

1 **High frequency, continuous measurements reveal strong diel and seasonal cycling of**  
2  **$p\text{CO}_2$  and  $\text{CO}_2$  flux in a mesohaline reach of the Chesapeake Bay**

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11 **Key Points:**

- 12 • Automated  $p\text{CO}_2$  measurements capture daily cycles and anomalous events in estuaries  
13 where  $p\text{CO}_2$  changes rapidly and across a wide range.
- 14 • Rhode River is net autotrophic (Dec-May), net heterotrophic (Jun-Nov), NEP is near  
15 balanced annually, but can reverse status during a single day.
- 16 • Year-round continuous measurements reveal that  $p\text{CO}_2$  and  $\text{CO}_2$  flux are mediated by  
17 temperature effects on biological activity and are inverse to the physical solubility of  
18  $\text{CO}_2$ .

19

20

21 **ABSTRACT**

22 We estimated hourly air-water gas transfer velocities ( $k_{600}$ ) for carbon dioxide in the Rhode  
23 River, a mesohaline subestuary of the Chesapeake Bay. Gas transfer velocities were calculated  
24 from estuary-specific parameterizations developed explicitly for shallow, microtidal estuaries in  
25 the Mid-Atlantic region of the United States, using standardized wind speed measurements.  
26 Combining the gas transfer velocity with continuous measurements of  $p\text{CO}_2$  in the water and in  
27 the overlying atmosphere, we determined the direction and magnitude of  $\text{CO}_2$  flux at hourly  
28 intervals across a 3 yr record (01 July 2018 to 01 July 2021). Continuous year-round  
29 measurements enabled us to document strong seasonal cycling whereby the Rhode River is  
30 primarily autotrophic during cold-water months (Dec–May), and largely net heterotrophic in  
31 warm-water months (Jun–Nov). Although there is inter-annual variability in  $\text{CO}_2$  flux in the  
32 Rhode River, the annual mean condition is near carbon neutral. Measurement at high temporal  
33 resolution across multiple years revealed that  $\text{CO}_2$  flux and apparent trophic status can reverse  
34 during a single 24 hr period.  $p\text{CO}_2$  and  $\text{CO}_2$  flux are mediated by temperature effects on  
35 biological activity and are inverse to temperature-dependent physical solubility of  $\text{CO}_2$  in water.  
36 Biological/biogeochemical carbon fixation and mineralization are rapid and extensive, so  
37 sufficient sampling frequency is crucial to capture unbiased extremes and central tendencies of  
38 these estuarine ecosystems.

39

40 **1. Introduction**

41 Understanding the air–sea exchange of gases and establishing methodologies for accurate  
42 measurements has been a decades-long focus of atmospheric scientists, oceanographers, and  
43 biogeochemists seeking to understand interactions between oceans and the atmosphere and how  
44 these interactions contribute to the global carbon cycle (Broecker et al., 1979; Wanninkhof,  
45 1992, 2013). Coastal oceans and estuaries are ecosystems of interest for understanding the  
46 complex nature and contribution of the land–sea interface to lateral mass transport of carbon  
47 (Abril & Borges, 2005; Cai & Wang, 1998; Frankignoulle et al., 1998; Song et al., 2023) but also  
48 with respect to the role these ecosystems play as both atmospheric  $\text{CO}_2$  sources and sinks (Abril  
49 & Borges, 2005; Chen et al., 2020; Dai et al., 2022; Jiang et al., 2008). The exchange of carbon  
50 dioxide, methane, and other greenhouse gases (GHGs) between Earth’s atmosphere and inland  
51 waters, estuaries, coastal oceans are well-documented but not fully quantified (Abril & Borges,

52 2005; Cai, 2011; Laruelle et al., 2017; Raymond & Cole, 2001; Raymond et al., 2013; Van Dam  
53 et al., 2019). CO<sub>2</sub> evasion from estuaries alone has been estimated at 15–17% of the total CO<sub>2</sub>  
54 input from oceans to the atmosphere (Chen et al., 2020; Laruelle et al., 2017), indicating the  
55 regional and global significance of estuaries (Bauer et al., 2013; Frankignoulle et al., 1998; Jiang  
56 et al., 2008). Yet, there is still great uncertainty surrounding the true net contributions of coastal  
57 oceans, estuaries, and inland water bodies to the atmospheric loading of GHGs (Borges, 2005;  
58 Chen et al., 2020; Herrmann et al., 2020; Joesoef et al., 2015; Laruelle et al., 2017; Raymond et  
59 al., 2013; Van Dam et al., 2019).

60  
61 To better understand the effects of estuaries on atmospheric GHG exchange and accumulation, it  
62 is imperative that we understand their capacity and function as carbon sources and sinks and  
63 ultimately how estuaries factor into the planet's overall global carbon budget (Herrmann et al.,  
64 2020; Laruelle et al., 2017; Van Dam et al., 2019). Many attempts to characterize CO<sub>2</sub> flux in  
65 estuaries and nearshore oceans (Chen et al., 2013; Herrmann et al., 2020; Rosentreter et al. 2021)  
66 have relied on direct measurements using floating domes, tracer gases or, more recently, eddy  
67 covariance methods (Laruelle et al., 2017; Van Dam et al., 2019). Because flux measurements are  
68 time intensive, they tend to be temporally and spatially limited (Herrmann et al., 2020; Klaus &  
69 Vachon, 2020). Using direct flux measurements to derive accurate gas transfer velocity  
70 constants ( $K_0$ , the velocity of gas crossing the air-water boundary) enables models to be  
71 parameterized to estimate  $K_0$  and compute gas flux. Thus, correlative models that incorporate  
72 simultaneous environmental measurements such as wind and/or water velocity, factors that affect  
73 turbulence at the air-water interface and promote gas exchange, have aided in the widespread  
74 accumulation of gas flux estimates (Raymond & Cole, 2001; Van Dam et al., 2019; Wanninkhof,  
75 2014). Gas transfer velocity constant models vary according to the habitat/system being observed  
76 and chemical, physical, and biological factors present in each (e.g., lakes, rivers/streams,  
77 estuaries, and oceans; Herrmann et al., 2020; Ho et al., 2016; Raymond & Cole, 2001; Van Dam  
78 et al., 2019; Wanninkhof, 1992). To reduce uncertainty of computed gas fluxes, it is critical that  
79 the appropriate  $K_0$  models are matched to a targeted ecosystem.

80  
81 Coastal oceans and estuaries are exceptionally complex, frequently characterized by their relative  
82 shallowness and how their freshwater inputs (riverine, surface, and groundwater) mix with salt

83 water (Chen et al., 2020). High nutrient and pollutant loading, due to urbanization and  
84 eutrophication by humans, also have important effects on estuaries and coastal oceans (Freeman  
85 et al., 2019). High spatial and temporal variability are hallmarks of estuaries.

86  
87 Here we present a 3 year data set that combines high frequency (1 min interval) measurements of  
88 dissolved and atmospheric CO<sub>2</sub> with co-located and continuous measurements of salinity, water  
89 temperature, tidal cycling, and wind velocity, recorded at the Smithsonian Environmental  
90 Research Center (SERC) dock, in the Rhode River, Maryland. To estimate hourly, daily,  
91 seasonal, and annual CO<sub>2</sub> flux rates, we applied a CO<sub>2</sub> gas velocity constant model developed by  
92 Van Dam et al. (2019) for the New River, North Carolina. This model is expressly designed for  
93 application to shallow, well-mixed, microtidal estuaries located in the Mid-Atlantic coast of the  
94 United States.

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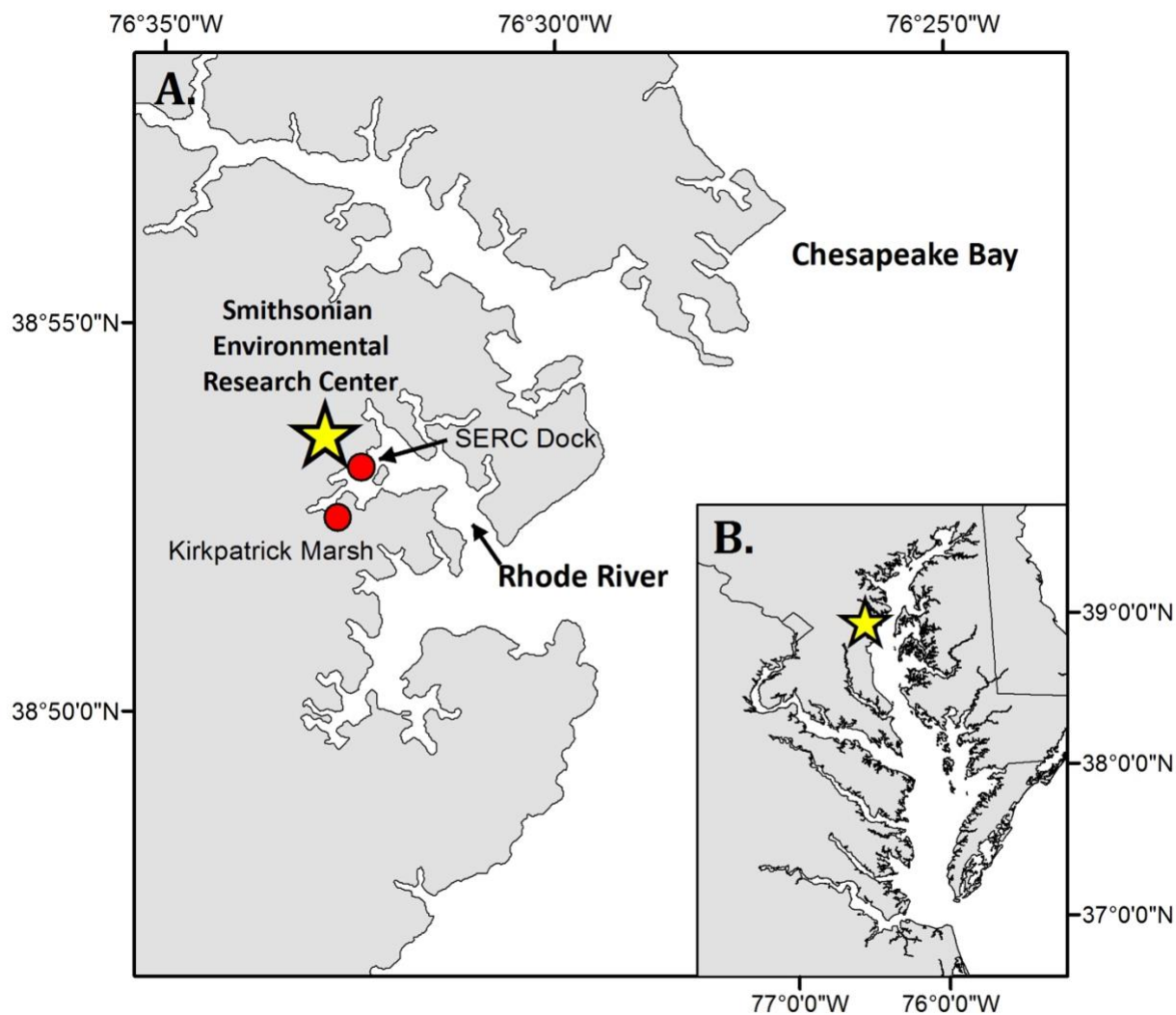
## 96 **2. Methods**

### 97 2.1 Study Location

98 The Rhode River is a tributary and subestuary of the Chesapeake Bay, a drowned river valley,  
99 coastal plain estuary (Fig. 1). The Rhode River has been studied extensively by SERC staff and  
100 colleagues for over 4 decades: nutrient chemistry (Jordan & Correll, 1991; Jordan et al., 1991),  
101 phytoplankton ecology (Gallegos et al., 2010), color dissolved organic matter distribution  
102 (Tzortziou et al., 2008; Tzortziou et al., 2011), and more recently, modeling of dissolved organic  
103 carbon (DOC) input from freshwater and tidal marsh sources (Clark et al., 2020). Located on the  
104 Bay's northwestern shore (38°52'N, 76°32'W), the Rhode River is bounded at its head by Muddy  
105 Creek and at its mouth by the mainstem of the Chesapeake Bay. The Rhode River is a shallow  
106 (mean depth = 2 m, max depth = 4.1 m), mesohaline (0 to 18 ppt), well-mixed, eutrophic  
107 tributary with a length of approximately 5 km; its surface area is approximately 5 km<sup>2</sup> with a  
108 shoreline perimeter of 39 km (Breitburg et al., 2008; Clark et al., 2018). A 0.21 km<sup>2</sup> tidal marsh  
109 (Kirkpatrick Marsh) fringes the estuary at the mouth of Muddy Creek (Fig. 1). Tides are semi-  
110 diurnal with a mean amplitude of approximately 30 cm, but water height can be strongly affected  
111 by wind and weather events. Muddy Creek is the main freshwater source of the Rhode River and  
112 has a maximum flow rate of 10.42 m<sup>3</sup> · s<sup>-1</sup> and mean flow rate 0.18 m<sup>3</sup> · s<sup>-1</sup> (mean flow = 15,552  
113 m<sup>3</sup> · d<sup>-1</sup>; Clark et al., 2020; Clark et al., 2018; Jordan et al., 1986). The mean daily volume of

114 freshwater inflow from Muddy Creek is approximately 0.5% of the mean daily tidal exchange  
115 volume, based on the Rhode River's area and mean tidal amplitude. In the absence of  
116 measurements of the pH or  $p\text{CO}_2$  of the freshwater entering the Rhode River from Muddy Creek  
117 or other lesser freshwater inputs to the estuary, we are unable to report these  $p\text{CO}_2$  or pH values.  
118 However, given the exceedingly small overall volume of freshwater input to the Rhode River  
119 from its surrounding watershed, it is not considered a river-dominated estuary so is not expected  
120 to be substantially influenced by the chemical characteristics of this input. This is not to say there  
121 is no freshwater influence, only that such influences are likely quite local when mixing with far  
122 larger volumes of water from the Chesapeake Bay and therefore beyond the resolution of this  
123 study.

