

1 **Expanding seawater carbon dioxide and methane measuring capabilities with a Seaglider**

2

3 Claudine Hauri^{1*}, Brita Irving¹, Dan Hayes², Ehsan Abdi^{3,4}, Jöran Kemme⁵, Nadja Kinski⁵, and

4 Andrew M. P. McDonnell^{6,7}

5

6 ¹International Arctic Research Center, University of Alaska Fairbanks, Fairbanks, AK 99775,

7 USA

8 ²Advanced Offshore Operations, Inc., Houston, TX 77004, USA

9 ³Cyprus Subsea Consulting and Services, Lakatamia 2326, Cyprus

10 ⁴now at Akvaplan-Niva, 9296 Tromsø, Norway

11 ⁵-4H-JENA engineering GmbH, 07745 Jena, Germany

12 ⁶College of Fisheries and Ocean Science, University of Alaska Fairbanks, Fairbanks, AK 99775,

13 USA

14 ⁷now at Alaska Renewables, Fairbanks, AK 99709, USA

15

16 Corresponding author: Claudine Hauri (chauri@alaska.edu)

17 **Abstract**

18 Warming, ocean acidification, and deoxygenation are increasingly putting pressure on
19 marine ecosystems. At the same time, thawing permafrost and decomposing hydrates in Arctic
20 shelf seas may release large amounts of methane (CH₄) into the water column, which could
21 accelerate local ocean acidification and contribute to climate change. The key parameters to
22 observing and understanding these complex processes and feedback mechanisms are vastly
23 undersampled throughout the oceans. We developed carbon dioxide (CO₂) and CH₄ gliders,
24 including standard operational procedures with the goal that CO₂ and CH₄ measurements become
25 more common for glider operations. The Seagliders with integrated Contros HydroC CO₂ or CH₄
26 sensors also include conductivity, temperature, depth, oxygen, chlorophyll-a, backscatter, and
27 fluorescent dissolved organic matter sensors. Communication via satellite allows for near-real
28 time data transmission, sensor adjustments, and adaptive sampling. Several sea trials with the
29 CO₂ Seaglider in the Gulf of Alaska and data evaluation with discrete water and underway
30 samples suggest near ‘weather quality’ CO₂ data as defined by the Global Ocean Acidification
31 Network. A winter mission in Resurrection Bay, Alaska provided first insights into the water
32 column inorganic carbon dynamics during this otherwise undersampled season. The CH₄
33 Seaglider passed its flight trials in Resurrection Bay, yet needs to be tested during a field mission
34 in an area with CH₄ concentrations beyond background noise. Both sensing systems are available
35 to the science community through the industry partners (Advanced Offshore Operations and -4H-
36 JENA engineering GmbH) of this project.

37

38 **1. Introduction**

39 Understanding the distribution and dynamics of carbon dioxide (CO₂) and methane (CH₄)
40 in the ocean is crucial for predicting and mitigating climate change and ocean acidification
41 impacts. Within the ocean, CO₂ levels (measured as the partial pressure of CO₂, *p*CO₂ and/or
42 fugacity of CO₂) are spatially and temporally variable as they are influenced by a myriad of
43 highly dynamic physical, chemical, and biological processes. On top of this natural variability,
44 the ocean has absorbed about one third of the CO₂ emitted by humans since the industrial
45 revolution (Sabine et al., 2004; Gruber et al., 2019). In doing so, it has played an important role
46 in mitigating climate change (Sabine and Tanhua, 2010). However, both the oceanic uptake of
47 anthropogenic CO₂ and climate change are altering the distribution of oceanic CO₂ and are
48 causing ocean acidification (Doney et al., 2009; Qi et al., 2022; Woosley and Millero, 2020). At
49 the same time, the oceans are warming and losing oxygen (Johnson and Lyman, 2020; Breitburg
50 et al., 2018), increasing the stress on marine ecosystems. As these long-term changes unfold,
51 marine heat waves, and high acidity or low oxygen extreme events will last longer, become more
52 intense, and happen more often and at the same time (Laufkötter et al., 2020; Gruber et al., 2021;
53 Hauri et al., 2024). Negative effects on certain organisms are even stronger if exposed to a
54 combination of different stressors (Breitberg et al., 2015; Kroeker et al., 2017).

55 Over the coming 100 years, CH₄ possesses a global warming potential approximately 28
56 times greater than that of CO₂ (IPCC AR5; Myhre et al., 2013). Sediments along the seafloor at
57 continental margins contain large amounts of CH₄, with about ten times as much carbon as the
58 atmosphere (Kessler, 2014). CH₄ is biologically produced in anoxic sediments and the surface
59 mixed layer or released from geological sources like hydrocarbon seeps and degrading methane
60 hydrate deposits (Barnes and Goldberg; Du et al, Skarke 2014). This powerful greenhouse gas is
61 emitted to the atmosphere through bubbling (ebullition) or diffusive gas transfer (Reeburgh,

62 2007; McGinnis et al., 2006), which is limited by rapid oxidation to CO₂ during transport
63 through the water column (Leonte et al., 2017). CH₄ occurs generally at low levels (background
64 concentrations) throughout oceans, unless close to a source. Positive feedback mechanisms, like
65 warming induced CH₄ seepage from destabilizing hydrates and thawing subsea permafrost, may
66 further accelerate ocean acidification and climate change (Garcia-Tigreros et al., 2021; Sparrow
67 et al., 2018; Shakhova et al., 2010; Rees et al., 2022).

68 To effectively observe and understand the complex processes and feedback mechanisms
69 regulating Earth's systems, certain key parameters, defined by the Global Ocean Observing
70 System as essential ocean variables, must be measured accurately. However, these variables are
71 often vastly undersampled across time and space due to traditional sampling methods, which rely
72 mainly on discrete water sample collections from dedicated research cruises, underway
73 measurements from transiting vessels, or time series measurements from in situ sensors on fixed
74 moorings. Although biogeochemical sensors deployed on autonomous platforms like moorings
75 and Argo floats have become more prevalent, challenges such as high power requirements,
76 sensor size, and data quality hinder their widespread use on underwater gliders. Autonomous,
77 spatially resolved surface measurements of *p*CO₂ and pH are commonly collected using wave
78 gliders and sail drones (Chavez et al., 2018; Nickford et al., 2022; Manley and Willcox, 2010).
79 The state-of-the-art biogeochemical (BGC) Argo floats measure variables like pH, O₂, NO₃,
80 chlorophyll-a, suspended particles, and downwelling irradiance in subsurface waters (Claustre et
81 al., 2020). These floats can last several years at low sampling resolutions, such as a 2000-meter
82 depth profile every ten days, or they can be programmed for high-resolution and shallow
83 sampling. They can even sample beneath seasonal sea ice (Briggs et al., 2018). Despite their

84 capabilities, their trajectory is hard to control, and they are usually not recovered after their
85 mission, which prevents sensor calibration and post-mission corrections.

86 Ocean gliders autonomously collect water column data along planned waypoints, which
87 allows for controlled exploration and adaptive sampling. To date, pH is the only carbon system
88 parameter that has been successfully integrated into ocean gliders (Hemming et al., 2017; Saba et
89 al., 2019; Possenti et al., 2021; Takeshita et al., 2021). The most promising results came from
90 ISFET based pH sensors (Saba et al., 2019; Wright-Fairbanks et al., 2020; Takeshita et al.,
91 2021). However, ISFET-based pH sensors require significant conditioning periods before
92 deployment, suffer from biofouling, require annual cleaning and calibration at the manufacturer,
93 and careful discrete sample collection at deployment and recovery to characterize and correct for
94 sensor drift (Thompson et al., 2021). There have been few attempts to integrate $p\text{CO}_2$ sensors
95 into gliders (Hemming et al., 2017; Hauri et al., 2018; von Oppeln-Bronikowski et al., 2021).
96 Hemming et al. (2017) did not publish the data because of low quality. Von Oppeln-Bronikowski
97 et al. (2021) integrated an Aanderaa CO_2 optode that measures $p\text{CO}_2$ by detecting the
98 luminescent quenching response from a CO_2 -sensitive membrane with a Slocum G2 glider but
99 suffered from instability, thermal-lag issues, variable conditioning periods (4 days to 1 month),
100 large offsets (> 1000 uatm), nonlinear temperature-dependent response time, and a high
101 dependence on prior foil calibration. Hauri et al. (2018) integrated the Pro Oceanus Mini Pro
102 CO_2 sensor with a Slocum G2. However, the Pro Oceanus Mini Pro CO_2 sensor used at the time
103 did not withstand the pressure changes imposed by glider missions. The Franatech METS CH_4
104 sensor has been integrated into Alseamar SeaExplorer and Teledyne Slocum gliders and
105 successfully used to generate concentration maps of a methane seep in a semi-quantitative way
106 (Meurer et al., 2021).

107 Here we integrated modified versions of the Contros HydroC CO₂ and CH₄ sensors with
108 a Seaglider® (registered trademark of the University of Washington). We discuss details of the
109 physical and software integration, present CO₂ and CH₄ data from tank experiments, evaluate the
110 quality of *p*CO₂ data collected during CO₂ Seaglider missions, and discuss highlights from
111 missions in Resurrection Bay, Alaska.

112

113 **2. Methods**

114 2.1 CO₂ Seaglider

115 We integrated a modified version (Seaglider (SG) HydroC CO₂) of the CONTROS
116 HydroC™ CO₂ sensor (-4H-JENA engineering GmbH, Kiel, Germany) with a Seaglider M1
117 (Figure 1 a and b). The Seaglider M1 was specifically designed for long endurance missions in
118 deep waters to 1000 m depth. The HydroC CO₂ sensor was outfitted with a semi-permeable
119 TOUGH membrane (Pinnau and Toy, 1996) that equilibrated dissolved CO₂ between the ambient
120 seawater and the headspace of the sensor, where the gas concentration was determined by
121 nondispersive infrared (NDIR) spectrometry.

122 Since the equilibration time (response time) of membrane-based sensors is affected by the
123 exchange of the water mass in front of the sensor head, we installed a Seabird Electronics (SBE)
124 5M pump next to the SG HydroC CO₂ sensor using tubing to transfer seawater from outside the
125 glider fairing to the membrane surface (Figure 1a). The response time was determined at the
126 manufacturer, verified in the field, and then used to correct for hysteresis during the post-
127 processing phase (see Section 2.7.2).

128 The form factor of the HydroC CO₂™ sensor and Seaglider were changed to achieve an
129 internal integration of the sensor with the Seaglider. The standard high-performance HydroC

130 CO₂TM sensor was changed from ø 89 x 380 mm to ø 136 x 294 mm by rearranging the gas-cycle
131 components and the control unit (Figure 1c). This new SG HydroC CO₂ sensor is available in
132 polyoxymethylene cladding rated to 300 m or a titanium housing rated to 1000 m to provide a
133 choice between a coastal mission and an offshore deeper mission. Use of the titanium housing
134 required a syntactic foam housing to compensate for the weight, whereas the polyoxymethylene
135 housing was integrated into the glider with simple brackets (Figure 2). Despite these adjustments
136 to the size of the sensor, to our knowledge, it is still the largest and heaviest sensor that has been
137 integrated with a Seaglider to date. The forward fairing of the Seaglider was extended by 40 cm
138 with a fiberglass cylindrical extension to create internal wet payload space for the sensor, pump,
139 and cables (Figure 1 a and b). The sensor was mounted with the membrane facing aft to ensure
140 that potential bubbles within the internal tubing of the sensor could escape the system during the
141 downcast of the first dive. In situ comparison of the orientation of the sensor and close
142 examination of *p*CO₂ and internal pressure data suggested the highest data quality was achieved
143 with this mounting design.

144 One of the advantages of using ocean gliders for ocean observing is the ability for real-
145 time communication of data and commands between the pilot and the glider. To take advantage
146 of this, modifications were needed to allow two-way communication between the Seaglider
147 firmware and the HydroC firmware. The Seaglider firmware has a feature to allow easy
148 integration of “logging devices,” which provides a way to build commands for the pilot on land
149 to switch the sensor on and off and change sampling strategy during the mission (on/off below or
150 above certain depth) when it comes to the surface for a communication session. The Seaglider
151 firmware can also automatically set the clock of the sensor on request at every surfacing and
152 send small samples of the data stream via Iridium along with the standard sensor data. This

153 required the writing and testing of a driver file (CNF file). However, to take full advantage of the
154 ability of the HydroC, a more-advanced electronic integration was carried out using Smart
155 Interoperable Real-time Maritime Assembly (SIRMA™, registered trademark of Cyprus Subsea
156 Consulting and Services, C.S.C.S., Ltd.). This small programmable electronic circuit contained
157 hardware elements to adapt the sensor power and communication requirements to those available
158 on the host platform. It also allowed for separate storage and processing capabilities to
159 supplement the main host processor that controls the flight, sampling, and telecommunications of
160 the host. Most importantly here, it was programmed to relay pilot commands to the SG HydroC
161 CO₂ for the built-in “zero” function, which isolated the internal gas circuit until there was no
162 CO₂ present, measured the concentration signal, and assigned a zero value. Then the gas circuit
163 was exposed to the headspace behind the diffusion membrane for in situ sampling. SIRMA was
164 also programmed to extract raw data from the HydroC and calculate the bin average of some of
165 the output fields, which were useful for real-time mission adaptation and confirmation of sensor
166 operation. Three levels of output were allowed, depending on how much surfacing time could be
167 tolerated before continuing the mission (Baud rate for Iridium is very low, on the order of 4800
168 bps). More detailed information can be found in the CO₂ Seaglider SOP (Irving et al., 2024).

169 In addition to the HydroC CO₂ sensor, the CO₂ Seaglider carried an Aanderaa 4831F
170 optode, a compact optical oxygen sensor, which works on the principle of luminescence
171 quenching by oxygen with a precision of 0.1 μM and an absolute accuracy of +/- 2 μM after
172 multipoint calibration. The 4831F was equipped with a fast response sensing foil with a well-
173 characterized response time of 8 seconds. The Aanderaa optode measured absolute oxygen
174 concentration and percentage saturation. It is the most widely used on ocean gliders and has been
175 integrated into both Slocum and Seagliders (OceanGliders Oxygen SOP, 2024; Bittig et al.,

176 2018). The OceanGliders community has developed a Standard Operating Procedure (SOP) that
177 details everything from mounting, calibration, available sensors, piloting tips, and response time
178 correction, to post-processing (OceanGliders Oxygen SOP, 2024). The CO₂ Seaglider was also
179 outfitted with an SBE CT sail and Wetlabs Ecopuck measuring chlorophyll fluorescence at 695
180 nm.

181

182 **2.2 CH₄ Seaglider**

183 We also integrated a modified version of the CONTROS HydroC CH₄ sensor (-4H-JENA
184 engineering GmbH, Kiel, Germany) with the Seaglider. The manufacturer's published
185 uncertainty of the HydroC CH₄ sensor is 2 µatm or ± 3 %, whichever is greater . The SG HydroC
186 CH₄ sensor had the same form factor as the SG HydroC CO₂ sensor. However, it was 0.5 kg
187 heavier due to its Tunable Diode Laser Absorption Spectroscopy (TDLAS) component, so the
188 SG HydroC CH₄ had to be integrated with changes to the glider's ballast.

