No.: egusphere-2024-2869

Title: Measurement report: Crustal materials play an increasing role in elevating

particle pH: Insights from 12-year records in a typical inland city of China.

Reviewer #2:

General Comments:

Zhang et al. analyzed the particle pH collected in a Chinese inland city. This analysis is based on

4228 filter samples collected from four seasons in 2011 – 2022. As a measurement report, the study

fits into the scope of Atmospheric Chemistry and Physics. The authors claim that the evolving

particle pH was driven by the interplay of declines in SO₄²-, increases in TNH_x, and rises in Ca²⁺

concentrations over time. The manuscript can be considered for publication once the comments below

are addressed.

Thank you for your careful reading of our paper and valuable comments and suggestions. We believe

that we have adequately addressed your comments. To facilitate your review, we used green highlights

for your comments, yellow highlights for Reviewer #1, and red color indicating our own corrections

in the manuscript.

Major Comment:

1. lines 111-112 and 120-121: It is unclear whether Ti was measured or not.

Response: Sorry for the mistake. We did not measure the element Ti. The sentence has been corrected:

"Elements were analyzed using a wavelength dispersive X-ray fluorescence spectrometer (S8 TIGER,

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Bruker, Germany) to determine concentrations of Fe, Na, Mg, Al, Si, Cl, K, Ca, V, Ni, Cu, Zn, Cr, Mn, Co, Cd, Ga, As, Se, Sr, Sn, Sb, Ba, and Pb (Tremper et al., 2018)."

2. Lines 135 – 136: The linear regression proposed by Wei et al. (2023) is based on the dataset collected in the wintertime but not from 2013 to 2020. This needs clarification. Since the regression was based on wintertime data, it is questionable if the relationship is still valid for data in seasons other than winter.

Response: Thanks for your comment. We have added a validation of the simulated NH₃: "To validate the applicability of Equation 4 for annual NH₃ estimation and pH simulation in Zhengzhou, this study utilized both observed NH₃ data (from a Thermo Scientific URG-9000D ambient ion monitor, USA) and calculated NH₃ values derived from Equation 4 at the same monitoring site throughout 2022, inputting them into the thermodynamic model for pH simulation. As shown in Figure S2, pH values calculated from observed and simulated NH₃ exhibit good agreement (r = 0.97, P < 0.01). Furthermore, NH₃ concentrations modeled by ISORROPIA demonstrate a significant correlation (r = 0.95, P < 0.01) with that simulated NH₃ by Equation 4. These results collectively demonstrate the reliability of the NH₃ estimation method in this study."

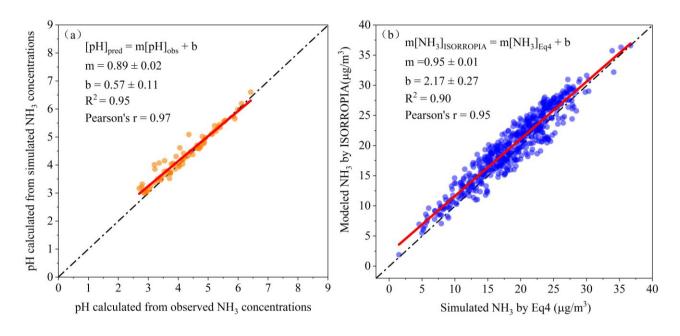


Figure S2. Verification of the NH₃ Method.

3. Section 2.2.3: Why was 1000 m chosen as the height in the HYSPLIT simulations? If 1000 m was already well above the boundary layer height, how do the simulated back trajectory simulations represent the ground-level measurements? In addition, how was the optimum number of clusters chosen by the authors? What clustering technique was used in the analysis? What data was used as the meteorological input?

Response: Thanks for your comments.

While the surface elevation of Zhengzhou is approximately 100 m above sea level (ASL), setting the height at 1000 m ASL takes into account the minimum altitude needed to traverse the average elevation of the Taihang Mountains (ranging from 1000 to 1500 m ASL). This ensures that the simulated trajectory paths over this topographical barrier are physically realistic.

The Angle Distance algorithm was used to cluster air mass trajectories, enabling the identification of dominant air mass directions and potential pollution sources affecting the study site during different periods. The optimal number of clusters was determined by evaluating the spatial variance (SPVAR)

of each trajectory from the cluster mean and the total spatial variance (TSV). The final classification was selected just before the second rapid increase in TSV. The underlying principle is that TSV initially rises sharply during clustering, then increases gradually; however, once the number of clusters reaches a certain threshold, TSV surges again, indicating that the merged clusters are highly dissimilar, marking the end of the classification process. The classification results correspond to the different air mass categories before this final merging step. The mean trajectories of these clusters represent the primary airflow patterns at the target site during the analysis period (Wang et al., 2009).

The HYSPLIT simulations utilized meteorological input data from the Global Data Assimilation System (GDAS) with 3D wind vectors, temperature, relative humidity, geopotential height, surface pressure, and boundary layer diagnostics.

In the revised version, these descriptions have been added in Section 2.2.3:

"Backward trajectories were calculated using the mixed-particle Lagrangian integrated trajectory method (HYSPLIT, https://www.readv.nnaa.gov/HYSPLIT trai.phr.). Meteorological input data were from the Global Data Assimilation System (GDAS) with 3D wind vectors, temperature, relative humidity, geopotential height, surface pressure, and boundary layer diagnostics. 24-h backward trajectories were simulated for air masses above 1000 m above ground level in Zhengzhou. While the surface elevation of Zhengzhou is approximately 100 m above sea level (ASL), setting the height at 1000 m ASL takes into account the minimum altitude needed to traverse the average elevation of the Taihang Mountains (ranging from 1000 to 1500 m ASL). This ensures that the simulated trajectory paths over this topographical barrier are physically realistic.

The Angle Distance algorithm was used to cluster air mass trajectories, enabling the identification

of dominant air mass directions and potential pollution sources affecting the study site during different periods. The optimal number of clusters was determined by evaluating the spatial variance (SPVAR) of each trajectory from the cluster mean and the total spatial variance (TSV). The final classification was selected just before the second rapid increase in TSV. The underlying principle is that TSV initially rises sharply during clustering, then increases gradually; however, once the number of clusters reaches a certain threshold, TSV surges again, indicating that the merged clusters are highly dissimilar, marking the end of the classification process. The classification results correspond to the different air mass categories before this final merging step. The mean trajectories of these clusters represent the primary airflow patterns at the target site during the analysis period (Wang et al., 2009). Subsequently, trajectories from two periods, 2013–2018 and 2019–2022, were clustered separately to analyze the variations between the two policy implementation periods."

4. Sections 3.1 and 3.2: How did the author estimate the decreases in the mass concentrations? Which year was chosen as the reference year? A proper analysis using Mann-Kendall and Sen's slope should be carried out here and in other places associated with trend analysis.

Response: Thank you for your comment. In the revised manuscript, the Mann - Kendall, Sen's slope, and Least - Squares slope methods were comprehensively used to analyze the inter - annual change trends. Please refer to Table S4 for details.

Table S4. Analysis of the inter - annual trends of CM and Ca²⁺ concentrations and pH during different periods using multiple methods.

	2011–2013			2013–2019			2019–2022		
	CM	Ca ²⁺	pH	CM	Ca ²⁺	<mark>рН</mark>	CM	Ca ²⁺	pH
MK-Z	3.01	2.70	1.41	- 9.74	-13.62	3.00	2.48	8.21	5.12
MK-p	0.003	0.007	0.159	<2.2 E–16	<2.2E–16	0.003	0.013	2.20E-16	2.99E-07
Sen's slope	0.082	0.023	7.10E-03	-0.015	-4.14E-03	9.15E-04	5.80E-03	5.42E-03	2.93E-03
LS slope	2.65	0.61	<mark>/</mark>	-0.81	-0.32	0.11	0.24	0.40	0.21

^{**} MK-Z and MK-p represent the trend (Z) and significance (p) calculated by the Mann - Kendall method using daily data, respectively; Sen's slope represents the Sen slope using daily data; LS slope represents the Least - Squares slope using annual data. All the above calculations were performed using the R language (R version 4.0.2).

5. Please briefly describe the Air Pollution Prevention and Control Action Plan and Three-Year Action Plan in the main text. In general, readers outside China have no idea about these policies.

Response: Thank you for your comments. We have added a description of the policies: "Over the past twelve years, the Chinese government implemented the Air Pollution Prevention and Control Action Plans (2013–2018) and the Three-Year Action Plan (2018–2020). The Air Pollution Prevention and Control Action Plan focused on reducing PM2.5 concentrations in key regions and aiming to cut PM2.5 levels by 10–25% in priority areas over five years. To achieve these goals, it adopted several measures. In terms of industrial restructuring, it mandated the elimination of a large amount of outdated production capacity in industries such as iron/steel and cement to optimize the industrial structure and reduce high-pollution production. For emission standards, it set strict requirements for multiple industrial sectors, especially coal-fired power plants, and gradually introduced ultra-low emission requirements to control pollutants like SO₂, NO_x, and PM. Regarding energy transition, it promoted a shift from coal to cleaner energy sources, including capping coal consumption in certain regions and restricting the construction of small-scale coal-fired boilers. Subsequently, the Three-Year Action Plan was carried out to continue improving air quality with a broader scope of regions under control, further reducing pollutant emissions and enhancing the overall air quality index. The measures included enhanced transportation controls, such as introducing stricter vehicle emission standards (like National VI standards for vehicles) and establishing diesel truck exclusion zones in many cities to reduce emissions from the transportation sector. It also adopted precision governance through grid-based environmental supervision, dividing areas into small grids for more accurate and efficient monitoring of pollution sources. Additionally, it strengthened the legal and institutional framework by revising relevant laws, such as the Air Pollution Prevention and Control Law, to strengthen legal penalties for environmental violations and implementing an environmental tax system to encourage enterprises to reduce emissions. Correspondingly, the annual average concentration of PM_{2.5} in Zhengzhou decreased from $212 \pm 102 \,\mu\text{g/m}^3$ in 2013 to $60 \pm 41 \,\mu\text{g/m}^3$ in 2022, representing a reduction of approximately 72%."

6. Please explain why Zhengzhou is representative of a typical inland city in North China in the main text. There is no background information about aerosol research conducted in Zhengzhou and how they compare with those findings in other typical Chinese megacities, such as Beijing or Shanghai.

Response: Thanks for your comment. We have added a summary section:

"Zhengzhou presents unique atmospheric chemistry that distinguishes it from other mega-cities in North China. As the capital of China's foremost agricultural province (Henan Province, contributing 18% of national NH₃ emissions), Zhengzhou's PM_{2.5} composition combined substantial crustal material (15 ± 3% in PM_{2.5} vs. <10% in Beijing) with exceptional ammonia abundance (Huang et al., 2012; Liu et al., 2018; Wang et al., 2018). This created distinct particle acidity characteristics, maintaining pH 4.5–6.0 compared to lower pH levels (3.3–5.4) in other cities like Beijing (Ding et al., 2019; Zhang et al., 2021). However, two critical research gaps persist: (1) the long-term evolution of CM under control policies remains unquantified; (2) the role of CM on pH buffer capacity in NH₃-enriched environments lacks systematic assessment.

To address these gaps, our study pioneers the first multi-decadal analysis (2011–2022) coupling PM_{2.5} components with thermodynamic modeling through three key innovations: (1) revealing the long-term trends of CM, (2) analyzing the variations of CM sources, and (3) exploring pH trend and

its relationship with CM. The resultant findings advance our understanding of urban aerosol acidity chemistry by underscoring the critical role of CM."

Minor Comment:

1. Figure S1: Please label the highways, coal-fired power plants, and gas-fired power plants on the map.

Response: Thanks for your comment. We have updated Figure S1:

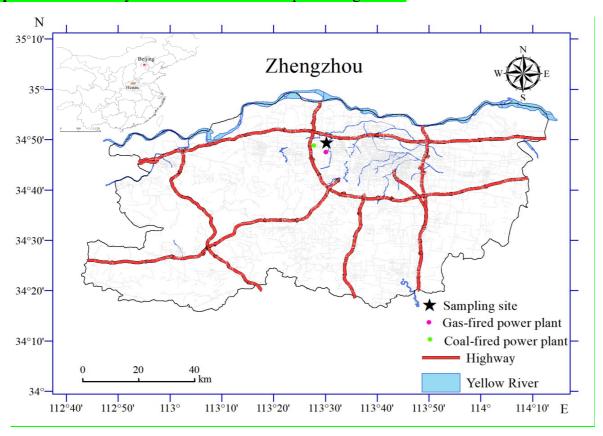


Figure S1.Sampling site in Zhengzhou, China. © 2019 National Geomatics Center of China. All

rights reserved.