124  
125 Although the Rhode River is a model ecosystem that has been studied intensively for several  
126 decades across many dimensions (Clark et al. 2018; Correll et al., 1992; Gallegos et al., 1992;  
127 Jordan et al. 1991; Rose et al. 2019), no work to date has expressly characterized the nature and  
128 dynamics of  $\text{CO}_2$  flux between the river and the atmosphere.



129  
 130 **Fig. 1.** Location of study site on the Rhode River, Edgewater MD (A), as situated in the  
 131 Chesapeake Bay (B). All  $p\text{CO}_2$  and related water quality values reported were measured from the  
 132 SERC dock, that extends approximately 75 m from shore on Rhode River. Red circles indicate  
 133 location of dock and a tidal creek that drains the Kirkpatrick saltmarsh (marsh area = 0.21 km<sup>2</sup>, 1  
 134 km up estuary from the dock).  
 135

136 2.2 In Situ Measurements, Calculated Parameters and Quantities

137 Continuous, automated environmental measurements were made in and above the Rhode River  
 138 during a 3 year period between 01 July 2018 and 01 July 2021. The purpose of these

139 measurements was to document fluctuations in aqueous  $p\text{CO}_2$  on a fine time scale, from which  
 140  $\text{CO}_2$  flux between the water and atmosphere could be calculated.

141 2.2.1 Aqueous  $\text{CO}_2$  ( $p\text{CO}_{2\text{water}}$ )

142 To measure the  $\text{CO}_2$  gradient ( $\Delta C = p\text{CO}_{2\text{water}} - p\text{CO}_{2\text{air}}$ ) across the Rhode River surface waters  
 143 and its overlying atmosphere, measurements of  $p\text{CO}_2$  were made with a non-dispersive infrared  
 144 (NDIR) detector. In the case of dissolved gas measurements, water was equilibrated continuously  
 145 with a spherical falling film equilibrator (Miller et al. 2019). Water from 1 m below the water's  
 146 surface was pumped and dispersed continuously over a 25.4 cm diameter sphere. The falling film  
 147 created on the sphere generates a gas exchange surface which forces  $\text{CO}_2$  in the equilibrator  
 148 headspace into equilibrium with the water's  $\text{CO}_2$  content (i.e. mole fraction =  $x\text{CO}_2$  ( $\mu\text{mol/mol}$ )).  
 149 Water exits the equilibrator via an airtight drain that prevents headspace contamination from  
 150 surrounding atmospheric air. Headspace gas circulates continuously in a closed loop through the  
 151 equilibrator, water trap and gas dehumidifier, past the NDIR, and back into the equilibrator.  
 152 Experimental observations concluded that spherical falling film equilibrators achieve 99%  
 153 equilibration of  $\text{CO}_2$  within 10–15 mins, depending on whether step changes are from low to  
 154 high or high to low; details of the operation and performance of the falling film equilibrator are  
 155 described in Miller et al. (2019). Measurements were made at 1 min intervals at a pressure equal  
 156 to the ambient barometric pressure.

157  
 158 Measured raw  $\text{CO}_2$  mole fractions ( $\mu\text{mol/mol}$ ) were converted to partial pressures ( $\mu\text{atm}$ ) using  
 159 equation 1. Minute-over-minute values were rounded down to the nearest hour and averaged to  
 160 provide hourly means. The mole fractions were then evaluated with corresponding water  
 161 temperature and salinity measurements following the methodology of Zeebe and Wolf-Gladrow  
 162 (2001) where saturation vapor pressure of water is calculated according to Weiss and Price  
 163 (1980) to determine  $p\text{CO}_{2\text{water}}$ .

164

$$165 \quad p\text{CO}_{2\text{water}} = x\text{CO}_2 \cdot (p - p\text{H}_2\text{O}) \quad (1)$$

166

167 where

168  $p\text{CO}_2$  = partial pressure of  $\text{CO}_2$  of water ( $\mu\text{atm}$ )

169  $x\text{CO}_2$  = mole fraction of  $\text{CO}_2$  in water ( $\mu\text{mol/mol}$ )

170  $p = \text{total pressure} = 1 \text{ atm}$

171  $p_{\text{H}_2\text{O}} = \text{saturation vapor pressure of water } (\mu\text{atm})$

172

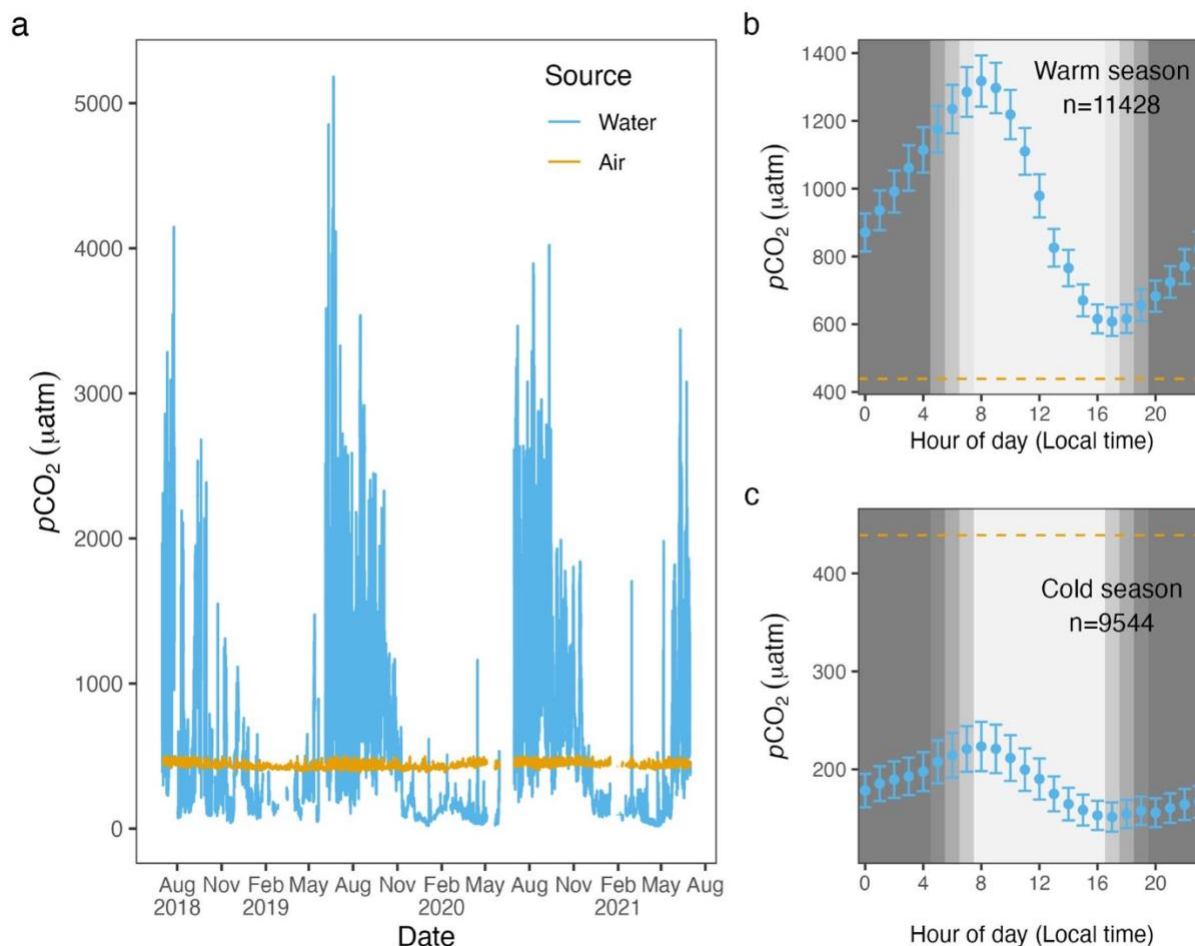
### 173 2.2.2 Atmospheric CO<sub>2</sub>

174 Every six hours, the sample gas stream was automatically diverted with programmed solenoid  
175 control valves from the equilibrator to an atmospheric port located approximately 5 m above the  
176 pier deck. During atmospheric sampling, 15 1-min interval measurements were made. To  
177 account for inaccuracies during the transition period from equilibrator to atmospheric sampling,  
178 the final eight measurements were averaged and the first seven were discarded. Similarly, the  
179 first 30 measurements following switchover from atmospheric port to equilibrator were  
180 discarded, to ensure measurements were fully equilibrated with water. For these atmospheric  
181 measurements, the contribution of the vapor pressure of water to the total atmospheric pressure  
182 of the open-air environment was considered negligible (i.e.  $p_{\text{H}_2\text{O}} = 0$  and  $p = 1$ ), such that  
183  $p\text{CO}_{2\text{atm}} = x\text{CO}_{2\text{atm}}$ . As such, any potential differences are expected to fall well within the  
184 measurement accuracy of the instrument (see below).

185

186 One advantage to using a shared NDIR sensor for aquatic and atmospheric samples is that any  
187 minor effects of instrument drift will be reflected in both data streams, as opposed to two sensors  
188 that drift independently of one another. Likewise, significant and sustained deviation from  
189 typical local atmospheric variability will be captured during atmospheric sampling and can signal  
190 the need for recalibration and assist with QA/QC of corresponding data from both streams. A  
191 disadvantage of using a common sensor for both dissolved and atmospheric CO<sub>2</sub> measurements  
192 is that it results in a mismatch in sampling frequency of the two. With this limitation in mind, we  
193 chose a higher sampling frequency for aquatic measurements to better describe the inherently  
194 higher variability in dissolved CO<sub>2</sub> in water vs. that in the atmosphere (Fig. 2).





195  
 196 **Fig. 2.** Hourly  $p\text{CO}_{2\text{water}}$  (blue) and  $p\text{CO}_{2\text{air}}$  (goldenrod) values from 01 July 2018 to 01 July  
 197 2021 (panel a). The air-water  $\text{CO}_2$  gradient,  $\Delta C = p\text{CO}_{2\text{water}} - p\text{CO}_{2\text{air}}$  describes the directionality  
 198 of gas diffusion. Negative  $\Delta C$  values ( $p\text{CO}_{2\text{water}}$  values falling below goldenrod demarcation)  
 199 represent gas movement from air to water and vice versa. Panels b and c depict mean  $p\text{CO}_{2\text{water}}$   
 200 values (95% CI) for each hour of the day for warm and cold seasons, with the dashed lines equal  
 201 to the mean 3 yr value of  $p\text{CO}_{2\text{air}}$ .  
 202

203 Given the 3 yr time series and strong diel cycling of  $p\text{CO}_{2\text{water}}$  (and dissolved oxygen (DO), see  
 204 Figs. S1 and S2) in the Rhode River, we chose to aggregate aqueous minute-over-minute  
 205 measurements to mean hourly averages. Owing to the relative lack of short-term variability in  
 206 local atmospheric  $\text{CO}_2$  concentrations (Fig. 2), we used linear interpolation to impute  
 207 atmospheric  $\text{CO}_2$  concentrations during hours in between actual readings (6 hr gaps between  
 208 atmospheric measurements), which we assumed to be more realistic and reliable than Last  
 209 Observation Carried Forward (LOCF) methods, where the last observation is repeated for all  
 210 gaps until the next measurement is encountered, a method that has fallen out of favor, especially

211 for environmental time series data (Lachin, 2016). To determine if any inadvertent bias was  
212 introduced by linear interpolation procedure, summary statistics of actual atmospheric readings  
213 and actual readings + imputed CO<sub>2</sub> values were compared statistically. This approach enabled us  
214 to take advantage of >25,000 time points throughout the 3 yr period of observation, providing  
215 hourly resolution. Mean  $p\text{CO}_{2\text{air}} = 437 \pm 20.0 \mu\text{atm}$  (Table 1), variability that falls well within  
216 manufacturer's specifications (see section 2.2.4).

217

### 218 2.2.3 CO<sub>2</sub> gradient ( $\Delta C$ )

219  $\Delta C$  was determined by subtraction,  $p\text{CO}_{2\text{water}} - p\text{CO}_{2\text{air}}$ , where positive  $\Delta C$  values correspond to  
220 higher CO<sub>2</sub> concentrations in the water, tending toward movement from water to air (outgassing  
221 or evasion, where Rhode River = CO<sub>2</sub> source), and negative values that signal CO<sub>2</sub> transport  
222 from air to water (dissolution, where Rhode River = CO<sub>2</sub> sink). Figure 2 shows  $p\text{CO}_{2\text{water}}$  and  
223  $p\text{CO}_{2\text{air}}$  plotted on an hourly basis for the 3 yr period beginning 01 July 2018 and ending 01 July  
224 2021. Across this period,  $\Delta C$  was predominantly negative during cold months and predominantly  
225 positive during warm months when  $p\text{CO}_{2\text{water}}$  tended to reach the highest values of the year, but  
226  $\Delta C$  sometimes reversed sign due to occasional extreme day-time photosynthetic drawdown of  
227 CO<sub>2</sub> (Fig. 2).

228

### 229 2.2.4 Accuracy of CO<sub>2</sub> measurements

230 Estimated accuracy of the spherical falling film equilibrator and NDIR sensor (SenseAir K30,  
231 <https://senseair.com/>) combination were experimentally determined in the lab and found to  
232 measure water equilibrated with known gas concentrations to be within the  $\pm 1\%$  uncertainty  
233 limits of the of certified standard gas mixtures used, and well within the published accuracy  
234 specification of the SenseAir K30 (i.e.,  $\pm 30 \text{ ppmv} \pm 3\%$  of instrument reading). Experimental  
235 analysis by Martin et al. (2017) report even higher accuracy when relative humidity and  
236 atmospheric pressure are controlled for. Details on performance of the spherical falling film  
237 equilibrator, such as accuracy, precision, and time constants can be found in Miller et al. (2019).  
238 Although SenseAir offers automated calibration via long term comparisons to atmospheric  
239 readings, this feature was deactivated. The K30 NDIR was periodically validated using standard  
240 zero CO<sub>2</sub> (nitrogen) and standard certified span gases at intervals of one to two months during  
241 the study period. Although the K30 was never observed to drift beyond its factory specifications,

242 the sensor was occasionally re-calibrated in the lab, and measured values were accepted without  
243 adjustment.