189

190 **2.3 Spring and winter CO₂ Seaglider missions**

191 Both versions of the CO₂ Seaglider (rated to 300 m versus 1000 m) were tested in
192 separate missions (Figure 3, Table S1) in spring (53 dives, 4 – 7 May 2022, Figure 4) and winter
193 (310 dives, 8 – 21 February 2023, Figure 5). The 300 m version with integrated
194 polyoxymethylene housing was tested during the 4 – 7 May 2022 mission. The glider followed
195 along a transect within Resurrection Bay. CTD casts near the glider path allowed for in-depth
196 evaluation of the data quality. The 1000 m depth-rated CO₂ Seaglider with integrated titanium
197 housing was tested in February 2023. Estimated energy consumption during the CO₂ Seaglider
198 missions was 19 out of 135 Ah and 75 out of 120 Ah for the 24 V which powered the SG

199 HydroC CO₂ sensor battery for the spring and winter missions, respectively. Before the February
200 mission, the on board modem was replaced with a newer model, with different input voltage
201 requirements, which were probably not met as the mission evolved. As a result, the glider could
202 not communicate and was lost. While this was an unfortunate mistake, the loss of the glider had
203 nothing to do with the HydroC CO₂ integration.

204

205 **2.4 Tank experiments**

206 Shortly before the May 2022 glider mission, the glider was kept in a flow-through tank at
207 the Alutiiq Pride Marine Institute for roughly 12 h for cross-calibration purposes. The flow-
208 through tank was fed with water from about 75 m depth and 91 m from the laboratory into
209 Resurrection Bay, near a freshwater source. During the tank experiment, SG HydroC CO₂T-
210 0718-001 (Figure 6b, blue line) was integrated into the Seaglider, and SG HydroC CO₂T-0422-
211 001 (Figure 6b, black line) and the SG HydroC CH₄ (Figure 6c) sensors were secured next to the
212 Seaglider. The water was kept in motion with a circulation pump. Triplicate discrete water
213 samples for dissolved inorganic carbon, pH, and CH₄ analysis were taken every four hours
214 (Table 1).

215

216 **2.5 Rosette package**

217 One of the SG HydroC CO₂ sensors (CO₂T-0422-001) was installed on an SBE-55 frame
218 ECO water sampler with six 4-liter sample bottles (Seabird Scientific) during the May 2022 trials
219 (Tables 2 and 3, Figure 7 and S1). The SBE-55 and SG HydroC CO₂ were powered by an SBE-
220 33 carousel deck unit. The SG HydroC CO₂ interfered with the communication stream and
221 thereby prevented real-time data acquisition and control of the SBE-55, however data were

222 internally logged. The depth of the rosette package was monitored directly on the winch and the
223 timing of firing of the sample bottles, after an approximate 15-minute hovering period (to allow
224 for equilibration), was programmed in advance based on time intervals. On May 3 (Table 2,
225 Figure 7) only samples from the upper 20 m of the water column were usable due to issues with
226 manually measuring the depths and the sample collection. On May 7 (Table 3, Figure S1) two
227 bottles that were intended to be fired while the rosette was stationary at depth, were instead fired
228 while the rosette was in motion.

229

230 **2.6 Discrete water samples**

231 2.6.1 Inorganic carbon chemistry

232 Discrete seawater samples were collected for sensor validation in two different cases in
233 May of 2022. Firstly, samples were taken alongside two SG HydroC CO₂ sensors during a tank
234 experiment at the Alutiiq Pride Marine Institute (Figure 6b, Table 1), from adjacent sample
235 bottles (Figure 1d). Secondly, samples were taken from bottles during a CTD cast within 1 km
236 and 4 hours of the HydroC measuring *p*CO₂ on the glider while conducting dives (Section 3.2).

237 Inorganic carbon sampling in the Gulf of Alaska's glaciated coastal regions required
238 methodological variations from open-ocean best practices to ensure that suspended mineral
239 particles do not compromise the instrumentation and/or bias measurements between sample
240 collection and analysis (Sejr et al., 2011). Given this, the discrete seawater samples were filtered
241 (replaceable 0.45 µm filter in a 47 mm polycarbonate in-line filter) with a peristaltic pump
242 straight from the Niskin bottles (see Bockmon and Dickson (2014) for detailed method), or tank,
243 into pre-cleaned 500 mL borosilicate bottles, and poisoned with 200 µL mercuric chloride
244 (HgCl₂) (Dickson et al., 2007). Samples were transported and stored at room temperature before

245 analysis. Samples were opened immediately (< 10 minutes) before concurrent analyses of pH
246 and DIC to limit gas exchange with ambient lab conditions. Samples were analyzed for DIC
247 using an Apollo SciTech, LLC Dissolved Inorganic Carbon Analyzer model AS-C6. All species
248 of dissolved inorganic carbon in a sample were converted to CO₂ by the addition of a strong acid.
249 The CO₂ gas was then purged from the sample through a drying system. The concentration of
250 CO₂ gas was measured using a non-dispersive infrared gas analyzer, the LI-7000 CO₂/H₂O
251 Analyzer. This method required Certified Reference Material (CRM, Batch #198 from A.
252 Dickson's Certified Reference Materials Laboratory) to create a three-point calibration line. The
253 calibration line was used to quantify the total amount of CO₂ in the sample as the integrated area
254 under the concentration-time curve. Apollo SciTech recommendations to improve analytical
255 accuracy were followed and included bubbling of CO₂ off the acid daily, allowing the analyzer to
256 warm up for at least 2 hours before measurements begin, measuring a set of standards at the
257 beginning and end of each day and every 9 samples, using Ultra High Purity (UHP) N₂ gas, and
258 filtering the N₂ gas with a PTFE filter, CO₂ scrubber (Ascarite II) and H₂O scrubber
259 (Mg(ClO₄)₂).

260 Samples were analyzed spectrophotometrically for pH with a CONTROS HydroFIA pH
261 (Aßmann et al., 2011) operating in discrete measurement mode using unpurified m-Cresol Purple
262 (mCP) as the indicator dye (Clayton and Byrne, 1993). Sample temperature was stabilized
263 at 25.00 +/- 0.01 °C during measurements using Peltier elements and 5 repetitive measurements
264 were taken for each sample. At the beginning of each day, the HydroFIA pH underwent a
265 conditioning period using seawater with similar properties until values stabilized. CRMs (known
266 TA and DIC concentration, Batch #198 from A. Dickson's Certified Reference Materials
267 Laboratory) were measured at the beginning and end of the day, as well as every 9 samples.

268 All data processing and analyses were done using an in-house MATLAB routine. In situ
269 pH and $p\text{CO}_2^{\text{disc}}$ were calculated from input pair pH_{lab} and DIC using CO2SYSv3 (Sharp et al.,
270 2023) with dissociation constants for carbonic acid of Sulpis et al. (2020), bisulfate of (Dickson,
271 1990), hydrofluoric acid of Perez and Fraga, (1987), and the boron-to-chlorinity ratio of (Lee et
272 al., 2010). Sulpis et al. (2020) found that the carbonic acid dissociation constants of Lueker et al.
273 (2000) may underestimate $p\text{CO}_2$ in cold regions (below $\sim 8^\circ\text{C}$) and, therefore, overestimate pH
274 and CO_3^{2-} . Differences between discrete $p\text{CO}_2$ calculated with the carbonic acid dissociation
275 constants by Lueker et al. (2000) (the standard in synthesis data products (e.g., Jiang et al., 2021,
276 Lauvset et al., 2022, Metzl et al., 2024) and the HydroC $p\text{CO}_2$ from the tank experiment were
277 found to be on average $4.6 \mu\text{atm}$ (1.6 %) and $4.2 \mu\text{atm}$ (0.7 %) greater for SN0422 and SN0718,
278 respectively, when compared with discrete $p\text{CO}_2$ based on carbonic acid dissociation constants
279 by Sulpis et al. (2020).

280 Discrete $p\text{CO}_2$ uncertainty (u_c) was calculated as the combined standard uncertainty from
281 *errors.m* (Orr et al., 2018) that propagates input uncertainties plus errors in the dissociation
282 constants. Input uncertainties for pH_{lab} and DIC were the standard uncertainties, defined as the
283 square root of the sum of the squared random uncertainty component plus the squared systematic
284 uncertainty components. For pH_{lab} the random uncertainty was the sample precision, or standard
285 deviation of the measurements. For DIC, the random uncertainty was the propagated error
286 calculated with the first-order Taylor series expansion (Equation 1, Orr et al. (2018)) and
287 assuming the correlation term was zero for the conversion of molarity ($\mu\text{moles l}^{-1}$) to molality
288 ($\mu\text{moles kg}^{-1}$). Systematic uncertainty components were the uncertainty in the CRM used for
289 instrument offset and drift correction, and the published instrument accuracy, or if available, the
290 daily instrument accuracy as defined below. Daily instrument accuracy was defined as the

291 maximum difference between the known CRM concentration, and the measured CRM
292 concentration after data were corrected for instrument drift and offset of all available CRM's not
293 used in the instrument drift and offset calculation. CRM pH_{lab} "known" values were calculated
294 using CO2SYSv3 (Sharp et al., 2023) with inputs pH and DIC. Nutrient concentrations (SiO_4^{-2} ,
295 PO_4^{-3}) were assumed to be negligible in the CO2SYS calculations (e.g. DeGrandpre et al., 2019;
296 Vergara-Jara et al., 2019; Islam et al., 2017).

297

298 2.6.2 Methane

299 Two sets of triplicate pCH_4 discrete water samples were collected during the tank
300 experiment for an initial evaluation of the SG HydroC CH_4 sensor (Table 1, Figure 6c). Seawater
301 was filtered from the tank into 250 mL vials. The vials were closed with a rubber stopper, topped
302 with an aluminum cap, and closed with a crimp immediately. A dry and clean syringe was
303 flushed with 10 mL of N_2 gas twice. The third fill was kept, and the syringe valve was closed. N_2
304 was then injected into the headspace while simultaneously pulling 10 mL seawater out of the vial
305 using a second syringe. 50 μ L mercuric chloride ($HgCl_2$) were added to the vial, which was then
306 shaken for about a minute and put into a fridge at 4°C for over 12 h to equilibrate the headspace.
307 The samples were then sent to the Kessler analytical laboratory at the University of Rochester for
308 analysis.

309

310 **2.7 Data post-processing**

311 *2.7.1 pCO_2 post-processing*

312 SG HydroC CO_2 data were post-processed using Jupyter Notebook scripts developed by -4H-
313 JENA engineering GmbH at the original resolution (2 seconds). SG HydroC CO_2 (SG HydroC

314 CO2T-0422-001) data from the tank experiment (Table 1, Figure 6) and rosette mounted CTD
315 casts (Table 2 and 3, Figure 7 and S1) were post-processed to correct for baseline drift (change in
316 the zero signal reference) and span drift (changes in the sensor's concentration dependent
317 characteristics) using pre- and post-calibration coefficients interpolated over the
318 deployment ([Fietzek et al., 2014](#)). For the May 2022 Seaglider integrated SG HydroC
319 CO₂ sensor (SG HydroC CO2T-0718-001, Table 3, Figures 4 and 8), data were post-processed
320 with pre-calibration coefficients only (no span drift correction) because the sensor was damaged
321 during the return shipment for post-calibration. Differences between sensors remained low
322 despite the difference in processing, with a mean difference during the tank experiment of $2.1 \pm$
323 $1.0 \mu\text{atm}$ (0.9%) and median difference of $2.0 \pm 1.0 \mu\text{atm}$ (0.9%) (Table 1, Figure 6b).

324 The $p\text{CO}_2$ data from February 2023 was collected with a sensor that was factory calibrated two
325 weeks prior to deployment (SG HydroC CO2T-0422-001) but were not post-processed because a
326 required parameter (p_NDIR) was not relayed in real-time and the glider was lost. Lack of post-
327 calibration most likely had no negative effect on the quality of data since the HydroC was only
328 collecting data for ~4 days during the spring mission and ~2 days during the winter mission.

329 HydroC $p\text{CO}_2$ and $p\text{CO}_2^{\text{RTC}}$ data at the original resolution (2 s) and RTC resolution (8 s)
330 were linearly interpolated onto the Seaglider timestamp and 1-meter binned data were calculated
331 by first averaging 1 meter (+/- 0.5 m) upcast and downcast data independently, linearly
332 interpolating over gaps, then averaging the interpolated 1-meter binned upcast and downcast
333 together.

334

335 *2.7.2 Response time correction*

336 The ability to determine the in situ response time (τ_{63} of the HydroC, which took into
337 account membrane characteristics and the rate of water exchange over the membrane, i.e.
338 pump characteristics) of the sensor made correction for hysteresis through data post processing
339 possible. This is critical for a sensor operating on profiling platforms, especially in the Gulf of
340 Alaska, where strong environmental gradients were encountered. Fiedler et al. (2013) used a
341 CONTROS HydroCTM CO₂ with a silicone, polydimethylsiloxane (PDMS) membrane and
342 reported a linear response time dependency on water temperature on the order of one second per
343 one °C. For this study, the SG HydroC CO₂ sensors were deployed with the new robust TOUGH
344 membrane, which had Teflon AF2400 as the active separation layer with a low temperature
345 dependence on the permeability coefficient (Pinnau and Toy, 1996). Response times determined
346 during calibration at -4H-JENA were used for response time correction (RTC) and found to be
347 106 seconds for the HydroC mounted on the rosette in May 2022 and 108 seconds when it was
348 integrated into the Seaglider in February 2023 (HydroC CO2T-0422-001). The response time of
349 the HydroC integrated into the Seaglider in May 2022 (HydroC CO2T-0718-001) was 109
350 seconds. Since field verification of the response time was recommended to ensure the highest
351 quality post-processed data product (because τ_{63} can be affected by the speed of water exchange
352 across the membrane due to pump speed, tube length, etc.), we verified the sensor response time
353 at deployment. After the glider was stationary for approximately 15 minutes, a zeroing interval
354 was performed with the HydroC CO₂. The response time was determined by reviewing the time
355 it took for the signal to recover to the ambient concentration. Our in situ response time tests were
356 suggested to be within 5 seconds of the response time found during calibration (not shown).
357 Before RTC was applied, HydroC CO₂ data were smoothed using a quadratic regression
358 (MATLAB's smoothdata.m function with the loess method) over a 2-minute window. This was

359 done to eliminate erroneous spikes in the RTC signal while retaining the original 2-second
360 resolution of the $p\text{CO}_2$ data. The RTC resolution of 8 seconds was determined with the L-curve
361 analysis included in the publicly available code from Dølven et al. (2022). The Dølven et al.
362 (2022) RTC method was used because it produced more realistic profiles than an RTC method
363 (Miloshevich et al., 2004, not shown) previously used for HydroC CO_2 correction from a
364 profiling float (Fielder et al. 2013). In addition, Dølven et al. (2022) developed their algorithm
365 with equilibrium-based sensors in mind and was proven with a sensor with a long response time
366 (HydroC CH_4 $\tau_{63} \cong 23$ minutes).

367

368 2.7.3 $p\text{CH}_4$ post-processing

369 SG HydroC $p\text{CH}_4$ data were response time corrected using a τ_{63} of 43 minutes (Dølven et
370 al., 2022; Figure 6c, blue line). Before RTC was applied, HydroC CH_4 data were smoothed using
371 a quadratic regression (MATLAB's smoothdata.m function with the Loess method) over a 2-
372 minute window to avoid erroneous spikes in the RTC data while retaining the original 2-second
373 resolution of the $p\text{CH}_4$ data. The RTC resolution of 30 seconds was determined with the L-curve
374 analysis included in the publicly available code from Dølven et al. (2022). Discrete CH_4 samples
375 were collected during the tank experiment (Table 1, Figure 6c red diamonds) and analyzed at
376 John Kessler's laboratory at the University of Rochester. Discrete CH_4 sample values were
377 converted from the concentration of dissolved gas in water (mol L^{-1}) to partial pressure ($p\text{CH}_4^{\text{disc}}$,
378 μatm) using the solubility coefficient following Sarmiento and Gruber (2006). $p\text{CH}_4^{\text{disc}}$
379 uncertainty (u ; Table 1, Figure 6c red error bars) was calculated as the square root of the sum of
380 the squared 1) mean of the standard deviations from each sample as returned from the lab and 2)
381 the standard deviation of the triplicates.

