



1 **Pelagic coccolithophore production and dissolution and their impacts on particulate**
2 **inorganic carbon cycling in the western North Pacific**

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10 **Abstract.** Coccolithophores, a type of single-celled phytoplankton that is abundant in global oceans, are closely associated
11 with the carbonate pump and thus play a crucial role in the marine carbon cycle. Here we investigated coccolithophore
12 abundances, species compositions, coccolithophore calcium carbonate (CaCO_3 as calcite) and particulate inorganic carbon
13 (PIC) concentrations in the upper water column of the western North Pacific Ocean, along a meridional transect spanning the
14 oligotrophic subtropical gyre and the nutrient-rich Kuroshio-Oyashio transition region. Our results revealed that
15 *Umbellosphaera tenuis* was the dominant coccolithophore species in the former, while *Emiliana huxleyi* and *Syracosphaera*
16 spp. dominated in the latter. Coccolithophore calcite contributed a major fraction of the PIC standing stocks above a depth of
17 150 m, among which *E. huxleyi* was the most important producer while less abundant and larger species also played a role.
18 The coccolithophore CaCO_3 production rate in the subtropical gyre ($0.62 \text{ mol m}^{-2} \text{ yr}^{-1}$) was ~5-fold higher than that in the
19 Kuroshio-Oyashio transition region ($0.14 \text{ mol m}^{-2} \text{ yr}^{-1}$), indicating that inorganic carbon metabolism driven by
20 coccolithophores is relatively strong in oligotrophic ocean waters. Using a box model including coccolithophore CaCO_3
21 production and metabolic calcite saturation state, we demonstrated that CaCO_3 dissolution associated with organic carbon
22 metabolism can generate excess alkalinity in the oversaturated upper water column of the western North Pacific Ocean. Results
23 of our study highlight the critical role of coccolithophores in CaCO_3 production and dissolution; knowledge of these processes
24 is important to assess PIC cycling and carbonate pump efficiency in the pelagic ocean.

25



26 **1 Introduction**

27 Calcium carbonate (CaCO_3) production and dissolution are two major processes associated with CaCO_3 cycling in the ocean,
28 and a key component of the global oceanic carbon cycle (Broecker and Peng, 1982) via the so-called carbonate pump (Volk
29 and Hoffert, 1985). Production of biogenic CaCO_3 by calcifying plankton in the euphotic zone elevates the partial pressure of
30 carbon dioxide (CO_2) in seawater (e.g., Feely et al., 2002), while ballasting of sinking particles can promote the transport of
31 carbon from the surface to deep sea and marine sediments (e.g., Armstrong et al., 2001; Klaas and Archer, 2002). Dissolution
32 of CaCO_3 in the water column acts as a buffer to facilitate ocean sequestration of atmospheric CO_2 and has the effect of
33 reducing the rate of ocean acidification (Feely et al., 2004; Barrett et al., 2014). This acidification feedback mechanism makes
34 it harder for calcifying organisms to produce their skeletons, and thus adversely affects marine ecosystems (Feely et al., 2004;
35 Steiner et al., 2018). Therefore, quantification of marine CaCO_3 production and dissolution is of vital importance in
36 determining the response of marine ecosystems to changes in the partial pressure of CO_2 .

37 Marine CaCO_3 occurs in the form of calcite, aragonite and high-magnesium calcite. Coccolithophores are a key, single-
38 celled phytoplankton taxonomic group, responsible for a large percentage (30–60 %) of modern oceanic CaCO_3 production
39 and 10–20 % of marine primary production on a global scale (Poulton et al., 2006, 2013). Coccolithophore calcite accounts
40 for a major fraction of the CaCO_3 exported to the deep sea and sediments (Broerse et al., 2000; Young and Ziveri, 2000; Rigual
41 Hernández et al., 2020). Data assessment along a northeast Pacific transect from Hawaii to Alaska suggested that
42 coccolithophore calcite comprises 90 % of the total CaCO_3 production in the euphotic zone, while pteropods and foraminifera
43 only play a minor role (Ziveri et al., 2023). However, large uncertainties remain in estimates of the production rate of CaCO_3
44 in the upper ocean (Balch et al., 2007; Berelson et al., 2007; Smith and Mackenzie, 2016; Ziveri et al., 2023). Based on a
45 global compilation of CaCO_3 production using in situ ^{14}C incubations, Daniels et al. (2018) found that calcification rate ranged
46 from 10 to 600 $\text{mg m}^{-2} \text{d}^{-1}$. A recent estimate of CaCO_3 biomass from three main pelagic calcifying plankton groups also
47 suggested large variation in CaCO_3 production in the eastern North Pacific Ocean, ranging from 110 to 729 $\text{mg m}^{-2} \text{d}^{-1}$ (Ziveri
48 et al., 2023). The generally low coverage of observations and the considerable spatiotemporal variation shown by available
49 data, along with the current scarcity of studies, most likely result in deviations among regional estimates.

50 CaCO_3 dissolution is generally assumed to mainly occur below the saturation horizon. However, this assumption has been



51 challenged by an increasing number of studies which suggest considerable dissolution in the oversaturated upper water-column
52 where CaCO_3 saturation state (Ω) is greater than 1 (Sabine et al., 2002; Chung et al., 2003; Berelson et al., 2007; Barrett et al.,
53 2014). Shallow-water CaCO_3 dissolution is supported by excess alkalinity/calcium (Feely et al., 2002, 2004; Cao and Dai,
54 2011; Subhas et al., 2022), decreased sinking fluxes of particulate inorganic carbon (PIC) measured using the ^{234}Th method,
55 and sediment trap data from the upper ocean (Dong et al., 2019; Roca-Martí et al., 2021). One possible mechanism for CaCO_3
56 dissolution in calcite-oversaturated shallow waters is the dissolution of more soluble forms of CaCO_3 including aragonitic
57 pteropods and high-Mg fish calcites (Honjo et al., 2008; Wilson et al., 2009; Dong et al., 2019; Folkerts et al., 2024; Oehlert
58 et al., 2024). An alternative hypothesis is that microbial oxidation of organic matter produces an acidic microenvironment
59 conducive for carbonate dissolution (Bishop et al., 1980; Milliman et al., 1999). The digestive system of grazing zooplankton
60 may also contain acidic conditions facilitating CaCO_3 dissolution (Pond et al., 1995; White et al., 2018).

61 The North Pacific Ocean is a vital region for modulating the carbon cycle, as it accounts for ~25 % of the global ocean sink
62 for atmospheric CO_2 (Takahashi et al., 2009). In the eastern North Pacific Ocean, CaCO_3 production, export, and dissolution
63 have been studied along a transect from Hawaii to Alaska (Dong et al., 2019, 2022; Naviaux et al., 2019; Subhas et al., 2022;
64 Ziveri et al., 2023), which revealed that depth-integrated CaCO_3 production in the nutrient-rich subpolar gyre is twice as high
65 as that in the nutrient-poor subtropical gyre. This contrast, however, is smaller than the sixfold to sevenfold difference based
66 on satellite estimates of surface PIC, indicating the importance of coccolithophore CaCO_3 production in deep waters. These
67 investigations also suggested that widespread shallow-water CaCO_3 dissolution is driven by metabolic activity along the
68 Hawaii-to-Alaska transect.

69 Here, we determined the abundances and species compositions of coccolithophore, as well as the concentrations of
70 coccolithophore calcite and PIC based on both Niskin bottle and in situ pump sampling in the upper water column of the
71 western North Pacific Ocean. Additionally, we conducted measurements of environmental conditions such as nutrient and
72 carbonate chemistry parameters. The aims of this research were to answer the following questions: (1) What is the distribution
73 of coccolithophore abundances and species compositions across the oligotrophic-nutrient replete environmental gradient? (2)
74 What is the contribution of coccolithophores to CaCO_3 production in the euphotic zone? (3) Does shallow-water CaCO_3
75 dissolution occur in the western North Pacific Ocean, and what is the significance of metabolic activities in driving dissolution



76 in oversaturated ambient conditions?

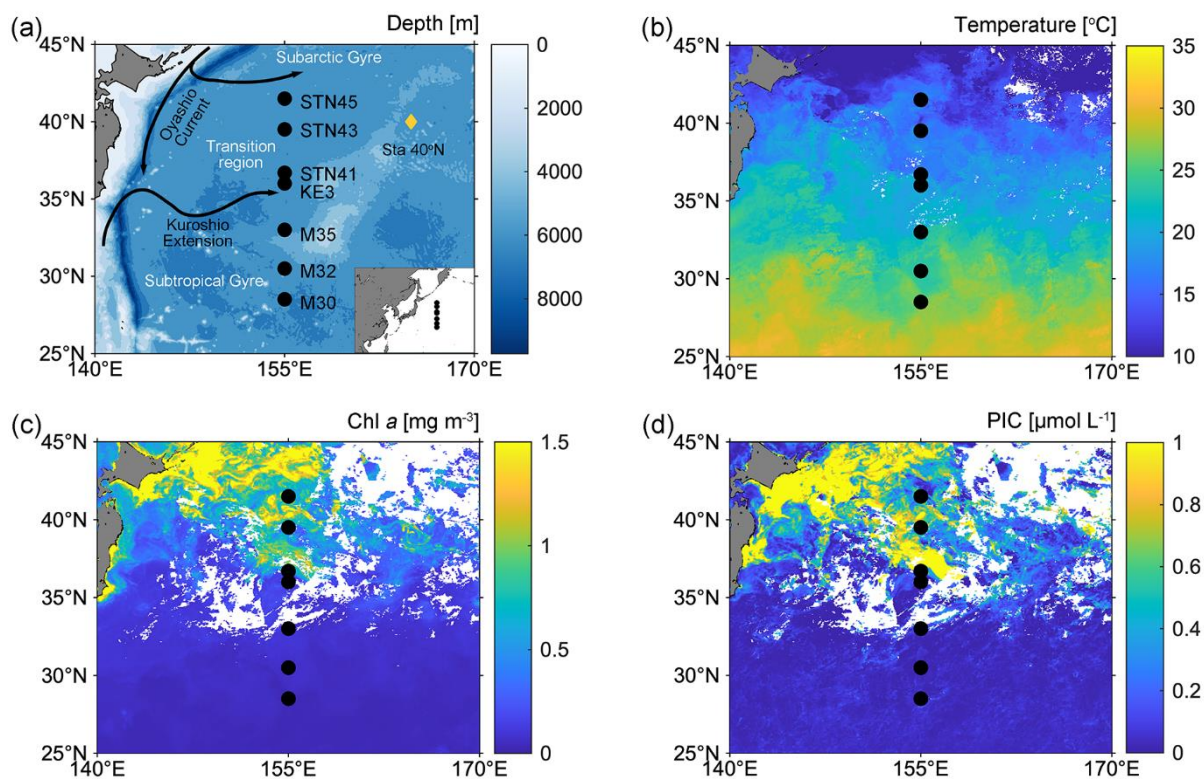
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78 **2 Methods**

