



# Molecular level characterization of supraglacial dissolved organic matter sources and exported pools the southern Greenland Ice Sheet

Eva L. Doting<sup>1,2,\*</sup>, Ian T. Stevens<sup>1</sup>, Anne M. Kellerman<sup>3</sup>, Pamela E. Rossel<sup>4</sup>, Runa Antony<sup>4,5</sup>, Amy M. McKenna<sup>6,7</sup>, Martyn Tranter<sup>1</sup>, Liane G. Benning<sup>4,8</sup>, Robert G. M. Spencer<sup>3</sup>, Jon R. Hawkings<sup>2,9</sup> and Alexandre M. Anesio<sup>1</sup>

<sup>1</sup> Department of Environmental Science, iClimate, Aarhus University, Frederiksborgvej 399, 4000 Roskilde, Denmark

<sup>2</sup> Department of Earth and Environmental Science, University of Pennsylvania, Philadelphia, PA, USA

<sup>3</sup> National High Magnetic Field Laboratory Geochemistry Group and Department of Earth, Ocean, and Atmospheric Science, Florida State University, Tallahassee, FL, USA

<sup>4</sup> Interface Geochemistry Section, German Research Centre for Geosciences, GFZ Potsdam, Telegrafenberg, 14473 Potsdam, Germany

<sup>5</sup> National Centre for Polar and Ocean Research, Ministry of Earth Sciences, Goa, India

<sup>6</sup> National High Magnetic Field Laboratory, Florida State University, Tallahassee, Florida, 32310-4005, USA

<sup>7</sup> Department of Soil and Crop Sciences, Colorado State University, Fort Collins, CO, 80523, USA

<sup>8</sup> Department of Earth Science, Freie Universität Berlin, 12249 Berlin, Germany

<sup>9</sup> iC3, Department of Geosciences, UiT The Arctic University of Norway, Tromsø, Norway

Correspondence to: Eva L. Doting ([edoting@sas.upenn.edu](mailto:edoting@sas.upenn.edu))

**Abstract.** During the ablation season, active microbial communities colonise large areas of the Greenland Ice Sheet surface and produce dissolved organic matter (DOM) that may be exported downstream by surface melt. Meltwater flow through the bare ice interfluvial area, characterized by a porous weathering crust, is slow ( $\sim 10^{-2} \text{ m d}^{-1}$ ), meaning that it presents a potential site for photochemical and/or microbial alteration of supraglacial DOM. Transformations of supraglacial DOM during transport through the supraglacial drainage system remain unexplored, limiting our understanding of supraglacial DOM inputs to downstream subglacial and coastal ecosystems. Here, we employ negative-ion electrospray ionization 21 tesla Fourier transform ion cyclotron resonance mass spectrometry to catalogue the molecular composition of DOM in supraglacial dark ice, weathering crust meltwater, and supraglacial stream water sampled in a hydrologically connected supraglacial micro-catchment to address this knowledge gap. Dark ice DOM contained significantly more aromatic ( $25 \pm 3 \%$ ) and less biolabile ( $13 \pm 4 \%$ ) DOM than weathering crust meltwater ( $3 \pm 0$  and  $50 \pm 0 \%$ , respectively), pointing to retention of DOM on the ice surface and microbial, as well as photochemical alteration of DOM during transit through the supraglacial drainage system. These findings have implications for our understanding of supraglacial biogeochemical cycling, highlighting the importance of including the weathering crust photic zone when assessing supraglacial inputs to subglacial and downstream ecosystems.



## 1 Introduction

Microbial blooms dominated by the algae *Ancylonema alaskanum* and *Ancylonema nordenskiöldii* (Procházková et al., 2021; Lutz et al., 2018) cover large areas of the Greenland Ice Sheet ablation zone during the summer melt season (Cook et al., 2020; Uetake et al., 2010; Stibal et al., 2012). These algae produce a purple-brown pigment, purpurogallin carboxylic acid-6-O- $\beta$ -D-glucopyranoside (Remias et al., 2012), which provides protection from UV radiation and significantly lowers the albedo of the Greenland Ice Sheet ablation zone (Uetake et al., 2010; Yallop et al., 2012; Stibal et al., 2017; Ryan, 2017, 2018). Ice surface microbial production was found to correlate with supraglacial concentrations of carbohydrates and low-molecular weight compounds (Musilova et al., 2017), indicating that microbial communities are likely the primary driver of biolabile dissolved organic matter (DOM) production on the ice surface. Previous studies have shown that supraglacial stream and supraglacial snowpack DOM contain a high relative abundance of biolabile (D'Andrilli et al., 2015) aliphatic and peptide-like molecular formulae (Stubbins et al., 2012; Kellerman et al., 2021; Hemingway et al., 2019; Antony et al., 2017; Lawson et al., 2014), which are likely of microbial origin (Kellerman et al., 2018; Spencer et al., 2015). Yet, to date, DOM associated with algal blooms on glacier surface ice has not been characterized at the molecular level, limiting our understanding of surface ice microbial contributions to the biolabile character of supraglacial stream DOM.

The biolability of glacial runoff has been shown to correlate with increasing  $^{14}\text{C}$  age, meaning that glaciers are a source of both ancient and biolabile DOM (Hood et al., 2009; Stubbins et al., 2012; Spencer et al., 2014b; Singer et al., 2012). Radiocarbon dating of DOC in glacier ice and meltwaters from Alaska, the European Alps, Greenland, the Tibetan Plateau, and Ecuador all confirmed the presence of ancient carbon in glacial runoff (Bhatia et al., 2013; Stubbins et al., 2012; Singer et al., 2012; Spencer et al., 2014a, b; Holt et al., 2023). Aged DOM may be delivered to supraglacial surfaces by atmospheric deposition of soil or combustion-derived organic matter (Hood et al., 2009; Barker et al., 2009; Bhatia et al., 2010; Singer et al., 2012; Fellman et al., 2015; Li et al., 2018; Price et al., 2009; Spencer et al., 2014b; Stubbins et al., 2012; Bardgett et al., 2007; Holt et al., 2023). However, this source material would be expected to be characterized by high aromaticity (Chen and Jaffé, 2014; Fellman et al., 2013; Hansen et al., 2016; Li et al., 2018; Masiello, 2004) and hence appears disconnected from the aliphatic and peptide-rich DOM observed in supraglacial runoff. However, Holt et al. (2021) showed that photochemical degradation of both modern and aged aromatic organic matter sources common to glacier environments produces aliphatic compounds, potentially explaining the composition of glacial DOM. It is not yet understood whether this potential photochemical alteration of allochthonous DOM occurs during transport to glacier surfaces, on the glacier surface itself, or both.

To assess potential transformations of allochthonous and autochthonous DOM, the transport of water and associated DOM through the supraglacial drainage network must be considered. On the Greenland Ice Sheet, the supraglacial drainage network is comprised of streams, rivers, lakes and more than 200,000 km<sup>2</sup> of bare ice surface (Ryan et al., 2019). The majority of surface melt is generated in this bare ice interfluvial area (Steger et al., 2017), which is characterised by the presence of a



65 porous weathering crust, which forms due to shortwave radiation penetration into the surface ice, the consequent near-surface  
melting that arises, and the rate of percolation away from the melt zone (Stevens et al., 2018). Interstitial waterflow is slow ( $\sim$   
 $10^{-2}$  m d<sup>-1</sup>), meaning that this weathering crust photic zone is a likely site for microbial and/or photochemical alteration of  
supraglacial DOM because the water residence time here is days to weeks (Irvine-Fynn et al., 2021; Stevens et al., 2018; Yang  
70 et al., 2018). Typically, surface melt ends up in supraglacial streams that flow into moulins and crevasses that drain to the  
glacier bed (Yang and Smith, 2016). An understanding of supraglacial DOM characteristics and sources, and its  
transformations during transport through the supraglacial drainage system is therefore critical to assessing the nature and  
biolability of supraglacial DOM inputs to downstream subglacial and coastal ecosystems.

Here, we assess changes in DOM composition as it is transported through the supraglacial drainage system on the Greenland  
75 Ice Sheet. We use negative electrospray ionization 21 Tesla Fourier transform ion cyclotron resonance mass spectrometry (21  
T FT-ICR MS) to determine the molecular composition of dark surface ice, laboratory-generated dark surface ice debris  
leachate, weathering crust meltwater, and supraglacial stream meltwater sampled from a hydrologically connected micro-  
catchment. We reveal significant differences between supraglacial DOM pools, demonstrating the importance of the  
weathering crust in supraglacial biogeochemical processes with respect to the composition of supraglacial DOM that is  
80 delivered to downstream subglacial and coastal ecosystems.

## 2 Methods

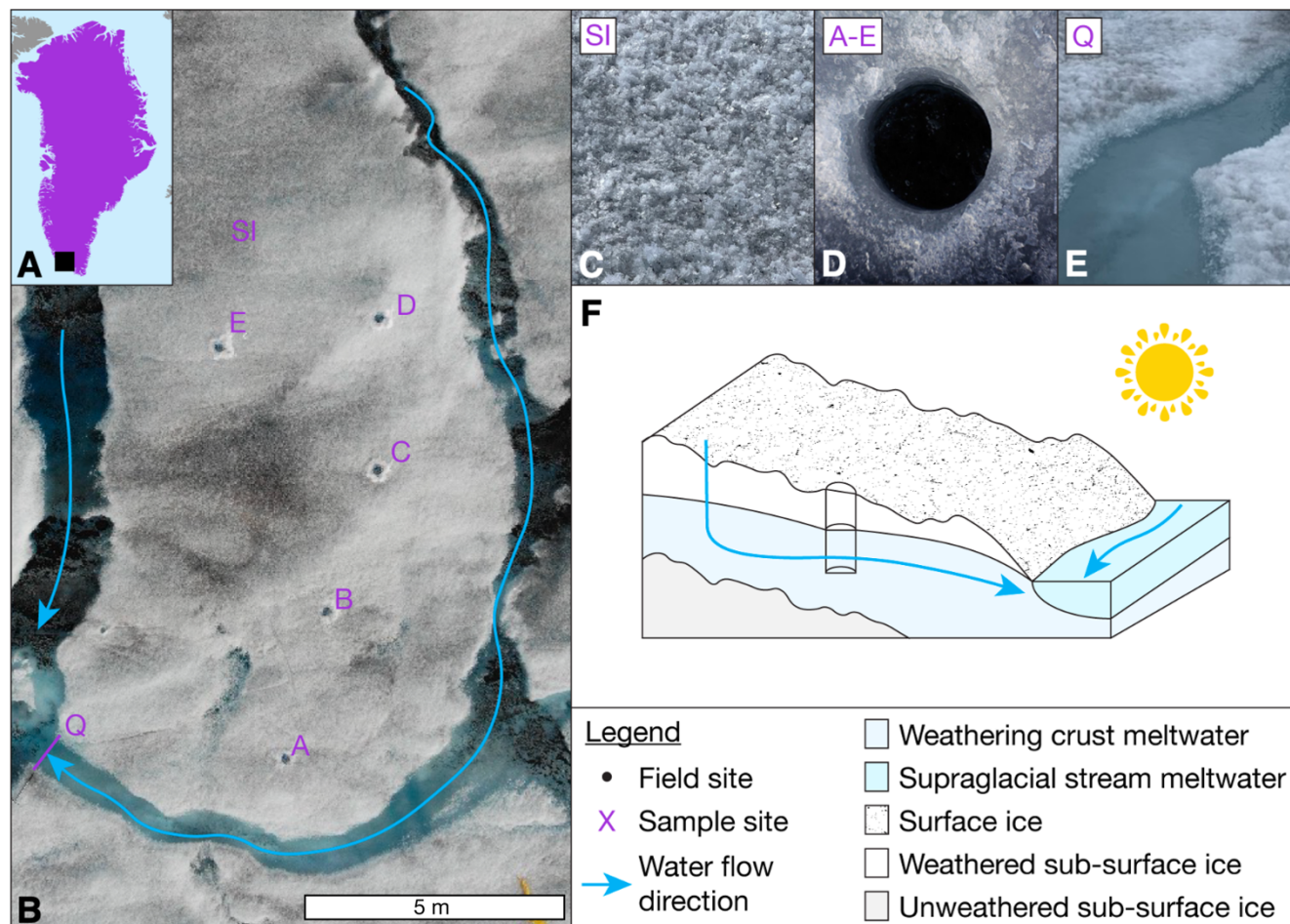
### 2.1 Site description, sample collection and field processing

