

1 **Variable contribution of wastewater treatment plant effluents to nitrous oxide**
2 **emissions**

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11
12 **Abstract**

13 Nitrous oxide (N₂O), a potent greenhouse gas and ozone-destroying agent, is produced during
14 nitrogen transformations in both natural and human-constructed environments. Wastewater
15 treatment plants (WWTPs) produce and emit N₂O into the atmosphere during the nitrogen removal
16 process. However, the impact of WWTPs on N₂O emissions in downstream aquatic systems
17 remains poorly constrained. By measuring N₂O concentrations at a monthly resolution over a year
18 in the Potomac River Estuary, a tributary of Chesapeake Bay in the eastern United States, we found
19 a strong seasonal variation in N₂O concentrations and fluxes: N₂O concentrations were larger in
20 fall and winter but the flux was larger in summer and fall. Observations at multiple stations across
21 the Potomac River Estuary revealed hotspots of N₂O emissions downstream of WWTPs. N₂O
22 concentrations were higher at stations downstream of WWTPs compared to other stations (median:
23 21.2 nM vs 16.2 nM) despite the similar concentration of dissolved inorganic nitrogen, suggesting
24 the direct discharge of N₂O from WWTPs into the aquatic system or a higher N₂O production yield
25 in waters influenced by WWTPs. Since wastewater production has increased substantially with the
26 growing population and is projected to continue to rise, accurately accounting for N₂O emissions
27 downstream of the WWTPs ~~is important for constraining and predicting future~~, global N₂O
28 emissions. Efficient N₂O removal, in addition to dissolved nitrogen removal, should be an essential
29 part of water quality control in WWTPs.

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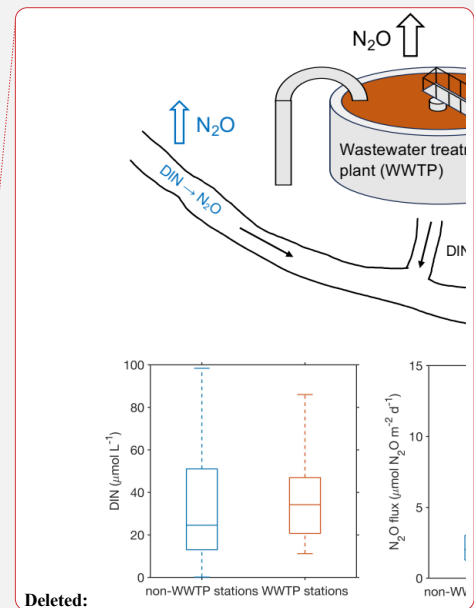
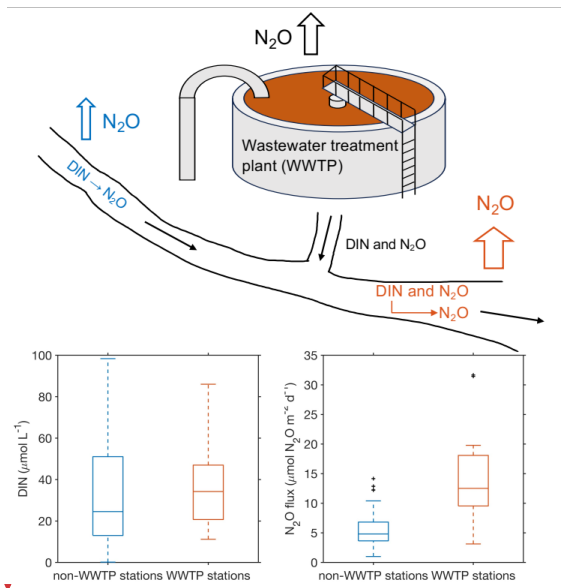
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33 Key words: nitrous oxide, greenhouse gas emission, nitrogen pollution, wastewater treatment
34 plants, spatial and seasonal variation

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36 Summary: Wastewater treatment plants (WWTPs) are known to be hotspots of greenhouse gas
37 emissions. However, the impact of WWTPs on the emission of the greenhouse gas N_2O in
38 downstream aquatic environments is less constrained. We found spatially and temporally variable
39 but overall higher N_2O concentrations and fluxes in waters downstream of WWTPs, pointing to
40 the need for efficient N_2O removal in addition to treating nitrogen in WWTPs.

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45 **Introduction**

46 Nitrogen (N) enters the aquatic environment from agricultural and urban runoff, atmospheric
47 deposition, and wastewater treatment plants (WWTPs), potentially leading to eutrophication,
48 especially in densely populated regions (Galloway et al., 2008; Morée et al., 2013). During
49 microbial transformations of N in aquatic systems (e.g., nitrification and denitrification), a
50 powerful greenhouse gas and ozone depleting agent – N₂O – is produced (Quick et al., 2019).
51 Aquatic systems are large but highly variable sources of N₂O to the atmosphere (Wang et al.,
52 2023). For example, on a global basis, 0.04 - 0.291 Tg N yr⁻¹ and ~~0.04 - 3.6~~ Tg N yr⁻¹ of N₂O is
53 estimated to outgas from rivers and estuaries, respectively (Murray et al., 2015; Maavara et al.,
54 2019; Yao et al., 2019; [Rosentreter et al., 2023](#)). The high end of the estimates in these inland and
55 coastal waters approaches the scale of the global marine N₂O emissions (2.5 - 4.3 Tg N yr⁻¹ in Tian
56 et al., 2020). The large uncertainty in the estimate of aquatic N₂O emission is partly due to high
57 spatial and temporal variabilities of N₂O flux within/across rivers and estuaries and the lack of
58 observations to capture such variability. Therefore, sampling and measurements of N₂O
59 concentration at high spatial and temporal resolutions would be desirable to constrain aquatic N₂O
60 emission.

61
62 The major factors that appear to correlate with N₂O concentration are dissolved inorganic nitrogen
63 (DIN) and oxygen status (Hu et al., 2016; Zheng et al., 2022). Waste and wastewater release large
64 amounts of DIN into the aquatic environment. ~~In addition~~, waste and wastewater emit ~0.3 Tg N
65 yr⁻¹ of N₂O (estimated from 2007-2016) into the atmosphere globally, an amount that is
66 continuously increasing at a rate of 0.04±0.01 Tg N yr⁻¹ per decade (Tian et al., 2020). N₂O
67 emission from WWTPs accounts for ~5.2% of total N₂O emission in [the United States](#)
68 (EPA, 2023). N₂O emissions from different WWTPs ~~are highly variable~~, and ~~are normally~~
69 calculated as a function of DIN loading into the WWTPs, using an N₂O emission factor
70 (Kampschreur et al., 2009). N₂O emission factors range from 0.16% to 4.5% (N₂O emitted/DIN
71 loading) ([Eggleston et al., 2006](#); De Haas and Andrews, 2022). In addition to direct emission from
72 the WWTPs, N₂O can be discharged via WWTP effluent and produced due to DIN release from
73 WWTP effluent into the creeks, rivers, and other downstream aquatic systems (McElroy et al.,
74 1978; Beaulieu et al., 2010; Masuda et al., 2018). However, the impact of WWTPs on downstream
75 N₂O concentration is less studied and the downstream N₂O emission remains poorly constrained.

