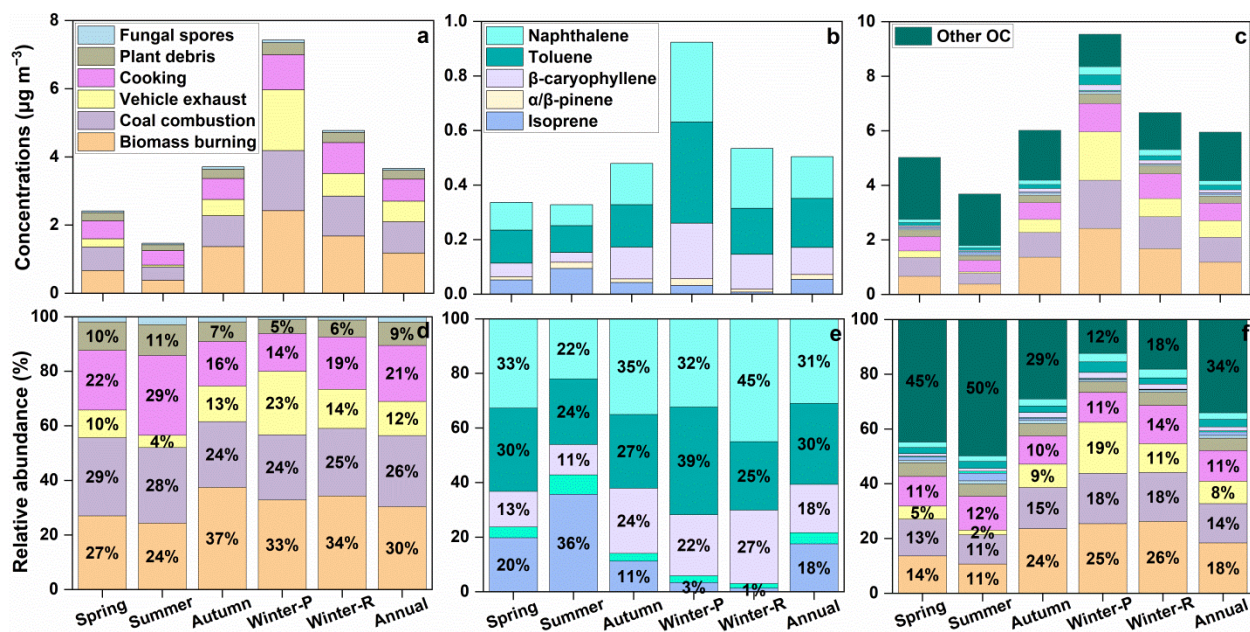


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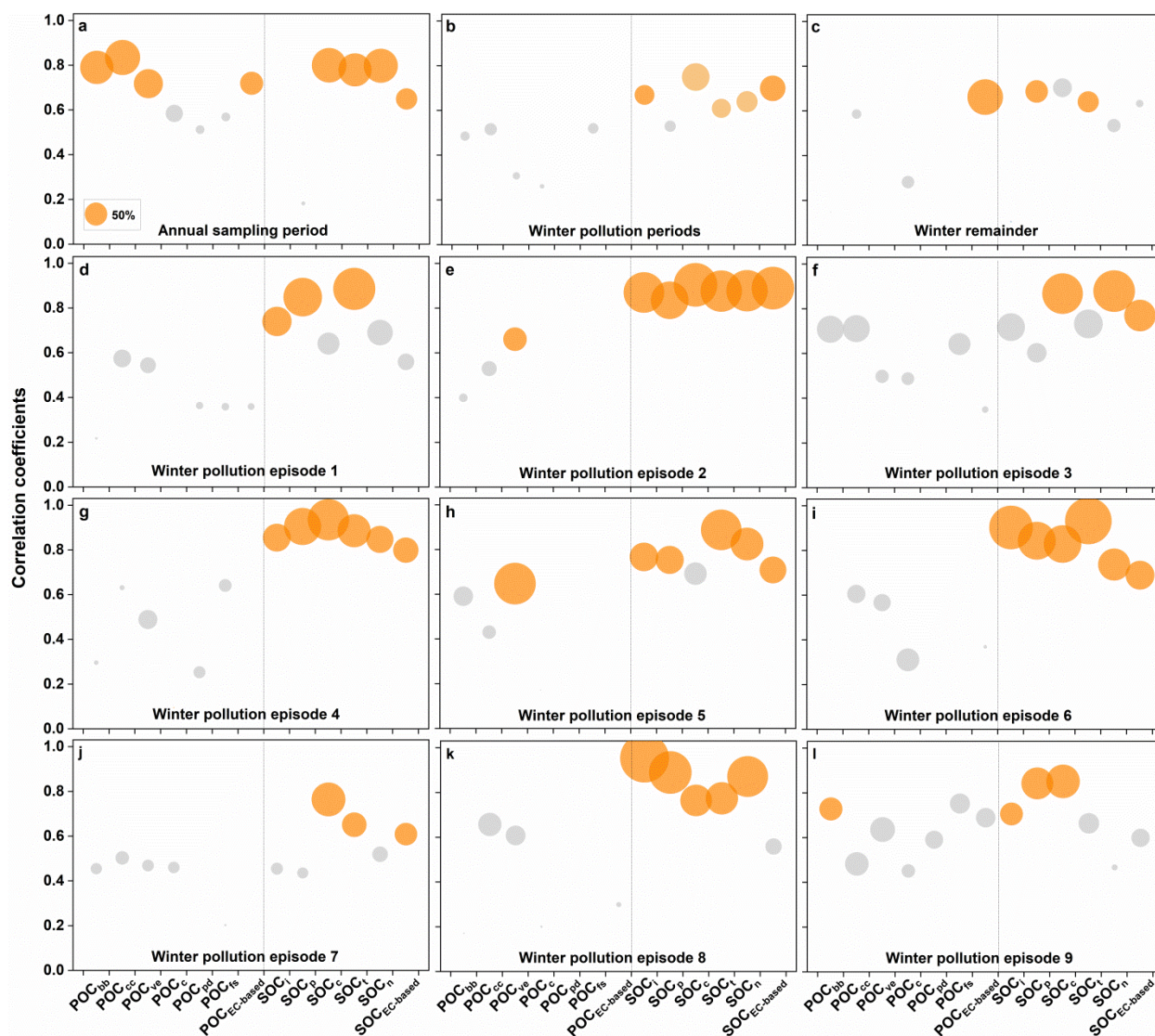


