Rare earth element resources on Fuerteventura, Canary 1 Islands, Spain: a geochemical and mineralogical approach 2

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18 Abstract. Rare earth elements (REEs) play a pivotal role in the ongoing energy and mobility transition 19 challenges. Given their critical importance, governments worldwide and especially from the European 20 Union, are actively promoting the exploration of REE resources. In this context, alkaline magmatic rocks 21 (including trachytes, phonolites, syenites, melteigites and ijolites), carbonatites and their associated 22 23 24 weathering products were subjected to a preliminary evaluation as potential targets for REE exploration on Fuerteventura Island (Canary Archipelago, Spain) based on mineralogical and geochemical studies. These lithologies show significant REE concentrations. However, only carbonatites exhibit the potential to host 25 26 27 28 29 economically viable REE mineral deposits. REE concentrations in carbonatites of about 10,300 ppm REY (REEs plus yttrium) have been detected, comparable to other locations hosting significant deposits of these critical elements worldwide. Conversely, alkaline magmatic rocks and the resulting weathering products display limited REE contents. Notably, REEs in carbonatites are associated with primary accessory phases such as REE-bearing pyrochlore and britholite, and secondary monazite. The carbonatites of Fuerteventura 30 hold promise as prospective REE deposits within a non-conventional geological setting (oceanic island). 31 However, due to intricate structural attributes, the irregular distribution of these mineralizations and 32 possible land use and environmental constraints, additional future detailed investigations are imperative to 33 ascertain their viability as substantial REE resources. 34

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Keywords. Rare earth elements, Carbonatites, Fuerteventura, Canary Islands, Weathering

37 1 Introduction

38 The implementation of actions to mitigate climate change is one of the major global challenges facing 39 society today. In this regard, the European Commission (EC) has adopted a series of economic and 40 technological proposals aimed at reducing greenhouse gas emissions and transforming Europe into the first 41 climate-neutral continent by 2050. These guiding principles, encompassed in the European Green Deal 42 (EGD) (European Commission, 2019), are directly linked to the establishment of an energy and mobility 43 transition based on green technologies that will replace the current fossil fuel-based model. However, in 44 order to achieve the ambitious targets, other finite resources will be essential: metals. Some of them, 45 commonly referred to as green metals, will play a crucial role in a successful energy and mobility transition 46 and, consequently, in achieving the neutrality goals outlined by the EGD (Graedel et al., 2015; Jyothi et al., 47 2020). Among the green metals is the group of rare earth elements (REEs) that are critical for a wide range 48 of high-tech applications such as wind turbines, electric vehicles, rechargeable batteries, energy-efficient 49 lighting, optical telecommunications, photovoltaic cells, solar energy harvesting or artificial photosynthesis 50 for "green hydrogen" (H2) generation (Méndez-Ramos et al., 2013; Acosta-Mora et al., 2018; Wondraczeck 51 et al, 2015). Such elements are also commonly known as the vitamins of modern industry (Alonso et al., 52 2012; Chakhmouradian and Wall, 2012; Massari and Ruberti, 2013; Charalampides et al., 2015; Weng et 53 al., 2015; Balaram, 2019).

According to the International Union of Pure and Applied Chemistry (IUPAC), REEs comprise a group of 17 chemical elements: scandium (Sc), yttrium (Y) and the 15 members of the lanthanide series (Connelly et al., 2005). The term "rare" is confusing because, even though REEs seldom occur in pure mineral phases, their average concentration in the Earth's crust is around 125 ppm, surpassing other common industrial metals such as copper, gold or platinum (Long et al., 2010; Rudnick and Gao, 2014).

Given their pivotal role in modern industry and green technologies, as well as the projected increase in demand for REEs in the coming years, governments worldwide are actively promoting the exploration of new REE resources (Barteková and Kemp, 2016). In line with this, the EC included light rare earth elements (LREEs) and heavy rare earth elements (HREEs) in the 2023 list of critical raw materials (CRMs), acknowledging them as essential and considering HREEs as the material with the highest supply risk (European Commission, 2023a). On March 16, 2023, the EC presented a bold initiative to the European Parliament: The European Critical
Raw Materials Act. This regulation aims to establish a comprehensive framework to ensure a secure and

67 sustainable supply of CRMs, including REEs, in the coming years (European Commission, 2023b).

68 The search for REEs in the geological environment has primarily centred on investigating non-conventional

HREE sources such as soils and weathering products (Braun et al., 1993; Wang et al., 2010, Wang et al.,

70 2013; Berger et al., 2014; Aiglsperger et al., 2016; Torró et al., 2017; Reinhardt et al., 2018; Borst et al.,

71 2020), but also traditional and well-known LREE-bearing lithologies, such as carbonatites (Goodenough et

72 al., 2016; Yang et al., 2019; Pirajno and Yu, 2022).

73 Carbonatites are igneous rocks formed by carbonate mantle melts and are genetically associated with a 74 wide range of mafic, ultramafic, and alkaline silicate rocks (Yaxley et al., 2002). Although carbonates such 75 as calcite or dolomite are their main forming minerals, a significant portion of carbonatites contain 76 accessory phases enriched in critical metals such as REEs (Christy et al., 2021). REEs can be contained in 77 fluorcarbonates (e.g., bastnäsite, parisite, huanghoite, synchysite), phosphates (e.g., monazite, 78 rhabdophane), silicates (e.g., allanite), or even oxides (e.g., REE-bearing pyrochlore, cerianite). These 79 accessory minerals make carbonatites the main current REE source, representing 86.5% of the deposits 80 under exploitation for these elements (Liu et al., 2023). However, although carbonatites are rare rocks, 81 predominantly found in continental rifts associated with cratons (Humphreys-Williams et al., 2021), they 82 have exceptionally been described in other geological contexts, most notably oceanic islands associated 83 with hotspots, such as Cape Verde (Mourão et al., 2010; De Ignacio et al., 2018;) or Fuerteventura in the 84 Canary Islands (Mangas et al., 1996; Carnevale et al., 2021).

85 The petrogenesis of carbonatites is still a debated topic (Anenburg et al., 2021; Yaxley et al., 2022). 86 Different processes have been proposed for their formation, although there is a consensus that they originate 87 from primary fusion processes derived from a carbonated mantle (Kamenetsky et al., 2021). For the specific 88 case of the oceanic carbonatites, this debate is even more lively. Doucelance et al. (2010) suggested a 89 shallow origin from low-degree partial melting at the base of the oceanic lithosphere. Other authors have 90 proposed the involvement of unmixing process linking to alkaline magma suites (Weidendorfer et al., 91 2016), the action of hydrothermal fluids of marine origin enriched in Ca that would have serpentinized the 92 mantle (Park and Rye, 2013) or even the contribution of recycled marine carbonates through subduction or 93 assimilated in shallow magma chambers (Démeny et al., 1998; Hoernle et al., 2002; Doucelance et al., 94 2014).

95 The present work concentrates on the mineralogical and geochemical study of carbonatites and associated 96 alkaline igneous rocks, along with their weathering products, in three different sectors in the western region 97 of Fuerteventura (Canary Islands, Spain; Figure 1). The primary goal of this research is to conduct an initial 98 evaluation of these materials, which are genetically associated with a volcanic island linked to an oceanic 99 intraplate magmatism. This assessment aims to enhance our understanding of REE accumulation while 100 appraising the potential of this peculiar geological environment as a non-conventional source of REE.