244  
245 CO<sub>2</sub> measurements were downloaded to a database at approximately two-week intervals during  
246 the observation period. Data were graphed and reviewed visually, in combination with twice  
247 weekly observations of equilibrator function recorded in an accompanying notebook. Anomalous  
248 data were flagged and excluded from data analysis (e.g., flooding or clogging events that  
249 interrupted proper equilibration.)

250

### 251 2.3 Co-located water quality and atmospheric measurements

252 This water quality station at the SERC dock is a long-term node of the Maryland Department of  
253 Natural Resources “Eyes on the Bay” Chesapeake Bay tidal water monitoring program, and has  
254 been operated by the SERC since 1986. Water quality and atmospheric data are maintained by  
255 the MarineGEO Upper Chesapeake Bay Observatory and can be accessed online (Benson et al.,  
256 2023). A YSI EXO2 sonde was positioned 1 m below the water’s surface and in proximity (~2.5  
257 m distance) to the submerged water pump that fed the *p*CO<sub>2</sub> equilibrator. Sonde measurements  
258 were made at 6 minute intervals and aggregated to 1 hr averages. The published accuracy  
259 specifications for the YSI sonde are as follows: temperature: ±0.01 °C (-5° to 35° C); salinity:  
260 ±1% of reading or 0.1 ppt (0–70 ppt); dissolved oxygen: ±0.1 mg/L or 1% of reading (0 to 20  
261 mg/L). Discrete measurements of temperature and salinity were made with a handheld YSI  
262 Professional Plus 2030 with Quattro Cable instrument, with the following specifications:  
263 temperature: ±0.02 °C (-5° to 70° C); salinity: ±1% of reading or 0.1 ppt (0–70 ppt); dissolved  
264 oxygen: ±0.2 mg/L or 2% of reading (0 to 20 mg/L). Equilibrator temperature was measured  
265 with a probe (EDS model OW-TEMP-B3-12xA) accurate to ±0.5 °C (-10° to 85 °C). Discrete  
266 measurements were routinely compared with the sonde to corroborate measurement agreement.  
267 Wind speed measurements were made using a sonic anemometer (Vaisala WXT-520 weather  
268 transmitter) mounted 7 m above the mean low tide height of the water and located directly above  
269 the *p*CO<sub>2</sub> equilibrator.

270

### 271 2.4 Data Processing

272 Data included in this study span 01 Jul 2018 to 01 Jul 2021.

273 2.4.1 Gas-specific solubility

274 To determine the purely physical effects of temperature and salinity on CO<sub>2</sub> solubility, gas-  
 275 specific solubility values  $K_0$  (mmol · m<sup>-3</sup> · μatm<sup>-1</sup>) were calculated across the 3 yr observation  
 276 period using water temperature and salinity measurements in combination with  $p\text{CO}_{2\text{water}}$  values,  
 277 according to Weiss and Price (1980) at 1 hour intervals.

278

279 2.4.2 Gas transfer velocity estimation ( $k$ )

280 Given the similarities between the Rhode River and New River estuaries (e.g., shallow,  
 281 microtidal estuaries with slow water velocity and strong diel cycles in  $p\text{CO}_2$  and DO), we chose  
 282 to parameterize gas transfer velocity  $k$  (cm · h<sup>-1</sup>) standardized to the unitless Schmidt number  
 283 600 ( $k_{600}$ ) according to the estuary-specific  $k$  parameterization model developed by Van Dam et  
 284 al. (2019). Van Dam et al. (2019) determined that  $k$  correlated with wind speed differently during  
 285 daytime versus nighttime hours (linear vs. parabolic relationships). Wind speed data were  
 286 collected during the 3 yr period from a sonic anemometer located on the SERC dock directly  
 287 above the equilibration system and approximately 7 m above the water's surface at mean low  
 288 tide height. For the analysis, windspeeds were standardized for a height of 10 m following a  
 289 power-law relationship,  $U_{10} = U_7 * (10/7)^{0.15}$  (Saucier, 2003). Following Van Dam et al.,  
 290 wind speed data were binned to 1.5 m s<sup>-1</sup> intervals for day and night readings and raw values  
 291 replaced by the mean wind speed for each bin. The median binned windspeed over the Rhode  
 292 River was 2.2 m s<sup>-1</sup>, regardless of time of day or season. Recorded windspeeds never exceeded  
 293 10m s<sup>-1</sup> and were dominated by much lower values (Fig. S1). Unlike the New River Estuary, the  
 294 Rhode River's windspeed profile does not differ much between day and night, nor across season.  
 295 For this reason, we chose to use the most conservative  $k_{600}$  formulation from Van Dam et al  
 296 (2019), that combines day and night winds to estimate  $k_{600}$ .

297

298 Wind speed was used to parameterize  $k_{600}$  as follows:

299

$$300 \quad k_{600} = 1.5 * U_{10} + 4.2 \quad (2)$$

301

302 where  $U_{10}$  = mean of binned wind speed at 10 m above the water's surface (m · s<sup>-1</sup>).

303

304 2.4.3 CO<sub>2</sub> flux

305 Using continuous, parallel 3 yr records (01 July 2018 to 01 July 2021) of dissolved and  
 306 atmospheric *p*CO<sub>2</sub>, water temperature, salinity, and wind speed (at standard 10m height, *U*<sub>10</sub>),  
 307 CO<sub>2</sub> flux was derived according to the equation:

309 
$$\text{CO}_2 \text{ flux} = k_{600} \cdot K_0 \cdot \Delta C \cdot (600 / Sc)^{-0.5} \quad (3)$$

310 where

311 CO<sub>2</sub> flux = the rate and direction of CO<sub>2</sub> mass moving between water and gas phases  
 312 (mmol · m<sup>-2</sup> · hr<sup>-1</sup>)

313 *k*<sub>600</sub> = gas transfer velocity (cm · hr<sup>-1</sup>), normalized to a common Schmidt number  
 314 (*Sc* = 600)

315 *K*<sub>0</sub> = gas-specific solubility for CO<sub>2</sub> (mmol · m<sup>-3</sup> · μatm<sup>-1</sup>)

316 Δ*C* = air-water concentration gradient (μatm)

317 *Sc* = Schmidt number

318

319 Note: CO<sub>2</sub> flux calculations require conversion from traditional *k*<sub>600</sub> units (cm · hr<sup>-1</sup>) to (m · hr<sup>-1</sup>)  
 320 from Δ*C* units (μatm) to (atm) prior to calculation.

321

322 2.4.4 Day/Night Designation

323 To differentiate daytime from nighttime hours, we used the position of the measurements  
 324 (latitude) in the Rhode River, combined with the local date and time. This approach enabled us to  
 325 uniformly designate various environmental measurements as happening during the day or night  
 326 (R package "LakeMetabolizer", Winslow et al., 2016).

327

328 2.4.5 Seasonality

329 We chose to break the year into two 6 month periods based seasonal water temperature shifts,  
 330 designating June–November as “warm-water months” when water temperatures averaged 23.2 ±  
 331 6.90 °C. (mean ± 1 sd) and December–May as “cold-water months”, 10.9 ± 5.66 °C (Figs. S1 and  
 332 S2).

333

334

#### 335 2.4.6 Effect size

336 Owing to the large number of observations available for comparison in this study, the likelihood  
337 of finding statistically significant results is quite high. Whether such statistical results by  
338 themselves connote practical and informative differences can be difficult to discern. Effect sizes  
339 (Omega-squared,  $\omega^2$ ) were calculated according to two-factor ANOVAs where independent  
340 variables were investigated by season (cold-water vs. warm-water season), day/night period and  
341 the interaction of season and day/night. The independent variables compared were:  $K_0$ , CO<sub>2</sub> flux,  
342  $\Delta p\text{CO}_2$ ,  $k_{600}$ ,  $p\text{CO}_{2\text{air}}$ ,  $p\text{CO}_{2\text{water}}$ , and wind speed. To account for temporal autocorrelation and  
343 lack of independence of observations that are typical of environmental time series data, we  
344 corrected for overinflation in the residual mean square used in the effect size calculations by  
345 removing the autocorrelation present within residuals, leaving the white-noise component as the  
346 unbiased estimate of residual variability (Cochrane-Orcutt procedure, R package "orcutt", Spada  
347 et al., 2018).

348

### 349 **3. Results and Discussion**

#### 350 3.1 Daily and Seasonal Cycling of $p\text{CO}_2$

351 Hourly averaged measurements of  $p\text{CO}_{2\text{water}}$  in the Rhode River across three years revealed  
352 strong diel and seasonal cycling (Fig. 2). Mean and maximum  $p\text{CO}_{2\text{water}}$  were significantly higher  
353 in warm-water vs. cold-water months (Table 1). During warm-water months (Jun–Nov) daily  
354 oscillations of  $p\text{CO}_2$  frequently transit from far above to below ambient atmospheric conditions  
355 over the course of the day, only to reverse direction (from low to high) during the nighttime  
356 hours (Fig. 2). During the summer,  $p\text{CO}_{2\text{water}}$  levels sometimes shifted by as much as 4500  $\mu\text{atm}$   
357 in both directions during a single 24 hr period (Fig. 2). This pattern is consistent with  
358 biologically driven cycling whereby very high early morning  $p\text{CO}_{2\text{water}}$  conditions are depleted  
359 by net photosynthetic activity (inorganic carbon fixation) over the course of the day, but high  
360  $p\text{CO}_{2\text{water}}$  is restored by respiration in the benthos and water column at night (Song et al. 2023).  
361 Comparing dissolved oxygen (DO) over the same period, similar harmonic cycling is observed,  
362 but maximums and minimums of  $p\text{CO}_2$  and DO were inversely related (Fig. S1), hallmarks of a  
363 production/respiration driven system (Herrmann et al., 2020; Van Dam et al., 2019).

364

365 **Table 1.** Descriptive statistics comparing seasonality of  $p\text{CO}_2$ ,  $\text{CO}_2$  flux and associated  
 366 parameters in cold-water (Dec–May) and warm-water seasons (Jun–Nov).  
 367

Season	Time Period	Variable	Units	N	Mean	Min	Max	SD
<b>overall</b>	-	<b><math>\text{CO}_2</math> flux</b>	<b><math>\text{mmol} \cdot \text{m}^{-2} \cdot \text{hr}^{-1}</math></b>	<b>20971</b>	<b>-0.09</b>	<b>-4.89</b>	<b>11.18</b>	<b>1.823</b>
cold	day	$\text{CO}_2$ flux	$\text{mmol} \cdot \text{m}^{-2} \cdot \text{hr}^{-1}$	4494	-1.39	-4.89	8.26	1.134
cold	night	$\text{CO}_2$ flux	$\text{mmol} \cdot \text{m}^{-2} \cdot \text{hr}^{-1}$	5050	-1.39	-4.66	5.24	0.927
warm	day	$\text{CO}_2$ flux	$\text{mmol} \cdot \text{m}^{-2} \cdot \text{hr}^{-1}$	6007	1.18	-3.95	11.18	1.731
warm	night	$\text{CO}_2$ flux	$\text{mmol} \cdot \text{m}^{-2} \cdot \text{hr}^{-1}$	5421	0.78	-3.97	8.05	1.467
<b>overall</b>	-	<b><math>K_0</math></b>	<b><math>\text{mmol} \cdot \text{m}^{-3} \cdot \mu\text{atm}^{-1}</math></b>	<b>20971</b>	<b>0.04</b>	<b>0.03</b>	<b>0.07</b>	<b>0.011</b>
cold	day	$K_0$	$\text{mmol} \cdot \text{m}^{-3} \cdot \mu\text{atm}^{-1}$	4494	0.05	0.03	0.07	0.009
cold	night	$K_0$	$\text{mmol} \cdot \text{m}^{-3} \cdot \mu\text{atm}^{-1}$	5050	0.05	0.03	0.07	0.008
warm	day	$K_0$	$\text{mmol} \cdot \text{m}^{-3} \cdot \mu\text{atm}^{-1}$	6007	0.03	0.03	0.06	0.007
warm	night	$K_0$	$\text{mmol} \cdot \text{m}^{-3} \cdot \mu\text{atm}^{-1}$	5421	0.04	0.03	0.07	0.008
<b>overall</b>	-	<b><math>k_{600}</math></b>	<b><math>\text{cm} \cdot \text{hr}^{-1}</math></b>	<b>20971</b>	<b>7.86</b>	<b>5.57</b>	<b>18.36</b>	<b>2.047</b>
cold	day	$k_{600}$	$\text{cm} \cdot \text{hr}^{-1}$	4494	8.71	5.57	16.33	2.251
cold	night	$k_{600}$	$\text{cm} \cdot \text{hr}^{-1}$	5050	7.74	5.57	18.36	2.081
warm	day	$k_{600}$	$\text{cm} \cdot \text{hr}^{-1}$	6007	7.92	5.57	18.36	1.868
warm	night	$k_{600}$	$\text{cm} \cdot \text{hr}^{-1}$	5421	7.20	5.57	18.36	1.751
<b>overall</b>	-	<b><math>\Delta\text{C}</math></b>	<b><math>\mu\text{atm}</math></b>	<b>20971</b>	<b>154</b>	<b>-436</b>	<b>4750</b>	<b>645.8</b>
cold	day	$\Delta\text{C}$	$\mu\text{atm}$	4494	-239	-436	1553	220.9
cold	night	$\Delta\text{C}$	$\mu\text{atm}$	5050	-256	-434	1204	164.2
warm	day	$\Delta\text{C}$	$\mu\text{atm}$	6007	570	-399	4750	745.5
<b>overall</b>	-	<b><math>p\text{CO}_{2\text{air}}</math></b>	<b><math>\mu\text{atm}</math></b>	<b>20971</b>	<b>437</b>	<b>387</b>	<b>500</b>	<b>20.0</b>
cold	day	$p\text{CO}_{2\text{air}}$	$\mu\text{atm}$	4494	430	390	497	16.0
cold	night	$p\text{CO}_{2\text{air}}$	$\mu\text{atm}$	5050	432	387	499	17.8
warm	day	$p\text{CO}_{2\text{air}}$	$\mu\text{atm}$	6007	439	390	499	20.7
warm	night	$p\text{CO}_{2\text{air}}$	$\mu\text{atm}$	5421	443	387	500	21.5
<b>overall</b>	-	<b><math>p\text{CO}_{2\text{water}}</math></b>	<b><math>\mu\text{atm}</math></b>	<b>20971</b>	<b>591</b>	<b>15</b>	<b>5182</b>	<b>651.8</b>
cold	day	$p\text{CO}_{2\text{water}}$	$\mu\text{atm}$	4494	191	15	1982	220.9
cold	night	$p\text{CO}_{2\text{water}}$	$\mu\text{atm}$	5050	176	17	1638	163.9
warm	day	$p\text{CO}_{2\text{water}}$	$\mu\text{atm}$	6007	1009	47	5182	752.6
warm	night	$p\text{CO}_{2\text{water}}$	$\mu\text{atm}$	5421	844	38	4855	632.2
<b>overall</b>	-	<b>wind speed</b>	<b><math>\text{m} \cdot \text{s}^{-1}</math></b>	<b>20971</b>	<b>2.4</b>	<b>0.1</b>	<b>9.8</b>	<b>1.42</b>
cold	day	wind speed	$\text{m} \cdot \text{s}^{-1}$	4494	3.1	0.3	8.9	1.53
cold	night	wind speed	$\text{m} \cdot \text{s}^{-1}$	5050	2.4	0.3	9.1	1.45
warm	day	wind speed	$\text{m} \cdot \text{s}^{-1}$	6007	2.5	0.1	9.8	1.28
warm	night	wind speed	$\text{m} \cdot \text{s}^{-1}$	5421	2.0	0.1	9.1	1.23

