Cr(VI) reduction, electricity production, and microbial resistance variation in

paddy soil under microbial fuel cell operation

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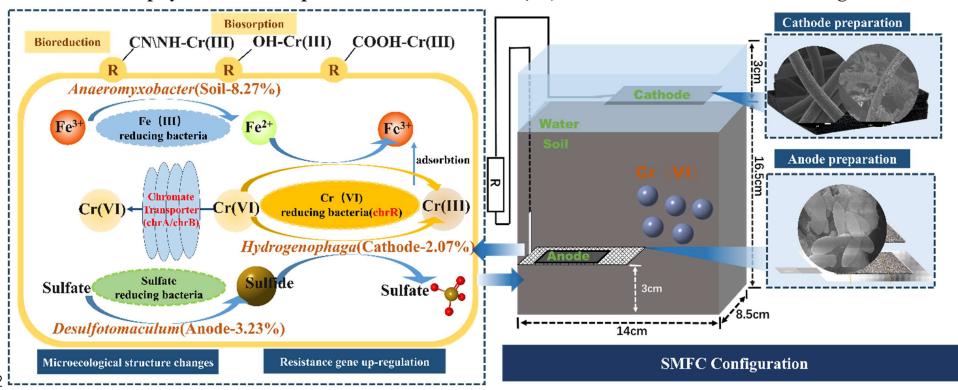
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# 1 Abstract figure:

# "physio-bio" adsorption and reduction of Cr(VI) under bioelectrochemical driving



<b>Abstract:</b> Microbial fuel cell (MFC) is an efficient in-situ approach to combat pollutants and generate
electricity. This study constructed a soil MFC (SMFC) to reduce Cr(VI) in paddy soil and investigate its
influence on microbial community and microbial resistance characteristics. Fe <sub>3</sub> O <sub>4</sub> nanoparticle as the
cathodic catalyst effectively boosted power generation (0.97 V, 102.0 mW/m²), whose porous structure
and reducibility also contributed to Cr reduction and immobilization. After 30 days, 93.67% of Cr(VI)
was eliminated. The bioavailable Cr decreased by 97.44% while the residual form increased by 88.89%.
SMFC operation greatly changed soil enzymatic activity and microbial structure, with exoelectrogens
like Desulfotomaculum (3.32% in anode) and Cr(VI)-reducing bacteria like Hydrogenophaga (2.07% in
cathode) more than 1000 folds of soil. In particular, SMFC operation significantly enhanced heavy metal
resistance genes (HRGs) abundance. Among them, <i>chrA</i> , <i>chrB</i> , <i>and chrR</i> increased by 99.54~3314.34%
in SMFC anode, probably attributed to the enrichment of potential tolerators like Acinetobacter,
Limnohabitans, and Desulfotomaculum. These key taxa were positively correlated with HRGs but
negatively correlated with pH, EC, and Cr(VI), which could have driven Cr(VI) reduction. This study
provided novel evidence for bioelectrochemical system application in contaminated paddy soil, which
could be a potential approach for environmental remediation and detoxification.

Keywords: Chromium; Microbial fuel cell; Microbial response; Metal resistance

Nomenclature	
SMFC	Soil microbial fuel cell
HRGs	Heavy metal resistance genes
HMs	Heavy metals
HGT	Horizontal gene transfer
EAB	Electrochemical active bacteria
GF	Graphite felt
ORR	Oxygen reduction reaction
WCV	Working circuit voltage
OCV	Open circuit voltage
ARGs	Antibiotic resistance genes
CMFC-A	Anode of the closed circuit group
OMFC-A	Anode of the open circuit group
CMFC-C	Cathode of the closed circuit group
OMFC-C	Cathode of the open circuit group
CMFC-S	Soil of the closed circuit group
OMFC-S	Soil of the open circuit group

Soil of the non-electrode control group

### 1. Introduction

NMFC-S

Chromium (Cr) is a main toxic heavy metal (HM), that enters the environment mainly due to its wide use in electroplating, tanning, and other industries (Coetzee et al., 2020). Mineral-sourced phosphate fertilizer also contains high-level Cr, promoting its spreading, migration and accumulation in soil and underground water (Chen et al., 2022a). The persistence and toxicity of HMs create permanent and selective stress on environmental microbes. Under the toxic stress, microorganisms have deployed various strategies, including active efflux of the toxic from the microbial cell, regulation of targets, and enzymatic modification of the toxic, presenting as elevation of heavy metal resistance gene (HRG) and tolerant microbe (Van Hoek et al., 2011). Even a sub-dose of Cr (especially the Cr(VI) state) can promote plasmid-mediated horizontal gene transfer (HGT) (Zhang et al., 2018), causing HRG enrichment, threatening environmental safety (Guo et al., 2021; Wang et al., 2020a; Wang et al., 2023b). Meanwhile, due to the co-selection effect, the long-term existence of HMs also causes the enrichment of antibiotic-

resistant bacteria (ARB), further increasing the resistance gene spreading risk in the environment (Men et al., 2018). Such a threat could be more fierce under the coexistence of HMs and antibitics (Ashbolt Nicholas et al., 2013). Hence, the enrichment of HRGs and ARB under Cr exposure has become an emerging concern.

Common remediation methods for Cr-influenced soil include chemical reduction and leaching (Cong et al., 2022), electrokinetic remediation (Morales-Benítez et al., 2023), and phytoaccumulation (Yaashikaa et al., 2022), which convert Cr(VI) into insoluble and low toxic forms (e.g., Cr(III)) by adsorption, ion exchange, and redox (Rani et al., 2022). Among them, the microbial approach using functional microbes is commonly used for the continuous treatment of soil-groundwater, which has a low cost without side effects (Fan et al., 2023). For example, *Comamonas testosteroni* bacterial strains are used to degrade hexachlorobenzene and *Acinetobacter calcoaceticus GSN3* strain to degrade phenol (Dimova et al., 2022; Irankhah et al., 2019). However, pollutants can be tightly adsorbed by soil particles and persistently remain (Wang et al., 2023c). The complex soil constituents and competition of indigenous microorganisms inhibit the colonization and development of functional microbes and limit their effectiveness (Guo et al., 2021).

Microbial fuel cell (MFC) technology can transform or immobilize HMs and generate electricity utilizing electrochemical active bacteria (EABs) (Gupta et al., 2023; Chen et al., 2022b), which have been used in sediment or soil to treat HMs and organics and monitor environmental toxicity (Li et al., 2023b). For example, soil microbial fuel cells can mitigate the accumulation of heavy metals in rice and promote the removal of atrazine in the soil (Gustave et al., 2020; Farkas et al., 2024). At present, soil MFC (SMFC) has been used for pollution control, focusing on pollutant content and forms as well as the

electrochemical properties (Hamdan and Salam, 2023; Liu et al., 2023a). There is a lack of systematic research about the MFC effect on soil microbial community structure shifting and resistance characteristics, especially under the HMs contamination circumstance.

In this study, an SMFC was constructed to remediate Cr(VI) contaminated paddy soil. EABs were pre-loaded on the SMFC anode to promote electricity production and Cr transformation. Ferroferric oxide (Fe<sub>3</sub>O<sub>4</sub>) nanoparticles, which can reduce and fix Cr(VI) directly, were used as a catalyst for cathodic oxygen reduction reaction (ORR) to improve SMFC performance due to its catalytic effect, non-toxicity, cost-effectiveness, and free of secondary pollution (Liu et al., 2023b). During operation, Cr(VI) was simultaneously reduced and immobilized by bio-physical adsorption and electrochemical-microbial reduction. The Cr(VI) reduction mechanism and the analysis of microbial community structure shaping and HRG variation were comprehensively studied. For the first time, SMFC-driven Cr(VI) reduction was associated with microbial resistance, which evolved along with microbial adaptation and development. This study not only provides a reference for the microbial remediation of polluted soil but also improves the practical field application of MFC. The method can be used for in-situ contaminant treatment in varie environments, such as aquaculture ponds, inland lakes, and wetlands.

### 2. Materials and Methods

# 2.1. Chemicals

- All the chemicals and reagents were analytical grade or premium pure from Kelong Chemical Reagent
- 71 Factory, Chengdu, China.