2. Section 2.1: What are the uncertainties in the measured concentrations for individual components?

Response: Thanks for your comment. The method detection limits and measurement uncertainties are

summarized in Table S2.

Table S2 The method detection limit (MDL) and measurement uncertainties (Unc) of individual

components

	MDLs (μg/m ³)	Unc (%)
EC	0.1	13.1
OC	0.1	<mark>9.8</mark>
Na ⁺	0.005	<mark>9.6</mark>
$\mathrm{NH_4}^+$	0.011	10.1
$rac{ extbf{K}^+}{ extbf{Mg}^{2^+}}$	<mark>0.006</mark>	<mark>9.5</mark>
Mg^{2+}	0.002	10.1 9.5 9.3 8.8 8.2 9.3
Ca^{2+}	0.017	<mark>8.8</mark>
<mark>F</mark>	0.001	8.2
C1 ⁻	0.001	9.3
NO_3^-	0.015	10.1
SO ₄ ²⁻	0.031	<mark>9.9</mark>
Na	0.003	10.9
Mg	0.002	10.6
Al	0.004	9.2
<u>S1</u>	0.005	9.2 9.3 9.5 9.4 9.4
<u>CI</u>	0.008	9.5
K	0.005	9.4
Ca V	0.01	9.4 57.0
AI Si CI K Ca V Ni Cr Mn	0.008	57.9
INI Cr	0.006	96.6 24.7
M _D	0.02 0.02	24.7 16.8
Fe	0.03	9.3
Co	0.009	79.6
C ₁	0.005	77.0 5.8
7.n	0.003	79.6 5.8 8.4
Co Cu Zn Ga	0.005	84.7
	0.008	27.4
Se	0.006	25.7
As Se Sr Cd Sn	0.006	22.8
Cd	0.03	68.6
<mark>Sn</mark>	0.02	42.0
<mark>Sb</mark>	0.02	73.6
<mark>Ba</mark>	0.02	15.6
Pb	0.02	13.4

3. Lines 114 – 115: Apart from just citing references here, please provide the details about analytical methods and quality control in the supplement.

Response: Thanks for your comments. We have added detailed analytical methods and quality control in both the Manuscript and Supplement material:

Manuscript

"Blank filters were also routinely analyzed with each batch of samples to detect sample contamination and provide quality assurance on the elemental concentrations. Detailed analytical methods and quality control are described in the supplement (Text S1). The method detection limits and measurement uncertainties are summarized in Table S2."

Supplement material

"Text S1 Instruments and Measurements

Samples were collected using a high-volume sampler (TE-6070D, Tisch, USA) and air particulate samplers (TH-16A, Tianhong, China) from April 2011 to December 2022. Two quartz filters and two Teflon filters were used daily from 10:00 AM to 9:00 AM the next day, resulting in a total of 5848 samples. After excluding abnormal data due to instrument malfunctions, 4228 valid samples were obtained. Detailed information on the samples is provided in Table S1.

Organic carbon (OC) and elemental carbon (EC) were analyzed using a carbon analyzer (Model 5L, Sunset Laboratory, USA). The analysis of EC and OC was conducted in two stages. In the first stage, the filter membrane was placed in a quartz heating furnace under a pure helium atmosphere. As the temperature gradually increased to approximately 580°C, OC was volatilized and released. In the second stage, heating continued in a mixed atmosphere of 2% oxygen and 98% helium. When the temperature reached approximately 870°C, EC underwent oxidative decomposition and was released.

During the helium flow transmission, OC and EC released at different temperatures were completely oxidized to CO₂ in a MnO₂ oxidation furnace and subsequently reduced to CH₄ for detection by a flame ionization detector (FID). The detection limits for both OC and EC were 0.2 μg/cm². Before each sample analysis, calibration was performed using a standard sucrose solution. Additionally, parallel tests were conducted every ten samples to ensure accuracy.

Water-soluble inorganic ions (Cl⁻, NO₃⁻, SO₄²⁻, Na⁺, NH₄⁺, K⁺, Mg²⁺, and Ca²⁺) were measured using ion chromatography (ICS-90 and ICS-900 models, Dionex, USA). Half of the PM_{2.5} filter was cut into pieces and ultrasonically extracted with 20 mL of Milli-Q water for 30 min, followed by filtering through a 0.45 mm polytetrafluoroethylene syringe filter before analysis. The cation concentrations were determined by an IonPacASII-HC4 mm anion separation column and an IonPacAGII-HC4 mm guard column, with 20 mM methane sulfonate as an eluent at 0.8 mL/min. The anions were measured by an IonPacCS12A cation separation column and an IonPacCG12A guard column, with a solution of 8.0 mM Na₂CO₃ + 1.0 mM NaHCO₃ as an eluent at 1.0 mL/min. The regression coefficients (R²) of the calibration curves were over 0.9996 for all ions, except NH₄⁺ (0.9988), which showed a quadratic response.

Elements were analyzed using a wavelength dispersive X-ray fluorescence spectrometer (S8 TIGER, Bruker, Germany) to determine concentrations of Fe, Na, Mg, Al, Si, Cl, K, Ca, V, Ni, Cu, Zn, Cr, Mn, Co, Cd, Ga, As, Se, Sr, Sn, Sb, Ba, and Pb (Tremper et al., 2018), which has been approved by the United States Environmental Protection Agency (Chow and Watson, 1994). The spectrometer was equipped with an X-ray tube featuring close coupling among the tube, sample, and detector, ensuring high efficiency and optimal excitation of elements within the sample. Before analysis, the instrument was calibrated using a series of high-quality, self-prepared standards. Calibration

procedures were conducted following established methods (Chow and Watson, 1994). To assess potential contamination and ensure data quality, blank filters were routinely analyzed alongside each batch of samples."

4. Section 2.2.2: What was the set activity coefficient of H in the model?

Response: With ISORROPIA, γH^+ and γOH^- are assumed to be equal to unity, whereas the activity coefficients for the other ionic pairs (e.g., H^+ – Cl^-) are calculated by Kusik-Meissner method (Fountoukis and Nenes, 2007).

Fountoukis, C. and Nenes, A.: ISORROPIA II: a computationally efficient thermodynamic equilibrium model for K^+ – Ca^{2+} – Mg^{2+} – NH_4^+ – Na^+ – SO_4^{2-} – NO_3^- – Cl^- – H_2O aerosols, Atmos. Chem. Phys., 7, 4639–4659, https://doi.org/10.5194/acp-7-4639-2007, 2007.

5. Line 172: There is no data showing that the WS was higher in spring and summer compared to autumn and winter.

Response: According to the comments of Reviewer 1, this part has been revised to:

"Seasonal trends (Fig. S5) reveal significant declines in CM during spring in 2013–2019 with WS decreasing from 2.2 m/s in 2013 to 1.4 m/s in 2019 (Fig. S6) and stable RH (Fig. S7). Similarly, summer CM reductions in 2013–2019 corresponded with WS declines. These patterns suggest spring-summer CM improvements resulted from the synergistic effects of meteorological changes and dust control policies. Conversely, autumn-winter seasons showed limited CM reductions despite comparable WS decreases in 2013–2019, highlighting the need for enhanced dust emission controls in

Zhengzhou during these seasons."

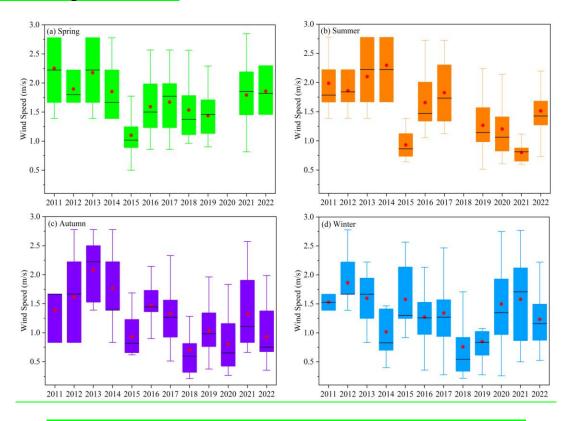


Figure S6 The variation in WS across different seasons from 2011 to 2022.

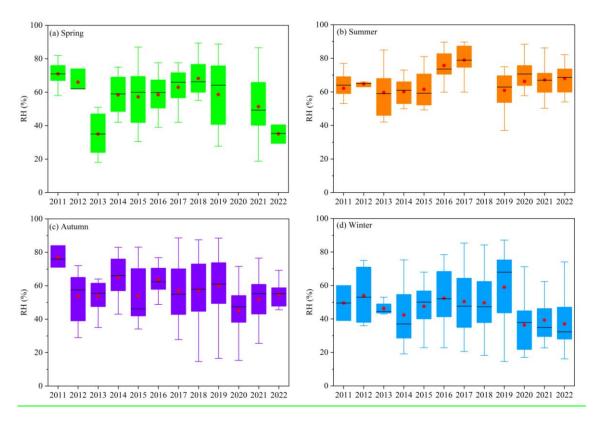


Figure S7 The variation in RH across different seasons from 2011 to 2022.

6. Figure S4: Is there any data for Ca²⁺ in 2015?

Response: Thank you for your comments. The relevant data for Ca²⁺ has been presented in Figure 2 of the main text.

7. Lines 229 – 231: Please provide a discussion about why the pH trend is similar to those in Beijing but different from those in Shanghai and Hong Kong.

Response: Thanks for your suggestion. We have added a discussion: "The increasing trend in pH values observed in this study is similar to the findings in Beijing (Song et al., 2019; Xie et al., 2020), presumably attributable to the comparable chemical composition trends and meteorological conditions. In contrast, Shanghai and Hong Kong display divergent trends (Nah et al., 2023; Zhou et al., 2022). This disparity might be ascribed to the stronger buffering effect exerted by NH₃ and dust in Zhengzhou than marine aerosols (Na⁺/Cl⁻) in these coastal cities (Shi et al., 2017; Liu et al., 2019). Moreover, the relatively higher temperatures and more abundant rainfall in Shanghai and Hong Kong could also contribute to the distinct trends observed in their pH values."

8. Figure 5: How did the authors estimate the contribution of a component to the changes in pH? **Response:** First, we conducted a sensitivity analysis using all the observational data from 2011 to 2022.

A given range for a variable (i.e., TNHx) along with the corresponding average values of other parameters (i.e., SO₄²⁻, NO₃⁻, Na⁺, Cl⁻, Ca²⁺, K⁺, Mg²⁺, RH, and T) was input into ISORROPIA-II to investigate the sensitivity of particle pH to this variable (i.e., TNHx), and the results are presented in

Fig. S10 in the Supplement. Based on the sensitivity analysis curves, the pH values corresponding to a variable in different years were calculated according to the average values of this variable in different years (Table S6). The difference in pH values of this variable between two adjacent years was defined as Δ pH. The above description has also been added to the main text:

"Sensitivity analyses were conducted to explore the dominant factors driving the elevated particle pH in Zhengzhou by giving a range for one parameter (i.e., TNHx) and average values for other parameters (i.e., SO₄²⁻, NO₃⁻, Na⁺, Cl⁻, Ca²⁺, K⁺, Mg²⁺, RH, and T) input into the ISORROPIA-II model."

"Based on the sensitivity analysis curves, the pH values corresponding to a variable in different years were calculated according to the average values of this variable in different years (Table S6). The difference in pH values of this variable between two adjacent years was defined as Δ pH which is illustrated in Fig. 5"

9. Where is the figure or table associated with Lines 256 - 266?

Response: Sorry for the misunderstanding. This sentence has been modified to: "During the period from 2020 to 2022, the influence of SO_4^{2-} on particle pH gradually decreased, with a decrease in concentration from 0.3 to 2.3 μ g/m³ (Table S6) only bringing about a pH decrease of 0.03 to 0.14 (Fig. 5).

