# 1 Understanding Absorption by Black Versusthe Mechanism and

# 2 Importance of Brown Carbon Bleaching Across the Visible

# 3 <u>Spectrum</u> in Biomass Burning Plumes from the WE-CAN 4 Campaign

5 Yingjie Shen<sup>1</sup>, Rudra P. Pokhrel<sup>1, \*</sup>, Amy P. Sullivan<sup>2</sup>, Ezra J. T. Levin<sup>2, \*</sup>, Lauren A. Garofalo<sup>3</sup>,

Delphine K. Farmer<sup>3</sup>, Wade Permar<sup>4</sup>, Lu Hu<sup>4</sup>, Darin W. Toohey<sup>5</sup>, Teresa Campos<sup>6</sup>, Emily V.
 Fischer<sup>2</sup>, Shane M. Murphy<sup>1</sup>

- <sup>1</sup>Department of Atmospheric Science, University of Wyoming, Laramie, WY 82071, USA.
- <sup>9</sup> <sup>2</sup>Department of Atmospheric Science, Colorado State University, Fort Collins, CO 80523, USA
- <sup>3</sup>Department of Chemistry, Colorado State University, Fort Collins, CO 80523, USA
- <sup>4</sup>Department of Chemistry and Biochemistry, University of Montana, Missoula, MT 59812, USA.
- <sup>5</sup>Department of Atmospheric and Oceanic Sciences, University of Colorado Boulder, Boulder, CO 80309, USA
- 13 <sup>6</sup>National Center for Atmospheric Research, Atmospheric Chemistry Division, Boulder, CO 80301, USA

\*now at Air Pollution Control Division, Colorado Department of Public Health and Environment, Denver, CO 80246,
 USA

16 *Corresponding author: Shane M. Murphy (shane.murphy@uwyo.edu)* 

- Abstract. Aerosol absorption of visible light has an important impact on global radiative forcing. Wildfires are one of the major sources of light-absorbing aerosol, but there remains significant uncertainty about the magnitude,
- 19 wavelength dependence, and bleaching of absorption from biomass burning aerosol. We collected and analyzed data
- from 21 Western United States wildfire smoke plumes during the 2018 WE-CAN airborne measurement campaign to
- 21 determine the contribution of black carbon (BC), brown carbon (BrC), and lensing to the aerosol mass absorption
- 22 cross-section (MAC). MAC<sub>BC</sub>, MAC of organics (MAC<sub>BrC+lensing</sub>), Comparison to commonly used parameterizations
- and modeling studies suggest model overestimation of absorption is likely due to incorrect BrC refractive indices.
- 24 Modelers (Wang et al. 2018; Carter et al. 2021) invoke a bleaching process that decreases the MAC of water soluble
- 25 BrC (MAC<sub>ws-BrC660</sub>) are calculated using Photoacoustic Absorption Spectrometer, Single Particle Soot Photometer and
- 26 Particle into Liquid Sampler measurements. MAC<sub>BC660</sub> does not change significantly with physical age, organic
- 27 aerosol (OA) concentration, to offset the overestimation of absorption in models. However, no evidence of decreasing
- 28 MAC is observed in individual WE-CAN fire plumes or in aged plumes from multiple fires. A decrease in OA mass
- and water-soluble organic carbon (WSOC), both normalized by CO to correct for dilution, is observed with increasing
- 30 oxygen to carbon (O:C) ratio (O:C), and decreasing gas-phase toluene:benzene ratio, modified combustion efficiency
- 31 (MCE), altitude, or temperature, and has a relatively stable when data from all fires is combined and in half of
- 32 individual fire plumes. This results in a strong decrease in total absorption at 405 nm and slight decrease at 660 nm
- 33 with these chemical markers. These results demonstrate that changes in absorption with chemical markers of plume
- 34 age are the result of decreasing OA rather than changes in the MAC of the organic material itself. While decreasing
- 35 MAC or OA mass with aging could both be called bleaching, and can both correct overestimation of absorption in
- 36 models, it's important to distinguish these two effects because decreasing OA mass will also decrease scattering, which

average, of 54% of non-BC absorption (23% total absorption) at 660 nm is from water-soluble BrC. MAC as Brecon is 38  $0.06 \pm 0.04 \text{ m}^2 \text{ g}^{-1}$  while MAC<sub>BrC+lensing</sub> is  $0.11 \pm 0.06 \text{ m}^2 \text{ g}^{-1}$  at 660 nm, increasing to  $0.59 \pm 0.19 \text{ m}^2 \text{ g}^{-1}$  at 405 nm. 39 MAC<sub>BrC+lensine</sub> is constant with physical age and MCE, but increases slightly with increasing O:C or decreasing 40 toluene:benzene, while total absorption (normalized to CO) slightly decreases with increasing O:C or decreasing 41 toluene:benzene due to decreasing OA. No evidence of , confirming that BrC bleaching is observed. Comparison to 42 commonly used parameterizations, modeling studies, and the FIREX AQ observations suggest model overestimation 43 44 of absorption is likely due to incorrect BrC refractive indices. absorption is important across the visible spectrum. Ouantification of significant brown carbon BrC in theat red wavelengths and the stability of MAC<sub>BC</sub>, the observation 45 of minimal bleaching, and the observation of being caused by changes in OA with O:C and toluene:benzene markers 46 47 all serve as important of plume age provide important improvements to our understanding of BrC and critical 48 constraints on aerosol absorption in regional and global climate models.

will cause a significantly different net radiative effect. We also find that an average value of  $10.9 \pm 2.1 \text{ m}^2 \text{ g}^4$ . On

#### 49 **1 Introduction**

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50 Atmospheric aerosol impact the climate system by directly scattering and absorbing solar radiation, by 51 indirectly changing cloud properties, and through deposition that changes the surface albedo (McConnell et al., 2007; 52 Sarangi et al., 2020). Biomass burning injects a large amount of primary organic aerosol (POA), secondary organic 53 aerosol (SOA) and black carbon (BC) into the atmosphere every year. BC is somewhat poorly defined, but is generally 54 considered to be insoluble and refractory and includes a variety of materials such as char, biochar, charcoal, elemental 55 carbon (EC), and soot (Wei et al., 2013). Although it only represents a small fraction of aerosol mass, BC has a significant impact on the global energy budget due to its ability to strongly absorb solar radiation- at all visible 56 57 wavelengths. While still important, positive radiative forcing of BC is lower in IPCC AR6 (2022) than in IPCC AR5 (2013). Bond et al. (2013) estimated the direct radiative forcing for BC from 1750 to 2005 at the top of the atmosphere 58 (TOA) to be +0.71 W m<sup>-2</sup>, with an uncertainty of 90% while the latest IPCC AR6 (2022) estimates effective radiative 59 60 forcing for BC from 1750 to 2019 to be +0.11 (-0.2 ~ +0.42) W m<sup>-2</sup>. It is important to note that AR5 reported direct radiative forcing while AR6 reports effective radiative forcing. While BC is emitted from nearly all combustion 61 62 processes, the largest global source of BC is thought to be biomass burning (Bond et al., 2013). Organic aerosol (OA) 63 also absorbs visible light, but its absorption strongly depends on the wavelength of light (Kirchstetter and Novakov, 64 2004). Non-BC light absorbing organic compounds are often called brown carbon (BrC) and they are usually co-65 emitted with BC or formed by secondary chemistry in biomass burning plumes (Andreae and Gelencsér, 2006). Unlike 66 BC, which absorbs light from the UV to the IR, BrC absorption sharply increases in the UV and shorter visible portions 67 of the spectrum and has been historically considered to be almost transparent near the red wavelengths (Andreae and 68 Gelencsér, 2006; Bahadur et al., 2012; Liu et al., 2020). The global-mean TOA direct radiative forcing from BrC also shows a large uncertainty, with estimates ranging from +0.03 W m<sup>-2</sup> to +0.57 W m<sup>-2</sup> (Saleh, 2020). Wildfires in the 69 70 Western U.S. have increased in recent decades (Westerling et al., 2006; Burke et al., 2021), and will continue 71 increasing according to model predictions (Yue et al., 2013; Hurteau et al., 2014; Ford et al., 2018; Neumann et al.,

2021). Therefore, quantitative studies of the radiative effects caused by BC and BrC emitted from wildfires are crucial
 for a better understanding of future climate and essential to improve climate models.

74 The large uncertainty in the radiative forcing from BC is caused both by uncertainties in emissions and by 75 uncertainty in properties that affect its optics, such as size distribution, morphology, refractive index, and mixing state 76 (Bond et al., 2006; Kleinman et al., 2020; Brown et al., 2021). For wildfires, most of the aerosol mass is organic 77 (Garofalo et al., 2019). When BC is internally mixed with OA, the BC is coated by other absorbing or non-absorbing 78 materials that cause more photons to interact with the BC core, and therefore enhance the absorption of the BC core. 79 This process is often called the lensing effect even though geometric lensing is not actually happening at these sizes 80 (Fuller et al., 1999). The absorption enhancement caused by the lensing effect is defined as the ratio of the absorption 81 cross-section of a coated BC particle to that of an equivalent uncoated BC particle (Lack and Cappa, 2010). Laboratory 82 experiments have shown a strong absorption enhancement of BC by a factor of two or more (Schnaiter et al., 2003; 83 Schnaiter et al., 2005; Bond and Bergstrom, 2006; Bond et al., 2006; Peng et al., 2016). While Observations of 84 absorption enhancement from ambient BC vary widely in field studies due to variations in coating thickness, coating 85 material, source type, or methodological differences, but it is often much lower than laboratory values (Liu et al., 2015, 86 2017; Cappa et al., 2012, 2019; Healy et al., 2015; Krasowsky et al., 2016). Cappa et al. (2019) summarized absorption 87 enhancements observed at the red end of the visible spectrum from 10 studies including ambient measurements, source 88 sampling, and lab experiments. The absorption enhancement reported by those measurements ranged from 1.1 to 2.8.

89 Three methods (referred to henceforth as core shell Mie theory, thermal denuder, and mass absorption cross-90 section) can be used to obtain estimates of absorption enhancement. Numerical solutions to Mie theory (Bohren and Huffman, 1983) have been used to model aerosol absorption for many years and can be adapted into a core shell 91 92 model, which is a simplified version of the complex mixing states in real atmospheric particles (Chylek et al., 2019). 93 The core shell model assumes particles are concentric spheres, where BC acts as a core and the other materials 94 (typically organics) act as a shell. This model can simulate the absorption enhancement with geometric and optical inputs (i.e., shell thickness, particle radius, refractive index). However, the assumptions made by Mie theory may have 95 significant errors for irregular particles, often found in fresh soot particles and when mixed BC and organics are not 96 concentric spheres. A thermodenuder can be used to remove volatile coating materials by heating them to a temperature 97 from 250-400 °C. The ratio of the absorption coefficient measured in ambient air and measured after passing particles 98 99 through the thermodenuder gives an empirical absorption enhancement. Liu et al. (2015) utilized a thermodenuder to find the average absorption enhancement at 405 nm and 781 nm to be 1.3 and 1.4, respectively, in the UK during 100 winter. Pokhrel et al. (2017) utilized a photoacoustic absorption spectrometer (PAS) with a thermodenuder and showed 101 that absorption enhancement determined in this manner depends on fuel type and combustion conditions, with 102 absorption enhancements ranging from 0.92 to 1.43 at 660 nm and reaching a maximum of 5.6 at 405 nm. However, 103 104 a thermodenuder cannot always remove coating materials completely and thus can lead to underestimates of absorption 105 enhancement. The mass absorption cross section of BC (MAC<sub>BC</sub>) is another way to describe the absorbing ability of 106 BC containing particles by describing the absorption per unit mass of BC. The mass absorption cross section of BC 107 (MAC<sub>BC</sub>) is a different way to describe the absorbing ability of BC containing particles versus absorption enhancement. 108 By describing the absorption per unit mass of BC, MAC<sub>BC</sub> can be a fundamental input in climate models to convert

- 109 mass concentration into absorption coefficients (Cho et al., 2019). MAC<sub>BC</sub> is the particulate absorption divided by the
- 110 mass of the pure BC at the same wavelength. In this way, the calculated  $MAC_{BC}$  will include absorption of the BC
- 111 core, along with the absorption and absorption enhancement caused by the coating material. Unfortunately, the MAC
- 112 of the overall BC particle, MAC<sub>BC<sub>a</sub></sub> in the ambient atmosphere continues to be not wellpoorly understood due to thea
- 113 lack of field measurements and instrumental limitations. The of filter-based instruments to measure this parameter.
- 114 <u>Processes that occur during atmospheric aging processes on of BC canalso</u> introduce uncertainties on in its absorption.
- Bond and Bergstrom (2006) suggested a MAC<sub>BC</sub> of  $7.5\pm1.2 \text{ m}^2 \text{ g}^{-1}$  at 550 nm for fresh BC. Subramanian et al. (2010)
- 116 reported a MAC<sub>BC</sub> of  $10.9\pm2.1 \text{ m}^2 \text{ g}^{-1}$  at 660 nm and 13.1 m<sup>2</sup> g<sup>-1</sup> at 550 nm over Mexico City when using a single
- 117 particle soot photometer (SP2) and the filter-based particle soot absorption photometer (PSAP) instrument during
- airborne measurements. Krasowsky et al. (2016) reported a MAC<sub>BC</sub> enhancement of  $1.03\pm0.05$  due to the coatings on
- BC. Zhang et al. (2017) found a MAC<sub>BC</sub> with a mean of 10 m<sup>2</sup> g<sup>-1</sup> and a standard deviation of 4 m<sup>2</sup> g<sup>-1</sup> at 660 nm by
- using both SP2 and PSAP measurements. Cho et al. (2019) summarized  $MAC_{BC}$  estimated from more than 10 studies
- in East and South Asia in both ambient conditions and laboratory experiments, and the values ranged from 4.6 to 11.3
- $122 m^2 g^{-1}$ .

123 The limitations of current measurement techniques bring major uncertainty into quantifying BrC absorption, because BrC is usually co-emitted with BC which makes it challenging to measure BrC absorption independently. 124 125 BrC absorption can be directly measured through the solvent-extraction method (Peltier et al., 2007; Zeng et al., 2021; Sullivan et al., 2022) or a thermodenuder (Cappa et al., 2012; Liu et al., 2015; Pokhrel et al., 2017). However, the 126 127 solvent-extraction method will miss BrC that's insoluble in water or organic solvents, and thermal denuders miss BrC 128 that is not volatile at the denuder temperature. BrC absorption can also be calculated from multi-wavelength total 129 absorption measurements, but this approach must assume the absorption Ångström exponent (AAE) for BC and assumes that BrC does not absorb at longer wavelengths, adding significant uncertainty. 130

131 To improve the understanding of the evolution of light-absorbing aerosol from biomass burning, smoke from 21 wildfires in the Western United States were measured near their sources and downwind onboard the NSF/NCAR 132 133 C-130 aircraft during the Western Wildfire Experiment for Cloud Chemistry, Aerosol Absorption and Nitrogen (WE-134 CAN) campaign. This campaign represented one of the firstan airborne attemptsattempt to fully characterize Western 135 U.S. wildfires from several different fuel types, locations, and fire stages (flaming vs. smoldering). This paper presents 136 novel observations about the absorbing properties of the aerosol and compares these observations to modeling studies conducted with the WE-CAN data and to results from the Fire Influence on Regional to Global Environment - Air 137 138 Quality (FIREX) study conducted in 2019 (Zeng et al., 2021).

# 139 2 Experimental Method

140 This work relies on measurements made during the WE-CAN field campaign, which sampled smoke emitted 141 by wildfires across the Western U.S. using the NSF/NCAR C-130 research aircraft. The goal of the campaign was to 142 make detailed observations of the physical, chemical, and optical evolution of aerosol in western wildfire smoke and 143 its impact on climate, air quality, weather, and nutrient cycles. The WE-CAN field campaign consisted of 19 research 144 flights that took place from Jul. 24 – Sep. 13, 2018. Data from 13 flights where all required instrumentation was

- 145 <u>available</u> were analyzed in this study. The flight path and dominant wildfire for each of these flights are shown in Fig.
- 146 1. The fire locations, fuel types for each fire during WE-CAN were characterized and summarized by Lindaas et al.

147 <u>(2021).</u>

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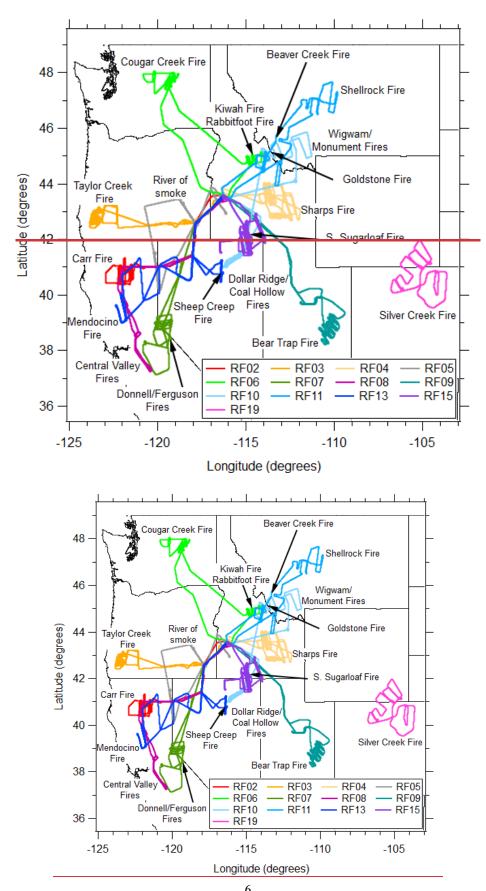


Figure 1: Flight paths and the sampled wildfires for the WE-CAN flights analyzed in this paper.

