

Contrasts in the marine inorganic carbon chemistry of the Benguela Upwelling System since the Last Glacial Maximum

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Abstract. Upwelling regions are dynamic systems where relatively cold, nutrient- and CO₂-rich waters reach to the surface from the deep. CO₂ sink or source properties of these regions are dependent not only on the dissolved inorganic carbon content of the upwelled waters, but also on the efficiency of the biological carbon pump which constrains the drawdown of atmospheric CO₂ in the surface waters. The Benguela Upwelling System (BUS) is a major upwelling region with one of the most productive marine ecosystems today. However, contrasting signals reported on the variation in upwelling intensities based on, for instance, foraminiferal and radiolarian indices over the last glacial cycle indicate that a complete understanding of (local) changes is currently lacking. To reconstruct changes in the CO₂ history of the Northern Benguela upwelling region over the last 27 kyrs, we used a box core (64PE450-BC6) and piston core (64PE450-PC8) from the Walvis Ridge. Here, we apply various temperature and *p*CO₂-proxies, representing both surface (U_{37}^K , $\delta^{13}C$ of alkenones) and subsurface (Mg/Ca, $\delta^{11}B$ in planktonic foraminiferal shells) processes. Reconstructed *p*CO₂ records suggest enhanced storage of carbon at depth during the Last Glacial Maximum. The offset between $\delta^{13}C$ of planktonic (high $\delta^{13}C$) and benthic foraminifera (low $\delta^{13}C$) suggests an evidence of a more efficient biological carbon pump, potentially fuelled by remote and local iron supply through aeolian transport and dissolution in the shelf regions, effectively preventing release of the stored glacial CO₂.

1 Introduction

Upwelling systems are crucial components in the global carbon cycle thanks to intense biogeochemical cycling and enhanced biological productivity (Turi et al., 2014). Upwelling zones return the cold, nutrient- and CO₂-rich waters from depth to the surface which is also reflected in regional changes in surface water inorganic carbon chemistry. The connection between the deep and surface ocean thereby provides a potential mechanism linking changes in ocean circulation and chemistry with the atmosphere. Still, the shoaling of the thermocline and nutricline in these regions also favours phytoplankton growth to such a degree that these areas represent majority of the most productive regions of the ocean (Fig. 1 a). Thus, the leakage of CO₂ from the depths to the atmosphere is negated by biological sequestration, simultaneously rendering its quantification a challenging undertaking. The surface waters of the upwelling system undergo an increase in the partial pressure of CO₂ (*p*CO₂) and decrease in pH due to upwelling of deep CO₂-rich water. In turn, the enhanced primary productivity due to increased nutrients result in drawdown of *p*CO₂ by converting CO₂ into organic carbon, after which it may be returned to the deep ocean via the biological carbon pump (BCP; Volk and Hoffert, 1985; Longhurst and Glen Harrison, 1989; Ducklow et al., 2001; Turi et al., 2014; Hales et al., 2005; Muller-Karger et al., 2005). Ultimately, the net CO₂ flux from the ocean to the atmosphere is a function of the balance between upwelling strength (increase in CO₂) and efficiency of the BCP (drawdown of CO₂). On geological time scales this efficiency may have varied, potentially modulating local air-sea CO₂ balance (Kohfeld et al., 2005; Kwon et al., 2009; Parekh et al., 2006; Hain et al., 2014).

The efficiency of the BCP determines how much of newly produced particulate organic carbon at the surface is transported to the deep (Volk and Hoffert, 1985; Hain et al., 2014). During primary production, nutrients are consumed (e.g., nitrate, phosphate; Redfield, 1958) from the surface ocean and dissolved inorganic carbon (DIC) is taken up in organic matter, which is also reflected by the enrichment in ¹³C of the surface DIC (Degens et al.,

1968). This implies that we can use seawater carbon isotopes as proxy for the efficiency of the BCP. Seawater carbon isotopes can be reconstructed using the carbon isotopic composition ($\delta^{13}\text{C}$) in shells of carbonate producers, such as foraminifera. During high productivity periods, the enhanced carbon uptake at the sea surface will enrich the shells of planktonic foraminifera in ^{13}C . At the same time, the ^{13}C -depleted carbon transported to the deep as organic matter will decrease the ^{13}C content of the deep water DIC pool, resulting in low $\delta^{13}\text{C}$ values in the benthic foraminifera shells (Fig. 1 b). Therefore, the stable carbon isotopic composition of these inorganic archives are imprinted by complex processes related to both surface to deep gradients and ocean circulations. Water masses of different origins carry distinct $\delta^{13}\text{C}$ compositions resulting in integrated signatures of air-sea exchange and production/remineralization related to different water masses within the foraminiferal shells. The difference between planktonic and benthic $\delta^{13}\text{C}$, however also records a measure for the efficiency of the BCP, where more divergent values between the surface and the deep indicate a more efficient BCP (Hilting et al., 2008).

This study focuses on the Benguela Upwelling System (BUS) as it is one of the major upwelling regions, where strength of the upwelling and productivity changed over glacial/interglacial timescales. Whether upwelling intensity was stronger during glacial periods (Oberhänsli, 1991; Little et al., 1997; Kirst et al., 1999; Mollenhauer et al., 2003) or interglacial periods (Diester-Haass et al., 1992; Des Combes and Abelmann, 2007) is, however, still debated. Inconsistencies in the published body of work is possibly caused by seasonal differences between proxy signal carriers and/or major spatial (depth) related gradients, which is especially true for regions with strong CO_2 flux dynamics (Fig. 1 c). Exchange of CO_2 between seawater and atmosphere at these regions may be constrained only by applying multiple proxies that comprise various living depths and seasonal preferences. Therefore here, we compare organic and inorganic proxies for temperature (U_{37}^K , Mg/Ca) and the carbon system (alkenone- $\delta^{13}\text{C}$, foraminiferal- $\delta^{11}\text{B}$) with reconstructed efficiency of the BCP in the Benguela upwelling area to unravel the potential role of such areas in known changes in atmospheric $p\text{CO}_2$ on glacial-interglacial time scales. At the same time this allows comparing proxies and investigate (in)consistencies between different carbon system- and temperature proxies.

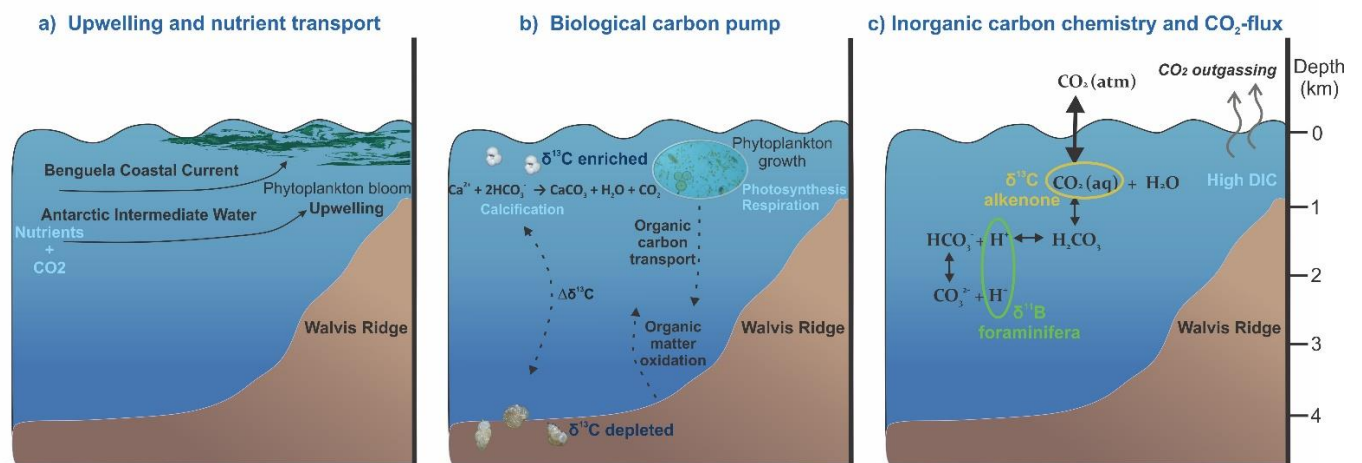


Figure 1: Cross sections of the Benguela Upwelling System depicting the characteristics of an upwelling region, where a) nutrient- and CO_2 -rich waters are upwelled to the surface, b) high productivity contributes to the drawdown of CO_2 in the surface layers via the biological carbon pump, and c) the upwelling strength

and efficiency of the biological carbon determines variations in the marine inorganic carbon chemistry, and hence CO₂-flux.

2 Oceanographic setting

The BUS is one of the four major Eastern Boundary Upwelling Systems, it is located between 15° and 34° S along the coastline of Africa (Hill, 1998; Hart and Currie, 1960). This region bears the highest productivity today among the Eastern Boundary Upwelling Systems, fuelled by nutrients transported mainly from the higher latitudes. Advection of the cold and nutrient rich water is a persistent phenomenon throughout the year (Carr, 2001; Chavez and Messié, 2009) and the magnitude of the particulate organic carbon (POC) flux from the surface to the deep exceeds 20 gC m⁻² yr⁻¹ (Henson et al., 2011; Laws et al., 2000; Devries and Weber, 2017).

The BUS is associated with the South Atlantic anticyclonic gyre which gives rise to upwelling on its southeastern flank where it meets the African continent (Peterson and Stramma, 1991). The low-pressure system over western South Africa causes a pressure gradient between the continent and the ocean and thereby strengthen the southerly wind stress off the coast of Angola and Namibia. The interplay between the equatorward trade winds, the Coriolis force, and the presence of the continental boundary lead to the offshore transport of surface waters. As such, this causes coastal upwelling of nutrient-rich South Atlantic Central Water (formed in the western South Atlantic; Stramma and England, 1999) and Antarctic Intermediate Water (AAIW). The upwelled waters are transported equatorward along the coast of Africa via the Benguela Current (BC) giving rise to high biological productivity. Filaments of productive waters can be seen extending from the African continent (Fig. 2). Finally, the Walvis Ridge potentially plays a role in affecting local hydrography and hence the position of the upwelling (Peterson and Stramma, 1991).

The BC with its two main branches, the Benguela Oceanic Current and the Benguela Coastal Current, is the major northward flowing component of the BUS which joins the poleward flowing Angola Current in the north (Stramma and England, 1999). This convergence zone is located between 15°S and 18°S and is known as the Angola-Benguela frontal zone. The upwelling zone is bounded by warm current systems, the Angola Current system in the north and the Agulhas Current system in the south (Shannon and Nelson, 1996; Shillington, 1998; Shannon and O'toole, 2003). Hence, the BC is composed of a mixture of waters originated not only from the mid-latitude surface waters of the Central Southern Atlantic Ocean and the Southern Ocean but also from the Indian Ocean (Gordon, 1986; Lutjeharms and Valentine, 1987). This creates a north-to-south decrease in surface water temperature and salinity in the region (Santana-Casiano et al., 2009). Hydrographic changes in the region over glacial cycles have been related to changes in the transfer of Indian Ocean waters through Agulhas leakage variability (Knorr and Lohmann, 2003; Peeters et al., 2004; Scussolini and Peeters, 2013).

The BUS region is characterized by year-round upwelling of varying intensity due to the seasonal shift of the South Atlantic gyre. This results in stronger upwelling intensities in June-August compared to the rest of the year (Santana-Casiano et al., 2009; Kämpf and Chapman, 2016). The spatial and temporal dynamics of the BUS result in large variability in the associated CO₂ flux. Predominantly, the surface waters within the BUS act as a CO₂ source (Laruelle et al., 2014; Brady et al., 2019; Roobaert et al., 2019), but this may be interrupted by periods

during which it acts as a CO₂ sink due to the high primary productivity (Gruber et al., 2009; Gregor and Monteiro, 2013).