Technical Comment:

1. Line 17: It is unclear what is "particle pH response"

Response: Thank you for your careful reading of our paper. We have implemented the following

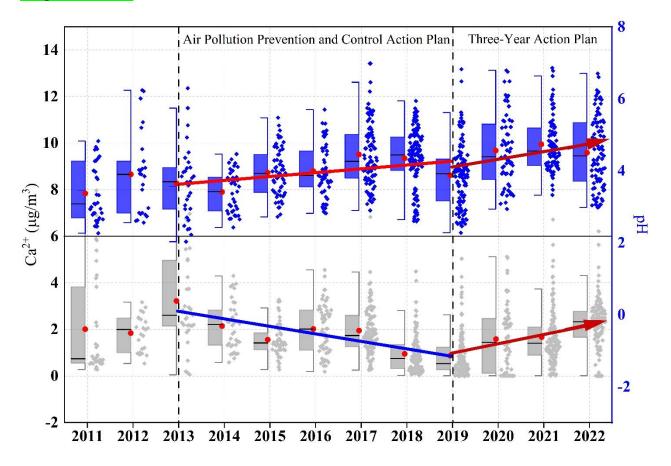
revisions: "Emerging concerns regarding aerosol acidity trends have been highlighted amid China's sustained initiatives to mitigate emissions of both acidic and alkaline precursors, especially in North China, which is significantly affected by dust aerosol."

2. Line 19: It should be "12-year".

Response: Done!

3. Graphical abstract: Please use a different color for the pH. Apparently, the color light blue is too bright for visualization.

Response: Done!



4. Line 57: Please use a word different than "research hotspot".

Response: OK. We have revised "research hotspot" to "research focus".

5. Line 64: Please use a different term instead of "fostered a persistent belief".

Response: This sentence has been deleted referring to Reviewer 1.

6. Eq 4: It should be "NH₄".

Response: Done!

7. Figures in main text and SI: the labels for x- and y- axis and annotations are too small to read.

All figures need to be modified for better visualization.

Response: Done!

8. Lines 173 and 174: more effective than what?

Response: Sorry for the misunderstanding. This sentence has been deleted referring to Reviewer 1.

9. Line 246: NO₃⁻ can be used to replace nitrate ions.

Response: Done!

10. Line 252: SO_4^{2-}/NO_3^{-} can means the ratio between SO_4^{2-} and NO_3^{-} . I assume the authors use "/" to represent "or" here. If so, please use the word "or"

Response: Done!

11. The format of "TNH_x" should be consistent throughout the manuscript.

Response: Done!

12. Some color choices in figures need to be reconsidered. Certain annotations in color are not

color-friendly for visualization.

Response: Thanks for your suggestion. The colors of the figures in both the main text and supplement

have been adjusted to enhance their readability.

Manuscript

- 2 Measurement report: Crustal materials play an increasing role
- 3 in elevating particle pH: Insights from 12-year records in a
- 4 typical inland city of China.
- 5 Hongyu Zhang^{1, 2}, Shenbo Wang^{2, 3*}, Zhangsen Dong^{1, 2*}, Xiao Li^{2, 3}, Ruiqin Zhang^{2, 3}
- ⁷ Collage of Chemistry, Zhengzhou University, Zhengzhou, 450000, China
- ² Research Institute of Environmental Sciences, Zhengzhou University, Zhengzhou
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- 11 China

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Abstract

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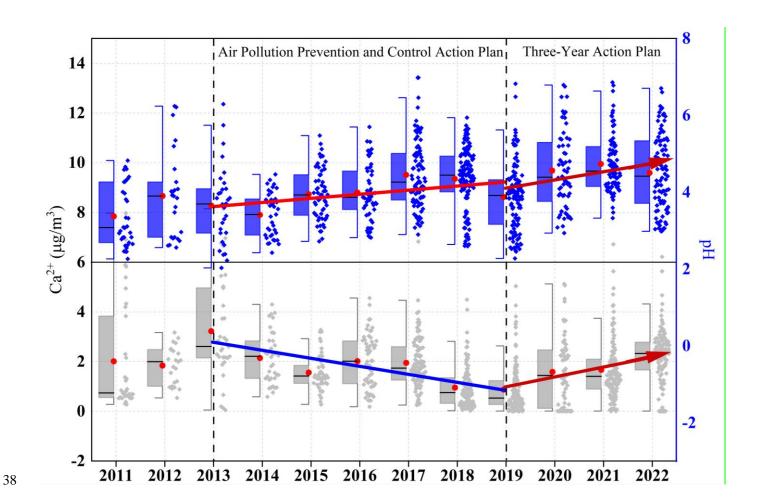
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Particle acidity serves as a key determinant in atmospheric chemical processes. Emerging concerns regarding aerosol acidity trends have been highlighted amid China's sustained initiatives to mitigate emissions of both acidic and alkaline precursors, especially in North China, which is significantly affected by dust aerosol. 12-year observational data in Zhengzhou reveal that the annual average PM_{2.5} concentration decreased from $162 \pm 81 \, \mu g/m^3$ in 2011 to $60 \pm 41 \, \mu g/m^3$ in 2022, with the largest reduction in sulfate (73%). Correspondingly, the annual particle pH increased by 0.10 units from 2011 to 2019. In addition, the elevated particle pH in 2015 and 2018 was notably influenced by the increase in TNH_x (NH₃ + NH₄⁺). Note that the crustal material concentrations and their proportions increased significantly during 2019–2022, which might be responsible for the resuspension of surrounding soil dust. Even though the TNH_x concentration was decreasing, the annual average growth rate of pH values increased to 0.21 units from 2019 to 2022. This phenomenon is not unique to Zhengzhou, as major cities in the North China Plain have also experienced a pronounced upward trend in coarse particles after 2019. Therefore, the long-term evolution of particle acidity in North China will require comprehensive consideration of synergistic effects involving acidic precursors, ammonia, and crustal materials.

32 **Keywords:** Dust, aerosol acidity, sources, North China Plain, control measurement

Synopsis: The future ammonia reduction policies in North China may not lead to a rapid increase in particle acidity in the presence of crustal materials., which further elevated the particle pH after 2019.

37 Graphical abstract:



Highlights:

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- Crustal material concentrations and their proportions increased significantly during 2019–2022;
- The resuspension of surrounding soil dust may determine the rebound of crustal material concentrations;
- Rebound in crustal material further elevated the particle pH.

44 1 Introduction

Particle acidity is a critical parameter that affects atmospheric chemistry, such as the gas-particle

partitioning of semi-volatile and volatile species (Surratt et al., 2010; Guo et al., 2016), the solubility of metals (Tao and Murphy, 2019), acid-catalyzed reactions (Rengarajan et al., 2011), and acid deposition (Mao et al., 2009), thereby determining aerosol concentration and chemical composition, as well as impacting human health, ecosystems, and climate (Li et al., 2017; Pye et al., 2020; Su et al., 2020; Nenes et al., 2021). Generally, the global fine particulate matter (PM_{2.5}, aerodynamic diameter \leq 2.5 µm) exhibits a bimodal pH distribution ranging from 1–3 (e.g., in the United States and Europe) (Guo et al., 2015; Battaglia et al., 2017; Masiol et al., 2020; Zhang et al., 2021) and 4-5 (e.g., in East Asia) (Kim et al., 2022; Sharma et al., 2022). The atmosphere rich in gaseous ammonia (NH₃) and crustal material (CM) shows significant pH buffering effects (Wang et al., 2020; Zheng et al., 2020; Karydis et al., 2021), which is a dominant factor that drives the high particle pH in East Asia (Karydis et al., 2021; Zhang et al., 2021; Kim et al., 2022; Sharma et al., 2022). In recent years, the changing trends in particle pH have become a research focus, especially in China, in response to air pollution control policies, i.e. Air Pollution Prevention and Control Action Plan (2013–2018) and Three-Year Action Plan (2018–2020). The annual average PM_{2.5} concentration in Beijing dropped by 64% from 89.5 µg/m³ in 2013 to 32 µg/m³ in 2023 (MEP, 2023), with a clear downward trend of sulfate concentration, and nitrate surpassing sulfate as the primary component (Zhai et al., 2019; Zhou et al., 2019; Li et al., 2023). The atmospheric behavior of ammonium, governed by gas-particle partitioning processes involving ammonia (NH₃) as the predominant alkaline gas, demonstrates notable stability in concentration levels, with observational records showing less than 5% interannual variation in NH₃ column densities over North China during 2015–2019 (Dong et al., 2023). Under such conditions, the dominant inorganic aerosol component transitions from ammonium sulfate

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to ammonium nitrate. This compositional shift enhances atmospheric particulate hygroscopicity due to ammonium nitrate's superior water uptake capability, ultimately elevating particle pH levels through aqueous-phase dilution mechanisms (Wexler and Seinfeld, 1991; Pinder et al., 2007, 2008; Heald et al., 2012; Weber et al., 2016). For instance, a significant increase in the nitrate-to-sulfate molar ratio from 2014–2017 in Beijing resulted in the particle pH increasing from 4.4 to 5.4 (Xie et al., 2020). Moreover, increased NH₃ concentrations raised particle pH by 0.3-0.4 units from 2014/2015 to 2018/2019 in Beijing (Song et al., 2019). Over Europe and North America, the pH has increased strongly from about 2.8 and 2.2 during the 1970s to 3.9 and 3.3 in 2020 respectively, especially during the 1990s, with significantly increasing NH₃ emission (Karydis et al., 2021). On the contrary, modeling results indicate a continuous decline in pH in East Asia from 1970 to 2020 due to sharp increases in SO₂ and NO_x emissions (Karydis et al., 2021). In addition, the PM_{2.5} pH showed a slight decrease of 0.13 from 2018 to 2022 summer in Beijing due to the change in total nitrate (NO₃⁻ + HNO₃) (Li et al., 2023). Moreover, Zhou et al. (2022) found a decreasing pH trend from 2011 to 2019 in eastern China, primarily influenced by temperature, followed by sulfate and non-volatile cations. Similarly, Nah et al. (2023) observed a decreasing pH trend from 2011 to 2020 in Hong Kong, attributing it to temperature and sulfate levels. Thus, concerns have been raised about the potential increase in the acidity of aerosol and precipitation due to China's ongoing efforts to reduce ammonia emissions, which pose severe health risks and acid deposition (Liu et al., 2019; Shi al., 2019). In addition to NH₃, CM is another key alkaline substance, that buffers particle pH. Ca²⁺ can form insoluble CaSO₄ with sulfate, reducing sulfate concentration in the aqueous phase of aerosol, and thus

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lowering H⁺ and aerosol liquid water content (ALWC) concentrations and enhancing particle pH (Ding

et al., 2019; Karydis et al., 2021). Moreover, non-volatile cations can lower the molar ratio of ammonia to sulfate, leading to an increase in particle pH (Zheng et al., 2022). Karydis et al. (2021) framework demonstrated that CM played a critical buffering role in sustaining aerosol pH around 7 across the Middle East arid regions. The model sensitivity tests revealed that under hypothetical dust-free conditions (CM = 0), aerosol acidity would escalate to pH~4 due to NH₄+/SO₄2- domination. Wang et al. (2022) reported that non-volatile cations accounted for approximately 8–17% of hourly aerosol pH variation. Li et al. (2023) indicated that the buffering effect of cations was the major reason for the relatively small pH changes from 2018 to 2022 in Beijing, emphasizing that reducing coarse particle emissions in the future could significantly decrease particle pH. In addition, there was a rising trend in the contribution of CM to particle pH in Tianjin, China (Shi et al., 2017). Therefore, it is evident that CM has a significant impact on the variation of particle pH, especially in North China, which is significantly affected by dust aerosol, but the trend of CM concentration and its long-term implication is still lacking unfortunately.

Zhengzhou presents unique atmospheric chemistry that distinguishes it from other mega-cities in North China. As the capital of China's foremost agricultural province (Henan Province, contributing 18% of national NH₃ emissions), Zhengzhou's PM_{2.5} composition combined substantial crustal material (15 ± 3% in PM_{2.5} vs. <10% in Beijing) with exceptional ammonia abundance (Huang et al., 2012; Liu et al., 2018; Wang et al., 2018). This created distinct particle acidity characteristics, maintaining pH 4.5–6.0 compared to lower pH levels (3.3–5.4) in other cities like Beijing (Ding et al., 2019; Zhang et al., 2021). However, two critical research gaps persist: (1) the long-term evolution of CM under control policies remains unquantified; (2) the role of CM on pH buffer capacity in NH₃-

enriched environments lacks systematic assessment.

