

1 **Variable contribution of wastewater treatment plant effluents to nitrous oxide**
2 **emissions**

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12 **Abstract**

13 Nitrous oxide (N₂O), a potent greenhouse gas and ozone-destroying agent, is produced during
14 nitrogen transformations in both natural and human-constructed environments. Wastewater
15 treatment plants (WWTPs) produce and emit N₂O into the atmosphere during the nitrogen removal
16 process. However, the impact of WWTPs on N₂O emissions in downstream aquatic systems
17 remains poorly constrained. By measuring N₂O concentrations at a monthly resolution over a year
18 in the Potomac River Estuary, a tributary of Chesapeake Bay in the eastern United States, we found
19 a strong seasonal variation in N₂O concentrations and fluxes: N₂O concentrations were larger in
20 fall and winter but the flux was larger in summer and fall. Observations at multiple stations across
21 the Potomac River Estuary revealed hotspots of N₂O emissions downstream of WWTPs. N₂O
22 concentrations were higher at stations downstream of WWTPs compared to other stations (median:
23 21.2 nM vs 16.2 nM) despite the similar concentration of dissolved inorganic nitrogen, suggesting
24 the direct discharge of N₂O from WWTPs into the aquatic system or a higher N₂O production yield
25 in waters influenced by WWTPs. Since wastewater production has increased substantially with the
26 growing population and is projected to continue to rise, accurately accounting for N₂O emissions
27 downstream of the WWTPs is important for constraining and predicting future global N₂O
28 emissions. Efficient N₂O removal, in addition to dissolved nitrogen removal, should be an essential
29 part of water quality control in WWTPs.

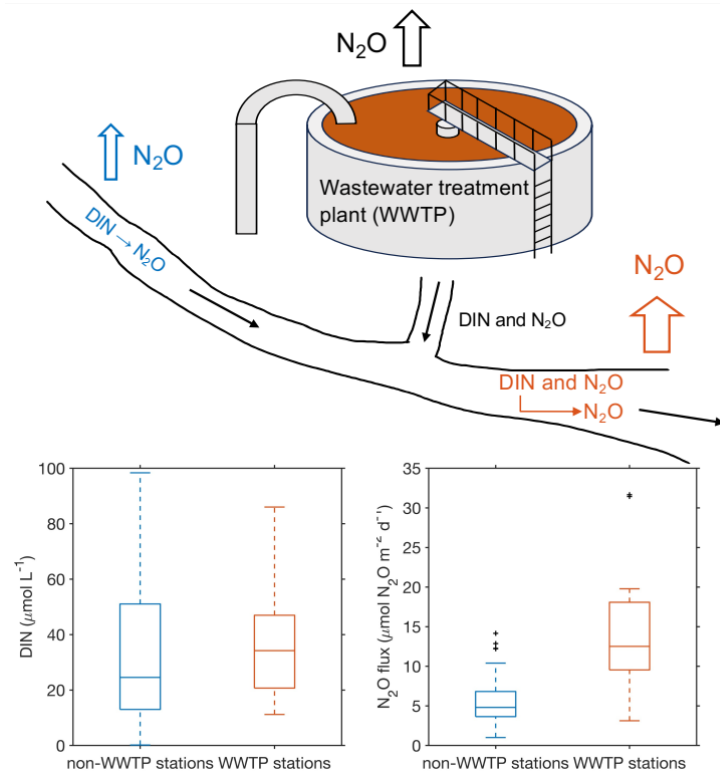
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31 Key words: nitrous oxide, greenhouse gas emission, nitrogen pollution, wastewater treatment
32 plants, spatial and seasonal variation

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34 Summary: Wastewater treatment plants (WWTPs) are known to be hotspots of greenhouse gas
35 emissions. However, the impact of WWTPs on the emission of the greenhouse gas N₂O in
36 downstream aquatic environments is less constrained. We found spatially and temporally variable
37 but overall higher N₂O concentrations and fluxes in waters downstream of WWTPs, pointing to
38 the need for efficient N₂O removal in addition to treating nitrogen in WWTPs.

39



42 **Introduction**

43 Nitrogen (N) enters the aquatic environment from agricultural and urban runoff, atmospheric
44 deposition, and wastewater treatment plants (WWTPs), potentially leading to eutrophication,
45 especially in densely populated regions (Galloway et al., 2008; Morée et al., 2013). During
46 microbial transformations of N in aquatic systems (e.g., nitrification and denitrification), a
47 powerful greenhouse gas and ozone depleting agent – N_2O – is produced (Quick et al., 2019).
48 Aquatic systems are large but highly variable sources of N_2O to the atmosphere (Wang et al.,
49 2023). For example, on a global basis, 0.04 - 0.291 Tg N yr^{-1} and 0.04 - 3.6 Tg N yr^{-1} of N_2O is
50 estimated to outgas from rivers and estuaries, respectively (Murray et al., 2015; Maavara et al.,
51 2019; Yao et al., 2019; Rosentreter et al., 2023). The high end of the estimates in these inland and
52 coastal waters approaches the scale of the global marine N_2O emissions (2.5 - 4.3 Tg N yr^{-1} in Tian
53 et al., 2020). The large uncertainty in the estimate of aquatic N_2O emission is partly due to high
54 spatial and temporal variabilities of N_2O flux within/across rivers and estuaries and the lack of
55 observations to capture such variability. Therefore, sampling and measurements of N_2O
56 concentration at high spatial and temporal resolutions would be desirable to constrain aquatic N_2O
57 emission.

58
59 The major factors that appear to correlate with N_2O concentration are dissolved inorganic nitrogen
60 (DIN) and oxygen status (Hu et al., 2016; Zheng et al., 2022). Waste and wastewater release large
61 amounts of DIN into the aquatic environment. In addition, waste and wastewater emit ~ 0.3 Tg N
62 yr^{-1} of N_2O (estimated from 2007-2016) into the atmosphere globally, an amount that is
63 continuously increasing at a rate of 0.04 ± 0.01 Tg N yr^{-1} per decade (Tian et al., 2020). N_2O
64 emission from WWTPs accounts for $\sim 5.2\%$ of total N_2O emission in 2021 in the United States
65 (EPA, 2023). N_2O emissions from different WWTPs are highly variable, and are normally
66 calculated as a function of DIN loading into the WWTPs, using an N_2O emission factor
67 (Kampschreur et al., 2009). N_2O emission factors range from 0.16% to 4.5% (N_2O emitted/DIN
68 loading) (Eggleston et al., 2006; De Haas and Andrews, 2022). In addition to direct emission from
69 the WWTPs, N_2O can be discharged via WWTP effluent and produced due to DIN release from
70 WWTP effluent into the creeks, rivers, and other downstream aquatic systems (McElroy et al.,
71 1978; Beaulieu et al., 2010; Masuda et al., 2018). However, the impact of WWTPs on downstream
72 N_2O concentration is less studied and the downstream N_2O emission remains poorly constrained.