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84 Here we specifically compared the N₂O concentration upstream and downstream of the WWTPs
85 in order to assess the impact of WWTPs on N₂O emission, which could help to constrain the
86 emission factor associated with the WWTPs effluents.

87
88 The Potomac River is a major tributary of the Chesapeake Bay – the largest estuary in the United
89 States. The Potomac River Estuary is located in a highly populated area, mainly surrounded by
90 Washington, D.C., and the states of Virginia and Maryland in the eastern United States. [The annual
91 mean discharge of Potomac River from 1895 to 2002 measured at Chain Bridge near Washington,
92 DC was 321 m³ s⁻¹ with a large interannual variability \(Jaworski et al., 2007\). The annual total
93 nitrogen loading was estimated to be around 27.7 × 10⁶ kg N year⁻¹ in 2008-2009 \(Bricker et al.,
94 2014\).](#) The Potomac River Estuary has experienced ecological degradation for decades partly due
95 to excess nutrient inputs including from the effluents of WWTPs (Bricker et al., 2014; Jaworski et
96 al., 2007). For example, the Blue Plains Advanced WWTP in Washington, D.C. is one of largest
97 WWTPs in the world, treating an average of ~1454 million liters of water per day. Pioneering
98 work in 1978 showed that Blue Plains WWTP was a large source of nitrogen to the Potomac River
99 Estuary, triggering high N₂O production and concentration downstream (McElroy et al., 1978).

100 Thanks to higher standards mandated by governmental agencies (nitrogen concentration [in
101 effluents](#) below 7.5 mg L⁻¹) starting in 1980s and the technical improvements in N removal from
102 the wastewater, the nitrogen concentration in effluents of WWTPs in the Potomac River has
103 decreased substantially (Pennino et al., 2016). However, the concurrent effect on N₂O
104 concentration is largely unknown. The Department of Environmental Quality (DEQ) of Virginia
105 maintains an approximately monthly routine monitoring program for water quality (e.g., nitrogen
106 concentration, phosphorus concentration, chlorophyll concentration) and physical properties (e.g.,
107 temperature, salinity, pH, and dissolved oxygen concentration) in the Potomac River Estuary but
108 not for N₂O. Therefore, we collaborated with DEQ of Virginia to measure the spatial and temporal
109 variation of N₂O concentrations in the Potomac River Estuary.

110

111 **Materials and Methods**

112 **Sample collection for N₂O and nutrients**

113 Surface waters at ~0.5 m depth at eleven stations in the tidal Potomac River Estuary were sampled
114 monthly or bimonthly (depending on the weather) on a vessel (Grady White 208) for the analysis
115 of DIN concentration, and both concentration and nitrogen isotopes of N₂O from April 2022 to
116 May 2023 (Figure 1). The eleven stations are characterized into 3 groups: embayment, downstream
117 of WWTPs, embayment not associated with WWTPs, and the central channel of the Potomac
118 River. Three embayment stations downstream of WWTPs are associated with three different
119 WWTPs: Noman Cole, Mooney and Aquia, all of which implement tertiary treatment of the
120 wastewater. We obtained the volume discharge and total N in treated water of each WWTP from
121 Discharge Monitoring Reporting required by Virginia Pollutant Discharge Elimination System
122 permit. Noman Cole WWTP discharges ~140.8 million liters of water and 370 kg N per day into
123 Pohick Creek. Mooney WWTP discharges ~54.9 million liters of water and 147 kg N per day into
124 the Neabsco Creek. Aquia WWTP discharges much less water and N into the Aquia Creek (~21.2
125 million liters per day and 35 kg N per day). The distances from the embayment stations
126 downstream of WWTPs to Noman Cole, Mooney, Aquia WWTPs were approximately 4, 1.8 and
127 5.8 km, respectively.

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132 The embayment stations were 2-3 meters deep while the average depth of central channel stations
 133 was around 8 meters. The embayment stations have been routinely sampled for water quality
 134 analyses by the DEQ of Virginia since the early 1970's. The central channel stations were added
 135 for this study. The purposes of this sampling design are to evaluate the impact of WWTPs on
 136 downstream distribution of DIN and N_2O , and to compare DIN and N_2O concentrations between
 137 edge and central channel of the river. The central channel is likely affected both by the Potomac
 138 mainstem flow and by the input from tributaries, while the embayment stations may be mainly
 139 affected by water flow from tributaries but also influenced by the tidal cycle (see the salinity
 140 change in Supplementary Figure 1b). While estuarine N_2O concentrations could be affected by
 141 tides (Gonçalves et al., 2015), sampling was not always conducted at the same tidal state due to
 142 logistic difficulties. Triplicate water samples for N_2O concentrations and isotopes were collected
 143 via a submersible pump into 60 mL serum bottles after overflowing three times the bottle's volume.
 144 After removing 3 mL water to create a 3 mL air headspace via a syringe, the serum bottles were
 145 immediately sealed with butyl stoppers and aluminum crimps and preserved with 0.5 mL of 10_M
 146 NaOH solution to stop biological activities. NaOH has been shown to be an effective and less
 147 environmentally hazardous preservative for N_2O and nutrient analysis (Frame et al., 2016; Wong
 148 et al., 2017).

