

1 Seasonal variations in composition and sources of atmospheric ultrafine particles 2 in urban Beijing based on near-continuous measurements

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17 **Abstract.** Understanding the composition and sources of atmospheric ultrafine particles (UFPs) is
18 essential in evaluating their exposure risks. It requires long-term measurements with high time
19 resolution, which are to date scarce. We performed near-continuous measurements of UFP composition
20 during four seasons in urban Beijing using a thermal desorption chemical ionization mass spectrometer,
21 accompanied by real-time size distribution measurements. We found that UFPs in urban Beijing are
22 dominated by organic components, varying seasonally from 68 to 81%. CHO organics (i.e., molecules
23 containing carbon, hydrogen, and oxygen) are the most abundant in summer, while sulfur-containing
24 organics, some nitrogen-containing organics, nitrate, and chloride are the most abundant in winter.
25 With the increase of particle diameter, the contribution of CHO organics decreases, while that of sulfur-
26 containing and nitrogen-containing organics, nitrate, and chloride increase. Source apportionment
27 analysis of the UFP organics indicates contributions from cooking and vehicle sources, photooxidation
28 sources enriched in CHO organics, and aqueous/heterogeneous sources enriched in nitrogen- and
29 sulfur-containing organics. The increased contributions of cooking, vehicle, and photooxidation
30 components are usually accompanied by simultaneous increases in UFP number concentrations related
31 to cooking emission, vehicle emission, and new particle formation, respectively. While the increased
32 contribution of the aqueous/heterogeneous composition is usually accompanied by the growth of UFP
33 mode diameters. The highest UFP number concentrations in winter are due to the strongest new particle
34 formation, the strongest local primary particle number emissions, and the slowest condensational
35 growth of UFPs to larger sizes. This study provides a comprehensive understanding of urban UFP
36 composition and sources and offers valuable datasets for the evaluation of UFP exposure risks.

37 **1. Introduction**

38 Ultrafine particles (UFPs, particles with diameters smaller than or equal to 100 nm) have significant
39 effects on human health (HEI, 2013; Ohlwein et al., 2019; WHO, 2013) and global climate (Kulmala
40 et al., 2004; Pierce and Adams, 2007). Their human exposure risks and climate effects are highly
41 related to their composition and size (Oberdorster et al., 2005; Pierce and Adams, 2007). To better
42 evaluate the exposure risks of UFPs and to formulate corresponding air quality guidelines, the World
43 Health Organization made several recommendations to guide measurements and regulations of UFPs
44 in 2021 (Organization, 2021). They emphasized that local understanding of UFP origins and their
45 chemical composition are scarce in most parts of the world.

46 Current field studies of atmospheric UFP composition and their source apportionment are mostly based
47 on offline sampling. These measurements usually use a size-resolving impactor to collect UFPs on
48 filters for tens of hours to several days (Cabada et al., 2004; Cass et al., 2000; Ham and Kleeman, 2011;
49 Herner et al., 2005; Kleeman et al., 2009; Massling et al., 2009; Xue et al., 2019; Xue et al., 2020a;
50 Xue et al., 2020b; Zhao et al., 2017). They found that organic carbon, sulfate, and nitrate could account
51 for 50-90% of the detected compounds, and the composition could vary greatly with UFP sizes due to
52 different sources and atmospheric evolutions (Cabada et al., 2004; Herner et al., 2005; Massling et al.,
53 2009). For source apportionment, the Chemical Mass Balance analysis was often used which relies
54 heavily on the source profiles and the limited numbers of molecular markers (Ham and Kleeman, 2011;
55 Kleeman et al., 2009; Xue et al., 2019). The identified sources include meat cooking, gasoline, diesel,
56 motor oil, and wood burning. However, these offline explorations can not capture the high temporal
57 variability in size-resolved UFP composition and sources, nor can they distinguish primary UFPs from
58 secondary sources, e.g., new particle formation, due to the lack of tracers and composition profiles for
59 secondary sources. Simultaneous real-time measurements of UFP composition and size distributions
60 are needed to accurately identify both primary and secondary sources.

61 Using a thermal desorption chemical ionization mass spectrometer (TDCIMS), the size-resolved
62 composition of UFPs can be measured with a resolution of tens of minutes (Li et al., 2021; Smith et
63 al., 2004). Previously, it has been used in a number of sites for short-term measurements and found
64 distinct characteristics for UFPs in urban (Li et al., 2022a; Li et al., 2021; Smith et al., 2008; Smith et
65 al., 2005), rural (Lawler et al., 2020; Smith et al., 2010), and remote areas (Glicker et al., 2019; Lawler
66 et al., 2018; Lawler et al., 2021; Lawler et al., 2014). For instance, UFPs in urban areas have more
67 nitrogen- and sulfur-containing organics (Smith et al., 2005; Winkler et al., 2012), while those at forest
68 sites have more CHO organics (i.e., molecules containing carbon, hydrogen, and oxygen) (Glicker et
69 al., 2019; Lawler et al., 2018). With these near-continuous measurements, unique sources such as
70 fungal bursts (Lawler et al., 2020) and sea-salt nanoparticles (Lawler et al., 2014) were identified and
71 the mechanisms of new particle formation were examined (Li et al., 2022a). Source apportionment
72 analysis was performed for the high time-resolution TDCIMS results in Amazon Basin to isolate

73 anthropogenic UFPs from background UFPs (Glicker et al., 2019). These analyses mainly focused on
74 short-term analysis covering several weeks. To address the UFP composition and sources from a more
75 comprehensive view, there is an urgent need for long-term and high-time-resolution measurements in
76 diverse environments.

