



# Impact of introducing electric vehicles on ground-level O<sub>3</sub> and PM<sub>2.5</sub> in the Greater Tokyo Area: Yearly trends and the importance of changes in the Urban Heat Island effect

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**Abstract.** Battery electric vehicles (BEVs) are considered a solution for global warming and air pollution, and several countries have announced a shift to BEVs in the 2030s. This study is an evaluation of changes in the urban heat island (UHI) effect in the Greater Tokyo Area (GTA) of Japan as a result of introducing BEVs using numerical weather prediction. The results indicated that the substitution of internal combustion vehicles with BEVs led to a maximum decrease of 0.2 °C in the local temperature for the metropolitan GTA. Estimation of the effects of introducing BEVs on tropospheric ozone (O<sub>3</sub>) and fine particulate matter (PM<sub>2.5</sub>) using a regional chemical transport model indicated that mitigating against the UHI effect led to a decline in the ground-level O<sub>3</sub> formation due to decreased atmospheric chemical reactions and lower biogenic volatile organic compounds (BVOC). The decrease in vehicular exhaust emissions and increase in power plant emissions were nonlinear, and the total change in O<sub>3</sub> depends on season and location. The temperature decrease that resulted from the mitigation of UHI resulted in enhanced particle coagulation, with an increase in ground-level PM<sub>2.5</sub> formation in several regions. Furthermore, a decrease in the BVOC emissions also resulted in increased PM<sub>2.5</sub> owing to enhancement of the OH + SO<sub>2</sub> reaction. The total prevented annual premature deaths of 175 and 77 that resulted from the changes in O<sub>3</sub> and PM<sub>2.5</sub>, respectively, indicate the positive effects of BEV introduction on air quality management in the GTA, and may be applicable to other megacities worldwide.

## 1 Introduction

Climate change has significantly impacted global temperatures, inducing natural hazards and health problems (Heidari and Pearce, 2016; Dubash et al., 2018; Iacobuta et al., 2018; Fox et al., 2019). In response, worldwide organizations are attempting to restrict greenhouse gas (GHG) emissions, with the target of minimizing the temperature increase from the era of the industrial revolution to only 1.5 °C or 2.0 °C by 2100 (Fawzy et al., 2020; Schreyer et al., 2020; Strapason et al., 2020), and



governments, especially in developed countries, have set future scenarios with the goal of achieving net zero-carbon societies. The transport sector is known to be a large GHG emission source; thus, the introduction of battery electric vehicles (BEVs) is considered one solution by which emissions can be minimized (Moro and Lonza, 2018; Shen et al., 2019; Andersson and Börjesson, 2021). The introduction of BEVs is also expected to decrease primary air pollutant emissions from engine exhaust and the evaporation of gasoline (Soet et al., 2014; Ferero et al., 2016; Requia et al., 2018). The nitrogen oxides (NO<sub>x</sub>) and volatile organic compounds (VOCs) that are emitted by vehicles are precursors of tropospheric ozone (O<sub>3</sub>) and fine particulate matter (PM<sub>2.5</sub>), which are harmful to animals, including humans (Finlayson-Pitts and Pitts Jr., 1993; Sillman, 1999; Volkamer et al., 2006; Holmes, 2007). The introduction of BEVs is thus expected to decrease these precursors. Aside from the direct exhaust emission issue, Muratori suggested that there would be an increased power demand due to BEV battery charging (Muratori, 2018), which suggests an increase in emissions of primary air pollutants from power plants attributable to the introduction of BEVs. Furthermore, the anthropogenic heat (AH) from vehicles is expected to decrease by the introduction of BEVs because of the mitigation of engine exhaust emissions. According to the review article of Ulpiani, the mechanisms of the impact of the UHI effect on air pollutants are associated with: (1) the change of the kinetics of O<sub>3</sub> (and PM<sub>2.5</sub>) formation, (2) the change of air mixing ratio arising from the change of ambient temperature, and (3) the change of biogenic VOC (BVOC) emissions (Ulpiani, 2021), which are all correlated with each other. Therefore, the detailed information concerning how the introduction of BEVs is likely to contribute to atmospheric pollution is particularly important in determining governmental strategies for the future.

The Greater Tokyo Area (GTA) of Japan is one of the largest regions in the world in terms of economy, population and human activities, industry, etc., and the air pollution caused by road traffic in the area remains a major concern. Previously, Hata and Tonokura conducted chemical transport modeling in the GTA to evaluate how ground-level O<sub>3</sub> concentrations change in response to the introduction of zero-emission vehicles (Hata and Tonokura, 2019). The study revealed ground-level O<sub>3</sub> increases in urban areas that were due to VOC-limited atmospheric conditions and the NO-titration effect (Santiago et al., 2022), and concluded that the health risk associated with increased O<sub>3</sub> surpasses that of influenza and heatstroke, which was not considered in the study conducted by Hata and Tonokura. An evaluation of the impact of nighttime BEV battery charging on ground-level O<sub>3</sub> by Kayaba and Kajino during the summer season in the GTA implied a non-negligible impact on the O<sub>3</sub> concentration (Kayaba and Kajino, 2023). However, despite the impactful findings surrounding the introduction of BEVs on the air quality in the GTA, the study targeted only O<sub>3</sub> as a secondary air pollutant and the evaluation of PM<sub>2.5</sub> was neglected, although the potential health risk of PM<sub>2.5</sub> is expected to be equal to or higher than O<sub>3</sub> (Poppe III et al., 2002; WHO Regional Office for Europe, Copenhagen, 2008). Some studies have also reported positive effects from introducing BEVs on local O<sub>3</sub> and PM<sub>2.5</sub>, based on the results of chemical transport modelling (Li et al., 2016; Pan et al., 2019; Lin et al., 2020).

Despite these findings, the effects of the change of AH, which finally causes change of the urban heat island (UHI) effect, attributable to the introduction of BEVs to ground-level O<sub>3</sub> and PM<sub>2.5</sub> were never considered. Xie et al. conducted a model calculation to evaluate the effect of AH on ground-level O<sub>3</sub> in southern China, with results showing a high impact on O<sub>3</sub> formation, corresponding to a 2.5 ppb increase (Xie et al., 2016), and other studies have also indicated a positive correlation



65 between  $O_3$  and the UHI effect (Stathopoulou et al., 2008; Wang et al., 2018; Ulpiani, 2021). The introduction of BEVs is expected to decrease the local UHI effect by reducing AH (Li et al., 2015). Thus, a more accurate evaluation of the impact of introducing BEVs on the local air quality that considers changes in the UHI is required to assist in policymaking.

The purpose of this study was therefore to clarify the yearly impact of the introduction of BEVs on ground-level  $O_3$  and  $PM_{2.5}$  while also examining the effect of the UHI mitigation on  $O_3$  and  $PM_{2.5}$  production in the GTA. The impact of changes to the UHI that result from the introduction of BEVs on the formation of the two atmospheric pollutants was evaluated using a chemical transport model coupled with numerical weather prediction. A detailed analysis of the changes in ground-level  $O_3$  and  $PM_{2.5}$  was conducted within the scope of atmospheric chemistry. Finally, the change in premature mortality in the GTA due to the introduction of BEVs was estimated as a representative health risk. To the best of our knowledge, this is the first study to consider the UHI effect that is caused by the shift to BEVs for road transportation on  $O_3$  and  $PM_{2.5}$  using chemical transport model. Although the target region of this study is the GTA in Japan, the results are expected to be referenced by other megacities, thus contributing to policymaking worldwide.

## 2 Methodology

### 2.1 Scenarios for the replacement of internal combustion vehicles with battery electric vehicles

The target of this study is the replacement of internal combustion vehicles (ICVs) with BEVs. Nine categories of vehicles were applied in the emissions inventory of this study: light and normal passenger cars (PV), buses, light and small tracks (TV), heavy-duty vehicles (HV), motorcycles, and vehicles used for specific purposes (Shibata and Morikawa, 2021). The worldwide strategy to mitigate GHG reduction focuses on the complete substitution of ICVs with BEVs over the period 2030 to 2035 (Mulrow and Grubert, 2023), with net zero emissions from all vehicles by 2050. Owing to technological issues related to the capacity of batteries used in BEVs, it remains unclear whether HVs can be completely electrified (Forrest et al., 2020), and several plans have suggested the introduction of renewable fuels instead of electrification for these vehicles (Hosseinzadeh-Bandbafha et al., 2021). For these reasons, BEV replacement concentrated only on PVs and TVs in this study. The electrification strategy for motorcycles and specific purpose vehicles is not clear, and the impact of these vehicles on air pollution is expected to be less than that of other vehicles in Japan; thus, the electrification for these vehicles also was not included in the study.

