

20 understanding of how terrestrial input affecting the marine environment near the coast.

21 I think the manuscript is suitable for publication in the ACP after a revision.

22 **Reply:** We sincerely appreciate the reviewer's valuable comments and suggestions,
23 which help us improve the manuscript. Following the reviewer's comments, we made
24 a revision to the manuscript. All changes in the revised manuscript are highlighted in
25 blue. We also provide below a point-to-point list regarding the changes made in the text
26 for the reviewer's convenience to review.

27 Major comments or suggestions:

28 1. The positive matrix factorization model is the methodology employed by the authors
29 to deconvolute the source factors of atmospheric NMHCs and assess their respective
30 contributions. The authors should provide a more detailed exposition on the principles
31 of this approach in Method section to enhance clarity and understanding.

32 **Reply:** Thanks for your valuable suggestion. Details for the PMF model have been
33 supplied in the manuscript, as indicated below:

34 Line 221-245:

35 "PMF model introduced in detail in the study of Paatero and Trapper (1994) was
36 applied to analyze the data of atmospheric NMHCs in the Yellow Sea and the East China
37 Sea. Based on a matrix consisting of the concentrations of diverse chemical species, the
38 objective of PMF is to determine the number of NMHCs source factors, the chemical
39 composition profile of each factor, and the contribution of each factor to species. The

40 matrix representation of this model is as follows:

$$41 \quad x_{ij} = \sum_{k=1}^p g_{ik} f_{kj} + e_{ij} \quad (9)$$

42 Where x_{ij} represents the concentration of species j measured on sample i , p denotes
43 the number of factors facilitating the samples. f_{kj} represents the concentration of
44 species j in factor profile k , g_{ik} denotes the relative contribution of factor k to sample
45 i , and e_{ij} represents the PMF model error of species j measured on sample i . The
46 factors resolved by PMF are typically interpreted as sources. The objective of this
47 algorithm is to find the values of f_{kj} , g_{ik} , and p that best reproduce x_{ij} , continuously
48 adjusting f_{kj} and g_{ik} until the minimum Q value for a given p is attained. Q is
49 defined as:

$$50 \quad Q = \sum_{i=1}^n \sum_{j=1}^m \left(\frac{e_{ij}}{\sigma_{ij}} \right)^2 \quad (10)$$

51 Where σ_{ij} represents the uncertainty of the concentration of the species j in sample i ,
52 n is the number of samples, and m is the number of species. In applying the PMF
53 model, the significance of missing data in the matrix was decreased by using the species
54 median. The uncertainty for normal data was estimated as 20 % of the NMHCs
55 concentrations because the analytical uncertainty was not available (Buzcu and Fraser,
56 2006). The model ran 20 times and we selected the result with the minimum Q value.
57 Besides, approximately 94 % of the scaled residuals given by PMF ranged from -3 to 3
58 (Fig. S1), suggesting a reasonable fit of the model result."

