

45 **Abstract**

46 Constraining dust models to understand and quantify the effect of dust upon climate and ecosystems
47 requires comprehensive analyses of the physiochemical properties of dust-emitting sediments in arid
48 regions. Following previous studies in the Moroccan Sahara and Iceland, here we analyse a diverse set
49 of crusts and aeolian ripples (n=55) from various potential dust-emitting basins within the Mojave
50 Desert, California, USA. We focus on the characterization of their particle size distribution (PSD),
51 mineralogy, aggregation/cohesion state and Fe mode of occurrence. Our results show differences in
52 fully and minimally dispersed PSDs, with crusts average median diameters (92 and 37 μm , respectively)
53 compared to aeolian ripples (226 and 213 μm , respectively). Mineralogical analyses unveiled strong
54 variations between crusts and ripples, with crusts being enriched in phyllosilicates (24 vs 7.8 %),
55 carbonates (6.6 vs 1.1 %), Na-salts (7.3 vs 1.1 %) and zeolites (1.2 and 0.12 %), and ripples being
56 enriched in feldspars (48 vs 37 %), quartz (32 vs 16 %), and gypsum (4.7 vs 3.1 %). The size fractions
57 from crust sediments display a homogeneous mineralogy, while those of aeolian ripples a more
58 heterogeneous one, mostly due to their different particle aggregation. Bulk Fe content analyses
59 indicate higher concentrations in crusts (3.0 ± 1.3 wt %) compared to ripples (1.9 ± 1.1 wt %), with similar
60 proportions in their Fe mode of occurrence: nano Fe-oxides/readily exchangeable Fe represent ~ 1.6
61 %, hematite/goethite ~ 15 %, magnetite/maghemite ~ 2.0 % and structural Fe in silicates ~ 80 % of the
62 total Fe. We identified segregation patterns in PSD and mineralogy differences in Na-salts content
63 within the Mojave basins, that can be explained by sediment transportation dynamics and precipitates
64 due to groundwater table fluctuations described in previous studies in the region. Mojave Desert
65 crusts show similarities with previously sampled crusts in the Moroccan Sahara for PSD and readily
66 exchangeable Fe, yet exhibit substantial differences in mineralogical composition, which should
67 significantly influence the emitted dust particle characteristics.

68

69 **Keywords:** Arid regions, dust sources, desert dust, dust-emitting sediment formation model, dust
70 mineralogy.

71 **1. Introduction**

72 Desert dust produced by wind erosion of arid and semi-arid surfaces has significant effects on climate
73 and ecosystems (Weaver et al., 2002; Goudie & Middleton, 2006; Sullivan et al., 2007; Crumeyrolle et
74 al., 2008; De Longeville et al., 2010; Karanasiou et al., 2012; Pérez García-Pando et al., 2014; among
75 others). Dust affects the energy and water cycles through its absorption and scattering of both
76 shortwave (SW) and longwave (LW) radiation (Perez et al., 2006; Miller et al., 2014), and exerts
77 influence on cloud formation, precipitation patterns, and the associated indirect radiative forcing by
78 serving as nuclei for liquid and ice clouds (e.g. Harrison et al., 2019). Dust also undergoes
79 heterogeneous chemical reactions in the atmosphere that enhance their hygroscopicity and modify
80 their optical properties (Bauer et al., 2005), and when deposited into ocean waters, its bioavailable
81 iron content acts as a catalyst for photosynthesis by ocean phytoplankton, thereby increasing carbon
82 dioxide uptake and influencing the global carbon cycle (Jickells et al., 2005). Dust primarily originates
83 from arid inland basins, which include various sedimentary environments such as aeolian deposits,
84 endorheic depressions, and fluvial- and alluvial-dominated systems (Bullard et al., 2011). Wind
85 typically mobilizes loose sand from adjacent ripples or dunes, which then erodes more consolidated
86 surfaces, typically paved sediments and crusts, to release dust (Stout and Lee, 2003; Shao et al., 2011).
87 Atmospheric dust emission models have improved by identifying preferential dust sources using
88 criteria like topography and hydrology (Ginoux et al., 2001). However, these models still struggle with
89 capturing small-scale variability partly due to the lack of relevant soil measurements in arid regions,
90 despite advancements in understanding the geomorphological and sedimentological factors
91 influencing dust emissions (Bullard et al., 2011). For instance, the particle size distribution (PSD) and
92 cohesion of the sediments affect saltation bombardment and aggregate disintegration processes
93 involved in dust emission (Shao et al., 1993).

94 Understanding the mineral composition of dust is also crucial for assessing its climate impact. Dust
95 contains various minerals such as quartz, clay minerals, feldspars, carbonates, salts, and iron oxides.
96 The climate effects of dust are influenced by these minerals' relative abundances, sizes, shapes, and
97 mixing states. For example, iron oxides control solar radiation absorption by dust (Formenti et al.,
98 2014; Engelbrecht et al., 2016; Di Biagio et al., 2019; Zubko et al., 2019), nano Fe oxides and easily
99 exchangeable Fe increase the fertilising effect of dust in ocean and terrestrial ecosystems
100 (Hettiarachchi et al., 2019; Baldo et al., 2020, Hettiarachchi et al., 2020), K-feldspar and quartz impact
101 ice nucleation in clouds (Atkinson et al., 2013; Harrison et al., 2019; Chatziparaschos et al., 2023), and
102 calcite influences acid reactions on dust surfaces (Paulot et al., 2016). The mineralogical composition
103 of dust can vary significantly across different regions due to geological and climatic factors (Claquin et
104 al., 1999; Journet et al., 2014). However, most models assume a globally uniform dust composition
105 due to limited global data on parent soil sources. Only a few models account for dust mineralogical
106 composition variations (e.g., Scanza et al., 2015; Perlwitz et al., 2015; Li et al., 2021; Gonçalves Ageitos
107 et al., 2023; Obiso et al., 2024) by using global soil type atlases and extrapolating from a limited
108 number of analyses (Claquin et al., 1999; Journet et al., 2014). These atlases rely on assumptions about
109 soil texture and color, often base their data on soil samples taken from depths deeper than those
110 relevant to wind erosion, and the method used to characterize particle size and associated mineralogy
111 fully breaks down natural soil aggregates.

112 Since 2022, the EMIT mission has been acquiring comprehensive measurements of surface
113 mineralogical composition for use in Earth System models (Green et al., 2020). EMIT employs imaging

114 spectroscopy across the visible to short wavelength infrared (VSWIR) spectral range from the
115 International Space Station to map the occurrence and estimate the abundance of ten key dust source
116 minerals. Additionally, EMIT has the potential to estimate surface soil texture. While identifying
117 dominant surface minerals has traditionally been a strength of spectrometers, quantifying these
118 minerals poses significant challenges. Factors such as mineral grain size and composition can affect
119 spectral absorptions, certain dominant materials like quartz and feldspar exhibit minimal absorption
120 features, and the presence of other materials can further complicate the analysis.

121 Overall, there is a notable lack of comprehensive measurements characterizing relevant properties of
122 surface sediments in dust source regions. This gap hampers our ability to evaluate and constrain
123 mineral abundance derived from reflectance spectroscopy and to improve dust emission modeling.
124 Addressing this issue, the FRontiers in dust minerAloGical coMposition and its Effects upoN climaTe
125 (FRAGMENT) project has, over recent years, conducted a series of coordinated and interdisciplinary
126 field campaigns across remote dust source regions. The project's objectives include enhancing the
127 understanding and quantification of dust-source properties, examining their relationship with emitted
128 dust characteristics, refining spaceborne spectroscopy retrievals of surface minerals (Green et al.,
129 2020, Clark et al., 2024), and improving the representation of dust mineralogy in Earth system models
130 (Perlwitz et al., 2015; Li et al., 2021; Gonçalves et al., 2023; Obiso et al., 2023). The FRAGMENT field
131 campaigns involved regional sediment sampling in several regions and detailed wind erosion and dust
132 emission measurements at selected sites. Studies stemming from these activities include those by
133 González-Romero et al. (2023), González-Flórez et al. (2022), and Panta et al. (2022), and González-
134 Romero et al. (2024). These campaigns have been executed in various geographic locations, such as
135 the Moroccan Sahara (2019), Iceland (2021), the Mojave Desert in the United States (2022), and
136 Jordan (2022). Through these efforts, the FRAGMENT project contributes to filling critical knowledge
137 gaps in dust-source characterization.

138 Following our previous studies in the Moroccan Sahara (González-Romero et al. 2023) and Iceland
139 González-Romero et al. (2024), this study focuses on the characterization of dust-emitting sediments
140 collected from the Mojave Desert in 2022. The Mojave Desert is a closed-basin wedge-shaped region
141 located in the southwestern United States, between California and Nevada. The region is surrounded
142 by mountain ranges and traversed by the Mojave river and other intermittent rivers for over 200 km
143 from the San Bernardino mountains to the east (Dibblee, 1967, Reheis et al., 2012). Despite its limited
144 global importance (dust emission from North America represents only ~3 % of the global dust
145 emission; Kok et al., 2021), the Mojave Desert is an important regional dust source (Ginoux et al.,
146 2012), with most emission occurring in the playa lakes and alluvium deposits near playa lakes (Reheis
147 & Kihl, 1995; Reheis et al. 2009, Urban et al., 2018). Reynolds et al. (2009) observed 71 days with dust
148 plumes during 37 months of camera recording at the Franklin Lake playa. According to remote sensing
149 data (MODIS), aerosol optical depth (AOD) is higher in spring and summer and reaches a minimum in
150 winter (Frank et al., 2007). However, from November to May, eastward flows of the jet-stream affect
151 the Mojave Desert, which, in combination with topography, favour the development of northerly
152 winds that can lead to dust emission (Urban et al., 2009). Up to 65 % of emission in the Mojave Desert
153 is estimated to be due to natural factors, whereas 35 % to anthropogenic activities, including off-road
154 recreation practices, mine operations, and military training and livestock grazing (Frank et al., 2007).
155 The AOD in this region is also affected by dust transported from other regions (Tong et al., 2012) and
156 pollution transported from the Los Angeles Basin (Frank et al., 2007, Urban et al., 2009). In the Mojave

157 Desert, Reynolds et al. (2009) noted an association between wet periods and dust emission, directly
158 related to the generation of new thin crusts and salt crust removal.

159 The Mojave Desert includes several significant playa lakes, such as Rogers and Rosemond, Owens Lake,
160 Death-Valley-Badwater, Panamint Valley, Bristol, Cadiz and Danby, Searles Lake, Soda Lake, and
161 Mesquite Lake, among others (Reheis & Kihl, 1995; Reheis, 1997; Potter and Coppernoll-Houston,
162 2019). Reynolds et al. (2007 and 2009) distinguished between two types of playa lakes: wet playas
163 influenced by groundwater, and dry playas, unaffected by groundwater, though both can experience
164 surface-water runoff. Goudie (2018) further delineated wet playas as having a groundwater table
165 within 5 m of the surface, while dry playas have a groundwater table deeper than 5 m. Additionally,
166 Goudie (2018), Buck et al. (2011), Nield et al. (2015) and Nield et al. (2016) observed that the
167 interaction between salt minerals and the groundwater table on wet playas lead to the formation of
168 fluffy surfaces through salt reworking by water during evapotranspiration.

169 In the Mojave Desert, three different Aridisols are present in the Rand mountains alluvial fan,
170 corresponding to Xerillic soils or Aridisols according to Eghbal & Southard (1993), typical in arid and
171 semi-arid regions, with low organic matter content and low structures. The uppermost layer of those
172 Aridisols, ranging from 0 to 1 cm in depth exhibited a texture of 15-30% gravel, 69-74 % sand and 10-
173 11 % clay. Reheis et al. (1995) described soils (<2 mm) primarily composed of silt (30-70 %) and clay
174 (20-45 %). The mineralogy of those samples was dominated by quartz, feldspars, amphiboles, and clay
175 minerals, including smectite, mica and kaolinite (Eghbal & Southard, 1993). The Cronese Lakes and
176 Soda Lake playas are documented to contain salt precipitates, but mineralogy is not specified.
177 Mesquite Lake playa is noted for its gypsum deposits (Reynolds et al., 2009). At Franklin Lake playa,
178 surfaces are characterized by silt-clay size particles (Goldstein et al., 2017) with mineralogical
179 descriptions provided in Reynolds et al. (2009) indicating fluffy surfaces comprised of halite,
180 thenardite, trona, burkeite, calcite, illite, smectite, and kaolinite. Similar mineralogical results are
181 described at Soda Lake by Reheis et al. (2009), with a higher proportion of Na-salts, quartz, gypsum
182 and carbonates. Furthermore, Goldstein et al. (2017) identified a diverse array of minerals at Franklin
183 Lake playa, including clays, zeolites, plagioclase, K-feldspar, quartz, calcite, dolomite and salt minerals
184 such as trona, halite, burkeite and thenardite.