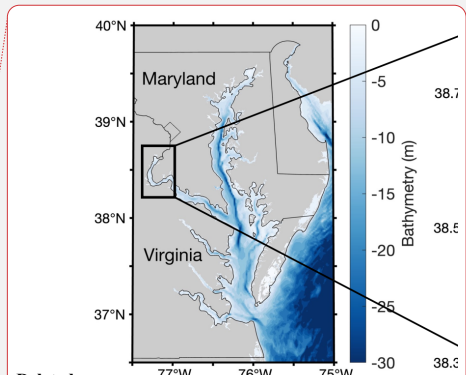
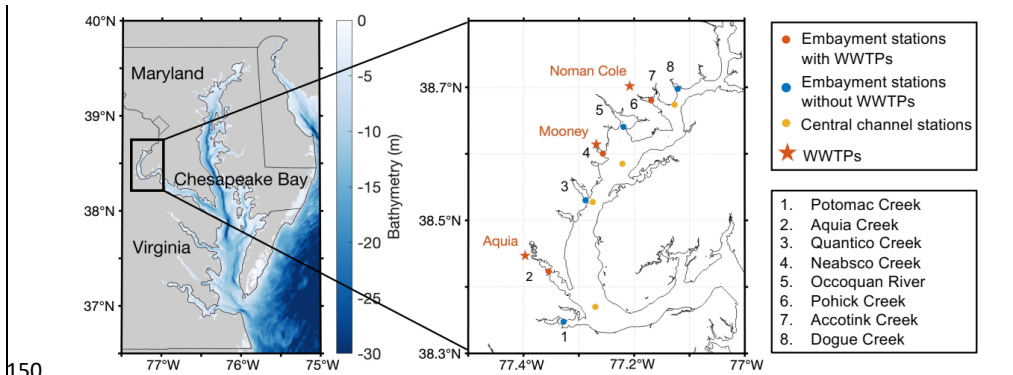
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151 Figure 1. Sampling stations in the Potomac River Estuary including embayment stations associated
 152 with WWTPs (red circles) and without WWTPs (blue circles), and central channel stations (yellow
 153 circles). Locations of WWTPs (Noman Cole, Mooney and Aquia) are shown in red stars.
 154 Creeks/ivers with sampling stations are numbered in the map with names shown in the legend.

161 Stream sampling sites upstream and downstream of WWTPs in creeks 4 – 7 are shown in Figure
162 4 below.

163
164 In addition to the routine sampling in the Potomac River Estuary, we also sampled its tributaries,
165 some of which were associated with the WWTPs, on May 18, 2023 (Figure 1) to specifically
166 evaluate the impact of WWTPs on downstream N₂O concentrations. Four creeks/ivers were
167 sampled including Neabsco Creek (5 stations: 2 stations upstream and 3 stations downstream of
168 Mooney WWTP), Occoquan River (3 stations, no WWTP), Pohick Creek (4 stations: 2 stations
169 upstream and 2 stations downstream of Noman Cole WWTP), and Accotink Creek (2 stations, no
170 WWTP). ~~Because Aquia WWTP discharges much less water and N into the Aquia Creek, its~~
171 ~~impact was not specifically investigated. Since water depths of these creeks/ivers were shallow,~~
172 ~~the water samples were collected by directly submerging 60 mL serum bottles into the surface~~
173 ~~water (~0.1 m) and preserving them as described above.~~

174
175 Besides N₂O sampling, temperature, salinity, and dissolved O₂ concentrations were recorded via a
176 YSI EXO1 sonde. Chlorophyll-a samples (300 mL) were filtered onto GF/F filters and kept on ice
177 in a cooler. The filters were then kept frozen at -20°C in the lab until analysis within 3 months
178 (Arar and Collins, 1997). Samples of total nitrogen and phosphorus (both particulate and
179 dissolved) were collected into 250 mL HDPE bottles and kept in ice in a cooler until analysis
180 within 48 hours on land (Rice et al., 2012; EPA, 1983).

181
182 **Measurement of N₂O and nutrient concentrations**

183 N₂O in the serum bottles was stripped by helium carrier gas into a Delta V Plus mass spectrometer
184 (Thermo) for the analyses of N₂O concentration and isotope ratio (m/z = 44, 45, 46) (Tang et al.,
185 2022). The total amount of N₂O in the serum bottles was determined using a standard curve of
186 N₂O peak area with N₂O standards containing a known amount of N₂O reference gas (0, 0.207,
187 0.415, 0.623, 0.831, 1.247 nmol N₂O). The total amount of N₂O dissolved in the water was
188 calculated after subtracting the amount of N₂O in 3 mL air headspace. Specifically, the monthly
189 atmospheric N₂O concentrations were obtained from the nearby atmospheric station in Brentwood,
190 Maryland (<https://gml.noaa.gov/>), (Andrews et al., 2023). The amount of N₂O in 3 mL air
191 headspace was generally less than 4% of the amount of N₂O dissolved in the 57 mL water samples.

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199 The N₂O concentration in samples was then calculated from the total amount of N₂O dissolved in
200 the water divided by the volume of water in the serum bottles. The detection limit and precision
201 of N₂O concentration measurement were 1.29 and 0.33 nM, respectively. We used N₂O produced
202 from nitrate isotope standards (USGS34 = -1.8‰ and IAEA = 4.7‰) to calibrate for δ¹⁵N of N₂O
203 samples.

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205 After analyzing N₂O concentration, samples were neutralized to pH ~7 by adding 10%
206 hydrochloric acid. NO₂⁻ + NO₃⁻ (NO_x⁻) concentration in these samples was measured using the
207 vanadium (III) reduction method by converting NO_x⁻ to NO, which was then quantified by
208 chemiluminescence analyzer (Braman and Hendrix, 1989). The detection limit of NO_x⁻
209 concentration was 0.15 μM. NH₄⁺ and NO₂⁻ concentrations were measured at a few selected
210 stations using the fluorometric orthophthalaldehyde method (Holmes et al., 1999) and the
211 colorimetric method (Hansen and Koroleff, 1999), respectively. Their concentrations were much
212 smaller than NO₃⁻ alone, mostly accounting for less than 10% of the DIN concentration. Therefore,
213 we only present NO_x⁻ data in this study.