368  
 369 On the seasonal timescale,  $p\text{CO}_2$  was consistently lowest and DO highest during cold-water  
 370 months of the year (Dec–May; Fig. S1). Importantly, for both gases the temporal variability (diel

371 cycling; Fig. S2) was most constrained during cold-water months across years, strongly  
372 suggesting that carbon fixation exceeds respiration for prolonged periods (weeks to months). In  
373 contrast, during warm-water months (Jun–Nov), photosynthesis/carbon fixation and respiration  
374 are more evenly balanced, compensating one another over 24 hr periods (i.e., respiration >  
375 productivity at night and productivity > respiration during daylight hours; Fig. 2).

376

### 377 3.2 Air-water concentration gradient = $\Delta C$ ( $\mu\text{atm}$ )

378 When hourly  $p\text{CO}_{2\text{water}}$  and  $p\text{CO}_{2\text{air}}$  values (composed of 4 hourly measurements and 20  
379 interpolated values per day) were plotted across the three years of observation, the diel and  
380 seasonal cycles of  $p\text{CO}_{2\text{water}}$  are evident. As expected, atmospheric concentrations of  $\text{CO}_2$   
381 remained relatively constant compared with aqueous loads. When the mean raw  $p\text{CO}_{2\text{air}}$   
382 measurements (mean = 435.1, 95% CI [434.4, 435.7]) were compared with raw  $p\text{CO}_{2\text{air}}$   
383 measurements + imputed estimates (mean = 435.4, 95% CI [435.2, 435.7]) no statistical  
384 difference was observed, indicating that no substantial bias was introduced by linear  
385 interpolation of atmospheric measurements.

386

387 Although nearshore atmospheric  $\text{CO}_2$  concentrations are expected to vary more than those in  
388 isolated well-mixed atmosphere (e.g., at the Mona Loa Observatory), annual mean values were  
389 consistent and within the published uncertainty of the K30 NDIR sensor, when compared with  
390 global measurements conducted at Mona Loa ([Thoning et al., 2023](#)). Local perturbations (e.g.,  
391 effects of terrestrial photosynthetic drawdown when wind is absent) were apparent in  
392 measurements (Fig. 2) but there were no instances when the measured local atmospheric values  
393 were suspiciously high or low for days on end, as compared with expected global mean  
394 atmospheric values for the time period (i.e., 408–416 ppmv; <https://www.co2.earth/annual-co2>,  
395 [Thoning et al., 2023](#)). This lack of sustained anomalous deviation served as additional  
396 confirmation that the K30 was functioning properly and had not drifted outside its calibration  
397 range. Importantly, given the extreme diel cycling and seasonal variability of the Rhode River's  
398  $p\text{CO}_{2\text{water}}$ , the absolute accuracy necessary for determining year-over-year changes in  
399 atmospheric or ocean  $p\text{CO}_2$  is not a requirement for these  $\text{CO}_2$  flux calculations which rely on  
400 relative differences between water and atmospheric measurements.

401

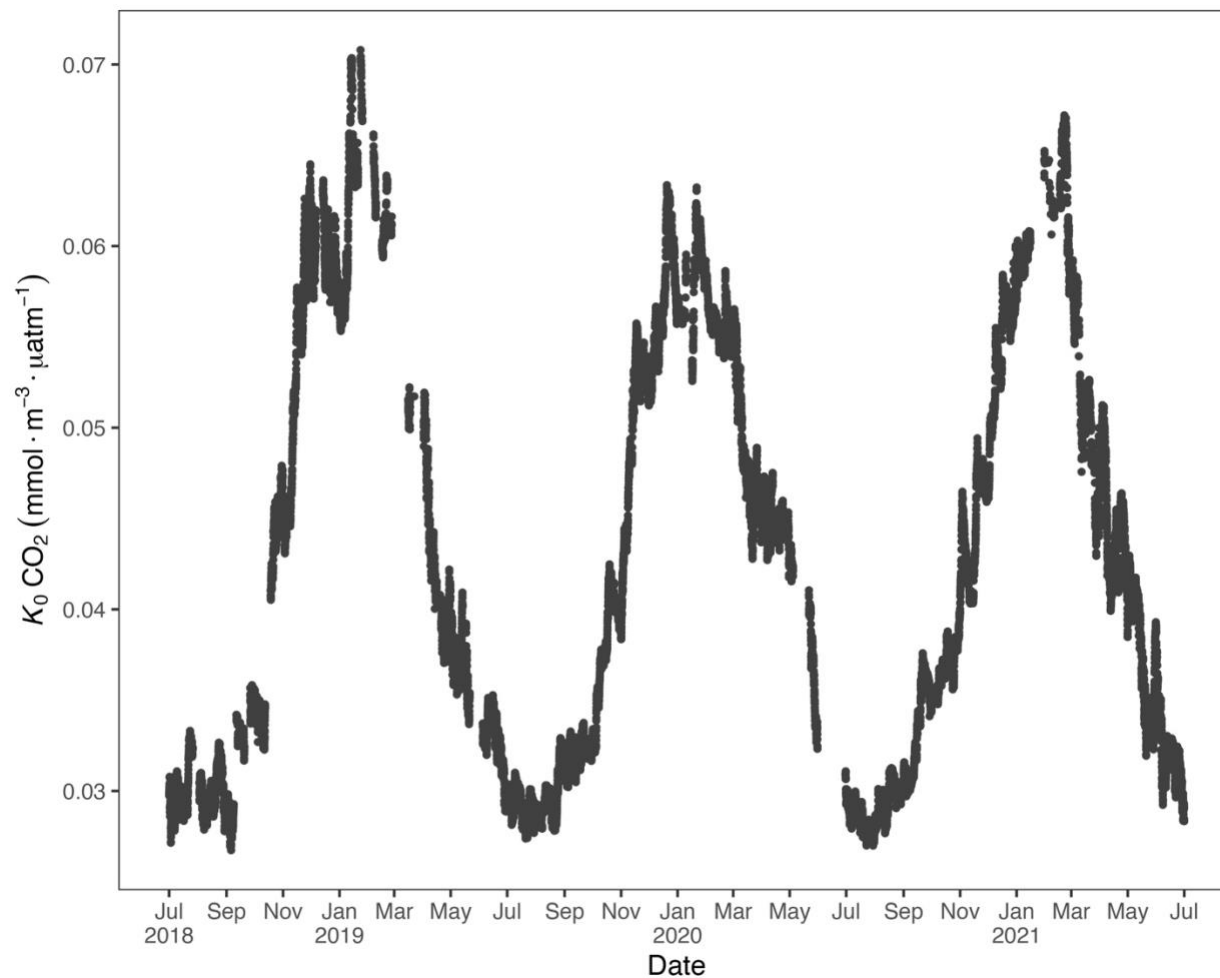


402 Hourly air-water concentration gradient values =  $\Delta C$  ( $\mu\text{atm}$ ) were calculated and plotted across  
403 the three years of study (Fig. 2). During warm months,  $p\text{CO}_{2\text{water}}$  routinely shifts from  
404 supersaturated to sub-atmospheric and back again, over the course of 24 hours (e.g., between  
405  $>2000 \mu\text{atm}$  and  $<410 \mu\text{atm}$  on a single day). These large daily swings in  $p\text{CO}_{2\text{water}}$  produced  
406 concomitant directional reversals of  $\Delta C$  ( $p\text{CO}_{2\text{water}} - p\text{CO}_{2\text{air}}$ ), which result in longer term  
407 averaged gradients (e.g., multi-day, multi-week averages) near zero (Fig. 2). In contrast, most of  
408 the time during cold-water months is spent in a state of sub-atmospheric  $p\text{CO}_{2\text{water}}$  (under-  
409 saturation with respect to overlying atmosphere), resulting in  $\Delta C$  values that are negative and  
410 which promote movement of  $\text{CO}_2$  from the atmosphere into the water.

411

### 412 3.3 Gas-specific solubility ( $K_0$ )

413 To account for the physical effects of temperature and salinity on the solubility of  $\text{CO}_2$  in  
414 estuarine water,  $K_0$  was calculated by methods of Weiss and Price (1980).  $K_0$  varied strongly  
415 across seasons over the 3 yr observation period. The maximum annual range = 0.027 to 0.071  
416  $\text{mmol} \cdot \text{m}^{-3} \cdot \mu\text{atm}^{-1}$ ; mean cold-water months = 0.051 and mean warm-water months = 0.035  
417  $\text{mmol} \cdot \text{m}^{-3} \cdot \mu\text{atm}^{-1}$ , confirming that  $\text{CO}_2$  was most soluble during winter and least soluble in  
418 summer (Fig. 3). This is inverse to observed dissolved  $\text{CO}_2$  values:  $p\text{CO}_{2\text{water}}$  was lowest and  
419 least variable during winter and highest and most variable during summer (Fig. 2, Table 1)  
420 suggesting that solubility, in and of itself, plays only a minor and non-limiting role in  $p\text{CO}_{2\text{water}}$   
421 in the Rhode River. Effect size ( $\omega^2$ ) estimates indicated that the greatest proportion of variability  
422 in  $K_0$  was associated with season, vs. day/night or the interaction of the two (Table 2).



423

424 **Fig. 3.** Gas-specific solubility ( $K_0$ ) for  $\text{CO}_2$  based on water temperature and salinity.

425 Units are  $\text{mmol m}^{-3} \mu\text{atm}^{-1}$  in the Rhode River (01 Jul 2018 to 01 Jul 2021).

426 **Table 2.** Contrast effect sizes based on two-factor ANOVA where independent variables were  
 427 compared by season (cold-water season = Dec–May vs warm-water season = Jun–Nov),  
 428 day/night period and the interaction of the two.  $\omega^2$  is a measure of effect size, estimating the  
 429 proportion of total variance explained by each parameter. Effect sizes were corrected for inherent  
 430 temporal autocorrelation using the Cochrane-Orcutt procedure (Spada et al., 2018).  
 431

Variable	Factor	Effect Size ( $\omega^2$ )
$K_0$	Season	0.0300
$K_0$	Day/Night	0.000575
$K_0$	Season:Day/Night	0.0000140
CO <sub>2</sub> flux	Season	0.415
CO <sub>2</sub> flux	Day/Night	0.00295
CO <sub>2</sub> flux	Season:Day/Night	0.00301
$\Delta C$	Season	0.310
$\Delta C$	Day/Night	0.00501
$\Delta C$	Season:Day/Night	0.00333
$k_{600}$	Season	0.00164
$k_{600}$	Day/Night	0.00269
$k_{600}$	Season:Day/Night	0.0000549
$pCO_{2air}$	Season	0.000137
$pCO_{2air}$	Day/Night	0.0000134
$pCO_{2air}$	Season:Day/Night	0.00000137
$pCO_{2water}$	Season	0.188
$pCO_{2water}$	Day/Night	0.00275
$pCO_{2water}$	Season:Day/Night	0.00191
wind speed	Season	0.00711
wind speed	Day/Night	0.0186
wind speed	Season:Day/Night	0.000182

432

### 433 3.4 Temperature/Biology ratio

434 To independently parse the magnitude of the physical versus biological forcing of  $pCO_{2water}$ , we  
 435 estimated the Takahashi’s Temperature/Biology ratio (Takahashi et al., 2002) to compare the  
 436 influence of temperature and biological activities on  $pCO_{2water}$ . Across the 3 year period, we  
 437 found that just  $26.0 \pm 4.0\%$  (mean  $\pm$  SD) of forcing was attributable to the effect of temperature  
 438 on solubility, confirming that the predominant driver of  $pCO_{2water}$  in the Rhode River is indeed  
 439 biological activity (75%, Table 3). These patterns demonstrate the outsized role that biological  
 440 processes play in shaping  $pCO_{2water}$  in nearshore marine and estuarine ecosystems (Dai et al.,  
 441 2022; Van Dam et al., 2019).

