



1     **Aerosol Composition Trends during 2000-2020: In depth insights from model**  
2                                    **predictions and multiple worldwide observation datasets**

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12    **Abstract**

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14    Atmospheric aerosols significantly impact Earth's climate and air quality. In  
15    addition to their number and mass concentrations, their chemical composition  
16    influences their environmental and health effects. This study examines global trends in  
17    aerosol composition from 2000 to 2020, using the EMAC atmospheric chemistry-  
18    climate model and a variety of observational datasets. These include PM<sub>2.5</sub> data from  
19    regional networks and 744 PM<sub>1</sub> datasets from AMS field campaigns conducted at 169  
20    sites worldwide. Results show that organic aerosol (OA) is the dominant fine aerosol  
21    component in all continental regions, particularly in areas with significant biomass  
22    burning and biogenic VOC emissions. EMAC effectively reproduces the prevalence of  
23    secondary OA but underestimates the aging of OA in some cases, revealing  
24    uncertainties in distinguishing fresh and aged SOA. While sulfate is a major aerosol  
25    component in filter-based observations, AMS and model results indicate nitrate  
26    predominates in Europe and Eastern Asia. Mineral dust also plays a critical role in  
27    specific regions, as highlighted by EMAC. The study identifies substantial declines in  
28    sulfate, nitrate, and ammonium concentrations in Europe and North America, attributed  
29    to emission controls, with varying accuracy in model predictions. In Eastern Asia,  
30    sulfate reductions due to SO<sub>2</sub> controls are partially captured by the model. OA trends  
31    differ between methodologies, with filter data showing slight decreases, while AMS  
32    data and model simulations suggest slight increases in PM<sub>1</sub> OA across Europe, North  
33    America, and Eastern Asia. This research underscores the need for integrating advanced  
34    models and diverse datasets to better understand aerosol trends and guide  
35    environmental policy.

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39 **1. Introduction**

40 Atmospheric aerosols are tiny solid or liquid particulate matter (PM) suspended in  
41 the air, ranging in size from a few nanometers to several micrometers. Atmospheric  
42 aerosol, especially fine particles with diameters less than 2.5 micrometers (PM<sub>2.5</sub>),  
43 poses health risks as it can penetrate deep into the respiratory system (Who, 2003).  
44 Long-term exposure to high levels of PM has been associated with respiratory and  
45 cardiovascular diseases (Brook et al., 2010; George et al., 2017). Dominici et al. (2006)  
46 and Pope et al. (2009) highlight the impact of PM on mortality and morbidity, while  
47 more recent studies have determined that the air pollution by PM<sub>2.5</sub> is responsible for  
48 more than 3 million premature deaths per year worldwide (Lelieveld et al., 2015; Who,  
49 2022). As a result, air pollution is recognized as the largest environmental threat to  
50 human health in the recent WHO report (Who, 2021). Furthermore, aerosols can  
51 directly influence the Earth's climate by scattering and absorbing sunlight, leading to  
52 changes in radiation balance (Haywood and Boucher, 2000; Ipcc, 2013). Aerosols can  
53 also affect the Earth's energy balance indirectly through interactions with clouds, i.e.,  
54 by serving as cloud condensation (CCN) and ice (IN) nuclei, affecting cloud formation,  
55 cloud properties, and precipitation patterns (Andreae and Rosenfeld, 2008). Beside the  
56 number and mass concentrations of atmospheric aerosol, its chemical composition  
57 determines its aerosol-related climatic (Klingmuller et al., 2019; Klingmüller et al.,  
58 2020; Kok et al., 2023) and health impacts (Lelieveld et al., 2015; Fang et al., 2017;  
59 Karydis et al., 2021).

60 Atmospheric aerosols have various precursors, and they can be categorized into  
61 primary and secondary aerosols based on their origin. Primary sources include natural  
62 processes such as volcanic eruptions, wildfires, and sea spray, as well as human  
63 activities like industrial emissions and transportation. Secondary aerosols are formed  
64 through the oxidation of gas phase pollutants in the atmosphere. Sulfate aerosols are  
65 formed through the oxidation of sulfur dioxide (SO<sub>2</sub>) which is primarily released from  
66 the burning of fossil fuels, particularly coal, and natural sources like volcanoes. Nitrate  
67 aerosols result from the atmospheric oxidation of nitrogen oxides (NO<sub>x</sub>) emitted from  
68 combustion processes, such as those in vehicles and power plants. Ammonium is  
69 formed by the reaction of ammonia (NH<sub>3</sub>), which is emitted from agricultural activities  
70 and waste treatment, with an acid. Secondary organic aerosols (SOA) can form through  
71 the oxidation of volatile organic compounds (VOCs), which are emitted from  
72 vegetation, industrial processes, and vehicle exhaust.



73 Several measures have been discussed and implemented to mitigate pollutants  
74 emitted from specific source sectors including transport, energy (power generation,  
75 industries etc.), waste management, urban planning and agriculture. A few of the most  
76 prominent global conferences that have taken place for the purpose of combating  
77 climate change and air pollution are the Conferences of the Parties (COP) since the  
78 early 90s, and the supreme decision-making body of the United Nations' Framework  
79 Convention on Climate Change (UNFCCC). Their passed agreements binding the  
80 parties to individual emission targets are for instance the Agenda 21 of 1992, the Kyoto  
81 Protocol of 1997 and its successor - the Paris Agreement of 2015. Besides these global  
82 agreements, the single parties had to implement national or continental plans to meet  
83 air quality requirements. The resulting emission trends have been so drastic that aerosol  
84 composition has been unevenly altered in different parts of the world. Most European  
85 countries are bound by the Gothenburg Protocol targets from 1999 and its amendment  
86 from 2012 and have in majority successfully reduced pollutant levels (Emep, 2021).  
87 SO<sub>x</sub> emissions have declined the most, by more than 80% in the last two decades. NO<sub>x</sub>  
88 emissions have declined significantly as well (by 50%), but for NH<sub>3</sub> only very small  
89 reductions have been achieved (~10%) (Hoesly et al., 2018; Emep, 2021). NMVOCs  
90 have also been significantly decreased due to emission controls to the transportation  
91 and the solvents sector (Hoesly et al., 2018). In the US, pollutant levels are controlled  
92 through regulations imposed by the National Ambient Air Quality Standards  
93 (NAAQS), the Regional Haze Rule and the US Clean Air act of 1970. The US and  
94 Canada are also part of the Gothenburg protocol. Over Asia, South Korea and China  
95 belong to the Newly Industrialized and high-growth economies. Especially from 1980  
96 to the mid-2000s, pollutants emissions grew in China (Hoesly et al., 2018; Zhai et al.,  
97 2019). However, in the face of the Beijing Olympic Games in 2008, there have been  
98 drastic endeavors of air pollution control in Beijing and neighboring administrative  
99 regions (Huang et al., 2010). In 2013, the first consistent and aggressive emission  
100 controls started under the Clean Air Action (Zhai et al., 2019). The Clean Air Action  
101 has identified three target regions, the megacity clusters of Beijing-Tianjin-Hebei,  
102 Yangtze River Delta and the Pearl River Delta, while in 2018, the latter was replaced  
103 by the Fenwei Plain (Zhai et al., 2019).

104 Air pollution concentration levels can vary by time of day, season, across large spans  
105 of time, based on meteorological factors, and in connection to climate change. Trends  
106 analysis of air pollution concentrations (Guerreiro et al., 2014; Lang et al., 2019) can



107 allow the assessment of the impact of various factors on air quality including changes  
108 in industrial activities, traffic patterns, or energy production. Analyzing trends in air  
109 pollutants enables comparisons between different regions or countries (Anttila and  
110 Tuovinen, 2010; Chow et al., 2022; Kyllönen et al., 2020) as well as between different  
111 datasets that provide information for the same pollutant. This can highlight areas that  
112 are successfully addressing air quality issues, provide benchmarks for others to follow  
113 but also highlight any kind of inability of each method to reproduce the concentration  
114 levels of the pollutants.

115 In this study, we use the comprehensive atmospheric chemistry-climate model  
116 EMAC to present 20-year global composition trends of fine aerosols in different regions  
117 of the planet. Here, for the first time, EMAC uses a computationally lite version of the  
118 organic aerosol module ORACLE (Tsimpidi et al., 2014) and the new highly  
119 computationally efficient module ISORROPIA-lite (Kakavas et al., 2022; Milousis et  
120 al., 2024). The large emission trends in our model are considered by employing the  
121 Copernicus Atmosphere Monitoring Service (CAMS) inventory for anthropogenic  
122 emissions (Granier et al., 2019). Model results are combined with a global observational  
123 aerosol composition dataset to provide insights into the large spatiotemporal changes  
124 in aerosol composition over the past two decades, driven by changes in aerosol  
125 precursor emissions. The dataset includes observations from regional filter-based  
126 monitoring networks that routinely collect  $PM_{2.5}$  (e.g. EMEP, IMPROVE, EPA,  
127 EANET, SPARTAN), and a unique comprehensive compilation of 744 individual  
128 Aerosol Mass Spectrometer (AMS) field campaigns worldwide that provide in-situ  
129 measurements of  $PM_1$  composition.

130

## 131 **2. Observational Dataset**

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### 133 **2.1 $PM_1$ Dataset**

134 Since the year 2000, the quadrupole-based Aerodyne aerosol mass spectrometer (Q-  
135 AMS) and its successors enjoy great popularity as a method for atmospheric aerosol  
136 sampling. A great advantage of AMS is its ability to deliver high-resolved real-time  
137 quantitative data on mass concentration of particles between  $\sim 0.05 - 1 \mu m$   
138 (Canagaratna et al., 2007) as a function of their non-refractory chemical composition  
139 (i.e., OA and inorganic  $SO_4^{2-}$ ,  $NO_3^-$ ,  $NH_4^+$ , and  $Cl^-$ ) (Jayne et al., 2000). Thus, over the  
140 years and numerous field campaigns, a lot of valuable chemical and microphysical



141 information about ambient aerosols has been obtained (Ng et al., 2011). During 2000s,  
142 these campaigns did not last more than a month, however, the development of the  
143 Aerosol Chemical Speciation Monitor (ACSM), a small and cost-efficient version of  
144 AMS, allowed the long-term monitoring of the PM<sub>1</sub> composition over several locations  
145 during the 2010s.

146

### 147 **2.1.1 AMS factor analysis techniques**

148 The AMS spectra of OA are often further analyzed via factor analysis techniques in  
149 order to extract detailed information about the OA composition as well. Among factor  
150 analysis techniques (e.g., ME-2 (Paatero, 1999); PCA (Zhang et al., 2013); MCA  
151 (Zhang et al., 2007; Cottrell et al., 2008)), the PMF (Paatero and Tapper, 1994; Paatero,  
152 1997) is the most popular technique, occasionally in combination with the ME-2.  
153 Overall, a mass spectrum that peaks at  $m/z = 44$  (or  $f_{44}$ ) and  $m/z = 43$  (or  $f_{43}$ ) is mostly  
154 dominated by the  $\text{CO}_2^+$  and  $\text{C}_2\text{H}_3\text{O}^+$  ions, respectively. The first is mostly linked to  
155 acidic groups (i.e., -COOH), typically associated with chemically aged and oxygenated  
156 organic aerosols (OOA), while the latter is dominated by non-acid oxygenates. OOA  
157 can be further categorized into different levels of aging and volatility stages. Most  
158 commonly, a less oxidized (semi-volatile) OA (L-OOA (Bougiatioti et al., 2014)) and  
159 a more oxidized (low-volatile) OA (M-OOA (Bozzetti et al., 2017)) are distinguished  
160 (Jimenez et al., 2009; Ng et al., 2010; Crippa et al., 2014; Stavroulas et al., 2019). The  
161 two OOA factors could be identified on the basis of the  $f_{44}$  to  $f_{43}$  ratio: M-OOA  
162 component spectra have a higher  $f_{44}$ , while L-OOA component spectra have slightly  
163 higher  $f_{43}$ . Besides these general factors, other oxygenated OA compounds have been  
164 resolved in some campaigns. One of the most important is the IEPOX-OA with  
165 abundant ions at  $m/z = 53$ , 75, or 82. This “isoprene” factor correlates strongly with  
166 molecular tracers of SOA that are derived from isoprene epoxydiols (Xu et al., 2015;  
167 Budisulistiorini et al., 2013; Budisulistiorini et al., 2016). Several campaigns in North  
168 America have found IEPOX-OA, as have campaigns in South America and Australia.  
169 Furthermore, methane-sulfonic acid (MSA) is often retrieved from datasets of marine  
170 sites (Crippa et al., 2014; Mallet et al., 2019). Some studies could identify a nitrogen-  
171 enriched OA-factor, NOA, mainly composed of amino compounds formed from  
172 industrial or marine emissions. A more local-SOA factor that is related to humic-like  
173 substances, termed as HULIS OA, found in the Netherlands (Schlag et al., 2016) and  
174 in Crete (Crippa et al., 2014). In Greece (Bougiatioti et al., 2014; Stavroulas et al., 2019;



175 Vasilakopoulou et al., 2023), in the Amazonian (De Sá et al., 2019) and often in Asia  
176 (Zhang et al., 2015b; Chakraborty et al., 2015; Du et al., 2015) OOA factors directly  
177 associated with biomass burning were found, that are processed from fresh biomass  
178 burning emissions. Furthermore, OOA compounds that are verifiable only biogenically  
179 oxygenated were also derived (Kostenidou et al., 2015).

180 Apart from the mass spectrum, OA types can also be distinguished by their oxygen  
181 to carbon ratio (O:C), which is an indicator of photochemical aging. Primary organic  
182 aerosol (POA) is fresh and has a lower oxygen content than OOA, therefore lower O:C  
183 ratios. Yet, it sometimes has the same dominant m/z peaks. Some of the most commonly  
184 resolved POA factors are the Hydrocarbon-like (HOA) and Biomass Burning (BBOA)  
185 OA. HOA has spectra that are distinguished by clear hydrocarbon signatures,  
186 dominated by the ion series  $C_nH_{2n+1}^+$  and  $C_nH_{2n-1}^+$  (Ng et al., 2010). HOA correlates  
187 with fossil fuel combustion tracers like  $NO_x$ , CO and elemental carbon (Lanz et al.,  
188 2008; Tsimpidi et al., 2016), therefore, is very often observed to be traffic-related and  
189 a rather dominant POA factor in urban areas (Crippa et al., 2014; Xu et al., 2015;  
190 Budisulistiorini et al., 2016). On the other hand, BBOA typically originates from forest  
191 and savanna fires as well as from anthropogenically induced agricultural fires (Hoesly  
192 et al., 2018) and residential wood burning for heating. This makes the contribution of  
193 BBOA to total OA highly episodic (Zhang et al., 2007) and seasonal, and in several  
194 cases underestimated due to the rapid physicochemical transformation of these  
195 emissions to OOA (Stavroulas et al., 2019; Vasilakopoulou et al., 2023). Typical tracers  
196 to identify BBOA in the spectra are gas-phase acetonitrile, particle-phase levoglucosan  
197 and potassium ( $K^+$ ) (Lanz et al., 2010; Crippa et al., 2014). However, its mass spectra  
198 are also highly variable since they can be affected by different types of wood and  
199 burning conditions (Crippa et al., 2014).

200 Furthermore, a coal combustion factor (CCOA) is often identified, which presents a  
201 dominant contribution to POA during the heating season, mostly in Eastern Asia (Sun  
202 et al., 2013; Zhang et al., 2014). In many cases, HOA shows remarkably similar spectral  
203 patterns as CCOA, so that these two factors could not be separated and, instead, are  
204 combined in a fossil fuel related OA factor (FFOA) (Sun et al., 2018; Xu et al., 2019).  
205 Another relatively frequent primary type resolved by the factor analysis is the cooking  
206 related OA (COA) (Mohr et al., 2012). Its spectral pattern is governed by OA from  
207 fresh cooking emissions and, fittingly, the spectral profiles have a distinct diurnal cycle  
208 which corresponds to typical (local) meal hours (Mohr et al., 2012; Sun et al., 2013;

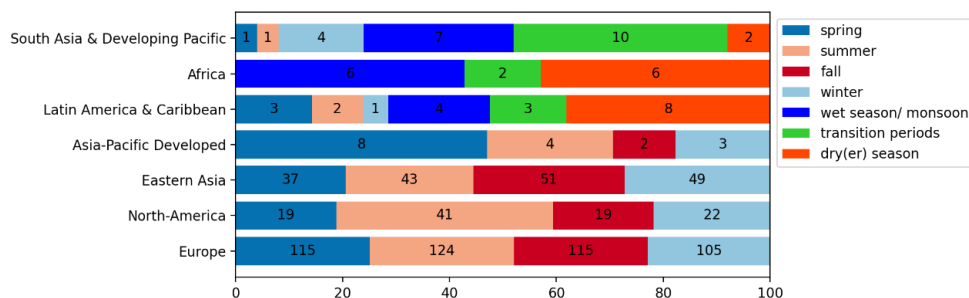


209 Stavroulas et al., 2019). Occasionally, special types of COA are also resolved, including  
210 coffee roastery OA (Timonen et al., 2013) and OA related to charbroiling (Lanz et al.,  
211 2007).

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### 213 2.1.2 AMS Dataset

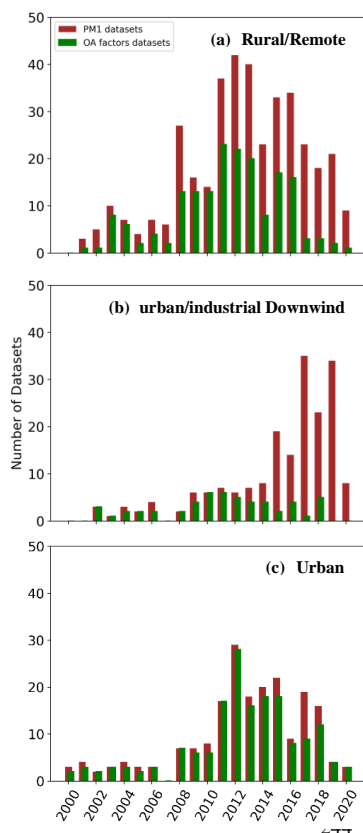
214 Here, a collection of AMS and ACSM field campaign datasets during the period of  
215 2000-2020 has been compiled. The dataset covers a wide range of environments and  
216 seasons from almost every continental region worldwide (Figure 1), characterized by a  
217 variety of atmospheric and climatological conditions as well as sources of pollutants.  
218 The selected field campaigns lasted from at least one full week to several months.  
219 Individual campaigns lasting more than one month are divided into shorter periods of  
220 preferably only one month. All of these individual periods of campaign data (thus  
221 covering a maximum of one month) are hereafter referred to as individual datasets.



**Figure 1:** Seasonal distribution of datasets per subcontinent. The colored bars indicate the relative proportions by season. The numbers in the colored boxes indicate the absolute number of field campaigns that occurred in each season.

222

223 The number of both PM<sub>1</sub> and OA composition datasets found for each year is  
224 increasing significantly for all regions through the years (Figure 2) due to the growing  
225 popularity of the AMS devices and the continuous improvement of the analysis



**Figure 2:** Total AMS (dark red) and factor analysis (green) datasets per year in (a) rural, (b) urban-downwind, and (c) urban regions

techniques. Especially during the second decade, the number of field campaigns increase drastically, supported by the use of ACSM devices since 2010. The long-term campaigns in South Africa (2010-2011; (Tiitta et al., 2014)) and the Southern Great Planes (2010- 2012; (Parworth et al., 2015)) belong to the very first where the ACSM has been utilized. Furthermore, campaigns in regions downwind of urban environments have gotten a growing attention mostly after 2014, primarily in Europe. However, usually these datasets are not factor analyzed and lack information for the OA composition. It is worth mentioning that the small number of downwind datasets available can partially attributed to the ambiguous definition of downwind sites, which might have led instead to the more conventional classifications of rural or urban locations in some cases.

