



- Ocean Acidification trends and Carbonate System dynamics in
- the North Atlantic Subpolar Gyre during 2009-2019
- 3 David Curbelo-Hernández<sup>1</sup>, Fiz F. Pérez<sup>2</sup>, Melchor González-Dávila<sup>1,\*</sup>, Sergey V.
- 4 Gladyshev<sup>3</sup>, Aridane G. González<sup>1</sup>, David González-Santana<sup>1</sup>, Antón Velo<sup>2</sup>, Alexey Sokov<sup>3</sup>,
- 5 and J. Magdalena Santana-Casiano<sup>1</sup>.
- 6 <sup>1</sup> Instituto de Oceanografía y Cambio Global (IOCAG), Universidad de Las Palmas de Gran
- 7 Canaria (ULPGC). Las Palmas de Gran Canaria, Spain.
- 8 <sup>2</sup> Instituto de Investigaciones Marinas (IIM), CSIC, Vigo, Spain.
- 9 <sup>3</sup> P. P. Shirshov Institute of Oceanology, Russian Academy of Sciences, Moscow, Russian
- 10 Federation
- \*Corresponding Author: Melchor González-Dávila (melchor.gonzalez@ulpgc.es)
- 12 Keypoints:
- 13 During the 2010s, the subpolar North Atlantic experienced a 50-86% increase in
- anthropogenic CO<sub>2</sub>, accelerating by <10% the acidification.
- Anthropogenic CO<sub>2</sub> contributed to acidification by 53-68% in upper layers and >82% in the
- 16 interior ocean.
- 17 The acidification trends (0.0006 and 0.0032 units yr<sup>-1</sup>) declined the  $\Omega_{\text{Ca}}$  and  $\Omega_{\text{Arag}}$  by 0.004-
- 18 0.021 and 0.003-0.0013 units yr<sup>-1</sup>, respectively.



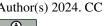


#### Abstract

19

20 The CO<sub>2</sub>-carbonate system dynamics in the North Atlantic Subpolar Gyre (NASPG) were 21 evaluated between 2009 and 2019. Data was collected aboard eight summer cruises through 22 the CLIVAR 59.5°N section. The Ocean Acidification (OA) patterns and the reduction in the saturation state of calcite ( $\Omega_{Ca}$ ) and aragonite ( $\Omega_{Arag}$ ) in response to the increasing 23 anthropogenic CO<sub>2</sub> (C<sub>ant</sub>) were assessed within the Irminger, Iceland and Rockall basins 24 25 during a poorly-assessed decade in which the physical patterns reversed in comparison with 26 previous well-known periods. The observed cooling, freshening and enhanced ventilation increased the interannual rate of accumulation of  $C_{\rm ant}$  in the interior ocean by 50-86% and 27 the OA rates by close to 10%. The OA trends were 0.0013-0.0032 units yr<sup>-1</sup> in the Irminger 28 and Iceland basin and 0.0006-0.0024 units yr<sup>-1</sup> in the Rockall Trough, causing a decline in 29  $\Omega_{\text{Ca}}$  and  $\Omega_{\text{Arag}}$  of 0.004-0.021 and 0.003-0.0013 units yr<sup>-1</sup>, respectively. The  $C_{\text{ant}}$ -driven rise 30 in total inorganic carbon (C<sub>T</sub>) was the main driver of the OA (contributed by 53-68% in upper 31 layers and >82% toward the interior ocean) and the reduction in  $\Omega_{Ca}$  and  $\Omega_{Arag}$  (>64%). The 32 transient decrease in temperature, salinity and A<sub>T</sub> collectively counteracts the C<sub>T</sub>-driven 33 34 acidification by 45-85% in the upper layers and in the shallow Rockall Trough and by <10% 35 in the interior ocean. The present investigation reports the acceleration of the OA within the 36 NASPG and expands knowledge about the future state of the ocean.

37 **Keywords:** Ocean Acidification, Anthropogenic Carbon, North Atlantic Subpolar Gyre.





#### 1. Introduction

The ocean uptake of approximately one-third of the CO<sub>2</sub> released into the atmosphere 39 40 (Friedlingstein et al., 2023; Gruber et al., 2019a) has an important role in the climate regulation causing changes in the marine carbonate chemistry. The exponential increase in the global 41 ocean CO<sub>2</sub> sink in phase with those of anthropogenic emissions (Friedlingstein et al., 2023) 42 has resulted in a long-term decrease in the concentration of carbonate ions ([CO<sub>3</sub><sup>2-</sup>]) and pH. 43 This process has been collectively referred to as Ocean Acidification (OA; Caldeira and 44 45 Wickett, 2005, 2003; Doney et al., 2009; Orr et al., 2005; Raven et al., 2005; Feely et al., 2009) and favour the dissolution of calcium carbonate (CaCO<sub>3</sub>). It affects not only calcifying marine 46 organisms and ecosystems which use the biogenic CaCO<sub>3</sub> forms of calcite and aragonite (e. g. 47 Gattuso et al., 2015; Langdon et al., 2000; Pörtner et al., 2004, 2019; Riebesell et al., 2000) 48 49 but also the global biogeochemical cycles (Gehlen et al., 2011; Matear and Lenton, 2014). 50 The absorption of anthropogenic CO<sub>2</sub> has reduced the pH of the global surface ocean by 0.1 51 units since preindustrial times, representing approximately a 30% increase in acidity (Caldeira 52 and Wickett, 2003). The model projections estimate that the pH could fall by 0.5 units by the end of the century if global CO<sub>2</sub> emissions continue to rise, while a drop of 0.2 units is expected 53 for the most conservative scenario (Caldeira and Wickett, 2005; Orr et al., 2005, 2011; Raven 54 et al., 2005). However, as the absorption and storing of anthropogenic carbon ( $C_{ant}$ ) within the 55 56 ocean is not uniform (Sabine et al., 2004a), OA rates may show a significant spatial variability 57 and should be regionally studied. The temporal evolution of the carbonate system variables in surface waters are monitored and assessed in several time-series stations located across 58 different ocean regions (Bates et al., 2014). The largest OA rates are expected to occur across 59 60 high northern and southern latitudes (Bellerby et al., 2005; Orr et al., 2005), where deep convective overturning and subduction occur favouring the entrance of Cant in the interior 61 ocean (Maier-Reimer and Hasselmann, 1987; Lazier et al., 2002; Sarmiento et al., 1992). 62 63 The North Atlantic is one of the strongest CO<sub>2</sub> sinks and stores over 25% of the C<sub>ant</sub> accumulated in the global ocean (e. g. Gruber et al., 2019; Khatiwala et al., 2013; Pérez et al., 64 65 2024, 2010, 2008, 2024; Sabine et al., 2004; Takahashi et al., 2009). The Atlantic Meridional Overturning Circulation (AMOC) plays a significant role by conveying acidified Cant-loaded 66 67 waters polewards and exporting them to the ocean interior across deep-water formation areas



Roberts et al., 2009).



68 (Lazier et al., 2002; Pérez et al., 2013, 2008; Steinfeldt et al., 2009). It contributes to homogenize the  $C_{\text{ant}}$  and pH in the whole water column in such regions and exported these 69 properties southwards to the global deep ocean (Perez et al., 2018). Thus, the North Atlantic 70 71 behaves as a crucial region for understanding the impacts of anthropogenic forcing on the 72 global ocean. 73 OA has been widely studied in the North Atlantic through the monitoring of the ocean 74 physicochemical properties at time-series stations (summarized by Bates et al., 2014) placed 75 in subtropical and subpolar latitudes: the European Station for Time series in the Ocean at the Canary Islands (ESTOC; 29.04°N, 15.50°W; González-Dávila et al., 2010; González-Dávila 76 and Santana-Casiano, 2023; Santana-Casiano et al., 2007), the Bermuda Atlantic Time-series 77 Study (BATS; 32.0°N, 64.0°W; Bates et al., 2012), the Irminger Sea Time Series (IRM-TS; 78 79 64.3°N, 28.0°W; Olafsson et al., 2010) and the Iceland Sea Time Series (IS-TS; 68.0°N, 12.66°W; Olafsson et al., 2009, 2010). OA rates has also been evaluated along transects 80 through repeated hydrographic cruises (i.e. Guallart et al., 2015; García-Ibáñez et al., 2016; 81 Vázquez-Rodríguez et al., 2012b) or even covered by volunteer observing ships (Fröb et al., 82 83 2019). These investigations have revealed a rate of decrease in pH of  $\sim 0.001$ -0.002 units yr<sup>-1</sup>. Moreover, González-Dávila and Santana-Casiano, (2023) has recently indicated that these 84 85 rates are increasing since 1995. 86 The assessment of OA is of especial interest across the North Atlantic Subpolar Gyre (NASPG; 87 50-60°N), where the atmospheric CO<sub>2</sub> sink is particularly strong and the deep-water formation processes favour the storage of  $C_{\text{ant}}$  through the whole water column (Gruber et al., 2019b; 88 Sabine et al., 2004b; Watson et al., 2009, Pérez et al. 2008). Likewise, the deep-water 89 90 formation processes create the largest and deepest ocean environments supersaturated for aragonite (at more than 2000 m depth; Feely et al., 2004; Jiang et al., 2015), which is the main 91 CaCO<sub>3</sub> mineral for Cold-water corals (CWC; Roberts et al., 2009) and some pteropods 92 (Bathmann et al., 1991; Urban-Rich et al., 2001). These deep biomes are predicted to be one 93 of the first in the global ocean affected by OA, mainly due to the shoaling of the Aragonite 94 95 Saturation Horizon and its progressive exposition to undersaturated conditions for aragonite at intermediate and deep waters (Raven et al., 2005; Guinotte et al., 2006; Turley et al., 2007; 96





98 The physical processes along the NASPG, which are subject to significant spatiotemporal 99 variability introduced by the atmospheric forcing and climatology on an interannual scale, directly influenced the biogeochemistry (Corbière et al., 2007; Fröb et al., 2019). The changes 100 101 in North Atlantic Current (NAC) modifies the poleward heat transport from subtropical 102 latitudes and the air-sea interactions, influencing temperature patterns (Josey et al., 2018; 103 Mercier et al., 2015). Recent studies noticed the surface cooling and freshening of the NASPG 104 in the 2010s (Holliday et al., 2020; Josey et al., 2018; Robson et al., 2016; Tesdal et al., 2018) 105 contrasting with the period of warming and salinification in the 1990s extended until 2005 106 (Häkkinen and Rhines, 2004; Hátún et al., 2005; Robson et al., 2014). Anomalously heat loss 107 and winter deep convection were found to be of high intense since 2008 contributing to the extreme cold anomaly along the NASPG (e. g. De Jong et al., 2012; de Jong and de Steur, 108 109 2016; Fröb et al., 2019, 2016; Gladyshev et al., 2016b, 2016a; Piron et al., 2017; Våge et al., 110 2009). These fluctuations in the vertical mixing and ocean circulation patterns introduces 111 changes in the distribution of the carbonate system variables. Several investigations have evaluated the drivers, trends and impacts of OA in the western 112 113 NASPG at the Irminger and Iceland basins (e. g. Fontela et al., 2020; García-Ibáñez et al., 114 2021, 2016; Perez et al., 2018; Pérez et al., 2021; Ríos et al., 2015), while few studies have 115 addressed it along the Rockall Trough (e. g. McGrath et al., 2013, 2012a, 2012b, Humphreys 116 et al., 2016) due to lack of repeated hydrographic sections or time-series stations and 117 subsequent limitation of continuous surface-to-bottom data. The high longitudinal variability 118 in the NASPG caused by the influence of different circulation patterns and water masses 119 (García-Ibáñez et al., 2018, 2015) introduced several physicochemical heterogeneities 120 between the Irminger and Iceland with the Rockall basin (Ellett et al., 1986; McGrath et al., 121 2013, 2012b; Holliday et al., 2000). These differences in the distributions of Marine Carbonate 122 System (MCS) variables should be considered to improve our understanding of OA in the 123 entire North Atlantic. 124 This study evaluated the OA in the NASPG across the Irminger, Iceland and Rockall basins 125 during the 2010s. High-quality direct measurements of CO<sub>2</sub> system variables from eight hydrographic cruises occupying 59.5°N between 2009 and 2019 were used to evaluate the 126 127 drivers and trends of pH, and the potential effects of OA on calcifying organisms of changes



132

150

151

152

153

154

155

156

157



- in calcite (ΩCa) and aragonite (ΩAr) saturation states. This study advances our understanding of the complexities associated with OA in the NASPG and supports ongoing efforts to model and predict future acidification scenarios in the North Atlantic and global ocean.
  - 2. Methodology

#### 2.1. Data collection

Data was collected along the hydrographic CLIVAR 59.5°N section (Daniault et al., 2016; 133 134 Gladyshev et al., 2016c, 2018, 2017; Sarafanov et al., 2018) from 8 summer cruises with dates spanning 11 years (2009-2019). The section covers the length of the North Atlantic at 59.5°N 135 136 between Scotland and Greenland (4.5-43.0°W), crossing the Irminger and Iceland basins and 137 the Rockall Trough (Figure 1). Generally, the sampling stations were equidistantly spaced 138 every 20 nmi apart (~1/3° longitude) and repeated in all the cruises except for the cruise of 139 2016, when the station spacing was decreased to 10 nmi over Reykjanes Ridge western and 140 eastern slopes. The distance between stations over the east Greenland slope and shelf always 141 decreased from 10 nmi to about 2 nmi. The surface-to-bottom sampling and in situ 142 measurements were performed by using a SBE 911plus CTD with SBE32 Carousel containing 143 24 Niskin bottles (10 L) with additional sensors for pressure (P), dual temperature (T) and salinity (S), and dissolved oxygen (DO). The eight cruises included in the new dataset are the 144 145 result of an international collaboration between researchers from the P. P. Shirshov Institute of 146 Oceanology at the Russian Academy of Science and the Marine Chemistry research group 147 from the Oceanography and Global Change Institute (QUIMA-IOCAG) at the University of Las Palmas de Gran Canaria (ULPGC). A detailed overview and metadata of the cruises is 148 149 given in Table 1.

# 2.1.1. CO<sub>2</sub> system variables measurements

The analysis of the MCS variables followed the same analytical methodology and provided high-quality  $CO_2$  measurements in all the hydrographic cruises. It includes the sampling and data collection techniques, quality control and calculation procedures published in the updated version of the DOE method manual for  $CO_2$  analysis in seawater given by Dickson et al., 2007. The seawater samples were onboard analysed for total alkalinity ( $A_T$ ) and total inorganic carbon ( $C_T$ ) determination by using a VINDTA 3C and following Mintrop et al., (2000). The  $A_T$  was analysed by potentiometric titration with HCl to the carbonic acid endpoint and



159

160



(CRMs; provided by A. Dickson at Scripps Institution of Oceanography), giving values with 161 an accuracy of  $\pm 1.5$  µmol kg<sup>-1</sup> for  $A_{\rm T}$  and  $\pm 1.0$  µmol kg<sup>-1</sup> for  $C_{\rm T}$ . 162 Spectrophotometric pH measurements (Clayton and Byrne, 1993) in total scale at constant 163 164 temperature of 15°C (pH<sub>T,15</sub>) were performed for the cruises between 2009 and 2016. A 165 spectrophotometric pH sensor (SP101-SM) developed by the QUIMA-IOCAG group at the ULPGC in collaboration with SensorLab (González-Dávila, 2014; González-Dávila et al., 166 167 2016) was used. The method uses 4 wavelengths analysis for the m-cresol purple, includes 168 auto-cleaning steps and performs a blank for pH calculation immediately after the dye 169 injection. The spectrophotometric sensor was in situ tested by using a TRIS seawater buffer and provided pH<sub>T15</sub> values with an accuracy of ±0.002 units. However, DelValls and Dickson, 170 (1998) reported an uncertainty of the spectrophotometric pH determination associated to the 171 172 TRIS used for calibration of -0.0047 units. Hence, the experimental pH values were corrected 173 by adding 0.0047 units. 2.1.2. Dissolved oxygen (DO) measurements 174 The WINKLER method introduced by Winkler (1888) and optimized by Carpenter (1965) and 175 Carrit and Carpenter (1966) was used to analytically determine the dissolved oxygen (DO) of 176 177 the seawater samples in all the cruises from 2009 to 2016. The seawater samples for DO 178 determination were collected from the bottle samples in pre-calibrated glass wide-neck bottles 179 avoiding bubble formation. The temperature of the water was recorded during the sampling. 180 All the reagents and solutions used for dissolved oxygen determination were prepared 181 following the procedures described by Dickson and Goyet (1994) and their possible impurities 182 were controlled by determining a blank every 2 days. 183 As DO could not be analytically measured during the cruise of 2019 (due to limitations 184 related with the oceanographic cruise plan), it was computed for this year by comparing the 185 performance of the DO sensor during the cruise of 2019 versus (1) DO data estimated by a 186 neural network for the cruises of 2016 and 2019 and (2) WINKLER-measured DO data in the cruise of 2016. The neural network ESPER NN (Empirical Seawater Property Estimation 187

determined through the developing of the full titration curve (Millero et al., 1993; Dickson and Goyet, 1994). The  $C_T$  was determined through coulometric titration (Johnson et al., 1993). The

VINDTA 3C was in situ calibrated through the titration of Certified Reference Material



191

217



188 Routine) introduced by Carter et al., (2021) was used for DO estimations. The computational

procedure is detailed in Appendix A.

# 2.2. Data processing

# 2.2.1. Evaluation of the internal consistency of the data using CANYON-B

The measured and determined data were compared with estimations given by the Bayesian 192 neural network "CANYON-B" (Bittig et al., 2018), a re-developed and more robust neural 193 194 network based on CANYON (CArbonate system and Nutrients concentration from 195 hYdrological properties and Oxygen using a Neural-network; Sauzède et al., 2017). 196 CANYON-B estimates the four MCS variables (A<sub>T</sub>, C<sub>T</sub>, pH<sub>T</sub> and pCO<sub>2</sub>) and macronutrients concentrations (PO<sub>4</sub><sup>3-</sup>, NO<sub>3</sub><sup>-</sup> and Si(OH)<sub>4</sub>, hereinafter PO<sub>4</sub>, NO<sub>3</sub> and Si(OH)<sub>4</sub>) as a function of 197 198 a simple set of input variables which include P, T, S, DO, latitude, longitude and date. This 199 neural network is trained on and validated against bottle and sensor data from GLODAPv2, 200 GO-SHIP and Argo profiles, and provides a local uncertainty for each variable. The standard errors of estimate reported for CANYON-B by Bittig et al., (2018) are 6.3  $\mu$ mol kg<sup>-1</sup> for  $A_T$ , 201 7.1  $\mu$ mol kg<sup>-1</sup> for  $C_T$ , 0.013 units for pH, 20  $\mu$ atm for  $pCO_2$ , 0.051  $\mu$ mol kg<sup>-1</sup> for PO<sub>4</sub>, 0.68 202 μmol kg<sup>-1</sup> for NO<sub>3</sub> and 2.3 μmol kg<sup>-1</sup> for Si(OH)<sub>4</sub>. The crossover analysis between measured 203 204 and estimated data did not show systematic differences but individual outliers. The measured 205 data that were higher/lower than the CANYON-B estimate by plus/minus twice the predicted 206 variable uncertainty of the neural network were considered as outliers and removed from the 207 dataset.

The total amount of measured data was 8974 for  $A_T$ , 7495 for  $C_T$ , 8706 for pH<sub>T</sub>, 9656 for DO, 9114 for PO<sub>4</sub> and 9192 for Si(OH)<sub>4</sub>. The difference between the measured and CANYON-B-estimated variables (referred hereinafter as canyon-estimated variables) were performed for each sample in which CANYON-B could be applied (samples with availability of T, S and DO measurements). The number of data, mean values and standard deviation of the measured variables for each cruise were summarized in Table S1. The average differences with the 95% confidence interval for each cruise are shown in Table S2. The average differences for the

entire period (2009-2019) were lower than 2.1  $\mu$ mol kg<sup>-1</sup> for  $A_T$ , 2  $\mu$ mol kg<sup>-1</sup> for  $C_T$ , 0.0002 for

216  $pH_T$ , 0.02  $\mu$ mol kg<sup>-1</sup> for PO4 and 0.25  $\mu$ mol kg<sup>-1</sup> for Si(OH)4.

#### 2.2.2. Computational methods





218 The computational procedures to calculate MCS system variables applied in this investigation used the CO<sub>2</sub>SYS programme developed by Lewis and Wallace, (1998) and run with the 219 MATLAB software (van Heuven et al., 2011; Orr et al., 2018; Sharp et al., 2023). The set of 220 221 constants used for computations includes the carbonic acid dissociation constants of Lueker et 222 al., (2000), the HSO<sub>4</sub> dissociation constant of Dickson, (1990), the HF dissociation constant 223 of Perez and Fraga, (1987) and the value of [B]<sub>T</sub> determined by Lee et al., (2010). The pH in 224 total scale at in situ temperature (pH<sub>T</sub>) was computed from the measured  $A_T$  and pH<sub>T,15</sub> (the computed C<sub>T</sub> was given as an output). The pH<sub>T</sub> for the cruise of 2019, in which direct pH 225 226 measurements were not performed, was computed from the measured  $A_T$  and  $C_T$ . 227 In addition, as three of the four MCS variables were measured in the rest of the cruises and due to gaps in data, an intercomparison between measured and computed  $C_T$  and pH<sub>T15</sub> was 228 229 performed. It considers the availability of measurements for each latitude, longitude and time and the differences between the measured and computed pH with the canyon-estimated pH<sub>T</sub>. 230 The use of measured or computed  $C_T$  followed these conditions: (1) If there is measured  $C_T$ 231 but not measured pH, measured  $C_T$  was used, (2) if there is measured pH but not measured  $C_T$ , 232 233 computed  $C_T$  was used, (3) and if there is measured  $C_T$  and pH, measured  $C_T$  was used when 234 the differences between measured and canyon pH<sub>T</sub> is lower than the differences between 235 computed and canyon-estimated  $pH_T$ , while computed  $C_T$  was used when the opposite happens. In total, 6375 measured and 2872 computed C<sub>T</sub> data were used in this study (69% 236 237 and 31%, respectively). The average differences in each cruise between the combined (measured and computed, also referred as " $C_{T \text{ (new)}}$ ") and canyon-estimated  $C_{T}$  variable is 238 provided in Table S2. The amount and percentage of measured and computed  $C_T$  data per 239 240 cruise is given in Table S3. As the measured  $C_{\rm T}$  was in average 1.9  $\mu$ mol kg<sup>-1</sup> higher than the canyon-estimated and the computed C<sub>T</sub> was in average 1.7 µmol kg<sup>-1</sup> lower, the new 241 compilation based on these previous conditions allowed to reduce the difference to 1.5 µmol 242 kg-1. 243

# 2.2.3. Anthropogenic $CO_2$ ( $C_{ant}$ ) calculation

The anthropogenic CO<sub>2</sub> ( $C_{ant}$ ) was estimated by using the biogeochemical back-calculation  $\phi C_T^o$  method, which has an overall estimated uncertainty of  $\pm 5.2 \,\mu$ mol kg<sup>-1</sup> (Pérez et al., 2008; Vázquez-Rodríguez et al., 2009). The method considers the change of  $C_T$  between the





preindustrial era (1750) and the time of the observations, as well as the processes involved in the uptake and distribution of  $C_{\text{ant}}$  (biogeochemistry, mixing processes and air-sea fluxes). The  $C_{\text{ant}}$  was calculated (Eq. 1) as the difference between the  $C_{\text{T}}$  at the time of observation, the  $C_{\text{T}}$ that the seawater would have in equilibrium with a preindustrial atmosphere (preformed  $C_T$ ;  $C_{\rm T}^{\rm pre}$ ), the offsets of such equilibrium values (air-sea CO<sub>2</sub> disequilibrium;  $\Delta C_{\rm T}^{\rm dis}$ ) and the changes in  $C_T$  due to the organic and carbonate pumps ( $\Delta C_T^{\text{bio}}$ ). The  $C_T$  and  $A_T$  at the time of observations and the preformed  $A_{\rm T}(A_{\rm T}^0)$  are needed as input parameters and the computational procedure was described by Vázquez-Rodríguez et al., (2012). 

$$C_{ant} = C_T - C_T^{pre} - \Delta C_T^{dis} - \Delta C_T^{bio}$$
 (1)

The  $\phi C_T^o$  method is an improved process-based  $C_{ant}$  estimation method tested and widely applied in the Atlantic Ocean (Vázquez-Rodríguez et al., 2009) which present distinctive characteristics relative to existing  $C_{ant}$  approaches, such as the classical  $\Delta C^*$  (GSS' 96; Gruber et al., 1996) and the TrOCA (Touratier et al., 2007). The main advantages of the  $\phi C_T^o$  method has been described by Pérez et al., (2008).