73 Here we specifically compared the N₂O concentration upstream and downstream of the WWTPs
74 in order to assess the impact of WWTPs on N₂O emission, which could help to constrain the
75 emission factor associated with the WWTPs effluents.

76
77 The Potomac River is a major tributary of the Chesapeake Bay – the largest estuary in the United
78 States. The Potomac River Estuary is located in a highly populated area, mainly surrounded by
79 Washington, D.C., and the states of Virginia and Maryland in the eastern United States. The annual
80 mean discharge of Potomac River from 1895 to 2002 measured at Chain Bridge near Washington,
81 DC was 321 m³ s⁻¹ with a large interannual variability (Jaworski et al., 2007). The annual total
82 nitrogen loading was estimated to be around 27.7 ×10⁶ kg N year⁻¹ in 2008-2009 (Bricker et al.,
83 2014). The Potomac River Estuary has experienced ecological degradation for decades partly due
84 to excess nutrient inputs including from the effluents of WWTPs (Bricker et al., 2014; Jaworski et
85 al., 2007). For example, the Blue Plains Advanced WWTP in Washington, D.C. is one of largest
86 WWTPs in the world, treating an average of ~1454 million liters of water per day. Pioneering
87 work in 1978 showed that Blue Plains WWTP was a large source of nitrogen to the Potomac River
88 Estuary, triggering high N₂O production and concentration downstream (McElroy et al., 1978).
89 Thanks to higher standards mandated by governmental agencies (nitrogen concentration in
90 effluents below 7.5 mg L⁻¹) starting in 1980s and the technical improvements in N removal from
91 the wastewater, the nitrogen concentration in effluents of WWTPs in the Potomac River has
92 decreased substantially (Pennino et al., 2016). However, the concurrent effect on N₂O
93 concentration is largely unknown. The Department of Environmental Quality (DEQ) of Virginia
94 maintains an approximately monthly routine monitoring program for water quality (e.g., nitrogen
95 concentration, phosphorus concentration, chlorophyll concentration) and physical properties (e.g.,
96 temperature, salinity, pH, and dissolved oxygen concentration) in the Potomac River Estuary but
97 not for N₂O. Therefore, we collaborated with DEQ of Virginia to measure the spatial and temporal
98 variation of N₂O concentrations in the Potomac River Estuary.

99

100 **Materials and Methods**

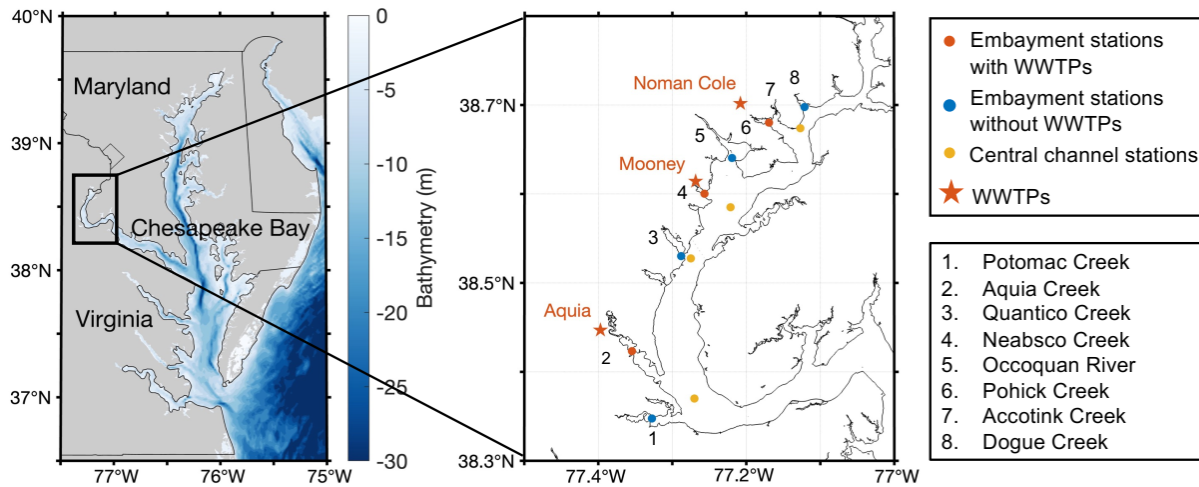
101 **Sample collection for N₂O and nutrients**

102 Surface waters at ~0.5 m depth at eleven stations in the tidal Potomac River Estuary were sampled
103 monthly or bimonthly (depending on the weather) on a vessel (Grady White 208) for the analysis

104 of DIN concentration, and both concentration and nitrogen isotopes of N₂O from April 2022 to
105 May 2023 (Figure 1). The eleven stations are characterized into 3 groups: embayment downstream
106 of WWTPs, embayment not associated with WWTPs, and the central channel of the Potomac
107 River. Three embayment stations downstream of WWTPs are associated with three different
108 WWTPs: Noman Cole, Mooney and Aquia, all of which implement tertiary treatment of the
109 wastewater. We obtained the volume discharge and total N in treated water of each WWTP from
110 Discharge Monitoring Reporting required by Virginia Pollutant Discharge Elimination System
111 permit. Noman Cole WWTP discharges ~140.8 million liters of water and 370 kg N per day into
112 Pohick Creek. Mooney WWTP discharges ~54.9 million liters of water and 147 kg N per day into
113 the Neabsco Creek. Aquia WWTP discharges much less water and N into the Aquia Creek (~21.2
114 million liters per day and 35 kg N per day). The distances from the embayment stations
115 downstream of WWTPs to Noman Cole, Mooney, Aquia WWTPs were approximately 4, 1.8 and
116 5.8 km, respectively.

