



The influence of extratropical cross-tropopause mixing on the correlation between ozone and sulfate aerosol in the lowermost stratosphere

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Abstract. The composition of the upper troposphere/lower stratosphere region (UTLS) is influenced by horizontal transport, vertical transport within convective systems and warm conveyor belts, rapid turbulent mixing, as well as photochemical production or loss of species. This results in the formation of the extratropical transition layer (ExTL), which has been defined by the vertical structure of CO profiles and studied by now mostly by means of trace gas correlations. Here, we extend the analysis to aerosol particles and derive the ozone to sulfate aerosol correlation in Central Europe from aircraft in-situ measurements during the CAFE-EU/BLUESKY mission, probing the UTLS during the COVID-19 period with significant reduced anthropogenic emissions. We operated a compact time-of-flight aerosol mass spectrometer (C-ToF-AMS) to measure the chemical composition of non-refractory aerosol particles in the size range from about 40 to 800 nm. In our study, we find a correlation between the ozone mixing ratio (O₃) and the sulfate mass concentration in the lower stratosphere. The correlation exhibits some variability over the measurement period exceeding the background sulfate to ozone correlation. Especially during one flight, we observed enhanced mixing ratios of sulfate aerosol in the lowermost stratosphere, where the analysis of trace gases shows tropospheric influence. Also, back trajectories indicate, that no recent mixing with tropospheric air occurred within the last 10 days. In addition, we analyzed satellite SO₂ retrievals from TROPOMI for volcanic plumes and eruptions. From these analyses, we conclude that gas-to-particle conversion of volcanic SO₂ leads to the observed enhanced sulfate aerosol mixing ratios.

1 Introduction

The chemical composition of upper tropospheric aerosol particles is highly variable because primary aerosols from natural and anthropogenic ground sources reach this altitude (Martinsson et al., 2019), and secondary aerosols are formed here from gas-to-particle conversion. However, the stratospheric aerosol composition is less complex as the main component is particu-

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late sulfate (SO_4^{2-}) with concentrations between 0.1 and 40 µg m⁻³ accompanied by minor tropospheric compounds (Deshler, 2008; Murphy et al., 2013; Brimblecombe, 2014; Kremser et al., 2016).

The stratospheric aerosol layer, also known as "Junge layer" (Junge and Manson, 1961), is part of the lowermost stratosphere (LMS) and is located roughly between 15 and 25 km (Hofmann and Rosen, 1981). The chemical composition of the aerosol layer underlies seasonal variations induced by the Brewer-Dobson circulation (Martinsson, 2005; Friberg et al., 2014) and volcanic activity, which may increase the aerosol optical depth (AOD) by up to 40 % (Friberg et al., 2018). Sulfate aerosol is formed due to oxidation of carbonyl sulfide (OCS) and sulfur dioxide (SO₂) (Crutzen, 1976; Brühl et al., 2012; Solomon et al., 2011; Kremser et al., 2016). Both precursor gases have their main sources in the troposphere. OCS is the main sulfur containing trace gas in the atmosphere and has biological sources, such as photochemical dissolving of organic matter and oceanic emissions (Andreae, 1990; Brühl et al., 2012; Kremser et al., 2016). SO₂ is primary emitted by industrial processes like the fossil fuel burning. While degassing volcanoes contribute to the tropospheric SO₂ budget (Voigt et al., 2014), explosive eruptions can directly inject SO₂ into the lower stratosphere (Kremser et al., 2016). Another direct sulfate aerosol source are aircraft which emit soot and volatile sulfate containing particles at cruise altitudes into the upper troposphere/lower stratosphere (Voigt et al., 2010; Williamson et al., 2021; Tomsche et al., 2022).

Transport processes of aerosol particles into the UTLS have been subject of several studies, for example with focus on tropical processes or the Asian tropopause aerosol layer (ATAL) (e.g., Appel et al., 2022; Fadnavis et al., 2013; Froyd et al., 2009; Höpfner et al., 2019). Tracer correlations of aerosol particles with ozone (O₃) and nitrous oxide (N₂O) based on high altitude in-situ measurements also have been used in the context of polar vortex dynamics after a major volcanic eruption (Borrmann et al., 1993, 1995). In those studies, the temporal evolution of the correlation between the mixing ratios of sulfate aerosol, surface area for aerosol particles with a size between roughly 10 nm and more than one µm, in addition to O₃ inside and outside of the polar vortex was analyzed over the course of 22 months after the Mt. Pinatubo eruption in 1991. The observations revealed a slow development of a linear correlation between the O₃ versus aerosol number and surface mixing ratios in the mid latitude UTLS, which degraded again later on. This demonstrated the suitability of aerosol properties as dynamical tracers once the microphysical processes like new particle formation, coagulation, and condensational growth following a volcanic eruption into the stratosphere have ceased. Furthermore, a temporally stable equilibrium of particle size/surface area has been established.

In the extratropics, the chemical composition of the tropopause region is influenced not only by the Brewer-Dobson circulation, but also by convection, mixing along the subtropical and polar frontal jet streams, breaking of gravity and Rossby waves, and vertical wind shear (e.g., Gettelman et al., 2011; Kaluza et al., 2021). This forms a transition layer above the tropopause, called the extratropical transition layer (ExTL), where tropospheric as well as stratospheric influence is observed (Hoor et al., 2004; Hegglin et al., 2009; Gettelman et al., 2011; Konopka and Pan, 2012; Barré et al., 2013). The effect of these small-scale mixing processes on the chemical composition of aerosol particles in the ExTL is not well known until now (Kunkel et al., 2019). The lifetime of atmospheric aerosol particles with diameters lower than one µm can reach one month or more (Jaenicke, 1980). This is sufficient for the particles to be transported up into the tropopause region over long distances and subsequently, by mixing processes, into the stratosphere. Furthermore, the lifetime of more than a month corresponds to the timescale that





gas-to-particle conversion needs to form sulfate aerosol particles from SO₂ as a precursor gas in the UTLS (Jurkat et al., 2010; Gorkavyi et al., 2021; Rollins et al., 2017).

In our study, we use tracer-tracer correlations as a tool for mixing diagnostics to identify stratospheric air masses and underlying mixing processes. The basic principle of this method is to use a tropospheric tracer with sources in the troposphere and a rather constant stratospheric background, e.g. carbon monoxide (CO) and water vapor (H_2O), and a tracer with only a stratospheric increase or decrease and a fairly constant mixing ratio in the troposphere, e.g. O_3 or N_2O . In a scatter plot with the tropospheric tracer on the abscissa and the stratospheric tracer on the ordinate, one would expect two separated reservoirs that are not connected if no mixing processes would occur. If mixing takes place, both reservoirs are be connected by mixing lines, where the mixing ratios are between the two regimes, depending on the state of mixing (Fischer et al., 2000; Hoor et al., 2002).

