



1 **Uncertainties in the effects of organic aerosol coatings on**
2 **polycyclic aromatic hydrocarbon concentrations and**
3 **their estimated health effects**

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29



30 **Abstract.**

31 We utilized the CAM5 model to examine how different degradation approaches for particle-bound
32 Polycyclic Aromatic Hydrocarbons (PAHs) affect the spatial distribution of benzo(a)pyrene (BaP). Three
33 approaches were evaluated: NOA (no OA coatings), Shielded (where viscous OA coatings protect PAHs
34 from oxidation), and ROI-T (where OA coatings influence PAHs through reactive oxygen intermediates
35 related to temperature). Our findings indicate that the seasonal variation of BaP is influenced by
36 emissions, deposition, and degradation approaches. All simulations predict higher population-weighted
37 global average (PWGA) fresh BaP concentrations during December-January-February (DJF) compared
38 to June-July-August (JJA), primarily due to increased emissions from household activities, less efficient
39 wet removal, and unfavourable winter conditions. The Shielded and ROI-T approaches show that viscous
40 OA coatings significantly inhibit BaP oxidation, leading to PWGA fresh BaP concentrations two to six
41 times higher in DJF than in NOA. The Shielded approach predicts the highest PWGA fresh BaP
42 concentration of 1.3 ng m^{-3} in DJF, with 90% of BaP protected from oxidation. In contrast, the ROI-T
43 approach forecasts lower concentrations in mid-to-low latitudes. Model evaluations indicate the Shielded
44 method performs best, with a normalized mean bias within $\pm 20\%$. The incremental lifetime cancer risk
45 ranges from 0.6 to 2 deaths per 100,000 persons based on fresh BaP exposure. Overall, the human health
46 risks from fresh and oxidized PAHs are comparable, underscoring the importance of including both
47 forms in risk assessments and highlighting the critical role of accurate degradation approaches in PAH
48 modelling.

49



50 1 Introduction

51 Polycyclic aromatic hydrocarbons (PAHs), emitted from incomplete combustion of biofuels and fossil
52 fuels, are persistent organic pollutants composed of multiple aromatic rings. Some of them are
53 contaminants of global concern due to their well-known carcinogenic and mutagenic properties, which
54 increase the risk to human health [Boffetta et al., 1997; Perera, 1997; Chen and Liao, 2006; IARC, 2010;
55 Kim et al., 2013; Muir et al., 2019]. For instance, in 1976, the United States Environmental Protection
56 Agency (US EPA) listed 16 PAHs as priority pollutants [Keith 2015]. Among these, particle-bound
57 PAHs are more carcinogenic than gas-phase PAHs [Y Liu et al., 2017]. Therefore, benzo(a)pyrene, one
58 of the most carcinogenic PAHs and predominantly existing in the particle phase, is often used as an
59 indicator of cancer risk resulting from exposure to PAH mixtures [EPA 2004; EPCEU 2004; MEPPRC
60 2009; CPCB, 2020; IARC, 2021]. Considering that lifetime exposure to 0.1 ng m^{-3} of BaP would increase
61 the additional lung cancer risk by one in 100,000 exposed persons, the World Health Organization
62 (WHO) recommends limiting BaP concentrations to 0.1 ng m^{-3} [WHO, 2000; Bostrom et al., 2002].

63 High levels of BaP in ambient air have been measured globally over the past two decades, ranging from
64 0.1 to 2.5 ng m^{-3} in Europe and North America, with even higher concentrations observed in rural areas
65 of China and India, exceeding 10 ng m^{-3} [Lee et al., 2011; W Wang et al., 2011; Kim et al., 2012; Brown
66 et al., 2013; Hu et al., 2017; Radonić et al., 2017; Hu et al., 2018; Ma et al., 2018; J Han et al., 2019;
67 Lhotka et al., 2019; Munyeza et al., 2019; Ahad et al., 2020; Kumar et al., 2020]. However, compared to
68 measurements, previous regional or global models suffer from large uncertainties, with biases spanning
69 several orders of magnitude, largely due to an incomplete understanding of the complex gas-particle
70 partitioning [Friedman et al., 2014; Galarneau et al., 2014; Lammel et al., 2015; Shrivastava et al., 2017;
71 Mu et al., 2018; F Han et al., 2022]. For example, Iakovides et al. (2021) reported that using an octanol-
72 air partition coefficient absorption model, such as the Junge-Pankow model, the gas-particle fractions of
73 simulated PAHs are more suitable for remote or rural areas but not for urban areas. To differentiate
74 between aerosols in European urban or rural areas, Arp et al. (2008) developed polyparameter linear free
75 energy relationships (ppLFER) equations. Shahpoury et al. (2016) reported that the ppLFER model can
76 distinguish a variety of organics, including liquid water-soluble/organic soluble organics, and
77 solid/semisolid organic polymers, as well as the inorganic phases of aerosols. Therefore, by adopting the
78 ppLFER scheme, the gas-particle partitioning of simulated PAHs in anthropogenically impacted areas is
79 improved, and the simulated PAHs show good agreement with observations [Tomaz et al., 2016; Kelly
80 et al., 2021].

81 The lack of clarity regarding the chemical loss of PAHs is a significant factor contributing to large
82 deviations in model-simulated BaP concentrations compared to measured values. As a semi-volatile
83 compound, BaP in the gas-phase undergoes degradation through various pathways, primarily involving
84 reactions with OH and NO_3 radicals, along with photolytic processes driven by light. In a particle-bound
85 state, while BaP can also be degraded by OH and NO_3 , this occurs at a much slower rate compared to
86 degradation by ozone, which serves as the primary mechanism in this phase [Keyte et al., 2013].
87 Laboratory studies have shown that particle-bound BaP can undergo rapid oxidation within hours
88 through heterogeneous chemical degradation of BaP on the surface of black carbon (BC), organic carbon
89 (OC), and sulfate aerosols [Pöschl et al., 2001; Kwamena et al., 2004; Kahan et al., 2006; Zhou et al.,
90 2012]. Despite laboratory findings, field measurements have revealed that BaP persists in the atmosphere
91 for extended periods and can be transported over long distances, reaching even the Arctic [Halsall et al.,



92 1997; Masclet et al., 2000; Schauer et al., 2003; Lohmann and Lammel, 2004; Van Overmeiren et al.,
93 2024]. A recent laboratory study demonstrated that the presence of secondary organic aerosol (SOA)
94 coatings could shield BaP from ozone oxidation [Zelenyuk et al., 2012]. Based on this, Friedman et al.
95 (2014) used an exponential decay function, assuming that 80% of SOA-bound PAHs were still present
96 after 24 hours. However, the shielding effectiveness of PAHs depends on the phase state of SOA, which
97 should be temperature- and relative humidity-dependent [Koop et al., 2011; Zhou et al., 2013;
98 Berkemeier et al., 2016; Shiraiwa et al., 2017; Shrivastava et al., 2017; Mu et al., 2018]. Shrivastava et
99 al. (2017) developed a new PAH modeling approach in the global Community Atmosphere Model,
100 assuming that viscous SOA can completely inhibit particle-bound PAHs (i.e., BaP) oxidation reactions
101 under cool or dry conditions. Implementing this approach significantly improved the agreement between
102 simulated and measured BaP concentrations at hundreds of locations worldwide compared to models
103 that ignored the shielding effects of SOA coatings. Meanwhile, Mu et al. (2018) suggested that shutting
104 off particle-bound BaP degradation based on the simple thresholds of temperature and relative humidity
105 used in Shrivastava et al. (2017) cannot represent the complex multiphase reactions of BaP. They
106 proposed a new ROI-T approach, accounting for the effects of temperature and humidity on SOA phase
107 state and BaP degradation chemical reaction rate. The BaP concentrations simulated using the ROI-T
108 approach exhibited the best agreement with measurements at Xianghe (China) and Gosan (South Korea)
109 sites [Mu et al., 2018]. However, their simulations still showed a significant underestimation of BaP
110 concentrations for European and Arctic background sites.

