1 Ocean Acidification trends and Carbonate System dynamics 2 across the North Atlantic Subpolar Gyre water masses during

3 **2009-2019**

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13 Keypoints:

- 14 During the 2010s, the subpolar North Atlantic experienced a 50-86% increase in
- anthropogenic CO_2 , accelerating by 7-10% the acidification.
- 16 Anthropogenic CO_2 contributed to acidification by 53-68% in upper layers and >82% in the 17 interior ocean.
- 18 The acidification trends (0.0006 and 0.0032 units yr⁻¹) declined the Ω_{Ca} and Ω_{Arag} by 0.004-
- 19 0.021 and 0.003-0.0013 units yr^{-1} , respectively.

20 Abstract

The CO₂-carbonate system dynamics in the North Atlantic Subpolar Gyre (NASPG) were 21 evaluated between 2009 and 2019. Data was collected aboard eight summer cruises through 22 the CLIVAR 59.5°N section. The Ocean Acidification (OA) patterns and the reduction in the 23 saturation state of calcite (Ω_{Ca}) and aragonite (Ω_{Arag}) in response to the increasing 24 anthropogenic CO_2 (C_{ant}) were assessed within the Irminger, Iceland and Rockall basins 25 26 during a poorly-assessed decade in which the physical patterns reversed in comparison with previous well-known periods. The observed cooling, freshening and enhanced ventilation 27 increased the interannual rate of accumulation of Cant in the interior ocean by 50-86% and 28 the OA rates by close to 10%. The OA trends were 0.0013-0.0032 units yr⁻¹ in the Irminger 29 and Iceland basin and 0.0006-0.0024 units yr⁻¹ in the Rockall Trough, causing a decline in 30 Ω_{Ca} and Ω_{Arag} of 0.004-0.021 and 0.003-0.0013 units yr⁻¹, respectively. The C_{ant}-driven rise 31 in total inorganic carbon ($C_{\rm T}$) was the main driver of the OA (contributed by 53-68% in upper 32 layers and >82% toward the interior ocean) and the reduction in Ω_{Ca} and Ω_{Arag} (>64%). The 33 transient decrease in temperature, salinity and $A_{\rm T}$ collectively counteracts the C_T-driven 34 acidification by 45-85% in the upper layers and in the shallow Rockall Trough and by <10%35 36 in the interior ocean. The present investigation reports the acceleration of the OA within the NASPG and expands knowledge about the future state of the ocean. 37

38 Keywords: Ocean Acidification, Anthropogenic Carbon, North Atlantic Subpolar Gyre.

39 **1. Introduction**

The ocean uptake of approximately one-third of the CO₂ released into the atmosphere 40 (Friedlingstein et al., 2023; Gruber et al., 2019a) has an important role in the climate regulation 41 causing changes in the marine carbonate chemistry. The exponential increase in the global 42 ocean CO₂ sink in phase with those of anthropogenic emissions (Friedlingstein et al., 2023) 43 has resulted in a long-term decrease in the concentration of carbonate ions ($[CO_3^{2-}]$) and pH. 44 This process has been collectively referred to as Ocean Acidification (OA; Caldeira and 45 Wickett, 2005, 2003; Doney et al., 2009; Orr et al., 2005; Raven et al., 2005; Feely et al., 2009) 46 and favour the dissolution of calcium carbonate (CaCO₃). It affects not only calcifying marine 47 organisms and ecosystems which use the biogenic CaCO₃ forms of calcite and aragonite (e.g. 48 Gattuso et al., 2015; Langdon et al., 2000; Pörtner et al., 2004, 2019; Riebesell et al., 2000) 49 but also the global biogeochemical cycles (Gehlen et al., 2011; Matear and Lenton, 2014). 50

The absorption of anthropogenic CO_2 has reduced the pH of the global surface ocean by 0.1 51 units since preindustrial times, representing approximately a 30% increase in acidity (Caldeira 52 and Wickett, 2003). According to the IPCC's Representative Concentration Pathways (RCPs) 53 scenarios (Van Vuuren et al., 2011; Moss et al., 2010), which project various future trajectories 54 of greenhouse gas concentrations, the model projections estimate a potential pH decrease of 55 0.3–0.4 units by the end of the century under the RCP8.5 scenario, which assumes continued 56 high CO₂ emissions. In contrast, the most conservative RCP2.6 scenario, which includes 57 significant emission reductions, anticipates a pH drop of 0.2–0.3 units (IPCC 2013 and 2021). 58 However, as the absorption and storing of anthropogenic carbon (C_{ant}), defined as the fraction 59 60 of inorganic carbon resulted from human emissions (Sarmiento et al., 1992), is not uniform 61 within the ocean (Sabine et al., 2004a), OA rates may show a significant spatial variability and should be regionally studied. The temporal evolution of the carbonate system variables in 62 surface waters are monitored and assessed in several time-series stations located across 63 different ocean regions (Bates et al., 2014). The largest OA rates are expected to occur across 64 high northern and southern latitudes (Bellerby et al., 2005; Orr et al., 2005), where deep 65 convective overturning and subduction occur favouring the entrance of C_{ant} in the interior 66 ocean (Maier-Reimer and Hasselmann, 1987; Lazier et al., 2002; Sarmiento et al., 1992). 67

The North Atlantic is one of the strongest CO₂ sinks and stores over 25% of the C_{ant} 68 accumulated in the global ocean (e. g. Gruber et al., 2019; Khatiwala et al., 2013; Pérez et al., 69 2024, 2010, 2008, 2024; Sabine et al., 2004; Takahashi et al., 2009). The Atlantic Meridional 70 Overturning Circulation (AMOC) plays a significant role by conveying acidified Cant-loaded 71 waters polewards and exporting them to the ocean interior across deep-water formation areas 72 73 (Lazier et al., 2002; Pérez et al., 2013, 2008; Steinfeldt et al., 2009). It contributes to 74 homogenize the C_{ant} and pH in the whole water column in such regions and exported these properties southwards to the global deep ocean (Perez et al., 2018). Thus, the North Atlantic 75 76 behaves as a crucial region for understanding the impacts of anthropogenic forcing on the global ocean. 77

OA has been widely studied in the North Atlantic through the monitoring of the ocean 78 physicochemical properties at time-series stations (summarized by Bates et al., 2014) placed 79 in subtropical and subpolar latitudes: the European Station for Time series in the Ocean at the 80 Canary Islands (ESTOC; 29.04°N, 15.50°W; González-Dávila et al., 2010; González-Dávila 81 and Santana-Casiano, 2023; Santana-Casiano et al., 2007), the Bermuda Atlantic Time-series 82 Study (BATS; 32.0°N, 64.0°W; Bates et al., 2012), the Irminger Sea Time Series (IRM-TS; 83 84 64.3°N, 28.0°W; Olafsson et al., 2010) and the Iceland Sea Time Series (IS-TS; 68.0°N, 12.66°W; Olafsson et al., 2009, 2010). OA rates has also been evaluated along transects 85 through repeated hydrographic cruises (i.e. Guallart et al., 2015; García-Ibáñez et al., 2016; 86 Vázquez-Rodríguez et al., 2012b) or even covered by volunteer observing ships (Fröb et al., 87 2019). These investigations have revealed a rate of decrease in pH of $\sim 0.001-0.002$ units yr⁻¹. 88 89 Moreover, González-Dávila and Santana-Casiano (2023) has recently indicated that these rates 90 are increasing since 1995.

The assessment of OA is of especial interest across the North Atlantic Subpolar Gyre (NASPG; 50-60°N), where the atmospheric CO₂ sink is particularly strong and the deep-water formation processes favour the storage of C_{ant} through the whole water column (Gruber et al., 2019b; Sabine et al., 2004b; Watson et al., 2009, Pérez et al. 2008). Likewise, the deep-water formation processes create the largest and deepest ocean environments supersaturated for aragonite (at more than 2000 m depth; Feely et al., 2004; Jiang et al., 2015), which is the main CaCO₃ mineral for Cold-water corals (CWC; Roberts et al., 2009) and some pteropods (Bathmann et al., 1991; Urban-Rich et al., 2001). These deep biomes are predicted to be one
of the first in the global ocean affected by OA, mainly due to the shoaling of the Aragonite
Saturation Horizon and its progressive exposition to undersaturated conditions for aragonite at
intermediate and deep waters (Gehlen et al 2014; Guinotte et al., 2006; Raven et al., 2005;
Roberts et al., 2009; Turley et al., 2007).

The physical processes along the NASPG, which are subject to significant spatiotemporal 103 104 variability introduced by the atmospheric forcing and climatology on an interannual scale, directly influenced the biogeochemistry (Corbière et al., 2007; Fröb et al., 2019). The changes 105 in North Atlantic Current (NAC) modifies the poleward heat transport from subtropical 106 latitudes and the air-sea interactions, influencing temperature patterns (Josey et al., 2018; 107 Mercier et al., 2015). Recent studies noticed the surface cooling and freshening of the NASPG 108 in the 2010s (Holliday et al., 2020; Josey et al., 2018; Robson et al., 2016; Tesdal et al., 2018) 109 contrasting with the period of warming and salinification in the 1990s extended until 2005 110 (Häkkinen and Rhines, 2004; Hátún et al., 2005; Robson et al., 2014). Anomalously heat loss 111 and winter deep convection were found to be of high intense since 2008 contributing to the 112 extreme cold anomaly along the NASPG (e. g. De Jong et al., 2012; de Jong and de Steur, 113 114 2016; Fröb et al., 2019, 2016; Gladyshev et al., 2016b, 2016a; Piron et al., 2017; Våge et al., 2009). These fluctuations in the vertical mixing and ocean circulation patterns introduces 115 changes in the distribution of the carbonate system variables. 116

The estimated OA trend over 1991-2011 for surface waters across the North Atlantic Subpolar 117 biome was -0.0020 ± 0.0001 units yr⁻¹ (Lauvset et al., 2015). Chau et al., 2024 recently reported 118 that the surface waters in the Irminger and Iceland basins has acidified over 1985-2021 at rates 119 of -0.0016 ± 0.0001 and -0.0014 ± 0.0001 units yr⁻¹. Several observation-based investigations 120 have evaluated the drivers, trends and impacts of OA through the entire water column in the 121 Irminger and Iceland basins (e. g. Fontela et al., 2020; García-Ibáñez et al., 2021, 2016; Perez 122 123 et al., 2018; Pérez et al., 2021; Ríos et al., 2015), while few studies have addressed it in the 124 Rockall Trough (e. g. McGrath et al., 2013, 2012a, 2012b, Humphreys et al., 2016) due to lack of repeated hydrographic sections or time-series stations and subsequent limitation of 125 continuous surface-to-bottom data. The high longitudinal variability in the NASPG caused by 126 the influence of different circulation patterns and water masses (García-Ibáñez et al., 2018, 127

2015) introduced several physicochemical heterogeneities between the Irminger and Iceland
with the Rockall basin (Ellett et al., 1986; McGrath et al., 2013, 2012b; Holliday et al., 2000).
These differences in the distributions of Marine Carbonate System (MCS) variables should be
considered to improve our understanding of OA in the entire North Atlantic.

This study evaluated the OA in the NASPG across the Irminger, Iceland and Rockall basins during the 2010s. High-quality direct measurements of CO₂ system variables from eight hydrographic cruises occupying 59.5°N between 2009 and 2019 were used to evaluate the drivers and trends of pH, and the potential effects of OA on calcifying organisms of changes in calcite (Ω Ca) and aragonite (Ω Ar) saturation states. This study advances our understanding of the complexities associated with OA in the NASPG and supports ongoing efforts to model and predict future acidification scenarios in the North Atlantic and global ocean.

- 139 **2.** Methodology
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2.1. Data collection

141 Data were collected from eight summer cruises conducted along the transverse hydrographic section at 59.5°N between 2009 and 2019 (Daniault et al., 2016; Gladyshev et al., 2016b, 2017, 142 143 2018; Sarafanov et al., 2018). This section is part of the World Climate Research Programme (WCRP) within the framework of the CLIVAR (Climate and Ocean: Variability, Predictability 144 145 and Change) project and covers the length of the Subpolar North Atlantic between Scotland and Greenland (4.5-43.0°W), crossing the Irminger and Iceland basins and the Rockall Trough 146 (Figure 1). Generally, the sampling stations were equidistantly spaced every 20 nmi apart 147 $(\sim 1/3^{\circ} \text{ longitude})$ and repeated in all the cruises except for the cruise of 2016, when the station 148 spacing was decreased to 10 nmi over Reykjanes Ridge western and eastern slopes. The 149 distance between stations over the east Greenland slope and shelf always decreased from 10 150 nmi to about 2 nmi. The surface-to-bottom sampling and in situ measurements were performed 151 152 by using a SBE 911plus CTD with SBE32 Carousel containing 24 Niskin bottles (10 L) with additional sensors for pressure (P), dual temperature (T) and salinity (S), and dissolved oxygen 153 (DO). The eight cruises included in the new dataset are the result of an international 154 collaboration between researchers from the P. P. Shirshov Institute of Oceanology at the 155 Russian Academy of Science and the Marine Chemistry research group from the 156

Oceanography and Global Change Institute (OUIMA-IOCAG) at the University of Las Palmas 157 de Gran Canaria (ULPGC). A detailed overview of the cruises is given in Table 1. 158

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2.1.1. CO₂ system variables measurements

The analysis of the MCS variables followed the same analytical methodology and provided 160 high-quality CO₂ measurements in all the hydrographic cruises. It includes the sampling and 161 data collection techniques, quality control and calculation procedures published in the updated 162 version of the DOE method manual for CO₂ analysis in seawater given by Dickson et al., 2007. 163 The seawater samples were onboard analysed for total alkalinity $(A_{\rm T})$ and total inorganic 164 carbon ($C_{\rm T}$) determination by using a VINDTA 3C and following Mintrop et al., (2000). The 165 $A_{\rm T}$ was analysed by potentiometric titration with HCl to the carbonic acid endpoint and 166 determined through the developing of the full titration curve (Millero et al., 1993; Dickson and 167 Goyet, 1994). The $C_{\rm T}$ was determined through coulometric titration (Johnson et al., 1993). The 168 VINDTA 3C was calibrated through the titration of Certified Reference Material (CRMs; 169 170 provided by A. Dickson at Scripps Institution of Oceanography), giving values with an accuracy of $\pm 1.5 \ \mu\text{mol kg}^{-1}$ for A_{T} and $\pm 1.0 \ \mu\text{mol kg}^{-1}$ for C_{T} . 171

Spectrophotometric pH measurements (Clayton and Byrne, 1993) in total scale at constant 172 temperature of 15°C (pH_{T,15}) were performed for the cruises between 2009 and 2016. A 173 spectrophotometric pH sensor (SP101-SM) developed by the QUIMA-IOCAG group at the 174 ULPGC in collaboration with SensorLab (González-Dávila, 2014; González-Dávila et al., 175 176 2016) was used. The method uses 4 wavelengths analysis for pH indicator dyes (m-cresol 177 purple), includes auto-cleaning steps and performs a blank for pH calculation immediately after the dye injection. The spectrophotometric sensor was in situ tested by using a TRIS 178 seawater buffer (Ramette et al., 1977) and provided pH_{T15} values with an accuracy of ± 0.002 179 units. To account for the systematic uncertainty reported by DelValls and Dickson (1998) 180 181 related to the pK* values of m-cresol purple, and in line with their recommendations, a correction of +0.0047 units was applied to the measured pH_{T15}. This adjustment ensures that 182 the calculated pH values are consistent with the more accurate pK* determinations. 183

2.1.2. Dissolved oxygen (DO) measurements 184

185 The WINKLER method introduced by Winkler (1888) and optimized by Carpenter (1965) and Carrit and Carpenter (1966) was used to analytically determine the dissolved oxygen (DO) of 186 the seawater samples in all the cruises from 2009 to 2016. The seawater samples for DO 187 determination were collected from the bottle samples in pre-calibrated glass wide-neck bottles 188 avoiding bubble formation. The temperature of the water was recorded during the sampling. 189 All the reagents and solutions used for dissolved oxygen determination were prepared 190 following the procedures described by Dickson, (1995) and their possible impurities were 191 controlled by determining a blank every 2 days. 192

As DO could not be analytically measured during the cruise of 2019 (due to limitations related with the oceanographic cruise plan), it was computed for this year by comparing the performance of the DO sensor during the cruise of 2019 versus (1) DO data estimated by a neural network for the cruises of 2016 and 2019 and (2) WINKLER-measured DO data in the cruise of 2016. The neural network ESPER_NN (Empirical Seawater Property Estimation Routine) introduced by Carter et al., (2021) was used for DO estimations. The computational procedure is detailed in Appendix A.