# 149 2.1 Instrumentation

The following instruments are a subset of those flown during the WE-CAN campaign and are utilized in this work. The full WE-CAN dataset is archived at <u>https://data.eol.ucar.edu/master\_lists/generated/we-can</u>. All aerosol instruments utilized in this paper, except the PILS, <u>used-pulled air from the same Solid Diffuser Inlet (SDI)</u> inlet. The PILS sampled from a Submicron Aerosol Inlet (SMAI) (Craig et al., 2013a, 2013b, 2014; Moharreri et al., 2014). <u>All</u> the measurements were converted to standard temperature and pressure (STP, 1 atm, 0°C) based on the measured temperature and pressure (Eq. 1) before data were uploaded.

156  $Variables_{STP} = Variables_{measured} \cdot \frac{Pressure_{STP}}{Pressure_{measured}} \cdot \frac{Temperature_{measured}}{Temperature_{STP}}$  (Eq. 1)

#### 157 **2.1.1** Photoacoustic Absorption Spectrometer (PAS)

158 Aerosol absorption coefficients were measured with the multi-wavelength PAS built by the University of 159 Wyoming (Foster et al., 2019), based on the design of Lack et al. (2012b). A PAS can directly measure the absorption 160 coefficient of dry aerosol. The PAS represents the only way to directly measure aerosol absorption other than thea photothermal interferometer (PTI, Sedlacek, 2007), which measures the change in the refractive index of the air near 161 162 particles caused by heating from absorption. Briefly, when modulated laser light (at the resonant frequency of the cell) 163 is absorbed by the aerosol<sub> $\tau$ </sub> it heats the surrounding air inducing pressure waves that are-ereated and amplified by the 164 cavity then detected by two microphones (Lack et al., 2006; Foster et al., 2019). The PAS used here has four cells that 165 measure the aerosol absorption coefficient from dry air at 405 nm and 660 nm and thermally denuded air at 405 nm 166 and 660 nm. The denuder was set to 300°C, with the goal of evaporating volatile organic aerosol which might have a 167 potential impact on light absorption. However, absorption from the denuded channels was not used in this study, 168 because the absorption enhancement calculated using the thermodenuder approach iswas much smaller than the approach taking the ratio of MAC<sub>BC</sub> to MAC<sub>BC-core</sub>, and we believe the discrepancy is due to the presence of significant 169 170 residual organic material after denuding. Two NO<sub>x</sub> denuders coated with potassium hydroxide, guaiacol and methanol 171 were installed on the PAS in front of the inlet to remove the absorption from gas-phase NO<sub>2</sub> (Williams and Grosjean 172 1990). No evidence of  $NO_2$  absorption (which would cause baseline shifts) was observed during filter measurements 173 that are acquired every few minutes. A 3 LPM PM<sub>2.5</sub> cyclone (URG-2000-30ED) was used on the PAS in front of the 174 inlet to provide a PM<sub>1.0</sub> cut-<u>under a total flow rate of 5.7 LPM</u>. In addition, a Nafion drier (Purma Pure PD-100T-175 24MPS) with 100 tubes was installed on the inlet system to dry sample to a relative humidity below 30%. The particle 176 loss (< 3%) in the drier was corrected during post-processing. The uncertainty in the absorption coefficient measured 177 by the PAS mainly comes from the calibration technique, in which the highly absorbing substance Regal Black and 178 the CAPS PM<sub>SSA</sub> were utilized (Foster et al., 2019). The PAS was routinely calibrated (after each flight or every other 179 day if there was a flight everyday) during WE-CAN with an accuracy of +/- 10%. 180 The PAS microphone shows a pressure-dependent response to pressure. To account for this behavior, we

performed pressure-dependent calibration of the PAS where the instrument pressure (both PAS and CAPS  $PM_{SSA}$ ) was dropped stepwise by ~50 torr from ambient to ~300 torr (typical minimum pressure level during WE-CAN). A

183 calibration was performed at each pressure step and the calibration constants were fitted with pressure to get a change

in calibration at a desired pressure. Pressure-dependent calibrations were repeated pre and post-campaign to capturevariability.

- 186 Aerosol optical properties (absorption and extinction) were converted to standard temperature and pressure
- 187 (STP, 1 atm, 0°C) before data were uploaded. We used temperature and pressure measured by the PAS and CAPS
- 188 PM<sub>SSA</sub> to convert optical properties to STP. For absorption, the sample temperature measured by the PAS RH sensor
- 189 (Vaisala RH probe) and the pressure measured by the temperature and pressure sensor of the CAPS PM<sub>SSA</sub> were used.
- 190 Whereas for extinction, temperature and pressure measured by the CAPS PM<sub>SSA</sub> were used.

#### 191 2.1.2 Cavity-Attenuated Phase Shift Spectrometer (CAPS PM<sub>SSA</sub>)

After pulling through the NO<sub>X</sub> denuder, the PM<sub>1.0</sub> cyclone, and the Nafion drier in front of the PAS inlet, the sampled air entered through the Aerodyne CAPS  $PM_{SSA_{450}}$  and CAPS  $PM_{SSA_{660}}$  to measure the aerosol scattering and extinction coefficients at 450 nm and 660 nm, respectively. CAPS  $PM_{SSA}$  instruments measure extinction by utilizing the cavity attenuated phase shift spectroscopy and measure scattering with an integrating sphere (Onasch et al., 2015). Ammonium sulfate particles were used to calibrate the scattering channel of the CAPS  $PM_{SSA}$  during WE-CAN with an accuracy of +/- 3%.

# 198 2.1.3 Particle-into-Liquid Sampler (PILS) systems

199 BrC absorption and water-soluble organic carbon (WSOC) were measured by a Particle-into-Liquid Sampler (PILS) system (Sullivan et al., 2022). The PILS continuously collects ambient particles into purified water and 200 201 provides a liquid sample with the aerosol particles dissolved in it for analysis (Orsini et al., 2003). The size-cut for the 202 PILS was provided by a nonrotating microorifice uniform deposit impactor (MOUDI) with a 50% transmission 203 efficiency of 1 µm (aerodynamic diameter) at 1 atmosphere ambient pressure (Marple et al., 1991). The total airflow 204 for the PILS was approximately 15 LPM. Upstream of the PILS was an activated carbon parallel plate denuder 205 (Eatough et al., 1993) to remove organic gases. In addition, a valve was manually closed periodically for 10 min 206 diverting the airflow through a Teflon filter before entering the PILS allowing for background measurements. The 207 liquid sample obtained from the PILS was pushed through a 0.2 µm PTFE liquid filter by a set of syringe pumps to 208 ensure insoluble particles were removed. The flow was then directed through a liquid waveguide capillary cell (LWCC) 209 and Total Organic Carbon (TOC) Analyzer for near real-time measurement of BrC absorption and WSOC, 210 respectively. More details and a schematic illustration can be found in Zeng et al. (2021).

For the absorption measurement, a 2.5 m path-length LWCC (World Precision Instruments, Sarasota, FL) was used. A dual deuterium and tungsten halogen light source (DH-mini, Ocean Optics, Largo, FL) and absorption spectrometer (FLAME-T-UV-VIS, Ocean Optics, Largo, FL) were coupled to the LWCC via fiber optic cables. Absorption spectra were recorded using the Oceanview Spectroscopy Software over a range from 200 to 800 nm. The wavelength-dependent absorption was calculated following the method outlined in Hecobian et al. (2010). For this study, a 16 s integrated measurement of absorption with a limit of detection (LOD) of 0.1 Mm<sup>-1</sup> was obtained (Sullivan et al., 2022). For the WSOC measurement, a Sievers Model M9 Portable TOC Analyzer (Suez Waters Analytical Instruments, Boulder, CO) was used. This analyzer works by converting the organic carbon in the liquid sample to carbon dioxide through chemical oxidation involving ammonium persulfate and ultraviolet light. The carbon dioxide formed was then measured by conductivity. The increase in conductivity observed was proportional to the amount of organic carbon in the liquid sample. The analyzer was run in turbo mode providing a 4 s integrated measurement of WSOC with a LOD of  $0.1 \ \mu g \ C/m^3$  (Sullivan et al., 2022).

#### 224 **2.1.4 Single Particle Soot Photometer (SP2)**

225 Refractory black carbon (rBC) number and mass concentrations were measured with a Single Particle Soot 226 Photometer (SP2; Droplet Measurement Technologies) which uses a continuous, 1064 nm Nd:YAG laser to heat 227 absorbing material, primarily rBC, to its vaporization temperature and measures the resulting incandescence (Schwarz 228 et al., 2006). Similar to the CAPS PM<sub>SSA</sub>, the sampled air was sent through the NO<sub>X</sub> denuder, PM<sub>1.0</sub> cyclone, and 229 Nafion drier in front of the PAS inlet before it went to the SP2. The SP2 was calibrated with PSL and size-selected 230 fullerene soot. On the C-130, the SP2 sample line was diluted with HEPA-filtered, pressured ambient air that was 231 passed through a mass flow controller to prevent signal saturation. During post-processing the data was corrected for 232 dilution back to ambient concentrations-then to STP.

# 233 2.1.5 Ultra-High Sensitivity Aerosol Spectrometer (UHSAS)

Particle number concentration was measured by a rack-mounted Ultra-High Sensitivity Aerosol Spectrometer (UHSAS). The flow rate of the rack-mounted UHSAS can be manually lowered by the in-flight operator when the aircraft flew across smoke plumes, so that the UHSAS can stay within its optimum concentration measurement range (Sullivan et al., 2022). The UHSAS was calibrated with ammonium sulfate. The particle mass concentration was calculated by applying these size bins and multiplying by a particle density of 1.4 g cm<sup>-3</sup> (Sullivan et al., 2022). The volume mean diameter of the particles for all the detected plumes range between 0.18 µm and 0.34 µm.

#### 240 2.1.6 Proton-Transfer-Reaction Time-of-Flight Mass Spectrometer (PTR-ToF-MS)

The University of Montana proton-transfer-reaction time-of-flight mass spectrometer (PTR-ToF-MS 4000, Ionicon Analytik) was utilized to report the VOC mixing ratios during WE-CAN (Permar et al., 2021). Only the toluene and benzene mixing ratio derived from the PTF-ToF-MS were used in this work; their overall uncertainty is <15%. More details of the operation, calibration, and validation on the PTR-ToF-MS during WE-CAN can be found in Permar et al. (2021).

#### 246 2.1.7 High-Resolution Aerosol Mass Spectrometry (HR-AMS)

Organic aerosol (OA) was detected by the high-resolution aerosol mass spectrometry (HR-AMS; Aerodyne Inc.). The description of the AMS operation during WE-CAN can be found in Garofalo et al. (2019). The atomic oxygen-to-carbon ratios (O:C) and organic mass-to-organic carbon ratio (OM:OC) used in this work were determined via the improved ambient elemental analysis method for the AMS (Canagaratna et al., 2015). Average (integrated) elemental ratios were obtained by averaging (integrating) elemental masses of carbon, hydrogen, and oxygen and recalculating elemental ratios.

#### 253 2.1.8 Quantum Cascade Laser (QCL) and Picarro Cavity Ring-Down spectrometer (Picarro)

The carbon monoxide (CO) mixing ratio was measured by both an Aerodyne quantum cascade laser instrument (CS-108 miniQCL) and a Picarro cavity ring-down spectrometer (G2401-m WS-CRD) (Garofalo et al., 2019). Because the QCL has better precision than the Picarro instrument, CO measurements from the QCL were preferentially used. However, CO measurements from the Picarro CO data were used for RF10 and RF13, because the CO-QCL was not operated during those two flights. The carbon dioxide (CO<sub>2</sub>) mixing ratio was also determined from the Picarro.

#### 260 2.2 Plume Physical Age

The physical age of the plume was calculated by dividing the distance the plume was sampled from the fire source by the in-plume average wind speed. The average wind speed was measured on the NSF/NCAR C-130 aircraft during each plume pass. The distance was estimated by using the longitude and latitude of the geometric center of the plume measured on the NSF/NCAR C-130 and the fire location provided by the U.S. Forest Service. <u>TheseThe</u> same plume ages weremethod was used by Garofalo et al. (2019), Peng et al. (2020), Lindaas et al. (2021), Permar et al. (2021), and Sullivan et al. (2022).(2022) and are also utilized here for consistency.

#### 267 2.3 Plume Integration Method

268 During the WE-CAN campaign, both the SP2 and PILS had a-significant hysteresis compared to other 269 instruments. In the SP2 this is because the sampled air was diluted with particle-free ambient air at various ratios to 270 prevent signal saturation. In the PILS this is because of the retention effect of liquid on the wetted component or within 271 dead volumes (Zeng et al., 2021). Therefore, it was most accurate to integrate properties across airborne transects of 272 wildfire plumes to avoid the impact of instrument hysteresis and measurement noise that can dramatically impact 273 instantaneous ratios. Pseudo-Lagrangian sampling was used during the flights for the WE-CAN campaign, the C-130 274 aircraft repeatedly crossed the smoke plume from a particular fire by traveling perpendicular to the prevailing winds, 275 crossing the plume, turning, then crossing the plume again further downwind. In this work, we manually identified 276 plume edges based on the inflection point when CO concentrations stopped rapidly changing as we entered and exited 277 the smoke plume. The outside of plume measurement periods had CO mixing ratios from 100 - 300 ppbv. The lowest 278 10% of each variable from outside plume segments were set to be the background of that variable. If the time between 279 two consecutive outside plume segments was larger than 20 s and the highest CO mixing ratio was 100 ppbv higher 280 than the outside plume CO criteria, this segment was chosen as a plume. The start and end point of each plume was 281 slightly adjusted manually based on the CO mixing ratio to make sure the entire plume was covered. A different start 282 and end point for the SP2 and PILS was adjusted manually based on the rBC mass concentrations and WSOC, 283 respectively.

#### 284 2.4 Absorption Enhancement and Mass Absorption Cross-section

Absorption enhancement  $(E_{abs})$  is the ratio of the absorption of the whole particleall particles (including BC core and coating materials) to the absorption of the BC corealone (Lack and Cappa, 2010).  $E_{abs}$  at a specific wavelength  $(E_{abs \lambda})$  was calculated in this study by Eq. <u>12</u>:

$$E_{abs_{\lambda}} = \frac{\beta_{Total_{\lambda}}}{M_{BC} * MAC_{BC\_core_{\lambda}}} \frac{Abs_{Total_{\lambda}}}{M_{BC} * MAC_{BC\_core_{\lambda}}}$$
(Eq. 12)

where  $\beta_{Total_{\lambda}}Abs_{Total_{\lambda}}$  is the total absorption coefficient at a wavelength of  $\lambda$  nm measured by the PAS,  $M_{BC}$  is the mass concentration of BC measured by the SP2, and  $MAC_{BC\_core\_\lambda}$  is the MAC of BC <u>corealone</u> (without any other coating material) at  $\lambda$  nm, which is set to be 6.3 m<sup>2</sup> g<sup>-1</sup> at 660 nm (Bond and Bergstrom, 2006; Subramanian et al., 2010).