382

383 **3. Results**

384 **3.1 Glider flight**

385 The Seaglider was able to “fly” properly, allowing the desired undisturbed flow, despite
386 the large payload and major changes to the vehicle fairing. Example flight profiles with the
387 polyoxymethylene and Titanium integrated sensors are shown in Figures 9 and 10, respectively.
388 Pitch and vertical velocity are in the stable range, and roughly symmetric between downcast and
389 upcast, indicating a nearly balanced glider. Heading varies around the targeted value as the roll
390 adjusted to heading errors. It should be noted that this level of variability is typical of standard
391 Seagliders. Operating Seagliders in shallow water (< 200 m) is risky because of the likelihood of
392 meeting depth-averaged currents of the same order of magnitude as the vehicle speed. A typical
393 single dive cycle of downcast and upcast shows that the sensor data are free of noise that could
394 be expected if there were recirculated water from the glider meeting the sensors. The expected
395 endurance of the CO₂ Seaglider is around 18 days and 15 days for the CH₄ Seaglider with
396 constant sampling at full depth.

397

398 **3.2 CO₂ Seaglider data evaluation**

399 The quality of the CO₂ Seaglider data was thoroughly tested with discrete measurements
400 during a tank experiment, nearby CTD cast, and glider missions.

401

402 **3.2.1 Tank experiment**

403 Discrete water samples show good agreement with the SG HydroC CO₂ sensors (Figure
404 6b, Table 1). The values of discrete water samples represent the average of triplicate samples

405 (Figure 6c, red diamonds). Differences between the SG HydroC CO₂ sensors remained low, with
406 a mean difference during the tank experiment of $2.1 \pm 1.0 \mu\text{atm}$ (0.9 %) and median difference of
407 $2.0 \mu\text{atm}$ (0.9 %; Table 1). Percent differences (Eq. 1) between the SG HydroC CO₂ sensors and
408 discrete water samples collected in the tank were between -1.4 and 1.9 % (Table 1, Figure 6).

$$409 \quad \% \text{ difference} = \frac{pCO_2^{\text{HydroC}} - pCO_2^{\text{disc}}}{pCO_2^{\text{disc}}} * 100\% \quad (\text{Equation 1})$$

410

411 3.2.2 Profiling experiment

412 Rosette-based profiles with the SG HydroC CO₂ sensor in combination with discrete
413 water samples were used to test and evaluate the response correction algorithm by Dølven et al.
414 (2022). The rosette was lowered into the water and kept at different depths for about 20 minutes
415 at a time (Figure 7a and Figure S1a). Sample bottles were programmed to collect seawater
416 toward the end of each hovering period. pCO_2 measured with the HydroC ranged from $218 \mu\text{atm}$
417 at the surface to $411 \mu\text{atm}$ at 80 m depth on 3 May (Figure 7b) and $231 \mu\text{atm}$ at the surface to
418 $382 \mu\text{atm}$ at 77 m depth on 7 May (Figure S1). Differences between the rosette mounted SG
419 HydroC CO₂ sensor and discrete samples ranged from $-3.3 \mu\text{atm}$ (-1.4 %) to $8.2 \mu\text{atm}$ (3.5 %)
420 with a lowest percent difference of 0.6 % (Table 2) on 3 May and from $-5.7 \mu\text{atm}$ (-1.6 %) to
421 $12.1 \mu\text{atm}$ (3.9 %) with a lowest percent difference of 0.3 % (Table 3) on 7 May.

422

423 3.2.3 Data evaluation during CO₂ Seaglider mission

424 The quality of the pCO_2 data from the CO₂ Seaglider was further evaluated during a 4 – 7
425 May sea trial mission in spring 2022 in Resurrection Bay, Alaska (Figure 3).

426 Discrete water samples were taken in proximity (1 km and within 4 hours) of the downcast of
427 dive #51 (Table 3, Figures 4a and 8). The response time corrected CO₂ Seaglider data compares
428 well with the discrete water samples (Figure 8), overestimating the discrete water samples
429 between 8.3 μatm (2.7 %) and 12.0 μatm (5.1 %) (Table 3). The mean difference between the
430 rosette mounted and Seaglider integrated SG HydroC CO₂ sensors during the 7 May cast at the
431 time of discrete samples was 8.5 μatm +/- 8.9 μatm (3.7 %). The larger difference between SG
432 HydroC CO₂ sensors compared to the difference during the tank experiment (see Section 3.2.1)
433 is unsurprising, given the spatial and temporal distance between sensors (Table 3). Collecting
434 more discrete samples throughout the water column and in closer proximity (within 100 m,
435 Thompson et al., 2021) to the CO₂ Seaglider conducting dives would allow a more tightly
436 constrained uncertainty estimate for response time corrected *p*CO₂ data collected on a glider and
437 should be a priority for future researchers.

438

439 **3.3 CH₄ Seaglider data evaluation**

440 3.3.1 Tank experiment

441 The SG HydroC CH₄ was also evaluated during the tank experiment described in section
442 2.4 (Figure 6c). Percent differences (Eq. 1) between discrete *p*CH₄ (average of triplicate samples)
443 and *p*CH₄^{RTC} were 6.6 to 15.8 % (Table 1). During the experiment, there was a decrease in
444 salinity from 30.95 to 29.88 where *p*CO₂ decreased by 80 μatm. The corresponding *p*CH₄^{RTC}
445 signal decreased by 25.4 μatm from 32.3 to 6.9 μatm. Although the triplicate discrete *p*CH₄ water
446 samples were slightly lower than the sensor-measured *p*CH₄ values, they also reflected this step
447 change.

448

449 3.4 Winter and springtime $p\text{CO}_2$ in Resurrection Bay, Alaska

450 The surface-to-subsurface $p\text{CO}_2$ gradient is much larger in spring than in winter (Figure
451 11). During the 4 – 7 May mission, the average surface $p\text{CO}_{2,\text{Seaglider}}^{\text{RTC}}$ was $240.7 \pm 16.5 \mu\text{atm}$
452 (mean \pm standard deviation at 2 meters) with an average temperature of $5.8 \pm 0.4 \text{ }^\circ\text{C}$ (Figures
453 4 and 11). In February, surface $p\text{CO}_{2,\text{Seaglider}}^{\text{RTC}}$ was near atmospheric $p\text{CO}_2$ ($427.4 \pm 13.0 \mu\text{atm}$,
454 temperature $4.1 \pm 0.3 \text{ }^\circ\text{C}$) and about $180 \mu\text{atm}$ higher than in May (Figures 5 and 11). NOAA's
455 moored sensor located in Sunny Cove ($59.911 \text{ }^\circ\text{N}$, $-149.35 \text{ }^\circ\text{W}$), near the CO_2 Seaglider trial site,
456 measured an average sea surface $p\text{CO}_2$ of $240.7 \pm 10.4 \mu\text{atm}$ during the time of the May 2022
457 mission (Monacci et al., 2023), which compared remarkably well with the Seaglider based
458 measurements. A minimum of $140 \mu\text{atm}$ was measured in Sunny Cove in mid-April (3-day
459 average) (Figure 12, Monacci et al., 2023), suggesting that the peak of the spring bloom
460 happened three weeks before the May 2022 glider mission. Since we don't have salinity data
461 from the May CO_2 Seaglider mission (conductivity sensor failure), we cannot disentangle the
462 contributions of freshwater or primary production on the low surface $p\text{CO}_2$ values observed
463 (Figure 4). The moored sensor in Sunny Cove measured an average sea surface $p\text{CO}_2$ of 416.4
464 $\pm 4.2 \mu\text{atm}$ during the time of the February mission, straddling the atmospheric $p\text{CO}_2$ values
465 (Monacci et al., 2023, Figure 12). Subsurface $p\text{CO}_{2,\text{Seaglider}}^{\text{RTC}}$ at 180 m was on average $545.6 \pm$
466 $16.9 \mu\text{atm}$ during the February mission and $518.2 \pm 37.4 \mu\text{atm}$ during the May 2022 mission
467 (Figure 11a). $p\text{CO}_2$ was much lower in May than in February throughout the upper water column
468 ($< 120 \text{ m}$), whereas there was not much of a seasonal difference at deeper depth. Some of the
469 fine scale features apparent in the May $p\text{CO}_2$ and O_2 profiles are likely due to various levels of
470 photosynthetic activity (Figure 11). As the glider transitioned into the open Gulf of Alaska
471 during the February mission, water with $\text{O}_2 < 150 \mu\text{M}$ shoaled into the upper 150 m of the water

472 column (Figure 5). Unfortunately, the HydroC CO₂ sensor was turned off at that stage of the
473 mission to conserve battery.

474

475 **4. Discussion**

476 The newly developed CO₂ Seaglider is the first of its kind to autonomously collect high
477 quality *p*CO₂ data. The tank and rosette experiments and in situ data evaluation suggest that the
478 post-processed data from the CO₂ Seaglider generally fall near the relative uncertainty of 2.5%,
479 which is a threshold defined as the “quality sufficient to identify relative spatial patterns and
480 short-term variation” (“weather quality”, Newton et al., 2015). This is the highest quality of
481 *p*CO₂ data that has been measured with a subsurface autonomous vehicle to this date and
482 therefore an important step towards filling the subsurface carbonate system data gap. -4H-JENA
483 is reassessing their sensor calibration methodology and data post-processing algorithm to further
484 improve the HydroC’s data accuracy.

485 The newly developed CO₂ Seaglider is suitable for data collection in open ocean or
486 coastal environments with bottom depths deeper than 300 m. However, the coastal Gulf of
487 Alaska is a highly dynamic environment, with strong freshwater and wind influence, and rugged
488 shallow (often < 200 m) bottom topography. Strong currents (up to 0.50 m s⁻¹) made the piloting
489 of the glider extremely difficult throughout the project and confirmed that the Seaglider cannot
490 reliably reach desired waypoints in these conditions. The current version of the CO₂ Seaglider is
491 also not suitable for operating in the coastal Gulf of Alaska in summer and early fall, due to
492 strong seasonal salinity gradients in this freshwater influenced area. Another issue we faced was
493 the fact that the forward-looking altimeter could not detect the Seafloor as it should in its
494 position behind the HydroC CO₂. In areas with detailed topography maps this would not be an

495 issue, but in the coastal Gulf of Alaska reliable topography information is not readily available
496 yet. An obvious next step is to integrate the SG HydroC CO₂ sensor into a newer glider platform,
497 such as the Seaglider SGX or Teledyne Slocum G3 glider. The extended energy bay, larger
498 buoyancy range, and thruster should make the operation of the coastal Slocum G3 with HydroC
499 sensors relatively easy and would allow for autonomous high-resolution water column
500 measurements of $p\text{CO}_2$ and $p\text{CH}_4$ in dynamic coastal environments. The integration of a HydroC
501 on a Slocum glider will require a custom-made wet-payload bay due to the size of this sensor.
502 For open ocean or deeper coastal regions, the integration with the Seaglider SGX, with 60%
503 higher energy capacity, would be effective and nearly identical to the work already done here.
504 The SG HydroC CH₄ was successfully integrated into the Seaglider as part of this project.
505 While tank experiments showed promising results, short field tests of the CH₄ Seaglider in
506 shallow water revealed low and patchy methane concentrations near the detection limit (not
507 shown). The CH₄ Seaglider requires further testing in environments with strong $p\text{CH}_4$ gradients
508 during longer and deeper dives (to allow for equilibration) to assess the accuracy of its
509 response time-corrected data in the field. The sensor's slow response time likely limits the glider
510 to providing qualitative rather than quantitative results. However, due to the scarcity of oceanic
511 CH₄ observations, deploying a CH₄ glider can help identify the location of methane sources and
512 guide the placement of in situ observations to conduct a more quantitative assessment of CH₄
513 fluxes and dynamics.

514 Ocean gliders are part of the Intergovernmental Oceanographic Commission (IOC-
515 UNESCO) Global Ocean Observing System (GOOS) through the OceanGliders program
516 (<https://www.oceangliders.org/>). Like other elements of the GOOS coordinated by OceanOPs of
517 the Observation Coordination Group (floats, buoys, moorings, ships, and tide gauges),

518 OceanGliders contributes to “Ocean Observation for Climate, Ocean Health and Real Time
519 Services”. CO₂ gliders are perfectly suited to contribute data for understanding relevant
520 inorganic carbon processes in coastal shelf and boundary regions where mesoscale or sub-
521 mesoscale variability dominates. The current work can also serve as a first step to bring together
522 interested scientists and engineers to further develop and improve the capability of gliders to
523 measure high-quality data. OceanGliders supports this effort by promoting the formation of
524 volunteer international task teams, for which a task team could be requested for oceanographic
525 greenhouse gas research. By doing this, the visibility and availability of data will be improved as
526 well, since GOOS provides an interactive data platform for all its programs ([https://www.ocean-](https://www.ocean-ops.org/board)
527 [ops.org/board](https://www.ocean-ops.org/board)). An OceanGliders task team could also be linked with the GOOS-sponsored
528 Global Climate Observing System (GCOS: <https://gcos.wmo.int/en/home>) program through their
529 Ocean Observations Physics and Climate Panel (OOPC): ”a scientific expert advisory group
530 charged with making recommendations for a sustained global ocean observing system for
531 climate.”

532

533 **5. Concluding Thoughts**

534 Near real-time and high-resolution water column data that can be retrieved from gliders
535 outfitted with sensors measuring salinity, temperature, inorganic carbon system parameters,
536 oxygen, and $p\text{CH}_4$ are key to tackling a variety of today’s climate change-related issues. These
537 datasets will become instrumental in advancing biogeochemical model forecasting and early
538 warning systems for extreme heat, acidity, and oxygen compound events that affect coastal
539 subsistence communities, commercial fisheries, and mariculture. Furthermore, using
540 biogeochemical gliders to monitor the environment of tagged organisms (e.g. crabs, fish) would

541 provide insight into the organism's position and behavior relative to important environmental
542 drivers across susceptible ecosystems. Such biogeochemical glider data will help bridge in situ
543 chemical and biological measurements, and environmental change to impacts on biology, and
544 thereby fill an important research gap (Widdicomb et al., 2023). Potentially large natural and
545 anthropogenic sources of CH₄ may become contributors to climate change, and if oxidized, to
546 ocean acidification (Garcia-Tigreros et al., 2021; Sparrow et al., 2018; Shakhova et al., 2010;
547 Rees et al., 2022). These CH₄ sources need to be properly assessed and quantified, and if
548 characterized as anthropogenic origin, emitters must be held accountable (Goodman et al., 2022).
549 Once the combined HydroC CH₄/CO₂ is available it will provide a new tool to co-measure *p*CH₄
550 and *p*CO₂ and give valuable insight into these processes and feedback mechanisms. Other
551 advancing fields, such as marine Carbon Dioxide Removal (mCDR) and monitoring,
552 verification, and reporting (MRV) thereof will also need detailed knowledge of the distribution
553 of CO₂ in the water column (National Academies of Sciences, Engineering, and Medicine.
554 2022).

555 The CO₂ Seaglider has been extensively tested and is ready to be used in open ocean
556 environments. An important next step will be to integrate the HydroC CO₂ and CH₄ sensors into
557 a glider platform that reliably functions in shallow, and freshwater-affected coastal areas, such as
558 the Gulf of Alaska, to be able to fill the large spatial and temporal data gap in these highly
559 dynamic areas.