To address these gaps, our study pioneers the first multi-decadal analysis (2011–2022) coupling PM_{2.5} components with thermodynamic modeling through three key innovations: (1) revealing the long-term trends of CM, (2) analyzing the variations of CM sources, and (3) exploring pH trend and its relationship with CM. The resultant findings advance our understanding of urban aerosol acidity chemistry by underscoring the critical role of CM.

2 Experiment and method

2.1 Instruments and Measurements

E) in Zhengzhou, China. The sampling site (Fig. S1), approximately 14 m above the ground, is primarily surrounded by residential areas with well-developed transportation networks and no significant industrial sources. There are two highways located 3 km to the south and 7 km to the east. Additionally, a coal-fired power plant located 6 km to the east was shut down in 2020, and a gas-fired power plant is situated 3 km to the south.

Samples were collected using a high-volume sampler (TE-6070D, Tisch, USA) and air particulate samplers (TH-16A, Tianhong, China) from April 2011 to December 2022. Two quartz filters and two Teflon filters were used daily from 10:00 AM to 9:00 AM the next day, resulting in a total of 5848 samples. After excluding abnormal data due to instrument malfunctions, 4228 valid samples were obtained. Detailed information on the samples is provided in Table S1. Organic carbon (OC) and

Sampling was conducted on the fourth-floor platform at Zhengzhou University (34.75° N, 113.61°

elemental carbon (EC) were analyzed using a carbon analyzer (Model 5L, Sunset Laboratory, USA).

Water-soluble inorganic ions (Cl⁻, NO₃⁻, SO₄²⁻, Na⁺, NH₄⁺, K⁺, Mg²⁺, and Ca²⁺) were measured using ion chromatography (ICS-90 and ICS-900 models, Dionex, USA) (Yu et al., 2017; Jiang et al., 2018). Elements were analyzed using a wavelength dispersive X-ray fluorescence spectrometer (S8 TIGER, Bruker, Germany) to determine concentrations of Fe, Na, Mg, Al, Si, Cl, K, Ca, V, Ni, Cu, Zn, Cr, Mn, Co, Cd, Ga, As, Se, Sr, Sn, Sb, Ba, and Pb (Tremper et al., 2018). Meteorological conditions, including temperature (T), relative humidity (RH), and wind speed (WS) were obtained using an automatic weather station (Wang et al., 2019). Blank filters were also routinely analyzed with each batch of samples to detect sample contamination and provide quality assurance on the elemental concentrations. Detailed analytical methods and quality control are described in the supplement (Text S1). The method detection limits and measurement uncertainties are summarized in Table S2. The quality assurance protocol excluded temporally discrete dust storm and precipitation periods to prevent contamination of the source analysis of CM and modeling particle pH, given that such events induce nonrepresentative extremes in both crustal element concentrations and pH values, coupled with elevated PM measurement uncertainties. The annual mean PM_{2.5} concentration data for cities in the North China Plain were obtained from the China National Environmental Monitoring Center (CNEMC), available at https://www.cnemc.cn/.

2.2 Data Analysis

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2.2.1 Mass reconstruction

The calculation method for CM is as follows (Tian et al., 2016):

$$[CM]=1.89\times[AI] + 2.14\times[Si] + 1.4\times[Ca] + 1.43\times[Fe] + 1.94[Ti]$$
 (1)

where [Al], [Si], [Ca], [Fe] and [Ti] represent the concentrations of the respective elements (μg/m³), but Ti was not measured.

2.2.2 Thermodynamic model

The particle pH was calculated using the ISORROPIA-II mode (version 2.1, http://isorropia.eas.gatech.edu). The input data (excluding RH \leq 30%), including SO₄²⁻, TNO₃ (HNO₃ + NO₃⁻), TNH_x (NH₃+NH₄⁺), Ca²⁺, K⁺, Na⁺, Mg²⁺, Cl⁻, RH and T, with the temporal resolution aligned with the sampling periods (from 10:00 AM to 9:00 AM the following day). Input data (excluding RH \leq 30%) included SO₄²⁻, TNO₃ (HNO₃ + NO₃⁻), TNH_x (NH₃+NH₄⁺), Ca²⁺, K⁺, Na⁺, Mg²⁺, Cl⁻, RH and T. The concentrations of hydrogen ions in air (H_{air}⁺) and ALWC were calculated using the aerosol equilibrium composition system Na⁺-K⁺-Ca²⁺-Mg²⁺-NH₄⁺-SO₄²⁻-NO₃⁻-Cl⁻-H₂O H_{air}⁺ (Fountoukis and Nenes, 2007). pH values were calculated using the following formula:

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$$pH = -\log_{10}H_{aq}^{+} \cong -\log_{10}\frac{1000H_{air}^{+}}{ALWC_{i} + ALWC_{o}} \cong -\log_{10}\frac{1000H_{air}^{+}}{ALWC_{i}}$$
 (2)

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$$ALWC_{o} = \frac{m_{org} \rho_{w}}{\rho_{w}} \frac{\kappa_{org}}{\left(\frac{1}{RH} - 1\right)}$$
 (3)

where ALWC_i and ALWC_o refer to the ALWC for inorganic and organic components, respectively. m_{org} denotes the mass of organic aerosol, ρ_w is the density of water (1.0 g/cm³), ρ_{org} is the density of organic material (1.4 g/cm³) (Guo et al., 2015), k_{org} is the hygroscopicity parameter for organic aerosol (0.087) (Chang et al., 2010; Li et al., 2016). The ISORROPIA-II model operated under metastable conditions in the forward mode. Due to the lack of measured data for gaseous HNO₃ and NH₃, TNO₃ was represented solely by NO₃⁻. The concentration of NH₃ was simulated based on a linear regression

equation proposed by Wei et al. (2023), who used the same data as this study from 2013 to 2020:

NH₃ = 19.909×RH+0.559×T-0.35×NH₄+0.123×NO₃+2.159×CI-0.224×SO₄²-154.923 (4) where NO₃-, SO₄²-, NH₄+, and CI⁻ correspond to their respective concentrations (μ g/m³). To validate the applicability of Equation 4 for annual NH₃ estimation and pH simulation in Zhengzhou, this study utilized both observed NH₃ data (from a Thermo Scientific URG-9000D ambient ion monitor, USA) and calculated NH₃ values derived from Equation 4 at the same monitoring site throughout 2022, inputting them into the thermodynamic model for pH simulation. As shown in Figure S2, pH values calculated from observed and simulated NH₃ exhibit good agreement (r=0.97, P<0.01). Furthermore, NH₃ concentrations modeled by ISORROPIA demonstrate a significant correlation (r=0.95, P<0.01) with that simulated NH₃ by Equation 4. These results collectively demonstrate the reliability of the NH₃ estimation method in this study.

2.2.3 HYSPLIT analysis

Backward trajectories were calculated using the mixed-particle Lagrangian integrated trajectory method (HYSPLIT, https://www.ready.noaa.gov/HYSPLIT traj.php). Meteorological input data were from the Global Data Assimilation System (GDAS) with 3D wind vectors, temperature, relative humidity, geopotential height, surface pressure, and boundary layer diagnostics. 24-h backward trajectories were simulated for air masses above 1000 m above ground level in Zhengzhou. While the surface elevation of Zhengzhou is approximately 100 m above sea level (ASL), setting the height at 1000 m ASL takes into account the minimum altitude needed to traverse the average elevation of the Taihang Mountains (ranging from 1000 to 1500 m ASL). This ensures that the simulated trajectory paths over this topographical barrier are physically realistic.

The Angle Distance algorithm was used to cluster air mass trajectories, enabling the identification of dominant air mass directions and potential pollution sources affecting the study site during different periods. The optimal number of clusters was determined by evaluating the spatial variance (SPVAR) of each trajectory from the cluster mean and the total spatial variance (TSV). The final classification was selected just before the second rapid increase in TSV. The underlying principle is that TSV initially rises sharply during clustering, then increases gradually; however, once the number of clusters reaches a certain threshold, TSV surges again, indicating that the merged clusters are highly dissimilar, marking the end of the classification process. The classification results correspond to the different air mass categories before this final merging step. The mean trajectories of these clusters represent the primary airflow patterns at the target site during the analysis period (Wang et al., 2009). Subsequently, trajectories from two periods, 2013–2018 and 2019–2022, were clustered separately to analyze the variations between the two policy implementation periods.

3 Results and discussion

3.1 Temporal variations in chemical components

Over the past twelve years, the Chinese government implemented the Air Pollution Prevention and Control Action Plans (2013–2018) and the Three-Year Action Plan (2018–2020). The Air Pollution Prevention and Control Action Plan focused on reducing PM_{2.5} concentrations in key regions and aiming to cut PM_{2.5} levels by 10–25% in priority areas over five years. To achieve these goals, it adopted several measures. In terms of industrial restructuring, it mandated the elimination of a large amount of outdated production capacity in industries such as iron/steel and cement to optimize the

industrial structure and reduce high-pollution production. For emission standards, it set strict requirements for multiple industrial sectors, especially coal-fired power plants, and gradually introduced ultra-low emission requirements to control pollutants like SO₂, NO_x, and PM. Regarding energy transition, it promoted a shift from coal to cleaner energy sources, including capping coal consumption in certain regions and restricting the construction of small-scale coal-fired boilers. Subsequently, the Three-Year Action Plan was carried out to continue improving air quality with a broader scope of regions under control, further reducing pollutant emissions and enhancing the overall air quality index. The measures included enhanced transportation controls, such as introducing stricter vehicle emission standards (like National VI standards for vehicles) and establishing diesel truck exclusion zones in many cities to reduce emissions from the transportation sector. It also adopted precision governance through grid-based environmental supervision, dividing areas into small grids for more accurate and efficient monitoring of pollution sources. Additionally, it strengthened the legal and institutional framework by revising relevant laws, such as the Air Pollution Prevention and Control Law, to strengthen legal penalties for environmental violations and implementing an environmental tax system to encourage enterprises to reduce emissions. The long-term trends in PM_{2.5} concentrations and its chemical components in Zhengzhou from

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The long-term trends in PM_{2.5} concentrations and its chemical components in Zhengzhou from 2011 to 2022 are depicted in Fig. 1, with annual average concentrations listed in Table 1. Correspondingly, the annual average concentration of PM_{2.5} in Zhengzhou decreased from 162 ± 81 µg/m³ in 2011 to 60 ± 41 µg/m³ in 2022, representing a reduction of approximately 63%. In particular, the reduction rate reached 72% after 2013. As for chemical components, the largest reductions were observed in SO₄²⁻ (79%), decreasing from 38.0 ± 19.9 µg/m³ in 2013 to 7.9 ± 4.5 µg/m³ in 2022,

followed by EC (76%). Additionally, the concentrations of NH₄⁺ and NO₃⁻ also significantly decreased by 68% and 56%, respectively. The proportion of each component in PM_{2.5} (Fig. S3) reveals a decrease in SO₄²⁻, K⁺, and Cl⁻, indicating effective control measures targeting coal and biomass combustion (Lei et al., 2021). However, the proportions of NO₃⁻ and OC in PM_{2.5} rose from 11% and 12% in 2013 to 13% and 17% in 2022, respectively, similar to the trend observed in the North China Plain (Wen et al., 2018; Zhai et al., 2019; Li et al., 2023).

