1 Complementary aerosol mass spectrometry elucidates sources

2 of wintertime sub-micron particle pollution in Fairbanks,

3 Alaska, during ALPACA 2022

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- 29 Fairbanks, Alaska, is a sub-arctic city that frequently suffers from non-attainment of national air quality
- 30 standards in the wintertime due to the coincidence of weak atmospheric dispersion and increased local

emissions. However, significant uncertainties exist about aerosol sources, formation, and chemical processes during cold winter periods. We identified and quantified various local sources using source apportionment approach and could determine the their chemical composition, size distribution, and concentrations of atmospheric sub-micron non-refractory particulate matter (NR-PM₁) emitted from them and quantify their sources in the urban centre of Fairbanks. As part of the Alaskan Layered Pollution and Chemical Analysis (ALPACA) campaign, we deployed a Chemical Analysis of Aerosol Online (CHARON) inlet coupled with a proton transfer reaction - time of flight mass spectrometer (PTR-ToF MS) and an Aerodyne high-resolution aerosol mass spectrometer (AMS) to measure organic aerosol (OA) and NR-PM₁, respectively. We deployed Positive Matrix Factorisation (PMF) analysis for source identification of the NR-PM₁. The PTR_{CHARON} factorisation identified four residential heating sources, with oil combustion accounting for $16 \pm 9\%$ and wood combustion contributing to $47 \pm 20\%$ of the OA. The analysis could further differentiate between hardwood and softwood combustion. In contrast, The AMS analysis identified three primary factors: a single biomass burning, a hydrocarbonlike and cooking factors which together accounted for 28, 38 and 11 % of the total OA, respectively. Additionally, a combined organic and inorganic PMF analysis revealed two further factors: one enriched in nitrates and another rich in sulphates of organic and inorganic origin. The PTR_{CHARON} factorization could identify four primary sources from residential heating - one from oil combustion and three wood combustion, categorised as -types (low temperature, softwood, and hardwood). Collectively, all residential heating factors accounted for 79% of the total OA_ eCooking and road transport were also recognised as primary contributors to overall emission profile provided by PTR_{CHARON}. All PMF could apportion a single secondary organic fraction. These results evidence the complementarity of the two instruments and their ability in describing the complex chemical composition of PM₁ and the related sources, These results demonstrate that This work further demonstrates the capability of PTR_{CHARON} to provide both generates qualitative and quantitative information offering with enhanced a comprehensive understanding of the organic aerosol sources. resolution of organic aerosol sources. Such When combined with suitable complementary instruments like the AMS, such evidence based insights into the sources of sub-micron aerosol pollution can assist environmental regulators and citizen efforts to improve for the improvement in air quality in Fairbanks and in the wider fast-urbanising regional sub-Arctic areas.

Keywords PM₁, mass spectrometry, source apportionment, <u>organic</u> and inorganic aerosol, Fairbanks,
 Arctic <u>cities</u>, air quality, CHARON PTR-ToF MS, HR-ToF AMS, <u>proton transfer reaction</u>residential
 heating, wood combustion

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1 Introduction

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Extremely cold urban regions of the Earth, such as in the Arctic, experience poor dispersion of atmospheric pollution, especially during the wintertime, when the unique meteorological characteristics, such as extremely low solar radiation and strong radiative cooling at the surface, are coupled with enhanced local anthropogenic emissions from heating, industry, and transport. A good example is the sub-arctic city of Fairbanks, Alaska, where air quality standards are frequently violated during the winters with concentrations of fine particulate matter (i.e., with aerodynamic diameters smaller than 2.5 µm; PM_{2.5}) exceeding the 24-h limit of 35 µg/m³ defined by EPA's National Ambient Air Quality Standards (Dunleavy and Brune, 2020; Epa, n.d.). Not only is Fairbanks one of the cities with the most polluted wintertime air in the US, but it has also been declared a 'moderate non-attainment area' since 2009, and due to the persistence of the problem, it was reclassified as a 'severe nonattainment area' in 2017. Increased local anthropogenic emissions and poor atmospheric dispersion due to strong surface-based temperature inversions (> 0.5°C/m in the lowest 10 m above the ground) are major causes of wintertime pollution in the region (Tran and Mölders, 2011; Mayfield and Fochesatto, 2013). Many research studies have recognised biomass combustion as the major source of aerosol in Fairbanks (Ward et al., 2012; Wang and Hopke, 2014; Kotchenruther, 2016; Ye and Wang, 2020; Haque et al., 2021) that drives overall PM_{2.5} concentrations across the city during strong temperature inversion conditions (Robinson et al., 2023). A comprehensive study covering three winters from 2008–2011 apportioned 60-80% of PM_{2.5} mass at four locations in Fairbanks to emissions from residential wood stoves, open burning of biomass, outdoor boilers, and other solid-fuel combustion. (Ward et al., 2012). Source apportionment of year-round PM_{2.5} in the past two decades [2008–2009 (Haque et al., 2021), 2005–2012 (Wang and Hopke, 2014), 2009–2014 (Kotchenruther, 2016), and 2013–2019 (Ye and Wang, 2020)] also revealed woodsmoke as a major contributor to PM_{2.5} loads [47.5% (Haque et al., 2021), 40.5% (Wang and Hopke, 2014), ~52% (Kotchenruther, 2016), and ~19% (Ye and Wang, 2020)]. Wildfire activity and residential wood combustion are the major sources in summer and winter, respectively. The persistent role of wood-burning emissions in shaping the air quality of Fairbanks during winters triggered the implementation of a two-stage burn restriction in 2015 by the Alaska Department of Environmental Conservation (ADEC). The ADEC advisories restricted the operation of solid-fuel heating devices and required alternative heat sources to be used on days with weak atmospheric dispersion and $PM_{2.5} > 25 \mu g/m^3$ are observed or forecasted (Fye et al., 2009; Czarnecki, 2017; Jentgen, 2022). Sulphate has been observed to be the second largest component of PM_{2.5} mass in Fairbanks (Ward et al., 2012; Wang and Hopke, 2014), forming ~33% of the annual average PM_{2.5} mass (Ye and Wang, 2020). Isotope analyses have revealed 62% of this PM_{2.5} sulphate to be primary (e.g., from residential heating oil combustion) during the winters (Moon et al., 2023).

The aforementioned studies on air quality in Fairbanks have focused on PM_{2.5} even though PM₁ has been recognised as the major cause of negative health effects (Wang et al., 2015; Mainka and Zajusz-Zubek, 2019) due to its capability to spread deeper into the respiratory or cardiovascular systems (Meng et al., 2013; Liu et al., 2013; Chen et al., 2017). Currently, PM₁ concentrations are not regulated globally, but its strong contribution to atmospheric PM_{2.5} loads and impacts has implications for the attainment of the latter's regulatory limits. efforts to monitor PM₁ are surprisingly scarce, even in-a 'non-attaining' eitycities, such as Fairbanks, underscoring the need for a better characterisation of sub-micron aerosols to understand local sourcesemissions, atmospheric processes, chemical composition and ultimately to inform public health and support policy decisions. even though characterising the chemical composition of sub-micron atmospheric aerosol and capturing the variation in their mass concentrations is key to unravelling the complexities of local emissions and their transformation in Fairbanks and, most importantly, to underscore the health and policy implications of atmospheric emissions.

Mass spectrometric techniques have advanced over the years, featuring greater mass accurancy, resolving power, and sensitivity. For instance, the Aerodyne high-resolution time-of-flight aerosol mass spectrometer (HR-ToF AMS; called AMS from hereon) is a well-established method for quantification of non-refractory NR-PM₁. Aerosol vapourisation at high temperatures and electron ionisation result in substantive molecular decomposition, facilitating quantification with high time resolution (Decarlo et al., 2006), but at the cost of molecular-level information resolution. The lack of molecular level information provided by AMS. This limitation has encourageds the rise of complementary techniques to better understand both primary aerosol sources and secondary aerosol formation. For instance, extractive electrospray ionisation (EESI)-ToF MS has been successfully deployed in Beijing (Tong et al., 2021) and in Zurich to resolve multiple OA sources (Stefenelli et al., 2019a; Oi et al., 2019). Although the instrument the EESI ToF MS provides molecular-level information, its quantitative response is variable and selective for polar species, preventing its independent application for ambient measurements. Other measurement methods, such as thermal desorption aerosol GC/MS flame ionisation detector (TAG)(Williams et al., 2006) and filter inlet for gases and aerosols chemical ionisation (FIGAERO-CIMS)-ToF MS (Lopez-Hilfiker et al., 2014), similarly offer better chemical resolution than the AMS, but a lower temporal resolution. Semi-continuous measurements, such as those from TAG and FIGAERO-CIMS, may not capture the rapid variation in sources. To improve the analysis of sub-micron OA in ambient air, a novel inlet system called the chemical analysis for aerosol online (CHARON) was developed to collect real-time measurements (Eichler et al., 2015). This inlet minimises thermal and ionisation-induced fragmentation of sampled OA by employing a low-temperature vapourisation system (150°C ≤) and coupling with a relatively softer and less selective-ionisation method, such as the proton-transfer reaction (PTR). The CHARON PTR-ToF MS (called PTR_{CHARON} from hereon) was first-successfully used for the characterisation of OA from ship

exhaust (Eichler et al., 2017), ambient OA in Lyon, France, and Valencia, Spain, and OA source apportionment in Innsbruck, Austria (MüŁler et al., 2017). Recently, the inlet was used to quantify individual compounds in laboratory-generated secondary organic aerosol (Lannuque et al., 2023) and complex mixtures, such as vehicular gasoline emissions and atmospheric organic matter (Piel et al., 2019; Kostenidou et al., 2024). Additionally, the analyser (commonly a PTR ToF MS) coupled to the CHARON inlet The system can measure gas-phase species as well, creating the opportunity to explore VOC precursor emissions or phase partitioning (Peng et al., 2023; Gkatzelis et al., 2018). Overall, PTR_{CHARON} and AMS are complementary techniques; that provide robust qualitative and quantitative infio. The former features molecular level information of the OA faction in contrast to the AMS, but has limited ability to analyse detect particles smaller thanbelow 150 nm (Eichler et al., 2015); the latter instrument—covers smaller particle size range (i.e., > 60 nm) and detects inorganic components too (Decarlo et al., 2006). Together they provide an excellent combination of real-time and quantitative data on atmospheric ambient aerosol.