560

561 **Data availability**

562 The CO₂ Seaglider data is publicly available (Hauri et al., 2022; 2023). The HydroC-
563 specific SIRMA code and CNF file are available on Github (Cyprus-Subsea, 2024a and 2024b).

564 More detailed information on the HydroC – glider integration and operation can be found in the
565 CO₂ Seaglider Standard Operating Procedures (Irving et al., 2024).

566

567 **Author contributions**

568 C.H. and A.M. developed the research ideas and the proposal that funded this work. C.H.
569 led the fieldwork and writing of this manuscript. B.I. led the preparation for fieldwork and glider
570 data processing and analysis. D.H. led glider piloting for all trials. D.H. and E.A. assisted with
571 data processing, sensor programming, mechanical integration, glider ballasting, deployment, and
572 recovery. N.K. and J.K. provided technical support for the HydroC sensors. All authors
573 contributed to the writing of this manuscript.

574

575 **Competing interests**

576 Authors Hayes and Abdi are employed by AOOI and CSCS (respectively) and their
577 objective is to support the ocean research community by providing innovative, cutting-edge
578 observing technological solutions. These include autonomous platforms and related services in
579 unique configurations. Through the support of the National Science Foundation and the National
580 Oceanographic Partnership Program, AOOI was able to jointly develop the CO₂ and CH₄ gliders
581 and prove and improve the scientific utility of this approach. Authors Kinski and Kemme are
582 employed by -4H-JENA engineering GmbH, the manufacturer of the HydroC CO₂ and CH₄
583 sensors. The objective of -4H-JENA engineering GmbH is to provide best possible accuracy of
584 dissolved gas measurements on any platform and at any environmental condition. Intensive
585 collaboration with scientists is essential for the development of these products.

586

587 **Acknowledgments**

588 The Seaglider field trials took place in the traditional and contemporary hunting grounds
589 of the Sugpiaq People. We also acknowledge that our Fairbanks-based offices are located on the
590 Native lands of the Lower Tanana Dena. The Indigenous Peoples never surrendered lands or
591 resources to Russia or the United States. We acknowledge this not only because we are grateful
592 to the Indigenous communities who have been in deep connection with the land and water for
593 time immemorial, but also in recognition of the historical and ongoing legacy of colonialism. We
594 are committed to improving our scientific approaches and working towards co-production for a
595 better future for everyone.

596 We would like to thank Jack Triest for his technical support throughout the project. We
597 are also grateful to Brian Mullaly, Captain of the RV Nanuq, and Seward Marine Center staff,
598 especially Pete Shipton, Ed DeCastro, Jenny Grischuk, and Jenny Elhard for their assistance
599 during the field trials in Seward. We are also grateful for the support from the Alutiiq Pride
600 Marine Institute, Alaska Sealife Center, and the Autonomous Remote Technology Lab. Finally,
601 we would like to express our gratitude to John Kessler and Katherine Gregory for analyzing our
602 CH₄ discrete water samples, guiding us through the sampling process and discussing CH₄
603 Seaglider missions with us. We would also like to thank for the support of Cyprus Subsea
604 engineers Sergey Vekli, Loizos Groutas, and Jerald Reodica in mechanical and electronic sensor
605 integration and piloting, as well as assisting with Cyprus sea testing of the HydroC sensors and
606 CO₂ Seaglider.

607

608 **Financial support**

609 We would like to thank the National Oceanographic Partnership Program and the
610 National Science Foundation for the support of this project (OCE-1841948).

611 **References**

612 Aßmann, S., Frank, C., and Körtzinger, A.: Spectrophotometric high-precision seawater
613 pH determination for use in underway measuring systems, *Ocean Sci.*, 7, 597–
614 607, <https://doi.org/10.5194/os-7-597-2011>, 2011.

615
616 Barnes, R. O. and Goldberg, E. D. Methane production and consumption in anoxic marine
617 sediments. *Geology* 4, 297–300, 1976.

618
619 Bittig, H. C., Körtzinger, A., Neill, C., van Ooijen, E., Plant, J. N., Hahn, J., Johnson, K. S.,
620 Yang, B., and Emerson, S. R.: Oxygen Optode Sensors: Principle, Characterization, Calibration,
621 and Application in the Ocean, *Front. Mar. Sci.*, 4, <https://doi.org/10.3389/fmars.2017.00429>,
622 2018.

623
624 Bockmon, E. E. and Dickson, A. G.: A seawater filtration method suitable for total dissolved
625 inorganic carbon and pH analyses, *Limnology and Oceanography Methods*, 12(4), 191–195,
626 <https://doi.org/10.4319/lom.2014.12.191>, 2014.

627
628 Breitberg, D., Salisbury, J., Bernhard, J., Cai, W.-J., Dupont, S., Doney, S., Kroeker, K., Levin,
629 L., Long, W. C., Milke, L., Miller, S., Phelan, B., Passow, U., Seibel, B., Todgham, A., and
630 Tarrant, A.: And on Top of All That... Coping with Ocean Acidification in the Midst of Many
631 Stressors, *Oceanography*, 25, 48–61, <https://doi.org/10.5670/oceanog.2015.31>, 2015.

632
633 Breitburg, D., Levin, L. A., Oschlies, A., Grégoire, M., Chavez, F. P., Conley, D. J., Garçon, V.,

634 Gilbert, D., Gutiérrez, D., Isensee, K., Jacinto, G. S., Limburg, K. E., Montes, I., Naqvi, S. W.
635 A., Pitcher, G. C., Rabalais, N. N., Roman, M. R., Rose, K. A., Seibel, B. A., Telszewski, M.,
636 Yasuhara, M., and Zhang, J.: Declining oxygen in the global ocean and coastal waters, *Science*,
637 359, 46, <https://doi.org/10.1126/science.aam7240>, 2018.

638

639 Briggs, E. M., Martz, T. R., Talley, L. D., Mazloff, M. R., and Johnson, K. S.: Physical and
640 Biological Drivers of Biogeochemical Tracers Within the Seasonal Sea Ice Zone of the Southern
641 Ocean From Profiling Floats, *J. Geophys. Res. Oceans*, 123, 746–758,
642 <https://doi.org/10.1002/2017JC012846>, 2018.

643

644 Chavez, F. P., Sevadjan, J., Wahl, C., Friederich, J., and Friederich, G. E.: Measurements
645 of $p\text{CO}_2$ and pH from an autonomous surface vehicle in a coastal upwelling system, *Deep Sea*
646 *Res. Part II Top. Stud. Oceanogr.*, 151, 137–146, <https://doi.org/10.1016/j.dsr2.2017.01.001>,
647 2018.

648

649 Claustre, H., Johnson, K. S., and Takeshita, Y.: Observing the Global Ocean
650 with Biogeochemical-Argo, *Annu. Rev. Mar. Sci.*, 12, 23–48, [https://doi.org/10.1146/annurev-](https://doi.org/10.1146/annurev-marine-010419-010956)
651 [marine-010419-010956](https://doi.org/10.1146/annurev-marine-010419-010956), 2020.

652

653 Clayton, T. D. and Byrne, R. H.: Spectrophotometric seawater pH measurements: total hydrogen
654 ion concentration scale calibration of m-cresol purple and at-sea results, *Deep Sea Res. Part*
655 *Oceanogr. Res. Pap.*, 40(10), 2115–2129, 1993.

656

657 Cyprus-Subsea: Smart-Cable-HydroC, GitHub repository [code], [https://github.com/Cyprus-](https://github.com/Cyprus-Subsea/Smart-Cable-HydroC)
658 [Subsea/Smart-Cable-HydroC](https://github.com/Cyprus-Subsea/Smart-Cable-HydroC) (last access: 14 June 2024), 2024a.
659
660 Cyprus-Subsea: CO₂ and CH₄ CNF files, GitHub repository [code], [https://github.com/Cyprus-](https://github.com/Cyprus-Subsea/Smart-Cable-HydroC/tree/main/docs)
661 [Subsea/Smart-Cable-HydroC/tree/main/docs](https://github.com/Cyprus-Subsea/Smart-Cable-HydroC/tree/main/docs) (last access: 7 July 2024), 2024b.
662
663 DeGrandpre, M. D., Lai, C. Z., Timmermans, M. L., Krishfield, R. A., Proshutinsky, A.,
664 and Torres, D.: Inorganic Carbon and *p*CO₂ Variability During Ice Formation in the Beaufort
665 Gyre of the Canada Basin, *J. Geophys. Res. Oceans*, 124, 4017–4028,
666 <https://doi.org/10.1029/2019JC015109>, 2019.
667
668 Dickson, A. G.: Thermodynamics of the dissociation of boric acid in synthetic seawater
669 from 273.15 to 318.15 K, *Deep Sea Res. Part Oceanogr. Res. Pap.*, 37, 755–
670 766, [https://doi.org/10.1016/0198-0149\(90\)90004-F](https://doi.org/10.1016/0198-0149(90)90004-F), 1990.
671
672 Dickson, A. G., Sabine, C. L., and Christian J.R.: Guide to Best Practices for Ocean
673 CO₂ Measurements, *PICES Spec. Publ.* 3, 191, 2007.
674
675 Dølven, K. O., Vierinen, J., Grilli, R., Triest, J., and Ferré, B.: Response time correction of slow-
676 response sensor data by deconvolution of the growth-law equation, *Geosci. Instrum. Methods*
677 *Data Syst.*, 11, 293–306, <https://doi.org/10.5194/gi-11-293-2022>, 2022.
678
679 Doney, S. C., Fabry, V. J., Feely, R. A., and Kleypas, J. A.: Ocean Acidification: The Other

680 CO₂ Problem, *Annu. Rev. Mar. Sci.*, 1, 169–192,
681 <https://doi.org/10.1146/annurev.marine.010908.163834>, 2009.
682
683 Du, M. et al. High resolution measurements of methane and carbon dioxide in surface waters
684 over a natural seep reveal dynamics of dissolved phase air–sea flux. *Environ. Sci. Technol.* 48,
685 10165–10173 (2014).
686
687 Fiedler, B., Fietzek, P., Vieira, N., Silva, P., Bittig, H. C., and Körtzinger, A.: In situ CO₂ and
688 O₂ measurements on a profiling float, *J. Atmospheric Ocean. Technol.*, 30, 112–
689 126, <https://doi.org/10.1175/JTECH-D-12-00043.1>, 2013.
690
691 Fietzek, P., Fiedler, B., Steinhoff, T., and Körtzinger, A.: In situ quality assessment of a novel
692 underwater *p*CO₂ sensor based on membrane equilibration and NDIR spectrometry, *J.*
693 *Atmospheric Ocean. Technol.*, 31, 181–196, <https://doi.org/10.1175/JTECH-D-13-00083.1>,
694 2014.
695
696 Garcia-Tigreros, F., Leonte, M., Ruppel, C. D., Ruiz-Angulo, A., Joung, D. J., Young, B.,
697 and Kessler, J. D.: Estimating the Impact of Seep Methane Oxidation on Ocean pH
698 and Dissolved Inorganic Radiocarbon Along the U.S. Mid-Atlantic Bight, *J. Geophys. Res.*
699 *Biogeosciences*, 126, e2019JG005621, <https://doi.org/10.1029/2019JG005621>, 2021.
700
701 Goodman, S., Davies, P., Maddox, M., and Durkee, J.: Arctic Methane – Situational Awareness,
702 Assessment and Policy Directions, Results of the June 23rd, 2022 Arctic Methane Workshop,

703 Summary Report, 2022.

704

705 Gruber, N., Clement, D., Carter, B. R., Feely, R. A., van Heuven, S., Hoppema, M., Ishii, M.,

706 Key, R. M., Kozyr, A., Lauvset, S. K., Lo Monaco, C., Mathis, J. T., Murata, A., Olsen, A.,

707 Perez, F. F., Sabine, C. L., Tanhua, T., and Wanninkhof, R.: The oceanic sink for anthropogenic

708 CO₂ from 1994 to 2007, *Science*, 363, 1193–1199,

709 <https://doi.org/10.1126/science.aau5153>, 2019.

710

711 Gruber, N., Boyd, P. W., Frölicher, T. L., and Vogt, M.: Biogeochemical extremes

712 and compound events in the ocean, *Nature*, 600, 395–407, <https://doi.org/10.1038/s41586-021->

713 03981-7, 2021.

714

715 Hauri, C., McDonnell, A., Winsor, P., Irving, B., and Statscewich, H.: Development of

716 an Autonomous Carbon Glider to Monitor Sea-Air CO₂ Fluxes in the Chukchi Sea, Bureau of

717 Ocean Energy Management, 2018.

718

719 Hauri, C., Irving, B., Hayes, D., Abdi, E., Kemme, J., Kinski, N., McDonnell,

720 A.M.P.: CO₂ Seaglider trajectory file from Gulf of Alaska 2022.

721 SEANOE. <https://doi.org/10.17882/100964>, 2022.

722

723 Hauri, C., Irving, B., Hayes, D., Abdi, E., Kemme, J., Kinski, N., McDonnell,

724 A.M.P.: CO₂ Seaglider trajectory file from Gulf of Alaska 2023.

725 SEANOE. <https://doi.org/10.17882/100965>, 2023.

726

727 Hauri, C., Pagès, R., Hedstrom, K., Doney, S. C., Dupont, S., Ferriss, B., and Stuecker, M. F.:
728 More Than Marine Heatwaves: A New Regime of Heat, Acidity, and Low Oxygen Compound
729 Extreme Events in the Gulf of Alaska, *AGU Adv.*, 5, e2023AV001039,
730 <https://doi.org/10.1029/2023AV001039>, 2024.

731

732 Hemming, M. P., Kaiser, J., Heywood, K. J., Bakker, D. C. E., Boutin, J., Shitashima, K., Lee,
733 G., Legge, O., and Onken, R.: Measuring pH variability using an experimental sensor on an
734 underwater glider, *Ocean Sci.*, 13, 427–442, <https://doi.org/10.5194/os-13-427-2017>, 2017.

735

736 Irving, B., Hauri, C., Hayes, D., Abdi, E., Kinski, N.: Carbon Dioxide SOP, Version 1.0.0.
737 (GitHub Repository, Carbon Dioxide
738 SOP, https://britairving.github.io/Carbon_Dioxide_SOP/README.html (last access: 7 July,
739 2024).

740

741 Islam, F., DeGrandpre, M. D., Beatty, C. M., Timmermans, M.-L., Krishfield, R. A., Toole, J.
742 M., and Laney, S. R.: Sea surface CO₂ and O₂ dynamics in the partially ice-covered Arctic
743 Ocean, *J. Geophys. Res. Oceans*, 122, 1425–1438, <https://doi.org/10.1002/2016JC012162>, 2017.

744

745 Jiang, L.-Q., Feely, R. A., Wanninkhof, R., Greeley, D., Barbero, L., Alin, S., Carter, B. R.,
746 Pierrot, D., Featherstone, C., Hooper, J., Melrose, C., Monacci, N., Sharp, J. D., Shellito, S., Xu,
747 Y.-Y., Kozyr, A., Byrne, R. H., Cai, W.-J., Cross, J., Johnson, G. C., Hales, B., Langdon, C.,
748 Mathis, J., Salisbury, J., and Townsend, D. W.: Coastal Ocean Data Analysis Product in North

749 America (CODAP-NA) – an internally consistent data product for discrete inorganic carbon,
750 oxygen, and nutrients on the North American ocean margins, *Earth Syst. Sci. Data*, 13, 2777–
751 2799, <https://doi.org/10.5194/essd-13-2777-2021>, 2021.

752

753 Johnson, G. C. and Lyman, J. M.: Warming trends increasingly dominate global ocean,
754 *Nat. Clim. Change*, 10, 757–761, <https://doi.org/10.1038/s41558-020-0822-0>, 2020.