442

443

444 **Table 3.** Takahashi Temperature/Biology Ratio (Eq. 5a From Takahashi et al. 2002).  
 445

Year	N	$\Delta p\text{CO}_2_{\text{bio}}$	$\Delta p\text{CO}_2_{\text{temp}}$	T/B ratio
2018	4416	3193.0	765.8	0.240
2019	8760	3669.8	1019.6	0.278
2020	8784	2772.1	846.0	0.305
2021	4345	2356.1	507.2	0.215
Overall	26305	3701.5	926.4	0.250

446

447

448 3.5 Gas transfer velocity ( $k_{600}$ )

449 Gas transfer velocity is affected by both mass transfer from molecular diffusion driven by  $\Delta C$   
450 (i.e.  $\text{CO}_2$  gradient between water and atmosphere) and momentum transfer linked to external  
451 environmental forces that enhance turbulence at the air-water boundary layer (Ho et al., 2016;  
452 Raymond & Cole, 2001; Van Dam et al., 2019). Van Dam et al. (2019) validated the use of wind  
453 speed at 10 m above the water's surface ( $U_{10}$ ) to estimate gas transfer velocities of  $\text{CO}_2$  that were  
454 standardized to a Schmidt number of 600 ( $k_{600}$ ) by comparing estimated values to  $k_{600}$  values  
455 derived directly from eddy covariance  $\text{CO}_2$  flux measurements. Given the relative uniformity of  
456 wind speed over the Rhode River where median binned  $U_{10}$  windspeed (converted from  $U_7$   
457 measurements) was  $2.2 \text{ m} \cdot \text{s}^{-1}$  regardless of time of day or season, and that maximum values  
458 rarely exceeded  $10 \text{ m} \cdot \text{s}^{-1}$  (Table 1, Fig. S1), we chose to use the most conservative estuarine-  
459 specific parameterization of  $k_{600}$  (Van Dam et al., 2019) (Eq. 2). The mean overall Rhode River  
460  $k_{600}$  value for  $\text{CO}_2$  (mean  $\pm$  SD,  $7.86 \pm 2.05 \text{ cm} \cdot \text{hr}^{-1}$ ) was of comparable magnitude to that of  
461 the New River Estuary, NC ( $9.37 \pm 9.47 \text{ cm} \cdot \text{hr}^{-1}$ ). However, wind speed varied far less on the  
462 Rhode River than the New River estuary and day/night explained more variability in wind speed  
463 than season. Because wind speed directly influenced the formulation of  $k_{600}$  (Eq. 2), the effect  
464 size of day/night is similarly greater than the seasonal effect on gas transfer velocity (Table 2).  
465 Nevertheless, effect sizes ( $\omega^2$ ) indicate that “season” explained at least 10 times more of the  
466 observed variance of  $p\text{CO}_2_{\text{water}}$ ,  $p\text{CO}_2_{\text{air}}$ , air-water concentration gradient,  $\text{CO}_2$  flux, and gas-  
467 specific solubility than “day/night” or their interaction (Table 2). Given the minor freshwater  
468 input and microtidal nature of the Rhode River, we do not believe that lateral water velocity and  
469 bottom turbulence appreciably affect the gas transfer velocity of  $\text{CO}_2$  here, although we did not  
470 investigate possible influences explicitly.

471

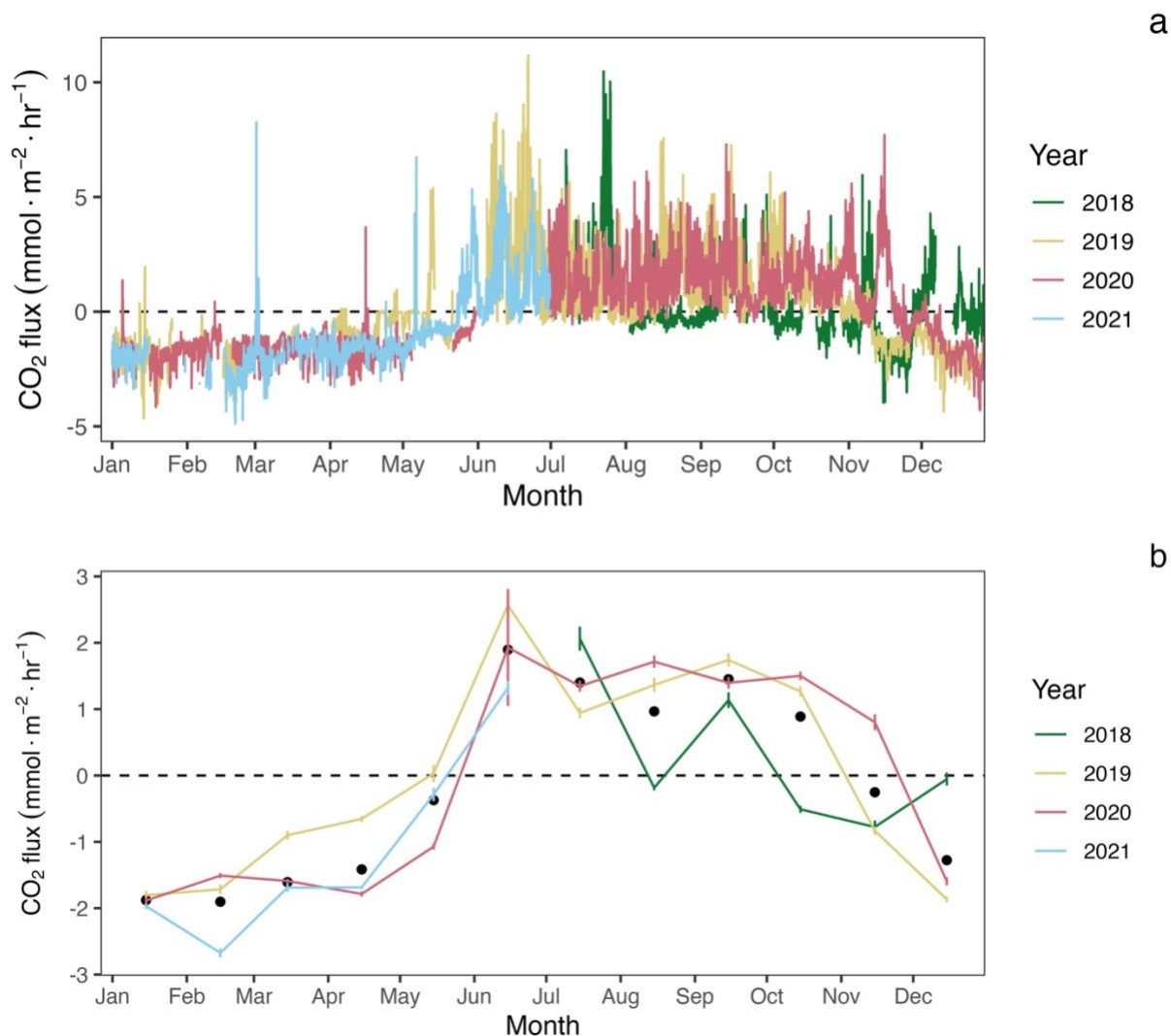
472 Importantly, in coastal marine and estuarine habitats,  $\Delta C$  can shift as much as several thousand  
473  $\mu\text{atm}$  per day due to diel cycling associated with  $\text{CO}_2$  production and depletion (Figs. 2, S2). The  
474 uncertainty surrounding gas transfer velocity parameterization can represent a major source of  
475 error in  $\text{CO}_2$  flux calculations (Frankignoulle et al., 1998; Upstill-Goddard, 2006; Wanninkhof &  
476 McGillis, 1999); however, small errors in  $k_{600}$  have far less effect on  $\text{CO}_2$  flux calculations in  
477 estuaries which experience  $p\text{CO}_2$  swings of several thousand  $\mu\text{atm}$  during a single day, compared  
478 with more stable conditions of the open ocean where interannual ranges of  $p\text{CO}_2$  are typically far  
479 less (Van Dam et al., 2019).

480

### 481 3.6 $\text{CO}_2$ flux - Seasonality and Interannual Variation

482  $\text{CO}_2$  flux was determined according to Eq. 3 using hourly  $\Delta C$  measurements,  $\text{CO}_2$  solubility  
483 values ( $K_0$ ) calculated according to temperature and salinity, and estuary-specific standardized  
484 gas transfer velocities ( $k_{600}$ ) of Van Dam et al. (2019).  $\text{CO}_2$  flux was plotted across the three  
485 years of observations at hourly and monthly intervals (Fig. 4a-b). As observed with  $p\text{CO}_2$ ,  $\text{CO}_2$   
486 flux in the Rhode River was shown to be strongly seasonal. Given the similarity in windspeed  
487 across seasons (Fig. S1), the effect of differential mean  $\Delta C$  and variation between warm- and  
488 cold-water seasons (Fig. 2, Table 1) almost certainly drives the observed seasonal differences in  
489  $\text{CO}_2$  flux (Fig. 4). Again, the specific solubility of  $\text{CO}_2$  is greatest at low temperatures, yet this is  
490 contrary to the observed mean  $p\text{CO}_{2\text{water}}$  patterns, pointing toward a biological mechanism for  
491  $p\text{CO}_2$ ,  $\Delta C$ , and ultimately,  $\text{CO}_2$  flux. The effect size of season on  $\text{CO}_2$  flux was two orders of  
492 magnitude greater than either day/night or the season day/night by interaction (Table 2).

493



494

495 **Fig. 4.** CO<sub>2</sub> flux estimates by year: a. Hourly, b. Monthly average CO<sub>2</sub> flux estimates with 95%  
 496 confidence limits. Black dots in panel b indicate mean monthly fluxes across years.

497

498 Among years,  $p\text{CO}_{2\text{water}}$  and CO<sub>2</sub> flux largely repeat themselves, with dissolved CO<sub>2</sub> becoming  
 499 consistently sub-atmospheric and CO<sub>2</sub> flux going negative (gas exchange from atmosphere to  
 500 water) between December and May and abruptly transitioning to much higher maximum, yet  
 501 variable  $p\text{CO}_{2\text{water}}$  values with net positive CO<sub>2</sub> fluxes from June through November (Figs. 2 and  
 502 4). Monthly averaged CO<sub>2</sub> fluxes are consistent among years (Fig. 4b), with net positive CO<sub>2</sub>  
 503 fluxes (heterotrophic conditions) between June and November and negative (autotrophic) fluxes  
 504 dominating when water temperatures are cold, between December and May. Despite the overall

505 similarities in seasonal CO<sub>2</sub> flux, inter-annual patterns can vary considerably. When hourly CO<sub>2</sub>  
506 flux values were averaged for the year, the Rhode River in 2019 was shown to have a net  
507 positive flux but a net negative flux in 2020. When scaled for the year, 2019 outgassed CO<sub>2</sub> from  
508 the water to the atmosphere at a rate of 2215.08 mmol · m<sup>-2</sup> · yr<sup>-1</sup> (95% CI = 1816.88, 2613.29).  
509 The annual net flux rate in 2020 was negative (i.e. CO<sub>2</sub> moved from the atmosphere into the  
510 river) at a rate of -1361.31 mmol · m<sup>-2</sup> · yr<sup>-1</sup> (95% CI = -1723.60, -999.01).

511  
512 At shorter time scales, such as comparing the same week of the year among years, we sometimes  
513 observed vast differences in the magnitude and direction of CO<sub>2</sub> flux (Fig. S3), signaling  
514 differences in seasonal conditions between years. Transient events can also result in deviations  
515 from otherwise typical CO<sub>2</sub> flux conditions. For example, the period from July 2018 to Jan 2019  
516 deviated from other years as CO<sub>2</sub> flux was more erratic, with intermittent periods of negative and  
517 positive CO<sub>2</sub> flux extending later into the winter season than in other years. When water  
518 temperatures are compared among years, 2018 was shown to be more inconsistent, with more  
519 pronounced temperature shifts and reversals than in 2019 or 2020 (Fig. S1). Salinities remained  
520 relatively low for the latter half of 2018 into early 2019, reflecting wetter conditions (Fig. S1).  
521 There were also two rapid salinity declines (>4 ppt reductions) in July and October 2018, likely  
522 associated with strong precipitation events. These events were both followed by immediate  
523 spikes in chlorophyll-*a* concentration to levels exceeding 200 μg · L<sup>-1</sup>, indicative of  
524 phytoplankton bloom conditions. From 2018 to 2021, chlorophyll-*a* levels of this magnitude and  
525 greater were generally confined to cold-water months (Dec–May; Fig. S1 Erratic water  
526 temperature and salinity are also reflected in more variable gas-specific solubility (*K*<sub>0</sub>) for CO<sub>2</sub> in  
527 2018 than later years (Fig. 3).

528  
529 Gallegos et al. (1992) documented predictable phytoplankton blooms associated with freshets in  
530 the Rhode River, when nutrient-rich freshwater inundates the estuary, not from point and non-  
531 point sources within the local Rhode River watershed, but instead from the enormous watershed  
532 that feeds the Susquehanna River, the primary source of freshwater input into the Chesapeake  
533 above the Potomac as well as >50% of the entire Bay's freshwater (U.S. Geological Survey,  
534 2023). Unlike river dominated estuaries, in the Rhode River estuary, volumetric influxes from  
535 the Chesapeake Bay end member far exceed freshwater input from the Muddy Creek and

536 secondary tributaries. In the Rhode River, phytoplankton blooms result in the temporary  
537 depletion of  $p\text{CO}_{2\text{water}}$ , followed by a spike, as phytoplankton senesce and organic carbon is  
538 decomposed/re-mineralized back into inorganic carbon. Episodic, short-lived occurrences like  
539 these demonstrate how immediate small scale biological forcing, can be coupled with, and  
540 catalyzed by, distant large-scale weather and hydrological events. These in turn can influence  
541  $p\text{CO}_2$  flux variations within seasons and among years (Fig. 3 and S3; and Chen et al., 2020).