With this study we want to introduce particulate sulfate as stratospheric tracer in the correlation with O₃ and find processes that are responsible for the variability of the correlation between sulfate aerosol and O₃ mixing ratios. Therefore, we use in-situ aircraft measurements from the CAFE-EU (Chemistry of the Atmosphere Field Experiment over Europe)/BLUESKY mission, conducted in spring 2020 from Oberpfaffenhofen, Germany.

70 2 Methods

2.1 Data overview

The CAFE-EU/BLUESKY measurement campaign was conducted with the research aircraft HALO (High Altitude and Long Range Research Aircraft) and DLR-Falcon, both operated by the German Aerospace Center (DLR). The measurement flights were performed over Central Europe and the North Atlantic between 16 May and 09 June 2020, partly co-located with both aircraft (see Fig. 1). The measurements were conducted during the first COVID-19 lockdown in Germany and Europe, such that the main goal of the campaign was to investigate the atmospheric changes during reduced industrial activity and lower emissions compared to other times (Voigt et al., 2022). During this time, air traffic was significantly reducd over Europe by up to 80 % (Schumann et al., 2021a, b). Krüger et al. (2022) found substantial reduction in aerosol particles in the lower troposphere in this period and Reifenberg et al. (2022) could explain the observed reduction in some tracer concentrations with the reduced emissions of pollutants. Tomsche et al. (2022) investigated the SO2 concentrations in the UTLS region above Europe, which was influenced by changes in sulfur sources such as aviation as well as sinks. Here we focus on the transport processes in the extratropical transition layer, which has been probed with a set of instrumentations onboard HALO and DLR-Falcon.

2.2 Instrumentation

For our study, we use the chemical composition of aerosol particles measured onboard HALO and trace gas measurements onboard HALO and DLR-Falcon. The chemical composition of non-refractory aerosol particles was measured on HALO with





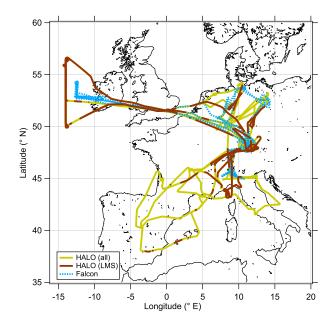


Figure 1. Overview map of all measurement flights performed during the CAFE-EU/BLUESKY measurement campaign between 16 May and 09 June 2020. The HALO flight path as solid line divided into the complete data set (yellow) and stratospheric (brown) segments, while the DLR-Falcon flight path is shown as a dashed blue line.

a compact time-of-flight aerosol mass spectrometer (C-ToF-AMS) for particles in the size range between 40 nm and 800 nm (Drewnick et al., 2005; Schulz et al., 2018) and with this within the same size range as previous studies (e.g., Borrmann et al., 1995). We obtain quantitative information on the mass concentration of sulfate, nitrate, ammonium, organic matter and chloride normalized to STP conditions. The measurement interval of the C-ToF-AMS during the campaign was 30 s, resulting in a spatial resolution of about 6 km in the UTLS region. The accuracy of the AMS is about 30 % (Canagaratna et al., 2007; Middlebrook et al., 2012). In the following, we use the mixing ratio instead of the mass concentration for comparison with the trace gas measurements

In addition to the aerosol chemical composition data, we use trace gas measurements, like SO_2 , CO, H_2O , O_3 and nitric acid (HNO_3) onboard HALO and DLR-Falcon.

CO measurements on HALO were performed with the quantum cascade laser absorption spectrometer TRISTAR (Tadic et al., 2017; Röder et al., 2023) with a total measurement uncertainty of 3 % at 10 s time resolution. O₃ on HALO was measured by FAIRO which measured on the basis of a UV photometer and chemiluminescence (Zahn et al., 2011). On board the DLR-Falcon, CO and O₃ were measured with a cavity ring-down spectrometer (PICARRO G2401) and a dual-cell UV photometer (TE 49C), respectively. Both instruments were calibrated before and after the flights with standards that can be traced back to the GAW Station Hohenpeissenberg. The precision/accuracy of the CO and O₃ measurements are 3 ppbv/ 5 ppbv and 3 %/5%, respectively. An additional in-situ data set is provided by the atmospheric chemical ionization mass spectrometer (AIMS) deployed on DLR-Falcon and includes information on gaseous SO₂ and HNO₃. For the detection of upper tropospheric and





lower stratospheric SO₂ and HNO₃ mixing ratios the AIMS uses SF₅⁻ reagent ions (Voigt et al., 2014; Jurkat et al., 2016; Marsing et al., 2019; Tomsche et al., 2022). The one sigma detection limit is 0.0006 to 0.0017 ppbv and 0.005 to 0.009 ppbv for SO₂ and HNO₃, respectively. The total uncertainty for SO₂ is 22.7 % (Tomsche et al., 2022) and 16 % for HNO₃ (Marsing, 2021).

2.3 Meteorology and trajectories

In addition to the in-situ measurement data, we use model data interpolated onto the flight path of both aircraft. For meteorological information, we use the ERA5 reanalysis data set with a temporal resolution of three hours and a grid spacing of 0.5° in the horizontal and a vertical spacing of approximately 500 m in the UTLS (Hersbach et al., 2020). Based on the native variables we additionally calculated potential vorticity and equivalent latitude.

For our analysis of the air mass origin and possible transport pathways, we use trajectories calculated with the Lagrangian analysis tool (LAGRANTO; Sprenger and Wernli, 2015). Therefore, we initialize a set of trajectories every 30 seconds along the flight path. The starting points of the trajectories of each set are placed in a three dimensional box around the actual point at the flight path, to gain a better statistic and to minimize interpolation errors between the measurements and the model grid. More specifically, we take the location of the aircraft and add five additional points every 0.01 degree in all horizontal directions and five points every one hPa above and below the altitude of the aircraft. Thus, a total of 231 trajectories are started at each release time, providing information for 10 days back in time with quantities such as potential temperature and potential vorticity.