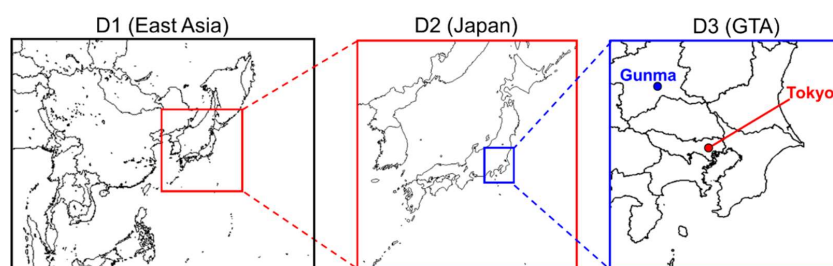
### 2.2 Numerical Weather Predictions

#### 2.2.1 General information

A Weather Research and Forecasting Model (WRF v4.3.3) was used to obtain meteorological conditions (Skamarock et al., 2024) from Dec. 1, 2016 to Dec. 31, 2017. The first month was treated as the spin-off period and the whole of 2017 as the analysis period. Fig. 1 shows the included domains: East Asia (D1:  $45 \times 45$  km), Japan (D2:  $15 \times 15$  km), and the GTA (D3:



95  $5 \times 5$  km). Objective analysis data describing the boundary conditions were obtained at a resolution of  $1^\circ \times 1^\circ$  from the Final Atmospheric Research provided by the National Center for Atmospheric Research (NCAR: [National Center for Atmospheric Research archives, 2024](#)). The boundary conditions of sea surface temperature for  $0.082^\circ \times 0.082^\circ$  were obtained from the National Oceanic and Atmospheric Administration (NOAA) ([Environmental Modeling Center Archives, 2024](#)). The modelled results were validated by comparison with observation data from the Japan Meteorological Agency ([Japan Meteorological Agency archives, 2024](#)). Two sites, the metropolitan areas of Tokyo and Gunma, for which further details are listed in Table S1 of the supporting information (SI), were selected to represent urban and suburban/rural areas, respectively. A simple single-layer Urban Canopy Model (UCM: [Kusaka et al., 2021](#)) was combined with the WRF calculations so that the UHI effect before and after the introduction of BEVs could be considered. UCM parameters obtained from a previous study ([Hara et al., 2010](#)) were used to determine the urban canopy parameters for the GTA. Further details of the simulation setup for the WRF are listed in Table S2 of the SI.



**Figure 1: Domains used for numerical weather prediction and chemical transport model. Prefectural areas of Tokyo and Gunma are in domain D3.**

### 2.2.2 Consideration of UHI effects after the introduction of BEVs

110 Several studies have suggested that UHIs are one of the results of AH from transportation ([Rosenfeld et al., 1998](#); [Nuruzzaman, 2015](#); [Rizvi et al., 2023](#)), and that the introduction of BEVs is expected to eradicate this issue. The changes in the total energy efficiency of BEVs as compared to ICVs affects the AH as well as vehicular exhaust emissions, leading to local changes in the UHI. The process used to evaluate AH following the introduction of BEVs is described in Appendix A. Briefly, two categories of passenger cars and one small truck listed were treated as model vehicles in this study (Tables A1 and A2), and using the fuel consumption for ICVs and electric consumption for BEVs obtained using catalogue data from the manufacturers, the differences in the energy efficiencies of ICV and BEV were calculated. Statistical data concerning the number of vehicles ([Database from Automobile Inspection & Registration Information Association, 2024](#)), average mileage per year ([Ministry of Land, Infrastructure, Transport, and Tourism, 2024](#)), and the heat enthalpy of gasoline and light fuel ([Agency for Natural Resources and Energy, 2024](#)) were then applied and the total AH produced by ICVs and BEVs were calculated. Data describing stationery and transportation sources in Tokyo in 2010 ([Ministry of the Environment, 2024](#)) were used to calculate the AH



from these sources. The results showed a reduction ratio of 0.813, or 18.7 %, for AH in the GTA in association with the introduction of BEVs.

## 2.3 Chemical transport modelling

### 2.3.1 Modelling description

125 A community multiscale air quality modeling system (CMAQ v5.3.3: [Murphy et al., 2021](#)) was used to calculate the air quality in Japan. Domain D3 was targeted for analysis and the year 2017 was examined, as described in Fig. 1. The period Jan. 1 to Mar. 31 was treated as winter, Apr. 1 to June 30 as spring, Jul. 1 to Sep. 30 as summer, and Oct. 1 to Dec. 31 as autumn. The meteorological inputs obtained from the results of the WRF calculations described in Sect. 2.2 were applied to the CMAQ calculation. SAPRC-07 was used to describe the gas-phase chemical mechanism ([Carter, 2010](#)), and the AERO6 module with  
130 ACM liquid chemistry as the aerosol chemical mechanism ([Binkowski and Roselle, 2003](#)). The emission inventories were provided by Chatani et al. ([Chatani et al., 2018](#)). Chemical transport modelling was validated by comparing the two observation sites listed in Table S1. The primary emissions, evaporative VOC emissions, and refueling processes associated with the use of ICVs were expected to decrease under the introduction of BEVs; however, charging batteries increases the requirement for power from plants, increasing the emissions from this source.

### 135 2.3.2 Emission changes resulting from the substitution of ICVs with BEVs

Three categories of vehicular emissions affect the environment: tailpipe engine exhaust emissions (both hot- and cold-start emissions), evaporative emissions from gasoline fleets (hot soak, diurnal breathing, and running losses), and particulate dust from tires (break, tire, and hoisting dust) ([Shibata and Morikawa, 2021](#)). Stationary sources such as VOC emissions during refueling ([Yamada et al., 2018](#)) and power plants that supply energy for battery charging ([Casals et al., 2016](#); [Kayaba and  
140 Kajino, 2023a](#)) also require consideration. In the BEV introduction scenario, all exhaust, evaporative, and refueling emissions from the targeted vehicles were set to zero. The change in the emissions of particulate dust was not considered in this study, although a previous study suggested that brake dust increases after the introduction of BEVs owing to the weight of the battery ([Kayaba and Kajino, 2023b](#)). The increase in primary emissions from power plants that is due to charging BEVs was estimated using the value of the charging voltage of specific electric vehicles, as described in Sect. A.3 of Appendix A. The results  
145 suggested a 33.8 % increase in the amount of air pollutants from powerplants for the GTA.

### 2.3.3 Scenarios for the chemical transport model

Details of the two major scenarios investigated in this study: BASE, a basic scenario without the introduction of BEVs, and ALL, an BEV-introduction scenario, are listed in Table 1. The ALL scenario considers the effects of introducing BEVs and includes the emission reductions from engine exhaust and evaporative VOCs, emission increases from power plants as a result  
150 of battery charging, and UHI changes. Four sensitivity scenarios: the introduction of BEVs ( $S_{EV}$ ), the increase in emissions



from power plants ( $S_{PP}$ ), the change in ambient temperature ( $S_{UHI}$ ), and the change in BVOC emissions resulting from changes to the UHI ( $S_{BVOC}$ ) were also generated to evaluate the contributions of emission reduction on the formation of ground-level  $O_3$  and  $PM_{2.5}$  in GTA. Note that  $S_{UHI}$  only considers the effects of temperature change, and that the change in BVOC emissions due to such temperature changes was not included. The BASE scenario was calculated with D1, D2, and D3, while all other scenarios were calculated only for the D3 domain using the initial and boundary conditions obtained from the BASE scenario. Changes in the annual mean primary  $NO_x$ , anthropogenic VOCs (AVOC), particulate matter (PM), ammonia ( $NH_3$ ), sulfur dioxide ( $SO_2$ ), and BVOC emissions can be seen in Figs. S1 to S4 of the SI.

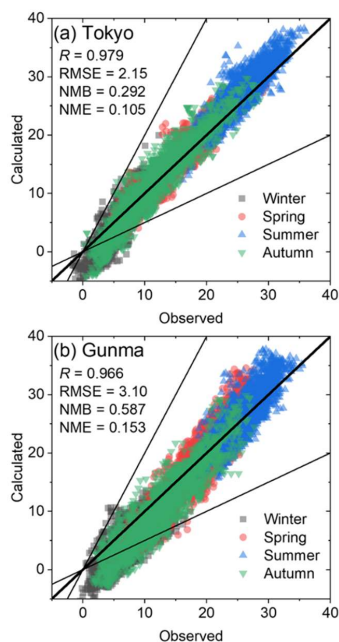
**Table 2: Six scenarios calculated using the chemical transport model.**

Scenario	Definition
BASE	Base scenario for 2017.
ALL	Includes total effects of substituting ICVs with BEVs.
$S_{EV}$	Sensitivity scenario clarifying the effect of reducing engine exhaust and evaporative emissions.
$S_{PP}$	Sensitivity scenario clarifying the effect of increasing power plant emissions.
$S_{UHI}$	Sensitivity scenario clarifying the effect of changes in the temperature as a result of AH reduction.
$S_{BVOC}$	Sensitivity scenario clarifying the effect of changes in the BVOCs emissions caused by the change in temperature.

### 160 3 Results

#### 3.1 Validation of modelled temperature, $O_3$ , and $PM_{2.5}$ with the observed results

Figure 2 shows the correlation between the observed and calculated results for the hourly ground temperature for winter, spring, summer, and autumn in Tokyo in 2017. The correlations are exhibited using four statistical factors: correlation factor ( $R$ ), root mean square error (RMSE), normalized mean bias (NMB), and normalized mean error (NME). Overall, the model was found to replicate the observed results well in all seasons, with  $R = 0.979$  for Tokyo and  $R = 0.966$  for Gunma. Underestimations that are associated with low temperatures and overestimations that correlate with hot temperatures are within the range validated in previous studies examining the same region (Hata et al., 2020).