111 Therefore, the oxidation chemistry of particle-bound BaP, highly dependent on the concentrations of
112 oxidants (primarily ozone) and the effectiveness of organic aerosol (OA) coatings, which are influenced
113 by temperature and relative humidity (RH). This dependence results in notable seasonal variations in
114 both fresh BaP concentrations and oxidized BaP. Considering that assessments of PAH-induced lung
115 cancer risks often rely on modeled BaP concentrations [Shen et al., 2014; Shrivastava et al., 2017; F Han
116 et al., 2020; Famiyeh et al., 2021; F Han et al., 2022; Li et al., 2022], uncertainties in modeled BaP
117 concentrations carry significant implications for estimates of PAH exposure and human health risks. This
118 study systematically investigates the uncertainty in simulated BaP concentrations due to varying
119 chemical mechanisms of BaP oxidation across different seasons and evaluates the appropriateness of
120 PAH modeling. This paper is organized as follows: Section 2 introduces the model, particle-bound BaP
121 degradation approaches, emissions, and observation data used in this study. Section 3 first presents the
122 simulated fresh and oxidized BaP concentrations in winter and summer, followed by a detailed
123 comparison between simulated BaP and measurements. Section 4 gives the conclusions and implications
124 for discussions.

125 2 Methods

126 2.1 Measurements

127 We collected observed fresh BaP concentrations at 66 background/remote sites and 208 non-
128 background sites worldwide (Table S1, S2). The observation data of fresh BaP were obtained from the
129 Integrated Atmospheric Deposition Network (IADN, available from <https://www.epa.gov/great-lakes-monitoring/great-lakes-integrated-atmospheric-deposition-network>), the European Monitoring and
130 Evaluation Programme (EMEP, available from <https://www.emep.int> [Tørseth et al., 2012]), the Global
131



132 Environmental Assessment Information System (GENASIS, available from <https://www.genasis.cz>), the
133 Arctic Monitoring and Assessment Programme (AMAP [Hung et al., 2010]), and previous studies [Shen
134 et al., 2014; Shrivastava et al., 2017]. For oxidized BaP, measurements were available from only two
135 locations: Grenoble—an urban site situated at 5.73°E, 45.16°N—in 2013, and SIRTA—a background
136 site located at 2.15°E, 48.71°N (<http://sirtaa.ipsl.fr/>)—in the years 2014-2015. However, due to the
137 measurement limitations, data on oxidized BaP (primarily nitro-BaP) were only available from two sites
138 in France. Therefore, this study only includes concentrations of oxidized BaP from Grenoble (an urban
139 site located at 5.73E, 45.16°N) in 2013 [Tomaz et al., 2016] and from the ACTRIS SIRTA atmospheric
140 supersite (Site Instrumental de Recherche par Télédétection Atmosphérique, which is representative of
141 the suburban background conditions in the Paris region, located at 2.15°E, 48.71°N; <http://sirtaa.ipsl.fr/>)
142 in 2014-2015 [Lanzafame et al., 2021].

143 2.2 Overview of the model

144 We employed the global Community Atmosphere Model version 5.2 (CAM5) to simulate the global
145 distribution of BaP concentrations. Tracer concentrations obtained from CAM5 simulations were
146 performed at a horizontal resolution of 1.9 ° latitude by 2.5 ° longitude, and a vertical resolution of 30
147 layers between the surface and 3.6 hPa. The Model for Ozone and Related Chemical Tracers (MOZART-
148 4) represented the gas-phase chemical mechanism [Emmons et al., 2010], while the properties and
149 processes of aerosol species were included in the Modal Aerosol Model (MAM3) [X Liu et al., 2012].
150 The model encompassed six aerosol species, including inorganic aerosols (e.g. mineral dust, black
151 carbon, sulfate, and sea salt) and organic aerosols (primary organic aerosol and secondary organic
152 aerosol). In addition, this study utilized an update of the volatility basis-set (VBS) approach developed
153 by Shrivastava et al. (2015). The VBS approach tracked SOA formation based on SOA precursor gas
154 sources, addressing both functionalization and fragmentation reactions during multi-generational aging
155 of SOA precursor gases, as well as oligomerization reactions of SOA. The ppLFER (polyparameter linear
156 free energy relationships) model was applied to the gas-particle partitioning of BaP, encompassing both
157 BaP absorption into organic aerosols and adsorption onto the surface of black carbon aerosol [Shahpoury
158 et al., 2016]. Following Shrivastava et al. (2017), we divided organic aerosols (OA) into the liquid water-
159 soluble/organic soluble phase and the solid/semi-solid organic polymer phase. More than 90% of
160 particle-bound BaP is absorbed within organic aerosols after applying the ppLFER model. The transport,
161 dry deposition, and wet removal of particle-bound BaP (including oxidized PAH) are treated similarly
162 to other aerosol species in CAM5 [X Liu et al., 2012].

163 2.3 BaP degradation

164 The model incorporates the gas-phase reaction of BaP with hydroxyl radicals (OH). Consistent with
165 previous studies, the second-order rate coefficient for the reaction of gaseous BaP with OH is set at 5×10^{11}
166 $\text{cm}^3 \text{molecules}^{-1} \text{s}^{-1}$ [Keyte et al., 2013; Shrivastava et al., 2017]. Heterogeneous reactions of
167 particulate-phase BaP with OH and ozone are also included in the model [Cazaunau et al., 2010; Zhou
168 et al., 2012; Keyte et al., 2013; Zhou et al., 2013]. The second-order rate coefficient for the reaction of
169 particle-bound BaP with OH is determined to be $2.9 \times 10^{-13} \text{cm}^3 \text{molecules}^{-1} \text{s}^{-1}$ [Esteve et al., 2006],
170 which is two orders of magnitude slower than the gas-phase reaction rate of BaP with OH. Conversely,
171 particle-bound BaP reacts rapidly with ozone within a few hours, representing the primary oxidation



172 pathway for BaP. Note that the photolysis of BaP is not included in this study, partly because its
173 photolysis rate constant is much lower compared to that of low molecular weight PAHs [Niu et al., 2007],
174 and the current model already underestimates BaP concentrations.

175 In this study, three approaches are implemented to estimate particle-bound BaP degradation, providing
176 insights into the uncertainty associated with this process.

177 (1) In the default NOA approach, the organic coating does not affect the BaP heterogeneous loss
178 process. The heterogeneous oxidation of particle-bound BaP follows the Langmuir-Hinselwood
179 mechanism, indicating that the first-order reaction rate (k) is variable dependent on ozone
180 concentrations, without diffusion limitations from thin liquid-like SOA coatings [Zhou et al., 2012;
181 Zhou et al., 2013].

182 (2) Following Shrivastava et al. (2017), the SOA Shielded approach is implemented, accounting for the
183 shielding of BaP by viscous SOA coatings. The kinetics of the heterogeneous oxidation of BaP with
184 ozone become much slower after absorption by organic aerosols, as thick OA coatings reduce the
185 kinetics of mass transfer of BaP from the interior of the particle to the particle surface. The
186 effectiveness of SOA shielding is related to its thickness and viscosity, influenced by temperature
187 and relative humidity [Zhou et al., 2012; Zhou et al., 2013]. In this approach, when SOA coatings
188 are less than 20 nm, we assume that SOA cannot effectively shield particle-bound BaP, and thus,
189 the heterogeneous oxidation kinetics remain the same as in the default NOA approach. Thick SOA
190 coatings (> 20 nm) can completely turn off the particle-bound BaP heterogeneous loss kinetics under
191 dry or cool conditions (relative humidity < 50% or temperature < 296 K). Different oxidation
192 kinetics with ozone are applied under humid and warm conditions with thick SOA coatings, where
193 the second-order rate coefficient for the reaction of particle-bound BaP with ozone is 14 and 6.2×10^{-15}
194 $\text{cm}^3 \text{molecules}^{-1} \text{s}^{-1}$ under moderate humidity ($50\% \leq \text{RH} < 70\%$) and high humidity conditions
195 ($\text{RH} \geq 70\%$), respectively [Zhou et al., 2013; Shrivastava et al., 2017].