#### 2.2.4. Water mass characterization

The characterization of the basins and water masses was done by considering the 2006-2021 mean combined CLIVAR 59.5°N section constructed with potential vorticity, dissolved oxygen and salinity together with the large-scale circulation in the North Atlantic (e. g. Lherminier et al., 2010; Pérez et al., 2021; Sarafanov et al., 2012; Schmitz and McCartney, 1993; Schott and Brandt, 2007; Sutherland and Pickart, 2008). A schematic diagram with the main surface and deep currents in the NASPG is depicted in Figure 1a. The basin division considered the NAC pathways and revealed a west-to-east distribution comprising the Irminger and Iceland basins and the Rockall Trough. The Iceland basin was delimited along its eastern boundary by the central NAC branches around the northern part of the Haton Bank and George Bligh Bank, and along its western boundary by the Return Current over the eastern flank of the Reykjanes Ridge slope. This suggest that the Iceland basin could be longitudinally separated in two subregions: the western Iceland basin (24.0-29.5°W) and the eastern Iceland basin (14.0-24.0°W).





276 The upper layers were mainly occupied by Subpolar Mode Waters (SPMW) and North Atlantic 277 Central Waters (NACW). SPMW is formed in the Iceland basin (McCartney and Talley, 1982; Brambilla and Talley, 2008; Tsuchiya et al., 1992; Van Aken and Becker, 1996), flow eastward 278 279 to the Rockall Trough and recirculate across the Reykjanes Ridge (Brambilla and Talley, 2008). 280 In the Irminger basin, SPMW flow with the Irminger Current to the north over the western 281 Reykjanes Ridge flank and to the south over the eastern Greenland slope (Figure 1a). Thus, 282 SPMW signal was detected in the western and eastern Irminger basin up to 400-700 m depth 283 and limited to subsurface depths in the central part of the basin. NACW were placed above 284 SPMW east of the Irminger basin and separated in two branches: Eastern North Atlantic 285 Central Water (ENACW), formed by winter convection in the intergyre region and moved 286 poleward from the Bay of Biscay through the Rockall Trough (Harvey, 1982; Pollard et al., 287 1996), and Western North Atlantic Central Water (WNACW), flowing northward with the 288 NAC along the western Iceland basin. The intermediate layers were mainly occupied by 289 Labrador Sea Water (LSW), formed in the Labrador Sea and transported eastward (e. g. Pickart 290 et al., 2003; Fröb et al., 2016). LSW path diverges into two cores when it reaches the Reykjanes 291 Ridge (Álvarez et al., 2004; Pickart et al., 2003): a fraction of LSW rapidly moved to the 292 Irminger basin and incorporated into the Deep Western Boundary Current (DWBC) (Bersch et 293 al., 2007) and a second LSW core was transported eastward into the Iceland and Rockall 294 basins. In the Irminger and western Iceland basin, LSW placed above Iceland-Scotland 295 Overflow Water (ISOW), which originated from the overflow of Norwegian Sea waters over 296 the Iceland-Scotland ridges and flowed southward and below 1500 m depth through the 297 western NASPG (van Aken and de Boer, 1995; Dickson et al., 2002; Fogelqvist et al., 2003). 298 The bottom of the western Irminger basin was occupied by Denmark Strait Overflow Water 299 (DSOW), recently formed from deep waters from the Nordic seas flowing southward over the 300 Greenland-Iceland ridge and sinking through the eastern Greenland slope (Read, 2000; 301 Stramma et al., 2004; Yashayaev and Dickson, 2008). LSW core transported eastwards rises 302 in depth through the western Haton Bank flank and occupy the bottom depths in the eastern 303 Iceland basin and in the Rockall Trough. A low-ventilated thermocline layer is placed between 304 SPMW and LSW in the eastern NASPG (García-Ibáñez et al., 2016), which represent the 305 product of mixing with waters coming from the south (i. e. Mediterranean Waters; MW).





334

307 water masses. In order to enhance the comprehension of the spatial distribution and trends of 308 the biogeochemical variables and to facilitate comparisons with previous studies along the 309 NASPG, the hydrographic characterization was simplified based on the following principles: 310 (1) the Iceland basin was not divided into its western and eastern parts and its longitudinal 311 span was delimited by the Reykjanes Ridge (29.5°W) and the Haton Bank (17°W), (2) upper 312 Labrador Sea Water (uLSW) was separated from deeper LSW (e. g. Stramma et al., 2004), (3) 313 the weak and spatially-limited influence of the return current and WNACW was removed by 314 considering the upper and intermediate layers of both the Irminger and Iceland basin fully 315 occupied by SPMW above uLSW, and (4) only the east branch of NACW (ENACW), placed 316 above SPMW, was contemplated for the upper Rockall Trough. 317 The whole water column was separated in layers delimited by potential density isopycnals at a reference pressure of 0 dbar following Azetsu-Scott et al. (2003), Kieke et al. (2007), Pérez 318 et al. (2008) and Yashayaev et al. (2008). The vertically distributed water masses separated in 319 320 density layers is represented for the entire section in Figure 1b. The vertical characterization in density layers allows to consistently compare the low-variable physical and chemical 321 322 properties within each water mass, enabling to assume linearity in the ocean CO<sub>2</sub> system. The 323 determination of the isopycnal limits between layers in the Irminger and Iceland basins 324 followed previous biogeochemical studies in the western boundary of the North Atlantic 325 (Fontela et al., 2020; García-Ibáñez et al., 2016; Pérez et al., 2010, 2008; Vázquez-Rodríguez et al., 2012a). The surface-to-bottom distribution of the main water masses in these basins 326 (with their respective  $\sigma_0$  lower limits shown in brackets) was SPMW (27.68 kg m<sup>-3</sup>), uLSW 327 328 (27.76 kg m<sup>-3</sup>), LSW (27.81 kg m<sup>-3</sup>) and ISOW (27.88 kg m<sup>-3</sup>). The low temperature and 329 salinity DSOW were considered at the bottom of the westernmost part of the Irminger basin. 330 The hydrography of the Rockall Trough has been characterized in previous studies in the 331 Northeast Atlantic (e. g. Ellett et al., 1986; Harvey, 1982; McGrath et al., 2012a, 2012b; 332 Holliday et al., 2000). The considered surface-to-bottom distribution of the main water masses was ENACW (27.35 kg m<sup>-3</sup>), SPMW (27.68 kg m<sup>-3</sup>) and LSW (bottom). 333

The physical and biogeochemical interannual changes were analysed in the main basins and

#### 2.2.5. pH<sub>T</sub> trends deconvolution





OA trends arise due to the combined variations in T, S,  $C_T$  and  $A_T$ . The influence of each driver on OA was analysed by assuming linearity and employing a first-order Taylor-series deconvolution to evaluate the pH<sub>T</sub> trends (Fröb et al., 2019; García-Ibáñez et al., 2016; Pérez et al., 2021; Takahashi et al., 1993; Tjiputra et al., 2014). Partial derivatives of pH<sub>T</sub> versus T, S, C<sub>T</sub> and A<sub>T</sub> were calculated based on mean properties of each layer by using the most recent equation (Eq. 2) given by Pérez et al., (2021). This equation introduced salinity-normalized C<sub>T</sub> and  $A_T$  ( $NX_T = X_T/S*35$ ) to remove the effect of the freshwater fluxes in the variation of  $A_T$ and  $C_{\rm T}$ .

$$\frac{dpH_T}{dt} = \frac{\partial pH_T}{\partial T}\frac{dT}{dt} + \left(\frac{\partial pH_T}{\partial S} + \frac{NC_T}{S_0}\frac{\partial pH_T}{\partial C_T} + \frac{NA_T}{S_0}\frac{\partial pH_T}{\partial A_T}\right)\frac{dS}{dt} + \frac{S}{S_0}\frac{\partial pH_T}{\partial C_T}\frac{dNC_T}{dt} + \frac{S}{S_0}\frac{\partial pH_T}{\partial A_T}\frac{dNA_T}{dt}$$
(2)

It is important to remark that the changes in  $NA_T$  and  $NC_T$  are linked with biogeochemical processes which have different influences: the processes involved in the organic carbon pump contribute to strongly change the  $NC_T$  weakly affecting the  $NA_T$ , while those involved in the carbonate pump affect the  $NA_T$  twice as much as  $NC_T$ . The complexity and heterogeneity of the processes that govern the pH<sub>T</sub> change were considered by this equation.

# 2.2.6. Calculation of the state of saturation of Calcite ( $\Omega_{Ca}$ ) and Aragonite ( $\Omega_{Arag}$ ): trends and drivers

The adverse impacts of OA on marine calcification processes and its correlation with the saturation states of Calcite ( $\Omega_{Ca}$ ) and Aragonite ( $\Omega_{Arag}$ ) has been commonly demonstrated (e. g. Gattuso et al., 2015; Langdon et al., 2000; Pörtner et al., 2004, 2019; Riebesell et al., 2000). The  $\Omega_{Ca}$  and  $\Omega_{Arag}$  were calculated as the product of the ion concentrations of calcium ([Ca<sup>2+</sup>]) and carbonate ([CO<sub>3</sub><sup>2-</sup>]) divided by the stoichiometry solubility products (K<sub>sp</sub>) for calcite (K<sub>Ca</sub>) and aragonite (K<sub>Arag</sub>) given by Mucci (1983) (Eq. 3 and 4). The  $\Omega_{Ca}$  and  $\Omega_{Arag}$  were calculated with the CO2SYS programme (Lewis and Wallace, 1998) for MATLAB (van Heuven et al., 2011; Orr et al., 2018; Sharp et al., 2023), applying the set of constants detailed in section 2.2.2.

$$\Omega_{Ca} = \frac{[Ca^{2+}][CO_3^{2-}]}{K_{Ca}}$$
 (3)

$$\Omega_{Arag} = \frac{[Ca^{2+}][CO_3^{2-}]}{K_{Arag}}$$
 (4)



363

364

365

366367

369

370



The collective temporal changes in the physico-chemical properties governing the OA influenced the  $\Omega_{Ca}$  and  $\Omega_{Arag}$  variations and were considered in this study. The influence of the potential drivers was analysed by employing a first-order Taylor-series deconvolution to evaluate the  $\Omega_{Ca}$  and  $\Omega_{Arag}$  trends, as done with the pH<sub>T</sub> (section 2.2.5). Likewise, the interannual changes of  $\Omega_{Ca}$  and  $\Omega_{Arag}$  were assumed linear and given by the sum of the partial derivates of  $\Omega_{Ca}$  and  $\Omega_{Arag}$  versus each driver (García-Ibáñez et al., 2021) in Eq. 5.

$$\frac{d\Omega}{dt} = \frac{\partial\Omega}{\partial T}\frac{dT}{dt} + \left(\frac{\partial\Omega}{\partial S} + \frac{NC_T}{S_0}\frac{\partial\Omega}{\partial C_T} + \frac{NA_T}{S_0}\frac{\partial\Omega}{\partial A_T}\right)\frac{dS}{dt} + \frac{S}{S_0}\frac{\partial\Omega}{\partial C_T}\frac{dNC_T}{dt} + \frac{S}{S_0}\frac{\partial\Omega}{\partial A_T}\frac{dNA_T}{dt}$$
 (5)

3. Results

# 3.1. Physicochemical characterization of the water column

371 The vertical distribution of the physical and biogeochemical variables is depicted in Figures 2, 372 3, S2 and S3. These figures exhibited the changes in the water-column properties throughout 373 the section between 2009 and 2016. The subsurface layers were characterized by warmer and 374 saltier waters than intermediate and deep layers among the three basins (Figure 2a and 2b). A 375 West-to-East increase in temperature and salinity throughout the water column was observed 376 in all the cruises. The temperature and salinity signals were highest in the Rockall Trough (4.5-11.0°C and 35.0-35.4, respectively), followed by the Iceland basin (3.0-7.5°C and 34.9-35.2, 377 respectively) and the Irminger basin (1.5-6.5°C and 34.8-35.1, respectively). The longitudinal 378 379 differences in temperature were more remarkable toward the upper layers through the SPMW 380 and uLSW. 381 The spatial variability in the physical properties introduced heterogeneities in the distribution 382 of the  $CO_2$  system variables. The  $A_T$  show a well-correlated direct relationship with salinity throughout the section (r<sup>2</sup>=0.89), with lower and vertically-homogenized average values in the 383 Irminger basin (2302.8-2307.3 µmol kg<sup>-1</sup> in subsurface waters and 2298.8-2301.0 µmol kg<sup>-1</sup> 384 in bottom waters) and Iceland basin (2308.7-2315.0 µmol kg<sup>-1</sup> in subsurface waters and 385 2305.2-2308.0 µmol kg<sup>-1</sup> in bottom waters) compared to the Rockall Trough (2317.9-2329.1 386 μmol kg<sup>-1</sup> in subsurface waters and 2308.5-2310.9 μmol kg<sup>-1</sup> in bottom waters). The upper 387 layers were characterized by low C<sub>T</sub> values (2153.7-2160.8 µmol kg<sup>-1</sup> at the Irminger basin, 388 2158.1-2168.4 µmol kg<sup>-1</sup> at the Iceland basin and 2120.1-2131.0 µmol kg<sup>-1</sup> at the Rockall 389 Trough), while a rapidly increment with depth was found below 100-200 m depth (2154.7-390





2171.2  $\mu$ mol kg<sup>-1</sup> throughout the section). The notable difference in the distribution of  $A_T$  and 391  $C_{\rm T}$  (Figure 2c and 3a, respectively) compared to those of  $NA_{\rm T}$  and  $NC_{\rm T}$  (Figure S2) elucidated 392 the remarkable significance of freshwater fluxes on the carbon variables fluctuations during 393 394 the period of study. The entrance of  $C_{\text{ant}}$  through the atmosphere-seawater interface caused higher  $C_{\text{ant}}$  values in the upper layers (higher than 50  $\mu$ mol kg<sup>-1</sup> in the first 1000 m depth; 395 Figure 3b). The natural component of the  $C_T$  ( $C_{nat}=C_T-C_{ant}$ ; Figure 3c) correlated with  $C_T$ 396 (r<sup>2</sup>=0.87), and show a distribution characterized by low surface (<2110 µmol kg<sup>-1</sup>) and high 397 bottom concentrations (>2130 µmol kg<sup>-1</sup>). 398 399 The pH<sub>T</sub> (Figure 2d) rapidly decreased with depth showing the effect of biological uptake in the upper layers and remineralization in deeper areas. The subsurface layer up to 100-200 m 400 depth exhibited pH<sub>T</sub> values higher than 8.025 units, which fell to 7.975 units at the bottom 401 402 layers. The pH<sub>T</sub> profiles reported an intrusion of remineralized and poorly oxygenated water between 500 and 1000 m depth with relatively low pH<sub>T</sub> (<7.975) compared to adjacent layers 403 in the Iceland basin and in the western part of the Rockall Trough. This thermocline layer was 404 405 previously observed at ~500 m depth by García-Ibáñez et al., (2016) along a more meridional 406 transect which crossed the Iceland basin northwest-southeast. It introduces differences in the 407 intermediate water masses between the Iceland and Rockall basins with the Irminger basin. 408 The spatial and interannual fluctuations in the ventilation rates through changes in the water 409 mass formation and respiration processes represent a source of variability in the 410 biogeochemical patterns. The apparent oxygen utilization (AOU), defined as the difference between saturated oxygen (calculated following Benson and Krause, 1984) and measured 411 oxygen, was used to assess the ventilation of the water masses (Figure 2e). The high AOU 412 413 values indicate low ventilation, while low AOU values indicate the opposite. The slow renewal of waters with high AOU favour the accumulation of the product of remineralization (de la 414 415 Paz et al. 2017). Thus, the areas with higher AOU (Figure 2e) were found to have high concentration of  $C_T$  and low pH<sub>T</sub> (Figures 3a and 2d, respectively). The near surface waters 416 permanently in contact with the atmosphere exhibited the lowest AOU values (<20 µmol kg<sup>-</sup> 417 418 1). The Irminger Basin presents the most significant water column ventilation among the entire section, with maximum AOU ranging from 35 to 50 µmol kg<sup>-1</sup> at the LSW and ISOW and the 419 remarkable intrusion of oxygenated DSOW (>260 µmol kg<sup>-1</sup> DO) over the continental slope 420

427

428 429

430 431

432 433

434 435

436

437

438

439 440

441

442

443

444 445

446 447

448

449

450





with AOU ranging from 30 to 40 μmol kg<sup>-1</sup>. The intermediate and deep layers of the Iceland 421 and Rockall basins were less ventilated, with AOU values higher than 45-50 µmol kg<sup>-1</sup>. The 422 thermocline layer placed between 500 and 1000 m depth along these two basins presented the 423 424 highest maximum AOU throughout the period (>60 µmol kg<sup>-1</sup>). The stagnation of these waters corresponds with the high  $C_T$  and low pH<sub>T</sub> (Figures 3a and 2d, respectively) encountered at 425

intermediate depths and should be considered in its temporal evolution.

#### 3.2. Temporal evolution of the physicochemical properties

The interannual trend in the distribution of the physicochemical properties was analysed through the whole water column across the Irminger, Iceland and Rockall basins by yearly averaging the variables for each layer, following previous studies in the NASPG (e.g. Fontela et al., 2020; García-Ibáñez et al., 2016) and applying linear regressions, where the ratios of interannual change were given by the values of the slopes. The temporal distribution and trends of the average physicochemical properties (Figures 4, 5, 6, S4, S5 and S6) revealed remarkable heterogeneities in their interannual evolution within the period 2009-2019 among the different basins and water masses. The mean properties were represented with error bars that are two times the error of the mean  $(2\sigma = 2 * (Standard Deviation / \sqrt{n}))$ , where n is the number of bottle samples in each layer and cruise. The interannual ratios are presented along with their respective standard error of estimate and correlation factors (r<sup>2</sup> and p-value) in Table 3 and S4. The standard errors of the slopes were calculated by considering the standard error of the annual mean values. The p-values  $\leq 0.01$  indicated that the trends were statistically significant at the 99% confidence level, the p-values  $\leq 0.05$  indicated that the trends were statistically significant at the 95% confidence level and the p-values  $\leq 0.1$  indicated that the trends were statistically significant at the 90% level. Trends with p-values > 0.1 were considered as not statistically significant but provided an estimation of the temporal evolution of the variables in their respective layers. These not statistically significant trends were explained by the high variability and changes in the low-limit depth of the layers encountered between consecutive years. As there was a lack of in situ measurements and sampling along the west half of the Irminger basin (36.5-42.5°W) in the cruise of 2019 (due to permit restrictions to study the national

waters of Denmark), the GO-SHIP A25-OVIDE data for the cruise of 2018 (available at



460



451 SEANOE [https://www.seanoe.org/], Pascale et al., 2022) were considered to adjust the 2019 452 data. The average values were calculated with both the available data in the easternmost part of the Irminger basin during the cruise of 2019 and the A25-OVIDE-2018 data available in 453 454 the same part of the section (29.6-36.5°W). The difference between these average values 455 provides the variation of each variable from 2018 to 2019, which can be extrapolated to the 456 western part of the Irminger basin by assuming linearity in the temporal evolution. Thus, the 457 average values for 2019 were adjusted by applying the product with the calculated change between 2018 and 2019. 458

#### 4. Discussion

# 4.1. Reversal of the physical trends during the 2010s

461 The present investigation revealed the cooling and freshening of the upper ocean in the 462 NASPG within the period 2009-2019 (Figure 4; Table 2), as recently reported since the reversal 463 of climatic trend and surface physical properties occurring after 2005 (Holliday et al., 2020; Josey et al., 2018; Robson et al., 2016; Tesdal et al., 2018). The temperature decreased in the 464 465 upper ocean (with more than 95% level of confidence in SPMW, while non statistically significant in ENACW) by 0.05-0.08 °C yr<sup>-1</sup> (Table 2), which is consistent with the ratio of 466 heat loss per decade among the first 700 m depth equivalent to approximately -0.45 °C decade 467 468 <sup>1</sup> (-0.045 °C yr<sup>-1</sup>) encountered over the period 2005-2014 (Robson et al., 2016). The interannual 469 temperature trends in subsurface layers (Table 2) similarly draw the cooling observed in the Irminger basin between 2008 and 2017 (-0.05 and -0.11 °C yr<sup>-1</sup> for summer and winter, 470 respectively; Leseurre et al., 2020) and the winter average surface cooling along the entire 471 472 NASPG between 2004 and 2017 (-0.08  $\pm$  0.02 °C yr<sup>-1</sup>; Fröb et al., 2019). The decrement in subsurface salinity (with more than 95% level of confidence in both SPMW and ENACW) of 473 0.006-0.018 yr<sup>-1</sup> (Table 2) agreed with the interannual rates provided by Tesdal et al., (2018) 474 for the Irminger basin  $(-0.007 \pm 0.002 \text{ yr}^{-1})$  and for the central-eastern NASPG  $(-0.020 \pm 0.003 \text{ m}^{-1})$ 475 yr<sup>-1</sup>) over the period 2004-2015. 476 477 The fluctuations in physical properties were linked to a decrease in oceanic heat transport and 478 storage within the NASPG, which has been attributed to changes in the AMOC over decadal to multidecadal timescales (Balmaseda et al., 2007; Desbruyères et al., 2013; Mercier et al., 479 480 2015; Smeed et al., 2018). However, the assessment of the temporal evolution of the AMOC