77 The primary and secondary sources of particles in urban atmospheres usually show significant seasonal
78 characteristics. Thus, addressing the seasonal variations of UFPs, as well as their governing factors, is
79 fundamental to evaluating their long-term impacts. For fine particles in Beijing, coal combustion is
80 more abundant in winter due to domestic heating in the surrounding regions (Sun et al., 2015; Zhang
81 et al., 2013), biomass burning is more abundant in harvest seasons (Zhang et al., 2008), and dust storms
82 are more frequent in spring (Xu et al., 2020; Zhang et al., 2013). Besides these primary sources,
83 previous studies on larger particles showed higher oxidation states of organic aerosols in summer due
84 to stronger photochemical processes (Hu et al., 2017; Ma et al., 2022; Sun et al., 2018). However,
85 considering the short lifetime of UFPs, seasonal variations of the composition and sources of UFPs are
86 likely different. For example, as an important source of UFPs, new particle formations in Beijing were
87 observed to be the weakest in summer and strongest in winter due to temperature variations (Deng et
88 al., 2020; Li et al., 2020; Wu et al., 2007). Similar to Beijing, seasonal variations of UFP composition
89 based on high time-resolution measurements are also scarce for other atmospheric sites.

90 Here, we performed near-continuous measurements of UFP composition and size distributions over
91 four seasons in a typical megacity of Beijing with ~22 million people. The UFP composition, its size
92 dependence, and seasonal variability were analyzed. Several molecular markers from cooking and
93 vehicle emissions were identified. These markers were combined with the Positive Matrix
94 Factorization (PMF) analysis to address contributions from primary and secondary sources of UFPs.
95 The aerosol General Dynamic Equation (GDE) was used to quantify the emission rates of primary
96 UFPs and the formation rates of secondary UFPs. The driving factors for the seasonal variations of
97 UFP composition and number concentrations were identified.

98 **2. Methods**

99 **2.1 Field measurements.**

100 The sampling site is on the fifth floor of a building on the west campus of Beijing University of
101 Chemical Technology (39°94'N, 116°30'E) (Liu et al., 2020). The site is a typical urban site,
102 surrounded by residential and commercial buildings. Three trafficked roads are 130~565 m away from
103 the sites. UFP composition, particle number-size distribution, trace gases, and meteorological
104 conditions were measured over four seasons between Dec. 2019 and Aug. 2021. An overall of 149
105 days' TDCIMS measurements were used for analysis, with at least three weeks' data for each season.
106 Details of the sampling periods are described in [Table S1](#).

107 UFP composition was measured by the TDCIMS using the “bulk collection mode” (Li et al., 2021;

108 Smith et al., 2004). The TDCIMS collects pre-charged particles on a high voltage-biased Pt filament
109 and then vaporizes the particles for analysis by a chemical ionization high-resolution time-of-flight
110 mass spectrometer (CI-HTOF, Aerodyne Research Inc. and ToFwerk AG). The particle electrostatic
111 collection efficiency on the filament decreases rapidly with increasing particle size due to decreased
112 electrical mobility, ensuring that the collected particle mass is mainly from UFPs (Li et al., 2021).
113 During the particle evaporation, an electrical current is applied to the metal filament to an estimated
114 temperature of ~ 600 °C within a minute. The observed compounds can be desorbed thoroughly within
115 the heating periods as indicated by the desorption profile. Using O_2^- as the reagent ion, sulfate, nitrate,
116 chloride, and most of the oxygenated organics can be measured, while black carbon, hydrocarbon
117 compounds, and bases such as ammonium and aminium are less likely to be detected due to lower
118 sensitivity. Every sampling cycle is followed by a background cycle where no voltages are applied to
119 the Pt filament for particle collection. The signals from the background cycle are subtracted from the
120 sampling cycle to exclude minor influences from the gas phase compounds. Each analysis cycle
121 (including a sample and a background cycle) is set to be 10-40 min, depending on the estimated sample
122 mass. The detailed principles, operations, and quantifications of the TDCIMS are the same as the “bulk
123 collection mode” described in our previous study (Li et al., 2021).

124 The particle number size distributions from 1 nm to 10 μm were measured using a home-built particle
125 size distribution system (PSD, 3 nm–10 μm) and a diethylene glycol scanning mobility particle
126 spectrometer (DEG-SMPS, 1–7.5 nm). The configuration and operation of the PSD are the same as
127 described in our previous studies (Cai et al., 2017). The time resolution of the measurement is 5 min.
128 The number and mass concentrations of atmospheric UFPs were estimated via the integration of size
129 distribution measurements, assuming spherical particles with a density of 1.4 g cm⁻³.

130 Other parameters used in this study include the meteorological conditions measured by the
131 meteorology stations (AWS310, Vaisala Inc., Finland) and trace gases measured by the trace gas
132 analyzers (TGA, Thermo Fisher). The mixing layer height (MLH) was estimated from the vertical
133 profiles measured by a ceilometer (CL51, Vaisala Inc., Finland) and a three-step idealized-profile
134 method was used to estimate the MLH (Eresmaa et al., 2012).

135 **2.2 Source apportionment of UFP composition.**

136 The Igor-based interface SoFi (solution finder, version 6.5) and ME-2 (Canonaco et al., 2013) were
137 used for the PMF analysis to analyze the sources of organics in UFPs. The integrated thermal
138 desorption signals of organic peaks with m/z between 100 and 300 measured by the TDCIMS were
139 used as the input data matrix. The integrated thermal desorption signals from the background samples
140 were used to derive the input error matrix. The best solution in each season was chosen according to
141 Q/Q_{exp} , the similarities between m/z profiles, time series, and diurnal variations of the factors. The
142 correlations between each factor and the measured key species, trace gas, and $PM_{2.5}$ were calculated
143 for better identification of the factors. It should be noted that there were also many peaks with m/z

144 below 100, but a large fraction of them was from thermal decomposition, and their inclusion would
 145 add great complexity to the factor assignments. The signal intensity instead of the mass concentration
 146 was used because sensitivity quantification of the TDCIMS was based on the calibration of limited
 147 numbers of compounds which may induce unknown uncertainties when quantifying the sources. As a
 148 result, the signal intensity measured by the TDCIMS is reported for reference and the relative
 149 variations of detected species are studied rather than their estimated ambient concentrations.