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215 N₂O flux calculation

216 Surface N₂O flux was calculated using the following equation: $Flux = k \times (N_{2O_{measured}} -$
217 $N_{2O_{equilibrium}})$. The equilibrium N₂O concentration ($N_{2O_{equilibrium}}$) was calculated based on the
218 solubility of N₂O (Weiss and Price, 1980). The gas transfer velocity (k) was estimated based on
219 three different parameterizations: $k_{600} = 1.91 \times e^{0.35 \times U}$ (Raymond and Cole, 2001); $k_{600} =$
220 $0.314 \times U^2 - 0.436 \times U + 3.99$ (Jiang et al., 2008); $k = 0.251 \times U^2 \times (\frac{Sc}{660})^{-0.5}$ (Wanninkhof,
221 2014). U is the wind speed at the 10 m height obtained from the National Centers for
222 Environmental Prediction (NCEP) reanalysis (Kalnay et al., 1996;
223 <https://psl.noaa.gov/data/gridded/data.ncep.reanalysis.html>), Sc is the Schmidt number that could
224 be estimated as a function of temperature (Wanninkhof, 2014). Since our samples have salinity
225 close to 0, we used the parameterization of Sc for freshwater. Average values of the three N₂O flux
226 estimates are presented in the paper and N₂O fluxes estimated by different parameterizations are
227 provided in the associated dataset. We acknowledge large variations in estimating k values in the
228 riverine and estuarine systems by using different empirical models (Raymond and Cole, 2001;

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240 Borges et al., 2004; ~~Rosentreter et al., 2021~~). For instance, the effect of water velocity and water
241 depth on gas transfer velocity was not considered in the parameterizations above. Therefore, we
242 focus on ~~evaluating the spatiotemporal variations in~~ N₂O fluxes and their driving factors instead
243 of their absolute magnitude.

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245 Results and discussion

246 Spatial and temporal variations of N₂O concentrations in the Potomac River Estuary

247 Along the roughly 50 km sampling transect in the Potomac River Estuary, NO_x⁻ concentration
248 decreased from 98 to <1 μM from upstream to downstream (Figure 2a). NO_x⁻ concentration
249 showed a clear seasonal pattern: higher in winter and spring while lower in summer and fall. The
250 spatial and temporal patterns ~~were~~ likely attributable to the distribution of nutrient sources into the
251 Potomac River, DIN uptake and other removal processes along the river (Glibert et al., 1995;
252 Carstensen et al., 2015). For example, the maximum N loading into the Chesapeake Bay occurs in
253 winter and spring (Da et al., 2018). Meanwhile, N₂O concentration decreased from approximately
254 40 to 10 nM along the sampling transect and was higher in the fall and winter (Figure 2b). Since
255 temperature decreased from ~31°C in summer to 4°C in winter (Supplementary Figure 1a), the
256 increase in N₂O solubility in colder water during winter partly explained the seasonal change. In
257 contrast, N₂O saturation had higher values in summer and fall (Figure 2c), suggesting a higher
258 N₂O production in summer and fall. It is worth noting that N₂O saturation was above 100% at all
259 sampling stations with a maximum reaching 500%, indicating the Potomac River Estuary was a
260 consistent and strong source of N₂O to the atmosphere. N₂O flux ranged from ~~1~~ to ~~31.7~~ μmol N₂O
261 m⁻² d⁻¹ (Figure 2d). N₂O concentration (median: 18.2 nM) and flux (median: ~~5.6~~ μmol N₂O m⁻² d⁻¹
262) in the Potomac River Estuary were substantially higher than in the mainstem of the Chesapeake
263 Bay (2.6 ~~to~~ 20.9 nM N₂O with a median value at 10.6 nM and -0.3 ~~to~~ 4.3 μmol N₂O m⁻² d⁻¹ with a
264 median at 0.5 μmol N₂O m⁻² d⁻¹ (Tang et al., 2022; Laperriere et al., 2019)). Therefore, the
265 tributaries of the Chesapeake Bay (i.e., Potomac River) are intense sources of N₂O to the
266 atmosphere.

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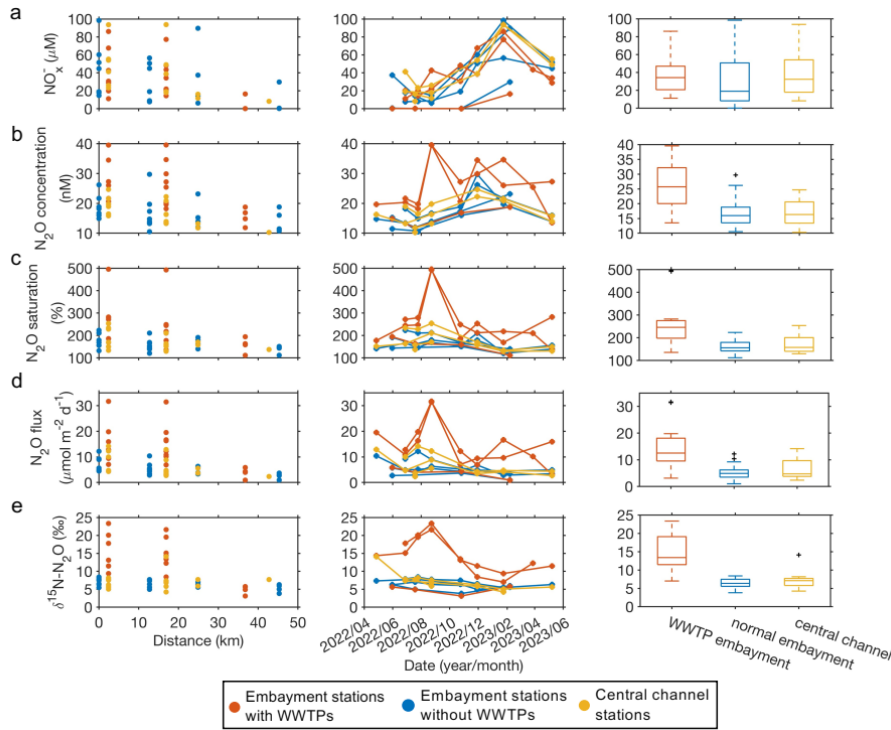
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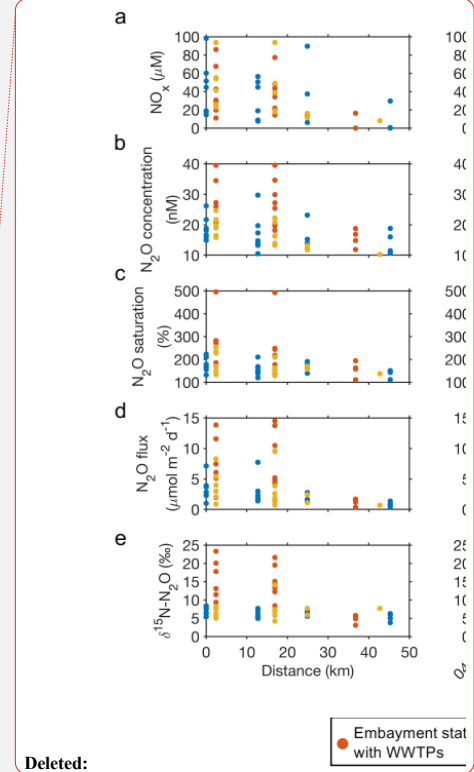


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279 Figure 2. Spatial and temporal variations of NO_x^- concentration (a), N_2O concentration (b), N_2O
 280 saturation (c), N_2O flux (d) and $\delta^{15}\text{N}$ of N_2O (e). The distance shows from upstream to downstream
 281 stations in the Potomac River. Embayment stations associated with WWTPs (red circles and lines)
 282 and without WWTPs (blue circles and lines), and central channel stations (yellow circles and
 283 lines). For the boxplots, the red line in each box is the median. The bottom and top of each box are
 284 the 25th and 75th percentiles of the observations, respectively. The error bars represent 1.5 times
 285 the interquartile range away from the bottom or top of the box, with black + signs showing outliers
 286 beyond that range. Embayment stations associated with WWTPs had significantly higher N_2O
 287 concentration, N_2O saturation, N_2O flux and $\delta^{15}\text{N}$ values compared to other stations ($p < 0.01$, t -
 288 test) but not significantly different NO_x^- concentration.