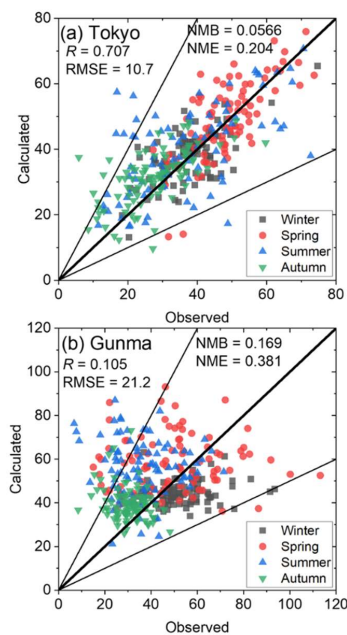


**Figure 2: Comparison of observed and calculated hourly-ground temperatures for four seasons in (a) Tokyo and (b) Gunma.**

170 Figure 3 shows the correlation between the observed and calculated (BASE) results and the 8-h daily maximum average (MDA8) O<sub>3</sub> concentrations. As seen in Fig. 3(a), the modeled O<sub>3</sub> replicated the observed results well, with most plots within a factor of two. However, Fig. 3(b) suggests less correlation for the modeled O<sub>3</sub> with the observed results for Gunma. Emery et al. proposed indicators that can be used to validate O<sub>3</sub> and PM<sub>2.5</sub> for chemical transport modeling (Emery et al., 2017). According to these indicators, the modeling performance for MDA8-O<sub>3</sub> should be  $R > 0.75$ ,  $NMB < \pm 0.05$ , and  $NME < 0.15$  while the criteria value should be  $R > 0.50$ ,  $NMB < \pm 0.15$ , and  $NME < 0.25$ . Tokyo fully meets these criteria and is close to

175 while the criteria value should be  $R > 0.50$ ,  $NMB < \pm 0.15$ , and  $NME < 0.25$ . Tokyo fully meets these criteria and is close to the described goals. However, Gunma did not meet the criteria for all indicators. This may be because Gunma is located in the countryside, where less primary air pollutant emissions are generated than in highly polluted areas such as Tokyo, and sources of O<sub>3</sub> formation in Gunma are BVOCs, transportation from other regions, and some local anthropogenic emissions, rendering it difficult to predict O<sub>3</sub> by CTM. Nevertheless, 70 % of the calculated results were within a factor of 2 and are thus considered

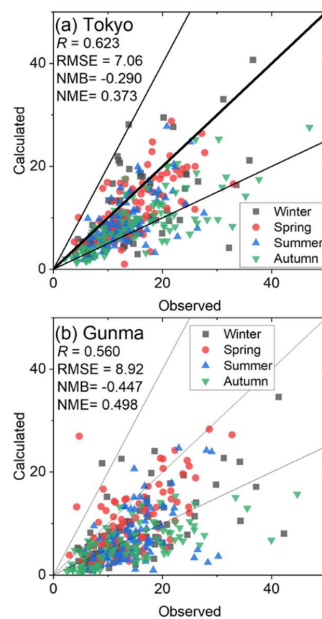
180 acceptable for analysis.



**Figure 3: Comparison of observed and calculated MDA8 O<sub>3</sub> concentrations for four seasons in (a) Tokyo and (b) Gunma.**

Figure 4 shows the correlation between the observed and calculated (BASE) results and the 24-h daily average (DA24) PM<sub>2.5</sub> concentrations. Emery et al. proposed that indicators to model DA24 PM<sub>2.5</sub> should be  $R > 0.70$ ,  $NMB < \pm 0.10$ , and  $NME < 0.35$  in terms of performance, while criteria should be  $R > 0.40$ ,  $NMB < \pm 0.30$ , and  $NME < 0.50$  (Emery et al., 2017). According to Fig. 4(a), the modeled results meet all the criteria, and some  $R$  and  $NME$  values are close to the goal. However, Fig. 4(b) shows that  $NMB$  overshoots the criteria, while  $R$  and  $NME$  meet the criteria. As mentioned above, Gunma is suburban to rural, which may explain the lower accuracy of the modeled results as compared to those obtained in urban areas. Analysis therefore considers the fact that relatively less accuracy was obtained for modeled O<sub>3</sub> and PM<sub>2.5</sub> in the Gunma region.





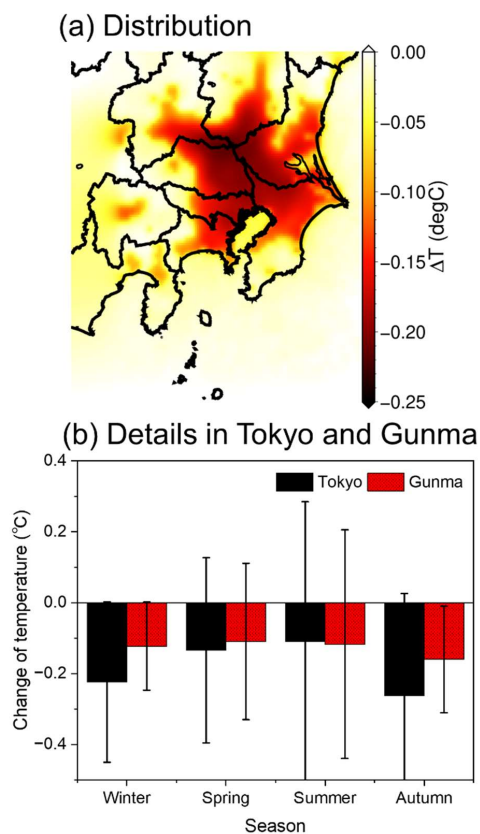
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**Figure 4: Comparison of observed and calculated 24-h averaged  $PM_{2.5}$  concentrations for four seasons in (a) Tokyo and (b) Gunma.**

### 3.2 UHI effect in the GTA

Figure 5 shows the average difference in the ground surface temperature ( $\Delta T$ ) between the BASE and ALL cases over four seasons as calculated using the WRF. The seasonal  $\Delta T$  distributions are described in Fig. S5 of the SI. According to Fig. 5(a), a decrease in the average  $\Delta T$  in the calculated domain is due to the decrease in AH that resulted from the introduction of BEV. According to Fig. S5, this decrease in  $\Delta T$  is high in winter and autumn, and low in summer, and is distributed from the center of Tokyo, where high AH is expected as a result of condensed anthropogenic activity. The seasonal changes in  $\Delta T$  for Tokyo and Gunma are extracted in Fig. 5(b), which shows a range of 0.1 °C to 0.2 °C for  $\Delta T$ , with relatively high error bars, particularly for spring and summer. This is likely because of the high planetary boundary layer (PBL) that forms in the warmer seasons, which enhances thermal convection and leads to the large discrepancies observed in the  $\Delta T$ . The change of the height of the PBL between ground surface and free troposphere due to the change of the UHI effect is described in Fig. S6 of the SI. According to Fig. S6 (a)–(c), the PBL height decreased by ~3% in the winter, spring, and summer seasons, but this change of PBL height reached ~30% in autumn. The change in the UHI effect is expected to affect air pollution via atmospheric conditions (changes in the rate constants of atmospheric chemical reactions and changes in the mixing ratio of air due to the change of PBL height) and emissions (BVOC emissions from biogenic sources). These effects are discussed in the following sections.

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**Figure 5: Changes in the ground temperature following the introduction of BEVs in the GTA: (a) yearly averaged distribution, and (b) seasonal variation in Tokyo and Gunma.**

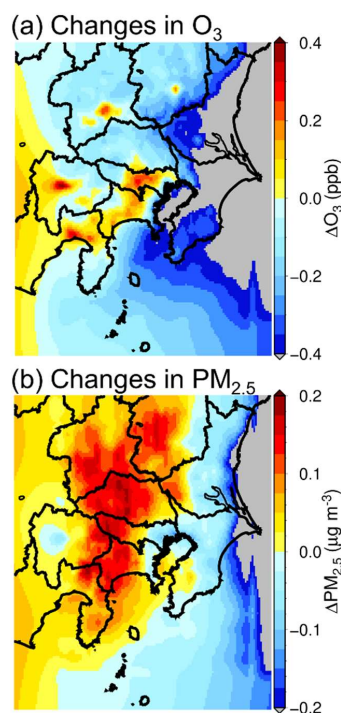
## 210 3.3 Effect of the introduction of BEVs on ground-level O<sub>3</sub> and PM<sub>2.5</sub> concentrations

### 3.3.1 Overall effects

Figure 6 shows the overall effects of introducing BEVs on the change in the yearly averaged O<sub>3</sub> and PM<sub>2.5</sub> concentrations in the GTA, which were evaluated by taking the difference in the results from the ALL and BASE scenarios. As shown in Fig. 6(a), O<sub>3</sub> increased around the central area of the GTA, as well as in several other sites for which consistent results were reported in a previous study (Hata et al., 2019; Kayaba and Kajino, 2023a). The seasonal changes in O<sub>3</sub> can be seen in Fig. S7 of the SI. According to the results, a strong increase in O<sub>3</sub> was estimated in the winter and autumn seasons, when a considerable decrease in temperature is estimated, affecting the UHI effect. However, a slight increase in O<sub>3</sub> in the spring and summer seasons



indicates that the UHI effect contributes significantly to the local formation of  $O_3$  in the winter and autumn seasons. According to Fig. 6(b), increases in  $PM_{2.5}$  also occurred on the western and northern sides of the GTA. The primary  $PM_{2.5}$  emission was expected to decrease as a result of the introduction of BEVs; thus, this increase is attributed to the enhancement of secondary products. Seasonal changes in  $PM_{2.5}$  are shown in Fig. S8 of the SI.