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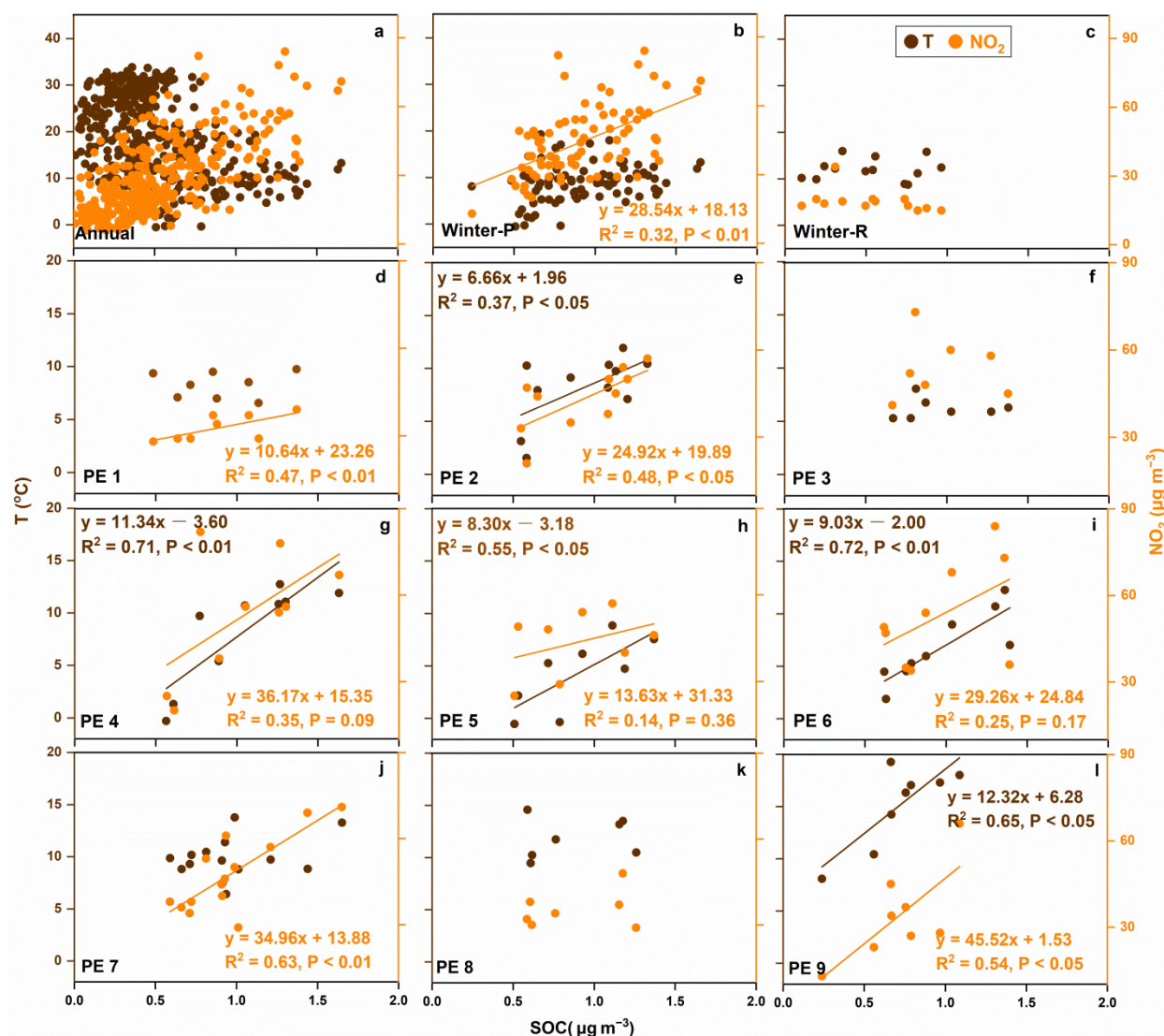


Figure 7. Linear correlations between temperature, NO₂, and SOC for the annual sampling periods (a), winter pollution periods (b), winter remainder (c), and individual winter pollution episodes (d–l). SOC concentrations were estimated using a tracer-based method. Winter-P refers to the winter pollution periods, while Winter-R denotes the winter remainder. PE1–PE9 correspond to winter pollution episodes 1–9. Before conducting linear correlation analyses, the Shapiro-Wilk test was used to assess the normality of the data, and linear correlations were conducted only for datasets significantly conforming to a normal distribution ($p > 0.05$).