3.2 Temporal variations in CM

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Notably, there is no clear declining trend in the CM concentration, with a rebound observed during 2020–2022 (Fig. 1i). Furthermore, the proportion of CM in PM_{2.5} exhibits a significant upward trend (Fig. S3). To further analyze its trend, sampling data were divided into three periods corresponding to governmental stages: 2011–2013, when no special control measures were implemented; 2013–2019, coinciding with the implementation of the Air Pollution Prevention and Control Action Plan; and 2019–2022, coinciding with the Three-Year Action Plan. During these periods, Henan Province and Zhengzhou City implemented several dust control policies summarized in Table S3. As shown in Fig. 2a and 2b, the mass concentration of CM peaked at $14.6 \pm 8.3 \,\mu\text{g/m}^3$ in 2013, accounting for 8% of PM_{2.5}. To evaluate the inter - annual change trend of CM, the Mann - Kendall method, Sen's slope, and Least - Squares (LS) slope were comprehensively used with the results presented in Table S4. From 2013 to 2019, the CM concentration notably decreased from 14.6 ± 8.3 to 8.5 ± 7.8 µg/m³, with an annual average decline rate of 0.81 μg/(m³·year) from LS slope [0.015 μg/(m³·year) from Sen's slope. Apart from control measures, the interannual meteorological analysis shows (Fig. S4) WS exhibited a declining trend, with a decrease rate of 43%, while RH showed an increasing trend at a rate of 8% from 2013 to 2019, under which conditions that were unfavorable for dust resuspension (Wang et al., 2013, 2018). Seasonal trends (Fig. S5) reveal significant declines in CM during spring in 2013– 2019 with WS decreasing from 2.2 m/s in 2013 to 1.4 m/s in 2019 (Fig. S6) and stable RH (Fig. S7). Similarly, summer CM reductions in 2013–2019 corresponded with WS declines. These patterns suggest spring-summer CM improvements resulted from the synergistic effects of meteorological changes and dust control policies. Conversely, autumn-winter seasons showed limited CM reductions despite comparable WS decreases in 2013–2019, highlighting the need for enhanced dust emission controls in Zhengzhou during these seasons. As for the individual crustal elements in Fig. S8, Ca exhibited the highest average annual decline rate of 33% during 2013–2019, followed by Al. Si showed a less pronounced decline, attributed to its association with soil dust, where control measures for exposed soil are lacking (Zhang et al., 2020). In addition, the Ca²⁺ concentration as depicted in Fig. 2c decreased from $3.2 \pm 2.1 \,\mu\text{g/m}^3$ in 2013 to $2.2 \pm 1.1 \,\mu\text{g/m}^3$ in 2019, with an approximate annual average decline rate of 0.32 μg/(m³·year) from LS slope [4.14E–03 μg/(m³·year) from Sen's slope] in Table S4, further demonstrating the decline in dust source. It was worth noting that the proportions of CM, Ca, Al, Fe, Si, and Ca²⁺ in PM_{2.5} have shown consecutive annual increases from 2013 to 2019, with CM proportion increasing from 8% in 2013 to 14% in 2019, indicating that CM reduction lagged behind PM_{2.5} reduction efforts in Zhengzhou during this period. Additionally, both concentration and proportion of Ca^{2+} in 2022 (2.2 ± 1.1 µg/m³ and 14%) were higher than in other cities of China, such as Beijing (1.0 μ g/m³ and 2.8%), Tianjin (0.5 μ g/m³ and 1.4%), and Xiamen (0.48 μ g/m³ and 1.5%) (Shi et al., 2017; Xu et al., 2025; Zhang et al., 2021). These results indicate that CM remained an important component of PM_{2.5} in Zhengzhou City.

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During 2019–2022, both CM and Ca²⁺ concentrations exhibited significant rebounds, with annual growth rates of 0.24 and 0.4 μg/(m³·year) from LS slope [5.80E–03 and 5.42E–03 μg/(m³·year) from Sen's slope], respectively, and their proportions increased from 14% and 2% in 2019 to 22% and 5% in 2022. CM concentrations rebounded in all seasons, particularly in winter (Fig. S5). Changes in meteorological conditions may be a significant factor contributing to these concentration rebounds, accompanied by the average WS increased by 0.14 m/s and RH decreased by 7% from 2020 to 2022 (Fig. S4, S6, and S7), facilitating dust resuspension. Furthermore, the lack of more effective dust control measures, as indicated by the absence of significant changes in the dust control policies from the Air Pollution Prevention and Control Action Plan and Three-Year Action Plan, may be another important factor contributing to the rebound of dust.

3.3 Sources of CM

Elemental ratios were employed to characterize the sources of CM, with the Ca/Al ratio widely recognized as a reliable indicator of sandy origin (Zhang et al., 2017). In addition, significant variations in Ca/Si ratios (Table S5) were observed among different dust sources (Road, Construction, Piles, Soil). Fig. 3a illustrates the trend in Ca/Si ratios from 2011 to 2022. After 2013, Ca/Si ratios showed a declining trend annually, with the average ratio decreasing from a peak of 1.6 in 2016 to a lowest of 0.4 in 2022. Compared with Ca/Si ratios from different types of dust sources, the effect of road and construction dust on CM has gradually decreased. This may be attributed to the implementation of dust control measures such as enclosure, shielding, and dust suppression at construction and demolition sites, as well as dust control on ground surfaces and roads (Table S5). During 2019–2022, the average Ca/Si ratio remained below 1, with a mean of 0.4 in 2022, indicating that soil dust predominantly

contributed to CM. Currently, measures for controlling soil-suspended dust are limited, primarily relying on long-term strategies such as afforestation and increasing urban green coverage, thus requiring a longer process and sustained investment.

Sand dust transport serves as a significant source of CM in the North China Plain (Zhang et al., 2024). The Ca/Al ratio from 2016 to 2022 (Fig. 3b) shows minimal variation, with annual averages ranging between 1.5 and 2.5, indicating no significant changes in the source regions of sand. The transport trajectories reveal that the predominant pathways for long-distance transport of sand dust originated from Inner Mongolia, passing through Shaanxi and Shanxi provinces. Compared to 2013–2018 (45%), the influence of long-distance transport decreased to 25% during 2019–2022. In contrast, local transport within Henan province and short-distance transport from Shandong province showed a noticeable increase. These findings suggest that the rebound in CM concentrations during 2019–2022 in Zhengzhou might be responsible for the resuspension of surrounding soil dust.

3.4 Long-term trend of particle pH

Are shown in Fig. 4 and Table S4, pH values showed a clearly increasing trend after 2014. From 2013 to 2019, the annual pH increased by 0.11 units from the LS slope [9.15E–04 units from Sen's slope], reaching a maximum median value of 4.45 (Mean: 4.35) in 2018. Note that the annual average growth rate of pH values increased to 0.21 units from LS slope [2.93E–03 units from Sen's slope] from 2019 to 2022, with a maximum median value of 4.42 (Mean: 4.51) in 2022. Seasonally, pH values showed increasing trends in spring, summer, and autumn, and notably increased in winter from 2020 to 2022 (Fig. S9). The increasing trend in pH values observed in this study is similar to the findings in Beijing (Song et al., 2019; Xie et al., 2020), presumably attributable to the comparable chemical

composition trends and meteorological conditions. In contrast, Shanghai and Hong Kong display divergent trends (Nah et al., 2023; Zhou et al., 2022). This disparity might be ascribed to the stronger buffering effect exerted by NH₃ and dust in Zhengzhou than marine aerosols (Na⁺/Cl⁻) in these coastal cities (Shi et al., 2017; Liu et al., 2019). Moreover, the relatively higher temperatures and more abundant rainfall in Shanghai and Hong Kong could also contribute to the distinct trends observed in their pH values.

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Sensitivity analyses were conducted to explore the dominant factors driving the elevated particle pH in Zhengzhou by giving a range for one parameter (i.e., TNH_x) and average values for other parameters (i.e., SO₄²⁻, NO₃⁻, Na⁺, Cl⁻, Ca²⁺, K⁺, Mg²⁺, RH, and T) input into the ISORROPIA-II model. Are shown in Fig. S10, particle pH increases with the cation concentrations (e.g., TNH_x, K⁺, Ca²⁺, Mg²⁺, and Na⁺) and decreases with anions concentrations (e.g., SO₄²⁻ and NO₃⁻). Additionally, RH does not significantly affect pH, whereas an increase in T leads to a noticeable decrease in particle pH. Based on the sensitivity analysis curves, the pH values corresponding to a variable in different years were calculated according to the average values of this variable in different years (Table S6). The difference in pH values of this variable between two adjacent years was defined as ΔpH which is illustrated in Fig. 5. According to Equation (2), in addition to H⁺ concentration, particle pH is primarily influenced by the dilution effect of ALWC. Moreover, ALWC affects the gas-particle partitioning of semi-volatile compounds, thereby influencing particle acidity (Zuend et al., 2010; Zuend and Seinfeld, 2012). As shown in Fig.5 and Table S6, only in 2015, 2019, and 2020 did the increases in ALWC concentration (17.6 μg/m³, 4.1 μg/m³, and 11.6 μg/m³, respectively) lead to pH increases of 0.22, 0.06, and 0.14 units. This clearly cannot fully explain the significant pH increase in Zhengzhou since 2013.

Notably, since 2013, H^+ concentration has shown a decreasing trend. Particularly, H^+ concentrations decreased by 7.6×10^{-6} , 11.2×10^{-6} , and 7.2×10^{-6} mol/L in 2013, 2015, and 2017, respectively, leading to pH increases of 0.21, 0.36, and 0.42 units. After 2019, a continuous decline in H^+ concentration was observed for three consecutive years, resulting in pH increases of 0.21, 0.13, and 0.2 units in 2020, 2021, and 2022, respectively. These findings indicate that the increase in pH from 2019 to 2022 in Zhengzhou was primarily driven by the reduction in H^+ concentration.

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The concentration of H⁺ in the aerosol liquid phase is influenced by both chemical composition and meteorological conditions. To further understand the factors affecting ΔpH , we analyzed the variations in PM_{2.5} chemical components and meteorological parameters. Results indicate that the decline in SO₄²⁻ from 2013 to 2018 was the primary cause of the increase in particle pH, as it decreased H⁺ and ALWC concentrations (Fig. S11) in aerosol (Ding et al., 2019; Zhang et al., 2021). The average SO_4^{2-} concentration decreased by 14.6 and 5.3 μ g/m³, resulting in a pH increase of 0.43 and 0.35 units from 2013 to 2014 and 2016 to 2017, respectively, which was comparable to an increased rate of 0.3 units in East Asia due to SO₂ emission controls since 2016 (Karydis et al., 2021). As another acidic ion, the decrease in nitrate concentration did not significantly contribute to the pH increases, consistent with findings from Ding et al. (2019) and Zhang et al. (2021). This is primarily because NO₃ declined more slowly compared to sulfate ions and exceeded sulfate concentrations after 2016, under which conditions that nitrate-rich particles can absorb twice the amount of water that sulfate-rich particles, leading to an increase in ALWC concentration and inhibiting pH decline (Lin et al., 2020; Xie et al., 2020). On the other hand, increases in particle pH in 2015 and 2018 were notably influenced by changes in TNH_x with concentrations increased by 5.5 and 1.3 μg/m³, respectively. Increased TNH_x concentrations could react with SO₄²⁻ or NO₃⁻ and consume a substantial amount of H⁺, thereby raising particulate matter pH values (Seinfeld et al., 1998; Zhang et al., 2021). Substantial decreases in T in 2015 (4.2°C), 2017 (4.9°C), and 2018 (2.8°C), favoring NH₃ partitioning into the particle phase and reducing H⁺ concentrations, drove increases in particle pH (Tao and Murphy, 2019).

During the period from 2020 to 2022, the influence of SO_4^{2-} on particle pH gradually decreased, with a decrease in concentration from 0.3 to 2.3 μ g/m³ (Table S6) only bringing about a pH decrease of 0.03 to 0.14 (Fig. 5). Moreover, a rebound in SO_4^{2-} concentration to $7.9 \pm 4.5 \,\mu$ g/m³ in 2022 even resulted in a decrease of 0.11 units in pH instead. On the other hand, TNHx began to show a slight annual decline (0.9 to 2.2 μ g/m³), resulting in a significant decrease in pH (0.21–0.35). Consequently, the increase in pH values was closely related to the rise in Ca^{2+} concentration. Ca^{2+} is less volatile and competes preferentially with NH₃ to neutralize anions such as SO_4^{2-} to form insoluble CaSO₄, which precipitates from the aerosol aqueous phase (Ding et al., 2019; Karydis et al., 2021), thereby reducing H⁺ concentrations (Fig. S11) and subsequently lowering particle acidity. Specifically, increases of 0.7 and 0.5 μ g/m³ in Ca^{2+} concentrations led to pH increases of 0.13 and 0.09 units in 2020 and 2022, respectively, making Ca^{2+} a primary controlling factor for pH elevation.

4 Conclusions

The annual average $PM_{2.5}$ concentration in Zhengzhou decreased from $212.4 \pm 101.5 \ \mu g/m^3$ in 2013 to $59.5 \pm 41.2 \ \mu g/m^3$ in 2022, with the largest reduction in SO_4^{2-} . As for CM, their concentrations notably decreased from 2013 to 2019, because of effective dust control measures, as well as decreased wind speed and increased relative humidity. However, the proportions of CM in $PM_{2.5}$ have shown

consecutive annual increases. In addition, CM concentrations and their proportions increased significantly during 2019–2022, which might be responsible for the resuspension of surrounding soil dust. Correspondingly, the annual pH increased by 0.11 units from 2013 to 2019 mainly due to the decline in SO_4^{2-} , increased TNH_x, or decreased temperature. During the period from 2020 to 2022, the annual average growth rate of pH values increased to 0.21 units from 2019 to 2022, which was determined by the rise in Ca^{2+} concentration.