196 (3) Following Mu et al. (2018), the ROI-T approach is implemented, accounting for the temperature
197 and humidity dependence of the phase state, diffusivity, and reactivity of particulate-phase BaP.
198 First-order reaction rate coefficients for BaP ozonolysis are sensitive to both temperature and RH
199 below room temperature (296 K), but are only temperature sensitive above room temperature [Mu
200 et al., 2018]. Under cool and dry conditions, the first-order reaction rate coefficients are four orders
201 of magnitude lower than those under warm conditions. Notably, the ROI-T approach yields a much
202 slower oxidation reaction of particle-bound BaP than the default NOA approach under cool and dry
203 conditions but a faster oxidation reaction rate under warm conditions.

204 2.4 Model Sensitivity Simulations

205 We conduct simulations using CAM5 to explore the uncertainty of seasonal variations in BaP
206 concentrations with different PAH oxidation approaches. Three effects of OA coatings on particle-bound
207 PAH oxidation, as detailed in section 2.3, are considered. Hence, sensitivity simulations are performed
208 as follows:

- 209 (1) NOA (OA coatings do not affect PAH oxidation);
- 210 (2) Shielded (viscous OA coatings turning off BaP reaction with ozone under cool and dry conditions)
211 [Shrivastava et al., 2017];



212 (3) ROI-T (the reaction rate between BaP and ozone varies with the effectiveness of OA coatings
213 based on temperature and RH) [Mu et al., 2018].
214 All simulations are conducted over two years, with the first year allocated for spin-up. Since most
215 observations occurred around 2004-2009, winds and temperature are nudged toward ERA-Interim data
216 from January 2007 to December 2008 in this study.

217 2.5 Emissions

218 This study utilizes the Global Emission Modeling System (GEMS) $0.1^\circ \times 0.1^\circ$ global BaP emission
219 inventory with temporal and spatial variations, which is available from gems.sustech.edu.cn. The
220 inventory includes data from all major fuel consumption sources and industrial processes [Shen et al.,
221 2013]. The spatiotemporal changes in global BaP emissions are detailed in our previous work
222 [Shrivastava et al., 2017; Lou et al., 2023]. Anthropogenic emissions, including BC, OC, and precursor
223 gases for both secondary aerosols and ozone, are sourced from the HTAP_v2.2 2008 emission inventory
224 [Janssens-Maenhout et al., 2015]. Additionally, emissions from agricultural waste burning and open
225 biomass burning emissions are obtained from the Emissions Database for Global Atmospheric Research
226 (EDGAR v4.3) and Global Fire Emissions Database (GFED3.0), respectively [van der Werf et al., 2010;
227 Crippa et al., 2018]. To maintain consistency with the modeling timeframe and facilitate comparison
228 with observations, all emissions are set at 2008 levels.

229 2.6 Global model downscaling formulation

230 To enhance the comparison between simulated BaP concentrations and measurements, particularly in
231 anthropogenically influenced sites such as those near cities, we implemented downscaling. This involved
232 reducing the model-calculated near-surface BaP concentrations from the coarse grid resolution of
233 $2.5^\circ \times 1.9^\circ$ to a higher resolution ($0.1^\circ \times 0.1^\circ$). Following the methodology outlined by Shen et al. (2014),
234 we assigned a weighting factor (W_i) for each $0.1^\circ \times 0.1^\circ$ receiving grid. This factor was determined by
235 summing the contributions of emissions from all $0.1^\circ \times 0.1^\circ$ emission grids within a nine-grid
236 neighborhood (one $2.5^\circ \times 1.9^\circ$ grid covering the $0.1^\circ \times 0.1^\circ$ receiving grid and the other eight surrounding
237 it). The formulation for W_i is as follows:

$$238 \quad W_i = \sum_{j=1}^n \frac{2.03Q_j f_j e^{-r_a t_{ji}}}{u_j \sigma z_j x_{ji}} \quad (1)$$

239 Here, Q_j (ng/s) represents the emissions density of the j th emission grid. f_j (dimensionless) and u_j
240 (m/s) are wind frequency (0-1) and wind speed in directions 1 to 16 in the j th emission grid, respectively,
241 taken from the ERA-interim reanalysis wind field. The degradation rate r_a (/s) involves the gas-phase
242 reaction with OH and the particle-phase heterogeneous reaction with ozone in the receiving grid based
243 on simulation output. Additionally, t_{ji} (s) and x_{ji} (m) denote the distance and transport time from the
244 j th emission grid to the i th receiving grid, and σ (m) is the vertical standard deviation of the
245 concentrations. Finally, the calculated W_i is used as a proxy to disaggregate the model-calculated
246 concentration of each $2.5^\circ \times 1.9^\circ$ model grid to a $0.1^\circ \times 0.1^\circ$ grid. Previous studies have reported substantial
247 improvements in the distribution and magnitude of observed BaP concentrations through this
248 downscaling process in similar simulations [Shen et al., 2014; Shrivastava et al., 2017; Lou et al., 2023].
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250 2.7 Incremental Lifetime Cancer Risk

251 The Incremental Lifetime Cancer Risk (ILCR) induced by exposure to PAHs in ambient air is calculated
252 using the following formula [Shen et al., 2014]:

$$253 \quad ILCR = CSF \times LADD \times SUS = CSF \times \frac{C \times IR \times y}{BW \times LE} \times SUS \quad (2)$$

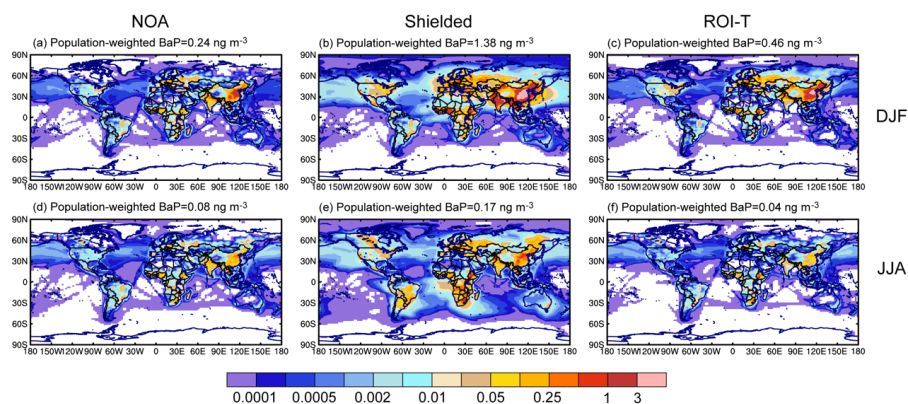
254 where CSF, LADD, and SUS represent the cancer slope factor, lifetime average daily dose, and a factor
255 describing individual susceptibility, respectively, depending on age, gender, ethnicity, and geographic
256 region. LADD is calculated from BaP exposure concentration (C , mg m^{-3}), which is downscaled from
257 model-predicted BaP concentrations in this study, inhalation rate (IR , m^3/day), exposure duration (y ,
258 year), body weight (BW), and average life expectancy of the global population (LE , 70 years). ILCR in
259 this study is a population-weighted average and represents the maximum likelihood estimate; the unit for
260 ILCR is one death per 100,000 persons.

261 3 Results

262 3.1 Simulation of seasonal variations in global fresh BaP

263 Given that lifetime exposure to 0.1 ng m^{-3} of BaP theoretically results in an additional lung cancer death
264 per 100,000 exposed persons, the WHO recommends a limit of 0.1 ng m^{-3} [Bostrom et al., 2002]. BaP
265 degradation approaches can significantly impact BaP concentrations, further influencing the assessment
266 of PAH exposure risks in various regions. Here, in this study, we investigated three different particle-
267 bound BaP degradation approaches related to the OA coating hypothesis to examine their effects on the
268 spatial distribution of BaP. Considering that the effectiveness of OA coatings is strongly dependent on
269 temperature and humidity variations, we analyzed the distribution of BaP concentrations under different
270 seasons.

