

1 **Measurement report: The variation properties of aerosol hygroscopic
2 growth related to chemical composition during new particle formation
3 days in a coastal city of southeast China**

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15
16 **Abstract.** The scattering of solar radiation by aerosol is significantly affected by
17 relative humidity (RH) due to the aerosol hygroscopicity. In order to better understand
18 the characteristics of aerosol scattering hygroscopic growth and its influencing factors
19 during new particle formation (NPF) days, we conducted the in-situ campaign from
20 February to April 2022 in Xiamen, a coastal city in southeastern China. The aerosol
21 scattering hygroscopic growth factor $f(RH)$, commonly used to describe the aerosol
22 indirectly hygroscopicity, varies greatly due to the influence of aerosol chemical
23 composition and size. In the relatively clean atmosphere of Xiamen, NPF events occur
24 frequently, and the variation in chemical composition during the events has a substantial
25 influence on the aerosol scattering hygroscopic growth. In this study, we investigated
26 the features and influencing factors of $f(RH)$ in the NPF days. The research results
27 emphasized that $f(RH)$ differed significantly between NPF and non-NPF days, mainly
28 impacted by the aerosol chemical compositions, especially sulfate and nitrate. In the
29 NPF days, sulfate was the dominant contributor to $f(RH)$, distinguishing from the non-
30 NPF days. Aerosol hygroscopicity-chemical composition closure demonstrated that
31 NH_4HSO_4 was the main driving force (30.78%) of the hygroscopicity parameter $\kappa_{f(RH)}$
32 when NPF events happened, while NH_4NO_3 played a dominant role in $\kappa_{f(RH)}$ (up to 35%)
33 for non-NPF days. Although the uncertainty of the organic aerosol (OA) to
34 hygroscopicity might exist due to the varieties of chemical components and oxidation
35 level, it was the crucial driving factor for the variation in aerosol hygroscopicity. The
36 findings of this study would be helpful for the further understanding about the
37 properties of aerosol hygroscopicity in the coastal area, as well as complementing the
38 hygroscopic growth factors to the models of air quality and climate change.
39

40 **1 Introduction**

41 Atmospheric aerosols have direct and indirect effects on atmospheric visibility, the
42 earth-atmosphere radiation budget, clouds and precipitation, which in turn affect
43 climate (Charlson et al., 1992); and these effects are strongly dependent on the
44 hygroscopic properties of the ambient aerosol and the relative humidity (RH). The
45 aerosol optical properties are key parameters for accurately estimating the direct
46 radiative forcing caused by aerosols in climate models (IPCC, 2021). The aerosol
47 hygroscopicity has a significant impact on the optical properties by altering particle size
48 and refractive index, and ultimately on the climatic and environmental effects of
49 aerosols (Covert et al., 1972; Tang, 1996; Malm and Day, 2001). Particles absorb water
50 through hygroscopic growth, growing to sizes that are more efficient for light scattering,
51 and their refractive index changes, resulting in enhanced aerosol scattering (Seinfeld et
52 al., 1998). The aerosol liquid water content (ALWC) increases as particles absorb water
53 in humid environment due to aerosol hygroscopicity. The condensed water in aerosols
54 serves as an effective medium for multiphase chemistry, promoting the transformation
55 of active gaseous pollutants into particles. Meanwhile, the newly formed hygroscopic
56 aerosol components, such as secondary aerosols, can also alter aerosol hygroscopic

57 behaviors and enhance aerosol extinction efficiency. These processes lead to regional
58 visibility impairment and accelerated formation of heavy haze. Therefore, aerosol
59 hygroscopicity can profoundly affect atmospheric chemical processes (Wu et al., 2018)
60 and air quality (Liu et al., 2020a).

61 Aerosol hygroscopic growth become the main factors affecting the aerosol optical
62 properties at high ambient humidity due to the enhanced aerosol hygroscopicity and
63 increased RH (Jin et al., 2022). The aerosol scattering hygroscopic growth factor ($f(RH)$)
64 is defined as the ratio of aerosol scattering coefficient at an elevated RH level ($\sigma_{sp}(RH, \lambda)$)
65 to that under dry condition ($\sigma_{sp}(RH_{dry}, \lambda)$) (usually $RH < 40\%$) at a given
66 wavelength. The scattering enhancement owing to hygroscopic growth strongly
67 depends on the source of the aerosol, which varies in chemical composition (Yan et al.,
68 2009; Sheridan et al., 2002; Fierz-Schmidhauser et al., 2010a; Kotchenruther and Hobbs,
69 1998). For example, marine aerosols tends to have higher $f(RH)$ than urban or
70 continental aerosols, and their $f(RH)$ decreases under strong anthropogenic influence
71 (Zieger et al., 2010; Yan et al., 2009; Sheridan et al., 2001). Mineral dust and freshly
72 emitted biomass combustion aerosols exhibit the lowest $f(RH)$ values among the
73 aerosol types studied (Sheridan et al., 2002; Pan et al., 2009; Fierz-Schmidhauser et al.,
74 2010a). Hydrophilic species such as secondary inorganic components, sea salts, and
75 water-soluble organics in the aerosol are the main contributors to the hygroscopic
76 growth (Li et al., 2021; Fierz-Schmidhauser et al., 2010a), while black carbon and some
77 organic carbons are the major proportion of the hydrophobic species (Liu and Zhang,
78 2010). Thus, discrepancies in the chemical composition of the aerosol, the fraction of
79 soluble and insoluble chemical components, i.e., lead to variations of $f(RH)$ (Malm et
80 al., 2005; Zieger et al., 2014; Zhang et al., 2015). Additionally, particle number size
81 distribution (PNSD) is another factor affecting $f(RH)$. For a fixed chemical composition,
82 $f(RH)$ decreases with increasing particle size (Fierz-Schmidhauser et al., 2010b; Zieger
83 et al., 2010; Baynard et al., 2006). **Particle morphology, including shape and mixing**
84 **state, significantly affects aerosol hygroscopicity.** Giordano et al. (2015) demonstrated
85 that the non-spherical shape of particles not only influences their hygroscopic behavior,
86 but also changes in the mixing state of particles—whether they are internally or
87 externally mixed—fluence their hygroscopic properties (Stevens and Dastoor, 2019).
88 Noted that the aerosol chemical compositions, particle sizes and morphology have a
89 combined impact on $f(RH)$ in the atmospheric environment, as the two are closely
90 related (Wu et al., 2017; Kreidenweis and Asa-Awuku, 2014).

91 The observation site is situated in Xiamen, a fast-urbanization coastal city in
92 southeastern China, at the junction of land and sea. **As a result of massive population**
93 **growth and rapid economic development, its atmospheric environment is subjected to**
94 **complex pollution situations, such as the increased atmospheric oxidation (Liu et al.,**
95 **2022) and relative high nitrogen oxide pollution (Chen et al., 2023).** Notably, despite
96 these conditions, aerosol concentrations remain generally lower compared to those in
97 highly polluted megacities in China, according to open-access data from China
98 Environmental Monitoring Centre (<https://www.zq12369.com/index.php>). On the other

99 hand, it is located at a subtropical city with a relative high air temperatures and high
100 RH. High RH not only directly increases light scattering, leading to the decline of
101 visibility (Won et al., 2021), but also affects the aerosol chemical processes involved in
102 the particle formation(Sun et al., 2013; Chen et al., 2021). New particle formation (NPF)
103 events occur frequently in this coastal city in southeast China with relative clean air
104 quality (Wang et al., 2022). NPF is a process that low-volatile compounds emitted from
105 natural or anthropogenic sources form into thermodynamically stable molecular
106 clusters and grow into larger particles via condensation or collision with other vapours
107 or particles (Holmes, 2007). When NPF occurs, both the PNSD and the chemical
108 composition of the aerosol undergo significant changes, which have a remarkable
109 influence on the aerosol hygroscopicity and $f(RH)$. Previous studies of $f(RH)$ had been
110 conducted mainly in the megacity agglomerations such as North China Plain, Yangtze
111 River Delta and Pearl River Delta (Liu et al., 2012; Xia et al., 2019; Ding et al., 2021;
112 Jin et al., 2022; Zhao et al., 2019a), while limited attention had been paid to the
113 southeast coastal areas with relative low level of particle and high RH. Meanwhile,
114 these studies have focused more on the effect of aerosol chemical composition on $f(RH)$
115 (Li et al., 2021; Wang et al., 2021; Jin et al., 2022), however, the exploration to the
116 variation of $f(RH)$ during NPF days is quite few in China.