59

60

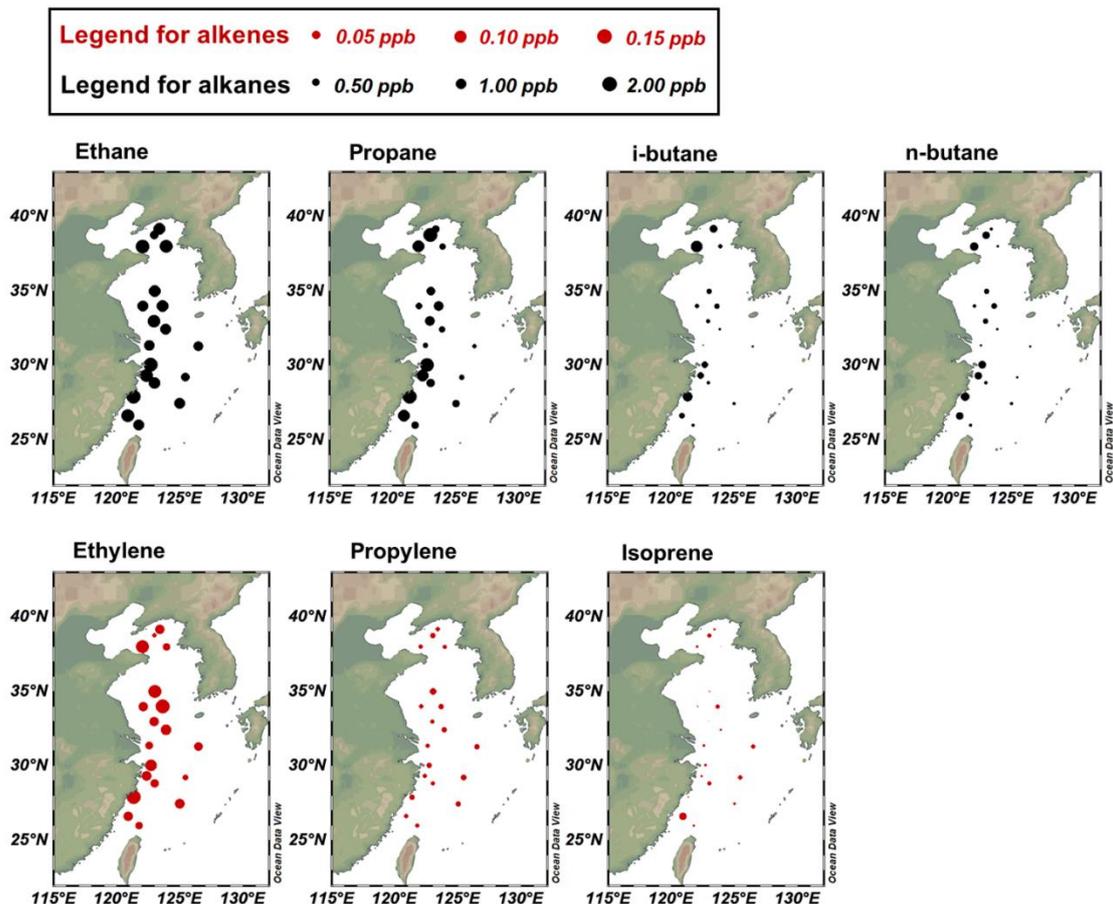
61

62 2. While the authors have made a clear statement regarding the significant difference in
63 NMHCs between urban and marine atmosphere, there is still a need for a more explicit
64 expression on the distribution characteristics of NMHCs, especially in the marine
65 atmosphere. It is advisable to make advancements in the description within the main
66 text, ideally supplemented with relevant Figure for a comprehensive view.

67 **Reply:** Thanks for your valuable suggestion. We have represented an explicit
68 description about the distribution of NMHCs in the marine atmosphere in the main text
69 and provided relevant Figures in supporting information, as indicated below:

70 Line 271-277:

71 “In spatial terms, multiple NMHCs (e.g. ethane, propane, i-butane, n-butane, and
72 ethylene) showed higher atmospheric concentrations in regions closer to the land. The
73 elevated concentrations are primarily concentrated along the coastal regions of the East
74 China Sea and the north Yellow Sea (Figure S3). The disparity in NMHCs
75 concentrations between land and ocean, as well as the distribution pattern of NMHCs
76 in the marine atmosphere, suggested the potential influence of terrestrial sources on the
77 oceanic NMHCs.”



78

79 **Figure S3** Distributions of alkanes (black dots) and alkenes (red dots) in the atmosphere over the
 80 **Yellow Sea and the East China Sea**

81 3. Obviously, the distance from the land to the oceanic station is a crucial parameter in
 82 the authors' discussion. However, I didn't find any information about the source of
 83 distance data or an introduction to the relevant calculation method. The authors should
 84 explicitly provide details on these aspects for clarity and transparency in the manuscript.