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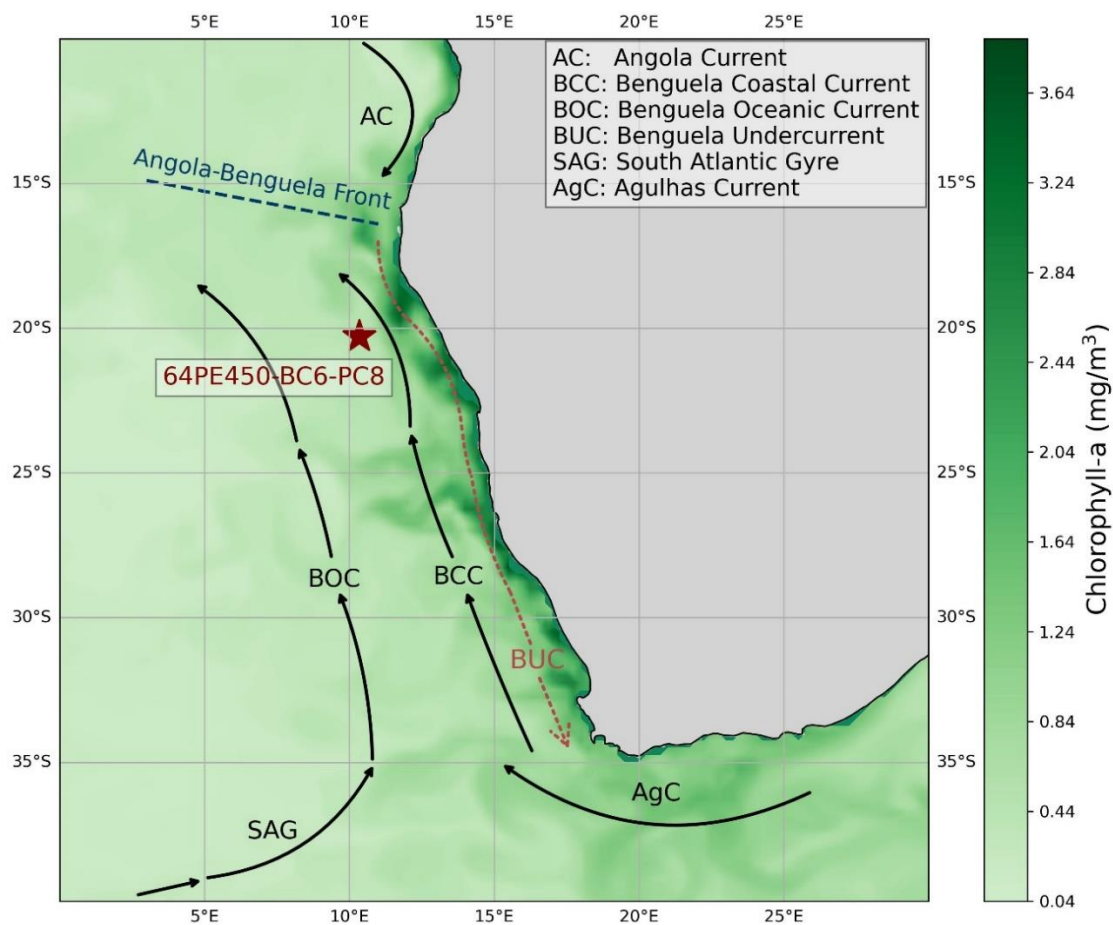


Figure 2: Map showing the location of sediment core 64PE450-BC6-PC8 and the dominant currents shaping the characteristics of the Benguela Upwelling System. The map is overlain with the distribution of surface water chlorophyll-a concentration of July 2023 obtained from Global Ocean Biogeochemistry Analysis and Forecast (E.U. Copernicus Marine Service Information; <https://doi.org/10.48670/moi-00015>). High chlorophyll-a concentrations indicate the high productivity and nutrient-rich upwelled waters of this region today.

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3 Proxy interpretation

Reconstruction of inorganic carbon chemistry can be used to constrain past changes in the CO₂ flux between the ocean and atmosphere. Reconstruction of the complete inorganic carbon system is based on at least two parameters of this system ($p\text{CO}_2$, $[\text{CO}_3^{2-}]$, $[\text{HCO}_3^-]$, pH, [DIC] and total alkalinity), as well as on the knowledge of temperature and salinity (Zeebe and Wolf-Gladrow, 2001). Commonly used tracers for constraining parameters of the marine inorganic carbon chemistry are based on both organic (e.g., $\delta^{13}\text{C}$ of alkenones; Pagani et al., 2002; Pagani, 2014; Popp et al., 1998; Laws et al., 1995) and inorganic (e.g., $\delta^{11}\text{B}$ of foraminifera shells; Hemming and Hanson, 1992; Palmer and Pearson, 2003; Foster and Rae, 2016) proxy signal carriers, although these proxies rarely agree completely for upwelling regions (Seki et al., 2010; Palmer et al., 2010) or in general (Rae et al., 2021).

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Proxies for seawater carbon chemistry have specific inherent complications and their application requires critical assumptions. For instance, previous studies have observed discrepancies between alkenone based $p\text{CO}_2$ reconstruction and ice core records (Palmer et al., 2010; Andersen et al., 1999; Zhang et al., 2013; Witkowski et al., 2020; Jasper et al., 1994), which could be related to disequilibrium between sea surface and the atmosphere, especially at dynamic sites like upwelling regions. However, it may also be explained by the process of CO_2 uptake in the algal cell, if passive diffusion is not the only way alkenone producers acquire CO_2 in the cell, as suggested by the traditional framework of this proxy (Bidigare et al., 1997). Alkenone-producers do use a carbon concentrating mechanism (CCM; Stoll et al., 2019; Reinfelder, 2011; Bolton and Stoll, 2013; Badger, 2021), which enables carbon acquisition in the cell through the active pumping of HCO_3^- to the chloroplast during low $p\text{CO}_2$ conditions. Also, $p\text{CO}_2$ reconstructions based on alkenone $\delta^{13}\text{C}$ values are subject to uncertainties related to the so-called b factor that expresses the effect of multiple parameters related to the physiology of the alkenone producers (Jasper et al., 1994; Rau et al., 1996; Popp et al., 1998). Application of the b factor for the reconstruction of $p\text{CO}_2$ is much debated (e.g., Wilkes and Pearson, 2019) and an adaptation of CCM by the alkenone producers inevitably hampers the application of the proxy. However, there are examples of alkenone-based $p\text{CO}_2$ reconstructions reliably reproducing glacial-interglacial $p\text{CO}_2$ variability (Palmer et al., 2010; Jasper and Hayes, 1990; Bae et al., 2015), potentially related to specific local conditions. As the b value is best represented by a linear relationship to nutrient availability (Bidigare et al., 1997), we here rely on the analysis of barium over calcium ratio (Ba/Ca) in planktonic foraminiferal shells that correlates with seawater Ba concentration, and hence is used as a proxy for seawater $[\text{PO}_4^{3-}]$ (Lea and Boyle, 1989; Lea and Boyle, 1990b, a; Hönisch et al., 2011).

4 Materials and methods

Samples were taken from box core 64PE450-BC6 and piston core 64PE450-PC8 retrieved from the south flank of the Walvis Ridge, both taken at the same location (approximately -20.29 S, 10.35 E) at a water depth of ~1375 mbsf. The box core consisted of 40.59 cm of sediment, whereas the piston core collected 1453 cm (cut into 15 sections, of which we dated the first 100 cm). The top of the piston core was missing and, hence, we used the BC to supplement the missing top of the PC and obtain a near continuous record, which we here refer to as 64PE450-BC6-PC8. To align the box core and the piston core, lightness reflectance data (L^*) was used here as an additional constraint to the radiocarbon dates (Supplementary Fig. S1). The detailed reflectance data (63 μm resolution) shows an overlap between the two cores. The top 4.24 cm of 64PE450-PC8 (later referred to as the disturbed core-top) overlaps with approximately 30.14 cm of 64PE450-BC6 (i.e., from 10.45 to 40.59 cm bsf). This suggests severe compression of the top of 64PE450-PC8, likely due to the piston coring, but the overlap can still be used to align the age model of the two cores.

The composite record was sampled with a resolution varying between 2 and 5 cm to optimize coverage of the glacial-interglacial transition. All samples were freeze-dried and subsequently split in sub-samples to obtain lipid biomarkers and foraminifera from the same core depth.

4.1 Foraminiferal sample cleaning

Due to their relatively high abundance in upwelling regions as well as common use in paleoclimate reconstructions (e.g., Spero and Lea, 1996), we here selected specimens of *Globigerina bulloides* for the planktonic foraminifera-based records. Freeze-dried samples were washed over a 63 μm sieve, dried and further dry sieved to separate size fractions 150-315 μm and 315-425 μm . Specimens of *G. bulloides* were picked from the latter size fraction for analysis of oxygen and carbon isotopes, minimizing any potential impacts of ontogeny. However, as much more specimens were needed, the smaller size fraction was used for radiocarbon, element / calcium (El/Ca), and boron isotope analysis. The foraminifer's size has been shown to affect boron isotopes of several symbiont-bearing planktonic foraminifera species (e.g., *T. sacculifer*, *G. ruber*, *O. universa*; Hönisch and Hemming, 2004; Hennehan et al., 2013; Hennehan et al., 2016), and hence size fraction need to be minimized to avoid introducing uncertainties related to ontogenetic variability. *G. bulloides* is a symbiont-barren species, and therefore, boron isotopes are not affected by pH change in their microenvironments related to the symbionts' physiological processes (i.e., respiration, photosynthesis). However, differences in shell size may correspond to different environmental conditions, and for instance, reflect changes in calcification depth and/or seasons (Jonkers et al., 2013; Osborne et al., 2016). Also, previous studies using *G. bulloides* to reconstruct pH relied on a narrow size fraction when the number of specimens allowed this (Raitzsch et al., 2018; Martinez-Boti et al., 2015). Alternatively, a combination of two or more parameters (e.g., temperature and productivity) may impact shell size, resulting in mixed isotopic signatures within size fractions (Metcalf et al., 2015). Any size-dependent bias on boron isotopes of (symbiont-barren) foraminifera still needs to be investigated, and hence we assume that the data presented here for *G. bulloides* reflect average conditions with respect to depth and seasonal variability. To construct a benthic foraminiferal carbon isotope record, specimens of *Cibicidoides wuellerstorfi* were picked from the 315-425 μm size fraction.

Foraminiferal samples were cleaned prior to the analysis of El/Ca ratios and stable isotopes, following an adapted protocol of Barker et al. (2003). This adapted protocol is as follows: for the analysis of the shells' element concentrations in solution and the boron isotopic composition, specimens were carefully cracked using a scalpel to open up the chambers and release potential clay content from the inside. The samples were subsequently transferred to acid cleaned 1.5 mL vials (Treff) and rinsed three times with deionized water (Milli-Q), twice with methanol, followed by another thorough rinse with deionized water, using ultra-sonication for each rinsing step. To remove all organic material from the shells, samples were placed in a hot block and oxidized with NH_4OH -buffered 1% H_2O_2 solution for 45 minutes at 90 °C. To ensure complete removal of organic material, this step was repeated up to three times based on visual inspection. After the oxidative cleaning, the samples were transferred to new pre-cleaned vials (Treff) and leached with diluted acid (1 mM HNO_3) followed by rinsing the samples three times with deionized water. Because the boron isotope analysis is very sensitive to contamination two additional leaching steps with 1% NH_4OH were added followed by rinsing with deionized water before the acid leaching. Samples for El/Ca and $\delta^{11}\text{B}$ analysis were finally dissolved in 500 μL 0.1 M ultra grade HNO_3 and in 75-80 μL 0.5 M ultra grade HNO_3 , respectively.

Specimens taken for the analysis of El/Ca ratios with LA-Q-ICP-MS and for the measurement of $\delta^{18}\text{O}$ and $\delta^{13}\text{C}$ were cleaned following the same clay removal and oxidative cleaning step as described above but without cracking the shells before the cleaning steps.

4.2 Radiocarbon analysis

220 Radiocarbon analysis ($^{14}\text{C}/^{12}\text{C}$) on 50-100 specimens of well-preserved shells of *G. bulloides* were performed at
Laboratory of Ion Beam Physics, ETH Zürich. The analysis of $^{14}\text{C}/^{12}\text{C}$ followed the protocol described in Wacker
et al. (2013; 2014). Briefly, samples were measured with a gas ion source in a Mini Carbon Dating System
(MICADAS; Synal et al., 2007) with an automated method for acid digestion of carbonates (Wacker et al., 2013).
225 Samples were placed in 4.5 mL exetainer vials (Labco Limited®, UK) and purged with a flow of 60 mL min⁻¹ of
helium for 10 minutes and subsequently leached with 100 µL 0.02 M ultrapure HCl with an automated syringe to
remove adsorbed contaminants. Analysis of the released CO₂ from both the leachate and remaining sample
provided confirmation for the near complete removal of contaminants. The released CO₂ from the leachate was
directly transported by helium to a zeolite trap and injected into the ion source for $^{14}\text{C}/^{12}\text{C}$ analysis. The remaining
230 leached sample was acidified with 100 µL ultrapure H₃PO₄ (85%) and heated at 60 °C for a minimum of 1 hour.
The released CO₂ was then injected in the ion source for analysis (Wacker et al., 2014; Fahrni et al., 2013). The
difference between the radiocarbon values of the leachate and leached samples were less than 5%. Radiocarbon
determinations are given in the conventional radiocarbon ages and corrected for isotopic fractionation via $^{13}\text{C}/^{12}\text{C}$
isotope ratios. Calibration was performed using the Marine20 calibration curve (Heaton et al., 2020) with a local
correction to the marine reservoir age (ΔR) of 146 ± 85 ^{14}C years (Dewar et al., 2012). These calculations were
235 computed using the Bayesian age-depth model in the Bacon v2.3 package for the R statistical programming
software (Blaauw and Christen, 2011).