117 **In order to investigate the characteristics of $f(RH)$ and its influencing factors during
118 the NPF days, the enhanced observations utilized a high-resolution, humidified
119 nephelometer system combined with the PNSD instruments to measure $f(RH)$ for RH
120 ranging from 40% to 91% in urban Xiamen. Routine measurements including other
121 aerosol chemical and physical properties were also synchronously conducted.
122 Differences and variations in $f(RH)$ between NPF and non-NPF days were explored and
123 the effect of aerosol chemical compositions on $f(RH)$ were also discussed. The research
124 was expected to characterize the properties of aerosol hygroscopicity during the NPF
125 and non-NPF days in coastal area with relative low level of particle and high RH, and
126 provide references to the model improvement for air quality and climate change.**
127

128 **2 Instruments and methods**

129 **2.1 Observation site**

130 The enhanced observations were carried out at the Institute of Urban Environment
131 of the Chinese Academy of Sciences in Xiamen (IUE, CAS), which is situated on the
132 west coast of the Taiwan Strait. The observation station (118°03'E, 24°36'N) was
133 located on the roof of an 80m-heigh building, a typical urban site surrounded by two
134 main trunk roads (Jimei main road and Haixiang express road), shopping malls,
135 educational institutions, and residential areas, and there was no apparent industrial
136 emission sources nearby. Thus, the collected data can accurately represent the average
137 air quality levels in the urban area of Xiamen. The observations were conducted
138 consecutively from the 1st February to the 30th April 2022.

139 **2.2 Observation instruments**

140 The $f(RH)$ values were obtained using a multi-band dual-nephelometer system

141 (PB-FRH100, BMET, China) comprising a nephelometer for aerosol scattering
142 coefficients under dry conditions and another nephelometer for humidified aerosols.
143 The airflow initially passed through two tandem Nafion dryers which could decrease
144 the RH of the airflow to less than 30%. After this, the airflow was divided into two
145 routes, one was directed straight into the nephelometer, while the other was humidified
146 via a Gore-Tex tube set in a stainless steel tube before flowing into the nephelometer.
147 The space between these two tubes contained circulating water, which was heated by
148 two water baths. The scattering coefficients of dry and humidified PM_{2.5} were measured
149 at three wavelengths (450, 525 and 635 nm) using two nephelometers. The detailed
150 principles and operation of the system has been described in Supplement material (Text
151 S1). This study set the minimum and maximum RH at 40% and 91%, respectively, with
152 a 45-minute cycle for humidification.

153 An integrating nephelometer (Aurora-3000, Ecotech, Australia) was used to
154 simultaneously and continuously measure the 5-min average σ_{sp} at the same three
155 wavelengths, and the σ_{sp} at 525 nm was appropriate for characterizing the aerosol
156 scattering coefficient in this study. A Scanning Mobility Particle Sizer (SMPS, model
157 3938 L50, TSI Inc., USA), integrated with a Differential Mobility Analyzer (DMA,
158 model 3082, TSI Inc., USA), a butanol-based Condensation Particle Counter (CPC,
159 model 3750, TSI Inc., USA) and an aerosol neutralizer, were used to continuously
160 measure the PNSD in the range of 7-300 nm over a 5-minute scanning interval during
161 the measurement. The hourly chemical composition of aerosol, including sulfate (SO₄),
162 nitrate (NO₃), ammonium (NH₄), chloride (Chl) and **organics (Org)**, was measured by
163 a high-resolution Aerodyne Aerosol Chemical Speciation Monitor (Q-ACSM). The
164 relative ion efficiency (RIE) for SO₄, NO₃, NH₄, Chl and **Org** was 0.53, 1.1, 5.49, 1.3
165 and 1.4, respectively. PMF/ME-2 models were performed to identify OA factors
166 resolving primary organic aerosol (POA) and secondary organic aerosol (SOA) in this
167 study. POA is the unoxygenated component, and SOA is a more oxygenated organic
168 aerosol. **More detailed descriptions on the PMF analysis are given in Text S2.** An AE-
169 31 aethalometer (Magee Scientific, USA) was used to measure the black carbon (BC)
170 aerosol concentrations. Hourly mass concentrations of PM_{2.5} and PM₁ were measured
171 by a Tapered Element Oscillating Microbalance (TEOM1405, Thermo Scientific Corp.,
172 USA). Ambient meteorological parameters, including air temperature (T), RH, wind
173 speed (WS) and wind direction (WD), were continuously monitored by an ultrasonic
174 weather station instrument (150WX, Airmar, USA). Gaseous pollutants (carbon oxide,
175 sulfur dioxide, and nitrogen dioxide) were measured by online Thermo Instruments TEI
176 48i, 49i, 43i, and 42i (Thermo Scientific Corp., USA).

177

178 **2.3 Identification and classification of new particle formation**

179 The NPF process involves nucleating and growing. The particles nucleated at a
180 critical size of approximately 1.5 ± 0.4 nm (Kulmala et al., 2012) and then could grow
181 into larger particles. During the sampling period, a total of 85 days of valid observations
182 were available for PNSD analysis. The particles were divided into three modes:

183 nucleation mode (< 25 nm), Aitken mode (25-100 nm) and accumulation mode (> 100
184 nm) in this study (Kalkavouras et al., 2021; Shen et al., 2022; Wang et al., 2022). To
185 identify NPF events, the visual analysis of PNSD data described by Dal Maso et al.
186 (2005) was used. If a new particle mode is observed in the nucleation mode within a
187 few hours and the mode shows clear signs of growth, characterized by a distinct
188 “banana” shape in the time series of PNSD, then the day can be classified as an NPF
189 event day. With the exception of NPF days, all other days are classified as non-NPF
190 days or Undefined days. Non-NPF days are confirmed when there is clearly no
191 evidence of NPF or when the above criteria are not met. Besides, days that can not fulfil
192 the criteria to be identified as NPF or non-NPF days were classified as Undefined days,
193 and they are characterised by the presence of some particles in the nucleation mode
194 with no visible signs of growth, or by the observation of growth not in the nucleation
195 mode.

196 **2.4 Data processing**

197 The scattering Ångström exponent (α) indicates the wavelength dependence of
198 aerosol scattering, and the parameters related to aerosol size are relatively low for large
199 particles and relatively high for small particles(Guan et al., 2021). It is expressed as
200 follows:

$$201 \alpha = \frac{\log(\sigma_{sp,\lambda_1}) - \log(\sigma_{sp,\lambda_2})}{\log(\lambda_1) - \log(\lambda_2)} \quad (1)$$

202 Where, $\lambda_1=450$ nm and $\lambda_2=635$ nm in this study.

203 The overall hygroscopicity parameter $\kappa_{f(RH)}$ can be obtained from the measured
204 $f(RH)$. The detailed calculation procedure of this method is shown in Supplement
205 material ([Text S3](#)).

206 For the calculation of the hygroscopicity parameter κ based on the measured
207 aerosol chemical-composition data, we used the mass concentrations of OA, SO_4^{2-} ,
208 NO_3^- , NH_4^+ and Cl^- provided by ACSM. In this study, a simplified ion pairing scheme
209 proposed by Gysel et al. (2007) was used to obtain the concentrations of AN, AS, ABS,
210 and AC by turning mass concentrations of ions into mass concentrations of the
211 corresponding inorganic salts. The κ values and densities of these salts are shown in
212 Table S1 (Liu et al., 2014; Wu et al., 2016; Kuang et al., 2020). The simple mixing rule,
213 called Zdanovsky-Stokes-Robinson (ZSR), is commonly used in κ_{chem} calculations,
214 therefore, the κ_{chem} of this study can be calculated on the basis of chemical volume
215 fractions ε_i (Petters and Kreidenweis, 2007) (see [Text S4](#) for a detailed process).

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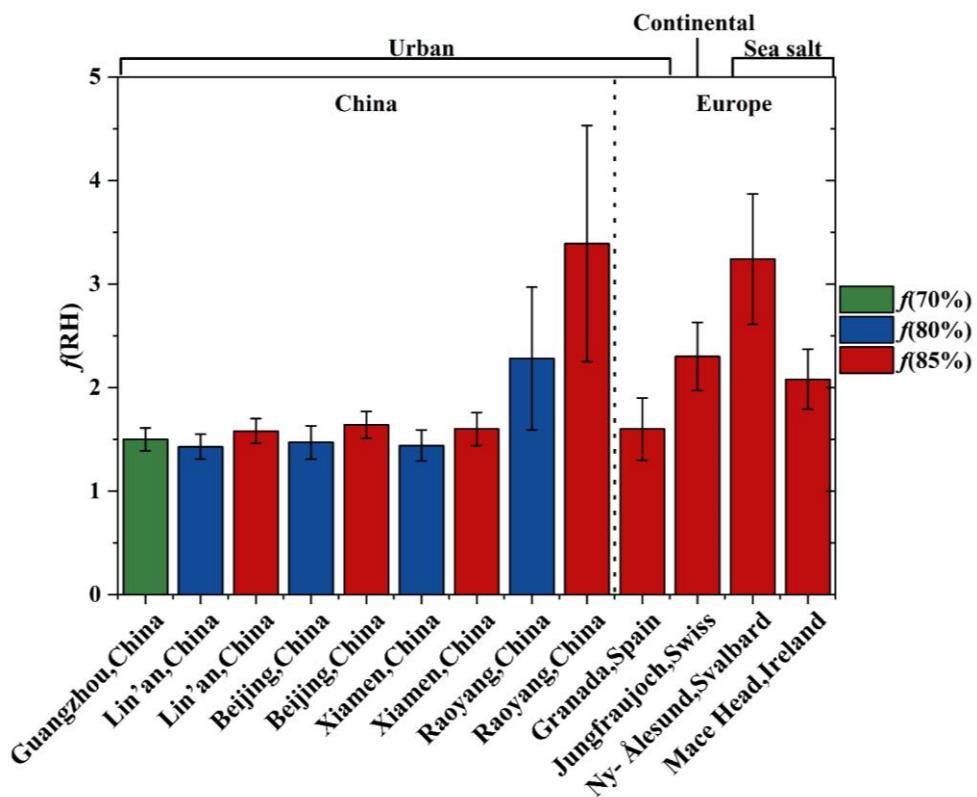
217 **3 Results and discussion**