A small supraglacial catchment on the southern Greenland Ice Sheet (61° 06' N, 46° 51' W; Fig. 1), located < 1 km from the  
QAS\_M PROMICE weather station (Fausto et al., 2021), was sampled on July 28, 2021 (Day of Year (DOY) 209). Within  
85 the catchment, samples were collected from surface ice, weathering crust meltwater (by sampling a refilled auger hole), and  
supraglacial stream water (Fig. 1B-E). These were supplemented with measurements of hydraulic conductivity and discharge  
in weathering crust auger holes A-E (Fig. 1B) and the supraglacial stream, respectively.

Samples for FT-ICR MS molecular level characterization were collected from surface ice between 14:00 and 15:00 (n = 4),  
90 weathering crust meltwater from auger hole D at 14:00 (n = 4), and supraglacial stream water at Q (n = 5) in Fig. 1. Surface  
ice samples (herein “dark ice”; Fig 1C) were collected from areas with visible debris and algae (observed by hand-held  
microscope in the field), using a sample-cleaned ice axe to scrape the top  $\sim$ 2 cm of ice into acid cleaned (1.2 M HCl) 1 L  
polycarbonate bottles. Ice was melted, filtered (combusted 47 mm, 0.7  $\mu$ m GF/F filter, Whatman), acidified (pH 2, HCl).  
Aliquots for DOC analysis were stored in furnace-dried 40 mL amber vials with acid-washed caps and PTFE-lined septa in the  
95 dark at 4 °C, while the rest of the acidified sample was stored in the dark in an acid cleaned PC bottle until solid phase extraction  
back in the home laboratory. Material retained on the filter (herein “surface debris”) was collected into 150 mL acid-cleaned



polycarbonate bottles and stored in the dark at  $-20\text{ }^{\circ}\text{C}$ . Weathering crust meltwater and supraglacial stream water samples were collected into 1 L PC bottles and processed as per the surface ice filtrate.



100 **Figure 1: (A) Map of Greenland indicating the approximate location of the study site; (B) drone image of the**  
**supraglacial catchment indicating water flow direction and sampling locations (indicated by Q for stream, A-E for**  
**weathering crust holes and SI for approximate area where dark surface ice was sampled); (C) close-up of typical dark**  
**surface ice, (D) close-up of typical weathering crust auger hole (diameter 14 cm); (E) close-up of the stream sampling**  
**location; and (F) schematic illustrating the sampling area with water flow direction, a weathering crust auger hole and**  
**the supraglacial stream.**

105

## 2.2 Near-surface hydrology

Recharge rate was measured in weathering crust auger holes at odd hours between 7:00 and 21:00, with a supplementary measurement at 14:00, using logging ultrasonic range finders. Hydraulic conductivity and water table height were calculated following Stevens et al. (2018). Combined with an uncrewed aerial vehicle (UAV)-derived orthophoto and digital elevation model (DEM), weathering crust meltwater flow direction and magnitude were modelled using the Spatial Analyst package in

110



ArcMap 10.8 (esri, USA), designed for terrestrial groundwater systems. A full description is included in the supplementary methods.

### 2.3 Preparation of laboratory leachates

115 Within four months of sample collection, surface debris samples, which had been kept at  $-20\text{ }^{\circ}\text{C}$  in the dark, were freeze-dried (ScanVac CoolSafe,  $-110\text{ }^{\circ}\text{C}$ ) and cryo-milled using stainless-steel sample holders (Retsch MM400). Sample holders were cleaned with Milli-Q and the first aliquot of milled sample was discarded. For each sample, 100 mg of milled material was weighed into acid-cleaned and combusted 4 mL glass vials, in triplicate, and 1 mL of Milli-Q was added to each triplicate to extract DOM that may feasibly be leached from debris on the ice surface. Vials were shaken at 500 rpm for 1 hour and then centrifuged at 3200 rpm for 10 minutes. To yield enough volume for DOC and FT-ICR MS analysis, supernatants were diluted  
120 in 50 mL Milli-Q water and filtered using an acid-washed and combusted glass syringe and pre-rinsed wwPTFE (Acrodisc One,  $0.2\text{ }\mu\text{m}$ , Pall) syringe filter. A sub-sample for DOC analysis was collected from each prepared sample before recombining the triplicates for solid phase extraction (Section 2.6). The laboratory-generated cold water surface ice debris leachate is referred to as ‘laboratory leachate’ throughout.

### 2.4 Dissolved organic carbon analysis

125 DOC measurements were carried out on a Shimadzu TOC-L<sub>CSH</sub> analyser with a high sensitivity platinum catalyst within four months of sample collection (samples were stored at  $4\text{ }^{\circ}\text{C}$  in the dark between collection and analysis). Samples were analysed in the non-purgeable organic carbon mode. Pre-acidified samples were sparged with carbon-free air for 2 min to eliminate inorganic carbon species before oxidizing the remaining DOC to  $\text{CO}_2$  through high-temperature combustion ( $680\text{ }^{\circ}\text{C}$ ), followed by non-dispersive infrared detection. Up to five replicate injections were made for each sample until the coefficient of variation  
130 (CV) for three of the replicate injections was  $\leq 2\%$ . Measurements were quantified using a potassium hydrogen phthalate (Sigma-Aldrich) calibration curve. The instrument quantification limit ( $27\text{ }\mu\text{g L}^{-1}$ ) was calculated from linear calibrations following the root mean square error method described by Corley (2003). Analytical precision calculated based on the standard error from seven repeat measurements of a  $100\text{ }\mu\text{g L}^{-1}$  standard was  $1.6\%$ .

### 2.5 Total carbon and total nitrogen analysis

135 Cryo-milled surface debris samples were analysed for total carbon (TC) and total nitrogen (TN) content on an Elemental analyser (Euro EA). Samples were oxidized in an oxygen atmosphere at a furnace temperature of  $950\text{ }^{\circ}\text{C}$ . After combustion, the resulting gases ( $\text{CO}_2$ ,  $\text{NO}_x$ ) were separated in a gas chromatography column at  $70\text{ }^{\circ}\text{C}$  and detected by thermal conductivity. The limit of quantification was  $0.15\%$  for both nitrogen and carbon. Overall precision for analyses of carbon and nitrogen was within  $5\%$  RSD. Results are reported in weight percent (wt. %).





## 140 2.6 Solid phase extraction

All DOM samples were solid phase extracted back in the home laboratory, following Dittmar et al. (2008), to remove inorganic interferences and concentrate the organic matter prior to FT-ICR MS analysis. Laboratory leachates were extracted on 3 mL, 100 mg, Bond Elut PPL cartridges. Weathering crust water, stream water and dark ice samples were extracted on 6 mL, 1 g, Bond Elut PPL cartridges. Samples were eluted with 6 mL of methanol into acid-soaked and combusted 10 mL amber glass  
145 vials. Eluates were dried under nitrogen flow and stored at -20 °C until analysis.

## 2.7 21 T Fourier Transform Ion Cyclotron Resonance Mass Spectrometry

Dried eluates were reconstituted in methanol prior to analysis, adjusting the volume to achieve a target concentration of 50 mg C L<sup>-1</sup>. DOM composition was analysed using a custom-built hybrid linear ion trap FT-ICR MS equipped with a 21 tesla superconducting solenoid magnet at the National High Magnetic Field Laboratory in Tallahassee, Florida (see supplementary  
150 methods; Hendrickson et al., 2015; Smith et al., 2018). Time-domain transients of 3.1 s (achieved mass resolving power,  $m/\Delta m_{50\%} > 2,000,000$  at  $m/z$  400) were conditionally co-added and acquired with the Predator data station, with 100 time-domain acquisitions averaged for all experiments (Blakney et al., 2011), phase-corrected (Xian et al., 2010), and internally-calibrated with 10-15 homologous series that span the entire molecular weight distribution based on the “walking” calibration method (Savory et al., 2011).

155 Mass spectra were calibrated, and molecular formulae were assigned (C<sub>1-100</sub>H<sub>4-200</sub>O<sub>1-30</sub>N<sub>0-4</sub>S<sub>0-2</sub>). Molecular formulae were classified by heteroatom (any atom other than carbon or hydrogen) type and number as oxygen only (CHO), oxygen and nitrogen (CHON), oxygen and sulphur (CHOS), and oxygen, nitrogen, and sulphur (CHONS). Molecular speciation was categorized based on neutral elemental ratios of H/C and O/C and the modified aromaticity index (AI<sub>mod</sub>), calculated from the neutral species following Eq. (1):

$$160 \quad AI_{mod} = \frac{1+C-\frac{1}{2}O-S-\frac{1}{2}(N+H)}{C-\frac{1}{2}O-N-S} \quad (1)$$

where C, H, O, N and S are the number of carbon, hydrogen, oxygen, nitrogen and sulphur atoms in a given molecular formulae (Koch and Dittmar, 2006, 2016). The nominal oxidation state of carbon (NOSC) was calculated following Eq. (2):

$$NOSC = 4 - \frac{(4C+H-3N-2O-2S)}{C} \quad (2)$$

Where C, H, O, N and S denote the number of atoms of each element in each formula (Riedel et al., 2012). Elemental  
165 compositions were classified into eight groups: condensed aromatics (AI<sub>mod</sub> ≥ 0.67), polyphenols (0.67 > AI<sub>mod</sub> > 0.50), peptide-like formulae (H/C ≥ 1.5, O/C ≤ 0.9, N > 0), sugar-like formulae (H/C ≥ 1.5, O/C > 0.9), and highly unsaturated and phenolic formulae (HUP; AI<sub>mod</sub> ≤ 0.50, H/C < 1.5) and aliphatics (H/C ≥ 1.5, O/C ≤ 0.9, N = 0), which were both separated



170 into high O/C ( $O/C > 0.5$ ) and low O/C ( $O/C < 0.5$ ) (Osterholz et al., 2016; Spencer et al., 2014b). The relative abundance (RA) of each assigned formula in a sample was obtained by dividing the signal magnitude of each individual  $m/z$  peak by the sum of all assigned signals in the sample. RA weighted metrics were calculated for the mass,  $AI_{mod}$ , NOSC, H/C and O/C. Calculated metrics can give a variety of information about the composition and lability of DOM in studied samples. For example, compounds with a  $H/C > 1.5$  have been related to more biolabile material, whereas compounds with a  $H/C < 1.5$  are considered less biolabile (D'Andrilli et al., 2015). A negative NOSC corresponds to more reduced compounds, whereas a positive NOSC corresponds to more oxidized ones. It has to be noted that these boundaries can be ambiguous as, for example, 175 the glacier ice algae polyphenolic pigment purpurogallin carboxylic acid-6-O- $\beta$ -D-glucopyranoside ( $C_{18}H_{18}O_{12}$ ) contains a two-ring aromatic core (Remias et al., 2012), but does not meet the  $AI_{mod}$  threshold for polyphenolics due to its glucopyranoside sidechain and is hence classified as a high O/C HUP. However, the formulae classifications outlined above have shown systematic and biogeochemical coherence with sources and degradation patterns of DOM (Stubbins et al., 2010; D'Andrilli et al., 2015; Kellerman et al., 2018; Spencer et al., 2015) and are therefore informative for use in the present study.