79 **2.1 Sample collection**

80 Sampling was conducted onboard R/V *Tan Kah Kee* during cruise NORC2022-306 from 09 June to 25 July 2022. The cruise
81 trajectory crossed from the oligotrophic North Pacific Subtropical Gyre (NPSG) to the relatively nutrient-rich Kuroshio-
82 Oyashio transition region along the 155°E meridian (Fig. 1a; Table S1). Seven sampling stations can be divided into those
83 located in the NPSG region, including stations M30, M32 and M35, characterized by high sea-surface temperature (SST) and
84 low surface chlorophyll *a* (Chl *a*) and PIC concentrations, and those located in the Kuroshio-Oyashio transition region,
85 including stations KE3, STN41, STN43 and STN45, featuring lower SST, but higher Chl *a* and PIC concentrations (Fig. 1b-
86 d).

87



88

89 **Fig 1.** (a) Map of the western North Pacific Ocean showing sampling stations (black filled circles) and major surface currents
90 (solid black lines). The yellow diamond indicates station 40° N at which particulate inorganic carbon (PIC) sinking flux was
91 investigated by Honda et al., (2002); (b–d) satellite-based temperature, chlorophyll *a* (Chl *a*) and particulate inorganic carbon
92 (PIC) concentrations in surface water in June 2022 (data from the Moderate Resolution Imaging Spectroradiometer (MODIS)-
93 Aqua satellite; <https://oceancolor.gsfc.nasa.gov/l3/>).

94

95 Water samples were collected within the water column above 300 m depth using Niskin bottles on a rosette system equipped
96 with SBE-911 conductivity-temperature-depth (CTD) sensors (Sea-Bird Electronics, Inc., Bellevue, WA, USA). For PIC
97 analyses, 24 L of seawater were collected using acid-cleaned fluorinated bottles and filtered through two quartz microfiber
98 (QMA) filters (1.0 μm pore size, 25 mm diameter). For coccolithophore analyses, 2–4 L of seawater were collected and gently
99 filtered through polycarbonate membranes (0.8 μm pore size, 25 mm diameter), using a vacuum pump at <20 mm Hg pressure.



100 Filters were oven-dried at 60°C and stored in plastic petri dishes.

101 Size-fractionated particles were collected using McLane Research in situ pumps. Filter holders were loaded with a 51 µm
102 Sefar polyester mesh prefilter followed by paired Whatman QMA filters. Hereafter, we refer to the two particle size fractions
103 as large and small size fractions (LSF, > 51 µm and SSF, 1–51 µm, respectively). A 1/4 subsample of the 51 µm polyester mesh
104 prefilter and two circles of 23 mm diameter subsample of the QMA filter were analyzed for LSF and SSF PIC concentrations,
105 respectively, and the sum of the two fractions yielded the total PIC (PIC_{total}) concentration.

106 2.2 Sample analyses

107 PIC concentrations were determined by measuring the amount of CO₂ released after acid treatment of the filters using a Thermo
108 Delta V Plus isotope ratio mass spectrometer (IRMS, Thermo Fisher, USA) coupled with a Thermo Gasbench II system at the
109 Center for Isotope Geochemistry and Geochronology of the Laoshan Laboratory (Li et al., 2021). International reference
110 materials of calcite NBS-18 and IAEA-603 were measured for calibration. The PIC analytical precision was better than 10 %
111 (one standard deviation, 1SD).

112 A portion of the filters was mounted with a carbon sticky tab on a stub and gold-coated prior to analysis using a Quanta 650
113 FEG field-emission scanning electron microscope (SEM). The coccosphere cell or detached coccolith concentrations (CC,
114 cells or coccoliths L⁻¹) were estimated as follows:

$$115 \text{CC} = (F * C)/(V * S) \quad (1)$$

116 where F is the effective filtration area (336.9 mm²), C is the total number of coccosphere cells or detached coccoliths, V is
117 the filtered seawater volume, and S is the total area of fields of view (mm²). This cell counting strategy gives a detection limit
118 of at least 1.87 cells mL⁻¹ (Bollmann et al., 2002). Coccolithophore species identification followed Young et al. (2003) and the
119 Nannotax3 website (<http://ina.tmsoc.org/Nannotax3/>). Individual coccolithophore calcite content was calculated by
120 multiplying the number of coccoliths per cell by the average coccolith calcite mass of a given species (Young and Ziveri, 2000;
121 Yang and Wei, 2003; Boeckel and Baumann, 2008; Beuvier et al., 2019; Jin et al., 2022). All the biometry work based on SEM
122 images was conducted using imageJ free software (imagej.nih.gov/ij/) and Coccobiom2-SEM measuring macro (Young, 2015).

123 2.3 Estimation of CaCO₃ production rate

124 CaCO₃ production rates above 150 m were determined by dividing measurements of the living CaCO₃ standing stock (which



125 only included whole coccosphere cells and excluded loose coccoliths) by the coccolithophore turnover time, which is 0.7–10
126 days with a growth rate ranging from 0.1 to 1.5 cell divisions day⁻¹ (Krumhardt et al., 2017; Ziveri et al., 2023). Uncertainty
127 in the CaCO₃ standing stock estimates, which were obtained by vertically integrating PIC concentrations above a depth of 150
128 m, was typically ±10 % (1SD).

129 A Monte Carlo-based probabilistic approach was used to determine the CaCO₃ production rate and the uncertainties
130 associated with the turnover time using a flat probability distribution in MATLAB visualized by a violin plot (Hoffmann, 2015).
131 To obtain an annual CaCO₃ production based on our field observations, we used the ratio of satellite-derived PIC for July 2022
132 to annual climatology PIC (data from the NASA Goddard Space Flight Center’s Ocean Ecology Laboratory) to calibrate for
133 potential seasonal variability (Ziveri et al., 2023).

134 **2.4 Influence of environmental conditions on coccolithophores**

135 The redundancy analysis (RDA) function in the vegan package in R, in combination with Monte Carlo permutation, were
136 performed to assess the relative importance of environmental variables in explaining the overall variation in the composition
137 of the coccolithophore community (Oksanen et al., 2007). The contribution of each environmental variable to community
138 variation was determined by hierarchical partitioning in canonical analysis via the ‘dbRDA’ function in the “rdacca.hp”
139 package in R (Lai et al., 2022). We further conducted random forest analyses to identify the main predictors of coccolithophore
140 abundance with the “randomForest” package in R (Liaw and Wiener, 2002).

141 **2.5 Evaluation of shallow-water CaCO₃ dissolution**

142 To estimate the effect of microenvironment undersaturation driven by microbial oxidation of organic matter, we assumed that
143 aerobic metabolic activity of marine particles consumes all ambient oxygen and alters ambient dissolved inorganic carbon
144 (DIC) and total alkalinity (TA). This process decreases microenvironment Ω and thus causes CaCO₃ dissolution. We calculated
145 Ω due to in situ metabolism (defined as Ω_{met}) using the concurrently altered DIC, TA and soluble reactive phosphate (SRP)
146 concentrations following Subhas et al., (2022):

$$147 \quad DIC_{met} = DIC + [O_2] * R_{CO} \quad (2a)$$

$$148 \quad TA_{met} = TA - [O_2] * R_{NO} \quad (2b)$$

$$149 \quad SRP_{met} = SRP + [O_2] * R_{PO} \quad (2c)$$



150 where R_{CO} , R_{NO} , and R_{PO} denote the Redfield ratio of carbon to oxygen (0.688), nitrate to oxygen (0.0941), and phosphate to
151 oxygen (0.0059), respectively (Anderson and Sarmiento, 1994).

152 A one-dimensional (1-D) model was used to diagnose the sinking flux and $CaCO_3$ dissolution rate in the upper 1000 m of
153 the water-column by coccolithophore $CaCO_3$ production, coccolith dissolution kinetics and Ω_{met} (Dong et al., 2019; Subhas et
154 al., 2022). The model assumes that dissolution of $CaCO_3$ particles occurs when the water column $\Omega_{met} < 1$. The magnitude of
155 dissolution thus depends on the initial sinking flux, Ω_{met} and particle residence time as follows:

$$156 \quad Flux_{zi} = Flux_{zi-1} * \left(1 - R_{diss} * \frac{Z_i - Z_{i-1}}{w} \right) \quad (3)$$

157 where $Flux_{zi}$ and $Flux_{zi-1}$ ($mmol \ m^{-2} \ d^{-1}$) denote the $CaCO_3$ sinking flux at a given depth Z_i and its overlying depth Z_{i-1} ,
158 respectively, R_{diss} ($g \ g^{-1} \ d^{-1}$) denotes $CaCO_3$ dissolution rate which is a function of the Ω_{met} at depth Z_{i-1} (Subhas et al., 2018;
159 Subhas et al., 2022) and w ($m \ d^{-1}$) denotes the particle sinking rate.