271 In DJF (December-January-February), population-weighted global average (PWGA) BaP concentrations
272 with different particle-bound BaP degradation approaches are predicted to be $0.24\text{-}1.38 \text{ ng m}^{-3}$,
273 consistently exceeding the WHO recommendation. High levels of BaP concentrations are simulated to
274 appear in East Asia, South Asia, North Africa, and Europe, with the peak BaP exposure in eastern China
275 exceeding 1.0 ng m^{-3} (Fig. 1). In contrast, BaP concentrations are much lower in JJA (June-July-August),
276 with population-weighted global average values of $0.04\text{-}0.17 \text{ ng m}^{-3}$. These results indicate that the
277 simulated BaP exhibits strong seasonality, primarily influenced by changes in emissions, deposition, and
278 BaP degradation chemistry. In 2008, residential biomass use contributed more than 60% of total
279 atmospheric BaP emissions for households cooking, heating, and lighting [Shen et al., 2013]. Since the
280 demand for heating and lightning is higher in winter than in summer, more residential biomass burning
281 is required in winter, inevitably producing BaP. In addition, less precipitation in the Northern
282 Hemisphere in winter compared to summer (Fig. S1), linked to less efficient wet removal, contributes to
283 the seasonal variations of BaP. Thus, without the impacts of OA coatings on BaP degradation, the
284 seasonal variations of BaP concentrations in simulations using the NOA approach primarily represent
285 the changes in emissions and deposition.



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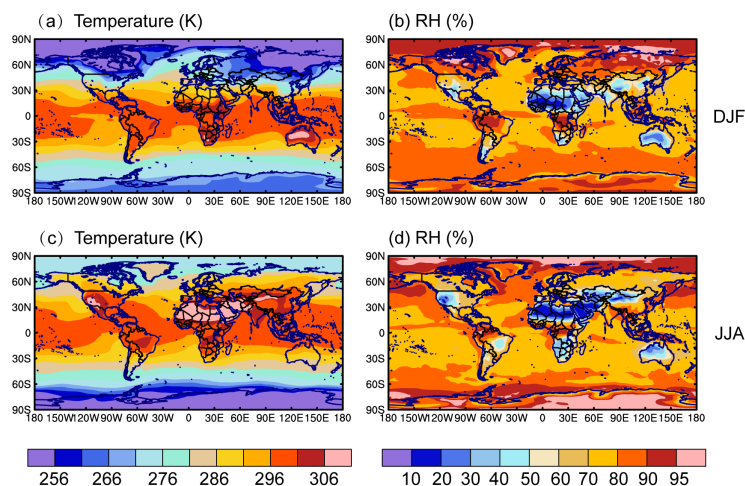
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Figure 1. The spatial distribution of simulated BaP concentrations in (a-c) DJF and (d-f) JJA. Fresh BaP concentrations with different heterogeneous reaction approaches for particle-bound BaP are shown in the left column (NOA), the middle column (Shielded), and the right column (ROI-T), respectively.

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Figure 2. The spatial distribution of surface-layer average temperature (left column, unit: K) and relative humidity (right column, unit: %) in DJF (December-January-February) and JJA (June-July-August), respectively.

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Compared with simulations using the NOA approach, those incorporating OA coatings can effectively impede the BaP loss process, leading to a significant increase in BaP concentrations during winter (Fig. 1). Upon absorption by organic aerosols, the presence of viscous OA coatings substantially hinders the mass transfer kinetics of BaP from the particle core to the surface. The denser the organic aerosol, the slower the diffusion of BaP, consequently slowing down the heterogeneous reactions of particle-bound BaP with ozone and OH on the aerosol surface. This effect is more pronounced in winter than in summer, attributed to cooler conditions that likely increase the viscosity of SOA [Shrivastava et al., 2017].

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303 For example, the PWGA BaP from the simulation using the Shielded approach is six times higher than
304 that in the simulation using the NOA approach in DJF. With the Shielded approach, the OA coating is
305 assumed to be sufficiently tacky to prevent BaP from undergoing heterogeneous reactions with ozone
306 completely under dry or cool conditions, thereby extending the lifetime of BaP. During Northern
307 Hemisphere winters, effective OA shielding occurs in areas characterized by cool temperatures (<296
308 K) or dry conditions (RH<50%), covering most of the regions with high BaP emission densities
309 [Shiraiwa et al., 2011; Saukko et al., 2012; Zhou et al., 2012; Bateman et al., 2015]. Furthermore, BaP
310 with OA coatings can be transported over long distances to remote areas, including the Arctic. Treating
311 OA coating effectiveness as the ROI-T approach, the BaP concentrations also increase, with PWGA BaP
312 estimated to be twice as high as in the simulation with the NOA approach during winter. Compared with
313 the Shielded simulation, BaP concentrations in the ROI-T simulation exhibit similar spatial patterns in
314 high latitudes such as Europe, northern China, and the Arctic, but lower concentrations in southern China,
315 South Asia, and North Africa. The ROI-T approach assumes that the diffusion coefficients of BaP and
316 ozone within OA coatings decrease with reducing temperature and relative humidity, thus reducing the
317 degradation rate of BaP. That is, under cold (<273 K) or dry (<50%) conditions, such as mid-to-high
318 latitudes in winter (Fig. 2a, b), the degradation rates of BaP in the ROI-T approach are two to four orders
319 of magnitude smaller than those without the OA coating effect. In contrast, the OA coating in southern
320 China, South Asia and Africa is not as effective as those in Europe, northern China, and the Arctic,
321 resulting in BaP concentrations similar to NOA simulation.

322 In JJA, BaP concentrations tend to concentrate near the source areas. While BaP concentrations in the
323 simulation using the Shielded approach are estimated to be higher than those in the NOA simulation, the
324 concentrations found in the simulation using the ROI-T approach are even lower. The ROI-T approach
325 assumes that the diffusion coefficients of BaP and ozone increase with temperature, leading to an
326 estimated faster degradation rate of BaP than in NOA and Shielded approach simulations at conditions
327 above room temperature. Our results are consistent with previous studies [Shrivastava et al., 2017; Mu
328 et al., 2018].

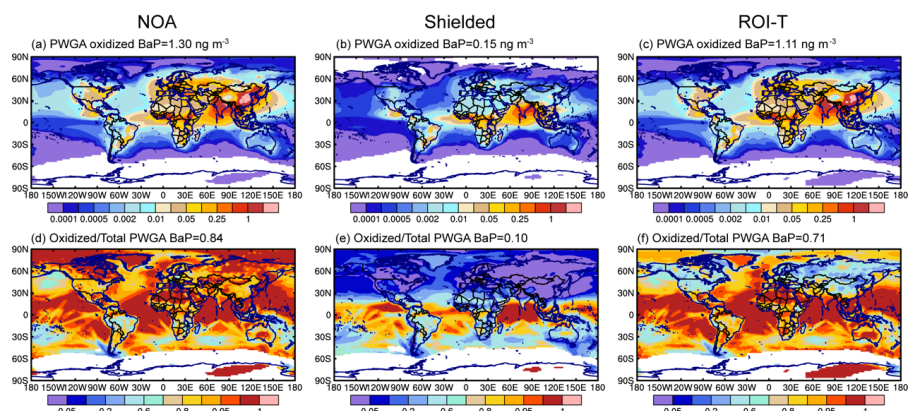
329 **3.2 Simulation of seasonal variations in global oxidized BaP**

330 Previous modeling studies assumed that fresh PAHs are completely degraded after oxidation [Schili and
331 Lammel, 2007; Matthias et al., 2009; Friedman et al., 2014]. However, laboratory experiments suggested
332 that several oxidized PAHs may remain particle-bound and even increase in molecular weight [Ringuet
333 et al., 2012; Zelenyuk et al., 2012; Jariyasopit et al., 2014]. Furthermore, not only fresh BaP but also
334 certain oxidized BaP species and derivatives exhibit toxicity [EHC 2003; Clergé et al., 2019; Hrdina
335 et al., 2022; Peng et al., 2023]. Therefore, it is essential to understand the impact of different BaP
336 degradation approaches on oxidized BaP.