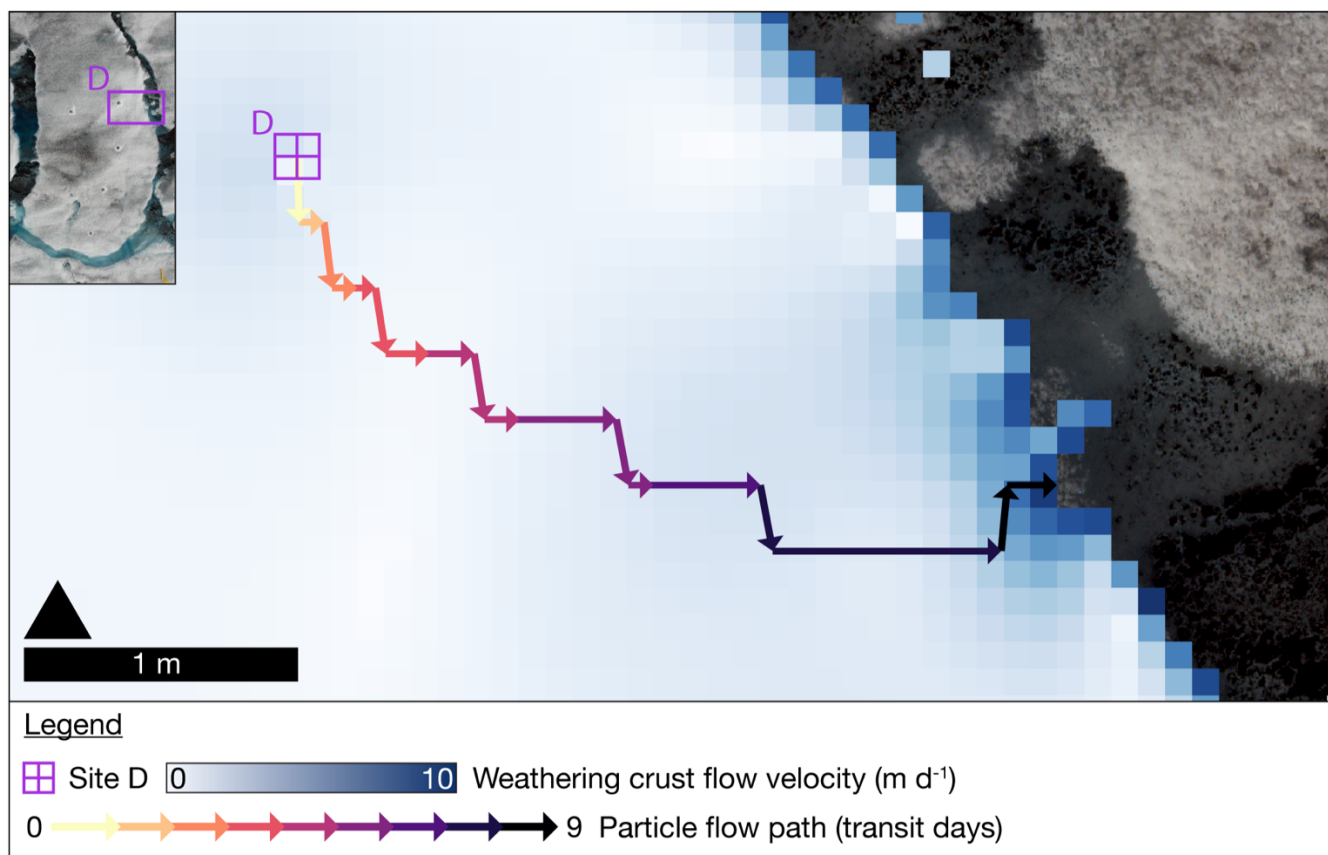
## 180 2.8 Statistics

All statistical analyses were performed in R (R Team, 2014). Peaks with a high intensity (relative abundance  $> 0.1\%$ ) in the procedural field blank were removed from the dataset as they were assumed to be contaminants, and data were renormalized to the total sum of assigned signals before exploring DOM characteristics. Pairwise comparisons were performed between samples grouped as laboratory leachate, dark ice, weathering crust meltwater, and supraglacial stream water. Bartlett's test 185 was used to assess homogeneity of variance for all variables. If variance was equal, a one-way ANOVA assuming equal variance was used, followed by pairwise comparison using a t-test if the ANOVA was significant, and p-values were adjusted using Bonferroni correction. If variance was unequal, a one-way ANOVA assuming unequal variance was used, followed by pairwise comparison using a t-test assuming unequal variance if the ANOVA was significant, and p-values were adjusted using Šidák correction. Prior to Principal Component Analysis (PCA) using the R package 'vegan' (Oksanen et al., 2011), variables 190 were unit variance scaled.

## 3 Results

### 3.1 Near-surface hydrology of the study site

Our hydrological modelling approach reveals that meltwater from Hole D, where weathering crust meltwater samples were collected, transits from the sampling point within the weathering crust in a south-easterly direction to the main supraglacial 195 stream over a period of nine days (Figure 2) assuming the prevailing weather crust state during the study period remains constant.



**Figure 2: Results of the flow direction and magnitude model, with pixel colour indicating weathering crust water flow velocity. Arrows indicate the modelled particle flow path through the weathering crust from Hole D to the supraglacial stream.**

### 200 3.2 Bulk DOC, TC and TN concentrations

Dark ice DOC concentrations were similar ( $0.90 - 1.01 \text{ mg L}^{-1}$ ) in the four samples collected within the micro-catchment and were significantly higher than the DOC concentrations in weathering crust meltwater and supraglacial stream water (Table 1). The surface ice debris, from which the laboratory leachate was generated, contained between 3.00 – 3.35 wt. % TC and between 0.25 – 0.28 wt. % TN. The mean coefficient of variation of DOC concentrations between triplicate extractions of surface debris  
205 of the same sample was 2.4%.

### 3.3 Molecular level composition of supraglacial DOM pools

A total of 24,578 unique molecular formulae were assigned across the dataset, with between 6,385 and 9,667 formulae assigned in individual samples (Table 1). There was no significant difference between sample types (laboratory leachate, dark ice, weathering crust and supraglacial stream water) in terms of the number of formulae assigned, and 2,885 formulae were shared  
210 between all samples in the dataset. For laboratory leachate samples, on average 63% (3,653 – 6,053 formulae) of the assigned formulae were also present in the corresponding dark ice sample. The mean RA weighted mass (Table 1) was higher in dark





ice ( $450 \pm 10$  Da) and supraglacial stream water ( $452 \pm 12$  Da) samples than in the laboratory leachates ( $407 \pm 10$  Da). Mean RA weighted NOSC and H/C ratio were significantly lower and higher, respectively, in laboratory leachate, weathering crust meltwater and supraglacial stream than in dark ice (Table 1), corresponding to a significantly lower prevalence of aliphatic and peptide-like compounds ( $13 \pm 4$  %RA) in dark ice DOM than in the rest of the dataset (32.8 – 68.3 % RA).

The %RA of heteroatom classes differed significantly between sample types (Table 1). Laboratory leachates were composed of approximately equal portions of CHO (35.9 – 58.0 %RA) and CHOS (33.3 – 55.2 %RA), with relatively minor contributions of CHON (4.9 – 10.6 %RA) and CHONS (0.25 – 0.47 %RA) formulae. Dark ice DOM was predominantly composed of CHO (91.2 – 96.5 %RA), with minor contributions of CHON (3.5 – 6.9 %RA) and CHOS (0 – 1.9 %RA) formulae. The prevalence of CHON in weathering crust meltwater and supraglacial stream samples was similar (13.9 – 15.5 and 14.9 – 16.7 %RA, respectively), but there was a significant difference in the contribution of CHO (63.3 – 70.5 and 73.6 – 83.8 %RA, respectively) and CHOS (14.5 – 22.1 and 0.6 – 11.5 %RA, respectively). No CHONS formulae were assigned in dark ice, weathering crust meltwater, or supraglacial stream samples.

Laboratory leachate, weathering crust meltwater, and supraglacial stream DOM was predominantly composed of aliphatic and HUP compounds (Table 1). Aliphatic compounds accounted for approximately half of the DOM in laboratory leachates (46.4 – 62.1 %RA), just under half in weathering crust meltwater samples (40.8 – 47.4 %RA) and roughly a third of DOM in supraglacial stream samples (28.4 – 38.3 %RA). Dark ice DOM was comprised predominantly of HUP (59.0 – 68.5 %RA) and aromatic (21.8 – 27.5 %RA) compounds. High O/C HUP compounds with the molecular formula  $C_{18}H_{18}O_{12}$ , which may include the algal pigment purpurogallin carboxylic acid-6-O- $\beta$ -D-glucopyranoside, accounted for 7.3 – 13.9 %RA of dark ice DOM, 2.1 – 7.4 %RA of laboratory leachate DOM, 0.02 – 0.04 %RA of weathering crust meltwater DOM, and 0.01 %RA of supraglacial stream DOM. Polyphenolic compounds with the molecular formula  $C_{12}H_8O_7$ , which may include the aglycone degradation product of the algal purpurogallin pigment, accounted for 5.7 – 8.8 %RA of dark ice DOM, 0.2 – 2.5 %RA of laboratory leachate DOM, 0.02 %RA of weathering crust meltwater DOM, and 0.01 %RA of supraglacial stream DOM. Aromaticity in dark ice was high ( $AI_{mod}$  0.305 – 0.340) compared to laboratory leachate, weathering crust meltwater, and supraglacial stream samples ( $AI_{mod}$  0.123 – 0.172), in which aromatic compounds made up less than 5 %RA of DOM.