101

102 2 Geological setting

103 2.1 The Canary Island Seamount Province

104 The Canary Islands archipelago, located between 27°N and 30°N of latitude, is part of the Canary Island 105 Seamount Province (CISP). This volcanic region forms a band of approximately 1300 km in length and 106 350 km in width, running parallel to the African continental margin. Within the CISP, there are over 100 107 seamounts and up to 8 emerged islands: El Hierro, La Palma, La Gomera, Tenerife, Gran Canaria, 108 Lanzarote, Fuerteventura and Savage islands (Courtillot et al., 2003; Schmincke and Sumita, 2010; van den 109 Bogaard, 2013). Based on magnetic anomaly measurements and dating of both emerged and submarine 110 igneous materials, volcanic activity in the CISP spans more than 142 Ma, from the Early Cretaceous to the 111 present day (Frisch, 2012; van den Bogaard, 2013; Longpré and Felpeto, 2021).

112 2.2 Fuerteventura Island

Fuerteventura, the easternmost island of the Canarian archipelago, along with Lanzarote, forms the emergent crest of the Eastern Canarian Volcanic Ridge, which is located approximately 100 km offshore from the Moroccan coast (Figure 1). Fuerteventura is the oldest island in the archipelago, with its initial stages of formation linked with submarine volcanic activity, dating to the Oligocene (~34 Ma). The first episodes of subaerial volcanism occurred around ~23 Ma (Coello, 1992; Ancochea et al., 1996; Pérez-Torrado et al., 2023).

- 119 Fuerteventura is characterized by the occurrence of three distinct main geological units, arranged in order 120 from oldest to youngest: the Fuerteventura basal complex (FBC), the Miocene subaerial volcanic units, and
- 121 the Pliocene-Quaternary volcano-sedimentary facies (Fúster et al., 1968; Le Bas et al., 1986; Muñoz et al.,
- 122 2005; Gutiérrez et al., 2006; Troll and Carracedo, 2016).
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124 **2.2.1** The Fuerteventura basal complex

125 The FBC unit mainly outcrops in the western part of the island (Figure 1). Two different groups of 126 lithofacies may be distinguished: (1) Early Jurassic to Late Cretaceous oceanic crust materials (Steiner et 127 al., 1998), constituted by mid-ocean ridge basalts and oceanic sediments; (2) Oligocene submarine and 128 transitional volcanic rocks associated with plutonic bodies and dyke swarms (Feraud et al., 1985; Hobson 129 et al., 1998; Gutiérrez et al., 2006). In this second group, a set of lithologies can be distinguished related to 130 an ultra-alkaline-carbonatitic magmatic pulse that occurred ~25 Ma (Le Bas, 1981; Barrera et al., 1986; 131 Balogh et al., 1999). Additionally, alkaline ultramafic, mafic and felsic plutonic rocks such as wehrlites, 132 pyroxenites, gabbros and syenites intruded the previously existing Oligocene materials, forming distinctive 133 ring complexes (Muñoz et al., 2005). These magmatic rocks, predominantly of Oligocene age, have been 134 interpreted as episodes of submarine and transitional growth in Fuerteventura (Le Bas et al., 1986; Gutiérrez 135 et al., 2006).

136 In general, outcrops related with the FBC intrusive assemblage exhibit significant variations and four

- 137 distinct morphologies and characteristic textures can be identified (Fúster et al., 1968; Barrera et al., 1986;
- 138 Le Bas et al., 1986; Fernández et al., 1997; Mangas et al., 1992, 1994, 1997; Ahijado 1999; Ancochea et
- 139 al., 2004; Ahijado et al., 2005; Muñoz et al., 2005):

- 140 (1) Basaltic, alkaline and carbonatitic dykes and veins of meter-scale, decimeter-scale, and
 141 centimeter-scale, that are randomly distributed, resulting in a chaotic arrangement (Figure 2a, b).
 142 Related to the carbonatite veins and dikes, an intense fenitization may occur.
- (2) Shear zones (Fernández et al., 1997), characterized by gradual or diffuse boundaries, which
 display assimilation structures between different rock bodies, along with the presence of
 mylonites, and brecciated textures resulting from deformation (Figure 2c).
- 146 (3) Pegmatitic textures developed within certain rock bodies, often containing centimeter-sized147 crystals of rock-forming minerals (Figure 2d).
- (4) Contact metamorphism and metasomatism, as well as skarn zones that occur in deformed or
 undeformed carbonatites, influenced by subsequent hydrothermal fluid circulation (Ahijado et al.,
 2005; Casillas et al., 2008, 2011).
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152 In addition, during Miocene magmatic pulses, alkaline plutons were formed in the central-western part of 153 Fuerteventura Island north of the locality of Pájara (sector 2, Figure 1). These intrusions constitute typical 154 ring complexes of alkaline magmatic rocks, including nepheline syenites, syenites, and trachytes (Muñoz, 155 1969). They are regarded as the most recent rocks in the FBC (Figure 1) and have been dated using the K-156 Ar method, yielding an approximate age of 20.6 ± 1.7 Ma (Le Bas et al., 1986; Holloway and Bussy, 2008).

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158 2.2.2 Miocene subaerial volcanic unit

159 During the Miocene, Fuerteventura witnessed the formation of up to three volcanic edifices (Figure 1; 160 Coello et al., 1992; Ancochea et al., 1996). The northern volcanic structure, referred to as the Tetir edifice, 161 experienced two volcanic construction phases between 22 and 12.8 Ma (Balcells et al., 1994). These 162 episodes involved the eruption of basalts, picritic basalts, oceanic basalts, trachybasalts and trachytes. In 163 the central part of the island, the Gran Tarajal edifice developed three different construction phases 164 spanning from 22.5 to 14.5 Ma (Balcells et al., 1994). On the Jandía Peninsula, in the southern part of the 165 island, a volcanic edifice comprising both basaltic and trachybasaltic materials emerged. It formed three 166 successive construction episodes occurring between 20.7 and 14.2 Ma ago (Balcells et al., 1994). Based on 167 their mineralogical and petrological features, the lithologies comprising this unit have not been considered 168 as potentially containing significant concentrations of REEs. Therefore, they have not been included in the 169 evaluation conducted in the present study.

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171 2.2.3 Pliocene and Quaternary volcano-sedimentary facies

172 After the subaerial volcanic activity during the Miocene, a period of volcanic quiescence ensued, leading 173 to the erosion of the previously formed volcanic edifices. Subsequently, during the Pliocene (between 5.3 174 and 2.6 Ma), a phase of magmatic rejuvenation began, characterized by scattered Strombolian eruptions 175 (Figure 1). Concurrently, various sedimentary formations emerged across the entire island, including littoral 176 and shallow-water marine deposits, as well as aeolian, colluvial, and alluvial subaerial sediments and 177 paleosols from the Pliocene to the Quaternary (Fúster et al., 1968; Zazo et al., 2002; Ancochea et al., 2004). 178 The soils on Fuerteventura are predominantly classified as eutric cambisols and lithosols-vitric andosols, 179 according to the FAO/UNESCO (1970) nomenclature. However, the current arid and deforested conditions 180 have led to extensive erosion of the weathered rock profiles present in different areas of the island. Edaphic 181 calcretes are abundant in Fuerteventura (Alonso-Zarza and Silva, 2002; Huerta et al., 2015), with their 182 primary source of calcium believed to be the Pliocene paleodunes formed by calcarenites, rather than the 183 parent igneous rock itself (Chiquet et al., 1999; Huerta et al., 2015; Alonso-Zarza et al., 2020). Interestingly, 184 the aeolian dust deposits predominantly originate from the Sahara Desert (Goudie and Middleton, 2001; 185 Menéndez et al., 2007; Scheuvens et al., 2013). 186 187 188 189 190 **3** Materials and Methods 191 **3.1 Sampling** 192 Alkaline magmatic rocks and especially carbonatites are considered potential targets for the exploration of rare earth elements (Goodenough et al., 2016; Balaram et al., 2019; Anenburg et al., 2021; Beland and 193 194 Jones, 2021). In Fuerteventura, these types of lithologies are found in two distinct geological areas: the 195 Oligocene (sectors 1 and 3; Figure 1) and the Miocene lithologies related with the FBC (sector 2, Figure 196 1). 197 Considering that weathering profiles may concentrate REE in larger quantities than primary bedrocks (Bao

and Zhao, 2008; Menéndez et al., 2019, Braga and Biondi, 2023; Chandler et al., 2024), these lithological