200 **2.2.** Data processing

201 2.2.1. Evaluation of the internal consistency of the data using CANYON-B

202 The measured and determined data were compared with estimations given by the Bayesian neural network "CANYON-B" (Bittig et al., 2018), a re-developed and more robust neural 203 204 network based on CANYON (CArbonate system and Nutrients concentration from hYdrological properties and Oxygen using a Neural-network; Sauzède et al., 2017). 205 206 CANYON-B estimates the four MCS variables (A_T , C_T , pH and pCO_2) and macronutrients concentrations (PO₄³⁻, NO₃⁻ and Si(OH)₄, hereinafter PO₄, NO₃ and Si(OH)₄) as a function of 207 a simple set of input variables which include P, T, S, DO, latitude, longitude and date. This 208 209 neural network is trained on and validated against bottle data from GLODAPv2 and recent GO-SHIP profiles and compared with sensor data from Argo floats. The standard errors of 210 estimate reported for CANYON-B by Bittig et al., (2018) are 6.3 μ mol kg⁻¹ for A_T, 7.1 μ mol 211 kg⁻¹ for C_T , 0.013 units for pH, 20 µatm for pCO₂, 0.051 µmol kg⁻¹ for PO₄, 0.68 µmol kg⁻¹ for 212 NO₃ and 2.3 µmol kg⁻¹ for Si(OH)₄. The crossover analysis between measured and estimated 213 data did not show systematic differences but individual outliers. The measured data that were 214

higher/lower than the CANYON-B estimate by plus/minus twice the predicted variableuncertainty of the neural network was considered as outliers and removed from the dataset.

The total amount of measured data was 8974 for A_T , 7495 for C_T , 8706 for pH_{T,15}, 9656 for 217 DO, 9114 for PO₄ and 9192 for Si(OH)₄. The difference between the measured and CANYON-218 B-estimated variables (referred hereinafter as canyon-estimated variables) were performed for 219 each sample in which CANYON-B could be applied (samples with availability of T, S and DO 220 221 measurements). The number of data, mean values and standard deviation of the measured variables for each cruise were summarized in Table S1. The average differences with the 95% 222 confidence interval for each cruise are shown in Table S2. The average differences for the 223 entire period (2009-2019) were lower than 2.1 μ mol kg⁻¹ for A_T , 2 μ mol kg⁻¹ for C_T , 0.0002 for 224 pH, 0.02 µmol kg⁻¹ for PO₄ and 0.25 µmol kg⁻¹ for Si(OH)₄. The minimal difference between 225 the measured and canyon-estimated pH infers confidence in the correction applied to the 226 measured pH following DelValls and Dickson (1998). 227

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2.2.2. Computational methods

The computational procedures to calculate MCS system variables applied in this investigation 229 used the CO_{2.SYS} programme developed by Lewis and Wallace, (1998) and run with the 230 MATLAB software (van Heuven et al., 2011; Orr et al., 2018; Sharp et al., 2023). The set of 231 constants used for computations includes the carbonic acid dissociation constants of Lueker et 232 al., (2000), the HSO₄⁻ dissociation constant of Dickson, (1990), the HF dissociation constant 233 of Perez and Fraga, (1987) and the value of [B]_T determined by Lee et al., (2010). The pH in 234 total scale at *in situ* temperature (pH_T) was computed from the measured A_T and pH_{T,15} (the 235 computed $C_{\rm T}$ was given as an output). The pH_T for the cruise of 2019, in which direct pH 236 measurements were not performed, was computed from the measured $A_{\rm T}$ and $C_{\rm T}$. 237

The saturation states of Calcite (Ω_{Ca}) and Aragonite (Ω_{Arag}), determined from the product of the ion concentrations of calcium ([Ca²⁺]) and carbonate ([CO₃²⁻]) divided by the stoichiometry solubility products (K_{sp}) for calcite (K_{Ca}) and aragonite (K_{Arag}) given by Mucci (1983), were generated as outputs of the CO_{2,SYS} computational routine. The decrease in Ω_{Ca} and Ω_{Arag} reports the adverse impacts of OA on marine calcification processes (e. g. Gattuso et al., 2015; Langdon et al., 2000; Pörtner et al., 2004, 2019; Riebesell et al., 2000).

An internal consistency test was conducted on the three measured MCS variables. The 244 measured variables were compared with canyon-estimated and CO_{2SYS}-computed variables. 245 The average differences and standard deviations were summarized in Table S2 and ensure the 246 consistency of the observations. In addition, due to gaps in data, an intercomparison between 247 measured and computed $C_{\rm T}$ and pH_{T15} was performed. It considers the availability of 248 measurements for each latitude, longitude and time and the differences between the measured 249 250 and computed pH with the canyon-estimated pH_T. The use of measured or computed $C_{\rm T}$ followed these conditions: (1) If there is measured $C_{\rm T}$ but not measured pH, measured $C_{\rm T}$ was 251 used, (2) if there is measured pH but not measured $C_{\rm T}$, computed $C_{\rm T}$ was used, (3) and if there 252 is measured $C_{\rm T}$ and pH, measured $C_{\rm T}$ was used when the differences between measured and 253 254 canyon pH_T is lower than the differences between computed and canyon-estimated pH_T, while computed C_T was used when the opposite happens. In total, 6375 measured and 2872 computed 255 256 $C_{\rm T}$ data were used in this study (69% and 31%, respectively). The average differences in each cruise between the combined (measured and computed, also referred as " $C_{T(new)}$ ") and canyon-257 258 estimated C_T variable is provided in Table S2. The amount and percentage of measured and computed $C_{\rm T}$ data per cruise is given in Table S3. As the measured $C_{\rm T}$ was in average 1.9 µmol 259 kg⁻¹ higher than the canyon-estimated and the computed $C_{\rm T}$ was in average 1.7 µmol kg⁻¹ lower, 260 the new compilation based on these previous conditions allowed to reduce the difference to 261 262 $1.5 \ \mu mol \ kg^{-1}$.

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2.2.3. Anthropogenic CO₂ (Cant) calculation

The anthropogenic CO_2 (C_{ant}) was estimated by using the biogeochemical back-calculation 264 ϕC_T^{o} method, which has an overall estimated uncertainty of $\pm 5.2 \mu mol kg^{-1}$ (Pérez et al., 2008; 265 266 Vázquez-Rodríguez et al., 2009). The method considers the change of $C_{\rm T}$ between the preindustrial era (1750) and the time of the observations, as well as the processes involved in 267 the uptake and distribution of C_{ant} (biogeochemistry, mixing processes and air-sea fluxes). The 268 C_{ant} was calculated (Eq. 1) as the difference between the C_{T} at the time of observation, the C_{T} 269 that the seawater would have in equilibrium with a preindustrial atmosphere (preformed $C_{\rm T}$; 270 $C_{\rm T}^{\rm pre}$), the offsets of such equilibrium values (air-sea CO₂ disequilibrium; $\Delta C_{\rm T}^{\rm dis}$) and the 271 changes in $C_{\rm T}$ due to the organic and carbonate pumps ($\Delta C_{\rm T}^{\rm bio}$). The $C_{\rm T}$ and $A_{\rm T}$ at the time of 272

observations and the preformed $A_T (A_T^0)$ are needed as input parameters and the computational procedure was described by Vázquez-Rodríguez et al., (2012).

275
$$C_{ant} = C_T - C_T^{pre} - \Delta C_T^{dis} - \Delta C_T^{bio}$$
(1)

The ϕC_T^{0} method is an improved process-based C_{ant} estimation method tested and widely applied in the Atlantic Ocean (Vázquez-Rodríguez et al., 2009) which present distinctive characteristics relative to existing C_{ant} approaches, such as the classical ΔC^* (GSS' 96; Gruber et al., 1996) and the TrOCA (Touratier et al., 2007). The main advantages of the ϕC_T^{0} method has been described by Pérez et al., (2008).

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2.2.4. Hydrographic characterization

282 The characterization of the basins and water masses was done by considering the 2009-2019 283 mean combined 59.5°N section constructed with potential vorticity, dissolved oxygen and salinity together with the large-scale circulation in the North Atlantic (e. g. Lherminier et al., 284 2010; Pérez et al., 2021; Sarafanov et al., 2012; Schmitz and McCartney, 1993; Schott and 285 Brandt, 2007; Sutherland and Pickart, 2008). A schematic diagram with the main surface and 286 287 deep currents in the NASPG is depicted in Figure 1a. The basin division considered the NAC 288 pathways and revealed a west-to-east distribution comprising the Irminger and Iceland basins and the Rockall Trough. The Iceland basin was delimited along its eastern boundary by the 289 central NAC branches around the northern part of the Haton Bank and George Bligh Bank, 290 and along its western boundary by the Return Current over the eastern flank of the Reykjanes 291 292 Ridge slope. This suggest that the Iceland basin could be longitudinally separated in two subregions: the western Iceland basin (24.0-29.5°W) and the eastern Iceland basin (14.0-293 24.0°W). 294

The upper layers were mainly occupied by Subpolar Mode Waters (SPMW) and North Atlantic Central Waters (NACW). SPMW is formed in the Iceland basin (McCartney and Talley, 1982; Brambilla and Talley, 2008; Tsuchiya et al., 1992; Van Aken and Becker, 1996), flow eastward to the Rockall Trough and recirculate across the Reykjanes Ridge (Brambilla and Talley, 2008). In the Irminger basin, SPMW flow with the Irminger Current to the north over the western Reykjanes Ridge flank and to the south over the eastern Greenland slope (Figure 1a). Thus, SPMW signal was detected in the western and eastern Irminger basin up to 400-700 m depth 302 and limited to subsurface depths in the central part of the basin. NACW were placed above SPMW east of the Irminger basin and separated in two branches: Eastern North Atlantic 303 Central Water (ENACW), formed by winter convection in the intergyre region and moved 304 poleward from the Bay of Biscay through the Rockall Trough (Harvey, 1982; Pollard et al., 305 1996), and Western North Atlantic Central Water (WNACW), flowing northward with the 306 NAC along the western Iceland basin. The intermediate layers were mainly occupied by 307 308 Labrador Sea Water (LSW), formed in the Labrador Sea and transported eastward (e.g. Pickart et al., 2003; Fröb et al., 2016). LSW path diverges into two cores when it reaches the Reykjanes 309 Ridge (Álvarez et al., 2004; Pickart et al., 2003): a fraction of LSW rapidly moved to the 310 Irminger basin and incorporated into the Deep Western Boundary Current (DWBC) (Bersch et 311 312 al., 2007) and a second LSW core was transported eastward into the Iceland and Rockall basins. In the Irminger and western Iceland basin, LSW placed above Iceland-Scotland 313 314 Overflow Water (ISOW), which originated from the overflow of Norwegian Sea waters over the Iceland-Scotland ridges and flowed southward and below 1500 m depth through the 315 316 western NASPG (van Aken and de Boer, 1995; Dickson et al., 2002; Fogelqvist et al., 2003). The bottom of the western Irminger basin was occupied by Denmark Strait Overflow Water 317 (DSOW), recently formed from deep waters from the Nordic seas flowing southward over the 318 Greenland-Iceland ridge and sinking through the eastern Greenland slope (Read, 2000; 319 320 Stramma et al., 2004; Yashayaev and Dickson, 2008). LSW core transported eastwards rises in depth through the western Haton Bank flank and occupy the bottom depths in the eastern 321 Iceland basin and in the Rockall Trough. A low-ventilated thermocline layer is placed between 322 323 SPMW and LSW in the eastern NASPG (García-Ibáñez et al., 2016), which represent the product of mixing with waters coming from the south (i. e. Mediterranean Waters; MW). 324

To enhance the comprehension of the spatial distribution and trends of the biogeochemical 325 variables and to facilitate comparisons with previous studies along the NASPG, the 326 327 hydrographic characterization was simplified based on the following principles: (1) the Iceland basin was not divided into its western and eastern parts and its longitudinal span was delimited 328 329 by the Reykjanes Ridge (29.5°W) and the Haton Bank (17°W), (2) upper Labrador Sea Water (uLSW) was separated from deeper LSW (e. g. Stramma et al., 2004), (3) the weak and 330 331 spatially-limited influence of the return current and WNACW was removed by considering the upper and intermediate layers of both the Irminger and Iceland basin fully occupied by SPMW 332

above uLSW, and (4) only the east branch of NACW (ENACW), placed above SPMW, wascontemplated for the upper Rockall Trough.

The whole water column was separated in layers delimited by potential density isopycnals at 335 a reference pressure of 0 dbar following Azetsu-Scott et al. (2003), Kieke et al. (2007), Pérez 336 et al. (2008) and Yashayaev et al. (2008). The vertically distributed water masses separated in 337 density layers is represented for the entire section in Figure 1b. The vertical characterization 338 339 in density layers allows to consistently compare the low-variable physical and chemical properties within each water mass, enabling to assume linearity in the ocean CO₂ system. The 340 determination of the isopycnal limits between layers in the Irminger and Iceland basins 341 followed previous biogeochemical studies in the western boundary of the North Atlantic 342 (Fontela et al., 2020; García-Ibáñez et al., 2016; Pérez et al., 2010, 2008; Vázquez-Rodríguez 343 et al., 2012a). The surface-to-bottom distribution of the main water masses in these basins 344 (with their respective σ_0 lower limits shown in brackets) was SPMW (27.68 kg m⁻³), uLSW 345 (27.76 kg m⁻³), LSW (27.81 kg m⁻³) and ISOW (27.88 kg m⁻³). The low temperature and 346 salinity DSOW were considered at the bottom of the westernmost part of the Irminger basin. 347 The hydrography of the Rockall Trough has been characterized in previous studies in the 348 Northeast Atlantic (e. g. Ellett et al., 1986; Harvey, 1982; McGrath et al., 2012a, 2012b; 349 Holliday et al., 2000), with the main water masses surface-to-bottom distributed as ENACW 350 (27.35 kg m⁻³), SPMW (27.68 kg m⁻³) and LSW (bottom). 351

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2.2.5. Data adjustment for trends computation

The interannual trends were analysed through the whole water column across the Irminger, 353 Iceland and Rockall basins by yearly averaging the variables for each layer, following previous 354 studies in the NASPG (e.g. Fontela et al., 2020; García-Ibáñez et al., 2016). Linear regressions 355 were applied to the mean values, in which the value of the slope give the ratios of interannual 356 changes. The errors of the means were calculated through the relation of the Standard 357 Deviation and the square root of the number of bottle samples in each layer and cruise 358 (Standard Deviation/ \sqrt{n}). The standard errors of the slopes were calculated by accounting 359 for the error propagation of the annual mean values. The Pearson correlation test was employed 360 to assess the strength and direction of the linear regressions and evaluate the significance of 361 the interannual trends. This test provided correlation coefficients (r^2) and corresponding p-362

values to determine statistical significance. The p-values ≤ 0.01 indicated that the trends were 363 statistically significant at the 99% confidence level, the p-values ≤ 0.05 indicated that the 364 trends were statistically significant at the 95% confidence level and the p-values ≤ 0.1 indicated 365 that the trends were statistically significant at the 90% level. Trends with p-values > 0.1 were 366 considered as not statistically significant but provided an estimation of the temporal evolution 367 of the variables in their respective layers. These not statistically significant trends were 368 369 explained by the high variability and changes in the low-limit depth of the layers encountered between consecutive years. 370

As there was a lack of in situ measurements and sampling along the west half of the Irminger 371 basin (36.5-42.5°W) in the cruise of 2019 (due to permit restrictions to study the national 372 waters of Denmark), the GO-SHIP A25-OVIDE data for the cruise of 2018 (available at 373 SEANOE [https://www.seanoe.org/], Lherminier et al., 2022) were considered to adjust the 374 2019 data. The average values were calculated with both the available data in the easternmost 375 part of the Irminger basin during the cruise of 2019 and the A25-OVIDE-2018 data available 376 in the same part of the section (29.6-36.5°W). The difference between these average values 377 provides the variation of each variable from 2018 to 2019, which can be extrapolated to the 378 379 western part of the Irminger basin by assuming linearity in the temporal evolution. Thus, the average values for 2019 were adjusted by applying the product with the calculated change 380 between 2018 and 2019. 381

382

2.2.6. Deconvolution of the trends

383 OA trends arise due to the combined variations in T, S, C_{T} and A_{T} . The influence of each driver on OA and subsequent impacts on marine calcification processes was analysed by assuming 384 linearity and employing a first-order Taylor-series deconvolution (Sarmiento and Gruber, 385 2006) to evaluate the trends for pH_T (Fröb et al., 2019; García-Ibáñez et al., 2016; Pérez et al., 386 387 2021; Takahashi et al., 1993; Tjiputra et al., 2014) and Ω (García-Ibáñez et al., 2021). The interannual rates of change of pH_T and Ω result from the sum of their partial derivatives versus 388 T, S, $C_{\rm T}$ and $A_{\rm T}$, calculated based on mean properties of each layer. The most recent equation 389 defined by Pérez et al., (2021) was used (Eq. 2), in which X represents pH_T , Ω_{Ca} and Ω_{Arag} and 390 391 salinity-normalized $C_{\rm T}$ and $A_{\rm T}$ (NC_T and NA_T, normalized to a constant salinity of 35) were 392 used to remove the effect of the freshwater fluxes and evaporation/precipitation effects.

$$393 \qquad \frac{dX}{dt} = \frac{\partial X}{\partial T}\frac{dT}{dt} + \left(\frac{\partial X}{\partial S} + \frac{NC_T}{S_0}\frac{\partial X}{\partial C_T} + \frac{NA_T}{S_0}\frac{\partial X}{\partial A_T}\right)\frac{dS}{dt} + \frac{S}{S_0}\frac{\partial X}{\partial C_T}\frac{dNC_T}{dt} + \frac{S}{S_0}\frac{\partial X}{\partial A_T}\frac{dNA_T}{dt}$$
(2)

It is important to remark that the changes in $NA_{\rm T}$ and $NC_{\rm T}$ are linked with biogeochemical processes which have different influences: the processes involved in the organic carbon pump contribute to strongly change the $NC_{\rm T}$ weakly affecting the $NA_{\rm T}$, while those involved in the carbonate pump affect the $NA_{\rm T}$ twice as much as $NC_{\rm T}$. The complexity and heterogeneity of the processes that govern the pH_T change were considered by this equation.

