

Supplemental Material to: Deployment and evaluation of an $\text{NH}_4^+/\text{H}_3\text{O}^+$ reagent-ion switching chemical ionization mass spectrometer for the detection of reduced and oxygenated gas-phase organic compounds

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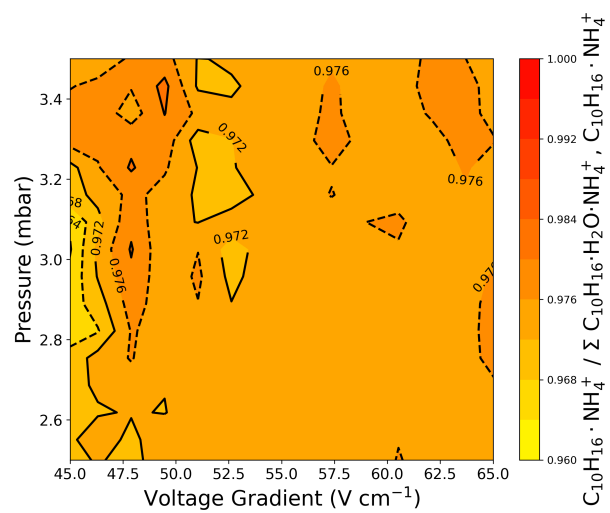


Figure S1. Contribution of the α -pinene water cluster ($\text{C}_{10}\text{H}_{16} \cdot \text{H}_2\text{O} \cdot \text{NH}_4^+$) to the sum of the water cluster and molecular ion with NH_4^+ .

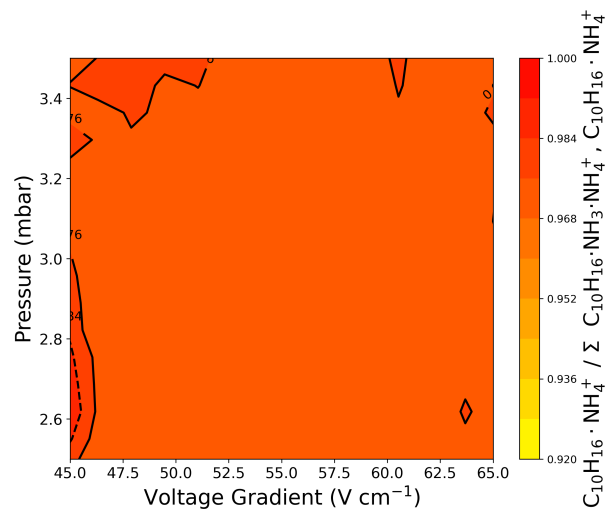


Figure S2. Contribution of the α -pinene ammonia cluster ($\text{C}_{10}\text{H}_{16} \cdot \text{NH}_3 \cdot \text{NH}_4^+$) to the sum of the ammonia cluster and molecular ion with NH_4^+ .

Table S1. Table of NH_4^+ and H^+ affinities obtained from the NIST Chemistry WebBook (Edward P. Hunter and Sharon G. Lias; Michael M. Meot-Ner (Mautner) and Sharon G. Lias).

Compound	NH_4^+ Affinity [kJ mol ⁻¹]	H^+ Affinity [kJ mol ⁻¹]
Ethylene	42	681
Acetone	118	812
Methylal	122	<i>no data</i>
2-Methyl-propene	146	802
1,2-Dimethoxy-ethane	160	858
Benzene	81	750
Cyclohexane	40	687
1,3,5-Trimethylbenzene	91	836
Hydrogen sulfide	48	705
Water	86	691
Ammonia	107	854