542  
543 Overall, except for wind speed, the effect sizes for the other six measured or calculated variables  
544 were shown to be greatest for season vs. day/night or the interaction of season x day/night, and in  
545 all cases the season effect was greater by at least 1 order of magnitude (Table 2). Seasonality has  
546 10 to 1000 times more explanatory power than other variables investigated as estimated by  $\omega^2$   
547 (Table 2).

548

### 549 3.7 Diel Cycling

550 The notion that estuaries are predominantly heterotrophic systems that invariably outgas more  
551  $\text{CO}_2$  to the atmosphere than they absorb has been a long-held view (Abril et al., 2000; Borges et  
552 al., 2004; Cai, 2011; Cai et al., 2000; Chen, 2013; Frankignoulle et al., 1998, Gattuso et al.,  
553 1998). However, more recently investigators have realized that physical and hydrological  
554 characteristics, geographical location, size, and biological and biogeochemical activities may  
555 individually, or together, influence  $\text{CO}_2$  flux in estuaries and therefore contributions to  
556 atmospheric chemistry (Brodeur et al. 2019; Caffrey, 2004; Chen et al., 2013, 2020; Herrmann et  
557 al., 2020). Furthermore, inadequate sampling can induce bias (e.g., upscaling from a small  
558 number of daytime samples taken during warm-water months can skew apparent patterns;  
559 Laruelle et al., 2017; Van Dam et al., 2019.) Using 1 minute sampling intervals, averaged to the  
560 hour, reveals patterns in the Rhode River that might otherwise be overlooked. We document the  
561 Rhode River as having strong seasonality in both  $p\text{CO}_2$  content as well as the extent and  
562 direction of  $\text{CO}_2$  flux (Figs. 2, S1, S2). Both measures are marked by daily oscillations,  
563 frequently reversing direction during a single 24 hr period in warm-water months (Figs. 2) but  
564 more stable and unidirectional during cold-water months (Figs. 2 and 5).

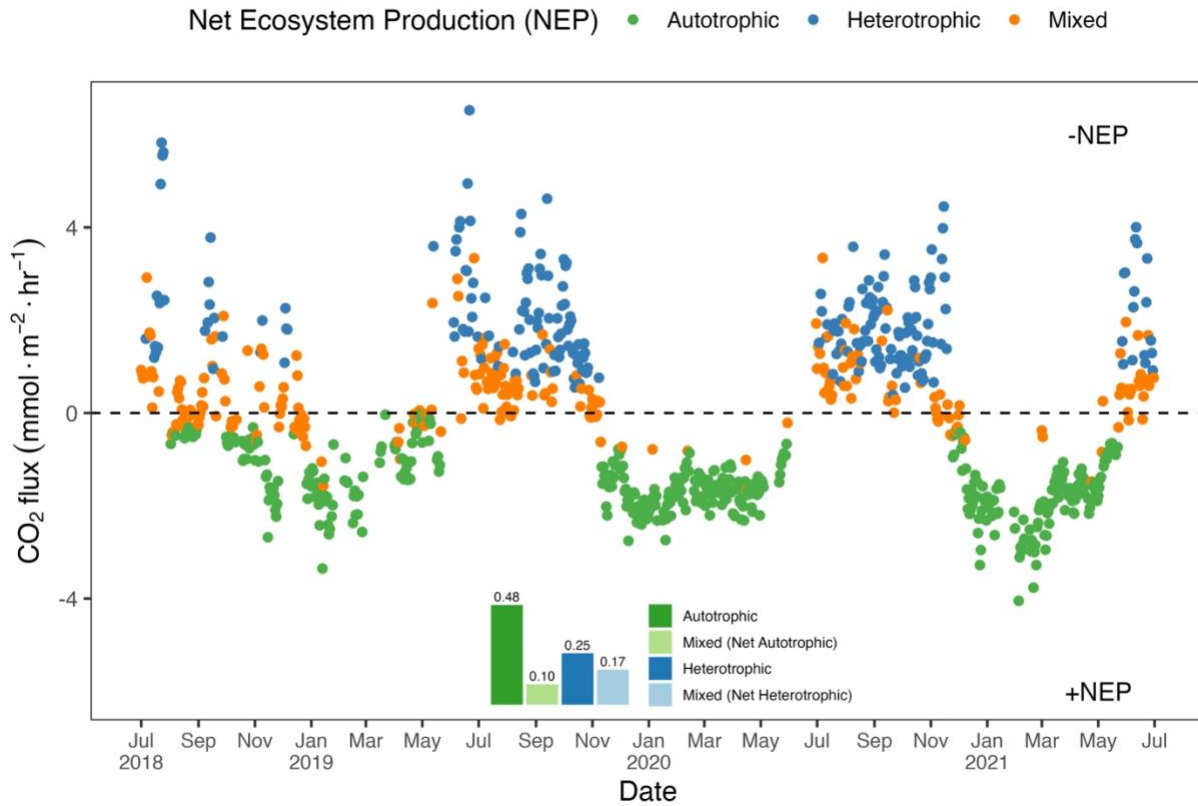
565

### 566 3.8 Shifting Net Ecosystem Production



567 To better understand how the net ecosystem production (NEP) of the Rhode River shifts  
568 throughout the year, where positive NEP indicates the river is storing carbon (autotrophic state)  
569 and negative NEP indicates it is releasing carbon to the atmosphere (heterotrophic state), we  
570 calculated hourly CO<sub>2</sub> flux values, averaged them by day (i.e. 24 hr period) and plotted each in  
571 relation to the  $\Delta C = 0$  reference. Each day of the 3 yr study was categorized as either net  
572 heterotrophic (CO<sub>2</sub> flux from water to atmosphere) or net autotrophic (CO<sub>2</sub> flux from atmosphere  
573 to water). Each day was then further identified as either purely heterotrophic (all 24 hours were  
574 heterotrophic), purely autotrophic, or mixed (some hours were heterotrophic and some were  
575 autotrophic but resulting in a net autotrophic or net heterotrophic state for the day) (Fig. 5). From  
576 July 2018 to July 2021, most 24 hr periods were categorized as pure autotrophic (444/920 =  
577 48%), while 25% (229/990) were purely heterotrophic, and the remainder of mixed trophic status  
578 (17% net heterotrophic and 10% net autotrophic; Fig. 5).  
579

580



581

582 **Fig. 5.** Daily mean CO<sub>2</sub> flux estimates (CO<sub>2</sub> gradient is CO<sub>2water</sub> – CO<sub>2air</sub>). Green dots indicate  
 583 days when all 24 hourly flux measurements were negative (autotrophic with +NEP); blue dots  
 584 indicate days on which all 24 hourly flux measurements were positive (heterotrophic with -NEP).  
 585 Orange dots indicate that hourly fluxes were both negative and positive, and the position of the  
 586 orange dot below or above the zero line indicates whether the day was net autotrophic or net  
 587 heterotrophic. Insert describes the proportion of days in each category indicating that during 58%  
 588 (0.48 + 0.10) of days across three years of observation, the Rhode River was a CO<sub>2</sub> sink.  
 589

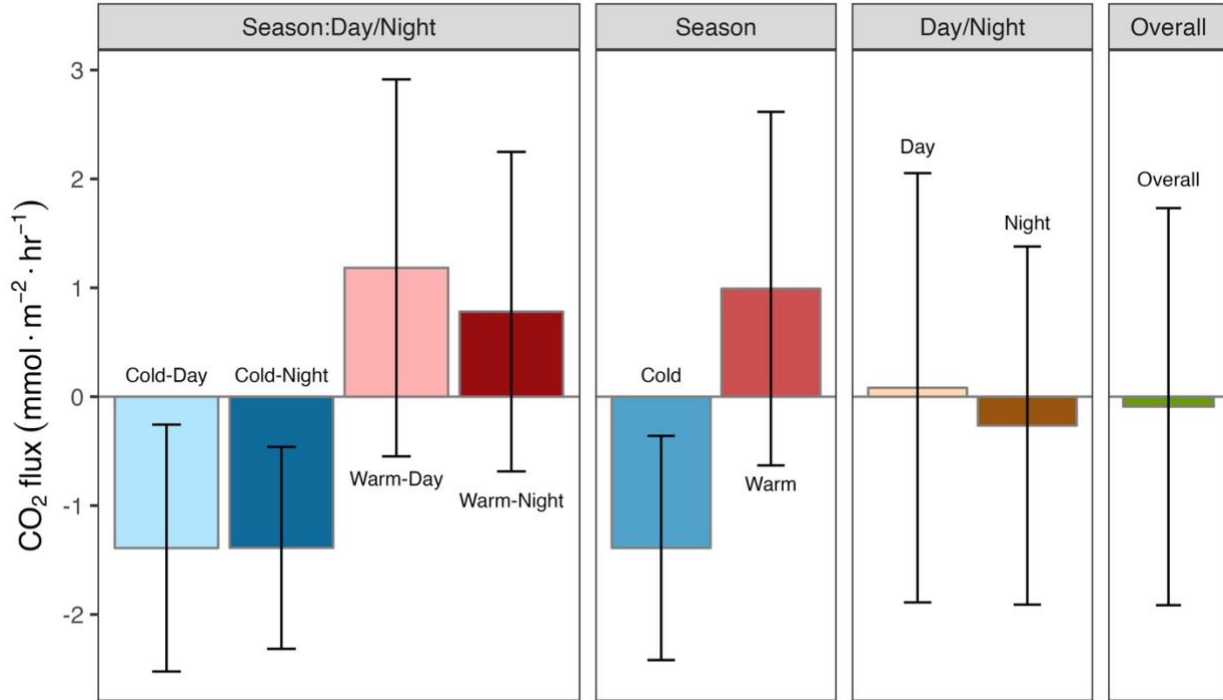
590 Altogether, the Rhode River was net autotrophic for (535 of 920 days = 58%) and net  
 591 heterotrophic for 42% (385 days) across three years. When CO<sub>2</sub> flux is integrated over all three  
 592 years, the Rhode River is shown to have near neutral NEP (Fig. 6). The effect size of season is  
 593 two orders of magnitude greater than either that of day/night or season:day/night interaction  
 594 (Table 2). Mean CO<sub>2</sub> flux values highlight the obvious correlation between season and NEP;  
 595 error bars (± 1 SD) reveal the importance of diel cycling where the magnitude and directionality  
 596 of day/night flux variability is approximately equal to the overall variability accrued across all  
 597 three years (Fig. 6). Although CO<sub>2</sub> flux is less variable and more autotrophic during cold-water

598 months than warm-months in the Rhode River, the range of possible values that occur across  
599 night and day, regardless of season, must be taken into consideration to minimize incidental  
600 sampling bias (Figs. 2 and 6).

601  
602 A multi-year investigation of CO<sub>2</sub> flux in the main stem of Chesapeake Bay by Chen et al.  
603 (2020) combined several bay-wide cruises that were distributed across seasons to collect discrete  
604 and underway *p*CO<sub>2</sub> data for CO<sub>2</sub> flux calculations. They concluded that the low salinity upper  
605 bay, which receives large volumes of freshwater from the Susquehanna River, was net  
606 heterotrophic; the mesohaline middle bay was net autotrophic, and the polyhaline lower bay was  
607 near carbon neutral. Chen et al. (2020) characterized Chesapeake Bay, on the whole, as a weak  
608 source of CO<sub>2</sub> to the atmosphere (net flux = 0.73 mol · m<sup>-2</sup> · yr<sup>-1</sup>) but suggested that during wet  
609 years, it may function as weak sink of CO<sub>2</sub>. Herrmann et al. (2020) also concluded that the  
610 Chesapeake Bay was a weak source of CO<sub>2</sub> to the atmosphere based on calculated *p*CO<sub>2</sub> values  
611 from long term pH and alkalinity measurements (net flux = 1.2 mol · m<sup>-2</sup> · yr<sup>-1</sup>mol). Brodeur and  
612 colleagues (2019) examined dissolved inorganic carbon (DIC) and total alkalinity along the  
613 mainstem of the Chesapeake Bay across the year in 2016 and concluded that DIC increases from  
614 north to south and from surface waters to depth, but that seasonal riverine input and biological  
615 cycling were significantly important, concluding that the Bay as a whole was a sink for CO<sub>2</sub>.

616  
617 When our annual mean *p*CO<sub>2</sub> values were compared with the Chen et al. (2020) survey, the  
618 Rhode River was shown to be higher on average and more variable than the mesohaline main  
619 stem of the Bay (591 ± 652 vs. 416 ± 167 μatm), including a substantially greater measured  
620 range (min = 15, max = 5182 μatm vs. 103 and 1033 μatm). These results suggest that water in  
621 the shallow and well mixed Rhode River, and DIC in particular, undergo more acute biological  
622 transformation than in the mesohaline main stem of Chesapeake Bay. Chen et al. (2020) point to  
623 a variety of factors that affect *p*CO<sub>2</sub> and CO<sub>2</sub> flux in the main stem bay, including temperature,  
624 depth, stratification, and freshwater input volume, some of which may attenuate biological  
625 forcing. Interannual variability was demonstrated in both the Rhode River (some years were net  
626 autotrophic and others heterotrophic, Figs. 4 and 5) and in the mesohaline main stem of the Bay;  
627 however, we attribute interannual variability in *p*CO<sub>2</sub> and CO<sub>2</sub> flux primarily to variation in  
628 water temperature that in turn drives biological activity. We conclude that seasonal variations the

629 Rhode River (and likely similar rivers in the mesohaline portion of the Chesapeake) are  
 630 significant and predictable, closely associated with water temperature, and that temperature  
 631 mediates NEP biologically rather than by changes to the solubility of CO<sub>2</sub>.