399 **3. Results**

The vertical distribution of the physical and biogeochemical variables is depicted for the 400 cruises of 2009 and 2016 in Figures 2, 3, S2 and S3. The subsurface layers were characterized 401 by warmer and saltier waters than intermediate and deep layers among the three basins (Figure 402 2a and 2b). A West-to-East increase in temperature and salinity throughout the water column 403 404 was observed in all the cruises. The temperature and salinity signals were highest in the Rockall Trough (4.5-11.0°C and 35.0-35.4, respectively), followed by the Iceland basin (3.0-405 7.5°C and 34.9-35.2, respectively) and the Irminger basin (1.5-6.5°C and 34.8-35.1, 406 407 respectively). The longitudinal differences in temperature were more remarkable toward the upper layers through the SPMW and uLSW. 408

- 409 The spatial variability in the physical properties introduced heterogeneities in the distribution 410 of the CO₂ system variables. The A_T show a well-correlated linear relationship with salinity
- 411 throughout the region ($A_T = 54.57 (\pm 0.36)$ Salinity + 396.7 (± 12.7); r²=0.90 and p-value <
- 412 0.01; Standard Error of Estimate of 2.9 μ mol kg⁻¹), with lower and vertically-homogenized
- 413 average values in the Irminger basin (2302.8-2307.3 µmol kg⁻¹ in subsurface waters and
- 414 2298.8-2301.0 μ mol kg⁻¹ in bottom waters) and Iceland basin (2308.7-2315.0 μ mol kg⁻¹ in
- 415 subsurface waters and 2305.2-2308.0 μ mol kg⁻¹ in bottom waters) compared to the Rockall 416 Trough (2317.9-2329.1 μ mol kg⁻¹ in subsurface waters and 2308.5-2310.9 μ mol kg⁻¹ in bottom
- 417 waters).

418 The upper layers were characterized by low C_T values (2153.7-2160.8 μ mol kg⁻¹ at the

419 Irminger basin, 2158.1-2168.4 μ mol kg⁻¹ at the Iceland basin and 2120.1-2131.0 μ mol kg⁻¹ at

- 420 the Rockall Trough), while a rapidly increment with depth was found below 100-200 m depth
- 421 (2154.7-2171.2 μ mol kg⁻¹ throughout the section). The notable difference in the distribution of

- 422 $A_{\rm T}$ and $C_{\rm T}$ (Figure 2c and 3a, respectively) compared to those of $NA_{\rm T}$ and $NC_{\rm T}$ (Figure S2) 423 elucidated the remarkable significance of freshwater fluxes on the carbon variables 424 fluctuations during the period of study. The entrance of $C_{\rm ant}$ through the atmosphere-seawater 425 interface caused higher $C_{\rm ant}$ values in the upper layers (higher than 50 µmol kg⁻¹ in the first 426 1000 m depth; Figure 3b). The natural component of the $C_{\rm T}$ ($C_{\rm nat}=C_{\rm T}$ - $C_{\rm ant}$; Figure 3c) correlated 427 with $C_{\rm T}$ (r²=0.87), and show a distribution characterized by low surface (<2110 µmol kg⁻¹) and 428 high bottom concentrations (>2130 µmol kg⁻¹).
- The pH_T (Figure 2d) rapidly decreased with depth showing the effect of biological uptake in 429 the upper layers and remineralization in deeper areas. The subsurface layer up to 100-200 m 430 depth exhibited pH_T values higher than 8.025 units, which fell to 7.975 units at the bottom 431 layers. The pH_T profiles reported an intrusion of remineralized and poorly oxygenated water 432 between 500 and 1000 m depth with relatively low pH_T (<7.975) compared to adjacent layers 433 in the Iceland basin and in the western part of the Rockall Trough. This thermocline layer was 434 previously observed at ~500 m depth by García-Ibáñez et al., (2016) along a more meridional 435 transect which crossed the Iceland basin northwest-southeast. It introduces differences in the 436 437 intermediate water masses between the Iceland and Rockall basins with the Irminger basin.

The spatial and interannual fluctuations in the ventilation rates through changes in the water 438 mass formation and respiration processes represent a source of variability in the 439 biogeochemical patterns. The apparent oxygen utilization (AOU), defined as the difference 440 441 between saturated oxygen (calculated following Benson and Krause, 1984) and measured oxygen, was used to assess the ventilation of the water masses (Figure 2e). The high AOU 442 443 values indicate low ventilation, while low AOU values indicate the opposite. The slow renewal 444 of waters with high AOU favour the accumulation of the product of remineralization (de la Paz et al. 2017). Thus, the areas with higher AOU (Figure 2e) were found to have high 445 concentration of $C_{\rm T}$ and low pH_T (Figures 3a and 2d, respectively). The near surface waters 446 permanently in contact with the atmosphere exhibited the lowest AOU values (<20 µmol kg⁻ 447 ¹). The Irminger Basin presents the most significant water column ventilation among the entire 448 section, with maximum AOU ranging from 35 to 50 µmol kg⁻¹ at the LSW and ISOW and the 449 remarkable intrusion of oxygenated DSOW (>260 µmol kg⁻¹ DO) over the continental slope 450 with AOU ranging from 30 to 40 µmol kg⁻¹. The intermediate and deep layers of the Iceland 451

and Rockall basins were less ventilated, with AOU values higher than 45-50 μ mol kg⁻¹. The thermocline layer placed between 500 and 1000 m depth along these two basins presented the highest maximum AOU throughout the period (>60 μ mol kg⁻¹). The stagnation of these waters corresponds with the high $C_{\rm T}$ and low pH_T (Figures 3a and 2d, respectively) encountered at intermediate depths and should be considered in its temporal evolution.

The temporal distribution and trends of the average physicochemical properties (Figures 4, 5, 6, S4, S5 and S6) revealed remarkable heterogeneities in their interannual evolution within the period 2009-2019 among the different basins and water masses. The interannual trends are presented along with their respective standard error of estimate and correlation factors (r^2 and p-value) in Table 2 and S4. The observed decrease in temperature and salinity, which was more pronounced in subsurface layers, and its implication on the MCS variations were discussed in section 4.

464 **4. Discussion**

465

4.1. Reversal of the physical trends during 2009-2019

The present investigation revealed the cooling and freshening of the upper ocean in the 466 NASPG within the period 2009-2019 (Figure 4; Table 2), as recently reported since the reversal 467 of climatic trend and surface physical properties occurring after 2005 (Holliday et al., 2020; 468 469 Josey et al., 2018; Robson et al., 2016; Tesdal et al., 2018). The temperature decreased in the upper ocean (with more than 95% level of confidence in SPMW, while non statistically 470 significant in ENACW) by 0.05-0.08 °C yr⁻¹ (Table 2), which is consistent with the ratio of 471 heat loss per decade among the first 700 m depth equivalent to approximately -0.45 °C decade⁻ 472 ¹ (-0.045 °C yr⁻¹) encountered over the period 2005-2014 (Robson et al., 2016). The interannual 473 temperature trends in subsurface layers (Table 2) similarly draw the cooling observed in the 474 Irminger basin between 2008 and 2017 (-0.05 and -0.11 °C yr⁻¹ for summer and winter, 475 respectively; Leseurre et al., 2020) and the winter average surface cooling along the entire 476 NASPG between 2004 and 2017 (-0.08 \pm 0.02 °C yr⁻¹; Fröb et al., 2019). The decrement in 477 subsurface salinity (with more than 95% level of confidence in both SPMW and ENACW) of 478 0.006-0.018 yr⁻¹ (Table 2) agreed with the interannual rates provided by Tesdal et al., (2018) 479 for the Irminger basin (-0.007 \pm 0.002 yr⁻¹) and for the central-eastern NASPG (-0.020 \pm 0.003 480 yr^{-1}) over the period 2004-2015. 481

482 The fluctuations in physical properties were linked to a decrease in oceanic heat transport and storage within the NASPG, which has been attributed to changes in the AMOC over decadal 483 484 to multidecadal timescales (Balmaseda et al., 2007; Desbruyères et al., 2013; Mercier et al., 2015; Smeed et al., 2018). However, the assessment of the temporal evolution of the AMOC 485 in high latitudes remains uncertain, and there is no evidence of its impact on physical patterns 486 across the NASPG on an interannual scale (Jackson et al., 2022). The changes in the 487 488 atmospherics forcing also account for the variability of the upper ocean physical properties and can have a cumulative effect over several years (Balmaseda et al., 2007; Böning et al., 489 490 2006; Eden and Willebrand, 2001; Marsh et al., 2005).

The distribution of the water mass properties, the processes of vertical and horizontal mixing 491 492 and the circulation patterns in the Irminger and Iceland basins were described by García-Ibáñez 493 et al., 2016 and 2018. The poleward path of the ENACW (Pollard et al., 1996) and its mixing with waters moving from the west across the NASPG (Ellett et al., 1986) accounted for the 494 495 highest subsurface temperature and salinity signals observed in the Iceland basin and even more in the Rockall Trough. The SPMW and LSW in the Rockall Trough exhibited higher 496 497 temperature and salinity signals in the respectively order of $\sim 1^{\circ}$ C and $\sim 0.05-0.1$ compared to 498 the Irminger and Iceland basins (Figure 4). The NASPG circulation patterns account for these differences by transporting eastward these water masses, which subduct below the ENACW 499 500 in the Rockall Trough and mixed with warmer and more saline intermediate waters (i.e. 501 Mediterranean Water) moving from the south (e. g. Ellett et al., 1986; Harvey, 1982; Holliday 502 et al., 2000).

503 The low temperature and salinity signals in the less-stratified Irminger basin (Figure 2) 504 experienced weaker interannual decreases in subsurface layers and higher rates of cooling and 505 freshening in intermediate and deep waters compared with the Iceland and Rockall basins (Figure 4; Table 2). These longitudinal thermohaline heterogeneities were related to the 506 507 enhancement of vertical mixing processes in areas of water mass formation along the western 508 NASPG (Fröb et al., 2016; García-Ibáñez et al., 2015; Pickart et al., 2003; Piron et al., 2017) and the water mass transformation along the NAC (Brambilla and Talley, 2008). The strongest 509 decrement in subsurface temperature and salinity along the Iceland and Rockall basins (Figure 510 4; Table 2) coincided with the significant event of heat loss and freshening observed by 511

Holliday et al., (2020) in the eastern NASPG over the period 2012-2016, so-called the Great
Salinity Anomaly. This pattern was not easily discernible in the Irminger basin due to the
transport of freshwater through the Fram Strait, as well as due the redirection of the Labrador
Current combined with changing wind stress curl (Holliday et al., 2020).

5164.2.Evaluation of the interannual trends in CT in response to changes in Cant and517Cnat

The changes in the physical patterns influenced the interannual variability of the MCS. The 518 increase in $C_{\rm T}$ expected in the upper ocean due to the atmospheric CO₂ uptake was offset by 519 the cooling and freshening (and dealkalinization) of the subsurface layers. The observed rates 520 of increase in $C_{\rm T}$ (Table 2) did not show notable differences with respect to the interannual 521 trends determined from previous decades at the Irminger and Iceland basins (0.62-0.82 and 522 0.38-0.64 µmol kg⁻¹ yr⁻¹, respectively; García-Ibáñez et al., 2016) and at IRM-TS and IS-TS 523 (0.49-0.71 and 0.39-0.94 µmol kg⁻¹ yr⁻¹, respectively; Pérez et al., 2021). The interannual 524 rates of increase in NCT were higher than those of CT in the subsurface layers, while the 525 trends were similar among intermediate and deep layers (Table 2). A detailed description of 526 the interannual A_T trends is provided in Appendix B. 527

The entrance of C_{ant} through the air-sea interface and its accumulation dominated the observed 528 increase in $C_{\rm T}$ (Figure 5 and Table 2). The increase in ventilation over 2009-2019, shown by 529 the negative AOU trends (Figure S6 and Table S4), favoured the vertical mixing. The upper 530 waters, due to be in contact with the atmosphere and have high biological production rates 531 during the warm months, show high Cant and low Cnat contents. The enhanced transport of 532 upper waters toward the interior ocean explained the rapid growth in C_{ant} at intermediate and 533 deep layers. The C_{ant} trends ranged between 0.85 and 1.77 µmol kg⁻¹ yr⁻¹ (statistically 534 significant at the 99% level). They were higher than the observed on a decadal to multidecadal 535 scale since the late 20th century in the Irminger and Iceland basins (0.21-0.89 µmol kg⁻¹ yr⁻¹ 536 during 1991-2015, García-Ibáñez et al., 2016; and 0.38-1.15 µmol kg⁻¹ yr⁻¹ during 1983-2013, 537 Pérez et al., 2021), which show the enhancement in the C_{ant} accumulation on interannual scales 538 during periods of high ventilation, as previously reported by Perez et al., (2008). The C_{nat} show 539 an inverse relationship with C_{ant} at intermediate and deep layers ($r^2 > 0.5$; statistically significant 540 541 at the 95% level of confidence) and weakly decreased across the western deep-convection

NASPG (Figure 5 and Table 2). The growth in phytoplankton biomass (Ostle et al., 2022), 542 together with the enhanced export toward the interior ocean under increasing ventilation, 543 account to the observed decrease in C_{nat} in upper waters. The C_{nat} showed a weaker decrease 544 at intermediate and deep layers due to the dominance of remineralization, which was not 545 intense enough at this time of the year to neutralize the downward transport of low-C_{nat} water 546 from the surface but accounted to partially compensate for its effect. The observed variations 547 548 in C_{nat} between years were strongly linked with fluctuations in the biological processes explained its non-significant trends at several layers. The changes in the circulation pattern of 549 550 the NASGP and thus in the horizontal advection related with the climatological forcing (Balmaseda et al., 2007; Desbruyères et al., 2013; Mercier et al., 2015; Thomas et al., 2008; 551 552 Xu et al., 2013) could behave as a source of variability for both Cant and Cnat and also infers differences between consecutive years. 553

554 The vertical distribution of C_{ant} and C_{nat} along the transect (Figure 3b and 3c) reflect the higher stratification in the Iceland and Rockall basin compared to the well vertically-mixed Irminger 555 556 basin. It represents a source of variability in the interannual changes of C_{ant} among the different layers and basins (Figure 4; Table 2). In the western NASPG, the surface heat loss and 557 558 enhanced deep convection processes favour the solubility and subsequent uptake of 559 atmospheric CO₂ and inject oxygenated and CO₂-rich waters into deeper layers (Messias et al., 2008). Its likely accounts for intermediate and deep layers in the Irminger basin exhibiting the 560 highest Cant accumulation rates in the NASPG (Figure 5; Table 2). The highest ventilation of 561 the interior ocean in the Irminger basin was demonstrated by its minimum AOU values (Figure 562 563 2 and S6). It induced a rapid surface-to-bottom transport of C_{ant} shown by its highest rates of increase in intermediate and deep waters throughout the region (Figure 5; Table 2). The high 564 565 C_{ant} values and its rapidly increment at DSOW were explained by the improved oxygenation of this layer at shallower depths (interannual AOU trends given in Table S4) and its subduction 566 567 through the continental slope below ISOW.