289

290 Stations close to each other had similar NO_x^- concentrations (e.g., upstream stations > downstream
 291 stations), regardless of station category (i.e., with WWTP, without WWTP, central channel of the



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293 Potomac River). In contrast, N₂O concentrations varied within locations according to the station
294 category: N₂O concentrations were substantially higher at stations downstream of WWTPs
295 ($p < 0.01$, t -test). N₂O concentrations were similar between stations in embayments without
296 WWTPs and the central channel (Figure 2). This suggests these WWTPs are efficient in removing
297 DIN from sewage and other sources but WWTPs may discharge N₂O directly into the effluent or
298 enhance downstream N₂O production (e.g., higher N₂O production yield from the same amount of
299 DIN). This effect extended to our sampling stations ~1.8-4 km downstream of the WWTPs.
300 However, the effect of WWTPs on downstream N₂O varied among stations. For example, elevated
301 N₂O concentrations were observed downstream from Noman Cole and Mooney WWTPs but not
302 downstream from Aquia WWTP. This difference may be related to the different N removal
303 processes of WWTPs that produce N₂O at different yields (de Haas and Andrews, 2022; Zhao et
304 al., 2024). However, we don't have detailed information about the three WWTPs other than that
305 they all implement tertiary treatment. In addition, the different dilution factors by riverine
306 discharges also matter. For example, the volume of effluent from Mooney WWTP was higher than
307 the discharge of Neabsco Creek while the volume of effluent from Aquia WWTP were generally
308 lower than the discharge of Aquia Creek (Supplementary Figure 2a-b). Particularly, the highest
309 N₂O concentration of up to 40 nM was found at two stations downstream of the Noman Cole and
310 Mooney WWTPs on August 23, 2022 when the river discharge was low (Supplementary Figure
311 2). Thus, the effect of WWTPs on downstream N₂O concentrations also varies seasonally (Schult
312 et al., 2023; Murray et al., 2020), with a relatively more important role in the dry season. Repeated
313 spatial and temporal sampling allowed us to capture these N₂O hotspots. Previous studies have
314 shown the impact of WWTPs on downstream N₂O concentration in aquatic environments. For
315 example, the highest N₂O concentration ~675 nM in the Potomac River was measured near the
316 discharge of the Blue Plains WWTP in 1977 (McElroy et al., 1978). Highest N₂O emissions in the
317 Ohio River near Cincinnati were attributed to direct input of N₂O from WWTPs' effluents
318 (Beaulieu et al., 2010).

319
320 In addition, a higher nitrogen isotopic signature ($\delta^{15}\text{N}$) of N₂O associated with WWTPs (median
321 $\delta^{15}\text{N}$ at 13‰) also suggests the distinct sources or cycling processes of N₂O compared to stations
322 of the central channel and without the influence of WWTPs (median $\delta^{15}\text{N}$ of N₂O at 6‰, Figure
323 2e) in the Potomac River Estuary. In comparison, the average $\delta^{15}\text{N}$ of N₂O in the tropospheric air

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332 is around 6.55‰ (Snider et al., 2015). $\delta^{15}\text{N}$ of N_2O for stations with the influence of WWTPs
333 showed a clear seasonal variation: higher in summer than winter (Figure 2e). This seasonal
334 difference may be related to the seasonal change in the relative importance of WWTPs' effluents
335 versus riverine discharge (Supplementary Figure 2c). For example, a relatively larger WWTPs'
336 effluent volume compared to the riverine discharge led to a larger $\delta^{15}\text{N}$ of N_2O in summer.
337 However, no clear seasonal pattern of $\delta^{15}\text{N}$ of N_2O was seen for stations without the influence of
338 WWTPs. $\delta^{15}\text{N}$ of N_2O produced in WWTPs depends on the treatment stages and aeration
339 conditions (Toyoda et al., 2011; Tumendelger et al., 2014). For example, the average $\delta^{15}\text{N}$ values
340 were reported to be -24.5‰ and 0‰ respectively for N_2O produced from nitrification during oxic
341 treatment versus N_2O produced from anaerobic denitrification in a California WWTP (Townsend-
342 Small et al., 2011). Our observed $\delta^{15}\text{N}$ of N_2O downstream of WWTPs was higher than the values
343 found in these urban WWTPs. One of the reasons for the increased $\delta^{15}\text{N}$ of N_2O may be partial
344 N_2O reduction via denitrification in the WWTPs, in downstream creeks, or in sediments; this
345 denitrification effect has been seen in the marine oxygen minimum zones (Kelly et al., 2021).
346 Denitrification as the cause of the elevated $\delta^{15}\text{N}$ is partly supported by the higher $\delta^{15}\text{N}$ of N_2O
347 when NO_x^- was reduced to less than 40 μM , suggesting the occurrence of N_2O reduction when the
348 concentration of other denitrification substrates became low (Supplementary Figure 3). However,
349 we do not know the exact locations where denitrification occurred (e.g., WWTPs, anoxic niches
350 in suspended particles, sediments), which deserves further investigations. The influence of
351 denitrification on unique isotopic signatures of N_2O produced from WWTPs has also been
352 observed in Tama River in Japan (Toyoda et al., 2009).