218 **3.1 Overview of $f(RH)$ and derived aerosol variables observations**

219 The typical levels of light scattering coefficients (σ_{sp}), $f(RH)$ values at RH = 80%
220 ($f(80\%)$), scattering Ångström exponents (α), $\text{PM}_{2.5}$ mass concentrations, and ambient
221 RH from February to April 2022 were displayed in [Figure S3](#). To evaluate the aerosol
222 hygroscopicity conveniently, $f(80\%)$ was often employed (Xia et al., 2023; Xia et al.,
223 2019; Zhao et al., 2019b; Wu et al., 2017). The scattering hygroscopic growth factor

224 $f(80\%)$ ranged from 1.00 to 2.48, with an average factor of 1.44 ± 0.15 during the whole
225 campaign. The mean concentration of $PM_{2.5}$ was $24.79 \pm 17.74 \mu g m^{-3}$, suggested the
226 $PM_{2.5}$ pollution was relative low in Xiamen referring to the air quality index (AQI)
227 grading standard of China ($PM_{2.5} \leq 35 \mu g m^{-3}$). During the period of observation, the
228 hourly mean σ_{sp} measured under dry conditions varied from 1.15 to $662.57 Mm^{-1}$, with
229 a mean of $135.50 \pm 108.78 Mm^{-1}$, and its maximum occurred at the peak of $PM_{2.5}$
230 concentrations. The α was usually considered as an indicator of particle size, which was
231 high when $PM_{2.5}$ level was low, indicating that there were more fine particles existed in
232 $PM_{2.5}$ with low concentration. At the same time, the variation trend of $f(80\%)$ was
233 similar to that of α . Ambient RH fluctuated considerably and was generally at a high
234 level. The wind direction was more evenly distributed, and the wind speed was
235 relatively stable, concentrated in 1-3 m/s. The comparison presented in Figure 1 and
236 Table S2 implied that $f(RH)$ varied widely across different regions, with consistently
237 higher values observed in Europe compared to China. The differences between $f(RH)$
238 in this study and in other regions of China were smaller than those outside of China. In
239 urban China (Lin'an, Beijing, Guangzhou and Xiamen), $f(RH)$ was generally small
240 (Zhang et al., 2015; Zhao et al., 2019b; Ren et al., 2021; Li et al., 2021); whereas in
241 Raoyang, greatly influenced by anthropogenic polluted aerosols, $f(RH)$ raised
242 significantly (Wu et al., 2017). Urban area in Europe (Granada, Spain) also displayed
243 similar values of $f(RH)$ (Titos et al., 2014). Nevertheless, the $f(RH)$ of continental
244 aerosols in Europe (Jungfraujoch, Swiss) obviously surpassed what had been observed
245 in China (Zieger et al., 2013). In the Arctic (Ny- Ålesund, Svalbard) and the shore of
246 the ocean (Mace Head), $f(RH)$ showed high values due to the sea salt (Zieger et al.,
247 2010; Zieger et al., 2013). However, Mace Head, which was also exposed to air
248 pollution from the urban areas, had lower $f(RH)$ than the undisturbed Arctic. Thus, the
249 large variability of $f(RH)$ across measurement sites was primarily attributed to the
250 different aerosol sources and chemical composition.

251



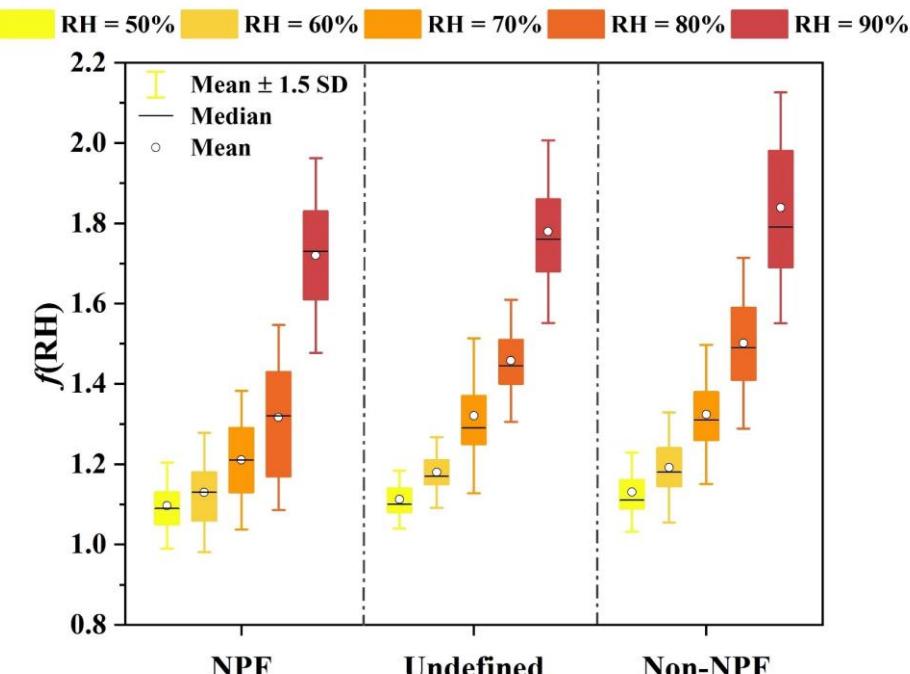
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Figure 1. Mean values of $f(80\%)$, $f(85\%)$ and $f(70\%)$ values in different observation sites. The error bars represent standard deviation. Blue, green, and red represent $f(80\%)$, $f(85\%)$ and $f(70\%)$, respectively.

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257

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Figure 2. The $f(RH)$ measured for a given RH in different days. The $f(RH)$ values at RH=50%, 60%, 70%, 80%, and 90% were counted for the three types of days

260 NPF, Undefined, and non-NPF, respectively. Different colours represent different RH.
261 The error bars represent the mean ± 1.5 standard deviation.

262
263 The entire campaign was devided into three type of days, NPF, Undefined, and
264 non-NPF, based on the above classification method. The PNSD spectrum and number
265 concentration of the example NPF and non-NPF days from February to April 2022 are
266 shown in [Figure S4](#) and [Text S5](#). 11 NPF days were identified, representing
267 approximately 12.94% of all observed days. In addition, we categorized 18 days as non-
268 NPF days, and the remaining days as Undefined. The statistical analysis of particle
269 number concentrations in different days during the observation period was summarized
270 in Table S3. The mean number concentration of nucleation, aitken and accumulation
271 mode was $1.66 \times 10^3 \pm 1.59 \times 10^3 \text{ cm}^{-3}$, $3.80 \times 10^3 \pm 2.80 \times 10^3 \text{ cm}^{-3}$ and $8.59 \times 10^2 \pm 4.04$
272 $\times 10^2 \text{ cm}^{-3}$ in NPF days, contributing 26.25%, 60.14% and 13.61% to total number
273 concentration, respectively. The majority of particle number concentrations were
274 comprised of nucleation and aitken mode particles. Figure 2 explores the variations of
275 $f(\text{RH})$ with RH growth in different days. Firstly, $f(\text{RH})$ emerged an approximately
276 exponential rise as RH increasing, with a significant growth when the RH ranged from
277 80% to 90%, the interval of which was the most beneficial to the aerosol scattering
278 hygroscopic growth (Liu et al., 2013). Moreover, Zhao et al. (2019b) found a prominent
279 difference in the aerosol hygroscopicity when RH exceeded 90%, with aerosol
280 hygroscopicity in this humidity range being lower than when RH was below 90%.
281 These results indicate that particles exhibited different hygroscopicity behaviors under
282 different RH conditions, which were probably related to the chemical composition,
283 particle size and morphology of aerosols. Secondly, the characteristics of $f(\text{RH})$ were
284 distinct among different days, especially between NPF days and non-NPF days ([Figure S5](#)).
285 When RH was below 80%, $f(\text{RH})$ was significantly lower during NPF days
286 compared to non-NPF and Undefined days. The $f(\text{RH})$ growth of NPF days was greater
287 than those of the other two sorts of days for RH between 80 and 90 %, and vice versa
288 for RH below 80%. Such a growth pattern caused the $f(\text{RH})$ of the NPF days reaching
289 a level equivalent to that of the other days as the RH rose to 90%. Moreover, the
290 fluctuations of $f(\text{RH})$ were larger in the NPF days than in the other two sorts of days,
291 indicating $f(\text{RH})$ in NPF days had a greater dispersion. These results might be related
292 to the dramatic increase in particle number concentrations and variations in chemical
293 composition during the NPF days.

294 The following discussion will focus on the differences in the aerosol scattering
295 hygroscopic growth and the aerosol hygroscopicity between NPF and non-NPF. As
296 Undefined days are in a transitional state, they do not accurately reflect the
297 characteristics of the NPF.