481 in high latitudes remains uncertain, and there is no evidence of its impact on physical patterns 482 across the NASPG on an interannual scale (Jackson et al., 2022). The changes in the atmospherics forcing also account for the variability of the upper ocean physical properties 483 and can have a cumulative effect over several years (Balmaseda et al., 2007; Böning et al., 484 485 2006; Eden and Willebrand, 2001; Marsh et al., 2005). 486 The distribution of the water mass properties, the processes of vertical and horizontal mixing 487 and the circulation patterns in the Irminger and Iceland basins were described by García-Ibáñez 488 et al., 2016 and 2018. The poleward path of the ENACW (Pollard et al., 1996) and its mixing with waters moving from the west across the NASPG (Ellett et al., 1986) accounted for the 489 490 highest subsurface temperature and salinity signals observed in the Iceland basin and even more in the Rockall. The SPMW and LSW in the Rockall Trough exhibited higher temperature 491 492 and salinity signals in the respectively order of  $\sim 1^{\circ}$ C and  $\sim 0.05-0.1$  compared to the Irminger 493 and Iceland basins (Figure 4). The NASPG circulation patterns account for these differences 494 by transporting eastward these water masses, which subduct below the ENACW in the Rockall 495 Trough and mixed with warmer and more saline intermediate waters (i.e. Mediterranean 496 Water) moving from the south (e. g. Ellett et al., 1986; Harvey, 1982; Holliday et al., 2000). 497 The low temperature and salinity signals in the less-stratified Irminger basin (Figure 2) 498 experienced weaker interannual decreases in subsurface layers and higher rates of cooling and 499 freshening in intermediate and deep waters compared with the Iceland and Rockall basins 500 (Figure 4; Table 2). These longitudinal thermohaline heterogeneities were related to the 501 enhancement of vertical mixing processes in areas of water mass formation along the western 502 NASPG (Fröb et al., 2016; García-Ibáñez et al., 2015; Pickart et al., 2003; Piron et al., 2017) 503 and the water mass transformation along the NAC (Brambilla and Talley, 2008). The strongest 504 decrement in subsurface temperature and salinity along the Iceland and Rockall basins (Figure 505 4; Table 2) coincided with the significant event of heat loss and freshening observed by Holliday et al., (2020) in the eastern NASPG over the period 2012-2016, so-called the Great 506 507 Salinity Anomaly. This pattern was not easily discernible in the Irminger basin due to the 508 transport of freshwater through the Fram Strait, as well as due the redirection of the Labrador Current combined with changing wind stress curl (Holliday et al., 2020). 509





4.2. 510 Evaluation of the interannual trends in C<sub>T</sub> in response to changes in C<sub>ant</sub> and  $C_{nat}$ 511 512 The changes in the physical patterns observed in the NASPG influenced the interannual 513 variability of the MCS. The increase in  $C_T$  expected in the upper ocean due to the atmospheric CO<sub>2</sub> uptake was offset by the cooling and freshening (and dealkalinization) of the subsurface 514 515 layers in the entire NASPG. The entrance of C<sub>ant</sub> through the air-sea interface and its 516 accumulation dominated the observed increase in  $C_{\rm T}$ , while the  $C_{\rm nat}$  experienced a slightly 517 decrease throughout the region (Figure 5 and Table 2). A detailed description of the interannual trends in C<sub>T</sub> and A<sub>T</sub> is provided in Appendix B. 518 519 The increase in the ventilation rates during this decade, shown by the negative AOU trends 520 (Figure S6 and Table S4), explained the higher growth in Cant than expected due to the 521 atmospheric CO<sub>2</sub> increase. It leads an enhancement in the vertical mixing processes which 522 drove the transport of  $C_T$ -rich subsurface waters toward deeper areas and the slightly decrease in  $C_{\text{nat}}$  through the whole water column. The trends of  $C_{\text{ant}}$  among the 2010s (0.85-1.77 µmol 523 kg<sup>-1</sup> yr<sup>-1</sup>; statistically significant at the 99% level) were higher than the observed on a decadal 524 to multidecadal scale since the late 20<sup>th</sup> century in the Irminger and Iceland basins (0.21-0.89 525 μmol kg<sup>-1</sup> yr<sup>-1</sup> during 1991-2015,García-Ibáñez et al., 2016; and 0.38-1.15 μmol kg<sup>-1</sup> yr<sup>-1</sup> 526 527 during 1983-2013, Pérez et al., 2021), which suggest an enhancement in the Cant accumulation 528 on interannual scales during periods of high ventilation, as previously reported by Perez et al., 529 (2008).530 The vertical distribution of  $C_{\text{ant}}$  and  $C_{\text{nat}}$  along the transect (Figure 3b and 3c) reflect the higher 531 stratification in the Iceland and Rockall basin compared with the well vertically-mixed 532 Irminger basin. It represents a source of variability in the interannual changes of  $C_{\text{ant}}$  among 533 the different layers and basins (Figure 4; Table 2). In the western NASPG, the surface heat loss 534 and enhanced deep convection processes favour the solubility and subsequent uptake of 535 atmospheric CO<sub>2</sub> and inject oxygenated and CO<sub>2</sub>-rich waters into deeper layers (Messias et al., 536 2008). Its likely accounts for intermediate and deep layers in the Irminger basin exhibiting the highest Cant accumulation rates in the NASPG (Figure 5; Table 2). The highest ventilation of 537 538 the interior ocean in the Irminger basin was demonstrated by its minimum AOU values (Figure 539 2 and S6). It induced a rapid surface-to-bottom transport of Cant shown by its highest rates of





540 increase in intermediate and deep waters throughout the region (Figure 5; Table 2). The high Cant values and its rapidly increment at DSOW were explained by the improved oxygenation 541 of this layer at shallower depths (interannual AOU trends given in Table S4) and its subduction 542 543 through the continental slope below ISOW. 544 In the eastern NASPG, the stratification weakened due to the path of the NAC warming eastward the upper water column and accounted to slowdown the increase in Cant in the Iceland 545 basin. An exception comes with the Rockall basin, in which the relatively warm and salty 546 ENACW (Figure 2 and 4) showed the maximum  $C_{ant}$  (58-68 µmol kg<sup>-1</sup>) and minimum  $C_T$ 547 (2120-2131  $\mu$ mol kg<sup>-1</sup>) and  $C_{nat}$  (2058-2070  $\mu$ mol kg<sup>-1</sup>) throughout the region (Figure 3 and 548 5). The strong stratification of the Rockall Trough due to the wide differences in the physical 549 properties between the ENACW with SPMW and LSW plays a crucial role. The lower AOU 550 encountered in ENACW (<20 µmol kg<sup>-1</sup>) compared with deeper layers (>30 µmol kg<sup>-1</sup>) suggest 551 552 that the enhanced ventilation processes were limited to the subsurface layer increasing the entrance of C<sub>ant</sub> through the air-sea interface. The strong oxygenation, which reach the oxygen 553 saturation after 2014, could be related with the high rates of renovation of ENACW due to its 554 path from the south (Pollard et al., 1996) and its mixing with waters moving eastward (Ellett 555 et al., 1986). As the NAC transports nutrient-rich waters northward and eastward into 556 557 subsurface layers in the Rockall Trough, biological production tends to increase and actively reduced the CO<sub>2</sub> excess from the ENACW (McGrath et al., 2012b), as proved by the observed 558 559 low C<sub>T</sub> and C<sub>nat</sub>. The strong interannual increase in the ENACW ventilation during this decade increase the  $C_{\rm ant}$  and decrease the  $C_{\rm nat}$  (Rodgers et al., 2009) keeping approximately constant 560 the  $C_{\rm T}$  (Table 3). The poorly ventilated thermocline (AOU > 60  $\mu$ mol kg<sup>-1</sup>), placed between 561 562 500-1000 m in the eastern NASPG, induced a  $C_{\text{nat}}$ -driven increase in  $C_T$  among the SPMW 563 and uLSW. However, its intrusion does not present relevant variations with time and thus does not introduce differences in the interannual trends of the biogeochemical properties. 564

# 4.3. Acidification trends

565

566

567

568

569

The interannual pH<sub>T</sub> trends (Figure 6, Table 2) exhibited the acidification of the whole water column in NASPG during the period 2009-2019. Despite the acidification rates observed in the most subsurface waters among the three basins were not significant at the 90% confidence level (Table 2), they were consistent in the interval of 0.001 units yr<sup>-1</sup> to those observed during





 $(0.0018 \pm 0.0002 \text{ units yr}^{-1} \text{ during } 1995-2014 \text{ and } 0.0020 \pm 0.0001 \text{ units yr}^{-1} \text{ during } 1995-2023$ 571 at ESTOC, González-Dávila and Santana-Casiano, 2023; and 0.0017 ± 0.0001 units yr<sup>-1</sup> during 572 1983-2014 at BATS, Bates et al., 2014) and subpolar latitudes (-0.0017 $\pm$  0.0002 units vr<sup>-1</sup> at 573 IRM-TS during 1983-2013 and  $-0.0026 \pm 0.0002$  units yr<sup>-1</sup> at IS-TS during 1985-2013, 574 575 summarized by Pérez et al., 2021). In addition, the changes in the surface pH<sub>T</sub> trends has been 576 reported by Leseurre et al., (2020) in the western NASPG within a wide latitudinal area (54-64°N) during the period 2008-2017 in comparison with the periods 1993-1997 and 2001-2007. 577 578 Although the highly significant cooling observed in SPMW, the year-to-year variations in 579 ventilation (shown by the annual average AOU and its trends in Figure S6) and thus in C<sub>nat</sub> and C<sub>ant</sub> (Figure 5), which could be related with fluctuations in the atmospheric forcing, introduced 580 relevant changes in pH<sub>T</sub> on an interannual scale and explained the low significant trends. This 581 582 behaviour was clearly reflected in the Irminger basin, where strong slowdowns in ventilation 583 were observed from 2009 to 2010 and from 2013 to 2014, resulted in a relatively increase in C<sub>nat</sub> and decrease in C<sub>ant</sub> observed in SPMW and extended with less intensity through the whole 584 585 water column. 586 The highest acidification rates were found through intermediate and deep waters in the Irminger and Iceland basins, coinciding with the highest rates of increase in  $C_{\rm ant}$  (Table 2, 587 trends statistically significant at more than 95% level of confidence). The exception comes 588 589 with the DSOW, which presented and interannual decrease in pH<sub>T</sub> in phase with those of the uLSW. This singularity was previously observed by García-Ibáñez et al., (2016), which noticed 590 591 the similar trends between the DSOW and LSW attributed to the recently formation and sink 592 through the continental slope of the DSOW. The acidification rates found among the uLSW, LSW and ISOW (0.0026-0.0032 units yr<sup>-1</sup>) experienced, on an interannual scale, an 593 acceleration in comparison with previous reported based on long-term records [e. g. 0.0009-594 0.0017 units yr<sup>-1</sup> estimated for 1981-2008 by Vázquez-Rodríguez et al., (2012b); 0.0013-595 596 0.0016 units yr<sup>-1</sup> estimated for 1991-2015 by García-Ibáñez et al., (2016); 0.0015-0.0019 units  $yr^{-1}$  estimated for 1983-2013 at the IRM-TS by Pérez et al., (2021); 0.0019 ± 0.0001 units yr-597 598 1 estimated for 1993-2017 by Leseurre et al., (2020)]. Contrasting the rates of change in pH<sub>T</sub> 599 during the decade of study with those encountered by these multidecadal evaluations (and 600 considering the total amount of years comprising each of the studies and the changes in the ion

larger periods at time-series stations located across the North Atlantic: at subtropical latitudes





601 hydrogen concentration- $[H_T^+]$ ), we estimate an acceleration in the rates of acidification of 0.4-602 5.4% in the Irminger basin and 1.0-9.0% in the Iceland basin during the 2010s since the late 20th century. This acceleration was mainly attributed to increased deep-water ventilation 603 604 (shown in the rapid decrease in AOU in Figure S6) favouring the progressively increase in the accumulation of Cant and Cnat toward intermediate a deep layers, in which cooling was not 605 606 significant in the Irminger basin and neither enough intense in both basins to compensate the 607 acidification. 608 Although the similarities encountered in the pH<sub>T</sub> trends among both basins, the average values 609 presented differences which may be closely linked with the transport and transformations of the water masses along the NASPG and mainly modulated by the Reykjanes Ridge (García-610 Ibáñez et al., 2015, 2016, 2018). The transformation of the SPMW formed in the Iceland 611 612 (McCartney and Talley, 1982; Brambilla and Talley, 2008; Tsuchiya et al., 1992; Van Aken and 613 Becker, 1996) and flowing with the NAC across the Reykjanes Ridge (Brambilla and Talley, 614 2008) accounted for the lower pH<sub>T</sub> values in the Irminger basin. The differences in pH<sub>T</sub> found 615 at intermediate and deep layers were related with the divergence of the LSW path into two 616 cores when it reaches the Reykjanes Ridge (Álvarez et al., 2004; Pickart et al., 2003) and the 617 ISOW path flowing southward along the western Iceland basin and recirculated northward into 618 the eastern Irminger basin (Dickson and Brown, 1994; Saunders, 2001). These differences in 619 the spreading of water masses enhanced the ventilation in the Irminger basin favouring the fall 620 in pH<sub>T</sub> compared with the Iceland basin. The rise in the ISOW following the Reykjanes Ridge slope through its eastern flank favoured a strong vertical mixing over and around the ridge 621 622 (Ferron et al., 2014) and a reduction of the LSW core in the Iceland basin (García-Ibáñez et 623 al., 2015), contributing to resemble pH<sub>T</sub> values and trends among the uLSW and LSW in this 624 basin. 625 The upper waters of the Rockall Trough presented the maximum pH<sub>T</sub> throughout the transect (8.02-8.08 units). The observed strong pH<sub>T</sub> fluctuations between years related with interannual 626 627 changes in the NAC do not allow to discern trends with a statistically interval of confidence 628 equal or higher than the 90% on a decadal scale. The interannual decrease in pH<sub>T</sub> in the ENACW (~0.001 units yr<sup>-1</sup>) was half than the observed along southernmost transects in the 629 Rockall Trough between 1991 and 2010 (~0.002 units yr<sup>-1</sup>, McGrath et al., 2012a). The 630



632 633

634 635

636

637

638 639

640

641

642

643 644

645

646

647 648

649

650

651

652

653

654

655

656

657

658

659

660



temporal distribution of the average pH<sub>T</sub> (Figure 6) highly influenced by changes in the ventilation (seen in AOU trends in Figure S6) allow to discern two periods: the approximately constant ventilation rates keep a steady state in terms of pH<sub>T</sub> during 2009-2011, while the progressively renewal of oxygenated water after 2012 (and peaking in this year) increase the pH<sub>T</sub>. The year-to-year variability in the biogeochemical patterns after 2012 may be attributed to the fluctuations in the spreading into the Rockall Trough of several water masses occupying different depths coming from the south and east (Ellett et al., 1986; Pollard et al., 1996). This contributed to enhance the oxygenation of the ENACW during the 2010s (seen in minimum AOU values highly variables between years and which tend to decrease with 99% statistical confidence; Figure S6 and Table S4) and the reduction of the injection of saline subsurface waters from subtropical latitudes (Holiday et al. 2020). The findings suggest that the strong decrease in  $A_T$  (Figure S4 and Table S4) due to the freshening and weak increase in  $C_T$  (Figure 5 and Table 2) due to enhanced ventilation counteract the acidification in the ENACW. The SPMW among the Iceland and Rockall basins showed similar pH<sub>T</sub> trends (Table 2) due to the emplacement of the poorly-oxygenated thermocline at these depths (García-Ibáñez et al., 2016). The approximately constant AOU at SPMW in the eastern NASPG (Figure S6) proved its steady ventilation, which can introduce differences in the acidification rates among the layers accomplishing the Rockall Trough. The influence of the cooling and freshening of deeper areas due to the spreading and horizontal mixing was notable in the LSW, which presented slightly higher pH<sub>T</sub> values in the Rockall respect to the adjacent Iceland basin.

# 4.4. Drivers pH

Due to the variety of processes involved in OA, a decomposition of the pH<sub>T</sub> trends into the individual components that govern its spatio-temporal variability was done (see section 2.2.5). The interannual pH<sub>T</sub> changes  $(\frac{dpH_T}{dt})$  explained by fluctuations in temperature  $(\frac{\partial pH_T}{\partial T} \frac{\partial T}{dt})$ , salinity  $(\frac{\partial pH_T}{\partial S} \frac{\partial S}{dt})$ , A<sub>T</sub>  $(\frac{\partial pH_T}{\partial A_T} \frac{\partial A_T}{dt})$  and C<sub>T</sub>  $(\frac{\partial pH_T}{\partial C_T} \frac{\partial C_T}{dt})$  were calculated for each layer and basin (Eq. 2) and summarized in Table 3. The positive contributions of each of the drivers indicate an increase in pH<sub>T</sub> while negative contributions the opposite. The cumulative pH<sub>T</sub> change resulting from the distinct drivers  $(\frac{dpH_T}{dt})$  (calculated) in Table 3) were consistent with the observed pH<sub>T</sub> trends  $(\frac{dpH_T}{dt})$  (obs) in Table 3, discussed in section 4.2), thereby instilling confidence in the methodology. The minimal differences between observed and calculated rates of change have





662

663

664

665

666

667

668

669

670

671

672673

674

675 676

677

678 679

680

681

682

683

684 685

686

687 688

689 690

added coherence to the non-significant trends identified for pH and its drivers in some basins and layers (Table 2, 3 and S4). In the entire section at SPMW, the  $\frac{dpH_T}{dt}$  (calculated), explained by the cumulative impact of its drivers (all of them statistically significant at the 95% level of confidence), aligns within a range of <0.0002 units yr-1 with  $\frac{dpH_T}{dt}$  (obs) (which was not significant). In the Irminger and Iceland basins at intermediate and deep layers, the  $\frac{dpH_T}{dt}$  (obs) (statistically significant at least at the 95% level of confidence) were consistent within the range of <0.001 units yr-1 with  $\frac{dpH_T}{dt}$  (calculated) (T, S and NA<sub>T</sub> shows non-significant trends at some of the intermediate and deep layers). The interannual variations were non-significant for pH<sub>T</sub> neither for its drivers in the Rockall Trough at LSW and ENACW. The high temporal dispersion of average data in these layers was mainly related to the rise in depth of LSW along the eastern continental slope and its mixing with shallower waters coming from subtropical latitudes (Ellett et al., 1986; Harvey, 1982; Holliday et al., 2000). The substantial variability introduced by these processes made it difficult to discern the pattern of acidification and its drivers on an interannual scale in the shallow Rockall Trough. Therefore, long-term monitoring and the development of multidecadal-scale studies are required in this area to derive significant conclusions. The cooling and freshening of the NASPG during the 2010s modified the physical-driven p $H_T$ changes compared with those encountered by García-Ibáñez et al., (2016) during previous decades in the western NASPG. The cooling contributed to increase the pH<sub>T</sub> and compensated the observed acidification rate. The increase in pH<sub>T</sub> due to temperature fluctuations was maximum at SPMW (~0.001 units yr<sup>-1</sup>) and was reduced an order of magnitude to negligible toward intermediate and deep layers (<0.0003 units yr<sup>-1</sup> at uLSW and below). The increase in pH<sub>T</sub> due to salinity fluctuations was minimal (<0.0001 units yr<sup>-1</sup>) through the whole water column in the three basins, reflecting that the observed freshening caused insignificant changes in pH<sub>T</sub>. The temperature and salinity contributed by 19.1-26.5% and 1.2-3.3%, respectively, in the total pH<sub>T</sub> change in the upper layers, while presented an influence three times lower in intermediate and deep layers (1.3-7.6% and <0.6%, respectively). The enhanced convective processes in the Irminger basin (e. g. Fröb et al., 2016; García-Ibáñez et al., 2015; Gladyshev et al., 2016a, 2016b; Piron et al., 2017) together with the rapid transport of LSW from the Labrador Sea to the Irminger basin (Yashayaev et al., 2007) introduced differences in the





691 thermal-driven pH<sub>T</sub> with the Iceland basin which has been previously reported by García-Ibáñez et al., (2016). The advection of LSW through the Greenland continental slope also 692 affected the DSOW (Read, 2000; Yashayaev and Dickson, 2008), which shows thermal-driven 693 pH<sub>T</sub> changes consistent with those encountered through the LSW in the Irminger basin. 694 695 Despite the negligible direct contribution of the salinity fluctuations over the pH<sub>T</sub> changes, the 696 freshwaters fluxes influence the distribution of  $A_T$  and  $C_T$  indirectly affecting pH<sub>T</sub> trends. Once 697 removed the effect of salinity by normalization (Pérez et al., 2021), the positive  $NA_{\rm T}$  trends 698 encountered in the upper layers lead a rise in pH<sub>T</sub>, while the diminished NA<sub>T</sub> contributed to decrease the pH<sub>T</sub> toward the interior ocean. The changes in  $NA_T$  described the 7.8-10.1% of 699 the total pH<sub>T</sub> change at SPMW. The NA<sub>T</sub>-driven pH<sub>T</sub> changes became insignificant with depth 700 701 (Table 3) due to the insignificantly interannual changes in NA<sub>T</sub> through LSW and ISOW (Table 702 S4). The weak contribution of  $NA_{\rm T}$  in these layers (1.3-5.1%) could be related to the difficulty 703 of reversing the large alkalinization until the 2000s resulted from the slowdown in the 704 formation of LSW since the mid-90s (Lazier et al., 2002; Yashayaev, 2007), which was 705 transmitted towards deeper overflow waters (Sarafanov et al., 2010). The substantial 706 interannual changes and the abrupt change between periods of increase and decrease of the 707 seawater properties at DSOW (Yashayaev et al., 2003; Stramma et al., 2004) linked with 708 changes in the LSW formation (Dickson et al., 2002) explained the rapidly decrease in  $NA_T$ 709 (Table S4), which described the 14.6% of the pH<sub>T</sub> declining. The increase in  $NC_T$  drove by the rise in  $C_{ant}$  was found to govern the acidification, with a 710 contribution higher than the 67% in the whole water column throughout the region. The  $NC_T$ -711 712 driven pH<sub>T</sub> declining was close to twice the observed and calculated acidification rates through 713 the SPMW (Table 3). However, the contribution of  $NC_T$  at SPMW (67-69%) was lower than the encountered toward the interior ocean (82-96%) due to the relevance of temperature and 714  $A_{\rm T}$  over pH<sub>T</sub> trends in the upper ocean. The cooling and increase in  $NA_{\rm T}$  counteracted the 715 acidification expected by the increasing  $C_T$  at SPMW by 28-34% and 11-15%, respectively. 716 717 The weaker cooling through the intermediate and deep layers leads a lower thermal-718 neutralization of the  $C_T$ -driven acidification (1.5-9.3%), while the decreasing  $NA_T$  contributed to decrease the pH<sub>T</sub> by < 2-12% in the uLSW, LSW and ISOW and by  $\sim 15\%$  in the DSOW. 719 720 The driver analysis also remarked that the role of freshening in counteract the acidification



724



was small in the upper layers (<6%) and becoming insignificant toward the interior ocean (<2%).