755

756 Kessler, J. Atlantic bubble bath. *Nature Geosci* 7, 625–
757 626, <https://doi.org/10.1038/ngeo2238>, 2014.

758

759 Kroeker, K. J., Kordas, R. L., and Harley, C. D. G.: Embracing interactions in
760 ocean acidification research: Confronting multiple stressor scenarios and context dependence,
761 *Biol. Lett.*, 13, <https://doi.org/10.1098/rsbl.2016.0802>, 2017.

762

763 Laufkötter, C., Zscheischler, J., and Frölicher, T. L.: High-impact marine heatwaves attributable
764 to human-induced global warming, *Science*, 369, 1621–
765 1625, <https://doi.org/10.1126/science.aba0690>, 2020.

766

767 Lauvset, S. K., Lange, N., Tanhua, T., Bittig, H. C., Olsen, A., Kozyr, A., Alin, S., Álvarez, M.,
768 Azetsu-Scott, K., Barbero, L., Becker, S., Brown, P. J., Carter, B. R., da Cunha, L. C., Feely, R.
769 A., Hoppema, M., Humphreys, M. P., Ishii, M., Jeansson, E., Jiang, L.-Q., Jones, S. D., Lo
770 Monaco, C., Murata, A., Müller, J. D., Pérez, F. F., Pfeil, B., Schirnick, C., Steinfeldt,
771 R., Suzuki, T., Tilbrook, B., Ulfsbo, A., Velo, A., Woosley, R. J., and Key, R. M.:

772 GLODAPv2.2022: the latest version of the global interior ocean biogeochemical data product,
773 Earth Syst. Sci. Data, 14, 5543–5572, <https://doi.org/10.5194/essd-14-5543-2022>, 2022.

774

775 Lee, K., Kim, T.-W., Byrne, R. H., Millero, F. J., Feely, R. A., and Liu, Y.-M.: The universal
776 ratio of boron to chlorinity for the North Pacific and North Atlantic oceans, *Geochim.*
777 *Cosmochim. Acta*, 74, 1801–1811, <https://doi.org/10.1016/j.gca.2009.12.027>, 2010.

778

779 Leonte, M., Kessler, J. D., Kellermann, M. Y., Arrington, E. C., Valentine, D. L., Sylva, S.
780 P.: Rapid rates of aerobic methane oxidation at the feather edge of gas hydrate stability in the
781 waters of Hudson Canyon, US Atlantic Margin. *Geochim. Cosmochim. Acta* 204, 375–
782 387, <https://doi.org/10.1016/j.gca.2017.01.009>, 2017.

783

784 Lueker, T. J., Dickson, A. G., and Keeling, C. D.: Ocean $p\text{CO}_2$ calculated from dissolved
785 inorganic carbon, alkalinity, and equations for K_1 and K_2 : validation based on laboratory
786 measurements of CO_2 in gas and seawater at equilibrium, *Mar. Chem.*, 70, 105–
787 119, [https://doi.org/10.1016/S0304-4203\(00\)00022-0](https://doi.org/10.1016/S0304-4203(00)00022-0), 2000.

788

789 Manley, J. and Willcox, S.: The Wave Glider: A persistent platform for ocean science,
790 in: OCEANS'10 IEEE SYDNEY, OCEANS'10 IEEE SYDNEY, 1–
791 5, <https://doi.org/10.1109/OCEANSSYD.2010.5603614>, 2010.

792

793 McGinnis, D. F., Greinert, J., Artemov, Y., Beaubien, S. E. & Wüest, A. Fate of rising methane
794 bubbles in stratified waters: How much methane reaches the atmosphere? *J. Geophys.*

795 Res. 111, <https://doi.org/10.1029/2005jc003183>, 2006.

796

797 Metzl, N., Fin, J., Lo Monaco, C., Mignon, C., Alliouane, S., Antoine, D., Bourdin, G., Boutin,
798 J., Bozec, Y., Conan, P., Coppola, L., Diaz, F., Douville, E., Durrieu de Madron, X., Gattuso, J.-
799 P., Gazeau, F., Golbol, M., Lansard, B., Lefèvre, D., Lefèvre, N., Lombard, F., Louanchi,
800 F., Merlivat, L., Olivier, L., Petrenko, A., Petton, S., Pujo-Pay, M., Rabouille, C., Reverdin, G.,
801 Ridame, C., Tribollet, A., Vellucci, V., Wagener, T., and Wimart-Rousseau, C.: A synthesis of
802 ocean total alkalinity and dissolved inorganic carbon measurements from 1993 to 2022: the
803 SNAPO-CO2-v1 dataset, *Earth Syst. Sci. Data*, 16, 89–120, [https://doi.org/10.5194/essd-16-89-](https://doi.org/10.5194/essd-16-89-2024)
804 2024, 2024.

805

806 Meurer, W. P., Blum, J., and Shipman, G.: Volumetric Mapping of Methane Concentrations at
807 the Bush Hill Hydrocarbon Seep, Gulf of Mexico, *Front. Earth Sci.*,
808 9, <https://doi.org/10.3389/feart.2021.604930>, 2021.

809

810 Monacci, N.M.; Bott, R.; Cross, J.N.; Dougherty, S.; Maenner, S.; Musielewicz, S.; Osborne, J.;
811 Sutton, A. (2023). High-resolution ocean and atmosphere $p\text{CO}_2$ time-series measurements from
812 mooring GAKOA_149W_60N. High-resolution ocean and atmosphere $p\text{CO}_2$ time-series
813 measurements from mooring GAKOA_149W_60N in the Gulf of Alaska (NCEI Accession
814 0116714). NOAA National Centers for Environmental
815 Information. Dataset. https://doi.org/10.3334/cdiac/otg.tsm_gakoa_149w_60n

816

817 Manley, J. and Willcox, S.: The Wave Glider: A persistent platform for ocean science, in:

818 OCEANS'10 IEEE SYDNEY, OCEANS'10 IEEE SYDNEY, 1–
819 5, <https://doi.org/10.1109/OCEANSSYD.2010.5603614>, 2010.
820
821 Miloshevich, L. M., Paukkunen, A., Vömel, H., and Oltmans, S. J.: Development and
822 Validation of a Time-Lag Correction for Vaisala Radiosonde Humidity Measurements,
823 J. Atmospheric Ocean. Technol., 21, 1305–1327, [https://doi.org/10.1175/1520-0426\(2004\)021<1305:DAVOAT>2.0.CO;2](https://doi.org/10.1175/1520-0426(2004)021<1305:DAVOAT>2.0.CO;2), 2004.
824
825
826 Myhre, G., Shindell, D., Bréon, F.-M., Collins, W., Fuglestedt, J., Huang, J., Koch, D.,
827 Lamarque, J.-F., Lee, D., Mendoza, B., Nakajima, T., Robock, A., Stephens, G., Takemura, T.,
828 and Zhang, H.: Anthropogenic and Natural Radiative Forcing, in: Climate Change 2013: The
829 Physical Science Basis, Contribution of Working Group I to the Fifth Assessment Report of
830 the Intergovernmental Panel on Climate Change, edited by: Stocker, T. F., Qin, D., Plattner, G.-
831 K., Tignor, M., Allen, S. K., Boschung, J., Nauels, A., Xia, Y., Bex, V., and Midgley, P.
832 M., Cambridge University Press, Cambridge, UK, New York, NY, USA, 2013.
833
834 National Academies of Sciences, Engineering, and Medicine: A Research Strategy for Ocean-
835 based Carbon Dioxide Removal and Sequestration. Washington, DC: The National
836 Academies Press. <https://doi.org/10.17226/26278>, 2022.
837
838 Newton, J. A., Feely, R. A., Jewett, E. B., Williamson, P., and Mathis, J.: Global
839 ocean acidification observing network: requirements and governance plan, GOA-
840 ON, Washington, 61 pp., 2015.

841

842 Nickford, S., Palter, J. B., Donohue, K., Fassbender, A. J., Gray, A. R., Long, J., Sutton, A. J.,
843 Bates, N. R., and Takeshita, Y.: Autonomous Wintertime Observations of Air-Sea Exchange in
844 the Gulf Stream Reveal a Perfect Storm for Ocean CO₂ Uptake, *Geophys. Res. Lett.*, 49,
845 e2021GL096805, <https://doi.org/10.1029/2021GL096805>, 2022.

846

847 von Oppeln-Bronikowski, N., de Young, B., Atamanchuk, D., and Wallace, D.: Glider-based
848 observations of CO₂ in the Labrador Sea, *Ocean Sci.*, 17, 1–16, [https://doi.org/10.5194/os-17-1-](https://doi.org/10.5194/os-17-1-2021)
849 2021, 2021.

850

851 OceanGliders Oxygen SOP: <https://nora.nerc.ac.uk/id/eprint/533559/>, last access: 24
852 January 2024.

853

854 Orr, J. C., Epitalon, J. M., Dickson, A. G., and Gattuso, J. P.: Routine uncertainty propagation
855 for the marine carbon dioxide system, *Mar. Chem.*, 207, 84–
856 107, <https://doi.org/10.1016/j.marchem.2018.10.006>, 2018.

857

858 Perez, F. F. and Fraga, F.: Association constant of fluoride and hydrogen ions in seawater, *Mar.*
859 *Chem.*, 21, 161–168, [https://doi.org/10.1016/0304-4203\(87\)90036-3](https://doi.org/10.1016/0304-4203(87)90036-3), 1987.

860

861 Pinnau, I., and Toy, L. G.: Gas and vapor transport properties of amorphous perfluorinated
862 copolymer membranes based on 2,2-bis(trifluoromethyl)-4,5-difluoro-1,3-
863 dioxole/tetrafluoroethylene, *Journal of Membrane Science*, 109 (1), 125-

864 133, [https://doi.org/10.1016/0376-7388\(95\)00193-X](https://doi.org/10.1016/0376-7388(95)00193-X), 1996.

865

866 Possenti, L., Humphreys, M. P., Bakker, D. C. E., Cobas-García, M., Fernand, L., Lee, G. A.,
867 Pallottino, F., Loucaides, S., Mowlem, M. C., and Kaiser, J.: Air-Sea Gas Fluxes and
868 Remineralization From a Novel Combination of pH and O₂ Sensors on a Glider, *Front. Mar. Sci.*,
869 8, 1–19, <https://doi.org/10.3389/fmars.2021.696772>, 2021.

870

871 Reeburgh, W. Oceanic methane biogeochemistry. *Am. Chem. Soc.* 107, 486–513, 2007.

872

873 Sarmiento, J. L. and Gruber, N.: *Ocean Biogeochemical Dynamics*, Princeton University Press,
874 Princeton, NJ, 526 pp., ISBN 9780691017075, 2006.

875

876 Qi, D., Ouyang, Z., Chen, L., Wu, Y., Lei, R., Chen, B., Feely, R. A., Anderson, L. G., Zhong,
877 W., Lin, H., Polukhin, A., Zhang, Y., Zhang, Y., Bi, H., Lin, X., Luo, Y., Zhuang, Y., He, J.,
878 Chen, J., and Cai, W. J.: Climate change drives rapid decadal acidification in the Arctic Ocean
879 from 1994 to 2020, *Science*, 377, 1544–1550, <https://doi.org/10.1126/science.abo0383>, 2022.

880

881 Rees, A. P., Bange, H. W., Arévalo-Martínez, D. L., Artioli, Y., Ashby, D. M., Brown, I.,
882 Campen, H. I., Clark, D. R., Kitidis, V., Lessin, G., Tarran, G. A., and Turley, C.: Nitrous oxide
883 and methane in a changing Arctic Ocean, *Ambio*, 51, 398–410, [https://doi.org/10.1007/s13280-](https://doi.org/10.1007/s13280-021-01633-8)
884 021-01633-8, 2022.

885

886 Saba, G. K., Wright-Fairbanks, E., Chen, B., Cai, W. J., Barnard, A. H., Jones, C. P., Branham,

887 C. W., Wang, K., and Miles, T.: The Development and Validation of a Profiling Glider Deep
888 ISFET-Based pH Sensor for High Resolution Observations of Coastal and Ocean Acidification,
889 *Front. Mar. Sci.*, 6, 1–17, <https://doi.org/10.3389/fmars.2019.00664>, 2019.

890

891 Sabine, C. L. and Tanhua, T.: Estimation of anthropogenic CO₂ inventories in the ocean., *Annu.*
892 *Rev. Mar. Sci.*, 2, 175–98, <https://doi.org/10.1146/annurev-marine-120308-080947>, 2010.

893

894 Sabine, C. L., Feely, R. A., Gruber, N., Key, R. M., Lee, K., Bullister, J. L., Wanninkhof, R.,
895 Wong, C. S., Wallace, D. W. R., Tilbrook, B., Millero, F. J., Peng, T.-H., Kozyr, A., Ono, T.,
896 and Rios, A. F.: The oceanic sink for anthropogenic CO₂, *Science*, 305, 367–
897 71, <https://doi.org/10.1126/science.1097403>, 2004.

898

899 Sejr, M. K., Krause-Jensen, D., Rysgaard, S., Sørensen, L. L., Christensen, P. B., and Glud, R.
900 N.: Air-sea flux of CO₂ in arctic coastal waters influenced by glacial melt water and sea ice,
901 *Tellus B*, 63, 815–822, <https://doi.org/10.1111/j.1600-0889.2011.00540.x>, 2011.

902

903 Sharp, J. D., Pierrot, D., Humphreys, M. P., Epitalon, J.-M., Orr, J. C., Lewis, E. R., and
904 Wallace, D. W. R.: CO2SYSv3 for MATLAB, , <https://doi.org/10.5281/zenodo.7552554>, 2023.

905

906 Shakhova, N., Semiletov, I., Salyuk, A., Yusupov, V., Kosmach, D., and Gustafsson, Ö.:
907 Extensive Methane Venting to the Atmosphere from Sediments of the East Siberian Arctic Shelf,
908 *Science*, 327, 1246–1250, <https://doi.org/10.1126/science.1182221>, 2010.

909

910 Skarke, A., Ruppel, C., Kodis, M., Brothers, D. and Lobecker, E. Widespread methane leakage
911 from the sea floor on the northern US Atlantic margin. *Nat. Geosci.* 7, 657 (2014).
912

913 Sparrow, K. J., Kessler, J. D., Southon, J. R., Garcia-Tigreros, F., Schreiner, K. M., Ruppel, C.
914 D., Miller, J. B., Lehman, S. J., and Xu, X.: Limited contribution of ancient methane to surface
915 waters of the U.S. Beaufort Sea shelf, *Sci. Adv.*, 4, eaao4842,
916 <https://doi.org/10.1126/sciadv.aao4842>, 2018.
917

918 Sulpis, O., Lauvset, S. K., and Hagens, M.: Current estimates of K and K appear inconsistent
919 with measured CO₂ system parameters in cold oceanic regions, *Ocean Sci.*, 16, 847–
920 862, <https://doi.org/10.5194/os-16-847-2020>, 2020.
921

922 Takeshita, Y., Jones, B. D., Johnson, K. S., Chavez, F. P., Rudnick, D. L., Blum, M., Conner, K.,
923 Jensen, S., Long, J. S., Maughan, T., Mertz, K. L., Sherman, J. T., and Warren, J. K.: Accurate
924 pH and O₂ Measurements from Spray Underwater Gliders, *J. Atmospheric Ocean. Technol.*, 38,
925 181–195, <https://doi.org/10.1175/JTECH-D-20-0095.1>, 2021.
926

927 Thompson, T., Saba, G. K., Wright-Fairbanks, E., Barnard, A. H., and Branham, C. W.:
928 Best Practices for Sea-Bird Scientific deep ISFET-based pH sensor integrated into a Slocum
929 Webb Glider, in: *OCEANS 2021: San Diego – Porto*, *OCEANS 2021: San Diego – Porto*, 1–8,
930 <https://doi.org/10.23919/OCEANS44145.2021.9706067>, 2021.
931

932 Vergara-Jara, M. J., DeGrandpre, M. D., Torres, R., Beatty, C. M., Cuevas, L. A., Alarcón, E.,

933 and Iriarte, J. L.: Seasonal changes in carbonate saturation state and air-sea CO₂ fluxes during an
934 annual cycle in a stratified-temperate fjord (Reloncaví Fjord, Chilean Patagonia), *J. Geophys.*
935 *Res. Biogeosciences*, 124, 2851–2865, <https://doi.org/10.1029/2019jg005028>, 2019.