292 MAC<sub>BC</sub> at  $\lambda$  nm was calculated following Eq. 23:

0

293 MAC<sub>BC λ</sub>

$$294 = \frac{\beta_{Total_{\lambda}}}{M_{BC}} \frac{Abs_{Total_{\lambda}}}{M_{BC}}$$
(Eq. 23)

295 MAC<sub>BC</sub> is utilized more often in this study than  $E_{abs}$  because there is not a widely accepted MAC for BC 296 emitted from wildfire. MAC of BrC and lensing is calculated at 405 and 660 nm (Eq. 34):

297  $MAC_{BrC+lensing_{\lambda}}$ 

$$298 = \frac{\beta_{Total_{\lambda}} - M_{BC} * MAC_{BC\_core_{\lambda}}}{M_{OA}} Abs_{Total_{\lambda}} - M_{BC} * MAC_{BC\_core_{\lambda}}}{M_{OA}}$$
(Eq. 34)

where  $M_{0A}$  is the organic mass measured by the AMS. Again, the <u>MAC of the BC-MAC<sub>BC core  $\lambda}</u> is set to be 6.3 and$ 10.2 <u>m<sup>2</sup> g<sup>-1</sup></u>, respectively, at 660 nm and 405 nm yielding an absorption Ångström exponent (AAE, the negative slopeof a logarithmic absorption coefficient against wavelength) of 0.99 for the BC core (Bond and Bergstrom, 2006;Subramanian et al., 2010; Liu, et al., 2015). It should be noted that both BrC and lensing contribute to the MAC<sub>BrC+lensing</sub>, $and cannot be separated using this approach. MAC of water-soluble BrC at <u>660</u><math>\lambda$  nm (MAC<sub>ws\_BrC\_660</sub> $\lambda$ ) is calculated using Eq. 4<u>5</u>:</u></sub>

$$305 \qquad MAC_{ws\_BrC\_660} = \frac{\beta_{ws\_BrC_660}}{WSOC} - MAC_{ws\_BrC\lambda}$$

$$306 = \frac{Abs_{ws\_BrC\_\lambda}}{WSOC * (WSOM: WSOC)} \qquad (Eq. 45)$$

$$307 \qquad \text{where } \beta_{ws\_BrC_660}Abs_{ws\_BrC\_660} \text{ is water-soluble light absorption and WSOC is water-soluble organic carbon mass,}$$

$$308 \qquad \text{which are both measured by the PILS system. WSOM: WSOC ratio is set to be 1.6 (Sullivan et al., 2022).}$$

$$309 \qquad \text{To investigate which contributes more to absorption enhancement at 660 nm, the absorption from BrC or the}$$

$$310 \qquad \text{lensing effect, the fractional absorption from BrC at 660 nm is calculated by Eq. 5}$$

$$311 \qquad Fractional Abs_{BrC} = \frac{\beta_{BrC\_660}}{MSOC} \qquad (Eq. 5)$$

$$\frac{1}{\beta_{Total_{660}} - M_{BC} * MAC_{BC_{core_{660}}}}$$

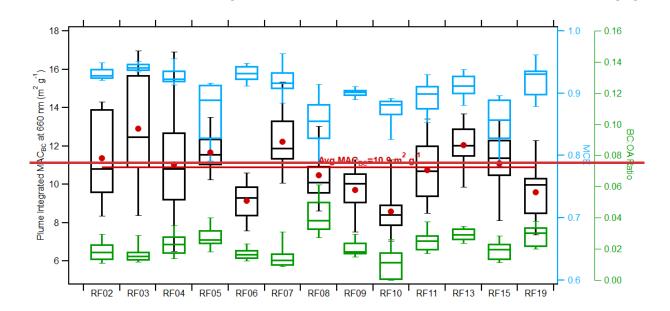
312 where  $\beta_{HTC.660}$  is the total BrC absorption coefficient at 660 nm. This is calculated from the water-soluble light 313 absorption provided by the PILS, where we convert absorption from water soluble BrC to total BrC by 314 WSOM:WSOC and OM:WSOM ratio, and correct absorption from liquid phase to particle phase via Mie 315 theory (more details in 3.1.4, Eq. 9-10). 2.5 Fractional non-BC Absorption from BrC 316 This approach assumes that water insoluble BrC has the same refractive index as water soluble BrC. This assumption would provide a lower estimation on the BrC contribution to the total absorption because Sullivan et al. (2022) found 317 that 45% of the BrC absorption at 405 nm in WE-CAN came from water soluble species, and Zeng et al. (2022) found 318 that insoluble BrC absorbs more at higher wavelengths than soluble BrC, and methanol insoluble BrC chromophores 319 caused 87% of the light absorption at 664 nm.  $\beta_{Total 660}$  is the total absorption coefficient at 660 nm which is 320 measured by the PAS, M<sub>RC</sub> is the mass concentration of BC which is measured by the SP2, and MAC<sub>RC core soft</sub> is the 321 MAC of the BC core at 660 nm which is set to be 6.3 m<sup>2</sup> g<sup>-4</sup> (Bond and Bergstrom, 2006; Subramanian et al., 2010). 322 323 2.5 Modified Combustion Efficiency (MCE) The variation of burn condition (e.g., flaming vs. smoldering) and fuel type can cause a significant difference 324 325 in BC emissions and changes in aerosol properties (Akagi et al., 2011; Andreae, 2019). Burn conditions can be estimated with the modified combustion efficiency (MCE), defined as Eq. 6: 326  $MCE = \frac{\Delta CO_2}{\Delta CO + \Delta CO_2}$ 327 (Eq. 6)where  $\Delta CO_2$  and  $\Delta CO$  are the background subtracted  $CO_2$  and CO mixing ratio. The background of  $CO_2$  and CO328 329 mixing ratio is obtained via the same process described in Section 2.3. 330 2.6 Absorption, Scattering and Single Scattering Albedo (SSA) 331 Plume integrated absorption and scattering were normalized (x/CO) by taking the ratio of backgroundsubtracted absorption or scattering ( $\Delta x$ ) to the background subtracted CO mixing ratio ( $\Delta CO$ ) (Eq. 7), so that the 332 changing of the normalized properties is not impacted by dilution of the plume with background air. 333  $x/CO = \frac{\Delta x}{\Delta CO}$ 334 (Eq. 7) 335 Although the CAPS PM<sub>SSA</sub> provides scattering directly, the scattering is only accurate if extinction is below 336 1000 Mm<sup>+</sup> (Onasch et al., 2015). We calculated scattering by subtracting absorption measured by the PAS from extinction measured by the CAPS PM<sub>SSA</sub> to avoid high uncertainty caused by extremely dark plumes. Similarly, SSA 337 338 at a specific wavelength ( $\lambda$ ) was also calculated by using both the PAS absorption ( $\beta_{Tatat,\lambda}$ ) and CAPS PM<sub>SSA</sub> 339 extinction (Ext.) (Eq. 8).  $SSA_{\lambda} = \frac{Ext_{\lambda} - \beta_{Total_{\lambda}}}{Ext_{T}}$ 340 (Eq.8)

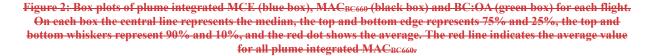
# 341 **3 Results and Discussion**

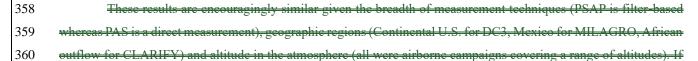
#### 342 3.1 Absorption of BC and BrC at Red Wavelengths

#### 343 3.1.1 Mass Absorption Cross-Section of Black Carbon at 660 nm

344 Plume integrated MCE, MAC<sub>BC</sub>-at 660 nm (MAC<sub>BC660</sub>) and BC:OA ratio from the 13 WE CAN research flights with clear plume transects of biomass burning plumes are shown in Fig. 2. Even fire plumes from individually 345 named fires are usually a mix of many different burning conditions, and it is hard to identify the exact source in most 346 wildfire smoke measurements, especially for well mixed plumes. Therefore flight to flight data is analyzed in this 347 348 study because each flight covered a region, and an overall behavior of absorbing aerosol from wildfire can be provided. MAC<sub>BC660</sub> varies between different flights with RF03 having the highest average MAC<sub>BC660</sub> of 12.9 m<sup>2</sup> g<sup>-1</sup>. (median 349 MCE of 0.94, median BC:OA of 0.015) and RF10 having the lowest average MAC<sub>BC660</sub> of 8.6 m<sup>2</sup>-g<sup>-1</sup>, (median MCE 350 median BC:OA of 0.011). The average of all plume integrated MAC<sub>BC666</sub> is 10.9 m<sup>2</sup> e<sup>4</sup>, with a standard 351 of 0.88. deviation of 2.1 similar to some other recent airborne measurements. Subramanian et al. (2010) 352 353 m<sup>2</sup> g<sup>+</sup> using a SP2 and PSAP operated during the MILAGRO campaign, which MAC<sub>BC660</sub> of 10.9 reported a included airborne measurements for biomass fires over Mexico. Similarly, Zhang et al. (2017) estimated a MACRECON 354 355 of 10 m<sup>2</sup> g<sup>+</sup> utilizing both SP2 and PSAP deployed on the NASA DC 8 research aircraft for the DC3 campaign, which measured the upper tropospheric BC over the central U.S. Taylor et al. (2020) calculated a MAC<sub>BC655</sub> of  $12 \pm 2$  m<sup>2</sup>g<sup>-1</sup> 356 over the southeast Atlantic Ocean, using airborne measurements from a SP2 and PAS in the CLARIFY 2017 campaign. 357







361 we apply 6.3 m<sup>2</sup> g<sup>-1</sup> as the MAC of a BC core at 660 nm (Bond and Bergstrom, 2006; Subramanian et al., 2010), then 362 the average absorption enhancement for the entire campaign is 1.7. This means the absorption of coated BC is 1.7 363 times higher than bare BC at 660 nm, which is close to the factor of -2 reported by laboratory experiments (Schnaiter 364 et al., 2005; Peng et al., 2016), larger than some field measurements (Cappa et al., 2012&2019; Healy et al., 2015), 365 but close to 1.85 ± 0.45 measured by Taylor et al. (2020) in African biomass burning plumes. The similarity to the 366 Taylor et al. (2020) result suggests global similarities in the MAC<sub>BC660</sub> from aerosol emitted from wildfires.

- 367 The variation of MCE in different flights is caused by the different fuel sources and burning characteristics 368 of the measured fires. Fires with high MCE tend to have more flaming combustion while those with lower MCE tend 369 to have more smoldering combustion. Because of this, fires with different MCE may produce different coating material 370 and thus changing MAC<sub>BC660</sub>. MCE can vary in the same flight (Fig. 2), for example RF05 (mixture of multiple fire 371 sources) and RF15 (single fire sources), because multiple fires were measured in some flights. Several other factors, 372 such as physical age and chemical age (discussed later) also may impact MAC<sub>BC660</sub>, but we first investigate if there 373 was a clear relationship between MCE and MAC<sub>BC660</sub>.
- 374 The comparison between plume integrated  $MAC_{BC660}$  and MCE is shown in Fig. S1. No clear relationship 375 between  $MAC_{BC660}$  and MCE can be seen from individual flights or amongst all the flights combined ( $R^2$ =0.01). This 376 result indicates that the combustion conditions (flaming or smoldering) does not have an easily described relationship 377 to  $MAC_{BC660}$ . This poor relation is similar to the relationship observed by Pokhrel et al. (2016). This lack of 378 relationship is likely due to the difficulty for MCE to predict aerosol properties such as BC:OA (Grieshop et al., 2009), 379 upon which effective organic aerosol absorptivity highly depends (Saleh et al., 2014).

#### 380 **3.1.2 Relationship of Bulk Optical Properties at 660 nm to Physical Age**

381 Figure 3 shows the evolution of MAC<sub>BC660</sub>, SSA, scattering and absorption at 660 nm versus the time since 382 emission, which will be referred to from now on as physical age. While some flights only intercepted fresh plumes 383 (for example, RF03 and RF15, with a physical age less than 200 min), others intercepted relatively old plumes (for example, RF02 and RF11, with a physical age of 600 min and 800 min, respectively). To eliminate the influence of 384 385 dilution of particles with time, scattering and absorption are normalized by taking the ratio of the enhancement of 386 scattering or absorption above the background to the enhancement of CO concentration above the background. CO is 387 a conserved tracer that does not react on timescales relevant to these observations. No clear trend between MACBC660 388 and physical age is apparent from individual flights (Fig. 3a), linear fitting of all flights combined (black solid line) 389 gives a slope close to zero demonstrating that the absorption enhancement changes little with physical age, which is 390 consistent with the results from Subramanian et al. (2010).

391 However, SSA at 660 nm (SSA<sub>660</sub>) shows a slightly increasing trend with physical age (Fig. 3b) increasing 392 from 0.94 to 0.97 in 10 hours, though the correlation is not very strong with a  $R^2$ -of 0.14. The increase of SSA<sub>660</sub> is 393 partly caused by the increase of scattering at 660 nm (Fig. 3c). The particle size increases with age (Fig. S2) due to 394 coagulation of small particles and condensation of vapors. The volume mean diameter of the particles increased from 395 on average 0.18 µm to 0.34 µm across all the plumes detected. Even for each individual flight, the increasing trend in 396 particle mean diameter is clear. Another contributor to increasing SSA is the decrease in absorption at 660 nm (Fig. 397 3d) with age for most fires. Overall, the trends in SSA, absorption, and scattering with physical age are quite weak

398 with poor correlation coefficients.

399

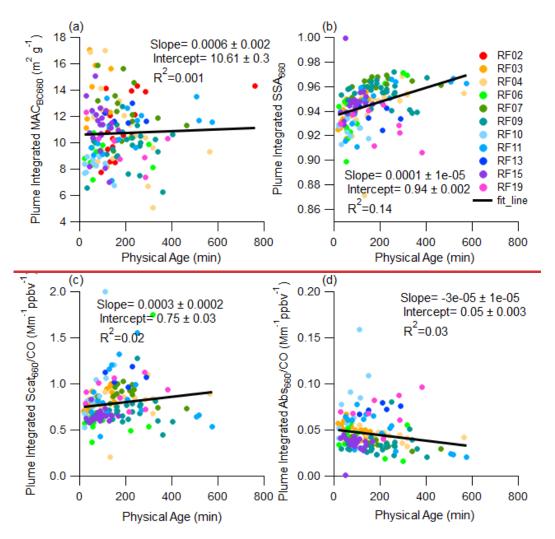


Figure 3: Time evolution of plume integrated (a) MAC<sub>BC660</sub>, (b) SSA<sub>660</sub>, (c) seattering, and (d) absorption at 660 nm.

### 400 3.1.3 Response of Optical Properties at 660 nm to Markers of Chemical Oxidation

401 MAC<sub>BC660</sub> was next compared to chemical markers of oxidation, since physical age did not provide strong 402 correlations and often does not do a good job of representing oxidation and photochemical reactions that occur in 403 plumes. Organic coatings of BC cores may be removed via these reactions, or new species may condense on the BC 404 core during chemical reactions, both of which would further change the optical properties of BC. Figure 4 shows the 405 evolution of plume integrated MAC<sub>BC660</sub> versus chemical clocks based on (a) the ratio of gas phase toluene:benzene 406 and (b) the particle phase oxygen to carbon (O:C) ratio. The toluene:benzene ratio decreases with photochemical 407 processing time since toluene is more reactive than benzene (Gouw et al., 2005), while The O:C ratio characterizes 408 the oxidation state of OA and typically increases with photochemical age (Aiken et al., 2008).

As shown in Fig. 4 the toluene:benzene ratio ranges from 0.33 to 0.88 across all flights while the O:C ratio 409 is between 0.35 and 0.72. It is difficult to discern any pattern of MAC<sub>BC660</sub> changing with either marker in an individual 410 flight. However, the trends are slightly clearer after combining plumes from all the flights. From the linear fit line 411 (black solid line) to data from all flights, the negative slope of 5.7 between MAC<sub>BC660</sub> and toluene:benzene ratio (Fig. 412 413 4a) infers that MAC<sub>BC660</sub> is larger when toluene:benzene is lower, which is typically thought to indicate more extensive 414 oxidation has occurred. The positive slope of 2.2 between MAC<sub>BC660</sub> and O:C ratio (Fig. 4b), supports the idea of larger MAC<sub>RC660</sub> with more oxidation. However, correlations are poor ( $R^2 < 0.2$ ) and because these trends are not 415 visible within a single plume, the explanation for the trends must be either that different fires emit different O:C and 416 toluene:benzene ratios, or that the chemistry that created the observed ratios occurred before the first transect of a 417 plume. Figure S3 supports this explanation in that while there is chemical aging within flights, the O:C and 418 toluene:benzene ratios are more variable from flight to flight. For example, RF06 got more chemical aged with time, 419 but the chemical markers for RF13 were flat with time. For either mechanism, the data shows that plumes that appear 420 421 "older" either by photochemical aging or because of more aged appearing emissions have a slightly higher MAC<sub>BC660</sub> 422 though the main point is that the MAC<sub>BC660</sub> does not change dramatically with either physical or chemical age for the observations during WE CAN. 423 Figure 4c and 4d show that the plume integrated mass concentration of BC (M<sub>BC</sub>, from the SP2) normalized 424 by CO (the ratio of background subtracted M<sub>RC</sub> to background subtracted CO) decreases with the toluene:benzene and 425 O:C ratios. One would expect a constant value of M<sub>BC</sub>/CO for single plumes from an individual fire because both are 426 primary and inert. Indeed, there is no obvious decreasing of  $M_{BC}$  ( $R^2 < 0.5$ ) within an individual flights except for 427 RF13. The decreasing of M<sub>BC</sub> with markers of chemical age amongst all the flights appears to be due to the different 428

429 properties of the different fires near the source. Different fires tend to have different O:C ratio and toluene:benzene

430 ratios, as shown in Fig. S4. Therefore, the decreasing trend of M<sub>BC</sub> with markers of chemical age is more likely caused

431 by fire to fire properties or rapid aging at the source rather than aging of the plume after the initial transect.

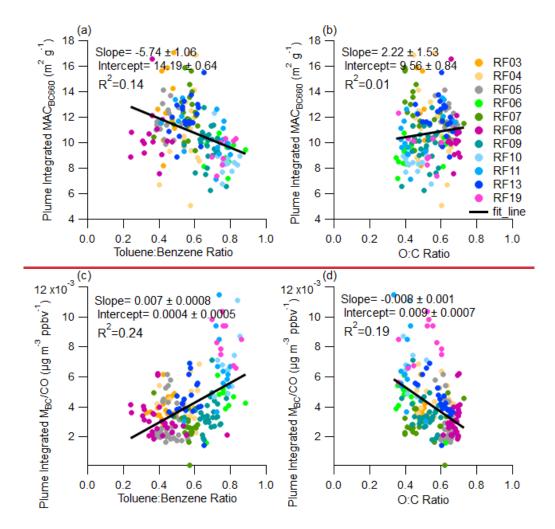


Figure 4: Plume integrated MAC<sub>BC660</sub> variations with (a) tolucne:benzene ratio and (b)O:C ratio; plume integrated M<sub>BC</sub> variations with (c) tolucne:benzene ratio and (d)O:C ratio.

