

# 1 **Measurement report: Oxidation potential of water-soluble** 2 **aerosol components in the southern and northern of Beijing**

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## 14 15 **Abstract**

16 Water-soluble components have significant contribution to the oxidative  
17 potential (OP) of atmospheric fine particles (PM<sub>2.5</sub>), while our understanding of  
18 water-soluble PM<sub>2.5</sub> OP and its sources, as well as its relationship with water-soluble  
19 components, is still limited. In this study, the water-soluble OP levels in wintertime  
20 PM<sub>2.5</sub> in the south and north of Beijing, representing the difference in sources, were  
21 measured with dithiothreitol (DTT) assay. The volume normalized DTT (DTT<sub>v</sub>) in the  
22 north ( $3.5 \pm 1.2 \text{ nmol min}^{-1} \text{ m}^{-3}$ ) was comparable to that in the south ( $3.9 \pm 0.9 \text{ nmol}$   
23  $\text{min}^{-1} \text{ m}^{-3}$ ), while the mass normalized DTT (DTT<sub>m</sub>) in the north ( $65 \pm 28 \text{ pmol min}^{-1}$   
24  $\mu\text{g}^{-3}$ ) was almost twice that in the south ( $36 \pm 14 \text{ pmol min}^{-1} \mu\text{g}^{-3}$ ). In both the south  
25 and north of Beijing, DTT<sub>v</sub> was better correlated with soluble elements instead of total  
26 elements. In the north, soluble elements (mainly Mn, Co, Ni, Zn, As, Cd and Pb) and  
27 water-soluble organic compounds, especially light-absorbing compounds (also known  
28 as brown carbon), had positive correlations with DTT<sub>v</sub>. However, in the south, the

29 DTT<sub>v</sub> was mainly related to soluble As, Fe and Pb. The sources of DTT<sub>v</sub> were further  
30 resolved using the positive matrix factorization (PMF) model. Traffic-related  
31 emissions (39%) and biomass burning (25%) were the main sources of DTT<sub>v</sub> in the  
32 south, and traffic-related emissions (> 50%) contributed the most of DTT<sub>v</sub> in the north.  
33 Our results indicate that vehicle emission was the important contributor to OP in  
34 Beijing ambient PM<sub>2.5</sub> and suggest that more study is needed to understand the  
35 intrinsic relationship between OP and light absorbing organic compounds.

36

## 37 **1 Introduction**

38 Atmospheric fine particulate matter (PM<sub>2.5</sub>) pollution is one of the major global  
39 environmental issues, affecting air quality, climate and human health (Huang et al.,  
40 2014; Burnett et al., 2018; An et al., 2019; Zheng et al., 2020). The exposure to PM<sub>2.5</sub>  
41 was estimated to be responsible for 8.9 million deaths worldwide in 2015, of which  
42 28% occurred in China (Burnett et al., 2018). Numerous studies have shown that  
43 oxidative stress is one possible mechanisms underlying the adverse effects of PM<sub>2.5</sub>  
44 on human health (Chowdhury et al., 2019; Lelieveld et al., 2021; Yu et al., 2022b;  
45 Guascito et al., 2023). When entering the human body, PM<sub>2.5</sub> can induce the  
46 production of excessive reactive oxygen species (ROS) (e.g., H<sub>2</sub>O<sub>2</sub>, ·OH and ·O<sub>2</sub><sup>-</sup>),  
47 leading to cellular redox imbalance and generating oxidative stress effects. The ability  
48 of PM<sub>2.5</sub> to cause oxidative stress is defined as oxidative potential (OP).

49 The methods to determine the OP of PM<sub>2.5</sub> include cellular and acellular assays,  
50 and acellular methods are more widely used than cellular methods (Charrier and  
51 Anastasio, 2012; Xiong et al., 2017; Calas et al., 2018; Bates et al., 2019; Wang et al.,  
52 2020b; Campbell et al., 2021; Oh et al., 2023). Among acellular methods, the  
53 dithiothreitol (DTT) assay is extensively applied to determine the OP of ambient  
54 particles (Charrier and Anastasio, 2012; Xiong et al., 2017; Liu et al., 2018; Wang et  
55 al., 2020b; Puthussery et al., 2022; Wu et al., 2022a). DTT is a surrogate of cellular  
56 reductants, and the consumption rate of DTT was used to assess the OP of PM<sub>2.5</sub>.  
57 Previous studies have shown that organic matters (e.g., water-soluble organic species

58 and PAHs) and some transition metals (e.g., Mn and Cu) are the important  
59 contributors to DTT consumption of PM<sub>2.5</sub> (Charrier and Anastasio, 2012; Verma et al.,  
60 2015; Bates et al., 2019; Wu et al., 2022a; Wu et al., 2022b). For example, Charrier  
61 and Anastasio (2012) measured the OP of PM<sub>2.5</sub> in San Joaquin Valley, California and  
62 reported that about 80% of DTT consumption was contributed by transition metals.  
63 Verma et al. (2015) measured the OP of water-soluble PM<sub>2.5</sub> in the southeastern  
64 United States and reported that about 60% of DTT activity was contributed by water-  
65 soluble organics. The mixtures of metals and organics may produce synergistic or  
66 antagonistic effects, such as  $\cdot\text{O}_2^-$  produced from oxidation of DTT by quinones is  
67 more efficiently transformed to  $\cdot\text{OH}$  in the presence of Fe, while the DTT  
68 consumption and  $\cdot\text{OH}$  generation of quinones are reduced in the presence of Cu  
69 (Xiong et al., 2017; Yu et al., 2018; Bates et al., 2019).

70 A number of studies have investigated the OP of water-soluble components in  
71 PM<sub>2.5</sub>, which show that the average water-soluble OP values in urban areas ranged  
72 from 0.1 to 10 nmol min<sup>-1</sup> m<sup>-3</sup> (Fang et al., 2016; Liu et al., 2018; Chen et al., 2019;  
73 Wu et al., 2022a; Yu et al., 2022a; Xing et al., 2023). Due to the complexity in  
74 chemical composition and sources of PM<sub>2.5</sub> that determine the OP levels, the sources  
75 of OP are also diverse (Verma et al., 2015; Bates et al., 2019; Tuet et al., 2019; Yu et  
76 al., 2019; Cao et al., 2021). Several studies have investigated the emission sources and  
77 ambient samples to identify the sources of OP (Tuet et al., 2019; Yu et al., 2019; Wang  
78 et al., 2020b; Cao et al., 2021), which include both primary and secondary sources.  
79 For example, Cao et al. (2021) measured the water-soluble OP of PM<sub>2.5</sub> samples from  
80 six biomass and five coal burning emissions in China, with average values of 4.5-7.4  
81 and 0.5-2.1 pmol min<sup>-1</sup> μg<sup>-1</sup>, respectively. Tong et al. (2018) investigated the OP of  
82 secondary organic aerosols (SOA) from oxidation of naphthalene, isoprene and β-  
83 pinene with  $\cdot\text{OH}$  or O<sub>3</sub>, which were 104 ± 7.6, 48 ± 7.9 and 36 ± 3.1 pmol min<sup>-1</sup> μg<sup>-1</sup>,  
84 respectively. Verma et al. (2014) identified the sources of water-soluble OP of PM<sub>2.5</sub>  
85 in Atlanta, United States from June 2012 to September 2013 with positive matrix  
86 factorization (PMF) and chemical mass balance (CMB) methods, of which biomass