185 This study characterises the particle size distribution, mineralogy and modes of occurrence of Fe of
186 selected potential dust-emitting sediment surfaces from the Mojave Desert. In addition, the
187 mineralogy of different size fractions is analysed, based on a sieving protocol that minimally disturbs
188 sediments. We further discuss the potential effect of sedimentary transport on the particle size and
189 mineralogy across the sampled basins building upon previous studies in the literature. Finally, our
190 results are broadly compared with current EMIT standard (semi-quantitative) products, and with those
191 obtained using similar protocols in previous FRAGMENT campaigns in other regions (González-Romero
192 et al., 2023; González-Romero et al., 2024).

193 **2. Methodology**

194 **2.1 Study area**

195 The Mojave Desert, located between California and Nevada, has a diverse geological history spanning
196 from the Cambrian and Precambrian eras to the Holocene (Figure 1). This geological complexity

197 encompasses volcanic, plutonic, metamorphic, and sedimentary units (Jennings et al., 1962; Miller et
198 al., 2014). In areas once submerged during the last glacial maximum (LGM), we now find ephemeral
199 playa lakes that have existed for thousands of years since the LGM, offering a glimpse into the region's
200 dynamic past (Miller et al., 2018). These playa lakes together with alluvial fans, floodplains and other
201 features are surrounded by a variety of source rocks, exhibit diverse particle sizes and compositions
202 and can potentially emit dust under favourable wind conditions.

203 The regional distribution of the annual Frequency of Occurrence (FoO) of dust events, with dust optical
204 depth exceeding 0.1, derived from MODIS Deep Blue C6.1 Level 2 data following the methodology of
205 Ginoux et al. (2012), is illustrated in Figure 2. The FoO provides an overall estimate of dust emission
206 frequency above a certain threshold at a resolution of 0.1° by 0.1° over the region. Sediment samples
207 were collected from various locations within the Mojave Desert region, including areas with relatively
208 high FoO (see locations in Figures 1 and 2). Among these locations is Soda Lake and its surroundings,
209 near Baker, CA, which is linked to Silver Lake to the north and is surrounded by igneous, volcanic, and
210 carbonate rocks, as well as dune fields to the south (Figure 1). The area is influenced by aeolian,
211 alluvial, and fluvial processes and experiences annual precipitation of 80-100 mm (Urban et al., 2018).
212 This ephemeral lake contains salts resulting from the evaporation of groundwater sourced from an
213 aquifer nestled in the Zzyzx Mountains (Honke et al., 2019). Dust emissions are a recurrent
214 phenomenon, originating from fine sediments accumulated in the lake's central areas during sporadic
215 floodings, from the white evaporite surfaces in the lake, and from the alluvial deposits to the south of
216 the playa lake (Urban et al., 2018). According to the FoO, the areas with higher dust emissions are the
217 southern part of the lake and the alluvial deposits to the southwest, extending up to Afton Canyon.

218 Samples were also collected from the Cronese lakes, Mesquite Lake, Ivanpah Lake and Coyote Lake
219 (Figure 1), which lie in areas with significant FoO signals (Figure 2) and have been documented as dust
220 sources in Reheis & Kihl (1995) and Reheis et al. (2009). The Cronese lakes are adjacent to the Soda
221 Lake area to the west, sharing a similar geologic context (Figures 1 and 2). Mesquite Lake, located on
222 the border between California and Nevada, is encircled by carbonate and igneous rocks, mirroring the
223 geological setting of the nearby Ivanpah Lake. Notably, Mesquite Lake playa is the only playa affected
224 by a gypsum mine pit, as documented by Reynolds et al. (2009). Further contributing to the diversity
225 of the region's geological makeup is Coyote Lake, flanked by Miocene and Pleistocene sediments.
226 These playa lakes, characterized as endorheic ephemeral lakes, receive groundwater inputs in some
227 cases, enriching the lakes with salts that subsequently precipitate on the surfaces of their central
228 regions (Whitney et al., 2015; Urban et al., 2018).

229 Other areas with relatively high FoO not sampled in our study include the Ashford Junction alluvial
230 deposits and the Fort Irwin area, where the northern valley, including Nelson Lake, may be more prone
231 to dust emission due to significant anthropogenic disturbance. It is important to note that the FoO
232 may tend to highlight areas such as playas and their surroundings, where in some cases the most dust
233 per unit area could be produced (Floyd and Gill, 2011; Baddock et al., 2016). However, some alluvial
234 regions with lower emission rates not surpassing the FoO threshold may produce more dust overall
235 due to their greater areal extent (Reheis and Kihl, 1995; Baddock et al., 2016). Additionally, many other
236 types of dust-producing surfaces active in the Mojave Desert, such as gravel roads, agricultural lands,
237 and recreational off-road tracks, are rarely observed by satellite retrievals (Urban et al., 2018).

238 The new EMIT sensor onboard the International Space Station offers a glimpse of the mineralogical
239 diversity in the Mojave Desert (Green et al., 2020). Figure 3 displays standard Tetracorder RGB color
240 composite semi-quantitative products for EMIT. Tetracorder is a software system that employs a set
241 of algorithms within an expert system decision-making framework to identify and map compounds
242 (Clark, 2024; Clark et al., 2024). Figure 3 shows a true color image, along with standard products for
243 Fe²⁺ and Fe³⁺ bearing mineral electronic absorptions (including hematite and goethite) in the visible to
244 very-near infrared spectral range. It also displays standard products for the EMIT-targeted minerals,
245 excluding hematite and goethite: calcite, chlorite/serpentine, dolomite, gypsum, illite/muscovite,
246 kaolinite-dioctahedral group, montmorillonite group, and vermiculite. These products highlight areas
247 where the presence of each mineral or component is significant, measured in terms of band depth*fit,
248 where the fit represents the least squares correlation coefficient from a feature fit of observed and
249 reference library spectra. These analyses reveal the widespread presence of phyllosilicates such as
250 kaolinite, smectite, montmorillonite, and illite across the area. The northeastern sector, particularly
251 around Mesquite Lake, exhibits notable concentrations of carbonates and gypsum. Additionally,
252 goethite and hematite are detected, with a more pronounced presence of goethite in the northern
253 portion and hematite in the southern part of the region. The detection of mixtures of Fe²⁺ and Fe³⁺
254 within various minerals enriches our understanding of the region's mineralogical diversity.

255 Quantitative surface mineralogy (mineral mass abundances of the 10 EMIT-targeted minerals) and soil
256 texture products are currently being developed by the EMIT team for use in Earth System Models.
257 Their publication and evaluation will be the focus of forthcoming publications. Thus, it is beyond the
258 scope of this study to perform a detailed quantitative comparison between our analyses and
259 comparable EMIT products. However, in the results section, we broadly compare these standard
260 products with the results of our in-situ analyses.

261 **2.2 Sampling**

262 Representative surfaces of dust-emitting sediments were sampled in May 2022, with depths of up to
263 3 cm, using a 5 cm² inox shovel. Samples were stored in a plastic bag, labelled, and documented with
264 photographs, descriptions, and coordinates, and transported to the laboratories for subsequent
265 analyses. The type of samples considered are crusts (semi-cohesive fine sediments accumulated
266 during floodings in depressions) and ripples (aeolian ripples that are built up under favourable winds
267 and supply sand for saltation) (Figure 4). A total of 55 surface sediments and ripples (32 from Soda
268 Lake, 9 from Mesquite Lake, 1 from Ivanpah Lake, 11 from the Cronese Lakes, and 2 from Coyote Lake)
269 were sampled for laboratory analysis. Once in the laboratory, the samples were dried for 24-48 h at
270 40-50 °C, sieved to pass through a 2 mm mesh.

271 Our rationale for selecting crusts and ripples is two-fold. On the one side, dust emission is primarily
272 driven by two mechanisms: saltation bombardment and aggregate disintegration. In saltation
273 bombardment, dust is ejected from soil aggregates (typically crusts and paved sediments rich in clay
274 and silt particles) when impacted by saltating sand particles. In aggregate disintegration, dust is
275 released from saltating soil aggregates (Shao et al., 1993; Alfaro et al., 1997; Shao, 2001). By
276 characterizing the PSD (both dry and wet sieved) and mineralogy of ripples (concentrating sand
277 particles) and crusts (concentrating clay and silt particles), we provide comprehensive and valuable
278 information for developing and refining dust emission models. On the other side, in arid regions,
279 quartz and feldspar typically dominate sediment mass. However, current spaceborne hyperspectral

280 instruments (such as EMIT) cannot directly identify feldspar and quartz because their absorption
281 features lie outside the instrument's spectral range. This poses a significant challenge in quantifying
282 surface mineral abundances from remote spectroscopy. At all FRAGMENT sampling locations
283 (Morocco, Iceland, US-Mojave, and Jordan), we measured reflectance spectra using an ASD Fieldspec
284 3. By characterizing and contrasting ripples (with high quartz and feldspar content and larger particle
285 sizes) and crusts, we aim to provide information to enhance understanding and improve modelling
286 assumptions for estimating surface mineral abundances and soil texture from remote spectroscopy in
287 subsequent studies.

288 We acknowledge that the limited number of samples collected may not fully represent the potential
289 variability among crusts and ripples within the studied locations due to varying conditions (Buck et al.,
290 2011). However, our samples broadly represent the composition and particle size distributions (PSDs)
291 of this type of sediments in these areas, allowing for meaningful comparisons with sediments from
292 other locations.

293 **2.3 Analyses**

294 2.3.1 Particle size distribution

295 Particle size distributions (PSD) of bulk samples (<2 mm) were analysed as described in González-
296 Romero et al. (2023) for the evaluation of the aggregation state. First, we conducted a minimally
297 dispersed PSD (MDPSD) analysis, which minimizes the breaking of the aggregates that are
298 encountered in natural conditions. Second, we conducted a fully dispersed PSD (FDPSD) analysis,
299 which breaks the aggregates. Wet dispersion was done according to Sperazza et al., (2004), using
300 water and sodium hexametaphosphate dispersion for 24 h. Both PSDs (MDPSD and FDPSD) were
301 obtained by a laser diffractometer with the Malvern Mastersizer 2000 Hydro G and Scirocco for the
302 fully and minimally dispersed conditions, respectively. We note that under wet dispersion, at least
303 some salt minerals may dissolve.

304 In addition, we separated 20 selected samples from different sources, including 16 crusts and 4 aeolian
305 ripples, into different size ranges to understand how mineral composition changes with size. We used
306 a series of sieves with mesh sizes of 2 mm, 1 mm, 500 μm , 250 μm , 80 μm , 63 μm , 40 μm , and 20 μm .
307 The sieving process involved hand shaking the full column for 1 minute, followed by ultrasound
308 sonication for 1 minute at the 500 μm , 80 μm , 40 μm , and 20 μm size fractions. This method ensured
309 the effective separation of the size fractions for subsequent mineralogical analysis.

310 2.3.2 Mineralogical composition

311 To quantify the different contents of crystalline minerals and amorphous components, X-Ray
312 Diffraction (XRD) coupled with a Rietveld quantitative method were used (Rietveld, 1969; Cheary and
313 Coelho, 1992; Young, 1995 and Topas, 2018). Adding a known amount of an internal standard material
314 allowed, via the Rietveld method, the quantification of a mixture of minerals and any non-crystalline
315 material in the mixture not included in the Rietveld method (De la Torre et al., 2001; Madsen, 2001,
316 Scarlett and Madsen, 2006; Machiels et al., 2010; Ibañez et al., 2013). For the analysis, a measured
317 amount of dry ground sample was mixed and dry ground again with 10-20 % of fluorite (CaF_2 powder,
318 Merck), used here as an internal standard for quantitative purposes. The XRD patterns of the samples

319 were analysed by a Bruker D8 A25 Advanced Powder X-ray diffractometer operated at 40kV and 40
320 mA with monochromatic Cu K α radiation ($\approx 1.5405 \text{ \AA}$). This device uses a Bragg-Brentano geometry
321 and a sensitive detector LynxEye 1D. Diffractograms were recorded from 4 to 120 $^\circ$ of 2 θ and steps of
322 0.015 $^\circ$ in 1s and maintained rotation (15/min). For the clay identification, samples were analysed using
323 the oriented aggregate method by XRD, decanting clay fractions from samples and smearing the
324 slurries in glass slides. After, three treatments were applied including air drying (AO), glycolation with
325 ethylene glycol (AG) and heating at 550 $^\circ\text{C}$ for 2h (AC) with its three different diffractograms. Finally,
326 the three diffractograms allows to corroborate the presence of Illite, Chlorite, Palygorskite and
327 Montmorillonite through Thorez (1976) and USGS Open File procedures. Data collected were
328 evaluated using the Bruker AXS DIFFRAC.EVA software package (Bruker AXS, Karlsruhe, Germany,
329 2000) and the Rietveld analyses performed with TOPAS 4.2 program (Bruker AXS, 2003-2009). A
330 Chebyshev function of level 5 was used to fit the background and abundances of crystalline phases
331 and amorphous phases were normalised to 100 wt %. Fits were evaluated by visual comparison, the
332 R_{wp} (R-weighted pattern), R_{exp} (R-expected), and Goodness of Fit (GOF).