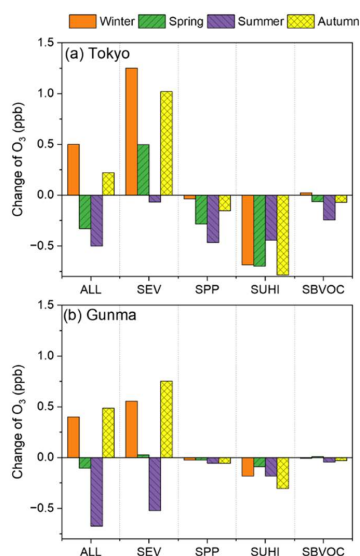
225 **Figure 6: Change in the yearly-averaged (a)  $O_3$  and (b)  $PM_{2.5}$  concentrations following the introduction of BEVs in the GTA. Results include the effects of all BEV-introduction scenarios, including the reduction of primary air pollutants from the transport sector, increase in the primary air pollutants due to the high energy demands of battery charging, UHI mitigation, and changes in the BVOC emissions as a result of UHI mitigation.**

### 3.3.2 Sensitivity analysis of the change in $O_3$

The contributions of the five scenarios that were subtracted from the BASE scenario: ALL,  $S_{EV}$ ,  $S_{PP}$ ,  $S_{UHI}$ , and  $S_{BVOC}$ , to changes in the  $O_3$  concentrations in Tokyo and Gunma can be seen in Fig. 7. Detailed definitions of the scenarios are described in Sect. 2.3.3 and Table 1. According to Fig. 7, the seasonal trends obtained using the five scenarios were similar for Tokyo and Gunma, whereas the absolute change in  $O_3$  was higher in Tokyo because of the condensed anthropogenic activities in the city. As mentioned in Sect. 3.3.1, the results of ALL show an increase in  $O_3$  during winter and autumn and decreases in spring and summer, with  $S_{EV}$  also showing a decrease in  $O_3$  for the summer season in Tokyo and Gunma. According to a previous



study (Kayaba and Kajino, 2023a), the atmospheric conditions in the urban area of the GTA are VOC-limited during summer; thus, the reduction of  $\text{NO}_x$  from vehicular emissions leads to an increase in  $\text{O}_3$  over most of the year in Tokyo and Gunma. Regime analyses in Sect. 4.1 clarify the ozone sensitivity regime aside in seasons other than the summer. Furthermore,  $S_{\text{PP}}$ ,  $S_{\text{UHI}}$ , and  $S_{\text{BVOC}}$  indicated negative impacts on the  $\text{O}_3$  formation in Tokyo and Gunma in all seasons. The increase in air pollutants from power plants, mainly  $\text{NO}_x$ , leads to  $\text{NO}$  titration, lowering the  $\text{O}_3$  concentration in Tokyo and Gunma. The power plants are located in the bay area of Tokyo; for which a detailed map is provided in a previous study (Kayaba and Kajino, 2023a). Notably, the mitigation effect of the UHI caused a significant decrease in the  $\text{O}_3$  formation in Tokyo, which is attributed to the kinetics of  $\text{O}_3$ -chemistry and a lowered PBL height, indicating the importance of the introduction of BEVs to mitigate  $\text{O}_3$  problems in urban areas, not only in terms of direct emissions but also the local temperature decrease. The UHI effect was considerably lower in Gunma than it was in Tokyo because Gunma is a suburban/rural site, and the decrease in temperature was not as significant in Gunma, as seen in Fig. 5. Mitigation of the UHI effect also caused a decrease in BVOC emissions in central GTA and an increase on the west side of the GTA. BVOCs have high  $\text{O}_3$ -formation potential; thus, the decrease in BVOC emissions in the center of the GTA lessened the  $\text{O}_3$  concentration in Tokyo and Gunma. The results of the  $S_{\text{UHI}}$  and  $S_{\text{BVOC}}$  scenarios indicate that the introduction of BEVs mitigates the UHI effect, leading to the deactivation of both  $\text{O}_3$  formation and BVOC emissions.

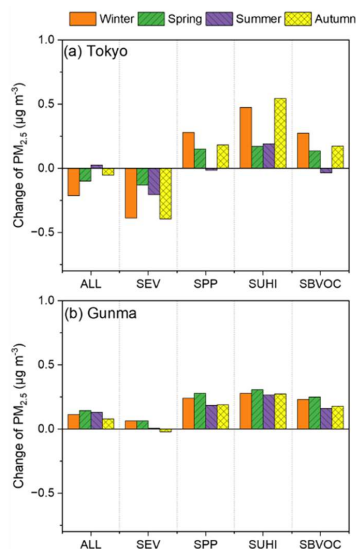


250 **Figure 7: Contributions of the five scenarios to  $\text{O}_3$  in (a) Tokyo and (b) Gunma.**



### 3.3.3 Sensitivity analysis of changes in PM<sub>2.5</sub>

Figure 8 shows the contributions of the five scenarios: ALL, SEV, SPP, SUHI, and SBVOC, to the observed change in the PM<sub>2.5</sub> concentration in Tokyo and Gunma. Unlike O<sub>3</sub>, the seasonal trends in Tokyo and Gunma show different behaviors. The SEV scenario negatively or slightly affected PM<sub>2.5</sub>, owing to the decrease in primary emissions from the exhaust. SEV positively affected PM<sub>2.5</sub> in Gunma, except during spring, while SPP, SUHI, and SBVOC caused an increase in PM<sub>2.5</sub> in both Tokyo and Gunma over most of the year. The intense increase in PM<sub>2.5</sub> that was associated with SPP, SUHI, and SBVOC is discussed in Sect. 4.2. Overall, unlike O<sub>3</sub>, UHIs positively affect PM<sub>2.5</sub> formation, and changes to the UHI are likely to have negative effects on the GTA.



260 **Figure 8: Contributions of the five scenarios to PM<sub>2.5</sub> in GTA in (a) Tokyo and (b) Gunma.**

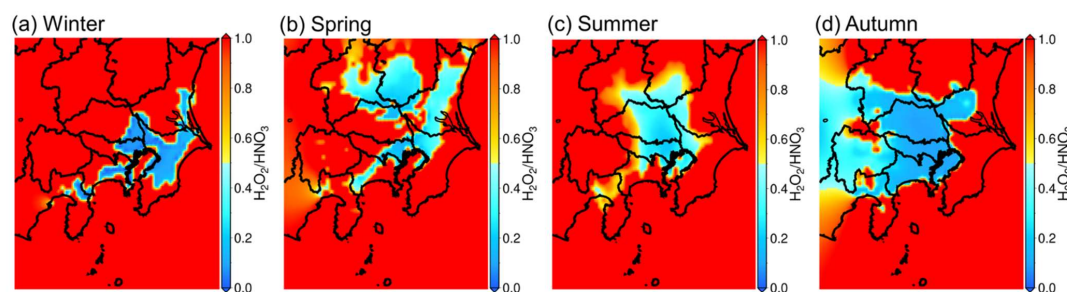
## 4 Discussion

### 4.1 Analysis of ozone formation sensitivity regime in the GTA

Figure 9 shows the seasonal average of the hydrogen peroxide (H<sub>2</sub>O<sub>2</sub>) to HNO<sub>3</sub> ratio (HNR) for the four seasons, each of which was averaged from 9 a.m. to 3 p.m. to represent the sensitivity of the daytime photochemical O<sub>3</sub> formation. Details of the HNR and related topics of O<sub>3</sub> formation-sensitive regimes are described in Appendix B. According to Fig. 9, the values of HNR around Tokyo are less than 0.5 for all seasons, meaning that the central Tokyo region of GTA is VOC-sensitive throughout the year. A relatively wide distribution of VOC-sensitive regimes was observed in spring, summer, and autumn, whereas a



narrow distribution was observed in the winter. According to Sect. 3.3.1 and 3.3.2, the sensitivity of the introduction of BEVs (scenario  $S_{EV}$ ) to  $O_3$  formation showed positive effects in winter and autumn. The value of  $\sim 0.1$  for the HNR in Tokyo in winter and autumn indicates strong VOC sensitivity in Tokyo, and the decrease in the  $NO_x$  might surpass the effect of the VOC reduction, resulting in an increase in  $O_3$  in the  $S_{EV}$  scenario for those seasons. Despite this, the degradation in the  $O_3$  formation that results from the mitigation of the UHI effect contributed to reducing the effect of the increase in  $O_3$ , resulting in a mild increase in  $O_3$  under the ALL scenario in winter and autumn. However, according to Fig. 9(b) and (c), the HNR was close to 0.5 for spring and summer, which is the border between the VOC-sensitive regime and  $NO_x$ -sensitive regime in and around central Tokyo. According to Sect. 3.3.1 and 3.3.2, the  $S_{EV}$  scenario results in a decrease in  $O_3$  formation during spring and summer. Particularly in summer, Fig. 7 shows a decrease in  $O_3$  under the  $S_{EV}$  scenario in both Tokyo and Gunma, despite the fact that Tokyo and Gunma are VOC-sensitive and transition regimes, respectively. This means that the effect of the VOC reduction might surpass the effect of the  $NO_x$  reduction in the warmer seasons, contributing to the decrease in  $O_3$ . Combined with the strong impact that UHI mitigation has as a result of the introduction of BEVs,  $O_3$  was estimated to significantly decrease in the warmer seasons (ALL scenario).  $S_{PP}$ , the scenario used to confirm the effect of the increase in power plant emissions, also contributed to the decrease of  $O_3$ , similar to the UHI mitigation ( $S_{UHI}$  and  $S_{BVOC}$ ).  $NO_x$  is the primary pollutant emitted by power plants. According to Fig. 9, central GTA was mostly VOC-limited, and the increase in  $NO_x$  emissions caused a decrease in  $O_3$ , leading to the observed decrease in  $O_3$  in both Tokyo and Gunma under the  $S_{PP}$  scenario.