120 3 Results

3.1 Part 1: Correlation of ozone and particulate sulfate

 O_3 is a suitable tracer to identify stratospheric air masses due to the photochemical production of O_3 in the stratosphere and its low abundance in the troposphere and its local chemical lifetime of years in the lower stratosphere. There are several ways to identify the tropopause and thus the lower boundary of the stratosphere from measured O_3 mixing ratios. For example, fixed threshold values of typically 70 ppbv or 100 ppbv have been used (e.g., Bethan et al., 1996; Staehelin, 2003). This method has the disadvantage to neglect the seasonal cycle of O_3 and thus the threshold value can be too low or too high when periods exceeding one month are analyzed. Another way to determine the O_3 -based tropopause is to take the seasonal cycle into account by using a daily threshold for the O_3 tropopause. This method is described in Zahn et al. (2004) and Thouret et al. (2005) on the basis of long-term observations. In our study we use the method of Zahn et al. (2004) to calculate daily O_3 thresholds for the tropopause. Figure 2 shows the seasonal cycle of the O_3 mixing ratio at the chemical tropopause and the 2 PVU (potential vorticity units, $10^{-6}m^2s^{-1}Kkg^{-1}$) dynamical tropopause, both calculated after Zahn et al. (2004). For the period of our study (16 May to 09 June), we only see small differences in the results between the chemical and the dynamical tropopause since both thresholds are close together around 120 ppbv O_3 during the time of our measurements. We extracted all stratospheric



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data (i.e. all data points with O₃ mixing ratios larger than the daily threshold) and calculated the frequency distribution of the potential vorticity from the ERA5 data set (Fig. 2) to verify that the chemical tropopause inferred from the measured O₃ mixing ratios and the calculated thresholds correspond to the dynamical tropopause. This shows that the vast majority of the stratospheric data points have PV values larger than 2 PVU. In Fig. 2, we observe two modes in the PV distribution of the stratospheric data which can be explained by the different stratospheric ages of the sampled air masses. In total, less than 4 % of the data points have PV values below the 2 PVU threshold. Thus we use the definition of the chemical tropopause based on O₃ mixing ratios after Zahn et al. (2004) in this study.

Besides O₃, there are other trace gases used as indicators of stratospheric air masses, e.g. H₂O or N₂O. Moreover, within lim-

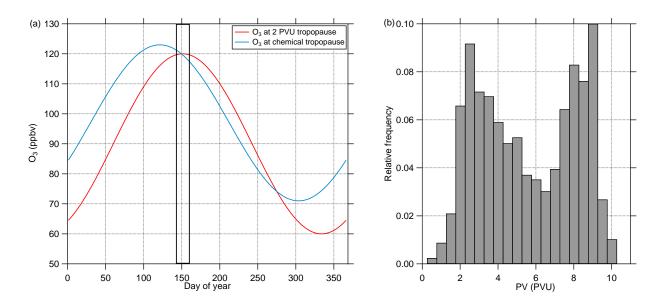


Figure 2. (a) Calculated seasonal cycle of O_3 mixing ratios at the 2 PVU dynamical tropopause and the chemical tropopause, both calculated after Zahn et al. (2004). The measurement period is marked with the solid frame. (b) Relative frequency of PV values the along the AMS data points from CAFE-EU/BLUESKY identified as stratospheric using the calculated O_3 mixing ratios to define the chemical tropopause. Less than 4 % of the data points show PV values lower than 2 PVU indicating a negligible amount of tropospheric air. The complete stratospheric data set holds 2049 data points which represent the 100 % of the subset.

its, aerosol particle properties can be applied as well (Borrmann et al., 1993, 1995). A good example is the mass concentration of particulate sulfate, which increases in the stratosphere due to the formation from precursors, which takes 30 to 60 days and reaches its maximum in the Junge layer, where particulate sulfate is present in the form of binary solution droplets with inclusions of sulfuric acid (Junge and Manson, 1961; Brühl et al., 2012; Kremser et al., 2016). Thus, we expect a positive correlation between particulate sulfate and O_3 in the stratosphere. The observations from the CAFE-EU/BLUESKY campaign confirm this (Fig. 3). Here, two distinct regimes appear in the correlation plot of O_3 and particulate sulfate. In the tropospheric regime, the sulfate mass concentration shows a high variability at O_3 levels below 100 ppbv, depending on the source regions within the the boundary layer, e.g. industrial areas. In the stratosphere, we observe a linear correlation between the two species with a



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slope of 900 to 2300 ppbv/ppbv, but with variations between the individual measurement flights, i.e. on short time scales of a few days. This analysis shows that the correlation between O₃ and particulate sulfate can be used as a tool for analysing

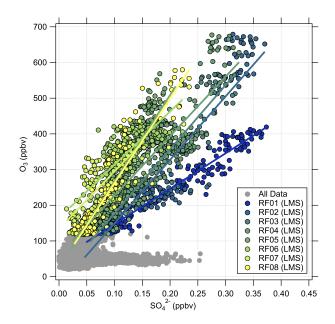


Figure 3. Correlation between O_3 mixing ratio and particulate sulfate mixing ratio for the full CAFE-EU/BLUESKY data set. The color-coded data points indicate the stratospheric data derived from the tropopause O_3 mixing ratios. The grey data shows the complete data set including the tropospheric data. The solid color-coded lines represent the linear regressions for the individual flights.

air masses in terms of their stratospheric character. The variations in the slope and compactness of the correlation appear on short timescales of a few days. The aim of this study is to understand these variations and link them to possible atmospheric processes. There are a number of different pathways by which sulfur species can be transported from the troposphere into the stratosphere and thus be a possible reason for the observed variability (e.g., Kremser et al., 2016). Feinberg et al. (2019) show the modelled atmospheric sulfur budget under volcanically quiescent conditions and the pathways that lead to the formation of particulate sulfate in the stratosphere. Among these pathways, the most efficient one is the mixing of precursor gases such as OCS and SO₂ into the stratosphere, where OCS is oxidised to SO₂ and SO₂ is further converted to sulfuric acid, forming sulfate aerosol. It is important to emphasise that this budget is valid for volcanically quiescent conditions, because in the presence of volcanic eruptions an additional large source of SO₂ adds up to the other pathways. In this case, SO₂ is transported in the eruption column up into the free troposphere or even the UTLS region; depending on the strength of the eruption. Then SO₂ is converted to sulfuric acid and particulate sulfate also in the upper troposphere (Kremser et al., 2016).