432 SSA, scattering, and absorption at 660 nm are compared with the toluene:benzene ratio and O:C ratio in Fig. 433 5. There is a slight trend of increasing SSA<sub>660</sub> with these ratios that would correspond to more oxidized aerosol, but 434 the correlation is poor ( $\mathbb{R}^2$ =0.04). The increase of SSA<sub>660</sub> with markers of chemical age is consistent with the work 435 from Kleinman et al. (2020) on Western U.S. wildfire emissions, although Log<sub>10</sub>(NO<sub>x</sub>/NO<sub>y</sub>) was used as the indicator 436 for photochemical age in their study.

437 Both scattering and absorption at 660 nm decrease with ratios corresponding to more chemical aging, which suggests that the amount of absorbing material has changed. The comparison between normalized OA and these 438 markers of chemical age demonstrates that this is indeed the case. As can be seen in Fig. 5 (g h), OA decreases with 439 increasing O:C ratio with an R<sup>2</sup> of 0.7. A similar relationship can be found between normalized WSOC chemical age 440 441 (Fig. 6) in that WSOC decreases with increasing O:C ratio with a  $\mathbb{R}^2$  of 0.3. However, it is key to note that this correlation is not derived from individual flights and in fact is not robust in each flight and is rather due to fire to fire 442 443 variation. Figure S5 shows the correlation between normalized OA and chemical age within each fire source, which shows that different fires emit different OA, and OA does not always decrease with chemical age within a single fire 444

445 (Kiwah fire and Rabbitfoot fire). Therefore, we believe that different OA is caused by fire properties or fast chemistry

446 near the source, but that these markers (O:C, toluene:benzene) provide a significant correlation with the amount of

447 organic aerosol observed in various plumes.

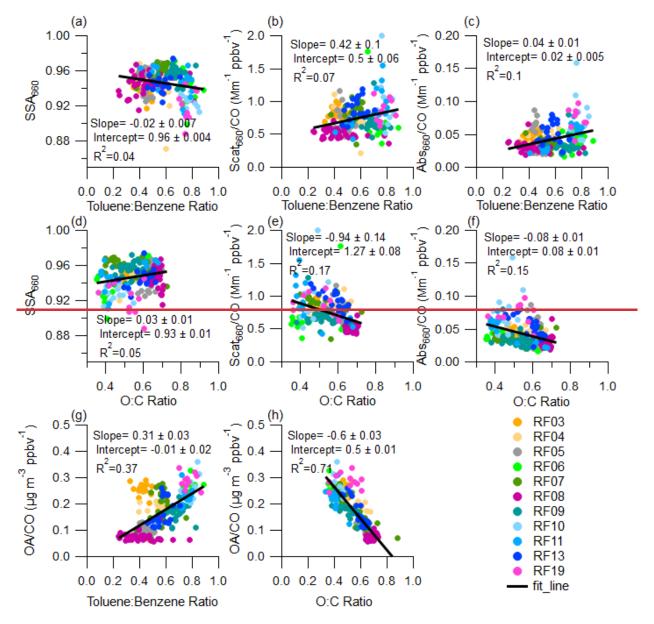


Figure 5: Plume integrated optical properties at 660 nm and normalized OA variation with chemical age. Top panels show (a) SSA, (b) Scattering, and (c) absorption variation with toluene:benzene ratio. Middle panels show (d) SSA, (c) Scattering, and (f) absorption variation with O:C ratio. Bottom panels show plume integrated normalized OA variation with (g) toluene:benzene ratio and (h) O:C ratio.

448

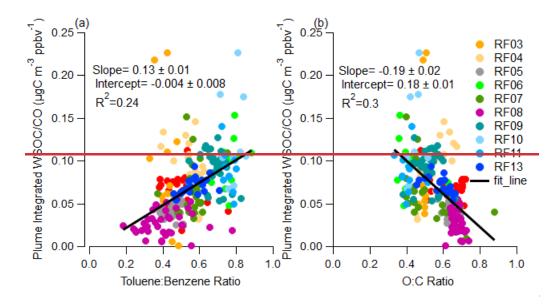


Figure 6: Plume integrated normalized WSOC variation with (a) toluene:benzene ratio and (b) O:C ratio.

The behavior of plume integrated normalized OA with altitude and temperature is shown in Fig. 7 a b. The
 trend is not significant, but the main reason is that for most flights we transected the plume, and caused the straight
 lines within the same colored marker. However, it is still clear that the smallest OA was captured in the plumes (RF08)

that have highest temperature (~305 K), and larger OA tends to be observed in the colder plumes (RF19). More studies
 are needed to determine whether OA is evaporated in high temperature plume, but it's beyond the scope of this work.

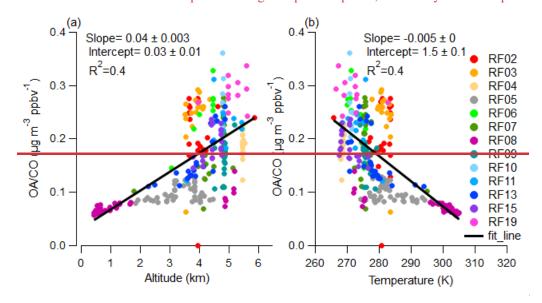


Figure 7: Plume integrated normalized OA variation with (a) altitude and (b) temperature.

454 RF05 and RF08 were chosen as case studies, to observe the optical properties of highly aged aerosol from 455 multiple fire sources to see if the optical properties of this aerosol were similar to those observed in the plume sampling 456 of individual fires at large chemical or physical age. RF08 was a flight through the Central Valley of California where 457 aged smoke from multiple fires that had settled into the valley was measured while RF05 was a flight in which smoke

from several California fires was observed in California, Oregon and Idaho roughly 300-600 miles from the fires 458 (flightpaths are shown in Fig. 1). MACRC666, CO mixing ratio, MCE, toluene:benzene ratio, O:C ratio and SSA666 are 459 displayed in Fig. 8a and 8b. The mixing ratio of CO is relatively low in these aged dilute smoke plumes vs. the plumes 460 near the sources analyzed earlier. CO mixing ratio is used as an indicator of smoke when it exceeds 150 ppb. 1 minute 461 averages of MAC<sub>BC660</sub> are calculated to reduce noise. Therefore, the 1-minute averages for MCE, toluene:benzene 462 463 ratio and O:C ratio were also calculated, and all the negative values were removed. As shown in Fig. 8, the smallest toluene:benzene ratio is ~0.35 in RF05, and is ~0.16 in RF08, while the largest O:C ratio is ~0.7 in both RF05 and 464 RF08, which indicates these two cases indeed captured plumes that appear chemically aged compared with the smallest 465 toluene:benzene ratio (0.33) and the largest O:C ratio (0.88) in near fire measurements shown in Fig. 4. 466

In RF05 (Fig. 8a), the weighted average O:C ratio over the entire flight was 0.64, the average MCE was 0.82 467 with a standard deviation of 0.1, the toluene:benzene ratio averaged 0.45 with a standard deviation of 0.05, and the 468 SSA666 averaged 0.95 with a standard deviation of 0.01. MCE has a few points because all the negative values were 469 470 removed, where either CO or CO2 is smaller than the background (CO < 150 ppb). MAC<sub>BC660</sub> varied from 8.9 m<sup>2</sup> g<sup>+1</sup> to 15.7 m<sup>2</sup> g<sup>-1</sup> with an average of 11.7 m<sup>2</sup> g<sup>-1</sup> and a standard deviation of 1.38 m<sup>2</sup> g<sup>-1</sup>. The reasonably large variation of 471 MCE may be caused by variability in the burn conditions of different fire sources, but the overall conclusion is that 472 these emissions, which were measured 300 to 600 miles away, have a very similar MAC<sub>BC660</sub> to that of the near source 473 474 flights.

The RF08 (Fig. 8b) results are similar to RF05, even though these emissions were smoke of mixed aged from 475 multiple fire sources in the Central Valley. The weighted average O:C ratio was 0.67 over the entire measurement, 476 average MCE was 0.84 with a standard deviation of 0.05, average toluene:benzene ratio was 0.41 with a standard 477 deviation of 0.15, and average SSA<sub>660</sub> was 0.94 with a standard deviation of 0.01. MAC<sub>BC660</sub> averaged 10.9 m<sup>2</sup> g<sup>4</sup> 478 with a standard deviation is 2.24 m<sup>2</sup> g<sup>4</sup>. There are several extreme values that exist in the dataset, probably because 479 of time alignment issues caused by variation in the dilution rate of the SP2 which cannot be totally eliminated from 480 the 1 minute average. In addition, the smoke from RF08 (Fig. 8b) is split into four regions based on observed CO 481 mixing ratios, and integrated MAC<sub>BC660</sub> is calculated for each region (purple star marker). Region edges are 482 represented by blue dashed lines. Region integrated MAC<sub>RC660</sub> is relatively stable with an average value of 10.2 m<sup>2</sup> g<sup>-</sup> 483 <sup>+</sup> and a standard deviation of 0.6 m<sup>2</sup> g<sup>-1</sup>. 484

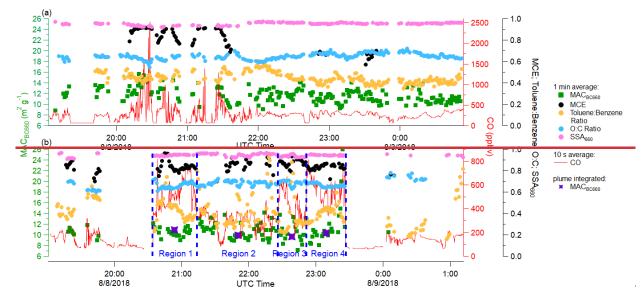
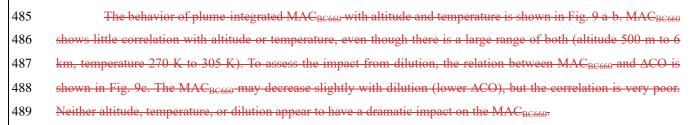


Figure 8: Time series of plume properties during (a) RF05, and (b) RF08(Central Valley of California). Different square and round markers indicate 1 min averages of different variables as shown in the legend, and the red solid-line represents 10 s averages of the mixing ratio of CO. Purple stars in RF08 indicate region integrated MAC<sub>BC660</sub> (individual regions are separated based on the concentration of CO, and indicated by blue dashed lines).



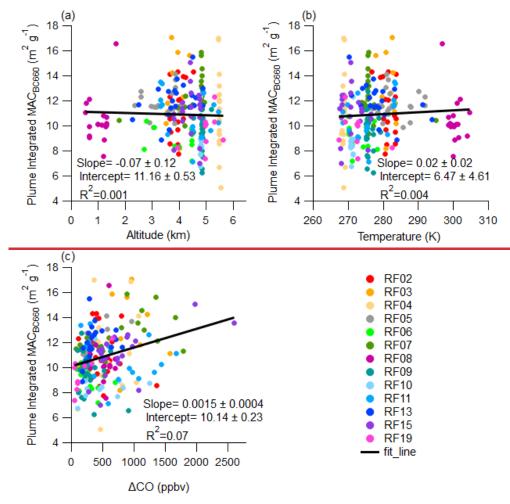


Figure 9: Plume integrated MAC<sub>BC660</sub> variations with (a) Altitude, (b) Temperature, and (c) ACO.

#### 490 3.1.4 Contribution of Brown Carbon Versus Lensing at 660 nm

491 Many previous studies of BrC assume that BrC does not absorb significant amounts of light at long 492 wavelengths (532~705 nm) (Wonaschütz et al., 2009; Lack et al., 2012a; Taylor et al., 2020; Zeng et al., 2021, Zhang 493 et al., 2022). In this study, a PILS system was used to quantify the absorption of light for water-soluble BrC at 660 494 nm. This absorption is not likely caused by traditional BC, which is insoluble and will be removed by the 0.2 µm filter 495 in the PILS (Peltier et al., 2007; Zeng et al., 2021).

496 One aim was-To investigate if which contributes more to absorption enhancement at 660 nm-was primarily
 497 due to, the lensing effect or due to absorption from BrC. The fraction of or the lensing effect, the fractional non-BC
 498 absorption from BrC at 660 nm wasis calculated by Eq. 6

499 Fractional 
$$Abs_{BrC} = \frac{Abs_{BrC_{660}}}{Abs_{Total_{660}} - M_{BC} * MAC_{BC_{core_{660}}}}$$
 (Eq.6)

500 where  $Abs_{Total_{660}}$  is the total absorption coefficient at 660 nm which is measured by the PAS,  $M_{BC}$  is the mass 501 concentration of BC which is measured by the SP2, and  $MAC_{BC\_core\_660}$  is the MAC of the BC core at 660 nm which

is set to be 6.3 m<sup>2</sup> g<sup>-1</sup> (Bond and Bergstrom, 2006; Subramanian et al., 2010). 5-Abs<sub>Brc 660</sub> is the total BrC absorption 502 coefficient at 660 nm, which is calculated from the water-soluble light absorption provided by the PILS, where we 503 504 convert absorption from water-soluble BrC to total BrC. More specifically, to convert the measured light absorption 505 by water-soluble organics into total BrC absorption in the ambient, it had to be multiplied by two factors. The first 506 factor converts absorption from water-soluble BrC into absorption from total BrC. This factor is obtained by taking 507 the ratio between total particulate organic mass and water-soluble particulate organic mass (OM: WSOCWSOM). Water-soluble organic mass is calculated from the PILS WSOC data using a WSOM:WSOC (water-soluble organic 508 509 mass : water-soluble organic carbon) ratio of 1.6 (Duarte et al., 2015 & 2019). Ambient organic mass is measured by 510 the AMS or calculated from the particle size distributions measured by the UHSAS assuming the particle mass all comes from organic material with a particle density of  $1.4 \text{ g cm}^{-3}$ . Both methods are used and compared in this paper. 511 512 The second factor accounts for the fact that particles absorb more light than the same substance in the bulk liquid 513 phase. Here we use Mie theory (Bohren and Huffman, 1983) to convert absorption from BrC in aqueous solution to 514 the absorption from BrC particles in the ambient (Liu et al., 2013; Zeng et al., 2020). The complex refractive index 515 (m = n + ik) was put into a Mie code (implemented into Igor by Ernie R. Lewis base on Bohren and Huffman, 1983) 516 to obtain the absorption efficiency (Q), and further used to calculate the absorption coefficient by Eq. 97 (Liu et al., 517 2013). The real part of the refractive index (n) is set to be 1.55, and the imaginary part is calculated by using Eq. 108518 (Liu et al., 2013).

$$520 = \frac{3}{2} \cdot \frac{Q \cdot WSOC}{D_p \cdot \rho}$$

$$(Eq.97)$$

 $\beta Abs(\lambda, D_n)$ 

519

$$521 \qquad \frac{\pi \cdot WSOC}{4\pi \cdot WSOC} \kappa$$

$$522 \qquad = \frac{\rho\lambda \cdot H_2 O_A bs(\lambda)}{4\pi \cdot WSOC} \qquad (Eq. 108)$$

where  $\lambda$  is the wavelength,  $D_p$  is the diameter of the particle,  $\beta Abs(\lambda, D_p)$  is absorption coefficient, Q is absorption efficiency, particle density ( $\rho$ ) is set to be 1.4 g cm<sup>-3</sup>, *WSOC* is the mass concentration of WSOC (µgC m<sup>-3</sup>) measured by the PILS, and  $H_2 O_{\beta}Abs(\lambda)$  is the water-soluble light absorption coefficient measured by PILS. The plume averaged particle size distribution was used in the calculation, then the absorption coefficient was calculated for each size bin of UHSAS to obtain the most accurate Mie factor for each plume.

The average OM: WSOCWSOM factor based on the UHSAS (UHSAS factor) for all the plumes is 2.36 with a standard deviation is 1.17. The averaged OM: WSOCWSOM based on the AMS (AMS factor) is 1.63 with a standard deviation of 0.74. The average Mie factor at 660 nm is 1.47 (standard deviation of 0.13), which is close to the factor of 1.36 found by Zeng et al. (2022) based on FIREX data. The Mie factor at 405 nm based on the WE-CAN data is also calculated, with an average of 1.83, which is similar to the factor that Zeng et al. (2022) determined at 405 nm (1.7) based on FIREX and Liu et al. (2013) determined at 450 nm (1.9) based on measurements in Atlanta.

Sensitivity tests were done on these factors by choosing reasonable ranges of particle density (1.1 g cm<sup>-3</sup>, 1.4 g cm<sup>-3</sup> and 1.7 g cm<sup>-3</sup>) and WSOM:WSOC ratio (1.5, 1.6 and 1.8) (Duarte et al., 2015 & 2019; Finessi, et al., 2012;
Sun et al., 2011) (Table <u>4S1</u>). Particle density only affects the Mie factor and UHSAS factor, while WSOM:WSOC

- ratio affects the AMS factor and UHSAS factor. As shown in Table  $\frac{4S1}{1}$ , the impact of particle density on the Mie
- factor (both at 660 nm and 405 nm) is negligible, WSOM:WSOC is the only component that affects the AMS factor
- 539 (ranging from 1.48 to 1.73), while the UHSAS factor is much more sensitive (ranging from 1.65 to 3.06) to both

540 particle density and WSOM:WSOC. Overall, Table <u>+S1</u> demonstrates that none of the factors other than the UHSAS

- 541 factor are sensitive to the exact parameters chosen for the calculation, giving confidence that the results presented are
- 542 robust.
- 543
   This approach assumes that water insoluble BrC has the same refractive index as water soluble BrC. This

   544
   assumption would provide a lower estimation on the BrC contribution to the total absorption because Sullivan et al.