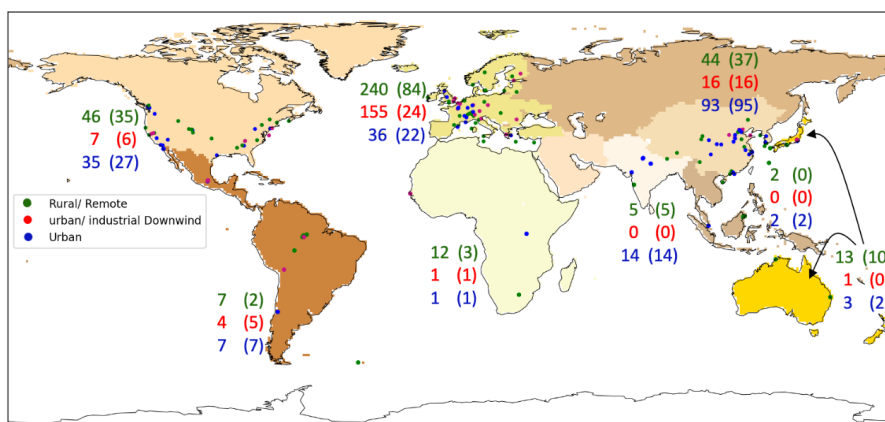
Overall, the compiled dataset includes PM<sub>1</sub> aerosol composition from 744 AMS field campaigns datasets at 169 observational sites around the world, while factor analysis has been used to estimate the OA composition in 398 cases

249 at 140 different observational sites (Table S1). The dataset includes an intermediate  
 250 level regional breakdown following the sixth assessment report of IPCC working group  
 251 III (Ipc, 2022) as shown in Figure 3. The most represented subcontinents are Europe,  
 252 Eastern Asia and North America. Datasets from these three northern-hemisphere  
 253 continents are more or less evenly distributed over the seasons with only a little  
 254 imbalance for North America which is over-represented during summer (Figure 1). The  
 255 rest of the regions include a significantly lower number of datasets; therefore, the  
 256 seasonal distribution is often very uneven. As an example, 50% of the data over the  
 257 Asia-Pacific Developed region has been collected during spring. On the contrary, the  
 258





259 changes between the wet and dry seasons are well represented over Africa where the  
260 ACSM has been employed for year-long campaigns (Tiitta et al., 2014).

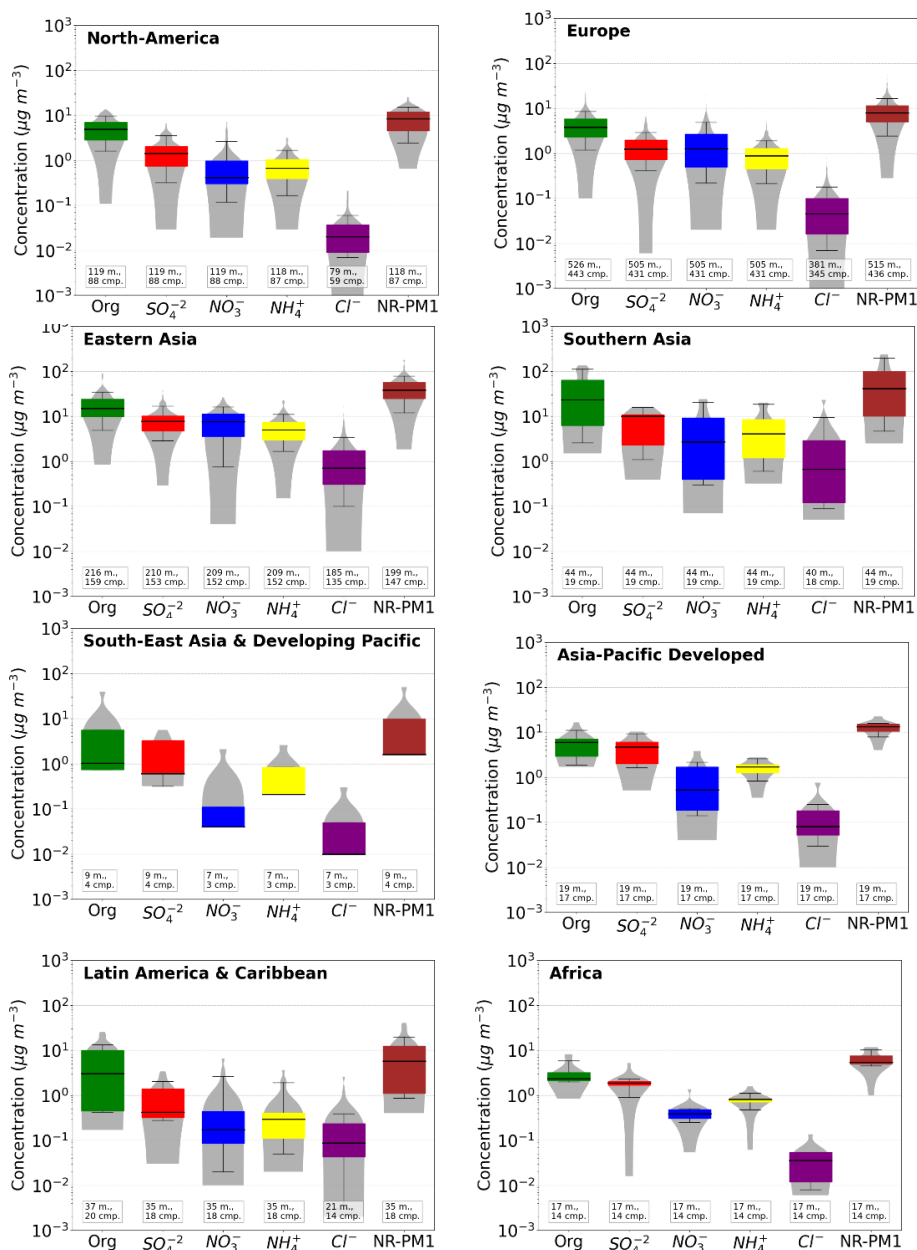


**Figure 3:** Worldwide distribution of AMS and ACSM datasets for the of period 2000 - 2020. The world map is colored according to the intermediate level regional breakdown of the sixth assessment report of IPCC working group III (IPCC, 2022). The rural (green), downwind (red) and urban (blue) campaign locations and the total number of PM<sub>1</sub> composition (and OA factor analysis in parenthesis) datasets for each region are also shown.

261

### 262 2.1.3 Observed PM<sub>1</sub> Aerosol Composition

263 The PM<sub>1</sub> aerosol composition derived from AMS field campaigns at 8 regions  
264 around the world is depicted in Figure 4. The analysis of the AMS dataset reveals that  
265 OA is the dominant component of PM<sub>1</sub> in all continental regions. Campaign data from  
266 tropical or subtropical regimes (e.g., Latin America and Southern/Southeast Asia) is  
267 strongly affected by biomass burning and biogenic VOC emissions, illustrating  
268 remarkably high OA fractions with regional means around 65% and a maximum of  
269 92% in the Amazonian. However, OA concentration shares up to 90% are also found  
270 over the Northern Hemisphere regions where the regional average OA contribution to  
271 PM<sub>1</sub> concentrations is around 50%. Overall, OA contributes between 17 - 92% (50%  
272 on average) of total PM<sub>1</sub>. This agrees well with the ranges reported by Kanakidou et al.  
273 (2005) (20%-90%) and Zhang et al. (2007) (18%-70% or 45% on average). Sulfate has  
274 been the dominant inorganic compound in the aerosol composition in most regions



**Figure 4:** Bar chart plots depicting the distribution (violin) and the 25<sup>th</sup>, 50<sup>th</sup> and 75<sup>th</sup> percentiles (box) of the mass concentration (in  $\mu\text{g m}^{-3}$ ) for the major PM<sub>1</sub> aerosol components, i.e., organic aerosol (green), sulfate (red), nitrate (blue), ammonium (yellow), chloride (purple), and the total non-refractive PM<sub>1</sub> (dark red). The 10<sup>th</sup> and 90<sup>th</sup> percentiles (whiskers) for each aerosol component are also shown. The number of total months (m.) with AMS data and the number of campaigns (cmp.) is written in small boxes under the violins.



276 (Figure 4). The highest regional average share of sulfate is found over Asia-Pacific  
277 Developed (37%) while the lowest over Europe (17%) where SO<sub>2</sub> has been drastically  
278 reduced due to strict air pollution mitigation strategies. Nitrate dominates over sulfate  
279 over Europe and Eastern Asia. However, it is surprising that the PM<sub>1</sub> inorganic  
280 composition of North America is dominated by sulfate, even though similar mitigation  
281 strategies have been enforced as in Europe. This might be due to an over-representation  
282 of summer data in North America (Figure 1) which resulted in lower nitrate  
283 concentrations since higher temperatures hinder the condensation of nitric acid in the  
284 aerosol phase. At the same time, sulfate concentrations are higher during summer due  
285 to the increased photochemical production of H<sub>2</sub>SO<sub>4</sub>. Overall, nitrate concentrations are  
286 highest in winter in Europe and North America, accounting for roughly a quarter of  
287 total PM<sub>1</sub> (Figures S1 and S2). A similar proportion is observed in spring, although the  
288 absolute concentration is lower. The lowest average nitrate concentrations and shares  
289 occur in summer, when sulfate peaks and dominates the inorganic composition.  
290 Although both sulfate and nitrate are generated through photochemical reactions, this  
291 seasonal shift is due to nitric acid remaining in the gas phase at higher temperatures.  
292 Additionally, the increased production of sulfuric acid reduces the amount of free  
293 ammonia available for ammonium nitrate formation, further contributing to the summer  
294 nitrate decline (Seinfeld and Pandis, 2006). Ammonium concentrations remain  
295 relatively stable throughout the seasons, presenting similar shares of PM<sub>1</sub> (Figures S1  
296 and S2). However, in contrast to Europe and North America, sulfate concentrations in  
297 East Asia are highest in winter, closely followed by summer (Figure S3). While  
298 photochemical reactions still dominate during warmer, sunnier seasons, aqueous phase  
299 reactions are more influential in East Asian winter, particularly under high relative  
300 humidity (RH) and severe haze conditions. These factors are often present in Chinese  
301 winters and likely explain this regional pattern (Zhang et al., 2015a; Zhou et al., 2020a).  
302 Over the southern regions, ammonium follows sulfate in the inorganic aerosol  
303 composition due to the high agricultural activities. Overall, the global average  
304 contribution of the inorganic compounds to total PM<sub>1</sub> concentration is 20%, 18%, and  
305 11%, and 1% by sulfate, nitrate, ammonium, and chloride, respectively. However,  
306 Zhang et al. (2007) reported much stronger contribution by sulfate (32%), less by nitrate  
307 (10%), and similar values of ammonium (13%) and chloride (1%). Given that Zhang et  
308 al. (2007) utilized AMS observations from the early 2000s, this is a first indication that  
309 the inorganic aerosol composition has been altered during the last 20 years.



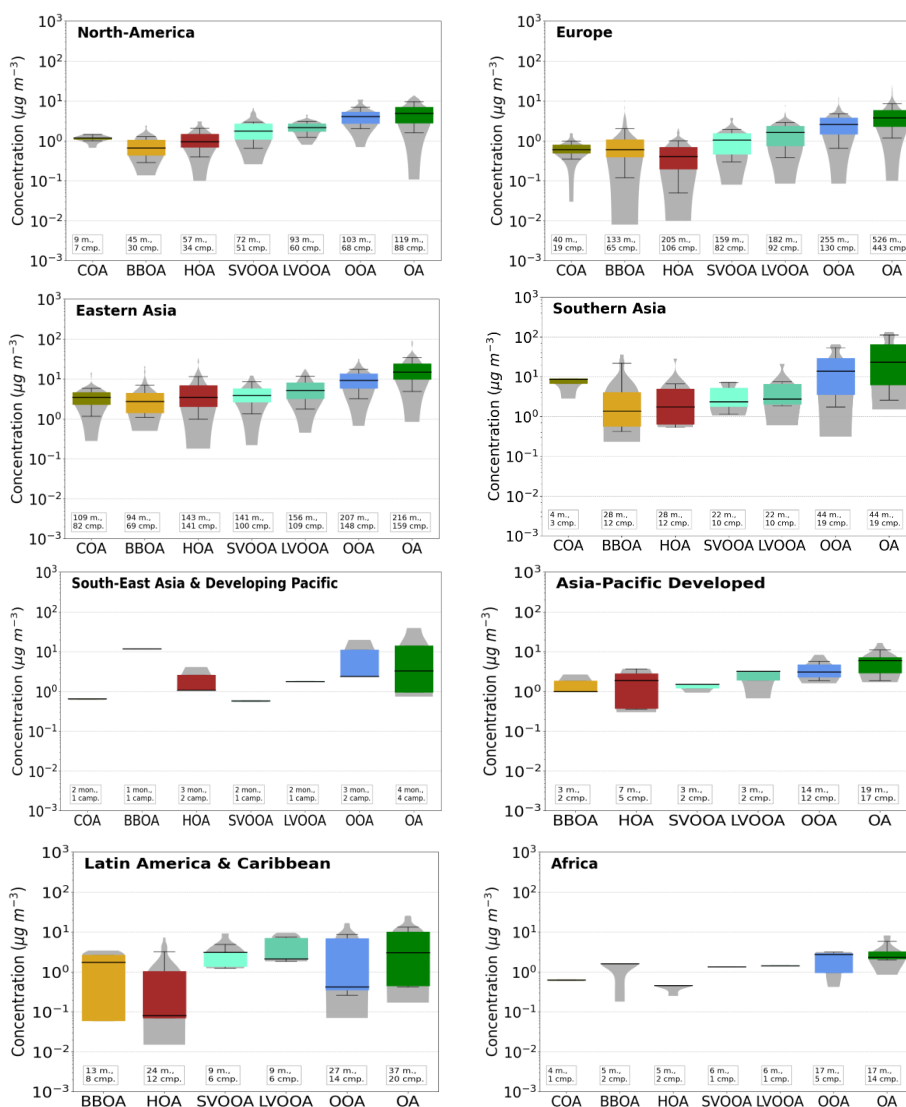
#### 310 **2.1.4 Observed PM<sub>1</sub> Organic Aerosol Composition**

311 HOA concentrations are observed to be higher over North America and Eastern Asia  
312 in comparison to Europe (Figure 5). This could be explained by the significant influence  
313 of traffic emissions on HOA in the vicinity of urban areas. While urban locations are  
314 equally represented with rural sites in the dataset collection of North America and  
315 Eastern Asia, in Europe, rural sites are immensely over-represented (3 times more than  
316 urban sites), diminishing the importance of HOA. On the other side, the over-  
317 representation of rural sites in the European dataset resulted in high concentrations of  
318 BBOA which is found to be the dominant primary source of OA in the region (Lanz et  
319 al., 2010). Here, BBOA originates mostly from domestic wood burning during the  
320 colder seasons in central Europe, including the Alps, rather than from open biomass  
321 burning. Even though a few campaigns took place in the European boreal forests, only  
322 very few factor analyses have distinguished BBOA as an individual component. Thus,  
323 the contribution of European boreal forests to total European BBOA is unfortunately  
324 not clear yet. Similarly, biomass burning is an important source of OA in North  
325 America and Eastern Asia (Rattanavaraha et al., 2017; Zhou et al., 2020b) but less  
326 important than HOA (Figure 5). Biomass burning also presents an especially important  
327 source in tropical and subtropical regions (i.e., South Asia and the Developing Pacific,  
328 Africa, and Latin America and Caribbean) due to episodic wildfires and harvest related  
329 burning (Budisulistiorini et al., 2018; Cash et al., 2021). Overall, the concentration  
330 range of BBOA is very high since it varies a lot with season. However, it should be  
331 emphasized that the availability of factor analysis datasets in equatorial and southern  
332 hemisphere continents is very low and therefore, there is not enough data available for  
333 statistically profound statements. The last primary type of OA, COA, is population  
334 dependent and therefore is mainly found in urban areas and highly populated regions  
335 (Zhou et al., 2020b). Cooking is a very constant and local source throughout the year  
336 with low variability and high contributions over Eastern Asia, Europe, North America,  
337 and South Asia and developing Pacific, especially in urban campaign sites.

338 OOA is unequivocally the dominant contributor to total OA with a mean share of  
339 60% in urban and 75% in rural regions. Overall, the OOA contribution range from 19%  
340 (urban minimum) to 99% (rural maximum). The extreme shares were both found during  
341 European campaigns. The mean OOA share in Europe however lies roughly in the same  
342 magnitude as the global mean (~70%). The dominant OOA subfactors resolved are L-  
343 OOA and M-OOA, while the more aged M-OOA dominates in the OA composition of



344 all examined regions (~60% of total OOA). This agrees with the findings of Ng et al.  
 345 (2010), who stated that OOA component spectra become increasingly similar to each  
 346 other with atmospheric oxidation, indicating that ambient OA converges towards highly  
 347 aged M-OOA.



**Figure 5:** Bar chart plots depicting the distribution (violin) and the 25<sup>th</sup>, 50<sup>th</sup> and 75<sup>th</sup> percentiles (box) of the mass concentration (in µg m<sup>-3</sup>) for the major PM<sub>10</sub> OA components calculated from the collected factor analysis datasets, i.e., COA (olive green), BBOA (orange), HOA (dark red), L-OOA (light turquoise), M-OOA (dark turquoise), OOA (blue), and total OA (green). The 10<sup>th</sup> and 90<sup>th</sup> percentiles (whiskers) for each aerosol component are also shown. The number of datasets (m.) and the number of campaigns (cmp.) is written in small boxes under the violins.

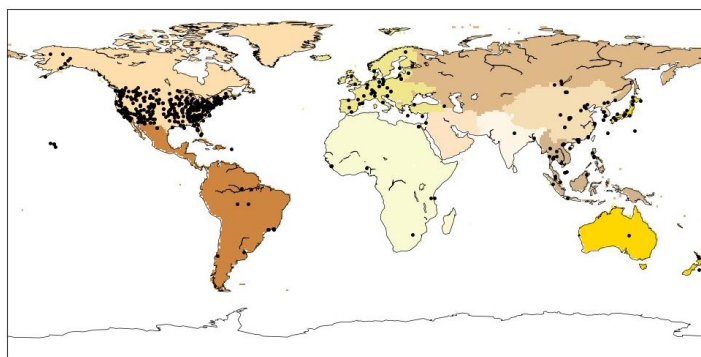


## 348 2.2 PM<sub>2.5</sub> Dataset

349 Routine filter measurement PM<sub>2.5</sub> data from large observational networks in East  
350 Asia, Europe and North America is used. The filter samplers have three modules that  
351 independently collect PM<sub>2.5</sub> species on a Teflon, a nylon and a quartz filter. The aerosol  
352 chemical composition is determined by further analysis of the filters in the laboratory  
353 via ion chromatography (inorganic ions), thermal-optical analysis (OC and EC), and X-  
354 ray fluorescence (XRF; trace elements) (Solomon et al., 2014). Potential difficulties  
355 that could arise when comparing on-line AMS and ACSM PM<sub>1</sub> composition to off-line  
356 filter based PM<sub>2.5</sub> composition, are discussed in section 5. The Environmental  
357 Protection Agency (EPA) network includes 211 monitor sites primarily in urban areas  
358 of North America. The data used here cover monthly averaged PM<sub>2.5</sub> aerosol  
359 component measurements during 2000-2018  
360 ([https://aqs.epa.gov/aqsweb/airdata/download\\_files.html](https://aqs.epa.gov/aqsweb/airdata/download_files.html)). The Interagency  
361 Monitoring of Protected Visual Environments (IMPROVE) network includes 198  
362 monitoring sites that are representative of the regional haze conditions over North  
363 America. IMPROVE samplers collect 24-hour samples, every three days. The data used  
364 here cover monthly averaged PM<sub>2.5</sub> aerosol component measurements during 2000-  
365 2018 (<http://views.cira.colostate.edu/fed/QueryWizard/Default.aspx>). It is worth  
366 mentioning that ammonium measurements by IMPROVE are only available until the  
367 year 2006. The European Monitoring and Evaluation Programme (EMEP) network  
368 monitors the long-range transmission of air pollutants in Europe and Eastern Eurasia  
369 (Figure 6). This network includes 70 monitoring sites. The data used here cover  
370 monthly averaged PM<sub>2.5</sub> aerosol component measurements during 2000-2018  
371 (<https://www.emep.int/>). Finally, the Acid Deposition Monitoring Network in East Asia  
372 (EANET) network includes 39 (18 remote, 10 rural, 11 urban) air concentration monitor  
373 sites in Eurasia, Eastern Asia, South-East Asia and Developing Pacific, and Asia-  
374 Pacific Developed. The data used here cover monthly averaged PM<sub>2.5</sub> aerosol  
375 component measurements during 2001-2017 (<https://www.eanet.asia/>). The global  
376 particulate matter network SPARTAN (Snider et al., 2015; Snider et al., 2016) includes  
377 a global federation of ground-level PM<sub>2.5</sub> monitors situated primarily in highly  
378 populated regions around the world (i.e., North America, Latin America and Caribbean,  
379 Africa, Middle East, Southern Asia, Eastern Asia, South-Eastern Asia and Developing  
380 Pacific) (Figure 6). The data used here covers monthly averaged PM<sub>2.5</sub> aerosol  
381 component measurements of sulfate, nitrate, ammonium and sodium during 2013-2019



382 (<https://www.spartan-network.org/>). Finally, PM<sub>2.5</sub> aerosol component measurements  
383 from individual observational field campaigns over Latin America and Caribbean,  
384 Africa, Europe, Eastern Asia, and Asia-Pacific Developed reported as campaign  
385 averages in the literature are used ( Wang et al., 2019; Radhi et al., 2010; Favez et al.,  
386 2008; Mkoma, 2008; Mkoma et al., 2009; Weinstein et al., 2010; Celis et al., 2004;  
387 Bourotte et al., 2007; Fuzzi et al., 2007; Mariani and De Mello, 2007; Martin et al.,  
388 2010; Souza et al., 2010; Gioda et al., 2011; Molina et al., 2010; Molina et al., 2007;  
389 Kuzu et al., 2020; Aggarwal and Kawamura, 2009; Batmunkh et al., 2011; Cho and  
390 Park, 2013; Feng et al., 2006; Li et al., 2010; Pathak et al., 2011; Zhang et al., 2012;  
391 Zhao et al., 2013).

