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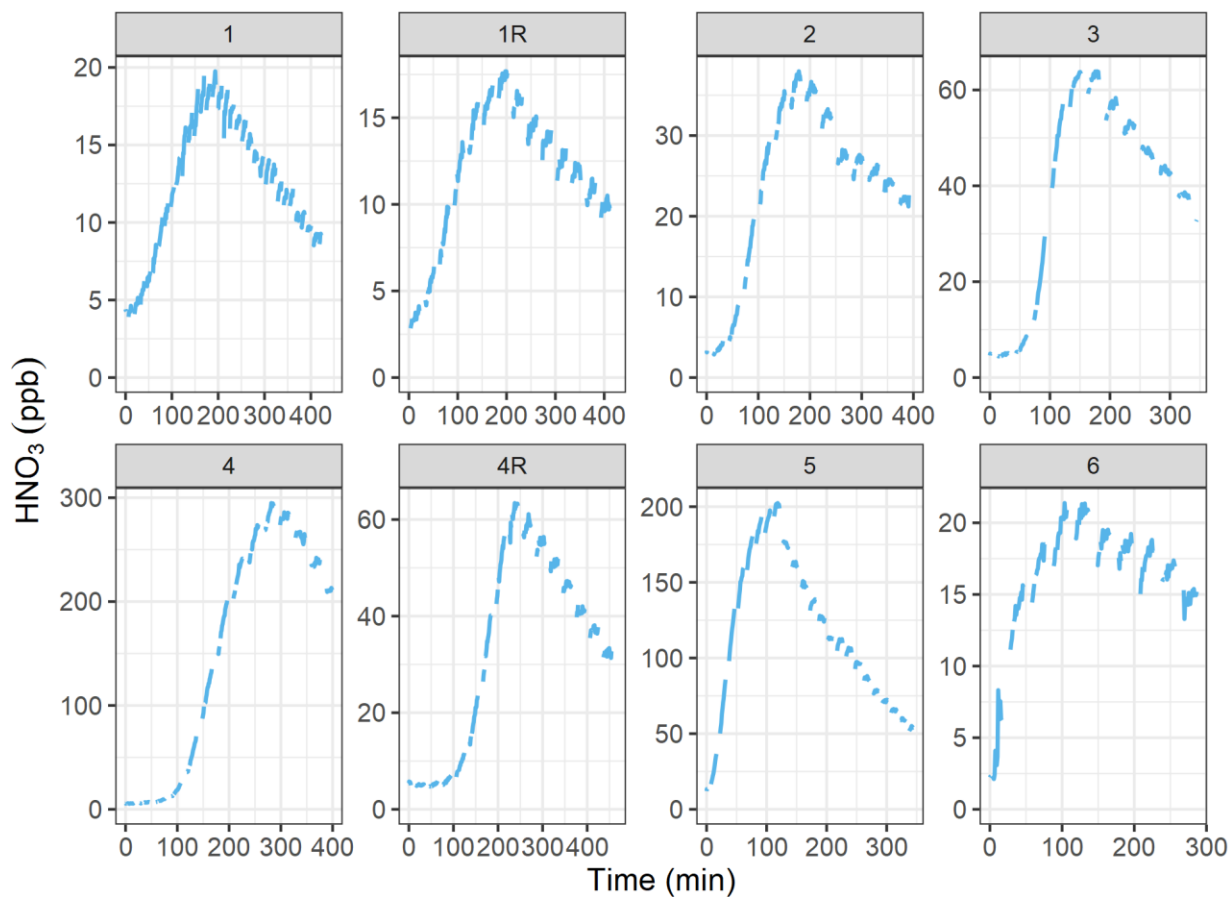


Fig. S1. The measured HNO₃ concentrations for the various conducted experiments. The measurements were made using CIMS. A substantial “chamber blank” was observed before the start (time = 0) of the experiments.