150 2.3 Quantifying source and loss terms of UFP number concentrations.

151 In the measured size distribution plots, there are usually abrupt increases in UFP number concentration.
 152 During new particle formation (NPF) periods, the abrupt increases of UFPs are usually accompanied
 153 by a burst of sub-3 nm particles and usually start from noontime. During non-NPF periods, the abrupt
 154 increases of UFPs are usually accompanied by an increase in primary emission tracers (as will be
 155 shown in Section 3.2). We apply the GDE to quantify the new particle formation rates (J) and primary
 156 particle number emission rates (E) at the observation site. The calculation of J follows those described
 157 in previous studies (Cai and Jiang, 2017; Cai et al., 2017). The calculation of $E_{[i,j]}$ ($\text{m}^{-3} \text{s}^{-1}$), the particle
 158 emission rates in the size range of $[d_i, d_j]$, follows Eq. 1 (Cai et al., 2018; Kontkanen et al., 2020)
 159 during non-NPF periods.

$$160 \quad E_{[i,j]} = \frac{dN_{[i,j]}}{dt} + GR(n_j - n_i) + CoagSnk_{[i,j]} - CoagSrc_{[i,j]} - TR_{[i,j]} \quad (\text{Eq. 1})$$

161 Where $\frac{dN_{[i,j]}}{dt}$ ($\text{m}^{-3} \text{s}^{-1}$) is the variation of the particle number concentration in the size range of $[d_i,$
 162 $d_j]$ during the period of dt (s^{-1}); $GR(n_j - n_i)$ ($\text{m}^{-3} \text{s}^{-1}$) is the net condensation growth term, GR ($\text{m} \text{s}^{-1}$) is
 163 the condensational growth rate of particle d_i , and n_i (m^{-4}) is the particle number size distribution
 164 function for particle d_i ; $CoagSrc_{[i,j]}$ and $CoagSnk_{[i,j]}$ ($\text{m}^{-3} \text{s}^{-1}$) are the coagulation source and sink terms;
 165 $TR_{[i,j]}$ is the transport term. Overall, 33 size bins were included in the size range of 3-50 nm.

166 The term $\frac{dN_{[i,j]}}{dt}$ and $CoagSnk_{[i,j]}$ can be directly calculated from the size distribution data (Cai et al.,
 167 2018). GR is calculated by the theoretical condensation of the condensable vapors, that is the sum of
 168 H_2SO_4 and condensable organic vapor concentrations. Here, we regard condensable organic vapors as
 169 oxygenated organic molecules (OOMs) with saturation vapor pressure lower than $0.3 \mu\text{g m}^{-3}$ as in our
 170 previous studies (Li et al., 2022a; Qiao et al., 2021). Since not all the observation days were equipped
 171 with the measurements of condensable vapors, we adopted seasonal-dependent GR derived from
 172 seasonal average condensable vapor concentrations reported in our previous study, that is 1.2×10^7 ,
 173 9.9×10^7 , 1.2×10^8 , and $5.0 \times 10^7 \text{ cm}^{-3}$ for winter, spring, summer, and autumn (Qiao et al., 2021),
 174 respectively. For particles smaller than 50 nm, $CoagSrc_{[i,j]}$ term can be neglected; for particles smaller
 175 than 5 nm, the uncertainties will be very large for E . We only calculated E_{3-50} in this study and

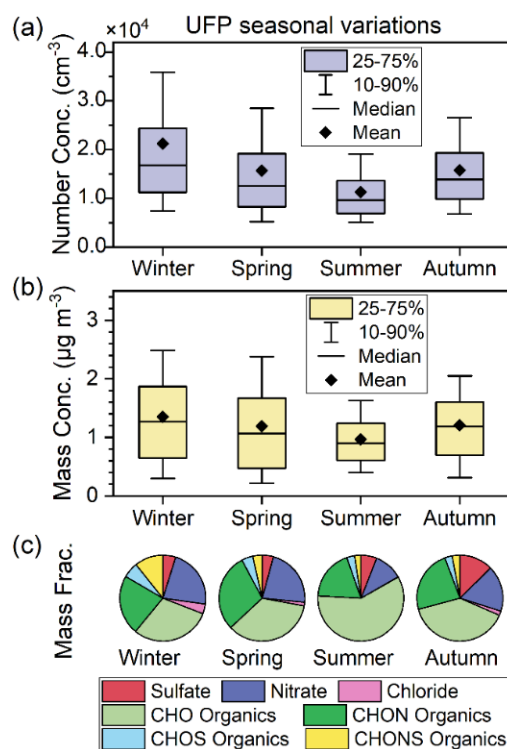
176 $CoagSrc_{[i,j]}$ was thus neglected. Generally, $TR_{[i,j]}$ term cannot be quantified using the mathematic
177 method. As our previous study has indicated there was no significant transport term on a long-term
178 time scale (Kontkanen et al., 2020), we initially assumed that $TR_{[i,j]}$ equals 0. Another assumption is
179 that the influences from the variation in MLH are neglected. We briefly explore how the assumptions
180 of $TR_{[i,j]}$ and MLH influence the results in the next paragraph.