The <u>detailed composition</u> of sub-micron aerosol pollution in the <u>deterioration of air quality</u> in Fairbanks —and other anthropogenically influenced regions of the sub-wider Arctic regions – is still not wellunderstood. To address this issue, we deployed a PTR_{CHARON} and an AMS in the urban centre of Fairbanks during the ALPACA (Alaskan Layered Pollution and Chemical Analysis) campaign as part of the French CASPA (Climate-Relevant Aerosol Sources and Processes in the Arctic) project in January–February 2022 (Simpson et al. 2024). We aimed to determine the composition, concentrations, and sources of atmospheric NR-PM₁. In this paper, we present: (i) an intercomparison of the performance of the two instruments focusing on OA quantitation, (ii) the identification of major OA sources in Fairbanks and their variation during the field campaign, and (iii) the source apportionment of organic and inorganic aerosol (e.g., ammonium, nitrate, and sulphursulphate). These findings highlight the synergistic benefits of combining multiple analytical techniques and emphasise how soft ionisation mass spectroscopic methods enhance molecular-level insights into particulate organic carbon. This integrated approach advances our understanding of the complex composition of particulate matter, offering valuable contributions to environmental characterisation and source apportionment studies. The results obtained here demonstrates the capability of technique a m/\Delta m of ~5000good mass resolution, combined with the soft ionisation of the PTR_{CHARON}, provide significant to capabilities for both molecular level qualitative and quantitative data, and allowings a better better ultimately contributing to a better understanding of the complex composition of particulate organic carbonand its related sources.

2 Methodology

2.1 Field campaign

- 171 The data presented in this study were collected during the ALPACA campaign in Fairbanks, Alaska,
- US from January 20 to February 26, 2022. ALPACA is an international collaborative field experiment
- to understand sources of outdoor and indoor air pollution in the cold and dark conditions of Fairbanks'
- winter. The scientific objectives and broad preliminary findings of the experiment were recently
- 175 reviewed (Simpson et al. 2024). All instruments used for this study were housed in a trailer parked at
- the Community and Technical College (CTC) of the University of Alaska, Fairbanks (64.84064°N,
- 177 147.72677°W; 136 m above sea level). The CTC is in the urban core of Fairbanks, close (within 40 m)
- to a major downtown road and parking area (Simpson et al. 2024); the west of this locality is dominated
- by residential activities, while the north and east have commercial activity.

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- 181 The trailer was equipped with a suite of particle counters and mass spectrometers with high temporal
- resolutions (varying from 10 seconds to 2 minutes). A scanning mobility particle sizer (SMPS) and a
- multi-angle absorption photometer (MAAP) were utilised to measure the distribution of particles sized
- 184 15.1 to 661.2 nm and black carbon concentrations, respectively. A separate inlet was used for
- PM₁/PM_{2.5}/PM₁₀ measurements conducted with a commercial optical particle counter (model OPC
- 1.109, Grimm Aerosol Technik) at a time resolution of 1 min. Two mass spectrometers, PTR_{CHARON}
- 187 (150-1000 nm) and AMS (60-700-1000 nm), were connected to the same inlet that sampled air at 3.5
- meters above ground level through a short (≈ 1 m) stainless tube with a 1/2" outer diameter extending
- through the trailer roof. A HEPA filter was placed upstream of the inlet <u>for an hour</u> at regular intervals
- 190 (twice a week) to measure the instrumental background. Additionally, meteorological data, including
- ambient temperatures at 3 and 23 m; wind speed and direction; and trace gases, namely CO, SO₂, O₃,
- 192 NO and NO₂, were recorded as described in a previous study associated with the campaign (Cesler-
- 193 Maloney et al., 2022).

2.2 Instrumentation

2.2.1 PTR-ToF MS: Operation operation and data processing

- The OA was quantified with a PTR-ToF MS (PTR-TOF 6000 X2, Ionicon Analytik GmbH, Austria)
- 197 coupled to a CHARON inlet in near real-time at 20-see temporal resolution, i.e., the PTR_{CHARON}. The
- 198 CHARON inlet has been described in detail by Eichler et al. (Eichler et al., 2015) and its applications
- were further evaluated and improved in subsequent studies (Müller et al., 2017; Leglise et al., 2019;
- Müller et al., 2019; Piel et al., 2019; Peng et al., 2023). Here, the PTR-ToF MS was configured to

alternate between direct-sampling of ambientof air to measure VOCs for 15 minutes (not included in the current study) and sampling of particulate matter through the CHARON inlet for 45 minutes. The instrument was operated at a low E/N of 65 Td (i.e., drift voltage/pressure; pressure, temperature, and voltage of the drift tube were set at 2.6 mbar, 120°C, and 265 V) and in RF mode for optimal sensitivity. The thermodesorber in the CHARON inlet was operated at 150°C and 8 mbar; this combination of moderate temperature with low pressure expands the range of detection to include ELVOCs as well (Piel et al. 2021). Raw data was obtained as described in **Section S1** and pre-processed with the Ionicon Data Analyzer (IDA, version 1.0.0.2), followed by post-processing (i.e., background subtraction, conversion of raw signal to mixing ratios, temporal averaging, PMF input generation) with an in-house data processing tool, PeTeR Toolkit (version 6.0; Igor 6.37). The error matrix was also calculated by PeTeR using uncertainties in ion counts and background signals. Among the resolved 1118 ions resolved spanning the range of m/z 50–425, only 336 were retained above the S/N with m/z 50–425, and 318 ions could be given a molecular formula based on the criteria described in Section S2. PTR ToF MS records raw signals in counts per second (cps) that were converted to mixing ratios according to the molecular identity determined for the detected ions and their protonation efficiencies (further details in Section S1). For comparison with the AMS, mixing ratios were converted to mass concentrations, i.e., µg/m³, using Equation S2. Mass concentrations calculated for the PTR_{CHARON} require a critical correction for the enrichment of sampled OA in the aerodynamic lens of the CHARON inlet (Eichler et al., 2015; Müller et al., 2017); further details are provided in Section S3. Total (or bulk) OA at a given point in time was the sum of mass concentrations of all ions, which was corrected for fragmentation using a previously reported method (Leglise et al., 2019), which increased the total OA mass concentrations by 17%.

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Species with m/z > 50 (the largest m/z detected above the S/N was 425) were retained for PMF of OA₂-Smaller as molecules of between m/z 18–50 were presented in low concentrations and; they are expected to be too volatile to be present in OA and were likely detected by PTR_{CHARON} as artefacts from the denuder function. Time series were averaged to 2 minutes (from 20 seconds) and two matrices ($m/z \times t$ time points) were extracted: (i) ion concentrations and (ii) their measurement uncertainties, using PeTeR version 6.0 in Igor 6.37. The final matrices — after removing empty rows and columns—had the following dimensions 336 × 17,986. Where required, ion intensities (in either ppb or $\mu g/m^3$) were normalised to the sum of all measured intensities.

2.2.2 HR-ToF-AMS: oOperation and data processing of the AMS

NR-PM₁ was were-monitored with 1-minute time resolution by a high-resolution time-of-flight mass spectrometer (AMS) -(Aerodyne Research Inc., Billerica, USA), extensively with spectral acquisition

at 1 minute intervals. The instrument has been described by previously (Decarlo et al., (2006) and ; Canagaratna et al., (2007). Briefly, ambient particles are sampled through a critical orifice, focused into a narrowed beam by an aerodynamic lens, accelerated toward a standard vapouriser heated element at (600°C) for flash vapourisation, and then ionised by electron impact (70 eV at 10⁻⁷ torr). Finally, the ions are analysed by a time-of-flight mass spectrometer. Standard calibrations were performed using 300 nm size-selected dried ammonium nitrate and ammonium sulphate particles at the beginning and the end of the campaign. Nitrate-equivalent values of sample mass concentrations were converted by applying relative ionisation efficiencies (RIEs) for organics, nitrates, ammonium, sulphursulphate, and chlorides (1.4, 1.1, 3.15, 1.93, and 1.3, respectively). For quantitative purposes, the eCollection efficiency (CE) of particles must be considered as strongly viscous particles in the sampled air are prone to bouncing off the vapouriserhas been calculated in PIKA using , thereby suffering from reduced detection. We used the time series of the —composition-dependent CE (CDCE) generated by PIKA following using a previously reported algorithm (Middlebrook et al., (2012) method. The calculated CE values, which ranged from 1.00 to 0.35.