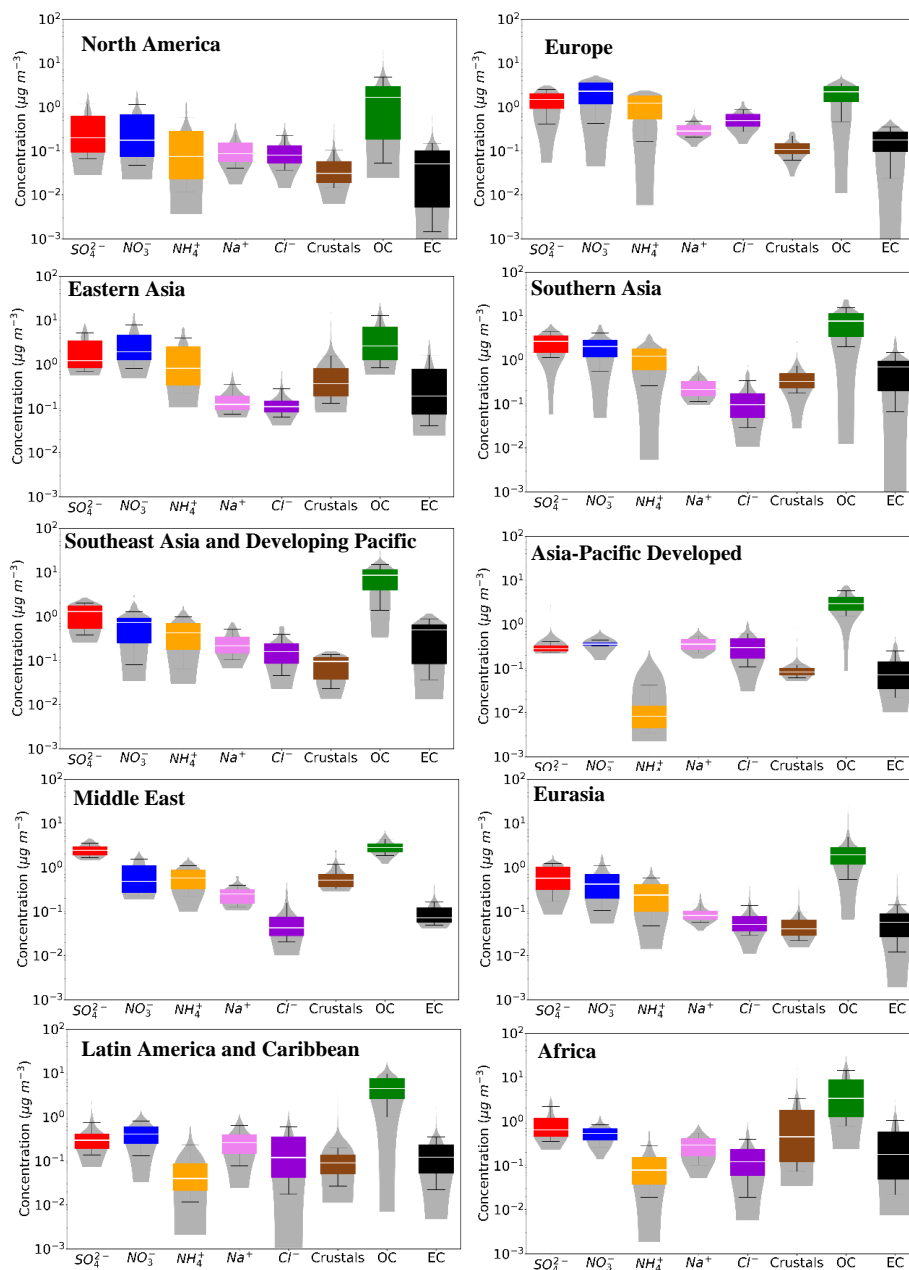


392

**Figure 6:** Worldwide distribution of filter-based observations for the period of 2000-2020. The world map is colored following the intermediate level regional breakdown of the sixth assessment report of IPCC working group III (IPCC, 2022). The black dots correspond to the location of the monitor stations.

### 393 2.2.1 PM<sub>2.5</sub> Aerosol Composition

394 The PM<sub>2.5</sub> aerosol composition derived from filter observations around the world is  
395 depicted in Figure 7. OA is the dominant component of PM<sub>2.5</sub> in most regions,  
396 especially over regions affected by the tropical forests of the southern hemisphere (e.g.,  
397 Latin America & Caribbean and Africa). Over the Northern Hemisphere, OA and EC  
398 dominate the aerosol composition in Eastern Asia (54% and 22% of total PM<sub>2.5</sub>,  
399 respectively) and contribute significantly to PM<sub>2.5</sub> over Europe (30% and 5% of total  
400 PM<sub>2.5</sub>, respectively). On the other hand, over North America, OA share is equally  
401 important to sulfate over rural areas (28% of total PM<sub>2.5</sub> each) and less important over  
402 urban areas (24% versus 33% of sulfate). Indeed, sulfate is the most important inorganic  
403 component of PM<sub>2.5</sub> around the world (~50% of the inorganic PM<sub>2.5</sub> mass on average)  
404 followed by nitrate and ammonium (~20% each). This contradicts the results from AMS



**Figure 7:** Bar chart plots depicting the distribution (violin) and the 25<sup>th</sup>, 50<sup>th</sup> and 75<sup>th</sup> percentiles (box) of the mass concentration (in  $\mu\text{g m}^{-3}$ ) for the major  $\text{PM}_{2.5}$  aerosol components, i.e., sulfate (red), nitrate (blue), ammonium (yellow), sodium (pink), chloride (purple), crustal ions (brown), organic aerosol (green), and elemental carbon (black). The 10<sup>th</sup> and 90<sup>th</sup> percentiles (whiskers) for each aerosol component are also shown.





406 campaigns showing that ammonium nitrate surpasses ammonium sulfate in the aerosol  
407 composition, especially over Europe and North America. However, filter measurements  
408 are prone to negative sampling artifacts due to evaporation losses of the semivolatile  
409 ammonium nitrate under warm and dry conditions (Ames and Malm, 2001), in contrast  
410 to the nonvolatile sulfate aerosols (Docherty et al., 2011). The contribution of sulfate  
411 to the measured inorganic PM<sub>2.5</sub> aerosol composition is highest over Middle East, while  
412 nitrate contributes significantly over Europe (Figure 7). The dominant inorganic ion  
413 varies with the season (Figures S1-S3). Nitrate is most important in winter, accounting  
414 for about a quarter of total PM<sub>2.5</sub>, while sulfate is the dominant PM<sub>2.5</sub> component in  
415 summer and spring. Over the 8 regions where all 7 components are measured, the  
416 average contribution of each species to total PM<sub>2.5</sub> concentration is 21%, 12%, 10%,  
417 2%, 3%, and 40%, and 12% by sulfate, nitrate, ammonium, sodium, chloride, OA, and  
418 EC respectively.

419

### 420 **3 Model calculated Dataset**

421 The ECHAM/MESSy Atmospheric Chemistry (EMAC) model is used, a numerical  
422 chemistry and climate simulation system that includes sub-models describing  
423 atmospheric processes from the troposphere to the mesosphere and their interaction  
424 with oceans, land, and human influences (Jöckel et al., 2006). EMAC uses the Modular  
425 Earth Submodel System (MESSy2) (Jöckel et al., 2010) to link the different sub-models  
426 with an atmospheric dynamical core, being an updated version of the 5th generation  
427 European Centre - Hamburg general circulation model (ECHAM5) (Roeckner et al.,  
428 2006). The EMAC model has been extensively described and evaluated against  
429 observations and satellite measurements and can be applied to a range of spatial  
430 resolutions (Tsimpidi et al., 2016; Karydis et al., 2016; Janssen et al., 2017; Tsimpidi  
431 et al., 2018; Pozzer et al., 2022; Milousis et al., 2024). The spectral resolution used in  
432 this study is T63L31, corresponding to a horizontal grid resolution of 1.875°x1.875°  
433 and 31 vertical layers extending to 10 hPa at about 25 km from the surface. The  
434 presented model simulations cover the period 2000–2020.

435

#### 436 **3.1 Model configuration**

437 In the model configuration used, EMAC calculates fields of gas phase species online  
438 through the Module Efficiently Calculating the Chemistry of the Atmosphere  
439 (MECCA) submodel (Sander et al., 2019). MECCA calculates the concentration of a



440 range of gases, including aerosol precursor species such as SO<sub>2</sub>, NH<sub>3</sub>, NO<sub>x</sub>, DMS,  
441 H<sub>2</sub>SO<sub>4</sub> and DMSO. The concentrations of the major oxidant species (OH, H<sub>2</sub>O<sub>2</sub>, NO<sub>3</sub>,  
442 and O<sub>3</sub>) are also calculated online. The loss of gas phase species to the aerosol through  
443 heterogeneous reactions (e.g., N<sub>2</sub>O<sub>5</sub> to form HNO<sub>3</sub>) is treated using the  
444 MECCA\_KHET submodel (Jöckel et al., 2010). The aqueous phase oxidation of SO<sub>2</sub>  
445 and the uptake of HNO<sub>3</sub> and NH<sub>3</sub> in cloud droplets are treated by the SCAV submodel  
446 (Tost et al., 2006; Tost et al., 2007).

447 Aerosol microphysics and gas/aerosol partitioning are calculated by the Global  
448 Modal-aerosol eXtension (GMXe) module (Pringle et al., 2010). The aerosol size  
449 distribution is described by 7 interacting lognormal modes (4 hydrophilic and 3  
450 hydrophobic modes). The modes cover the aerosol size spectrum (nucleation, Aitken,  
451 accumulation and coarse). The aerosol composition within each mode is uniform with  
452 size (internally mixed), though can vary between modes (externally mixed). The  
453 removal of gas and aerosol species through dry deposition is calculated within the  
454 DRYDEP submodel (Kerkweg et al., 2006) based on the big leaf approach. The  
455 sedimentation of aerosols is calculated within the SEDI submodel (Kerkweg et al.,  
456 2006) using a first order trapezoid scheme. Cloud properties and microphysics are  
457 calculated by the CLOUD submodel utilizing the detailed two-moment microphysical  
458 scheme of Lohmann and Ferrachat (2010) and considering a physically based treatment  
459 of the processes of liquid (Karydis et al., 2017) and ice crystal (Bacer et al., 2018)  
460 activation.

461

### 462 **3.2 State of the art modules for the inorganic thermodynamics**

463 The inorganic aerosol composition is computed with the ISORROPIA-lite  
464 thermodynamic equilibrium model (Kakavas et al., 2022) as implemented in EMAC by  
465 Milousis et al. (2024). ISORROPIA-lite is an accelerated and simplified version of the  
466 widely used ISORROPIA-II aerosol thermodynamics model which calculates the  
467 gas/liquid/solid equilibrium partitioning of the K<sup>+</sup>-Ca<sup>2+</sup>-Mg<sup>2+</sup>-NH<sub>4</sub><sup>+</sup>-Na<sup>+</sup>-SO<sub>4</sub><sup>2-</sup>-NO<sub>3</sub><sup>-</sup>-  
468 Cl<sup>-</sup>-H<sub>2</sub>O aerosol system. ISORROPIA-lite assumes that the aerosol is always in a  
469 metastable state (i.e., it is composed only of a supersaturated aqueous phase) and uses  
470 binary activity coefficients from precalculated look-up tables to minimize the  
471 computational cost. ISORROPIA-lite provides almost identical results with  
472 ISORROPIA-II in a metastable mode and reduces its computational cost by 35%  
473 (Kakavas et al., 2022). The application of ISORROPIA-lite in EMAC improved the



474 computational speed of the model by 4% (Milousis et al., 2024). The assumption of  
475 thermodynamic equilibrium is a good approximation for fine mode aerosols which can  
476 reach equilibrium within the time frame of one model timestep. However, the  
477 equilibrium timescale for large particles is typically larger than the timestep of the  
478 model (Meng and Seinfeld, 1996). To account for kinetic limitations, the process of  
479 gas/aerosol partitioning is calculated in two stages (Pringle et al., 2010). In the first  
480 stage the amount of the gas phase species that are able to kinetically condense onto the  
481 aerosol phase within the model timestep is calculated assuming diffusion limited  
482 condensation (Vignati et al., 2004). In the second stage ISORROPIA-lite re-distributes  
483 the mass between the gas and the aerosol phase assuming instant equilibrium between  
484 the two phases.

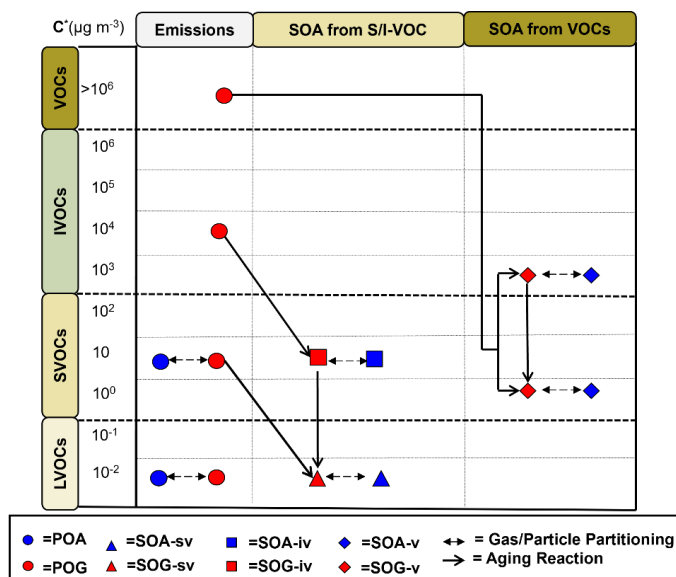
485

### 486 **3.3 State of the art module for organic aerosol**

487 The organic aerosol composition and evolution in the atmosphere is calculated by  
488 the ORACLE module (Tsimpidi et al., 2024). ORACLE is a computationally efficient  
489 version of the ORACLE module (Tsimpidi et al., 2014) which simulates a wide variety  
490 of semi-volatile organic products separating them into bins of logarithmically spaced  
491 effective saturation concentrations. ORACLE minimizes the number of surrogate  
492 species used to describe POA and SOA formation from different emission sources,  
493 while at the same time it reproduces similar total organic aerosol mass concentrations  
494 with the ORACLE module (Tsimpidi et al., 2024). In this application ORACLE uses  
495 three surrogate species with effective saturation concentration at 298 K of  $C^* = 10^{-2}$ ,  
496  $10^1$ , and  $10^4 \mu\text{g m}^{-3}$  to cover the volatility range of LVOCs, SVOCs and IVOCs  
497 emissions from biomass burning and other combustion sources (biofuel and fossil fuel  
498 combustion, and other urban sources). These organic compounds are allowed to  
499 partition between the gas and aerosol phases resulting in the formation of POA. The  
500 least volatile fraction, at  $10^{-2} \mu\text{g m}^{-3}$ , describes the low volatility organics in the  
501 atmosphere that are mostly in the particulate phase even in remote locations. The  $10 \mu\text{g}$   
502  $\text{m}^{-3}$  volatility bin describes the semivolatile organics in the atmosphere which partition  
503 between the particle and gas phase at atmospheric conditions. Finally, even under  
504 highly polluted conditions the majority of the material in the  $10^4 \mu\text{g m}^{-3}$  volatility bin  
505 will exist almost exclusively in the vapor phase. Photochemical reactions that modify  
506 the volatility of the emitted organic compounds that remain in the gas phase are taken  
507 into account and the oxidation products are simulated separately in the module to keep



508 track of the SOA formation from SVOC and IVOC emissions. LVOCs are not allowed  
 509 to participate in photochemical reactions since they are already in the lowest volatility  
 510 bin. A similar approach is followed for SOA formed from VOCs. In the this version of  
 511 ORACLE, it is assumed that the oxidation of the anthropogenic and biogenic VOC  
 512 species results in two products for each precursor distributed in two volatility bins with  
 513 effective saturation concentrations at 298 K equal to 1 and  $10^3 \mu\text{g m}^{-3}$  at 298 K. Overall,  
 514 we have assumed that functionalization and fragmentation processes after any  
 515 subsequent photochemical aging as a result of the reaction with OH results in a net  
 516 average decrease of volatility by a factor of  $10^3$  for SOA produced by SVOC/IVOC and  
 517 anthropogenic VOC, without a net average change of volatility for SOA produced by  
 518 biogenic VOC (Tsimpidi et al., 2024). In total 18 organic compounds are simulated  
 519 explicitly, i.e., 9 in each of the gas and aerosol phases. Based on the saturation



**Figure 8:** Schematic of the VBS resolution and the formation procedure of POA and SOA from LVOCs, SVOCs, IVOCs and VOCs emissions in ORACLE-lite. Red indicates that the compound is in the vapor phase and blue in the particulate phase. The circles correspond to primary organic material that can be emitted either in the gas or in the aerosol phase. The triangles indicate the formation of SOA from SVOCs by fuel combustion and biomass burning sources, while the squares show SOA from IVOCs by fuel combustion and biomass burning sources, and the diamonds the formation of SOA from anthropogenic and biogenic VOC sources. The partitioning processes, the aging reactions and the names of the species used to track all compounds are also shown.



520 concentration of each organic compound, ORACLE calculates the partitioning between  
521 the gas and particle phases by assuming bulk equilibrium and that all organic  
522 compounds form a pseudo-ideal solution. A schematic overview of the ORACLE  
523 module and the different aerosol types and chemical processes considered here is  
524 provided in Figure 8. More details about ORACLE can be found in Tsimpidi et al.  
525 (2024).