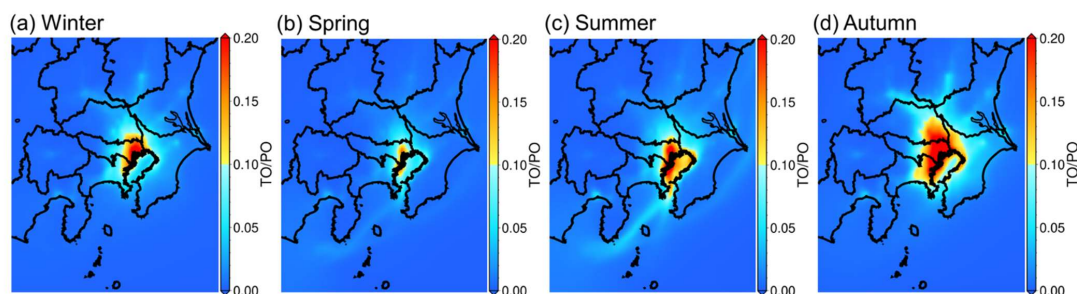


285 **Figure 9: Seasonal trends in the ozone sensitivity regime defined by the ratio of  $H_2O_2/HNO_3$  (HNR) for all four seasons. Analysis was conducted by averaging the results for 9:00 a.m. to 3:00 p.m. each day. Values  $> 0.5$  are  $NO_x$ -limited and  $< 0.5$  are VOC-limited.**

In addition to  $O_3$  formation sensitivity, the effect of NO titration is important when discussing the impact of changes in primary emissions on direct  $O_3$  formation (Akimoto et al., 2015). The concentration of total produced ozone, or potential ozone (PO), is defined using  $[PO] = [O_3] + [TO]$ . TO is the titrated- $O_3$  from NO titration with concentration defined by  $[TO] = [NO_2] - \alpha[NO_x]$ , where  $\alpha$  is the ratio of  $O_3$  formed by the reaction of  $NO + O_3 \rightarrow NO_2$  (NO-titration). The parameter  $\alpha$  is the ratio of  $O_3$  that is scavenged by the titration of NO, for which a value of 0.1 has been widely accepted in Japan (Itano et al., 2007; Akimoto et al., 2015). Figure 10 shows the seasonal variation in the ratio of TO to PO (TO/PO), which exhibits the contribution of the NO titration effect to the total  $O_3$  formation. The effect of NO titration is strong in the central area of Tokyo throughout the year because of high  $NO_x$  emissions and relatively high  $O_3$  concentrations. Compared with the spring and summer, a wider



295 distribution of high TO/PO values is observed for winter and autumn, more relevant NO titration effects in these seasons. According to Fig. 7, a positive increase in  $O_3$  was observed in winter and autumn with the introduction of BEVs, which might be partly attributed to the NO titration effect that results from  $NO_x$ -reduction.

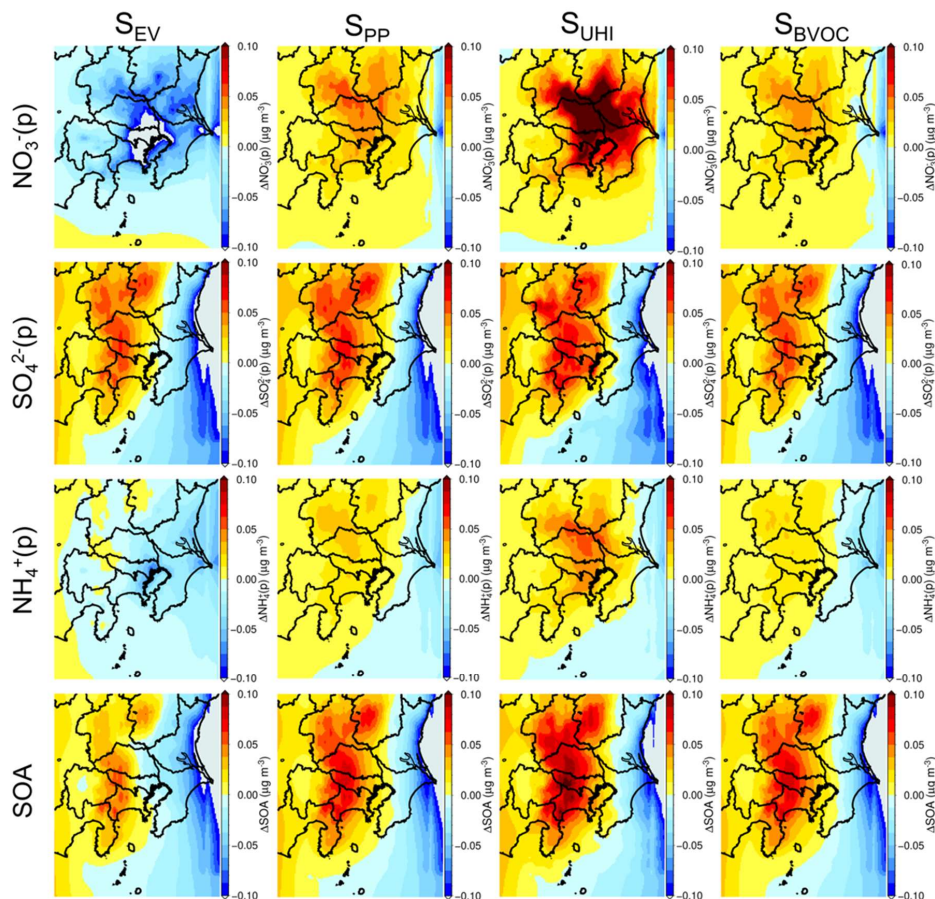


300 **Figure 10: Seasonal trend in the effect of NO-titration defined by the ratio of titrated- $O_3$ /potential- $O_3$  (TO/PO) for the four seasons. NO-titration is strong in regions where the value of TO/PO is high.**

#### 4.2 Composition analysis of the change in $PM_{2.5}$ concentration resulting from the introduction of BEVs

The annual distributions of the four components of  $PM_{2.5}$ ,  $NO_3^-(p)$ ,  $SO_4^{2-}(p)$ ,  $NH_4^+(p)$ , and secondary organic aerosols (SOA) for the four sensitivity scenarios are shown in Fig. 11. In terms of  $S_{EV}$ , a large decrease in  $NO_3^-(p)$  was estimated due to the reduction in the primary emission of NO. On the other hand,  $SO_4^{2-}(p)$  increased because of the increase in the  $O_3$  level, enhancing the OH formation and increasing the  $SO_4^{2-}(p)$  that is attributed to the  $SO_2 + OH$  reaction. Increases in  $SO_4^{2-}(p)$  enhance SOA formation, leading to a large increase (Budisulistiorini et al., 2017). The  $S_{UHI}$  sensitivity scenario shows an increase in all components. It is well known that  $NO_3^-(p)$  is in equilibrium with  $HNO_3(g)$  in the atmosphere, and that the equilibrium shifts to the  $NO_3^-(p)$  form when the temperature increases, which is why a strong increase in  $NO_3^-(p)$  occurs in the  $S_{UHI}$  sensitivity scenario (Morino et al., 2006). The decrease in temperature that results from the introduction of BEVs in the  $S_{UHI}$  scenario enhances the condensation of SOA, leading to an increase in the remaining components (Sheehan and Bowman, 2001). The most important result in the  $S_{BVOC}$  scenario is the increase in the components of  $PM_{2.5}$  because changes in the UHI leads to changes in the BVOC in this scenario; however, because no temperature change was considered an increase in  $PM_{2.5}$  was not expected. According to Fig. S4, the BVOC emissions decreased owing to changes in the UHI effect in similar regions, where an increase in the  $SO_4^{2-}(p)$  was observed, as seen in Fig. 11. This is unexpected because BVOC is considered a precursor to  $SO_4^{2-}(p)$  in terms of OH generation. Possible reasons for this are discussed in Sect 4.3. The  $S_{EV}$  scenario also assumes the reduction of primary emissions, meaning that a decrease in  $SO_4^{2-}(p)$  was expected. Nevertheless, all components of  $PM_{2.5}$ , especially  $SO_4^{2-}(p)$  and SOA, were observed to increase considerably in the western and northern areas of the GTA, corresponding with the suburban to rural areas.