Figure 3 shows that direct mixing of sulfate aerosol was not observed during the campaign, due to the low sulfate mixing ratios at the chemical tropopause, so some other processes need to be taken into account. Previous studies with focus on the Raikoke eruption in 2019 determined no significant contribution from this volcanic eruption (Tomsche et al., 2022; Reifenberg et al.,



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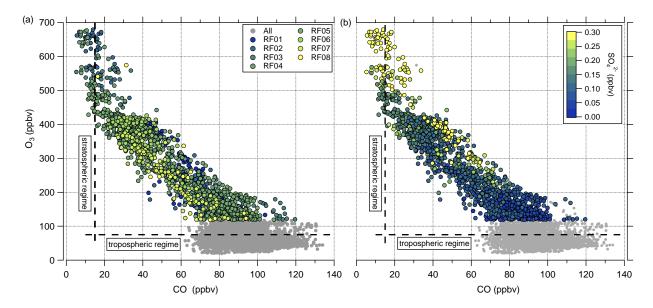


Figure 4. (a) CO-O₃ correlation for all data points in grey and the stratospheric data color-coded with the flight numbers. (b) The same correlation but the stratospheric data is color-coded with the sulfate aerosol mixing ratio. In addition, the dashed lines indicate the mean tropospheric and stratospheric regimes without mixing processes.

2022). The following section will show the influence of cross-tropopause mixing, especially of the precursor gas SO_2 and the potential influence of a more current volcanic eruption.

3.2 Part 2: Case study on aerosol chemical composition related to mixing processes

Volcanic influence is one of the possibilities that can explain the observed variability in the correlation. One major eruption occurred in 2019 by the Raikoke volcano. However, Tomsche et al. (2022) and Reifenberg et al. (2022) showed that this eruption does not have a significant impact on the measurements during CAFE-EU/BLUESKY. In the following, we focus on a case study to explain the variability, especially for research flight RF01, as the slope in the O₃-SO₄²⁻ correlation for this flight clearly differs from the other flights (see Fig. 3).

The correlation between CO and O_3 can be used to identify mixing processes between the troposphere and stratosphere (Fischer et al., 2000; Hoor et al., 2002). The presence of mixing lines connecting the sampled tropospheric and the stratospheric air masses indicates recent mixing processes. In Fig. 4, most of the stratospheric data of the whole campaign data set lie on such mixing lines. Thus, it can be concluded that the measured air masses in the stratosphere are influenced by tropospheric air that was mixed across the tropopause. As expected, the correlation in Fig. 4b shows much higher particulate sulfate mixing ratios at higher O_3 levels in the stratosphere and lower mixing ratios close to the tropopause. However, we can identify an anomaly of high particulate sulfate mixing ratios (yellow points) at about 400 ppbv O_3 and 40 ppbv CO (see also Fig. C1). Here, the measured mixing ratios of sulfate aerosol are in the range we would expect higher up in the stratosphere.



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In the following, we will analyze whether this anomaly is caused by a cross-tropopause mixing event and if such events can explain the observed variability in the SO_4 - O_3 correlation. For this analysis, we binned our data set along equivalent latitude, which can be used as a dynamical tracer (Butchart and Remsberg, 1986; Hegglin, 2005), and potential temperature to see where the anomaly is located (see Fig. 5). The observed sulfate anomaly occurs in Fig. 5b between 40° N and 45° N at potential temperatures between 345 K and 350 K and is not connected to the stratosphere. The slope of the potential vorticity contour lines indicates that this region is in the vicinity of the jet stream. The sulfate anomaly can not be identified in the O_3 distribution (Fig. 5a), but we can observe the expected increase from the troposphere to the stratosphere. This location of the anomaly is in good agreement with the previous observation that the anomaly is located on a mixing line in the $CO-O_3$ correlation in a transition regime between the troposphere and the stratosphere.

We further investigate the meteorology over the campaign period to determine whether the anomaly is induced by mixing processes. In particular, we use the vertical wind shear S^2 and the static stability N^2 (see also Appendix B) to identify regions with higher potential for mixing processes. Kaluza et al. (2021) showed in their study that in regions with high vertical wind shear ($S^2 > 4 \cdot 10^{-4} \text{ s}^{-2}$), conditions are favourable for rapid mixing. Figure 6 shows the analysis of the stability parameters mentioned above along with the resulting gradient Richardson number. The vertical wind shear shows high values in the region of the sulfate anomaly (see Fig. 6b), exceeding the mentioned threshold for enhanced mixing. This also results in a reduction of the gradient Richardson number in the same bin to values around 1 (see Fig. 6c), which indicates favourable conditions for turbulence. The static stability shows the expected transition from tropospheric to stratospheric values (see Fig. 6a). These findings suggest that mixing occurred in this region and plays a role in the variation of the SO_4 - SO_3 correlation. To prove this, we investigated the data in the region of the sulfate anomaly in more detail. Therefore, we extracted the data points that contribute to the anomaly. In total there are 45 data points in the bin between 40 and 50° N and 345 to 350 K. The majority of these points (41 points) was sampled during research flight RF01 in a time span of 20 minutes (see Fig. 3). The flight was conducted on 23 May 2020 over Germany, while the anomaly was measured in an area over Lower Saxony (see also Fig. C1 and Fig. C2). A time series of the measurements from this flight is shown in Fig. 7, while the period of the anomaly is marked with a colored box.

Figure 7 reveals the correlation between CO and particulate sulfate in the lower troposphere, as well as the positive correlation between O_3 and particulate sulfate and the anti-correlation between sulfate and CO in the stratosphere. To identify mixing from the measurements, we use two types of scatterplots. The first is the previously introduced scatterplot using CO and O_3 with different color coding (Fig. 8a-c), the second is the scatterplot with H_2O and O_3 (Hegglin et al., 2009) (see Fig. 8d-f). This figure contains only data measured on 23 May 2020. The H_2O-O_3 method follows the same principle, with high water vapor mixing ratios in the troposphere and a constant stratospheric background value around five ppmv as described in Hegglin et al. (2009). All of these scatterplots show two separate branches of mixing lines. The observed sulfate anomaly is located on the upper branch of the mixing lines.