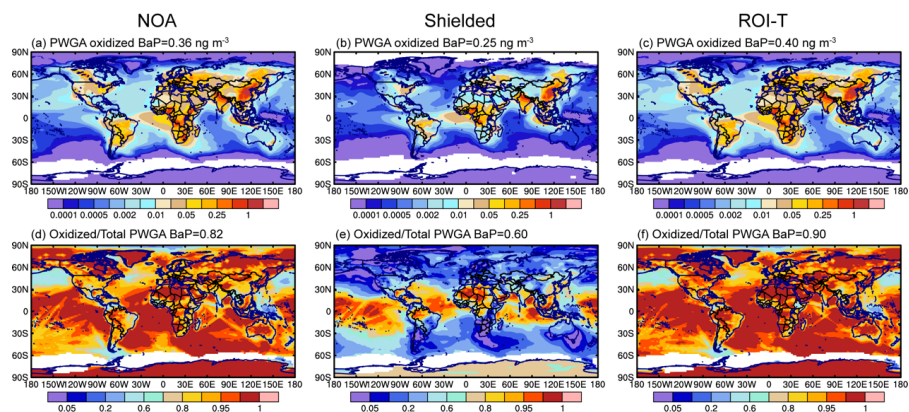
337 In this study, we tracked oxidized particle-bound BaP, which is formed through the heterogeneous
338 reactions of particulate-phase BaP with OH and ozone. Figure 3 shows the spatial distribution of
339 simulated oxidized BaP concentrations in DJF. In the absence of an OA coating, particle-bound BaP is
340 always available to react with O₃. Therefore, most particle-bound BaP is rapidly oxidized near source
341 areas, with a PWGA oxidized BaP concentration of 1.3 ng m⁻³, or 82% of the total (sum of fresh and
342 oxidized) BaP (Fig. 3a, d). If oxidized BaP is as toxic as fresh BaP, the oxidized BaP concentration in
343 NOA globally exceeds the WHO recommendation of 0.1 ng m⁻³ by a wide margin. In comparison, the



344 Shielded simulation predicts that high levels of oxidized BaP only appear in the tropics in winter (Fig.
 345 3b), because OA coatings are less effective at protecting BaP from ozone attack under high temperature
 346 and high RH conditions (Fig. 2a, b). Since the viscous OA coatings completely shut down the particle-
 347 bound BaP oxidation reaction under cool or dry conditions, most fresh BaP can stay in the atmosphere
 348 for several days, with only 10% of the total BaP being oxidized. Surprisingly, although the OA coating
 349 slowed the diffusion of particle-bound BaP from inside the interior of the OA to the particle surface in
 350 the ROI-T simulation, 71% of the total BaP was still oxidized on a global basis.
 351



352
 353 **Figure 3.** The spatial distribution of simulated (a-c) oxidized BaP concentrations and (d-f) the ratio of oxidized
 354 BaP to the total (fresh+oxidized) BaP in DJF. Simulations with the different heterogeneous reactions of
 355 particle-bound BaP approaches are shown in the left (NOA simulation), middle (Shielded), and right (ROI-T)
 356 columns, respectively.



358
 359 **Figure 4.** Same as Figure 3 but for JJA.

360
 361 Due to the less effectiveness of OA coatings under warm and moist conditions, all simulations with
 362 different BaP degradation approaches predict that oxidized BaP contributes to more than 90% of the total



363 BaP concentrations in JJA in the tropics (30°S-30°N). At mid-to-high latitudes, the oxidized BaP varies
364 greatly with the effect of OA coatings on the BaP oxidation reactions. More than 80% of total BaP is
365 oxidized in mid-to-high latitudes applying the NOA or ROI-T approaches, with peaks exceeding WHO
366 recommendations in most of East Asia, West Europe, and North America. Meanwhile, the Shielded
367 approach assumes that OA coatings largely limit the particle-bound BaP oxidation reaction, whereas the
368 oxidized BaP contributes no more than 40% to the total BaP. Our results indicate that current model
369 estimates of human exposure to fresh or oxidized PAHs are highly sensitive to assumptions about PAH
370 degradation processing, especially during North Hemisphere winter.

371 3.3. Model Evaluation

372 3.3.1 Fresh BaP

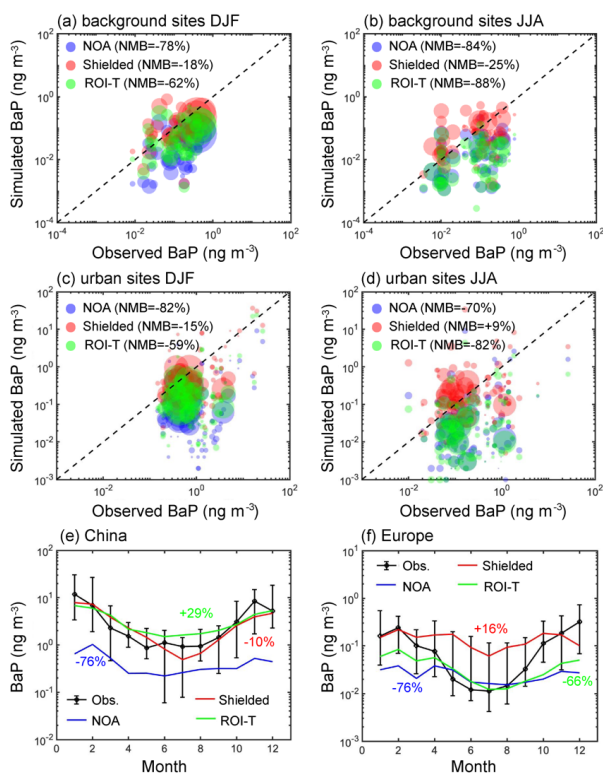
373 To assess simulated BaP concentrations, we select surface BaP measurements from 66
374 background/remote sites and 208 urban-affected sites worldwide (Table S1, S2), covering the period
375 1997 to 2014, with a focus on the years between 2004 and 2011. Median BaP observations at each site
376 are compared with simulated BaP in DJF and JJA, respectively. Given the global model's horizontal grid
377 spacing of approximately 200 km, we specifically compare simulated BaP concentrations with
378 measurements from background sites. To address the limitations of the coarse global model, we
379 downscale the simulated BaP grid to a higher resolution of 10 km based on factors such as wind speed,
380 wind direction and frequency, emission density, and gas/particle BaP degradation rates to account for
381 strong gradients and high BaP concentrations near urban areas. This downscaled approach aims to
382 account for strong gradients and high BaP concentrations near source areas. Our previous studies
383 reported that while the coarse-grid model significantly underestimates concentrations in urban-affected
384 regions, the downscaled BaP vastly improves the comparison between the model and observations
385 [Shrivastava et al., 2017; Lou et al., 2023].

386 Figure 5a compares measured and model-predicted concentrations at 66 background sites around the
387 world in DJF. The model-estimated BaP for the Shielded approach during the same time and locations
388 of the measurements agrees best with observations of global BaP concentrations, with a normalized mean
389 bias (NMB) of -18%. In contrast, without the effect of OA coating on the degradation of particle-bound
390 BaP, NOA predictions are 78% lower than observed BaP globally (Fig. 5b). Comparisons between
391 measured and downscaled simulated BaP at urban-affected sites show similar results, as the OA shielding
392 approach significantly improves the model's ability to predict fresh BaP concentrations.