632  
 633

634 **Fig. 6.** Mean CO<sub>2</sub> flux  $\pm$  1 SD ( $\text{mmol} \cdot \text{m}^{-2} \cdot \text{hr}^{-1}$ ) plotted by day/night cycling, cold-water/warm-  
 635 water season, season by day/night interaction, and overall CO<sub>2</sub> flux across three years of  
 636 observation.

637

638 In the Rhode River, we find that CO<sub>2</sub> flux reverses itself daily for part of the year (June–  
 639 November) yielding some days that are characterized as a net sink (net autotrophic) and others  
 640 that are a net source (net heterotrophic). From December to May, diel cycling is minimal and the  
 641 river is almost exclusively a net sink, autotrophic both day and night. Finally, although CO<sub>2</sub> flux  
 642 is pronounced but variable across seasons, the net CO<sub>2</sub> flux of the Rhode River on an annual  
 643 basis is near neutral.

644

645

646

647 3.9 Lateral transport

648 Tidal cycling has been shown to liberate and laterally transport DOC from brackish marshes to  
649 adjacent estuaries (Cai, 2011; Herrmann, 2015) and therefore is of great importance to carbon  
650 cycling and budgets of wetlands and estuaries (Najjar et al., 2020). DOC outwelling from the  
651 Kirkpatrick Marsh (hereafter KPM), a 0.21 km<sup>2</sup> tidal marsh located approximately 1 km up  
652 estuary from our primary study site at the SERC Dock (Fig. 1), into the Rhode River has been  
653 measured and modeled extensively in recent years (Clark et al., 2020; Menendez et al., 2022;  
654 Tzortziou et al., 2011; Tzortziou et al., 2008). These studies indicate that the KPM is responsible  
655 for a large portion of overall DOC input to the Rhode River, as well as significant export from  
656 the river to the mainstem of Chesapeake Bay. Model generation and validation by Clark et al.  
657 (2020) indicate that up to 13.1% of the total DOC input to the Rhode River originates in the  
658 KPM. Another important source (53% of total) is DOC derived from phytoplankton and is  
659 therefore labile and readily biodegraded and remineralized into DIC. Furthermore, large  
660 quantities of other, semi-labile forms of DOC are exported from the KPM, which are themselves  
661 subject to photochemical and biodegradation and remineralization (Clark et al., 2020).  
662 Importantly, each of these DOC streams provides a potential source of DIC, including  $p\text{CO}_2$ , to  
663 the Rhode River.

664  
665 Dissolved inorganic carbon generated in brackish tidal wetlands is also outwelled directly into  
666 estuaries (e.g., Cai et al., 2000; Chu et al., 2018; Song et al., 2023). Recent work by Song et al.  
667 (2023) demonstrates that  $p\text{CO}_2$  in a salt marsh tidal creek in Waquoit Bay, MA was regulated by  
668 both tide height (inversely) and the day/night cycle, with nighttime low tides resulting in the  
669 highest  $p\text{CO}_2$  values, signaling a strong local effect from respiration and photosynthesis in  
670 combination with tidal outwelling.

671  
672 In the Rhode River watershed  $p\text{CO}_2$  was measured continuously in the single tidal creek that  
673 drains the KPM using the same methods as at our primary study location. We observed that the  
674 KPM tidal creek  $p\text{CO}_2$  follows the tidal cycle exclusively, yet outside the mouth of the tidal  
675 creek, in the estuary proper, day/night cycling overwhelms this marsh tidal signal. Simultaneous  
676  $p\text{CO}_2$  measurements from the SERC dock follow a strict day/night cycle (Fig. S4). However,  
677 while peak levels of dissolved  $\text{CO}_2$  in the Kirkpatrick Marsh creek occur at low tide and can  
678 reach values nearly 20 times greater than highs at the SERC dock (Fig. S4) there is no obvious

679 evidence of this tidal DIC input at the dock site. Remineralization of DOC exported from the  
680 KPM, as well as DOC originating in other locations within the watershed are important sources  
681 of DIC in the river, but given the relative volumes of these sources to that of the much larger  
682 estuary, as well as the physical distance (~1 km) from SERC dock, these input signals should be  
683 expected to be lagged and damped inside the estuary and not tightly coupled with tidal cycles.  
684 Instead,  $p\text{CO}_2$  exported from the KPM is expected to undergo significant dilution effects, be  
685 partially off-gassed to the atmosphere, and be metabolized via photosynthesis, reducing its  
686 influence on downstream sites. These findings suggest that despite periodic extreme  $p\text{CO}_2$  in  
687 KPM tidal creek (>30,000 ppmv), the overall mass of  $\text{CO}_2$  export is not sufficient to have  
688 measurable effects on the deeper, well-mixed portions of the Rhode River.

689  
690 Thus, although land – sea interfaces and outwelling of DOC and DIC are important in estuaries  
691 and coastal ecosystems, the relative sizes of wetlands and adjacent water bodies and the overall  
692 volume of water moving between the two are also important factors. In eutrophic estuaries like  
693 the Rhode River, biological forcing can rapidly assimilate DIC and degrade and mineralize labile  
694 forms of DOC, as evidenced by extensive diel cycling in these systems (e.g., Brodeur et al. 2019;  
695 Song et al. 2023, and the present study.) The much larger and complex Chesapeake Bay  
696 generally follows seasonal changes in  $p\text{CO}_2$  and  $\text{CO}_2$  flux, but these appear to be most  
697 predictable in the upper oligohaline portion and the polyhaline region of the bay near the mouth,  
698 where freshwater and oceanic end-member effects are most pronounced (Brodeur et al. 2019;  
699 Chen et al., 2020). The central mesohaline part of Chesapeake Bay comprises numerous discrete  
700 and unique watersheds and subestuaries/ivers, each of which exchanges water with the bay.  
701 Elucidating spatial and temporal patterns of  $p\text{CO}_2$  and  $\text{CO}_2$  flux are vital for understanding each  
702 one's role as an atmospheric source or sink, but also could provide better insight into how each  
703 may be influenced by global increases in atmospheric  $\text{CO}_2$  (i.e., acidification and its influences  
704 on estuarine metabolism, and the local biota, fisheries, and habitats each support.) Collectively,  
705 these and other subestuaries will have cumulative effects on the overall water quality of  
706 Chesapeake Bay, including cycling of DOC and DIC, which in turn affect  $p\text{CO}_2$  and  $\text{CO}_2$  flux.

707

708

709 **4. Conclusion**

710 As indicated in this study and others, the role that biological processes play in estuaries to either  
711 fix CO<sub>2</sub> (autotrophy) or liberate CO<sub>2</sub> (heterotrophy) are extensive, complex, and can be quite  
712 variable over space and time (Brodeur et al. 2019; Chen et al., 2020; Herrmann et al., 2020;  
713 Rosentreter et al., 2021). High frequency automated measurements revealed strong seasonal  
714 contrasts in dissolved CO<sub>2</sub> content and CO<sub>2</sub> flux between water and atmosphere of the Rhode  
715 River, a shallow mesohaline reach of the Chesapeake Bay. Importantly, only through high  
716 frequency, multi-year measurements could diel and seasonal cycling be fully discerned. The  
717 timing and frequency of measurements are critical and have potential for strong and misleading  
718 biases if sampling is insufficient. In contrast, cold-water months coincide with long periods  
719 (weeks to months) of continuous sub-atmospheric sink conditions for CO<sub>2</sub>. Using these  
720 measurements, we estimated the direction and magnitude of CO<sub>2</sub> flux in hourly, daily, and  
721 annual terms. In the Rhode River CO<sub>2</sub> flux reverses itself daily for part of the year (June through  
722 November) yielding some days that are characterized as net sink (net autotrophic and NEP > 0)  
723 and others that are net source (net heterotrophic and NEP < 0). From December to May diel  
724 cycling is minimal, and the river is almost exclusively a CO<sub>2</sub> sink with +NEP both day and night.  
725 Although CO<sub>2</sub> flux is pronounced but variable across seasons, the net CO<sub>2</sub> flux of the Rhode  
726 River on an annual basis is near carbon neutral, although some years are net heterotrophic and  
727 others net autotrophic.

728  
729 High frequency sampling of *p*CO<sub>2</sub>, although typically confined spatially, is one approach to  
730 understanding fundamental aspects of estuarine metabolic states and CO<sub>2</sub> flux that may  
731 otherwise go undetected (Song et al., 2023). To address the spatial complexity of estuarine,  
732 nearshore, and inland waters, more observation locations are required. As with any  
733 environmental or ecological question, careful sampling design is critical to balance efficiency  
734 and statistical power.

735  
736 As the largest and arguably most complex estuary in the United States, the Chesapeake Bay is  
737 the subject of extensive ecosystem management efforts and ranks among the most studied and  
738 monitored estuaries in the world (Boesch and Goldman 2009). Yet, information on CO<sub>2</sub> and  
739 GHG fluxes continue to be limited (Brodeur 2019; Chen et al., 2020; Herrmann et al., 2020).  
740 Given the extensive coordinated monitoring programs that either make real-time water quality

741 measurements and/or maintain routine water sampling schedules (e.g., Maryland DNR “Eyes on  
742 the Bay” program) in this region, existing water quality observation assets and sampling  
743 programs could be leveraged to more fully characterize and quantify CO<sub>2</sub> and other GHG  
744 dynamics and flux in the Bay and elsewhere (see Saba et al. 2019). For example, coordinated  
745 deployment of additional automated sampling devices (e.g., robust air-water equilibrators and  
746 traditional atmospheric gas sensors) in key locations would enable estimates of CO<sub>2</sub> flux, and if  
747 combined with pH, DIC, or total alkalinity measurements, carbonate chemistry calculations as  
748 well. Importantly, such installations need not be permanent. Instead, a small group of  
749 instruments could be systematically deployed across an existing observation network, co-located  
750 with other water quality instruments using a stratified sampling approach to capture spatial  
751 variability. For example, a set of shifting two to four week deployments during summer and  
752 winter months could yield sufficient data to advance our understanding of Chesapeake Bay-wide  
753 CO<sub>2</sub> flux significantly in a single year. Such information would complement underway transects  
754 that are vital, but which tend to underestimate temporal variability in any given location. In the  
755 case of dissolved GHGs, liquid-air equilibration techniques are being used to measure multiple  
756 GHGs (Call et al. 2015; Hartmann 2018; Gülzow et al. 2011; Miller et al. 2019; Xiao et al.  
757 2020).

758  
759 Understanding the GHG dynamics in estuaries is a vital component to generating accurate global  
760 budgets (Maher & Eyre, 2012) as well as informing where emerging carbon capture  
761 technologies, including nature-based solutions, might be best located (Bradshaw & Dance, 2005;  
762 Sun et al., 2021). In the case of estuaries, there have been extensive global losses of seagrasses  
763 due to habitat degradation, pollution, and disease (Waycott et al. 2009). In addition to many  
764 other ecosystem service benefits, restoration of seagrass and submerged aquatic vegetation has  
765 the potential to restore and enhance natural carbon sequestration (i.e. blue carbon; Kennedy et al.  
766 2022; Macreadie et al. 2022; Unsworth et al. 2022). In Virginia, U.S.A., Oreska et al. (2020)  
767 demonstrated how the functional benefits of a restored seagrass meadow habitat can be  
768 quantified ecologically in terms of their ability to sequester carbon and affect GHG fluxes  
769 between the estuary and atmosphere. Uniquely, these investigators then monetized the costs and  
770 benefits of habitat restoration and function as CO<sub>2</sub> offset credits, as part of a GHG budget, and



771 demonstrated how such approaches can be used to incentivize habitat restoration (Oreska et al.  
772 (2020).

773  
774 Increasing the completeness and utility of global GHG budgets, as they relate to human activities  
775 and ecosystem functions, are necessary steps toward combating global climate change.  
776 Measurement of GHGs at high spatial and temporal resolution using economical, automated  
777 measurement solutions can increase our understanding of GHG dynamics at small ecologically  
778 significant scales, as well as at the larger ecosystem level of an estuary.

779

#### 780 **Data Availability**

781 Hourly means of  $p\text{CO}_2$  and associated environmental data used in the analyses are available at  
782 the Smithsonian Figshare repository <https://doi.org/10.25573/serc.22491655> via under Creative  
783 Commons license [CC BY-NC 4.0](https://creativecommons.org/licenses/by-nc/4.0/).

784

#### 785 **Author Contributions**

786 AWM contributed to project Conceptualization, Funding acquisition, Investigation,  
787 Methodology, Project Administration, Resources, Supervision and Writing – Original Draft.  
788 JRM contributed to Data Curation, Formal Analysis, Software and Visualization. ACR  
789 contributed to Data Curation, Investigation, Methodology and Project Administration. MSM  
790 contributed to Conceptualization, Supervision and Visualization. KJK contributed to  
791 Conceptualization, Data Curation, Software, Validation. All authors contributed to Writing –  
792 review and editing.