   545
   (2022) found that 45% of the BrC absorption at 405 nm in WE-CAN came from water-soluble species, and Zeng et
- 546 <u>al. (2022) found that insoluble BrC absorbs more at higher wavelengths than soluble BrC, and methanol-insoluble</u>
- 547 <u>BrC chromophores caused 87% of the light absorption at 664 nm.</u>
- 548 **2.6 Absorption of BrC and Water-soluble BrC**
- 549 The bulk absorption coefficient of water-soluble BrC at a specific wavelength ( $Abs_{ws_BrC\lambda}$ ) is measured by 550 PILS system directly. The bulk absorption coefficient of BrC is calculated from Eq. 9:
- 555 Then the plume integrated absorption and scattering were normalized (x/CO) by taking the ratio of 556 background-subtracted absorption or scattering  $(\Delta x)$  to the background-subtracted CO mixing ratio  $(\Delta CO)$  (Eq. 10), 557 so that the changing of the normalized properties is not impacted by dilution of the plume with background air.

558 
$$x/CO = \frac{\Delta x}{\Delta CO}$$
 (Eq. 10)

559 2.7 Modified Combustion Efficiency (MCE)

<u>The variation of burn condition (e.g., flaming vs. smoldering) and fuel type can cause a significant difference</u>
 <u>in BC emissions and changes in aerosol properties (Akagi et al., 2011; Andreae, 2019). Burn conditions can be</u>
 <u>estimated with the modified combustion efficiency (MCE), defined as Eq. 11:</u>

563 
$$MCE = \frac{\Delta CO_2}{\Delta CO + \Delta CO_2}$$
(Eq.11)

564 where  $\Delta CO_2$  and  $\Delta CO_2$  are the background-subtracted  $CO_2$  and  $CO_2$  mixing ratio. The background of  $CO_2$  and  $CO_3$  mixing ratio is obtained via the same process described in Section 2.3.

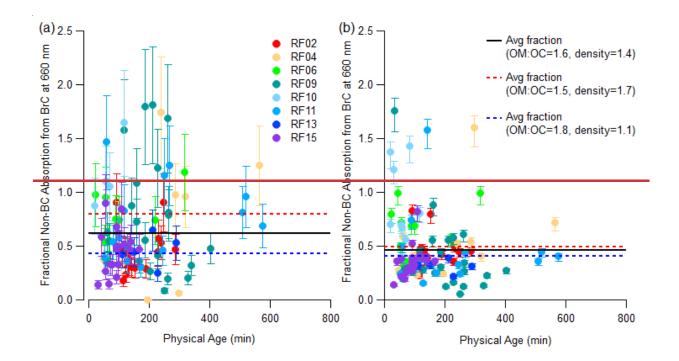
#### 566 <u>3 Results and Discussion</u>

Table 1: Average values and standard deviation of AMS factor, Mie factor at 660 nm and UHSAS factor for all the
 integrated plumes when using different particle density and WSOM: WSOC ratio. Unit of particle density is g cm<sup>-3</sup>.

Factor	Particle Density	WSOM:WSOC	Average	Standard Deviation
AMS factor	<del>N/A</del>	<del>1.5</del>	<del>1.73</del>	<del>0.79</del>
	<del>N/A</del>	<del>1.6</del>	<del>1.63</del>	<del>0.74</del>
	<del>N/A</del>	<del>1.8</del>	<del>1.48</del>	<del>0.79</del>
Mie factor at 660 nm	1.1	N/A	1.47	0.13
	1.4	N/A	<del>1.47</del>	0.13
	<del>1.7</del>	N/A	<del>1.47</del>	0.13
Mie factor at 405 nm	1.1	N/A	1.83	<del>0.89</del>
	1.4	<del>N/A</del>	<del>1.83</del>	<del>0.89</del>
	<del>1.7</del>	N/A	<del>1.83</del>	<del>0.89</del>
UHSAS factor	1.1	1.8	<del>1.65</del>	<del>0.82</del>
	<del>1.4</del>	<del>1.6</del>	<del>2.36</del>	<del>1.17</del>
	<del>1.7</del>	1.5	<del>3.06</del>	1.52

569

Figure 10 shows the time evolution of the fraction of non BC absorption from BrC at 660 nm for the biomass 570 burning plumes encountered during WE CAN. Assuming a MAC of the BC core of 6.3 m<sup>2</sup> g<sup>-1</sup>, BrC contributes roughly 571 572 the same amount of absorption at 660 nm as lensing (62% UHSAS method, 46% AMS method). This means that 19% 573 (based on the AMS) to 26% (based on the UHSAS) of the total absorption at 660 nm comes from BrC. When different particle density and WSOM:WSOC ratios are considered (top and bottom whiskers, as well as red and blue dashed 574 lines), the fraction of non BC absorption is 43 80% for the UHSAS approach and 41 49% for the AMS approach 575 based on different OM:OC and density. While there is considerable variability between flights, a rule of thumb that 576 577 roughly half of the non BC absorption at red wavelengths is from absorbing organic material seems reasonable. To our knowledge, this is the first-attempt to differentiate between lensing and absorbing organics in the red wavelengths. 578 This approach assumes that water insoluble BrC has the same refractive index as water soluble BrC. 579



# 580 **3.1.5 Aging of Water Soluble BrC at 660 nm**

581 To further analyze the evolution of BrC at 660nm in wildfire emissions, the MAC of water soluble BrC at 660 nm (MAC<sub>ws BrC660</sub>) was calculated by taking the ratio of water soluble light absorption ( $\beta_{ws BrC660}$ ) and WSOC 582 provided by the PILS (Eq. 4). Similar to MAC<sub>BC660</sub>, MAC<sub>we BrC660</sub> is relatively flat with physical age (Fig. 11a), with 583 an average of 0.06 m<sup>2</sup> g<sup>-1</sup> and a standard deviation of 0.04 m<sup>2</sup> g<sup>-1</sup>, but most of the plumes measured were less than 10 584 585 hours old. Interestingly, the fit lines for correlations with markers of chemical age suggest that MACws BrC660 tends to be larger when these markers indicate a more oxidized plume (Fig. 11 b c), which is distinct from what has been found 586 587 in previous studies that BrC at shorter wavelengths decays with chemical age. The trend of increasing MAC<sub>we BrC660</sub> is not clear in each flight and is only observed when properties are compared between fires. While the correlation 588 589 coefficients are low, it can be stated that MAC<sub>we BrC660</sub> is consistently larger in more oxidized plumes and there is not 590 a decrease with increased oxidation or chemical aging. Given that all the observed smoke plumes were of similar 591 physical ages, this again leads to the idea that these properties are the result of different emissions or fast chemistry 592 that occurs before the plumes are first observed by the aircraft.

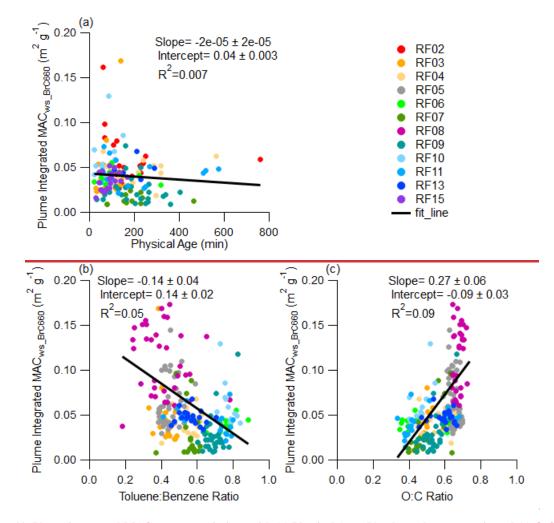
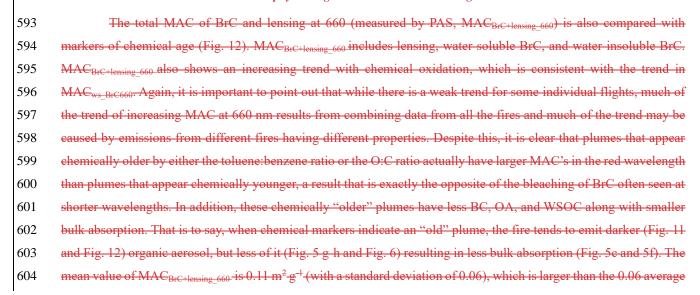


Figure 11: Plume integrated MAC<sub>ws\_BrC660</sub> variations with (a) Physical Age; (b) toluene:benzene ratio and (c) O:C ratio. RF05 and RF08 were measuring mixed aged smokes, during which the fire source cannot be identified, and therefore the physical age is unavailable for both flights.



605 of MAC<sub>ws BrC660</sub>, a result we have attributed to the lensing effect, but which could also partially be the result of water-

606 insoluble BrC having a higher MAC than water soluble BrC.

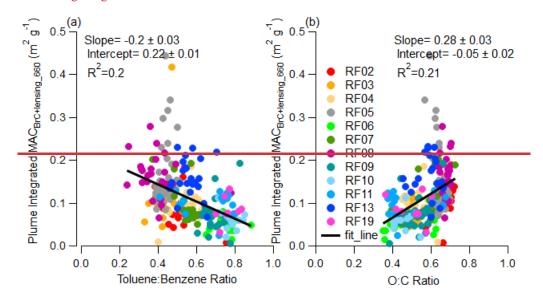


Figure 12: Plume integrated MAC<sub>BrC+lensing 660</sub> variations with (a) toluene:benzene ratio and (b) O:C ratio.

607The ratio of BC:OA has been shown to correlate to the optical properties of biomass burning aerosol (Pokhrel608et al., 2016; Saleh et al., 2014). During WE-CAN, MAC<sub>BrC+lensing\_660</sub> showed an increasing trend with increasing BC:OA609ratio (Fig. S6), which is similar to Saleh et al. (2014), who found that the imaginary part of the refractive index of OA610at 550 nm increases with the BC:OA ratio. This consistency suggests that MAC of BrC at 550 nm and 660 nm have611the same behavior in that BrC grows more absorbing as the BC:OA ratio increases.

612 3.2 Characteristics of BrC at 405 nm

#### 613 3.2.1 Behavior of MAC<sub>BrC+kensing\_405</sub>

614 Palm et al. (2020) utilized data from both WE CAN and Monoterpene and Oxygenated aromatic Oxidation 615 at Night and under LIGHTs (MOONLIGHT) campaigns and found that evaporated biomass burning POA is the 616 dominant source of the biomass burning SOA in wildfire plumes, which happened in the first a few hours after 617 emission. They also found that for those SOA that formed from oxidation, phenolic compounds contribute  $29 \pm 15\%$ 618 of BrC absorption at 405 nm. In this section, characteristics of BrC were also analyzed at 405 nm to understand if the 619 behavior was similar or different to BrC decay at 660 nm. BrC at 405 nm was calculated in the same way that it was calculated at 660 nm, following Eq. 3, and therefore it also has a contribution from the lensing effect. Figure 13 shows 620 621 the plume integrated MAC<sub>BrC+lensing 405</sub> variations with (a) Physical Age and (b) MCE. Similar to 660 nm, MAC BrC+lensing 405 varies from fire to fire and no clear trend can be found with increasing physical age or MCE. Similar 622 behavior was also observed in Western wildfires at 405 nm in FIREX-AQ (Zeng et al., 2022). The MAC BIC Hensing 4057 623 varies from 0.08 m<sup>2</sup> e<sup>+</sup> to 1.6 m<sup>2</sup> e<sup>+</sup> with a mean value of 0.59 m<sup>2</sup> e<sup>-1</sup> and a standard deviation of 0.19. The largest 624 625 values are from RF05, the flight through California, Oregon, and Idaho, where aged smoke from different fires was mixed. The large MACBrC+lensing 405 in RF05 is related to relatively small OA (Fig. 5 g h), which occurred when the 626

627 plane left the smoke-filled boundary layer during RF05. If we exclude  $MAC_{BrC+lensing\_405}$  from RF05, the values range 628 from 0.08 m<sup>2</sup>-g<sup>-1</sup> to 1.09 m<sup>2</sup>-g<sup>-1</sup>, but still have a mean value of 0.59 m<sup>2</sup>-g<sup>-1</sup> and a standard deviation of 0.15. Again, we 629 note that this value includes the contribution of lensing. Despite this, our results lie in the same range as those measured 630 without the contribution of lensing of 0.31 ± 0.09 m<sup>2</sup>-g<sup>-1</sup> measured in CLARIFY 2017 (Taylor, 2020), 0.13 - 2.0 m<sup>2</sup>-g<sup>-1</sup>

631 measured in FIREX AQ (Zeng et al., 2022), and 0.25 1.18 m<sup>2</sup> g<sup>-1</sup> measured in ORACLES (Zhang et al., 2022).

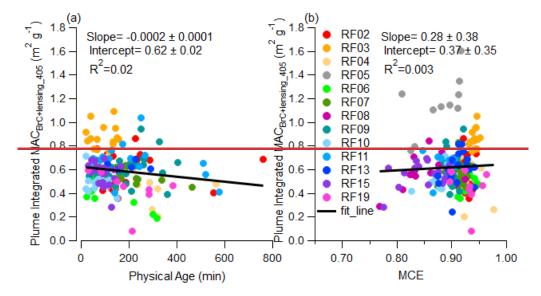


Figure 13: Plume integrated MAC<sub>BrC+lensing</sub> 405 variations with (a) Physical Age and (b) MCE

Figure 14 shows the behavior of brown carbon at 405 nm with markers of chemical aging. Very weak or non-632 trends are observed. If there is any trend, it is a slight increase in MACBrC+lensing 405-with decreasing toluene:benzene 633 ratio, which is consistent with the results for MACBrC+lensing 660 (Fig. 12) and MACws BrC660 (Fig. 11). The flat or slightly 634 increasing trend with increasing oxidation shows that the decrease in total aerosol absorption with markers of chemical 635 age is due to a decrease in OA (Fig. 5 g h) because the BrC is actually darker in oxidized plumes at all wavelengths. 636 It is important to remember that most of the trends observed in WE CAN are caused by emissions from different fires 637 638 versus variations within a fire, which tend to be quite small. Consistent results for the behavior of MAC<sub>BrC</sub> at different wavelengths derived using different instruments (PAS and PILS) is further evidence that BrC decay does not occur in 639 the WE-CAN dataset, or at least that plume integrated results cannot capture the BrC decay that might be occurring at 640 the edges of the plume. While further research focused on the edge of the plumes, which often appear highly oxidized, 641 is needed, this is beyond the scope of the current work. Despite this, it is the plume integrated results that are relevant 642 643 for climate impacts and for comparison to model output, discussed in the following section.