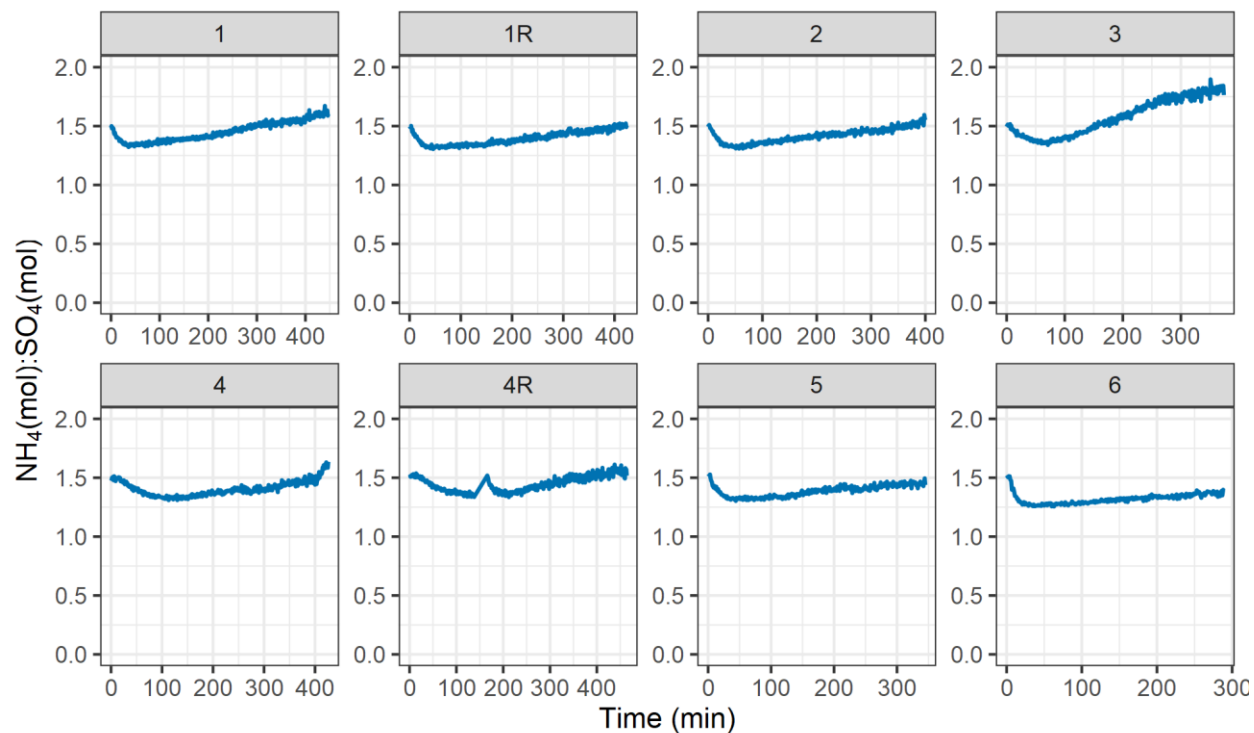


Fig. S2. The measured $\text{NH}_4(\text{mol}):\text{SO}_4(\text{mol})$ ratio for each conducted experiment from the HR-ToF-AMS. The ratios remained consistent and below 2 during the experiment indicative of minor contributions of pNO_3 derived from HNO_3 .