The analysis of the changes in  $\Omega_{Ca}$  and  $\Omega_{Arag}$  hold significance in elucidating the potential

# 4.5. Interannual changes in $\Omega_{Ca}$ and $\Omega_{Arag}$

725 effects of OA over the CaCO3 species calcite and aragonite, thereby offering insights into their 726 potential implications for marine calcifying organisms and ecosystems. The vertical 727 distribution of  $\Omega_{Ca}$  and  $\Omega_{Arag}$  is presented in Figure S3. The upper and intermediate layers up 728 to 2100-2400 m depth of the Irminger and Iceland and the whole Rockall basin were 729 supersaturated for aragonite ( $\Omega_{Arag} > 1$ ), while the DSOW was undersaturated ( $\Omega_{Arag} < 1$ ). The 730 ISOW, with  $\Omega_{Arag}$  ranged between 1.0 and 1.1 at the beginning of the decade, crossed to 731 undersaturated conditions at the end of the period due to the progressively rise of the aragonite 732 saturation horizon (depth in which  $\Omega_{Arag}=1$ ). The whole water column throughout the section 733 was supersaturated for calcite ( $\Omega_{Ca}$ >1) due to its lower solubility (Mucci, 1983). The  $\Omega_{Ca}$  and 734  $\Omega_{\text{Arag}}$  in the SPMW (2.2-2.7 and 1.4-1.7 units, respectively) were lower than the encountered 735 equatorward in the subsurface Atlantic (>4.0 and >2.5 units, respectively; González-Dávila et 736 al., 2010; González-Dávila and Santana-Casiano, 2023). The poleward pathway of low-737 latitude upper waters through the Rockall Trough explained the higher  $\Omega_{Ca}$  and  $\Omega_{Arag}$  found in 738 the ENACW (3.0-3.6 and 1.8-2.3 units, respectively). The reduction in  $\Omega_{Ca}$  and  $\Omega_{Arag}$  towards 739 higher latitudes in upper and intermediate layers smooth the vertical gradients in the NASPG 740 compared with the subtropical latitudes (González-Dávila et al., 2010; González-Dávila and 741 Santana-Casiano, 2023). 742 The interannual trends in  $\Omega_{Ca}$  and  $\Omega_{Arag}$  (Figure 7, Table 2) exhibited the decrement through 743 the whole water column along the NASPG with a level of statistical confidence generally 744 higher than the 90%. The rates of declining for  $\Omega_{Ca}$  and  $\Omega_{Arag}$  in the SPMW (0.011-0.021 and 0.007-0.013 units yr-1; respectively) were consistent with the trends observed up to 100 m 745 746 depth at ESTOC between 1995 and 2023 (0.019  $\pm$  0.001 and 0.012  $\pm$  0.001 units yr<sup>-1</sup>, 747 respectively; González-Dávila and Santana-Casiano, 2023) and in surface waters at the IS-TS between 1985 and 2008 (0.0117  $\pm$  0.0011 and 0.0072  $\pm$  0.0007 units yr<sup>-1</sup>, respectively; 748 Olafsson et al., 2009). The declining in  $\Omega_{Arag}$  in the SPMW accelerated by ~26% and ~51% in 749 750 the Irminger and Iceland basins, respectively, in comparison with the trends given for the



752

753 754

755

756

757

758

759

760

761

762

763

764

765

766

767

768 769

770

771

772

773

774

775

776 777

778

779

780

781



period 1991-2018 (0.0052 ± 0.0006 and 0.0049 ± 0.0015 units yr<sup>-1</sup>, respectively; García-Ibáñez et al., 2021). The observed decrease in  $\Omega_{Arag}$  in the SPMW was ~23% faster in the Rockall Trough than in the adjacent Iceland basin. The interannual declining for  $\Omega_{Ca}$  and  $\Omega_{Arag}$  in the ENACW (0.012 and 0.008 units yr<sup>-1</sup>, respectively) agreed with these previous observations but were not statistically significances likely due to the high variability modifying the changes in pH<sub>T</sub> in this layer (see section 4.2). Despite the acceleration of the acidification rates toward intermediate and deep layers, the declining rates weakened for  $\Omega_{Ca}$  and even more for  $\Omega_{Arag}$ (Table 2). Moreover, the vertical profiles were approximately constant throughout the section in contrast with the heterogeneous vertical distribution of pH<sub>T</sub> between basins. This behaviour was previously observed in the Irminger and Iceland basins by García-Ibáñez et al., (2021) and explained by pressure and temperature-induced changes in the speciation of the CO<sub>2</sub>-carbonate chemistry species (Jiang et al., 2015) and in the solubility of calcite and aragonite (Mucci, 1983). Their combined action counterbalanced the alterations in  $\Omega$  resulting from acidification, particularly in colder deep waters where the solubility of calcite and aragonite was reduced (García-Ibáñez et al., 2021). However, the fall down in  $\Omega_{Ca}$  and  $\Omega_{Arag}$  along the uLSW, LSW and ISOW accelerated by 40-75% in relation with the trends reported by García-Ibáñez et al., (2021) for the Irminger and Iceland basins. The LSW and ISOW presented faster declining rates for  $\Omega_{Ca}$  and  $\Omega_{Arag}$  in the Irminger (Table 2), which may be caused by the enhanced ventilation of the interior ocean which accelerated the acidification (see section 4.2). The westward rise in depth of these layers along the Greenland continental slope, accompanied by a subsequent elevation in the horizons of solubility, resulted in reduced buffering capacity against acidification effects in the Irminger basin when compared to the Iceland basin. In contrast, the rise in depth of LSW in the Rockall Trough favour the increment of ~0.2 units in  $\Omega_{\rm Ca}$  and  $\Omega$ Arag with respect to the Iceland basin but had not influence on the interannual trends, which were coinciding. The  $\Omega_{Ca}$  and  $\Omega_{Arag}$  in the DSOW, despite showed a trend accelerated by ~30% compared to the observed by García-Ibáñez et al., (2021), presented the weakest interannual decreases throughout the section  $(0.004 \pm 0.003)$  and  $0.002 \pm 0.001$  units yr<sup>-1</sup>, respectively) due to the high pressure and low temperatures compensating the rapidly acidification (Figure 6, Table 2). A driver analysis enabled the assessment of the impact of individual processes involved in OA on the variations in  $\Omega_{\text{Ca}}$  and  $\Omega_{\text{Arag}}$  (see section 2.2.6). The correlation of  $\Omega$  with pH<sub>T</sub> ( $r^2$ =0.90)





782 with a level of significance higher than the 99% explained that the individual components driving OA accompanied the declining in  $\Omega$ . The interannual  $\Omega$  variations  $(\frac{d\Omega}{dt})$  explained by 783 fluctuations in temperature  $(\frac{\partial \Omega}{\partial T} \frac{\partial T}{\partial t})$ , salinity  $(\frac{\partial \Omega}{\partial S} \frac{\partial S}{\partial t})$ ,  $A_T (\frac{\partial \Omega}{\partial A_T} \frac{\partial NA_T}{\partial t})$  and  $C_T (\frac{\partial \Omega}{\partial C_T} \frac{\partial NC_T}{\partial t})$  were calculated 784 785 for each layer and basin (Eq. 5) and summarized in Table 4. The sum of changes in  $\Omega$  due to the distinct drivers  $(\frac{d\Omega}{dt})$  (calculated) in Table 4) agreed with observed  $\Omega$  trends  $(\frac{d\Omega}{dt})$  (obs) in Table 786 4) in all the basin and layers except for the DSOW, in which the strong  $NA_T$  decrease had a 787 crucial influence on declining  $\Omega$ . The driver analysis, as mentioned when was applied for pH<sub>T</sub>, 788 789 contributes to add coherence and consistency to those non-significant trends identified and/or 790 its drivers in some basins and layers (Table 2, 3 and S4) The  $C_{\text{ant}}$ -driven rise in  $NC_T$  governed the decrease in  $\Omega$  with a contribution of 79-83% in the 791 SPMW which reached  $\sim 97\%$  toward deeper waters. The increase in  $NA_T$  in the SPMW 792 793 accounted by 10.4-13.0% in the  $\Omega$  trends and counteracted its NC<sub>T</sub>-driven decrease by 12.6-794 16.2%. The contribution of  $NA_T$  fall and reversed toward deeper waters, explained <6% of the 795 decline in  $\Omega$  in the uLSW, LSW and ISOW in the Irminger basin and <11% in the Iceland basin. The pronounced impact of the rapid decrease in  $NA_{\rm T}$  on the acidification of the DSOW 796 797 (see section 4.3) depicted the greater contribution of  $NA_{\rm T}$  encountered among the Irminger basin (16%) and compensated the  $C_T$ -driven decrease in  $\Omega$  by 36.4%. In the Rockall Trough, 798 the contribution of  $NC_T$  changes on  $\Omega$  was reduced at LSW (78.2-79.0%) compared to the 799 800 Irminger basin (94.5%) while the effect of  $NA_{\rm T}$  fluctuations tripled until reach 12.6-12.7%. 801 Despite the evaluated crucial role of cooling in counteracting the acidification, the temperature fluctuations have an opposite effect on  $\Omega$  owing to the thermodynamic relationship inherent in 802 803 the acid-base equilibrium of the CO<sub>2</sub>-carbonate system (Dickson and Millero, 1987). In the 804 Irminger and Iceland basins, the observed decreasing temperatures negligibly contributed to 805 fall down the  $\Omega$  (3.6% in the SPMW and <2% in intermediate and deep waters). The influence of salinity, as occurred with the pH<sub>T</sub> trends, was minimal: the observed freshening contributed 806 to elevate the  $\Omega$  trends and compensated its declining by 4.6-4.7% at SPMW, 1.1-2.1% at 807 uLSW and LSW and 0.5-1.2% at ISOW and DSOW. Even the slightly faster cooling and 808 freshening observed in the Rockall Trough, the contributions of temperature and salinity on 809 the  $\Omega$  did not exceed the 7% in each of its layers. 810





811 The driver analysis exhibited the strongest interannual decrease in  $\Omega$  in the upper layers governed by the uptake of  $C_{ant}$  weakly compensated by the increase in  $NA_T$  and favoured by 812 the cooling and freshening. The decrease in  $\Omega$  could have severe consequences on organisms 813 814 reliant on aragonite, which is less resistant to dissolution than calcite (Mucci, 1983; Broecker 815 and Peng, 1983) and thus expected to experience relatively higher susceptibility to the effects 816 of OA over shorter time scales (Raven et al., 2005). The progressive reduction in  $\Omega_{Arag}$  is 817 driving a long-term decrease in the depth of the aragonite saturation horizon ( $\Omega_{Arag}=1$ ) by 80-400 m since the preindustrial era (Álvarez et al., 2003; Feely et al., 2004; Pérez et al., 2013, 818 819 2018; Pérez et al., 2013; Tanhua et al., 2007; Wallace, 2001) and is projected to shoal by more 820 than 2000 m by the end of the century under the IS92a scenario (Orr et al., 2005). Likewise, 821 Orr et al., (2005) suggested that high-latitudes surface waters could become undersaturated 822 when the atmospheric CO<sub>2</sub> concentration double the preindustrial concentration within the 823 next 50 years. It would reduce the calcification rates in some shallow calcifying organism by 824 more than the 50% (Feely et al., 2004). 825 The planktonic aragonite-producers pteropods (e. g. Limacina helicina, Clio pyramidata), 826 which have high population densities in subpolar regions up 300 m depth (Bathmann et al., 827 1991; Urban-Rich et al., 2001) and play a key role in the export flux of both carbonate and 828 organic carbon (Accornero et al., 2003; Collier et al., 2000), are expected to be highly 829 vulnerable to OA if the aragonite saturation horizon continue to shoal (Orr et al., 2005). The 830 undersaturation toward intermediate and upper layers negatively influence the aragonite-based 831 CWC (e. g. Lophelia pertusa, Madrepora oculate), which show their highest diversity and 832 population along the NASPG between 200 and 1000 m depth among the global ocean (Roberts 833 et al., 2009). In fact, several studies reported that CWC ecosystems are anticipated to be among 834 the first deep-sea ecosystems to experience acidification threats (Guinotte et al., 2006; Maier 835 et al., 2009; Raven et al., 2005; Roberts et al., 2009; Turley et al., 2007), particularly in the 836 North Atlantic (Perez et al., 2018). The findings presented here contribute to a deeper 837 understanding of the biological impacts of OA along the NASPG.

# 5. Conclusions

838

839

840

This research has evaluated the interannual changes in the basin-wide CO<sub>2</sub>-carbonate system dynamics along the NASPG during the 2010s. Despite the observational period is relatively





842 has allowed to evaluate the ocean response, in terms of carbonate system dynamics and on an interannual scale, to changes in deep-water convection and to isolate events affecting the 843 844 physical patterns. The present study improved the comprehension of how the processes 845 modifying the rates of accumulation of Cant and acidification on an interannual scale could 846 have a relevance impact on its decadal and multidecadal trends. 847 The assessment of OA within the Irminger and Iceland basins was enhanced by supplying 848 novel data and trends spanning a decade in which the physical patterns reversed. Additionally, 849 the study provides an unprecedent analysis of the physico-chemical variations in the Rockall 850 Trough, which is crucial for the assessment of the entire longitudinal span of the NASPG and 851 advancing our understanding of OA in the North Atlantic and Global Ocean. The data and 852 results given in this article could be used for modelling and compared with other repeated 853 hydrographic section data at mid and high latitudes in the North Atlantic, such as the A02, 854 A25, AR07E and AR28 framed in the GO-SHIP program, as well as used in conjunction to 855 develop future studies focused on the transport of  $C_{ant}$ -loaded and acidified waters. The 856 observational period is relatively short to quantify long-term trends and to formulate 857 significant future projections. The acceleration in surface warming and consequent changes in 858 fCO<sub>2</sub> and pH observed during 2010s may be linked to isolated extreme events such as marine 859 heat waves and are not necessarily indicative of prolonged behaviours over time. Overall, the entrance and accumulation of  $C_{\text{ant}}$  and interannual acidification trends were 860 strongly affected by the cooling, freshening and enhancement in the oxygenation of the whole 861 862 water column during the 2010s. The interannual acidification trends ranged between 0.0013 and 0.0032 units yr<sup>-1</sup> in the Irminger basin, 0.0023 and 0.0029 units yr<sup>-1</sup> in the Iceland basin 863 and 0.0006 and 0.0024 units yr<sup>-1</sup> in the Rockall Trough. The convective processes increased 864 the accumulation rates of  $C_{\rm ant}$  in the interior ocean by 50-86% and accelerated the acidification 865 rates by around 10% compared to previous decades in the Irminger and Iceland basins. In the 866 867 eastern NASPG, the shallower hydrography of the Rockall Trough and the poleward 868 circulation patterns accounted for differences in the acidification rates respect to surrounding waters. The high variability of this area explained the non-significant trends at interannual 869 870 timescales and support the necessity of assess the evolution of its carbonate system properties

short to quantify long-term trends and to formulate significant future projections, the finding





over larger time periods. However, the low NA<sub>T</sub> content of ENACW due to the spreading of 871 872 subtropical subsurface waters into higher latitudes was suggested as the main process 873 decelerating the acidification trends in the upper Rockall Trough. The improved oxygenation 874 of LSW decreasing the  $C_{\text{nat}}$  and thus compensating the  $C_{\text{ant}}$ -driven increase in  $C_{\text{T}}$  may 875 contributed to slowdown the declining in pH<sub>T</sub> in relation to the Iceland basin. The acidification 876 of the NASPG was accompanied by a decline in the  $\Omega_{Ca}$  and  $\Omega_{Arag}$  of 0.004-0.011 and 0.003-877 0.009 units yr<sup>-1</sup>, respectively, in the Irminger basin; 0.007-0.016 and 0.005-0.010 units yr<sup>-1</sup>, respectively, in the Iceland basin; and 0.008-0.021 and 0.005-0.013 units yr<sup>-1</sup>, respectively, in 878 879 the Rockall Trough. 880 The rise in  $NC_T$ , mainly explained by the increasing uptake of  $C_{ant}$ , was found to govern the acidification of the NASPG with a contribution ranged between 53% and 68% in the upper 881 882 water column and higher than 82% toward the interior ocean. The increase in  $NC_T$  was also the main driver of  $\Omega_{Ca}$  and  $\Omega_{Arag}$  trends, with contributions higher than 82% in the Irminger 883 basin, 79% in the Iceland basin and 64% in the Rockall Trough. The combined effect of the 884 885 decreasing temperature, salinity and NA<sub>T</sub> neutralized the 45-49% of the C<sub>T</sub>-driven acidification 886 along the entire longitudinal span of the SPMW. The cooling drove this compensation (27-887 50%) followed by the decrease in  $NA_{\rm T}$  (11-33%), while the freshening had a minimal influence 888 (<6%). The deep-water ventilation processes slowdown the cooling and freshening toward the 889 interior ocean in the Irminger and Iceland and drove the progressively interannual increase in 890  $NA_{\rm T}$ . Thus, the  $NA_{\rm T}$  contributed to acidification by <11% within the intermediate and deep 891 layers and the physical counteraction of the  $C_T$ -driven acidification fall to <10%. In contrast, 892 the cooling weakly promoted the decline in  $\Omega$  (<7% in the upper water column and <2% toward 893 the interior ocean), being only efficiently counteracted in subsurface layers by the increase in 894  $NA_T$  (12-16%) and the freshening (3-5%). 895 The present investigation pretended to emphasize the progressively increase in the uptake and accumulation of Cant and subsequent acceleration of OA along the NASPG. The longitudinal 896 897 span of the NASPG and the differences in circulation patterns, water masses and bathymetry 898 along the section behave as a relevant source of spatio-temporal variability. The enhanced convective processes in the western NASPG were found to favour the entrance of Cant in 899 900 intermediate and deep-layers and this its acidification, as well as influence the carbonate



902

903 904

905

906

907

908

909

910

930



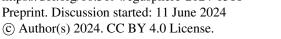
system dynamics in the eastern NASPG. The advancement of comprehensive basin-wide longitudinal evaluations, as the presented here, facilitates a more accurate understanding of the mechanisms dictating basin-scale acidification processes. Furthermore, this promotes the improvement of the projections pertaining to the future state of the oceans run by models and forecast. Considering the important variability in the mechanism controlling the distribution of the physico-biogeochemical properties and particularly the OA in the North Atlantic, this research aims to highlight the necessity of continue monitoring and sampling the whole water column through repeated hydrographic sections, especially through the highly variable but less assess easternmost part.

# Appendix A: Correction of Dissolved Oxygen records for the cruise of 2019

911 The sensor-measured DO data for the cruise of 2019 were corrected by considering the DO 912 output data given by the neural network ESPER NN (Carter et al., 2021) for the cruises of 913 2016 and 2019 (hereinafter ESPER-estimated DO) and the WINKLER-measured DO during 914 the cruise of 2016. Among the 16 equations provided by the ESPER NN that differently 915 combines seawater properties as predictors, we use the equation 8 which only need as inputs 916 the T and S (due to lack of measured macronutrients during the cruise of 2019) along with 917 latitude, longitude, depth and date (see Table 2 in Carter et al., 2021). The reported Root Mean 918 Squared Error (RMSE) of equation 8 for DO estimations in the global ocean is  $\pm 9.7 \mu mol \, kg$ 1, which is reduced for intermediate waters (1000-1500 m) to  $\pm$  5.9  $\mu$ mol kg<sup>-1</sup> (see Table 7 in 919 Carter et al., 2021). Additionally, a new set of DO for 2019 based on WINKLER data for 2016 920 was computed, which was referred in this study as "pseudo-WINKLER" data. The difference 921 922 between WINKLER-measured and ESPER-estimated DO during 2016 was interpolated to the 923 longitudes and depths of the samples of 2019 by applying Delaunay Triangulation. The pseudo-WINKLER data was described as the sum of these interpolated differences and the 924 925 ESPER-estimated DO data for 2019. The longitudinal distribution of measured and ESPERestimated DO data for 2016 and 2019 is depicted in Figure S1a and S1b. The interpolated 926 927 pseudo-WINKLER data for the cruise of 2019 were included in Figure S1a. The sensor records of DO in 2019 were in average 4.90 µmol kg<sup>-1</sup> lower than the ESPER-928 estimated and 10.31 µmol kg<sup>-1</sup> lower than the pseudo-WINKLER. A higher discrepancy was 929

observed in the average sensor-measured DO in the east part (237.60  $\pm$  15.00  $\mu$ mol kg<sup>-1</sup>)







compared with the west part  $(281.40 \pm 14.75 \, \mu mol \, kg^{-1})$ . The average differences (measured 931 932 minus ESPER-estimated DO and measured minus pseudo-WINKLER DO, ΔDO<sub>meas-ESPER</sub> and 933 ΔDO<sub>meas-pseudoWINLKER</sub>, respectively; Figure S2c and S1d) shows that the sensor records were strongly underestimated in the east part (-20.98  $\pm$  10.91 and -28.77  $\pm$  12.60  $\mu$ mol kg<sup>-1</sup>, 934 935 respectively) and weakly overestimated in the west part (8.59  $\pm$  8.53 and 5.18  $\pm$  12.02  $\mu$ mol kg<sup>-1</sup>, respectively) during the cruise of 2019. These differences were corrected separately west 936 and east of 21.5°W by using the relationship  $\frac{\Delta DO_{meas-pseudoWINKLER}}{measured\ DO}$ . The averages of this 937 relationship in the west and east part of the transect (0.016 and -0.12 µmol kg<sup>-1</sup>, respectively) 938 939 were used as corrector factors. The corrected DO values were given by the product between the measured DO and  $\left(1 - \frac{\Delta DO_{meas-pseudoWINKLER}}{measured\ DO}\right)$ . 940 Appendix B: Interannual trends of  $C_T$ ,  $NC_T$ ,  $A_T$  and  $NA_T$ 941 942 The observed rates of increase in  $C_T$  (Table 2) did not show notable differences with respect 943 to the interannual trends determined from previous decades at the Irminger and Iceland basins (0.62-0.82 and 0.38-0.64 μmol kg<sup>-1</sup> yr<sup>-1</sup>, respectively; García-Ibáñez et al., 2016) and at IRM-944 TS and IS-TS (0.49-0.71 and 0.39-0.94 μmol kg<sup>-1</sup> yr<sup>-1</sup>, respectively; Pérez et al., 2021). The 945 946 interannual rates of increase in  $NC_T$  were higher than those of  $C_T$  in the subsurface layers, 947 while the trends were similar among intermediate and deep layers (Table 2). The interannual trends of  $A_T$  (Figure S4 and Table S4) was found to be highly impacted by 948 949 freshening, with decreasing rates ranging from -0.33 to -0.71 µmol kg<sup>-1</sup> yr<sup>-1</sup> among the SPMW and ENACW and from -0.01 to -0.18 µmol kg<sup>-1</sup> yr<sup>-1</sup> within the uLSW, LSW, ISOW 950 and DSOW. It contrasts with the minimal interannual changes and slight rates of increase in 951  $A_T$  encountered among the different layers by García-Ibáñez et al., (2016) from 1991 to 2015 952 953 in the Irminger basin (between 0.10 and 0.28 µmol kg<sup>-1</sup> yr<sup>-1</sup>) and Iceland basin (between -0.04 and 0.07 µmol kg<sup>-1</sup> yr<sup>-1</sup>), and with the trends reported for the period 1983-2013 by Pérez 954 et al., (2021) at the IRM-TS (between 0.13 and 0.22 µmol kg<sup>-1</sup> yr<sup>-1</sup>) and at the IS-TS (between 955 -0.04 and 0.15  $\mu$ mol kg<sup>-1</sup> yr<sup>-1</sup>). These heterogeneities in the temporal evolution of the  $A_T$  were 956 957 driven by the decadal salinification of the whole water column observed since the late 20th 958 century and interrupted by interannual freshening episodes such as during the 2010s.