333 2.3.3 Mode of occurrence of Fe

334 As XRD is not precise enough for Fe-oxide quantification, wet chemistry and sequential extractions of
335 Fe are needed for quantification of the mode of occurrence of Fe (González-Romero et al., 2023;
336 2024). Samples were analysed with a two-step acid digestion for the total Fe (FeT) content following
337 the procedure by Querol et al. (1993, 1997). A reference material (NIST-1633b, coal fly ash) was used
338 for quality control in every batch. The sequential extraction presented in Shi et al. (2009), Baldo et al.
339 (2020) and González-Romero et al. (2024) were used to quantify readily exchangeable Fe ions and
340 nano Fe oxides (FeA), the amount of crystalline Fe oxides as goethite and hematite (FeD), and
341 crystalline magnetite (FeM). For the 1st extraction, 30 mg sample were leached with 10 ml of an
342 ascorbate solution (extractant solution) and shaken in dark conditions for 24 h and filtered. Another
343 30 mg were leached with 10 ml of a dithionite solution (extractant solution), shaken for 2 h in dark
344 conditions and filtered for the 2nd extraction. The solid residue was then leached again in 10 ml of an
345 oxalate solution for 6 h in dark conditions and filtered for the 3rd extraction. The extracted solution
346 of each phase (FeT, FeA, FeD and FeM) was analysed to quantify dissolved Fe by Inductively Coupled
347 Plasma Atomic Emission Spectrometry (ICP-AES). FeA is obtained with the 1st extraction, FeD is
348 obtained subtracting from the 2nd extraction the amount of Fe from the 1st extraction. Finally, the
349 FeM is related to the 3rd extraction. At the end, the equivalent to the Fe as structural Fe was obtained:
350 $\text{FeS} = \text{FeT} - \text{FeA} - \text{FeD} - \text{FeM}$ which is included in other minerals and amorphous phases. To test
351 accuracy, 30 mg of Arizona Test Dust (ATD; ISO 12103-1, A1 Ultrafine Test Dust; Powder Technology
352 Inc.) was subjected to the same extraction procedure in every batch and extraction.

353 The averaged Fe content of the reference material 1633b was $7.6 \pm 0.5 \%$ (certified 7.8%).
354 Furthermore, the average values of the sequential Fe extraction of the ATD reference material were
355 0.073 ± 0.012 , 0.47 ± 0.01 , and $0.042 \pm 0.002 \%$ for FeA, FeA+FeD and FeM, respectively, while the
356 certified contents are 0.067, 0.48, and 0.047 %, respectively.

357 **3. Results**

358 **3.1. Particle size distribution**

359 The PSD and median particle diameter of fully and minimally disturbed samples are key parameters
360 for understanding the cohesion and aggregation state of sediments (González-Romero et al., 2024).
361 We note that in the Mojave Desert some basins are enriched in salts, which can cause some artefacts
362 in the FDPD. The dissolution of salts during wet dispersion for bulk PSD analysis (<2 mm) can remove
363 aggregating agents. These salt cementation of the crusts might reduce the dust emission potential of
364 the surface.

365 The average PSDs of crusts across different basins exhibit remarkable similarity, yet disparities
366 between FDPDs and MDPSDs are pronounced, indicating varying degrees of particle cohesion and
367 aggregation at Cronese, Mesquite, Ivanpah and Coyote lakes. In these locations, FDPDs feature a
368 dominant mode at 8-10 μm alongside a coarser mode at 100 μm , while MDPSDs are characterized by
369 a dominant coarser mode (Figure 5). In contrast, Soda Lake crusts exhibit similarity between FDPDs
370 and MDPSDs. Averaged FDPDs and MDPSDs of aeolian ripples from the Mojave Desert are found to
371 be similar, typically featuring a major size mode between 100-300 μm . However, distinctions arise
372 when analysing specific lakes. Aeolian ripples from Soda, Cronese, and Coyote lakes showcase a
373 dominant coarse mode at 200-300 μm , whereas those from Mesquite Lake show a dominant mode at
374 a finer scale, approximately at 100 μm (Figure 5).

375 The crusts' mean of all median particle diameters (mean median) in the analyzed Mojave Desert dust
376 source sediments reveal a coarser MDPSD compared to FDPD, with values of 92 and 37 μm ,
377 respectively. In contrast, the mean median particle diameter is similar for aeolian ripples (226 and 213
378 μm , respectively) (Table S1). Analysing specific locations, the mean median particle diameter from the
379 MDPSD of crusts varies, with the finest crust observed at Ivanpah Lake (35 μm) and the coarsest at
380 Mesquite Lake (141 μm). For FDPD, the finest crust originates from Coyote Lake (8.4 μm), while the
381 coarsest is from Soda Lake (52 μm) (Table S1). Similarly, for aeolian ripples, the mean median particle
382 diameters for both MDPSD and FDPD are finer at Mesquite Lake (167 and 67 μm , respectively) and
383 coarser at Cronese lakes (264 and 234 μm , respectively) (Table S1). The high degree of particle
384 aggregation observed in crusts, contrasting with the lower aggregation state in ripples, aligns with
385 findings reported for dust-emitting sediments from Morocco by González-Romero et al. (2023).

386 The mean median particle diameters of crusts collected in the Mojave Desert are similar to those from
387 the Morocco described by González-Romero et al. (2023). Specifically, the mean median MDPSD
388 diameter for the Mojave Desert ($92 \pm 74 \mu\text{m}$) closely resembles that of the Morocco Draâ Lower basin
389 ($113 \pm 79 \mu\text{m}$), albeit slightly finer, and is notably coarser than that of Iceland ($55 \pm 62 \mu\text{m}$) (González-
390 Romero et al., 2023, 2024). Furthermore, while the finest crust sampled in the Mojave Desert (Ivanpah
391 with 35 μm) is almost twice as coarse as the finest from Morocco (L'Bour with 20 μm). For FDPD, the
392 Icelandic top sediment surface is the coarsest ($56 \pm 69 \mu\text{m}$), followed by both Morocco and Mojave
393 crusts (37 ± 77 and $37 \pm 48 \mu\text{m}$, respectively). Additionally, average MDPSD median diameters of
394 aeolian ripples from the Mojave Desert sources samples closely resemble those from Morocco (226
395 and 221 μm , respectively), while those from Iceland are slightly coarser (280 μm).

396 Dry sieved size fractions of dust-emitting sediments show the highest percentage of mass in the 250-
397 500 and 80-250 μm fractions, with minimal mass at 500-1000 μm and 1-2 mm and in the finer fractions
398 (20-40 and $<20 \mu\text{m}$) (Figure 6, Table S2). In both cases, the size fractions from 80 to 500 μm
399 accumulated a total of 75 to 90 % of the total mass fraction (Table S2).

400 Close to the centre of the Soda Lake, where numerous crust samples were collected, before reaching
401 massive crust cementation by evaporite minerals, the FDPD median diameter reaches very fine sizes
402 (8-15 μm) (Figure S1). In contrast, towards the edges of the basin (closer to the mountains surrounding
403 this endorheic lake), the size markedly increases, ranging from 22 to 87 μm (Figure S1). Similar
404 patterns, yet with coarser sizes, are observed for the MDPD. As described in previous studies, the
405 fluctuation of the groundwater table in the centre of the basin can lead to a massive precipitation of
406 salts, resulting in the formation of compact crusts (Figure 4) (Reynolds et al., 2007; Nield et al., 2016a;
407 Nield et al., 2016b; Urban et al., 2018) that should effectively reduce dust emission. However, at the
408 edges, where the precipitation of salts is less frequent, and reworking of the crusts by fluctuations in
409 the groundwater occurs, salty and spongy crusts are formed (Figure 4) (Nield et al., 2016a and Nield
410 et al., 2016b). These spongy crusts, being less compact, are more easily broken by saltating particles,
411 potentially leading to high-salt dust emissions.

412 The slight particle size segregation, with finer particles accumulating towards the center of the lake,
413 can be attributed to the transport of sediments from the surrounding mountains to the lake's center
414 by runoff waters during rain episodes. Initially, the coarser particles are deposited, followed by the
415 finer particles that remain suspended in the water for a longer duration. Nevertheless, the crusts in
416 the surroundings alluvial fans of the Soda Lake (22-87 μm in the edges compared to 8-15 μm in the
417 centre, Figure S1) are fine enough and surrounded by dunes (availability of saltators for saltation
418 bombardment) to have a high potential dust emission under favourable conditions (Reynolds et al.,
419 2006; Reheis et al., 2009; Urban et al., 2018).

420 **3.2. Mineralogy**

421 Dust-emitting sediments from the Mojave Desert primarily consist of feldspars ($41 \pm 12 \%$, including
422 albite/anorthite and microcline), quartz ($22 \pm 11 \%$) and clay minerals ($18 \pm 12 \%$, such as kaolinite,
423 montmorillonite and illite). Additionally, minor contents of carbonate minerals ($6.6 \pm 6.6 \%$),
424 amphibole (pargasite) $4.1 \pm 1.5 \%$, and iron oxides (maghemite/magnetite) ($0.77 \pm 0.54 \%$) are
425 observed (Figure 7, Tables 2 and S3). At Soda, Mesquite and Cronese lakes, Na-salts such as halite,
426 thenardite, trona, and burkeite are also present, with an average salt content of $5.0 \pm 11 \%$.
427 Additionally, zeolites ($0.77 \pm 1.1\%$ to 8.5%) including laumontite and analcime are detected at Soda,
428 Cronese, and Coyote lakes (the southern sites), with the highest content observed at Coyote Lake.
429 High amounts of gypsum are found at Mesquite Lake ($15 \pm 29 \%$) (Figure 7, Tables 2 and S3). Moreover,
430 Mesquite Lake crusts exhibit high contents of dolomite and calcite ($15 \pm 11\%$) compared to other
431 basins ($3.6 \pm 2.6\%$ to 7.2%) (Table 2).

432 The overall mineral composition of the dust-emitting sediments originates primarily from the source
433 rocks prevalent in the region. These include dominant of Mesozoic granitic rocks, as well as pre-
434 Tertiary, Tertiary and Quaternary volcanic rocks, and Pre-Cambrian and Mesozoic metamorphic rocks
435 (Figure 1). In the northern, northeastern, and eastern areas of Mesquite Lake, an important limestone
436 and dolostone massif from the Palaeozoic era contributes notably to the high content of calcite and

437 dolomite in the sediments of this lake (Figure 1). Zeolite content in the sediments may be attributed
438 to the weathering of volcanic outcrops in the region or to precipitation in alkaline lakes. This diverse
439 bedrock mineralogy results in a wide variety of minerals in the dust-emitting sediments. The form of
440 iron oxide detected in the samples, identified via XRD, is maghemite. However, distinguishing between
441 maghemite and magnetite using XRD is challenging (Vandenberghe et al., 2000), and magnetite has
442 been found to be ubiquitous in Mojave dust (Reheis et al., 2009; Reynolds et al., 2006). Therefore, we
443 refer hereafter to maghemite/magnetite to account for the potential presence of both minerals in the
444 samples. In comparison to aeolian ripples, the average composition of the crusts shows enrichment in
445 clay minerals (24 ± 11 versus 7.8 ± 2.3 % in crust and ripples, respectively), carbonates (6.6 ± 6.6 versus
446 1.1 ± 2.2 %), Na-salts (7.3 ± 13 versus 1.1 ± 3.7 %), zeolites (1.2 ± 1.9 versus 0.12 ± 0.52 %) and
447 maghemite/magnetite (0.92 ± 0.59 versus 0.49 ± 0.28 %), while being depleted in quartz (16 ± 7.2
448 versus 32 ± 9.5 %), feldspars (37 ± 9.7 versus 48 ± 13 %) and gypsum (3.1 ± 14 versus 4.7 ± 20 %), with
449 similar amphibole content (4.1 ± 1.5 versus 4.1 ± 1.6 %) (Figure 7, Tables 2 and S3). These mineral
450 enrichment and depletion trends in crusts are observed in all the playa lakes, except for Mesquite
451 Lake, which is discussed below.

452 In Soda Lake, the concentration of Na-salts in crusts increases towards the inner part of the lake,
453 ranging from 5-10 % at the margins to 45-50 % in the centre, where compact and fully salt-cemented
454 crusts form. This phenomenon is illustrated in Figure 8, which presents a geological and mineralogical
455 cross-section of Soda Lake. In addition to the water transport to this central part of the basin during
456 the rain episodes, groundwater discharge from the Zzyzx mountains occurs. There, the groundwater
457 table is close to the surface, and its interaction with the surface causes the massive mobilisation of
458 Na-salts that consolidate the crusts (Figure 4) (Nield et al., 2016b). Cycles of precipitation and
459 dissolution of the salts yield salty spongy crusts (Figure 4) at the edges of these massive crusts, with
460 higher dust emission potential in the degraded salty crusts (Nield et al., 2016a). The very high content
461 of Na-salts content in Soda Lake is attributed to the continuous high Na-S-Cl groundwater interaction
462 in the vicinity of Zzyzx, defining Soda Lake as a wet playa lake according to Reynolds et al. (2009) and
463 Urban et al. (2018). On the other hand, Cronese, Coyote, and Ivanpah are categorized as dry lakes.