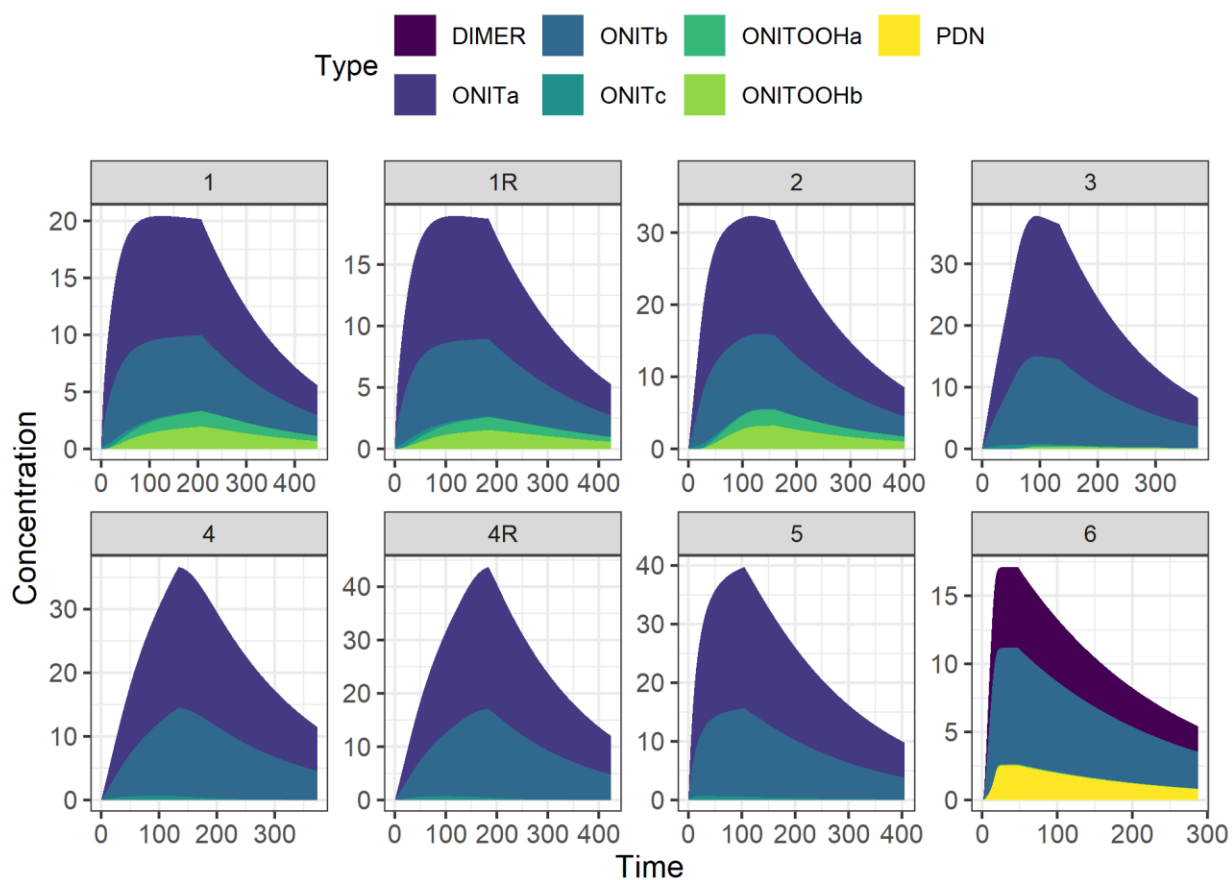


Fig. S3. The simulated types of organic nitrate and isomers formed during the various α -pinene/ NO_x oxidation experiments using the USC-API mechanism. The mechanism includes 2 pinene hydroxyl-nitrate (ONITa (tertiary) and ONITb (secondary)), pinene carbonyl nitrate (ONITc), pinene nitrooxy-hydroperoxide (ONITOOHa (tertiary) and ONITOOHb (secondary)), dimer (DIMER), and pinene dinitrate (PDN).

Table S1. Summary of the USC-API Species List.

Species	Description	SMILES
Inorganic		
CO	Carbon monoxide	C=O
CO2	Carbon dioxide	O=C=O
H2	Hydrogen	[H][H]
H2O2	Hydrogen peroxide	OO
HNO3	Nitric acid	O=N(=O)O
HO2	Hydroperoxy radical	[O]O
HO2NO2	Pernitric acid	OO[N+](=O)[O-]
HONO	Nitrous acid	O=N[O]
N2O5	Dinitrogen pentoxide	O=[N+](=O)O[N+](=O)[O-]
NO	Nitric oxide	[N]=O
NO2	Nitrogen dioxide	[O-][N+](=O)
NO3	Nitrate radical	[O-][N+](=O)O
O1D	Excited state oxygen atom O(¹ D)	[O]
O3	Ozone	O=O=O
O3P	Ground state oxygen atom O(₃ P)	[O]
OH	Hydroxyl radical	[OH]
Organic		
ACT	Acetone	CC(=O)C
API	A-pinene	C/C1=C/CC2CC1C2(C)C
APINAO	alpha-pinene alkoxy radical	OC1CC2CC(C1(C)[O])C2(C)C
APINAO2	tertiary (major) peroxy radical from APIN + OH + O2	[O]OC1(C)C(O)CC2CC1C2(C)C
APINAOOH	pinene-derived hydroxy hydroperoxide	OOC1(C)C(O)CC2CC1C2(C)C
APINBO	alpha-pinene alkoxy radical	[O]C1CC2CC(C1(C)O)C2(C)C
APINBO2	secondary (minor) peroxy radical from APIN + OH + O2	[O]OC1CC2CC(C1(C)O)C2(C)C
APINBOOH	pinene-derived hydroxy hydroperoxide	OOC1CC2CC(C1(C)O)C2(C)C
APINCO	alpha-pinene alkoxy radical	CC1=CCC(CC1O)C(C)(C)[O]
APINCO2	tertiary (minor) peroxy radical from APIN + OH + O2	[O]OC(C)(C)C1CC=C(C)C(O)C1
APINCOOH	pinene-derived hydroxy hydroperoxide	OOC(C)(C)C1CC=C(C)C(O)C1