The number of formulae, or molecular diversity, of aromatic compounds assigned was highest for dark ice (765 – 957), followed by weathering crust meltwater (491 – 547), supraglacial stream water (271 – 420), and was lowest in laboratory leachates (153 – 440). The number of biolabile (aliphatic and peptide-like) formulae was highest in laboratory leachate and weathering crust samples (2,984 – 3,719 and 3,124 – 3,341, respectively), and lowest in dark ice (1,584 – 2,640). The number of biolabile formulae assigned in supraglacial stream samples (2,498 – 2,890) was similar to weathering crust meltwater samples. The %RA of peptide-like compounds was significantly lower in dark ice (1.3 – 2.8 %RA) than in other sample types (Table 1).



250

**Table 1: Dissolved organic carbon concentrations and dissolved organic matter composition for each sample type, expressed as mean (standard deviation). %RA denotes percent relative abundance, # denotes number of formulae. Significance (adjusted p-value < 0.05) of pairwise comparisons is denoted by capital letters, where values that have at least one letter in common are not significantly different from one another.**

	Laboratory leachate	Dark ice	Weathering crust	Stream
<b>DOC (mg L<sup>-1</sup>)</b>	386 (85) <sup>A*</sup>	0.94 (0.05) <sup>B</sup>	0.18 (0.04) <sup>C</sup>	0.14 (0.01) <sup>C</sup>
<b>Formulae (#)</b>	8,403 (864) <sup>A</sup>	7,570 (965) <sup>A</sup>	9,008 (587) <sup>A</sup>	8,343 (1022) <sup>A</sup>
<b>Mass<sup>wa</sup> (Da)</b>	407 (10) <sup>A</sup>	450 (10) <sup>BC</sup>	429 (9) <sup>AB</sup>	452 (12) <sup>C</sup>
<b>AI<sub>mod</sub><sup>wa</sup></b>	0.149 (0.022) <sup>A</sup>	0.326 (0.016) <sup>B</sup>	0.158 (0.003) <sup>A</sup>	0.162 (0.006) <sup>A</sup>
<b>NOSC<sup>wa</sup></b>	-0.652 (0.120) <sup>A</sup>	0.112 (0.061) <sup>B</sup>	-0.569 (0.045) <sup>AC</sup>	-0.470 (0.043) <sup>C</sup>
<b>H/C<sup>wa</sup></b>	1.471 (0.056) <sup>A</sup>	1.072 (0.033) <sup>B</sup>	1.438 (0.013) <sup>A</sup>	1.405 (0.019) <sup>A</sup>
<b>O/C<sup>wa</sup></b>	0.369 (0.039) <sup>A</sup>	0.582 (0.019) <sup>B</sup>	0.406 (0.018) <sup>AC</sup>	0.445 (0.014) <sup>C</sup>
<b>CHO (%RA)</b>	47 (9) <sup>A</sup>	94 (2) <sup>B</sup>	68 (3) <sup>C</sup>	81 (4) <sup>D</sup>
<b>CHON (%RA)</b>	8 (2) <sup>A</sup>	5 (2) <sup>A</sup>	15 (1) <sup>B</sup>	16 (1) <sup>B</sup>
<b>CHOS (%RA)</b>	44 (9) <sup>A</sup>	1 (1) <sup>B</sup>	17 (3) <sup>C</sup>	3 (5) <sup>B</sup>
<b>CHONS (%RA)</b>	0.37 (0.09) <sup>A</sup>	0 (0) <sup>B</sup>	0 (0) <sup>B</sup>	0 (0) <sup>B</sup>
<b>Aliphatic High O/C (%RA)</b>	5 (2) <sup>AB</sup>	4 (1) <sup>B</sup>	7 (1) <sup>AC</sup>	9 (1) <sup>C</sup>
<b>Aliphatic Low O/C (%RA)</b>	51 (7) <sup>A</sup>	7 (2) <sup>B</sup>	37 (3) <sup>C</sup>	25 (4) <sup>D</sup>
<b>HUP High O/C (%RA)</b>	21 (8) <sup>A</sup>	50 (7) <sup>A</sup>	21 (3) <sup>A</sup>	25 (3) <sup>A</sup>
<b>HUP Low O/C (%RA)</b>	15 (2) <sup>AB</sup>	12 (3) <sup>B</sup>	26 (1) <sup>A</sup>	34 (2) <sup>A</sup>
<b>Peptide-like (%RA)</b>	5 (2) <sup>A</sup>	2 (1) <sup>B</sup>	7 (1) <sup>A</sup>	5 (1) <sup>A</sup>
<b>Condensed aromatic (%RA)</b>	0 (0) <sup>AC</sup>	8 (1) <sup>B</sup>	1 (0) <sup>C</sup>	0 (0) <sup>A</sup>
<b>Polyphenolic (%RA)</b>	2 (2) <sup>A</sup>	17 (2) <sup>B</sup>	3 (0) <sup>CD</sup>	2 (0) <sup>AD</sup>
<b>Sugar (%RA)</b>	0.17 (0.10) <sup>A</sup>	0.03 (0.02) <sup>B</sup>	0.03 (0.02) <sup>B</sup>	0.04 (0.02) <sup>B</sup>
<b>Aromatic compounds (%RA)</b>	3 (2) <sup>AB</sup>	25 (3) <sup>C</sup>	3 (0) <sup>A</sup>	2 (0) <sup>B</sup>
<b>Aliphatic + peptide-like (%RA)</b>	62 (8) <sup>A</sup>	13 (4) <sup>B</sup>	50 (2) <sup>C</sup>	38 (4) <sup>D</sup>
<b>Aromatic compounds (#)</b>	286 (134) <sup>A</sup>	862 (80) <sup>B</sup>	523 (24) <sup>C</sup>	362 (70) <sup>AC</sup>
<b>Aliphatic + peptide-like (#)</b>	3,395 (320) <sup>A</sup>	2,088 (464) <sup>B</sup>	3,228 (119) <sup>AC</sup>	2,698 (148) <sup>C</sup>

\* DOC concentration in supernatant (1 mL) after cold water extraction of 100 mg cryo-milled dark surface ice debris

<sup>wa</sup> Relative abundance weighted average

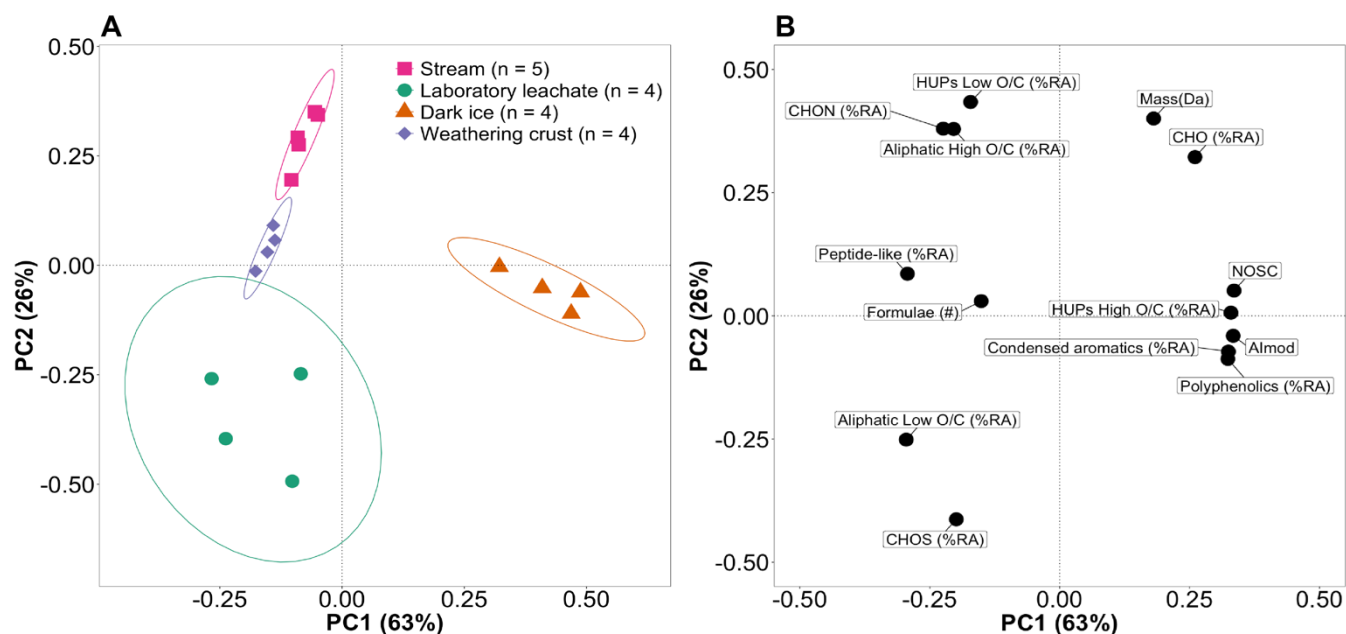


### 3.3 Compositional differences between hydrologically connected DOM pools

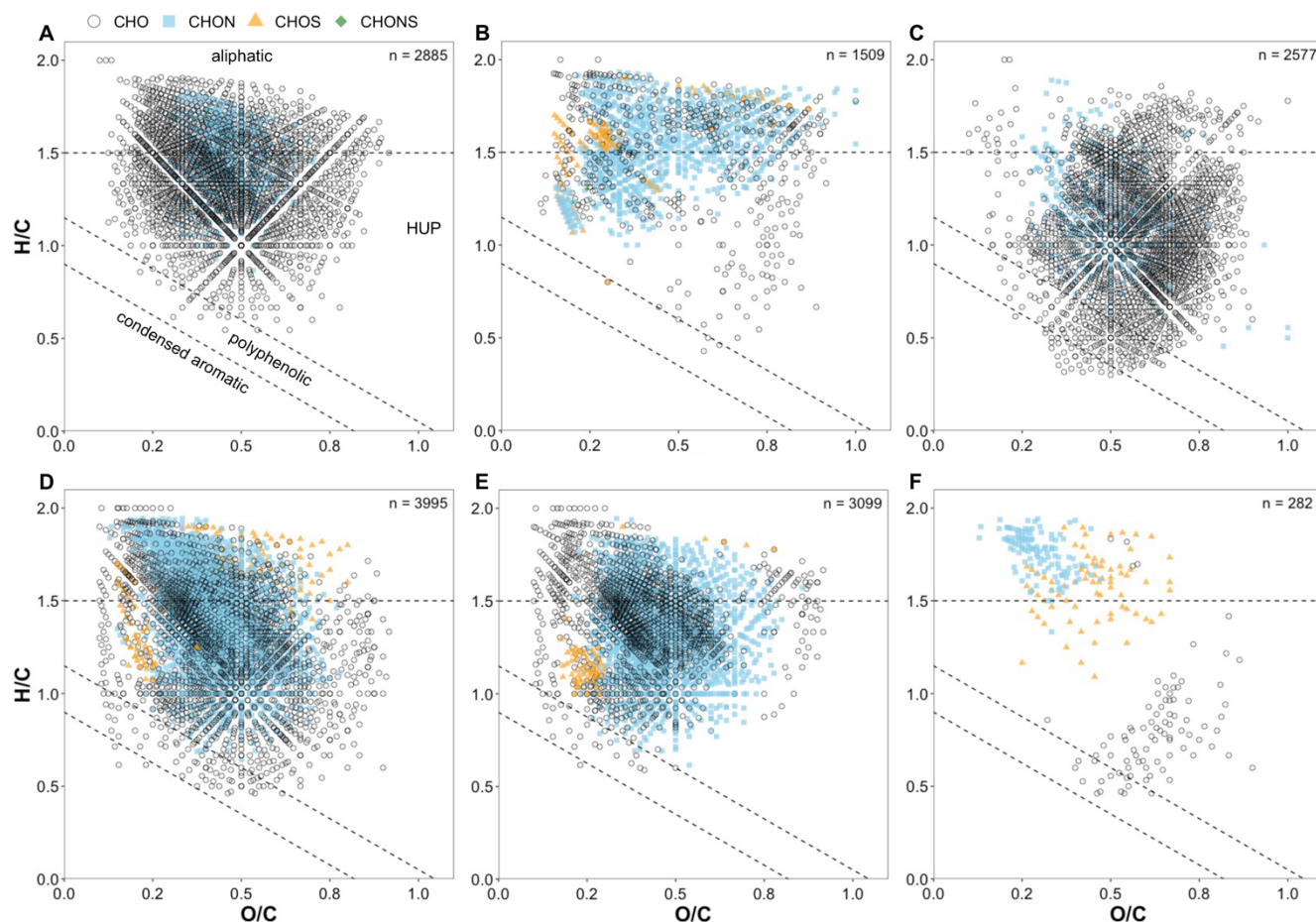
255 Principal component (PC) analysis was used to assess DOM parameters that distinguish the different sample groups in more detail (Fig. 3). PC1 explained 63% of variance in the data and correlated positively with NOSC,  $AI_{mod}$ , high O/C HUP, condensed aromatic and polyphenolic compounds, as well as %RA of CHO, and mean RA weighed mass. PC1 correlated negatively with molecular diversity (number of formulae), peptide-like compounds, low and high O/C aliphatic compounds, low O/C HUP compounds, and %RA of CHON and CHOS. Dark ice samples separated from the other sample groups along