85 **Reply:** We are sorry for our negligence about the introduction of the calculation method
 86 for the distance from the land to the oceanic stations. It has been supplied in the
 87 manuscript, as indicated below:

88 **Line 196-202:**

89 **“2.7 Calculation of the shortest distance from the sampling station to the land**

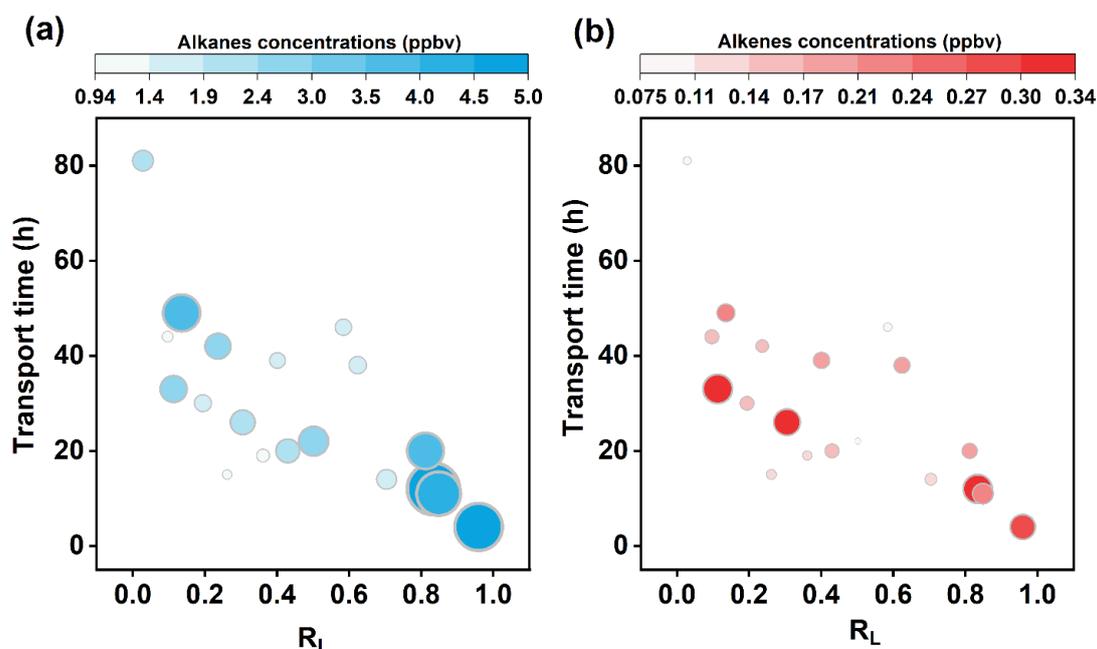
90 Coastline latitude and longitude data near the study area (20-45°N, 110-130°E) were
91 extracted from the World Vector Shorelines (downloaded from
92 <https://www.ngdc.noaa.gov/mgg/shorelines/data/gshhg/latest/>). Subsequently,
93 distances from the maritime sampling stations to all coastal locations were computed.
94 The minimum value among these distances was selected as the shortest distance to the
95 land (listed in Table S9).”

96 4. The authors have employed an innovative approach in assessing the impact of sea-
97 air flux on the marine NMHCs, by calculating the atmospheric lifetime-weighted
98 concentrations of different gases. In this way, the variabilities of atmospheric
99 reactivities of different gases were considered simultaneously when examining the
100 relationship between flux and concentration. It is found intriguing and seemed effective.
101 Furthermore, the authors could extend this novel idea to the discussion about the air
102 mass. Both of land retention and transport time of air masses serve as indicators of air
103 mass characteristics, reflecting the impact of land-based inputs. Combining these two
104 parameters to collectively explore the impact of air masses on NMHCs might offer a
105 new perspective, potentially leading to fresh discoveries

106 **Reply:** Thanks for your valuable suggestion. As combining the effects of R_L and
107 transport time of air mass on marine NMHCs, it emphasized the possibility of the
108 terrestrial influence on the marine atmospheric environment.

109 Line 343-347:

110 “Notably, elevated alkane concentrations were affected by those air masses with larger
111 R_{L-mean} (>0.8) and shorter transport time (<20 h) (Figure S4). This emphasized the
112 terrestrial influence on alkanes in the marine atmosphere, since both R_L and transport
113 time serve as indicators of air mass terrestrial characteristics.”