### 2.2 Construction of SMFC

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73 2.2.1 Soil 74 Paddy soil from Jintang County, Chengdu, China (30°74' N, 104°59' W) was collected and used to 75 construct SMFC. The soil has organic matter, organic carbon, and total nitrogen of 8.84±0.02%, 76 1.74±0.01%, and 321.67±1.25 mg/kg, respectively. Referring to the previous studies and the Chinese 77 Soil Environmental Quality Standard (GB15618-2018), 120.0 mg/kg potassium dichromate was added 78 to the soil and a final Cr(VI) concentration of 118.8 mg/kg was achieved before use (Liu et al., 2020; 79 Mandal et al., 2017; Li et al., 2024). 80 2.2.2 Electrodes Preparation 81 Aluminum foam (66.0×54.0×5.0 mm, porosity 60-80%, bulk density 0.50-1.10 g/cm<sup>3</sup>) (SANZHENG 82 Metal material, Chengdu, China) was used as anode. The anode microflora was derived from municipal 83 sludge (Chengdu Sixth Sewage Treatment Plant, China) after acclimating with 100 mg/L Cr(VI). Before 84 assembling, the aluminum foam was cultivated in the anode microflora for 2 weeks. Then the anode was 85 tied to titanium mesh tightly with titanium wire. Graphite felt (GF) (100.0×50.0×3.0 mm, bulk density 86 0.10-0.15 g/cm<sup>3</sup>) was used as the cathode (Table S1). Before use, it was cleaned, dried, and loaded with 87 Fe<sub>3</sub>O<sub>4</sub> as the ORR catalyst, as detailed in section 1 of the supplementary material. For characterization, 88 we utilized a scanning electron microscope (SEM) to examine the structure and morphology of the 89 electrode surface. In addition, we performed X-ray photoelectron spectroscopy (XPS) and energy 90 dispersive spectroscopy (EDS) to analyze the valence state and element composition. The phase

composition was determined using an X-ray diffractometer (XRD).

#### 2.2.3 SMFC construction

A plastic box (140.0×85.0×165.0 mm) was used as the SMFC reactor, with 1.50 kg air-dried soil and overlying water of 3.0 cm to simulate the flooded state during rice planting. The cathode was floated on the water surface while the anode was buried (about 3.0 cm from the bottom). The cathode and anode were connected to a 2000  $\Omega$  resistor using titanium wire. The water level was kept constant by daily replenishment (Fig.S1).

### 2.3 Design and Operation

Three treatments were set up as shown in Fig. S1, and three parallels were set for each group. NMFC: The control group with no electrode, only contains an equal amount of overlying water and paddy soil. OMFC: The open circuit group with disconnected electrodes and equal amounts of overlying water and paddy soil. CMFC: The complete closed-circuit SMFC capable of producing electricity, with electrodes connected by a  $2000\Omega$  resistor, and equal amounts of overlying water and paddy soil.

The experiments were conducted at 25°C. A Raspberry Pi data acquisition system (ARMv7 architecture) was connected at both ends of the resistor of CMFC to monitor the voltage. A multimeter was used for verification. The detailed code information can be found in section 2 of the supplementary material.

Soil and water samples were taken every 5 days until day 35, and the operation continued for another 10 days until day 45 (Specific sampling details are given in the supplementary file). The electrochemical properties of the SMFC including the polarization curve and power density curve were tested using an electrochemical workstation on days 15 and 30 (Ch660e, Shanghai Chenhua Instrument Co., Ltd., Shanghai, China) (Chen et al., 2022b).

### 2.4 Cr Migration and Transformation

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To determine total Cr, 0.50 g soil was subjected to acid digestion (HCl-HNO<sub>3</sub>-HClO<sub>4</sub>) before measurement using flame atomic absorption spectrometry (FAAS) (PinAAcle 900T AA Spectrometer, PerkinElmer, America). To determine Cr speciation, BCR sequential extraction was applied to divide Cr into HOAc extractable, reducible, oxidizable, and residual fractions with mobility and availability from high to low (Wang et al., 2020a). Also, Cr(VI) concentration in overlying water was determined by a spectrophotometer at 540 nm, while Cr(VI) in soil was determined using FAAS after alkaline digestion (Fan et al., 2021). Duplicates, method blanks, and standard reference materials were used for quality control. Cr recovery in standard reference materials was 92~108%.

### 2.5 Microbial response during operation

- 2.5.1 Soil biochemical response
- Soil dehydrogenase (DHA) activity was measured using 2, 3, 5-triphenyl tetrazolium chloride (TTC)
- method. Urease activity was determined by the phenol sodium hypochlorite colorimetric assay. Invertase
- activity was determined by the 3, 5-dinitro salicylic acid colorimetric assay. The acid phosphatase (ACP)
- activity was determined by the p-nitrophenyl disodium phosphate colorimetric assay (Wang et al., 2019;
- 128 Wang et al., 2017).
- 2.5.2 Microbial community structure
- The microbial community structure of the electrodes and soil was determined by high-throughput
- sequencing (Faust and Raes, 2012; Deng et al., 2012; Bokulich et al., 2018). Majorbio (Shanghai, China)
- performed 16S rRNA gene sequencing using the Illumina HiSeq platform. 0.50 g of fresh homogenized
- samples were used to extract the total bacterial DNA with a universal DNA Kit (Omega Biotek Inc.,

gene were amplified with primer pairs 338F and 806R. After sequencing, the operational taxonomic units (OTUs) with a 97 % similarity cutoff were clustered using UPARSE version 7.1, and chimeric sequences were identified and removed. The taxonomy of each OTU representative sequence was analyzed by RDP Classifier version 2.2 against the 16S rRNA database using a confidence threshold of 0.7. The alpha diversity, beta diversity, microbial community structure change, and environmental factor correlation analysis were conducted. (Wang et al., 2023a) 2.5.3 HRG Fluctuation Microbial DNA Rapid extraction kit (Shenggong Bioengineering Co., LTD., Shanghai, China) was used to extract total DNA from fresh samples (The extraction method is shown in the supplementary material). The abundance of HRGs in the surface soil of SMFC and OMFC anode after operation was analyzed using an SYBR Green real-time fluorescence quantitative PCR system (7500, Thermo Fisher, USA) (Wang et al., 2023c). The soil of OMFC was used for comparison. The detected genes included HRGs (chrA, chrB, chrR, recG, nfsA, zupT, fpvA) and MGEs (intI, tnpA02, tnpA04, tnpA05). The primer sequences are provided in Table S2. The specific detection steps were as follows: pre-denaturation at 95 °C for 30 s, denaturation at 95°C for 5 s, and annealing and extension at 60°C for 30 s. Forty cycles were performed to make three replicates, and 16S rRNA was used as the internal reference gene. The relative gene expression results were analyzed using the  $2^{-4}$  method, which is commonly used for relative quantification, where  $\Delta\Delta$ Ct = (Ct target gene - Ct internal reference gene) experimental group -

USA). After amplification and purification, the V3-V4 hypervariable regions of the bacterial 16S rRNA

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(Ct target gene - Ct internal reference gene) control group.

### 2.6 Data analysis

The experimental data were evaluated using one-way analysis of variance (ANOVA) based on three tests. The mean values and standard deviations were calculated using SPSS 22.0 software (IBM, USA). A significance level of P<0.05 was considered statistically significant, while P<0.01 was defined as highly statistically significant. Graphs were plotted using Origin 2022 software.

#### 3. Results

### 3.1 Electrodes characterization

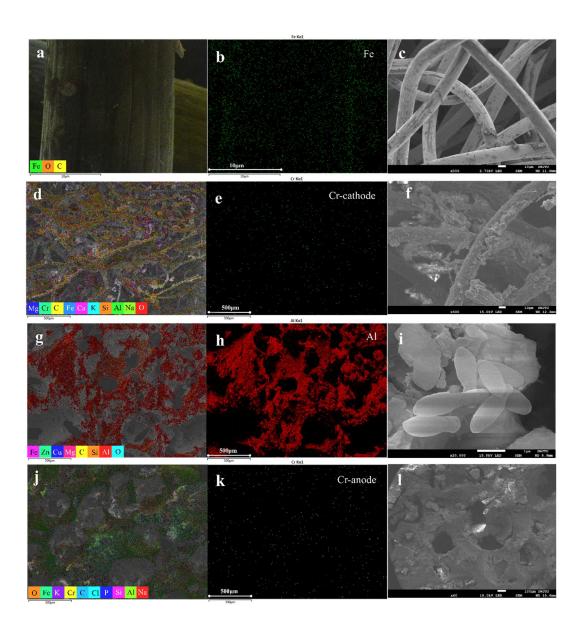
formation on the cathode (Fig. 1d-f).

As demonstrated in Fig. S2, Fig. 1, and Fig. 2, the raw GF had smooth surfaces with C, and O as the main elements (Fig. S2). After Fe<sub>3</sub>O<sub>4</sub> loading, black patches constituted with spherical particles appeared, bringing Fe (13.17%) and O (19.97%) on the GF surface (Fig. 1a-c and Table S3). XPS revealed that peaks of the cathode GF loaded with catalyst at 710.8 and 724.4 eV were consistent with typical Fe<sub>3</sub>O<sub>4</sub> peaks (Fig. 2a-b), indicating its successful loading. The CV curves of the cathode (Fig. 2c) presented an obvious oxidation peak at 0.85 V, indicating its excellent electrochemical performance.

Chromium exists mainly in the III and VI oxidation states in soil. The peaks at 576.9eV and 579eV are typical peaks of Cr(III) and Cr(VI), respectively. After operation, the typical peaks of Cr(III) and Cr(VI) can be observed on the X-ray energy spectra of both cathode and anode, and Cr(VI) and Cr(III) are present on the cathode and cathode of SMFC (Fig. 2d-e). The results showed that at least the valence state of Cr changed on the electrode and Cr(VI) was reduced and fixed by the electrode (Kim et al., 2017). GF was found loaded with many soil elements including Cr, Na, Mg, and Ca (Table S3). SEM also

observed many microorganism cells and extracellular organic-like substances, implying the biofilm

As presented in Fig. S2C, the raw aluminum foam showed a rough porous structure with mainly Al and O on the surface (Table S3). After loading EAB, many spherical and rod-shaped bacteria were observed, indicating that aluminum foam has a good capacity to carry microorganisms (Fig. 1g-i). After the operation, many millimeter-scale soil particles were embedded in the anode interspace, indicating the intense mass transfer between the anode and soil (Fig. 1j-l).