87 burning was the largest contributor. Wang et al. (2020b) quantified the sources of  
88 water-soluble OP of PM<sub>2.5</sub> in Xi'an, China in 2017 using PMF and multiple linear  
89 regression (MLR) methods, with significant contributions from secondary sulfates,  
90 vehicle emissions and coal combustion. Some studies have also measured the OP of  
91 particles with different particle sizes, and reported that smaller size fractions typically  
92 have higher ROS activity compared to large PM size fractions (Saffari et al., 2014;  
93 Shafer et al., 2016; Basis et al., 2023). For example, Basis et al. (2023) measured the  
94 OP of water-soluble fraction of size segregated PM (< 0.49, 0.49-0.95, 0.95-1.5, 1.5-  
95 3.0, 3.0-7.2 and > 7.2 μm) collected during the cold and warm periods at an urban site  
96 in Thessaloniki, northern Greece, and the results showed that the total DTT activity of  
97 the PM < 3 μm size fraction were higher (2-5 times) than that of PM > 3 μm size  
98 fraction in both warm and cold periods. Despite these efforts, comparative studies on  
99 the differences in pollution levels and sources of PM<sub>2.5</sub> OP in different districts are  
100 still limited.

101 In this study, the DTT activity of water-soluble matter in PM<sub>2.5</sub> samples collected  
102 simultaneously in the southern and northern of Beijing in January 2018 were  
103 measured. The concentration and light absorption of water-soluble organic carbon  
104 (WSOC), as well as the concentrations of 14 trace elements and 7 light-absorbing  
105 nitroaromatic compounds (NACs) were quantified. The sources of DTT activity were  
106 then identified with PMF model. The results acquired in this study provide a  
107 comparison of PM<sub>2.5</sub> OP in different districts of Beijing and its connection with  
108 organic compounds, trace elements and sources, which could be helpful for further  
109 study of the regional differences in the effects of PM<sub>2.5</sub> on human health.

110

## 111 **2 Materials and methods**

### 112 **2.1 Sampling**

113 Ambient 24 h integrated PM<sub>2.5</sub> filter samples were collected from January 1 to 31,  
114 2018 simultaneously in the south (the Dingfuzhuang village (DFZ), Daxing district;  
115 39.61°N, 116.28°E) and north (the National Center for Nanoscience and Technology

116 (NCNT), Haidian district; 39.99°N, 116.32°E) of Beijing (Figure S1). The distance  
117 between the two sampling sites is about 42 km. The south site is surrounded by  
118 agricultural, industrial, and transportation areas, and the north site is surrounded by  
119 residential, transportation and commercial areas. PM<sub>2.5</sub> samples were collected on pre-  
120 baked (780 °C, 3 h) quartz-fiber filters (20.3 × 25.4 cm; Whatman, QM-A, Clifton, NJ,  
121 USA) using high-volume PM<sub>2.5</sub> samplers (1.13 m<sup>3</sup> min<sup>-1</sup>; Tisch, Cleveland, OH, USA)  
122 which were placed on the roof of buildings at heights of about 5 m (south) and 20 m  
123 (north) above the ground. 31 samples were collected at each site. After collection, the  
124 samples were wrapped in baked aluminum foils and stored in a freezer (-20 °C) until  
125 further analysis.

## 126 **2.2 Chemical analysis**

127 The mass of PM<sub>2.5</sub> on the filter was measured by a digital microbalance with a  
128 precision of 0.1 mg (LA130S-F, Sartorius, Germany) after 24-h equilibration at a  
129 constant temperature (20-23 °C) and humidity (35-45%) chamber. Each filter was  
130 weighted at least two times, and the deviations for blank and sampled filters among  
131 the repetitions were less than 5 and 10 µg, respectively. The PM<sub>2.5</sub> mass concentration  
132 was calculated by dividing the weight difference before and after sampling by the  
133 volume of sampled air.

134 For WSOC analysis, one punch (1.5 cm<sup>2</sup> for concentration analysis and 0.526  
135 cm<sup>2</sup> for light absorption measurement) of filter was taken from each sample and  
136 extracted ultrasonically with ultrapure water (> 18.2 MΩ cm) for 30 min. After, the  
137 extracts were filtered with a 0.45 µm PVDF pore syring filter to remove insoluble  
138 substances. Finally, the concentration of WSOC was measured with a total organic  
139 carbon-total nitrogen analyzer (TOC-L, Shimadzu, Japan; (Ho et al., 2015)) and the  
140 light absorption of WSOC was measured by an UV-Vis spectrophotometer (300-700  
141 nm; Ocean Optics, USA) equipped with a liquid waveguide capillary cell (LWCC-  
142 3100, World Precision Instruments, Sarasota, FL, USA; (Yuan et al., 2020)). The  
143 absorption coefficient (Abs) of WSOC was calculated according to formula S1 in the  
144 Supporting Information (SI).

145 The total concentration and soluble fraction concentration of 14 trace elements  
146 (i.e., Ti, V, Cr, Mn, Fe, Co, Ni, Cu, Zn, As, Sr, Cd, Ba, and Pb) were quantified by an  
147 inductively coupled plasma mass spectrometer (ICP-MS, 7700x, Agilent Technologies,  
148 USA), and the details are shown in the SI. For soluble fraction concentration analysis,  
149 a punch of filter (47 mm diameter) was extracted with ultrapure water and then  
150 centrifuged from residues. For total concentration analysis, another 47 mm diameter  
151 filter of the same sample was used and digested with 10 mL HNO<sub>3</sub> and 1 mL HF at  
152 180 °C for 12 h. The extracts were then heated and concentrated to ~ 0.1 mL, and  
153 diluted to 5 mL with 2% HNO<sub>3</sub>. Afterwards, the diluents were filtered with a 0.22 μm  
154 PTFE pore syring filter and stored in a freezer (-4 °C) until further ICP-MS analysis.