Regardless of the type of scatterplot, we observe an increase in the particulate sulfate mixing ratio and potential temperature along the mixing line, starting at the tropopause and extending into the stratosphere (Fig. 8). In contrast, we also observe a decrease in the CO mixing ratio. This result is consistent with the assumption that tropospheric air enters the stratosphere



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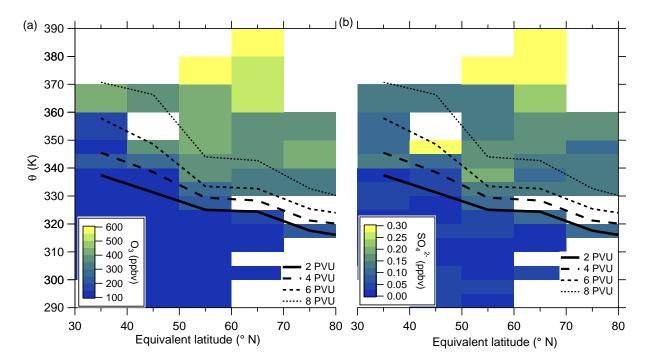


Figure 5. Median of O_3 mixing ratio (a) and sulfate aerosol mixing ratio (b) in the θ -equivalent latitude space. Below 350 K, we use 5 K vertical resolution and 10° equivalent latitude bins. Above 350 K, we enlarge the vertical bins to 10 K to obtain a higher statistical evaluation basis (see Fig. A1). In general, only bins with more than 10 data points are evaluated. We added the 2 to 8 PVU lines to show the location of the dynamical tropopause and the extent of the ExTL into the stratosphere. We only show the region of interest for our study between 30 and 80 degrees north.

at lower potential temperatures with lower amounts of sulfate aerosol, accordingly higher mixing ratios of precursor gases, and higher CO. During the slow ascent was consequence of the Brewer-Dobson circulation and the processing of air masses, gas-to-particle conversion of OCS and SO₂ will lead to an increase of particulate sulfate. CO will decrease by dilution and also by photochemical destruction, with an atmospheric lifetime of one to three months (Seinfeld and Pandis, 2016). This is almost the same time as the calculated e-folding time of gas-to-particle conversion of SO₂ to sulfate aerosol in the mid-latitude LMS region (Jurkat et al., 2010).

Further evidence that the anomaly is caused by tropospheric influence are the lower O_3 values and the water vapor mixing ratios. If the air mass were of stratospheric origin, we would expect O_3 mixing ratios higher than 400 ppbv (see Fig. 8) and a water vapor mixing ratio of 10 to 15 ppmv.

Similarly to the HALO measurements, we analysed measurements conducted on the DLR-Falcon for this analysis. The DLR-Falcon performed a measurement flight on the same day and sampled in the area where the anomaly was encountered just 40 min later than HALO (Fig. C2). The DLR-Falcon does not reach the same high altitudes as HALO, so the air masses were probed slightly below and thus show higher CO mixing ratios. This time, we use the scatterplot of CO and HNO₃ to identify





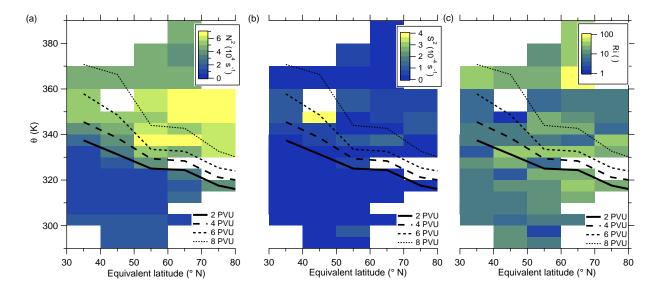


Figure 6. Median of static stability N^2 (a), vertical wind shear S^2 (b) and gradient Richardson number (c) in the θ -equivalent latitude space. Below 350 K, we use 5 K vertical resolution and 10° equivalent latitude bins. Above 350 K, we enlarge the vertical bins to 10 K to obtain a higher statistical evaluation basis (see Fig. A1). In general, only bins with more than 10 data points are evaluated. We added the 2 to 8 PVU lines to show the location of the dynamical tropopause and the extent of the ExTL into the stratosphere. We only show the region of interest for our study between 30 and 80 degrees north.

mixing in combination with gas-to-particle conversion (Fig. 9). HNO₃ was already introduced as a stratospheric tracer by Proffitt et al. (1989) or Arnold et al. (1989) and utilised because the O₃ data for the DLR-Falcon are not available for this flight. We identify a mixing line in the scatterplot, connecting troposphere and stratosphere. The SO₂ mixing ratios on this mixing line are elevated with respect to the measured stratospheric values around 0.04 ppbv but already reduced from their measured tropospheric maxima of 0.01 ppbv, which is an indication for gas-to-particle conversion along this mixing line.

As a complement to the measurement data, we analyzed the meteorological situation along the flight path for the period of the anomaly (Fig. B1). The flight path was located just above the maximum of the subtropical jet stream and a layer of strong vertical wind shear (Fig. B1a). Further, we see that the flight path crossed a layer of low gradient Richardson number (Fig. B1)b, and later continued slightly above this layer. This indicated a region of instability which is an important factor for mixing processes. The preceding discussion has shown that very likely mixing between tropospheric and stratospheric air masses occurred prior to the in-situ measurements.

Hereafter, we examine the origin of the air masses comprised of high mixing ratios of sulfate aerosol. Therefore, we exploit LAGRANTO backward trajectories starting at a grid around the flight path with 230 trajectories for each measurement point. The trajectories are calculated 10 days backward, to see whether the air masses ascended in regions with higher concentrations of precursor gases or even crossed such regions in the upper troposphere. For our analysis, we filtered the trajectories by selecting only those with a minimum potential temperature below 345 K and with an increase of at least 5 K potential



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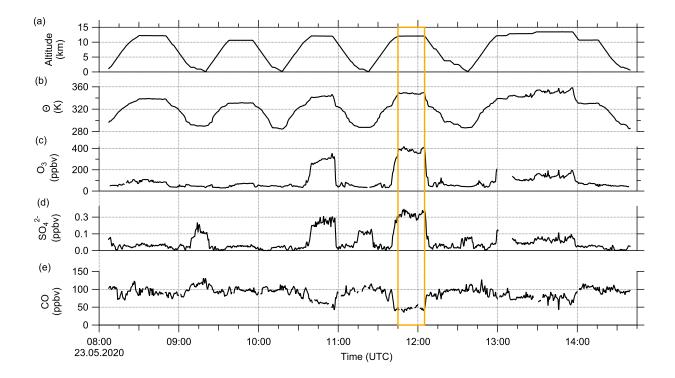


Figure 7. Time series of in-situ measurements during research flight RF01 on 23 May 2020 in 30 s time steps. (a) Altitude, (b) potential temperature θ , (c) O₃ mixing ratios, (d) sulfate aerosol mixing ratios and (e) CO mixing ratios. The orange box marks the period with the observed sulfate anomaly described in the text and shown in Fig. C1.

temperature. Figure 10 shows that the selected trajectories are close together and move with the jet stream. The trajectories cross the region of East Asia within 10 days before the measurement and most of them crossed China and its megacities like Chengdu and Shanghai. The time series of potential temperature shows that no boundary layer air masses were transported to the measurement region in the last 10 days (Fig. 10a), considering the assumption that the trajectories can resolve convective uplift. As a consequence, the enhanced particulate sulfate needs to be older than 10 days and was most likely not directly mixed into the LMS as particulate sulfate.