DIMER	Dimers from nRO ₂ + RO ₂ reactions	CC3(C)C4CC(O[N+](=O)O)C(OOC1(C)C(O[N+](=O)O)CC2CC1C2(C)C)C3C4
MO2	Generic RO ₂	N/A
NAPINAO2	tertiary (major) peroxy radical from APIN + NO ₃ + O ₂ ;	[O]OC1CC2CC(C1(C)ON(=O)=O)C2(C)C
NAPINBO2	secondary (minor) peroxy radical from APIN + NO ₃ + O ₂ ;	[O]OC1(C)C(ON(=O)=O)CC2CC1C2(C)C
ONITa	Tertiary (major) alpha-pinene hydroxynitrate	O=N(=O)OC1(C)C(O)CC2CC1C2(C)C
ONITb	Secondary (minor) alpha-pinene hydroxynitrate	O=N(=O)OC1CC2CC(C1(C)O)C2(C)C
ONITc	Tertiary alpha-pinene carbonyl nitrate	O=N(=O)OC(C)(C)C1CC=C(C)C(O)C1
ONITOOHa	Tertiary alpha-pinene nitrate hydroperoxide	OOC1CC2CC(C1(C)ON(=O)=O)C2(C)C
ONITOOHb	Secondary alpha-pinene nitrate hydroperoxide	OOC1(C)C(ON(=O)=O)CC2CC1C2(C)C
P	Generic Product	N/A
PAN	Peroxyacetyl nitrate	CC(=O)OON(=O)=O
PDN	Pinene dinitrate	CC1(C)C2CC(O[N+](=O)O)C(C)(O[N+](=O)O)C1C2
PINAL	pinonaldehyde	CC(=O)C1CC(CC=O)C1(C)C
PINO3	pinonaldehyde-derived acyl peroxy radical	[O]OC(=O)CC1CC(C(=O)C)C1(C)C
PINPAN	C10 peroxyacyl nitrate	O=N(=O)OOC(=O)CC1CC(C(=O)C)C1(C)C
RO	Generic Aloyx Radical	N/A
RO3	Generic acyl peroxy radical	N/A

Table S2. Summary of the USC-API Mechanism.

Label	Reaction	Reaction Rate (<i>k</i>)
Inorganic Photolysis		
R001	$O_3 \rightarrow O_3P + O_2$	JO3P*
R002	$O_3 \rightarrow O1D + O_2$	JO1D*
R003	$H_2O_2 \rightarrow OH + OH$	JH2O2*
R004	$NO_2 \rightarrow O_3P + NO$	JNO2*
R005	$NO_3 \rightarrow O_2 + NO$	JNO3 NO*
R006	$NO_3 \rightarrow O_3P + NO_2$	JNO3 NO2*
R007	$HONO \rightarrow OH + NO$	JHONO*
R008	$HNO_3 \rightarrow OH + NO_2$	JHNO3*
R009	$HO_2NO_2 \rightarrow 0.2 OH + 0.2 NO_3 + 0.8 HO_2 + 0.8 NO_2$	JHO2NO2*
Inorganic Reactions		
R010	$O_3 + OH \rightarrow HO_2 + O_2$	$1.70E-12 * \exp(-940 / T)$
R011	$O_3 + HO_2 \rightarrow OH + O_2 + O_2$	$1.00E-14 * \exp(-490 / T)$
R012	$O_3 + NO \rightarrow NO_2 + O_2$	$1.40E-12 * \exp(-1310 / T)$
R013	$NO_2 + O_3 \rightarrow NO_3 + O_2$	$1.40E-13 * \exp(-2470 / T)$
R014	$O_3P + O_2 \rightarrow O_3$	$M * 5.60E-34 * (T / 300)^{-2.6} * 0.21 * M$
R015	$O_3P + O_3 \rightarrow O_2 + O_2$	$8.00E-12 * \exp(-2060 / T)$
R016	$O1D + O_2 \rightarrow O_3P + O_2$	$3.20E-11 * 0.21 * M$
R017	$O1D + N_2 \rightarrow O_3P + N_2$	$1.80E-11 * \exp(107 / T) * 0.78 * M$
R018	$O1D + H_2O \rightarrow OH + OH$	$2.20E-10 * H_2O$
R019	$OH + H_2 \rightarrow HO_2 + H_2O$	$7.70E-12 * \exp(-2100 / T)$
R020	$OH + HO_2 \rightarrow H_2O + O_2$	$4.80E-11 * \exp(250 / T)$
R021	$HO_2 + HO_2 \rightarrow H_2O_2 + O_2$	$2.20E-13 * \exp(600 / T) + 1.90E-33 * M * \exp(980 / T)$
R022	$HO_2 + HO_2 + H_2O \rightarrow H_2O_2 + H_2O + O_2$	$(3.08E-34 * \exp(2800 / T) + 2.59E-54 * M * \exp(3180 / T)) * H_2O$
R023	$H_2O_2 + OH \rightarrow HO_2 + H_2O$	$2.90E-12 * \exp(-160 / T)$
R024	$NO + O_3P \rightarrow NO_2$	K_O3P_NO**
R025	$NO + OH \rightarrow HONO$	K_OH_NO**
R026	$HO_2 + NO \rightarrow OH + NO_2$	$3.45E-12 * \exp(270 / T)$
R027	$HO_2 + NO \rightarrow HNO_3$	K_HO2_NO_HNO3**