181 On the particle size distribution plots, we notice that the abrupt appearance of particles during non-
182 NPF days usually happens in the early morning (6:00-9:00) and late afternoon (17:00-20:00) (Figure
183 S1). The afternoon peak is accompanied by a decrease in MLH and an increase in particle numbers in
184 all sizes within the range of 3-50 nm, thus the abrupt increase in particle number concentration could
185 be due to the combined effects of MLH, transport, and emission. The morning peak is accompanied
186 by the increase in MLH, which should decrease particle number concentrations, and the increase in
187 particle number is only observed for 3-30 nm particles but not for 30-50 nm. Thus, the increasing
188 morning peak could only be caused by the primary emission of 3-30 nm particles. As a result, the 3-
189 30 nm particle emission rate during 6:00-9:00 is calculated to represent the average primary particle
190 emission rates for each day. It should be noted that the emission rates during 6:00-9:00 may be
191 underestimated due to the increase of MLH, and the emission rates only represent the increasing rates
192 of primary particles at the observational site, not the direct emission rates from the sources.

193 **3. Results and discussion**

194 **3.1 UFP concentration, composition, and seasonal variability**

195 The overall concentration of UFPs is the highest in winter and the lowest in summer. The UFP number
196 concentrations expressed in mean \pm standard deviation are $(1.7\pm 1.2)\times 10^4$, $(1.5\pm 1.1)\times 10^4$,
197 $(1.1\pm 0.7)\times 10^4$, and $(1.5\pm 0.9)\times 10^4$ cm^{-3} (Figure 1a) and the UFP mass concentrations are 1.3 ± 0.9 ,
198 1.2 ± 0.9 , 1.0 ± 0.6 , and 1.2 ± 0.7 $\mu\text{g m}^{-3}$ (Figure 1b) for winter, spring, summer, and autumn, respectively.
199 The seasonal variations are partly caused by the variation in MLH (Figure S2), while the other driving
200 factors are related to the source and loss terms of UFPs and will be further discussed in Section 3.3.



201

202 **Figure 1.** Seasonal variations of UFP concentrations and composition in urban Beijing. (a) UFP
 203 number concentrations in the size range of 3-100 nm. (b) UFP mass concentrations integrated from
 204 size distribution measurements, assuming spherical particles with a density of 1.4 g cm⁻³. (c) Mass
 205 fractions of the components measured by the TDCIMS in negative ion mode.

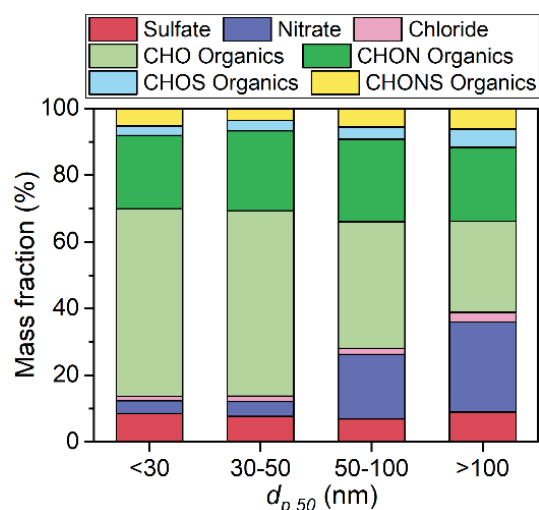
206 The detected UFP composition is dominated by organics (68-81% for mass fraction), with minor
 207 contributions from nitrate (11-22%), sulfate (4-13%), and chloride (0.1-4%) over all four seasons
 208 (Figure 1c). The organic species include CHO, CHON, CHOS, and CHONS organics, contributing 30-
 209 59%, 19-29%, 3-6%, and 2-11% mass concentrations of the detected UFP compounds, respectively.
 210 The detected particulate species are similar for all four seasons as indicated by the similarities in the
 211 mass defect plots (Figure S3). The measured composition is consistent with the offline results from
 212 Beijing, which showed that organics were the most abundant in UFPs (Massling et al., 2009; Zhao et
 213 al., 2017). It should be noted that the collected mass integrated from the TDCIMS signals is ~50% of
 214 the total collected mass estimated from the size distributions (Figure S4). This is possibly due to the
 215 uncertainties in the quantification methods or because some UFP compounds (e.g., ammonia, amines,
 216 black carbon, and alkanes) cannot be ionized by O₂⁻ in the TDCIMS. However, as the mass estimated
 217 from the two methods are in good correlation, we assume that the TDCIMS-measured composition is
 218 representative of UFP composition. As some of the particulate CHON, CHOS, and CHONS organics
 219 would decompose to CHO fragments in the TDCIMS during the thermal desorption process, there may
 220 be some underestimation of CHON and CHOS/CHONS organics and overestimation of CHO organics.

221 A major seasonal difference in UFP chemical composition is that the highest fractions of slow-

222 desorbed CHO organics are observed in summer (59%), which may be related to the strongest solar
223 radiation and lowest NO_x concentrations. On the one hand, the appearance of most CHO organic ions
224 during temperature ramping of the Pt wire occurs at higher temperatures compared to nitrate and
225 chloride, while at slightly lower temperatures compared to sulfate (Figure S5). The higher temperature
226 desorption, which we refer to as “slowly-desorbed,” indicates that these species must be low-volatility
227 compounds or the corresponding thermal decomposition fragments. On the other hand, the overall
228 CHO organic mass has an afternoon peak at ~14:00, and its diurnal variation is consistent with that for
229 O₃ in summer (Figure S6), indicating they might be related to photooxidation chemistry. Based on
230 these, we hypothesize that CHO organics in the UFPs are mostly from the partitioning of low-volatility
231 compounds originating from the gas-phase oxidation. Thus, the higher CHO fractions in summer are
232 due to the strongest solar radiation, which benefits the gaseous photooxidation, and the lowest NO_x
233 (Figure S2), which contributes to the formation of CHO organics over CHON organics (Yan et al.,
234 2020; Ye et al., 2019).