Alternatively, the sulfate aerosol could originate from gas-to-particle conversion of SO_2 that was mixed into the LMS. To examine this hypothesis, we start from Fig. 9, which shows that close to the tropopause, the SO_2 values are quite high and are decreasing along the mixing line. This is an indicator for ongoing gas-to-particle conversion in combination with cross-tropopause mixing. A possible source for the high SO_2 mixing ratios in the troposphere and close to the tropopause could be volcanic activities. Therefore, we searched for volcanic eruptions in the period of two months before the measurements, corresponding to the e-folding time of about 50 to 60 days (Jurkat et al., 2010). For the analysis, we used daily TROPOMI retrievals in different volcanically active regions to identify possible source regions. Thereby, we identified the Kamchatka Peninsula in Russia as an origin of enhanced SO_2 emissions in the beginning of April 2020. This corresponds to the archived reports by the



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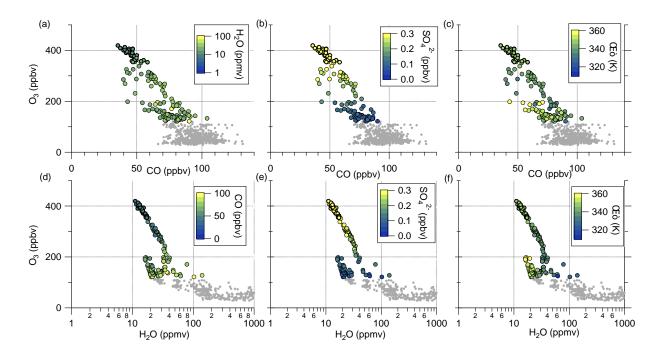


Figure 8. Tracer-tracer correlations for 23 May 2020 to identify mixing processes. The CO-O₃ correlations are color-coded with (a) SO_4^{2-} , (b) θ and (c) H₂O. The H₂O-O₃ correlations are color-coded with (d) SO_4^{2-} , (e) θ and (f) CO. The data correspond to the blue colored data in Fig. 3 and the sulfur anomaly in Fig. 5.

Kamchatka Volcanic Eruption Response Team (KVERT) (Institute of Volcanology and Seismology FEB RAS, 2023). In the report for 08 April 2020, an explosive eruption of the Sheveluch volcano is described, with a volcanic cloud height reaching up to 10 km and thus into the tropopause region. We performed HYSPLIT (Stein et al., 2015) forward dispersion simulations of the Sheveluch volcano plume and analyzed the eruption plume in the model in different heights to get a broad overview of the distribution. Therefore, we calculated forward trajectories in heights between 9000 and 11000 m with a vertical resolution of 500 m. In the following, we only consider the levels up to 9000 m (see Fig. D1) and between 10000 and 11000 m (see Fig. 11), because here the model shows differences in the plume and indicates possible mixing into the stratosphere. We observe a large volcanic plume up to 9000 m which is distributed, spread and stretched within the first three days over the Northern Pacific reaching over Alaska and also eastwards, close to the Hudson Bay, Canada (see Fig. D1). For the layer between 10000 and 11000 m, the results look completely different (see Fig. 11). Here, we observe no signal of the volcanic plume on the first day. However, more areas with volcanic plume influence occur over Alaska reaching towards Canada during the second day. These affected areas increase over time. Thus, also the HYSPLIT dispersion model results support the hypothesis of mixing volcanic emissions into the stratosphere within the first three days. This is an additional indicator for the mixing of SO₂ into the LMS, the process of gas-to-particle conversion starts and forms particulate sulfate aerosol over several weeks. This results in the





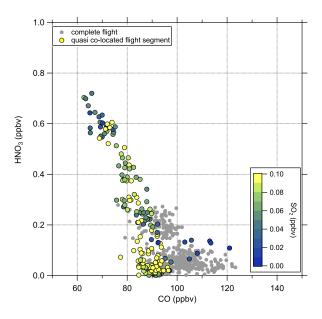


Figure 9. Scatterplot of CO and HNO₃ for the DLR-Falcon flight on 23 May 2020. The quasi co-located measurements with HALO are color-coded with SO2. Here, a mixing event was observed in the HALO data. The DLR-Falcon did not reach the high altitudes of HALO, so the lowest CO values are higher.

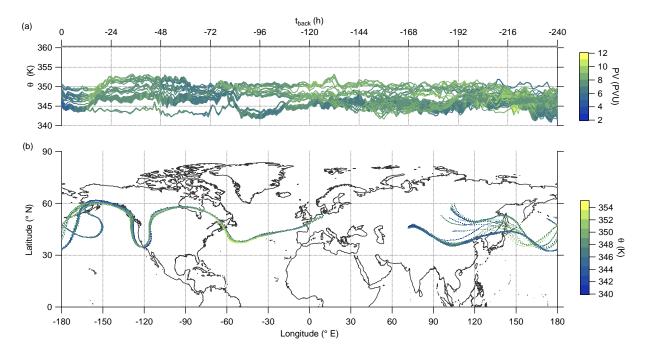


Figure 10. LAGRANTO 10 days back trajectories for cases with enhanced SO_4^{2-} in the lowermost stratosphere. The upper panel (a) shows a time series of θ color-coded with PV. The lower panel (b) shows the position of the trajectories on the map and θ as color-code.