117
118 The embayment stations were 2-3 meters deep while the average depth of central channel stations
119 was around 8 meters. The embayment stations have been routinely sampled for water quality
120 analyses by the DEQ of Virginia since the early 1970's. The central channel stations were added
121 for this study. The purposes of this sampling design are to evaluate the impact of WWTPs on
122 downstream distribution of DIN and N₂O, and to compare DIN and N₂O concentrations between
123 edge and central channel of the river. The central channel is likely affected both by the Potomac
124 mainstem flow and by the input from tributaries, while the embayment stations may be mainly
125 affected by water flow from tributaries but also influenced by the tidal cycle (see the salinity
126 change in Supplementary Figure 1b). While estuarine N₂O concentrations could be affected by
127 tides (Gonçalves et al., 2015), sampling was not always conducted at the same tidal state due to
128 logistic difficulties. Triplicate water samples for N₂O concentrations and isotopes were collected
129 via a submersible pump into 60 mL serum bottles after overflowing three times the bottle's volume.
130 After removing 3 mL water to create a 3 mL air headspace via a syringe, the serum bottles were
131 immediately sealed with butyl stoppers and aluminum crimps and preserved with 0.5 mL of 10 M
132 NaOH solution to stop biological activities. NaOH has been shown to be an effective and less
133 environmentally hazardous preservative for N₂O and nutrient analysis (Frame et al., 2016; Wong
134 et al., 2017).

135



136

137 Figure 1. Sampling stations in the Potomac River Estuary including embayment stations associated
138 with WWTPs (red circles) and without WWTPs (blue circles), and central channel stations (yellow
139 circles). Locations of WWTPs (Noman Cole, Mooney and Aquia) are shown in red stars.
140 Creeks/ivers with sampling stations are numbered in the map with names shown in the legend.
141 Stream sampling sites upstream and downstream of WWTPs in creeks 4 – 7 are shown in Figure
142 4 below.

143

144 In addition to the routine sampling in the Potomac River Estuary, we also sampled its tributaries,
145 some of which were associated with the WWTPs, on May 18, 2023 (Figure 1) to specifically
146 evaluate the impact of WWTPs on downstream N₂O concentrations. Four creeks/ivers were
147 sampled including Neabsco Creek (5 stations: 2 stations upstream and 3 stations downstream of
148 Mooney WWTP), Occoquan River (3 stations, no WWTP), Pohick Creek (4 stations: 2 stations
149 upstream and 2 stations downstream of Noman Cole WWTP), and Accotink Creek (2 stations, no
150 WWTP). Because Aquia WWTP discharges much less water and N into the Aquia Creek, its
151 impact was not specifically investigated. Since water depths of these creeks/ivers were shallow,
152 the water samples were collected by directly submerging 60 mL serum bottles into the surface
153 water (~0.1 m) and preserving them as described above.

154

155 Besides N₂O sampling, temperature, salinity, and dissolved O₂ concentrations were recorded via a
156 YSI EXO1 sonde. Chlorophyll-a samples (300 mL) were filtered onto GF/F filters and kept on ice
157 in a cooler. The filters were then kept frozen at -20°C in the lab until analysis within 3 months
158 (Arar and Collins, 1997). Samples of total nitrogen and phosphorus (both particulate and
159 dissolved) were collected into 250 mL HDPE bottles and kept in ice in a cooler until analysis
160 within 48 hours on land (Rice et al., 2012; EPA, 1983).

161

162 **Measurement of N₂O and nutrient concentrations**

163 N₂O in the serum bottles was stripped by helium carrier gas into a Delta V Plus mass spectrometer
164 (Thermo) for the analyses of N₂O concentration and isotope ratio (m/z = 44, 45, 46) (Tang et al.,
165 2022). The total amount of N₂O in the serum bottles was determined using a standard curve of
166 N₂O peak area with N₂O standards containing a known amount of N₂O reference gas (0, 0.207,
167 0.415, 0.623, 0.831, 1.247 nmol N₂O). The total amount of N₂O dissolved in the water was
168 calculated after subtracting the amount of N₂O in 3 mL air headspace. Specifically, the monthly
169 atmospheric N₂O concentrations were obtained from the nearby atmospheric station in Brentwood,
170 Maryland (<https://gml.noaa.gov/>) (Andrews et al., 2023). The amount of N₂O in 3 mL air
171 headspace was generally less than 4% of the amount of N₂O dissolved in the 57 mL water samples.
172 The N₂O concentration in samples was then calculated from the total amount of N₂O dissolved in
173 the water divided by the volume of water in the serum bottles. The detection limit and precision
174 of N₂O concentration measurement were 1.29 and 0.33 nM, respectively. We used N₂O produced
175 from nitrate isotope standards (USGS34 = -1.8‰ and IAEA = 4.7‰) to calibrate for δ¹⁵N of N₂O
176 samples.

177

178 After analyzing N₂O concentration, samples were neutralized to pH ~7 by adding 10%
179 hydrochloric acid. NO₂⁻ + NO₃⁻ (NO_x⁻) concentration in these samples was measured using the
180 vanadium (III) reduction method by converting NO_x⁻ to NO, which was then quantified by
181 chemiluminescence analyzer (Braman and Hendrix, 1989). The detection limit of NO_x⁻
182 concentration was 0.15 μM. NH₄⁺ and NO₂⁻ concentrations were measured at a few selected
183 stations using the fluorometric orthophthalaldehyde method (Holmes et al., 1999) and the
184 colorimetric method (Hansen and Koroleff, 1999), respectively. Their concentrations were much

185 smaller than NO_3^- alone, mostly accounting for less than 10% of the DIN concentration. Therefore,
186 we only present NO_x^- data in this study.

187

188 **N₂O flux calculation**

189 Surface N₂O flux was calculated using the following equation: $Flux = k \times (N_2O_{measured} -$
190 $N_2O_{equilibrium})$. The equilibrium N₂O concentration ($N_2O_{equilibrium}$) was calculated based on the
191 solubility of N₂O (Weiss and Price, 1980). The gas transfer velocity (k) was estimated based on
192 three different parameterizations: $k_{600} = 1.91 \times e^{0.35 \times U}$ (Raymond and Cole, 2001); $k_{600} =$
193 $0.314 \times U^2 - 0.436 \times U + 3.99$ (Jiang et al., 2008); $k = 0.251 \times U^2 \times (\frac{Sc}{660})^{-0.5}$ (Wanninkhof,
194 2014). U is the wind speed at the 10 m height obtained from the National Centers for
195 Environmental Prediction (NCEP) reanalysis (Kalnay et al., 1996;
196 <https://psl.noaa.gov/data/gridded/data.ncep.reanalysis.html>). Sc is the Schmidt number that could
197 be estimated as a function of temperature (Wanninkhof, 2014). Since our samples have salinity
198 close to 0, we used the parameterization of Sc for freshwater. Average values of the three N₂O flux
199 estimates are presented in the paper and N₂O fluxes estimated by different parameterizations are
200 provided in the associated dataset. We acknowledge large variations in estimating k values in the
201 riverine and estuarine systems by using different empirical models (Raymond and Cole, 2001;
202 Borges et al., 2004; Rosentreter et al., 2021). For instance, the effect of water velocity and water
203 depth on gas transfer velocity was not considered in the parameterizations above. Therefore, we
204 focus on evaluating the spatiotemporal variations in N₂O fluxes and their driving factors instead
205 of their absolute magnitude.

