## Response to Reviewers' Comments on Manuscript ID egusphere-2024-3210: "Observation and modelling of atmospheric OH and HO<sub>2</sub> radicals at a subtropical rural site and implications for secondary pollutants"

We sincerely appreciate the editor and the three referees for thoughtful comments and valuable suggestions which has helped us to improve our manuscript. In response to your comments, we have undertaken a comprehensive revision of the manuscript.

Please find our itemized responses below and revisions in the re-submitted files. We use *italicized* text for your questions, **blue** for our response, and **red** to indicate where changes have been made in the manuscript. The changes inside the manuscript will be highlighted in yellow.

## **Reviewer** 3

The manuscript reports about CIMS measurements of OH and HO2 in a subtropical rural site and the comparison of them with the results of MCM box model. Although in general simultaneous measurements of OH and HO2 provide very helpful information for understanding of radical budgets and the article reporting such measurements would potentially be of an interest, the present study cannot be published in its present form because it actually does not report HO2 measurements. The method used in this study for detection of peroxy radicals by their conversion to OH in reaction with added NO will result in about the same conversion efficiency for both HO2 and organic peroxy radicals RO2. To distinguish between the HO2 and RO2 radicals using the conversion scheme with NO several groups previously developed CIMS methods based on a modulation of chemical conditions in their instruments to measure either HO2 or RO2, predominantly (Hanke et al., 2002; Edwards et al., 2003; Hornbrook et al., 2011). In brief, HO2 mode requires efficient suppression of RO2 to HO2 conversion in reaction of alkoyy (RO) radicals with O2 in favor of the formation of alkyl nitrites:

RO2 + NO => RO + NO2

RO + O2 => R'O + HO2

RO + NO + M => RONO + M

Although the authors of the present manuscript make reference to the study of Edwards et al., 2003, they use NO concentration of 1.2 ppm leading to about 90% conversion of RO to HO2 and resulting in similar contribution of HO2 and RO2 to the detected signal, assuming their similar ambient concentrations. Referring in the manuscript supplement to the study of Fuchs, 2014 as an example of using the same NO concentration of 1.2 ppm for HO2 detection the authors do not take into account low pressure in the FAGE conversion stage, hence low O2, making RO+O2 reaction negligible and allowing HO2 measurements with low interference from RO2, although not for all of RO2 (Fuchs et al., 2011). Finally, the authors do mention once "interference" from RO2 by saying that "To access HO2 interference caused by the ambient RO2 conversion, the model underwent a three-days spins-up to simulate the ambient RO2 concentration". However, it doesn't seem to be a correct procedure to make correction using the model and after that compare the measurements corrected in this way with the same model.

The present OH and "HO2" measurements may still be of value and present the basis of an important publication. However, for this the measurements should be correctly interpreted and presented with detailed description of a calibration procedure of the peroxy radicals.

**Response:** Thank you for your insightful comment. In our study, both the experimental and modelling results did not show significant  $RO_2$  interference under our environmental conditions. Therefore, we consider our measurements to be representative of ambient HO<sub>2</sub> concentration.

The details were added to the revised manuscript on page 7:

"Interference from  $RO_2$  can affect  $HO_2$  measurements, potentially resulting in an overestimation of ambient  $HO_2$  levels (Edwards et al., 2003; Fuchs et al., 2014; Hanke et al., 2002). In our study, both the experimental and modelling results did not show significant interference under our environmental conditions (Text S4.3)."

The contents were added an Text S4.3 on page 11 on SI as below:

"To assess the possible HO<sub>2</sub> interference from RO<sub>2</sub>, we first simulated with a box model production of HO<sub>2</sub> from RO<sub>2</sub> in the inlet system with addition of 1.2 ppm NO to ambient air composition observed in previous field study in Hok Tsui (a coastal site in Hong Kong) by our team in 2020 (Zou et al., 2023) before this observation. The observationconstrained MCM model (described in the Text S6) was run for three days, and the RO<sub>2</sub> outputs were taken as used as the initial concentrations entering the inlet. Then another model run was conducted by setting j-values setting to zero and reaction time as the residence time (47 ms) to simulate the conversion of RO<sub>2</sub> by NO in the CIMS inlet. We compared the OH concentrations (from RO<sub>2</sub> conversion to HO<sub>2</sub> and then to OH) at the outlet with the total concentration of HO<sub>2</sub>+OH after spinning up. The result shows a difference of less than 2% suggesting negligible conversation of RO<sub>2</sub> to HO<sub>2</sub> in the inlet at 1.2 ppm NO injection. Similar model tests with real time conditions were also done for Conghua study after field study and show less than 2% interference.

To verity the model results, experiment tests were conducted in both laboratory and field settings (Hok Tsui in Hong Kong and Conghua) by comparing the HO<sub>2</sub> calibration factor obtained in synthetic air (with minimal interference of RO<sub>2</sub> due to very low VOCs concentrations in the synthetic air) and that conducted in indoor and outdoor air (with potential interference due to presence of VOCs). The results in Table S6 show difference of 1% - 3% between the HO<sub>2</sub> calibration factor in the synthetic air with that in the lab indoor air and that in the ambient air at Hok Tsui (with [O<sub>3</sub>] <70 ppb [NO<sub>x</sub>] < 10 ppb) as well as Cong Hua (with [O<sub>3</sub>] <60 ppb [NO<sub>x</sub>] < 10 ppb), confirming little interference of RO<sub>2</sub> to HO<sub>2</sub> measurements (See Table S6 below). These results might be due to the low concentrations of biogenic volatile organic compounds (BVOCs) in our two study sites ([C<sub>5</sub>H<sub>8</sub>] <0.2 ppb) as previous studies show large interference of BVOC to HO<sub>2</sub> measurements (Fuchs et al., 2014).