260 PC1, reflecting the higher aromaticity and lower relative abundance of biolabile peptide-like and aliphatic compounds. PC2 explained a further 26% of variance in the data, correlating positively with low O/C HUP and high O/C aliphatic compounds, %RA of CHO, %RA of CHON, and mean RA weighed mass. Finally, PC2 correlated negatively with low O/C aliphatic compounds and %RA CHOS. Laboratory leachate, weathering crust meltwater and supraglacial stream samples separate along PC2, with the former containing a higher %RA of CHOS formulae and low OC aliphatic compounds. Weathering crust

265 meltwater and supraglacial stream samples formed two separate clusters, driven by differences in prevalence of low O/C aliphatic compounds and sulphur-containing compounds. Overall, the four sample types present significantly different DOM compositions.



270 **Figure 3: (A) PC analysis scores plot of DOM composition in laboratory leachate (green circles), dark ice (orange triangles), weathering crust meltwater (purple diamonds) and supraglacial stream water (pink squares) samples, with ellipses representing 90% confidence intervals; and (B) loadings plot of the variables included in the PC analysis.**



275

Figure 4: van Krevelen diagrams showing molecular formulae assigned (A) in all samples in the dataset; (B) in all laboratory leachates; (C) in all dark ice samples; (D) in all weathering crust meltwater samples; (E) in all supraglacial stream samples; and (F) in all weathering crust meltwater samples but not in any supraglacial stream sample. In panels B-E, formulae presented in panel A are excluded. The total number of molecular formulae displayed in each diagram is denoted in the top right of each panel, and data points are coloured according to their assigned heteroatom class with CHO compounds in black open circles, CHON in light blue squares, CHOS in orange triangles and CHONS in green diamonds. Note that only two CHONS formulae are shown in this figure, both in panel B.

280

Data were plotted in van Krevelen space to visualize the molecular composition and diversity of the core supraglacial DOM pool (Fig. 4A) and the different sample types (Fig. 4B-E). The core supraglacial DOM signature, consisting of 2,885 molecular formulae that were assigned in every sample in the dataset, was comprised mainly of HUP and aliphatic compounds with the heteroatomic formula CHO or CHON (Fig. 4A). To visualize the difference between laboratory leachates, dark ice, weathering crust meltwater and stream water, molecular formulae that were assigned in all samples, but were not present in the core supraglacial DOM signature, were plotted in Fig. 4B-E, respectively. The laboratory leachate signature contained 1,509 molecular formulae, that predominantly occupied the aliphatic and HUP space in the van Krevelen diagram (Fig. 4B). The dark ice signature contained 2,577 formulae, with most formulae falling in the HUP region of the van Krevelen diagram (Fig.

290



4C) and a clear contribution of polyphenolic and condensed aromatic formulae. The weathering crust meltwater signature was the most molecularly diverse with an additional 3,995 formulae present in all samples, falling mostly in the aliphatic and HUP space, with ~52% of formulae containing nitrogen. The supraglacial stream signature was relatively similar to the weathering crust meltwater signature, but with fewer formulae (3,099) in all regions of the van Krevelen diagram. To further examine these differences, molecular formulae present in all weathering crust meltwater samples but not in any of the supraglacial stream samples are highlighted in Fig. 4F, where approximately 60% of the formulae fall in the aliphatic region and contain sulphur or nitrogen. Of the 282 formulae in Fig. 4F, a total of 182 formulae had also been assigned in at least one dark ice or laboratory leachate sample. Across the entire dataset, only 24 formulae were present in all supraglacial stream samples but not in any weathering crust meltwater samples.

## 4 Discussion

### 4.1 Dark ice, surface ice debris, and laboratory generated leachates

Dark ice samples analysed in this study were collected by scraping off the top 2 cm of surface ice in patches with a visible debris loading. The DOC concentrations in these dark ice filtrates ( $0.90 - 1.01 \text{ mg C L}^{-1}$ ) fell within the range of values reported by Lutz et al. (2017) for 12 glaciers in Svalbard and Arctic Sweden ( $0.27 - 2.33 \text{ mg C L}^{-1}$ ), but was higher than those reported for dark ice surfaces ( $0.17 - 0.32 \text{ mg C L}^{-1}$ ) on Leverett glacier in southwest Greenland (Musilova et al., 2017). Surface ice DOC concentrations depend on local atmospheric deposition (Stubbins et al., 2012), microbial abundance and productivity (Musilova et al., 2017), mineral dust or particulate loadings (McCutcheon et al., 2021), and DOC contained within the ablating glacier surface. In addition, lysis of cells during sampling, thawing, or filtering might contribute to dark ice DOC concentrations or DOM composition. However, as lysis of microbial cells also takes place in undisturbed ice surface communities, for example as a result of fungal infections of glacier ice algal cells (Fiołka et al., 2021), the potential effect of cell lysis during sampling cannot be quantified or mitigated, but is unlikely to significantly alter the dark ice DOM composition. If cell lysis during sampling was the main source of surface ice DOM, we would expect to see similarities in dark ice and laboratory water leachate DOM composition, but these two sample groups present distinct DOM signatures (Table 1, Fig. 3A, Fig 4B-C) and we therefore assume that the dark ice DOM composition reported here is not a sampling artifact.

Surface ice debris removed during filtration of dark ice samples was used to determine TC ( $3.00 - 3.35 \%$ ) and TN ( $0.25 - 0.28 \%$ ) content, and to extract a laboratory-generated cold-water leachate for DOM characterisation. For all debris samples, TC and TN values were in the lower end of the range of values reported for ice with a high biomass loading (TC:  $2.59 - 8.45 \%$ , TN:  $0.20 - 0.87 \%$ ; McCutcheon et al., 2021), likely because our sample collection took place before the peak of the typical ablation season microbial bloom and because the ice surface in the sampling region had a relatively high particulate loading (microscope observations in the field). Interestingly, laboratory leachate and dark ice DOM compositions were significantly different (Table 1, Fig. 3, Fig. 4B-C) despite originating from the same sample. Laboratory leachate DOM had a lower mean





RA weighted mass and aromaticity and a higher %RA of aliphatic and peptide-like compounds than dark ice DOM (Table 1). The freeze-drying and cryo-milling of surface debris prior to preparation of the leachate likely lysed cells within the debris samples, yielding DOM with a higher prevalence of aliphatic and peptide-like compounds, which are typically of microbial origin (Kellerman et al., 2018; Spencer et al., 2015). When comparing the molecular signature of laboratory leachate and dark ice DOM in Fig. 4B and Fig. 4C, the more aromatic signature of dark ice compared to leachate DOM is also apparent.

The lack of aromatic compounds in laboratory leachate DOM either suggests that: i) surface debris is not a source of aromatics; ii) oxidation needs to occur to solubilize aromatics from the particulate OM; or iii) aromatics were already preferentially leached from the surface debris before it was collected. Over 80% of the condensed aromatic formulae found in dark ice have an O/C > 0.4, corresponding to  $6.4 \pm 0.8$  %RA of the total dark ice DOM pool, indicating that microbial oxidation may be taking place on the surface (Antony et al., 2017). Furthermore, Antony et al. (2017) showed that microbial reworking of DOM in the snowpack of a coastal Antarctic site resulted in DOM with a higher aromaticity and a higher magnitude and number of nitrogen-, sulphur- and phosphorus-containing formulae. Similar microbial reworking in surface ice may partially explain the higher aromaticity of dark ice DOM compared to laboratory leachate, which was generated to approximate dark ice DOM prior to microbial or photochemical alteration. A likely pathway towards higher aromaticity of dark ice DOM due to microbial degradation is the deglycosylation of the algal pigment purpurogallin carboxylic acid-6-O- $\beta$ -D-glucopyranoside, which is classified as a HUP compound. It has previously been shown that fungal infections of glacier ice algae can cause algae to lose their pigmentation (Fiołka et al., 2021), and that the Greenland Ice Sheet surface fungal species *P. anthracinoglaciei* is able to convert this pigment into its aglycone derivative (Perini et al., 2023), likely utilising the sugar moiety as an energy source. The resulting compound has the polyphenolic molecular formula  $C_{12}H_8O_7$ , which accounts for 5.7 – 8.8 %RA of dark ice DOM. Atmospheric deposition resulting from biomass burning or anthropogenic activity may also deliver aromatic OM to supraglacial surfaces (Stubbins et al., 2012). Aromatic DOM is susceptible to photochemical degradation (Spencer et al., 2009; Stubbins et al., 2010), and recent work has shown that a 28-day irradiance experiment using burnt biomass and anthropogenic OM leachates selectively removed aromatic compounds and produced aliphatic compounds (Holt et al., 2021). The elevated aromaticity of dark ice suggests that photodegradation of aromatic DOM is not taking place on the ice surface, and that aromatic compounds are both degraded and produced at comparable rates or that surface ice DOM residence times are not long enough to achieve complete photochemical degradation of aromatics. Absence of photodegradation is unlikely given the high levels of irradiance the surface ice receives during the ablation season. Instead, we suggest that the elevated aromaticity of dark ice DOM is likely due to microbial reworking of surface ice debris or DOM.