5 Implication

Control measures implemented by the Chinese government have proven effective in reducing dust, but this study reveals that the crustal materials in PM_{2.5} rebounded after 2019. This phenomenon is not unique to Zhengzhou, as major cities in the North China Plain have also experienced a pronounced upward trend in coarse particles after 2019 (Fig. S12). Thus, crustal materials persist as a substantial constituent of atmospheric aerosols in North China, sustaining elevated particle pH levels. Extensive research has established that heightened particle pH inhibits nitrate reduction in aerosols (Ding et al., 2019; Lin et al., 2020; Wen et al., 2018), particularly significant given nitrate's predominant role in haze formation within this region. Notably, while moderately acidic aerosols demonstrate reduced health impacts, particles with pH < 3 exhibit substantially greater health risks (Shi et al., 2019). Consequently, future environmental management strategies must prioritize real-time assessment of regulatory impacts on particle acidity. This necessitates an integrated approach that simultaneously addresses acidic precursors, alkaline precursors, and crustal material contributions to atmospheric acid chemistry.

Data availability

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397 All the data presented in this article can be accessed through 398 https://doi.org/10.5281/zenodo.14032007 (Zhang, 2024).

Supporting Information

Additional data, figures, and tables, some of which are referenced directly within the manuscript, and detailed descriptions of field measurements and samples.

Author contributions

Shenbo Wang designed this study. Hongyu Zhang and Zhangsen Dong analyzed the data and prepared the manuscript with the contributions of all coauthors. Xiao Li conducted measurements. Ruiqin Zhang provided funding acquisition. All authors have read and agreed to the published version of the manuscript.

Competing interests

The authors declare that they have no conflict of interest.

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636 Figures

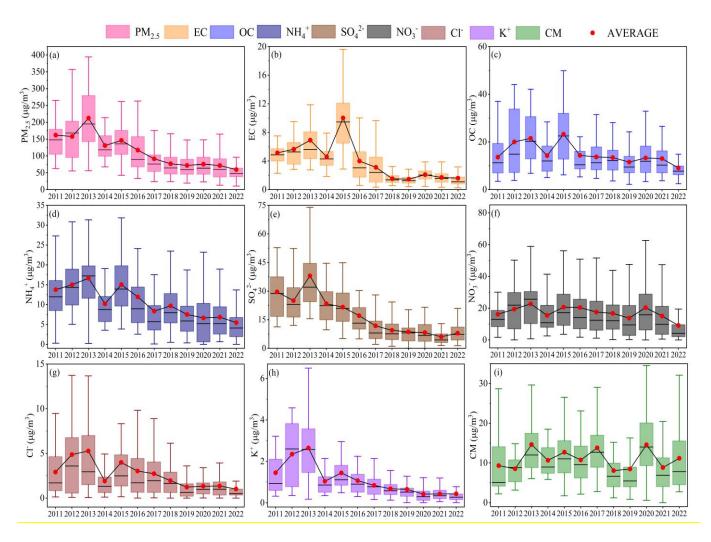


Figure 1. Long-term trends in the concentrations of PM_{2.5} and its chemical components in from 2011 to 2022 in Zhengzhou. Box plots depict annual averages (red dots) and medians (black lines), the top, middle, and bottom lines represent the 75, 50, and 25 percentiles of statistical data, respectively, and the upper and lower whiskers represent the 90 and 10 percentiles of statistical data, respectively.

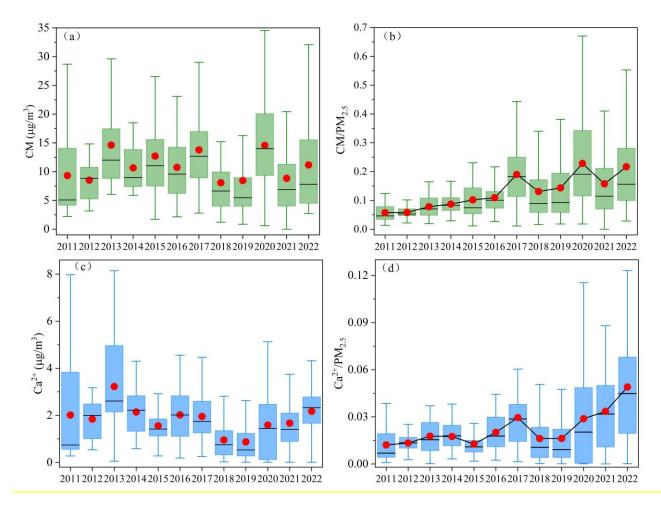


Figure 2. (a) and (c) Long-term trends in CM and Ca²⁺ concentrations in Zhengzhou from 2011 to 2022, respectively. Box plots depict annual averages (red dots) and medians (black lines). (b) and (d) Long-term trends in the proportions of CM and Ca²⁺ in PM_{2.5}, respectively.

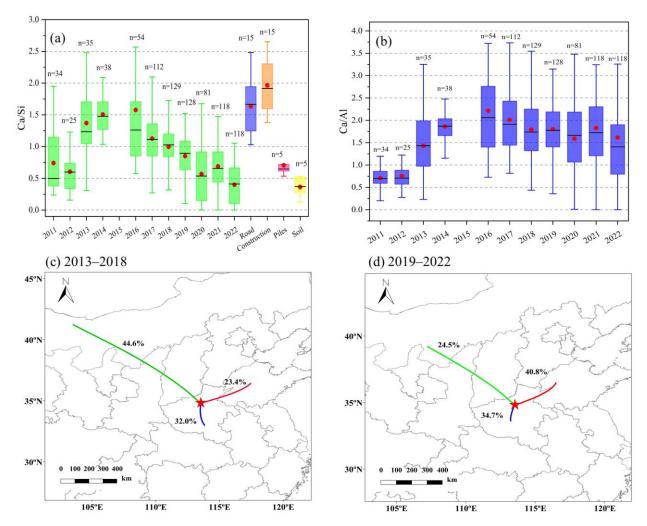


Figure 3. (a) The annual Ca/Si ratios in Zhengzhou from 2011 to 2022 compared with those in various dust sources (specific values and references in Table S5). The red dots and black lines in the box plots represent the annual averages and medians, respectively, with n indicating the sample size. (b) The Ca/Al ratios in Zhengzhou from 2011 to 2022. The red dots and black lines in the box plots represent the annual averages and medians, respectively, with n indicating the sample size. (c) and (d) The transport pathways of CM during 2013–2018 and 2019–2022, respectively.

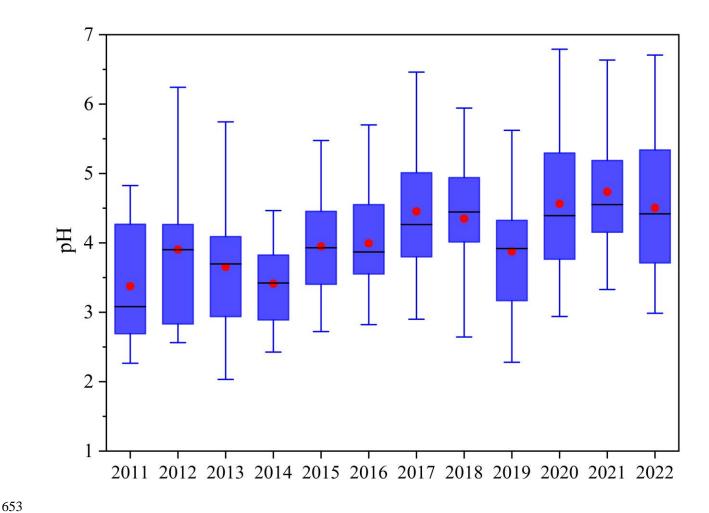


Figure 4. The time series of particle pH in Zhengzhou from 2011 to 2022. In the boxplots, red dots and black lines represent the annual mean and median values, respectively.

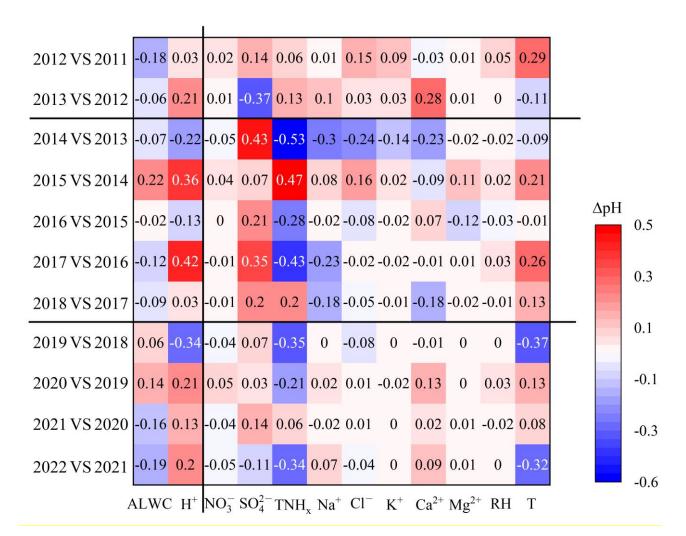


Figure 5. Contribution of each component to the changes in pH (Δ pH) between adjacent years. The difference between component concentrations and meteorological parameters between adjacent years is listed in Table S6.

Table

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Table 1. Annual average concentrations of PM_{2.5} and its components from 2011 to 2022 in Zhengzhou, China (μg/m³).

Years	PM _{2.5}	EC	OC	NO_3^-	SO ₄ ²⁻	$\mathrm{NH_4}^+$	CM	Ca ²⁺
2011	161.9 ± 81.4	5.1 ± 2.1	13.6 ± 8.6	16.2 ± 11.2	29.6 ± 14.3	13.8 ± 8.3	9.3 ± 7.3	2.0 ± 2.2
2012	157.9 ± 71.2	5.6 ± 2.5	20.0 ± 13.4	20.2 ± 13.7	25.0 ± 11.2	15.0 ± 7.1	8.5 ± 3.4	1.8 ± 0.8
2013	212.4 ± 101.5	6.9 ± 3.8	21.5 ± 10.4	22.7 ± 13.2	38.0 ± 19.9	17.1 ± 6.9	14.6 ± 8.3	3.2 ± 2.1
2014	130.8 ± 48.7	4.6 ± 2.0	14.2 ± 8.2	15.5 ± 10.8	23.4 ± 9.3	10.2 ± 6.2	10.7 ± 4.4	2.1 ± 1.0
2015	146.1 ± 61.0	10.0 ± 4.7	23.2 ± 11.6	20.6 ± 14.5	21.6 ± 9.8	15.7 ± 7.5	12.7 ± 6.8	1.6 ± 0.7
2016	117.4 ± 73.5	4.0 ± 2.8	14.4 ± 10.0	20.4 ± 18.7	17.1 ± 11.3	11.9 ± 10.6	10.8 ± 5.3	2.0 ± 1.1
2017	91.5 ± 61.1	3.1 ± 2.5	13.7 ± 7.5	17.6 ± 15.9	11.8 ± 11.6	8.4 ± 7.9	13.8 ± 6.5	2.0 ± 1.0
2018	76.8 ± 41.6	1.5 ± 0.7	13.4 ± 7.3	16.7 ± 13.5	9.4 ± 6.0	9.7 ± 6.1	8.1 ± 5.7	1.0 ± 0.8
2019	68.4 ± 34.8	1.5 ± 0.8	11.5 ± 6.8	13.8 ± 13.9	8.6 ± 6.4	7.5 ± 6.1	8.5 ± 7.8	0.9 ± 0.9
2020	75.5 ± 31.8	2.1 ± 0.9	13.3 ± 7.9	18.6 ± 14.2	8.3 ± 5.6	6.7 ± 6.6	14.6 ± 7.6	1.6 ± 1.4
2021	71.5 ± 45.9	1.7 ± 0.9	13.0 ± 8.0	15.1 ± 15.1	6.1 ± 4.5	6.8 ± 6.0	8.9 ± 7.0	1.7 ± 1.2
2022	59.5 ± 41.1	1.6 ± 1.5	9.1 ± 8.1	10.0 ± 14.4	7.9 ± 4.5	5.5 ± 5.4	11.2 ± 8.3	2.2 ± 1.1

Supplement materials:

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Text S1 Instruments and Measurements.