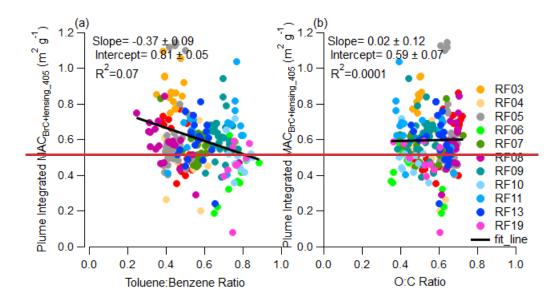
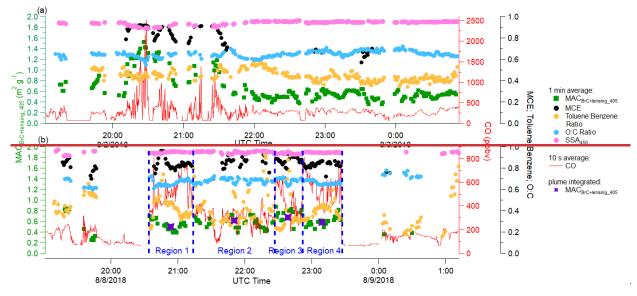


Figure 14: Plume integrated MAC<sub>BrC+lensing</sub> 405 variations with (a) toluene:benzene ratio and (b) O:C ratio

RF05 and RF08 are presented as case studies again to investigate the behavior of MAC<sub>BrCHensing 405</sub> in mixed 644 plumes emitted from different fire sources. Figure 15 is similar to Fig. 645 8. but with MACBrC+lensing 405 instead of MAC<sub>BC666</sub>. For the case of RF05 (Fig. 15a) MAC<sub>BrCHensing</sub> 405 varied from 0.36 m<sup>2</sup>-g<sup>-1</sup> to 1.52 m<sup>2</sup>-g<sup>-1</sup> with an average 646 of 0.66 m<sup>2</sup> g<sup>-1</sup> and a standard deviation of 0.26 m<sup>2</sup> g<sup>-1</sup>, the SSA<sub>450</sub> averaged 0.94 with a standard deviation of 0.02, 647 648 which is similar to SSA<sub>660</sub>. The MAC<sub>BrCHensine</sub> 405- is larger when CO mixing ratio is higher, but does not have a significant correlation with any other variables shown in Fig. 15. For the case of RF08 (Fig. 15b) MAC BrC+lensing 405 is 649 more stable than in RF05, and varied from 0.25 m<sup>2</sup> g<sup>-1</sup> to 0.88 m<sup>2</sup> g<sup>-1</sup> with an average of 0.59 m<sup>2</sup> g<sup>-1</sup> and a standard 650 deviation of 0.14 m<sup>2</sup> g<sup>-1</sup>, the SSA<sub>450</sub> average was 0.95 with a standard deviation of 0.01. The regional integrated 651 MAC<sub>BrC+lensine 405</sub> is even more stable with an average value of 0.59 m<sup>2</sup> g<sup>-1</sup> and a standard deviation of 0.07 m<sup>2</sup> g<sup>-1</sup>. 652



 $\frac{653}{\text{When comparing MAC}_{\text{BrC+lensing}_405} \text{ with altitude and temperature (Fig. 16), it has the same behavior with}}{\frac{654}{\text{MAC}_{\text{BcC660}} \text{ in that MAC}_{\text{BrC+lensing}_405} \text{ stays relatively constant with both altitude and temperature. In addition, when}}{\frac{655}{\text{MAC}_{\text{BrC+lensing}_405} \text{ is plotted vs. ACO (not shown), no clear change in MAC}_{\text{BrC+lensing}_405} \text{ is seen due to dilution.}}}{\frac{656}{\text{C+Brc+lensing}_405} \text{ and MAC}_{\text{BcC660}} \text{ do not appear to be affected by altitude or temperature during WE}}}{\frac{657}{\text{CAN}}}$ 

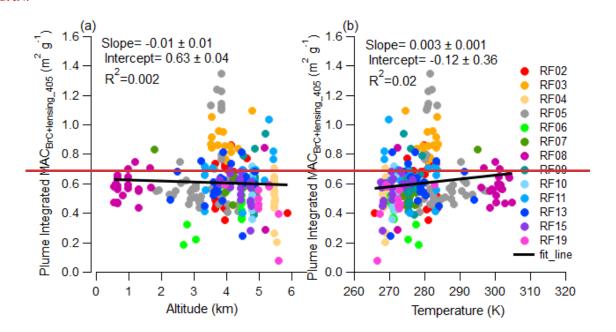


Figure 16: Plume integrated MAC<sub>BrC+lensing\_405</sub> variations with (a) Altitude and (b) Temperature

When comparing the relationship between MAC<sub>BrC+lensing\_405</sub> with the BC:OA ratio (Fig. S7), there is not a clear increasing trend, and the correlation is worse than that at 660 nm (Fig. S6). However, the increasing trend still exists in most individual flights (not in RF04 and RF10). The increasing trend-is not as clear as in Saleh et al. (2014), most probably because the range of BC:OA ratios observed during WE CAN (0.007-0.061) is much smaller than that (0.005-0.7) observed in Saleh's work. Even in their work, the increasing trend is not very clear if one only focuses on the region where the BC:OA ratio is less than 0.03. Also, the Saleh et al.-results were obtained from laboratory burns and not wildfires, which might also cause a discrepancy.

# 665 <u>3.2.23.1</u> Comparison of WE-CAN <u>ResultsMAC<sub>BrC</sub></u> to Modeling Studies

The BC:OA ratios measured during WE CAN were utilized in the Saleh et al. parameterization (2014), It is 666 challenging for climate models to simulate absorption from BrC, especially because it is highly wavelength dependent 667 668 and may change with chemical age (Liu et al., 2020). Recently the Saleh et al. (2014) parameterization has been implemented in models in an attempt to better parameterize the imaginary part of the BrC refractive index (Wang et 669 670 al., 2018; Carter et al., 2021). To test how accurately the Saleh parameterization matched WE-CAN data, the BC:OA ratios measured during WE-CAN were input into the Saleh parameterization, which provides an imaginary part for 671 the refractive index of BrC ( $k_{BrC,\lambda}$ ) as a function of the BC:OA ratio. The meanplume integrated BC:OA ratio for each 672 673 plume was used in the parameterization, which gave an average  $k_{BrC}$  of 0.025, 0.013, 0.009, respectively, at 405 nm,

550 nm and 660nm. Mie theory (Bohren and Huffman, 1983) was then used to calculate the MAC for BrC, in which, 674 675 To do the Mie calculations we assumed a real part of the refractive index of 1.7 for BrC (same as Saleh et al., 2014), 676 aused volume mean diameterdiameters measured for each plume, and aused an organic density of 1.4 g cm<sup>-3</sup>. Figure 677 172 compares the observed MAC<sub>BrC+lensing</sub> (Eq. 4) and MAC<sub>ws BrC</sub> (Eq. 5) with the value calculated from the Saleh 678 parameterization with inputs from WE-CAN. In both the observations and the parameterization, the MAC<sub>BrC</sub> decreases 679 as wavelength increases. However, the Saleh parameterization is always significantly larger than the observations. 680 The MAC<sub>BrC</sub> from the Saleh parameterization, which does not include lensing effects, is a factor of 3.4 and 2.8 larger 681 than the observed MAC<sub>BrC+lensing</sub> at 405 nm and 660 nm, respectively. The range of BC:OA ratios during WE-CAN  $(0.007 \sim 0.061)$  is much smaller than that on the very small end of the range  $(0.005 \sim 0.7)$  used in Saleh's work, and the 682 683 parameterization failed to capture absorbing aerosol properties for this study. The discrepancy could also be partly 684 because the data Saleh et al. used for their parameterization comes from controlled laboratory burns and not wildfiresor because of the sensitivity of  $MAC_{Brc}$  to density when using. It is worth noting that the Saleh parameterization-685 686 When we increase of MAC<sub>BrC</sub> is very sensitive to organic aerosol density. If particle density is increased from 1.4 g cm<sup>-3</sup> to 1.7 g cm<sup>-3</sup>, the Saleh parameterization median MAC<sub>BrC</sub> decreases to 1.6 m<sup>2</sup> g<sup>-1</sup> and 0.24 m<sup>2</sup> g<sup>-1</sup>, respectively, at 687 688 405 nm and 660 nm (a factor of 2.8 and 2.3, respectively, compared to observed MAC<sub>BrC</sub> at 405 nm and 660 nm). This 689 suggests The fact that the Saleh parameterization overestimates the absorption property of biomass aerosol especially 690 for fresh emitted aerosols and more suggests that different parameterizations need to be developed are needed for the Western U.S.. Carter et al. (2021) utilized the Saleh parameterization for BrC absorption in the GEOS-Chem model 691 692 and also found that the Saleh model overestimated BrC absorption for WE-CAN. It was hypothesized that the 693 overestimation was due to the lack of a bleaching process for BrC in the model and offset part of the overestimation 694 by bringing in bleaching into the model. However, our results show that the overestimation in the model is caused by 695 an incorrect refractive index.

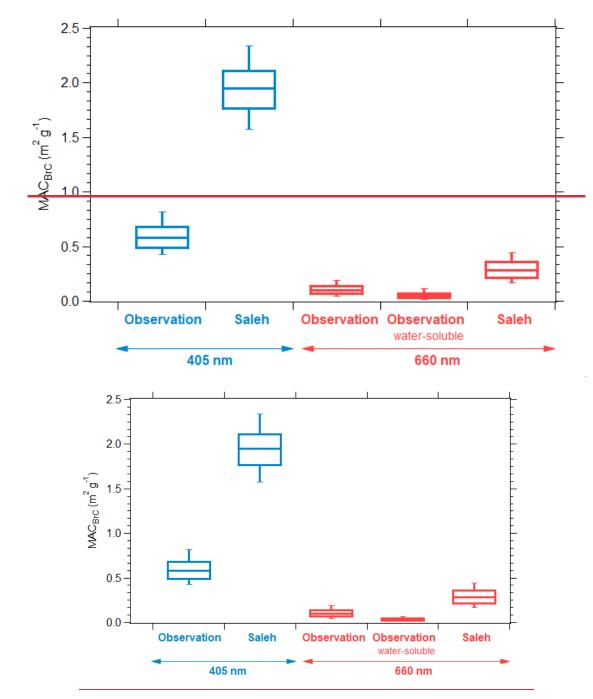


Figure 472: Boxplot summary for observed and parameterized (Saleh) MAC<sub>BrC</sub> at 405 nm (blue) and 660 nm (red). For each box, the central line represents the median, the top and bottom edges represent the 75<sup>th</sup> and 25<sup>th</sup> percentile, and the top and bottom whiskers represent the 90<sup>th</sup> and 10<sup>th</sup> percentile of the data.

# 696 <u>3.2 Investigation of BrC Bleaching at Visible Wavelengths</u>

A limited number of field measurements have shown BrC decay with chemical age (Forrister et al., 2015;
 Wang et al., 2016). Despite a relatively poor understanding of the mechanism of bleaching or whitening of BrC, this
 process has been implemented in numerous model simulations (Brown et al., 2018; Wang et al., 2018; Carter et al.,

700 2021). The definition of bleaching or whitening is unclear in previous literature, models tend to treat bleaching as the 701 change of refractive index or decreasing of MAC (Brown et al., 2018; Wang et al., 2018; Carter et al., 2021), while 702 observations or lab experiments mostly link bleaching to the decrease of absorption coefficient (Forrister et al., 2015; 703 Palm et al., 2020; Zeng et al., 2022). It is important to distinguish between these two, because the decrease of 704 absorption coefficient can also be caused by loss of absorbing organic material, which will also change the scattering 705 coefficient and radiative forcing. Therefore, the MAC of BrC and the absorption coefficient of BrC at visible 706 wavelengths were calculated and analyzed together with two chemical clocks (O:C and toluene:benzene ratio) and 707 organic mass, to determine whether BrC bleached during the WE-CAN campaign, and whether the bleaching was 708 caused by the less organic mass or the changing of refractive index. Because all large wildfire emissions are a mix of 709 different regions that are burning slightly different fuels at different combustion efficiencies and because models treat 710 regions, not individual fires, we identify relationships in this paper that hold true across all the flight data collected 711 during WE-CAN. These types of broad correlations are much more useful than individual case studies yielding results

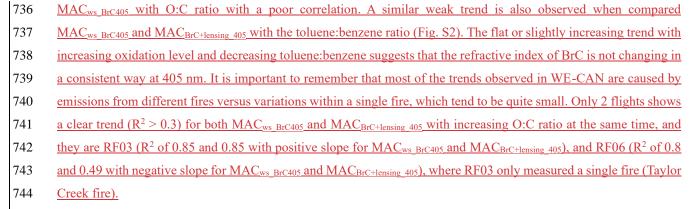
712 <u>that only hold true sometimes.</u>

#### 713 <u>3.2.1 Consistency of The Mass Absorption Cross-Section of BrC at 405 nm</u>

714 Palm et al. (2020) combined data from WE-CAN and the Monoterpene and Oxygenated aromatic Oxidation 715 at Night and under LIGHTs (MOONLIGHT) chamber experiment and found that evaporated biomass-burning POA 716 is the dominant source of biomass-burning SOA in wildfire plumes during the first a few hours after emission. They 717 also found that of the SOA formed from oxidation, phenolic compounds contribute  $29 \pm 15\%$  of BrC absorption at 718 405 nm. In this section, we analyze the characteristics of BrC at 405 nm to understand the average properties of BrC 719 and to understand the balance of BrC formation versus bleaching during WE-CAN. The MAC of BrC is calculated 720 following Eq. 4 and therefore it includes a contribution from the lensing effect. The MAC of water-soluble BrC is calculated following Eq. 5. Figure 3 shows the MAC of BrC at 405 nm versus the aerosol oxidation level (O:C ratio), 721 722 while Fig. S2 is a similar plot that uses a simple photochemical clock, the gas-phase toluene:benzene ratio. The O:C 723 ratio characterizes the oxidation state of OA and typically increases with photochemical age (Aiken et al., 2008), while 724 the toluene:benzene ratio decreases with photochemical processing time since toluene is more reactive than benzene 725 (Gouw et al., 2005). Both markers are two commonly used markers to indicate the chemical age of smoke, and they 726 correlated well with each other during WE-CAN (Fig. S1). The MAC<sub>BrC+lensing</sub> 405, varies from 0.08 m<sup>2</sup> g<sup>-1</sup> to 1.6 m<sup>2</sup> g<sup>-1</sup> with a mean value of 0.59 m<sup>2</sup> g<sup>-1</sup> and a standard 727 deviation of 0.19. The largest values are from RF05, the flight through California, Oregon, and Idaho, where aged 728 smoke from different fires was mixed. The large MAC<sub>BrC+lensing 405</sub> occurred when the aircraft left the smoke-filled 729

730 boundary layer during RF05. If we exclude MAC<sub>BrC+lensing 405</sub> from RF05, the values range from 0.08 m<sup>2</sup> g<sup>-1</sup> to 1.09

- 731  $\underline{m^2 g^{-1}}$ , but still have a mean value of 0.59  $\underline{m^2 g^{-1}}$  and a standard deviation of 0.15. Again, we note that this value
- 732 includes the contribution of lensing. Despite this, our results lie in the same range as those measured without the
- 733 contribution of lensing of  $0.31 \pm 0.09 \text{ m}^2 \text{ g}^{-1}$  measured in CLARIFY-2017 (Taylor, 2020), 0.13-2.0 m<sup>2</sup> g<sup>-1</sup> measured
- in FIREX-AQ (Zeng et al., 2022), and 0.25-1.18 m<sup>2</sup> g<sup>-1</sup> measured in ORACLES (Zhang et al., 2022). Very weak or
- 735 non-trends are observed versus the chemical markers of aging (Fig. 3). If there is any trend, it is a slight increase in



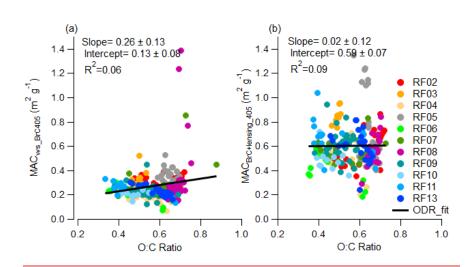
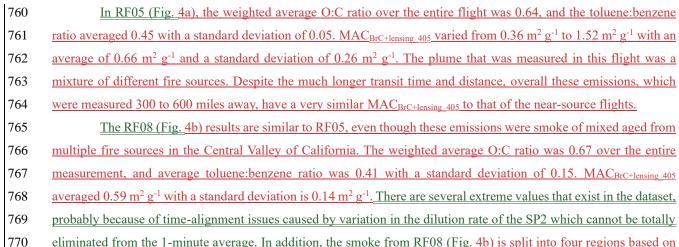


Figure 3: Plume integrated (a) MAC<sub>ws BrC405</sub> and (b)MAC<sub>BrC+lensing 405</sub> variations with the organic aerosol O:C ratio

746	RF05 and RF08 were chosen as case studies, to observe the optical properties of highly aged aerosol to see
747	if the optical properties of this aerosol were similar to those observed in the near-field plume sampling of individual
748	fires at the longest chemical or physical age observed, which was roughly 6-24 hours of physical aging. RF08 was a
749	flight through the Central Valley of California where aged smoke from multiple fires that had settled into the valley
750	was measured while RF05 was a flight in which smoke from several California fires was observed in California,
751	Oregon and Idaho roughly 300~600 miles from the fires (flightpaths are shown in Fig. 1). MAC <sub>BrC+lensing_405</sub> , CO
752	mixing ratio, toluene:benzene ratio, and O:C ratio are displayed in Fig. 4a and 4b. The mixing ratio of CO is relatively
753	low in these aged dilute smoke plumes vs. the plumes near the sources analyzed earlier. 1-minute averages of
754	MAC <sub>BrC+lensing 405</sub> are calculated to reduce noise and 1-minute-averages for toluene:benzene ratio and O:C ratio were
755	calculated and all the negative values were removed. As shown in Fig. 4, the smallest toluene:benzene ratio is ~0.35
756	in RF05, and is ~0.16 in RF08, while the largest O:C ratio is ~0.7 in both RF05 and RF08, which indicates these two
757	cases indeed captured plumes that appear chemically aged to similar extent to the other near-field flights where the
758	smallest toluene:benzene ratio was 0.33 and the largest O:C ratio was 0.88 in near-fire measurements (Fig. 3 and Fig.
759	<u>S2).</u>
•	



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771 varying observed CO mixing ratios, and integrated MAC<sub>BrC+lensing\_405</sub> is calculated for each region (purple star marker).

The regional edges are represented by blue dashed lines. Integrated  $MAC_{BrC+lensing 405}$  for all of these variable CO

regions is relatively stable with an average value of  $0.59 \text{ m}^2 \text{ g}^{-1}$  and a standard deviation of  $0.07 \text{ m}^2 \text{ g}^{-1}$ .

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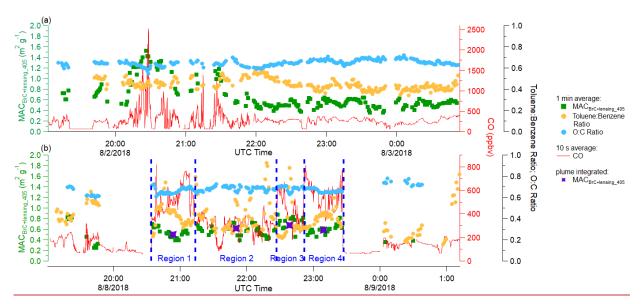


Figure 4: Time series of plume properties during (a) RF05 (measurements far from fire source), and (b) RF08 (Central Valley of California). Different square and round markers indicate 1 min averages of different variables as shown in the legend, and the red solid line represents 10 s averages of the mixing ratio of CO. Purple stars in RF08 indicate region integrated MAC<sub>BrC+lensing 405</sub> (individual regions are separated based on the concentration of CO, and indicated by blue dashed lines).