4.3 Analysis of stable oxygen and carbon isotopes

Pre-weighed 20-40 µg of the shells of *G. bulloides* were dissolved in orthophosphoric acid and analysed at 71 °C
by a Kiel IV device coupled to a MAT 253 Isotope Ratio Mass Spectrometer (IRMS, Thermo Fischer Scientific®)
240 at the NIOZ. Analyses were calibrated using standard bracketing (NBS-19) and the NIOZ house standard (NFHS-
1; Mezger et al., 2016) was used to monitor drift. Accuracy and precision for $\delta^{13}\text{C} = 0.814 \pm 0.04$ ‰ and $\delta^{18}\text{O} =$
 1.024 ± 0.12 ‰ were calculated across several analytical runs of NFHS-1 ($\pm 1\sigma$ SD, n = 64).

4.4 Analysis of foraminiferal Element/Calcium ratios

Prior to the analysis of the samples in solution, a few planktonic foraminifera specimens were screened for
245 preservation to minimize the possibility of diagenetic overprint affecting the geochemical signature of the shells.
For this, the ratios of $^{23}\text{Na}/^{43}\text{Ca}$, $^{24}\text{Mg}/^{43}\text{Ca}$, $^{25}\text{Mg}/^{43}\text{Ca}$, $^{27}\text{Al}/^{43}\text{Ca}$, $^{55}\text{Mn}/^{43}\text{Ca}$ $^{88}\text{Sr}/^{43}\text{Ca}$ were simultaneously
monitored during the ablation of single chambers of *G. bulloides* by Laser Ablation Quadrupole Inductively
Coupled Plasma Mass Spectrometer (LA-Q-ICP-MS). Laser ablation data was acquired on 60-µm diameter spots
with a repetition rate of 4 Hz and a laser energy density of ~ 1 J cm⁻². The JCp (*Porites* sp. coral) nano-pellet was
250 used to monitor instrumental drift and Jct (*Tridacna gigas* giant clam; Okai et al., 2004), MACS-3 and the NIOZ
Foraminifera House Standard-2-Nano-Pellet (NFHS-2-NP; Boer et al., 2022) provided further quality control on
the measurement. NIST SRM610 was used as calibration standard. Data was evaluated both as profiles and shell
averages.

Approximately 40-50 specimens of *G. bulloides* were dissolved for solution analyses using a Sector-Field Inductively Coupled Plasma Mass Spectrometer (SF-ICP-MS, Thermo Fischer Scientific® Element-2). Applied cleaning procedure is based on Barker et al. (2003) as discussed above. A pre-scan of calcium concentrations ($[Ca^{2+}]$) was performed on an aliquot of 30 μ L of the dissolved samples and based on that data subsequently all samples were diluted to match $[Ca^{2+}]$ (100 ppm) for element analyses. Isotopes of ^{25}Mg , ^{138}Ba were measured in low resolution. All samples were measured against 4 ratio calibration standards (De Villiers et al., 2002) and alternated with 0.1 M HNO_3 in between samples to increase the efficiency of wash-out. All samples are drift-corrected using the NFHS-1 standard (Mezger et al., 2016) and three additional standards: NFHS-2 (Boer et al., 2022), JCp, and JCt (Okai et al., 2004), to evaluate accuracy and precision of the analytical runs. Uncertainty from the internal precision on the basis of short term stability is $< 2\%$ for both Mg, and Ba. Samples were analysed in replicates yielding an uncertainty of $< 0.02\text{ mmol mol}^{-1}$ for Mg, and $< 0.14\text{ }\mu\text{mol mol}^{-1}$ for Ba.

4.5 Micro-distillation and boron isotope analysis

Approximately 150 specimens of *G. bulloides* were cleaned for the analysis of boron isotopes. Boron was separated from the calcium carbonate matrix via the micro-distillation technique (Gaillardet et al., 2001; Wang et al., 2010; Misra et al., 2014). 70 μ L of the sample was placed on the lid of a Teflon® fin-legged conical beaker (5 mL) and placed upside down on a hotplate at 100 °C for 20-24 hours. The fin-legged vials were wrapped in aluminium foil to provide a heat gradient for a more efficient separation of boron. Once the micro-distillation was complete, the vials were carefully removed from the hotplate while turning them over and subsequently left for cooling. Sample residue was removed by putting new lids on the beakers and each sample was diluted with 0.2 M HF + 0.2 M HNO_3 for a pre-scan of the boron concentration ($[B]$). Based on the results of the pre-scan, a final dilution was made to set $[B]$ at 5 ppb for the analysis of $\delta^{11}B$.

Analysis of the micro-distilled samples was performed at the NIOZ on a Neptune Plus Multi-Collector Inductively Coupled Mass Spectrometer (MC-ICP-MS, Thermo Fisher Scientific®) equipped with high performance extraction cones (Jet sample cone and 'X' skimmer cone) to maximize sensitivity for boron. Samples were injected using a Savillex® 50 μ L min^{-1} C-flow nebulizer and Teflon® Scott type spray chamber. Beams of ^{10}B and ^{11}B were measured on L3 and H3 Faraday cups equipped with amplifiers using $10^{13}\Omega$ resistors (Misra et al., 2014; Lloyd et al., 2018). The instrument was tuned to obtain a stable sensitivity, typically 15-25 mV ppb^{-1} B.

Solutions of 0.2 M HF + 0.2 M HNO_3 were used for rinsing throughout the analytical run between analyses, and as matrix for each sample and standard. The analysis followed the approach of sample-standard bracketing using NIST 951 as reference standard. All samples and quality control standards were analysed in duplicates and thus here average values with $\pm 2\sigma$ standard deviations are reported. Samples with a replicate precision higher than $\pm 0.6\text{ ‰}$ (2σ) were excluded from this study. A coral standard (Chanakya and Misra, 2022) was treated with the complete carbonate cleaning and micro-distillation procedure for each analytical sequence and repeatedly analysed to monitor long term precision ($\delta^{11}B = 24.57 \pm 0.65\text{ ‰ } 2\sigma$, $n = 64$). Additionally, non-micro-distilled AE-121 standard was analysed within each run for quality control ($\delta^{11}B = 19.48 \pm 0.33\text{ ‰ } 2\sigma$, $n = 46$).

In addition to the coral standard, the initial test analysis to validate the boron purification method and instrumental accuracy and precision also included repeated measurements of seawater (Southern Ocean, $\delta^{11}\text{B} = 39.72 \pm 0.25$ ‰ 2σ , $n = 5$) and a boron standard (AE-121) mixed with CaCO_3 (trace metal basis, Acros Organics®) to mimic foraminiferal calcium concentrations ($\delta^{11}\text{B} = 19.53 \pm 0.25$ ‰ 2σ , $n = 18$).

4.6 Estimating past salinity and foraminifera-based temperatures, pH, and $p\text{CO}_2$

Sea surface temperatures (SST) were calculated from foraminiferal Mg/Ca values using the species specific temperature calibration of Mashiotto et al. (1999),

$$\text{Mg/Ca} = 0.47 (\pm 0.03)^{0.107(\pm 0.003) * \text{SST}}, \quad (1)$$

where propagated error was calculated based on 1 standard deviation of the duplicate analysis of Mg/Ca and the uncertainty derived from the calibration equation. Mg/Ca values of planktonic foraminifera are known to be affected by salinity and pH changes as well (Gray et al., 2018; Dueñas-Bohórquez et al., 2009; Gray and Evans, 2019), however, a correction for these effects requires independent estimates for salinity and pH with species-specific calibrations. For calculating past carbon chemistry, salinity is an important parameter and it was estimated based on its conservative relationship with relative sea level change (Waelbroeck et al., 2002). Modern seawater salinity of the BUS (35.43 ± 0.30) was derived from the WOCE Global Data Version 3.0 (Schlitzer, 2000) based on the five closest datapoints to the location of core 64PE450-BC6-PC8. Using these salinity estimates, the effect of salinity on Mg/Ca-based SST was evaluated, and found to have only a small offset in SST values (< 0.4 °C). As the effect of salinity is relatively minor and adding it would also introduce additional uncertainties, we here decided to refrain from correcting for salinity and pH when calculating past temperatures.

The measured $\delta^{11}\text{B}$ values of *G. bulloides* were converted into pH (Hemming and Hanson, 1992) using Eq. (2):

$$\text{pH} = \text{pK}_\text{B}^* - \log(-(\delta^{11}\text{B}_\text{sw} - \delta^{11}\text{B}_\text{borate}) / (\delta^{11}\text{B}_\text{sw} - \alpha * \delta^{11}\text{B}_\text{borate} - \epsilon)), \quad (2)$$

where the equilibrium constant, pK_B^* (Dickson, 1990), was calculated for each sample based on SST derived from the Mg/Ca values of *G. bulloides* and salinity based on sea level. The fractionation factor between $\text{B}(\text{OH})_3$ and $\text{B}(\text{OH})_4^-$, expressed here as α , is 1.0272 ± 0.0006 , from which fractionation, ϵ , is derived as 27.2 ± 0.6 (Klochko et al., 2006). Boron isotopic composition of seawater, $\delta^{11}\text{B}_\text{sw}$ is 39.61 ± 0.2 ‰, based on a large range of temperature, salinity and depth conditions (Foster et al., 2010), and the $\delta^{11}\text{B}$ of borate was calculated from the measured $\delta^{11}\text{B}$ of *G. bulloides* using the species-specific core-top calibration from Raitzsch et al. (2018).

Constraining $p\text{CO}_2$ based on seawater pH requires knowledge of a second independent parameter (Zeebe and Wolf-Gladrow, 2001). For this purpose, total alkalinity was assumed to be the same as today's value at the BUS. Taking the five nearest available datapoint at the core location from the GLODAPv2023 dataset (Lauvset et al., 2024) defines an average value of 2349 ± 11 $\mu\text{mol kg}^{-1}$, which was used to supplement the pH-based $p\text{CO}_2$ reconstruction throughout the 27 kyrs record.

Uncertainty on the reconstructed pH value was determined for each sample through error propagation that considered the above described uncertainties of pK_B^* , α , ϵ , $\delta^{11}\text{B}_\text{sw}$ and the standard deviation (external uncertainty) based on duplicate or triplicate analysis of foraminiferal $\delta^{11}\text{B}$.

Concentrations of CO₂ are based on pH and inorganic carbon chemistry calculations using the PyCO2SYS package (Humphreys et al., 2022) in Python version 3.11.2. Uncertainty is propagated for each computed carbon chemistry parameter as described in Humphreys et al. (2022).

335 4.7 Lipid extraction and alkenone analyses

Lipids were extracted from the freeze-dried and homogenized sediment samples using an accelerated solvent extractor (ASE[®] 350, DIONEX[®]) at the NIOZ. Samples were extracted with dichloromethane (DCM) and methanol (9:1, v/v) at 100 °C to obtain the total lipid content and subsequently dried under N₂ gas at 35 °C in a Caliper TurboVap LV Evaporator. Samples were then redissolved in DCM and run through an Na₂SO₄ column to
 340 eliminate excess water. The extract was passed through an alumina (Al₂O₃) column and separated into apolar, ketone and polar fractions using a mixture of hexane : DCM (9:1, v/v), hexane : DCM (1:1, v/v), and DCM : methanol (1:1, v/v), respectively. All extracts were dried under N₂ and the ketone fraction was further utilized to obtain the relative abundance and δ¹³C values of the long chain alkenones.

345 Ketone fractions were dissolved in ethyl acetate and concentrations of alkenones were measured using a gas chromatograph with flame ionization detection (GC-FID, Agilent[®] 6890N) equipped with silica capillary column (CP-Sil 5 CB; 50 m x 0.32 mm, 0.12 µm film thickness). The temperature program of the GC-FID analyses used an initial temperature of 70 °C that increased with a rate of 20 °C min⁻¹ to 200 °C followed instantly by heating at a rate of 3 °C min⁻¹ until it reached 320 °C where it remained constant for 10 minutes.