526

### 527 **3.4 Emissions**

528 Fuel combustion and agriculture related emissions are based on the high resolution  
529 ( $0.1^\circ \times 0.1^\circ$ ) Copernicus Atmosphere Monitoring Service global anthropogenic  
530 emission inventory applied at monthly intervals, CAMSv4.2 (Granier et al., 2019). The  
531 emission factors used for the distribution of traditional POA emissions from fuel  
532 combustion and open biomass burning sources into the three volatility bins considered  
533 by ORACLE are based on the work of Tsimpidi et al. (2024). These emission factors  
534 account additionally for IVOC emissions that are not included in the original emission  
535 inventories. We assume that the missing IVOC emissions from anthropogenic  
536 combustion are 1.5 times the traditional OA emissions included in the inventory.  
537 LVOCs and SVOCs are assumed to be emitted in the aerosol phase, while IVOCs are  
538 emitted in the gas phase. Then, they are allowed to partition between the gas and particle  
539 phase. Figure S4 shows the temporal evolution of anthropogenic emissions of inorganic  
540 ( $\text{SO}_2$ ,  $\text{NH}_3$ ,  $\text{NO}_x$ ) and organic (LVOC, SVOC, IVOC, VOC) aerosol precursors over  
541 the last 20 years, while Table S5 shows their decadal percentage change between the  
542 2000s and 2010s. Open biomass burning emissions are calculated online based on the  
543 dry matter burned from observations (Kaiser et al., 2012) and the fire type which affect  
544 the emission factors for the different tracers (Akagi et al., 2011). Similar to POA  
545 emissions from fuel combustion, POA from biomass burning is distributed to LVOC,  
546 SVOC, and IVOC emissions, however, no additional IVOC emissions are assumed for  
547 open biomass burning and therefore the sum for the biomass burning emission factors  
548 is unity (Tsimpidi et al., 2016).

549 Biogenic emissions of isoprene and terpenes are calculated online using the Model  
550 of Emissions of Gases and Aerosol from Nature (MEGANv2.04; Guenther et al., 2012)  
551 with an average emission flux of 454 and  $81.7 \text{ Tg yr}^{-1}$ , respectively. The natural  
552 emissions of  $\text{NH}_3$  are based on the GEIA database (Bouwman et al., 1997) and include  
553 excreta from domestic animals, wild animals, synthetic nitrogen fertilizers, oceans,



554 biomass burning, and emissions from soils under natural vegetation. NO<sub>x</sub> produced by  
555 lightning is calculated online and distributed vertically based on the parameterization  
556 of Price and Rind (1992). The emissions of NO from soils are calculated online based  
557 on the algorithm of Yienger and Levy (1995). Eruptive and non-eruptive volcanic  
558 degassing emissions of SO<sub>2</sub> are based on the AEROCOM data set (Dentener et al.,  
559 2006). The oceanic DMS emissions are calculated online by the AIRSEA submodel  
560 (Pozzer et al., 2006). Emission fluxes of sea spray aerosols are calculated online (Guelle  
561 et al., 2001) assuming a composition of 55% Cl<sup>-</sup>, 30.6% Na<sup>+</sup>, 7.7% SO<sub>4</sub><sup>2-</sup>, 3.7% Mg<sup>2+</sup>,  
562 1.2% Ca<sup>2+</sup>, 1.1% K<sup>+</sup> (Seinfeld and Pandis, 2006). The average global emission flux of  
563 sea spray aerosols is 5910 Tg yr<sup>-1</sup>. Dust emission fluxes are calculated online by using  
564 the meteorological fields calculated by the EMAC model (temperature, pressure,  
565 relative humidity, soil moisture and the surface friction velocity) together with specific  
566 input fields for soil properties (i.e., the geographical location of the dust sources, the  
567 clay fraction of the soils, the rooting depth, and the monthly vegetation area index)  
568 (Astitha et al., 2012). The average global emission flux of dust particles is 5684 Tg yr<sup>-1</sup>.  
569 Emissions of individual crustal species (Ca<sup>2+</sup>, Mg<sup>2+</sup>, K<sup>+</sup>, Na<sup>+</sup>) are estimated as a  
570 constant fraction of mineral dust emissions. This fraction is determined based on the  
571 geological information that exists for the different dust source regions of the planet  
572 (Karydis et al., 2016) and is applied online on the calculated mineral dust emission  
573 fluxes based on the location of the grid cell (Klingmuller et al., 2018).

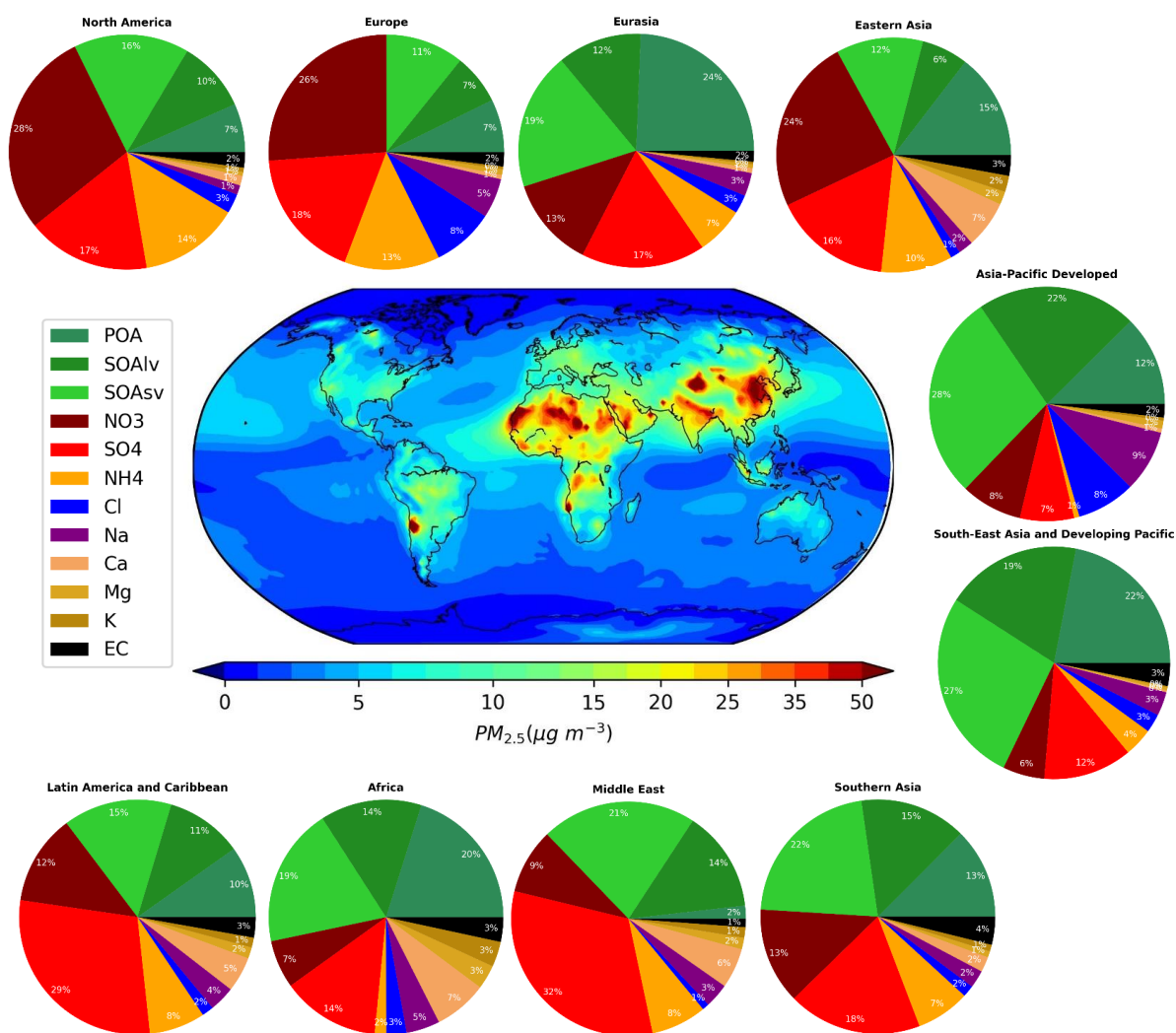
574

### 575 **3.5 Model calculated aerosol composition**

576 The EMAC simulation corroborates the findings based on filters and AMS  
577 observations that OA is the dominant component of fine atmospheric aerosols in all  
578 continental regions (Figure 9). The strongest OA contribution to total PM<sub>2.5</sub> (more than  
579 50%) is calculated over regions affected by biomass burning and biogenic VOC  
580 emissions: the tropical forests and savannas of Africa, Latin America and Caribbean,  
581 Southern Asia, and Southeast Asia and Developing Pacific, as well as the boreal forests  
582 of Eurasia. Considerable OA shares (30-35%) are also calculated over the industrialized  
583 regions of the Northern Hemisphere (i.e., North America, Europe, Eastern Asia) and  
584 the Middle East, where strong fossil and biofuel combustion related sources are located.  
585 OA shares peak in the summer over Europe and North America and in the winter over  
586 East Asia (Figures S1-S3). EMAC is also able to reproduce the dominance of SOA  
587 (resolved by the AMS as OOA) in all regions, even in regions with strong primary



588 emissions, e.g., close to tropical forests or industrial areas. However, EMAC cannot  
 589 reproduce the dominance of aged SOA in many cases (resolved as M-OOA by the  
 590 AMS), especially over Eastern Asia, revealing weaknesses in the oxidation scheme of  
 591 its organic module (e.g., including missing sources and formation pathways). POA has  
 592 the strongest contribution (more than 20%) over heavily forested areas (e.g., Africa and  
 593 Eurasia) and the lowest (less than 10%) over highly industrialized regions (e.g., Europe



**Figure 9:** Pie charts showing the simulated 20-year average chemical composition of  $PM_{2.5}$  in the 10 regions considered according to WGIII AR6. The central world map shows the simulated average near-surface concentration of  $PM_{2.5}$  (in  $\mu g m^{-3}$ ) during the period 2000-2020.



594 and Middle East). Regarding the inorganic aerosol composition, the EMAC model is  
595 not always consistent with the filter-based observations since in many regions it reveals  
596 that nitrate overpasses sulfate in the aerosol composition, which is also supported by  
597 the AMS results. These regions are Europe, North America, and Eastern Asia, where  
598 nitrate accounts for 25-30% of total PM<sub>2.5</sub>, with higher contributions in winter and  
599 lower contributions in summer (Figures S1-S3). Sulfate becomes the dominant  
600 inorganic aerosol component only during winter over North America (Figures S1-S3).  
601 On the other side, sulfate contribution is stronger over the Middle East and Latin  
602 America and Caribbean (~30%). Ammonium follows the spatial distribution of sulfate  
603 and nitrate with high contributions to PM<sub>2.5</sub> composition (~10-15%) over the highly  
604 populated and agriculturally intensive regions of North America, Europe, Eastern Asia  
605 and Southern Asia. Mineral dust is simulated to be a significant natural contributor to  
606 aerosol composition in some regions. Here we only focus on the chemically active  
607 components of mineral dust, which are the crustal cations of calcium, potassium,  
608 sodium, and magnesium. Their total share to PM<sub>2.5</sub> composition is around 15% in  
609 regions affected by desert emissions (e.g., Africa, Middle East, Eastern Asia) while in  
610 other areas their contribution is limited (~ 1%). Finally, sodium and chloride from sea  
611 salt emissions are found to be high over regions with long coastlines per land area. Most  
612 notably, chloride consists of 8% of the total PM<sub>2.5</sub> over the Asia Pacific Developed  
613 region, while sodium is the dominant inorganic component in the same region with a  
614 share of 8.5%.

615

#### 616 **4 In depth model Evaluation**

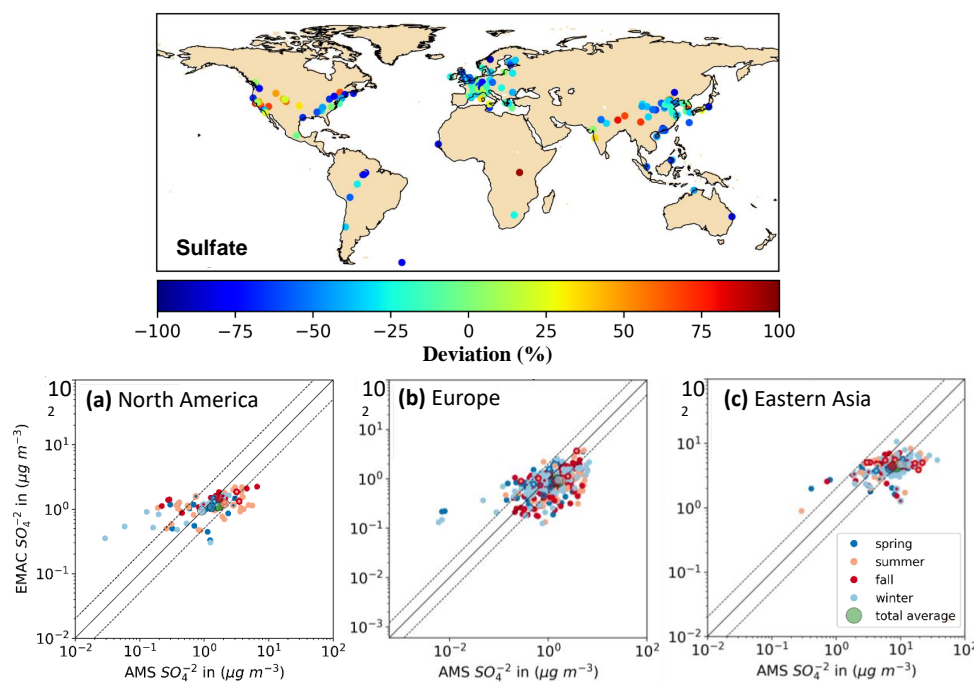
##### 617 **4.1 Sulfate**

618 The EMAC performance for sulfate is best over North America, where the model  
619 tends to underpredict its concentrations with a MB of -0.45  $\mu\text{g m}^{-3}$  (Figure 10a). The  
620 model performs better over rural regions with very low NMB (-8%) and worst over  
621 urban locations (NMB=-40%). This performance can be attributed to the low spatial  
622 resolution used and to possible errors in the assumed injection height of SO<sub>2</sub> (Yang et  
623 al., 2019) which can affect sulfate concentrations close to sources. Furthermore, EMAC  
624 tends to overestimate sulfate over the Midwest, while underestimating its  
625 concentrations over the Eastern states (Figure 10). The coarse resolution of the model  
626 cannot reproduce the orography of the mountainous Midwest and therefore  
627 overestimates the sulfate concentrations at high altitude sites. On the other hand, due to





628 its coarse resolution, it underestimates the sulfate concentrations over the urban areas  
 629 of the densely populated Eastern states. Therefore, the model underpredicts  
 630 observations over the Eastern US, where sulfate concentrations are high, and  
 631 overpredicts observations over the Midwest, where sulfate concentrations are low. As  
 632 a result, the model produces a quite narrow range of concentrations (i.e., 0.3 - 2.5  $\mu\text{g}$   
 633  $\text{m}^{-3}$ ) over the North America in contrast to the AMS observations which cover almost  
 634 three orders of magnitude, ranging from 0.1 to 10  $\mu\text{g}$   $\text{m}^{-3}$ . The seasonal pattern of both  
 635 measured and observed data shows clear differences between summer and winter. The  
 636 model calculates the highest sulfate concentrations in autumn, in contrast to the AMS  
 637 observations which show a peak in summer. The lowest sulfate concentrations are  
 638 observed in winter which are well captured by the model at most sites (Figure 10a).



**Figure 10:** Deviations (in %) between EMAC results and the AMS and ACSM datasets over the period 2000 – 2020 (top). Negative values (blue colors) correspond to underprediction of sulfate concentrations by the model. Scatter plots comparing model results for  $\text{PM}_{10}$  sulfate concentrations (in  $\mu\text{g}$   $\text{m}^{-3}$ ) with AMS and ACSM observations (bottom) over (a) North America, (b) Europe, and (c) Eastern Asia. Each point represents the data set mean and is colored based on the season of the field campaign. Also shown are the 1:1, 2:1, and 1:2 lines.



639 In Europe, the model underpredicts sulfate in all types of environments and all  
 640 seasons by about 40% due to errors in emissions and an underestimation of the  
 641 oxidation capacity of the atmosphere (Emep, 2021). However, a few overpredictions  
 642 are calculated over Italy and Greece. Around 65% of the simulated sulfate  
 643 concentrations over Europe are within a factor of 2 compared to measurements (Figure  
 644 10b). The performance of the model does not exhibit any clear seasonal pattern except  
 645 a slight tendency towards higher underpredictions during summer when the observed  
 646 sulfate concentrations are the highest of the year. Over Asia, sulfate concentrations are  
 647 significantly higher than over Europe and North America, however, the performance of  
 648 the model is similar. Sulfate is underpredicted most of the time (Figure 10c, Table 1).  
 649 The model performs better over rural locations (NME=-38%) and worst over urban  
 650 areas (50%). Furthermore, while the model underpredicts sulfate concentrations during  
 651 all seasons, its performance is worst in winter when sulfate exhibits its annual peak  
 652 concentrations (Figure 10c) due to its multiphase formation during haze events, a  
 653 pathway not accurately resolved by the model. Furthermore, similar to North America,  
 654 the concentration range of the simulated sulfate over Eastern Asia is much narrower  
 655 than the observed, covering little more than one order of magnitude compared to two  
 656 orders of magnitude reported by the AMS. Over the tropical and subtropical regions,  
 657 sulfate is underestimated again, mostly over the Asian regions (NME  $\approx$  -45%) and less  
 658 over Africa and Latin America and Caribbean (NME  $\approx$  -30%) (Table S2, Figure S5).

**Table 1:** Statistical evaluation of EMAC PM<sub>1</sub> sulphate concentrations against AMS and ACSM datasets over Europe, North America, and Eastern Asia for the period of 2000–2020.

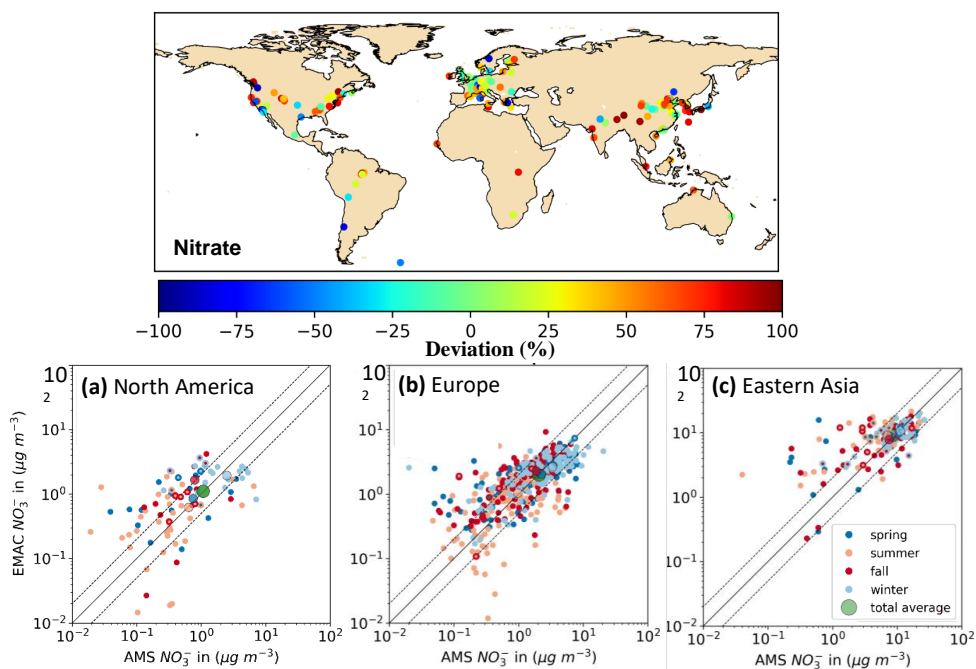
Continent	Region	Number of data sets	Mean observed [ $\mu\text{gm}^{-3}$ ]	Mean predicted [ $\mu\text{gm}^{-3}$ ]	MAGE [ $\mu\text{gm}^{-3}$ ]	MB [ $\mu\text{gm}^{-3}$ ]	NME [%]	NMB [%]	RMSE [ $\mu\text{gm}^{-3}$ ]
Europe	all	431	1.54	0.91	0.79	-0.63	51.08	-41.17	1.22
	rural	240	1.41	0.8	0.76	-0.61	53.99	-43.35	1.18
	DW	155	1.71	1.07	0.82	-0.65	47.9	-37.72	1.3
	urban	36	1.71	0.95	0.84	-0.75	48.9	-44.15	1.07
North-America	all	88	1.63	1.18	0.85	-0.45	52.37	-27.72	1.27
	rural	46	1.14	1.05	0.71	-0.09	62.28	-8.25	1.05
	DW	7	1.8	1.29	0.6	-0.5	33.53	-28.05	0.85
	urban	35	2.23	1.32	1.09	-0.91	48.75	-40.74	1.57
Eastern-Asia	all	153	8.54	4.52	4.44	-4.02	52.05	-47.12	6.47
	rural	44	7.15	4.44	3.42	-2.71	47.77	-37.93	4.71
	DW	16	7.93	4.31	4.04	-3.61	50.94	-45.58	4.55
	urban	93	9.3	4.59	5.0	-4.71	53.77	-50.69	7.41



659 **4.2 Nitrate**

660 The model is able to capture the observed average nitrate concentrations over the  
661 different regions and seasons with very low NMB (below 10%). However, the NME is  
662 high over all regions (40-80%) indicating that the discrepancy between model results  
663 and observations is highly scattered and not systematically biased (Table 2). The  
664 accurate prediction of nitrate concentrations is rather complex. Nitrate is typically  
665 formed in areas characterized by high ammonia and nitric acid concentrations and low  
666 sulfate concentrations. At the same time, the thermodynamic equilibrium of ammonium  
667 nitrate varies several orders of magnitude under typical atmospheric conditions  
668 (Seinfeld and Pandis, 2006). This variation causes significant challenges in the  
669 calculation of nitrate concentrations since small errors in RH and T can shift the  
670 equilibrium of nitric acid to the gas or the aerosol phase. Therefore, even though the  
671 scatter is not negligible, it is encouraging that the EMAC model seems to perform  
672 surprising well under diverse environments and atmospheric conditions (Figure 11).  
673 The scatter is more intense over North America (NME=88%), especially during the  
674 summer season where the occurrence of high temperatures and the semi-volatile nature  
675 of  $\text{NH}_4\text{NO}_3$  hinder the model's ability to capture the observations accurately (Figure  
676 11a). However, the model is still able to capture the seasonality of nitrate concentrations  
677 well with the highest concentrations calculated during the periods with the lowest  
678 temperatures (i.e., winter), when almost all the nitric acid that is available is transferred  
679 to the particulate phase.