In the eastern NASPG, the stratification weakened due to the path of the NAC warming eastward the upper water column and accounted to slowdown the increase in C_{ant} in the Iceland basin. An exception comes with the Rockall basin, in which the relatively warm and salty ENACW (Figure 2 and 4) showed the maximum C_{ant} (58-68 µmol kg⁻¹) and minimum C_T

(2120-2131 μ mol kg⁻¹) and C_{nat} (2058-2070 μ mol kg⁻¹) throughout the region (Figure 3 and 572 5). The enhanced oxygenation of the ENACW (AOU $\leq 20 \mu$ mol kg⁻¹ and reaching the oxygen 573 saturation after 2014) was related with its high rates of renovation due to its path from the 574 south (Pollard et al., 1996) and its mixing with waters moving eastward (Ellett et al., 1986). 575 This favoured the transport subsurface waters with relatively high C_{ant} content from lower 576 latitudes into the Rockall Trough and introduced wide differences respect to adjacent deeper 577 578 layers moved from the western NASPG which strength the stratification. As the NAC transports nutrient-rich waters northward and eastward into subsurface layers in the Rockall 579 Trough, biological production tends to increase and actively reduced the CO₂ excess from the 580 ENACW (McGrath et al., 2012b), as proved by the observed low $C_{\rm T}$ and $C_{\rm nat}$. The ENACW 581 582 presented relatively low C_{nat} and C_T (Figure 5) and high A_T and NA_T in 2014. These variations indicated that the increase in carbonate and bicarbonate concentrations rising A_T and NA_T was 583 compensated by the depletion in dissolved CO_2 . The relatively high temperature and NA_T in 584 2014 likely indicated an improved spreading of subsurface waters from subtropical latitudes 585 586 into the Rockall Trough. The enhanced biological production in these waters, together with the reduction in solubility due to warming which favour the CO₂ evasion to the atmosphere, 587 account for decreasing C_{nat} and thus C_T . 588

The strong interannual increase in the ENACW ventilation during this decade increase the C_{ant} and decrease the C_{nat} (Rodgers et al., 2009) keeping approximately constant the C_T (Table 2). The poorly ventilated thermocline (AOU > 60 µmol kg⁻¹), placed between 500-1000 m in the eastern NASPG, induced a C_{nat} -driven increase in C_T among the SPMW and uLSW. However, its intrusion does not present relevant variations with time and thus does not introduce differences in the interannual trends of the biogeochemical properties.

595

4.3. Acidification trends

The interannual pH_T trends (Figure 6, Table 2) exhibited the acidification of the whole water column in NASPG during the period 2009-2019. Despite the acidification rates observed in the most subsurface waters among the three basins were not significant at the 90% confidence level (Table 2), they were consistent in the interval of 0.001 units yr⁻¹ to those observed during larger periods at time-series stations located across the North Atlantic: at subtropical latitudes $(0.0018 \pm 0.0002 \text{ units yr}^{-1} \text{ during } 1995-2014 \text{ and } 0.0020 \pm 0.0001 \text{ units yr}^{-1} \text{ during } 1995-2023$

at ESTOC, González-Dávila and Santana-Casiano, 2023; and 0.0017 ± 0.0001 units yr⁻¹ during 602 1983-2014 at BATS, Bates et al., 2014) and subpolar latitudes (-0.0017 \pm 0.0002 units yr⁻¹ at 603 IRM-TS during 1983-2013 and -0.0026 ± 0.0002 units yr⁻¹ at IS-TS during 1985-2013, 604 summarized by Pérez et al., 2021). In addition, the changes in the surface pH_T trends has been 605 reported by Leseurre et al., (2020) in the western NASPG within a wide latitudinal area (54-606 64°N) during the period 2008-2017 in comparison with the periods 1993-1997 and 2001-2007. 607 608 Although the highly significant cooling observed in SPMW, the year-to-year variations in ventilation (shown by the annual average AOU and its trends in Figure S6) and thus in C_{nat} and 609 C_{ant} (Figure 5) introduced relevant changes in pH_T on an interannual scale and explained the 610 low significant trends. The extreme negative NAO index of 2009-2010 (Jung et al., 2011) 611 612 weakened the wind forcing, which infers variability in the circulation patterns and physical properties of the surface waters, consequently reducing deep convection. This was observed 613 614 in the slowdown in ventilation from 2009 to 2010 (Figure S6) in the Irminger and Iceland basins which caused a relatively increase in C_{nat} and decrease in C_{ant} (Figure 5). 615

616 The highest acidification rates were found through intermediate and deep waters in the Irminger and Iceland basins, coinciding with the highest rates of increase in C_{ant} (Table 2, 617 trends statistically significant at more than 95% level of confidence). The exception comes 618 with the DSOW, which presented and interannual decrease in pH_T in phase with those of the 619 uLSW. This singularity was previously observed by García-Ibáñez et al., (2016), which noticed 620 the similar trends between the DSOW and LSW attributed to the recently formation and sink 621 through the continental slope of the DSOW. The acidification rates found among the uLSW, 622 LSW and ISOW (0.0026-0.0032 units yr⁻¹) experienced, on an interannual scale, an 623 acceleration in comparison with previous reported based on long-term records [e. g. 0.0009-624 0.0017 units yr⁻¹ estimated for 1981-2008 by Vázquez-Rodríguez et al., (2012b); 0.0013-625 0.0016 units yr⁻¹ estimated for 1991-2015 by García-Ibáñez et al., (2016); 0.0015-0.0019 units 626 yr⁻¹ estimated for 1983-2013 at the IRM-TS by Pérez et al., (2021); 0.0019 ± 0.0001 units yr-627 1 estimated for 1993-2017 by Leseurre et al., (2020)]. Contrasting the rates of change in pH_T 628 during the decade of study with those encountered by these multidecadal evaluations (and 629 considering the total amount of years comprising each of the studies and the changes in the ion 630 631 hydrogen concentration- $[H_T^+]$), we estimate an acceleration in the rates of acidification of 0.4-5.4% in the Irminger basin and 1.0-9.0% in the Iceland basin during the 2010s since the late 632

 20^{th} century. This acceleration was mainly attributed to increased deep-water ventilation (shown in the rapid decrease in AOU in Figure S6) favouring the progressively increase in the accumulation of C_{ant} and C_{nat} toward intermediate a deep layers, in which cooling was not significant in the Irminger basin and neither enough intense in both basins to compensate the acidification.

Although the similarities encountered in the pH_T trends among both basins, the average values 638 639 presented differences which may be closely linked with the transport and transformations of the water masses along the NASPG and mainly modulated by the Reykjanes Ridge (García-640 Ibáñez et al., 2015, 2016, 2018). The transformation of the SPMW formed in the Iceland 641 (McCartney and Talley, 1982; Brambilla and Talley, 2008; Tsuchiya et al., 1992; Van Aken and 642 Becker, 1996) and flowing with the NAC across the Reykjanes Ridge (Brambilla and Talley, 643 2008) accounted for the lower pH_T values in the Irminger basin. The differences in pH_T found 644 at intermediate and deep layers were related with the divergence of the LSW path into two 645 cores when it reaches the Reykjanes Ridge (Álvarez et al., 2004; Pickart et al., 2003) and the 646 ISOW path flowing southward along the western Iceland basin and recirculated northward into 647 the eastern Irminger basin (Dickson and Brown, 1994; Saunders, 2001). These differences in 648 649 the spreading of water masses enhanced the ventilation in the Irminger basin favouring the fall in pH_T compared with the Iceland basin. The rise in the ISOW following the Reykjanes Ridge 650 slope through its eastern flank favoured a strong vertical mixing over and around the ridge 651 (Ferron et al., 2014) and a reduction of the LSW core in the Iceland basin (García-Ibáñez et 652 653 al., 2015), contributing to resemble pH_T values and trends among the uLSW and LSW in this 654 basin.

655 The upper waters of the Rockall Trough presented the maximum pH_T throughout the transect (8.02-8.08 units). The observed strong pH_T fluctuations between years related with interannual 656 changes in the NAC do not allow to discern trends with a statistically interval of confidence 657 equal or higher than the 90%. The interannual decrease in pH_T in the ENACW (~0.001 units 658 yr⁻¹) was half than the observed along southernmost transects in the Rockall Trough between 659 1991 and 2010 (~0.002 units yr⁻¹, McGrath et al., 2012a). The temporal distribution of the 660 average pH_T (Figure 6) highly influenced by the high-ventilation (seen in minimum AOU 661 662 values highly variables between years and which tend to decrease with 99% statistical

confidence: Figure S6 and Table S4) allow to discern two periods: the approximately constant 663 ventilation rates keep a steady state in terms of pH_T during 2009-2011, while the progressively 664 renewal and oxygenation of subsurface waters after 2012 (and peaking in this year) increase 665 the pH_T. The renewal of waters in the shallow Rockall Trough, in contrast with the 666 westernmost NASPG, was not primarily driven by vertical but by lateral advection. The 667 modifications of the ENACW through air-sea exchange and mixing with adjacent waters 668 669 modulated its properties at different time scales (Holliday et al., 2000) and caused the observed variations in the MCS. The variations in pH_T between consecutive years after 2012 may be 670 671 attributed to the fluctuations in the spreading into the Rockall Trough of several water masses occupying different depths coming from the south and east (Ellett et al., 1986; Pollard et al., 672 673 1996). Holliday et al., 2020 reported the reduction in the spreading of saline subsurface waters from subtropical latitudes and diversion of Arctic freshwater from the western boundary into 674 675 the eastern NASPG during 2012-2016. The subsequent freshening of the ENACW compensated for the increase in AT expected without the effect of salinity (see in the decreasing 676 677 AT against the increasing NAT; Figure S4 and Table S4) and weakened the increase in CT expected due to poleward advection (see in the slowdown in the rise of CT in comparison with 678 those of NCT; Figure 5 and S5 and Table 2 and S4). The C_T remains approximately constant 679 (Figure 5 and Table 2) due to the increase in C_{ant} (0.85 ± 0.11 µmol kg⁻¹ yr⁻¹; p-value < 0.01) 680 was neutralized by the decrease in C_{nat} (-0.84 ± 0.50 µmol kg⁻¹ yr⁻¹; p-value < 0.1). These 681 findings suggest that the atmospheric CO₂ invasion was offset by the growing phytoplankton 682 biomass favouring its biological uptake (Ostle et al., 2022) and the weakening transport of 683 remineralized and saline water from the south (Holliday et al., 2020), thus compensating the 684 acidification of the ENACW. 685

The SPMW among the Iceland and Rockall basins showed similar pH_T trends (Table 2) due to the emplacement of the poorly-oxygenated thermocline at these depths (García-Ibáñez et al., 2016). The approximately constant AOU at SPMW in the eastern NASPG (Figure S6) proved its steady ventilation, which can introduce differences in the acidification rates among the layers accomplishing the Rockall Trough. The influence of the cooling and freshening of deeper areas due to the spreading and horizontal mixing was notable in the LSW, which presented slightly higher pH_T values in the Rockall respect to the adjacent Iceland basin.

693 4.4. Interannual changes in Ω_{Ca} and Ω_{Arag}

The analysis of the changes in Ω_{Ca} and Ω_{Arag} hold significance in elucidating the potential 694 effects of OA over the CaCO₃ species calcite and aragonite, thereby offering insights into their 695 potential implications for marine calcifying organisms and ecosystems. The vertical 696 distribution of Ω_{Ca} and Ω_{Arag} is presented in Figure S3. The upper and intermediate layers up 697 to 2100-2400 m depth of the Irminger and Iceland and the whole Rockall basin were 698 supersaturated for an agonite ($\Omega_{\text{Arag}} > 1$), while the DSOW was undersaturated ($\Omega_{\text{Arag}} < 1$). The 699 ISOW, with Ω_{Arag} ranged between 1.0 and 1.1 at the beginning of the decade, crossed to 700 undersaturated conditions at the end of the period due to the progressively rise of the aragonite 701 saturation horizon (depth in which $\Omega_{Arag}=1$). The whole water column throughout the section 702 was supersaturated for calcite (Ω_{Ca} >1) due to its lower solubility (Mucci, 1983). The Ω_{Ca} and 703 Ω_{Arag} in the SPMW (2.2-2.7 and 1.4-1.7 units, respectively) were lower than the encountered 704 equatorward in the subsurface Atlantic (>4.0 and >2.5 units, respectively; González-Dávila et 705 al., 2010; González-Dávila and Santana-Casiano, 2023). The poleward pathway of low-706 latitude upper waters through the Rockall Trough explained the higher Ω_{Ca} and Ω_{Arag} found in 707 the ENACW (3.0-3.6 and 1.8-2.3 units, respectively). The reduction in Ω_{Ca} and Ω_{Arag} towards 708 709 higher latitudes in upper and intermediate layers smooth the vertical gradients in the NASPG compared with the subtropical latitudes (González-Dávila et al., 2010; González-Dávila and 710 Santana-Casiano, 2023). 711

The correlation of Ω with pH_T (r²=0.90) with a level of significance higher than the 99% 712 explained that the individual components driving OA accompanied the declining in Ω . The 713 714 interannual trends in Ω_{Ca} and Ω_{Arag} (Figure 7, Table 2) exhibited the decrement through the whole water column along the NASPG with a level of statistical confidence generally higher 715 than the 90%. The rates of declining for Ω_{Ca} and Ω_{Arag} in the SPMW (0.011-0.021 and 0.007-716 0.013 units yr-1; respectively) were consistent with the trends observed up to 100 m depth at 717 ESTOC between 1995 and 2023 (0.019 ± 0.001 and 0.012 ± 0.001 units yr⁻¹, respectively; 718 González-Dávila and Santana-Casiano, 2023) and in surface waters at the IS-TS between 1985 719 and 2008 (0.0117 ± 0.0011 and 0.0072 ± 0.0007 units yr⁻¹, respectively; Olafsson et al., 2009). 720 The Ω_{Arag} trend estimated for SPMW in the Irminger basin (-0.007 ± 0.003 units yr⁻¹) is 721 consistent with that reported for surface waters by Bates et al., (2014) over 1983-2014 (-0.008 722

 ± 0.004 units yr⁻¹) and fall within the range of those estimated during summer by Leseure et 723 al., 2020 over 2008-2017 (-0.005 \pm 0.001 units yr⁻¹). Chau et al., 2014 recently deduced from 724 reconstructed products a slower decrease (-0.004 ± 0.001 units yr⁻¹), highlighting the large 725 uncertainty in the estimations of interannual trends for pH and Ω_{Arag} across the NASPG due to 726 the low-data sampling frequency at their monitoring sites. The declining in Ω_{Arag} in the SPMW 727 accelerated by ~26% and ~51% in the Irminger and Iceland basins, respectively, in comparison 728 729 with the trends given for the period 1991-2018 (0.0052 ± 0.0006 and 0.0049 ± 0.0015 units yr⁻ ¹, respectively; García-Ibáñez et al., 2021). The observed decrease in Ω_{Arag} in the SPMW was 730 ~23% faster in the Rockall Trough than in the adjacent Iceland basin. The interannual declining 731 for Ω_{Ca} and Ω_{Arag} in the ENACW (0.012 and 0.008 units yr⁻¹, respectively) agreed with these 732 733 previous observations but were not statistically significances likely due to the high variability modifying the changes in pH_T in this layer (see section 4.2). Despite the acceleration of the 734 735 acidification rates toward intermediate and deep layers, the declining rates weakened for Ω_{Ca} and even more for Ω_{Arag} (Table 2). Moreover, the vertical profiles were approximately constant 736 737 throughout the section in contrast with the heterogeneous vertical distribution of pH_T between basins. This behaviour was previously observed in the Irminger and Iceland basins by García-738 Ibáñez et al., (2021) and explained by pressure and temperature-induced changes in the 739 speciation of the CO₂-carbonate chemistry species (Jiang et al., 2015) and in the solubility of 740 741 calcite and aragonite (Mucci, 1983). Their combined action counterbalanced the alterations in Ω resulting from acidification, particularly in colder deep waters where the solubility of calcite 742 and aragonite was reduced (García-Ibáñez et al., 2021). However, the fall down in Ω_{Ca} and 743 Ω_{Arag} along the uLSW, LSW and ISOW accelerated by 40-75% in relation with the trends 744 reported by García-Ibáñez et al., (2021) for the Irminger and Iceland basins. The LSW and 745 746 ISOW presented faster declining rates for Ω_{Ca} and Ω_{Arag} in the Irminger (Table 2), which may be caused by the enhanced ventilation of the interior ocean which accelerated the acidification 747 (see section 4.2). The westward rise in depth of these layers along the Greenland continental 748 slope, accompanied by a subsequent elevation in the horizons of solubility, resulted in reduced 749 750 buffering capacity against acidification effects in the Irminger basin when compared to the Iceland basin. In contrast, the rise in depth of LSW in the Rockall Trough favour the increment 751 of ~0.2 units in Ω_{Ca} and Ω Arag with respect to the Iceland basin but had not influence on the 752 interannual trends, which were coinciding. The Ω_{Ca} and Ω_{Arag} in the DSOW, despite showed a 753

trend accelerated by ~30% compared to the observed by García-Ibáñez et al., (2021), presented the weakest interannual decreases throughout the section $(0.004 \pm 0.003 \text{ and } 0.002 \pm 0.001$ units yr⁻¹, respectively) due to the high pressure and low temperatures compensating the rapidly acidification (Figure 6, Table 2).