350 Based on the initial concentration measurement, samples were diluted with ethyl acetate to allow stable carbon isotope analysis using a gas-chromatography-isotope ratio-mass spectrometer (GC-IRMS, Thermo Fisher Scientific[®] Delta V Advantage Trace[®] 1310). The GC-IRMS was equipped with crossbond trifluoropropylmethyl polysiloxane columns (Rtx-200; 60m x 0.32 mm, 0.5 µm film thickness) and helium as a carrier gas. Each sample
 355 was manually injected on the GC-IRMS. The starting temperature of the GC-IRMS was 70 °C which then increased with 18 °C min⁻¹ until reaching 250 °C. After reaching that temperature heating continued with 1.5 °C min⁻¹ until 320 °C, where it was kept stable for 25 minutes. Samples were analyzed for carbon isotopes in duplicates and instrumental accuracy was monitored through measurement of the of B5 *n*-alkane mixture standard (provided by A. Schimmelmann, Indiana University) every day (i.e. after every 6-7 samples). The isolink II
 360 combustion reactor was oxidized for 10 minutes every day before the start of standard and sample analysis. Each analysis was followed by 2 minutes of seed oxidation to maintain the reactor oxygenated.

4.8 Calculation of alkenone based temperatures and pCO₂

Alkenone-based sea surface temperatures were derived from ketone unsaturation index (U^{K'}₃₇), where U^{K'}₃₇ is defined as the relative abundance of di- and tri-unsaturated C₃₇ methyl alkenones (Prahl and Wakeham, 1987):

$$365 \quad U_{37}^{K'} = C_{37:2} / (C_{37:3} + C_{37:2}). \quad (3)$$

Sea surface temperature was then calculated using the alkenone temperature calibration model developed for the Atlantic region (Conte et al., 2006).

Fractionation of stable carbon isotopes during photosynthesis ($\epsilon_{p37:2}$) can be computed based on the difference between the carbon isotopic ratio of aqueous carbon dioxide ($\delta^{13}\text{C}_{\text{CO}_2}$) and the organic biomass ($\delta^{13}\text{C}_{\text{org}}$):

$$\epsilon_{p37:2} = [(\delta^{13}\text{C}_{\text{CO}_2} + 1000) / (\delta^{13}\text{C}_{\text{org}} + 1000) - 1] \times 1000. \quad (4)$$

$\delta^{13}\text{C}_{\text{CO}_2}$ was derived from the carbon isotopes of planktonic foraminifera, *G. bulloides* corrected for the temperature dependent fractionation during calcite precipitation (Romanek et al., 1992) and the fractionation between dissolved and gaseous carbon dioxide (Mook et al., 1974).

$\delta^{13}\text{C}_{\text{org}}$ was calculated from the carbon isotopes of di-unsaturated alkenones ($\delta^{13}\text{C}_{37:2}$) as

$$\delta^{13}\text{C}_{\text{org}} = [(\delta^{13}\text{C}_{37:2} + 1000) * (\Delta\delta^{13}\text{C}_{\text{org}} + 1)] - 1000, \quad (5)$$

where $\Delta\delta^{13}\text{C}_{\text{org}}$ expresses the carbon isotopic difference between $\text{C}_{37:2}$ and DIC, that has been defined between 3 – 6 ‰ based on culture experiment (Riebesell et al., 2000; Schouten et al., 1998; Van Dongen et al., 2002). We here take the commonly applied value of 4.2 ‰ (Bijl et al., 2010; Pagani et al., 2005; Pagani et al., 2010; Pagani et al., 2011; Seki et al., 2010; Palmer et al., 2010).

Based on $\epsilon_{p37:2}$, aqueous CO_2 ($[\text{CO}_2]_{\text{aq}}$) can be reconstructed as followed (Hayes, 1993; Pagani et al., 2002):

$$[\text{CO}_2]_{\text{aq}} = b / (\epsilon_f - \epsilon_{p37:2}), \quad (6)$$

where ϵ_f stands for the carbon isotopic fractionation associated with carbon fixation estimated as 25 ‰ (e.g., Popp et al., 1998). Parameter b expresses all physiological factors affecting total carbon isotope fractionation that includes cell shape and size, membrane permeability as well as the algae's growth rate (Jasper et al., 1994; Rau et al., 1996; Popp et al., 1998; Conte et al., 1994; Riebesell et al., 2000). Earlier studies using phytane (Bice et al., 2006; Damsté et al., 2008) and alkenone (Witkowski et al., 2018) to reconstruct $p\text{CO}_2$ estimated b for a mean value of 165 - 170 ‰ $\text{kg } \mu\text{M}^{-1}$. As growth rate and thereby nutrient availability have a large influence on the physiological factors and, accordingly, b values are highly correlated to $[\text{PO}_4^{3-}]$ (Bidigare et al., 1997), b can be best described at our core site by estimating past changes in $[\text{PO}_4^{3-}]$ (Pagani et al., 2005). Here, $[\text{PO}_4^{3-}]$ is estimated based on the Ba/Ca ratio of planktonic foraminifera, *G. bulloides* (Lea and Boyle, 1989; Lea and Boyle, 1990b, a; Martin and Lea, 1998; Lea and Boyle, 1991). We therefore constrain past changes in b as (Pagani et al., 2005):

$$b = [118.52 \times (\text{Ba/Ca} \times [\text{PO}_4^{3-}]_{\text{modern}} / \text{Ba/Ca}_{\text{modern}}) + 84.07]. \quad (7)$$

Average modern PO_4^{3-} concentration ($[\text{PO}_4^{3-}]_{\text{modern}}$) in the BUS is 0.63 $\mu\text{mol kg}^{-1}$ (obtained from GLODAPv2023; Lauvset et al., 2024; Olsen et al., 2016; Key et al., 2015) whereas the corresponding modern foraminiferal Ba/Ca value ($\text{Ba/Ca}_{\text{modern}}$) was analyzed here (19.08 $\mu\text{mol mol}^{-1}$). Eq. (7) basically assumes a constant and proportional relation between Ba and $[\text{PO}_4^{3-}]$. This seems reasonable for our purposes as surface water Ba concentration has been shown to be reflected proportionally in foraminiferal Ba/Ca (Lea and Boyle, 1991; Hönisch et al., 2011) and the cold nutrient rich surface waters are generally enriched in dissolved barium (e.g., Davis et al., 2020).

To calculate atmospheric $p\text{CO}_2$ from aqueous concentrations of CO_2 , Henry's law was applied using the temperature and salinity dependent solubility constant, K_0 .

$$p\text{CO}_2 = [\text{CO}_2]_{\text{aq}} / K_0. \quad (8)$$

Uncertainty propagation for the calculated $p\text{CO}_2$ values was based on the errors derived from 1 standard deviation of duplicate analysis of $\delta^{13}\text{C}$ of the extracted ketone fraction, $\delta^{13}\text{C}$ of foraminifera and Ba/Ca values of foraminifera. The largest uncertainty in alkenone based $p\text{CO}_2$ reconstructions originates from the estimated past $[\text{PO}_4^{3-}]$ and to incorporate potential variation in nutrient levels during the deglaciation, an additional uncertainty of $0.2 \mu\text{mol kg}^{-1}$ was assigned to the known modern values of $[\text{PO}_4^{3-}]$. This uncertainty is based on the gradient measured today in upper 50 meters of the water column, which is more than the variability observed in surface water today, but also includes potential changes in the upwelled waters.

5 Results

5.1 Radiocarbon ages

The calibrated mean radiocarbon ages generally increase with depth in both BC and PC cores used here. Sediment core 64PE450-BC6 comprises 40.59 cm, where the core-top sample was dated at 4.863 ka BP, and an age of 9.551 ka BP at 40 cm bsf (Fig. 3 a). This suggests an average sedimentation rate of about 10 cm kyr^{-1} , with somewhat higher values ($> 10 \text{ cm kyr}^{-1}$) at the top 12 cm. Radiocarbon dates at the top of the box core (0-11 cm) indicate reversed ages (4.9 ka at 11 cm and 5.2 ka at 1 cm uncalibrated ages). This interval also corresponds to elevated Ca/Al, Ti/Al, and Si/Al ratios measured through X-Ray Fluorescence (XRF)-core-scanning (Supplementary Fig. S2; using the method described in Weltje and Tjallingii, 2008). The enrichment of elements commonly found in coarse fractions and heavy minerals is likely due to the removal of fine fraction material by winnowing, which may have also contributed to the loss of the last 4.8 kyrs in the sedimentary record. Alternatively, the upper 10 cm bsf have constant ages due to bioturbation. Radiocarbon analyses from sediment core 64PE450-PC8 included 6 samples of the upper 100 cm of sediment collected. The age-depth model for this core suggests 9.994 ka BP years at 5 cm bsf depth (Fig. 3 b) and indicates the presence of a disturbed core-top. Low sedimentation rates (2 cm kyr^{-1}) characterize the top 10 cm bsf of this core which is approximately in line with the sedimentation rate at the deepest parts of box core 64PE450-BC6 ($\sim 6 \text{ cm kyr}^{-1}$). However, average sedimentation rate in 64PE450-PC8 is lower (4 cm kyr^{-1}) compared to the average values observed in the box core, which, in part, might also be due to compaction with increasing depth. The top 60 cm bsf of the core shows a steady increase in sedimentation rate ($2 - 7 \text{ cm kyr}^{-1}$) and therefore the low average values may be attributed to a relatively abrupt decrease in sedimentation rate at 60 cm bsf in the core, which correspond to an age of 23.586 ka BP. Between 60 and 100 cm bsf depth sedimentation rates remain $1 - 2 \text{ cm kyr}^{-1}$. Due to the uncertainty related to sediment deposition between 60 and 100 cm bsf of the piston core, this study focuses only on the 40.59 cm bsf of core 64PE450-BC6 and the upper 65.5 cm bsf of core 64PE450-PC8.

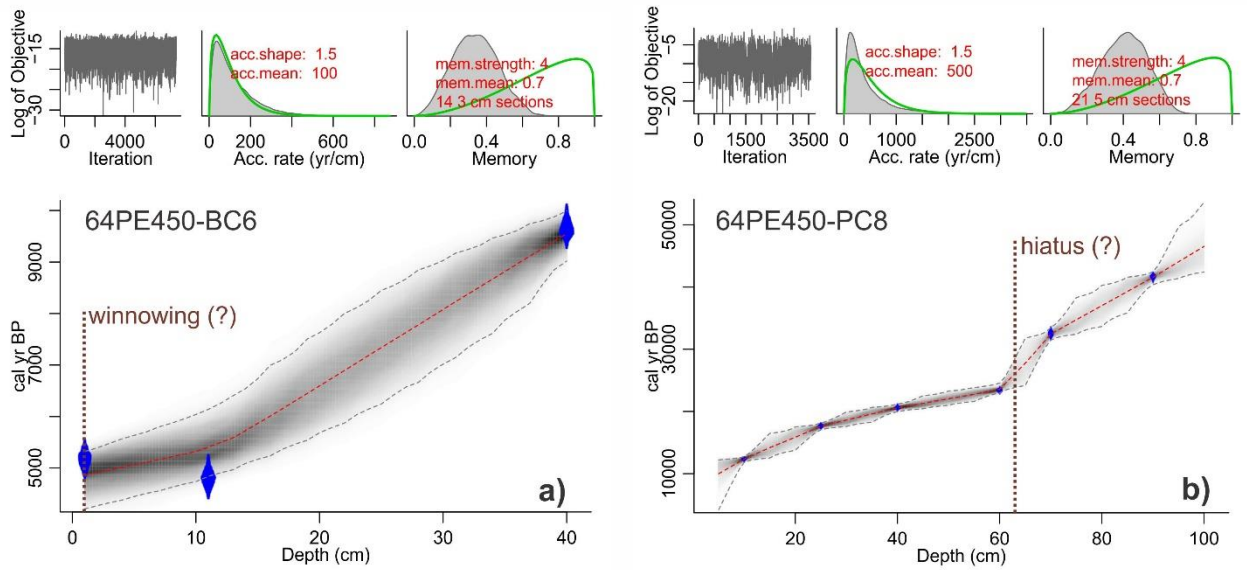


Figure 3: Age-depth model of a) 64PE450-BC6 (box core) and b) 64PE450-PC8 (piston core) based on radiocarbon dates, where blue diamonds indicate the sampling depth for ^{14}C analysis. The calibration of radiocarbon ages and the figure was generated using the Bacon v2.3 package for the R statistical programming software (Blaauw and Christen, 2011). Calibration was performed using the Marine20 calibration curve (Heaton et al., 2020) with a local carbon reservoir correction (ΔR) of 146 ± 85 ^{14}C years (Dewar et al., 2012). Red dashed lines show mean values of the best fitted model and grey dashed lines indicate 95% confidence interval. Note, that due to the disturbed core-top and potential hiatus in the piston core as indicated in panel b), only the interval from 4.25 to 65.5 cm of 64PE450-PC8 was used for temperature and carbon system reconstruction in this study.