Samples were collected using a high-volume sampler (TE-6070D, Tisch, USA) and air particulate samplers (TH-16A, Tianhong, China) from April 2011 to December 2022. Two quartz filters and two Teflon filters were used daily from 10:00 AM to 9:00 AM the next day, resulting in a total of 5848 samples. After excluding abnormal data due to instrument malfunctions, 4228 valid samples were obtained. Detailed information on the samples is provided in Table S1. Organic carbon (OC) and elemental carbon (EC) were analyzed using a carbon analyzer (Model 5L, Sunset Laboratory, USA). The analysis of EC and OC was conducted in two stages. In the first stage, the filter membrane was placed in a quartz heating furnace under a pure helium atmosphere. As the temperature gradually increased to approximately 580°C, OC was volatilized and released. In the second stage, heating continued in a mixed atmosphere of 2% oxygen and 98% helium. When the temperature reached approximately 870°C, EC underwent oxidative decomposition and was released. During the helium flow transmission, OC and EC released at different temperatures were completely oxidized to CO₂ in a MnO₂ oxidation furnace and subsequently reduced to CH₄ for detection by a flame ionization detector (FID). The detection limits for both OC and EC were 0.2 µg/cm². Before each sample analysis, calibration was performed using a standard sucrose solution. Additionally, parallel tests were conducted every ten samples to ensure accuracy. Water-soluble inorganic ions (Cl⁻, NO₃⁻, SO₄²⁻, Na⁺, NH₄⁺, K⁺, Mg²⁺, and Ca²⁺) were measured using ion chromatography (ICS-90 and ICS-900 models, Dionex, USA). Half of the PM_{2.5} filter was cut into pieces and ultrasonically extracted with 20 mL of Milli-Q water for 30 min, followed by filtering through a 0.45 mm polytetrafluoroethylene syringe filter before analysis. The cation concentrations were determined by an IonPacASII-HC4 mm anion separation column and an IonPacAGII-HC4 mm guard column, with 20 mM methane sulfonate as an eluent at 0.8 mL/min. The anions were measured by an IonPacCS12A cation separation column and an IonPacCG12A guard column, with a solution of 8.0 mM Na₂CO₃ + 1.0 mM NaHCO₃ as an eluent at 1.0 mL/min. The regression coefficients (R²) of the calibration curves were over 0.9996 for all ions, except NH₄⁺ (0.9988), which showed a quadratic response.

Elements were analyzed using a wavelength dispersive X-ray fluorescence spectrometer (S8 TIGER, Bruker, Germany) to determine concentrations of Fe, Na, Mg, Al, Si, Cl, K, Ca, V, Ni, Cu, Zn, Cr, Mn, Co, Cd, Ga, As, Se, Sr, Sn, Sb, Ba, and Pb (Tremper et al., 2018), which has been approved by the United States Environmental Protection Agency (Chow and Watson, 1994). The spectrometer was equipped with an X-ray tube featuring close coupling among the tube, sample, and detector, ensuring high efficiency and optimal excitation of elements within the sample. Before analysis, the instrument was calibrated using a series of high-quality, self-prepared standards. Calibration procedures were conducted following established methods (Chow and Watson, 1994). To assess potential contamination and ensure data quality, blank filters were routinely analyzed alongside each batch of samples.

Figures

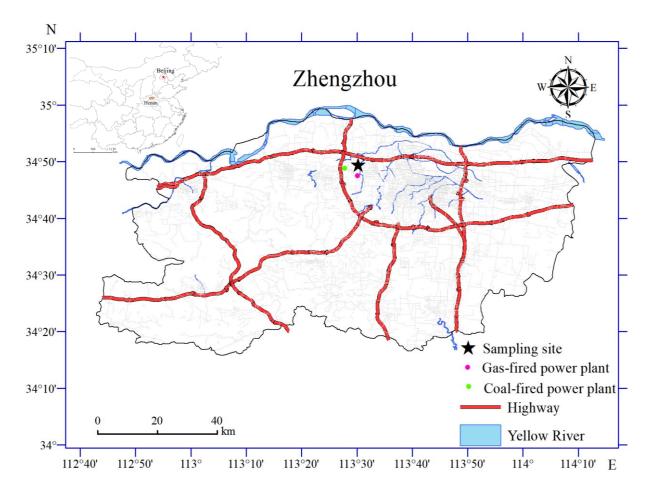


Figure S1.Sampling site in Zhengzhou, China. © 2019 National Geomatics Center of China. All rights reserved.

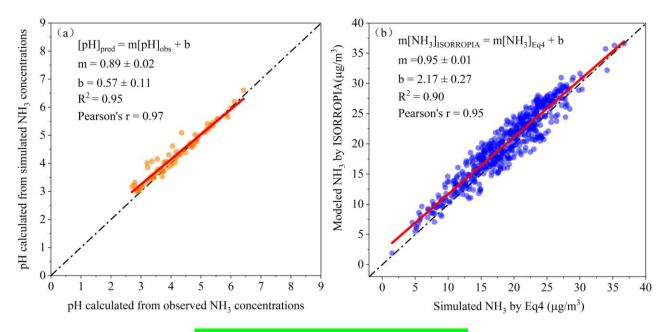


Figure S2. Verification of the NH₃ Method.

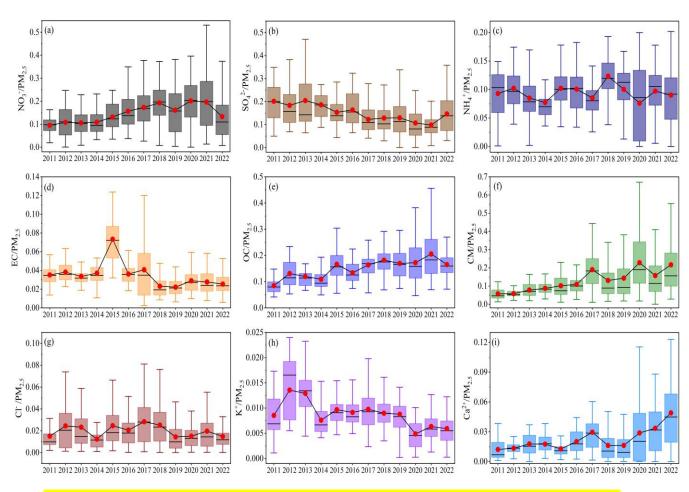


Figure S3. Trends in the proportions of chemical components in PM_{2.5} from 2011 to 2022.

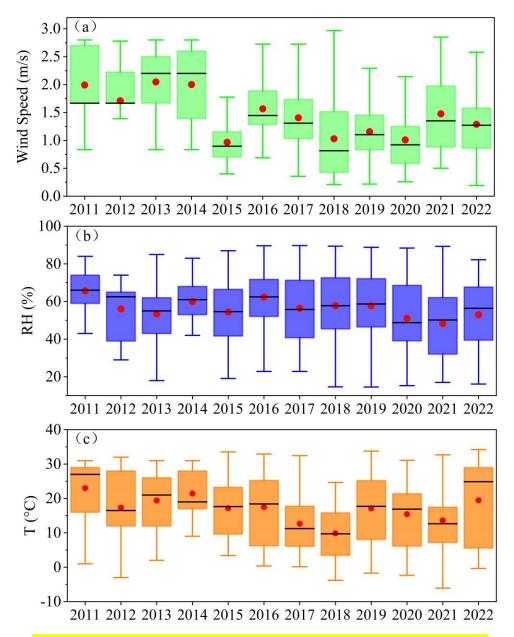


Figure S4. Trends in the meteorological parameters from 2011 to 2022.

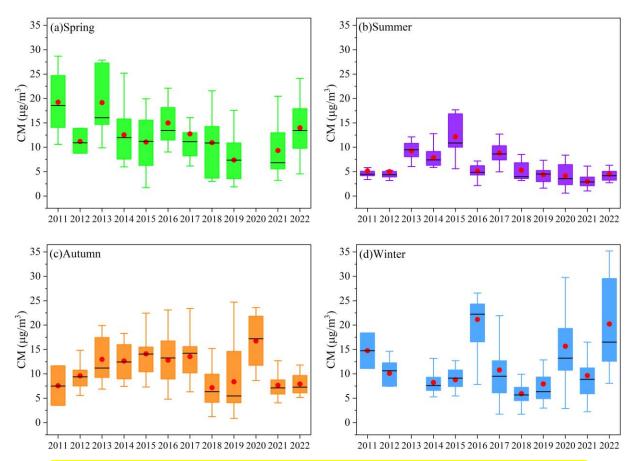


Figure S5. Trends in the CM concentrations in different seasons from 2011 to 2022.

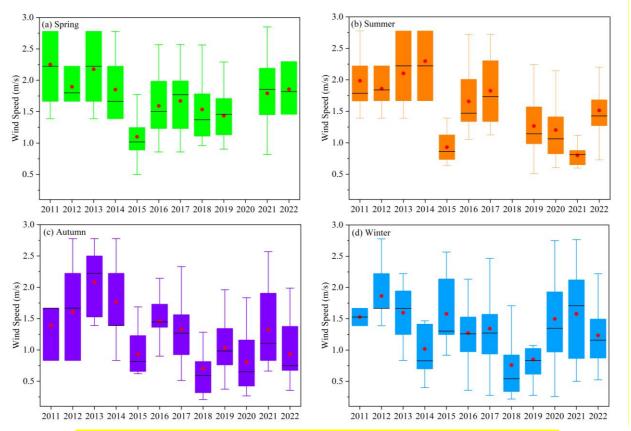


Figure S6. The variation in WS across different seasons from 2011 to 2022.

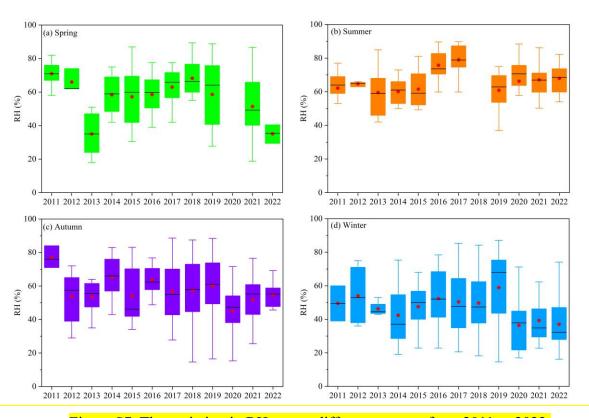


Figure S7. The variation in RH across different seasons from 2011 to 2022.

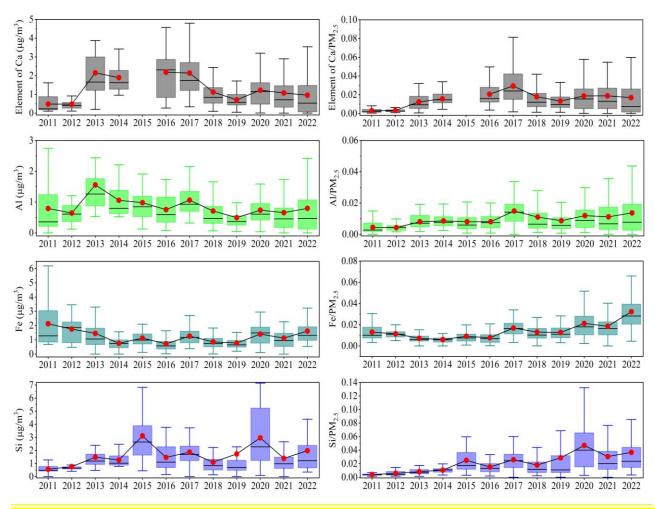


Figure S8. Trends in the concentrations of crustal elements and their proportions in PM_{2.5} from 2011 to 2022.