The decrease in Ω could have severe consequences on organisms reliant on aragonite, which 758 is less resistant to dissolution than calcite (Mucci, 1983; Broecker and Peng, 1983) and thus 759 760 expected to experience relatively higher susceptibility to the effects of OA over shorter time scales (Raven et al., 2005). The progressive reduction in Ω_{Arag} is driving a long-term decrease 761 in the depth of the aragonite saturation horizon ($\Omega_{\text{Arag}}=1$) by 80-400 m since the preindustrial 762 era (Álvarez et al., 2003; Feely et al., 2004; Pérez et al., 2013, 2018; Pérez et al., 2013; Tanhua 763 764 et al., 2007; Wallace, 2001) and is projected to shoal by more than 2000 m by the end of the century under the IS92a scenario (Orr et al., 2005). The vertical section of Ω_{Arag} in Figure S3 765 shows the shallower aragonite saturation horizon during 2009 and 2016 compared to 766 767 preindustrial times. Likewise, Orr et al., (2005) suggested that high-latitudes surface waters could become undersaturated when the atmospheric CO₂ concentration double the 768 769 preindustrial concentration within the next 50 years. It would reduce the calcification rates in some shallow calcifying organism by more than the 50% (Feely et al., 2004). 770

771 The planktonic aragonite-producers pteropods (e. g. Limacina helicina, Clio pyramidata), 772 which have high population densities in subpolar regions up 300 m depth (Bathmann et al., 773 1991; Urban-Rich et al., 2001) and play a key role in the export flux of both carbonate and organic carbon (Accornero et al., 2003; Collier et al., 2000), are expected to be highly 774 775 vulnerable to OA if the aragonite saturation horizon continue to shoal (Orr et al., 2005). The 776 undersaturation toward intermediate and upper layers negatively influence the aragonite-based 777 CWC (e. g. Lophelia pertusa, Madrepora oculate), which show their highest diversity and population along the NASPG between 200 and 1000 m depth among the global ocean (Roberts 778 779 et al., 2009). In fact, several studies reported that CWC ecosystems are anticipated to be among 780 the first deep-sea ecosystems to experience acidification threats (Gehlen et al 2014; Guinotte et al., 2006; Maier et al., 2009; Raven et al., 2005; Roberts et al., 2009; Turley et al., 2007), 781 particularly in the North Atlantic (Perez et al., 2018). The findings presented here contribute 782 to a deeper understanding of the biological impacts of OA along the NASPG. 783

784 4.5. Processes controlling OA and Ω trends

Due to the variety of processes involved in OA, a decomposition of the pH_T and Ω trends into 785 786 the individual components that govern their spatio-temporal variability was done (see section 2.2.6). The interannual variations in pH_T ($\frac{dpH_T}{dt}$) and Ω ($\frac{d\Omega}{dt}$) explained by fluctuations in 787 temperature $\left(\frac{\partial pH_T}{\partial T}\frac{\partial T}{dt}\right)$ and $\frac{\partial'\Omega}{\partial T}\frac{\partial T}{dt}$, salinity $\left(\frac{\partial pH_T}{\partial S}\frac{\partial S}{dt}\right)$ and $\frac{\partial'\Omega}{\partial S}\frac{\partial S}{dt}$, $A_T \left(\frac{\partial pH_T}{\partial A_T}\frac{\partial A_T}{dt}\right)$ and C_T 788 $\left(\frac{\partial pH_T}{\partial C_T}\frac{\partial C_T}{dt}\right)$ and $\frac{\partial \Omega}{\partial c_T}\frac{\partial C_T}{dt}$ were calculated for each layer and basin (Eq. 2) and summarized in Table 789 3 and 4. The positive contributions of each of the drivers indicate increments while negative 790 contributions the opposite. The cumulative changes resulting from the distinct drivers (referred 791 to with the subscript "calculated" in Table 3 and 4) were consistent with the observed pH_T 792 trends (referred to with the subscript "obs" in Table 3 and 4), thereby instilling confidence in 793 the methodology. An exception was found at the DSOW, in which the strong $NA_{\rm T}$ decrease had 794 a crucial influence on declining Ω . 795

796 The minimal differences between observed and calculated rates of change have added coherence to the non-significant trends identified for pH_T and Ω trends and/or its drivers in 797 some basins and layers (Table 2, 3 and S4). In the entire section at SPMW, the $\frac{dpH_T}{dt}$ (calculated), 798 explained by the cumulative impact of its drivers (all of them statistically significant at the 799 95% level of confidence), aligns within a range of <0.0002 units yr⁻¹ with $\frac{dpH_T}{dt}$ (obs) (which 800 was not significant). In the Irminger and Iceland basins at intermediate and deep layers, the 801 $\frac{dpH_T}{dt}$ (obs) (statistically significant at least at the 95% level of confidence) were consistent 802 within the range of <0.001 units yr⁻¹ with $\frac{dpH_T}{dt}$ (calculated) (T, S and NA_T shows non-significant 803 trends at some of the intermediate and deep layers). The interannual variations were non-804 significant for pH_T neither for its drivers in the Rockall Trough at LSW and ENACW. The high 805 temporal dispersion of average data in these layers was mainly related to the rise in depth of 806 807 LSW along the eastern continental slope and its mixing with shallower waters coming from subtropical latitudes (Ellett et al., 1986; Harvey, 1982; Holliday et al., 2000). The substantial 808 809 variability in the Rockall Trough made it difficult to discern OA patterns and its drivers on an interannual scale. Therefore, long-term monitoring and the development of multidecadal-scale 810 811 studies are required in this area to derive significant conclusions.

812 The cooling and freshening modified the physical-driven pH_T changes compared with those 813 encountered by García-Ibáñez et al., (2016) during previous decades in the western NASPG. The cooling contributed to increase the pH_T and compensated the observed acidification rate. 814 The increase in pH_T due to temperature fluctuations was maximum at SPMW (~ 0.001 units yr⁻ 815 ¹) and negligible in deeper layers (< 0.0003 units yr⁻¹ at uLSW and below). The increase in pH_T 816 due to salinity fluctuations was minimal (<0.0001 units yr⁻¹) through the whole water column 817 818 in the three basins, reflecting that the observed freshening caused insignificant changes in pH_T. The temperature and salinity contributed by 19.1-26.5% and 1.2-3.3%, respectively, in the total 819 pH_T change in the upper layers, while presented an influence three times lower toward the 820 interior ocean (1.3-7.6% and <0.6%, respectively). The enhanced convective processes in the 821 822 Irminger basin (e. g. Fröb et al., 2016; García-Ibáñez et al., 2015; Gladyshev et al., 2016a, 2016b; Piron et al., 2017) together with the rapid transport of LSW from the Labrador Sea to 823 824 the Irminger basin (Yashayaev et al., 2007) introduced differences in the thermal-driven pH_T with the Iceland basin, as previously reported by García-Ibáñez et al., (2016). The advection 825 826 of LSW through the Greenland continental slope also affected the DSOW (Read, 2000; Yashayaev and Dickson, 2008), which shows thermal-driven pH_T changes consistent with 827 those encountered through the LSW in the Irminger basin. 828

829 Despite the negligible direct contribution of the salinity fluctuations over the pH_T changes, the freshwaters fluxes influence the distribution of $A_{\rm T}$ and $C_{\rm T}$ indirectly affecting pH_T trends. After 830 831 removing salinity effects, NA_T show positive trends in subsurface layers and negative trends 832 toward the interior ocean (Figure S4 and Table S4; detailed in Appendix B). The changes in $NA_{\rm T}$ described the 7.8-10.1% of the total pH_T change at SPMW. The $NA_{\rm T}$ -driven pH_T changes 833 weakened with depth (Table 3) due to the insignificantly interannual changes in NA_T through 834 LSW and ISOW (Table S4). The weak contribution of *NA*_T in these layers (1.3-5.1%) could be 835 related to the difficulty of reversing the large alkalinization until the 2000s resulted from the 836 837 slowdown in the formation of LSW since the mid-90s (Lazier et al., 2002; Yashayaev, 2007), which was transmitted towards deeper overflow waters (Sarafanov et al., 2010). The 838 839 substantial interannual changes and the abrupt change between periods of increase and decrease of the seawater properties at DSOW (Yashayaev et al., 2003; Stramma et al., 2004) 840 841 linked with changes in the LSW formation (Dickson et al., 2002) explained the rapidly decrease in NA_T (Table S4), which described the 14.6% of the pH_T declining. 842

The increase in $NC_{\rm T}$ drove by the rise in $C_{\rm ant}$ was found to govern the acidification, with a 843 contribution higher than the 67% across the entire water column. The NC_T-driven pH_T 844 declining was close to twice the observed and calculated acidification rates through the SPMW 845 (Table 3). However, the contribution of NC_T at SPMW (67-69%) was lower than the 846 encountered toward the interior ocean (82-96%) due to the relevance of temperature and $A_{\rm T}$ 847 over pH_T trends in the upper layers. The cooling and increase in NA_T counteracted the 848 849 acidification expected by the increasing $C_{\rm T}$ at SPMW by 28-34% and 11-15%, respectively. In intermediate and deep layers, the thermal-neutralization of the $C_{\rm T}$ -driven acidification was 850 851 weaker (1.5-9.3%) and the decreasing NA_T contributed to decrease the pH_T by < 15%. Freshening played a minor role in countering acidification (<6% in upper layers and <2% in 852 853 the interior ocean).

In line with declining pH_T, 79-83% of the decrease in Ω in subsurface layers was attributed to 854 the C_{ant} -driven rise in NC_T , with this influence reaching up to 97% in deeper waters. The 855 increase in NA_T in the SPMW accounted by 10.4-13.0% in the Ω trends and counteracted its 856 *NC*_T-driven decrease by 12.6-16.2%. The contribution of *NA*_T fall and reversed toward deeper 857 waters, explained <6% of the decline in Ω in the uLSW, LSW and ISOW in the Irminger basin 858 859 and <11% in the Iceland basin. The pronounced impact of the rapid decrease in $NA_{\rm T}$ on the acidification of the DSOW (see section 4.3) depicted the greater contribution of NA_{T} 860 encountered among the Irminger basin (16%) and compensated the $NC_{\rm T}$ -driven decrease in Ω 861 by 36.4%. In the Rockall Trough, the contribution of $NC_{\rm T}$ changes on Ω was reduced at LSW 862 863 (78.2-79.0%) compared to the Irminger basin (94.5%) while the effect of NA_T fluctuations 864 tripled until reach 12.6-12.7%.

865 Despite the crucial role of cooling in mitigating acidification, temperature fluctuations have an opposite effect on Ω due to the thermodynamic relationship inherent in the acid-base 866 equilibrium of the CO₂-carbonate system (Dickson and Millero, 1987). In the Irminger and 867 Iceland basins, the observed decrease in temperature contributed negligibly to the decline in 868 Ω (3.6% in the SPMW and less than 2% in intermediate and deep waters). The influence of 869 salinity, as with the pH_T trends, was minimal: the observed freshening slightly elevated the 870 Ω trends, offsetting the decline by 4.6-4.7% in the SPMW, 1.1-2.1% in the uLSW and LSW, 871 and 0.5-1.2% in the ISOW and DSOW. Even with the slightly faster cooling and freshening 872

observed in the Rockall Trough, the contributions of temperature and salinity to Ω did not exceed 7% in any of its layers.

875 **5.** Conclusions

This research has evaluated the interannual changes in the basin-wide MCS dynamics along 876 the NASPG during 2009-2019. Despite the observational period is relatively short to quantify 877 long-term trends and to formulate significant future projections, the finding has allowed to 878 evaluate the ocean response, in terms of MCS dynamics and on an interannual scale, to changes 879 880 in deep-water convection and to isolate events affecting the physical patterns. The assessment of OA within the Irminger and Iceland basins was enhanced by supplying novel data and trends 881 882 spanning a decade in which the physical patterns reversed. Additionally, the study provides an unprecedent analysis of the physico-chemical variations in the Rockall Trough, which is 883 884 crucial for the assessment of the entire longitudinal span of the NASPG. It facilitates a more 885 accurate understanding of the mechanisms dictating basin-scale acidification processes and 886 advances our understanding of OA in the North Atlantic and Global Ocean.

Overall, the entrance and accumulation of Cant and interannual acidification trends were 887 strongly affected by the cooling, freshening and enhancement in the oxygenation during this 888 decade. The longitudinal span of the NASPG and the differences in circulation patterns, water 889 masses and bathymetry behaved as a source of spatio-temporal variability. The interannual 890 acidification trends of the main water masses across the NASPG ranged between 0.0006-891 892 0.0032 units yr-1 and caused a decline in the Ω Ca and Ω Arag of 0.004-0.021 and 0.003-0.013 893 units yr-1, respectively. The convective processes increased the accumulation rates of Cant in the interior ocean by 50-86% and accelerated the acidification rates by around 10% compared 894 895 to previous decades in the Irminger and Iceland basins. The shallower hydrography of the Rockall Trough and the poleward circulation patterns accounted for differences in the 896 acidification rates respect to surrounding waters. 897

The Cant-driven increase in NCT was found to govern the acidification of the NASPG with contributions exceeding 60%. The combined effect of the decreasing temperature, salinity and NAT neutralized close to one-half of the acidification along the entire longitudinal span of the SPMW. The enhanced deep-water ventilation in the western NASPG slowdown the cooling

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and freshening toward the interior ocean, weakening the physical counterbalance ofacidification.