5.2 Stable isotopes

Carbon isotope values of the planktonic foraminifer, *G. bulloides*, vary between -1.4 ‰ and 0.6 ‰ (VPDB). The glacial part of the record is marked by relatively high $\delta^{13}\text{C}$ values with a maximum of 0.6 ‰ at 23.3 ka BP. After that, there is a rapid decrease to a value of -1.1 ‰ at 22 ka BP, followed by a period of varying $\delta^{13}\text{C}$ values between -0.6 - 0.1 ‰ until 18 ka BP. Between 18 and 16 ka BP, $\delta^{13}\text{C}$ values decrease to a minimum value of -1.4 ‰ and then with a slight increase values stabilize around -1.1 ‰ until 11 ka BP. During the Holocene, $\delta^{13}\text{C}$ values of *G. bulloides* vary between -1.3 and -0.2 ‰ (Fig. 4 a). The $\delta^{13}\text{C}$ values of the benthic foraminifer, *C. wuellerstorfi*, although measured at somewhat lower resolution, range between 0.5 ‰ and 1.0 ‰. It appears that benthic $\delta^{13}\text{C}$ values were on average 0.2 ‰ heavier during the Holocene compared to the glacial (Fig. 4 a).

The $\delta^{18}\text{O}$ values of *G. bulloides* range from 0.0 ‰ (VPDB) to 3.2 ‰ with the most depleted values at ~8 ka BP (Fig. 4 b). The highest values of $\delta^{18}\text{O}$ can be seen at 23.6 ka BP (peak glacial), after which values show rapid changes and subsequently, continue to decrease gradually before reaching a plateau at about 10 ka BP. During the Holocene these values stayed relatively stable and varied only between 0.0 ‰ and 0.6 ‰ (VPDB). The trend differs from the lower resolution benthic record of $\delta^{18}\text{O}$ measured on *C. wuellerstorfi* (Fig. 4 b). The benthic foraminiferal $\delta^{18}\text{O}$ values are consistently higher compared to the *G. bulloides* values, which is in line with lower

bottom water temperatures. This difference, however, appears smaller during the end glacial than during the Holocene.

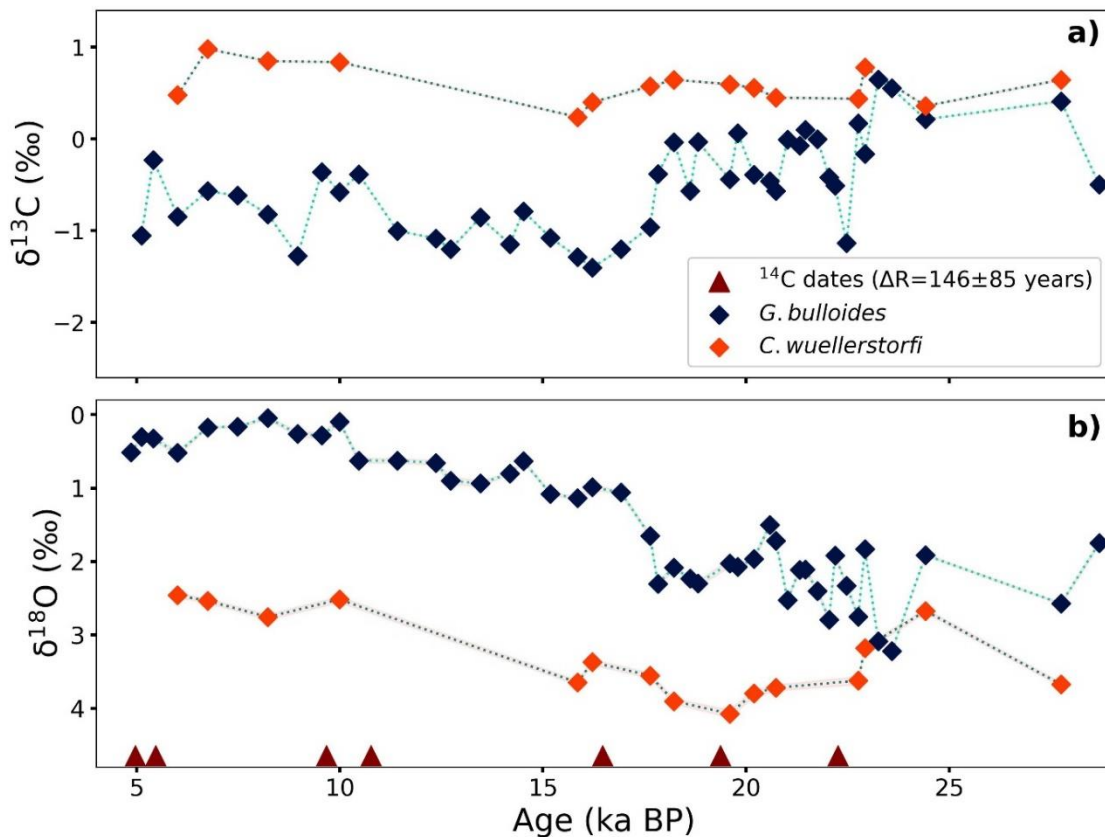


Figure 4: a) $\delta^{13}\text{C}$ and b) $\delta^{18}\text{O}$ values of planktonic (*G. bulloides*) and benthic (*C. wuellerstorfi*) foraminifera plotted with the age model. Red triangles indicate the ages tied with radiocarbon dates.

The boron isotopic composition of the planktonic foraminifer, *G. bulloides*, ranges between 15.1 and 17.0 ‰ (relative to NIST 951) with larger variations during the last 6-5 kyrs (Fig. 5 a). The lowest $\delta^{11}\text{B}$ values were observed at 5.4 ka BP, whereas $\delta^{11}\text{B}$ values reach a maximum at 13.5 ka BP. Prior to this maximum value, $\delta^{11}\text{B}$ values show an increasing trend from 27.8 to 13.5 ka BP (Fig. 5 a).

The carbon isotopic composition of the alkenones shows its heaviest value (-22.4 ‰) at 19.6 ka BP. After this peak, $\delta^{13}\text{C}$ values reach a minimum (-23.4 ‰) at 15.9 ka, then increasing again towards the most recent values (-22.7 to -22.9 ‰; Fig. 5 b).

5.3 Element/Ca ratios in the planktonic foraminifer, *G. bulloides*

Mg/Ca reaches maximum values of 2.85 and 2.81 mmol mol^{-1} at 16.9 and 6.7 ka BP, respectively (Fig. 5 c). Substantially lower values characterise the interval between 16.9 and 6.7 ka BP, when Mg/Ca ranges between 2.46 and 2.69 mmol mol^{-1} . The lowest values (2.30-2.40 mmol mol^{-1}) were found at 4.9-5.4 ka BP and 21.3-27.8 ka BP.

The general trend in foraminiferal Ba/Ca shows an increase from glacial to recent (Fig. 5 d). The oldest part of the record shows relatively stable Ba/Ca values at around 6 $\mu\text{mol mol}^{-1}$. During the last 15 kyrs, however, more variability is observed for Ba/Ca. The highest Ba/Ca value (19.1 $\mu\text{mol mol}^{-1}$) is observed in the top of the record and the lowest values, 4.02 and 3.94 $\mu\text{mol mol}^{-1}$ at 17.6 and 27.8 ka BP, respectively.

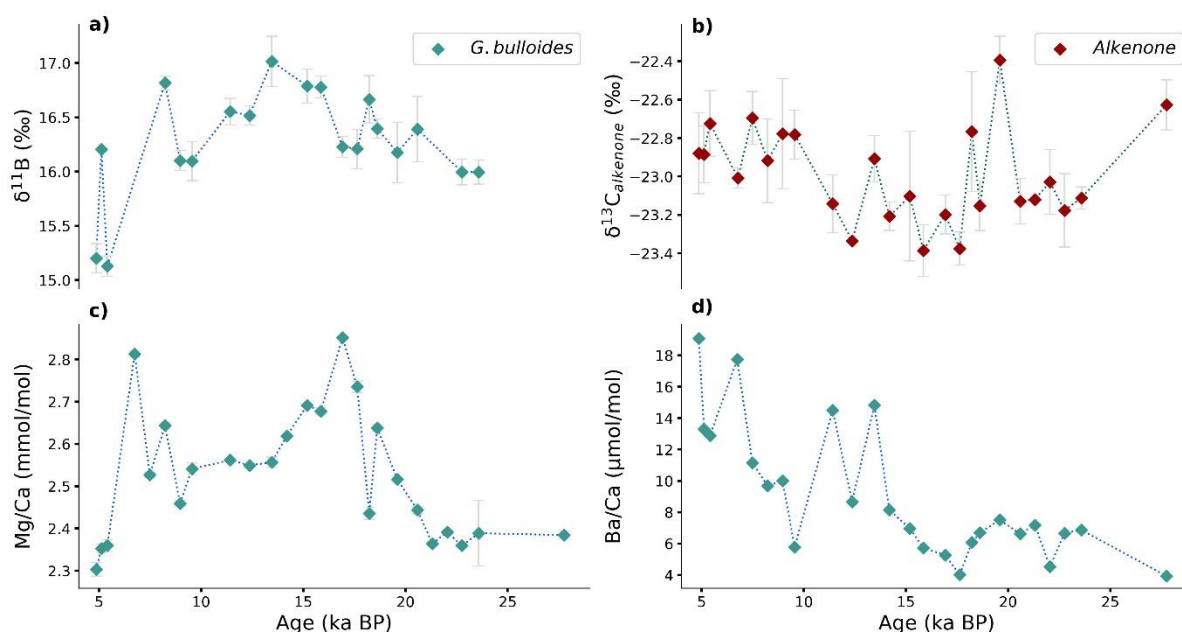


Figure 5: Measured a) foraminiferal $\delta^{11}\text{B}$, b) alkenone $\delta^{13}\text{C}$, and foraminiferal element concentrations: c) Mg/Ca, d) Ba/Ca plotted over the past 27.8 kyrs at the Benguela Upwelling System. Error bars show $\pm 1\sigma$ standard deviation. When error bars are not shown, the error of the duplicate measurement is smaller than the symbol.

5.4 $U_{\mathcal{T}}^K$ sea surface temperatures

The alkenone based $U_{\mathcal{T}}^K$ record shows continues increase throughout the deglaciation (Supplementary Fig. S3). The lowest value (0.56) was measured at 27.8 ka BP which increases to 0.58 until 23.6 ka BP. The interval of 22.0 - 19.6 ka BP records again low $U_{\mathcal{T}}^K$ values, which then follows a steady increase until 9.6 ka BP. During the early Holocene $U_{\mathcal{T}}^K$ values stabilize around 0.7 until 6.8 ka BP when values slightly start to decrease reaching 0.67 in the uppermost part of the record.

6 Discussion

6.1 Local temperatures over the LGM, last deglaciation, and Holocene

The proxy-based temperature reconstructions resemble to the Southern Hemisphere climate responses based on the gradual temperature increase from 23 ka BP onwards (e.g., Petit et al., 1999; Suggate and Almond, 2005; Clark et al., 2009). However, comparing these reconstructions with both Northern (NGRIP; North Greenland Ice Core Project Members, 2004) and Southern Hemisphere (EPICA-Dome C; EDC; Jouzel et al., 2007) records, it is evident that individual climate events show similarities also to the trends observed in the Northern Hemisphere (Fig. 6). This suggests that the location of 64PE450-BC6-PC8 reflects a complex structure of the water column temperature, potentially affected by both Northern and Southern Hemisphere climatic changes.