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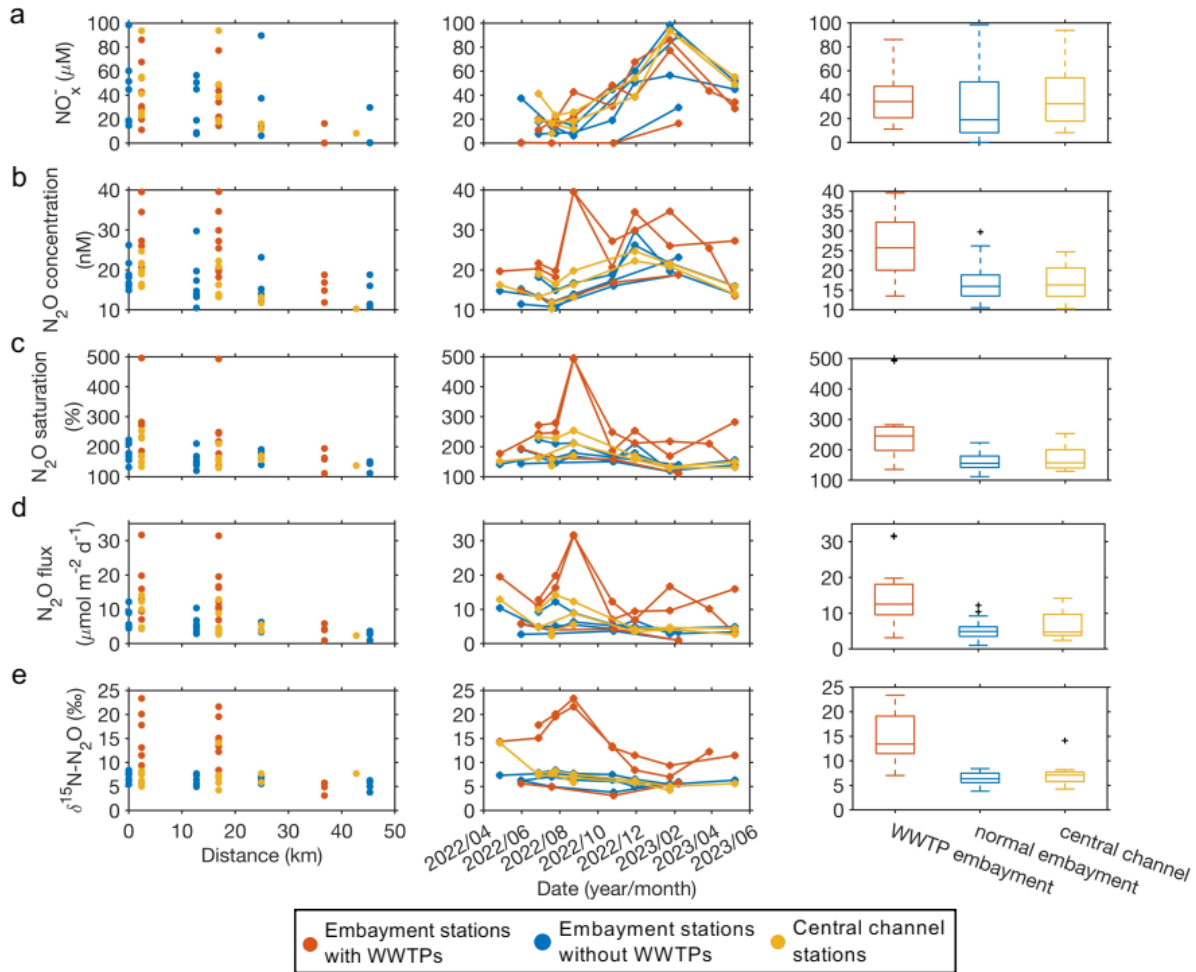
207 **Results and discussion**

208 **Spatial and temporal variations of N₂O concentrations in the Potomac River Estuary**

209 Along the roughly 50 km sampling transect in the Potomac River Estuary, NO_x^- concentration
210 decreased from 98 to $<1 \mu\text{M}$ from upstream to downstream (Figure 2a). NO_x^- concentration
211 showed a clear seasonal pattern: higher in winter and spring while lower in summer and fall. The
212 spatial and temporal patterns were likely attributable to the distribution of nutrient sources into the
213 Potomac River, DIN uptake and other removal processes along the river (Glibert et al., 1995;
214 Carstensen et al., 2015). For example, the maximum N loading into the Chesapeake Bay occurs in

215 winter and spring (Da et al., 2018). Meanwhile, N₂O concentration decreased from approximately
216 40 to 10 nM along the sampling transect and was higher in the fall and winter (Figure 2b). Since
217 temperature decreased from ~31°C in summer to 4°C in winter (Supplementary Figure 1a), the
218 increase in N₂O solubility in colder water during winter partly explained the seasonal change. In
219 contrast, N₂O saturation had higher values in summer and fall (Figure 2c), suggesting a higher
220 N₂O production in summer and fall. It is worth noting that N₂O saturation was above 100% at all
221 sampling stations with a maximum reaching 500%, indicating the Potomac River Estuary was a
222 consistent and strong source of N₂O to the atmosphere. N₂O flux ranged from 1 to 31.7 μmol N₂O
223 m⁻² d⁻¹ (Figure 2d). N₂O concentration (median: 18.2 nM) and flux (median: 5.6 μmol N₂O m⁻² d⁻¹)
224 in the Potomac River Estuary were substantially higher than in the mainstem of the Chesapeake
225 Bay (2.6 to 20.9 nM N₂O with a median value at 10.6 nM and -0.3 to 4.3 μmol N₂O m⁻² d⁻¹ with a
226 median at 0.5 μmol N₂O m⁻² d⁻¹ (Tang et al., 2022; Laperriere et al., 2019)). Therefore, the
227 tributaries of the Chesapeake Bay (i.e., Potomac River) are intense sources of N₂O to the
228 atmosphere.