199 formations were included in the present evaluation study and sampling was conducted on a selection of six

different profiles: (1) Agua Salada ravine (sector 1) and (2) Aulagar ravine (sector 3), developed on
carbonatites, (3) the FV-30 road, (4) Las Peñitas quarry, (5) Palomares ravine and (6) the Pájara profiles,

202 on syenite bedrock (Figure 1; Table S1).

Accordingly, a systematic sampling campaign was conducted in three different sectors of Fuerteventura, targeting alkaline and carbonatitic igneous rocks and their associated weathering products. The specific locations of these predetermined sectors are outlined in Figure 1. As a result, a set of 29 representative samples of potentially REE-enriched magmatic rocks, along with 21 samples of associated weathering products, were collected for further analysis and evaluation (Table S1). For the weathering products, we conducted six sampling profiles (labelled A to F; Figure 1) at various suitable points to compare the mineralogical and geochemical changes resulting from weathering of the primary magmatic rocks.

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211 **3.2 Petrographic and mineralogical studies**

212 Selected samples of magmatic rocks were prepared in thin sections for textural and mineralogical analysis 213 at the Laboratory of Geological and Paleontological Preparation of the Natural Sciences Museum of 214 Barcelona (LPGiP-MCNB; Barcelona, Spain). A representative subset of these samples was also examined 215 using a JEOL JSM-7100 field emission scanning electron microscope (FE-SEM) at the Scientific and 216 Technological Centers of the Universitat de Barcelona (CCiTUB). The FE-SEM system is equipped with 217 an INCA Pentaflex EDS (energy dispersive spectroscopy) detector (Oxford Instruments, England), which 218 allowed for the acquisition of semi-quantitative analyses of mineral phases. The general operating 219 conditions for the FE-SEM were a 15-20 kV accelerating voltage and a 5 nA beam current.

220 To achieve accurate and precise mineralogical identification and characterization of the weathering 221 magmatic rocks and calcretes, X-ray powder diffraction (XRPD) measurements were performed using a 222 PANalytical Empyrean powder diffractometer equipped with a PIXcel1D Medipix 3 detector at the 223 Integrated XRD Service of the General Research Support Service of La Laguna University, Spain. The 224 diffractometer employed incident Cu K_{α} radiation at 45 kV and 40 mA, along with an RTMS (real-time 225 multiple strip) PIXcel1D detector with an amplitude of 3.3473° 20. The diffraction patterns were obtained 226 by scanning random powders in the 2θ range from 5° to 80°. Data sets were generated using a scan time of 227 57 seconds and a step size of 0.0263° (20), with a $1/16^{\circ}$ divergence slit. Mineral identification and semi-228 quantitative results were obtained using the PANalytical's HighScore Plus search-match software (v. 4.5) 229 with a PDF+ database.

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231 **3.3 Geochemical analyses** 232 The major elements composition of carbonates from carbonatites was studied using an electron probe 233 microanalyzer (EPMA) system. The EPMA analyses were conducted on a JEOL JXA-8230 electron 234 microprobe, equipped with five wavelength-dispersive spectrometers and a silicon-drift detector EDS, 235 located at the CCiTUB. The spot mode was employed for the analyses and the electron column was set to 236 an accelerating voltage of 15 kV and a beam current of 10 nA. Standard counting times of 10 seconds were 237 used, along with a focused beam, to achieve the highest possible lateral resolution. The analytical standards 238 employed during the analysis process were: celestine (PETJ, Sr K_a) wollastonite (PETL, Ca K_a), periclase 239 (TAPH, Mg K_a), hematite (LiFH, Fe K_a), rhodonite (LiFH, Mn K_a) and albite (TAPH, Na K_a). 240 Bulk-rock geochemical data of major and trace element composition were obtained by X-ray fluorescence 241 (XRF) and inductive coupled plasma (ICP)-emission spectrometry. The samples were prepared by lithium 242 metaborate/tetraborate fusion and nitric acid digestion at the ACTLABS Activation Laboratories Ltd. 243 (Ancaster, Canada). 244 245 **4 Results** 246 4.1 Petrography and mineralogy 247 4.1.1 Alkaline magmatic rocks and carbonatites 248 The primary lithologies under study, consist of Oligocene (~25 Ma) alkaline igneous and carbonatitic rocks, 249 as well as Miocene alkaline lithologies (K-Ar age of 20.6±1.7 Ma; Le Bas et al., 1986), that form part of 250 the FBC. Their outcrops extend across kilometer-scale areas but exhibit high heterogeneity at a detailed 251 level due to the occurrence of numerous small intrusions, ranging in size from metric to decimetric 252 dimensions (Figs. 2a, b). 253 At a mineralogical level, separation of the different types of alkaline rocks found in the FBC is complex 254 because these lithologies are intimately associated and infiltrate diffusely, leading to the formation of hybrid 255 intrusions. The materials with the most mafic composition correspond to pyroxenites and melteigites, and 256 their formation is associated with the earliest magmatic fractions. However, these are commonly spatially 257 associated with more differentiated rocks, mainly ijolites, nepheline syenites, and syenites. All these 258 lithologies have a relatively simple mineralogy, characterized by varying proportions of nepheline (10-30%)

259 modal) and potassium feldspar (50-80% modal), associated with aegirine-augite and biotite (10-30%

260 modal). A set of accessory minerals with varying proportions (always less than 5% modal) also occur,
261 including ilmenite, titanite, zircon, and fluorapatite.

At a textural level, the alkaline series lithologies of the FBC present granular textures with millimeter-sized euhedral grains. However, in some of the intrusions in sectors 2 and 3, pegmatitic syenites-ijolites were detected with centimeter-sized grains characterized by the presence of large aegirine-augite crystals.

265 Some of the intrusions described in the three sectors show aphanitic textures caused by faster cooling,

266 resulting in rocks with similar mineralogy but extrusive-type textural characteristics. Therefore, due to their

textural features, some dikes and apophyses, although mineralogically equivalent, should be classified astrachytes and phonolites.

Carbonatitic intrusions commonly co-occur with the alkaline rocks, sharing similar morphology, textures, and spatial distribution within the outcrops (Figure 2e). Furthermore, alkaline and/or carbonatitic intrusions can be occasionally associated with mafic intrusions, primarily pyroxenites and alkaline gabbros. In addition, a subsequent set of mafic dikes with basaltic composition overlaps the previous intrusive bodies (Figs. 2a, b).