**Fig. 1** Characterization of electrode materials before and after operation by EDS and SEM. (a-c) EDS and SEM images of cathode loaded with Fe<sub>3</sub>O<sub>4</sub>; (d-f) EDS and SEM

images of cathode after the SMFC operation; (g-h) EDS and SEM images of anode microorganisms; (j-l) EDS and SEM images of the anode after SMFC operation.



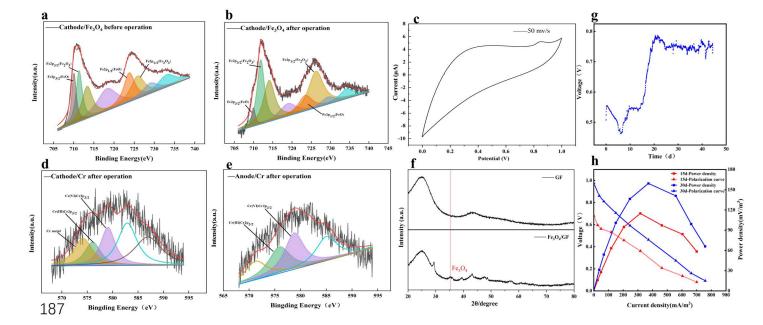


Fig. 2 Characterization of electrode materials and power generation performance of SMFC. (a-b) Fe2p spectra of cathode/Fe<sub>3</sub>O<sub>4</sub> composite cathode, (c) cyclic voltammetry (CV) curve of cathode/Fe<sub>3</sub>O<sub>4</sub>, (d-e) Cr2p spectra of GF composite cathode and Anodic Aluminum foam after operation, (f) XRD spectrum of the cathode-Fe<sub>3</sub>O<sub>4</sub>; (g) SMFC output voltage distribution, (h) 15-day vs. 30-day polarization curves and power density curves of the SMFC.

# 3.2 Electricity Generation Performance

Initially, CMFC showed a working circuit voltage (WCV) of 0.55 V and an open circuit voltage (OCV) of 0.68 V (Fig. 2g). In the first week, WCV dropped quickly to 0.45 V but bounced back and stabilized at 0.75 V on day 25, implying the adaption process of the anode microbes in the soil. We reasonably concluded that in the complex heterogeneous environment of soil, the anodic EAB needs some time to

adapt to fluctuating environmental conditions facing environmental disturbances. During SMFC operation, the anode microbial community could be gradually selected and enriched, and a stable adaptive community is formed, so the SMFC voltage tends to be stable. This conjecture is also reflected in subsequent results.

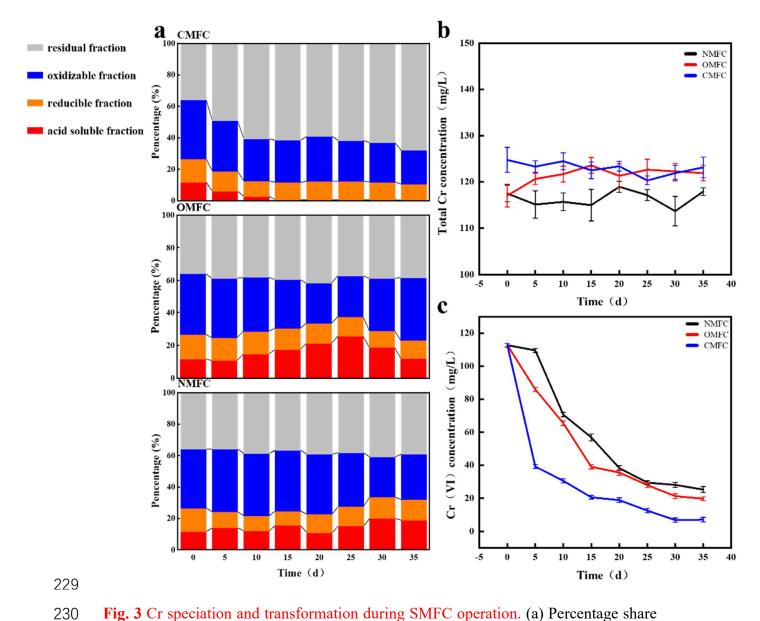
On day 15 (OCV of 0.67 V) and day 30 (OCV of 0.97 V), a series of resistors ( $50\sim10~000~\Omega$ ) was connected to the electrodes to determine the polarization curves and power density of the SMFC. As shown in Fig. 2h, the power density increased and decreased with the elevation of external resistance. At 510  $\Omega$ , the power density reached a maximum of 114.9 mW/m³ ( $73.5~\text{mW/m}^2$ ) on day 15 and 159.4 mW/m³ ( $102.0~\text{mW/m}^2$ ) on day 30 (Table S4). The result indicated that the electrochemical performance of SMFC enhanced gradually probably due to the microbial adaption. Even after 45 days, the CMFC still had a WCV of around 0.75 V, indicating its substantial and stable electricity-producing capacity. Compared with the literature, the SMFC in the current work has an outstanding power generation capacity (Table S5).

### 3.3 Cr(VI) reduction and immobilization during operation

The forms and speciations of HMs determine their bioavailability and toxicity (Jia et al., 2022). After operation, Cr forms in soils were significantly changed (P<0.05) (Fig. 3). In CMFC, the acid-soluble Cr decreased substantially by 97.44%, the oxidizable and reducible fractions did not change significantly, while the residual form of Cr increased by 88.89% (Fig. 3a). However, in NMFC, the acid-soluble Cr increased substantially by 61.54% on day 35. In OMFC, the acid-soluble Cr increased before decreasing, which was opposite to its oxidizable state. On day 35, the Cr bioavailability of OMFC (11.9%) and NMFC (18.9%) was 3866-6200 folds of CMFC (0.3%). It is inferred that the electric field and the

220	microbial communities' evolvement may lead to better Cr immobilization and transform Cr into less toxic
221	forms.

In the meantime, the Cr(VI) concentration (Fig. 3c) dropped in all the groups, and Cr (VI) in CMFC soil was significantly lower than OMFC and NMFC (P<0.05) after the experiment. On day 35, CMFC showed 13.59% and 20.87% higher Cr(VI) elimination than OMFC and NMFC, respectively. The overlying water was initially free of Cr. During the experiment (Fig. S3), 0.21~12.72 mg/L Cr was determined, which could be released from the soil. A low level of Cr(VI) (less than 3.15 mg/L) was detected but vanished later (day 15), which could be attributed to the dynamic adsorption-desorption of soil particles and electrodes.



**Fig. 3** Cr speciation and transformation during SMFC operation. (a) Percentage share of Cr in different chemical fractions in CMFC, OMFC, and NMFC soil; Changes in soil (b) total chromium and (c) Cr(VI) concentrations during SMFC operation.

# 3.4 Soil properties

3.4.1 Soil Physicochemical Property

The soil pH decreased in all the groups (Fig. S4A). The soil pH of CMFC decreased fastest from the initial 7.71 to about 6.83 on day 35, with a minimum of 6.77 on day 30, which was 0.14-7.87% lower than others (P<0.05). During the experiment, oxygen in the flooded soil decreased rapidly due to

microbial consumption, and acidic products (e.g., low-molecule organic acids) were produced to increase soil acidity. Microorganisms (especially EABs) decompose soil organic matter and release a large number of electrons and protons, making oxidizing substances such as nitrate and high valence metals (Fe(III), Mn(IV), and Cr(VI)) to accept electrons for reduction, causing protons (H<sup>+</sup>) accumulation (He et al., 2016). Such a phenomenon was more intense in CMFC due to the rapid electron transfer through wire to the cathode, leaving protons elevated near the anode.

In all the groups, EC increased rapidly from the initial 1.55 ms/cm before stabilizing (Fig. S4B), which maximized 2.6, 2.4, and 2.4 ms/cm in CMFC, OMFC, and NMFC (*P*>0.05), respectively. The rapid increase in EC could be attributed to the inundation that increased the soluble salt content of the soils. The electromigration in the MFC electric field may also increase soil mass transfer and positively affect soil electrical conductivity (Zhang et al., 2020).