464 Mesquite Lake features extensive gypsum deposits at the surface, which are a major component of
465 both dunes and crusts. A small gypsum mine operates in Mesquite Lake. The gypsum content in crusts
466 is notably higher at the center (80%) compared to the margins (3-11%). In contrast, the contents of
467 Na-salts and carbonates are greater at the margins (30% and 12-18%, respectively) than at the center
468 (7.5-14% and <0.1-6.9%, respectively). Aeolian ripples at the center of Mesquite Lake exhibit a very
469 high gypsum content, whereas at the margins, these ripples contain higher amounts of quartz,
470 feldspars, and clays than at the center. Despite the presence of the disturbed mine area, most large
471 dust events at Mesquite Lake have been observed to originate from natural (undisturbed) playa
472 surfaces near the margins (Reynolds R., personal communication).

473 Amphiboles in the Mojave Desert, sourced from metamorphic rocks of the area, are homogeneous
474 and can serve as a marker for emitted desert dust in the region. Comparing mineralogy from Mojave
475 Desert crusts to Moroccan surface samples (González-Romero et al., 2023), the former are largely
476 enriched in feldspars, clay minerals, Na-salts, and gypsum, and depleted in quartz and carbonates,
477 with trace proportions of amphibole, zeolites, and maghemite/magnetite. Ripples in the Mojave
478 Desert are depleted in quartz and carbonates, enriched in feldspars, clay minerals, Na-salts, and
479 gypsum, with traces of amphibole, maghemite/magnetite, and zeolites compared to Moroccan

480 ripples. The mineralogy of the Mojave Desert is markedly different from that of Iceland, due to
481 differences in bedrock geology, although both contain feldspars, zeolites, and maghemite/magnetite
482 (González-Romero et al., 2024).

483 Particle aggregation of the dust-emitting sediments from the Mojave Desert samples is similar to those
484 described by González-Romero et al. (2023) for Moroccan samples, likely due to the presence of clays,
485 Na-salts and precipitated carbonates. This aggregation inhibits aerodynamic entrainment and dust
486 emission should be primarily controlled by saltation bombardment (Shao et al., 1993). According to
487 the XRD analysis, the occurrence of crystalline Fe oxides is limited to maghemite/magnetite in contrast
488 to the hematite/goethite content found in Moroccan crusts (González-Romero et al., 2023). However,
489 due to the XRD low precision on the detection of low contents of minerals such as hematite and
490 goethite, their presence in the samples cannot be ruled out. In fact, both the EMIT standard products
491 (Figure 3), and the Fe mode of occurrence analysis discussed in the next section suggest the presence
492 of hematite and goethite.

493 The EMIT standard products (Figure 3) indicate the presence of phyllosilicates such as kaolinite,
494 smectite, montmorillonite, and illite, broadly consistent with our results. Specifically, around Mesquite
495 Lake, where elevated levels of gypsum and carbonates were detected, the EMIT results corroborate
496 the significance of these minerals in the same vicinity. Similarly, in Coyote, Ivanpah, and Cronese Lakes,
497 there is agreement regarding the prevalence of illite and muscovite as the major clay minerals,
498 alongside kaolinite. However, discrepancies arise in Soda Lake, where EMIT identifies a dominant
499 presence of montmorillonite, contrasting with our XRD results indicating a predominance of illite,
500 muscovite, and kaolinite. While Tetracorder identified predominant montmorillonite, illite, muscovite
501 and kaolinite could be on the order of 30% of the montmorillonite abundance and not show in the
502 EMIT spectra without a more sophisticated non-linear radiative transfer model to find the relative
503 abundances of these two minerals. This is due to the relative absorption strengths of the spectral
504 features of these minerals relative to those in montmorillonite. While our XRD analyses highlight the
505 presence of maghemite/magnetite, these minerals do not present clear absorbing features in the
506 spectral range of the EMIT instrument and are not considered within the 10 EMIT standard minerals.
507 In contrast to the XRD results, EMIT highlights the significant presence of goethite in the northern
508 sources (Mesquite and Ivanpah Lakes). Conversely, in the southern sources (Soda, Cronese, and
509 Coyote Lakes), EMIT highlights a major mixture of Fe²⁺ and Fe³⁺ species. The limited precision of XRD
510 for low proportions of Fe oxides, underscores the need for complementary techniques and analyses
511 to bolster our findings.

512 The mineralogical composition of the dry size-segregated fractions of the dust-emitting sediments is
513 outlined in Table S4. The findings indicate that there is no significant size enrichment process in crusts;
514 rather, there exists a relatively uniform distribution of quartz, feldspars, zeolites, and Fe oxides across
515 all size fractions (Figure 6). A slight, albeit not significant, enrichment of carbonates and clays is
516 observed, along with a slight depletion of Na-salts and gypsum in the finer fractions (<20 µm).
517 Additionally, pargasite shows a slight enrichment in the 40-80 µm fraction. In contrast, for aeolian
518 ripples, quartz exhibits significant enrichment in the coarser fraction (250-500 µm) and depletion in
519 the finer ones (<80 µm). Regarding carbonates, clays, and Fe oxides, there is an enrichment towards
520 the finer fractions (<20 µm), while the content of feldspars remains relatively homogeneous. Pargasite
521 content increases in the 40-80 µm fraction, and Na-salts and gypsum are either not detected or
522 present in trace amounts (Figure 6). The notable disparity in the enrichment factor between crusts

523 and aeolian ripples is partly attributed to the reduced amount of sand and the differing cohesion
524 states: crusts exhibit high cohesion, resulting in a homogenized mineralogy across size fractions (as
525 aggregates form a homogeneous concretion of minerals), while aeolian ripples display lower or
526 negligible aggregation, leading to a slightly more heterogeneous mineralogy across size fractions
527 compared to crusts.

528 **3.3. Mode of occurrence of Fe**

529 The average content of FeT in the crusts is 3.0 ± 1.3 wt %, while for aeolian ripples is 1.9 ± 1.1 wt %.
530 Among these crusts, 1.8 ± 0.92 % of the FeT occurs as FeA, 17 ± 7.2 % as FeD, 2.1 ± 1.2 as FeM and 79
531 ± 8.5 % as FeS (Tables 3 and S5). Aeolian ripples have very similar contents and modes of occurrence
532 of Fe in the analysed samples of the Mojave Desert.

533 Among the crusts, Ivanpah has the highest FeT content at 4.9 %, followed by Cronese and Coyote lakes
534 (3.7 ± 1.2 % and 3.5 %, respectively), with Soda Lake showing a similar content (3.1 ± 1.2 %). Mesquite
535 has the lowest FeT (1.6 ± 0.53 %), probably due to dilution of detrital Fe-bearing minerals with salts
536 and gypsum. FeS is the dominant mode of occurrence in most lakes, ranging from 68 % (1 sample) at
537 Ivanpah, to 74 ± 3.5 and 74 ± 13 % at Mesquite and Cronese, and to 83 ± 2.8 and 82 % at Soda and
538 Coyote lakes. The FeD is higher at Ivanpah (29 %), Cronese Lakes (21 ± 11 %) and Mesquite Lake ($20 \pm$
539 2.7 %) than at Soda and Coyote lakes (14 ± 2.5 and 14 %). FeM is higher at Mesquite Lake (3.7 ± 1.2
540 %), followed by Cronese and Coyote lakes (2.3 ± 1.1 and 2.4 %), and Soda (1.5 ± 0.49 %) and Ivanpah
541 Lakes (0.82 %). Finally, FeA is higher at Cronese Lake (2.4 ± 0.99 %), compared to Coyote, Mesquite,
542 Soda and Ivanpah lakes (1.8, 1.8 ± 0.93 , 1.5 ± 0.81 and 1.4 %) (Tables 3 and S5). Crusts are enriched in
543 FeT, FeD and FeA compared to ripples, while ripples are enriched in FeM and FeS (Tables 3 and S5).

544 Overall, the bulk Fe content in crusts is driven by structural Fe from clays and amphiboles (as deduced
545 from the high correlation shown in Figure 9a), followed by small proportions of hematite and goethite
546 (not detected by XRD), which are clearly higher at the northern lakes Ivanpah and Mesquite lakes
547 (consistent with the highlighted presence of goethite in the EMIT standard products), probably due to
548 the Precambrian and Cambrian metamorphic rocks that supply sediments. Furthermore, the easily
549 exchangeable Fe is also driven by clay minerals (Figure 9b).

550 Compared to crusts in other arid regions analysed by González-Romero et al. (2023, 2024), the
551 analysed Mojave Desert crusts have similar FeT content than Moroccan crusts but are much lower
552 than the Iceland top sediments (loose surface sediments in Iceland according to González-Romero et
553 al. (2024)) (3.0 ± 1.3 , 3.6 ± 0.71 and 9.5 ± 0.39 %, for Mojave, Morocco, and Iceland respectively). The
554 proportion of FeS in FeT is similar to the Icelandic sediments but higher than in Moroccan samples (79
555 ± 8.5 and 79 ± 6.5 %, and 67 ± 2.4 , respectively). The proportion of FeM is clearly lower than that of
556 Iceland, but higher than that of Morocco (2.1 ± 1.2 and 16 ± 5.4 %, for Mojave and Iceland; Morocco
557 proportion is negligible). The FeD proportion is intermediate between Morocco and Iceland (17 ± 7.2 ,
558 31 ± 2.3 , 3.5 ± 1.5 %, respectively), while the FeA proportion is similar to both Morocco and Iceland
559 crusts (1.8 ± 0.92 , 1.3 ± 0.39 and 1.9 ± 0.55 %, respectively) (Figure 10).

560 **4. Discussion and conclusions**

561 The playa lakes sampled within the Mojave Desert can serve as significant dust-emitting sources in the
562 region. Descriptions provided by Urban et al. (2018) and satellite imagery (Figure 2) confirm the

563 presence of desert dust emissions originated from these areas. The lithology, geological/tectonic
564 evolution, and past and current climate conditions collectively contribute to the formation of these
565 dust sources in the Mojave Desert.

566 Dust-emitting sediments in this region predominantly stem from substratum rocks, comprising mainly
567 granitic and volcanic formations, along with metamorphic Pre-Cambrian, Cambrian, Paleozoic, and
568 Mesozoic rocks. Endorheic basins, shaped by faulting during the Tertiary-Quaternary period,
569 accumulated fine sediments through erosion, transportation, and deposition processes. Wetter
570 conditions prevailing during the Pleistocene epoch led to the formation of deep lakes within the
571 basins, which gradually desiccated as the climate evolved. These arid conditions rendered the playa
572 lakes susceptible to dust emission under specific atmospheric conditions. Notably, a particle size
573 segregation is observed, transitioning from coarser sediments in the proximal alluvial areas towards
574 finer particle crusts within the central regions of the lakes. In the playa lakes, finer sediments
575 accumulate towards the centre of the lakes due to flood events inundating the central areas and
576 ponding, which facilitates the deposition of coarser particles followed by top finer sediment sizes.

577 According to the conceptual model depicted in Figure 11, the finer dust particle size distributions
578 range from 8.4 to 99 μm inside Soda Lake and 46 to 111 μm outside Soda Lake, underscoring this
579 sedimentation process. Comparisons with conceptual models proposed for other regions, such as
580 those by González-Romero et al. (2023, 2024) for locations in Morocco and Iceland, reveal a similar
581 transport fractionation phenomenon occurring in the Mojave Desert. These crusts, observed within
582 Soda Lake, show enrichment in clay minerals, carbonate minerals, salts, and iron oxides, while
583 experiencing depletion in coarser constituents such as feldspars and quartz.

584 In the Mojave Desert, there are two distinct types of playa lakes, characterized as wet and dry,
585 depending on the regime of the groundwater table and its relationship with the surface, as described
586 by Reynolds et al. (2007 and 2009), Buck et al. (2011), Nield et al. (2016a and b), Urban et al. (2018)
587 and Goudie (2018). Understanding the groundwater table regime is fundamental in this region due to
588 its profound relation on the porosity of the crust and its consequential impact on mineralogy, including
589 the precipitation and enrichment of salts (Figure 11). This dynamic contrasts sharply with other
590 conceptual models, where the relationship between crust formation and the groundwater table is
591 either minimal or absent entirely. For instance, there is no or little relation between crusts and
592 groundwater table in Morocco, and in Iceland, where the water regime is largely influenced by
593 floodings from glaciers (González-Romero et al., 2023, 2024). In wet playa lakes like Soda Lake, the
594 presence of salty crusts, whether massive or spongy, is significantly pronounced. Conversely, in dry
595 playa lakes such as Ivanpah, Coyote, and Cronese, the relationship of salt crusts is notably less
596 prominent as the proportion of Na-salts is lower (see Figure 11).

597 At Soda Lake, a hard crust, measuring up to 0.5 meters in thickness (Figure 3), forms through the
598 extensive precipitation of Na-salts, particularly near the Zzyzx area, where a relatively constant
599 mobilisation of salts is due to the water table evaporation or vapour discharge from deeper parts of
600 the sediment towards the surface (Nield et al., 2015, 2016a and b). Along the edges of this massive
601 crusty area, the frequent oscillation of the water table may result in the precipitation and dissolution
602 of salts in lower quantities compared to the centre, leading to the formation of weaker crusts
603 characterized by high porosity. These porous crusts may contribute to an increased dust emission rate
604 compared to the hard salt crusts found in the centre. Dry lakes such as Ivanpah, Cronese, and Coyote

605 do not exhibit the formation of spongy crusts due to the low concentrations of salts. In wet playas,
606 strong dust emission may happen when very strong winds rip off thin crusts exposing the fine-grained
607 sediment beneath including lithogenic and salt mineral particles (Rich Reynolds, personal
608 communication).