All carbonatites described in different outcrops from sectors 1 and 3 are predominantly composed of calcite (95% modal) and can thus be classified as calciocarbonatites (Le Maitre, 2005). None of the studied samples shows the occurrence of ferromagnesian carbonates such as ankerite, dolomite, and/or siderite, as well as REE carbonates. Texturally, calcite occurs as euhedral crystals ranging in size from millimetres to centimetres, often recrystallized and exhibiting polysynthetic twinning. In some cases, a secondary micritic calcite matrix is present, filling interstitial spaces and fractures.

The major element composition of calcite is relatively consistent across all the carbonatite samples.
Notably, there are significant contents of SrO, with values of up to 5.43 wt%, while REEs are absent from
the carbonate composition (Table S2).

The accessory mineralogy (~5% modal) comprises disseminated phases within the calcium carbonate. Among them, the occurrence of minerals from the spinel group, including magnetite (Figure 3a), and primarily jacobsite, occurring as subhedral crystals of up to 50 μ m (Figure 3b). Another characteristic mineral is perovskite, occurring as subhedral crystals of up to 100 μ m. These grains are remarkable for their significant Nb contents, as described in other carbonatitic localities worldwide (Torró et al., 2012). Britholite also occurs as subhedral crystals of up to 100 μ m (Figure 3b). This primary britholite contains significant LREE content (Figure 4), and its alteration leads to the formation of secondary REE-enriched phosphates, mainly monazite-Nd (Figure 3c), which also contains substantial amounts of La and Ce (Figure 4). REEs, in addition to occurring in primary britholite and secondary monazite, were also detected in tiny pyrochlore grains, heterogeneously disseminated in the calcite groundmass (Figure 3b). In some cases, pyrochlore forms euhedral crystals of up to 20 µm, also included in calcite (Figure 3d). This pyrochlore shows slight zoning towards plumbopyrochlore (Christy and Atencio, 2013), with significant enrichment in Pb observed at grain borders (Figure 3d).

296 Carbonatites can be affected by certain contact metamorphism, especially in sectors 1 and 3 (Figure 1) and 297 may exhibit a slightly different mineralogy from the one described thus far. This is characterized by the 298 occurrence of skarn-type metamorphic minerals, formed due to the interaction between carbonatites and 299 spatially associated silica-rich rocks. Among these minerals, there are subhedral crystals of andradite, up 300 to 30 µm in size, implanted in a matrix of secondary calcite and phlogopite, exhibiting pronounced zoning 301 with kerimasite cores (Figure 3e). In these areas, the occurrence of REE mineralizations associated with 302 allanite (Figure 3f) is also typical. Allanite occurs as granular aggregates associated with hydrothermal 303 secondary sulfates, primarily baryte (Figure 3f), but occasionally celestine (Figure 3c).

304 This particular mineralogy, typically associated with skarn formations, emerges from the interaction

305 between a carbonatite intrusion and surrounding silicate rocks, in contrast to the typical process. It has

306 recently gained attention from several researchers in various carbonatite locations worldwide, who have

- 307 coined the term antiskarn to describe it (Anenburg and Mavrogenes, 2018; Yaxley et al., 2022).
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309 4.1.2 Weathering products

In certain areas within the three studied sectors (Figure 1), there is evidence of the development of characteristic shallow geological formations consistently associated with weathering, which affect the outlined magmatic lithologies (Figs. 5, 6). These geological products were studied through the analysis of six alteration profiles, developed on carbonatites (Agua Salada and Aulagar) and syenites (Palomares ravine, FV-30 road, Las Peñitas quarry, and Pájara) (Figure 1). The carbonatite-calcrete sections generally consist of centimetre-scale calcrete veins injected into the

bedrock, seemingly without any apparent connection to the current upward lithosol (Figure 5). In general,

317 the development of soils or weathering products was not detected on carbonatites in any of the studied

318 sectors of the FBC.

Weathering products developed on syenite bedrock are generally more abundant, and the corresponding alteration profiles are better preserved than in carbonatites. The cambic B horizon displays reddish to yellowish colorations (5YR6/6), with a thickness of up to 20-30 cm. Additionally, it is common to find BC

322 horizons instead of B horizons, while the C horizon is well-developed, reaching a 30-40 cm thickness at

323 certain levels of the profile (Figure 6). Furthermore, except for the Las Peñitas profile (E profile, Figure 1),

324 centimetre-scale calcrete bands (Bk; Jahn et al., 2006) were also detected in deeper layers across all the

325 studied profiles.

In terms of mineralogical composition, carbonatite profiles exhibit significant changes due to weathering. In general, weathering processes lead to a reduction in calcite, the disappearance of fluorapatite, and the formation of secondary minerals like palygorskite (Figure 7). The contribution from lateral slope movement is also evident through the presence of residual plagioclase and clinopyroxene.

330 In the case of syenite weathering profiles, illite/chlorite and kaolinite are the predominant secondary 331 products, followed by muscovite and palygorskite (Figure 7). Other minerals such as quartz were also 332 detected, even in the C horizons.

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334 4.2 Bulk-rock and mineral geochemistry

Chemical analysis of the major, minor and trace elements were carried out in order to evaluate the geochemical features and the distribution of REEs, on 25 representative samples of igneous rocks from the FBC, including trachytes, phonolites, syenites, ijolites and carbonatites (Table S3). In addition, we also analysed 21 samples of weathering products (Table S4).

339 The total REY (REEs plus yttrium) content in the FBC igneous rocks exhibits widespread and significant 340 enrichment in comparison to the average crustal values (~125 ppm, Rudnick and Gao, 2014). Notably, the 341 extrusive and magmatic alkaline lithologies (trachytes and phonolites as well as syenites and ijolites) show 342 variable REY values ranging between about 230 and 1,400 ppm (Table S3). In contrast, the carbonatitic 343 rocks exhibit REY content more than ten times greater than the alkaline lithologies, with specific samples 344 reaching maximum values of up to about 10,300 ppm, as evidenced in sample 85a sourced from a 345 carbonatite outcrop in sector 1 (Table S3). 346 The weathered magmatic rocks, though moderately significant in REY content relative to the average

- 347 crustal values (Table S4), still exhibit slightly lower levels compared to the content observed in the
- 348 associated alkaline and carbonatitic protoliths (Table S3). A contrasting pattern emerges in the calcretes,

349 where REY values experience a sharp reduction, presenting virtually negligible values ranging between 20

and 72 ppm REY. These levels are significantly below the average Earth's crust values (Rudnick and Gao,

351 2014) and are markedly lower than those observed in both the alkaline lithologies and, particularly, the352 carbonatites of the FBC.

REE normalized diagrams further underscore this distribution, portraying elevated content in the carbonatites, followed by the alkaline rocks (Figure 8a). Meanwhile, the weathered magmatic rocks and calcretes (Figure 8b) display significantly lower values. All studied lithologies exhibit clear negative patterns, indicative of enrichment in LREEs relative to HREEs. Notably, carbonatites and alkaline rocks (Figure 8a) exhibit a flattening of these negative patterns in the final segment, indicating a certain degree of HREE enrichment.