R028	$\text{NO} + \text{NO} \rightarrow \text{NO}_2 + \text{NO}_2$	$3.30\text{E-}39 * \exp(530 / \text{T}) * 0.21 * \text{M}$
R029	$\text{HONO} + \text{OH} \rightarrow \text{NO}_2 + \text{H}_2\text{O}$	$2.50\text{E-}12 * \exp(260 / \text{T})$
R030	$\text{O}_3\text{P} + \text{NO}_2 \rightarrow \text{NO} + \text{O}_2$	$5.50\text{E-}12 * \exp(188 / \text{T})$
R031	$\text{O}_3\text{P} + \text{NO}_2 \rightarrow \text{NO}_3$	$\text{K_O}_3\text{P_NO}_2^{**}$
R032	$\text{OH} + \text{NO}_2 \rightarrow \text{HNO}_3$	$\text{K_OH_NO}_2\text{HONO}_2^{**}$
R033	$\text{OH} + \text{HNO}_3 \rightarrow \text{NO}_3 + \text{H}_2\text{O}$	K_OH_HNO_3^{**}
R034	$\text{OH} + \text{NO}_3 \rightarrow \text{HO}_2 + \text{NO}_2$	$2.00\text{E-}11$
R035	$\text{HO}_2 + \text{NO}_3 \rightarrow 0.7 \text{OH} + 0.7 \text{NO}_2 + 0.3 \text{HNO}_3$	$4.00\text{E-}12$
R036	$\text{NO} + \text{NO}_3 \rightarrow \text{NO}_2 + \text{NO}_2$	$1.80\text{E-}11 * \exp(110 / \text{T})$
R037	$\text{NO}_2 + \text{NO}_3 \rightarrow \text{NO} + \text{NO}_2 + \text{O}_2$	$4.50\text{E-}14 * \exp(-1260 / \text{T})$
R038	$\text{NO}_3 + \text{NO}_3 \rightarrow \text{NO}_2 + \text{NO}_2 + \text{O}_2$	$8.50\text{E-}13 * \exp(-2450 / \text{T})$
R039	$\text{NO}_2 + \text{NO}_3 \rightarrow \text{N}_2\text{O}_5$	$\text{K_NO}_2\text{NO}_3^{**}$
R040	$\text{N}_2\text{O}_5 \rightarrow \text{NO}_2 + \text{NO}_3$	$\text{K_N}_2\text{O}_5^{**}$
R041	$\text{N}_2\text{O}_5 + \text{H}_2\text{O} \rightarrow \text{HNO}_3 + \text{HNO}_3$	$2.50\text{E-}22 * \text{H}_2\text{O}$
R042	$\text{HO}_2 + \text{NO}_2 \rightarrow \text{HO}_2\text{NO}_2$	$\text{K_HO}_2\text{NO}_2^{**}$
R043	$\text{HO}_2\text{NO}_2 \rightarrow \text{HO}_2 + \text{NO}_2$	$\text{K_HO}_2\text{NO}_2^{**}$
R044	$\text{OH} + \text{HO}_2\text{NO}_2 \rightarrow \text{NO}_2 + \text{H}_2\text{O} + \text{O}_2$	$1.30\text{E-}12 * \exp(380 / \text{T})$
R045	$\text{OH} + \text{CO} \rightarrow \text{HO}_2 + \text{CO}_2$	K_OH_CO^{**}
API Oxidation		
R046	$\text{API} + \text{OH} \rightarrow 0.572 \text{APINAO}_2 + 0.353 \text{APINBO}_2 + 0.075 \text{APINCO}_2$	$1.2\text{e-}11 * \exp(440 / \text{T})$
R047	$\text{API} + \text{O}_3 \rightarrow \text{RO}_2 + \text{OH}$	$8.05\text{e-}16 * \exp(-640 / \text{T})$
R048	$\text{API} + \text{NO}_3 \rightarrow 0.65 \text{NAPINAO}_2 + 0.35 \text{NAPINBO}_2$	$1.2\text{e-}12 * \exp(490 / \text{T})$
APIO2 Rxns		
R049	$\text{APINAO}_2 + \text{NO} \rightarrow 0.23 \text{ONITa} + 0.77 \text{PINAL} + 0.77 \text{HO}_2 + 0.77 \text{NO}_2$	$2.70\text{e-}12 * \exp(360 / \text{T})$
R050	$\text{APINBO}_2 + \text{NO} \rightarrow 0.23 \text{ONITb} + 0.77 \text{PINAL} + 0.77 \text{HO}_2 + 0.77 \text{NO}_2$	$2.70\text{e-}12 * \exp(360 / \text{T})$
R051	$\text{APINCO}_2 + \text{NO} \rightarrow 0.125 \text{ONITc} + 0.875 \text{ACT} + 0.875 \text{P} + 0.875 \text{NO}_2$	$2.70\text{e-}12 * \exp(360 / \text{T})$
R052	$\text{APINAO}_2 + \text{HO}_2 \rightarrow \text{APINAOOH}$	$2.91\text{e-}13 * \exp(1300 / \text{T}) * 0.914$
R053	$\text{APINBO}_2 + \text{HO}_2 \rightarrow \text{APINBOOH}$	$2.91\text{e-}13 * \exp(1300 / \text{T}) * 0.914$
R054	$\text{APINCO}_2 + \text{HO}_2 \rightarrow \text{APINCOOH}$	$2.91\text{e-}13 * \exp(1300 / \text{T}) * 0.914$
R055	$\text{APINAO}_2 + \text{NO}_3 \rightarrow \text{APINAO} + \text{NO}_2$	$\text{KRO}_2\text{NO}_3^{**}$
R056	$\text{APINBO}_2 + \text{NO}_3 \rightarrow \text{APINBO} + \text{NO}_2$	$\text{KRO}_2\text{NO}_3^{**}$
R057	$\text{APINCO}_2 + \text{NO}_3 \rightarrow \text{APINCO} + \text{NO}_2$	$\text{KRO}_2\text{NO}_3^{**}$