The interannual  $NA_T$  trends reversed in comparison with those of  $A_T$  along the SPMW and 959 ENACW (Figure S4 and Table S4). This increment in NAT was related with the stagnation of 960  $A_T$ -rich subtropical waters in the upper layers due to the slowdown of the NASPG since the 961 mid-90s (e. g. Böning et al., 2006; Häkkinen and Rhines, 2004). 962 963 **Code Availability** 964 **MATLAB** R codes for **CANYON-B** available 965 https://github.com/HCBScienceProducts/CANYON-B. MATLAB and R code for ESPER NN 966 are available at https://github.com/BRCScienceProducts/ESPER. MATLAB code for carbon available 967 anthropogenic calculation is http://oceano.iim.csic.es/ media/cantphict0 toolbox 20190213.zip. The CO<sub>2</sub>SYS programme 968 for MATLAB is available at https://github.com/jonathansharp/CO2-System-Extd. 969 970 **Data Availability Statement** The measured surface-to-bottom CLIVAR data (2009-2019) used in this investigation are 971 published in open-access at Zenodo (DOI: 10.5281/zenodo.10276221). The GO-SHIP A25-972 available 973 **OVIDE** data for the cruise of 2018 at **SEANOE** 974 (https://www.seanoe.org/data/00762/87394/). **Author contribution** 975 976 DCH contributed with data analysis and wrote the manuscript. FFP, DCH, AV, DGS, AGG, MGD and JMSC worked on the design, conceptualization and data preparation. SG, AS, MGD, 977 JMSC, AGG and DGS participated in 8, 4, 7, 7, 2 and 2 cruises, respectively. SG and AS were 978 979 the Chief Scientist in all cruises and responsible for the operational and maintenance 980 procedures for the CTD and additional sensors and thus for physical and sensor-measured variables. MGD and JMSC got the funding acquisition and provision of resources for the 981 982 Spanish team from the ULPGC. SG and AS got the funding for ship time and provision of 983 resources for all the cruise participants. All authors critically revised the manuscript. 984 **Competing interest** 985 The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest. 986 987 Acknowledgement 988 The participation on the cruises for the Spanish Team from the ULPGC was funded by the 989 Science Spanish Ministry under the Complimentary Actions CTM2008-05255, CTM2010-990 09514-E and CTM2011-12984-E (years 2009-2011), the FP7 European project https://doi.org/10.5194/egusphere-2024-1388 Preprint. Discussion started: 11 June 2024 © Author(s) 2024. CC BY 4.0 License.





991 CARBOCHANGE under grant agreement no. 264879 and by the Spanish Innovation and 992 Science Ministry through the Projects EACFe (CTM2014-52342-P) and ATOPFe (CTM2017-83476-P). SG and AS were supported by FMWE-2023-0002. FFP and AV were supported by 993 (PID2019-104279GB-C21) 994 BOCATS2 project funded MCIN/AEI/10.13039/501100011033 and by EuroGO-SHIP project (Horizon Europe 995 #101094690). The participation of DCH was funded by the PhD grant PIFULPGC-2020-2 996 997 ARTHUM-2. Special thanks go to the technician and researchers Adrian Castro Álamo (2 998 cruises), Anna Barrera Galderique (3 cruises), Rayco Alvarado Medina (2 cruises) and Pilar 999 Aparicio Rizzo (1 cruise) who helped with in situ analysis. We also thanks technicians at the 1000 P. P. Shirshov Institute of Oceanology from the Russian Academy of Science for the in situ 1001 analysis of dissolved oxygen and nutrients.





# Legend for figures

1003 Figure 1. (a) Map of the North Atlantic Subpolar Gyre (NASPG) with the schematic diagram 1004 of the surface and deep circulation patterns compiled from Lherminier et al., (2010); Pérez et 1005 al., (2021); Sarafanov et al., (2012); Schmitz and McCartney, (1993); Schott and Brandt, (2007) and Sutherland and Pickart, (2008). The acronyms are defined as follow: the 1006 1007 bathymetric features are shown in grey (RR: Reykjanes Ridge, HB: Haton Bank, GBB: George 1008 Bligh Bank, CGFZ: Charlie-Gibbs Fracture Zone, GIR: Greenland-Iceland Ridge, and GSR: 1009 Greenland-Scotland Ridge), the surface currents are shown in orange (NAC: North Atlantic Current, and IC: Irminger Current) and the deep water circulation is shown in blue and purple 1010 1011 (ISOW: Iceland-Scotland Overflow Water, DSOW: Denmark Strait Overflow Water, LSW: Labrador Sea Water, and DWBC: Deep Western Boundary Current). The longitudinal 1012 1013 distribution of the surface-to-bottom sampling stations along the cruise track of 2016 (repeated 1014 throughout the cruises) is shown with red dots. The black lines along the cruise track delimited 1015 the three basins. (b) Vertical distribution of the water masses considered in this study for each of the basins. The isopycnals, plotted over the salinity distribution for the cruise of 2016, show 1016 1017 the limits of the layers and were defined by potential density (in kg m<sup>-3</sup>) referred to 0 dbar ( $\sigma_0$ ). The vertical gray lines show the limits between basins. The acronyms of the water masses and 1018 1019 the selection of potential density values delimiting the layers are detailed in section 2.2.4. 1020 Figure produced with Ocean Data View (Schlitzer, 2021).

- 1021 Figure 2. Water-column distribution along the longitudinal transect of (a) temperature, (b)
- salinity, (c) A<sub>T</sub>, (d) pH<sub>T</sub> and I) AOU for the cruises of 2009 (left plots) and 2016 (right plots).
- The vertical while lines show the limits between basins. Figure produced with Ocean Data
- 1024 View (Schlitzer, 2021).
- Figure 3. Water-column distribution along the longitudinal transect of (a) C<sub>T</sub>, (b) C<sub>ant</sub> and (c)
- 1026 C<sub>nat</sub> for the cruises of 2009 (left plots) and 2016 (right plots). The vertical while lines show the
- limits between basins. Figure produced with Ocean Data View (Schlitzer, 2021).
- 1028 Figure 4. Temporal distribution (2009-2019) of the average temperature and salinity in each of
- the layers considered for the Irminger (left plot column), Iceland (central plot column) and
- 1030 Rockall basins (right plot column). The average values were calculated for each cruise and
- layer and represented with coloured points together with their respective error bars at the time
- of each cruise (the method used for calculations was described in section 3.2). In the Irminger
- plots, the empty points represent the average values for 2019 calculated with the measured
- data available in the easternmost part of the basin (sampled part during this cruise), while the
- coloured points for 2019 represent the average values corrected with A25-OVIDE-2018 data.
- 1036 The interannual trends were given by linear regression of the average values, with the values
- of the slope, the standard error of estimate and the  $r^2$  presented in Table 2.
- 1038 Figure 5. Temporal distribution (2009-2019) of the average C<sub>T</sub>, C<sub>ant</sub> and C<sub>nat</sub> in each of the
- layers considered for the Irminger (left plot column), Iceland (central plot column) and Rockall
- basins (right plot column). The average values were calculated for each cruise and layer and
- represented with coloured points together with their respective error bars at the time of each
- cruise (the method used for calculations was described in section 3.2). In the Irminger plots,
- the empty points represent the average values for 2019 calculated with the measured data available in the easternmost part of the basin (sampled part during this cruise), while the
- coloured points for 2019 represent the average values corrected with A25-OVIDE-2018 data.





- The interannual trends were given by linear regression of the average values, with the values 1046
- 1047 of the slope, the standard error of estimate and the  $r^2$  presented in Table 2.
- Figure 6. Temporal distribution (2009-2019) of the average pH<sub>T</sub> (in situ temperature) in each 1048
- 1049 of the layers considered for the Irminger (left plot column), Iceland (central plot column) and
- 1050 Rockall basins (right plot column). The average values were calculated for each cruise and
- layer and represented with coloured points together with their respective error bars at the time 1051
- 1052 of each cruise (the method used for calculations was described in section 3.2). In the Irminger
- 1053 plots, the empty points represent the average values for 2019 calculated with the measured
- 1054 data available in the easternmost part of the basin (sampled part during this cruise), while the
- coloured points for 2019 represent the average values corrected with A25-OVIDE-2018 data. 1055
- The interannual trend were given by linear regression of the average values, with the values of 1056
- the slope, the standard error of estimate and the  $r^2$  presented in Table 2. 1057
- Figure 7. Temporal distribution (2009-2019) of the average  $\Omega$ Ca and  $\Omega$ Arag in each of the 1058
- 1059 layers considered for the Irminger (left plot column), Iceland (central plot column) and Rockall
- 1060 basins (right plot column). The average values were calculated for each cruise and layer and
- represented with coloured points together with their respective error bars at the time of each 1061
- 1062 cruise (the method used for calculations was described in section 3.2). In the Irminger plots,
- the empty points represent the average values for 2019 calculated with the measured data 1063
- available in the easternmost part of the basin (sampled part during this cruise), while the 1064
- coloured points for 2019 represent the average values corrected with A25-OVIDE-2018 data. 1065
- The interannual trends were given by linear regression of the average values, with the values 1066
- 1067 of the slope, the standard error of estimate and the  $r^2$  presented in Table 2.

## **Legend for Tables**

1068

- 1069 Table 1. Metadata list of hydrographic cruises.
- Table 2. Interannual trends of temperature, salinity,  $C_T$ ,  $C_{ant}$ ,  $C_{nat}$ ,  $pH_T$ ,  $\Omega Ca$  and  $\Omega Arag$  in each 1070
- 1071 of the layers and basins. The ratios of change were based on linear regressions applied to the
- 1072 average values (as represented in Figures 4-7) and presented together with its Standard error
- of estimate. The correlation coefficients r<sup>2</sup> and p-values were also provided. Values in bold 1073
- denote trends statistically significant at the 95% level of confidence. 1074
- 1075
- Table 3. Temporal changes in pH<sub>T</sub> (**in 10<sup>-3</sup> units yr<sup>-1</sup>**) explained by fluctuations in temperature  $\left(\frac{\partial pH_T}{\partial T}\frac{\partial T}{\partial t}\right)$ , salinity  $\left(\frac{\partial pH_T}{\partial S}\frac{\partial S}{\partial t}\right)$ , A<sub>T</sub>  $\left(\frac{\partial pH_T}{\partial A_T}\frac{\partial NA_T}{\partial t}\right)$ , and C<sub>T</sub>  $\left(\frac{\partial pH_T}{\partial C_T}\frac{\partial NC_T}{\partial t}\right)$  in each of the layers considered for the Irminger, Iceland and Rockall basins during the period 2009-2019. The sum 1076
- 1077
- 1078 of changes explained by the individual drivers represents the calculated interannual  $pH_T$
- 1079
- change  $\left(\frac{dpH_T}{dt} \text{ calculated}\right)$ , as detailed in section 2.2.5. The observed interannual pH<sub>T</sub> trends  $\left(\frac{dpH_T}{dt} \text{ observed}\right)$ , shown in Figure 7 and provided in Table 2, were also added to the table for 1080
- 1081 comparison.
- 1082
- Table 4. Temporal changes in  $\Omega$ Ca and  $\Omega$ Arag (in 10<sup>-3</sup> units yr<sup>-1</sup>) explained by fluctuations in temperature  $\left(\frac{\partial \Omega}{\partial T}\frac{\partial T}{\partial t}\right)$ , salinity  $\left(\frac{\partial \Omega}{\partial S}\frac{\partial S}{\partial t}\right)$ , A<sub>T</sub>  $\left(\frac{\partial \Omega}{\partial A_T}\frac{\partial NA_T}{\partial t}\right)$ , and C<sub>T</sub>  $\left(\frac{\partial \Omega}{\partial C_T}\frac{\partial NC_T}{\partial t}\right)$  in each of the layers 1083
- considered for the Irminger, Iceland and Rockall basins during the period 2009-2019. The sum 1084

https://doi.org/10.5194/egusphere-2024-1388 Preprint. Discussion started: 11 June 2024 © Author(s) 2024. CC BY 4.0 License.





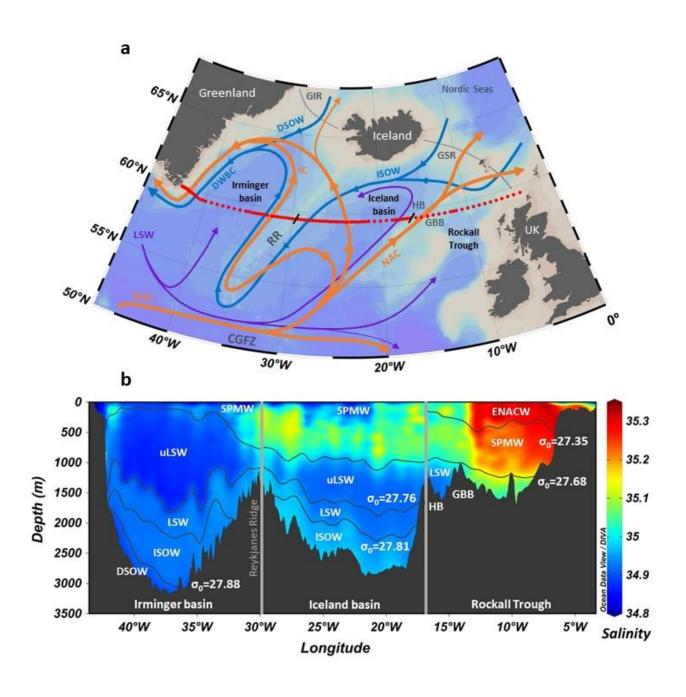
1085	of changes explained by the individual drivers represents the calculated interannual $\Omega$ change
1086	$\left(\frac{d\Omega}{dt} \text{ calculated}\right)$ , as detailed in section 2.2.6. The observed interannual $\Omega$ trends
1087	$\left(\frac{d\Omega}{dt}\right)$ observed, shown in Figure 6 and provided in Table 2, were also added to the table for
1088	comparison.





1089 Figures

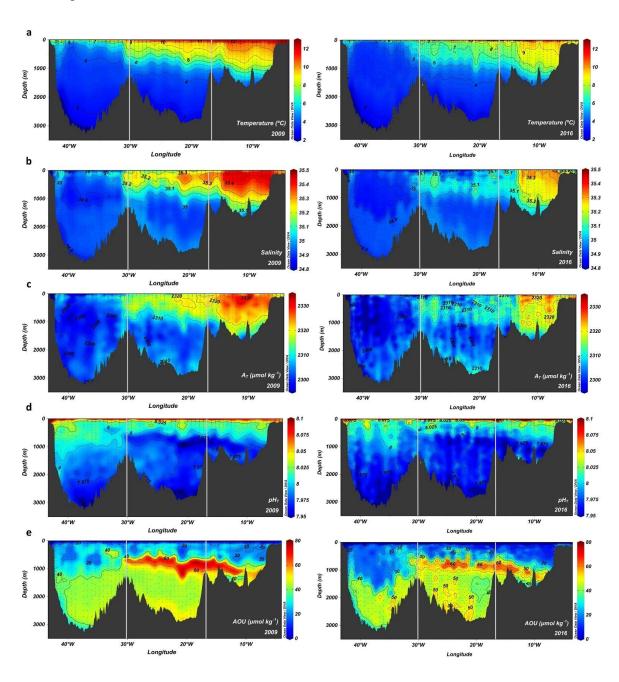
1090 Fig. 1







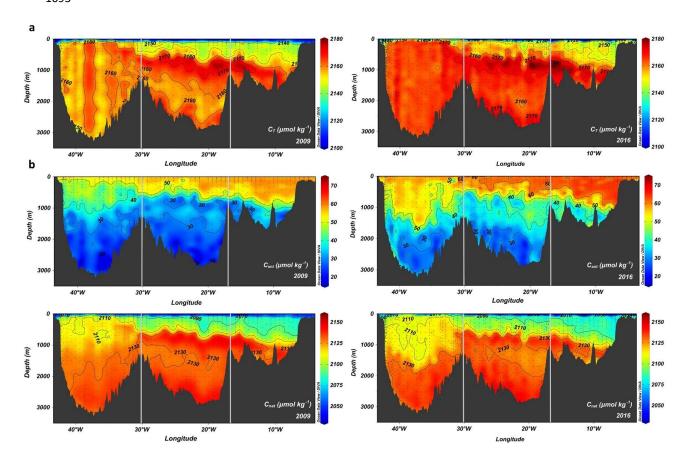
1091 Fig. 2







1092 Fig. 31093







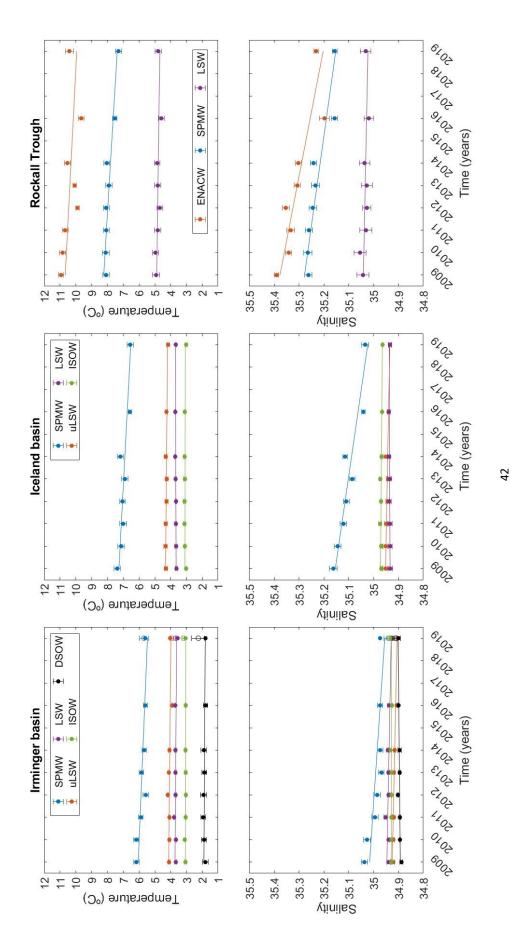
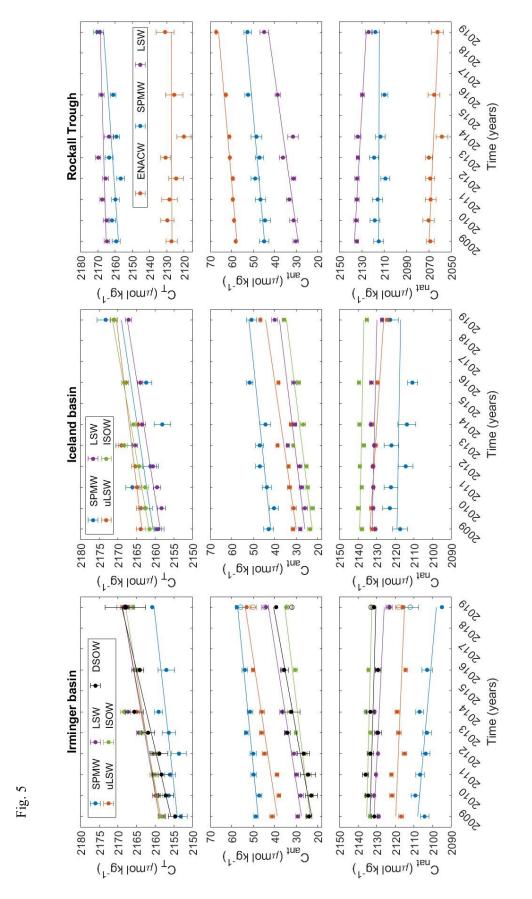


Fig. 4



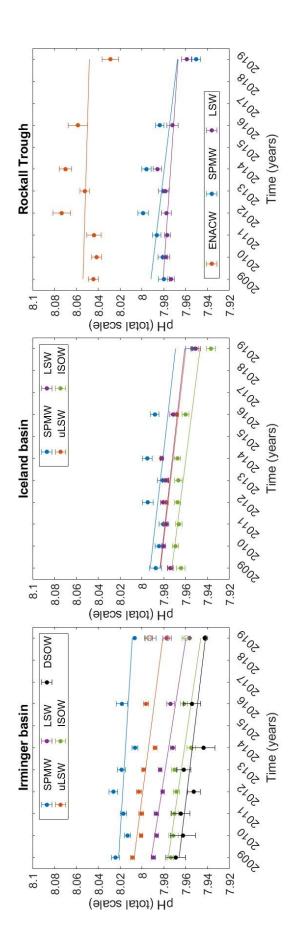




43



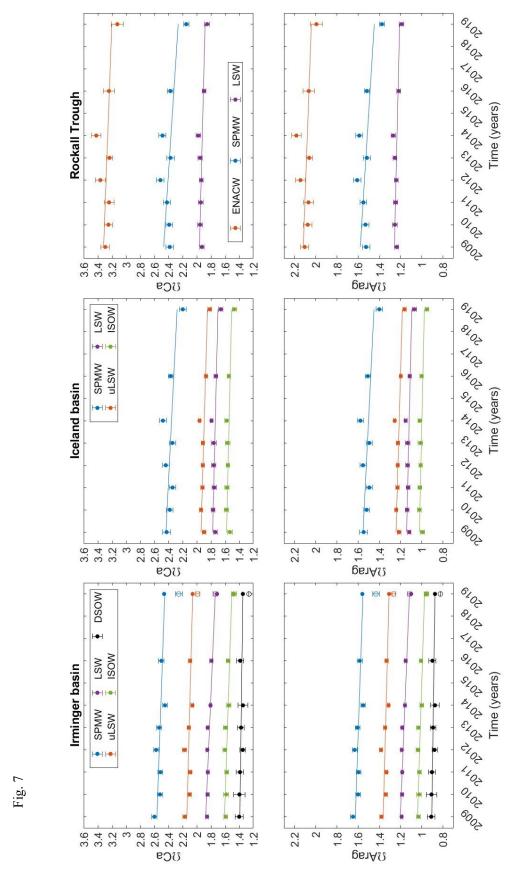




44







45





Year	Cruise ID	Date	Research Vessel (R/V)	Chief Scientist
2009	AI28	Aug 15-Sept 27	Akademik Ioffe	A. Sokov
2010	AI31	Sep 2-Sep 27	Akademik Ioffe	A. Sokov
2011	SV33	Sep 9-Sep 28	Akademik Sergey Vavilov	A. Sokov
2012	AI38	May 25-Jul 1	Akademik Ioffe	S. Gladyshev
2013	AI41	Jun 26-Jul 23	Akademik Ioffe	S. Gladyshev
2014	AI44	Jun 27-Jul 20	Akademik Ioffe	S. Gladyshev
2016	AI51	Jun 3-Jul 13	Akademik Ioffe	S. Gladyshev
2019	AMK77	Aug 8-Sep 10	Akademik Mstislav Keldysh	S. Gladyshev