155 The concentrations of organic markers (including levoglucosan, mannosan,  
156 galactosan, hopanes (including 17α(H)-22,29,30-trisnorhopane, 17α(H),21β(H)-30-  
157 norhopane, 17β(H),21α(H)-30-norhopane, 17β(H),21α(H)-hopane, 17β(H),21α(H)-  
158 hopane and 17β(H),21α(H)-hopane), picene, phthalic acid, isophthalic acid and  
159 terephthalic acid) and light-absorbing NACs (including 4-nitrophenol (4NP), 2-  
160 methyl-4-nitrophenol (2M4NP), 3-methyl-4-nitrophenol (3M4NP), 4-nitrocatechol  
161 (4NC), 3-methyl-5-nitrocatechol (3M5NC), 4-methyl-5-nitrocatechol (4M5NC) and  
162 4-nitro-1-naphthol (4N1N)) were determined by a gas chromatograph–mass  
163 spectrometer (GC-MS; Agilent Technologies, Santa Clara, CA, USA) following the  
164 method described elsewhere (Wang et al., 2020a), and more details about the analysis  
165 can be found in SI. All of the results reported in this study were corrected for blanks.

### 166 **2.3 Oxidative potential**

167 The DTT assay was applied to determine the oxidative potential of water-soluble  
168 components in PM<sub>2.5</sub> according to the method by Gao et al. (2017). In brief, a quarter  
169 of a 47 mm filter was ultrasonically extracted with 5 mL ultrapure water for 30 min  
170 and then filtered with a 0.45 μm PVDF pore syring filter to remove insoluble  
171 substances. Several studies have shown that ultrasonic treatment of samples can lead  
172 to an increase in their OP values (Miljevic et al., 2014; Jiang et al., 2019), however,  
173 there was also a study showed that the difference in OP values of water-soluble PM<sub>2.5</sub>

174 measured by DTT assay was small for samples extracted by ultrasonic and shaking  
175 (Gao et al., 2017). Consistent with the extraction methods for organic markers and  
176 trace elements, ultrasonic method was used to extract samples for DTT analysis.  
177 Afterwards, 0.5 mL of the extract was mixed with 1 mL of potassium phosphate  
178 buffer (pH = 7.4) and 0.5 mL of 2 mM DTT in a brown vial, and then placed in a  
179 water bath at 37 °C. Then, 20  $\mu$ L of this mixture was taken at designated time  
180 intervals (2, 7, 13, 20, and 28 min) and mixed with 1 mL trichloroacetic acid (TCA;  
181 1% w/v) in another brown vial to terminate the reaction. Then, 0.5 mL of 5,5'-  
182 dithiobis-(2-nitrobenzoic acid) (DTNB; 2.5  $\mu$ M) and 2 mL of tris buffer (pH = 8.9)  
183 were added to form 2-nitro-5-thiobenzonic acid (TNB) which has light absorption at  
184 412 nm. Finally, the absorption of TNB was measured by a LWCC-UV-Vis. The DTT  
185 consumption rate was quantified by the remaining DTT concentration at different  
186 reaction times. Daily solution blanks and filter blanks were analyzed in parallel with  
187 samples to evaluate the consistency of the system performance. Besides, for every 10  
188 samples, one sample was chosen to be measured three times to check the  
189 reproducibility, and the relative standard deviation was lower than 5%. Ambient  
190 samples were corrected for filter blank. The DTT activities were normalized by the  
191 volume of sampled air ( $\text{DTT}_v$ ,  $\text{nmol min}^{-1} \text{m}^{-3}$ ) and the mass concentration of  $\text{PM}_{2.5}$   
192 ( $\text{DTT}_m$ ,  $\text{pmol min}^{-1} \mu\text{g}^{-1}$ ).

193 Considering that for samples containing a significant amount of substances  
194 whose DTT response is non-linear with  $\text{PM}_{2.5}$  concentration (e.g., Cu, Mn), the  $\text{DTT}_m$   
195 value depends on the concentration of  $\text{PM}_{2.5}$  added to the reaction solution (Charrier  
196 et al., 2016). The response of  $\text{DTT}_m$  to  $\text{PM}_{2.5}$  concentration added to the reaction  
197 solution was analyzed using sample containing high concentrations of soluble Cu and  
198 Mn (Figure S2). When the  $\text{PM}_{2.5}$  concentration added to the reaction solution is less  
199 than  $150 \mu\text{g mL}^{-1}$ , the  $\text{DTT}_m$  response is greatly affected by the difference in added  
200  $\text{PM}_{2.5}$  concentration; however, when the  $\text{PM}_{2.5}$  concentration added to the reaction  
201 solution is greater than  $150 \mu\text{g mL}^{-1}$ , the  $\text{DTT}_m$  response is less affected by the  
202 difference in  $\text{PM}_{2.5}$  concentration ( $< 12\%$ ). In this study, the concentration of  $\text{PM}_{2.5}$

203 added to the reaction solution of most samples from the two sites was greater than 150  
204  $\mu\text{g mL}^{-1}$  (ranged from 79 to 749  $\mu\text{g mL}^{-1}$ , with an average of  $409 \pm 164$  and  $207 \pm 95$   
205  $\mu\text{g mL}^{-1}$  in the south and north, respectively), therefore, the difference in  $\text{PM}_{2.5}$   
206 concentration added to the reaction solution of different samples should had a  
207 relatively small impact on the difference in  $\text{DTT}_m$  values of different samples. This  
208 study did not consider the impact of metal precipitation in phosphate matrix on the  
209 measured DTT values, as there is not a straightforward method to correct the artifact  
210 caused by this phenomenon (Yalamanchili et al., 2023).

## 211 **2.4 Source apportionment**

212 The sources of DTT activities were identified and quantified using PMF model  
213 implemented by the multilinear engine (ME-2; (Paatero, 1997)) following the method  
214 described in our previous studies (Huang et al., 2014; Yuan et al., 2020). For each site,  
215 31 samples (a total of 62 samples) and 23 species were input into PMF model. The  
216 number of samples is higher than the number of species. The input data include  
217 species concentration (including  $\text{DTT}_v$ , 14 trace elements and 8 organic markers) and  
218 uncertainties. The species-specific uncertainties were calculated following Liu et al.  
219 (2017). For a clear separation of sources profiles, the contribution of corresponding  
220 markers was set to 0 in the sources unrelated to the markers (see Table S1). More  
221 details are described in SI (PMF analysis).