PINAL Chemistry		
R058	PINAL + hv → RO2 + CO + HO2	J.JPINAL*
R059	PINAL + OH → PINO3 + RO2	5.2e-12 * exp(600 / T)
R060	PINAL + NO3 → PINO3 + HNO3	2.00E-14
R061	PINO3 + NO → RO2 + NO2	7.5E-12 * exp(290 / T)
R062	PINO3 + NO3 → RO2 + NO2	4.00E-12
R063	PINO3 + NO2 → PINPAN	K_ACO3_NO2**
R064	PINO3 + HO2 → 0.44 RO2 + 0.44 OH + 0.15 O3 + 0.55 P	5.20E-13 * exp(980 / T)
R065	PINPAN → PINO3 + NO2	K_PAN**
General RO2 Rxns		
R066	MO2 + NO → NO2 + RO	2.7e-12 * exp(360 / T)
R067	MO2 + HO2 → P	2.91e-13 * exp(1300 / T)
R068	MO2 + NO3 → RO + NO2	2.30E-12
R069	MO2 + MO2 → P	1.30E-12
R070	MO2 + PINO3 → P	5.00E-12
ACT Chemistry		
R071	ACT + OH → RO2	1.39E-13 + 3.72E-11 exp(-2044 / T)
R072	ACT + hv → RO3 + RO2	JACT*
PAN Chemistry		
R073	RO3 + NO2 → PAN	K_ACO3_NO2**
R074	PAN → RO3 + NO2	K_PAN**
R075	PAN + hv → RO3 + NO2	JPAN1*
R076	PAN + hv → MO2 + NO3 + CO2	JPAN2*
RO3 Chemistry		
R077	RO3 + NO → RO2 + NO2	8.10E-12 * exp(270 / T)
R078	RO3 + HO2 → P	4.30E-13 * exp(1040 / T)
R079	RO3 + RO2 → HO2 + P	2.00E-12 * exp(500 / T)
R080	RO3 + RO3 → P	2.50E-12 * exp(500 / T)
APIOOH Chemistry		
R081	APINAOOH + hv → P + OH	J.JAPINAOOH*
R082	APINAOOH + OH → RO2	1.83E-11
R083	APINBOOH + hv → P + OH	J.JAPINAOOH*
R084	APINBOOH + OH → P + OH	3.28E-11
R085	APINCOOH + hv → P + OH	J.JAPINAOOH*
R086	APINCOOH + OH → RO2	1.03E-10
NAPINO2 Chemistry		
R087	NAPINAO2 + HO2 → 0.37 ONITa + 0.63 PINAL + 0.63 HO2 + 0.63 OH	2.66e-13 * exp(1300 / T)