114
115 **Figure S4** Impacts of air mass (indicated by transport time and R_L) on atmospheric alkanes (a) and
116 alkenes (b) over the Yellow Sea and the East China Sea

117
118 **Minor comments:**

119 1. Show the standard deviation when you mentioned average, like line 266 “the mean
120 (range) concentration of ethane, propane, i-butane, and n-butane was 2.26 (0.277-5.72),
121 2.95 (0.149-20.1), 2.57 (BD-27.6), and.....”

122 **Reply:** Thanks for your valuable suggestion. It has been revised in the manuscript, as
123 indicated below:

124 Line 251-254:

125 “In urban atmosphere (n = 14), the mean (range) concentration of ethane, propane, i-
126 butane, and n-butane was 2.26 ± 1.66 (0.277-5.72), 2.95 ± 5.12 (0.149-20.1), $2.57 \pm$
127 6.99 (BD-27.6), and 3.29 ± 7.68 (0.018-30.2) ppb, respectively (Table 1).”

128 Line 257-260:

129 “For alkene species in the urban atmosphere (n = 14), the mean (range) of ethylene,
130 propylene, and isoprene was 0.180 ± 0.126 (0.035-0.390), 0.036 ± 0.040 (BD-0.129),
131 and 0.046 ± 0.072 (0.006-0.250) ppb, respectively.”

132 Line 262-266:

133 “In the marine atmosphere (n = 19), the mean (range) concentration of ethane, propane,
134 i-butane, n-butane, ethylene, propylene, and isoprene was 1.24 ± 0.298 (0.686-1.72),
135 0.822 ± 0.518 (0.226-1.79), 0.283 ± 0.302 (BD-1.17), 0.256 ± 0.214 (0.025-0.694),
136 0.151 ± 0.077 (0.028-0.295), 0.033 ± 0.009 (0.022-0.060), and 0.008 ± 0.010 (BD-
137 0.043) ppb, respectively.”

138 Line 366-370:

139 “The mean (range) of sea-to-air fluxes of ethane, propane, i-butane, n-butane, ethylene,
140 propylene, and isoprene was 44.6 ± 35.0 (0.2-118), 41.5 ± 39.9 (0.2-157), 31.7 ± 38.2
141 $(0.1-146)$, 10.9 ± 25.4 (-0.8-96.1), 321 ± 294 (1.7-775), 56.1 ± 55.2 (0.2-212), and 112
142 ± 134 (0.5-468) $\text{nmol m}^{-2} \text{d}^{-1}$, respectively, in the Yellow Sea and the East China Sea
143 (Table 1).”

144 2. line 358 “ethane possesses an atmospheric lifetime of approximately 78 d at 24 h
145 [\bullet OH] concentration of 6×10^5 molecules cm^{-3}”. References are needed to
146 illustrate the source and credibility of the data used here.

147 **Reply:** Thanks for your valuable suggestion. We have cited the relevant references as
148 indicated below:

149 “For instance, ethane possesses an atmospheric lifetime of approximately 78 d at 24 h
150 [\bullet OH] concentration of 6×10^5 molecules cm^{-3} (Jobson et al., 1999), using the rate
151 constant with \bullet OH at 288 K taken from Atkinson et al. (1997).”

152 3 line 468 “ozone formation potential (OFP) of NMHCs was calculated using $\text{OFP} =$
153 $\text{MIR} \times \text{C} \dots$ ” The calculation description should be in the method section and clearly
154 present the specific constants used in the equation and their literature sources.

155 **Reply:** Thanks for your valuable suggestion. The calculation description has been
156 composed in the method section 2.5 and the specific constants have been presented in
157 Table S11.