359 The FCB carbonatites exhibit a depletion in some critical elements commonly associated with this lithology 360 such as Nb or Ta (Table S3). Negative anomalies of both Nb and Ta are clearly observed in the multi-361 element diagrams of carbonatite samples (Figure 9a). However, given the presence of pyrochlore in the 362 carbonatites, these anomalies in Nb and Ta are likely not indicative. We interpret that the low concentrations 363 of these elements could be attributed to an analytical artifact that would underestimate the contents of High 364 Field Strength Elements (HFSE) due to the challenge of pyrochlore dissolution in the analytical digestion 365 protocols employed. These protocols have been primarily devised to assess the contents of REEs rather 366 than HFSE. Additionally, alkaline rock patterns also show a distinctive negative anomaly in Sr (Figure 9b). 367 As for the weathering products, their contents of other minor elements do not indicate significant 368 concentrations of metals or critical elements like Nb or Ta (Table S4). The multi-element diagrams for the 369 calcretes exhibit a negative Ta anomaly (Figure 9c), while the patterns of weathered magmatic rocks do not 370 reveal notable anomalies in any group of elements (Figure 9d).

A specific geochemical study of REE distribution in the six studied weathering profiles was also conducted
(Figure 10). The main objective was to evaluate the geochemical interactions between the protolith and the
related weathering lithologies, with the aim of detecting potential REE enrichments or depletions caused
by weathering processes.

In the exchange patterns of calcretes spatially associated with carbonatitic protoliths, as analyzed in the Agua Salada and Aulagar ravine profiles (Figs. 10A, B), REE concentrations are two orders of magnitude lower than in the carbonatite (Figure 10A), as also previously determined from the REE diagrams (Figure 8). Notably, it was found that the REE concentration is directly proportional to the distance from the 379 protolith (Figure 10B), and calcrete samples with the highest REE concentrations (sample 14; Figure 10) 380 were found in closer proximity to the primary carbonatites than more REE depleted samples (samples 15 381 and 18; Figure 10B). In addition, although the values of all elements are depleted in the calcrete patterns, 382 there is a greater depression in LREE than in HREE relative to the protolith, resulting in typically positive 383 patterns, except for sample 76 from the Agua Salada ravine, where a clear inverse trend is observed (Figure 384 10A).

385 In general, the diagrams in Figure 10 show that weathering products on syenites exhibit enrichment relative 386 to the protolith (green areas in Figs. 10C, D, E). However, calcrete samples, whether derived from 387 carbonatites or syenites, consistently show depletions compared to the protolith contents (reddish areas in 388 Figs. 10C, D, F). The diagrams corresponding to the weathering products generated on syenites exhibit 389 similar morphologies (Figs. 10C, D, E, F). Overall, these lithologies are characterized by enrichment in 390 REEs relative to the protolith as well as V-shaped patterns, featured by the presence of a negative anomaly 391 in Eu, which is also reported in all C and B horizons developed on syenites, except sample 61 (Figure 10C), 392 and is likely related to plagioclase crystallization.

393

394 **5 Discussion**

395 5.1 REE evaluation of the FBC magmatic rocks

396 The FBC magmatic rocks, in the three study sectors, encompass alkaline lithologies (trachytes, phonolites, 397 syenites, melteigites, and ijolites) as well as carbonatites. Regarding the group of alkaline rocks, the 398 detected REE content varies between 214 and 1,330 ppm (Table S3), significantly higher than the average 399 concentration determined in the Earth's crust (~125 ppm, Rudnick and Gao, 2014). However, this finding 400 is not surprising, and the observed values in Fuerteventura are not anomalous, as these types of lithologies 401 typically exhibit REE concentrations within this range (Dostal, 2017). Therefore, the measured REE 402 concentrations are neither significant nor sufficiently elevated to hypothetically consider these lithologies 403 as a potential non-conventional deposit of these critical elements in the FBC.

404 On the other hand, FBC carbonatites present significantly higher values in terms of REE content. In the 405 studied carbonatite samples from sectors 1 and 3 (carbonatites do not outcrop in sector 2), REE content 406 ranges between about 1300 ppm and 10,300 ppm. The latter value corresponds to the richest REE-detected 407 sample in the entire FBC, which is located in the Agua Salada ravine area of sector 1 (Table S3; Figure 1).

408 The reported REE content values in the FBC carbonatites are similar to the general average concentrations 409 found in other locations worldwide where carbonatites are exploited for REE extraction. This is the case, 410 for example, of Bayan Obo, the largest REE mine in the world in terms of reserves and production (Lai et 411 al., 2015; Liu et al., 2018). In this locality, high-grade carbonatites exhibit average concentrations of 2880 412 ppm (Wu et al., 2008; Smith et al., 2015, 2016), which are equivalent to those measured in some of the 413 samples from Fuerteventura. It should be noted that low-grade carbonatite ore from Bayan Obo presents 414 extremely high values in comparison to the FBC, with REE concentrations reaching 30,750 ppm (Chao et 415 al., 1997; Smith et al., 2016).

Another significant example is the Mountain Pass carbonatite in California, USA, regarded as the largest
REE mine in the American continent, intermittently in operation for REE extraction since 1954 (Olson et
al., 1954; Haxel, 2005). In this REE deposit, average value across the whole complex are around 2580 ppm
(Castor et al., 2008; Mariano and Mariano, 2012; Smith et al., 2016), also in line with REE concentrations
detected in the present study for the FBC carbonatites.

421 This comparative analysis can also be carried out using normalized REE values (Figure 11). In this regard, 422 FBC carbonatites are significantly depleted in LREE compared to those from Bayan Obo (Yang et al., 423 2019) and Mountain Pass (Castor et al., 2008), although they show similar values to other REE deposits 424 associated with carbonatites, such as those in Ashram, Canada (Beland and Jones, 2021) and Bear Lodge, 425 USA (Moore et al., 2015; Smith et al., 2016; Figure 11). However, the pattern of the Fuerteventura 426 carbonatites exhibits a slightly less pronounced slope, indicating a higher relative content of HREE, which 427 are considered the materials with the highest risk of supply among all the CRMs defined by the EC 428 (European Commission, 2023a). In fact, in the FBC carbonatites, the normalized HREE values are 429 equivalent to those reported in the primary carbonatitic rocks from the deposits of Bayan Obo (China) and 430 Mountain Pass (USA) (Figure 11). The relative significant HREE content reported in FBC carbonatites 431 holds particular significance for several economic and technological reasons. The use of HREEs, such as 432 Yb, Er, and Tm, is of particular interest in cutting-edge photonic and nanotechnology applications. 433 At the mineralogical level, it was observed that, in the FBC carbonatites, the main REE-hosting minerals

434 are accessory phases; primarily minerals from the pyrochlore group, found as disseminated euhedral micro-

435 crystals implanted in primary calcite (Figs. 3b, d). Another REE-bearing mineral in the FBC carbonatites

436 is britholite, which exhibits significant LREE content. However, this mineral is commonly altered to

437 monazite (Figs. 3c, 4), interpreted as a secondary phase but also a carrier of these critical elements (Chen438 et al., 2017).

Another noteworthy aspect is the lack of REE **[luorcarbonates**] like bastnäsite REE(CO₃)F, parisite Ca(REE)₂(CO₃)₃F₂, synchysite Ca(REE)(CO₃)₂F or huanghoite Ba(REE)(CO₃)₂F. They do not occur in the FBC, as they do in other REE deposits associated with, for example, the Bayan Obo carbonatite or the Sulphide Queen carbonatite from Mountain Pass (Castor et al., 2008; Smith et al., 2015, 2016). This point is crucial for a future hypothetical evaluation of the FBC carbonatites, as the processing of oxides and phosphates for REE extraction is a much more complex and expensive treatment process than for REEbearing carbonates (McNulty et al., 2022).