609 Particle aggregation facilitated by diagenetic salts and carbonate minerals is prevalent in the samples
610 dust-emitting sediments of the Mojave Desert, akin to the equivalent sediments found in the
611 Moroccan Sahara. The average grain size of the crusts from both regions is similar, with MDPSD values
612 of $113 \pm 79 \mu\text{m}$ for Morocco and $92 \pm 74 \mu\text{m}$ for the Mojave Desert, and FDPD values of $37 \pm 77 \mu\text{m}$
613 and $37 \pm 48 \mu\text{m}$, respectively. These patterns contrast with the lower aggregation state and finer
614 MDPSD observed in Icelandic dust ($55 \pm 62 \mu\text{m}$) (Table 4).

615 In terms of mineralogy, crusts from the Mojave Desert are enriched in feldspars, clay minerals, Na-
616 salts, and gypsum, whereas crusts from the Moroccan Sahara are enriched in quartz and carbonates
617 (Table 4). The mineralogy of Icelandic top sediments (loose surface sediments in Iceland according to
618 González-Romero et al. (2024)) differs due to their volcanic origin; however, both the Mojave Desert
619 crusts and Icelandic top sediments contain similar amounts of zeolites. Salt enrichment in the crusts
620 is primarily attributed to interactions with the groundwater table as shown in previous studies (Nield
621 et al. (2016a and b), Urban et al. (2018) and Goudie (2018) (Figure 11).

622 The total iron content (FeT) remains consistent throughout the samples collected in the Mojave
623 Desert, with slightly higher levels observed in the Ivanpah crust, albeit diluted by the high salt content
624 in the wet playa lake crusts or the elevated gypsum content in the Mesquite Lake. While the total Fe
625 content is comparable between the Mojave Desert and Moroccan Sahara crusts (3.0 and 3.6 wt %,
626 respectively), it is substantially lower than in Icelandic top sediments (9.3 wt %). Exchangeable Fe
627 proportions in FeT are similar among the three environments. The proportion of Fe from hematite and
628 goethite in Mojave Desert crusts fall between those of Moroccan Sahara crusts and Icelandic top
629 sediments (17, 31, and 0.5 wt %, respectively). The proportion of maghemite/magnetite in Mojave
630 Desert crusts is much lower compared to Icelandic top sediments (2.1 and 15 %, respectively). Finally,
631 the proportion of structural Fe in the samples is similar across the three environments.

632 In conclusion, the dust-emitting sediments collected from the Mojave Desert exhibit distinct
633 signatures in mineralogy and modes of occurrence of Fe compared to those from the Moroccan
634 Sahara, despite similar particle sizes. These differences can influence emitted dust properties, and
635 associated impacts. Similarities in fully disturbed and minimally disturbed particle size distributions
636 support comparable dust emission mechanisms, with saltation bombardment playing a prominent
637 role.

638 **Code availability.** The Tetracorder code used in this paper is provided by Clark (2023,
639 <https://github.com/PSI-edu/spectroscopy-tetracorder>).

640 **Data availability.** Data used in this paper are given in the main paper itself and in the Supplement. If
641 needed, data are also available upon request by emailing the authors.

642 **Author contribution.** Sample permits were obtained by BLE, RG and AK. The samples were collected
643 by CPG-P, AGR, AK, RG and XQ and analysed by AGR, MHC and NM. EMIT mineralogy maps were
644 produced by RG, PB and RC. PG provided the FoO map. AGR analyzed the data and wrote of the original

645 draft manuscript supervised by CPG-P and XQ. CPG-P and XQ re-edited the manuscript and all authors
646 contributed to data discussion, reviewing and manuscript finalization.

647 **Competing interests.** At least one of the (co-)authors is a member of the editorial board of
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949 **Figure captions:**

950 **Figure 1:** Study area map including the playa lakes studied together with a geologic map, simplified
951 from Jennings et al. (1962) and Miller et al. (2014). The star represents the Zzyzx complex and green
952 dots the samples used in this study. Basemap: Imagery data from © Google Earth Pro v: 7.3.6.9345.

953 **Figure 2:** Map of Frequency of Occurrence (FoO) of dust optical depth > 0.1 over the study region
954 derived from MODIS C6.1 Aqua (1:30PM equatorial passing time) Level 2 Deep Blue aerosol products
955 at 0.1 degree resolution. A dust occurrence is counted when DOD > 0.1, Angstrom Exponent < 0.3 and
956 DOD at 412 nm > DOD at 470 nm. Blue iso-contours represent 5 and 10 % of daily occurrences per
957 year averaged over 20 years (2003-2022). Green dots represent the samples collected and used in this
958 study. Basemap: Imagery data from © Google Earth Pro v: 7.3.6.9345.

959 **Figure 3:** EMIT scenes emit20231015T215209_color-visRGB and emit20230728T214142_color-visRGB
960 at 60 m per pixel showing the diversity of Fe²⁺, Fe³⁺ and Fe²⁺ and Fe³⁺ bearing minerals and the
961 carbonates, salt and phyllosilicates minerals.

962 **Figure 4:** Examples of samples collected in the Mojave Desert including crusts (a), aeolian ripples (b),
963 massive compact crust (c) and a salty spongy crust (d).

964 **Figure 5:** Fully dispersed particle size distribution (FDPD) and minimally dispersed particle size
965 distribution (MDPSD) for crusts and aeolian ripples from the Mojave Desert (median PSD from all the
966 samples), Soda, Mesquite, Cronese, Ivanpah and Coyote Lakes. In shaded blue and brown the standard
967 deviation of each PSD (n° of samples used in Table 1), except for Ivanpah and Coyote Lakes (only 1
968 sample each).

969 **Figure 6:** % of mass fractions from the dry sieved size fractions (250-50, 80-250, 63-80, 40-63, 20-40
970 and <20 µm). The range of the enrichment factors of each mineral group for each dry size fraction of
971 the 16 crust samples (blue) and for the 4 aeolian ripples samples (red).

972 **Figure 7:** Box-plot showing averaged mineral contents for all samples, crusts and aeolian ripples (wt
973 %).

974 **Figure 8:** Geological cross section and mineralogy of the crusts of the Soda Lake. Top panel represent
975 major mineralogy composition. Mid panel represents the position of the samples, the Zzyzx complex,
976 and the path of the cross section. Bottom schematic cross section simplifying the position in the basin.
977 Basemap: Imagery data from © Google Earth Pro v: 7.3.6.9345.

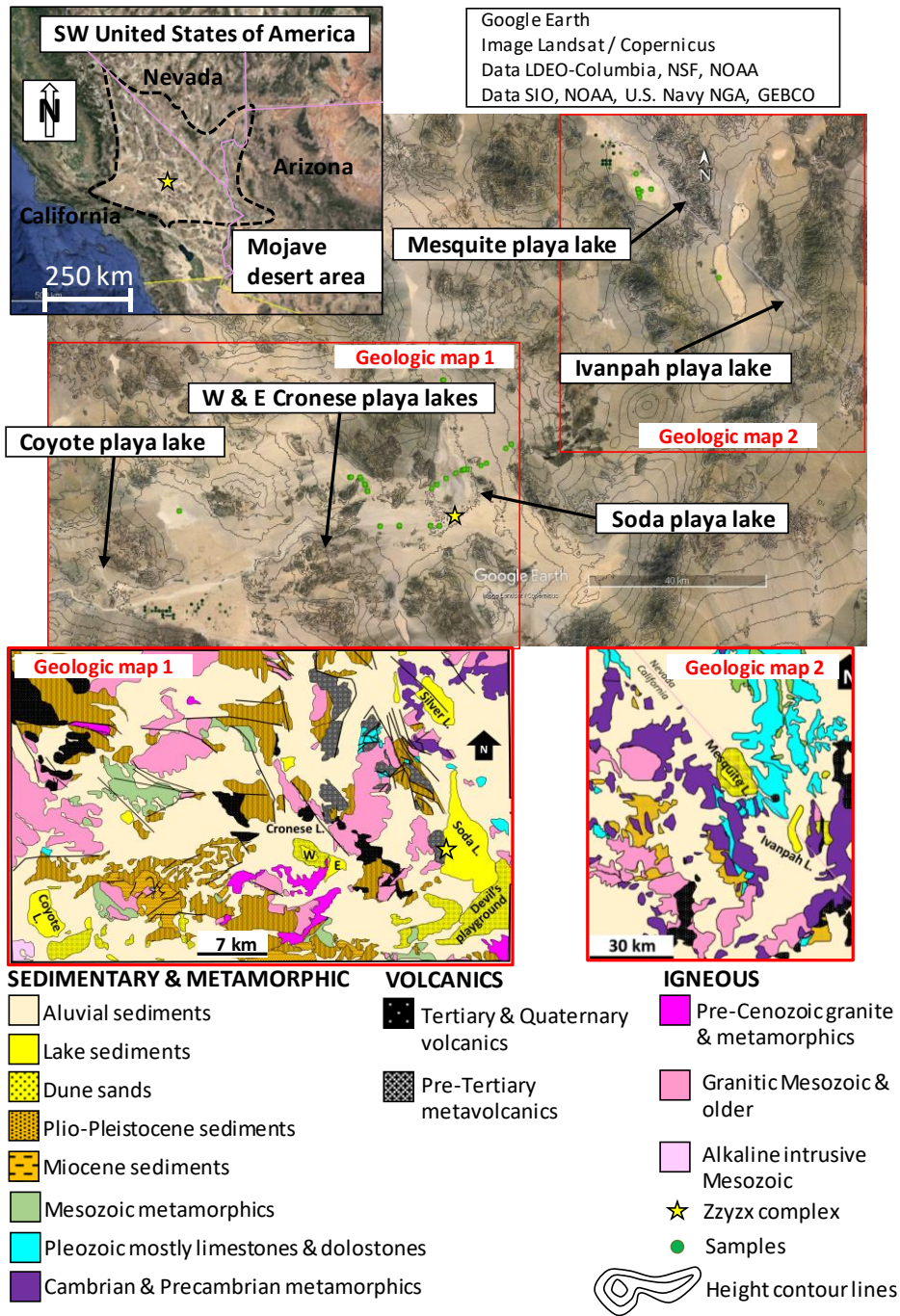
978 **Figure 9:** Cross-correlation plots of the clay contents and amphibole with the FeT (a) and clay minerals
979 and FeA (b), all in wt % in crusts.

980 **Figure 10:** Modes of occurrence of Fe comparison between the crusts (C) playa lakes analysed in this
981 study, the average of the crusts and ripples (R) at Mojave Desert, Morocco and Iceland Top surface
982 (TS). FeA is referred to the exchangeable Fe and nano Fe oxides, FeD is the Fe content in hematite and
983 goethite, FeM is the Fe content in maghemite/magnetite and FeS is the Fe content in Fe bearing
984 minerals.

985 **Figure 11:** Conceptual model of wet and dry playa lakes differences due to groundwater differences
986 and how this can affect the mineralogy of the surface in the playa lakes. Also illustrated is the expected
987 dust emission rate, major mineralogy and modes of occurrence of Fe differences expected in the
988 emitted dust.

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990 Figure 1.