Table S2. Vapor pressure estimates of certified gas standard analytes at 25°C.^a

Analyte	Formula	Vapor Pressure $\mu\text{g m}^{-3}$	Vapor Pressure Pa	Method
dimethyl sulfide	C ₂ H ₆ S	1.60E+9	6.38E+4	Mean VP of Antoine & Grain methods
methane thiol	CH ₄ S	3.92E+9	2.02E+5	Mean VP of Antoine & Grain methods
dimethyl sulfoxide	C ₂ H ₆ SO	2.61E+6	82.9	Mean VP of Antoine & Grain methods
benzene	C ₆ H ₆	3.66E+8	1.16E+4	Mean VP of Antoine & Grain methods
toluene	C ₇ H ₈	1.17E+8	3.16E+3	Mean VP of Antoine & Grain methods
1,3,5-trimethylbenzene	C ₉ H ₁₂	1.30E+7	268	Mean VP of Antoine & Grain methods
phenol	C ₆ H ₆ O	1.63E+6	43	Modified Grain method
isoprene	C ₅ H ₈	2.02E+9	7.35E+4	Mean VP of Antoine & Grain methods
limonene	C ₁₀ H ₁₆	1.06E+7	193	Mean VP of Antoine & Grain methods
acetone	C ₃ H ₆ O	7.78E+8	3.32E+4	Mean VP of Antoine & Grain methods
hydroxyacetone	C ₃ H ₆ O ₂	6.93E+6	232	Mean VP of Antoine & Grain methods
methyl ethyl ketone	C ₄ H ₈ O	3.81E+8	1.31E+4	Mean VP of Antoine & Grain methods
methyl vinyl ketone	C ₄ H ₆ O	3.45E+8	1.22E+4	Mean VP of Antoine & Grain methods
3-hexanone	C ₆ H ₁₂ O	8.73E+7	2.16E+3	Mean VP of Antoine & Grain methods
2-octanone	C ₈ H ₁₆ O	1.27E+7	246	Mean VP of Antoine & Grain methods
camphor	C ₁₀ H ₁₆ O	8.72E+4	1.42	Modified Grain method
acetaldehyde	C ₂ H ₄ O	2.15E+9	1.21E+5	Mean VP of Antoine & Grain methods
trans-2-hexenal	C ₆ H ₁₀ O	2.49E+7	629	Mean VP of Antoine & Grain methods
beta-cyclocitral	C ₁₀ H ₁₆ O	1.48E+6	24.1	Mean VP of Antoine & Grain methods
trans-3-hexenol	C ₆ H ₁₂ O	5.05E+6	125	Mean VP of Antoine & Grain methods
acetonitrile	C ₂ H ₃ N	4.42E+8	2.67E+4	Mean VP of Antoine & Grain methods
propane-1,2-diol	C ₃ H ₈ O ₂	4.54E+5	14.8	Mean VP of Antoine & Grain methods
D5-Siloxane	C ₁₀ H ₃₀ O ₅ Si ₅	4.35E+6	29.1	Mean VP of Antoine & Grain methods

^a Estimated using EPI Suite (US EPA).

Table S3. Comparison of sensitivities calculated in this study with those reported by Xu et al. (2022).^a

Analyte	This Study counts s ⁻¹ ppt _v ⁻¹	Xu et al. (2022) counts s ⁻¹ ppt _v ⁻¹
Acetonitrile	0.85	0.55
Acetaldehyde	<0.1	0.021
Acetone	0.98	1.2
Isoprene	<0.1	0.028
Methyl vinyl ketone	1.5	1.5
Methyl ethyl ketone	1.3	1.6
Hydroxyacetone	1.6	2.1
Benzene	<0.1	<0.001
Phenol	<0.1	0.19
Hexanone ^b	3.4	3.8
Trimethylbenzene ^c	<0.1	<0.001

^a We are using a Vocus-S and Xu et al. (2022) report using a Vocus-2R which have different time-of-flight region lengths. The instruments also differ in extraction frequency (i.e., 25 kHz here, and 17.5 kHz for Xu et al. (2022)).

^b This study used 3-Hexanone and Xu et al. (2022) used 2-Hexanone.

^c This study used 1,2,5-Trimethylbenzene and Xu et al. (2022) used 1,3,5-Trimethylbenzene.

