

# Supplement to: Temperature-dependent rate coefficients for the reaction of OH radicals with selected alkanes, aromatic compounds and monoterpenes

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## S1 Data of reaction coefficient measurements

**Table S1.** OH reaction rates of methane measured in this study.

canister A		canister B	
<i>T</i> / K	<i>k</i> / 10 <sup>-14</sup> cm <sup>3</sup> s <sup>-1</sup>	<i>T</i> / K	<i>k</i> / 10 <sup>-14</sup> cm <sup>3</sup> s <sup>-1</sup>
340.3 ± 0.6	1.40 ± 0.07	340.7 ± 0.6	1.32 ± 0.06
331.1 ± 0.5	1.12 ± 0.05	331.4 ± 0.5	1.14 ± 0.05
321.7 ± 0.5	0.99 ± 0.05	322.0 ± 0.5	0.96 ± 0.04
312.4 ± 0.4	0.83 ± 0.04	312.5 ± 0.4	0.84 ± 0.04
303.1 ± 0.4	0.73 ± 0.03	303.2 ± 0.4	0.71 ± 0.03
293.8 ± 0.5	0.62 ± 0.03	293.8 ± 0.4	0.60 ± 0.03
286.0 ± 0.6	0.50 ± 0.02	284.4 ± 0.5	0.48 ± 0.03

**Table S2.** OH reaction rates of ethane measured in this study.

canister A		canister B		canister C	
$T/K$	$k/10^{-13}\text{cm}^3\text{s}^{-1}$	$T/K$	$k/10^{-13}\text{cm}^3\text{s}^{-1}$	$T/K$	$k/10^{-13}\text{cm}^3\text{s}^{-1}$
$340.2 \pm 0.6$	$3.85 \pm 0.14$	$340.4 \pm 0.5$	$3.84 \pm 0.08$	$340.4 \pm 0.6$	$3.73 \pm 0.07$
$331.0 \pm 0.5$	$3.66 \pm 0.21$	$331.0 \pm 0.5$	$3.44 \pm 0.11$	$331.2 \pm 0.5$	$3.41 \pm 0.12$
$321.7 \pm 0.4$	$3.22 \pm 0.15$	$321.7 \pm 0.4$	$3.14 \pm 0.09$	$321.8 \pm 0.4$	$3.21 \pm 0.12$
$312.4 \pm 0.4$	$3.07 \pm 0.15$	$312.3 \pm 0.4$	$2.85 \pm 0.13$	$312.4 \pm 0.4$	$2.94 \pm 0.08$
		$303.0 \pm 0.4$	$2.62 \pm 0.11$	$303.1 \pm 0.4$	$2.65 \pm 0.06$
		$293.7 \pm 0.5$	$2.31 \pm 0.06$	$293.7 \pm 0.4$	$2.36 \pm 0.04$
		$284.4 \pm 0.6$	$2.06 \pm 0.05$	$284.6 \pm 0.5$	$2.09 \pm 0.06$

**Table S3.** OH reaction rates of propane measured in this study. Propane was sampled from the same canister in all measurements.

series A		series B		series C	
$T/K$	$k/10^{-12}\text{cm}^3\text{s}^{-1}$	$T/K$	$k/10^{-12}\text{cm}^3\text{s}^{-1}$	$T/K$	$k/10^{-12}\text{cm}^3\text{s}^{-1}$
$340.4 \pm 0.7$	$1.46 \pm 0.05$	$339.9 \pm 0.7$	$1.45 \pm 0.04$	$340.2 \pm 0.6$	$1.52 \pm 0.07$
$331.2 \pm 0.5$	$1.35 \pm 0.04$	$321.4 \pm 0.4$	$1.26 \pm 0.03$	$331.0 \pm 0.4$	$1.39 \pm 0.07$
				$321.7 \pm 0.4$	$1.30 \pm 0.07$
$312.2 \pm 0.4$	$1.19 \pm 0.03$			$312.4 \pm 0.4$	$1.22 \pm 0.06$
		$302.8 \pm 0.5$	$1.12 \pm 0.03$	$303.1 \pm 0.4$	$1.16 \pm 0.06$
				$293.7 \pm 0.5$	$1.08 \pm 0.06$
				$284.8 \pm 0.5$	$1.01 \pm 0.06$
series D		series E			
$T/K$	$k/10^{-12}\text{cm}^3\text{s}^{-1}$	$T/K$	$k/10^{-12}\text{cm}^3\text{s}^{-1}$		
$340.4 \pm 0.6$	$1.45 \pm 0.03$	$340.4 \pm 0.6$	$1.47 \pm 0.05$		
$321.8 \pm 0.4$	$1.29 \pm 0.04$	$321.7 \pm 0.4$	$1.33 \pm 0.05$		
$303.0 \pm 0.4$	$1.16 \pm 0.04$	$303.0 \pm 0.4$	$1.17 \pm 0.03$		
$284.7 \pm 0.6$	$0.99 \pm 0.04$	$284.4 \pm 0.6$	$0.98 \pm 0.04$		