# 775 <u>3.2.2 Decrease in Absorption at 405 nm Observed with Markers of Chemical Oxidation</u>

Although neither the MAC<sub>BrC+lensing 405</sub> nor MAC<sub>ws BrC405</sub> decreases with O:C or Toluene:Benzene, Fig. 5
 shows that BrC bleaching is observed in terms of decreased total absorption. Figure 5a and 5c show the behavior of
 BrC absorption at 405 nm with markers of the aerosol oxidation level (O:C) and photochemistry (toluene:benzene).
 The absorption coefficient of BrC shown in Fig. 5 is calculated by Eq. 9-10, which cannot separate the absorption

780 caused by the BrC and lensing effect. To confirm that observed trends are not the result of changing lensing, the 781 absorption coefficient of water-soluble BrC measured by the PILS, which does not include lensing effects, is also 782 compared in Fig. 6a and 6c. The average water-soluble BrC absorption at 405 nm (Abs<sub>ws BrC405</sub>, 0.02 Mm<sup>-1</sup> ppbv<sup>-1</sup>) 783 is only 20% of the total absorption from BrC plus lensing (Abs<sub>BrC+lensing 405</sub>, 0.11 Mm<sup>-1</sup> ppbv<sup>-1</sup>). However,  $Abs_{BrC+lensing 405}$  and  $Abs_{ws BrC405}$  both decrease with increasing O:C (R<sup>2</sup> = 0.65 and R<sup>2</sup> = 0.3, respectively for 784 785 Abs<sub>BrC+lensing 405</sub> and Abs<sub>ws BrC405</sub>) and decreasing toluene:benzene ratio, which suggest a similar level of 786 decreasing BrC absorption for all the fires observed in WE-CAN from numerous locations in the western U.S.. This relationship holds despite differences in fuel type, burn conditions, meteorology, etc. between all of these fires. The 787 observed trends are mostly due to the decreasing of both total OA mass (Fig. 5b and 5d) and WSOC (Fig. 6b and 6d) 788 789 with the increasing O:C ratio ( $R^2 = 0.8$  and  $R^2 = 0.4$ , respectively for OA and WSOC) and decreasing toluene:benzene ratio ( $R^2 = 0.64$  and  $R^2 = 0.44$ , respectively for OA and WSOC). Overall, the organic aerosol O:C ratio better predicts 790 791 BrC evolution than toluene:benzene ratio, probably because it is a particle-phase property rather than a gas-phase one. 792 Again, it is important to clarify if BrC "bleaching" is caused by decreasing BrC absorption coefficient or decreasing 793 of BrC refractive index (or MAC). In this study, decreasing MACBrC is not observed, rather the BrC absorption 794 coefficient decreases significantly with the the simple O:C and Toluene:Benzene chemical clocks due to loss of OA 795 mass. Less OA mass also causes decrease in bulk scattering coefficient (Fig. S3), leading to a very different net 796 radiative effect than reducing MACBrC. 797 It is important to recall that we aimed to find general trends that hold for all fires in the western U.S., and the above trend is significant when all fires are grouped together, although the trend is, in fact, not robust in each flight 798 799 and is rather due to variations between the fire plumes rather than variation within a single fire plume. Figure 7 shows 800 the correlation between normalized OA and chemical age for each fire source. It demonstrates that different fires show 801 different relationships and that OA does not always decrease with oxidation level/chemical aging within a single fire 802 (Kiwah fire and Rabbitfoot fire), though increasing O:C ratio does correlate well ( $R^2 > 0.3$ ) with decreasing OA mass in 7 fires (with R<sup>2</sup> of 0.94 for Taylor Creek fire, 0.87 for Carr fire, 0.86 for Beaver Creek fire, 0.8 for Coal Hollow 803 fire, 0.76 for Bear Trap fire, 0.35 for Sharps fire, and 0.31 for Sugarloaf fire). Toluene:benzene ratio didn't track OA 804

- as good as O:C ratio, and decreasing toluene:benzene ratio correlates well ( $R^2 > 0.3$ ) with decreasing OA mass in 3
- 806 fires (with R<sup>2</sup> of 0.87 for Rabbitfoot fire, 0.85 for Coal Hollow fire, and 0.84 for Bear Trap fire).
- 807

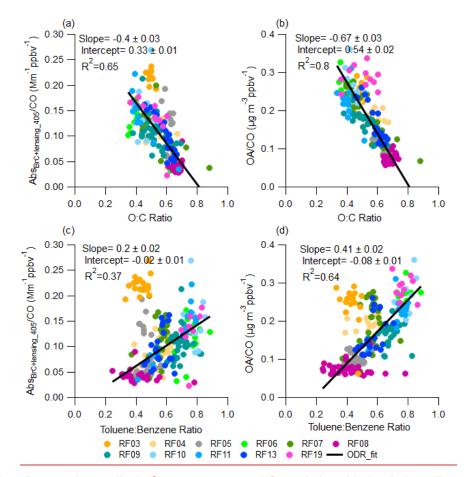


Figure 5: Plume integrated normalized  $Abs_{BrC+lensing_405}$ , and OA variation with chemical age. Top panels show (a) plume integrated normalized  $Abs_{BrC+lensing_405}$ , and (b) plume integrated normalized OA variation with O:C ratio. Bottom panels show (c) plume integrated normalized  $Abs_{BrC+lensing_405}$ , and (d) plume integrated normalized OA variation with toluene:benzene ratio. Data from RF03 was excluded from the ODR fit with toluene:benzene ratio, because RF03 sampled the injection of fresh smoke into the free troposphere, where gas species reacted more rapidly than particles and toluene:benzene ratio failed to keep track of aerosol evolution.

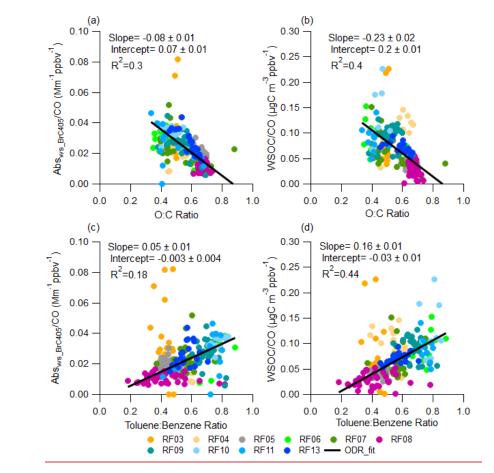
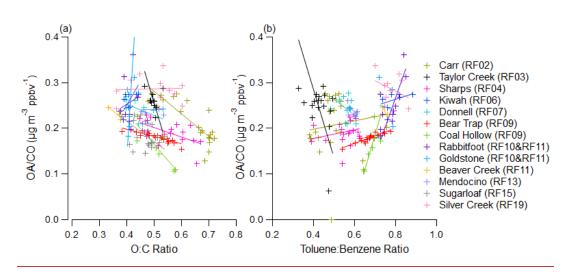
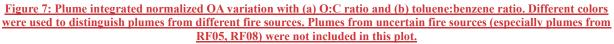


Figure 6: Similar to Fig. 5, but with plume integrated normalized Abs<sub>ws\_BrC405</sub>, from PILS in (a) and (c), and WSOC in (b) and (d)





811 No trend is observed in CO normalized OA mass with plume physical age (Fig. S4), which is consistent with 812 the result from Garofalo et al. (2019) in that no net OA mass change was observed in individual plumes during WE-813 CAN when they are characterized by physical age, although more data from additional fires were included in the current work. Plume-integrated CO-normalized OA also shows weak or no trend with altitude and temperature (Fig. 814 815 S5). However, we note that the smallest OA:CO was captured in the plumes (RF08) that have highest temperature 816 (~305 K), and larger OA:CO tends to be observed in the colder plumes (RF19). More studies are needed to determine how much OA is evaporated in high temperature plumes because the WE-CAN dataset does not capture enough 817 818 variation of temperature within plumes to make a robust conclusion. No clear trend was found between 819 MAC<sub>BrC+lensing</sub> 405 and physical age or MCE (Fig. S6). Similar behavior was also observed in Western wildfires at 405 820 nm in FIREX-AQ (Zeng et al., 2022). Part of the reason is that for most fires, we only captured the first few hours (< 821 15 h), and MCE do not have a robust capability to predict biomass burning particle properties (McClure et al., 2020). No trend is found between MAC<sub>BrC+lensing 405</sub> and altitude or temperature (Fig. S7). The trend with BC:OA ratio (Fig. 822 823 S8) is not as clear as in Saleh et al. (2014), most probably because the range of BC:OA ratios observed during WE-824 CAN (0.007~0.061) is much smaller than that (0.005~0.7) observed in Saleh's work. Even in their work, the increasing trend is not very clear if one only focuses on the region where the BC:OA ratio is less than 0.03. Also, the Saleh et al. 825

826 (2014) results were obtained from laboratory burns and not wildfires, which might also cause a discrepancy.

# 827 <u>3.2.3 Mass Absorption Cross-Section and Optical Properties of BrC at 660 nm</u>

828 BrC is defined as OA that has strong absorption at UV and shorter visible portions of the spectrum and has been historically considered to be almost transparent near the red wavelengths (Andreae and Gelencsér, 2006; Bahadur 829 830 et al., 2012; Liu et al., 2020). However, during WE-CAN, we were able to quantify Abs<sub>ws BrC660</sub> with the PILS 831 instrument. We know that absorption observed in the PILS at 660 nm is not BC because BC is insoluble and will be 832 removed by the 0.2 µm filter in the instrument. Next, we investigate the behavior of BrC absorption at 660 nm to see 833 if BrC has a similar behavior at the long versus short ends of the visible spectrum. 834 Figure 8 shows the behavior of brown carbon at 660 nm vs. the O:C ratio. Similar to 405 nm, no bleaching 835 in terms of decreased MAC is observed at 660 nm. If there is any trend, it is increasing MAC<sub>ws BrC660</sub> and MAC<sub>BrC+lensing</sub> 660 with organic aerosol O:C ratio. Similar trends are observed, though with lower correlation, versus 836 837 the toluene:benzene ratio (Fig. S9). The mean value of MAC<sub>BrC+lensing 660</sub> is 0.11 m<sup>2</sup> g<sup>-1</sup> (with a standard deviation of 0.06), which is much larger than the 0.03 average of MAC<sub>ws BrC660</sub>, a result we have attributed to the lensing effect, 838 but which could also partially be the result of water-insoluble BrC having a higher MAC than water-soluble BrC. 839 These results for the behavior of MACBrc at different wavelengths derived using different instruments (PAS 840 841 and PILS) is further evidence that MACBIC does not decrease with physical or chemical age in the WE-CAN dataset.

- 842 At a minimum, the plume integrated results, which represent total optical properties relevant to climate models, do
- 843 not capture any  $MAC_{BrC}$  decay that might be occurring at the edges of the plume.

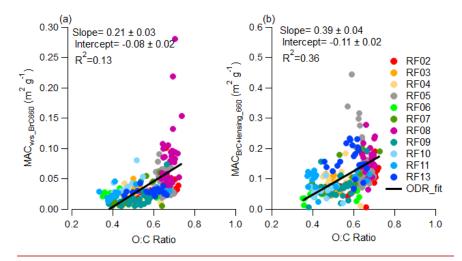


Figure 8: Plume integrated (a) MACws BrC660 and (b)MACBrC+lensing 660 variations with O:C ratio

845 Similar to our analysis at 405 nm, RF05 and RF08 are presented as case studies to investigate the behavior of MAC<sub>BrC+lensing 660</sub> in aged plumes emitted from different fire sources. Figure 9 is similar to Fig. 4, but with 846 MACBrC+lensing 660 instead of MACBrC+lensing 405. For the case of RF05 (Fig. 9a) MACBrC+lensing 660 varied from 0.04 m<sup>2</sup> 847 g<sup>-1</sup> to 0.40 m<sup>2</sup> g<sup>-1</sup> with an average of 0.15 m<sup>2</sup> g<sup>-1</sup> and a standard deviation of 0.07 m<sup>2</sup> g<sup>-1</sup>. The MAC<sub>BrC+lensing 660</sub> tends 848 to be larger when CO mixing ratio is higher, but does not have a significant correlation with any marker of oxidation 849 level or photochemistry shown in Fig. 9. For the case of RF08 (Fig. 9b) MAC<sub>BrC+lensing 660</sub> is more stable than in RF05, 850 and varied from 0.04 m<sup>2</sup> g<sup>-1</sup> to 0.37 m<sup>2</sup> g<sup>-1</sup> with an average of 0.18 m<sup>2</sup> g<sup>-1</sup> and a standard deviation of 0.06 m<sup>2</sup> g<sup>-1</sup>. The 851 852 regional integrated MAC<sub>BrC+lensing 660</sub> is even more stable with an average value of  $0.16 \text{ m}^2 \text{ g}^{-1}$  and a standard deviation of 0.01 m<sup>2</sup> g<sup>-1</sup>. Similar to the results at 405 nm, we observe that the MAC in these very aged plumes is very similar to 853 the average MAC observed in the near field. 854 855

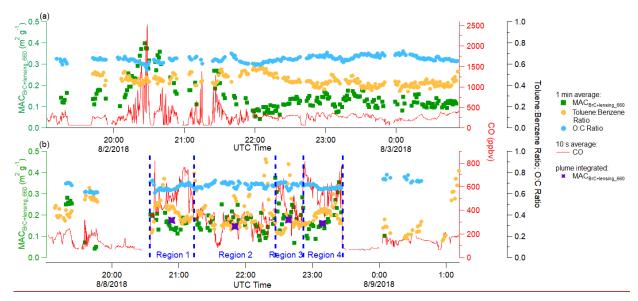


Figure 9: Time series of plume properties during (a) RF05, and (b) RF08(Central Valley of California). Different square and round markers indicate 1 min averages of different variables as shown in the legend, and the red solid line represents 10 s averages of the mixing ratio of CO. Purple stars in RF08 indicate region integrated MAC<sub>BrC+lensing</sub> 660 (individual regions are separated based on the concentration of CO, and indicated by blue dashed lines).

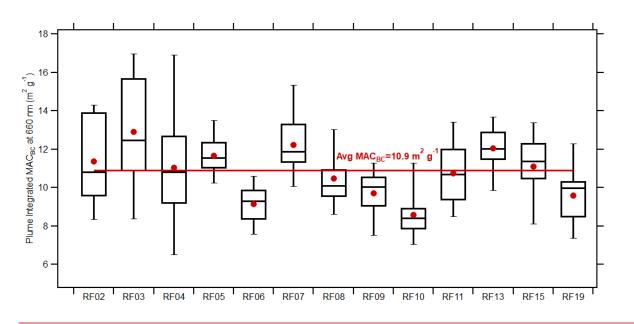
Normalized Abs<sub>BrC+lensing\_660</sub> and total scattering coefficient at 660 nm (Fig. S10), as well as normalized 856 Absws BrC660 (Fig. S11) were also investigated to see if they decreased with markers of chemical age similar to the 857 results seen at 405 nm. However, the correlation between these BrC optical properties with O:C ratio or 858 859 toluene:benzene ratio at 660 nm is much weaker and flatter than they are at 405 nm. Perhaps this is because BrC 860 absorption is very small at 660 nm, and a large uncertainty is brought in from the assumptions required for calculation of this property and instrumental uncertainties. The average normalized Abs<sub>BrC+lensing\_660</sub> is 0.02 Mm<sup>-1</sup> ppbv<sup>-1</sup>, which 861 is 5 times lower than the absorption at 405 nm; while the average normalized Abs<sub>ws BrC660</sub> is a order of magnitude 862 lower than Abs<sub>ws BrC405</sub>. The MAC<sub>BrC+lensing 660</sub> (Fig. S12) shows better correlation with BC:OA ratio than 863 MAC<sub>BrC+lensing</sub> 405, though the increasing trend is still not as significant as Saleh et al. (2014) due to a much smaller 864 865 BC:OA ratio during WE-CAN.