904 The present investigation emphasizes the progressively increase in the uptake and accumulation of Cant and subsequent acceleration of OA along the NASPG. Novel data and 905 906 results provided could be compared with other repeated hydrographic section data at mid and high latitudes in the North Atlantic, such as the A02, A25, AR07E and AR28 framed in the 907 908 GO-SHIP program, as well as used in conjunction to develop future investigations. Additionally, they contribute to the improvement of the projections pertaining to the future 909 state of the oceans run by models and forecast. Considering the important variability in the 910 mechanism controlling the distribution of the physico-biogeochemical properties and 911 912 particularly the OA in the North Atlantic, this research aims to highlight the necessity of continue monitoring and sampling the whole water column through repeated hydrographic 913 sections, especially through the highly variable but less assess easternmost part. 914

915 Appendix A: Correction of Dissolved Oxygen records for the cruise of 2019

916 The sensor-measured DO data for the cruise of 2019 were corrected by considering the DO 917 output data given by the neural network ESPER NN (Carter et al., 2021) for the cruises of 2016 and 2019 (hereinafter ESPER-estimated DO) and the WINKLER-measured DO during 918 919 the cruise of 2016. Among the 16 equations provided by the ESPER NN that differently combines seawater properties as predictors, we use the equation 8 which only need as inputs 920 921 the T and S (due to lack of measured macronutrients during the cruise of 2019) along with 922 latitude, longitude, depth and date (see Table 2 in Carter et al., 2021). The reported Root Mean Squared Error (RMSE) of equation 8 for DO estimations in the global ocean is \pm 9.7 µmol kg-923 1, which is reduced for intermediate waters (1000-1500 m) to \pm 5.9 µmol kg⁻¹ (see Table 7 in 924 Carter et al., 2021). Additionally, a new set of DO for 2019 based on WINKLER data for 2016 925 926 was computed, which was referred in this study as "pseudo-WINKLER" data. The difference between WINKLER-measured and ESPER-estimated DO during 2016 was interpolated to the 927 longitudes and depths of the samples of 2019 by applying Delaunay Triangulation. The 928 pseudo-WINKLER data was described as the sum of these interpolated differences and the 929 930 ESPER-estimated DO data for 2019. The longitudinal distribution of measured and ESPER-

estimated DO data for 2016 and 2019 is depicted in Figure S1a and S1b. The interpolatedpseudo-WINKLER data for the cruise of 2019 were included in Figure S1a.

The sensor records of DO in 2019 were in average 4.90 µmol kg⁻¹ lower than the ESPER-933 estimated and 10.31 µmol kg⁻¹ lower than the pseudo-WINKLER. A higher discrepancy was 934 observed in the average sensor-measured DO in the east part (237.60 \pm 15.00 μ mol kg⁻¹) 935 compared with the west part (281.40 \pm 14.75 µmol kg⁻¹). The average differences (measured 936 minus ESPER-estimated DO and measured minus pseudo-WINKLER DO, $\Delta DO_{meas-ESPER}$ and 937 $\Delta DO_{meas-pseudoWINLKER}$, respectively; Figure S2c and S1d) shows that the sensor records were 938 strongly underestimated in the east part (-20.98 \pm 10.91 and -28.77 \pm 12.60 μ mol kg⁻¹, 939 respectively) and weakly overestimated in the west part (8.59 ± 8.53 and $5.18 \pm 12.02 \mu mol$ 940 kg⁻¹, respectively) during the cruise of 2019. These differences were corrected separately west 941 and east of 21.5°W by using the relationship $\frac{\Delta DO_{meas-pseudoWINKLER}}{measured DO}$. The averages of this 942 relationship in the west and east part of the transect (0.016 and -0.12 µmol kg⁻¹, respectively) 943 were used as corrector factors. The corrected DO values were given by the product between 944 the measured DO and $\left(1 - \frac{\Delta DO_{meas-pseudoWINKLER}}{measured DO}\right)$. 945

946 Appendix B: Interannual trends of A_T and NA_T

The interannual trends of $A_{\rm T}$ (Figure S4 and Table S4) was found to be highly impacted by 947 freshening, with decreasing rates ranging from -0.33 to -0.71 µmol kg⁻¹ yr⁻¹ among the 948 SPMW and ENACW and from -0.01 to -0.18 µmol kg⁻¹ yr⁻¹ within the uLSW, LSW, ISOW 949 and DSOW. It contrasts with the minimal interannual changes and slight rates of increase in 950 A_T encountered among the different layers by García-Ibáñez et al., (2016) from 1991 to 2015 951 in the Irminger basin (between 0.10 and 0.28 µmol kg⁻¹ yr⁻¹) and Iceland basin (between -952 0.04 and 0.07 µmol kg⁻¹ yr⁻¹), and with the trends reported for the period 1983-2013 by Pérez 953 et al., (2021) at the IRM-TS (between 0.13 and 0.22 µmol kg⁻¹ yr⁻¹) and at the IS-TS (between 954 -0.04 and 0.15 μ mol kg⁻¹ vr⁻¹). These heterogeneities in the temporal evolution of the A_T were 955 956 driven by the decadal salinification of the whole water column observed since the late 20th 957 century and interrupted by interannual freshening episodes such as during the 2010s.

958 The interannual increase in $NA_{\rm T}$ in upper layers could be related to acidification, which favour 959 the dissolution of carbonates, combined with increasing biological production reported for

- 960 upper layers across the NASPG (Ostle et al, 2022). It contrasts with the constant to weakly 961 decrease in $NA_{\rm T}$ at intermediate and deep layers, in which the accelerated acidification was
- 962 compensated by the dominance of remineralization processes over lower biological uptake.
- 963 Consequently, the positive NA_{T} trends encountered in the upper layers lead a rise in pH_T,
- while the diminished $NA_{\rm T}$ contributed to decrease the pH_T toward the interior ocean.
- 965 The A_T/S relationship has increased at a rate of $0.5 \pm 0.2 \mu$ mol kg⁻¹ yr⁻¹ (p-value < 0.05) due
- to the combined action of the freshening (Figure 4) and the progressive increase of A_T -rich
- 967 water inflows through upper layers (observed in the positive trends of NA_T in SPMW and
- 968 ENACW; Figure S4). This was likely associated with the stagnation of A_T -rich subtropical
- 969 waters in the upper layers due to the slowdown of the NASPG since the mid-90s (e.g., Böning
- et al., 2006; Häkkinen and Rhines, 2004), along with changes in the spreading of waters from
- 971 higher latitudes influenced by melting.

972 Code Availability

973 MATLAB and R codes for CANYON-B available are at https://github.com/HCBScienceProducts/CANYON-B. MATLAB and R code for ESPER NN 974 are available at https://github.com/BRCScienceProducts/ESPER. MATLAB code for 975 976 anthropogenic carbon calculation is available at http://oceano.iim.csic.es/ media/cantphict0 toolbox 20190213.zip. The CO₂SYS programme 977 for MATLAB is available at https://github.com/jonathansharp/CO2-System-Extd. 978

979 Data Availability Statement

The measured surface-to-bottom CLIVAR data (2009-2019) used in this investigation are 980 published in open-access at Zenodo (DOI: 10.5281/zenodo.10276221). The GO-SHIP A25-981 the data for available 982 **OVIDE** cruise of 2018 is at SEANOE (https://www.seanoe.org/data/00762/87394/). 983

984 Author contribution

985 DCH contributed with data analysis and wrote the manuscript. FFP, DCH, AV, DGS, AGG, 986 MGD and JMSC worked on the design, conceptualization and data preparation. SG, AS, MGD, JMSC, AGG and DGS participated in 8, 4, 7, 7, 2 and 2 cruises, respectively. SG and AS were 987 the Chief Scientist in all cruises and responsible for the operational and maintenance 988 989 procedures for the CTD and additional sensors and thus for physical and sensor-measured variables. MGD and JMSC got the funding acquisition and provision of resources for the 990 Spanish team from the ULPGC. SG and AS got the funding for ship time and provision of 991 992 resources for all the cruise participants. All authors critically revised the manuscript.

993 **Competing interest**

994 The authors declare that the research was conducted in the absence of any commercial or 995 financial relationships that could be construed as a potential conflict of interest.

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1011 Legend for figures

Figure 1. (a) Map of the North Atlantic Subpolar Gyre (NASPG) with the schematic diagram 1012 of the surface and deep circulation patterns compiled from Lherminier et al., (2010); Pérez et 1013 al., (2021); Sarafanov et al., (2012); Schmitz and McCartney, (1993); Schott and Brandt, 1014 (2007) and Sutherland and Pickart, (2008). The acronyms are defined as follow: the 1015 bathymetric features are shown in grey (RR: Reykjanes Ridge, HB: Haton Bank, GBB: George 1016 Bligh Bank, CGFZ: Charlie-Gibbs Fracture Zone, GIR: Greenland-Iceland Ridge, and GSR: 1017 Greenland-Scotland Ridge), the surface currents are shown in orange (NAC: North Atlantic 1018 Current, and IC: Irminger Current) and the deep water circulation is shown in blue and purple 1019 (ISOW: Iceland-Scotland Overflow Water, DSOW: Denmark Strait Overflow Water, LSW: 1020 Labrador Sea Water, and DWBC: Deep Western Boundary Current). The longitudinal 1021 distribution of the surface-to-bottom sampling stations along the cruise track of 2016 (repeated 1022 throughout the cruises) is shown with red dots. The black lines along the cruise track delimited 1023 the three basins. (b) Vertical distribution of the water masses considered in this study for each 1024 of the basins. The isopycnals, plotted over the salinity distribution for the cruise of 2016, show 1025 the limits of the layers and were defined by potential density (in kg m⁻³) referred to 0 dbar (σ_0). 1026 The vertical gray lines show the limits between basins. The water masses (ENACW: Eastern 1027 North Atlantic Central Water, SPMW: Subpolar Mode Water, uLSW: upper Labrador Sea 1028 Water, LSW: Labrador Sea Water, ISOW: Iceland-Scotland Overflow Water, DSOW: Denmark 1029 Strait Overflow Water) and the selection of potential density values delimiting the layers are 1030 1031 described in section 2.2.4. Figure produced with Ocean Data View (Schlitzer, 2021).

Figure 2. Water-column distribution along the longitudinal transect of (a) temperature, (b)
salinity, (c) A_T, (d) pH_T and (e) AOU for the cruises of 2009 (left plots) and 2016 (right plots).
The vertical white lines show the limits between basins. Figure produced with Ocean Data
View (Schlitzer, 2021).

Figure 3. Water-column distribution along the longitudinal transect of (a) C_T, (b) C_{ant} and (c)
 C_{nat} for the cruises of 2009 (left plots) and 2016 (right plots). The vertical white lines show the
 limits between basins. Figure produced with Ocean Data View (Schlitzer, 2021).

Figure 4. Temporal distribution (2009-2019) of the average temperature and salinity in each of 1039 1040 the layers considered for the Irminger (left plot column), Iceland (central plot column) and Rockall basins (right plot column). The average values were calculated for each cruise and 1041 layer and represented with coloured points together with their respective error bars at the time 1042 of each cruise (the method used for calculations was described in section 3.2). In the Irminger 1043 plots, the empty points represent the average values for 2019 calculated with the measured 1044 data available in the easternmost part of the basin (sampled part during this cruise), while the 1045 coloured points for 2019 represent the average values corrected with A25-OVIDE-2018 data. 1046 The interannual trends were given by linear regression of the average values, with the values 1047 of the slope, the standard error of estimate and the r^2 presented in Table 2. 1048

Figure 5. Temporal distribution (2009-2019) of the average C_T, C_{ant} and C_{nat} in each of the layers considered for the Irminger (left plot column), Iceland (central plot column) and Rockall basins (right plot column). The average values were calculated for each cruise and layer and represented with coloured points together with their respective error bars at the time of each cruise (the method used for calculations was described in section 3.2). In the Irminger plots, the empty points represent the average values for 2019 calculated with the measured data available in the easternmost part of the basin (sampled part during this cruise), while the coloured points for 2019 represent the average values corrected with A25-OVIDE-2018 data. The interannual trends were given by linear regression of the average values, with the values of the slope, the standard error of estimate and the r^2 presented in Table 2.

Figure 6. Temporal distribution (2009-2019) of the average pH_T (in situ temperature) in each 1059 of the layers considered for the Irminger (left plot column), Iceland (central plot column) and 1060 Rockall basins (right plot column). The average values were calculated for each cruise and 1061 layer and represented with coloured points together with their respective error bars at the time 1062 of each cruise (the method used for calculations was described in section 3.2). In the Irminger 1063 plots, the empty points represent the average values for 2019 calculated with the measured 1064 data available in the easternmost part of the basin (sampled part during this cruise), while the 1065 coloured points for 2019 represent the average values corrected with A25-OVIDE-2018 data. 1066 The interannual trend were given by linear regression of the average values, with the values of 1067 the slope, the standard error of estimate and the r^2 presented in Table 2. 1068

Figure 7. Temporal distribution (2009-2019) of the average Ω Ca and Ω Arag in each of the 1069 layers considered for the Irminger (left plot column), Iceland (central plot column) and Rockall 1070 1071 basins (right plot column). The average values were calculated for each cruise and layer and 1072 represented with coloured points together with their respective error bars at the time of each cruise (the method used for calculations was described in section 3.2). In the Irminger plots, 1073 1074 the empty points represent the average values for 2019 calculated with the measured data available in the easternmost part of the basin (sampled part during this cruise), while the 1075 coloured points for 2019 represent the average values corrected with A25-OVIDE-2018 data. 1076 The interannual trends were given by linear regression of the average values, with the values 1077 of the slope, the standard error of estimate and the r^2 presented in Table 2. 1078

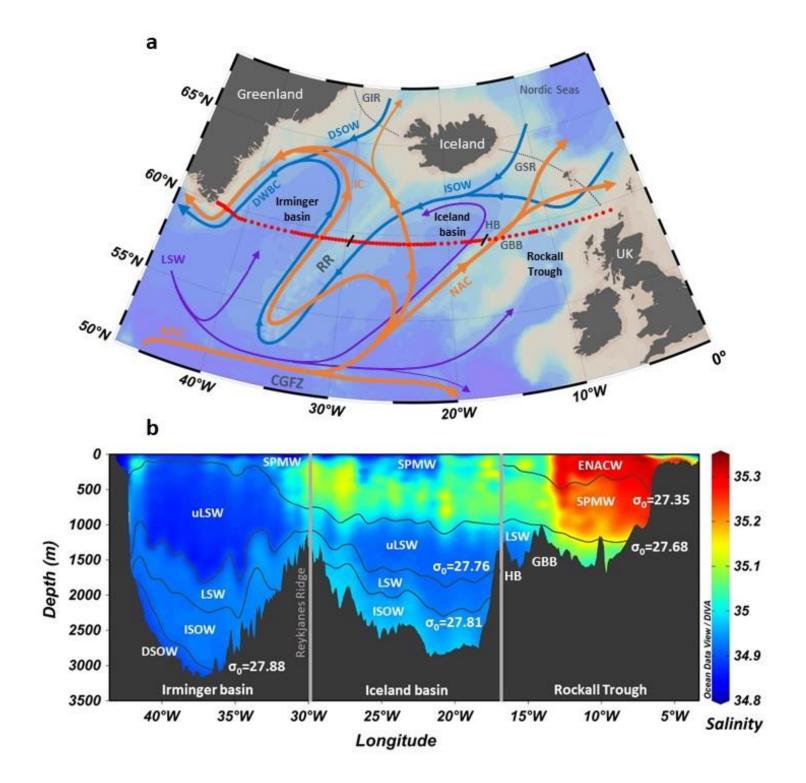
1079 Legend for Tables

- 1080 Table 1. Metadata list of hydrographic cruises.
- 1081 Table 2. Interannual trends of temperature, salinity, C_T , C_{ant} , C_{nat} , pH_T , ΩCa and $\Omega Arag$ in each 1082 of the layers and basins. The ratios of change were based on linear regressions applied to the 1083 average values (as represented in Figures 4-7) and presented together with its Standard error 1084 of estimate. The correlation coefficients r^2 and p-values were also provided. Values in bold 1085 denote trends statistically significant at the 95% level of confidence.
- 1086Table 3. Temporal changes in pHT (in 10-3 units yr-1) explained by fluctuations in temperature1087 $\left(\frac{\partial pH_T}{\partial T}\frac{\partial T}{dt}\right)$, salinity $\left(\frac{\partial pH_T}{\partial S}\frac{\partial S}{dt}\right)$, AT $\left(\frac{\partial pH_T}{\partial A_T}\frac{\partial NA_T}{dt}\right)$, and CT $\left(\frac{\partial pH_T}{\partial C_T}\frac{\partial NC_T}{dt}\right)$ in each of the layers1088considered for the Irminger, Iceland and Rockall basins during the period 2009-2019. The sum1089of changes explained by the individual drivers represents the calculated interannual pHT1090change $\left(\frac{dpH_T}{dt}$ calculated), as detailed in section 2.2.5. The observed interannual pHT trends1091 $\left(\frac{dpH_T}{dt}$ observed), shown in Figure 7 and provided in Table 2, were also added to the table for1092comparison.
 - 37