**Table S4.** OH reaction rates of n-butane measured in this study.

canister A		canister B		canister C	
$T/K$	$k/10^{-12}\text{cm}^3\text{s}^{-1}$	$T/K$	$k/10^{-12}\text{cm}^3\text{s}^{-1}$	$T/K$	$k/10^{-12}\text{cm}^3\text{s}^{-1}$
$340.1 \pm 0.7$	$2.82 \pm 0.11$	$340.3 \pm 0.6$	$2.98 \pm 0.08$	$340.5 \pm 0.6$	$3.17 \pm 0.08$
$331.1 \pm 0.6$	$2.84 \pm 0.07$	$331.2 \pm 0.5$	$2.88 \pm 0.08$	$331.3 \pm 0.4$	$2.98 \pm 0.12$
$321.9 \pm 0.5$	$2.71 \pm 0.06$	$321.8 \pm 0.4$	$2.80 \pm 0.06$	$322.0 \pm 0.4$	$3.08 \pm 0.14$
$312.5 \pm 0.4$	$2.60 \pm 0.07$	$312.5 \pm 0.4$	$2.67 \pm 0.07$	$312.6 \pm 0.4$	$2.78 \pm 0.11$
$303.1 \pm 0.4$	$2.55 \pm 0.05$	$303.1 \pm 0.4$	$2.60 \pm 0.05$	$303.2 \pm 0.4$	$2.58 \pm 0.09$
$293.9 \pm 0.5$	$2.43 \pm 0.04$	$293.7 \pm 0.5$	$2.52 \pm 0.06$	$293.8 \pm 0.4$	$2.56 \pm 0.14$
$285.3 \pm 0.6$	$2.29 \pm 0.05$	$284.5 \pm 0.5$	$2.33 \pm 0.06$		

**Table S5.** OH reaction rates of methyl vinyl ketone (MVK) measured in this study.

canister A		canister B		canister C	
$T/K$	$k/10^{-11}\text{cm}^3\text{s}^{-1}$	$T/K$	$k/10^{-11}\text{cm}^3\text{s}^{-1}$	$T/K$	$k/10^{-11}\text{cm}^3\text{s}^{-1}$
$340.6 \pm 0.5$	$1.51 \pm 0.03$	$340.4 \pm 0.5$	$1.58 \pm 0.04$	$340.5 \pm 0.5$	$1.53 \pm 0.04$
$331.3 \pm 0.5$	$1.62 \pm 0.04$	$331.3 \pm 0.5$	$1.69 \pm 0.04$	$331.3 \pm 0.5$	$1.67 \pm 0.05$
$321.9 \pm 0.4$	$1.72 \pm 0.05$	$321.8 \pm 0.4$	$1.80 \pm 0.04$	$321.9 \pm 0.4$	$1.78 \pm 0.04$
$312.5 \pm 0.4$	$1.84 \pm 0.05$	$312.5 \pm 0.4$	$1.93 \pm 0.05$	$312.5 \pm 0.4$	$1.91 \pm 0.06$
$303.1 \pm 0.4$	$1.99 \pm 0.04$	$303.1 \pm 0.4$	$2.05 \pm 0.05$	$303.1 \pm 0.4$	$1.98 \pm 0.04$
$293.8 \pm 0.5$	$2.15 \pm 0.05$	$293.8 \pm 0.4$	$2.26 \pm 0.06$	$293.8 \pm 0.5$	$2.21 \pm 0.05$
$285.2 \pm 0.5$	$2.21 \pm 0.11$	$284.5 \pm 0.5$	$2.35 \pm 0.06$	$284.7 \pm 0.5$	$2.42 \pm 0.06$