#### 4.2 Near-surface hydrology and the DOM composition of meltwater in the supraglacial drainage system

In the studied micro-catchment, modelled lateral interstitial flow velocity through the weathering crust was in the order of decimetres per day, corresponding with estimates from Stevens et al. (2018), Irvine-Fynn et al. (2021) and Yang et al. (2018). The particle track from Hole D (Fig. 2) confirmed the assumed hydrological connections between sampling locations. We



355 assume that water sampled from the weathering crust auger holes was comprised of a mixture of recent melt, originating near  
the hole, and meltwater already transiting within the weathering crust. We also assume the addition of further meltwater,  
percolating from the unsaturated zone of the weathering crust, along the transit pathway from Hole D to the supraglacial stream  
(Fig. 1E). The presence of a core supraglacial DOM signature, with 2,885 molecular formulae shared across all samples in the  
dataset (Fig. 4A), supports the assumption that the dark ice, weathering crust meltwater and supraglacial stream environments,  
360 and hence DOM pools, are hydrologically connected. The indicative value of nine days for meltwater flow through the  
weathering crust to the supraglacial stream allows the possibility of modification of the DOM pool via microbial reworking  
and/or photochemical degradation (Riedel et al., 2016; Antony et al., 2017, 2018; Holt et al., 2021).

Given the hydrological connectivity of surface ice with weathering crust meltwaters, a higher degree of convergence between  
dark ice and weathering crust DOM may be expected. The higher concentration of DOC in dark ice relative to weathering  
365 crust meltwater may be due to dilution of dark ice DOC in the weathering crust by meltwater produced in the subsurface and  
percolating from the unsaturated zone of the weathering crust, or from surface ice with a low debris loading, which likely  
contains less aromatic DOM. Yet, given the differences in the number and elemental composition of aromatic formulae  
assigned in dark ice and weathering crust meltwater, retention of DOM on the ice surface, potentially via association with  
extracellular polymeric substances (EPS) as suggested by Holland et al. (2019), is a more likely explanation. EPS has been  
370 shown to play a role in the formation of granules in supraglacial cyanobacteria communities (Yallop et al., 2012; Stibal et al.,  
2012; Langford et al., 2010) and Perini et al. (2023) observed glacier ice algae embedded in EPS substances during co-  
cultivation with the surface ice fungi *Articulospora* sp. To date, the chemical composition and role of EPS in microbial  
communities on the ice surface (c.f. cryoconite granules) remains unknown. We recommend further characterisation of organic  
matter delivered by atmospheric deposition and of EPS associated with surface ice communities, combined with controlled  
375 incubation experiments, to assess supraglacial DOM sources, retention, photodegradation, and microbial reworking on the ice  
surface.

In addition to the differences between dark ice and weathering crust DOM composition, the differences in heteroatomic  
composition, mean RA weighted mass, and the %RA of low O/C aliphatic and condensed aromatic compounds between  
weathering crust meltwater and supraglacial stream DOM (Table 1, Fig. 3) also point to potential microbial and/or  
380 photochemical alteration of DOM in the near-surface. These compositional differences are unlikely to be the result of upstream  
contributions to supraglacial stream DOM, as only 12 formulae were uniquely assigned to all supraglacial stream samples  
(accounting for 0.02 – 0.05 %RA).. The high prevalence of aliphatic and peptide-like compounds, and the increase in N- and  
S- containing compounds relative to dark ice DOM, suggest that microbial reworking of DOM takes place within the  
weathering crust. The small but statistically significant decrease in %RA of aromatic DOM in supraglacial stream samples  
385 compared to weathering crust meltwater (Table 1, Fig. 4D-F) indicates that further photodegradation of aromatic DOM might  
also take place during transport through the supraglacial drainage system.



## 5 Conclusions

The ice surface and the weathering crust photic zone present important sites for transformations of supraglacial DOM, altering the composition of DOM in supraglacial streams that drain to the glacier bed and deliver nutrients to subglacial and downstream ecosystems. To our knowledge, this study presents the first characterization of DOM associated with microbial communities and weathering crust meltwater in the Greenland Ice Sheet bare ice ablation zone. The distinct composition of dark ice DOM relative to weathering crust and supraglacial stream DOM highlights the importance of future research into the role of atmospheric deposition, surface ice DOM retention by EPS, and supraglacial water residence times with regards to their effects on the resulting DOM composition in supraglacial runoff. This is particularly important as rainfall over the Greenland Ice Sheet is becoming more prevalent (Niwano et al., 2021), consequently increasing the frequency of weathering crust degradation events (Müller and Keeler, 1969). Temporary removal of the weathering crust, which presents a site for microbial and/or photochemical alteration of DOM, could potentially result in the rapid export of more aromatic, and likely less biolabile, DOM from ice surfaces during or following rain events. Our findings have implications for the understanding of supraglacial biogeochemical cycling, emphasizing the importance of including the weathering crust photic zone when assessing supraglacial inputs to subglacial and downstream ecosystems.

## Data availability

All FT-ICR MS data used in this study can be found in the Open Science Framework Repository via DOI 10.17605/OSF.IO/JRBTH.

## Author contributions

Study design, conceptualisation, and sample collection was done by ELD and ITS. Field processing was done by PER. DOC, TC and TN analysis was done by RA. Lab set-up for DOM extractions was done by ELD and PER. Sample preparation was done by ELD. FT-ICR MS data acquisition was done by AMM. Molecular formula assignment and classification was done by AMK. Hydrological data processing was done by ITS. Data analysis and manuscript preparation was done by ELD. All authors contributed to the final manuscript with discussion and revisions.

**Competing interests:** The authors declare that they have no conflict of interest.

## Acknowledgements

This study was financially supported by the European Research Council (ERC) Synergy Grant DEEP PURPLE under the European Union's Horizon 2020 research and innovation programme (grant agreement No 856416), the Aarhus University Research Foundation through a Starting Grant for AMA (AUFF-2018), the Aarhus University Interdisciplinary Centre for



415 Climate Change (iClimate), and the network programme of the Danish Agency for Science and Higher Education (9096-  
00101B) and the Helmholtz Recruiting Initiative grant (award # I-044-16-01 to LGB). RA acknowledges funding from the  
Alexander von Humboldt Foundation, and RGMS would like to acknowledge NSF DEB 1145932. A portion of this work was  
performed in the Ion Cyclotron Resonance User Facility at the National High Magnetic Field Laboratory, which is supported  
by the National Science Foundation Division of Chemistry and Division of Materials Research through DMR-2128556 and  
420 the State of Florida. All 21 T FT-ICR MS files are publicly available via the Open Science Framework through DOI  
10.17605/OSF.IO/JRBTH. Predator analysis and PetroOrg© software is publicly available for ICR facility users at  
<https://nationalmaglab.org/user-facilities/icr/icr-software>. Finally, the authors would like to thank the entire DEEP PURPLE  
team, especially those involved in the 2021 field campaign.

## References

- 425 Antony, R., Willoughby, A. S., Grannas, A. M., Catanzano, V., Sleighter, R. L., Thamban, M., Hatcher, P. G., and Nair, S.:  
Molecular Insights on Dissolved Organic Matter Transformation by Supraglacial Microbial Communities, *Environ. Sci.*  
*Technol.*, 51, 4328–4337, [https://doi.org/10.1021/ACS.EST.6B05780/SUPPL\\_FILE/ES6B05780\\_SI\\_002.XLS](https://doi.org/10.1021/ACS.EST.6B05780/SUPPL_FILE/ES6B05780_SI_002.XLS), 2017.
- Antony, R., Willoughby, A. S., Grannas, A. M., Catanzano, V., Sleighter, R. L., Thamban, M., and Hatcher, P. G.: Photo-  
biochemical transformation of dissolved organic matter on the surface of the coastal East Antarctic ice sheet,  
430 *Biogeochemistry*, 141, 229–247, <https://doi.org/10.1007/S10533-018-0516-0/FIGURES/3>, 2018.
- Bardgett, R. D., Richter, A., Bol, R., Garnett, M. H., Bäuml, R., Xu, X., Lopez-Capel, E., Manning, D. A. C., Hobbs, P. J.,  
Hartley, I. R., and Wanek, W.: Heterotrophic microbial communities use ancient carbon following glacial retreat, *Biol. Lett.*,  
3, 487–490, <https://doi.org/10.1098/RSBL.2007.0242>, 2007.
- Barker, J. D., Sharp, M. J., and Turner, R. J.: Using synchronous fluorescence spectroscopy and principal components  
435 analysis to monitor dissolved organic matter dynamics in a glacier system, *Hydrol. Process.*, 23, 1487–1500,  
<https://doi.org/10.1002/HYP.7274>, 2009.
- Bhatia, M. P., Das, S. B., Longnecker, K., Charette, M. A., and Kujawinski, E. B.: Molecular characterization of dissolved  
organic matter associated with the Greenland ice sheet, *Geochim. Cosmochim. Acta*, 74, 3768–3784,  
<https://doi.org/10.1016/j.gca.2010.03.035>, 2010.
- 440 Bhatia, M. P., Das, S. B., Xu, L., Charette, M. A., Wadham, J. L., and Kujawinski, E. B.: Organic carbon export from the  
Greenland ice sheet, *Geochim. Cosmochim. Acta*, 109, 329–344, <https://doi.org/10.1016/j.gca.2013.02.006>, 2013.
- Blakney, G. T., Hendrickson, C. L., and Marshall, A. G.: Predator data station: A fast data acquisition system for advanced  
FT-ICR MS experiments, *Int. J. Mass Spectrom.*, 306, 246–252, <https://doi.org/10.1016/j.ijms.2011.03.009>, 2011.
- Chen, M. and Jaffé, R.: Photo- and bio-reactivity patterns of dissolved organic matter from biomass and soil leachates and  
445 surface waters in a subtropical wetland, *Water Res.*, 61, 181–190, <https://doi.org/10.1016/J.WATRES.2014.03.075>, 2014.
- Cook, J. M., Tedstone, A. J., Williamson, C., McCutcheon, J., Hodson, A. J., Dayal, A., Skiles, M., Hofer, S., Bryant, R.,  
McAree, O., McGonigle, A., Ryan, J., Anesio, A. M., Irvine-Fynn, T. D. L., Hubbard, A., Hanna, E., Flanner, M., Mayanna,