Data was averaged to 2 minutes and extracted as concentration and measurement uncertainty matrices ($m/z \times time$ points) using SQUIRREL version 1.65 and PIKA version 1.25 in Igor 8.04. Separate matrices (and subsequently PMF) were prepared for organic only (abbreviated AMS_{org}) and by combining organic and inorganic species (abbreviated AMS_{org+inorg}). The inorganic species included in the analyses were nitrates (m/z 30, NO⁺ and 46, NO₂⁺), sulphur-sulphate (m/z 48, SO⁺; 64, SO₂⁺; 80, SO₃⁺; 81, HSO₃⁺; and 98, H₂SO₄⁺), ammonium (m/z 15, NH⁺; 16, NH₂⁺; and 17, NH₃⁺), and chlorides (m/z 35, Cl⁺ and 36, HCl⁺). Error matrices were calculated by PIKA based on uncertainty in ion counts, background signal, air beam correction, and electronic noise (Sueper, 2014). Atomic O/C and H/C ratios were calculated based on established methods (Aiken et al., 2007; Aiken et al., 2008; Canagaratna et al., 2015). Where needed for comparison with the PTR_{CHARON}, mass concentrations of PAHs were estimated from fragments as described previously (Herring et al., 2015), and levoglucosan was estimated as detailed in **Section S4**.

Species with m/z 12–120 were retained for PMF in this study, excluding important PAHs detected up to m/z 252; such PAHs were used as external tracers for factor identification. All PAHs were included in total OA quantification and associated comparisons. This exclusion is expected to cause underestimation below the 2% (by <2%) of the mass of some factors, particularly HOA (hydrocarbon-like organic aerosol) and BBOA (biomass-burning organic aerosol). After removing empty rows and eolumns, Final matrices from AMS_{org} and AMS_{org+inorg} analyses had the following dimensions: 193 × 24,762 and 205 × 24,762, respectively.

2.3 Source apportionment: Positive matrix factorisation

271 Source apportionment was performed using a PMF implemented in the multilinear engine (ME-2) 272 (Paatero, 1997a, 1999). The PMF was configured and analysed using the SoFi (Source Finder) Pro 273 interface (Canonaco et al., 2013) (version 8.4.1.9.1; Igor 8.04). PMF is a descriptive mathematical 274 algorithm that describes the input data, i.e. measurements of several variables collected over time (here, 275 $m/z \times \text{sampling time points}$), as a linear combination of factors that have constant mass spectra 276 associated with temporally varying concentrations of the spectral constituents (Paatero, 1997b; Paatero 277 and Tapper, 1994); each of the factors is representative of an emission source. The mathematical 278 expressions and functions of the PMF algorithm have been exhaustively detailed in previous studies 279 (e.g., refs. (Tong et al., 2021; Stefenelli et al., 2019a; Chen et al., 2022; Chazeau et al., 2022), etc.). 280 Below we summarise the user-defined configurations applied in SoFi Pro to optimise the PMF of our 281 datasets, PTR_{CHARON}, AMS_{org}, and AMS_{org+inorg}. The results were compared in terms of identified sources 282 and the mass of OA (or total NR-PM₁) apportioned to each source.

2.3.1. General methodology for PMF analysis

Preliminary PMF was performed without using a priori information, i.e., the so-called unconstrained factorisation, to understand the dataexplore the structure of the dataset, potential factor variability, preliminary source contributions, and guide the selection of an optimal solution before applying constraints. These unconstrained trials explored tested We considered solutions ranging from 3 three to 13 factors, applying a Cell-wise, step-wise, cell-wise-down-weighting was applied approach; whereby variables with S/N < 0.2 ("bad" variables) were down-weighted by a factor of 10, while those with Θ 0.2 < S/N < 2 ("weak" variables) were down-weighted by a factor of $\frac{10 \text{ and-}2}{C}$, respectively (Paatero and Hopke, 2003; Ulbrich et al., 2009). Upon establishing some that primary factors, e.g., such as cooking and biomass burning, which could were successfully identified readily be factorised in unconstrained trials, we narrowed explored only a subset of the range of possible solutions by applying the a-value approach, which allows for directing the PMF toward meaningful solutions with the a-value approach. For this approach, the user can improve factorisation results by constraining the PMF with external data, if when available (Canonaco et al., 2013; Paatero, 1999). For instance, a factor profile from a PMF trial in the same experiment, a time series from an external tracer from the same campaign, or a well-established factor profile for a source from another experiment may be provided to the PMF as an 'anchor/vector' around which it can build a factor in its overall solution. The extent to which each PMF factor can diverge from the anchor is defined by the value of a (Tong et al., 2021), which varies from 0 to 1, where 0 = no divergence and 1 = up to 100% divergence. This anchor can be provided for one or multiple factors and has been proven to improve the quality of PMF solutions compared to unconstrained trials (Tong et al., 2021; Stefenelli et al., 2019a; Chen et al., 2022).

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Currently, there are no fully objective criteria for choosing the best number of factors in a solution, but some criteria have been suggested in the literature to make an appropriate selection (Chen et al., 2022; Zhang et al., 2011; Ulbrich et al., 2009; Crippa et al., 2014). The PMF solutions reported here were primarily selected based on their -physical meaninginterpretability, which was in turn, determined by the presence of the distribution of known tracer compounds in the factors, temporal correlation with colocated measurements of external tracers (e.g., NO_x, SO₂)., and the temporal agreement of factors determined by the two instruments. We selected an esolved eight, four, and six factors solution from PTR_{CHARON}, AMS_{org}, and AMS_{org+inorg}, respectively. The justification for these solutions choices is presented in Table S2. Once the most suitable solution, i.e., the base-case, was established, bootstrap analyses were performed to assess its stability, evaluate uncertainties, and conduct a sensitivity analysis on the range of a-values used. In an unblocked bootstrapping approach, the original matrices (both data and error) are perturbed by random resampling of the rows to create a new input of the same dimensions, resulting in some duplications and deletions throughout the input (Paatero et al., 2014). The need and application of this approach differed between the PTR_{CHARON} and the two AMS datasets as discussed in Sections S5 and S6, respectively. Ancillary data on particle size distribution in-have been associated to mass spectrometry data in an additional PMF analysis each factor was generated by a fully constrained PMF or simple linear regressions of the SMPS datasets (Section S7). Finally, the quality of solutions was gauged by the Q/Q_{exp} values and from key diagnostic plots of residuals and the statistical stability across multiple runs (Figure S5-S7).

3 Results and Discussion

3.1 Campaign overview

Figure 1 <u>summarises depicts a summary</u>-meteorological conditions, chemical composition and particle size distribution of NR-PM₁ observed from January 20 to February 26, 2022. High aerosol loads coincided with poor atmospheric dispersion, due to low wind speeds (<2 m/s), low temperature (below - 10°C) associated with strong surface temperature inversions, as temperature differences between 23 and 3 m above sea level ranging from 3°C to 10°C. The <u>campaign</u>-average <u>valuess</u> of BC and NR-PM₁ measured with the MAAP and AMS were 1.4 ± 1.4 μg/m³ and 8.3 ± 9.3 μg/m³, respectively. <u>Intense During intense</u> pollution <u>events episodes occurred from Jan 31 to Feb 02, during which the daily average concentrations of NR-PM₁ were 24–27 μg/m³. For this polluted period, another ALPACA study <u>During the same sampling period at NCore site (Fairbanks) reported campaign averaged PM_{2.5} <u>value</u> of ~25 and ~29 μg/m³ <u>were reported at NCore (a monitoring station located approximately 580 m from the CTC) using a beta attenuation mass monitor and a nearby site of Downtown using a DustTrak DRX</u></u></u>

aerosol monitor (Robinson et al., 2023). In our study, the hourly PM₁-concentrations measured at the CTC site comprised up to 99% of the PM_{2.5}-measured with an OPC, warranting that future studies in Fairbanks must explore the distribution, dynamics, and impacts of sub-micron aerosol to gauge the need for its targeted mitigation. Ancillary OPC measurements at the CTC site showed that the hourly PM₁ mass comprised up to 99% of the PM_{2.5}.

Organics were the predominant component of NR-PM₁ throughout the campaign, constituting $\sim 66 \pm$ 11% of its total mass, while chloride, ammonium, nitrate, and sulphursulphate-based inorganics contributed 2 ± 3 , 3 ± 3 , 6 ± 4 , and $22 \pm 10\%$. This is in line with previous studies in Fairbanks, where OA was the largest component of PM_{2.5} mass (Ward et al., 2012; Ye and Wang, 2020; Robinson et al., 2024). Specifically, according to a recent study from 2020 to 2021, ACSM analysis during the wintertime demonstrated inorganics to form less than 25% of the PM_{2.5} mass only, with sulphate (~10%) and nitrate (~8%) being the predominant components (Robinson et al., 2024). Despite the different average concentrations, the fractional contributions of these non-refractory components remained almost invariable throughout the campaign (Figure 1D). Detailed molecular-level composition of organics with the PTR_{CHARON} reveals a large majority of organics to comprise only C, H, and/or O atoms, while only $\sim 9 \pm 4\%$ of the OA_{CHARON} mass measured with this instrument was attributable to heteroatomic species molecules, including organonitrates and organosulphates (Figure 4 and S8). Generally, heteroatomic species cannot be distinguished at a resolving power of 5000 FWHM in complex environmental mixtures, such as atmospheric aerosol (Reemtsma, 2009). In this study, based on the low formula error and lack of an appropriate alternate, we gave 53 low-concentration ions (< 2%of the total signal) CHOS or CHNO identities, but due to the low confidence in their formula assignments, they were not considered for factor identification. Prominent peaks include m/z 217.09 $(C_{12}H_{12}N_2O_2)$, 219.09 $(C_{15}H_{10}N_2)$, 123.05 $(C_4H_{10}O_2S)$, and 151.08 $(C_6H_{14}O_2S)$.