229



230
 231 Figure 2. Spatial and temporal variations of NO_x^- concentration (a), N_2O
 232 saturation (c), N_2O flux (d) and $\delta^{15}\text{N}$ of N_2O (e). The distance shows from upstream to downstream
 233 stations in the Potomac River. Embayment stations associated with WWTPs (red circles and lines)
 234 and without WWTPs (blue circles and lines), and central channel stations (yellow circles and
 235 lines). For the boxplots, the red line in each box is the median. The bottom and top of each box are
 236 the 25th and 75th percentiles of the observations, respectively. The error bars represent 1.5 times
 237 the interquartile range away from the bottom or top of the box, with black + signs showing outliers
 238 beyond that range. Embayment stations associated with WWTPs had significantly higher N_2O
 239 concentration, N_2O saturation, N_2O flux and $\delta^{15}\text{N}$ values compared to other stations ($p < 0.01$, t -
 240 test) but not significantly different NO_x^- concentration.

241
 242 Stations close to each other had similar NO_x^- concentrations (e.g., upstream stations > downstream
 243 stations), regardless of station category (i.e., with WWTP, without WWTP, central channel of the

244 Potomac River). In contrast, N₂O concentrations varied within locations according to the station
245 category: N₂O concentrations were substantially higher at stations downstream of WWTPs
246 ($p < 0.01$, t -test). N₂O concentrations were similar between stations in embayments without
247 WWTPs and the central channel (Figure 2). This suggests these WWTPs are efficient in removing
248 DIN from sewage and other sources but WWTPs may discharge N₂O directly into the effluent or
249 enhance downstream N₂O production (e.g., higher N₂O production yield from the same amount of
250 DIN). This effect extended to our sampling stations ~1.8-4 km downstream of the WWTPs.
251 However, the effect of WWTPs on downstream N₂O varied among stations. For example, elevated
252 N₂O concentrations were observed downstream from Noman Cole and Mooney WWTPs but not
253 downstream from Aquia WWTP. This difference may be related to the different N removal
254 processes of WWTPs that produce N₂O at different yields (de Haas and Andrews. 2022; Zhao et
255 al., 2024). However, we don't have detailed information about the three WWTPs other than that
256 they all implement tertiary treatment. In addition, the different dilution factors by riverine
257 discharges also matter. For example, the volume of effluent from Mooney WWTP was higher than
258 the discharge of Neabsco Creek while the volume of effluent from Aquia WWTP were generally
259 lower than the discharge of Aquia Creek (Supplementary Figure 2a-b). Particularly, the highest
260 N₂O concentration of up to 40 nM was found at two stations downstream of the Noman Cole and
261 Mooney WWTPs on August 23, 2022 when the river discharge was low (Supplementary Figure
262 2). Thus, the effect of WWTPs on downstream N₂O concentrations also varies seasonally (Schult
263 et al., 2023; Murray et al., 2020), with a relatively more important role in the dry season. Repeated
264 spatial and temporal sampling allowed us to capture these N₂O hotspots. Previous studies have
265 shown the impact of WWTPs on downstream N₂O concentration in aquatic environments. For
266 example, the highest N₂O concentration ~675 nM in the Potomac River was measured near the
267 discharge of the Blue Plains WWTP in 1977 (McElroy et al., 1978). Highest N₂O emissions in the
268 Ohio River near Cincinnati were attributed to direct input of N₂O from WWTPs' effluents
269 (Beaulieu et al., 2010).

270

271 In addition, a higher nitrogen isotopic signature ($\delta^{15}\text{N}$) of N₂O associated with WWTPs (median
272 $\delta^{15}\text{N}$ at 13‰) also suggests the distinct sources or cycling processes of N₂O compared to stations
273 of the central channel and without the influence of WWTPs (median $\delta^{15}\text{N}$ of N₂O at 6‰, Figure
274 2e) in the Potomac River Estuary. In comparison, the average $\delta^{15}\text{N}$ of N₂O in the tropospheric air

275 is around 6.55‰ (Snider et al., 2015). $\delta^{15}\text{N}$ of N_2O for stations with the influence of WWTPs
276 showed a clear seasonal variation: higher in summer than winter (Figure 2e). This seasonal
277 difference may be related to the seasonal change in the relative importance of WWTPs' effluents
278 versus riverine discharge (Supplementary Figure 2c). For example, a relatively larger WWTPs'
279 effluent volume compared to the riverine discharge led to a larger $\delta^{15}\text{N}$ of N_2O in summer.
280 However, no clear seasonal pattern of $\delta^{15}\text{N}$ of N_2O was seen for stations without the influence of
281 WWTPs. $\delta^{15}\text{N}$ of N_2O produced in WWTPs depends on the treatment stages and aeration
282 conditions (Toyoda et al., 2011; Tumendelger et al., 2014). For example, the average $\delta^{15}\text{N}$ values
283 were reported to be -24.5‰ and 0‰ respectively for N_2O produced from nitrification during oxic
284 treatment versus N_2O produced from anaerobic denitrification in a California WWTP (Townsend-
285 Small et al., 2011). Our observed $\delta^{15}\text{N}$ of N_2O downstream of WWTPs was higher than the values
286 found in these urban WWTPs. One of the reasons for the increased $\delta^{15}\text{N}$ of N_2O may be partial
287 N_2O reduction via denitrification in the WWTPs, in downstream creeks, or in sediments; this
288 denitrification effect has been seen in the marine oxygen minimum zones (Kelly et al., 2021).
289 Denitrification as the cause of the elevated $\delta^{15}\text{N}$ is partly supported by the higher $\delta^{15}\text{N}$ of N_2O
290 when NO_x^- was reduced to less than 40 μM , suggesting the occurrence of N_2O reduction when the
291 concentration of other denitrification substrates became low (Supplementary Figure 3). However,
292 we do not know the exact locations where denitrification occurred (e.g., WWTPs, anoxic niches
293 in suspended particles, sediments), which deserves further investigations. The influence of
294 denitrification on unique isotopic signatures of N_2O produced from WWTPs has also been
295 observed in Tama River in Japan (Toyoda et al., 2009).