The U_{37}^K based sea surface temperature reconstruction shows low temperatures (18.2 – 18.5 °C) between 23.6 and 18.6 ka BP (Fig. 6 a) which corresponds to the relatively low temperatures in the record based on the $\delta^{18}O$ values from the shells of *G. bulloides*. These temperature minima and the relatively high variability (1.5 – 3.2 ‰) shown by the $\delta^{18}O$ record between 24 and 21 ka BP indicate the Last Glacial Maximum (LGM; e.g., Clark et al., 2009; Hughes et al., 2013) within this record. While the here observed trends are in agreement with the general deglacial temperature records from higher latitudes, such a pattern is not reflected by the temperature record based on Mg/Ca ratios of *G. bulloides*. In fact, the Mg/Ca-based temperature trend is only between 23 and 16 ka BP comparable to the other local (U_{37}^K , foraminifera- $\delta^{18}O$) and high latitude reconstructions (ice core- $\delta^{18}O$; e.g., Jouzel et al., 2007), after which this trend deviates from the classic deglacial pattern showing decreasing temperatures towards the Holocene.

While all these proxies have their inherent complications, the discrepancies between these records likely also reflect changes in the interaction between the core site and the higher latitudes. For instance, foraminiferal $\delta^{18}O$ values not only depend on temperature, but also on the stable oxygen isotopic signature of the water and hence salinity. This may result in a signal of increasing temperatures in the U_{37}^K and Mg/Ca records, but not in the $\delta^{18}O$ record (e.g., between 20 to 18 ka BP), due to a change in salinity that may compensate the impact of warming on $\delta^{18}O$ values. Shifts in e.g., river runoff, might also have impacted local seawater $\delta^{18}O$ and hence foraminiferal oxygen isotopes. Similarly, multiple studies suggested an overall reduced Agulhas leakage during the LGM compared to deglacial levels (Pether, 1994; Flores et al., 1999; Rau et al., 2002; Peeters et al., 2004; Charles and Morley, 1988; Wefer et al., 1996; Franzese et al., 2006), and changes therein could also have resulted in enhanced $\delta^{18}O$ variability during the LGM.

Not only the trends but also absolute values differ between the temperature reconstructions (i.e., U_{37}^K and Mg/Ca), which is likely reflecting a difference in water depth where the proxy signal carriers lived. Alkenone-based temperature reconstruction agrees with reported modern SSTs from the upper 50 m of the Northern Benguela (Santana-Casiano et al., 2009), while the foraminifer-based temperature (Mg/Ca) corresponds to the values observed somewhat deeper (100-150 m, GLODAPv2023; Lauvset et al., 2024; Supplementary Fig. S4). The vertical dispersion of *G. bulloides* may be large, but the 100-150 m living habitat corresponds well with previously reported living depths for this species (Tapia et al., 2022; Lessa et al., 2020; Rebotim et al., 2017). The offset between U_{37}^K and foraminiferal Mg/Ca values during the LGM is about 3.0 °C, and this gradually increases to 5.6 °C during the early Holocene. This implies that foraminiferal Mg/Ca values are affected by other factors than local SST alone. Partial dissolution of foraminiferal shells at depth may affect Mg/Ca values and thereby bias

545 reconstructed temperatures through the preferential loss of Mg^{2+} (Dekens et al., 2002; Regenberg et al., 2006).
However, the impact of dissolution is probably minor only as the water depth at site 64PE450-PC8-BC6 is less
than 1.4 km and *G. bulloides* is reported to be less sensitive to dissolution compared to other surface dwellers
(Mekik et al., 2007). Whereas Mg/Ca values may also be affected by early diagenesis (Hover et al., 2001; Kozdon
et al., 2013; Ni et al., 2020; Panieri et al., 2017; Sexton et al., 2006; Stainbank et al., 2020), El/Ca ratios in the
550 shell's profile obtained through laser ablation did not show any evidence for such diagenetic effects.

Inconsistent trends and minor changes in absolute temperature values over the last 27 kyrs in this record make it
challenging to identify millennial scale climate events with confidence. The last deglacial was marked by large
climate fluctuations primarily linked to the release of meltwater in the Northern Hemisphere weakening Atlantic
555 Meridional Overturning Circulation (AMOC; Mcmanus et al., 2004; Rahmstorf, 2002; Denton et al., 2010; Hodell
et al., 2017; Pöppelmeier et al., 2023). This, in turn changed global heat distribution causing cooling in the North
Atlantic region that may have been accompanied by warming in the Southern Hemisphere (Broecker, 1998;
Stocker, 1998). Such well-known climate events include the Heinrich Stadial 1 (HS1; ~17.8-14.5 ka BP; e.g.,
Bond et al., 1993; Cacho et al., 1999; Mcmanus et al., 1994; Calvo et al., 2007; Wang et al., 2013) and the Younger
560 Dryas (YD; ~12.8-11.6 ka BP; Rühlemann et al., 1999; Kaplan et al., 2010; Panmei et al., 2017; Blunier and
Brook, 2001; Alley, 2000), which were interrupted by a warming known as the Bølling-Allerød Northern
Hemisphere warming (B-A; ~14.6-12.9 ka BP; e.g., Pedro et al., 2016; Blunier et al., 1997; Lamy et al., 2007;
Vandergoes et al., 2008). While high latitude Southern Hemisphere records present warming signals during HS1
and YD, our temperature records remain relatively constant at the time of these events, and only the alkenone
565 based temperatures indicate slight warming at 12.4-11.4 ka BP. Deviating trends shown by the different proxy
signal carriers are evident at the time of the B-A Northern Hemisphere warming, when Southern Hemisphere
temperature reconstructions are also inconsistent at other locations (Lamy et al., 2007; Vandergoes et al., 2008).

Comparing the trends observed in the different proxies points to the dynamic glacial-deglacial history of the BUS,
570 which was likely shaped by a varying influence of Southern and Northern sourced waters. Also, minor offsets
might be explained by changes in either seasonality between haptophytes and *G. bulloides* (Leduc et al., 2010),
or differences in water depth where the signals were recorded. Hence, such differences between the records could
be due to a shift in productive season and/or a shift in depth habitat. As unravelling these signals is highly
speculative, we here focus on the glacial vs. interglacial contrasts observed.

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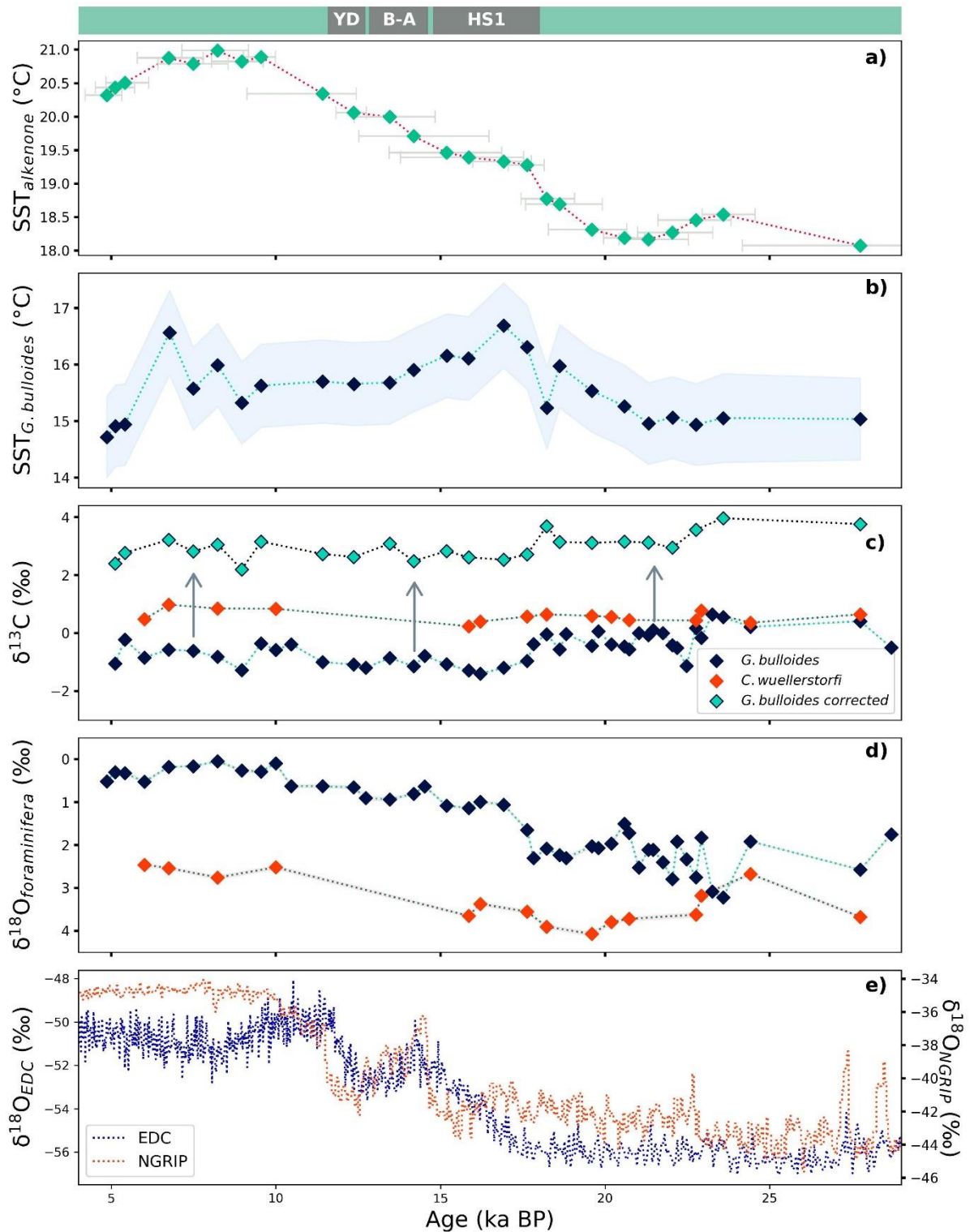


Figure 6: Reconstructed sea surface temperatures (SST) based on a) the alkenone unsaturation index, U_{37}^K , and b) foraminiferal Mg/Ca, c) $\delta^{13}\text{C}$ analysed in benthic (*C. wuellerstorfi*) and planktonic (*G. bulloides*) foraminifera with corrected values, d) $\delta^{18}\text{O}$ of benthic (*C. wuellerstorfi*) and planktonic (*G. bulloides*) foraminifera, and e) $\delta^{18}\text{O}$ ice core record from EPICA-Dome C (EDC; Jouzel et al., 2007) and North Greenland Ice Core Project (NGRIP; North Greenland Ice Core Project Members, 2004) shown for the past 29 kyrs. Corrected $\delta^{13}\text{C}$ values of *G. bulloides* marked with green diamonds in panel c) are based on temperature (derived from Mg/Ca ratios; Bemis et al., 2000) and $[\text{CO}_3^{2-}]$ values (derived from pH and TA; Bijma et al., 1999), and arrows indicate the direction of the correction. Modern day SST at core site

64PE450-BC6-PC8 is approximately 20.7 °C (GLODAPv2023; Lauvset et al., 2024; Santana-Casiano et al., 2009). Major climate events (YD: Younger Dryas; B-A: Bølling-Allerød interval; HS1: Heinrich Stadial 1) are marked above the top panel as reference. Horizontal error bars in panel a) show the age uncertainty based on 95 % confidence interval of the calibrated age. Blue shaded area in panel b) indicates the error propagated from temperature calibration uncertainty and $\pm 1\sigma$ standard deviation of the duplicate measurement of the samples. Analysis of the stable isotopes (panel c and d) provided an error smaller than the symbols shown on the figure.

6.2 Biological carbon pump

Comparing benthic (*C. wuellerstorfi*) and planktonic (*G. bulloides*) trends in $\delta^{13}\text{C}$ shows they have similar values during the last glacial and higher $\delta^{13}\text{C}$ values for the benthic than for the planktonic foraminifera during the interglacial. This is in contrast to what would be expected if BCP determined the foraminiferal carbon isotope signatures (Hain et al., 2014; Hilting et al., 2008). Generally, DIC in surface waters is enriched in ^{13}C as the BCP results in preferential export of ^{12}C rich organic matter to the deeper water masses, where it is released through remineralization. The efficiency and strength of the BCP is known to be affected by multiple processes, such as the formation, sinking, and interaction of aggregates with other minerals (Fowler and Knauer, 1986; Alldredge and Silver, 1988; Armstrong et al., 2001; Francois et al., 2002; Klaas and Archer, 2002; De La Rocha and Passow, 2007; Turner, 2015), and the efficiency is generally reflected by the offset in ^{13}C between the surface and deep water ($\Delta\delta^{13}\text{C}$). However, species specific offsets from equilibrium values between seawater DIC $\delta^{13}\text{C}$ and foraminiferal carbonate $\delta^{13}\text{C}$ can challenge interpretation of the $\Delta\delta^{13}\text{C}$ and are likely responsible for the here observed lower $\delta^{13}\text{C}$ values of *G. bulloides* compared to *C. wuellerstorfi*.