160 TA regeneration rate (R_{TA} , $\mu mol \ kg^{-1} \ yr^{-1}$) at depth Z_i resulting from shallow-water $CaCO_3$ dissolution was calculated based
161 on the following equation:

$$162 \quad R_{TA_{zi}} = \frac{2 * Flux_{zi} * R_{diss}}{\rho * w} \quad (4)$$

163 where ρ denotes the density of seawater ($1029 \ kg \ m^{-3}$).

164

165 **3 Results**

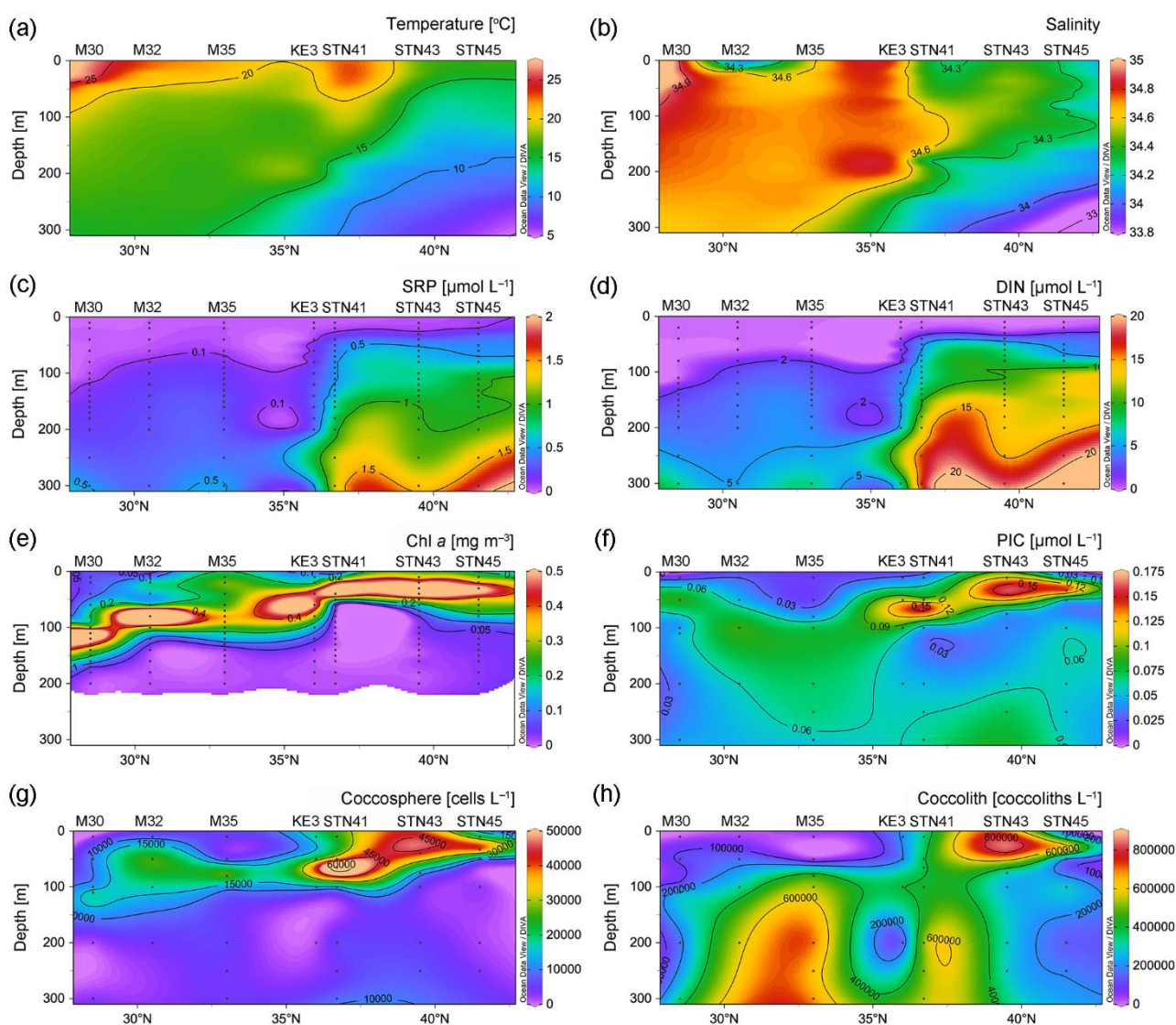
166 **3.1 Hydrography and nutricline**

167 Temperature and salinity were high in surface water and gradually decreased with increasing depth above 300 m.
168 Hydrochemical variables clearly exhibited a south to north trend. Temperature and salinity were highest at the surface of station
169 M30, due to strong net evaporation in the subtropical gyre (Fig. 2a and b). There was a northward decrease in temperature and
170 salinity due to the influence of upwelling in the subarctic gyre. The surface mixed layer depth, defined as the depth where
171 potential density increases by $0.03 \ kg \ m^{-3}$ compared to that at the sea surface, varied around 11–25 m.

172 In contrast to temperature and salinity and as expected, the distribution of dissolved inorganic nitrogen (DIN, nitrate plus
173 nitrite), SRP and dissolved silicate (DSi) showed a generally northward increasing pattern (Figs. 2c–d and S1). Surface DIN



174 concentrations were below the detection limit in the NPSG region and averaged $0.02 \mu\text{mol L}^{-1}$ in the Kuroshio-Oyashio
175 transition region. The top of the nutricline, defined as the depth where DIN concentrations reach $0.1 \mu\text{mol L}^{-1}$, ranged from
176 110 m at station M30 to 20 m at station STN45. The ammonium (NH_4^+) concentration above 100 m at station STN45 was
177 notably higher than that at other stations (Fig. S1b). Similar to the nutricline distribution, the deep chlorophyll maximum
178 (DCM) depth gradually shoaled northward from 110 m at station M30 in the NPSG region to 33 m at station STN45 in the
179 Kuroshio-Oyashio transition region (Fig. 2e).



180



181 **Fig. 2.** Vertical depth distributions of (a) temperature, (b) salinity and concentrations of (c) soluble reactive phosphate (SRP),
182 (d) dissolved inorganic nitrogen (DIN, nitrate plus nitrite), (e) Chlorophyll *a* (Chl *a*), (f) particulate inorganic carbon (PIC),
183 (g) coccosphere cell and (h) detached coccolith in the upper 300 m of the water column in the study area.

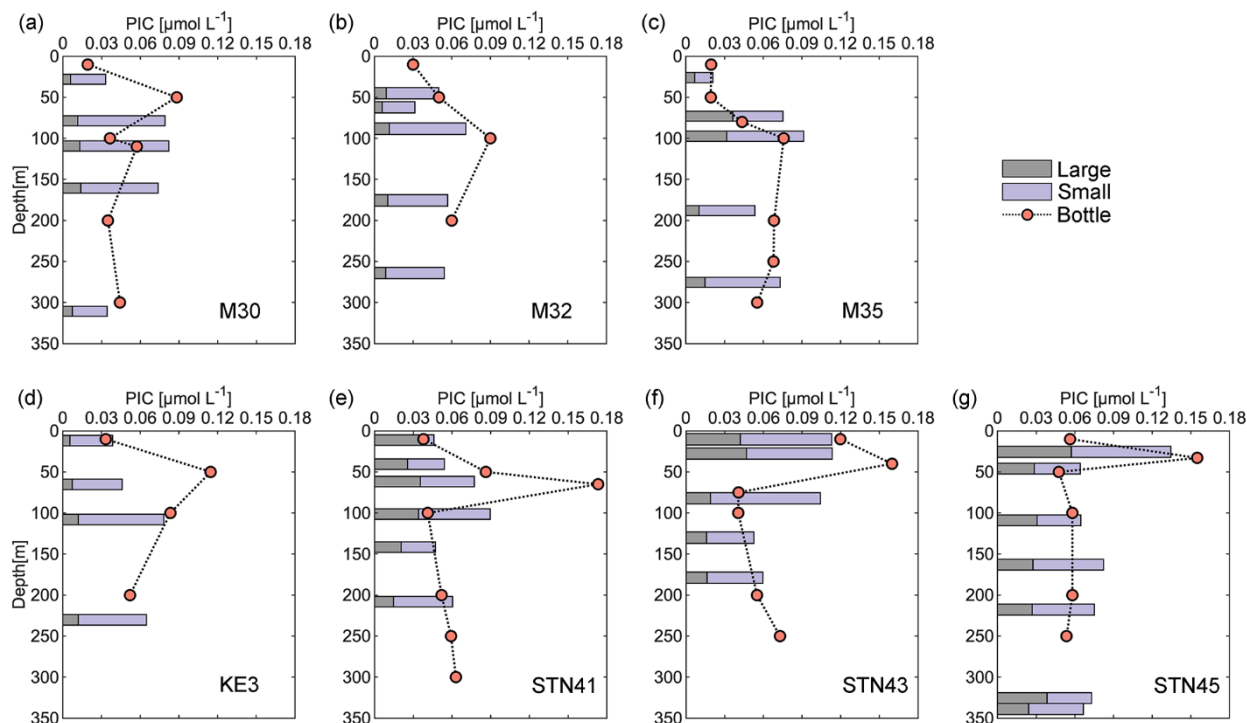
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185 **3.2 Vertical distribution of PIC and coccolithophore concentrations**

186 PIC concentrations along the 155°E transect ranged from 0.02 to 0.17 $\mu\text{mol L}^{-1}$, with an average of $0.06 \pm 0.04 \mu\text{mol L}^{-1}$ in
187 the upper 300 m of the water column (Fig. 2f). Generally, PIC concentrations were lower at the surface and increased with
188 increasing depth to attain a maximum in the DCM layer, and decreased with depth thereafter. In the DCM layer, PIC
189 concentrations ranged from 0.06 $\mu\text{mol L}^{-1}$ at 110 m of station M30 in the subtropical gyre to 0.16 $\mu\text{mol L}^{-1}$ at 33 m of station
190 STN45 in the Kuroshio-Oyashio transition region. The vertical distribution pattern of bottle-derived PIC and coccosphere cell
191 concentrations overall followed that of Chl *a*, showing a northward shoaling of the subsurface maximum.