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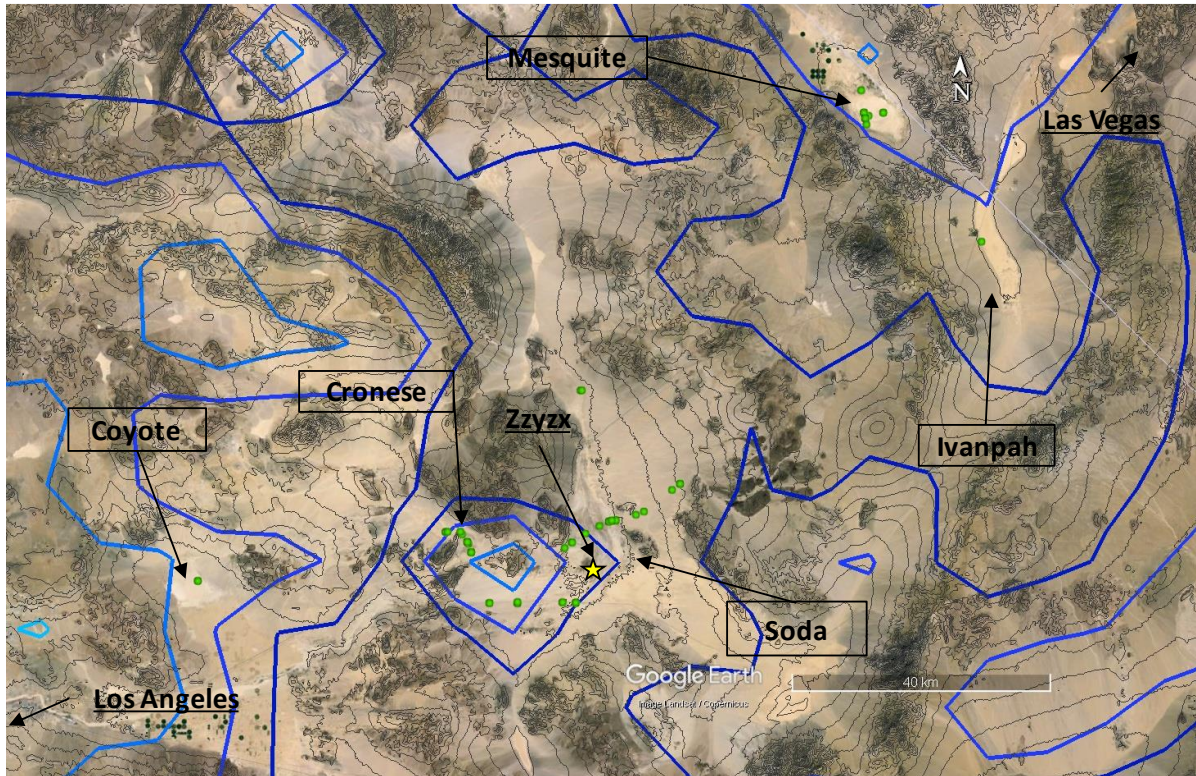
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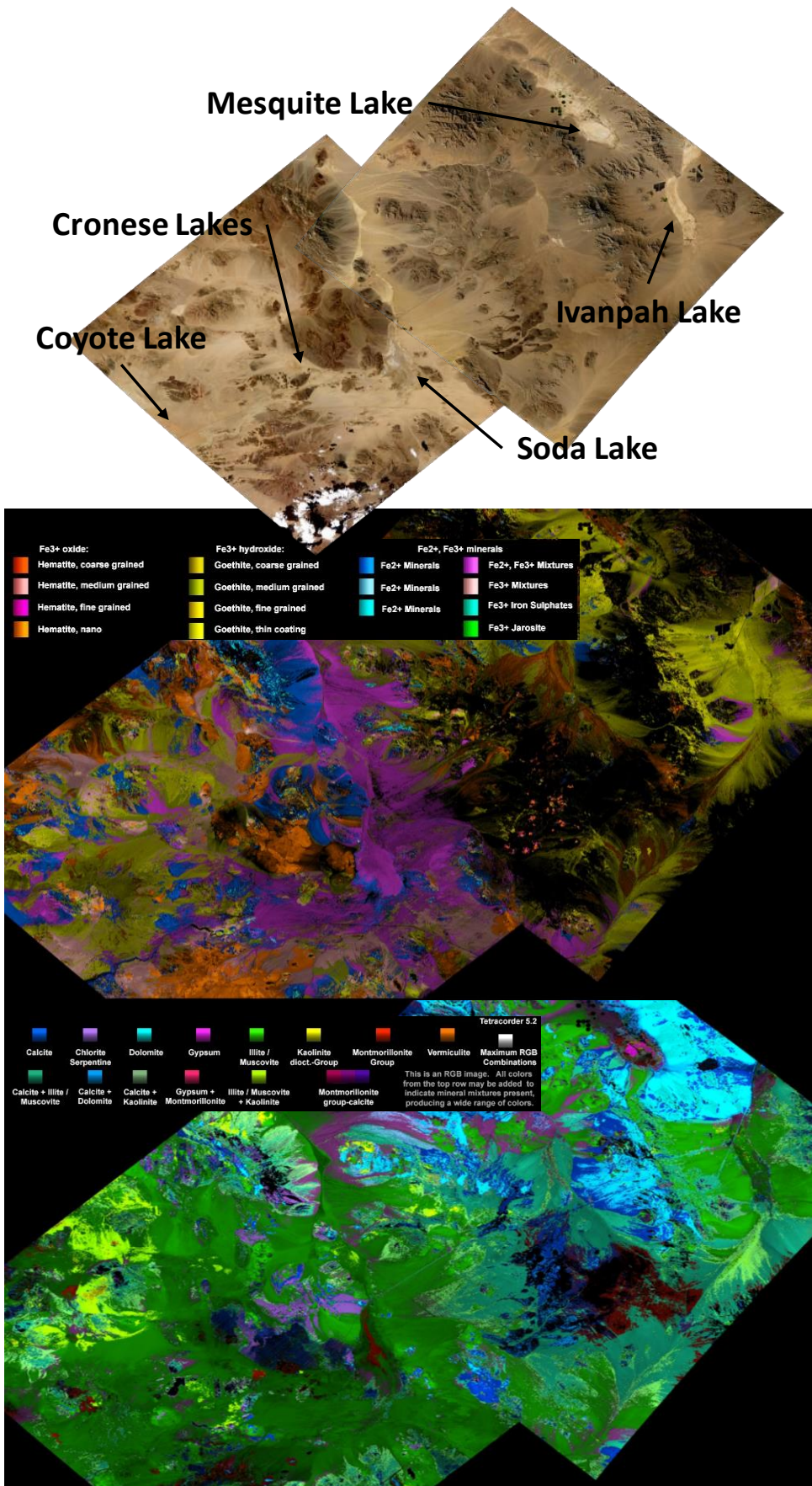
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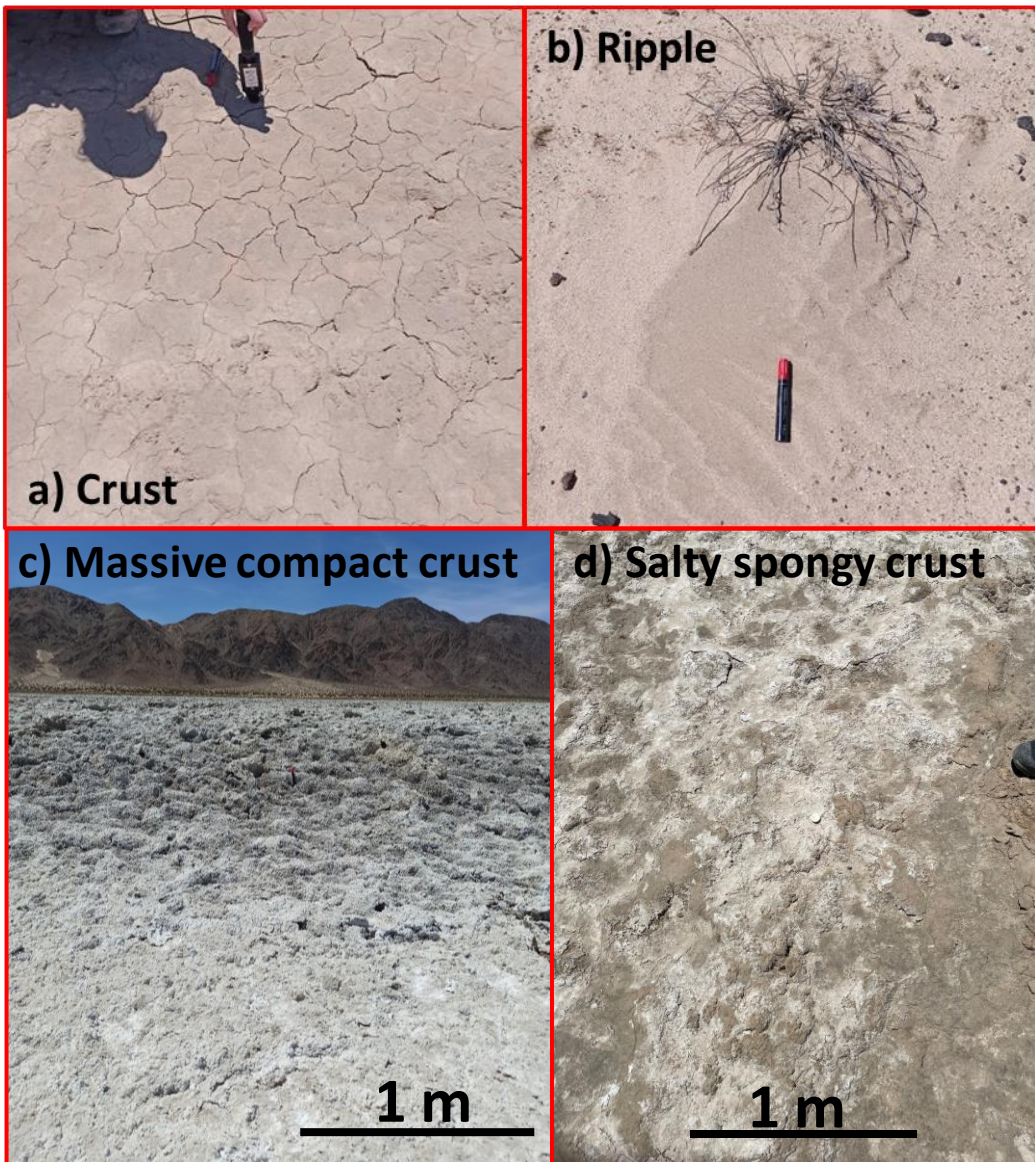
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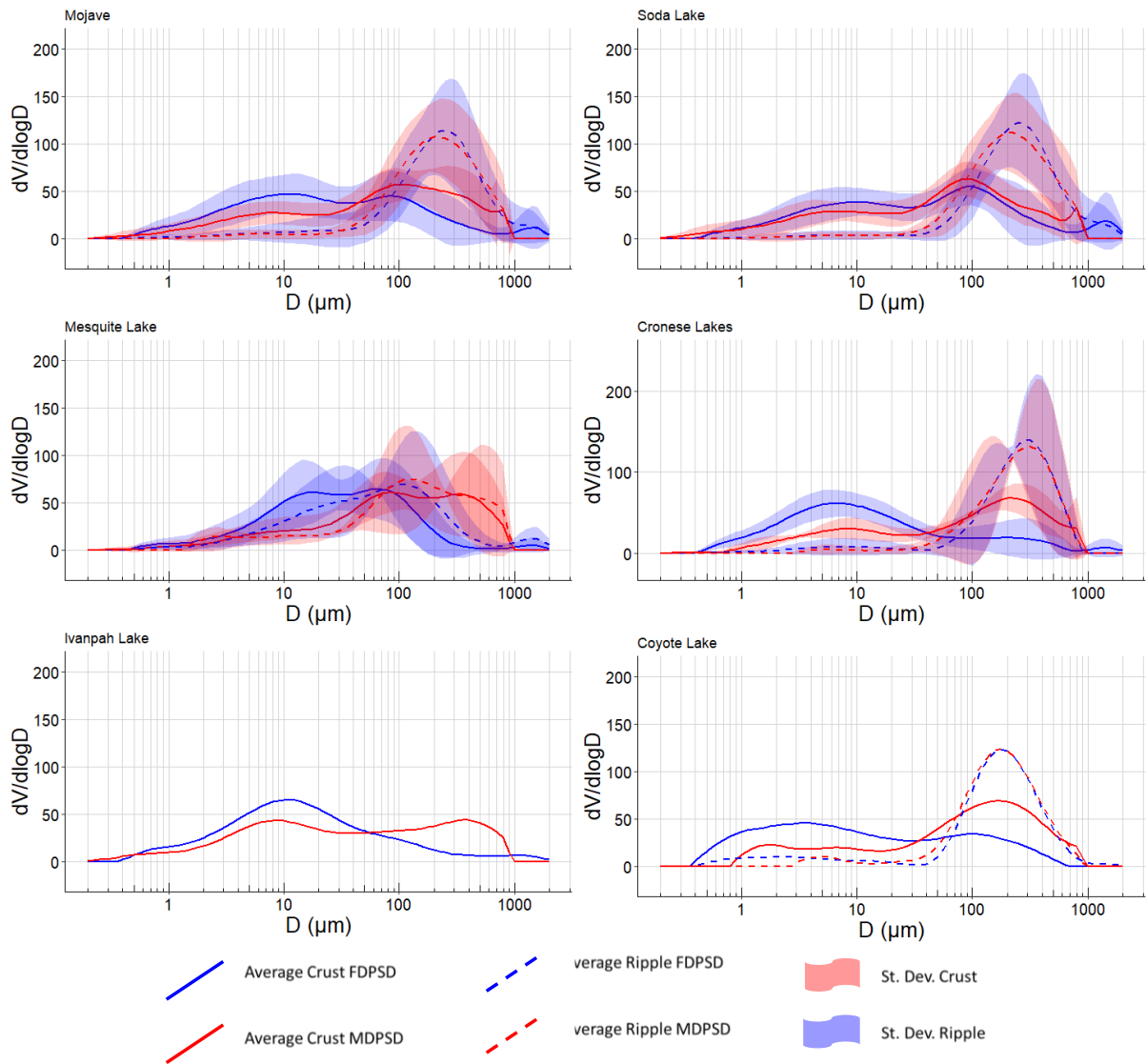
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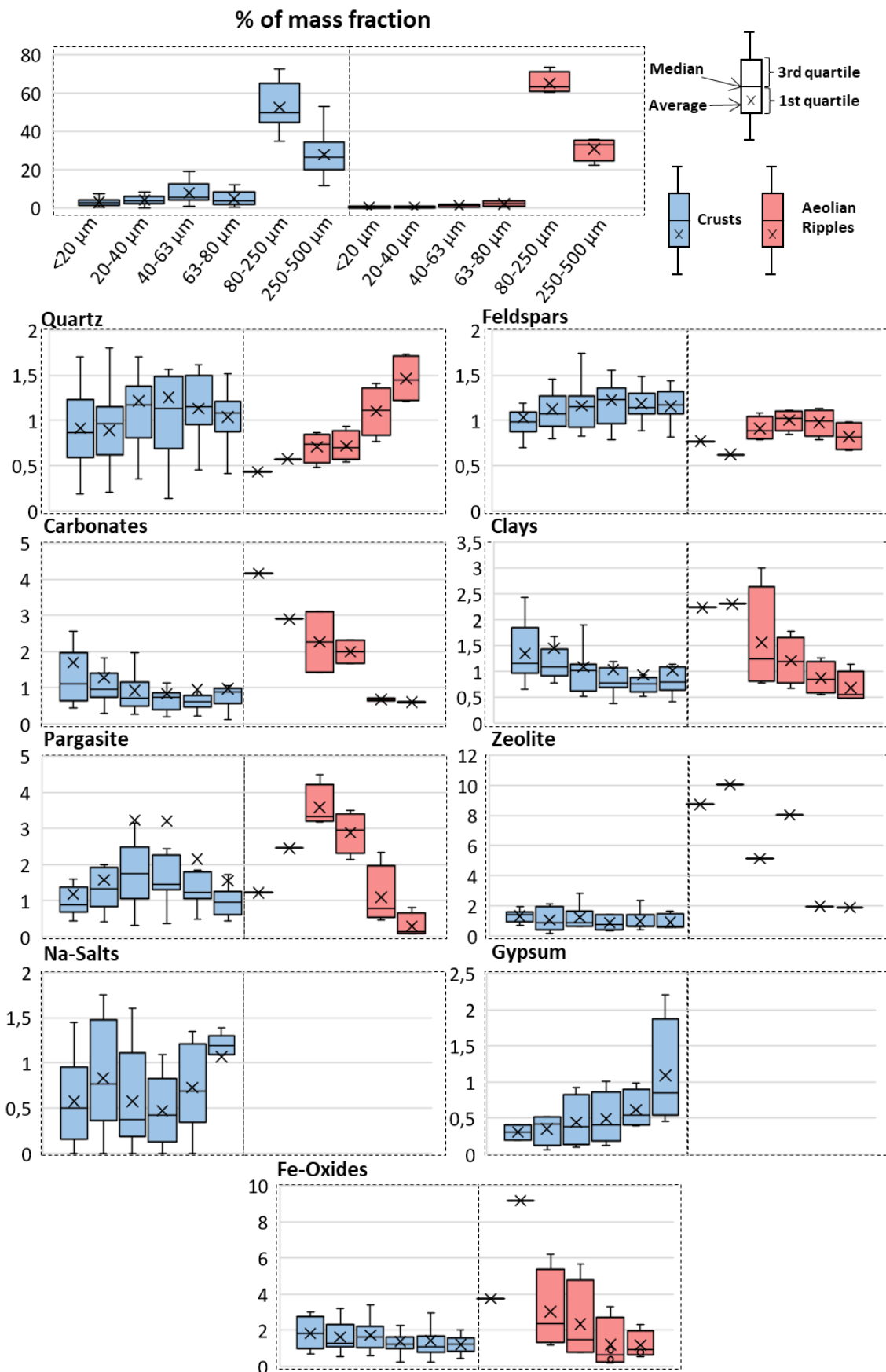
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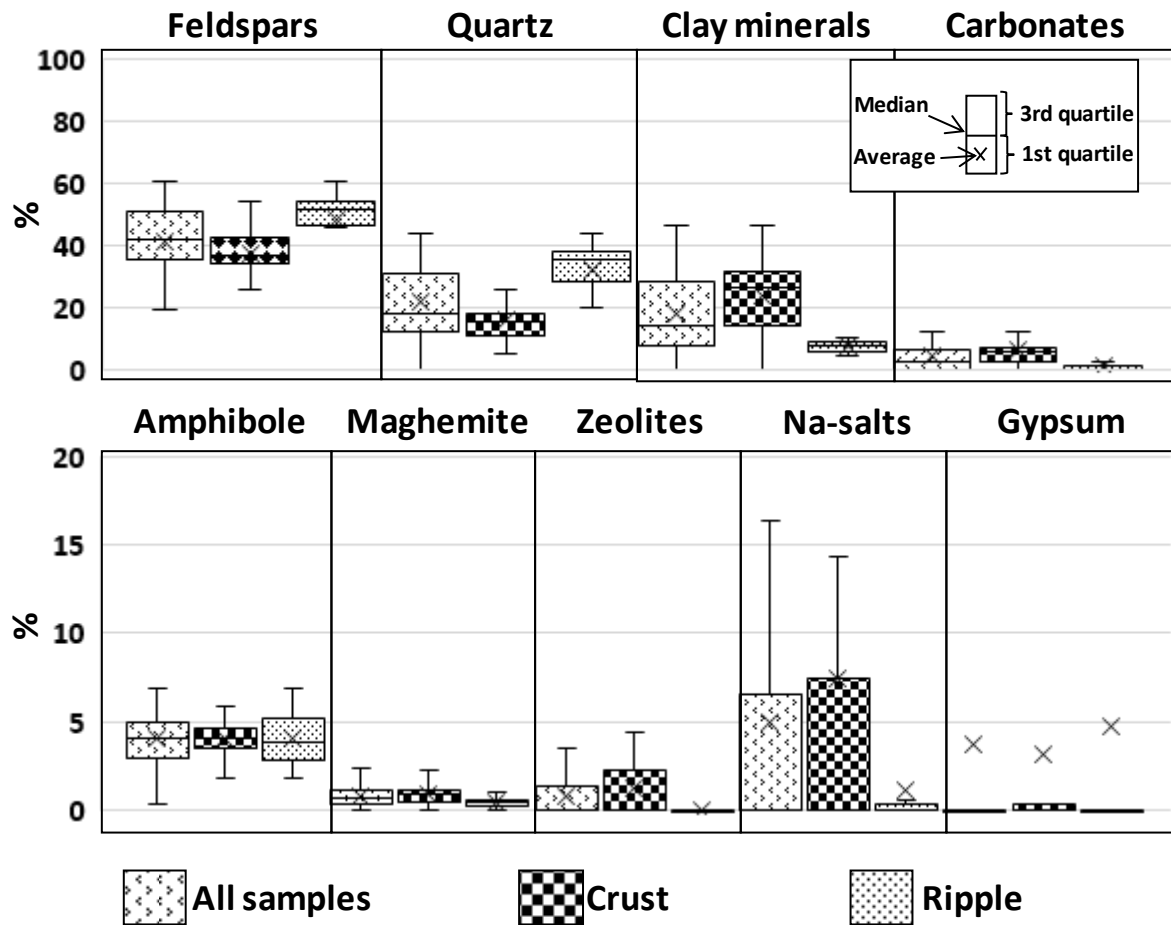
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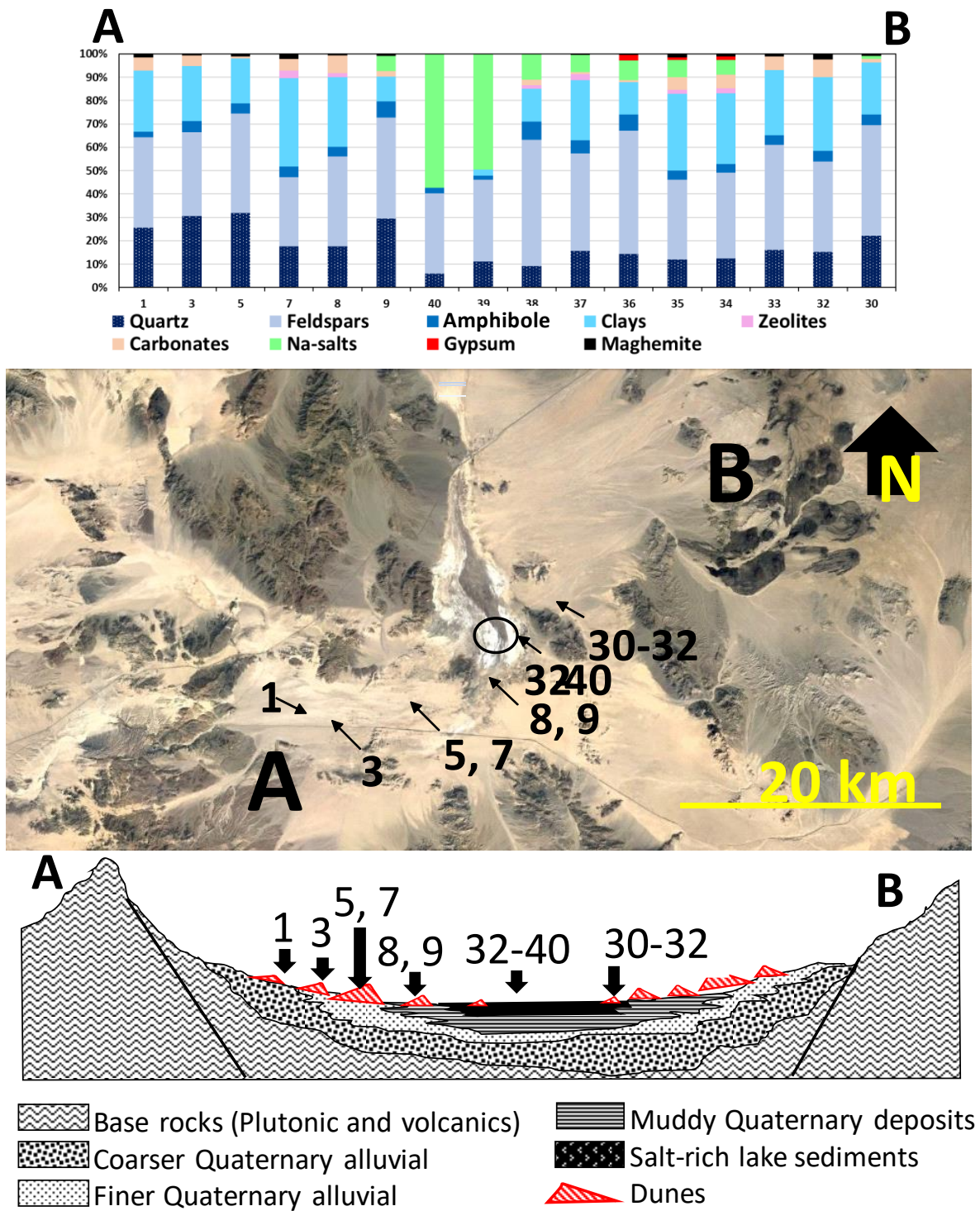