Application of foraminiferal $\Delta\delta^{13}\text{C}$ as their proxy for BCP efficiency requires a direct relation with the $\delta^{13}\text{C}$ of DIC. Benthic foraminiferal $\delta^{13}\text{C}$ values, and in particular $\delta^{13}\text{C}$ values of the epifaunal *C. wuellerstorfi*, are generally considered faithful recorders of the $\delta^{13}\text{C}$ values of DIC, with carbonate $\delta^{13}\text{C}$ values being close to equilibrium (Thomas and Shackleton, 1996; Hilting et al., 2008). The stable carbon isotopic composition of DIC today is approximately 0.5 – 0.7 ‰ at a depth of 1.3 km along the latitude 20° S (Kroopnick, 1980; Kroopnick, 1985; Sarnthein et al., 1994; Curry and Oppo, 2005; Schmittner et al., 2013) which agrees well with the value inferred from *C. wuellerstorfi* in the uppermost sample (6.0 ka BP) of core 64PE450-BC6-PC8.

Stable carbon isotopic values from planktonic foraminifera have been shown to be generally lower with respect to the equilibrium values of DIC (Kahn, 1979; Kahn and Williams, 1981; Oppo and Fairbanks, 1989; Spero, 1992), indicating a strong biological impact (i.e. the vital effect; e.g., Spero, 1992; De Nooijer et al., 2014; Erez, 2003). For instance, symbiont-bearing species such as *O. universa* and *T. sacculifer* show offsets in $\delta^{13}\text{C}$ as much as 1.8 and 1.4 ‰ respectively, depending on irradiance level (Spero, 1992; Spero and Lea, 1993). Although *G. bulloides* lacks algal symbionts (Hemleben et al., 1989) it has been shown to deviate even more from the ambient seawater $\delta^{13}\text{C}$ values (Kahn and Williams, 1981; Spero and Lea, 1996). Several factors likely add together to the observed offset, such as carbon chemistry ($[\text{CO}_3^{2-}]$; Spero et al., 1997; Bijma et al., 1999), temperature (Bemis et al., 2000), and respiration (Zeebe et al., 1999). The range of $[\text{CO}_3^{2-}]$ and temperature observed within our 27 kys record may yield a $\delta^{13}\text{C}$ -offset of 0.6-1.4 ‰ and 2.4-2.6 ‰, respectively. More recently, Bird et al. (2017)

suggested that also bacterial symbiosis may partly explain the observed offset for $\delta^{13}\text{C}$ in *G. bulloides*. While symbiont photosynthesis contributes to elevating foraminiferal $\delta^{13}\text{C}$ due to preferential ^{12}C removal, geochemical signature of *G. bulloides* is more likely to be controlled by the respiration of photoautotrophic cyanobacteria that produces depleted CO_2 and hence, decreases shell $\delta^{13}\text{C}$ values (Bird et al., 2017). Irrespective of the process involved, a substantial correction has to be applied to the $\delta^{13}\text{C}$ values of *G. bulloides* to approach the heavier seawater DIC $\delta^{13}\text{C}$ values. Increasing temperature (Bemis et al., 2000) and $[\text{CO}_3^{2-}]$ (Bijma et al., 1999) will also increase the offset between $\delta^{13}\text{C}$ values of *G. bulloides* and seawater [DIC] suggesting that larger corrections are required during the Holocene than during the last glacial. Still, when the corrections for changes in temperature and $[\text{CO}_3^{2-}]$ are applied individually or combined, trends remain the same, showing the highest $\delta^{13}\text{C}$ values of planktonic foraminifera during the LGM (Supplementary Fig. S5). Hence, despite the uncertainties in interpreting the absolute planktonic $\delta^{13}\text{C}$ values, the trend in $\Delta\delta^{13}\text{C}$ should still provide a measure for changes in the efficiency of the BCP. Therefore, we here applied a combined correction for both temperature (Bemis et al., 2000) and $[\text{CO}_3^{2-}]$ (Bijma et al., 1999), derived from foraminiferal Mg/Ca and $\delta^{11}\text{B}$, respectively (Fig. 6 c). The offset of the $\delta^{13}\text{C}$ value of the core-top sample with the modern $\delta^{13}\text{C}$ values of the DIC is approximately 2.4 ‰ (Kroopnick, 1985), which agrees with the applied corrections based on temperature and $[\text{CO}_3^{2-}]$ in the most recent samples (2.4 – 2.8 ‰; 4.9-5.4 ka BP).

Offsets between the $\delta^{13}\text{C}$ of the planktonic and benthic foraminifera reflect differences in the BCP, but potentially also changes in the dominant water mass at the cores' locations. Intermediate depths of the South Atlantic are dominated today by the Antarctic Intermediate Water and this likely remained the major water mass over the last glacial cycle (Pahnke et al., 2008; Howe et al., 2016; Gu et al., 2017). However, it is unclear whether the depth range of the AAIW increased (Muratli et al., 2010) or decreased (Ronge et al., 2015; Li et al., 2021) during the LGM compared to the present day. In the western Atlantic, $\delta^{13}\text{C}$ values of benthic foraminifera suggest persistence of AAIW water masses at the depth of our core site (e.g., Curry and Oppo, 2005). As our values correspond to those found in the western Atlantic (Curry and Oppo, 2005; Lacerra et al., 2019; Umling et al., 2019), a sustained influence of southern water masses is likely, with the $\delta^{13}\text{C}$ value of DIC in the AAIW during the LGM remaining relatively similar to the present day.

Variations in the stable carbon isotopic composition of surface seawater DIC are attributed to the changes in biological activity and air-sea exchange (Lynch-Stieglitz et al., 1995). While enhanced biological activity will result in an increase of $\delta^{13}\text{C}$ values of DIC, more intense air-sea exchange will contribute to a decrease. In upwelling regions, the upwelled light carbon may still result in a net decrease in $\delta^{13}\text{C}$ values despite the enhanced biological activity. The analysed $\delta^{13}\text{C}$ values of benthic foraminifera from the LGM in this study show on average values that are 0.2 ‰ lower than those during the Holocene, with a minimum-glacial to a maximum-Holocene range of 0.4 – 1 ‰. This increasing trend of mid-depth, benthic $\delta^{13}\text{C}$ values from glacial to interglacial agrees with the trends reported from the Brazilian Margin, where potentially both air-sea CO_2 exchange (Umling et al., 2019) and remineralization (Lacerra et al., 2019) affect the $\delta^{13}\text{C}$ signal. Estimates of $[\text{PO}_4^{3-}]$ based on foraminiferal Ba/Ca show a general increase from glacial to interglacial with relatively low values between 18 to 15 ka BP corresponding to the observed planktonic $\delta^{13}\text{C}$ minimum (Supplementary Fig. S6). Although we cannot rule out

a contribution of air-sea gas exchange, especially on the time scale of short-term climatic changes, we believe that the observed glacial-interglacial trend was primarily driven by change in the BCP efficiency.

Summarizing the observed impacts, assuming a more or less stable presence of AAIW at intermediate depth and minimal impact of change in air-sea CO₂ exchange on $\delta^{13}\text{C}$ values, after correcting for offsets, a larger difference between planktonic and benthic foraminiferal $\delta^{13}\text{C}$ values during the LGM compared to the Holocene is evident (Fig. 6 c; and Supplementary Fig. S5), suggesting a more efficient BCP.

6.3 $p\text{CO}_2$ record of the BUS over the last 27 kyrs

Although $\delta^{11}\text{B}$ in foraminifera shells and $\delta^{13}\text{C}$ of alkenones are the most commonly applied methods to reconstruct $p\text{CO}_2$, these proxies are only very rarely compared in the same record. Since these proxies are recording different components of the speciation of carbon in seawater and have different biases, they do not necessarily have to show similar results. Here, we observe only a modest change in pH ($8.08\text{--}8.23 \pm 0.07$, derived from foraminiferal $\delta^{11}\text{B}$) during the last deglaciation, whereas $p\text{CO}_2$ values show a change from 180 to 280 ppm (± 42 ppm, derived from $\delta^{13}\text{C}$ of alkenones; Fig. 7).

Minor variability in pH was reported previously by Raitzsch et al. (2018) for the Walvis Ridge for the same time interval, although reconstructed pH values were slightly higher ($0.10\text{--}0.14$ pH units). Although only minor, the offset is in line with the core studied here being closer to the upwelling area, as the major upwelling area extends only about 200 km out of the coast today (Lutjeharms and Meeuwis, 1987; Lutjeharms and Stockton, 1987). With lowest pH values in the core of the upwelling area and values increasing towards the open ocean, the trend in the offset between the two areas is minor but in the right direction.

Using pH and total alkalinity, $p\text{CO}_2$ can be calculated, suggesting higher $p\text{CO}_2$ compared to the known atmospheric values over the past 27 kyrs (Fig. 7; Petit et al., 1999). Calculated $p\text{CO}_2$ based on the foraminiferal $\delta^{11}\text{B}$ only match seawater equilibrium values at 13.5 and 8.2 ka BP. Whereas the Bølling-Allerød event marks an AMOC amplification, AMOC was reduced around 8.2 ka BP due to the meltwater input in the North Atlantic (Matero et al., 2017; Barber et al., 1999; Pedro et al., 2016; Blunier et al., 1997). However, the calculated $p\text{CO}_2$ values are associated with considerable uncertainty for a large part related to total alkalinity being ill-constrained. Using estimates of total alkalinity based on relative sea level change is debated (e.g., De La Vega et al., 2023) as this does not account for all changes in alkalinity on glacial-interglacial timescales. Because total alkalinity calculated based on the relative sea level change during the last deglaciation results in only a small (less than 10 ppm) offset, we here used constant alkalinity ($2349 \pm 11 \mu\text{mol kg}^{-1}$, GLODAPv2023; Lauvset et al., 2024) in combination with boron isotope-based pH to determine $p\text{CO}_2$. The trends observed here are not affected by the alkalinity values used.

The seawater $p\text{CO}_2$ reconstruction based on the $\delta^{13}\text{C}$ of alkenones follows past atmospheric $p\text{CO}_2$ known from the Vostok ice core record well (Petit et al., 1999). This suggests that over the interval studied here, the BUS remained more or less in equilibrium with the atmosphere with regard to CO₂ and did not act as an appreciable source or sink. An offset is observed (about 65 ppm) during the Holocene, between 11 and 7 ka BP, when

alkenone-based reconstruction suggests somewhat lower $p\text{CO}_2$ values compared to the ice core record, although this difference falls well within the uncertainty of the proxy. This could indicate a temporary transition of the area to a CO_2 sink as the seawater becomes undersaturated with respect to CO_2 .

The good agreement between alkenone-based and ice core $p\text{CO}_2$ records suggests that previously recognized complications in applying this proxy may not be relevant in the BUS region. Adaptation of CCM by the alkenone producers can hamper the use of the proxy under low $p\text{CO}_2$ conditions (Badger, 2021). However, the mechanisms that may control CCM in the alkenone producers are not fully constrained (e.g., Reinfelder, 2011), and effectiveness of the CCM may differ between species (e.g., Goudet et al., 2020; Heurreux et al., 2017). As the reconstruction of $p\text{CO}_2$ values based on alkenone $\delta^{13}\text{C}$ provided a reasonable record here and in another upwelling region (Palmer et al., 2010), application of this proxy in upwelling sites may be able to rely on the classical concept of passive diffusion of CO_2 (Bidigare et al., 1997; Laws et al., 1995). Additional uncertainty in the alkenone-based $p\text{CO}_2$ reconstruction may derive from the estimation of the b factor. Often modern, constant $[\text{PO}_4^{3-}]$ is assumed to estimate the b factor for reconstructing $p\text{CO}_2$ (Pagani et al., 1999; Zhang et al., 2013; Pagani et al., 2005; Witkowski et al., 2020), or assuming that the membrane permeability has not changed significantly, one can correct for the growth rate effects of the alkenone producers (Zhang et al., 2019; Zhang et al., 2020). Here, we used foraminiferal Ba/Ca, which is suggested to reflect nutrient ($[\text{PO}_4^{3-}]$) variations but does not vary with temperature, salinity or carbon chemistry parameters (Lea and Spero, 1994; Hönisch et al., 2011) unlike other suggested nutrient proxies such as Cd/Ca (Oppo and Rosenthal, 1994; Allen et al., 2016) and Zn/Ca (Van Dijk et al., 2017). Using the Ba/Ca approach to constrain the b factor yields here 10 to 105 ppm lower $p\text{CO}_2$ values compared to the approach of using a constant $[\text{PO}_4^{3-}]$ based on modern value over the last 27 kyrs (Supplementary Fig. S7). These approaches give nonnegligible difference in our conclusions, as the results based on the use of constant $[\text{PO}_4^{3-}]$ imply constant net CO_2 outgassing in the BUS over the last glacial and deglacial. However, variability in surface $p\text{CO}_2$ remains minor, causing only 32 ppm change in $p\text{CO}_2$ in the BUS over 27 kyrs, which is unlikely considering the highly dynamic properties of the region on glacial-interglacial timescale (Mollenhauer et al., 2002; McKay et al., 2016; Romero et al., 2003). Also, previous studies pointed out an overestimation of surface $p\text{CO}_2$ when constant nutrient levels are applied to constrain the b factor (Zhang et al., 2019 and references therein), which is in line with our observation when the two approaches (Ba/Ca and constant- $[\text{PO}_4^{3-}]$) are compared. While the Ba/Ca method still awaits refinements in future studies, this approach may provide an efficient way to address some of the uncertainties originating from local conditions, that are not targeted when constant $[\text{PO}_4^{3-}]$ values are assumed.