**Table S6.** OH reaction rates of myrcene measured in this study.

canister A		canister B		canister C	
$T/K$	$k/10^{-10}\text{cm}^3\text{s}^{-1}$	$T/K$	$k/10^{-10}\text{cm}^3\text{s}^{-1}$	$T/K$	$k/10^{-10}\text{cm}^3\text{s}^{-1}$
$340.5 \pm 0.6$	$1.14 \pm 0.04$	$340.6 \pm 0.5$	$1.20 \pm 0.02$	$340.7 \pm 0.5$	$1.17 \pm 0.03$
$331.3 \pm 0.4$	$1.25 \pm 0.03$	$331.3 \pm 0.4$	$1.32 \pm 0.03$	$331.4 \pm 0.4$	$1.33 \pm 0.04$
$321.9 \pm 0.4$	$1.40 \pm 0.03$	$322.0 \pm 0.4$	$1.46 \pm 0.07$	$322.1 \pm 0.4$	$1.42 \pm 0.06$
$312.6 \pm 0.4$	$1.53 \pm 0.04$	$312.6 \pm 0.4$	$1.57 \pm 0.04$	$312.8 \pm 0.4$	$1.51 \pm 0.05$
$303.1 \pm 0.4$	$1.64 \pm 0.04$	$303.3 \pm 0.4$	$1.62 \pm 0.05$	$303.3 \pm 0.4$	$1.68 \pm 0.07$
$293.8 \pm 0.5$	$1.77 \pm 0.03$	$293.9 \pm 0.5$	$1.76 \pm 0.06$	$293.9 \pm 0.5$	$1.73 \pm 0.04$
$284.5 \pm 0.6$	$1.99 \pm 0.06$	$284.8 \pm 0.5$	$1.92 \pm 0.07$	$285.4 \pm 0.5$	$1.88 \pm 0.07$

**Table S7.** OH reaction rates of  $\Delta^3$ -carene measured in this study.

canister A		canister B		canister C	
$T/K$	$k/10^{-11}\text{cm}^3\text{s}^{-1}$	$T/K$	$k/10^{-11}\text{cm}^3\text{s}^{-1}$	$T/K$	$k/10^{-11}\text{cm}^3\text{s}^{-1}$
$340.8 \pm 0.6$	$7.3 \pm 0.3$	$340.6 \pm 0.5$	$7.53 \pm 0.17$	$340.6 \pm 0.6$	$7.65 \pm 0.13$
$331.4 \pm 0.5$	$7.25 \pm 0.14$	$331.2 \pm 0.5$	$7.33 \pm 0.13$	$331.3 \pm 0.5$	$7.54 \pm 0.17$
$322.0 \pm 0.4$	$7.43 \pm 0.17$	$321.9 \pm 0.4$	$7.76 \pm 0.15$	$321.9 \pm 0.4$	$7.6 \pm 0.3$
$312.6 \pm 0.4$	$7.55 \pm 0.14$	$312.5 \pm 0.4$	$8.02 \pm 0.16$	$312.5 \pm 0.4$	$8.1 \pm 0.3$
$303.3 \pm 0.4$	$7.79 \pm 0.15$	$303.1 \pm 0.4$	$8.29 \pm 0.17$	$303.2 \pm 0.4$	$8.2 \pm 0.2$
$293.9 \pm 0.4$	$8.29 \pm 0.16$	$293.9 \pm 0.4$	$8.7 \pm 0.3$	$293.8 \pm 0.4$	$8.69 \pm 0.16$
$284.5 \pm 0.5$	$8.5 \pm 0.3$	$284.4 \pm 0.5$	$9.3 \pm 0.3$	$284.5 \pm 0.5$	$9.0 \pm 0.6$