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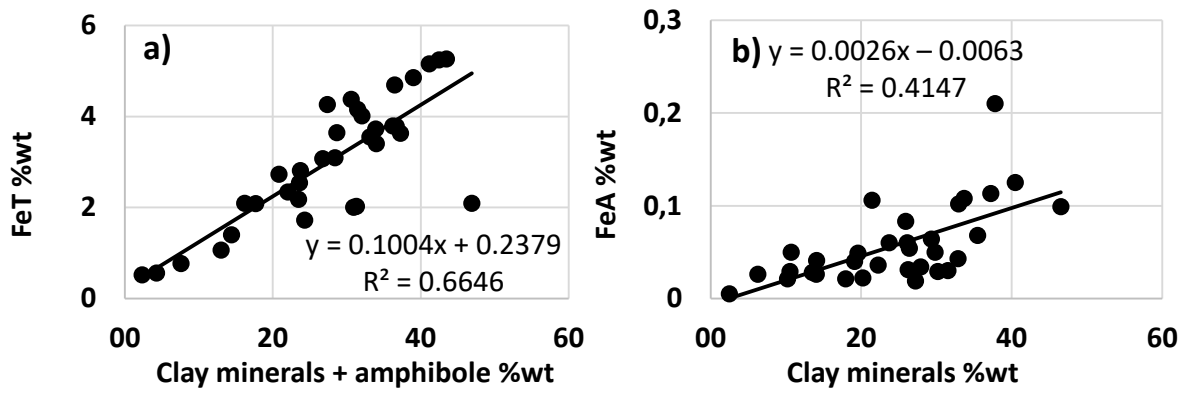
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1083 Figure 9.