680 Over Europe, despite some widely dispersed points, the majority of datapoints (70%)  
681 lie within a factor of two compared to observations (Figure 11b). Similar to North  
682 America, the seasonality is very well captured, and the model predictions are mostly  
683 scattered during the warmer seasons. However, the overall performance is better here  
684 with NMB = -4% and NME = 53%. Over Eastern Asia, the overestimation appears to  
685 be more systematic, especially during the summer and fall (Figure 11c). However, with  
686 an overall NMB of 7.7%, the performance can still be considered very good (Table 2).  
687 Nitrate levels are significantly overestimated by the model, especially over the west  
688 coast of South Korea and the Chinese inlands (Figure 11). However, Eastern China and  
689 especially the coastal regions are well described by the model. The contribution of sea  
690 salt to nitrate formation is important in these coastal regions due to their proximity to  
691 the Pacific Ocean (Bian et al., 2017). Therefore, the overestimation of nitrate levels on  
692 the west coast of Korea, in contrast to the well captured east coast, could be caused by



**Figure 11:** Deviations (in %) between EMAC results and the AMS and ACSM datasets over the period 2000 – 2020 (top). Negative values (blue colors) correspond to underprediction of nitrate concentrations by the model. Scatter plots comparing model results for PM<sub>1</sub> nitrate concentrations (in  $\mu\text{g m}^{-3}$ ) with AMS and ACSM observations (bottom) over (a) North America, (b) Europe, and (c) Eastern Asia. Each point represents the data set mean and is colored based on the season of the field campaign. Also shown are the 1:1, 2:1, and 1:2 lines.

693 the dominant west-east winds in the Yellow Sea simulated by the model, leading to an  
 694 overestimation of the sea salt content that can contribute to nitrate formation. Over the  
 695 tropical and subtropical regions, the discrepancies between the simulated and observed  
 696 nitrate concentrations are less dispersed with a tendency towards overprediction by the  
 697 model in most regions (Figure S5; Table S2). Over Latin America and the Caribbean,  
 698 the model underpredicts nitrate (NMB = -50%) except for a few strong overpredictions,  
 699 mostly during the wet season, suggesting possible errors in simulated wet deposition  
 700 (Figure S5). On the other hand, over Africa, the model overpredicts nitrate during the  
 701 dry season, especially over Welegund, an observation site downwind of Johannesburg.  
 702 Nitrate is strongly overpredicted over the Asia Pacific Developed region, especially  
 703 over the industrialized regions of Japan and Australia. On the contrary, the model  
 704 performance for nitrate is good over the Southeast Asia and the Developing Pacific



705 (NMB = -3%) with few random over- and underpredictions during the monsoon and  
 706 the transition periods towards that season.

**Table 2:** Statistical evaluation of EMAC PM<sub>1</sub> nitrate concentrations against AMS and ACSM datasets over Europe, North America, and Eastern Asia for the period of 2000–2020.

Continent	Region	Number of data sets	Mean observed [ $\mu\text{gm}^{-3}$ ]	Mean predicted [ $\mu\text{gm}^{-3}$ ]	MAGE [ $\mu\text{gm}^{-3}$ ]	MB [ $\mu\text{gm}^{-3}$ ]	NME [%]	NMB [%]	RMSE [ $\mu\text{gm}^{-3}$ ]
Europe	all	431	2.07	1.98	1.09	-0.09	52.54	-4.17	1.74
	rural	240	1.57	1.56	1.01	-0.02	64.02	-1.02	1.77
	DW	155	2.88	2.61	1.24	-0.27	42.95	-9.52	1.76
	urban	36	1.86	2.12	0.96	0.26	51.71	13.73	1.36
North-America	all	88	1.07	1.1	0.94	0.04	87.79	3.35	1.45
	rural	46	0.81	0.84	0.66	0.03	81.44	3.59	0.97
	DW	7	0.98	1.18	1.11	0.2	114.09	20.62	1.53
	urban	35	1.43	1.44	1.27	0.01	88.93	0.82	1.88
Eastern-Asia	all	152	8.47	9.12	3.58	0.65	42.3	7.67	4.81
	rural	43	6.44	7.36	4.11	0.92	63.8	14.21	5.17
	DW	16	4.74	7.49	3.67	2.75	77.54	58.0	5.08
	urban	93	10.05	10.22	3.33	0.17	33.08	1.65	4.58

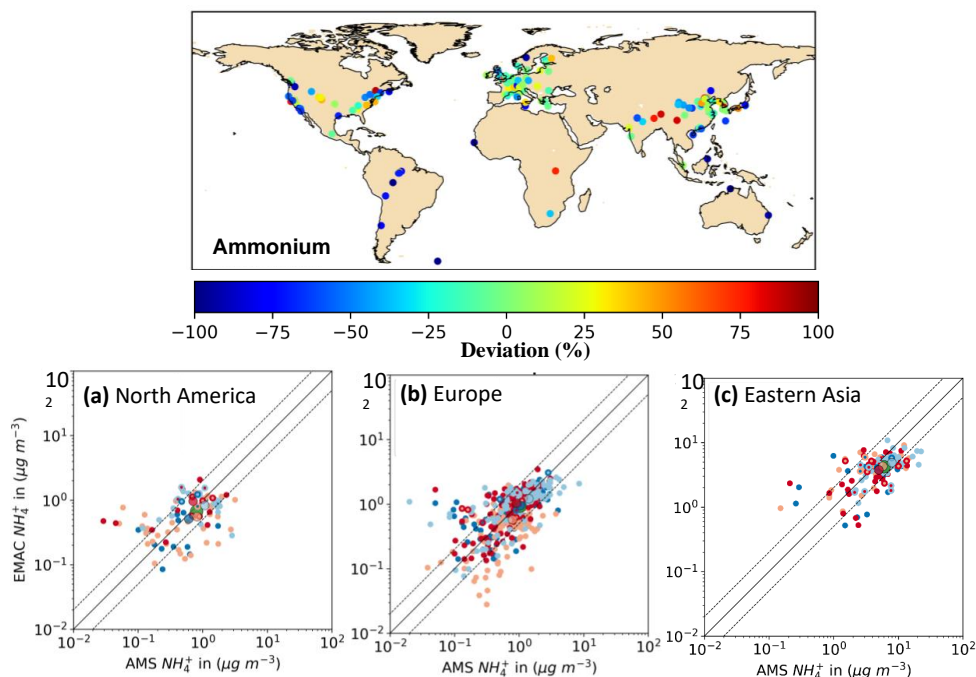
707

### 708 4.3 Ammonium

709 EMAC tends to underpredict ammonium over the three main subcontinents of the  
 710 Northern Hemisphere, however, its performance is considered satisfactory with  
 711 relatively low bias and scatter (Table 3). The model evaluation exhibits a large scatter  
 712 only over North America (NME = 63%), where 50% of the comparison sites are beyond  
 713 the factor 2 intervals (Figure 12a). Ammonium tends to be overestimated during autumn  
 714 and underestimated during the rest of the seasons; especially during the summer (Figure  
 715 12a). Over Europe, the model exhibits its best performance with low NMB (-9%). The  
 716 average deviation from the observations is also relatively low (Figure 12) and 75% of  
 717 the model results diverge less than a factor of two from measurements. Surprisingly,  
 718 the model performance is best over the Benelux region (Figure 12) where NH<sub>3</sub>  
 719 emissions are the highest over Europe. While the good model performance for  
 720 ammonium over Europe indicates an accurate emission inventory for agricultural and  
 721 livestock NH<sub>3</sub>, the overprediction of nitrate and underprediction of sulfate suggest that  
 722 the model overpredicts the fraction of ammonium that exists as ammonium nitrate  
 723 (instead of ammonium sulfate). Over Asia, the model strongly underestimates  
 724 ammonium (NMB = -30%), especially over Eastern China (Figure 12). While this



725 underestimation can be partially attributed to sulfate underpredictions, the simultaneous  
726 overestimation of nitrate over the same areas indicates errors in the  $\text{NH}_3$  emission  
727 inventory. On the other hand, ammonium is overpredicted close to the deserts of Inland  
728 China (e.g., over Tibet) and over South Korea (Figure 12). Over the Tropics and the  
729 southern continents, ammonium is underestimated to a higher extent than in the  
730 northern continents (with NMB from -40 to -60%). The main problem in model  
731 performance is over Asia Pacific Developed and Africa, where the model predicts low  
732 ammonium shares that are not supported by AMS observations (Figure S2). On the  
733 other hand, EMAC has the largest underprediction and highest NMB over Latin  
734 America. Nevertheless, here and over South Asia, EMAC and AMS agree that  
735 ammonium has the smallest fraction of  $\text{PM}_{10}$ . Overall, deviations in ammonium can be  
736 traced back to global livestock emission inventory uncertainties as criticized by Hoesly  
737 et al. (2018).



**Figure 12:** Deviations (in %) between EMAC results and the AMS and ACSM datasets over the period 2000 – 2020 (top). Negative values (blue colors) correspond to underprediction of ammonium concentrations by the model. Scatter plots comparing model results for  $\text{PM}_{10}$  ammonium concentrations (in  $\mu\text{g m}^{-3}$ ) with AMS and ACSM observations (bottom) over (a) North America, (b) Europe, and (c) Eastern Asia. Each point represents the data set mean and is colored based on the season of the field campaign. Also shown are the 1:1, 2:1, and 1:2 lines.



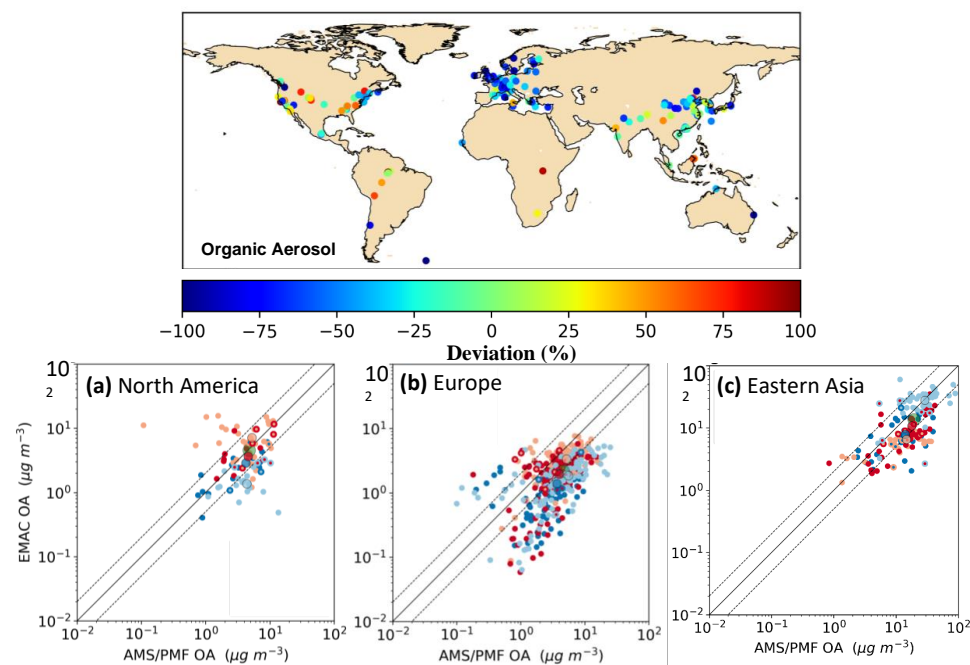
**Table 3:** Statistical evaluation of EMAC PM<sub>1</sub> ammonium concentrations against AMS and ACSM datasets over Europe, North America, and Eastern Asia for the period of 2000–2020.

Continent	Region	Number of data sets	Mean observed [ $\mu\text{gm}^{-3}$ ]	Mean predicted [ $\mu\text{gm}^{-3}$ ]	MAGE [ $\mu\text{gm}^{-3}$ ]	MB [ $\mu\text{gm}^{-3}$ ]	NME [%]	NMB [%]	RMSE [ $\mu\text{gm}^{-3}$ ]
Europe	all	431	1.03	0.93	0.44	-0.1	42.33	-9.38	0.68
	rural	240	0.8	0.76	0.42	-0.05	52.55	-5.62	0.69
	DW	155	1.36	1.2	0.47	-0.15	34.67	-11.39	0.69
	urban	36	1.13	0.94	0.38	-0.19	33.48	-16.78	0.5
North-America	all	87	0.81	0.67	0.51	-0.14	62.97	-17.29	0.71
	rural	46	0.54	0.49	0.36	-0.05	65.9	-8.6	0.45
	DW	7	0.85	0.74	0.44	-0.11	51.52	-12.93	0.54
	urban	34	1.16	0.89	0.73	-0.27	62.85	-23.41	0.97
Eastern-Asia	all	152	5.99	4.21	2.63	-1.78	43.96	-29.76	3.79
	rural	43	4.91	3.64	2.49	-1.27	50.77	-25.83	3.66
	DW	16	3.8	3.51	1.6	-0.29	42.07	-7.51	1.96
	urban	93	6.87	4.59	2.88	-2.28	41.89	-33.18	4.08

738

#### 739 4.4 Organic aerosol

740 The model performance for total OA concentration varies significantly between the  
 741 three continents. Over North America, the simulated mean OA represents well the  
 742 observed OA by AMS (NMB = -4%). However, the comparison exhibits a significant  
 743 scatter (NME = 64%) since the model tends to overpredict OA over rural locations  
 744 (NMB = 37%) and underpredict it over and downwind of urban sites (NMB = -28%).  
 745 The model roughly captures the seasonality of OA concentrations over North America,  
 746 with high OA concentrations in summer and autumn and lower concentrations in spring  
 747 and winter. OA concentrations peak during summer due to enhanced biogenic VOC  
 748 emissions and photochemistry (Goldstein and Galbally, 2007; Tsimpidi et al., 2016),  
 749 however, EMAC tends to overpredict some low OA concentrations measured by AMS  
 750 over a few rural locations during summertime (Figure 13a). Over Europe, the model  
 751 tends to underestimate OA during all seasons, except summer (Figure 13b). The model  
 752 performance is worst during wintertime, where sources from biomass burning,  
 753 particularly by domestic wood burning, and their dark oxidation have been recently  
 754 identified as a major source of model bias over Europe during wintertime (Tsimpidi et  
 755 al., 2016; Kodros et al., 2020). This also affects the simulated OA seasonality over  
 756 Europe where the model estimates higher OA concentrations during summer over all  
 757 types of environments, while the AMS observations reveal that this is true only over  
 758



**Figure 13:** Deviations (in %) between EMAC results and the AMS and ACSM datasets over the period 2000 – 2020 (top). Negative values (blue colors) correspond to underprediction of organic aerosol concentrations by the model. Scatter plots comparing model results for PM<sub>1</sub> organic aerosol concentrations (in  $\mu\text{g m}^{-3}$ ) with AMS and ASCM observations (bottom) over (a) North America, (b) Europe, and (c) Eastern Asia. Each point represents the data set mean and is colored based on the season of the field campaign. Also shown are the 1:1, 2:1, and 1:2 lines.

759 rural locations. According to AMS, over and downwind of urban areas, OA  
760 concentrations peak during wintertime. Over Eastern Asia, the model exhibits its best  
761 performance with relatively low bias (NMB = -29%) and scatter (NME = 49%). In  
762 contrast to Europe, the wintertime OA is well captured by the model even over urban  
763 locations (Table 4). The model has excellent performance over rural and urban-  
764 downwind locations with 75% of the datapoints lying within a factor of two compared  
765 to observations. However, as it is typical for every global model (Tsigaridis et al.,  
766 2014), the model fails to reproduce some of the high OA concentrations observed over  
767 large urban centers due to its limited spatial resolution. Over the rest of the continental  
768 regions, the overall performance of the model is satisfying for OA. EMAC tends to  
769 underpredict OA over the tropical regions of South Asia and Developing Pacific and  
770 over the more urbanized regions of the Asia Pacific Developed, without any clear  
771 seasonal pattern (Figure S5). In contrast, simulated OA are overestimated over Africa,





772 mostly during the dry season. Over Latin America and Caribbean, the evaluation  
 773 datapoints are more scattered with a few significant overestimations during the  
 774 Amazonian wet season and underestimations during the dry season.

**Table 4:** Statistical evaluation of EMAC PM<sub>1</sub> OA concentrations against AMS and ACSM datasets over Europe, North America, and Eastern Asia during 2000–2020.

Continent	Region	Number of data sets	Mean observed [ $\mu\text{gm}^{-3}$ ]	Mean predicted [ $\mu\text{gm}^{-3}$ ]	MAGE [ $\mu\text{gm}^{-3}$ ]	MB [ $\mu\text{gm}^{-3}$ ]	NME [%]	NMB [%]	RMSE [ $\mu\text{gm}^{-3}$ ]
Europe	all	442	4.59	2.18	2.73	-2.41	59.54	-52.56	3.95
	rural	247	3.63	1.96	2.11	-1.66	58.16	-45.93	3.08
	DW	156	5.93	2.45	3.65	-3.48	61.58	-58.63	5.08
	urban	39	5.33	2.45	3.01	-2.88	56.39	-54.12	3.73
North-America	all	86	4.77	4.56	3.05	-0.2	64.1	-4.24	4.29
	rural	46	3.29	4.51	2.95	1.22	89.79	37.28	4.52
	DW	7	5.58	4.6	2.29	-0.97	40.98	-17.42	2.92
Eastern Asia	urban	33	6.66	4.63	3.36	-2.03	50.53	-30.46	4.2
	all	159	19.3	13.64	9.41	-5.65	48.74	-29.3	13.19
	rural	44	12.77	10.75	6.72	-2.02	52.64	-15.81	11.55
	DW	16	11.38	9.79	6.0	-1.59	52.73	-13.97	8.87
	urban	99	23.48	15.55	11.15	-7.93	47.49	-33.76	14.41

775

#### 776 4.4.1 POA

777 The simulated POA concentrations are compared with the sum of the AMS HOA  
 778 and BBOA concentrations. POA concentrations are mostly underestimated by the  
 779 model over North America and Europe (NMB  $\approx$  -45%) and significantly overestimated  
 780 over Eastern Asia (NMB = 98%). In North American rural regions, POA simulated  
 781 concentrations are highest during spring and winter and lowest during fall, consistent  
 782 with the observed POA levels. However, during summer, most of the observed data is  
 783 underestimated by the model (Figure 14). Over urban locations, POA is more severely  
 784 underestimated (NMB = -68%) due to the coarse spatial resolution of the model and the  
 785 evaporation of organic compounds upon emission. POA concentrations are also  
 786 underestimated over European urban regions (NMB = -52%), however, to a lesser  
 787 extent than over North America. Over rural locations, the model performance is  
 788 scattered during all seasons with a few cases of strong over and under predictions (NME  
 789 = 62%). Over Eastern Asia, a pronounced overestimation during winter is striking,  
 790 especially over mega-city clusters (NMB = 106%; Table 5) such as around Hong Kong  
 791 and Shanghai. This discrepancy can be related to overestimations in the emission  
 792 inventory (e.g., not including the emission reductions in the frame of the Chinese



793 control action plans) but also to the overestimated partition of the freshly emitted  
 794 SVOCs to the aerosol phase during the low winter temperatures. Tsimpidi et al. (2016)  
 795 has also reported POA overestimations over Eastern Asia due to too high simulated  
 796 bbPOA transported from the surrounding boreal forests. Since in ORACLE POA do  
 797 not participate in aqueous phase and other heterogeneous reactions, they do not convert  
 798 to SOA via these pathways, which can explain part of the positive model bias during  
 799 winter.