On average, the OA mass loading recovered by PTR_{CHARON} (i.e., OA_{CHARON}) accounted for approximately 85% of the OA mass measured by the AMS (i.e., OA_{AMS}). While the two instruments showed a strong-good temporal agreement (R² = 0.60) as depicted in **Figures 2A–B**, measurements were biased either toward the AMS_{org} or the PTR_{CHARON} (i.e., distributed away from the 1:1 line in the scatter plot of **Figure 2C**) during different periods of the campaign. These trends could be explained by the variation in relative contributions of two major emission sources identified by both instruments in this study: on-road traffictransport and biomass burning. OA_{CHARON} was comparable to OA_{AMS}, when the relative contribution of BBOA_{AMS,org} was more than 50% of total OA_{AMS} and HOA_{AMS,org} (i.e., traffictransport_{CHARON}) was less than 10% (**Figure 2D–E**). Similar trends were observed for some major constituents of BBOA, e.g., levoglucosan and a PAH (C₂₀H₁₂) as shown in **Figure S9**. Part of such discrepancy This relationship of instrument performance with the source-can be traced back to the size

373 transmission of particles, where sub-100 nm urban vehicular emissions are underestimated by the 374 PTR_{CHARON} (Guo et al., 2020; Pikridas et al., 2015; Louis et al., 2017; Kostenidou et al., 2020), and 375 larger than 100 nm biomass burning emissions (Reid et al., 2005) are estimated well (Janhäll et al., 376 2010). Another pPart of the quantitative difference between the two instruments can also be explained 377 by the detection abilities of PTR limitation in ionisation and the ionisation-induced fragmentation of 378 analyte ions, during PTR ionisation that introduces a negative biaThis Only abias has been reported to 379 be small underestimation of for oxidised organic compounds has been reported with the PTR_{CHARON} that 380 can be corrected using previously proposed methods (Leglise et al., 2019). Contrarily, Additional Ttests 381 carried outconducted in our laboratory with five C₁₆-C₂₆ alkanes as markers of vehicle emissions 382 revealed that they undergo extensive fragmentation, increases dramatically and resultings in a-2-4 times 383 underestimation of their actual concentrations. In line with this, tThe ineffective ionisation of saturated 384 alkanes by PTR (Ellis and Mayhew, 2014) and their tendency of alkanes from vehicular exhausts to 385 undergo dissociative PTR-ionisation has also been reported previously (Gueneron et al., 2015) hashave 386 also been reported.

3.2 Source apportionment

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3.2.1. Overview of source apportionment

389 A four-factor solution was selected for the AMS_{org} measurements with three primary factors (i.e., HOA, 390 COA, and BBOA) and an oxygenated or aged OA factor (i.e., OOA). The mass spectra and time series 391 are presented in the supplement (Figure S10). Counterparts of these four factors were diagnosed in $AMS_{org+inorg}$ based on a high temporal correlation ($R^2 > 0.9$; Table S4), along with two additional factors: 392 393 a sulphur-rich factor (labelled sulph-OA) and a nitrate-rich factor (labelled AmNi) (Figure 3). An eight-394 factor solution was selected for PTR_{CHARON} and is summarised in Figures 4 and 5. To differentiate 395 between corresponding factors retrieved from the different datasets, they have been given unique 396 subscripts, e.g. COA_{AMS,org}, COA_{AMS,org+inorg}, COA_{AMS} (i.e., referring to both AMS datasets), or 397 COA_{CHARON}. Amongst the three datasets COA, HOA (labelled 'traffietransport' in PTR_{CHARON} 398 analyses), and OOA were common. A single BBOA factor was observed in AMS_{org} and AMS_{org+inorg}, 399 while four chemically distinct, but closely co-varying counterparts were detected by PTR_{CHARON}. 400

3.2.2. Organic aerosol from residential heating.

Both AMS analyses indicate that biomass burning is among the major sources of PM₁ during the ALPACA campaign. On average, BBOA contributed 1.5 \pm 1.9 μ g/m³ (28 \pm 18% of total OA_{AMS}) and $1.6 \pm 2.2 \,\mu\text{g/m}^3 \,\text{NR-PM}_1 \,(19 \pm 14\% \,\text{of total NR-PM}_1 \,\text{mass})$. The mass spectra of BBOA_{AMS} featured a strong peak at m/z, 60 (C₂H₄O₂⁺) and 73 (C₃H₅O₂⁺) (**Figure S10A–B**). These fragments are markers of anhydrosugars in wood-forming polymers, such as cellulose(Tobler et al., 2021). Wood combustion has 12

previously been estimated to be the largest emitter of aerosols in Fairbanks and surrounding areas, where it may produce as much as 80% of the aerosol load-(Haque et al., 2021; Ward et al., 2012; Wang and Hopke, 2014; Kotchenruther, 2016). Wood burning emissions are also the major driver of the spatial variability of PM_{2.5} and BC in Fairbanks during strong atmospheric temperature inversions-(Robinson et al., 2023). Other typical residential heating sources of emissions in Fairbanks include coal, gas, and fuel oil-(Simpson et al., 2019).

The BBOA_{AMS} factor was strongly correlated with PAHs ($R^2 \ge 0.7$) while a . In addition, a moderate correlation was observed with SO₂ ($R^2 = 0.4$) (**Table 1**). While PAHs are a major component of biomass combustion emissions, the emission of SO₂ is largely associated with coal and oil combustion (Smith et al., 2011; Dunleavy and Brune, 2019). However, the AMS was unable to distinguish between multiple combustion-related sources. As shown in the diurnal plots in **Figure 3**, the concentration of the BBOA_{AMS} factor enhanced at ~1800 AKST, stayed stable through the night and then decreased in the early morning. Its lowest mass concentrations occurred during the afternoon (1300–1500 AKST). Therefore, BBOA_{AMS} was could be associated with residential heating, i.e., the combustion of a variety of fuels by residents within their homes (non-commercially), such as in wood-burning stoves, furnaces, boilers, etc. for heating living space. We did not find evidence of OA or NR-PM₁ from commercial heat providers, such as power plants, likely due to their small contribution to surface-level aerosol due to smokestacks lying above the inversion layer.

 PTR_{CHARON} apportioned **2.6** \pm **3.4** μ g/m³ of OA_{CHARON}, on average, to four distinct residential heating-related sources expressed as ResH1–4 (**62** \pm **26%** of total OA_{CHARON}). These factors closely co-varied in time and were correlated reasonably well (R² = 0.5–0.7; **Table S5**) with the BBOA_{AMS} factor. In addition, combining all four residential heating-related factors in PTR_{CHARON} into a composite factor increased the correlation (R²) with AMS_{org} and AMS_{org+inorg} to 0.79 and 0.82, respectively, suggesting that PMF was not able to effectively separate these closely co-varying residential heating factors when their molecular signatures were weakened due to the extensive EI-induced fragmentation-in AMS dataset.

The four factors from PTR_{CHARON} were identified as different sources based on the distribution of key marker species and correlation with external (e.g., trace gases, etc.) and internal (e.g., PAHs). measured with co-located instruments; particle size distribution) tracers. The IL evoglucosan signal is used here as an internal tracer of biomass burning being because it is relatively stable under typical atmospheric conditions (Fraser and Lakshmanan, 2000). A majority of the signal from pProtonated levoglucosan (m/z 163) and its fragments (at m/z 85, 127, and 145) were found appeared in ResH1, ResH4, and ResH2 with 30, 26, 14% of the total signal respectively (Figure S11) in the same order), with only minor

association with ResH3accounted for suggesting that they the former three to originate from biomass burning more specifically, wood-burning (Figure 4 and S11). These three wood-burning related factors collectively accounted for produced an average of $2.1 \pm 2.5 \,\mu\text{g/m}^3$ (of OA_{CHARON}).

> **ResH1** includes low temperature combustion markers: this factor is small as it contributes to only an average of $0.5 \pm 0.5 \,\mu g/m^3$ (14%) of the total OA _{CHARON}, but it contains the highest fraction of levoglucosan (~30%). Approximately 65% of the total signal of ResH1 is due to compounds with six or fewer carbon atoms, compared to heavier species present in the other factors (Figure S13). The most abundant species are at m/z 69.03 (C₄H₄O; furan) (Palm et al., 2020; Jiang et al., 2019), m/z 87.04 (C₄H₆O₂; oxobutanal) (Brégonzio-Rozier et al., 2015), m/z 97.03 (C₅H₄O₂; furfural), m/z 109.0286 $(C_6H_4O_2; benzoquinone)$ (Stefenelli et al., 2019b) and m/z 115.04 ($C_5H_6O_3$; methyl-dihydrofuran) (Koss et al., 2018). Consistent with these molecular formulae, the concentration-weighted average O/C of ResH1 was relatively higher (i.e., 0.42) compared to other residential heating factors (O/C = 0.2-0.3). The most abundant species observed in ResH1 can be related to depolymerisation reactions occurring during low temperature and early stages of the combustion process (Collard and Blin, 2014; Sekimoto et al., 2018). ResH1 includes mixed wood-burning OA: Although ResH1 had the strongest levoglucosan signal, it contributed the least OA with an average of $0.5 \pm 0.5 \,\mu g/m^3$ and did not feature any other prominent wood burning tracers, such as PAHs. As shown in Figure S13, ~65% of the total signal of ResH1 came from compounds with six or fewer carbon atoms, compared to heavier species in other factors. Many species with the greatest concentrations in ResH1, relative to other factors, have been reported as oxidation products of BBOA ageing in previous studies, such as m/z 69.03 (C₄H₄O; furan) (Palm et al., 2020; Jiang et al., 2019), m/z 87.04 (C4H6O2; oxobutanal) (Brégonzio-Rozier et al., 2015), m/z 97.03 (C₅H₄O₂; furfural), and m/z 109.0286 (C₆H₄O₂; benzoquinone)(Stefenelli et al., 2019b). Collectively, ResH1 comprises OA from the combustion of a variety of mixed wood based solid fuels as evidenced by the presence of levoglucosan, but it also likely includes OA in the early stages of processing.