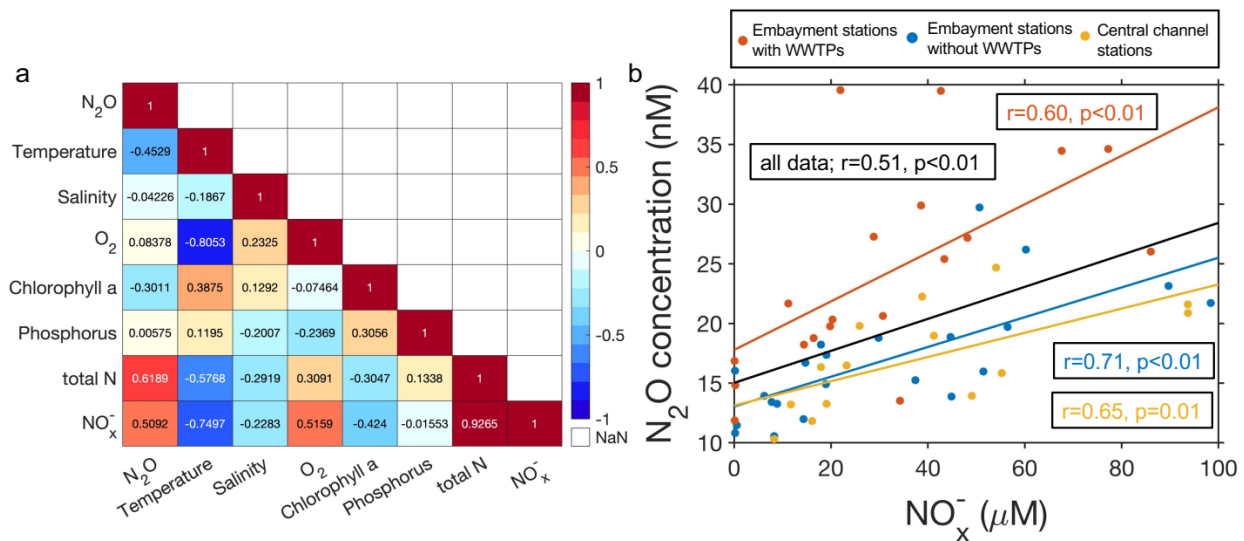
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297 **Environmental controls on N_2O concentrations**

298 N_2O concentrations showed positive correlations with total N ($r=0.62$, $p<0.01$) and NO_x^-
299 concentrations ($r=0.51$, $p<0.01$) (Figure 3a). Correlation analyses done separately for stations with
300 or without WWTPs had similar patterns (Supplementary Figure 4). A better correlation between
301 the N_2O concentration and total N may indicate the contribution of other N sources besides NO_x^-
302 to N_2O production. N_2O could be produced from nitrification in the process of oxidizing NH_4^+ to
303 NO_x^- in the oxic environment as previously shown in the oxygenated mainstem of the Chesapeake
304 Bay (Tang et al., 2022). However, we can't exclude the possibility of N_2O production from
305 denitrification associated with anaerobic microsites in particles or in sediment (Beaulieu et al.,

306 2011; Wan et al., 2023). Future investigations with ^{15}N tracers should be conducted to differentiate
 307 N_2O production pathways around the WWTPs. Furthermore, N_2O concentration was negatively
 308 correlated with temperature since higher temperature reduced the N_2O solubility. Although
 309 previous studies have showed dissolved oxygen to be an important driver of N_2O concentrations
 310 or fluxes in rivers and estuaries (Rosamond et al., 2012; Wang et al., 2015; Zheng et al., 2022), we
 311 did not find a strong dependence of N_2O on oxygen concentrations in the Potomac River Estuary
 312 (Figure 3a). This lack of strong dependence is probably because of the overall oxygenated
 313 conditions (Supplementary Figure 1c), and opposite correlations found in stations without WWTPs
 314 (positive) versus in stations with WWTPs (negative) (Supplementary Figure 4), which may be
 315 influenced by the different N_2O production pathways.

316



317

318 Figure 3. (a) Correlation coefficients among different environmental factors and N_2O
 319 concentrations. (b) Relationship between N_2O and NO_x^- concentrations at different categories of
 320 sampling stations.

321

322 The significant positive relationship between N_2O and NO_x^- concentration existed for samples
 323 collected at stations from all three different categories (Figure 3b). N_2O concentrations at stations
 324 downstream of WWTPs were notably higher than at other stations not associated with WWTPs
 325 even under the similar range of NO_x^- concentration. The larger slope of N_2O concentration versus
 326 NO_x^- concentration at stations downstream of WWTPs may be related to the direct input of N_2O
 327 from WWTPs into the downstream waters or different N_2O production pathways and production

328 yields that deserve further investigations. The DIN concentration has been found to be a good
329 predictor of N₂O concentration and emission in many other rivers and estuaries (Murray et al.,
330 2015; Reading et al., 2020; Zheng et al., 2022;). However, the correlation varied spatially, which
331 may be affected by the variable N₂O emission factors from DIN cycling. The emission factors are
332 affected by temperature, concentration and forms of N, oxygen, organic carbon concentration and
333 many other factors (Hu et al., 2016). The external N₂O input (e.g., input from WWTPs) could also
334 affect the relationship between N₂O and DIN concentrations (Dong et al., 2023). Compared to DIN
335 (~28 to 71 μM) and N₂O concentrations (~16 to 61 nM) measured approximately 45 years ago in
336 the same section of the Potomac River (McElroy et al., 1978), current DIN and N₂O concentrations
337 have slightly decreased. Thus, an additional benefit of nutrient regulation is the reduction of
338 greenhouse gas - N₂O - emissions, beyond improving water quality.

339

340 Since N₂O concentrations had the strongest correlation with total N concentrations (reflecting the
341 N₂O production potential) and temperature (affecting N₂O solubility), we developed a predictive
342 model of N₂O concentration based on total N and temperature. Predictions were performed
343 separately for stations with WWTPs ($N_2O \text{ concentration} = 0.115 \times \text{total N} - 0.241 \times$
344 $\text{temperature} + 17.185$, n=18, r=0.78; p<0.01) and without WWTPs including central channel
345 stations ($N_2O \text{ concentration} = 0.049 \times \text{total N} - 0.298 \times \text{temperature} + 18.888$, n=23,
346 r=0.81, p<0.01). The observed N₂O variability was generally captured by these simple linear
347 models (Supplementary Figure 5) but there were variabilities in the observations remaining to be
348 explained. Addition of other predictors did not significantly improve the model performance, so
349 we chose the simple predictive model that is mechanistically understandable. We then applied the
350 two predictive models separately to estimate N₂O concentrations at the embayment station in the
351 Pohick Bay (with WWTP) and the embayment station in the Occoquan Bay (without WWTP)
352 using total N concentration and temperature that were measured since 2008 by the DEQ of Virginia
353 monitoring program (Supplementary Figures 6 and 7). Predicted N₂O concentrations showed a
354 clear seasonality: higher in winter and lower in summer. N₂O concentrations in the Pohick Bay
355 decreased substantially (-0.9 nM/year) possibly due to the nutrient reduction (total N concentration
356 decreasing at 8.8 μM/year) over the last 14 years (Supplementary Figure 6). However, N₂O
357 concentrations in the Occoquan Bay only decreased slightly (-0.1 nM /year, not statistically
358 significant) along with the minor nutrient reduction (total N concentration decreasing at non-