222

## 223 **3 Results and discussion**

### 224 **3.1 DTT activity and concentrations of water-soluble $\text{PM}_{2.5}$ components**

225 Figure 1 shows the daily variation of DTT activity, light absorption of WSOC at  
226 wavelength 365 nm ( $\text{Abs}_{365}$ ), together with the concentrations of  $\text{PM}_{2.5}$ , WSOC,  
227 NACs and total elements in the south and north of Beijing. Their average values are  
228 shown in Table S2. Generally, the average values of  $\text{PM}_{2.5}$ , WSOC,  $\text{Abs}_{365}$ , NACs and  
229 total elements were higher in the south than in the north. Specifically, the  
230 concentrations of  $\text{PM}_{2.5}$  and WSOC in the south ( $122 \pm 49 \mu\text{g m}^{-3}$  and  $8.1 \pm 5.0 \mu\text{gC}$   
231  $\text{m}^{-3}$ , respectively) were both about two times higher than that in the north ( $62 \pm 28 \mu\text{g}$



232  $\text{m}^{-3}$  and  $4.0 \pm 2.0 \mu\text{gC m}^{-3}$ , respectively), indicating that the proportion of WSOC in  
233  $\text{PM}_{2.5}$  was similar in the south and north. However, the  $\text{Abs}_{365}$  in the south was about  
234 three times that in the north, indicating that the chemical composition of WSOC was  
235 different between the south and north. Previous studies have reported that NACs are  
236 the main water-soluble light-absorbing organic compounds (also known as brown  
237 carbon, BrC) of  $\text{PM}_{2.5}$  (Lin et al., 2017; Huang et al., 2020; Li et al., 2020). For the 7  
238 NACs quantified in this study, the total concentration of nitrophenols (4NP, 2M4NP  
239 and 3M4NP), nitrocatechols (4NC, 3M5NC and 4M5NC), and 4N1N in the south  
240 ( $108 \pm 73 \text{ ng m}^{-3}$ ,  $118 \pm 91 \text{ ng m}^{-3}$  and  $12 \pm 8.2 \text{ ng m}^{-3}$ , respectively) was about three,  
241 five and four times, respectively, those in the north ( $35 \pm 22 \text{ ng m}^{-3}$ ,  $24 \pm 30 \text{ ng m}^{-3}$   
242 and  $3.1 \pm 3.0 \text{ ng m}^{-3}$ , respectively). These results indicate that the sources and  
243 emission strength of water-soluble organic compounds were different in the south and  
244 north of Beijing, suggesting the different contribution of water-soluble organic  
245 compounds to DTT activity. The concentration trends of total trace elements were also  
246 different between the south and north of Beijing, with  $\text{Fe} > \text{Zn} > \text{Ti} > \text{Mn} > \text{Cu} > \text{Ba} >$   
247  $\text{Pb} > \text{Sr} > \text{Cr} > \text{As} > \text{V} > \text{Ni} > \text{Cd} > \text{Co}$  in the south, and  $\text{Fe} > \text{Ti} > \text{Zn} > \text{Ba} > \text{Mn} >$   
248  $\text{Pb} > \text{Cu} > \text{Cr} > \text{Sr} > \text{As} > \text{Ni} > \text{V} > \text{Cd} > \text{Co}$  in the north. It should be noted that  
249 although the contents of  $\text{PM}_{2.5}$ , WSOC and total elements measured in this study were  
250 higher in the south than in the north, the average  $\text{DTT}_v$  value in the south ( $3.9 \pm 0.9$   
251  $\text{nmol min}^{-1} \text{ m}^{-3}$ ) was comparable to that in the north ( $3.5 \pm 1.2 \text{ nmol min}^{-1} \text{ m}^{-3}$ ),  
252 meanwhile, the average  $\text{DTT}_m$  value was much higher (1.8 times) in the north ( $65 \pm$   
253  $28 \text{ pmol min}^{-1} \mu\text{g}^{-1}$ ) than in the south ( $36 \pm 14 \text{ pmol min}^{-1} \mu\text{g}^{-1}$ ). Ahmad et al. (2021)  
254 also reported that the concentrations of  $\text{PM}_{2.5}$ , WSOC, and most elements in Lahore,  
255 Pakistan, were higher than those in Peshawar, Pakistan, while the  $\text{DTT}_v$  values of the  
256 two sites were similar, and the  $\text{DTT}_m$  value in Peshawar was higher than that in  
257 Lahore. The lower  $\text{DTT}_m$  in the south than in the north may be due to the increased  
258  $\text{PM}_{2.5}$  in the south containing more substances with no or little contribution to DTT  
259 activity, and indicates that the intrinsic OP of water-soluble components of  $\text{PM}_{2.5}$  was  
260 higher in the north than in the south. The similar  $\text{DTT}_v$  values in the south and north

261 indicate that the exposure-relevant OP of water-soluble components of PM<sub>2.5</sub> was  
262 comparable in the two sites, and the water-soluble DTT<sub>v</sub> was not consistent with the  
263 content of water-soluble substances. Due to the complex chemical composition of  
264 PM<sub>2.5</sub>, there may also be antagonistic and synergistic effects, contributing to the  
265 inconsistent relationship between DTT activity and compounds content (Xiong et al.,  
266 2017; Lionette et al., 2021).

267 Figure 2 shows the comparison of water-soluble PM<sub>2.5</sub> DTT activity measured in  
268 this study with those measured in other regions of Asia during similar periods. It can  
269 be seen that the DTT<sub>v</sub> values measured in Beijing in this study were lower than that in  
270 Jinzhou, Tianjin, Yantai, and Shanghai in China, Lahore and Peshawar in Pakistan,  
271 and Delhi in India (Liu et al., 2018; Ahmad et al., 2021; Puthussery et al., 2022; Wu et  
272 al., 2022a), higher than that in Xi'an, Nanjing, Hangzhou, Guangzhou, and Shenzhen  
273 in China (Wang et al., 2019; Wang et al., 2020b; Ma et al., 2021; Yu et al., 2022c;  
274 Xing et al., 2023), and comparable with that in Ningbo, China (Chen et al., 2022).  
275 Different from DTT<sub>v</sub>, the DTT<sub>m</sub> value measured in NCNT in Beijing was similar with  
276 that in Jinzhou, Tianjin, Yantai, Shanghai and Ningbo in China (Liu et al., 2018; Chen  
277 et al., 2022; Wu et al., 2022a), and higher than that in other regions. The differences in  
278 water-soluble DTT activity of PM<sub>2.5</sub> in different regions can be explained by the  
279 differences in chemical composition, sources and atmospheric formation processes  
280 (Tong et al., 2017; Wong et al., 2019; Daellenbach et al., 2020; Wang et al., 2020b;  
281 Cao et al., 2021). For example, Cao et al. (2021) reported the water-soluble DTT  
282 activity of PM<sub>2.5</sub> from biomass and coal burning emissions in China, and the average  
283 value of biomass burning (4.5-7.4 pmol min<sup>-1</sup> μg<sup>-1</sup>) was much higher than that of coal  
284 burning (0.5-2.1 pmol min<sup>-1</sup> μg<sup>-1</sup>). Tuet et al. (2017) measured the water-soluble DTT  
285 activity of SOA generated under different precursors and reaction conditions, with  
286 SOA from naphthalene photooxidation under RO<sub>2</sub> + NO-dominant dry reaction  
287 conditions had the highest DTT activity.