Basin	Layer	Temperature	ıre	Salinity			$C_{\mathrm{T}}$				C <sub>mt</sub>			$C_{nat}$	Ħ			$\mathrm{ph}_{\mathrm{T}}$			Œ	Ωса		<u> </u>	$\Omega$ Arag	ag	
		ratio ( ${}^{\circ}C$ yr $^{-1}$ ) r $^{2}$	p-value	ratio ( $^{\circ}$ C yr $^{-1}$ ) $^{-2}$ p-value	ılue ratio	(umol kg	, yr	r 5	p-value ratio (µmol kg '' r	µmol kg	yr ) r	2 p-value		ol kg 'ı	·-1 r 2	p-value	ratio ( $\mu$ mol kg $^{-1}$ yr $^{-1}$ ) r $^2$ p-value ratio ( $10^3$ units yr $^{-1}$ )	nits yr )	2 4	p-value ra	ratio (units yr )	r 5	p-value		ratio (units yr <sup>-1</sup> )	r 5	p-value
	SPMW	<b>-0.058</b> $\pm$ <b>0.024</b> 0.60 0.02	0.02	<b>-0.006</b> $\pm$ <b>0.003</b> 0.59 0.03	-	0.62 ±	0.23 0	0.66 0.02	2 0.95	+1	0.17 0.89	10.0> 68	-1.00	± 0.42	09:0 7:	0.02	-1.25 ±	0.93	0.32 0	0.14 -0.	$-0.011 \pm 0.006$	06 0.50	0.05	-0.007	± 0.003	0.53	0.04
	nLSW	$-0.014 \pm 0.011 \ 0.36$	0.16	$-0.014 \pm 0.011 \ 0.30 \ 0.16$ $-0.002 \pm 0.001 \ 0.59 \ 0.03$		1.02 ±	0.18 0	0.89 <0.01	1.48	+1	0.29 0.87	87 <0.01	-0.47	± 0.38	8 0.28	0.18	-2.62 ±	69.0	> 62.0	<0.01 -0.	-0.008 ± 0.005	05 0.40	60.00	-0.006	± 0.003	0.44	80.0
Irminger	TSW	$-0.010 \pm 0.008 \ 0.31$	0.15	$-0.010 \pm 0.008$ 0.31 0.15 $-0.002 \pm 0.001$ 0.50 0.05		0.98 ±	0.26 0	0.78 <0.01	1.53	+1	0.23 0.9	0.92 <0.01	-0.54	± 0.30	0 0.46	0.06	-3.17 ±	0.52	0.91	<0.01	$-0.014 \pm 0.003$	03 0.85	6 <0.01	-0.009	± 0.002	0.85	<0.01
	ISOW	$-0.002 \pm 0.003 0.11$	0.42	$-0.002 \pm 0.003 \ 0.11 \ 0.42$ 0.000 $\pm 0.000 \ 0.00 \ 0.99$	<b>06.0</b> 66	+1	0.34 0	0.64 0.02	2 1.18	+1	0.29 0.81	81 <0.01	-0.27	± 0.20	0.32	0.14	-2.97 ±	0.70	0.83	<0.01	$-0.010 \pm 0.003$	03 0.73	3 <0.01	-0.007	± 0.002	0.74	<0.01
	DSOW	$-0.008 \pm 0.008 \ 0.22$	0.25	$0.001 \pm 0.001 \ 0.43 \ 0.08$		1.32 ±	0.23 0	0.90 <0.01	1.77	+1	0.32 0.89	89 <0.01	-0.32	± 0.33	3 0.19	0.28	-2.41 ±	0.87	> 79.0	<0.01 -0.	$-0.004 \pm 0.003$	03 0.39	0.10	-0.003	± 0.002	0.46	0.07
	SPMW	-0.074 ± 0.022 0.74	10:0> 1	<b>-0.074</b> $\pm$ <b>0.022</b> 0.74 <0.01 <b>-0.013</b> $\pm$ <b>0.002</b> 0.89 <0.01	.01 0.85	+1	0.64 0	0.32 0.15	5 1.02	+1	<b>0.31</b> 0.74	74 <0.01	-0.19	± 0.74	4 0.02	0.75	-2.32 ±	1.63	0.34 (	0.13 -0.	-0.016 ± 0.010	10 0.37	, 0.11	-0.010	± 0.007	0.39	0.10
	uLSW	<b>-0.012</b> $\pm$ <b>0.005</b> 0.63 0.02	3 0.02	<b>-0.002</b> ± <b>0.000</b> 0.76 <0.01		+ 89.0	0.22 0	0.71 <0.01	1.42	+1	0.38 0.7	0.78 <0.01	-0.74	± 0.21	1 0.75	<0.01	-2.31 ±	1.01	0.58	0.03 -0.	-0.009 ± 0.005	05 0.46	0.07	-0.006	± 0.003	0.47	90.0
Iceland	LSW	$0.005 \pm 0.003 \ 0.43 \ 0.08$	8 0.08	$0.000 \pm 0.000 \ 0.28 \ 0.18$		+ 88.0	0.22 0	0.80 <0.01	1.18	+1	0.35 0.7	0.75 <0.01	-0.26	± 0.26	6 0.20	0.27	-2.26 ±	1.06	0.54 (	0.04	-0.008 ± 0.005	05 0.41	0.00	-0.005	± 0.003	0.41	0.09
	ISOW	$-0.003 \pm 0.006 \ 0.05 \ 0.61$	0.61	<b>-0.001</b> $\pm$ <b>0.000</b> 0.47 0.05		0.98 ±	0.17 0	0.89 <0.01	1.20	+1	<b>0.32</b> 0.79	79 <0.01	-0.23	± 0.21	1 0.23	0.23	-2.58 ±	0.99	0.64	<0.01 -0.	$-0.007 \pm 0.004$	04 0.42	80.0	-0.005	± 0.003	0.43	80.0
	ENACW	<b>ENACW</b> $-0.073 \pm 0.061 \ 0.27 \ 0.19$	0.19	$-0.017 \pm 0.004 \ 0.80 \ < 0.01$	.01 0.05	+1	0.57 0	0.00 0.92	2 0.85	+1	0.11 0.9	0.94 <0.01	-0.84	± 0.50	0 0.43	0.08	-0.58 ±	2.31	0.02	0.77 -0.	$-0.012 \pm 0.013$	13 0.18	3 0.30	-0.008	± 0.008	0.19	0.28
Rockall	SPMW		40.01	<b>-0.085</b> $\pm$ <b>0.019</b> 0.84 <0.01 <b>-0.013</b> $\pm$ <b>0.003</b> 0.85 <0.01	.01 0.86	+1	0.46 0	0.48 0.05	5 0.87	+1	<b>0.18</b> 0.86	86 <0.01	-0.07	± 0.59	00.00	0.88	-2.43 ±	1.90	0.30	0.16 -0.	$-0.021 \pm 0.013$	13 0.38	0.10	-0.013	± 0.008	0.39	0.10
	TSW	-0.020 ± 0.016 0.29	0.17	$-0.020 \pm 0.016 \ 0.29 \ 0.17$ $-0.002 \pm 0.001 \ 0.30 \ 0.16$	16 0.35	+1	0.29 0	0.27 0.19	9 1.38	+1	0.34 0.81	81 <0.01	-1.05	± 0.24	4.0.84	<0.01	-1.36 ±	0.97	0.34	0.13 -0.	-0.008 ± 0.004	04 0.45	0.07	-0.005	± 0.003	0.45	0.07

47





Basin	Layer	$\frac{\partial p H_T}{\partial T} \frac{\partial T}{\partial t}$	$\frac{\partial pH_T}{\partial S} \frac{\partial S}{\partial t}$	$\frac{\partial p H_T}{\partial A_T} \frac{\partial N A_T}{dt}$	$\frac{\partial pH_T}{\partial C_T}\frac{\partial NC_T}{dt}$	$\frac{dpH_T}{dt} (obs)$	$\frac{dpH_T}{dt} (calc$	(calculated)
	SPMW	$0.91 \pm 0.38$	$0.05 \pm 0.02$	$0.31 \pm 0.43$	$-2.67 \pm 0.63$	$-1.25 \pm 0.93$	-1.41 ±	0.85
	uLSW	$0.22  \pm  0.17$	$0.02 \pm 0.01$	$-0.10 \pm 0.40$	$-2.99 \pm 0.53$	$-2.62 \pm 0.69$	-2.86 ±	89.0
Irminger	$\Gamma$ SW	$0.16 \pm 0.12$	$0.01 \pm 0.01$	$-0.04 \pm 0.39$	$-2.85 \pm 0.62$	$-3.17 \pm 0.52$	-2.72 ±	0.74
	MOSI	$0.03  \pm  0.05$	$0.00 \pm 0.00$	$-0.13 \pm 0.30$	$-2.38 \pm 0.88$	$-2.97 \pm 0.70$	-2.48 ±	0.93
	DSOW	$0.13 \pm 0.12$	$-0.01 \pm 0.00$	$-0.60 \pm 0.18$	$-3.41 \pm 0.62$	$-2.41 \pm 0.87$	-3.90 ±	99.0
	SPMW	$1.15 \pm 0.35$	$0.10 \pm 0.02$	$0.61 \pm 0.19$	$-4.14 \pm 1.76$	$-2.32 \pm 1.63$	-2.27 ±	1.81
[00]	uLSW	$0.19 \pm 0.08$	$0.01 \pm 0.00$	$-0.24 \pm 0.45$	$-2.08 \pm 0.66$	$-2.31 \pm 1.01$	-2.12 ±	08.0
ıcelalıd	$\Gamma$ SW	$-0.08 \pm 0.05$	$0.00 \pm 0.00$	$-0.04 \pm 0.44$	$-2.26 \pm 0.57$	$-2.26 \pm 1.06$	-2.38 ±	0.72
	MOSI	$0.04 \pm 0.10$	$0.01 \pm 0.00$	$0.12 \pm 0.40$	$-2.70 \pm 0.43$	$-2.58 \pm 0.99$	-2.53 ±	09.0
	ENACW	$1.13 \pm 0.94$	$0.14 \pm 0.04$	$0.73 \pm 0.66$	$-2.25 \pm 1.39$	$-0.58 \pm 2.31$	-0.25 ±	1.80
Rockall	SPMW	$1.31  \pm  0.29$	$0.10 \pm 0.02$	$0.47  \pm  0.22$	$-3.84 \pm 1.23$	$-2.43 \pm 1.90$	-1.96 ±	1.28
	LSW	$0.30 \pm 0.24$	$0.01 \pm 0.01$	$-0.14 \pm 0.37$	$-0.94 \pm 0.86$	$-1.36 \pm 0.97$	± 92.7e	96.0

œ





.;			эл эт	Seve	$\partial\Omega$ $\partial NA_T$	$\partial\Omega$ $\partial NC_T$	1 `	υp		
Basin	Layer		$\overline{\partial T} \overline{dt}$	$\overline{\partial S} \overline{dt}$	$\overline{\partial A_T}$ $dt$	$\overline{\partial C_T}$ $dt$	$\overline{dt}^{(obs)}$	$\overline{dt}$	(calculatea <sub>)</sub>	atea)
	CDMATE	Calcite	$-0.57 \pm 0.24$	$-0.43 \pm 0.18$	$1.68 \pm 2.37$	$-13.35 \pm 3.14$	$-11.03 \pm 5.57$	7 -12.67	± L	3.94
	SPIM W	Aragonite	$-0.49 \pm 0.20$	$-0.29 \pm 0.12$	$1.07 \pm 1.50$	$-8.47 \pm 1.99$	$-7.17 \pm 3.46$	6 -8.17	7 +	2.50
	WO I.	Calcite	$-0.17 \pm 0.13$	$-0.12 \pm 0.05$	$-0.46 \pm 1.82$	$-12.61 \pm 2.24$	$-8.28 \pm 5.16$	•	+ 9	2.89
	uL.5 W	Aragonite	$-0.13 \pm 0.10$	$-0.08 \pm 0.03$	$-0.29 \pm 1.16$	$-8.03 \pm 1.43$	$-5.55 \pm 3.21$		3	1.84
1	1 5117	Calcite	$-0.15 \pm 0.11$	$-0.09 \pm 0.05$	$-0.17 \pm 1.55$	$-10.42 \pm 2.27$	$-13.54 \pm 2.88$	8 -10.83	3 +	2.75
Hillinger	LS W	Aragonite	$-0.11 \pm 0.08$	$-0.06 \pm 0.03$	$-0.11 \pm 0.99$	$-6.69 \pm 1.45$	$-8.65 \pm 1.83$		7 +	1.76
	16011	Calcite	$-0.04 \pm 0.05$	$0.00 \pm 0.01$	$-0.44 \pm 1.03$	$-7.48 \pm 2.75$	$-10.35 \pm 3.23$	3 -7.96	+1	2.94
	ISO W	Aragonite	$-0.02 \pm 0.04$	$0.00 \pm 0.01$	$-0.29 \pm 0.67$	$-4.84 \pm 1.78$	$-6.66 \pm 2.04$		+1	1.90
	Deow.	Calcite	$-0.13 \pm 0.12$	$0.03 \pm 0.02$	$-1.78 \pm 0.52$	+1	$-4.30 \pm 2.76$	•	1 +	1.77
	DSOW	Aragonite	$-0.09 \pm 0.09$	$0.02 \pm 0.01$	$-1.16 \pm 0.34$	$-6.01 \pm 1.10$	$-3.02 \pm 1.68$	8 -7.24	<b>1</b> ±	1.15
	CDAMA	Calcite	$-0.88 \pm 0.26$	$-0.86 \pm 0.16$	$3.16 \pm 1.00$	$-19.59 \pm 8.35$	$-15.77 \pm 10.40$		± ∠	8.42
	OF IVI W	Aragonite	$-0.72 \pm 0.22$	$-0.58 \pm 0.10$	$2.02 \pm 0.64$	$-12.48 \pm 5.32$	$-10.37 \pm 6.5$	·	± 7	5.37
	WP I.	Calcite	$-0.17 \pm 0.07$	$-0.09 \pm 0.03$	$-1.02 \pm 1.89$	$-7.98 \pm 2.52$	$-9.18 \pm 5.11$	1 -9.26	+1	3.15
[colon]	ur.s w	Aragonite	$-0.12 \pm 0.05$	$-0.06 \pm 0.02$	$-0.65 \pm 1.21$	+1	+1		+1	2.02
ıcciaiiu	I CW	Calcite	$0.08 \pm 0.05$	$0.02 \pm 0.01$	$-0.15 \pm 1.70$	$-7.92 \pm 2.00$	$-7.53 \pm 4.64$		7 ±	2.63
	LS W	Aragonite	$0.06 \pm 0.03$	$0.01 \pm 0.01$	$-0.09 \pm 1.09$	$-5.10 \pm 1.29$	+1		2 +	1.69
	\X\O31	Calcite	$-0.04 \pm 0.10$	$-0.03 \pm 0.02$	$0.41 \pm 1.37$	$-8.38 \pm 1.33$	$-7.22 \pm 4.34$		+1	1.91
	NOCI	Aragonite	$-0.03 \pm 0.07$	$-0.02 \pm 0.01$	$0.27 \pm 0.89$	$-5.43 \pm 0.86$	$-4.72 \pm 2.76$		2 ±	1.24
	EMACW.	Calcite	$-0.82 \pm 0.69$	$-1.50 \pm 0.38$	$5.16 \pm 4.63$	$-14.21 \pm 8.78$	$-11.60 \pm 12.67$	57 -11.37	± L	9.95
	EINACW	Aragonite	$-0.79 \pm 0.66$	$-1.00 \pm 0.25$	$3.29 \pm 2.95$	$-9.06 \pm 5.60$	$-7.66 \pm 7.96$		7 ±	6.37
Dooleall	CDMIX	Calcite	$-1.15 \pm 0.26$	$-0.82 \pm 0.18$	$2.44 \pm 1.15$	$-18.21 \pm 5.83$	$-20.57 \pm 13.40$	•	4 +	5.95
NOCKAII	V IVI V	Aragonite	$-0.93 \pm 0.21$	$-0.55 \pm 0.12$	$1.56 \pm 0.74$	$-11.66 \pm 3.73$	$-13.24 \pm 8.47$	7 -11.58	+1	3.81
	I CW	Calcite	$-0.28 \pm 0.22$	$-0.10 \pm 0.08$	$-0.58 \pm 1.57$	$-3.62 \pm 3.30$	$-7.88 \pm 4.41$	.1 -4.59	+ +	3.66
	LO W	Aragonite	$-0.21 \pm 0.16$	$-0.07 \pm 0.05$	$-0.37 \pm 1.01$	$-2.33 \pm 2.12$	$-4.97 \pm 2.82$	.2 -2.97	7 ±	2.35

49





## References

Accornero, A., Manno, C., Esposito, F., & Gambi, M.C. (2003). The vertical flux of particulate matter in the polynya of Terra Nova Bay. Part II. Biological components. Antarctic Science, 15, 175–188. DOI: 10.1017/S0954102003001214

Álvarez, M., Pérez, F.F., Bryden, H., & Ríos, A.F. (2004). Physical and biogeochemical transports structure in the North Atlantic subpolar gyre. Journal of Geophysical Research: Ocean, 109. DOI: 10.1029/2003jc002015

Álvarez, M., Ríos, A.F., Pérez, F.F., Bryden, H.L., & Rosón, G. (2003). Transports and budgets of total inorganic carbon in the subpolar and temperate North Atlantic. Global Biogeochemical Cycles, 17(2), 2-1-2–21. DOI: 10.1029/2002gb001881

Anderson, L.G. (2001). of the Surface Ocean CO2 System in the Nordic Seas and Northern North Atlantic to Climate Change Using mixed-layer salinity and temperature Model and empirical relationships between sea surface temperature and surface water CO2 fugacity, and between sea.

Azetsu-Scott, K., Jones, E. P., Yashayaev, I., & Gershey, R. M. (2003). Time series study of CFC concentrations in the Labrador Sea during deep and shallow convection regimes (1991–2000). Journal of Geophysical Research: Oceans, 108(C11).

Balmaseda, M.A., Smith, G.C., Haines, K., Anderson, D., Palmer, T.N., & Vidard, A. (2007). Historical reconstruction of the Atlantic Meridional Overturning Circulation from the ECMWF operational ocean reanalysis. Geophysical Research Letters, 34, 1–6. DOI: 10.1029/2007GL031645

Bates, N.R., Astor, Y.M., Church, M.J., Currie, K., Dore, J.E., González-Dávila, M., Lorenzoni, L., Muller-Karger, F., Olafsson, J., & Santana-Casiano, J.M. (2014). A timeseries view of changing surface ocean chemistry due to ocean uptake of anthropogenic CO2 and ocean acidification. Oceanography, 27, 126–141. DOI: 10.5670/oceanog.2014.16

Bates, N.R., Best, M.H.P., Neely, K., Garley, R., Dickson, A.G., & Johnson, R.J. (2012). Detecting anthropogenic carbon dioxide uptake and ocean acidification in the North Atlantic Ocean. Biogeosciences, 9, 2509–2522. DOI: 10.5194/bg-9-2509-2012

Bathmann, U. V., Noji, T.T., & von Bodungen, B. (1991). Sedimentation of pteropods in the Norwegian Sea in autumn. Deep Sea Research Part A. Oceanographic Research Papers, 38, 1341–1360. DOI: 10.1016/0198-0149(91)90031-A

Benson, B.B., & Krause, D. (1984). The concentration and isotopic fractionation of oxygen dissolved in freshwater and seawater in equilibrium with the atmosphere. Deep Sea Research Part B. Oceanographic Literature Review, 31, 859. DOI: 10.1016/0198-0254(84)93289-8

Bersch, M., Yashayaev, I., & Koltermann, K. P. (2007). Recent changes of the thermohaline circulation in the subpolar North Atlantic. Ocean Dynamics, 57, 223-235.





Bittig, H.C., Steinhoff, T., Claustre, H., Fiedler, B., Williams, N.L., Sauzède, R., Körtzinger, A., & Gattuso, J.P. (2018). An alternative to static climatologies: Robust estimation of open ocean CO2 variables and nutrient concentrations from T, S, and O2 data using Bayesian neural networks. Frontiers in Marine Science, 5, 1–29. DOI: 10.3389/fmars.2018.00328

Böning, C.W., Scheinert, M., Dengg, J., Biastoch, A., & Funk, A. (2006). Decadal variability of subpolar gyre transport and its reverberation in the North Atlantic overturning. Geophysical Research Letters, 33. DOI: 10.1029/2006GL026906

Brambilla, E., & Talley, L.D. (2008). Subpolar mode water in the northeastern Atlantic: 1. Averaged properties and mean circulation. Journal of Geophysical Research: Ocean, 113, 1–18. DOI: 10.1029/2006JC004062

Broecker, W.S., & Peng, T.H. (1983). Tracers in the sea: W. S. Broecker and T. H. Peng. Eldigio Press Lamont Doherty Geological Observatory, 1982, 690 pages (300 figures and tables; 740 commented bibliographic references), US \$35.00. Geochimica et Cosmochimica Acta, 47, 1336. DOI: 10.1016/0016-7037(83)90075-3

Bryden, H.L., King, B.A., Mccarthy, G.D., & Mcdonagh, E.L. (2014). Impact of a 30 % reduction in Atlantic meridional overturning during 2009 – 2010 683–691. DOI: 10.5194/os-10-683-2014

Caldeira, K., & Wickett, M. (2005). Ocean model predictions of chemistry changes from carbon dioxide emissions to the atmosphere and ocean. Journal of Geophysical Research C: Oceans, 110, 1–12. DOI: 10.1029/2004JC002671

Caldeira, K., & Wickett, M.E. (2003). Anthropogenic carbon and ocean pH. Nature, 425, 365. DOI: 10.1038/425365a

Carpenter, J. H. (1965). The accuracy of the Winkler method for dissolved oxygen analysis 1. Limnology and Oceanography, 10(1), 135-140.