353

354 Environmental controls on N_2O concentrations

355 N_2O concentrations showed positive correlations with total N ($r=0.62$, $p<0.01$) and NO_x^-
356 concentrations ($r=0.51$, $p<0.01$) (Figure 3a). Correlation analyses done separately for stations with
357 or without WWTPs had similar patterns (Supplementary Figure 4). A better correlation between
358 the N_2O concentration and total N may indicate the contribution of other N sources besides NO_x^-
359 to N_2O production. N_2O could be produced from nitrification in the process of oxidizing NH_4^+ to
360 NO_x^- in the oxic environment as previously shown in the oxygenated mainstem of the Chesapeake
361 Bay (Tang et al., 2022). However, we can't exclude the possibility of N_2O production from
362 denitrification associated with anaerobic microsites in particles or in sediment (Beaulieu et al.,

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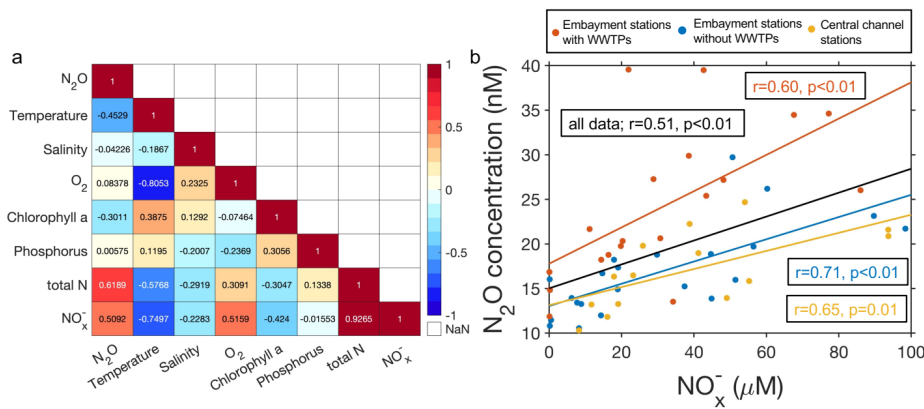
367 2011; Wan et al., 2023). Future investigations with ^{15}N tracers should be conducted to differentiate
 368 N_2O production pathways around the WWTPs. Furthermore, N_2O concentration was negatively
 369 correlated with temperature since higher temperature reduced the N_2O solubility. Although
 370 previous studies have showed dissolved oxygen to be an important driver of N_2O concentrations
 371 or fluxes in rivers and estuaries (Rosamond et al., 2012; Wang et al., 2015; Zheng et al., 2022), we
 372 did not find a strong dependence of N_2O on oxygen concentrations in the Potomac River Estuary
 373 (Figure 3a). This lack of strong dependence is probably because of the overall oxygenated
 374 conditions (Supplementary Figure 1c), and opposite correlations found in stations without WWTPs
 375 (positive) versus in stations with WWTPs (negative) (Supplementary Figure 4), which may be
 376 influenced by the different N_2O production pathways.

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379 Figure 3. (a) Correlation coefficients among different environmental factors and N_2O
 380 concentrations. (b) Relationship between N_2O and NO_x^- concentrations at different categories of
 381 sampling stations.

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382

383 The significant positive relationship between N_2O and NO_x^- concentration existed for samples
 384 collected at stations from all three different categories (Figure 3b). N_2O concentrations at stations
 385 downstream of WWTPs were notably higher than at other stations not associated with WWTPs
 386 even under the similar range of NO_x^- concentration. The larger slope of N_2O concentration versus
 387 NO_x^- concentration at stations downstream of WWTPs may be related to the direct input of N_2O
 388 from WWTPs into the downstream waters or different N_2O production pathways and production

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395 yields that deserve further investigations. The DIN concentration has been found to be a good
396 predictor of N₂O concentration and emission in many other rivers and estuaries (Murray et al.,
397 2015; Reading et al., 2020; Zheng et al., 2022). However, the correlation varied spatially, which
398 may be affected by the variable N₂O emission factors from DIN cycling. The emission factors are
399 affected by temperature, concentration and forms of N, oxygen, organic carbon concentration and
400 many other factors (Hu et al., 2016). The external N₂O input (e.g., input from WWTPs) could also
401 affect the relationship between N₂O and DIN concentrations (Dong et al., 2023). Compared to DIN
402 (~28 to 71 μM) and N₂O concentrations (~16 to 61 nM) measured approximately 45 years ago in
403 the same section of the Potomac River (McElroy et al., 1978), current DIN and N₂O concentrations
404 have slightly decreased. Thus, an additional benefit of nutrient regulation is the reduction of
405 greenhouse gas - N₂O - emissions, beyond improving water quality.

406
407 Since N₂O concentrations had the strongest correlation with total N concentrations (reflecting the
408 N₂O production potential) and temperature (affecting N₂O solubility), we developed a predictive
409 model of N₂O concentration based on total N and temperature. Predictions were performed
410 separately for stations with WWTPs ($N_2O \text{ concentration} = 0.115 \times \text{total N} - 0.241 \times$
411 $\text{temperature} + 17.185$, $n=18$, $r=0.78$; $p<0.01$) and without WWTPs including central channel
412 stations ($N_2O \text{ concentration} = 0.049 \times \text{total N} - 0.298 \times \text{temperature} + 18.888$, $n=23$,
413 $r=0.81$, $p<0.01$). The observed N₂O variability was generally captured by these simple linear
414 models (Supplementary Figure 5) but there were variabilities in the observations remaining to be
415 explained. Addition of other predictors did not significantly improve the model performance, so
416 we chose the simple predictive model that is mechanistically understandable. We then applied the
417 two predictive models separately to estimate N₂O concentrations at the embayment station in the
418 Pohick Bay (with WWTP) and the embayment station in the Occoquan Bay (without WWTP)
419 using total N concentration and temperature that were measured since 2008 by the DEQ of Virginia
420 monitoring program (Supplementary Figures 6 and 7). Predicted N₂O concentrations showed a
421 clear seasonality: higher in winter and lower in summer, N₂O concentrations in the Pohick Bay
422 decreased substantially (-0.9 nM/year) possibly due to the nutrient reduction (total N concentration
423 decreasing at 8.8 μM/year) over the last 14 years (Supplementary Figure 6). However, N₂O
424 concentrations in the Occoquan Bay only decreased slightly (-0.1 nM /year, not statistically
425 significant) along with the minor nutrient reduction (total N concentration decreasing at non-