3.4.2 Soil Biochemical Response

Soil enzyme activity is an important index to evaluate soil environmental change. Soil enzymes, as biocatalysts involved in biochemical reactions, play an important role in nutrient mineralization, decomposition of organic matter, and nutrient cycling (Chen et al., 2024; Liu et al., 2018). The DHA activity increased significantly in all three groups (2 244.0~3 138.0% higher than the initial value) and continuously (*P*<0.05) (Fig. S5A). Under flooding, microbial activity changed from aerobic to anaerobic, leading to a sharp decline in soil redox potential, accompanied by the stimulation of soil DHA (Sardans and Peñuelas, 2005). During operation, urease activity in CMFC showed a gradual increase (2.70~12.40% higher than day 0 from days 10~35), while it in OMFC and NMFC showed a slight decrease (6.10~7.10% lower than day 0 from days 5~35) (Fig. S5B). SMFC electric field and Fe(II) promote extracellular

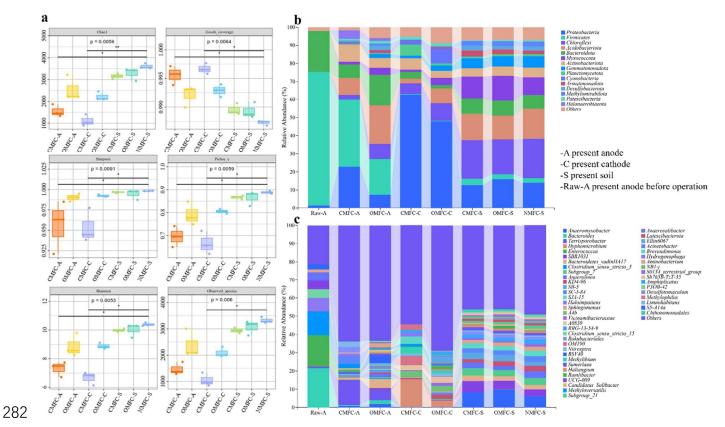
electron transfer (EET) (Chen et al., 2023a), which promotes the enrichment of ammonia-nitrogen transforming bacteria in soil could have caused the higher urease activity in CMFC than NMFC and OMFC (P<0.05). Soil invertase activity decreased initially but increased later for CMFC and OMFC, but showed an opposite trend for NMFC. After operation, CMFC had a significantly higher invertase activity than others (P<0.05) (Fig. S5C). Soil ACP showed a similar trend with urease, with CMFC continuously increasing by 13.20~48.90% and considerably higher than OMFC and NMFC (Fig. S5D). The dynamic measurement of soil enzyme activity in SMFC helps us to understand and analyze the changing state of soil microecology.

# 3.5 SMFC operation reshaped soil microbial community

Microbial community structures in the electrodes were analyzed, which obtained 15 dominant phyla and 50 dominant genera (>1.0%). Overall, *Firmicutes* (73.93%), *Proteobacteria* (62.53%), and *Chloroflexi* (21.73%) were found the dominant phyla, while *Bacteroides* (21.48%), *Enterococcus* (17.26%), and *Hyphomicrobium* (15.34%) were the dominant genera.

The alpha diversity analysis indicated a significant difference in the microbial community among the samples (Fig. 4a). The higher chao1 index in soils than the electrodes demonstrated higher microbial richness. Most of the alpha index in OMFC-A and OMFC-C were significantly higher than CMFC-A and CMFC-C, demonstrating a higher microbial richness and diversity in OMFC. The results indicated the different microbial evolution patterns in different electrodes and the selection effect of the electricity field during CMFC operation. The Venn diagram (Fig. S6) found no OTU coincidence among the samples, indicating their obvious specificity. In comparison, 2130 OTUs were shared by CMFC-S, OMFC-S, and NMFC-S, indicating the similarity of the soil microbial community (Fig. S6B). 45 OTUs

were shared by Raw-A, CMFC-A, and OMFC-A, accounting for 45.92%, 1.95%, and 1.16%, respectively, indicating the successful colonization and development of the preloaded EABs.



**Fig. 4** Microbial response during SMFC operation. (a) Alpha diversity analysis of the electrodes and soils; Microbial community structure is based on (b) the phylum level and (c) the genus level.

3.5.1 Soil microbial community reshaping on phylum level

At the phylum level (Fig. 4b), before the operation, the anode was dominated by *Firmicutes* (73.93%), *Bacteriodota* (22.74%), and *Proteobacteria* (1.27%). After the operation, *Firmicutes* and *Bacteriodota* decreased by 49.83% and 66.84% in CMFC-A, and 73.20% and 26.52% in OMFC-A, respectively. While *Proteobacteria* increased by 1 698.43% in CMFC-A and 475.59% in OMFC-A. Besides, many other phyla emerged, including *Acidobacteriota* (9.41%~21.40%), *Actinobacteriota* (0.23%~9.52%), *Halanaerobiaeota* (1.48%~4.77%), *Myxococcota* (1.51%~3.93%), *Chloroflexi* (0.01%~2.57%),

#### indicating the increased microbial diversity, probably due to the penetration of soil indigenous microbe.

The cathode was free of microorganisms initially. However, many phyla were observed after the operation. The CMFC-C was dominated by *Proteobacteria* (62.53%), *Actinobacteriota* (6.24%), *Planctomycetota* (6.04%), *Chloroflexi* (5.88%), and *Bacteroidota* (3.81%), while OMFC-C was dominated by *Proteobacteria* (47.93%), *Chloroflexi* (9.75%), *Cyanobacteria* (8.30%), *Acidobacteriota* (8.01%), *Actinobacteriota* (3.97%), *Myxococcota* (3.92%), and *Gemmatimonadota* (2.87%). The *Proteobacteria* phylum was rich in EABs, its advantage in both electrodes of CMFC indicated that SMFC operation was favorable for EAB colonization and development. All the soils were dominated by *Chlorobacteria*, *Acidobacteria*, *Proteobacteria*, *Bacteroidetes*, and *Myxococcota*, and the difference was not significant.

303 3.5.2 Soil microbial community reshaping on genus level

At the genus level (Fig. 4c), MFC operation presented a selection effect, with *Terrisporobacter* increasing from 0.81% to 13.71% and *Bacteroides* decreasing from 12.48% to 0.53% in CMFC anode. Compared with the Raw-A, many EABs in CMFC-A decreased, including *Clostridium\_sensu\_stricto\_5* (from 12.99% to 0.052%), *Clostridium\_sensu\_stricto\_15* (from 4.70% to 0.47%), *Enterococcus* (from 17.26% to 0.03%) (Choi, 2022; Zhang et al., 2023).

However, the *Desulfotomaculum* in CMFC-A increased to 3.32% compared with 0.003% in the soil (CMFC-S). Besides, soil indigenous bacteria including *Ramlibacter, Methyloversatilis*, and *Acinetobacter* colonized in the anode and elevated by 4.89~1 579 fold compared with soil. Nevertheless, multiple dominant genera in the soils decreased in CMFC-A than in OMFC-A. For example, *SBR1031*, *Bacteroidetes vadinHA17*, and *Anaerolinea* were significantly increased in soil, but less in CMFC-A

and OMFC-A. The electric field action to a certain extent helped the anode to resist external microbial intrusion to ensure the stability of the anodic microbial community.

During operation, the prolonged interaction between the soil and water phases resulted in the gradually evolving unique biofilm structure of the cathode. For instance, *Hyphomicrobium* (3.56~15.34% in soils), an aerobic chemoheterotroph capable of degrading a wide range of organics, accounted for 15.34% and 3.56% of CMFC-C and OMFC-C, respectively (He et al., 2019). *Hydrogenophaga*, a gramnegative bacteria capable of denitrification and Cr(VI) reduction, accounted for 2.07% of CMFC-C (Wang et al., 2022). Meanwhile, the SMFC operation caused the enrichment of several resistant bacteria. *Subgroup\_7*, a typical HM-tolerant bacterium (Li et al., 2023a), was enriched in both cathode and soil. *Acinetobacter* and *Limnohabitans*, also tolerators that carry HRGs and ARGs, were found 4.31% and 3.03% in CMFC-A (Dahal et al., 2023; Zhang et al., 2021).

The increase of iron in soil and water due to the use of Fe<sub>3</sub>O<sub>4</sub> as cathode catalyst may be responsible for the enrichment of *Terrisporobacter* and *Anaeromyxobacter* in the CMFC-A and OMFC-A. They were found closely associated with Fe<sup>3+</sup> reduction to gain energy in various environments (Lin et al., 2007; Wang et al., 2020b).

329 3.5.3 Soil metal resistance gene variation

Under Cr(VI) stress, certain microbes would utilize pathways like specific or non-specific Cr(VI) reduction, free radical detoxification, DNA damage repair, etc. to survive in toxic environments (Morais et al., 2011). Using qPCR analysis, the abundance of typical HRGs and MGEs in the anodic soils was determined (Fig. 5a), among which *chrA*, *chrB*, *and chrR*, are typical HRGs involved in Cr response, while *IntI*, *tnpA02*, *and tnpA05* are common mobile genetic elements (MGEs) that varied greatly during

operation (*P*<0.05). Compared with OMFC and NMFC, *chrA* in CMFC increased by 237.83% and 3414.34%, *chrB* by 141.52% and 153.63%, *chrR* by 221.86% and 839.41%, *IntI* by 151.77% and 167.91%, *tnpA02* by 331.86% and 1118.97%, and *tnpA05* by 416.91% and 99.54%. Changes in soil resistance characteristics during SMFC operation may be directly or indirectly caused by environmental factors and resistant microbial communities.