298 299 **3.2 Parameterization of the $f(\text{RH})$**

300 To better characterize the dependence of $f(\text{RH})$ on RH, many different empirical
301 expressions have been applied in previous studies to fit the measurements of $f(\text{RH})$

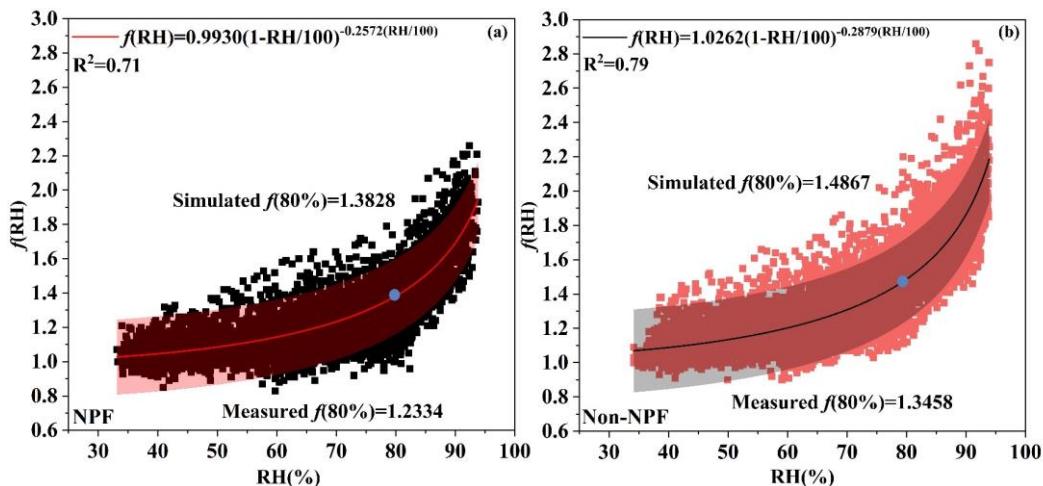
302 (Kotchenruther et al., 1999; Kotchenruther and Hobbs, 1998; Gasso et al., 2000; Carrico
303 et al., 2003; Pan et al., 2009; Zieger et al., 2014; Yu et al., 2018). We fitted four
304 commonly-used empirical equations to the $f(RH)$ values, and compared the results to
305 find that Eq. (2) (Chen et al., 2014) was the most suitable for describing the enhanced
306 scattering caused by the monotonic hygroscopic growth (see [Figure S6 and Text S6](#) for
307 a detailed comparison).

308
$$f(RH) = a(1 - \frac{RH}{100})^{-b(\frac{RH}{100})} \quad (2)$$

309 where a is a coefficient that reflects the level of $f(RH)$ values, and b is the parameter
310 for the magnitude of scattering enhancement unaffected by RH that quantifies aerosol
311 hygroscopicity to some extent. Higher $f(RH)$ is related to higher “ a ” and “ b ” values.

312 The fitted $f(RH)$ curves between NPF and non-NPF days over the entire
313 observation period are presented in Figure 3. For these days, Eq.(2) was a difference in
314 the fitting results, with fitting degrees being better in non-NPF days and relatively
315 worse in NPF days. This reflects the fact that $f(RH)$ on NPF days was influenced by
316 more complex factors than on non-NPF days, including the source, composition, and
317 morphology of the aerosols. The fitted $f(RH)$ curve shown in Figure 3(a) is apparently
318 below that shown in Figure 3(b). Similarly, both the observed $f(80\%)$ and the simulated
319 $f(80\%)$ in the NPF days were lower compared to those in the non-NPF days. The b was
320 lower for the NPF days, indicating that the aerosol hygroscopicity of the NPF days was
321 weaker than those of non-NPF days. In this work, b was lower than that in the study
322 done by Zhao et al. (2019b), but slightly higher than the findings of Chen et al. (2014)
323 (Table S4), even though both of their studies were conducted in the North China Plain
324 (NCP), where $PM_{2.5}$ concentrations and $f(RH)$ were higher than those in Xiamen. Due
325 to the proximity of a , a smaller b value resulted in the $f(RH)$ being lower in this study
326 compared to those in the NCP. It should be noted that the aerosol scattering hygroscopic
327 growth does not necessarily weaken even in atmospheric environments with light
328 particle pollution. This also shows that aerosol scattering hygroscopic growth is mainly
329 controlled by the aerosol properties, such as aerosol chemical composition, etc., which
330 are strongly related to the particle formation mechanism and the source of fine particles
331 (Li et al., 2021; Chen et al., 2022a).

332



333

334 **Figure 3. The $f(RH)$ curves fitted by Eq.(2) on NPF and non-NPF days.** (a)
 335 belongs to NPF days, (b) belongs to non-NPF days. Red and black lines are the curves
 336 fitted by Eq. (2), and blue dots represent the simulated values of $f(80\%)$. The light-color
 337 shaded areas show the 95 % prediction bands for the fits.

338

339 **3.3 Distribution characteristics of $f(RH)$ and aerosol chemical compositions**

340 The mass concentration of the chemical components in NR-PM₁ (nonrefractory
 341 submicron particles), including SO₄, NO₃, NH₄, Chl and Org, and BC in aerosols were
 342 displayed in Table S5. SO₄, NO₃ and NH₄ (SNA) constituted the majority of the
 343 inorganic ions, which were converted from gaseous precursors by complex chemical
 344 processes in the atmosphere. Compared with ten years ago, the concentrations of all
 345 SNA in Xiamen have significantly decreased (Deng et al., 2016). The decline ratio in
 346 sulfate concentration reached 84.0%, indicating the effectiveness of sulfate control
 347 measures in recent years. **In Xiamen, the main measures to control sulfate include the**
 348 **application of desulfurization technology in the flue gas of power plants and coal**
 349 **combustion boilers, as well as the promotion of clean energy for seaships.** However, in
 350 contrast to the studies in 2011-2013 and 2017 (Wu et al., 2015a; Wu et al., 2020), the
 351 ratio of nitrate to sulfate had increased in this study, suggesting that nitrate pollution
 352 has become more prominent in recent years. The concentration of Org decreased in
 353 comparison to the previous study, while its proportion in aerosol remained unchanged
 354 (Chen et al., 2022b). Additionally, BC is regarded as hydrophobic species (Zieger et al.,
 355 2014).

356 The diurnal variation of $f(80\%)$ and chemical mass fractions are displayed in Figure
 357 4. The diurnal variation of $f(RH)$ was significantly related to the mass fraction of
 358 chemical components in particle. NPF days had an obvious lower values of $f(80\%)$
 359 compared with the other event. The mass concentration of PM₁ in the non-NPF days
 360 (12.00 $\mu\text{g m}^{-3}$) was slightly lower than those in the NPF days (12.40 $\mu\text{g m}^{-3}$). The mass
 361 fraction of Org in the aerosol was higher during the NPF period than during the non-
 362 NPF period, and $f(80\%)$ in the NPF period was smaller than that in the non-NPF period.
 363 Based on the wind direction and the homology of Org with BC and CO (Figure S7, S8),

364 the increasing fraction of **Org** might be attributed to the emissions from heavy truck in
365 the major roads near the observation site, which had a weaker impact on the aerosol
366 hygroscopic growth than the SNA.