### 288 **3.2 Correlation between DTT activity and water-soluble PM<sub>2.5</sub> components**

289 Figure 3 shows the correlations of DTT<sub>v</sub> with PM<sub>2.5</sub>, WSOC and Abs<sub>365</sub> in the

290 south and north of Beijing. It can be seen that the correlation coefficient between  
291  $DTT_v$  and  $PM_{2.5}$  was moderate in both the south ( $r = 0.42$ ) and north ( $r = 0.45$ ),  
292 indicating that the OP of particles cannot be evaluated solely by the total  $PM_{2.5}$   
293 concentration. The correlations between  $DTT_v$  with WSOC and  $Abs_{365}$  were strong in  
294 the north ( $r$  of 0.69 and 0.70, respectively), while relatively weak in the south ( $r$  of  
295 0.41 and 0.40, respectively). The high correlations between  $DTT_v$  with WSOC and  
296  $Abs_{365}$  in the north of Beijing qualitatively agree with previous studies in Xi'an, China  
297 and Atlanta, United States (Verma et al., 2012; Chen et al., 2019), and suggest that  
298 water-soluble organic matter, especially BrC, has a significant contribution to DTT  
299 consumption in the north. Light-absorbing BrC typically has conjugated electrons,  
300 making it more likely to transport electrons for catalytic reactions, thereby  
301 contributing to DTT activity (Chen et al., 2019; Wu et al., 2022). Further, in the north,  
302 the  $DTT_v$  was closely related to the concentrations of NACs ( $r$  of 0.57 to 0.79) (Figure  
303 S3), suggesting that NACs may be important contributors to DTT consumption. Feng  
304 et al. (2022) reported the positive correlations between NACs and biomarkers in  
305 saliva and urine (interleukin-6 and 8-hydrox-2'-deoxyguanosine). Zhang et al. (2023)  
306 also reported that NACs are major proinflammatory components in organic aerosols,  
307 contributing about 24% of the interleukin-8 response of all compounds detected by  
308 Fourier transform ion cyclotron resonance mass spectrometry (FT-ICR-MS) in  
309 electrospray ionization negative mode (ESI-). Certainly, it may also be other  
310 substances related to NACs that contribute to the DTT activity, including those not  
311 detected in this study, driving the good correlation between NACs and  $DTT_v$  in the  
312 north of Beijing, which is worth studying in the future.

313 The correlation coefficients between  $DTT_v$  and 14 trace elements are shown in  
314 Figure 4. Generally, the correlations between  $DTT_v$  and soluble elements were higher  
315 than that between  $DTT_v$  and total elements in both the south and north of Beijing. For  
316 soluble elements, in the south, the  $DTT_v$  showed positive correlations with Mn, Fe, Cr,  
317 Co, As and Pb ( $r > 0.5$ ), while in the north, it exhibited strong positive correlations  
318 with Mn, Co, Ni, Zn, As, Cd and Pb ( $r > 0.7$ ), indicating the different sources of  $DTT_v$

319 in the south and north of Beijing. It is worth noting that the concentrations of all  
320 soluble elements were higher in the south than in the north (Figure S4), while the  
321 correlation between  $DTT_v$  and most soluble elements was lower in the south than in  
322 the north (Figure 4). The high correlations between  $DTT_v$  and soluble elements in the  
323 north of Beijing suggests that soluble elements also had a significant contribution to  
324 DTT consumption. The low correlations between  $DTT_v$  and soluble elements in the  
325 south of Beijing may be due to the nonlinear relationship between DTT consumption  
326 and element concentration (Charrier and Anastasio, 2012; Wu et al., 2022a). As shown  
327 in Figure S5, the relationship between most soluble trace elements and  $DTT_v$  was  
328 more non-linear than linear. As the concentration of soluble elements increases, the  
329 growth rate of  $DTT_v$  obviously decreases.

330 In addition to being associated with individual water-soluble species, the  
331 interactions between metals and organic compounds also affect the consumption of  
332 DTT (Xiong et al., 2017; Wu et al., 2022b), with both synergistic and antagonistic  
333 effects. For example, Wu et al. (2022b) measured the DTT consumption of Fe(III) and  
334 Cu(II) interacting with 1,4-naphthoquinone, 9,10-phenanthraquinone, citric acid, and  
335 4-nitrocatechol, respectively. Their results showed that Cu(II) had antagonistic effects  
336 in interacting with most organics except for citric acid, and Fe(III) had an additive  
337 effect on DTT consumption of 1,4-naphthoquinone and citric acid, while it had an  
338 antagonistic effect on 1,4-naphthoquinone and 9,10-phenanthraquinone. Due to the  
339 complex composition of water-soluble organic aerosols, the knowledge about the  
340 effects of organics and metal-organic interactions on DTT activity are still limited,  
341 especially the effects of BrC chromophores and their interactions with metals.

### 342 **3.3 Sources of DTT activity**

343 This study analyzed eight organic markers (including levoglucosan, mannosan,  
344 and galactosan for biomass burning, hopanes for vehicle emissions, picene for coal  
345 combustion, and phthalic acid, isophthalic acid and terephthalic acid for secondary  
346 formation) to help identify the sources of DTT activity. The average concentrations of  
347 these organic markers are shown in Table S2. The correlation coefficients between

348 DTT<sub>v</sub> and organic markers are shown in Figure S6. In the south, levoglucosan,  
349 mannosan, galactosan, and hopanes had moderate correlation with DTT<sub>v</sub> (r of 0.41 to  
350 0.48); phthalic acid, isophthalic acid and terephthalic acid had low to moderate  
351 correlation with DTT<sub>v</sub> (r of 0.28 to 0.54); picene had low correlation with DTT<sub>v</sub> (r of  
352 0.21). These results suggest that biomass burning and vehicle emissions could have  
353 significant contribution to water-soluble PM<sub>2.5</sub> OP in the south. In the north, hopanes  
354 had the highest correlation with DTT<sub>v</sub> (r = 0.70), indicating that vehicle emissions  
355 could have an important contribution. Levoglucosan, mannosan, galactosan, phthalic  
356 acid, isophthalic acid, terephthalic acid, and picene had moderate to high correlations  
357 with DTT<sub>v</sub> in the north, suggesting that biomass and coal burning, and secondary  
358 formation may also have certain contribution to water-soluble PM<sub>2.5</sub> OP.