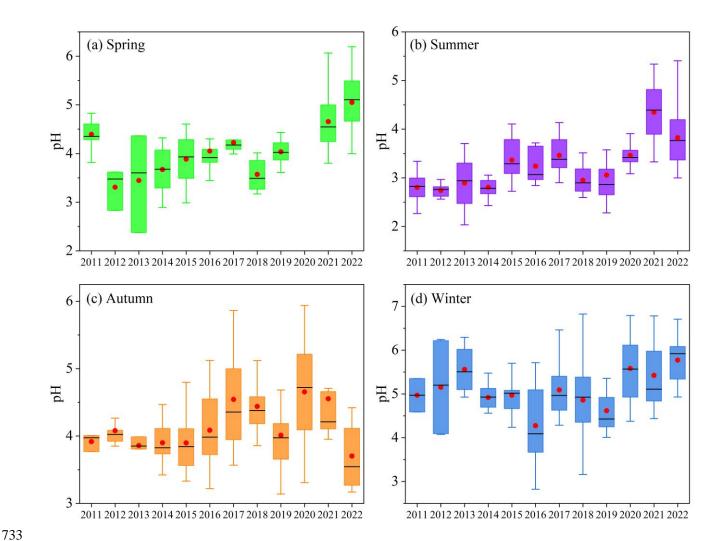


Figure S9. Trends in the particle pH in different seasons from 2011 to 2022.

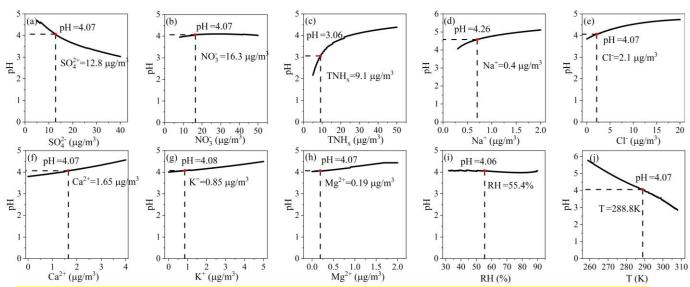


Figure S10. Sensitivity analysis of input parameters to particle pH. The dashed line represents the average of the observational data from 2011 to 2022.

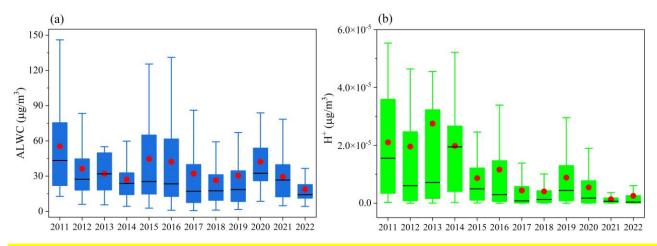


Figure S11. Trends in aerosol liquid water content (ALWC) and H⁺ concentrations from 2011 to 2022

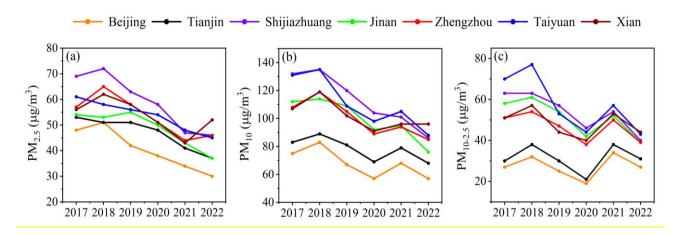


Figure S12. Trends in the annual average concentrations of PM_{2.5}, PM₁₀, and PM_{10-2.5} in provincial capitals in the North China Plain.

Tables

Table S1. Information on sampling date and numbers.

Years	Sampling date	The	effective	number	of
		samp	oles		
	April 7–20				
2011	July 1–31	188			
	October 28–December 2				
	December 11-November 23				
	February 25–26				
2012	April 21–May 6	140			
	July 22–August 2				
	October 17–November 1				
	December 8–25				
	February 25–March 6				
2013	April 1–May 1	184			
	June 5–July 30				
	September 20-October 13				
	December 2–18				
	April 1–May 5				
2014	June 18–July 20	180			
	October 7–24				
	December 30–31				
	January 1–15				
2015	April 1–20	248			
	July 1–20				
	October 9–24				
	January 6–22				
2016	April 8–30	252			
	July 9–31				
	October 1–20				
	December 29–31				
	January 1–20				
2017	April 18–May 4	480			
	July 1–26				
	October 14-December 31				
	January 1–31				
2018	April 1–30	600			
	July 1–31				
	October 9–December 31				

	January 1–31	
2019	April 1–30	592
	July 1–31	
	September 2-October 31	
	November 12–30	
	December 21–31	
	January 1–20	
2020	June 5–July 31	332
	October 6-November 13	
	December 15–31	
	January 1–31	
2021	March 16-April 30	540
	July 1–August 8	
	October 17–December 31	
	January 1–4	
2022	April 1–May 3	492
	July 1–August 11	
	September 5-October 11	
	December 10-31	
Total		4228

Table S2. The method detection limit (MDL) and measurement uncertainties (Unc) of individual

components	
MDLs $(\mu g/m^3)$	Unc (%)
0.1	13.1
0.1	<mark>9.8</mark>
0.005	<mark>9.6</mark>
0.011	10.1
0.006	<mark>9.5</mark>
0.002	9.3
0.017	8.8
0.001	8.2
0.001	<mark>9.3</mark>
0.015	10.1
0.031	<mark>9.9</mark>
0.003	10.9
0.002	10.6
0.004	<mark>9.2</mark>
0.005	9.3
	MDLs (μg/m³) 0.1 0.1 0.005 0.011 0.006 0.002 0.017 0.001 0.001 0.001 0.0015 0.031 0.003 0.002 0.004

<mark>C1</mark>	0.008	<mark>9.5</mark>
C1 K	0.005	9.5 9.4 9.4 57.9
<mark>Ca</mark> <mark>V</mark>	0.01	<mark>9.4</mark>
V	0.008	<mark>57.9</mark>
<mark>Ni</mark>	0.006	<mark>96.6</mark>
<mark>Cr</mark>	0.02	24.7
Ni Cr Mn	0.02	16.8
<mark>Fe</mark> <mark>Co</mark> Cu <mark>Zn</mark>	0.03	16.8 9.3
Co	0.009	<mark>79.6</mark>
Cu	0.005	5.8
Zn	0.003	79.6 5.8 8.4
<mark>Ga</mark>	0.005	84.7
<mark>As</mark>	0.008	27.4
<mark>Se</mark>	0.006	25.7
<mark>Sr</mark>	0.006	22.8
As Se Sr Cd Sn Sb	0.03	<mark>68.6</mark>
<mark>Sn</mark>	0.02	42.0
	0.02	73.6
<mark>Ba</mark>	0.02	15.6
Pb Pb	0.02	13.4

Table S3. Control measures for dust implemented by Henan Province and Zhengzhou government

Release time	Policies	Regulatory focus
2013.9	Regulations on Reducing Pollutant Emissions in Henan Province	Road, Construction
2014.8	Temporary Regulations on Dust Control Management at Construction Sites in Henan Province	Construction
2016.7	Implementation Plan for Controlling Dust Pollution in Henan Province	Road, Construction
2018.2	Regulations on the Prevention and Control of Atmospheric Pollution in Henan Province	Road, Construction, Piles
2019.4	Special Action Plan for Fine Management of Dust Pollution Prevention and Control at Construction Sites in Zhengzhou City, 2019	Construction

2019.8	Enhanced Action Plan for Intensive Dust Control at Construction Sites in 2019	Construction
2021.1	Special Governance Plan for Key Project Dust Pollution in Zhengzhou	Road, Construction, Piles

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Table S4. Analysis of the inter - annual trends of CM and Ca²⁺ concentrations and pH during different periods using multiple methods.

	2011–2013			2013–2019			2019–2022		
	CM	Ca^{2+}	<mark>рН</mark>	- CM	Ca ²⁺	pН	CM	Ca ²⁺	pH
MK-Z	3.01	2.70	1.41	- 9.74	-13.62	3.00	2.48	8.21	5.12
MK-p	0.003	0.007	0.159	<2.2 E–16	<2.2E-16	0.003	0.013	2.20E-16	2.99E-07
Sen's slope	0.082	0.023	7.10E-03	-0.015	-4.14E-03	9.15E-04	5.80E-03	5.42E-03	2.93E-03
LS slope	2.65	0.61	<mark>/</mark>	-0.81	-0.32	0.11	0.24	0.40	0.21

^{**} MK-Z and MK-p represent the trend (Z) and significance (p) calculated by the Mann - Kendall method using daily data, respectively; Sen's

slope represents the Sen slope using daily data; LS slope represents the Least - Squares slope using annual data. All the above calculations were

performed using the R language (R version 4.0.2).

Dust source	City	Ca/Si	Reference
Road dust	Xi'an	2.04	http://www.klacp.ac.cn/
	Yinchuan	2.48	wgPMzypfypk/ycy/2017
	Lanzhou	1.67	06/t20170610_375562.ht
	Beijing	1.25	ml
	Tianjin	1.03	
	Baoding	1.16	
	Shijiazhuang	1.98	
	Handan	1.83	
	Shenyang	1.81	
	Changsha	1.92	
	Chongqing	1.38	
	Chengdu	1.17	
	Kunming	1.94	
	Taiyuan	1.55	
	Nanjing	1.28	
Construction dust	Xi'an	1.69	http://www.klacp.ac.cn/
	Yinchuan	1.84	wgPMzypfypk/ycy/2017
	Lanzhou	2.33	06/t20170610_375562.h
	Beijing	2.65	ml
	Tianjin	1.46	
	Baoding	1.58	
	Shijiazhuang	1.38	
	Handan	1.86	
	Shenyang	1.92	
	Changsha	2.30	
	Chongqing	2.52	
	Chengdu	2.15	
	Kunming	1.60	
	Taiyuan	1.92	
	Nanjing	2.26	
Piles dust	Xi'an	0.72	(Yang, 2016)
	Tianjin	0.57	(Zhang et al., 2018)
	Taiyuan	0.61	(Bi et al., 2007)
	Jinan	1.01	(Bi et al., 2007)
	1	0.65	http://www.nkspap.com:
		-	9091/Index.aspx
Soil dust	Nanchang	0.37	(Xu et al., 2019)
.	Xi'an	0.27	(Yang, 2016)
	Jincheng	0.13	(Wang et al., 2016)
	Wuhan	0.52	(Gong and Luo, 201

		8)
/	0.53	http://www.nkspap.com:
		9091/Index.aspx

Table S6. The difference between component concentrations (μg/m³) and meteorological parameters between adjacent years. 50_{4}^{2-} $\overline{\text{TNH}_{\text{X}}}$ C1⁻ K^+ Ca^{2+} Mg^{2+} **Years ALWC** $H^{+}(10^{-6})$ $NO_3^ Na^+$ **RH(%)** $T(^{\circ}C)$ -0.2 2012VS2011 <mark>-19.0</mark> -1.5 **4.0** <mark>-4.6</mark> 1.3 0.022.0 <mark>0.9</mark> 0.04 <mark>-9.6</mark> **-5.7 -4.6 -2.6** 2.1 2013VS2012 -7.6 2.6 13.0 2.1 0.2 <mark>0.4</mark> 0.3 **1.4** 0.1 2014VS2013 **-4.5** 7.9 -7.3 -14.6 **-6.9** -0.4 -3.4 -1.6 -1.1 -0.2 <mark>6.6</mark> 2.0 2015VS2014 **5.2** -1.8 **5.5** 0.1 2.1 0.4 -0.6 0.6 -5.6 -4.2 **17.6** -11.2 2016VS2015 **-2.3** 3.0 -0.2 **-4.5** -3.7 -0.03 -0.1 -0.4 0.5 **-0.7** 8.0 0.3 -7.2 **-2.9 -5.3** -3.6 -0.2 -0.3 -0.2 **-0.1** <mark>-6.0</mark> <mark>-4.9</mark> 2017VS2016 **-10.0** 0.1 2018VS2017 **-5.8** -0.3 **-0.8 -2.4** 1.3 -0.1 **-0.8** -0.2 **-0.1 -0.1** 1.4 **-2.8** 4.1 4.8 -3.0 **-0.8 -2.2 -0.7 -0.1** 7.3 2019VS2018 -0.04 -0.03 **-0.1 -0.01** -3.4 <mark>4.9</mark> **-0.3** <mark>-0.9</mark> 0.1 0.1 **-0.2** 0.7 <mark>-6.6</mark> **-2.1** 2020VS2019 11.6 0.022021VS2020 -12.7 **-1.5** -3.6 **-2.3** 0.2 **-0.01** 0.03 0.01 0.1 0.04**-2.8 -1.5**

-1.5

-10.6

2022VS2021

1.9

<mark>-1.4</mark>

-5.1

0.03

-0.3

0.01

0.5

<mark>4.7</mark>

0.04

5.8

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