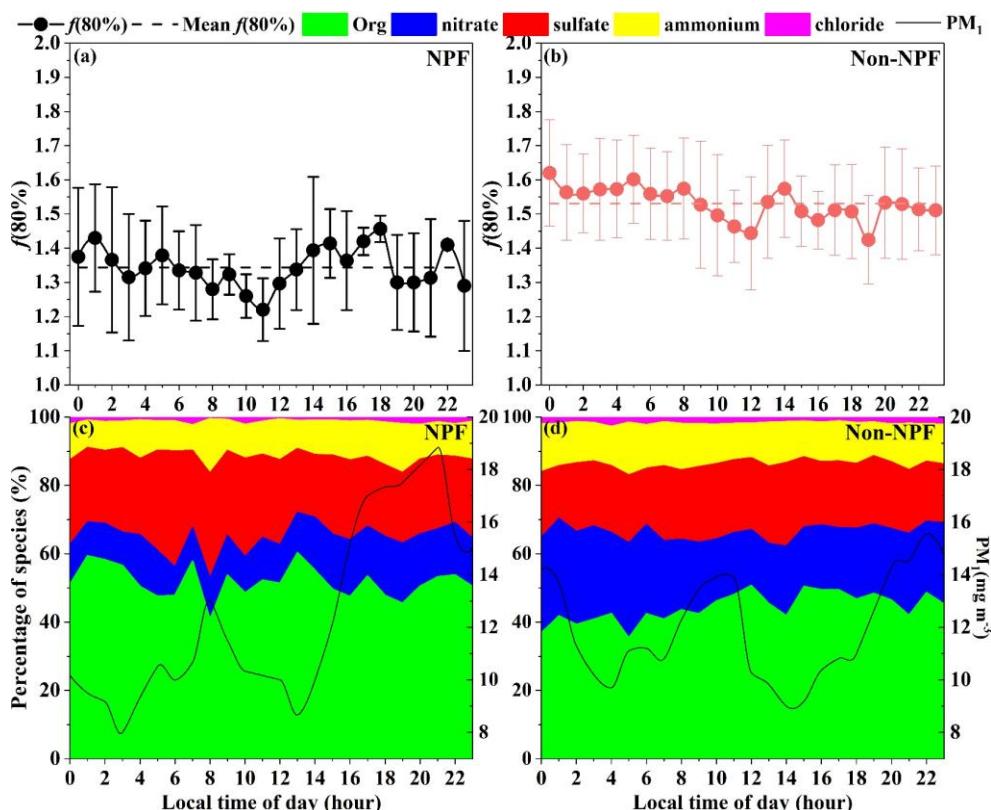
367 The $f(80\%)$ showed the same pattern as the mass fraction of SNA, especially for
368 nitrate, and the amount of nitrate was significantly low during NPF period. The mass
369 fraction of sulfate was much higher than that of nitrate on NPF days compared to non-
370 NPF days. The SO_2 concentrations on NPF days were clearly higher than that on non-
371 NPF days (Figure S9). Moreover, SO_2 concentrations increased from the morning and
372 peaked around 15:00 on NPF days, which was consistent with the increasing trend of
373 particle number concentration on NPF days. The reaction of SO_2 and OH radicals is
374 considered as the primary pathway of sulfuric acid generation (Sipilä et al., 2010).
375 Condensation mode reactions could occur due to the large amount of sulfuric acid in
376 the atmosphere during NPF events. It is hypothesized that condensation mode reactions
377 occurring simultaneously with NPF events were mainly responsible for the higher mass
378 fraction of sulfate during NPF days observed at our site (Yue et al., 2010). Noted that
379 the lower $f(80\%)$ in NPF period was probably attributed to the low mass fraction of
380 nitrate and high mass fraction of sulfate in SNA. Chen et al. (2022a) reported that
381 aerosol hygroscopic growth in Shanghai and Guangzhou during NPF days were lower
382 than those during non-NPF days, which were consistent with the results of this study,
383 but the opposite pattern occurred in the NCP. This illuminates that aerosols had a lower
384 hygroscopicity during the NPF event in Xiamen. The study by Liu et al. (2021) also
385 found that the hygroscopicity of 40 nm organic aerosol (OA) was significantly
386 enhanced during NPF days in urban Beijing, which could be derived from different
387 precursors and accounted for the formation of OA during the NPF process. **On the basis**
388 **of aerosol chemical compositions during NPF, it could be speculated that when particle**
389 **formation occurs in NPF days, the condensation of large quantities of sulfuric acid and**
390 **organic vapours onto the pre-existing particles results in the conversion of mixed state**
391 **on the surface of particles from external mixture to internal mixture. These processes**
392 **alter the optical and chemical properties of particles, which in turn might change the**
393 **aerosol scattering hygroscopicity growth.(Wu et al., 2016)**

394 The sulfur/nitrogen oxidation ratios (SOR/NOR) used to assess the extent of
395 secondary sulphate and nitrate formation are shown in Figure 5, which were calculated
396 from the formulas of $\text{NOR}=[\text{NO}_3]/([\text{NO}_3]+[\text{NO}_2])$ and $\text{SOR}=[\text{SO}_4]/([\text{SO}_4]+[\text{SO}_2])$,
397 respectively (Liu et al., 2020b). The levels of SOR and NOR were subjected to the
398 regulation by photochemical reactions, exhibiting an increase during daytime.
399 Moreover, the variations in SOR/NOR levels and $f(80\%)$ across NPF days were found
400 to be coincident, as SOR/NOR levels dropped down, particularly for NOR, $f(80\%)$ also
401 displayed a low level. We assume that nitrate was essential for aerosol scattering and
402 hygroscopic growth. This assumption was confirmed by comparing Figure 4(a) (c) with
403 Figure 5(a), which show a significant decline in $f(\text{RH})$ when both nitrate content and
404 NOR were low, especially during NPF days. The sharp rise in NOR in the afternoon
405 during the NPF days resulted in a significant increase in the relative amount of nitrate

406 in the aerosol and $f(80\%)$, indicating the rapid response of the aerosol hygroscopic
 407 growth to nitrate, which can be interpreted as a stronger hygroscopicity of nitrate
 408 compared to sulfate. In non-NPF days, SOR and NOR were enhanced compared to NPF
 409 days, which might be resulted from the aqueous phase reaction at relatively high RH
 410 (Sun et al., 2013; Ge et al., 2012). In non-NPF days with high RH and $f(RH)$, water
 411 vapour condensed on the particles and aerosol hygroscopic growth occurred (Martin,
 412 2000). The increase in aerosol liquid water content (ALWC) on the surface of particles
 413 is crucial for heterogeneous reactions in the atmosphere (Mogili et al., 2006). The
 414 heterogeneous hydrolysis of N_2O_5 (Pathak et al., 2009; Brown et al., 2005) and the
 415 aqueous-phase oxidation of SO_2 (Seinfeld et al., 1998; Sun et al., 2013) are important
 416 pathways for nitrate and sulfate formation, respectively. In other words, aqueous-phase
 417 reactions contribute to the production of secondary aerosol (Ge et al., 2012; Xu et al.,
 418 2017a). Thus, the elevation of RH during non-NPF days might promote the
 419 transformation of NO_2 and SO_2 to nitrate and sulfate via aqueous-phase reactions,
 420 manifesting as the enhancement of NOR and SOR.

421 In brief, the influence of SNA on $f(RH)$, particularly nitrate, is significant. Sulfate
 422 dominated the SNA during the NPF days, characterized by weaker aerosol hygroscopic
 423 growth compared to non-NPF days, indicating the remarkably different bulk aerosol
 424 compositions and condensation mode of aerosol formation mechanisms between NPF
 425 and non-NPF days.

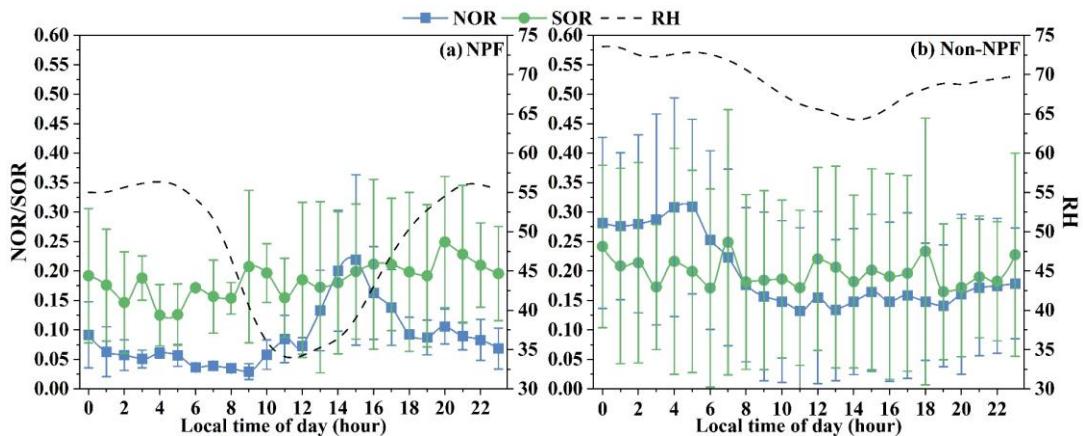
426



427

428 **Figure 4. Diurnal variations of $f(80\%)$ and aerosol chemical mass fractions on**
 429 **NPF and non-NPF days. (a) and (c) belong to NPF days, (b) and (d) non-NPF days.**

430 These chemical compositions include Org, NO_3 , SO_4 , NH_4 , Chl. Solid lines are the mass
 431 concentrations of PM_{1} , dashed lines are the mean value of $f(80\%)$.
 432

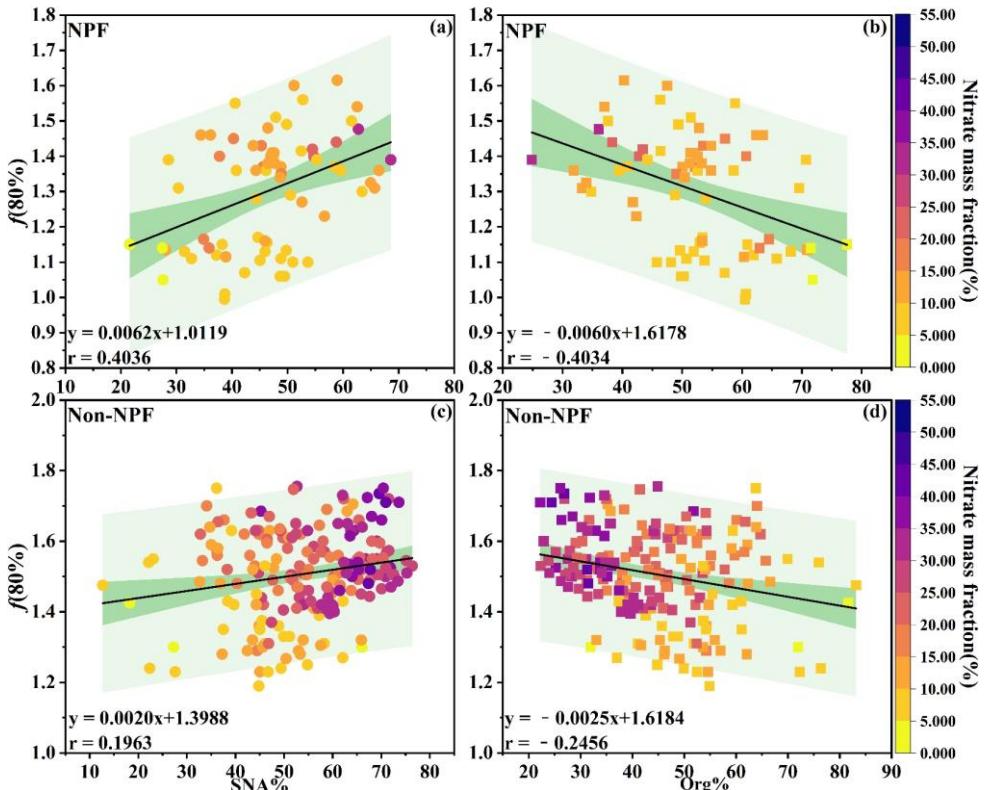


433
 434 **Figure 5. Diurnal variations of NOR, SOR and RH on NPF and non-NPF days.**
 435 (a) belongs to NPF days, (b) belongs to non-NPF days. Blue represents NOR, green
 436 represents SOR, dashed lines are RH.