The elevation of HRGs and MGEs could be due to the enrichment of multiple metal-resistant bacteria (MRB) such as *Acinetobacter, Limnohabitans*, and *Brevundimonas*. Moreover, the anodic *Desulfotomaculum*, which accounted for 3.23% of CMFC-A, is a typical sulfate-reducing bacterium (SRB) that produces H<sub>2</sub>S, a natural signaling molecule that contributes to tolerance triggering, maintenance, and diffusion through community sensing, which facilitates HRG elevation through HGT (Shatalin et al., 2021). Besides, Cr(VI) reducing bacteria like *Hydrogenophaga* (1.31% in CMFC-A) may also up-regulate the Cr reductase gene *chrR* (Sundarraj et al., 2023).

Furthermore, pH changes may also affect soil resistance characteristics. Liu et al. (2023c) observed the abundance of multidrug efflux pump genes in the acid soil was significantly positively correlated with soil acidity. The intensified proton generation and accumulation in CMFC could have led to the HRG elevation. In addition, HMs toxicity exerts direct selective pressure, which affects microbial community structure and their function, leading to the thriving of tolerators like *Desulfotomaculum sp*, *Hydrogenophaga*, and *Methylophilus* (Hernández-Ramírez et al., 2018), hence the spontaneous HRG elevation (Wang et al., 2023c).

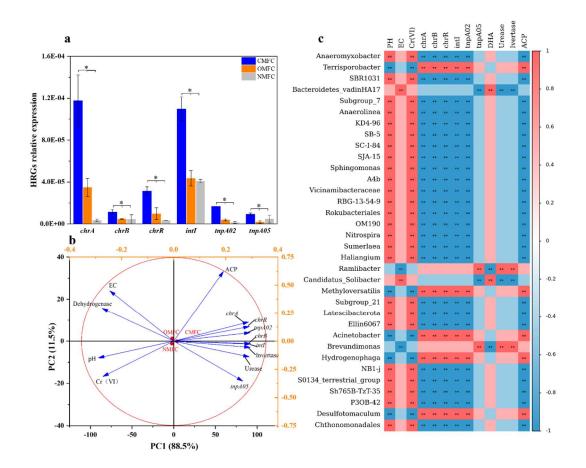


Fig. 5 HRG variation and correlation analysis of dominant bacteria, HRG, and soil enzymatic activity. (a)qPCR results of HRG changes in soil around anode-soil after SMFC operation (*chrA*, *chrB*, *chrR*, *intI*, *tnpA02*, *tnpA05*). Different letters denote significant differences among treatments (P < 0.05); (b) principal component analysis (PCA) of SMFC soil physicochemical properties with Enzyme activity, HRGs, and native bacterial genera and (c) Spearman's correlation heatmap (mean relative abundance > 1%). \* P < 0.05, according to LSD test (mean  $\pm$  S.E., n = 3); \*\* P < 0.01.

# 3.6 Correlation analysis

To visually analyze the correlation between bacterial communities and environmental factors, spearman correlation analysis and principal component analysis (PCA) were conducted (Fig. 5b). Spearman

correlation analysis (Fig. 5c ) isolated four main bacterial genera clusters, which further revealed the correlation of environmental factors (pH, EC), soil enzyme activities (DHA, Urease, Invertase, ACP), Cr(VI), and HRGs (chrA, chrB, intI, tnpA02, tnpA05) with the microbial community. Cluster 1 (Terrisporobacter, Methyloversatilis, Acinetobacter, Hydrogenophaga, and Desulfotomaculum) was positively correlated with HRGs (chrA, chrB, intI, tnpA02) and ACP (P<0.01), but negatively correlated with pH and Cr(VI) (P<0.01), suggesting they may contribute to HGT and HRGs enrichment. Cluster 2 (Anaeromyxobacter, Subgroup 7, Anaerolinea, SB-5, Sphingomonas, etc.) was negatively correlated (P<0.01) with HRGs (chrA, chrB, intl, tnpA02) and ACP, but positively correlated with pH and Cr(VI) (P<0.01). Cluster 3 (Bacteroidetes vadinHA17 and Candidatus Solibacter) was positively correlated (P<0.01) with soil EC and DHA. Cluster 4 (Ramlibacter, and Brevundimonas) were positively correlated with urease and invertase but negatively correlated with EC. PCA analysis (Fig. 5b) also indicated a close correlation between environmental factors, in which soil pH, EC, Cr(VI) concentration, and DHA were positively correlated with each other but negatively correlated with urease, ACP, invertase, and HRGs. Especially, Cr(VI) was significantly negatively correlated with HRGs (P<0.01), which could be partially explained by the tolerators thriving and Cr(VI) reduction during SMFC operation.

# 4. Discussion

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In this study, Cr(VI) reduction, microbial community variation, and HRG fate in SMFC were investigated for the first time. The results proved that SMFC was an effective method to eliminate Cr(VI) (93.76%), immobilize Cr (97.44%), and generate electricity (0.97 V).

Microorganisms have developed efficient detoxification strategies to counteract the toxic effects of

heavy metal stress (Rouch et al., 1995; Tan et al., 2020). In the SMFC system, Cr(VI) reduction was a synergic result of adsorption/biosorption (including the adsorption of anode and cathode materials, surface catalyst adsorption, and microbial membrane adsorption), bioelectrochemistry reduction, and microbial reduction (including intracellular sequestration, export, reduced permeability, extracellular sequestration, and extracellular detoxification) (Fig. 6). The preloading of Fe<sub>3</sub>O<sub>4</sub> and EABs on the electrodes significantly improved Cr(VI) reduction and power generation by accelerating SMFC stabilization. The detailed explanation is as follows:

The electricity-producing process of SMFC can inhibit HMs' release and migration in soil (Zhu et al., 2019; Feng et al., 2024). In this study, Cr forms changed greatly from acid-soluble to a more stable residual fraction with low toxicity. The Fe<sub>3</sub>O<sub>4</sub>-modified cathode not only directly adsorbs or reduces Cr(VI) due to the high specific area and ferrous iron, but also enhances the electrochemical effect of the system. The electrons derived from anodic microbial metabolism can directly reduce Cr(VI) in the soil to Cr(III), while part of them is transmitted to the cathode, where Cr(VI) in the overlying water compete with oxygen as electron acceptors and complete the current loop (Thapa et al., 2022).

In addition, microorganisms can also directly or indirectly reduce or fix Cr. Biosorption, sulfide, and hydroxide precipitation are the main immobilization mechanisms of HMs by microorganisms (Ma et al., 2024). For example, *Desulfotomaculum sp.*, a typical SRB, enriched to 3.23% in CMFC-A, may produce sulfide ions by reducing alienated sulfate, thus forming highly insoluble metal sulfide to fix Cr through microorganism-induced sulfide precipitation (MISP). *Hydrogenophaga*, which dominated in both electrodes, was a known Cr-reducing bacteria. Some of iron-reducing bacteria present in this study (e.g., *Anaeromyxobacter* and *Terrisporobacter*) may also contribute to Cr(VI) reduction by participating

in the Fe cycle through EET, while the ferrous iron reduces Cr(VI). Additionally, the CMFC in this work contains many genera capable of transforming nitrogen. For example, Hyphomicrobium dominated in CMFC-C (15.34%), a typical denitrifying bacterium, can effectively reduce nitrate and nitrite (Ernst et al., 2021). Methylophilus accounted for 2.93% of CMFC-A but was much lower in other groups, which is a methylotrophic microorganism (Yang et al., 2020). The bacteria mentioned above were found with high urease-producing ability, whose enrichment not only improves soil urease activity and nutrient cycling but also immobilizes Cr through microorganism-induced carbonate precipitation (MICP) (Qian et al., 2017). In addition to the reduction of Cr during SMFC operation, under electrochemical selection and HMs stress, the microbial community gradually evolved with higher richness and diversity, along with the HRG enrichment and nutrient cycling variation. Firstly, some soil indigenous bacteria were much lower in CMFC-A than OMFC-A, indicating the electric field contributed to the anode stability by preventing external bacteria intrusion and is less vulnerable to environmental fluctuations. The microbial community change is significantly related to HRG enrichment. Many EABs and MRBs are significantly enriched. For example, Desulfotomaculum, an SRB with a dual role of electroproduction and HMs reduction (Jiang et al., 2020; Yin et al., 2021). Other examples also include cumulative-resistant bacteria like Acinetobacter and Limnohabitans which are only enriched in the CMFC-A (Dahal et al., 2023; Al-Jabri et al., 2018). Their enrichment directly causes vertical gene transfer and HRG elevation. The Cr forms, soil physicochemical properties, soil enzyme activities, and microecological structure mirrored each other, helping to understand Cr transformation patterns and target the key factors affecting metal resistance changes.

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HMs existence in soil can induce HGT occurrence and cause ARG elevation, which has become a major concern (Chen et al., 2023b; Fu et al., 2023). Sub-lethal levels of metal ions can increase mutation rates and enrich de novo mutants with significant resistance to multiple antibiotics (Li et al., 2019). This study focused on the toxic alleviation of a single HM (Cr) in SMFC, during which tolerator accumulation caused considerable HRG enrichment. SMFC is an eco-friendly and cost-effective technology for the insitu bioremediation of contaminated soil/sediment and powering environmental sensors in remote areas. It has the potential to be used as a novel early warning system for soil environmental hazards. Nevertheless, before the commercialization of large-scale applications in the field, significant efforts should be made to reveal the HRG enrichment mechanism during SMFC operation and pay attention to ARG change under HMs contamination or HMs-antibiotic co-contamination.