Table S4. Vapor pressure estimates of potential biogenic ROC analytes at 25°C.^a

Analyte <i>MCM name</i>	Formula	Vapor Pressure $\mu\text{g m}^{-3}$	Vapor Pressure Pa	Method
LIMCOOH	C ₁₀ H ₁₈ O ₃	4.40E+2	0.00586	Modified Grain method
LIMAOH	C ₁₀ H ₁₈ O ₂	3.55E+3	0.0517	Modified Grain method
LIMBCO	C ₁₀ H ₁₆ O ₂	7.47E+3	0.11	Modified Grain method
LIMONONIC	C ₁₀ H ₁₆ O ₃	8.77E+3	0.118	Modified Grain method
APINBCO	C ₁₀ H ₁₆ O ₂	1.76E+4	0.26	Modified Grain method
PINAL	C ₁₀ H ₁₆ O ₂	3.62E+5	5.34	Modified Grain method
PINONIC	C ₁₀ H ₁₆ O ₃	6.15E+3	0.0828	Modified Grain method
C109OH	C ₁₀ H ₁₆ O ₃	4.66E+2	0.00627	Modified Grain method
C107OH	C ₁₀ H ₁₆ O ₃	1.99E+3	0.0268	Modified Grain method
HCO5	C ₅ H ₈ O ₂	2.07E+6	51.2	Mean VP of Antoine & Grain methods
LIMALNO3	C ₁₀ H ₁₇ NO ₆	1.16E+1	0.000116	Modified Grain method
NLIMALOH	C ₁₀ H ₁₇ NO ₆	2.32E+1	0.000233	Modified Grain method
MBOANO3	C ₅ H ₁₁ NO ₅	7.79E+3	0.117	Modified Grain method

^a Estimated using EPI Suite (US EPA).