ResH2 and ResH4 include OA from hardwood and pinewood combustion, respectively: Two more factors associated with wood-burning were ResH2 and ResH4. Their average OACHARON-concentrations were 1.1 ± 1.9 and 0.8 ± 0.9 μg/m³, respectively, corresponding to 28 and 20% of the OACHARON (Figure 6). As shown in Figure 6A, The ResH2 was dominant factor in the PMF of PTR_{CHARON} thatand could reach ~37 μg/m³ of OACHARON alone during the most severe pollution episodes. Not only did these Both factors correspond to were associated to OA particles sizes greater than 300 nm (Figure S12), which is typical of woodsmoke (Glasius et al., 2006), and but they also presented unique molecular signatures of different wood types as shown in (Figure S11) and discussed next. Generally, the specific nature of

wood cannot be inferred unambiguously because the emissions of known marker species, such as levoglucosan or methoxy phenols, vary not just with fuel used and its quality, but also with the type of heating appliance, operational conditions, appliance efficiency, and stage in the combustion cycle (Fine et al., 2002; Alves et al., 2017). Regardless, several studies (Fine et al., 2002; Schauer and Cass, 2000; Kawamoto, 2017)_have distinguished between softwood from hardwood by investigating the presence of marker compounds that were observed in our study as well, such as substituted phenols and resin acids (Figure S11).

ResH2 featured an abundance of of prominent methoxy phenols, including $C_7H_8O_2$ (guaiacol), $C_8H_{10}O_3$ (syringol), $C_{10}H_{10}O_3$ (coniferaldehyde), $C_6H_6O_2$ (benzenediol (catechol)—or methylfurfural), and $C_8H_{10}O_2$ (creosol), that where they collectively accounted for ~9% of the total signal, compared to 1, 2, and 2% in ResH1, ResH3, and ResH4, respectively. These compounds are important products of lignin pyrolysis in birch, aspen, and spruce and are usually found detected in the gas phase at mild ambient temperatures (Kong et al., 2021). Guaiacol and syringol are depolymerisation products of guaiacyl and syringyl units of lignin at 200–400°C, and they rapidly transition to catechols, cresols, and phenols during secondary pyrolysis reactions at 400–450°C, eventually leading to enhanced PAH formation at >700°C (Kawamoto, 2017). While guaiacols are emitted to some extent by the burning of both hardwood and softwood, semi- or low-volatility substituted syringols that primarily exist in the condensed phase are emitted in much higher amounts by hardwood combustion (Kawamoto, 2017; Fine et al., 2002, 2001; Schauer and Cass, 2000). In this study, derivatives of guaiacols, including $C_{10}H_{12}O_2$ (eugenol), $C_{10}H_{14}O_2$ (4-propyl guaiacol), and $C_{10}H_{10}O_3$ (coniferaldehyde) presented much higher "relative concentration" concentrations (Equation S4) of 0.56–1.41 for ResH2 and ResH4 compared to <0 for ResH1 (<0). Other compounds, such as $C_8H_8O_3$ (vanillin), $C_9H_{10}O_3$ (acetovanillone), $C_{10}H_{12}O_3$ (propiovanillone), and $C_{10}H_{12}O_4$ (methyl homovanillate) were predominantly found in ResH2. Similarly, substituted syringols, i.e., $C_{11}H_{14}O_3$ (methoxy eugenol), $C_{10}H_{12}O_4$ (acetosyringone), and C₁₁H₁₄O₄ (syringyl acetone, propionyl syringol, or sinapyl alcohol) were almost entirely associated with ResH2 as well. These compounds have been reported as markers of hardwood burning (Fine et al., 2001), implying a potentially greater contribution of hardwood emissions smoke to the ResH2 factor. In Alaska, relevant hardwood species include deciduous leafy trees, i.e., paper birch, balsam poplar, quaking aspen, etc (Adec, 2023).

For The ResH4, in addition to the levoglucosan marker ions, a predominance of largepresented a unique fingerprint characterised by, oxygenated molecules with more bearing more than 13 carbon atoms was observed (**Figure S13**), such as $C_{16}H_{30}O_6$ (m/z 319.21), $C_{20}H_{28}O_2$ (m/z 301.21), $C_{22}H_{18}O$ (m/z 299.14), $C_{20}H_{18}O_4$ (m/z 323.12), and $C_{20}H_{30}O_2$ (m/z 303.24) and $C_{22}H_{18}O$ (m/z 299.14), in addition to the levoglucosan marker ions (26% of the total signal). Amongst these, more than 60% of the The intense

signals -from m/z 301 (C₂₀H₂₈O₂) and m/z 303 (C₂₀H₃₀O₂) was associated with ResH4 (Figure S11). These species are likely related to resin acids, dehydroabietic acid and abietic acid, respectively, which are almost exclusively emitted from the thermal alteration of resins in coniferous species, and thus, are indicative of softwood burning (Simoneit, 2002, 1999). Owing Due to the presence of these compounds, ResH4 was identified interpreted as an OA factor influenced by softwood combustion. Softwood species in Alaska include trees with needles and cones, e.g., hemlock, cedar, and spruce (Adec, 2023).

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ResH3 includes OA from heating oil combustion: this factor A factor, labelled ResH3, contributed to $16 \pm 9\%$ of the total OA_{CHARON} ($0.6 \pm 0.6 \, \mu g/m^3$) and showed the characteristic diurnal pattern of residential heating as it . It correlated quite well ($R^2 = 0.56$) with BBOA_{AMS,org}. However, its chemical composition was very different from the other residential heating factors. Notably, levoglucosan contributed to a smaller fraction of the total signal of ResH3 (i.e., 9%) compared to other residential heating factors (1314–2930%; Figure S11), but while PAHs represented a much larger fraction of its total signal (for instance, 30, 31, and 29% of $C_{16}H_{10}$ (m/z 203.09), $C_{18}H_{12}$ (m/z 229.10), and $C_{20}H_{12}$ (m/z253.10); **Figure S13**). These PAHs could be fluoranthene (or pyrene), naphthacene (or benzo[x]anthracene, chrysene), and benzo(x)pyrene (or benzo(x)fluoranthene)), which have been reported in emissions of light oil combustion (Bari et al., 2009). Additionally, ResH3 was strongly correlated with SO_2 ($R_2 = 0.61$) during the campaign, compared to a moderate lower correlation (\mathbb{R}^2 of < 0.47) with the other the remaining residential heating factors. Residential combustion of heating oil is an important source of SO₂ in Fairbanks, compared to wood and coal, due to $\sim 2/3^{\rm rd}$ of the households using oil-fired space heaters and the high sulphur content of > 1600 ppm in fuel oils commonly consumed here (e.g., #1 and #2 fuel oil and waste motor oil are relevant in Fairbanks)(Dunleavy and Brune, 2019). Consistent with the possibility of the ResH3 factor denoting being associated with fuel oil emissions, the factor is characterised by particles smaller than 100 nm (Figure S12) and due to the CHARON inlet's cut-off, its mass concentration was possibly fuel oil emissions, a fully constrained PMF on SMPS measurements matched it with particles smaller than 100 nm (Figure S12). Due to the small particle size, it is possible that mass concentrations of OACHARON were under apportioned toestimated. ResH3; this possibility is discussed in detail for the on road traffic factor in the next section.

3.2.3. Hydrocarbon-like and cooking organic aerosol

- The HOA_{AMS} factors were characterised by notable peaks at m/z 43 (C₃H₇⁺), 57 (C₄H₉⁺), 71 (C₅H₁₁⁺),
- 85 ($C_6H_{13}^+$), and 99 ($C_7H_{15}^+$), belonging to [C_nH_{2n+1}] series, that are typical of n- and branched alkanes.
- 547 There were also Additional peaks at m/z 55 (C₄H₇⁺), 69 (C₅H₉⁺), 81 (C₆H₉⁺), 83 (C₆H₁₁⁺), 95 (C₇H₁₁⁺),
- 548 97 $(C_7H_{13}^+)$, 107 $(C_8H_{11}^+)$, 109 $(C_8H_{13}^+)$, and 111 $(C_8H_{15}^+)$ that belong to represented $[C_nH_{2n-1}]^+$ and

[C_nH_{2n-3}]⁺ series, <u>related to which are typical indicative of cycloalkanes (McLafferty et al. 1993)</u>. These are key ions <u>are associated with engine-lubricating oils</u>, vehicular exhaust, and diesel fuel (Canagaratna et al., 2004). The HOA_{AMS,org} and HOA_{AMS,org+inorg} factors contributed $38 \pm 20\%$ (of the OA_{AMS}) and $21 \pm 14\%$ (of the OA_{AMS} and total-NR-PM₁) mass, respectively (**Figures 6 and S14**).