320 **Figure 11: Contributions of the four sensitivity scenarios to the annual average PM<sub>2.5</sub> composition in the GTA.**

### 4.3 Correlation between PM<sub>2.5</sub> increase, the change in BVOC emissions, and the UHI effect

The sensitivity scenarios for  $S_{BVOC}$  described in Sect. 4.2 showed an unreasonable increase in  $SO_4^{2-}(p)$  and a simultaneous increase in SOA. This indicates that not only the change in temperature and PBL height, but also the change in BVOC emissions caused by the mitigation of the UHI effect, led to increases in  $SO_4^{2-}(p)$  and SOA. According to Fig. S4, a decrease in the BVOC emissions was observed in some parts of the central to northern regions of the GTA, with an increase observed in other areas. Considering the change in the BVOC emissions, the change in  $SO_4^{2-}(p)$  in the  $S_{BVOC}$  scenario can be explained by atmospheric-chemical discussions that are divided into daytime and nighttime, as seen in the following.



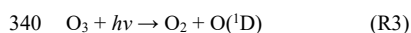


#### 4.3.1 Sensitivity analysis of daytime PM<sub>2.5</sub> changes

First, considering the daytime S<sub>BVOC</sub> scenario, the simplified oxidation reactions of BVOC and SO<sub>2</sub> are described in the following chemical reactions. BVOC oxidation by O<sub>3</sub> is not considered in this discussion because the rate of oxidation by OH is much faster than that associated with O<sub>3</sub>.



In both R1 and R2, the particle formation schemes are simplified. The atmospheric oxidizer OH is generated by the following photochemical reactions during the daytime:



Assuming a steady-state approximation of the intermediate radicals, the rates of O(<sup>1</sup>D) and OH are described by Eqs. 1 and 2:

$$\frac{d[\text{O}(^1\text{D})]}{dt} = k_{\text{R3}}[\text{O}_3] - k_{\text{R4}}[\text{O}(^1\text{D})][\text{H}_2\text{O}] = 0 \quad (1)$$

$$\frac{d[\text{OH}]}{dt} = 2k_{\text{R4}}[\text{O}(^1\text{D})][\text{H}_2\text{O}] - k_{\text{R1}}[\text{BVOC}][\text{OH}] - k_{\text{R2}}[\text{SO}_2][\text{OH}] = 0 \quad (2)$$

where  $k_{\text{R1}}$ ,  $k_{\text{R2}}$ ,  $k_{\text{R3}}$ , and  $k_{\text{R4}}$  are the rate constants in Eq. R1, R2, R3, and R4. Combining Eqs. 1 and 2, the rate of SO<sub>4</sub><sup>2-</sup>(p) formation is described by Eq. 3, as follows:

$$\frac{d[\text{SO}_4^{2-}(\text{p})]}{dt} = k_{\text{R2}}[\text{SO}_2][\text{OH}] = \frac{2k_{\text{R2}}k_{\text{R3}}[\text{SO}_2][\text{O}_3]}{k_{\text{R1}}[\text{BVOC}] + k_{\text{R2}}[\text{SO}_2]} \quad (3)$$

According to SAPRC-07 applied in the chemical transport model in this study,  $k_{\text{R1}}$  lies in the order  $\sim 10^{-11}$  cm<sup>3</sup> molecule<sup>-1</sup> s<sup>-1</sup> and  $k_{\text{R2}}$  around  $\sim 10^{-13}$  cm<sup>3</sup> molecule<sup>-1</sup> s<sup>-1</sup>. The BVOC concentration is much higher than the SO<sub>2</sub> concentration in suburban and rural areas, meaning that  $k_{\text{R1}}[\text{BVOC}]$  is much higher than  $k_{\text{R2}}[\text{SO}_2]$ . Thus, Eq. 3 can be transformed into Eqs. 4, as follows:



$$r_{\text{SO}_4} \equiv \frac{d[\text{SO}_4^{2-}(\text{p})]}{dt} = k_{\text{R}_2}[\text{SO}_2][\text{OH}] = \frac{2k_{\text{R}_2}k_{\text{R}_3}[\text{SO}_2][\text{O}_3]}{k_{\text{R}_1}[\text{BVOC}]} \quad (4)$$

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According to Eq. 4, the rate of  $\text{SO}_4^{2-}(\text{p})$  formation is proportional to  $\text{O}_3$  and  $\text{SO}_2$ , and inversely proportional to BVOC. In terms of the chemical explanation, if BVOC emissions decrease due to changes in the UHI effect, the rate of reaction R1 is weakened; and the redundant OH enhances reaction R2, which is why  $r_{\text{SO}_4}$  is inversely proportional to BVOC. According to Fig. 7, mitigation of the UHI effect led to a decrease in the  $\text{O}_3$  in the  $\text{S}_{\text{BVOC}}$  scenario, degrading the chemical reaction, and the decrease in BVOC was caused by the mitigation of UHI via the introduction of BEVs. Therefore, if the rate of  $\text{SO}_4^{2-}(\text{p})$  formation is calculated using Eq. 4, increases occur in the  $\text{S}_{\text{BVOC}}$  scenario as compared with the BASE scenario, indicating that  $\text{SO}_4^{2-}(\text{p})$  formation is enhanced because the decrease in  $\text{O}_3$  is lower than the decrease in BVOC. BVOC was assumed to be isoprene (because ~70 % of the BVOC emissions in Japan are isoprene (Hata et al., 2023)), and the ratio of  $r_{\text{SO}_4}$  in BASE and  $\text{S}_{\text{BVOC}}$  ( $\eta = r_{\text{SO}_4}^{\text{SBVOC}}/r_{\text{SO}_4}^{\text{BASE}}$ ) in daytime from 9:00 a.m. to 3:00 p.m. was calculated, with the results shown in Fig. 12(a). The  $\eta$  increases in the western and northern regions of GTA correspond to the distribution of the  $\text{SO}_4^{2-}(\text{p})$  increase described in Fig. 11. More qualitatively, the reduction in the BVOC emissions reduced OH consumption, and the resulting OH enhanced  $\text{SO}_4^{2-}(\text{p})$  formation in the western and northern regions of the GTA. Similarly, the decrease in primary emissions in the  $\text{S}_{\text{EV}}$  also led to a decrease in the  $\text{NO}_x$  and VOCs emissions, which might have enhanced the  $\text{SO}_2 + \text{OH}$  reaction, causing an increase in  $\text{SO}_4^{2-}(\text{p})$  and ultimately enhancing the formation of SOA (Sheehan and Bowman, 2001).

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#### 375 4.3.2 Sensitivity analysis of nighttime $\text{PM}_{2.5}$ changes

The photochemical reactions R3 and R4 do not occur during the night, when OH is generated from different sources. One of the well-known sources of OH at night is alkene ozonolysis, leading to the formation of Criegee intermediates (CIs), and both the vibrationally excited and stabilized CIs form OH via unimolecular decomposition (Vereecken et al., 2017). These reaction pathways can be appended and described, as seen in reaction R5:

380



where  $\nu$  is a stoichiometric number that depends on the type of alkene. Alkenes also consume OH radicals via reaction R6, and the rates of OH consumption by alkenes are higher than those of other types of VOCs, such as alkanes and aromatics.

385



$\text{SO}_4^{2-}(\text{p})$  is formed by reaction R2. Note that stabilized-CIs are also known to be strong oxidizers of  $\text{SO}_2$  while the effect of  $\text{SO}_4^{2-}(\text{p})$  formation from stabilized CIs is almost negligible in the GTA (Nakamura et al., 2023). The steady-state approximation of the OH formation rate is given by Eq. 7:

390



$$\frac{d[\text{OH}]}{dt} = \nu k_{R5}[\text{Alkene}][\text{O}_3] - k_{R6}[\text{Alkene}][\text{OH}] - k_{R2}[\text{SO}_2][\text{OH}] = 0 \quad (7)$$

where  $k_{R5}$  and  $k_{R6}$  are the rate constants of R5 and R6, respectively. Coupled with reaction R2, nighttime  $r_{\text{SO}_4}$ , assuming simply  
395 that  $\nu = 1$ , is formulated as follows:

$$\frac{d[\text{SO}_4^{2-}(\text{p})]}{dt} = k_{R2}[\text{SO}_2][\text{OH}] = \frac{2k_{R2}k_{R5}[\text{Alkene}][\text{SO}_2][\text{O}_3]}{k_{R1}[\text{Alkene}] + k_{R2}[\text{SO}_2]} \quad (8)$$

Assuming  $[\text{Alkene}] \gg [\text{SO}_2]$  in the countryside, Eq. 8 can be transformed into Eq. 9 as follows:  
400

$$r_{\text{SO}_4} \equiv \frac{d[\text{SO}_4^{2-}(\text{p})]}{dt} = k_{R2}[\text{SO}_2][\text{OH}] = \frac{2k_{R2}k_{R3}}{k_{R1}}[\text{SO}_2][\text{O}_3] \quad (9)$$