936

937 Widdicombe, S., Isensee, K., Artioli, Y., Gaitán-Espitia, J. D., Hauri, C., Newton, J. A., et
938 al.: Unifying biological field observations to detect and compare ocean acidification impacts
939 across marine species and ecosystems: What to monitor and why. *Ocean Science*, 19(1), 101–
940 119. <https://doi.org/10.5194/os-19-101-2023>, 2023.

941

942 Woosley, R. J. and Millero, F. J.: Freshening of the western Arctic negates anthropogenic carbon
943 uptake potential, *Limnol. Oceanogr.*, <https://doi.org/10.1002/lno.11421>, 2020.

944

945 Wright-Fairbanks, E. K., Miles, T. N., Cai, W.-J., Chen, B., and Saba, G. K.:

946 Autonomous Observation of Seasonal Carbonate Chemistry Dynamics in the Mid-Atlantic

947 Bight, *J. Geophys. Res. Oceans*, 125,

948 e2020JC016505, <https://doi.org/10.1029/2020JC016505>, 2020.

949

950

951

952

953 **Tables**

954 **Table 1. Tank experiment.** Evaluation of SG HydroC CO₂ and SG HydroC CH₄ sensors
 955 compared to reference discrete $p\text{CO}_2^{\text{disc}}$ and $p\text{CH}_4^{\text{disc}}$. Units of $p\text{CO}_2$ and $p\text{CH}_4$ are μatm except
 956 when shown as percent difference in parenthesis (Equation 1). Columns with subscripts sn422
 957 and sn0718 indicate data from sensors HydroC CO2T-0422-001 and HydroC CO2T-0718-001,
 958 respectively. Superscript RTC indicates response time corrected values following Dølvén et al.
 959 (2022). $p\text{CO}_2^{\text{disc}}$ and $p\text{CH}_4^{\text{disc}}$ values are the average of triplicate bottles and are shown in Figure
 960 6.

961

Triplicate Date Time (UTC)	$p\text{CO}_2^{\text{disc}} \pm \text{uc} (\mu\text{atm})$	$p\text{CO}_{2,\text{sn422}}^{\text{RTC}} - p\text{CO}_2^{\text{disc}}$	$p\text{CO}_{2,\text{sn0718}}^{\text{RTC}} - p\text{CO}_2^{\text{disc}}$	$p\text{CH}_4^{\text{disc}} \pm \text{u} (\mu\text{atm})$	$p\text{CH}_4^{\text{RTC}} - p\text{CH}_4^{\text{disc}}$
5/2/2022 3:25	298.7 ± 10.2	-0.9 (-0.3%)	-	-	-
5/2/2022 7:32	227.1 ± 7.8	4.3 (1.9%)	2.4 (1.1%)	-	-
5/2/2022 11:27	223.3 ± 7.7	0.7 (0.3%)	-2.6 (-1.2%)	-	-
5/2/2022 15:30	227.8 ± 7.9	-1.1 (-0.5%)	-3.3 (-1.4%)	-	-
5/2/2022 00:11	-	-	-	25.4 ± 2.1	4.0 (15.8 %)
5/2/2022 12:06	-	-	-	7.3 ± 1.3	0.5 (6.6 %)

962

963

964 **Table 2. Profiling experiment.** Evaluation of SG HydroC CO₂ sensor compared to reference
 965 discrete $p\text{CO}_2^{\text{disc}}$. Units of $p\text{CO}_2$ are μatm except when shown as percent difference in
 966 parenthesis (Eq. 1). $p\text{CO}_2$ with subscripts sn422 indicate data from the HydroC installed on the
 967 rosette (HydroC CO2T-0422-001). The superscript RTC indicates response time corrected values
 968 following Dølven et al. (2022).

Discrete Date Time (UTC)	Discrete Depth (m)	$p\text{CO}_2^{\text{disc}} \pm \text{uc}$ (μatm)	$p\text{CO}_{2,\text{sn422}}^{\text{RTC}} - p\text{CO}_2^{\text{disc}}$
5/3/2022 21:21	2.5	214.5 ± 7.5	5.4 (2.5%)
5/3/2022 21:39	19.9	246.8 ± 8.5	1.6 (0.6 %)
5/3/2022 22:33	9.6	244.4 ± 8.5	-3.3 (-1.4 %)
5/3/2022 22:34	9.7	234.7 ± 8.1	8.2 (3.5 %)

969

970

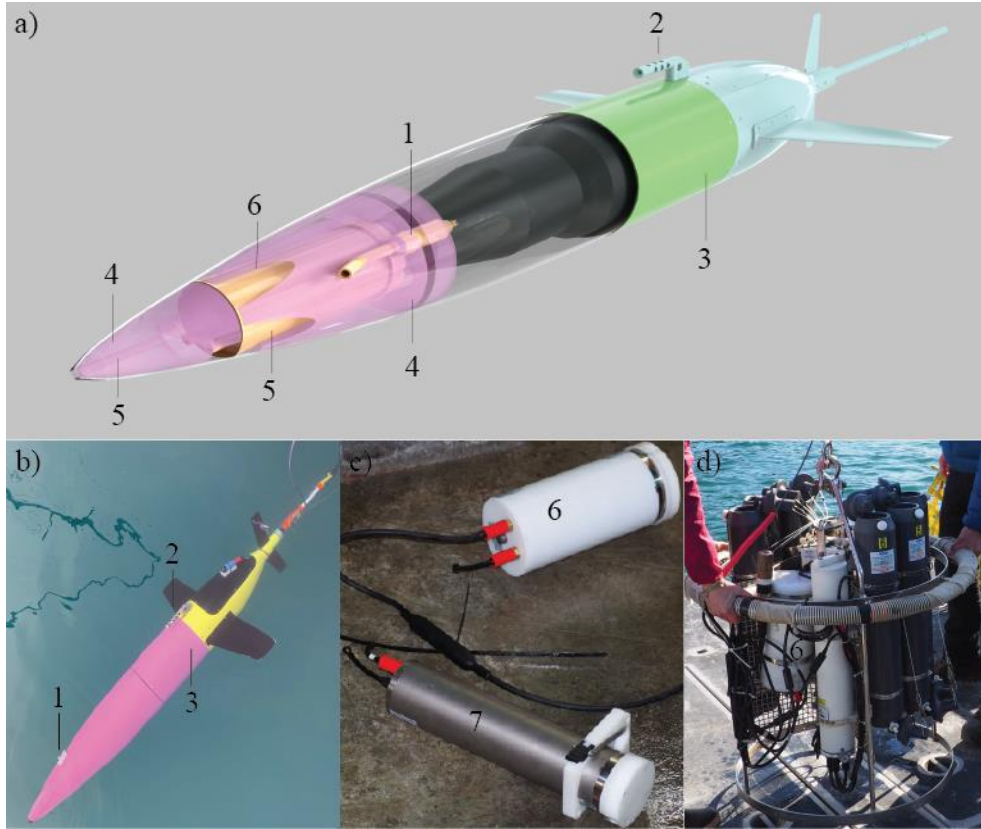
971 **Table 3. Seaglider HydroC evaluation with a nearby cast.** Evaluation of Seaglider integrated
972 and rosette mounted SG HydroC CO₂ sensors compared to $p\text{CO}_2^{\text{disc}}$ collected from a nearby cast.
973 Units of $p\text{CO}_2$ are μatm except when shown as percent difference in parenthesis (Eq. 1) and
974 differences between $p\text{CO}_{2,\text{Seaglider}}^{\text{RTC}}$ were calculated with the average (upcast and downcast
975 combined) 1-meter binned data. The superscript RTC indicates response time corrected values
976 following Dølven et al. (2022), and subscripts Rosette and Seaglider indicate the SG HydroC
977 CO₂ sensor mounted on the rosette (SG HydroC CO2T-0422-001) and integrated into the
978 Seaglider (SG HydroC CO2T-0718-001), respectively. Time delay (HH:MM) and spatial
979 distance (km) columns represent the distance between $p\text{CO}_{2,\text{Seaglider}}^{\text{RTC}}$ measured at the discrete
980 depth and the discrete date time. The asterisk (*) indicates the comparison with $p\text{CO}_{2,\text{Rosette}}^{\text{RTC}}$
981 taken as nearest in time before sensor zeroing (Figure S1).

Discrete Date Time (UTC)	Discrete Depth (m)	$p\text{CO}_2^{\text{disc}} \pm \text{uc}$ (μatm)	$p\text{CO}_{2,\text{Rosette}}^{\text{RTC}} - p\text{CO}_2^{\text{disc}}$	Delay (HH:MM)	Distance (km)	$p\text{CO}_{2,\text{Seaglider}}^{\text{RTC}} - p\text{CO}_2^{\text{disc}}$
5/7/2022 18:06	71.8	349.7 ± 7.8	-5.7 (-1.6 %)	02:47	0.4	10.2 (2.9 %)
5/7/2022 18:24	57.1	313.8 ± 6.7	12.1 (3.9 %)	03:05	0.6	8.3 (2.7 %)
5/7/2022 18:42	19.8	285.3 ± 6.1	0.8 (0.3 %)	03:23	0.8	8.6 (3.0 %)
5/7/2022 19:00	1.6	233.4 ± 5.0	-2.3 (-1.0 %)*	03:41	0.9	12.0 (5.1 %)

982

983

984 **Figures**



985

986 **Figure 1. CO₂ Seaglider.** CO₂ Seaglider a) schematic rendering and b) picture in Resurrection
987 Bay, Seward, Alaska, during a checkout dive on 6 February 2023, before beginning the first
988 winter mission collecting high resolution *p*CO₂ data. Highlighted are 1) SeaBird 5M pump, 2)
989 conductivity and temperature sail, 3) extension, 4) syntactic foam, 5) water flow channels, and 6)
990 SG HydroC CO₂ in a titanium housing, enabling *p*CO₂ observations down to 1000 m. c) Picture
991 of new SG HydroC CO₂ in a polyoxymethylene housing (6, rated to 300 m depth) and original
992 CONTROS HydroC™ CO₂ (7). d) Picture of rosette set up for profiling experiment.

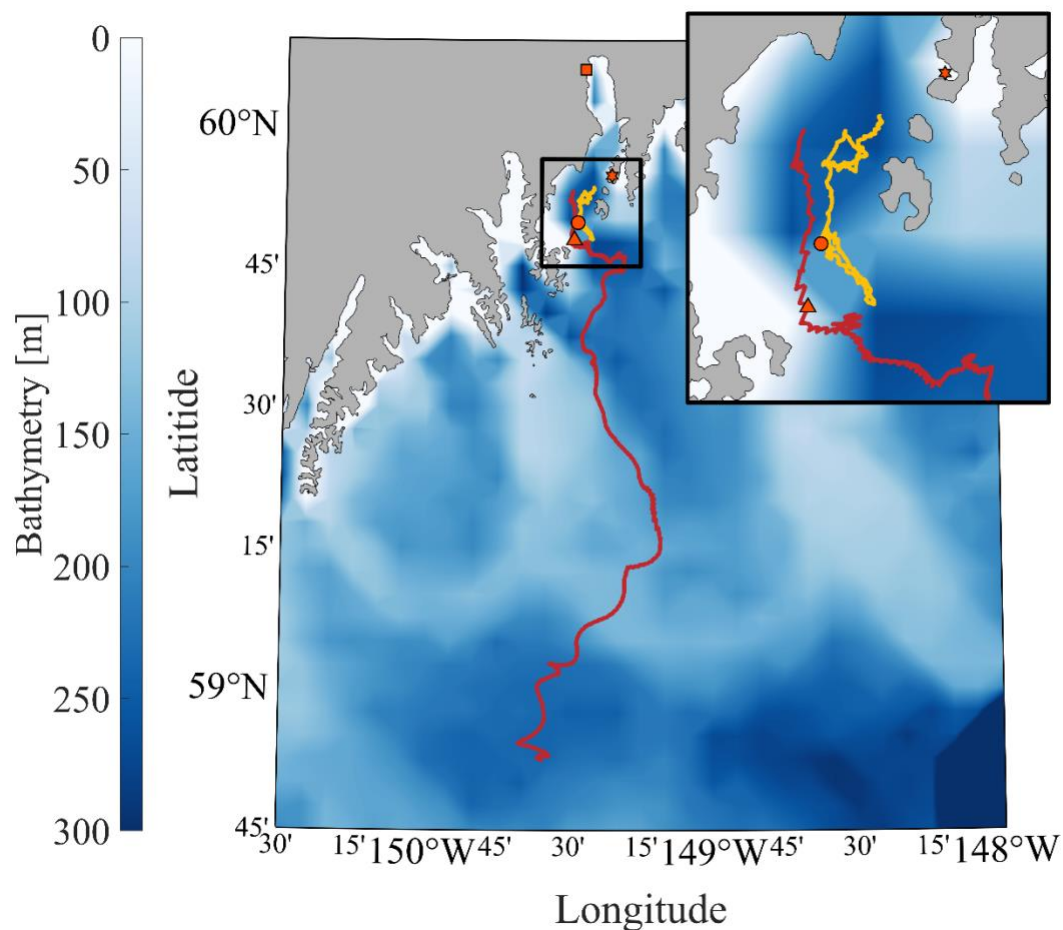


993

994 **Figure 2. SG HydroC CO₂ sensor mounting designs.** a) Titanium SG HydroC CO₂ (rated to

995 1000m) in a custom syntactic foam coat and b) polyoxymethylene SG HydroC CO₂ (rated to

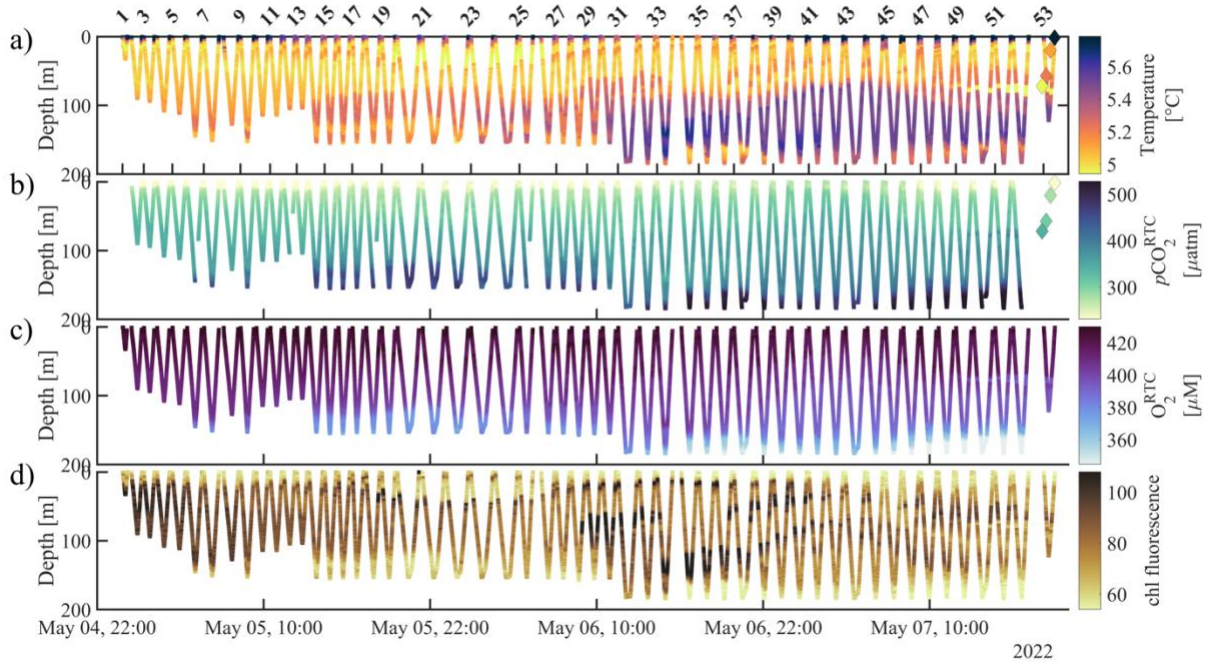
996 300m) with brackets.