# 866 <u>3.3 Relative Importance of BrC vs. the Lensing Effect at 660 nm</u>

867 Plume integrated MAC<sub>BC</sub> at 660 nm (MAC<sub>BC660</sub>) from the 13 WE-CAN research flights with clear plume transects of biomass burning plumes are shown in Fig. 10. The MAC<sub>BC660</sub> discussed in this section is calculated from 868 Eq. 3, and has contributions from absorption from the BC core, the BrC shell, and the lensing effect. Again, even fire 869 plumes from individually named fires are usually a mix of many different burning conditions, and it is hard to identify 870 the exact source in most wildfire smoke measurements, especially for well mixed plumes. Therefore flight-to-flight 871 872 data is analyzed because each flight covered a region, and an overall behavior of absorbing aerosol from wildfire can be provided. MAC<sub>BC660</sub> varies between different flights with RF03 having the highest average MAC<sub>BC660</sub> of 12.9 m<sup>2</sup> 873 g<sup>-1</sup>, and RF10 having the lowest average MAC<sub>BC660</sub> of 8.6 m<sup>2</sup> g<sup>-1</sup>. Even in highly aged plumes with emissions mixed 874 from multiple fires (RF05 and RF08), the MAC<sub>BC660</sub> is similar in magnitude and consistency with an average of 11.3 875

876  $\pm$  1.8 m<sup>2</sup> g<sup>-1</sup>. The average of all plume-integrated MAC<sub>BC660</sub> is 10.9 m<sup>2</sup> g<sup>-1</sup>, with a standard deviation of 2.1 m<sup>2</sup> g<sup>-1</sup>. This result is similar to some other recent airborne measurements. Subramanian et al. (2010) reported a MAC<sub>BC660</sub> of 877 878  $10.9 \pm 2.1 \text{ m}^2 \text{ g}^{-1}$  using a SP2 and PSAP operated during the MILAGRO campaign, which included airborne 879 measurements of biomass burning over Mexico. Similarly, Zhang et al. (2017) estimated a MAC<sub>BC660</sub> of 10 m<sup>2</sup> g<sup>-1</sup> 880 utilizing both SP2 and PSAP deployed on the NASA DC-8 research aircraft for the DC3 campaign, which measured 881 the upper tropospheric BC over the central U.S. Taylor et al. (2020) calculated a MAC<sub>BC655</sub> of  $12 \pm 2 \text{ m}^2\text{g}^{-1}$  for biomass burning emissions from Africa over the southeast Atlantic Ocean, using airborne measurements from a SP2 and PAS 882 883 in the CLARIFY-2017 campaign. 884 These results are encouragingly similar given the breadth of measurement techniques (PSAP is filter-based whereas PAS is a direct measurement), geographic regions (Continental U.S. for DC3, Mexico for MILAGRO, African 885 886 outflow for CLARIFY) and altitude in the atmosphere (all were airborne campaigns covering a range of altitudes). If 887 we apply 6.3 m<sup>2</sup> g<sup>-1</sup> as the MAC of a BC core at 660 nm (Bond and Bergstrom, 2006; Subramanian et al., 2010), then 888 the average absorption enhancement for the entire WE-CAN campaign is 1.7. This means the absorption of coated BC 889 is 1.7 times higher than bare BC at 660 nm, which is somewhat close to the factor of  $\sim 2$  reported by laboratory 890 experiments (Schnaiter et al., 2005; Peng et al., 2016), larger than some field measurements (Cappa et al., 2012&2019; 891 Healy et al., 2015), but close to  $1.85 \pm 0.45$  measured by Taylor et al. (2020) in African biomass burning plumes. The 892 similarity to the Taylor et al. (2020) result suggests global similarities in the MAC<sub>BC660</sub> from aerosol emitted from 893 wildfires.

894



<u>Figure 10: Box plots of plume integrated MAC<sub>BC660</sub> for each flight.</u> On each box the central line represents the median, the top and bottom edge represents 75% and 25%, the top and bottom whisker represents 90% and 10%. edges represent the 75<sup>th</sup> and 25<sup>th</sup> percentile, and the top and bottom whiskers represent the 90<sup>th</sup> and 10<sup>th</sup> percentile of the data. The red dot shows the average, and the red line indicates the average value for all plume integrated MAC<sub>BC660</sub>.

895 <u>MAC<sub>BC660</sub> is also compared with the physical age and MCE (Fig. S13), the O:C and toluene:benzene chemical</u>
 896 clocks (Fig. S14), and the altitude, temperature and dilution (ΔCO) (Fig. S15). However, no clear trend is be found in
 897 these comparisons.

898 The average absorption enhancement of 1.7 at 660 nm in this study indicates that, on average, 41% of total absorption at 660 nm is caused by lensing and absorbing organics, instead of BC itself. Figure 11 shows the fraction 899 900 of non-BC absorption from BrC at 660 nm for the biomass burning plumes encountered during WE-CAN using Eq. 901 6-8 with OM calculated from the AMS. The figure is plotted versus plume physical age to allow visualization of the 902 variability, though there is no clear trend with physical age other than perhaps a decrease in variability with increasing 903 physical age. Figure S16 shows a similar result by using OM calculated from the UHSAS. More details on the calculation and the AMS vs. UHSAS methods are explained in section 2.5. Assuming a MAC of the BC core of 6.3 904  $m^2$  g<sup>-1</sup>, BrC contributes roughly the same amount of absorption at 660 nm as lensing (46% from the AMS method, 62% 905 906 from theUHSAS method). This means that 19% (AMS method) to 26% (UHSAS method) of the total absorption at 907 660 nm comes from BrC. When different particle density and WSOM:WSOC ratios are considered (top and bottom 908 whiskers, as well as red and blue dashed lines), the fraction of non-BC absorption is 41-49% for the AMS approach 909 (Fig. 11) and 43-80% for the UHSAS approach (Fig. S16) based on different OM:OC and density. The UHSAS approach shows larger uncertainty because it's sensitive to the particle density when calculating particulate mass 910 911 (Table S1). While there is considerable variability between flights, a rule of thumb that roughly half of the non-BC 912 absorption at red wavelengths is from absorbing organic material seems reasonable. To the best of our knowledge, this 913 is the first observation-based attempt to differentiate between lensing and absorbing organics in the red wavelengths. 914 This approach assumes that water insoluble BrC has the same refractive index as water soluble BrC.

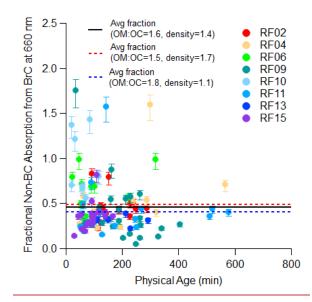


Figure 11: Time evolution of the fraction of non-BC absorption from BrC at 660 nm with AMS and Mie factor. Markers were calculated using a density of 1.4 g cm<sup>-3</sup> and WSOM:WSOC ratio of 1.6. The top whiskers represent sensitive test values using a density of 1.7 g cm<sup>-3</sup> and WSOM:WSOC ratio of 1.5, while the bottom whiskers represent sensitive test values using a density of 1.1 g cm<sup>-3</sup> and WSOM:WSOC ratio of 1.8. The averaged fraction of non-BC absorption from BrC from all the plumes are shown in black solid lines, while the range of this result from sensitivity tests are shown in red and blue dashed lines.

#### 916 <u>4 Conclusion</u>

#### 917 4-Conclusion

In this study, we presented results that help usenable a better understandunderstanding of the ability of 918 919 wildfire aerosol emissions from wildfires to absorb visible light and how those properties change after emission. We 920 presented mass absorption coefficients (MAC) for blackBC and brown carbonBrC from Western United States 921 wildfires measured during the WE-CAN campaign at both short and long visible wavelengths (MAC<sub>BC660</sub>, 922 MAC<sub>BrC+lensing</sub> 660, MAC<sub>ws</sub> BrC660, MAC<sub>BrC+lensing</sub> 405). We also investigated single scattering albedo (SSA), total 923 scattering and total absorption at 660. We observed that the mass absorption coefficient of black carbon stayed 924 relatively constant across all plumes measured and at all physical ages (ages up to 15 hours observed), with an averaged 925  $MAC_{BC660}$  of  $10.9 \pm 2.1 \text{ m}^2$  g + (average  $\pm$  standard deviation). This average showed no variation with altitude or 926 temperature, and we saw no evidence that MAC<sub>BC660</sub> is correlated to MCE. We also investigated the bulk absorption 927 coefficient for BrC and bulk scattering coefficient for total aerosol at both short and long visible wavelengths. General 928 trends that held for all the fire sources are derived, which should be valid throughout the western U.S. given the Even in highly aged plumes with emissions mixed from multiple fires (RF05 and RF08), the MAC<sub>BC666</sub> is similar in 929 magnitude and consistency with an average of  $11.3 \pm 1.8$  m<sup>2</sup> g<sup>-1</sup>. Both the fact that this MAC is significantly larger 930 931 than the MAC for uncoated BC (often cited to be - 6.3 m<sup>2</sup> g<sup>+</sup>) and the fact that the MAC remains relatively constant

932 across different fires and different plume ages are key insights that can improve models of acrossl optical properties
 933 in wildfire emissions.

934 We find that total organic aerosol (OA) and water soluble organic carbon (WSOC) are strongly correlated with markers of chemical age. OA and WSOC (both normalized to CO) decrease with decreasing toluene:benzene 935 ratio or increasing O:C ratio. However, this phenomenon is observed in variations between different fire sources rather 936 937 than during the aging of individual fire plumes. We interpret this variability to mean the fires either had different emission ratios of toluene:benzene and O:C or the smoke underwent rapid secondary chemistry prior to the first plume 938 939 pass in WE CAN. Regardless, the correlations are relatively strong (R<sup>2</sup> of 0.24 to 0.71) and provide a potential link between chemical markers and total organic aerosol amounts across a wide range of fires. While OA and WSOC 940 decrease with decreasing toluene:benzene or increasing O:A, MAC<sub>BC660</sub> actually shows a weak increasing trend with 941 these same markers of aging, showing that while the total amount of organic aerosol is decreasing, the ability of the 942 organic to absorb per mass is staying relatively constant, or even increasing. variety of emissions used to develop them. 943 944 Through a novel use of PILS data, we find that BrC contributes 41 80% of non BC absorption at 660 nm 945 (assuming 6.3 m<sup>2</sup>-g<sup>+</sup>-as the MAC of BC core at 660 nm). BrC contributes, on average, 26% of total absorption, but the absorption cross section of water soluble BrC is relatively small at 660 nm, with a MAC  $_{\rm m}$  BrC660 of 0.06  $\pm$  0.04 m<sup>2</sup> 946 g<sup>-1</sup>, which does not change with physical age. The average MAC<sub>BrC+lensing 660</sub> derived from the PAS (which includes 947 both brown carbon absorption and lensing of black carbon) is  $0.11 \pm 0.06 \text{ m}^2 \text{ g}^4$ . 948

949 In the blue visible wavelengths where brown carbon is more often thought about,  $MAC_{BrC+lensing\_405}$  is  $0.59 \pm$ 950  $0.19 \text{ m}^2 \text{ g}^{-1}$  and shows little variation with physical age or MCE. There are weak increasing trends in all the  $MAC_{BrC}$ 951 data we obtained  $(MAC_{ws\_BrC660}, MAC_{BrC+lensing\_660} \text{ and } MAC_{BrC+lensing\_405})$  with markers of chemical age 952 (toluene:benzene, O:C), while bulk absorption of total aerosol decreases with these same markers of chemical age. In 953 highly aged plumes from multiple fires (RF05 and RF08), the  $MAC_{BrC+lensing\_405}$  has an average value of  $0.63 \pm 0.2 \text{ m}^2$ 954  $g^+$ , suggesting that brown earbon remains significantly absorbing even at relatively longer ages.

955 By utilizing a common parameterization for BrC refractive index from Saleh et al. (2014), with measured 956 inputs for the BC:OA ratio and particle size, we calculated the theoretical MAC<sub>BrC660</sub> and MAC<sub>BrC405</sub>, and they were 957 2.3~3.4 times larger than the measured MAC<sub>BrC+lensing</sub> during WE-CAN. While this discrepancy has been resolved 958 previously by implementing bleaching into model schemes, we show that this is probably the incorrect explanation 959 given the MAC of brown carbon is actually higherBrC either remains constant or slightly increases when chemical markers (O:C, toluene:benzene) suggest more oxidation- has occurred. We suggest a new BrCdifferent 960 961 parameterization of the refractive index is needed to represent wildfire optical properties in the Western United States rather than using bleaching to decrease the mass absorption cross section (MAC) of the Saleh parameterization. We 962 also note that there needs to be better terminology to distinguish between decreasing absorption caused by losses of 963 organic aerosol mass versus decreasing absorption caused by changes in the mass absorption cross section (MAC) of 964 the aerosol. Finally, these results are based on the plume integration method, which might neglect aerosol decay at the 965 966 edge of the plume where oxidation and evaporation are more rapid compared to the center of the plume. While this 967 effect may be important for studying mechanisms of smoke evolution, it does not affect mean properties, which are what ultimately affect climate and are comparable to modeling resultsMAC of the aerosol. 968

969 In the blue visible wavelengths, where BrC is more often thought about,  $MAC_{BrC+lensing_405}$  is  $0.59 \pm 0.19 \text{ m}^2$ 970  $g^{-1}$  and shows little variation with physical age, MCE, altitude, temperature or BC:OA ratio. There isn't any decreasing 971 trends in all the MAC<sub>BrC</sub> data we obtained (MAC<sub>ws BrC405</sub>, MAC<sub>BrC+lensing\_405</sub>, MAC<sub>ws BrC660</sub>, and MAC<sub>BrC+lensing\_660</sub>) 972 with markers of chemical age (toluene:benzene, O:C), but bulk absorption of BrC does decrease with these same 973 markers. In highly aged plumes from multiple fires (RF05 and RF08), the MAC<sub>BrC+lensing\_405</sub> has an average value of 974  $0.63 \pm 0.2 \text{ m}^2 \text{ g}^{-1}$ , suggesting that brown carbon remains significantly absorbing even at relatively longer ages.

We find that total organic aerosol (OA) and water-soluble organic carbon (WSOC) are strongly correlated 975 976 with chemical markers of oxidative age. OA and WSOC (both normalized to CO) decrease with decreasing toluene:benzene ratio and increasing O:C ratio. However, this phenomenon is only clearly observed when data from 977 all the observed fires is included rather than during the aging of individual fire plumes. This could mean that the fires 978 979 either had different emission ratios of toluene:benzene and O:C or the smoke underwent rapid secondary chemistry 980 prior to the first plume pass in WE-CAN. Regardless, the correlations are fairly robust ( $R^2$  of 0.4 to 0.8) given the 981 many variables (MCE, fuel type, etc.) that are changing in the dataset and provide a potential link between chemical 982 markers and total organic aerosol amounts across a wide range of fires. While OA and WSOC decrease with decreasing 983 toluene: benzene or increasing O:C, MAC<sub>BrC</sub> actually shows a weak increasing trend with these same markers of aging, showing that while the total amount of organic aerosol is decreasing, the ability of the organic to absorb per mass is 984 985 staying relatively constant, or even increasing. We also found that the bulk scattering coefficient (normalized to CO) decreases with decreasing toluene:benzene ratio or increasing O:C ratio due to less OA being present, which leads to 986 987 a very different net radiative effect than that which results from just changing the refractive index of BrC.

988 In the red visible wavelengths, where BrC is often less noticed, we observed that the MAC of BC stayed relatively constant across all plumes measured and at all physical ages (ages up to 15 hours observed), with an averaged 989  $MAC_{BC660}$  of  $10.9 \pm 2.1$  m<sup>2</sup> g<sup>-1</sup> (average ± standard deviation), which includes the contribution from both lensing effect 990 991 and absorbing organics. This average showed no clear trends with altitude or temperature, and we saw no evidence that MAC<sub>BC660</sub> is correlated to MCE. Even in highly aged plumes with emissions mixed from multiple fires (RF05 992 and RF08), the MAC<sub>BC660</sub> is similar in magnitude to the near-source plumes with an average of  $11.3 \pm 1.8$  m<sup>2</sup> g<sup>-1</sup>. Both 993 994 the fact that this MAC is significantly larger than the MAC for uncoated BC (often cited to be ~  $6.3 \text{ m}^2 \text{ g}^{-1}$ ) and the 995 fact that the MAC remains relatively constant across different fires and different plume ages are key insights that can 996 improve models of aerosol optical properties in wildfire emissions.

997Through a novel use of PILS data, we find that BrC contributes 41-80% of non-BC absorption at 660 nm998(assuming 6.3 m² g⁻¹ as the MAC of BC core at 660 nm). BrC contributes, on average, 26% of total absorption, but999the absorption cross section of water-soluble BrC is relatively small at 660 nm, with a MAC<sub>ws BrC660</sub> of 0.03 ± 0.02 m²900g⁻¹, which does not change with physical age, and no trend with MCE is observed. The average MAC<sub>BrC+lensing 660</sub>001derived from the PAS (which includes both brown carbon absorption and lensing of black carbon) is  $0.11 \pm 0.06$  m²

 $002 g^{-1}$ .

# 1003 Data Availability

1004 The WE-CAN data can be found at http://data.eol.ucar.edu/master\_lists/generated/we-can/.

- 1005 The DOI for each data set used in this work are:
- 1006 PAS and CAPS PM<sub>SSA</sub>: https://doi.org/10.26023/K8P0-X4T3-TN06
- 1007 PILS1: https://doi.org/10.26023/9H07-MD9K-430D and https://doi.org/10.26023/CRHY-NDT9-C30V
- 1008 PILS2: https://doi.org/10.26023/7TAN-TZMD-680Y
- 1009 SP2: https://doi.org/10.26023/P8R2-RAB6-N814
- 1010 UHSAS: https://doi.org/10.26023/BZ4F-EAC4-290W
- 1011 PTR-ToF-MS: https://doi.org/10.26023/K9F4-2CNH-EQ0W
- 1012 HR-AMS: https://doi.org/10.26023/MM2Y-ZGFQ-RB0B
- 1013 Picarro: https://doi.org/10.26023/NNYM-Z18J-PX0Q
- 1014 miniQCL: https://doi.org/10.26023/Q888-WZRD-B70F

# 1015 Author Contributions

- 1016 SMM designed the project. YS wrote the paper. YS, RPP, APS, EJTL, LAG, DKF, WP, LH, DWT, TC, EVF, and SMM
- 1017 collected and analyzed data.

### 1018 Competing Interests

1019 The authors declare that they have no conflict of interest.

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