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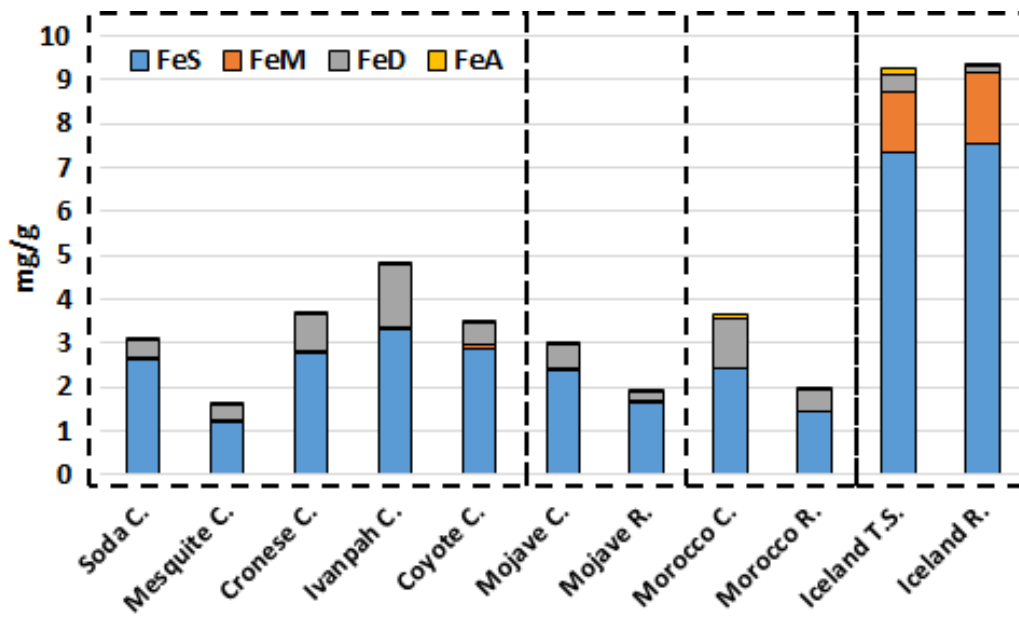
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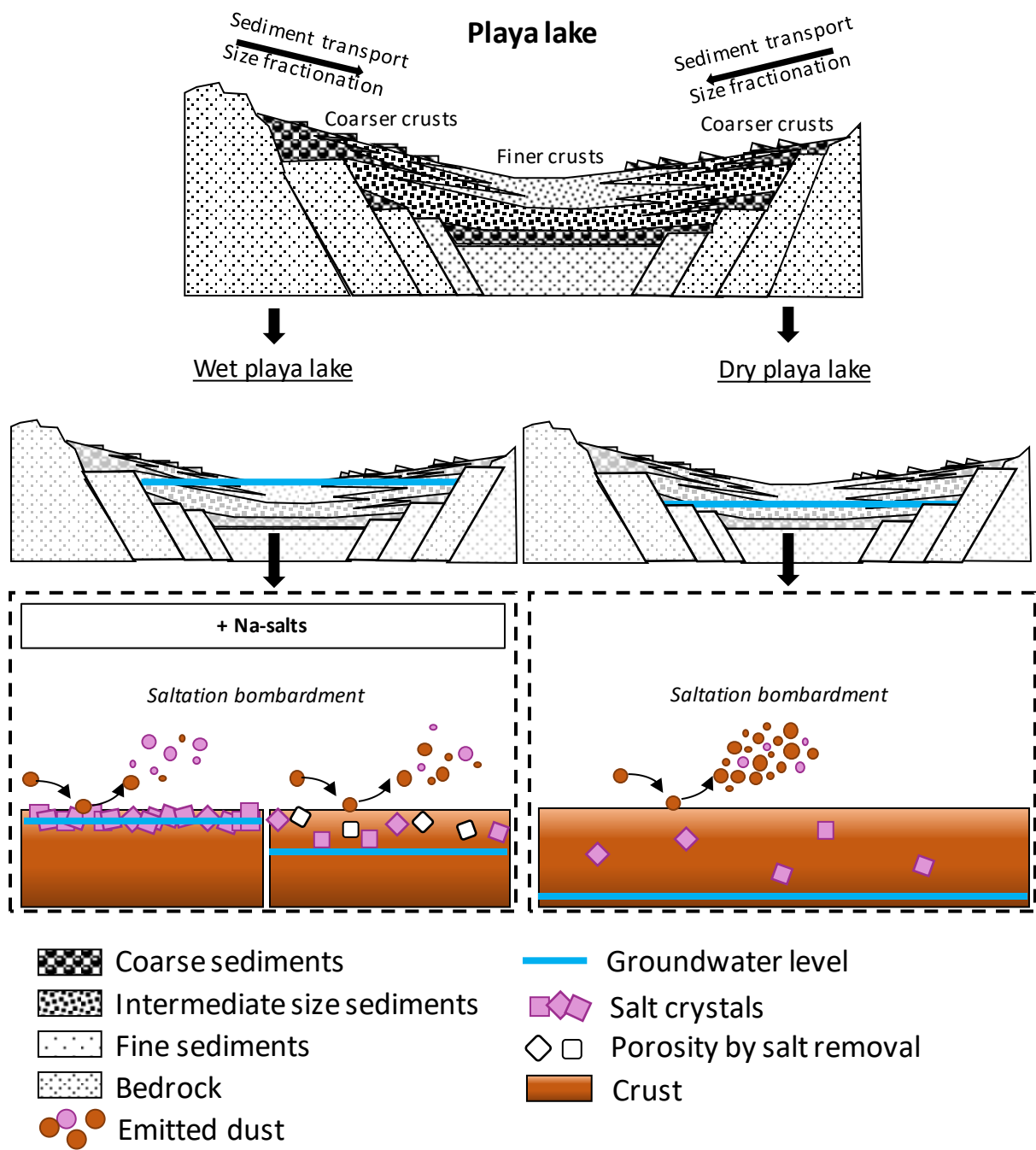
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1127 Figure 11.



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1138 Table 1. Full range (<2000 μm), <63 μm and >63 to 2000 μm mean diameter, standard deviation, min.,
 1139 and max for minimally dispersed particle size distribution (MDPSD) and fully dispersed particle size
 1140 distribution (FDPSD).

Surface	Location	N	MDPSD		
			Full range	$\leq 63 \mu\text{m}$	>63 to 2000 μm
			Mean of medians \pm Std. Dev. [Min,Max]		
Crusts	Mojave	35	92 \pm 74 [10,349]	22 \pm 6.4 [11,34]	254 \pm 71 [155,489]
Ripples		20	226 \pm 88 [88, 418]	37 \pm 6.0 [20,46]	276 \pm 80 [130,424]
Crusts	Soda	17	63 \pm 47 [10,156]	21 \pm 6.5 [11,31]	234 \pm 82 [155,489]
	Cronese	9	109 \pm 60 [35,195]	18 \pm 2.2 [15,22]	280 \pm 40 [238,357]
	Mesquite	7	141 \pm 117 [31,349]	28 \pm 5.6 [21,34]	257 \pm 79 [157,387]
	Ivanpah	1	35 \pm NA [35,35]	16 \pm NA [16,16]	314 \pm NA [314,314]
	Coyote	1	101 \pm NA [101,101]	20 \pm NA [20,20]	254 \pm NA [254,254]
Ripples	Soda	15	231 \pm 87 [88,418]	39 \pm 3.5 [29,43]	275 \pm 77 [130,424]
	Cronese	2	264 \pm 147 [160,368]	40 \pm 8.8 [34,46]	292 \pm 120 [208,377]
	Mesquite	2	167 \pm 112 [110,225]	26 \pm 8.9 [20,32]	286 \pm 146 [183,389]
	Ivanpah	0	NA	NA	NA
	Coyote	1	179 \pm NA [179,179]	32 \pm NA [32,32]	236 \pm NA [236,236]
			FDPSD		
Surface	Location	N	Full range	$\leq 63 \mu\text{m}$	>63 to 2000 μm
			Mean of medians \pm Std. Dev. [Min,Max]		
Crusts	Mojave	35	37 \pm 48 [4.9,240]	18 \pm 6.6 [8.4,35]	306 \pm 237 [106,1093]
Ripples		20	213 \pm 92 [28,362]	29 \pm 8.3 [15,48]	335 \pm 99 [213,561]
Crusts	Soda	17	52 \pm 61 [8.4,240]	19 \pm 5.3 [12,27]	321 \pm 212 [113,815]
	Cronese	9	17 \pm 23 [4.9,77]	12 \pm 3.1 [8.4,19]	381 \pm 345 [144,1093]
	Mesquite	7	34 \pm 28 [11,91]	24 \pm 7.7 [16,35]	185 \pm 104 [106,336]
	Ivanpah	1	12 \pm NA [21,21]	15 \pm NA [15,15]	347 \pm NA [347,347]
	Coyote	1	8.4 \pm NA [8.4,8.4]	12 \pm NA [12,12]	187 \pm NA [187,187]
Ripples	Soda	15	234 \pm 82 [92,362]	31 \pm 7.9 [21,48]	346 \pm 97 [238,561]
	Cronese	2	236 \pm 126 [147,325]	18 \pm NA [18,18]	295 \pm 108 [219,371]
	Mesquite	2	67 \pm 56 [28,107]	27 \pm 3.5 [24,29]	336 \pm 173 [213,458]
	Ivanpah	0	NA	NA	NA
	Coyote	1	156 \pm NA [156,156]	15 \pm NA [15,15]	245 \pm NA [245,245]

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1151 Table 2. Average and standard deviations of the mineral contents (wt %) from crust and aeolian ripple
 1152 samples from the Mojave Desert and the different study basins. NaN, not a number. Maghemite is
 1153 considered as Maghemite and Magnetite

	Clays	Carbonate	Salts	Zeolites	Maghemite	Quartz	Feldspars	Gypsum	Amphibole
CRUSTS	24±11	6.6±6.6	7.3±13	1.2±1.9	0.92±0.59	16±7.2	37±9.7	3.1±14	4.1±1.5
Soda	22±11	3.6±2.6	8.9±17	0.77±1.1	0.97±0.66	18±7.7	40±6.7	0.29±0.68	4.5±1.6
Cronese	31±11	5.4±1.8	2.2±3.4	2.4±1.7	1.0±0.28	14±7.3	40±5.5	<0.1	3.4±1.5
Coyote	28	7,2	1,2	8,5	0,48	11	37	<0.1	5.6
Ivanpah	36	6.9	<0.1	<0.1	1,2	15	36	<0.1	3.5
Mesquite	17±8.2	15±11	12±14	<0.1	0.71±0.75	14±5.8	24±12	15±29	2.8±1.4
RIPPLES	7.8±2.3	1.1±2.2	1.1±3,7	0.12±0.52	0.49±0.28	32±9.5	48±13	4.7±20	4.1±1.6
Soda	7.4±1.8	0.47±0.73	0.19±0.46	<0.1	0.49±0.25	35±4.5	52±4.7	<0.1	4.3±1.5
Cronese	8.4±0.60	1.2±1.7	<0.1	<0.1	0.83±0.33	32±9.0	53±0.03	<0.1	4.7±3.2
Coyote	7.9	2.3	<0.1	2,3	0.60	28	52	<0.1	3.5
Ivanpah	NaN	NaN	NaN	NaN	NaN	NaN	NaN	NaN	NaN
Mesquite	10±6.1	4.8±6.8	9.4±9.9	<0.1	0.19±0.27	10±14	15±21	47±60	3.7±1.5

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1182 Table 3. Fe content in wt % for total Fe (FeT) content, and in % for ascorbate Fe (FeA), dithionite (FeD),
1183 oxalate Fe (FeM) and structural Fe (FeS). NaN not a number.

	FeT	FeA %	FeD %	FeM %	FeS %
CRUSTS	3.0±1.3	1.8±0.92	17±7.2	2.1±1.2	79±8.5
Soda	3.1±1.2	1.5±0.81	14±2.5	1.5±0.49	83±2.8
Cronese	3.7±1.2	2.4±0.99	21±11	2.3±1.1	74±13
Coyote	3.5	1.8	14	2.4	82
Ivanpah	4.9	1.4	29	0.82	68
Mesquite	1.6±0.53	1.8±0.93	20±2.7	3.7±1.2	74±3.5
RIPPLES	1.9±1.1	1.4±1.2	12±5.6	2.4±1.8	84±7.5
Soda	2.0±1.2	0.98±0.39	10±3.4	2.1±1.8	87±4.4
Cronese	2.3±1.5	1.4±0.35	14±9.3	2.8±2.9	82±12
Coyote	1.3	3.4	26	3.0	68
Ivanpah	NaN	NaN	NaN	NaN	NaN
Mesquite	1.0±1.1	3.6±3.0	20±1.2	4.4±1.2	73±4.1

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1211 Table 4. Summarise MDPSD (μm), FDPSD (μm) median particle diameter, Quartz (Qtz, wt %), feldspars
 1212 (Feld., wt %), clay mineral (Clay, wt %), carbonates (Carb., wt %), Na-salts (Na-S, wt %), Gypsum (Gp,
 1213 wt %), total Fe content (FeT, wt %), exchangeable Fe (FeA, wt %), dithionite Fe (Hematite and Goethite,
 1214 wt %), oxalate Fe (FeM, wt %) and structural Fe (FeS, wt %) for Mojave and Morocco crusts and Iceland
 1215 top sediments. NaN not a number.
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	MDPSD d(0.5)	FDPSD d(0.5)	Qtz	Feld.	Clay	Carb.	Na-S	Gp	FeT	FeA	FeD	FeM	FeS
Mojave	92	37	16	37	24	6.6	7.3	3.1	3.0	0.06	0.53	0.06	2.4
Morocco	113	37	48	9.4	17	22	7.0	0.64	3.6	0.07	1.1	NaN	2.4
Iceland	55	56	0.21	20	NaN	NaN	NaN	NaN	9.3	0.15	0.43	1.4	7.3

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