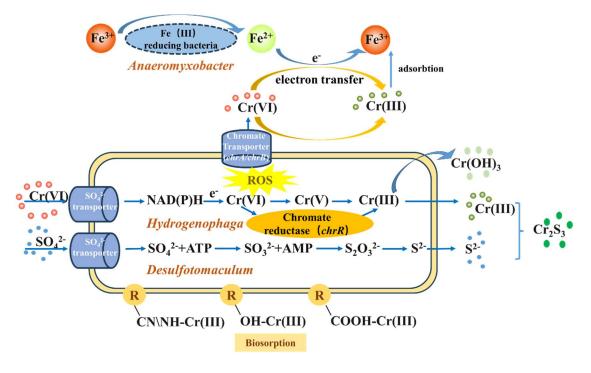


Fig. 6 Cr(VI)reduction Mechanism of during SMFC operation

### 5. Conclusion

During SMFC operation, the close interaction among the soil physicochemical properties, enzyme

activities, resistance genes, and microbial community structure determined the performance. The preloading of Fe<sub>3</sub>O<sub>4</sub> in the cathode and EAB in the anode greatly contributed to power production and Cr(VI)
elimination. Anodic microbial metabolism, cathodic redox, and the MFC electric field reduced or
immobilized Cr(VI) to eliminate its risk. The enrichment of multiple MRBs, such as *Acinetobacter*; *Limnohabitans*, and *Desulfotomaculum*, resulted in HRG elevation, which contributes to microbial
adaptation and function but brings concerns for future application. This study provides a reference for
the remediation of HM-contaminated soil using MFC, which is conducive to promoting the practical
application of bioelectrochemical technology in the field.

#### **Author contribution:**

HN: Conceptualization, Investigation, and Writing; XL: Investigation, Visualization; PL: Investigation, Visualization; HQ: Methodology; LJ: Writing-review and editing; SM: Writing-review and editing; MW: Writing-review; FX: Funding acquisition, Supervision; HX: Funding acquisition, Supervision; CW:

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# **Declaration of interests**

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- The authors declare that they have no known competing financial interests or personal relationships
- that could have appeared to influence the work reported in this paper.

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751	Supplementary materials for
752	Cr(VI) reduction, electricity production, and microbial resistance variation in paddy
753	soil under microbial fuel cell operation
754	Sections
755	Section 1: Preparation of cathode supported by Fe <sub>3</sub> O <sub>4</sub> catalyst
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769	Fig. S2 Electrode material characterization. (A, B) SEM images of GF without catalyst
770	loading; (C, D) SEM images of aluminum foam.
771	Fig. S3 Variation of (A) total chromium and (B) Cr(VI) in overlying water during

- 772 SMFC operation.
- 773 **Fig. S4** pH (A), EC (B) variation curves of soil.
- 774 Fig. S5 Changes in soil enzyme activities during SMFC operation (A) Dehydrogenase
- 775 (B) Urease (C) Invertase (D) Acid Phosphatase.
- 776 **Fig. S6** Venn diagram on OTU level in different treatments.
- Fig. S7 Characterization of electrode materials before and after operation by EDS mapping. (A) EDS image of cathode loaded with Fe<sub>3</sub>O<sub>4</sub>; (B) EDS image of cathode
- after the SMFC operation; (C) EDS image of anode microorganisms; (D) EDS image
- 780 of the anode after SMFC operation.

- **Section 1:** Preparation of cathode supported by Fe<sub>3</sub>O<sub>4</sub> catalyst
- Cathode preparation: 100×50×3 mm graphite felt (GF) (purchased from Jiangsu

  784 Xinye Electronic Materials Factory, Jiangsu, China) was selected as the cathode
- material of the SMFC, which was ultrasonicated in ethanol for 50 min to remove
- 786 impurities, removed, and washed. Then put it into an oven and dried at 60°C for 12 h.
- 787 It was put into the hydrothermal synthesis reactor and added concentrated nitric acid,
- and reacted at 90°C for 9 h. After the reaction was completed, the GF was rinsed
- 789 continuously with ultrapure water to ensure that the pH of the rinsed water was neutral,
- and dried at  $60^{\circ}$ C for 12 h to complete the pre-activation of the GF. The Fe<sub>3</sub>O<sub>4</sub> catalyst
- 791 was obtained by dissolving 3.6 g of ferric chloride hexahydrate, 6 g of sodium acetate,
- and 1 g of sodium citrate in 140 ml of ethylene glycol, and repeatedly ultrasonicating

for 2 h to form a homogeneous solution. The pre-activated GF was put into a hydrothermal synthesis reactor: the  $Fe_3O_4$  catalyst was added and soaked for half an hour and then heated at 200°C for 8 h. After heating, the reactor was cooled down to room temperature, washed repeatedly with ultrapure water and ethanol, and then put into an oven at 60°C to dry for 12 h. The reaction was then dried at 60°C.

## Section 2: Python code of Raspberry Pi voltage acquisition system

```
800
            1.
                 # -*- coding:UTF-8 -*-
801
                  from threading import Timer
802
                 import time
             3.
803
                 import RPi.GPIO as GPIO
             4.
804
                 import datetime
             5.
805
             6.
                 import os
806
             7.
                 import sys
807
                  import xlwt
808
             9.
809
             10. sys.path.append('./modules/')
810
             11.
811
              12. from GetVoltage import get_voltage
812
              13. from modules.ADS1263 import ADS1263
813
             14.
814
              15. GPIO.setmode(GPIO.BCM)
815
              16. CollectTimes = 150
816
             17.
817
             18.
                 def get average list():
                    """smooth"""
818
             19.
819
             20.
                    if CollectTimes <= 0:
820
             21.
                      print("Error number for the array to averaging!/n")
821
             22.
                      return -1
822
             23.
                    elif CollectTimes <= 5:</pre>
823
             24.
                      return sum(VoltageArray) / CollectTimes
824
             25.
                    else:
825
             26.
                      return sum(VoltageArray) / CollectTimes
826
             27.
827
             28. # clean txt
828
             29. book = xlwt.Workbook(encoding='utf-8',style compression=0)
829
             30. sheet = book.add sheet('MFCdata',cell overwrite ok=True)
830
             31. col = ('current time','MFC1','MFC2','MFC3')
831
             32. for i in range(0,4):
832
             33.
                    sheet.write(0,i,col[i])
833
             34.
834
             35. x=0
835
             36. while(1):
836
             37.
                    x=x+1
837
             38.
838
             39.
                    time.sleep(1)
839
             40.
```

```
840
           41.
                 VoltageArray = []
841
           42.
                 for i in range(CollectTimes):
842
           43.
                  VoltageArray.append(get_voltage(0))
843
           44.
                 adc1 = _get_average_list()
844
           45.
845
           46.
                 846
           47.
847
           48.
                 VoltageArray = []
848
           49.
                 for i in range(CollectTimes):
849
           50.
                  VoltageArray.append(get voltage(1))
850
           51.
                 adc2 = _get_average_list()
851
           52.
852
           53.
                 853
           54.
854
           55.
                 VoltageArray = []
855
           56.
                 for i in range(CollectTimes):
856
           57.
                  VoltageArray.append(get_voltage(2))
857
           58.
                 adc3 = _get_average_list()
858
           59.
859
           60.
                 860
           61.
861
           62.
                 curr time = datetime.datetime.now()
862
           63.
                 time_str = datetime.datetime.strftime(curr_time,'%Y-%m-%d %H:%M:%S')
863
           64.
864
                 datalist = [time str,str(adc1),str(adc2),str(adc3)]
           65.
865
           66.
                 print(datalist)
866
           67.
867
           68.
                 for j in range(0, 4):
868
           69.
                  sheet.write(x, j, datalist[j])
869
           70.
                  savepath = '/home/pi/excel.xls'
870
           71.
                  book.save(savepath)
871
           72.
872
         73.
                 time.sleep(598.5)
```

## **Section 3: Sampling method**

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We used a plastic cylindrical straw with a diameter of 0.4 cm and a length of 16 cm as the sediment sampler. The SMFC sediment part is inserted by the sampler vertically at a specific time, and then quickly removed, and the upper, middle, and lower parts of the sampler are mixed as a determination sample. And the fresh sample each time is only 8-16 g, only 0.2-0.4% of the SMFC. The total sampling amount shall not exceed 5% of the total population. Because the sampler is much smaller than SMFC, the disturbance is avoided to a great extent and the normal operation of SMFC is guaranteed. Section 4: Soil DNA extraction method Microbial DNA Rapid extraction kit (Shenggong Bioengineering Co., LTD., Shanghai, China) was used to extract total DNA from fresh samples. Specifically, 0.50 g sample, 0.50 g magnetic beads, and 1.0 ml SLX-Mlus Buffer were added in a 2.0 ml Eppendorf tube, and ground for 250 s under 45 HZ. Then added and mixed with 100 µl DS Buffer, and cultivated under 70 °C for 10 min and then 90°C for 2 min. Then the mixture was centrifuged at 10000 g for 5 min at room temperature. 800 µl supernatant was moved to a new tube and added with 270 µl P2 buffer and 100 µl HTR reagent, and then cultivated under -20 °C for 5 min and then centrifuged again at 10000 g for 5 min. The supernatant was then moved to a new 2 ml tube added with the same amount of XP5 buffer and mixed upside down for 8 min. After magnetic rack adsorption, discard the residual liquid, remove the tube, add 500 µL XP5 Buffer, and mix well. Then adsorbed again with a magnetic rack, discard the residual liquid, remove the tube, add 500 µL