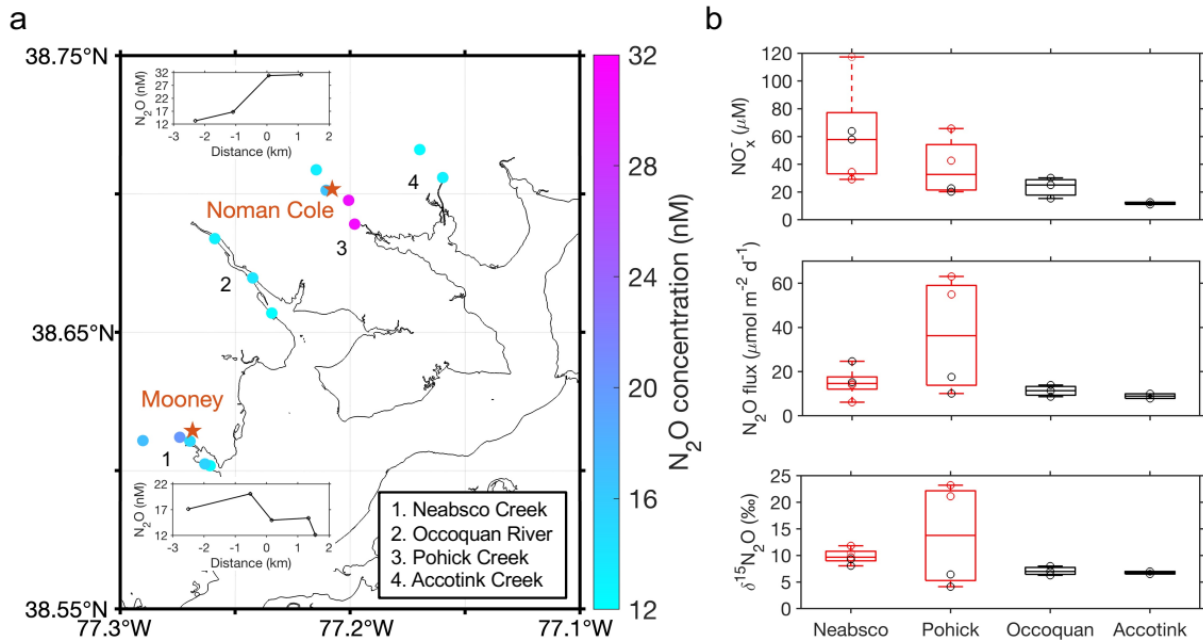
359 statistically significant rate of 0.5 $\mu\text{M}/\text{year}$) (Supplementary Figure 7). Continuation of
360 environmental monitoring in the Potomac River (e.g., N nutrients and temperature), which is much
361 easier than sampling and measuring N_2O gas, could be used to indirectly estimate the changes in
362 N_2O concentrations in the future. These predictors are likely to be important in other estuaries, but
363 the weighting would vary among locations.

364

365 **Impact of wastewater treatment plants on N_2O concentrations and emissions**

366 To further evaluate how WWTPs affect the N_2O distribution in the Potomac River, we measured
367 N_2O concentrations upstream and downstream of the two WWTP effluents (Mooney and Noman
368 Cole in Neabsco Creek and Pohick Creek, respectively) and compared them to N_2O concentrations
369 measured in two creeks that do not have WWTPs. Interestingly, the N_2O concentration at the
370 station upstream of Mooney WWTP in Neabsco Creek was higher than the N_2O concentration at
371 the station downstream of Mooney WWTP (20.1 vs 15.0 nM) (Figure 4a). The exact mechanisms
372 were not clear but one of the potential reasons could be the influence by tidal cycles: high tide
373 during the sampling time (salinity was 0.17 instead of 0) may have reversed the water flow and
374 diluted the WWTP effluent with low N_2O concentration Potomac water (12.1 nM at the outflow
375 of Neabsco Creek into the Potomac River Estuary). In contrast, we found substantially higher N_2O
376 concentration downstream of the Noman Cole WWTP (30.8 nM downstream vs 16.7 nM
377 upstream) in the Pohick Creek, which is less affected by the tidal cycle due to its semi-closed
378 geography (salinity was 0.12). The high downstream N_2O concentration may suggest the direct
379 addition of N_2O from WWTP effluent to the downstream environment. Furthermore, $\delta^{15}\text{N}$ of N_2O
380 in stations downstream of WWTPs were generally higher than the other two creeks that do not
381 have WWTPs (Figure 4b), confirming the distinct source of N_2O production by WWTPs found in
382 the Potomac River Estuary. Overall, the influence of WWTP effluents on downstream distribution
383 of N_2O is variable, and could be affected by the physical movement of water.

384



385
 386 Figure 4. (a) Color-coded N_2O concentration at creek sampling stations on May 18, 2023. WWTPs
 387 (Mooney and Noman Cole) are shown in red stars. The insert figures show the change in N_2O
 388 concentrations as a function of distance up or down stream from the WWTPs. Creeks/river with
 389 sampling stations are numbered in the map with names shown in the legend. (b) Box plots of NO_3^-
 390 , N_2O flux and $\delta^{15}N$ of N_2O comparing four creeks. Neabsco and Pohick Creeks with WWTPs are
 391 displayed with red color boxes. Red and black circles in the boxplots show the data points of
 392 stations downstream and upstream/or without WWTPs, respectively. NO_3^- , N_2O flux and $\delta^{15}N$ of
 393 N_2O were clearly higher at stations downstream from the WWTP in Pohick Creek.

394
 395 Dong et al. (2023) evaluated the potential impact of wastewater nitrogen discharge on estuarine
 396 N_2O emissions globally. Here we compiled data from previous studies with direct N_2O
 397 measurements in aquatic systems associated with WWTPs (not included in Dong et al., 2023) to
 398 assess the global impact of WWTPs on aquatic N_2O concentrations or emissions (McElroy et al.,
 399 1978; Hemond and Duran, 1989; Toyoda et al., 2009; Beaulieu et al., 2010; Rosamond et al., 2012;
 400 Chun et al., 2020; Masuda et al., 2021; Masuda et al., 2018; Dylla, 2019). WWTP effluents or
 401 water downstream of the WWTPs contain some of the highest N_2O concentrations and fluxes
 402 observed in the aquatic system (Table 1 and Supplementary Figure 8). For example, up to
 403 12,411.4% saturation of N_2O was measured in the effluent of WWTPs in the Tama River in Japan