192 Concentrations of coccosphere cells ranged from ca. 970 to 75,000 cells L^{-1} (Fig. 2g). Along the transect, a subsurface
193 maximum was evidenced around the DCM layer with an average of 42,000 cells L^{-1} , followed by a steep decrease below 100
194 m. The highest coccosphere cell concentration was observed at 65 m of station STN41, corresponding to the highest PIC
195 concentration. The average coccosphere cell concentration was notably lower in the NPSG region (9,800 cells L^{-1}) than in the
196 transition region (18,000 cells L^{-1}). The detached coccolith concentration averaged 340,000 coccoliths L^{-1} , with a range of
197 11,000 to 800,000 coccoliths L^{-1} (Fig. 2h). The highest concentration was observed around 10–40 m of station STN43. High
198 coccolith concentrations were also observed below 100 m at stations M32, M35 and STN41.

199 Size-fractionated PIC concentrations from in situ pumps varied from 0.01 to 0.09 $\mu\text{mol L}^{-1}$ in the SSF and from 0.01 to 0.06
200 $\mu\text{mol L}^{-1}$ in the LSF. $\text{PIC}_{\text{total}}$ concentrations averaged $0.07 \pm 0.02 \mu\text{mol L}^{-1}$, and were comparable to bottle-derived PIC
201 concentrations (Fig. 3). Roughly 70 % of the PIC was contributed by the SSF at each sampling station. Generally, LSF PIC
202 concentrations increased northward from stations M30–M35 to stations KE3–STN45 and accounted for 22 % and 36 % of
203 $\text{PIC}_{\text{total}}$ concentrations in the NPSG region and the Kuroshio-Oyashio transition region, respectively. The maximum
204 concentration of LSF PIC (0.06 $\mu\text{mol L}^{-1}$) was observed at 26 m of station STN45 (Fig. 3g).



205

206 **Fig. 3.** Vertical depth distributions of particulate inorganic carbon (PIC) concentrations derived from sampling using both
207 Niskin bottles and in situ pumps (small size fraction of 1–51 μm and large size fraction of $> 51 \mu\text{m}$) in the upper 350 m of the
208 water column at sampling stations in the study area.

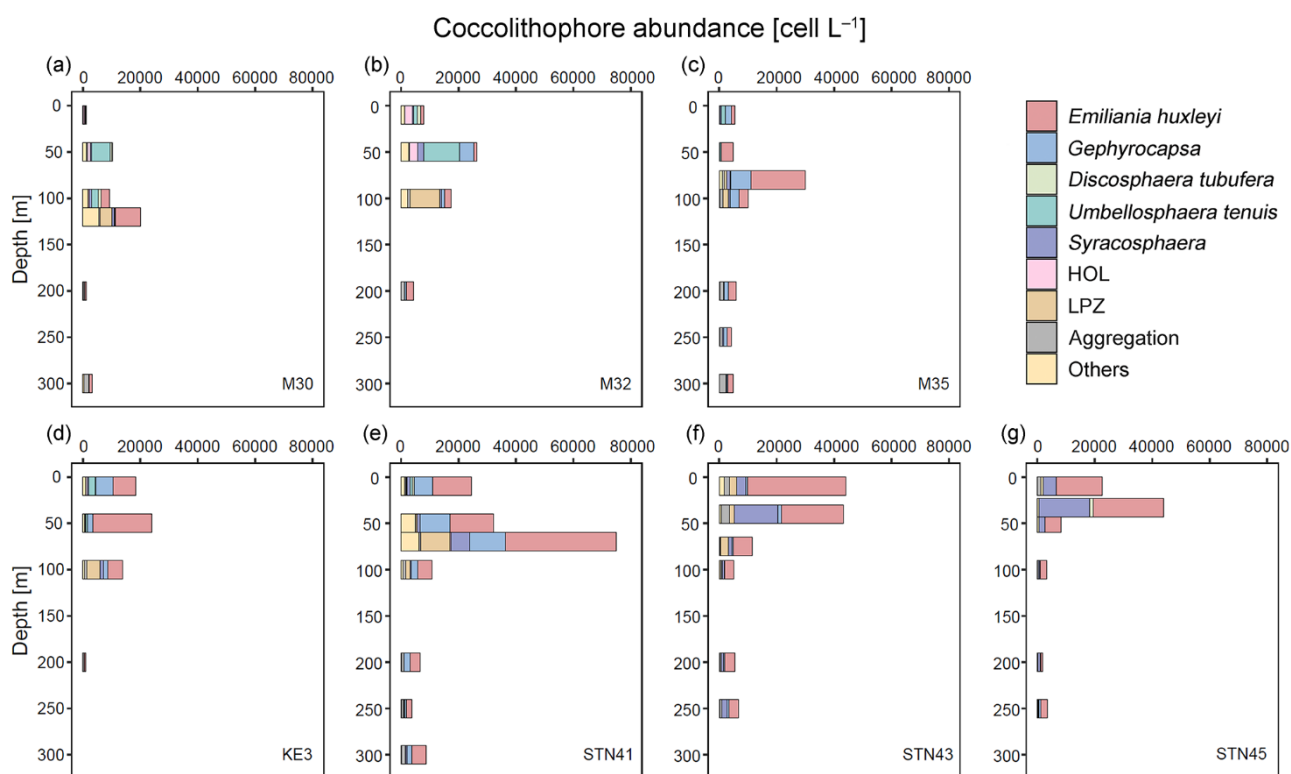
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210 3.3 Characteristics of the coccolithophore assemblage

211 Coccolithophore populations were predominantly represented by *Emiliania huxleyi*, *Gephyrocapsa ericsonii*, *Gephyrocapsa*
212 *oceanica*, *Umbellosphaera tenuis*, *Syracosphaera* spp., holo-coccolithophores (HOL), *Algirosphaera robusta*, and
213 *Florisphaera profunda* (comprising $> 1\%$ of total coccosphere abundance; Fig. 4). In surface water, coccolithophore cells
214 were dominated by *Dicosphaera tubifera*, *U. tenuis* and HOL at stations M30 and M32 (Fig. 4a and b) and by *G. ericsonii* at
215 stations M35, KE3 and STN41 (Fig. 4c, d and e), while high abundance of *E. huxleyi* and *Syracosphaera* spp. was clearly
216 observed at stations STN43 and STN45 (Fig. 4f and g). It is noteworthy that *E. huxleyi* contributed the largest fraction (50%)
217 to the total coccolithophore assemblage and was also found to be the dominant species in the DCM layer. *U. tenuis* was mainly



218 observed in subtropical gyre waters, with peak abundance at 50 m and lower abundance at the surface and in DCM waters
219 (Fig. 4a and b). Lower euphotic zone (LPZ) coccolithophore species (including *A. robusta* and *F. profunda*) were commonly
220 found in the subsurface population below 50 m, accounting for 7 % of the entire coccolithophore community. Overall,
221 coccolithophores were scarce in the NPSG region and dominated by *U. tenuis*, whereas their abundance notably increased in
222 the Kuroshio-Oyashio transition region where it was dominated by *E. huxleyi*, *Gephyrocapsa* and *Syracosphaera* spp..
223



224
225 **Fig. 4.** Abundance of different coccolithophore groups in the upper 300 m of the water column at sampling stations in the
226 study area. Lower euphotic zone (LPZ) species include *Florisphaera profunda* and *Algirosphaera robusta*; HOL indicates
227 holo-coccolithophores.

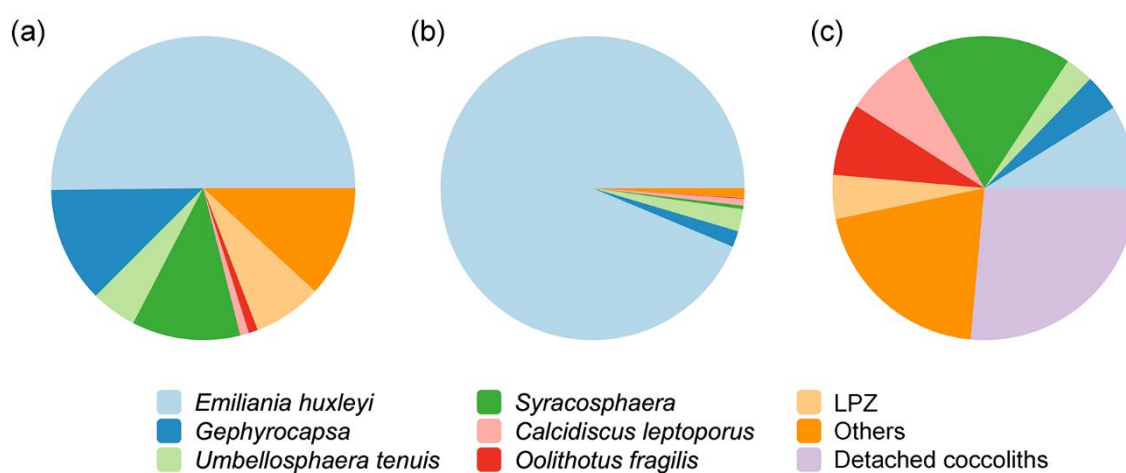
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229 The estimated coccolithophore calcite concentrations ranged from 0.00 to 0.23 $\mu\text{mol L}^{-1}$ averaging $0.05 \pm 0.04 \mu\text{mol L}^{-1}$



230 above 300 m along the 155°E transect. *E. huxleyi* accounted for 50 % of the total coccolithophore assemblage but represented
231 only 9 % of the coccolithophore calcite concentration (Fig. 5a and c). The less abundant (<1 %) species *Calcidiscus leptoporus*
232 and *Oolithotus fragilis* accounted for 7.5 % and 7.7 % of the coccolithophore calcite concentration, respectively.
233 *Syracosphaera* spp. was the largest contributor, accounting for 17.7 % of the coccolithophore calcite concentration (Fig. 5c).
234 Additionally, *E. huxleyi* detached coccoliths, comprising 94 % of the total detached coccoliths, made up ~16 % of the total
235 coccolithophore calcite concentration (Fig. 5b and c).