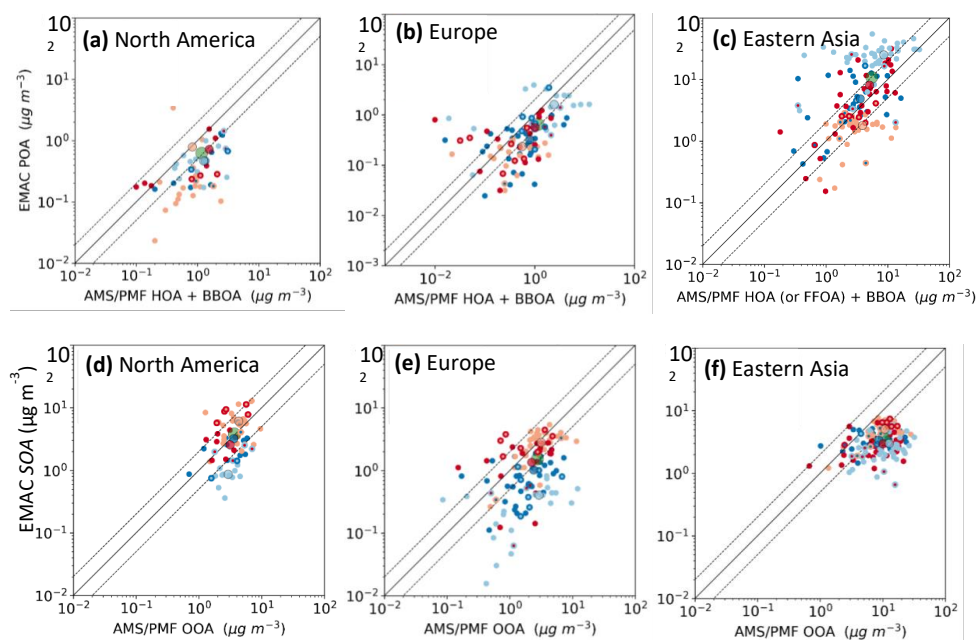
**Table 5:** Statistical evaluation of EMAC PM<sub>1</sub> POA concentrations against AMS and ACSM datasets over Europe, North America, and Eastern Asia during 2000–2020.

Continent	Region	Number of data sets	Mean observed [ $\mu\text{g m}^{-3}$ ]	Mean predicted [ $\mu\text{g m}^{-3}$ ]	MAGE [ $\mu\text{g m}^{-3}$ ]	MB [ $\mu\text{g m}^{-3}$ ]	NME [%]	NMB [%]	RMSE [ $\mu\text{g m}^{-3}$ ]
Europe	all	106	1.18	0.67	0.81	-0.51	68.45	-43.27	1.6
	rural	62	0.9	0.49	0.56	-0.4	62.5	-44.78	1.13
	DW	23	1.7	1.12	1.38	-0.58	81.24	-34.19	2.6
	urban	21	1.45	0.69	0.91	-0.75	62.87	-52.2	1.32
North-America	all	50	1.17	0.63	0.95	-0.54	80.93	-46.2	1.49
	rural	21	0.62	0.64	0.65	0.02	106.18	3.64	1.59
	DW	3	0.41	1.27	1.16	0.86	284.25	212.29	1.76
Eastern-Asia	all	26	1.7	0.55	1.16	-1.16	67.93	-67.93	1.38
	all	129	4.88	9.91	6.89	5.03	141.12	103.18	11.4
	rural	35	5.1	9.28	6.11	4.18	119.82	82.03	11.38
Eastern-Asia	DW	13	3.68	5.91	3.8	2.23	103.0	60.49	5.82
	urban	81	4.98	10.83	7.72	5.85	155.08	117.62	12.07

800

**Table 6:** Statistical evaluation of EMAC PM<sub>1</sub> SOA concentrations against AMS and ACSM datasets over Europe, North America, and Eastern Asia during 2000–2020.

Continent	Region	Number of data sets	Mean observed [ $\mu\text{g m}^{-3}$ ]	Mean predicted [ $\mu\text{g m}^{-3}$ ]	MAGE [ $\mu\text{g m}^{-3}$ ]	MB [ $\mu\text{g m}^{-3}$ ]	NME [%]	NMB [%]	RMSE [ $\mu\text{g m}^{-3}$ ]
Europe	all	129	2.77	1.51	1.69	-1.26	61.01	-45.6	2.44
	rural	84	2.53	1.52	1.53	-1.01	60.54	-39.93	2.34
	DW	24	3.43	1.19	2.41	-2.24	70.17	-65.19	3.15
	urban	21	2.98	1.81	1.5	-1.16	50.55	-39.09	1.83
North-America	all	67	3.66	3.83	2.34	0.17	63.89	4.52	3.0
	rural	35	3.16	3.51	2.11	0.35	66.92	11.08	2.79
	DW	6	5.27	4.23	2.33	-1.05	44.25	-19.87	3.05
Eastern-Asia	urban	26	3.97	4.17	2.65	0.2	66.66	4.97	3.24
	all	147	10.65	3.54	7.23	-7.11	67.86	-66.74	9.24
	rural	36	8.18	3.38	5.29	-4.8	64.62	-58.69	7.15
Eastern-Asia	DW	16	7.84	4.03	3.82	-3.82	48.68	-48.68	5.23
	urban	95	12.06	3.52	8.53	-8.53	70.79	-70.79	10.41



**Figure 14:** Scatter plots comparing model results for PM<sub>1</sub> primary organic aerosol (a-c) and secondary organic aerosol (d-f) concentrations (in  $\mu\text{g m}^{-3}$ ) with AMS and ASCM observations of HOA+BBOA and OOA, respectively, over North America (a, d), Europe (b, e), and Eastern Asia (c, f). Each point represents the data set mean and is colored based on the season of the field campaign. Also shown are the 1:1, 2:1, and 1:2 lines.

801

#### 802 4.4.2 SOA

803 The model simulated OOA concentrations over North America are in very good  
804 agreement with the OOA derived by the PMF analysis of the AMS observations (NMB  
805 = 4.5%). The model performs well over both urban and rural areas and during all  
806 seasons, except winter when it tends to underpredict the AMS-OOA estimations (Table  
807 6; Figure 14c). L-OOA concentrations are reproduced by the model particularly well  
808 (Figure S6a), however, M-OOA concentrations are slightly underestimated during  
809 spring and fall and severely underpredicted during winter (Figure S6d). Similarly, the  
810 model performance for all OOA types over Europe is best during summer and worst  
811 during winter when it underpredicts the AMS estimations, especially for the M-OOA  
812 (Figure S6e). During summer, the high temperatures enhance the biogenic VOC  
813 emissions from vegetation and, more importantly, the more abundant solar radiation  
814 increase the transformation of gas phase organic compounds through photochemical  
815 processing into particulate OOA (Seco et al., 2011; Xu et al., 2017; Tsimpidi et al.,



816 2016). The model performance during summer suggests that the model can accurately  
817 represent this process. In winter, however, photochemical processing has lower impact  
818 on OOA formation and evolution (Xu et al., 2017). Therefore, in seasons with  
819 decreasing temperatures and/or photochemical activity, the model performance is  
820 worsening, strongly suggesting that other processes become increasingly more  
821 important. Missing SOA formation processes are related to heterogeneous reactions  
822 like oligomerization or aqueous phase processing (Hallquist et al., 2009; Tsimpidi et  
823 al., 2016). Under high RH, aqueous phase processing can rapidly result in highly  
824 oxidized OOA (i.e., M-OOA with high oxygen to carbon ratio, O:C), while the impacts  
825 on fresher, less oxygenated OOA (i.e., L-OOA) are minor. For the latter, photochemical  
826 aging processes under low RH are more important (Xu et al., 2017). Such processes  
827 occur during all seasons, however, the meteorological conditions during winter favor  
828 the formation of M-OOA from aqueous phase chemistry against the L-OOA formation  
829 from gas-phase photochemical oxidation processes (Xu et al., 2017; Mortier et al.,  
830 2020; Pozzer et al., 2022). Therefore, this missing formation pathway becomes  
831 gradually more important from spring and fall to winter. Additionally, recent studies  
832 have identified high production of SOA during wintertime which can be attributed to  
833 the rapid oxidation of biomass burning OA by the  $\text{NO}_3$  radical during nighttime (Kodros  
834 et al., 2020; Paglione et al., 2020; Liu, 2024). Since residential heating from woodstoves  
835 is not included in the model and ORACLE includes only the predominant  
836 photochemical processing of BBOA by OH, a non-consideration of dark chemical  
837 processing of BBOA can lead to substantial underprediction of OOA during the cold  
838 seasons. Over Eastern Asia, OOA is underestimated even during summer (Figure 14f),  
839 mainly due to the underestimation of M-OOA since L-OOA is relatively well  
840 represented during all seasons (Figure S6). In fact, Eastern Asia is characterized by high  
841 RH even during summer, corroborating our hypothesis that aqueous phase processes  
842 may be an important missing piece in simulating the SOA formation. Recent studies  
843 have provided strong evidence that the uptake of water-soluble gas-phase oxidation  
844 products (even small carbonyls like formaldehyde and acetic acid) to the aqueous phase  
845 and their subsequent oxidation and oligomerization can lead to significant increases of  
846 SOA mass during pollution events (Gkatzelis et al., 2021). Overall, EMAC performs  
847 best over the Eastern Asian rural areas during summer and spring and worst in the  
848 vicinity of urban regions during fall and winter. Especially during wintertime, while the  
849 model simulates well the total OA, it significantly overpredicts POA (Figure 14c) and

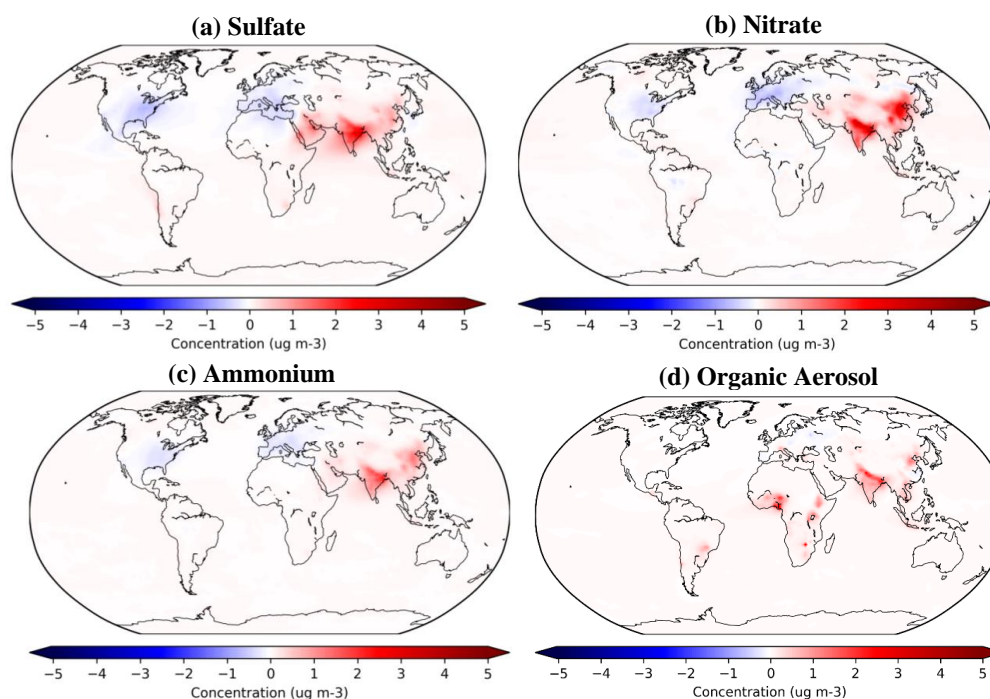


850 at the same time underpredicts SOA (especially M-OOA). This disagreement can be  
851 due to an overestimation of the POA formation from the emitted SVOC species, but  
852 also due to a missing mechanism that can significantly transform POA to SOA in the  
853 aerosol phase during winter.

854

## 855 **5 Aerosol Trends**

856 Here, the simulated 20-year global aerosol composition trends of fine aerosols are  
857 presented and discussed against trends calculated based on observational data. For this,  
858 it is vital to have data well distributed spatially and measured consistently in a  
859 comparable way at all observational sites within a region (Tørseth et al., 2012; Hand et  
860 al., 2011). These conditions, unfortunately, cannot be satisfied by the available PM<sub>1</sub>  
861 datasets (Figure 2). Instead, here we summarize the available observational data from  
862 each region for the 1<sup>st</sup> versus the 2<sup>nd</sup> decade of the examined period. This allows a rough  
863 statistical comparison between the two decades and can give insights on the overall  
864 tendency of the observed aerosol composition trends for each region. These trends are  
865 compared against the simulated PM<sub>1</sub> trends based on the respective spatiotemporal  
866 model data, as well as based on all the available model data for the entire model domain  
867 over the complete 20-year period (Figure 15). As the spatial and temporal AMS  
868 campaign distribution is much higher for regions in the northern than the southern  
869 hemisphere, only PM<sub>1</sub> data of the former is plotted here. PM<sub>2.5</sub> data from the large  
870 monitoring networks is also used to calculate the aerosol composition trends within the  
871 regions of North America, Europe, and Eastern Asia. These networks present  
872 cooperative measurement efforts that, among others, provide routinely filter based  
873 measured data of aerosol composition. Even though not every element is always  
874 measured at all sites and despite data gaps for some places, collectively, the networks'  
875 datasets provide the consistency and duration requirements mentioned above. The  
876 calculated trends are compared against PM<sub>2.5</sub> simulated results based on the respective  
877 spatiotemporal model data. It is worth noting that a comparison of filter PM<sub>2.5</sub> to AMS  
878 detected PM<sub>1</sub> is not completely straightforward. First, as seen in Sections 2.1.3 and  
879 2.2.1, there are expected compositional differences between the two size ranges,  
880 especially in polluted regions (Sun et al., 2020; Petit et al., 2015). Second, instrumental  
881 differences of the real-time on-line AMS (Decarlo et al., 2006) versus the non-real-time  
882 off-line filter instruments (Docherty et al., 2011; Hand et al., 2011) can manipulate the  
883 measurements in different ways, as discussed in the following sections.

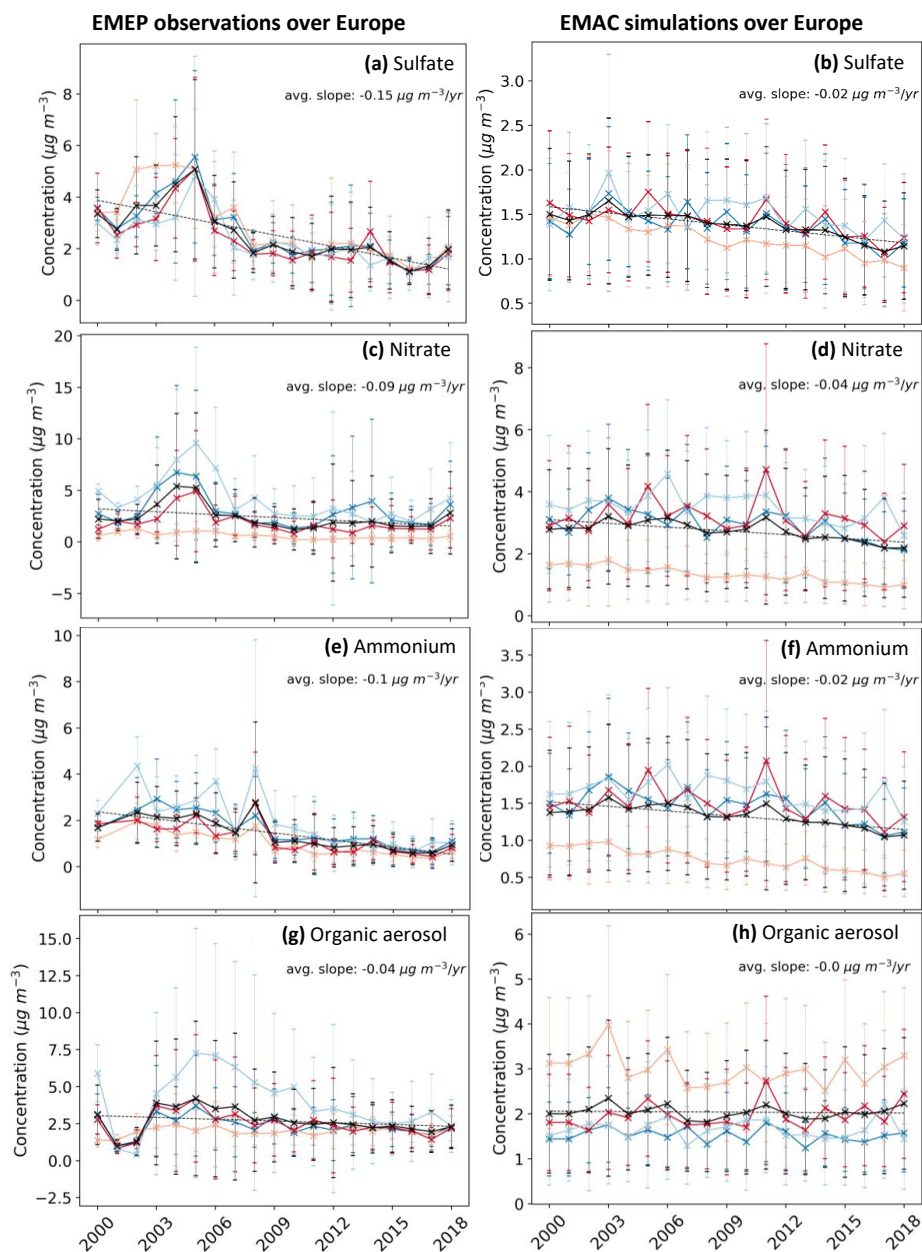


**Figure 15:** Simulated decadal change in (a) sulfate, (b) nitrate, (c) ammonium, and (d) anthropogenic organic aerosol concentrations between the 2000s and 2010s.