393 It's worth noting that the effectiveness of OA coatings depends largely on temperature and humidity,
394 which are related to the meteorological characteristics of different regions. We, therefore, compare
395 measured and model-simulated BaP concentrations at different latitudes, namely relatively high latitudes
396 (measured locations north of 40°N) and low latitudes (measured locations between 40°S and 40°N),
397 respectively. Figures 6a and 6b demonstrate that the OA shielding particle-bound BaP approach increases
398 the simulated BaP concentrations in much better agreement with the measured values than without the
399 OA coating effect. This improvement is not sensitive to latitude. For the ROI-T treatment, although
400 predicted fresh BaP concentrations at locations above 40°N were two or three times higher than the
401 treatment without OA coating effects, the simulation still substantially underestimates the BaP
402 concentrations in these regions by 50% (Fig. 6c). Moreover, model-estimated BaP concentrations in



403 ROI-T perform even worse at low latitudes compared to high latitudes. On a global average, the ROI-T
 404 approach, accounting for the temperature and humidity dependence of the phase state, diffusivity, and
 405 reactivity of particulate-bound BaP, underestimates BaP by ~60% in DJF (Fig. 5a, c).
 406

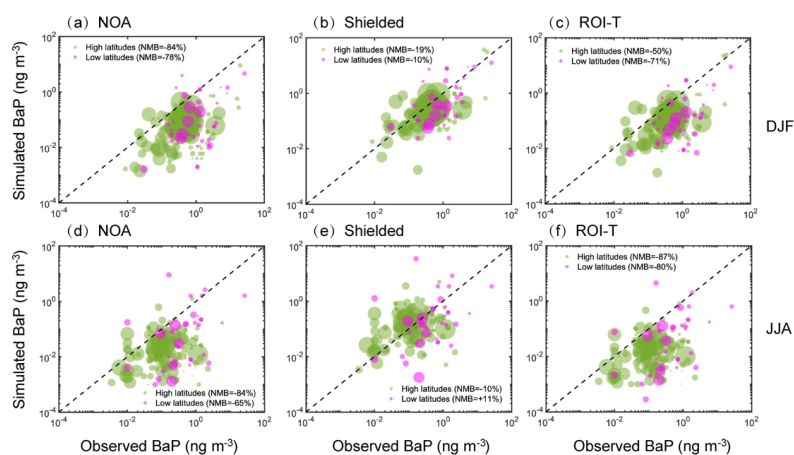


407
 408 **Figure 5. Comparison between simulated surface BaP concentrations from NOA, Shielded, and ROI-T**
 409 **simulations and ground-based measurements (a-b) for 66 background sites, (c-d) for 208 urban impacted**
 410 **sites. The circle area is proportional to the number of days sampled at each site. Annual variation of measured**
 411 **and simulated BaP concentrations at (e) 18 sites (6 background and 12 urban impacted sites) in China, and**
 412 **(f) 18 background sites in Europe. Black lines represent measured values (median and 15th and 85th**
 413 **percentiles of site monthly means), while blue (NOA), red (Shielded), and green (ROI-T) lines represent the**
 414 **median of the monthly model-simulated BaP means for these sites.**

415
 416 In JJA, both NOA and ROI-T simulations struggle to capture BaP concentrations, underestimating
 417 observations by more than 70% (Fig. 5b and 5d). However, similar to DJF, the model-predicted
 418 concentrations in simulation using the Shielded approach exhibit the best agreement with ground
 419 measurements in JJA, showing a normalized mean bias of -25% and +9% at the background and urban-
 420 affected sites, respectively. Interestingly, the ROI-T approach deviates more from the actual observed
 421 values, especially at measurement locations between 40°N and 40°S (Fig. 6d and 6f).



422 Figure 5e indicates that both the Shielded simulation (red line) and ROI-T simulation (green line) capture
423 the magnitude and seasonal variations of BaP concentrations compared with monthly observations at 18
424 sites in China (black lines). The simulated and observed BaP concentrations peak in winter but are lowest
425 in summer. As mentioned in section 3.1, the predicted monthly variations in BaP concentrations are due
426 to the seasonality of BaP emissions and BaP oxidation rates. For instance, residential emissions in China
427 are four times higher in winter than in summer, contributing 78% of BaP emissions in winter and 56%
428 in summer [Shen et al., 2013]. Furthermore, lower wintertime temperatures favor more viscous OA
429 coatings to reduce BaP diffusion and decrease oxidation rates, while more liquid-like OA coatings in
430 summer have a minor effect on BaP oxidation reactions. In contrast, although the models show a similar
431 seasonal cycle to observations, fresh BaP concentrations are largely underestimated throughout the year
432 in the absence of the OA coating effect (NOA).
433



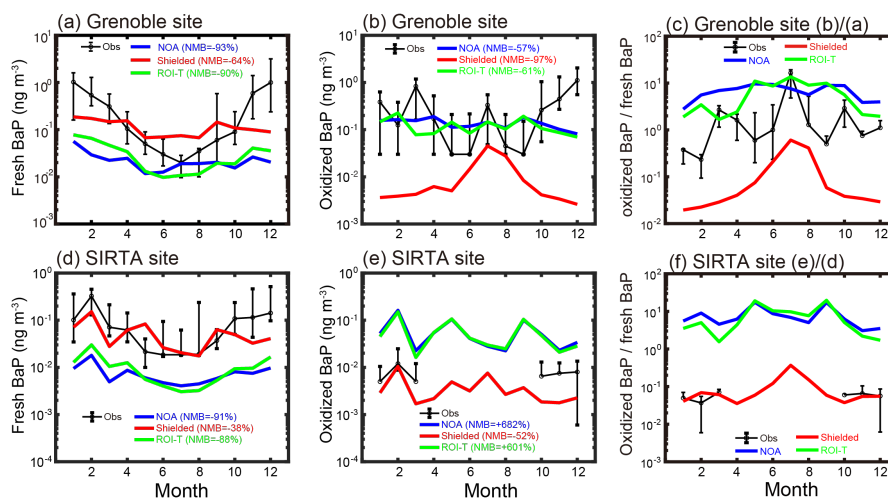
434
435 **Figure 6. Comparison between simulated surface BaP concentrations at relatively high latitude regions**
436 **(marked as olive, measured locations north of 40°N) and low latitude regions (marked as magenta, measured**
437 **locations between 40°N and 40°S) in (a-c) DJF and (d-f) JJA, respectively. The circle area is proportional to**
438 **the number of days sampled at each site. Both background and urban sites are included.**

439
440 In Europe, although the simulated BaP for the Shielded approach also exhibits the best agreement with
441 observations at 18 sites throughout the year, with a normalized mean bias of +16%, the simulated BaP
442 fails to capture the magnitude of the measured BaP concentrations during the warm season (Fig. 5f).
443 From April to October, model-predicted BaP concentrations in the Shielded simulation are overestimated
444 by 88%. In contrast, the simulated BaP concentrations for the ROI-T approach are consistent with the
445 monthly variation of the measured BaP concentrations, despite showing a 66% underestimated annual
446 mean, which is mainly due to the significant underestimation in cool season. Our results suggest that
447 while the Shielded simulation is likely closer to actual BaP magnitudes at mid- and low-latitudes, the
448 ROI-T approach may better represent seasonal variation at mid-and high-latitudes but overestimates the
449 coefficient of BaP multiphase degradation rates.



450 3.3.2 Oxidized BaP

451 Due to limited observations of oxidized BaP, specifically 1-, 3-, and 6-nitrobenzo(a)pyrene, we assess
452 monthly changes in BaP for three different particle-bound BaP degradation approaches performed at two
453 sites, Grenoble and SIRTA. In this study, we compare both simulated fresh BaP and oxidized BaP with
454 in situ measurements. Note that the simulated fresh/oxidized BaP concentrations used for comparison in
455 Grenoble are values after downscaling, as the site is centrally located and represents a location with
456 significant urban influence. In contrast, the SIRTA site is located 25 km southwest of central Paris and
457 is considered representative of regional background air quality. We, therefore, compare simulated
458 fresh/oxidized BaP concentrations with coarse horizontal resolution to measurements at the SIRTA site.
459



460

461 **Figure 7. Monthly comparison between simulated surface fresh and oxidized BaP from NOA, Shielded, and**
462 **ROI-T simulations and ground-based measurements at (a-c) Grenoble and (e-f) SIRTA sites. Black lines**
463 **represent measured values (median and 15th and 85th percentiles of each site), while blue (NOA), red**
464 **(Shielded), and green (ROI-T) lines represent the model-simulated median per month. The ratio of oxidized**
465 **BaP to fresh BaP are represent in (c) and (f). From April to September at SIRTA site, observed concentrations**
466 **of oxidized BaP were below LQ and therefore not presented on the graph.**

467

468 The measurement site at the sampling station of “Les Frenes” in Grenoble (5.73°E, 45.16°N, France)
469 represents the most densely populated urban area in Europe. Although the simulated concentrations
470 applying the Shielded approach best match the observed fresh BaP concentrations in Grenoble among
471 the three approaches, the model largely underestimates winter BaP concentrations but overestimates
472 summer concentrations (Fig. 7a). Therefore, the assumption that viscous organic aerosol coatings
473 completely shut off the reaction of fresh particle-bound BaP with ozone under cool and dry conditions
474 is somewhat distorted and fails to capture seasonal variation in fresh BaP in Grenoble. The relatively low
475 ratio of oxidized to fresh BaP using the Shielded approach in Fig. 7c indicates that the chemical scheme
476 overly protects fresh PAHs from oxidation. Consequently, it underestimates oxidized BaP by one order



477 of magnitude. In contrast, despite the overall underestimation, the ROI-T simulation captures the
478 seasonal variations in fresh BaP concentrations in Grenoble (Fig. 7a). However, the magnitude of the
479 simulated oxidized BaP concentration is very similar to the observed values (Fig. 7b). As we mentioned
480 above, the ratio of oxidized to fresh BaP in Fig. 7c reveals that the oxidation rate of BaP from fresh to
481 oxidized is too fast under ROI-T treatment, especially during cold season.