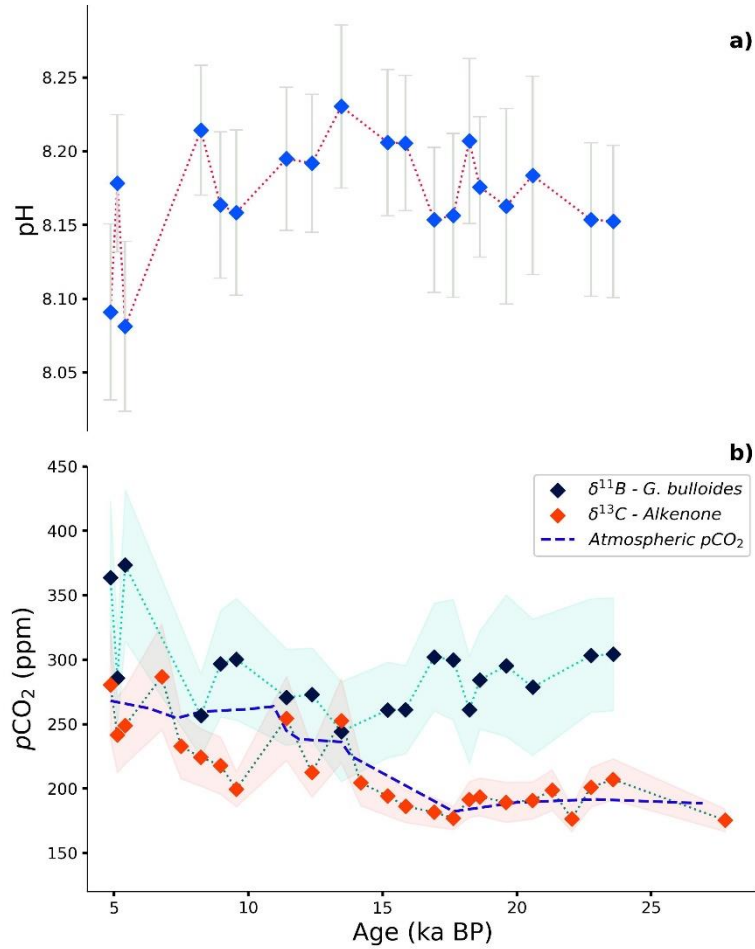


Figure 7: Reconstruction of a) pH based on $\delta^{11}\text{B}$ of *G. bulloides*, and b) $p\text{CO}_2$ based on $\delta^{11}\text{B}$ of *G. bulloides* combined with a constant total alkalinity value of $2349 \pm 11.07 \mu\text{mol kg}^{-1}$ (dark blue diamonds) and $\delta^{13}\text{C}$ of alkenones with Ba/Ca-based $[\text{PO}_4^{3-}]$ estimate (red diamonds). Modern day $p\text{CO}_2$ value of the AAIW is approximately 326 ppm (Lauvset et al., 2024; Salt et al., 2015). Blue dashed line shows the Vostok ice core record of $p\text{CO}_2$ (Petit et al., 1999). Light green and red shaded area represent propagated error for the foraminifera and alkenone based reconstructions, respectively. See further details on uncertainty propagation in the text.

6.4 Change in the efficiency of BCP and CO_2 disequilibrium

Most obvious from comparing the alkenone and foraminifera-based $p\text{CO}_2$ reconstruction is the difference in amplitude of change on a glacial-interglacial time scale. Whereas the alkenone-based reconstruction closely mimics atmospheric changes, the foraminifera-based reconstruction shows a constant $p\text{CO}_2$. This results in an interglacial difference in $p\text{CO}_2$ ($\Delta p\text{CO}_2 = p\text{CO}_{2(\text{foraminifera})} - p\text{CO}_{2(\text{alkenone})}$) of about 64 ± 20 ppm between the alkenone and foraminifera-based reconstructions, while during glacial times $\Delta p\text{CO}_2$ increases to approximately 94 ± 20 ppm (Fig. 8). Because the *G. bulloides* are proliferating during the upwelling season, they likely primarily reflect the deeper upwelled water (i.e., subsurface) compared to the alkenones which are synthesized e.g., by the surface-dwelling coccolithophorids. Note that *G. bulloides* may migrate between approximately 50 and 400 m (Rebotim et al., 2017), which can affect the calculated $p\text{CO}_2$ gradients. Still, *G. bulloides* represents a larger

average depth than the alkenone-based record, and hence, carbon system conditions that are closer to those of the upwelled intermediate waters, i.e., the AAIW in the BUS.

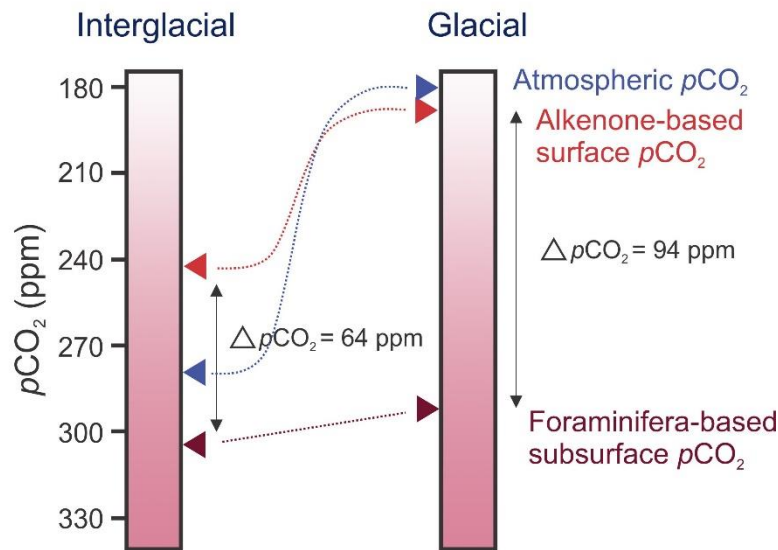


Figure 8: Schematic comparison of interglacial and glacial $p\text{CO}_2$ values. Red arrows mark average interglacial (left) and glacial (right) values calculated from the alkenone- and planktonic foraminifera- (pH and total alkalinity) based proxies in this study.

Studies using foraminifera-based proxies suggested more intense upwelling during glacial times (Oberhänsli, 1991; Little et al., 1997), but at the same time radiolarian-based upwelling proxies indicate reduced upwelling (Des Combes and Abelman, 2007). Due to its location and the influence of water masses both from the north and the south, cells of the BUS are characterized by different environmental conditions (e.g., temperature and nutrients; Emeis et al., 2018). During the LGM, cold source waters likely impacted the northern cells of the BUS more than its central and southern parts (Des Combes and Abelman, 2007) affirming complexity of this upwelling system. While we may conclude that upwelling intensities were different from one cell to another, potentially also impacted by the offshore transition of the modern strong upwelling cells (e.g., Mollenhauer et al., 2002), increased cold water input does not necessarily correlate with stronger upwelling (Des Combes and Abelman, 2007), potentially explaining conflicting interpretations based on different proxies.

Atmospheric $p\text{CO}_2$ was significantly reduced during the LGM, hence the presence of an increased amount of CO_2 at subsurface depths implies either enhanced upwelling or that the upwelled waters were richer in CO_2 or both. Lower atmospheric CO_2 during the glacial is likely explained by multiple processes. The larger extent of sea ice over the glacial Southern Ocean prevented CO_2 escaping from seawater in an area today acting as a major CO_2 exchange region (Stephens and Keeling, 2000), whereas enhanced iron fertilization likely contributed to more efficient utilization and transport of carbon and nutrients to the deep (Martin, 1990; Martínez-García et al., 2014). Aeolian transport and dissolution in the shelf regions might have provided important sources of iron at that time (Martin, 1990; Tian et al., 2023), which would locally influence air-sea carbon balance, still with minimum impact on global atmospheric $p\text{CO}_2$ due to adjacent regions where excess carbon can be utilized. Also locally at the BUS,

aeolian transport presumably increased due to the intensified trade winds (Stuut et al., 2002), although a more humid climate may (Stuut et al., 2002; Cockcroft et al., 1987) or may not (Shi et al., 1998; Partridge et al., 1999) have prevailed in southwest Africa during the LGM. Therefore it remains speculative whether stronger winds may have provided sufficient iron for phytoplankton growth locally, or potentially excess iron input in the sub-Antarctic region provided a source for additional mid-depth CO₂ storage (Martínez-García et al., 2014). Still, a more efficient biological carbon pump, as indicated by the offset between the planktonic and benthic foraminiferal carbon isotope records (Fig. 6 c), suggests that an increased supply of carbon in the upwelling areas from intermediate depths to the surface, may have been effectively counterbalanced.

Based on comparing *p*CO₂ proxies, with *G. bulloides* recording primarily the upwelled waters and alkenones the surface waters, we see evidence for enhanced storage of carbon at depth during the glacial. The resulting mid-depth high CO₂ waters provide also at that time the source for upwelled waters in the BUS, which could have resulted in the local release of (part of the) stored CO₂ if not prevented by an efficient biological carbon pump. An increased biological pump acted as an effective cap on the stored carbon and hence contributed to preventing the release of mid-depth CO₂ during the glacial.

7 Conclusions

Carbon system proxies were applied to demonstrate changes in inorganic carbon chemistry of the Northern Benguela Upwelling System over the last 27 kyrs. Temperature reconstructions based on both organic and inorganic proxies indicate that the BUS may be associated with climatic changes observed both in the Northern and Southern Hemisphere. While surface values of *p*CO₂ reconstructed from $\delta^{13}\text{C}$ of alkenones generally track atmospheric *p*CO₂, the foraminifera-based reconstruction suggests minor variation in *p*CO₂ in the subsurface since the Last Glacial Maximum until present. Hence, the increased gradient of *p*CO₂ between the surface waters and depth observed for the last glacial period provides evidence for enhanced storage of carbon in the Antarctic Intermediate Waters. Outgassing of CO₂, however, could be effectively prevented by the biological carbon pump as also indicated by the offset in the $\delta^{13}\text{C}$ of planktonic and benthic foraminifera.

Data availability

All data used in this study can be obtained from the NIOZ Data Archive System at <https://doi.org/10.25850/nioz/7b.b.lh> (Karancz et al., 2024): Table 1. Uncalibrated and calibrated radiocarbon ages; Table 2. Single foraminifera analysis (LA-Q-ICP-MS); Table 3. $\delta^{18}\text{O}$ and $\delta^{13}\text{C}$ of planktonic and benthic foraminifera, El/Ca and $\delta^{11}\text{B}$ of planktonic foraminifera, $\text{U}_{\text{37}}^{\text{K}}$ and $\delta^{13}\text{C}$ of alkenones; Table 4. XRF and lightness reflectance data.

Author contributions

SK, LJdN, SS, GJR designed the study. RH and ZE collected the sample material and SK prepared and processed the samples. SK, BwdW, MTJvdM, SM were responsible for the analysis of foraminifera and alkenones. JL and

NH conducted the radiocarbon analysis. SK interpreted and visualized the data under the supervision of LJdN, SS, and GJR, and drafted the paper with contribution from all co-authors.

Competing interests

The authors declare that they have no conflict of interest.

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