446

447 **5.2 REE evaluation of associated weathering products**

448 The weathering materials developed on magmatic rocks, also analysed for their REE concentrations, 449 constitute the remnants of soils that were interpreted as developed under wetter conditions during a humid 450 phase of the oxygen isotope stage 2, spanning from 29 to 20 thousand years BP (Huerta et al., 2016). This 451 period aligns with the last glacial maximum, marked by heightened humidity in the Canary Islands, 452 resulting in slope erosion and the formation of talus flatiron (Gutiérrez-Elorza et al., 2013). Over time, these 453 materials have undergone substantial volume reduction due to human-driven deforestation and erosion, 454 primarily before the 15th century (Machado-Yanes, 1996). Notably, topography plays an essential role in 455 the distribution of these weathering profiles and influences specific physical attributes such as slope 456 (FAO/UNESCO, 1974).

457 The studied weathering products developed on syenite rocks (profiles C, D, E, F; Figs. 1, 7) are classified 458 by FAO/UNESCO (1974) as eutric cambisols, reflecting a Mediterranean climate condition. Indeed, on the 459 African continent, which is adjacent to the Canary Islands, eutric cambisols are primarily found within the 460 tropical subhumid zone, gradually transitioning into the semi-arid zone (FAO/UNESCO, 1974). These 461 syenite weathering profiles exhibit better-preserved characteristics and a more significant extent compared 462 to those studied in carbonatites (profiles A and B; Figs. 1, 7). In general, intensive weathering plays a 463 crucial role in the formation of REE deposits, as these elements tend to be concentrated in such geological 464 formations compared to others leached during the weathering process. This phenomenon is exemplified in 465 several locations worldwide, where REE deposits associated with weathering products occur, for instance, 466 Bear Lodge in the USA (Andersen et al., 2017), Chuktukon and Tomtor in Russia (Kravchenko and

467 Pokrovsky, 1995; Kravchenko et al., 2003; Chebotarev et al., 2017), Las Mercedes in the Dominican 468 Republic (Torró et al., 2017), Araxá in Brazil (Braga and Biondi, 2023), and Mount Weld in Australia 469 (Zhukova et al., 2021; Chandler et al., 2024), among many others. However, the weathering processes on 470 Fuerteventura are characterized by fluctuating climatic conditions and intense erosion in the context of a 471 typical Mediterranean climate, which is in turn characterized by drier conditions and a lower propensity for 472 intense weathering compared to tropical climates. The weathering processes on Fuerteventura do not 473 therefore typically lead to the development of laterites and mature weathering profiles, since these 474 conditions do not favor the formation and subsequent preservation of these products, particularly within 475 the carbonatite bedrock areas. Consequently, this constraint substantially reduces the capacity of the FBC 476 to potentially contain economically valuable REE concentrations within the associated weathering 477 products.

478

479 5.3 Fuerteventura carbonatites as potential REE source

480 Based on the mineralogical and geochemical data, it can be concluded that, among the lithologies studied 481 in the FBC, only the carbonatites are favorable targets for further characterization and evaluation of their 482 potential economic viability as an REE source. Therefore, the primary alkaline rocks, as well as the entire 483 suite of corresponding secondary weathering products, can be ruled out.

484 The geochemical data obtained from the oceanic carbonatites of Fuerteventura, exemplified in multielement 485 and REE diagrams (Figure 8), suggest a petrogenetic affinity with carbonatites associated with 486 intracontinental rift geological settings. This similarity has also been previously highlighted by other 487 authors such as Carnevale et al. (2021) who, based on stable isotope data (δ^{13} C and δ^{18} C) and noble gases 488 isotopic composition (He, Ne, Ar), suggested that oceanic and continental carbonatites were comparable in 489 petrogenetic terms. Therefore, despite the lingering questions about the formation processes of oceanic 490 carbonatites, their assessment as a possible source of critical metals, especially REEs, should be fully 491 considered in the same way as their continental counterparts.

However, when considering a more detailed assessment of the sectors where the FBC carbonatites outcrop, it is essential to note that the distribution of these outcrops and thus potential REE mineralization is not straightforward. The carbonatite outcrops have a very limited surface distribution, in the order of meters (Figure 2e), and exhibit complex structural features influenced by shear metamorphism (Figure 2c) and overlapping episodes of intrusive activity that resulted in swarms of dikes with intricate distributions (Figs. 497 2a, b). Hence, these general features of the carbonatite outcrops make it imperative to validly estimate their 498 volume and to carry out more precise studies of their depth distribution, which likely involve drilling and 499 geophysical techniques. These prospective hypothetical findings would provide a deeper understanding of 500 the morphology and dimensions of the carbonatitic bodies, enhancing the ability to calculate resources and 501 reserves while refining the general metallogenic modeling.

- 502 However, it is important to highlight that any attempt to assess potential REE deposits linked to FBC
- 503 carbonatites must consider the irregular distribution of these mineralizations. In addition, it should also be
- 504 considered the existence of regulatory constraints that may stem from the allocation of land for strategic
- 505 military activities, as well as environmental considerations to safeguard natural and marine-coastal areas,
- specially bearing in mind that Fuerteventura is a UNESCO biosphere reserve territory. This latter point is
 particularly pertinent for a specific area within sector 3 (Figure 1). Therefore, any comprehensive analysis
- 508 of the potential of FBC carbonatites as REE sources must also factor in these potential restrictions tied to
- 509 land use regulations aimed at upholding the broader socio-economic, environmental, and societal interests
- 510 inherent to a distinctive site like the island of Fuerteventura.
- 511

512 6 Conclusions

A preliminary evaluation of rare earth element (REE) content was conducted through a mineralogical and geochemical study of alkaline and carbonatitic igneous rocks within the Fuerteventura basal complex (FBC), along with associated weathering products. Based on the gathered data and their corresponding interpretations, our findings can be summarized as follows:

- 517 (i) The concentrations of REEs present in the alkaline and carbonatitic rocks of the FBC are
 518 significant and exceed the average values attributed to the Earth's crust.
- 519 (ii) The weathering products developed on these magmatic rocks do not exhibit significant REE
 520 enrichment.
- 521 (iii) Calcified horizons (Bk, calcretes), spatially related with carbonatites, have practically
 522 negligible concentrations of REE elements. Colluvial processes may have influenced the
 523 lateral transport and accumulation of REEs in Pleistocene-Holocene deposits distant from the
 524 source area.
- 525 (iv) Among the magmatic rocks, carbonatites are the only lithology studied within the FBC with 526 a real potential to host REE mineral resources. The detected concentrations of REY in

- 527 carbonatites range up to about 10,300 ppm, which is a comparable concentration to other 528 locations hosting significant deposits of these critical elements worldwide.
- (v) Within carbonatites, REEs are primarily hosted in two accessory mineral phases: (1) oxides
 belonging to the pyrochlore group; and (2) phosphates. In this second group, primary phases
 such as REE-bearing britholite can be distinguished, as well as monazite generated as a
 secondary product from the britholite alteration.
- (vi) Primary calcite in the Fuerteventura carbonatites is not the predominant host of REEs. It
 displays a highly homogeneous composition with insignificant Fe-Mg content and negligible
 REEs.
- (vii) The carbonatites within the FBC could be considered potential REE resources associated with
 a non-conventional geological setting. However, the complex structural features of the studied
 FBC outcrops (deformation, metamorphism, swarms of dikes from different intrusive
 pulses...) make it essential to conduct more detailed studies to quantify the real economic
 possibilities of this lithology as an REE source.
- (viii) All the studied sectors contain outcrops located in restricted areas due to environmental or
 military use concerns. Any further detailed evaluation of the FBC carbonatites must take into
 account the environmental, socio-economic, and geostrategic factors that will significantly
 limit the real potential extension of REE deposits, considering a hypothetical exploitation.
- 545

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558 the elaboration of polished thin sections.