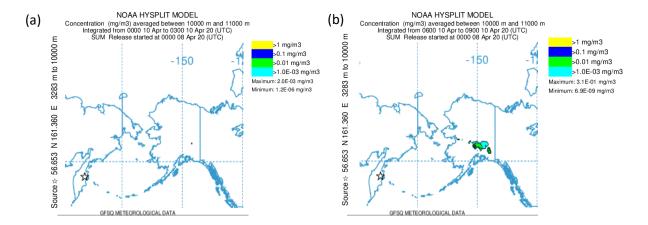


Figure 11. NOAA HYSPLIT dispersion model simulation (Stein et al., 2015) for the Sheveluch eruption on 08 April 2020 on basis of GFS meteorological data. The particle concentration between 10000 and 11000 m is depicted. (a) shows the averaging period on 10 April 2020 from 00 UTC until 03 UTC and (b) shows the next period on 10 April 2020 from 06 UTC until 09 UTC.

observation of higher mixing ratios of particulate sulfate in the LMS roughly seven weeks after the eruption, especially for research flight RF01 where the flight path was close to the jet stream and crossed one filament of volcanically influenced air masses.

4 Conclusions

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Usually trace gas correlations, such as O_3 -CO or O_3 -H₂O, have been used to study mixing processes between the troposphere and stratosphere. In our study, we showed that in addition to trace gas measurements, also aerosol measurements, especially particulate sulfate, can be applied to identify troposphere-stratosphere exchange. Furthermore, we showed that the method to define the chemical tropopause proposed by Zahn et al. (2004) is in agreement with the dynamical tropopause definition for our campaign and thus suitable for the separation of stratospheric and tropospheric air masses. Similar to the correlation between CO and O_3 , the correlation of SO_4^{2-} and O_3 in the lowermost stratosphere also showed some variability induced by mixing events. In a case study during the CAFE-EU/BLUESKY mission, we observed air masses with high sulfate mixing ratios in the lowermost stratosphere, reaching values that are typically found at higher altitudes in the stratospheric aerosol layer. Meteorological and dynamical parameters such as vertical wind shear and gradient Richardson number indicated that mixing across the tropopause occurred in this region and along the transport and air mass history. Additionally, we found that this anomaly of higher particulate sulfate in the lowermost stratosphere occurred during one single flight. During this flight, we found one single mixing line in the CO-O₃ correlation with increasing sulfate aerosol mixing ratios towards the stratosphere. In addition, we used measurements of the quasi co-located DLR-Falcon aircraft. In the same tracer-tracer correlation framework, using HNO₃ instead of O₃, we also found one mixing line. In contrast to the increasing sulfate for the HALO measurements, we observed decreasing mixing ratios of SO₂ which is a precursor gas for particulate sulfate. This led to the hypothesis of upward



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mixing of precursor gases and on-going gas-to-particle conversion in the lowermost stratosphere. Here, we could identify volcanic activities on the Kamchatka Peninsula, Russia, and the explosive eruption of the Sheveluch volcano as a possible source for the SO₂. The Sheveluch eruption injected SO₂ directly into the upper troposphere from where it was mixed into the stratosphere, with subsequent gas-to-particle conversion to sulfate aerosol. We can thus conclude that the chemical composition of the aerosol in the lowermost stratosphere is affected by small-scale mixing processes, and that the ExTL can thus also be characterized by aerosol properties. In addition to direct mixing of aerosol particles, the process of mixing of precursor gases with subsequent gas-to-particle conversion needs also to be considered, as we showed in our case study. We intend to use this method in the future with data obtained during previous airborne measurements in the UTLS to extend the analysis to a larger scale. Furthermore, we aim to compare the results with chemistry climate model studies to see whether chemical transport models can represent small-scale mixing across the tropopause and the associated gas-to-particle conversion processes. Another study should investigate how models represent the influence of volcanic eruptions on the lowermost stratosphere.

Data availability. Data measured onboard HALO are available on request at the HALO database at https://halo-db.pa.op.dlr.de/mission/120. The DLR-Falcon basic aircraft data and the AIMS data are available on request at the HALO database at https://halo-db.pa.op.dlr.de/mission/119 (Re3data.Org, 2016) or on request to the authors.

Author contributions. PJ set up the study together with JS, performed the data analysis and prepared the manuscript. KK and JS provided the sulfate data and supported the analysis. LT, AM and CV provided the SO₂ and HNO₃ data. HS provided CO and O₃ data from DLR-Falcon. AZ provided the O₃ data from HALO. HF and LR provided CO data from HALO. HCL and DK provided model data and backward trajectories as well as the code for the cross sections along the flight path. All co-authors commented on the manuscript and discussed the presented results.

Competing interests. The contact author has declared that none of the authors has any competing interests.

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 $The \ authors \ thank \ NOAA \ Air \ Resources \ Laboratory \ for \ providing \ the \ HYSPLIT \ dispersion \ calculations \ on \ their \ website.$





Appendix A: Supporting information about the 2-D binned cross sections

In the following, we explain the adjustment of our bin scheme for the two dimensional binning analysis. Figure A1b shows the data distribution for the evenly distributed bin scheme in the vertical. Here, we see that many bins in the LMS contain less than 10 data points, so we could expect some bias in the median. Therefore, we enlarge the vertical bins at 350 K and above to 10 K to include more data points in one vertical bin and increase the statistic of the analysis without losing information.

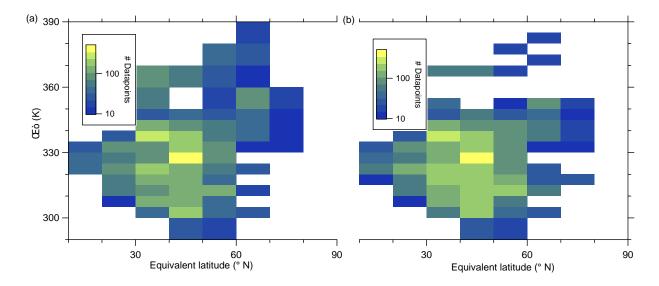


Figure A1. Number of data points in the 2-D cross sections (used in Fig. 5 and Fig. 6). (a) The used bin scheme with larger vertical bins starting at $\theta > 350$ K. (b) The evenly data distribution without any adjustment on the bin scheme.