437
 438 **3.4 Relationships between $f(\text{RH})$ and aerosol chemical compositions**

439 Figure 6 and Figure 7 exhibits $f(80\%)$ as a function of the mass fractions of SNA
 440 and Org between NPF and non-NPF days, revealing the effect of chemical compositions
 441 on the aerosol scattering hygroscopic growth. The SNA fraction showed a positive
 442 correlation with $f(80\%)$ as a result of its high hygroscopicity, whereas the Org fraction
 443 demonstrated a negative correlation with $f(80\%)$ due to its relative lower hygroscopicity
 444 compared to SNA, in line with findings from previous studies (Zhang et al., 2015; Wu
 445 et al., 2017; Ren et al., 2021; Zieger et al., 2014). The magnitude of R for the linear
 446 regression was higher during the NPF period compared to the non-NPF period,
 447 illuminating a stronger correlation between $f(80\%)$ and the mass fraction of SNA or Org
 448 specifically during NPF days. The proportion of aerosol chemical compositions
 449 remained relatively stable during the non-NPF days, which could be accounted for the
 450 limited correlation between $f(80\%)$ and their mass fractions during this period.
 451 Comparing Figure 6(a), 6(b) with Figure 7(a), 7(b), there was a stronger relationship
 452 between mass fraction of sulfate and $f(80\%)$ in NPF days. Conversely, a stronger
 453 association of $f(80\%)$ with nitrate was observed in non-NPF days. This indicated that
 454 sulfate had an important influence on the aerosol hygroscopicity enhancement during
 455 NPF period, while nitrate was the primary contributor for non-NPF days. This
 456 phenomenon could be explained by that the mass fraction of sulfate in the SNA was
 457 highest when the NPF event occurred, yet the differences between the mass fractions
 458 of sulfate and nitrate were slight in non-NPF days. The role of nitrate in the aerosol
 459 hygroscopic properties will be discussed in the following paragraphs.

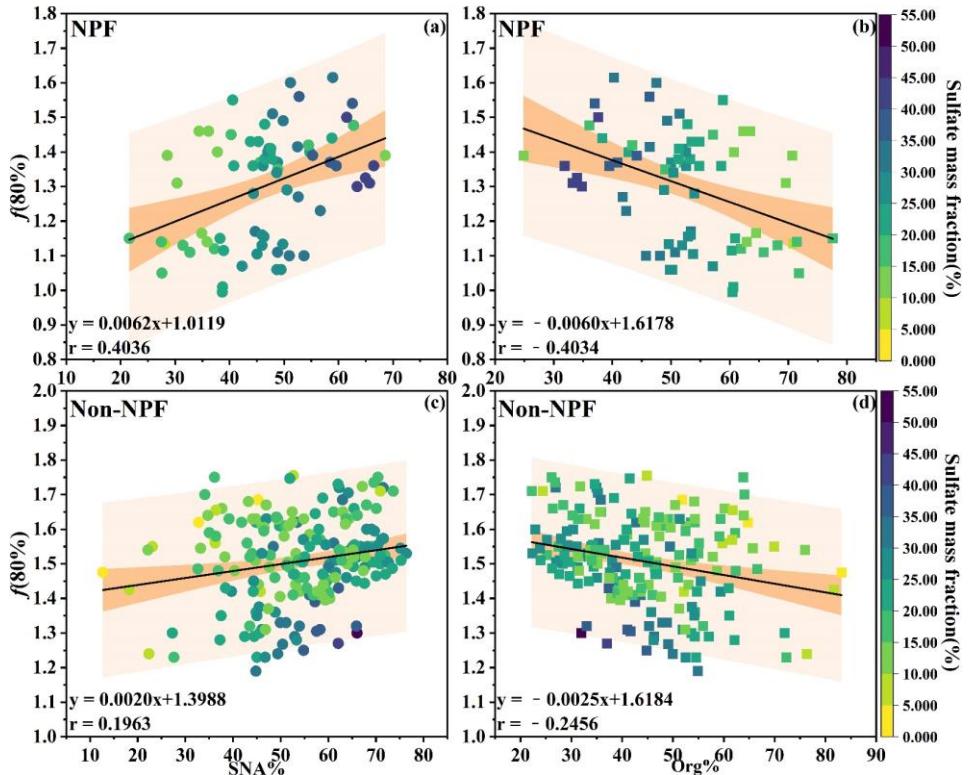
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461

462 **Figure 6. The aerosol scattering hygroscopic growth factor $f(80\%)$ as a**
 463 **function of SNA and Org mass fraction colored by the nitrate mass fraction. (a)**
 464 **and (b) belong to NPF days; (c) and (d) belong to non-NPF days. The linear regression**
 465 **function and Pearson's correlation coefficient (R) are given in each panel. The dark-**
 466 **color shaded areas denote 95 % confidence levels, and the light-color shaded areas**
 467 **show the 95 % prediction bands for the fits.**

468



469

470 **Figure 7. The aerosol scattering hygroscopic growth factor $f(80\%)$ as a**
 471 **function of SNA and Org mass fraction colored by the sulfate mass fraction.** (a)
 472 and (b) belong to NPF days; (c) and (d) belong to non-NPF days. The linear regression
 473 function and Pearson's correlation coefficient (R) are given in each panel. The dark-
 474 color shaded areas denote 95 % confidence levels, and the light-color shaded areas
 475 show the 95 % prediction bands for the fits.

476

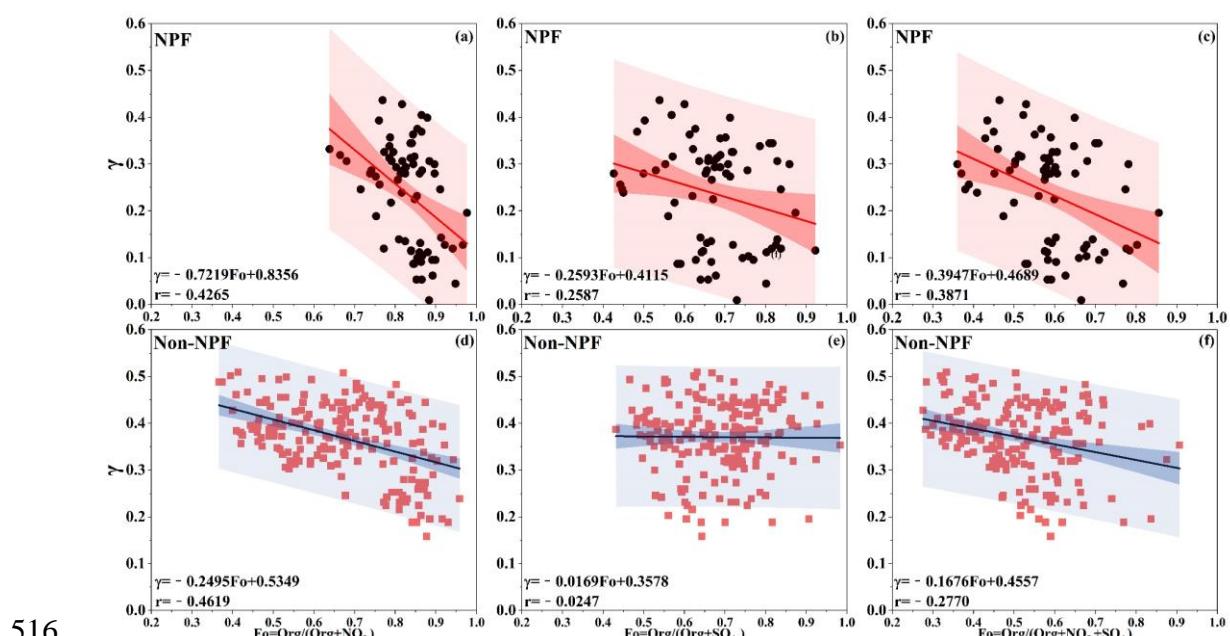
477 As mentioned above, the aerosol hygroscopic growth was significantly influenced
 478 by the proportion of SNA and Org in aerosols. The fitting parameter γ , which depends
 479 on the aerosol hygroscopicity, is defined as $\gamma = \ln f(RH) / \ln((100/RH_{ref}) / (100 - RH))$
 480 (Quinn et al., 2005; Zhang et al., 2015). Here γ was based on $RH_{ref} = 40\%$ and $RH = 80\%$. γ can be employed to characterize the relationship between aerosol hygroscopic
 481 growth and SNA. The relative quantity of Org and inorganic matters can be expressed
 482 as $F_o = \text{Org} / (\text{Org} + C_i)$, where C_i is the mass concentration of SNA. We chose NO_3 , SO_4
 483 and $\text{NO}_3 + \text{SO}_4$ as different SNA constitute in this study, respectively. γ and F_o were
 484 negatively correlated for all scatter plots, and linear regressions of γ versus F_o were
 485 fitted (Figure 8). γ and $F_o = \text{Org} / (\text{Org} + \text{NO}_3)$ (abbreviated as F_{o+N} , Figure 8a, d) were
 486 more strongly correlated than γ and $F_o = \text{Org} / (\text{Org} + \text{SO}_4)$ (abbreviated as F_{o+S} , Figure
 487 8b, e). Additionally, the correlation between γ and $F_o = \text{Org} / (\text{Org} + \text{NO}_3 + \text{SO}_4)$
 488 (abbreviated as F_{o+N+S} , Figure 8c, f) was observed to be smaller than that of the
 489 correlation between γ and F_{o+N} . This is yet more evidence that NO_3 played more
 490 important role than SO_4 in determining the aerosol hygroscopicity in Xiamen, contrary
 491 to the conclusion of Quinn et al. (2005), Malm et al. (2005) and Yan et al. (2009). This