559

560 Statements and Declarations

561 Data availability statement

- 562 The authors confirm that the data supporting the findings of this study are available within the article and
- 563 its supplementary materials.

564

565 Competing interests

- 566 The authors declare no competing interests. The funders had no role in the design of the study, in the
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569 Author contributions

- 570 Conceptualization: MC, IM, LQ, JY, JM; fieldwork and sampling: MC, IM, LQ, JY, RC, AA, JM;
- 571 methodology: MC, IM, JY, JM; validation of results: MC, IM, LQ, JY, RC, JMR, JM; data curation: MC,
- 572 IM, JM; writing-original draft preparation: MC, IM, JY, JM; writing-review editing: MC, IM, LQ, JY, RC,
- 573 JMR, JM; supervision: IM, JY, JM; project administration: JMR, JM; funding acquisition: IM, JY, RC,
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575

576 Additional information

577 Supplementary tables are available in the online version at https: <u>XXXXX</u>

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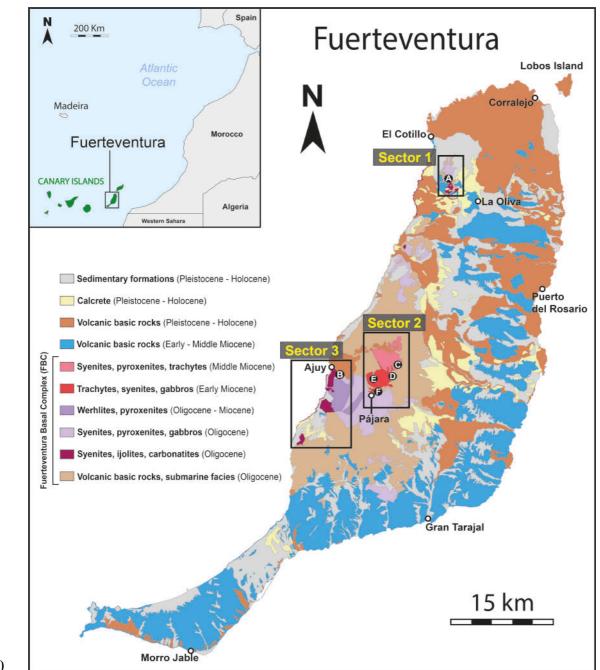
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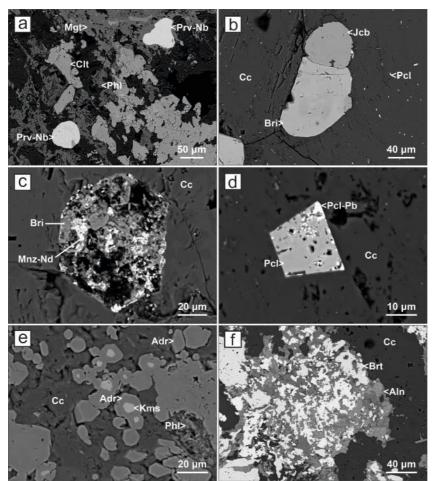
Figure 1

932 Figure 1: Simplified geological map of Fuerteventura Island (modified from Balcells et al., 2006) showing 933 the location of the three study sectors for the assessment of REE content in the FBC. Additionally, the 934 studied weathering profiles are also indicated, as: (A) Agua Salada ravine; (B) Aulagar ravine; (C) 935 Palomares ravine; (D) FV-30 road; (E) Peñitas (F) Pájara. Las quarry; 936

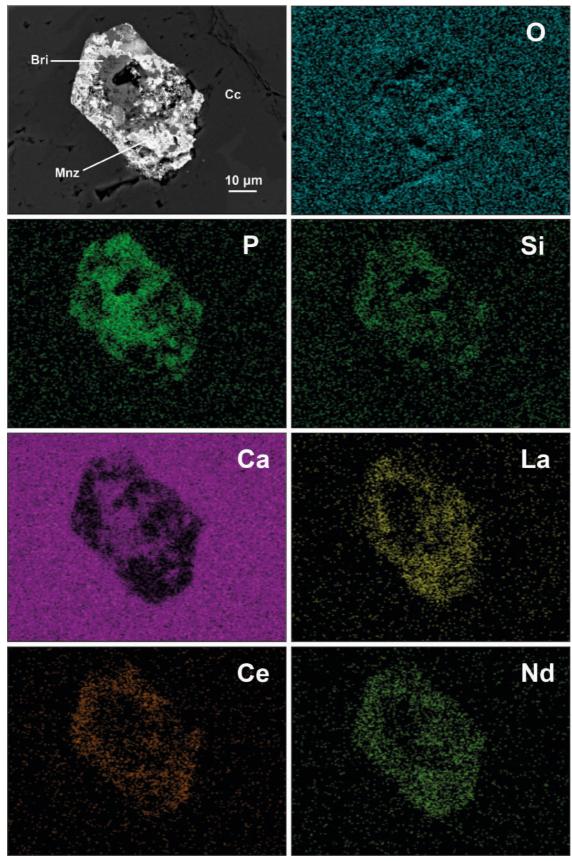
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941 Figure 2: (a), (b) Images showing typical outcrops of the FBC in the southern area of Ajuy (sector 3; 942 Figure 1). The images highlight characteristic swarms of alkaline and carbonatitic intrusions (whitish) 943 intersected by later-intruded basaltic dikes (black colour). (c) Detailed view of a carbonatitic dike located 944 in a shear zone of sector 3, exhibiting distinct linear sigmoidal structures resulting from deformation. (d) 945 Detailed view of centimetre-sized phlogopite crystals within a carbonatitic dike outcropping in sector 3, 946 displaying a typical pegmatitic texture. (e) Overview of an outcrop of metric-scale carbonatitic dikes in the 947 sector 1 area.



950 Figure 3: SEM (backscattered electron, BSE) images of the Fuerteventura carbonatites. (a) Subhedral 951 crystals of niobium-rich perovskite (Prv-Nb) associated with phlogopite (Phl) and magnetite (Mgt) 952 aggregates. The association has been affected by secondary hydrothermal processes, leading to the 953 formation of celestine (Clt). (b) Typical subhedral crystal of jacobsite (Jcb) associated with britholite (Bri). 954 Both crystals are hosted in magmatic calcite (Cc), with numerous disseminated microcrystals of pyrochlore 955 (Pcl). (c) Partially altered subhedral grain of britholite (Bri) hosted in magmatic calcite (Cc). The alteration 956 process led to the formation of secondary REE phosphates such as monazite-Nd (Mnz-Nd). (d) Euhedral 957 crystal of pyrochlore (Pcl) hosted in calcite (Cc). Brighter areas developed on the grain's borders correspond 958 to plumbopyrochlore (Pcl-Pb) zonation. (e) Typical mineral association related to small skarn like areas 959 associated with carbonatites. Subhedral zoned crystals of andradite (Adr), hosted in calcite (Cc) and 960 phlogopite (Phl), with a significant Zr zoning leading to kerimasite (Kms) cores. (f) Typical low-961 metamorphic alteration developed on carbonatites composed of allanite (Aln) aggregates hosted in calcite 962 (Cc) and associated with secondary baryte (Brt). Abbreviations of mineral names in all the pictures follow 963 the criteria proposed by Warr (2021).