R088	$\text{NAPINAO}_2 + \text{NO} \rightarrow 0.9 \text{ PINAL} + 1.9 \text{ NO}_2 + 0.1 \text{ ONITa}$	$2.55\text{e-}12 * \exp(380 / \text{T})$
R089	$\text{NAPINAO}_2 + \text{NO}_3 \rightarrow 0.1 \text{ PDN} + 0.9 \text{ PINAL} + 1.8 \text{ NO}_2$	2.30E-12
R090	$\text{NAPINAO}_2 + \text{NAPINAO}_2 \rightarrow 0.16 \text{ DIMER} + 1.68 \text{ PINAL} + 1.68 \text{ NO}_2$	1.00E-14
R091	$\text{NAPINAO}_2 + \text{NAPINBO}_2 \rightarrow 0.34 \text{ ONITb} + 0.08 \text{ DIMER} + 1.34 \text{ PINAL} + 0.08 \text{ DIMER} + 1.34 \text{ NO}_2$	1.00E-14
R092	$\text{NAPINBO}_2 + \text{HO}_2 \rightarrow \text{ONITOOHb}$	$2.66\text{e-}13 * \exp(1300 / \text{T})$
R093	$\text{NAPINBO}_2 + \text{NO} \rightarrow 0.9 \text{ PINAL} + 1.9 \text{ NO}_2 + 0.1 \text{ ONITb}$	$2.55\text{e-}12 * \exp(380 / \text{T})$
R094	$\text{NAPINBO}_2 + \text{NO}_3 \rightarrow 0.1 \text{ PDN} + 0.9 \text{ PINAL} + 1.8 \text{ NO}_2$	2.30E-12
R095	$\text{NAPINBO}_2 + \text{NAPINBO}_2 \rightarrow 0.68 \text{ ONITb} + 0.16 \text{ DIMER} + \text{PINAL} + \text{NO}_2$	1.00E-14
RO2 Cross Reactions		
R096	$\text{APINAO}_2 + \text{APINAO}_2 \rightarrow 2 \text{ PINAL} + 2 \text{ HO}_2$	1.00E-14
R097	$\text{APINAO}_2 + \text{APINBO}_2 \rightarrow 2 \text{ PINAL} + 2 \text{ HO}_2$	1.00E-14
R098	$\text{APINAO}_2 + \text{APINCO}_2 \rightarrow 2 \text{ PINAL} + 2 \text{ HO}_2$	1.00E-14
R099	$\text{APINBO}_2 + \text{APINBO}_2 \rightarrow 2 \text{ PINAL} + 2 \text{ HO}_2$	1.00E-14
R0100	$\text{APINBO}_2 + \text{APINCO}_2 \rightarrow 2 \text{ PINAL} + 2 \text{ HO}_2$	1.00E-14
R101	$\text{APINCO}_2 + \text{APINCO}_2 \rightarrow 2 \text{ PINAL} + 2 \text{ HO}_2$	1.00E-14
R102	$\text{RO} + \text{O}_2 \rightarrow \text{MO}_2$	$1\text{E-}11 * \text{M} * 0.21$
ONIT Reactions		
R103	$\text{ONITa} + \text{OH} \rightarrow \text{PINAL} + \text{NO}_2$	5.50E-12
R104	$\text{ONITb} + \text{OH} \rightarrow \text{P} + \text{NO}_2$	3.64E-12
R105	$\text{ONITc} + \text{OH} \rightarrow \text{ACT} + \text{P} + \text{NO}_2$	9.87E-11
R106	$\text{ONITc} + \text{h}\nu \rightarrow \text{ACT} + \text{RO}_2 + \text{NO}_2$	JONIT*

*The photolysis frequencies (J) were calculated inline using the F0AM Box model based on the MCMv3.2 cross-sections

**The complex reaction rate constants (K) were from the MCMv3.2

Table S3. Summary of the organic nitrate hydrolysis reactions incorporated into the USC-API (“USC-API-Hydro”) mechanism that was utilized for a sensitivity simulation. Note: the DIMER and pinene dinitrate were not included in the hydrolysis scheme since the nighttime oxidation experiment exhibited low organic nitrate filter hydrolysis efficiency.