236



237

238 **Fig. 5.** Contribution of different coccolithophore groups to (a) coccosphere cell abundance, (b) detached coccolith abundance
239 and (c) coccolithophore calcite concentration in the upper 300 m of the water column. Lower euphotic zone (LPZ) species
240 include *Florisphaera profunda* and *Algirosphaera robusta*.

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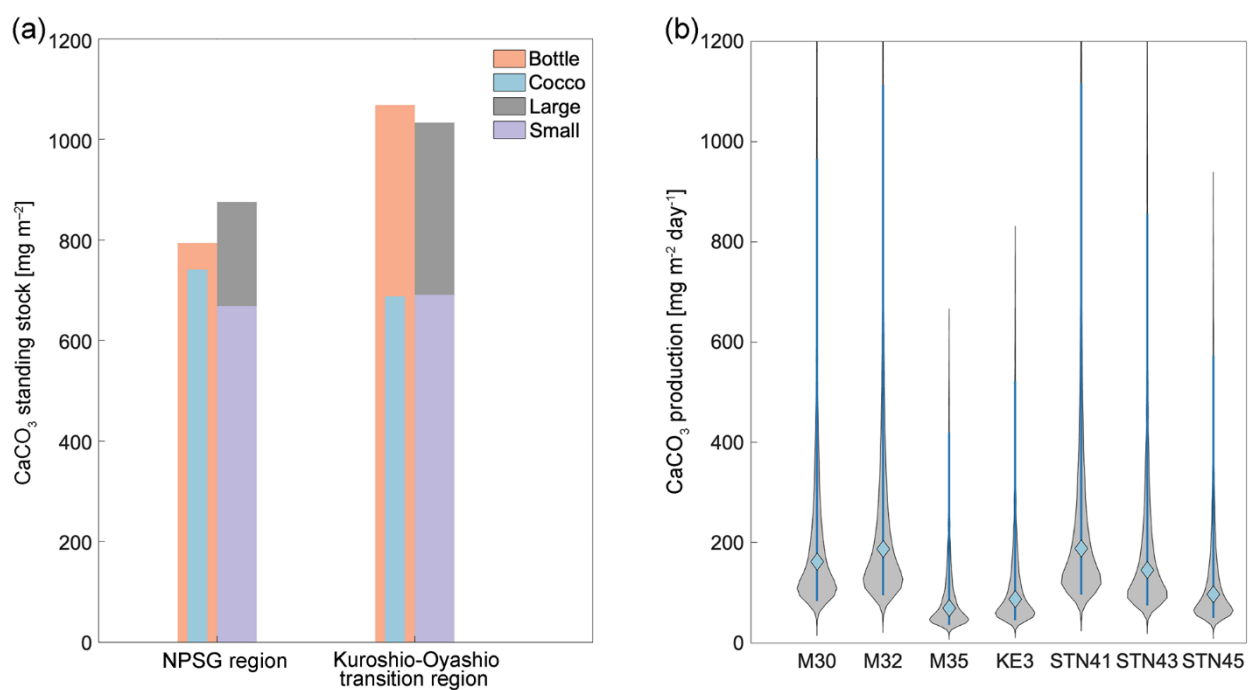
242 3.4 CaCO₃ standing stock and production

243 The standing stock of CaCO₃ was determined by considering the first shallow sampling depth to a consistent depth of 150 m
244 (Figs. 6a and S2a). CaCO₃ standing stock derived from Niskin bottle-sampling ranged from 660 to 1,200 mg m⁻², and was
245 lower in the oligotrophic NPSG region (790 mg m⁻²) than in the relatively nutrient-high Kuroshio-Oyashio transition region
246 (1,100 mg m⁻²). Based on the estimated coccolithophore calcite concentrations, CaCO₃ standing stocks ranged from 370 to



247 1,000 mg m⁻² and peaked at station STN41. Calcite from coccolithophores comprised on average 76 % of the CaCO₃ standing
248 stock from Niskin bottle samples, and the contribution was higher in the NPSG region (91 %) than in the Kuroshio-Oyashio
249 transition region (65 %) (Fig. 6a), demonstrating the vital role of coccolithophores in CaCO₃ production, particularly in
250 oligotrophic ocean waters.

251



252

253 **Fig. 6.** (a) Calcium carbonate (CaCO₃) standing stock in the upper 150 m water column estimated from Niskin bottle particulate
254 inorganic carbon (PIC), total calcite (Cocco) and size-fractionated (large and small fractions indicate > 51 and 1–51 μm,
255 respectively) PIC concentrations in the North Pacific Subtropical Gyre (NPSG) and Kuroshio-Oyashio transition regions; (b)
256 CaCO₃ production by coccolithophores in the upper 150 m water column at indicated sampling stations in June-July 2022.

257

258 Total CaCO₃ standing stock derived from in situ pump samples ranged from 820 to 1,300 mg m⁻², averaging 880 mg m⁻² in
259 the subtropical gyre and 1,030 mg m⁻² in the transition region, consistent with results from Niskin bottle samples. The CaCO₃
260 standing stock of the SSF ranged from 514 to 904 mg m⁻² and accounted for 71 % of the total standing stock in the entire



261 research domain (Fig. 6a).

262 Given that coccolithophores have a turnover time of 0.7–10 days (Krumhardt et al., 2017; Ziveri et al., 2023), CaCO₃
263 production rate in the upper 150 m of the water column ranged from 70 to 190 mg m⁻² d⁻¹ during the sampling period (Fig.
264 6b). Generally, the coccolithophore CaCO₃ production was comparable in the subtropical gyre and the Kuroshio-Oyashio
265 transition region, averaging 140 ± 62 and 130 ± 46 mg m⁻² d⁻¹, respectively. Coccolithophore CaCO₃ production was maximal
266 at station STN41 where it reached 190 mg m⁻² d⁻¹, corresponding to the maximum coccosphere cell concentration (Fig. 2e).
267 The lowest coccolithophore CaCO₃ production of 70 mg m⁻² d⁻¹ was observed at M35 in the NPSG region.

268

269 **4 Discussion**

270 **4.1 Coccolithophore responses to environmental factors**

271 Coccolithophores are an important component of phytoplankton biomass and fill a variety of ecological niches in global oceans.
272 They inhabit different marine environments from oligotrophic to eutrophic, warm to cold, euphotic to aphotic, and stratified
273 to mixed (Balch, 2018). It is of critical importance to evaluate the response of coccolithophore species to different
274 environmental conditions to better understand changes in coccolithophore species diversity, community composition, and their
275 role in the oceanic carbon cycle.

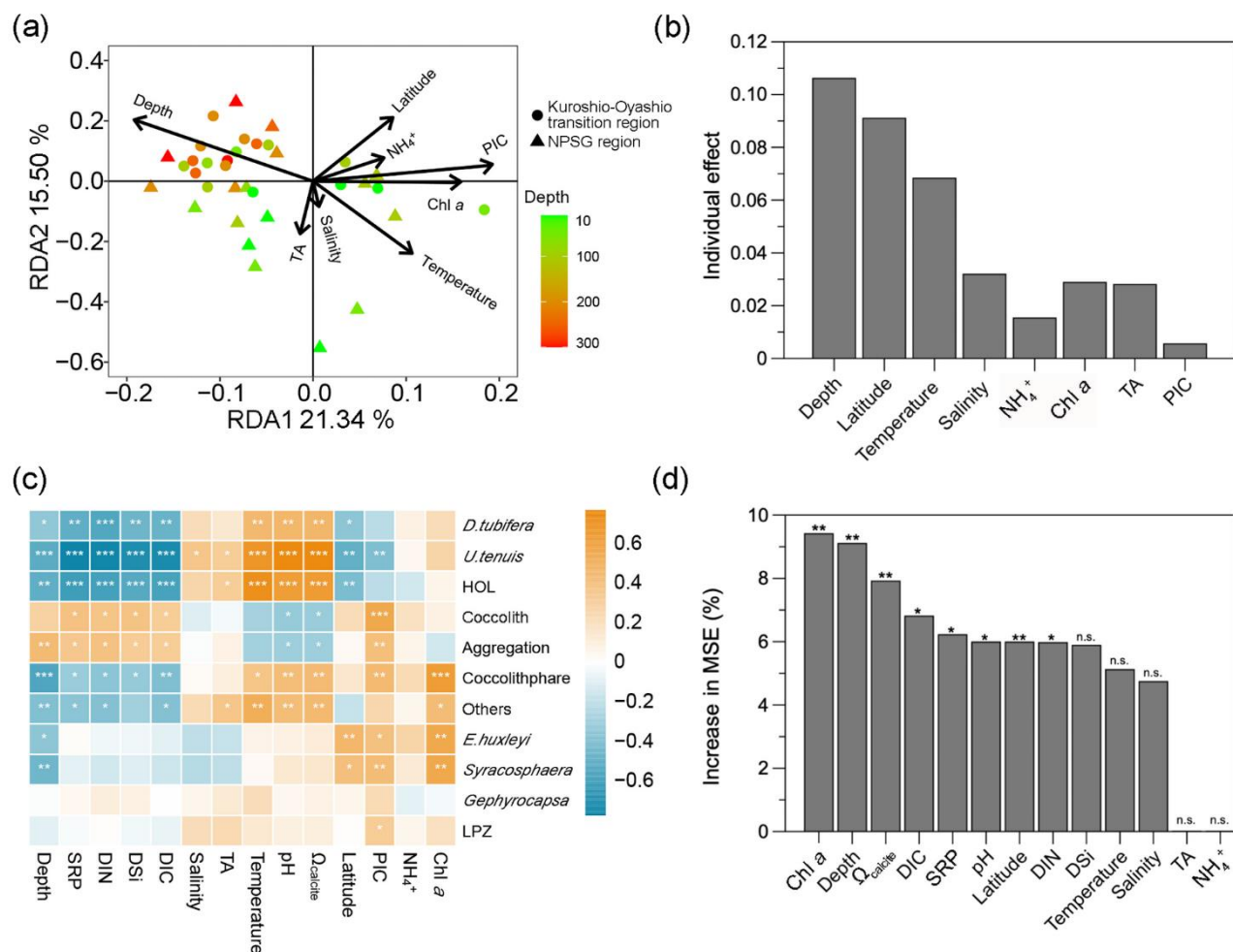
276 Although biogeographical zones of coccolithophores in the North and Central Pacific were identified a couple of decades
277 ago, few studies have investigated coccolithophore distributions in the North Pacific over the recent two decades (Okada and
278 Honjo, 1973; Hagino et al., 2005). In the western North Pacific Ocean, higher diversity and less abundant coccolithophore
279 assemblages were observed in the oligotrophic subtropical gyres, whereas the Kuroshio-Oyashio transition region tended to
280 exhibit a lower diversity corresponding to higher PIC and coccolithophore concentrations (Figs. 2 and S3). This finding is
281 consistent with results from the Atlantic Ocean, and a result of the different survival strategies of various coccolithophore
282 species (Balch et al., 2019). Coccolithophores are nutrient stress tolerant and have lower iron cell quotas, and are thus generally
283 abundant in the open ocean (Gregg and Casey, 2007; Brun et al., 2015). When nutrients and light are plentiful, the heavy
284 coccoliths of this group of phytoplankters pose a selective disadvantage over diatoms and chlorophytes (Gregg and Casey,
285 2007). The majority of coccolithophore species are K-selected, characterized by relatively slow-growth, large cell size and are