884

## 885 5.1 Europe

886 Figure 16 depicts the interannual and seasonal concentration change of filter  
887 measured  $\text{PM}_{2.5}$  components with a polynomial fitted trendline, in comparison to the  
888 corresponding concentration trends as calculated by the EMAC model. Both  
889 observations and the model reveal a concentration decrease for the three main inorganic  
890 components of  $\text{PM}_{2.5}$ , following the emission reductions during the last 20 years.  
891 Sulfate concentrations have decreased drastically during the last decade (i.e., -46%  
892 compared to 2000s). However, the simulated reduction is not so apparent mainly  
893 because filter observations show much higher concentrations during the first half of the  
894 2000s than model simulations. Until 2005, observed sulfate concentrations rose during  
895 all seasons, however, they rapidly dropped under the 2000 levels in the following years.  
896 The average decline rate is  $-0.15 \mu\text{g m}^{-3} \text{ yr}^{-1}$ , compared to the simulated rate of  $-0.02$   
897  $\mu\text{g m}^{-3} \text{ yr}^{-1}$ . AMS measurements (Figure 17) corroborate the findings of filter  
898 observations, revealing a drastic decrease in  $\text{PM}_1$  sulfate concentrations during the  
899 decade of 2010s (i.e., -18% compared to 2000s). EMAC underestimates European  $\text{PM}_1$



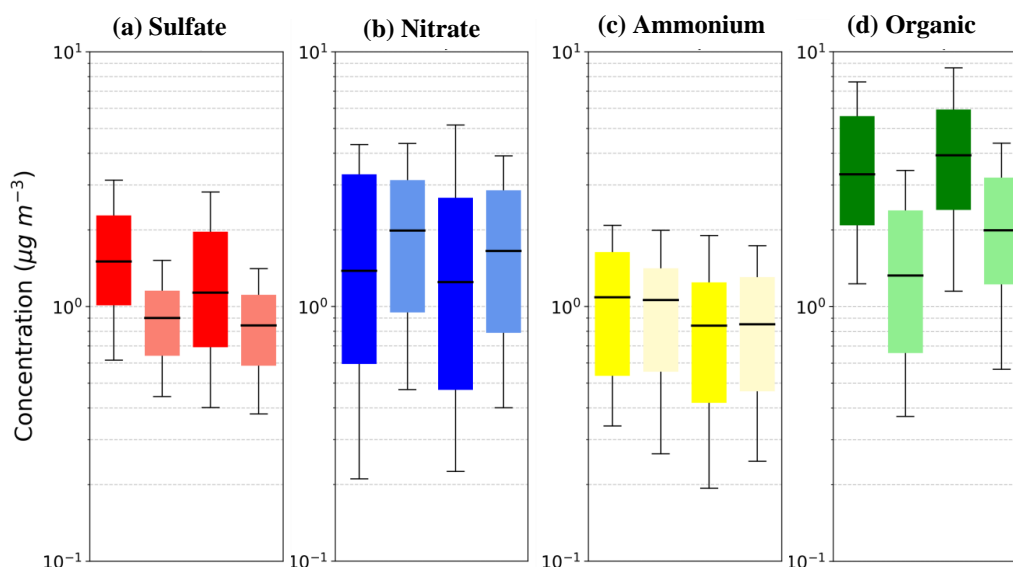
**Figure 16:** Temporal evolution of the observed (a, c, e, and g subplots on the left) and simulated (b, d, f, h subplots on the right) concentrations of  $\text{PM}_{2.5}$  sulfate (a, b), nitrate (c, d), ammonium, (e, f), and organic aerosol (g, h) during the period 2000–2018 over Europe. Black lines show the annual trend while the dark blue, light blue, orange, and red lines represent the seasonal trends during winter, spring, summer, and autumn. Ranges represent the  $1\sigma$  SD (standard deviation).



901 sulfate (Figure 10b) resulting in a less pronounced negative trend in its concentrations  
902 (i.e., -11%) since the model underestimation is more pronounced during the 2000s. The  
903 average simulated decadal change in sulfate  $PM_1$  concentrations for the entire European  
904 domain is -15% (Figure 15). Similar to sulfate, filter measured nitrate concentrations  
905 rose until 2005 (except during summer where they remain in low levels) and then  
906 quickly dropped again with an average rate of  $-0.09 \mu\text{g m}^{-3} \text{yr}^{-1}$  (Figure 16c). The high  
907 observed nitrate concentrations during the first half of the 2000s results in an average  
908 decrease of -35% between the two decades. On the other hand, the calculated change  
909 of AMS- $PM_1$  nitrate concentrations between the 2000s and the 2010s is -10%, which  
910 is similar to the simulated drop of -12%. However, it is worth mentioning that the model  
911 significantly overestimates the nitrate concentrations both in comparison to AMS  
912 measurements (Figure 11b) and to filter observations, especially during summer  
913 (Milousis et al., 2024). The analysis of model simulation and observations (both by  
914 AMS and filters) reveal that ammonium concentrations exhibit strong reductions  
915 between the decades of 2000s and 2010s. The average concentration reduction between  
916 the two decades is -21% based on the AMS observations, -13% based on the EMAC  
917 results for  $PM_1$  (or -16% for the entire European domain), and -56% for the  $PM_{2.5}$  filter  
918 observations. Therefore, the reduction of ammonium is much stronger based on the  
919 filter observations (i.e.,  $-0.1 \mu\text{g m}^{-3} \text{yr}^{-1}$ ) than based on AMS measurements or modeled  
920 data (i.e.,  $-0.02 \mu\text{g m}^{-3} \text{yr}^{-1}$ ). It is worth emphasizing that ammonium is clearly  
921 declining, even though  $NH_3$  emissions have only been slightly reduced. This apparent  
922 inconsistency can be attributed to the strong reductions of  $SO_2$  and  $NO_x$ . This results in  
923 reduced availability of acids (i.e.,  $H_2SO_4$  and  $HNO_3$ ) preventing the formation of  
924 ammonium and allowing the  $NH_3$  to reside in the gas phase. This is also verified by  
925  $NH_3$  observations, where no significant trends, and even statistical increases, have been  
926 observed despite reported reductions in  $NH_3$  emissions (Fagerli et al., 2016; Liu et al.,  
927 2024).

928 The downward trend of organic aerosol calculated based on the filter observations  
929 ( $-0.04 \mu\text{g m}^{-3} \text{yr}^{-1}$ ) is milder than that of inorganic components and differs between  
930 seasons (Figure 16e). During summer, there is no clear trend observed, while in winter,  
931 OC concentration soars after 2003 until 2005 when it starts to gradually drop until it  
932 reaches the concentration levels of the other seasons during the second half of 2010s.  
933 Irregularities in the early first decade could be owed to a lack of OC data (Fagerli et al.,  
934 2016). OC data during spring and autumn shows a mild downward trend after 2005 as





**Figure 17:** Decadal  $\text{PM}_{10}$  concentration trends in Europe expressed by the bar plots of the mass concentration (in  $\mu\text{g m}^{-3}$ ) for (a) sulfate, (b) nitrate, (c) ammonium, and (d) OA during the periods 2000 - 2010 (left) and 2011 - 2020 (right) as calculated from the AMS observational dataset (dark colors) and the corresponding simulation values (light colors). The upper and lower whiskers range from 10-90%, the quartiles from 25-75% of the dataset. The black line is the median.

935 well. Overall, the average difference of OC concentration between the two decades is -  
936 22%. However, model data does not corroborate this reduction; on the opposite a slight  
937 increase is calculated by the model during the last five years (Figure 16h). This agrees  
938 with the AMS observations which predict a positive OA trend (Figure 17d) with an  
939 average increase of  $+0.44 \mu\text{g m}^{-3}$  (or 10%) from the first to the second decade. Despite  
940 the prominent underestimation of  $\text{PM}_{10}$  OA by the model, the simulated  $\text{PM}_{10}$  OA trend  
941 is also positive with an average decadal increase of  $+0.55 \mu\text{g m}^{-3}$  (or 31%). Overall,  
942 inconsistencies between AMS and filter observations can be attributed to instrumental  
943 differences. First, is the size of particulate matter observed which is  $2.5 \mu\text{m}$  for filters  
944 and up to  $1 \mu\text{m}$  for the AMS. The size distribution of OA can be affected by multiple  
945 factors, including RH and chemical composition. Sun et al. (2020) have shown that the  
946  $\text{PM}_{10}/\text{PM}_{2.5}$  SOA ratio increases when RH is below 60% and the contribution of  
947 inorganic components in the aerosol decreases. This increase is related to differences  
948 in aerosol water content due to changes in aerosol hygroscopicity and phase state.  
949 Simulated data reveals that the frequency of RH dropping below 60% over European  
950 locations has marginally increased (by 1%) during the decade of 2010s. However, the

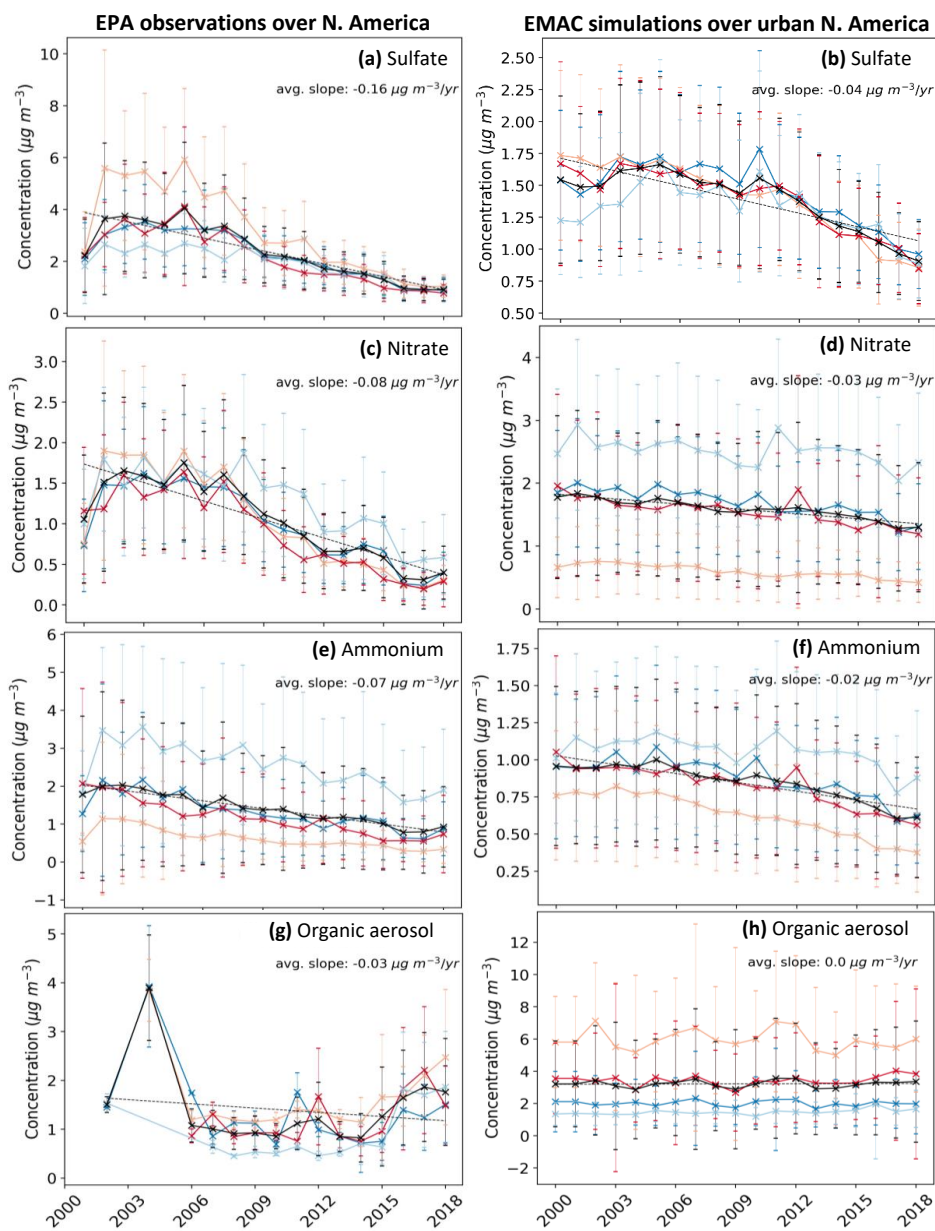


951 drastic reduction of sulfate and nitrate levels during the same period can explain the  
952 increase in PM<sub>1</sub> OA, as measured by the AMS, as opposed to the decrease in PM<sub>2.5</sub> OA  
953 observed by filters. Another important difference between the AMS and the filters is  
954 that the latter, in contrast to AMS, only detects the carbonaceous fraction (OC) of OA.  
955 Then, the ratio of the total organic mass (OM) to OC must be considered when  
956 comparing the measured OC to AMS or simulated OA. However, the OM:OC is  
957 broadly debated in literature. OM:OC is closely correlated to the oxygen to carbo ratio  
958 (O:C) and therefore it is dependent on the chemical aging degree of OA. For the range  
959 of SOA found in the atmosphere, Aiken et al. (2008) calculated the OM/OC ratios  
960 between 1.9 to 2.5. Similarly, the ratio for POA varies depending on the source and  
961 composition between 1.3 and 1.5 (Aiken et al., 2008). As the EMEP stations in Europe  
962 are a mix of urban and rural locations, the measured OC concentrations are typically  
963 multiplied by a median OM:OC value of 1.7. However, the oxidation capacity of the  
964 atmosphere has increased as anthropogenic emissions such as SO<sub>2</sub> have decreased  
965 (Dalsøren et al., 2016), leading to an increased oxidation rate of organic compounds  
966 and the formation of SOA. Consequently, a growing SOA fraction over the last 20 years  
967 would have been accompanied by a rising OM:OC ratio. It can be assumed that while  
968 the OC measured by the filters showed a slight downward trend (Figure 16g), a  
969 conversion into OA via adapted gradually increasing OM:OC ratios could have  
970 compensated the OC reduction and show a better matching trend compared to the AMS  
971 and EMAC OA.

972

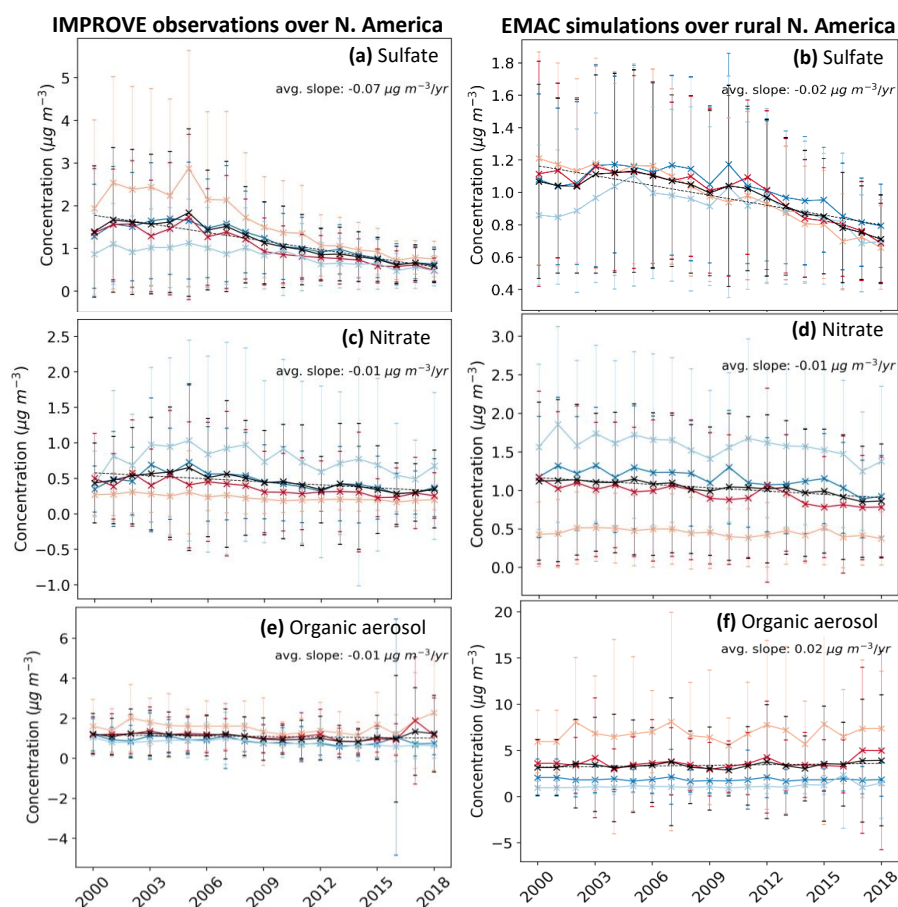
## 973 **5.2 North America**

974 Over North America, the filter measured inorganic aerosol compound  
975 concentrations declined strongly during the last 20 years, following their precursor  
976 emission reductions, with higher reductions over urban locations (Figure 18) and less  
977 over rural regions (Figure 19). Nitrate reductions are more pronounced over urban  
978 regions ( $-0.07 \mu\text{g m}^{-3} \text{yr}^{-1}$ ), especially during winter, while over rural locations, the  
979 decline is imperceptible ( $-0.01 \mu\text{g m}^{-3} \text{yr}^{-1}$ ) since the abundance of NH<sub>3</sub> have  
980 decelerated the decrease of NH<sub>4</sub>NO<sub>3</sub>. On the other hand, the drastic decrease of SO<sub>2</sub>  
981 emissions (Table S5, Figure S4) resulted in strong reductions of sulfate concentrations  
982 primarily over urban areas ( $-0.16 \mu\text{g m}^{-3} \text{yr}^{-1}$ ) but also over remote regions ( $-0.07 \mu\text{g m}^{-3}$   
983  $\text{yr}^{-1}$ ), especially during the summer seasons. Following the reductions of sulfate and



984

**Figure 18:** Temporal evolution of the observed (a, c, e, and g subplots on the left) and simulated (b, d, f, h subplots on the right) concentrations of PM<sub>2.5</sub> sulfate (a, b), nitrate (c, d), ammonium, (e, f), and organic aerosol (g, h) during the period 2000–2018 over urban locations in North America. Black lines show the annual trend while the dark blue, light blue, orange, and red lines represent the seasonal trends during winter, spring, summer, and autumn. Ranges represent the 1 $\sigma$  SD (standard

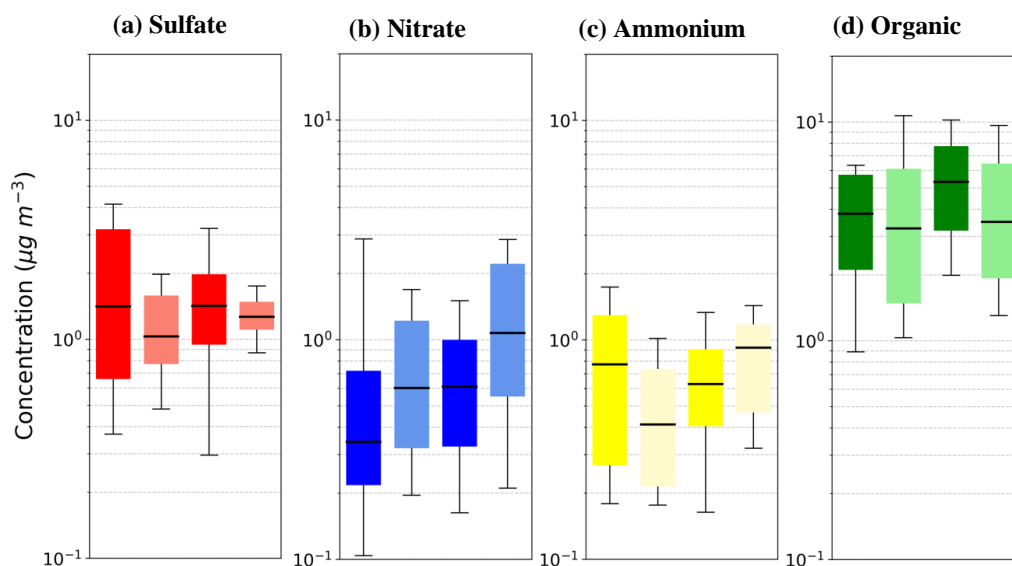


**Figure 19:** Temporal evolution of the observed (a, c, e, and g subplots on the left) and simulated (b, d, f, h subplots on the right) concentrations of PM<sub>2.5</sub> sulfate (a, b), nitrate (c, d), and organic aerosol (e, f) during the period 2000–2018 over rural locations in North America. Black lines show the annual trend while the dark blue, light blue, orange, and red lines represent the seasonal trends during winter, spring, summer, and autumn. Ranges represent the 1 $\sigma$  SD (standard deviation).