Label	Reaction	Rate (<i>k</i>)
Hydro-01	ONITa=HNO3+P	2.78E-04
Hydro-02	ONITb=HNO3+P	2.78E-04
Hydro-03	ONITc=HNO3+P	2.78E-04
Hydro-04	ONITOOHa=HNO3+P	2.78E-04
Hydro-05	ONITOOHb=HNO3+P	2.78E-04

Table S4. Summary of the measured NO_y isotope data ($\Delta^{17}\text{O}$, $\delta^{18}\text{O}$, and $\delta^{15}\text{N}$) and uncertainties ($\pm\sigma$) for the various conducted experiments.

Exp	NO ₂ (‰)			HNO ₃ (‰)			pNO ₃ (‰)		
	$\delta^{15}\text{N}$	$\Delta^{17}\text{O}$	$\delta^{18}\text{O}$	$\delta^{15}\text{N}$	$\Delta^{17}\text{O}$	$\delta^{18}\text{O}$	$\delta^{15}\text{N}$	$\Delta^{17}\text{O}$	$\delta^{18}\text{O}$
1	-67.8±1.7	13.4±0.7	43.6±1.8	-37.0±1.7	15.6±0.7	49.0±1.8	-79.6±2.4	5.2±0.8	21.7±0.7
1R	-63.3±1.7	10.8±0.7	45.3±1.8	-37.7±1.7	16.6±0.7	49.0±1.8	-76.3±2.6	5.6±0.8	22.3±0.8
2	-68.3±1.7	18.9±0.7	65.8±1.8	-34.3±1.7	13.4±0.7	43.7±1.8	-78.5±1.1	9.2±0.7	29.2±0.6
3	-70.2±1.7	32.4±0.7	92.6±1.8	-32.0±1.7	16.5±0.7	49.5±1.8	-73.1±1.8	11.1±0.8	37.7±0.8
	-71.8±1.7	32.6±0.7	86.6±1.8	-20.1±1.7	16.3±0.7	48.5±1.8			
	-68.7±1.7	31.2±0.7	80.6±1.8	-19.1±1.7	16.7±0.7	49.3±1.8			
4	-67.8±1.7	40.8±0.7	108.7±1.8	-32.8±1.7	18.1±0.7	48.1±1.8	-90.3±1.9	12.0±0.7	42.4±0.8
	-71.0±1.7	39.3±0.7	107.8±1.8	-47.0±1.7	21.1±0.7	60.2±1.8			
	-71.4±1.7	38.6±0.7	112.1±1.8	-49.7±1.7	21.7±0.7	61.3±1.8			
4R	-67.6±1.7	39.6±0.7	107.8±1.8	-29.6±1.7	17.8±0.7	53.9±1.8	-87.5±4.1	11.8±0.9	40.4±1.4
	-62.9±1.7	39.4±0.7	115.3±1.8	-30.6±1.7	18.1±0.7	55.0±1.8			
	-68.2±1.7	40.7±0.7	108.5±1.8	-26.8±1.7	17.9±0.7	52.2±1.8			
	-68.7±1.7	41.2±0.7	108.7±1.8	-26.8±1.7	18.4±0.7	54.7±1.8			
5	-5.5±1.7	39.1±0.7	103.0±1.8	-8.0±1.7	14.7±0.7	48.5±1.8	-22.5±1.0	15.6±0.9	51.0±1.4
	-6.0±1.7	36.9±0.7	100.8±1.8	-6.3±1.7	15.7±0.7	48.3±1.8			
	-6.0±1.7	37.0±0.7	98.0±1.8	-4.0±1.7	15.3±0.7	48.7±1.8			
	-7.1±1.7	38.1±0.7	101.0±1.8	-4.7±1.7	15.2±0.7	46.8±1.8			
6	-46.0±1.7	7.0±0.7	31.2±1.8	-24.8±1.7	14.5±0.7	45.4±1.8	NA	NA	NA
	-45.8±1.7	7.3±0.7	31.7±1.8	-21.6±1.7	15.3±0.7	47.0±1.8			
	-45.5±1.7	7.4±0.7	32.7±1.8	-24.4±1.7	15.3±0.7	45.8±1.8			