286 more competitive in low-nutrient and well-stratified regions (Brand, 1994), whereas only few r-selected species (such as the
287 fast-growing and small-sized *E. huxleyi*) mainly survive in relatively dynamic and nutrient-rich regions (Charalampopoulou,
288 2011; Brun et al., 2015; O'Brien et al., 2016). In the present study, the most abundant and widely distributed coccolithophore
289 species was *E. huxleyi* which showed increasing abundance northward along the study transect (Fig. 4). This is consistent with
290 prior observations demonstrating that *E. huxleyi* is the most abundant coccolithophore species in the subarctic, subantarctic
291 and bordering transitional regions (Saavedra-Pellitero et al., 2014).

292 According to RDA results, environmental variables accounted for 47.6 % of the total variation in coccolithophore
293 community composition (Fig. 7a). The first two RDA axes suggested that there were significant spatial differences in the
294 coccolithophore community across depths and regions (Fig. S4). Correspondingly, hierarchical partitioning analysis showed
295 that depth and latitude had a significant effect on coccolithophore community variation ($p < 0.05$). Other environmental factors,
296 such as temperature, salinity, Chl *a* and TA were also important influencing the coccolithophore community (Fig. 7b).

297 Based on Spearman's correlation analysis, coccolithophore abundance showed a significant positive relationship with
298 temperature, Ω_{calcite} and pH, and a significant negative relationship with depth, DIC and macro-nutrient concentrations,
299 especially for *D. tubifera*, *U. tenuis* and HOL that are more sensitive to environmental factors (Fig. 7c). The positive correlation
300 with temperature is consistent with field observations and model simulations pointing to a general trend of increasing
301 coccolithophore abundance in the context of global warming (Rivero-Calle et al., 2015; Rousseaux and Gregg, 2015). More
302 abundant species like *E. huxleyi* and *Syracosphaera* spp., however, only showed a highly positive correlation with depth,
303 latitude and Chl *a* concentration, suggesting that these species are more adaptable to varying environmental conditions
304 (Schlüter et al., 2014). Using random forest analysis, we also determined that the best predictors of coccolithophore abundance
305 were Chl *a* concentration, depth and Ω_{calcite} ($p < 0.01$; Fig. 7d).



306

307 **Fig. 7.** (a) Redundancy analysis (RDA) diagram illustrating the relationship between the coccolithophore community and
 308 environmental factors; (b) independent contribution of each environmental factor to coccolithophore community variation
 309 using hierarchical partitioning-based canonical analysis; (c) correlations between coccolithophore groups and environmental
 310 factors with color gradients denoting the significance of the Spearman's correlation coefficient r . Asterisks represent the
 311 statistical significance (** $p < 0.01$, * $p < 0.05$); (d) random forest mean predictor importance, i.e., the percentage
 312 of increase in the mean variance error (MSE) of environmental factors on coccolithophore abundance (** $p < 0.01$, * $p < 0.05$,
 313 n.s., non-significant). Chl *a*: chlorophyll *a*, DIC: dissolved inorganic carbon, TA: total alkalinity, Ω_{calcite} : saturation state with
 314 respect to calcite, PIC: particulate inorganic carbon, DIN: dissolved inorganic nitrogen (nitrate plus nitrite), NH₄⁺: ammonium,

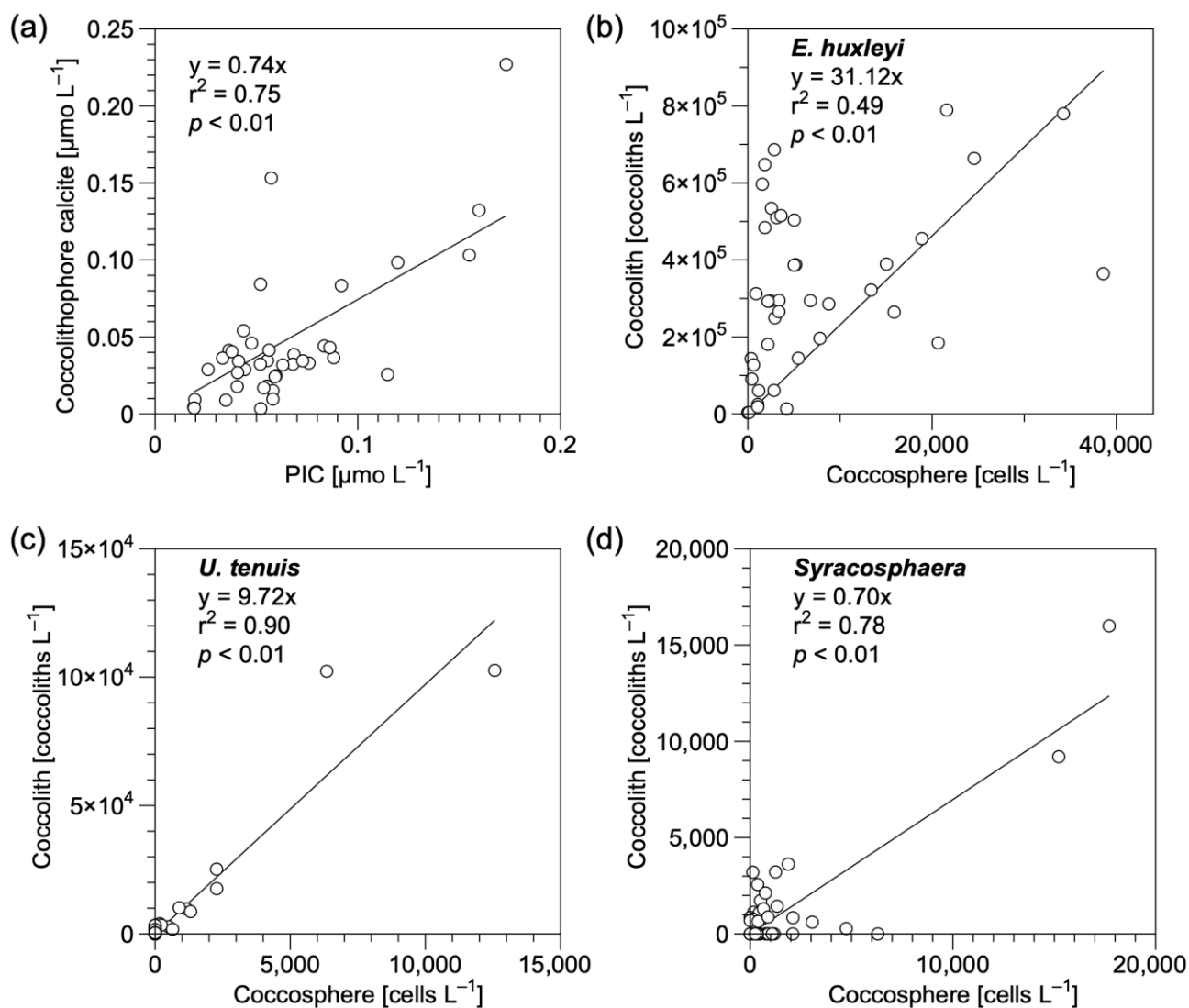


315 SRP: soluble reactive phosphate, DSi: dissolved silicate, HOL: holo-coccolithophores and LPZ: lower euphotic zone species
316 *Florisphaera profunda* and *Algirosphaera robusta*.