Year of experiements	Calibration Conditions	Calibratio OH	n Factor (cm <sup>3</sup> ) HO <sub>2</sub>	Notes
2021	Synthetic air in lab	7.912E-10	9.156E-10	On the afternoon of
	Indoor air in lab	8.146E-10	9.275E-10	Nov 20, 2021
	Synthetic air at Hok Tsui	8.212E-10	9.181E-10	On the morning of
	Outdoor air at Hok Tsui	8.252E-10	9.378E-10	Dec 23, 2021
2022	Synthetic air in lab	1.043E-09	1.080E-09	On the morning of
	Indoor air in lab	1.035E-09	1.119E-09	May 04, 2022
	Synthetic air at Conghua	1.033E-09	1.085E-09	On the morning of
	Outdoor air at Conghua	1.025E-09	1.092E-09	Nov 17, 2022

**Table S5** Calibration factors of OH and HO<sub>2</sub> obtained in different conditions to estimate the RO<sub>2</sub> interference.

Notes:

The difference between calibration factors obtained at 2022 and 2020 is due to the changes of CIMS's settings

 $\begin{array}{l} \mbox{Chemical Condition of outdoor air of Hok Tsui } [O_3] < 70 \mbox{ ppb, } [NO_x] < 10 \mbox{ ppb, } [C_5H_8] < 0.2 \mbox{ ppb} \\ \mbox{Chemical Condition of outdoor air of Cong Hua } [O_3] < 60 \mbox{ ppb, } [NO_x] < 10 \mbox{ ppb, } C_5H_8] < 0.2 \mbox{ ppb} \\ \mbox{Chemical Condition of outdoor air of Cong Hua } [O_3] < 60 \mbox{ ppb, } [NO_x] < 10 \mbox{ ppb, } C_5H_8] < 0.2 \mbox{ ppb} \\ \mbox{Chemical Condition of outdoor air of Cong Hua } [O_3] < 60 \mbox{ ppb, } [NO_x] < 10 \mbox{ ppb, } C_5H_8] < 0.2 \mbox{ ppb} \\ \mbox{Chemical Condition of outdoor air of Cong Hua } [O_3] < 60 \mbox{ ppb, } [NO_x] < 10 \mbox{ ppb, } [C_5H_8] < 0.2 \mbox{ ppb} \\ \mbox{Chemical Condition of outdoor air of Cong Hua } [O_3] < 60 \mbox{ ppb, } [NO_x] < 10 \mbox{ ppb, } [C_5H_8] < 0.2 \mbox{ ppb} \\ \mbox{Chemical Condition of outdoor air of Cong Hua } [O_3] < 60 \mbox{ ppb, } [NO_x] < 10 \mbox{ ppb, } [C_5H_8] < 0.2 \mbox{ ppb} \\ \mbox{Chemical Condition of outdoor air of Cong Hua } [O_3] < 60 \mbox{ ppb, } [NO_x] < 10 \mbox{ ppb, } [O_3] < 0.2 \mbox{ ppb} \\ \mbox{Chemical Condition of outdoor air of Cong Hua } [O_3] < 0.2 \mbox{ ppb, } [NO_x] < 0.2 \mbox{ ppb, } [$ 

As to the referee's comment on our correction of interference by comparing model and observations, the correction affected concentrations of less than 2%. We agree with the referee that this correction is take account for partial interference. However, the small correction is consistent with the test results shown above indicating minimal interference of  $RO_2$  to  $HO_2$  during the field study.

The following modification have been made on page 14 of SI as below:

## "Text S6 The model efforts to correct measurement interference.

Ambient NO can cause interference to OH measurement by concerting HO<sub>2</sub> to OH in the inlet system. To assess and correct this effect, model simulations were conducted, which also simulated conversion of RO<sub>2</sub> to HO<sub>2</sub> by NO (i.e., interference to HO2 measurements as discussed before). We first constrained all measured species (except OH and HO<sub>2</sub>) in the model, and a three-day spin-up was used to simulate the chemical conditions of the sampled air during measurement. Then the outputs were used as inputs for another simulation with the injection gases (SO<sub>2</sub> and/or NO) to simulate chemical reactions in the inlet with reaction time of 47 ms to match the reaction time in the CIMS. Photolysis frequencies were maintained at zero to reflect the dark environment of the inlet. The modeled OH concentrations without NO addition and OH concentrations with NO addition represent ambient NO interference to OH and HO<sub>2</sub>, respectively.

The calculated interferences for OH and HO<sub>2</sub> measurements were in the range  $1 \times 10^4$  cm<sup>-3</sup> to  $1 \times 10^5$  cm<sup>-3</sup> (mean:  $3 \times 10^4$ ) and  $8 \times 10^5$  cm<sup>-3</sup> to  $2 \times 10^6$  cm<sup>-3</sup> (mean:  $1.2 \times 10^6$ ), respectively. These lead to correction of measurement data of OH and HO<sub>2</sub> less than 2%."

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