HOA is generally associated with linked to vehicular emissions but from on road traffic, it which were The unconstrained PTR_{CHARON} analysis was not able to apportion a road transport-related factor, was not observed in the unconstrained PMF of PTR_{CHARON}. Hhowever, an factor for on road traffic factor was 'artificially' diagnosed in the PTR_{CHARON} analysis by constraining the its factorisation with the time series of a mobile gasoline factor, identified in the gas-phase PTR-ToF MS analyses of ALPACA campaign (Temime Roussel et al., 2022), a small road transport-associated factor was identified. This approach was effective, revealing success of constraining this factor was evident in characteristics typical of on road traffic, such as For instance, this latter it was strongly correlationed with black carbon and NO_x (R² of 0.58 and 0.66; **Table 1**) and featured high contributions of C_8H_{10} (xylene; ethylbenzene: $\frac{2\%}{2}$), C_7H_8 (toluene: $\frac{4\%}{2}$), and C_6H_6 (benzene: $\frac{0.5\%}{2}$) to its total mass concentrations (Figures 4 and S11). Despite exhibiting some reasonable diurnal trend peaking at In additionNotably. peaks in the daily average mass concentrations of the traffic HARON factor coincided with the morning (0900 AKST) and evening (1700–1600 AKST) rush hours (Figure 5), However, the traffic (HARON-the factor had accounted for negligible concentrations (< 1 µg/m³) and presented some it contained implausible unlikely species, such as m/z 315.22 (C₂₁H₃₀O₂; possibly cannabidiol) that would otherwise (e.g., inabsent in the unconstrained PMF, trials) appear as PMF residuals, making its environmental representativeness suspicious, large discrepancy between the OA apportioned to HOA by PTR_{CHARON} and AMS_{org}. For instance, on average, $2.1 \pm 3.0 \,\mu\text{g/m}^3$ of OA was associated with HOA_{AMS,org}-during the campaign, compared to only $0.1 \pm 0.1 \,\mu g/m^3$ for the road traffic ransport (Figure 6). We speculate that the These discrepancies shortcomings seen in OA mass measuredments by the from PTR_{CHARON} relative compared to the AMS_{ore} were are largely instrumental, particularly partly duedue to such as the poor transmission low sensitivities of the PTR_{CHARON} for small particles (<100 nm) by the CHARON inlet and the limited sensitivity towards hydrocarbons by PTR, but other possible biases can be due to heating oil OA signal interfering with the HOA_{AMS}, as discussed in S8. Previous studies using the PTR_{CHARON} in Innsbruck, Austria, successfully observed a traffic factor, but did not detect no cooking emissions despite sampling operating at in an urban locality area (MüLler et al., 2017). A vVarietyous of environmental and user biases could also may be involved, such as the contribution of non-vehicular sources to the HOAAMS factors and the choiceuse of suboptimal conversion coefficients (e.g., RIE) in the AMS analyses (see Sections S8 and S9 for details). These are important considerations are critical for ambient air analysis in employing the with PTR_{CHARON}, as obtaining for ambient air analyses because a full picture of the sources involved, especially in urban regions influenced by

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primary OA emissions of smaller particle sizes, may not be possible without necessitate complementary measurements.

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3.2.4. Cooking organic aerosol

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Another primary factor identified in Fairbanks was cooking, likely arising which could either be from residential or commercial activities around the CTC. Both-The COA_{AMS} factors featured a high abundance of $C_vH_v^+$ ions, along with prominent O_1 fragments at m/z, 55 ($C_3H_3O^+$), 84 ($C_5H_8O^+$), and 98 $(C_6H_{10}O^+)$, which originating from organic acids (Mohr et al., 2009) and These fragments have been reported serve used as diagnostic spectral markers of COA in urban settings (Sun et al., 2011). The f55/f57 value (i.e., the ratio of fractions of $C_4H_7^+$ to $C_4H_9^+$) was ~3.00 for COA_{AMS}, compared to ~1.04 in for HOA_{AMS} (Figure S10D). Although A aAs a reliable tracer for COA remains unidentified in the AMS spectrum, a high-f55/f57 ratio of->1 is considered a characteristic feature of COA-(Katz et al., 2021; Sun et al., 2011) because a reliable external tracer for it is yet to be identified. The PMF analysis of PTR_{CHARON} also contained revealed a distinct COA factor dominated by long-chain fatty acids. C₁₈H₃₂O₂, C₁₈H₃₄O₂, and C₁₈H₃₆O₂, identified here as linoleic, oleic, and stearic acids contributing to that which contributed 11, 16, and 4% to the total COA_{CHARON} mass (Figure 4 and S11). These fatty acids are commonly found-markers of in OA from cooking oil and meat (Katz et al., 2021; Mohr et al., 2009). Throughout Across the whole campaign, The COACHARON made peaked its highest contributions of at contributed to a maximum of ~9% to of the total OA_{CHARON} mass and exhibited a unique diurnal pattern visualised in Figure 5 with a minor maximumum in the afternoon (lunchtime) and a second maximum in the evening (dinnertime). , creating resulting in the a unique diurnal pattern visualised in Figure 5. Similarly to the HOA and traffic_{CHARON} factors The average absolute concentrations of COA AMS or and COACHARON were quite disparate between the two instruments were (0.6 ± 0.8) for the AMS and $0.1 \pm 0.2 \,\mu g/m^3$ for the CHARON, such a discrepancy can be explained by the same reason discussed above and detailed), respectively for reason discussed above in the section as well as in Sections S8 and S9. (Figure 6).

3.2.45. Oxygenated organic aerosol

It is common in <u>PIn the past</u> source apportionment studies <u>have reported</u>, it has been common to report multiple OOA factors <u>differing that differ</u> in volatilities or oxygenation levels (e.g., Stefenelli et al., 2019a; Kumar et al., 2022; Cash et al., 2020). Here, <u>but</u> we diagnosed only a single OOA factor in either AMS or PTR_{CHARON} measurements. <u>Specifically</u>, <u>t</u>The OOA_{AMS} factors <u>in AMS analyses was</u> were identified based on a prominent peak at m/z 43 (C₂H₃O⁺) which is a tracer of less oxygenated OA, —and m/z 29 (CHO⁺; **Figure S10A**) and . This factor It correlated showed a strong correlation (R² 0.74)

with OOA_{CHARON}, with an R² of 0.74. The whe average absolute concentrations of OOA_{CHARON} and OOA_{AMS.org} were 0.4 ± 0.6 and $1.0 \pm 2.1 \,\mu\text{g/m}^3$, respectively.: -Notably, tsSome of the most intense ions in the mass spectra spectrum of OOA_{CHARON}, relative to other factors, have been tentatively assigned toincluded m/z 73.03 (C₃H₅O₂, e.g., methylglyoxal), m/z 99.01–04 (C₄H₂O₃C₅H₆O₂, e.g., maleic anhydride oxo-pentanal), m/z 113.06 ($C_6H_8O_2$, e.g., methyl-oxo-pentanal), m/z 127.08 ($C_7H_{10}O_2$; e.g., heptadienoic acid), m/z 137.06 (C₈H₈O₂; e.g., methoxy-benzaldehyde), m/z 167.10 (C₁₀H₁₄O₂), m/z171.07 (C₈H₁₀O₄) and m/z 185.10 (C₁₃H₁₂O; e.g., benzyl phenol). Among these compounds some have previously been associated with atmospheric oxidation or photolysis of BBOA (Montoya-Aguilera et al., 2017; Lignell et al., 2013; Smith et al., 2020), other could be due to oxidation aromatic VOCs detected originating from road transport (Temime-Roussel et al., 2022). Few Several, as well as some other species that overlapped with the residential heating factorstracers, notably m/z 163.06 (C₆H₁₀O₅; e.g., levoglucosan), m/z 179.08 ($C_{10}H_{10}O_3$; e.g., coniferaldehyde), and m/z 301.21 ($C_{20}H_{28}O_2$; e.g., dehydroabietic acid). Some of these species compounds (e.g., C₄H₂O₃, C₁₀H₁₄O₂, C₂H₁₀O₂) have previously been associated with atmospheric oxidation or photolysis of BBOA (Montoya Aguilera et al., 2017; Lignell et al., 2013; Smith et al., 2020). But Ggiven the prominence of wood-burning as the a major primary emissions in ALPACA, the OOA is likely linked to BBOA. A recent study in Fairbanks identified wintertime OOA as a mixture of BBOA and SOA formed from non-photochemical processing using an ACSM (Robinson et al., 2024). The examination of f44 versus f60 in the AMS_{ore} dataset plot (Figure S10C) is consistence with aged OOA derived by biomass burning, as previously demonstrated by Xu et al., (2023). Another recent source apportionment of NR-PM₄study measured with the HR-ToF AMS at a site close to the CTC did not reveal-identified an OOA factor at all, while BBOA, HOA, and a mixed primary factor (HOA, COA, etc.) comprised 45, 25, and 31% of total OA, on average, during the campaign (Yang et al., 2024). Minimal processing, and thus, ILA limited OOA formation is plausible due to reduced -short solar light exposure periods and pollution residence in this period of the year in Fairbanks (Cesler-Maloney et al., 2024), however, but a the complete disappearance absence of OOA is more likely a result to be a consequence of it remaining an unresolved organic fractionunder the factorisation method used. Another recent study in Fairbanks using the ACSM identified wintertime OOA as a mixture of real BBOA and SOA formed from non-photochemical processing (Robinson et al., 2024). This aspect was investigated via an f44 versus f60 plot for AMS_{ers} that supports some influence of biomass burning at all levels of oxidation of OA (Figure S10C). The placement of OOA_{AMS ore} toward the left edge of the f44 versus f60 plot is consistent with aged OA from wood burning (Xu et al., 2023), but an urban influence cannot be ruled out in field settings (Cubison et al., 2011), especially when m/z 60 and 73 are only 0.2 and 0.4% of the total OOA_{AMS,org} signal (**Figure S10B**).