317

318 **4.2 Contribution of coccolithophore calcite to PIC**

319 In this study, bottle- and pump-derived PIC concentrations generally agreed with each other (Fig. 3), and both were comparable
320 to and of the same order of magnitude as the suspended PIC concentrations detected in the Atlantic, Indian and Pacific Oceans
321 (Beaufort et al., 2008; Barrett et al., 2014; Lam et al., 2015, 2018; Marañón et al., 2016). Coccolithophore calcite
322 concentrations showed a significant positive correlation with PIC concentrations ($r^2 = 0.75$, $p < 0.01$, $n = 40$; Fig. 8a),
323 highlighting the major contribution of coccospheres and detached coccoliths (68 %) to the total CaCO_3 in the upper 300 m of
324 the water column. This is consistent with findings from the eastern North Pacific Ocean where coccolithophores dominate
325 CaCO_3 production (Ziveri et al., 2023). It is noteworthy that detached coccolith concentrations of *E. huxleyi*, *U. tenuis* and
326 *Syracosphaera* spp. showed a significant positive relationship with their coccosphere cell concentrations (Fig. 8b–d),
327 indicating that those detached particles were likely to have originated from living cells.



328
329 **Fig. 8.** Relationship of (a) coccolithophore calcite vs particulate inorganic carbon (PIC) concentrations and (b–d) detached
330 coccolith vs coccosphere cell concentrations for (b) *Emiliana huxleyi*, (c) *Umbellosphaera tenuis* and (d) *Syracosphaera* spp.
331 in the upper 300 m water column at the study site. Equations describing the fitted straight lines are also shown.

332
333 Abundant coccolithophore groups, including *E. huxleyi*, *Syracosphaera* spp., LPZ species and aggregation, showed a
334 significant positive relationship with PIC concentration (Fig. 7c), but less abundant species like *C. leptopus* and *O. fragilis*
335 also made a large contribution to calcite concentrations. It has been reported that despite the relatively low abundance (< 2 %)



336 of the coccolithophore community, some larger species such as *C. leptoporus*, *Helicosphaera carteri* and *Coccolithus pelagicus*
337 could account for most of the coccolithophore CaCO₃ flux to the deep ocean (Rigual Hernández et al., 2020). Some rare
338 coccolithophore species with high coccolith and coccosphere cell concentrations have also been identified as important
339 contributors to both upper-ocean calcite production (Daniels et al., 2016) and deep-sea calcite fluxes (Ziveri et al., 2007). Thus,
340 although *E. huxleyi* is one of the most abundant species in the ocean, larger coccolithophore species can also play an important
341 role in CaCO₃ export.

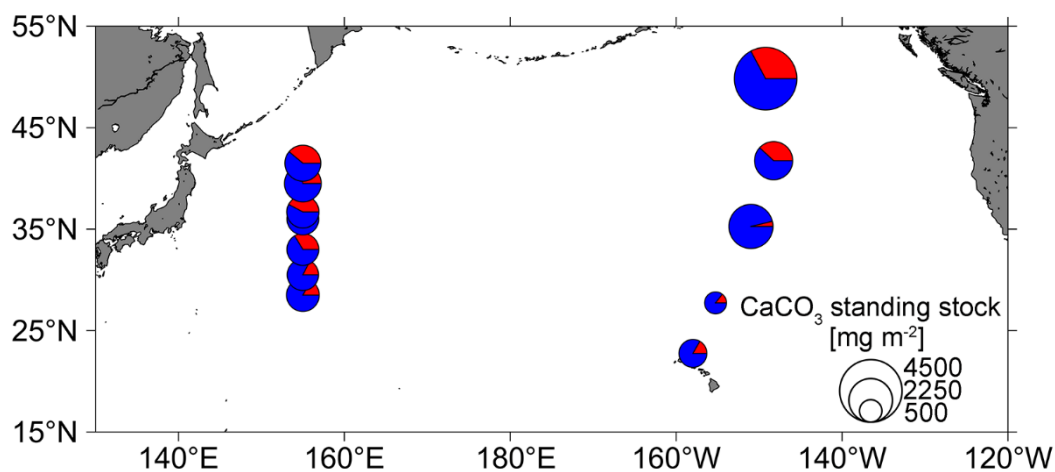
342 Higher CaCO₃ standing stock in the upper 150 m of the Kuroshio-Oyashio transition region (Fig. 6a) is consistent with
343 satellite observations suggesting that higher surface PIC concentrations occur at high latitudes (Balch et al., 2005; Berelson et
344 al., 2007). In the present study, however, the relative contribution of coccolithophores to the CaCO₃ standing stock was higher
345 in the NPSG region (~91 %) than in the Kuroshio-Oyashio transition region (~65 %) (Fig. 6a). To date, most studies estimated
346 CaCO₃ standing stock using satellite-derived data, which might be challenging to use in subtropical gyres where the DCM
347 depth usually lies below 100 m. Because coccolithophore CaCO₃ is largely produced in the lower layer of the euphotic zone
348 and is thus difficult to detect by satellites, coccolithophore contributions could be underestimated.

349 In oligotrophic ocean gyres, subsurface CaCO₃ production could still occur even if surface PIC is low (Balch et al., 2018).
350 Along our studied transect, maximum coccolithophore abundances increased about twofold from the subtropical gyre to the
351 transition region (Fig. 2e), while a much smaller difference was found in the integrated coccolithophore CaCO₃ between the
352 two regions (Fig. 6a). This suggests that subsurface coccolithophore CaCO₃ contributed substantially to the total upper water
353 column PIC flux in the NPSG. Coccolithophore groups were diverse in the subtropical gyre, where the environmental
354 conditions favor slow-growing, large and heavy species, which account for a large fraction of CaCO₃ production. In the
355 Southern Ocean, coccolithophores contribute to the highest annual CaCO₃ export in waters with low algal biomass
356 accumulations (Rigual Hernández et al., 2020). Given that low surface PIC regions (< 0.1 mmol m⁻³) occupy ~87 % of the
357 global ocean surface (Ziveri et al., 2023), our data suggest that coccolithophore CaCO₃ production in subsurface waters
358 constitutes an important part of global pelagic CaCO₃ production.

359 Size-fractionated PIC concentrations showed a smaller contribution of coccolithophores to the CaCO₃ standing stock in the
360 Kuroshio-Oyashio transition region (67 %) than in the NPSG region (76 %) (Fig. 6a), consistent with observations from the



361 eastern North Pacific Ocean (Fig. 9). The contribution of LSF PIC (e.g., zooplanktonic foraminifera, pteropods and heteropods)
362 to CaCO_3 standing stock is higher in the subpolar gyre (35 %) than in the subtropical gyre (16 %) of the eastern North Pacific
363 Ocean (Ziveri et al., 2023). Betzer et al., (1984) reported that foraminifera calcite is more abundant in northern regions (north
364 of 42°N) of the western North Pacific. At Ocean Station Papa in the northeast Pacific (50°N , 145°W), model results showed
365 that foraminifera calcite accounts for only 18–30 % of the total CaCO_3 production, whereas coccolithophores are the major
366 producer, contributing up to 59–77 % of the total CaCO_3 production (Fabry, 1989). In the Atlantic Ocean, coccolithophore
367 calcite fluxes and species richness are higher in subtropical than in temperate waters, which is ascribed to the reduced
368 competition with diatoms in the former (Broerse et al., 2000). Based on these findings, we suggest that differences in ecosystem
369 structure among sites modulate the relative contribution of various calcifiers to pelagic PIC production. The higher abundance
370 of non-calcareous phytoplankton (e.g., diatoms) in the transition zone could also reduce coccolithophore biomass via resource
371 competition (Quere et al., 2005; Sinha et al., 2010) and stimulate the growth of foraminifera (Schiebel et al., 2017), resulting
372 in the observed decreased contribution of small coccolithophores to total CaCO_3 production.



374 **Fig. 9.** Pie charts showing the composition of the total calcium carbonate (CaCO_3) standing stock in the upper 150 m of the
375 water column in the western (this study) and eastern North Pacific Ocean (data from the CDisK-IV cruise; Ziveri et al., 2023).
376 Red represents the standing stock of large size-fractionated ($> 51 \mu\text{m}$) CaCO_3 from this study, and planktonic foraminifera,
377



378 pteropods and heteropods from the CDisK-IV cruise. Blue represents the standing stock of small size-fractionated (1–51 μm)
379 CaCO_3 from this study and coccolithophores from the CDisK-IV cruise.

380

381 **4.3 CaCO_3 production compared with the eastern North Pacific**

382 Using a seasonal-correction method (Ziveri et al., 2023), the average coccolithophore CaCO_3 production above 150 m was
383 estimated to be $0.35 \text{ mol m}^{-2} \text{ yr}^{-1}$ for the entire research domain, which agrees well with the global estimate of 0.4 mol m^{-2}
384 yr^{-1} (Balch et al., 2007) and model result of $0.3 \text{ mol m}^{-2} \text{ yr}^{-1}$ in the North Pacific (Hopkins and Balch, 2018). In particular, this
385 production was $0.62 \text{ mol m}^{-2} \text{ yr}^{-1}$ in the subtropical gyre and $0.14 \text{ mol m}^{-2} \text{ yr}^{-1}$ in the Kuroshio-Oyashio transition region (Fig.
386 S2b). However, the latter is much lower than the recent estimate of $0.9\text{--}1.0 \text{ mol m}^{-2} \text{ yr}^{-1}$ by Ziveri et al. (2023) based on data
387 from the transition zone and subpolar gyre in the eastern North Pacific Ocean using the same seasonal-correction method.