359 To further quantify the sources of DTT activity in the south and the north of  
360 Beijing, the PMF model, which was widely used for the source apportionment of  
361 PM<sub>2.5</sub> OP (Liu et al., 2018; Shen et al., 2022; Cui et al., 2023), was applied. The input  
362 species include DTT<sub>v</sub>, soluble elements and organic markers, and five to seven factors  
363 were examined. Due to the oil factor mixed with vehicle emissions factor in the five-  
364 factor solution, and there was no new reasonable factor when increasing the factor  
365 number to seven in the PMF analysis (Figure S7). Finally, six factors were resolved  
366 and quantified using PMF model in the south and north of Beijing, including biomass  
367 burning, coal burning, traffic-related, dust, oil combustion, and secondary formation,  
368 and the profiles of these sources are shown in Figure S8. The uncertainties of PMF  
369 analysis for these sources were 2-14%. Factor 1 is characterized by high contribution  
370 of levoglucosan, mannosan, and galactosan, mainly from biomass burning (Huang et  
371 al., 2014; Chow et al., 2022). The DTT activity of biomass burning organic aerosol  
372 was measured by Wong et al. (2019), which was  $48 \pm 6$  pmol min<sup>-1</sup> μg<sup>-1</sup> of WSOC.  
373 Liu et al. (2018) quantified the sources of DTT<sub>v</sub> in coastal cities (Jinzhou, Tianjin, and  
374 Yantai) in China with PMF model and multiple linear regression method, and the  
375 results showed that biomass burning contributed 28% on average in winter. Factor 2  
376 exhibits a large fraction of picene, Zn, Mn, Cd, As, and Pb, which is considered to be

377 coal burning (Huang et al., 2014; Huang et al., 2018). Joo et al. (2018) measured the  
378 DTT activity of PM<sub>2.5</sub> emitted from coal combustion at different temperatures, with  
379 the highest values of  $26 \pm 21 \text{ pmol min}^{-1} \mu\text{g}^{-1}$  and  $0.10 \pm 0.06 \text{ nmol min}^{-1} \text{ m}^{-3}$   
380 occurring at 550 °C. Factor 3 is identified as traffic-related emissions, which is  
381 characterized by the higher loading of hopanes, Ba, Sr, Cu and Ni (Huang et al., 2018;  
382 Chow et al., 2022). Vreeland et al. (2017) measured the DTT activity of PM<sub>2.5</sub> emitted  
383 by side street and highway vehicles in Atlanta, with values of  $0.78 \pm 0.60 \text{ nmol min}^{-1}$   
384  $\text{m}^{-3}$  and  $1.1 \pm 0.60 \text{ nmol min}^{-1} \text{ m}^{-3}$ , respectively. Ting et al. (2023) reported that the  
385 DTT activity of PM<sub>2.5</sub> from vehicle emissions in Ziqing tunnel in Taiwan, China, was  
386 0.15-0.46  $\text{nmol min}^{-1} \text{ m}^{-3}$ . Factor 4, secondary formation, which is identified by high  
387 levels of phthalic acid, isophthalic acid, and terephthalic acid (Al-Naiema and Stone,  
388 2017; Wang et al., 2020a). Verma et al. (2014) reported that secondary formation  
389 contributed about 30% to the water-soluble DTT activity of PM<sub>2.5</sub> in urban Atlanta. It  
390 is worth noting that the DTT activity of SOA generated from different precursors is  
391 different (Tuet et al., 2017; Tong et al., 2018). For example, the DTT activity of SOA  
392 from naphthalene was higher than that from isoprene (Tuet et al., 2017; Tong et al.,  
393 2018). Factor 5 is dominated by crustal elements Fe and Ti, mainly from dust (Huang  
394 et al., 2018). The DTT activity of atmospheric particulate matter during dust periods  
395 were reported in previous studies (Chirizzi et al., 2017; Khoshnamvand et al., 2023)  
396 and it has a low contribution in this study. Factor 6 is identified as oil combustion  
397 because of the high levels of V and Ni (Moreno et al., 2011; Minguillón et al., 2014;  
398 Huang et al., 2018).

399 The source contributions of DTT<sub>v</sub> in the south and north of Beijing are shown in  
400 Figure 5, exhibiting obvious district differences. In the south, traffic-related emissions  
401 (39%) and biomass burning (25%) had the most contribution to DTT<sub>v</sub>, followed by  
402 secondary formation (17%), coal burning (15%), dust (2%), and oil combustion (2%).  
403 In the north, traffic-related emissions (52%) had the highest contribution to DTT<sub>v</sub>,  
404 followed by coal burning (20%), secondary formation (13%), biomass burning (8%),  
405 oil combustion (4%), and dust (3%). The absolute contribution of each source to

406 DTT<sub>v</sub> varies by 1.2-3.4 times between the south and north of Beijing (Table S3). The  
407 large district differences in sources of DTT<sub>v</sub> of water-soluble PM<sub>2.5</sub> call for more  
408 research on the relationship between sources, chemical composition, formation  
409 processes and OP of PM<sub>2.5</sub>.

410

#### 411 **4 Conclusions**

412 In this study, the water-soluble OP of ambient PM<sub>2.5</sub> collected in winter in the  
413 south and north of Beijing were quantified, together with the concentration and light  
414 absorption of WSOC, and concentrations of 7 light-absorbing NACs and 14 trace  
415 elements. The average DTT<sub>v</sub> value was comparable in the south ( $3.9 \pm 0.9 \text{ nmol min}^{-1}$   
416  $\text{m}^{-3}$ ) and north ( $3.5 \pm 1.2 \text{ nmol min}^{-1} \text{ m}^{-3}$ ), while the DTT<sub>m</sub> was higher in the north ( $65$   
417  $\pm 28 \text{ pmol min}^{-1} \mu\text{g}^{-1}$ ) than in the south ( $36 \pm 14 \text{ pmol min}^{-1} \mu\text{g}^{-1}$ ), indicating that the  
418 exposure-relevant OP of water-soluble components of PM<sub>2.5</sub> was similar in the two  
419 sites and that the intrinsic OP of water-soluble components of PM<sub>2.5</sub> was higher in the  
420 north than in the south. The correlation between DTT<sub>v</sub> and soluble elements was  
421 higher than that between DTT<sub>v</sub> and total elements in both the south and north. In the  
422 north, the DTT<sub>v</sub> was strongly correlated with soluble Mn, Co, Ni, Zn, As, Cd and Pb  
423 ( $r > 0.7$ ), and in the south it positively correlated with Mn, Fe, Cr, Co, As and Pb ( $r >$   
424  $0.5$ ). In addition, in the north the DTT<sub>v</sub> was also positively correlated with WSOC,  
425 Abs<sub>365</sub> and NACs ( $r$  of 0.56 to 0.79), while in the south it was weakly correlated ( $r \leq$   
426  $0.4$ ). These results indicate that in the north trace elements and water-soluble organic  
427 compounds, especially BrC chromophores, both had significant contributions to DTT  
428 consumption, and in the south the consumption of DTT may be mainly from trace  
429 elements. Six sources of DTT<sub>v</sub> were resolved with the PMF model, including biomass  
430 burning, coal burning, traffic-related, dust, oil combustion, and secondary formation.  
431 On average, traffic-related emissions (39%) and biomass burning (25%) were the  
432 major contributors of DTT<sub>v</sub> in the south, and traffic-related emissions (52%) was the  
433 predominated source in the north. The differences in DTT<sub>v</sub> sources in the south and  
434 north of Beijing suggest that the relationship between source emissions and

435 atmospheric processes and PM<sub>2.5</sub> OP deserve further exploration in order to better  
436 understand the regional differences of health impacts of PM<sub>2.5</sub>. Besides, in order to  
437 gain a more comprehensive understanding of the regional differences in PM<sub>2.5</sub> OP,  
438 sources and its relationship with chemical composition, longer periods and different  
439 seasonal datasets are also need to be studied in the future.