997
 998 **Figure 3. Map of CO₂ Seaglider study area.** The bathymetry of the Gulf of Alaska is shown in
 999 color with zoomed in section of the head of Resurrection Bay (outlined black square and inset
 1000 map). Tracks of the CO₂ Seaglider from the 4 – 7 May 2022 and 8 – 21 February 2023 missions
 1001 are shown in yellow and red, respectively. Orange markers outlined in black show the location of
 1002 the Alutiiq Pride Marine Institute (square), National Oceanic and Atmospheric Administration’s
 1003 Gulf of Alaska Ocean Acidification mooring (star), 7 May CTD cast (circle), and last location
 1004 where $p\text{CO}_2$ was collected during the February 2023 mission (triangle).

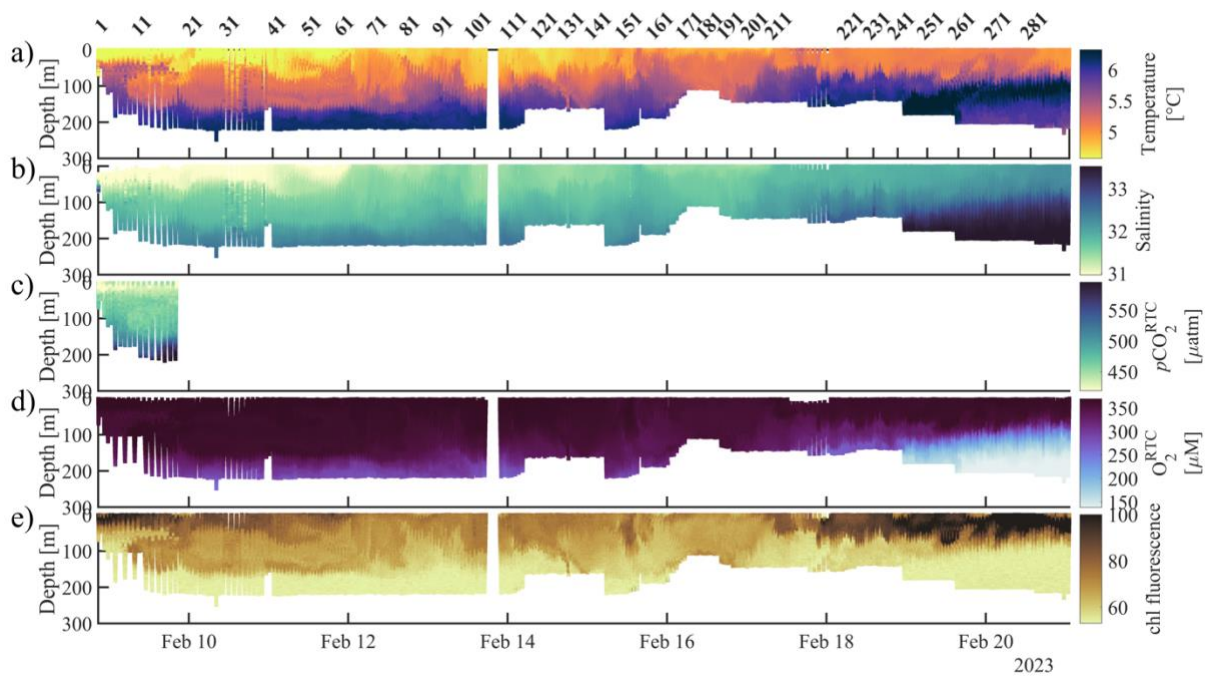
1005

1006 moving median filter applied to the RTC data. Black diamonds with red filling show discrete
1007 $p\text{CH}_4^{\text{disc}}$ and all discrete values of $p\text{CO}_2^{\text{disc}}$ and $p\text{CH}_4^{\text{disc}}$ are the average of triplicate bottles.



1008

1009 **Figure 4. CO₂ Seaglider data from the 4 – 7 May 2022 mission in Resurrection Bay,**
1010 **Seward, Alaska.** Depth profiles of a) Temperature [°C], b) RTC $p\text{CO}_2$ [μatm] c) RTC O_2 [μM],
1011 and d) raw chlorophyll fluorescence. The diamonds show discrete values that were taken during
1012 a CTD cast (Table 3).



1013

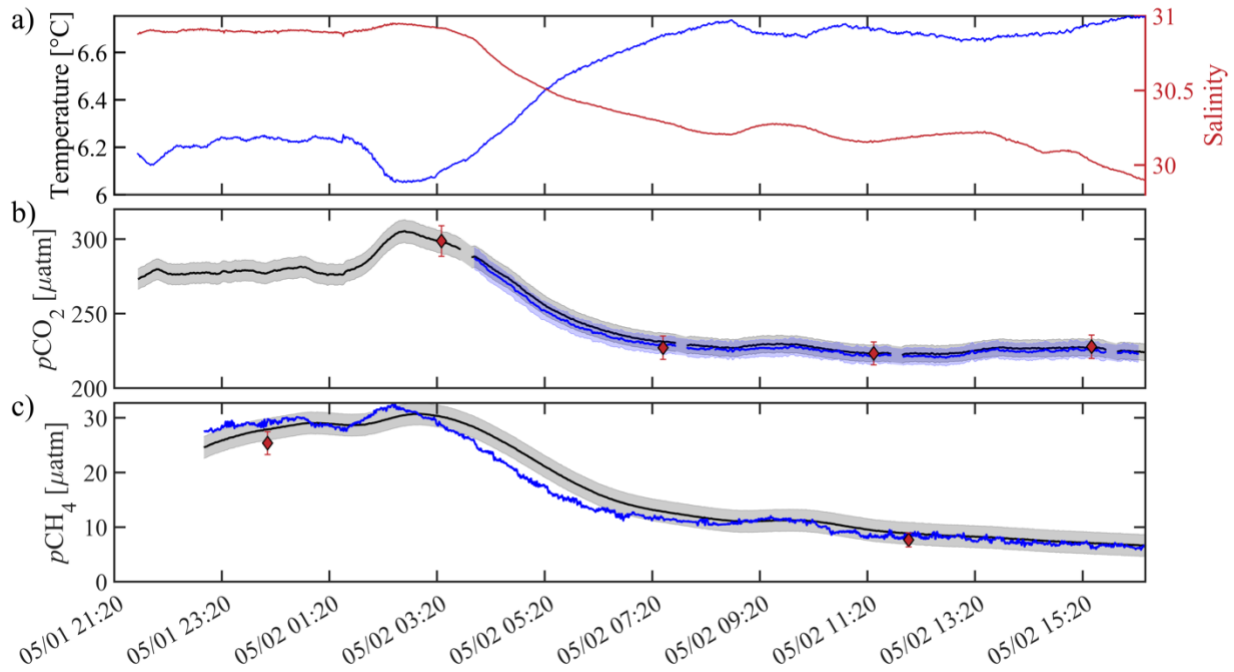
1014 **Figure 5. CO₂ Seaglider data collected during the 8 – 21 February 2023 winter mission.**

1015 Shown are a) temperature (°C), b) salinity, c) response time corrected $p\text{CO}_2$ ($p\text{CO}_2^{\text{RTC}}$, μatm), d)

1016 response time corrected oxygen (O_2^{RTC} , μM), and e) raw chlorophyll fluorescence (chl

1017 fluorescence) as time/dive number vs. pressure.

1018



1019

1020 **Figure 6. Sensor validation during a tank experiment at the Alutiiq Pride Marine Institute**

1021 **on 1 – 2 May 2022.** a) temperature (blue line) and salinity (red line) from a recently calibrated

1022 Sea-Bird Scientific SBE37. b) black (blue) lines show $p\text{CO}_2$ in μatm from HydroC CO2T-0422-

1023 001 (HydroC CO2T-0718-001) with the shaded gray (blue) areas showing a relative uncertainty

1024 of 2.5% (weather quality goal; Newton et al., 2015). Black circles with red filling show discrete

1025 $p\text{CO}_2^{\text{disc}}$ with error bars showing the combined standard uncertainty from *errors.m* (Orr et al.,

1026 2018). HydroC $p\text{CO}_2$ data are shown at 1 minute resolution with a 2-minute moving median

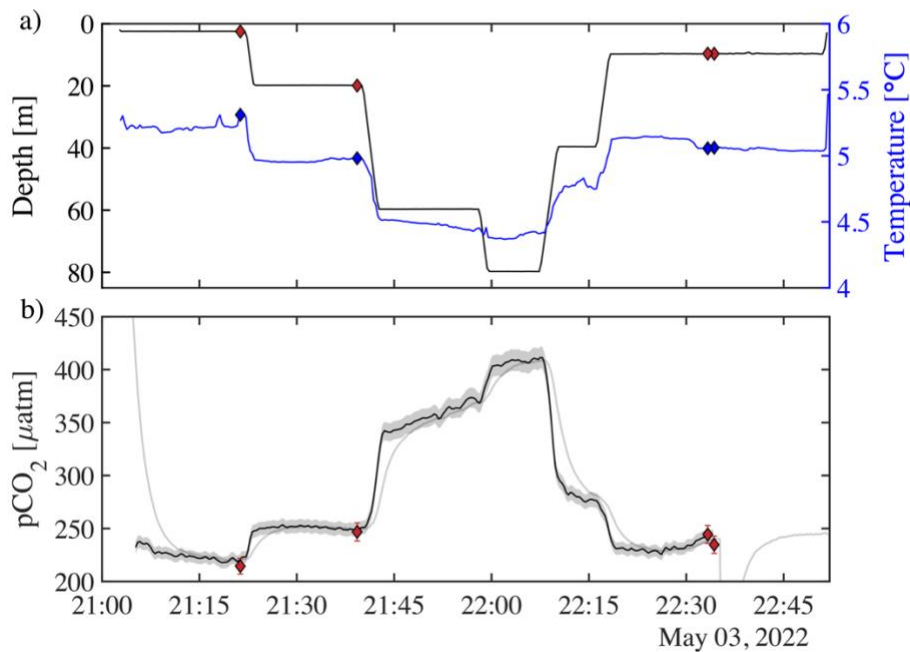
1027 filter applied and have not been corrected for response time, but differences were negligible (<

1028 0.1 μatm). c) Black line shows $p\text{CH}_4$ in μatm from HydroC CH4T-0422-001 with the shaded

1029 gray bar showing an uncertainty of 2 μatm . The blue line is the response time corrected signal

1030 with a response time of 43 minutes following Dølven et al., (2022). HydroC $p\text{CH}_4$ data are

1031 shown at 1 minute resolution with a 2-minute moving median filter applied to the raw data and a
1032 10-minute

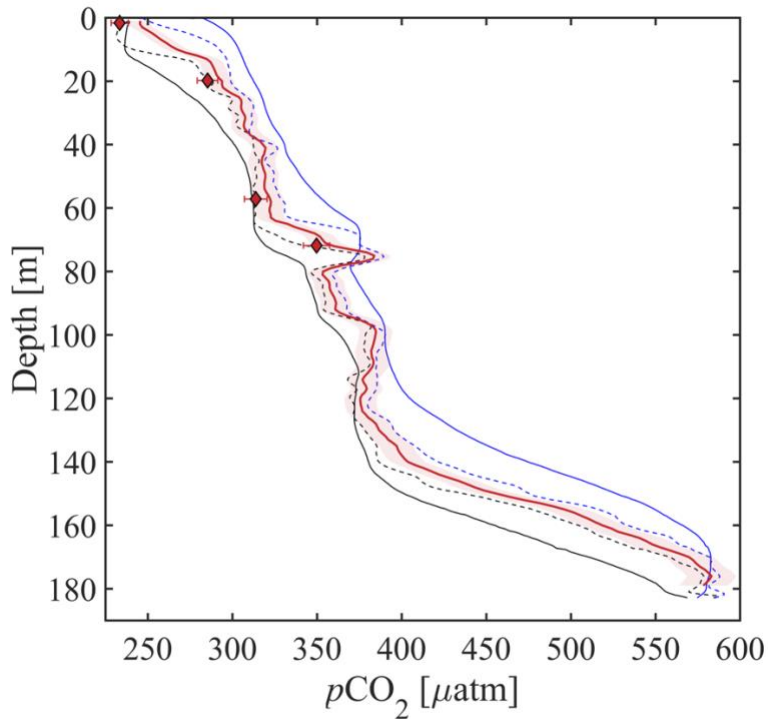


1033
1034 **Figure 7. Profiling experiments from 3 May 2022 with HydroC CO2T-0422-001 sensor**
1035 **mounted on the rosette.** a) Pressure vs time on the left (black) axis with diamonds showing
1036 rosette CTD values of pressure (red filled diamond), and temperature vs time on the right (blue)
1037 axis and temperature (blue filled diamond) at the time of the bottle fire. b) $p\text{CO}_2$ measured by
1038 the rosette mounted SG HydroC CO_2 sensor as raw (gray line) and response time corrected
1039 signal (thick black line; $p\text{CO}_{2,\text{sn422}}^{\text{RTC}}$ in Table 2) with shaded relative uncertainty of 2.5%
1040 (weather goal; Newton et al., 2015). $p\text{CO}_2^{\text{disc}}$ shown as red diamonds with vertical red error bars
1041 showing combined standard uncertainty (Orr et al., 2018). Table 2 shows differences between
1042 discrete $p\text{CO}_2^{\text{disc}}$ and $p\text{CO}_{2,\text{sn422}}^{\text{RTC}}$. The SG HydroC CO_2 sensor started a zeroing interval at

1043 22:35 on 3 May 2022, so $p\text{CO}_{2,\text{sn422}}^{\text{RTC}}$ is not shown after that time but signal recovery can be
1044 seen in the uncorrected signal (gray line).