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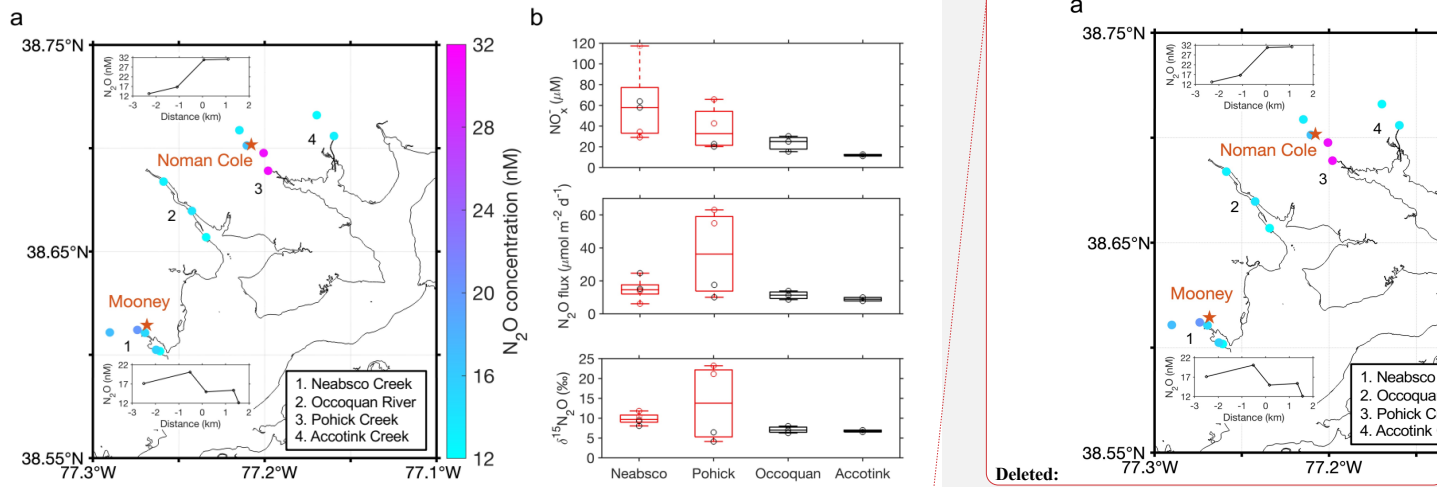
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440 statistically significant rate of 0.5 $\mu\text{M}/\text{year}$ (Supplementary Figure 7). Continuation of
441 environmental monitoring in the Potomac River (e.g., N nutrients and temperature), which is much
442 easier than sampling and measuring N_2O gas, could be used to indirectly estimate the changes in
443 N_2O concentrations in the future. These predictors are likely to be important in other estuaries, but
444 the weighting would vary among locations.

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446 **Impact of wastewater treatment plants on N_2O concentrations and emissions**

447 To further evaluate how WWTPs affect the N_2O distribution in the Potomac River, we measured
448 N_2O concentrations upstream and downstream of the two WWTP effluents (Mooney and Noman
449 Cole in Neabsco Creek and Pohick Creek, respectively) and compared them to N_2O concentrations
450 measured in two creeks that do not have WWTPs. Interestingly, the N_2O concentration at the
451 station upstream of Mooney WWTP in Neabsco Creek was higher than the N_2O concentration at
452 the station downstream of Mooney WWTP (20.1 vs 15.0 nM) (Figure 4a). The exact mechanisms
453 were not clear but one of the potential reasons could be the influence by tidal cycles: high tide
454 during the sampling time (salinity was 0.17 instead of 0) may have reversed the water flow and
455 diluted the WWTP effluent with low N_2O concentration Potomac water (12.1 nM at the outflow
456 of Neabsco Creek into the Potomac River Estuary). In contrast, we found substantially higher N_2O
457 concentration downstream of the Noman Cole WWTP (30.8 nM downstream vs 16.7 nM
458 upstream) in the Pohick Creek, which is less affected by the tidal cycle due to its semi-closed
459 geography (salinity was 0.12). The high downstream N_2O concentration may suggest the direct
460 addition of N_2O from WWTP effluent to the downstream environment. Furthermore, $\delta^{15}\text{N}$ of N_2O
461 in stations downstream of WWTPs were generally higher than the other two creeks that do not
462 have WWTPs (Figure 4b), confirming the distinct source of N_2O production by WWTPs found in
463 the Potomac River Estuary. Overall, the influence of WWTP effluents on downstream distribution
464 of N_2O is variable, and could be affected by the physical movement of water.



467

468 Figure 4. (a) Color-coded N₂O concentration at creek sampling stations on May 18, 2023. WWTPs
 469 (Mooney and Noman Cole) are shown in red stars. The insert figures show the change in N₂O
 470 concentrations as a function of distance up or down stream from the WWTPs. Creeks/ivers with
 471 sampling stations are numbered in the map with names shown in the legend. (b) Box plots of NO₃⁻,
 472 N₂O flux and δ¹⁵N of N₂O comparing four creeks. Neabsco and Pohick Creeks with WWTPs are
 473 displayed with red color boxes. Red and black circles in the boxplots show the data points of
 474 stations downstream and upstream/or without WWTPs, respectively. NO₃⁻, N₂O flux and δ¹⁵N of
 475 N₂O were clearly higher at stations downstream from the WWTP in Pohick Creek.

476

477 Dong et al. (2023) evaluated the potential impact of wastewater nitrogen discharge on estuarine
 478 N₂O emissions globally. Here we compiled data from previous studies with direct N₂O
 479 measurements in aquatic systems associated with WWTPs (not included in Dong et al., 2023) to
 480 assess the global impact of WWTPs on aquatic N₂O concentrations or emissions (McElroy et al.,
 481 1978; Hemond and Duran, 1989; Toyoda et al., 2009; Beaulieu et al., 2010; Rosamond et al., 2012;
 482 Chun et al., 2020; Masuda et al., 2021; Masuda et al., 2018; Dylla, 2019). WWTP effluents or
 483 water downstream of the WWTPs contain some of the highest N₂O concentrations and fluxes
 484 observed in the aquatic system (Table 1 and Supplementary Figure 8). For example, up to
 485 12,411.4% saturation of N₂O was measured in the effluent of WWTPs in the Tama River in Japan