Thus, nighttime  $r_{\text{SO}_4}$  is directly proportional to the concentration of  $\text{O}_3$ . Chemically, nighttime alkene ozonolysis increases the number of OH radicals through CI-chemistry, indicating that  $\text{O}_3$  concentration is the key factor for OH concentration and  $r_{\text{SO}_4}$ .  
405 The value of  $\eta$  ( $= r_{\text{SO}_4}^{\text{SBVOC}}/r_{\text{SO}_4}^{\text{BASE}}$ ) in nighttime from 9:00 p.m. to 3:00 a.m. was calculated using Eq. 9, and the results shown in Fig. 12(b). The fact that  $\eta$  takes a value of  $> 1.0$  in central to north regions of GTA means that  $r_{\text{SO}_4}$  will increase in  $\text{S}_{\text{BVOC}}$  for most of the GTA. The increase in nighttime  $r_{\text{SO}_4}$  is almost equivalent to the increase in daytime  $r_{\text{SO}_4}$ , and  $r_{\text{SO}_4}$  increases in the  $\text{S}_{\text{BVOC}}$  scenario compared with BASE in both daytime and nighttime. As mentioned in Sect. 4.3.1, similar arguments can be made for the  $\text{S}_{\text{EV}}$  scenario, where OH is treated as the source for the oxidation of alkenes. The nitrate radical ( $\text{NO}_3$ ) that is  
410 formed by the reaction of  $\text{NO}_2$  and  $\text{O}_3$  is also known as an important oxidizer of alkenes at night (Ng et al., 2017).  $\text{NO}_3$  adducts the C=C double bond of alkenes, finally forming organic peroxides ( $\text{RO}_2$ ), which affect the nocturnal  $\text{HO}_x$  cycle. For this reason, the reaction of alkenes with  $\text{NO}_3$  is also expected to enhance OH formation and subsequent  $\text{SO}_2$  oxidation; however, discussion of  $\text{NO}_3$  is omitted in this study for simplicity. As a conclusion of Sect. 4.3, mitigation of the UHI caused an increase in  $\text{SO}_4^{2-}(\text{p})$  and subsequent  $\text{PM}_{2.5}$  formation; however, the change in the UHI affected the BVOC emissions in the GTA, and  
415 the decrease in BVOC emissions itself also caused an increase in the  $\text{SO}_4^{2-}(\text{p})$  formation due to enhancement of the  $\text{SO}_2 + \text{OH}$  reaction.

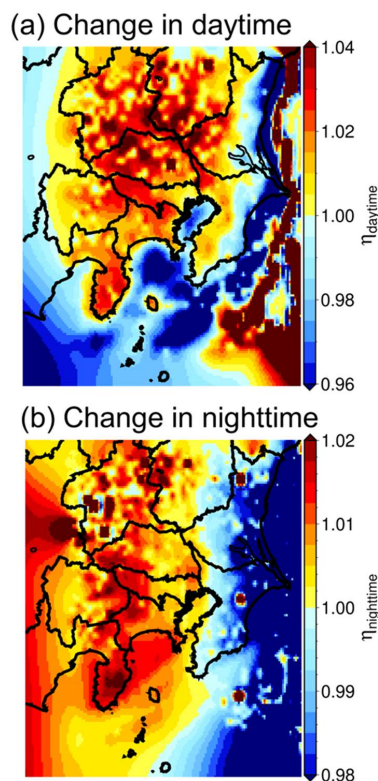


Figure 12: Annual mean ratio of  $r_{SO_4}$ ,  $\eta = r_{SO_4}^{SBVOC}/r_{SO_4}^{BASE}$  obtained by BASE and SBVOC for (a) daytime from 9:00 a.m. to 3:00 p.m., and (b) nighttime from 9:00 p.m. to 3:00 a.m.

#### 420 4.4 Impact of the introduction of BEVs on premature mortality

Previous sections have suggested that the introduction of BEVs in the GTA leads to an increase in  $O_3$  in the urban area and an increase in  $PM_{2.5}$  in the suburban/rural area. To evaluate how these effects impact human health, the changes in premature mortality that resulted from changes in  $O_3$  and  $PM_{2.5}$  were estimated. The change in the number of deaths,  $\Delta P$ , was calculated using Eq. 5, as follows:

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$$\Delta P = \text{sgn}(\Delta C) \cdot (1 - e^{-\beta|\Delta C|}) \cdot P_d \quad (5)$$

where  $\text{sgn}$  is a sign function,  $\Delta C$  is the change in concentration of air pollutants, and  $P_d$  is the base number of deaths in the calculated region.  $\beta$  is an epidemiological parameter that is defined by Eq. 6, as follows:



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$$\beta = \frac{\ln(\text{RR})}{\Delta C_{\text{RR}}} \quad (6)$$

where RR is the relative risk from exposure to air pollutants and  $\Delta C_{\text{RR}}$  is the change in the concentration of a pollutant. Values of 1.003 for MDA8 O<sub>3</sub> change and 1.04 for daily average PM<sub>2.5</sub> with  $\Delta C_{\text{RR}} = 10 \mu\text{g m}^{-3}$  were used in this study (Poppe III et al., 2002; WHO Regional Office for Europe, 2008). The calculated target was the changes before and after the introduction of BEVs (the ALL scenario) for the entire region of the GTA (D3 in Fig. 1). The results showed that the change in the number of deaths caused by the change in O<sub>3</sub> concentration was -175 ( $-1.41 \times 10^{-4}$  % of the total population), while the changes in the PM<sub>2.5</sub> concentration led to a decrease of -77 ( $-6.21 \times 10^{-5}$  % of the total), indicating that the number of deaths decreased after the introduction of BEVs with regards to both O<sub>3</sub> and PM<sub>2.5</sub>. Premature mortality in association with changes in O<sub>3</sub> after the introduction of BEVs has been estimated previously for the urban part of the GTA; however, the effects in the rural area were not considered (Hata et al., 2019). According to the results of this study, the O<sub>3</sub>-related premature deaths over the entire GTA following the introduction of BEVs saved 175 lives in the region. Furthermore, PM<sub>2.5</sub> concentration increases were observed mainly in the suburban/rural areas, which is due to the change in kinetics before and after the introduction of BEVs. Nevertheless, the population is condensed within the urban areas, where the estimated decrease in PM<sub>2.5</sub> was due to the decrease in primary emissions, and 77 deaths were prevented in the GTA. Therefore, the introduction of BEVs in the GTA is effective, not only in supporting GHG emission issues but also in mitigating the health impacts that result from air pollution. The change in UHI after the introduction of BEVs is one of the most important factors in the change in O<sub>3</sub> and PM<sub>2.5</sub>. Nevertheless, it should be noted that in terms of PM<sub>2.5</sub>, 122 more premature deaths were estimated for rural/suburban areas due to the increase in PM<sub>2.5</sub>, mainly because of changes in the UHI and the increase in emissions from power plants, indicating that local health risks are likely to be induced by the introduction of BEVs.

## 5 Summary

The impacts of the introduction of BEVs on ground-level O<sub>3</sub> and PM<sub>2.5</sub> were evaluated by chemical transport modeling in the GTA, with a particular focus on the effect of the change in UHI. The results suggest O<sub>3</sub> increases in urban areas and PM<sub>2.5</sub> increases in suburban/rural areas. The change in the UHI contributed to a decrease in O<sub>3</sub> formation due to the degradation of O<sub>3</sub>-related atmospheric chemistry, while it contributed to an increase in PM<sub>2.5</sub> due to the enhancement of particle condensation and, in some regions, a decrease in BVOCs, which led to the enhancement of SO<sub>4</sub><sup>2-(p)</sup> formation. The change in the number of premature deaths after introducing BEVs was calculated to be -175 in association with O<sub>3</sub> and -77 for PM<sub>2.5</sub> over the entire GTA, suggesting positive impacts on human health. Previous studies have mainly focused on the effects of the decrease in primary emissions due to the shift from ICVs to BEVs on ground-level O<sub>3</sub>; however, no studies have shown clear evidence of how the mitigation of the UHI by the introduction of BEVs has a local impact on O<sub>3</sub> and PM<sub>2.5</sub>. This study showed the relevant



impact of UHI mitigation on the decrease in O<sub>3</sub> and increase in PM<sub>2.5</sub>, for which the absolute contributions were as high as the change in primary emissions. Despite these findings, this study only focused on road transportation; however, a similar treatment could be considered for all stationary sources and other transportation vehicles such as ships, to clarify how electrization changes the local UHI effect its effect on air pollution. Finally, this study focused on 2017, but the entire transformation from ICVs to BEVs is expected to reach completion, such as the 2050s, when the atmosphere will have changed; thus, the O<sub>3</sub> formation sensitivity regime will also differ. Future research should therefore implement emission inventories and meteorology for different time periods to comprehensively evaluate the effect of emission strategies and the UHI effect.

### Appendix A: Estimation of the changes in anthropogenic heat as a result of converting ICVs to BEVs in 2017

#### A.1 Modelled vehicles

Tables A1 and A2 show the specifications of the modeled vehicles used to estimate the change in anthropogenic heat after substitution with BEVs. Three types of vehicles were compared in this study: light passenger cars, normal passenger cars, and small heavy-duty vehicles. Each type of vehicle was compared, and the same model used for ICVs and BEVs, with differences thus primarily involving the energy source (engine or battery). Based on the differences between ICVs and BEVs listed in Tables A1 and A2, the change in anthropogenic heat was estimated as described in the following sections.