493 findings also underscored the substantial impact of nitrate on aerosol properties,
 494 aligning with recent research conducted in various regions of China (Zhang et al., 2015;
 495 Sun et al., 2020; Liao et al., 2020; Jin et al., 2022).

496 Over the recent decades, the Chinese government has attached great importance to
 497 the air pollution control, and the prominent results have been achieved in reducing SO_2
 498 emissions (Zhang et al., 2019; Zheng et al., 2018). As the concentration of SO_2
 499 decreases, there might be an increasing trend for NH_3 to combine with NO_3^- to form
 500 NH_4NO_3 , thereby enhancing the role of nitrate in atmospheric processes. The relative
 501 low value of $f(80\%)$ for NPF days can be explained by the fact that the organic and
 502 inorganic aerosol fractions were distinct, with sulfate being the predominant component
 503 of the inorganic aerosol during this period. During NPF days, there was a stronger
 504 correlation between γ and $F_{\text{O}+\text{S}}$, while the correlation of γ and $F_{\text{O}+\text{S}}$ was extremely weak
 505 in non-NPF days, which could be related to the role of sulfuric acid in atmospheric
 506 nucleation in NPF days. Li et al. (2021) proposed the parameterization of γ according
 507 to its linear relationship between F and TC, and the R^2 was 0.92, where, F was
 508 determined as the mass ratio of $F = \text{TC}/(\text{TC} + \text{NO}_3^- + \text{SO}_4^{2-} + \text{NH}_4^+)$. However, the
 509 correlation between γ and $F_{\text{O}+\text{N}+\text{S}}$ in this study was lower than that reported by Li et al.
 510 (2021). As the comparison with these parameterizations suggested, the comprehensive
 511 chemical composition information was likely in favor of accurate γ parameterizations
 512 and subsequent $f(\text{RH})$ determinations.

513 Overall, SNA are more effective in promoting aerosol hygroscopic growth than that
 514 of Org , with nitrate having the strongest impact.

515



517 **Figure 8. Scatter plots of γ versus the relative amount of Org and inorganics
 518 (F_{O}).** (a, d) $F_{\text{O}} = \text{Org}/(\text{Org} + \text{NO}_3^-)$, (b, e) $F_{\text{O}} = \text{Org}/(\text{Org} + \text{SO}_4^{2-})$, and (c, f) $F_{\text{O}} = \text{Org}/(\text{Org} +$
 519 $\text{NO}_3^- + \text{SO}_4^{2-})$. (a), (b) and (c) belong to NPF event; (d), (e) and (f) belong to non-NPF
 520 event. Solid lines represent the linear fit. The dark-color shaded areas denote 95 %

521 confidence levels, and the light-color shaded areas show the 95 % prediction bands for
522 the fits.

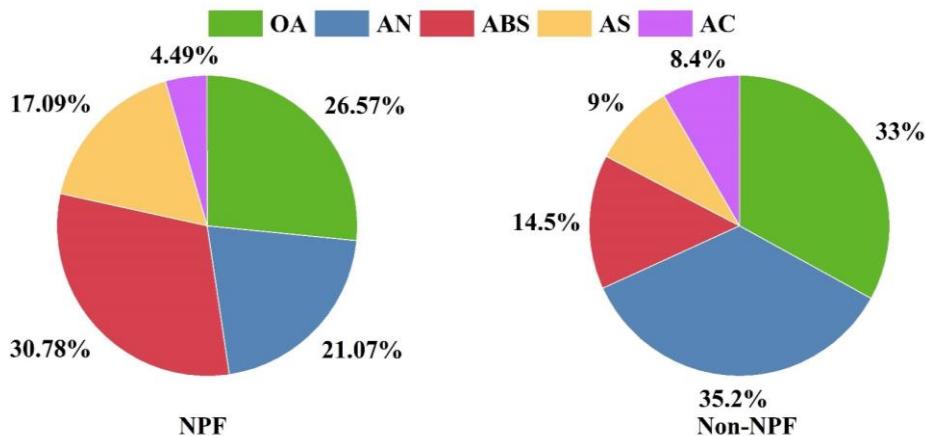
523

524 **3.5 Aerosol hygroscopicity–chemical composition closure**

525 The hygroscopicity parameter κ serves as a highly effective parameter for
526 investigating aerosol hygroscopicity and can be assessed using the humidified
527 nephelometer systems for aerosol light scattering hygroscopic growth factors. The $\kappa_{f(RH)}$,
528 obtained from the measured aerosol scattering hygroscopic growth factor, is a function
529 of the overall hygroscopicity of aerosol particles (Chen et al., 2014; Kuang et al., 2017;
530 Kuang et al., 2020) and widely used to explain the influence of aerosol hygroscopic
531 growth on aerosol optical properties (Tao et al., 2014; Kuang et al., 2015; Brock et al.,
532 2016). Moreover, κ can be estimated based on the mass concentration and chemical
533 composition of particles, commonly known as κ_{chem} . According to Kuang et al. (2020),
534 $\kappa_{f(RH)}$ accurately represents κ_{chem} and can therefore be used for the aerosol
535 hygroscopicity-chemistry composition closure, which was used to investigate the effect
536 of aerosol chemical composition on the overall aerosol hygroscopicity in this study. **The**
537 **average $\kappa_{f(RH)}$ from 60% to 90% was used as a proxy for hygroscopicity closure.**

538 Figure 9 shows the contribution of chemical composition to $\kappa_{f(RH)}$, calculated as the
539 product of κ and its volume fraction for each aerosol chemical composition. During the
540 observation period, the inorganic salts, especially SNA, played a major role in the
541 aerosol hygroscopicity, while some previous studies found that low water-soluble
542 compounds, most likely secondary organic species, predominantly promote new
543 particle formation and have an effect on aerosol hygroscopicity and their cloud
544 condensation nuclei (CCN) activity (Levin et al., 2012; Wu et al., 2015b; Väkevä et al.,
545 2002). For aerosol hygroscopicity during NPF days, NH_4HSO_4 (ABS) was the most
546 dominant contributor followed by NH_4NO_3 (AN). Although $(\text{NH}_4)_2\text{SO}_4$ (AS) ranked
547 third place to aerosol hygroscopicity in NPF, it was still higher than its contributions to
548 non-NPF days, which was consistent with the mentioned above that sulfate was the
549 main factor influencing the aerosol scattering hygroscopicity growth during NPF period.
550 The contribution of AN to aerosol hygroscopicity was predominant during non-NPF
551 days, while the contribution of AS and ABS decreased substantially compared to those
552 during NPF. The facilitation of organic aerosol (OA) to the aerosol hygroscopicity
553 fluctuated considerably due to variations of OA composition under different events. The
554 26.57% of $\kappa_{f(RH)}$ attributed to OA in NPF was lower than that in non-NPF days. This
555 phenomenon was probably associated with the variety of OA components, for instance,
556 secondary organic aerosols (SOAs) exhibiting limited hygroscopicity (Wang et al.,
557 2024). Furthermore, during non-NPF days, despite the increased contribution of OA to
558 $\kappa_{f(RH)}$, $f(RH)$ was higher than that of NPF days, which might be related to the increase
559 in the overall aerosol hygroscopicity during this period and the change in physical
560 properties of particles during these days (Wu et al., 2016).