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442

443 **Date availability.** Raw data used in this study can be obtained from the following  
444 open link: <https://doi.org/10.5281/zenodo.10791126> (Yuan et al., 2024). It is also  
445 available on request by contacting the corresponding author.

446

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448

449 **Author contributions.** RJH designed the study. Data analysis was done by WY, CL,  
450 LY, HY and RJH. WY, CL, LY, HY and RJH interpreted data, prepared the display  
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453

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455

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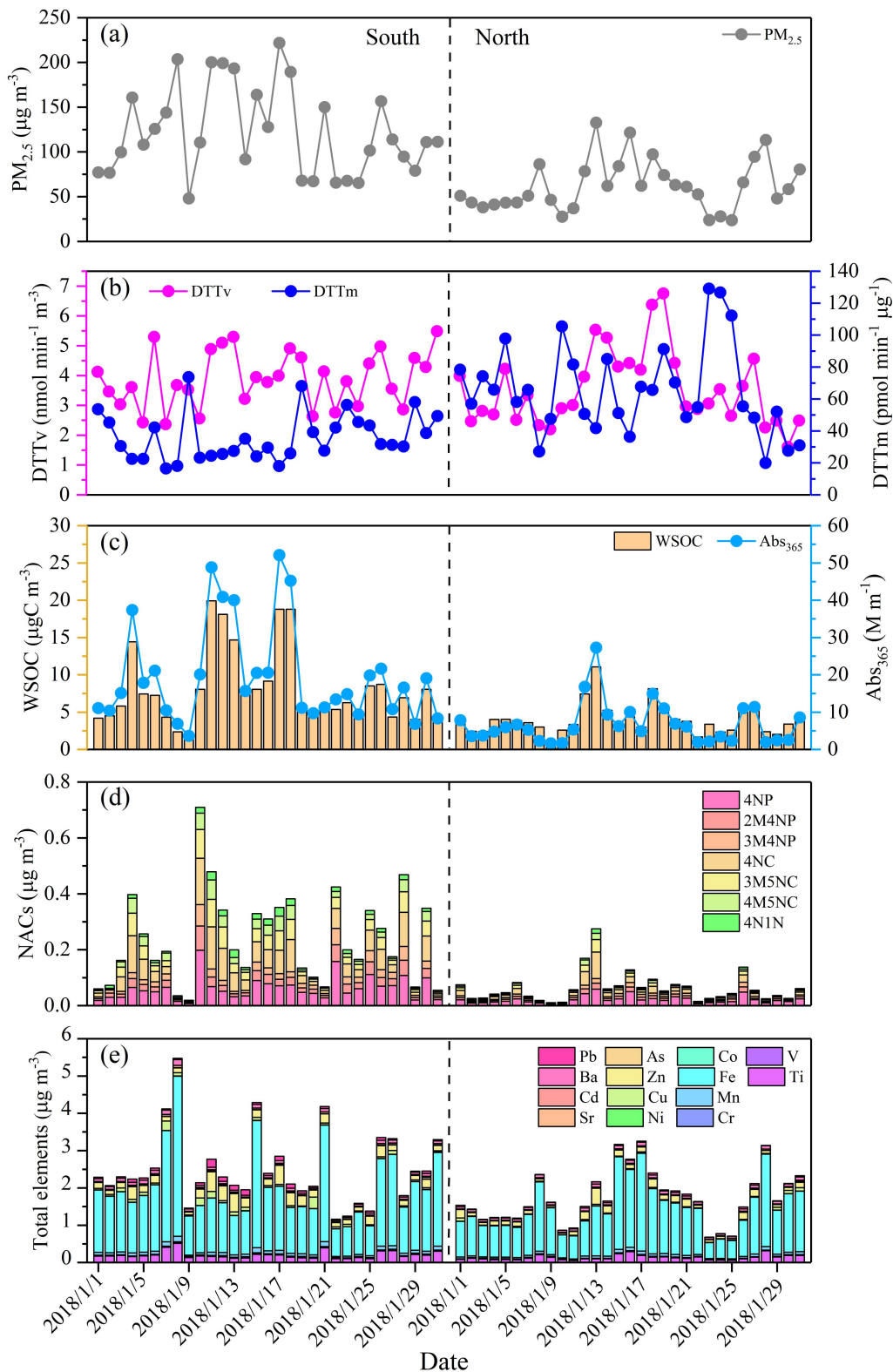
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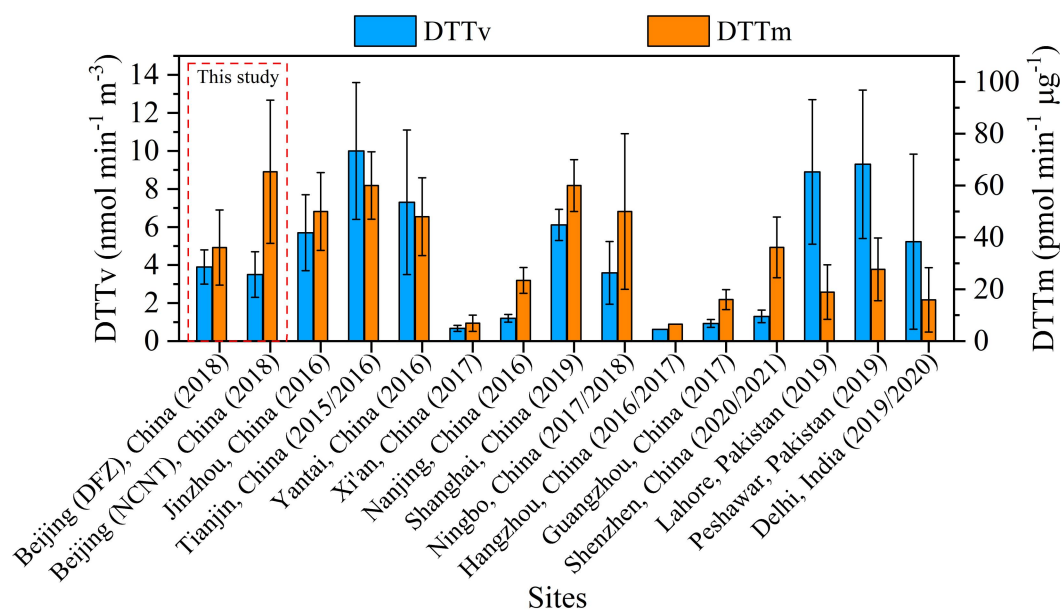
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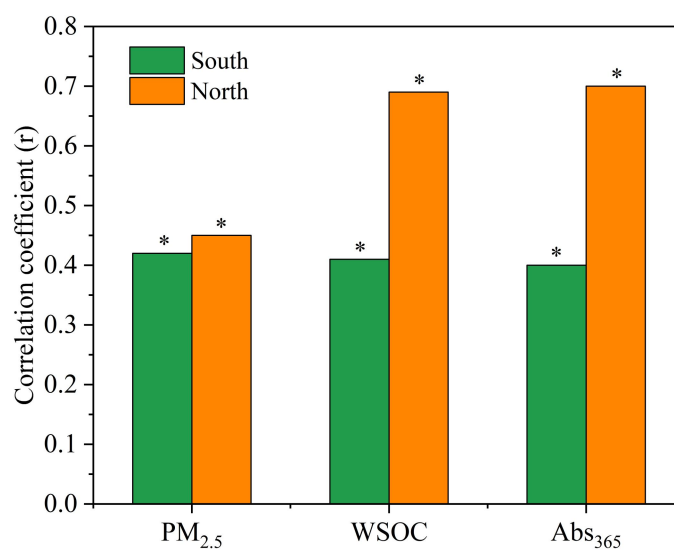
805 **Figure 1.** Time series of (a)  $PM_{2.5}$  concentration, (b)  $DTT_v$  and  $DTT_m$ , (c)  
 806 concentration and light absorption at wavelength 365 nm ( $Abs_{365}$ ) of WSOC,  
 807 concentrations of (d) NACs and (e) total elements.