- S., Benning, L. G., Van As, D., Yallop, M., McQuaid, J. B., Gribbin, T., and Tranter, M.: Glacier algae accelerate melt rates on the south-western Greenland Ice Sheet, *Cryosphere*, 14, 309–330, <https://doi.org/10.5194/tc-14-309-2020>, 2020.
- 450 Corley, J.: Best practices in establishing detection and quantification limits for pesticide residues in foods, in: *Handbook of residue analytical methods for agrochemicals*, Wiley, 59–74, 2003.
- D’Andrilli, J., Cooper, W. T., Foreman, C. M., and Marshall, A. G.: An ultrahigh-resolution mass spectrometry index to estimate natural organic matter lability, *Rapid Commun. Mass Spectrom.*, 29, 2385–2401, <https://doi.org/10.1002/RCM.7400>, 2015.
- 455 Dittmar, T., Koch, B., Hertkorn, N., and Kattner, G.: A simple and efficient method for the solid-phase extraction of dissolved organic matter (SPE-DOM) from seawater, *Limnol. Oceanogr. Methods*, 6, 230–235, <https://doi.org/10.4319/lom.2008.6.230>, 2008.
- Fausto, R. S., Van As, D., Mankoff, K. D., Vandecrux, B., Citterio, M., Ahlstrøm, A. P., Andersen, S. B., Colgan, W., Karlsson, N. B., Kjeldsen, K. K., Korsgaard, N. J., Larsen, S. H., Nielsen, S., Pedersen, A., Shields, C. L., Solgaard, A. M., and Box, J. E.: Programme for Monitoring of the Greenland Ice Sheet (PROMICE) automatic weather station data, *Earth Syst. Sci. Data*, 13, 3819–3845, <https://doi.org/10.5194/ESSD-13-3819-2021>, 2021.
- 460 Fellman, J. B., Petrone, K. C., and Grierson, P. F.: Leaf litter age, chemical quality, and photodegradation control the fate of leachate dissolved organic matter in a dryland river, *J. Arid Environ.*, 89, 30–37, <https://doi.org/10.1016/J.JARIDENV.2012.10.011>, 2013.
- 465 Fellman, J. B., Hood, E., Raymond, P. A., Stubbins, A., and Spencer, R. G. M.: Spatial Variation in the Origin of Dissolved Organic Carbon in Snow on the Juneau Icefield, Southeast Alaska, *Environ. Sci. Technol.*, 49, 11492–11499, [https://doi.org/10.1021/ACS.EST.5B02685/SUPPL\\_FILE/ES5B02685\\_SI\\_001.PDF](https://doi.org/10.1021/ACS.EST.5B02685/SUPPL_FILE/ES5B02685_SI_001.PDF), 2015.
- Fiołka, M. J., Takeuchi, N., Sofińska-Chmiel, W., Wójcik-Mieszawska, S., Irvine-Fynn, T., and Edwards, A.: Morphological and spectroscopic analysis of snow and glacier algae and their parasitic fungi on different glaciers of Svalbard, *Sci. Rep.* 2021 111, 11, 1–18, <https://doi.org/10.1038/s41598-021-01211-8>, 2021.
- 470 Hansen, A. M., Kraus, T. E. C., Pellerin, B. A., Fleck, J. A., Downing, B. D., and Bergamaschi, B. A.: Optical properties of dissolved organic matter (DOM): Effects of biological and photolytic degradation, *Limnol. Oceanogr.*, 61, 1015–1032, <https://doi.org/10.1002/LNO.10270>, 2016.
- Hemingway, J. D., Spencer, R. G. M., Podgorski, D. C., Zito, P., Sen, I. S., and Galy, V. V.: Glacier meltwater and monsoon precipitation drive Upper Ganges Basin dissolved organic matter composition, *Geochim. Cosmochim. Acta*, 244, 216–228, <https://doi.org/10.1016/j.gca.2018.10.012>, 2019.
- 475 Hendrickson, C. L., Quinn, J. P., Kaiser, N. K., Smith, D. F., Blakney, G. T., Chen, T., Marshall, A. G., Weisbrod, C. R., and Beu, S. C.: 21 Tesla Fourier Transform Ion Cyclotron Resonance Mass Spectrometer: A National Resource for Ultrahigh Resolution Mass Analysis, *J. Am. Soc. Mass Spectrom.*, 26, 1626–1632, <https://doi.org/10.1007/s13361-015-1182-2>, 2015.
- 480 Holland, A. T., Williamson, C. J., Sgouridis, F., Tedstone, A. J., McCutcheon, J., Cook, J. M., Poniecka, E., Yallop, M. L., Tranter, M., and Anesio, A. M.: Dissolved organic nutrients dominate melting surface ice of the Dark Zone (Greenland Ice Sheet), *Biogeosciences*, 16, 3283–3296, <https://doi.org/10.5194/bg-16-3283-2019>, 2019.
- Holt, A. D., Kellerman, A. M., Li, W., Stubbins, A., Wagner, S., McKenna, A., Fellman, J., Hood, E., and Spencer, R. G. M.: Assessing the Role of Photochemistry in Driving the Composition of Dissolved Organic Matter in Glacier Runoff, *J. Geophys. Res. Biogeosciences*, 126, e2021JG006516, <https://doi.org/10.1029/2021JG006516>, 2021.
- 485





- Holt, A. D., Kellerman, A. M., Battin, T. I., McKenna, A. M., Hood, E., Andino, P., Crespo-Pérez, V., Peter, H., Schön, M., De Staercke, V., Styllas, M., Tolosano, M., and Spencer, R. G. M.: A Tropical Cocktail of Organic Matter Sources: Variability in Supraglacial and Glacier Outflow Dissolved Organic Matter Composition and Age Across the Ecuadorian Andes, *J. Geophys. Res. Biogeosciences*, 128, 1–18, <https://doi.org/10.1029/2022JG007188>, 2023.
- 490 Hood, E., Fellman, J., Spencer, R. G. M., Hernes, P. J., Edwards, R., Damore, D., and Scott, D.: Glaciers as a source of ancient and labile organic matter to the marine environment, *Nature*, 462, 1044–1047, <https://doi.org/10.1038/nature08580>, 2009.
- Irvine-Fynn, T. D. L., Edwards, A., Stevens, I. T., Mitchell, A. C., Bunting, P., Box, J. E., Cameron, K. A., Cook, J. M., Naegeli, K., Rassner, S. M. E., Ryan, J. C., Stibal, M., Williamson, C. J., and Hubbard, A.: Storage and export of microbial biomass across the western Greenland Ice Sheet, *Nat. Commun.* 2021 121, 12, 1–11, <https://doi.org/10.1038/s41467-021-24040-9>, 2021.
- 495 Kellerman, A. M., Guillemette, F., Podgorski, D. C., Aiken, G. R., Butler, K. D., and Spencer, R. G. M.: Unifying Concepts Linking Dissolved Organic Matter Composition to Persistence in Aquatic Ecosystems, *Environ. Sci. Technol.*, 52, 2538–2548, [https://doi.org/10.1021/ACS.EST.7B05513/SUPPL\\_FILE/ES7B05513\\_SI\\_001.PDF](https://doi.org/10.1021/ACS.EST.7B05513/SUPPL_FILE/ES7B05513_SI_001.PDF), 2018.
- 500 Kellerman, A. M., Vonk, J., McColaugh, S., Podgorski, D. C., van Winden, E., Hawkings, J. R., Johnston, S. E., Humayun, M., and Spencer, R. G. M.: Molecular Signatures of Glacial Dissolved Organic Matter From Svalbard and Greenland, *Glob. Biogeochem. Cycles*, 35, e2020GB006709, <https://doi.org/10.1029/2020GB006709>, 2021.
- Koch, B. P. and Dittmar, T.: From mass to structure: an aromaticity index for high-resolution mass data of natural organic matter, *Rapid Commun. Mass Spectrom.*, 20, 926–932, <https://doi.org/10.1002/RCM.2386>, 2006.
- 505 Koch, B. P. and Dittmar, T.: From mass to structure: an aromaticity index for high-resolution mass data of natural organic matter, *Rapid Commun. Mass Spectrom.*, 30, 250–250, <https://doi.org/10.1002/RCM.7433>, 2016.
- Langford, H., Hodson, A., Banwart, S., and Bøggild, C.: The microstructure and biogeochemistry of Arctic cryoconite granules, *Ann. Glaciol.*, 51, 87–94, <https://doi.org/10.3189/172756411795932083>, 2010.
- Lawson, E. C., Bhatia, M. P., Wadham, J. L., and Kujawinski, E. B.: Continuous summer export of nitrogen-rich organic matter from the greenland ice sheet inferred by ultrahigh resolution mass spectrometry, *Environ. Sci. Technol.*, 48, 14248–14257, [https://doi.org/10.1021/ES501732H/SUPPL\\_FILE/ES501732H\\_SI\\_001.PDF](https://doi.org/10.1021/ES501732H/SUPPL_FILE/ES501732H_SI_001.PDF), 2014.
- 510 Li, C., Chen, P., Kang, S., Yan, F., Tripathee, L., Wu, G., Qu, B., Sillanpää, M., Yang, D., Dittmar, T., Stubbins, A., and Raymond, P. A.: Fossil Fuel Combustion Emission From South Asia Influences Precipitation Dissolved Organic Carbon Reaching the Remote Tibetan Plateau: Isotopic and Molecular Evidence, *J. Geophys. Res. Atmospheres*, 123, 6248–6258, <https://doi.org/10.1029/2017JD028181>, 2018.
- 515 Lutz, S., Anesio, A. M., Edwards, A., and Benning, L. G.: Linking microbial diversity and functionality of arctic glacial surface habitats, *Environ. Microbiol.*, 19, 551–565, <https://doi.org/10.1111/1462-2920.13494>, 2017.
- Lutz, S., McCutcheon, J., McQuaid, J. B., and Benning, L. G.: The diversity of ice algal communities on the Greenland Ice Sheet as revealed by oligotyping, *Microb. Genomics*, 4, e000159, <https://doi.org/10.1099/MGEN.0.000159/CITE/REFWORKS>, 2018.
- 520 Masiello, C. A.: New directions in black carbon organic geochemistry, *Mar. Chem.*, 92, 201–213, <https://doi.org/10.1016/J.MARCHEM.2004.06.043>, 2004.