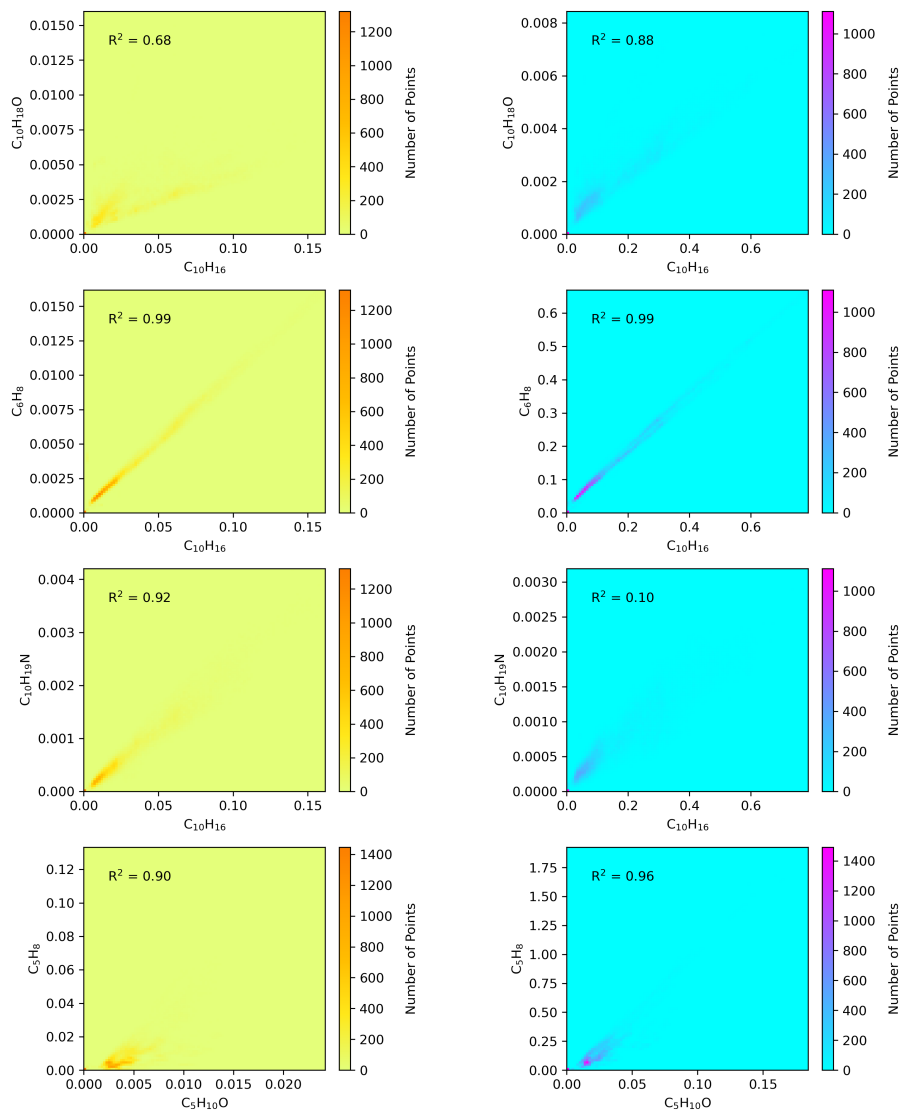


Figure S3. 2D-histograms of ion signals for a selection of ions detected by NH_4^+ (orange/yellow, left) and H_3O^+ (blue/purple, right). Color bars show frequency per bin for the 100 x100 bin grid. Signals for the ions are in counts extraction⁻¹ at 25 kHz extraction frequency.

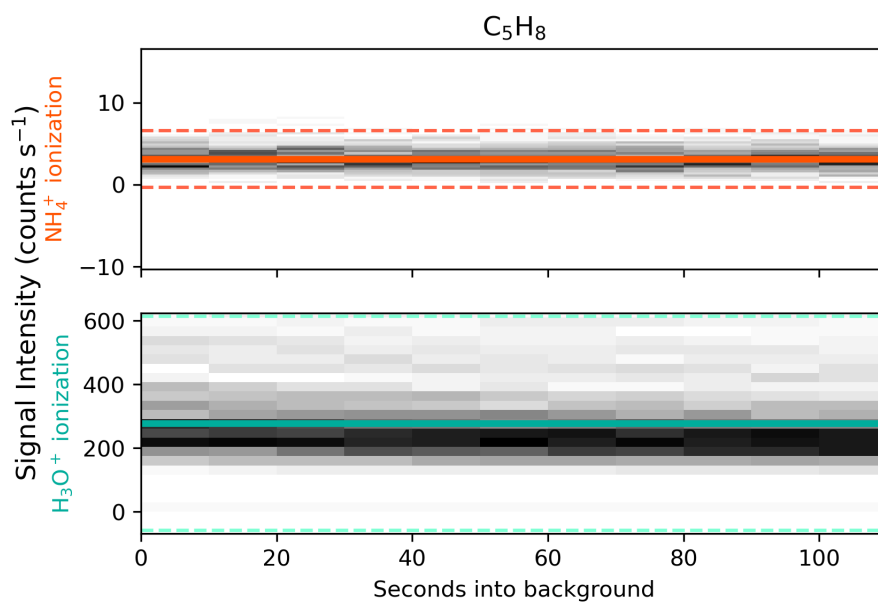


Figure S4. 2D-histograms of campaign zero periods for the C_5H_8 ion detected with NH_4^+ (orange, top) and H_3O^+ (blue, bottom). Solid horizontal lines represent mean signal during zero periods and dashed lines represent 3σ deviation from the mean.

Table S5. Potential structures and literature precedent for organic nitrate peaks.

Chemical Formula	Name in MCM ^a	Reported in Fry et al. (2013)?	Potential Assignment
C ₅ H ₉ NO ₅	NMBOBCO C4MCONO3OH	no	232-MBO nitrate/ Isoprene nitrate
C ₅ H ₁₁ NO ₅	MBOANO3	no	232-MBO nitrate
C ₁₀ H ₁₅ NO ₃		no	Dehydration fragment of C ₁₀ H ₁₇ NO ₄
C ₁₀ H ₁₅ NO ₄	NC101CO NC91CHO	yes (night)	Terpene nitrate (carbonyl)
C ₁₀ H ₁₅ NO ₅	PINALNO3	no	Fragment from LIMALNO3? Faxon et al. (2018) speculate that it could be a fragment of dimers. Oxidation of primary emissions of terpenoid oxygenates.
C ₁₀ H ₁₅ NO ₆	C10PAN2 C923PAN C918PAN C108NO3	no	Terpene nitrate (PAN/carbonyl nitrate)
C ₁₀ H ₁₇ NO ₄	APINCNO3 BPINBNO3 LIMANO3 LIMCNO3 APINANO3 BPINANO3	yes (day and night)	Terpene nitrate (alcohol)
C ₁₀ H ₁₇ NO ₅	NBPINAOOH NAPINBOOH NLIMOOH	yes (night)	Terpene nitrate (hydroperoxy)
C ₁₀ H ₁₇ NO ₆	LIMALNO3 NLIMALOH	no	Limonene nitrate
C ₁₀ H ₁₆ N ₂ O ₆		no	Proposed as α -pinene oxidation product (Bates et al., 2022)

^aExploration of the MCM is non-exhaustive (Saunders et al., 2003; Jenkin et al., 2015).