158 **“5 Calculation of ozone formation potential of NMHCs**

$$159 \text{OFP}_i = \text{MIR}_i \times C_i \quad (5)$$

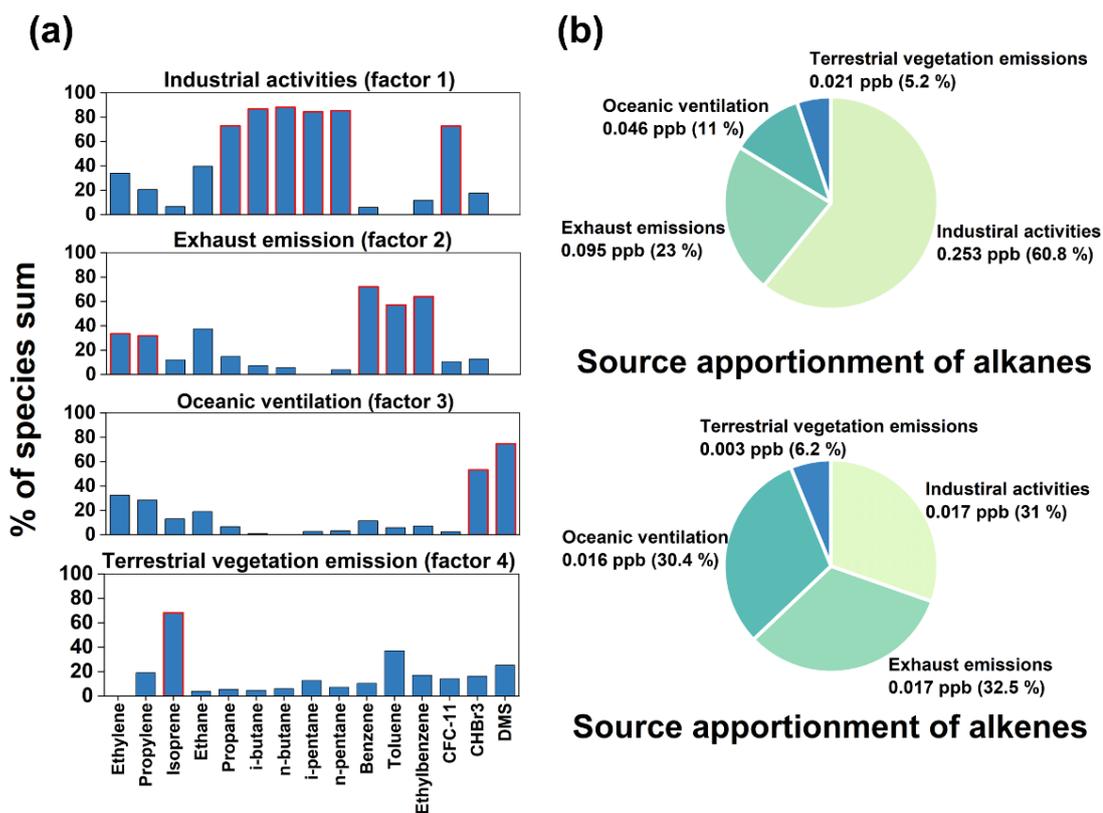
160 To assess the environmental implications of different sources, the ozone formation
161 potential (OFP) of NMHCs was calculated using Eq. (5), where MIR_i depicts the
162 maximum incremental reactivity and C_i represents the concentration of NMHCs
163 (Carter, 1994). Specific data was listed in supplementary Table S11.”

164 4 Is it necessary to include both the full term and abbreviation ‘NMHCs’ in the caption
 165 of each Figure? Generally, after the initial mention in the text, subsequent references
 166 can use the abbreviation alone for conciseness.

167 **Reply:** Thanks for your valuable suggestion. After the first appearance of the full name,
 168 we use abbreviations in the following text.

169 5 Increase the font size of the text in the Figures to make them more readable.

170 **Reply:** We feel sorry for the blurred figures due to the inappropriate font size. We have
 171 increased the font size of the text in the Figures, e.g. Figure 7.



172
 173 **Figure 7** Representative factor profiles from the PMF model (panel a) and relative contributions of
 174 different factors/sources to the alkanes and alkenes in the oceanic atmosphere (panel b). NMHCs in

175 panel a marked with red rim are selected as indicators for the specific factors.