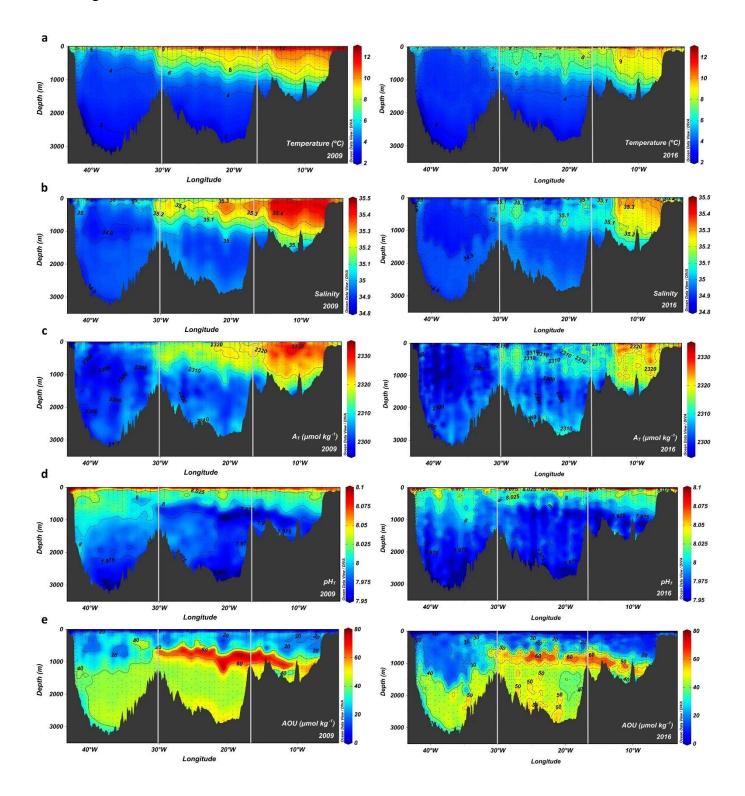
Table 4. Temporal changes in Ω Ca and Ω Arag (in 10⁻³ units yr⁻¹) explained by fluctuations in temperature $\left(\frac{\partial \Omega}{\partial T} \frac{\partial T}{\partial t}\right)$, salinity $\left(\frac{\partial \Omega}{\partial S} \frac{\partial S}{\partial t}\right)$, A_T $\left(\frac{\partial \Omega}{\partial A_T} \frac{\partial NA_T}{\partial t}\right)$, and C_T $\left(\frac{\partial \Omega}{\partial C_T} \frac{\partial NC_T}{\partial t}\right)$ in each of the layers considered for the Irminger, Iceland and Rockall basins during the period 2009-2019. The sum of changes explained by the individual drivers represents the calculated interannual Ω change $\left(\frac{d\Omega}{dt}$ calculated), as detailed in section 2.2.6. The observed interannual Ω trends $\left(\frac{d\Omega}{dt}$ observed), shown in Figure 6 and provided in Table 2, were also added to the table for comparison.

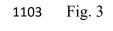
| 1100 Figures |
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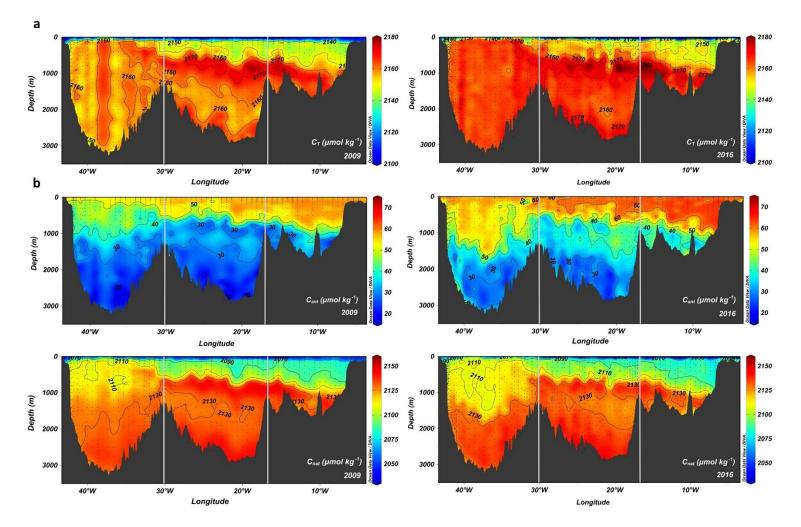
1101 Fig. 1



1102 Fig. 2







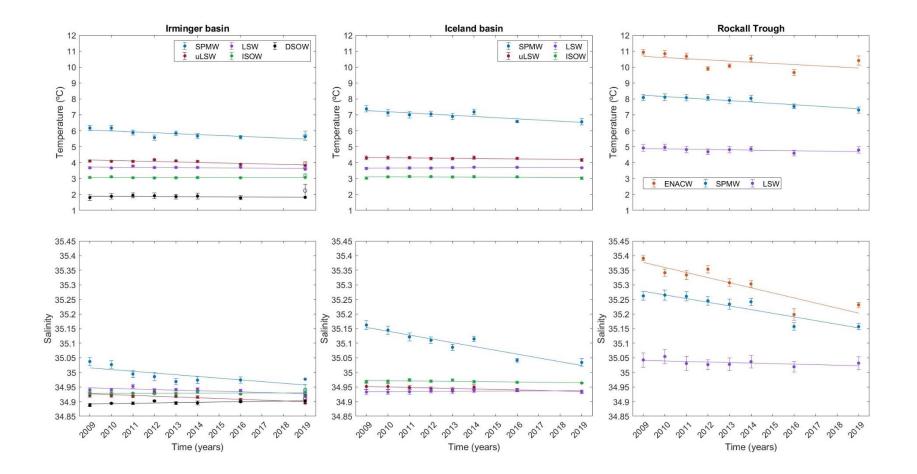


Fig. 4



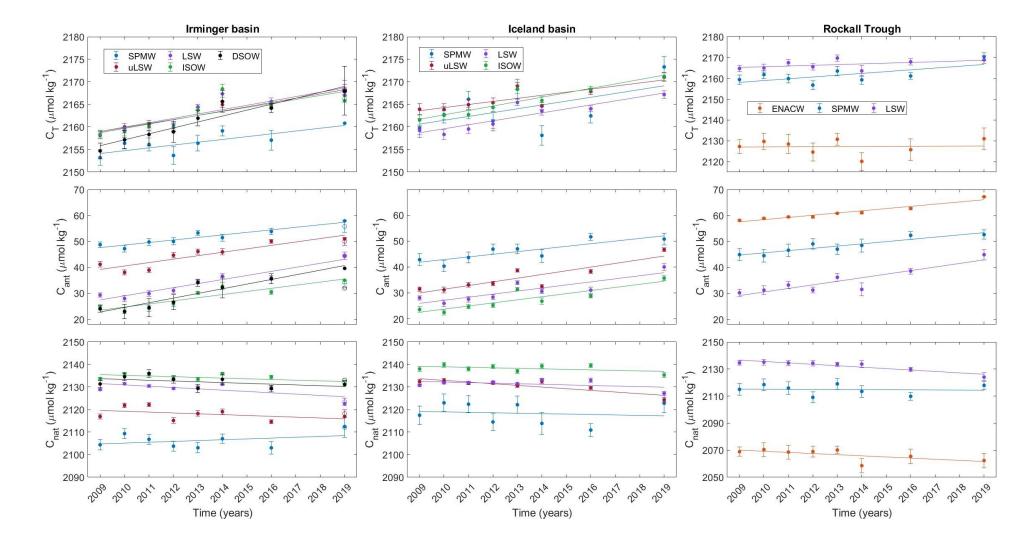




Fig. 6

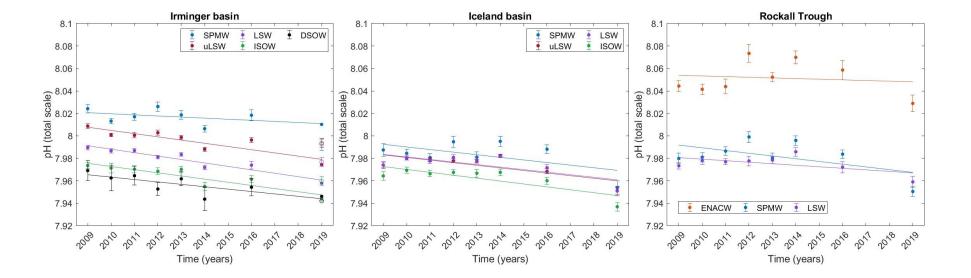
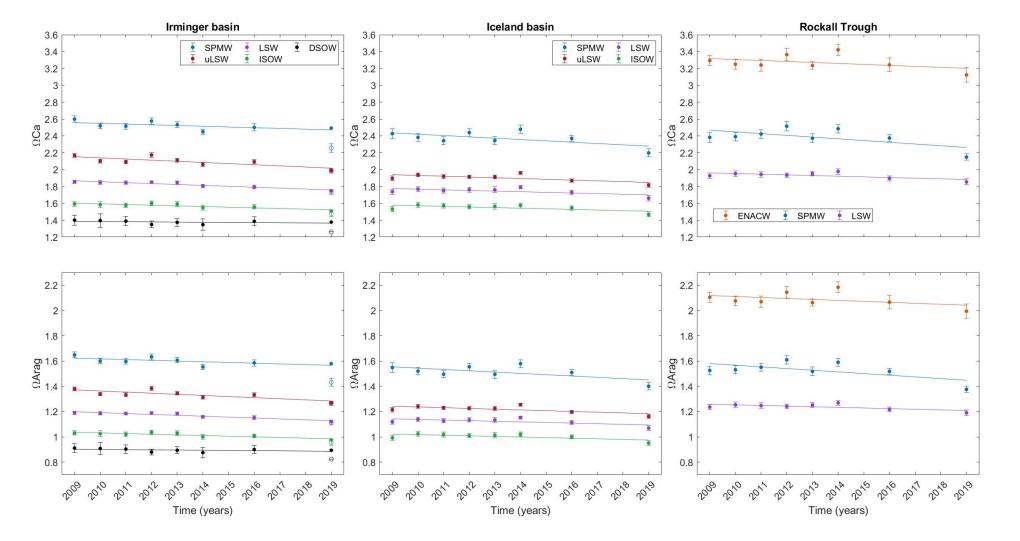


Fig. 7



| Tal | hl | e | 1 |
|-----|----|---|---|
| Ia | U | | T |

| Year | Cruise ID | Date | Research Vessel (R/V) | Chief Scientist | Number of stations | MCS measured variables | |
|------|-----------|----------------|---------------------------|-----------------|--------------------|--|--|
| 2009 | AI28 | Aug 15-Sept 27 | Akademik Ioffe | A. Sokov | 67 | A_T , C_T and pH | |
| 2010 | AI31 | Sep 2-Sep 27 | Akademik Ioffe | A. Sokov 84 | | A_T , C_T and pH | |
| 2011 | SV33 | Sep 9-Sep 28 | Akademik Sergey Vavilov | A. Sokov | 98 | A_T , C_T and pH | |
| 2012 | AI38 | May 25-Jul 1 | Akademik Ioffe | S. Gladyshev | 66 | A _T , C _T and pH | |
| 2013 | AI41 | Jun 26-Jul 23 | Akademik Ioffe | S. Gladyshev | 75 | A _T , C _T and pH | |
| 2014 | AI44 | Jun 27-Jul 20 | Akademik Ioffe | S. Gladyshev | 76 | A_T , C_T and pH | |
| 2016 | AI51 | Jun 3-Jul 13 | Akademik Ioffe | S. Gladyshev | 104 | A_T , C_T and pH | |
| 2019 | AMK77 | Aug 8-Sep 10 | Akademik Mstislav Keldysh | S. Gladyshev | 47 | A _T and C _T | |

| Tal | ble | 2 |
|-----|-----|---|
|-----|-----|---|

| Basin | Layer | Temperature | Salinity | C _T | C _{ant} | C _{nat} | рН _т | Ωca | ΩArag |
|----------|-------|--|---|--|--|--|--|---|--|
| | | ratio (°C yr ⁻¹) r^2 p-value | ratio (psu yr ⁻¹) r^2 p-value | ratio (μ mol kg ⁻¹ yr ⁻¹) r ² p-value | ratio (μ mol kg ⁻¹ yr ⁻¹) r ² p-value | ratio (μ mol kg ⁻¹ yr ⁻¹) r ² p-value | ratio $(10^{-3} \text{ units yr}^{-1})$ r ² p-value | ratio (units yr ⁻¹) r^2 p-value | ratio (units yr^{-1}) r^2 p-value |
| | SPMW | -0.058 ± 0.024 0.60 0.02 | -0.006 ± 0.003 0.59 0.03 | 0.62 ± 0.23 0.66 0.02 | 0.95 ± 0.17 0.89 <0.01 | -1.00 ± 0.42 0.60 0.02 | $-1.25 \pm 0.93 0.32 0.14$ | -0.011 ± 0.006 0.50 0.05 | -0.007 ± 0.003 0.53 0.04 |
| | uLSW | $-0.014 \pm 0.011 \ 0.30 \ 0.16$ | -0.002 ± 0.001 0.59 0.03 | 1.02 ± 0.18 0.89 <0.01 | 1.48 ± 0.29 0.87 <0.01 | -0.47 \pm 0.38 0.28 0.18 | -2.62 ± 0.69 0.79 <0.01 | $-0.008 \ \pm \ 0.005 \ \ 0.40 \ \ 0.09$ | $-0.006 \ \pm \ 0.003 \ \ 0.44 \ \ 0.08$ |
| Irminger | LSW | $-0.010~\pm~0.008~~0.31~~0.15$ | -0.002 ± 0.001 0.50 0.05 | 0.98 ± 0.26 0.78 <0.01 | 1.53 ± 0.23 0.92 <0.01 | $-0.54 \pm 0.30 0.46 0.06$ | -3.17 ± 0.52 0.91 <0.01 | -0.014 ± 0.003 0.85 <0.01 | -0.009 ± 0.002 0.85 <0.01 |
| | ISOW | $-0.002 \ \pm \ 0.003 \ \ 0.11 \ \ 0.42$ | $0.000 \ \pm \ 0.000 \ \ 0.00 \ \ 0.99$ | 0.90 ± 0.34 0.64 0.02 | 1.18 ± 0.29 0.81 <0.01 | -0.27 \pm 0.20 0.32 0.14 | -2.97 ± 0.70 0.83 <0.01 | -0.010 ± 0.003 0.73 <0.01 | -0.007 ± 0.002 0.74 <0.01 |
| | DSOW | $-0.008 \pm 0.008 \ 0.22 \ 0.25$ | $0.001 \ \pm \ 0.001 \ \ 0.43 \ \ \ 0.08$ | 1.32 ± 0.23 0.90 <0.01 | 1.77 ± 0.32 0.89 <0.01 | $\begin{array}{rrrr} \text{-}0.32 & \pm & 0.33 & 0.19 & 0.28 \end{array}$ | -2.41 ± 0.87 0.67 <0.01 | $-0.004 \ \pm \ 0.003 \ \ 0.39 \ \ 0.10$ | $-0.003 \ \pm \ 0.002 \ 0.46 \ 0.07$ |
| | SPMW | -0.074 ± 0.022 0.74 <0.01 | -0.013 ± 0.002 0.89 <0.01 | $0.85 \pm 0.64 0.32 0.15$ | 1.02 ± 0.31 0.74 <0.01 | -0.19 \pm 0.74 0.02 0.75 | $-2.32 \pm 1.63 0.34 0.13$ | -0.016 \pm 0.010 0.37 0.11 | $-0.010 \ \pm \ 0.007 \ \ 0.39 \ \ 0.10$ |
| Tecleud | uLSW | -0.012 ± 0.005 0.63 0.02 | -0.002 ± 0.000 0.76 <0.01 | 0.68 ± 0.22 0.71 <0.01 | 1.42 ± 0.38 0.78 <0.01 | -0.74 ± 0.21 0.75 <0.01 | -2.31 ± 1.01 0.58 0.03 | $-0.009 \ \pm \ 0.005 \ 0.46 \ 0.07$ | $-0.006 \ \pm \ 0.003 \ \ 0.47 \ \ 0.06$ |
| Iceland | LSW | $0.005 ~\pm~ 0.003 ~~ 0.43 ~~ 0.08$ | $0.000 \ \pm \ 0.000 \ \ 0.28 \ \ \ 0.18$ | 0.88 ± 0.22 0.80 <0.01 | 1.18 ± 0.35 0.75 <0.01 | -0.26 \pm 0.26 0.20 0.27 | -2.26 ± 1.06 0.54 0.04 | $\text{-}0.008 \hspace{0.2cm} \pm \hspace{0.2cm} 0.005 \hspace{0.2cm} 0.41 \hspace{0.2cm} 0.09$ | $-0.005 \ \pm \ 0.003 \ \ 0.41 \ \ 0.09$ |
| | ISOW | $-0.003 \pm 0.006 \ 0.05 \ 0.61$ | -0.001 ± 0.000 0.47 0.05 | 0.98 ± 0.17 0.89 <0.01 | 1.20 ± 0.32 0.79 <0.01 | -0.23 \pm 0.21 0.23 0.23 | -2.58 ± 0.99 0.64 <0.01 | $-0.007 \ \pm \ 0.004 \ \ 0.42 \ \ 0.08$ | $-0.005 \ \pm \ 0.003 \ \ 0.43 \ \ 0.08$ |
| | ENACW | $-0.073 \pm 0.061 \ 0.27 \ 0.19$ | -0.017 ± 0.004 0.80 <0.01 | $0.05 \ \pm \ 0.57 \ 0.00 \ 0.92$ | 0.85 ± 0.11 0.94 <0.01 | $-0.84 \pm 0.50 0.43 0.08$ | $-0.58 \pm 2.31 0.02 0.77$ | $\begin{array}{rrrr} \text{-}0.012 & \pm & 0.013 & 0.18 & 0.30 \end{array}$ | $-0.008 \ \pm \ 0.008 \ \ 0.19 \ \ 0.28$ |
| Rockall | SPMW | -0.085 ± 0.019 0.84 <0.01 | -0.013 ± 0.003 0.85 <0.01 | 0.86 ± 0.46 0.48 0.05 | 0.87 ± 0.18 0.86 <0.01 | -0.07 \pm 0.59 0.00 0.88 | $-2.43 \pm 1.90 0.30 0.16$ | $\textbf{-0.021} \hspace{0.2cm} \pm \hspace{0.2cm} 0.013 \hspace{0.2cm} 0.38 \hspace{0.2cm} 0.10$ | $-0.013 \ \pm \ 0.008 \ \ 0.39 \ \ 0.10$ |
| | LSW | $-0.020 \pm 0.016 \ 0.29 \ 0.17$ | $-0.002 \pm 0.001 \ 0.30 \ 0.16$ | $0.35 \pm 0.29 0.27 0.19$ | 1.38 ± 0.34 0.81 <0.01 | -1.05 ± 0.24 0.84 <0.01 | $-1.36 \ \pm \ 0.97 0.34 0.13$ | $-0.008 \ \pm \ 0.004 \ 0.45 \ 0.07$ | $-0.005 \ \pm \ 0.003 \ 0.45 \ 0.07$ |