967 of britholite (Bri) hosted in calcite (Cc) and partially transformed into secondary monazite (Mnz).



Figure 5: (a) General view of a typical surface outcrop of Quaternary calcrete located in the Aulagar ravine
area (profile B, sector 3; Fig.1). (b) Centimetre-thick calcrete layer filling a fracture between two
carbonatitic dikes in the Aulagar ravine area (profile B, sector 3; Fig.1). (c) Calcrete layer developed within
fractures between carbonatitic rocks in the Agua Salada ravine area (profile A, sector 1; Fig.1).



988 Figure 6: (a) General view of Las Peñitas quarry syenite outcrop (profile E, sector 2; Fig.1) where it is 989 possible to distinguish different fractures filled by injected secondary weathering products. (b) Syenite 990 weathering profile in Las Peñitas quarry (profile E, sector 2; Figure 1) showing surface erosion and B, BC 991 and C horizons injected in the syenite bedrock (R). (c) Weathering profile displaying the development of 992 C and BC horizons associated with a syenite protolith (R), located in Las Peñitas quarry (profile E, sector 993 2; Figure 1). (d) Weathering profile developed on syenite in the FV-30 road area (profile D, sector 2), 994 exhibiting the development of C, B and calcrete (Bk) horizons. (e) Weathering profile on syenite protolith 995 (R) displaying a metric sized C horizon in the Pájara area (profile F, sector 2; Figure 1).



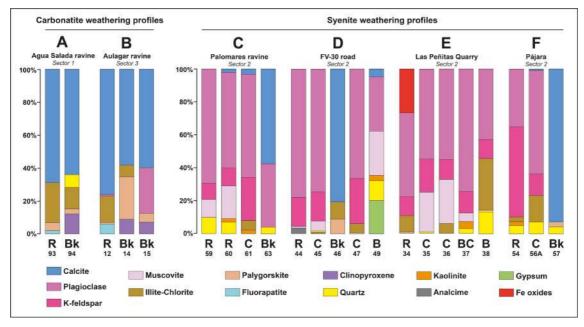


Figure 7: Graphical mineralogical quantification of the studied weathering profiles: (A) Agua Salada ravine; (B) Aulagar ravine; (C) Palomares ravine; (D) FV-30 road; (E) Las Peñitas quarry; (F) Pájara. The corresponding class assigned to the edaphic horizons (B, BC, Bk, C, R) and the sample number are shown at the foot of the columns.

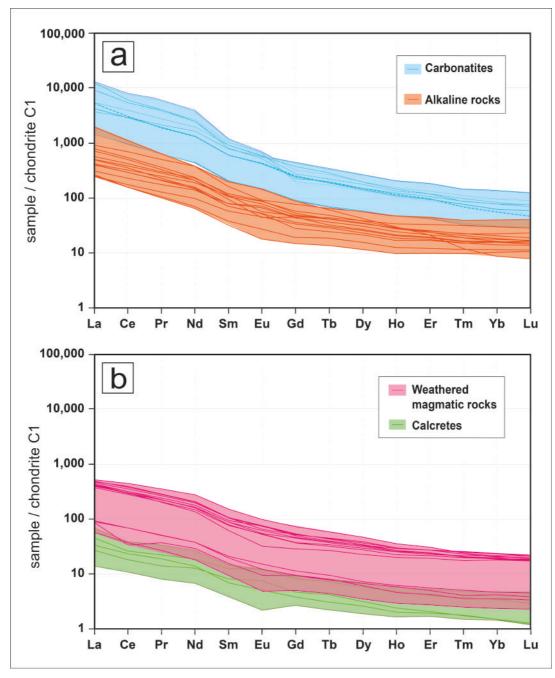




Figure 8: REE plots of the studied Fuerteventura lithologies normalised to C1 chondrites. Normalisation
values are from McDonough and Sun (1995).

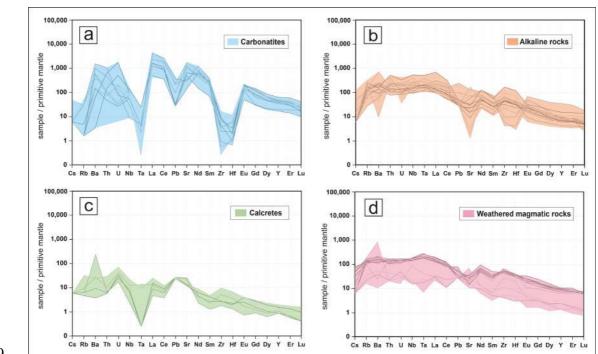
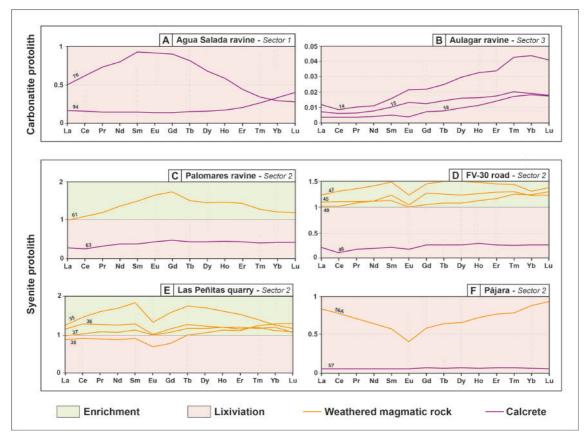




Figure 9

1030 Figure 9: Multi-elemental trace element plots of Fuerteventura intrusive lithologies normalised to the

- 1031 primitive mantle. Normalisation values from McDonough and Sun (1995).



1050 Figure 10: REE weathering enrichment/leaching diagrams between primary magmatic protoliths 1051 (carbonatites and syenites) and the associated weathering products from the studied profiles (Figure 1). The 1052 sample number is labelled on the corresponding pattern line.

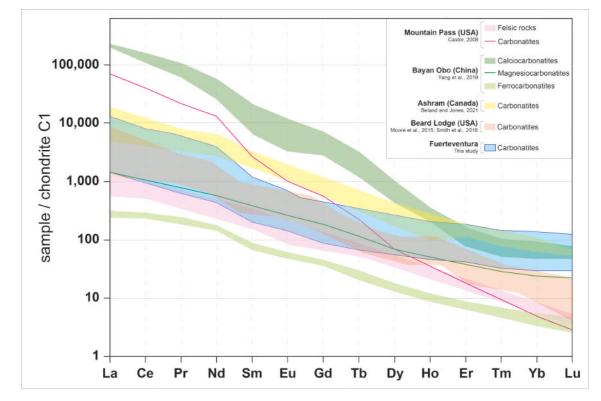


Figure 11: REE plot of the studied Fuerteventura carbonatites compared to other carbonatitic localities
worldwide where REE deposits have been reported. REE contents for comparison are from Castor (2008),
Yang et al. (2019), and Beland and Jones (2021). Normalisation values are from McDonough and Sun
(1995).