235 Another seasonal difference is that higher fractions of fast-desorbed species are measured in winter,
236 including nitrate, chloride, and some CHON (e.g., C₆H₄NO₃⁻, nitrophenols) organic compounds. These
237 species are all desorbed at lower temperatures (Figure S5) and their concentrations in UFPs are
238 negatively correlated to ambient temperature (Figure S7), indicating their relatively higher volatility.
239 Thus, the higher fractions in winter are mainly governed by the temperature-dependent partitioning of
240 these compounds. It should be noted that CHONS organic (e.g., deprotonated aminomethanesulfonic
241 acid CH₄NSO₃⁻ and deprotonated taurine C₂H₆NSO₃⁻) fractions also increase in winter. Previously,
242 CH₄NSO₃⁻ and C₂H₆NSO₃⁻ were reported to be formed in the gas phase through the reaction between
243 SO₃ and amines under dry conditions (Li et al., 2018; Sarkar et al., 2019). Their gaseous formation are
244 likely happen in winter Beijing due to the low ambient relative humidity. The seasonal variations of
245 these CHONS species are different from those in larger particles where S-containing organics are
246 mainly organosulfates from primary emissions or heterogeneous/aqueous reactions (Ma et al., 2022).

247 The composition of UFPs also varies greatly with particle size. As shown in Figure 2, the most
248 significant size-dependent variations were observed for nitrate and CHO organics. The nitrate fraction
249 increases significantly with increasing particle diameter, probably due to the Kelvin effects that prevent
250 it from partitioning to small particles, or due to an increase in aqueous/heterogeneous processes at
251 larger particle sizes. The CHO organic fraction decreases significantly with increased particle size,
252 possibly due to its low volatility that favors smaller particles compared to the high-volatility
253 compounds. Compared to CHO organics, the relative contributions of N- and S-containing organics
254 increase with particle size, possibly due to higher volatility or the aqueous/heterogeneous formation as
255 particles grow. The sulfate fraction does not change significantly with particle sizes, possibly due to
256 the opposite size-dependence of condensational growth of H₂SO₄ and the aqueous/heterogeneous
257 formation of sulfate.

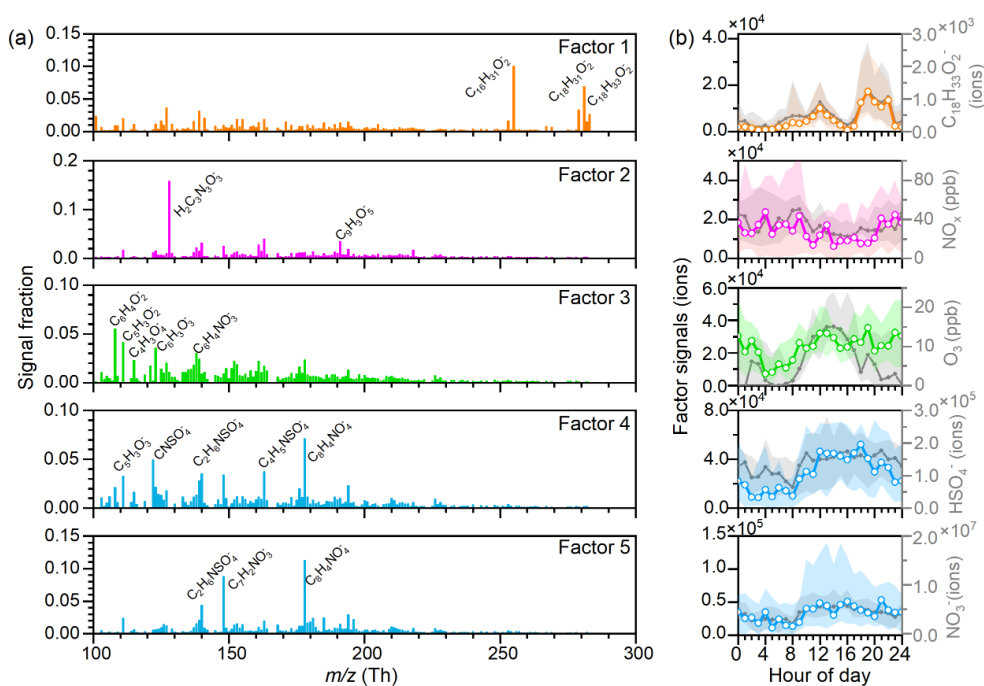


258

259 **Figure 2.** Size-dependent composition of UFPs. UFP composition mass fraction variation with the
 260 representing particle size $d_{p,50}$. $d_{p,50}$ corresponds to 50% volume mean diameter of particles collected
 261 on the TDCIMS filament.

262 3.2 Sources of UFP organics and their seasonal variabilities

263 As organics are the main components of UFPs, PMF source apportionments were performed for the
 264 organic compounds. Five factors were identified in each season. The factor profiles and their diurnal
 265 variations in winter are shown in Figure 3, and the results in other seasons are shown in Figure S8-10.
 266 The correlations between PMF factors and key UFP components, trace gases, meteorology parameters,
 267 and $PM_{2.5}$ for the four seasons are shown in Figure S11.



268

269 **Figure 3.** Source apportionment of the UFP organic composition (m/z 100-300) measured by the
 270 TDCIMS in winter. (a) m/z profiles of the five PMF factors; (b) diurnal variations of each factor and

271 their related terms.