1045

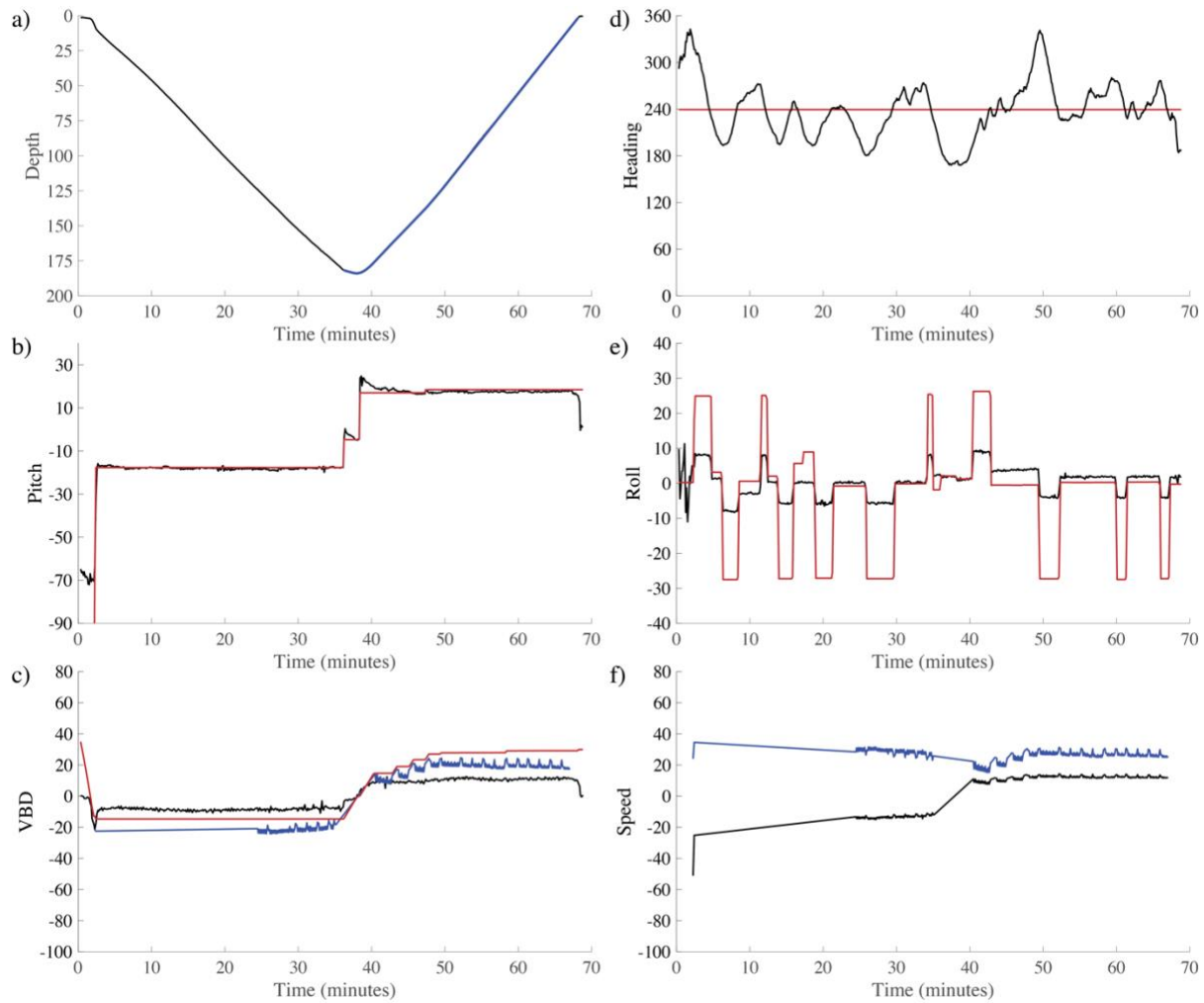
1046



1047

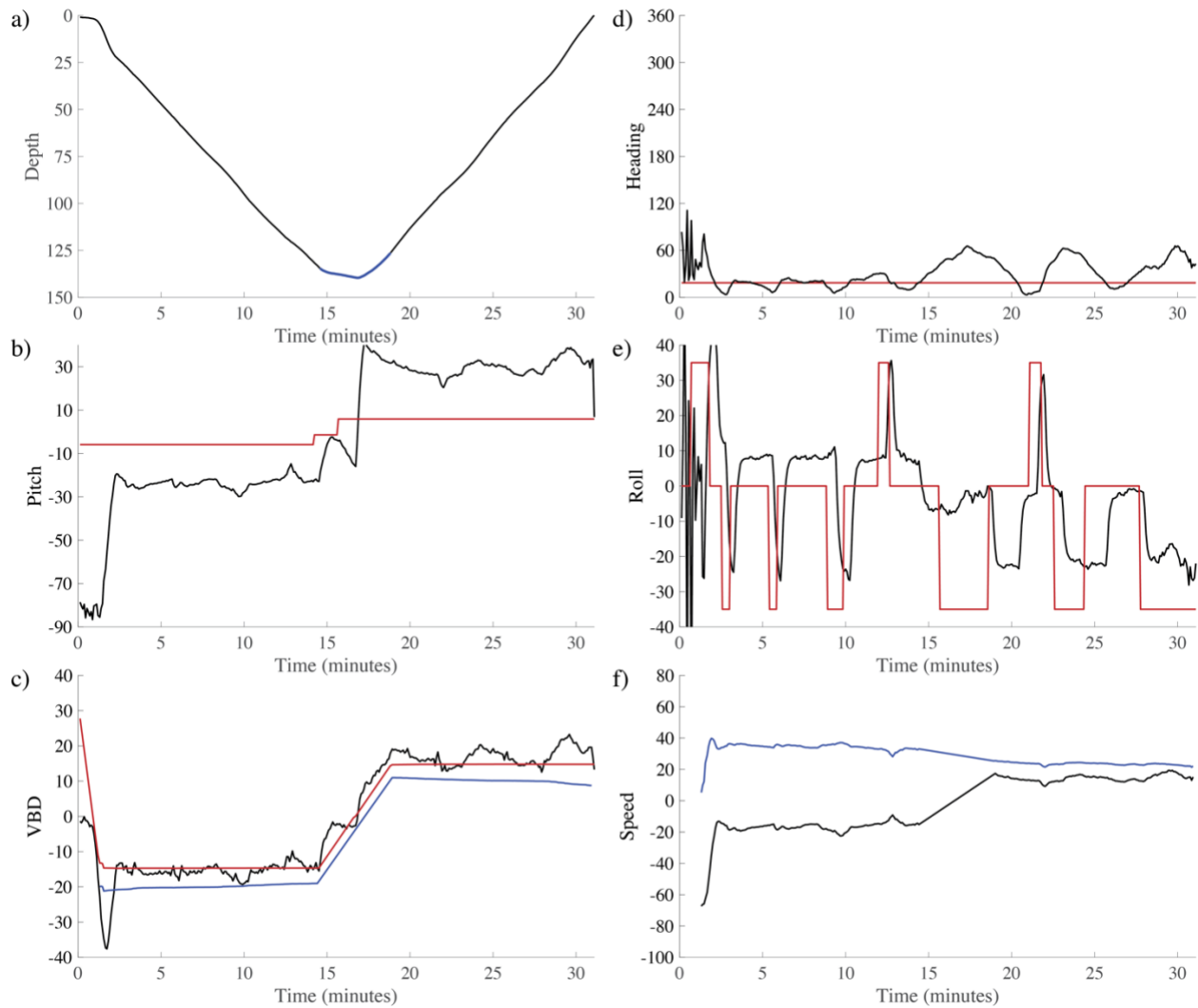
1048 **Figure 8. CO₂ Seaglider data from a single dive during the 4 – 7 May 2022 mission in**
1049 **Resurrection Bay, Seward, Alaska.** Depth profile of $p\text{CO}_2$ in μatm showing the original
1050 resolution smoothed $p\text{CO}_2$ used in the RT correction (downcast = solid black, upcast = solid
1051 blue), RTC $p\text{CO}_2$ following Dølven et al. (2022) (dashed black line = downcast, dashed blue line
1052 = upcast), and 1-meter binned RTC profile (thick red line) with red shading showing the relative
1053 uncertainty of 2.5 %. Discrete $p\text{CO}_2^{\text{disc}}$ shown as red diamonds with horizontal red error bars
1054 showing combined standard uncertainty (Orr et al., 2018). Differences between $p\text{CO}_2^{\text{disc}}$ and
1055 $p\text{CO}_{2,\text{Seaglider}}^{\text{RTC}}$ are shown in Table 3.

1056



1057

1058 **Figure 9. Dive details for the 300 m rated CO₂ Seaglider (dive# 51).** a) Depth (black line,
 1059 meters), b) pitch (black line, degrees) with pitch control (red line, mm of battery shift), c)
 1060 Change in displacement of Variable Buoyancy Drive (VBD) (red line, units of 10 cc), vertical
 1061 velocity from pressure measurements (black line, cm/s), and buoyancy (blue line, units of 10 g),
 1062 d) heading in (desired red line, measured black line, degrees), e) roll (battery roll position red
 1063 line, glider measured roll black line, degrees), and f) vertical speed (calculated from buoyancy
 1064 and pitch, black line, cm/s) and horizontal speed (calculated from buoyancy and pitch, blue line,
 1065 cm/s).



1066

1067 **Figure 10. Dive details for the 1000 m rated CO₂ Seaglider (dive# 203).** a) Depth (black line,

1068 meters), b) pitch (black line, degrees) with pitch control (red line, mm of battery shift), c)

1069 Change in displacement of Variable Buoyancy Drive (VBD) (red line, units of 10 cc), vertical

1070 velocity from pressure measurements (black line, cm/s), and buoyancy (blue line, units of 10 g),

1071 d) Heading (desired heading red line, measured heading black line, degrees) e) roll (battery roll

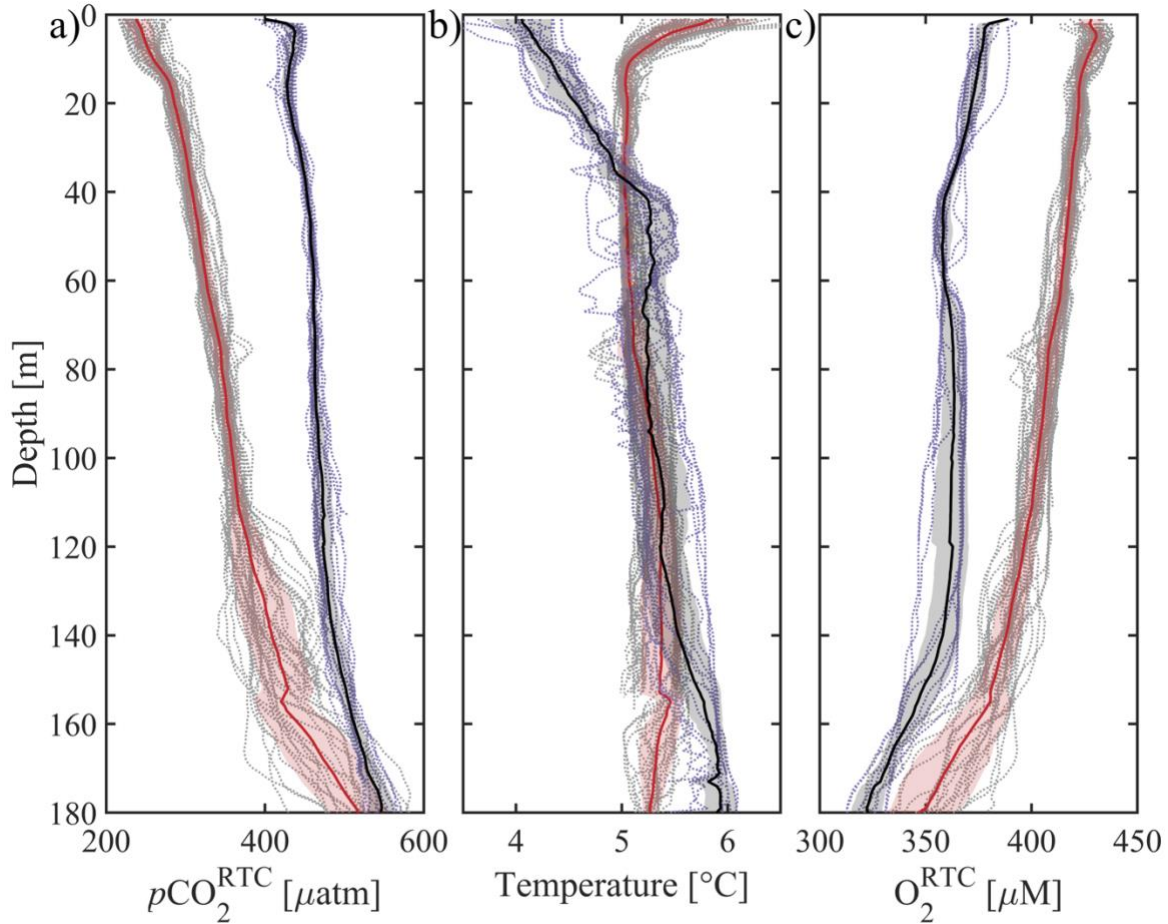
1072 position red line, glider measured roll black line, degrees), and f) vertical speed (calculated from

1073 buoyancy and pitch, black line, cm/s) and horizontal speed in cm/s (calculated from buoyancy

1074 and pitch, blue line, cm/s).

1075

1076



1077

1078

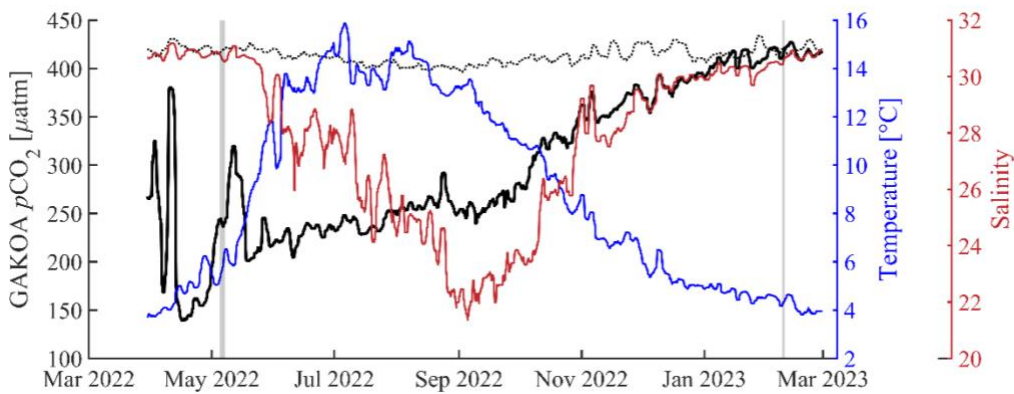
1079 **Figure 11. Averaged CO₂ Seaglider profiles from the 4 – 7 May 2022 and 8 – 21 February**
1080 **2023 missions in Resurrection Bay, Seward, Alaska.** Depth profiles of all 1-meter binned
1081 dives (dotted gray), average 1-meter binned dive from May 2022 mission (red thick line, dive#1-
1082 51, 00:01 5 May to 16:37 7 May, 2022) and February 2023 mission (black thick line, dive#1-17,
1083 20:50 8 February to 19:54 9 February, 2023) with shading showing the standard deviation of the

1084 values in each bin added and subtracted from the average. a) Response time corrected $p\text{CO}_2$
1085 ($p\text{CO}_2^{\text{RTC}}$, μatm), b) temperature [$^{\circ}\text{C}$], and c) response time corrected oxygen (O_2^{RTC} , μM).

1086

1087

1088



1089

1090 **Figure 12. National Oceanic Atmospheric Administration’s Gulf of Alaska ocean**
1091 **acidification surface time-series from March 2022 - 2023.** Left axis sea surface (dotted black
1092 line) and air (black line, 4 meter above sea level) $p\text{CO}_2$ [μatm] and right axes sea surface
1093 temperature (blue, $^{\circ}\text{C}$) and sea surface salinity (red). All data shown as 3 day running mean.
1094 Vertical shaded gray areas highlight the CO₂ Seaglider missions in May 2022 and February
1095 2023. The mooring is located at 59.911 $^{\circ}\text{N}$, -149.35 $^{\circ}\text{W}$ (Monacci et al., 2023).

1096

1097