**Table A3: Specifications of passenger cars modeled in this study (L/100 km for ICVs and Wh/km for BEVs).**

Property	Light passenger car		Normal passenger car	
	Conventional	BEV	Conventional	BEV
Category				
Manufacturer	Nissan	Nissan	Toyota	Nissan
Model	SBA-B43W	ZAA-B6AW	BA-MZEA17W-AWXNB	ZAA-ZE1
Total weight (kg)	1060	1290	1535	1795
WLTC (L/100km and Wh/km)*	4.31	124	5.24	155
Urban (L/100km and Wh/km)*	5.15	100	7.35	133
Rural (L/100km and Wh/km)*	4.02	113	5.03	145
Highway (L/100km and Wh/km)*	4.15	142	4.46	171

**Table A2: Specifications of small heavy-duty vehicles modeled in this study (L/100km for ICV and Wh/km for BEV).**

Property	Small heavy-duty vehicle	
	Conventional	BEV
Manufacturer	Mitsubishi	Mitsubishi



Model	2RG-FBAV0	ZAB-FEAVK
Total weight (kg)	4495	5015
JH25 (L/100km and Wh/km)*	8.46	347

### A.2 Estimation of the changes in exhaust emissions for the three types of vehicles

480 The change in the emissions of primary air pollutants due to the substitution of ICVs with BEVs was estimated by multiplying the emission factors with the sources of engine exhaust emissions, evaporative emissions from vehicles (hot-soak loss, diurnal breathing loss, and running loss), and evaporative emissions from gasoline service stations as defined in the emission inventory provided by the Ministry of the Environment (Shibata and Morikawa, 2021). Light and normal passenger cars and small heavy-duty ICVs were assumed to be substituted by BEVs. The emissions from engine exhaust, evaporation from vehicles, and 485 evaporation from service stations were set to zero by multiplying the emission factor of 0 by the emission inventories. The primary emissions of particles from brake and tire dust were not examined in this study.

### A.3 Estimation of the change of primary emissions from powerplants in GTA

The change in the emissions of primary air pollutants from power plants as a result of substituting ICVs with BEVs was estimated based on the catalog data of the three BEVs listed in Tables A1 and A2, statistical data of the number of vehicles in 490 the GTA (Database from Automobile Inspection & Registration Information Association, 2024) and the average mileage per year (Ministry of Land, Infrastructure, Transport, and Tourism, 2024) listed in Table A3.

**Table A3: Number of vehicles in GTA and the average mileage per year set in this study.**

Category	Number of vehicles	Mileage (km/y)
Light passenger	11 286 521	10 575
Normal passenger	6 181 929	10 575
Small HV	3 053 453	14 325

Multiplying the number of vehicles in the GTA, mileage per year, and battery consumption listed in Tables A1 and A2 by the 495 transmission loss of electricity from power plants to charging stations (1.038) resulted in a total energy demand of 49.6 TWh/y for charging BEVs in the GTA. The total energy demand for electricity in the GTA in 2017 was estimated at 147 TWh/y. Thus, the energy demand in powerplants in GTA after substituting ICVs with BEVs increased  $49.6/147 \times 100 = 34 \%$ , and we assumed that the increase in the primary emissions of air pollutants from powerplants was proportional, at 34 %, in the BEV-introduction scenario.



#### 500 **A.4 Estimation of the change in AH for the three types of vehicles**

The change in anthropogenic heat caused by substituting ICVs with BEVs was estimated for each vehicle type based on the energy efficiencies listed in Tables A1 and A2. The thermal enthalpies of gasoline (passenger cars) and light fuel (small heavy-duty vehicles) ICVs were assumed to be 31 250 kJ/L and 35 770 kJ/L, respectively (Agency for Natural Resources and Energy, 2024), allowing the ratio of AH between the ICVs and BEVs to be calculated. For example, the fuel consumption of conventional light passenger cars listed in Table A1 was 5.15 L/100 km. Multiplying the gasoline thermal enthalpy (31250 kJ/L by 5.15 L/100 km, results in a total energy consumption of 1347 kJ/km. The battery consumption of the light passenger car BEV listed in Table A1 is 124 Wh/km = 446 kJ/km. Thus, the ratio of the decrease in AH caused by the introduction of light passenger BEVs was calculated to be  $446/1347 = 0.33$ . Similarly, the ratios of the decrease in AH by the introduction of BEV normal passenger cars and small heavy-duty vehicles were calculated to be 0.34 and 0.41 respectively. Then, by applying statistical data on the number of vehicles (Database from Automobile Inspection & Registration Information Association, 2024) listed in Table A3, the number-weighted ratio of the decrease in AH was calculated to be 0.35. This indicates that the change in AH before and after the introduction of BEVs contributed to the 35 % decrease.

The Ministry of the Environment released the estimated data of UHI-related parameters, including AH, from stationery and transportation sources for the metropolitan area of Tokyo (Ministry of the Environment, 2010). According to this report, the AH from the transportation sector was 505.8 TJ/d, and the AH from all sectors was 1574.3 TJ/d. This study focuses on the electrification of passenger cars and small heavy-duty vehicles, and the ratio of the number of these vehicles in GTA is 89 %, meaning that the AH from the targeted vehicles is calculated at  $505.8 \times 0.892 = 451.5$  TJ/d. The ratio of the decrease in the AH after the introduction of BEVs was calculated to be 0.35; therefore, the decrease in AH after the introduction of BEVs is  $505.8 \times 0.892 \times 0.35 = 157.6$  TJ/d, and the AH after the introduction of BEVs in the transportation sector is  $505.8 - 451.5 + 157.6 = 211.9$  TJ/d. The AH from all sectors after the introduction of BEVs is calculated to be  $1574.3 - 505.8 + 211.9 = 1280.4$  TJ/d. Finally, the ratio of the decrease in the total AH after the introduction of BEVs was calculated to be  $1280.4 / 1574.3 = 0.813$ . Therefore, a total decrease of 81.3 % AH is assumed for the GTA as a result of introducing BEVs under the assumptions made in these discussions.

#### **Appendix B: Details of methods used to determine the ozone formation sensitivity regime**

525 The formation of  $O_3$  is dependent on the concentration of  $NO_x$ , and the rate of  $O_3$  formation is determined by the concentration of VOCs. The balance of  $NO_x$  and VOCs concentrations thus determines whether  $O_3$  formation is positive or negative when emission reduction strategies are implemented. Atmospheric conditions in high- $NO_x$  regions, such as highly urbanized areas, are considered VOC-limited, while conditions in high-VOC regions such as forest, are  $NO_x$ -limited (Sillman, 1999). In the VOC-limited regime, the reduction in VOC emissions responded to a decrease in  $O_3$  whereas the reduction in  $NO_x$  emissions responded to an increase in  $O_3$ . In contrast, in the  $NO_x$ -limited regime, a reduction in the  $NO_x$  emissions corresponded to a decrease in  $O_3$ , with almost no response to VOC in terms of  $O_3$  formation. Several indicators enabled the determination of the





O<sub>3</sub>-sensitivity regime in the targeted region. For example, the formaldehyde (HCHO) to NO<sub>2</sub> ratio (FNR) is commonly used to distinguish between VOC-sensitive (FNR < 1.0), NO<sub>x</sub>-sensitive (FNR > 2.0), and transition regimes (1.0 ≤ FNR ≤ 2.0) (Tonnesen and Dennis, 2000; Martin et al., 2004; Duncan et al., 2010). The analytical hypothesis of the FNR came from the fact that HCHO is an intermediate product of atmospheric oxidation; thus, the amount of HCHO can be an indicator of VOC. FNR is commonly used in several studies, especially because of the availability of data on the column density of HCHO and NO<sub>2</sub> from satellites such as the Ozone Monitoring Instrument (Vander, 2010) and TROPOspheric Monitoring Instrument (van Geffen et al., 2020). Nevertheless, HCHO is produced not only by the oxidation of VOCs, but also by primary emission, and the value of the FNR strongly depends on the location (Jin et al., 2017). In addition to FNR, the hydrogen peroxide (H<sub>2</sub>O<sub>2</sub>) to HNO<sub>3</sub> ratio (HNR) has been proposed as an indicator of the ozone sensitivity regime (Sillman, 1995). The analytical hypothesis of the HNR was based on the fact that both H<sub>2</sub>O<sub>2</sub> and HNO<sub>3</sub> are end products of the HO<sub>x</sub> cycle and are produced via the termination reaction of two OH radicals to form H<sub>2</sub>O<sub>2</sub> and the reaction of OH and NO<sub>2</sub> to form HNO<sub>3</sub>. According to Kayaba et al., regimes are distinguished as VOC-sensitive (HNR < 0.5) and NO<sub>x</sub>-sensitive (HNR > 0.5) in the GTA (Kayaba et al., 2023a). Unlike the FNR, the HNR does not directly depend on primary emissions; therefore, an accurate analysis can be performed without considering location. Therefore, in this study, HNR was used to discuss the O<sub>3</sub> sensitivity regime.

#### Data availability

All the data simulated in this study are available upon request from the corresponding author.

#### Competing interests

The authors declare that they have no conflict of interest.

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#### Author contributions

HH and TI proposed the research concept. NM calculated the energy consumption and anthropogenic heat from the data of conventional and electric vehicles. HH performed the numerical simulations and analyses and drafted the manuscript. NM and TI reviewed the manuscript for accuracy.



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