Appendix B: Supporting information about the meteorological cross sections

This section offers some meteorological analysis for the flight segment with the observed particulate sulfate anomaly (see Fig. B1). Afterwards, we show some additional data along the back trajectories. More precisely, we show the time series of the Richardson number to indicate potential instable regions the air masses have crossed before the measurement (see Fig. B2). First we want to introduce the used variables for our stability analysis, similar to e.g., Kaluza et al. (2019) or Kunkel et al. (2019):

$$N^2 = -\frac{g}{\rho_0} \frac{\partial \rho(z)}{\partial z} \quad \text{squared Brunt-V\"{a}is\"{a}l\"{a} frequency} \tag{B1}$$

$$S^{2} = \left(\frac{\partial u}{\partial z}\right)^{2} + \left(\frac{\partial v}{\partial z}\right)^{2} \quad \text{vertical wind shear}$$
 (B2)

$$Ri = \frac{N^2}{S^2}$$
 gradient Richardson number (B3)

335 Turbulence may occur at a gradient Richardson number lower than 0.25

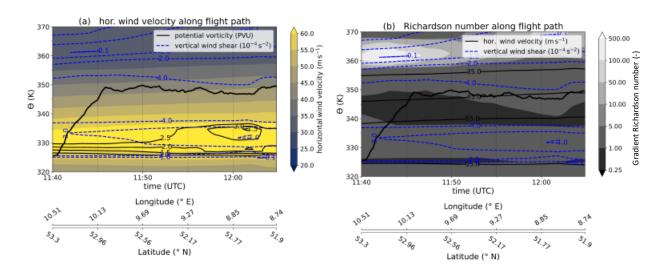


Figure B1. Meteorological cross sections along the flight path, calculated on the basis of ERA5-reanalysis data (Hersbach et al., 2020). (a) contains the horizontal wind speed as filled contour, the flight altitude as black bold solid line, the PV contour as thin black lines and the vertical wind shear as dashed blue contour lines. In (b) the filled contour changed to the gradient Richardson number and the thin black lines are contour lines of the horizontal wind speed. Latitude and longitude values are added to the time series for reference.

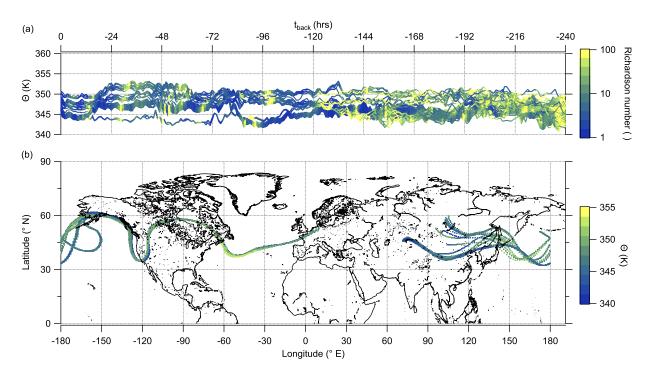


Figure B2. LAGRANTO 10 days back trajectories for cases with enhanced SO_4^{2-} in the lowermost stratosphere starting in Northern Germany. (a) Time series of θ color-coded with the gradient Richardson number as marker for potentially instable regions along the trajectories. (b) The position of the trajectories on the map and the potential temperature as color-code.





Appendix C: Supporting information about the sulfate anomaly

The following figures are supporting information on the observed anomaly of higher mixing ratios of particulate sulfate. This includes a detailed view on the research flight RF01 where the anomaly was observed (Fig. C1). Figure C2 locates the flight segment of the anomaly on the map including the quasi co-located DLR-Falcon flight path and the sulfur dioxide mixing ratio measured on the DLR-Falcon. The last figure (Fig. C3) highlights two selected research flights and their corresponding accuracy within the correlation of the complete data set.

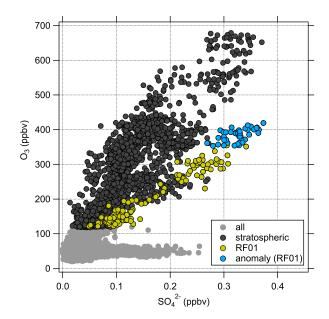


Figure C1. Scatter plot of the in-situ measured SO_4^{2-} and O_3 containing/including the whole campaign data in bright grey. Stratospheric data points are colored in dark grey. The mixing event of research flight RF01 is indicated in yellow and the sulfate anomaly identified during this flight in blue.





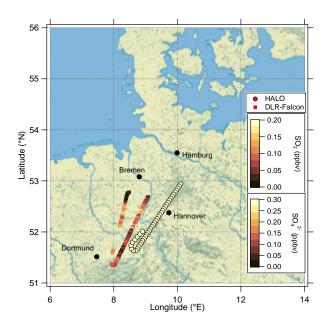


Figure C2. Location of the analyzed flight segment during the research flight on 23 May 2020 between 11:45 and 12:05 UTC where the mixing event in the vicinity of the jetstream was observed. The flight path of HALO are the filled circles with color-coded with the sulfate mixing ratio and the flight path of DLR-Falcon are the filled squares color-coded with the SO₂ mixing ratio. The map was created from public domain GIS data found on the Natural Earth website (http://www.naturalearthdata.com; last access: 02 January 2024)

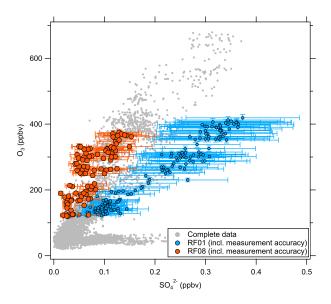


Figure C3. Scatterplot of particulate sulfate against O_3 for the complete campaign data set. The data points for flights RF01 and RF08 are highlighted together with their measurement accuracy as examples.





Appendix D: Supporting information about the HYSPLIT dispersion simulation

The figure shown in this section is in addition to Fig. 11 and shows the similar variables but for the altitude range from sea level up to $9000 \,\mathrm{m}$ to show the entire volcanic main plume and its distribution.

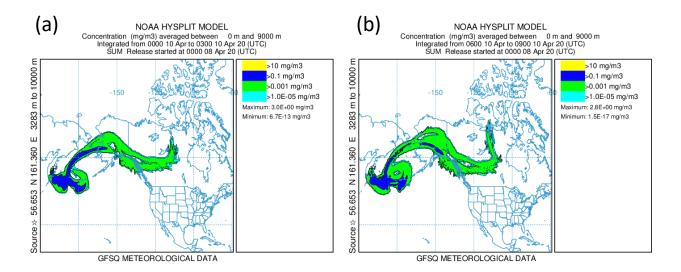


Figure D1. NOAA HYSPLIT dispersion model simulation for the Sheveluch eruption on 08 April 2020 on basis of GFS meteorological data (Stein et al., 2015). The location of Sheveluch volcano is given by a little star. Further, the particle concentration averaged between sea level and 9000 m is shown. (a) includes the averaging period on 10 April 2020 from 00 UTC to 03 UTC and (b) shows the next period on the same day from 06 UTC to 09 UTC.





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