Carrit, D. E., & Carpenter, J. H. (1966). Recommendation procedure for Winkler analyses of sea water for dissolved oxygen. Journal of Marine Research, 24, 313-318.

Carter, B.R., Bittig, H.C., Fassbender, A.J., Sharp, J.D., Takeshita, Y., Xu, Y.Y., Álvarez, M., Wanninkhof, R., Feely, R.A., & Barbero, L. (2021). New and updated global empirical seawater property estimation routines. Limnology and Oceanography: Methods, 19, 785–809. DOI: 10.1002/lom3.10461

Clayton, T.D., & Byrne, R.H. (1993). Spectrophotometric seawater pH measurements: total hydrogen ion concentration scale calibration of m-cresol purple and at-sea results. Deep Sea Research Part I: Oceanographic Research Papers, 40, 2115–2129. DOI: 10.1016/0967-0637(93)90048-8

Collier, R., Dymond, J., Honjo, S., Manganini, S., Francois, R., & Dunbar, R. (2000). The vertical flux of biogenic and lithogenic material in the Ross Sea: Moored sediment trap





observations 1996-1998. Deep Sea Research Part II: Topical Studies in Oceanography, 47, 3491–3520. DOI: 10.1016/S0967-0645(00)00076-X

Corbière, A., Metzl, N., Reverdin, G., Brunet, C., & Takahashi, T. (2007). Interannual and decadal variability of the oceanic carbon sink in the North Atlantic subpolar gyre. Tellus, Series B: Chemical and Physical Meteorology, 59, 168–178. DOI: 10.1111/j.1600-0889.2006.00232.x

Daniault, N., Mercier, H., Lherminier, P., Sarafanov, A., Falina, A., Zunino, P., Pérez, F.F., & Ríos, A.F. (2016). The northern North Atlantic Ocean mean circulation in the early 21st century. Progress in Oceanography, 146, 142–158. DOI: 10.1016/j.pocean.2016.06.007

de la Paz, M., García-Ibáñez, M. I., Steinfeldt, R., Ríos, A. F., & Pérez, F. F. (2017). Ventilation versus biology: What is the controlling mechanism of nitrous oxide distribution in the North Atlantic?. Global Biogeochemical Cycles, 31(4), 745–760. DOI: 10.1002/2016GB005507

Dickson, A. G., & Goyet, C. (1994). Handbook of methods for the analysis of the various parameters of the carbon dioxide system in sea water. V. 2. United States. DOI: 10.2172/10107773

Dickson, A. G., Sabine, C. L., & Christian, J. R. (2007). Guide to best practices for ocean CO2 measurements. PICES Special Publication 3, 191 pp.

Dickson, B., Yashayaev, I., Meincke, J., Turrell, B., Dye, S., & Holfort, J. (2002). Rapid freshening of the deep North Atlantic Ocean over the past four decades. Nature, 416(6883), 832-837. DOI: 10.1038/416832a

De Jong, M. F., & de Steur, L. (2016). Strong winter cooling over the Irminger Sea in winter 2014–2015, exceptional deep convection, and the emergence of anomalously low SST. Geophysical Research Letters, 43, 7106–7113. DOI: 10.1002/2016GL069596

De Jong, M. F., Van Aken, H. M., Våge, K., & Pickart, R. S. (2012). Convective mixing in the central Irminger Sea: 2002-2010. Deep Sea Research Part I: Oceanographic Research Papers, 63, 36–51. DOI: 10.1016/j.dsr.2012.01.003

DelValls, T. A., & Dickson, A. G. (1998). The pH of buffers based on 2-amino-2-hydroxymethyl-1,3-propanediol ('tris') in synthetic sea water. Deep Sea Research Part I: Oceanographic Research Papers, 45, 1541–1554. DOI: 10.1016/S0967-0637(98)00019-3

Desbruyères, D., Thierry, V., & Mercier, H. (2013). Simulated decadal variability of the meridional overturning circulation across the A25-Ovide section, 462–475. DOI: 10.1029/2012JC008342

Dickson, A. G. (1990). Standard potential of the reaction: AgCl(s) + 1/2H2(g) = Ag(s) + HCl(aq), and the standard acidity constant of the ion HSO4- in synthetic sea water from 273.15 to 318.15 K. J. Chem. Thermodyn., 22, 113–127. DOI: 10.1016/0021-9614(90)90074-Z





Dickson, A. G., & Millero, F. J. (1987). A comparison of the equilibrium constants for the dissociation of carbonic acid in seawater media. Deep Sea Research Part A: Oceanographic Research Papers, 34, 1733–1743. DOI: 10.1016/0198-0149(87)90021-5

Dickson, R. R., & Brown, J. (1994). The production of North Atlantic Deep Water: Sources, rates, and pathways. J. Geophys. Res. Ocean., 99, 12319–12341. DOI: 10.1029/94JC00530

Doney, S. C., Fabry, V. J., Feely, R. A., & Kleypas, J. A. (2009). Ocean Acidification: The Other CO2 Problem. Ann. Rev. Mar. Sci., 1, 169–192. DOI: 10.1146/annurev.marine.010908.163834

Eden, C., & Willebrand, J. (2001). Mechanism of interannual to decadal variability of the North Atlantic circulation. J. Clim., 14, 2266–2280. DOI: 10.1175/1520-0442(2001)014<2266:MOITDV>2.0.CO;2

Ellett, D. J., Edwards, A., & Bowers, R. (1986). The hydrography of the Rockall Channel—an overview. Proc. R. Soc. Edinburgh. Sect. B. Biol. Sci., 88, 61–81. DOI: 10.1017/s0269727000004474

Feely, R. A., Sabine, C. L., Lee, K., Berelson, W., Kleypas, J., Fabry, V. J., Millero, F. J. (2004). Impact of anthropogenic CO2 on the CaCO3 system in the oceans. Science, 305, 362–366. DOI: 10.1126/SCIENCE.1097329

Ferron, B., Kokoszka, F., Mercier, H., Lherminier, P. (2014). Dissipation rate estimates from microstructure and finescale internal wave observations along the A25 Greenland-Portugal OVIDE line. J. Atmos. Ocean. Technol., 31, 2530–2543. DOI: 10.1175/JTECH-D-14-00036.1

Fogelqvist, E., Blindheim, J., Tanhua, T., Østerhus, S., Buch, E., & Rey, F. (2003). Greenland–Scotland overflow studied by hydro-chemical multivariate analysis. Deep Sea Research Part I: Oceanographic Research Papers, 50(1), 73-102.

Fontela, M., Pérez, F. F., Carracedo, L. I., Padín, X. A., Velo, A., García-Ibañez, M. I., Lherminier, P. (2020). The Northeast Atlantic is running out of excess carbonate in the horizon of cold-water corals communities. Sci. Rep., 10. DOI: 10.1038/s41598-020-71793-2

Friedlingstein, P., Sullivan, M. O., Jones, M. W., Andrew, R. M., Gregor, L., Hauck, J., Quéré, C. Le, Luijkx, I. T., Olsen, A., Peters, G. P., Peters, W. (2022). Global Carbon Budget 2022, 4811–4900.

Fröb, F., Olsen, A., Becker, M., Chafik, L., Johannessen, T., Reverdin, G., Omar, A. (2019). Wintertime fCO2 Variability in the Subpolar North Atlantic Since 2004. Geophys. Res. Lett., 46, 1580–1590. DOI: 10.1029/2018GL080554

Fröb, F., Olsen, A., Våge, K., Moore, G. W. K., Yashayaev, I., Jeansson, E., Rajasakaren, B. (2016). Irminger Sea deep convection injects oxygen and anthropogenic carbon to the ocean interior. Nat. Commun., 7. DOI: 10.1038/ncomms13244





García-Ibáñez, M. I., Bates, N. R., Bakker, D. C. E., Fontela, M., Velo, A. (2021). Cold-water corals in the Subpolar North Atlantic Ocean exposed to aragonite undersaturation if the 2 °C global warming target is not met. Global Planetary Change, 201. DOI: 10.1016/j.gloplacha.2021.103480

García-Ibáñez, M. I., Pardo, P. C., Carracedo, L. I., Mercier, H., Lherminier, P., Ríos, A. F., Pérez, F. F. (2015). Structure, transports and transformations of the water masses in the Atlantic Subpolar Gyre. Progress in Oceanography, 135, 18-36. DOI: 10.1016/j.pocean.2015.03.009

García-Ibáñez, M. I., Pérez, F. F., Lherminier, P., Zunino, P., Mercier, H., Tréguer, P. (2018). Water mass distributions and transports for the 2014 GEOVIDE cruise in the North Atlantic. Biogeosciences, 15, 2075-2090. DOI: 10.5194/bg-15-2075-2018

García-Ibáñez, M. I., Zunino, P., Fröb, F., Carracedo, L. I., Ríos, A. F., Mercier, H., Olsen, A., Pérez, F. F. (2016). Ocean acidification in the subpolar North Atlantic: Rates and mechanisms controlling pH changes. Biogeosciences, 13, 3701-3715. DOI: 10.5194/bg-13-3701-2016

Gladyshev, S. V., Gladyshev, V. S., Falina, A. S., Sarafanov, A. A. (2016a). Winter convection in the Irminger Sea in 2004–2014. Oceanology, 56, 326-335. DOI: 10.1134/S0001437016030073

Gladyshev, S. V., Gladyshev, V. S., Gulev, S. K., Sokov, A. V. (2016b). Anomalously deep convection in the Irminger Sea during the winter of 2014–2015. Doklady Earth Sciences, 469, 766-770. DOI: 10.1134/S1028334X16070229

Gladyshev, S. V., Gladyshev, V. S., Gulev, S. K., Sokov, A. V. (2016c). Anomalously deep convection in the Irminger Sea during the winter of 2014–2015. Doklady Earth Sciences, 469, 766-770. DOI: 10.1134/S1028334X16070229

Gladyshev, S. V., Gladyshev, V. S., Member, C., Gulev, R. A. S. S. K., Sokov, A. V. (2018). Structure and Variability of the Meridional Overturning Circulation in the North Atlantic Subpolar Gyre, 2007–2017. Doklady Earth Sciences, 483, 1524-1527. DOI: 10.1134/S1028334X18120024

Gladyshev, S. V., Gladyshev, V. S., Member, C., Gulev, R. A. S. S. K., Sokov, A. V. (2017). Subpolar Mode Water Classes in the Northeast Atlantic: Interannual and Long-Term Variability. Doklady Earth Sciences, 476, 1203-1206. DOI: 10.1134/S1028334X17100166

González-Dávila, M., Santana-Casiano, J. M., Petihakis, G., Ntoumas, M., Suárez de Tangil, M., Krasakopoulou, E. (2016). Seasonal pH variability in the Saronikos Gulf: A year-study using a new photometric pH sensor. Journal of Marine Systems, 162, 37-46. DOI: 10.1016/j.jmarsys.2016.03.007

González-Dávila, M., Santana-Casiano, J. M., Prêcheur-Massieu, H. (2014). New pH sensor for monitoring ocean acidification. Sea Technology, 55, 36-40.





González-Dávila, M., Santana-Casiano, J.M., Rueda, M.J., Llinás, O. (2010). The water column distribution of carbonate system variables at the ESTOC site from 1995 to 2004. Biogeosciences, 7, 3067-3081. DOI: 10.5194/bg-7-3067-2010

Gruber, N., Clement, D., Carter, B.R., Feely, R.A., van Heuven, S., Hoppema, M., Ishii, M., Key, R.M., Kozyr, A., Lauvset, S.K., Monaco, C. Lo, Mathis, J.T., Murata, A., Olsen, A., Perez, F.F., Sabine, C.L., Tanhua, T., Wanninkhof, R. (2019a). The oceanic sink for anthropogenic CO2 from 1994 to 2007. Science, 363, 1193

Gruber, N., Sarmiento, J.L., Stocker, T.F. (1996). An improved method for detecting anthropogenic CO2 in the oceans. Global Biogeochem. Cycles. https://doi.org/10.1029/96GB01608

Guallart, E.F., Fajar, N.M., Padín, X.A., Vázquez-Rodríguez, M., Calvo, E., Ríos, A.F., Hernández-Guerra, A., Pelejero, C., Pérez, F.F. (2015). Ocean acidification along the 24.5°N section in the subtropical North Atlantic. Geophys. Res. Lett., 42, 450–458. https://doi.org/10.1002/2014GL062971

Häkkinen, S. (2002). Surface salinity variability in the northern North Atlantic during recent decades. J. Geophys. Res. Ocean, 107, SRF 4-1. https://doi.org/10.1029/2001JC000812

Häkkinen, S., Rhines, P.B. (2004). Decline of Subpolar North Atlantic Circulation during the 1990s. Science (80-.), 304, 555–559. https://doi.org/10.1126/science.1094917

Harvey, J. (1982). Theta–S relationships and water masses in the eastern North Atlantic. Deep-Sea Res. Part a-Oceanogr. Res. Pap., 29 (8), 1021–1033.

Hátún, H., Sande, A.B., Drange, H., Hansen, B., Valdimarsson, H. (2005). Influence of the Atlantic subpolar gyre on the thermohaline circulation. Science (80-.), 309, 1841–1844. https://doi.org/10.1126/science.1114777

Holliday, N.P., Bersch, M., Berx, B., Chafik, L., Cunningham, S., Florindo-López, C., Hátún, H., Johns, W., Josey, S.A., Larsen, K.M.H., Mulet, S., Oltmanns, M., Reverdin, G., Rossby, T., Thierry, V., Valdimarsson, H., Yashayaev, I. (2020). Ocean circulation causes the largest freshening event for 120 years in eastern subpolar North Atlantic. Nat. Commun., 11. https://doi.org/10.1038/s41467-020-14474-y

Holliday, P. N., Pollard, R.T., Read, J.F., Leach, H. (2000). Water mass properties and fluxes in the Rockall Trough, 1975-1998. Deep-Sea Research Part I: Oceanographic Research Papers. <a href="https://doi.org/10.1016/S0967-0637(99)00109-0">https://doi.org/10.1016/S0967-0637(99)00109-0</a>

Humphreys, M. P., et al. (2016). Multidecadal accumulation of anthropogenic and remineralized dissolved inorganic carbon along the Extended Ellett Line in the northeast Atlantic Ocean. Global Biogeochem. Cycles. 30, 293–310, doi:10.1002/2015GB005246.

IPCC (2007). Climate Change 2007: The Physical Science Basis. Contribution of Working Group I to the Fourth Assessment Report of the Intergovernmental Panel on Climate Change.





IPCC (2021). Climate Change 2021: The Physical Science Basis. Contribution of Working Group I to the Sixth Assessment Report of the Intergovernmental Panel on Climate Change.

IPCC Working Group (2013). Climate Change 2013: The Physical Science Basis. Working Group I Contribution to the Fifth Assessment Report of the Intergovernmental Panel on Climate Change (Cambridge Univ. Press).

Jackson, L.C., Biastoch, A., Buckley, M.W., Desbruyères, D. G., Frajka-Williams, E., Moat, B., and Robson, J. (2022). The evolution of the North Atlantic Meridional Overturning Circulation since 1980. Nat. Rev. Earth Environ. 3, 241–254. https://doi.org/10.1038/s43017-022-00263-2

Jiang, L.Q., Feely, R.A., Carter, B.R., Greeley, D.J., Gledhill, D.K., Arzayus, K.M. (2015). Climatological distribution of aragonite saturation state in the global oceans. Global Biogeochem. Cycles, 29, 1656–1673. https://doi.org/10.1002/2015GB005198

Johnson, K.M., Wills, K.D., Butler, D.B., Johnson, W.K., Wong, C.S. (1993). Coulometric total carbon dioxide analysis for marine studies: maximizing the performance of an automated gas extraction system and coulometric detector. Mar. Chem., 44, 167–187. https://doi.org/10.1016/0304-4203(93)90201-X

Josey, S.A., Hirschi, J.J.M., Sinha, B., Duchez, A., Grist, J.P., Marsh, R. (2018). The recent Atlantic cold anomaly: Causes, consequences, and related phenomena. Ann. Rev. Mar. Sci., 10, 475–501. https://doi.org/10.1146/annurev-marine-121916-063102

Khatiwala, S., Tanhua, T., Mikaloff Fletcher, S., Gerber, M., Doney, S.C., Graven, H.D., Gruber, N., McKinley, G.A., Murata, A., Ríos, A.F., Sabine, C.L. (2013). Global ocean storage of anthropogenic carbon. Biogeosciences, 10, 2169–2191. https://doi.org/10.5194/bg-10-2169-2013

Kieke, D., Rhein, M., Stramma, L., Smethie, W. M., Bullister, J. L., & LeBel, D. A. (2007). Changes in the pool of Labrador Sea Water in the subpolar North Atlantic. Geophysical Research Letters, 34(6).

Knight, J.R., Folland, C.K., Scaife, A.A. (2006). Climate impacts of the Atlantic multidecadal oscillation. Geophys. Res. Lett., 33, 2–5. https://doi.org/10.1029/2006GL026242

Langdon, C., Takahashi, T., Sweeney, C., Chipman, D., Atkinson, J. (2000). Rate of an experimental coral reef responds to manipulations in the concentrations of both CaCO3. Global Biogeochem. Cycles, 14, 639–654.

Lazier, J., Hendry, R., Clarke, A., Yashayaev, I., Rhines, P. (2002). Convection and restratification in the Labrador Sea, 1990–2000. Deep Sea Res. Part I Oceanogr. Res. Pap., 49, 1819–1835. https://doi.org/10.1016/S0967-0637(02)00064-X

Lee, K., Kim, T.W., Byrne, R.H., Millero, F.J., Feely, R.A., Liu, Y.M. (2010). The universal ratio of boron to chlorinity for the North Pacific and North Atlantic oceans. Geochim. Cosmochim. Acta, 74, 1801–1811. https://doi.org/10.1016/j.gca.2009.12.027





Leseurre, C., Lo Monaco, C., Reverdin, G., Metzl, N., Fin, J., Olafsdottir, S., Racapé, V. (2020). Ocean carbonate system variability in the North Atlantic Subpolar surface water (1993–2017). Biogeosciences, 17, 2553–2577. https://doi.org/10.5194/bg-17-2553-2020

Lewis, E., Wallace, D. (1998). Program Developed for CO2 System Calculations ORNL/CDIAC-105, Carbon Dioxide Information Analysis Centre.

Lherminier, P., Mercier, H., Huck, T., Gourcuff, C., Perez, F. F., Morin, P., ... & Falina, A. (2010). The Atlantic Meridional Overturning Circulation and the subpolar gyre observed at the A25-OVIDE section in June 2002 and 2004. Deep Sea Research Part I: Oceanographic Research Papers, 57(11), 1374-1391.

Lueker, T.J., Dickson, A.G., Keeling, C.D. (2000). Ocean pCO2 calculated from dissolved inorganic carbon, alkalinity, and equations for K1 and K2: Validation based on laboratory measurements of CO2 in gas and seawater at equilibrium. Mar. Chem., 70, 105–119. https://doi.org/10.1016/S0304-4203(00)00022-0

Marsh, R., de Cuevas, B.A., Coward, A.C., Bryden, H.L., Álvarez, M. (2005). Thermohaline circulation at three key sections in the North Atlantic over 1985–2002. Geophys. Res. Lett., 32, 1–4. https://doi.org/10.1029/2004GL022281

Matear, R.J., Lenton, A. (2014). Quantifying the impact of ocean acidification on our future climate. Biogeosciences, 11, 3965–3983. https://doi.org/10.5194/bg-11-3965-2014

Maier-Reimer, E., & Hasselmann, K. (1987). Transport and storage of CO2 in the ocean: an inorganic ocean-circulation carbon cycle model. Climate dynamics, 2, 63-90.

Maier, C., Hegeman, J., Weinbauer, M. G., & Gattuso, J. P. (2009). Calcification of the cold-water coral Lophelia pertusa, under ambient and reduced pH. Biogeosciences, 6(8), 1671-1680. https://doi.org/10.5194/bg-6-1671-2009

Mauritzen, C., Häkkinen, S. (1999). On the relationship between dense water formation and the "Meridional Overturning Cell" in the North Atlantic Ocean. Deep. Res. Part I Oceanogr. Res. Pap., 46, 877–894. https://doi.org/10.1016/S0967-0637(98)00094-6

McDonagh, E.L., King, B.A., Bryden, H.L., Courtois, P., Szuts, Z., Baringer, M., Cunningham, S.A., Atkinson, C., McCarthy, G. (2015). Continuous estimate of Atlantic oceanic freshwater flux at 26.5°N. Journal of Climate, 28, 8888–8906. https://doi.org/10.1175/JCLI-D-14-00519.1

McGrath, T., Kivimäe, C., McGovern, E., Cave, R.R., Joyce, E. (2013). Winter measurements of oceanic biogeochemical parameters in the Rockall Trough (2009-2012). Earth System Science Data, 5, 375–383. https://doi.org/10.5194/essd-5-375-2013

McGrath, T., Kivimäe, C., Tanhua, T., Cave, R.R., McGovern, E. (2012a). Inorganic carbon and pH levels in the Rockall Trough 1991-2010. Deep Sea Research Part I: Oceanographic Research Papers, 68, 79–91. https://doi.org/10.1016/j.dsr.2012.05.011





McGrath, T., Nolan, G., McGovern, E. (2012b). Chemical characteristics of water masses in the Rockall Trough. Deep Sea Research Part I: Oceanographic Research Papers, 61, 57–73. https://doi.org/10.1016/j.dsr.2011.11.007

McCartney, M. S., & Talley, L. D. (1982). The subpolar mode water of the North Atlantic Ocean. Journal of Physical Oceanography, 12(11), 1169-1188.

Mercier, H., Lherminier, P., Sarafanov, A., Gaillard, F., Daniault, N., Desbruyères, D., Falina, A., Ferron, B., Gourcuff, C., Huck, T., Thierry, V. (2015). Variability of the meridional overturning circulation at the Greenland – Portugal OVIDE section from 1993 to 2010. Progress in Oceanography, 132, 250–261. https://doi.org/10.1016/j.pocean.2013.11.001

Messias, M.J., Watson, A.J., Johannessen, T., Oliver, K.I.C., Olsson, K.A., Fogelqvist, E., Olafsson, J., Bacon, S., Balle, J., Bergman, N., Budéus, G., Danielsen, M., Gascard, J.C., Jeansson, E., Olafsdottir, S.R., Simonsen, K., Tanhua, T., Van Scoy, K., Ledwell, J.R. (2008). The Greenland Sea tracer experiment 1996–2002: Horizontal mixing and transport of Greenland Sea Intermediate Water. Progress in Oceanography, 78, 85–105. https://doi.org/10.1016/J.POCEAN.2007.06.005

Millero, F.J., Zhang, J., Lee, K., Campbell, D.M. (1993). Titration alkalinity of seawater. 44, 153–165.