985 nitrate, ammonium decrease strongly over urban locations by  $-0.08 \mu\text{g m}^{-3} \text{yr}^{-1}$ ,  
 986 especially during the 2010s (Figure 18), even though  $\text{NH}_3$  emissions remain practically  
 987 unchanged (Figure S4). Similarly, over Canada, strong reductions in sulfate and nitrate  
 988 concentrations were observed by the Canadian Air and Precipitation Monitoring  
 989 Network (CAPMoN), driven by significant decreases in  $\text{SO}_2$  and  $\text{NO}_x$  emissions (Cheng  
 990 et al., 2022). While PM<sub>2.5</sub> concentrations decreased in eastern Canada, as observed by  
 991 the National Air Pollution Surveillance (NAPS), emission reductions were less  
 992 effective in the west, where large-scale wildfires overwhelmed these improvements and  
 993 even led to occasional increases in PM<sub>2.5</sub> concentrations (Yao and Zhang, 2024). These



994 regional differences over Canada are also captured by the EMAC model (Figure 15).  
995 Furthermore, EMAC simulates a weaker decline of sulfate concentrations over both  
996 rural and urban locations (Figures 18 and 19), mainly due to its tendency to  
997 underestimate sulfate concentrations during the 2000s and especially during summer.  
998 Reductions on the simulated nitrate and ammonium concentrations are also noticeable  
999 but to a lesser extent than on the filter observations (Figures 18 and 19). The observed  
1000 OA concentrations over urban regions decrease until 2009, however, they gradually  
1001 increase during 2010s by  $0.11 \mu\text{g m}^{-3} \text{ yr}^{-1}$ . On the other hand, the model calculated OA  
1002 concentration levels remain practically unchanged during the simulated period. Both  
1003 the simulated and the observed OA concentration trends are also very weak over the  
1004 rural and remote regions (Figure 19).



**Figure 20:** Decadal  $\text{PM}_{10}$  concentration trends in North America expressed by the bar plots of the mass concentration (in  $\mu\text{g m}^{-3}$ ) for (a) sulfate, (b) nitrate, (c) ammonium, and (d) OA during the periods 2000 - 2010 (left) and 2011 - 2020 (right) as calculated from the AMS observational dataset (dark colors) and the corresponding simulation values (light colors). The upper and lower whiskers range from 10-90%, the quartiles from 25-75% of the dataset. The black line is the median.

1005 Figure 20 depicts the decadal  $\text{PM}_{10}$  concentration trends in North America between  
1006 2000s and 2010s. The AMS data for  $\text{PM}_{10}$  aerosol composition is composed of  
1007 observational datasets from 30 field campaigns during the 2000s and 58 during the  
1008 2010s (Figure 2). This uneven distribution can statistically manipulate the calculations



1009 and hinder the extraction of valid statements for trends over North America. Sulfate  
1010 concentrations exhibit a tighter distribution during the 2<sup>nd</sup> decade (Figure 20); however,  
1011 the mean concentration remains unchanged between the two decades. On the other  
1012 hand, the simulated sulfate concentrations increase during the 2010s, mainly due to the  
1013 larger proportion of urban field campaigns during the second decade. Indeed, the model  
1014 simulates a reduction of the continental average sulfate concentrations by 20%, with  
1015 maximum differences exceeding  $1 \mu\text{g m}^{-3}$  over the Southeast US (Figure 15). This  
1016 contradicted behavior is also mirrored on nitrate concentrations where both the AMS  
1017 dataset and the corresponding simulated results produce a positive trend between the  
1018 two decades, while the simulated continental average nitrate concentrations decrease  
1019 (Figure 15). Furthermore, compared to AMS observations, the model tends to  
1020 underpredict sulfate concentrations and overpredict nitrate. This results in a strong  
1021 correlation of the simulated ammonium with nitrate exhibiting a significant positive  
1022 trend, which is not observed in the AMS dataset (Figure 20). Finally, as for PM<sub>2.5</sub> OA,  
1023 the observed and, to a lesser extent, the simulated PM<sub>1</sub> OA concentrations increase  
1024 slightly during the 2010s.

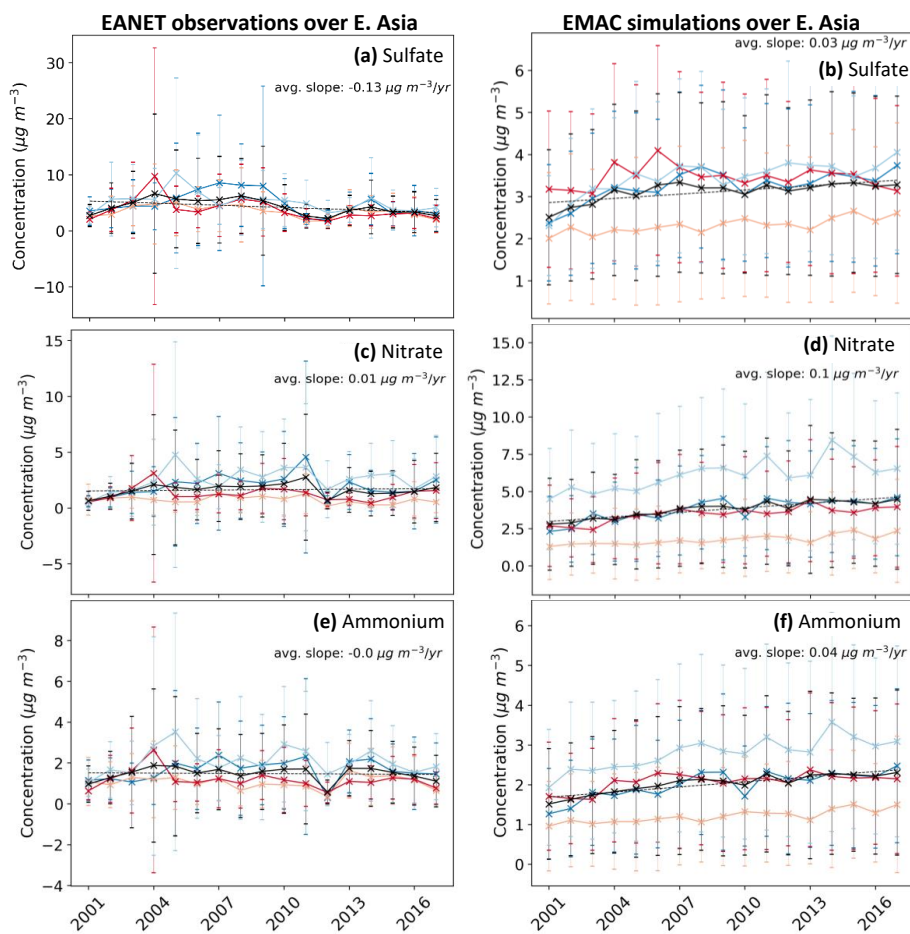
1025

### 1026 **5.3 Eastern Asia**

1027 EANET observations of PM<sub>2.5</sub> sulfate reveal a significant increase of its  
1028 concentrations until 2007 (Figure 21). However, in view of the upcoming Beijing  
1029 Olympic Games in 2008, the first SO<sub>2</sub> emission controls have started to be  
1030 implemented, and sulfate gradually reduced by  $-0.27 \mu\text{g m}^{-3} \text{ yr}^{-1}$ . By the end of 2017,  
1031 SO<sub>2</sub> emissions have been declined by 59% following the Clean Air Action (Zhai et al.,  
1032 2019), however, observed sulfate concentrations have decreased by only 23% due to an  
1033 increased dry deposition and oxidation rate of SO<sub>2</sub> during the same period (Fagerli et  
1034 al., 2016). EMAC fails to reproduce the reduction of sulfate concentrations after 2008  
1035 since the CAMS emission inventory assumes only a stabilization of SO<sub>2</sub> emissions after  
1036 the year 2013, instead of a strong decline (Figure S4). At the same period, NO<sub>x</sub> was  
1037 reduced by 21% and NH<sub>3</sub> by just 3% (Zhai et al., 2019). This however is not mirrored  
1038 in the observed nitrate trends (Figure 21), where nitrate reduces by only  $-0.05 \mu\text{g m}^{-3}$   
1039  $\text{yr}^{-1}$  after 2007. The strong SO<sub>2</sub> reduction hinders the decline of nitrate since reductions  
1040 in (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> release NH<sub>3</sub> to react with HNO<sub>3</sub> and form NH<sub>4</sub>NO<sub>3</sub>. In contrast to  
1041 observations, the simulated nitrate and ammonium continues to increase until the end  
1042 of 2010s following the trends in NO<sub>x</sub> emissions used as input in the model (Figure S4).



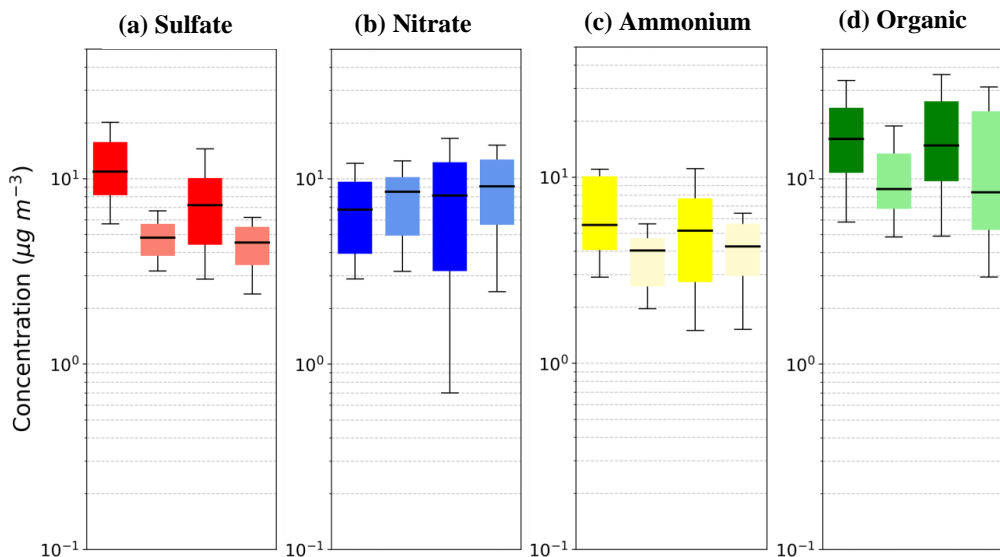
1043 The frequency of AMS field-campaigns started to grow significantly in Eastern Asia  
1044 only after 2008, while after 2013, the first consistent and aggressive emission controls  
1045 started in China under the Clean Air Action (Zhai et al., 2019). Thus, since 2013 marks  
1046 a significant year for Eastern Asia and due to the lack of AMS campaigns prior to 2006  
1047 in the region, the decade comparison for Eastern Asia is done for the periods of 2006-  
1048 2012 and 2013-2020. Between these two periods, AMS observations reveal a -17%  
1049 decline for sulfate, while the corresponding simulated sulfate concentrations reduce by  
1050 just -5% (Figure 22). Similar to  $PM_{2.5}$ , the average  $PM_1$  nitrate concentrations remain



**Figure 21:** Temporal evolution of the observed (a, c, e, and g subplots on the left) and simulated (b, d, f, h subplots on the right) concentrations of  $PM_{2.5}$  sulfate (a, b), nitrate (c, d), and ammonium (e, f) during the period 2000–2018 over Eastern Asia. Black lines show the annual trend while the dark blue, light blue, orange, and red lines represent the seasonal trends during winter, spring, summer, and autumn. Ranges represent the  $1\sigma$  SD (standard deviation).



1051 the same between the two periods with a marginal decline observed by the AMS and a  
1052 marginal increase simulated by EMAC, while the observed ammonium reduces by 18%  
1053 following the reduction in sulfate concentrations (Figure 22). In contrast to inorganic  
1054 aerosol precursors, the anthropogenic VOC emissions over Eastern Asia continue to  
1055 increase even after 2013, mostly due to the use of solvents but also due to the energy  
1056 transformation and industrial sector (Hoesly et al., 2018). Thus, both the observed and  
1057 the simulated PM<sub>1</sub> OA concentrations increase between the two examined periods by  
1058 15% and 33%, respectively (Figure 22).



**Figure 22:** Decadal PM<sub>1</sub> concentration trends in Eastern Asia expressed by the bar plots of the mass concentration (in  $\mu\text{g m}^{-3}$ ) for (a) sulfate, (b) nitrate, (c) ammonium, and (d) OA during the periods 2006 - 2012 (left) and 2013 - 2020 (right) as calculated from the AMS observational dataset (dark colors) and the corresponding simulation values (light colors). The upper and lower whiskers range from 10-90%, the quartiles from 25-75% of the dataset. The black line is the median.

1059

## 1060 6 Conclusion

1061 This study investigates global trends in atmospheric aerosol composition over the  
1062 past two decades, using the EMAC chemistry-climate model and the CAMS  
1063 anthropogenic emissions inventory. Results integrate model outputs with global  
1064 observational data from 2000-2020, covering PM<sub>2.5</sub> composition from regional  
1065 monitoring networks (e.g., EMEP in Europe) and PM<sub>1</sub> composition from 744 AMS  
1066 observational datasets at 169 sites worldwide. Findings reveal substantial regional





1067 variations in aerosol composition driven by industrial activities, energy production, and  
1068 air quality regulations, highlighting the complexity of air pollution dynamics and its  
1069 management.

1070 AMS field campaign data show that OA are the dominant  $PM_{10}$  component globally,  
1071 especially in tropical and subtropical regions affected by biomass burning and biogenic  
1072 VOC emissions. Sulfate is the primary inorganic compound across most areas, though  
1073 nitrate predominates in Europe and Eastern Asia. Notably, North America shows  
1074 unexpected sulfate dominance, likely due to seasonal sampling bias. HOA levels are  
1075 higher in North America and Eastern Asia, while BBOA is prominent in rural Europe  
1076 and tropical regions. OOA, particularly aged M-OOA, is the largest OA contributor in  
1077 rural regions across all studied areas.

1078 For  $PM_{2.5}$  composition, global filter observations indicate OA as the primary  
1079 component in most regions, notably in Southern Hemisphere tropical forests. In Eastern  
1080 Asia, OA and elemental carbon (EC) are prominent, while OA and sulfate have similar  
1081 importance in rural North America. Globally, sulfate constitutes roughly 50% of the  
1082 inorganic  $PM_{2.5}$  mass, followed by nitrate and ammonium. However, sulfate dominance  
1083 observed in filter samples contrasts with AMS findings, likely due to sampling artifacts.  
1084 Regionally, sulfate is highest in the Middle East, while nitrate plays a significant role  
1085 in Europe. Across eight regions,  $PM_{2.5}$  averages are: 21% sulfate, 12% nitrate, 10%  
1086 ammonium, 2% sodium, 3% chloride, 40% OA, and 12% EC.

1087 The EMAC model confirms OA as the dominant component of fine aerosols  
1088 globally, with the highest concentrations in regions influenced by biomass burning,  
1089 such as tropical forests and savannas. Northern industrialized regions exhibit  
1090 substantial OA levels (30-35%) from fossil and biofuel combustion. While EMAC  
1091 successfully reproduces the prominence of SOA, it struggles to accurately simulate  
1092 aged SOA in areas like Eastern Asia. The model further suggests that nitrate surpasses  
1093 sulfate in  $PM_{2.5}$  composition in Europe, North America, and Eastern Asia, consistent  
1094 with AMS findings but differing from some filter observations. Ammonium mirrors  
1095 sulfate and nitrate distribution, with significant contributions in populated and  
1096 agricultural regions. Mineral dust and sea salt emissions also play key roles regionally.  
1097 Overall, EMAC provides valuable insights into global fine aerosol composition, while  
1098 indicating areas for model refinement.

1099 This study presents a 20-year analysis of global trends in fine aerosol composition,  
1100 comparing EMAC model simulations with observed trends. Given limited and



1101 inconsistent  $PM_1$  datasets, the analysis focuses on broad regional trends across the first  
1102 and second decades, using primarily Northern Hemisphere AMS campaign data and  
1103  $PM_{2.5}$  data from major monitoring networks in North America, Europe, and East Asia.  
1104 While these comparisons offer insights, they are complicated by compositional  
1105 differences between  $PM_1$  and  $PM_{2.5}$  and by differences between real-time AMS and  
1106 non-real-time filter-based methods.

1107 Both filter-based data and EMAC simulations show a major decline in key inorganic  
1108 components over Europe, especially in sulfate, which dropped by 46% in the last  
1109 decade. The EMAC model, however, underestimates the sulfate reduction due to initial  
1110 discrepancies in early 2000s concentrations. Nitrate and ammonium also declined  
1111 significantly, though the model overestimates nitrate levels. Organic aerosol (OA)  
1112 trends vary by method: filter data indicate a slight decrease, while AMS data and  
1113 simulations suggest a mild OA increase in  $PM_1$ , likely due to differences in particle size  
1114 ( $PM_{2.5}$  vs.  $PM_1$ ) and instrument detection capabilities (filter-based OC vs. AMS OA).

1115 In North America, filter-based measurements reveal sharp declines in inorganic  
1116 aerosol compounds, particularly in urban areas. Nitrate and sulfate concentrations  
1117 decreased significantly due to lower  $SO_2$  and  $NO_x$  precursor emissions, with  
1118 ammonium levels following this trend, although ammonia itself remained stable in the  
1119 2010s. The EMAC model, however, simulates a weaker sulfate and nitrate decline,  
1120 underestimating sulfate in the early 2000s while overestimating nitrate. Observed OA  
1121 concentrations in urban North America decreased until 2009, then rose in the 2010s, a  
1122 trend only partially captured by the model.  $PM_1$  sulfate and nitrate levels from AMS  
1123 data show inconsistent trends, with the model generally underestimating sulfate and  
1124 overestimating nitrate, leading to a positive ammonium trend in the model not observed  
1125 in AMS data.

1126 In Eastern Asia, EANET  $PM_{2.5}$  data show rising sulfate concentrations until 2007,  
1127 followed by a decline as  $SO_2$  emission controls implemented prior to the 2008 Beijing  
1128 Olympics. Despite a 59% reduction in  $SO_2$  emissions by 2017, sulfate concentrations  
1129 fell by only 23%, likely due to increased dry deposition and oxidation rates. The EMAC  
1130 model does not fully capture this trend, as it assumes stable  $SO_2$  emissions post-2013  
1131 rather than a steep decline. Similarly, while observed nitrate and ammonium levels  
1132 show minimal reductions after 2007, the model inaccurately projects continued  
1133 increases, reflecting discrepancies in  $NO_x$  emissions trends. AMS data indicate a 17%  
1134 reduction in  $PM_1$  sulfate from 2006–2012 to 2013–2020, compared to a 5% reduction



1135 in the model, with observed PM<sub>1</sub> OA concentrations increasing by 15% and model  
1136 predictions showing a 33% rise, driven by sustained VOC emissions from solvents and  
1137 industrial sources.

1138 Overall, despite the complexities and inconsistencies in long-term aerosol trend  
1139 analysis due to instrumental and methodological differences, this study highlights the  
1140 importance of consistent, long-term global aerosol trend analysis. By integrating model  
1141 results and observational data over 20 years, the study reveals significant  
1142 spatiotemporal changes in atmospheric aerosol composition over different regions of  
1143 the planet, largely driven by recent changes in aerosol precursor emissions.

1144

1145 **Code and data availability.** The usage of MESSy (Modular Earth Submodel System)  
1146 and access to the source code is licensed to all affiliates of institutions which are  
1147 members of the MESSy Consortium. Institutions can become a member of the MESSy  
1148 Consortium by signing the “MESSy Memorandum of Understanding”. More  
1149 information can be found on the MESSy Consortium website: [http://www.messy-](http://www.messy-interface.org)  
1150 [interface.org](http://www.messy-interface.org) (last access: 8 November 2024). The data produced in the study are  
1151 available from the author upon request

1152

1153

1154 **Authors contribution:** APT designed the research with contributions from VAK. APT  
1155 and VAK developed ORACLE-lite. AM and VAK implemented ISOROPIA-lite in  
1156 EMAC. SS selected all AMS observations and NM provided specific observations from  
1157 sites over the Mediterranean. APT performed the simulations. APT and SS analyzed  
1158 the results. APT, SS and VAK wrote the manuscript with contributions from NM and  
1159 AM. All co-authors made revisions and corrections.

1160

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