404 (Toyoda et al., 2009). In addition, N₂O flux up to 40,800 μmol N₂O-N m⁻² d⁻¹ was found
 405 downstream of the Regina WWTP in the Wascana Creek in Canada (Dylla, 2019). The
 406 downstream N₂O flux was >300 times higher than the N₂O flux upstream of the Regina WWTP.
 407 In comparison, the maximum N₂O saturation and flux previously reported in a global riverine N₂O
 408 dataset were around 2,500% and 12,754 N₂O-N m⁻² d⁻¹ (Hu et al., 2016). Across the sites listed in
 409 Table 1, N₂O concentration/saturation/flux downstream of the WWTPs was 1.45 to 374-fold of
 410 the upstream waters. The only exception was our observed decrease in N₂O concentrations
 411 downstream of Mooney WWTP on May 18, 2023, which was likely influenced by the tidal cycle.
 412 The wide range of apparent WWTP effect is related to many factors including the variable N₂O
 413 emission factors in the WWTPs, the ratio of WWTP effluent volume to riverine discharge, the
 414 distance from the WWTPs where measurements were conducted, and the direction of water flow
 415 (e.g., tidal cycle). In addition, the estuarine type, mixing regime, and stratification are also
 416 important factors controlling N₂O emissions (Brown et al., 2022). Overall, failing to account for
 417 N₂O emissions downstream of the WWTPs and their variability would substantially bias estimates
 418 of aquatic N₂O emissions. This uncertainty is increased by the fact that only a few observations
 419 are available (all in the northern hemisphere) (Supplementary Figure 8) compared to >58 000
 420 WWTPs present globally (Ehalt Macedo et al., 2022). It is also important to restrict the N₂O
 421 emission via efficient N₂O reduction in the WWTPs considering the projected increase in future
 422 wastewater production (Qadir et al., 2020).

423
 424 Table 1. Global N₂O observations in aquatic systems associated with wastewater treatment plants.
 425 N₂O data are presented in concentration (nM), saturation (%) or flux (μmol N₂O-N m⁻² d⁻¹)
 426 according to how they are reported in different studies.

River/location	WWTP	N ₂ O upstream or in tributaries without WWTP	N ₂ O in WWTP effluent	N ₂ O downstream or in tributaries with WWTP	Average fold change (downstream vs upstream)	Reference
Potomac River/ Washington, D.C., USA	Blue Plains WWTP	11-34 nM		147-318 nM	9.3	McElroy et al., 1978
Assabet River/ Massachusetts, USA	Westborough WWTP	~10 nM	1045 nM	163 nM	16.3	Hemond and Duran. 1989
Tama River/ Tokyo, Japan	Plant 1 Plant 2	350.7% saturation 219.3%	12411.4% saturation 3326.2%	3454.8% saturation 1029.6%	9.8 4.7	Toyoda et al., 2009

Ohio River/ Cincinnati, USA		27.9 $\mu\text{mol N}_2\text{O-N m}^{-2} \text{d}^{-1}$		1068 $\mu\text{mol N}_2\text{O-N m}^{-2} \text{d}^{-1}$	38.2	Beaulieu et al., 2010
Grand River/ Ontario, Canada	e.g., Kitchener WWTP	4-12 $\mu\text{mol N}_2\text{O-N m}^{-2} \text{d}^{-1}$		9-113 $\mu\text{mol N}_2\text{O-N m}^{-2} \text{d}^{-1}$	9.4	Rosamond et al., 2012
Wascana Creek/ Saskatchewan, Canada	Regina WWTP	-32.5 to 109 $\mu\text{mol N}_2\text{O-N m}^{-2} \text{d}^{-1}$	227 to 72800 $\mu\text{mol N}_2\text{O-N m}^{-2} \text{d}^{-1}$	398 to 40800 $\mu\text{mol N}_2\text{O-N m}^{-2} \text{d}^{-1}$	374	Dylla. 2019
Han River/ Seoul, Korea	JNW	39.7 nM	602.1 nM	441.6 nM	11.1	Chun et al., 2020
A-river	A-WWPT	61 nM	493 nM	180 nM	3	Masuda et al., 2021
B-river	B-WWTP	95	246	286	3	Masuda et al., 2018
C-river/Miyagi, Japan	C-WWTP	100	319	145	1.45	
Potomac River Estuary /Virginia, USA	Noman Cole Mooney Aquia	10.8-29.7 nM		11.87-39.5 nM	1.6	This study
Neabsco Creek/ Virginia, USA	Mooney	20.1 nM		15.0 nM	0.75	This study
Pohick Creek/ Virginia, USA	Noman Cole	16.7 nM		30.8 nM	1.84	This study

427

428 **Conclusion**

429 Taking advantage of the routine water monitoring program by the DEQ of Virginia, we detected
430 strong spatial and temporal variabilities of N_2O concentrations and emissions in the Potomac River
431 Estuary, a major tributary of Chesapeake Bay. Observations across the Potomac River Estuary also
432 allowed us to identify hotspots of N_2O emissions associated with WWTPs effluents. Higher N_2O
433 concentrations downstream of WWTPs compared to regions with similar nitrogen nutrient
434 concentrations suggested the direct discharge of dissolved N_2O from WWTPs and/or intense N_2O
435 production. A survey of globally available data shows N_2O concentrations or emissions are
436 consistently elevated in waters downstream from WWTPs. Future ^{15}N tracer incubations would
437 help to explain the high N_2O concentration downstream of WWTPs by disentangling the N_2O
438 production pathways. In addition, concurrent measurements of the N flux and N_2O concentration
439 downstream of WWTPs will help to constrain overall N_2O emission factors associated with
440 WWTPs. Our work could encourage potential collaborations between scientific community and
441 governmental agencies/the public to better observe the environmental pollution or quality, e.g.,
442 increasing the frequency and resolution of observations for N_2O and other greenhouse gases along
443 with many regularly monitored environmental factors like temperature and nutrients. Such efforts

444 may identify previously overlooked sources of N₂O emission and help to better estimate N₂O
445 emissions from aquatic systems.

446

447 **Data availability**

448 Data presented in this study has been deposited in Zenodo repository:
449 <https://doi.org/10.5281/zenodo.10775250>.

450

451 **Author contribution**

452 W.T. conceived the study. J.T., T.J., and W.T. collected N₂O samples from the Potomac River
453 Estuary. W.T. analyzed samples and interpreted data with other coauthors. W.T. wrote the first
454 draft of the manuscript with input from B.B.W. All coauthors contributed to the result discussion
455 and manuscript writing.

456

457 **Competing interests**

458 The authors declare that they have no conflict of interest.

459

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467

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