| Table 3 | Tabl | le | 3 |
|---------|------|----|---|
|---------|------|----|---|

| Basin | Layer | $\frac{\partial pH_T}{\partial T}\frac{\partial T}{dt}$ | $\frac{\partial pH_T}{\partial S}\frac{\partial S}{dt}$ | $\frac{\partial pH_T}{\partial A_T}\frac{\partial NA_T}{dt}$ | $\frac{\partial pH_T}{\partial C_T}\frac{\partial NC_T}{dt}$ | $\frac{dpH_T}{dt} (obs)$ | $\frac{dpH_T}{dt} (calculated)$ |
|----------|-------|---|---|--|--|--|---------------------------------|
| | SPMW | $0.91 \hspace{.1in} \pm \hspace{.1in} 0.38$ | 0.05 ± 0.02 | $0.31 \hspace{.1in} \pm \hspace{.1in} 0.43$ | -2.67 ± 0.63 | -1.25 ± 0.93 | -1.41 ± 0.85 |
| Irminger | uLSW | $0.22 \hspace{.1in} \pm \hspace{.1in} 0.17$ | $0.02 \hspace{0.1in} \pm \hspace{0.1in} 0.01$ | $\textbf{-0.10} \hspace{0.1in} \pm \hspace{0.1in} \textbf{0.40}$ | $-2.99 \hspace{0.1 in} \pm \hspace{0.1 in} 0.53$ | $-2.62 \hspace{0.1in} \pm \hspace{0.1in} 0.69$ | -2.86 ± 0.68 |
| | LSW | $0.16 \hspace{0.2cm} \pm \hspace{0.2cm} 0.12$ | $0.01 \hspace{0.1in} \pm \hspace{0.1in} 0.01$ | $\textbf{-0.04} \hspace{0.2cm} \pm \hspace{0.2cm} \textbf{0.39}$ | $-2.85 \hspace{0.1 in} \pm \hspace{0.1 in} 0.62$ | $\textbf{-3.17} \hspace{0.2cm} \pm \hspace{0.2cm} \textbf{0.52}$ | -2.72 ± 0.74 |
| | ISOW | $0.03 \hspace{0.1in} \pm \hspace{0.1in} 0.05$ | $0.00 \hspace{0.1in} \pm \hspace{0.1in} 0.00$ | $-0.13 \hspace{0.1in} \pm \hspace{0.1in} 0.30$ | $-2.38 \hspace{0.2cm} \pm \hspace{0.2cm} 0.88$ | $-2.97 \hspace{0.1in} \pm \hspace{0.1in} 0.70$ | -2.48 ± 0.93 |
| | DSOW | $0.13 \hspace{.1in} \pm \hspace{.1in} 0.12$ | -0.01 ± 0.00 | $\textbf{-0.60} \hspace{0.1in} \pm \hspace{0.1in} \textbf{0.18}$ | -3.41 ± 0.62 | -2.41 ± 0.87 | -3.90 ± 0.66 |
| Iceland | SPMW | $1.15 \hspace{0.2cm} \pm \hspace{0.2cm} 0.35$ | 0.10 ± 0.02 | $0.61 \hspace{0.1in} \pm \hspace{0.1in} 0.19$ | -4.14 ± 1.76 | -2.32 ± 1.63 | -2.27 ± 1.81 |
| | uLSW | $0.19 \hspace{0.2cm} \pm \hspace{0.2cm} 0.08$ | $0.01 \hspace{0.1in} \pm \hspace{0.1in} 0.00$ | $-0.24 \hspace{0.2cm} \pm \hspace{0.2cm} 0.45$ | $-2.08 \hspace{0.1in} \pm \hspace{0.1in} 0.66$ | -2.31 ± 1.01 | -2.12 ± 0.80 |
| | LSW | $\textbf{-0.08} \hspace{0.2cm} \pm \hspace{0.2cm} 0.05$ | $0.00 \hspace{0.1in} \pm \hspace{0.1in} 0.00$ | $\textbf{-0.04} \hspace{0.2cm} \pm \hspace{0.2cm} \textbf{0.44}$ | $-2.26 \hspace{0.1in} \pm \hspace{0.1in} 0.57$ | -2.26 ± 1.06 | -2.38 ± 0.72 |
| | ISOW | 0.04 ± 0.10 | $0.01 \hspace{0.1in} \pm \hspace{0.1in} 0.00$ | 0.12 ± 0.40 | $-2.70 \hspace{0.1 in} \pm \hspace{0.1 in} 0.43$ | -2.58 ± 0.99 | -2.53 ± 0.60 |
| | ENACW | 1.13 ± 0.94 | 0.14 ± 0.04 | 0.73 ± 0.66 | -2.25 ± 1.39 | -0.58 ± 2.31 | -0.25 ± 1.80 |
| Rockall | SPMW | $1.31 \hspace{.1in} \pm \hspace{.1in} 0.29$ | $0.10 \hspace{0.1in} \pm \hspace{0.1in} 0.02$ | $0.47 \hspace{0.2cm} \pm \hspace{0.2cm} 0.22$ | -3.84 ± 1.23 | $-2.43 \hspace{0.1 in} \pm \hspace{0.1 in} 1.90$ | -1.96 ± 1.28 |
| | LSW | $0.30 \hspace{0.1in} \pm \hspace{0.1in} 0.24$ | $0.01 \hspace{0.1in} \pm \hspace{0.1in} 0.01$ | -0.14 ± 0.37 | -0.94 ± 0.86 | $-1.36 \hspace{0.1in} \pm \hspace{0.1in} 0.97$ | -0.76 ± 0.96 |

Table 4

| Basin | Layer | | $\frac{\partial \Omega}{\partial T} \frac{\partial T}{dt}$ | $\frac{\partial \Omega}{\partial S} \frac{\partial S}{dt}$ | $\frac{\partial \Omega}{\partial A_T} \frac{\partial NA_T}{dt}$ | $\frac{\partial \Omega}{\partial C_T} \frac{\partial N C_T}{dt}$ | $\frac{d\Omega}{dt} (obs)$ | $\frac{d\Omega}{dt}$ (calcu | ılated) |
|----------|-----------|-----------|--|--|---|--|----------------------------|-----------------------------|---------|
| | CDMW | Calcite | -0.57 ± 0.24 | $-0.43 ~\pm~ 0.18$ | 1.68 ± 2.37 | -13.35 ± 3.14 | -11.03 ± 5.57 | -12.67 | ± 3.94 |
| | SPMW | Aragonite | $\textbf{-0.49} ~\pm~ 0.20$ | $\textbf{-0.29} \hspace{0.2cm} \pm \hspace{0.2cm} \textbf{0.12}$ | $1.07 \hspace{0.2cm} \pm \hspace{0.2cm} 1.50$ | -8.47 ± 1.99 | -7.17 ± 3.46 | -8.17 | ± 2.50 |
| | uLSW | Calcite | -0.17 ± 0.13 | $\textbf{-0.12} ~\pm~ 0.05$ | $\textbf{-0.46}~\pm~1.82$ | -12.61 ± 2.24 | -8.28 ± 5.16 | -13.36 | ± 2.89 |
| | uls w | Aragonite | -0.13 ± 0.10 | $\textbf{-0.08} ~\pm~ 0.03$ | $\textbf{-0.29} ~\pm~ 1.16$ | -8.03 ± 1.43 | -5.55 ± 3.21 | -8.53 | ± 1.84 |
| Irminger | LSW | Calcite | $\textbf{-0.15} ~\pm~ \textbf{0.11}$ | $\textbf{-0.09} ~\pm~ 0.05$ | $\textbf{-0.17} ~\pm~ \textbf{1.55}$ | -10.42 ± 2.27 | -13.54 ± 2.88 | -10.83 | ± 2.75 |
| miniger | | Aragonite | $\textbf{-0.11} ~\pm~ \textbf{0.08}$ | $-0.06~\pm~0.03$ | $\textbf{-0.11} ~\pm~ \textbf{0.99}$ | -6.69 ± 1.45 | -8.65 ± 1.83 | -6.97 | ± 1.76 |
| | ISOW | Calcite | $-0.04~\pm~0.05$ | $0.00 ~\pm~ 0.01$ | $-0.44 ~\pm~ 1.03$ | -7.48 ± 2.75 | -10.35 ± 3.23 | -7.96 | ± 2.94 |
| | 150 W | Aragonite | -0.02 ± 0.04 | $0.00 ~\pm~ 0.01$ | $\textbf{-0.29} ~\pm~ \textbf{0.67}$ | -4.84 ± 1.78 | -6.66 ± 2.04 | -5.15 | ± 1.90 |
| | DSOW | Calcite | -0.13 ± 0.12 | $0.03 \hspace{0.2cm} \pm \hspace{0.2cm} 0.02$ | $-1.78~\pm~0.52$ | -9.23 ± 1.68 | -4.30 ± 2.76 | -11.11 | ± 1.77 |
| | DSOW | Aragonite | -0.09 ± 0.09 | $0.02 \ \pm \ 0.01$ | $-1.16~\pm~0.34$ | -6.01 ± 1.10 | -3.02 ± 1.68 | -7.24 | ± 1.15 |
| Iceland | SPMW | Calcite | $\textbf{-0.88} ~\pm~ 0.26$ | $\textbf{-0.86} ~\pm~ 0.16$ | $3.16 ~\pm~ 1.00$ | -19.59 ± 8.35 | -15.77 ± 10.40 | -18.17 | ± 8.42 |
| | | Aragonite | $\textbf{-0.72} ~\pm~ 0.22$ | $\textbf{-0.58} ~\pm~ 0.10$ | $2.02 \ \pm \ 0.64$ | -12.48 ± 5.32 | -10.37 ± 6.55 | -11.77 | ± 5.37 |
| | uLSW | Calcite | $\textbf{-0.17} ~\pm~ \textbf{0.07}$ | $-0.09 ~\pm~ 0.03$ | $-1.02 \hspace{0.1in} \pm \hspace{0.1in} 1.89$ | -7.98 ± 2.52 | -9.18 ± 5.11 | -9.26 | ± 3.15 |
| | | Aragonite | $-0.12 ~\pm~ 0.05$ | $-0.06~\pm~0.02$ | $\textbf{-0.65} ~\pm~ \textbf{1.21}$ | -5.11 ± 1.61 | -5.92 ± 3.23 | -5.95 | ± 2.02 |
| | LSW | Calcite | $0.08 \hspace{0.2cm} \pm \hspace{0.2cm} 0.05$ | $0.02 \hspace{0.2cm} \pm \hspace{0.2cm} 0.01$ | $-0.15 ~\pm~ 1.70$ | -7.92 ± 2.00 | -7.53 ± 4.64 | -7.97 | ± 2.63 |
| | LSW | Aragonite | $0.06 ~\pm~ 0.03$ | $0.01 \hspace{0.2cm} \pm \hspace{0.2cm} 0.01$ | $\textbf{-0.09} \hspace{0.1in} \pm \hspace{0.1in} 1.09$ | -5.10 ± 1.29 | -4.83 ± 2.96 | -5.12 | ± 1.69 |
| | ISOW | Calcite | -0.04 ± 0.10 | $-0.03 ~\pm~ 0.02$ | $0.41 \hspace{0.2cm} \pm \hspace{0.2cm} 1.37$ | -8.38 ± 1.33 | -7.22 ± 4.34 | -8.05 | ± 1.91 |
| | 150W | Aragonite | -0.03 ± 0.07 | $-0.02 ~\pm~ 0.01$ | $0.27 \hspace{0.2cm} \pm \hspace{0.2cm} 0.89$ | -5.43 ± 0.86 | -4.72 ± 2.76 | -5.22 | ± 1.24 |
| | ENACW | Calcite | $\textbf{-0.82} ~\pm~ \textbf{0.69}$ | $-1.50~\pm~0.38$ | $5.16 ~\pm~ 4.63$ | -14.21 ± 8.78 | -11.60 ± 12.67 | -11.37 | ± 9.95 |
| | ENACW | Aragonite | $\textbf{-0.79} ~\pm~ 0.66$ | $-1.00~\pm~0.25$ | $3.29 \ \pm \ 2.95$ | -9.06 ± 5.60 | -7.66 ± 7.96 | -7.57 | ± 6.37 |
| Rockall | SPMW | Calcite | $\textbf{-1.15}~\pm~0.26$ | $\textbf{-0.82} \hspace{0.2cm} \pm \hspace{0.2cm} \textbf{0.18}$ | $2.44 \hspace{0.2cm} \pm \hspace{0.2cm} 1.15$ | -18.21 ± 5.83 | -20.57 ± 13.40 | -17.74 | ± 5.95 |
| NUCKall | ST 1VI VV | Aragonite | $\textbf{-0.93}~\pm~0.21$ | $\textbf{-0.55} ~\pm~ 0.12$ | $1.56 ~\pm~ 0.74$ | -11.66 ± 3.73 | -13.24 ± 8.47 | -11.58 | ± 3.81 |
| | ISW | Calcite | $\textbf{-0.28} ~\pm~ 0.22$ | $\textbf{-0.10} ~\pm~ \textbf{0.08}$ | $\textbf{-0.58} ~\pm~ 1.57$ | -3.62 ± 3.30 | -7.88 ± 4.41 | -4.59 | ± 3.66 |
| | LSW | Aragonite | $\textbf{-0.21} ~\pm~ 0.16$ | $\textbf{-0.07} ~\pm~ 0.05$ | $\textbf{-0.37} ~\pm~ 1.01$ | -2.33 ± 2.12 | -4.97 ± 2.82 | -2.97 | ± 2.35 |

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