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809 **Figure 2.** Comparison of DTT<sub>v</sub> and DTT<sub>m</sub> values of water-soluble PM<sub>2.5</sub> measured in  
 810 this study with those measured in other areas of Asia during similar period.

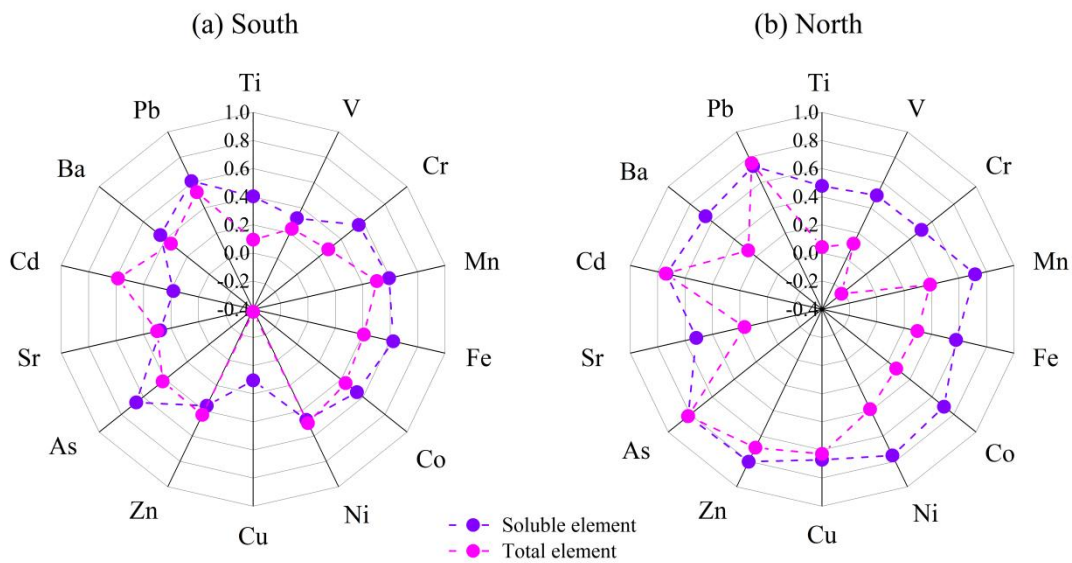
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813 **Figure 3.** Correlation coefficients between DTT<sub>v</sub> and PM<sub>2.5</sub>, WSOC, and Abs<sub>365</sub> in the  
 814 south and north of Beijing (\* indicates correlation is significant at the 0.05 level).

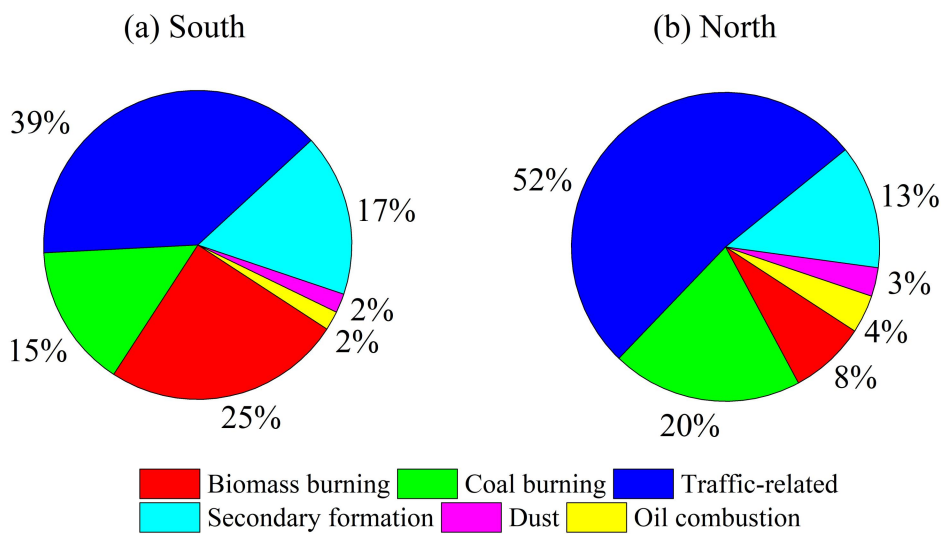
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817 **Figure 4.** Correlation coefficients between  $DTT_v$  and elements in the (a) south and (b)  
 818 north of Beijing.

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821 **Figure 5.** Contributions of resolved sources to  $DTT_v$  in the (a) south and (b) north of  
 822 Beijing.