388 Several factors may lead to the above discrepancy. First, CaCO_3 production rate on the present study was estimated based
389 only on coccolithophores, whereas estimates by Ziveri et al. (2023) also included the contribution from planktonic foraminifera,
390 pteropods and heteropods. Second, in the CDisK-IV cruise to the eastern North Pacific Ocean, coccolithophore calcite
391 concentrations were significantly higher than suspended seawater PIC concentrations collected by in situ pumps in the
392 transition zone and subpolar gyre (Fig. S5; Dong et al., 2019, 2022). Calculations based on these apparently inconsistent data
393 may result in an overestimation of actual CaCO_3 production. Third, high spatial and seasonal variations in PIC production
394 might occur between the two oceanic environments, in particular the high dynamics in the Kuroshio-Oyashio transition region
395 may have essentially altered the coccolithophore community and associated CaCO_3 production.

396 **4.4 Shallow-water CaCO_3 dissolution in the western North Pacific**

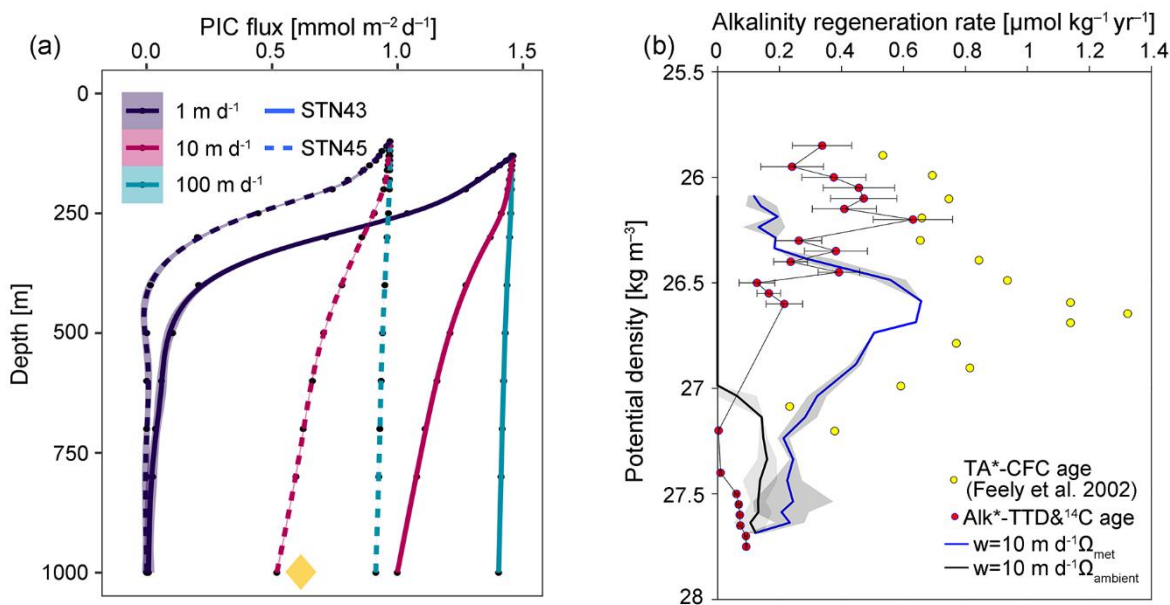
397 We used a box model to calculate the magnitude of metabolic CaCO_3 dissolution in shallow waters of the western North Pacific
398 Ocean (Eqs. 3 and 4; Dong et al., 2019, 2022). Focusing on CaCO_3 produced by coccolithophores and assuming that production
399 occurs in the euphotic zone, we applied the model to stations STN43 and STN45 because the calculated metabolic calcite
400 saturation horizon varied from 100 m to 150 m in the Kuroshio-Oyashio transition region. The results suggest that CaCO_3
401 might start to dissolve in setting marine particles after sinking out of the euphotic zone at the two stations, despite apparently
402 oversaturated ambient conditions (Fig. S6). Three different particle sinking rates (1, 10 and 100 m d^{-1}) were used in our model



403 calculations (Fig. 10a). To examine the possible influence by lateral transport around the Kuroshio Extension, our data in the
404 upper 1200 m were compared with those obtained during the CDisK-IV cruise (Fig. S5d–f). There was no significant difference
405 in PIC distribution patterns between western and eastern basins across 27°N–42°N in the North Pacific Ocean (one-way
406 ANOVA, $p > 0.05$), with the latter having relatively limited water parcel transport in the horizontal direction. We thus contend
407 that in our first order estimation, the effect of lateral transport on PIC distributions could be considered negligible (Dong et al.,
408 2019; Subhas et al., 2022).

409 Assuming that all coccolithophore production is exported out of the euphotic zone, our model results show that the PIC
410 sinking flux using the 10 m d⁻¹ sinking rate agrees well with data obtained from sediment traps deployed in July 1999 at a
411 40°N station (Honda et al., 2002) near our research domain (Fig. 10a). At this rate, the R_{TA} driven by Ω_{met} showed a vertical
412 distribution pattern similar to TA*-chlorofluorocarbon (CFC) age-based R_{TA} in the entire North Pacific Ocean (Feely et al.,
413 2002), with both displaying a maximum at a density of 26.58 kg m⁻³ corresponding to a water depth of 300 m (Fig. 10b). In
414 addition, our R_{TA} varied within a magnitude comparable to Alk*-transit time distribution (TTD) and ¹⁴C age-based R_{TA} values.
415 Instead, those driven by ambient Ω were essentially equal to zero with densities < 27.0 kg m⁻³ and became considerable with
416 further increasing densities below 600 m. Therefore, our model results indicate that shallow-water CaCO₃ dissolution indeed
417 occurs in the western North Pacific Ocean mainly as a result of metabolic acidification in the particulate microenvironment.

418



419

420 **Fig. 10.** (a) Particulate inorganic carbon (PIC) fluxes above 1,000 m at stations STN43 and STN45 generated by a box model
421 using three different sinking rates (1, 10 and 100 m d^{-1}); (b) vertical distribution of the alkalinity regeneration rate (R_{TA}) at
422 stations STN43 and STN45 generated by a 1-D model driven by metabolism-altered calcite saturation state (Ω_{met}) and ambient
423 calcite saturation state (Ω_{ambient}) at a sinking rate of 10 m d^{-1} . Lines indicate R_{TA} binned and averaged at 0.05 kg m^{-3} potential
424 density intervals. In (a), the yellow diamond marks the PIC flux at 1,000 m at station 40°N, 165°E in the western North Pacific
425 Ocean (Honda et al., 2002; Fig. 1a). In (b), TA*-chlorofluorocarbon (CFC) age represents the excess alkalinity-based R_{TA} from
426 (Feely et al., 2002) and Alk*-TTD& ^{14}C age represents the excess alkalinity-based R_{TA} using transit-time distribution (TTD)
427 ages and ^{14}C ages combined with Alk* (Key et al., 2004; Gebbie and Huybers, 2012; Carter et al., 2014; Jeansson et al., 2021;
428 Sulpis et al., 2021). Details of the estimation method are provided in the Supplementary Materials.

429

430 This finding is consistent with that obtained in the eastern North Pacific Ocean (Subhas et al., 2022), which suggests widely
431 occurring shallow-water dissolution throughout the entire North Pacific Ocean associated with organic carbon respiration. The
432 maximum R_{TA} value of 0.66 $\mu\text{mol kg}^{-1} \text{yr}^{-1}$ at a density of 26.58 kg m^{-3} in our study aligns exactly with that of 0.6 $\mu\text{mol kg}^{-1}$
433 yr^{-1} at a density of 26.54 kg m^{-3} in the eastern North Pacific Ocean (Subhas et al., 2022). It is noteworthy that the R_{TA}



434 discrepancy between this and previous studies might be ascribed to various dissolution mechanisms (Fig. 10b), since our model
435 only accounted for coccolithophore calcite excluding other calcifying plankton groups such as those producing aragonite and
436 high-Mg calcite. Jansen and Wolf-Gladrow (2001) also suggested that dissolution within zooplankton guts may account for
437 25 % of the shallow-water dissolution signal.

438

439 **5 Conclusions**

440 We have demonstrated that coccolithophore abundances and compositions had distinct geographic and vertical distribution
441 patterns, with *U. tenuis* dominated in the NPSG region while *E. huxleyi* and *Syracosphaera* spp. in the Kuroshio-Oyashio
442 transition region. The environmental variables that best described varying coccolithophore communities were depth and
443 latitude. Calcite derived from coccolithophores contributed on average ~76 % of the PIC standing stocks above 150 m, with a
444 relatively greater contribution in the subtropical gyre than in the transition region. Our results suggests that coccolithophore
445 CaCO₃ production was ~5-fold higher in the former than in the latter, which highlights the importance of coccolithophores in
446 oligotrophic environments.

447 This study also inferred that extensive CaCO₃ dissolution occurs above the ambient calcite saturation horizon, and is
448 primarily driven by the metabolic activity associated with organic carbon respiration. Given the important role of CaCO₃
449 production and dissolution in the marine alkalinity and carbon cycles, additional studies are required that target
450 coccolithophore production at different scales from seasonal to annual and from regional to global, as well as processes leading
451 to CaCO₃ dissolution in the apparently oversaturated upper ocean.



452 *Data availability.* Data for temperature, salinity, coccolithophore cell and coccolith abundances, coccolithophore calcite and
453 PIC concentrations can be downloaded from the Science Data Bank (<https://www.scidb.cn/en>). Satellite-based temperature,
454 Chl *a* and PIC concentration data were obtained from the MODIS-Aqua satellite (<https://oceancolor.gsfc.nasa.gov/l3/>).

455
456 *Supplement link.*

457
458 *Author Contributions.* YH, ZC, and MD conceived and designed the study. YH, ZS, DF, and JC contributed to data acquisition
459 and analysis. YH, ZS, ZC, and MD wrote the first draft of the manuscript. YH, ZS, ZC, JY, and MD discussed results and
460 edited the paper. All authors read and approved the final version of the manuscript.

461
462 *Competing interests.* The authors declare that they have no conflict of interests.

463
464 *Disclaimer.*

465
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