- 525 McCutcheon, J., Lutz, S., Williamson, C., Cook, J. M., Tedstone, A. J., Vanderstraeten, A., Wilson, S. A., Stockdale, A., Bonneville, S., Anesio, A. M., Yallop, M. L., McQuaid, J. B., Tranter, M., and Benning, L. G.: Mineral phosphorus drives glacier algal blooms on the Greenland Ice Sheet, *Nat. Commun.* 2021 121, 12, 1–11, <https://doi.org/10.1038/s41467-020-20627-w>, 2021.
- Müller, F. and Keeler, C. M.: Errors in Short-Term Ablation Measurements on Melting Ice Surfaces, *J. Glaciol.*, 8, 91–105, <https://doi.org/10.3189/S0022143000020785>, 1969.
- 530 Musilova, M., Tranter, M., Wadham, J., Telling, J., Tedstone, A., and Anesio, A. M.: Microbially driven export of labile organic carbon from the Greenland ice sheet, *Nat. Geosci.*, 10, 360–365, <https://doi.org/10.1038/ngeo2920>, 2017.
- Niwano, M., Box, J. E., Wehrlé, A., Vandecrux, B., Colgan, W. T., and Cappelen, J.: Rainfall on the Greenland Ice Sheet: Present-Day Climatology From a High-Resolution Non-Hydrostatic Polar Regional Climate Model, *Geophys. Res. Lett.*, 48, e2021GL092942, <https://doi.org/10.1029/2021GL092942>, 2021.
- 535 Oksanen, A. J., Blanchet, F. G., Friendly, M., Kindt, R., Legendre, P., Mcglinn, D., Minchin, P. R., Hara, R. B. O., Simpson, G. L., Solymos, P., Stevens, M. H. H., and Szocs, E.: *Vegan: Community ecology package*, 2011.
- Osterholz, H., Kirchman, D. L., Niggemann, J., and Dittmar, T.: Environmental drivers of dissolved organic matter molecular composition in the Delaware estuary, *Front. Earth Sci.*, 4, 95, <https://doi.org/10.3389/FEART.2016.00095/BIBTEX>, 2016.
- 540 Perini, L., Gostinčar, C., Likar, M., Frisvad, J. C., Kostanjšek, R., Nicholes, M., Williamson, C., Anesio, A. M., Zalar, P., and Gunde-Cimerman, N.: Interactions of Fungi and Algae from the Greenland Ice Sheet, *Microb. Ecol.*, 86, 282–296, <https://doi.org/10.1007/s00248-022-02033-5>, 2023.
- Price, P. B., Rohde, R. A., and Bay, R. C.: Fluxes of microbes, organic aerosols, dust, sea-salt Na ions, non-sea-salt Ca ions, and methanesulfonate onto Greenland and Antarctic ice, *Biogeosciences*, 6, 479–486, <https://doi.org/10.5194/BG-6-479-2009>, 2009.
- 545 Procházková, L., Řezanka, T., Nedbalová, L., and Remias, D.: Unicellular versus Filamentous: The Glacial Alga *Ancylonema alaskana* comb. et stat. nov. and Its Ecophysiological Relatedness to *Ancylonema nordenskiöldii* (Zygnematophyceae, Streptophyta), *Microorg.* 2021 Vol 9 Page 1103, 9, 1103, <https://doi.org/10.3390/MICROORGANISMS9051103>, 2021.
- 550 Remias, D., Schwaiger, S., Aigner, S., Leya, T., Stuppner, H., and Lütz, C.: Characterization of an UV- and VIS-absorbing, purpurogallin-derived secondary pigment new to algae and highly abundant in *Mesotaenium berggrenii* (Zygnematophyceae, Chlorophyta), an extremophyte living on glaciers, *FEMS Microbiol. Ecol.*, 79, 638–648, <https://doi.org/10.1111/j.1574-6941.2011.01245.x>, 2012.
- Riedel, T., Biester, H., and Dittmar, T.: Molecular fractionation of dissolved organic matter with metal salts, *Environ. Sci. Technol.*, 46, 4419–4426, [https://doi.org/10.1021/ES203901U/SUPPL\\_FILE/ES203901U\\_SI\\_001.PDF](https://doi.org/10.1021/ES203901U/SUPPL_FILE/ES203901U_SI_001.PDF), 2012.
- 555 Riedel, T., Zark, M., Vähätalo, A. V., Niggemann, J., Spencer, R. G. M., Hernes, P. J., and Dittmar, T.: Molecular signatures of biogeochemical transformations in dissolved organic matter from ten world rivers, *Front. Earth Sci.*, 4, 85, <https://doi.org/10.3389/FEART.2016.00085/BIBTEX>, 2016.
- Ryan, J.: Derivation of high spatial resolution albedo from UAV digital imagery: application over the Greenland Ice Sheet, *Front Earth Sci*, 5, 00040, <https://doi.org/10.3389/feart.2017.00040>, 2017.



- 560 Ryan, J. C.: Dark zone of the Greenland Ice Sheet controlled by distributed biologically-active impurities, *Nat Commun*, 9, <https://doi.org/10.1038/s41467-018-03353-2>, 2018.
- Ryan, J. C., Smith, L. C., Van As, D., Cooley, S. W., Cooper, M. G., Pitcher, L. H., and Hubbard, A.: Greenland Ice Sheet surface melt amplified by snowline migration and bare ice exposure, *Sci. Adv.*, 5, [https://doi.org/10.1126/SCIADV.AAV3738/SUPPL\\_FILE/AAV3738\\_SM.PDF](https://doi.org/10.1126/SCIADV.AAV3738/SUPPL_FILE/AAV3738_SM.PDF), 2019.
- 565 Savory, J. J., Kaiser, N. K., Mckenna, A. M., Xian, F., Blakney, G. T., Rodgers, R. P., Hendrickson, C. L., and Marshall, A. G.: Measurement Accuracy with a “Walking” Calibration Equation, *Anal. Chem.*, 83, 1732–1736, 2011.
- Singer, G. A., Fasching, C., Wilhelm, L., Niggemann, J., Steier, P., Dittmar, T., and Battin, T. J.: Biogeochemically diverse organic matter in Alpine glaciers and its downstream fate, *Nat. Geosci.*, 5, 710–714, <https://doi.org/10.1038/ngeo1581>, 2012.
- 570 Smith, D. F., Podgorski, D. C., Rodgers, R. P., Blakney, G. T., and Hendrickson, C. L.: 21 Tesla FT-ICR Mass Spectrometer for Ultrahigh-Resolution Analysis of Complex Organic Mixtures, *Anal. Chem.*, 90, 2041–2047, [https://doi.org/10.1021/ACS.ANALCHEM.7B04159/ASSET/IMAGES/MEDIUM/AC-2017-04159Q\\_0010.GIF](https://doi.org/10.1021/ACS.ANALCHEM.7B04159/ASSET/IMAGES/MEDIUM/AC-2017-04159Q_0010.GIF), 2018.
- Spencer, R. G. M., Stubbins, A., Hernes, P. J., Baker, A., Mopper, K., Aufdenkampe, A. K., Dyda, R. Y., Mwamba, V. L., Mangangu, A. M., Wabakanghanzi, J. N., and Six, J.: Photochemical degradation of dissolved organic matter and dissolved lignin phenols from the Congo River, *J. Geophys. Res. Biogeosciences*, 114, 3010, <https://doi.org/10.1029/2009JG000968>, 2009.
- 575 Spencer, R. G. M., Vermilyea, A., Fellman, J., Raymond, P., Stubbins, A., Scott, D., and Hood, E.: Seasonal variability of organic matter composition in an Alaskan glacier outflow: Insights into glacier carbon sources, *Environ. Res. Lett.*, 9, <https://doi.org/10.1088/1748-9326/9/5/055005>, 2014a.
- 580 Spencer, R. G. M., Guo, W., Raymond, P. A., Dittmar, T., Hood, E., Fellman, J., and Stubbins, A.: Source and biolability of ancient dissolved organic matter in glacier and lake ecosystems on the Tibetan Plateau, *Geochim. Cosmochim. Acta*, 142, 64–74, <https://doi.org/10.1016/J.GCA.2014.08.006>, 2014b.
- Spencer, R. G. M., Mann, P. J., Dittmar, T., Eglinton, T. I., McIntyre, C., Holmes, R. M., Zimov, N., and Stubbins, A.: Detecting the signature of permafrost thaw in Arctic rivers, *Geophys. Res. Lett.*, 42, 2830–2835, <https://doi.org/10.1002/2015GL063498>, 2015.
- 585 Steger, C. R., Reijmer, C. H., and Van Den Broeke, M. R.: The modelled liquid water balance of the Greenland Ice Sheet, *Cryosphere*, 11, 2507–2526, <https://doi.org/10.5194/tc-11-2507-2017>, 2017.
- Stevens, I. T., Irvine-Fynn, T. D. L., Porter, P. R., Cook, J. M., Edwards, A., Smart, M., Moorman, B. J., Hodson, A. J., and Mitchell, A. C.: Near-surface hydraulic conductivity of northern hemisphere glaciers, *Hydrol. Process.*, 32, 850–865, <https://doi.org/10.1002/HYP.11439>, 2018.
- 590 Stibal, M., Šabacká, M., and Žárský, J.: Biological processes on glacier and ice sheet surfaces, *Nat. Geosci.*, 5, 771–774, <https://doi.org/10.1038/ngeo1611>, 2012.
- 595 Stibal, M., Box, J. E., Cameron, K. A., Langen, P. L., Yallop, M. L., Mottram, R. H., Khan, A. L., Molotch, N. P., Christmas, N. A. M., Cali Quaglia, F., Remias, D., Smeets, C. J. P. P., van den Broeke, M. R., Ryan, J. C., Hubbard, A., Tranter, M., van As, D., and Ahlström, A. P.: Algae Drive Enhanced Darkening of Bare Ice on the Greenland Ice Sheet, *Geophys. Res. Lett.*, 44, 11,463–11,471, <https://doi.org/10.1002/2017GL075958>, 2017.



- Stubbins, A., Spencer, R. G. M., Chen, H., Hatcher, P. G., Mopper, K., Hernes, P. J., Mwamba, V. L., Mangangu, A. M., Wabakanghanzi, J. N., and Six, J.: Illuminated darkness: Molecular signatures of Congo River dissolved organic matter and its photochemical alteration as revealed by ultrahigh precision mass spectrometry, *Limnol. Oceanogr.*, 55, 1467–1477, <https://doi.org/10.4319/LO.2010.55.4.1467>, 2010.
- 600 Stubbins, A., Hood, E., Raymond, P. A., Aiken, G. R., Sleighter, R. L., Hernes, P. J., Butman, D., Hatcher, P. G., Striegl, R. G., Schuster, P., Abdulla, H. A. N., Vermilyea, A. W., Scott, D. T., and Spencer, R. G. M.: Anthropogenic aerosols as a source of ancient dissolved organic matter in glaciers, *Nat. Geosci.* 2012 53, 5, 198–201, <https://doi.org/10.1038/ngeo1403>, 2012.
- 605 Uetake, J., Naganuma, T., Hebsgaard, M. B., Kanda, H., and Kohshima, S.: Communities of algae and cyanobacteria on glaciers in west Greenland, *Polar Sci.*, 4, 71–80, <https://doi.org/10.1016/j.polar.2010.03.002>, 2010.
- Xian, F., Hendrickson, C. L., Blakney, G. T., Beu, S. C., and Marshall, A. G.: Automated broadband phase correction of fourier transform ion cyclotron resonance mass spectra, *Anal. Chem.*, 82, 8807–8812, <https://doi.org/10.1021/ac101091w>, 2010.
- 610 Yallop, M. L., Anesio, A. M., Perkins, R. G., Cook, J., Telling, J., Fagan, D., MacFarlane, J., Stibal, M., Barker, G., Bellas, C., Hodson, A., Tranter, M., Wadham, J., and Roberts, N. W.: Photophysiology and albedo-changing potential of the ice algal community on the surface of the Greenland ice sheet, *ISME J.* 2012 612, 6, 2302–2313, <https://doi.org/10.1038/ismej.2012.107>, 2012.
- Yang, K. and Smith, L. C.: Internally drained catchments dominate supraglacial hydrology of the southwest Greenland Ice Sheet, *J. Geophys. Res. Earth Surf.*, 121, 1891–1910, <https://doi.org/10.1002/2016JF003927>, 2016.
- 615 Yang, K., Smith, L. C., Karlstrom, L., Cooper, M. G., Tedesco, M., van As, D., Cheng, X., Chen, Z., and Li, M.: A new surface meltwater routing model for use on the Greenland Ice Sheet surface, *The Cryosphere*, 12, 3791–3811, <https://doi.org/10.5194/tc-12-3791-2018>, 2018.