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488 (Toyoda et al., 2009). In addition, N₂O flux up to 40,800 μmol N₂O-N m⁻² d⁻¹ was found
 489 downstream of the Regina WWTP in the Wascana Creek in Canada (Dylla, 2019). The
 490 downstream N₂O flux was >300 times higher than the N₂O flux upstream of the Regina WWTP.
 491 In comparison, the maximum N₂O saturation and flux previously reported in a global riverine N₂O
 492 dataset were around 2,500% and 12,754 N₂O-N m⁻² d⁻¹ (Hu et al., 2016). Across the sites listed in
 493 Table 1, N₂O concentration/saturation/flux downstream of the WWTPs was 1.45 to 374-fold of
 494 the upstream waters. The only exception was our observed decrease in N₂O concentrations
 495 downstream of Mooney WWTP on May 18, 2023, which was likely influenced by the tidal cycle.
 496 The wide range of apparent WWTP effect is related to many factors including the variable N₂O
 497 emission factors in the WWTPs, the ratio of ~~WWTP effluent volume to riverine discharge~~,
 498 the distance from the WWTPs where measurements were conducted, and the direction of water flow
 499 (e.g., tidal cycle). ~~In addition, the estuarine type, mixing regime, and stratification are also~~
 500 ~~important factors controlling N₂O emissions (Brown et al., 2022)~~. Overall, failing to account for
 501 ~~N₂O emissions~~ downstream of the WWTPs and ~~their~~ variability would substantially bias estimates
 502 of aquatic N₂O emissions. This uncertainty is increased by the fact that only a few observations
 503 are available (all in the northern hemisphere) (Supplementary Figure 8) compared to >58 000
 504 WWTPs present globally (Ehalt Macedo et al., 2022). It is also important to restrict the N₂O
 505 emission via efficient N₂O reduction in the WWTPs considering the projected increase in future
 506 wastewater production (Qadir et al., 2020).

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 508 Table 1. Global N₂O observations in aquatic systems associated with wastewater treatment plants.
 509 N₂O data are presented in concentration (nM), saturation (%) or flux (μmol N₂O-N m⁻² d⁻¹)
 510 according to how they are reported in different studies.

River/location	WWTP	N ₂ O upstream or in tributaries without WWTP	N ₂ O in WWTP effluent	N ₂ O downstream or in tributaries with WWTP	Average fold change (downstream vs upstream)	Reference
Potomac River/ Washington, D.C., USA	Blue Plains WWTP	11-34 nM		147-318 nM	9.3	McElroy et al., 1978
Assabet River/ Massachusetts, USA	Westborough WWTP	~10 nM	1045 nM	163 nM	16.3	Hemond and Duran. 1989
Tama River/ Tokyo, Japan	Plant 1 Plant 2	350.7% saturation 219.3%	12411.4% saturation 3326.2%	3454.8% saturation 1029.6%	9.8 4.7	Toyoda et al., 2009

Ohio River/ Cincinnati, USA		27.9 $\mu\text{mol N}_2\text{O-N m}^{-2} \text{ d}^{-1}$		1068 $\mu\text{mol N}_2\text{O-N m}^{-2} \text{ d}^{-1}$	38.2	Beaulieu et al., 2010
Grand River/ Ontario, Canada	e.g., Kitchener WWTP	4-12 $\mu\text{mol N}_2\text{O-N m}^{-2} \text{ d}^{-1}$		9-113 $\mu\text{mol N}_2\text{O-N m}^{-2} \text{ d}^{-1}$	9.4	Rosamond et al., 2012
Wascana Creek/ Saskatchewan, Canada	Regina WWTP	-32.5 to 109 $\mu\text{mol N}_2\text{O-N m}^{-2} \text{ d}^{-1}$	227 to 72800 $\mu\text{mol N}_2\text{O-N m}^{-2} \text{ d}^{-1}$	398 to 40800 $\mu\text{mol N}_2\text{O-N m}^{-2} \text{ d}^{-1}$	374	Dylla, 2019
Han River/ Seoul, Korea	JNW	39.7 nM	602.1 nM	441.6 nM	11.1	Chun et al., 2020
A-river	A-WWPT	61 nM	493 nM	180 nM	3	Masuda et al., 2021
B-river	B-WWTP	95	246	286	3	Masuda et al., 2018
C-river/Miyagi, Japan	C-WWTP	100	319	145	1.45	
Potomac River Estuary /Virginia, USA	Noman Cole Mooney Aquia	10.8-29.7 nM		11.87-39.5 nM	1.6	This study
Neabsco Creek/ Virginia, USA	Mooney	20.1 nM		15.0 nM	0.75	This study
Pohick Creek/ Virginia, USA	Noman Cole	16.7 nM		30.8 nM	1.84	This study

515

516 Conclusion

517 Taking advantage of the routine water monitoring program by the DEQ of Virginia, we detected
518 strong spatial and temporal variabilities of N_2O concentrations and emissions in the Potomac River
519 Estuary, a major tributary of Chesapeake Bay. Observations across the Potomac River Estuary also
520 allowed us to identify hotspots of N_2O emissions associated with WWTPs effluents. Higher N_2O
521 concentrations downstream of WWTPs compared to regions with similar nitrogen nutrient
522 concentrations suggested the direct discharge of dissolved N_2O from WWTPs and/or intense N_2O
523 production. A survey of globally available data shows N_2O concentrations or emissions are
524 consistently elevated in waters downstream from WWTPs. Future ^{15}N tracer incubations would
525 help to explain the high N_2O concentration downstream of WWTPs by disentangling the N_2O
526 production pathways. In addition, concurrent measurements of the N flux and N_2O concentration
527 downstream of WWTPs will help to constrain overall N_2O emission factors associated with
528 WWTPs. Our work could encourage potential collaborations between scientific community and
529 governmental agencies/the public to better observe the environmental pollution or quality, e.g.,
530 increasing the frequency and resolution of observations for N_2O and other greenhouse gases along
531 with many regularly monitored environmental factors like temperature and nutrients. Such efforts

532 may identify previously overlooked sources of N₂O emission and help to better estimate N₂O
533 emissions from aquatic systems.

534

535 **Data availability**

536 Data presented in this study has been deposited in Zenodo repository:
537 <https://doi.org/10.5281/zenodo.10775250>.

538

539 **Author contribution**

540 W. T. conceived the study. J. T., T. J., and W. T. collected N₂O samples from the Potomac River
541 Estuary. W. T. analyzed samples and interpreted data with other coauthors. W. T. wrote the first
542 draft of the manuscript with input from B. B. W. All coauthors contributed to the result discussion
543 and manuscript writing.

544

545 **Competing interests**

546 The authors declare that they have no conflict of interest.

547

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550 on May 18, 2023. We thank Elizabeth Wallace and Lindsay Pagaduan for analyzing the nutrient
551 samples. We thank Virginia Department of Environmental Quality for maintaining the routine
552 sampling and for providing the opportunity to collect N₂O samples in the Potomac River Estuary.

553 We thank Virginia Pollutant Discharge Elimination System for providing water discharge and
554 quality data of wastewater treatment plants. This study is supported by Princeton University.

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