PHB, and mix well. Then adsorbed again with a magnetic rack, discard the residual liquid, remove the tube, add 500  $\mu$ L SPW Wash Buffer, and mix well (repeat this step twice). Then the mixture was adsorbed again with a magnetic rack, discard the residual liquid, was centrifuged in the tube under 10000 g for 10 s. Then the beads were adsorbed again with a magnetic rack, discard the residual liquid, and let stand for 8 min. After that, the beads were added with 100  $\mu$ L elution buffer, mixed, and let stand for 5 min. Finally, after adsorbing with a magnetic rack, the supernatant was moved to a new 1.0 ml Eppendorf tube, and total DNA was obtained for further use. The PCR reaction system was constructed.

Table S1 Main material dimensions

Material	Length/cm	Width/cm	Area/cm <sup>2</sup>
Collector plate	10.5	5.5	57.75
Aluminum foam	6.6	5.4	35.64
GF	10.0	5.0	50.00

Table S2 Primer sequence of HRGs and MGEs

Gene	Primer sequence	Function description	Ref
		chrA encodes a transporter	(Rivera
chrA-F	TCCTTCGGCGGCCCTGCCGGNCARATHGC	protein involved in chromate	et al.,
chrA-R	GTAGGTGGCCAGCTGCTNGCYTCNGGNCC	ca	2000)
		efflux.	2008)

			(Branco
chrB-F	CCGGAATTCATGCGTGTCTGGCGAACCCTGA	chrB genes regulate the	and
chrB-R	CCCAAGCTTTCACTCTGCGGAAGAACGA	transcription of genes in the	Morais,
		transporter protein complex.	2013)
			(Nepple
			(r.veppre
		The chromate reductase chrR is	et al.,
ChrR-F	AGGAACTTCTGCGTGCCCTC	the best-known of the reductases	2000;
ChrR-R	TACGGTGACAGTGCGTTTGC	that catalyze the reduction of	Baldiris
		$Cr^{6+}$ to $Cr^{3+}$ .	et al.,
			2018).
IntI-F	CGAACGAGTGGCGGAGGGTG	MGEs such as integrons,	(Wu et
IntI-R	TACCCGAGAGCTTGGCACCCA	plasmids, and transposons play a	al.,
tnpA02-F	GGGCGGTCGATTGAAA	key role in the transfer of	2022;
tnpA02-R	GTGGGCGGGATCTGCTT	resistance genes between	Wu et
tnpA05-F	GCCGCACTGTCGATTTTTATC	different microorganisms in the	al.,
tnpA05-R	GCGGGATCTGCCACTTCTT	environment	2023) <del>.</del>

Table S3 Distribution percentage of EDS elements in electrode materials.

_	Samples					
Elements	Cathode	Fe <sub>3</sub> O <sub>4</sub> -	Cathode after	Anode	Anode with	Anode after

		cathode	operation		EAB-loading	operation
C	93.74%	66.85%	33.83%	6.7%	14.65%	14.35%
O	6.26%	19.97%	38.96%	47.57%	18.49%	50.92%
Fe	n.d.	13.17%	8.07%	n.d.	n.d.	0.88
Mg	n.d.	n.d.	0.44%	n.d.	0.41%	n.d.
Al	n.d.	n.d.	3.34%	37.04%	53.72%	17.56%
Cr	n.d.	n.d.	0.08%	n.d.	n.d.	0.12%
Cu	n.d.	n.d.	n.d.	n.d.	1.59%	n.d.
Zn	n.d.	n.d.	n.d.	n.d.	1.81%	n.d.
Si	n.d.	n.d.	7.16%	8.68%	9.32%	2.58%
Na	n.d.	n.d.	2.18%	n.d.	n.d.	4.55%
Ca	n.d.	n.d.	4.81%	n.d.	n.d.	n.d.
K	n.d.	n.d.	1.12%	n.d.	n.d.	0.58%
C1	n.d.	n.d.	n.d.	n.d.	n.d.	2.67%
P	n.d.	n.d.	n.d.	n.d.	n.d.	5.81%

Table S4 SMFC power generation performance on 15-day and 30-day

Resistor ( $\Omega$ )	15d-Current density	15d-Power density	30d-Current density	30d-Power density
	(mA/m <sup>2</sup> )	$(mW/m^2)$	$(mA/m^2)$	$(mW/m^2)$
51	448.57	37.43	485.23	42.21
100	386.53	53.84	424.57	63.47
200	296.55	62.24	357.26	90.22
510	202.88	73.52	238.08	102.02
1000	131.18	60.60	157.54	87.35
2000	75.99	40.43	95.91	64.58
5100	31.55	17.77	43.88	34.41

10000	17.01	10.13	23.95	20.09

**Table S5** Performance comparison of various configurations of SMFC.

Reactor configuration	Chamber volume (L)	Output voltage	Maximum power density (mW/m²)	Reference
Single-chamber SMFC	0.72	0.33	17.3	(Li et al., 2016)
Single-chamber SMFC	2.16	0.297	12.1	(Yu et al., 2017)
Two-chamber SMFC	4.2	0.399	29.78	(Srivastava et al., 2019)
Single-chamber circle SMFC	0.95	0.345	24	(Yu et al., 2021)
Single-chamber SMFC	8	0.4-0.6	0.2 mW	(Yoon et al., 2023)
Single-chamber circle SMFC	n.m.	n.m.	25.51	(Wang et al., 2023)
Single-chamber SMFC	9.7	n.m.	70.4±1.4	(Dhillon et al., 2023)
Two-chamber SMFC	0.05	n.m.	$71.00 \pm 0.82$	(Zhang et al., 2023a)
Constructed wetland MFC	38.85	0.197	12.5	(Tao et al., 2023)
Constructed wetland MFC	4.08	0.425	28.1	(Niu et al., 2023)
Plant-SMFC	0.769	0.51	46.8	(V et al., 2023)
Single-chamber circle SMFC	2	0.17	10	(Youssef et al., 2023)
Constructed wetland MFC	2.7	0.394	4468.4	(Zhang et al., 2023b)
Two-chamber SMFC	0.5	0.306	20.35	(Zhang et al., 2024)
Constructed wetland MFC	13	0.167	1.4	(Dai et al., 2024)
Plant-SMFC	1.05	0.55	8957.7	(Chen et al., 2024)
Single-chamber circle SMFC	1.6	0.4	472.52±14.2	(Zhao et al., 2024)
Single-chamber circle SMFC	1.4	0.528	178.17	(Sun and Wang, 2024)

Single-chamber SMFC 2.3 0.75 102 This study

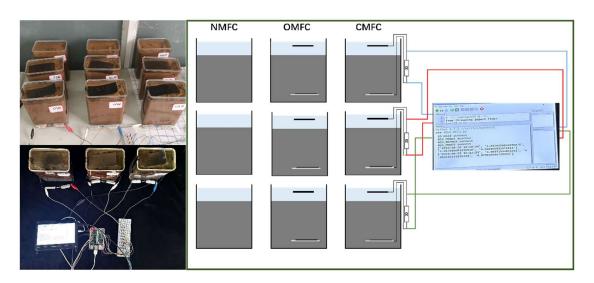
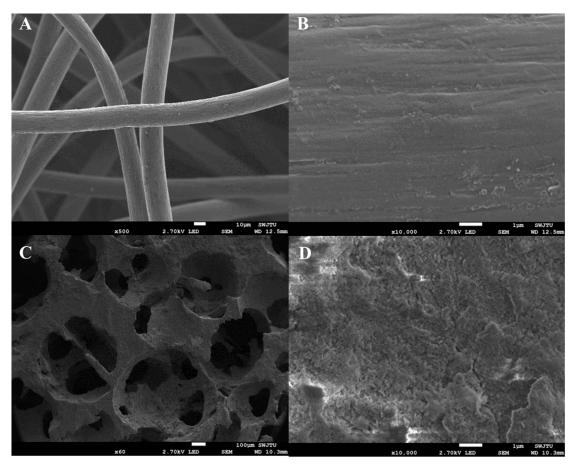


Fig. S1 SMFC structure and experimental grouping.

The left part of the picture is the SMFC photo, and the right part is the model groups. A plastic box ( $140.0 \times 85.0 \times 165.0$  mm) was used as the SMFC reactor, with 1.50 kg soil and overlying water of 3.0 cm to simulate the flooded state during rice planting. The cathode was floated on the water surface while the anode was buried (about 3.0 cm from the bottom). The cathode and anode were connected to a  $2000~\Omega$  resistor using titanium wire. The water level was kept constant by daily replenishment.