Mintrop, L., Pérez, F.F., González-Dávila, M., Santana-Casiano, J.M., Körtzinger, A. (2000). Alkalinity determination by potentiometry: Intercalibration using three different methods. Ciencias Marinas, 26, 23–37. https://doi.org/10.7773/cm.v26i1.573

Mucci, A. (1983). The solubility of calcite and aragonite in seawater at various salinities, temperatures, and one atmosphere total pressure. American Journal of Science, 283(7), 780-799.

Olafsson, J., Olafsdottir, S.R., Benoit-Cattin, A., Danielsen, M., Arnarson, T.S., Takahashi, T. (2009). Rate of Iceland Sea acidification from time series measurements. Biogeosciences, 6, 2661–2668. https://doi.org/10.5194/bg-6-2661-2009

Olafsson, J., Olafsdottir, S.R., Benoit-Cattin, A., Takahashi, T. (2010). The Irminger Sea and the Iceland Sea time series measurements of seawater carbon and nutrient chemistry 1983-2008. Earth System Science Data, 2, 99–104. https://doi.org/10.5194/essd-2-99-2010

Orr, J. C. (2011). Recent and future changes in ocean carbonate chemistry. In Ocean Acidification (Vol. 1), Oxford University Press, 41–66.

Orr, J.C., Epitalon, J.-M., Dickson, A. G., Gattuso, J.-P., (2018). Routine uncertainty propagation for the marine carbon dioxide system. Marine Chemistry 207, 84-107. https://doi.org/10.1016/j.marchem.2018.10.006

Orr, J.C., Fabry, V.J., Aumont, O., Bopp, L., Doney, S.C., Feely, R.A., Gnanadesikan, A., Gruber, N., Ishida, A., Joos, F., Key, R.M., Lindsay, K., Maier-Reimer, E., Matear, R., Monfray, P., Mouchet, A., Najjar, R.G., Plattner, G.K., Rodgers, K.B., Sabine, C.L., Sarmiento, J.L.,





Schlitzer, R., Slater, R.D., Totterdell, I.J., Weirig, M.F., Yamanaka, Y., Yool, A. (2005). Anthropogenic ocean acidification over the twenty-first century and its impact on calcifying organisms. Nature, 437, 681–686. [https://doi.org/10.1038/nature

Pascale, L., Perez, F. F., Branellec, P., Mercier, H., Velo, A., Messias, M. J., Castrillejo, M., Reverdin, G., Fontela, M., Baurand, F. (2022). GO-SHIP A25 - OVIDE 2018 Cruise data. SEANOE. https://doi.org/10.17882/87394

Pérez, F.F., Fontela, M., García-Ibáñez, M.I., Mercier, H., Velo, A., Lherminier, P., Zunino, P., De La Paz, M., Alonso-Pérez, F., Guallart, E.F., Padin, X.A. (2018). Meridional overturning circulation conveys fast acidification to the deep Atlantic Ocean. Nature, 554, 515–518. https://doi.org/10.1038/nature25493

Pérez, F.F., Fraga, F. (1987). Association constant of fluoride and hydrogen ions in seawater. Marine Chemistry, 21, 161–168. https://doi.org/10.1016/0304-4203(87)90036-3

Pérez, F.F., Mercier, H., Vázquez-Rodríguez, M., Lherminier, P., Velo, A., Pardo, P.C., Rosón, G., Ríos, A.F. (2013). Atlantic Ocean CO2 uptake reduced by weakening of the meridional overturning circulation. Nature Geoscience, 6, 146–152. https://doi.org/10.1038/ngeo1680

Pérez, F.F., Olafsson, J., Ólafsdóttir, S.R., Fontela, M., Takahashi, T. (2021). Contrasting drivers and trends of ocean acidification in the subarctic Atlantic. Scientific Reports, 11, 1–16. https://doi.org/10.1038/s41598-021-93324-3

Pérez, F.F., Vázquez-Rodríguez, M., Louarn, E., Padín, X.A., Mercier, H., Ríos, A.F. (2008). Temporal variability of the anthropogenic CO2 storage in the Irminger Sea. Biogeosciences, 5, 1669–1679. https://doi.org/10.5194/bg-5-1669-2008

Pérez, F.F., Vázquez-Rodríguez, M., Mercier, H., Velo, A., Lherminier, P., Ríos, A.F. (2010). Trends of anthropogenic CO2 storage in North Atlantic water masses. Biogeosciences, 7, 1789–1807. <a href="https://doi.org/10.5194/bg-7-1789-2010">https://doi.org/10.5194/bg-7-1789-2010</a>

Perez, F. F., Becker, M., Goris, N., Gehlen, M., Lopez-Mozos, M., Tjiputra, J., Olsen, A., Müller, J. D., Huertas, I. E., Chau, T. T. T., Cainzos, V., Velo, A., Benard, G., Hauck, J., Gruber, N., Wanninkhof, R. (2024). An assessment of CO2 storage and sea-air fluxes for the Atlantic Ocean and Mediterranean Sea between 1985 and 2018. Global Biogeochemical Cycles, 38, e2023GB007862. https://doi.org/10.1029/2023GB007862

Pickart, R.S., Spall, M.A., Ribergaard, M.H., Moore, G.W.K., Milliff, R.F. (2003). Deep convection in the Irminger Sea forced by the Greenland tip jet. Nature, 424, 152–156. https://doi.org/10.1038/nature01729

Piron, A., Thierry, V., Mercier, H., Caniaux, G. (2017). Gyre-scale deep convection in the subpolar North Atlantic Ocean during winter 2014–2015. Geophysical Research Letters, 44, 1439–1447. https://doi.org/10.1002/2016GL071895

Pollard, R.T., Griffiths, M.J., Cunningham, S.A., Read, J.F., Pérez, F.F., Ríos, A.F. (1996). Vivaldi 1991 - A study of the formation, circulation, and ventilation of Eastern North Atlantic





Central Water. Progress in Oceanography, 37, 167–172. https://doi.org/10.1016/S0079-6611(96)00008-0

Portner, H.O., Langenbuch, M., & Reipschläger, A. (2004). Biological Impact of Elevated Ocean CO2 Concentrations: Lessons from Animal Physiology and Earth History. Journal of Oceanography, 60, 705–718.

Pörtner, H. O., et al. (2019). IPCC Special Report on the Ocean and Cryosphere in a Changing Climate. Wiley IPCC Intergovernmental Panel on Climate Change.

Raven, J., Caldeira, K., Elderfield, H., Hoegh-Guldberg, O., Liss, P., Riebesell, U., Shepherd, J., Turley, C., & Watson, A. (2005). Ocean acidification due to increasing. Coral Reefs, 12/05, 68.

Read, J.F. (2000). CONVEX-91: Water masses and circulation of the Northeast Atlantic subpolar gyre. Progress in Oceanography, 48, 461–510. https://doi.org/10.1016/S0079-6611(01)00011-8

Riebesell, U., Zondervan, I., Rost, B., Tortell, P.D., Zeebe, R.E., Morel, F.M.M. (2000). Reduced calcification of marine plankton in response to increased atmospheric CO2. Nature, 407, 364–366. https://doi.org/10.1038/35030078

Ríos, A.F., Resplandy, L., García-Ibáñez, M.I., Fajar, N.M., Velo, A., Padin, X.A., Wanninkhof, R., Steinfeldt, R., Rosón, G., Pérez, F.F., Morel, F.M.M. (2015). Decadal acidification in the water masses of the Atlantic Ocean. Proceedings of the National Academy of Sciences, U.S.A., 112, 9950–9955. https://doi.org/10.1073/pnas.1504613112

Roberts, J.M., Wheeler, A.J., Freiwald, A., Cairns, S. (2009). Cold-Water Corals: The Biology and Geology of Deep-Sea Coral Habitats. Cambridge University Press. https://doi.org/10.1017/CBO9780511581588.

Robson, J., Hodson, D., Hawkins, E., Sutton, R. (2014). Atlantic overturning in decline? Nature Geoscience, 7, 2–3. https://doi.org/10.1038/ngeo2050

Robson, J., Ortega, P., Sutton, R. (2016). A reversal of climatic trends in the North Atlantic since 2005. Nature Geoscience, 9, 513–517. https://doi.org/10.1038/ngeo2727

Rodgers, K.B., Key, R.M., Gnanadesikan, A., Sarmiento, J.L., Aumont, O., Bopp, L., Doney, S.C., Dunne, J.R., Glover, D.M., Ishida, A., Ishii, M., Jacobson, A.R., Monaco, C. Lo, Maier-Reimer, E., Mercier, H., Metzl, N., Pérez, F.F., Rios, A.F., Wanninkhof, R., Wetzel, P., Winn, C.D., Yamanaka, Y. (2009). Using altimetry to help explain patchy changes in hydrographic carbon measurements. Journal of Geophysical Research: Ocean, 114, 1–20. https://doi.org/10.1029/2008JC005183

Sabine, C.L., Feely, R.A., Gruber, N., Key, R.M., Lee, K., Bullister, J.L., Wanninkhof, R., Wong, C.S., Wallace, D.W.R., Tilbrook, B., Millero, F.J., Peng, T.H., Kozyr, A., Ono, T., Rios, A.F. (2004a). The oceanic sink for anthropogenic CO2. Science, 305, 367–371. https://doi.org/10.1126/science.1097403





Sabine, C.L., Feely, R.A., Gruber, N., Key, R.M., Lee, K., Bullister, J.L., Wanninkhof, R., Wong, C.S., Wallace, D.W.R., Tilbrook, B., Millero, F.J., Peng, T.H., Kozyr, A., Ono, T., Rios, A.F. (2004b). The oceanic sink for anthropogenic CO2. Science, 305, 367–371. https://doi.org/10.1126/SCIENCE.1097403/SUPPL FILE/SABINE.SOM.PDF

Santana-Casiano, J. M., González-Dávila, M., and Curbelo-Hernández, D. (2023). Surface-to-bottom data of total alkalinity, total inorganic carbon, pH and dissolved oxygen in the subpolar North Atlantic along the CLIVAR 59.5N hydrographic section during 2009-2019. [Data set]. Zenodo. https://doi.org/10.5281/zenodo.10276222

Santana-Casiano, J.M., González-Dávila, M., Rueda, M.-J., Llinás, O., González-Dávila, E.-F. (2007). The interannual variability of oceanic CO2 parameters in the northeast Atlantic subtropical gyre at the ESTOC site. Global Biogeochemical Cycles, 21. https://doi.org/10.1029/2006GB002788

Sarafanov, A., Falina, A., Mercier, H., Sokov, A., Lherminier, P., Gourcuff, C., Gladyshev, S., Gaillard, F., and Daniault, N. (2012). Mean full-depth summer circulation and transports at the northern periphery of the Atlantic Ocean in the 2000s. Journal of Geophysical Research: Oceans, 117(C1).

Sarafanov, A., Falina, A., Sokov, A., Zapotylko, V., Gladyshev, S. (2018). Ship-Based Monitoring of the Northern North Atlantic Ocean by the Shirshov Institute of Oceanology. The Main Results. In: Velarde, M., Tarakanov, R., Marchenko, A. (eds) The Ocean in Motion. Springer Oceanography. Springer, Cham. https://doi.org/10.1007/978-3-319-71934-4 25

Sarafanov, A., Mercier, H., Falina, A., and Sokov, A. (2010). Cessation and partial reversal of deep water freshening in the northern North Atlantic: observation-based estimates and attribution. Tellus A: Dynamic Meteorology and Oceanography, 62:1, 80-90. https://doi.org/10.1111/j.1600-0870.2009.00418.x

Sarmiento, J.L., Orr, J.C., Siegenthaler, U. (1992). A perturbation simulation of CO2 uptake in an ocean general circulation model. Journal of Geophysical Research: Ocean, 97, 3621–3645. https://doi.org/10.1029/91JC02849

Saunders, P. M., (2001). Chapter 5.6 The dense northern overflows. Int. Geophys., 77, 401–417. <a href="https://doi.org/10.1016/S0074-6142(01)80131-5">https://doi.org/10.1016/S0074-6142(01)80131-5</a>

Sauzède, R., Bittig, H.C., Claustre, H., de Fommervault, O.P., Gattuso, J.P., Legendre, L., Johnson, K.S. (2017). Estimates of water-column nutrient concentrations and carbonate system parameters in the global ocean: A novel approach based on neural networks. Frontiers in Marine Science, 4, 1–17. https://doi.org/10.3389/fmars.2017.00128

Schlitzer, R. (2021). Ocean Data View. Available at: https://odv.awi.de.

Schmitz Jr, W. J., & McCartney, M. S. (1993). On the north Atlantic circulation. Reviews of Geophysics, 31(1), 29-49.





Schott, F. A., & Brandt, P. (2007). Circulation and deep water export of the subpolar North Atlantic during the 1990's. Washington DC American Geophysical Union Geophysical Monograph Series, 173, 91-118.

Sharp, J.D., Pierrot, D., Humphreys, M.P., Epitalon, J.-M., Orr, J.C., Lewis, E.R., Wallace, D.W.R. (2023, Jan. 19). CO2SYSv3 for MATLAB (Version v3.2.1). Zenodo. http://doi.org/10.5281/zenodo.3950562

Smeed, D.A., Josey, S.A., Beaulieu, C., Johns, W.E., Moat, B.I., Frajka-Williams, E., Rayner, D., Meinen, C.S., Baringer, M.O., Bryden, H.L., McCarthy, G.D. (2018). The North Atlantic Ocean Is in a State of Reduced Overturning. Geophysical Research Letters, 45, 1527–1533. https://doi.org/10.1002/2017GL076350

Steinfeldt, R., Rhein, M., Bullister, J.L., Tanhua, T. (2009). Inventory changes in anthropogenic carbon from 1997 – 2003 in the Atlantic Ocean between 20 ° S and 65 ° N, 23, 1–11. https://doi.org/10.1029/2008GB003311

Stephens, T. (2021). Ocean Acidification. In: Research Handbook on Law, Governance and the Planetary Boundaries. 22, 295–308. https://doi.org/10.4337/9781789902747.00025

Stramma, L., Kieke, D., Rhein, M., Schott, F., Yashayaev, I., Koltermann, K.P. (2004). Deep water changes at the western boundary of the subpolar North Atlantic during 1996 to 2001. Deep Sea Research Part I: Oceanographic Research Papers, 51, 1033–1056. https://doi.org/10.1016/J.DSR.2004.04.001

Sutherland, D. A., and Pickart, R. S. (2008). The East Greenland coastal current: Structure, variability, and forcing. Progress in Oceanography, 78(1), 58-77.

Takahashi, T., Olafsson, J., Goddard, J.G., Chipman, D.W., Sutherland, S.C. (1993). Seasonal variation of CO2 and nutrients in the high-latitude surface oceans: A comparative study. Global Biogeochemical Cycles, 7, 843–878. https://doi.org/10.1029/93GB02263

Takahashi, T., Sutherland, S.C., Wanninkhof, R., Sweeney, C., Feely, R.A., Chipman, D.W., Hales, B., Friederich, G., Chavez, F., Sabine, C., Watson, A., Bakker, D.C.E., Schuster, U., Metzl, N., Yoshikawa-Inoue, H., Ishii, M., Midorikawa, T., Nojiri, Y., Körtzinger, A., Steinhoff, T., Hoppema, M., Olafsson, J., Arnarson, T.S., Tilbrook, B., Johannessen, T., Olsen, A., Bellerby, R., Wong, C.S., Delille, B., Bates, N.R., de Baar, H.J.W. (2009). Climatological mean and decadal change in surface ocean pCO2, and net sea-air CO2 flux over the global oceans. Deep Sea Research Part II: Topical Studies in Oceanography, 56, 554–577. https://doi.org/10.1016/j.dsr2.2008.12.009

Tanhua, T., Körtzinger, A., Friis, K., Waugh, D.W., Wallace, D.W.R. (2007). An estimate of anthropogenic CO2 inventory from decadal changes in oceanic carbon content. Proceedings of the National Academy of Sciences, U.S.A., 104, 3037–3042. https://doi.org/10.1073/pnas.0606574104





Tesdal, J.E., Abernathey, R.P., Goes, J.I., Gordon, A.L., Haine, T.W.N. (2018). Salinity trends within the upper layers of the subpolar North Atlantic. Journal of Climate, 31, 2675–2698. https://doi.org/10.1175/JCLI-D-17-0532.1

Thomas, H., Prowe, A.E.F., Lima, I.D., Doney, S.C., Wanninkhof, R., Greatbatch, R.J., Schuster, U., Corbière, A. (2008). Changes in the North Atlantic Oscillation influence CO2 uptake in the North Atlantic over the past 2 decades. Global Biogeochemical Cycles, 22, 1–13. https://doi.org/10.1029/2007GB003167

Tjiputra, J.F., Olsen, A., Bopp, L., Lenton, A., Pfeil, B., Roy, T., Segschneider, J., Totterdell, I., Heinze, C. (2014). Long-term surface pCO2 trends from observations and models. Tellus, Series B: Chemical and Physical Meteorology, 66, 1–24. https://doi.org/10.3402/tellusb.v66.23083

Touratier, F., Azouzi, L., Goyet, C. (2007). CFC-11, Δ14C and 3H tracers as a means to assess anthropogenic CO2 concentrations in the ocean. Tellus, Series B: Chemical and Physical Meteorology, 59, 318–325. https://doi.org/10.1111/j.1600-0889.2006.00247.x

Tsuchiya, M., Talley, L.D., McCartney, M.S. (1992). An eastern Atlantic section from Iceland southward across the equator. Deep Sea Research Part A: Oceanographic Research Papers, 39, 1885–1917. https://doi.org/10.1016/0198-0149(92)90004-D

Urban-Rich, J., Dagg, M., Peterson, J. (2001). Copepod grazing on phytoplankton in the Pacific sector of the Antarctic polar front. Deep Sea Research Part II: Topical Studies in Oceanography, 48, 4223–4246. https://doi.org/10.1016/S0967-0645(01)00087-X

Våge, K., Pickart, R.S., Thierry, V., Reverdin, G., Lee, C.M., Petrie, B., Agnew, T.A., Wong, A., Ribergaard, M.H. (2009). Surprising return of deep convection to the subpolar North Atlantic Ocean in winter 2007-2008. Nature Geoscience, 2, 67–72. https://doi.org/10.1038/ngeo382

Van Aken, H.M., and Becker, G. (1996). Hydrography and through-flow in the north-eastern North Atlantic Ocean: the NANSEN project. Prog. Oceanogr. 38, 297–346. https://doi.org/10.1016/S0079-6611(97)00005-0

Van Aken, H. M., and De Boer, C. J. (1995). On the synoptic hydrography of intermediate and deep water masses in the Iceland Basin. Deep Sea Research Part I: Oceanographic Research Papers, 42(2), 165-189.

Van Heuven, S., D. Pierrot, J.W.B. Rae, E. Lewis, and D.W.R. Wallace (2011). MATLAB Program Developed for CO2 System Calculations. ORNL/CDIAC-105b. Carbon Dioxide Information Analysis Center, Oak Ridge National Laboratory, U.S. Department of Energy, Oak Ridge, Tennessee. https://doi.org/10.3334/CDIAC/otg.CO2SYS MATLAB v1.1

Vázquez-Rodríguez, M., Padin, X.A., Pardo, P.C., Ríos, A.F., Pérez, F.F. (2012). The subsurface layer reference to calculate preformed alkalinity and air-sea CO2 disequilibrium in the Atlantic Ocean. J. Mar. Syst. 94, 52–63. https://doi.org/10.1016/j.jmarsys.2011.10.008





Vázquez-Rodríguez, M., Pérez, F.F., Velo, A., Ríos, A.F., Mercier, H. (2012). Observed acidification trends in North Atlantic water masses. Biogeosciences 9, 5217–5230. https://doi.org/10.5194/bg-9-5217-2012

Vázquez-Rodríguez, M., Touratier, F., Monaco, C. Lo, Waugh, D.W., Padin, X.A., Bellerby, R.G.J., Goyet, C., Metzl, N., Ríos, A.F., Pérez, F.F. (2009). Anthropogenic carbon distributions in the Atlantic Ocean: Data-based estimates from the Arctic to the Antarctic. Biogeosciences 6, 439–451. https://doi.org/10.5194/BG-6-439-2009

Wallace, D. W. (2001). Storage and transport of excess CO2 in the oceans: The JGOFS/WOCE global CO2 survey. In International Geophysics (Vol. 77, pp. 489-L). Academic Press. https://doi.org/10.1016/S0074-6142(01)80136-4

Watson, A.J., Schuster, U., Bakker, D.C.E., Bates, N.R., Corbière, A., González-Davila, M., Friedrich, T., Hauck, J., Heinze, C., Johannessen, T., Körtzinger, A., Metzl, N., Olafsson, J., Olsen, A., Oschlies, A., Antonio Padin, X., Pfeil, B., Magdalena Santana-Casiano, J., Steinhoff, T., Telszewski, M., Rios, A.F., Wallace, D.W.R., Wanninkhof, R. (2009). Tracking the variable North Atlantic sink for atmospheric CO2. Science (80-.). 326, 1391–1393. https://doi.org/10.1126/science.1177394

Winkler, L. W. (1888). Die bestimmung des im wasser gelösten sauerstoffes. Berichte der deutschen chemischen Gesellschaft, 21(2), 2843-2854.

Xu, X., Hurlburt, H.E., Jr, W.J.S., Zantopp, R., Fischer, J., Hogan, P.J. (2013). On the currents and transports connected with the atlantic meridional overturning circulation in the subpolar North Atlantic. J. Geophys. Res. 118, 502–516. https://doi.org/10.1002/jgrc.20065

Yashayaev, I. (2007). Hydrographic changes in the Labrador Sea, 1960–2005. Progress in Oceanography, 73(3-4), 242-276. https://doi.org/10.1016/j.pocean.2007.04.015

Yashayaev, I., Dickson, B. (2008). Transformation and Fate of Overflows in the Northern North Atlantic. In: Dickson, R.R., Meincke, J., Rhines, P. (eds) Arctic–Subarctic Ocean Fluxes. Springer, Dordrecht. https://doi.org/10.1007/978-1-4020-6774-7 22

Yashayaev, I., Holliday, N. P., Bersch, M., and van Aken, H. M. (2008). The history of the Labrador Sea Water: Production, spreading, transformation and loss. Arctic–Subarctic Ocean Fluxes: Defining the Role of the Northern Seas in Climate, 569-612.

Yashayaev, I., Lazier, J., & Clarke, R. (2003). Temperature and salinity in the central Labrador Sea during the 1990s and in the context of the longer-term change. Retrieved from https://doi.org/10.17895/ices.pub.19271729