272 Factor 1 and factor 2 are identified as cooking-related and vehicle-related sources, respectively. Factor
273 1 is enriched in $C_{16}H_{31}O_2^-$, $C_{18}H_{31}O_2^-$, and $C_{18}H_{33}O_2^-$ (assigned to deprotonated palmitic acid, linoleic
274 acid, and oleic acids, respectively). Previous studies have revealed that saturated and unsaturated fatty
275 acids are the major constituents in cooking emissions, accounting for 73-85% of the cooking organic
276 matter, among which palmitic acid and oleic acid can be treated as the unique fingerprints of
277 atmospheric cooking particles (Zhao et al., 2007b, a). Factor 1 and the tracers show clear morning,
278 noon, and evening peaks, corresponding to breakfast, lunch, and dinner times. Factor 2 is enriched in
279 $C_3N_3O_3H_2^-$ (assigned to deprotonated cyanuric acid). Cyanuric acid was previously found with the
280 biggest emission in the urea-based selective catalytic reduction (SCR) technology for the reduction of
281 NO_x from the exhaust of diesel-powered vehicles (Yassine et al., 2012). Factor 2 and the tracer show
282 clear morning peaks corresponding to the morning rush hours, consistent with the diurnal variation of
283 NO_x . These two factors are also identified in the other three seasons (Figure S8-10).

284 Besides these two primary sources, trace amounts of a biomass-burning tracer $C_6H_9O_5^-$ (assigned to
285 deprotonated levoglucosan) were also observed. However, its contribution to the total signal is small
286 and could not be separated into individual factors in the PMF analysis. We thus conclude that the
287 contribution of biomass burning to UFPs might be small in urban Beijing. This is understandable since
288 the burning of high-polluting fuels has been phased out in urban Beijing by the People's Government
289 of Beijing Municipality since 2014 (Municipality, 2014). Although large particles in urban Beijing
290 could be influenced by biomass burning and coal combustion through transport from surrounding
291 regions (Li et al., 2022b; Sun et al., 2015; Zhang et al., 2013; Zhang et al., 2008), UFPs could hardly
292 survive after long-distance transport due to their short lifetime.

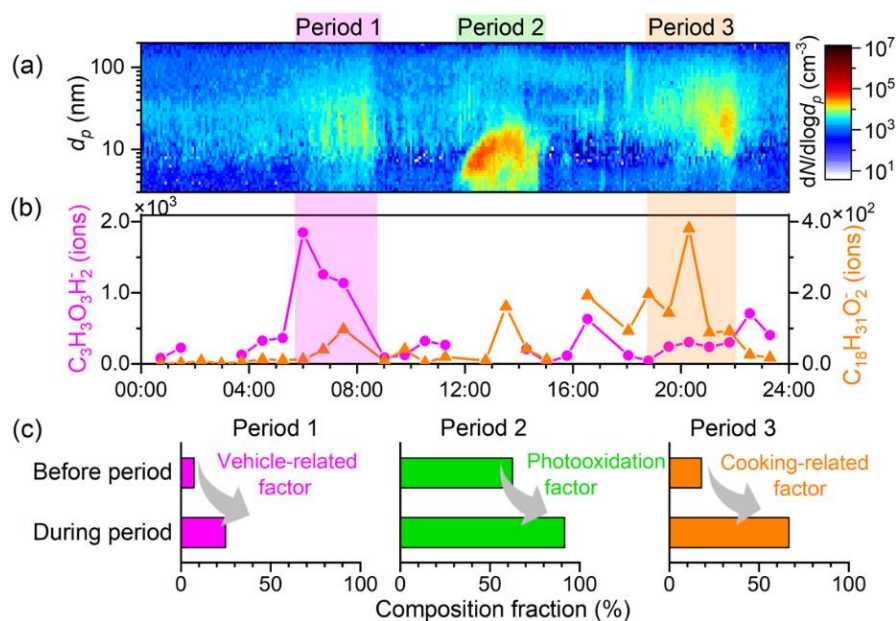
293 Factors 3-5 are identified as secondary sources related to photooxidation formation or
294 aqueous/heterogeneous formation. In winter (Figure 3), factor 3 is enriched in slowly-desorbed, low-
295 volatility CHO organics and has daytime peaks at ~12:00-18:00, which is consistent with the diurnal
296 variation of O_3 . They should come from gas-phase photooxidation followed by gas-particle
297 partitioning. Factor 4 and factor 5 are enriched in N- or S-containing organics. Their time series and
298 diurnal variability are highly correlated with sulfate, nitrate, $PM_{2.5}$, and relative humidity, indicating
299 the aqueous/heterogeneous formation pathway. Similarly, in spring and autumn, factor 3 is identified
300 as a photooxidation factor that is enriched in CHO organics, and factors 4-5 are identified as
301 aqueous/heterogeneous factors that are enriched in N- or S-containing organics (Figure S8-10).
302 Differently, in summer, factor 4 is identified as a photooxidation factor.

303 Clear seasonal variability of the sources was observed, with the contribution of primary emission
304 factors and aqueous/heterogeneous factors higher in winter and autumn, and the contribution of
305 photooxidation factors higher in summer (Figure S11). The sum of cooking and vehicle sources
306 contributed to 10-35% of the total organic signals in the m/z range of 100-300. The fractions of these

307 primary emissions are higher in winter and autumn, possibly indicating higher emissions. Another
308 possibility is that the oxidation degradation of these primary emissions is faster in summer and spring
309 due to higher oxidants and ambient temperature. The contributions of photooxidation factors are 20-
310 70% to the total organic signals in the m/z range of 100-300, with the highest in summer, and lowest
311 in winter. This is consistent with the highest CHO organic fractions in UFPs in summer in [Figure 1](#)
312 and is attributed to the strongest solar radiation. The contributions of aqueous/heterogeneous sources
313 are 15-60% to the total organic signals in the m/z range of 100-300, with the highest in winter and
314 lowest in summer.