793

#### 794 **Competing Interests**

795 The corresponding author has declared that none of the authors has any competing interests.

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799

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805

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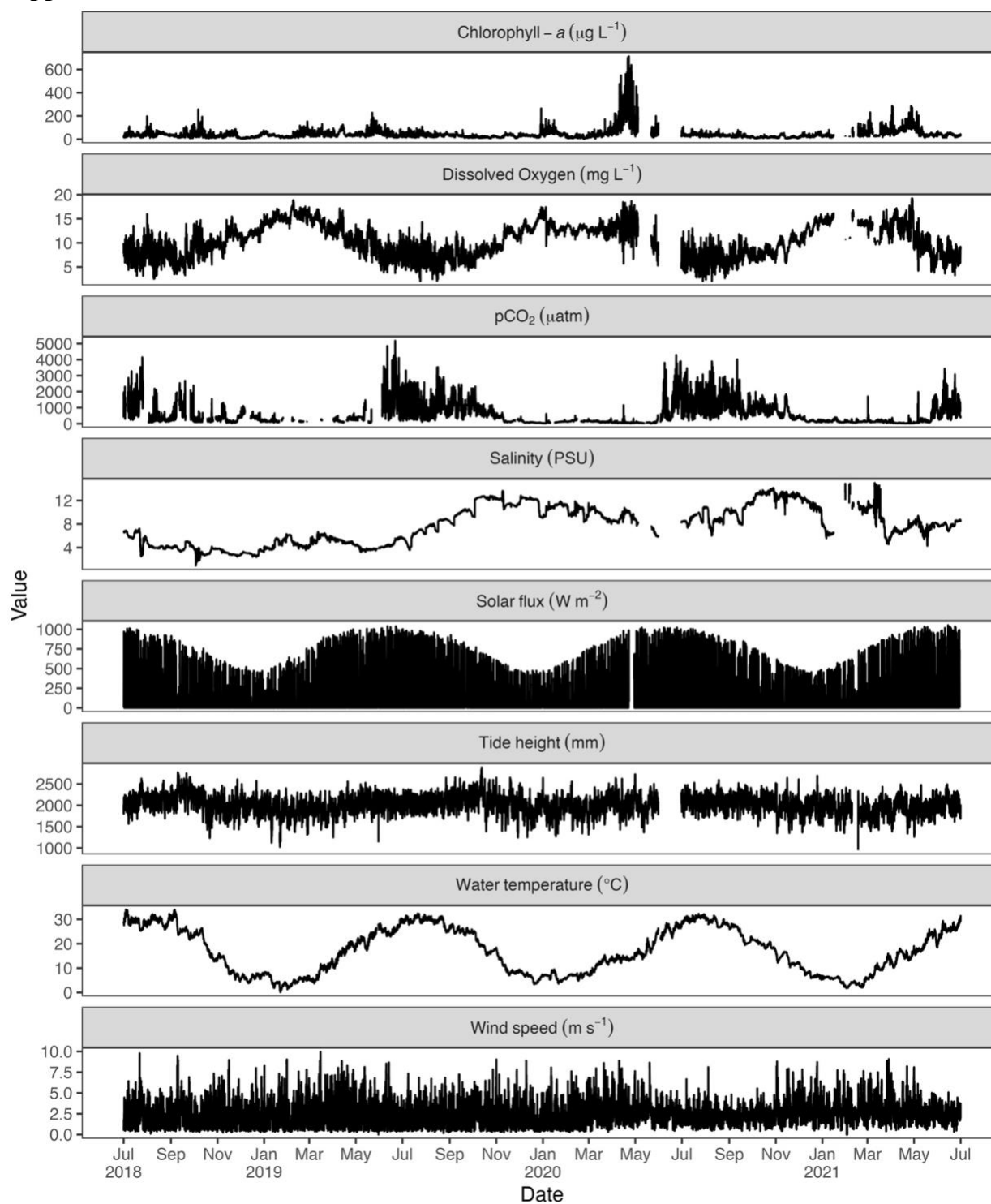


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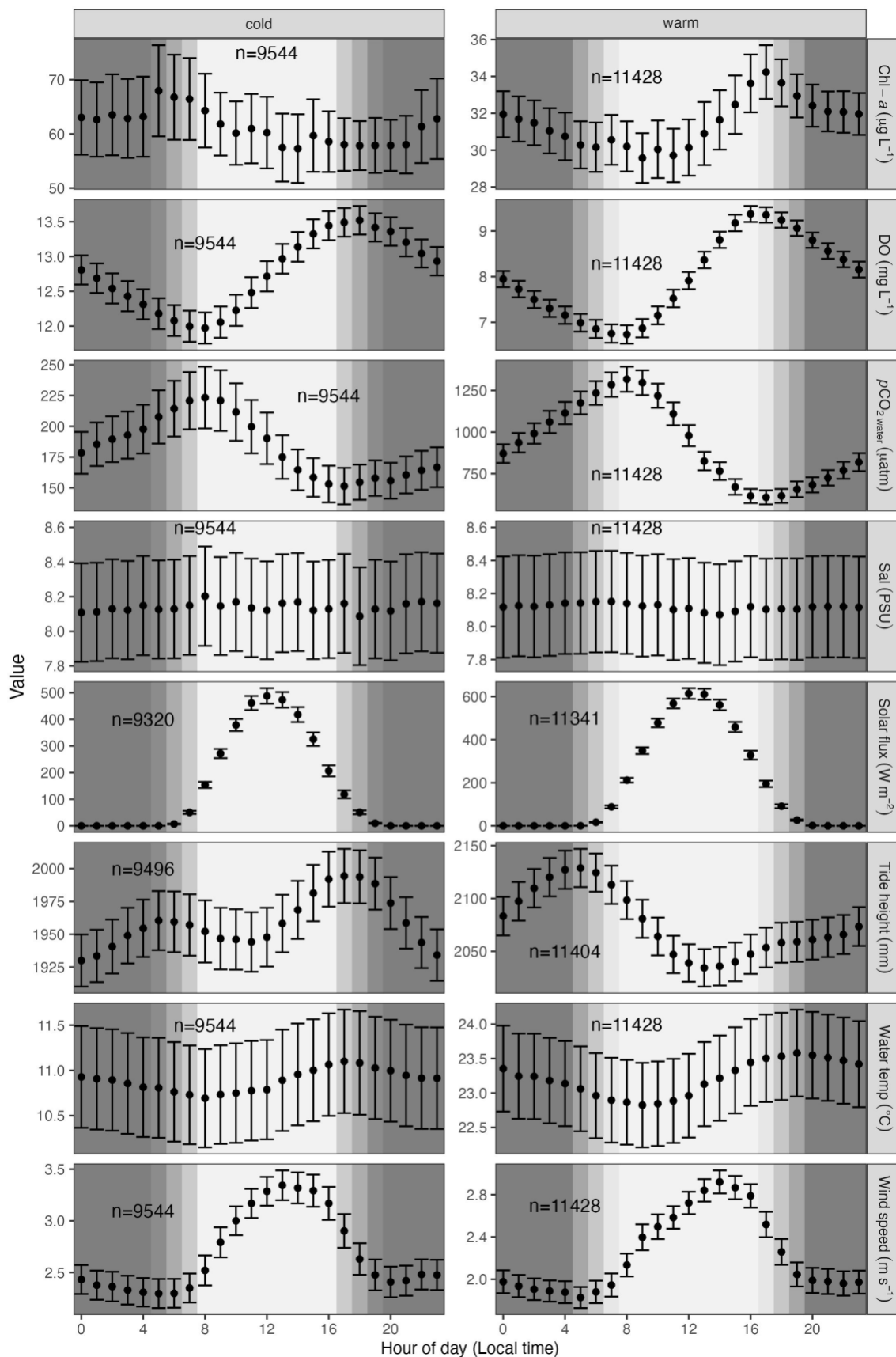
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1061 **Supplemental**

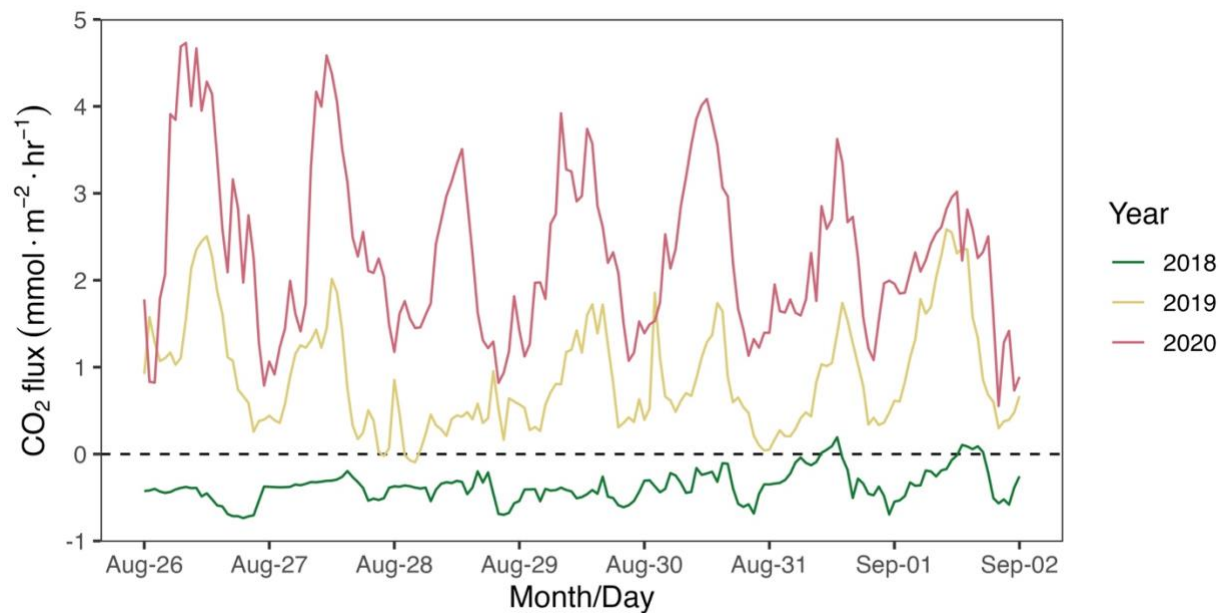


1062 **Fig. S1.** Plot of all raw values from environmental variables for the same time period as CO<sub>2</sub> flux  
 1063 (July 2018–July 2021).  
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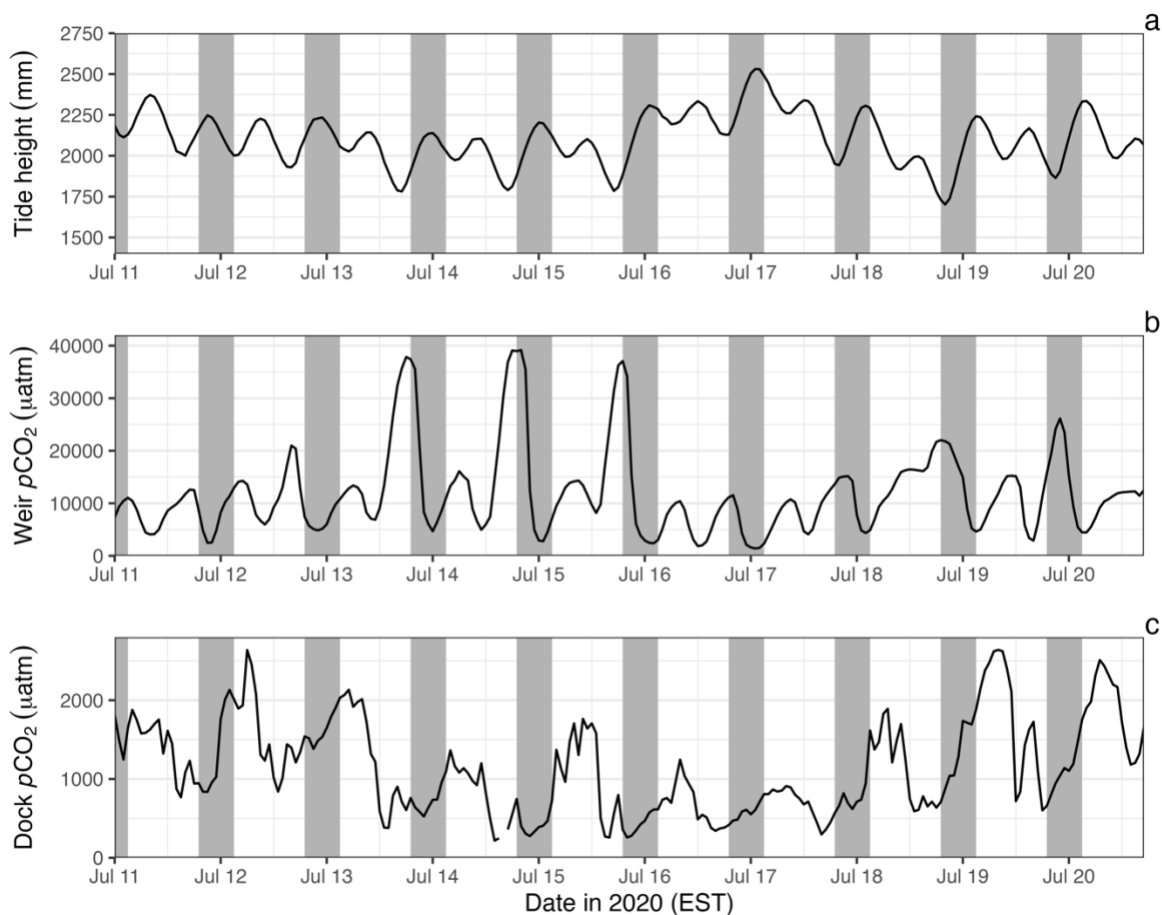


1066 **Fig. S2.** Average hourly values (95% CI) of environmental variables across 24 hours of the day  
 1067 (July 2018–July 2021) in cold and warm seasons. Light/dark background indicates day/night  
 1068 conditions.

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1071 **Fig. S3.** Hourly CO<sub>2</sub> flux estimates for the week of August 26 to September 2 where CO<sub>2</sub> flux  
 1072 status differs among years.  
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1075  
1076 **Fig. S4.** Simultaneous  $p\text{CO}_2$  measurements (1 hr intervals) from SERC dock (panel c) and the  
1077 mouth of the single tidal creek that drains the Kirkpatrick Marsh (panel b) (11–20 Jul 2020)  
1078 indicate that dissolved  $\text{CO}_2$  varies at the dock according to a day/night cycle while  $\text{CO}_2$  in the  
1079 marsh tidal creek rises and falls inversely with tide height (panel a), indicating outwelling of  
1080 marsh derived  $\text{CO}_2$  (e.g., root respiration, pore and groundwater).