482 In addition, the underestimation of both fresh and oxidized BaP concentrations may be partly due to the
483 coarse horizontal resolution of simulated BaP, and inaccurate urban PAH emissions. We use a
484 downscaling formulation to convert the 200 km grid resolution to a ~10 km grid resolution, but the spatial
485 distribution of BaP obtained in this way is highly dependent on accurate emissions and meteorological
486 fields. Previous studies have reported that traditional biomass combustion for residential heating is the
487 main source of PM_{2.5} in France in winter and including in the Grenoble area [Favez et al., 2009; 2021;
488 Srivastava et al., 2018; Weber et al., 2019; Zhang et al., 2020a], thus inevitably emitting large amounts
489 of BaP. Considering the underestimation of both fresh and oxidized BaP concentrations at the Grenoble
490 site in winter, there is a large uncertainty in the emission and spatial distribution of PAHs in urban areas
491 (Fig. 7a, b).

492 For the SIRTa site, the simulated BaP from the Shielded simulation shows good performance compared
493 to the observed concentrations of fresh and oxidized BaP in winter (Fig. 7d, e). The performance of
494 Shielded approach in summer remains unclear due to the lack of observed concentrations of oxidized
495 BaP (Fig. 7e). However, the underestimation of fresh BaP concentrations and overestimation of oxidized
496 BaP concentrations in ROI-T and NOA suggest that the particle-bound PAH degradation rate is too fast
497 for these two approaches (Fig. 7c-d).

498 According to the ROI-T approach, once BaP is absorbed by organic aerosols, it can only be oxidized
499 when it comes to the surface through bulk diffusion or O₃ absorption from the gas sorption layer to bulk
500 layers. The changes in the BaP degradation rate coefficient are highly dependent on variations in
501 temperature and relative humidity [Mu et al., 2018]. Considering that RH in the French winter is
502 generally higher than 70% (Fig. 2b), the BaP degradation rate coefficient decreases by only one order of
503 magnitude for every 20 K drop in temperature from around 293 K. Therefore, the oxidation rate of ROI-
504 T for particle-bound PAHs is reduced by no more than 50% when the temperature is around 280 K in
505 the French winter (Fig. 2a). Our results suggest that at higher humidity, the ROI-T approach
506 underestimates the impact of OA coatings on PAH degradation effectiveness. Thus, the model's ability
507 to simulate fresh BaP is not significantly enhanced over the default NOA when the ROI-T approach is
508 elected, as relative humidity is significantly higher than 70% in mid- and high-latitude winters (Figs. 5f,
509 6c).

510 3.4. Lung-cancer risk of PAH mixture

511 As an indicator of cancer risk from PAH mixtures, previous studies calculated PAH-associated health
512 risks based on exposure to BaP concentrations using a method grounded in epidemiological data
513 [Bostrom et al., 2002; Zhang et al., 2009; Shen et al., 2014; Shrivastava et al., 2017; T Wang et al., 2017;
514 Kelly et al., 2021]. These studies primarily considered fresh PAHs when assessing PAH-associated
515 health risks. In this study, we follow the approach of previous studies to estimate ILCR [Shen et al.,
516 2014; Shrivastava et al., 2017; Lou et al., 2023].



517 Figure 8a illustrates that global and regional population-weighted ILCR varies significantly across
518 simulations when only considering exposure to fresh PAH. This variation is due to the substantial impact
519 of PAH degradation approaches on fresh BaP concentrations. On a global population-weighted basis, the
520 ILCR is predicted to be $\sim 0.6 \times 10^{-5}$ from the NOA and ROI-T simulations, falling within WHO-acceptable
521 risk levels for PAH exposure. However, based on the Shielded simulation, the global population-
522 weighted ILCR is predicted to be $\sim 2 \times 10^{-5}$, exceeding the acceptable limit of 1 death per 100,000 persons.
523 Moreover, without the heterogeneous oxidation of BaP, Shen et al. (2014) predicted an even higher
524 global population-weighted ILCR of 3×10^{-5} . These results underscore the high sensitivity of global ILCR
525 estimates to the choice of PAH degradation approaches.

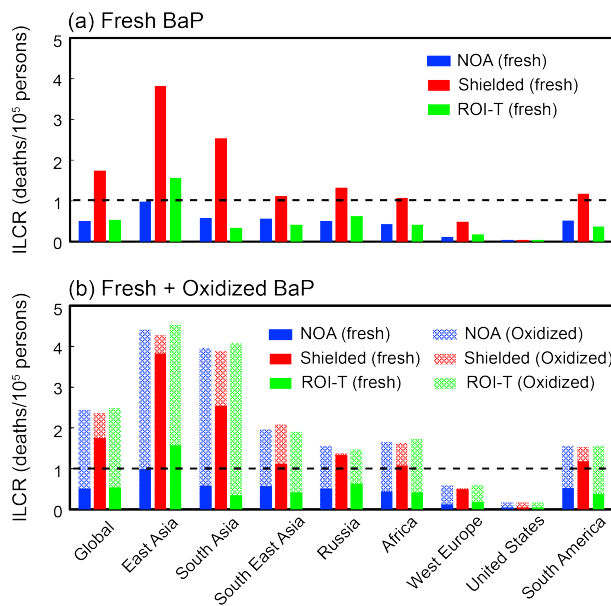
526 The variations in fresh BaP exposure and population-weighted ILCR are even more important for
527 regional estimation. Using the Shielded approach, the regional average population-weighted ILCR is
528 predicted to exceed 1×10^{-5} over East Asia, South Asia, Southeast Asia, Russia, Africa, and South
529 America. In contrast, ILCR for NOA and ROI-T simulations suggests a 3-4 times lower lung cancer risk
530 in these regions, expected to be below 1×10^{-5} except in East Asia (Fig. 8a). Due to the high emission
531 levels in 2008, the ROI-T simulation estimates a 50% higher ILCR than the NOA simulation, also
532 exceeding the WHO acceptable limit in the East Asia.

533 Furthermore, recent laboratory studies suggest that oxidized PAHs persist in the particle-phase and often
534 appear as higher molecular weight peaks in particle mass spectra [Ringuet et al., 2012; Zelenyuk et al.,
535 2012; Keyte et al., 2013; Jariyasopit et al., 2014]. Some PAH oxidation products may even be more toxic
536 than their parent compounds [EHC 2003; Clergé et al., 2019; Hrdina et al., 2022; Peng et al., 2023]. A
537 quantitative understanding of the toxicity of these products is lacking, as each parent PAH could be
538 oxidized into hundreds of products. In this study, we conduct a conservative first-order calculation of
539 lung cancer risk associated with oxidized PAHs, assuming that PAH oxidation products have the same
540 toxicity as their parents. On a global population-weighted basis, the ILCR is projected to 2.5 deaths per
541 100,000 persons when exposure to oxidized BaP is added to our previous calculations of fresh BaP
542 exposure for all three simulation approaches (Fig. 8b). While three-quarters of the global population-
543 weighted ILCR for the Shielded simulation is contributed to fresh PAH, oxidized PAHs contribute
544 approximately 40% the ILCR in warm and humid regions such as South Asia, Southeast Asia, and Africa.
545 In comparison, NOA and ROI-T simulations predict a dominant contribution of ILCR from PAH
546 oxidation products compared to fresh PAHs over most regions of the globe. For example, over East Asia
547 and South Asia, the NOA and ROI-T simulations predict that the regional population-weighted ILCR
548 will exceed 3 deaths per 100,000 persons resulting from the oxidized PAHs alone, compared to ~ 1 death
549 per 100,000 persons from oxidized PAHs in the Shielded simulation. The oxidized-fresh PAH ILCR split
550 is much greater in the NOA and ROI-T simulations compared to the Shielded simulation.