315 To identify the sources for UFP numbers, we further combined the source analysis with variation in
316 particle size distributions. Among the identified four classes of composition sources, some are related
317 to the increase of UFP number concentrations, while others are related to the increase of UFP diameters.

318 The increase in UFP numbers is usually accompanied by the enhanced contribution of cooking- or
319 vehicle-related components or new particle formation events. An example is shown in [Figure 4](#), a
320 relatively clean day with little interference from background aerosols. There are three periods where
321 UFP bursts were observed. During period 1, i.e., 6:00-9:00, a mode with a peak diameter at ~20 nm
322 appeared with a rapid increase in the vehicle tracer, $C_3N_3O_3H_2^-$. Compared to that before period 1, the
323 contribution of the vehicle-related factor increased from 7% to 25%. During period 2, i.e., 12:00-15:00,
324 new particle formation happens with a burst of particles at a peak diameter of 5-10 nm. Compared to
325 that before period 2, the contribution of the photooxidation-related factor increased from 64% to 92%.
326 This is consistent with our previous studies that slowly-desorbed CHO organics were the most
327 abundant compounds during NPF periods (Li et al., 2022a). During period 3, i.e., 18:00-22:00, a mode
328 with peak diameter at ~30 nm bursts, with a rapid increase in the cooking tracer, $C_{18}H_{31}O_2^-$. $C_{18}H_{31}O_2^-$
329 also has two minor peaks in the morning and noon time, consistent with the cooking activities.
330 Compared to that before period 3, the contribution of the cooking-related factor increased from 18%
331 to 67%. Thus, we conclude that the increase in UFP numbers in the three periods is mainly attributed
332 to the increase in vehicle emissions, new particle formations, and cooking emissions, respectively.



333

334 **Figure 4.** The particle number size distributions (a) and UFP composition variability (b-c) on April
 335 16th, 2020. The three periods with the abrupt appearance of UFP particles on this day are identified as
 336 vehicle-related (period 1), NPF-related (period 2), and cooking-related (period 3) according to
 337 TDCIMS composition measurements.

338 The morning, noon, and evening peaks in UFP numbers were widely observed during the observation
 339 days in all four seasons. During non-NPF days, the UFP number concentration peak mainly appears in
 340 the morning and evening time, corresponding to the primary emissions, and we choose the morning
 341 periods to calculate the daily-averaged E . During NPF days, the UFP number peak mainly appears in
 342 the noon time, and the daily-averaged J was calculated during these periods. These further indicate
 343 that cooking emissions, vehicle emissions, and new particle formation are the main sources of UFP
 344 number concentrations.

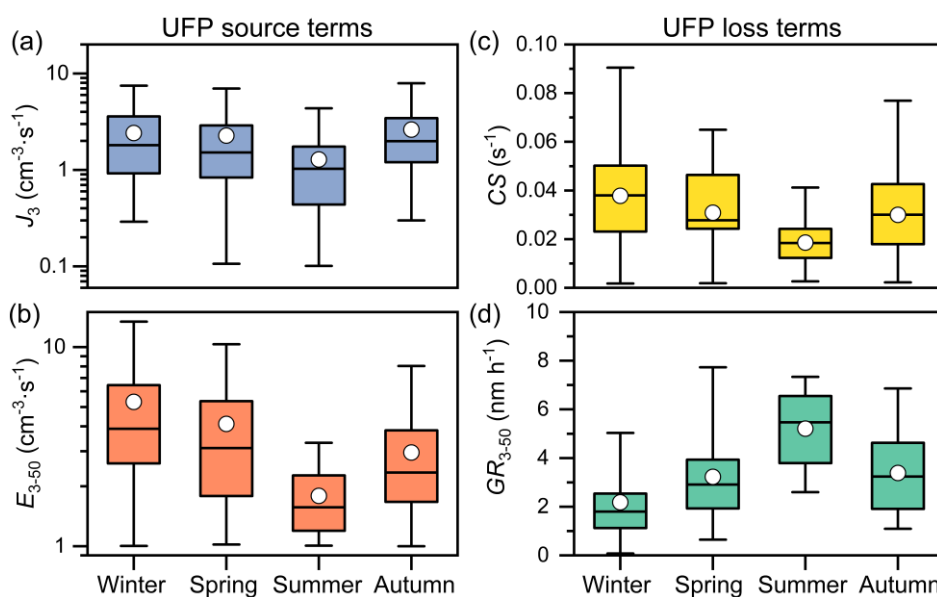
345 Different from these three factors, the increased contribution of the aqueous/heterogeneous factor is
 346 not accompanied by the increase of UFP number concentrations but by the increase of UFP mode
 347 diameters. The contribution of aqueous/heterogeneous factor to sub-50 nm particles is only ~20%. For
 348 example, in the day presented in Figure 4, aqueous/heterogeneous factor accounted less than ~10% for
 349 the three bursts of UFP number concentration. However, it starts to dominate the organic composition
 350 when UFP particles grow above 50 nm (Figure S12), indicating an important role of
 351 aqueous/heterogeneous processes in the growth of particles larger than 50 nm in diameter.

352 3.3 Driving factors for the seasonal variability of UFPs.

353 As we have identified the main sources for UFP number concentrations in Section 3.2, we can further
 354 address the reasons for the significant seasonal differences in UFP number concentrations as has been
 355 shown in Figure 1, according to the variations in their sources and losses. The source terms mainly

356 include new particle formation rates (here represented by J_3) and primary emission rates (here
 357 represented by E_{3-50}); the loss terms are presented by condensational growth rates (GR_{3-50}) and
 358 coagulation sinks. Here, we apply the condensation sink (CS) to evaluate the strength of coagulation
 359 loss.