**Table S8.** OH reaction rates of  $\gamma$ -terpinene measured in this study.

canister A		canister B		canister C	
$T/K$	$k/10^{-10}\text{cm}^3\text{s}^{-1}$	$T/K$	$k/10^{-10}\text{cm}^3\text{s}^{-1}$	$T/K$	$k/10^{-10}\text{cm}^3\text{s}^{-1}$
$340.5 \pm 0.5$	$1.06 \pm 0.02$	$340.5 \pm 0.5$	$1.06 \pm 0.03$	$340.4 \pm 0.5$	$1.10 \pm 0.03$
$331.2 \pm 0.4$	$1.21 \pm 0.03$	$331.1 \pm 0.4$	$1.22 \pm 0.03$	$331.2 \pm 0.5$	$1.25 \pm 0.03$
$321.9 \pm 0.4$	$1.34 \pm 0.04$	$321.8 \pm 0.4$	$1.29 \pm 0.04$	$321.9 \pm 0.4$	$1.36 \pm 0.03$
$312.5 \pm 0.4$	$1.42 \pm 0.03$	$312.5 \pm 0.4$	$1.45 \pm 0.03$	$312.4 \pm 0.4$	$1.48 \pm 0.05$
$303.1 \pm 0.4$	$1.61 \pm 0.03$	$303.1 \pm 0.4$	$1.51 \pm 0.03$	$303.1 \pm 0.4$	$1.56 \pm 0.04$
$293.9 \pm 0.4$	$1.68 \pm 0.03$	$293.7 \pm 0.5$	$1.65 \pm 0.03$	$293.8 \pm 0.4$	$1.67 \pm 0.04$
$284.6 \pm 0.5$	$1.77 \pm 0.05$	$284.5 \pm 0.5$	$1.97 \pm 0.05$	$284.5 \pm 0.5$	$1.86 \pm 0.05$

**Table S9.** OH reaction rates of toluene measured in this study.

canister A		canister B		canister C	
$T/K$	$k/10^{-12}\text{cm}^3\text{s}^{-1}$	$T/K$	$k/10^{-12}\text{cm}^3\text{s}^{-1}$	$T/K$	$k/10^{-12}\text{cm}^3\text{s}^{-1}$
$340.5 \pm 0.5$	$4.73 \pm 0.11$	$340.4 \pm 0.5$	$4.62 \pm 0.15$	$340.5 \pm 0.5$	$4.77 \pm 0.11$
$331.1 \pm 0.5$	$5.13 \pm 0.13$	$331.2 \pm 0.5$	$5.03 \pm 0.10$	$331.2 \pm 0.5$	$5.12 \pm 0.16$
$321.8 \pm 0.4$	$5.36 \pm 0.12$	$321.8 \pm 0.4$	$5.54 \pm 0.18$	$321.9 \pm 0.4$	$5.44 \pm 0.14$
$312.4 \pm 0.4$	$5.58 \pm 0.15$	$312.5 \pm 0.4$	$5.77 \pm 0.20$	$312.6 \pm 0.4$	$5.63 \pm 0.24$
$303.2 \pm 0.4$	$5.77 \pm 0.09$	$303.2 \pm 0.4$	$6.09 \pm 0.18$	$303.2 \pm 0.4$	$5.87 \pm 0.14$
$293.9 \pm 0.4$	$6.01 \pm 0.11$	$293.8 \pm 0.4$	$6.11 \pm 0.31$	$293.8 \pm 0.4$	$6.01 \pm 0.14$
$284.5 \pm 0.5$	$6.14 \pm 0.11$	$284.5 \pm 0.5$	$6.08 \pm 0.23$	$285.5 \pm 0.5$	$5.90 \pm 0.18$

**Table S10.** OH reaction rates of mesitylene measured in this study.

canister A		canister B		canister C	
$T/K$	$k/10^{-11}\text{cm}^3\text{s}^{-1}$	$T/K$	$k/10^{-11}\text{cm}^3\text{s}^{-1}$	$T/K$	$k/10^{-11}\text{cm}^3\text{s}^{-1}$
$340.6 \pm 0.5$	$3.99 \pm 0.08$	$340.5 \pm 0.5$	$4.03 \pm 0.08$	$340.5 \pm 0.6$	$4.06 \pm 0.12$
$331.1 \pm 0.4$	$4.35 \pm 0.09$	$331.2 \pm 0.4$	$4.48 \pm 0.09$	$331.3 \pm 0.4$	$4.56 \pm 0.09$
$321.8 \pm 0.4$	$4.92 \pm 0.13$	$321.8 \pm 0.4$	$4.85 \pm 0.10$	$321.9 \pm 0.4$	$4.98 \pm 0.11$
$312.5 \pm 0.4$	$5.25 \pm 0.11$	$312.4 \pm 0.4$	$5.28 \pm 0.12$	$312.5 \pm 0.4$	$5.37 \pm 0.13$
$303.1 \pm 0.4$	$5.57 \pm 0.17$	$303.2 \pm 0.4$	$5.69 \pm 0.10$	$303.1 \pm 0.4$	$5.77 \pm 0.10$
$293.8 \pm 0.4$	$6.41 \pm 0.16$	$293.9 \pm 0.4$	$6.17 \pm 0.15$	$293.8 \pm 0.4$	$6.27 \pm 0.14$
$284.5 \pm 0.5$	$6.64 \pm 0.12$	$284.6 \pm 0.5$	$6.63 \pm 0.13$	$284.6 \pm 0.5$	$6.65 \pm 0.17$