References

- Bates, K. H., Burke, G. J. P., Cope, J. D., and Nguyen, T. B.: Secondary organic aerosol and organic nitrogen yields from the nitrate radical (NO₃) oxidation of alpha-pinene from various RO₂ fates, *Atmospheric Chemistry and Physics*, 22, 1467–1482, <https://doi.org/10.5194/acp-22-1467-2022>, 2022.
- 5 Edward P. Hunter and Sharon G. Lias: Proton Affinity Evaluation, in: NIST Chemistry WebBook, NIST Standard Reference Database Number 69, Eds. P.J. Linstrom and W.G. Mallard, National Institute of Standards and Technology, Gaithersburg MD, 20899, <https://doi.org/https://doi.org/10.18434/T4D303>.
- Faxon, C., Hammes, J., Le Breton, M., Pathak, R. K., and Hallquist, M.: Characterization of organic nitrate constituents of secondary organic aerosol (SOA) from nitrate-radical-initiated oxidation of limonene using high-resolution chemical ionization mass spectrometry, *Atmospheric Chemistry and Physics*, 18, 5467–5481, <https://doi.org/10.5194/acp-18-5467-2018>, 2018.
- 10 Fry, J. L., Draper, D. C., Zarzana, K. J., Campuzano-Jost, P., Day, D. A., Jimenez, J. L., Brown, S. S., Cohen, R. C., Kaser, L., Hansel, A., Cappellin, L., Karl, T., Hodzic Roux, A., Turnipseed, A., Cantrell, C., Lefer, B. L., and Grossberg, N.: Observations of gas- and aerosol-phase organic nitrates at BEACHON-RoMBAS 2011, *Atmospheric Chemistry and Physics*, 13, 8585–8605, <https://doi.org/10.5194/acp-13-8585-2013>, 2013.
- 15 Jenkin, M. E., Young, J. C., and Rickard, A. R.: The MCM v3.3.1 degradation scheme for isoprene, *Atmospheric Chemistry and Physics*, 15, 11 433–11 459, <https://doi.org/10.5194/acp-15-11433-2015>, 2015.
- Michael M. Meot-Ner (Mautner) and Sharon G. Lias: Binding Energies Between Ions and Molecules, and The Thermochemistry of Cluster Ions, in: NIST Chemistry WebBook, NIST Standard Reference Database Number 69, Eds. P.J. Linstrom and W.G. Mallard, National Institute of Standards and Technology, Gaithersburg MD, 20899, <https://doi.org/https://doi.org/10.18434/T4D303>.
- 20 Saunders, S. M., Jenkin, M. E., Derwent, R. G., and Pilling, M. J.: Protocol for the development of the Master Chemical Mechanism, MCM v3 (Part A): tropospheric degradation of non-aromatic volatile organic compounds, *Atmospheric Chemistry and Physics*, 3, 161–180, <https://doi.org/10.5194/acp-3-161-2003>, 2003.
- US EPA: Estimations Programs Interface Suite for Microsoft Windows, <https://www.epa.gov/tsca-screening-tools/epi-suitetm-estimation-program-interface>.
- 25 Xu, L., Coggon, M. M., Stockwell, C. E., Gilman, J. B., Robinson, M. A., Breitenlechner, M., Lamplugh, A., Crounse, J. D., Wennberg, P. O., Neuman, J. A., Novak, G. A., Veres, P. R., Brown, S. S., and Warneke, C.: Chemical ionization mass spectrometry utilizing ammonium ions (NH₄⁺ CIMS) for measurements of organic compounds in the atmosphere, *Atmospheric Measurement Techniques*, 15, 7353–7373, <https://doi.org/10.5194/amt-15-7353-2022>, 2022.