360 The main sources of the UFP number concentration, J_3 , and E_{3-50} , are both higher in winter and lower
 361 in summer (Figure 5a-b), which are presumably caused by temperature effects. The temperature effect
 362 on J_3 is mainly due to the temperature-dependent cluster evaporation rates as reported in our previous
 363 study (Deng et al., 2020). This seasonal dependence of atmospheric UFPs attributed to vehicle
 364 emissions and its underlining reasons have not been revealed before. On the one hand, the low ambient
 365 temperature will largely increase the vehicle emission factors for particle numbers and gaseous
 366 hydrocarbons (Suarez-Bertoa and Astorga, 2018; Wen et al., 2021). On the other hand, a large fraction
 367 of the nanoparticles from vehicle emissions has been proposed to be formed by nucleation of the
 368 emitted hydrocarbon vapors or their oxidation products (Rönkkö and Timonen, 2019). The high
 369 ambient temperatures in summer may suppress the formation of these vehicle-related particles, just
 370 like it suppresses J_3 during ambient NPF.



371

372 **Figure 5.** Seasonal variations of the main source and loss terms of UFP number concentration. (a) New
 373 particle formation rates for 3 nm particles (J_3); (b) daily average primary particle emission rates for 3-
 374 50 nm particles (E_{3-50}) during 6:00-9:00; (c) condensation sink (CS); (d) growth rates for 3-50 nm
 375 particles (GR_{3-50}) during new particle formation events using the mode fitting method.

376 The lowest GR of UFP occurs in winter (Figure 5d), which further contributes to high wintertime UFP
 377 number concentrations. CS and GR_{3-50} have opposite trends, with CS being the highest in winter while
 378 GR_{3-50} being the highest in summer (Figure 5c-d). The highest GR_{3-50} in summer is due to the highest
 379 condensable vapor concentrations in summer caused by strong solar radiation and high temperature

380 favoring the formation of condensable OOMs (Li et al., 2022a; Qiao et al., 2021). The theoretical
381 condensational GR by OOMs and H_2SO_4 for 20 nm particles are 1.1, 3.0, 4.0, and 1.8 $nm \cdot h^{-1}$ in winter,
382 spring, summer, and autumn, respectively, and they are close to GR_{3-50} derived during NPF events
383 using the mode-fitting method as shown in Figure 5d. Under the observed average GR_{3-50} , the time
384 needed for sub-3 nm to grow above 50 nm is the shortest in summer (~ 9 h) and the longest in winter
385 (~ 24 h). Thus, the lower GR in winter also contributes to the highest UFP number concentrations in
386 winter.

387 4. Conclusions

388 In this study, we explored the UFP composition and sources in typical polluted urban environments
389 based on near-continuous measurements of UFP composition and size distributions in Beijing over
390 four seasons. We observed that UFP composition varied with seasons and particle diameter, indicating
391 their different sources. Specifically, photooxidation processes generate more CHO organics, leading
392 to higher CHO fractions in summer. While aqueous/heterogeneous processes generate more N- and S-
393 containing organics, leading to higher N- and S-containing organic fractions in above-50 nm particles
394 than sub-50 nm particles. Combining the PMF analysis for UFP organics and the size distribution
395 analysis, we found that vehicle and cooking emissions are two of the most important primary sources
396 of UFP number concentrations in urban Beijing, while new particle formation is the most important
397 secondary source of UFP number concentrations and would increase the contribution of CHO organics
398 to UFP composition. The aqueous/heterogeneous sources would not increase UFP number
399 concentration but would increase UFP mode diameters and mass concentrations. For the seasonal
400 variations, we found that UFP number concentrations are the highest in winter. This is mainly due to
401 the highest primary particle emissions, the highest new particle formation rates, and the lowest particle
402 growth rates in winter. Further controlling of UFPs in urban Beijing needs to focus on vehicle
403 emissions, and the gas precursors related to secondary sources of UFPs.

404 The observed distinct seasonal variabilities of UFP composition and their size dependence emphasize
405 the importance of long-term and high-time-resolution measurements of both UFP composition and size
406 distributions. This could provide valuable datasets for the evaluation of UFP's long-term exposure risks.
407 The high time-resolution measurements combined with PMF analysis can also help identify the
408 secondary UFP sources, which contribute the major fraction of organic signals but could not be
409 identified from previous offline UFP measurements. Further addressing the UFP composition and
410 sources on the regional scale still requires measurements at sites with different distances from the
411 emission sources due to the short lifetime of UFPs.

412

413 **Data availability.**

414 Data are available upon request from the corresponding authors.

415 **Supplement.**

416 The contents of the supporting information include the diurnal variations of E_{3-50} during non-NPF days
417 over four seasons (Figure S1); the diurnal variations of MLH, UVB, T, RH, O_3 , NO_x , and $PM_{2.5}$ in the
418 four seasons (Figure S2); details of the measured UFP composition during four seasons (Figure S3);
419 seasonal variation of the UFP mass estimated from TDCIMS and PSD (Figure S4); the averaged
420 thermal desorption profiles of different UFP composition (Figure S5); the diurnal variation of CHO
421 organics in the four seasons (Figure S6); temperature dependence of some fast-desorbed UFP
422 composition (Figure S7); the spectra of five PMF-factors during spring, summer, and autumn (Figure
423 S8-10); summary of PMF factors during the four seasons (Figure S11); the contribution of different
424 factors as a function of particle sizes (Figure S12); summary of sampling periods (Table S1).

425 **Author contributions.**

426 XL, JJ, and JS designed the study. XL, YC, YYL, RC, YRL, CD, and JW participated in data collection
427 and performed the data analysis. XL prepared the manuscript with contributions from all co-authors.
428 All authors approved the final version of the manuscript.

429 **Competing interests.**

430 The authors declare that they have no conflict of interest.

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435

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