**Table S11.** OH reaction rates of m-xylene measured in this study.

canister A		canister B		canister C	
$T/K$	$k/10^{-11}\text{cm}^3\text{s}^{-1}$	$T/K$	$k/10^{-11}\text{cm}^3\text{s}^{-1}$	$T/K$	$k/10^{-11}\text{cm}^3\text{s}^{-1}$
$340.5 \pm 0.5$	$1.66 \pm 0.04$	$340.4 \pm 0.5$	$1.72 \pm 0.06$	$340.6 \pm 0.5$	$1.70 \pm 0.06$
$331.2 \pm 0.5$	$1.80 \pm 0.05$	$331.3 \pm 0.4$	$1.92 \pm 0.05$	$331.5 \pm 0.4$	$1.94 \pm 0.06$
$321.8 \pm 0.4$	$2.01 \pm 0.05$	$321.9 \pm 0.4$	$2.05 \pm 0.05$	$322.2 \pm 0.4$	$2.03 \pm 0.08$
$312.5 \pm 0.4$	$2.16 \pm 0.06$	$312.5 \pm 0.4$	$2.22 \pm 0.05$	$312.8 \pm 0.4$	$2.22 \pm 0.09$
$303.1 \pm 0.4$	$2.41 \pm 0.06$	$303.2 \pm 0.4$	$2.37 \pm 0.06$	$303.2 \pm 0.4$	$2.30 \pm 0.06$
$293.9 \pm 0.5$	$2.62 \pm 0.07$	$293.9 \pm 0.4$	$2.60 \pm 0.04$	$294.1 \pm 0.4$	$2.56 \pm 0.07$
$285.0 \pm 0.5$	$2.79 \pm 0.11$	$285.0 \pm 0.6$	$2.72 \pm 0.07$		

**Table S12.** OH reaction rates of o-xylene measured in this study.

canister A		canister B		canister C	
$T/K$	$k/10^{-11}\text{cm}^3\text{s}^{-1}$	$T/K$	$k/10^{-11}\text{cm}^3\text{s}^{-1}$	$T/K$	$k/10^{-11}\text{cm}^3\text{s}^{-1}$
$340.5 \pm 0.5$	$1.05 \pm 0.03$	$340.6 \pm 0.6$	$1.06 \pm 0.03$	$340.6 \pm 0.6$	$1.03 \pm 0.03$
$331.3 \pm 0.4$	$1.13 \pm 0.04$	$331.3 \pm 0.5$	$1.13 \pm 0.03$		
$321.9 \pm 0.4$	$1.22 \pm 0.04$	$321.9 \pm 0.4$	$1.22 \pm 0.03$	$322.1 \pm 0.4$	$1.16 \pm 0.03$
$312.5 \pm 0.4$	$1.28 \pm 0.03$	$312.6 \pm 0.4$	$1.28 \pm 0.04$	$312.7 \pm 0.4$	$1.29 \pm 0.04$
$303.1 \pm 0.4$	$1.34 \pm 0.03$	$303.2 \pm 0.4$	$1.34 \pm 0.03$	$303.4 \pm 0.4$	$1.31 \pm 0.04$
$293.8 \pm 0.4$	$1.43 \pm 0.03$	$293.9 \pm 0.4$	$1.43 \pm 0.03$	$294.0 \pm 0.5$	$1.40 \pm 0.04$
$285.1 \pm 0.5$	$1.44 \pm 0.05$	$285.6 \pm 0.5$	$1.43 \pm 0.05$	$286.2 \pm 0.5$	$1.41 \pm 0.06$