Sulphate and OOA. Much more An intriguing insight interesting information about regarding the the OOA factor emerged was gleaned from the AMS_{org+inorg} measurements, which revealed it to be rich

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indicating a significant content of sulphur in sulphur containing compounds (Figure S15). The AMS does not quantitatively distinguish among the different sulphur-containing species, but such as hydroxymethane sulphonate (HMS: CH₂(OH)SO⁻³), SO₂²-(sulphite), HSO₂-(bisulphite), and SO₄²-(sulphate), or between organic and inorganic sulphur, but we have tried to use following guidelines from previous works (Chen et al., 2019; Schueneman et al., 2021) we could explored the ratio of sulphur fragments to investigate speculate on the presence of different species forms of sulphur present such as hydroxymethane sulphonate (HMS; CH₂(OH)SO⁻³), HSO₄⁻¹ (bisulphite), SO₄²⁻¹ (sulphate) and (H2SO4) sulphuric acid. In light of previous studies on sulphur source apportionment with the AMS and source differentiation based on fragmentation patterns, (Chen et al., 2019; Schueneman et al., $\frac{2021}{100}$. An organosulphate content of $\sim 0.8 \pm 1.3$ ug/m³ ($\sim 20 \pm 16\%$) was then derived using the ratios of SO⁺ and SO₂⁺ ions against SO₃⁺, HSO₃⁺, and H₂SO₄⁺ ions as detailed by Song et al., (2019). This value is in good agreement with previous reports from the same field campaign (Campbell et al., 2022; Robinson et al., 2024). Additionally, to mimic potential matrix effects of wood burning OA on sulphate fragmentation patterns, we performed calibrations on the AMS spectra from a solution of with pure (NH₄)₂SO₄ mixed with various amounts of levoglucosan (i.e., 0–80% in mass). This mixture was used to assess the matrix effect that can potentially impact sulphur fragmentation patterns in an environment affected by wood smoke as previously demonstrated with other organic matrices. We were compared to ambient data and PMF factors as shown compared the fractions of HSO₃+ to H₂SO₄+ fragments normalised to the fractions H₂SO₄⁺ and HSO₃⁺ for pure (NH₄)₂SO₄ as suggested by Chen et al., (2019). Results are shown in Figure S16A. Results Among the sulphate-rich factors the are shown in Figure S16A where the OOA_{AMS.org+inorg} factor exhibited much lower HSO₃⁺ to H₂SO₄⁺ intensities suggesting which is indicative of an influence of an higher fraction of organosulphate -compounds, as it is also evidenced in influence Figure S16D-E by the strong correlation between derived organo-sulpahur fraction and sulphate-ions in the OOA_{AMS,org+inorg} factor ($R^2 = 0.85$, slope = 0.57). (Figure S16D E). The average total concentration of sulphursulphate related fragments in OOA_{AMS ore timors} was **0.9 ± 1.8** µg/m³, on average, and which accounted for 26 ± 23% of the total sulphur sulphate measured with the AMS. This finding is consistent, which agrees with the theoretical estimations of organosulphur content (Song et al., 2019).

Further information on chemical composition was gathered by comparing this factor AMS results with IC measurements from PM_{0.7} filters sampleds analysed as part of another ALPACA study (Dingilian et al., 2024). Despite the good correlations between the two datasets, shown in Figure 7A, the AMS results underestimated - Although there is a notable underestimation Both methods (IC and AMS) correlated well, despite a negative bias against in the AMS analysis that underestimated the sums of approximately by 30%, 26%, and 35% lower sulphur sulphate, ammonium, and nitrate related fragments, respectively (see Section 2.2.2 for fragments included). the correlation between the two

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methods remains strong by ~ 31, 26%, and 35% compared to the IC analyses (Figure 7A). Both the total estimated organosulphur fraction and the OOA_{AMS,org+inorg} factor presented very strongrobust correlations ($R^2 > 0.90$) with the $S_{(IV)}$ and HMS ions from filter analysis (Figures 7B and S16F I), and a somewhat with a relatively weaker $\frac{1}{2}$ but still strong correlation (R² > 0.61–0.68), with the SO₄²⁻ ion (Figures 7 and S16F-I). In additionAdditionally, the The OOA_{AMS,org+inorg} this-factor was also very strongly correlated with total-ammonium -(AMS data $R^2 = 0.8695$, Table 1; filter IC $R^2 = 0.77$, Figure **S16D-E7B**), potentially promoting the formation of S_(IV) species under favourable meteorological conditions and aerosol composition (Campbell et al., 2024). This author also reported that S_(IV) species, including HMS, have been represented the observed the major secondary organosulphur component of PM_{2.5} in Fairbanks during wintertime, with average concentrations of 0.29 and 0.34 μg/m³ recorded with IC in 2020 and 2021, respectively, and contributing to 26–41% of total sulphate (Campbell et al., 2022). Recently, co-varying HMS and S_(IV) species, were distinguished in Fairbanks, and identifying the non-HMS S_(IV) were reported to be as aldehyde S_(IV) compounds (Dingilian et al., 2024). In addition, this factor was very strongly correlated with total ammonium ($R^2 = 0.95$, Table 1; Figure S16D E) which could raise acrosol pH, favouring the formation of S_(IV) species under appropriate meteorological conditions and acrosol composition (Campbell et al., 2024). Therefore, the presence of HMS and other organic S_(IV) species in the AMS_{ore+inore} factor is well substantiated. Overall, the molecular level composition of OOA from PTR_{CHARON} and the inorganic chemical information from AMS_{org-inorg}, as well as diurnal patters with indicating enhanced peak concentrations in the afternoon (Figure 3), are that is indicative of chemical daytime processing that the wintertime OOA in Fairbanks is not solely comprised of HMS; . Instead, it is instead a complex mixture of secondary non-heteroatomic organic matter and organosulphur compounds, which underscoring the need for further exploration of the hints toward its formation from complex atmospheric processing pathways involved that needs further exploration.

3.2.56. Additional insights from combined analysis of organic and inorganic in-AMS data measurements

Two additional factors, sulph-OA (i.e., sulphur-rich OA) and AmNi (i.e., ammonium nitrate), were observed from the PMF of AMS_{org+inorg} (Figure 3). Approximately 40–60% of these factors' masses comprised sulphur and nitrogen species (Figure S15).

Sulphur-rich organic aerosol: Sulph-OA is composed by sulphate 60%, organics 30%, ammonium 6% and nitrate 4%. Like the OOA_{AMS,org+inorg} factor, sulph OA was also sulphur rich. Its The chemical composition was <u>further</u> explored via the fHSO₃/fH₂SO₄ analysis detailed in Section 3.2.4 and as shown in This factor lay in the upper right quadrant of Figure S16A, the factor is positioned between pure

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H₂SO₄ and and/or (NH₄)₂SO₄. The measured [NH₄]/[SO₄] ratio for sulph-OA-was_0.07, which is much considerably lower than the theoretical mass ratio of 0.38 and 0.18 for of (NH₄)₂SO₄ and NH₄HSO₄, respectively, indicating . Therefore, this factor is inferred to have an acidic nature of Sulph-OA (Chen et al. 2019). The sulph-OA factor was well strongly-correlated with SO₂ (R² = 0.6), which is majorly a primary product of residential heating oil (Dunleavy and Brune, 2020) and moderately correlated with the ResH3 factor (R² of 0.33). The factor was also associated with . Therefore, it is likely that sulph-OA comprises primary ultrafine particles emissions in the 50–80 nm range from heating oil combustion (Figure S12D). Surprisingly, sulph-OA was only moderately correlated with the ResH3 factor (R² of 0.33), which was identified as heating oil OA in the PTR_{CHARON} analysis. Regardless of the low correlation, we speculate that ResH3 and sulph-OA represent originated from the same source, i.e., residential heating fuel oil combustion, and their temporal disagreement may result from instrumental biases of the CHARON inlet in quantifying particles smaller than 100 nm (Figures S12B and D). For instance, as shown in Figures S12E-F, the organic—only ResH3 supersedes sulph-OA concentrations, when larger particles are abundant, and it has lower concentrations for smaller particles.

This factor contained $0.6 \pm 0.5 \,\mu\text{g/m}^3$ of sulphur. Despite the low concentrations, sulph OA made up (~58 ± 26%) of total sulphur sulphate measured with the AMS and because it dominated during the low-pollution periods, which were more frequent and lasted longer than the high-pollution periods events (Figure 1). Other primary factors, HOA_{AMS,org+inorg}, COA_{AMS,org+inorg}, and BBOA_{AMS,org+inorg}, contained an additional $11 \pm 9\%$ of the sulphur sulphate ($0.2 \pm 0.2 \,\mu\text{g/m}^3$), so collectively, Pprimary sulphur factors collectively made up $69 \pm 24\%$ ($0.7 \pm 0.6 \,\mu\text{g/m}^3$) of total sulphur sulphate. This value is in close agreement with a previous ALPACA study that reported ~62 ± 12% of total SO₄²⁻ mass to be of primary origin and associated with particles of smaller than 700 nm ($2.1 \pm 1.4 \,\mu\text{g/m}^3$ in PM_{0.7}) (Moon et al., 2023).