551 Despite differences in organic coating effectiveness and heterogeneous reactivity between the NOA,
552 Shielded, and ROI-T simulations, all schemes suggest that oxidized PAHs are crucial for lung cancer
553 risk and cannot be neglected. If the toxicity of oxidized PAHs is similar to fresh PAHs, the total ILCR
554 (fresh+oxidized) is comparable in the three approaches. However, oxidized PAHs could be much more
555 important in certain regions (such as Southeast Asia, South Asia, and Africa), depending on their
556 composition/toxicity and where the organic coatings are less effective in shielding them from
557 heterogeneous reactivity. Considering the high levels of oxidized PAHs in mid-to-low latitudes, the



558 measurements for oxidized PAHs, as well as human health exposure to oxidized PAHs, are necessary
559 for further studies [Kelly et al., 2021].
560



561
562 **Figure 8. Regional PAH-associated ILCR from NOA, Shielded, and ROI-T simulations. Note that solid bars**
563 **in (a, b) and shaded bars in (b) represent ILCR calculated from exposure to fresh and oxidized PAHs,**
564 **respectively.**

565 4. Conclusion and Discussion

566 This study uses the CAM5 model to investigate the impact of particle-bound PAH degradation
567 approaches on the spatial distribution of BaP, considering the presence or absence of OA coatings. The
568 three PAH degradation approaches are (1) OA coatings do not affect particle-bound BaP oxidation, (2)
569 the impact of OA coating on particle-bound BaP, assuming viscous OA coatings completely shield the
570 reaction of BaP with ozone under cool and dry conditions, and (3) the influence of OA coating on
571 particle-bound BaP degradation, assuming OA coatings slow down the oxidation reaction rate as a
572 function of temperature and humidity.

573 In general, the seasonal variation of BaP is highly dependent on changes in emissions, deposition, and
574 the chosen BaP degradation approach. In DJF, the demand for household activities such as cooking,
575 heating, and lighting increases, contributing to high BaP emissions. Additionally, less efficient wet
576 removal processes further enhance simulated BaP levels in DJF compared to JJA. The PWGA fresh BaP
577 concentration is predicted to be 0.24 ng m^{-3} without considering OA coatings, notably underestimating
578 measurements worldwide. In the absence of OA coatings, 82% of total BaP is rapidly oxidized near
579 source areas in DJF. The presence of viscous OA coatings slows down the oxidation process of fresh
580 BaP, resulting in a substantial increase in concentrations, ranging from 2 to 6 times in DJF. The Shielded



581 approach predicts the highest PWGA fresh BaP concentration of 1.30 ng m^{-3} in DJF. According to the
582 Shielded simulation, most fresh BaP can persist in the atmosphere for several days, with only 10% of the
583 total BaP being oxidized globally, mainly in the tropics. While the magnitudes and spatial distribution
584 of fresh BaP concentrations are similar at high latitudes (e.g., Europe and North America) between the
585 Shielded and the ROI-T approaches, the ROI-T approach, considering decreasing OA coating
586 effectiveness with reduced temperature and RH, leads to lower fresh BaP concentrations in Africa, South
587 Asia, and East Asia compared to the Shielded. Surprisingly, the ROI-T approach still predicts 71% of the
588 total BaP to be oxidized globally in DJF.

589 In JJA, BaP concentrations are concentrated near the source areas. While the Shielded approach still
590 predicts a much higher PWGA fresh BaP concentration than that without OA coating affect, the ROI-T
591 approach predicts the lowest PWGA fresh BaP concentration of 0.04 ng m^{-3} due to even faster first-order
592 reaction rate coefficients under warm and humidity concentrations. In comparison, all simulations predict
593 that more than 90% of the total BaP is oxidized in JJA in the tropics (30°S - 30°N). At mid-to-high
594 latitudes, the oxidized BaP varies greatly with the OA coating assumptions. While 80-95% of the total
595 BaP is oxidized in NOA or ROI-T simulations, the oxidized BaP contributes no more than 40% to the
596 total BaP in the Shielded simulation.

597 In our model evaluation, we compare simulated fresh BaP concentrations with observations at 66
598 background/remote sites and 208 non-background sites globally. On a global basis, both NOA and ROI-
599 T simulations perform poorly, underestimating fresh BaP concentrations by 60-80%. The Shielded
600 approach predicts the best agreement with observations, with normalized mean bias always within $\pm 20\%$.
601 The choice of BaP degradation approach has significantly different effects on improving BaP simulations
602 in various regions worldwide. For instance, both Shielded and ROI-T approaches improve the magnitude
603 and seasonal variations of BaP concentrations in China. However, while the Shielded simulation is more
604 aligned with the actual concentrations of BaP in Europe, our results indicate that the Shielded approach
605 is somewhat distorted and fails to capture seasonal variation in fresh BaP.

606 Additionally, due to measurement limitations, concentrations of oxidized BaP from Grenoble in 2013
607 and from SIRTA in 2014-2015 are also used for model evaluation. Our results indicate that while the
608 Shielded approach agrees best with the measured magnitude of BaP concentrations, this approach
609 underestimates the oxidation rates, especially during the warm season. In contrast, although the ROI-T
610 approach overstates the oxidation rates of BaP in multiphase environments, it offers a better depiction of
611 seasonal variations in fresh BaP concentrations. Therefore, to improve global PAH simulations, further
612 studies are needed to better understand the impact of OA coatings on PAH degradation effectiveness,
613 either for the ROI-T approach at higher humidity or for the Shielded approach at room temperature. A
614 broader range of observational data, including both ground-based and satellite-derived information (such
615 as water-soluble organic aerosols [Zhang et al., 2020b]), could also be utilized to validate the model.

616 We also employ a methodology based on epidemiological data to estimate the PAH-associated ILCR.
617 When calculated solely based on exposure to fresh BaP, the population-weighted ILCR varies largely
618 among the three PAH degradation approach simulations. On a global population-weighted basis, the
619 ILCR ranges from $\sim 0.6 \times 10^{-5}$ for the NOA and ROI-T simulations to $\sim 2 \times 10^{-5}$, exceeding the acceptable
620 limit of 1 death per 100,000 persons. Furthermore, when considering the toxicity of oxidized PAHs to
621 be similar to their parent PAHs, the total ILCR (fresh+oxidized) remain comparable across the three
622 approach simulations, amounting to 2.5 deaths per 100,000 persons. Oxidized PAHs may also be



623 important, depending on factors like their composition/toxicity and the effectiveness of OA coatings in
624 shielding fresh PAHs from heterogeneous reactivity. This study underscores the significance of
625 considering both fresh and oxidized PAHs in assessing cancer risk, particularly in regions where oxidized
626 PAHs may play a substantial role.
627

628 **Data availability**

629 The ERA-Interim reanalysis data is available from
630 <https://www.ecmwf.int/en/forecasts/datasets/reanalysis-datasets/era-interim>. The GEMS $0.1^\circ \times 0.1^\circ$
631 global BaP emission inventory is available from gems.sustech.edu.cn. The long-term observation data
632 are obtained from IADN (<https://www.epa.gov/great-lakes-monitoring/great-lakes-integrated-atmospheric-deposition-network>),
633 EMEP (<https://www.emep.int>), and GENASIS
634 (<https://www.genasis.cz>).
635

636 **Competing interests**

637 Manish Shrivastava is a member of the editorial board of ACP.

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646
647

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