561



562
Figure 9. Contribution of aerosol chemical composition to $\kappa_{f(RH)}$ on NPF and
563 **non-NPF days.** Green, blue, red, yellow, and purple code represent organic aerosol
564 (OA), NH_4NO_3 (AN), NH_4HSO_4 (ABS), $(\text{NH}_4)_2\text{SO}_4$ (AS), and NH_4Cl (AC),
565 respectively.

566
567 Due to the uncertainty in the hygroscopicity of OA, we investigated the
568 characteristics of the hygroscopicity parameter κ_{OA} for organic aerosol, as well as the
569 relationship between κ_{OA} and the oxidative properties of OA (Figure 10). Non-NPF
570 days had the highest value of κ_{OA} , and NPF days had the lowest. The κ_{OA} values were
571 greater than those of the previous study in China (Wu et al., 2016; Kuang et al., 2020;
572 Kuang et al., 2021), but similar to the findings of Chang et al. (2010) at rural site in
573 Ontario, Canada (Table 1). The proportion of POA and SOA in OA in our study (Figure
574 S11) showed higher SOA mass fractions on non-NPF days than on NPF days. It is
575 evident that high SOA mass fractions on non-NPF days corresponded to high κ_{OA} values.
576 The results from these previous studies (Wu et al., 2016; Kuang et al., 2020; Kuang et
577 al., 2021; Chang et al., 2010) also highlighted that SOA and oxygenated organic
578 aerosols, were likely to be the determinants of κ_{OA} . Furthermore, SOA dominated OA
579 mass both in this study and previous studies; however, κ_{OA} values differed much across
580 studies. Noted that the hygroscopicity of SOA might vary significantly under different
581 emission and atmospheric conditions due to variations in VOC precursors and SOA
582 formation pathways (Kuang et al., 2021). The discrepancy in κ_{OA} suggests that using
583 a constant κ_{OA} value to calculate κ might lead to a large bias.

584
585 **Table 1. Comparisons of the average κ_{OA} in different study.**

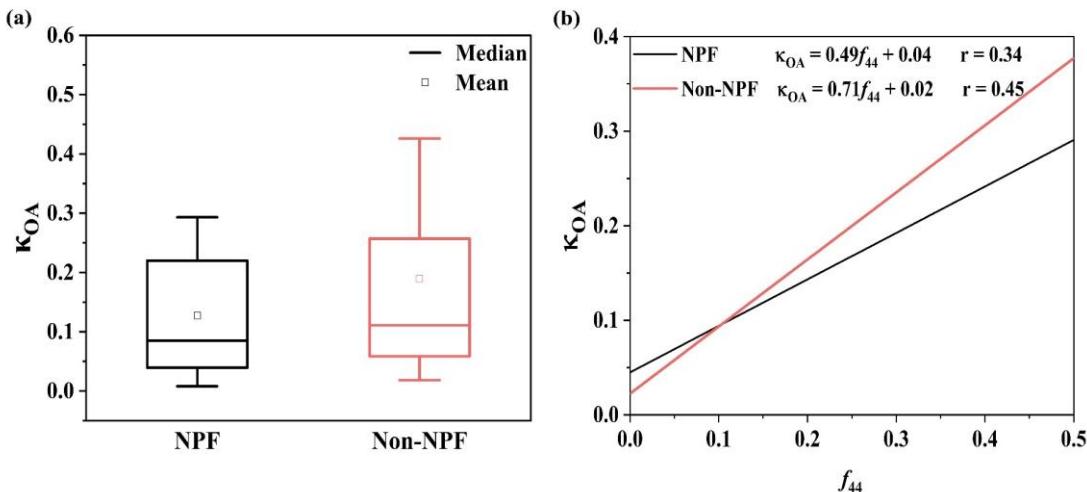
Study area	Periods	Remarks	Mean κ_{OA}	Reference
Beijing, China	Summer, 2014	urban	0.06	Wu et al. (2016)
Dingxing, China	2018/11/11-12/24	rural	0.08 ± 0.06	Kuang et al. (2020)
Heshan, China	2018/9/30-11/17	rural	0.085 ± 0.05	Kuang et al. (2021)

Ontario, Canada	Spring, 2007	rural	0.22 ± 0.04	Chang et al. (2010)
			0.13 ± 0.11	
Xiamen, China	2022/2-4	urban	(NPF) 0.19 ± 0.21 (non-NPF)	This study

587

588 To further investigate the factors affecting κ_{OA} , we compared the effect of OA
 589 oxidation level on κ_{OA} , where, f_{44} was used to represent the level of OA oxidation
 590 (Kuang et al., 2020; Chen et al., 2017). The values of f_{44} in the component mass
 591 spectrometry were ratio of m/z 44 to total signals, reflecting the absolute oxidation
 592 degree of aerosols (Chen et al., 2022b). The results showed a slightly weak correlation
 593 between κ_{OA} and the oxidation level of OA, indicating that the degree of oxidation was
 594 one of significant parameters in determining the hygroscopicity of OA. For both NPF
 595 days and non-NPF days, the hygroscopicity of OA enhanced with its oxidation level.
 596 Most of the previous studies on κ_{OA} had shown that the hygroscopicity of OA usually
 597 increased with the uplift of oxidation degree of OA, which was also found in this study.
 598 Nevertheless, a more pronounced increase in κ_{OA} with elevated f_{44} was observed in non-
 599 NPF days, attributing to the OA components, formation mechanisms, and the species
 600 of VOCs among different events, which can also be accounted by the fact that less
 601 hygroscopic OA is produced during the NPF process in Xiamen, contrasted with the
 602 finding of Liu et al. (2021). Thereby, it could be found that alterations in the component
 603 and oxidation of OA might regulate the variation in κ_{OA} (Timonen et al., 2013; Xu et al.,
 604 2014; Xu et al., 2017b). In addition, the κ_{OA} values in NPF and non-NPF days were
 605 lower than those in the studies of Chen et al. (2017), Liu et al. (2021) and Kuang et al.
 606 (2020). These results reflect that the use of parameters related to oxidative properties,
 607 such as f_{44} or the ratio of O and C alone is not sufficient to characterize the
 608 hygroscopicity of OA, and that the molecular information of OA and the special
 609 formation mechanisms should also be considered (Liu et al., 2021). However, OA
 610 components is still one of the main driving force of the variation in aerosol
 611 hygroscopicity.

612



613
614 **Figure 10. The derived κ_{OA} and f_{44} on NPF and non-NPF days.** (a) Variation of
615 κ_{OA} between NPF and non-NPF days. (b) The relationship between derived κ_{OA} and f_{44} .
616

617 **4 Conclusions**

618 In this study, the aerosol scattering hygroscopic growth during the new particle
619 formation in Xiamen, the coastal city of southeast China, was clearly depicted by in
620 situ observations. The distinct aerosol hygroscopic behaviors were evident during NPF
621 events in the urban environment characterized by the elevated ambient temperatures,
622 high levels of relative humidity, and low pollution. The aerosol scattering hygroscopic
623 growth of NPF days was weaker than those of non-NPF days. However, under high RH
624 conditions (80% to 90%), the $f(RH)$ growth of NPF days exceeded that of non-NPF
625 days. Furthermore, it was determined that one of the two-parameter fitting equations
626 was more adept at accurately representing the observed $f(RH)$.

627 The notable variations in $f(RH)$ between NPF and non-NPF days were impacted by
628 changes in SNA and **Org** of aerosols. SNA had a more important effect on $f(RH)$ and a
629 higher efficacy in enhancing aerosol hygroscopic growth than that of **Org**, with nitrate
630 exhibiting the most pronounced impact. Sulfate was highlighted as the dominance in
631 SNA during NPF days, with weaker $f(RH)$ compared to non-NPF days. It is likely that
632 the condensation mode reactions occurring simultaneously with NPF events changed
633 the aerosol chemical composition and had an obvious effect on $f(RH)$. The contribution
634 of SNA to aerosol hygroscopicity surpassed that of **Org**, with ABS and AN dominating
635 in NPF days and non-NPF days respectively, revealed by the aerosol hygroscopicity–
636 chemical composition closure. The estimated κ_{OA} exhibited a decrease in NPF days
637 compared to non-NPF days, and it showed an increase corresponding to the level of OA
638 oxidation in both two types of days. The uncertainty in OA hygroscopicity resulted from
639 variations in its components and oxidation states, however it was believed to be an
640 important driver of the alteration in aerosol hygroscopicity. The findings in this study
641 may provide a better explanation of aerosol hygroscopicity properties in the coastal city,
642 which could also offer valuable insights into the use of hygroscopic growth factors in
643 the models of air quality and climate change.

644

645 **Data availability**

646 The dataset for this paper can be accessed at
647 <https://doi.org/10.5281/zenodo.13756825> (Li et al., 2024).

648

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657 Research Station of Fujian Province.

658

659 **Author contributions**

660 ML and JC conceived the conceptual development of the paper. LL made
661 measurements, analyzed the data and wrote the paper. ML designed the project and
662 conducted the field campaigns. ML, XF and JC directed the manuscript development
663 and editing. YC made measurements. All authors contributed to discussion and review.

664

665 **Competing interests**

666 The authors declare that they have no known competing financial interests or
667 personal relationships that could have appeared to influence the work reported in this
668 paper.

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