AmNi factor includes atmospherically processed vehicular emissions. The second inorganic factor was composed of 35% nitrates, 14% ammonium, and 43% organics. It accounted accounting for 71 ± 23% of the total nitrate measured by the AMS in NR PM1 (R² = 0.98). The average concentrations of this factor and the nitrate species in it were 1.1 ± 1.6 μg/m³ and 0.4 ± 0.5 μg/m³. The factor was more abundant when NOx concentrations were high (above 130 ppbv)(Figure S17B), its diurnal trend peaked around 1400 AKST (Figure 3), roughly 3-4 hours after the morning peak of HOA_{AMS} and was associated with relatively small particles of 110 nm (Figure S12D). A high contribution of aliphatic moieties characterised the organic fraction, and according to the difference in mass concentrations of HOA_{AMS,org} and HOA_{AMS,org+inorg} of 13% (Figure S17A), we speculate that some organic components of HOA_{AMS,org}

were transferred to the AmNi factor (**Figure 6**). All these elements suggest a probable contribution from the vehicular emissions to this factor. The presence of inorganic compounds provided more variables to the PMF, and thus, improved the resolution of factors into distinct AmNi and HOA_{AMS.org+inorg} factors. It presented a distinct peak from ~1200 1800 hrs and then stable, low concentrations throughout the night (Figure 3). This peak followed 3-4 hours after the peak in the mass concentrations of HOAAMS (or trafficCHARON) during the morning, implying its probable origin from vehicular NOx, which was supported by the highest contributions of this factor coinciding with peaks in NOx concentrations (Figure S17B). Generally, during the ALPACA campaign, the AmNi factor had much lower concentrations than HOAAMS,org+inorg; however, they were both associated with the highest recorded ambient temperatures (5 to -10°C) and solar radiations (as per jNO2 values)(Figure S17C - D). According to atmospheric modelling studies in Fairbanks (Joyce et al., 2014), the formation of NO3 from NOx via the nocturnal reactions slows at temperatures below 15°C, causing them to have higher concentrations during warmer periods. Interestingly, according to the difference in mass concentrations of HOAAMS,org and HOAAMS,org inorg and its correlation with the AmNi factor (Figure S17A), we speculate that some portion of the organic components of of the AmNi factor were apportioned to HOAAMS,org causing it to have higher contributions than HOAAMS,org+inorg (Figure 6). The inclusion of inorganics provided more variables to the PMF, and thus, improved the resolution of factors into distinct AmNi and HOAAMS.org+inorg factors.

4. Local environmental implications and conclusive remarks

For instance During the period of the campaign, during this study, 12–48-hour-long ADEC advisories for wood-burning restrictions were implemented seven times. Variation in the relative contributions of ResH1–4 during these advisories is depicted in **Figures 8** and **S18–21**. For all advisory events, ResH2 and ResH4, i.e., woodsmoke, were the predominant contributors *before* and *after* the advisories were in place. ResH2 (i.e., hardwood-related fuels) remained a prominent contributor to OA_{CHARON} during the 3rd (Stage 2), 4th (Stage 1), and 5th (Stage 1) advisories. A notable increase was observed in ResH3 contribution, i.e., heating oil, at least once during the 2nd (Stage 1), 5th (Stage 1), 6th (Stage 1), and 7th advisory events. While -ResH2 (i.e., hardwood-related fuels) remained a prominent contributor to OA_{CHARON} during the 3rd (Stage 2), 4th (Stage 1), and 5th (Stage 1) advisories. Most households in Fairbanks use heating oil (~72% of residents), followed by wood (~22% of residents) (Dunleavy and Brune, 2019), which was not reflected here in the relative contributions of apportioned to ResH3. This can be linked to a higher PM₁ release from wood combustion per given volume of fuel compared to other commonly used sources, including heating oil and/or an especially under less than optimal combustion conditions (e.g., moist wood) or with inefficient appliances. There is also the possibility

that due to the typical particle size of ResH3 emissions underestimation of ResH3 by PTR_{CHARON} being associated with smaller than 100 nm (**Figure S12**).

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All seven ADEC advisories coincided with the coldest periods of the campaign (Figure 1). Therefore, the response of Fairbanks' residents to ADEC advisories cannot be assessed independently from their response to increased need for heating or the dynamics of OA under the unique meteorology (i.e., low temperatures/low solar radiations/strong inversions) during sampling. In our study As expected, the absolute average concentrations of all factors were inversely related to ambient temperature, but the percent change differed considerably across factors. Specifically, as temperatures decreased from -10°C to below -25°C, the average absolute concentrations for traffictransport_{CHARON}, COA_{CHARON}, OOA_{CHARON}, ResH1-4 increased $0.25 \times$, $0.75 \times$, $9.0 \times$, $1.4 \times$, $25.1 \times$, $3.0 \times$, and $2.9 \times$, respectively (**Figure S22**). The steep increase in the relative contribution of ResH2, was associated with hardwood-tracers, contrasts with previous reportsbased fuels. In contrast, based on surveys (Dunleavy and Brune, 2019) and ratios of organic tracers in ambient air samples (Haque et al., 2021), previous studies reported indicating birch and spruce whichare widely found in Alaskan boreal forests, as the most popular firewood in Fairbanks. Laboratory studies have shown that the burning of softwood pellets of Douglas Fir or eastern white pine emits less PM than hardwood pellets of the same volume, and this response varies based on the moisture content of the wood and the heating appliance used (Morin et al., 2022). High PM emission per volume burned could also be the reason behind hardwood burning being the dominant contributor of PM in our analysis. Also ResH2 comprises a broader spectrum of volatile and semi-volatile substituted phenolic species, and thus, it is likely to undergo efficient gas-to-particle partitioning at low temperatures toward increasing OA loads (Ijaz et al., 2025). Overall, investigating the variation in the emission patterns, especially in response to regulations, such as the ADEC burn restrictions, is a complex issue that requires appropriately acknowledging the influence of meteorology, the physicochemical nature of the emissions, and change in emissions at the source. Based on the observations in this study, it cannot be conclusively inferred that either hardwood or softwood based solid fuels are more popularly consumed wood types in Fairbanks, but they are certainly among the largest contributors to sub-micron OA emissions. These findings are critical to addressing air pollution in Fairbanks, which has been a persistent issue for a long time, by guiding policies and citizen action

5. Conclusion

- A CHARON inlet coupled with PTR-ToF MS and HE-ToF-AMS were deployed during the Alaskan
- 825 <u>Layered Pollution and Chemical Analysis (ALPACA) campaign. The PMF analysis of AMS data</u>
- 826 revealed three primary factors: biomass burning, hydrocarbon-like and cooking factors accounting for

827 28, 38 and 11 % of the total OA, respectively. A combined organic and inorganic PMF analysis provided 828 additional insights and revealed the presence of an organo-sulphate compounds mostly associated to 829 the OOA factor and of another sulphate-rich factor of acidic nature. A nitrate factor, associated with 830 hydrocarbon-like OA and high NOx levels, was interpreted as aged road transport emissions. The 831 PTR_{CHARON} PMF analysis could differentiate four residential heating sources—one oil combustion and 832 three wood combustion types, associated with low temperature, softwood, and hardwood combustion. 833 Such factorisation was achieved with the support of specific tracers that CHARON could successfully 834 identify, as furans, aromatic alcohols (resorcinol, guaiacol, eugenol, syringiol), aldehydes (furfural, 835 coniferaldehyde), acids (benzoic, dehydroabietic, abietic, linoleic, oleic, and stearic) and various PAHs. 836 Collectively all residential heating factors accounting for 79% of the total OA_{CHARON}. Cooking and road 837 transport were also recognised as primary sources by PTR_{CHARON}. All PMF analyses could apportion a single secondary organic fraction accounting for 11-19% of the total OA. This work evidence the 838 839 complementarity of the two instruments and their ability in describing the complex chemical 840 composition of PM₁ and the related sources. The enhanced deconvolution of closely co-varying sources 841 of ambient pollution epitomises the novelty of our study and demonstrates the capability of PTR_{CHARON} 842 to deliver detailed qualitative and quantitative insights, thus enabling a comprehensive understanding 843 of organic aerosol sources. These advances can assist environmental regulators and citizen efforts to 844 improve air quality in Fairbanks and the fast-urbanising regional sub-Arctic areas.

Data availability

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Supporting text, figures, and tables are available in the Supplementary Material.

Author contributions

The manuscript was written with the contributions of all authors. BT-R and BDA set up, ran, and maintained the instrumentation during the campaign in Fairbanks. SA, NB, and ED aided during the campaign. MC-S collected and contributed meteorological and trace gas data. BA, RJW, KD, and AM provided data on ion chromatography analysis of offline filter samples. BT-R and AI processed and analysed the data with help from BC. WS and KS -coordinated the ALPACA and CASPA projects. KL, BDA, BB, SB, JF, JM, and JS contributed to funding acquisition for the CASPA project. BDA supervised the project reported here.

856 Competing interests

The authors declare that they have no conflict of interest.

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