**Fig. S2** Electrode material characterization. (A, B) SEM images of GF without

catalyst loading; (C, D) SEM images of aluminum foam.

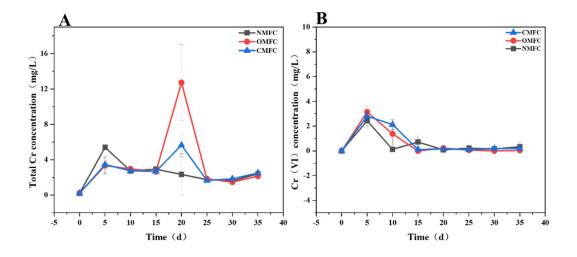


Fig. S3 Variation of (A) total chromium and (B) Cr(VI) in overlying water during

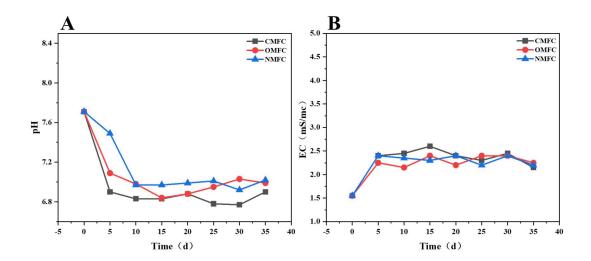


Fig. S4 pH (A), EC (B) variation curves of soil.

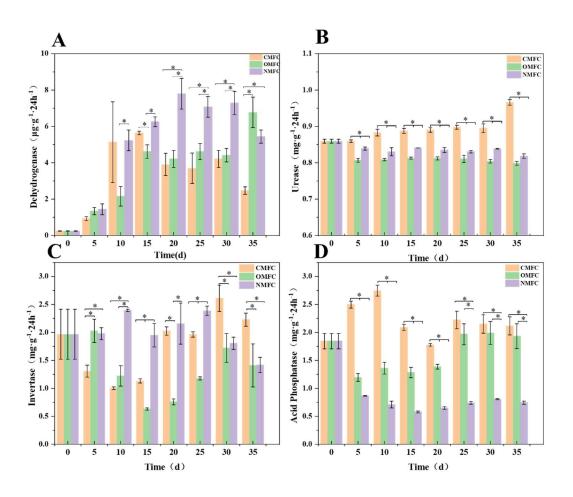


Fig. S5 Changes in soil enzyme activities during SMFC operation (A) Dehydrogenase

(B) Urease (C) Invertase(D) Acid Phosphatase.

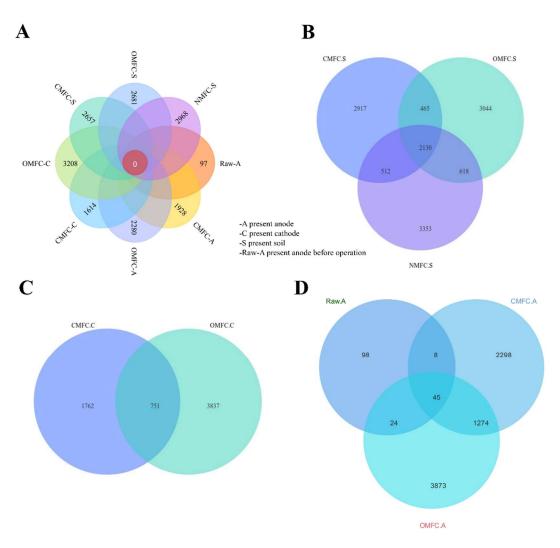
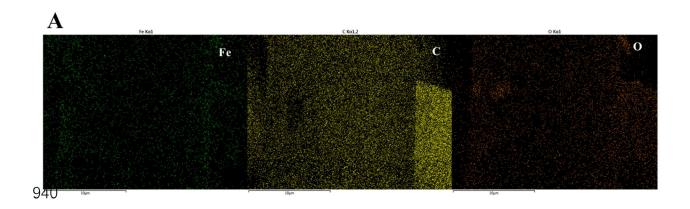
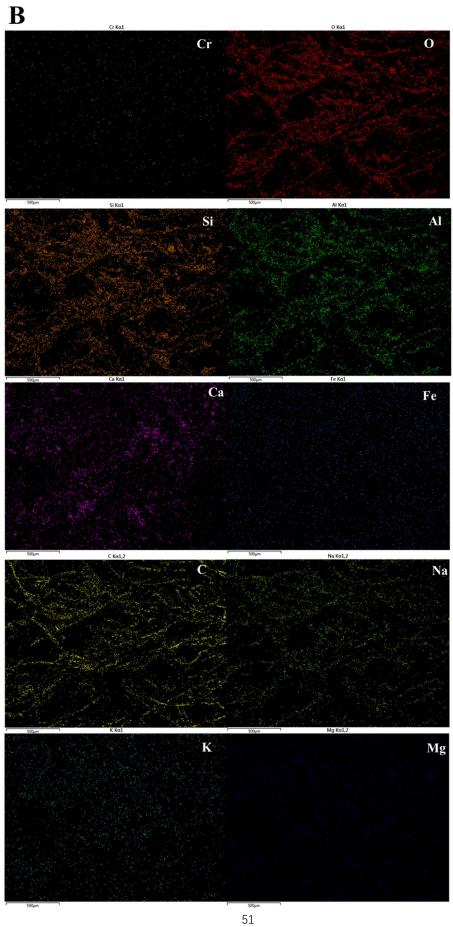
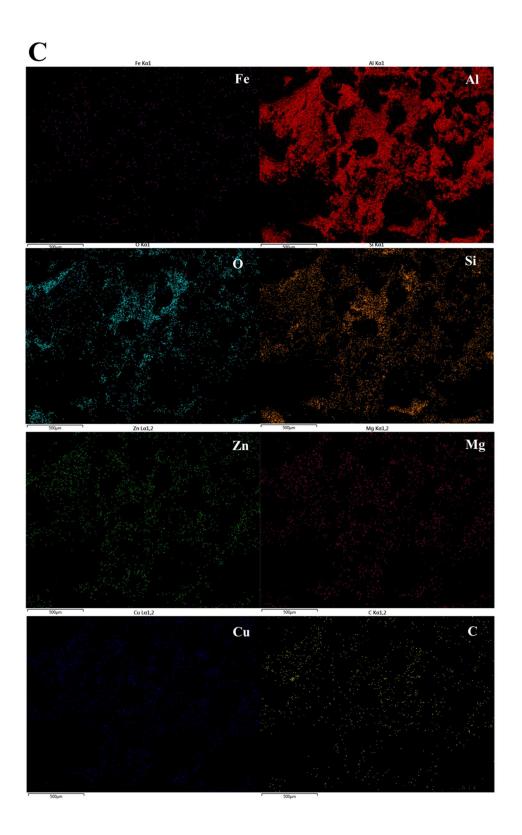


Fig. S6 Venn diagram on OTU level in different treatments







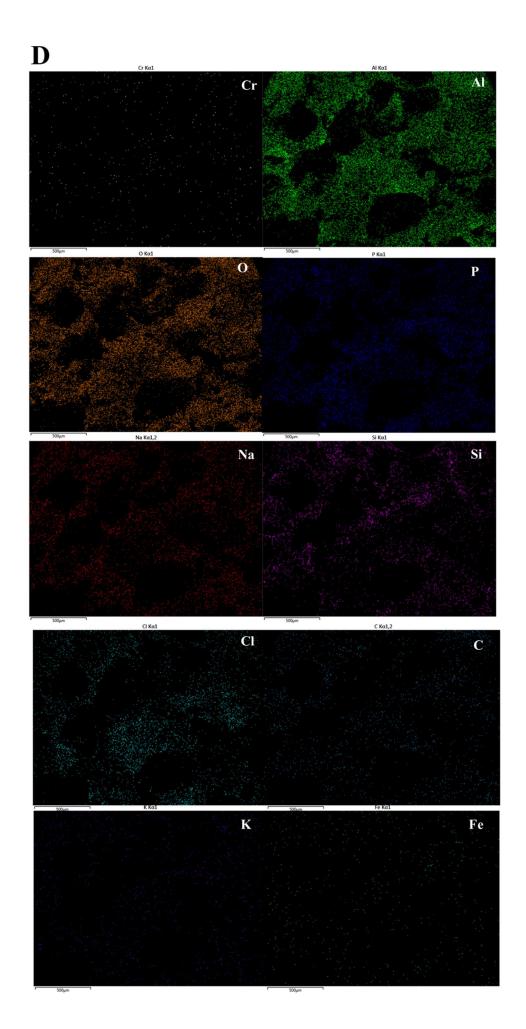


Fig. S7 Characterization of electrode materials before and after operation by EDS mapping. (A) EDS image of cathode loaded with Fe<sub>3</sub>O<sub>4</sub>; (B) EDS image of cathode after the SMFC operation; (C) EDS image of anode microorganisms; (D) EDS image of the anode after SMFC operation.

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